

Anonymous Referee #2

The effects of aerosol chemical composition on the relative humidity dependence of light scattering are presented for a site in Beijing. Parameterizations of $f(\text{RH})$ are developed for different observed conditions (e.g., very clean, moderately polluted, polluted based on measured light scattering levels). The paper is very well written and the figures (with one exception) clearly convey the results of the study. I only have minor comments – see below.

Line 39: change to “that REDUCES the amount”. Also, please add a brief description of how SO₂ control reduces the amount of sulfate.

Reply: Revised. L. Zhang et al. (2015) studied the relationship between the scattering enhancement factor and chemical composition in Lin’an, China, finding that nitrate has a stronger effect on aerosol hygroscopicity than sulfate has, partially due to the rigid control of SO₂ that reduces the amount of sulfate and increases the content of nitrite (Morgan et al., 2010). Apart from sea salt emissions and gypsum dust emissions during construction containing sulfate, sulfate is mainly formed by the oxidation of its gaseous precursor, SO₂, in the atmosphere. In recent years, SO₂ emissions have been reduced substantially through a series of effective measures taken in China, like controlling the burning of loose coal and desulfurizing industrial equipment (Q. Zhang et al., 2019). Reducing SO₂ in the atmosphere thus directly affects the reduction in the sulfate content of aerosols. The saturated vapor pressure of nitric acid (HNO₃) is higher than that of sulfuric acid (H₂SO₄), so the availability of ammonia (NH₃) is key to the partitioning of HNO₃. HNO₃ is often neutralized by NH₃ after H₂SO₄. Therefore, a reduction in SO₂ means that more NH₃ can be used to neutralize HNO₃, leading to higher nitrate concentrations, such as ammonium nitrate (NH₄NO₃), in aerosols (Monks et al., 2019).

[Lines 36-46]

Lines 109 – 111: Why is the absorption coefficient at 880 nm transformed into those at 525 nm? Doesn't the 7-wavelength aethalometer have a measurement wavelength near to 525 nm?

Reply: The absorption coefficient at 520 nm measured by the 7-wavelength aethalometer is more appropriate than using the absorption coefficient at 880 nm. So we chose absorption at 520 nm then converted it to 525 nm. The single-scattering albedo ($\omega_{0(525\text{nm})}$) was also recalculated. Also updated were Figs. 2b, 2c, 3a and Table 1.

[Lines 130-132]

Lines 163 – 164: It is stated that “the proportion of organic matter and BC with weak hygroscopic abilities was low” from the southeast sector. Figure 3d indicates that mass fractions of BC were high in the southeast sector which seems to contradict this statement. Please clarify in the text.

Reply: The text has been revised as: “Figure 3c reveals that strongly hygroscopic aerosols with high values of $f(\text{RH} = 85\%, 525 \text{ nm})$ primarily came from the southeast sector. The proportion of secondary inorganics with strong hygroscopic abilities in aerosols from this direction was high, while the proportion of organic matter with weak hygroscopic abilities was low (Figs. 3e-f). Figure 3d indicates that the mass fraction of BC with weak hygroscopicity was slightly low in the southeast sector when wind speeds were lower than 4 m s⁻¹. However, when wind speeds were higher than 4 m s⁻¹, the mass fraction of BC was relatively high in this direction. Of all data associated with

southeast winds, identified were only three cases with wind speeds higher than 4 m s⁻¹, likely winds of short duration so not representative.”

[Lines 180-186]

Figure 6: The inset figures showing organic mass fraction vs. f(RH) are difficult to read because of their size – especially if a reader is looking at a print version of the paper. I recommend putting the insets into a separate figure.

Reply: We have put the inset figures into a separate figure: Figure S6.

Lines 258 – 260: Please report the mass fractions of organics, SO₄, and NO₃ if they were provided in Malm et al. (2003), Pan et al. (2009), Quinn et al. (2005), and Yan et al. (2009). It is difficult to assess differences in the role of NO₃ versus SO₄ in determining f(RH) in these different regions without knowing the chemical composition reported in these previously published papers.

Reply: The proportions of organics, SO₄²⁻, and NO₃⁻ provided in Malm et al. (2003, 2005), Yan et al. (2009), Pan et al. (2009), and this study are listed in Table R1.

Table R1: Proportions of sulfate, nitrate, and organic matter reported in previous studies and this study.

Site	Ammoniated sulfate or SO ₄ ²⁻ /FM [%]	NH ₄ NO ₃ or NO ₃ ⁻ /FM [%]	OMC/FM [%]	Reference
BBNP	51.0		21.0	Malm et al. (2003, 2005)
GC	31	4.8	44	
GSM	63	0.8	25	
BJ-1	28.0	16.0	34.0	Yan et al. (2009)
BJ-2	23.0	12.0	36.0	Sun et al. (2004)
BJ-3	23.0	14.0	29.0	
XA#1	15.3	7.6	29.8	Pan et al. (2009)
XA#2	17.7	9.4	24.4	
XA#3	11.2	5.2	37.4	
XA#4	5.3	1.5	40.0	
XA#5	9.5	5.7	26.0	
XA#6	6.1	1.2	36.7	
XA#7	8.9	1.1	39.4	
XA#8	10.8	3.2	43.0	
BJ-CMA	19.0	21.0	39.0	This study

Noet: BBNP was the observation site at the Big Bend National Park, Texas.

GC was the site at the Grand Canyon.

GSM was the site at the Great Smoky Mountains.

BJ-1, BJ-2, and BJ-3 were, respectively, affected by traffic emissions, industrial emissions, and anthropogenic emissions.

XA was the site at the Xin’An weather operational station in Baodi County. The symbols ‘#1’ - ‘#8’ represent different sampling dates.

FM: The FM at the XA site represents PM_{2.1}. The FMs at the other sites represent PM_{2.5}.

Lines 261 – 263: Does this mean the Chinese government has made more efforts to control SO₂ emissions than other governments or has made more efforts to control SO₂ than NO_x emissions? Please clarify in the text.

Reply: To address this comment, the following has been added to the revised manuscript:

“In recent years, the Chinese government has made more efforts to control SO₂ emissions (Q. Zhang et al., 2019), e.g., adjusting and optimizing industrial capacities. Clean fuels have also been promoted in the residential sector, with trials for using clean energy in heating in northern China carried out in all "2+26" cities and in the Fenhe and Weihe River Plains. In addition, compliance with industrial emission standards has been strengthened. Desulfurization technology has been also applied to many heavy industrial facilities. However, China has many small-scale manufacturing enterprises, so it is much more difficult to regulate NO_x emissions than SO₂ emissions. H. Li et al. (2109) have reported that emissions of SO₂ and NO_x in 2017 dropped by 79.9% and 38.1%, respectively, from 2014 levels in Beijing, China. In 2020, SO₂ and primary PM_{2.5} emissions dropped to one million tons, while NO_x and volatile organic compound emissions were still ten million tons.”

[Lines 287-295]

Line 315: Please define “DF”.

Reply: Done. DF is the difference between $f(\text{RH} = 85\%, 525 \text{ nm})$ and $f(\text{RH} = 80\%, 525 \text{ nm})$, i.e., $f(\text{RH} = 85\%, 525 \text{ nm})$ minus $f(\text{RH} = 80\%, 525 \text{ nm})$.

[Lines 347-348]

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