# 1 Trends, composition, and sources of carbonaceous aerosol at

# 2 the Birkenes Observatory, Northern Europe, 2001-2018

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#### 20 Abstract

- 21 We present 18 years (2001–2018) of aerosol measurements: organic and elemental carbon (OC and EC),
- 22 organic tracers (levoglucosan, arabitol, mannitol, trehalose, glucose, 2-methyltetrols), trace elements
- 23 and ions at the Birkenes Observatory (Southern Norway), a site representative of the Northern European
- region. The OC/EC (2001-2018) and the levoglucosan (2008-2018) time series are the longest in
- 25 Europe, with OC/EC available for the PM<sub>10</sub>, PM<sub>2.5</sub> (fine) and PM<sub>10-2.5</sub> (coarse) size fractions, providing
- 26 the opportunity for a nearly two-decade long assessment. Using positive matrix factorization (PMF) we
- 27 identify seven carbonaceous aerosol sources at Birkenes: Mineral dust dominated (MIN),
- 28 traffic/industry-like (TRA/IND), short range transported biogenic secondary organic aerosol
- 29 (BSOA<sub>SRT</sub>), primary biological aerosol particles (PBAP), biomass burning (BB), ammonium nitrate
- 30 dominated ( $NH_4NO_3$ ), and (one low carbon fraction) sea salt (SS).

- 32 yr<sup>-1</sup>), and a smaller decline in levoglucosan (-2.8% yr<sup>-1</sup>), suggesting that OC/EC from traffic and industry
- is decreasing, while abatement of OC/EC from biomass burning has been slightly less successful. EC
- 34 abatement of anthropogenic sources is further supported by decreasing EC fractions in  $PM_{2.5}$  (-3.9% yr<sup>-</sup>
- <sup>1</sup>) and PM<sub>10</sub> (-4.5% yr<sup>-1</sup>). PMF apportioned 72% of EC to fossil fuel sources, further supported by PMF
- 36 applied to absorption photometer data, which yielded a two-factor solution with a low aerosol Ångstrøm
- 37 exponent (AAE=0.93) fraction assumed to be equivalent black carbon from fossil fuel combustion
- 38 (eBC<sub>ff</sub>), contributing 78% to eBC mass. The higher AAE fraction (AAE=2.04) is likely eBC from BB

<sup>31</sup> We observed significant (p<0.05), large decreases of EC in  $PM_{10}$  (-3.9% yr<sup>-1</sup>) and  $PM_{2.5}$  (-4.2%

- 39 (eBC<sub>bb</sub>). Source receptor model calculations (FLEXPART) showed that Continental Europe and western
- 40 Russia were the main source regions both of elevated  $eBC_{bb}$  and  $eBC_{ff}$ .
- Dominating biogenic sources explain why there was no downward trend for OC. A relative increase in the OC fraction in  $PM_{2.5}$  (+3.2% yr<sup>-1</sup>) and  $PM_{10}$  (+2.3% yr<sup>-1</sup>) underscores the importance of biogenic sources at Birkenes (BSOA and PBAP), which were higher in the vegetative season and dominated both fine (53%) and coarse (78%) OC. Furthermore, 77–91% of OC in  $PM_{2.5}$ ,  $PM_{10-2.5}$  and  $PM_{10}$  was attributed to biogenic sources in summer vs. 22-37% in winter. The coarse fraction had the highest share of biogenic sources regardless of season and was dominated by PBAP, except in winter.
- 47 Our results show a shift in aerosol composition at Birkenes and thus also in the relative source 48 contributions. The need for diverse off-line and on-line carbonaceous aerosol speciation to understand 49 carbonaceous aerosol sources, including their seasonal, annual, and long-term variability has been 50 demonstrated.
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### 52 1. Introduction

53 Carbonaceous aerosol has been studied intensively over the last 20 years due to its influence on radiative forcing (Bond et al., 2013; Myhre and Samset, 2015; Lund et al., 2018), both directly by 54 55 scattering and absorption of sunlight, and semi directly and indirectly by influencing cloud properties 56 (Boucher et al., 2013; Hodnebrog et al., 2014; Myhre et al., 2013). It also contributes to the burden of 57 respiratory and cardiovascular disease (Janssen et al., 2012; WHO, 2013). Consequently, carbonaceous 58 aerosol [here: elemental carbon (EC) and organic carbon (OC)] is measured regularly in major air 59 monitoring networks such as e.g., EMEP and IMPROVE (e.g., Malm et al. 1994; Tørseth and Hov, 2003; Tørseth et al., 2012; UNECE, 2019; Hjellbrekke, 2020). Carbonaceous aerosol has an atmospheric 60 61 lifetime of days to a few weeks and is thus relevant for atmospheric long-range transport. Accordingly, 62 the European Monitoring and Evaluation Programme (EMEP) included OC/EC measurements in 2004 63 after a pioneering measurement campaign at 12 European sites from 2002-2003 (Yttri et al., 2007a; 64 Tørseth et al., 2012), showing that carbonaceous aerosol was a major constituent of the ambient aerosol in the European rural background environment, accounting for 9-37% (OM = organic matter) and 1-65 5% (EC) of PM<sub>10</sub>, and that OM was more abundant than sulfate (SO<sub>4</sub><sup>2-</sup>) at sites reporting both variables 66 67 (Yttri et al., 2007a). Similar conclusions were found from another long-term campaign, CARBOSOL 68 (Gelencsér et al., 2007; Pio et al., 2007), which monitored atmospheric aerosol and its components for 69 two years at six sites along a west-east transect extending from the Azores, in the mid-Atlantic Ocean, 70 to K-Kuszta (Hungary), in centra Europe.

# There are numerous carbonaceous aerosol sources, both anthropogenic, e.g. emissions from combustion of fossil fuel and biomass, and biogenic, e.g. vegetation emitted terpene/isoprene oxidation, and primary biological aerosol particles (PBAP) from e.g. plants and fungi (Bauer et al., 2002; Donahue et al., 2009; Hallquist et al., 2009; Fröhlich-Nowoisky et al., 2016).

75 Detailed source apportionment and quantification of carbonaceous aerosol is challenging due to 76 it numerous sources, the complexity of atmospheric formation and the vast number of organic 77 compounds associated with carbonaceous aerosols. A few studies have addressed carbonaceous aerosol 78 sources in the European rural background environment using source-specific organic tracers (Gelencsér 79 et al., 2007; Szidat et al., 2009; Genberg et al., 2011; Gilardoni et al., 2011; Yttri et al., 2011a,b). These 80 consistently show that residential wood burning dominates OC in winter, whereas biogenic secondary 81 organic aerosol (BSOA) is the major source in summer. PBAP makes a significant contribution to  $PM_{10}$ 82 in the vegetative season in the Nordic countries, second only to BSOA (Yttri et al., 2011a,b). Fossil fuel 83 sources typically dominate EC regardless of season but residential wood burning emissions can be 84 equally important and occasionally dominate in the heating season (Zotter et al., 2014; Yttri et al., 2019). 85 On-line high time-resolution measurements by aerosol mass spectrometer (AMS) and aerosol chemical 86 speciation monitors (ACMS) have become available in recent years, complementing off-line analysis of 87 organic tracers. In the comprehensive study by Crippa et al. (2014), including 15 European rural 88 background sites and 2 urban sites, covering winter, spring and fall, hydrocarbon-like organic aerosol 89 (OA) (11±5%) and biomass burning OA (12±5%) contributed almost equally to the total OA 90 concentration. The vast majority was however attributed to secondary sources; i.e., semi volatile 91 oxygenated OA ( $34\pm11\%$ ) and low-volatility oxygenated OA ( $50\pm16\%$ ). Secondary oxygenated OA 92 (OOA) can be both anthropogenic and biogenic; however Crippa et al. (2014) did not draw any 93 conclusions on this. Results presented by Bougiatioti et al. (2014) show how freshly emitted biomass 94 burning OA can be transformed to more oxidized OOA after just a short time in the atmosphere when 95 subject to high temperatures and high solar radiation.

96 Over the last decades, European anthropogenic emissions of secondary inorganic aerosol precursors, 97 e.g. ammonia (NH<sub>3</sub>) and nitrogen oxides (NO<sub>x</sub>), and non-methane volatile organic compounds 98 (NMVOC) have stabilized, and those of sulfur dioxide (SO<sub>2</sub>) significantly reduced, following 99 implementation of the Gothenburg Protocol (Reis et al., 2012; UNECE, 2013; Matthews et al., 2020). 100 The anthropogenic carbonaceous aerosol is not regulated by any binding international protocol, although 101 co-benefit is expected from the regulation of NO<sub>x</sub> and NMVOC, which act as precursors of secondary 102 organic aerosol (Hallquist et al., 2009). PM<sub>2.5</sub> was included in the revised version of the Gothenburg 103 protocol (UNECE, 2013) in 2012, which states that effort should be directed towards sources that also 104 emit black carbon (BC), which inevitably also will influence OC.

Residential wood burning is a major source of carbonaceous aerosol in circumpolar countries (e.g.
Yttri et al., 2014) and even considered the most important source in Norway, accounting for 48% (2017)
of PM<sub>2.5</sub> (Grythe et al., 2019). This region also regularly experiences major wild and agricultural fires
(e.g. Stohl et al., 2006 and 2007). A growing number of studies show that residential wood burning is
more widespread in continental Europe than previously assumed and that its contribution to the ambient
carbonaceous aerosol can be substantial (Sillanpää et al., 2005; Gelencsér et al., 2007; Puxbaum et al.,
2007; Lanz et al., 2010; Maenhaut et al., 2012; Genberg et al., 2011; Fuller et al., 2014; Yttri et al.,

2019) and even dominating (Szidat et al., 2007; Herich et al., 2014). Residential wood burning is a
decentralized source in Europe and combustion typically takes place in small units where the emissions
are emitted without after-treatment. An economic downturn in Greece compelled households to burn

115 firewood and waste material as fuel costs rose, increasing residential wood burning emissions in urban

areas by 30% (Saffari et al., 2013). Future increases in European wood burning emissions might occur

117 due to climate change mitigation policies supporting the use of renewable and biofuels (van der Gon et

al., 2015). Denier van der Gon et al. (2015) conclude that European emissions from residential wood

burning are significantly underestimated, thus it appears timely to address how ambient carbonaceous

120 aerosol -particularly from biomass burning -has developed over the last two decades.

121 Kahnert et al. (2004) and Tørseth et al. (2012) highlight the importance of long-term measurements 122 (> 10 years) of carbonaceous aerosol. The Birkenes Observatory in southern Norway holds the longest 123 time series of OC and EC in Europe, dating back to 2001, including measurements in both the  $PM_{10}$  and 124 the  $PM_{2.5}$  fractions. Infleunced by major anthropogenic emission regions in Europe, the Birkenes 125 Observatory is well suited to monitor air pollution from Continental Europe.

Here we apply positive matrix factorization (PMF) to identify sources of carbonaceous aerosol at the Birkenes Observatory. Measurements of complementary species accompany OC/EC monitoring, allowing us to understand these sources, their contribution and variability at time scales from minutes to decades: organic tracers for biomass burning (levoglucosan), PBAP (arabitol, mannitol, trehalose and glucose) and BSOA (2-methyltetrols), as well as high time resolution equivalent black carbon resulting from biomass (eBC<sub>bb</sub>) and fossil (eBC<sub>ff</sub>) fuel combustion, derived from multiwavelength aethalometer measurements.

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#### 134 **2.** Methodology

#### 135 2.1 Sampling site

136 The Birkenes Observatory (58°23'N, 8°15'E, 219 m above sea level, asl) is an EMEP/GAW (Global 137 Atmospheric Watch) supersite in southern Norway (Figure 1) situated 100 m south-east of the old Birkenes site, initiating measurements in 2009. The observatory is in the Boreo-nemorale zone with 138 139 mixed coniferous and deciduous trees (65% of the land use near the site); the remainder being meadows (10%), low intensity agricultural areas (10%), and freshwater lakes (15%). Close to the Skagerrak coast 140 141  $(\sim 20 \text{ km})$  and at low altitude, the observatory experiences a maritime climate with relatively mild winters 142 and moderately warm summers. The prevailing wind is westerly/south westerly. Figure S 1 shows 143 ambient temperature and precipitation (2001-2018) at Birkenes. The nearest city is Kristiansand 144 (population  $\sim 61\ 000$ ) 25 km to the south/south-west.

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#### 146 **2.2 Measurements and procedures**

#### 147 **2.2.1 Off-line filter measurements**

148 We collected OC/EC, organic tracers and PM mass filter samples using two low-volume samplers with

149 a PM<sub>10</sub> and a PM<sub>2.5</sub> inlet. Quartz fiber filters (Whatman QM-A; 47 mm in diameter) were pre-fired (850 150 °C; 3 h). We conditioned the filters  $[20 \pm 1^{\circ}C; 50 \pm 5\% \text{ RH} (relative humidity)]$  for 48 h before and after 151 exposure and weighed them to obtain PM mass. We kept filters in petri slides and stored them at 4 °C 152 after weighing and before OC/EC analysis. After OC/EC analysis and prior to organic tracer analysis 153 the samples were stored at -18 °C. Two field blanks were assigned to each month of sampling and were 154 treated in exactly the same manner regarding preparation, handling, transport and storage as the exposed 155 filters, except that they were not inserted in the samplers. We collected one sample per sampler per week 156 (168 hours), except for 14 August 2002–17 September 2008, when two samples were collected per 157 sampler per week; at 24 h and 144 h intervals. The sampling inlets are 2 m above the Observatory roof, 158 5 m above the ground level (~226 m asl). The OC/EC and PM mass time series date back to February 159 2001 and organic tracers back to January 2008 (monosaccharide anhydrides) and January 2016 (sugars, 160 sugar-alcohols and 2-methyltetrols).

161 We performed thermal-optical analysis (TOA, Sunset Laboratory OC/EC instrument), using 162 transmission for charring correction. We used the Quartz temperature programme in 2001–2008 and 163 EUSAAR-2 (Cavalli et al., 2010) from 2008. We compare the two temperature programmes for  $PM_{2.5}$ 164 samples collected in 2014 in Supplementary Sect. S1. OC/EC instrument performance is regularly inter-165 compared under the joint EMEP/ACTRIS quality assurance and quality control effort (e.g. Cavalli et 166 al., 2013).

167 Until 2014, we determined monosaccharide anhydrides (levoglucosan, mannosan, galactosan) 168 in PM<sub>10</sub> using high-performance liquid chromatography high-resolution time-of-flight mass 169 spectrometry (HPLC-HR-TOFMS) in negative electrospray ionization mode according to the method 170 of Dye and Yttri (2005). After 2014, we use ultra-performance liquid chromatography (UPLC), with 171 two Waters columns (2 x 2.1 x 150 mm HSS T3, 1.8 µm, Waters Inc.). Changing the column improved 172 the chromatographic resolution, allowing the analysis of sugars, sugar-alcohols and 2-methyltetrols. We 173 identified the monosaccharide anhydrides based on retention time and mass spectra (accurate mass and 174 isotope pattern) of authentic standards (Table S 1). Isotope-labelled standards of levoglucosan, 175 galactosan, arabitol, mannitol, trehalose and glucose were used as internal recovery standard (Table S 176 1).

Weekly OC/EC, PM<sub>10</sub>, PM<sub>2.5</sub> are publicly available on EBAS (<u>http://ebas.nilu.no</u>). Mean values
(daily/weekly/seaonal/annual) used below, merging of data from the old and new Birkenes sites, and
quality assurance of the filter data are detailed in Sect. S1. We used the Mann-Kendall test (Mann, 1945;
Kendall, 1975; Gilbert, 1987) to identify significant trends in the filter based measurements, and the
Theil-Sen slope (Theil, 1950; Sen, 1968; Gilbert, 1987) to quantify the trends (Sect. S2).

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### 183 2.2.2 Online measurement and source apportionment of absorption coefficients

We determined Absorption coefficients (B<sub>Abs</sub>) using a multi-wavelength absorption photometer (AE33
Aethalometer, Magee Scientific). Here we performed source apportionment using the aethalometer

- 186 model (Sandradewi et al., 2008) to determine eBC<sub>bb</sub>/eBC<sub>ff</sub>. However, the aethalometer model requires *a-priori* knowledge of the aerosol Ångstrøm exponents (AAE), uncertainties which can lead to large 187 188 variation in the magnitude of the resulting time series and negative concentrations during some periods. 189 Often, the aethalometer model yields negative concentrations for any single input AAE pair. Therefore, 190 we also used a novel **PMF** application (Platt et al., in prep.) finding two factors, a low AAE factor (0.9) 191 and a higher AAE factor (2.04) identified as eBCff and eBCbb, respectively. Uncertainties were assessed 192 using bootstrapping (n=2000). The advantages of the PMF are that no a-priori knowledge of the factor 193 AAEs is required, no periods of negative concentration result, deviations from a strict power-law 194 dependence of B<sub>Abs</sub> on wavelength (e.g. due to degradation of light absorbing components in the 195 atmosphere or instrument errors/bias) are permitted, and poorly fitting data are assigned to a residual. 196 Meanwhile bootstrapping allows estimation of uncertainties, the methodology of the PMF analysis and 197 aethalometer data post-processing are detailed in Sect. S3 (Table S 2).
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### 199 2.3 FLEXPART model simulations

200 We investigated the origin of the observed eBC with a Lagrangian transport model (FLEXPART v10.4, 201 Pisso et al., 2019). The model, powered by European Centre for Medium-Range Weather Forecasts with 137 vertical layers and a horizontal resolution of 0.1°×0.1° tracks simulated particles arriving at the 202 203 receptor 30 days backwards in time (retro-plume mode) and accounts for gravitational settling, dry and 204 wet deposition (Grythe et al., 2017), turbulence (Cassiani et al., 2014), unresolved mesoscale motions 205 (Stohl et al., 2005) and includes a deep convection scheme (Forster et al., 2007). Output consists of an emission sensitivity ( $0.5^{\circ} \times 0.5^{\circ}$  resolution), a quantitative measure for the particle mass concentration at 206 207 the receptor resulting from a unit emission flux at the Earth's surface. The emission sensitivity can also 208 be interpreted as a probability distribution field of the particle's origin, used in the present study to 209 identify possible source regions of eBC.

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## 211 **2.4 Positive Matrix Factorization analysis on filter data**

We performed PMF ME2 (Canonaco et al., 2013) (See Sect. S3 for a description of the analysis principle 212 213 and S4 for its application to filter data) for samples collected in 2016-2018 (151 samples), using the 214 following as input data: OC (in  $PM_{2.5}$  and  $PM_{10-2.5}$ ), EC (in  $PM_{10}$ ), levoglucosan, mannosan, galactosan, arabitol, mannitol, trehalose, glucose, 2-methylerythritol, 2-methylthreitol, V, Mn, Ti, Fe, Co, Ni, Cu, 215 Zn, As, Cd, and Pb (all in PM<sub>10</sub>), SO<sub>4</sub><sup>2</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, K<sup>+</sup>, Na<sup>+</sup>, Cl<sup>-</sup> (from open filter face). 216 217 Table S 3 shows miscellaneous settings of the PMF analysis of these data including missing data 218 treatment and an assessment of the PMF performance. The input data and error estimates were prepared 219 using the procedure suggested by Polissar et al. (1998) and Norris et al. (2014), see Sect. S3. 220 Source apportionment by PMF is based on the temporal variability of the components. It is 221 expected that significant contributions to carbonaceous aerosol at Birkenes is via long-range 222 atmospheric transport (LRT), alongside more local sources. Local and LRT sources will have different temporal variability, and significant mixing of air masses and chemical transformation is expected for the latter, i.e., factor profiles at Birkenes are expected to differ somewhat from emission profiles at the source, even though the profile is distinctive enough for source attribution. Because of this we did not attempt to constrain factor profiles via e.g. ME-2 (Canonaco et al., 2013) since Birkenes, as a relatively clean rural background site, is unlikely to receive unprocessed emissions. Furthermore, mixed contributions to a factor can in some cases be resolved *a posteriori* for source quantification (i.e. if it is clear where mass should be reassigned), without potentially perturbing the output factor time series.

- Two previous studies have used factor analysis to study PM sources at Birkenes (Aamundsen et al. (1992; Maenhaut, 2018). The present study focuses on carbonaceous aerosol, using OC, EC and highly source specific organic tracers as input in addition to inorganic species and elements used by Aamundsen et al. (1992) and Maenhaut (2018). This provides a different set of factors, based on different input, hampering any reliable comparison of these studies.
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### 236 2.4.1 Identification of PMF factors

The biomass burning (BB) factor appears well confined in the PMF solution (Figure 2, Table S 4), explaining all the monosaccharide anhydrides (95–98%). OC<sub>BB</sub> was almost exclusively (87%) in the fine fraction of PM<sub>10</sub>. Other key qualifiers derived from the BB factor are the ratios listed in Table 1, which are highly comparable to the results obtained by <sup>14</sup>C-analysis reported in the comprehensive study by Zotter et al. (2014). The BB factor is elevated in the heating season and peaks in winter, pointing to residential heating as the major source.

243 The TRA/IND factor explained most EC (50%), the majority of the trace elements Pb (84%), 244 Zn (82%), Cd (81%), As (78%), V (70%), Ni (69%), Cu (62%) and Co (42%) and a noticeable fraction 245 of SO<sub>4</sub><sup>2-</sup> (20%), which suggests influence of various anthropogenic emissions. TRA/IND explained a 246 small fraction of fine OC (10%) and a negligible fraction of coarse OC (4%). The majority of OC (88%) 247 resides in the fine fraction, which is in line with its combustion-derived origin. The high EC fraction 248 unambiguously points to combustion processes, and the low OC/EC ratio (1.4 for PM2.5) towards a 249 substantial, but not exclusive, influence from vehicular traffic. Cu and Zn result from brake wear (Fomba 250 et al., 2018), whereas tire wear is an additional source of Zn (Pacyna et al., 1986), corroborating the influence of vehicular traffic to the TRA/IND factor. Ni and V are commonly associated with 251 252 combustion of heavy oil (Viana et al., 2008), As, Cd and Pb with combustion of coal, and to a lesser 253 extent oil, but also from metallurgic activity (Pacyna et al., 1986). The TRA/IND factor has a minimum 254 in summer and shows minor variability for the rest of the year. A similar drop in the vehicular traffic 255 factor in summer for Helsinki was shown by Saarikoski et al. (2008).

The PMF analysis confined the majority of coarse OC (53%) and essentially all (82–93%) of the PBAP tracers (arabitol, mannitol, trehalose, and glucose) within one factor (PBAP). The PBAP factor has a pronounced seasonal variability with increased levels in the vegetative season and nearly absent outside of it, as previously described for coarse OC (Yttri et al., 2007a) and PBAP tracers (Yttri et al., 2007b) at Birkenes.

261 2-methyltetrols (92–96%) are oxidation products of isoprene (Claeys et al., 2004) and are almost exclusively attributed to the BSOA<sub>SRT</sub> (SRT=Short Range Transport) factor, which explains 9% of fine 262 OC and 13% of coarse OC. The complete absence of EC and the presence of SO<sub>4</sub><sup>2-</sup> (17%) underpins the 263 secondary nature of this factor, which is present in summer with tail ends in late spring an early fall. The 264 265 BSOA<sub>SRT</sub> time series increases abruptly in the transition May/June, as leaves unfold, and subsides 266 equally rapid in the beginning of October when trees shed their leaves. The near absence of 2-267 methyltetrols prior to May/June suggests that the 0.5-1.5 months earlier onset of the vegetative season 268 in Continental Europe (Rötzer and Chmielewski, 2001) is not reflected by the 2-methyltetrols 269 observations at Birkenes, indicating a short atmospheric lifetime for 2-methyltetrols. Consequently, 270 local isoprene emissions likely explain the observed concentrations of 2-methyltetrols at Birkenes, 271 questioning to what extent the BSOA<sub>SRT</sub> factor includes a continental BSOA contribution. Similar 272 sources (deciduous and coniferous trees), temperature dependent emissions, and formation rates, suggest 273 that particulate phase oxidation products of mono- and sesquiterpenes are accounted for by the isoprene-274 derived BSOA<sub>SRT</sub>-factor as well, but with a similar issue concerning local versus LRT contribution, as 275 proposed for the 2-methyltetrols.

276 The MIN factor is defined by its content of Ti (93% of total), Fe (75%), Mn (52%) and Ca<sup>2+</sup> 277 (39%) (Figure 2, Table S 4), well-known constituents of mineral dust (e.g. Alastuey et al., 2016). It also contains some of the elements that dominate the TRA/IND factor, including Co (43%), Cu (20%), Ni 278 (17%) and V (14%), indicating anthropogenic influence. Notably, 31% of fine OC is attributed to the 279 280 MIN factor, whereas it explains 13% of coarse OC. This corresponds to that reported by Kyllönen et al. 281 (2020) for the Subarctic site Pallas (Finland) where 29% of the fine OC was apportioned to the mineral 282 dust factor. Waked et al. (2014) found a similar result for Lens (France) where the mineral dust factor 283 explained 15% of OC. No information on the size distribution was available in Kyllönen et al. (2020) 284 and Waked et al. (2014), whereas in the present study 86% of OC in the MIN factor resides in the fine 285 fraction of PM<sub>10</sub>. Since mineral dust typically resides in the coarse fraction of PM<sub>10</sub> (Ripoll et al., 2015), one would expect the same for its carbon content, e.g. as CaCO<sub>3</sub>. More efficient deposition of coarse 286 287 mode mineral dust during LRT is one possible explanation but mixing of air masses is more likely, as 288 13% of the EC also resides in this factor. The high OC/EC ratio in the unweighted MIN factor profile 289 (18 for  $PM_{2.5}$ ) indicates a minor primary combustion particle influence, and the absence of levoglucosan 290 shows that the EC content originates from fossil fuel combustion (consistent with some TRA/IND 291 influence). Using Eq. (1), 8% of the MIN factor's fine OC content is attributed to combustion of fossil fuel OC (OC<sub>PrimFF</sub>), whereas the corresponding percentage for  $PM_{10}$  OC is 7%. If all Ca<sup>2+</sup> and Mg<sup>2+</sup> in 292 the MIN factor was present as either Calcite (CaCO<sub>3</sub>) or Dolomite CaMg(CO<sub>3</sub>)<sub>2</sub>, the CO<sub>3</sub><sup>2-</sup>-carbon would 293 account for no more than 3% of the factor's  $PM_{10}$  OC content, and 22% if all reside in its coarse fraction. 294 295 This shows that the OC content of the MIN factor mostly originates from other sources than mineral 296 dust and combustion of fossil fuel. The MIN factor is most abundant in spring and early summer, as

seen by Waked et al. (2014), and is associated with southern air masses, as seen for the dry and warm period in the transition of May/June 2018 (Figure 3) when there was a pronounced peak in the MIN factor time series (Figure 2). Indeed, the mean ambient temperature was 4°C higher in May 2018 than for May 2001–2018, whereas it was 2.4°C higher for June 2018 than for June 2001–2018. We thus suggest that the climatological conditions that activate mineral dust sources also favour BSOA formation and that the majority of both fine (92%) and coarse fraction (78%) OC in the MIN factor is LRT BSOA (OC<sub>BSOA,LRT</sub>).

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$$OC_{Fossil,primary,MIN} = [EC_{MIN}] x \left(\frac{OC}{EC}\right)_{TRA/IND}, \left(\frac{OC}{EC}\right)_{TRA/IND} = 1.4$$
 Eq.1

The majority of  $NH_4^+$  (77%) and  $NO_3^-$  (68%) reside in the  $NH_4NO_3$  factor, which points to secondary inorganic aerosol (SIA) formation during LRT. This is supported by a noticeable contribution of  $SO_4^{2-}$  (35%) to the  $NH_4NO_3$  factor, as well. The factors content of  $NO_2$  (30%) points towards a combustion-derived origin of  $NO_3$ , as does EC (13%). The factor's OC content is comparable to that seen for the BB factor. The factor is most pronounced in winter and spring.

The sea salt (SS) aerosol factor was recognized by its high Cl<sup>-</sup> (96%), Na<sup>+</sup> (87%) and Mg<sup>2+</sup> (79%) fractions. The K<sup>+</sup>/Na<sup>+</sup> (0.036), Ca<sup>2+</sup>/Na<sup>+</sup> (0.034) and SO<sub>4</sub><sup>2-</sup>/Na<sup>+</sup> (0.282) ratios derived from the SS factor closely resemble these ratios in sea water (0.037, 0.038 and 0.252) (Stumm and Morgan, 1996), further demonstrating the successful separation of this factor.

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#### 317 **3. Results and discussion**

#### 318 **3.1** Levels and trends of carbonaceous aerosol and organic tracers

- 319 Annual mean carbonaceous aerosol concentrations at Birkenes (2001–2018) are among the lowest in Europe (Yttri et al., 2007a; Yttri et al., 2019), with OC from 0.56–1.07 µg C m<sup>-3</sup> for PM<sub>10</sub> and 0.50– 320 0.93  $\mu$ g C m<sup>-3</sup> for PM<sub>2.5</sub>, and EC from 0.05–0.15  $\mu$ g C m<sup>-3</sup> (Figure 4; Table S 4). EC, being from 321 322 combustion that generates fine PM, was almost exclusively associated with PM<sub>2.5</sub>, whereas OC was 323 abundant also in the coarse fraction ( $PM_{10-2.5}$ ), particularly in summer and fall (Figure 4). The 324 correlation between OC and EC varied by season (Table S 6) and was highest in the heating season, 325 reflecting the contribution of biogenic, non-EC sources, such as BSOA and PBAP in the vegetative season. The higher R<sup>2</sup>-values for PM<sub>2.5</sub> compared to PM<sub>10</sub> can partly be attributed to PBAP, which 326
- 327 mainly resides in  $PM_{10-2.5}$ .

The variability of the annual mean OC (15–22%) and EC (27%) concentrations was comparable to the major secondary inorganic aerosol (SIA) (SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>) and sea salt (SS) aerosol species (Na<sup>+</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>) (25–31%). A difference of > 60% between consecutive years was observed for OC and EC in PM<sub>10</sub> and PM<sub>2.5</sub>, whereas 160% was seen for OC in PM<sub>10-2.5</sub>. It is important to note that despite decades of SO<sub>2</sub>, NH<sub>3</sub> and NO<sub>x</sub> mitigation efforts, SIA dominates PM<sub>10</sub> mass (29–52%) most years, followed by carbonaceous aerosol (24–40%) and SS aerosol (10–28%) (Figure 5; Table S 7). SIA

- 334 constituents were also the largest PM<sub>10</sub> fraction during air pollution episodes (Table S 8), reflecting that
- 335 Birkenes is affected by major SIA precursor emission regions in Continental Europe.
- Levels of total carbon (TC) and PM fractions are shown in Table S 9 and Table S 10, respectively
   for completeness. In the following sections we discuss the OC and EC fractions separately in detail.
- 339 3.1.1 Organic carbon
- We found no significant trend for OC in  $PM_{10}$  (OC<sub>PM10</sub>). For fine OC in  $PM_{2.5}$  (OC<sub>PM2.5</sub>) there was a minor decrease (-0.8% yr<sup>-1</sup>), whereas there was a minor increase for coarse OC (OC<sub>PM10-2.5</sub>) (0.8% yr<sup>-1</sup>) (Table S 11). The anthropogenic fraction of OC observed at Birkenes likely has a downward trend as found for EC (Sect. 3.1.2), but the substantial influence of natural sources demonstrated in the present, as well as in previous, studies (Yttri et al., 2011b), explains the general lack of trends for OC.
- The OC time series are characterized by two years where the annual mean was substantially 345 higher (2006) and lower (2012) than the proceeding and the following year (Figure 4). The increased 346 347 level in 2006 was most pronounced in the fine fraction and in all seasons except spring, whereas the 348 drop in 2012 mainly was attributed to the coarse fraction and was observed in all seasons. The OC<sub>PM10</sub>-349 2.5 annual mean time series is characterised by a stepwise increase from 2001 up to, and including, 2006, 350 after which the concentration dropped and showed minor annual variability, except for the very low 351 annual mean of 2012. After 2015, there are indications of a similar stepwise increase as seen for 2001– 352 2006.
- 353 The OCPM<sub>10-2.5</sub> contribution to OC<sub>PM10</sub> ranged from 18–35% on an annual basis (2001 excluded due to
- data capture <50%), and levels were highest in summer and fall. Previous studies (Simpson et al.,
- 355 2007; Yttri et al., 2011a,b) showed that BSOA largely dominates the fine carbonaceous aerosol in
- 356 summer at Birkenes, whereas the present study shows that Birkenes regularly experiences major air
- pollution events in spring, as a result of LRT (Table S 4, Table S 7 and Table S 8). Hence, both
- biogenic sources and LRT explain the observed seasonality of fine OC.
- 359 We attribute elevated OC<sub>PM2.5</sub> in winter 2010 to residential wood burning emissions as discussed in Sect. 3.2.1. Only on five occasions did the seasonal mean of OC<sub>PM2.5</sub> exceed 1 µg C m<sup>-3</sup>, four of those 360 in the first three years of the time series. The highest mean was observed in summer 2002 (1.4 µg C m<sup>-</sup> 361 362 <sup>3</sup>) when wildfires in Eastern Europe influenced Birkenes (Yttri et al., 2007a). The four other occasions, 363 spring (2001, 2002, 2003 and 2018), also saw prolonged episodes of PM air pollution with the hallmark of LRT; i.e., elevated SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>. According to our PMF analysis (See Sect. 3.2) there are 364 several anthropogenic and biogenic sources likely to contribute to fine OC at Birkenes, whereas coarse 365 366 fraction OC is dominated by a single source, PBAP (Yttri et al., 2007 a,b; Yttri et al., 2011 a,b; Glasius et al., 2018). Hence, it is not surprising that OC<sub>PM2.5</sub> was the dominant OC fraction, accounting for 70-367 89% OC<sub>PM10</sub> on an annual basis. 368
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#### 370 3.1.2 Elemental carbon

- 371 Notably, EC levels dropped from 2007–2008, contrasting with the annual mean OC time series,
- 372 (Figure 4 and Table S 5). This major downward trend of EC clearly points to changing source
- 373 contributions to EC at Birkenes. We rarely observed seasonal means exceeding 0.15 µg C m<sup>-3</sup>; only in
- 374 winter 2006, 2007 and 2010, spring 2001, 2003 and 2007, and fall 2005 and 2011. Weekly samples
- 375 exceeded 0.5 µg C m<sup>-3</sup> for three samples only, all associated with LRT.
- A statistically significant reduction was calculated for EC in  $PM_{10}$  (-3.9% yr<sup>-1</sup>) and  $PM_{2.5}$  (-4.2% 376 yr<sup>-1</sup>) (Table S 11), corresponding well with  $SO_4^{2-}$  (-3.8% yr<sup>-1</sup>) and  $PM_{2.5}$  (-4.0% yr<sup>-1</sup>). The trend for EC 377 378 was most pronounced in spring and summer  $(-4.0 - -5.9\% \text{ yr}^{-1})$  (Table S 12), as seen for SO<sub>4</sub><sup>2-</sup> (-4.2 - -379 6.4% yr<sup>-1</sup>) and PM<sub>2.5</sub> (-3.0 – -4.4% yr<sup>-1</sup>) (Table S 12). The EMEP model finds a somewhat lower 380 reduction for EC (-3.0 % yr<sup>-1</sup>) for 2001–2017 (EEA, 2020) with the largest emission reductions for the 381 road transport (83 kt; -3.6% yr<sup>-1</sup>) and off-road categories (44 kt; -3.7% yr<sup>-1</sup>) (https://www.ceip.at), which are sectors with a minor seasonal variability. We suggest that these sectors explain the downward trend 382 383 observed for EC at Birkenes, and that the seasonality of the EC trend is due to the substantial contribution from less abated sources, such as domestic heating in winter and fall. Notably, modelled EC emissions 384 are unchanged for the category other stationary combustion for 2001-2016 (-1 kt; -0.08% yr<sup>-1</sup>) 385 386 (https://www.ceip.at), which includes residential heating, and wood burning in particular.
- Effective abatement of SIA precursors and fossil EC, along with a high natural source contribution to OC, largely explains why the OC fraction increased significantly for  $PM_{2.5}$  (+3.2% yr<sup>-1</sup>) and  $PM_{10}$  (+2.4% yr<sup>-1</sup>), whereas it decreased for the EC fraction (-3.9 – -4.5% yr<sup>-1</sup>) (Table S 13). The largest increase (OC) and decrease (EC) was seen in the vegetative season (Table S 14) when BSOA and PBAP increase and the influence of poorly abated sources such as domestic heating is low. Consequently, these results demonstrate a long-term change in the aerosol chemical composition at Birkenes and thus also in the relative source composition of PM.
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#### 395 3.1.3 Levoglucosan

Levels of levoglucosan and other organic tracers are given in Table S 15, whereas other organic tracers(arabitol, mannitol, trehalose, glucose, and 2-methyltetrols) are discussed in Sect. S6.

The statistically significant decrease of levoglucosan (-2.8% yr<sup>-1</sup>) at Birkenes for 2008–2018 398 399 (Figure 6; Table S 11), and the fact that biomass burning levels observed at Birkenes are largely 400 explained by continental emissions (Figure 7) might indicate that wood burning emissions in continental 401 Europe are declining. However, surprisingly, we find no significant trend for levoglucosan on a seasonal 402 basis (Table S 12). Furthermore, and although one should be careful drawing conclusions from non-403 significant outcomes, it is worth noting that the levoglucosan to EC ratio most likely increased (+2.8%  $yr^{-1}(PM_{10})$ , CI (Confidence Interval) = -3.5 - +6.5%  $yr^{-1}$  and +2.3%  $yr^{-1}(PM_{2.5})$ , CI = -2.2 - 5.0 %  $yr^{-1}$ ) 404 for the period 2008–2018, whereas it most likely decreased (-1.8% yr<sup>-1</sup> (PM<sub>10</sub>), CI = -10.6 - +1.8 and -405 3.6% yr<sup>-1</sup> (PM<sub>2.5</sub>), CI = -9.8 - +1.3% yr<sup>-1</sup>) for the levoglucosan to OC ratio (Table S 13). A more 406

407 efficient abatement of fossil sources than biomass burning would explain the levoglucosan to EC

increase, whereas we fail to see a similar trend for the levoglucosan to OC ratio, as prevailing naturalsources mask the assumed reduction in fossil OC of anthropogenic origin.

410 The levoglucosan time-series provides a hitherto unprecedented opportunity to validate European 411 residential wood burning emission inventories at a decadal time basis. Unfortunately, the inventories 412 suffer from non-harmonized emission reporting and lack of condensable organics (van der Gon et al., 413 2015, Simpson et al., 2019), which hampers any reliable attempt for such validation. Given the 414 uncertainties in the trend calculations (i.e. annual vs. seasonal trends), more work is needed to 415 investigate trends in levoglucosan and biomass burning, foremost by continuation of the actual time 416 series. Such efforts should be initiated immediately given the numerous studies that point to residential 417 wood burning as a major source of air pollution in Europe (e.g. Denier van der Gon et al., 2015; Yttri et 418 al., 2019).

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#### 420 **3.2** Sources of carbonaceous aerosol at Birkenes

421 We used PMF to apportion carbonaceous aerosol at Birkenes for 2016-2018. The time period was 422 restricted by organic tracer data availability. Carbonaceous aerosol annual means for 2016-2018 were 423 within the long-term annual mean (±SD) for OC, and only slightly lower for EC in 2016 and 2017 and 424 are thus representative of the longer time series. Six out of seven factors identified in contribution-425 weighted relative profiles from PMF (Figure 2; Table S 4) were associated with significant amounts of 426 carbonaceous aerosol. This includes factors for mineral dust-dominated (MIN), which OC content is 427 associated mainly with LRT BSOA (BSOALRT), traffic/industrial-like (TRA/IND), biogenic secondary 428 organic aerosol (BSOA<sub>SRT</sub>), which is short-range transported, primary biological aerosol particles 429 (PBAP), biomass burning (BB), and ammonium nitrate dominated (NH<sub>4</sub>NO<sub>3</sub>). The sea salt aerosol factor 430 (SS) had a negligible (<1%) carbonaceous aerosol content.

The MIN factor (31%) explained the largest fraction of fine OC, whereas BB (17%), NH4NO<sub>3</sub> (17%) and PBAP (16%) had almost equally large shares, as did TRA/IND (10%) and BSOA<sub>SRT</sub> (9%) (Figure 8). Coarse OC was by far most abundant in the PBAP factor (53%), whereas BSOA<sub>SRT</sub> (13%), MIN (13%) and NH4NO<sub>3</sub> (12%) explained almost equally large shares. For the other factors, coarse OC was minor. EC was apportioned to only five factors of which TRA/IND (50%) dominated by far. BB made a 21% contribution and MIN and NH4NO<sub>3</sub> equally large shares (13%). The 3% apportioned to PBAP is an assumed analytical artefact (See Sect. 3.2.2 for details).

The BB, NH<sub>4</sub>NO<sub>3</sub> and TRA/IND factors are considered entirely anthropogenic, BSOA<sub>SRT</sub> and PBAP exclusively natural, whereas MIN is mixed (Figure 8). Natural (54%) and anthropogenic (46%) sources contributed almost equally to fine OC (Figure 8) annually, so also in spring and fall (51% natural), whereas natural sources prevailed in summer (77%) and anthropogenic in winter (78%). Natural sources dominated coarse OC annually (78%) and in all seasons (70–91%), except winter (37%). We consider the minor fraction of coarse OC attributed to carbonate-carbon (3%) to be of natural origin. The findings for OC in PM<sub>10</sub> are rather like that of PM<sub>2.5</sub>, only that the natural contribution is somewhat

more pronounced due to the influence from a mostly naturally influenced coarse OC fraction.

- 447 **3.2.1** Anthropogenic carbonaceous aerosol sources
- 448 According to PMF, BB accounted for 14–17% of OC annually, considering both PM<sub>2.5</sub> and PM<sub>10</sub> vs. 449 only 6% of coarse OC. BB was by far the major contributor to OC in winter (35-37%) and by far the 450 most minor contributor in summer (2-3%) (not considering SS). Spring and fall are transition seasons 451 where BB still made a substantial 14-19% contribution to OC. BB explained 22% of EC annually 452 (excluding  $EC_{PBAP}$ , which we assume is an analytical artefact, see Sect. 3.2.2), hence fossil fuel 453 combustion (78%) was the major source. Emissions from residential wood burning increased in the 454 heating season but fossil fuel sources dominated EC even in winter (66%). It cannot be excluded that 455 part of levoglucosan originates from wildfires in summer, spring, and fall, though this itself may be due 456 to anthropogenic activity. However, the levoglucosan/mannosan (L/M) ratio indicates minor variability 457 in the source composition throughout the year (See Sect. S5), suggesting one dominating source. 458 The 78%:22% split of EC into fossil fuel combustion and biomass burning derived from PMF is 459 supported by high time resolved concentrations of eBC<sub>BB</sub> and eBC<sub>FF</sub> derived from multiwavelength 460 aethalometer measurements of the absorption coefficient, following the PMF-approach of Platt et al. 461 (in prep.). With this approach we find eBC<sub>BB</sub>/eBC<sub>TOT</sub>=28% (Table 2). Meanwhile, using the 462 aethalometer model and AAE<sub>FF</sub>=0.9 and AAE<sub>BB</sub>=1.68 (Zotter et al. 2017) as input we find eBC<sub>BB</sub>/eBC<sub>TOT</sub>=48%, however the aethalometer model is extremely sensitive to the input AAE and the 463
- 464 AAE values suggested by Zotter et al. (2017) are only recommended where no *a priori* information on
- the AAEs is available and a significant advantage of the PMF approach by Platt et al. (in prep.) is that
- the AAE is an output.

467 Source regions of elevated (70<sup>th</sup> percentile) and low (30<sup>th</sup>) winter and summertime eBC<sub>BB</sub> (and eBC<sub>FF</sub>) observed at Birkenes for 2018 were studied using the approach of Hirdman et al. (2010). The 468 469 results show that Birkenes is a receptor of LRT exclusively from Continental Europe for elevated eBC<sub>BB</sub> 470 and eBC<sub>FF</sub> levels (Figure 7), both in summer and winter. This is consistent with a lack of diurnal 471 variation in either eBC<sub>BB</sub> or eBC<sub>FF</sub>, likely because there are few local sources at Birkenes. The main 472 source regions extend from the Atlantic coast in the west to the Ural Mountains in winter, whereas the 473 regions in summer are confined to Eastern Europe and western Russia (but not as far east as the Urals). 474 Notably, the Nordic countries do not contribute to elevated levels except for southern parts of Finland 475 in summer. The footprints are almost identical for eBC<sub>BB</sub> and eBC<sub>FF</sub> both for summer and winter. High 476 similarity in winter is not a surprise, as the footprint covers such a wide area and because wood burning 477 for residential heating is common in several European countries. The summertime footprint is a 478 subsection of the wintertime footprint that covers an area well-known for severe wildfires and 479 agricultural fires (Stohl et al., 2007 and Yttri et al., 2007a), and thus agrees with previous studies. 480 Further, Sciare et al. (2008) point to the European countries bordering the Black Sea as having high 481 carbonaceous aerosol of fossil origin. Low  $eBC_{BB}$  and  $eBC_{FF}$  levels at Birkenes are consistent with air 482 masses that have an oceanic or terrestrial origin at high latitudes, mainly from the Arctic. Notably, the 483 30% highest values explain 74% of  $eBC_{BB}$  at Birkenes for the actual period, hence LRT is decisive not 484 only for episodes of high concentrations but also largely explains the mean concentration. All  $eBC_{BB}$ 485 and  $eBC_{FF}$  observations included in the 70<sup>th</sup> percentile was made in winter despite the less pronounced 486 seasonality of  $eBC_{FF}$  compared to  $eBC_{BB}$ .

487 To generate a longer BB time series of  $OC_{BB}$  and  $EC_{BB}$  we combine the levoglucosan time series 488 (2008–2018) with levoglucosan/OC and levoglucosan/EC ratios derived from the BB factor of the PMF 489 analysis (Table 1; See Sect. S 5 for details). Depletion of levoglucosan by OH oxidation is more likely 490 in summer (Hoffmann et al., 2010; Yttri et al., 2014), still we assume that levels mostly reflect biomass 491 burning emissions in all seasons.

EC<sub>BB</sub> levels were elevated in the heating season (Figure 9; Table S 16). A strong temperature 492 influence is illustrated by a 9°C difference in the 25<sup>th</sup> percentile of wintertime temperatures in 2015 (-493 494 0.3°C) and 2010 (-9.3°C) (Figure S 1), which experienced the lowest (19 ng m<sup>-3</sup>) and the highest (84 ng 495 m<sup>-3</sup>) winter-time mean concentration of EC<sub>BB</sub>, respectively. Winter 2010 was exceptionally cold due to a negative North Atlantic Oscillation, and the only occasion when EC<sub>BB</sub> exceeded EC<sub>FF</sub>, with annual 496 497 mean EC<sub>BB</sub>>60% higher than the long-term mean. Pronounced interannual variability was seen for the 498 wood burning contribution in winter, from 21-60% to EC, with the lowest fractions occasionally 499 matched by those in spring and fall, typically ranging between 20–30%. EC<sub>BB</sub>/EC was small in summer 500 (4–15%), considerably less than other seasons, except in 2008, where we calculate a substantial 30–40% 501 contribution. Levoglucosan cannot be used to differentiate emissions from residential wood burning, 502 wildfires and agricultural fires; exceptions are major wildfire and agricultural fire episodes identifiable 503 by unusual high concentrations and traced by source receptor models/satellite data for plumes/burnt 504 areas (Yttri et al., 2007a, Stohl et al., 2007). Influence from major wildfires in Eastern Europe caused a 505 summertime peak in fine OC and EC in 2002 at Birkenes (Yttri et al., 2007a). In June 2008, the largest wildfire in Norway since the Second World War raged 25 km northeast of the Birkenes Observatory, 506 with an area of 30 km<sup>2</sup> burnt. The observatory was downwind of the fire on only one day, according to 507 508 FLEXPART (Figure S 2). Despite this, the levoglucosan concentration for the weekly filter sample was 153 ng m<sup>-3</sup>, by far the highest in one decade of sampling. Notably, the annual mean concentration of 509 510 levoglucosan for 2008 increased by nearly 35% and EC<sub>bb</sub> contributed significantly to EC for summer 511 2008.

The seasonality of  $OC_{BB}$  (Figure 9) was like  $EC_{bb}$ . Mean wintertime  $OC_{BB}/OC$  was 39–40% and >50% in 2010 and 2012, considering both  $PM_{10}$  and  $PM_{2.5}$ . The summertime contribution was typically <5%, reflecting both low levoglucosan levels and major influences from BSOA and PBAP, which peak in summer. Notably, five of the seven highest weekly OC concentrations for the  $PM_{10}$  time-series were attributed to emissions from major wildfires in Eastern Europe, i.e., August 2002, and May/September 2006, and thus prior to the initiation of the levoglucosan time series. The local wildfire episode in summer 2008 caused a substantial increase in OC<sub>bb</sub>/OC (13–18%), which is within the lower range of that observed for spring (12–27%) and fall (13–39%).

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#### 521 3.2.2 Biogenic carbonaceous aerosol sources

The general lack of PBAP tracers in the MIN (<1%) and SS (<2%) factors and no sea salt and Ti in the 522 523 PBAP factor, implies that soil and sea spray aerosol do not contribute to PBAP at Birkenes, although 524 this has been shown elsewhere (O'Dowd et al., 2004; Jia and Fraser, 2011). PBAP represented by 525 glucose, arabitol and mannitol appears to be associated with leaves rather than soil material and to be a 526 source of local origin (Samaké et al., 2019). However, even large PBAP, such as birch pollen (avg. diam. 527 22 µm), has a potential for long range atmospheric transport of 1000 km due to its low density, 528 hydrophobic nature, release during favourable dispersion conditions, and (often) emission height > 10529 m (e.g. Sofiev et al., 2006; Skjøth et al., 2007).

530 The PBAP factor concentration was nearly one order of magnitude higher in summer and fall 531 than in winter and was the major contributor to coarse OC for all seasons except winter, particularly in 532 summer (54%) and fall (69%) (Figure 8). These are conservative estimates, as 3–9% of the PBAP tracers 533 reside in the BSOA<sub>SRT</sub> factor, likely due to co-variability, as there is no scientific evidence linking biologically formed sugars and sugar-alcohols to abiotic formation of BSOA. Notably, the PBAP factor 534 535 explained 20–26% of fine OC in summer and fall, being the major contributor in fall. Consequently, PBAP was the major contributor to OC even in PM<sub>10</sub> in summer (31%) and fall (40%). The PBAP factor 536 even explained 16% of fine OC (Figure 8) annually, corresponding to 0.084 µg C m<sup>-3</sup>, which is 537 538 marginally lower than the factor's content of coarse OC (0.113 µg C m<sup>-3</sup>). Combined, this made PBAP 539 the most abundant contributor to OC in PM<sub>10</sub> along with the MIN factor (both 26%). Some PBAP tracers 540 partly reside in the fine mode (Carvalho et al., 2003; Yttri et al., 2007b) but the 43% OC<sub>PBAP</sub> found in 541 the fine fraction in the present study is higher than what has previously been reported for the actual 542 PBAP tracers at Birkenes; i.e. 6–7% (arabitol and mannitol), 20% (trehalose), and 33% (glucose) (Yttri 543 et al., 2007b). It cannot be excluded that the PBAP factor contains some fine OC from other sources e.g. 544 due to condensation, but although there is a seasonal co-variability with the BSOA<sub>SRT</sub> factor, only 2– 545 3% of the 2-methyltetrols were explained by the PBAP factor and there was a low correlation between the PBAP and the BSOA<sub>SRT</sub>, which questions this hypothesis. 546

547 Arabitol and mannitol are well-known tracers of fungal spores (Bauer et al., 2008), one of the 548 most abundant sources of PBAP (Elbert et al., 2007). Applying an OC to mannitol ratio of 5.2-10.8 for 549 fungal spores (Bauer et al., 2008; Yttri et al., 2011a), we estimate that 11-22% of OC<sub>PBAP</sub> (in PM<sub>10</sub>) 550 comes from this source. Glucose is one of the primary molecular energy sources for plants and animals, 551 a building block of natural dimers and polymers (e.g. sucrose and cellulose), and thus ubiquitous in 552 nature and considered a PBAP tracer of general character, and clearly important for allocation of carbon 553 mass to PBAP. Nevertheless, a wider range of organic tracers ought to be tested in future PMF studies 554 to explore the potential of further separation of the highly heterogenic PBAP source, including cellulose, 555 but also amino acids. A greater diversity of PBAP tracers may also provide a more correct PBAP 556 estimate. The PMF approach used in the present study gives a somewhat higher, but overlapping, 557 estimate of OC<sub>PBAP</sub> at Birkenes for August 2016–2018 than Latin Hypercube sampling (LHS) for August 558 2009 (Yttri et al., 2011b) (Table 3). The LHS approach was based on a priori emission ratios, with 559 uncertainty ranges estimated in a similar way to a Monte Carlo analysis (though less computationally 560 extensive), and considered only the sum of fungal spores and plant debris as OC<sub>PBAP</sub>, based on mannitol 561 (fungal spores) and cellulose (plant debris), whereas the PMF approach may pick other contributing, i.e. 562 co-varying, sources.

563 The 3% EC in the PBAP factor is substantially less than the 16% reported by Waked et al. 564 (2014), which stated that atmospheric mixing, PMF limitations and artifacts caused by thermal-optical 565 analysis could be plausible explanations. In the present study, low levels of coarse fraction EC occasionally appear in summer and fall (Table S 5), following the seasonality of PBAP. This finding 566 does not exclude any of the three possibilities proposed by Waked et al., (2014), but supports the 567 suggestion by Dusek et al. (2017) that PBAP, or at least some types of PBAP, chars and evolves as 568 569 modern carbon EC during thermal-optical analysis. If EC<sub>PBAP</sub> indeed is an analytical artefact, then 570 constraining the PBAP factor to contain no EC, as suggested by Weber et al. (2019), should be done 571 with caution, as it will wrongfully apportion pyrolytic carbon generated from PBAP as EC to another 572 source. Thus, EC<sub>PBAP</sub> should rather be interpreted as OC<sub>PBAP</sub>. With no EC<sub>SS</sub>, no EC<sub>BSOA,SRT</sub> and EC<sub>PBAP</sub> an assumed analytical artefact, EC can be apportioned into a fossil fuel category (ECFF), consisting of 573 574 the MIN, TRA/IND, and NH<sub>4</sub>NO<sub>3</sub> factors (explains 0.2% of levoglucosan), and a non-fossil biomass 575 burning category ( $EC_{BB}$ ), the BB factor. Some EC has been reported from meat cooking (Rogge et al., 576 1991), which is a non-fossil source, but its influence is minor at Birkenes, as it has not been observed 577 based on concurrent ACSM-measurements and is not accounted for by levoglucosan.

578 Our PMF results support the use of OC<sub>PM10-2.5</sub> as a proxy of OC<sub>PBAP</sub>, which has a pronounced 579 seasonality (Figure 4) with the highest seasonal mean concentration observed in summer for 15 of the 580 studied years and in fall for the three others (Table S 5). The seasonal mean exceeded 0.5  $\mu$ g C m<sup>-3</sup> on two occasions only; fall 2005 and fall 2006. With a few exceptions, OC<sub>PM10-2.5</sub> contributed more than 581 582 30% to  $OC_{PM10}$  in summer and fall. The highest relative contribution (45–50%) to  $OC_{PM10}$  were 583 exclusively observed in fall (2004, 2005, 2006, 2008, 2014, 2017), likely reflecting a combination of 584 high OC<sub>PM10-2.5</sub> concentrations and fine fraction OC<sub>BSOA</sub> declining at this time of the year. OC<sub>PM10-2.5</sub> 585 made a substantially lower contribution to OC<sub>PM10</sub> in winter (mean: 13%) and in spring (mean: 19%) compared to summer and fall, although contributions exceeding 25% were observed in spring for certain 586 587 years. Notably however, the PBAP factor explains 16% of fine OC, which would not be accounted for 588 using coarse OC as a proxy of  $OC_{PBAP}$ .

These numbers suggest that PBAP is a major, continuous contributor to OC in PM<sub>10</sub> at Birkenes for a period of nearly two decades, and that it largely explains the seasonality. Estimates of PBAP levels in the continental European rural background environment are largely lacking and should be undertaken to explore PBAPs potential importance. With a longer vegetative season and a different climate, the PBAP flux might be larger in more southerly countries, although the relative contribution might be lower due higher overall OC levels. Waked et al. (2014) found that  $OC_{PBAP}$  accounted for 17% of OC in PM<sub>10</sub> on an annual basis for an urban background site in Lens (Northern France), and between 5–6% in winter/spring and 27–37% in summer/fall using PMF for source apportionment. These fractions are comparable to those observed in the present study, albeit concentrations calculated by Waked et al. (2014) were higher.

599 PBAP is a large OC source not included in many models. OC model closure, both for overall 600 levels and seasonality, would thus likely be improved in many cases by its inclusion. This appears to be 601 particularly important for regions with low anthropogenic influence. Birkenes is situated in the Boreo-602 nemorale zone, a transition zone of the Nemorale and the Boreal zone, hence, findings made for this site 603 likely gives an indication of what can be expected for this scarcely populated, circumpolar region, which 604 by far is the largest terrestrial biome of the Northern Hemisphere. Hence, measurements in unperturbed 605 areas should include PBAP for a better understanding of background conditions. In turn, such 606 measurements may improve e.g. climate models; i.e., the aerosol climate effect under relatively clean 607 conditions.

608 Modelled estimates suggest a 10-40% contribution of BSOA to fine OC annually at Birkenes 609 (Simpson et al., 2007; Bergström et al., 2012). Hence, the 9% contribution of OC<sub>BSOA,SRT</sub> to fine OC, 610 and the 13% contribution to coarse OC (10% to OC in PM<sub>10</sub>) found in the present study by PMF, appears 611 to be in the lower range. Further, 3–9% of the PBAP tracers reside in the BSOA<sub>SRT</sub> factor, hence some 612 of its OC content may rather be attributed to PBAP, further lowering the OC content of the BSOA<sub>SRT</sub> 613 factor but strengthening coarse OC as a proxy of PBAP. BSOA<sub>SRT</sub> made a negligible contribution to 614 fine, coarse and PM<sub>10</sub> OC in all seasons, except in summer (22–25%), apparently contradicting previous 615 studies that unambiguously point to BSOA as the major carbonaceous aerosol source at Birkenes in the 616 vegetative season (Simpson et al., 2007; Yttri et al., 2011b). Note that a prevailing BSOA source in 617 summer is considered a normal situation also for European rural background environment in general 618 (e.g. Gelenscer et al., 2007), not only for Birkenes. Table 3 shows that OC<sub>BSOA.SRT</sub> obtained by PMF for 619 August 2016–2018 is substantially lower than that obtained by LHS for August 2009 (Yttri et al., 620 2011b). Although not obtained for the same year, we argue that methodology rather than climatology 621 explains most of the difference. OC<sub>BSOA,LHS</sub> provides an upper estimate including all modern carbon, 622 local and from LRT (excluding biomass burning and PBAP fungal spores and plant debris), whereas 623 OC<sub>BSOA,SRT</sub> gives a lower estimate accounting for locally formed BSOA.

It is less likely that anthropogenic secondary organic aerosol (ASOA) resides in the BSOA<sub>SRT</sub>factor, as ASOA precursors result from combustion processes and evaporative losses. Further, Yttri et al. (2011a) found higher ASOA concentrations in the Norwegian rural background environment in winter compared to summer, which is opposite of  $BSOA_{SRT}$ , hence co-variation and/or apportionment to the same factor do not appear likely. ASOA is less abundant than BSOA at Birkenes, as calculated by Simpson et al. (2007) and Bergström et al. (2012), but the estimates vary substantially and are very uncertain (Spracklen et al., 2011), particularly for ASOA (from 1% to 10–20%). It is difficult to predict which PMF factor(s) accounted for ASOA, but for the sake of separating OC into a natural and an anthropogenic fraction we assume that ASOA is not part of neither BSOA<sub>SRT</sub> nor the PBAP factor, which we consider as exclusively natural factors. To provide an upper estimate of the natural sources (Figure 8), we neither consider it part of the MIN factor.

- 635 With 90% (in PM<sub>10</sub>) and 92% (in PM<sub>2.5</sub>) of the MIN factor's OC content attributed to LRT BSOA (OC<sub>BSOA,LRT</sub>) (See Sect. 2.4.1), the combined contribution of locally formed BSOA (OC<sub>BSOA,SRT</sub>) 636 637 and OC<sub>BSOA,LRT</sub> to OC in PM<sub>10</sub> and PM<sub>2.5</sub> would be 34-38% on an annual basis, 37-41% in spring and 638 50-57% in summer. From this we can deduct that 1/3 of BSOA is of local origin, whereas 2/3 are longrange transported. For August 2016–2018, the joint contribution of OC<sub>BSOA,SRT</sub> and OC<sub>BSOA,LRT</sub> to OC in 639 PM<sub>10</sub> is 31%, corresponding better with the LHS estimate (Table 3) but still noticeably lower. Notably, 640 OC<sub>BSOA,SRT</sub>, OC<sub>BSOA,LRT</sub> and OC<sub>PBAP</sub> combined contributed 79% to OC in PM<sub>10</sub> in August 2016–2018, 641 642 which exactly matches the sum of OC<sub>BSOA,LHS</sub> and OC<sub>PBAP,LHS</sub> to OC in PM<sub>10</sub> in August 2009. This 643 suggests that LHS and PMF apportion an equally large amount of OC to natural sources but that the 644 split between BSOA and PBAP likely differs. It is evident that the LHS-approach provides an upper 645 estimate of BSOA (Gelenscer et al., 2007; Yttri et al. 2011a), whereas the great diversity of PBAP likely 646 is underestimated by just accounting for plant debris and fungal spores. The lower estimate of OC<sub>BSOA</sub> and the higher estimate of OCPBAP provided by PMF in the present study is in line with this and 647 648 encourage further effort to apportion these major carbonaceous aerosol sources correctly. Inclusion of 649 monoterpene and sesquiterpene oxidation products (Kleindienst et al., 2007) to PMF would possibly 650 improve our understanding of the SOA apportionment, as would knowledge about their atmospheric 651 lifetime.
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#### 653 4. Conclusions

The carbonaceous aerosol time-series at the Birkenes Observatory initiated in 2001 is unique due to its unprecedented length in Europe and because measurements are performed both for  $PM_{10}$  and  $PM_{2.5}$ . Such long-time series are of utmost importance, e.g. for the evaluation of projections, air-quality models, and climate models. The need for concurrent and diverse off-line and on-line carbonaceous aerosol speciation measurements for understanding of carbonaceous aerosol sources, seasonal, annual, and longterm variability has been utterly demonstrated.

660 Statistically significant and comparably large reductions ( $\sim -4\%$  yr<sup>-1</sup>) were calculated for EC 661 and PM<sub>2.5</sub> at the Birkenes Observatory for 2001–2018, with EC reductions largely attributed to road 662 transportation. No significant declining trend was calculated for OC, likely because prevailing natural 663 sources masked any reduction in anthropogenic sources. Further reduction of carbonaceous aerosol may 664 be hampered by poorly abated sources such as domestic heating, though more work is needed to assess 665 this. We emphasize the importance of establishing long lasting, high quality carbonaceous aerosol and

- organic tracers time series at several sites across Europe for this purpose. The OC fraction of PM<sub>10</sub> 666
- (+2.3% yr<sup>-1</sup>) and PM<sub>2.5</sub> (+3.2% yr<sup>-1</sup>) increased significantly from 2001–2018, whereas the EC fraction 667

- decreased  $(-4.0 -4.7\% \text{ yr}^{-1})$ , causing a successive change in the aerosol chemical composition and in 668

669 the relative source composition.

- 670 Source apportionment using PMF identified seven factors, six of which were carbonaceous 671 dominated: Mineral dust dominated (MIN), traffic/industrial-like (TRA/IND), biogenic secondary 672 organic aerosol (BSOA), primary biological aerosol particles (PBAP), biomass burning (BB) and 673 ammonium nitrate dominated (NH<sub>4</sub>NO<sub>3</sub>). Carbonaceous material was negligible in the sea salt (SS) 674 factor. Combustion of fossil fuel (78%) was the major source of EC and TRA/IND (50%) the key factor. 675 Emissions from residential wood burning increased in the heating season but fossil fuel sources 676 dominated EC even in winter (66%). Continental Europe and western parts of Russia were the main source regions of elevated levels of eBC, both for biomass burning and for combustion of fossil fuels. 677 678 Natural sources dominated both fine (53%) and coarse (78%) fraction OC, thus also OC in  $PM_{10}$  (60%). 679 The natural fraction increased substantially in the vegetative season due to biogenic secondary organic 680 aerosol and primary biological aerosol particles, confined to the BSOA, PBAP and MIN factors. 77-91% of OC was attributed to natural sources in summer and 22-37% in winter. The coarse fraction 681 682 showed the highest share of natural sources regardless of season and was dominated by PBAP, except 683 in winter. Notably, PBAP (26%) made a larger contribution to OC in PM<sub>10</sub> than BB (14%), and an equally large contribution as BB (17%) in PM<sub>2.5</sub>. 684
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#### 686 **Author contribution**

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#### 697 Acknowledgements

Time series used in the present study, except for the organic tracers, were obtained as part of the 698 699 Norwegian national monitoring program (Aas et al., 2020). The monosaccharide anhydrides, the sugar-700 alcohols and the 2-methyltetrols (organic tracers) time series were funded by the Norwegian Research 701 Council through the Strategic Institute Projects "Observation and Modelling Capacities for Northern 702 and Polar Climate and Pollution" and the "Studying sources, formation and transport of short-lived *climate forcers by advanced high-time resolution measurements*". All data are reported to the EMEP
monitoring programme (Tørseth et al., 2012) and are available from the database infrastructure EBAS
(http://ebas.nilu.no/) hosted at NILU.

The research leading to these results has benefited from Aerosols, Clouds, and Trace gases Research InfraStructure (ACTRIS), funding from the European Union Seventh Framework Programme (FP7/2007–2013) under ACTRIS-2 and the grant agreement no. 262254, and the COST Action CA16109, Chemical On-Line cOmpoSition and Source Apportionment of fine aerosol-COLOSSAL; i.e., for participation in interlaboratory comparison for thermal-optical analysis and QA/QC of

711 measurements.

OC/EC and mass concentration were measured as part of the Norwegian national monitoring
programme (Aas et al., 2020), whereas monosaccharide anhydrides were analysed as part of the SACC
(Strategic Aerosol Observation and Modelling Capacities for Northern and Polar Climate and Pollution)
and SLCF (Describing sources, formation, and transport of short lived climate forcers using advanced,
novel measurement techniques) projects.

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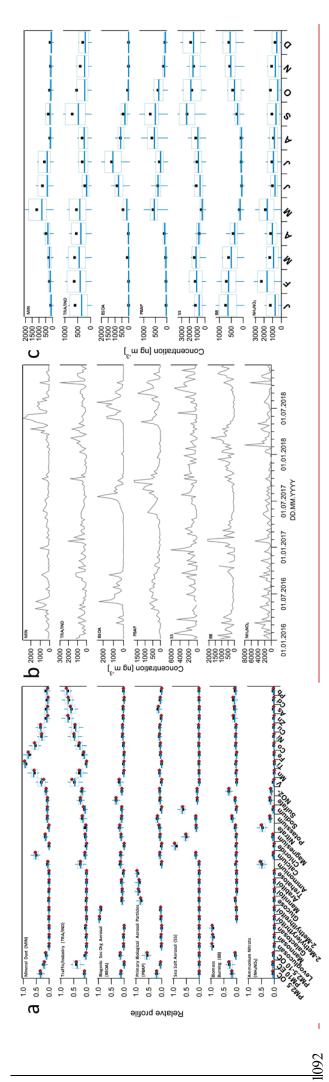
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## 1087 Figures



1088

1089Figure 1: The Birkenes Observatory (58°23' N, 8°15' E; 219 m asl) lies in the Boreo-nemoral zone, 20 km from the1090Skagerrak coastline in Southern Norway.



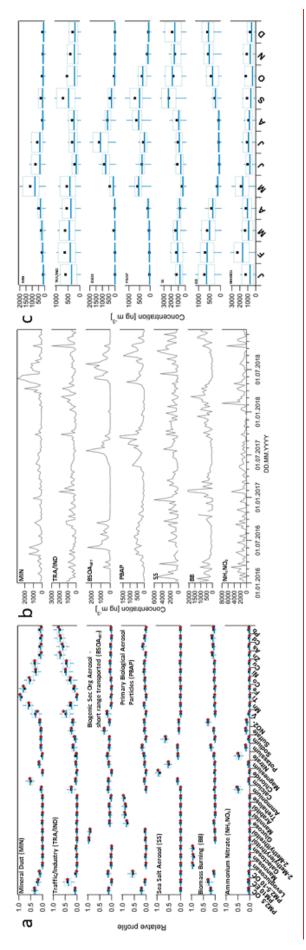
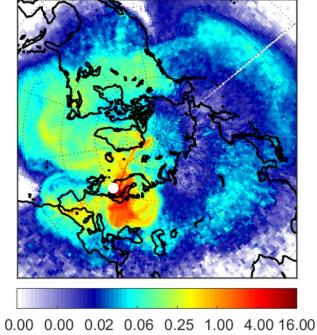
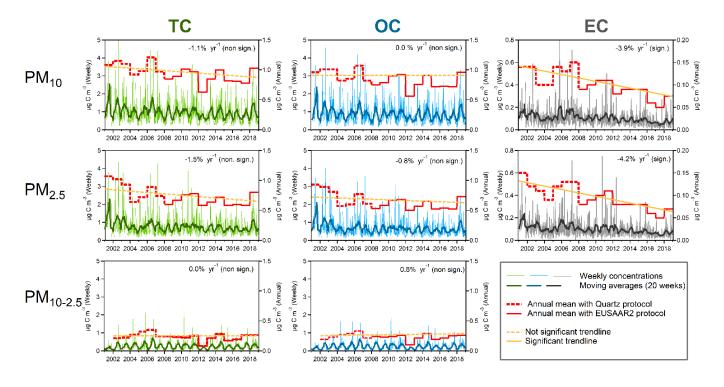


Figure 2: Results from PMF for Birkenes 2016–2018. a) Factor profiles b) Factor time series c) aggregated monthly data. Boxes in a), c) show statistics from bootstrapped solutions (n=5000): percentiles 25/75 (box), median (horizontal line) and 10/90 (whiskers). Black markers are the means. Red markers in a) show the base factor profiles.  $\frac{1093}{1095}$ 

Footprint: 30-May-2018 to 06-Jun-2018



- Averaged emission sensitivity [ns m<sup>-2</sup>] 1097
- Figure 3: Footprint emission sensitivities calculated using the FLEXPART model for the period 30 May–6 June 2018 at the Birkenes Observatory. 1098 1099



1103 Figure 4: Ambient aerosol concentrations of TC, OC and EC in PM10 (Upper panels), in PM2.5 (Mid-Panels), and TC

and OC in PM10-2.5 (Lower panels), presented as weekly (168 h) and annual mean concentrations for the Birkenes Observatory for 2001–2018. The trendlines account for the protocol shift.

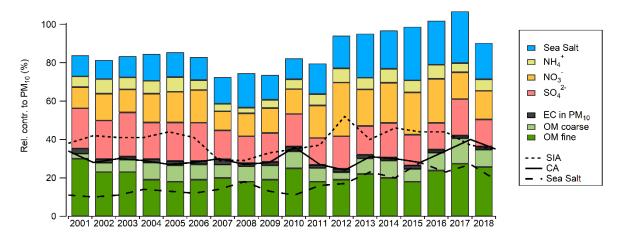
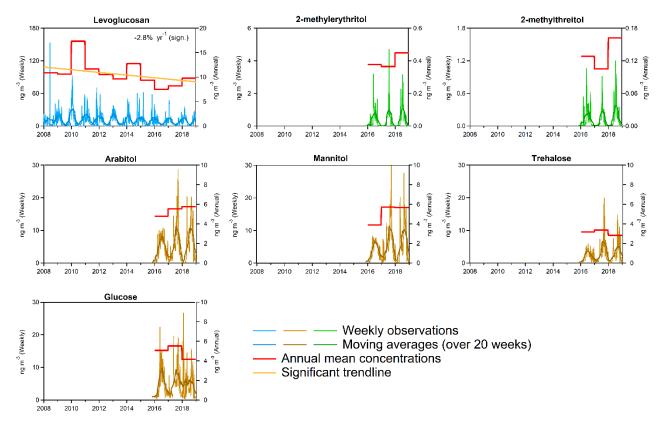




Figure 5: Mass closure of PM<sub>10</sub> for Birkenes for the period 2001–2018 (Unit: %). Notation: Sea salt = Sum of Na<sup>+</sup>, Mg<sup>2+</sup>,

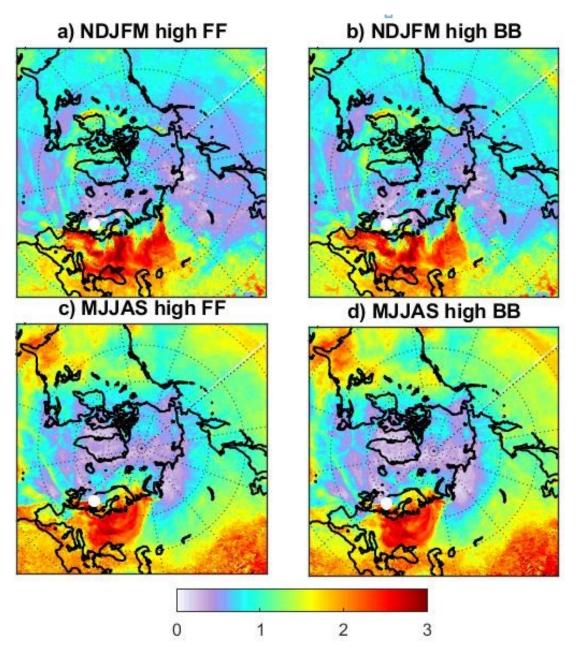
1107 1108 1109 Cl<sup>+</sup>; SIA = Secondary inorganic aerosol (SIA) = Sum of SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>; NH<sub>4</sub><sup>+</sup>; CA = Carbonaceous aerosol; OM = Organic matter. OM is calculated using OC:OM=1.9 (Yttri et al., 2011a).



1111

1112 Figure 6: Ambient aerosol concentrations of organic tracers in PM<sub>10</sub>. Levoglucosan, 2-methylerythritol and 2-11113 methylthreitol (Upper panels), arabitol, mannitol, trehalose (Mid-Panels), and glucose (Lower panel), presented as

1114 weekly (168 h) and annual mean concentrations for the Birkenes Observatory for the period 2008–2018.



1116 1117 1118 1119 Figure 7: 70th percentiles of eBCff (left panels, a and c) and eBCbb (right panels, b and d) for winter (NDJFM) and summer (MJJAS). The color-coding shows the ratio of residence times for footprint sensitivities during measurements

- exceeding the 70th percentile and the average footprint sensitivity.
- 1120

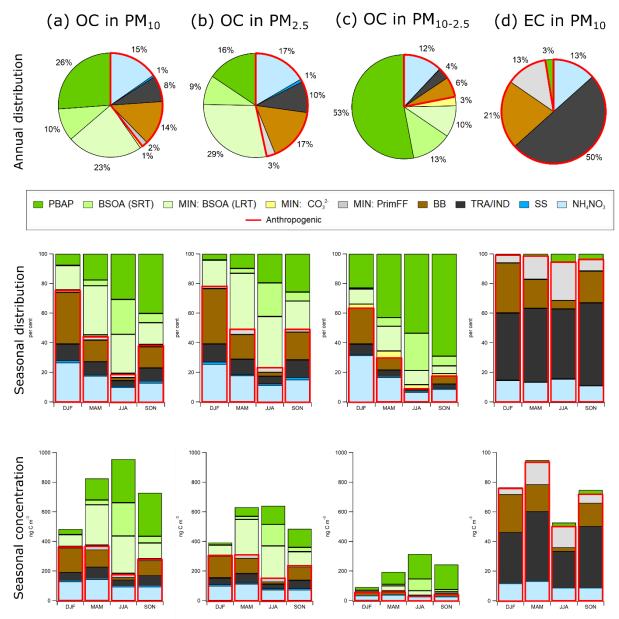
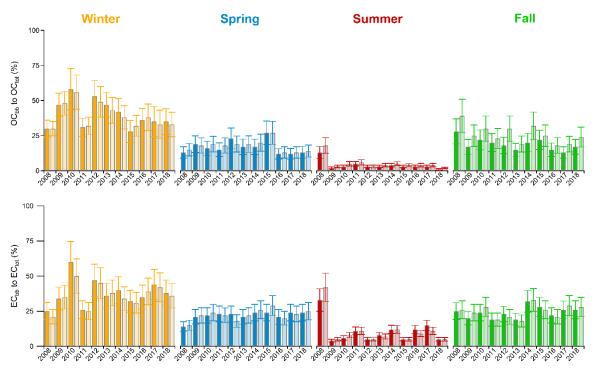


Figure 8: Factor contributions to OC in PM<sub>10</sub> (a), PM<sub>2.5</sub> (b), PM<sub>10-2.5</sub> (c), and EC in PM<sub>10</sub> (d) at Birkenes (2016–2018) (upper panels), and divided into seasons (middle and lower panels), as determined by positive matrix factorization. The factors enclosed by the full red line represents anthropogenic sources. The OC content of the MIN factor is divided into long range transported BSOA (OC<sub>BSOA,LRT</sub>) and primary OC from fossil fuel combustion (OC<sub>PrimFF</sub>) following Eq. (1), and carbonate carbon (OC<sub>CO32-</sub>) (Sect. 2.4.1).



1129 Figure 9: Relative contribution of  $OC_{bb}$  to  $OC_{tot}$  (upper panel) and  $EC_{bb}$  to  $EC_{tot}$  (lower panel) in PM<sub>10</sub> (dark colors) and in PM<sub>2.5</sub> (light colors), as a function of season at Birkenes for 2008–2018 (DJF = Winter; MAM = Spring; JJA =

- 1131 Summer; SON = Fall).

#### Tables

### 1136 Table 1: Variables describing the biomass burning source derived from the PMF BB factor in the present study, and comparable variables obtained by <sup>14</sup>C-analysis reported by Zotter et al. (2014).

Present study	Zotter et al. (2014) <sup>1</sup>
$OC/Levoglucosan$ (in $PM_{10}$ ) = 12.7	$OC_{NF}/Levoglucosan$ (in $PM_{10}$ ) = 12.6 ± 3.1
$OC/Levoglucosan$ (in $PM_{2.5}$ ) = 11.1	
EC/Levoglucosan (in $PM_{10}$ ) = 1.96	$EC_{NF}/Levoglucosan$ (in $PM_{10}$ ) = 1.72 ± 0.59
OC/EC (in $PM_{10}$ ) = 6.5	$OC/_{NF}EC_{NF}$ (in $PM_{10}$ ) = 7.7 ± 2.1
$OC/EC$ (in $PM_{2.5}$ ) = 5.7	

1138 1139 <sup>1</sup>North of the Alps Notation:  $OC_{NF} = Non-fossil OC$ ;  $EC_{NF} = Non-fossil EC$ 

Table 2: Biomass burning fraction derived from the PMF and the aethalometer model. Aethalometer model 1 shows the biomass burning fraction obtained by the default pair of Absorption Ångstrøm Exponents (AAE) suggested by

Zotter et al. (2014), whereas Aethalometer model 2 show the biomass burning fraction obtained using the pair of AAEs

1141 1142 1143 1144 derived from PMF.

	PMF	Aethalometer model 1	Aethalometer model 2
Biomass burning fraction	0.27	0.48	0.28
Fossil AAE	0.93	0.9 (Zotter et al., 2014)	0.93 (from PMF)
Biomass burning AAE	2.04	1.68 (Zotter et al., 2014)	2.04 (from PMF)

Table 3: OC<sub>BSOA</sub>, short-range transported (SRT) and long-range transported (LRT), and OC<sub>PBAP</sub> concentrations and their relative contribution to OC in PM<sub>10</sub> at Birkenes in August, as obtained by Latin Hypercube Sampling (Yttri et al., 2011b) and by PMF (present study). 1146 1147

	Reference	Approach	Approach OC <sub>BSOA.SRT</sub> (ng C m <sup>-3</sup> )	OC <sub>BSOA.SRT</sub> /OC	OC <sub>BSOA.SRT+LRT</sub> (ng C m <sup>-3</sup> )	OC <sub>BSOA,SRT+LRT</sub> /OC	OC <sub>PBAP</sub> (ng C m <sup>-3</sup> )	OC <sub>PBAP</sub> /OC
August 2009	Yttri et al. (2011b)	$\Gamma HS^3$			$505^{1} (408 - 598)^{2,3}$		$290^{1} (213 - 380)^{2}$	$0.31^{1} (0.22 - 0.40)^{2}$
August 2016	Present study	PMF	115	0.19	173	0.28	318	0.52
August 2017	Present study	PMF	183	0.19	252	0.26	553	0.57
August 2018	Present study	PMF	159	0.20	316	0.40	287	0.36
August 2016–2018	Present study	PMF	152	0.19	247	0.31	386	0.48

50<sup>th</sup> percentile 10<sup>th</sup>-90<sup>th</sup> percentile LHS-approach includes both OC<sub>BSOA.SRT</sub> and OC<sub>BSOA.LRT</sub>

#### 1 SUPPLEMENT TO

- 2 Trends, composition and sources of carbonaceous aerosol in the last 18 years at the Birkenes
  3 Observatory, Northern Europe, by K. E. Yttri et al.
- 4

#### 5 S1. Quality assurance

6 The OC/EC data are not field blank corrected, in accordance with the standard operating procedure 7 provided by EMEP (Yttri et al., 2007a; EMEP, 2014). The positive sampling artefact of OC for weekly 8 samples collected at Birkenes has been quantified on a campaign basis using the QBQ (Quartz fibre 9 filter Behind Quartz fibre filter) approach (McDow and Huntzicker, 1990; Turpin et al., 1994) in summer (18±4%; Yttri et al., 2011b), fall (19±7%; Yttri et al., 2019), and winter/spring (24±13%; Yttri et al., 10 11 2019) but only for  $PM_{10}$ . For OC in  $PM_{2.5}$ , which at Birkenes is obtained from an identical and co-12 located sampler, operating at the same filter face velocity as the  $PM_{10}$  sampler, the positive sampling 13 artefact is considered equally large, whereas its relative importance is slightly higher. The negative 14 sampling artefact has not been addressed.

OC/EC analysis was performed within 2 months after the filter samples were collected and according to the Quartz (2001–2008) and the EUSAAR-2 (from 2008) temperature programs. EUSAAR-2 is designed to reduce the inherited uncertainties associated with splitting of OC and EC, e.g. by preventing premature burn-off of EC (Cavalli et al., 2010). The uncertainty associated with repeated OC/EC analyzes of a filter sample is typically <10%, which includes both analytical uncertainty and heterogenic distribution of the deposited aerosol particles on the filter sample.

21 The laser's ability to detect changes in the transmittance of a filter sample high in initial EC is 22 crucial to obtain a correct value for EC (and OC). 15 µg EC cm<sup>-2</sup> has been suggested as an upper limit (Subramanian et al., 2004; Wallén et al., 2010) but this value is likely to vary. The nine filter samples 23 24 (out of nearly 1800) with an EC content exceeding 15 µg C cm<sup>-2</sup> in the current dataset were considered 25 valid. Further, a non-biased separation between OC and EC requires that either pyrolytic carbon (PC) 26 evolves before EC during analysis or that PC and EC have the same light absorption coefficient. It is 27 well known that this is not always the case (Yang and Yu, 2002) and there is a lack of information on 28 the magnitude of this imperfection.

Deviation from the protocol-defined temperature steps will affect the analysis results of the TOA instrument (Chow et al., 2005; Panteliadis et al., 2015) and temperature offsets ranging from -93 °C to +100 °C per temperature step have been reported (Panteliadis et al., 2015). Thus, calibration by the temperature calibration kit available from the instrument manufacturer (Sunset laboratory Inc) since 2012 is strongly recommended. Temperature calibration was implemented as part of the regular QA/QC procedures for thermal-optical analysis in 2013.

A comparison of the two temperature programmes used for the Birkenes time series was performed for PM<sub>2.5</sub> filter samples collected at Birkenes in 2014, using temperature calibrated versions of both Quartz and EUSAAR-2. There was a good agreement between the two temperature programs for TC and OC, i.e. close to the expected uncertainty associated with analysis and sampling, whereas for EC the difference was pronounced (Table S 17), although in close correspondence with that previously reported by Panteliadis et al. (2015). Note that OC and EC data for the period 2001–2007 discussed in the main are text not corrected according to Eq. (S 18–20) (Table S 17), except for the purpose of trend calculations.

Field blanks did not contain monosaccharide anhydrides, sugars, sugar-alcohols or 2methyltetrols in noticeable amounts. Filter samples for which the content was below the limit of detection (LOD) but > 0, were considered valid and included when calculating the annual and seasonal means. Organic tracers were analyzed within 1 year after collection of the aerosol filter samples. The uncertainty (analytical and sampling uncertainty) associated with measurements of monosaccharide anhydrides is within 10 - 15 % (Yttri et al., 2015). A similar range of uncertainty is expected for the other organic tracers.

50 Mass concentrations of  $PM_{10}$  and  $PM_{2.5}$  were field blank corrected. The overall uncertainty 51 associated with determination of the  $PM_{10}$  and  $PM_{2.5}$  mass concentration is < 5%. The monitoring of 52 major ions and trace elements follows the guidelines by EMEP (EMEP, 2014) and are within the data 53 quality objective of the network: 15–25% uncertainty for the combined sampling and analysis of major 54 ions and 30% for heavy metals.

- 55
- 56

#### S2. Calculation of trends - Statistical approach

The Mann-Kendall test (Mann, 1945; Kendall, 1975; Gilbert, 1987) was used for calculating the significance of the trend and if a significant trend was found, the Theil-Sen slope (Theil, 1950; Sen, 1968; Gilbert, 1987) was calculated. This procedure has been widely used in atmospheric science, like in the recent TOAR project analysing global surface ozone trends (e.g. Fleming et al., 2018; Lefohn et al., 2018), in the review of the EMEP observations (Tørseth et al., 2012) and in numerous other observation based papers (Aas et al. 2019; Ciarelli et al., 2019; Theobald et al., 2019; Masiol et al., 2019; Collaud Coen et al., 2020).

The Mann-Kendall test is a non-parametric test that does not rely on any assumptions of distribution and is therefore well suited for atmospheric data that often deviate from normality and contain outliers that would hamper a standard linear regression. The basics of the Mann-Kendall test is to count the signs of all forward concentration differences in time, and if there is a sufficient overweight of positive or negative differences, the 0-hypothesis ( $H_0$ ) of no trend could be rejected. The S statistic given below contains the sum of all the signs based on the observed values  $y_i$  at time *i*:

70

71 
$$S = \sum_{i=1}^{n-1} \sum_{j=i+1}^{n} sign(y_j - y_i)$$
 Eq.S1

This statistic together with the number of samples and the number of ties in the data were used to calculate the *p* value as given by Gilbert (1987). In our work, we assumed significant trends when p < 0.05.

With  $p < 0.05 H_0$  was rejected and the value of the trend was estimated by the Theil-Sen slope estimator:

78 
$$\beta = median\left(\frac{y_j - y_i}{t_j - t_i}\right), \quad j > i$$
 Eq.S2

79

80 where  $t_i$  denotes the time *i* of the observed value  $y_i$ .

81 The Theil-Sen slope is simply the median of all the forward concentration gradients. In addition 82 to the slope, the  $2\sigma$  confidence intervals were calculated according to Gilbert (1987), providing the 95 83 % confidence range of the slopes.

The Mann-Kendall test and Theil-Sen slope estimation were applied to all species and ratios discussed in this work. These calculations were based on the seasonal and annual mean values, separately, as presented below. For the ratios, r = x/y (e.g. the fraction of NO<sub>3</sub><sup>-</sup> in PM<sub>10</sub>), we based the calculations on the ratios of the seasonal means and not on the seasonal means of the ratios, i.e.:

88

89 
$$r = \frac{x}{y}$$
, where  $x = \frac{1}{n} \sum (x_i)$  and  $y = \frac{1}{n} \sum (y_i)$  Eq. S3

90

For all cases where the 0-hypothesis (H<sub>0</sub>) could be rejected, the Theil-Sen slopes were calculated, and this slope was further transferred into the relative trend by dividing the trend ( $\beta$ ) by the mean of the observed values:

94

95 
$$\beta_{rel} = \frac{\beta}{\left[\frac{1}{n}\Sigma(y_i)\right]}$$
, where  $y_i = observed$  concentration or ratio at time i Eq. S4

96

# 97 S3. Absorption coefficent measurements and source apportionment

98 The absorption coefficient ( $B_{Abs}$ ) was measured using the multi wavelength ( $\lambda$ =370; 470; 520; 590; 660; 99 880; 950 nm) aethalometer (AE33, Magee Scientific), operating behind a PM<sub>10</sub> inlet. We calculate 100 absorption coefficients ( $B_{Abs}$ ) according to Drinovec et al. (2015):

101

102 
$$B_{Abs}(\lambda) = \frac{A \cdot \left(\frac{ATN_{t_2}(\lambda) - ATN_{t_1}(\lambda)}{100}\right)}{Q \cdot C \cdot (1 - \zeta) \cdot \left(1 - k(\lambda) \cdot \left(ATN_{t_2}(\lambda) - ATN_{ref}(\lambda)\right)\right) \cdot (t_2 - t_1)} \qquad Eq. S5$$

103 where ATN = attenuation at time t =1 and t =2, and of the reference spot *ref*, Q is the instrument flow 104 rate on spot 1, A is the filter spot area, k is the loading compensation parameter from the 2 spot 105 compensation algorithm. Here we neglect lateral air flow losses ( $\zeta$ ) and the scattering compensation C since these are not wavelength dependent in Eq. (S5) and hence do not affect source apportionment
based on wavelength dependence, while conversion to eBC via co-located filter measurements of EC
also results in compensation of these parameters using:

110  $eBC(\lambda) = B_{Abs}(\lambda) / \alpha_{effective}(\lambda)$  Eq. S6

111 where  $\alpha_{\text{effective}}$  is an effective mass absorption cross section ( $\alpha$ ) incorporating scattering and lateral flow 112 losses:

113 
$$\alpha_{effective}(\lambda) = \alpha(\lambda) \times c \times (1 - \zeta)$$
 Eq. S7

114 Hence  $\alpha_{effective}$  is a conversion factor between  $B_{Abs}$  and eBC and has no physical meaning beyond 115 this.

116 The AE33 of this study automatically generates  $B_{Abs}(\lambda)$  at 1-minute resolution. However, as 117 discussed by Springston et al. (2007) and Backman et al. (2017), the time interval  $(t_2 - t_1)$  Eq.(S5) can be adjusted to any integer multiple of the base resolution in post-processing. Here we adopt the approach 118 119 of Backman et al. (2017), fixing the time interval to 1 hour and calculating  $B_{Abs}(\lambda)$  according to Eq. 120 (S5). In case one or more filter advances occurred within the one-hour interval, data from each individual 121 filter spot falling within the interval were treated separately and a time-weighted average recorded for 122 that hour. The advantage of this technique is enhanced noise reduction, i.e. using the one-hour interval 123 approach the noise reduction is proportional to as much as l/n (where n are the measurement points), 124 rather than *l/sqrt(n)*, attainable via signal averaging.

125 Here we performed source apportionment of aethalometer data using the *aethalometer model* 126 (Sandradewi et al., 2008). Assuming two sources contribute to total Babs ( $B_{Abs,Tot}$ ), i.e. fossil fuel 127 combustion ( $B_{Abs,ff}$ ) and biomass burning ( $B_{Abs,bb}$ ):

128

109

$$129 \qquad B_{Abs,Tot} = B_{Abs,ff} + B_{Abs,bb} \qquad \qquad Eq. S8$$

130 Then, using a wavelength pair, here  $\lambda_1$ =470 nm and  $\lambda_2$ =880 nm,

131 
$$B_{Abs,bb}(\lambda_2) = \frac{B_{Abs}(\lambda_1) - B_{Abs}(\lambda_2) \cdot \left(\frac{\lambda_1}{\lambda_2}\right)^{-\alpha_{ff}}}{\left(\frac{\lambda_1}{\lambda_2}\right)^{-\alpha_{ff}} - \left(\frac{\lambda_1}{\lambda_2}\right)^{-\alpha_{ff}}} \qquad Eq. \ S9 \text{ and}$$

132 
$$B_{Abs,ff}(\lambda_2) = \frac{B_{Abs}(\lambda_1) - B_{Abs}(\lambda_2) \cdot \left(\frac{\lambda_1}{\lambda_2}\right)^{-\alpha_{bb}}}{\left(\frac{\lambda_1}{\lambda_2}\right)^{-\alpha_{ff}} - \left(\frac{\lambda_1}{\lambda_2}\right)^{-\alpha_{bb}}} \qquad Eq. S10$$

- where  $\alpha_{ff}$  and  $\alpha_{bb}$  are the absorption Ångstrøm exponents (AAE) for fossil fuel and biomass burning, respectively. Note that when using this approach, the AAEs must be assumed *a priori*, while the data are not fitted or error weighted, which can lead to negative values in the resulting time series of the factors due to uncertainty in the AAEs e.g. Grange et al. (2020).
- Here we also used positive matrix factorisation (PMF) to distinguish between the two sources
  in Eq. (S8). The theory of PMF is detailed elsewhere (Paatero and Tapper, 1994) Briefly, a matrix of
  measurement data X is represented by a bilinear model comprising factor profiles F (rows), factor time

- 140 series *G* (columns) and a residual matrix *E*:
- 141

$$142 X = G \cdot F + E Eq. S11$$

144 In PMF factors are found using a least-squares fitting routine in which the object function Q, i.e. the 145 square of residuals e weighted to uncertainty  $\sigma$ , is minimised across all cells (rows *i*-*m*, columns *j*-*n*) 146

147 
$$Q^m = \sum_{i=1}^m \sum_{j=1}^n \left(\frac{e_{ij}}{\sigma_{ij}}\right)^2 \qquad Eq. S12$$

Here, we use the source finder (SoFi, (Canonaco et al., 2013)) toolkit ref, to call PMF (To model the error matrix  $\sigma_{ij}$  we use the clean air test function of the AE33 to determine the standard deviation of the attenuation of the blank  $\delta_{ATN_{air}}$ , calculating  $\sigma_{ij}$ , using:

151

152 
$$\sigma_{ij} = \sqrt{f_A^2 + f_Q^2 + 2\left(\frac{\delta_{ATN_{air}}(\lambda_j)}{ATN_i(\lambda_j)}\right)^2 + \left(\frac{\delta_{ATN_{air}}(\lambda_j)}{ATN_{i-1}(\lambda_j)}\right)^2 + \left(\frac{\delta_{ATN_{air}}(\lambda_j)}{ATN_{ref}(\lambda_J)}\right)^2} \cdot B_{Abs,i}(\lambda_j) \quad Eq.S13$$

153

154 where  $f_A$  and  $f_Q$  are the fractional uncertainties in the spot area and the flow rate, respectively (both 155 0.015 according to Backman et al., 2017). Clean air tests were performed only periodically. Therefore, 156 to generate an error estimate for all time points, we interpolated (bilinear interpolation) between the 157 clean air tests to generate the full error matrix, accounting for drift in  $\delta_{ATN_{air}}$ . Points before and after 158 the last clean air test were calculated using the first and last values of  $\delta_{ATN_{air}}$ , respectively.

159 According to Eq. (S11), X could be represented by any combination of G and F, i.e. the PMF 160 model has *rotational ambiguity*. In practice, many rotations produce negative values and are thus 161 forbidden. Nevertheless, many rotations and local minima in Eq. (S11) are likely to exist. To assess this, 162 we generated multiple (n=2000) bootstrap replacement matrices (block size 24 to conserve diurnal 163 variation if present), running PMF on each matrix 5 times for a total of 10000 runs. PMF settings are 164 shown in Table S 2.

165 We import all 2000 files generated using SoFi for each factor solution. To map the factors, we 166 calculated an effective AAE from the factor profiles  $\alpha_F$ , using

- 167
- 168  $\alpha_F$

169 
$$= -\frac{\log \left(\frac{F_{j=2}}{F_{j=6}}\right)}{\log(\frac{470}{880})}$$
 Eq. S14

170 sorting factors and time series from each run from low to high with respect to  $\alpha_F$ . Binning the effective 171 AAEs from each factor also provides a convenient means to investigate the solution space for rotational ambiguity.

173

#### 174 S4. Positive matrix factorisation applied to filter data

- 175 We performed PMF for samples collected in 2016-2018 (151 samples), with the following as input
- 176 data: OC (in PM<sub>2.5</sub> and PM<sub>10-2.5</sub>), EC (in PM<sub>10</sub>), levoglucosan, mannosan, galactosan, arabitol,
- 177 mannitol, trehalose, glucose, V, Mn, Ti, Fe, Co, Ni, Cu, Zn, As, Cd, and Pb (all in PM<sub>10</sub>), SO<sub>4</sub><sup>2</sup>, NO<sub>3</sub><sup>-</sup>,
- 178  $NH_4^+$ ,  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $K^+$ ,  $Na^+$ ,  $Cl^-$  (open filter face).
- 179 Table S 3 shows miscellaneous settings of the PMF analysis of these data. The input data and error
- 180 estimates were prepared using the procedure suggested by Polissar et al. (1998) and Norris et al.
- 181 (2014), see also Table S 3 for miscellaneous settings including missing data treatment and assessment
- 182 of the PMF performance.
- 183 If the concentration was greater than the LOD, the calculation was based on a user provided184 fraction of the concentration and LOD:

185 
$$Unc = \sqrt{(Error Fraction \times Concentration)^2 + (\frac{1}{2} \times LOD)^2}$$
 Eq.S15

186

187 The analytical uncertainties (20%) as error fraction of OC, EC, organic tracers, ions, and elements 188 were used to determine the corresponding error estimates. Based on given understanding of OC sources, 189 2–10 factors with random seeds were examined, and 7 factors were determined based on: 1) The 190 decrease in Q/Qexp was larger than the relative change in number of factors up to 7; 2) All factors could 191 be interpreted; 3) All factors were distinct.

- To assess the statistical uncertainty in the model we performed repeated analyses on bootstrapresampled matrices. A base profile was generated from a manually mapped average of 50 runs. From each bootstrap run, we fitted all 7 bootstrap factors vs all 7 factors from the base run profile (representing a  $7 \times 7$  matrix of r<sup>2</sup> values). We then mapped the bootstrap factors in order of the r<sup>2</sup> value: The highest value was assumed to be a match, then then the next highest value excluding both previously mapped factors to any other factor (representing a  $6 \times 6$  matrix of r<sup>2</sup> values), and so on. This was to avoid any factors being mapped twice.
- The minimal robust and true Q values of the base run were 5507.9 and 5580.8, respectively. All the (error) scaled residuals were within  $\pm 5$  and > 97.8% within  $\pm 3$ , normally distributed and centred around zero. The average Q/Qexp was 1.2. We also observe no structure in the residuals, which were evenly distributed between measurements from different instruments (i.e. we did not observe factors representing groups of compounds by instrument type, Figure S 3).
- 204

## 205 S5. Emission ratios used to calculate OC and EC from biomass burning

- 206 Emission ratios derived from ambient data are a good alternative to direct emission measurements, 207 accounting for the aggregate effects of fuel type and combustion conditions, but results will nevertheless 208 vary from region to region (e.g. Zotter et al., 2014). Here, we used ratios from our PMF analysis 209 (Table 1) to calculate carbonaceous aerosol from biomass burning for 2008–2018. The levoglucosan to 210 mannosan ratio is rather consistent between seasons, with the values for summer  $(5.1\pm0.9)$  and fall 211  $(5.2\pm0.7)$  being slightly lower than for winter  $(5.4\pm0.8)$  and spring  $(6.0\pm0.7)$ . This might indicate that 212 emissions from one source of biomass burning (wood burning for residential heating) dominate for all 213 seasons, supporting the use of one levoglucosan to OC (and EC) ratio for calculations. The lower 214 levoglucosan to mannosan ratio observed in summer and fall might indicate increased influence of wild 215 and agricultural fires, but the magnitude of these sources remains speculative, except during severe 216 episodes, e.g. in August 2002, May and September 2006, and June 2008.
- 217

# 8 S6. Levels of PBAP and BSOA organic tracers

The annual mean concentration of the PBAP tracers ranged from 2.8–3.4 ng m<sup>-3</sup> (trehalose) to 4.8–5.8 ng m<sup>-3</sup> (arabitol) (2016–2018) (Figure 6, Table S 15). Levels were elevated in the vegetative season, particularly in summer and fall. Mannitol and arabitol were highly correlated ( $R^2$ =0.85), underlining their common origin, and the mannitol to arabitol ratio (0.9±0.2) corresponds well with previously reported results for these fungal spore tracers (e.g. Bauer et al., 2008; Yttri et al., 2007b; Yttri et al. 2011 a, b).

The annual mean concentration of 2-methylerythritol  $(0.365-0.441 \text{ ng m}^{-3})$  (2016–2018) was higher than that of 2-methylthreitol  $(0.105-162 \text{ ng m}^{-3})$ , and the two isomers were highly correlated  $(R^2=0.915)$ , which is consistent with other studies (e.g., Ion et al., 2005; Kourtchev et al., 2005; Edney et al., 2005; El Haddad et al., 2011; Alier et al., 2013). 2-methyltetrols were elevated in the period when deciduous trees have leaves (transition May/June to early October).

230

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#### **Supplementary Figures**

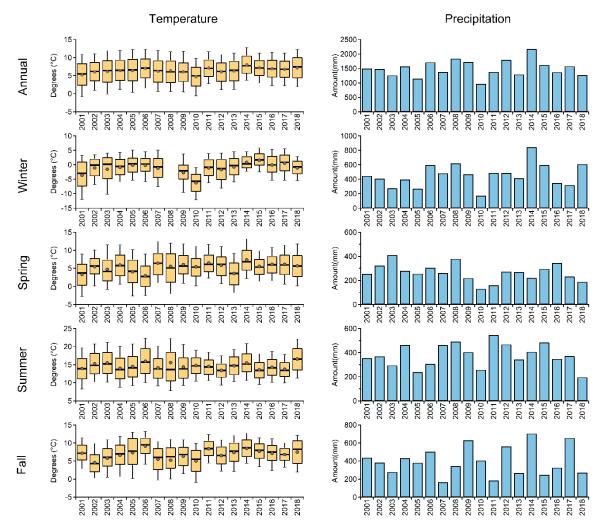


Figure S 1: Annual and seasonal ambient mean (point), 10<sup>th</sup> to 25<sup>th</sup> percentile (bar), 50<sup>th</sup> percentile (line), 75<sup>th</sup> to 90<sup>th</sup> percentile (whisker) temperature (left panel) and precipitation (right panel) at the Birkenes Observatory, 2001–2018.

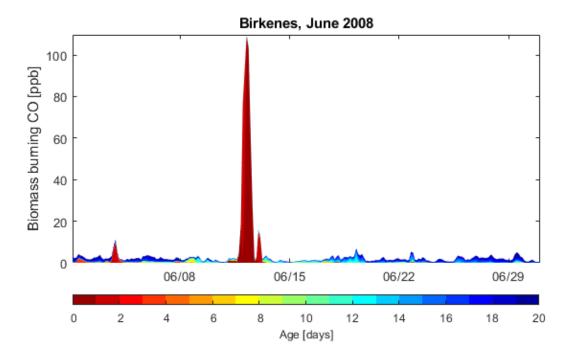


Figure S 2: Age spectra for June 2008 for the CO tracer originating from wildfires calculated for the Birkenes Observatory. We ran a 20 days FLEXPART simulation backwards in time, releasing 40 000 particles for the Birkenes Observatory on a 3-hourly basis, using ECMWF meteorology. With daily MODIS information of burned area we constructed a CO emission inventory, which we combined with the model simulation. With this approach we achieved a time series of CO from wildfires with a 3-hourly time resolution. Additionally, we split the modeled CO concentrations by age. The spectrum goes from 1 to 20 days after release according to the bar. This approach is described in more detail in Stohl et al. (2007). For most of June 2008, concentrations of a few ppb were calculated for the Birkenes Observatory, except for 11–12 of June (100 ppb). The age of the airmasses were only 1 day, which means that the CO was released on a location less than 24 hours before it reached the site.

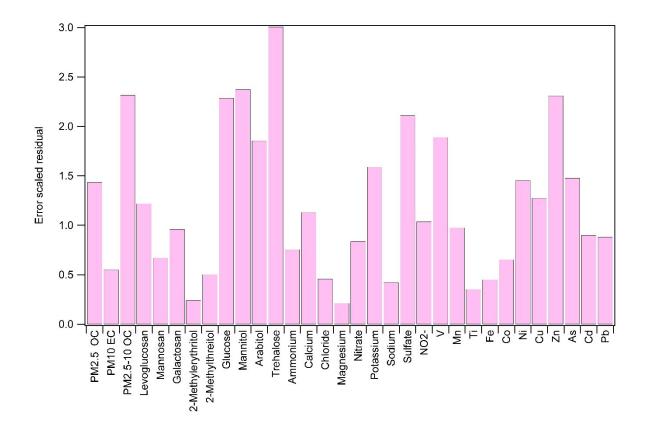


Figure S 3: Average error scaled residuals for each of the variables in the PMF solution presented in this paper

# 1 Supplementary Tables

Table S 1: Species quantification ion, molecular weight and formula, and internal/recovery and quantification
 standards used for identification and quantification.

Species	Quantitation ion M-H(+)	Molecular weight	Molecular formula	Internal/ Recovery standard	Quantification standard
Monosaccharide anhydrides					
Galactosan	161.046	162.141	$C_{6}H_{10}O_{5}$	<sup>13</sup> C <sub>6</sub> -Galactosan (CIL; 98%; Andover, MA)	Galactosan (Sigma; purity not given; Product of England)
Mannosan	161.046	162.141	$C_{6}H_{10}O_{5}$	<sup>13</sup> C <sub>6</sub> -Levoglucosan <sup>1</sup>	Mannosan (Sigma; Approx
				<sup>13</sup> C <sub>6</sub> -Galactosan <sup>1</sup>	98%; Product of England)
Levoglucosan	161.046	162.141	$C_{6}H_{10}O_{5}$	<ul> <li><sup>13</sup>C<sub>6</sub>-Levoglucosan (CIL;</li> <li>98%; city not given)</li> </ul>	Levoglucosan (Aldrich; 99%; Product of Switzerland)
Sugar-alcohols					
Mannitol	181.072	182.172	$C_6H_{14}O_6$	<ul> <li><sup>13</sup>C<sub>6</sub>-Mannitol (Omnicron Biochemicals Inc; 99.70%; South Bend, Indiana)</li> </ul>	Mannitol (ICN Biochemicals; ACS reagent grade; Aurora, Ohio)
Arabitol	151.061	152.15	$C_5H_{12}O_6$	<ul> <li><sup>13</sup>C<sub>5</sub>-Arabitol (Omnicron Biochemicals Inc; 99.60%; South Bend, Indiana)</li> </ul>	Arabitol (ICN Biochemicals; purity not given; Aurora, Ohio)
2-methyltetrols					, 
2-Methylerythritol	135.066	136.147	$C_5H_{12}O_4$	<sup>13</sup> C <sub>6</sub> -Galactosan (CIL; 98%; Andover, MA)	2-Methylerythritol, Produced at UNC <sup>2</sup>
2-Methylthreitol	135.066	136.147	C5H12O4	<sup>13</sup> C <sub>6</sub> -Galactosan (Brand; purity; city)	2-Methylthreitol, Produced at UNC <sup>2</sup>
Dimeric sugars				12	
Trehalose	341.109	342.296	$C_{12}H_{22}O_{11}$	<ul> <li><sup>13</sup>C<sub>12</sub>-Trehaose (Omnicron Biochemicals Inc; 99.70%; South Bend, Indiana)</li> </ul>	Trehalose (Fluka; <99.5%; Packed in Switzerland)
Monomeric sugars					
Glucose	179.0561	180.16	C <sub>6</sub> H <sub>12</sub> O <sub>6</sub>	<sup>13</sup> C <sub>1</sub> -Glucose (CIL; 99%; Andover, MA)	Glucose (Sigma; purity not given; city not given)

1.C-labelled mannosan is not commersially available, hence we used the average of  ${}^{13}C_6$ -Levoglucosan  ${}^{13}C_6$ -Galactosan to calculate the recovery of mannosan.

2.Standard produced by University of North Carolina (UNC)

6 Table S 2: Settings used for absorption coefficient PMF analysis.

Parameter	Setting 7
Data matrix dimensions i×j	5240×7
Missing data treatment	Rows removed
Number of factors	2
Factor constraints	None
Robust mode setting	4
Seed	Random
Bootstrap replacement runs	2000
Block size	24
Repeat runs (per bootstrap)	5

10 Table S 3: Miscellaneous settings of PMF analysis.

Parameter	Setting
Data matrix dimensions i×j	151×34
Missing data and below detection limit data	Replaced with geometric mean concentration
Missing data error estimate	Replaced with 4 $\times$ geometric mean concentration
Missing data error estimate	$5/6 \times \text{limit of detection}$
Number of factors	7
Factor contsraints	None
Robust mode setting	4
Seed	Random
Bootstrap replacement runs	5000
Block size	1 row
Repeat runs (per bootstrap)	5

	Mineral Dust (MIN)	Traffic/Industry (TRA/IND)	Biogenic Secondary Organic Aerosol (BSOA <sub>SRT</sub> )	Primary Biological Aerosol Particle (PBAP)	Sea salt aerosol (SS)	Biomass burning (BB)	Ammonium Nitrate (NH4NO3)
PM2.5 OC	31.3	9.7	9.1	15.6	0.9	16.7	16.6
PM <sub>10</sub> EC	12.8	50.0	0.0	2.6	0.0	21.2	13.4
PM <sub>10-2.5</sub> OC	12.6	3.5	12.5	53.0	0.1	6.3	12.0
Levoglucosan	0.2	0.0	1.5	0.5	0.0	97.8	0.0
Mannosan	0.0	0.5	1.5	1.5	0.0	95.6	0.9
Galactosan	0.0	3.5	0.0	0.0	1.1	95.5	0.0
2-methylerythritol	0.5	0.7	95.9	1.8	0.0	0.4	0.7
2-methylthreitol	0.6	1.3	91.5	3.2	0.7	1.5	1.2
Glucose	1.0	0.6	6.5	81.6	2.0	5.2	3.1
Mannitol	0.1	0.3	6.1	91.3	1.7	0.0	0.5
Arabitol	0.0	0.0	8.6	89.5	1.3	0.5	0.0
Trehalose	0.5	0.0	3.3	93.5	1.1	0.7	0.9
Ammonium	2.1	13.9	4.8	0.0	1.5	1.0	76.7
Calcium	39.0	6.0	7.7	3.9	34.9	1.0	7.6
Chloride	0.0	0.0	0.0	0.0	96.2	0.2	3.5
Magnesium	5.8	2.7	3.9	1.8	79.0	0.6	6.2
Nitrate	0.8	3.6	5.7	3.4	17.1	1.7	67.8
Potassium	3.6	12.7	4.6	8.3	28.4	10.4	32.0
Sodium	2.2	5.1	2.7	0.0	86.8	0.3	3.0
Sulfate	3.9	19.6	17.2	3.3	17.4	3.4	35.2
NO <sub>2</sub> -	7.2	15.0	4.6	4.6	20.0	19.0	29.7
V	14.1	70.1	10.1	0.0	0.0	0.0	5.8
Mn	51.9	38.6	0.6	5.7	2.4	0.8	0.0
Ti	93.4	0.5	0.6	0.0	3.7	1.8	0.0
Fe	74.5	18.4	0.0	3.2	0.7	1.0	2.1
Со	42.6	42.1	1.3	3.3	5.0	2.4	3.4
Ni	17.1	68.6	3.0	4.5	1.7	2.1	2.9

12 Table S 4: Contribution weighted relative profiles for PMF-derived factors (%).

	Mineral Dust (MIN)	Traffic/Industry (TRA/IND)	Biogenic Secondary Organic Aerosol (BSOAsrt)	Primary Biological Aerosol Particle (PBAP)	Sea salt aerosol (SS)	Biomass burning (BB)	Ammonium Nitrate (NH4NO3)
Cu	19.9	61.9	3.3	3.7	5.7	1.3	4.2
Zn	4.3	81.5	1.4	0.6	0.0	5.8	6.4
As	3.2	78.4	1.7	6.2	0.0	4.8	5.7
Cd	5.4	80.5	0.1	2.3	1.6	6.4	3.8
Pb	4.5	83.9	0.8	0.0	1.5	2.8	6.4

Table S 5: Annual and seasonal mean concentrations of EC and OC in PM<sub>10</sub>, PM<sub>2.5</sub> and PM<sub>10-2.5</sub> at Birkenes for 2001–2018 (Unit:  $\mu g \ C \ m^{-3}$ ).

14 15

2001 DJF MAM JJA SON 2002 DJF MAM JJA SON 2003 DJF MAM	<b>OC</b> 0.96 0.60 0.98 2.34 0.58 <b>1.01</b> 0.53 1.30 1.77 0.59 <b>1.01</b>	Capture 56 59 92 26 48 85 95 95 92	<b>EC</b> <b>0.14</b> 0.12 0.16 0.16 0.10 <b>0.14</b> 0.12	Capture           56           59           92           26           48	OC 0.08 NA 0.03	<b>Capture</b> <b>45</b> 34	OC 0.93 0.64	<u>Capture</u> 56 59	EC 0.15 0.12	<u>Capture</u> 56 59
DJF MAM JJA SON <b>2002</b> DJF MAM JJA SON <b>2003</b> DJF	0.60 0.98 2.34 0.58 <b>1.01</b> 0.53 1.30 1.77 0.59	59 92 26 48 <b>85</b> 95 92	0.12 0.16 0.16 0.10 <b>0.14</b>	59 92 <mark>26</mark>	NA 0.03	34	0.64			
MAM JJA SON <b>2002</b> DJF MAM JJA SON <b>2003</b> DJF	0.98 2.34 0.58 <b>1.01</b> 0.53 1.30 1.77 0.59	92 26 48 <b>85</b> 95 92	0.16 0.16 0.10 <b>0.14</b>	92 26	0.03			39	0.17	14
JJA SON 2002 DJF MAM JJA SON 2003 DJF	2.34 0.58 <b>1.01</b> 0.53 1.30 1.77 0.59	26 48 85 95 92	0.16 0.10 <b>0.14</b>	26			1 00			
SON 2002 DJF MAM JJA SON 2003 DJF	0.58 <b>1.01</b> 0.53 1.30 1.77 0.59	48 <b>85</b> 95 92	0.10 <b>0.14</b>		0.24	73	1.00	92	0.18	92
2002 DJF MAM JJA SON 2003 DJF	<b>1.01</b> 0.53 1.30 1.77 0.59	<b>85</b> 95 92	0.14	48	0.24	26	2.1	26	0.20	26
DJF MAM JJA SON <b>2003</b> DJF	0.53 1.30 1.77 0.59	95 92			0.09	48	0.49	48	0.11	48
MAM JJA SON <b>2003</b> DJF	1.30 1.77 0.59	92		85	0.190	<b>80</b>	0.89	<i>91</i>	0.12	<i>91</i>
JJA SON <b>2003</b> DJF	1.77 0.59			95	0.05	88	0.49	95 92	0.12	95
SON <b>2003</b> DJF	0.59		0.17	92 69	0.12	92	1.19	92	0.14	92
<b>2003</b> DJF		68	0.20	68	0.48	68	1.4	91	0.14	91
DJF	1.01	86	0.10	86	0.16	73	0.49	85	0.09	85
	0.00	<i>82</i>	0.10	<i>82</i>	0.23	77	0.81	<u>81</u>	0.11	81
MAM	0.83	85	0.11	85	0.06	76	0.84	85	0.12	85
	1.13	86	0.13	86	0.21	76	1.02	86	0.15	86
JJA	1.26	85	0.08	85	0.44	85	0.82	85	0.09	85
SON	0.75	70	0.09	70	0.22	70	0.53	70	0.09	70
2004	0.82	85	0.10	85	0.27	81	0.57	84	0.09	84
DJF	0.57	86	0.08	86	0.09	86	0.48	86	0.08	86
MAM	1.13	86	0.12	86	0.28	73	0.79	86	0.12	86
JJA	0.99	79	0.10	79	0.38	79	0.61	79	0.08	79
SON	0.71	86	0.08	86	0.32	86	0.39	86	0.08	80
2005	0.85	<b>79</b>	0.14	<b>79</b>	0.29	75	0.60	80	0.12	80
DJF	0.46	64	0.10	64	0.06	51	0.53	71	0.11	71
MAM	0.78	85	0.13	86	0.15	85	0.63	85	0.12	85
JJA	0.92	86	0.09	85	0.33	86	0.59	86	0.08	86
SON	1.16	79	0.25	79	0.53	79	0.62	79	0.16	79
2006	1.07	78	0.13	78	0.33	75	0.77	78	0.13	78
DJF	0.79	86	0.12	86	0.08	79	0.71	86	0.17	86
MAM	0.95	85	0.08	85	0.16	78	0.82	85	0.14	85
JJA	1.43	57	0.12	57	0.48	57	0.96	57	0.10	57
SON	1.24	86	0.19	86	0.60	86	0.64	86	0.11	86
2007	0.82	<b>79</b>	0.15	77	0.21	77	0.61	77	0.13	77
DJF	0.58	58	0.17	58	0.08	58	0.50	58	0.17	58
MAM	0.99	86	0.18	85	0.17	86	0.82	86	0.15	86
JJA	1.03	86	0.13	86	0.39	79	0.66	79	0.10	79
SON	0.60	86	0.13	79	0.18	86	0.41	86	0.10	86
2008	0.75	87	0.09	84	0.24	75	0.53	88	0.08	88
DJF	0.44	90	0.08	83	0.07	82	0.38	90	0.09	84
MAM	0.79	86	0.11	92	0.20	79	0.61	86	0.10	86
JJA	1.27	86	0.08	86	0.46	86	0.81	86	0.06	86
SON	0.51	90	0.09	82	0.23	82	0.32	90	0.08	97
2009	0.79	<b>98</b>	0.10	<b>98</b>	0.23	77	0.58	96	0.09	96
DJF	0.56	100	0.12	100	0.06	69	0.47	92	0.11	92
MAM	0.81	92	0.11	85	0.11	85	0.74	92	0.10	92
JJA	1.1	100	0.09	100	0.40	100	0.71	100	0.07	100
SON	0.68	100	0.09	100	0.28	100	0.40	100	0.07	100
2010	0.90	94	0.11	94	0.24	79	0.67	96	0.10	96
DJF	0.95	92	0.14	92	0.09	40	0.86	92	0.16	92
MAM	0.82	85	0.09	85	0.18	82	0.61	100	0.08	100
JJA	1.02	100	0.09	100	0.34	100	0.68	100	0.07	100
SON	0.79	100	0.11	100	0.24	100	0.51	92	0.09	92
2011	0.92	98	0.11	92	0.26	94	0.68	98	0.11	96
DJF	0.60	100	0.11	100	0.09	92	0.50	92	0.11	100
MAM	0.00	93	0.10	70	0.07	93	0.73	100	0.10	85
JJA	1.07	100	0.07	100	0.37	100	0.69	100	0.07	100
SON	1.07	100	0.17	100	0.30	92	0.77	100	0.16	100
<i>2012</i>	0.56	<b>89</b>	0.08	<b>86</b>	0.30 0.10	79	0.50	<i>90</i>	0.10 0.08	<i>90</i>

	PM <sub>10</sub>	C (	EG	<b>C</b> (	PM <sub>10-2.5</sub>	<b>G</b> (	PM <sub>2.5</sub>	<b>C</b> (	EG	<b>C</b> (
DIE	OC	Capture	EC	Capture	OC	Capture	OC	Capture	EC	Captur
DJF	0.52	100	0.09	100	0.04	84	0.49	100	0.09	100
MAM	0.58	85	0.09	85	0.03	70	0.63	70	0.11	70
JJA	0.78	70	0.07	70	0.18	70	0.59	100	0.07	100
SON	0.56	100	0.07	92	0.14	92	0.30	92	0.07	92
2013	0.76	<i>92</i>	0.09	96	0.21	90	0.57	<b>98</b>	0.08	96
DJF	0.49	92	0.10	92	0.05	68	0.47	91	0.09	91
MAM	0.79	100	0.10	100	0.15	100	0.63	100	0.09	100
JJA	1.16	100	0.07	92	0.37	100	0.79	100	0.07	92
SON	0.58	100	0.07	100	0.22	92	0.39	100	0.07	100
2014	0.91	<u>98</u>	0.09	100	0.29	94	0.65	96	0.08	96
DJF	0.61	100	0.10	100	0.08	76	0.59	84	0.11	84
MAM	0.91	100	0.10	100	0.23	100	0.69	100	0.09	100
JJA	1.10	100	0.05	100	0.35	100	0.75	100	0.05	100
SON	1.20	100	0.10	100	0.47	100	0.55	100	0.09	100
2015	0.72	<b>98</b>	0.09	<b>98</b>	0.19	85	0.52	88	0.08	88
DJF	0.44	100	0.06	100	0.11	100	0.34	100	0.06	100
MAM	0.59	92	0.10	92	0.11	79	0.50	92	0.08	92
JJA	1.01	100	0.09	100	0.35	63	0.66	63	0.08	63
SON	0.83	100	0.10	100	0.22	99	0.62	99	0.10	99
2016	0.73	100	0.06	100	0.21	95	0.54	100	0.06	100
DJF	0.44	100	0.07	100	0.07	86	0.37	100	0.06	100
MAM	0.83	100	0.07	100	0.21	92	0.64	100	0.07	100
JJA	0.98	100	0.04	100	0.33	100	0.65	100	0.05	100
SON	0.68	100	0.07	100	0.20	100	0.48	100	0.07	100
2017	0.72	94	0.05	94	0.25	<b>79</b>	0.52	94	0.05	94
DJF	0.57	100	0.07	100	0.11	63	0.53	100	0.07	100
MAM	0.65	92	0.05	92	0.14	84	0.52	92	0.05	92
JJA	0.92	86	0.03	86	0.34	86	0.58	86	0.04	86
SON	0.77	100	0.06	100	0.38	83	0.47	100	0.05	100
2018	0.96	100	0.08	100	0.26	90	0.73	100	0.07	100
DJF	0.49	100	0.07	100	0.08	77	0.45	100	0.07	100
MAM	1.32	100	0.11	100	0.28	92	1.06	100	0.10	100
JJA	1.20	100	0.05	100	0.32	100	0.90	100	0.05	100
SON	0.81	100	0.08	100	0.31	100	0.50	100	0.07	100

19 Table S 6: R<sup>2</sup>-values for OC versus EC as a function of size fraction and season.

	Winter	Spring	Summer	Fall
<b>PM</b> <sub>10</sub>	0.66	0.58	0.51	0.64
PM <sub>2.5</sub>	0.75	0.69	0.58	0.76

	OM/PM <sub>10</sub>	EC/PM <sub>10</sub>	SO4 <sup>2-</sup> /PM <sub>10</sub>	NO <sub>3</sub> -/PM <sub>10</sub>	$\rm NH_4^+/PM_{10}$	SS/PM <sub>10</sub>	OM/PM <sub>2.5</sub>	EC/PM <sub>2.5</sub>	OM /PM10-2.22
2001	$31\pm 4$	$2.7{\pm}0.4$	21	11	9	11	38±5	$3.6 \pm 0.5$	$8.9\pm1.8^{1}23$
2002	$26\pm4$	$2.1{\pm}0.3$	20	14	8	10	$30\pm4$	$2.3 \pm 0.3$	$16{\pm}3$
2003	$29\pm4$	$1.6\pm0.2$	23	12	9	11	32±5	$2.5 \pm 0.4$	$18\pm4$
2004	27±4	$1.9\pm0.3$	19	15	L	14	32±5	$2.9{\pm}0.4$	$21\pm4$
2005	25±4	$2.4{\pm}0.3$	20	16	8	13	$29\pm4$	$3.3{\pm}0.5$	$21\pm4$
2006	$26\pm4$	$1.8{\pm}0.3$	20	17	5	12	$31\pm4$	$3.0 \pm 0.4$	$18\pm4$
2007	27±4	$2.9\pm0.4$	15	10	4	14	35±5	$4.3 \pm 0.6$	$16{\pm}3$
2008	25±4	$1.7{\pm}0.2$	14	12	С	18	$36\pm 5$	$3.1 {\pm} 0.4$	$17 \pm 3$
2009	$26\pm4$	$3.7{\pm}1.9$	15	13	4	13	$31\pm4$	$2.8 \pm 0.4$	$19\pm4$
2010	$34\pm 5$	$2.4{\pm}0.3$	17	13	5	11	37±5	$3.2 \pm 0.5$	$21\pm4$
2011	25±4	$1.7{\pm}0.2$	14	17	9	16	32±4	$3.0{\pm}0.4$	$15 \pm 3$
2012	22±3	$1.8{\pm}0.3$	17	28	L	17	$33\pm 5$	$3.0{\pm}0.4$	$10\pm 2$
2013	29±4	$2.0 \pm 0.3$	15	19	9	23	37±5	$3.0{\pm}0.4$	$18\pm4$
2014	$28\pm4$	$1.6\pm0.2$	18	21	L	20	$36\pm 5$	$2.6 \pm 0.4$	$20\pm4$
2015	$26\pm4$	$1.9\pm0.3$	16	22	9	28	$36\pm 5$	$3.3{\pm}0.5$	$20\pm3$
2016	32±5	$1.5 {\pm} 0.2$	14	23	7	23	$41{\pm}6$	$2.6\pm0.4$	$21\pm4$
2017	$38\pm 5$	$1.5 {\pm} 0.2$	19	14	5	27	49±7	$2.8{\pm}0.4$	$28\pm 6$
2018	$34\pm 5$	$1.6 {\pm} 0.2$	14	15	6	19	46±7	$2.6 \pm 0.4$	$20\pm4$

<b>%)</b> <sup>1</sup>
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<sup>&</sup>lt;sup>1</sup> 1) Data capture below 50%.
2) Notation: Conversion factors applied OM = OC x 1.9; EC = EC x 1.1

	<b>РМ</b> 10 МАХ		MO	EC	504 <sup>2-</sup>	NO <sup>3-</sup>	NH₄⁺	SS	<b>РМ</b> 2.5 мах		ΜO	EC
	µg т <sup>-3</sup>	Season <sup>1)</sup>	%	%	%	%	%	%	µg т <sup>-3</sup>	Season <sup>1)</sup>	%	%
2001	$12.4{\pm}1.6$	234	<b>4</b> 3±16	$3\pm 1$	$14{\pm}10$	4±3	4±3	3±3	12.2±2.4	234	$50{\pm}10$	5±2
2002	$23.8 \pm 3.6$	223	26±11	$1.8\pm0.7$	$14\pm9$	$13 \pm 14$	$8\pm4$	2±2	$18.8 \pm 2.9$	223	29±12	$1.7{\pm}0.5$
2003	$17.9 \pm 2.7$	222	$16\pm3$	$1.3 \pm 0.4$	17±3	$18\pm 8$	$11\pm 2$	8±3	$12.0 \pm 1.9$	222	$19\pm3$	$1.6 \pm 0.3$
2004	$17.0 {\pm} 6.8$	222	23±9	$1.5 {\pm} 0.4$	21±3	$11{\pm}10$	$8\pm4$	$14\pm18$	$11.6 \pm 6.5$	222	27±9	$2.4{\pm}0.8$
2005	$16.9 \pm 2.8$	244	26±6	$3.4{\pm}1.6$	15±3	20±3	$2\pm 0$	5±3	12.4±3.1	224	27±10	$2.9 \pm 0.4$
2006	$26.5 \pm 3.0$	144	23±8	$4.1 \pm 1.6$	19±7	$17\pm4$	8±2	7±5	$15.5 \pm 0.9$	124	$30{\pm}10$	$2.3 \pm 0.7$
2007	$13.8 \pm 3.5$	223	40±2	$4.2 \pm 1.5$	12±0	6±2	$4\pm1$	$1{\pm}0$	$10.7 \pm 3.4$	223	48±2	$4.1 \pm 1.5$
2008	$12.5 \pm 3.0$	222	12±5	$1.5 {\pm} 0.5$	12±3	20±7	5±3	15±15	7.0±2.7	223	$16\pm3$	$1.9 \pm 0.6$
2009	$14.8 \pm 5.7$	222	$21\pm1$	$1.8\pm0.0$	12±5	$13\pm9$	5±3	$3\pm 2$	$10.6 \pm 4.1$	122	28±3	$2.5 {\pm} 0.5$
2010	$12.0\pm 2.4$	344	$21\pm 6$	$1.9\pm0.6$	14±3	$13\pm 2$	$3\pm 1$	$23 \pm 11$	$10.7 \pm 3.7$	144	42±9	$5.0 {\pm} 0.6$
2011	$16.7 \pm 1.2$	224	23±13	$2.0 \pm 1.3$	23±2	$17\pm4$	$12\pm0$	$2\pm 1$	12.1±1.5	224	27±13	$3.0 \pm 1.9$
2012	$11.9 \pm 1.7$	122	<b>29</b> ±21	$3.9 \pm 3.4$	$10 \pm 9$	24±25	3±2	$1{\pm}0$	7.2±1.4	123	30±26	3.0±2.7
2013	$9.4{\pm}0.6$	222	$12\pm4$	$1.4{\pm}0.5$	$9\pm 2$	7±2	$8{\pm}0$	$13\pm 5$	$5.7 {\pm} 0.6$	223	$30{\pm}9$	$2.6 \pm 1.0$
2014	$17.7 \pm 3.3$	224	$18\pm 8$	$1.5 {\pm} 0.2$	15±5	25±6	$10\pm 1$	$14\pm 8$	$8.1{\pm}0.9$	122	$21\pm 6$	$2.5 \pm 0.5$
2015	$12.0 \pm 4.0$	122	$21 \pm 10$	$2.2 \pm 1.3$	12±7	$23 \pm 10$	$11\pm 2$	12±12	$6.9{\pm}1.7$	122	$31{\pm}20$	$3.3{\pm}2.0$
2016	9.2.±0.2	223	30±22	$1.3 \pm 0.3$	6±5	$18 \pm 17$	7±7	$6\pm 1$	6.7±1.5	223	37±25	$1.9 \pm 0.3$
2017	9.8.±2.7	114	33±8	$2.3 \pm 1.1$	19±3	$14\pm7$	$7\pm 1$	$13 \pm 13$	$6.3 \pm 2.0$	112	29±8	$2.7 \pm 0.3$
2018	12.8.±3.1	124	32±24	$1.3 {\pm} 0.5$	$11 \pm 3$	$20\pm15$	8±5	$5\pm 1$	$9.0{\pm}1.0$	144	$21 \pm 11$	$1.7 \pm 1.2$

Table S 8: Mean (±SD) relative chemical composition of the 3 weekly samples with the highest PM mass concentration (PMMAX) per year for the period 2001–2018.<sup>2</sup>

<sup>2</sup> 1 = DJF; 2 = MAM; 3 = JJA; 4 = SON

26 27 Table S 9: Annual and seasonal mean concentrations of TC in PM10, PM2.5 and PM10-2.5 at Birkenes for 2001–2018 (Unit: μg C m<sup>-3</sup>).

	<b>PM</b> <sub>10</sub>		PM <sub>10-2.5</sub>		PM <sub>2.5</sub>	
	TC	Capture	TC	Capture	TC	Capture
2001	1.09	63	0.07	48	1.08	63
DJF	0.72	59	0	34	0.75	59
MAM	1.14	92	0.02	65	1.19	92
JJA	2.50	26	0.20	26	2.30	26
SON	0.68	48	0.08	48	0.60	48
2002	1.15	85	0.00 0.21	80	1.01	91
DJF	0.65	95	0.05	88	0.60	95
MAM	1.47	92	0.14	92	1.33	92
JJA	1.96	68	0.53	68	1.50	91
SON	0.68	86	0.18	73	0.59	86
2003	<i>1,12</i>	82	<i>0,23</i>	78	0,93	82
DJF	0.94	85	0.05	82	0.95	85
MAM	1.26	86	0.05	76	1.16	85
JJA	1.20	85	0.20	85	0.91	85
SON	0.84	83 70	0.43	83 70	0.91	83 70
2004	0.84 0.91	84	0.22 0.28	70 79	0.62 0.65	84
2004 DJF		<b>8</b> 6				
	0.65		0.09	86	0.56	86
MAM	1.13	86 70	0.32	66 70	0.91	86 70
JJA	1.09	79	0.41	79	0.68	79 86
SON	0.79	86	0.33	86	0.47	86
2005	0.99	<b>79</b>	0.32	75	0.71	80
DJF	0.56	64	0.06	51	0.64	71
MAM	0.91	85	0.16	85	0.75	85
JJA	1.01	86	0.34	86	0.67	86
SON	1.41	79	0.62	79	0.79	79
2006	1.20	78	0.35	70	0.90	78
DJF	0.91	86	0.08	66	0.88	86
MAM	1.03	85	0.10	72	0.96	85
JJA	1.55	57	0.49	57	1.06	57
SON	1.43	86	0.68	86	0.75	86
2007	0.99	77	0.24	76	0.74	77
DJF	0.76	58	0.09	58	0.67	58
MAM	1.17	86	0.20	86	0.97	86
JJA	1.16	86	0.41	79	0.77	79
SON	0.76	79	0.21	79	0.52	86
2008	0.85	84	0.25	81	0.60	86
DJF	0.47	84	0.05	76	0.43	84
MAM	0.90	86	0.21	79	0.71	86
JJA	1.35	86	0.47	86	0.87	86
SON	0.65	82	0.24	82	0.39	90
2009	0.89	<u>98</u>	0.23	<i>92</i>	0.67	96
DJF	0.68	100	0.05	84	0.58	92
MAM	0.92	92	0.12	85	0.85	92
JJA	1.19	100	0.41	100	0.78	100
SON	0.76	100	0.29	100	0.47	100
2010	1.00	94	0.21	87	<b>0.</b> 77	96
DJF	1.09	92	0.06	69	1.03	92
MAM	0.91	85	0.17	85	0.69	100
JJA	1.11	100	0.36	100	0.75	100
SON	0.90	100	0.26	92	0.61	92
2011	0.99	98	0.25	92 92	0.80	100
DJF	0.71	100	0.09	92 92	0.64	100
MAM	0.88	92	0.09	85	0.84	100
JJA	1.13	100	0.20	100	0.34	100
SON	1.13	100	0.37	92	0.70	100
2012	0.64	<b>89</b>	0.32 0.10	92 79	0.93 0.58	92
2012	0.04	07	0.10	/ 7	0.30	74

	<b>PM</b> <sub>10</sub>		PM <sub>10-2.5</sub>		PM <sub>2.5</sub>	
	TC	Capture	ТС	Capture	TC	Capture
DJF	0.61	100	0.03	92	0.58	100
MAM	0.67	85	0	68	0.73	77
JJA	0.85	70	0.25	56	0.66	100
SON	0.51	100	0.14	92	0.37	92
2013	0.84	<b>98</b>	0.21	<i>92</i>	0.65	98
DJF	0.59	92	0.04	76	0.56	91
MAM	0.89	100	0.17	100	0.72	100
JJA	1.22	100	0.37	100	0.86	100
SON	0.66	100	0.23	92	0.46	100
2014	1.00	100	0.30	94	0.73	96
DJF	0.71	100	0.07	76	0.70	84
MAM	1.02	100	0.24	100	0.78	100
JJA	1.16	100	0.35	100	0.80	100
SON	1.12	100	0.48	100	0.64	100
2015	0.81	<b>98</b>	0.19	88	0.60	88
DJF	0.50	100	0.11	100	0.39	100
MAM	0.70	92	0.10	92	0.60	92
JJA	1.10	100	0.36	63	0.74	63
SON	0.94	100	0.23	96	0.71	96
2016	0.80	100	0.21	94	0.6	100
DJF	0.51	100	0.08	82	0.43	100
MAM	0.90	100	0.22	92	0.71	100
JJA	1.02	100	0.32	100	0.70	100
SON	0.76	100	0.21	100	0.55	100
2017	0.78	94	0.26	78	0.58	<i>94</i>
DJF	0.63	100	0.11	59	0.60	100
MAM	0.70	92	0.14	84	0.58	92
JJA	0.95	86	0.34	86	0.61	86
SON	0.84	100	0.40	83	0.51	100
2018	1.03	100	0.26	90	0.8	100
DJF	0.56	100	0.08	77	0.52	100
MAM	1.43	100	0.30	92	1.06	100
JJA	1.25	100	0.32	92	0.95	100
SON	0.88	100	0.32	100	0.56	100

Notation: Red numbers indicate annual or seasonal means based on < 50% data capture.

Table S 10: Annual and seasonal mean mass concentrations of PM<sub>10</sub>, PM<sub>2.5</sub> and PM<sub>10-2.5</sub> at Birkenes for 2001–2018 (Unit: 31 µg m<sup>-3</sup>).

	<b>PM</b> <sub>10</sub>	Capture	PM2.5	Capture	PM10-2.5	Capture
2001	5.8	56	4.6	58	1.7	54
DJF	4.5	59	3.2	59	1.3	59
MAM	6.6	92	5.5	100	1.8	92
JJA	7.9	26	7.3	26	1.3	19
SON	4.9	48	2.9	48	1.9	48
2002	7.5	83	5.7	91	2.2	80
DJF	5.6	88	3.8	95	1.8	88
MAM	11.0	92	8.5	92	2.4	92
JJA	10.0	68	7.3	95	2.9	68
SON	3.9	86	2.8	86	1.5	73
2003	6.7	78	4.8	82	2.4	74
DJF	5.9	76	4.4	85	2.2	76
MAM	9.3	82	6.7	86	3.4	76
JJA	5.9	85	4.4	81	1.7	79
SON	5.4	70	3.5	70	2.3	64
2004	5.7	84	3.4	84	2.4	<i>83</i>
DJF	4.5	86	2.8	86	1.7	86
MAM	8.2	86	5.4	85	3.1	79
JJA	5.7	79	3.3	79	2.4	79
SON	4.3	86	2.0	86	2.3	86
2005	6.5	<b>79</b>	4.0	80	2.6	75
DJF	5.2	64	3.1	71	2.8	51
MAM	6.9	85	4.9	85	1.9	85
JJA	5.5	86	3.4	86	2.1	86
SON	8.0	79	4.4	79	3.6	79
2006	7.8	78	4.7	77	3.4	73
DJF	7.4	86	4.5	79	3.0	79
MAM	6.2	85	4.6	86	2.1	72
JJA	8.3	57	5.4	57	3.0	57
SON	9.5	86	4.5	86	5.0	86
2007	5.8	72	3.3	74	2.5	72
DJF	4.2	38	2.0	38	2.2	38
MAM	7.5	86	4.4	86	3.1	86
JJA	6.1	79	3.5	86	2.5	79
SON	4.4	86	2.4	86	2.0	86
2008	5.7	77	2.8	86	2.0 2.7	75
DJF	5.1	84	2.5	90	2.8	84
MAM	7.1	79	4.1	73	2.0	73
JJA	6.1	86	3.1	86	2.9	86
SON	4.0	58	1.9	97	2.9	58
2009	5.8	85	3.6	94	2.1 2.3	76
DJF	<b>4.4</b>	96	3	92	2.3 1.4	88
MAM	9.4	74	5.3	100	3.4	74
JJA	6.1	71	3.8	92	3.4 2.2	63
SON	4.3	99	2.2	92	2.2	83

	<b>PM</b> <sub>10</sub>	Capture	PM2.5	Capture	PM10-2.5	Capture
2010	5.1	88	3.4	94	2.2	77
DJF	3.8	100	3.4	87	0.55	64
MAM	5.4	71	3	98	2.3	68
JJA	6.7	90	3.7	100	3.1	91
SON	4.8	92	3.5	92	2.5	85
2011	7.0	<b>98</b>	4.1	100	3.2	96
DJF	5.6	100	3.1	100	2.5	100
MAM	8.2	93	5.5	100	3.9	85
JJA	5.2	100	3.5	100	1.7	100
SON	9.0	100	4.4	100	4.6	100
2012	4.9	89	2.9	<i>92</i>	2.0	85
DJF	4.4	100	2.7	100	1.7	99
MAM	6.7	85	3.2	100	3.3	70
JJA	4.6	70	3.8	70	1.4	70
SON	4.0	100	2.0	100	2.0	100
2013	4.9	<i>92</i>	2.9	86	2.2	84
DJF	3.8	69	2.6	44	1.7	44
MAM	6.4	100	3.7	100	2.8	100
JJA	5.4	100	3.4	100	2.0	100
SON	3.6	100	1.8	100	2.0	92
2014	6.1	<i>98</i>	3.4	96	2.0 2.7	<i>94</i>
DJF	5.7	100	3.5	84	2.4	84
MAM	7.2	92	3.7	100	3.3	92
JJA	5.2	100	3.2	100	2.0	100
SON	6.3	100	3.3	100	3.0	100
2015	5.3	<b>98</b>	2.7	88	2.5	88
DJF	5.5	100	2.5	100	31	100
MAM	5.2	100	2.8	100	2.5	100
JJA	5.5	92	2.9	63	1.9	63
SON	5.0	100	2.8	99	2.4	99
2016	4.3	100	2.5	100	1.9	100
DJF	4.1	100	2.1	100	2.0	100
MAM	4.8	100	3.1	100	1.7	100
JJA	4.3	100	2.6	100	1.7	100
SON	4.1	100	2.0	100	2.1	100
2017	3.8	90	2.0	94	1.7	90
DJF	3.7	100	2.2	100	1.4	100
MAM	3.8	92	2.2	92	1.4	92
JJA	3.8 3.9	86	2.3	86	1.5	86
SON	3.9	85	2.1 1.4	100	2.4	85
2018	5.8 5.4	100	<b>3.0</b>	98	2.4 2.5	98
DJF	<b>5.4</b> 3.8	100	<b>3.0</b> 2.4	92	2.5 1.4	92
MAM	5.8 6.5	100	2.4 4.2	100	1.4 2.3	100
JJA	6.5 5.4	100	4.2 3.1	100		100
SON	5.4 6.2	100	3.1 2.2	100	2.3 3.9	100

 $\frac{32}{\text{Red numbers indicate annual or seasonal means based on < 50% data capture.}}$ 

Table S 11: Sen slope of annual means and corresponding confidence intervals for significant slopes (p=0.05), as well as change in annual mean presented as percentage change per year and as percentage change for the period 2001–2018. Non-significant values in red.

34 35 36

	Slope (% yr <sup>-1</sup> )	CI-1	CI-2	Change 2001-2018 (%)
<b>PM</b> <sub>10</sub>	-2.2	-3.7	-0.7	-38
PM <sub>2.5</sub>	-4.0	-5.7	-2.2	-69
PM <sub>10-2.5</sub>	-0.1	-2.2	1.4	-2.4
OC in PM <sub>10</sub>	0	-1.4	0.9	0
OC in PM <sub>2.5</sub>	-0.8	-2.8	0.7	-13
<b>OC in PM</b> <sub>10-2.5</sub>	0.8	-1.7	3.4	13
EC in PM <sub>10</sub>	-3.9	-5.8	-1.9	-66
EC in PM <sub>2.5</sub>	-4.2	-6.2	-2.6	-71
TC in PM <sub>10</sub>	-1.1	-2.0	0.0	-19
TC in PM <sub>2.5</sub>	-1.5	-3.5	0.0	-26
<b>TC in PM</b> <sub>10-2.5</sub>	0.0	-2.4	1.9	0
<b>SO</b> <sub>4</sub> <sup>2-</sup>	-3.8	-6.1	-1.8	-65
NO <sub>3</sub> -	0.8	-2.5	4.3	14
NH4 <sup>+</sup>	-2.7	-5.8	0.5	-47
SS	2.2	0.6	4.5	38
Levoglucosan <sup>1)</sup>	-2.8	-8.8	-0.2	-28

37 Notation: Trends for levoglucosan are calculated for the period 2008-2018.

		DJF			MAM			JJA			SON	
	Slope (% yr <sup>-1</sup> )	CI-1	CI-2	Slope (% yr <sup>-1</sup> )	CI-1	CI-2	Slope (% yr <sup>-1</sup> )	CI-1	CI-2	Slope (% yr <sup>-1</sup> )	CI-1	CI-2
PM <sub>10</sub>	-1.6	-3.3	-0.1	-3.3	-5.6	-0.9	-2.4	-5.1	-0.9	-0.5	-3.3	2.2
$PM_{2.5}$	-2.4	4.6	-0.8	-4.4	-7.0	-3.0	-3.0	-7.1	-1.4	-2.9	-6.5	0.6
$PM_{10-2.5}$	0.0	-2.6	3.0	-0.4	-4.2	2.3	-2.3	-4.5	0.6	0.8	-1.2	4.3
OC in PM <sub>10</sub>	-0.2	-2.8	1.4	-1.0	-3.1	1.2	-0.7	-3.2	1.4	1.7	-1.3	4.0
OC in PM <sub>2.5</sub>	-0.8	-3.7	1.4	-1.9	-3.7	0.5	-0.6	-2.8	2.1	1.5	-1.2	4.3
OC in PM <sub>10-2.5</sub>	3.2	0.0	6.7	3.7	-1.7	7.2	-1.4	-2.5	0.1	1.2	-2.4	5.4
EC in PM <sub>10</sub>	-2.8	-4.9	-0.7	-4.0	-6.0	-1.6	-5.9	-9.8	-3.3	-2.3	-9.2	0.0
EC in PM <sub>2.5</sub>	-3.1	-5.4	-0.7	-4.6	-6.7	-3.0	-4.1	-6.7	-2.2	-2.0	-5.5	0.0
TC in PM <sub>10</sub>	-1.0	-3.5	0.7	-2.0	-3.4	-0.1	-1.2	-3.6	0.9	0.9	-2.8	3.5
TC in PM <sub>2.5</sub>	-1.4	-3.5	0.7	-2.6	-4.2	-0.9	-1.0	-3.6	1.3	0.5	-2.5	2.7
TC in PM <sub>10-2.5</sub>	3.4	-0.3	6.7	3.1	-2.6	7.9	-1.7	-3.2	-0.7	1.1	-3.3	4.9
$SO_4^{2-}$	-3.0	-6.1	-0.2	-6.4	-9.0	-3.8	-4.2	-5.9	-2.9	-2.4	-5.3	0.7
SS	2.9	-1.4	7.4	1.0	-1.6	3.9	3.7	2.3	5.6	3.1	0.0	4.8
Levoglucosan <sup>1)</sup>	-3.3	-15.9	6.7	1.9	-6.7	5.3	-5.7	-18.1	1.4	-1.4	-12.3	7.2

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42 43 44 Table S 13: Sen slope of annual mean ratios and corresponding confidence intervals for significant slopes (p=0.05), as

well as change in annual mean presented as percentage change per year and as percentage change for the period 2001-

2018 (2008–2018 for levoglucosan). Non-significant values in red.

	Slope (% yr <sup>-1</sup> )	CI-1	<b>CI-2</b>	Change 2001-2018 (%)
OC <sub>PM10</sub> to PM <sub>10</sub>	2.4	0.7	3.4	41
OC <sub>PM2.5</sub> to PM <sub>2.5</sub>	3.2	1.7	4.4	55
<b>OC</b> <sub>PM10-2.5</sub> to <b>PM</b> <sub>10-2.5</sub>	1.1	-1.3	2.5	17
EC <sub>PM10</sub> to PM <sub>10</sub>	-4.5	-7.1	-2.8	-77
EC <sub>PM2.5</sub> to PM <sub>2.5</sub>	-3.9	-5.8	-1.9	-66
TC <sub>PM10</sub> to PM <sub>10</sub>	1.8	0.3	2.6	30
TC <sub>PM2.5</sub> to PM <sub>2.5</sub>	2.6	1.4	3.7	44
TC <sub>PM10-2.5</sub> to PM <sub>10-2.5</sub>	0.4	-1.8	1.8	7.1
SO4 <sup>2-</sup> to PM <sub>10</sub>	-2.1	-3.4	-0.4	-35
$NO_3^-$ to $PM_{10}$	3.8	0.8	6.3	64
NH <sub>4</sub> <sup>+</sup> to PM <sub>10</sub>	-0.7	-3.1	2.4	-12
SS to PM <sub>10</sub>	4.4	3.0	6.7	75
Levoglucosan to OC <sub>PM10</sub>	-1.8	-10.6	1.8	-18
Levoglucosan to OC <sub>PM2.5</sub>	-3.6	-9.8	1.3	-36
Levoglucosan to EC <sub>PM10</sub>	2.8	-3.5	6.5	28
Levoglucosan to EC <sub>PM2.5</sub>	2.3	-2.2	5.0	24
Levoglucosan to TC <sub>PM10</sub>	-1.1	-9.0	2.7	-11
Levoglucosan to TC <sub>PM2.5</sub>	-3.1	-8.1	2.0	-31

45 Notation: Trends for levoglucosan are calculated for the period 2008–2018.

Table S 14:Sen slope of seasonal mean ratios and corresponding co per year and as percentage change for the period 2001 – 2018 (2008
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		DJF			MAM			JJA			SON	
	Slope (% yr <sup>-1</sup> )	CI-1	CI-2	Slope (% yr <sup>-1</sup> )	CI-1	CI-2	Slope (% yr <sup>-1</sup> )	CI-1	CI-2	Slope (% yr <sup>-1</sup> )	CI-1	CI-2
<b>OCPMI0</b> to PM10	1.8	-0.6	3.7	1.7	0.4	3.7	2.3	1.0	3.6	1.7	0.4	3.7
<b>OCPM2.5</b> to PM2.5	1.8	-0.4	3.9	3.1	0.9	5.8	3.9	2.4	5.1	3.1	0.9	5.8
<b>OC</b> <sub>PM10-2.5</sub> to PM <sub>10-2.5</sub>	2.8	-0.9	5.8	-0.3	-2.5	2.6	1.6	-1.2	3.1	-0.3	-2.5	2.6
ECPM10 to PM10	-4.6	-7.2	-2.7	-4.7	-7.9	-2.5	-7.1	-10.7	-3.4	-4.7	-7.9	-2.5
ECPM2.5 to PM2.5	-3.9	-5.4	-2.5	-3.6	-5.3	-1.2	-4.3	-7.5	-1.7	-3.6	-5.3	-1.2
TCPM10 to PM10	0.9	-1.2	3.2	1.1	-0.2	2.7	1.7	0.6	2.8	1.1	-0.2	2.7
TCPM2.5 to PM2.5	1.2	-0.6	3.3	-0.7	-3.6	1.6	0.7	-1.4	1.9	-0.7	-3.6	1.6
TCPM10-2.5 to PM10-2.5	2.2	-1.6	4.4	2.4	0.5	4.6	3.2	2.1	4.1	2.4	0.5	4.6
SO4 <sup>2-</sup> to PM <sub>10</sub>	-2.2	4.0	0.1	-1.3	-3.4	-0.4	-1.3	-3.1	0.0	-1.3	-3.4	-0.4
NO <sup>3-</sup> to PM <sub>10</sub>	7.3	3.2	11.1	1.2	-2.4	4.1	4.4	-0.4	7.1	1.2	-2.4	4.1
NH4 <sup>+</sup> to PM <sub>10</sub>	2.6	-1.7	5.4	-1.4	-6.5	3.0	-0.6	-4.8	3.6	-1.4	-6.5	3.0
SS to PM <sub>10</sub>	4.1	-0.3	7.4	4.8	1.8	7.6	6.2	4.3	8.0	4.8	1.8	7.6
Levoglucosan to OC <sub>PM10</sub>	-5.9	-7.9	2.5	4.9	-9.2	0.0	0.0	-13.6	0.0	-4.9	-9.2	0.0
Levoglucosan to OC <sub>PM2.5</sub>	-2.7	-6.5	1.8	4.2	-10.8	0.0	0.0	-8.7	3.7	4.2	-10.8	0.0
Levoglucosan to EC <sub>PM10</sub>	3.1	-6.9	8.6	1.1	-6.5	5.1	8.2	-2.6	13.6	1.1	-6.5	5.1
Levoglucosan to EC <sub>PM2.5</sub>	2.5	-3.7	6.4	-0.4	-8.7	5.6	3.5	-5.1	10.7	-0.4	-8.7	5.6
Levoglucosan to TC <sub>PM10</sub>	-3.5	-7.4	3.5	-5.6	-8.7	0.0	0.0	-13.6	0.0	-5.6	-8.7	0.0
Levoglucosan to TCPM2.5	-3.2	-6.4	2.1	-2.1	-6.6	1.8	0.0	-11.9	7.4	-2.1	-6.6	1.8

Notation: Trends for levoglucosan are calculated for the period 2008–2018

2008 10		Cap Man-		Cap (	Galact-	Cap	Levo/	Cap	Ara-	Cap	Man-	Cap	2-methyl-	Cap	2-methyl-	Cap	Tre-	Cap	Glu-	Cap
	san		nosan	-	osan		Mann		bitol		nitol		erythritol		threitol		halose		cose	
	10.90 90	) 2.68			0.91	90	4.7	75												
	0.24 90				.69	90	4.7	69												
					0.40	86	5.5	82												
					2.00	90	3.4	58												
	1.34 97	7 2.63			0.53	76	4.5	89												
				-	0.57	98	5.9	90												
					1.05	92	5.6	83												
		~		-	).69	100	6.6	100												
				-	0.06	100	5.4	94												
				-	).50	100	5.4	85												
					11.1	100	5.2	85												
					2.85	100	4.9	92												
					0.79	100	4.6	71												
					).11	100	5.2	84												
					0.73	100	5.7	92												
					1.83	100	4.7	90												
					1.13	100	4,4	84												
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					1.12	100	4.5	85												
					0.73	100	4.4	90												
					1.43	100	4.8	100												
					).61	100	5.5	100												
					).20	100	3.2	70												
					).69	100	3.8	92												
	<b>62</b> 100	0 1.57		100 0	).50	100	5.6	92												
					).85	100	5,3	92												
	).88 100				).62	100	6.7	100												
					).15	100	5.8	92												
	66 100				).36	100	5.5	85												
					).60	100	S	96												
					).92	100	4.2	92												
					).62	100	5.8	92												
	15 100				).14	100	5.8	100												
					).89	100	4.1	100												
					1.43	98	6.4	80												
					.45	98	6.9	80												
_	12.31 92			92 0	0.63	92	9.9	61												
		Ŭ			0.07	100	5.6	84												
				-	).61	100	6.1	97												

Table S 15: Annual and seasonal mean mass concentrations of levoglucosan, mannosan and galactosan (2008–2018), arabitol, mannitol, trehalose, glucose, 2-methylerythritol and 2-

	Levo-	Cap	Man-	Cap	Galact-	Cap	Levo/	Cap	Ara-	Cap	Man-	Cap	2-methyl-	Cap	2-methyl-	Cap	Tre-	Cap	Glu-	Cap
	glucosan		nosan		0San		Mann		bitol		nitol		erythritol		threitol		halose		cose	
2016	7.52	76	1.21	76	0.30	94	5.7	92	4.78	66	3.89	66	0.38	66	0.13	66	3.17	66	5.07	66
DJF	12.54	94	1.87	94	0.47	94	6.4	93	1.70	94	1.09	94	0.02	94	0.01	94	0.76	94	1.47	94
MAM	7.62	100	1.25	100	0.31	100	5.9	100	4.67	100	3.40	100	0.21	100	0.11	100	5.82	100	4.60	100
JJA	2.37	100	0.47	100	0.09	77	5.3	100	8.29	100	6.69	100	1.01	100	0.29	100	3.73	100	9.93	100
SON	7.92	92	1.30	92	0.28	100	5.5	76	4.26	100	4.17	100	0.24	100	0.09	100	2.19	100	4.04	100
2017	8.24	94	1.35	95	0.32	94	5.6	92	5.52	94	5.70	94	0.37	94	0.10	94	3.37	94	5.17	94
DJF	15.77	100	2.49	100	0.65	100	5.6	100	1.16	100	1.40	100	0.01	100	0.01	100	1.03	100	2.03	100
MAM	6.22	92	0.98	92	0.23	92	9.9	84	3.90	92	3.65	92	0.05	92	0.02	92	1.58	92	4.11	92
JJA	2.27	86	0.48	86	0.06	86	4,8	86	9.15	86	8.45	86	1.32	86	0.35	86	4.21	86	7.09	86
SON	7.84	100	1.31	100	0.29	100	5.5	100	8.18	100	9.49	100	0.17	100	0.06	100	6.61	100	7.58	100
2018	9.77	100	1.62	100	0.39	100	6.1	98	5.76	100	5.65	100	0.45	98	0.16	98	2.83	100	4.16	100
DJF	13.50	100	2.21	100	0.60	100	5.9	100	0.72	100	0.94	100	0.01	92	0.01	92	0.60	100	2.74	100
MAM	13.64	100	2.04	100	0.54	100	6.8	100	4.18	100	4.21	100	0.24	100	0.08	100	1.91	100	3.59	100
JJA	1.38	100	0.25	100	0.04	100	5.8	100	8.66	100	7.94	100	1.24	100	0.42	100	2.90	100	4.52	100
SON	10.64	100	2.01	100	0.41	100	5.5	92	9.38	100	9.47	100	0.25	100	0.12	100	5.91	100	5.79	100

Table S 16: Seasonal mean ( $\pm$ SD) concentrations of TC<sub>bb</sub>, OC<sub>bb</sub> and EC<sub>bb</sub> in PM<sub>10</sub> and PM<sub>2.5</sub> at Birkenes 2008–2018. (Unit:  $\mu$ g C m<sup>-3</sup>).

	TCbb PM10	OCbb PM10	ECbb PM10	TCbb PM2.5	OCbb PM2.5	ECbb PM2.5
	TCbb PM10	OCbb PM10	ECbb PM10	<b>ТС</b> <sub>bb</sub> <b>РМ</b> <sub>2.5</sub>	OCbb PM2.5	ECbb PM2.5
2008	0.160±0.042	0.138±0.042	0.021±0.005	$\textbf{0.141} \pm \textbf{0.034}$	$0.121\pm0.034$	$0.020\pm0.004$
DJF	$0.150 \pm 0.039$	$0.130 \pm 0.039$	$0.020 \pm 0.004$	$0.133 \pm 0.032$	$0.114 \pm 0.032$	$0.019 \pm 0.004$
MAM	0.117±0.031	0.102±0.030	0.016±0.003	$0.104{\pm}0.025$	$0.089 \pm 0.025$	0.015±0.003
JJA	0.196±0.05	$0.170 \pm 0.051$	$0.026 \pm 0.006$	$0.174 \pm 0.042$	$0.149 \pm 0.042$	0.025±0.006
SON	0.166±0.043	$0.144 \pm 0.043$	$0.022 \pm 0.005$	$0.147 \pm 0.036$	0.126±0.036	0.021±0.005
2009	0.156±0.041	0.135±0.040	0.021±0.005	0.138±0.034	0.118±0.034	0.020±0.004
DJF	$0.301 \pm 0.078$	$0.261 \pm 0.078$	$0.040 \pm 0.009$	$0.266 \pm 0.065$	0.228±0.065	0.038±0.008
MAM	0.174±0.045	$0.150 \pm 0.045$	$0.023 \pm 0.005$	$0.154{\pm}0.037$	$0.131 \pm 0.037$	$0.022 \pm 0.003$
JJA	$0.028 \pm 0.007$	$0.024 \pm 0.007$	$0.004 \pm 0.001$	$0.025 \pm 0.006$	$0.021 \pm 0.006$	$0.004 \pm 0.00$
SON	0.133±0.035	0.115±0.035	$0.018 {\pm} 0.004$	$0.118 \pm 0.029$	$0.101 \pm 0.029$	$0.017 \pm 0.004$
2010	0.254±0.066	0.220±0.066	0.034±0.008	0.225±0.055	0.192±0.055	0.032±0.007
DJF	0.631±0.164	$0.547 \pm 0.164$	$0.084 \pm 0.019$	0.558±0.136	0.478±0.136	$0.080 \pm 0.018$
MAM	$0.150 \pm 0.039$	$0.130{\pm}0.039$	$0.020 \pm 0.004$	$0.132 \pm 0.032$	$0.113 \pm 0.032$	$0.019 \pm 0.004$
JJA	$0.043 \pm 0.011$	$0.038 \pm 0.011$	$0.006 \pm 0.001$	$0.038 \pm 0.009$	$0.033 \pm 0.009$	$0.006 \pm 0.001$
SON	0.201±0.052	$0.174 \pm 0.052$	$0.027 \pm 0.006$	$0.178 \pm 0.043$	0.152±0.043	0.026±0.000
2011	0.171±0.044	0.148±0.044	0.023±0.005	0.151±0.037	0.129±0.037	0.022±0.00
DJF	0.216±0.056	$0.187 \pm 0.056$	$0.029 \pm 0.006$	$0.191 {\pm} 0.047$	$0.164 \pm 0.047$	$0.027 \pm 0.000$
MAM	0.173±0.045	$0.150 \pm 0.045$	$0.023 \pm 0.005$	$0.153 {\pm} 0.037$	0.131±0.037	0.022±0.003
JJA	$0.059 \pm 0.015$	$0.051 \pm 0.015$	$0.008 \pm 0.002$	$0.052 \pm 0.013$	$0.044 \pm 0.013$	$0.007 \pm 0.002$
SON	$0.238 \pm 0.062$	$0.206 \pm 0.062$	$0.032 \pm 0.007$	0.210±0.051	$0.180 \pm 0.051$	$0.030 \pm 0.00^{\circ}$
2012	0.155±0.040	0.134±0.040	0.021±0.005	0.137±0.033	0.117±0.033	0.020±0.004
DJF	$0.319{\pm}0.083$	$0.276 \pm 0.083$	$0.043 \pm 0.009$	$0.282 \pm 0.069$	$0.241 \pm 0.069$	$0.040 \pm 0.009$
MAM	$0.156{\pm}0.041$	$0.136{\pm}0.041$	$0.021 \pm 0.005$	$0.138 \pm 0.034$	$0.118 \pm 0.034$	$0.020 \pm 0.004$
JJA	$0.026{\pm}0.007$	$0.022 \pm 0.007$	$0.003 {\pm} 0.001$	$0.023 \pm 0.006$	$0.019 \pm 0.006$	$0.003 {\pm} 0.00$
SON	$0.118{\pm}0.031$	$0.102{\pm}0.031$	$0.016 \pm 0.004$	$0.105 \pm 0.026$	$0.090 \pm 0.025$	$0.015 \pm 0.002$
2013	0.141±0.037	0.122±0.037	0.019±0.004	0.125±0.030	0.107±0.030	0.018±0.004
DJF	$0.267 \pm 0.069$	$0.231 \pm 0.069$	$0.036 \pm 0.008$	$0.236{\pm}0.058$	$0.202 \pm 0.058$	$0.034 \pm 0.003$
MAM	$0.159{\pm}0.041$	$0.138{\pm}0.041$	$0.021 \pm 0.005$	$0.141 \pm 0.034$	$0.121 \pm 0.034$	$0.020 \pm 0.004$
JJA	$0.040 \pm 0.010$	$0.035 \pm 0.010$	$0.005 \pm 0.001$	$0.035 {\pm} 0.009$	$0.030 \pm 0.009$	$0.005 \pm 0.00$
SON	$0.098 {\pm} 0.025$	$0.085 \pm 0.025$	$0.013 \pm 0.003$	$0.086 \pm 0.021$	$0.074 \pm 0.021$	$0.012 \pm 0.003$
2014	0.187±0.049	0.162±0.049	0.025±0.006	0.166±0.040	$0.142 \pm 0.040$	0.024±0.005
DJF	$0.295 {\pm} 0.077$	$0.256{\pm}0.077$	$0.040 \pm 0.009$	$0.261 \pm 0.064$	$0.224 \pm 0.064$	$0.037 \pm 0.003$
MAM	$0.182 \pm 0.047$	$0.158{\pm}0.047$	$0.024 \pm 0.005$	$0.161 \pm 0.039$	$0.138 \pm 0.039$	0.023±0.003
JJA	$0.046 \pm 0.012$	$0.040 \pm 0.012$	$0.006 \pm 0.001$	$0.041 \pm 0.010$	$0.035 {\pm} 0.010$	$0.006 \pm 0.00$
SON	$0.236 \pm 0.061$	$0.204{\pm}0.061$	$0.032 \pm 0.007$	$0.209{\pm}0.051$	$0.179 \pm 0.051$	$0.030 \pm 0.00^{\circ}$
2015	0.137±0.036	0.119±0.036	0.018±0.004	0.121±0.030	0.104±0.030	0.017±0.004
DJF	$0.144 \pm 0.038$	$0.125 \pm 0.038$	$0.019{\pm}0.004$	$0.128{\pm}0.031$	$0.109{\pm}0.031$	$0.018 \pm 0.004$
MAM	$0.180 \pm 0.047$	$0.156{\pm}0.047$	$0.024 \pm 0.005$	$0.160{\pm}0.039$	$0.137 \pm 0.039$	0.023±0.003
JJA	$0.033 {\pm} 0.009$	$0.029 \pm 0.009$	$0.004{\pm}0.001$	$0.029 \pm 0.007$	$0.025 \pm 0.007$	$0.004 \pm 0.001$
SON	$0.206 \pm 0.054$	$0.178 \pm 0.054$	$0.028 {\pm} 0.006$	$0.182 \pm 0.044$	$0.156 \pm 0.044$	0.026±0.000
2016	0.110±0.029	0.096±0.029	0.015±0.003	0.098±0.024	0.084±0.024	0.014±0.003
DJF	$0.184 \pm 0.048$	$0.159{\pm}0.048$	$0.025 \pm 0.005$	$0.162 \pm 0.040$	$0.139 \pm 0.040$	0.023±0.005
MAM	0.112±0.029	$0.097 \pm 0.029$	0.015±0.003	$0.099 \pm 0.024$	0.085±0.024	$0.014 \pm 0.003$

	TC <sub>bb</sub> PM <sub>10</sub>	OC <sub>bb</sub> PM <sub>10</sub>	EC <sub>bb</sub> PM <sub>10</sub>	TC <sub>bb</sub> PM <sub>2.5</sub>	OCbb PM2.5	EC <sub>bb</sub> PM <sub>2.5</sub>
JJA	$0.035 \pm 0.009$	$0.030 \pm 0.009$	$0.005{\pm}0.001$	$0.031{\pm}0.007$	$0.026{\pm}0.007$	$0.004{\pm}0.001$
SON	$0.116 \pm 0.030$	$0.101 \pm 0.030$	$0.016{\pm}0.003$	$0.103{\pm}0.025$	$0.088 {\pm} 0.025$	$0.015 {\pm} 0.003$
2017	0.121±0.031	0.105±0.031	0.016±0.004	0.107±0.026	0.092±0.026	0.015±0.003
DJF	$0.231 \pm 0.060$	$0.200 \pm 0.060$	$0.031{\pm}0.007$	$0.204{\pm}0.050$	$0.175 {\pm} 0.050$	$0.029{\pm}0.007$
MAM	$0.091 {\pm} 0.024$	$0.079 \pm 0.024$	$0.012{\pm}0.003$	$0.081 {\pm} 0.020$	$0.069 \pm 0.020$	$0.012{\pm}0.003$
JJA	$0.033 \pm 0.009$	$0.029{\pm}0.009$	$0.004{\pm}0.001$	$0.029{\pm}0.007$	$0.025 {\pm} 0.007$	$0.004{\pm}0.001$
SON	$0.115 \pm 0.030$	$0.100 \pm 0.030$	$0.015{\pm}0.003$	$0.102{\pm}0.025$	$0.087 \pm 0.025$	$0.015{\pm}0.003$
2018	0.143±0.037	0.124±0.037	0.019±0.004	0.127±0.031	0.108±0.031	$0.018 \pm 0.004$
DJF	$0.198 \pm 0.052$	$0.171 \pm 0.051$	$0.026{\pm}0.006$	$0.175 {\pm} 0.043$	$0.150{\pm}0.043$	$0.025{\pm}0.006$
MAM	$0.200 \pm 0.052$	$0.173 \pm 0.052$	$0.027 {\pm} 0.006$	$0.177 {\pm} 0.043$	$0.151 \pm 0.043$	$0.025 \pm 0.006$
JJA	$0.020 \pm 0.005$	$0.018 \pm 0.005$	$0.003{\pm}0.001$	$0.018{\pm}0.004$	$0.015 \pm 0.004$	$0.003{\pm}0.001$

57 Table S 17: Equations showing the relationship between EC, OC and TC for ambient PM<sub>2.5</sub> aerosol filter samples 58 collected at Birkenes in 2014, obtained by temperature calibrated Quartz and EUSAAR-2 temperature programs.

EC	$EC_{EUSAAR-2, TOT} = EC_{QUARTZ, TOT} \times 1.6118$	$R^2 = 0.876$	n = 50	Eq. S 16
OC	$OC_{EUSAAR-2, TOT} = OC_{QUARTZ, TOT} \ge 0.8687$	$R^2 = 0.977$	n = 50	Eq. S 17
TC	$TC_{EUSAAR-2, TOT} = TC_{QUARTZ, TOT} \times 0.9151$	$R^2 = 0.976$	n = 50	Eq. S 18