

Supplement of

Measurement report: Source characteristics of water-soluble organic carbon in PM_{2.5} at two sites in Japan, as assessed by long-term observation and stable carbon isotope ratio

Nana Suto and Hiroto Kawashima

Correspondence to: Nana Suto (nsuto@jari.or.jp) and Hiroto Kawashima (kawashima@akita-pu.ac.jp)

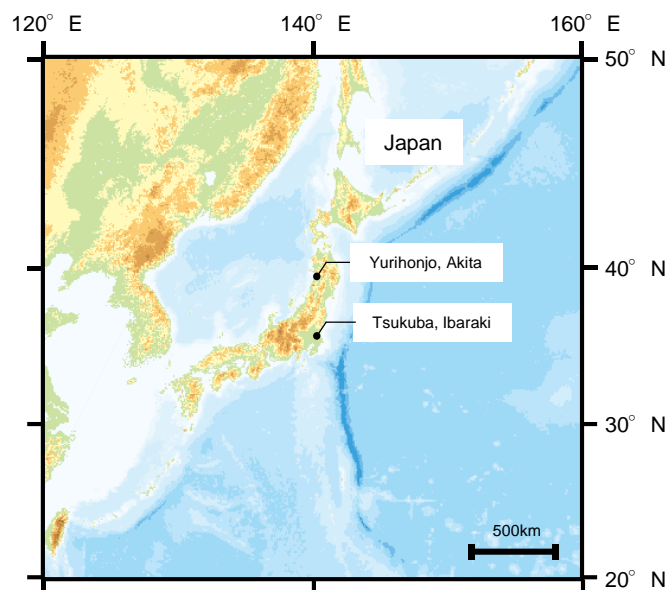


Figure S1. Sampling locations at the Japan Automobile Research Institute (Tsukuba, Ibaraki) and Akita Prefectural University (Yurihonjo, Akita). This map plots the sampling locations on a standard map provided by the Geospatial Information Authority of Japan.

Table S1. $\delta^{13}\text{C}_{\text{wsoc}}$ in $\text{PM}_{2.5}$ determined in the present and other published studies.

Sampling site	Site type	Particle size	Sampling period	<i>n</i>	Analytical method	$\delta^{13}\text{C}_{\text{wsoc}}$ (‰)		Reference
						Average \pm SD	Range	
Tsukuba, Japan	Suburban	$\text{PM}_{2.5}$	Jul 2017–Jul 2019	62	Wet oxidation/IRMS	-25.2 ± 1.1	-26.7 to -21.8	This study
Yurihonjo, Japan	Rural	$\text{PM}_{2.5}$	Aug 2017–Jul 2019	45	Wet oxidation/IRMS	-24.6 ± 2.4	-28.4 to -19.8	This study
Delhi, India	Urban	$\text{PM}_{2.5}$	Jan–Mar 2016	7	Combustion-EA/IRMS	-25.4 ± 1.0	–	Dasari et al. (2019)
Bhola, Bangladesh	Rural	$\text{PM}_{2.5}$	Jan–Mar 2016	12	Combustion-EA/IRMS	-24.2 ± 0.6	–	Dasari et al. (2019)
Hanimaadhoo, Maldives	Rural	$\text{PM}_{1.0}$	Jan–Mar 2016	15	Combustion-EA/IRMS	-20.9 ± 0.6	–	Dasari et al. (2019)
Seoul, Korea	Urban	TSP	Mar 2015–Jan 2016	78	TOC analyzer/IRMS	-24.0 ± 1.5	-27.5 to -21.0	Han et al. (2020)
Nanjing, China	Suburban	$\text{PM}_{2.5}$	Jan 2015	–	GasBench/IRMS	–	-26.24 to -23.35	Zhang et al. (2019)
Beijing, China	Urban	$\text{PM}_{2.5}$	Jan, Jun, 2013	10	Combustion-EA/IRMS	-22.51 ± 0.49	–	Yan et al. (2017)
						-25.40 ± 0.46	–	Yan et al. (2017)
Hanimaadhoo, Maldives	Rural	$\text{PM}_{2.5}$	Feb–Mar 2012	14	Combustion-EA/IRMS	-20.8 ± 0.7	-22.13 to -19.64	Bosch et al. (2014)
Jeju, Korea	Rural	TSP/ $\text{PM}_{2.5}$	Mar 2011	10	Combustion-EA/IRMS	–	–	Kirillova et al. (2014a)
New Delhi, India	Urban	$\text{PM}_{2.5}$	Oct 2010–Mar 2011	20	Combustion-EA/IRMS	-24.1 ± 0.9	-26.3 to -22.4	Kirillova et al. (2014b)
Sapporo, Japan	Rural	TSP	Sep 2009–Oct 2010	21	Combustion-EA/IRMS	-24.2 ± 1.59	-26.7 to -21.2	Pavuluri and Kawamura (2017)
Sapporo, Japan	Forest	TSP	Jun 2009–Dec 2010	–	Combustion-EA/IRMS	–	–	Miyazaki et al. (2012)
Stockholm, Sweden	Forest	TSP	Aug–Oct 2009	3	Combustion-EA/IRMS	–	-25.6 to -25.1	Kirillova et al. (2010)
Hanimaadhoo, Maldives	Rural	TSP	Jan 2008–Apr 2009	12	Combustion-EA/IRMS	-18.4 ± 0.5	-20.8 to -17.5	Kirillova et al. (2013)
Sinhagad, India	Rural	TSP	Jan 2008–Apr 2009	12	Combustion-EA/IRMS	-20.4 ± 0.5	-23.7 to -19.8	Kirillova et al. (2013)
Millbrook, USA	Rural	TSP	Mar, May, Aug 2007	3	Combustion-EA/IRMS	-24.7	-25.1 to -24.4	Wozniak et al. (2012b)
Harcum, USA	Rural	TSP	Feb, Apr, Aug 2007	3	Combustion-EA/IRMS	-25.4	-26.1 to -24.8	Wozniak et al. (2012b)
Millbrook, USA	Rural	TSP	May 2006–May 2007	9	Combustion-EA/IRMS	-25.2 ± 0.2	-26.0 to -23.9	Wozniak et al. (2012a)
Harcum, USA	Rural	TSP	Jun 2006–Jun 2007	10	Combustion-EA/IRMS	-25.3 ± 0.6	-27.4 to -21.1	Wozniak et al. (2012a)

Supplement S1 Water-soluble ion analysis

A portion of each quartz fiber filter (1.44 cm²) was extracted in 3 mL of Milli-Q water under ultrasonic agitation for 15 min. The extract was filtered through a syringe filter (Chromatodisc Type A 0.45 μm, GL Sciences, Japan) to remove insoluble materials. Anion concentrations were determined in the filtrate using an IonPac AS17-C column and IonPac AG17-C guard column (Thermo Fisher Scientific Inc.), with a 1–40 mM gradient of potassium hydroxide as the eluent. Cation concentrations were determined using a CS12A column and CG12A guard column (Thermo Fisher Scientific Inc.), with 20 mM methanesulfonic acid as the eluent. Calibration curves were prepared using cation mixed standard solution 2 and anion mixed standard solution 4 (Kanto Chemical Co., Inc., Tokyo, Japan). The coefficient of determination was >0.999 for all compounds, and the detection limits were 4 ppb (Cl⁻), 5 ppb (NO₂⁻), 12 ppb (NO₃⁻), 8 ppb (SO₄²⁻), 8 ppb (Na⁺), 5 ppb (NH₄⁺), 15 ppb (K⁺), 15 ppb (Mg²⁺), and 20 ppb (Ca²⁺). These values were comparable to those used in other research (Shen et al., 2009).

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