

# Source Apportionment of Fine Organic Carbon at an Urban Site of Beijing using a Chemical Mass Balance Model

Jingsha Xu<sup>1</sup>, Di Liu<sup>1,2</sup>, Xuefang Wu<sup>1,3</sup>, Tuan V. Vu<sup>1,4</sup>, Yanli Zhang<sup>5</sup>, Pingqing Fu<sup>6</sup>, Yele Sun<sup>7</sup>, Weiqi Xu<sup>7</sup>, Bo Zheng<sup>8,9</sup>, Roy M. Harrison<sup>1,\*</sup>, Zongbo Shi<sup>1,\*</sup>

1 School of Geography Earth and Environmental Science, University of Birmingham, Birmingham, B15 2TT, UK

2 Now at: Institute of Atmospheric Physics, Chinese Academy of Sciences, Beijing, 100029, China

3 School of Geology and Mineral Resources, China University of Geosciences Xueyuan Road 29, Beijing, 100083, China

4 Now at: Faculty of Life Sciences & Medicine, King's College London, London, WC2R 2LS, UK

5 Guangzhou Institute of Geochemistry, Chinese Academy of Sciences, Guangzhou, 510640, China

6 Institute of Surface-Earth System Science, Tianjin University, Tianjin, 300072, China

7 State Key Laboratory of Atmospheric Boundary Layer Physics and Atmospheric Chemistry, Institute of Atmospheric Physics, Chinese Academy of Sciences, Beijing, 100029, China

8 State Key Joint Laboratory of Environment Simulation and Pollution Control, School of Environment, Tsinghua University, Beijing, 100084, China

9 Now at: Laboratoire des Sciences du Climat et de l'Environnement, CEA-CNRS-UVSQ, UMR8212, Gif-sur-Yvette, France

\*Correspondence: Zongbo Shi (Z.Shi@bham.ac.uk) and Roy Harrison (r.m.harrison@bham.ac.uk)

## Abstract

Fine particles were sampled from 9<sup>th</sup> November to 11<sup>th</sup> December 2016 and 22<sup>nd</sup> May to 24<sup>th</sup> June 2017 as part of the Atmospheric Pollution and Human Health in a Chinese megacity (APHH-China) field campaigns in urban Beijing, China. Inorganic ions, trace elements, OC, EC, and organic compounds including biomarkers, hopanes, PAHs, n-alkanes and fatty acids, were determined for source apportionment in this study. Carbonaceous components contributed on average 47.2% and 35.2% of total reconstructed PM<sub>2.5</sub> during the winter and summer campaigns, respectively. Secondary inorganic ions (sulfate, nitrate, ammonium- SNA) accounted for 35.0% and 45.2% of total PM<sub>2.5</sub> in winter and summer. Other components including inorganic ions (K<sup>+</sup>, Na<sup>+</sup>, Cl<sup>-</sup>), geological minerals, and trace metals only contributed 13.2% and 12.4% of PM<sub>2.5</sub> during the winter and summer campaigns. Fine OC was explained by seven primary sources (industrial/residential coal burning, biomass burning, gasoline/diesel vehicles, cooking and vegetative detritus) based on a chemical mass balance (CMB) receptor model. It explained an average of 75.7% and 56.1% of fine OC in winter and summer, respectively. Other (unexplained) OC was compared with the secondary OC (SOC) estimated by the EC-tracer method, with correlation coefficients (R<sup>2</sup>) of 0.58 and 0.73, and slopes of 1.16 and 0.80 in winter and summer, respectively. This suggests that the unexplained OC by CMB was mostly associated with SOC. PM<sub>2.5</sub> apportioned by CMB showed that the SNA and secondary organic matter were the highest two contributors to PM<sub>2.5</sub>. After these, coal combustion and biomass burning were also significant sources of PM<sub>2.5</sub> in winter. The CMB results were also compared with results from Positive Matrix Factorization (PMF) analysis of co-located Aerosol Mass Spectrometer (AMS) data. The CMB was found to resolve more primary OA sources than AMS-PMF but the latter could apportion secondary OA sources. The AMS-PMF results for major components, such as coal combustion OC and oxidized OC correlated well with the results from CMB. However, discrepancies and poor agreements were found for other OC sources, such as biomass burning and cooking, some of which were not identified in AMS-PMF factors.

51

52 **Keywords:** PM<sub>2.5</sub>, Beijing, mass closure, CMB, AMS-PMF, source apportionment

## 53 **1 Introduction**

54 Beijing is the capital of China and a hotspot of particulate matter pollution. It has been  
55 experiencing severe PM<sub>2.5</sub> (particulate matter with an aerodynamic diameter of  
56  $\leq 2.5\mu\text{m}$ ) pollution in recent decades, as a result of rapid urbanization and  
57 industrialization, and increasing energy consumption (Wang et al., 2009). High PM<sub>2.5</sub>  
58 pollution from Beijing could have significant impact on human health (Song et al.,  
59 2006a; Li et al., 2013). A case study in Beijing revealed that a  $10\ \mu\text{g m}^{-3}$  increase of  
60 ambient PM<sub>2.5</sub> concentration will correspondingly increase 0.78%, 0.85% and 0.75%  
61 of the daily mortality of the circulatory diseases, cardiovascular diseases and  
62 cerebrovascular diseases, respectively (Dong et al., 2013). Furthermore, PM<sub>2.5</sub> causes  
63 visibility deterioration in Beijing. A better understanding of PM<sub>2.5</sub> sources in Beijing is  
64 essential, as it can provide important scientific evidence to develop measures to control  
65 PM<sub>2.5</sub> pollution.

66 Many studies have identified the possible sources of fine particulate matter in Beijing  
67 using various methods (Zheng et al., 2005; Song et al., 2006a; Song et al., 2006b; Li et  
68 al., 2015; Zhang et al., 2013; Yu and Wang, 2013). Song et al. (2006a) applied  
69 two eigenvector models, principal component analysis/absolute principal component  
70 scores (PCA/APCS) and UNMIX to study the sources of PM<sub>2.5</sub> in Beijing. Some studies  
71 used elemental tracers to do source apportionment of PM<sub>2.5</sub> by applying positive matrix  
72 factorization (PMF) (Song et al., 2006b; Li et al., 2015; Zhang et al., 2013; Yu and  
73 Wang, 2013). This approach has some underlying challenges. For example, PMF  
74 requires a relatively large sample size and a “best” solution of achieved factors requires  
75 critical assessment of its mathematical parameters and evaluation of the physical  
76 reasonability of the factor profiles (de Miranda et al., 2018; Ikemori et al., 2021; Oduber  
77 et al., 2021); secondly, many important PM<sub>2.5</sub> emission sources do not have a unique  
78 elemental composition. Hence, an elemental tracer-based method cannot distinguish  
79 sources such as cooking or vehicle exhaust, as they emit mainly carbonaceous  
80 compounds (Wang et al., 2009). Generally, organic matter (OM) is composed of  
81 primary organic matter (POM) and secondary organic matter (SOM). POM is directly  
82 emitted and SOM is formed through chemical oxidation of volatile organic compounds  
83 (VOCs) (Yang et al., 2016). OM was the largest contributor to PM<sub>2.5</sub> mass, which was  
84 reported to account for 30%-50% of PM<sub>2.5</sub> in some Chinese cities such as Beijing,  
85 Guangzhou, Xi’an and Shanghai (Song et al., 2007; He et al., 2001; Huang et al., 2014),  
86 and can contribute up to 90% of submicron PM mass in Beijing (Zhou et al., 2018).  
87 Furthermore, many organic tracers are more specific to particular sources, making them  
88 more suitable to identify and quantify different source contributions to carbonaceous  
89 aerosols and PM<sub>2.5</sub>.

90 Chemical Mass balance (CMB) model has been used for source apportionment of  
91 PM worldwide, including in the US (Antony Chen et al., 2010), UK (Yin et al., 2015),  
92 and China (Chen et al., 2015b). The CMB model assumes that source profiles remain  
93 unchanged between the emitter and receptor (Sarnat et al., 2008; Viana et al., 2008).  
94 The good performance of CMB and its comparability with other receptor modelling  
95 techniques was demonstrated in an intercomparison exercise conducted in Beijing (Xu  
96 et al., 2021). A few studies also have applied a CMB model for source apportionment  
97 of PM in Beijing (Zheng et al., 2005; Liu et al., 2016; Guo et al., 2013; Wang et al.,  
98 2009). For example, Zheng et al. (2005) investigated sources of PM<sub>2.5</sub> in Beijing, but

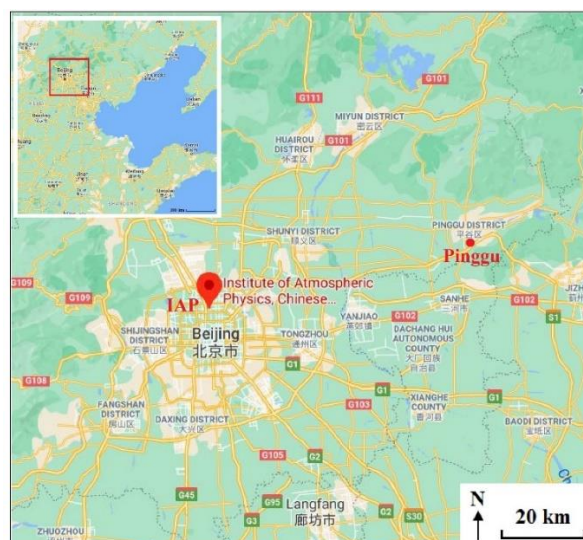
99 the source profiles they used were mainly derived in the United States, which were less  
100 representative of the local sources. Liu et al. (2016) and Guo et al. (2013) apportioned  
101 the sources of PM<sub>2.5</sub> in a typical haze episode in winter 2013 in Beijing during the  
102 Olympic Games period in summer 2008, respectively. Wang et al. (2009) apportioned  
103 the sources of PM<sub>2.5</sub> in both winter and summer. A major challenge of the CMB model  
104 is that it cannot quantify the contributions of secondary organic aerosol and unknown  
105 sources, which are often lumped as “unexplained OC”.

106 In this study, PM<sub>2.5</sub> samples were collected at an urban site of Beijing in winter 2016  
107 and summer 2017. OC, EC, PAHs, alkanes, hopanes, fatty acids and monosaccharide  
108 anhydrides in the PM<sub>2.5</sub> samples were determined, and applied in the CMB model for  
109 apportioning the organic carbon sources. To ensure that the source profiles used in the  
110 CMB model are representative, we mainly selected data which had been determined in  
111 China. The objectives of this study are: 1) to quantify the contributions of pollution  
112 sources to OC by applying a CMB model and compare them with those at a rural site  
113 of Beijing; 2) to compare the source apportionment results by CMB with those from  
114 Aerosol Mass Spectrometer-PMF analysis (AMS-PMF), to improve our understanding  
115 of different sources of OC.

## 116 **2 Methodology**

### 117 **2.1 Aerosol sampling**

118 PM<sub>2.5</sub> was collected at an urban sampling site (116.39E, 39.98N) - the Institute of  
119 Atmospheric Physics (IAP) of the Chinese Academy of Sciences in Beijing, China from  
120 9<sup>th</sup> November to 11<sup>th</sup> December 2016 and 22<sup>nd</sup> May to 24<sup>th</sup> June 2017, as part of the  
121 Atmospheric Pollution and Human Health in a Chinese megacity (APHH-China) field  
122 campaigns (Shi et al., 2019). The sampling site (Fig. 1) is located in the middle between  
123 the North 3<sup>rd</sup> Ring Road and North 4<sup>th</sup> Ring Road and approximately 200 m from a  
124 major highway. Hence, it is subject to many local sources, such as traffic, cooking, etc.  
125 The location of a rural site in Beijing - Pinggu during the APHH-China campaigns is  
126 also shown in Fig. 1. The rural site in Xibaidian village in Pinggu is about 60 km away  
127 from IAP and 4 km north-west of the Pinggu town centre. It is surrounded by trees and  
128 farmland with several similar small villages nearby. A provincial highway is  
129 approximately 500 m away on its eastside running north-south. This site is far from  
130 industrial sources and located in a residential area. Other information regarding the  
131 sampling site is described elsewhere (Shi et al., 2019).



132

133 **Figure 1.** Locations of the sampling sites in Beijing (IAP - urban site: Institute of Atmospheric Physics  
 134 of the Chinese Academy of Sciences; Pinggu - rural site) (source: © Google Maps).

135 PM<sub>2.5</sub> samples were collected on pre-baked (450°C for 6h) large quartz filters  
 136 (Pallflex, 8×10 inch) by Hi-Vol air sampler (Tisch, USA) at a flow rate of 1.1 m<sup>3</sup> min<sup>-1</sup>.  
 137 A Medium-Vol air sampler (Thermo Scientific Partisol 2025i) was also deployed at  
 138 the same location to collect PM<sub>2.5</sub> samples simultaneously on 47 mm PTFE filters at a  
 139 flow rate of 15.0 L min<sup>-1</sup>. Field blanks were also collected with the pump turned off  
 140 during the sampling campaign. Before and after sampling, all filters were put in a  
 141 balance room and equilibrated at a constant temperature and relative humidity (RH) for  
 142 24h prior to any gravimetric measurements, which were 22°C and 30% RH for summer  
 143 samples, 21°C and 33% RH for winter samples. PM<sub>2.5</sub> mass was determined through  
 144 the weighing of PTFE filters using a microbalance (Sartorius model MC5, precision: 1  
 145 µg). After that, filters were wrapped separately with aluminum foil and stored at under  
 146 -20°C in darkness until analysis. The large quartz filters were analyzed for OC, EC,  
 147 organic compounds and ion species, while small PTFE filters were used for the  
 148 determination of PM<sub>2.5</sub> mass and metals. Online PM<sub>2.5</sub> were determined by the TEOM  
 149 FDMS 1405-DF instrument at IAP with filter equilibrating and weighing conditions  
 150 comparable with the United States Federal Reference Method (RH: 30-40%;  
 151 temperature; 20-23°C) (Le et al., 2020; U.S.EPA, 2016).

152 **2.2 Chemical Analysis**

153 **2.2.1 OC and EC**

154 A 1.5 cm<sup>2</sup> punch from each large quartz filter sample was taken for organic carbon (OC)  
 155 and elemental carbon (EC) measurements by a thermal/optical carbon analyzer (model  
 156 RT-4, Sunset Laboratory Inc., USA) based on the EUSAAR2 (European Supersites for  
 157 Atmospheric Aerosol Research) transmittance protocol (Cavalli et al., 2010; Chen et al.,  
 158 2015a). Replicate analyses of OC and EC were conducted once every ten samples. The  
 159 uncertainties from duplicate analyses of filters were <10%. All sample results were  
 160 corrected by the values obtained from field blanks, which were 0.40 and 0.01 µg m<sup>-3</sup>  
 161 for OC and EC, respectively. Details of the OC/EC measurement method can be found  
 162 elsewhere (Paraskevopoulou et al., 2014). The instrumental limits of detection of OC

163 and EC in this study were estimated to be 0.03 and 0.05  $\mu\text{g m}^{-3}$ , respectively.  
164

### 165 **2.2.2 Organic compounds**

166 Organic tracers, including 11 n-alkanes ( $\text{C}_{24}$ - $\text{C}_{34}$ ), 2 hopanes (17a (H) -22, 29, 30-  
167 Trisnorhopane, 17b (H), 21a (H) -Norhopane), 17 PAHs (retene, phenanthrene,  
168 anthracene, fluoranthene, pyrene, benz(a)anthracene, chrysene, benzo(b)fluoranthene,  
169 benzo(k)fluoranthene, benzo(e)pyrene, benzo(a)pyrene, perylene, Indeno(1,2,3-  
170 cd)pyrene, dibenz(a,h)anthracene, benzo(ghi)perylene, coronene, picene), 3  
171 anhydrosugars (levoglucosan, mannosan, galactosan), 2 fatty acids (palmitic acid,  
172 stearic acid) and cholesterol in the  $\text{PM}_{2.5}$  samples were determined in this study.  $9\text{ cm}^2$   
173 of the large quartz filters were extracted 3 times with dichloromethane/methanol (HPLC  
174 grade, v/v: 2:1) under ultrasonication for 10 minutes. The extracts were then filtered  
175 and concentrated using a rotary evaporator under vacuum, and blown down to dryness  
176 with pure nitrogen gas.  $50\ \mu\text{L}$  of N,O-bis-(trimethylsilyl)trifluoroacetamide (BSTFA)  
177 with 1% trimethylsilyl chloride and  $10\ \mu\text{L}$  of pyridine were then added to the extracts,  
178 which were left reacting at  $70\ ^\circ\text{C}$  for 3 h to derivatize -COOH to TMS esters and -OH  
179 to TMS ethers. After cooling to room temperature, the derivatives were diluted with  
180  $140\ \mu\text{L}$  of internal standards ( $\text{C}_{13}$  n-alkane,  $1.43\ \text{ng}\ \mu\text{L}^{-1}$ ) in n-hexane prior to GC-MS  
181 analysis. The final solutions were analyzed by a gas chromatography mass spectrometry  
182 system (GC/MS, Agilent 7890A GC plus 5975C mass-selective detector) fitted with a  
183 DB-5MS column ( $30\ \text{m} \times 0.25\ \text{mm} \times 0.25\ \mu\text{m}$ ). The GC temperature program and MS  
184 detection details were reported in Li et al. (2018). Individual compounds were identified  
185 through the comparison of mass spectra with those of authentic standards or literature  
186 data (Fu et al., 2016). Recoveries for these compounds were in a range of 70-100%,  
187 which were obtained by spiking standards to pre-baked blank quartz filters followed by  
188 the same extraction and derivatization procedures. Field blank filters were analyzed the  
189 same way as samples for quality assurance, but no target compounds were detected.

### 190 **2.2.3 Inorganic components**

191 Half of the PTFE filter was extracted with 10 mL ultrapure water for the analysis of  
192 inorganic ions. Major inorganic ions including  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{NH}_4^+$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  were  
193 determined by using an ion chromatograph (IC, Dionex, Sunnyvale, CA, USA), the  
194 detection limits (DLs) of them were 0.032, 0.010, 0.011, 0.076, 0.138, 0.240 and 0.142  
195  $\mu\text{g m}^{-3}$  respectively. The analytical uncertainty was less than 5% for all inorganic ions.  
196 An intercomparison study showed that our IC analysis of the above-mentioned ions  
197 agreed well with those of the other laboratories (Xu et al., 2020). Trace metal including  
198 Al (DLs in  $\mu\text{g m}^{-3}$ , 0.221), Si (0.040), Ca (0.034), Ti (0.003) and Fe (0.044) were  
199 determined by X-ray fluorescence spectrometer (XRF). Other elements including V, Cr,  
200 Co, Mn, Ni, Cu, Zn, As, Sr, Cd, Sb, Ba and Pb were analyzed by Inductively-coupled  
201 plasma-mass spectrometer (ICP-MS) after extraction of 1/2 PTFE filter by diluted acid  
202 mixture ( $\text{HNO}_3/\text{HCl}$ ), and the detection limits of them were 1.32, 0.25, 0.04, 0.06, 2.05,  
203 1.25, 1.22, 1.74, 0.02, 0.03, 0.11, 0.06 and  $0.04\ \text{ng m}^{-3}$ , respectively. Mass  
204 concentrations of all inorganic ions and elements in this study were corrected for the  
205 field blank values, and the methods were quality assured with standard reference  
206 materials.

207

### 208 **2.3 Chemical Mass Closure (CMC) Method**

209 A Chemical Mass Closure analysis was carried out, which includes secondary inorganic  
210 ions (sulfate, nitrate, ammonium; SNA), sodium, potassium and chloride salts,  
211 geological minerals, trace elements, organic matter (OM), EC and bound water in  
212 reconstructed PM<sub>2.5</sub>. Geological minerals were calculated applying the equation (Eq. 1)  
213 (Chow et al., 2015):

$$214 \text{ Geological minerals} = 2.2\text{Al} + 2.49\text{Si} + 1.63\text{Ca} + 1.94\text{Ti} + 2.42\text{Fe} \quad (1)$$

215 Trace elements were the sum of all analysed elements excluding Al, Si, Ca, Ti and  
216 Fe. The average OM/OC ratios of organic aerosols (OA) from AMS elemental analysis  
217 were applied to calculate OM, which were 1.75±0.16 and 2.00±0.19 in winter and  
218 summer, respectively. Based on the concentrations of inorganic ions and gas-phase NH<sub>3</sub>,  
219 particle bound water was calculated by ISORROPIA II model (available  
220 at <http://isorro피아.eas.gatech.edu>) in forward mode and thermodynamically metastable  
221 phase state (Fountoukis and Nenes, 2007). Two sets of calculations were done for online  
222 and offline data, differing at the temperature and relative humidity as specified above.

### 223 **2.4 Chemical Mass Balance (CMB) model**

224 The chemical mass balance model (US EPA CMB8.2) was applied in this study to  
225 apportion the sources of OC by utilizing a linear least squares solution. Both  
226 uncertainties in source profiles and ambient measurements were taken into  
227 consideration in this model. The source profiles applied here were from local studies in  
228 China to better represent the source characteristics, including straw burning (wheat,  
229 corn, rice straw burning) (Zhang et al., 2007b), wood burning (Wang et al., 2009),  
230 gasoline and diesel vehicles (including motorcycles, light- and heavy-duty gasoline and  
231 diesel vehicles) (Cai et al., 2017), industrial and residential coal combustion (including  
232 anthracite, sub-bituminite, bituminite, and brown coal) (Zhang et al., 2008), and  
233 cooking (Zhao et al., 2015), except vegetative detritus (Rogge et al., 1993; Wang et al.,  
234 2009). The source profiles with EC and organic tracers used in the CMB model were  
235 provided in Table S1 of Wu et al. (2020). The selected fitting species were EC,  
236 levoglucosan, palmitic acid, stearic acid, fluoranthene, phenanthrene, retene,  
237 benz(a)anthracene, chrysene, benzo(b)fluoranthene, benzo(k)fluoranthene,  
238 benzo[ghi]perylene, picene, 17a (H) -22, 29, 30-trisnorhopane, 17b (H), 21a (H) -  
239 norhopane and n-alkanes (C24-C33), the concentrations of which are provided in Table  
240 1. The essential criteria in this model were met to ensure reliable fitting results. For  
241 instance, in all samples, R<sup>2</sup> were >0.80 (mostly >0.9), Chi<sup>2</sup> were <2, T<sub>stat</sub> values were  
242 mostly greater than 2 except the source of vegetative detritus, and C/M ratios (ratio of  
243 calculated to measured concentration) for all fitting species were in range of 0.8-1.2 in  
244 this study.

245

### 246 **2.5 Positive Matrix Factorization analysis of data obtained from Aerosol Mass** 247 **Spectrometer (AMS-PMF)**

248 An Aerodyne AMS with a PM<sub>1</sub> aerodynamic lens was deployed on the roof of the  
249 neighboring building- the Tower branch of IAP for real-time measurements of non-

250 refractory (NR) chemical species from 16<sup>th</sup> November to 11<sup>th</sup> December 2016 and 22<sup>nd</sup>  
251 May to 24<sup>th</sup> June 2017. The detailed information of the sampling sites is given  
252 elsewhere (Xu et al., 2019b). The submicron particles were dried and sampled into the  
253 AMS at a flow of  $\sim 0.1 \text{ L min}^{-1}$ . NR-PM<sub>1</sub> can be quickly vaporized by the 600 °C  
254 tungsten vaporizer and then the NR-PM<sub>1</sub> species including organics, Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>  
255 and NH<sub>4</sub><sup>+</sup> were measured by AMS in mass sensitive V mode (Sun et al., 2020). Details  
256 of AMS data analysis, including the analysis of organic aerosol (OA) mass spectra can  
257 be found elsewhere (Xu et al., 2019b). The source apportionment of organics in NR-  
258 PM<sub>1</sub> was carried out by applying PMF to the high-resolution mass spectra of OA, while  
259 that of fine OC in this study was conducted by applying source profiles along with an  
260 offline chemical speciation dataset. The procedures of the pretreatment of spectral data  
261 and error matrices can be found elsewhere (Ulbrich et al., 2009). It is noted that the data  
262 were missing during the period 09<sup>th</sup> - 15<sup>th</sup> November 2016 due to the malfunction of the  
263 AMS.

264

### 265 **3 Results and discussion**

#### 266 **3.1 Characteristics of PM<sub>2.5</sub> and Carbonaceous Compounds**

267 Mean concentrations of PM<sub>2.5</sub>, OC, EC and organic tracers during wintertime (9<sup>th</sup>  
268 November to 11<sup>th</sup> December 2016) and summertime (22<sup>nd</sup> May to 24<sup>th</sup> June 2017) at the  
269 IAP site are summarized in Table 1 and Fig. S1. The average PM<sub>2.5</sub> concentration was  
270  $94.8 \pm 64.4 \mu\text{g m}^{-3}$  during the whole winter sampling campaign. The winter sampling  
271 period was divided into haze (daily PM<sub>2.5</sub> >  $75 \mu\text{g m}^{-3}$ ) and non-haze days (< $75 \mu\text{g m}^{-3}$ ),  
272 based on the National Ambient Air Quality Standard Grade II of the limit for 24-  
273 hour average PM<sub>2.5</sub> concentration. The differentiation between haze and non-haze days  
274 enabled us to study the major sources contributing to the haze formation. The average  
275 daily PM<sub>2.5</sub> was  $136.7 \pm 49.8$  and  $36.7 \pm 23.5 \mu\text{g m}^{-3}$  on haze and non-haze days,  
276 respectively. Daily PM<sub>2.5</sub> in the summer sampling period was  $30.2 \pm 14.8 \mu\text{g m}^{-3}$ ,  
277 comparable with that on winter non-haze days.

278 OC concentrations ranged between  $3.9\text{-}48.8 \mu\text{g m}^{-3}$  (mean:  $21.5 \mu\text{g m}^{-3}$ ) and  $1.8\text{-}12.7$   
279  $\mu\text{g m}^{-3}$  (mean:  $6.4 \mu\text{g m}^{-3}$ ) during winter and summer, respectively. They are  
280 comparable with the OC concentrations in winter ( $23.7 \mu\text{g m}^{-3}$ ) and summer ( $3.78 \mu\text{g m}^{-3}$ )  
281 in Tianjin, China during an almost simultaneous sampling period (Fan et al., 2020),  
282 but much lower than the OC concentration ( $17.1 \mu\text{g m}^{-3}$ ) in summer 2007 in Beijing  
283 (Yang et al., 2016). The average OC concentration during haze days ( $29.4 \pm 9.2 \mu\text{g m}^{-3}$ )  
284 was approximately three times that of non-haze days ( $10.7 \pm 6.2 \mu\text{g m}^{-3}$ ) during winter.  
285 The average EC concentration during winter was  $3.5 \pm 2.0 \mu\text{g m}^{-3}$ ; its concentration was  
286  $4.6 \pm 1.3 \mu\text{g m}^{-3}$  on haze days, approximately 2.4 times that on winter non-haze days  
287 ( $1.9 \pm 1.6 \mu\text{g m}^{-3}$ ) and 5 times that ( $0.9 \pm 0.4 \mu\text{g m}^{-3}$ ) during the summer sampling period.  
288 The OC and EC concentrations in this study were comparable with the OC ( $27.9 \pm 23.4$   
289  $\mu\text{g m}^{-3}$ ) and EC ( $6.6 \pm 5.1 \mu\text{g m}^{-3}$ ) concentrations in winter Beijing in 2016 (Qi et al.,  
290 2018), but much lower than those in an urban area of Beijing during winter (OC and  
291 EC:  $36.7 \pm 19.4$  and  $15.2 \pm 11.1 \mu\text{g m}^{-3}$ ) and summer ( $10.7 \pm 3.6$  and  $5.7 \pm 2.9 \mu\text{g m}^{-3}$ ) in  
292 2002 (Dan et al., 2004).



293 On average, OC and EC concentrations in winter were 3.3 and 3.9 times those in  
 294 summer. Additionally, OC and EC were well-correlated in this study, with  $R^2$  values of  
 295 0.85 and 0.63 during winter and summer, respectively, suggesting similar paths of OC  
 296 and EC dispersion and dilution, and/or similar sources of carbonaceous aerosols, especially  
 297 in winter. Less correlated OC and EC in summer could be a result of SOC formation.  
 298 SOC in this study was estimated and is discussed in section 3.3.7.

299 **Table 1.** Summary of measured concentrations at IAP site in winter and summer.

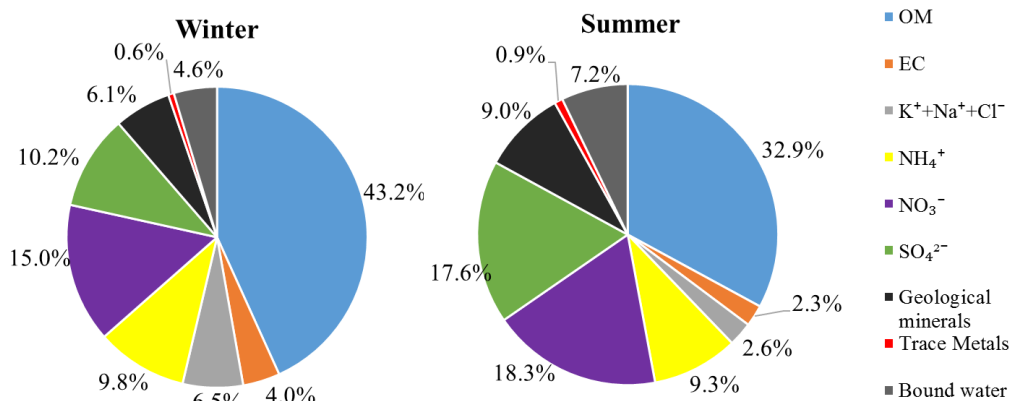
Compounds <sup>a</sup> / ng m <sup>-3</sup>	Winter		Winter (n=31)	Summer (n=34)
	Haze <sup>d</sup> (n=18)	Non-haze <sup>e</sup> (n=13)		
PM <sub>2.5</sub> (µg m <sup>-3</sup> )	136.7±49.8 (80.5-239.9) <sup>b</sup>	36.7±23.5 (10.3-72)	94.8±64.4 (10.3-239.9)	30.2±14.8 (12.2-78.8)
OC (µg m <sup>-3</sup> )	29.4±9.2 (13.7-48.8)	10.7±6.2 (3.9-21.5)	21.5±12.3 (3.9-48.8)	6.4±2.3 (1.8-12.7)
EC (µg m <sup>-3</sup> )	4.6±1.3 (1.6-6.6)	1.9±1.6 (0.3-5.2)	3.5±2.0 (0.3-6.6)	0.9±0.4 (0.2-1.7)
SOC <sup>c</sup> (µg m <sup>-3</sup> )	10.3±5.7 (2.9-24.6)	2.9±1.4 (0.0-5.5)	7.2±5.7 (0.0-24.6)	2.3±1.4 (0.0-6.0)
Levoglucosan	348.2±148.0 (83.1-512.5)	195.0±163.7 (19.1-539.5)	278.5±171.4 (19.1-539.5)	26.1±28.3 (2.9-172.2)
Palmitic acid	376.2±234.9 (44.5-1089.6)	278±280.6 (33.8-1137.2)	335±255.3 (33.8-1137.2)	25.2±11.9 (9.4-68)
Stearic acid	207.1±181.4 (23-846.7)	163.6±228.1 (17.3-903.2)	188.8±199.8 (17.3-903.2)	16.0±7.2 (5.6-36.4)
Phenanthrene	8.6±6.1 (1.8-19)	5.6±6.1 (1-24.8)	7.3±6.2 (1-24.8)	0.7±0.7 (0-3.8)
Fluoranthene	25.1±19.6 (4.2-76.2)	16.1±21.3 (4.2-84.3)	21.3±20.5 (4.2-84.3)	0.4±0.2 (0-0.9)
Retene	16±14.9 (2-52.2)	11.1±12.1 (0.5-45.5)	13.9±13.8 (0.5-52.2)	0±0 (0-0.1)
Benz(a)anthracene	21.5±16.5 (0.3-62.7)	10.8±9.3 (1.4-30.5)	17±14.8 (0.3-62.7)	0.2±0.1 (0-0.5)
Chrysene	22.6±14.1 (3.7-47.3)	13.6±15.6 (0.1-59.5)	18.8±15.2 (0.1-59.5)	0.2±0.1 (0-0.3)
Benzo(b)fluoranthene	52.6±29 (10.7-98)	28.1±31 (2.4-113.6)	42.3±31.8 (2.4-113.6)	0.7±0.5 (0-2)
Benzo(k)fluoranthene	12.2±8 (0-25.3)	6.7±6.8 (0-23.7)	9.9±7.9 (0-25.3)	0.2±0.1 (0-0.4)
Picene	0.8±0.8 (0-2.6)	0.3±0.5 (0-1.3)	0.6±0.7 (0-2.6)	0±0 (0-0)
Benzo(ghi)perylene	7.0±4.7 (0-13.6)	4.0±4.1 (0-14.0)	5.6±4.6 (0-14.0)	0±0.1 (0-0.3)
17a (H) -22, 29, 30- Trisnorhopane	2.7±1.6 (0.6-6.7)	1.6±1.5 (0.3-6)	2.2±1.6 (0.3-6.7)	0±0.1 (0-0.4)
17b (H), 21a (H) - Norhopane	3.1±1.6 (0.9-6.6)	1.8±1.8 (0.3-7.3)	2.6±1.8 (0.3-7.3)	0±0 (0-0.2)
C24	26.3±15.3 (7.8-55.5)	18±19.2 (2.1-71.2)	22.5±17.4 (2.1-71.2)	1.4±0.6 (0.5-3.3)
C25	28.2±15.6 (8.5-59)	19.5±20.5 (2.3-76.2)	24.2±18.3 (2.3-76.2)	2.9±1.5 (0.5-6.5)
C26	18.9±10.2 (5.8-40.2)	13±13.1 (1.8-48.2)	16.2±11.8 (1.8-48.2)	1.6±0.7 (0.3-4.3)
C27	20.4±9.2 (6.1-37.1)	13.8±12.5 (2.2-43.5)	17.4±11.2 (2.2-43.5)	4.4±2 (0.6-11.7)
C28	10.6±4.8 (3.2-19.2)	6.9±5.7 (1.5-19.3)	8.9±5.5 (1.5-19.3)	1.4±0.6 (0.3-2.9)
C29	22.3±10.1 (5.9-39.7)	14.3±12.6 (3-39)	18.7±11.9 (3-39.7)	5.2±3.3 (0.4-20.7)
C30	6.8±2.9 (2.2-11.4)	4.5±3.1 (1-9.7)	5.7±3.2 (1-11.4)	1±0.4 (0.2-2)
C31	11.6±4.2 (3.5-17.7)	7.7±5.8 (1.2-18.7)	9.8±5.3 (1.2-18.7)	4.3±3.2 (0.4-20)
C32	6.1±2.6 (1.7-9.3)	3.9±2.6 (0.7-8.2)	5.1±2.8 (0.7-9.3)	0.9±0.4 (0.2-1.7)
C33	5.8±2.7 (1.7-11.5)	3.9±3.1 (0.9-9.6)	4.9±3 (0.9-11.5)	1.8±1.1 (0.1-6.3)
C34	2.1±2.1 (0-5.5)	1.2±1.4 (0-4)	1.7±1.8 (0-5.5)	0.3±0.3 (0-0.9)

300 <sup>a</sup> The unit is ng m<sup>-3</sup> for all organic compounds and µg m<sup>-3</sup> for PM<sub>2.5</sub>, OC, EC and SOC; <sup>b</sup> mean±SD  
 301 (min-max); <sup>c</sup> SOC concentration was calculated by EC-tracer method; <sup>d</sup> Haze days: PM<sub>2.5</sub>≥75 µg  
 302 m<sup>-3</sup>; <sup>e</sup> Non-haze days: PM<sub>2.5</sub><75 µg m<sup>-3</sup>;

### 303 3.2 Chemical Mass Closure (CMC)

304 The composition of PM<sub>2.5</sub> applying the chemical mass closure method is plotted in Fig.2  
 305 and summarized in Table S1. Because the gravimetrically measured mass (offline PM<sub>2.5</sub>)  
 306 differs slightly from online PM<sub>2.5</sub> (Fig. S2), the regression analysis results between mass  
 307 reconstructed using mass closure (reconstructed PM<sub>2.5</sub>) and both measured PM<sub>2.5</sub>  
 308 (offline PM<sub>2.5</sub>/ online PM<sub>2.5</sub>) were investigated and plotted in Fig. 3.

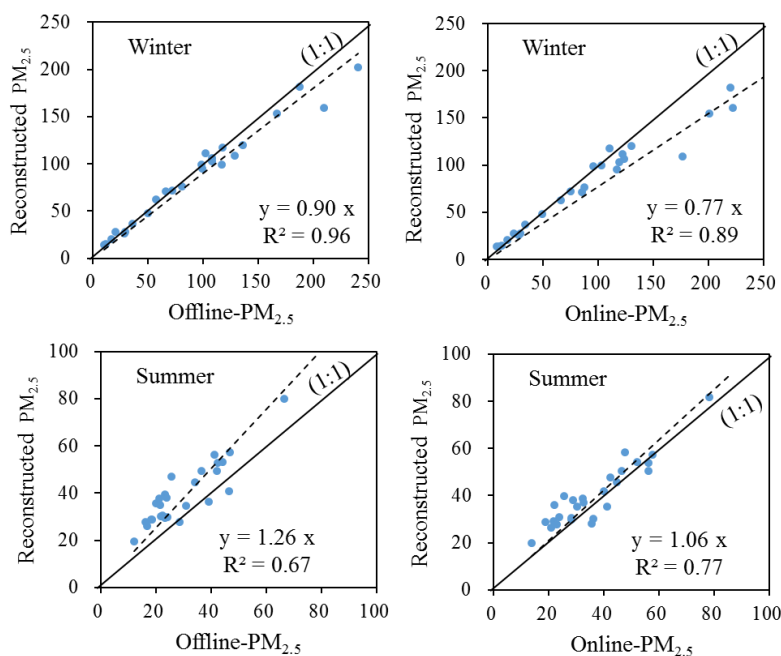
309



310  
311 **Figure 2.** Chemical components of reconstructed PM<sub>2.5</sub> (offline) applying mass closure method.

312

313



314

315 **Figure 3.** Regression results between reconstructed PM<sub>2.5</sub> and offline/online PM<sub>2.5</sub> by chemical mass  
316 closure method.

317 As shown in Fig. 3, measured offline/online PM<sub>2.5</sub> were moderately well correlated  
318 with the reconstructed PM<sub>2.5</sub> with slopes of 0.77~1.26 and R<sup>2</sup> of 0.67~0.96. In winter,  
319 the regression results were good between reconstructed PM<sub>2.5</sub> and offline-PM<sub>2.5</sub>. For  
320 online-PM<sub>2.5</sub>, it was much higher than the reconstructed PM<sub>2.5</sub> when the mass was over  
321 170 μg m<sup>-3</sup>. After excluding the outliers (2 outliers of offline-PM<sub>2.5</sub> > 200 μg m<sup>-3</sup> and 4  
322 outliers of online-PM<sub>2.5</sub> > 170 μg m<sup>-3</sup>), the regression results improved with both slopes  
323 and R<sup>2</sup> approaching unity (Fig. S3). This could indicate some uncertainties in offline  
324 and/or online PM<sub>2.5</sub> measurement for heavily polluted samples, or the applied OM/OC  
325 ratio in winter was not suitable for converting OC to OM in heavily polluted samples.  
326 During the summer campaign, the slope of the reconstructed PM<sub>2.5</sub> and online-PM<sub>2.5</sub>  
327 was close to 1, but that of reconstructed PM<sub>2.5</sub> and offline-PM<sub>2.5</sub> was 1.26. This could  
328 be due to the loss of semi-volatile compounds from PTFE filters or the positive artefacts

329 of quartz filters for chemical analyses, which can absorb more organics than PTFE  
 330 filters that are used for PM weighing. To avoid loss of semi-volatiles, all collected  
 331 samples were stored in cold conditions, including during shipment. The datapoints were  
 332 more scattered in summer, which could result from the large difference in OM-OC  
 333 relationships from day to day. The reconstructed inorganics (reconstructed PM<sub>2.5</sub>  
 334 excluding OM) correlated well with offline-PM<sub>2.5</sub>, but OM did not (Fig. S4). Hence,  
 335 the discrepancies of between reconstructed PM<sub>2.5</sub> and offline/online PM<sub>2.5</sub> in summer  
 336 may be mainly attributable to variable OM/OC ratios.

337 During the winter campaign, the carbonaceous components (OM & EC) accounted  
 338 for 47.2% of total reconstructed PM<sub>2.5</sub>, followed by the secondary inorganic ions (NH<sub>4</sub><sup>+</sup>,  
 339 SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>) (35.0%). In summer, on the contrary, secondary inorganic salts  
 340 represented 45.2% of PM<sub>2.5</sub> mass, followed by carbonaceous components (35.2%).  
 341 Bound water contributed 4.6% and 7.2% of PM<sub>2.5</sub> during the winter and summer,  
 342 respectively. All other components combined accounted for 13.2% and 12.4% of PM<sub>2.5</sub>  
 343 during the winter and summer campaigns, respectively.

344

### 345 3.3 Source apportionment of fine OC in urban Beijing applying a CMB model

346 The CMB model resolved seven primary sources of OC in winter and summer,  
 347 including vegetative detritus, straw and wood burning (biomass burning, BB), gasoline  
 348 vehicles, diesel vehicles, industrial coal combustion (Industrial CC), residential coal  
 349 combustion (Residential CC) and cooking. It explained an average of 75.7% (45.3-  
 350 91.3%) and 56.1% (34.3-76.3%) of fine OC in winter and summer, respectively. The  
 351 averaged CMB source apportionment results in winter and summer are presented in  
 352 Table 2. Daily source contribution estimates to fine OC and the relative abundance of  
 353 different sources contributions to OC in winter and summer are shown in Fig. 4.

354 During the winter campaign, coal combustion (industrial and residential CC, 7.5 μg  
 355 m<sup>-3</sup>, 35.0% of OC) was the most significant contributor to OC, followed by Other OC  
 356 (5.3 μg m<sup>-3</sup>, 24.8%), biomass (3.8 μg m<sup>-3</sup>, 17.6%), traffic (gasoline and diesel vehicles,  
 357 2.6 μg m<sup>-3</sup>, 11.9%), cooking (2.2 μg m<sup>-3</sup>, 10.3%), vegetative detritus (0.09 μg m<sup>-3</sup>, 0.4%).  
 358 On winter haze days, industrial coal combustion, cooking and Other OC were  
 359 significantly higher (nearly tripled) compared to non-haze days. During the summer  
 360 campaign, Other OC (2.9 μg m<sup>-3</sup>, 45.6%) was the most significant contributor to OC,  
 361 followed by coal combustion (2.0 μg m<sup>-3</sup>, 31.1%), cooking (0.7 μg m<sup>-3</sup>, 10.3%), traffic  
 362 (0.4 μg m<sup>-3</sup>, 6.1%), biomass burning (0.3 μg m<sup>-3</sup>, 5.3%), and vegetative detritus (0.1 μg  
 363 m<sup>-3</sup>, 1.7%).

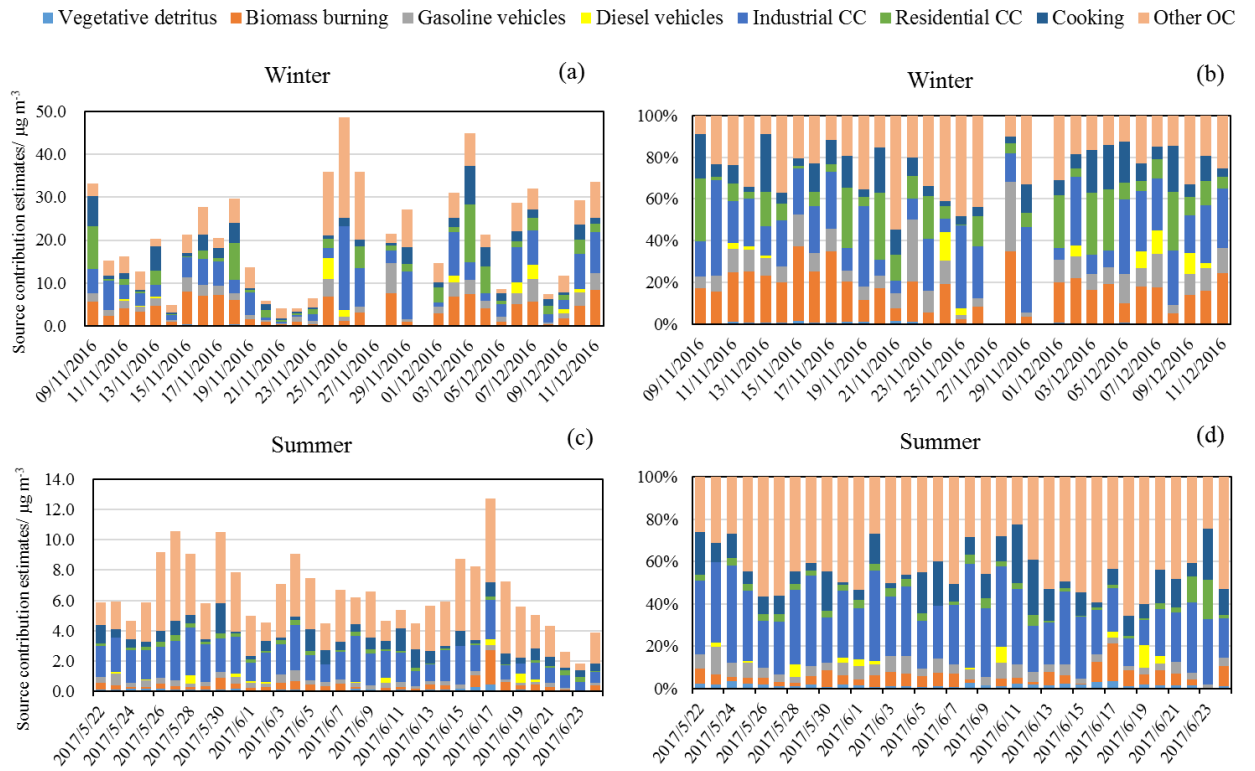
364 **Table 2.** Source contribution estimates (SCE, μg m<sup>-3</sup>) for fine OC in urban Beijing  
 365 during winter and summer from the CMB model

Sources	Winter		Winter (n=31)	Summer (n=34)
	Haze (n=18)	Non-haze (n=13)		
Vegetative detritus	0.11±0.08	0.07±0.08	0.09±0.08	0.11±0.08
Biomass burning	4.80±2.23	2.38±2.57	3.78±2.64	0.34±0.39
Gasoline vehicles	2.35±1.27	1.59±1.85	2.03±1.56	0.31±0.16
Diesel vehicles	0.83±1.43	0.14±0.33	0.54±1.15	0.08±0.16
Industrial coal combustion	7.09±4.17	1.95±1.36	4.94±4.15	1.82±0.72
Residential coal combustion	3.64±3.72	1.16±0.96	2.60±3.12	0.18±0.11

Cooking	3.23±2.30	0.85±0.52	2.23±2.13	0.66±0.43
Other OC <sup>a</sup>	7.4±5.6	2.5±1.4	5.3±4.9	2.9±1.5
Calculated OC <sup>b</sup>	22.0±6.5	8.2±5.3	16.2±9.1	3.5±1.2
Measured OC	29.4±9.2	10.7±6.2	21.5±12.3	6.4±2.3

366 <sup>a</sup> Other OC is calculated by subtracting calculated OC from measured OC;

367 <sup>b</sup> Calculated OC is the sum of OC from all seven primary sources: vegetative detritus, biomass burning,  
368 gasoline vehicles, diesel vehicles, industrial coal combustion, residential coal combustion and cooking.



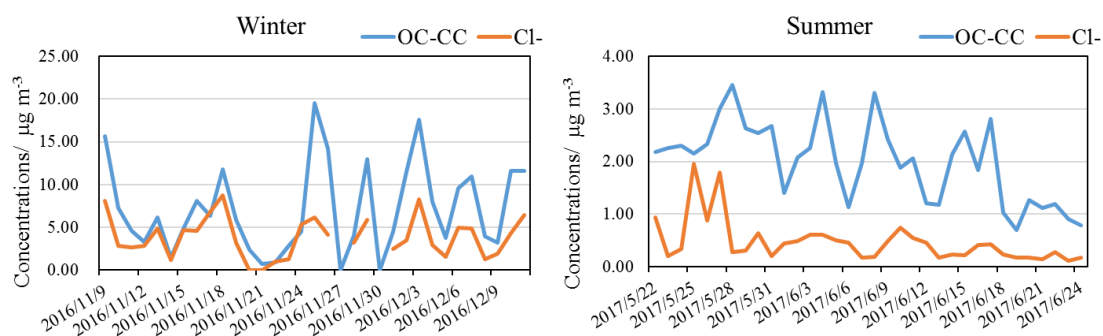
369

370 **Figure 4.** Daily source contribution estimates to fine OC in (a) winter and (c) summer  
 371 and their relative abundance in winter (b) and summer (d)

372 **3.3.1 Industrial and residential coal combustion**

373 In China, a large amount of coal is used in thermal power plant, industries, urban and  
 374 rural houses in northern China, especially during the heating period (mid-November to  
 375 mid-March) (Huang et al., 2017; Yu et al., 2019). But urban household coal use  
 376 experienced a remarkable drop of 58% during 2005-2015, which is much higher than  
 377 that of rural household coal use (5% of decrease) (Zhao et al., 2018). In this study, coal  
 378 combustion is the single largest source that contributed to primary OC in both winter  
 379 and summer. In addition, industrial CC was a more significant source of OC than  
 380 residential CC in urban Beijing. On average, coal combustion related OC was  $7.5 \pm 5.0$   
 381  $\mu\text{g m}^{-3}$  ( $34.5 \pm 9.8\%$  of OC) in winter, which was more than 3 times of that in summer -  
 382  $2.0 \pm 0.8 \mu\text{g m}^{-3}$  ( $32.3 \pm 10.2\%$  of OC), but the percentage contribution is similar. A  
 383 similar seasonal trend was also found in other studies in Beijing (Zheng et al., 2005;  
 384 Wang et al., 2009), but the relative contribution of coal combustion was much lower  
 385 than in this study. Industrial CC derived OC was  $4.94 \pm 4.15$  and  $1.82 \pm 0.72 \mu\text{g m}^{-3}$  in  
 386 winter and summer, respectively. Residential CC derived OC was  $2.60 \pm 3.12$  and  
 387  $0.18 \pm 0.11 \mu\text{g m}^{-3}$  in winter and summer, respectively. Residential CC was much higher  
 388 in winter compared to that in summer. On haze days, industrial CC and residential CC  
 389 derived OC were 3.6 and 3.1 times that on non-haze days, respectively, indicating an  
 390 important contribution to haze formation from industrial CC.

391 Coal combustion is also a major source for particulate chloride (Chen et al., 2014).  
 392 Because Beijing is an inland city, the contribution of marine aerosols to particulate  $\text{Cl}^-$   
 393 is considered minor, which is also supported by the higher  $\text{Cl}^-/\text{Na}^+$  mass ratios in winter  
 394 ( $10.1 \pm 4.8$ ) and summer ( $2.7 \pm 1.8$ ) than sea water (1.81), indicative of significant  
 395 contributions from anthropogenic sources (Bondy et al., 2017). Yang et al. (2018) also  
 396 reported that the contribution of sea-salt aerosol to fine particulate chloride was  
 397 negligible in China inland areas even during summer. Hence,  $\text{Cl}^-$  in this study was  
 398 mainly from anthropogenic sources. The time series of OC from coal combustion (OC-  
 399 CC) and  $\text{Cl}^-$  during winter and summer of Beijing are shown in Fig. 5. OC-CC and  $\text{Cl}^-$   
 400 exhibited similar trends in both seasons. The correlation coefficient ( $R^2$ ) between OC-  
 401 CC and  $\text{Cl}^-$  during winter was 0.62, which could be attributed to enhanced coal  
 402 combustion activities in this season. No significant correlation between the two was  
 403 found during the summer campaign, indicating the abundance of  $\text{Cl}^-$  in summer was  
 404 more influenced by other sources, probably including biomass burning. In addition, due  
 405 to the semi-volatility of ammonium chloride, it is liable to evaporate in summer (Pio  
 406 and Harrison, 1987). A similar phenomenon has been observed in Delhi (Pant et al.,  
 407 2015).



408

409 **Figure 5.** Time series of OC from coal combustion (OC-CC) and Cl<sup>-</sup> in winter and  
410 summer in Beijing

411

### 412 **3.3.2 Biomass burning**

413 Biomass burning (BB), including straw and wood burning, is an important source of  
414 atmospheric fine OC, which ranked as the second highest primary source of OC, after  
415 industrial coal combustion during the winter campaign, and third highest during the  
416 summer campaign after industrial CC and cooking. As shown in Fig. 4, the relative  
417 abundance of BB derived-OC during the winter campaign is much higher than the  
418 summer campaign. BB-derived OC from the CMB results was  $3.78 \pm 2.64 \mu\text{g m}^{-3}$  and  
419  $0.34 \pm 0.39 \mu\text{g m}^{-3}$  in winter and summer, contributing 17.6% and 5.3% of OC in these  
420 two seasons, respectively. These results are lower than those in 2005-2007 Beijing  
421 when BB accounted for 26% and 11% of OC in winter and summer, respectively (Wang  
422 et al., 2009). The BB-derived OC on winter haze days ( $4.80 \pm 2.23 \mu\text{g m}^{-3}$ ) was  
423 approximately double that of non-haze days ( $2.38 \pm 2.57 \mu\text{g m}^{-3}$ ), accounting for 16.3%  
424 and 22.2% of OC on haze and non-haze days, respectively.

425 Levoglucosan is widely used as a key tracer for biomass burning emissions (Bhattarai  
426 et al., 2019; Cheng et al., 2013; Xu et al., 2019a). Based on a levoglucosan to OC ratio  
427 of 8.2 % (Zhang et al., 2007a; Fan et al., 2020), the BB-derived OC was  $3.40 \pm 2.09 \mu\text{g}$   
428  $\text{m}^{-3}$  and  $0.32 \pm 0.35 \mu\text{g m}^{-3}$  during the winter and summer campaigns, respectively. These  
429 results are comparable to BB-derived OC from the CMB in this study. The estimated  
430 BB-derived OC concentration are also comparable with the BB-derived OC during the  
431 same sampling periods in Tianjin (Fan et al., 2020), but higher than those at IAP in  
432 2013-2014 (Kang et al., 2018).. Both of the studies applied the levoglucosan/OC ratio  
433 method to estimate the BB-derived OC although the actual ratio in Beijing air may be  
434 very different to 8.2%. The heavily elevated OC concentration in winter compared to  
435 summer could be a result of increased biomass burning activities for house heating and  
436 cooking in Beijing in addition to the unfavorable dispersion conditions under stagnant  
437 weather conditions in the winter.

438 In summer, the total OC concentration was highest on 17<sup>th</sup> June. The sudden rise of  
439 OC on this day was attributed to the enhanced biomass burning activities, which led to  
440 the highest level of BB-derived OC and highest BBOC to OC abundance. The  
441 levoglucosan concentration on this day was also the highest in summer, which reached  
442  $172 \text{ ng m}^{-3}$ .

### 443 **3.3.3 Gasoline and diesel vehicles**

444 OC and EC are the key components of traffic emissions (gasoline vehicles & diesel  
445 engines) (Chen et al., 2014; Chuang et al., 2016). Traffic related OC, as represented by  
446 the total sum of OC from gasoline and diesel vehicles, was  $2.4 \pm 2.3$  and  $0.39 \pm 0.22 \mu\text{g}$   
447  $\text{m}^{-3}$ , and contributed  $12.1 \pm 7.8\%$  and  $6.1 \pm 3.3\%$  of OC in winter and summer,  
448 respectively. These results are lower than the contribution of vehicle emissions to OC  
449 (13-20%) in Beijing during 2005 and 2006 (Wang et al., 2009), suggesting traffic  
450 emissions may be a less significant contributor to fine OC in the atmosphere in Beijing  
451 in 2016/2017. By multiplying by OM/OC factors of 2.39 and 1.47 in winter and summer,  
452 respectively, as mentioned in section 2.3, traffic related organic aerosol contributed

453 8.2±6.5% and 2.3±1.7% of PM<sub>2.5</sub> in winter and summer, respectively. The summer  
454 result was comparable with the vehicular emissions contribution to PM<sub>2.5</sub> (2.1%) in  
455 summer in Beijing, but higher than that in winter (1.5%) in Beijing estimated by using  
456 a PMF model (Yu et al., 2019). Gasoline vehicles dominated the traffic emissions;  
457 gasoline vehicle-derived OC was 2.03±1.56 and 0.31±0.16 μg m<sup>-3</sup> in winter and  
458 summer, respectively, which are approximately four times than that in winter  
459 (0.54±1.15 μg m<sup>-3</sup>) and summer (0.08±0.16 μg m<sup>-3</sup>) attributed to diesel vehicles. On  
460 haze days, gasoline- and diesel-derived OC were 2.35±1.27 and 0.83±1.43 μg m<sup>-3</sup>,  
461 respectively, much higher than gasoline- (1.59±1.85 μg m<sup>-3</sup>) and diesel-derived  
462 (0.14±0.33 μg m<sup>-3</sup>) OC on non-haze days. Even though diesel vehicles played a less  
463 important role in OC emissions, diesel-derived OC on haze days increased by around 6  
464 times above that of non-haze days, and such an increase was much higher than for  
465 gasoline, suggesting a potentially important role of diesel emissions on haze formation.

#### 466 **3.3.4 Cooking**

467 Cooking is expected to be an important contributor of fine OC in densely populated  
468 Beijing, which has a population of over 21 million. The cooking source profile was  
469 selected from a study which was carried out in the urban area of another Chinese  
470 megacity- Guangzhou, which includes fatty acids, sterols, monosaccharide anhydrides,  
471 alkanes and PAHs in particles from the Chinese residential cooking (Zhao et al., 2015).  
472 The resultant cooking related OC concentrations were 2.23±2.13 μg m<sup>-3</sup> and 0.66±0.43  
473 μg m<sup>-3</sup> in winter and summer, respectively, and both accounted for about 10% to total  
474 OC. Cooking OC was 3.23±2.30 μg m<sup>-3</sup> on winter haze days, around four times higher  
475 than that on non-haze days (0.85±0.52 μg m<sup>-3</sup>).

#### 476 **3.3.5 Vegetative detritus**

477 Vegetative detritus made a minor contribution to fine particle mass. Its concentration  
478 was 0.09±0.08 μg m<sup>-3</sup> (0.4%) and 0.11±0.08 μg m<sup>-3</sup> (1.7%) of OC during the winter and  
479 summer campaigns, respectively. These contributions are comparable with that in  
480 winter (0.5%), but higher than that in summer (0.3%) in urban Beijing during 2006-  
481 2007 (Wang et al., 2009). These results are also higher than the plant debris-derived  
482 OC in Tianjin in winter 2016 (0.02 μg m<sup>-3</sup>) and summer 2017 (0.01 μg m<sup>-3</sup>), which were  
483 calculated based on the relationship of glucose and plant debris and a OM/OC ratio of  
484 1.93 (Fan et al., 2020).

#### 485 **3.3.6 Other OC**

486 The Other OC was calculated by subtracting the calculated OC (the sum of OC from  
487 seven main sources) from measured OC concentrations. As shown in Table S2, there  
488 are four major source categories of OC in Beijing based on the Multi-resolution  
489 Emission Inventory for China (MEIC), which include power, industry, residential and  
490 transportation (Zheng et al., 2018). In the “industry” category, industrial coal  
491 combustion has been resolved by the CMB model. The local emissions of OC from  
492 industrial coal in Beijing were zero (shown in Table S2), and hence, the resolved POC  
493 from industrial coal combustion in Beijing should be regionally-transported. The MEIC  
494 data also show a small industrial oil combustion source. Since the tracers for this are  
495 likely to be the same as those for petroleum-derived road traffic emissions in CMB, this



496 may result in a small overestimation of the latter source. For the industrial processes  
497 related OC which have not been resolved by the CMB model, the annual average OC  
498 emissions in Beijing were 1161 and 1083 tonnes in 2016 and 2017 respectively, which  
499 accounted for 7.7% and 9.0% of the total OC emissions (POC). Therefore, the  
500 contribution from industrial processes to the total OC in the atmosphere (POC+SOC)  
501 was considered relatively small. The Other OC in this study is likely to be a mixture of  
502 predominantly SOC and a small portion of POC from sources such as industrial  
503 processes.

504 The Other OC was  $5.3 \pm 4.9$  and  $2.9 \pm 1.5 \mu\text{g m}^{-3}$  in winter and summer, respectively,  
505 contributing 24.8% and 43.9% of total measured OC. This is in good agreement with  
506 the Other OC estimated by CMB in another study in urban Beijing, for which Other OC  
507 contributed 22% and 44% of OC in winter and summer, respectively (Wang et al., 2009).  
508 SOC/OC in summer was more than 10% higher than that in summer 2008 in Beijing  
509 estimated using a tracer yield method, with the SOC derived from specific VOC  
510 precursors (toluene, isoprene,  $\alpha$ -pinene and  $\beta$ -caryophyllene) accounting for 32.5% of  
511 OC (Guo et al., 2012).

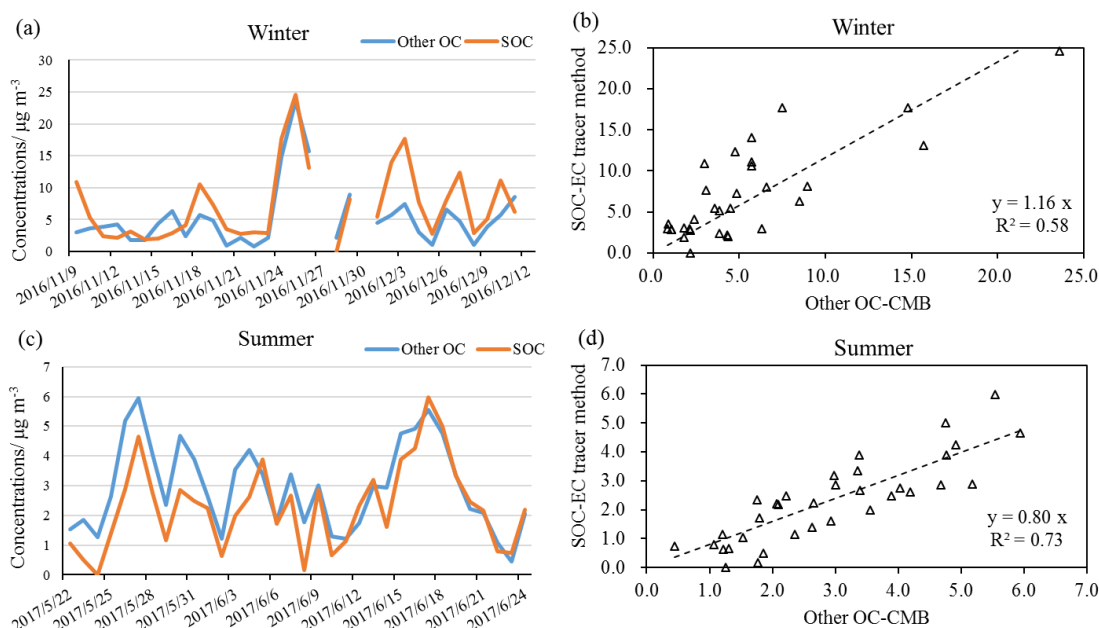
512 Even though the Other OC concentration was lower in summer, its relative  
513 abundance was higher than that in winter, suggesting relatively higher efficiency of  
514 SOA formation in summer due to more active photochemical processes under higher  
515 temperature and strong radiation. The Other OC on winter haze days was  $7.4 \pm 5.6 \mu\text{g m}^{-3}$ ,  
516 approximately 3 times of that on non-haze days ( $2.5 \pm 1.4 \mu\text{g m}^{-3}$ ). Other OC is also  
517 compared with the SOC estimated by EC-tracer method below.

### 518 3.3.7 SOC calculated based on the EC-tracer method

519 EC is a primary pollutant, while OC can originate from both primary sources and form  
520 in the atmosphere from gaseous precursors, namely primary organic carbon (POC) and  
521 SOC, respectively (Xu et al., 2018). The OC/EC ratios can be used to estimate the  
522 primary and secondary carbonaceous aerosol contributions. Usually, OC/EC ratios  $>$   
523 2.0 or 2.2 have been applied to identify and estimate SOA (Liu et al., 2017). In this  
524 study, all samples were observed with higher OC/EC ratios ( $>2.2$ ). SOC in this study  
525 was estimated using the equation below, assuming EC comes 100% from primary  
526 sources and the OC/EC ratio in primary sources is relatively constant (Turpin and  
527 Huntzicker, 1995; Castro et al., 1999):

$$528 \text{SOC}_i = \text{OC}_i - \text{EC}_i \times (\text{OC/EC})_{\text{pri}} \quad (4)$$

529 where  $\text{SOC}_i$ ,  $\text{OC}_i$  and  $\text{EC}_i$  are the ambient concentrations of secondary organic  
530 carbon, organic carbon and elemental carbon of sample  $i$ , respectively.  $(\text{OC/EC})_{\text{pri}}$  is  
531 the OC/EC ratio in primary aerosols. It is difficult to accurately determining the ratio  
532 of  $(\text{OC/EC})_{\text{pri}}$  for a given area.  $(\text{OC/EC})_{\text{pri}}$  varies with the contributions of different  
533 sources and can also be influenced by meteorological conditions (Dan et al., 2004). In  
534 this work,  $(\text{OC/EC})_{\text{pri}}$  was determined based on the lowest 5% of measured OC/EC  
535 ratios for the winter and summer campaigns, respectively (Pio et al., 2011). The average  
536 SOC concentrations during summer and winter were calculated and are shown in Table  
537 1. Daily concentrations of Other OC estimated by CMB and SOC estimated by the EC-  
538 tracer method in winter and summer are plotted in Fig. 6, as well as their correlation  
539 relationship.



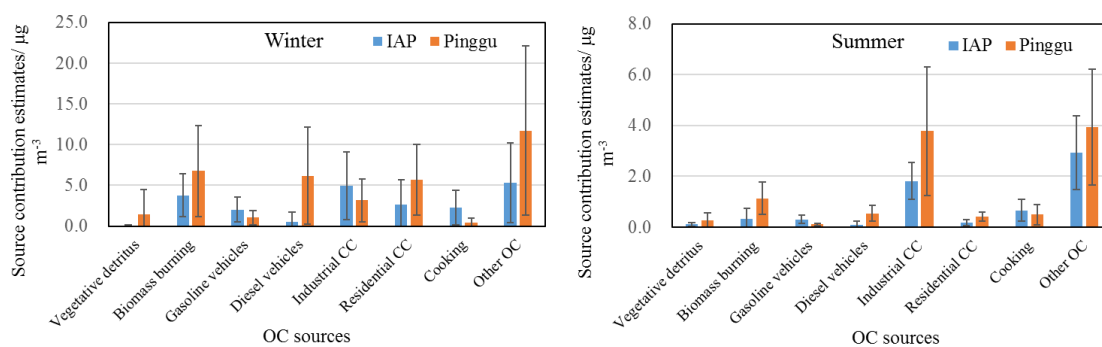
540  
 541 **Figure 6.** Time series of mean values for Other OC estimated by CMB and SOC  
 542 estimated by the EC-tracer method in winter (a) and summer (c); Correlation  
 543 relationship between Other OC estimated by CMB and SOC estimated by the EC-tracer  
 544 method in winter (b) and summer (d).

545 The average SOC concentrations in winter and summer are presented in Table 1. The  
 546 average SOC concentration during winter was  $7.2 \pm 5.7 \mu\text{g m}^{-3}$ , accounted for  
 547  $36.6 \pm 15.9\%$  of total OC. The average SOC concentration during summer was one third  
 548 of that in winter, which was  $2.3 \pm 1.4 \mu\text{g m}^{-3}$ , accounting for  $36.2 \pm 16.0\%$  of total OC.  
 549 The mean SOC concentrations during winter haze and non-haze periods were  $10.3 \pm 5.7$   
 550  $\mu\text{g m}^{-3}$  and  $2.9 \pm 1.4 \mu\text{g m}^{-3}$ , contributing to  $34.0 \pm 12.0\%$  and  $40.5 \pm 20.4\%$  of OC during  
 551 haze and non-haze episodes, respectively. As shown in Fig. 6, the SOC estimated by  
 552 the EC tracer method followed a similar trend to the Other OC calculated by the CMB  
 553 model. They were well-correlated in both seasons with  $R^2$  of 0.58 and 0.73 in winter  
 554 and summer samples, respectively and gradients of 1.16 and 0.80. This suggests that  
 555 the estimates of Other OC calculated from the CMB outputs were reasonable and  
 556 mainly represented the secondary organic aerosol.

### 557 3.4 Comparison with the source apportionment results in rural Beijing

558 The OC source apportionment results in this study are also compared with those in  
 559 another study conducted at a rural site of Beijing - Pinggu during APHH-Beijing  
 560 campaigns (Wu et al., 2020). CMB was run based on the results from high-time  
 561 resolution  $\text{PM}_{2.5}$  samples that were collected in Pinggu during the same sampling period,  
 562 but not on identical days. It is valuable to study both rural and urban sites, as both  
 563 exceed health-based guidelines and require evidence-based mitigation policies which  
 564 may differ depending on the source apportionment at each. Furthermore, urban air  
 565 pollution may affect the pollution levels in rural areas (Chen et al., 2020b), and  
 566 domestic heating and cooking led to high emissions of particles and precursor gases,  
 567 which may contribute to air pollution in the cities (Liu et al., 2021). The comparison of  
 568 results is presented in Fig. 7 and Table S3.

569 As shown in Fig. 7 and Table S3, slightly more OC was explained by CMB at the  
 570 urban site (75.7%) than the rural site (69.1%) during winter, but less OC was explained  
 571 at the urban site (56.1%) than the rural site (63.4%) during summer. As at the urban site,  
 572 biomass burning and coal combustion are important primary sources in rural Beijing.  
 573 Diesel contributed more to OC at the rural site, while cooking contributed more at the  
 574 urban site. The rural site also had a larger contribution from vegetative detritus to OC  
 575 than the urban site. The source contribution estimates from biomass burning at the rural  
 576 site was approximately 2 and 4 times that at the urban site during winter and summer.  
 577 In winter, biomass burning contributed a similar percentage of OC at both sites. A  
 578 higher percentage of OC from biomass burning was found at the rural site than the  
 579 urban site in summer, possibly because of use of biomass for cooking. For traffic  
 580 emitted OC, gasoline exceeded diesel at the urban site, while the rural site by contrast  
 581 has a larger diesel contribution. Industrial CC emitted OC is higher at the urban site  
 582 during winter, but lower in summer compared to the rural site. The source contribution  
 583 estimates of residential CC at the urban site is only half that of the rural site in both  
 584 seasons, and its relative contribution to OC was also lower at the urban site. Coal is  
 585 widely used for cooking and heating at the villages around the rural site at the time of  
 586 observations. Cooking accounted for over 10% of OC at the urban site, but less than 5%  
 587 at the rural site, which is plausible as the urban site is more densely populated.



588

589 **Figure 7.** Comparison of the source contribution estimates (SCE in  $\mu\text{g m}^{-3}$  (%OC)) at  
 590 IAP with those at a rural site in Beijing- Pinggu

### 591 3.5 Comparison with source apportionment results from AMS-PMF

592 Results from AMS-PMF were compared with the CMB source apportionment results  
 593 to investigate the consistency and potential uncertainties of both methods, and also to  
 594 provide supplemental source apportionment results (Ulbrich et al., 2009; Elser et al.,  
 595 2016). Similar comparisons have yielded valuable insights in earlier studies (Aiken et  
 596 al., 2009; Yin et al., 2015). It is noteworthy that the CMB model was applied to  $\text{PM}_{2.5}$   
 597 samples, while AMS-PMF was applied for  $\text{NR-PM}_1$  species. This may consequently  
 598 cause differences in the chemical composition and source attribution between the two  
 599 methods, as larger particles were not captured by AMS. However, as mentioned in the  
 600 study of Aiken et al. (2009), the mass concentration between  $\text{PM}_1$  and  $\text{PM}_{2.5}$  was small  
 601 with a reduced fraction of OA and increased fraction of dust. In addition, OC fractions  
 602 in fine particles were found mostly concentrated in particles  $<1 \mu\text{m}$  (Chen et al., 2020a;  
 603 Zhang et al., 2018; Tian et al., 2020). Hence, the bias was expected to be relatively  
 604 small. Six factors in non-refractory (NR)- $\text{PM}_1$  from the AMS were identified based on  
 605 the mass spectra measured in winter at IAP by applying a PMF model, including coal  
 606 combustion OA (CCOA-AMS), cooking OA (COA-AMS), biomass burning OA

607 (BBOA-AMS) and 3 secondary factors of oxidized primary OA (OPOA-AMS), less-  
 608 oxidized OA (LOOOA-AMS), and more-oxidized OA (MOOOA-AMS). In summer,  
 609 the PMF analysis resulted in 5 factors including 2 primary factors of hydrocarbon-like  
 610 OA (HOA-AMS), cooking OA (COA-AMS) and 3 secondary factors of oxygenated  
 611 OA (OOA-AMS): OOA1, OOA2, OOA3. These OOA factors were identified by PMF  
 612 based on diurnal cycles, mass spectra and the correlations between OA factors and other  
 613 measured species. Three OOA factors showed significantly elevated O/C ratios (0.67-  
 614 1.48), and correlated well with SIA ( $R=0.52-0.69$ ). Hence, OOA1, OOA2 and OOA3  
 615 represent three types of SOA. Compared to OOA2 and OOA3, OOA1 showed relatively  
 616 higher  $f_{43}$  (fraction of  $m/z$  43 in OA). In addition, the concentrations of OOA1 and  
 617 OOA3 were higher in daytime, implying the effect of photochemical processing. The  
 618 variations of OOA2 tracked well with  $C_2H_2O_2^+$  ( $R=0.89$ ), an aqueous-processing  
 619 related fragment ion (Sun et al., 2016), indicating that OOA2 was an OA factor  
 620 associated with aqueous-phase processing. Previous studies suggested that aqueous-  
 621 phase processing plays an important role in the formation of nitrogen-containing  
 622 compounds (Xu et al., 2017). The fact that OOA2 with relatively high N/C ratios  
 623 (0.046) was correlated with several N-containing ions (e.g.  $CH_4N^+$ ,  $C_2H_6N^+$ ,  $R=0.71-$   
 624  $0.77$ ) further supports the above argument. The factor profiles of AMS-PMF in winter  
 625 and summer are provided in Figs. S5 and S6, respectively.

626 In order to compare with the source apportionment results of OC in this study from the  
 627 CMB model, the OA concentrations from the AMS-PMF were converted to OC based  
 628 on various OA/OC ratios measured in Beijing: 1.35 for CCOA/CCOC (coal combustion  
 629 organic carbon), 1.31 for HOA/HOC (hydrocarbon-like organic carbon) (Sun et al.,  
 630 2016), 1.38 for COA/COC (cooking organic carbon), 1.58 for BBOA/BBOC (biomass  
 631 burning organic carbon) (Xu et al., 2019b), and 1.78 for OOA/OOC (Huang et al.,  
 632 2010). The concentrations of OA and corresponding OC from AMS-PMF analysis are  
 633 presented in Table 3. As the AMS data were missing during the period 09<sup>th</sup> - 15<sup>th</sup>  
 634 November 2016, the comparison of the AMS-PMF and CMB results for this period has  
 635 been excluded.

636 **Table 3.** Source contributions of OA and OC ( $\mu\text{g m}^{-3}$ ) from AMS-PMF results in urban  
 637 Beijing during winter and summer

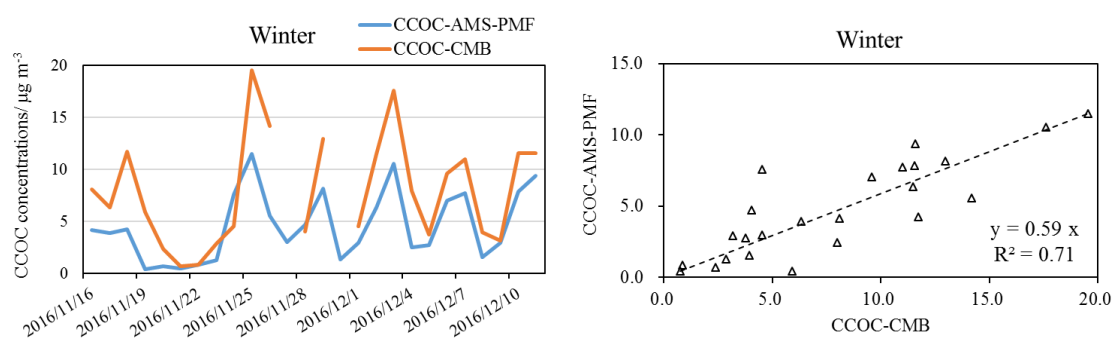
Winter			
Factors	Concentrations/ $\mu\text{g m}^{-3}$	Factors	Concentrations/ $\mu\text{g m}^{-3}$
CCOA	6.2±4.4	CCOC	4.6±3.3
COA	5.9±4.1	COC	4.3±3.0
BBOA	6.5±5.8	BBOC	4.1±3.7
OPOA	4.6±2.1	OPOC	2.6±1.2
LOOOA	5.2±5.2	LOOOC	2.9±2.9
MOOOA	8.1±7.0	MOOOC	4.6±4.0
OOA <sup>a</sup>	18.0±13.2	OOC <sup>d</sup>	10.1±7.4
OM <sup>b</sup>	36.7±24.0		
Summer			
Factors	Concentrations/ $\mu\text{g m}^{-3}$	Factors	Concentrations/ $\mu\text{g m}^{-3}$
HOA	0.7±0.4	HOC	0.5±0.3
COA	1.8±1.0	COC	1.3±0.7
OOA1	3.3±1.4	OOC1	1.9±0.8

OOA2	2.4±2.4	OOC2	1.4±1.3
OOA3	1.9±1.1	OOC3	1.1±0.6
OOA <sup>c</sup>	7.6±3.7	OOC	4.3±2.1
OM	10.1±3.9		

638 <sup>a</sup> OOA=OPOA+LOOOA+MOOOA; <sup>b</sup> OM is organics measured by AMS; <sup>c</sup> OOA=OOA1+OOA2+OOA3;  
 639 <sup>d</sup> OOC=OOC1+OOC2+OOC3

640 The CCOA-AMS factor was mainly characterized by m/z of 44, 73 and 115 (Sun et  
 641 al., 2016). In winter, CCOA-AMS was 6.2±4.4 μg m<sup>-3</sup>, contributing 16.9% of OM.  
 642 CCOC-AMS was 4.6±3.3 μg m<sup>-3</sup>, which was much lower than the estimated coal  
 643 combustion OC (7.9±5.2 μg m<sup>-3</sup>, industrial and residential coal combustion OC) by  
 644 CMB (CCOC-CMB). The time series of CCOC-CMB and CCOC-AMS in Fig. 8  
 645 showed a similar trend with relatively good correlation of R<sup>2</sup> = 0.71, but coal  
 646 combustion estimated by CMB was consistently higher than by AMS-PMF, probably  
 647 because AMS-PMF only resolved the sources of NR-PM<sub>1</sub>, and some coal combustion  
 648 particles are larger (Xu et al., 2011). The correlation coefficients (R<sup>2</sup>) of CCOC-AMS  
 649 with Cl<sup>-</sup> and NR-Cl<sup>-</sup> were 0.49 and 0.65, respectively in the winter data.

650



651

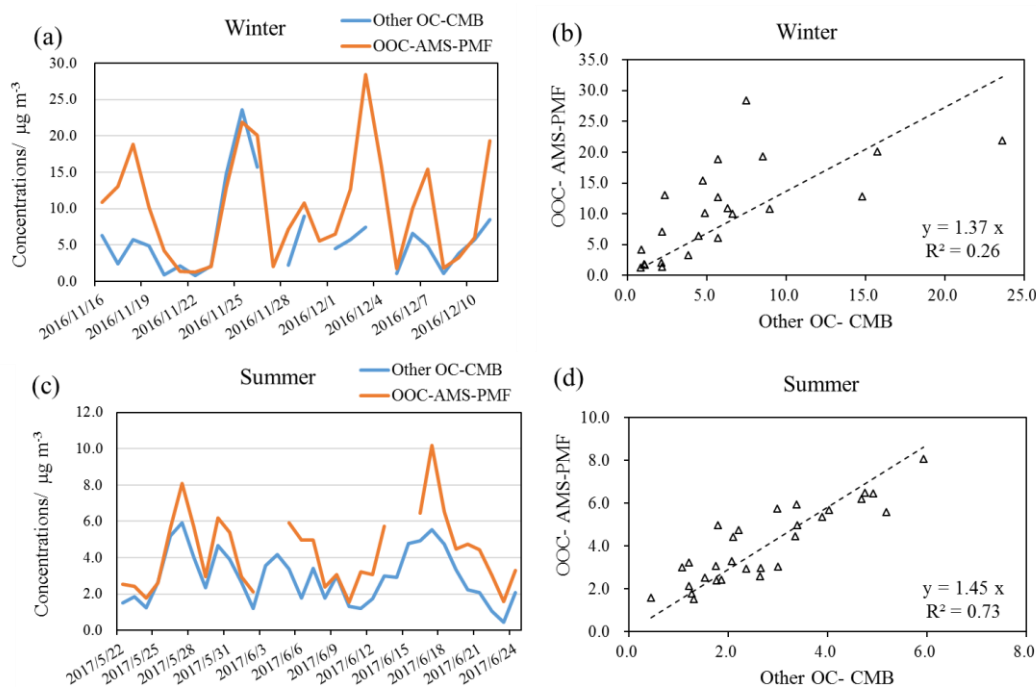
652 **Figure 8.** Time series and correlation of coal combustion related OC (CCOC) estimated  
 653 by CMB and CCOC from AMS-PMF analysis

654 BBOA-AMS in winter was 6.5±5.8 μg m<sup>-3</sup>, contributing 17.7% of OM. This BBOA-  
 655 AMS factor included a high proportion of m/z 60 and 73, which are typical fragments  
 656 of anhydrous sugars like levoglucosan (Srivastava et al., 2019). BBOC-AMS was  
 657 4.1±3.7 μg m<sup>-3</sup>, which was very close to the estimated BBOC-CMB (3.72±2.79 μg m<sup>-3</sup>,  
 658 16.4% of OC) during the same period.

659 COA-AMS is as a common factor identified in both winter and summer results. It is  
 660 characterized by high m/z of 55 and 57 in the mass spectrum (Sun et al., 2016). COA-  
 661 AMS was 5.9±4.1 and 1.8±1.0 μg m<sup>-3</sup> in winter and summer, respectively, contributing  
 662 16.1% and 17.8% of OM. COC-AMS was 4.3±3.0 and 1.3±0.7 μg m<sup>-3</sup> in winter and  
 663 summer, respectively, which were almost 2 times of the COC-CMB results for winter  
 664 (2.20±1.97 μg m<sup>-3</sup>) and summer (0.66±0.43 μg m<sup>-3</sup>). Yin et al. (2015) also reported that  
 665 COC-AMS was about 2 times of COC-CMB. The overestimation of cooking OC by  
 666 AMS-PMF could be due to a low relative ionization efficiency (RIE) for cooking OAs  
 667 (1.4) in AMS while the actual RIE could be higher, such as 1.56-3.06 (Reyes-Villegas  
 668 et al., 2018), and/or the use of a relatively low OA/OC ratio for cooking (Xu et al.,  
 669 2021).

670 HOA-AMS was  $0.7 \pm 0.4 \mu\text{g m}^{-3}$  in summer, accounting for 6.9% of OM. HOA-AMS  
 671 is usually identified based on the high contribution of aliphatic hydrocarbons in this  
 672 factor, particularly  $m/z$  of 27, 41, 55, 57, 69 and 71 (Aiken et al., 2009). This result is  
 673 lower than that (17% of OM) in rural Beijing during summer 2015 (Hua et al., 2018).  
 674 HOC-AMS was  $0.5 \pm 0.3 \mu\text{g m}^{-3}$  in summer, which is higher than the traffic  
 675 (gasoline+diesel) emitted OC ( $0.4 \pm 0.2 \mu\text{g m}^{-3}$ ) from the CMB model. No obvious  
 676 correlation was observed between HOC with nitrate and traffic emitted OC from the  
 677 CMB model during summer.

678 OOA-AMS concentrations (the sum of all oxidized OA) were  $18.0 \pm 13.2$  and  $7.6 \pm 3.7$   
 679  $\mu\text{g m}^{-3}$  in winter and summer, respectively, accounting for 49.0% and 75.2% of OM.  
 680 The derived OOC-AMS concentrations in winter and summer were  $10.1 \pm 7.4$  and  
 681  $4.3 \pm 2.1 \mu\text{g m}^{-3}$  in winter and summer, respectively, higher than the Other OC estimated  
 682 by CMB (Other OC-CMB) in winter ( $6.1 \pm 5.5 \mu\text{g m}^{-3}$ ) and summer ( $2.9 \pm 1.5 \mu\text{g m}^{-3}$ ) in  
 683 this study. This could be because AMS-PMF did not resolve HOC in winter and CCOC  
 684 in summer, which may be mixed with the OOA factors and lead to overestimation of  
 685 OOC concentrations. The time series and correlation of Other OC-CMB and OOC-  
 686 AMS is plotted in Fig. 9. A similar temporal trend was found between them, especially  
 687 in summer, which was also observed with a better correlation ( $R^2=0.73$ ).



688  
 689 **Figure 9.** Time series of mean values for Other OC estimated by CMB, and OOC  
 690 estimated by AMS-PMF in winter (a) and summer (c); Correlation relationship between  
 691 Other OC estimated by CMB and OOC estimated by AMS-PMF in winter (b) and  
 692 summer (d).

693 In summary, CMB is able to resolve almost all major known primary OA sources, but  
 694 AMS-PMF can resolve more secondary OA sources. The AMS-PMF results for major  
 695 components, such as CCOC-AMS and OOC-AMS agreed well with the results from  
 696 CMB in the winter. However, discrepancies or poor agreement was found for other  
 697 sources, such as BBOA-AMS and COA-AMS, although the temporal features were  
 698 very similar. Furthermore, AMS-PMF did not identify certain sources, probably due to

699 their relatively small contribution to particle mass. Overall, CMB and AMS-PMF  
700 offered complementary data to resolve both primary and secondary sources.

### 701 **3.6 Source contributions to PM<sub>2.5</sub> from the CMB model**

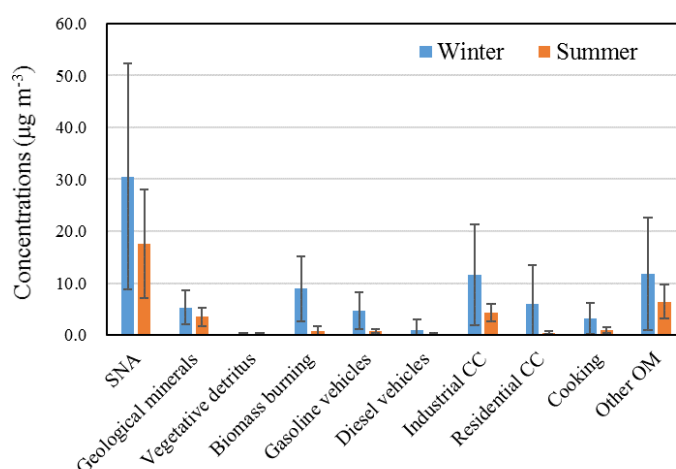
702 The source contributions to PM<sub>2.5</sub> were calculated by multiplication of the fine OC  
703 source estimates from CMB by the ratios of fine OC to PM<sub>2.5</sub> mass (Table S4), which  
704 were obtained from the same source profiles used for the OC apportionment by CMB  
705 (Zhang et al., 2007b; Wang et al., 2009; Cai et al., 2017; Zhang et al., 2008). For cooking,  
706 vegetative detritus and secondary organic aerosols, OM/OC ratios were applied  
707 considering the low contribution of inorganic species to PM<sub>2.5</sub> mass from these sources  
708 (Zhao et al., 2007; Bae et al., 2006b). The OM/OC ratios for oxygenated OA were in  
709 the range of 1.85-2.3 (Zhang et al., 2005; Aiken et al., 2008), and the OM/OC ratio was  
710 2.17 in secondary organic aerosols of PM<sub>2.5</sub> (Bae et al., 2006a). Therefore, an OM/OC  
711 ratio of 2.2 is applied in this study to convert the Other OC to OM. Due to the variability  
712 of the OC/PM<sub>2.5</sub> ratio in the source profiles, the application using the average OC/ PM<sub>2.5</sub>  
713 ratio of each source to convert the OC to PM<sub>2.5</sub> in all samples may be subject to  
714 uncertainties, as both organic species and PM<sub>2.5</sub> mass measurements are subject to  
715 analytical imprecision. Unfortunately, insufficient data are available for a formal  
716 analysis of uncertainty, but errors of around +/- 10% seem very probable. In addition,  
717 instead of OC/PM<sub>2.5</sub>, applying an OM/OC ratio to cooking and vegetative detritus  
718 sources for the calculation may result in an underestimation of PM<sub>2.5</sub> source  
719 contributions from these sources, because they can also emit inorganic pollutants.  
720 However, cooking emissions are mostly organic and the contribution from vegetative  
721 detritus to PM<sub>2.5</sub> is very small, so their effects on source contribution estimation here  
722 are considered negligible. The daily PM<sub>2.5</sub> contribution estimates and seasonal average  
723 source contributions are provided in Fig. S7 and Fig. 10, respectively. Detailed data and  
724 their relative abundance in the reconstructed PM<sub>2.5</sub> are summarized in Table S5.

725 As shown in Table S5, PM<sub>2.5</sub> mass was well explained by those sources which  
726 accounted for 91.9±24.1% and 99.0±19.1% of online PM<sub>2.5</sub> in winter and summer,  
727 respectively. In the summer, the offline PM<sub>2.5</sub> is lower than online observations. Thus,  
728 the CMB-based source contributions are more than offline PM<sub>2.5</sub> mass (121.7±26.6%).  
729 On average, the source contributions in winter ranked as SNA > coal combustion >  
730 Other OM > biomass burning > gasoline & diesel > geological minerals > cooking >  
731 vegetative detritus; in summer these ranked as SNA > other OM > coal combustion >  
732 geological minerals > cooking > gasoline & diesel > biomass burning > vegetative  
733 detritus.

734 Zheng et al. (2005) investigated the seasonal trends of PM<sub>2.5</sub> source contributions in  
735 Beijing during 2000 applying a CMB model. In winter (January), the contributions from  
736 coal combustion, biomass burning, diesel & gasoline, vegetative detritus to PM<sub>2.5</sub> were  
737 9.55 μg m<sup>-3</sup> (16% of PM<sub>2.5</sub> and hereafter), 5.8 μg m<sup>-3</sup> (9%), 3.85 μg m<sup>-3</sup>, 0.33 μg m<sup>-3</sup>,  
738 respectively. Contributions from gasoline, diesel, coal combustion and biomass burning  
739 were enhanced in Beijing during winter in 2016 compared to 2000, while the  
740 contribution from vegetative detritus basically remained similar. In summer (July) 2000,  
741 coal combustion contributed 2% of PM<sub>2.5</sub> (2.39 μg m<sup>-3</sup>), much less than that in summer  
742 2016 of this study. The contribution from diesel & gasoline (7.78 μg m<sup>-3</sup>, Zheng et al.,  
743 2005) was approximately 10 times of that in 2016 (0.8 μg m<sup>-3</sup>). Similarly, contributions  
744 from vegetative detritus and biomass burning were small and insignificant.



745 Zhou et al. (2017) estimated that coal combustion contributions in winter and  
 746 summer of Beijing-Tianjin-Hebei area in 2013 were  $15.9 \mu\text{g m}^{-3}$  and  $2.1 \mu\text{g m}^{-3}$ ,  
 747 respectively, which are comparable with those in this study. These results are also  
 748 comparable with the PMF-resolved coal and oil combustion in Beijing during winter  
 749 ( $17.4 \mu\text{g m}^{-3}$ ) and summer ( $2.2 \mu\text{g m}^{-3}$ ) in 2010 (Yu et al., 2013). SNA contributed 52.7  
 750 and  $26.4 \mu\text{g m}^{-3}$  of  $\text{PM}_{2.5}$  during winter (January) and summer (July), respectively (Yu  
 751 et al., 2013), which are much higher than those in this study. It is noteworthy that a  
 752 severe haze pollution event occurred during January 2013, which was characterized by  
 753 high concentrations of sulfate and nitrate in several studies (Zhou et al., 2017; Han et  
 754 al., 2016). The contribution from biomass burning in winter is consistent ( $8.5 \mu\text{g m}^{-3}$ )  
 755 with this study ( $8.9 \mu\text{g m}^{-3}$ ), but higher in summer ( $2.6 \mu\text{g m}^{-3}$ ) ( $0.8 \mu\text{g m}^{-3}$ ). The  
 756 cooking source contributed 4.8 and  $1.3 \mu\text{g m}^{-3}$  in  $\text{PM}_{2.5}$  during winter and summer 2013,  
 757 respectively, which is also comparable with this study.



758

759 **Figure 10.** Seasonal average  $\text{PM}_{2.5}$  source contribution estimates from the CMB  
 760 model

761

## 762 4 Conclusions

763 Carbonaceous aerosols contributed approximately 59% and 41% of reconstructed  $\text{PM}_{2.5}$   
 764 in winter and summer at the urban IAP site in Beijing. The OC and EC concentrations  
 765 were comparable with more recent studies (Fan et al., 2020; Qi et al., 2018), but lower  
 766 than those before 2013 (Yang et al., 2016; Dan et al., 2004), suggesting the  
 767 effectiveness of air pollution control measures since 2013 (Vu et al., 2019; Zhang et al.,  
 768 2019). CMB modelling showed that in the winter 2016, the top three primary  
 769 contributors to  $\text{PM}_{2.5}$ -OC were coal combustion (35%), biomass burning (17%), and  
 770 traffic (12%); these were in the same order with that at the rural site during the same  
 771 study period: coal combustion (29%), biomass burning (18%), and traffic (17%) (Wu  
 772 et al., 2020). In the summer 2017, the top three primary contributors to  $\text{PM}_{2.5}$ -OC were  
 773 coal combustion (32%), cooking (11%), and traffic (6%); these were different to that at  
 774 the rural site during the same study period: coal combustion (38%), biomass burning  
 775 (11%), and traffic (7%) (Wu et al., 2020). The Other OC, which was well-correlated  
 776 ( $R^2$ : 0.6~0.7; slope: 0.8~1.2) with the secondary OC (SOC) estimated based on the EC-  
 777 tracer method, accounted for 25% and 44% of OC at urban site and 31% and 37% of



778 OC at rural site during winter and summer, respectively. Although the annual average  
779 PM<sub>2.5</sub> levels in Beijing reduced from 88 µg m<sup>-3</sup> in year 2013 to 58 µg m<sup>-3</sup> in year 2017  
780 (Vu et al., 2019), and the deweathered concentration of PM<sub>1</sub> decreased by -38% in 2017  
781 comparing to 2007 (Zhang et al., 2020), our CMB modelling results indicate that the  
782 coal combustion and biomass burning still remained the dominant primary OC sources  
783 in winter 2016 and summer 2017, with road traffic ranked as the third highest. Cooking  
784 was a more significant source of OC than biomass burning at the urban site during  
785 summer. Compared to other CMB studies in Beijing, our study revealed an increase of  
786 the contributions from coal combustion, biomass burning and traffic to PM<sub>2.5</sub> in winter  
787 2016 compared to winter 2000, while those in this study remained similar compared to  
788 winter 2013. Sulfate, nitrate and ammonium concentrations were significantly lower in  
789 this study compared to 2013 (Zheng et al., 2005; Zhou et al., 2017). It is however  
790 notable that there is a broad consistency in the findings of the CMB studies, whereas  
791 the more numerous studies which have used PMF come to rather diverse conclusions  
792 (Srivastava et al., 2020).

793

794 *Data availability.* The data in this article are available from the corresponding authors  
795 upon request.

796

797 *Author contributions.* JX did the CMB modelling and drafted the paper with the help  
798 of ZS, RMH and all co-authors. DL, TVV conducted the laboratory analysis of organics  
799 and inorganics, respectively. XW, YZ provided the CMB source profiles. YS provided  
800 the AMS-PMF data.

801

802 *Competing interests.* The authors have no conflict of interests.

803

804 *Acknowledgement.* This research was funded by the UK Natural Environment Research  
805 Council (NERC, NE/N007190/1; NE/R005281/1) and Royal Society Advanced  
806 Fellowship (grant no: NAF\R1\191220).

807

## 808 **Reference**

809 Aiken, A. C., DeCarlo, P. F., Kroll, J. H., Worsnop, D. R., Huffman, J. A., Docherty, K. S., Ulbrich, I. M.,  
810 Mohr, C., Kimmel, J. R., Sueper, D., Sun, Y., Zhang, Q., Trimborn, A., Northway, M., Ziemann, P. J.,  
811 Canagaratna, M. R., Onasch, T. B., Alfarra, M. R., Prevot, A. S. H., Dommen, J., Duplissy, J., Metzger,  
812 A., Baltensperger, U., and Jimenez, J. L.: O/C and OM/OC Ratios of Primary, Secondary, and Ambient  
813 Organic Aerosols with High-Resolution Time-of-Flight Aerosol Mass Spectrometry, *Environ. Sci.*  
814 *Technol.*, 42, 4478-4485, 10.1021/es703009q, 2008.  
815 Aiken, A. C., Salcedo, D., Cubison, M. J., Huffman, J. A., DeCarlo, P. F., Ulbrich, I. M., Docherty, K. S.,  
816 Sueper, D., Kimmel, J. R., Worsnop, D. R., Trimborn, A., Northway, M., Stone, E. A., Schauer, J. J.,  
817 Volkamer, R. M., Fortner, E., de Foy, B., Wang, J., Laskin, A., Shutthanandan, V., Zheng, J., Zhang, R.,  
818 Gaffney, J., Marley, N. A., Paredes-Miranda, G., Arnott, W. P., Molina, L. T., Sosa, G., and Jimenez, J.  
819 L.: Mexico City aerosol analysis during MILAGRO using high resolution aerosol mass spectrometry at  
820 the urban supersite (T0) – Part 1: Fine particle composition and organic source apportionment, *Atmos.*

821 Chem. Phys., 9, 6633-6653, 10.5194/acp-9-6633-2009, 2009.

822 Antony Chen, L. W., Watson, J. G., Chow, J. C., DuBois, D. W., and Herschberger, L.: Chemical mass  
823 balance source apportionment for combined PM<sub>2.5</sub> measurements from U.S. non-urban and urban long-  
824 term networks, Atmospheric Environment, 44, 4908-4918,  
825 <https://doi.org/10.1016/j.atmosenv.2010.08.030>, 2010.

826 Bae, M.-S., Demerjian, K. L., and Schwab, J. J.: Seasonal estimation of organic mass to organic carbon  
827 in PM<sub>2.5</sub> at rural and urban locations in New York state, Atmospheric Environment, 40, 7467-7479,  
828 <https://doi.org/10.1016/j.atmosenv.2006.07.008>, 2006a.

829 Bae, M.-S., Schauer, J. J., and Turner, J. R.: Estimation of the Monthly Average Ratios of Organic Mass  
830 to Organic Carbon for Fine Particulate Matter at an Urban Site, Aerosol Sci. Technol., 40, 1123-1139,  
831 10.1080/02786820601004085, 2006b.

832 Bhattarai, H., Saikawa, E., Wan, X., Zhu, H., Ram, K., Gao, S., Kang, S., Zhang, Q., Zhang, Y., Wu, G.,  
833 Wang, X., Kawamura, K., Fu, P., and Cong, Z.: Levoglucosan as a tracer of biomass burning: Recent  
834 progress and perspectives, Atmos. Res., 220, 20-33, <https://doi.org/10.1016/j.atmosres.2019.01.004>,  
835 2019.

836 Bondy, A. L., Wang, B., Laskin, A., Craig, R. L., Nhliziyo, M. V., Bertman, S. B., Pratt, K. A., Shepson,  
837 P. B., and Ault, A. P.: Inland Sea Spray Aerosol Transport and Incomplete Chloride Depletion: Varying  
838 Degrees of Reactive Processing Observed during SOAS, Environ. Sci. Technol., 51, 9533-9542,  
839 10.1021/acs.est.7b02085, 2017.

840 Cai, T., Zhang, Y., Fang, D., Shang, J., Zhang, Y., and Zhang, Y.: Chinese vehicle emissions characteristic  
841 testing with small sample size: Results and comparison, Atmospheric Pollution Research, 8, 154-163,  
842 <https://doi.org/10.1016/j.apr.2016.08.007>, 2017.

843 Castro, L. M., Pio, C. A., Harrison, R. M., and Smith, D. J. T.: Carbonaceous aerosol in urban and rural  
844 European atmospheres: estimation of secondary organic carbon concentrations, Atmospheric  
845 Environment, 33, 2771-2781, [https://doi.org/10.1016/S1352-2310\(98\)00331-8](https://doi.org/10.1016/S1352-2310(98)00331-8), 1999.

846 Cavalli, F., Viana, M., Yttri, K. E., Genberg, J., and Putaud, J. P.: Toward a standardised thermal-optical  
847 protocol for measuring atmospheric organic and elemental carbon: the EUSAAR protocol, Atmos. Meas.  
848 Tech., 3, 79-89, 10.5194/amt-3-79-2010, 2010.

849 Chen, C., Zhang, H., Li, H., Wu, N., and Zhang, Q.: Chemical characteristics and source apportionment  
850 of ambient PM<sub>1.0</sub> and PM<sub>2.5</sub> in a polluted city in North China plain, Atmospheric Environment, 242,  
851 117867, <https://doi.org/10.1016/j.atmosenv.2020.117867>, 2020a.

852 Chen, L. W. A., Chow, J. C., Wang, X. L., Robles, J. A., Sumlin, B. J., Lowenthal, D. H., Zimmermann,  
853 R., and Watson, J. G.: Multi-wavelength optical measurement to enhance thermal/optical analysis for  
854 carbonaceous aerosol, Atmos. Meas. Tech., 8, 451-461, 10.5194/amt-8-451-2015, 2015a.

855 Chen, P., Wang, T., Hu, X., and Xie, M.: Chemical Mass Balance Source Apportionment of Size-  
856 Fractionated Particulate Matter in Nanjing, China, Aerosol Air Qual. Res., 15, 1855-1867,  
857 10.4209/aaqr.2015.03.0172, 2015b.

858 Chen, W.-N., Chen, Y.-C., Kuo, C.-Y., Chou, C.-H., Cheng, C.-H., Huang, C.-C., Chang, S.-Y., Roja  
859 Raman, M., Shang, W.-L., Chuang, T.-Y., and Liu, S.-C.: The real-time method of assessing the  
860 contribution of individual sources on visibility degradation in Taichung, Science of The Total  
861 Environment, 497-498, 219-228, <https://doi.org/10.1016/j.scitotenv.2014.07.120>, 2014.

862 Chen, Y., Shi, G., Cai, J., Shi, Z., Wang, Z., Yao, X., Tian, M., Peng, C., Han, Y., Zhu, T., Liu, Y., Yang,  
863 X., Zheng, M., Yang, F., Zhang, Q., and He, K.: Simultaneous measurements of urban and rural particles  
864 in Beijing – Part 2: Case studies of haze events and regional transport, Atmospheric Chemistry and  
865 Physics, 20, 9249-9263, 10.5194/acp-20-9249-2020, 2020b.

866 Cheng, Y., Engling, G., He, K. B., Duan, F. K., Ma, Y. L., Du, Z. Y., Liu, J. M., Zheng, M., and Weber,  
867 R. J.: Biomass burning contribution to Beijing aerosol, Atmospheric Chemistry And Physics, 13, 7765-  
868 7781, 10.5194/acp-13-7765-2013, 2013.

869 Chow, J. C., Lowenthal, D. H., Chen, L. W. A., Wang, X. L., and Watson, J. G.: Mass reconstruction  
870 methods for PM<sub>2.5</sub>: a review, Air Quality Atmosphere And Health, 8, 243-263, 10.1007/s11869-015-  
871 0338-3, 2015.

872 Chuang, M.-T., Chen, Y.-C., Lee, C.-T., Cheng, C.-H., Tsai, Y.-J., Chang, S.-Y., and Su, Z.-S.:  
873 Apportionment of the sources of high fine particulate matter concentration events in a developing  
874 aerotropolis in Taoyuan, Taiwan, Environmental Pollution, 214, 273-281,  
875 <https://doi.org/10.1016/j.envpol.2016.04.045>, 2016.

876 Dan, M., Zhuang, G., Li, X., Tao, H., and Zhuang, Y.: The characteristics of carbonaceous species and  
877 their sources in PM<sub>2.5</sub> in Beijing, Atmospheric Environment, 38, 3443-3452,  
878 <https://doi.org/10.1016/j.atmosenv.2004.02.052>, 2004.

879 de Miranda, R. M., de Fatima Andrade, M., Dutra Ribeiro, F. N., Mendonça Francisco, K. J., and Pérez-  
880 Martínez, P. J.: Source apportionment of fine particulate matter by positive matrix factorization in the

881 metropolitan area of São Paulo, Brazil, *Journal of Cleaner Production*, 202, 253-263,  
882 <https://doi.org/10.1016/j.jclepro.2018.08.100>, 2018.

883 Dong, F.-m., Mo, Y.-z., Li, G.-x., Xu, M.-m., and Pan, X.-c.: [Association between ambient PM10/PM2.5  
884 levels and population mortality of circulatory diseases: a case-crossover study in Beijing], *Beijing Da  
885 Xue Xue Bao Yi Xue Ban*, 45, 398-404, 2013.

886 Fan, Y., Liu, C. Q., Li, L., Ren, L., Ren, H., Zhang, Z., Li, Q., Wang, S., Hu, W., Deng, J., Wu, L., Zhong,  
887 S., Zhao, Y., Pavuluri, C. M., Li, X., Pan, X., Sun, Y., Wang, Z., Kawamura, K., Shi, Z., and Fu, P.: Large  
888 contributions of biogenic and anthropogenic sources to fine organic aerosols in Tianjin, North China,  
889 *Atmos. Chem. Phys.*, 20, 117-137, 10.5194/acp-20-117-2020, 2020.

890 Fountoukis, C., and Nenes, A.: ISORROPIA II: a computationally efficient thermodynamic equilibrium  
891 model for  $K^+-Ca^{2+}-Mg^{2+}-NH_4^+-Na^+-SO_4^{2-}-NO_3^- -Cl^- -H_2O$  aerosols, *Atmos. Chem. Phys.*, 7,  
892 4639-4659, 10.5194/acp-7-4639-2007, 2007.

893 Fu, P., Zhuang, G., Sun, Y., Wang, Q., Chen, J., Ren, L., Yang, F., Wang, Z., Pan, X., Li, X., and  
894 Kawamura, K.: Molecular markers of biomass burning, fungal spores and biogenic SOA in the  
895 Taklimakan desert aerosols, *Atmospheric Environment*, 130, 64-73, 10.1016/j.atmosenv.2015.10.087,  
896 2016.

897 Guo, S., Hu, M., Guo, Q., Zhang, X., Zheng, M., Zheng, J., Chang, C. C., Schauer, J. J., and Zhang, R.:  
898 Primary Sources and Secondary Formation of Organic Aerosols in Beijing, China, *Environ. Sci. Technol.*,  
899 46, 9846-9853, 10.1021/es2042564, 2012.

900 Guo, S., Hu, M., Guo, Q., Zhang, X., Schauer, J. J., and Zhang, R.: Quantitative evaluation of emission  
901 controls on primary and secondary organic aerosol sources during Beijing 2008 Olympics, *Atmos. Chem.  
902 Phys.*, 13, 8303-8314, 10.5194/acp-13-8303-2013, 2013.

903 Han, B., Zhang, R., Yang, W., Bai, Z., Ma, Z., and Zhang, W.: Heavy haze episodes in Beijing during  
904 January 2013: Inorganic ion chemistry and source analysis using highly time-resolved measurements  
905 from an urban site, *Science of The Total Environment*, 544, 319-329,  
906 <https://doi.org/10.1016/j.scitotenv.2015.10.053>, 2016.

907 He, K., Yang, F., Ma, Y., Zhang, Q., Yao, X., Chan, C. K., Cadle, S., Chan, T., and Mulawa, P.: The  
908 characteristics of PM2.5 in Beijing, China, *Atmospheric Environment*, 35, 4959-4970,  
909 [https://doi.org/10.1016/S1352-2310\(01\)00301-6](https://doi.org/10.1016/S1352-2310(01)00301-6), 2001.

910 Hua, Y., Wang, S., Jiang, J., Zhou, W., Xu, Q., Li, X., Liu, B., Zhang, D., and Zheng, M.: Characteristics  
911 and sources of aerosol pollution at a polluted rural site southwest in Beijing, China, *Science of The Total  
912 Environment*, 626, 519-527, <https://doi.org/10.1016/j.scitotenv.2018.01.047>, 2018.

913 Huang, R.-J., Zhang, Y., Bozzetti, C., Ho, K.-F., Cao, J.-J., Han, Y., Daellenbach, K. R., Slowik, J. G.,  
914 Platt, S. M., Canonaco, F., Zotter, P., Wolf, R., Pieber, S. M., Bruns, E. A., Crippa, M., Ciarelli, G.,  
915 Piazzalunga, A., Schwikowski, M., Abbaszade, G., Schnelle-Kreis, J., Zimmermann, R., An, Z., Szidat,  
916 S., Baltensperger, U., Haddad, I. E., and Prévôt, A. S. H.: High secondary aerosol contribution to  
917 particulate pollution during haze events in China, *Nature*, 514, 218-222, 10.1038/nature13774, 2014.

918 Huang, X., Liu, Z., Liu, J., Hu, B., Wen, T., Tang, G., Zhang, J., Wu, F., Ji, D., Wang, L., and Wang, Y.:  
919 Chemical characterization and source identification of PM2.5 at multiple sites in the Beijing-Tianjin-  
920 Hebei region, China, *Atmos. Chem. Phys.*, 17, 12941-12962, 10.5194/acp-17-12941-2017, 2017.

921 Huang, X. F., He, L. Y., Hu, M., Canagaratna, M. R., Sun, Y., Zhang, Q., Zhu, T., Xue, L., Zeng, L. W.,  
922 Liu, X. G., Zhang, Y. H., Jayne, J. T., Ng, N. L., and Worsnop, D. R.: Highly time-resolved chemical  
923 characterization of atmospheric submicron particles during 2008 Beijing Olympic Games using an  
924 Aerodyne High-Resolution Aerosol Mass Spectrometer, *Atmos. Chem. Phys.*, 10, 8933-8945,  
925 10.5194/acp-10-8933-2010, 2010.

926 Ikemori, F., Uranishi, K., Asakawa, D., Nakatsubo, R., Makino, M., Kido, M., Mitamura, N., Asano, K.,  
927 Nonaka, S., Nishimura, R., and Sugata, S.: Source apportionment in PM2.5 in central Japan using positive  
928 matrix factorization focusing on small-scale local biomass burning, *Atmospheric Pollution Research*,  
929 <https://doi.org/10.1016/j.apr.2021.01.006>, 2021.

930 Kang, M., Ren, L., Ren, H., Zhao, Y., Kawamura, K., Zhang, H., Wei, L., Sun, Y., Wang, Z., and Fu, P.:  
931 Primary biogenic and anthropogenic sources of organic aerosols in Beijing, China: Insights from  
932 saccharides and n-alkanes, *Environmental Pollution*, 243, 1579-1587,  
933 <https://doi.org/10.1016/j.envpol.2018.09.118>, 2018.

934 Le, T.-C., Shukla, K. K., Chen, Y.-T., Chang, S.-C., Lin, T.-Y., Li, Z., Pui, D. Y. H., and Tsai, C.-J.: On  
935 the concentration differences between PM2.5 FEM monitors and FRM samplers, *Atmospheric  
936 Environment*, 222, 117138, <https://doi.org/10.1016/j.atmosenv.2019.117138>, 2020.

937 Li, L., Ren, L., Ren, H., Yue, S., Xie, Q., Zhao, W., Kang, M., Li, J., Wang, Z., Sun, Y., and Fu, P.:  
938 Molecular Characterization and Seasonal Variation in Primary and Secondary Organic Aerosols in  
939 Beijing, China, 2018.

940 Li, P., Xin, J., Wang, Y., Wang, S., Li, G., Pan, X., Liu, Z., and Wang, L.: The acute effects of fine particles

941 on respiratory mortality and morbidity in Beijing, 2004–2009, *Environ. Sci. Pollut. Res.*, 20, 6433–6444,  
942 10.1007/s11356-013-1688-8, 2013.

943 Li, X., Nie, T., Qi, J., Zhou, Z., and Sun, X. S.: [Regional Source Apportionment of PM<sub>2.5</sub> in Beijing in  
944 January 2013], *Huan jing ke xue= Huanjing kexue*, 36, 1148–1153, 2015.

945 Liu, B., Wu, J., Zhang, J., Wang, L., Yang, J., Liang, D., Dai, Q., Bi, X., Feng, Y., Zhang, Y., and Zhang,  
946 Q.: Characterization and source apportionment of PM<sub>2.5</sub> based on error estimation from EPA PMF 5.0  
947 model at a medium city in China, *Environmental Pollution*, 222, 10–22,  
948 <https://doi.org/10.1016/j.envpol.2017.01.005>, 2017.

949 Liu, L., Zhang, J., Zhang, Y., Wang, Y., Xu, L., Yuan, Q., Liu, D., Sun, Y., Fu, P., Shi, Z., and Li, W.:  
950 Persistent residential burning-related primary organic particles during wintertime hazes in North China:  
951 insights into their aging and optical changes, *Atmos. Chem. Phys.*, 21, 2251–2265, 10.5194/acp-21-2251-  
952 2021, 2021.

953 Liu, Q., Baumgartner, J., Zhang, Y., and Schauer, J. J.: Source apportionment of Beijing air pollution  
954 during a severe winter haze event and associated pro-inflammatory responses in lung epithelial cells,  
955 *Atmospheric Environment*, 126, 28–35, <https://doi.org/10.1016/j.atmosenv.2015.11.031>, 2016.

956 Oduber, F., Calvo, A. I., Castro, A., Blanco-Alegre, C., Alves, C., Calzolari, G., Nava, S., Lucarelli, F.,  
957 Nunes, T., Barata, J., and Fraile, R.: Characterization of aerosol sources in León (Spain) using Positive  
958 Matrix Factorization and weather types, *Science of The Total Environment*, 754, 142045,  
959 <https://doi.org/10.1016/j.scitotenv.2020.142045>, 2021.

960 Pant, P., Shukla, A., Kohl, S. D., Chow, J. C., Watson, J. G., and Harrison, R. M.: Characterization of  
961 ambient PM<sub>2.5</sub> at a pollution hotspot in New Delhi, India and inference of sources, *Atmospheric  
962 Environment*, 109, 178–189, <https://doi.org/10.1016/j.atmosenv.2015.02.074>, 2015.

963 Paraskevopoulou, D., Liakakou, E., Gerasopoulos, E., Theodosi, C., and Mihalopoulos, N.: Long-term  
964 characterization of organic and elemental carbon in the PM<sub>2.5</sub> fraction: the case of Athens,  
965 Greece, *Atmos. Chem. Phys.*, 14, 13313–13325, 10.5194/acp-14-13313-2014, 2014.

966 Pio, C., and Harrison, R.: Vapour pressure of ammonium chloride aerosol: Effect of temperature and  
967 humidity, *Atmospheric Environment (1967)*, 21, 2711–2715, 10.1016/0004-6981(87)90203-4, 1987.

968 Pio, C., Cerqueira, M., Harrison, R. M., Nunes, T., Mirante, F., Alves, C., Oliveira, C., Sanchez de la  
969 Campa, A., Artífano, B., and Matos, M.: OC/EC ratio observations in Europe: Re-thinking the approach  
970 for apportionment between primary and secondary organic carbon, *Atmospheric Environment*, 45, 6121–  
971 6132, <https://doi.org/10.1016/j.atmosenv.2011.08.045>, 2011.

972 Qi, M., Jiang, L., Liu, Y., Xiong, Q., Sun, C., Li, X., Zhao, W., and Yang, X.: Analysis of the  
973 Characteristics and Sources of Carbonaceous Aerosols in PM<sub>2.5</sub> in the Beijing, Tianjin, and Langfang  
974 Region, China, *Int J Environ Res Public Health*, 15, 1483, 10.3390/ijerph15071483, 2018.

975 Rogge, W. F., Hildemann, L. M., Mazurek, M. A., Cass, G. R., and Simoneit, B. R. T.: Sources of fine  
976 organic aerosol. 4. Particulate abrasion products from leaf surfaces of urban plants, *Environ. Sci. Technol.*,  
977 27, 2700–2711, 10.1021/es00049a008, 1993.

978 Sarnat, J. A., Marmur, A., Klein, M., Kim, E., Russell, A. G., Sarnat, S. E., Mulholland, J. A., Hopke, P.  
979 K., and Tolbert, P. E.: Fine Particle Sources and Cardiorespiratory Morbidity: An Application of  
980 Chemical Mass Balance and Factor Analytical Source-Apportionment Methods, *Environmental Health  
981 Perspectives*, 116, 459–466, doi:10.1289/ehp.10873, 2008.

982 Shi, Z., Vu, T., Kotthaus, S., Harrison, R. M., Grimmond, S., Yue, S., Zhu, T., Lee, J., Han, Y., Demuzere,  
983 M., Dunmore, R. E., Ren, L., Liu, D., Wang, Y., Wild, O., Allan, J., Acton, W. J., Barlow, J., Barratt, B.,  
984 Beddows, D., Bloss, W. J., Calzolari, G., Carruthers, D., Carslaw, D. C., Chan, Q., Chatzidiakou, L., Chen,  
985 Y., Crilley, L., Coe, H., Dai, T., Doherty, R., Duan, F., Fu, P., Ge, B., Ge, M., Guan, D., Hamilton, J. F.,  
986 He, K., Heal, M., Heard, D., Hewitt, C. N., Hollaway, M., Hu, M., Ji, D., Jiang, X., Jones, R., Kalberer,  
987 M., Kelly, F. J., Kramer, L., Langford, B., Lin, C., Lewis, A. C., Li, J., Li, W., Liu, H., Liu, J., Loh, M.,  
988 Lu, K., Lucarelli, F., Mann, G., McFiggans, G., Miller, M. R., Mills, G., Monk, P., Nemitz, E., O'Connor,  
989 F., Ouyang, B., Palmer, P. I., Percival, C., Popoola, O., Reeves, C., Rickard, A. R., Shao, L., Shi, G.,  
990 Spracklen, D., Stevenson, D., Sun, Y., Sun, Z., Tao, S., Tong, S., Wang, Q., Wang, W., Wang, X., Wang,  
991 X., Wang, Z., Wei, L., Whalley, L., Wu, X., Wu, Z., Xie, P., Yang, F., Zhang, Q., Zhang, Y., Zhang, Y.,  
992 and Zheng, M.: Introduction to the special issue “In-depth study of air pollution sources and processes  
993 within Beijing and its surrounding region (APHH-Beijing)”, *Atmos. Chem. Phys.*, 19, 7519–7546,  
994 10.5194/acp-19-7519-2019, 2019.

995 Song, Y., Xie, S., Zhang, Y., Zeng, L., Salmon, L. G., and Zheng, M.: Source apportionment of PM<sub>2.5</sub> in  
996 Beijing using principal component analysis/absolute principal component scores and UNMIX, *Science  
997 of The Total Environment*, 372, 278–286, <https://doi.org/10.1016/j.scitotenv.2006.08.041>, 2006a.

998 Song, Y., Zhang, Y., Xie, S., Zeng, L., Zheng, M., Salmon, L. G., Shao, M., and Slanina, S.: Source  
999 apportionment of PM<sub>2.5</sub> in Beijing by positive matrix factorization, *Atmospheric Environment*, 40,  
1000 1526–1537, <https://doi.org/10.1016/j.atmosenv.2005.10.039>, 2006b.

1001 Song, Y., Tang, X., Xie, S., Zhang, Y., Wei, Y., Zhang, M., Zeng, L., and Lu, S.: Source apportionment  
1002 of PM<sub>2.5</sub> in Beijing in 2004, *Journal of Hazardous Materials*, 146, 124-130,  
1003 <https://doi.org/10.1016/j.jhazmat.2006.11.058>, 2007.

1004 Srivastava, D., Favez, O., Petit, J. E., Zhang, Y., Sofowote, U. M., Hopke, P. K., Bonnaire, N., Perraudin,  
1005 E., Gros, V., Villenave, E., and Albinet, A.: Speciation of organic fractions does matter for aerosol source  
1006 apportionment. Part 3: Combining off-line and on-line measurements, *Science of The Total Environment*,  
1007 690, 944-955, <https://doi.org/10.1016/j.scitotenv.2019.06.378>, 2019.

1008 Srivastava, D., Xu, J., Liu, D., Vu, T. V., Shi, Z., and Harrison, R. M.: Insight into PM<sub>2.5</sub> sources by  
1009 applying Positive Matrix factorization (PMF) at urban and rural sites of Beijing, *Atmospheric Chemistry  
1010 and Physics* (under review), 2020.

1011 Sun, Y., Du, W., Fu, P., Wang, Q., Li, J., Ge, X., Zhang, Q., Zhu, C., Ren, L., Xu, W., Zhao, J., Han, T.,  
1012 Worsnop, D. R., and Wang, Z.: Primary and secondary aerosols in Beijing in winter: sources, variations  
1013 and processes, *Atmos. Chem. Phys.*, 16, 8309-8329, 10.5194/acp-16-8309-2016, 2016.

1014 Sun, Y., He, Y., Kuang, Y., Xu, W., Song, S., Ma, N., Tao, J., Cheng, P., Wu, C., Su, H., Cheng, Y., Xie,  
1015 C., Chen, C., Lei, L., Qiu, Y., Fu, P., Croteau, P., and Worsnop, D. R.: Chemical Differences between  
1016 PM<sub>1</sub> and PM<sub>2.5</sub> in Highly Polluted Environment and Implications in Air Pollution Studies, *Geophysical  
1017 Research Letters*, n/a, e2019GL086288, 10.1029/2019gl086288, 2020.

1018 Tian, Y., Feng, Y., Liang, Y., Li, Y., Xue, Q., Shi, Z., Xu, J., and Harrison, R. M.: Size distributions of  
1019 inorganic and organic components in particulate matter from a megacity in northern China: dependence  
1020 upon seasons and pollution levels (In preparation), 2020.

1021 Turpin, B. J., and Huntzicker, J. J.: Identification of secondary organic aerosol episodes and quantitation  
1022 of primary and secondary organic aerosol concentrations during SCAQS, *Atmospheric Environment*, 29,  
1023 3527-3544, [https://doi.org/10.1016/1352-2310\(94\)00276-Q](https://doi.org/10.1016/1352-2310(94)00276-Q), 1995.

1024 U.S.EPA: Monitoring PM<sub>2.5</sub> in ambient air using designated reference or class I equivalent methods, in:  
1025 *Quality Assurance Handbook for Air Pollution Measurement Systems*, 2-12, 2016.

1026 Ulbrich, I. M., Canagaratna, M. R., Zhang, Q., Worsnop, D. R., and Jimenez, J. L.: Interpretation of  
1027 organic components from Positive Matrix Factorization of aerosol mass spectrometric data, *Atmos.  
1028 Chem. Phys.*, 9, 2891-2918, 10.5194/acp-9-2891-2009, 2009.

1029 Viana, M., Kuhlbusch, T. A. J., Querol, X., Alastuey, A., Harrison, R. M., Hopke, P. K., Winiwarter, W.,  
1030 Vallius, M., Szidat, S., Prévôt, A. S. H., Hueglin, C., Bloemen, H., Wählin, P., Vecchi, R., Miranda, A. I.,  
1031 Kasper-Giebl, A., Maenhaut, W., and Hitzenberger, R.: Source apportionment of particulate matter in  
1032 Europe: A review of methods and results, *Journal of Aerosol Science*, 39, 827-849,  
1033 <https://doi.org/10.1016/j.jaerosci.2008.05.007>, 2008.

1034 Vu, T. V., Shi, Z. B., Cheng, J., Zhang, Q., He, K. B., Wang, S. X., and Harrison, R. M.: Assessing the  
1035 impact of clean air action on air quality trends in Beijing using a machine learning technique,  
1036 *Atmospheric Chemistry And Physics*, 19, 11303-11314, 10.5194/acp-19-11303-2019, 2019.

1037 Wang, Q., Shao, M., Zhang, Y., Wei, Y., Hu, M., and Guo, S.: Source apportionment of fine organic  
1038 aerosols in Beijing, *Atmos. Chem. Phys.*, 9, 8573-8585, 10.5194/acp-9-8573-2009, 2009.

1039 Wu, X., Chen, C., Vu, T. V., Liu, D., Baldo, C., Shen, X., Zhang, Q., Cen, K., Zheng, M., He, K., Shi, Z.,  
1040 and Harrison, R. M.: Source apportionment of fine organic carbon (OC) using receptor modelling at a  
1041 rural site of Beijing: Insight into seasonal and diurnal variation of source contributions, *Environmental  
1042 Pollution*, 266, 115078, <https://doi.org/10.1016/j.envpol.2020.115078>, 2020.

1043 Xu, J., Jia, C., He, J., Xu, H., Tang, Y.-T., Ji, D., Yu, H., Xiao, H., and Wang, C.: Biomass burning and  
1044 fungal spores as sources of fine aerosols in Yangtze River Delta, China – Using multiple organic tracers  
1045 to understand variability, correlations and origins, *Environmental Pollution*, 251, 155-165,  
1046 <https://doi.org/10.1016/j.envpol.2019.04.090>, 2019a.

1047 Xu, J., Song, S., Harrison, R. M., Song, C., Wei, L., Zhang, Q., Sun, Y., Lei, L., Zhang, C., Yao, X., Chen,  
1048 D., Li, W., Wu, M., Tian, H., Luo, L., Tong, S., Li, W., Wang, J., Shi, G., Huangfu, Y., Tian, Y., Ge, B.,  
1049 Su, S., Peng, C., Chen, Y., Yang, F., Mihajlidi-Zelić, A., Đorđević, D., Swift, S. J., Andrews, I., Hamilton,  
1050 J. F., Sun, Y., Kramawijaya, A., Han, J., Saksakulkrai, S., Baldo, C., Hou, S., Zheng, F., Daellenbach, K.  
1051 R., Yan, C., Liu, Y., Kulmala, M., Fu, P., and Shi, Z.: An inter-laboratory comparison of aerosol inorganic  
1052 ion measurements by Ion Chromatography: implications for aerosol pH estimate, *Atmos. Meas. Tech.  
1053 Discuss.*, 2020, 1-36, 10.5194/amt-2020-156, 2020.

1054 Xu, J., Srivastava, D., Wu, X., Hou, S., Vu, Tuan V., Liu, D., Sun, Y., Vlachou, A., Moschos, V., Salazar,  
1055 G., Szidat, S., Prévôt, A. S. H., Fu, P., Harrison, R. M., and Shi, Z.: An evaluation of source apportionment  
1056 of fine OC and PM<sub>2.5</sub> by multiple methods: APHH-Beijing campaigns as a case study, *Faraday  
1057 Discussions*, 10.1039/D0FD00095G, 2021.

1058 Xu, M., Yu, D., Yao, H., Liu, X., and Qiao, Y.: Coal combustion-generated aerosols: Formation and  
1059 properties, *Proceedings of the Combustion Institute*, 33, 1681-1697,  
1060 <https://doi.org/10.1016/j.proci.2010.09.014>, 2011.

1061 Xu, W., Sun, Y., Wang, Q., Du, W., Zhao, J., Ge, X., Han, T., Zhang, Y., Zhou, W., Li, J., Fu, P., Wang,  
1062 Z., and Worsnop, D. R.: Seasonal Characterization of Organic Nitrogen in Atmospheric Aerosols Using  
1063 High Resolution Aerosol Mass Spectrometry in Beijing, China, *ACS Earth and Space Chemistry*, 1, 673-  
1064 682, 10.1021/acsearthspacechem.7b00106, 2017.

1065 Xu, W., Sun, Y., Wang, Q., Zhao, J., Wang, J., Ge, X., Xie, C., Zhou, W., Du, W., Li, J., Fu, P., Wang, Z.,  
1066 Worsnop, D. R., and Coe, H.: Changes in Aerosol Chemistry From 2014 to 2016 in Winter in Beijing:  
1067 Insights From High-Resolution Aerosol Mass Spectrometry, *Journal of Geophysical Research:*  
1068 *Atmospheres*, 124, 1132-1147, 10.1029/2018jd029245, 2019b.

1069 Xu, X., Zhang, H., Chen, J., Li, Q., Wang, X., Wang, W., Zhang, Q., Xue, L., Ding, A., and Mellouki, A.:  
1070 Six sources mainly contributing to the haze episodes and health risk assessment of PM<sub>2.5</sub> at Beijing  
1071 suburb in winter 2016, *Ecotoxicology and Environmental Safety*, 166, 146-156,  
1072 <https://doi.org/10.1016/j.ecoenv.2018.09.069>, 2018.

1073 Yang, F., Kawamura, K., Chen, J., Ho, K., Lee, S., Gao, Y., Cui, L., Wang, T., and Fu, P.: Anthropogenic  
1074 and biogenic organic compounds in summertime fine aerosols (PM<sub>2.5</sub>) in Beijing, China, *Atmospheric*  
1075 *Environment*, 124, 166-175, <https://doi.org/10.1016/j.atmosenv.2015.08.095>, 2016.

1076 Yang, X., Wang, T., Xia, M., Gao, X., Li, Q., Zhang, N., Gao, Y., Lee, S., Wang, X., Xue, L., Yang, L.,  
1077 and Wang, W.: Abundance and origin of fine particulate chloride in continental China, *Science of The*  
1078 *Total Environment*, 624, 1041-1051, <https://doi.org/10.1016/j.scitotenv.2017.12.205>, 2018.

1079 Yin, J., Cumberland, S. A., Harrison, R. M., Allan, J., Young, D. E., Williams, P. I., and Coe, H.: Receptor  
1080 modelling of fine particles in southern England using CMB including comparison with AMS-PMF  
1081 factors, *Atmos. Chem. Phys.*, 15, 2139-2158, 10.5194/acp-15-2139-2015, 2015.

1082 Yu, L., and Wang, G.: Characterization and Source Apportionment of PM<sub>2.5</sub> in an Urban Environment  
1083 in Beijing, *Aerosol Air Qual. Res.*, 13, 10.4209/aaqr.2012.07.0192, 2013.

1084 Yu, L. D., Wang, G. F., Zhang, R. J., Zhang, L. M., Song, Y., Wu, B. B., Li, X. F., An, K., and Chu, J. H.:  
1085 Characterization and Source Apportionment of PM<sub>2.5</sub> in an Urban Environment in Beijing, *Aerosol Air*  
1086 *Qual. Res.*, 13, 574-583, 10.4209/aaqr.2012.07.0192, 2013.

1087 Yu, S., Liu, W., Xu, Y., Yi, K., Zhou, M., Tao, S., and Liu, W.: Characteristics and oxidative potential of  
1088 atmospheric PM<sub>2.5</sub> in Beijing: Source apportionment and seasonal variation, *Science of The Total*  
1089 *Environment*, 650, 277-287, <https://doi.org/10.1016/j.scitotenv.2018.09.021>, 2019.

1090 Zhang, Q., Worsnop, D. R., Canagaratna, M. R., and Jimenez, J. L.: Hydrocarbon-like and oxygenated  
1091 organic aerosols in Pittsburgh: insights into sources and processes of organic aerosols, *Atmospheric*  
1092 *Chemistry And Physics*, 5, 3289-3311, 10.5194/acp-5-3289-2005, 2005.

1093 Zhang, Q., Zheng, Y., Tong, D., Shao, M., Wang, S., Zhang, Y., Xu, X., Wang, J., He, H., Liu, W., Ding,  
1094 Y., Lei, Y., Li, J., Wang, Z., Zhang, X., Wang, Y., Cheng, J., Liu, Y., Shi, Q., Yan, L., Geng, G., Hong, C.,  
1095 Li, M., Liu, F., Zheng, B., Cao, J., Ding, A., Gao, J., Fu, Q., Huo, J., Liu, B., Liu, Z., Yang, F., He, K.,  
1096 and Hao, J.: Drivers of improved PM<sub>2.5</sub> air quality in China from 2013 to 2017,  
1097 *Proceedings of the National Academy of Sciences*, 116, 24463-24469, 10.1073/pnas.1907956116, 2019.

1098 Zhang, R., Jing, J., Tao, J., Hsu, S. C., Wang, G., Cao, J., Lee, C. S. L., Zhu, L., Chen, Z., Zhao, Y., and  
1099 Shen, Z.: Chemical characterization and source apportionment of PM<sub>2.5</sub> in Beijing:  
1100 seasonal perspective, *Atmos. Chem. Phys.*, 13, 7053-7074, 10.5194/acp-13-7053-2013, 2013.

1101 Zhang, X. Y., Zhuang, G. S., Guo, J. H., Yin, K. D., and Zhang, P.: Characterization of aerosol over the  
1102 Northern South China Sea during two cruises in 2003, *Atmospheric Environment*, 41, 7821-7836,  
1103 10.1016/j.atmosenv.2007.06.031, 2007a.

1104 Zhang, Y.-x., Shao, M., Zhang, Y.-h., Zeng, L.-m., He, L.-y., Zhu, B., Wei, Y.-j., and Zhu, X.-l.: Source  
1105 profiles of particulate organic matters emitted from cereal straw burnings, *Journal of Environmental*  
1106 *Sciences*, 19, 167-175, [https://doi.org/10.1016/S1001-0742\(07\)60027-8](https://doi.org/10.1016/S1001-0742(07)60027-8), 2007b.

1107 Zhang, Y., Schauer, J. J., Zhang, Y., Zeng, L., Wei, Y., Liu, Y., and Shao, M.: Characteristics of Particulate  
1108 Carbon Emissions from Real-World Chinese Coal Combustion, *Environ. Sci. Technol.*, 42, 5068-5073,  
1109 10.1021/es7022576, 2008.

1110 Zhang, Y., Lang, J., Cheng, S., Li, S., Zhou, Y., Chen, D., Zhang, H., and Wang, H.: Chemical  
1111 composition and sources of PM<sub>1</sub> and PM<sub>2.5</sub> in Beijing in autumn, *Science of The Total Environment*,  
1112 630, 72-82, <https://doi.org/10.1016/j.scitotenv.2018.02.151>, 2018.

1113 Zhang, Y., Vu, T. V., Sun, J., He, J., Shen, X., Lin, W., Zhang, X., Zhong, J., Gao, W., Wang, Y., Fu, T.  
1114 M., Ma, Y., Li, W., and Shi, Z.: Significant Changes in Chemistry of Fine Particles in Wintertime Beijing  
1115 from 2007 to 2017: Impact of Clean Air Actions, *Environ. Sci. Technol.*, 54, 1344-1352,  
1116 10.1021/acs.est.9b04678, 2020.

1117 Zhao, B., Zheng, H., Wang, S., Smith, K. R., Lu, X., Aunan, K., Gu, Y., Wang, Y., Ding, D., Xing, J., Fu,  
1118 X., Yang, X., Liou, K.-N., and Hao, J.: Change in household fuels dominates the decrease in  
1119 PM<sub>2.5</sub> exposure and premature mortality in China in 2005–2015, *Proceedings of the*  
1120 *National Academy of Sciences*, 115, 12401-12406, 10.1073/pnas.1812955115, 2018.

1121 Zhao, X., Hu, Q., Wang, X., Ding, X., He, Q., Zhang, Z., Shen, R., Lü, S., Liu, T., Fu, X., and Chen, L.:  
1122 Composition profiles of organic aerosols from Chinese residential cooking: case study in urban  
1123 Guangzhou, south China, *Journal of Atmospheric Chemistry*, 72, 1-18, 10.1007/s10874-015-9298-0,  
1124 2015.  
1125 Zhao, Y., Hu, M., Slanina, S., and Zhang, Y.: Chemical Compositions of Fine Particulate Organic Matter  
1126 Emitted from Chinese Cooking, *Environ. Sci. Technol.*, 41, 99-105, 10.1021/es0614518, 2007.  
1127 Zheng, B., Tong, D., Li, M., Liu, F., Hong, C., Geng, G., Li, H., Li, X., Peng, L., Qi, J., Yan, L., Zhang,  
1128 Y., Zhao, H., Zheng, Y., He, K., and Zhang, Q.: Trends in China's anthropogenic emissions since 2010 as  
1129 the consequence of clean air actions, *Atmos. Chem. Phys.*, 18, 14095-14111, 10.5194/acp-18-14095-  
1130 2018, 2018.  
1131 Zheng, M., Salmon, L. G., Schauer, J. J., Zeng, L., Kiang, C. S., Zhang, Y., and Cass, G. R.: Seasonal  
1132 trends in PM<sub>2.5</sub> source contributions in Beijing, China, *Atmospheric Environment*, 39, 3967-3976,  
1133 <https://doi.org/10.1016/j.atmosenv.2005.03.036>, 2005.  
1134 Zhou, J., Xiong, Y., Xing, Z., Deng, J., and Du, K.: Characterizing and sourcing ambient PM<sub>2.5</sub> over key  
1135 emission regions in China II: Organic molecular markers and CMB modeling, *Atmospheric Environment*,  
1136 163, 57-64, <https://doi.org/10.1016/j.atmosenv.2017.05.033>, 2017.  
1137 Zhou, W., Wang, Q., Zhao, X., Xu, W., Chen, C., Du, W., Zhao, J., Canonaco, F., Prévôt, A. S. H., Fu, P.,  
1138 Wang, Z., Worsnop, D. R., and Sun, Y.: Characterization and source apportionment of organic aerosol at  
1139 260 m on a meteorological tower in Beijing, China, *Atmos. Chem. Phys.*, 18, 3951-3968,  
1140 10.5194/acp-18-3951-2018, 2018.  
1141