Source apportionment of atmospheric mercury in the remote marine atmosphere: Mace Head GAW station, Irish west coast

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10 Abstract

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We examined recent atmospheric mercury concentrations measured with a high temporal resolution of 15 min. at Mace Head, a GAW station on the west coast of Ireland. We attributed a direct contribution of 34% (0.44 ng m⁻³) to primary sources. Additionally, a steep decline (0.05 ng year⁻¹) in mercury concentrations was observed between 2013 and 2018.

Using a stereo algorithm we reconstructed 99.9% of the atmospheric mercury. A conservative analysis demonstrated no decreasing of TGM associated with atmospheric species typically used as tracers for oceanic emissions. The results show that the atmospheric mercury mass is mainly loaded in a baseline factor with an on-going decline. Moreover, we exploit temporal variation and wind pattern effects in the measured atmospheric species, the results show that the diurnal variation and seasonality in TGM observed in Mace Head is closely related to other species linked to primary sources and can be explained by transport from continental areas.

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24 **1. Introduction**

Atmospheric mercury is a bioaccumulative, toxic pollutant with the potential to be transported over large distances that poses a significant public health and environmental problem (WHO, 2007).

27 Despite efforts by governments and international agencies as well as the private sector to reduce 28 mercury release into the environment, current environmental levels are often still of concern.

Atmospheric mercury is emitted from both natural and anthropogenic sources as well as through recycling of past emissions. Natural sources are comprised of release from volcanoes, weathering of

31 rocks, forest fires and oceanic emissions. Anthropogenic sources are related to fossil fuel combustion,

32 cement production, industrial activities, mining and municipal or medical waste incineration. Mercury

- 33 is also reintroduced into the atmosphere through natural processes such as oceanic evaporation after
- 34 reduction of inorganic oxidized Hg in anaerobic environments, which leads to global cycling of this
- 35 element (Corbitt et al., 2011; Streets et al., 2011). The source contribution, as well as the life-time of
- 36 atmospheric mercury, is only roughly estimated.
- The 2018 Global Mercury Assessment (UN, 2018) reveals that primary anthropogenic mercury emissions into the air are 2220 t/y, also indicating an increase of 20% from such sources in recent

39 years. The 2018 UNEP Report (AMAP/UNEP, 2018) presents an inventory for the year 2015, which

40 indicates that the greatest atmospheric mercury emissions resulted from combustion of fossils fuels,

41 mainly coal. While mercury in the atmosphere is chemically inert, once released into this environment,

42 all sources are of concern.

To compile a global assessment based on inventories requires a number of assumptions and generalizations (AMAP/UNEP, 2018). Several discrepancies are observed in the mass balance-based estimation: there can be large differences between estimates, and it is important to recognize that there are sources of error in all methods for estimating mercury emissions.

- Here we report concentrations of atmospheric mercury (TGM: total gaseous mercury) measured from January 2013 to March 2018 at Mace Head. Mace Head station is located within the central North Eastern Atlantic region and based on a GEOS-Chem simulation it is one of the most influenced region by a decreasing mercury trend in ocean surface water, according to Soerensen et al.
- 51 (2012).
- 52 Using the relationship between mercury and other chemical atmospheric trace species (O₃, CFC-12,
- 53 CCl₄, N₂O, CH₄, CHCl₃, CO and H₂) and meteorological data (wind speed and direction), we performed

54 a mass balance to reconstruct atmospheric mercury. Solved by positive matrix factorization, the total

55 mercury mass was distributed into four different factors, classified as baseline, combustion, oceanic 56 and a fourth factor and then each of them was assessed for source trends.

- 57 Time series analysis of atmospheric mercury concentrations at Mace Head were already 58 reported by Weigelt et al. (2015) and Ebinghaus et al. (2011).
- In this work we apply a new approach for source apportionment and extend the time seriesanalysis up to March 2018.
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62 **2. Experimental Setup**

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64 **2.1. Sampling site and analytical methods**

65 Mace Head atmospheric research station is located on the west coast of Ireland at 53.33°N and 9.54°W,

55 km from Galway (80,000 inhabitants), the nearest city with significant industrial activity. It is a GAW

baseline station, exposed to the North Atlantic Ocean and is an ideal location to study both natural
and anthropogenic trace constituents in marine and continental air masses (Stanley et al., 2018).

69 In addition to atmospheric mercury, meteorological parameters are routinely monitored

70 (https://www.met.ie/). Atmospheric CFC-11, CFC-12, CHCl₃, CCl₄, N₂O, CH₄, CO and H₂ are measured

71 (Figure S1) as part of the AGAGE project (https://agage.mit.edu/).

72 TGM is monitored by an automated dual channel, single amalgamation, cold vapour atomic

73 fluorescence analyser (Tekran Analyzer Model 2537B, Tekran Inc., Toronto, Canada) described by

Ebinghaus et al. (2011). At some some level, instrument failure is inevitable, they are susceptible to malfunctions that can result in lost or poor-quality data. Some data quality control steps are taken to minimize the risk of loss and to improve the overall quality of data. Validation process, In order to ensure data reliability and comparability of Mace Head mercury data follows GMOS-Data Quality Management (G-DQM) protocol described by D`Amore et al. (2015) through a human check at Helmholtz-Zentrum Geesthacht.

The air-sampling inlet is located on a tower at 10m agl (18m amsl) with a rain shield only. Air is sampled at a flowrate of 1 L/min through unheated PTFE tubing (1/4" O.D.) to the instrument, which is located in an air-conditioned laboratory. As reported by Weigelt (2015), a PTFE pre-filter (pore size 0.2 mm) at the inlet of the instrument protects the sampling cartridges from contamination by particles. The device is operated with a temporal resolution of 15 minutes, calibrated every 25 hours using an internal mercury permeation source. The device has a detection limit of ~0.1 ng m⁻³ (Weigelt et al., 2015).

87 The wind streamlines for near surface level conditions were assessed from 88 https://earth.nullschool.net/ and long-range transport of air pollutants was calculated using the 89 HYSPLIT model (Draxler and Rolph, 2003) from NOAA (National Oceanic and Atmospheric 90 Administration).

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92 **2.2.** Source assessment / Probability mass function

Apportionment of atmospheric species is often performed by receptor models that are based on themass conservation principle:

95 The inclusion of the potential rotated infinity matrices transformation produces factors that 96 appear to be closer to realistic chemical profiles of sources:

$$x_{ij} = \sum_{k=1}^{p} g_{ik} f_{jk}$$
i=1,2,...,*m j*=1, 2.... *n* (1)

98 where x_{ij} is the concentration of the species *j* in the *i*th sample, g_{ik} is the contribution of the factor 99 (associated to a source) k^{in} in the *i*th sample and f_{jk} is the concentration of the species *j* in factor *k* as 100 *presented* by Paatero and Hopke (2003) and described by *Comero et al. (2009)*. This equation can be 101 solved by the probability mass function in *positive matrix factorization* (PMF) (Paatero and Tapper, 102 1994) with the Multilinear Engine (ME-2) developed by Paatero (1999) and implemented in Version 5 103 of the US EPA PMF (https://www.epa.gov/air-research/positive-matrix-factorization-model-104 environmental-data-analyses).

PMF is a stereo algorithm where analytical data sets are combined to create fingerprints and the profile is used to assess the contribution of each source based on the mass load, also providing a robust uncertainty estimation and source diagnostics. In this study, PMF was applied to the Mace Head 108 dataset with an hourly time resolution for the period 2013 to 2018. The results were constrained to 109 provide positive factor contribution. The uncertainty input in the matrix was estimated based on the 110 analytical accuracy of each individual species reported in Stanley et al. (2018) and Weigelt et al

111 (2013).

- The method provides a better qualitative solutions and time resolution of sources than principal component analysis (PCA) (Huang et al., 1999) or chemical mass balance (CMB) since PMF can generate source profiles ("learning algorithm") and let input of uncertainties which allow individual treatment
- 115 of matrix elements.
- 116 In the PMF the weighted factorization regression analysis is based on positive rotable factorization of
- 117 non-singular matrix T;
- 118 $X = FG + E = GTT^{-1}F + E = \overline{G}\overline{F} + E$,

119 where the new rotated factors are

120 $\overline{G} = G T$ and $\overline{F} = T^{-1} F$ as reported by Comore et al. (2009), then the factors are no-negatively 121 constrained.

Factors contributions are chosen on the basis of a matching strength score by using a form of discrete correlation. At the first interaction any matches which have the highest matching strength for primitives mass reconstruction that formed them are immediately chosen as reconstructed. Then, in accordance with the uniqueness constraint, all other matches associated with the primitives that have been formed for each chosen match are eliminated from further consideration. This allows further matches that were not either previously accepted or eliminated to propagate the process of PMF to a satisfactory solution if the propagation converges.

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130 **3. Results and discussion**

131 Time series of TGM concentrations composed of 48,914 hours of measurements covering the period 132 from January 2013 to March 2018 are given in Figure 1. Concentrations range from 0.9 to 3.3 ng m⁻³, 133 displaying a central tendency of 1.3 ± 0.2 ng m⁻³. TGM concentrations in the northern hemisphere have 134 been decreasing in recent decades (Ebinghaus et al., 2011; Slemr et al., 2003). For instance, Ebinghaus 135 et al. (2011) reported a decline trend of 0.028 ± 0.01 ng m⁻³ yr⁻¹ from 1996 to 2009. Account the more 136 recent years (1996 to 2018, March), this decline continued with approximately 0.025 ± 0.04 ng m⁻³ yr⁻¹ 137 ¹, figure 2. This observation could reflect a trend in global emissions, as mercury, roughly, has an 138 atmospheric lifetime of 0.5 to 1 year (Holmes et al., 2006; Lindberg et al., 2007; Si and Ariya 2018). The 139 increasing improvement of manufacturing processes involving mercury and regulations limiting the 140 emissions from coal-fired power plants since the 1980s (Hylander and Meili, 2003; Pirrone et al., 2009) 141 could be a possible reason for this observed decline at Mace Head. Jiskra et al. (2018) report the Hg⁰

(2)

- uptake by vegetation as an alternative mechanism for driving mercury depletion in the NorthernHemisphere atmosphere over the past 20 years.
- 144 However, this decreasing trend is inconsistent with the increased emissions from 1990 to 2015, as
- 145 indicated by anthropogenic Hg emission inventories (e.g., UN, 2018 and AMAP/UNEP, 2018). This
- 146 conundrum related to increasing global emissions on one hand and measured declines in
 - 147 atmospheric mercury is discussed by Zhang et al. (2016). They state that the inventories do not
 - account for the decline in the atmospheric release of Hg from commercial products, and do not
 - 149 properly account for the change in Hg^0/Hg^{II} speciation of emissions from coal-fired utilities
 - 150 after implementation of gases emission controls.

$151 \qquad \textbf{3.1. Temporal and wind pattern effects in mercury concentrations}$

152 Plots of TGM as a function of wind speed and direction can be seen in Figure 3 as well as the polar 153 frequency plot of wind direction. Concentrations of mercury are higher when winds come from the 154 east (continental air masses) and lower for winds from the west and northwest (Atlantic air masses). 155 The higher concentrations to the east are likely to be influenced by urban agglomerations, such as in 156 Galway, Dublin or even the UK and continental Europe. These higher levels observed to the east are 157 associated with relatively strong wind speeds of 15ms⁻¹, which could indicate a relatively distant 158 source. Furthermore, an increase of TGM with strong winds of 20 ms⁻¹ was observed, indicating sources 159 at further distances in air masses coming from westerly and south-westerly directions. 96-hour back 160 trajectories show that these high TGM concentrations at Mace Head were affected by air mass 161 transport from the Iberian Peninsula and long-range transport from North America.

Higher mercury concentrations under the influence of easterly and strong westerly/southwesterly winds closely resemble those of other pollutants that are also closely linked to anthropogenic emissions, such as carbon monoxide, and suggest TGM enrichment from continental air masses.

166 The polar plot shows low concentrations of mercury associated with strong and weak winds 167 coming from the North Sea and nearby land air masses, with in < 10 m s⁻¹.

The diurnal cycle of elemental mercury (Hg⁰) has been discussed extensively (Laurier et al., 2003; Weiss-Penzias et al., 2003; Laurier and Mason, 2007; Xia et al., 2010; Obrist et al., 2011; Moore et al., 2013; Wang et al., 2014; Ci et al., 2015; Wang et al., 2017; Castagna et al., 2018, Jiskra et al., 2018). Kalinchuk et al. (2019) reported solar radiation-driven increase and decrease of mercury concentrations in the Sea of Japan and in the Sea of Okhotsk, respectively. They assumed that the decrease in Hg⁰ concentrations in the marine boundary layer during daytime is mainly caused by its oxidation, catalyzed by active halogen species (mainly by atomic bromine radicals), which are released

- 175 from sea salt aerosols as Br₂ and could be transformed into reactive forms as a result of photolysis
- 176 (Holmes et al., 2009; Sprovieri et al., 2010; Mao and Talbot, 2012; Moore et al., 2013; Si and Ariya,

- 177 2018). However, the absence of a diurnal cycle for mercury is reported in several studies and more
 178 research should be done to confirm the catalytic photolysis oxidation, as large uncertainties exist in
 179 the gas-phase reaction of mercury (Si and Ariya, 2018).
- 180 With a standard electrode potential (E^{0}) of +0.85 V and a kinetic coefficient of reactivity of <9.8 × 10⁻¹³

to 2.1×10^{-12} cm³ molec⁻¹ s⁻¹, at 1 atm and 298 K (Khalizov et al., 2003; Shepler et al., 2007; Subir at al., 2011; Sun et al., 2016), Hg⁰ is a chemically relatively inert towards gas-phase oxidation, and a significant daily mass depletion by photooxidation is very unlikely.

- Seasonality and diurnal patterns for mercury concentrations at Mace Head have been detected, but similar patterns were observed for CO. As presented in Figure 4, wind direction was a driving factor for diurnal cycling of TGM at Mace Head as well as for CO and CHCl₃. Winds from the east (land breezes) showed sharp increases of TGM, CO, CFC-12 and CCl₄ (figure 3 and Figure S3). Conversely, an increase
- 188 of $CHCl_3$ in offshore winds (sea breezes) was observed.
- 189 Mace Head is mostly influenced by air masses from the Atlantic Ocean, however, as a coastal site can 190 be affected by on-shore breezes blowing from land to the North Atlantic. Daily fluctuations of wind
- 191 speed and direction in coastal areas are a result of differences in air pressure created by the different
- 192 heat capacities of water and dry land (Yan Y.Y., 2005).
- 193 Decrease of atmospheric mercury concentrations during warm periods has often been linked to 194 increased Hg^{2+} by catalytic mercury oxidation in the surface layer of the sea due to several chemical 195 and biological processes, mainly controlled by solar radiation (Kalinchuk et al., 2019 and references 196 therein). Si and Ariya (2018) and references therein reported maximum oxidation of mercury in 197 summer based on several atmospheric models but failed to reconstruct observed summer depletion 198 of atmospheric mercury at monitoring sites in North America and Europe. Furthermore, deposition 199 models could not predict the observed large seasonal variability of either Hg oxidation or wet 200 deposition flux (Travnikov et al., 2017).
- Figure 4 shows that the decrease of TGM during summer is closely related to CO depletion in this season.

203 In addition, it was observed similarity among TGM depletion during summer, enhancement during 204 autumn and seasonality of chloroform (CHCl₃). Decreased emissions of CHCl₃ from seawater or more 205 intense depletion by photooxidation during summer may be possible explanations. It should be noted 206 that any photochemical pattern of those species must be considered with caution because $CHCI_3$ is a 207 shorter-lived species (lifetime ~0.5yr), mainly produced in the ocean by biological processes that follow 208 a different oxidation pathway than mercury (Khalil and Rasmussen, 1999). It should also be noted that 209 wind pattern differences were observed within one year for Mace Head: strong winds during winter 210 predominately comes from the sea, and relatively calm winds during summer (Figure S2). This should 211 also be reflected in the observed seasonality of TGM concentrations.

- Figure 3, 4 and S3, show that the seasonality in TGM observed in Mace Head is closely related to other
- 213 species linked to primary sources and can be explained by transport from continental areas.
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215 **3.2. Source apportionment**

216 Figure 1 shows the set of four factors reconstructing atmospheric mercury concentrations obtained 217 from the PMF solution. As reported by Henry (1991), the first set of natural physical constraints of the 218 system to be considered in any approach for identifying and quantifying source mass contributions 219 must be the reconstruction of the original data set by the algorithm—that is, the solution must explain 220 the observations. Figure 5 shows that the sum of the predicted elemental mass contributions for all 221 sources is almost the same as the total TGM measured. Lower reconstruction performance was 222 observed in particular for concentrations higher than 2 ng m⁻³, which make up 0.44% of the 223 observations. One factor with a high load of O₃ and CO was found by the PMF solution which appeared 224 to be irrelevant for the mercury mass balance, as its load was just 0.003 ng m⁻³ (~ 0 %). However, for 225 atmospheric mercury concentrations higher than 2 ng m⁻³ this factor had a load of 0.57 ng m⁻³, and was 226 labeled as fourth factor.

The first factor with a loading of 66% of TGM mass (0.88 ng m⁻³) was labelled as baseline because it does not show any wind pattern, carries high loads of long-lived species such as CFCs and low loads of CO or sea-borne trace gas species. The PMF results show a statistically significant decrease in the baseline factor that could explain almost all of the trend changes in atmospheric mercury. This suggests a major decrease of anthropogenic inputs on a global scale. Slemr et al. (2011) reported a worldwide trend of atmospheric mercury, showing an equally strong decrease in the northern and southern hemispheres, which supports the argument of baseline-driven TGM decline.

According to Streets et al. (2011), anthropogenic Hg emissions in the USA and Europe decreased by 20% and 40%, respectively, from 1990 to 2008. However, emissions on a global scale, particularly from East Asia, are poorly reported (UN, 2018), even for most of the countries that are signatories of Minamata convention (UN, 2019). Moreover, the total emissions from small scale artisanal gold mining are highly uncertain estimates.

Another possible explanation for the declining trend may be the Hg^0 atmospheric life-cycling reduction due to atmospheric acidification caused by CO_2 increase and its potential (E°) to force elemental mercury oxidation. As reported by Slemr et al. (2011) and references therein, an increase in the atmospheric reactivity can induce large decreasing trends in the concentration of many long-lived substances. Clerbaux and Cunnold, (2007) did not observe lifetime changes for halogenated and other greenhouse gases, however, changes in oxidation rates of elemental mercury in the atmosphere could follow different kinetics. Furthermore, the increasing UV radiation and the shifting solar radiation to shorter wavelengths could also intensify the oxidation of elemental mercury into Hg^{2+} (IPCC, 2007; Qureshi et al., 2010). Based on a global box-model of mercury biogeochemical cycling Streets et al. (2011) present a trend of atmosphere mercury from 1850 to 2008 showing the increase of Hg^{2+} in the atmosphere in recent decades. Jiskra et al. (2018), on the other hand, hypothesise that increased vegetation uptake could be a reason for decreasing atmospheric mercury concentrations in recent years.

- 252 A second factor that contributes to mercury with 0.27 \pm 0.13 ng m⁻³ (21 %) and is characterised by a 253 high load of CO and labelled as combustion. The load of mercury in combustion factor increase to 0.53 254 ng m⁻³ for mercury concentrations higher than 2 ng m⁻³ being twice as high as for concentrations below 255 2ng/m⁻³ in this sector (Figure 8). A decreasing trend was observed in this factor, but this is a more 256 complex case because a higher load of Hg in the combustion factor could be strongly influenced by 257 wind direction, as shown in Figure 6. Moreover, seasonality observed in the factors fingerprinted by 258 CHCl₃ and CO (Figure 7) should, however, be considered with caution because those short-lived species 259 (CHCl₃ 4-5 months and CO 1-3 months) have lifetimes that vary by season, which can dampen mercury 260 load into its factor during summer.
- For the potential seasonality, significant trends are also difficult to establish due to the relatively short time series. The Global Mercury Assessment inventory (UN, 2018) estimates the contribution of combustion sources to atmospheric mercury at 24%.
- The wind patterns for the baseline, combustion and sea factors (discussed below) as displayed in the polar plot of Figure 6 indicate an interpretation of the PMF profile with "combustion" being mostly associated with easterly transport, "sea" being linked to north-westerly and south-westerly winds. The "baseline" factor does not correlate with any significant wind patterns.
- Another hand, no seasonality was observed for the baseline factor, linking lower concentrations of mercury in the warm season mainly to transport or evasion patterns and less to deposition by oxidation. For instance, no evidences of photooxidation increase in growing season was reported by Weigelt at al. (2013) which shows no significant seasonality in gaseous elemental mercury and gaseous oxidised mercury in a remote rural environment in Germany.
- Human activity has substantially increased the ocean mercury reservoirs and consequently the fluxes
 between the ocean and atmosphere (Strode et al., 2007; Smith-Downey et al., 2010).
- 275 The residence time of mercury in the ocean is substantially longer than in the atmosphere, ranging
- from years to decades or millennia (Strode et al., 2007; Primeau and Holzer, 2006). Acidification of
- 277 oceans, climate change, excess nutrient inputs, and pollution are fundamentally changing the ocean's
- 278 biogeochemistry (Doney, 2010) and will certainly also influence mercury ocean-air fluxes (Slemr et al.,
- 279 2011). The extent, however, and even the direction of the change is unknown.

- 280 Mason et al. (2012) estimate that global oceanic Hg⁰ evasion to be comparable to anthropogenic 281 emissions, and Sunderland and Mason (2007) attributed the mercury emitted from seawater in the 282 North Atlantic to the legacy of 20th-century anthropogenic sources in Europe and North America.
- This study shows an oceanic contribution (based on ocean factor solved by PMF) of 13% (0.17 \pm 0.07 ng m⁻³) to atmospheric TGM at Mace Head station. Based on atmospheric mercury concentration trends in the subsurface seawater Soerensen et al. (2012) predicted a decrease of approximately 0.045 ng m⁻³ yr⁻¹ of oceanic mercury emissions into the air over the North Atlantic. They also argued, based on cruise data that the decrease of oceanic emissions is forcing the atmospheric trend down. In this study, based on the PMF solution, we found no evidence for a decreasing mercury load in the oceanic factor, which could be traced by CHCl₃ and CH₄ concentrations.
- 290 Moreover, we find from the PMF solution that the decrease of atmospheric mercury is linked less to 291 oceanic emissions and is explained mainly by a baseline factor with a low load of short-lived species 292 with significant anthropogenic sources, such as CO and O₃, as well as a low load of sea trace species, 293 such as CHCl₃ and CH₄.
- A decrease in mercury is observed in the factor with higher loading of long-lived species such as CFCs. However, the presented solution for apportionment of atmospheric mercury has restrictions and requires further consideration, as the mercury sources are complex and numerous, and merely a few source tracers were used in this study.
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4. Conclusions

- This study presents a comprehensive source assessment of atmospheric mercury measured at Mace Head, a baseline station with a long-term decreasing trend of TGM. Positive matrix factorization was applied to a set of atmospheric mercury data from 2013 to 2018 with high temporal resolution. The profiles of source factor contributions indicate that baseline (0.86 ng m⁻³, 66%) and combustion processes (0.27 ng m⁻³, 21%) are the controlling factors of mercury in the atmosphere at this remote coastal measurement location. The high load of mercury in the baseline factor reflects the relatively long lifetime of this species in the atmosphere.
- Biogenic activities in the ocean were identified as another primary source, contributing 13 %
 (0.17 ng m⁻³).
- Therefore, based on the analysis of temporal changes in the sources, no decreasing in the oceanic factor in the period of this study could be detected. The decrease in atmospheric mercury concentrations was linked to the baseline factor. Source contributions by wind sector were also exploited, based on directional wind dependence of source loadings from the PMF

- analysis. The patterns are also consistent with the location of the sources: oceanic sources
- 315 coming from the west (Atlantic) and anthropogenic sources coming from east (Europe) of
- 316 Mace Head. Furthermore, more extensive and detailed descriptions concerning mercury
- 317 sources is needed to confirm and evaluate the reported trends, which then can have great
- 318 relevance for policy and regulations in light of the Minamata convention.
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- 330 References
- AMAP/UNEP: Technical Background Report for the Global Mercury Assessment 2018. United Nations
 Environment Programme (UNEP), 2018.
- Castagna, J., Bencardino, M., D'Amore, F., Esposito, G., Pirrone, N., Sprovieri, F.: Atmospheric mercury species
 measurements across the Western Mediterranean region: Behaviour and variability during a 2015 research
 cruise campaign. Atmos. Environ. 173, 108e126. https://doi.org/10.1016/J.ATMOSENV.2017.10.045, 2018.
- Ci, Z., Wang, C., Wang, Z., Zhang, X.: Elemental mercury (Hg⁰) in air and surface waters of the Yellow Sea during
 late spring and late fall 2012: Concentration, spatial-temporal distribution and air/sea flux. Chemosphere 119,
 199e208. https://doi.org/10.1016/j.chemosphere.2014.05.064, 2015.
- Clerbaux, C., and Cunnold, D.M.: Long-lived compounds, in: "Scientific Assessment of Ozone Depletion: 2006",
 WMO, Geneva, 2007.
- Comero, S., Capitani, L., Gawlik, B.M.: Positive Matrix Factorisation (PMF): An introduction to the chemometric
 evaluation of environmental monitoring data using PMF. European Commission. EUR 23946 EN, ISBN 978-92 79-12954-4 ISSN 1018-5593, DOI 10.2788/2497, 2009.
- Corbitt, E. S., D. J. Jacob, C. D. Holmes, D. G. Streets, and E. M. Sunderland: Global source receptor relationships
 for mercury deposition under present-day and 2050 emissions scenarios, Environ. Sci. Technol., 45, 10,477–
 10,484, doi:10.1021/es202496y, 2011.
- D'Amore, F., Bencardino, M., Sergio Cinnirella, S., Sprovieria, F., Pirrone, N.: Data quality through a web-based
 QA/QC system: implementation for atmospheric mercury data from the Global Mercury Observation
 System. Environmental Science Processes Impacts, 17, 1482–1491. DOI: 10.1039/c5em00205b, 2015.
- Doney, S. C.: The growing human footprint on coastal and openocean biogeochemistry, Science, 328, 1512–
 1516, 2010.
- 353Draxler, R. R., and G. D. Rolph: HYSPLIT (HYbrid Single-ParticleLagrangian Integrated Trajectory) Model access via354NOAAARLREADYWebsite,NOAAAirResour.Lab.,SilverSpring,Md.(Availableat355http://www.arl.noaa.gov/ready/hysplit4.html), 2003.
- Ebinghaus, R., Jennings, S.G., Kock, H.H., Derwent, R.G., Manning, A.J., Spain, T.G.: Decreasing trend in total
 gaseous mercury observations in baseline air at Mace Head, Ireland, from 1996 to 2009. Atmos. Environ. 45,
 3475e3480, 2011.
- Henry, R.C.: Multivariate Receptor Models, In: Receptor Modeling for Air Quality Management, P.K. Hopke, ed.,
 Elsevier Science Publishers, Amsterdam, 117-147, 1991.

- Holmes, C. D, Jacob, D. J. Yang, X.: Global lifetime of elemental mercury against oxidation by atomic bromine in
 the free troposphere, Geophys. Res. Lett., 33, L20808, doi:10.1029/2006GL027176, 2006.
- Holmes, C.D., Jacob, D.J., Mason, R.P., Jaffe, D.A.: Sources and deposition of reactive gaseous mercury in the
 marine atmosphere. Atmos. Environ. 43, 2278e2285. <u>https://doi.org/10.1016/J.ATMOSENV.2009.01.051</u>,
 2009.
- Huang, S., Rahn, K.A. Arimoto, R.: Testing and Optimizing Two Facot-Analysis Techniques on Aerosol at
 Narragansett, Rhode Island, Atmospheric Environ. 33:2169-2185, 1999.
- 368 IPCC: Climate Change 2007: Synthesis Report, Geneva, Switzerland, 2007.
- Hylander, L. D. Meili, M.: 500 years of mercury production: global annual inventory by region until 2000 and
 associated emissions, Sci. Total Environ. 304, 13–27, 2003.
- Kalinchuk, V., Aksentov, K., Karnaukh, V.: Gaseous elemental mercury (Hg(0)) in the surface air over the Sea of
 Japan, the Sea of Okhotsk and the Kuril-Kamchatka sector of the Pacific Ocean in August-September 2017.
 Chemosphere 224, 668e679, 2019.
- 374 Khalil, M.A.K., Rasmussen, R.A,: Atmospheric chloroform. 7, 1151-1158, 1999.
- Khalizov, A.F., Viswanathan, B., Larregaray, P., Ariya, P.A.: Theoretical Study on the Reactions of Hg with
 Halogens: Atmospheric Implications. J. Phys. Chem. A, 107, 6360–6365, 2003.
- Jiskra, M., Sonke, J.E., Obrist, D., Bieser, J., Ebinghaus, R., Myhre, C.L., Pfaffhuber, K.A., Wängberg, I., Kyllönen,
 K., Worthy, D., Martin, L.G., Labuschagne, C., Mkololo, T., Ramonet, M., Magand O., Dommergue. A.: A
 vegetation control on seasonal variations in global atmospheric mercury concentrations. NATURE
 GEOSCIENCE | VOL 11 | APRIL 2018 | 244–250, 2018.
- Laurier, F., Mason, R.: Mercury concentration and speciation in the coastal and open ocean boundary layer. J.
 Geophys. Res. 112, D06302. <u>https://doi.org/10.1029/2006JD007320</u>, 2007.
- Laurier, F.J.G., Mason, R.P., Whalin, L.: Reactive gaseous mercury formation in the North Pacific Ocean's marine
 boundary layer: A potential role of halogen, 2003.
- Lindberg, S., Bullock, R., Ebinghaus, R., Engstrom, D., Feng, X., Fitzgerald, W., Pirrone, N., Prestbo, E., and
 Seigneur, Ch.: A synthesis of progress and uncertainties in attributing the sources of mercury in deposition,
 Ambio 36, 19–32, 2007.
- Mao, H., Talbot, R.: Speciated mercury at marine, coastal, and inland sites in New England-Part 1: Temporal
 variability. Atmos. Chem. Phys. 12, 5099e5112. <u>https://doi.org/10.5194/acp-12-5099-2012,</u> 2012.
- Mason, R. P., Choi, A.L., Fitzgerald, W.F., Hammerschmidt, C.R., Lamborg, C.H., Soerensen, A.L., Sunderland, E.
 M.: Mercury bio-geochemical cycling in the ocean and policy implication, Environ. Res.,
 doi:10.1016/j.envres.2012.03.013, in press, 2012.
- Moore, C.W., Obrist, D., Luria, M.: Atmospheric mercury depletion events at the Dead Sea: Spatial and temporal
 aspects. Atmos. Environ. 69, 231e239. <u>https://doi.org/10.1016/J.ATMOSENV.2012.12.020</u>, 2013.
- Obrist, D., Tas, E., Peleg, M., Matveev, V., Faïn, X., Asaf, D., Luria, M.: Bromineinduced oxidation of mercury in
 the mid-latitude atmosphere. Nat. Geosci. 4, 22e26. <u>https://doi.org/10.1038/ngeo1018</u>, 2011.
- Paatero, P.: The multilinear engine a table-driven least squaresprogram for solving multilinear problems,
 including the n-wayparallel factor analysis model, J. Comput. Graph. Stat., 8, 854–888, 1999.
- Paatero, P. and Hopke, P. K.: Discarding or downweighting highnoise variables in factor analytic models, Anal.
 Chim. Acta, 490, 277–289, 2003.
- 401 Paatero, P. and Tapper, U.: Positive matrix factorization: A non-negative factor model with optimal utilization of
 402 error estimates of data values, Environmetrics, 5, 111–126, 1994.
- Pirrone, N., Cinnirella, S., Feng, X., Finkelman, R. B., Friedli, H. R., Leaner, J., Mason, R., Mukherjee, A. B., Stracher,
 G., Streets, D. G., and Telmer, K.,: Global mercury emissions to the atmosphere from natural and
 anthropogenic sources, in: Mercury Fate and Transport in the Global Atmosphere, edited by: Pirrone, N.,
 Mason, R., 3–49, Springer, Dordrecht, 2009.
- 407 Primeau, F. W. and Holzer, M.: The ocean's memory of the atmosphere: Residence-time and ventilation-rate
 408 distributions of water masses, J. Phys. Oceanography 36, 1439–1456, 2006.
- 409 Qureshi, A., O'Driscoll, N. J., MacLeod, M., Neuhold, Y.-M., and Hungerbuhler, K.: Photoreactions of mercury in 410 surface ocean water: Gross reaction kinetics and possible pathways, Environ. Sci. Technol. 44, 644–649, 2010.

- Smith-Downey, N. V., Sunderland, E. M., and Jacob, D. J.: Anthropogenic impacts on global storage and emissions
 of mercury from terrestrial soils: Insights from a new global model, J. Geophys. Res., 115, G03008,
 doi:10.1029/2009JG001124, 2010.
- 414 Shepler, B.C., Balabanov, N.B., Peterson, K.A.: Hg+Br–>HgBr recombination and collision induced dissociation 415 dynamics. J. Chem. Phys., 127, 164–304, 2007.
- 416 Si, L., Ariya, P.A.: Recent Advances in Atmospheric Chemistry of Mercury. Atmosphere (Basel) 9, 76. 417 <u>https://doi.org/10.3390/atmos9020076, 2018</u>.
- Slemr, F., Brunke, E.G., Ebinghaus, R., and Kuss, J.: Worldwide trend of atmospheric mercury since 1995, Atmos.
 Chem. Phys., 11, 4779 4787, doi:10.5194/acp-11-4779-2011, 2011.
- Slemr, F., Brunke, E.-G., Ebinghaus, R., Temme, C., Munthe, J., W€angberg, I., Schroeder, W., Steffen, A., Berg, T.:
 Worldwide trend of atmospheric mercury since 1977. Geophys. Res. Lett. 30 (10), 1516.
 http://dx.doi.org/10.1029/2003GL016954, 2003.
- Soerensen, A.L., Jacob, D.J., Streets, D.G., Witt, M.L.I., Ebinghaus, R., Mason, R.P., Andersson, M., Sunderland,
 E.M.: Multi-decadal decline of mercury in the North Atlantic atmosphere explained by changing subsurface
 seawater concentrations. Geophys. Res. Lett. 39, L21810. http://dx.doi.org/10.1029/2012GL053736, 2012.
- Sprovieri, F., Hedgecock, I.M., Pirrone, N.: An investigation of the origins of reactive gaseous mercury in the
 Mediterranean marine boundary layer. Atmos. Chem. Phys. 10, 3985e3997. <u>https://doi.org/10.5194/acp-10-</u>
 3985-2010, 2010.
- Stanley, K.M., Grant, A., O'Doherty, S., Young, D., Manning, A.L., Stavert, A.R., Gerard Spain, T.G., Salameh, P.K.,
 Harth, C.M., Simmonds, P.G., Sturges, W.T., Oram, D.E. and Derwent, R.G.: Greenhouse gas measurements
 from a UK network of tall towers: technical description and first results. Atmos. Meas. Tech., 11, 1437–1458,
 https://doi.org/10.5194/amt-11-1437-2018.
- Subir, M.; Ariya, P.A.; Dastoor, A.P.: A review of uncertainties in atmospheric modeling of mercury chemistry I.
 Uncertainties in existing kinetic parameters—Fundamental limitations and the importance of heterogeneous
 chemistry. Atmos. Environ., 45, 5664–5676, 2011.
- Sun, G.; Sommar, J.; Feng, X.; Lin, C.-J.; Ge, M.; Wang, W.; Yin, R.; Fu, X.; Shang, L.: Mass-dependent and independent fractionation of mercury isotope during gas-phase oxidation of elemental mercury vapor by
 atomic Cl and Br. Environ. Sci. Technol., 50, 9232–9241, 2016.
- Sunderland, E. M., and Mason, R.P.: Human impacts on open ocean mercury concentrations, Global Biogeochem.
 Cycles, 21, GB4022, doi:10.1029/2006GB002876, 2007.
- Streets, D. G., M. K. Devane, Z. Lu, T. C. Bond, Sunderland, E.M., and Jacob, D.J.: All-time releases of mercury to
 the atmosphere from human activities, Environ. Sci. Technol., 45 (24), 10,485–10,491,
 doi:10.1021/es202765m, 2011.
- Travnikov, O., Angot, H., Artaxo, P., Bencardino, M., Bieser, J., D'Amore, F., Dastoor, A., Simone, F.D., Diéguez,
 M.d.C., Dommergue, A., et al.: Multi-model study of mercury dispersion in the atmosphere: Atmospheric
 processes and model evaluation. Atmos. Chem. Phys., 17, 5271–5295, 2017.
- Strode, S. A., Jaegl'e, L., Selin, N. E., Jacob, D. J., Park, R. J., Yantoska, R. M., Mason, R. P., and Slemr, F.: Air-sea
 exchange in the global mercury cycle, Global Biogeochem. Cycles 21, GB1017, doi:10.1029/2006GB002766,
 2007.
- UN, Global Mercury Assessment: Environment Programme. Chemicals and Health Branch Geneva_ Switzerland,
 ISBN: 978-92-807-3744-8, 2018.
- 452UN,MinamataConventiononMercury:UnitedNationEnvironmentalProgram.453http://www.mercuryconvention.org/Countries/Parties/tabid/3428/language/en-US/Default.aspx2019.
- Wang, Y., Liu, R., Li, Y., Cui, X., Zhou, J., Liu, S., Zhang, Y.: GEM in the marine atmosphere and air-sea exchange of
 Hg during late autumn and winter cruise campaigns over the marginal seas of China. Atmos. Res. 191, 84e93.
 https://doi.org/10.1016/j.atmosres.2017.03.004, 2017.
- Wang, F., Saiz-Lopez, A., Mahajan, A.S., Gomez Martín, J.C., Armstrong, D., Lemes, M., Hay, T., Prados-Roman,
 C.: Enhanced production of oxidised mercury over the tropical Pacific Ocean: a key missing oxidation pathway.
 Atmos. Chem. Phys. 14, 1323e1335. <u>https://doi.org/10.5194/acp-14-1323-2014</u>, 2014.
- Weiss Penzias, P., Jaffe, D.A., McClintick, A., Prestbo, E.M., Landis, M.S.: Gaseous Elemental Mercury in the
 Marine Boundary Layer: Evidence for Rapid Removal in Anthropogenic Pollution, pp. 3755e3763.
 https://doi.org/10.1021/es0341081, 2003.

- Weigelt, A., Ebinghaus, R., Manning, A.J., Derwent, R.G., Simmonds, P., Spain, T.G, Jennings, S.G., Slemr, F.:
 Analysis and interpretation of 18 years of mercury observations since1996 at Mace Head, Ireland.
 Atmospheric Environment 100, 85 e 93, 2015.
- Weigelt, A., Temme, C., Bieber, E., Schwerin, A., Schuetze, M., Ebinghaus, R., and Kock, H.H.: Measurements of
 atmospheric mercury species at a German rural background site from 2009 to 2011 methods and results. *Environ. Chem.10*, 102–110, 2013.
- Wilson, S., Munthe, J., Sundseth, K., Kindbom, K., Maxson, P., Pacyna, P., Steenhuisen, F.: Updating historical
 global inventories of anthropogenic mercury emissions to air, AMAP Tech. Rep. 3, 14 pp., Arct. Monit. and
- 471 Assess. Programme, Oslo, 2010.
- WHO -World Health Organization: EXPOSURE TO MERCURY: A MAJOR PUBLIC HEALTH CONCERN. 20 Avenue
 Appia, CH-1211 Geneva-27, Switzerland (Document available online), 2007.
- 474Xia, C., Xie, Z., Sun, L.: Atmospheric mercury in the marine boundary layer along a cruise path from Shanghai,475ChinatoPrydzBay,Antarctica.Atmos.Environ.44,1815e1821.476https://doi.org/10.1016/J.ATMOSENV.2009.12.039, 2010.
- 477 Yan Y.Y.: Land and Sea Breezes. In: Oliver J.E. (eds) Encyclopedia of World Climatology. Encyclopedia of Earth
 478 Sciences Series. Springer, Dordrecht, 2005.
- 479 Zhang, Y., Jacob, D. J., Horowitz, H. M., Chen, L., Amos, H.M., Krabbenhoft, D. P., Slemr, F., St. Louis, V. L.,
- 480 Sunderland, E. M.: Observed decrease in atmospheric mercury explained by global declining anthropogenic 481 emissions. Proc. Natl. Acad. Sci. U. S. A.,113, 526–531, 2016.
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 Figure 1. TGM hourly variations measured at Mace Head, from 2013 to 2018 (bottom), time series of mercury

- 485 attributed to each factor (center) and time series of sea and combustion only (top).
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487 488 Figure 2. Time series decomposition of TGM (monthly averages) measured at Mace Head from 1996 to February 489 2018. From top to bottom it is presented the monthly time series followed by the patterns of deconstructed

490 components, trend, seasonality and radon. * TGM in ng m⁻³.



Figure 3. Polar plots for TGM (left) and polar wind frequency (right) at Mace Head. * TGM in ng m⁻³ and wind

- 493 speed in ms⁻¹.
- 494



495hourmonthhourmonth496Figure 4: Diurnal cycle and seasonal cycle of mercury and species loaded in the PMF matrix. The shaded areas497are the 95% confidence intervals in the mean. *Wind direction is normalised with West 90° as -1 and east (270°)498as 1.



Figure 5: Correlation among total elemental mercury measured and mercury reconstructed by the PMF solution and conditional quantiles plot showing the difference between PMF solution and observation. The observations are split up into bins according to correspondent reconstructed value. The median prediction line together with the 25/75th and 10/90th quantile values are plotted together with a line showing a "perfect model". It also shown is a histogram of reconstructed values (shaded grey) and a histogram of observed values (shown as a blue line).



Figure 6. Polar plots for the factors obtained in the PMF solution. The plots show variations of mercury (ng m⁻³)

loaded in each factor as a function of wind direction (°) and speed (ms⁻¹).

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564 closure for mercury concentration greater than 2 ng m⁻³ (right).