



1 **Urban organic aerosol composition in Eastern China differs from North to South: Molecular**
2 **insight from a liquid chromatography-Orbitrap mass spectrometry study**

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27 **Abstract:**

28 Particulate air pollution in China is influencing human health, ecosystem and climate. However,
29 the chemical composition of particulate aerosol, especially of the organic fraction, is still not well
30 understood. In this study, particulate aerosol samples with a diameter $\leq 2.5 \mu\text{m}$ ($\text{PM}_{2.5}$) were
31 collected in January 2014 in three cities located in Northeast, East and Southeast China, i.e.,
32 Changchun, Shanghai and Guangzhou, respectively. Organic aerosol (OA) in the $\text{PM}_{2.5}$ samples
33 was analyzed by ultrahigh performance liquid chromatography (UHPLC) coupled to high-
34 resolution Orbitrap mass spectrometry in both negative mode (ESI⁻) and positive mode
35 electrospray ionization (ESI⁺). After a non-target screening including molecular formula
36 assignments, compounds were classified into five groups based on their elemental composition, i.e.,
37 CHO, CHON, CHN, CHOS and CHONS. The CHO, CHON and CHN compounds present the
38 dominant signal abundances of 81–99.7% in the mass spectra and the majority of these compounds
39 were assigned to mono- and polyaromatics, suggesting that anthropogenic emissions are a large
40 source of urban OA in all three cities. However, the chemical characteristics of these compounds
41 varied among different cities. The degree of aromaticity and the number of polyaromatic
42 compounds were significantly higher in samples from Changchun, which could be attributed to the
43 large emissions from residential heating (i.e., coal combustion) during winter time in Northeast
44 China. Moreover, the ESI⁻ analysis showed higher H/C and O/C ratios for organic compounds in
45 Shanghai and Guangzhou compared to samples from Changchun, indicating that OA in lower
46 latitude regions of China experiences more intense photochemical oxidation processes. The
47 majority of sulfur-containing compounds (CHOS and CHONS) in all cities were assigned to
48 aliphatic compounds with low degrees of unsaturation and aromaticity. Again, samples from
49 Shanghai and Guangzhou exhibit a larger chemical similarity but largely differ from those from
50 Changchun.

51 **1. Introduction**

52 In the last decades, China has experienced rapid industrialization and urbanization accompanied by
53 severe and persistent particulate air pollution (Huang et al., 2014; Sun et al., 2014; Ding et al., 2016;
54 Song et al., 2018; Shi et al., 2019; Xu et al., 2019). These particulate air pollution extremes can not
55 only influence the regional air quality and human health in China, but also lead to a global
56 environmental problem due to long-distance transport of pollutants. To better understand the effects
57 of air pollution on air quality and human health, chemical characterization of fine particle
58 (particulate matter with an aerodynamic diameter of less than $2.5 \mu\text{m}$, or $\text{PM}_{2.5}$) is crucial. However,



59 the chemical composition of PM_{2.5} in China is still poorly understood due to a wide variety of
60 natural and anthropogenic sources as well as complex multiphase chemical reactions (Lin et al.,
61 2012a; Huang et al., 2014; Ding et al., 2016; Wang et al., 2017; Wang et al., 2018; An et al., 2019;
62 Wang et al., 2019a; Wang et al., 2019b). In particular, compared to the fairly well understood nature
63 of the inorganic fraction of aerosol, the organic fraction, also named organic aerosol (OA), is
64 considerably less comprehended in terms of chemical composition, corresponding precursors,
65 sources and formation mechanisms (Huang et al., 2017).

66 During pollution events in China, OA accounts for as high as more than 50% of the total mass of
67 fine particle (An et al., 2019). Chemical compounds in OA cover a large complexity of species
68 including alcohols, aldehydes, carboxylic acids, imidazoles, organosulfates, organonitrates and
69 polycyclic aromatic hydrocarbons (PAHs) (Lin et al., 2012a; Rincón et al., 2012; Kourtchev et al.,
70 2014; Wang et al., 2018; Elzein et al., 2019; Wang et al., 2019a). Thus, the capacity of traditional
71 analytical techniques is limited to identify the compounds in OA and the majority (> 70%) of OA
72 has not been identified yet as specific compounds (Hoffmann et al., 2011). The insufficient
73 knowledge of chemical composition of OA hinders a better understanding of the sources, formation
74 and atmospheric processes of air pollution in China.

75 Recently, ultrahigh resolution mass spectrometry (UHRMS), such as Fourier transform ion
76 cyclotron resonance mass spectrometry (FTICR-MS) and Orbitrap-MS, coupled with soft
77 ionization sources (e.g., electrospray ionization (ESI) and atmospheric pressure chemical ionization
78 (APCI)) have been introduced to elucidate the molecular composition of OA (Nizkorodov et al.,
79 2011; Lin et al., 2012a; Lin et al., 2012b; Rincón et al., 2012; Noziere et al., 2015; Kourtchev et al.,
80 2016; Tong et al., 2016; Tu et al., 2016; Brüggemann et al., 2017; Wang et al., 2017; Fleming et
81 al., 2018; Laskin et al., 2018; Song et al., 2018; Wang et al., 2018; Brüggemann et al., 2019;
82 Daellenbach et al., 2019; Ning et al., 2019; Wang et al., 2019a). Due to the two outstanding features
83 of high resolving power and high mass accuracy, UHRMS can give precise elemental compositions
84 of individual organic compounds. However, UHRMS studies on Chinese urban OA are very limited.
85 Wang et al. (Wang et al., 2017) characterized OA in Shanghai and showed variations in chemical
86 composition among different months and between daytime and nighttime. Our recent Orbitrap MS
87 study (Wang et al., 2018) showed that wintertime OA in PM_{2.5} collected in Beijing, China and
88 Mainz, Germany were very different in terms of chemical composition. In contrast, for summertime
89 OA from Germany and China, Brüggemann et al. (2019) found similar compounds and
90 concentrations of terpenoid organosulfates in PM₁₀, demonstrating that biogenic emission can
91 significantly affect OA composition at both locations. Ning et al. (2019) analyzed the OA collected



92 in a coastal Chinese city (Dalian) and found that more organic compounds were identified in haze
93 days compared to non-haze days. Nonetheless, since severe particulate pollution in China occurs
94 on a large-scale, more UHRMS studies are needed to fully elucidate the chemical composition of
95 OA in different Chinese cities.

96 In this study, PM_{2.5} aerosol samples were collected in three Chinese cities, i.e., Changchun,
97 Shanghai and Guangzhou, and their organic fraction was analyzed using ultra-high-performance
98 liquid chromatography (UHPLC) coupled with Orbitrap–MS. The Chinese cities of Changchun,
99 Shanghai and Guangzhou are located in the Northeast, East and Southeast of China, which are
100 major populated regions in China with a population of 7.5, 24 and 15 million, respectively. The
101 geographic locations of these three cities cover a large latitude spanning from 23.12° N to 43.53° N
102 resulting in different meteorological conditions, including intensity and duration of sunlight,
103 average daily temperature and monsoon climate. In addition, the industrial structure, energy
104 consumption and energy sources in these three cities are different, which can cause difference in
105 anthropogenic emissions, and thus, influence the chemical composition of the urban OA. For
106 example, OA is strongly affected by residential coal combustion during winter in Northeast China
107 (Huang et al., 2014; An et al., 2019). Therefore, this study presents a comprehensive overview of
108 chemical composition of OA in three representative Chinese cities during pollution episodes, which
109 eventually can improve our understanding of OA effects on climate and public health and also
110 provide a chemical database for haze mitigation strategies in China.

111 **2. Experimental**

112 **2.1 PM_{2.5} samples**

113 Three 24-h integrated urban PM_{2.5} samples were collected during severe haze pollution events with
114 daily average PM_{2.5} mass concentration higher than 115 $\mu\text{g m}^{-3}$ in each of the three Chinese cities:
115 Changchun (43.54° N, 125.13° E, 1.5 m above the ground), Shanghai (31.30° N, 121.50° E, 20 m
116 above the ground) and Guangzhou (23.07° N, 113.21° E, 53 m above the ground), which are located
117 in the Northeast, East and Southeast regions of China, respectively (see Fig. 1). Samples in
118 Changchun were collected on 4, 24 and 29 of January 2014 with PM_{2.5} mass concentrations of
119 185–222 $\mu\text{g m}^{-3}$, samples in Shanghai were collected on 1, 19 and 20 of January 2014 with PM_{2.5}
120 mass concentrations of 159–172 $\mu\text{g m}^{-3}$ and samples in Guangzhou were collected on 5, 6 and 11
121 of January 2014 with PM_{2.5} mass concentrations of 138–152 $\mu\text{g m}^{-3}$. Further details (e.g., the daily
122 average concentrations of PM_{2.5}, SO₂, NO₂, CO and O₃, and average temperature during sampling
123 dates) are presented in Table S1. All PM_{2.5} samples were collected on prebaked quartz-fiber filters



124 (20.3×25.4 cm) using a high-volume PM_{2.5} sampler at a flow rate of 1.05 m³ min⁻¹ (Tisch
125 Environmental, USA) and at each sampling site field blanks were taken. After sample collection,
126 filters were stored at -20 °C until analysis.

127 **2.2 Sample analysis**

128 Detailed description on the filter sample extraction and UHPLC–Orbitrap MS analysis can be found
129 in our previous studies (Wang et al., 2018; Wang et al., 2019a). Briefly, a part of the filters (around
130 1.13 cm², corresponding to about 600 µg particle mass in each extracted filter) was extracted three
131 times with 1.0–1.5 mL of acetonitrile-water (8/2, v/v) in an ultrasonic bath. The extracts were
132 combined, filtered through a 0.2 µm Teflon syringe filter and evaporated to almost dryness under
133 a gentle nitrogen stream. Finally, the residue was redissolved in 1000 µL acetonitrile-water (1/9,
134 v/v) to reach the total particulate mass concentration of around 600 µg mL⁻¹ for the following
135 analysis.

136 The analytes were separated using a Hypersil Gold column (C18, 50 x 2.0 mm, 1.9 µm particle size)
137 with mobile phases consisting of (A) 0.04% formic acid and 2% acetonitrile in MilliQ water and
138 (B) 2% water in acetonitrile. Gradient elution was applied with the A and B mixture at a flow rate
139 of 500 µL min⁻¹ as follows: 0–1.5 min 2% B, 1.5–2.5 min from 2% to 20% B (linear), 2.5–5.5 min
140 20% B, 5.5–6.5 min from 20% to 30% B (linear), 6.5–7.5 min from 30% to 50% B (linear), 7.5–8.5
141 min from 50% to 98% B (linear), 8.5–11.0 min 98% B, 11.0–11.05 min from 98% to 2% B (linear),
142 and 11.05–11.1 min 2% B. The Q Exactive Hybrid Quadrupole-Orbitrap MS was equipped with a
143 heated ESI source at 120 °C, applying a spray voltage of -3.3 kV and 4.0 kV for negative ESI mode
144 (ESI-) and positive ESI mode (ESI+), respectively. The mass scanning range was set from m/z 50
145 to 500 with a resolving power of 70,000 @ m/z 200. The Orbitrap MS was externally calibrated
146 before each measurement sequence using an Ultramark 1621 solution (Sigma–Aldrich, Germany)
147 providing mass accuracy of the instrument lower than 3 ppm. Each sample was measured in
148 triplicate with an injection volume of 10 µL.

149 **2.3 Data processing**

150 A non-target peak picking software (SIEVE[®], Thermo Fisher Scientific, Germany) was used to find
151 significant peaks in the LC-MS dataset and to calculate all mathematically possible chemical
152 formulas for ions signals with a sample-to-blank abundance ratio ≥ 10 using a mass tolerance of \pm
153 2 ppm. The permitted maximum elemental number of atoms was set as follows: ¹²C (39), ¹H (72),
154 ¹⁶O (20), ¹⁴N (7), ³²S (4), ³⁵Cl (2) and ²³Na (1) (Kind and Fiehn, 2007; Lin et al., 2012a; Wang et



155 al., 2018). To remove the chemically unreasonable formulas, further constraint was applied by
156 setting H/C, O/C, N/C, S/C and Cl/C ratios in the ranges of 0.3–3, 0–3, 0–1.3, 0–0.8 and 0–0.8
157 (Kind and Fiehn, 2007; Lin et al., 2012a; Rincón et al., 2012; Wang et al., 2018; Zielinski et al.,
158 2018), respectively. For chemical formula $C_cH_hO_oN_nS_sCl_x$, the double bond equivalent (DBE) was
159 calculated by the equation: $DBE = (2c + 2 - h - x + n) / 2$, and the aromaticity equivalent (X_c) as
160 a modified index for aromatic compounds was obtained using the equation: $X_c = [3(DBE - (p \times o$
161 $+ q \times n)) - 2] / [DBE - (p \times o + q \times n)]$, where p and q , respectively, refer to the fraction of oxygen
162 and sulfur atoms involved in the π -bond structure of a compound. $X_c \geq 2.50$ and $X_c \geq 2.71$ have
163 been suggested as unambiguous minimum criteria for the presence of monoaromatics and
164 polyaromatics, respectively (Yassine et al., 2014). Further details on the data processing can be
165 found in our previous studies (Wang et al., 2018; Wang et al., 2019a) and is also presented in the
166 Supporting Information (SI).

167 3. Results and discussion

168 3.1 General characteristics

169 The main purpose of this study was to tentatively identify and compare the chemical composition
170 of organic compounds in the $PM_{2.5}$ samples collected in the three Chinese cities: Changchun,
171 Shanghai and Guangzhou during pollution episodes. The number of organic compounds detected
172 in each city and the abundance-weighted average values of molecular mass (MM_{avg}), elemental
173 ratios, DBE and X_c for each subgroup are listed in Table 1. It should be noted that in this study we
174 focus solely on organic compounds with elevated signal abundances, and thus, presumably rather
175 high concentrations. In contrast to our previous study (Wang et al., 2018), compounds with low
176 concentrations were excluded by increasing the reconstitution volume from 500 μ L to 1000 μ L,
177 reducing the sample injection volume from 20 μ L to 10 μ L, and increasing the sample-to-blank
178 ratio from 3 to 10 during data processing.

179 Overall, 416–769 and 687–2943 organic compounds in different samples were determined in ESI–
180 and ESI+, respectively. The largest number of organic compounds was observed in Changchun
181 samples in both ESI– and ESI+, indicating that OA collected during winter season in Northeast
182 China was more complex compared to urban OA in East and Southeast China. This increased
183 number of compounds can possibly be explained by the large residential coal combustion emissions
184 in winter in North China (Huang et al., 2014; Song et al., 2018; An et al., 2019). In addition, ambient
185 temperatures were lowest during the sampling period in Changchun (i.e., -14 °C to -9 °C, Table
186 S1), which likely led to a decreased boundary layer height and therefore enhanced accumulation of



187 pollutants and enhanced formation of secondary organic aerosol through for example gas-to-
188 particle partitioning.

189 As shown in Table 1, the abundance-weighted average values of MM_{avg} , H/C and O/C ratio of the
190 total assigned formulas for Changchun samples detected in negative mode (Changchun⁻) are 169,
191 1.03 and 0.58, respectively, which are significantly lower compared to those for Shanghai⁻ (MM_{avg}
192 = 176, H/C = 1.05 and O/C = 0.69) and for Guangzhou⁻ (MM_{avg} = 183, H/C = 1.14 and O/C =
193 0.74). On the contrary, the aromaticity equivalent X_c for organics detected in Changchun⁻,
194 $X_c(\text{Changchun}^-) = 2.13$, is higher than that for Shanghai⁻, $X_c(\text{Shanghai}^-) = 1.92$, and
195 Guangzhou⁻, $X_c(\text{Guangzhou}^-) = 1.65$. These observations indicate that urban OA in Northeast
196 China features a lower degree of oxidation and a higher degree of aromaticity compared to urban
197 OA in East and Southeast China. The different chemical composition of the samples is probably
198 caused by the rather low ambient temperatures and decreased photochemical processing of organic
199 compounds in Northeast China, slowing down oxidation processes and leading to a larger number
200 of PAHs, which are mainly emitted from coal burning (Huang et al., 2014; Song et al., 2018).

201 Figure 1 shows the reconstructed mass spectra of organic compounds detected in ESI⁻ and ESI⁺.
202 It should be noted that uncertainties always exist when comparing signal abundances of organic
203 compounds due to their different ionization efficiencies and the gradient elution applied in the
204 separation method (Perry et al., 2008). In this study, we assume that all organic compounds have
205 roughly the same mass spectral response and we compare the peak abundances of organic
206 compounds among the different samples. A major fraction organic species detected in ESI⁻ are
207 attributed to CHO⁻ and CHON⁻, accounting for 30–42% and 39–55% in terms of peak abundance,
208 respectively, and comprising 39–45% and 23–33% in terms of peak numbers, respectively. This is
209 consistent with previous studies on Chinese urban OA by Wang et al. (2017 and 2018) and
210 Brüggemann et al. (2019). Comparing the organic compounds detected in ESI⁻ for the three cities,
211 139 formulas were observed in all cities as common formulas (Fig. 2a), accounting for 35–51%
212 and 78–87% of all assigned formulas in terms of peak numbers and peak abundance, respectively.
213 Despite the above-mentioned differences in chemical composition for OA from Changchun
214 compared to OA from Shanghai and Guangzhou, these results demonstrate that still a large number
215 of common organic compounds exist in Chinese urban OAs collected in different cities, in
216 particular for organics with higher signal abundances. Furthermore, as shown by the pie chart in
217 Fig. 2b, these common formulas are dominated by CHON⁻ and CHO⁻, accounting for 59% and
218 33% of the total common formulas in terms of peak abundance, respectively.



219 As it is commonly known, ESI exhibits different ionization mechanisms in negative and positive
220 ionization modes. While ESI⁻ is especially sensitive to deprotonatable compounds (e.g., organic
221 acids), ESI⁺ is more sensitive to protonatable compounds (e.g., organic amines) (Ho et al., 2003).
222 Due to the different ionization mechanisms, clear differences were observed in the mass spectra
223 (Fig. 1) and chemical characteristics (Table 1) from ESI⁻ and ESI⁺ measurements. For example,
224 CHO compounds were preferentially detected in ESI⁻, accounting for a relatively large fraction of
225 30–42% of all detected compounds, compared to merely 4–13% for such CHO compounds in ESI⁺.
226 In contrast, CHN compounds were only observed in ESI⁺, yielding a rather large peak abundance
227 fraction of 40–71%. This observation indicates that most CHO compounds with high
228 concentrations are probably organic acids, whereas the majority of CHN compounds likely belong
229 to the group of organic amines, which is in good agreement with previous studies (Lin et al., 2012a;
230 Wang et al., 2017; Wang et al., 2018). Organic compounds in ESI⁺ are dominated by CHN⁺ and
231 CHON⁺ compounds in terms of both peak number and peak abundance and these compounds are
232 characterized by rather high H/C ratio and low O/C ratios (Table 1), indicating a low degree of
233 oxidation. The Venn diagram presented for ESI⁺ measurements in Fig. 2a shows that out of a total
234 of 383–679 formulas, 168 formulas were found in samples from all three cities. Such common
235 formulas, thus, account for 25–44% and 65–90% of all assigned formulas in terms of peak numbers
236 and peak abundance, respectively. Among these common formulas, CHN⁺ and CHON⁺ exhibit the
237 highest abundance fractions of 61% and 35%, respectively (Fig. 2b).

238 In the following, we will compare and discuss the chemical properties in detail for the three cities,
239 including degrees of oxidation, unsaturation and aromaticity of each organic compound class (i.e.,
240 CHO, CHON, CHN, CHOS and CHONS). It should be noted that the chlorine-containing
241 compounds were not discussed in this study due to the very low MS signal abundance.

242 **3.2 CHO compounds**

243 CHO compounds have been widely observed in urban OA, accounting for a substantial fraction
244 (8–67%) of OA (Rincón et al., 2012; Tao et al., 2014; Wang et al., 2017; Wang et al., 2018).
245 Previous studies have shown that a large fraction of CHO compounds in urban OA is composed of
246 organic acids, containing deprotonatable carboxyl functional groups, which are detected
247 preferentially in negative ionization mode when using ESI⁻MS. As shown in Table 1, a total of
248 346, 164, and 196 CHO⁻ compounds were detected in ESI⁻ in the OA samples collected in
249 Changchun, Shanghai and Guangzhou, accounting for 30%, 40% and 42% of the overall peak
250 abundance in each sample, respectively. Out of all detected compounds, 52 common CHO⁻



251 formulas were observed for all cities, accounting for 13–17% and 74–88% of all identified CHO–
252 formulas in terms of numbers and abundance, respectively.

253 Despite this similarity, OA samples from Changchun– (i.e. in negative ionization mode) exhibit
254 certain differences compared to samples from Shanghai– and Guangzhou–. The average H/C
255 values for CHO– compounds are in a similar range for the three locations (i.e., 0.96–1.10), however,
256 the average O/C values for O/C(Shanghai–) = 0.59 and O/C(Guangzhou–) = 0.65 are rather high
257 compared to the average O/C ratio for Changchun–, O/C(Changchun–) = 0.41. Furthermore, the
258 van Krevelen diagram in Fig. S1 also shows that more CHO– compounds with higher O/C ratios
259 were found in Shanghai– and Guangzhou– compared to Changchun–. Altogether, these results
260 indicate that CHO– compounds in urban OA from East and Southeast China experienced more
261 intense oxidation and aging processes.

262 Similarly, as shown in Fig. 3, the abundance-weighted average molecular formulas for CHO–
263 compounds in Changchun–, Shanghai– and Guangzhou– are $C_{8.58}H_{7.86}O_{3.22}$ ($MM_{avg}(\text{Changchun–})$
264 = 162), $C_{8.01}H_{7.27}O_{4.22}$ ($MM_{avg}(\text{Shanghai–})$ = 171) and $C_{7.70}H_{8.04}O_{4.48}$ ($MM_{avg}(\text{Guangzhou–})$ = 172),
265 respectively. Again, these average formulas show that CHO– compounds from OA in Shanghai–
266 and Guangzhou– experienced more intense oxidation processes, indicated by the larger abundance-
267 weighted MM_{avg} with a higher degree of oxygenation. In contrast, CHO– compounds from OA
268 samples in Changchun– exhibit a lower abundance-weighted MM_{avg} with a decreased oxygen
269 content.

270 Besides oxygenation, the aromaticity of the detected CHO– compounds exhibits remarkable
271 differences in these three cities. In all cities, the CHO– compounds with high peak abundance were
272 mainly assigned to monoaromatics with $2.5 \leq X_c < 2.7$ (purple circles in Fig. 3) in the region of
273 7–12 carbon atoms per compound and DBE values of 5–7. The fraction of monoaromatics in total
274 CHO– compounds is 67% in Changchun, which is higher compared to 64% in Shanghai and 49%
275 in Guangzhou. In addition, 14% of CHO– compounds in Changchun were identified as
276 polyaromatic compounds with $X_c \geq 2.7$ (red circles in Fig. 3), which are significantly higher than
277 8% in Shanghai and 4% in Guangzhou. These observations indicate that CHO– compounds in the
278 three Chinese cities are highly affected by aromatic precursors (e.g., benzene, toluene and
279 naphthalene), in particular for Changchun aerosol samples.

280 Besides the monoaromatics and polyaromatics, the rest of the detected CHO– compounds was
281 assigned to aliphatic compounds with an X_c lower than 2.5 (grey circles in Fig. 3). Interestingly,
282 these aliphatic compounds account for about 47% of CHO– compounds for Guangzhou– samples



283 in terms of peak abundance, whereas samples from Changchun– and Shanghai– exhibit only rather
284 small fractions of such CHO– compounds, i.e., 19% and 28%, respectively. Such aliphatic
285 compounds are commonly derived from biogenic precursors (Kourtchev et al., 2016) and vehicle
286 emission (Tao et al., 2014; Wang et al., 2017) and/or generated by intense oxidation processes of
287 aromatic precursors, indicating the different biogenic and anthropogenic emission sources and
288 chemical reaction processes for OAs in the three cities.

289 In addition, through mass spectrometric analysis of individual compounds, we find that for the
290 Changchun– samples, formulas of $C_8H_6O_4$, $C_7H_6O_2$, $C_7H_6O_3$, $C_8H_8O_2$, and $C_8H_8O_3$ with DBE values
291 of 6, 5, 5, 5, and 5 dominate the assigned CHO formulas with respect to peak abundance. According
292 to previous studies, $C_8H_6O_4$, $C_7H_6O_2$ and $C_7H_6O_3$ are suggested to be phthalic acid, benzoic acid
293 and monohydroxy benzoic acid, respectively, which are derived from naphthalene (Kautzman et
294 al., 2010; Riva et al., 2015; Wang et al., 2017; He et al., 2018; Huang et al., 2019). $C_8H_8O_2$ and
295 $C_8H_8O_3$ are likely 4-hydroxy acetophenone and 4-methoxybenzoic acid which could be derived
296 from estragole (Lee et al., 2006; Pereira et al., 2014). For the Shanghai– samples, besides $C_8H_6O_4$,
297 $C_7H_6O_3$ and $C_7H_6O_2$, formulas of $C_6H_8O_7$ and $C_9H_8O_4$ with DBE values of 3 and 6 were observed
298 with high peak abundances. $C_6H_8O_7$ was identified as citric acid in the pollen sample and mountain
299 particle sample in previous studies (Fu et al., 2008; Wang et al., 2009; Jung and Kawamura, 2011)
300 and $C_9H_8O_4$ are probably homophthalic acid derived from e.g. estragole (Pereira et al., 2014). For
301 the Guangzhou– samples, besides the formulas of $C_8H_6O_4$ and $C_6H_8O_7$ discussed above, $C_4H_6O_4$
302 and $C_4H_6O_5$ with low DBE values of two were detected with high abundances and are suggested to
303 be succinic acid and malic acid, respectively (Claeys et al., 2004; Wang et al., 2017).

304 3.3 CHON compounds

305 A large amount of nitrogen-containing organic compounds was detected in these three cities,
306 accounting for 39–55% and 25–47% of total peak abundance detected in ESI– and ESI+,
307 respectively. Out of all detected compounds, 51 common CHON– and 89 common CHON+
308 formulas were observed in all cities, accounting for 90–96% and 61–75% of all CHON compounds
309 detected in ESI– and ESI+ in terms of peak abundance, respectively. This large fraction of common
310 formulas indicates that the chemical composition of CHON compounds with higher signal
311 abundances, and thus, presumably high concentrations are quite similar in all three Chinese cities.

312 The CHON compounds were further classified into different subgroups according to their O/N
313 ratios. As shown in Fig. 4, the majority (84–96%) of CHON– compounds exhibit O/N ratios ≥ 3 ,
314 allowing the assignment of one nitro ($-NO_2$) or nitroxy ($-ONO_2$) group for these formulas, which



315 are preferentially ionized in ESI⁻ mode (Lin et al., 2012b; Wang et al., 2017; Song et al., 2018;
316 Wang et al., 2018). CHON⁻ formulas with O/N ratios ≥ 4 suggest the presence of further
317 oxygenated functional groups, such as a hydroxyl group (-OH) or a carbonyl group (C=O). 59%
318 of CHON⁻ compounds with O/N ratios ≥ 4 were found in in Guangzhou⁻, which is significantly
319 higher than 51% in Changchun⁻ and 45% in Shanghai⁻, indicating that CHON⁻ compounds in
320 Southeast China show a higher degree of oxidation compared to those in Northeast and East China.
321 Not surprisingly, CHON⁺ compounds generally exhibit lower O/N ratios (Fig. S2), as they
322 probably contain reduced nitrogen functional group (e.g., amines) which are preferably detected in
323 ESI⁺. As shown in Fig. S2, CHON⁺ compounds with O/N ratio of 1 are dominant in Changchun⁺,
324 whereas CHON⁺ compounds in Shanghai⁺ and Guangzhou⁺ show a broader range of O/N ratios
325 from 1 to 3. Moreover, the average O/C ratios (0.27–0.45) in Shanghai⁺ and Guangzhou⁺ (Table
326 1) are much greater than that (0.19) in Changchun⁺. Consistent with the observations for CHO
327 compounds, these results indicate again that CHON⁺ compounds in the OA of East and Southeast
328 China experienced more intensive photooxidation, which is probably due to the higher intensity
329 and longer duration of sunlight in lower latitude parts of China during winter.

330 Figure 5 shows the DBE versus C number of CHON⁻ compounds for the three cities. The majority
331 of CHON⁻ compounds lie in the region of 5–15 C atoms and 3–10 DBEs. 67% of CHON⁻
332 compounds were assigned to mono or polyaromatics in Shanghai⁻, which is significant higher than
333 52% in Guangzhou⁻ and 55% in Changchun⁻. Additionally, the average DBE value for Shanghai⁻,
334 $\text{DBE}_{\text{avg}}(\text{Shanghai}^-) = 5.67$, is larger than corresponding values for Changchun⁻ and Guangzhou⁻
335 with $\text{DBE}_{\text{avg}}(\text{Changchun}^-) = 5.24$ and $\text{DBE}_{\text{avg}}(\text{Guangzhou}^-) = 5.56$, respectively. These
336 observations indicate that CHON⁻ compounds are dominated with aromatic compounds in all cities,
337 while they have higher degree of aromaticity and unsaturation in Shanghai⁻ compared with those
338 in Changchun⁻ and Guangzhou⁻. The abundance-weighted average molecular formulas for
339 CHON⁻ compounds in Changchun⁻, Shanghai⁻ and Guangzhou⁻ are $\text{C}_{7.10}\text{H}_{6.76}\text{O}_{3.56}\text{N}_{1.03}$,
340 $\text{C}_{7.07}\text{H}_{6.03}\text{O}_{3.80}\text{N}_{1.24}$ and $\text{C}_{7.12}\text{H}_{6.36}\text{O}_{3.99}\text{N}_{1.24}$, respectively, showing that CHON⁻ formulas in
341 Shanghai⁻ and Guangzhou⁻ contain more O and N atoms on average than those for Changchun⁻.
342 Formulas of $\text{C}_6\text{H}_5\text{O}_3\text{N}_1$, $\text{C}_6\text{H}_5\text{O}_4\text{N}_1$, $\text{C}_7\text{H}_7\text{O}_3\text{N}_1$, $\text{C}_7\text{H}_7\text{O}_4\text{N}_1$, $\text{C}_8\text{H}_9\text{O}_3\text{N}_1$, and $\text{C}_8\text{H}_9\text{O}_4\text{N}_1$ were detected
343 with the highest abundance in all cities. These molecular formulas are in line with nitrophenol or
344 nitrocatechol analogs, which have been identified in a previous urban OA study (Wang et al., 2017).
345 Furthermore, these nitrooxy-aromatic compounds were shown to enhance light absorbing
346 properties of OA (Laskin et al., 2015; Lin et al., 2015). In addition, it should be noted that the Xc
347 values for $\text{C}_6\text{H}_5\text{O}_4\text{N}_1$, $\text{C}_7\text{H}_7\text{O}_4\text{N}_1$ and $\text{C}_8\text{H}_9\text{O}_4\text{N}_1$ were calculated to be lower than 2.5, suggesting



348 that the fraction of aromatics in CHON⁻ compounds was underestimated. This is because that for
349 nitrocatechol analogs with formulas of C₆H₅O₄N₁, C₇H₇O₄N₁ and C₈H₉O₄N₁, only one oxygen atom
350 is involved in the π-bond structure corresponding to the p value of 0.25 in the Xc calculation
351 equation, which is lower than the p value of 0.5 applied for the Xc calculation in this study. The
352 diagram of DBE versus C number for CHON⁺ compounds observed in the three locations
353 (presented in Fig. S3 in SI) shows that more aromatic CHON⁺ compounds with relatively lower
354 degree of oxidation were assigned in Changchun⁺ samples compared to Shanghai⁺ and
355 Guangzhou⁺ samples.

356 3.4 CHN⁺ compounds

357 205–696 CHN⁺ compounds were detected in ESI⁺, which are likely amines according to previous
358 studies (Rincón et al., 2012; Wang et al., 2017; Wang et al., 2018). The number of CHN⁺
359 compounds accounts for 24%, 36% and 30% of the total organic compounds in Changchun⁺,
360 Shanghai⁺ and Guangzhou⁺, respectively, whereas the peak abundance of these compounds
361 accounts for 40%, 71% and 62%, respectively. Comparing the CHN⁺ compounds for the three
362 cities, 58 common CHN⁺ formulas were observed in all cities, which contribute to as much as
363 83–98% of the total abundance of CHN⁺ formulas. This large percentage indicates that CHN⁺
364 compounds with presumably high concentrations in Changchun⁺, Shanghai⁺ and Guangzhou⁺
365 exhibit similar chemical composition. However, again OA samples from Changchun show some
366 distinct differences to samples from Guangzhou and Shanghai, giving the smallest ratio for number
367 and peak abundance of CHN⁺ compounds.

368 A van Krevelen diagram of CHN⁺ compounds detected in the three samples is shown in Fig. 6,
369 illustrating H/C ratios as a function of N/C ratio. In this plot, major parts of the CHN⁺ compounds
370 are found in a region, which is constraint by H/C ratios between 0.5 and 2 and N/C ratios lower
371 than 0.5. Moreover, the pie charts show that the majority (83–87%) of these CHN⁺ compounds can
372 be assigned to mono- and polyaromatics with Xc ≥ 2.5. In addition, as shown in Table 1, the
373 average DBE and Xc values of CHN⁺ compounds are the highest among all organic species. These
374 observations imply that CHN⁺ compounds exhibit the highest degree of aromaticity of all organics
375 in the Chinese urban OA samples, which is consistent with previous studies (Lin et al., 2012b;
376 Rincón et al., 2012; Wang et al., 2018). Polyaromatic compounds with Xc ≥ 2.7 are displayed in
377 the lower left corner of the van Krevelen diagram, accounting for 41% of CHN⁺ compounds
378 detected in Changchun⁺, but merely for 9–10% in Shanghai⁺ and Guangzhou⁺. For example,
379 formulas of C₁₁H₁₁N₁ (Xc = 2.7), C₁₀H₉N₁ (Xc = 2.7), and C₁₂H₁₃N₁ (Xc = 2.7), which are assigned



380 to be naphthalene core structure-containing compounds, have relatively higher abundance in
381 Changchun+ than in Shanghai+ and Guangzhou+. Moreover, the average DBE and Xc values of
382 CHN+ compounds in Changchun+ are significantly higher than those in Shanghai+ and
383 Guangzhou+, further indicating that CHN+ compounds in Changchun+ show a higher degree of
384 aromaticity, which can be caused by large coal combustion emissions in the winter in Changchun.
385 According to a previous smog chamber study (Laskin et al., 2010), most CHN+ aromatics are
386 probably generated from biomass burning through the addition of reduced nitrogen (e.g., NH₃) to
387 the organic molecules via imine formation reaction, indicating that biomass burning probably made
388 a certain contribution to the formation of CHN+ compounds observed in the three urban OA
389 samples in our study.

390 **3.5 CHOS⁻ compounds**

391 In this study, 75–155 CHOS⁻ compounds were observed, accounting for 10%, 12% and 14% of
392 the total peak abundance of all organics in Changchun⁻, Shanghai⁻ and Guangzhou⁻, respectively.
393 Around 89–96% of these CHOS⁻ compounds were found to fulfill the O/S ≥ 4 criterion allowing
394 the assignment of at least one –OSO₃H functional group, and thus, a tentative classification to
395 organosulfates (OSs) (Lin et al., 2012a; Lin et al., 2012b; Tao et al., 2014; Wang et al., 2016; Wang
396 et al., 2017; Wang et al., 2018; Wang et al., 2019a). OSs were shown to affect the surface activity
397 and hygroscopic properties of the aerosol particles, leading to potential impacts on climate (Hansen
398 et al., 2015; Wang et al., 2019a). Out of all formulas, 28 common CHOS⁻ formulas were detected
399 for the three sample locations, accounting for 39%, 68% and 65% of the CHOS⁻ peak abundance
400 in Changchun⁻, Shanghai⁻ and Guangzhou⁻, respectively. However, 40 common CHOS⁻
401 formulas were found between Shanghai⁻ and Guangzhou⁻, accounting for 60–65% and 70–83%
402 in terms of the CHOS⁻ compounds number and peak abundance, respectively. This indicates that
403 the chemical composition of the major CHOS⁻ compounds of Shanghai⁻ and Guangzhou⁻ are
404 quite similar, while they show significant chemical differences for samples from Changchun⁻.

405 Figure 7 shows the DBEs as a function of carbon number for all CHOS⁻ compounds detected for
406 the three cities. The CHOS⁻ compounds exhibit a DBE range from 0 to 10 and carbon number
407 range of 2–15. However, the majority of CHOS⁻ compounds with elevated peak abundances
408 concentrate in a region with rather low DBE values of 0–5. The average H/C ratios of CHOS⁻
409 compounds are in the range of 1.56–1.85, and thus, higher than for any other compound class,
410 whereas the average DBE values of 1.71–2.55 are the lowest among all classes. This indicates that
411 CHOS⁻ compounds in the OA from the three Chinese cities are characterized by a low degree of



412 unsaturation. Moreover, the pie charts in Fig. 7 show that aliphatic compounds with $X_c \leq 2.5$ are
413 dominant in CHOS⁻ compounds with a fraction of 96–99%, which is significantly higher than that
414 (13–48%) for CHO, CHON and CHN species. Aliphatic CHOS⁻ compounds with $C \leq 10$ can be
415 formed from biogenic and/or anthropogenic precursors (Hansen et al., 2014; Glasius et al., 2018;
416 Wang et al., 2019a), such as C₂H₄O₆S₁ (derived from glyoxal) (Lim et al., 2010; McNeill et al.,
417 2012), C₃H₆O₆S₁ (derived from isoprene) (Surratt et al., 2007) and C₈H₁₆O₄S₁ (derived from α -
418 pinene) (Surratt et al., 2007). However, more CHOS⁻ compounds with $C > 10$ and with DBEs
419 lower than 1 are observed in Changchun⁻, such as C₁₄H₂₈O₅S₁, C₁₃H₂₆O₅S₁, C₁₂H₂₄O₅S₁,
420 C₁₁H₂₂O₅S₁ and C₁₁H₂₀O₆S₁. These high-carbon-number-containing CHOS⁻ compounds are likely
421 formed from long-alkyl-chain compounds with less oxygenated functional groups, which were
422 previously suggested to be emitted from traffic (Tao et al., 2014) or derived from sesquiterpene
423 emissions (Brüggemann et al., 2019). However, as sesquiterpene emissions can be expected to be
424 very low in wintertime at Changchun, the presence of these compounds further underlines the
425 strong impact of anthropogenic emissions on CHOS⁻ formation in Changchun⁻. In this study,
426 (O–3S)/C ratio was used instead of traditional O/C ratio to present the oxidation state of CHOS⁻
427 compounds, since the sulfate functional group contains three more oxygen atoms than common
428 oxygen-containing groups (e.g., hydroxyl and carbonyl), which makes no contribution to the
429 oxidation state of the carbon backbone of the CHOS⁻ compounds. Comparing average values for
430 H/C, (O–3S)/C and DBEs of CHOS⁻ for the three sample locations (see Table 1), we find that the
431 H/C ratios (1.85) and (O–3S)/C ratios (0.61–0.71) for Shanghai⁻ and Guangzhou⁻ samples are
432 larger than those for Changchun⁻ samples (H/C = 1.56 and (O–3S)/C = 0.52), whereas the DBE
433 values (1.71–1.79) in Shanghai⁻ and Guangzhou⁻ are lower than those for Changchun⁻ (2.55).
434 These observations indicate that CHOS⁻ compounds in urban OA from Northeast China are less
435 oxidized but more unsaturated compared to those in East and Southeast China, likely due to
436 enhanced emissions from residential heating during winter in North China.

437 3.6 CHONS compounds

438 4–5% of the total organics detected in ESI⁻ were identified as CHONS⁻ compounds. In contrast,
439 CHONS⁺ compounds account merely for 0.3–1% of all organics detected in ESI⁺. The average
440 MM_{avg} of the CHONS⁻ compounds for the three sample locations ranges from 214 to 293 Da,
441 generally showing larger molecular masses than compounds of any other class because of the likely
442 presence of both nitrate and sulfate functional groups. In total, only 8 common CHONS⁻ formulas
443 were detected for all three sample locations, accounting for 8%, 58% and 56% of the CHONS⁻
444 peak abundance in Changchun⁻, Shanghai⁻ and Guangzhou⁻, respectively. As already observed



445 for other compound classes, these percentages imply that the CHONS⁻ compounds in urban OA
446 of Shanghai⁻ and Guangzhou⁻ exhibit a rather similar chemical composition, whereas such
447 compounds are significantly different for Changchun⁻.

448 In the OA samples of Shanghai⁻ and Guangzhou⁻, 78–87% of CHONS⁻ formulas have 7 or more
449 O atoms, allowing the assignment of one –OSO₃H and one –NO₃ functional groups in the molecular
450 structures, thus, classifying them as potential nitrooxy-organosulfates. In contrast to Shanghai⁻ and
451 Guangzhou⁻, only 26% of CHONS⁻ compounds were assigned to such nitrooxy-organosulfates
452 for Changchun⁻, indicating that most of the N atoms in the CHONS⁻ compounds are present in a
453 reduced oxidation state, e.g., in the form of amines. The average DBE and Xc values of CHONS⁻
454 compounds in Shanghai⁻ and Guangzhou⁻ are 3.3–3.45 and 0.43–0.44, respectively. Again these
455 values differ significantly for the Changchun⁻ samples with an increased average DBE of 3.75 and
456 an average Xc of 1.06, indicating that CHONS⁻ compounds in Changchun⁻ possess on average a
457 higher degree of unsaturation and aromaticity compared to such compounds in Shanghai⁻ and
458 Guangzhou⁻ samples. Interestingly, the compound with formula C₁₀H₁₇O₇NS has the highest
459 relative peak abundance (32%) in Shanghai⁻ and Guangzhou⁻, whereas in Changchun⁻ the
460 compound with formula C₂H₃O₄NS is dominant. C₁₀H₁₇O₇NS has previously been identified as
461 pinanediol mononitrate generated from α/β-pinene (Iinuma et al., 2007; Surratt et al., 2008; Lin et
462 al., 2012b; Wang et al., 2017), while C₂H₃O₄NS may be assigned as a cyanogroup-containing
463 sulfate. This observation is comparable to our previous study (Wang et al., 2019a), which found
464 that C₁₀H₁₇O₇NS was dominant for CHONS⁻ compounds in low-concentration aerosol samples
465 collected in Beijing (China) and Mainz (Germany). Consistently, a C₂H₃O₄NS compound had the
466 highest abundance among CHONS⁻ compounds in polluted Beijing aerosol samples. This
467 agreement can be explained by the adjacent locations of Beijing (39.99° N, 116.39° E) and
468 Changchun (43.54° N, 125.13° E) and similar residential heating patterns by coal combustion during
469 wintertime. In conclusion, these results further demonstrate that the precursors for CHONS⁻
470 compounds in Shanghai⁻ and Guangzhou⁻ are different from those in Changchun⁻, which is
471 probably due to differences in anthropogenic emissions.

472 **4 Conclusion**

473 The molecular composition of the organic fraction of PM_{2.5} samples collected in three Chinese
474 megacities (Changchun, Shanghai and Guangzhou) was investigated using a UHPLC-Orbitrap
475 mass spectrometer. In total, 416–769 (ESI⁻) and 687–2943 (ESI⁺) organic compounds were
476 observed and separated into five subgroups: CHO, CHN, CHON, CHOS and CHONS. Specifically,



477 139 common formulas were detected in ESI⁻ and 168 common formulas in ESI⁺ for all sample
478 locations, accounting for 78–87% and 65–90% in terms of peak abundance, respectively. Overall,
479 we found that urban OA in Changchun, Shanghai and Guangzhou shows a quite similar chemical
480 composition for organic compounds of high concentrations. The majority of these organic species
481 was assigned to mono-aromatic or poly-aromatic compounds, indicating that anthropogenic
482 emissions are the major source for urban OA in all three cities.

483 Despite the chemical similarity of the three sample locations for major organic compounds in urban
484 OA, remarkable differences were found in chemical composition of the remaining particle
485 constituents, in particular for OA samples from Changchun. In general, a larger amount of
486 polyaromatics was observed for Changchun samples, most likely due to emissions from coal
487 combustion during wintertime residential heating period. Moreover, the abundance-weighted
488 average DBE and average X_c values of the total organic compounds in Changchun were found to
489 be larger than those for Shanghai and Guangzhou, showing that organic compounds in Changchun
490 possess a higher degree of unsaturation and aromaticity. For average H/C and O/C ratios a similar
491 trend was observed. While average H/C and O/C ratios detected in ESI⁻ were found to be highest
492 for Guangzhou samples, significantly lower values were observed for Shanghai and Changchun
493 samples, indicating that OA collected in lower latitude regions of China experiences more intense
494 photochemical oxidation processes.

495 **Author contributions.** RJH, TH and KW conducted the study design. LY, HN, JG and MW
496 collected the PM_{2.5} filter samples. KW and YZ carried out the experimental work and data analysis.
497 KW wrote the manuscript. KW, TH, RJH, M. Brüggemann, YZ, JH, M. Bilde and MG interpreted
498 data and edited the manuscript. All authors commented on and discussed the manuscript.

499 **Competing interests.** The authors declare that they have no conflict of interest.

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509 **References**

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511 An, Z., Huang, R. J., Zhang, R., Tie, X., Li, G., Cao, J., Zhou, W., Shi, Z., Han, Y., Gu, Z., and Ji, Y.: Severe
512 haze in northern China: A synergy of anthropogenic emissions and atmospheric processes, *Proc Natl Acad*
513 *Sci U S A*, 116, 8657-8666, 10.1073/pnas.1900125116, 2019.

514 Brüggemann, M., Poulain, L., Held, A., Stelzer, T., Zuth, C., Richters, S., Mutzel, A., van Pinxteren, D.,
515 Iinuma, Y., Katkevica, S., Rabe, R., Herrmann, H., and Hoffmann, T.: Real-time detection of highly
516 oxidized organosulfates and BSOA marker compounds during the F-BEACH 2014 field study, *Atmos.*
517 *Chem. Phys.*, 17, 1453-1469, 10.5194/acp-17-1453-2017, 2017.

518 Brüggemann, M., van Pinxteren, D., Wang, Y., Yu, J. Z., and Herrmann, H.: Quantification of known and
519 unknown terpenoid organosulfates in PM10 using untargeted LC-HRMS/MS: contrasting summertime
520 rural Germany and the North China Plain, *Environmental Chemistry*, -, <https://doi.org/10.1071/EN19089>,
521 2019.

522 Claeys, M., Graham, B., Vas, G., Wang, W., Vermeylen, R., Pashynska, V., Cafmeyer, J., Guyon, P., Andre,
523 M., Artaxo, P., and Maenhaut, W.: Formation of secondary organic aerosol through photooxidation of
524 isoprene, *Science*, 303, 1173-1175, 10.1126/science.1092805, 2004.

525 Daellenbach, K. R., Kourchev, I., Vogel, A. L., Bruns, E. A., Jiang, J., Petäjä, T., Jaffrezo, J.-L., Aksoyoglu,
526 S., Kalberer, M., Baltensperger, U., El Haddad, I., and Prévôt, A. S. H.: Impact of anthropogenic and
527 biogenic sources on the seasonal variation in the molecular composition of urban organic aerosols: a field
528 and laboratory study using ultra-high-resolution mass spectrometry, *Atmospheric Chemistry and Physics*,
529 19, 5973-5991, 10.5194/acp-19-5973-2019, 2019.

530 Ding, X., Zhang, Y.-Q., He, Q.-F., Yu, Q.-Q., Shen, R.-Q., Zhang, Y., Zhang, Z., Lyu, S.-J., Hu, Q.-H., Wang,
531 Y.-S., Li, L.-F., Song, W., and Wang, X.-M.: Spatial and seasonal variations of secondary organic aerosol
532 from terpenoids over China, *J. geophys. Res.-Atoms.*, 121, 14661-14678, doi:10.1002/2016JD025467,
533 2016.

534 Elzein, A., Dunmore, R. E., Ward, M. W., Hamilton, J. F., and Lewis, A. C.: Variability of polycyclic
535 aromatic hydrocarbons and their oxidative derivatives in wintertime Beijing, China, *Atmospheric*
536 *Chemistry and Physics*, 19, 8741-8758, 10.5194/acp-19-8741-2019, 2019.

537 Fleming, L. T., Lin, P., Laskin, A., Laskin, J., Weltman, R., Edwards, R. D., Arora, N. K., Yadav, A.,
538 Meinardi, S., Blake, D. R., Pillarisetti, A., Smith, K. R., and Nizkorodov, S. A.: Molecular composition
539 of particulate matter emissions from dung and brushwood burning household cookstoves in Haryana,
540 India, *Atmos. Chem. Phys.*, 18, 2461-2480, 10.5194/acp-18-2461-2018, 2018.

541 Fu, P., Kawamura, K., Okuzawa, K., Aggarwal, S. G., Wang, G., Kanaya, Y., and Wang, Z.: Organic
542 molecular compositions and temporal variations of summertime mountain aerosols over Mt. Tai, North
543 China Plain, *J. Geophys. Res.*, 113, 10.1029/2008jd009900, 2008.

544 Glasius, M., Hansen, A. M. K., Claeys, M., Henzing, J. S., Jedynska, A. D., Kasper-Giebl, A., Kistler, M.,
545 Kristensen, K., Martinsson, J., Maenhaut, W., Nøjgaard, J. K., Spindler, G., Stenström, K. E., Swietlicki,
546 E., Szidat, S., Simpson, D., and Yttri, K. E.: Composition and sources of carbonaceous aerosols in
547 Northern Europe during winter, *Atmos. Environ.*, 173, 127-141, 10.1016/j.atmosenv.2017.11.005, 2018.

548 Hansen, A. M. K., Kristensen, K., Nguyen, Q. T., Zare, A., Cozzi, F., Noejgaard, J. K., Skov, H., Brandt, J.,
549 Christensen, J. H., Strom, J., Tunved, P., Krejci, R., and Glasius, M.: Organosulfates and organic acids in
550 Arctic aerosols: speciation, annual variation and concentration levels, *Atmos. Chem. Phys.*, 14, 7807-
551 7823, <https://doi.org/10.5194/acp-14-7807-2014>, 2014.

552 Hansen, A. M. K., Hong, J., Raatikainen, T., Kristensen, K., Ylisirniö, A., Virtanen, A., Petäjä, T., Glasius,
553 M., and Prisle, N. L.: Hygroscopic properties and cloud condensation nuclei activation of limonene-
554 derived organosulfates and their mixtures with ammonium sulfate, *Atmos. Chem. Phys.*, 15, 14071-14089,
555 <https://doi.org/10.5194/acp-15-14071-2015>, 2015.

556 He, X., Huang, X. H. H., Chow, K. S., Wang, Q., Zhang, T., Wu, D., and Yu, J. Z.: Abundance and Sources
557 of Phthalic Acids, Benzene-Tricarboxylic Acids, and Phenolic Acids in PM2.5 at Urban and Suburban
558 Sites in Southern China, *ACS Earth and Space Chemistry*, 2, 147-158,
559 10.1021/acsearthspacechem.7b00131, 2018.

560 Ho, C. S., Lam, C. W. K., Chan, M. H. M., Cheung, R. C. K., Law, L. K., Suen, M. W. M., and Tai, H. L.:
561 Electrospray ionisation mass spectrometry: principles and clinical application, *Clin. Biochem. Rev.*, 24,
562 10, 2003.

563 Hoffmann, T., Huang, R. J., and Kalberer, M.: Atmospheric analytical chemistry, *Anal. Chem.*, 83, 4649-



- 564 4664, 10.1021/ac2010718, 2011.
- 565 Huang, G., Liu, Y., Shao, M., Li, Y., Chen, Q., Zheng, Y., Wu, Z., Liu, Y., Wu, Y., Hu, M., Li, X., Lu, S.,
566 Wang, C., Liu, J., Zheng, M., and Zhu, T.: Potentially Important Contribution of Gas-Phase Oxidation of
567 Naphthalene and Methylanthalene to Secondary Organic Aerosol during Haze Events in Beijing,
568 *Environ Sci Technol*, 53, 1235-1244, 10.1021/acs.est.8b04523, 2019.
- 569 Huang, R. J., Zhang, Y., Bozzetti, C., Ho, K. F., Cao, J. J., Han, Y., Daellenbach, K. R., Slowik, J. G., Platt,
570 S. M., Canonaco, F., Zotter, P., Wolf, R., Pieber, S. M., Bruns, E. A., Crippa, M., Ciarelli, G., Piazzalunga,
571 A., Schwikowski, M., Abbaszade, G., Schnelle-Kreis, J., Zimmermann, R., An, Z., Szidat, S.,
572 Baltensperger, U., El Haddad, I., and Prevot, A. S.: High secondary aerosol contribution to particulate
573 pollution during haze events in China, *Nature*, 514, 218-222, 10.1038/nature13774, 2014.
- 574 Huang, R. J., Cao, J. J., and Worsnop, D.: Sources and Chemical Composition of Particulate Matter During
575 Haze Pollution Events in China, in: *Air pollution in Eastern Asia: an integrated perspective*, edited by
576 Bouarar, I., Wang, X. M., and Brasseur, G. P., Springer, Cham, Switzerland, 49-68, 2017.
- 577 Inuma, Y., Müller, C., Berndt, T., Böge, O., Claeys, M., and Herrmann, H.: Evidence for the existence of
578 organosulfates from β -pinene ozonolysis in ambient secondary organic aerosol, *Environ. Sci. Technol.*,
579 41, 6678-6683, 10.1021/es070938t, 2007.
- 580 Jung, J., and Kawamura, K.: Enhanced concentrations of citric acid in spring aerosols collected at the Gosan
581 background site in East Asia, *Atmos. Environ.*, 45, 5266-5272, 10.1016/j.atmosenv.2011.06.065, 2011.
- 582 Kautzman, K. E., Surratt, J. D., Chan, M. N., Chan, A. W., Hersey, S. P., Chhabra, P. S., Dalleska, N. F.,
583 Wennberg, P. O., Flagan, R. C., and Seinfeld, J. H.: Chemical composition of gas- and aerosol-phase
584 products from photooxidation of naphthalene, *J. Phys. Chem. A*, 114, 913-934, 10.1021/jp908530s, 2010.
- 585 Kind, T., and Fiehn, O.: Seven Golden Rules for heuristic filtering of molecular formulas obtained by
586 accurate mass spectrometry, *BMC Bioinformatics*, 8, 10.1186/1471-2105-8-105, 2007.
- 587 Kourtchev, I., O'Connor, I. P., Giorio, C., Fuller, S. J., Kristensen, K., Maenhaut, W., Wenger, J. C., Sodeau,
588 J. R., Glasius, M., and Kalberer, M.: Effects of anthropogenic emissions on the molecular composition of
589 urban organic aerosols: An ultrahigh resolution mass spectrometry study, *Atmo. Environ.*, 89, 525-532,
590 10.1016/j.atmosenv.2014.02.051, 2014.
- 591 Kourtchev, I., Godoi, R. H. M., Connors, S., Levine, J. G., Archibald, A. T., Godoi, A. F. L., Paralovo, S. L.,
592 Barbosa, C. G. G., Souza, R. A. F., Manzi, A. O., Seco, R., Sjøstedt, S., Park, J.-H., Guenther, A., Kim,
593 S., Smith, J., Martin, S. T., and Kalberer, M.: Molecular composition of organic aerosols in central
594 Amazonia: an ultra-high-resolution mass spectrometry study, *Atmos. Chem. Phys.*, 16, 11899-11913,
595 <https://doi.org/10.5194/acp-16-11899-2016>, 2016.
- 596 Laskin, A., Laskin, J., and Nizkorodov, S. A.: Chemistry of atmospheric brown carbon, *Chem. Rev.*, 115,
597 4335-4382, 10.1021/cr5006167, 2015.
- 598 Laskin, J., Laskin, A., Roach, P. J., Slysz, G. W., Anderson, G. A., Nizkorodov, S. A., Bones, D. L., and
599 Nguyen, L. Q.: High-Resolution Desorption Electrospray Ionization Mass Spectrometry for Chemical
600 Characterization of Organic Aerosols, *Anal. Chem.*, 82, 2048-2058, 10.1021/ac902801f, 2010.
- 601 Laskin, J., Laskin, A., and Nizkorodov, S. A.: Mass Spectrometry Analysis in Atmospheric Chemistry, *Anal.*
602 *Chem.*, 90, 166-189, 10.1021/acs.analchem.7b04249, 2018.
- 603 Lee, A., Goldstein, A. H., Kroll, J. H., Ng, N. L., Varutbangkul, V., Flagan, R. C., and Seinfeld, J. H.: Gas-
604 phase products and secondary aerosol yields from the photooxidation of 16 different terpenes, *J. Geophys.*
605 *Res.*, 111, 10.1029/2006jd007050, 2006.
- 606 Lim, Y. B., Tan, Y., Perri, M. J., Seitzinger, S. P., and Turpin, B. J.: Aqueous chemistry and its role in
607 secondary organic aerosol (SOA) formation, *Atmos. Chem. Phys.*, 10, 10521-10539, 10.5194/acp-10-
608 10521-2010, 2010.
- 609 Lin, P., Rincon, A. G., Kalberer, M., and Yu, J. Z.: Elemental composition of HULIS in the Pearl River Delta
610 Region, China: results inferred from positive and negative electrospray high resolution mass
611 spectrometric data, *Environ. Sci. Technol.*, 46, 7454-7462, 10.1021/es300285d, 2012a.
- 612 Lin, P., Yu, J. Z., Engling, G., and Kalberer, M.: Organosulfates in humic-like substance fraction isolated
613 from aerosols at seven locations in East Asia: a study by ultra-high-resolution mass spectrometry, *Environ.*
614 *Sci. Technol.*, 46, 13118-13127, 10.1021/es303570v, 2012b.
- 615 Lin, P., Laskin, J., Nizkorodov, S. A., and Laskin, A.: Revealing Brown Carbon Chromophores Produced in
616 Reactions of Methylglyoxal with Ammonium Sulfate, *Environ. Sci. Technol.*, 49, 14257-14266,
617 10.1021/acs.est.5b03608, 2015.
- 618 McNeill, V. F., Woo, J. L., Kim, D. D., Schwier, A. N., Wannell, N. J., Sumner, A. J., and Barakat, J. M.:



- 619 Aqueous-phase secondary organic aerosol and organosulfate formation in atmospheric aerosols: a
620 modeling study, *Environ Sci Technol*, 46, 8075-8081, 10.1021/es3002986, 2012.
- 621 Ning, C., Gao, Y., Zhang, H., Yu, H., Wang, L., Geng, N., Cao, R., and Chen, J.: Molecular characterization
622 of dissolved organic matters in winter atmospheric fine particulate matters (PM_{2.5}) from a coastal city of
623 northeast China, *Sci Total Environ*, 689, 312-321, 10.1016/j.scitotenv.2019.06.418, 2019.
- 624 Nizkorodov, S. A., Laskin, J., and Laskin, A.: Molecular chemistry of organic aerosols through the
625 application of high resolution mass spectrometry, *Phys. Chem. Chem. Phys.*, 13, 3612-3629,
626 10.1039/c0cp02032j, 2011.
- 627 Noziere, B., Kalberer, M., Claeys, M., Allan, J., D'Anna, B., Decesari, S., Finessi, E., Glasius, M., Grgic, I.,
628 Hamilton, J. F., Hoffmann, T., Iinuma, Y., Jaoui, M., Kahnt, A., Kampf, C. J., Kourtchev, I., Maenhaut,
629 W., Marsden, N., Saarikoski, S., Schnelle-Kreis, J., Surratt, J. D., Szidat, S., Szmigielski, R., and
630 Wisthaler, A.: The molecular identification of organic compounds in the atmosphere: state of the art and
631 challenges, *Chem. Rev.*, 115, 3919-3983, 10.1021/cr5003485, 2015.
- 632 Pereira, K. L., Hamilton, J. F., Rickard, A. R., Bloss, W. J., Alam, M. S., Camredon, M., Muñoz, A., Vázquez,
633 M., Borrás, E., and Ródenas, M.: Secondary organic aerosol formation and composition from the photo-
634 oxidation of methyl chavicol (estragole), *Atmos. Chem. Phys.*, 14, 5349-5368, 10.5194/acp-14-5349-
635 2014, 2014.
- 636 Perry, R. H., Cooks, R. G., and Noll, R. J.: ORBITRAP MASS SPECTROMETRY: INSTRUMENTATION,
637 ION MOTION AND APPLICATIONS, *Mass Spectrometry Reviews*, 27, 661-699, 10.1002/mas.20186,
638 2008.
- 639 Rincón, A. G., Calvo, A. I., Dietzel, M., and Kalberer, M.: Seasonal differences of urban organic aerosol
640 composition - an ultra-high resolution mass spectrometry study, *Environ. Chem.*, 9, 298,
641 10.1071/en12016, 2012.
- 642 Riva, M., Tomaz, S., Cui, T., Lin, Y.-H., Perraudin, E., Gold, A., Stone, E. A., Villenave, E., and Surratt, J.
643 D.: Evidence for an Unrecognized Secondary Anthropogenic Source of Organosulfates and Sulfonates:
644 Gas-Phase Oxidation of Polycyclic Aromatic Hydrocarbons in the Presence of Sulfate Aerosol, *Environ.*
645 *Sci. Technol.*, 49, 6654-6664, 10.1021/acs.est.5b00836, 2015.
- 646 Shi, Z., Vu, T., Kotthaus, S., Harrison, R. M., Grimmond, S., Yue, S., Zhu, T., Lee, J., Han, Y., Demuzere,
647 M., Dunmore, R. E., Ren, L., Liu, D., Wang, Y., Wild, O., Allan, J., Acton, W. J., Barlow, J., Barratt, B.,
648 Beddows, D., Bloss, W. J., Calzolari, G., Carruthers, D., Carslaw, D. C., Chan, Q., Chatzidiakou, L., Chen,
649 Y., Crilley, L., Coe, H., Dai, T., Doherty, R., Duan, F., Fu, P., Ge, B., Ge, M., Guan, D., Hamilton, J. F.,
650 He, K., Heal, M., Heard, D., Hewitt, C. N., Hollaway, M., Hu, M., Ji, D., Jiang, X., Jones, R., Kalberer,
651 M., Kelly, F. J., Kramer, L., Langford, B., Lin, C., Lewis, A. C., Li, J., Li, W., Liu, H., Liu, J., Loh, M.,
652 Lu, K., Lucarelli, F., Mann, G., McFiggans, G., Miller, M. R., Mills, G., Monk, P., Nemitz, E., amp, apos,
653 Connor, F., Ouyang, B., Palmer, P. I., Percival, C., Popoola, O., Reeves, C., Rickard, A. R., Shao, L., Shi,
654 G., Spracklen, D., Stevenson, D., Sun, Y., Sun, Z., Tao, S., Tong, S., Wang, Q., Wang, W., Wang, X.,
655 Wang, X., Wang, Z., Wei, L., Whalley, L., Wu, X., Wu, Z., Xie, P., Yang, F., Zhang, Q., Zhang, Y.,
656 Zhang, Y., and Zheng, M.: Introduction to the special issue "In-depth study of air pollution sources and
657 processes within Beijing and its surrounding region (APHH-Beijing)", *Atmospheric Chemistry and*
658 *Physics*, 19, 7519-7546, 10.5194/acp-19-7519-2019, 2019.
- 659 Song, J., Li, M., Jiang, B., Wei, S., Fan, X., and Peng, P.: Molecular Characterization of Water-Soluble
660 Humic like Substances in Smoke Particles Emitted from Combustion of Biomass Materials and Coal
661 Using Ultrahigh-Resolution Electrospray Ionization Fourier Transform Ion Cyclotron Resonance Mass
662 Spectrometry, *Environ. Sci. Technol.*, 52, 2575-2585, 10.1021/acs.est.7b06126, 2018.
- 663 Sun, Y., Jiang, Q., Zhang, Z., Fu, P., Li, J., Yang, T., and Yin, Y.: Investigation of the sources and evolution
664 processes of severe haze pollution in Beijing in January 2013, *J. Geophys. Res.-Atmos.*, 119, 4380-4389,
665 10.1002/, 2014.
- 666 Surratt, J. D., Gomez-Gonzalez, Y., Chan, A. W., Vermeylen, R., Shahgholl, M., Kleindienst, T. E., Jaoui,
667 M., Maenhaut, W., Claeys, M., Flagan, R. C., and Seinfeld, J. H.: Evidence for Organosulfate in
668 Secondary Organic Aerosol, *Environ. Sci. Technol.*, 41, 517-527, 10.1021/es062081q, 2007.
- 669 Surratt, J. D., Gómez-González, Y., Chan, A. W., Vermeylen, R., Shahgholi, M., Kleindienst, T. E., Edney,
670 E. O., Offenberg, J. H., Lewandowski, M., Jaoui, M., Maenhaut, W., Claeys, M., Flagan, R. C., and
671 Seinfeld, J. H.: Organosulfate Formation in Biogenic Secondary Organic Aerosol, *J. Phys. Chem. A*, 112,
672 8345-8378, 2008.
- 673 Tao, S., Lu, X., Levac, N., Bateman, A. P., Nguyen, T. B., Bones, D. L., Nizkorodov, S. A., Laskin, J., Laskin,



- 674 A., and Yang, X.: Molecular Characterization of Organosulfates in Organic Aerosols from Shanghai and
675 Los Angeles Urban Areas by Nanospray-Desorption Electrospray Ionization High-Resolution Mass
676 Spectrometry, *Environ. Sci. Technol.*, 48, 10993-11001, 10.1021/es5024674, 2014.
- 677 Tong, H., Kourtchev, I., Pant, P., Keyte, I. J., O'Connor, I. P., Wenger, J. C., Pope, F. D., Harrison, R. M.,
678 and Kalberer, M.: Molecular composition of organic aerosols at urban background and road tunnel sites
679 using ultra-high resolution mass spectrometry, *Faraday Discuss.*, 189, 51-68, 10.1039/c5fd00206k, 2016.
- 680 Tu, P., Hall, W. A. t., and Johnston, M. V.: Characterization of Highly Oxidized Molecules in Fresh and
681 Aged Biogenic Secondary Organic Aerosol, *Anal. Chem.*, 88, 4495-4501,
682 10.1021/acs.analchem.6b00378, 2016.
- 683 Wang, G., Kawamura, K., Umemoto, N., Xie, M., Hu, S., and Wang, Z.: Water-soluble organic compounds
684 in PM_{2.5} and size-segregated aerosols over Mount Tai in North China Plain, *J. Geophys. Res.*, 114,
685 10.1029/2008jd011390, 2009.
- 686 Wang, K., Zhang, Y., Huang, R.-J., Cao, J., and Hoffmann, T.: UHPLC-Orbitrap mass spectrometric
687 characterization of organic aerosol from a central European city (Mainz, Germany) and a Chinese
688 megacity (Beijing), *Atmos. Environ.*, 189, 22-29, 10.1016/j.atmosenv.2018.06.036, 2018.
- 689 Wang, K., Zhang, Y., Huang, R.-J., Wang, M., Ni, H., Kampf, C. J., Cheng, Y., Bilde, M., Glasius, M., and
690 Hoffmann, T.: Molecular characterization and source identification of atmospheric particulate
691 organosulfates using ultrahigh resolution mass spectrometry, *Environ. Sci. Technol.*,
692 10.1021/acs.est.9b02628, 2019a.
- 693 Wang, M., Huang, R.-J., Cao, J., Dai, W., Zhou, J., Lin, C., Ni, H., Duan, J., Wang, T., Chen, Y., Li, Y.,
694 Chen, Q., Haddad, I. E., and Hoffmann, T.: Determination of n-alkanes, PAHs and hopanes in
695 atmospheric aerosol: evaluation and comparison of thermal desorption GC-MS and solvent extraction
696 GC-MS approaches, *Atmos. Meas. Tech. Discuss.*, 1-21, 10.5194/amt-2019-4, 2019b.
- 697 Wang, X. K., Rossignol, S., Ma, Y., Yao, L., Wang, M. Y., Chen, J. M., George, C., and Wang, L.: Molecular
698 characterization of atmospheric particulate organosulfates in three megacities at the middle and lower
699 reaches of the Yangtze River, *Atmos. Chem. Phys.*, 16, 2285-2298, [https://doi.org/10.5194/acp-16-2285-](https://doi.org/10.5194/acp-16-2285-2016)
700 [2016](https://doi.org/10.5194/acp-16-2285-2016), 2016.
- 701 Wang, X. K., Hayeck, N., Brüggemann, M., Yao, L., Chen, H. F., Zhang, C., Emmelin, C., Chen, J. M.,
702 George, C., and Wang, L.: Chemical characterization of organic aerosol in: A study by Ultrahigh-
703 Performance Liquid Chromatography Coupled with Orbitrap Mass Spectrometry, *J. Geophys. Res.-Atmos.*,
704 122, 703-722, <https://doi.org/10.1002/2017JD026930>, 2017.
- 705 Xu, W., Sun, Y., Wang, Q., Zhao, J., Wang, J., Ge, X., Xie, C., Zhou, W., Du, W., Li, J., Fu, P., Wang, Z.,
706 Worsnop, D. R., and Coe, H.: Changes in Aerosol Chemistry From 2014 to 2016 in Winter in Beijing:
707 Insights From High-Resolution Aerosol Mass Spectrometry, *J. Geophys. Res.-Atmos.*, 124, 1132-1147,
708 10.1029/2018jd029245, 2019.
- 709 Yassine, M. M., Harir, M., Dabek-Zlotorzynska, E., and Schmitt-Kopplin, P.: Structural characterization of
710 organic aerosol using Fourier transform ion cyclotron resonance mass spectrometry: aromaticity
711 equivalent approach, *Rapid Commun. Mass Spectrom.*, 28, 2445-2454, 10.1002/rcm.7038, 2014.
- 712 Zielinski, A. T., Kourtchev, I., Bortolini, C., Fuller, S. J., Giorio, C., Popoola, O. A. M., Bogialli, S., Tapparo,
713 A., Jones, R. L., and Kalberer, M.: A new processing scheme for ultra-high resolution direct infusion
714 mass spectrometry data, *Atmos. Environ.*, 178, 129-139, 10.1016/j.atmosenv.2018.01.034, 2018.

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721 **Table legend**

722 Table 1: Number of organic compounds in each subgroup and the abundance-weighted average
723 values of molecular mass (MM_{avg}), elemental ratios, double bond equivalent (DBE) and
724 aromaticity equivalent (X_c) for detected organic compounds in ESI⁻ and ESI⁺ in the three Chinese
725 cities.

726 **Figure legends**

727 Figure 1: Mass spectra of detected organic compounds reconstructed from extracted ion
728 chromatograms in ESI⁻ and ESI⁺. X axis refers to the molecular mass (Da) of the identified species.
729 Y axis refers to the relative peak abundance of each individual compound to the compound with
730 the greatest peak abundance. The pie charts show the percentage of each organic compound
731 subgroup (i.e., CHO, CHON, CHOS, CHONS and CHN) in each sample in terms of peak
732 abundance. The map in the lower right corner shows the locations of these three megacities in China.

733 Figure 2: (a) Venn diagrams showing the number distribution of all molecular formulas detected in
734 ESI⁻ and ESI⁺ for all sample locations. (b) Peak abundance contribution of each elemental formula
735 category to the total common formulas.

736 Figure 3: Double bond equivalent (DBE) versus carbon number for all CHO⁻ compounds for all
737 sample locations. The molecular formula represents the abundance-weighted average CHO⁻
738 formula and the area of the circles is proportional to the fourth root of the peak abundance of an
739 individual compound. The color bar denotes the aromaticity equivalent (gray with $X_c < 2.50$, purple
740 with $2.50 \leq X_c < 2.70$ and red with $X_c \geq 2.70$). The pie charts show the percentage of each X_c
741 category (i.e., gray color-coded compounds, purple color-coded compounds and red color-coded
742 compounds) in each sample in terms of peak abundance.

743 Figure 4: Classification of CHON⁻ compounds into different subgroups according to O/N ratios in
744 their formulas. The y-axis indicates the relative contribution of each specific O/N ratio subgroup to
745 the sum of peak abundances of CHON⁻ compounds.

746 Figure 5: Double bond equivalent (DBE) versus carbon number for all CHON⁻ compounds for all
747 sample locations. The molecular formula represents the abundance-weighted average CHON⁻
748 formula and the area of circles is proportional to the fourth root of the peak abundance of an
749 individual compound. The color bar denotes the aromaticity equivalent (gray with $XC < 2.50$,
750 purple with $2.50 \leq XC < 2.70$ and red with $XC \geq 2.70$). The pie charts show the percentage of each
751 X_c category (i.e., gray color-coded compounds, purple color-coded compounds and red color-
752 coded compounds) in each sample in terms of peak abundance.

753 Figure 6: Van Krevelen diagrams for CHN⁺ compounds in Changchun, Shanghai and Guangzhou
754 samples. The area of circles is proportional to the fourth root of the peak abundance of an individual
755 compound and the color bar denotes the aromaticity equivalent (gray with $X_c < 2.50$, purple with
756 $2.50 \leq X_c < 2.70$ and red with $X_c \geq 2.70$). The pie charts show the percentage of each X_c category
757 (i.e., gray color-coded compounds, purple color-coded compounds and red color-coded compounds)
758 in each sample in terms of peak abundance.

759 Figure 7: Double bond equivalent (DBE) versus carbon number for all CHOS⁻ compounds for all
760 sample locations. The molecular formula represents the abundance-weighted average CHOS⁻
761 formula and the area of circles is proportional to the fourth root of the peak abundance of an
762 individual compound. The color bar denotes the aromaticity equivalent (gray with $X_c < 2.50$, purple
763 with $2.50 \leq X_c < 2.70$ and red with $X_c \geq 2.70$). The pie charts show the percentage of each X_c
764 category (i.e., gray color-coded compounds, purple color-coded compounds and red color-coded



765 compounds) in each sample in terms of peak abundance.

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795 Table 1. Number of organic compounds in each subgroup and the abundance-weighted average
 796 values of molecular mass (MM_{avg}), elemental ratios, double bond equivalent (DBE) and
 797 aromaticity equivalent (X_c) for detected organic compounds in ESI⁻ and ESI⁺ in the three Chinese
 798 cities.

Sample ID	Subgroup	Number of compounds	Relative abundance		MM_{avg}	H/C	O/C*	DBE	X_c
			(%)						
Changchun ⁻	total	769	100		169	1.03	0.58	5.02	2.13
	CHO ⁻	346	30		162	0.96	0.41	5.65	2.28
	CHON ⁻	180	55		163	0.94	0.51	5.24	2.44
	CHOS ⁻	155	10		198	1.56	1.17(0.52)	2.55	0.50
	CHONS ⁻	88	5		214	1.35	1.07(-1.4)	3.75	1.06
Shanghai ⁻	total	416	100		176	1.05	0.69	4.99	1.92
	CHO ⁻	164	40		171	0.97	0.59	5.37	1.94
	CHON ⁻	135	44		169	0.86	0.56	5.67	2.47
	CHOS ⁻	75	12		190	1.85	1.41(0.61)	1.79	0.34
	CHONS ⁻	42	4		266	1.56	1.00(0.11)	3.30	0.44
Guangzhou ⁻	total	488	100		183	1.14	0.74	4.55	1.65
	CHO ⁻	196	42		172	1.10	0.65	4.68	1.57
	CHON ⁻	161	39		173	0.89	0.58	5.56	2.41
	CHOS ⁻	86	14		201	1.85	1.48(0.71)	1.71	0.21
	CHONS ⁻	45	5		293	1.56	0.82(0.06)	3.45	0.43
Changchun ⁺	total	2943	100		160	1.21	0.13	5.58	2.36
	CHO ⁺	609	13		174	0.94	0.28	6.55	2.22
	CHN ⁺	696	40		154	1.22	0.00	5.84	2.60
	CHON ⁺	1594	46.5		161	1.27	0.19	5.11	2.22
	CHONS ⁺	44	0.5		196	1.91	0.70	2.64	0.09
Shanghai ⁺	total	704	100		162	1.37	0.09	4.91	2.32
	CHO ⁺	87	4		184	1.13	0.43	5.46	1.46
	CHN ⁺	253	71		159	1.38	0.00	5.08	2.55
	CHON ⁺	350	24.7		167	1.40	0.27	4.34	1.81
	CHONS ⁺	14	0.3		241	1.17	0.61	5.32	0.91
Guangzhou ⁺	total	687	100		161	1.41	0.17	4.58	2.07
	CHO ⁺	125	8		185	1.12	0.42	5.19	1.20
	CHN ⁺	205	62		156	1.42	0.00	4.80	2.47
	CHON ⁺	336	29		165	1.47	0.45	4.00	1.51
	CHONS ⁺	21	1		209	1.84	0.71	3.05	0.31

799 *The values in brackets indicate the (O-3S)/C and (O-3S-2N)/C ratios for CHOS and CHONS compounds, respectively,
 800 detected in ESI⁻ mode.

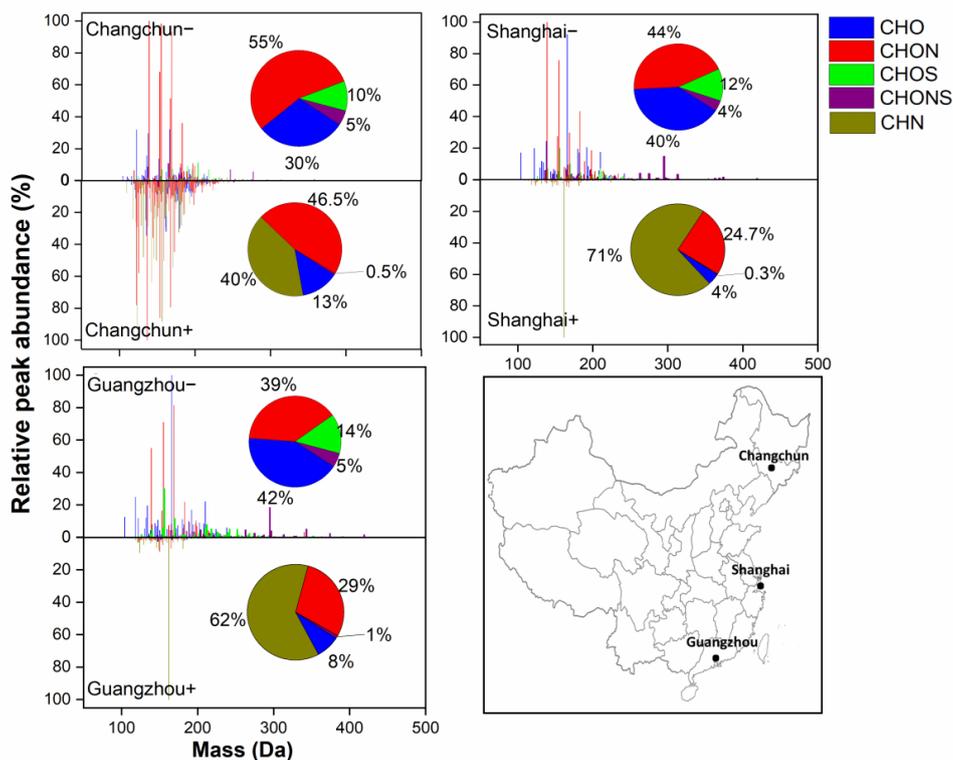
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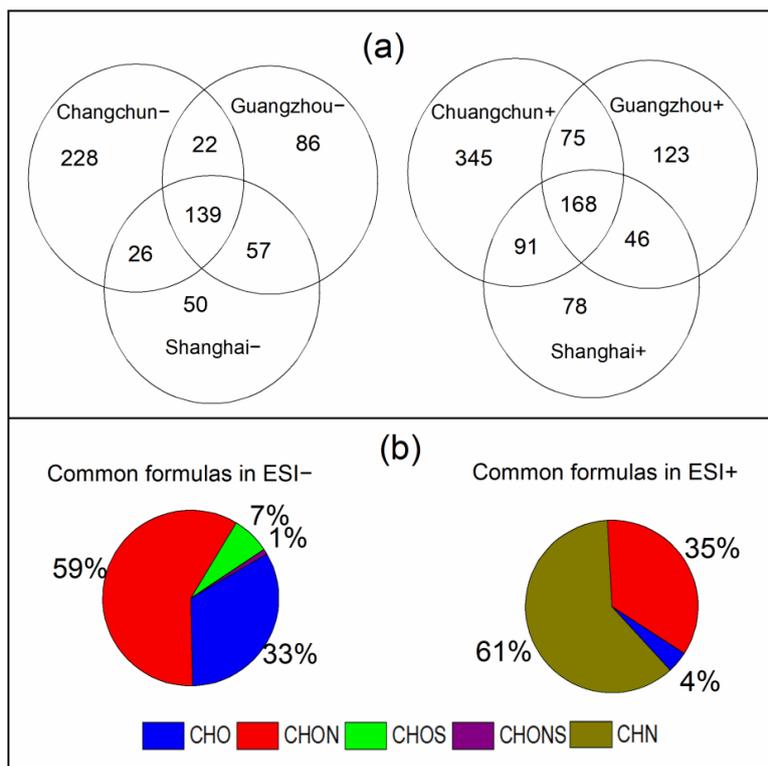
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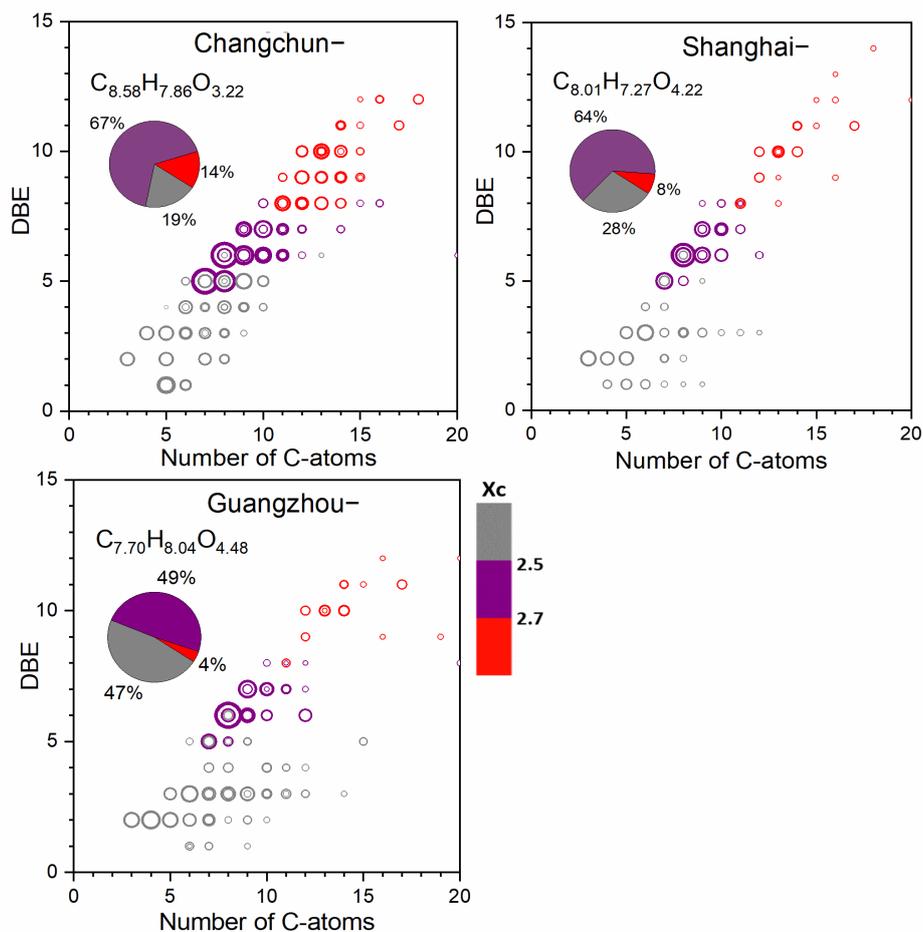
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Figure 1. Mass spectra of detected organic compounds reconstructed from extracted ion chromatograms in ESI⁻ and ESI⁺. X axis refers to the molecular mass (Da) of the identified species. Y axis refers to the relative peak abundance of each individual compound to the compound with the greatest peak abundance. The pie charts show the percentage of each organic compound subgroup (i.e., CHO, CHON, CHOS, CHONS and CHN) in each sample in terms of peak abundance. The map in the lower right corner shows the locations of these three megacities in China.



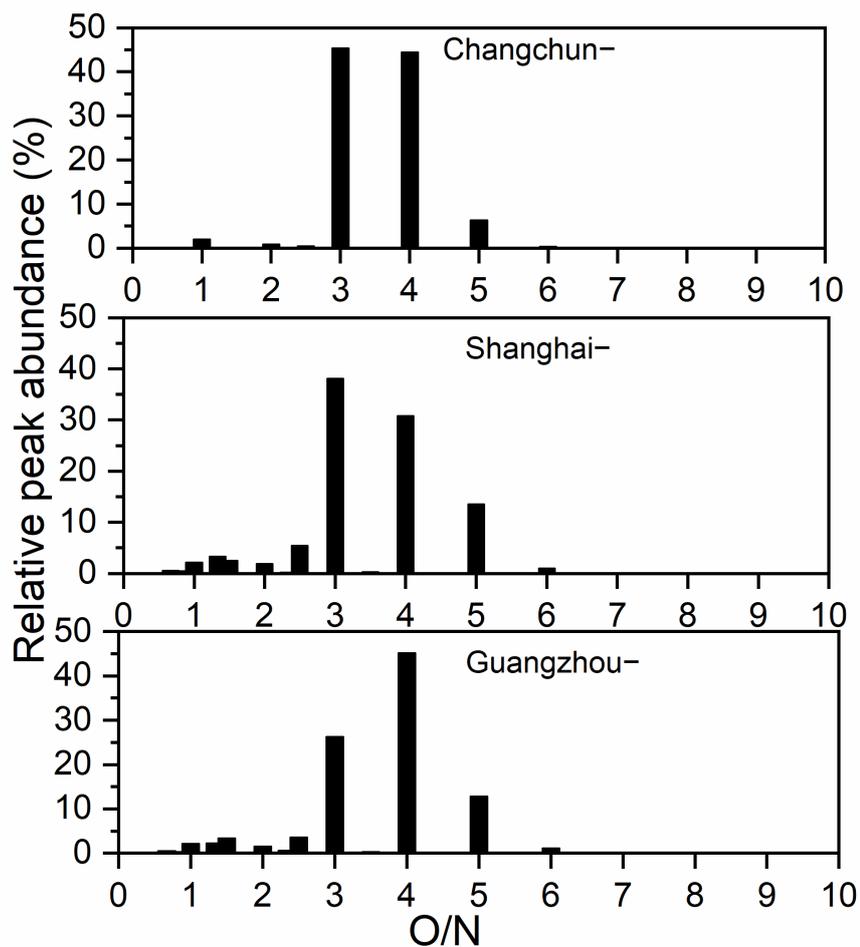
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813 Figure 2. (a) Venn diagrams showing the number distribution of all molecular formulas detected in
814 ESI- and ESI+ for all sample locations. (b) Peak abundance contribution of each elemental formula
815 category to the total common formulas.

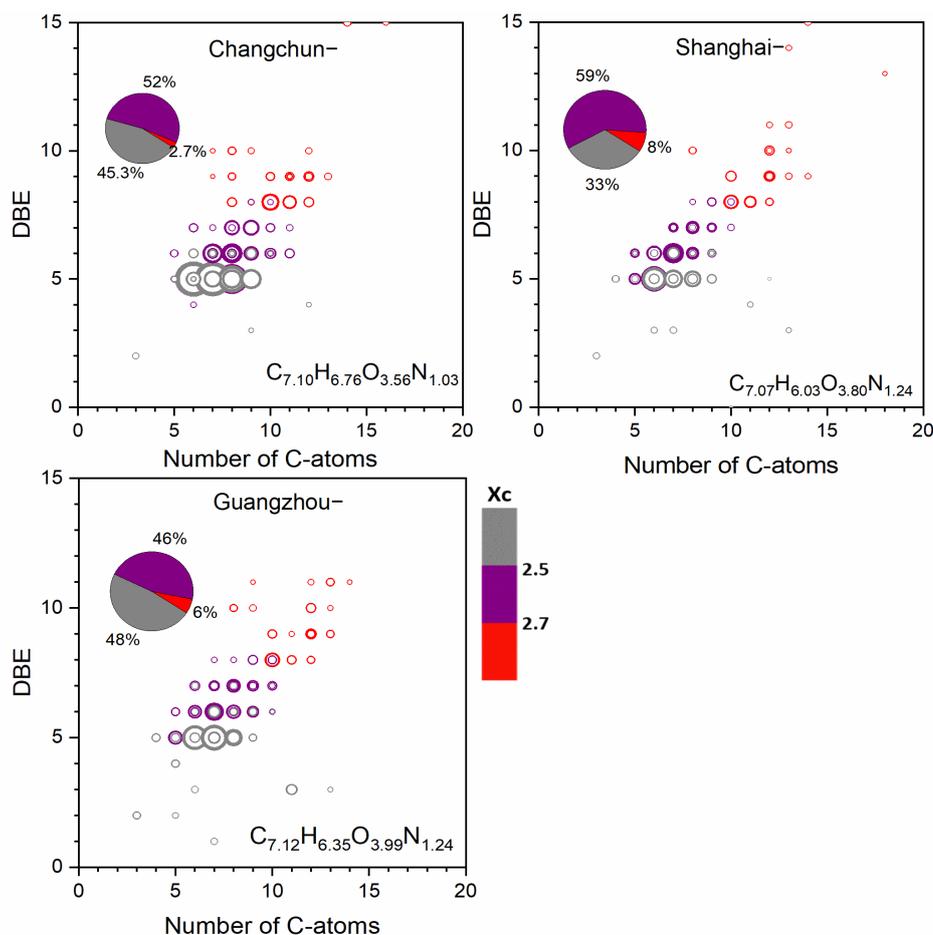


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817 Figure 3. Double bond equivalent (DBE) versus carbon number for all CHO- compounds for all
818 sample locations. The molecular formula represents the abundance-weighted average CHO-
819 formula and the area of the circles is proportional to the fourth root of the peak abundance of an
820 individual compound. The color bar denotes the aromaticity equivalent (gray with $X_c < 2.50$, purple
821 with $2.50 \leq X_c < 2.70$ and red with $X_c \geq 2.70$). The pie charts show the percentage of each X_c
822 category (i.e., gray color-coded compounds, purple color-coded compounds and red color-coded
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824
825 Figure 4. Classification of CHON- compounds into different subgroups according to O/N ratios in
826 their formulas. The y-axis indicates the relative contribution of each specific O/N ratio subgroup to
827 the sum of peak abundances of CHON- compounds.



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Figure 5. Double bond equivalent (DBE) versus carbon number for all CHON⁻ compounds for all sample locations. The molecular formula represents the abundance-weighted average CHON⁻ formula and the area of circles is proportional to the fourth root of the peak abundance of an individual compound. The color bar denotes the aromaticity equivalent (gray with XC < 2.50, purple with 2.50 ≤ XC < 2.70 and red with XC ≥ 2.70). The pie charts show the percentage of each Xc category (i.e., gray color-coded compounds, purple color-coded compounds and red color-coded compounds) in each sample in terms of peak abundance.

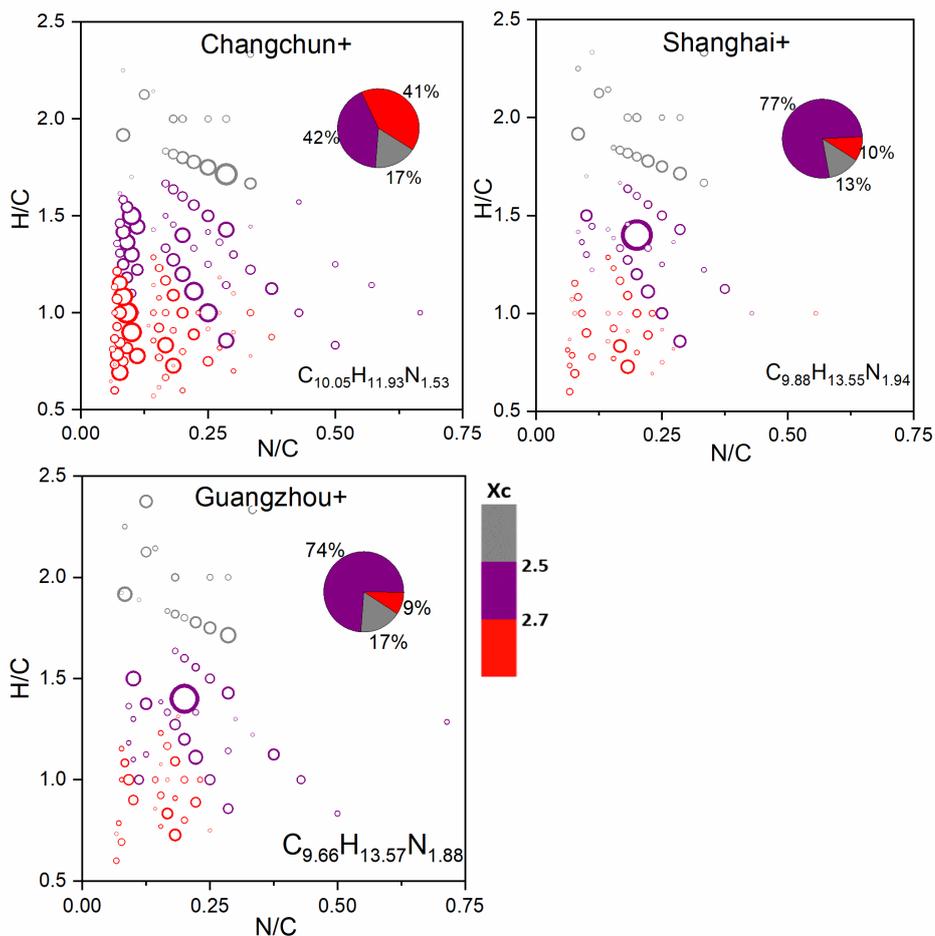
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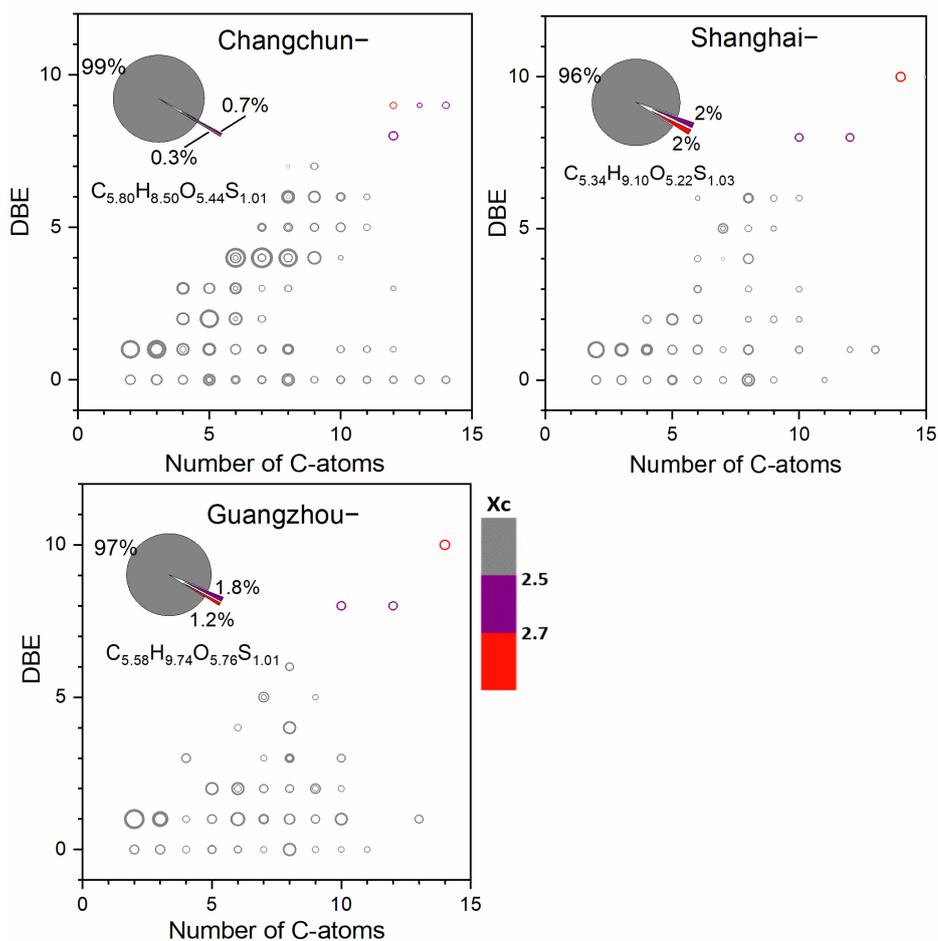
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837 Figure 6. Van Krevelen diagrams for CHN+ compounds in Changchun, Shanghai and Guangzhou
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839 compound and the color bar denotes the aromaticity equivalent (gray with $X_c < 2.50$, purple with
840 $2.50 \leq X_c < 2.70$ and red with $X_c \geq 2.70$). The pie charts show the percentage of each X_c category
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842 in each sample in terms of peak abundance.



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844 Figure 7. Double bond equivalent (DBE) versus carbon number for all CHOS-
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848 with $2.50 \leq X_c < 2.70$ and red with $X_c \geq 2.70$). The pie charts show the percentage of each X_c
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