The comment by **reviewer #1** is reproduced in a black font below. Our response follows the comment in a blue font. Text additions to the manuscript, for example, significantly modified sentences, appear in the revised manuscript in red color.

The section on limitations has answered some of my reservations. But I would still like to know what the variability is between the three samples at each location as this will improve the comparison between locations. This was not answered in the response.

We thank the reviewer for pointing out this variability between the three samples at each location. The main purpose of this study was to compare the chemical composition of organic compounds in the PM2.5 samples collected in the three Chinese cities. Therefore, to reduce the uncertainty caused by the variability between the three samples of each city, only organic compounds measured in all three samples of each city are used for intercity comparison. We have now added the statement in the revised manuscript (Page 7, Line 205-207) as follows:

'To reduce the uncertainty caused by the variability between the samples collected at each location, only organic compounds measured in all three samples of each city are used for intercity comparison in this study.'

1	Urban organic aerosol	composition in Eastern	China differs from	North to South: Molecular

2	insight from a lig	uid chromatogra	nhv-Orhitran	mass spectrometry	v studv
2	maight nom a ny	ulu chi omatogi a	phy-Orbitrap	mass specificity	siuuy

3	Kai Wang <sup>1,2,4</sup> , Ru-Jin Huang <sup>1</sup> , Martin Brüggemann <sup>3</sup> , Yun Zhang <sup>2</sup> , Lu Yang <sup>1</sup> , Haiyan Ni <sup>1</sup> , Jie Guo <sup>1</sup> ,							
4	Meng Wang <sup>1</sup> , Jiajun Han <sup>5</sup> , Merete Bilde <sup>4</sup> , Marianne Glasius <sup>4</sup> , and Thorsten Hoffmann <sup>2</sup>							
5	<sup>1</sup> State Key Laboratory of Loess and Quaternary Geology (SKLLQG), Center for Excellence in							
6	Quaternary Science and Global Change, and Key Laboratory of Aerosol Chemistry and Physics,							
7	Institute of Earth and Environment, Chinese Academy of Sciences, Xi'an 710061, China							
8	<sup>2</sup> Institute of Inorganic and Analytical Chemistry, Johannes Gutenberg University Mainz,							
9	Duesbergweg 10-14, Mainz 55128, Germany							
10	<sup>3</sup> Atmospheric Chemistry Department (ACD), Leibniz Institute for Tropospheric Research							
11	(TROPOS), Permoserstraße 15, 04318 Leipzig, Germany							
12	<sup>4</sup> Department of Chemistry, Aarhus University, Langelandsgade 140, DK-8000 Aarhus C, Denmark							
13	<sup>5</sup> Department of Chemistry, University of Toronto, 80 St. George Street, M5S3H6 Toronto, Canada							
14	Corresponding Author: Ru-Jin Huang (rujin.huang@ieecas.cn) and Thorsten Hoffmann							
15	(t.hoffmann@uni-mainz.de)							
16								
17								
18								
19								
20								
21								
22								
23								
24								
25								
26								

### 27 Abstract:

28 Air pollution by particulate matter in China affects human health, the ecosystem and the climate. 29 However, the chemical composition of particulate aerosol, especially of the organic fraction, is still not well understood. In this study, particulate aerosol samples with a diameter of  $\leq 2.5 \,\mu m \,(PM_{2.5})$ 30 31 were collected in January 2014 in three cities located in Northeast, East and Southeast China, 32 namely Changchun, Shanghai and Guangzhou. Organic aerosol (OA) in the PM<sub>2.5</sub> samples was 33 analyzed by ultrahigh performance liquid chromatography (UHPLC) coupled to high-resolution 34 Orbitrap mass spectrometry in both negative mode (ESI-) and positive mode electrospray 35 ionization (ESI+). After non-target screening including the assignment of molecular formulas, the 36 compounds were classified into five groups based on their elemental composition, i.e., CHO, 37 CHON, CHN, CHOS and CHONS. The CHO, CHON and CHN groups present the dominant signal 38 abundances of 81–99.7% in the mass spectra and the majority of these compounds were assigned 39 to mono- and polyaromatics, suggesting that anthropogenic emissions are a major source of urban 40 OA in all three cities. However, the chemical characteristics of these compounds varied between 41 the different cities. The degree of aromaticity and the number of polyaromatic compounds were 42 substantially higher in samples from Changchun, which could be attributed to the large emissions 43 from residential heating (i.e. coal combustion) during winter time in Northeast China. Moreover, 44 the ESI- analysis showed higher H/C and O/C ratios for organic compounds in Shanghai and 45 Guangzhou compared to samples from Changchun, indicating that OA undergoes more intense 46 photochemical oxidation processes in lower latitude regions of China and/or is affected to a larger 47 degree by biogenic sources. The majority of sulfur-containing compounds (CHOS and CHONS) in 48 all cities were assigned to aliphatic compounds with low degrees of unsaturation and aromaticity. 49 Here again, samples from Shanghai and Guangzhou show a greater chemical similarity but differ 50 largely from those from Changchun. It should be noted that the conclusions drawn in this study are 51 mainly based on comparison of molecular formulas weighted by peak abundance, and thus, are 52 associated with inherent uncertainties due to different ionization efficiencies for different organic 53 species.

# 54 1. Introduction

In the last decades, China has experienced rapid industrialization and urbanization accompanied by severe and persistent particulate air pollution (Huang et al., 2014; Sun et al., 2014; Ding et al., 2016; Song et al., 2018; Shi et al., 2019; Xu et al., 2019). These particulate air pollution extremes can not only influence the regional air quality and human health in China, but also lead to a global 59 environmental problem due to long-distance transport of pollutants. To better understand the effects 60 of air pollution on air quality and human health, chemical characterization of fine particle 61 (particulate matter with an aerodynamic diameter of less than 2.5  $\mu$ m, or PM<sub>2.5</sub>) is crucial. However, the chemical composition of  $PM_{2.5}$  in China is still poorly understood due to a wide variety of 62 63 natural and anthropogenic sources as well as complex multiphase chemical reactions (Lin et al., 64 2012a; Huang et al., 2014; Ding et al., 2016; Wang et al., 2017; Wang et al., 2018; An et al., 2019; 65 Tong et al., 2019; Wang et al., 2019a; Wang et al., 2019b). In particular, compared to the fairly 66 well understood nature of the inorganic fraction of aerosol, the organic fraction, also named organic 67 aerosol (OA), is considerably less understood in terms of chemical composition, corresponding 68 precursors, sources and formation mechanisms (Huang et al., 2017).

69 During pollution events in China, OA accounts for as high as more than 50% of the total mass of 70 fine particle (An et al., 2019). Chemical compounds in OA cover a large complexity of species 71 including alcohols, aldehydes, carboxylic acids, imidazoles, organosulfates, organonitrates and 72 polycyclic aromatic hydrocarbons (PAHs) (Lin et al., 2012a; Rincón et al., 2012; Kourtchev et al., 73 2014; Wang et al., 2018; Elzein et al., 2019; Wang et al., 2019a). Thus, the capacity of traditional 74 analytical techniques is limited to identify the compounds in OA and the majority (> 70%) of OA 75 has not been identified yet as specific compounds (Hoffmann et al., 2011). The insufficient 76 knowledge of chemical composition of OA hinders a better understanding of the sources, formation 77 and atmospheric processes of air pollution in China.

78 Recently, ultrahigh resolution mass spectrometry (UHRMS), such as Fourier transform ion 79 cyclotron resonance mass spectrometry (FTICR-MS) and Orbitrap-MS, coupled with soft 80 ionization sources (e.g., electrospray ionization (ESI) and atmospheric pressure chemical ionization 81 (APCI)) have been introduced to elucidate the molecular composition of OA (Nizkorodov et al., 82 2011; Lin et al., 2012a; Lin et al., 2012b; Rincón et al., 2012; Noziere et al., 2015; Kourtchev et al., 83 2016; Tong et al., 2016; Tu et al., 2016; Brüggemann et al., 2017; Wang et al., 2017; Fleming et 84 al., 2018; Laskin et al., 2018; Song et al., 2018; Wang et al., 2018; Brüggemann et al., 2019; 85 Daellenbach et al., 2019; Ning et al., 2019; Wang et al., 2019a). Due to the two outstanding features 86 of high resolving power and high mass accuracy, UHRMS can give precise elemental compositions 87 of individual organic compounds. However, UHRMS studies on Chinese urban OA are very limited. 88 Wang et al. (Wang et al., 2017) characterized OA in Shanghai and showed variations in chemical 89 composition among different months and between daytime and nighttime. Our recent Orbitrap MS 90 study (Wang et al., 2018) showed that wintertime OA in PM<sub>2.5</sub> collected in Beijing, China and 91 Mainz, Germany were very different in terms of chemical composition. In contrast, for summertime

92 OA from Germany and China, Brüggemann et al. (2019) found similar compounds and 93 concentrations of terpenoid organosulfates in  $PM_{10}$ , demonstrating that biogenic emission can 94 significantly affect OA composition at both locations. Ning et al. (2019) analyzed the OA collected 95 in a coastal Chinese city (Dalian) and found that more organic compounds were identified in haze 96 days compared to non-haze days. Nonetheless, since severe particulate pollution in China occurs 97 on a large-scale, more UHRMS studies are needed to fully elucidate the chemical composition of 98 OA in different Chinese cities.

99 In this study, PM<sub>2.5</sub> aerosol samples were collected in three Chinese cities, i.e., Changchun, 100 Shanghai and Guangzhou, and their organic fraction was analyzed using ultra-high-performance 101 liquid chromatography (UHPLC) coupled with Orbitrap-MS. The Chinese cities of Changchun, 102 Shanghai and Guangzhou are located in the Northeast, East and Southeast of China, which are major populated regions in China with a population of 7.5, 24 and 15 million, respectively. The 103 104 geographic locations of these three cities cover a large latitude spanning from 23.12° N to 43.53° N 105 resulting in different meteorological conditions, including intensity and duration of sunlight, 106 average daily temperature and monsoon climate. In addition, the industrial structure, energy 107 consumption and energy sources in these three cities are different, such as much more heavy 108 industries (e.g., coal chemical industry and steelworks) in Northeast China (Zhang, 2008), which 109 can cause difference in anthropogenic emissions, and can therefore influence the chemical 110 composition of urban OA. Moreover, OA is strongly affected by residential coal combustion during 111 winter in Northeast China (Huang et al., 2014; An et al., 2019). Therefore, this study presents a comprehensive overview of chemical composition of OA in three representative Chinese cities 112 113 during pollution episodes, which eventually can improve our understanding of OA effects on 114 climate and public health and also provide a chemical database for haze mitigation strategies in 115 China.

## 116 **2. Experimental**

## 117 **2.1 PM**<sub>2.5</sub> samples

118 Three 24–h integrated urban  $PM_{2.5}$  samples were collected during severe haze pollution events with 119 daily average  $PM_{2.5}$  mass concentration higher than 115 µg m<sup>-3</sup> in each of the three Chinese cities: 120 Changchun (43.54° N, 125.13° E, 1.5 m above the ground), Shanghai (31.30° N, 121.50° E, 20 m 121 above the ground) and Guangzhou (23.07° N, 113.21° E, 53 m above the ground), which are located 122 in the Northeast, East and Southeast regions of China, respectively (see Fig. 1). Samples in 123 Changchun were collected on 4, 24 and 29 of January 2014 with  $PM_{2.5}$  mass concentrations of

185–222 µg m<sup>-3</sup>, samples in Shanghai were collected on 1, 19 and 20 of January 2014 with PM<sub>25</sub> 124 mass concentrations of 159–172  $\mu$ g m<sup>-3</sup> and samples in Guangzhou were collected on 5, 6 and 11 125 of January 2014 with PM<sub>2.5</sub> mass concentrations of  $138-152 \ \mu g \ m^{-3}$ . Further details (e.g., the daily 126 127 average concentrations of  $PM_{2.5}$ ,  $SO_2$ ,  $NO_2$ , CO and  $O_3$ , the average temperature and the daily solar 128 radiation value during sampling dates) are presented in Table S1, the 48 hours back trajectories of 129 air arriving at the three sampling sites during the sampling periods are shown in Fig. S1. All PM<sub>2.5</sub> 130 samples were collected on prebaked quartz-fiber filters ( $20.3 \times 25.4$  cm) using a high-volume PM<sub>2.5</sub> sampler at a flow rate of 1.05 m<sup>3</sup> min<sup>-1</sup> (Tisch Environmental, USA) and at each sampling site field 131 132 blanks were taken. After sample collection, filters were stored at -20 °C until analysis.

### 133 **2.2 Sample analysis**

134 Detailed description on the filter sample extraction and UHPLC–Orbitrap MS analysis can be found 135 in our previous studies (Wang et al., 2018; Wang et al., 2019a). Briefly, a part of the filters (around 1.13 cm<sup>2</sup>, corresponding to about 600 µg particle mass in each extracted filter) was extracted three 136 137 times with 1.0-1.5 mL of acetonitrile-water (8/2, v/v) in an ultrasonic bath. The extracts were 138 combined, filtered through a 0.2 µm Teflon syringe filter and evaporated to almost dryness under 139 a gentle nitrogen stream. Finally, the residue was redissolved in 1000 µL acetonitrile-water (1/9, 140 v/v) to reach the total particulate mass concentration of around 600 µg mL<sup>-1</sup> for the following 141 analysis.

142 Compared to the direct infusion method applied in other UHRMS studies (Lin et al., 2012a; Lin et 143 al., 2012b; Rincón et al., 2012; Kourtchev et al., 2016; Fleming et al., 2018), the UHPLC technique 144 was used in this study, which could separate and concentrate the compounds before they entered 145 the ion source, reducing the ionization suppression and increasing the sensitive of the measurement. 146 In addition, it can provide separation of some compounds and information of retention time of the 147 compounds, which is useful for the identification of the compounds and the separation of isomers. The analytes were separated using a Hypersil Gold column (C18, 50 x 2.0 mm, 1.9 µm particle size) 148 149 with mobile phases consisting of (A) 0.04% formic acid and 2% acetonitrile in MilliQ water and 150 (B) 2% water in acetonitrile. Gradient elution was applied with the A and B mixture at a flow rate 151 of 500 µL min<sup>-1</sup> as follows: 0–1.5 min 2% B, 1.5–2.5 min from 2% to 20% B (linear), 2.5–5.5 min 152 20% B, 5.5–6.5 min from 20% to 30% B (linear), 6.5–7.5 min from 30% to 50% B (linear), 7.5–8.5 min from 50% to 98% B (linear), 8.5–11.0 min 98% B, 11.0–11.05 min from 98% to 2% B (linear), 153 154 and 11.05–11.1 min 2% B. The Q Exactive Hybrid Quadrupole-Orbitrap MS was equipped with a heated ESI source at 120 °C, applying a spray voltage of -3.3 kV and 4.0 kV for negative ESI mode 155

- 156 (ESI-) and positive ESI mode (ESI+), respectively. The mass scanning range was set from m/z 50
- to 500 with a resolving power of 70,000 @ m/z 200. The Orbitrap MS was externally calibrated
- before each measurement sequence using an Ultramark 1621 solution (Sigama–Aldrich, Germany)
- 159 providing mass accuracy of the instrument lower than 3 ppm. Each sample was measured in
- 160 triplicate with an injection volume of  $10 \,\mu$ L.

# 161 **2.3 Data processing**

A non-target peak picking software (SIEVE<sup>®</sup>, Thermo Fisher Scientific, Germany) was used to find 162 significant peaks in the LC-MS dataset and to calculate all mathematically possible chemical 163 164 formulas for ions signals with a sample-to-blank abundance ratio  $\geq 10$  using a mass tolerance of  $\pm$ 2 ppm. The permitted maximum elemental number of atoms was set as follows: <sup>12</sup>C (39), <sup>1</sup>H (72), 165 <sup>16</sup>O (20), <sup>14</sup>N (7), <sup>32</sup>S (4), <sup>35</sup>Cl (2) and <sup>23</sup>Na (1) (Kind and Fiehn, 2007; Lin et al., 2012a; Wang et 166 al., 2018). To remove the chemically unreasonable formulas, further constraint was applied by 167 setting H/C, O/C, N/C, S/C and Cl/C ratios in the ranges of 0.3-3, 0-3, 0-1.3, 0-0.8 and 0-0.8 168 169 (Kind and Fiehn, 2007; Lin et al., 2012a; Rincón et al., 2012; Wang et al., 2018; Zielinski et al., 170 2018), respectively. For chemical formula  $C_cH_hO_oN_nS_sCl_x$ , the double bond equivalent (DBE) was 171 calculated by the equation: DBE = (2c + 2 - h - x + n) / 2. The aromaticity equivalent (X<sub>C</sub>) as a 172 modified index for aromatic compounds was obtained using the equation:  $Xc = [3(DBE - (p \times o +$ 173  $(q \times n) - 2 / [DBE - (p \times o + q \times n)]$ , where p and q, respectively, refer to the fraction of oxygen 174 and sulfur atoms involved in the  $\pi$ -bond structure of a compound. As such the values of p and q 175 vary between compound categories (Yassine et al., 2014). For example, carboxylic acids and esters 176 are characterized using p = q = 0.5, while p = q = 1 and p = q = 0 are used for carbonyl and hydroxyl, 177 respectively. Since it is impossible to identify the structurers of the hundreds of formulas observed 178 in this study, we cannot know the exact values of p and q in an individual compound. Therefore, in this study, p = q = 0.5 was applied for compounds detected in ESI- as carboxylic compounds are 179 180 preferably ionized in negative mode. However, because of the high complexity of the mass spectra 181 in ESI+, p = q = 1 was used in ESI+ to avoid an overestimation of the amount of aromatics. Moreover, for DBE  $\leq (p \times o + q \times n)$  or  $X_C \leq 0$ ,  $X_C$  was defined as zero. Furthermore, in ESI–, for 182 183 odd numbers of  $(p \times o + q \times n)$ , the value of  $(p \times o + q \times n)$  was rounded down to the lower integer. 184  $Xc \ge 2.50$  and  $Xc \ge 2.71$  have been suggested as unambiguous minimum criteria for the presence 185 of monoaromatics and polyaromatics, respectively (Yassine et al., 2014).

Comparing the peak abundance has been used in recent UHRMS studies (Wang et al., 2017;
Fleming et al., 2018; Song et al., 2018; Ning et al., 2019) to illustrate the relative importance of

188 specific types of compounds. However, it should be noted that different organic compounds have 189 different signal response in the mass spectrometer due to the differences in ionization and

transmission efficiencies (Schmidt et al., 2006; Leito et al., 2008; Perry et al., 2008; Kruve et al.,

191 2014). Therefore, uncertainties may exist when comparing the peak areas among compounds. In

- this work, we assume that all organic compounds have the same peak abundance response in the
- 193 mass spectrometer. The peak abundance-weighted average molecular mass (MM), elemental ratios,
- 194 DBE, and Xc for formula  $C_cH_hO_oN_nS_sCl_x$  were calculated using following equations:
- 195  $MM_{avg} = \sum (MM_i \times A_i) / \sum A_i$
- 196  $O/C_{avg} = \sum (O/C_i \times A_i) / \sum A_i$
- 197  $H/C_{avg} = \sum (H/C_i \times A_i) / \sum A_i$
- 198  $DBE_{avg} = \sum (DBE_i \times A_i) / \sum A_i$
- 199  $Xc_{avg} = \sum (Xc_i \times A_i) / \sum A_i$
- 200 where  $A_i$  is the peak abundance for each individual compound i.

# 201 3. Results and discussion

## 202 **3.1 General characteristics**

203 The main purpose of this study was to tentatively identify and compare the chemical composition of organic compounds in the PM<sub>2.5</sub> samples collected in the three Chinese cities: Changchun, 204 205 Shanghai and Guangzhou during pollution episodes. To reduce the uncertainty caused by the 206 variability between the samples collected at each location, only organic compounds measured in 207 all three samples of each city are used for intercity comparison in this study. The number of organic 208 compounds and molecular formulas detected in each city, the peak abundance-weighted average 209 values of molecular mass (MM<sub>avg</sub>), elemental ratios, DBE, Xc and the isomer number fraction 210 (meaning the percentage of formula numbers that have isomers among all assigned formulas) for 211 each subgroup are listed in Table 1. It should be noted that in this study we focus solely on organic 212 compounds with elevated signal abundances, and thus, presumably rather high concentrations. In 213 contrast to our previous study (Wang et al., 2018), compounds with low concentrations were 214 excluded by increasing the reconstitution volume from 500  $\mu$ L to 1000  $\mu$ L, reducing the sample 215 injection volume from 20  $\mu$ L to 10  $\mu$ L, and increasing the sample-to-blank ratio from 3 to 10 during 216 data processing.

217 Overall, 416–769 (assigned to 272–415 molecular formulas) and 687–2943 (assigned to 383–679 218 molecular formulas) organic compounds in different samples were determined in ESI- and ESI+, 219 respectively. The largest number of organic compounds was observed in Changchun samples in 220 both ESI- and ESI+, indicating that OA collected during winter season in Northeast China was 221 more complex compared to urban OA in East and Southeast China. This increased number of 222 compounds can possibly be explained by the large residential coal combustion emissions in winter 223 in North China (Huang et al., 2014; Song et al., 2018; An et al., 2019). In addition, ambient temperatures were lowest during the sampling period in Changchun (i.e., -14 °C to -9 °C, Table 224 S1), which likely led to a decreased boundary layer height and therefore enhanced accumulation of 225 226 pollutants and enhanced formation of secondary organic aerosol through for example gas-to-227 particle partitioning.

228 As shown in Table 1, the abundance-weighted average values of  $MM_{avg}$  and O/C ratio of the total 229 assigned formulas for Changchun samples dectected in negative mode (Changchun-) are 169 and 230 0.58, respectively, which are lower than those for Shanghai– ( $MM_{avg} = 176$  and O/C = 0.69) and 231 for Guangzhou– ( $MM_{avg} = 183$  and O/C = 0.74). On the contrary, the aromaticity equivalent Xc for 232 organics detected in Changchun–, Xc(Changchun–) = 2.13, is substantially higher than that for 233 Shanghai–, Xc(Shanghai–) = 1.92, and Guangzhou–, Xc(Guangzhou–) = 1.65. These observations 234 indicate that urban OA in Northeast China features a lower degree of oxidation and a higher degree 235 of aromaticity compared to urban OA in East and Southeast China. Furthermore, the relative peak abundance fraction of compounds with  $O/C \ge 0.6$ , which are considered as highly oxidized 236 237 compounds (Tu et al., 2016), is 31% in Changchun-, and higher in Shanghai- (46%) and 238 Guangzhou– (51%). The different chemical composition of the samples is probably caused by the 239 rather low ambient temperatures and decreased photochemical processing of organic compounds 240 in Northeast China (indicated by the lower solar radiation in Northeast China, see Table S1), 241 slowing down oxidation processes and leading to a larger number of PAHs, which are mainly 242 emitted from coal burning (Huang et al., 2014; Song et al., 2018) or by different 243 biogenic/anthropogenic precursors. In addition, long-range transport of air masses (see the 48 hours 244 back trajectories in Fig. S1) may have a certain effect on the chemical properties of aerosol samples 245 collected in the three cities.

Figure 1 shows the reconstructed mass spectra of organic compounds detected in ESI- and ESI+.

247 A major fraction organic species detected in ESI- are attributed to CHO- and CHON-, accounting

for 30–42% and 39–55% in terms of peak abundance, respectively, and comprising 39–45% and

249 23–33% in terms of peak numbers, respectively. This is consistent with previous studies on Chinese

250 urban OA by Wang et al. (2017 and 2018) and Brüggemann et al. (2019). Comparing the organic 251 compounds detected in ESI- for the three cities, 120 formulas were observed in all cities as 252 common formulas (which refer to the compounds detected in all cities with the same molecular 253 formulas and with the same retention times (retention time difference  $\leq 0.1$  min)) (Fig. 2a), 254 accounting for 29-44% and 57-71% of all assigned formulas in terms of formula numbers and 255 peak abundance, respectively. Despite the above-mentioned differences in chemical composition 256 for OA from Changchun compared to OA from Shanghai and Guangzhou, these results demonstrate 257 that still a large number of common organic compounds exist in Chinese urban OAs collected in 258 different cities, in particular for organics with higher signal abundances. Furthermore, as shown by 259 the pie chart in Fig. 2b, these common formulas are dominated by CHON- and CHO-, accounting 260 for 62% and 30% of the total common formulas in terms of peak abundance, respectively.

261 As it is commonly known, ESI exhibits different ionization mechanisms in negative and positive 262 ionization modes. While ESI- is especially sensitive to deprotonatable compounds (e.g., organic 263 acids), ESI+ is more sensitive to protonatable compounds (e.g., organic amines) (Ho et al., 2003). 264 Due to the different ionization mechanisms, clear differences were observed in the mass spectra 265 (Fig. 1) and chemical characteristics (Table 1) from ESI- and ESI+ measurements. For example, 266 CHO compounds were preferentially detected in ESI-, accounting for a relatively large fraction of 267 30-42% of all detected compounds in terms of peak abundance, compared to merely 4-13% for 268 such CHO compounds in ESI+. In contrast, CHN compounds were only observed in ESI+, yielding 269 a rather large peak abundance fraction of 40–71%. In particular, as can be seen in Fig.1, several 270 peaks of CHN+ compounds in Shanghai+ and Guangzhou+ have much higher abundance compared 271 to other organic species, probably due to their high concentrations and/or high ionization 272 efficiencies in the positive mode. This observation indicates that most CHO compounds with high 273 concentrations are probably organic acids, whereas the majority of CHN compounds likely belong 274 to the group of organic amines, which is in good agreement with previous studies (Lin et al., 2012a; 275 Wang et al., 2017; Wang et al., 2018). Organic compounds in ESI+ are dominated by CHN+ and 276 CHON+ compounds in terms of both peak numbers and peak abundance and these compounds are 277 characterized by rather high H/C ratio and low O/C ratios (Table 1), indicating a low degree of 278 oxidation. The Venn diagram presented for ESI+ measurements in Fig. 2a shows that out of a total 279 of 383-679 formulas, 129 formulas were found in samples from all three cities. Such common 280 formulas, thus, account for 19-34% and 30-75% of all assigned formulas in terms of formula 281 numbers and peak abundance, respectively. Among these common formulas, CHN+ and CHON+ exhibit the highest abundance fractions of 72% and 26%, respectively (Fig. 2b). 282

283 In the following, we will compare and discuss the chemical properties in detail for the three cities,

284 including degrees of oxidation, unsaturation and aromaticity of each organic compound class (i.e.,

285 CHO, CHON, CHN, CHOS and CHONS). It should be noted that the chlorine-containing

286 compounds were not discussed in this study due to the very low MS signal abundance. In addition,

- 287 since peak abundances for the formula can vary by orders of magnitude, the area of the circles
- 288 presented in the Figure 3 and Figures 5–7 is proportional to the fourth root of the peak abundance
- of each formula to reduce the size difference of the circles. For a more detailed comparison, figures
- 289
- 290 with the circle size related to the absolute peak abundances are presented in the SI.

#### 291 **3.2 CHO compounds**

292 CHO compounds have been widely observed in urban OA, accounting for a substantial fraction 293 (8-67%) of OA (Rincón et al., 2012; Tao et al., 2014; Wang et al., 2017; Wang et al., 2018). 294 Previous studies have shown that a large fraction of CHO compounds in urban OA is composed of 295 organic acids, containing deprotonatable carboxyl functional groups, which are detected 296 preferentially in negative ionization mode when using ESI-MS. As shown in Table 1, a total of 297 346, 164, and 196 CHO- compounds were detected in ESI- in the OA samples collected in 298 Changchun, Shanghai and Guangzhou, accounting for 30%, 40% and 42% of the overall peak 299 abundance in each sample, respectively. Out of all assigned formulas, 47 common CHO– formulas 300 were observed for all cites, accounting for 35-52% and 42-68% of all identified CHO- formulas 301 in terms of formula numbers and peak abundance, respectively.

302 Despite this similarity, OA samples from Changchun- (i.e. in negative ionization mode) exhibit 303 certain differences compared to samples from Shanghai- and Guangzhou-. The average H/C 304 values for CHO- compounds are in a similar range for the three locations (i.e., 0.96–1.10), however, 305 the average O/C values for O/C(Shanghai–) = 0.59 and O/C(Guangzhou–) = 0.65 are rather high 306 compared to the average O/C ratio for Changchun–, O/C(Changchun–) = 0.41. Furthermore, the 307 relative peak abundance fraction of CHO- compounds with  $O/C \ge 0.6$ , which are considered as 308 highly oxidized compounds (Tu et al., 2016), is 14% in Changchun and somewhat higher in 309 Shanghai–(34%) and Guangzhou–(45%). Altogether, these results indicate that CHO– compounds 310 in urban OA from East and Southeast China experienced more intense oxidation and aging 311 processes and/or were affected to a larger degree by biogenic sources.

- 312 Similarly, as shown in Fig. 3, the abundance-weighted average molecular formulas for CHO-
- 313 compounds in Changchun–, Shanghai– and Guangzhou– are C<sub>8.58</sub>H<sub>7.86</sub>O<sub>3.22</sub> (MM<sub>avg</sub>(Changchun–)
- = 162), C<sub>8.01</sub>H<sub>7.27</sub>O<sub>4.22</sub> (MM<sub>avg</sub>(Shanghai-) = 171) and C<sub>7.70</sub>H<sub>8.04</sub>O<sub>4.48</sub> (MM<sub>avg</sub>(Guangzhou-) = 172), 314

- 315 respectively. Again, these average formulas show that CHO- in Shanghai- and Guangzhou-
- 316 experienced more intense oxidation processes and/or were affected to a larger degree by biogenic
- 317 precursors, indicated by the larger abundance-weighted MM<sub>avg</sub> with a higher degree of oxygenation.
- 318 In contrast, CHO- compounds from OA samples in Changchun- exhibit a lower abundance-
- 319 weighted MM<sub>avg</sub> with a decreased oxygen content.
- 320 Besides oxygenation, the aromaticity of the detected CHO- compounds exhibits remarkable 321 differences in these three cities. In all cities, the CHO– compounds with high peak abundance were 322 mainly assigned to monoaromatics with  $2.5 \le Xc < 2.7$  (purple circles in Fig. 3) in the region of 323 7–12 carbon atoms per compound and DBE values of 5–7. The relative peak abundance fraction 324 of monoaromatics in total CHO- compounds is 67% in Changchun, which is higher compared to 325 64% in Shanghai and 49% in Guangzhou. In addition, 14% of CHO- compounds in Changchun 326 were identified as polyaromatic compounds with  $Xc \ge 2.7$  (red circles in Fig. 3), which is higher 327 than the 8% in Shanghai and 4% in Guangzhou. These observations indicate that CHO- compounds 328 in the three Chinese cities are highly affected by aromatic precursors (e.g., benzene, toluene and 329 naphthalene), in particular for the Changchun aerosol samples.
- 330 Besides the monoaromatics and polyaromatics, the rest of the detected CHO- compounds were 331 assigned to aliphatic compounds with an Xc lower than 2.5 (grey circles in Fig. 3). Interestingly, 332 these aliphatic compounds account for about 47% of all CHO- compounds for Guangzhou-333 samples in terms of peak abundance, whereas samples from Changchun- and Shanghai- exhibit 334 only rather small fractions of such CHO- compounds, i.e., 19% and 28%, respectively. Such 335 aliphatic compounds are commonly derived from biogenic precursors (Kourtchev et al., 2016) and 336 vehicle emission (Tao et al., 2014; Wang et al., 2017) and/or generated by intense oxidation 337 processes of aromatic precursors, indicating the different biogenic and anthropogenic emission 338 sources and chemical reaction processes for OAs in the three cities.
- In addition, through the analysis of individual formulas, we find that for the Changchun– samples,
- 340 formulas of  $C_8H_6O_4$ ,  $C_7H_6O_2$ ,  $C_7H_6O_3$ ,  $C_8H_8O_2$ , and  $C_8H_8O_3$  with DBE values of 6, 5, 5, 5, and 5
- dominate the assigned CHO formulas with respect to peak abundance. According to previous studies,  $C_8H_6O_4$ ,  $C_7H_6O_2$  and  $C_7H_6O_3$  are suggested to be phthalic acid, benzoic acid and
- 342 studies,  $C_8H_6O_4$ ,  $C_7H_6O_2$  and  $C_7H_6O_3$  are suggested to be phthalic acid, benzoic acid and 343 monohydroxy benzoic acid, respectively, which are derived from naphthalene (Kautzman et al.,
- 2010; Riva et al., 2015; Wang et al., 2017; He et al., 2018; Huang et al., 2019).C<sub>8</sub>H<sub>8</sub>O<sub>2</sub> is likely 4-
- 345 hydroxy acetophenone, which could be derived from estragole (Pereira et al., 2014), while  $C_8H_8O_3$
- is suggested to be either 4-methoxybenzoic acid generated from estragole (Pereira et al., 2014) or

- 347 vanillin emitted from biomass burning (Li et al., 2014). For the Shanghai– samples, besides  $C_8H_6O_4$ ,
- 348  $C_7H_6O_3$  and  $C_7H_6O_2$ , formulas of  $C_6H_8O_7$  and  $C_9H_8O_4$  with DBE values of 3 and 6 were observed
- 349 with high peak abundances.  $C_6H_8O_7$  was identified as citric acid in the pollen sample and mountain
- 350 particle sample in previous studies (Fu et al., 2008; Wang et al., 2009; Jung and Kawamura, 2011)
- and  $C_{9}H_{8}O_{4}$  are probably homophthalic acid derived from e.g. estragole (Pereira et al., 2014). For
- 352 the Guangzhou– samples, besides the formulas of  $C_8H_6O_4$  and  $C_6H_8O_7$  discussed above,  $C_4H_6O_4$
- and  $C_4H_6O_5$  with low DBE values of two were detected with high abundances and are suggested to
- be succinic acid and malic acid, respectively (Claeys et al., 2004; Wang et al., 2017).

### 355 3.3 CHON compounds

A large amount of nitrogen-containing organic compounds was detected in these three cities, accounting for 39–55% and 25–47% of total peak abundance detected in ESI– and ESI+, respectively. Out of all assigned formulas, 45 common CHON– and 62 common CHON+ formulas were observed in all cities, accounting for 65–82% and 25–44% of all CHON compounds detected in ESI– and ESI+ in terms of peak abundance, respectively. It indicates that a large amount of CHON compounds in all three Chinese cities show similar properties of chemical composition.

362 The CHON compounds were further classified into different subgroups according to their O/N 363 ratios (Fig. 4 for CHON- and Fig. S3 for CHON+) or according to the number of nitrogen atoms in their molecular formulas (see Fig. S4 for CHON– and S5 for CHON+). As shown in Fig. 4, the 364 365 majority (84–96% in terms of peak abundance) of CHON– compounds exhibited O/N ratios  $\ge 3$ , 366 allowing the assignment of one nitro  $(-NO_2)$  or nitrooxy  $(-ONO_2)$  group for these formulas, which 367 are preferentially ionized in ESI- mode (Lin et al., 2012b; Wang et al., 2017; Song et al., 2018; 368 Wang et al., 2018). CHON– formulas with O/N ratios  $\geq 4$  suggest the presence of further oxygenated functional groups, such as a hydroxyl group (-OH) or a carbonyl group (C=O). In 369 370 terms of peak abundance, 59% of CHON- compounds observed in Guangzhou- exhibited formulas 371 with O/N ratios  $\ge 4$ , which is higher than 51% in Changchun– and 45% in Shanghai–, indicating 372 that CHON- compounds in Southeast China show a higher degree of oxidation compared to those 373 in Northeast and East China. Not surprisingly, CHON+ compounds generally exhibit lower O/N 374 ratios (Fig. S3), as they probably contain reduced nitrogen functional group (e.g., amines) which are preferably detected in ESI+. As shown in Fig. S3, CHON+ compounds with O/N ratio of 1 are 375 376 dominant in Changchun+, whereas CHON+ compounds in Shanghai+ and Guangzhou+ show a 377 broader range of O/N ratios from 1 to 3. Moreover, the average O/C ratios (0.27–0.45) in Shanghai+ 378 and Guangzhou+ (Table 1) are much greater than that (0.19) in Changchun+. Consistent with the 379 observations for CHO compounds, these results indicate again that CHON+ compounds in the OA

380 of East and Southeast China experienced more intensive photooxidation and/or were affected to a

381 larger degree by biogenic precursors.

382 Figure 5 shows the DBE versus C number of CHON– compounds for the three cities. The majority 383 of CHON- compounds lie in the region of 5-15 C atoms and 3-10 DBEs. 67% of CHON-384 compounds in terms of peak abundance were assigned to mono or polyaromatics in Shanghai-, 385 which is higher than 52% in Guangzhou- and 55% in Changchun-. It indicates that CHON-386 compounds are dominated with aromatic compounds in all cities, while relatively higher peak 387 abundance weighted fraction of aromatic CHON- compounds were observed in Shanghai. The 388 peak abundance-weighted average molecular formulas for CHON- compounds in Changchun-, 389 Shanghai and Guangzhou are  $C_{7,10}H_{6,76}O_{3,56}N_{1,03}$ ,  $C_{7,07}H_{6,03}O_{3,80}N_{1,24}$  and  $C_{7,12}H_{6,36}O_{3,99}N_{1,24}$ , 390 respectively, showing that CHON- formulas in Shanghai- and Guangzhou- contain more O and 391 N atoms on average than those for Changchun–. Formulas of  $C_6H_5O_3N_1$ ,  $C_6H_5O_4N_1$ ,  $C_7H_7O_3N_1$ , 392  $C_7H_7O_4N_1$ ,  $C_8H_9O_3N_1$ , and  $C_8H_9O_4N_1$  were detected with the highest abundance in all cities. These 393 molecular formulas are in line with nitrophenol or nitrocatechol analogs, which have been identified 394 in a previous urban OA study (Wang et al., 2017). Furthermore, these nitrooxy-aromatic 395 compounds were shown to enhance light absorbing properties of OA (Laskin et al., 2015; Lin et al., 396 2015). In addition, it should be noted that the Xc values for  $C_6H_5O_4N_1$ ,  $C_7H_7O_4N_1$  and  $C_8H_9O_4N_1$ 397 were calculated to be lower than 2.5, suggesting that the fraction of aromatics in CHON-398 compounds was underestimated. This is because that for nitrocatechol analogs with formulas of 399  $C_6H_5O_4N_1$ ,  $C_7H_7O_4N_1$  and  $C_8H_9O_4N_1$ , only one oxygen atom is involved in the  $\pi$ -bond structure 400 corresponding to the p value of 0.25 in the Xc calculation equation, which is lower than the p value 401 of 0.5 applied for the Xc calculation in this study. The diagram of DBE versus C number for 402 CHON+ compounds observed in the three locations (presented in Fig. S7 in SI) shows that more 403 aromatic CHON+ compounds with relatively lower degree of oxidation were assigned in 404 Changchun+ samples compared to Shanghai+ and Guangzhou+ samples.

## 405 **3.4 CHN+ compounds**

406 696 CHN+ compounds were detected in Changchun+ samples in ESI+, which is higher than in 407 Shanghai+ (253) and Guangzhou (205). These CHN+ compounds are likely assignable to amines 408 according to previous studies (Rincón et al., 2012; Wang et al., 2017; Wang et al., 2018). The 409 number of CHN+ compounds accounts for 24%, 36% and 30% of the total organic compounds in 410 Changchun+, Shanghai+ and Guangzhou+, respectively, whereas the peak abundance of these 411 compounds accounts for 40%, 71% and 62%, respectively. The majority (> 97% in terms of peak 412 abundance) of CHN+ compounds have one or two nitrogen atoms in their molecular formulas (see 413 Fig. S9). Comparing the CHN+ compounds for the three cities, 51 common CHN+ formulas were 414 observed in all cities, which contribute to as much as 43-89% of the total abundance of CHN+ 415 formulas. This large percentage indicates that CHN+ compounds with presumably high 416 concentrations in Changchun+, Shanghai+ and Guangzhou+ exhibit similar chemical composition. 417 However, again OA samples from Changchun show some distinct differences to samples from 418 Guangzhou and Shanghai.

A van Krevelen diagram of CHN+ compounds detected in the three samples is shown in Fig. 6, 419 420 illustrating H/C ratios as a function of N/C ratio. In this plot, major parts of the CHN+ compounds 421 are found in a region, which is constraint by H/C ratios between 0.5 and 2 and N/C ratios lower 422 than 0.5. Moreover, the pie charts show that the majority (83-87%) in terms of peak abundance and 423 72-90% in terms of peak numbers) of these CHN+ compounds can be assigned to mono- and 424 polyaromatics with  $Xc \ge 2.5$ . In addition, as shown in Table 1, the average DBE and Xc values of 425 CHN+ compounds are the highest among all organic species. These observations imply that CHN+ 426 compounds exhibit the highest degree of aromaticity of all organics in the Chinese urban OA 427 samples, which is consistent with previous studies (Lin et al., 2012b; Rincón et al., 2012; Wang et 428 al., 2018). Polyaromatic compounds with  $Xc \ge 2.7$  are displayed in the lower left corner of the 429 van Krevelen diagram, accounting for 41% in terms of peak abundance (48% in terms of peak 430 numbers) of CHN+ compounds detected in Changchun+, but merely for 9-10% in terms of peak 431 abundance (27–31% in terms of peak numbers) in Shanghai+ and Guangzhou+. For example, 432 formulas of  $C_{11}H_{11}N_1$  (Xc = 2.7),  $C_{10}H_9N_1$  (Xc = 2.7), and  $C_{12}H_{13}N_1$  (Xc = 2.7), which are assigned 433 to be naphthalene core structure-containing compounds, have relatively higher abundance in 434 Changchun+ than in Shanghai+ and Guangzhou+. Moreover, the average DBE and Xc values of 435 CHN+ compounds (see Table 1) in Changchun+ are substantially higher than those in Shanghai+ 436 and Guangzhou+, further indicating that CHN+ compounds in Changchun+ show a higher degree 437 of aromaticity, which can be caused by large coal combustion emissions in the winter in Changchun. 438 Remarkably, as can be seen in Fig. 6, the abundance of CHN+ compounds in Changchun+ 439 distributes evenly among different individual CHN+ compounds, while in Shanghai+ and Guangzhou+ they are dominated by the formula of  $C_{10}H_{14}N_2$  (the biggest purple circle in Fig. 6) 440 with DBE value of 5, which probably has high concentration and/or high ionization efficiency in 441 442 the positive ESI mode. According to a previous smog chamber study (Laskin et al., 2010), most CHN+ aromatics are probably generated from biomass burning through the addition of reduced 443

nitrogen (e.g., NH<sub>3</sub>) to the organic molecules via imine formation reaction, indicating that biomass
burning probably made a certain contribution to the formation of CHN+ compounds observed in
the three urban OA samples in our study.

### 447 **3.5 CHOS**- compounds

In this study, 75-155 CHOS- compounds were observed, accounting for 10%, 12% and 14% of 448 449 the total peak abundance of all organics in Changchun–, Shanghai– and Guangzhou–, respectively. 450 Around 89–96% of these CHOS– compounds were found to fulfill the O/S  $\ge$  4 criterion allowing the assignment of at least one  $-OSO_3H$  functional group, and thus, a tentative classification to 451 452 organosulfates (OSs) (Lin et al., 2012a; Lin et al., 2012b; Tao et al., 2014; Wang et al., 2016; Wang 453 et al., 2017; Wang et al., 2018; Wang et al., 2019a). OSs were shown to affect the surface activity 454 and hygroscopic properties of the aerosol particles, leading to potential impacts on climate (Hansen et al., 2015; Wang et al., 2019a). Out of all formulas, 23 common CHOS- formulas were detected 455 456 for the three sample locations, accounting for 28%, 58% and 52% of the CHOS- peak abundance 457 in Changchun-, Shanghai- and Guangzhou-, respectively. However, 40 common CHOS-458 formulas were found between Shanghai- and Guangzhou-, accounting for 60-65% and 78-81% 459 in terms of the CHOS- formula numbers and peak abundance, respectively. This indicates that the 460 chemical composition of the major CHOS- compounds of Shanghai- and Guangzhou- are quite 461 similar, while they show substantial chemical differences for samples from Changchun-.

462 Figure 7 shows the DBEs as a function of carbon number for all CHOS- compounds detected for 463 the three cities. The CHOS- compounds exhibit a DBE range from 0 to 10 and carbon number 464 range of 2-15. However, the majority of CHOS- compounds with elevated peak abundances 465 concentrate in a region with rather low DBE values of 0-5. The average H/C ratios of CHOS-466 compounds are in the range of 1.56-1.85, and thus, higher than for any other compound class, 467 whereas the average DBE values of 1.71–2.55 are the lowest among all classes. This indicates that 468 CHOS- compounds in the OA from the three Chinese cities are characterized by a low degree of 469 unsaturation. Moreover, the pie charts in Fig. 7 show that aliphatic compounds with  $Xc \le 2.5$  are 470 dominant in CHOS- compounds with a fraction of 96-99% in terms of peak abundance, which is 471 substantially higher than that (13-48%) for CHO, CHON and CHN species. Aliphatic CHOS-472 compounds with  $C \leq 10$  can be formed from biogenic and/or anthropogenic precursors (Hansen 473 et al., 2014; Glasius et al., 2018; Wang et al., 2019a), such as C<sub>2</sub>H<sub>4</sub>O<sub>6</sub>S<sub>1</sub> (derived from glyoxal) 474 (Lim et al., 2010; McNeill et al., 2012), C<sub>3</sub>H<sub>6</sub>O<sub>6</sub>S<sub>1</sub> (derived from isoprene) (Surratt et al., 2007) and 475  $C_8H_{16}O_4S_1$  (derived from  $\alpha$ -pinene). However, more CHOS- compounds with C > 10 and with 476 DBEs lower than 1 are observed in Changchun-, such as  $C_{14}H_{28}O_5S_1$ ,  $C_{13}H_{26}O_5S_1$ ,  $C_{12}H_{24}O_5S_1$ , 477  $C_{11}H_{22}O_5S_1$  and  $C_{11}H_{20}O_6S_1$ . These high-carbon-number-containing CHOS- compounds are likely 478 formed from long-alkyl-chain compounds with less oxygenated functional groups, which were 479 previously suggested to be emitted from traffic (Tao et al., 2014) or derived from sesquiterpene 480 emissions (Brüggemann et al., 2019). However, as sesquiterpene emissions can be expected to be 481 very low in wintertime at Changchun, the presence of these compounds further underlines the 482 strong impact of anthropogenic emissions on CHOS- formation in Changchun-. In this study, 483 (O-3S)/C ratio was used instead of traditional O/C ratio to present the oxidation state of CHOS-484 compounds, since the sulfate functional group contains three more oxygen atoms than common 485 oxygen-containing groups (e.g., hydroxyl and carbonyl), which makes no contribution to the 486 oxidation state of the carbon backbone of the CHOS- compounds. Comparing average values for 487 H/C, (O-3S)/C and DBEs of CHOS- for the three sample locations (see Table 1), we find that the 488 H/C ratios (1.85) and (O-3S)/C ratios (0.61-0.71) for Shanghai- and Guangzhou- samples are 489 larger than those for Changchun– samples (H/C = 1.56 and (O-3S)/C = 0.52), whereas the DBE 490 values (1.71–1.79) in Shanghai– and Guangzhou– are lower than those for Changchun– (2.55). 491 These observations indicate that CHOS- compounds in urban OA from Northeast China are less 492 oxidized but more unsaturated compared to those in East and Southeast China, likely due to 493 enhanced emissions from residential heating during winter in North China.

## 494 **3.6 CHONS compounds**

495 4-5% of the total organics detected in ESI- were identified as CHONS- compounds in terms of 496 peak abundance. In contrast, CHONS+ compounds account merely for 0.3-1% of all organics 497 detected in ESI+. The average MM<sub>avg</sub> of the CHONS- compounds for the three sample locations 498 ranges from 214 to 293 Da, generally showing larger molecular masses than compounds of any 499 other class because of the likely presence of both nitrate and sulfate functional groups. In total, only 500 5 common CHONS- formulas were detected for all three sample locations, accounting for 4%, 21% 501 and 20% of the CHONS- peak abundance in Changchun-, Shanghai- and Guangzhou-, 502 respectively. As already observed for other compound classes, these percentages imply that the 503 CHONS- compounds in urban OA of Shanghai- and Guangzhou- exhibit a rather similar chemical 504 composition, whereas such compounds are different for Changchun-.

505 In the OA samples of Shanghai– and Guangzhou–, 78-87% of CHONS– compounds in terms of 506 peak abundance have 7 or more O atoms in their formulas, allowing the assignment of one –OSO<sub>3</sub>H 507 and one -NO<sub>3</sub> functional groups in the molecular structures, thus, classifying them as potential

508 nitrooxy-organosulfates. In contrast to Shanghai- and Guangzhou-, only 26% of CHONS-509 compounds were assigned to such nitroooxy-organosulfates for Changchun-, indicating that most 510 of the N atoms in the CHONS- compounds are present in a reduced oxidation state, e.g., in the 511 form of amines. The average DBE and Xc values of CHONS- compounds in Shanghai- and 512 Guangzhou- are 3.3-3.45 and 0.43-0.44, respectively. Again these values differ for the 513 Changchun– samples with an increased average DBE of 3.75 and an average Xc of 1.06, indicating 514 that CHONS- compounds in Changchun- possess on average a higher degree of unsaturation and 515 aromaticity compared to such compounds in Shanghai- and Guangzhou- samples. Interestingly, 516 the compound with formula  $C_{10}H_{17}O_7NS$  has the highest relative peak abundance (32%) in 517 Shanghai– and Guangzhou–, whereas in Changchun– the compound with formula  $C_2H_3O_4NS$  is 518 dominant.  $C_{10}H_{17}O_7NS$  has previously been identified as mononitrate organosulfate generated from 519  $\alpha/\beta$ -pinene (Iinuma et al., 2007; Surratt et al., 2008; Lin et al., 2012b; Wang et al., 2017), while 520  $C_2H_3O_4NS$  may be assigned as a cyanogroup-containing sulfate. This observation is comparable to 521 our previous study (Wang et al., 2019a), which found that  $C_{10}H_{17}O_7NS$  was dominant for CHONS-522 compounds in low-concentration aerosol samples collected in Beijing (China) and Mainz 523 (Germany). Consistently, a  $C_2H_3O_4NS$  compound had the highest abundance among CHONS-524 compounds in polluted Beijing aerosol samples. This agreement can be explained by the adjacent locations of Beijing (39.99° N, 116.39° E) and Changchun (43.54° N, 125.13° E) and similar 525 526 residential heating patterns by coal combustion during wintertime. In conclusion, these results 527 further demonstrate that the precursors for CHONS- compounds in Shanghai- and Guangzhou-528 are different from those in Changchun, which is probably due to differences in anthropogenic 529 emissions.

# 530 4 Conclusion

531 The molecular composition of the organic fraction of  $PM_{2.5}$  samples collected in three Chinese 532 megacities (Changchun, Shanghai and Guangzhou) was investigated using a UHPLC-Orbitrap 533 mass spectrometer. In total, 416-769 (ESI-) and 687-2943 (ESI+) organic compounds were 534 observed and separated into five subgroups: CHO, CHN, CHON, CHOS and CHONS. Specifically, 535 120 common formulas were detected in ESI- and 129 common formulas in ESI+ for all sample 536 locations, accounting for 57–71% and 30–75% in terms of peak abundance, respectively. Overall, 537 we found that urban OA in Changchun, Shanghai and Guangzhou shows a quite similar chemical 538 composition for organic compounds of high concentrations. The majority of these organic species 539 was assigned to mono-aromatic or poly-aromatic compounds, indicating that anthropogenic 540 emissions are the major source for urban OA in all three cities.

541 Despite the chemical similarity of the three sample locations for organic compounds in urban OA, 542 remarkable differences were found in chemical composition of the remaining particle constituents, 543 in particular for OA samples from Changchun. In general, a larger amount of polyaromatics was 544 observed for Changchun samples, most likely due to emissions from coal combustion during 545 wintertime residential heating period. Moreover, the peak abundance-weighted average DBE and 546 average Xc values of the total organic compounds in Changchun were found to be larger than those 547 for Shanghai and Guangzhou, showing that organic compounds in Changchun possess a higher degree of unsaturation and aromaticity. For average H/C and O/C ratios a similar trend was 548 549 observed. While average H/C and O/C ratios detected in ESI- were found to be highest for 550 Guangzhou samples, relatively lower values were observed for Shanghai and Changchun samples, 551 indicating that OA collected in lower latitude regions of China experiences more intense 552 photochemical oxidation processes and/or are affect to a larger degree by biogenic sources.

# 553 **5 Limitations**

554 In this study, we used the peak abundance-weighted method to illustrate the difference in chemical 555 formulas assigned by Orbitrap mass spectrometry. This comparison was made based on the 556 assumption that the measured organic compounds have same peak abundance response in the mass 557 spectrometer. However, this assumption can bring some uncertainties because the ionization 558 efficiencies vary between different compounds (Schmidt et al., 2006; Leito et al., 2008; Perry et al., 559 2008; Kruve et al., 2014). For example, the ionization efficiencies of nitrophenol species detected 560 in negative ESI mode can vary by a large degree depending on the position of the substituents at 561 the nitrobenzene ring (Schmidt et al., 2006; Kruve et al., 2014) and the ionization efficiencies of 562 carboxylic acids can also vary by several orders of magnitude depending on the structures (Kruve 563 et al., 2014). Nonetheless, it is a challenging analytical task to identify and quantify all compounds 564 in ambient OA due to the high chemical complexity of OA and the limits in authentic standards of 565 OA. Despite the inherent uncertainties, the peak abundance-weighted comparison of molecular 566 formulas provides an overview of the difference in chemical composition of OA in these three 567 representative Chinese cities. In particular, the chemical formulas assigned in this study can be 568 validated in future studies by authentic standards and the difference in ionization efficiencies can 569 be further evaluated.

570

571 **Author contributions.** RJH, TH and KW conducted the study design. LY, HN, JG and MW 572 collected the PM<sub>2.5</sub> filter samples. KW and YZ carried out the experimental work and data analysis.

- 573 KW wrote the manuscript. KW, TH, RJH, M. Brüggemann, YZ, JH, M. Bilde and MG interpreted
- 574 data and edited the manuscript. All authors commented on and discussed the manuscript.
- **Competing interests.** The authors declare that they have no conflict of interest.

Acknowledgements. This study was supported by the National Natural Science Foundation of China (NSFC, Grant No. 41925015, No. 91644219 and No. 41877408), the Chinese Academy of Sciences (No. ZDBS-LY-DQC001), the National Key Research and Development Program of China (No. 2017YFC0212701), and the German Research Foundation (Deutsche Forschungsgemeinschaft, DFG) under Grant No. INST 247/664-1 FUGG. K. Wang and Y. Zhang acknowledge the scholarship from Chinese Scholarship Council (CSC) and Max Plank Graduate Center with Johannes Gutenberg University of Mainz (MPGC) and thanks Prof. Ulrich Pöschl, Dr. Christopher J. Kampf and Dr. Yafang Cheng for their helpful suggestion on this study. K. Wang also thanks Dr. Huanfeng Dong from Zhejiang University for the great support on the programming of data process.

### References

- An, Z., Huang, R. J., Zhang, R., Tie, X., Li, G., Cao, J., Zhou, W., Shi, Z., Han, Y., Gu, Z., and Ji, Y.: Severe haze in northern China: A synergy of anthropogenic emissions and atmospheric processes, Proc Natl Acad Sci U S A, 116, 8657-8666, 10.1073/pnas.1900125116, 2019.
- Brüggemann, M., Poulain, L., Held, A., Stelzer, T., Zuth, C., Richters, S., Mutzel, A., van Pinxteren, D.,
  Iinuma, Y., Katkevica, S., Rabe, R., Herrmann, H., and Hoffmann, T.: Real-time detection of highly
  oxidized organosulfates and BSOA marker compounds during the F-BEACh 2014 field study, Atmos.
  Chem. Phys., 17, 1453-1469, 10.5194/acp-17-1453-2017, 2017.
- Brüggemann, M., van Pinxteren, D., Wang, Y., Yu, J. Z., and Herrmann, H.: Quantification of known and unknown terpenoid organosulfates in PM10 using untargeted LC–HRMS/MS: contrasting summertime rural Germany and the North China Plain, Environmental Chemistry, -, <u>https://doi.org/10.1071/EN19089</u>, 2019.
- Claeys, M., Graham, B., Vas, G., Wang, W., Vermeylen, R., Pashynska, V., Cafmeyer, J., Guyon, P., Andre,
   M., Artaxo, P., and Maenhaut, W.: Formation of secondary organic aerosol through photooxidation of
   isoprene, Science, 303, 1173-1175, 10.1126/science.1092805, 2004.
- Daellenbach, K. R., Kourtchev, I., Vogel, A. L., Bruns, E. A., Jiang, J., Petäjä, T., Jaffrezo, J.-L., Aksoyoglu,
  S., Kalberer, M., Baltensperger, U., El Haddad, I., and Prévôt, A. S. H.: Impact of anthropogenic and
  biogenic sources on the seasonal variation in the molecular composition of urban organic aerosols: a field
  and laboratory study using ultra-high-resolution mass spectrometry, Atmospheric Chemistry and Physics,
  19, 5973-5991, 10.5194/acp-19-5973-2019, 2019.
- Ding, X., Zhang, Y.-Q., He, Q.-F., Yu, Q.-Q., Shen, R.-Q., Zhang, Y., Zhang, Z., Lyu, S.-J., Hu, Q.-H., Wang,
  Y.-S., Li, L.-F., Song, W., and Wang, X.-M.: Spatial and seasonal variations of secondary organic aerosol
  from terpenoids over China, J. geophy. Res.-Atoms., 121, 14661-14678, doi:10.1002/2016JD025467,
  2016.
- Elzein, A., Dunmore, R. E., Ward, M. W., Hamilton, J. F., and Lewis, A. C.: Variability of polycyclic
  aromatic hydrocarbons and their oxidative derivatives in wintertime Beijing, China, Atmospheric
  Chemistry and Physics, 19, 8741-8758, 10.5194/acp-19-8741-2019, 2019.
- Fleming, L. T., Lin, P., Laskin, A., Laskin, J., Weltman, R., Edwards, R. D., Arora, N. K., Yadav, A.,
  Meinardi, S., Blake, D. R., Pillarisetti, A., Smith, K. R., and Nizkorodov, S. A.: Molecular composition
  of particulate matter emissions from dung and brushwood burning household cookstoves in Haryana,
  India, Atmos. Chem. Phys., 18, 2461-2480, 10.5194/acp-18-2461-2018, 2018.
- Fu, P., Kawamura, K., Okuzawa, K., Aggarwal, S. G., Wang, G., Kanaya, Y., and Wang, Z.: Organic
  molecular compositions and temporal variations of summertime mountain aerosols over Mt. Tai, North
  China Plain, J. Geophy. Res., 113, 10.1029/2008jd009900, 2008.
- Glasius, M., Hansen, A. M. K., Claeys, M., Henzing, J. S., Jedynska, A. D., Kasper-Giebl, A., Kistler, M.,
  Kristensen, K., Martinsson, J., Maenhaut, W., Nøjgaard, J. K., Spindler, G., Stenström, K. E., Swietlicki,
  E., Szidat, S., Simpson, D., and Yttri, K. E.: Composition and sources of carbonaceous aerosols in
  Northern Europe during winter, Atmos. Environ., 173, 127-141, 10.1016/j.atmosenv.2017.11.005, 2018.
- Hansen, A. M. K., Kristensen, K., Nguyen, Q. T., Zare, A., Cozzi, F., Noejgaard, J. K., Skov, H., Brandt, J.,
  Christensen, J. H., Strom, J., Tunved, P., Krejci, R., and Glasius, M.: Organosulfates and organic acids in
  Arctic aerosols: speciation, annual variation and concentration levels, Atmos. Chem. Phys., 14, 7807https://doi.org/10.5194/acp-14-7807-2014, 2014.
- Hansen, A. M. K., Hong, J., Raatikainen, T., Kristensen, K., Ylisirniö, A., Virtanen, A., Petäjä, T., Glasius,
   M., and Prisle, N. L.: Hygroscopic properties and cloud condensation nuclei activation of limonenederived organosulfates and their mixtures with ammonium sulfate, Atmos. Chem. Phys., 15, 14071-14089,
   https://doi.org/10.5194/acp-15-14071-2015, 2015.
- He, X., Huang, X. H. H., Chow, K. S., Wang, Q., Zhang, T., Wu, D., and Yu, J. Z.: Abundance and Sources
  of Phthalic Acids, Benzene-Tricarboxylic Acids, and Phenolic Acids in PM2.5 at Urban and Suburban

- Sites in Southern China, ACS Earth and Space Chemistry, 2, 147-158,
  10.1021/acsearthspacechem.7b00131, 2018.
- Ho, C. S., Lam, C. W. K., Chan, M. H. M., Cheung, R. C. K., Law, L. K., Suen, M. W. M., and Tai, H. L.:
  Electrospray ionisation mass spectrometry: principles and clinical application, Clin. Biochem. Rev., 24, 10, 2003.
- Hoffmann, T., Huang, R. J., and Kalberer, M.: Atmospheric analytical chemistry, Anal. Chem., 83, 46494664, 10.1021/ac2010718, 2011.
- Huang, G., Liu, Y., Shao, M., Li, Y., Chen, Q., Zheng, Y., Wu, Z., Liu, Y., Wu, Y., Hu, M., Li, X., Lu, S.,
  Wang, C., Liu, J., Zheng, M., and Zhu, T.: Potentially Important Contribution of Gas-Phase Oxidation of
  Naphthalene and Methylnaphthalene to Secondary Organic Aerosol during Haze Events in Beijing,
  Environ Sci Technol, 53, 1235-1244, 10.1021/acs.est.8b04523, 2019.
- Huang, R. J., Zhang, Y., Bozzetti, C., Ho, K. F., Cao, J. J., Han, Y., Daellenbach, K. R., Slowik, J. G., Platt,
  S. M., Canonaco, F., Zotter, P., Wolf, R., Pieber, S. M., Bruns, E. A., Crippa, M., Ciarelli, G., Piazzalunga,
  A., Schwikowski, M., Abbaszade, G., Schnelle-Kreis, J., Zimmermann, R., An, Z., Szidat, S.,
  Baltensperger, U., El Haddad, I., and Prevot, A. S.: High secondary aerosol contribution to particulate
  pollution during haze events in China, Nature, 514, 218-222, 10.1038/nature13774, 2014.
- Huang, R. J., Cao, J. J., and Worsnop, D.: Sources and Chemical Composition of Particulate Matter During
  Haze Pollution Events in China, in: Air pollution in Eastern Asia: an integrated perspective, edited by
  Bouarar, I., Wang, X. M., and Brasseur, G. P., Springer, Cham, Switzerland, 49-68, 2017.
- 692 Iinuma, Y., Müller, C., Berndt, T., Böge, O., Claeys, M., and Herrmann, H.: Evidence for the existence of
  693 organosulfates from β-pinene ozonolysis in ambient secondary organic aerosol, Environ. Sci. Technol.,
  694 41, 6678-6683, 10.1021/es070938t, 2007.
- Jung, J., and Kawamura, K.: Enhanced concentrations of citric acid in spring aerosols collected at the Gosan
   background site in East Asia, Atmos. Environ., 45, 5266-5272, 10.1016/j.atmosenv.2011.06.065, 2011.
- Kautzman, K. E., Surratt, J. D., Chan, M. N., Chan, A. W., Hersey, S. P., Chhabra, P. S., Dalleska, N. F.,
  Wennberg, P. O., Flagan, R. C., and Seinfeld, J. H.: Chemical composition of gas- and aerosol-phase
  products from photooxidation of naphthalene, J. Phys. Chem. A, 114, 913-934, 10.1021/jp908530s, 2010.
- Kind, T., and Fiehn, O.: Seven Golden Rules for heuristic filtering of molecular formulas obtained by
   accurate mass spectrometry, BMC Bioinformatics, 8, 10.1186/1471-2105-8-105, 2007.
- Kourtchev, I., O'Connor, I. P., Giorio, C., Fuller, S. J., Kristensen, K., Maenhaut, W., Wenger, J. C., Sodeau,
  J. R., Glasius, M., and Kalberer, M.: Effects of anthropogenic emissions on the molecular composition of urban organic aerosols: An ultrahigh resolution mass spectrometry study, Atmo. Environ., 89, 525-532,
  10.1016/j.atmosenv.2014.02.051, 2014.
- Kourtchev, I., Godoi, R. H. M., Connors, S., Levine, J. G., Archibald, A. T., Godoi, A. F. L., Paralovo, S. L.,
  Barbosa, C. G. G., Souza, R. A. F., Manzi, A. O., Seco, R., Sjostedt, S., Park, J.-H., Guenther, A., Kim,
  S., Smith, J., Martin, S. T., and Kalberer, M.: Molecular composition of organic aerosols in central
  Amazonia: an ultra-high-resolution mass spectrometry study, Atmos. Chem. Phys., 16, 11899-11913,
  https://doi.org/10.5194/acp-16-11899-2016, 2016.
- Kruve, A., Kaupmees, K., Liigand, J., and Leito, I.: Negative electrospray ionization via deprotonation:
   predicting the ionization efficiency, Anal Chem, 86, 4822-4830, 10.1021/ac404066v, 2014.
- Laskin, A., Laskin, J., and Nizkorodov, S. A.: Chemistry of atmospheric brown carbon, Chem. Rev., 115,
  4335-4382, 10.1021/cr5006167, 2015.
- Laskin, J., Laskin, A., Roach, P. J., Slysz, G. W., Anderson, G. A., Nizkorodov, S. A., Bones, D. L., and
  Nguyen, L. Q.: High-Resolution Desorption Electrospray Ionization Mass Spectrometry for Chemical
  Characterization of Organic Aerosols, Anal. Chem., 82, 2048-2058, 10.1021/ac902801f, 2010.
- Laskin, J., Laskin, A., and Nizkorodov, S. A.: Mass Spectrometry Analysis in Atmospheric Chemistry, Anal.
   Chem., 90, 166-189, 10.1021/acs.analchem.7b04249, 2018.
- Lee, A., Goldstein, A. H., Kroll, J. H., Ng, N. L., Varutbangkul, V., Flagan, R. C., and Seinfeld, J. H.: Gas phase products and secondary aerosol yields from the photooxidation of 16 different terpenes, J. Geophy.
   Res., 111, 10.1029/2006jd007050, 2006.
- Leito, I., Herodes, K., Huopolainen, M., Virro, K., Kunnapas, A., Kruve, A., and Tanner, R.: Towards the
   electrospray ionization mass spectrometry ionization efficiency scale of organic compounds, Rapid
   Commun Mass Spectrom, 22, 379-384, 10.1002/rcm.3371, 2008.
- Li, Y. J., Huang, D. D., Cheung, H. Y., Lee, A. K. Y., and Chan, C. K.: Aqueous-phase photochemical oxidation and direct photolysis of vanillin – a model compound of methoxy phenols from biomass burning,

- 728 Atmospheric Chemistry and Physics, 14, 2871-2885, 10.5194/acp-14-2871-2014, 2014.
- Lim, Y. B., Tan, Y., Perri, M. J., Seitzinger, S. P., and Turpin, B. J.: Aqueous chemistry and its role in secondary organic aerosol (SOA) formation, Atmos. Chem. Phys., 10, 10521-10539, 10.5194/acp-10-10521-2010, 2010.
- Lin, P., Rincon, A. G., Kalberer, M., and Yu, J. Z.: Elemental composition of HULIS in the Pearl River Delta
   Region, China: results inferred from positive and negative electrospray high resolution mass
   spectrometric data, Environ. Sci. Technol., 46, 7454-7462, 10.1021/es300285d, 2012a.
- Lin, P., Yu, J. Z., Engling, G., and Kalberer, M.: Organosulfates in humic-like substance fraction isolated
  from aerosols at seven locations in East Asia: a study by ultra-high-resolution mass spectrometry, Environ.
  Sci. Technol., 46, 13118-13127, 10.1021/es303570v, 2012b.
- Lin, P., Laskin, J., Nizkorodov, S. A., and Laskin, A.: Revealing Brown Carbon Chromophores Produced in
   Reactions of Methylglyoxal with Ammonium Sulfate, Environ. Sci. Technol., 49, 14257-14266,
   10.1021/acs.est.5b03608, 2015.
- McNeill, V. F., Woo, J. L., Kim, D. D., Schwier, A. N., Wannell, N. J., Sumner, A. J., and Barakat, J. M.:
  Aqueous-phase secondary organic aerosol and organosulfate formation in atmospheric aerosols: a
  modeling study, Environ Sci Technol, 46, 8075-8081, 10.1021/es3002986, 2012.
- Ning, C., Gao, Y., Zhang, H., Yu, H., Wang, L., Geng, N., Cao, R., and Chen, J.: Molecular characterization
  of dissolved organic matters in winter atmospheric fine particulate matters (PM2.5) from a coastal city of
  northeast China, Sci Total Environ, 689, 312-321, 10.1016/j.scitotenv.2019.06.418, 2019.
- Nizkorodov, S. A., Laskin, J., and Laskin, A.: Molecular chemistry of organic aerosols through the application of high resolution mass spectrometry, Phys. Chem. Chem. Phys., 13, 3612-3629, 10.1039/c0cp02032j, 2011.
- Noziere, B., Kalberer, M., Claeys, M., Allan, J., D'Anna, B., Decesari, S., Finessi, E., Glasius, M., Grgic, I.,
  Hamilton, J. F., Hoffmann, T., Iinuma, Y., Jaoui, M., Kahnt, A., Kampf, C. J., Kourtchev, I., Maenhaut,
  W., Marsden, N., Saarikoski, S., Schnelle-Kreis, J., Surratt, J. D., Szidat, S., Szmigielski, R., and
  Wisthaler, A.: The molecular identification of organic compounds in the atmosphere: state of the art and
  challenges, Chem. Rev., 115, 3919-3983, 10.1021/cr5003485, 2015.
- Pereira, K. L., Hamilton, J. F., Rickard, A. R., Bloss, W. J., Alam, M. S., Camredon, M., Muñoz, A., Vázquez,
  M., Borrás, E., and Ródenas, M.: Secondary organic aerosol formation and composition from the photooxidation of methyl chavicol (estragole), Atmos. Chem. Phys., 14, 5349-5368, 10.5194/acp-14-53492014, 2014.
- Perry, R. H., Cooks, R. G., and Noll, R. J.: ORBITRAP MASS SPECTROMETRY: INSTRUMENTATION,
   ION MOTION AND APPLICATIONS, Mass Spectrometry Reviews, 27, 661-699, 10.1002/mas.20186,
   2008.
- Rincón, A. G., Calvo, A. I., Dietzel, M., and Kalberer, M.: Seasonal differences of urban organic aerosol
  composition an ultra-high resolution mass spectrometry study, Environ. Chem., 9, 298,
  10.1071/en12016, 2012.
- Riva, M., Tomaz, S., Cui, T., Lin, Y.-H., Perraudin, E., Gold, A., Stone, E. A., Villenave, E., and Surratt, J.
  D.: Evidence for an Unrecognized Secondary Anthropogenic Source of Organosulfates and Sulfonates:
  Gas-Phase Oxidation of Polycyclic Aromatic Hydrocarbons in the Presence of Sulfate Aerosol, Environ.
  Sci. Technol., 49, 6654-6664, 10.1021/acs.est.5b00836, 2015.
- Schmidt, A. C., Herzschuh, R., Matysik, F. M., and Engewald, W.: Investigation of the ionisation and
  fragmentation behaviour of different nitroaromatic compounds occurring as polar metabolites of
  explosives using electrospray ionisation tandem mass spectrometry, Rapid Commun Mass Spectrom, 20,
  2293-2302, 10.1002/rcm.2591, 2006.
- Shi, Z., Vu, T., Kotthaus, S., Harrison, R. M., Grimmond, S., Yue, S., Zhu, T., Lee, J., Han, Y., Demuzere,
  M., Dunmore, R. E., Ren, L., Liu, D., Wang, Y., Wild, O., Allan, J., Acton, W. J., Barlow, J., Barratt, B.,
  Beddows, D., Bloss, W. J., Calzolai, G., Carruthers, D., Carslaw, D. C., Chan, Q., Chatzidiakou, L., Chen,
  Y., Crilley, L., Coe, H., Dai, T., Doherty, R., Duan, F., Fu, P., Ge, B., Ge, M., Guan, D., Hamilton, J. F.,
  He, K., Heal, M., Heard, D., Hewitt, C. N., Hollaway, M., Hu, M., Ji, D., Jiang, X., Jones, R., Kalberer,
- 778 M., Kelly, F. J., Kramer, L., Langford, B., Lin, C., Lewis, A. C., Li, J., Li, W., Liu, H., Liu, J., Loh, M.,
- Lu, K., Lucarelli, F., Mann, G., McFiggans, G., Miller, M. R., Mills, G., Monk, P., Nemitz, E., amp, apos,
- 780 Connor, F., Ouyang, B., Palmer, P. I., Percival, C., Popoola, O., Reeves, C., Rickard, A. R., Shao, L., Shi,
- 781 G., Spracklen, D., Stevenson, D., Sun, Y., Sun, Z., Tao, S., Tong, S., Wang, Q., Wang, W., Wang, X.,
- 782 Wang, X., Wang, Z., Wei, L., Whalley, L., Wu, X., Wu, Z., Xie, P., Yang, F., Zhang, Q., Zhang, Y.,

- Zhang, Y., and Zheng, M.: Introduction to the special issue "In-depth study of air pollution sources and
  processes within Beijing and its surrounding region (APHH-Beijing)", Atmospheric Chemistry and
  Physics, 19, 7519-7546, 10.5194/acp-19-7519-2019, 2019.
- Song, J., Li, M., Jiang, B., Wei, S., Fan, X., and Peng, P.: Molecular Characterization of Water-Soluble
  Humic like Substances in Smoke Particles Emitted from Combustion of Biomass Materials and Coal
  Using Ultrahigh-Resolution Electrospray Ionization Fourier Transform Ion Cyclotron Resonance Mass
  Spectrometry, Environ. Sci. Technol., 52, 2575-2585, 10.1021/acs.est.7b06126, 2018.
- Sun, Y., Jiang, Q., Zhang, Z., Fu, P., Li, J., Yang, T., and Yin, Y.: Investigation of the sources and evolution processes of severe haze pollution in Beijing in January 2013, J. Geophy. Res.-Atmos., 119, 4380-4389, 10.1002/, 2014.
- Surratt, J. D., Gomez-Gonzalez, Y., Chan, A. W., Vermeylen, R., Shahgholl, M., Kleindienst, T. E., Jaoui,
  M., Maenhaut, W., Claeys, M., Flagan, R. C., and Seinfeld, J. H.: Evidence for Organosulfate in
  Secondary Organic Aerosol, Environ. Sci. Technol., 41, 517-527, 10.1021/es062081q, 2007.
- Surratt, J. D., Gómez-González, Y., Chan, A. W., Vermeylen, R., Shahgholi, M., Kleindienst, T. E., Edney,
  E. O., Offenberg, J. H., Lewandowski, M., Jaoui, M., Maenhaut, W., Claeys, M., Flagan, R. C., and
  Seinfeld, J. H.: Organosulfate Formation in Biogenic Secondary Organic Aerosol, J. Phys. Chem. A, 112,
  8345-8378, 2008.
- Tao, S., Lu, X., Levac, N., Bateman, A. P., Nguyen, T. B., Bones, D. L., Nizkorodov, S. A., Laskin, J., Laskin,
   A., and Yang, X.: Molecular Characterization of Organosulfates in Organic Aerosols from Shanghai and
   Los Angeles Urban Areas by Nanospray-Desorption Electrospray Ionization High-Resolution Mass
   Spectrometry, Environ. Sci. Technol., 48, 10993-11001, 10.1021/es5024674, 2014.
- Tong, H., Kourtchev, I., Pant, P., Keyte, I. J., O'Connor, I. P., Wenger, J. C., Pope, F. D., Harrison, R. M.,
  and Kalberer, M.: Molecular composition of organic aerosols at urban background and road tunnel sites
  using ultra-high resolution mass spectrometry, Faraday Discuss., 189, 51-68, 10.1039/c5fd00206k, 2016.
- Tong, H., Zhang, Y., Filippi, A., Wang, T., Li, C., Liu, F., Leppla, D., Kourtchev, I., Wang, K., Keskinen, H.
  M., Levula, J. T., Arangio, A. M., Shen, F., Ditas, F., Martin, S. T., Artaxo, P., Godoi, R. H. M.,
  Yamamoto, C. I., de Souza, R. A. F., Huang, R. J., Berkemeier, T., Wang, Y., Su, H., Cheng, Y., Pope,
  F. D., Fu, P., Yao, M., Pohlker, C., Petaja, T., Kulmala, M., Andreae, M. O., Shiraiwa, M., Poschl, U.,
  Hoffmann, T., and Kalberer, M.: Radical Formation by Fine Particulate Matter Associated with Highly
  Oxygenated Molecules, Environ Sci Technol, 53, 12506-12518, 10.1021/acs.est.9b05149, 2019.
- Tu, P., Hall, W. A. t., and Johnston, M. V.: Characterization of Highly Oxidized Molecules in Fresh and
  Aged Biogenic Secondary Organic Aerosol, Anal. Chem., 88, 4495-4501,
  10.1021/acs.analchem.6b00378, 2016.
- Wang, G., Kawamura, K., Umemoto, N., Xie, M., Hu, S., and Wang, Z.: Water-soluble organic compounds
  in PM2.5and size-segregated aerosols over Mount Tai in North China Plain, J. Geophy. Res., 114,
  10.1029/2008jd011390, 2009.
- Wang, K., Zhang, Y., Huang, R.-J., Cao, J., and Hoffmann, T.: UHPLC-Orbitrap mass spectrometric characterization of organic aerosol from a central European city (Mainz, Germany) and a Chinese megacity (Beijing), Atmos. Environ., 189, 22-29, 10.1016/j.atmosenv.2018.06.036, 2018.
- 822 Wang, K., Zhang, Y., Huang, R.-J., Wang, M., Ni, H., Kampf, C. J., Cheng, Y., Bilde, M., Glasius, M., and 823 Hoffmann, T.: Molecular characterization and source identification of atmospheric particulate 824 using ultrahigh resolution mass organosulfates spectrometry, Environ. Sci. Technol., 825 10.1021/acs.est.9b02628, 2019a.
- Wang, M., Huang, R.-J., Cao, J., Dai, W., Zhou, J., Lin, C., Ni, H., Duan, J., Wang, T., Chen, Y., Li, Y.,
  Chen, Q., Haddad, I. E., and Hoffmann, T.: Determination of n-alkanes, PAHs and hopanes in
  atmospheric aerosol: evaluation and comparison of thermal desorption GC-MS and solvent extraction
  GC-MS approaches, Atmos. Mea. Tech. Discuss., 1-21, 10.5194/amt-2019-4, 2019b.
- Wang, X. K., Rossignol, S., Ma, Y., Yao, L., Wang, M. Y., Chen, J. M., George, C., and Wang, L.: Molecular
  characterization of atmospheric particulate organosulfates in three megacities at the middle and lower
  reaches of the Yangtze River, Atmos. Chem. Phys., 16, 2285-2298, <u>https://doi.org/10.5194/acp-16-2285-2016</u>, 2016.
- Wang, X. K., Hayeck, N., Brüggemann, M., Yao, L., Chen, H. F., Zhang, C., Emmelin, C., Chen, J. M.,
  George, C., and Wang, L.: Chemical characterization of organic aerosol in: A study by UltrahighPerformance Liquid Chromatography Coupled with Orbitrap Mass Spectrometry, J. Geophy. Res.-Atmos.,
  122, 703-722, https://doi.org/10.1002/2017JD026930, 2017.
  - 23

- Xu, W., Sun, Y., Wang, Q., Zhao, J., Wang, J., Ge, X., Xie, C., Zhou, W., Du, W., Li, J., Fu, P., Wang, Z.,
  Worsnop, D. R., and Coe, H.: Changes in Aerosol Chemistry From 2014 to 2016 in Winter in Beijing:
  Insights From High-Resolution Aerosol Mass Spectrometry, J. Geophy. Res.-Atmos., 124, 1132-1147,
  10.1029/2018jd029245, 2019.
- Yassine, M. M., Harir, M., Dabek-Zlotorzynska, E., and Schmitt-Kopplin, P.: Structural characterization of
  organic aerosol using Fourier transform ion cyclotron resonance mass spectrometry: aromaticity
  equivalent approach, Rapid Commun. Mass Spectrom., 28, 2445-2454, 10.1002/rcm.7038, 2014.
- Zhang, P.: Revitalizing old industrial base of Northeast China: Process, policy and challenge, Chin. Geogra.
  Sci., 18, 109-118, 10.1007/s11769-008-0109-2, 2008.
- Zielinski, A. T., Kourtchev, I., Bortolini, C., Fuller, S. J., Giorio, C., Popoola, O. A. M., Bogialli, S., Tapparo,
  A., Jones, R. L., and Kalberer, M.: A new processing scheme for ultra-high resolution direct infusion
- 849 mass spectrometry data, Atmos. Environ., 178, 129-139, 10.1016/j.atmosenv.2018.01.034, 2018.

873

874

Table 1. Number of organic compounds and molecular formulas in each subgroup and the peak abundance-weighted average values of molecular mass ( $MM_{avg}$ ), elemental ratios, double bond equivalent (DBE), aromaticity equivalent (Xc) and isomer number fraction (meaning the percentage of formula numbers that have isomers among all assigned formulas) for detected organic compounds in ESI– and ESI+ in the three Chinese cities.

	0.1	Number of	Relative abundance		ШC	0/0**	DDE	V	Isomer number fraction
Sample ID	Subgroup	compounds <sup>*</sup>	(%)	MM <sub>avg</sub>	H/C	O/C**	DBE	Xc	(%) 34
Changchun-	total	769(415)	100	169	1.03	0.58	5.02	2.13	54 52
	CHO-	346(136)	30	162	0.96	0.41	5.65	2.28	36
	CHON-	180(96)	55	163	0.94	0.51	5.24	2.44	28
	CHOS-	155(105)	10	198	1.56	1.17(0.52)	2.55	0.50	8
<u>a</u> , , ,	CHONS-	88(78)	5	214	1.35	1.07(-1.4)	3.75	1.06	31
Shanghai-	total	416(272)	100	176	1.05	0.69	4.99	1.92	41
	CHO-	164(90)	40	171	0.97	0.59	5.37	1.94	41 37
	CHON-	135(89)	44	169	0.86	0.56	5.67	2.47	15
	CHOS-	75(62)	12	190	1.85	1.41(0.61)	1.79	0.34	13
	CHONS-	42(31)	4	266	1.56	1.00(0.11)	3.30	0.44	
Guangzhou-	total	488(304)	100	183	1.14	0.74	4.55	1.65	34
	CHO-	196(110)	42	172	1.10	0.65	4.68	1.57	44
	CHON-	161(98)	39	173	0.89	0.58	5.56	2.41	35
	CHOS-	86(67)	14	201	1.85	1.48(0.71)	1.71	0.21	21
	CHONS-	45(29)	5	293	1.56	0.82(0.06)	3.45	0.43	28
Changchun+	total	2943(679)	100	160	1.21	0.13	5.58	2.36	56
	CHO+	609(162)	13	174	0.94	0.28	6.55	2.22	50
	CHN+	696(126)	40	154	1.22	0.00	5.84	2.60	77
	CHON+	1594(352)	46.5	161	1.27	0.19	5.11	2.22	55
	CHONS+	44(39)	0.5	196	1.91	0.70	2.64	0.09	13
Shanghai+	total	704(383)	100	162	1.37	0.09	4.91	2.32	32
	CHO+	87(67)	4	184	1.13	0.43	5.46	1.46	19
	CHN+	253(84)	71	159	1.38	0.00	5.08	2.55	54
	CHON+	350(218)	24.7	167	1.40	0.27	4.34	1.81	30
	CHONS+	14(14)	0.3	241	1.17	0.61	5.32	0.91	0
Guangzhou+	total	687(412)	100	161	1.41	0.17	4.58	2.07	30
-	CHO+	125(87)	8	185	1.12	0.42	5.19	1.20	26
	CHN+	205(78)	62	156	1.42	0.00	4.80	2.47	54
	CHON+	336(227)	29	165	1.47	0.45	4.00	1.51	26

	CHONS+	21(20)	1	209	1.84	0.71	3.05	0.31	5
381	*The values in brackets	indicate the number	of unique	molecular	formulas.	**The values	in brackets	indicate the	

\*The values in brackets indicate the number of unique molecular formulas. \*\*The values in brackets in
 (O-3S)/C and (O-3S-2N)/C ratios for CHOS and CHONS compounds, respectively, detected in ESI- mode

883

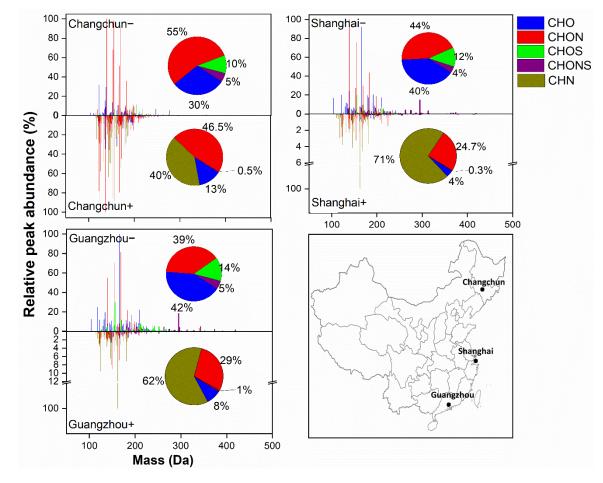
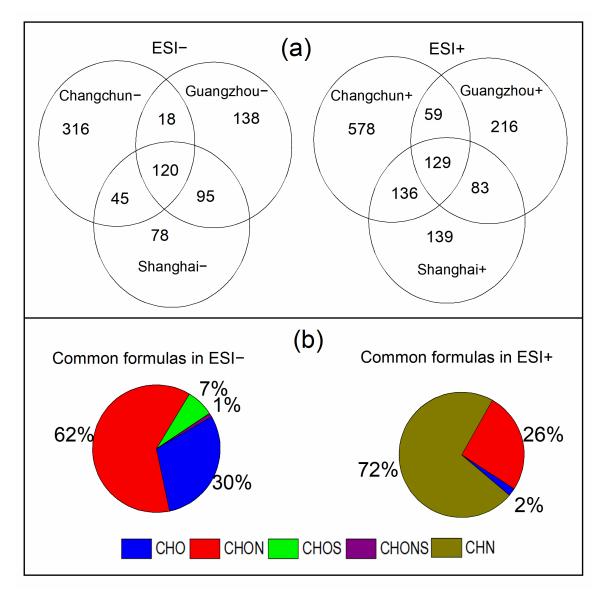


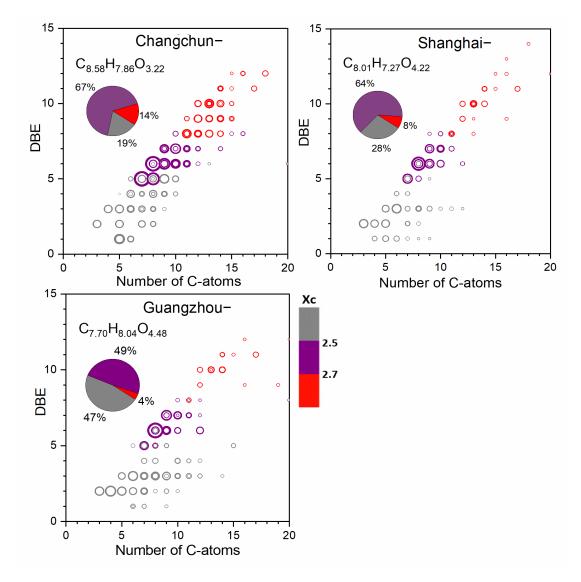
Figure 1. Mass spectra of detected organic compounds reconstructed from extracted ion chromatograms in ESI– and ESI+. The horizontal axis refers to the molecular mass (Da) of the identified species. The vertical axis refers to the relative peak abundance of each individual compound to the compound with the greatest peak abundance. The pie charts show the percentage of each organic compound subgroup (i.e. CHO, CHON, CHOS, CHONS and CHN) in each sample in terms of peak abundance. The map in the lower right corner shows the locations of these three megacities in China.

892



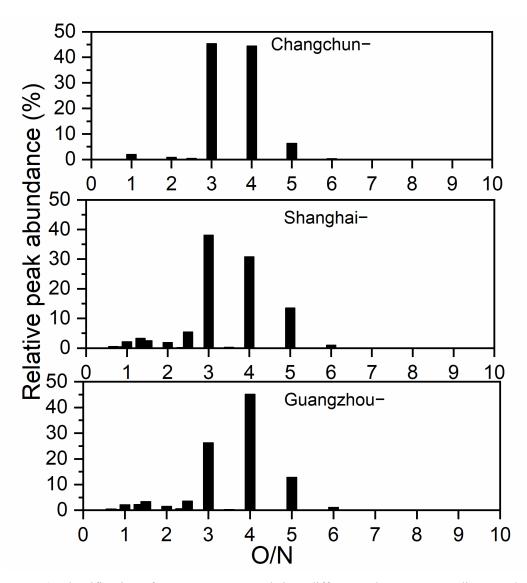
893

Figure 2. (a) Venn diagrams showing the number distribution of all molecular formulas detected in ESI- and ESI+ for all sample locations. The overlapping molecular formulas refer to the compounds detected in each city with the same molecular formulas and with the same retention times (retention time difference  $\leq 0.1$  min). (b) Peak abundance contribution of each elemental formula category to the total common formulas.

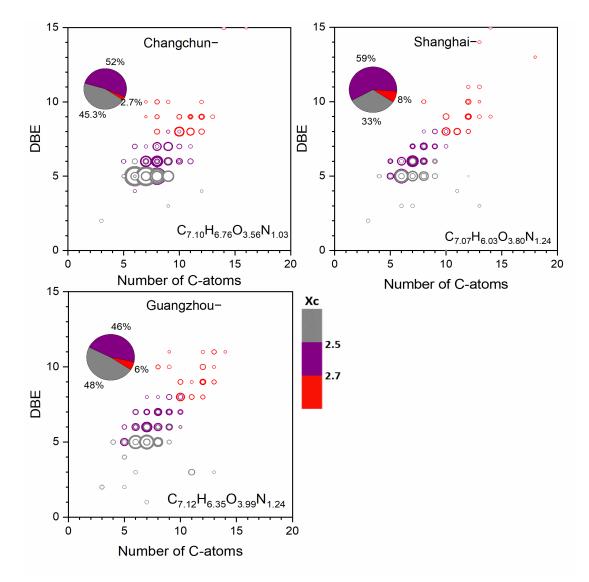


899

900 Figure 3. Double bond equivalent (DBE) versus carbon number for all CHO- compounds for all 901 sample locations. The molecular formula represents the abundance-weighted average CHO-902 formula and the area of the circles is proportional to the fourth root of the peak abundance of an 903 individual compound (a diagram with circle areas related to the absolute peak abundances is 904 presented in Fig. S2). The color bar denotes the aromaticity equivalent (gray with Xc < 2.50, purple 905 with  $2.50 \le Xc < 2.70$  and red with  $Xc \ge 2.70$ ). The pie charts show the percentage of each Xc 906 category (i.e., gray color-coded compounds, purple color-coded compounds and red color-coded 907 compounds) in each sample in terms of peak abundance.

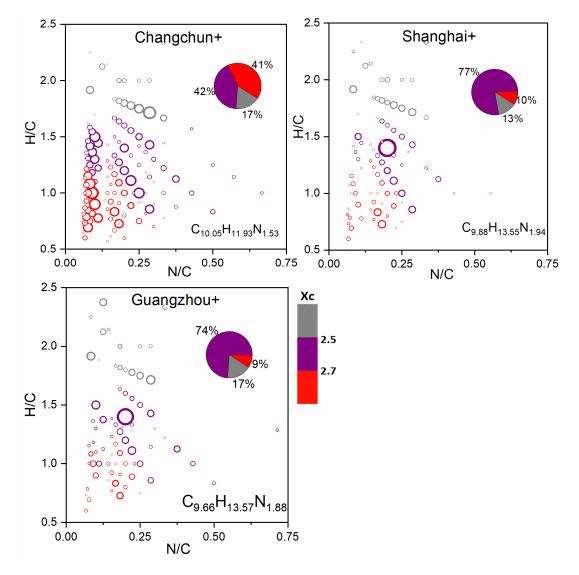


909 Figure 4. Classification of CHON- compounds into different subgroups according to O/N ratios in
910 their formulas. The y-axis indicates the relative contribution of each specific O/N ratio subgroup to
911 the sum of peak abundances of CHON- compounds.



912

913 Figure 5. Double bond equivalent (DBE) versus carbon number for all CHON- compounds for all 914 sample locations. The molecular formula represents the abundance-weighted average CHON-915 formula and the area of circles is proportional to the fourth root of the peak abundance of an individual compound (a diagram with circle areas related to absolute peak abundances is presented 916 917 in Fig. S6). The color bar denotes the aromaticity equivalent (gray with Xc < 2.50, purple with 2.50  $\leq$  Xc < 2.70 and red with Xc  $\geq$  2.70). The pie charts show the percentage of each Xc category (i.e., 918 919 gray color-coded compounds, purple color-coded compounds and red color-coded compounds) in 920 each sample in terms of peak abundance.



922Figure 6. Van Krevelen diagrams for CHN+ compounds in Changchun, Shanghai and Guangzhou923samples. The area of circles is proportional to the fourth root of the peak abundance of an individual924compound (a diagram with circle areas related to absolute peak abundances is presented in Fig.925S10) and the color bar denotes the aromaticity equivalent (gray with Xc < 2.50, purple with  $2.50 \le$ 926Xc < 2.70 and red with Xc  $\ge$  2.70). The pie charts show the percentage of each Xc category (i.e.,927gray color-coded compounds, purple color-coded compounds and red color-coded compounds) in928each sample in terms of peak abundance.

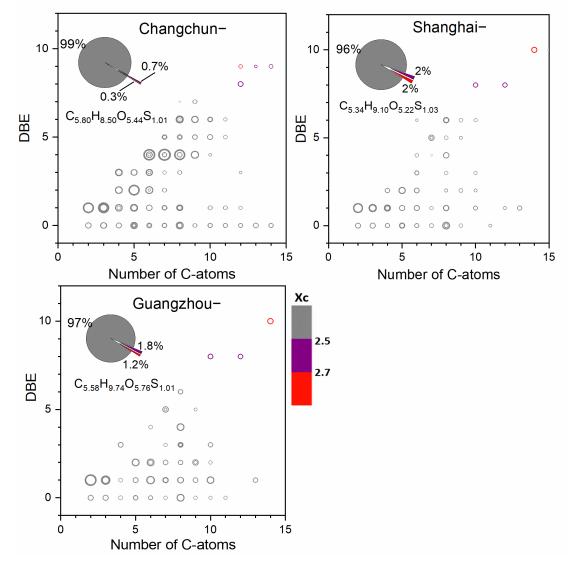


Figure 7. Double bond equivalent (DBE) versus carbon number for all CHOS- compounds for all sample locations. The molecular formula represents the abundance-weighted average CHOS-formula and the area of circles is proportional to the fourth root of the peak abundance of an individual compound (a diagram with circle areas related to absolute peak abundances is presented in Fig. S11). The color bar denotes the aromaticity equivalent (gray with Xc < 2.50, purple with  $2.50 \le Xc < 2.70$  and red with  $Xc \ge 2.70$ ). The pie charts show the percentage of each Xc category (i.e., gray color-coded compounds, purple color-coded compounds and red color-coded compounds) in each sample in terms of peak abundance.