

Author's response by Wenjie Wang et al.

Corresponding to mshao@pku.edu.cn.

We greatly appreciate the time and efforts that the Referees spent in reviewing our manuscript. The comments are really thoughtful and helpful to improve the quality of our paper. We have addressed each comment below, with the Referee comment in black text, our response in blue text, and relevant manuscript changes noted in red text.

In this work you have used in situ measurements for the retrieval of SSA at 525nm.

It would be much more appropriate to use aeronet based SSA since:

- they represent the aerosol column that is what you need for the particular application
- they are more representative for the whole time period.

Response: Many thanks. I agree with you that it would be much more appropriate to use aeronet based SSA. I have added it in Figure 6.

Line 209-210:

AE and SSA (440nm) were also acquired from ARONET.

Line 383-387:

Figure 6 (a) presents the relationship between AERONET based SSA (440nm) and AOD during 2012-2015. The result suggests a significant positive correlation between SSA and AOD. Additionally, Figure 6 (b) also presents a significant positive correlation between near-ground SSA (525nm) from the measurement campaign in August 2012 and AOD.

Line 1200-1201:

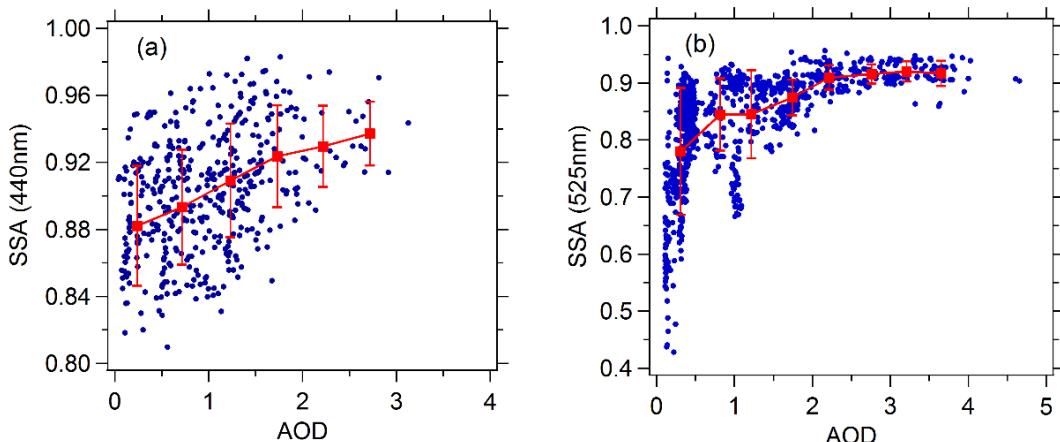


Figure 6. Correlation between SSA and AOD (380nm). (a) SSA (440nm) is acquired from AERONET during 2012-2015. (b) SSA (525nm) is acquired from near-ground measurement campaign in August 2012.

So what the reviewer is stating is to provide

- what is the SSA used from the in situ at 525nm.
- what is the difference with the aeronet based in situ at 440nm
- What is the effect of the above difference in the Photolysis Rates (PFs) ?

Mention also the possible uncertainty on the spectral dependence of the SSA as either it is used from 525nm nor from 440nm they both differ from the 305-315nm range of interest, here.

Response:

(1) For the SSA used from the in situ at 525nm, the absorption and scattering coefficients were measured with an Aethalometer (AE-31, Magee) and a Single Wavelength Integrating Nephelometer (Aurora-1000), respectively, with a time resolution of 1 minute. As aerosol particles were dried by decreasing RH to <40% when SSA was measured, we used the measured hygroscopic factor (Liu et al., 2009) and measured RH to correct the SSA.

(2) The mean near-ground SSA (525nm) in August 2012 (0.88 ± 0.08) is significantly lower than the mean AERONET based SSA (440nm) in the same period and in summer (0.94 ± 0.02). The different wavelength plays a minor role in the different SSA according to the wavelength dependence of AERONET based SSA (Figure S3). This difference is possibly due to the uncertainty of AERONET based SSA and the poor vertical representative of near-ground SSA.

(3) The effect of the difference in SSA (0.88 vs 0.94) results in photolysis frequencies changing by 11%-16% according to TUV model. The AERONET based SSA generally reproduces well the slope of $j(O^1D)$ versus AOD in spite that it significantly underestimates the absolute value of the slope at low AOD range (AOD<0.7), which is probably due to the uncertainty of AERONET based SSA in low AOD range.

(4) Thank you. I have mentioned the possible uncertainty on the spectral dependence of the SSA by adding the sentence “In addition to the uncertainty of SSA, both of SSA at 440nm and at 525nm differ from the 305-315nm range of $j(O^1D)$, which is likely to lead to some uncertainties for the analysis of the relationship between $j(O^1D)$ and AOD.”

Line 209-216: In addition, the SSA (525nm) data were derived from a field campaign

undertaken in August 2012. The absorption and scattering coefficients were measured with an Aethalometer (AE-31, Magee) and a Single Wavelength Integrating Nephelometer (Aurora-1000), respectively, with a time resolution of 1 minute. As aerosol particles were dried by decreasing relative humidity (RH) to <40% when SSA was measured, we used the measured hygroscopic factor (Liu et al., 2009) and measured RH to correct the SSA.

Line 409-421: It worth noting that the mean near-ground SSA (525nm) in August 2012 (0.88 ± 0.08) is significantly lower than the mean AERONET based SSA (440nm) in the same period (0.94 ± 0.02) and in summer (0.94 ± 0.02). The different wavelength plays a minor role in the different SSA according to the wavelength dependence of AERONET based SSA (Figure S3). This difference is possibly due to the uncertainty of AERONET based SSA and the poor vertical representative of near-ground SSA. The AERONET based SSA generally reproduces well the slope of $j(O^1D)$ versus AOD in spite that it significantly underestimates the absolute value of the slope at low AOD range ($AOD < 0.7$), which is probably due to the uncertainty of AERONET based SSA in low AOD range. In addition to the uncertainty of SSA, both of SSA at 440nm and at 525nm differ from the 305-315nm range of $j(O^1D)$, which is likely to lead to some uncertainties for the analysis of the relationship between $j(O^1D)$ and AOD.

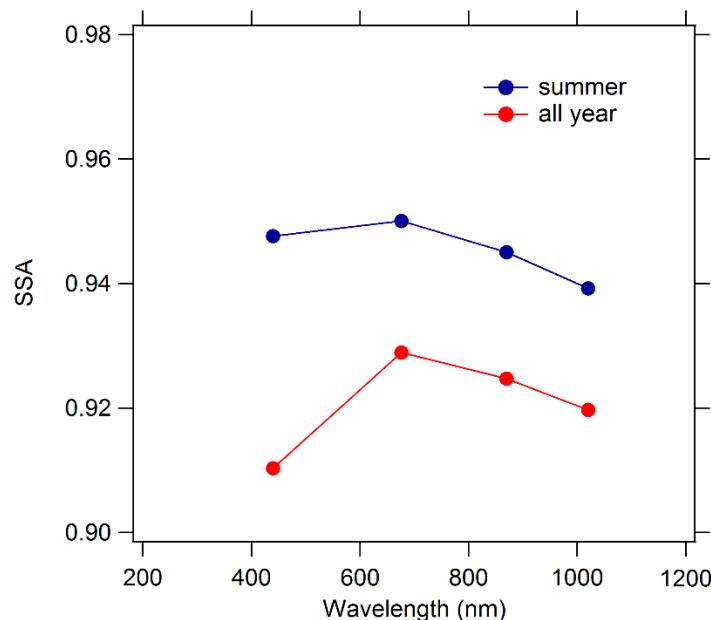


Figure S3. The dependence of AERONET based SSA on wavelength.

Additionally, I make other revisions in the manuscript.

Line 393-395: ~~However, due to absence of more SSA data of the period 2012–2015, we can't give more sufficient evidence for the dependence of SSA on AOD.~~

Line 399-402: Table S1 indicates that SSA in summer is higher significantly than in winter. High AOD levels often appeared in summer and low AOD levels occurred mostly in winter (Figure 2 and Table S1), another fact that may also explains the rapidly reduced slope of $j(O^1D)$ vs AOD with AOD.

Table S1. The seasonal mean AOD and SSA of AERONET.

Seasons	Spring	Summer	Autumn	Winter
AOD	0.83 ± 0.72	0.99 ± 0.90	0.59 ± 0.70	0.47 ± 0.47
SSA	0.91 ± 0.03	0.94 ± 0.02	0.91 ± 0.03	0.88 ± 0.03

1 **The impact of aerosols on photolysis frequencies and ozone**
2 **production in urban Beijing during the four-year period**
3 **2012–2015**

4 Wenjie Wang¹, Min Shao^{1,2*}, Xin Li^{1*}, Min Hu¹, Limin Zeng¹, Yusheng Wu¹, Tianyi
5 Tan¹

6

7 1 State Joint Key Laboratory of Environmental Simulation and Pollution Control,
8 College of Environmental Sciences and Engineering, Peking University, Beijing
9 100871, China

10 2 Institute for Environmental and Climate Research, Jinan University, Guangzhou
11 511443, China

12

13

14

15

16 *** Correspondence to:**

17 Prof. Min SHAO

18 College of Environmental Sciences and Engineering, Peking University, Beijing
19 100871, China

20 Tel: +86-10-62757973; Fax: +86-10-62757973

21 Email: mshao@pku.edu.cn

22

23 **Abstract**

24 During the period 2012-2015, the photolysis frequencies were measured at the
25 Peking University site (PKUERS), a representative site of urban Beijing. We present
26 a study of the effects of aerosols on two key photolysis frequencies, $j(O^1D)$ and
27 $j(NO_2)$. Both $j(O^1D)$ and $j(NO_2)$ display significant dependence on AOD (380nm)
28 with a nonlinear negative correlation. With the increase in AOD, the slopes of
29 photolysis frequencies vs AOD decrease, which indicates that the capacity of
30 aerosols to reduce the actinic flux decreases with AOD. The absolute values of
31 slopes are equal to $4.2\text{-}6.9 \cdot 10^{-6} \text{ s}^{-1}$ and $3.2 \cdot 10^{-3} \text{ s}^{-1}$ per AOD unit for $j(O^1D)$ and
32 $j(NO_2)$ respectively at SZA of 60° and AOD smaller than 0.7, both of which are
33 larger than those observed in a similar, previous study in the Mediterranean. This
34 indicates that the aerosols in urban Beijing have a stronger extinction effect on
35 actinic flux than absorptive dust aerosols in the Mediterranean. Since the photolysis
36 frequencies strongly depended on the AOD and the solar zenith angle (SZA), we
37 established a parametric equation to quantitatively evaluate the effect of aerosols on
38 photolysis frequencies in Beijing. According to the parametric equation, aerosols
39 lead to a decrease in seasonal mean $j(NO_2)$ by 24% and 30% for summer and winter,
40 respectively, and the corresponding decrease in seasonal mean $j(O^1D)$ by 27% and
41 33% respectively, compared to an aerosol-free atmosphere (AOD = 0). Based on an
42 observation campaign in August 2012, we used the photochemical box model to
43 simulate the ozone production rate ($P(O_3)$). The simulation results shows that the
44 monthly mean daytime net ozone production rate is reduced by up to 25% due to the

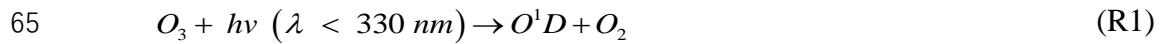
45 light extinction of aerosols. Through further in-depth analysis, it was found that
46 particulate matter concentrations maintain high level under the condition of high
47 concentrations of ozone precursors (VOCs and NO_x), which inhibits the production
48 of ozone to a large extent. This phenomenon implies a negative feedback mechanism
49 in the atmospheric environment of urban Beijing.

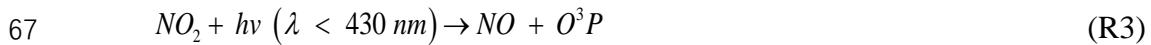
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51

52 **1. Introduction**

53 Solar radiation plays an important role in atmospheric photochemistry, driving
54 the photolysis of many key species. The photolysis of ozone (O₃), gaseous nitrous
55 acid (HONO), and carbonyl species, which contributes to the primary production of
56 HO_x (Volkamer et al., 2010). The photolysis of ozone produces O¹D, which then
57 reacts with H₂O to form OH radicals; these radicals are the main source of OH
58 radicals in the troposphere, as shown by reactions R1 and R2. The strong
59 dependence of OH concentration on j(O¹D) was found in a number of field
60 measurements (Ehhalt et al., 2000; Rohrer et al., 2014; Stone et al., 2012). In
61 addition, the photolysis of NO₂ produces O³P, and then O³P reacts with O₂ to
62 produce O₃, as shown by reactions R3 and R4, which is the only significant chemical
63 source of ozone in the troposphere (Finlayson-Pitts et al., 2000). The photolysis
64 frequencies of R1 and R3 are j(O¹D) and j(NO₂), respectively.





69 The photolysis frequencies are calculated by the following formula:

$$j = \int_{\lambda_1}^{\lambda_2} F(\lambda) \sigma(\lambda, T) \varphi(\lambda, T) d\lambda \quad (\text{E1})$$

71 $F(\lambda)$ is the actinic flux dependent on wavelength. Since the photolysis rates are
72 proportional to the actinic flux and not all stations acquire a 2π spectroradiometer or
73 chemical actinometers for J measurements, several methods have been developed to
74 determine actinic flux and photolysis frequencies from ground based measurements
75 of irradiance (Kylling et al 2003, Kazadzis et al. 2000, 2004, Topaloglou et al. 2005,
76 Trebs et al. 2009). $\sigma(\lambda, T)$ is the absorption cross section of the species that absorbs
77 in the wavelength range $\lambda_1-\lambda_2$. $\phi(\lambda, T)$ is the quantum yield of the photodissociation
78 reaction product. λ , and T represent wavelength, species and temperature,
79 respectively.

The effect of aerosols on photolysis frequencies depends on the aerosol optical properties, SZA and altitude (Liao et al., 1999). The aerosol optical depth (AOD) characterizes the integral of the extinction coefficient of aerosols in the vertical direction. The light extinction of aerosols includes scattering and absorption, which have different effects on the actinic flux. Previous studies showed that scattering aerosols can enhance the actinic flux throughout the troposphere, while absorptive aerosols reduce the actinic flux throughout the boundary layer (Jacobson, 1998; Dickerson et al., 1997; Castro et al., 2001; Flynn et al., 2010). To distinguish

88 between these two components, single scattering albedo (SSA) is defined as the ratio
89 of the scattering coefficient to the total extinction coefficient. In areas with severe
90 aerosol pollution, aerosols can significantly affect photolysis frequencies and ozone
91 production. Studies in Los Angeles (Jacobson, 1998), Mexico City (Castro et al.,
92 2001; Raga et al., 2001; Li et al., 2011), São Paulo (de Miranda et al., 2005), Huston
93 (Flynn et al., 2010), Europe (Real et al., 2011) and Russia (Pere et al., 2015) have
94 found that aerosols reduce ozone concentration by 5-30% by attenuating photolysis
95 frequencies. Studies in the eastern United States have shown that scattering aerosols
96 increase ozone concentration by 5-60% by increasing the photolysis frequencies
97 (Dickerson et al., 1997; He and Carmichael, 1999). Therefore, it is necessary to
98 quantitatively evaluate the effect of aerosols on photolysis frequencies for a better
99 understanding of ozone formation under highly polluted conditions.

100 Currently, the methods for quantitatively evaluating the influence of aerosols on
101 photolysis frequencies mainly include radiative transfer model and parameterization
102 method (Madronich et al., 1993). Radiative transfer model is based on an algorithm
103 for calculating solar radiation and photolysis frequencies (Madronich et al., 1999).
104 The observed data of related influential factors of the photolysis frequencies are
105 taken as the model's input to calculate the photolysis frequencies. This method
106 comprehensively considers the influence of aerosol optical properties on the
107 photolysis frequencies, but it does not necessarily reflect the true quantitative
108 relationship in the atmosphere due to complicated environmental conditions and thus
109 the simulated results don't necessarily reproduce observed values well (Lefer et al.,

110 2003; Shetter et al., 2003; Hofzumahaus et al., 2004). For example, the simulated
111 slope of $j(O^1D)$ vs AOD by Fast-JX algorithm within the CHIMERE model was
112 significantly smaller than the observed slope, particularly for the high SZA values
113 (Mailler et al., 2016). The parameterization method is based on the observation data
114 taken from a certain region and is used to establish the parameterized relationship
115 between the photolysis frequencies and optical properties of aerosols (such as AOD).
116 The method can reflect the actual atmospheric environment conditions; it also
117 considers less influential factors and thus is easy to apply (Casasanta et al., 2011;
118 Gerasopoulos et al., 2012). The disadvantage of this method is that the established
119 parametric equations apply only to a specific region and cannot be extended to other
120 regions.

121 With rapid economic development and urbanization in past decades, China's
122 atmospheric pollution has become more and more severe, characterized by high
123 concentrations of particulate matter and ozone. Satellite observations indicates that
124 both the particulate matter and the ozone of eastern China are at higher levels
125 compared with other locations in the globe (Verstraeten et al., 2015; Ma et al., 2014).
126 Levels of pollution in the Beijing–Tianjin–Hebei are even more severe (Chang et al.,
127 2009; Che et al., 2008; Zhang et al., 2014, Zhang et al., 2016). Therefore, it is
128 necessary to study the effects of aerosols on photolysis frequencies and ozone
129 production in the urban areas of China.

130 Previous model studies have shown that aerosols in China can affect ozone
131 production by changing the photolysis frequencies. Tang et al. (2004) used a sulfur

132 transmission–emission model (STEM) to discover that ozone concentration in
133 northeastern China was reduced by 0.1–0.8% in the sandstorm due to the change in
134 photolysis frequencies . Tie et al. (2005) used a global aerosol–chemical model to
135 show that aerosols caused $j(O^1D)$ and $j(NO_2)$ to decrease in winter by 20%-30% and
136 10%-30%, respectively, and in summer by 5%-20% and 1%-10%, respectively,
137 resulting in 2%-5% and 2% reductions in ozone concentration in winter and summer,
138 respectively. Li et al. (2011) used an air quality model to estimate the changes in the
139 photolysis frequencies caused by sulfate, nitrate, ammonium, and mineral dust
140 aerosols in the central and eastern regions of China from June 1 to June 12, 2006.
141 This study showed that the daily average $j(O^1D)$ in the troposphere at the altitude of
142 1 km, 3 km, and 10 km from the ground was reduced by 53%, 37-%, and 21%,
143 respectively, resulting in a decrease in the ozone concentration by 5.4%, 3.8%, and
144 0.10% in the three layers. Lou et al (2014) found that with aerosols, annual mean
145 photolysis frequencies , $j(O^1D))$ and $j(NO_2)$, were simulated to be reduced by 6-18%
146 in polluted eastern China, leading to reductions in O_3 of up to 0.5 ppbv in those
147 regions in spring and summer by using the global chemical transport model
148 (GEOS-Chem). However, all of these studies base their results on model simulations.
149 Research using long-term observational data to evaluate the effects of aerosols on
150 photolysis frequencies and ozone production in China has not yet been published.

151 Our overall goal is to quantitatively evaluate the effect of aerosols in urban
152 Beijing on photolysis frequencies and thus on ozone production. First, the
153 relationship between PM_{2.5} and AOD was investigated. Second, based on long-term

154 observations (2012-2015) of photolysis frequencies, we discussed the impact of
155 AOD on photolysis frequencies ($j(O^1D)$ and $j(NO_2)$) in urban Beijing in detail. The
156 relationship between photolysis frequencies and AOD is adequately compared with
157 previous study in the Mediterranean (Casasanta et al., 2011; Gerasopoulos et al.,
158 2012). Then, the quantitative relationship between photolysis frequencies, AOD, and
159 SZA was acquired by the parameterization method, which could be used to
160 quantitatively evaluate the effect of AOD on photolysis frequencies in Beijing.
161 Finally, a photochemistry box model was used to evaluate the effect of aerosols on
162 ozone production.

163 **2. Methodology**

164 **2.1. Measurement**

165
166 From August 2012 to December 2015, $j(O^1D)$ and $j(NO_2)$ were measured
167 continuously at PKUERS site. The data of the period during October 2012 to March
168 2013 and August 2015 are missed due to instrument maintenance and other
169 measurement campaigns. The site (39.99°N, 116.31°E) is located on the sixth floor
170 of a campus building at the Peking University, 20 km northwest of Tiananmen
171 Square. The height from the ground is about 30 m. The sampling point is surrounded
172 by classroom buildings. Concentration level and composition of air pollutants were
173 thought to be similar to the downtown so as to be representative for the whole of
174 Beijing (Wang et al., 2010; Xu et al., 2011; Zhang et al., 2012; Zhang et al., 2014).

175 The actinic flux was measured using a spectroradiometer and the photolysis
176 frequencies were calculated from the absorption cross section and quantum yield of
177 each species (Shetter and Müller, 1999). The spectroradiometer consisted of a single
178 monochromator with a fixed grating (CARL ZEISS), an entrance optic with a 2π
179 steradian (sr) solid angle quartz diffusor and a 2048×64 -pixel photodiode array
180 detector. The spectral measurements were performed with a wavelength resolution of
181 2 nm, covering a wavelength range of 290-650 nm (Hofzumahaus et al., 1999). A
182 1000 W National Institute of Standard and Technology (NIST) traceable lamp was
183 used for calibration under laboratory conditions (Bohn et al., 2008). The measured
184 spectra were corrected for dark signal and stray light. For $j(O^1D)$, the quantum yields
185 used were taken from Matsumi et al.(2002), while the ozone cross section was
186 derived from Daumont et al. (1992) and Malicet et al. (1995). Measured
187 temperatures were used to retrieve ozone absorption cross section and quantum yield.
188 For $j(NO_2)$, the quantum yields used were taken from Bass et al. (1976) and
189 Davenport et al. (1978), while the cross section was derived from Jones and Bayes
190 (1973), Harker et al. (1977) and Davenport (1978). The calculated photolysis
191 frequencies had a time resolution of 10 s and an accuracy of $\pm 10\%$ including
192 uncertainties associated with the quartz receiver and stray-light effects (Edwards and
193 Monks, 2003).

194 The optical properties of aerosols were measured by a CIMEL solar photometer
195 (AERONET level 2 data collection, <http://aeronet.gsfc.nasa.gov/>) and the site
196 selected is the Beijing-CAMS site ($39.93^\circ N$, $116.32^\circ E$), which is 6.4km from the

197 PKUERS site. The CIMEL solar photometer is an automatic solar-sky scanning
198 radiometer that uses selected spectral channels. The instrumentation, data acquisition,
199 retrieval algorithms, and calibration procedure conform to the standards of the
200 AERONET global network and have been described in detail by Fotiadi et al. (2006).
201 The solar extinction measurement was performed every 3 minutes in the spectral
202 range 340–1020 nm for the calculation of AOD at wavelengths 340, 380, 440, 500,
203 675, 870, 970, and 1020 nm. Under cloudless conditions, the overall uncertainty of
204 AOD data is ± 0.01 at $\lambda > 440$ nm and ± 0.02 at shorter wavelengths. In this study,
205 AOD at the wavelength of 380 nm was chosen for analysis. This wavelength was
206 selected as it is more representative of $j(\text{NO}_2)$. Additionally, at this wavelength we
207 can better compare with the results of Gerasopoulos et al. (2012). The daytime
208 clear-sky conditions were identified according to the presence of AOD data of
209 AERONET since AOD data are unavailable under cloudy conditions. AE and SSA
210 (440nm) were also acquired from ARONET. In addition, the SSA (525nm) data were
211 derived from a field campaign undertaken in August 2012. The absorption and
212 scattering coefficients were measured with an Aethalometer (AE-31, Magee) and a
213 Single Wavelength Integrating Nephelometer (Aurora-1000), respectively, with a
214 time resolution of 1 minute. As aerosol particles were dried by decreasing relative
215 humidity (RH) to <40% when SSA was measured, we used the measured
216 hygroscopic factor (Liu et al., 2009) and measured RH to correct the SSA.
217 Five-minute averages of AOD, SSA, and photolysis frequencies were analyzed in
218 this study. The total ozone column was obtained by OMI (Ozone Monitoring

219 Instrument) for the year 2012-2015, using overpass data
220 (<http://www.temis.nl/protocols/O3global.html>) (Henk et al., 2003). In addition,
221 meteorological parameters such as temperature, relative humidity, and pressure were
222 simultaneously observed at this site. Table 1 presents total O₃ column, temperature,
223 relative humidity, daytime clear-sky fraction and respective standard deviation for
224 different seasons.

225 The analysis of the effects of aerosols on ozone production (Section 3.4) was
226 based on the field campaign undertaken in August 2012. The relevant contents and
227 methods of observation are shown in Table 2. Since the time resolution of VOCs is 1
228 hour, all data analyzed in Section 3.4 was processed as 1-hour average values. In this
229 study, we focused on the effects of aerosols on photolysis frequencies and ozone
230 production under cloudless conditions.

231 **2.2 Radiative Transfer Model Description**

232 We use the Tropospheric Ultraviolet and Visible (TUV) radiation model
233 (version 5.3) provided by Sasha Madronich (Madronich, 1993). In order to solve the
234 radiative transfer equation, TUV uses the discrete-ordinates algorithm (DISORT)
235 with 4 streams and calculates the actinic flux spectra with wavelength range of
236 280-420 nm in 1 nm steps and resolution. Measured temperatures were used to
237 calculate the absorption cross sections and quantum yields. The key aerosol optical
238 properties including AOD, SSA and AE were input into the model to test the effect
239 of aerosols on photolysis frequencies. AE(380/550nm) is taken from AERONET and
240 the mean value of 1.2 during June 2012 - December 2015 is used in TUV model.

241

242 **2.3 Photochemical box model**

243 The photochemical box model used in this study is based on a regional
244 atmospheric chemical mechanism (RACM2) described by Goliff et al. (2013). The
245 mechanism includes 17 stable inorganic compounds, 4 intermediate inorganic
246 compounds, 55 stable organic compounds, and 43 intermediate organic compounds.
247 Compounds not specifically treated in RACM are incorporated into species with
248 similar functional groups. The isoprene-related mechanism used in this model is
249 LIM mechanism proposed by Peeters et al. (2009). In this study, the observed NO₂,
250 CO, SO₂, C₂–C₁₂ NMHCs, HCHO, photolysis frequencies, temperature, pressure,
251 and relative humidity were used as constraints to simulate the concentrations of
252 reactive radicals (RO₂, HO₂, and OH), intermediate species, and associated reaction
253 rate constants. HONO wasn't measured during the period and was calculated
254 according to the concentration of NO₂ and the observed ratio of HONO to NO₂ at an
255 urban site in Beijing, which had a marked diurnal cycle, a maximum in the early
256 morning (ratio values up to ~0.05–0.08 in summer) and a decrease during daytime to
257 values around 0.01–0.02 (Hendrick et al., 2014). The model was spun up for two
258 days once it started running in order to ensure that the simulation was stable. It was
259 assumed that the lifetime of simulated species removed by dry deposit was 24 hours.
260 The lifetime corresponds to the assumed deposit rate of 1.2 cm s⁻¹ and a well-mixed
261 boundary layer height of about 1 km (Lu et al., 2012). Net ozone production is equal

262 to the reaction rate between peroxy radicals (RO_2 and HO_2) and NO minus the loss
263 rate of NO_2 and O_3 as shown in E2, E3, and E4 as derived by Mihelcic et al. (2003).
264 The ozone production rate ($P(\text{O}_3)$), the ozone loss rate ($D(\text{O}_3)$), and the net $P(\text{O}_3)$
265 were calculated from the simulation results.

266

267
$$P(\text{O}_3) = k_{\text{HO}_2+\text{NO}} [\text{HO}_2][\text{NO}] + \sum (k^i_{\text{RO}_2+\text{NO}} [\text{RO}_2^i][\text{NO}]) \quad (\text{E2})$$

268

269
$$D(\text{O}_3) = (\theta j (\text{O}^1\text{D}) + k_{\text{OH}+\text{O}_3} [\text{OH}] + k_{\text{HO}_2+\text{O}_3} [\text{HO}_2] + \sum (k^j_{\text{alkene}+\text{O}_3} [\text{alkene}^j])) [\text{O}_3] + k_{\text{OH}+\text{NO}_2} [\text{OH}] [\text{NO}_2] \quad (\text{E3})$$

270

271
$$\text{net } P(\text{O}_3) = P(\text{O}_3) - D(\text{O}_3)$$

272 (E4)

273 where θ is the fraction of O^1D from ozone photolysis that reacts with water vapor. i
274 and j represent the number of species of RO_2 and alkenes, respectively.

275

276 **3. Results and discussion**

277

278 **3.1 The correlation between PM_{2.5} and AOD**

279 Compared with AOD, PM_{2.5} is a more common proxy to evaluate the level of
280 particulate matter pollution in spite that AOD is a more closely related parameter of
281 photolysis frequencies. As a result, we attempted to analyze the quantitative
282 relationship between PM_{2.5} and AOD to evaluate the influence of PM_{2.5} on AOD and

thus on photolysis frequencies. The factors that affect this relationship include aerosol type, aerosol size distribution, aerosol distribution in the vertical direction, relative humidity (RH) and planetary boundary layer height (PBLH) (van Donkelaar et al., 2010). Figure 1 shows the correlation between AOD and PM_{2.5} in four different seasons. The determination coefficient (r^2) is 0.53, 0.58, 0.62 and 0.59 for spring (March, April and May), summer (June, July and August), autumn (September, October and November) and winter (December, January and February), respectively. Meanwhile, the correlation exhibits significant seasonal differences, having relatively smaller slope (23.56) in summer and relatively larger slope (73.76) in winter. This implies that PM_{2.5} in summer has stronger light extinction capacity than in winter. One reason for the seasonal differences is the variation in RH among different seasons (Table 1). There is higher RH in summer (57.2% on average) than in winter (30.4% on average), leading to stronger hygroscopic growth of aerosol particles, and thus resulting in higher scattering ability of aerosol particles. According to another study in urban Beijing, the higher the RH, the smaller the slope, and the higher the PBLH, the smaller the slope (Zheng, C. W et al., 2017). In addition, the slope was smaller for scattering-dominant aerosols than for absorbing-dominant aerosols, and smaller for coarse mode aerosols than for fine mode aerosols (Zheng, C. W et al., 2017). The slopes of the correlation between AOD (at 550nm) and PM_{2.5} in this study in summer and winter are equal to 42.2 $\mu\text{g m}^{-3}$ and 119.2 $\mu\text{g m}^{-3}$, respectively, close to that from Ma et al. (2016) (54.9 $\mu\text{g m}^{-3}$ and 110.5 $\mu\text{g m}^{-3}$) and Xin et al. (2016) (55.2 $\mu\text{g m}^{-3}$ and 93.4 $\mu\text{g m}^{-3}$), but smaller

305 significantly than that from Zheng et al. (2017) ($65\text{--}74\mu\text{g m}^{-3}$ and $143\text{--}158\mu\text{g m}^{-3}$).
306 The differences mainly depend on the aerosol composition and size distribution at
307 different observational sites in Beijing. Compared with other cities in North China
308 (Tianjin, Shijiazhuang and Baoding) (Ma et al., 2016), the slope in Beijing for winter
309 is significantly higher. Consequently, using $\text{PM}_{2.5}$ to estimate AOD has a large
310 uncertainty due to multiple interference factors.

311
312

313 **3.2 Seasonal and diurnal variability of AOD and photolysis frequencies**

314 The diurnal cycles of AOD are shown in Figure 2. AOD displays obvious
315 diurnal variation, with relatively high level at noon and low level at dawn and
316 evening. The diurnal variation of $\text{PM}_{2.5}$ is significantly different from AOD. In
317 addition, AOD has obvious seasonal differences, with the highest AOD in summer
318 and the lowest AOD in winter. Conversely, $\text{PM}_{2.5}$ in winter ($42\mu\text{g m}^{-3}$) is
319 significantly higher than in summer ($35\mu\text{g m}^{-3}$). In spite of lower $\text{PM}_{2.5}$ in summer,
320 AOD in summer is higher due to stronger extinction capacity of $\text{PM}_{2.5}$ as discussed
321 in 3.1. Figure 3 shows the diurnal variation of the photolysis frequencies under
322 cloudless conditions for each season. $j(\text{O}^1\text{D})$ and $j(\text{NO}_2)$ are both highest in summer,
323 followed by spring and autumn, and lowest in winter. This seasonal difference is
324 mainly determined by the difference in SZA for the four seasons.

325 The observed mean daily maxima of photolysis frequencies at this site are
326 lower than that observed in the eastern Mediterranean (Crete, Greece, $35^\circ20'\text{N}$, 25°

327 40°E) (Gerasopoulos et al., 2012) by $7.8 \times 10^{-6} \pm 5.5 \times 10^{-6}$ s⁻¹ and $5.5 \times 10^{-6} \pm 1.8 \times$
328 10⁻⁶ s⁻¹ for j(O¹D), and $1.9 \times 10^{-3} \pm 1.2 \times 10^{-3}$ s⁻¹ and $3.3 \times 10^{-3} \pm 1.0 \times 10^{-3}$ s⁻¹ for
329 j(NO₂), in summer and winter respectively. The corresponding lower photolysis
330 frequencies of Beijing than the eastern Mediterranean due to SZA difference is $1.7 \times$
331 10⁻⁶ s⁻¹ and 3.0×10^{-6} s⁻¹ for j(O¹D), and 8.0×10^{-5} s⁻¹ and 6.6×10^{-4} s⁻¹ for j(NO₂)
332 according to TUV model under aerosol-free conditions, which are significantly
333 lower than observed decreased magnitudes. Additionally, we know that the
334 temperature is lower in Beijing during the winter compared to conditions in Crete.
335 The measured mean temperature in Beijing during winter is equal to 0.53 ± 4.2 °C
336 (Table 1). When we consider the temperature in Crete is 10 °C higher than in Beijing,
337 the lower j(O¹D) of Beijing than Crete is 5.5×10^{-7} s⁻¹, which is also not able to
338 compensate the j(O¹D) gap between the two sites during winter. Taking into account
339 the similar levels of ozone column concentration in the two sites, the large gap of
340 photolysis frequencies in the two sites is mainly attributed to the higher AOD in
341 Beijing (0.76 ± 0.75) than in the eastern Mediterranean (0.27 ± 0.13).

342 It can be seen from Figure 3 that the difference between winter and summer for
343 j(O¹D) is significantly larger than that for j(NO₂), where the summer midday
344 averages of j(O¹D) and j(NO₂) are 5 times and 2 times those of winter, respectively.
345 There are two reasons for this phenomenon. One, compared with j(NO₂), j(O¹D) is
346 more sensitive to the change in SZA and the same change in SZA results in a larger
347 change in j(O¹D) than j(NO₂). Two, the main influential factors of j(NO₂) under
348 cloudless conditions are SZA and AOD, and the influence of ozone column

349 concentration and temperature on $j(\text{NO}_2)$ is negligible. However, $j(\text{O}^1\text{D})$ is affected
350 significantly by the ozone column concentration and temperature, in addition to SZA
351 and AOD. The higher ozone column concentration and lower temperature in winter
352 than in summer lead to the difference in $j(\text{O}^1\text{D})$ further increasing (Table 1).

353 **3.3 The correlation between photolysis frequencies and AOD**

354 **3.3.1 The correlation between $j(\text{O}^1\text{D})$ and AOD**

355 In order to rule out the effect of SZA on photolysis frequencies, we chose SZA
356 equal to 30° and 60° ($\pm 1^\circ$) for analysis. Figure 4 presents the dependence of $j(\text{O}^1\text{D})$
357 on AOD at different levels of ozone column concentration at SZA of 30° and 60° (\pm
358 1°). The ozone column concentration has a classification width of 30 DU. This
359 relatively large classification width is chosen to make sure that there are enough
360 points to fit the relationship between $j(\text{O}^1\text{D})$ and AOD. $j(\text{O}^1\text{D})$ exhibits a clear
361 dependence on AOD, with a nonlinear negative correlation. The scatter of these
362 points is mainly due to variations in ozone column and temperature. As AOD
363 increased, the slope of $j(\text{O}^1\text{D})$ -AOD gradually decreases, indicating that the ability of
364 aerosols to reduce $j(\text{O}^1\text{D})$ gradually decreases with AOD. This result differs from
365 that found in Mediterranean, where $j(\text{O}^1\text{D})$ was linearly negatively correlated with
366 AOD (Casasanta et al., 2011; Gerasopoulos., 2012). A larger variation range of AOD
367 in Beijing (0-3) compared with Mediterranean (0-0.6) is one reason for the
368 difference.

369 For further analysis, the observed relation between $j(O^1D)$ and AOD was
370 compared with TUV-simulated results. Panels a and b of Figure 5 present the
371 comparison between observed and TUV-simulated $j(O^1D)$ against AOD at a SZA of
372 30° and 60° respectively and ozone column concentration of 330-360 DU. The
373 observed $j(O^1D)$ was at ozone column of 330-360DU and were scaled to the
374 temperature of 298K. $AE(380/550\text{nm}) = 1.2$, ozone column = 345 and Temperature =
375 298K were used in TUV model for all simulations. Mean Earth-Sun distance was
376 used in the calculations of TUV model and measured j -values were scaled to the
377 mean Earth-Sun distance. At low AOD level (< 0.8), the observed slope of $j(O^1D)$ vs
378 AOD is significantly larger than the simulated slope at SSA of 0.95, and slightly
379 larger than the simulated slope at SSA of 0.85. With AOD increasing, the observed
380 slope decreases rapidly to the level smaller than the simulated slopes.

381 The rapid change of the slope with AOD can be related to the variation of SSA
382 at different AOD level. Figure 6 presents the relationship between SSA and AOD-
383 ~~based on observed data in August 2012. Figure 6 (a) presents the relationship~~
384 between AERONET based SSA (440nm) and AOD during 2012-2015. The result
385 suggests a significant positive correlation between SSA and AOD. Additionally,
386 Figure 6 (b) also presents a significant positive correlation between near-ground SSA
387 (525nm) from measurement campaign in August 2012 and AOD. With the increase
388 in AOD, SSA is elevated; meanwhile, the slope of SSA vs AOD is gradually reduced.
389 Similar results in other regions have been obtained by Bais et al., 2005, Krotkov et
390 al., 2005 and Kazadzis et al., 2012. SSA characterizes the ratio of the scattering

extinction coefficient to the total extinction coefficient (scattering extinction coefficient plus absorptive extinction coefficient) of aerosols. The smaller the SSA, the higher the absorptive component and lower the scattering component of the aerosol, and the stronger the ability of the aerosol to reduce the actinic flux (Dickerson et al., 1997). Figure 6 indicates that aerosols in Beijing under low AOD conditions had a higher proportion of absorptive aerosol components than under high AOD conditions, and, as a result, had a stronger ability to reduce the photolysis frequencies, which contributed to the rapidly reduced slope of $j(O^1D)$ vs AOD with AOD. However, due to absence of more SSA data of the period 2012-2015, we can't give more sufficient evidence for the dependence of SSA on AOD. For another perspective, Owing to the biomass burning and soot emission generated from heating, the fine mode heavily-absorbing aerosol percentage is higher in winter than in summer (Zheng et al., 2017; Liu et al., 2016; Zhang et al., 2013), and thus aerosols in winter have stronger ability to reduce the photolysis frequencies. [Table S1](#) indicates that SSA in summer is higher significantly than in winter. High AOD levels often appeared in summer and low AOD levels occurred mostly in winter (Figure 2 and [Table S1](#)), another fact that may also explains the rapidly reduced slope of $j(O^1D)$ vs AOD with AOD.

It worth noting that the mean near-ground SSA (525nm) in August 2012 (0.88 ± 0.08) is significantly lower than the mean AERONET based SSA (440nm) in the same period (0.94 ± 0.02) and in summer (0.94 ± 0.02). The different wavelength plays a minor role in the different SSA according to the wavelength dependence of

413 AERONET based SSA (Figure S3). This difference is possibly due to the uncertainty
414 of AERONET based SSA and the poor vertical representative of near-ground SSA.
415 The AERONET based SSA generally reproduces well the slope of $j(O^1D)$ versus
416 AOD in spite that it significantly underestimates the absolute value of the slope at
417 low AOD range ($AOD < 0.7$), which is probably due to the uncertainty of AERONET
418 based SSA in low AOD range. In addition to the uncertainty of SSA, both of SSA at
419 440nm and at 525nm differ from the 305-315nm wavelength range of $j(O^1D)$, which
420 is likely to lead to some uncertainties for the analysis of the relationship between
421 $j(O^1D)$ and AOD.

422 Comparing panels a and b of Figure 4, we see that at AOD smaller than 1, the
423 slope of $j(O^1D)$ vs AOD exhibits a significant dependence on SZA and the slope at
424 30° of SZA is about 1.5-2.0 times larger than that at 60° of SZA. This result is
425 similar to that of the observations made in the central Mediterranean (Casasanta et
426 al., 2011). For the purpose of comparison with the study in the Mediterranean, the
427 slope of $j(O^1D)$ vs AOD was calculated at AOD smaller than 0.7.

428 Table 3 presents slope, intercept and the determination coefficient (r^2) of linear
429 fits of correlation between $j(O^1D)$ and AOD for each ozone column class at AOD
430 smaller than 0.7. At SZA of 60° and O₃ column concentration of 300-330 DU, the
431 respective slope of the linear regression indicates a reduction of $j(O^1D)$ by $4.2 \cdot 10^{-6}$
432 s⁻¹ per AOD unit. Gerasopoulos et al. (2012) reported that the observed slope in the
433 eastern Mediterranean was equal to $2.4 \cdot 10^{-6}$ s⁻¹ at O₃ column of 300-320 DU.
434 Casasanta et al. (2011) reported that the observed slope in the central Mediterranean

435 varied from $2.7 \cdot 10^{-6} \text{ s}^{-1}$ to $3.9 \cdot 10^{-6}$ at O₃ column of 300-330 DU. All of these results
436 are smaller than the value of the present study, indicating that aerosols in urban
437 Beijing had a stronger extinction capacity on j(O¹D) than those in the Mediterranean
438 that was influenced by both natural absorptive aerosols and anthropogenic aerosols.
439 Previous study indicated that SSA in Beijing ranged from 0.80 to 0.86 (Garland et al.,
440 2009; Han et al., 2015b; Han et al., 2017; Tian et al., 2015). The relatively low SSA
441 in Beijing could be an important reason for the stronger extinction capacity.

442

443

444 **3.3.2 The correlation between j(NO₂) and AOD**

445 Unlike j(O¹D), j(NO₂) is negligibly affected by ozone column concentration and
446 depends mainly on AOD and SZA under cloudless conditions. Figure 7 presents the
447 dependence of j(NO₂) on AOD at different SZA levels under cloudless conditions.
448 The cosine of SZA (cos (SZA)) is categorized according to a width of 0.2. In the
449 same category of cos (SZA), j(NO₂) displays a strong dependence on AOD. The
450 scatter of these points is due to the relatively large classification width of SZA to a
451 large extent. When cos(SZA) is at its maximum level (0.8-1), the correlation
452 between j(NO₂) and AOD is close to linear. When cos (SZA) decreases, the
453 correlation tends to be nonlinear. Similar to j(O¹D), the observed slopes of j(NO₂) vs
454 AOD are also larger than TUV-simulated slope at SSA of 0.95 and 0.85 when AOD
455 is smaller than 0.8, and decreased rapidly with increasing AOD (panels c and d of

456 Figure 5). The reason for this result is the same with that for $j(O^1D)$ as explained
457 above.

458

459 Table 4 presents the slope, intercept and the determination coefficient (r^2) of
460 linear fits of correlation between $j(NO_2)$ and AOD for each ozone column class at
461 AOD smaller than 0.7. The slope of $j(NO_2)$ vs AOD also displays a significant
462 dependence on $\cos(SZA)$. The slope increases as $\cos(SZA)$ increases from 0 to 0.5
463 and then decreases as $\cos(SZA)$ increases from 0.5 to 1. At SZA of $60^\circ \pm 1$
464 ($\cos(SZA)=0.5 \pm 0.015$), the respective slope of the linear regression indicates a
465 reduction of $j(NO_2)$ by $3.2 \cdot 10^{-3} s^{-1}$ per AOD unit. This result is larger than the value
466 for non-dust aerosols ($2.2 \cdot 10^{-3} s^{-1}$) and close to the value for dust aerosols ($3.1 \cdot 10^{-3}$
467 s^{-1}) in the eastern Mediterranean reported by Gerasopoulos et al. (2012).

468

469 **3.4 The parameterization relationship between photolysis frequencies, AOD,**
470 **and SZA**

471

472 As analyzed above, the photolysis frequencies ($j(O^1D)$ and $j(NO_2)$) strongly
473 depended on AOD and $\cos(SZA)$ and could be fit into expression E5 using a
474 quadratic polynomial form. The fitting parametric equations for $j(NO_2)$ is shown in
475 Table 5. For $j(O^1D)$, both of O_3 column and temperature affect $j(O^1D)$ significantly.
476 Figure S1 presents the dependence of $j(O^1D)$ on ozone column at low AOD level

477 (AOD<0.3) and SZA of (a) $30^\circ \pm 1^\circ$ and (b) $60^\circ \pm 1^\circ$, respectively. Ozone column
 478 ranging from 270 to 400 DU leads to $j(O^1D)$ reducing about 50%. In order to
 479 evaluate the impact of temperature on $j(O^1D)$, we calculated the ratio of $j(O^1D)$ at
 480 measured temperature to $j(O^1D)$ at temperature = 298K ($j(O^1D)/j(O^1D)_{T=298K}$)
 481 (Figure S2). $j(O^1D)/j(O^1D)_{T=298K}$ varied from 0.82 to 1.03 indicating that temperature
 482 changed $j(O^1D)$ by no more than 21%. Therefore, temperature played a minor role in
 483 changing $j(O^1D)$ compared with ozone column. As a result, when we fitted the
 484 relationship among $j(O^1D)$, AOD and $\cos(SZA)$, the effect of ozone column is
 485 considered but the effect of temperature is not considered. By fitting the relationship
 486 at different ozone classes (classification width=30DU), we found that ozone column
 487 increasing by 30DU results in $j(O^1D)$ at a constant SZA and AOD decreasing by
 488 18%. Therefore, the parametric equation for $j(O^1D)$ is transformed into the form E6,
 489 which reflects the influence of ozone column. The parameters a_1-a_6 correspond to
 490 ozone column range = 300-330 DU, thus we use 315 DU as the weighted standard of
 491 ozone column. The fitting parameters a_1-a_6 for $j(O^1D)$ is shown in Table 6.

492
$$j(NO_2) = a_1 + a_2 AOD + a_3 \cos(SZA) + a_4 (AOD)^2 + a_5 AOD \cos(SZA) + a_6 (\cos(SZA))^2$$

 493E5

494
$$j(O^1D) = [a_1 + a_2 AOD + a_3 \cos(SZA) + a_4 (AOD)^2 + a_5 AOD \cos(SZA) + a_6 (\cos(SZA))^2] \times [1 + (315 - O_3 \text{ column}) \times 0.006]$$

495E6

496 The coefficients of determination of the fitting equations are greater than 0.95
 497 for $j(NO_2)$ and $j(O^1D)$ at a certain O_3 column, indicating that both of the photolysis
 498 frequencies strongly depended on AOD and $\cos(SZA)$, and the effect of other factors

499 such as SSA and AE are integrated into the constant term in the parametric equation.
500 Since the ozone column concentration has greater influence on $j(O^1D)$ than on
501 $j(NO_2)$, the parameters of fitting equations for $j(NO_2)$ are similar, but the parameters
502 of fitting equations for $j(O^1D)$ have a large fluctuation at different O_3 column ranges
503 (especially a_1 and a_2). The parametric equations can be used to quantitatively
504 evaluate the effect of AOD on photolysis frequencies in Beijing. According to the
505 parametric equations, aerosols lead to a decrease in seasonal mean $j(NO_2)$ by 24%
506 and 30% and a decrease in seasonal mean $j(O^1D)$ by 27% and 33% in summer and
507 winter under clear-sky conditions, respectively, compared to an aerosol-free
508 atmosphere. The decreasing ratio of the photolysis frequencies in winter is higher
509 than in summer mainly due to the higher SZA in winter.

510

511 The effect of aerosols on photolysis frequencies in Beijing is compared with
512 other studies. Real and Sartelet (2011) reported a reduction in $j(NO_2)$ and $j(O^1D)$ of
513 13%-14% due to aerosols by using the radiative transfer code Fast-J during summer
514 2001 over European regions. Flynn et al (2010) reported that aerosols reduced $j(NO_2)$
515 by 3% in Huston during 2006 by using TUV model. Gerasopoulos et al (2012)
516 reported that aerosols reduced $j(NO_2)$ and $j(O^1D)$ by 5%-15% with 5-yr mean AOD
517 at 380nm equal to 0.27. All of these results are lower than the reduction ratio of this
518 study mainly due to higher aerosol level in Beijing (4-yr mean AOD equal to $0.76 \pm$
519 0.75). Hodzic et al. (2007) simulated a 15–30% $j(NO_2)$ photolysis reduction during
520 the 2003 European summer heatwave in the case of absorbing biomass burning

521 aerosols with AOD at 550 nm equal to 0.7-0.8 and SSA at 532 nm equal to 0.83-0.87.
522 The result of Hodzic et al. (2007) is comparable with the reduction ratio of this study
523 possibly due to the equivalent levels of AOD and SSA. In addition, Péré et al (2015)
524 simulated a higher reduction (20–50%) in $j(\text{NO}_2)$ and $j(\text{O}^1\text{D})$ along the transport of
525 the aerosol plume during the 2010 Russian summer wildfires episode. The higher
526 reduction is due to the higher level of AOD (peak value of AOD at 400nm reached
527 2-4), even though SSA is very high (0.97).

528

529 **3.5 The influence of AOD on ozone production**

530

531 In order to explain the effect of aerosol light extinction on ozone production, we
532 used the data from the field observation campaign undertaken in August 2012.
533 Ozone production depends on its precursors (NO_x and VOCs), meteorological
534 factors, and solar radiation. Solar radiation is the driving force for tropospheric
535 photochemical reactions, in which $j(\text{O}^1\text{D})$ and $j(\text{NO}_2)$ are both important for ozone
536 production. On the one hand, the increase in $j(\text{NO}_2)$ promotes the photolysis of NO₂,
537 thereby accelerating the formation of ozone. On the other hand, the increase in $j(\text{O}^1\text{D})$
538 accelerates the photolysis of ozone. In addition, the increase in the photolysis
539 frequencies will accelerate the photolysis of OVOC (especially formaldehyde and
540 acetaldehyde), HONO, and H₂O₂, resulting in increases in OH and HO₂, which will
541 promote the reaction between OH and VOCs and thus produce more RO₂. As a result,

542 more ozone is produced by increasing the reaction rate between RO₂ (or HO₂) and
543 NO. However, the increase in OH and HO₂ also consumes ozone and NO₂, which
544 contributes to the increase in D(O₃). In brief, the overall effect of changes in
545 photolysis frequencies on sources and sinks of ozone determines the change in the
546 net ozone production rate.

547 Ozone production (HO₂ + NO, RO₂ + NO), ozone loss (O¹D + H₂O, HO₂ + O₃,
548 O₃ + OH, NO₂ + OH, and O₃ + alkenes), and net ozone production rate during
549 August 2012 were calculated by using the box model. We used the observed
550 photolysis frequencies (i.e. j_{obs}) and the calculated photolysis frequencies by
551 parametric equation under the condition of AOD equal to 0 (i.e. j_{AOD=0}), were
552 used to constrain the box model. The difference of simulated results in the two
553 scenarios can be attributed to the effect of aerosol light extinction. As a result, the
554 presence of aerosols causes a decrease in both ozone production rate and loss rate, as
555 is shown in Figure 8. Since the decreasing amplitude of the daytime ozone
556 production rate is far larger than that of the daytime ozone loss rate, the mean
557 daytime net production rate of ozone is reduced by 25%. This reduction is
558 comparable with the results of the study in Mexico City, where aerosols caused a 20%
559 reduction in the ozone concentrations (Castro et al., 2001). Studies in Houston and
560 Crete have shown that aerosols cause ozone production rates to decrease by about 4%
561 and 12%, respectively, which are lower than that found in this study (Flynn et al.,
562 2010; Gerasopoulos et al., 2012).

563 The ratio of the observed photolysis frequencies to the photolysis frequencies at

564 AOD equal to 0 is defined as JIF (Flynn et al., 2010). A JIF of less than 1 indicates
565 that the aerosols cause a decrease in the photolysis frequencies. Figure 9 shows the
566 relation between $P(O_3)_{j_obs}/P(O_3)_{j_AOD=0}$ (or $D(O_3)_{j_obs}/D(O_3)_{j_AOD=0}$) and JIF. The
567 majority of JIF values were less than 1, with an average of 0.72, indicating that
568 aerosols greatly attenuated photolysis frequencies due to high levels of AOD
569 (average of 1.07) and low levels of SSA (average of 0.84) during the observation
570 period. $P(O_3)_{j_obs}/P(O_3)_{j_AOD=0}$ and $D(O_3)_{j_obs}/D(O_3)_{j_AOD=0}$ are both linearly
571 positively correlated with JIF and the scatters are mostly above the 1:1 line. As can
572 be seen from the figure 9, a 30% reduction in photolysis frequencies (JIF = 0.7) due
573 to the presence of aerosols results in a decrease in ozone production rate and loss rate
574 by about 26% and 15%, respectively. The decreasing amplitude in the ozone
575 production rate is greater than the decrease in the ozone loss rate because the
576 corresponding processes of ozone production are all light-driven, but the
577 corresponding processes of ozone loss are not all light-driven because the reaction of
578 O_3 with alkenes does not depend on solar radiation. According to the simulated
579 results, the reaction of ozone with alkenes during this campaign accounts for 17% of
580 total ozone loss.

581 The diurnal profile of the mean ozone production and loss rate is shown in
582 Figure 10. $P(O_3)$ peak midday in the 12:00-14:00 local hours at 31 ppb/h without
583 aerosol impact and 23 ppb/h with aerosol impact. The maximum $D(O_3)$ also occurs
584 between 12:00 and 14:00 at 4.2 ppb/h without aerosol impact and 3.5 ppb/h with
585 aerosol impact. There is little difference between aerosol-impact and aerosol-free

586 P(O₃) (or D(O₃)) in the hours of 6:00-11:00, but the difference in the afternoon
587 (12:00-18:00) is large, indicating that the reduction effect of aerosol on ozone
588 production mainly occurs during the afternoon.

589 The above analysis focuses on the effect of aerosol on the ozone production due
590 to aerosol light extinction. However, it does not consider the close relationship
591 between aerosol and ozone's gaseous precursors in the actual atmosphere. To explain
592 this problem, we chose two adjacent days (small SZA effect) with obviously
593 different AOD levels: a clean day (A day: August 21, 2012; AOD = 0.21, PM_{2.5}=21.6
594 μg m⁻³) and a day with high aerosol pollution (B day; August 26, 2012; AOD = 3.2,
595 PM_{2.5}=125.0 μg m⁻³) (Table 7). The difference in AOD between the two days can be
596 taken to represent the maximum daytime gap of AOD for this month. The ozone
597 column concentrations for these two days were 302 DU and 301 DU, respectively, of
598 which the effect on j(O¹D) is negligible. Under these conditions, the j(O¹D) value at
599 noon time decreases from $3.23 \times 10^5 \text{ s}^{-1}$ on A day to $1.29 \times 10^5 \text{ s}^{-1}$ on B day (i.e., a 60%
600 reduction) and the j(NO₂) value at noon time decreases from $8.26 \times 10^{-3} \text{ s}^{-1}$ on A day
601 to $4.19 \times 10^{-3} \text{ s}^{-1}$ on B day (i.e., a 49.2% reduction). As shown in Table 7, B day has
602 higher AOD and higher concentrations of gaseous pollutants. The concentrations of
603 CO, NO₂, HCHO and the OH reactivity of VOCs in B day are much higher than in A
604 day, with the ratio of 3.6, 2.3, and 2.0, respectively. The simultaneous increases of
605 gaseous pollutants and AOD are due to the fact that gaseous pollutants (NO_x, SO₂,
606 and VOCs) emitted by major pollution sources in Beijing, including traffic and
607 industry, have undergone the processes of gas-phase oxidation and nucleation to

608 generate secondary particulate matter that contributes to aerosol light extinction.
609 Previous studies have reported that secondary particulate matter has accounted for
610 more than 60% of total particulate matter during severe smog pollution in Beijing
611 summers (Han et al., 2015a; Guo et al., 2014). In addition, several studies have
612 shown that secondary components in particulate matter (especially secondary
613 organics and ammonium sulfate) have dominated the aerosol light extinction (Han et
614 al., 2014; Han et al., 2017; Wang et al., 2015). Observations made in Beijing during
615 the summer of 2006 showed that ammonium sulfate and ammonium nitrate
616 contributed 44.6% and 22.3%, respectively, to the total extinction coefficient during
617 a severe period of smog (Han et al., 2014); in the summer of 2014 in Beijing,
618 ammonium sulfate, secondary organic aerosols, and ammonium nitrate contributed
619 30%, 22%, and 18%, respectively, to the total extinction coefficient (Han et al.,
620 2017).

621 As shown in Figure 11, the simulation results indicate that the net $P(O_3)$ of B
622 day is 36.2% higher than that of A day due to higher concentrations of ozone
623 precursors on B day. This result is consistent with the observed ozone concentrations,
624 of which the observed ozone concentration in B day is 2.2 times higher than that of
625 A day. If we adjust the photolysis frequencies level of B day to the level of A day, the
626 net $P(O_3)$ increases by 70.0%, which indicates that the high level of particulate
627 matter in B day greatly inhibits ozone production. This result means that the system
628 is under negative feedback, thus keeping O_3 at a relatively stable level. Table 8
629 summarizes the average levels of gaseous pollutants and photolysis frequencies for

630 AOD less than 1 and greater than 1, as measured during August 2012. It shows that,
631 the concentrations of ozone's precursors are higher and the photolysis frequencies
632 are lower at high AOD levels ($AOD > 1$) than those at low AOD level ($AOD < 1$).
633 This result means that the negative feedback mechanism is prevalent throughout the
634 whole campaign period. Therefore, the prevention and control measures of air
635 pollution in Beijing need to incorporate this coupling mechanism between particulate
636 matter and ozone to achieve effective control of these two main pollutants.

637 **4. Conclusion**

638 Photolysis reactions are important driving forces for tropospheric
639 photochemical oxidation processes and ozone production. In this study, we explored
640 in detail the effects of aerosols on photolysis frequencies and ozone production in
641 Beijing, based on a long observation period of 4 years. We have found that:

642 (1) There is a strong correlation between $PM_{2.5}$ and AOD, and the slope in
643 summer is smaller significantly than in winter, which indicates that aerosols
644 in summer have a more efficient extinction capacity than in winter.

645 (2) As AOD increased, the extinction effect of aerosol on photolysis
646 frequencies was decreased; this result was probably related to a higher
647 proportion of scattering aerosols under high AOD conditions than under low
648 AOD conditions. The slope of the correlation between photolysis
649 frequencies and AOD indicates that the aerosols in urban Beijing have a

650 stronger extinction on actinic flux than absorptive dust aerosols in the
651 Mediterranean.

652 (3) The influence of AOD on photolysis frequencies was evaluated
653 quantitatively by establishing parametric equations. According to the
654 parametric equation, aerosols lead to a decrease in seasonal mean $j(\text{NO}_2)$ by
655 24% and 30% for summer and winter, respectively, and the corresponding
656 decrease in seasonal mean $j(\text{O}^1\text{D})$ by 27% and 33% respectively, compared
657 to an aerosol-free atmosphere.

658 (4) In order to evaluate the effects of aerosols on ozone production rate, we
659 carried out an observation campaign in August 2012. The results show that
660 aerosols reduced the net ozone production rate by 25% by reducing the
661 photolysis frequencies. High concentrations of ozone gaseous precursors
662 were often accompanied by high concentrations of particulate matter, which,
663 to a large extent, inhibited excessive levels of ozone generation and
664 reflected the negative feedback effect of the atmospheric system. Therefore,
665 the influence of aerosol on photolysis frequencies and thus on the rate of
666 oxidation of VOCs and NOx to ozone and secondary aerosol is important
667 for determining the atmospheric effects of controlling the precursor
668 emissions of these two important air pollutants (aerosols and ozone).

669

670 **Author contribution**

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Author	Contribution
Wenjie Wang	acquisition of data; analysis and interpretation of data; drafting the article and revising it critically
Min Shao	substantial contributions to conception and design; revising the article critically
<u>Xin Li</u>	<u>substantial contributions to conception and design; revising the article critically</u>
Min Hu	collection of data
Limin Zeng	collection of data
Yusheng Wu	collection of data
<u>Tianyi Tan</u>	<u>collection of data</u>

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1037 Table 1. O₃ column concentration, temperature, relative humidity, daytime clear-sky
1038 fraction and respective standard deviation for different seasons (spring: March, April
1039 and May; summer: June, July and August; autumn: September, October and
1040 November; winter: December, January and February).

Season	O ₃ column (DU)	Temperature (°C)	Relative humidity (%)	Clear-sky fraction(%)
Spring	355±37	16±7.8	33±18	41
Summer	310±24	28±4.2	57±18	36
Autumn	304±23	16±7.4	46±21	42
Winter	347±28	0.53±4.2	30±18	41

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1060 Table 2. Instruments deployed in the field campaign undertaken in August 2012 and
1061 used for data analysis.

Parameters	Measurement technique	Time resolution	Detection limit	Accuracy
j(O ¹ D) and j(NO ₂)	Spectroradiometer	10 s	/	± 10%
O ₃	UV photometry	60 s	0.5 ppbv	± 5%
NO	Chemiluminescence	60 s	60 pptv	± 20%
NO ₂	Chemiluminescence	60 s	300 pptv	± 20%
CO	IR photometry	60 s	4 ppb	± 5%
SO ₂	Pulsed UV fluorescence	60 s	0.1 ppbv	± 5%
HCHO	Hantzsch fluorimetry	60 s	25 pptv	± 5%
VOCs	GC-FID/MS	1 h	20-300 pptv	± 15~20%

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1076 Table 3. Slope, intercept and the square of correlation coefficient (r^2) of linear fits of
1077 correlation between $j(O^1D)$ and AOD for each ozone column class at AOD smaller
1078 than 0.7.

O_3 column (DU)	SZA=30°			SZA=60°		
	Slope ($10^{-6}s^{-1}$)	Intercept ($10^{-6}s^{-1}$)	r^2	Slope ($10^{-6}s^{-1}$)	Intercept ($10^{-6}s^{-1}$)	r^2
300-330	-6.2±1.5	26±1	0.34	-4.2±0.4	7.7±0.3	0.41
330-360	-6.5±1.4	23±1	0.40	-5.0±0.3	7.1±0.2	0.52
360-390	-9.5±1.6	21±1	0.52	-6.9±0.6	7.6±0.3	0.66

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1097 Table 4. Slope, intercept and the square of correlation coefficient (r^2) of linear fits of
1098 correlation between $j(\text{NO}_2)$ and AOD for each ozone column class at AOD smaller
1099 than 0.7.

cos(SZA)	Slope (10^{-3} s^{-1})	Intercept (10^{-3} s^{-1})	r^2
0-0.2	-1.3±0.1	1.5±0.0	0.52
0.2-0.4	-2.4±0.1	3.4±0.0	0.41
0.4-0.6	-3.2±0.1	5.5±0.0	0.49
0.6-0.8	-2.1±0.1	7.2±0.1	0.38
0.8-1.0	-1.8±0.1	8.1±0.1	0.26

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1115 Table 5. The fitting parameters a_1-a_6 and determination coefficients of E5 for $j(\text{NO}_2)$.

a_1	a_2	a_3	a_4	a_5	a_6	r^2
$\times 10^{-3}$						
-0.46 ± 0.05	-2.0 ± 0.03	13 ± 0.2	0.22 ± 0.01	0.32 ± 0.05	-4.0 ± 0.1	0.96

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1120 Table 6. The fitting parameters a_1-a_6 and determination coefficients of E6 for $j(\text{O}^1\text{D})$

1121 at ozone column range = 300-330 DU.

a_1	a_2	a_3	a_4	a_5	a_6	r^2
$\times 10^{-6}$						
1.1 ± 0.3	0.58 ± 0.17	-8.7 ± 0.9	0.63 ± 0.05	-7.5 ± 0.3	43 ± 1	0.96

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1132 **Table 7.** Mean and standard deviation of observed data during daytime (6:00–18:00)
 1133 for A day and B day.

Observed data	A day: August 21, 2012	B day: August 26, 2012
AOD	0.21 ± 0.05	3.2 ± 0.4
PM _{2.5} ($\mu\text{g m}^{-3}$)	22 ± 9	125 ± 16
O ₃ column (Du)	302 ± 3	301 ± 3
Temperature($^{\circ}\text{C}$)	28 ± 3	28 ± 3
Relative humidity (%)	48 ± 10	55 ± 12
j(O ¹ D)(s^{-1})	$1.6 \times 10^{-5} \pm 1.2 \times 10^{-5}$	$6.9 \times 10^{-6} \pm 5.2 \times 10^{-6}$
j(NO ₂)(s^{-1})	$5.4 \times 10^{-3} \pm 2.9 \times 10^{-3}$	$2.9 \times 10^{-3} \pm 1.7 \times 10^{-3}$
O ₃ (ppb)	40 ± 17	87 ± 53
NO ₂ (ppb)	11 ± 5	25 ± 10
CO (ppm)	0.24 ± 0.05	0.85 ± 0.14
VOC reactivity (s^{-1})	3.0 ± 0.7	6.4 ± 1.7
HCHO (ppb)	2.7 ± 1.1	7.4 ± 1.9

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1142 Table 8. Monthly mean and standard deviation of observed data during daytime

1143 (6:00–18:00) under the condition of AOD less than 1 and larger than 1 in August

1144 2012

Observed data	AOD<1	AOD>1
AOD	0.43 ± 0.24	2.0 ± 0.8
PM _{2.5} ($\mu\text{g m}^{-3}$)	26 ± 12	77 ± 47
O ₃ column (Du)	303 ± 4	302 ± 5
Temperature($^{\circ}\text{C}$)	30 ± 4	29 ± 4
Relative humidity (%)	42 ± 16	57 ± 13
j(O ¹ D)(s^{-1})	$1.6 \times 10^{-5} \pm 1.1 \times 10^{-5}$	$1.0 \times 10^{-5} \pm 0.7 \times 10^{-5}$
j(NO ₂)(s^{-1})	$5.6 \times 10^{-3} \pm 2.4 \times 10^{-3}$	$3.8 \times 10^{-3} \pm 1.7 \times 10^{-3}$
O ₃ (ppb)	52 ± 34	68 ± 46
NO ₂ (ppb)	16 ± 7.8	24 ± 9
CO (ppm)	0.47 ± 0.20	0.95 ± 0.47
VOC reactivity (s^{-1})	4.3 ± 1.7	6.2 ± 2.2
HCHO (ppb)	4.0 ± 1.4	6.5 ± 1.9

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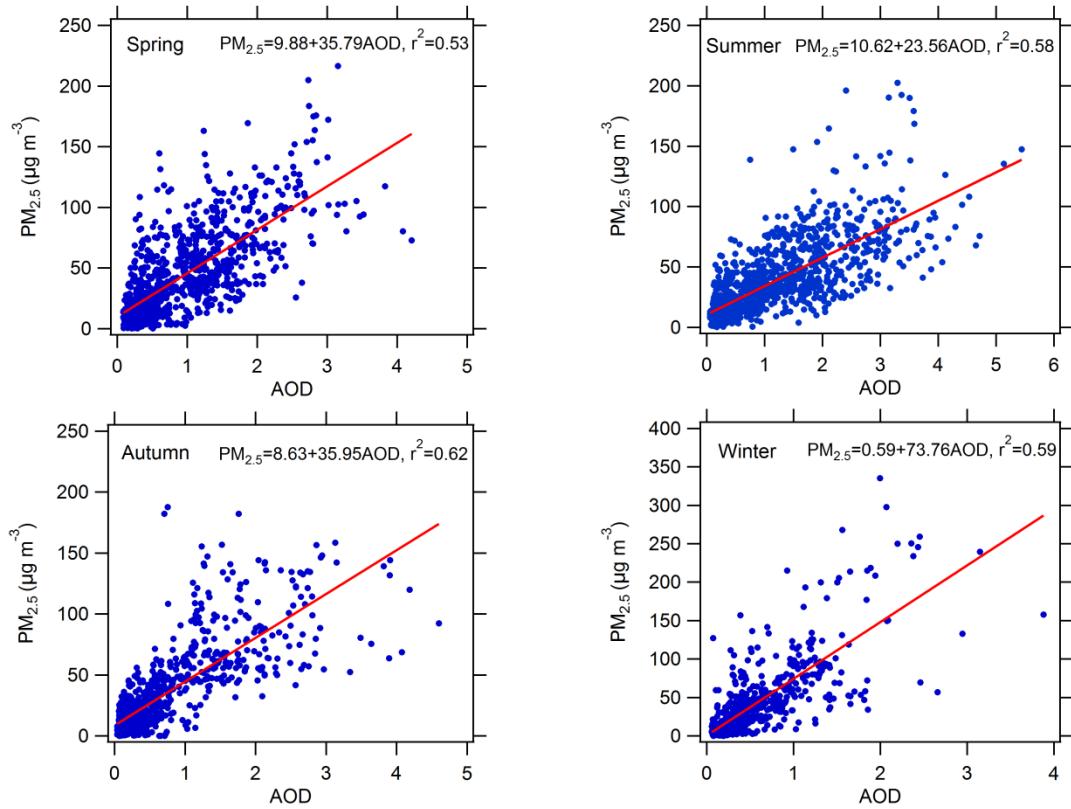
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Figure 1. Scatter plots between AOD at 380nm and PM_{2.5} in four different seasons.

1155 The slope, intercept and determination coefficient (r^2) were calculated.
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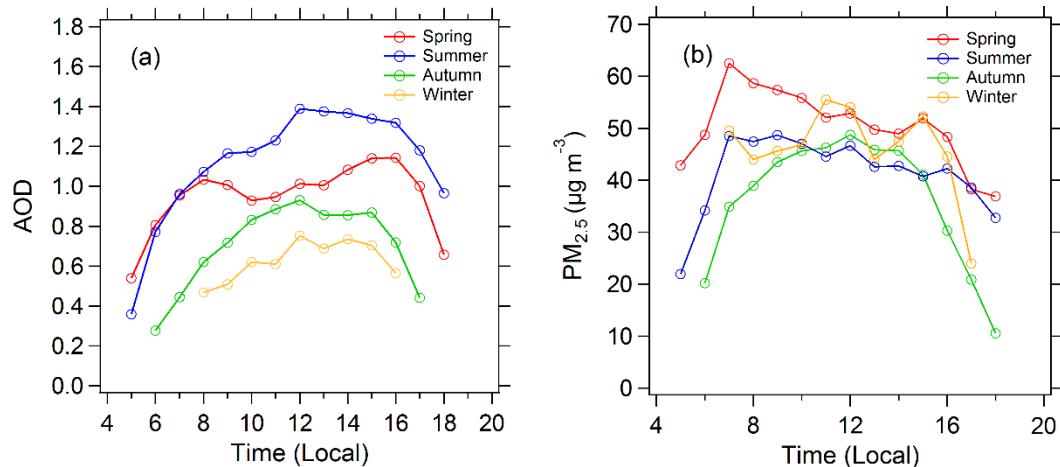
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1174 Figure 2. Diurnal cycles of (a) mean AOD and (b) mean $\text{PM}_{2.5}$ in the four seasons

1175 under cloudless conditions.

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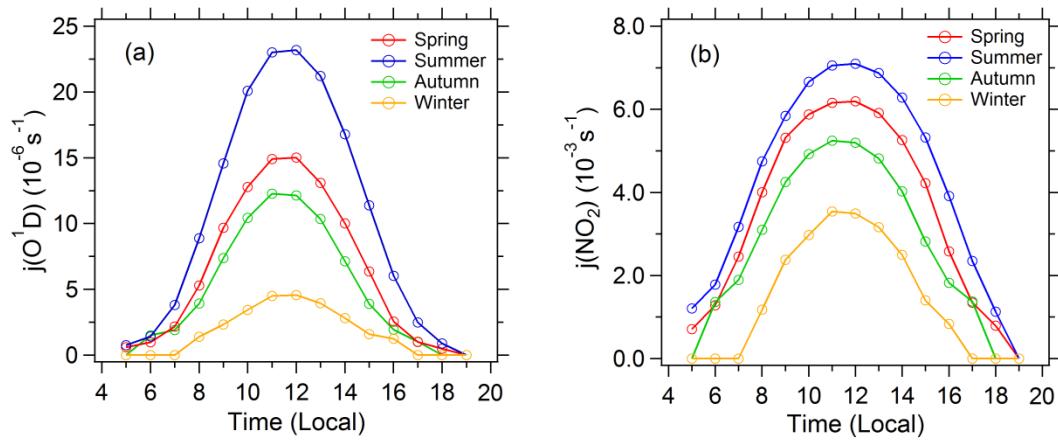
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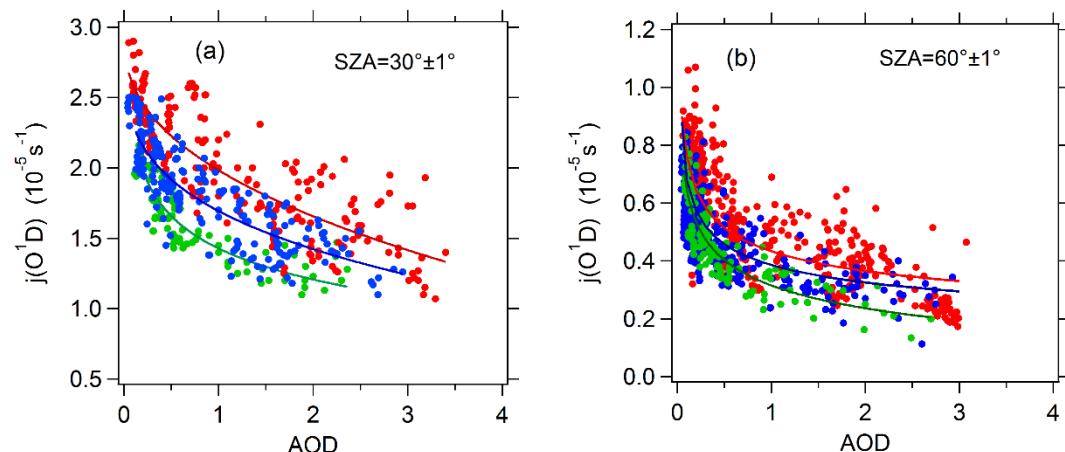
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Figure 3. Diurnal cycles of (a) mean $j(O^1D)$ and (b) mean $j(NO_2)$ in the four seasons

under cloudless conditions.

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1217 Figure 4. Dependence of $j(O^1D)$ on AOD (380nm) at SZA of (a) 30° and (b) 60° and
1218 at different classes of ozone column concentration: 300-330 DU (red), 330-360 DU
1219 (blue), and 360-390 DU (green). The full lines are fitted by exponential function.

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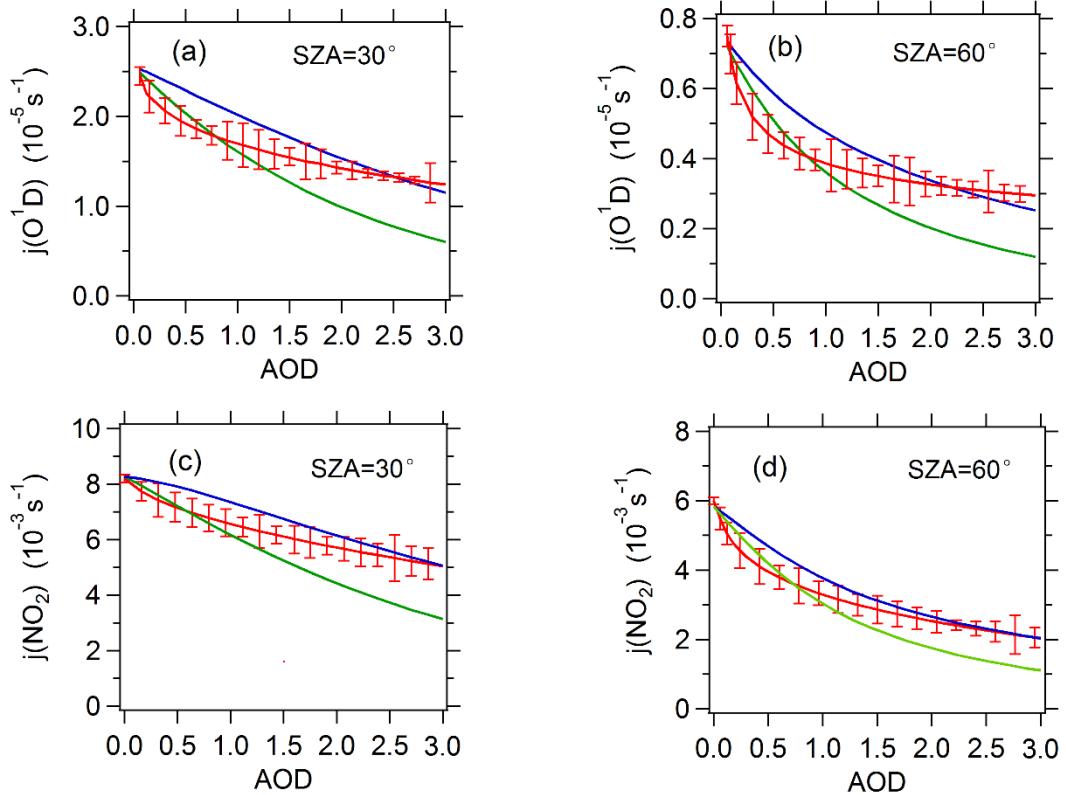
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1233 Figure 5. The relationship between observed or TUV-simulated photolysis

1234 frequencies and AOD (380nm) at SZA of 30° and 60°. For $j(O^1D)$, total ozone

1235 column classification of 330-360 DU is chosen. The red line represents observed

1236 average photolysis frequencies; the blue line and green line represents

1237 TUV-simulated average photolysis frequencies at SSA of 0.95 and 0.85 respectively.

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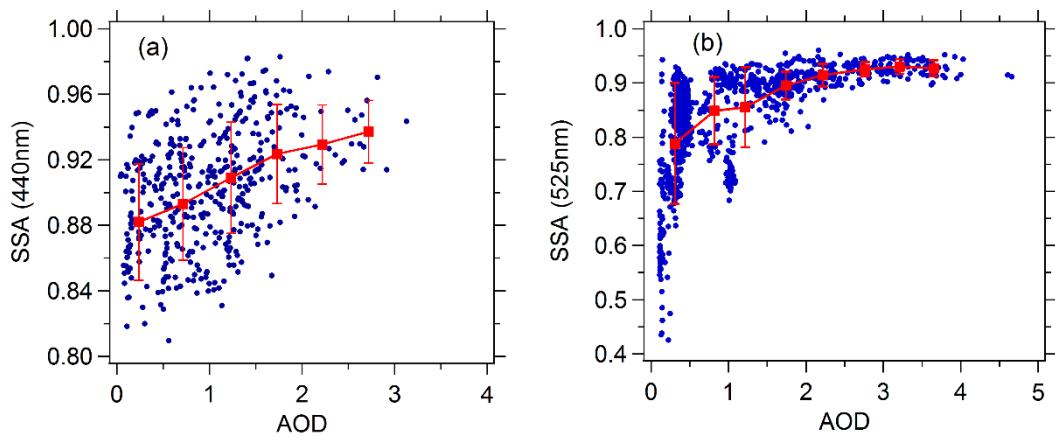
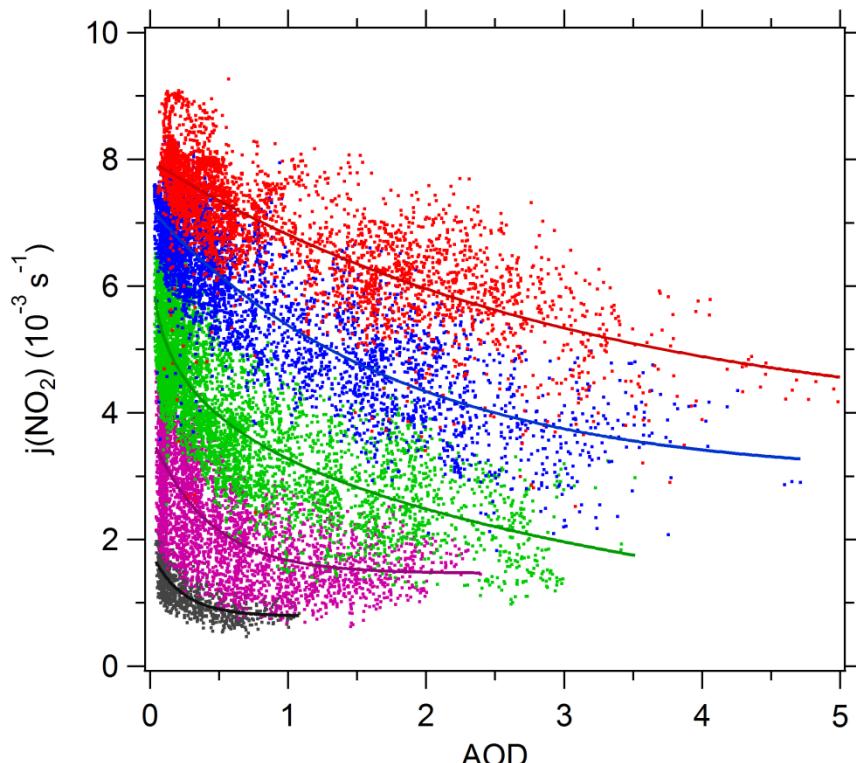


Figure 6. Correlation between SSA and AOD (380nm). (a) SSA (440nm) is acquired from AERONET during 2012-2015. (b) SSA (525nm) is acquired from near-ground measurement campaign in August 2012.



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1267 Figure 7. Dependence of $j(\text{NO}_2)$ on AOD (380nm) at different SZA classes. The
 1268 classes of $\cos(\text{SZA})$ are 0–0.2 (black), 0.2–0.4 (purple), 0.4–0.6 (green), 0.6–0.8
 1269 (blue), and 0.8–1 (red). The full lines are fitted by exponential function.

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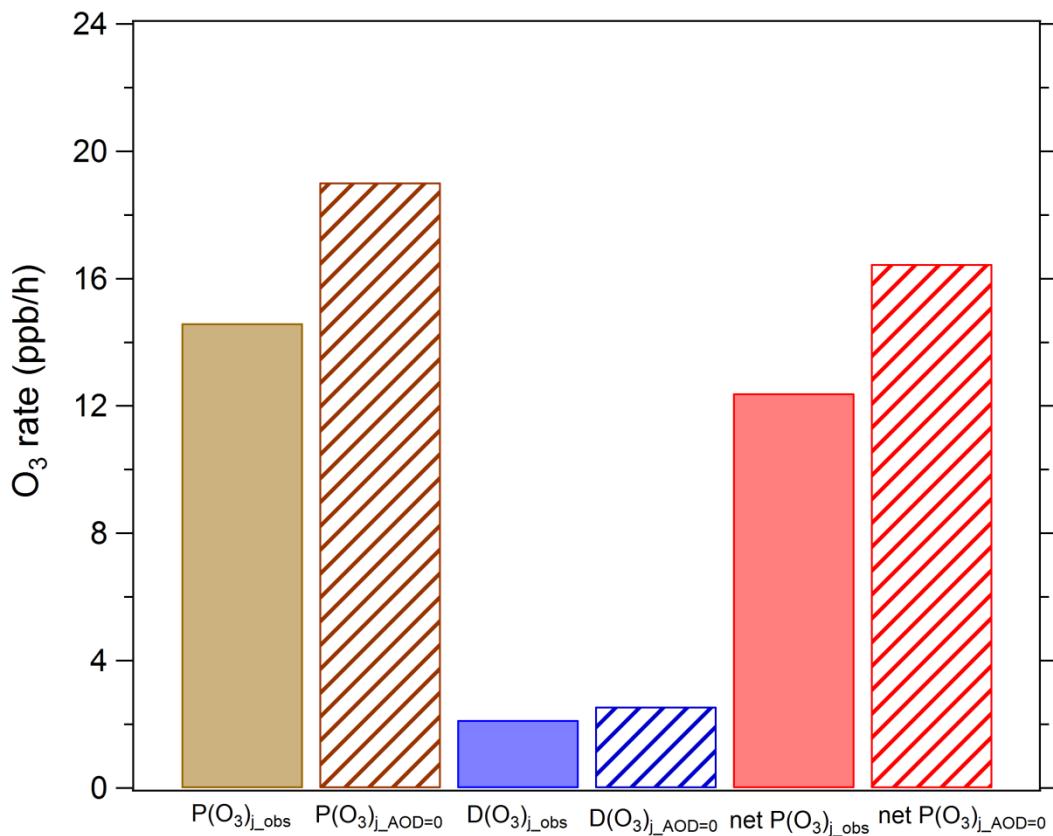
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1286 Figure 8. Mean daytime ozone production and loss terms in August 2012. P(O₃)_j_obs,
1287 D(O₃)_j_obs and net P(O₃)_j_obs represents ozone production rate, ozone loss rate, and
1288 net ozone production rate under observed photolysis frequencies; P(O₃)_j_AOD=0, D(O₃)
1289 _j_AOD=0 and net P(O₃)_j_AOD=0 represents ozone production rate, ozone loss rate, and
1290 net ozone production rate under calculated photolysis frequencies when AOD is
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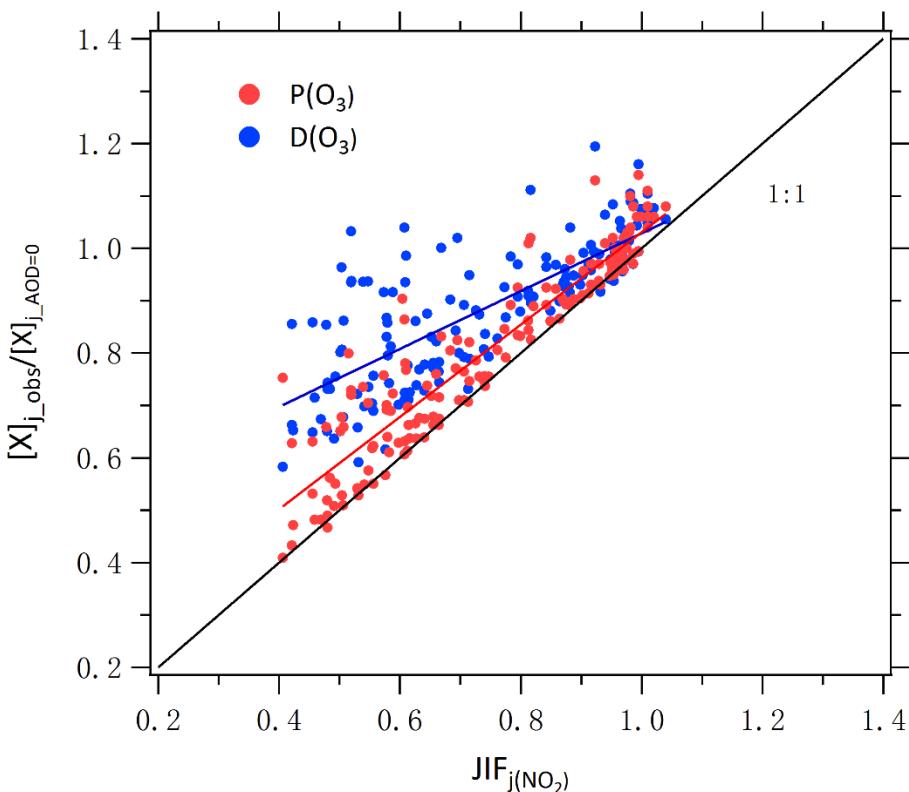
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1299 Figure 9. Correlation between $P(O_3)_{j_obs}/P(O_3)_{j_AOD=0}$ (or $D(O_3)_{j_obs}/D(O_3)_{j_AOD=0}$)
1300 and JIF of $j(NO_2)$. Single data point represent daytime hourly mean value.

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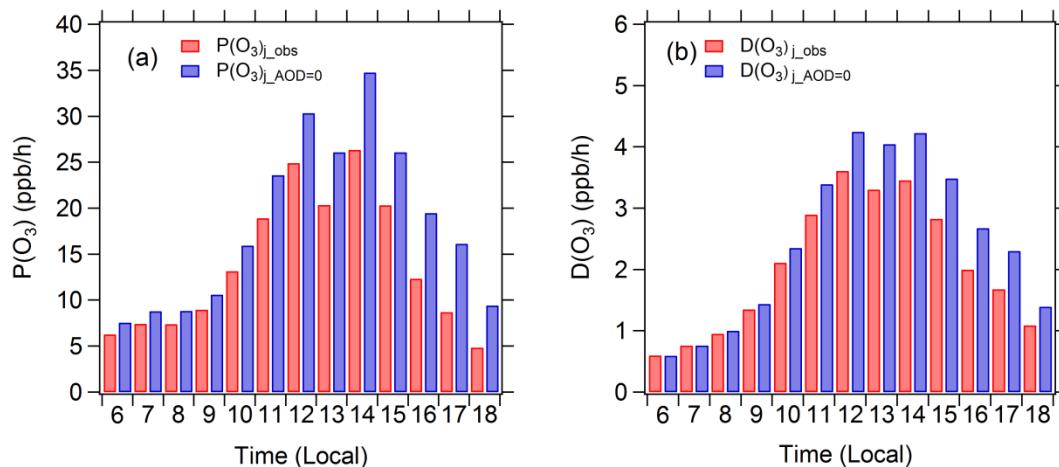
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1316 Figure 10. Diurnal profiles of mean $P(O_3)_j_{obs}$, $P(O_3)_j_{AOD=0}$, $D(O_3)_j_{obs}$, and

1317 $D(O_3)_j_{AOD=0}$ in August 2012 under clear-sky conditions.

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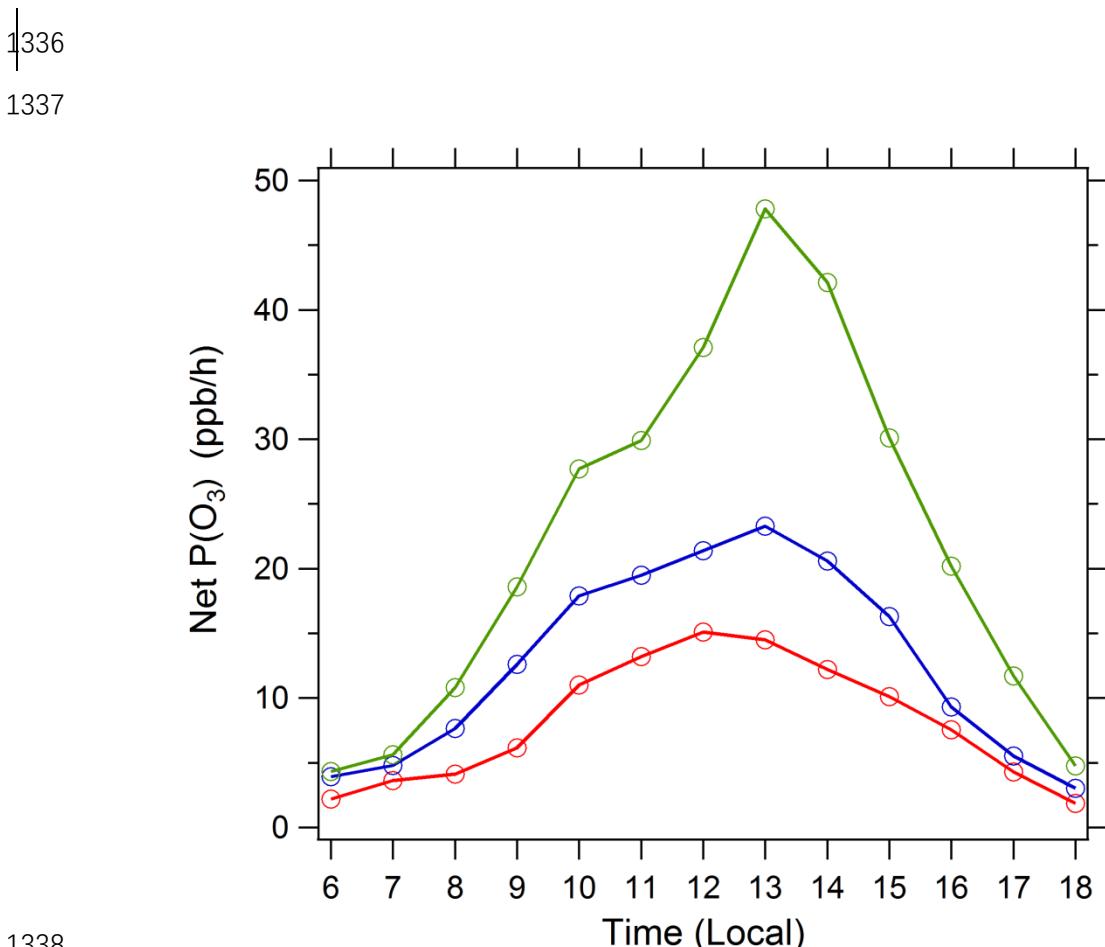
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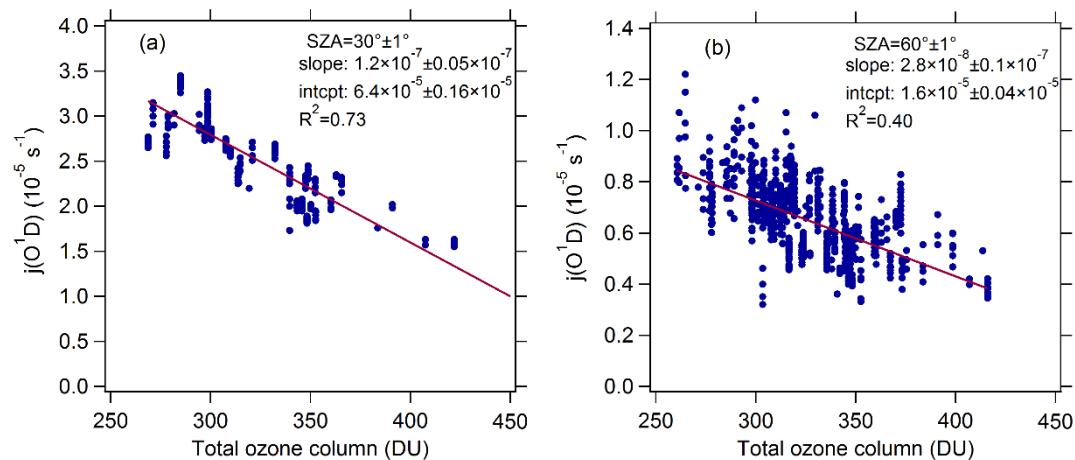
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1357 Figure S1. Dependence of $j(O^1D)$ on AOD (380nm) at low AOD level ($AOD < 0.3$) and
1358 SZA of (a) $30^\circ \pm 1^\circ$ and (b) $60^\circ \pm 1^\circ$, respectively.

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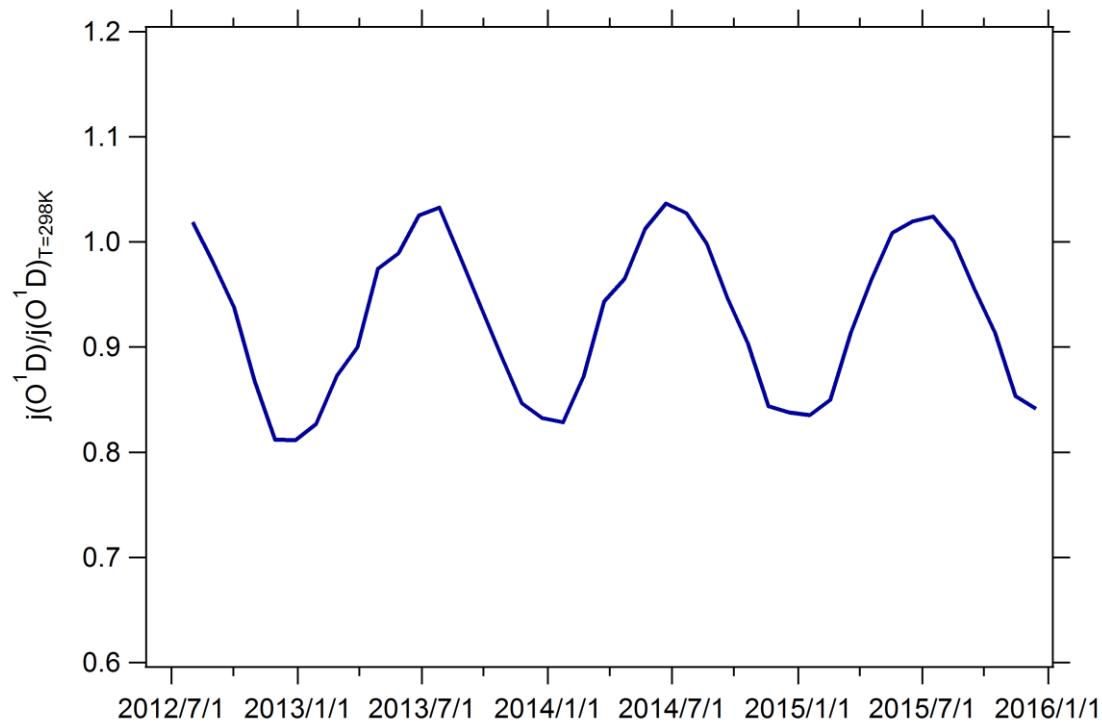
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1379 Figure S2. The time series of the monthly mean ratio of $j(O^1D)$ to $j(O^1D)_{T=298K}$
1380 ($j(O^1D)/j(O^1D)_{T=298K}$) from August 2012 to December 2015.

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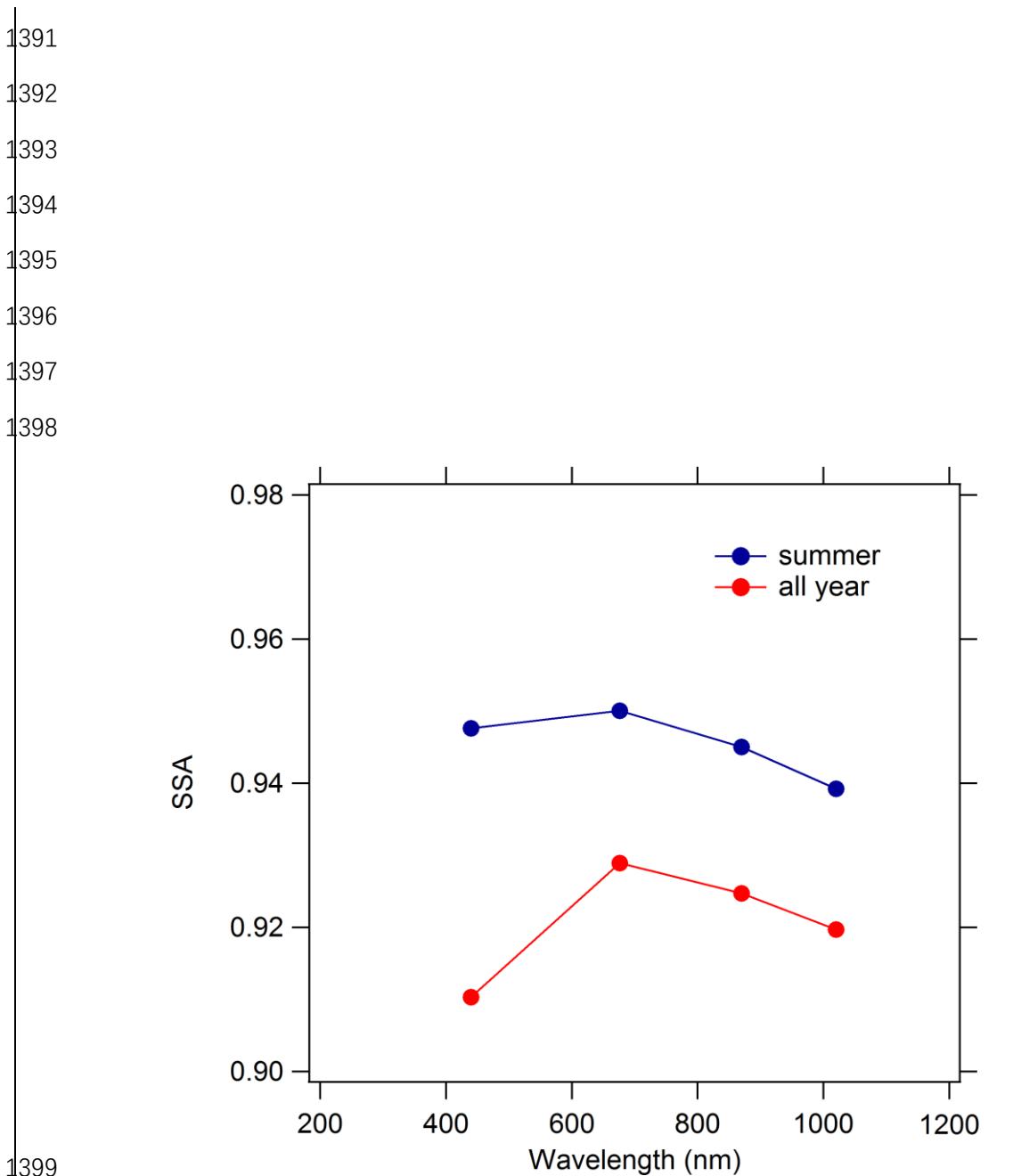
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1400 **Figure S3. The dependence of AERONET based SSA on wavelength.**
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Table S1. The seasonal mean AOD and SSA of AERONET.

<u>Seasons</u>	<u>Spring</u>	<u>Summer</u>	<u>Autumn</u>	<u>Winter</u>
<u>AOD</u>	<u>0.83 ± 0.72</u>	<u>0.99 ± 0.90</u>	<u>0.59 ± 0.70</u>	<u>0.47 ± 0.47</u>
<u>SSA</u>	<u>0.91 ± 0.03</u>	<u>0.94 ± 0.02</u>	<u>0.91 ± 0.03</u>	<u>0.88 ± 0.03</u>

