

1 An optimized tracer-based approach for estimating organic carbon emissions from
2 biomass burning in Ulaanbaatar, Mongolia

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13 Abstract

14 The impact of biomass burning (BB) on atmospheric particulate matter of $<2.5 \mu\text{m}$
15 diameter ($\text{PM}_{2.5}$) at Ulaanbaatar, Mongolia, was investigated using an optimized tracer-
16 based approach during winter and spring, 2017. Integrated 24 h $\text{PM}_{2.5}$ samples were
17 collected on quartz fiber filters using a 30 L min^{-1} air sampler at an urban site in
18 Ulaanbaatar. The aerosol samples were analyzed for organic carbon (OC) and elemental
19 carbon (EC), anhydrosugars (levoglucosan, mannosan, and galactosan), and water-
20 soluble ions. OC was found as the predominant species, contributing 64% and 56% to
21 the quantified aerosol components in $\text{PM}_{2.5}$ in winter and spring, respectively. BB was
22 identified as a major source of $\text{PM}_{2.5}$, followed by dust and secondary aerosols.
23 Levoglucosan/mannosan and levoglucosan/ K^+ ratios indicate that BB in Ulaanbaatar
24 was mainly originated from burning of softwood. Because of the large uncertainty
25 associated with quantitative estimation of OC emitted from BB (OC_{BB}), a novel
26 approach was developed to optimize the OC/levoglucosan ratio for estimating OC_{BB} .
27 The optimum OC/levoglucosan ratio in Ulaanbaatar was obtained by regression analysis
28 between $\text{OC}_{\text{non-BB}}$ ($\text{OC}_{\text{total}} - \text{OC}_{\text{BB}}$) and levoglucosan concentrations that gives the lowest
29 coefficient of determination (R^2) and slope. The optimum OC/levoglucosan ratio was
30 found to be 27.6 and 18.0 for winter and spring, respectively, and these values were
31 applied in quantifying OC_{BB} . It was found that 68% and 63% of the OC were emitted
32 from BB during winter and spring, respectively. This novel approach can also be
33 applied to other study sites to quantify OC_{BB} using their own chemical measurements.
34 In addition to OC_{BB} , sources of $\text{OC}_{\text{non-BB}}$ were also investigated through multivariate
35 correlation analysis. It was found that $\text{OC}_{\text{non-BB}}$ was originated mainly from coal burning,
36 vehicles, and vegetative emissions.

37

38 **Keywords:** Source identification, Biomass burning, Optimized organic-
39 carbon/levoglucosan ratio

40 **1. Introduction**

41 Organic aerosol (OA) contributes a significant fraction (10%–90%) of atmospheric
42 particulate matter (PM), which can affect human health and air quality (Jimenez et al.,
43 2009; Maenhaut et al., 2011; Fu et al., 2012; Allan et al., 2014; Chen et al., 2018). An
44 understanding of the sources of PM is highly relevant for air-quality remediation.
45 Biomass burning (BB) is a major source of organic carbon (OC) in PM_{2.5} (PM with
46 aerodynamic diameter ≤ 2.5 μm) and it may become more significant in the future as air-
47 quality regulations restrict other anthropogenic emissions (Davy et al., 2011; Allan et al.,
48 2014; Sullivan et al., 2019). Coal combustion, thermal **power plants**, and traffic
49 emissions also make potential contributions to the OC content of PM (Watson et al.,
50 2001a, b; Pei et al., 2016; Deshmukh et al., 2019; Haque et al., 2019), modifying PM
51 characteristics such as hygroscopicity, light-attenuating properties, and health impacts
52 (Jung et al., 2009; Sullivan et al., 2019). Previous studies have observed that the toxicity
53 of PM_{2.5} increases with the oxidation potential of BB species because of the water-
54 soluble fraction of OC (Verma et al., 2014).

55 Previous studies have identified and quantified OC emitted from BB (OC_{BB}) using
56 the BB tracers (levoglucosan, mannosan, galactosan, and K⁺). Levoglucosan is
57 produced from the pyrolysis of cellulose at temperatures of $>300^\circ\text{C}$ (Simoneit et al.,
58 1999; Claeys et al., 2010; Maenhaut et al., 2011; Nirmalkar et al., 2015; Achad et al.,
59 2018); and two isomers of levoglucosan, mannosan and galactosan are produced by the
60 burning of hemicellulose (Reche et al., 2012). The atmospheric concentration of
61 levoglucosan is higher **than that** of the two isomers because of the lower content of
62 hemicellulose (20%–30%, dry weight) than cellulose (40%–50%) in softwood and
63 hardwood (Reche et al., 2012; Sharma et al., 2015). Water-soluble K⁺ can also be used

64 as a BB tracer (Pio et al., 2008; Cheng et al., 2013; Nirmalkar et al., 2015; Chen et al.,
65 2018; Chantara et al., 2019). The proportion of these BB tracers in PM depends on
66 various factors such as the type of biomass (softwood, hardwood, crop, grass, etc.),
67 where it is burnt (traditional stoves, fireplaces, field burning, burning in closed
68 chambers, etc.), the type of burning (smoldering, flaming, etc.), and the burning season
69 (Fu et al., 2012; Cheng et al., 2013; Jung et al., 2014). Levoglucosan/mannosan,
70 levoglucosan/K⁺, and OC/levoglucosan ratios were used to identify major biomass types
71 and quantify OC_{BB} (Reche et al., 2012; Cheng et al., 2013; Jung et al., 2014; Chen et al.,
72 2018). However, OC/levoglucosan ratios are quite variable even with the same type of
73 BB because of variations in burning type, place, and season (Cheng et al., 2013;
74 Thepnuan et al., 2019 and references therein). It is therefore essential to optimize the
75 OC/levoglucosan ratio to better estimate OC_{BB}.

76 Ulaanbaatar, with a population of about 1 million, is an atmospheric pollution
77 ‘hotspot’ because of its topography, being situated in the Tuul river valley and
78 surrounded by the Khentei mountains, with a high elevation (1300 m–1949 m above sea
79 level) and large variations in temperature (–28°C to +16°C) and relative humidity
80 (17.7%–72.7%; Table 1; Batmunkh et al., 2013; Jung et al., 2014). As the world’s
81 coldest capital city during winter, it requires additional fuel for space heating. The
82 topography and low-temperature conditions cause an increase in PM concentrations,
83 which are exacerbated by low wind speeds and atmospheric temperature inversions
84 (Jung et al., 2010).

85 **Half of the** residents in Ulaanbaatar lives in 160,000 Gers (traditional Mongolian
86 **dwelling**s) (Guttikunda and Jawahar, 2014). Biomass is used as fuel for cooking and
87 heating in many of low-income Gers at Ulaanbaatar. The common tree species in

88 Mongolia are larch, pine, cedar, spruce, birch; these are mostly softwood
89 (<http://www.fao.org/3/w8302e/w8302e05.htm>; <http://www.fao.org/3/a-am616e.pdf>,
90 excess date 17-12-2019). Each Ger burns an average of 3 m³ of wood per year
91 (Guttikunda, 2008; Zhamsueva et al., 2018). Organic carbon (OC) has severe effects on
92 human health and global climate change (Sun et al., 2019). But there are very few
93 estimates of OC emitted from biomass burning (OC_{BB}) in Ulaanbaatar. Few studies have
94 investigated the chemical characteristics of aerosol in Ulaanbaatar (Jung et al., 2010;
95 Davy et al., 2011; Batmunkh et al., 2013), with none examining the contribution of
96 OC_{BB} and type of biomass. Therefore, this study estimated appropriate concentration of
97 OC_{BB} and identified the type of biomass at Ulaanbaatar, Mongolia.

98 In this study, we quantified the BB tracers levoglucosan, mannosan, galactosan, K⁺,
99 and other chemical species. Potential sources of PM_{2.5} were identified by principal
100 component analysis (PCA), with levoglucosan/K⁺ and levoglucosan/mannosan ratios
101 being used to identify major biomass types. OC_{BB} can be quantified from
102 OC/levoglucosan ratios and levoglucosan concentrations in PM. However, uncertainties
103 of OC_{BB} are high because OC/levoglucosan ratios can vary depending on fuel type,
104 burning conditions, and burning place (Duan et al., 2004; Cheng et al., 2013; Jung et al.,
105 2014). Therefore, it is required to determine the most suitable OC/levoglucosan ratio of
106 BB emissions for estimating appropriate concentration of OC_{BB}. Here, for the first time,
107 optimized OC/levoglucosan ratios were investigated for estimating concentrations of
108 OC_{BB} during winter and spring. OC_{non-BB} sources were also investigated using
109 multivariate correlation analysis with ions and elemental carbon (EC).

110

111 2. Methods

112 2.1 Sampling site and aerosol sampling

113 Aerosol sampling was carried out in Ulaanbaatar during the winter (17 January to
114 03 February) and spring (17 April to 4 May) of 2017, with 24 h periods commencing
115 daily at 11:00 local time. An aerosol sampler was installed on the rooftop of the
116 National Agency for Meteorology and Environmental Monitoring station in Ulaanbaatar
117 (47°92' N, 106°90' E, Fig. 1), 10 m above ground level. The sampling site was located
118 at 8 km–10 km far from two coal based thermal power plants to the west (Chung and
119 Chon, 2014). PM_{2.5} samples were collected on 47 mm diameter quartz fiber filters (Pall-
120 Life Sciences, USA) using an aerosol sampler (Murata Keisokuki Service, Japan) at a
121 flow rate of 30 L min⁻¹. Field blank filter was collected during winter (n=1) and spring
122 (n=1). The quartz fiber filter was loaded in the sampler for 5 minutes without operating
123 a pump. The concentration of all chemical **analytes has** been corrected using blank
124 filters concentration. Sampled filters were wrapped in aluminum foil and heated at
125 550°C for 12 h to remove adsorbed impurities before use and stored at -20°C before
126 and after sampling.

127

128 2.2 Filter analysis

129 A one-fourth part of each quartz fiber filter sample was extracted in 10 mL
130 ultrapure water (resistivity 18.2 MΩ, total OC content < 1 ppb,) under ultrasonication
131 for 30 min. The water extract was then filtered using a syringe filter (Millipore,
132 Millex-GV, 0.45μm) and stored at 4°C pending analysis. Water-soluble cations (K⁺,
133 Na⁺, Ca²⁺, Mg²⁺, and NH₄⁺) were quantified by an ion chromatograph (Dionex ICS
134 5000, Thermo Fisher Scientific, USA). Water-soluble cations were separated using an
135 IonPac CS-12A column (Thermo Fisher Scientific, USA) with 20 mM methanesulfonic

136 acid as eluent at a flow rate of 1.0 mL min⁻¹. Water-soluble anions (Cl⁻, NO₃⁻, and
137 SO₄²⁻) were separated using an IonPac AS-15 column (Thermo Fisher Scientific, USA)
138 with 40 mM KOH as eluent at a flow rate of 1.2 mL min⁻¹. The detection limits for
139 major inorganic ions based on 3σ of blanks were 0.01 μg m⁻³, 0.01 μg m⁻³, and 0.03 μg
140 m⁻³ for NO₃⁻, SO₄²⁻, and NH₄⁺, respectively.

141 Levoglucosan, mannosan, and galactosan were measured by a high-performance
142 anion-exchange chromatograph (Dionex, ICS-5000, Thermo Fisher Scientific, USA)
143 with pulsed amperometric detection involving an electrochemical detector with a gold
144 working electrode. Details of the method are given elsewhere (Jung et al., 2014). In
145 brief, separation involved a CarboPak MA1 (4 × 250 mm, Thermo Fisher Scientific,
146 USA) analytical column and NaOH eluent (360 mM, 0.4 mL min⁻¹). Limits of detection
147 were 3.0 ng m⁻³, 0.7 ng m⁻³, and 1.0 ng m⁻³ for levoglucosan, mannosan, and galactosan,
148 respectively.

149 Aerosol samples were analyzed for OC and EC using a thermal optical OC/EC
150 analyzer (Sunset Laboratory Inc. Forest Grove, OR, USA) with laser transmittance-
151 based correction of pyrolysis. Details of the analyzer and quality-control parameters are
152 reported elsewhere (Jung et al., 2014). In brief, 1.5 cm² punch samples of the quartz
153 fiber filter were placed in a quartz dish inside the thermal desorption oven of the
154 analyzer. OC and EC were quantified using a temperature program developed by the US
155 National Institute for Occupational Safety and Health (NIOSH) in an inert atmosphere
156 (100% He) and in an oxidizing atmosphere (98% He + 2% O₂), respectively. Detection
157 limits of OC and EC were 0.04 and 0.01 μg C m⁻³, and analytical uncertainties of them
158 were 1.3% and 3.7%, respectively.

159

160 2.3. Conditional Probability Function

161 The Conditional Probability Function (CPF) calculates the probability that a source
162 is located within a particular wind direction sector, $\Delta\Theta$:

$$163 \quad CPF = \frac{m_{\Delta\Theta}}{n_{\Delta\Theta}}$$

164 where $n_{\Delta\Theta}$ is the number of times that the wind passed through direction sector $\Delta\Theta$,
165 and $m_{\Delta\Theta}$ is the number of times that the source contribution peaked while the wind
166 passed through sector $\Delta\Theta$ (Ashbaugh et al., 1985). To use CPF with the Ulaanbaatar
167 data, the 24 h averaged source contribution data have been applied to all 1 h wind
168 direction averages recorded at the site for each date. The angular interval $\Delta\Theta$ was set at
169 10° . To calculate $m_{\Delta\Theta}$, the 75th percentiles of source contribution concentrations were
170 counted. CPF is useful in determining the direction of a source from a receptor site;
171 however, it cannot determine the actual location of the source.

172

173 2.4 Principal component analysis

174 In order to identify the source groupings of chemical species in PM_{2.5}, principal
175 component analysis (PCA) was applied. PCA is done using a commercially available
176 software package (SPSS, version 10.0). PCA applies projection dimension reduction
177 methods, converting several concentrations sets into significant sets of columns
178 (principal components, PCs) without damaging the original data. PCA is a widely used
179 statistical technique to quantitatively identify a small number of independent factors
180 among the species concentrations, which can explain the variance of the data, by using
181 the eigenvector decomposition of a matrix of pair-wise correlations. PCA with varimax
182 rotation and retention of principal components having eigenvalues >1.0 was used to

183 identify major species associated with different sources. It was widely used for
184 identification of pollution sources in the atmosphere (Fang et al., 2003, Nirmalkar et al.,
185 2015).

186

187 **3. Results and Discussion**

188 **3.1 Chemical characteristics of PM_{2.5} and source identification**

189 Mass concentrations of carbonaceous aerosol, BB tracers, and water-soluble ions in
190 PM_{2.5} samples collected at Ulaanbaatar during winter and spring of 2017 are
191 summarized in Table 1. OC contributed $64 \pm 5.1\%$ and $56 \pm 6.0\%$ of the quantified
192 aerosol components in PM_{2.5} in winter and spring, respectively (Table 1). Average
193 concentrations of OC during winter were five times those obtained in spring (Fig. 2).

194 **Previously, OC has been observed as major component in PM_{2.5} in Ulaanbaatar during**
195 **winter period (Jung et al., 2010; Batmunkh et al., 2013).** This may be attributed to
196 additional BB emission for home heating, and temperature inversions with low wind
197 speeds (average wind speed of $1.43 \pm 0.73 \text{ m s}^{-1}$; Table 1 and Fig. 3a). OC
198 concentrations decreased with increasing wind speed during winter (Fig. 3a) and spring
199 (Fig. 3b), over all air temperature ranges. The inverse relationship between OC and
200 wind speed during winter (Fig. 3a) and spring (Fig. 3b) suggests a predominance of
201 local sources, with higher wind speeds flushing air pollutants out of the area whereas
202 low wind speeds allow them to accumulate (Khan et al., 2010; Wang et al., 2018).

203 Average concentration of EC during winter ($1.71 \pm 0.58 \mu\text{g m}^{-3}$) was higher than
204 that in spring ($1.11 \pm 0.42 \mu\text{g m}^{-3}$) (Table 1), consistent with general urban observations
205 in cities of China (Ji et al., 2016) and India (Panda et al., 2016). During both winter and
206 spring, EC concentrations at the study site were lower and having different trends

207 compared to those observed in a suburban site ($2.3 \pm 1.0 \mu\text{g m}^{-3}$ and $3.1 \pm 1.5 \mu\text{g m}^{-3}$,
208 respectively) and an urban site ($2.3 \pm 1.0 \mu\text{g m}^{-3}$ and $3.3 \pm 1.2 \mu\text{g m}^{-3}$, respectively) in
209 Shanghai, China (Feng et al., 2009).

210 The potential source direction of EC during winter and spring was west as shown in
211 Fig. 5; this can be explained by the influence of emission from thermal power plants.
212 Correlation of EC was strong with Ca^{2+} during spring as shown in Fig. 4. CPF analysis
213 suggested that potential source direction of EC and Ca^{2+} was similar (Fig. 5). High
214 abundances of Ca^{2+} and EC is observed from stack emission of coal fired thermal power
215 plants (Pei et al., 2016; Zhang et al., 2015). Thus, EC and Ca^{2+} in Ulaanbaatar might be
216 strongly related to emission from thermal power plants.

217 Daily concentrations of levoglucosan, mannosan and galactosan have similar trends
218 during winter and spring (Fig. 2), possibly because of combustion of similar biomass
219 fuels in both seasons. Changes in concentrations of these BB tracers might be attributed
220 to changes in relative proportions of cellulose and hemicellulose in different biomass
221 fuels (Zhu et al., 2015; Nirmalkar et al., 2015). Concentrations of anhydrosugars were
222 four times higher in winter than in spring (Table 1) due to increased heating
223 requirements in winter. The higher relative humidity (58.5%–72.7%) and lower
224 temperature (-10.5°C to -27.8°C ; Table 1) in winter can also contribute to longer
225 atmospheric residence times due to increased levoglucosan stability (Lai et al., 2014).
226 Higher concentrations of BB tracers in winter than spring have previously been
227 observed in Beijing, China, (Liang et al., 2016) and were attributed to meteorological
228 conditions similar to those of Ulaanbaatar. Average K^+ concentration ($0.08 \pm 0.05 \mu\text{g m}^{-3}$)
229 in this study is significantly lower than the K ($0.32 \mu\text{g m}^{-3}$) observed in Ulaanbaatar
230 during 2004-2008 (Davy et al., 2011).

231 Among water-soluble ions, SO_4^{2-} ($9.7 \pm 3.4 \mu\text{g m}^{-3}$) was the most dominant $\text{PM}_{2.5}$
232 species during winter, followed by NH_4^+ ($6.2 \pm 2.4 \mu\text{g m}^{-3}$) and NO_3^- ($4.2 \pm 1.7 \mu\text{g m}^{-3}$),
233 whereas SO_4^{2-} ($1.9 \pm 0.5 \mu\text{g m}^{-3}$) was the dominant species during spring, followed by
234 Ca^{2+} ($0.9 \pm 0.4 \mu\text{g m}^{-3}$) and NH_4^+ ($0.7 \pm 0.3 \mu\text{g m}^{-3}$). The total $\text{SO}_4^{2-} + \text{NH}_4^+ + \text{NO}_3^-$
235 content accounted for 27% and 23% of the total measured chemical species during
236 winter and spring, respectively (Fig. 2 and Table 1). SO_4^{2-} is the most prevalent water-
237 soluble ion in $\text{PM}_{2.5}$ in Wuhan, Guangzhou, and Tianjin (China) due to industrial
238 emissions and coal burning (Gu et al., 2011; Tao et al., 2014; Huang et al., 2016; Pei et
239 al., 2016). This suggests that the higher SO_4^{2-} concentration in Ulaanbaatar may be
240 attributable to emissions from the three major coal-fired thermal power plants near the
241 study site.

242 The atmospheric concentrations of OC ($11\text{--}17 \mu\text{g m}^{-3}$) and levoglucosan (0.46--
243 $0.73 \mu\text{g m}^{-3}$) were higher for samples collected during 27–30 April 2017 than on almost
244 all remaining days in spring (Fig. 2b). Backward atmospheric trajectories based on the
245 Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model provided by
246 the US National Oceanic and Atmospheric Administration (NOAA) Air Resources
247 Laboratory (ARL) indicate that during those days' air masses originated from a region
248 where a significant number of fires were detected [US Fire Information for Resource
249 Management System (FIRMS); National Aeronautics and Space Administration
250 (NASA); Fig. 6a, b)]. Thus, the elevated OC and levoglucosan concentrations during
251 27–30 April might be influenced by long-range transport of BB from north of Mongolia.

252

253 3.2 Principal Component Analysis

254 Principal component analysis (PCA) is a useful tool for reducing the dimensionality

255 of large aerosol datasets to principal components using varimax rotation for source
256 identification (Cao et al., 2005; Lin et al., 2018; Nirmalkar et al., 2019). Four principal
257 components (PCs) in winter and three in spring were identified with eigenvalues >1
258 after varimax rotation explaining 96% and 92%, respectively, of the total variance
259 (Tables 2 and 3). The PCs were categorized on the basis of loadings of chemical
260 components as follows. In winter, PC1 includes BB characterized by high loadings of
261 levoglucosan, mannosan, and galactosan; PC2 includes dust characterized by Ca^{2+} and
262 Mg^{2+} content; PC3 includes secondary formation characterized by SO_4^{2-} , NO_3^- , and
263 NH_4^+ content; and PC4 includes fossil fuel combustion characterized by EC. In spring,
264 PC1 includes BB (levoglucosan, mannosan, and galactosan); PC2 includes dust (Ca^{2+}
265 and Mg^{2+}) and fossil fuel combustion (EC); and PC3 includes secondary formation
266 (SO_4^{2-} , NO_3^- , and NH_4^+). The PCA results show that the chemical components of $\text{PM}_{2.5}$
267 in Ulaanbaatar were mainly affected by BB during winter and spring. Further, OC was
268 primarily influenced by BB because it correlated well with the total variance of PC1
269 during winter (0.82; Table 2) and spring (0.77; Table 3).

270

271 3.3 Relationship among BB tracers

272 The correlations among the three BB tracers levoglucosan, mannosan, and
273 galactosan are shown in Fig. 7a (winter) and 7b (spring). The correlations between
274 levoglucosan and mannosan and between levoglucosan and galactosan are strong during
275 winter ($R^2 = 0.99$ for both pairs) and spring ($R^2 = 0.95$ and 0.83 , respectively; Fig. 7a, b).
276 Concentrations of levoglucosan and OC are strongly correlated during both winter ($R^2 =$
277 0.78) and spring ($R^2 = 0.86$; Fig. 8a), suggesting that a major fraction of OC might be
278 originated from BB in Ulaanbaatar. The similar strong correlation and steep slope

279 observed in OC–levoglucosan plots for PM collected in Chiang Mai Province (Thailand)
280 and Daejeon (Korea) were attributed mainly to BB (Jung et al., 2014; Thepnuan et al.,
281 2019).

282 Fine mode K^+ is considered as biomass burning tracer in previous studies (Louie et
283 al., 2005; Deshmukh et al., 2011; Cheng et al., 2013). The moderate correlation between
284 levoglucosan and K^+ concentrations ($R^2 = 0.68$) in winter indicates that they are
285 produced from similar sources (Fig. 8b), with BB contributing most of the K^+ . However,
286 the correlation between levoglucosan and K^+ was weak in spring ($R^2 = 0.49$; Fig. 8b).
287 Because K^+ is typically emitted at a higher mass fraction in flaming phase combustion
288 compared to smoldering (Lee et al., 2010), smoldering combustion tends to have higher
289 levoglucosan/ K^+ emission ratio compared to flaming combustion (Schkolnik et al., 2005;
290 Gao et al., 2003). High levoglucosan/ K^+ ratio was observed during winter (8.92)
291 compared to spring (4.21) in this site. Thus, weak correlation between levoglucosan and
292 K^+ concentrations at Ulaanbaatar in spring can be explained by mixed burning condition
293 such as smoldering and flaming.

294 OC and K^+ concentrations correlated well during winter ($R^2 = 0.79$; Fig. 9a) and
295 spring ($R^2 = 0.73$; Fig. 9b), suggesting that they might be originated from similar
296 sources. Because most of the aerosol particles emitted from BB belong to $PM_{2.5}$, the
297 correlation between OC and K^+ as well as levoglucosan suggests that BB is one of the
298 potential sources of OC in winter and spring. Because biomass fuel is burned in
299 traditional stoves with no pollution control devices in Ulaanbaatar (Batmunkh et al.,
300 2013), soil and ash particles are entrained in convective processes and uplifted in the
301 atmosphere together with smoke particles (Deshmukh et al., 2011; Nirmalkar et al.,
302 2019).

303

304 3.4 Tracing the source of BB aerosol

305 OC is a major contributor of the quantified aerosol components in PM_{2.5} in
306 Ulaanbaatar during spring and winter (Table 1). To quantify the OC_{BB}, it is necessary to
307 identify the BB fuel type. Several investigators used levoglucosan/mannosan and
308 levoglucosan/K⁺ ratios to identify BB fuel types (Puxbaum et al., 2007; Cheng et al.,
309 2013; Jung et al., 2014; Chen et al., 2018; Thepnuan et al., 2019).

310 The levoglucosan/mannosan ratio is source-specific and can be used to identify BB
311 fuel types due to the unique cellulose and hemicellulose compositions of different
312 biomass fuels (Zhang et al., 2007; Cheng et al., 2013). A previous study suggested that
313 the levoglucosan/mannosan ratio is strongly dependent on wood type, rather than on the
314 site where the wood is grown (Cheng et al., 2013). Therefore, the
315 levoglucosan/mannosan ratio was used to trace the type of wood burnt during winter
316 and spring for indoor heating and cooking purposes. Previous studies have used
317 levoglucosan/mannosan ratios to investigate the BB fuel types (Cheng et al., 2013; Jung
318 et al., 2014).

319 However, the levoglucosan/mannosan ratio can't distinguish crop residuals (29 ± 15)
320 (Sheesley et al., 2003, Sullivan et al., 2008, Engling et al., 2009, Oanh et al., 2011) and
321 hardwood (28 ± 28) (Fine et al. 2001, 2002, 2004a, b; Engling et al., 2006; Schmidl et
322 al., 2008; Bari et al., 2009; Goncalves et al., 2010) due to the overlap of ratios between
323 these fuel types (Cheng et al., 2013; Fine et al. 2001, 2002, 2004a, b; Engling et al.,
324 2006). However, levoglucosan/K⁺ ratio can distinguish between the two groups (Jung et
325 al., 2014, Chen et al., 2018). Both levoglucosan/mannosan and levoglucosan/K⁺ ratios
326 are therefore useful in distinguishing various types of fuel (Cheng et al., 2013; Puxbaum

327 et al., 2007).

328 A levoglucosan/mannosan–levoglucosan/K⁺ scatter plot based on results of the
329 present and previous studies is shown in Fig. 10, using data from Schauer et al. (2001),
330 Fine et al. (2001, 2002, 2004a, b), and Engling et al. (2006) for hardwood grown in the
331 USA; Schauer et al. (2001), Hays et al. (2002), Fine et al. (2001, 2002, 2004a, b), and
332 Engling et al. (2006) for US softwood; Schmidl et al. (2008), Bari et al. (2009) and
333 Goncalves et al. (2010) for hardwood grown in Europe; Iinuma et al. (2007), Schmidl et
334 al. (2008), and Goncalves et al. (2010) for European softwood; Engling et al. (2006) and
335 Sullivan et al. (2008) for needles and duff found in the USA; Sullivan et al. (2008) for
336 US grass; and from Sheesley et al. (2003), Sullivan et al. (2008), Engling et al. (2009)
337 and Oanh et al. (2011) for Asian rice straw.

338 The average levoglucosan/mannosan ratio was 3.6 ± 0.2 (range: 3.4 – 4.1) in winter
339 and 4.1 ± 1.0 (2.12 – 7.05) in spring, whereas the levoglucosan/K⁺ ratio was 8.9 ± 1.8
340 (5.5 – 12.4) in winter and 4.2 ± 2.1 (0.58 – 7.49) in spring at the study site (Fig. 10),
341 within the ranges reported for softwood burning sources (2.5 – 6.7 and 4.6 – 261,
342 respectively) (Fine et al., 2001; Schauer et al., 2001; Fine et al., 2002, 2004a, b; Hays et
343 al., 2002; Engling et al., 2006; Iinuma et al., 2007; Schmidl et al., 2008; Goncalves et al.,
344 2010; Cheng et al., 2013). During winter and spring, the levoglucosan/K⁺ and
345 levoglucosan/mannosan ratios in Ulaanbaatar appeared in the softwood region (Fig. 10).

346 Therefore, softwood burning seems to be the major source of BB aerosol in
347 Ulaanbaatar during both winter and spring, consistent with previously reported
348 softwood-burning emissions from fireplaces of northern and southern regions of the
349 USA (Fine et al., 2001, 2002), from household combustion in Zhengzhou, China (Chen
350 et al., 2018), and from stove wood combustion in the mid-European region (Austria;

351 Schmidl et al., 2008).

352

353 3.5 Optimization of OC/levoglucosan ratio for estimating OC_{BB} emission

354 OC_{BB} was estimated by multiplying OC/levoglucosan ratio and levoglucosan
355 concentration. Previous studies have used the OC/levoglucosan ratio obtained from
356 sources of BB aerosol to estimate OC_{BB}. A ratio of 7.35 reported for burning of four
357 types of US hardwood (Fine et al., 2002) was used for estimating OC_{BB} at four
358 background sites in Europe (Puxbaum et al., 2007). Later, mean value of 11.2 of
359 OC/levoglucosan ratio derived from ratios ranged between 4.5 – 24.6 was used for
360 estimating OC_{BB} in the UK (Harrison et al., 2012). However, such estimates may not be
361 accurate as the OC/levoglucosan ratio is highly variable in BB emissions. For example,
362 the average OC/levoglucosan ratio from softwood burning (23.8) is much higher than
363 that of hardwood burning (7.35) (Fine et al., 2002; Schmidl et al., 2008), differences are
364 more than ten-fold among studies of softwood-burning OC/levoglucosan ratios (Fine et
365 al., 2002; Hays et al., 2002; Engling et al., 2006; Iinuma et al., 2007; Goncalves et al.,
366 2010). Combustion conditions may also significantly influence OC/levoglucosan ratios.
367 For example, the OC/levoglucosan ratio varied by a factor of about seven between
368 burning the same wood (Loblolly pine) in a fireplace (27.6; Fine et al., 2002) and in a
369 stove (3.4; Fine et al., 2004b). Therefore, it is necessary to optimize the
370 OC/levoglucosan ratio for use in estimating OC_{BB}.

371 This study has used an optimized OC/levoglucosan ratio to estimate precise
372 concentration of OC_{BB} for the Ulaanbaatar study site. We have used a range of different
373 OC/levoglucosan ratios obtained from previous literatures (Fig. 11) for regression
374 analysis with measured levoglucosan concentrations to estimate optimum

375 OC/levoglucosan ratio (Fig. 12a, b). First, candidate OC_{BB} (Fig. 11) in this study was
376 estimated from OC/levoglucosan ratios for softwood burning in a previous chamber
377 experiments (Cheng et al., 2013; Schauer et al., 2001; Hays et al., 2002; Fine et al.,
378 2001, 2002, 2004a, b; Engling et al., 2006; Iinuma et al., 2007; Schmidl et al., 2008;
379 Goncalves et al., 2010, Fig 11) and measured levoglucosan concentration at this site.
380 Second, OC_{non-BB} concentration was calculated by subtracting OC_{BB} from corresponding
381 total OC. If calculated OC_{non-BB} doesn't contain OC_{BB} , both regression slope and R^2
382 between OC_{non-BB} versus levoglucosan will be close to zero. As shown in Fig. 12a and
383 12b, the lowest R^2 and regression slope were observed when OC/levoglucosan ratios of
384 27.6 and 18.0 in winter and spring, respectively. Thus, the optimized OC/levoglucosan
385 ratios for our site were determined to be 27.6 and 18.0 in winter and spring, respectively.

386 During winter higher optimum ratio of OC/levoglucosan might be due to
387 incomplete combustion during smoldering phenomena. As smoldering fires are
388 characterized by lower temperatures and thus have lower combustion efficiency, they
389 release more un-combusted condensable products, resulting in the production of more
390 unbroken organic compounds (Engling et al., 2006). Smoldering combustion generally
391 leads to increased emissions of volatile organic compounds (VOCs) and particulate
392 organic matter (OM) (Obrist et al., 2007). In contrast, the relatively lower optimum ratio
393 of OC/levoglucosan during spring might be due to the higher combustion efficiency
394 during flaming phenomena.

395 The OC_{BB} concentrations at the Ulaanbaatar study site were calculated from the
396 optimized OC/levoglucosan ratios (winter: 27.6 and spring: 18.0) and measured
397 levoglucosan concentrations. The OC_{BB} concentration was estimated to be 33.1 ± 11.9
398 $\mu\text{g C m}^{-3}$ (range 16.0–58.5 $\mu\text{g C m}^{-3}$) and $5.64 \pm 3.29 \mu\text{g C m}^{-3}$ (range 0.57–13.1 $\mu\text{g C}$

399 m^{-3}), accounting for 68% and 63% of the total OC in winter and spring, respectively
400 (Fig. 13). The average of previously published OC/levoglucosan ratios, 10.1 ± 7.9
401 (range 1.90 – 27.6), gives an estimated OC_{BB} concentration of $12.1 \pm 4.4 \mu\text{g C m}^{-3}$
402 (range 5.9–21.4 $\mu\text{g C m}^{-3}$) and $3.2 \pm 1.8 \mu\text{g C m}^{-3}$ ($0.32\text{--}7.34 \mu\text{g C m}^{-3}$) in winter and
403 spring, respectively. Their values are 2.7 (winter) and 1.8 (spring) times lower than
404 values estimated using our optimized OC/levoglucosan ratio.

405 Our estimated contribution of OC_{BB} was higher than that in Daejeon, South Korea
406 (24%–68% of total OC, mean $45\% \pm 12\%$; Jung et al., 2014) and Beijing, China (50%
407 of total OC; Cheng et al., 2013), where BB aerosols are produced mainly by the burning
408 of crop residues. The contribution of OC_{BB} to total OC is 57% and 31% during heating
409 (average temperature 0.6°C) and non-heating (average temperature 14°C) seasons in
410 Krynica Zdroj, Poland (Klejnowski et al., 2017), significantly lower than that of
411 Ulaanbaatar during both winter (average temperature -21°C) and spring (average
412 temperature 6°C). Such high concentrations of OC_{BB} in Ulaanbaatar and Krynica Zdroj
413 are likely due to intense wood burning for heating during winter.

414

415 3.6 Tracing sources of $\text{OC}_{\text{non-BB}}$

416 High concentration of $\text{OC}_{\text{non-BB}}$ was found during winter compared to spring (Fig.
417 13). Elevated $\text{OC}_{\text{non-BB}}$ could be attributed to enhanced emission from combustions and
418 favorable meteorological conditions (cold temperatures and inversion conditions, etc.)
419 during the winter. There is strong correlation between $\text{OC}_{\text{non-BB}}$ and SO_4^{2-} , NH_4^+ , and K^+
420 in winter and $\text{OC}_{\text{non-BB}}$ and NO_3^- , Na^+ , K^+ , Mg^{2+} , Ca^{2+} , and EC in spring (Table 4).
421 Residential combustion of coal emits significant amounts of OC, EC, and inorganic
422 species (SO_4^{2-} and metals) due to incomplete combustion and lack of pollution control

423 devices (Garcia et al., 1992; Li et al., 2016; Watson et al., 2001a, b). Garcia et al. (1992)
424 studied emissions of volatile organic compounds from coal burning and vehicle engines.

425 In Ulaanbaatar, the use of wood and coal for cooking and heating, and emissions
426 from old vehicles are reported as potential sources of OC (Batmunkh et al., 2013;
427 Zhamsueva et al., 2018). The three thermal power plants in Ulaanbaatar are point
428 sources for emissions of carbonaceous aerosol (Batmunkh et al., 2013), burning ~5
429 million tons of coal per year (Batmunkh et al., 2013). High concentrations of anions
430 (SO_4^{2-} and NO_3^-) and cations (NH_4^+ and Na^+) are reported in China (Zhou et al., 2003),
431 the USA (Caiazzo et al., 2013), Brazil (Flues et al., 2002), India (Guttikunda and
432 Jawahar, 2014), Korea (Park and Kim, 2004; Park et al., 2015), and Spain (Alastuey et
433 al., 1999) near coal-fired thermal power plants. Emissions of volatile organic
434 compounds from vegetation have also been observed in previous studies (Fehsenfeld et
435 al., 1992; Shao et al., 2001; Acton et al., 2016). The correlations of $\text{OC}_{\text{non-BB}}$ with ions
436 and EC are thus likely due to volatile organic compounds emitted from coal-burning
437 and vehicles, and vegetative emissions.

438

439 **4. Conclusions**

440 BB was identified as a major source of the quantified aerosol components in $\text{PM}_{2.5}$
441 in Ulaanbaatar, Mongolia, during the winter and spring of 2017, based on PCA. OC was
442 the major component of the quantified aerosol components during the entire sampling
443 period, winter and spring. For determination of OC_{BB} , the fuel type must be identified
444 and levoglucosan/mannosan and levoglucosan/ K^+ ratios obtained from previous studies
445 and our on-site measurements were used for this purpose.

446 Softwood burning was identified as a major source of OC_{BB} . However,

447 OC/levoglucosan ratios from softwood burning are highly variable, and an optimum
448 ratio was derived by regression analysis between daily concentrations of $OC_{\text{non-BB}}$ and
449 levoglucosan, yielding values of 27.6 and 18.0 for winter and spring, respectively. The
450 application of these ratios indicates that 68% and 63% of the OC originated from BB
451 during winter and spring, respectively, which is about double that estimated using
452 average values of previous studies. The atmospheric concentration of OC_{BB} was higher
453 in winter than in spring **mainly due to** additional BB for heating and cooking. BB
454 aerosols in Ulaanbaatar originate mainly from local softwood burning. The approach
455 developed here may be applied elsewhere for screening region-specific
456 OC/levoglucosan ratios for estimating atmospheric appropriate concentrations of OC_{BB} ,
457 aiding the establishment of BB control measures.

458

459 **Author contribution**

460 Jinsang Jung and Tsatsral Batmunkh designed the study and carried out the field
461 work. Jinsang Jung performed chemical analyses and quality-control measures. Jayant
462 Nirmalkar wrote the manuscript under the guidance of Jinsang Jung. All authors
463 commented on and discussed the manuscript.

464

465 **Competing interests**

466 The authors declare that they have no conflict of interests.

467

468 **Acknowledgments**

469 This work was funded by a grant (19011057) from the Korea Research Institute of

470 Standards and Science (KRISS) under the Basic R&D Project of Quantification of local
471 and long-range transported pollutants during a severe haze episode over the Korean
472 Peninsula. The authors gratefully acknowledge the NOAA Air Resources Laboratory for
473 the provision of the HYSPLIT transport and dispersion model and access to the READY
474 website (<http://www.arl.noaa.gov/ready.html>) and the Fire Information for Resource
475 Management System (FIRMS) of the National Aeronautics and Space Administration
476 (NASA), United States (<https://firms.modaps.eosdis.nasa.gov/alerts/>) used in this study.

477

478 **Data availability**

479 The data used in this study are available from the corresponding author upon
480 request (jsjung@kriss.re.kr).

481

482

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Table 1. Concentrations ($\mu\text{g m}^{-3}$) of organic carbon, elemental carbon, levoglucosan, mannosan, galactosan, and water-soluble ions in $\text{PM}_{2.5}$ samples collected from Ulaanbaatar, Mongolia, during the winter (n = 17) and spring (n = 17) of 2017.

| | OC | EC | Levoglucosan | Mannosan | Galactosan | Cl ⁻ | SO ₄ ²⁻ | NO ₃ ⁻ | Na ⁺ | NH ₄ ⁺ | K ⁺ | Mg ²⁺ | Ca ²⁺ | Temperature (°C) | Wind Speed (m sec ⁻¹) | RH (%) |
|---------------|-------|------|--------------|----------|------------|-----------------|-------------------------------|------------------------------|-----------------|------------------------------|----------------|------------------|------------------|---------------------|-----------------------------------------|-----------|
| Winter | | | | | | | | | | | | | | | | |
| Mean | 49.06 | 1.71 | 1.20 | 0.33 | 0.24 | 1.69 | 9.74 | 4.17 | 0.64 | 6.18 | 0.13 | 0.05 | 0.60 | -20.8 | 1.36 | 66.1 |
| SD | 17.32 | 0.58 | 0.43 | 0.13 | 0.09 | 0.76 | 3.37 | 1.69 | 0.44 | 2.42 | 0.04 | 0.02 | 0.24 | 4.74 | 0.73 | 4.56 |
| Min | 24.62 | 0.79 | 0.58 | 0.15 | 0.10 | 0.26 | 2.17 | 0.76 | 0.10 | 3.16 | 0.08 | 0.02 | 0.22 | -27.8 | 0.41 | 58.5 |
| Max | 79.07 | 3.34 | 2.12 | 0.61 | 0.43 | 2.89 | 16.06 | 7.51 | 1.34 | 11.59 | 0.18 | 0.08 | 1.04 | -10.5 | 3.55 | 72.7 |
| Spring | | | | | | | | | | | | | | | | |
| Mean | 8.50 | 1.11 | 0.31 | 0.08 | 0.04 | 0.30 | 1.90 | 0.70 | 0.13 | 0.74 | 0.08 | 0.04 | 0.93 | 6.11 | 2.60 | 35.1 |
| SD | 3.55 | 0.42 | 0.18 | 0.04 | 0.02 | 0.11 | 0.50 | 0.32 | 0.04 | 0.28 | 0.05 | 0.02 | 0.36 | 6.16 | 0.79 | 13.9 |
| Min | 2.80 | 0.60 | 0.03 | 0.01 | 0.00 | 0.11 | 1.04 | 0.10 | 0.07 | 0.33 | 0.02 | 0.02 | 0.48 | -1.52 | 1.64 | 17.8 |
| Max | 16.63 | 2.03 | 0.73 | 0.15 | 0.08 | 0.51 | 3.02 | 1.40 | 0.21 | 1.47 | 0.22 | 0.08 | 1.61 | 15.9 | 4.56 | 65.2 |

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835 Table 2. Source identification of chemical species using principal component (PC) analysis and varimax rotation at Ulaanbaatar,
 836 Mongolia, during winter of 2017.

| Winter Chemical species | Component | | | |
|-------------------------------|--------------------------|---------------|------------------------------|---------------------------------|
| | PC1 (Biomass Burning) | PC2 (Dust) | PC3 (Secondary formation) | PC4 (Fossil fuel combustion) |
| Levoglucosan | 0.96 | -0.06 | 0.24 | 0.06 |
| Mannosan | 0.95 | -0.08 | 0.27 | 0.06 |
| Galactosan | 0.95 | -0.07 | 0.28 | 0.04 |
| Cl ⁻ | 0.19 | 0.94 | -0.05 | -0.07 |
| SO ₄ ²⁻ | 0.43 | 0.01 | 0.88 | 0.09 |
| NO ₃ ⁻ | 0.28 | 0.20 | 0.87 | 0.20 |
| Na ⁺ | -0.27 | 0.87 | -0.33 | -0.17 |
| NH ₄ ⁺ | 0.48 | -0.12 | 0.86 | 0.07 |
| K ⁺ | 0.70 | 0.11 | 0.61 | 0.25 |
| Mg ₂ ⁺ | -0.15 | 0.90 | 0.25 | 0.26 |
| Ca ₂ ⁺ | -0.12 | 0.92 | 0.19 | 0.24 |
| OC | 0.82 | -0.17 | 0.47 | 0.07 |
| EC | 0.14 | 0.14 | 0.19 | 0.95 |
| Eigenvalues | 4.54 | 3.44 | 3.30 | 1.20 |
| % of Variance | 34.95 | 26.49 | 25.37 | 9.21 |
| Cumulative % | 34.95 | 61.44 | 86.81 | 96.02 |

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839 Table 3. Source identification of chemical species using PCA and varimax rotation at Ulaanbaatar, Mongolia, during spring of 2017.

| Spring | Component | | |
|-------------------------------|--------------------------|---------------------------------------------|------------------------------|
| | PC1 (Biomass Burning) | PC2 (Dust and Fossil fuel combustion) | PC3 (Secondary formation) |
| Levoglucosan | 0.88 | 0.13 | 0.39 |
| Mannosan | 0.94 | 0.00 | 0.30 |
| Galactosan | 0.95 | -0.11 | 0.20 |
| Cl ⁻ | 0.81 | 0.32 | -0.03 |
| SO ₄ ²⁻ | 0.18 | 0.12 | 0.93 |
| NO ₃ ⁻ | 0.59 | 0.54 | 0.52 |
| Na ⁺ | 0.08 | 0.91 | -0.09 |
| NH ₄ ⁺ | 0.44 | 0.05 | 0.88 |
| K ⁺ | 0.41 | 0.67 | 0.55 |
| Mg ²⁺ | 0.05 | 0.90 | 0.35 |
| Ca ²⁺ | 0.10 | 0.97 | 0.15 |
| OC | 0.77 | 0.41 | 0.46 |
| EC | 0.10 | 0.94 | 0.01 |
| Eigenvalues | 4.59 | 4.53 | 2.87 |
| % of Variance | 35.30 | 34.84 | 22.04 |
| Cumulative % | 35.30 | 70.14 | 92.18 |

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Table 4. Correlation coefficients (r) from Spearman correlation analysis for OC_{non-BB} and water-soluble ions during winter and spring of 2017 at Ulaanbaatar, Mongolia.

| | | Cl ⁻ | SO ₄ ²⁻ | NO ₃ ⁻ | Na ⁺ | NH ₄ ⁺ | K ⁺ | Mg ²⁺ | Ca ²⁺ | EC |
|----------------------|--------|-----------------|-------------------------------|------------------------------|-----------------|------------------------------|----------------|------------------|------------------|--------|
| OC _{non-BB} | Winter | -0.26 | 0.71** | 0.44 | -0.58* | 0.72** | 0.64** | -0.16 | -0.16 | 0.15 |
| | Spring | 0.29 | 0.37 | 0.59* | 0.74** | 0.23 | 0.65** | 0.78** | 0.77** | 0.74** |

*Correlation is significant at the .05 level (2-tailed); **Correlation is significant at the .01 level (2-tailed).

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843 **Figure captions**

844 Fig. 1 Sampling site in Ulaanbataar, Mongolia (<https://www.google.com/earth/versions/#earth-pro>, © Google Earth).

845 Fig. 2 Daily variations in atmospheric concentrations ($\mu\text{g m}^{-3}$) of chemical species in Ulaanbaatar during winter (a) and spring (b) of
846 2017.

847 Fig. 3 Daily atmospheric concentrations of OC ($\mu\text{g C m}^{-3}$) as a function of wind speed (m s^{-1}) and temperature ($^{\circ}\text{C}$) during winter (a)
848 and spring (b) of 2017.

849 Fig. 4 Relationship between $\text{PM}_{2.5}$ concentrations of Ca^{2+} and EC ($\mu\text{g m}^{-3}$) during spring of 2017.

850 Fig. 5 Conditional Probability Function (CPF) of levoglucosan (levo), OC, K^{+} , EC, Ca^{2+} during winter (a) and spring (b) of 2017.

851 Fig. 6 (a) Five-day backward air-mass trajectories (<https://ready.arl.noaa.gov/HYSPLIT.php>) and (b) FIRMS fire counts
852 (<https://firms.modaps.eosdis.nasa.gov/alerts/>) around Ulaanbaatar during spring of 2017.

853 Fig. 7 Correlations of $\text{PM}_{2.5}$ concentrations ($\mu\text{g m}^{-3}$) of mannosan and galactosan with levoglucosan during winter (a) and spring (b) of
854 2017.

855 Fig. 8 Correlation between $\text{PM}_{2.5}$ concentrations of (a) OC ($\mu\text{g C m}^{-3}$) and levoglucosan ($\mu\text{g m}^{-3}$) and (b) K^{+} and levoglucosan ($\mu\text{g m}^{-3}$)
856 during winter and spring of 2017.

857 Fig. 9 Correlation between $\text{PM}_{2.5}$ concentrations of OC ($\mu\text{g C m}^{-3}$) and K^{+} ($\mu\text{g m}^{-3}$) during winter (a) and spring (b) of 2017.

858 Fig. 10 Scatter plot of levoglucosan/ K^+ versus levoglucosan/mannosan from different types of BB emissions, including those measured
859 in Ulaanbaatar (blue circles and red squares).

860 Fig. 11 Comparison of previously reported OC/levoglucosan ratios for softwood burning.

861 Fig. 12 Graphical determination of optimized OC/levoglucosan ratios used to estimate $PM_{2.5}$ concentrations of OC_{BB} in Ulaanbaatar in
862 winter (a) and spring (b) of 2017.

863 Fig. 13 Relative contributions ($\mu\text{g C m}^{-3}$) of OC_{BB} and $OC_{\text{non-BB}}$ to $PM_{2.5}$ in Ulaanbaatar during winter and spring of 2017.

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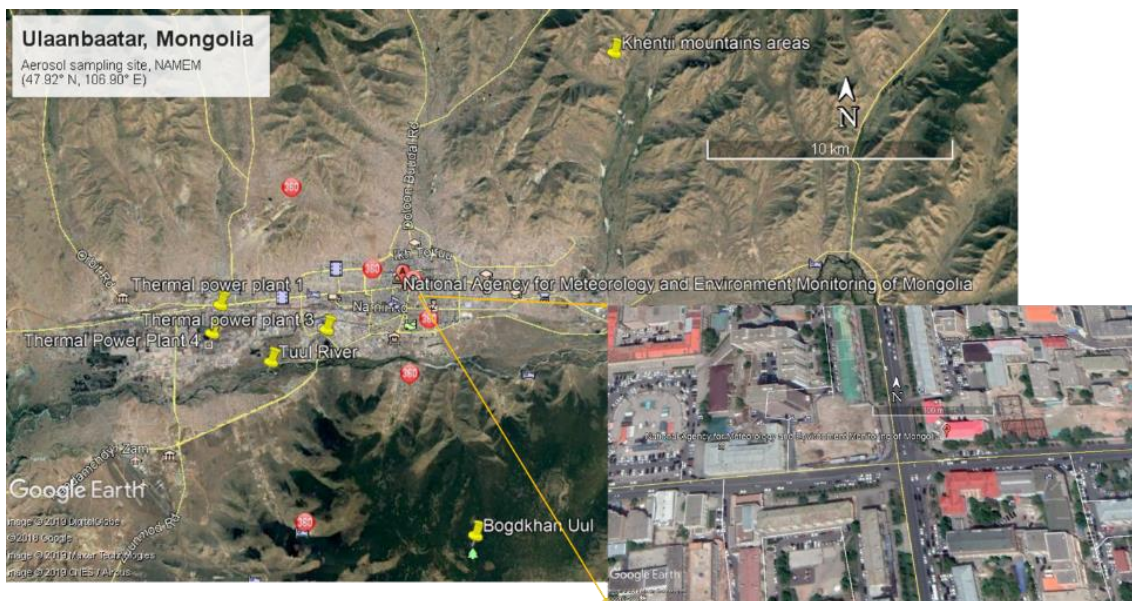
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Fig. 1

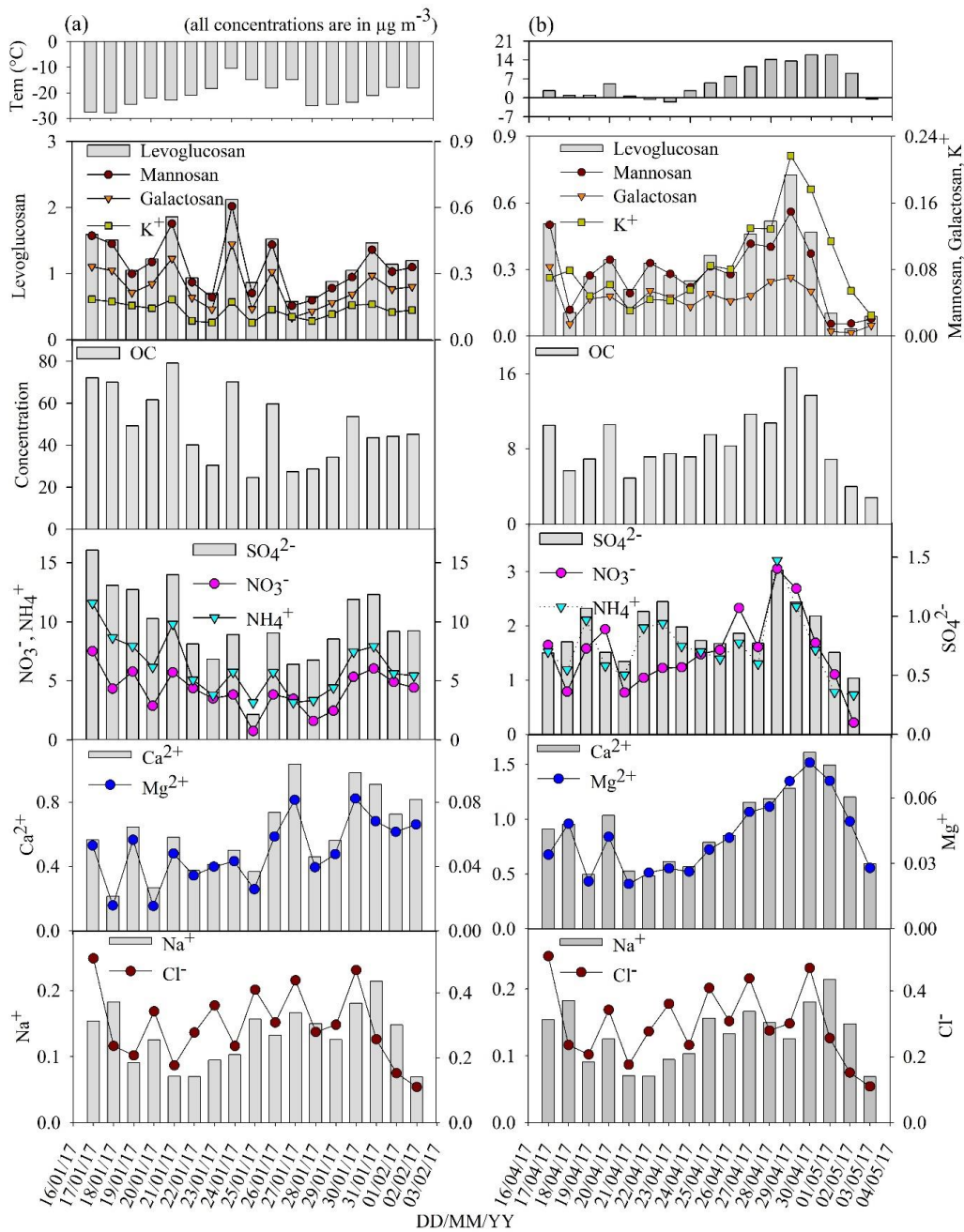


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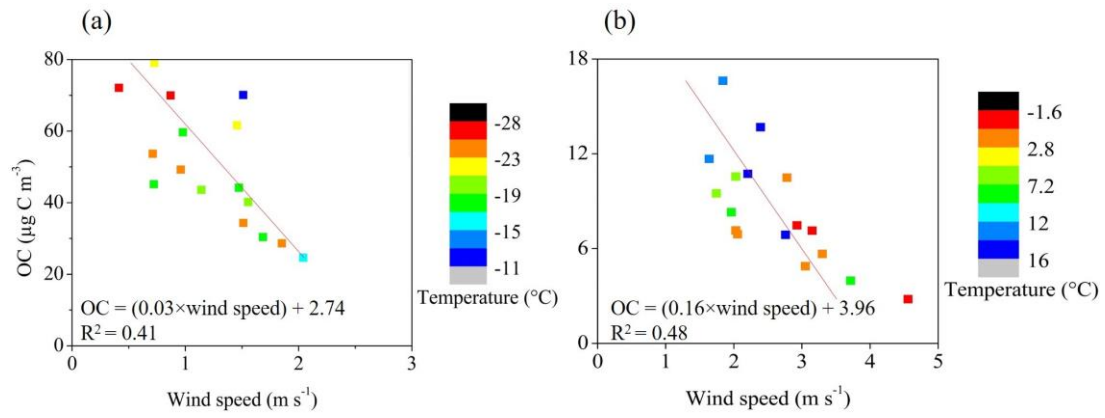
Fig. 2



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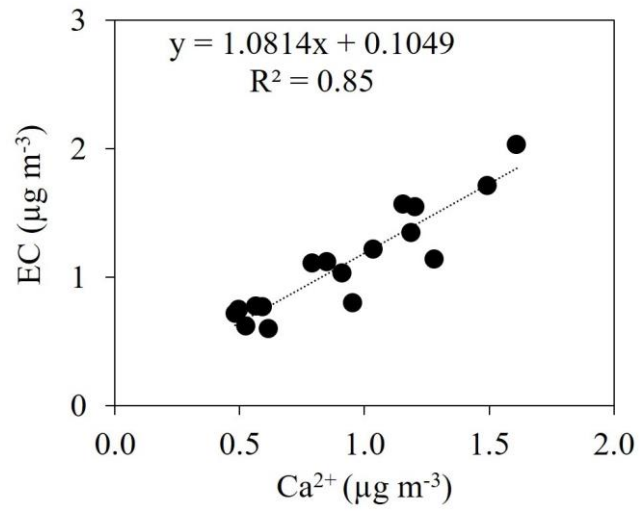
Fig. 3



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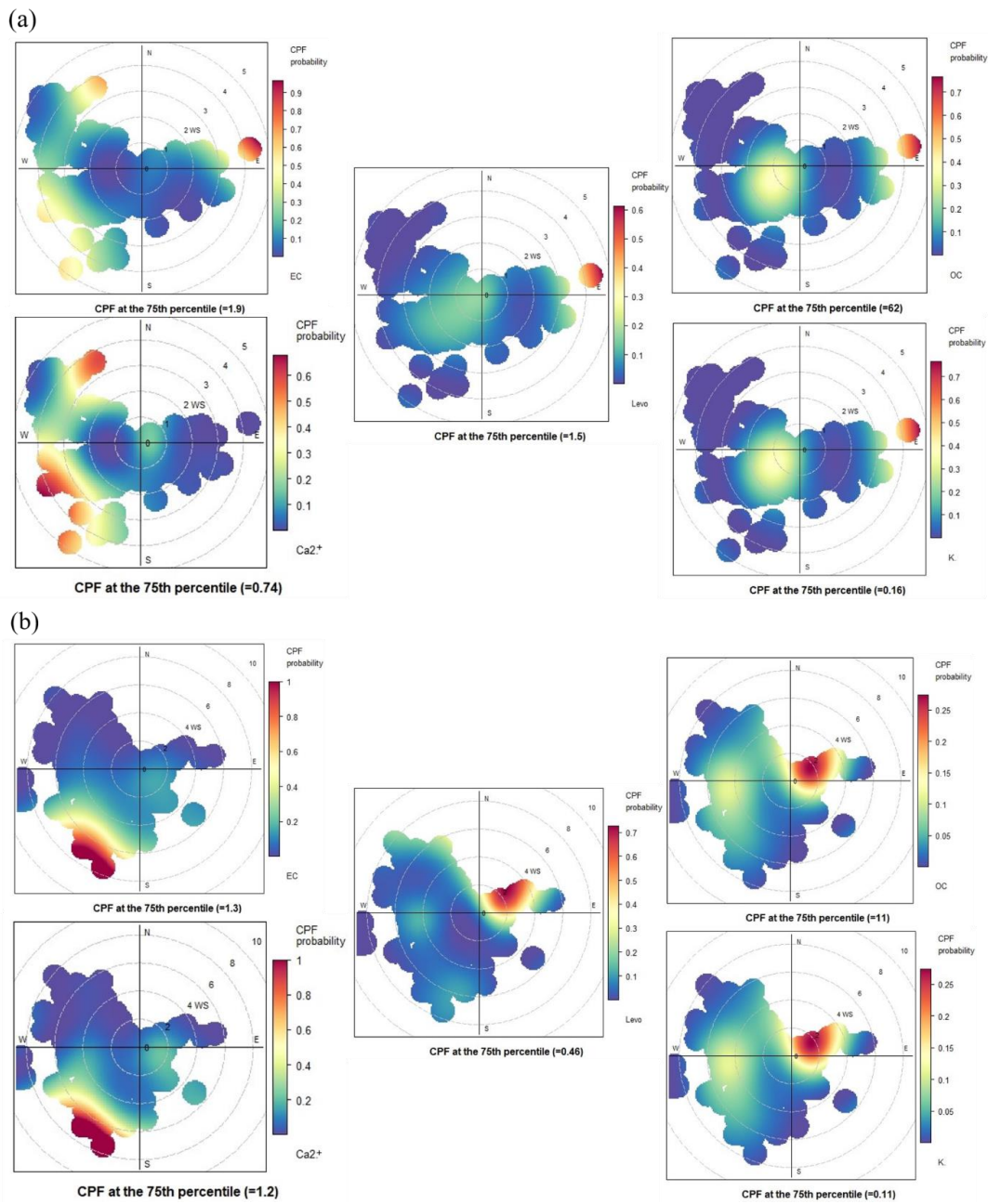
Fig. 4



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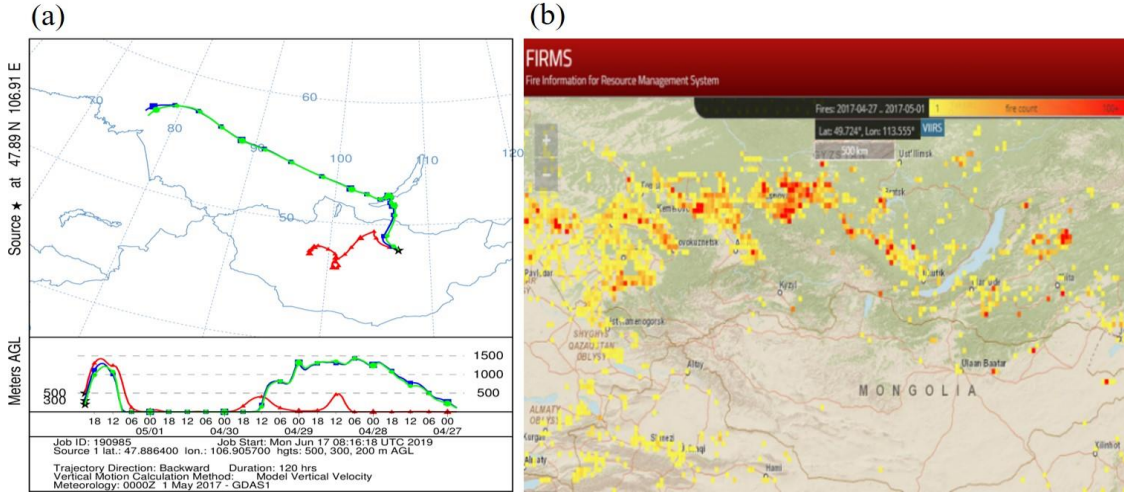
Fig. 5



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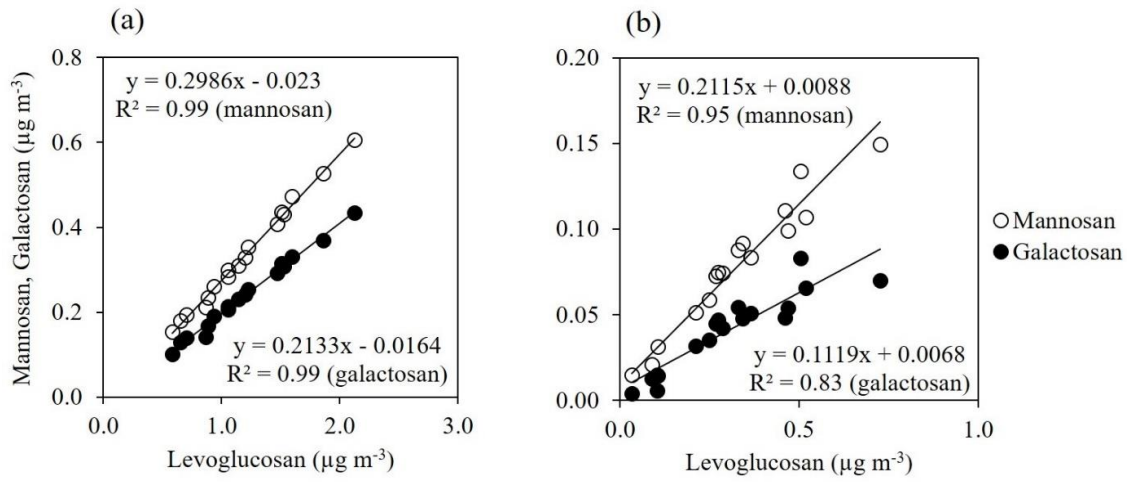
Fig. 6



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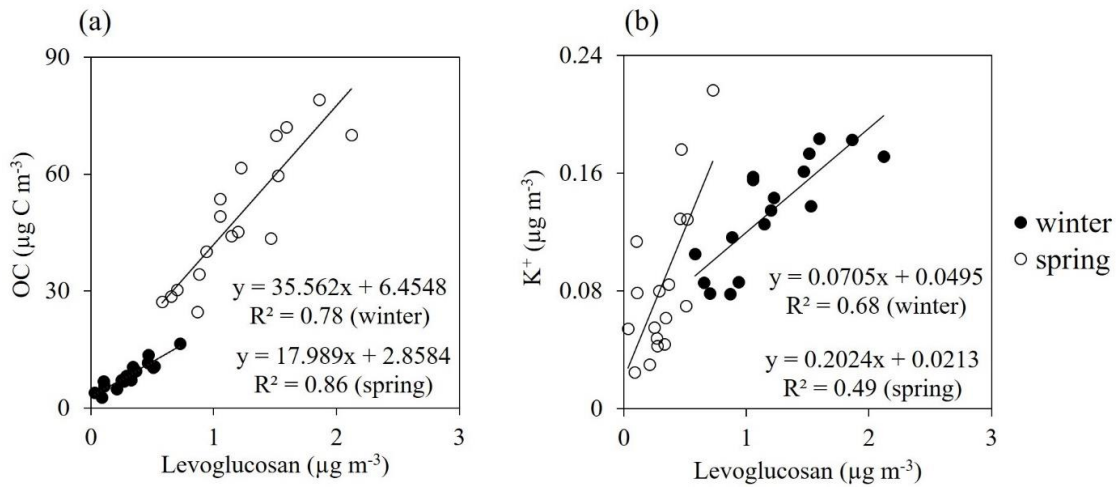
Fig. 7



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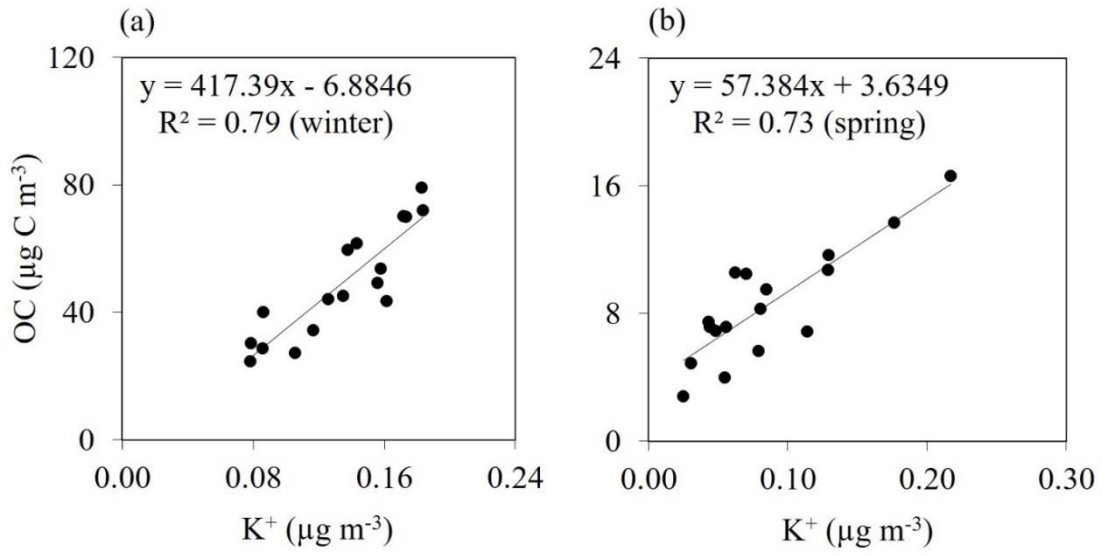
Fig. 8



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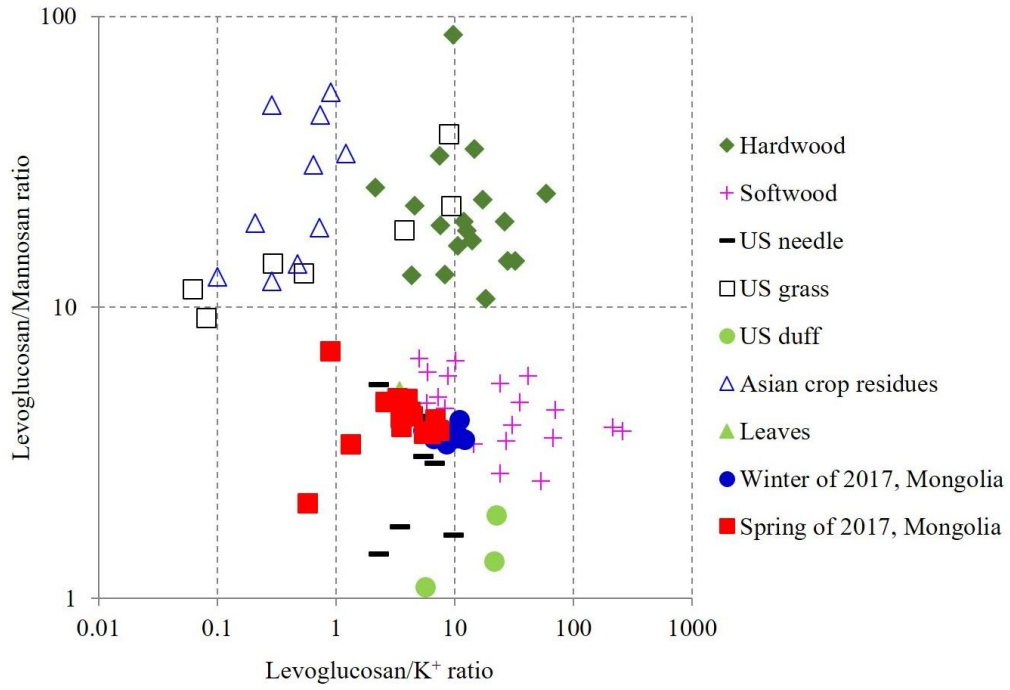
Fig. 9



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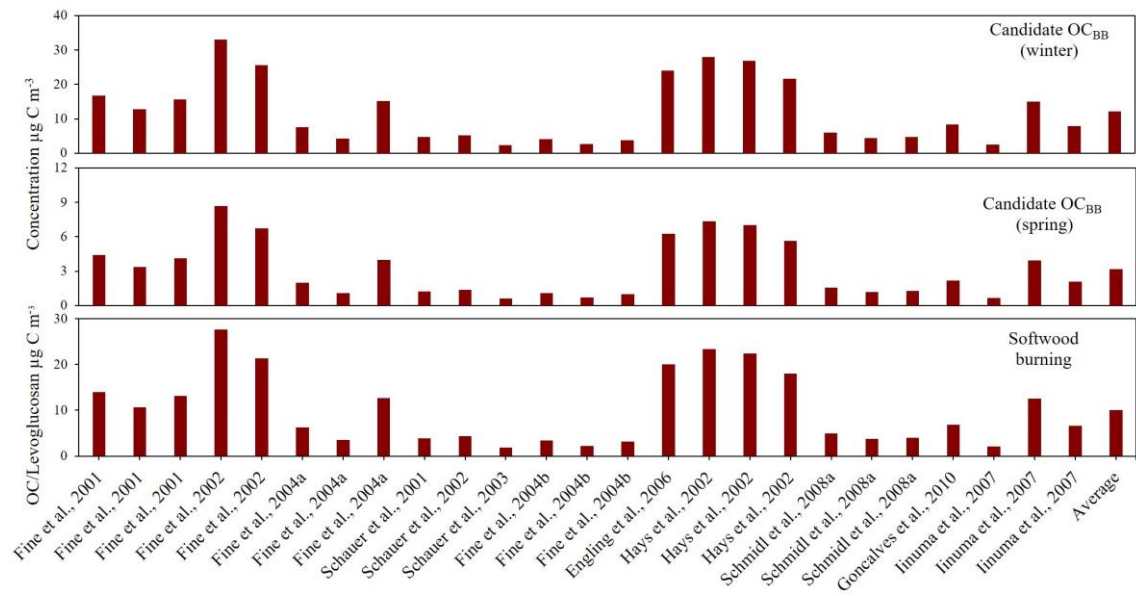
Fig. 10



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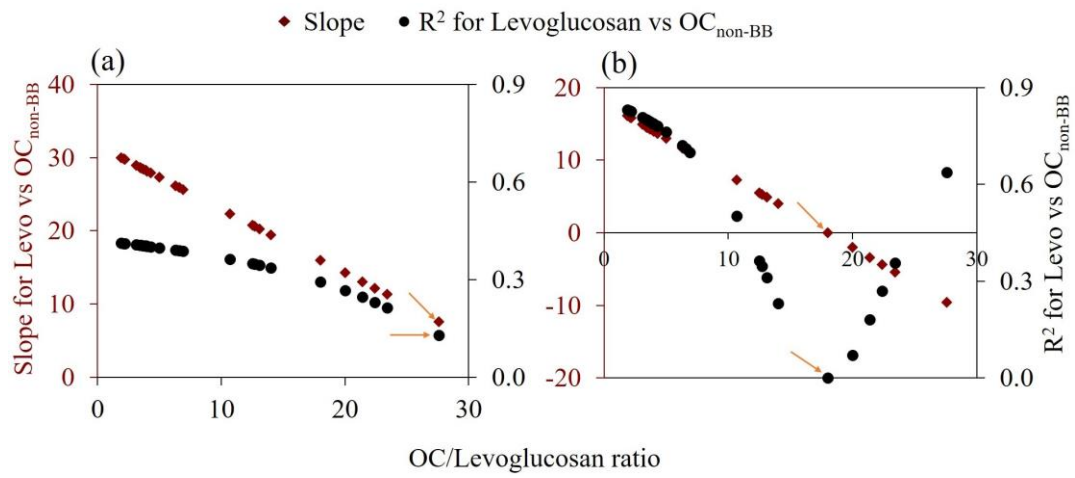
Fig. 11



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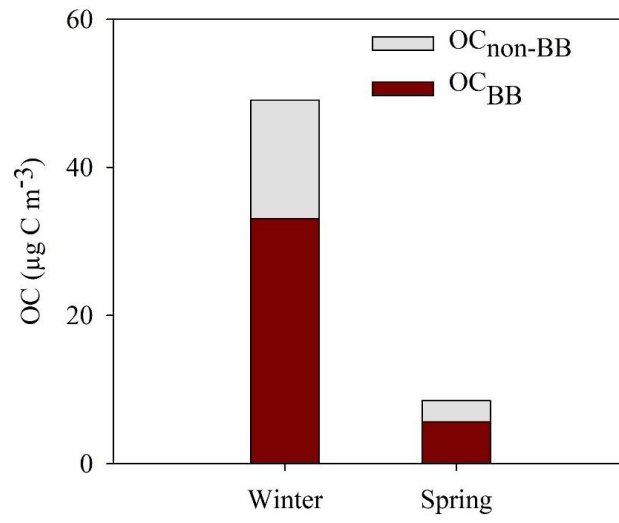
Fig. 12



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Fig. 13



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Report #1

Comments:

Thank very much for the reviewer for the appreciating comments and important inputs to improve the quality of the manuscript. We follow all the reviewer's comments very carefully and answer accordingly. We have incorporated all the modifications in the revised manuscript (RMS). Please refer to the revised manuscript where we highlighted the changes by [turquoise color](#). We provide here below a reply to the specific comments and modifications made in the revised manuscript based on the line number provided in RMS. Please follow the line numbers to reviewing the changes in RMS.

1. The revised Fig 2: It is noteworthy to me that the coldest periods in the winter do not coincide with the highest concentrations of levoglucosan, mannosan etc (e.g 16-17th Jan and 29th Jan). If my reading of the paper is correct and biomass burning for heating is important in Ulaanbaatar, then one would expect the highest concentration of these tracer to be high when it is cold. Perhaps the authors could comment.

Reply: We thank the reviewer for highlighting this point. During winter the ambient temperature was consistently low (less than -10 °C) in Ulaanbaatar therefore residential biomass burning occurred continuously for space heating. Thus, there would be no dependence of emission rates of levoglucosan and mannosan with ambient temperature during the sampling period. The concentration of levoglucosan and mannosan may be influenced by local wind speed as shown in Fig. 3 rather than ambient temperature.

2. Revised Fig 5: I am not sure that the authors have enough samples to perform CPF analysis. I think they need to justify applying a 24 hr average concentration measurement to each 1 hour wind measurement in order to do this analysis. I would have instead done a simpler analysis, like a wind rose plot with concentration data.

Reply: Thank you very much for the suggestion. Already we have calculated CPF using 24h concentrations of measured species and, hourly wind speed and wind direction data in revised manuscript. Also we already mentioned this phrase in section 2.3 of first revised manuscript.

[Following phrase added in first revised manuscript.](#)

[“To use CPF with the Ulaanbaatar data, the 24 h averaged source contribution data have been applied to all 1 h wind direction averages recorded at the site for each date.”](#)

[Please see line number 166-168](#)

3. I am still not sure I quite follow the method for determining the optimized OC/levoglucosan (Section 3.5). Is it that you selected a range of different OC/levoglucosan ratios from the literature (as shown in Fig 11?) to calculate the OC from biomass burning using the measured levoglucosan. Then the slope and r² of the relationship between the resultant OC_{non-bb} and levoglucosan was calculated to find the one that gave the minimum? This may need further clarification in the text, including indicating which ratio were used in Fig 12.

If that is correct why did you settle on the ratio for winter (27.8) when it did not give a slope of zero unlike observed in the spring (Fig 12)? Why not try a higher ratio to get to zero?

1034 Reply: Thank for the reviewer’s suggestions. We have selected a different
1035 OC/levoglucosan ratios of softwood burning from the literature (Fig. 11) to calculate the
1036 OC from biomass burning (OC_{BB}) from multiplying with measured levoglucosan
1037 concentration. The highest value of OC/levoglucosan ratio of softwood burning was
1038 reported to be 27.6 from the previous chamber experiment as shown in Fig. 11. Thus,
1039 we did not include OC/levoglucosan ratio of higher than 27.6 for optimum
1040 OC/levoglucosan ratios estimation.

1041 OC_{non-BB} is calculated by subtracting estimated OC_{BB} with measured total OC of each
1042 corresponding day. Then the slope and R^2 of the relationship between resultant OC_{non-BB}
1043 and measured levoglucosan concentration is calculated to find the one OC/levoglucosan
1044 ratio that gave the minimum. That OC/levoglucosan ratio is treated as optimized ratio.
1045 Finally, this optimized OC/levoglucosan ratio is used for estimating OC_{BB} for this site.
1046 OC/levoglucosan ratios of 27.6 and 18.0 have optimized for winter and spring,
1047 respectively, for this site to estimate OC_{BB} . We have modified phrase in section 3.5 for
1048 clarity about the approach for obtaining optimized OC/levoglucosan ratio.

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1050 Following sentences have been added and modified:

1051 “We have used a range of different OC/levoglucosan ratios obtained from previous
1052 literatures (Fig. 11) for regression analysis with measured levoglucosan concentrations
1053 to estimate optimum OC/levoglucosan ratio (Fig. 12a, b).”

1054 “The OC_{BB} concentrations at the Ulaanbaatar study site were calculated from the
1055 optimized OC/levoglucosan ratios (winter: 27.6 and spring: 18.0) and measured
1056 levoglucosan concentrations.”

1057 *Please see line number 372-375, 379, 381, 395-397*

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Report #2

Comments:

I was Referee #2 for the ACPD manuscript. The authors have reasonably addressed the first 3 of my 4 major comments. As to my major comment 4, reference to the study of Davy et al. (full reference: Davy, P.K., Gunchin, G., Markwitz, A., Trompeter, W.J., Barry, B.J., Shagjjamba, D., Lodoysamba, S., 2011. Air particulate matter pollution in Ulaanbaatar, Mongolia: determination of composition, source contributions and source locations, Atmos. Poll. Res., 2, 126-137) should definitely be made. The authors' explanation for the large discrepancy between the contribution from biomass burning in PM_{2.5} in that study and in the current study is not very convincing. But if the authors do not want to elaborate on this discrepancy in their manuscript, they should at least refer to Davy et al. (2011) in the Introduction as an earlier study in Ulaanbaatar, together with the studies of Jung et al. (2010) and Batmunkh et al. (2013), thus in lines 91-92. Since OC was not measured in Davy et al. (2011), "organic aerosol" in line 91 could then be replaced by "aerosol". Also, further within the manuscript's section 3.1, the authors could compare the PM_{2.5} K data of Davy et al. (2011) and the PM_{2.5} OC data of Jung et al. (2010) and of Batmunkh et al. (2013) with their own data.

Reply: Thank for the reviewers' comments and suggestion for improving the quality of manuscript. Please refer to the revised manuscript where we highlighted the changes by green color. We provide here below a reply to the specific comments and modifications made in the revised manuscript based on the line number provided in RMS. Please follow the line numbers to reviewing the changes in RMS.

We have replaced "organic aerosol" in revised manuscript by "aerosol". We have modified and added following sentences in revised manuscript as per reviewer suggestions:

"Few studies have investigated the chemical characteristics of aerosol in Ulaanbaatar (Jung et al., 2010; Davy et al., 2011; Batmunkh et al., 2013), with none examining the contribution of OCBB and type of biomass."

Please see line number 93-96.

"Previously, OC has been observed as major component in PM_{2.5} in Ulaanbaatar during winter period (Jung et al., 2010; Batmunkh et al., 2013)."

Please see line number 194-195.

"Average K⁺ concentration (0.08±0.05 μg m⁻³) in this study is significantly lower than the K (0.32 μg m⁻³) observed in Ulaanbaatar during 2004-2008 (Davy et al., 2011)."

Please see line number 228-230.

Besides, I have the following (minor) comments:

Reply: We thank to reviewer for his appreciable comments. We have attempted all the suggestions and modified the revised manuscript accordingly.

1. Line 33: replace "study site" by "study sites".

Reply: Replacement done, please see line number 33.

2. Line 48: replace "power plant" by "power plants".

Reply: Replacement done, please see line number 48.

- 1123
1124 3. *Line 60: replace "than those" by "than that".*
1125 Reply: Replacement done, please see line number 61.
1126
1127 4. *Line 82: replace "A half of" by "Half of the".*
1128 Reply: Replacement done, please see line number 85.
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1130 5. *Line 83: replace "dwelling)" by "dwellings)".*
1131 Reply: Replacement done, please see line number 86.
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1133 6. *Line 85: replace "birch these" by "birch; these".*
1134 Reply: Replacement done, please see line number 88.
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1136 7. *Line 89: replace "is very few estimate" by "are very few estimates".*
1137 Reply: Replacement done, please see line number 92-93.
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1139 8. *Line 93: replace "and identify" by "and identified".*
1140 Reply: Replacement done, please see line number 97.
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1142 9. *Line 120: replace "analyte has" by "analytes has".*
1143 Reply: Replacement done, please see line number 123.
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1145 10. *Line 136: I disagree with the authors' reply that "'were analyzed" is more*
1146 *appropriate thus we retain it in revised MS'; samples are analyzed, but species, as is*
1147 *the case here, are measured; therefore, "were analyzed" should be replaced by*
1148 *"were measured".*
1149 Reply: Replacement done, please see line number 141.
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1151 11. *Line 164: replace "percentile of" by "percentiles of".*
1152 Reply: Replacement done, please see line number 169.
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1154 12. *Line 173: replace "PC)" by "PCs)".*
1155 Reply: Replacement done, please see line number 178.
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1157 13. *Line 204: replace "5 that can" by "5; this can".*
1158 Reply: Replacement done, please see line number 211.
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1160 14. *Line 208: replace "plant (Pei" by "plants (Pei".*
1161 Reply: Replacement done, please see line number 215.
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1163 15. *Line 255: replace "results showed" by "results show".*
1164 Reply: Replacement done, please see line number 266.
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1166 16. *Line 270: replace "as biomass burning tracers in" by "as a biomass burning tracer*
1167 *in".*
1168 Reply: Replacement done, please see line number 282.
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- 1170 17. *Line 276: replace "tended to" by "tends to".*
1171 Reply: Replacement done, please see line number 288.
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- 1173 18. *Line 284: replace "belongs to" by "belong to".*
1174 Reply: Replacement done, please see line number 296.
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- 1176 19. *Line 359: replace "experiment (Cheng" by "experiments (Cheng".*
1177 Reply: Replacement done, please see line number 377.
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- 1179 20. *Line 370: replace "thus it has lower" by "thus have lower".*
1180 Reply: Replacement done, please see line number 388.
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- 1182 21. *Line 433: replace "mainly to" by "mainly due to".*
1183 Reply: Replacement done, please see line number 453.
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- 1185 22. *Line 433: replace "Ashbaugh, L.L., Malm, W.C., Sadeh, W.Z." by "Ashbaugh, L. L.,*
1186 *Malm, W. C., and Sadeh, W. Z."*
1187 Reply: Replacement done, please see line number 502-504.
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- 1189 23. *Lines 584-588: "Guttikanda, S., 2008" should come before "Guttikanda and*
1190 *Jawahar, 2014"; furthermore, for "Guttikanda, S., 2008", the authors should insert*
1191 *"SIM Working Paper No. 2008-005," after "Ulaanbaatar, Mongolia, "*
1192 Reply: Modification done, please see line number 615-619.
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- 1194 24. *Line 687: replace "Q. and Yan" by "Q., and Yan".*
1195 Reply: Modification done, please see line number 718.
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