



1 **Characterization of Organic Aerosol across the Global**
2 **Remote Troposphere: A comparison of ATom**
3 **measurements and global chemistry models**

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32 **Abstract.**

33 The spatial distribution and properties of submicron organic aerosols (OA) are among the
34 key sources of uncertainty in our understanding of aerosol effects on climate.
35 Uncertainties are particularly large over remote regions of the free troposphere and
36 Southern Ocean, where very little data has been available, and where OA predictions from
37 AeroCom Phase II global models span a factor of 400-1000, greatly exceeding the model
38 spread over source regions. The (nearly) pole-to-pole vertical distribution of non-refractory
39 aerosols was measured with an aerosol mass spectrometer onboard the NASA DC8
40 aircraft as part of the Atmospheric Tomography (ATom) mission during the northern
41 hemisphere summer (August 2016) and winter (February 2017). This study presents the
42 first extensive characterization of OA mass concentrations and their level of oxidation in
43 the remote atmosphere. OA and sulfate are the major contributors by mass to submicron
44 aerosols in the remote troposphere, together with sea salt in the marine boundary layer.
45 Sulfate was dominant in the lower stratosphere. OA concentrations have a strong
46 seasonal and zonal variability, with the highest levels measured in the summer and over
47 the regions influenced by the biomass burning from Africa (up to $10 \mu\text{g sm}^{-3}$). Lower
48 concentrations ($\sim 0.1\text{-}0.3 \mu\text{g sm}^{-3}$) are observed in the northern mid- and high- latitudes
49 and very low concentrations ($< 0.1 \mu\text{g sm}^{-3}$) in the southern mid- and high- latitudes. The
50 ATom dataset is used to evaluate predictions of eight current global chemistry models that
51 implement a variety of commonly used representations of OA sources and chemistry, as
52 well as of the AeroCom-II ensemble. The current model ensemble captures the average
53 vertical and spatial distribution of measured OA concentrations, and the spread of the
54 individual models remains within a factor of 5. These results are significantly improved
55 over the AeroCom-II model ensemble, which shows large overestimations over these
56 regions. However, some of the improved agreement with observations occurs for the
57 wrong reasons, as models have the tendency to greatly overestimate the primary OA
58 fraction, and underestimate the secondary fraction. Measured OA in the remote free
59 troposphere are highly oxygenated with organic aerosol to organic carbon (OA/OC) ratios
60 of $\sim 2.2\text{-}2.8$ and are 30-60% more oxygenated than in current models, which can lead to
61 significant errors in OA concentrations. The model/measurement comparisons presented
62 here support the concept of a more dynamic OA system as proposed by Hodzic et al.
63 (2016), with enhanced removal of primary OA, and a stronger production of secondary OA
64 in global models needed to provide a better agreement with observations.



65 1 Introduction

66 Organic aerosols (OA) are a complex mixture of directly emitted primary OA (POA) and
67 chemically produced secondary OA (SOA) from anthropogenic and biogenic emission
68 sources. They are associated with adverse health effects (Mauderly and Chow, 2008,
69 Shiraiwa et al., 2017) and a major contribution to radiative forcing in the climate system
70 (Boucher et al., 2013). The currently limited understanding of processes involved in the
71 formation, ageing, and removal of organic compounds results in large uncertainties in (i)
72 the predicted global OA burden, (ii) relative contributions of emissions vs. chemistry to OA
73 formation, (iii) spatial distribution, and (iv) impacts on radiation and clouds (Kanakidou et
74 al., 2005, Hallquist et al., 2009, Heald et al., 2011, Spracklen et al., 2011, Tsigaridis et al.,
75 2014, Hodzic et al., 2016, Shrivastava et al., 2017, Tsigaridis and Kanakidou, 2018). The
76 uncertainties are particularly large in the estimated global burden of SOA that range from
77 12 to 450 Tg y⁻¹ (see Fig. 9 of Hodzic et al., 2016), and in their direct and indirect radiative
78 forcing that range from -0.08 to -0.33 W m⁻², and -0.60 to -0.77 W m⁻², respectively
79 (Spracklen et al., 2011, Myhre et al., 2013, Scott et al., 2014, Hodzic et al., 2016, Tsigaridis
80 and Kanakidou, 2018). Reducing these uncertainties is becoming more important as OA
81 is on a path to becoming the dominant fraction of the submicron anthropogenic aerosol
82 mass globally due to the ongoing efforts to reduce SO₂ emissions and associated sulfate
83 aerosols.

84 Model performance has been especially poor in the remote regions of the atmosphere
85 where OA measurements available for model evaluation have been sparse (especially
86 aloft). Using data from 17 aircraft campaigns mostly located in the northern hemisphere
87 Heald et al. (2011) showed that the skill of the global GEOS-Chem model in predicting the
88 vertical distribution of OA was significantly decreased in remote regions compared to
89 polluted near-source regions. The study pointed out the limitations of commonly used SOA
90 formation mechanisms that are based on chamber data; these have the tendency to
91 underpredict OA in source regions and overpredict OA in the remote troposphere. For the
92 same field campaigns, Hodzic et al. (2016) showed that OA is likely a more dynamic
93 system than represented in chemistry-climate models, with both stronger production and
94 stronger removals. These authors suggested that additional removal mechanisms via e.g.
95 photolytic or heterogeneous reactions of OA particles are needed to explain low OA
96 concentrations observed in the upper troposphere where direct cloud scavenging is less
97 efficient. The recent global multi-model comparison study (Tsigaridis et al., 2014) within



98 the AeroCom Phase II project illustrates well the amplitude of model uncertainties
99 simulating OA mass concentrations and the contrast in model performance between near-
100 source and remote regions. The results indicate that model dispersion (the spread
101 between the models with the lowest and highest predicted OA concentrations) increases
102 with altitude from roughly 1 order of magnitude near the surface to 2-3 orders of magnitude
103 in the upper troposphere. Our own analyses of the AeroCom-II results shown in Figure 1a
104 indicate that model dispersion increases not only with altitude but also with distance from
105 the northern mid-latitude source (and data-rich) regions. The model spread is a factor of
106 10-20 in the free troposphere between the equator and northern mid-latitudes, and
107 increases to a factor of 200-800 over the Southern Ocean and near the tropopause. It is
108 not surprising that model spread is lower closer to source regions where it is mostly driven
109 by uncertainties in emissions and SOA production yields. Spread is expected to be larger
110 in remote regions where models are also impacted by uncertainties in transport, chemical
111 ageing and removal. The lowest model dispersion also coincides with the regions of the
112 northern hemisphere (NH) or the African biomass burning outflow where models have
113 been evaluated the most (Figure 1b), emphasizing the need for further model/observation
114 comparison studies in remote regions (of the southern hemisphere (SH) in particular).

115 Here, we present a unique data set of airborne aerosol mass spectrometer measurements
116 of OA mass concentrations collected onboard the NASA DC-8 as part of the Atmospheric
117 Tomography (ATom) mission. The aircraft sampled the vertical structure of the
118 atmosphere from near-surface (0.2 km) to the lower-stratosphere (LS) regions (12 km
119 altitude) over both the Pacific and Atlantic basins (to limit the influence of source regions)
120 with a quasi-global spatial coverage from 82°N to 67°S. This dataset is used to perform
121 the first systematic global-scale multi-model evaluation of the chemistry-climate models
122 focusing on OA in the remote troposphere over the remote oceans. We focus on the NH
123 summer (August 2016, ATom-1) and NH winter (February 2017, ATom-2) deployments.
124 Overall these ATom missions sampled the marine boundary layer (MBL) for 10% of the
125 flight tracks, 12% of the time the remote lower stratosphere, and the rest the free
126 troposphere. The model-observation comparisons are aimed at identifying discrepancies
127 in terms of OA mass concentrations and vertical distribution, their fractional contribution
128 to submicron aerosols, and their oxidation level in global models.

129 The modeling framework is described in Section 2. Section 3 describes the ATom dataset
130 and the spatial and vertical distributions of OA over the Atlantic and Pacific regions.



131 Section 4 presents the comparisons of ATom-1 and -2 data to multi-model predictions
132 from both the AeroCom-II models, and the ensemble of eight current model simulations of
133 the ATom campaign. Section 5 presents the conclusions of the study and discusses its
134 implications.

135 **2 Modeling framework**

136 **2.1 ATom models**

137 ATom measurements were compared with results of eight global models that simulated
138 the time period of the ATom-1 and 2 campaigns (August 2016 and February 2017), using
139 reanalysis meteorology (and a spin-up time of at least six to twelve months). These are
140 referred hereafter as ATom models and include the NASA global Earth system model
141 GEOS5-GOCART, the aerosol-climate model ECHAM-HAM, three versions of the NCAR
142 Community Earth System Model (CESM), and three versions of the global chemistry
143 GEOS-Chem model. Simulations were performed at various horizontal resolutions ranging
144 from relatively high ~50km (GEOS5-GOCART) and ~100km (CESM2 models) resolutions
145 to somewhat coarser grids of ~200km (CESM1-CARMA, GEOS-Chem) and ~400km for
146 GC10-TOMAS. The advantage of using the same host model (in the cases of variants of
147 CESM2 and GEOS-Chem) is that the dynamics and emissions remain comparable.
148 Models differ greatly in their treatment of emissions, gas-phase chemistry, aerosol
149 chemistry and physical processes, and aerosol coupling with radiation and clouds, among
150 others. Table 1 describes the configuration of various models (e.g. meteorology,
151 emissions), and their treatment of OA. In this section we only summarize the main features
152 and parameters directly impacting the OA simulations. Some models do not include SOA
153 chemistry and instead assume that SOA is directly emitted proportional to the emissions
154 of its precursors (ECHAM6-HAM, CESM2-SMP, GEOS5-GOCART, GC10-TOMAS),
155 while others have more complex treatments of organic compounds, their chemistry, and
156 partitioning into particles (GC12-REF, GC12-DYN, GC10-TOMAS, CESM1-CARMA,
157 CESM2-DYN). It should be noted that models that directly emit SOA assume that SOA is
158 a non-volatile species that remains irreversibly in the particle phase. There are some
159 commonalities between simulations for the treatment of biogenic emissions, which are
160 based in all models on the Model of Emissions of Gases and Aerosols from Nature
161 (MEGAN, Guenther et al., 2012) to generate meteorology-dependent emissions of volatile
162 organic compounds. None of the models includes the marine production of OA which is
163 estimated to be ~3 orders of magnitude smaller than the continental production of OA from



164 both isoprene and monoterpene precursors (Kim et al., 2017). This contribution could
165 however be larger for sea-spray biological material from phytoplankton with predicted
166 contributions of 0.01 to 0.1 $\mu\text{g m}^{-3}$ to surface submicron aerosol over remote oceanic
167 regions (Vergara-Temprado et al., 2017, Middlebrook et al., 1998). Below we only provide
168 a brief description of most important processes that influence OA for each model.

169 GEOS5-GOCART was run in a configuration similar to Bian et al. (2019) using the
170 anthropogenic emissions from HTAP v2 (Janssens-Maenhout et al., 2015) and biomass
171 burning emissions from the Quick Fire Emission Dataset (QFED v2.54). Aerosols are
172 simulated within the GOCART bulk aerosol module and include externally mixed particles
173 of black carbon (BC), organic carbon (OC), sulfate, ammonium, nitrate, dust and sea salt
174 (Colarco et al., 2010, Bian et al., 2017). The formation of SOA is based on a prescribed
175 10% formation yield from the monoterpene emissions. The primary emitted OC and SOA
176 are separated into hydrophobic (50%) and hydrophilic (50%) species, with a 2.5 days e-
177 folding time conversion from hydrophobic to hydrophilic organic particles. Both types of
178 organic particles are dry deposited, while the hydrophilic organic fraction undergoes rain
179 scavenging in convective updrafts, and in large-scale precipitation.

180 The ECHAM-HAM version ECHAM6.3-HAM2.3 (Tegen et al., 2019) was run using
181 updated anthropogenic emissions (Schacht et al., 2019) combining the ECLIPSE (Klimont
182 et al., 2017) emissions, with the Russian anthropogenic BC emissions from Huang et al.
183 (2015). For biomass burning the Global Fire Assimilation System (GFAS, Kaiser et al.,
184 2012) biomass burning emissions are used, however, without the scaling factor of 3.4
185 suggested by Kaiser et al. (2012). Aerosol composition and processes are simulated using
186 the Hamburg Aerosol Model (HAM2, Zhang et al., 2012), that considers an aerosol internal
187 mixture of sulfate, BC, OC, sea salt, and mineral dust. The aerosol population and their
188 microphysical interactions are simulated using seven log-normal modes, including the
189 nucleation mode, soluble and insoluble Aitken, accumulation and coarse modes. In the
190 model configuration used in this publication the formation of SOA is based on a prescribed
191 15% mass yield from monoterpene emissions only (Dentener et al., 2006).

192 The two simulations with the GEOS-Chem 12.0.1 global chemistry model (Bey et al.,
193 2001) use emissions based on CMIP6 global inventory with regional improvements for
194 anthropogenic sources, and on GFED v.4 for biomass burning emissions (Giglio et al.,
195 2013). Both simulations use the bulk aerosol representation and differ only in the treatment
196 of SOA formation and removal. The first configuration (called hereafter GC12-REF)



197 includes the default (<http://wiki.seas.harvard.edu/geos-chem/index.php>) representation of
198 SOA formation based on the volatility basis set (VBS) of Pye et al. (2010), and the isoprene
199 SOA formation based on Marais et al. (2016). The second configuration (referred to as
200 GC12-DYN) includes a more dynamic representation of the SOA lifecycle based on
201 Hodzic et al. (2016), with the exception of the treatment of isoprene SOA that is formed in
202 the aqueous aerosols as in Marais et al. (2016). As in Hodzic et al. (2016) the GC12-DYN
203 model version includes updated VBS SOA parameterization, updated dry and wet removal
204 of organic vapors, and photolytic removal of SOA. SOA formation is based on wall-
205 corrected chamber yields (Zhang et al., 2014) for the traditional precursors (isoprene,
206 monoterpenes, sesquiterpenes, benzene, toluene, xylene) and on yields derived from an
207 explicit chemical mechanism for higher molecular weight n-alkanes and n-alkenes species
208 (Hodzic et al., 2016). The removal of gas-phase oxidized volatile organics uses updated
209 Henry's law solubility coefficients from Hodzic et al. (2014), and photolytic removal of SOA
210 (Hodzic et al., 2015). In addition to OA, the model includes BC and dust, and simulates
211 the chemistry and gas-particle partitioning of inorganic compounds such as sulfate,
212 ammonium, nitrate and sea salt using the ISORROPIA II thermodynamic model
213 (Fountoukis and Nenes, 2007). In both GEOS-Chem configurations, BC and primary OC
214 are simulated with a hydrophobic and hydrophilic fraction for each. At the time of emission,
215 80% of BC and 50% of primary OC are considered as hydrophobic. Hydrophobic aerosols
216 are converted to hydrophilic aerosols with an e-folding lifetime of 1.15 days. An OA/OC
217 ratio of 2.1 is assumed to convert POC to POA, and SOA is simulated as OA mass (i.e.
218 no OA/OC ratio assumption is needed for SOA, except for comparison with OC
219 measurements). Soluble gases and aerosols are removed by both dry and wet deposition.
220 Wet deposition includes scavenging in convective updrafts, and in-cloud and below-cloud
221 scavenging from large-scale precipitation (Liu et al., 2001). Hydrophobic aerosols (BC and
222 POA) are scavenged in convective updrafts following Wang et al. (2014).

223 GC10-TOMAS is based on the GEOS-Chem version 10.01 coupled with Two Moment
224 Aerosol Sectional microphysics scheme (TOMAS) and ran in a similar configuration to that
225 described in Kodros et al. (2016). The model computes the evolution of sulfate, sea salt,
226 primary and secondary OA, BC, and dust aerosols described by 15 internally mixed size
227 bins (of which six were analyzed for these comparisons, cf. Table 1). Anthropogenic
228 emissions are based on the EDGAR v4 global inventory with regional improvements, while
229 the biomass burning emissions are from GFED v3. SOA are irreversibly made from the



230 emitted parent precursor, considering a 10% mass yield from monoterpene emissions,
231 and an emission flux of 0.2 Tg of SOA per Tg of CO for the anthropogenic CO emissions.

232 Simulations based on the CESM2.0 Earth system model use the Whole Atmosphere
233 Community Climate Model (WACCM6, Gettelman et al., 2019, Emmons et al., 2019).
234 Details on the specific of the model configurations are described in detail in Tilmes et al.
235 (2019) i.e. CESM2-SMP and CESM2-DYN correspond to the specified dynamics
236 WACCM6-SOAG and WACCM6-VBSext simulations described in that work, respectively.
237 Emissions are based on the CMIP6 global inventory for the year 2014 for anthropogenic
238 sources, and on the QFED version 2.4 for the wildfires inventory. Aerosols are represented
239 with the modal aerosol scheme (MAM4, Liu et al., 2012) that includes BC, primary and
240 secondary OA, sulfate, dust and sea salt. Four modes are considered including Aitken,
241 accumulation and coarse size modes, and an additional primary carbon mode. Only the
242 accumulation mode was used in this work. The CESM2-SMP and CESM2-DYN
243 simulations differ in their treatment of OA. CESM2-SMP forms OA directly using fixed
244 mass yields from primary emitted precursors (isoprene, monoterpenes, aromatics) without
245 explicitly simulating their oxidation and partitioning. These mass yields are increased by a
246 factor of 1.5 to match the anthropogenic aerosol indirect forcing (Liu et al., 2012). The
247 second configuration (referred to as CESM2-DYN) includes the formation and removal
248 parameterizations of organics of Hodzic et al. (2016), as implemented into CESM2 by
249 Tilmes et al. (2019) for all species based on low-NO_x VBS yields only. This is a similar
250 SOA scheme as used in GC12-DYN (with the exception of the detailed isoprene-SOA
251 scheme that is included in GC12-DYN). Organic gases and aerosols undergo dry and wet
252 deposition as described in Liu et al. (2012). It should be noted that CESM2-SMP does not
253 include deposition of intermediate organic vapors. Aerosol wet scavenging considers in-
254 cloud scavenging (the removal of cloud-borne particles that were activated at the cloud
255 base) and below-cloud scavenging for both convective and grid-scale clouds.

256 CESM1-CARMA simulations use the configuration described in Yu et al. (2019) which is
257 based on CESM1 and the sectional Community Aerosol and Radiation Model for
258 Atmospheres (CARMA v3.0). Anthropogenic emissions are those from the Greenhouse
259 gas-Air pollution Interactions and Synergies (GAINS) model, and biomass burning
260 emissions are from the Global Fire Emission Database (GFED v3, van der Werf et al.,
261 2010). In CARMA, 20 size bins are used for both pure sulfate particles (bins from 0.2 nm
262 to 1.3 μm in radius, only used up to 500 nm) and mixed aerosols composed of BC, primary



263 and secondary OC, dust, sea salt, and sea-spray sulfate (bins from 0.05–8.7 μm in radius,
264 again, only analyzed up to 500 nm). SOA formation is based on the VBS approach from
265 Pye et al. (2010). The removal of OA occurs only by dry and wet deposition. Compared to
266 the CESM2-SMP and CESM2-DYN simulations, the convective removal of aerosols uses
267 the modified scheme described in Yu et al. (2019) which accounts for aerosol secondary
268 activation from the entrained air above the cloud base, and the scavenging of activated
269 aerosols in convective updrafts. The default CESM can transport aerosols from the cloud
270 base to the top of the cloud in strong convective updrafts in one time step without
271 scavenging them, while the new scheme allows for a more efficient removal of all aerosols
272 inside convective clouds. A sensitivity simulation is performed for ATom-1 to quantify the
273 effect of this improved removal on OA concentrations (Section 4.5).

274 **2.2 AeroCom-II models**

275 The ATom measurements are also compared in a climatological way to the global model
276 OA predictions generated within the Phase II Aerosol Comparisons between Observations
277 and Models (AeroCom-II) project (Schulz et al., 2009). We consider the results of 28 global
278 models, which is a subset of those presented in Tsigaridis et al. (2014), based on the
279 availability of model results. It should be noted that the meteorological forcing used in
280 these models is mostly based on the year 2006, while the anthropogenic and biomass
281 burning emissions are mostly representative of the year 2000. For comparison purposes,
282 the monthly mean model outputs for the months of August (ATom-1) and February (ATom-
283 2) are interpolated along the flight path (latitude, longitude, and altitude), and averaged
284 the same way as the measurements (see section 3.2).

285 **3 Description of ATom measurements**

286 **3.1 Submicron aerosol data**

287 The measurements of non-refractory submicron aerosols were performed onboard the
288 NASA DC8 aircraft as part of the ATom field study (Wofsy et al., 2018) using the University
289 of Colorado Aerodyne High-Resolution Time-of-Flight Aerosol Mass Spectrometer (AMS
290 in the following, Canagaratna et al., 2007, DeCarlo et al., 2006).

291 We use measurements from both the NH summer (August 2016, ATom-1) and winter
292 (February 2017, ATom-2) deployments. Figure 2a shows the flight path and the vertical
293 extent of the ATom-1 dataset colored by OA mass concentrations (see Figure S1 for
294 ATom-2). The aircraft performed systematic vertical sampling with ~ 140 vertical profiles



295 per campaign throughout the troposphere from the near surface ~0.2 km to the upper
296 troposphere/lower stratosphere region at ~13 km altitude. Details on the operation of the
297 CU AMS on board the DC-8 are reported in Schroder et al. (2018), Nault et al. (2018), and
298 Guo et al. (2019). For ATom, the AMS reported the standard non-refractory aerosol
299 species OA, sulfate, nitrate, ammonium, and chloride, with the response for all the
300 nominally inorganic species characterized by in-field calibrations. In addition, it also
301 reported methanesulfonic acid (MSA, Hodshire et al., 2019a describes the AMS MSA
302 methods and calibrations for ATom) and sea salt for for $D_{\text{geo}} < 450$ nm (based on the
303 method of Ovadnevaite et al., 2012). Both of these species were important to achieve
304 closure with the volume calculated from the on-board sizing instruments in the marine
305 boundary layer (Guo et al., 2019). Another important refractory submicron species not
306 captured by the AMS measurements is BC. This was measured on ATom with the NOAA
307 SP2 instrument (Katich et al., 2018). It should be noted that aircraft measurements of
308 aerosol mass concentrations are given in $\mu\text{g sm}^{-3}$ (i.e., under standard conditions of 1 atm
309 and 273.15 K).

310 For ATom the AMS measured particles with geometric diameters (based on the campaign-
311 wide average density of 1640 kg m^{-3} , Guo et al., 2019) of between $D_{\text{geo}} \sim 60$ and 295 nm
312 with ~100% efficiency (and between 35 and 460 nm with 50% efficiency). Here we denote
313 the AMS aerosol data as “submicron” mass (based on the more usual definition using
314 aerodynamic diameter, which is larger than the geometric diameter; DeCarlo et al., 2004),
315 with the assumption that non-refractory aerosol are small contributors to mass above the
316 AMS size range. As shown in Brock et al., 2019 and Guo et al., 2019, the accumulation
317 mode for the ATom sampling environment only extended up to 500 nm, and hence, as
318 expected for a background tropospheric environment, this approximation is appropriate.
319 Very good agreement was observed with the integrated volume calculated from the
320 number size distributions for ATom (Brock et al., 2019). A low bias compared to a typical
321 submicron definition can occur in thick biomass burning plumes and in the lower
322 stratosphere at times (Guo et al., 2019). As detailed in Table 1, the accumulation mode
323 for the bulk models discussed in this study overlaps with the size range of the AMS, and
324 for the sectional models (CESM1-CARMA, GEOS-Chem-TOMAS, ECHAM6-HAM) only
325 the bins that match the AMS size range were used. As expected based on the previous
326 discussion, however, a comparison of the total OA calculated by these sectional models
327 with the modeled OA inside the AMS size-range showed small differences (Slopes for



328 ATom-1 linear regressions: CARMA:0.91, TOMAS: 0.94, ECHAM6-HAM 1.00) mostly
329 influenced by the high points in the biomass plumes off Africa).

330 Refractory and non-refractory aerosol composition was also measured using the NOAA
331 Particle Analysis by Laser Mass Spectrometry (PALMS) instrument. PALMS classifies
332 individual aerosol particles into compositional classes including biomass burning (Hudson
333 et al., 2004), sea salt (Murphy et al., 2019), mineral dust (Froyd et al., 2019), and
334 others. Mass concentrations for these particles types are derived by combining PALMS
335 composition data with aerosol size distribution measurements (Froyd et al., 2019). Good
336 agreement overall was found for OA, sulfate and seasalt between the two particle mass
337 spectrometers during ATom once the AMS and PALMS instrument transmissions were
338 accounted for (Guo et al., 2019). For all PALMS data used in this work (biomass burning
339 fraction and dust) the AMS transmission function was applied.

340 For a particular airmass, the mass fraction of biomass burning (BB) aerosol reported by
341 the PALMS instrument $f(\text{BB})_{\text{PALMS}}$ (Thompson and Murphy, 2000; Froyd et al., 2019) was
342 then used to evaluate the degree of BB influence. This parameter correlates quite well
343 with other gas-phase BB tracers, and is more useful as a particle tracer. Importantly, it is
344 not impacted by the long lifetimes of the gas-phase tracers (e.g. 9 months for CH_3CN) and
345 has hence a much higher contrast ratio in the ATom dataset. An airmass was classified
346 as non-BB influenced when $f(\text{BB})_{\text{PALMS}}$ was lower than 0.30 (Hudson et al, 2004) as shown
347 in Figure 2b. It was also used to assess the impact of POA on the total OA burden (next
348 section). For ATom-1 (ATom-2), 76% (97%) of measurements were classified as not
349 influenced by biomass burning.

350 **3.2 Estimation of the POA fraction for the ATom dataset**

351 For model evaluation purposes, it is important to know whether the source of OA is primary
352 or secondary. For ground studies close to sources (e.g. Jimenez et al., 2009) Positive
353 Matrix Factorization of AMS mass spectra (PMF, Ulbrich et al., 2009) can be used to
354 estimate the contribution of primary sources (mostly from transportation, heating, cooking,
355 and biomass burning) to total OA. This approach is not suitable for ATom, since the
356 aerosol is too aged to be effectively classified into POA and SOA by PMF. However, PMF
357 analyses can still be useful for other purposes, such as identifying MSA, as previously
358 observed in remote environments (Hodshire et al., 2019a).



359 Instead, in this work we have estimated POA based on the fact that it is co-emitted with
360 BC as part of the combustion processes releasing both species in source regions, and
361 that BC is not impacted by chemical aging processes over the lifetime of the airmass. We
362 assume non-differential removal (and transport) of the BC fraction relative to the rest of
363 the POA (the two are generally internally mixed, Lee et al., 2015). Table S1 summarizes
364 recent POA/BC and POC/EC emission ratio determinations for urban background sites,
365 which best represent real mixes of pollution sources, and for individual sources of POA
366 (from mobile sources – commonly referred as HOA – and cooking aerosol – COA). Based
367 on Table S1 data, we assume POA to be co-emitted with BC for anthropogenic fossil fuel
368 / urban region POA (herein called FF_{ratio} for simplicity, even though much of it is non-fossil,
369 Zotter et al., 2014; Hayes et al., 2015) at a ratio of 1.55 (average of all studies). Mobile
370 source measurements in general exhibit lower ratios (POA/OA ratio 0.5-1.5) while COA
371 determination typically ranges from 2 to 3. Hence, the ratio used here is a good estimate
372 for a diverse mix of urban sources as appropriate for ATom. For biomass burning sources,
373 we use a value of POA/BC = 13.5 (BB_{ratio}), based on the average of the recent review by
374 Andreae (2019), which included over 200 previous determinations for a variety of fuels
375 and burning conditions. We note the measured total OA/BC of ~3.5 (conservatively
376 assuming that all OA is POA) observed on both ATom missions for the large African-
377 sourced BB plumes over the Equatorial Atlantic. We note that using the larger BB_{ratio} from
378 Andreae (2019) leads to a POA fraction $\gg 100\%$ in the ATom African plumes. We also
379 perform sensitivity studies with values of both FF_{ratio} and BB_{ratio} within the literature range.
380 The PALMS determined mass fraction of biomass impacted aerosol ($f(BB)_{PALMS}$) can then
381 be used to determine a total POA contribution from both types of sources:

$$382 \quad POA = [BC] * (FF_{ratio} + (BB_{ratio} - FF_{ratio}) * f(BB)_{PALMS}) \quad (Eq. 1)$$

383 Further detail is provided in Table S2, which summarizes the POA/BC ratios used in the
384 emission inventories implemented in current models. Overall, there is reasonable
385 agreement with the measurements in Table S1, with FF_{ratio} ranging from ~0.5 for diesel
386 fuels, to >2 for energy production and ~5 for residential emissions (which include some
387 BB). On the other hand, for biomass burning, the emission inventories ratios range from
388 ~5 for crop, to ~15 for forest, and up to ~50 for peatland. While generally consistent with
389 the values discussed by Andreae (2019), they are on the lower end of the ranges
390 discussed in that work. Overall, within the uncertainties of both the measured and model-
391 implemented emission ratios no significant bias is apparent.



392 PALMS detection efficiency increases with size across the accumulation mode, and
393 therefore the $f(\text{BB})$ number fraction is weighted to the larger size end of the accumulation
394 mode. In very clean regions of the upper troposphere (typically $<0.15 \mu\text{g sm}^{-3}$ submicron
395 mass) particles below the PALMS size range can contribute significantly to aerosol mass
396 (Williamson et al., 2019; Guo et al., 2019). If BB particles are not evenly distributed across
397 the entire accumulation mode (due to preferential removal in convective updrafts of
398 primary aerosol, cf. Yu et al., 2019 and Section 4.5; and preferential condensation of SOA
399 on smaller particles), then the $f(\text{BB})$ reported by PALMS will be an overestimation. For the
400 final analysis these periods were left in the dataset, and therefore for the LS the reported
401 POA is likely overestimated for these regions, although their impact on the mass-weighted
402 campaign average is negligible.

403 **3.3 Data processing for comparisons**

404 For the comparisons between the measurements and the various global models, data
405 were averaged both vertically and zonally to minimize the impact of smaller plumes or
406 vertical gradients in aerosol concentrations that might not be captured by coarse resolution
407 models. For the same reason, all data near airports was removed from the datasets prior
408 to analysis (up to about 3 km on the climb in/out). In order to restrict this analysis to the
409 remote troposphere, the last leg of the ATom-1 mission (over the continental US) was
410 taken out of the dataset as well. Data was binned into 5 large latitude regions as shown in
411 Figure 2a including southern polar ($55\text{-}80^\circ\text{S}$, "S.Polar"), southern mid-latitudes ($25\text{-}55^\circ\text{S}$,
412 "S.Mid"), equatorial ($25^\circ\text{S}\text{-}25^\circ\text{N}$, "Equatorial"), northern mid-latitudes ($25\text{-}55^\circ\text{N}$, "N.Mid"),
413 northern polar ($55\text{-}80^\circ\text{N}$, "N.Polar") and analyzed separately for the Pacific and Atlantic
414 basins. For data in each of these latitude regions, altitude profiles were calculated with a
415 constant 600 m altitude resolution. According to both variability in the cleanest air and
416 statistical analysis of the organic background subtraction (Drewnick et al. 2009), the 1σ
417 precision at low concentrations for one-minute data ranged between 20 and 50 ng sm^{-3} ,
418 or a 3σ detection limit between 60 and 150 ng sm^{-3} for the one-minute data (confirmed by
419 frequent filter blanks). On average, each individual point in the profiles represents the
420 average of about 25 min of ATom flight data. At that time resolution, the OA 1σ precision
421 was about 10 ng sm^{-3} . Hence with very few exceptions (10 points for both missions
422 combined), the OA concentrations in the averaged profiles reported are well above the
423 instrumental detection limit in those regions. For model-measurement comparisons along
424 flight tracks, model outputs and measurements were considered at 1-minute time



425 resolution, which corresponds to ~0-700 m vertical resolution and ~0.05-0.15 degrees
426 horizontal resolution. Note that a large fraction of the 1-minute OA values in the remote
427 free troposphere were below the local 3σ detection limit and require averaging before use.
428 We therefore caution future data users that the reported data should be averaged as
429 needed, as replacing below-detection limit values by other values introduces biases on
430 averages. For fractional ratio analysis, measurements were averaged to 5-minute time
431 resolution to reduce the noise in the ratios due to noise in the denominator. The results
432 are not very sensitive to the 5-minute averaging (compared to 1-minute) as shown in
433 Figure S11 for OA to sulfate ratios.

434 Some of the performed analysis required separating the dataset into vertical subsets. In
435 this manuscript, we define the marine boundary layer (MBL) as the region below 1.5 times
436 the calculated boundary layer height in the NCEP global model reanalysis. The free
437 troposphere (FT) includes all data points between the top of MBL and the NCEP
438 tropopause height, and the LS region includes all points above the NCEP tropopause
439 height.

440 **3.4 Submicron aerosol composition**

441 Figure 2b shows that during both NH summer and winter ATom deployments, OA is one
442 of the three dominant components of the measured submicron aerosol in the remote
443 troposphere, together with sulfate and sea salt. During ATom-1, average submicron
444 aerosol concentrations were close to $0.8 \mu\text{g sm}^{-3}$ in the marine boundary layer and
445 biomass burning outflow regions, and ~2 times lower in the free troposphere and lower
446 stratosphere regions. ATom-2 had overall lower average concentrations below $0.4 \mu\text{g sm}^{-3}$
447 (vs. $0.5 \mu\text{g sm}^{-3}$ for ATom-1). As expected, sulfate (sulfuric acid in the lower stratosphere)
448 is the dominant constituent in the MBL (~50%) and LS (50-70%), while the OA contribution
449 is generally below 10% and 40%, respectively in those regions. A large fraction of sea salt
450 aerosol is found in the MBL especially during the NH winter deployment (~30%, see
451 Murphy et al., 2019).

452 OA is found to be a major constituent (~50%) of submicron aerosol in the clean (non-BB
453 influenced) free troposphere. The contribution of OA is 1.4 times larger than that of sulfate
454 during the NH summer, and 1.2 times lower than that of sulfate during the NH winter,
455 which is likely due to a large contribution of the NH sources to SOA production in the NH
456 summer. Biomass-burning events increase the OA contribution relative to that of sulfate,



457 and lead to a higher contribution of OA to total during the ATom-1 mission (stronger BB
458 influence).

459 **3.5 Spatial and vertical distribution of OA**

460 Figure 2a (and Fig. S1) shows the spatial and vertical distribution of OA mass
461 concentrations measured during ATom-1 (and ATom-2) campaigns. Most data were taken
462 over remote oceanic regions (and a few remote continental regions, primarily over the
463 Arctic). The measured OA varies between extremely clean conditions ($< 0.1 \mu\text{g sm}^{-3}$)
464 encountered mostly in the Pacific and Southern Ocean regions and moderately polluted
465 conditions ($> 2 \mu\text{g sm}^{-3}$) in the biomass burning outflow regions. During ATom-1 (August
466 2016), a strong BB influence is observed in the lower troposphere (below 6 km) over the
467 Atlantic basin off the African coast and over California with OA concentrations exceeding
468 $10 \mu\text{g sm}^{-3}$. OA associated with biomass burning is also present in the upper troposphere
469 over equatorial regions and over Alaska, associated with the deep convective transport of
470 biomass burning aerosols. The biomass burning contribution to carbonaceous aerosols in
471 those regions during ATom-1 was also apparent in the black carbon measurements
472 (Katich et al., 2019). ATom-2 was generally less polluted than ATom-1 due to a more
473 limited global influence of biomass burning emissions during that period, and also to a less
474 active photochemistry during winter months in the NH.

475 The measured OA is characterized by a strong latitudinal gradient. Figure 2c shows the
476 average vertical profiles of measured OA over the selected latitudinal bands during August
477 2016. The cleanest airmasses are observed over the remote oceanic regions of the
478 Southern Hemisphere (SH, 25-80°S) with OA mass concentrations below $0.06 \mu\text{g sm}^{-3}$.
479 These extremely low OA concentrations can be explained by the very low influence from
480 continental emission sources, and presumably low marine POA and SOA precursor
481 emissions. This is consistent with low concentrations of gas-phase pollutants (e.g. CO,
482 ethane, propane). An enhancement can be noticed above 10 km in the lower stratosphere.
483 In some cases, this could be related to the long-range transport of biomass burning
484 aerosols from the tropics. By comparison, the Arctic region is more polluted with an order
485 of magnitude higher OA levels compared to its analog of the SH (i.e. OA loadings ranging
486 from 0.1 to $0.5 \mu\text{g sm}^{-3}$). These concentrations are comparable to FT levels measured in
487 the extratropical regions (25-55°N) of the NH. The equatorial marine regions (25°S-25°N)
488 display the highest OA concentrations with a strong gradient between lower and upper
489 troposphere. In the lower troposphere OA, concentrations are close to $1 \mu\text{g sm}^{-3}$, and



490 decrease down to $0.1 \mu\text{g sm}^{-3}$ at altitudes above 4km. The highest OA levels are
491 associated with the African outflow over the southeastern Atlantic Ocean, which results
492 from the transport of the biomass burning smoke from the sub-Saharan regions and
493 increasing urban and industrial air pollution in southern West Africa (Flamant et al., 2018).
494 Figure 2d shows that the Atlantic basin is often more polluted than the Pacific basin, not
495 only because of the African biomass burning influence but also due to the contribution of
496 anthropogenic pollution in the lower troposphere of the NH. It should be noted that while
497 Asian pollution was intermittently sampled in the North Pacific Basin, no broad Asian
498 outflow features were present in these two ATom deployments. Several-fold higher OA
499 concentrations are found near the surface (below 1km) over the southern Pacific
500 compared to that same location in the southern Atlantic, which could be indicative of the
501 stronger emission of marine OA in the Pacific basin.

502 In addition to spatial gradients, a strong summer-to-winter contrast is observed in OA
503 concentrations. Figure 2e shows the ratio between OA vertical profiles measured in the
504 NH summer ATom-1 vs. in the NH winter ATom-2. The NH is more polluted during the NH
505 summer due to the photochemical production of SOA, as well as biomass burning
506 emissions, leading to the tripling of OA concentrations in the extratropical regions (25-
507 80°N) on average regardless of altitude. The doubling of OA loading in the lower
508 troposphere at the equator (25°S-25°N) in the NH summer (August, ATom-1) is strongly
509 influenced by the biomass burning activity in the sub-Saharan African region as already
510 mentioned above. Likewise, OA concentrations are found to be generally higher in the SH
511 during the SH summer. These zonal trends are broadly similar to the ones described in
512 Katich et al (2018) for BC.

513 **4 Model-measurement comparisons**

514 **4.1 Evaluation of predicted OA concentrations**

515 Prior to evaluating model performance in simulating OA, we have assessed the ATom
516 models' ability to simulate sulfate aerosols. According to the model evaluation shown in
517 Table S3, the predicted sulfate concentrations are generally within 40% of the measured
518 values, which is comparable to the AMS measurement uncertainties. The only exception
519 is found for the ECHAM6-HAM model, which overestimates sulfate aerosols by a factor of
520 two. These results imply that most ATom models capture relatively well the overall sulfate
521 burden. However, large root mean square error ($\text{RMSE} > 0.4 \mu\text{g sm}^{-3}$ for ATom-1 and >



522 0.2 $\mu\text{g sm}^{-3}$ for ATom-2) is indicative of their limited skill in reproducing the observed
523 variability in sulfate concentrations.

524 For OA, model evaluation metrics for the entire ATom-1 and ATom-2 campaigns are given
525 in Table 2 for the eight ATom models and their ensemble, as well as the AeroCom-II
526 ensemble. The results show that the normalized mean bias is substantially lower for the
527 ATom model ensemble compared to AeroCom-II decreasing from 74% to 4% for ATom-1
528 and from 137% to 23% to ATom-2, which is within the measurement uncertainty range.
529 The mean temporal correlations are substantially improved from 0.31 (0.38) for AeroCom-
530 II to 0.66 (0.48) for ATom model ensemble during ATom-1 (ATom-2). However, results
531 vary strongly among ATom models. Models using prescribed emissions of non-volatile
532 SOA have the tendency to overestimate the OA concentrations (with ~35-60%
533 overestimation for CESM2-SMP, ~70-100% for ECHAM6-HAM, and up to 150% for GC10-
534 TOMAS during ATom-2), with the exception of the GEOS5-GOCART model that on the
535 contrary underestimates OA concentrations by 5-25%. Models using the VBS
536 parameterization from Pye et al. (2010) tend to underpredict the OA concentrations by 40-
537 50% for GC12-REF and ~20-30% for CESM1-CARMA for ATom-1, most likely due to the
538 excessive evaporation of the formed SOA in remote regions and low yields for
539 anthropogenic SOA (Schroder et al., 2018; Shah et al., 2019). Models using the VBS
540 parameterization from Hodzic et al. (2016) (CESM2-DYN and GC12-DYN) where OA is
541 less volatile and also OA yields are corrected for wall losses in general show an improved
542 agreement with observations especially for CESM2-DYN (within 5-30% for ATom-1 and
543 15-25% during ATom-2).

544 Figure 3 compares the average median ratios between modeled and observed OA
545 concentrations for the ATom and AeroCom-II model ensembles for different regions (BB,
546 MBL, FT, LS). The results show that the median ratio for the ATom model ensemble is
547 close to unity in all regions. This is at least a factor of two improvement compared to
548 AeroCom-II models, which were almost always biased high for the remote regions
549 sampled in ATom. The model spread has also been reduced by a factor of 2-3 in all
550 regions. This reduction in the ensemble spread may partially be explained by a smaller
551 size of the ATom model ensemble, which also includes models that tend to have more
552 focus on OA simulation. In order to explore this further, results for a subset of AeroCom-
553 II models (using models related to those used in ATom) show only a slight reduction
554 (~10%) in the model spread, with however some regional differences i.e. an improved



555 agreement with observations in the MBL, but an increase in the model bias and spread in
556 the LS (Figure S2).

557 **4.2 Evaluation of predicted OA vertical distribution**

558 Figure 4 compares the mean vertical profiles of OA measured during ATom-1 and -2 with
559 the predictions of the model ensemble average based on the eight ATom models (Table
560 1) and 28 AeroCom-II models for the different latitudinal regions of the Pacific and Atlantic
561 basins. Note that the use of a wide logarithmic scale may make the observed differences
562 appear small, although they often reach factors of 2-10 and larger. For AeroCom-II, large
563 latitudinal differences exist in the results with a better performance closer to source
564 regions and large disagreement in the lower stratosphere and remote regions, as already
565 suggested by the mission medians shown in Figure 3. The best AeroCom-II model
566 performance is found over the equator in both basins, where the model ensemble captures
567 within a factor of 2 the observed OA concentrations throughout the troposphere in the
568 Pacific basin, and matches remarkably well the observations in the lower troposphere of
569 the Atlantic basin that is heavily influenced by biomass burning emissions. Reasonable
570 agreement is found for the OA vertical distribution over the NH Atlantic and Pacific oceans,
571 especially in the lower troposphere (< 4 km). The largest model discrepancies (1-2 orders
572 of magnitude) are found in the remote regions of the Southern Ocean and SH mid-latitudes
573 during both seasons and basins. The model overestimation is also large over the NH mid-
574 latitude Pacific basin in the upper troposphere. A spread of 2-3 orders of magnitude is
575 observed around the ensemble average indicating a very large variability in individual
576 model predictions. This evaluation of AeroCom-II models in remote regions is an extension
577 of that performed at the surface for urban and remote stations by Tsigaridis et al. (2014)
578 (as in that previous study, the data and model simulations compared are not synchronous
579 in time). The tendency of the model ensemble to overpredict OA concentrations by a factor
580 of 2 on average in the remote regions is consistent with the transition from the large
581 underprediction in OA near the source region to a slight overprediction of OA in remote
582 continental sites that was reported for most AeroCom-II models (Tsigaridis et al., 2014),
583 and also observed for default parameterizations in other studies (Heald et al., 2011;
584 Hodzic et al., 2016).

585 By comparison, the results of the ATom model ensemble show a much better agreement
586 with observations. The model spread is still substantial, but mostly below a factor of 5.
587 Figures S5 and S7 show OA vertical profiles for individual ATom models and the spread



588 in their results. In most regions, the ATom model ensemble captures reasonably well both
589 the absolute concentrations as well as the shape of the vertical profiles. In the biomass
590 burning outflow and NH mid-latitude regions, the ATom ensemble average better captures
591 the higher OA concentrations in the boundary layer and lower OA concentrations in the
592 lower stratosphere than the AeroCom-II ensemble.

593 **4.3 Oxidation level of organic aerosols (OA/OC ratios)**

594 In addition to OA mass concentrations, we also evaluate the model's ability to simulate
595 their degree of oxygenation, an indicator of their oxidation and aging (Aiken et al., 2008;
596 Kroll et al., 2011). Ambient measurements of the oxidation level of organic particles are
597 limited (Aiken et al., 2008, Canagaratna et al., 2015), and the ATom dataset provides the
598 first global distribution of O/C and OA/OC ratios for the remote aerosol. The OA/OC ratio
599 is an estimate of the average molecular weight of organic matter per carbon weight, and
600 it mostly depends on the oxygen content (i.e. the O/C ratio), in the absence of significant
601 concentrations of organonitrates and -sulfates. It is needed to compare measurements of
602 organic aerosol mass (from e.g. AMS) with organic carbon measurements (from e.g.
603 thermo-optical methods). It is also needed to compare the various types of measurements
604 to model concentrations, which are sometimes carried internally as OA and sometimes as
605 OC. A low OA/OC ratio is indicative of freshly emitted OA from combustion (typically ~1.4),
606 and its value increases with increased processing of organics in the atmosphere. Figure
607 5 shows that in the remote regions the bulk of measured OA/OC ratios during ATom-1
608 and -2 range between 2.2 and 2.5, and are larger than values of 2.1 ± 0.2 found in the
609 polluted US continental outflow regions that were sampled during SEAC4RS, WINTER
610 and DC3 field campaigns (Schroder et al., 2018). These values indicate that remote OA
611 is highly oxidized and chemically processed.

612 Importantly, this ratio is also used to calculate the total OA mass concentration for models
613 that provided their outputs in terms of organic carbon concentrations ($[OA]_i = [OC]_i \times$
614 OA/OC_{ratio}). Most Models use a constant OA/OC ratio, but the value used varies
615 substantially. OA/OC of 1.4 is used in ECHAM6-HAM, whereas 1.8 is used in GEOS5-
616 GOCART and GC10-TOMAS simulations for both POA and SOA. Other models calculated
617 directly SOA concentrations without applying this conversion (CESM1-CARMA, CESM2-
618 SMP, CESM2-DYN, GC12-REF and GC12-DYN), but for POA used the ratio of 1.8
619 (CESM1-CARMA, CESM2-DYN) and 2.1 (GC12-REF and GC12-DYN). Most of the
620 AeroCom-II models used the ratio of 1.4 for all primary and secondary OA (Tsigaridis et



621 al., 2014). The comparison with measurements shows that the measured values are ~40%
622 larger than those assumed in some of the AToM models, and 60-80% larger than used in
623 AeroCom-II models. The comparison between the observed and predicted OA/OC vertical
624 profiles (Fig. S3) shows that AeroCom-II models tend to generally underpredict this ratio,
625 and do not capture its increase in remote regions. As a result, this underestimation of
626 OA/OC ratios and the use of a constant value could substantially impact the comparisons
627 of OA mass concentrations for several models considered in this study (ECHAM6-HAM,
628 GEOS5-GOCART, CESM1-CARMA and GC10-TOMAS). If we correct for the
629 underestimated OA/OC ratio using the AToM measured values of 2.2 (to be conservative)
630 and compare to previously discussed biases in Table 2, the overprediction of the
631 ECHAM6-HAM model is increased to ~110-160%, and that of GC10-TOMAS to 180%
632 during AToM-2 while having ~15% bias in AToM-1, whereas GEOS5-GOCART results
633 now overestimate up to 30% during AToM1, and perform much better during AToM-2.

634 These results demonstrate that current global chemistry-climate models use unrealistically
635 low OA/OC ratios, which results in a large underestimate of the degree of oxidation of OA
636 in remote regions. Inaccurate prediction of OA oxidation as it ages could impact not only
637 the calculations of OA burden, but also its optical properties as the absorption of OA
638 changes with its degree of oxidation (through the formation and destruction of brown
639 carbon, Laskin et al., 2015, Forrister et al., 2015). However, models used in this study did
640 not include these effects.

641 **4.4 Contribution of primary vs. secondary OA**

642 We further assess whether global models can adequately predict the relative contributions
643 of primary and secondary OA. We strive to quantify these fractions with the most
644 straightforward methods (with the fewest assumptions) for both models and
645 measurements. POA concentrations were estimated from the BC measurements by using
646 an emission ratio appropriate to the air mass origin (biomass burning vs. anthropogenic),
647 using the $f(\text{BB})$ mass fraction from the PALMS single particle instrument (see Section 3.2).
648 POA is assumed to be chemically inert, while in reality it can slowly be lost to the gas-
649 phase by heterogeneous chemistry (e.g. George and Abbatt, 2010; Palm et al., 2018).
650 Thus, the observation-based method provides an upper limit to the fraction of POA. The
651 model/measurement comparison is only shown for the CESM and GEOS-Chem model
652 variants, as other participating models do not separate or did not report their POA and
653 SOA fractions. In all simulations, POA was treated as a chemically inert directly emitted



654 primary aerosol species that only undergoes transport, transformation from hydrophobic
655 to hydrophilic state with ageing (1-2 days typically), coagulation, and dry and wet
656 deposition.

657 Figure 6 compares the vertical profiles of measurement-derived POA during ATom-1 and
658 predicted by the CESM2-DYN model over clean remote regions of the Pacific basin and
659 northern polar Atlantic that are not influenced by biomass burning. Comparisons for other
660 models are similar (not shown). Observations show that POA is extremely small in remote
661 regions, whereas the model predicts that about half of the OA is made of POA in those
662 areas. Although the model reproduces quite well the measured total OA, it tends to
663 severely overpredict the amount of POA and underpredict that of SOA over clean remote
664 regions (with the two errors canceling each other when it comes to total OA). Over the
665 biomass burning regions (not shown here) it can be difficult to directly quantify POA and
666 SOA with this method, as total OA remains about constant, while POA decreases with
667 aging and SOA increases (Cubison et al., 2011; Jolleys et al., 2015; Hodshire et al.,
668 2019b). However, given this evolution the method used here would lead to an
669 overestimate of POA for this reason.

670 A more general comparison is made in Figure 7, using the frequency distributions of the
671 measured and simulated fraction of POA/OA, for the free troposphere only (Figure S8
672 shows the corresponding cumulative distributions). Observations indicate that most
673 remote FT airmasses contain less than 10% POA, except for biomass burning plumes that
674 are considered mostly primary. A slightly higher proportion of POA is seen in ATom-2,
675 which is consistent with a slower photochemical production of SOA during NH winter.
676 These results indicate that the remote OA is consistently dominated by SOA regardless
677 of the season and location. The comparison with models reveals a very large discrepancy
678 in the predicted vs. measured POA vs. SOA contributions. Models have a general
679 tendency to severely overpredict the fraction of POA and underpredict that of SOA,
680 displaying a much wider frequency distribution than the measurements (as also shown for
681 POA and SOA vertical profiles for individual models on Figures S5 and S7). In GC12-REF,
682 CESM2-DYN and CESM1-CARMA (without improved in-cloud removal) predictions for
683 ATom-1, more than a half of the remote OA is POA, while that is very rarely observed in
684 the free troposphere (possibly only during strong biomass burning events). Most models
685 fail to reproduce the overwhelming dominance of SOA that is seen in the measurements
686 during ATom-1, while the discrepancies are less severe during NH winter (ATom-2).



687 These seasonal differences suggest that model errors could be partially due to inefficient
688 production of SOA, although removal errors also probably play a major role (see next
689 section).

690 The differences are so large that they are pretty insensitive to details of the POA estimation
691 method from the measurements. As Figure S9 illustrates, the choice of FF_{ratio} has very
692 little impact on the overall distribution of POA. On the other hand, while the BB_{ratio} does
693 impact the overall distribution of POA, it mostly affects the points in the vicinity of the large
694 Atlantic plumes. Since the POA/BC ratio in those plumes is fairly low, (see Section 3.2),
695 using a very large BB_{ratio} mostly leads to an increase of the fraction of the points where
696 $POA > 100\%$. While the large range of published BB_{ratio} for different sources precludes a
697 more accurate estimation by our method, for the purposes of the comparison with the
698 model results we emphasize that even using the largest BB_{ratio} , $f(\text{SOA})$ is still significantly
699 larger in the ATom dataset than in any of the models.

700 A comparison between simulations that have the same treatment of POA, and only differ
701 in their chemistry and removal of SOA (e.g. CESM2-SMP vs. CESM2-DYN; GC12-REF
702 vs. GC12-DYN) indicate that a more complex SOA treatment does not always result in a
703 more accurate simulation of the primary/secondary character of OA, a result that was also
704 found in the AeroCom-II multi-model intercomparison (Tsigaridis et al., 2014).

705 **4.5 Sensitivity to OA formation and removal**

706 In this section, we further investigate some of the possible reasons for the incorrect model
707 predictions of the relative contributions of POA and SOA in remote regions. Given the
708 tendency of models to underestimate OA close to anthropogenic source regions and
709 overestimate OA downwind in past studies (e.g. Heald et al., 2011; Tsigaridis et al., 2014;
710 Hodzic et al., 2016), in this section we investigate the sensitivity of OA to increasing
711 sources and increasing removals. We have performed two additional model simulations
712 to test the sensitivity of the POA/SOA fractions to uncertainties in the representation of (i)
713 wet scavenging, based on the CESM1-CARMA simulations in which we have removed
714 the improvements in the aerosol removal by the convective updrafts (Yu et al., 2019); and
715 of (ii) SOA formation based on the GC12-REF simulations in which we have replaced the
716 SOA formation VBS mechanism (Pye et al., 2010) by an updated VBS mechanism that
717 uses chamber wall-loss corrected SOA yields (Hodzic et al., 2016, same formation
718 scheme that is used in GC12-DYN and CESM2-DYN runs, but with removals kept identical
719 to GC12-REF). The results of these two sensitivity simulations are displayed on Figure 8,



720 which shows measured and predicted mass concentrations of OA, POA, SOA and sulfate
721 for ATom-1 as a function of the number of days since the air mass was processed through
722 convection. One should keep in mind that this is an averaged plot that included airmasses
723 from various regions and altitudes, and not a Lagrangian plot following the same airmass.

724 **Sensitivity to in-cloud scavenging in convective clouds.** Inefficient wet removal of
725 primary OA could contribute to the POA overprediction in global models, especially in the
726 tropics. Previous global model studies have reported two to three orders of magnitude
727 overestimation of primary carbonaceous species such as BC in the free troposphere when
728 the removal in the convective updrafts was not included (e.g. Schwarz et al., 2013, Yu et
729 al., 2019). A strong reduction due to convective removal is also expected for POA
730 concentrations, as POA is a primary species co-emitted with BC at the surface and
731 internally mixed with it (Lee et al., 2015), and that is typically coated by secondary
732 inorganics and organics over short timescales (Petters et al., 2006; Jiang et al., 2010;
733 Wang et al., 2010). Figures 7a and 8 compare the simulations of CESM1-CARMA with
734 and without improved convective in-cloud scavenging during ATom-1. The improved in-
735 cloud scavenging scheme considers aerosol activation into cloud droplets from entrained
736 air above the cloud base, which is more realistic and results in a more efficient removal of
737 aerosols in the upper troposphere by convection. E.g. a two order of magnitude reduction
738 in BC in the upper FT was reported by Yu et al. (2019), resulting in much improved
739 agreement with observations. Similar results were observed for sea salt aerosols in
740 Murphy et al. (2019). Figure 8 shows that all submicron aerosol species simulated in
741 CESM1-CARMA are strongly impacted by the in-cloud removal above the cloud base.
742 POA concentrations are reduced by an order of magnitude while sulfate is reduced by
743 30% leading in both cases to a much-improved agreement with observations. SOA is
744 reduced by ~30% as well, which leads to an underprediction of measured SOA
745 concentrations. The overall impact on OA concentrations is a significant reduction, which
746 leads to ~20% underestimation of OA in the aged remote air during ATom-1.

747 For the CESM2-DYN model that does not have improved in-cloud removal, the reasonable
748 agreement (within 20%) with the observed OA concentrations thus results from
749 coincidental error compensation between the overpredicted POA and underpredicted
750 SOA. The prescribed SOA formation and the artificial 50% adjustment of SOA emissions
751 based on Liu et al. (2012) leads to an overestimation of observed SOA in aged remote
752 airmasses.



753 **Sensitivity to SOA formation.** In addition, we have also tested the sensitivity of the OA
754 composition to the choice of the SOA formation mechanism. Figure 8 compares the results
755 of the GC12-REF model that uses SOA formation yields derived from traditional chamber
756 experiments (Pye et al., 2010) and those corrected for losses of organic vapors onto
757 chamber walls as proposed in Hodzic et al. (2016). Previous studies have reported that
758 chamber wall losses could lead up to a factor of 4 underprediction of formed SOA (Zhang
759 et al., 2014; Krechmer et al., 2016). The comparison shows a factor of 3 increase in SOA
760 concentrations when the updated SOA formation is considered leading to a much better
761 agreement with the observed SOA as well as the observed total OA. GC12-REF predicts
762 well the amount of POA and overpredicts somewhat the amount of sulfate aerosols, which
763 is expected as it already includes the improved aerosol removal in convective updrafts
764 (Wang et al., 2014). Figure S6 also shows that POA vertical distribution is well captured
765 in GEOS-Chem in most regions, except over the polar north Pacific. It should be noted
766 that these results are consistent with the POA/OA frequency distribution shown in Figure
767 7 (e.g. GC12-REF POA/OA is shifted rightward compared to observations in Figure 7,
768 which is consistent with the fact that POA is about the right amount, and OA is
769 underpredicted in Figure 8).

770 These sensitivity simulations suggest that a stronger convective removal of POA and a
771 stronger production of SOA might be needed to correctly represent not only the total OA
772 concentrations but also its primary and secondary nature in remote free troposphere and
773 remote ocean regions. Accurate predictions of the OA concentration, composition, and
774 source contributions for the right reasons are key for accurately predicting their lifecycle
775 and radiative impacts. Only when there is confidence that the sources are accurately
776 predicted, we can have confidence in OA predictions for pre-industrial and future
777 conditions, as well as to evaluate PM mitigation strategies.

778 **4.6 OA and sulfate relative contributions in FT**

779 Finally, we assess the model ability to predict relative amounts of OA and sulfate in the
780 free troposphere where they are the two major constituents of the submicron aerosol
781 (Figure 2b). Accurate predictions of their relative contributions are crucial to determine the
782 hygroscopicity of the submicron aerosol, and its ability to serve as a cloud condensation
783 nuclei (CCN) in the remote free troposphere (Carslaw et al., 2013; Brock et al., 2016).

784 Figure 9a compares the average measured relative fractions of sulfate (36%) and
785 carbonaceous aerosols (OA=59% and BC=5%) in the FT with those predicted by



786 individual models during ATom-1. The CESM2 models best reproduce the observed
787 relative contributions, with a slight underestimation of OA (57% instead of 59%) for
788 CESM2-DYN, and a slight overestimation of OA (63% instead of 59%) for CESM2-SMP.
789 GEOS5-GOCART has 15% more OA relative to sulfate than observed. All other models
790 underestimate both OA and BC relative fractions. For instance, in GC12-REF and -DYN,
791 both the BC and OA fractions are ~40% (relative) lower than observed.

792 Figure 9b shows the frequency distribution of observed and predicted fractions of OA
793 relative to sulfate during ATom-1 and -2 in the free troposphere. Most models fail to
794 reproduce the relatively uniform nature of the observed distributions during ATom-1, with
795 typically narrower model shapes around a preferred ratio. The NH summer measurements
796 indicate that OA > sulfate in ~55% of the samples (consistent with Fig. 2b), while models
797 generally tend to underestimate the relative OA contribution. In particular, GEOS-Chem
798 and ECHAM6-HAM tend to overestimate the relative contribution of sulfate. A better
799 agreement is found for GEOS5-GOCART, CESM1-CARMA and CESM2-DYN, which
800 follow more closely the shape of the observed distribution. The comparisons also suggest
801 that the more complex SOA treatment of SOA formation and removal proposed by Hodzic
802 et al. (2016) in the same host model leads to an improved agreement with observations
803 (e.g. CESM2-DYN vs. CESM2-SMP; GC12-DYN vs. GC12-REF). It should be noted that
804 CESM2-SMP uses fixed SOA yields that were increased by 50% as suggested by Liu et
805 al. (2012), leading to an overestimation of the relative contribution of OA compared to that
806 of sulfate in the free troposphere. During the NH winter (ATom-2), measurements show a
807 somewhat higher proportion of sulfate aerosols (vs. ATom-1), which is consistent with a
808 slower production of SOA in the NH during winter and a reduced influence of biomass
809 burning. Similar conclusions are found for the evaluation of different models. It is worth
810 mentioning that the comparison performed for the whole ATom-1 and 2 dataset (not
811 shown) leads to similar results with even slightly stronger overestimation of the sulfate
812 relative contribution compared to OA.

813 The discrepancies between the observed and predicted composition of submicron aerosol
814 over remote regions can be quite large for other constituents as well. Figure 10 shows the
815 comparison of measured and predicted composition of the submicron aerosol over the
816 Southern Ocean (during the NH winter) where the disagreement in simulated sea salt,
817 nitrates, ammonium, and MSA often exceeds the contribution of OA. While the
818 observations show a more uniform distribution of non-marine aerosol with higher values



819 in the mid and upper troposphere, respectively, most models tend to simulate highest
820 fractions of sulfate and OA towards the tropopause. This may also be explained by the
821 uncertainties in modeled wet removal of aerosol that has been discussed above.

822

823 **5 Conclusions and implications**

824 Our understanding and representation in global models of the lifecycle of the OA remain
825 highly uncertain, especially in remote regions where constraints from measurements have
826 been very sparse. We have performed a systematic evaluation of the performance of eight
827 global chemistry climate models and of 28 AeroCom-II models in simulating the latitudinal
828 and vertical distribution of OA and its composition in the remote regions of the Atlantic and
829 Pacific marine boundary layer, free troposphere and lower stratosphere, using the unique
830 measurements from the ATom campaign. Our simulations are conducted for both ATom-
831 1 and ATom-2 deployments that took place in August 2016 and February 2017,
832 respectively. The main conclusions of the comparison are as follows:

- 833 • The AeroCom-II ensemble average tends to be biased high by a factor of 2-5 in
834 comparison to measured vertical OA profiles in the remote atmosphere during both
835 NH summer and NH winter. The ensemble spread increases from a factor of 40 in the
836 NH source regions to a factor of 1000 in remote regions of the Southern Ocean. The
837 evaluation of AeroCom-II models in the remote regions provides an extension of the
838 previous evaluation with continental ground data by Tsigaridis et al. (2014). We note
839 that the data from the AeroCom-II models were based on monthly mean values from
840 a different simulated year than the ATom campaigns; however, the consistent model
841 biases are strong enough that we would not expect our conclusions to change for a
842 different modeled year.
- 843 • The results of the ATom model ensemble used in this work show a much better
844 agreement with the OA observations in all regions and reduced model variability.
845 However, some of the agreement is for the wrong reasons, as most models severely
846 overestimate the contribution of POA and underestimate the contribution of SOA to
847 total OA. Sensitivity simulations indicate that the POA overestimate in CESM could be
848 due to an inadequate representation of primary aerosol removal by convective clouds,
849 (additional convective removal per Yu et al. (2019) in CESM1-CARMA led to a better
850 agreement with observations). Most models have insufficient production of SOA, and



851 sensitivity studies indicate that a stronger production of SOA is needed to capture the
852 measured concentrations. The photochemical ageing of POA which was not
853 considered here (unlike for SOA) could also contribute to the model overestimation.
854 The compensation between errors in POA and SOA in remote regions is however a
855 recurring issue in OA modeling (de Gouw and Jimenez, 2009). For instance, it was
856 found in the urban outflow regions such as Mexico City during MILAGRO 2006 field
857 campaign (Fast et al., 2009; Hodzic et al., 2009); Paris during MEGAPOLI 2009
858 (Zhang et al., 2013); the Los Angeles area during CalNex-2010 (Baker et al., 2015;
859 Woody et al., 2016); the NE US outflow during WINTER 2015 (Schroder et al., 2018;
860 Shah et al., 2019).

861 • Additional errors in simulated OA concentrations can arise from the use of too low
862 OA/OC ratios when model results (often calculated as OC) are converted to OA for
863 comparison with measurements. We note that OA is the most atmospherically-relevant
864 quantity, while OC is an operational quantity, partially a relic from a period in which
865 only OC could be separately quantified (although also of some use for carbon budget
866 studies). It should also be noted that most emission inventories still use OC as the
867 primary variable, which is why the use of accurate OA/OC ratios is still key for all
868 models. We show that the OA/OC ratio used in most models is too low compared to
869 measured values that range mostly from 2.2 to 2.5, resulting in errors in OA mass of
870 ~70% for AeroCom-II models and ~30% for current models that use organic carbon to
871 track OA mass. Remote OA is thus highly oxidized and chemically processed. These
872 results demonstrate that current global chemistry-climate models underestimate the
873 degree of oxidation of OA in remote regions and need to consider further chemical
874 ageing of OA, which could impact the calculations of its burden, and optical and
875 hygroscopic properties.

876 • The results also show that in most models (except CESM2) the predicted OA
877 contribution to the total submicron aerosol is underestimated relative to sulfate in the
878 remote FT where OA and sulfate are the dominant submicron aerosols (important for
879 climate). Accurate predictions of composition of submicron particles remains
880 challenging in remote regions and should be the topic of future studies.

881 Key implications of our results are: (i) Model errors on the relative contribution of POA and
882 SOA to OA reduce our confidence on the ability to simulate radiative forcing over time or
883 OA health impacts; (ii) Model errors for the relative contributions of sulfate and organics
884 to the submicron aerosol in the free troposphere could lead to errors in the predicted CCN



885 or radiative forcing of aerosols as inorganics are more hygroscopic than OA; (iii) the OA
886 system seems to be more dynamic with a need for an enhanced removal of primary OA,
887 and a stronger production of secondary OA in global models to provide a better agreement
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907 comparisons, performed certain model simulations, and wrote most of the manuscript.
908 Campuzano-Jost P. performed the analysis of the models/measurements comparisons,
909 performed AMS measurements, and contributed to the manuscript. Jimenez J. contributed
910 to the interpretation of the results and the elaboration of the manuscript. All other authors
911 provided either measurements or model outputs that are used in the paper, and feedback
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913 **References**

914 Aiken, A.C., P.F. DeCarlo, J.H. Kroll, D.R. Worsnop, J.A. Huffman, K. Docherty, I.M.
915 Ulbrich, C. Mohr, J.R. Kimmel, D. Sueper, Q. Zhang, Y. Sun, A. Trimborn, M. Northway,
916 P.J. Ziemann, M.R. Canagaratna, T.B. Onasch, R. Alfarra, A.S.H. Prevot, J. Dommen, J.



- 917 Duplissy, A. Metzger, U. Baltensperger, and J.L. Jimenez. O/C and OM/OC Ratios of
918 Primary, Secondary, and Ambient Organic Aerosols with High Resolution Time-of-Flight
919 Aerosol Mass Spectrometry *Environmental Science & Technology*, 42, 4478–4485, doi:
920 10.1021/es703009q, 2008.
- 921 Andreae, M. O.: Emission of trace gases and aerosols from biomass burning – An updated
922 assessment, *Atmos. Chem. Phys. Discuss.*, 1–27, doi:10.5194/acp-2019-303, 2019.
- 923 Bey, I., Jacob, D. J., Yantosca, R. M., and Logan, J. A.: Global modeling of tropospheric
924 chemistry with assimilated meteorology: model description and evaluation, *J Geophys.*
925 *Res.*, 106, 23073–23095, 2011.
- 926 Baker, K. R., Carlton, A. G., Kleindienst, T. E., Offenberg, J. H., Beaver, M. R., Gentner,
927 D. R., Goldstein, A. H., Hayes, P. L., Jimenez, J. L., Gilman, J. B., de Gouw, J. A., Woody,
928 M. C., Pye, H. O. T., Kelly, J. T., Lewandowski, M., Jaoui, M., Stevens, P. S., Brune, W.
929 H., Lin, Y.-H., Rubitschun, C. L., and Surratt, J. D.: Gas and aerosol carbon in California:
930 comparison of measurements and model predictions in Pasadena and Bakersfield, *Atmos.*
931 *Chem. Phys.*, 15, 5243-5258, <https://doi.org/10.5194/acp-15-5243-2015>, 2015.
- 932 Bian, H., Froyd, K., Murphy, D. M., Dibb, J., Chin, M., Colarco, P. R., Darmenov, A., da
933 Silva, A., Kucsera, T. L., Schill, G., Yu, H., Bui, P., Dollner, M., Weinzierl, B., and Smirnov,
934 A.: Observationally constrained analysis of sea salt aerosol in the marine atmosphere,
935 *Atmos. Chem. Phys. Discuss.*, <https://doi.org/10.5194/acp-2019-18>, 2019.
- 936 Bian, H., Chin, M., Hauglustaine, D. A., Schulz, M., Myhre, G., Bauer, S. E., Lund, M. T.,
937 Karydis, V. A., Kucsera, T. L., Pan, X., Pozzer, A., Skeie, R. B., Steenrod, S. D., Sudo, K.,
938 Tsigaridis, K., Tsimpidi, A. P., and Tsyro, S. G.: Investigation of global nitrate from the
939 AeroCom Phase III experiment, *Atmos. Chem. Phys.*, 17, 12911-12940,
940 <https://doi.org/10.5194/acp-17-12911-2017>, 2017.
- 941 Boucher, O., D. Randall, P. Artaxo, C. Bretherton, G. Feingold, P. Forster, V.-M.
942 Kerminen, Y. Kondo, H. Liao, U. Lohmann, P. Rasch, S.K. Satheesh, S. Sherwood, B.
943 Stevens and X.Y. Zhang: Clouds and Aerosols. In: *Climate Change 2013: The Physical*
944 *Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the*
945 *Intergovernmental Panel on Climate Change* [Stocker, T.F., D. Qin, G.-K. Plattner, M.
946 Tignor, S.K. Allen, J. Boschung, A. Nauels, Y. Xia, V. Bex and P.M. Midgley (eds.)].
947 Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 2013.



- 948 Bowman, K. P.: Large-scale isentropic mixing properties of the Antarctic polar vortex from
949 analyzed winds. *J. Geophys. Res.*, 98 (D12), 23 013–23 027, 1993.
- 950 Brock, C. A., Wagner, N. L., Anderson, B. E., Attwood, A. R., Beyersdorf, A., Campuzano-
951 Jost, P., Carlton, A. G., Day, D. A., Diskin, G. S., Gordon, T. D., Jimenez, J. L., Lack, D.
952 A., Liao, J., Markovic, M. Z., Middlebrook, A. M., Ng, N. L., Perring, A. E., Richardson, M.
953 S., Schwarz, J. P., Washenfelder, R. A., Welti, A., Xu, L., Ziemba, L. D., and Murphy, D.
954 M.: Aerosol optical properties in the southeastern United States in summer – Part 1:
955 Hygroscopic growth, *Atmos. Chem. Phys.*, 16, 4987-5007, [https://doi.org/10.5194/acp-16-](https://doi.org/10.5194/acp-16-4987-2016)
956 4987-2016, 2016.
- 957 Brock, C. A., Williamson, C., Kupc, A., Froyd, K. D., Erdesz, F., Wagner, N., Richardson,
958 M., Schwarz, J. P., Gao, R.-S., Katich, J. M., Campuzano-Jost, P., Nault, B. A., Schroder,
959 J. C., Jimenez, J. L., Weinzierl, B., Dollner, M., Bui, T. and Murphy, D. M.: Aerosol size
960 distributions during the Atmospheric Tomography Mission (ATom): methods,
961 uncertainties, and data products, *Atmos. Meas. Tech.*, 12(6), 3081–3099,
962 [doi:10.5194/amt-12-3081-2019](https://doi.org/10.5194/amt-12-3081-2019), 2019.
- 963 Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q.,
964 Onasch, T. B., Drewnick, F., Coe, H., Middlebrook, A. M., Delia, A., Williams, L. R.,
965 Trimborn, A. M., Northway, M. J., DeCarlo, P. F., Kolb, C. E., Davidovits, P., and Worsnop,
966 D. R.: Chemical and microphysical characterization of ambient aerosols with the Aerodyne
967 Aerosol Mass Spectrometer, *Mass Spectrom. Rev.*, 26, 185–222,
968 <https://doi.org/10.1002/mas.20115>, 2007.
- 969 Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P.,
970 Hildebrandt Ruiz, L., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue, N.
971 M., Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of organic
972 compounds using aerosol mass spectrometry: characterization, improved calibration, and
973 implications, *Atmos. Chem. Phys.*, 15, 253-272, <https://doi.org/10.5194/acp-15-253-2015>,
974 2015.
- 975 Carslaw, K. S., Lee, L. a, Reddington, C. L., Pringle, K. J., Rap, A., Forster, P. M., Mann,
976 G. W., Spracklen, D. V, Woodhouse, M. T., Regayre, L. a and Pierce, J. R.: Large
977 contribution of natural aerosols to uncertainty in indirect forcing., *Nature*, 503(7474), 67–
978 71, [doi:10.1038/nature12674](https://doi.org/10.1038/nature12674), 2013.



- 979 Colarco, P., da Silva, A., Chin, M., and Diehl, T.: Online simulations of global aerosol
980 distributions in the NASA GEOS-4 model and comparisons to satellite and ground based
981 aerosol optical depth, *J. Geophys. Res.*, 115, D14207, doi:10.1029/2009JD012820, 2010.
- 982 Cubison, M. J., Ortega, A. M., Hayes, P. L., Farmer, D. K., Day, D., Lechner, M. J., Brune,
983 W. H., Apel, E., Diskin, G. S., Fisher, J. A., Fuelberg, H. E., Hecobian, A., Knapp, D. J.,
984 Mikoviny, T., Riemer, D., Sachse, G. W., Sessions, W., Weber, R. J., Weinheimer, A. J.,
985 Wisthaler, A., and Jimenez, J. L.: Effects of aging on organic aerosol from open biomass
986 burning smoke in aircraft and laboratory studies, *Atmos. Chem. Phys.*, 11, 12049-12064,
987 <https://doi.org/10.5194/acp-11-12049-2011>, 2011.
- 988 DeCarlo, P., Slowik, J., Worsnop, D., Davidovits, P., Jimenez, J., Stainken, K., Williams,
989 L., Jayne, J., Kolb, C. and Rudich, Y.: Particle Morphology and Density Characterization
990 by Combined Mobility and Aerodynamic Diameter Measurements. Part 1: Theory, *Aerosol*
991 *Sci. Technol.*, 38(12), 1185–1205, doi:10.1080/027868290903907, 2004.
- 992 DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C.,
993 Gonin, M., Fuhrer, K., Horvath, T., Docherty, K. S., Worsnop, D. R., and Jimenez, J. L.:
994 Field-Deployable, High-Resolution, Time-of-Flight Aerosol Mass Spectrometer, *Anal.*
995 *Chem.*, 78, 8281–8289, <https://doi.org/10.1021/ac061249n>, 2006.
- 996 de Gouw, J., and J.L. Jimenez. Organic Aerosols in the Earth's Atmosphere.
997 *Environmental Science & Technology*, 43, 7614–7618, 2009. DOI: 10.1021/es9006004
- 998 Dentener, F., Kinne, S., Bond, T., Boucher, O., Cofala, J., Generoso, S., Ginoux, P., Gong,
999 S., Hoelzemann, J. J., Ito, A., Marelli, L., Penner, J. E., Putaud, J.-P., Textor, C., Schulz,
1000 M., van der Werf, G. R., and Wilson, J.: Emissions of primary aerosol and precursor gases
1001 in the years 2000 and 1750 prescribed data-sets for AeroCom, *Atmospheric Chemistry*
1002 *and Physics*, 6, 4321–4344, [https://doi.org/10.5194/acp-](https://doi.org/10.5194/acp-6-4321-2006)
1003 <https://www.atmos-chem-phys.net/6/4321/2006/>, 2006.
- 1004 Drewnick, F., Hings, S. S., Alfarra, M. R., Prevot, A. S. H. and Borrmann, S.: Aerosol
1005 quantification with the Aerodyne Aerosol Mass Spectrometer: detection limits and ionizer
1006 background effects, *Atmos. Meas. Tech.*, 2(1), 33–46, 2009.
- 1007 Emmons, L. K., Orlando, J. J., Tyndall, G., Schwantes, R. H., Kinnison, D. E., Marsh, D.
1008 R., Mills, M. J., Tilmes, S., and Lamarque, J.-F.: The MOZART Chemistry Mechanism in
1009 the Community Earth System Model version 2 (CESM2), to be Submitted to *J. Adv.*
1010 *Modeling Earth Systems*, 2019.



- 1011 Fast, J., Aiken, A. C., Allan, J., Alexander, L., Campos, T., Canagaratna, M. R., Chapman,
1012 E., DeCarlo, P. F., de Foy, B., Gaffney, J., de Gouw, J., Doran, J. C., Emmons, L., Hodzic,
1013 A., Herndon, S. C., Huey, G., Jayne, J. T., Jimenez, J. L., Kleinman, L., Kuster, W., Marley,
1014 N., Russell, L., Ochoa, C., Onasch, T. B., Pekour, M., Song, C., Ulbrich, I. M., Warneke,
1015 C., Welsh-Bon, D., Wiedinmyer, C., Worsnop, D. R., Yu, X.-Y., and Zaveri, R.: Evaluating
1016 simulated primary anthropogenic and biomass burning organic aerosols during
1017 MILAGRO: implications for assessing treatments of secondary organic aerosols, *Atmos.*
1018 *Chem. Phys.*, 9, 6191-6215, <https://doi.org/10.5194/acp-9-6191-2009>, 2009.
- 1019 Flamant, C., Knippertz, P., Fink, A. H., Akpo, A., Brooks, B., Chiu, C. J., Coe, H., Danuor,
1020 S., Evans, M., Jegede, O., Kalthoff, N., Konaré, A., Lioussé, C., Lohou, F., Mari, C.,
1021 Schlager, H., Schwarzenboeck, A., Adler, B., Amekudzi, L., Aryee, J., Ayoola, M.,
1022 Batenburg, A. M., Bessardon, G., Borrmann, S., Brito, J., Bower, K., Burnet, F., Catoire,
1023 V., Colomb, A., Den-jean, C., Fosu-Amankwah, K., Hill, P. G., Lee, J., Lathon, M.,
1024 Maranan, M., Marsham, J., Meynadier, R., Ngamini, J.-B., Rosenberg, P., Sauer, D.,
1025 Smith, V., Stratmann, G., Taylor, J. W., Voigt, C., and Yoboué, V.: The Dynamics-Aerosol-
1026 Chemistry- Cloud Interactions in West Africa field campaign: Overview and research
1027 highlights, *B. Am. Meteorol. Soc.*, 99, 83–104, [https://doi.org/10.1175/BAMS-D-16-](https://doi.org/10.1175/BAMS-D-16-1028)
1028 0256.1, 2018.
- 1029 Forrister, H., Liu, J., Scheuer, E., Dibb, J., Ziemba, L., Thornhill, K. L., Anderson, B.,
1030 Diskin, G., Perring, A. E., Schwarz, J. P., Campuzano-Jost, P., Day, D. A., Palm, B. B.,
1031 Jimenez, J. L., Nenes, A. and Weber, R. J.: Evolution of brown carbon in wildfire plumes,
1032 *Geophys. Res. Lett.*, 42(11), doi:10.1002/2015GL063897, 2015.
- 1033 Fountoukis, C. and Nenes, A.: ISORROPIA II : a computationally efficient thermodynamic
1034 equilibrium model for K^+ - Ca^{2+} - Mg^{2+} - Na^+ - SO_4^{2-} - NO_3^- - Cl^- - H_2O aerosols, *Atmos. Chem.*
1035 *Phys.*, 7, 4639–4659, 2007.
- 1036 Froyd, K. D., Murphy, D. M., Brock, C. A., Campuzano-Jost, P., Dibb, J. E., Jimenez, J.-
1037 L., Kupc, A., Middlebrook, A. M., Schill, G. P., Thornhill, K. L., Williamson, C. J., Wilson,
1038 J. C., and Ziemba, L. D.: A new method to quantify mineral dust and other aerosol species
1039 from aircraft platforms using single particle mass spectrometry, *Atmos. Meas. Tech.*
1040 *Discuss.*, <https://doi.org/10.5194/amt-2019-165>, in review, 2019.
- 1041 George, I. J. and Abbatt, J. P. D.: Heterogeneous oxidation of atmospheric aerosol
1042 particles by gas-phase radicals, *Nature Chemistry*, 2, 713–722, 2010.



- 1043 Gettelman, A., Mills, M. J., Kinnison, D. e., Garcia, R. R., Smith, A. K., Marsh, D. R.,
1044 Tilmes, S., Vitt, F., Bardeen, C. G., McInerny, J., Liu, H.-L., Solomon, S. C., Polvani, L.
1045 M., Emmons, L. K., Lamarque, J.-F., Richter, J. H., Glanville, A. S., Bacmeister, J. T.,
1046 Phillips, A. S., Neale, R. B., Simpson, I. R., DuVivier, A. K., Hodzic, A., Randel, W. J.: The
1047 Whole Atmosphere Community Climate Model Version 6 (WACCM6), *J. of Geophys. Res.*,
1048 in review, 2019.
- 1049 Giglio, L., Randerson, J. T. and Werf, G. R.: Analysis of daily, monthly, and annual burned
1050 area using the fourth-generation global fire emissions database (GFED4), *J. Geophys.*
1051 *Res. Biogeosciences*, 118, 317–328, 2013.
- 1052 Guo H, Campuzano-Jost, P, Nault, B.A. Schroder, J.C. Day, D.A. Williamson, C, Kupc, A,
1053 Brock, C.A. Froyd, K. Jimenez, J.L.: Evaluating the Aerosol Mass Spectrometer Quantified
1054 Submicron Aerosol Volume during the Atmospheric Tomography Mission (ATom), to be
1055 submitted to AMTD, 2019.
- 1056 Guenther, A. B., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L. K.,
1057 and Wang, X.: The Model of Emissions of Gases and Aerosols from Nature version 2.1
1058 (MEGAN2.1): an extended and updated framework for modeling biogenic emissions,
1059 *Geosci. Model Dev.*, 5, 1471–1492, <https://doi.org/10.5194/gmd-5-1471-2012>,
1060 <https://www.geosci-model-dev.net/5/1471/2012/>, 2012.
- 1061 Hayes, P. L., Carlton, A. G., Baker, K. R., Ahmadov, R., Washenfelder, R. A., Alvarez, S.,
1062 Rappenglück, B., Gilman, J. B., Kuster, W. C., de Gouw, J. A., Zotter, P., Prévôt, A. S. H.,
1063 Szidat, S., Kleindienst, T. E., Offenberg, J. H., Ma, P. K., and Jimenez, J. L.: Modeling the
1064 formation and aging of secondary organic aerosols in Los Angeles during CalNex 2010,
1065 *Atmos. Chem. Phys.*, 15, 5773-5801, <https://doi.org/10.5194/acp-15-5773-2015>, 2015.
- 1066 Hallquist, M., Wenger, J. C., Baltensperger, U., Rudich, Y., Simpson, D., Claeys, M.,
1067 Dommen, J., Donahue, N. M., George, C., Goldstein, A. H., Hamilton, J. F., Herrmann, H.,
1068 Hoffmann, T., Iinuma, Y., Jang, M., Jenkin, M. E., Jimenez, J. L., Kiendler-Scharr, A.,
1069 Maenhaut, W., McFiggans, G., Mentel, Th. F., Monod, A., Prévôt, A. S. H., Seinfeld, J. H.,
1070 Surratt, J. D., Szmigielski, R., and Wildt, J.: The formation, properties and impact of
1071 secondary organic aerosol: current and emerging issues, *Atmos. Chem. Phys.*, 9, 5155-
1072 5236, <https://doi.org/10.5194/acp-9-5155-2009>, 2009.
- 1073 Heald, C. L., Coe, H., Jimenez, J. L., Weber, R. J., Bahreini, R., Middlebrook, A. M.,
1074 Russell, L. M., Jolleys, M., Fu, T.-M., Al-lan, J. D., Bower, K. N., Capes, G., Crosier, J.,



- 1075 Morgan, W. T., Robinson, N. H., Williams, P. I., Cubison, M. J., DeCarlo, P. F., and Dunlea,
1076 E. J.: Exploring the vertical profile of atmospheric organic aerosol: comparing 17 aircraft
1077 field campaigns with a global model, *Atmos. Chem. Phys.*, 11, 12673–12696,
1078 doi:10.5194/acp-11-12673-2011, 2011.
- 1079 Hodshire, A. L., Campuzano-Jost, P., Kodros, J. K., Croft, B., Nault, B. A., Schroder, J.
1080 C., Jimenez, J. L. and Pierce, J. R.: The potential role of methanesulfonic acid (MSA) in
1081 aerosol formation and growth and the associated radiative forcings, *Atmos. Chem. Phys.*,
1082 19(5), 3137–3160, doi:10.5194/acp-19-3137-2019, 2019a.
- 1083 Hodshire, A., Akherati, M., Alvarado, B., Brown-Steiner, S., Jathar, J.L., Jimenez, S.
1084 Kreidenweis, C., Lonsdale, T., Onasch, A., Ortega, J., Pierce, J. R.: Aging Effects on Biomass
1085 Burning Aerosol Mass and Composition: A Critical Review of Field and Laboratory
1086 Studies. *Environ. Sci. Technol.*, submitted, 2019b.
- 1087 Hodzic, A., Jimenez, J. L., Madronich, S., Aiken, A. C., Bessagnet, B., Curci, G., Fast, J.,
1088 Lamarque, J.-F., Onasch, T. B., Roux, G., Schauer, J. J., Stone, E. A., and Ulbrich, I. M.:
1089 Modeling organic aerosols during MILAGRO: importance of biogenic secondary organic
1090 aerosols, *Atmos. Chem. Phys.*, 9, 6949–6981, <https://doi.org/10.5194/acp-9-6949-2009>,
1091 2009.
- 1092 Hodzic, A., Aumont, B., Knote, C., Lee-Taylor, J., Madronich, S., and Tyndall, G.: Volatility
1093 dependence of Henry's law constants of condensable organics: Application to estimate
1094 depositional loss of secondary organic aerosols, *Geophys. Res. Lett.*, 41, 4795–4804,
1095 doi:10.1002/2014GL060649, 2014.
- 1096 Hodzic, A., Madronich, S., Kasibhatla, P. S., Tyndall, G., Aumont, B., Jimenez, J. L., Lee-
1097 Taylor, J., and Orlando, J.: Organic photolysis reactions in tropospheric aerosols: effect
1098 on secondary organic aerosol formation and lifetime, *Atmos. Chem. Phys.*, 15, 9253–9269,
1099 <https://doi.org/10.5194/acp-15-9253-2015>, 2015.
- 1100 Hodzic, A., Kasibhatla, P. S., Jo, D. S., Cappa, C. D., Jimenez, J. L., Madronich, S., and
1101 Park, R. J.: Rethinking the global secondary organic aerosol (SOA) budget: stronger
1102 production, faster removal, shorter lifetime, *Atmos. Chem. Phys.*, 16, 7917–7941,
1103 <https://doi.org/10.5194/acp-16-7917-2016>, 2016.
- 1104 Huang, K., Fu, J. S., Prikhodko, V. Y., Storey, J. M., Romanov, A., Hodson, E. L., Cresko,
1105 J., Morozova, I., Ignatieva, Y., and Cabaniss, J.: Russian anthropogenic black carbon:



- 1106 Emission reconstruction and Arctic black carbon simulation, *J. Geophys. Res. Atmos.*,
1107 120, 11,306– 11,333, doi:10.1002/2015JD023358, 2015.
- 1108 Hudson, P. K., Murphy, D. M., Cziczo, D. J., Thomson, D. S., de Gouw, J. A., Warneke,
1109 C., Holloway, J., Jost, H. J. and Hübler, G.: Biomass-burning particle measurements:
1110 Characteristics composition and chemical processing, *J. Geophys. Res. D Atmos.*,
1111 109(23), 1–11, doi:10.1029/2003JD004398, 2004.
- 1112 Janssens-Maenhout, G., Crippa, M., Guizzardi, D., Dentener, F., Muntean, M., Pouliot,
1113 G., Keating, T., Zhang, Q., Kurokawa, J., Wankmüller, R., Denier van der Gon, H.,
1114 Kuenen, J. J. P., Klimont, Z., Frost, G., Darras, S., Koffi, B., and Li, M.: HTAP_v2.2: a
1115 mosaic of regional and global emission grid maps for 2008 and 2010 to study hemispheric
1116 transport of air pollution, *Atmos. Chem. Phys.*, 15, 11411–11432,
1117 <https://doi.org/10.5194/acp-15-11411-2015>, 2015.
- 1118 Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prevot, a. S. H., Zhang, Q., Kroll, J.
1119 H., DeCarlo, P. F., Allan, J. D., Coe, H., Ng, N. L., Aiken, a. C., Docherty, K. S., Ulbrich, I.
1120 M., Grieshop, A. P., Robinson, a. L., Duplissy, J., Smith, J. D., Wilson, K. R., Lanz, V. a.,
1121 Hueglin, C., Sun, Y. L., Tian, J., Laaksonen, A., Raatikainen, T., Rautiainen, J.,
1122 Vaattovaara, P., Ehn, M., Kulmala, M., Tomlinson, J. M., Collins, D. R., Cubison, M. J.,
1123 Dunlea, E. J., Huffman, J. A., Onasch, T. B., Alfarra, M. R., Williams, P. I., Bower, K.,
1124 Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K., Salcedo,
1125 D., Cottrell, L., Griffin, R., Takami, A., Miyoshi, T., Hatakeyama, S., Shimono, A., Sun, J.
1126 Y., Zhang, Y. M., Dzepina, K., Kimmel, J. R., Sueper, D., Jayne, J. T., Herndon, S. C.,
1127 Trimborn, a. M., Williams, L. R., Wood, E. C., Middlebrook, A. M., Kolb, C. E.,
1128 Baltensperger, U., Worsnop, D. R., Worsnop, D. R., Dunlea, J., Huffman, J. A., Onasch,
1129 T. B., Alfarra, M. R., Williams, P. I., Bower, K., Kondo, Y., Schneider, J., Drewnick, F.,
1130 Borrmann, S., Weimer, S., Demerjian, K., Salcedo, D., Cottrell, L., Griffin, R., Takami, A.,
1131 Miyoshi, T., Hatakeyama, S., Shimono, A., Sun, J. Y., Zhang, Y. M., Dzepina, K., Kimmel,
1132 J. R., Sueper, D., Jayne, J. T., Herndon, S. C., Trimborn, a. M., Williams, L. R., Wood, E.
1133 C., Middlebrook, A. M., Kolb, C. E., Baltensperger, U., Worsnop, D. R., Dunlea, E. J., et
1134 al.: Evolution of Organic Aerosols in the Atmosphere, *Science* 80., 326(5959), 1525–1529,
1135 doi:10.1126/science.1180353, 2009.
- 1136 Jolleys, M. D., Coe, H., McFiggans, G., Taylor, J. W., O’Shea, S. J., Le Breton, M.,
1137 Bauguitte, S. J.-B., Moller, S., Di Carlo, P., Aruffo, E., Palmer, P. I., Lee, J. D., Percival,
1138 C. J. and Gallagher, M. W.: Properties and evolution of biomass burning organic aerosol



- 1139 from Canadian boreal forest fires, *Atmos. Chem. Phys.*, 15(6), 3077–3095,
1140 doi:10.5194/acp-15-3077-2015, 2015.
- 1141 Kaiser, J. W., Heil, A., Andreae, M. O., Benedetti, A., Chubarova, N., Jones, L., Morcrette,
1142 J.-J., Razinger, M., Schultz, M. G., Suttie, M., and van der Werf, G. R.: Biomass burning
1143 emissions estimated with a global fire assimilation system based on observed fire radiative
1144 power, *Biogeosciences*, 9, 527–554, <https://doi.org/10.5194/bg-9-527-2012>,
1145 <https://www.biogeosciences.net/9/527/2012/>, 2012.
- 1146 Kanakidou, M., Seinfeld, J. H., Pandis, S. N., Barnes, I., Dentener, F. J., Facchini, M. C.,
1147 Van Dingenen, R., Ervens, B., Nenes, A., Nielsen, C. J., Swietlicki, E., Putaud, J. P.,
1148 Balkanski, Y., Fuzzi, S., Horth, J., Moortgat, G. K., Winterhalter, R., Myhre, C. E. L.,
1149 Tsigaridis, K., Vignati, E., Stephanou, E. G., and Wilson, J.: Organic aerosol and global
1150 climate modelling: a review, *Atmos. Chem. Phys.*, 5, 1053–1123,
1151 <https://doi.org/10.5194/acp-5-1053-2005>, 2005.
- 1152 Katich, J. M., Samset, B. H., Paul Bui, T., Dollner, M., Froyd, K.D., Campuzano-Jost, P.,
1153 Nault, B. A., Schroder, J. C., Weinzierl, B., Schwarz, J. P.: Strong Contrast in Remote
1154 Black Carbon Aerosol Loadings Between the Atlantic and Pacific Basins, *J. Geophys. Res.*
1155 *Atmos.*, 123 (23), pages 13,386–13,395, <https://doi.org/10.1029/2018JD029206>, 2018.
- 1156 Kim, M. J., G. A. Novak, M. C. Zuerb, M. Yang, B. W. Blomquist, B. J. Huebert, C. D.
1157 Cappa, and T. H. Bertram: Air-Sea exchange of biogenic volatile organic compounds and
1158 the impact on aerosol particle size distributions, *Geophys. Res. Lett.*, 44, 3887–3896,
1159 doi:10.1002/2017GL072975, 2017.
- 1160 Klimont, Z., K. Kupiainen, C. Heyes, P. Purohit, J. Cofala, P. Rafaj, J. Borken-Kleefeld,
1161 and W. Schöpp.: Global anthropogenic emissions of particulate matter including black
1162 carbon, *Atmos. Chem. Phys.*, 17(14), 8681–8723, doi: 10.5194/acp-17-8681-2017, 2017.
- 1163 Kodros, J. K., Cucinotta, R., Ridley, D. A., Wiedinmyer, C., and Pierce, J. R.: The aerosol
1164 radiative effects of uncontrolled combustion of domestic waste, *Atmos. Chem. Phys.*, 16,
1165 6771–6784, <https://doi.org/10.5194/acp-16-6771-2016>, 2016.
- 1166 Krechmer, J.E., D. Pagonis, P.J. Ziemann, and J.L. Jimenez. Quantification of gas-wall
1167 partitioning in Teflon environmental chambers using rapid bursts of low-volatility oxidized
1168 species generated in-situ. *Environmental Science and Technology*, 50, 5757–5765,
1169 doi:10.1021/acs.est.6b00606, 2016.



- 1170 Kroll, J. H., Donahue, N. M., Jimenez, J. L., Kessler, S. H., Canagaratna, M. R., Wilson,
1171 K. R., Altieri, K. E., Mazzoleni, L. R., Wozniak, A. S., Bluhm, H., Mysak, E. R., Smith, J.
1172 D., Kolb, C. E. and Worsnop, D. R.: Carbon oxidation state as a metric for describing the
1173 chemistry of atmospheric organic aerosol., *Nat. Chem.*, 3(2), 133–9,
1174 doi:10.1038/nchem.948, 2011.
- 1175 Laskin, A., Laskin, J., and Nizkorodov, S.: Chemistry of Atmospheric Brown Carbon,
1176 *Chem. Rev.* 2015, 115, 10, 4335-4382, 2015.
- 1177 Lee, A. K. Y., Willis, M. D., Healy, R. M., Onasch, T. B., and Abbatt, J. P. D.: Mixing state
1178 of carbonaceous aerosol in an urban environment: single particle characterization using
1179 the soot particle aerosol mass spectrometer (SP-AMS), *Atmos. Chem. Phys.*, 15, 1823-
1180 1841, <https://doi.org/10.5194/acp-15-1823-2015>, 2015.
- 1181 Liu, H. Y., D. J. Jacob, I. Bey, and R. M. Yantosca: Constraints from Pb-210 and Be-7 on
1182 wet deposition and transport in a global three-dimensional chemical tracer model driven
1183 by assimilated meteorological fields, *J. Geophys. Res.*, 106(D11), 12,109–12,128, 2001.
- 1184 Liu, X., Easter, R. C., Ghan, S. J., Zaveri, R., Rasch, P., Shi, X., Lamarque, J.-F.,
1185 Gettelman, A., Morrison, H., Vitt, F., Conley, A., Park, S., Neale, R., Hannay, C., Ekman,
1186 A. M. L., Hess, P., Mahowald, N., Collins, W., Iacono, M. J., Bretherton, C. S., Flanner, M.
1187 G., and Mitchell, D.: Toward a minimal representation of aerosols in climate models:
1188 description and evaluation in the Community Atmosphere Model CAM5, *Geosci. Model*
1189 *Dev.*, 5, 709-739, <https://doi.org/10.5194/gmd-5-709-2012>, 2012.
- 1190 Marais, E. A., Jacob, D. J., Jimenez, J. L., Campuzano-Jost, P., Day, D. A., Hu, W.,
1191 Krechmer, J., Zhu, L., Kim, P. S., Miller, C. C., Fisher, J. A., Travis, K., Yu, K., Hanisco, T.
1192 F., Wolfe, G. M., Arkinson, H. L., Pye, H. O. T., Froyd, K. D., Liao, J., and McNeill, V. F.:
1193 Aqueous-phase mechanism for secondary organic aerosol formation from isoprene:
1194 application to the southeast United States and co-benefit of SO₂ emission controls,
1195 *Atmos. Chem. Phys.*, 16, 1603-1618, <https://doi.org/10.5194/acp-16-1603-2016>, 2016.
- 1196 Myhre, G., Samset, B. H., Schulz, M., Balkanski, Y., Bauer, S., Bernsten, T. K., Bian, H.,
1197 Bellouin, N., Chin, M., Diehl, T., Easter, R. C., Feichter, J., Ghan, S. J., Hauglustaine, D.,
1198 Iversen, T., Kinne, S., Kirkevåg, A., Lamarque, J.-F., Lin, G., Liu, X., Lund, M. T., Luo, G.,
1199 Ma, X., van Noije, T., Penner, J. E., Rasch, P. J., Ruiz, A., Seland, Ø., Skeie, R. B., Stier,
1200 P., Takemura, T., Tsigaridis, K., Wang, P., Wang, Z., Xu, L., Yu, H., Yu, F., Yoon, J.-H.,
1201 Zhang, K., Zhang, H., and Zhou, C.: Radiative forcing of the direct aerosol effect from



- 1202 AeroCom Phase II simulations, *Atmos. Chem. Phys.*, 13, 1853-1877,
1203 <https://doi.org/10.5194/acp-13-1853-2013>, 2013.
- 1204 Mauderly, J. L., & Chow, J. C: Health Effects of Organic Aerosols, *Inhalation Toxicology*,
1205 20:3, 257-288, DOI: 10.1080/08958370701866008, 2008.
- 1206 Middlebrook, A. M., Murphy, D. M. and Thomson, D. S.: Observations of organic material
1207 in individual marine particles at Cape Grim during the First Aerosol Characterization
1208 Experiment (ACE 1), *J. Geophys. Res. Atmos.*, 103(D13), 16475–16483,
1209 doi:10.1029/97JD03719, 1998.
- 1210 Murphy, D. M., Froyd, K. D., Bian, H., Brock, C. A., Dibb, J. E., DiGangi, J. P., Diskin, G.,
1211 Dollner, M., Kupc, A., Scheuer, E. M., Schill, G. P., Weinzierl, B., Williamson, C. J., and
1212 Yu, P.: The distribution of sea-salt aerosol in the global troposphere, *Atmos. Chem. Phys.*,
1213 19, 4093-4104, <https://doi.org/10.5194/acp-19-4093-2019>, 2019.
- 1214 Nault, B. A., Campuzano-Jost, P., Day, D. A., Schroder, J. C., Anderson, B., Beyersdorf,
1215 A. J., Blake, D. R., Brune, W. H., Choi, Y., Corr, C. A., de Gouw, J. A., Dibb, J., DiGangi,
1216 J. P., Diskin, G. S., Fried, A., Huey, L. G., Kim, M. J., Knote, C. J., Lamb, K. D., Lee, T.,
1217 Park, T., Pusede, S. E., Scheuer, E., Thornhill, K. L., Woo, J.-H., and Jimenez, J. L.:
1218 Secondary organic aerosol production from local emissions dominates the organic aerosol
1219 budget over Seoul, South Korea, during KORUS-AQ, *Atmos. Chem. Phys.*, 18, 17769-
1220 17800, <https://doi.org/10.5194/acp-18-17769-2018>, 2018.
- 1221 Ovadnevaite, J., Ceburnis, D., Canagaratna, M., Berresheim, H., Bialek, J., Martucci, G.,
1222 Worsnop, D. R. and O'Dowd, C.: On the effect of wind speed on submicron sea salt mass
1223 concentrations and source fluxes, *J. Geophys. Res. Atmos.*, 117(D16), 1–11,
1224 doi:10.1029/2011JD017379, 2012.
- 1225 Petters, M. D., Prenni, A. J., Kreidenweis, S. M., DeMott, P. J., Matsunaga, A., Lim, Y. B.,
1226 and Ziemann, P. J. Chemical aging and the hydrophobic-to-hydrophilic conversion of
1227 carbonaceous aerosol, *Geophys. Res. Lett.*, 33, L24806, doi:10.1029/2006GL027249,
1228 2006.
- 1229 Palm, B. B., de Sá, S. S., Day, D. A., Campuzano-Jost, P., Hu, W., Seco, R., Sjostedt, S.
1230 J., Park, J.-H., Guenther, A. B., Kim, S., Brito, J., Wurm, F., Artaxo, P., Thalman, R., Wang,
1231 J., Yee, L. D., Wernis, R., Isaacman-VanWertz, G., Goldstein, A. H., Liu, Y., Springston,
1232 S. R., Souza, R., Newburn, M. K., Alexander, M. L., Martin, S. T., and Jimenez, J. L.:
1233 Secondary organic aerosol formation from ambient air in an oxidation flow reactor in



- 1234 central Amazonia, *Atmos. Chem. Phys.*, 18, 467–493, [https://doi.org/10.5194/acp-18-467-](https://doi.org/10.5194/acp-18-467-1235)
1235 2018, 2018.
- 1236 Pye, H. O. T., Chan, A. W. H., Barkley, M. P., and Seinfeld, J. H.: Global modeling of
1237 organic aerosol: The importance of reactive nitrogen (NO_x and NO₃), *Atmos. Chem. Phys.*,
1238 10, 11,261–11,276, doi:10.5194/acp-10-11261-2010, 2010.
- 1239 Robinson, A. L., Donahue, N. M., Shrivastava, M. K., Weitkamp, E. A., Sage, A. M.,
1240 Grieshop, A. P., Lane, T. E., Pandis, S. N., and Pierce, J. R.: Rethinking organic aerosols:
1241 Semivolatile emissions and photochemical aging, *Science*, 315, 1259–1262, 2007.
- 1242 Schroder, J. C., Campuzano-Jost, P., Day, D. A., Shah, V., Larson, K., Sommers, J. M.,
1243 Sullivan, A. P., Campos, T., Reeves, J. M., Hills, A., Hornbrook, R. S., Blake, N. J.,
1244 Scheuer, E., Guo, H., Fibiger, D. L., McDuffie, E. E., Hayes, P. L., Weber, R. J., Dibb, J.
1245 E., Apel, E. C., Jaeglé, L., Brown, S. S., Thornton, J. A. and Jimenez, J. L.: Sources and
1246 Secondary Production of Organic Aerosols in the Northeastern United States during
1247 WINTER, *J. Geophys. Res. Atmos.*, 123(14), 7771–7796, doi:10.1029/2018JD028475,
1248 2018.
- 1249 Schacht, J., Heinold, B., Quaas, J., Backman, J., Cherian, R., Ehrlich, A., Herber, A.,
1250 Huang, W. T. K., Kondo, Y., Massling, A., Sinha, P.R., Weinzierl, B., Zannata, M., and Tegen,
1251 I.: The importance of the representation of air pollution emissions for the modeled
1252 distribution and radiative effects of black carbon in the Arctic, *Atmos. Chem. Phys.*
1253 *Discuss.*, 2019, 1-39, doi: 10.5194/acp-2019-71, 2019.
- 1254 Schulz, M., Chin, M., and Kinne, S.: The aerosol model comparison project, AeroCom,
1255 phase II: clearing up diversity, *IGAC newsletter*, No. 41, 2–11, 2009.
- 1256 Scott, C. E., Rap, A., Spracklen, D. V., Forster, P. M., Carslaw, K. S., Mann, G. W., Pringle,
1257 K. J., Kivekäs, N., Kulmala, M., Lihavainen, H., and Tunved, P.: The direct and indirect
1258 radiative effects of biogenic secondary organic aerosol, *Atmos. Chem. Phys.*, 14, 447-
1259 470, <https://doi.org/10.5194/acp-14-447-2014>, 2014.
- 1260 Shah, V., Jaeglé, L., Jimenez, J. L., Schroder, J. C., Campuzano-Jost, P., Campos, T. L.,
1261 et al.: Widespread pollution from secondary sources of organic aerosols during winter in
1262 the Northeastern United States. *Geophysical Research Letters*, 46, 2974– 2983.
1263 <https://doi.org/10.1029/2018GL081530>, 2019.



- 1264 Shiraiwa, M., Ueda, K., Pozzer, A., Lammel, G., Kampf, C. J.: Aerosol Health Effects from
1265 Molecular to Global Scales, *Environ. Sci. Technol.* 51 (23), pp 13545–13567, 2017.
- 1266 Shrivastava, M., Cappa, C.D., Fan, J., et al: Recent advances in understanding secondary
1267 organic aerosol: Implications for global climate forcing, *Rev. Geophys.*, 55, 509–559,
1268 doi:10.1002/2016RG000540, 2017.
- 1269 Spracklen, D. V., Jimenez, J. L., Carslaw, K. S., Worsnop, D. R., Evans, M. J., Mann, G.
1270 W., Zhang, Q., Canagaratna, M. R., Allan, J., Coe, H., McFiggans, G., Rap, A., and
1271 Forster, P.: Aerosol mass spectrometer constraint on the global secondary organic aerosol
1272 budget, *Atmos. Chem. Phys.*, 11, 12109–12136, doi:10.5194/acp-11-12109-2011, 2011.
- 1273 Tegen, I., Neubauer, D., Ferrachat, S., Siegenthaler-Le Drian, C., Bey, I., Schutgens, N.,
1274 Stier, P., Watson-Parris, D., Stanelle, T., Schmidt, H., Rast, S., Kokkola, H., Schultz, M.,
1275 Schroeder, S., Daskalakis, N., Barthel, S., Heinold, B., and Lohmann, U.: The global
1276 aerosol–climate model ECHAM6.3–HAM2.3 – Part 1: Aerosol evaluation, *Geosci. Model
1277 Dev.*, 12, 1643–1677, <https://doi.org/10.5194/gmd-12-1643-2019>, 2019.
- 1278 Thomson, D. S., Schein, M. E. and Murphy, D. M.: Particle Analysis by Laser Mass
1279 Spectrometry WB-57F Instrument Overview, *Aerosol Sci. Technol.*, 33(1–2), 153–169,
1280 doi:10.1080/027868200410903, 2000.
- 1281 Tilmes, S., Hodzic, A., Emmons, L.K., Mills, M.J., Gettelman, A., Kinnison, D.E., Park, M.,
1282 Lamarque J.-F., Vitt, F., et al.: Climate forcing and trends of organic aerosols in the
1283 Community Earth System Model (CESM2), to be submitted to JAMES, 2019.
- 1284 Tsigaridis, K., Daskalakis, N., Kanakidou, M., Adams, P. J., Artaxo, P., Bahadur, R.,
1285 Balkanski, Y., Bauer, S. E., Bellouin, N., Benedetti, A., Bergman, T., Berntsen, T. K.,
1286 Beukes, J. P., Bian, H., Carslaw, K. S., Chin, M., Curci, G., Diehl, T., Easter, R. C., Ghan,
1287 S. J., Gong, S. L., Hodzic, A., Hoyle, C. R., Iversen, T., Jathar, S., Jimenez, J. L., Kaiser,
1288 J. W., Kirkevåg, A., Koch, D., Kokkola, H., Lee, Y. H., Lin, G., Liu, X., Luo, G., Ma, X.,
1289 Mann, G. W., Mihalopoulos, N., Morcrette, J.-J., Müller, J.-F., Myhre, G., Myriokefalitakis,
1290 S., Ng, N. L., O'Donnell, D., Penner, J. E., Pozzoli, L., Pringle, K. J., Russell, L. M., Schulz,
1291 M., Sciare, J., Seland, Ø., Shindell, D. T., Sillman, S., Skeie, R. B., Spracklen, D.,
1292 Stavrou, T., Steenrod, S. D., Takemura, T., Tiitta, P., Tilmes, S., Tost, H., van Noije, T.,
1293 van Zyl, P. G., von Salzen, K., Yu, F., Wang, Z., Wang, Z., Zaveri, R. A., Zhang, H., Zhang,
1294 K., Zhang, Q., and Zhang, X.: The AeroCom evaluation and intercomparison of organic



- 1295 aerosol in global models, *Atmos. Chem. Phys.*, 14, 10845–10895, doi:10.5194/acp-14-
1296 10845-2014, 2014.
- 1297 Tsigaridis, K., and Kanakidou, M.: The Present and Future of Secondary Organic Aerosol
1298 Direct Forcing on Climate, *Current Climate Change Reports*, 2018, Volume 4, Issue 2, pp
1299 84–98, 2018.
- 1300 Turpin, B. J. and Lim, H. J.: Species contributions to PM_{2.5} mass concentrations:
1301 Revisiting common assumptions for estimating organic mass, *Aerosol Sci. Tech.*, 35, 602–
1302 610, doi:10.1080/02786820152051454, 2001.
- 1303 Ulbrich, I. M., Canagaratna, M. R., Zhang, Q., Worsnop, D. R. and Jimenez, J. L.:
1304 Interpretation of organic components from Positive Matrix Factorization of aerosol mass
1305 spectrometric data, *Atmos. Chem. Phys.*, 9(9), 2891–2918, doi:10.5194/acp-9-2891-2009,
1306 2009.
- 1307 van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla, P. S.,
1308 Morton, D. C., DeFries, R. S., Jin, Y., and van Leeuwen, T. T.: Global fire emissions and
1309 the contribution of deforestation, savanna, forest, agricultural, and peat fires (1997–2009),
1310 *Atmos. Chem. Phys.*, 10, 11707-11735, <https://doi.org/10.5194/acp-10-11707-2010>,
1311 2010.
- 1312 Vergara-Temprado, J., Murray, B. J., Wilson, T. W., O'Sullivan, D., Browse, J., Pringle, K.
1313 J., Ardon-Dryer, K., Bertram, A. K., Burrows, S. M., Ceburnis, D., DeMott, P. J., Mason,
1314 R. H., O'Dowd, C. D., Rinaldi, M., and Carslaw, K. S.: Contribution of feldspar and marine
1315 organic aerosols to global ice nucleating particle concentrations, *Atmos. Chem. Phys.*, 17,
1316 3637-3658, <https://doi.org/10.5194/acp-17-3637-2017>, 2017.
- 1317 Wang, J., Cubison, M. J., Aiken, A. C., Jimenez, J. L., and Collins, D. R.: The importance
1318 of aerosol mixing state and size-resolved composition on CCN concentration and the
1319 variation of the importance with atmospheric aging of aerosols, *Atmos. Chem. Phys.*, 10,
1320 7267-7283, <https://doi.org/10.5194/acp-10-7267-2010>, 2010.
- 1321 Williamson, C. J., Kupc, A., Axisa, D., Bilsback, K.R., Bui, T., Campuzano-Jost, P.,
1322 Dollner, M., Froyd, K., Hodshire, A. L., Jimenez, J. L., Kodros, J. K., Luo, G., Murphy, D.
1323 M., Nault, B. A., Ray, E. A., Weinzierl, B. B., Wilson, J. C., Yu, F., Yu, P., Pierce, J.F.,
1324 Brock C. A.: A Large Source of Cloud Condensation Nuclei from New Particle Formation
1325 in the Tropics, submitted to *Nature*, 2019.



- 1326 Wofsy, S.C., S. Afshar, H.M. Allen, E. Apel, E.C. Asher, B. Barletta, J. Bent, H. Bian, B.C.
1327 Biggs, D.R. Blake, N. Blake, I. Bourgeois, C.A. Brock, W.H. Brune, J.W. Budney, T.P. Bui,
1328 A. Butler, P. Campuzano-Jost, C.S. Chang, M. Chin, R. Commane, G. Correa, J.D.
1329 Crouse, P. D. Cullis, B.C. Daube, D.A. Day, J.M. Dean-Day, J.E. Dibb, J.P. DiGangi,
1330 G.S. Diskin, M. Dollner, J.W. Elkins, F. Erdesz, A.M. Fiore, C.M. Flynn, K. Froyd, D.W.
1331 Gesler, S.R. Hall, T.F. Hanisco, R.A. Hannun, A.J. Hills, E.J. Hints, A. Hoffman, R.S.
1332 Hornbrook, L.G. Huey, S. Hughes, J.L. Jimenez, B.J. Johnson, J.M. Katich, R.F. Keeling,
1333 M.J. Kim, A. Kupc, L.R. Lait, J.-F. Lamarque, J. Liu, K. McKain, R.J. Mclaughlin, S.
1334 Meinardi, D.O. Miller, S.A. Montzka, F.L. Moore, E.J. Morgan, D.M. Murphy, L.T. Murray,
1335 B.A. Nault, J.A. Neuman, P.A. Newman, J.M. Nicely, X. Pan, W. Paplawsky, J. Peischl,
1336 M.J. Prather, D.J. Price, E. Ray, J.M. Reeves, M. Richardson, A.W. Rollins, K.H. Rosenlof,
1337 T.B. Ryerson, E. Scheuer, G.P. Schill, J.C. Schroder, J.P. Schwarz, J.M. St.Clair, S.D.
1338 Steenrod, B.B. Stephens, S.A. Strode, C. Sweeney, D. Tanner, A.P. Teng, A.B. Thames,
1339 C.R. Thompson, K. Ullmann, P.R. Veres, N. Vieznor, N.L. Wagner, A. Watt, R. Weber, B.
1340 Weinzierl, P. Wennberg, C.J. Williamson, J.C. Wilson, G.M. Wolfe, C.T. Woods, and L.H.
1341 Zeng. 2018. ATom: Merged Atmospheric Chemistry, Trace Gases, and Aerosols. ORNL
1342 DAAC, Oak Ridge, Tennessee, USA. <https://doi.org/10.3334/ORNLDAAC/1581>, 2018.
- 1343 Woody, M. C., Baker, K. R., Hayes, P. L., Jimenez, J. L., Koo, B., and Pye, H. O. T.:
1344 Understanding sources of organic aerosol during CalNex-2010 using the CMAQ-VBS,
1345 Atmos. Chem. Phys., 16, 4081-4100, <https://doi.org/10.5194/acp-16-4081-2016>, 2016.
- 1346 Yu, P., O. B. Toon, C. G. Bardeen, M. J. Mills, T. Fan, J. M. English, and R. R. Neely,
1347 Evaluations of tropospheric aerosol properties simulated by the community earth system
1348 model with a sectional aerosol microphysics scheme, J. Adv. Model. Earth Syst., 7, 865–
1349 914, doi:10.1002/2014MS000421, 2015.
- 1350 Yu, P., Froyd, K. D., Portmann, R. W., Toon, O. B., Freitas, S. R., Bardeen, C. G., et al.:
1351 Efficient in-cloud removal of aerosols by deep convection. Geophysical Research Letters,
1352 46. <https://doi.org/10.1029/2018GL080544>, 2019.
- 1353 Zhang, K., O'Donnell, D., Kazil, J., Stier, P., Kinne, S., Lohmann, U., Ferrachat, S., Croft,
1354 B., Quaas, J., Wan, H., Rast, S., and Feichter, J.: The global aerosol-climate model
1355 ECHAM-HAM, version 2: sensitivity to improvements in process representations, Atmos.
1356 Chem. Phys., 12, 8911-8949, <https://doi.org/10.5194/acp-12-8911-2012>, 2012.



- 1357 Zhang, Q. J., Beekmann, M., Drewnick, F., Freutel, F., Schneider, J., Crippa, M., Prevot,
1358 A. S. H., Baltensperger, U., Poulain, L., Wiedensohler, A., Sciare, J., Gros, V., Borbon, A.,
1359 Colomb, A., Michoud, V., Doussin, J.-F., Denier van der Gon, H. A. C., Haeffelin, M.,
1360 Dupont, J.-C., Siour, G., Petetin, H., Bessagnet, B., Pandis, S. N., Hodzic, A., Sanchez,
1361 O., Honoré, C., and Perrussel, O., 2013. Formation of organic aerosol in the Paris region
1362 during the MEGAPOLI summer campaign: evaluation of the volatility-basis-set approach
1363 within the CHIMERE model, *Atmos. Chem. Phys.*, 13, 5767-5790, doi:10.5194/acp-13-
1364 5767-2013.
- 1365 Zhang, X., Cappa, C. D., Jathar, S. H., McVay, R. C., Ensberg, J. J., Kleeman, M. J., and
1366 Seinfeld, J. H.: Influence of vapor wall loss in laboratory chambers on yields of secondary
1367 organic aerosol, *P. Natl. Acad. Sci. USA*, 111, 5802–5807, 2014.
- 1368 Zotter, P., I. El-Haddad, Y. Zhang, P.L. Hayes, X. Zhang, Y.H. Lin, L. Wacker, J. Schnelle-
1369 Kreis, G. Abbaszade, R. Zimmermann, J.D. Surratt, R. Weber, J.L. Jimenez, S. Szidat, U.
1370 Baltensperger, A.S.H. Prévôt. Diurnal cycle of fossil and non-fossil carbon using
1371 radiocarbon analyses during CalNex. *Journal of Geophysical Research-Atmospheres*,
1372 119, 6818–6835, doi:10.1002/2013JD021114, 2014.
- 1373



1374 Tables

1375 Table 1: ATom global model configurations and their treatment of the most important processes affecting organic aerosols.

Models & horizontal res. & config. reference	Aerosol module	Submicron size ⁽⁶⁾ OA (dust/seasalt)	SOA precursors ⁽¹⁾				SOA production	Emission	Removal			
			ISO	MT	SQ	ANT			C ₁₂	Standard ⁽²⁾	Improved	Photolytic
CESM1-CARMA (1.9° lon x 2.5° lat) (Yu et al. 2019)	20 bins	< 500 nm (< 800 nm)	x	x		x	Semi-volatile using VBS (Pye et al. 2010)	GAIS and GFED v3	x	For convective updrafts (Yu et al. 2019) ⁽³⁾		
CESM2-DYN (0.9° lon x 1.25° lat) (Times et al. 2019)	4 modes	< 270 nm (< 800 nm)	x	x	x	x	Semi-volatile using VBS (Hodzic et al. 2016)	CMIP6 and QFED v2.4	x	Water solubility of organic gases per Hodzic et al. (2014)		For SOA (Hodzic et al. 2016)
CESM2-SMP (0.9° lon x 1.25° lat) (Times et al. 2019)	4 modes	< 270 nm (< 800 nm)	x	x		x	Non-volatile with prescribed mass yields for all precursors ⁽⁴⁾	CMIP6 and QFED v2.4	x			
ECHAM6-HAM (1.87°lon x 1.87°lat) (Tegen et al. 2019)	7 modes	< 500 nm (< 500 nm)		x			Non-volatile with 15% prescribed mass yields (Dentener et al. 2006)	ECLIPSE ⁽⁵⁾ and GFAS	x			
GC12-REF (2° lon x 2.5° lat) (Bey et al. 2001)	Bulk	Bulk (< 500 nm)	x	x	x		Semi-volatile using VBS (Pye et al. 2010); non-volatile isoprene-SOA	CMIP6 and GFED v4	x	For convective updrafts per Wang et al. 2014		



GC12-DYN (2° lon x 2.5° lat) (Bey et al. 2001)	Bulk	Bulk (< 500 nm)	x	x	x	x	(Marais et al. 2016)	Semi-volatile using VBS (Hodzic et al. 2016); non- volatile isoprene-SOA (Marais et al. 2016)	CMIP6 and GFED v4	x	For convective updrafts (Wang et al. 2014); Water solubility of organic gases (Hodzic et al. 2014)	For SOA (Hodzic et al. 2016)
GC10-TOMAS (5° lon x 4° lat) (Kodros et al. 2016)	15 bins	< 316 nm (< 316 nm)	x	x	x		Non-Volatile using 10% mass yields for MT, 0.2 Tg SOA per Tg CO for anthropogenic emissions	EDGAR v4 and GFED v3	x	For convective updrafts (Wang et al. 2014)		
GEOS5- GOCART (0.5° lon x 0.625°lat) (Bian et al. 2019)	Bulk	bulk (< 730 nm for dust, 316 nm for seasalt)	x	x	x		Non-Volatile, 10% mass yields for all precursors	HTAP and QFED v2.54	x			

- 1376 (1) SOA precursors include isoprene (ISO), monoterpenes (MT), sesquiterpenes (SQ), anthropogenics (ANT) including aromatics such as
 1377 benzene, toluene and xylene, as well as lumped shorter chain alkanes and alkenes; and higher molecular weight n-alkanes and n-alkenes
 1378 (C>12).
 1379 (2) Standard removal includes dry deposition and sedimentation, as well as convective and large-scale scavenging of soluble organic gases
 1380 and aerosols, and below-cloud scavenging of aerosols.
 1381 (3) A sensitivity simulation is performed with CESM1-CARMA without the improved scavenging in convective updrafts.
 1382 (4) 5% for lumped C<12 alkanes, 5% for lumped C<12 alkenes, 15% for aromatics, 4% for isoprene, 25% for monoterpenes.
 1383 (5) Anthropogenic BC emission are replaced in Russia with the dataset of Huang et al. (2015).
 1384 (6) Submicron size range (diameter) used in various models for comparison with the AMS data.



1385 Table 2: Comparison of observed and simulated OA concentrations along ATom-1 and
 1386 ATom-2 flights for eight global model simulations and their ensemble. The results of the
 1387 model ensemble are also indicated. The statistical indicators are calculated as normalized
 1388 mean bias $NMB(\%) = 100 \times \sum_i (M_i - O_i) / \sum_i O_i$; normalized mean error $NME(\%) =$
 1389 $100 \times \sum_i |M_i - O_i| / \sum_i O_i$; root mean square error $RMSE(\mu g m^{-3}) =$
 1390 $\sqrt{(1/N) \sum_i (M_i - O_i)^2}$ and correlation coefficient (R^2) between modeled (M_i) and observed
 1391 (O_i) data points. The mean of ATom-1 observations is $\sim 0.23 \mu g m^{-3}$ and for ATom-2 is
 1392 $0.11 \mu g m^{-3}$. Figure S4 shows the normalized mean bias for all individual ATom model
 1393 simulations for various latitudinal regions and for both the Atlantic and Pacific basins.

Organic aerosols	Avg.Mod. ($\mu g m^{-3}$)	NMB (%)	NME (%)	RMSE ($\mu g m^{-3}$)	R^2	Avg.Mod. ($\mu g m^{-3}$)	NMB (%)	NME (%)	RMSE ($\mu g m^{-3}$)	R^2
Model	<i>ATom-1 scores (August 2016)</i>					<i>ATom-2 scores (February 2017)</i>				
AeroCom-II Ens.	0.400	74.2	127.3	0.560	0.31	0.254	137	175	0.278	0.38
AeroCom-II Sub. ⁽¹⁾	0.335	47.0	111	0.557	0.28	0.242	127	178	0.290	0.27
ATom Ensemble	0.239	-4.5	64.6	0.372	0.66	0.139	23	92.6	0.224	0.48
CESM2-DYN	0.268	4.6	83.7	0.867	0.47	0.140	25.6	111.7	0.317	0.36
CESM2-SMP	0.349	36.3	94.3	0.556	0.51	0.175	57.2	125.4	0.299	0.31
CESM1-CARMA	0.155	-33.2	93.8	0.603	0.12	0.131	22.6	119.6	0.244	0.31
ECHAM6-HAM	0.400	73.6	143.6	0.714	0.24	0.214	100	184.0	0.363	0.23
GC12-DYN	0.142	-32.6	79.4	0.560	0.16	0.174	14.7	96.6	0.312	0.39
GC12-REF	0.122	-43.0	76.5	0.536	0.18	0.147	3.6	96.3	0.292	0.35
GC10-TOMAS	0.218	-14.4	86.5	0.644	0.16	0.313	150.0	223.7	0.537	0.12
GEOS5-GOCART	0.242	-5.4	86.6	0.975	0.38	0.084	-24.9	86.4	0.268	0.29

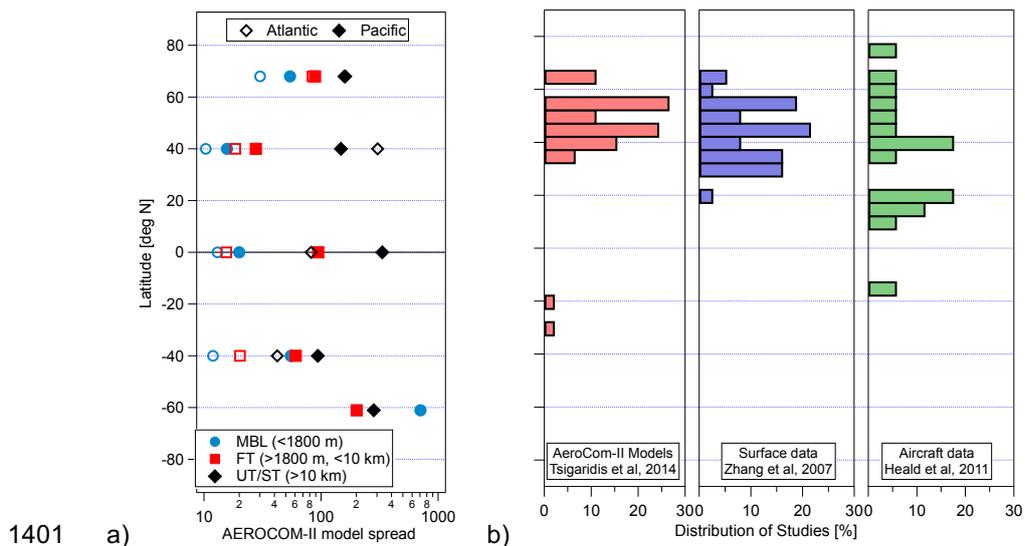
1394 (1) This is the subset of AeroCom-II model ensemble that includes only seven
 1395 models that are similar to those that are included in the ATom ensemble (either
 1396 the same model, or an older model version, or the same aerosol module).
 1397 AeroCom-II Sub. includes CAM5-MAM3, CCSM4-hem, ECHAM5-HAM2,



1398 GEOSChem-APM 8.2, GEOSChem 9, GISS-TOMAS and GMI (see Tsigaridis
1399 et al., 2014 for their description).



1400 Figures:



1401 a) b)
1402 Figure 1: (a, left) The ratio between the highest and the lowest OA concentrations as
1403 predicted among 28 global chemistry transport models participating in the AeroCom phase
1404 II intercomparison study (Tsigaridis et al. 2014); (b, right) Percent distribution of AeroCom-
1405 II models (author affiliations) and of measurements from two literature overview studies
1406 (Zhang et al., 2007; Heald et al., 2011) for the OA ground and aircraft AMS as a function
1407 of latitude. For the aircraft campaigns, the average latitude for the full deployment was
1408 taken.

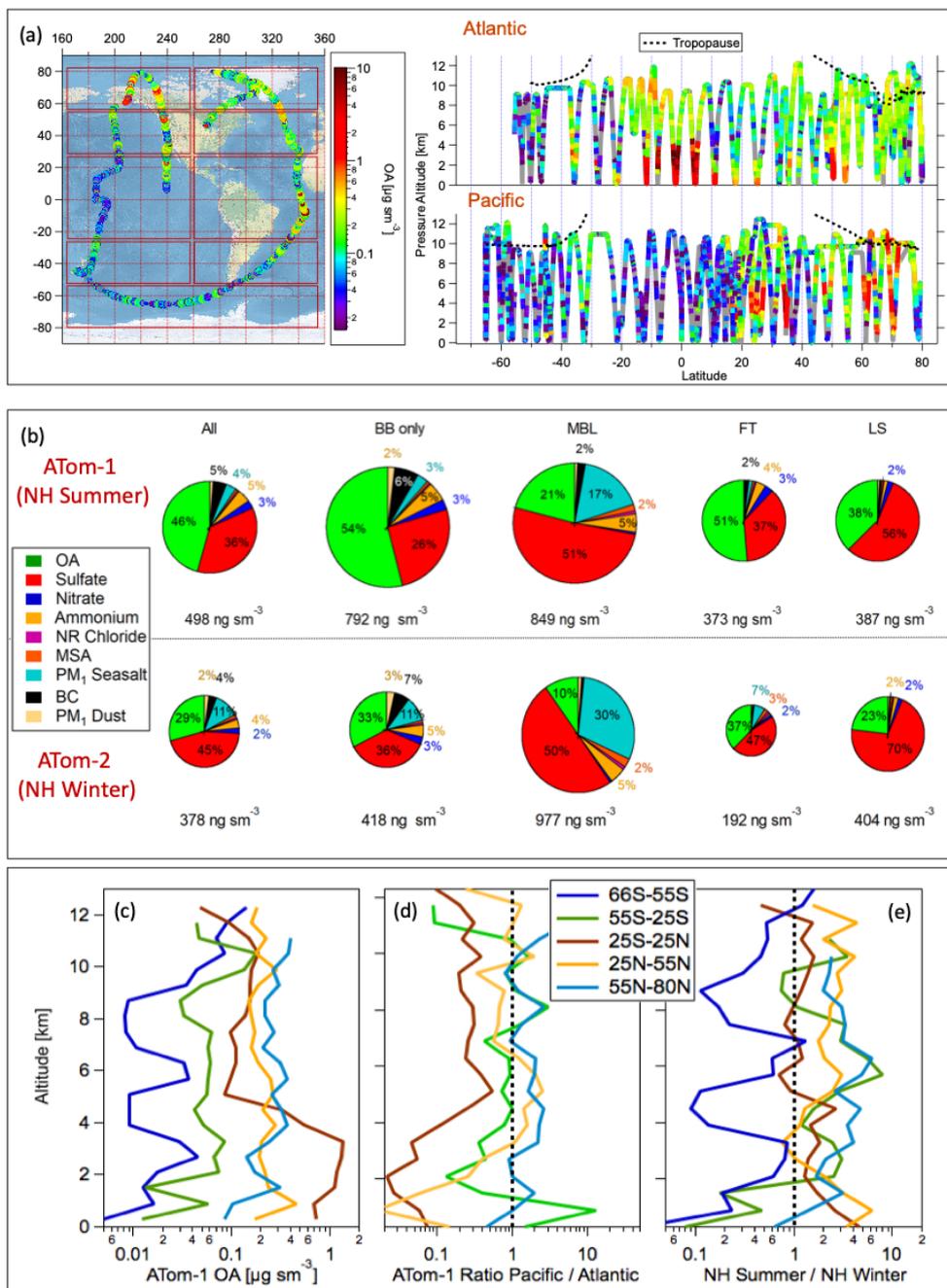
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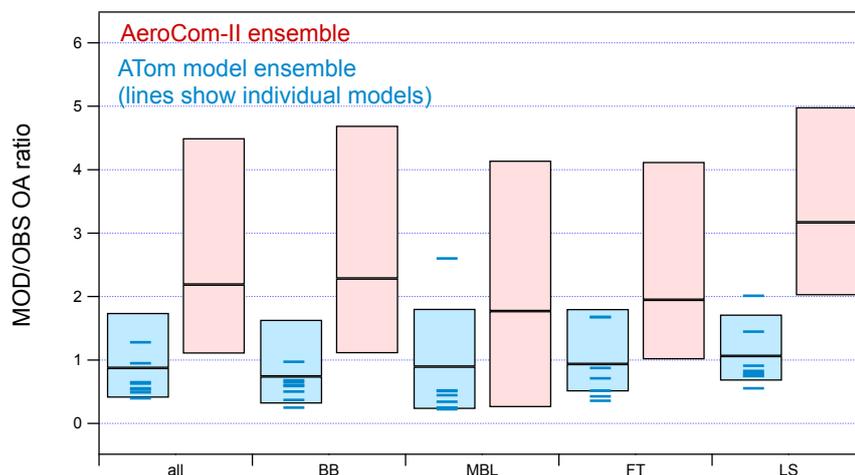
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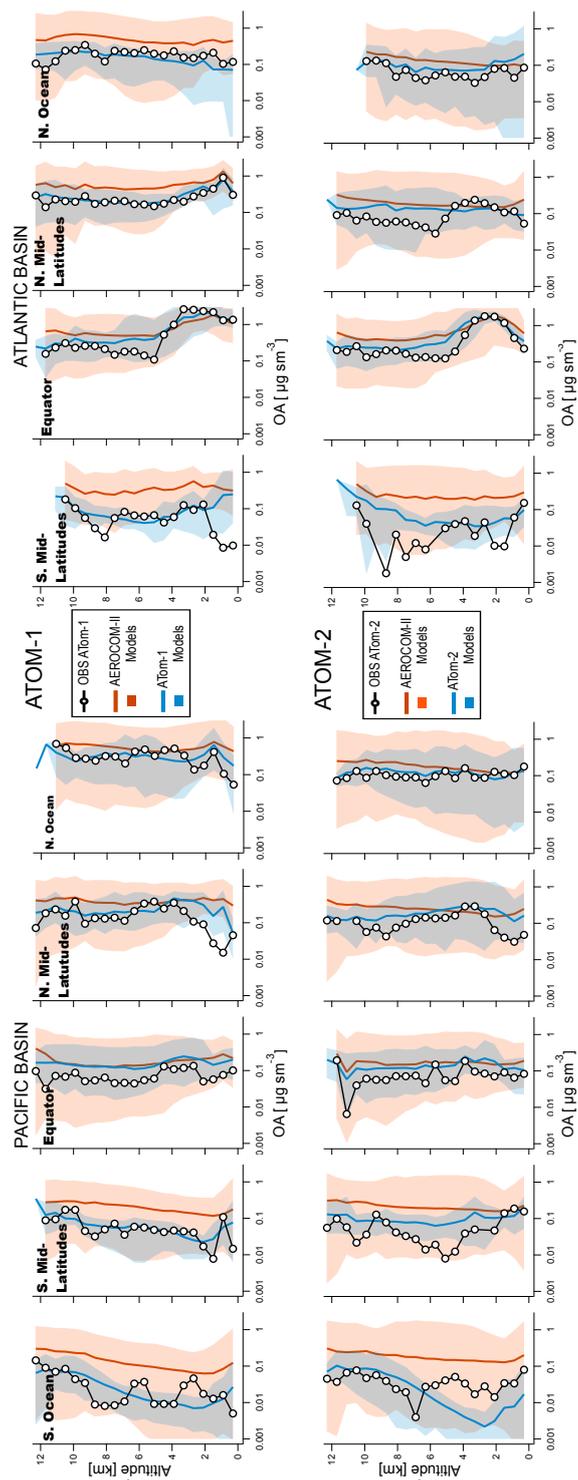


1415 Figure 2: (a, left) ATom-1 DC-8 flights during the August 2016 deployment. Red boxes
1416 indicate regions used for the latitude averaging of the model results. (a, right) Vertical
1417 distribution of OA concentrations ($\mu\text{g sm}^{-3}$) along ATom-1 flight tracks (b) Average
1418 submicron aerosol composition as measured in the biomass-burning influenced regions
1419 (BB only), and the non-BB influenced regions including the marine boundary layer (MBL),
1420 free troposphere (FT), and lower stratosphere (LS) for ATom-1 (upper plots) and ATom-2
1421 (lower plots). The BB influenced airmasses were filtered using the PALMS data (see
1422 section 3.1). Contributions below 2% are shown but not labeled on the pie chart graph. (c)
1423 The average OA vertical profiles are shown for each latitude region as well as (d) the ratios
1424 between the Pacific and Atlantic Oceans in each region. (e) The seasonal contrast in OA
1425 concentrations as calculated as the ratio in OA concentrations between the NH summer
1426 (ATom-1) and NH winter (ATom-2) campaigns. The corresponding plots for ATom-2 can
1427 be found in Fig. S1.



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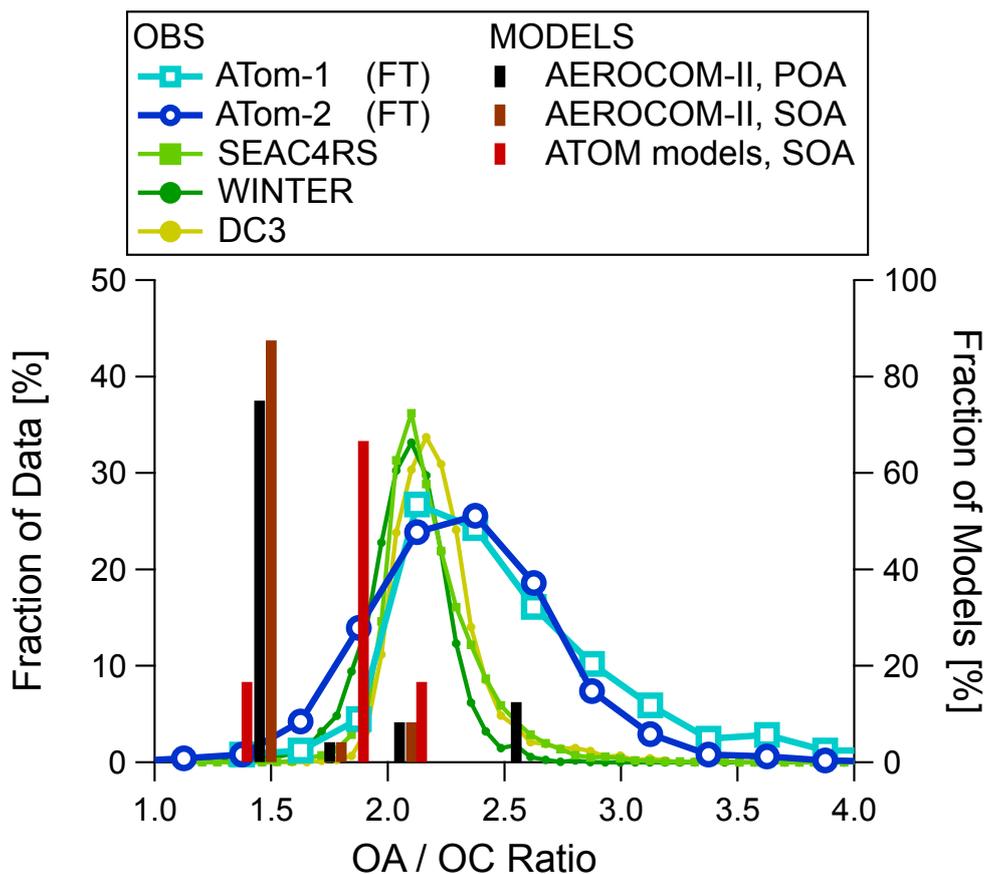
1429 Figure 3: Ratios between predicted and observed OA concentrations for all ATom-1 flights
1430 as calculated for the ATom and AeroCom-II model ensembles in different regions (“BB”
1431 biomass burning influenced regions; “MBL” clean marine boundary layer; “FT” clean free
1432 troposphere’ and “LS” lower stratosphere). Median of the ensemble ratio is shown as a
1433 horizontal line, while the boxes indicate 25th and 75th percentiles. Medians for the
1434 individual models included in the current ATom model ensemble are also shown as blue
1435 lines.



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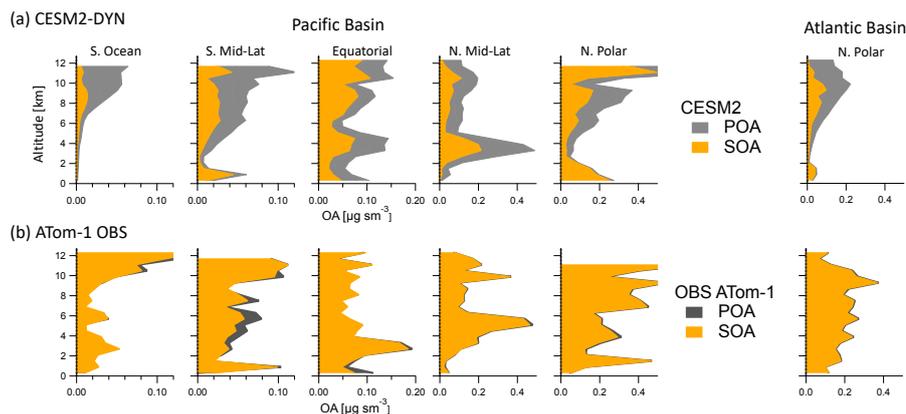
1437

1438 Figure 4: Comparison of latitude-averaged predicted OA vertical profiles with ATom-1 and -2 measurements taken over the Pacific (left
 1439 side) and Atlantic (right side) basins. Results of the AeroCom-II model ensemble average are shown in red while those of the ATom
 1440 model ensemble are shown in blue. Shaded areas indicate the variability (two standard deviations) within each model ensemble.



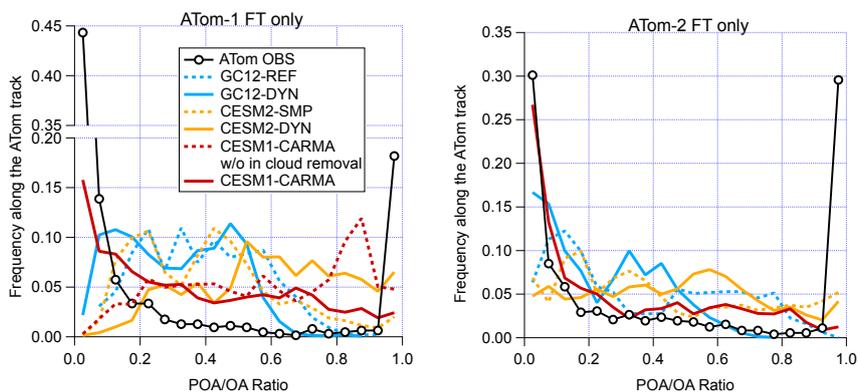
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1442 Figure 5: Distribution of the OA / OC ratio as measured during ATom-1 and -2. Values for
1443 the recent aircraft campaigns (SEAC4RS, DC3 and WINTER) that took place over
1444 continental US regions closer to continental source regions are also shown (Schroder et
1445 al., 2018). The bars (right axis) show the OA/OC used for SOA and POA by the models
1446 included in the AeroCom and ATom ensemble, with OA/OC=1.4 being the modal value for
1447 the former and 1.8 for the latter.



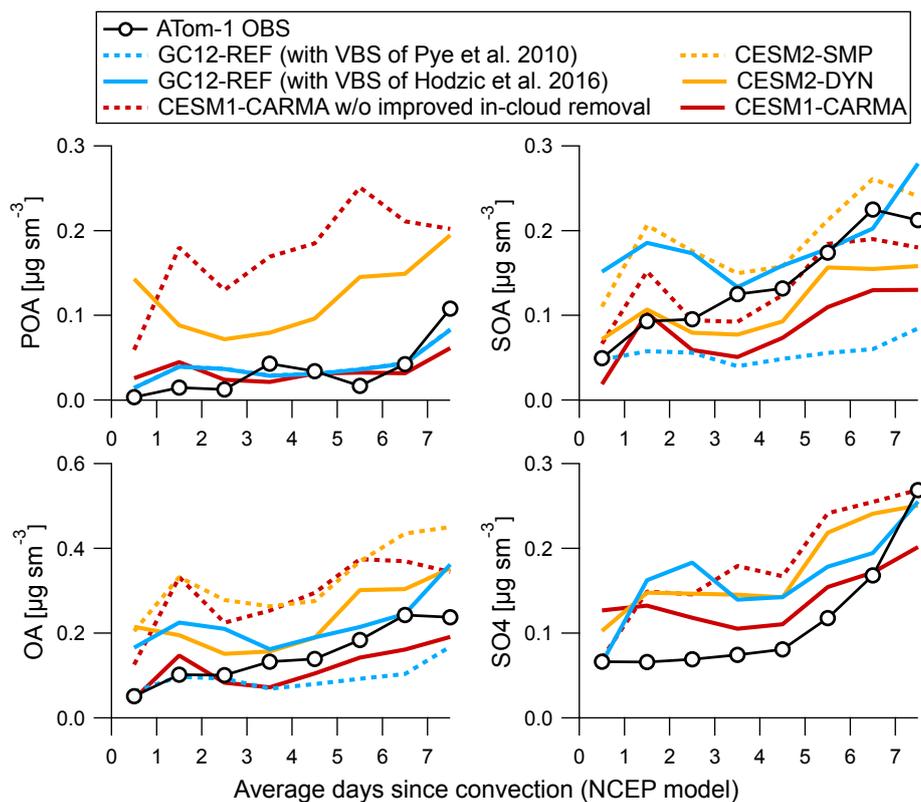
1448

1449 Figure 6: Comparison of averaged POA and SOA vertical profiles as observed during
 1450 ATom and as predicted by the CESM2-DYN model over the non-BB influenced Pacific
 1451 and Atlantic basins. The comparison is not shown for the strongly biomass burning
 1452 influenced regions as all the OA is conservatively allocated to POA in those regions.



1453

1454 Figure 7: Frequency distribution of observed and simulated ratio of POA to total OA in the
 1455 free troposphere during ATom-1 and ATom-2 as computed by the GCM12-, CEM2-, and
 1456 CEM1-CARMA models.



1457

1458 Figure 8: Measured and predicted mass concentrations of POA, SOA, OA and sulfate
1459 aerosols during ATom-1 as a function of the number of days since the air mass was
1460 processed through convection (based on a trajectory model from Bowman, 1993, and
1461 satellite cloud data from NASA Langley, <https://clouds.larc.nasa.gov/>). CESM2-SMP and
1462 CESM2-DYN have the same emissions and processing of POA and sulfate, and thus
1463 similar concentrations. The same is true for the two versions of GC12.

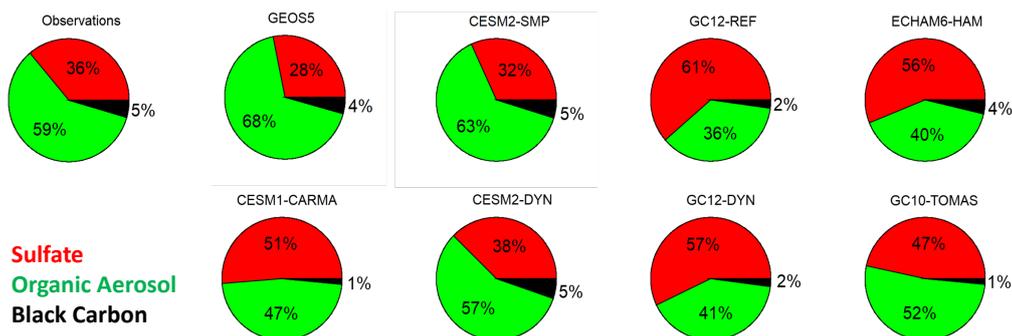
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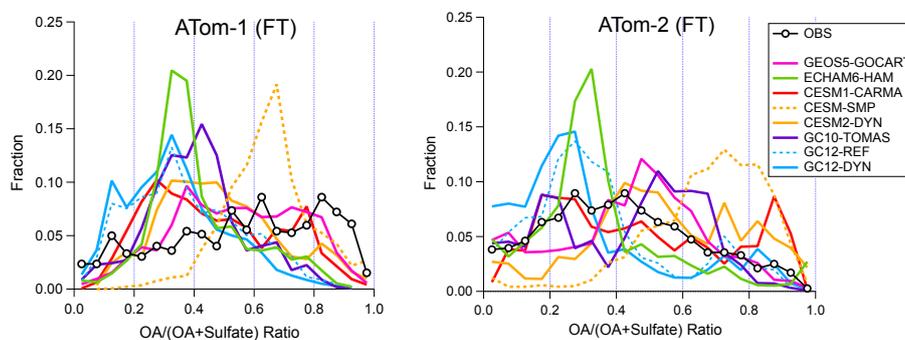


1467 (a)



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1469 (b)

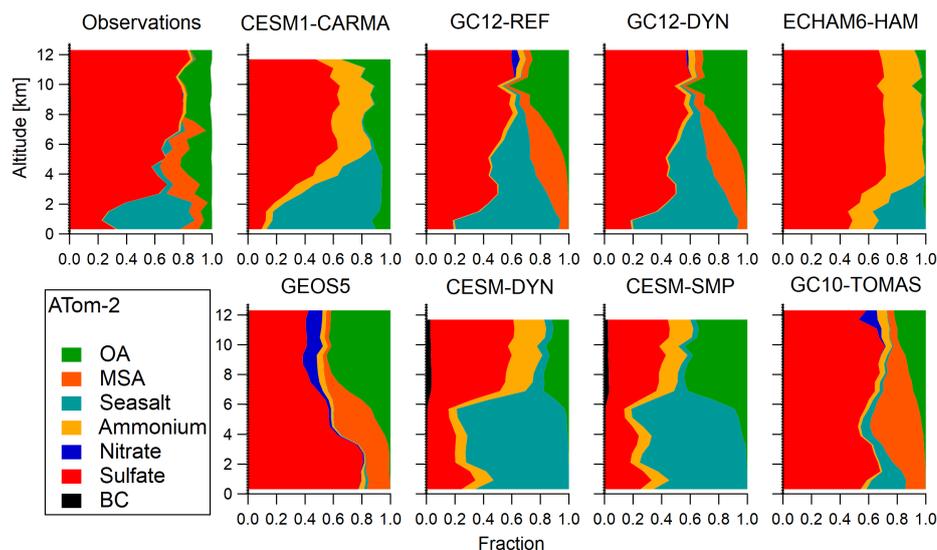


1470

1471 Figure 9: (a) Predicted and measured composition of submicron aerosols in the free
 1472 troposphere as a function of the submicron aerosol mass concentrations during ATom-1.
 1473 (b) Frequency distribution of observed and simulated ratio of organic to organic plus
 1474 sulfate aerosols in the free troposphere during ATom-1 and -2.

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1478 Figure 10: Comparison of measured and predicted composition of submicron aerosols as
1479 a function of altitude over the remote Southern Ocean region during NH Winter (ATom-2).
1480 For models that do not calculate ammonium in the aerosol (such as CESM1-CARMA,
1481 CESM2-SMP, CESM2-DYN and ECHAM6-HAM), ammonium was estimated from the
1482 sulfate mass assuming the formation of ammonium sulfate. Note that based on the
1483 PALMS measurements there is a small contribution of dust (~2% in the LS, negligible
1484 elsewhere) to the AMS size range, but since the accumulation mode dust in the models
1485 presented contains much larger sizes, it is omitted here to provide a fair comparison.