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A Machine Learning Examination of Hydroxyl Radical Differences Among Model Simulations for CCMI-1

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Abstract. Hydroxyl radical (OH) plays critical roles within the troposphere, such as determining the lifetime of methane (CH₄), yet is challenging to model due to its fast cycling and dependence on a multitude of sources and sinks. As a result, the reasons for variations in OH and the resulting CH₄ lifetime (τ_{CH_4}), both between models and in time, are difficult to

40 diagnose. We apply a neural network (NN) approach to address this issue within a group of models that participated in the Chemistry-Climate Model Initiative (CCMI). Analysis of the historical specified dynamics simulations performed for CCMI





indicates that the primary drivers of τ_{CH_4} differences among ten models are the flux of UV light to the troposphere (indicated by the photolysis frequency JO¹D) due mostly to clouds, mixing ratio of tropospheric ozone (O₃), the abundance of nitrogen oxides (NO_x=NO+NO₂), and details of the various chemical mechanisms that drive OH. Water vapor, carbon monoxide

- 45 (CO), the ratio of NO:NO_x, and formaldehyde (HCHO) explain moderate differences in τ_{CH_4} , while isoprene, CH₄, the photolysis frequency of NO₂ by visible light (JNO₂), overhead O₃ column, and temperature account for little-to-no model variation in τ_{CH_4} . We also apply the NNs to analysis of temporal trends in OH from 1980 to 2015. All models that participated in the specified dynamics historical simulation for CCMI demonstrate a decline in τ_{CH_4} during the analysed timeframe. The significant contributors to this trend, in order of importance, are tropospheric O₃, JO¹D, NO_x, and H₂O, with
- 50 CO also causing substantial interannual variability in OH burden. Finally, the identified trends in τ_{CH_4} are compared to calculated trends in the tropospheric mean OH concentration from previous work, based on analysis of observations. The comparison reveals a robust result for the effect of rising water vapor on OH and τ_{CH_4} , imparting an increasing and decreasing trend of about 0.5 % decade⁻¹, respectively. The responses due to NO_x, O₃ column, and temperature are also in reasonably good agreement between the two studies, though a discrepancy in the CH₄ response highlights a need for further

55 examination of the CH₄ feedback on the abundance of OH.

1 Introduction

Hydroxyl radical (OH) is a key species of interest for numerous tropospheric chemistry studies over the past several decades. As a result of its role as the primary daytime oxidant in the lower atmosphere, OH determines how quickly many tropospheric gases and aerosols degrade or transform chemically. Notably, loss of atmospheric methane (CH₄) is dominated
by its reaction with OH. Uncertainties in the abundance of OH at the global scale, coupled with source terms of CH₄ that are difficult to quantify, have driven disagreement in the causes of recent variations in the CH₄ growth rate (Nisbet et al., 2019; Turner et al., 2019). As a key element in the CH₄ budget, tropospheric OH must be studied further to clarify its present-day abundance as well as its variability over time.

- Numerous studies have sought to constrain the OH abundance and resulting CH₄ lifetime (τ_{CH_4}) using observations, global atmospheric models, and combinations of the two. Historically, chemical inversion of methyl chloroform (MCF: CH₃CCl₃) comprised the primary method capable of gleaning information about global-scale OH burdens (Bousquet et al., 2005; Krol et al., 1998; Lovelock, 1977; Montzka et al., 2000; Prinn et al., 1987; Ravishankara and Albritton, 1995; Spivakovsky et al., 2000), though additional species that are lost by reaction with OH were also tested for this purpose (Jöckel et al., 2002; Liang et al., 2017; Miller et al., 1998; Nisbet et al., 2016, 2019; Singh, 1977; Weinstock and Niki, 1969). Models have
- ⁷⁰ likewise been relied upon to derive tropospheric OH abundance and its evolution. Notably, the Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP) generated both historical (Naik et al., 2013) and future (Voulgarakis et al., 2013) simulations from numerous chemistry-climate models, revealing large discrepancies not only in present-day τ_{CH_4}





(with values ranging from 7.1 to 14.0 years) but also in how τ_{CH_4} is expected to vary given common emissions scenarios. Note that, here and throughout, τ_{CH_4} refers to the lifetime of CH₄ due to reaction with tropospheric OH only. Most recently,

- 75 the confluence of observations with advanced modelling techniques have enabled sophisticated analyses of global OH (Holmes et al., 2013; McNorton et al., 2016; Prather et al., 2012; Rigby et al., 2017; Turner et al., 2017). Despite the advent of numerous observing systems for species with some bearing on OH chemistry in the last several decades, it is widely acknowledged that current observations are insufficient to unambiguously derive current trends in OH (Nisbet et al., 2019; Prather & Holmes, 2017; Turner et al., 2019, 2017).
- 80 While global models are insufficient for clarifying the outstanding questions regarding OH and τ_{CH_4} on their own, they can serve as valuable testbeds in which to evaluate the factors influencing OH chemistry. The dominant reactions responsible for producing, cycling, and sequestering OH (see, e.g., Spivakovsky et al. (2000)) are well characterized and represented, to varying degrees of explicitness, in modern chemical mechanisms. Despite general consensus on the immediate drivers of OH chemistry, large differences in OH can manifest due to infrequently diagnosed differences in, e.g., ultraviolet (UV) flux
- 85 to the troposphere (needed to initiate ozone (O₃) photolysis for subsequent OH primary production) due to variations in cloud parameterizations and radiative transfer codes. Similarly, differences in the representations of volatile organic compound (VOC) oxidation pathways can influence the extent to which OH is recycled following reactions with hydrocarbons. Such nuances in the chemistry of OH make OH differences between models notoriously difficult to attribute. With properly coordinated simulations and sufficient model output, however, we have demonstrated that the barriers posed
- 90 by complex, non-linear chemistry can be overcome.

The multi-dimensional system that describes OH behaviour is well-suited for study via machine learning approaches. We have previously demonstrated the utility of neural networks (NNs) for quantifying differences in OH among a small group of chemical transport models (CTMs), which rely on the specification of meteorological conditions (Nicely et al., 2017). Other groups have similarly shown the promise of machine learning techniques to better parameterize within models such complex

- 95 processes as convection (Gentine et al., 2018), radiative transfer (Krasnopolsky et al., 2009), ozone production (Nowack et al., 2018) and deposition (Silva et al., 2019), and to replace the numerical integrators that simulate chemistry within models (Keller and Evans, 2019). The community continues to develop best practices for harnessing the power of machine learning for applications in atmospheric science. We build here on the specific application of NNs to better understand model representations of OH.
- 100 In this study, we apply an NN approach to quantifying the causes of OH differences to the large group of models that participated in the Chemistry-Climate Model Initiative. We repeat our earlier analysis that identifies the primary drivers of OH and τ_{CH_4} differences among model simulations conducted with specified dynamics, for a single year. We then expand the approach to study temporal variations in OH for 1980-2015, allowing for attribution of trends and interannual variability





in τ_{CH_4} to specific parameters. Finally, we compare the derived trends in OH simulated by the CCMI models to trends derived from a previous observation-based study.

2 Model Simulations

The Chemistry-Climate Model Initiative (CCMI), carried out as an official activity of the International Global Atmospheric Chemistry (IGAC) and the Stratospheric Processes And their Role in Climate (SPARC) communities, seeks to further intermodel evaluation of chemistry-climate models (Eyring et al., 2013). Phase 1 of CCMI has designed a set of simulations,

- 110 covering both historical and future timeframes, with prescribed emissions inventories such that the interactive chemistry and its interplay with dynamical and radiative processes can be robustly compared between models. The analysis presented here focuses on one simulation, the historical specified dynamics (SD) simulation from 1980 to 2010 (REF-C1SD) (Hegglin and Lamarque, 2015; Morgenstern et al., 2017). Details of the emissions inventories recommended for this simulation can be found in Eyring et al. (2013). We have also performed the inter-model comparison portion of this analysis (Section 3.2) for
- the historical free-running simulation conducted from 1960 to 2010 (REF-C1). However, since a comprehensive examination of OH within the REF-C1 simulations was conducted by Zhao et al. (2019), those results are presented in the Supplement. We also include output from models that are not formal participants in CCMI, but provided simulations comparable to those being used here. These additional models are described below. Monthly mean fields are used for the various chemical, physical, and radiative parameters necessary for evaluating OH, described in Section 3. We analyse all models that include and provided output for the complete list of these variables.

Models that participated in the REF-C1SD simulation were nudged toward reanalysis meteorological fields such that dynamical conditions are represented with historical accuracy. The details of how nudging – of the winds, temperature, and sometimes pressure and water vapor fields – is conducted can be found in Morgenstern et al. (2017), Table S30. Models that produced REF-C1SD simulations for CCMI and provided the necessary output to complete this analysis include:
125 CAM4-Chem (Tilmes et al., 2016), EMAC-L47MA, EMAC-L90MA (Jöckel et al., 2016), MOCAGE, MRI-ESM1r1, and WACCM. For both configurations of the EMAC model, the simulations that included nudging of wave-0 temperatures were used (Jöckel et al., 2016).

Four models also contributed SD-type simulations to be analysed alongside the REF-C1SD CCMI simulations. The Goddard Earth Observing System (GEOS) model (Molod et al., 2015) conducted a "Replay" run, meaning the general

130 circulation model computes its own meteorological fields for a 3 hour simulation period, then calculates the increment necessary to match a pre-existing reanalysis data set, in this case the Modern Era Retrospective Analysis for Research Applications version 2 (MERRA-2). The increment is then applied as a forcing to the meteorology at every time step during a second run of the same simulation period. This simulation includes full interactive tropospheric and stratospheric chemistry from the Goddard Modeling Initiative (GMI) chemical mechanism (Nielsen et al., 2017) with output for years





- 135 1980-2018 at 0.625° × 0.5° horizontal resolution and 72 vertical levels (Orbe et al., 2017; Stauffer et al., 2019; Wargan et al., 2018). This simulation is referred to as "GEOS Replay." Additionally, three chemical transport models (CTMs), which directly rely on established meteorological fields such as MERRA-2 rather than calculate them, provided output used in this analysis. The OsloCTM and GEOS-Chem CTMs output all required variables for year 2000, while the GMI CTM (Strahan et al., 2013) simulated the full 1980-2015 period. We note that, while the GEOS Replay simulation described above used the GMI chemistry package, all discussion of the simulation from "GMI" refer to the separate, standalone CTM. While CTMs
- read in and use external meteorological fields rather than "nudging" or "replaying" internally calculated fields, we expect them to similarly represent realistic meteorological conditions for a given year. As such, we group them with the REF-C1SD simulations from CCMI, bringing the total number of SD-type simulations analysed to ten.

3 Methods

145 **3.1** Neural network setup

Neural networks are generated to mimic the tropospheric chemistry with respect to OH for a given model following the method outlined in Nicely et al. (2017). Briefly, one NN is trained for one model, for one simulation month at a time. To reduce computational demands, we establish NNs for four months, one for each season: January, April, July, and October. Separate NNs are trained for the SD (main text) and free-running (Supplement) simulations, and all training is performed

- 150 with output from year 2000. The training process adjusts weighting factors such that mixing ratios of OH are predicted accurately when 3-D fields of the following variables are input to the NN: pressure, latitude, temperature (T), ozone (O₃), specific humidity (H₂O), methane (CH₄), the sum of nitrogen oxide and nitrogen dioxide (NO_x≡NO+NO₂), the ratio NO:NO₂, carbon monoxide (CO), isoprene ("ISOP"=C₅H₈), formaldehyde (HCHO), the photolysis frequency of NO₂ ("JNO₂"), the photolysis frequency of O₃ to excited state O(¹D) ("JO¹D"), and stratospheric O₃ column ("O₃COL"). Note
- 155 that many of the inputs covary with one another depending on the chemical regime or meteorological conditions. A strength of the NN approach is that the inputs chosen need not be independent of each other. All chemical species are input to the NN as unitless mixing ratios, except for CH₄, which is normalized by the maximum tropospheric value. This normalization enables direct comparison of CH₄ distributions between models, despite the fact that the use of boundary conditions sometimes results in substantially different amounts of CH₄ between models. (While the CCMI models generally used
- 160 roughly consistent boundary conditions, the additional simulations that were not formally part of CCMI exhibit CH_4 concentrations mutually exclusive to those in the CCMI models.) Pressure is provided in units of hPa, temperature in K, photolysis frequencies in s⁻¹, and O₃ COL in Dobson Units (DU). Three of the inputs HCHO, NO:NO₂, and O₃ COL have been introduced to this analysis since Nicely et al. (2017), due to availability of output from all models and to the added information they encompass that may be relevant for OH chemistry. For instance, having knowledge of the partitioning of
- 165 NO_x likely enables one to more accurately predict OH quantities compared to knowing just the total abundance of NO_x. Likewise, the introduction of O_3 COL is somewhat redundant when its primary effect on OH is through attenuation of



170



ultraviolet (UV) flux to the troposphere, which is already encompassed by the input JO¹D. However, JO¹D is also altered by other factors such as clouds, which cannot as easily be included as an input for this analysis (some models provide 2-D cloud fraction fields, others output 3-D fields, and still others do not give any indication of their cloud fields). Whether strong differences in JO¹D are caused by clouds or overhead O_3 should be clarified by inclusion of O_3 COL as an input.

The neural network architecture is consistent with that of Nicely et al. (2017) and is shown in **Figure 1**. However, the number of computational nodes was doubled from 15 to 30 given the availability of more powerful computing resources. Two hidden layers each containing 30 nodes provided strong performance of the NN in reproducing the OH mixing ratios from a given model. The training process and evaluation metrics are carried out as in Nicely et al. (2017); a detailed explanation of all procedures and methods may be found therein.

3.2 Inter-model comparison approach

Once NNs are established for each model, an analysis is conducted to quantify the OH and τ_{CH_4} differences attributable to individual input terms. To accomplish this, each model, *A*, is paired with another model, *B*, such that one input to the NN of model *A* is substituted with the same field from model *B*. All other inputs are held fixed, using fields from model *A* for year

- 180 2000. Any resulting changes in OH can then be directly attributed to the substituted variable. The "swaps" that are performed in this manner undergo a process we refer to as "extrapolation control," which restricts the substituted variable from leaving the range of values over which the native model's NN was trained. If, e.g., O₃ is being substituted from CAM4-Chem into the GMI NN, we not only check that a given CAM4-Chem O₃ value lies within the minimum and maximum GMI tropospheric values, but also that the GMI value of CO at that gridpoint can be associated with the new
- 185 CAM4-Chem O₃ value. This check is performed across all variables, and essentially prevents the substitutions from venturing too far outside of the chemical regimes simulated within the native model. In the case that a swapped variable exceeds the acceptable range of values, it is revised up or down accordingly. We find that extrapolation control is critical to achieve meaningful results with the NN method, though it necessarily forces the attributed changes in OH and τ_{CH_4} to be conservative estimates.
- 190 Metrics used to evaluate the results of variable swaps include tropospheric OH integrated columns for visualization and changes in τ_{CH_4} for a globally-summed quantity. Tropospheric columns are integrated vertically and weighted by the mass of CH₄ and the temperature-dependent rate constant of reaction between OH and CH₄. The global mean lifetime of CH₄ is found using Eq. 1:

$$\tau_{CH_4} = \frac{\sum M_{air} \times \chi CH_4}{\sum [OH] \times k_{OH+CH_4} \times M_{air} \times \chi CH_4},\tag{1}$$



by ~1.2 years.



- 195 where M_{air} is the mass of air within a grid box, brackets denote number density, χ denotes mixing ratio, k_{OH+CH4} is the reaction rate constant for the OH + CH₄ reaction calculated for each grid box temperature, and summations are performed over all tropospheric model grid boxes. This formulation is equivalent to the standard lifetime calculation of burden divided by loss rate, adapted to the quantities most directly related to model outputs available (Chipperfield et al., 2014). Again, we note that this is strictly the atmospheric lifetime of CH₄ with respect to loss by tropospheric OH. If one additionally includes all stratospheric grid boxes within the above summation, annual average lifetimes of almost all models consistently increase
 - **3.3** Time series evaluation approach

A new element of this analysis applies the already-established NNs of each model to examine the time evolution of OH over several decades of simulation. For this, we focus on the REF-C1SD simulation set, as it contains the most realistic 205 representation of historical emissions and meteorological conditions, and thus is most likely to resemble true OH variations. All models that provided SD-type simulations as described in Sections 2.2 and 2.3 are included, with the exception of GEOS-Chem and OsloCTM, both of which only provided output for year 2000. Using a similar swapping technique as described in Section 3.2, the NN for a given model is used to quantify the effect of substituting individual inputs from different years. No inter-model substitutions are conducted; instead, a single input is taken from the various years of the

- simulation (1980-2015) while all other inputs are fixed to their 2000 values. Because all swaps are performed on an intramodel basis, extrapolation control is largely unnecessary, since that model's chemical regimes do not vary drastically from the original year 2000 training output. However, we do see some instances, noted in Section 4.3, of anomalous behaviour in the τ_{CH_4} results because some variables undergo significant changes, particularly between the 1980s and training year 2000. Overall, the NN technique should be sufficiently generalizable to provide meaningful results even when using inputs lying
- 215 modestly outside of the range of training values. Robustness of the results is demonstrated by the emergence of several consistent features between the eight models examined, as discussed in Section 4.

4 Results and Discussion

4.1 Native model and NN performance

Figure 2 shows values of τ_{CH_4} found for all models that produced SD-type simulations. Annually and globally averaged lifetimes vary from 6.59 years (OsloCTM) to 8.41 years (GMI). All models exhibit the expected seasonal variation in τ_{CH_4} , with minimum values in the Northern Hemisphere (NH) summer months due to higher OH at this time of year. Specifically, the seasonal variation in the global mean is a result of greater anthropogenic influence in the NH and resulting increases in concentration of two OH precursors: O₃ and NO_x.





An example of NN performance is shown for the January WACCM model in Figure 3, relative to the native model OH
fields. Tropospheric OH columns are shown for the model and NN alongside the absolute value of the difference between the two. In general, the NNs from all models show similar magnitudes and spatial patterns in their calculated OH field, with errors somewhat randomly scattered and maximizing locally to values of ~10% of the total column value. Performance of all model NNs for year 2000 is strong, with values of *τ*_{CH4} calculated from the NN-generated OH field within 0.006 years of the parent model's *τ*_{CH4} on average. The maximum error in *τ*_{CH4}, an overestimate by 0.012 years, occurs for the MRIESM1r1 model in the month of January. Performance is generally poorest in boreal winter, with average offsets in *τ*_{CH4} of 0.007 years, and strongest in boreal summer, for which the mean bias is only 0.004 years.

4.2 Inter-model comparison

The inter-model comparison component of this analysis can be understood fundamentally by the OH and τ_{CH_4} differences generated by substituting input fields between models. An example of the OH column and τ_{CH_4} changes that are calculated

235 through individual variable swaps is shown in **Figure 4**. The two models with the highest and lowest values of τ_{CH_4} , GMI and OsloCTM, respectively, are chosen for this example. Swaps performed between the two models for the month of January reveal that local O₃, JO¹D, HCHO, and NO_x account for the largest differences in τ_{CH_4} for this particular model pairing. A complete budgeting of the changes in τ_{CH_4} attributable to all inputs for GMI and OsloCTM is shown in **Table 1**. Note that the values of τ_{CH_4} shown in Table 1 correspond to lifetimes for the month of January rather than annual averages and so will differ from the lifetimes noted at the beginning of Section 4.1.

It is worth discussing several features that are evident in the visualized OH changes shown in Fig. 4. First is the spatial distribution of the OH variations. Depending on how the sink or source term undergoing the swap affects OH chemistry, the strongest impacts may occur in localized areas or may distribute evenly over the globe. For instance, varying local O₃ and NO_x (Fig. 4a, b and 4g, h, respectively) exert the greatest influence on OH over the continental climatological tropics. This

- 245 is likely a result of the anthropogenic or biomass burning emissions sources, which limit the largest differences in O₃ and NO_x between the two models to areas proximate to the South American, African, and Indonesian source regions for the month of January. The OH changes resulting from substitutions of the inputs JO¹D and HCHO, however, are distributed over oceans as well as over land masses and, in the case of HCHO, are strongest in remote marine regions. This pattern is common for species that influence OH chemistry through mechanisms that are largely independent of local emissions. In the
- 250 case of HCHO, its role as a secondary source of OH through CH₄ oxidation is relatively more important in the absence of large VOC concentrations, thus its stronger influence is seen away from terrestrial vegetation.

The second feature to note in Fig. 4 is the symmetry between input swaps in opposing directions. In other words, the swap of an input from OsloCTM into the GMI NN generally yields OH column and τ_{CH_4} changes that are equal but opposite to the



285



changes resulting from use of a GMI input in the OsloCTM NN. With few exceptions, almost all regions of OH increase (red) in one model's NN are matched by OH decreases (blue) in the other model's NN in Fig. 4. The changes in τ_{CH_4} are 255 correspondingly similar in magnitude but opposite in sign. This behaviour is expected because a swap that may, e.g., increase an OH precursor and subsequently cause an increase in OH for one model will manifest as a decrease in that same precursor when the substitution occurs in the NN of the other model. While this pattern occurs for the vast majority of cases across all model pairings and swaps performed for this analysis, there are instances when symmetry is not maintained. This could happen for two reasons. First, the sensitivities of the two models to a particular change in an OH precursor or sink 260 could differ. For example, one model may be sensitive to an increase in CH₄, causing OH concentrations to drop in response. Another model may incorporate buffering effects that allow OH to be recycled following its reaction with CH4, causing it to be less sensitive to the same change in CH4. We refer to these variations in model sensitivities as chemical mechanism differences, as they are most likely a result of the chemical reactions, species representations, or reaction rates 265 implemented within a model's chemical mechanism. The second explanation for lack of symmetry in the OH response to a model swap is a forced asymmetry in the swapped inputs themselves, imposed by the extrapolation control technique described in Section 3.2. It is possible that the swap of an input in one direction, i.e. from Model A into Model B, could proceed with no alteration to the substituted variable, while the swap in the other direction, i.e. from B to A, results in the variable lying outside the trained range of Model A. The extrapolation control process will revise the substitute variable field from Model B, such that the difference between it and the native field from Model A is lessened. As such, the first 270 swap into the NN of Model B will yield a larger magnitude change in the input as compared to the swap into the NN of Model A. The impact of these factors is indirectly quantified through a remainder term that falls out of a full budgeting analysis, described below.

A third consideration in interpreting the information presented in Fig. 4 is the conditions that must be met in order for a large change in OH to manifest through this analysis. First, the two models between which a swap is conducted must exhibit differences in the parameter of interest. Should the two models exhibit, e.g., very similar O₃ fields, then swapping one model's O₃ with the other's will produce little difference in the NN-calculated OH. Second, the model must have some OH sensitivity to the variable being swapped. If a model is insensitive to changes in CH₄, swapping in a drastically different CH₄ field may not cause a perceivable difference in OH. Therefore, the absence of an OH response does not necessarily mean that input fields are similar between models. Conversely, the existence of large OH changes indicates that differences

in the swapped input field exist between the two models *and* that the native model demonstrates a dependence of OH on that input variable.

The final point of interest in Fig. 4 is the general consistency in the signs of OH and τ_{CH_4} changes for each model. The substitutions of all four variables generally cause an increase in OH within the GMI NN (and corresponding decrease in τ_{CH_4}) and a decrease in OH (increase in τ_{CH_4}) within the OSloCTM. This feature is most pronounced for this particular pair



"Mech."



of models due to our reasoning for choosing them: they exhibit the largest difference in τ_{CH_4} among our group of 10 models. Because the native GMI model has a longer τ_{CH_4} value compared to OsloCTM, it makes sense that incorporation of OsloCTM's various OH precursor and sink fields into the GMI NN will tend to decrease the GMI τ_{CH_4} , bringing it into closer agreement with that of OsloCTM. This characteristic points to the utility of this analysis as a budgeting tool for quantifying the cause of the difference in τ_{CH_4} between two models. The τ_{CH_4} accounting for the GMI and OsloCTM set of swaps conducted for January is shown in Table 1. When considering all 12 variable swaps that were performed, the NN analysis more than explains the original gap in τ_{CH_4} between the two models. The GMI January lifetime of 9.24 years is decreased to 6.71 years ($\tau_{ORIG} + \Delta \tau$) after summing all $\Delta \tau$ values, while the OsloCTM lifetime is increased from 7.18 years to 9.48. This budgeting rarely provides a perfect accounting of the τ_{CH_4} gap due to the same reasons that give rise to asymmetric OH responses to a given swap: chemical mechanism differences and asymmetric swaps of inputs due to extrapolation control. As a result, a remainder term, found as the difference between the other model's τ_{ORIG} and the present model's $\tau_{ORIG} + \Delta \tau$, is attributed to these factors. This term is listed in the last row of Table 1 with the label

Results from analysing individual model pairs reveal a multitude of insights regarding idiosyncrasies in emissions of, global distributions of, and OH sensitivities to the various input parameters. These results, available at our FTP site provided in Data Availability, may be especially useful to the reader with an interest in a particular species or model. However, with over 4000 plots (12 species × 10 models × 9 sub models × 4 months = 4320) and 180 τ_{CH_4} budget tables generated, it is beyond the scope of this paper to highlight and explain every interesting feature. Instead, we aggregate the results across all models to identify some primary conclusions. **Figure 5** shows the change in τ_{CH_4} for a specific model and substituted input variable, averaged over all nine pairings. For example, the data point shown for CAM4-Chem JO¹D is calculated from the nine $\Delta \tau_{CH_4}$ values obtained when swapping the JO¹D fields from the other nine models into the CAM4-Chem NN. The circular point represents the mean of those nine values, while the whiskers indicate one standard deviation about the mean. Aggregate results shown in this manner are compiled both for individual months (available on the FTP site noted above) as well as for annually averaged output. The latter is calculated as the average of the four monthly mean and standard deviation values, and is shown in Fig. 5.

As with the individual OH tropospheric column change plots (Fig. 4), numerous conclusions can be drawn by studying the aggregated results in Fig. 5. The method for reading the data in Fig. 5 is demonstrated in the following example. The mean $\Delta \tau_{CH_4}$ value attributable to JO¹D for the WACCM model is +0.99 years. This indicates that use of JO¹D fields from other models causes τ_{CH_4} to increase by ~1 year, meaning the native JO¹D field from WACCM imparts a low bias to τ_{CH_4} of 1

315 year, relative to the other models. A low τ_{CH_4} would result from OH concentrations being too high. Since OH and JO¹D are positively correlated (i.e., JO¹D can be thought of as a source for OH) the too-high OH is an indication of too-high JO¹D. In



examination.

320



general, positive values of $\Delta \tau_{CH_4}$ correspond to relative high biases in input parameters that are source terms for OH and to low biases for species that instead serve as sinks. This reasoning is less straightforward for species such as HCHO, which can both produce and consume OH, while it is also produced by OH-initiated oxidation. We stress that these comparisons are strictly relative to other models, not to any observation or other indication of truth. So, points that appear as outliers in Fig. 5 should not necessarily be interpreted as an erroneous result, but rather should be considered as an area for further

The ordering of variables along the x-axis of Fig. 5 denotes the average magnitude of $\Delta \tau_{CH_4}$ values across all models, with parameters on the left accounting for the largest τ_{CH_4} differences. As such, JO¹D is the largest driver of OH differences in 325 the CCMI SD model simulations, followed by local O₃ and NO_x. The subsequent variables H₂O, CO, the NO:NO_x ratio, and HCHO cause moderate variations in tropospheric OH, while ISOP, CH₄, JNO₂, O₃ COL, and T are not responsible for intermodel spread in τ_{CH_4} . We note that T differences between the SD simulations are likely limited due the meteorological constraints imposed on the models. However, examination of the free-running simulations, discussed in the Supplementary Material, also shows practically no impact of T on OH. Thus, we conclude that the effect of temperature on OH chemistry is

- 330 likely indirect, acting through pathways embodied by other variables, such as H₂O and species that exhibit strongly temperature-dependent reaction rates. Finally, the Mech. term, described in the discussion of Table 1, appears on the far right, indicating its origins as a remainder term from the budget analysis of individual model pairs. The magnitudes of $\Delta \tau_{CH_4}$ values attributed to chemical mechanism differences and asymmetric swaps between models are large enough to consistently rank the Mech. term third, between O₃ and NO_x, in terms of importance for OH in this analysis. Especially in model
- 335 simulations conducted with common emissions inventories, we expect some of the disparity in a short-lived species like OH to emerge from differences in chemical mechanism implementations. In other words, when responses in OH to a given change in a source or sink term differ between two models, the remainder term will increase, representing variations in the sensitivity of OH that presumably arise due to the two different implementations of the chemical mechanism. It is possible that other factors are represented by this term; e.g., other chemical species that influence OH chemistry but are not
- 340 considered in the NN analysis could contribute to the Mech. term. However, previous analysis using a 0-D chemical box model as a "standard" mechanism in Nicely et al. (2017) suggested a correlation between actual biases in OH imparted by a given model's chemical mechanism and the remainder term resulting from the NN analysis. Therefore, we have some confidence that the Mech. term is meaningful, though significant further study would be required to parse the actual mechanistic differences responsible for imparting bias in OH calculations.
- 345 Significant inter-model differences in the largest driver of τ_{CH_4} spread, JO¹D, could arise from two possible sources. The amount of solar UV light penetrating down to the troposphere is largely dictated by the stratospheric column O₃ amount. However, the differences in total O₃ column are generally small and insufficient to cause the variations in JO¹D seen among the CCMI models. Rather, JO¹D likely varies to a great extent due to differences in cloud cover, and dissimilar treatments of



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clouds within model photolysis codes. Supplementary Figure S1 highlights this effect by showing the ratio of JO^1D at the 350 surface to JO^1D in the upper troposphere (UT) for each model. The relatively small column amounts of O_3 within the troposphere should account for very little absorbed UV light, making it much more likely that deviations in this ratio from 1.0 are driven by scattering due to clouds and possibly aerosols. The fact that models show large spatial differences in this ratio is a strong indication that clouds underlie the model differences in JO^1D .

While the model differences in JO¹D, O₃, NO_x, and chemical mechanisms appear to drive the bulk of the τ_{CH_4} spread among this group of CCMI models, we emphasize that individual models may not adhere to these conclusions. As such, any efforts to improve a particular model should instead focus on the results specific to that model. For instance, HCHO plays a very small role in describing inter-model differences in OH on average, but for the OsloCTM model, HCHO is a much more important factor. Thus, we refrain from offering an across-the-board solution for remedying the large model spread in τ_{CH_4} and instead suggest a more individualized approach of studying plots such as those shown in Fig. 4 for more spatially and temporally resolved information. Visualizations of all model swaps, for all months and species, are available at our FTP site

360 temporally resolved information. Visualizations of all model swaps, for all months and species, are available at our FTP site provided in Data Availability for this purpose.

There are several other qualifications to note when considering the results of the inter-model comparison. One is the negating effect between the JO¹D and tropospheric O₃ variables. Many, but not all, model $\Delta \tau_{CH_4}$ values for JO¹D in Fig. 5 are opposite in sign to the $\Delta \tau_{CH_4}$ values attributed to O₃. Physically, photolysis of tropospheric O₃ by light at wavelengths below 336 nm to form excited state O(¹D) and subsequent reaction with H₂O to form OH is a loss pathway for O₃.

- Therefore, more UV flux will tend to decrease tropospheric O₃ concentrations while increasing OH, and vice versa. This physical mechanism, then, can explain the frequent cancellation of the $\Delta \tau_{CH_4}$ values attributed to these two factors. Should a modeler attempt to alter a model's OH field by forcing adjustments in its JO¹D, the opposing impact of tropospheric O₃ may result in no change for the value of τ_{CH_4} . However, this does not preclude the finding that both JO¹D and tropospheric O₃
- are substantially different in the models for reasons we do not fully understand. Tropospheric O₃ can also vary between models for reasons external to the radiative environment. For instance, differences in the stratosphere-troposphere exchange, wet and dry deposition, and lightning NO_x emissions can each cause substantial variations in tropospheric O₃ among models (Wild, 2007). Further parsing of the reasons for the O₃ differences seen among the CCMI models is difficult without specialized output, including tracers such as ozone of stratospheric origin and NO_x generated by lightning. We recommend a
- 375 targeted study to address the underlying reasons for the variations in tropospheric O₃.

The other qualification concerns the issue of causation versus correlation. Machine learning techniques, and NNs in particular, are generally more adept at identifying the predictors of a certain phenomenon than traditional methods, such as multiple linear regression. However, it is still possible that an input that is tightly correlated with the output may be misidentified as a driver of variations in the output. This is particularly relevant to keep in mind for species that serve as





380 sinks of OH, such as CO and CH₄. Whether a decline in OH initiates or results from an increase in its sinks is difficult to differentiate, even with advanced analysis methods. Therefore, descriptions of CO and CH₄ as drivers of OH variations in this text may just as well be interpreted conversely, as downstream indicators of the change in oxidizing capacity.

4.3 **Time series evaluation**

- The second half of our NN analysis interrogates temporal trends in OH and τ_{CH_A} . Figure 6 shows the evolution of τ_{CH_A} in 385 the SD-type simulations conducted for 1980-2010. Two models, GEOS-Chem and OsloCTM, only provided output for year 2000, and so only appear as single points in Fig. 6. In addition, some models provided output beyond year 2010; output from years through the end of 2015 was included when available. The lifetimes all show a general downward trend over time, consistent with the upward trend in global mean tropospheric OH concentration shown by Zhao et al. (2019) (their figure 4).
- Swaps of input variables to a NN are conducted on an intra-model basis, with the goal of determining which OH precursors 390 and sinks are responsible for OH variations over time. The results of these swaps are shown for each model in Figure 7. Changes in τ_{CH_4} attributable to each parameter are displayed as a function of year. Because we use the same NNs established for the inter-model comparison described in Section 3.2 trained on output from year 2000, the values of $\Delta \tau_{CH_A}$ for all species in year 2000 of Fig. 7 is zero by design. As an input field from another year is swapped into the NN, however, 395 OH differences manifest and are denoted by the corresponding change in τ_{CH_A} .

While significant diversity in the drivers of OH variability across models is evident from Fig. 7, there are also several distinctive features that appear repeatedly. For instance the response of τ_{CH_4} to changes in CO shows a prominent peak in year 1998 in all models except one. To gauge the role of emissions in this response, we show in Supplemental Figures S2-6 the time series of CO mixing ratios and other parameters averaged for the region most impactful to τ_{CH_4} : the tropical lower troposphere (latitudes between 30°S and 30°N, pressures greater than or equal to 700 hPa). Indeed, CO mixing ratios 400 maximize in almost all models in year 1998, likely as a result of the emissions inventory reflecting the extreme biomass burning and strong El Niño Southern Oscillation (ENSO) event during that and the preceding year (Duncan et al., 2003 and references therein). The increase in τ_{CH_4} can thus be explained by the increased CO sink of OH, causing a temporary depletion of the oxidant. In addition, less distinctive peaks in τ_{CH_4} due to CO are identified in other years with strong El Niño conditions, notably 1982-1983, 1987, and 1991-1992 (Duncan et al., 2003).

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The impacts of several other variables on τ_{CH_4} also demonstrate behaviour with reasonably identifiable causes. A prolonged decrease in τ_{CH_4} due to JO¹D from 1992 to 1998 is evident in the analysis of the CAM4-Chem, GEOS Replay, GMI, MRI-ESM1r1, and WACCM NNs. This corresponds with both the eruption of Mount Pinatubo in 1991 and the prolonged ENSO event of 1990-1995 (Allan & D'arrigo, 1999; Duncan et al., 2003). The former event likely impacted JO¹D through the





- 410 decrease in stratospheric O₃ that resulted (Aquila et al., 2013; Tie and Brasseur, 1995), while the latter caused reduction in cloud cover due to drought conditions (Duncan et al., 2003). Both effects will tend to increase the flux of UV light to the troposphere, increasing the primary production of OH and decreasing τ_{CH4}, as seen in Fig. 7. Interestingly, the τ_{CH4} response to H₂O is moderately anticorrelated with CO. This is particularly evident for year 1998 in many of the models, when the biomass burning mentioned above was driven specifically by El Niño conditions. Although a strong El Niño causes drought
 415 conditions over some regions, it is more fundamentally associated with warming sea surface temperatures and increased evaporation, particularly in the tropical Pacific Ocean. Thus, it is reasonable that larger values of specific humidity will tend to increase OH primary production during an El Niño year, as suggested by the decrease in τ_{CH4} shown in Fig. 7. An apparent increase in O₃ also coincides with the 1998 ENSO event, determined by the decreasing component of τ_{CH4}. The
- prevalence of biomass burning would indeed cause increases in tropospheric O₃ through increased emissions of its 420 precursors, CO, VOCs, and NO_x. Additionally, the τ_{CH_4} response to O₃ shows the most distinguishable trend of all the variables over the full 1980-2015 period. Steady decreases in τ_{CH_4} due to O₃ imply an increasing tropospheric O₃ burden, a modelling result supported by observations (Verstraeten et al., 2015).

We also note the appearance of spurious results in several cases. The τ_{CH_4} responses to CH₄ in EMAC-L47MA and EMAC-L90MA as well as to O₃ COL in MOCAGE extend to very large negative values in the early part of the time series. 425 Chemical conditions during the 1980s would differ most markedly from the regimes simulated in year 2000, on which the NNs are based. Particularly for concentrations of CH₄, which underwent monotonic rise aside from a stabilisation period from 2000 to 2007 (Turner et al., 2019), conditions in 1980 could be quite different. However, as was noted in Section 3.1, CH₄ inputs to the NNs are normalized against the maximum tropospheric value. The field of CH₄ for each year is likewise normalized against the maximum CH₄ for that year, so a strong response in τ_{CH_4} must indicate a significant change in the 430 distribution of CH₄, not just in changes in its concentration over time. Indeed, Supplementary Figure S7 shows the normalized CH₄ values used as input to the NNs for the pressure level closest to the surface. For each EMAC configuration (for the month in which the τ_{CH_4} response shown in Fig. 7 is largest and most unphysical), the CH₄ distributions in the 1980s

Southern Hemisphere drop relative to the larger concentrations in the Northern Hemisphere. Other models, such as WACCM shown in the bottom panels of Fig. S7, show practically no inter-annual change in the CH₄ distribution for a given month. This behaviour in the EMAC model appears to result from implementation of a Newtonian relaxation scheme to determine a time-varying, latitude-dependent lower boundary condition for CH₄ (Jöckel et al., 2016). Our spurious NN result may indeed be explained by a slowdown in the rate of increase in CH₄ concentrations at the lower boundary initiated in 1980, evident in supplementary figure E1 of Jöckel et al. (2016). While this method of determining boundary conditions

do show notable change from the year 2000 distribution used for training. Specifically, relative CH₄ mixing ratios in the

440 generally represents a more sophisticated treatment of CH₄, within the context of this analysis, it imparts an artificially strong signal in OH and τ_{CH_4} .





For the other occurrence of anomalous behaviour, MOCAGE shows an unrealistically large response of τ_{CH_4} to O₃ COL in the 1980s (Fig. 7f), a result not corroborated by any other model. Supplementary Figure S8 illustrates the likely cause of this behaviour. While most models exhibit modest changes in total O₃ COL between 1980 and 2000, including GEOS Replay 445 shown in the top set of panels, the MOCAGE model (bottom panels) shows much larger column amounts in year 1980. These values fall well outside the range of O₃ COL amounts on which the NN was trained, so unrealistic behaviour of the NN in this case is not surprising. Due to the anomalous NN behaviour in these three cases, the responses to EMAC CH₄ and MOCAGE O_3 COL are excluded from subsequent analysis of the NN time series results.

Figure 8 shows the multi-model mean attribution of variations in τ_{CH_4} . Many of the same features identified in Fig. 7 also 450 emerge here: clear definition of strong ENSO years in the CO response, apparent Mt. Pinatubo effects in the JO¹D response, and a general downward trend in τ_{CH_4} due to O₃ are all observed. Also, as might be expected from the inter-model comparison results discussed in the prior section, JO¹D, O₃, NO_x, H₂O, and CO account for many of the strongest OH variations over time (Fig. 7) as well as between models (Fig. 5).

We also perform linear fits to each response time series in Fig. 8. The resulting trends in τ_{CH_4} are shown in Figure 9, panel

455 (a). The interannual variability of τ_{CH_A} is also calculated as the standard deviation of the detrended time series, shown in Fig. 9b. Negative trends in τ_{CH_4} due to H₂O, JO¹D, NO_x, and O₃ stand out as largest in magnitude. The sum of all factors shown in Fig. 9a is -1.9% decade⁻¹, which is comparable to the mean downward trend in τ_{CH_4} seen in Fig. 6, -1.8% decade⁻¹. Time series of the model input variable fields show corresponding trends, with parameters that serve as source terms of OH increasing over time (Supplemental Figures S2-5). Tropospheric O_3 and NO_x show clear upward trends over time, while

- 460 H_2O and JO^1D show upward trends with more variability, which is also conveyed by the error bars in Fig. 9a. The interannual variability attributed to CO in Fig. 9b is also consistent with the large year-to-year swings in tropical lower tropospheric CO mixing ratios shown in Supplemental Figure S2. While Fig. 9a suggests that CO exhibits very little overall trend between 1980 and 2015, we note there is a discernible increase in CO prior to ~1998 in Fig. S2 followed by a steady decline thereafter. This is consistent with remote site measurements that show significant negative trends in CO since the 465 late 1990s (Zeng et al., 2012).

Finally, the attributed trends in τ_{CH_4} from the CCMI models (Fig. 9a) are compared in Figure 10 to trends in tropospheric mean OH concentration ("[OH]^{TROP}") from a previous observation-based analysis (Nicely et al., 2018). In that work, TOMS/OMI/SBUV observations of total column O₃ were used to infer radiative effects on the OH burden, while water vapor from the AIRS instrument, CH₄ from surface observations, NO_x from a global model simulation constrained to realistic emissions, and temperature from the MERRA-2 reanalysis were analysed to calculate chemical impacts on [OH]^{TROP}. In

470 Nicely et al. (2018), the trend in [OH]^{TROP} due to NO_x encompassed the effects of both the total abundance and the partitioning of NOx, while the O3 COL factor encompassed all radiative effects on OH. Thus, to perform a "like for like"



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comparison, the τ_{CH_4} trends due to NO_x and NO:NO_x are combined, as are the trends due to O₃ COL and JO¹D shown in Fig. 9a. Error bars shown in Fig. 10 represent the 1 σ uncertainty in the slope of the linear fit and, in the case of combined trends, are found by summing in quadrature the individual uncertainties. Because τ_{CH_4} varies with the inverse of OH concentration, note that the x-axis of Fig. 10 is inverted and a –1:1 line is shown in grey.

The trends in τ_{CH_4} from this analysis and in [OH]^{TROP} from Nicely et al. (2018) are in reasonably good agreement for H₂O, NO_x, O₃ COL, and temperature. In particular, the two trends due to H₂O agree within the uncertainties, with τ_{CH_4} decreasing by ~0.5 % decade⁻¹ and [OH]^{TROP} increasing at almost the same rate. The impacts of NO_x and O₃ COL are found to increase

- 480 OH concentrations in both studies, though the impacts on τ_{CH_4} from the CCMI models are found to be larger in magnitude than the observational estimate. The small impact of temperature, tending to lessen the OH burden, is also in close agreement between the two studies, with the CCMI models again showing a slightly stronger response. The response of OH to CH₄ however, is in poor agreement between the two studies. The previously determined observation-based estimate of $[OH]^{TROP}$ trend due to CH₄ was -1.01 ± 0.05 % decade⁻¹ while the CCMI model-based trend in τ_{CH_4} is only $+0.06\pm0.07$ %
- 485 decade⁻¹. On one hand, the treatment of CH₄ as a normalized value within the NN analysis, as noted above in the discussion of Fig. 7, precludes a realistic estimate of the OH response to changes in CH₄. Rather, the trend estimate calculated by the NN analysis of CCMI models represents the impact on OH of changes in the distribution of CH₄ within the troposphere. Since the source regions of CH₄ are not expected to change substantially over the 1980-2015 period, it is not surprising that the CCMI model-based trend is small. Meanwhile, the other estimate of the [OH]^{TROP} trend due to CH₄ from Nicely et al.
- 490 (2018) is not without limitations. As was acknowledged in that paper, the box model method used to estimate the sensitivities of OH to CH₄ (among other species) is inherently inadequate for capturing complex coupling of chemical systems and downstream effects. For example, the box modelled sensitivity of OH to variations in CH₄ were found for a range of latitude, pressure, and NO_x values (since the latter determines whether CH₄ oxidation consumes or regenerates OH radicals). To maintain the characteristics of those chemical regimes, then, O₃ was input and held fixed in the box model
- 495 simulation. As a result, especially in the relatively low-NO_x conditions prevalent throughout much of the troposphere, an increase in CH_4 would tend to consume OH without the corresponding increase in O_3 expected to result from greater CH_4 oxidation. That increase in tropospheric O_3 would offset some of the OH loss by increasing primary production, a process that should be captured in a fully coupled chemistry-climate model like those participating in CCMI.

We encourage the further examination of the response of OH to CH4, known as the CH4 feedback, on the global scale. This

500 topic has been of interest for some time (Holmes et al., 2013; Prather et al., 2001), though the necessity of providing boundary conditions for surface CH₄ rather than fluxes in models hampers our ability to realistically simulate CH₄. Regardless, a model approach using fully-coupled tropospheric chemistry, such as that performed by Holmes et al. (2013) for three CTMs, would provide a more direct measure of the CH₄ feedback on OH than both approaches depicted in Fig. 10.





Except for trends attributed to CH₄, the results depicted in Fig. 10 show relatively robust findings regarding the responses of [OH]^{TROP} and τ_{CH_4} to the factors examined through two independent studies.

5 Conclusions

We perform a neural network analysis of the monthly mean output from historical simulations of ten models that participated in CCMI for the purposes of understanding OH and τ_{CH_4} differences and temporal trends. NNs are trained to reproduce OH mixing ratios for a given model using 3-D fields of 12 OH precursor and sink parameters. Performing swaps of the NN 510 inputs between models produces a quantitative estimate of the difference in τ_{CH_4} that can be attributed to variations in the substituted variable. Among the ten models that we examine, on average, variations in JO¹D, local O₃, NO_x, and chemical mechanisms account for the largest differences in τ_{CH_4} . Model diversity in representations of H₂O, CO, the partitioning of NO_x, and HCHO is responsible for moderate OH differences, while isoprene, CH₄, JNO₂, overhead O₃ column, and temperature account for little-to-no variation in OH. However, the relative importance of a particular variable is highly 515 model-dependent, so any effort to improve the representation of OH within a given model should be guided by that particular

515 model-dependent, so any effort to improve the representation of OH within a given model should be guided by that particular model's results.

We also analyse time series of τ_{CH_4} using the year 2000 NNs generated for the first half of the study. All models exhibit a downward trend in τ_{CH_4} between 1980 and 2015, varying from -0.54 % decade⁻¹ to -2.97 % decade⁻¹ (average of -1.83 % decade⁻¹). Swaps of NN inputs are conducted between years rather than between models, so attributions of the factors influencing trends in τ_{CH_4} are found for each model and then combined into a multi-model mean result. This analysis indicates that the largest contributors to the decreasing trend in τ_{CH_4} are O₃, JO¹D, NO_x, and H₂O, while CO also imparts a large degree of interannual variability. Features due to strong ENSO events and associated biomass burning as well as the eruption of Mount Pinatubo are discernible in the time series of attributed variations in τ_{CH_4} . In particular, the species CO, H₂O, and O₃ instigate prominent responses during strong El Niño years. Finally, the attributed trends in τ_{CH_4} from the NN 525 analysis of CCMI model output are compared to trends in tropospheric mean OH concentration found previously in the observation-based study of Nicely et al. (2018). While the strong response of τ_{CH_4} to increasing H₂O over time appears to be a robust result, disagreement on the CH₄ feedback on OH between the two studies highlights limitations in the approaches of both, in addition to more systemic issues in the community's ability to model CH₄.

The NN and machine learning methods in general provide a valuable tool for performing insightful model intercomparisons of complex systems in a computationally-efficient manner. These approaches, however, must be undertaken with care to avoid erroneous results and recognition of their limitations. At present, we have devised a method to identify the drivers of OH variations, whether between models or between years, at coarse temporal resolution. Much future work is needed, though: observations must be incorporated to introduce a ground truth element to this analysis in a manner that either adjusts





for or avoids disconnects between coarse versus local/instantaneous spatiotemporal scales and appropriately accounts for 535 measurement uncertainty; an analysis of model output with much higher temporal frequency is needed to identify exactly where model differences in chemical mechanisms lie; and subsequent studies of why the various OH precursor and sink fields differ are required to make this analysis of greatest utility for improving model representations of τ_{CH_4} . While these challenges are significant, they are not insurmountable, especially as machine learning and other advanced statistical analysis techniques continue to be developed and honed.

540 Data Availability

All output from most of the models that participated in CCMI is available at the Centre for Environmental Data Analysis (CEDA), the Natural Environment Research Council's Data Repository for Atmospheric Science and Earth Observation, at http://data.ceda.ac.uk/badc/wcrp-ccmi/data/CCMI-1/output. WACCM and CAM4-Chem output for CCMI is available for download http://www.earthsystemgrid.org. at For instructions for access to both archives see http://blogs.reading.ac.uk/ccmi/badc-data-access. Output from the models that were not formal participants in CCMI Phase 545 1 is available from the co-authors who performed the model simulations; please contact the corresponding author with requests. A complete set of figures and tables generated by the model intercomparison and time series analyses is available at the FTP site https://acd-ext.gsfc.nasa.gov/anonftp/acd/atmos/jnicely/. In the event that this site is no longer active, please contact the corresponding author for access to all results.

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Author contributions

JMN and RJS conducted initial design of the method. JMN carried out the analysis. Development and refinement of the analysis were further guided by BND, GMW, and TFH. All other authors provided model output central to the analysis. JMN drafted the manuscript, and all co-authors provided assistance in finalizing the figures and text.

Competing interests

The authors declare that they have no conflicts of interest.

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| | | GMI | OsloCTM |
|---|--------------------|-------|---------|
| $\tau_{CH4,ORIG}^{a}$ (year) | | 9.24 | 7.18 |
| $\Delta 	au_{CH4}$ due to ^b : | O3 | -0.91 | +0.79 |
| | JO^1D | -0.59 | +0.60 |
| | НСНО | -0.64 | +0.51 |
| | NO _x | -0.45 | +0.33 |
| | JNO ₂ | -0.34 | +0.15 |
| | Isoprene | -0.03 | +0.28 |
| | CO | +0.19 | -0.07 |
| | H ₂ O | +0.10 | -0.13 |
| | CH4 | +0.11 | -0.06 |
| | NO/NO _x | +0.07 | -0.05 |
| | O3 COL | -0.02 | -0.06 |
| | Т | -0.02 | +0.00 |
| $\Delta 	au_{CH4,TOT}$ ° | | -2.52 | +2.30 |
| $\tau_{CH4,ORIG} + \Delta \tau_{CH4,TOT}$ | | 6.71 | 9.48 |
| Mech. ^d | | +0.47 | -0.24 |

Table 1. Accounting of CH₄ lifetime differences between GMI and OsloCTM simulations for January, 2000.

 ${}^{a}\tau_{CH4,ORIG}$ represents value of τ_{CH4} evaluated directly from the model.

^b $\Delta \tau_{CH4}$ calculated from output of NN when noted variable is substituted with values from the other model.

°Sum of all $\Delta \tau_{CH4}$ values calculated for each input substitution.

^dRemainder of original τ_{CH4} difference not accounted for by NN substitutions; calculated as $\tau_{CH4,ORIG}$ (model A) – [$\tau_{CH4,ORIG}$ (model B) + $\Delta \tau_{CH4,TOT}$ (model B)].







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Figure 1. Architecture for neural networks generated in this study. Blue boxes designate inputs (left) and output (right), red triangles indicate bias terms, green circles indicate nodes at which activation functions are performed, and grey arrows represent the weighting terms, which are optimized through the training process. For full details of the neural network setup and training, we refer readers to Nicely et al. [2017]. Although 15 nodes are shown here in each hidden layer, 30 were actually used for all NNs in this study.

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Figure 2. Seasonal variation in CH₄ lifetime for year 2000 for the CCMI specified dynamics (REF-C1SD) and chemical transport model simulations.







Figure 3. Tropospheric OH columns for the WACCM model REF-C1SD simulation, January 2000. (a) Columns calculated directly from the WACCM output; (b) columns calculated from the output from the WACCM January NN run with inputs from the native model; (c) difference in column values, (NN – model). Methane lifetime values calculated from 3-D OH fields from WACCM and from the WACCM NN are inscribed in panels (a) and (b), respectively. The methane lifetime difference, (NN – model), is noted in panel (c).







Figure 4. Changes in tropospheric OH column resulting from swap of indicated variable from another model into the NN of the native model for the specified dynamics simulation of January, 2000. Swaps of the inputs O_3 (a, b), $J(O_3 \rightarrow O^1D)$ (c, d), HCHO (e, f), and NO_x (g, h) are shown for the GMI (left) and OsloCTM (right) NNs.







Figure 5. Averaged changes in CH₄ lifetime accrued for a specified model (color), across all swaps of the indicated variable (xaxis) from all other models. Results are shown annually averaged for year 2000 of the specified dynamics REF-C1SD CCMI and chemical transport model simulations. Circle indicates the mean change in CH₄ lifetime; bars represent the 1σ standard deviation from all model pairings. Variables along the x-axis are ranked by averaged magnitude of the $\Delta \tau_{CH4}$ values (i.e., inputs located farther left are responsible for larger differences in CH₄ lifetime), except for the "Mech.+Nonlin." term, which is shown last to indicate its role as a remainder term. Model name abbreviations are "CAM4" for CAM4-Chem, "EM47" for EMAC-L47MA, "EM90" for EMAC-L90MA, "GRep" for GEOS Replay, "GCHM" for GEOS-Chem, "GMI" for GMI, "MOC" for MOCAGE,

"MRI" for MRI-ESM1r1, "OSLO" for OsloCTM, and "WACC" for WACCM.







775 Figure 6. Time series of CH₄ lifetime from REF-C1SD models. Only one year of output was available for two models (OsloCTM and GEOS-Chem), so their results are shown only as a single data point at year 2000.







780 Figure 7. Attributions of changes in CH₄ lifetime relative to year 2000 of the REF-C1SD simulations. Within the NN of a given model, use of individual inputs (indicated by color) from years other than 2000 result in a change and OH and subsequent CH₄ lifetime, shown here.







Figure 8. Same as Figure 7, but the average across all eight models, except with the CH₄ line from EMAC-L47MA and EMAC-L90MA and O₃ Column from MOCAGE removed.







Figure 9. Multi-model mean linear trend (a) and interannual variability (b) in τ_{CH4} attributed to each variable examined through the NN method.







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Figure 10. Comparison of the attributed trends in τ_{CH4} found in this work according to the REF-C1SD simulations performed for CCMI (y-axis) to the attributed trends in tropospheric mean OH ("[OH]^{TROP}") found based on observations in Nicely et al. [2018]. The grey dashed line indicates the –1:1 line, as values should be anti-correlated. The τ_{CH4} trend numbers from this work for NO_x combine the NO_x total abundance and partitioning (NO/NO_x) values from Figure 9, and for O₃ Column combine the J(O¹D) and O₃ Column values, as both effects are encompassed in the determination of [OH]^{TROP}.