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We thank you very much for the comments. In this manuscript, we checked the text for grammar mistakes carefully. Please see the details in the following marked-up manuscript.

Anything about our paper, please feel free to contact me at <u>ghwang@geo.ecnu.edu.cn</u>.

Best regards,

Sincerely yours

Gehui Wang Jul. 28, 2019

1	Molecular characteristics and diurnal variations of organic aerosols
2	at a rural site in the North China Plain with implications for the
3	influence of regional biomass burning
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#### 29 Abstract

30	Field burning of crop residue in early summer releases into the atmosphere a
31	large amount of pollutants with significant impacts on the air quality and aerosol
32	properties in the North China Plain (NCP). In order to investigate the influence of this
33	regional anthropogenic activity on molecular characteristics of organic aerosols, we-
34	collected PM <sub>2.5</sub> filter samples every-were collected with a <u>3 hours3-hr interval</u> at a
35	rural site of NCP during from June 10 <sup>th</sup> to 25 <sup>th</sup> , 2013, and analyzed them for more
36	than 100 organic tracer compounds, including both primary (n-alkanes, fatty
37	acids/alcohols, sugar compounds, polycyclic aromatic hydrocarbons, hopanes, and
38	phthalate esters) and secondary (phthalic acids, isoprene , $\alpha$ / $\beta$ pinene, $\beta$ -
39	caryophyllene, and toluene derived products) organic aerosol (SOA) tracers (phthalic
40	acids, isoprene-, $\alpha$ -/ $\beta$ -pinene, $\beta$ -caryophyllene, and toluene-derived products), -as
41	well as for-organic carbon (OC), elemental carbon (EC), and water-soluble organic
42	carbon (WSOC). Total concentrations of the measured organics ranged from 177 to
43	6248 ng m^-3 (mean 1806 $\pm$ 1308 ng m^-3) during the study period, most of which were
44	contributed by sugar compounds, followed by fatty acids and fatty alcohols.
45	Levoglucosan (240 $\pm$ 288 ng m <sup>-3</sup> ) was the most abundant single compound and
46	strongly correlated with OC and WSOC, suggesting that biomass burning (BB) is an
47	important source of summertime organic aerosols in this rural region. Based on the
48	analysis of fire spots and backward trajectories of air masses, two representative
49	periods were classified, which are (1) Period 1 (P1), Jun 13th 21:00-16th 15:00, when
50	air masses were uniformly from the southeast part of NCP, where intensive open-field
51	biomass burning biomass burning (BB) occurred; and (2) Period 2 (P2), Jun 22 <sup>nd</sup>

52	12:00-24 <sup>th</sup> 06:00, which is representative of local emission. Nearly all the measured
53	PM components showed much higher concentrations in P1 than in P2. Although <i>n</i> -
54	alkanes, fatty acids, and fatty alcohols presented similar temporal/diurnal variations as
55	those of levoglucosan throughout the entire period, their molecular distributions were
56	more dominated by high molecular weight (HMW) compounds in P1, demonstrating
57	an enhanced contribution from BB emissions. In contrast, intensive BB emission in
58	P1 seems to have limited influences on the concentrations of polycyclic aromatic
59	hydrocarbons (PAHs), hopanes and phthalate esters. Both 3-hydroxyglutaric acid and
60	$\beta$ -caryophyllinic acid showed strong linearly correlations with levoglucosan (R <sup>2</sup> =0.72
61	and 0.80, respectively), indicating that BB is also an important source for terpene-
62	derived SOA formation. A tracer-based method was used to estimate the distributions
63	of biomass-burning OC, fungal-spore OC and secondary organic carbon (SOC)
64	derived from isoprene, $\alpha$ -/ $\beta$ -pinene, $\beta$ -caryophyllene, and toluene in the different
65	periods. The results showed that the contribution of biomass-burning OC to total OC
66	in P1 (27.6%) was 1.7 times of that in P2 (17.1%). However, the contribution of SOC
67	from oxidation of the four kinds of volatile organic compounds (VOCs) increased
68	slightly from 16.3% in P1 to 21.1% in P2.
69	Key words: Organic aerosols; Molecular composition; North China Plain; Biomass

- 70 Burning

#### 72 1. Introduction

73	Organic aerosols (OA, i.e., the organic fraction of particles) constitute a substantial	
74	fraction (~10-90%) (Jimenez et al., 2009;Zhang et al., 2007a;Hallquist et al., 2009)-of	域代码已更改
75	atmospheric particles (Jimenez et al., 2009;Zhang et al., 2007a;Hallquist et al., 2009),	域代码已更改
76	and have significant effects on global and regional climate (Venkataraman et al.,	
77	2005;Kanakidou et al., 2005), air quality (Aggarwal et al., 2013;Wang et al., 2006b),	
78	human health (Lelieveld et al., 2015), and ecosystems (Tie et al., 2016). Organic	
79	aerosols in the atmosphere can be emitted directly from various sources, such as	
80	combustion of fossil fuels combustion, biomass burning, plant emission, and so on,	

which is defined as primary organic aerosols (POA). On the other hand, atmospheric 81 82 secondary OA (SOA) are produced from photochemical oxidation products of volatile organic compounds (VOCs) via gas-particle conversion processes such as nucleation, 83 condensation and heterogeneous chemical reactions (Hallquist et al., 2009). These 84 organic species could modify physicochemical characteristics of atmospheric aerosols 85 such as hygroscopicity, albedo, and oxidation state (Dinar et al., 2008;Chan et al., 86 2005;Fu et al., 2010). Thus, a thorough understanding of molecular composition and 87 source of organic aerosols is necessary in order to address aerosol related environmental 88 issues and to improve the accuracy of modelling studies. 89

Tremendous amounts of air pollutants including both particulate matters (PM) and their gaseous precursors (e.g., SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub>, and VOCs) are emitted into the atmosphere from power plants, industries and vehicles due to rapid economy development in China, leading to serious air <u>conditions-pollution</u> in the recent decades

94	(Zhang et al., 2009;Guo et al., 2014;Wang et al., 2016;Huang et al., 2014;Li et al., 2017).
95	The North China Plain (NCP) has been recognized as one of the most polluted regions
96	in the world, with very high concentrations of $\text{PM}_{2.5}$ on the ground surface (van
97	Donkelaar et al., 2010). The NCP is also one of the most significant aerosol sources in
98	the world, which has a significant impact on the East China Sea and Western North
99	Pacific (Andreae and Rosenfeld, 2008). Thus, extensive efforts have been made in
100	recent years to characterize the sources, properties, and processes of PM in the NCP.
101	Most of these results concluded that the severe air pollution in the region is related to
102	the source strength and frequently happens under stagnant weather conditions. Recently,
103	it has been studies shown showed that the rapid growth of secondary aerosols could
104	lead to an severe haze event in China under certain meteorological conditions (Wang et
105	al., 2016;Sun et al., 2014;Quan et al., 2013).
106	In the rural area of NCP, biomass burning for domestic cooking and heating, and
107	agricultural waste disposal is an important source of atmospheric PM (Wang et al.,
107 108	agricultural waste disposal is an important source of atmospheric PM (Wang et al., 2009b;Li et al., 2010;Zhang et al., 2016). Particularly, the open-field burning is still a
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116	erop residual on ozone, CO, black carbon (BC) and organic carbon (OC)
117	concentrations in NCP. Moreover, both off-line (Fu et al., 2012;Wang et al.,
118	2009b;Wang et al., 2011) and on-line (Sun et al., 2016) observations indicated that the
119	intensive emission from wheat straw burning in the region could change the molecular
120	distribution of organic aerosols of <u>in</u> the downwind urban or mountain areas.
121	During June 10th to 25th of 2013, we conducted a continuous sampling campaign
122	at a rural site in the northern part of NCP. PM <sub>2.5</sub> filter samples were collected with a
123	3-hour time resolution and determined for more than 100 organic compounds
124	including aliphatic lipids, sugar compounds, hopanes, polycyclic aromatic
125	hydrocarbons (PAHs), phthalate esters, and secondary oxidation products. The first
126	objective of this study was to get an overall understanding of temporal/diurnal
127	variation and molecular distribution of summertime OA in the rural region. The
128	second objective was to compare the results in two representative periods to
129	investigate the influence of regional field burning of wheat straw on the molecular
130	characteristics of organic aerosols.
131	2. Experimental section
132	2.1 Sample collection
133	The sampling was performed at the Integrated Ecological-Meteorological
134	Observation and Experiment Station of Chinese Academy of Meteorological Sciences
135	(39°08' N, 115°40' E, 15.2 m a <u>.s.l.</u> ), which is located in a rural area of Gucheng,
136	Hebei Province. Detailed information of the station and sampling campaign was
137	described in Li et al. (2018). Briefly, time-resolved (06:00-09:00, 09:00-12:00,

138	12:00–15:00, 15:00–18:00, 18:00–21:00, 21:00–24:00, 00:00–03:00, and 03:00–	
139	06:00, Beijing time) $PM_{2.5}$ samples were collected on the rooftop (about 10 m above	
140	the ground) of a three-story building on the campus of the Gucheng station. The	
141	sampling was conducted by using a high volume (1.13 m <sup>3</sup> min <sup>-1</sup> ) sampler (Anderson)	
142	with a $PM_{2.5}$ inlet from June 10 <sup>th</sup> to 25 <sup>th</sup> , 2013. This period was chosen, because open-	
143	field burning of wheat straw in NCP mainly occur in the mid of June. All samples	
144	were collected onto pre-baked (450 °C, 6-8 hr) quartz fiber filters. Field blank samples	
145	were also collected by mounting blank filters onto the sampler for about 15 min	
146	without pumping any air. After sampling, the sample filter was individually sealed in	
147	aluminum foil bags and stored in a freezer (-20 °C) prior to analysis.	
148	2.2 Organic compounds determination	
149	A size of 12.5-25 cm <sup>2</sup> of the filter sample was cut and extracted with a mixture of	
150	dichloromethane and methanol (2:1, v/v) under ultrasonication. The extracts were	
151	concentrated using a rotary evaporator under vacuum conditions and then blow down	
152	to dryness using pure nitrogen. After reaction with N,O-bis-(trimethylsilyl)	
153	trifluoroacetamide (BSTFA) at 70 $^\circ\!\!\mathbb{C}$ for 3 hrs., the derivatives were determined	
154	using gas chromatography/electron ionization mass spectrometry (GC/EI-MS) (Li et	
155	al., 2013b).	
156	GC/EI-MS analysis of the derivatized fraction was performed using an Agilent	
157	7890A GC coupled with an Agilent 5975C MSD. The GC separation was carried out	
158		
	on a DB-5MS fused silica capillary column with the GC oven temperature	

160	$min^{-1}$ with a final isothermal hold at 300 $^\circ\!\!C$ $$ for 16 min. The sample was injected in a
161	splitless mode at an injector temperature of 280 $^\circ\!\mathrm{C},$ and scanned from 50 to 650
162	Daltons using electron impact (EI) mode at 70 eV.
163	GC/EI-MS response factors of all the target compounds were determined using
164	authentic standards except several isoprene-derived SOA tracers. Response factors of
165	isoprene-derived SOA tracers were substituted by those of related surrogated
166	standards, which were described in Li et al. (2018). No significant contamination
167	(<5% of those in the samples) was found in the blanks. Recoveries of all the target
168	compounds ranged from 80% to 120%. Data presented were corrected for the field
169	blanks but not corrected for the recoveries.
170	2.3 OC, EC, and WSOC analysis
171	OC (organic carbon) and EC (elemental carbon) were analyzed using DRI Model
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171 172 173 174	OC (organic carbon) and EC (elemental carbon) were analyzed using DRI Model 2001 Carbon Analyzer following the Interagency Monitoring of Protected Visual Environments (IMPROVE) thermal/optical reflectance (TOR) protocol. A size of 0.526 cm <sup>2</sup> sample filter was placed in a quartz boat inside the analyzer and stepwise
171 172 173 174 175	OC (organic carbon) and EC (elemental carbon) were analyzed using DRI Model 2001 Carbon Analyzer following the Interagency Monitoring of Protected Visual Environments (IMPROVE) thermal/optical reflectance (TOR) protocol. A size of 0.526 cm <sup>2</sup> sample filter was placed in a quartz boat inside the analyzer and stepwise heated to temperatures of 140 °C (OC1), 280 °C (OC2), 480 °C (OC3), and
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171 172 173 174 175 176 177	OC (organic carbon) and EC (elemental carbon) were analyzed using DRI Model 2001 Carbon Analyzer following the Interagency Monitoring of Protected Visual Environments (IMPROVE) thermal/optical reflectance (TOR) protocol. A size of 0.526 cm <sup>2</sup> sample filter was placed in a quartz boat inside the analyzer and stepwise heated to temperatures of 140 °C (OC1), 280 °C (OC2), 480 °C (OC3), and 580 °C (OC4) in a non-oxidizing helium (He) atmosphere, and 580 °C (EC1), 740 °C (EC2), and 840 °C (EC3) in an oxidizing atmosphere of 2% oxygen in
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171 172 173 174 175 176 177 178 179 180	OC (organic carbon) and EC (elemental carbon) were analyzed using DRI Model 2001 Carbon Analyzer following the Interagency Monitoring of Protected Visual Environments (IMPROVE) thermal/optical reflectance (TOR) protocol. A size of 0.526 cm <sup>2</sup> sample filter was placed in a quartz boat inside the analyzer and stepwise heated to temperatures of 140 °C (OC1), 280 °C (OC2), 480 °C (OC3), and 580 °C (OC4) in a non-oxidizing helium (He) atmosphere, and 580 °C (EC1), 740 °C (EC2), and 840 °C (EC3) in an oxidizing atmosphere of 2% oxygen in helium. Pyrolyzed carbon (PC) is determined by reflectance and transmittance of 633 nm light. One sample was randomly selected from every 10 samples and re-analyzed. Differences determined from the replicate analyses were <5% for TC, and <10% for

182	Another aliquot of filter sample was extracted with organic-free Milli-Q water
183	under ultrasonication (15 min each, repeated 3 times) and filtered through a PTFE
184	filter to remove any particles and filter debris. Then the water-extract was analyzed
185	for water-soluble organic carbon (WSOC) using a TOC analyzer (TOC-L CPH,
186	Shimadzu, Japan). The difference between OC and WSOC was considered as water-
187	insoluble OC (WIOC). All carbonaceous components data reported here were
188	corrected by the field blanks.

189 **3. Results and discussion** 

#### 190 **3.1 Fire spots and air masses**

At present, open-field burning is still a common activity for disposal of crop 191 residue in the rural area of the North China Plain, especially during wheat harvest period 192 from the end of May to the middle of June (Fu et al., 2012). These extensive emissions 193 from regional biomass burning in the provinces of Anhui, Jiangsu, Shandong, Henan 194 195 and Hebei in NCP can cause severe air pollution on a local and regional scale. In our previous study, the fire spots in NCP during the sampling period were provided based 196 on the NASA satellite observation (https://firms.modaps.eosdis.nasa.gov/firemap/). 197 Combining 198 with information on air mass back-trajectories (http://ready.arl.noaa.gov/HYSPLIT.php), the sampling period was divided into two 199 200 sections: (1) June 10-18, when air masses were mainly transported via long distances from the southeast part of NCP, where intensive emissions from wheat straw burning 201 202 occurred; (2) June 19-25, when air masses were mostly influenced by local emissions 203 and regional emission from biomass burning decreased dramatically (Li et al., 2018).

204	In this study, we further selected two representative periods to estimate the contribution
205	of regional biomass burning. Period 1 (P1) designates 13th Jun 21:00 pm to 16th Jun
206	15:00 pm, during which air masses were influenced by intensive biomass burning and
207	transported uniformly from the southeast part of NCP (Figure 1 a and b, and Figure S1).
208	Period 2 (P2) designates 22 <sup>nd</sup> Jun 12:00 pm to 24 <sup>th</sup> Jun 06:00 am, during which fire
209	spots in the regions were relatively scarce and air masses came predominantly from the
210	surrounding areas of the sampling site (Figure 1 c and d). In addition, there were several
211	intermittent rainfalls during June $20^{\text{th}}_{\underline{a}}$ , which are favorable for wet deposition of
212	atmospheric pollutants. Thus, aerosols collected in P2 are well representative of local
213	fresh emission. It is worthwhile to note that the two samples collected during 21st June
214	18:00-24:00 pm were excluded from P2, because they were highly affected by near-site
215	biomass burning emission (detailed information is provided in Section 3.3).
216	3.2 Concentrations of PM <sub>2.5</sub> , OC, EC, WSOC and WIOC
217	Concentrations of PM <sub>2.5</sub> and carbonaceous components are presented in Table 1.
218	$PM_{2.5}$ concentrations ranged from 21 to 395 $\mu g$ m $^{-3}$ with a mean value at 159 $\pm$ 89 $\mu g$
219	$m^{-3}$ during the whole sampling period. As shown in Figure 2, $PM_{2.5}$ concentrations in
220	P1 (average $\pm 1\sigma = 231 \pm 89 \ \mu g \ m^{-3}$ ) increased continuously from around 150 $\mu g$
221	$m^{-3}$ to higher than 300 $\mu g\ m^{-3},$ indicating the occurrence of a severe air pollution
222	episode. In contrast, PM_{2.5} concentration during P2 is as low as $43\pm14~\mu g~m^{-3}.$
223	Similarly, the average concentration of OC is 29.4 $\pm$ 7.8 $\mu g$ m $^{-3}$ in P1, which is more
224	than 5 time higher than that in P2 (5.5 $\pm$ 1.7 $\mu g$ m^-3) . EC concentrations also
225	decreased dramatically from P1 (12.1 $\pm4.0~\mu g~m^{-3})$ to P2 (1.5 $\pm1.5~\mu g~m^{-3}).$ The

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226	average OC/EC ratio is $3.0\pm0.9$ for the whole sampling period, but the ratio in P2
227	$(3.8\pm1.0)$ was is higher in P2 $(3.8\pm1.0)$ than that in P1 $(2.5\pm0.4)$ , mainly due to the
228	high SOA formation activities in the rural areas of NCP in summer. It's worthwhile to
229	note that the average OC/EC ratio during the BB-influenced P1 is much lower than
230	the results reported for wheat straw burning in combustion chamber (12.9 $\pm$ 2.1) (Tian
231	et al., 2017) or and residential stove (6.3-11.1) (Li et al., 2009). The first reason is that
232	fossil fuel burning (such as coal burning and vehicle exhaust) with lower OC/EC ratio
233	(Tian et al., 2017) are still important sources in the region, which are also discussed in
234	Section 3.4. On the other hand, this phenomenon may also be related to different
235	combustion conditions of agricultural residuals in the open field. Li et al. (2009)
236	found that the flaming fire from biomass burning would result in more EC emission
237	and lower OC/EC ratio compared to smoldering fire. Actually, Hays et al. (2005)
238	obtained similar low OC/EC ratio of 2.4 for open wheat straw burning smoke from
239	Washington, US. Thus, the lower ratio of OC/EC observed in this work may indicate
240	that wheat straw combustion in NCP during P1 mainly occurred mainly in the flaming
241	phase.
242	As shown in Figure 2 and 3, the concentrations of WSOC show a consistent
243	temporal variation as those of OC ( $R^2=0.82$ ), highlighting the fact that WSOC is an
244	important fraction of OC in this region. In addition, the average ratio of WSOC/OC is
245	higher during P1 ( $0.62\pm0.16$ ) than during P2 ( $0.48\pm0.12$ ), mainly due to enhanced
246	emissions of water-soluble organic compounds (such as sugars, fatty alcohols/acids)

247 from biomass burning during P1. Due to the removal effect of the intermittent raining,

than those in P1 (10.3  $\pm$  4.4 µg m<sup>-3</sup>). 249 250 The diurnal variation profiles of EC/OC and WSOC/OC are shown in Figure 4. 251 EC/OC is generally lower in daytime and the lowest value occurred during 12:00-15:00 pm, mainly due to enhanced daytime formation of SOC. Previous studies have 252 253 shown that SOAsecondary organic aerosols are mainly composed of water-soluble 254 compounds, e.g., polyacids/polyalcohols and phenols (Kondo et al., 2007; Wang et al., 255 2009a). However, these compounds can come be emitted from primary emissions as well, especially from biomass burning (Shen et al., 2017;Fu et al., 2012). In this study, 256 the WSOC/OC presents lower value during daytime, especially in the afternoon when 257 photo-chemical oxidation is favorable. In addition, the diurnal variation pattern of 258 259 WSOC/OC is similar to that of levoglucosan/OC. Given that levoglucosan is a marker 260 of biomass burning emissions (Simoneit et al., 1999;Simoneit et al., 2004a), and many 261 kinds of SOA could be produced in the biomass burning plumes during the long-range 262 transport (detailed discussions are given in Section 3.3), <u>these These</u> results indicate 263 that particulate WSOC in the region is mostly derived from biomass burning activities 264 including direct emission and secondary oxidation.

concentrations of water-insoluble OC in P2 ( $3.0 \pm 1.3 \ \mu g \ m^{-3}$ ) are also much lower

## 265 3.3 Organic molecular composition

248

More than 100 organic species were detected in the aerosol samples, and their concentrations are shown in Table 2 and S1. In this study, these organic compositions are grouped into 10 compound classes based on functional groups and sources. Total concentrations of the measured organics ranged from 177 to 6248 ng m<sup>-3</sup> (average =

270	$1806\pm1308$ ng m^-3) during the whole sampling period with the predominance of
271	sugar compounds, followed by fatty acids and fatty alcohols. The temporal variation
272	profiles of the determined organic groups are shown in Figure 5. Nearly all the
273	measured organic species, especially <i>n</i> -alkanes, fatty acids, fatty alcohols, sugar
274	compounds, and PAHs, show much higher concentrations in P1 than in P2 (Figure
275	S2), indicating an important influence of regional biomass burning on airborne
276	organic aerosols in NCP.
277	3.3.1 Biomass-burning tracers
278	As described in Section 3.1, intensive emissions of open biomass burning were
279	observed in the southern part of NCP during June 13-16 (P1), which is an important
280	reason for the severe regional air pollution during this period. Levoglucosan, which is
281	produced in large quantities during pyrolysis of cellulose, is a key tracer for biomass
282	burning emissions (Simoneit, 2002;Simoneit et al., 1999). As shown in Table 2,
283	levoglucosan is the most abundant single compound in the whole sampling period,
284	ranged from 5.6 to 1447 ng m <sup>-3</sup> with a mean concentration of $240 \pm 288$ ng m <sup>-3</sup> .
285	Levoglucosan shows good positive correlations with both OC ( $R^2=0.61$ ) and WSOC
286	$(R^2=0.65)$ (Figure 3), confirming that biomass burning is an important source of both
287	aerosol OC and WSOC in the rural region of NCP during the sampling period. As
288	clearly shown in Figure 6, the concentrations of levoglucosan present a continual
289	increasing trend during P1 with a mean value of $404 \pm 344$ ng m <sup>-3</sup> . However, the tracer
290	presents very low concentrations (11-123 ng m <sup>-3</sup> ) for the most of the time during Jun
291	20-22. Interestingly, the concentration of levoglucosan suddenly increased by more

292	than 10 times at $21^{st}$ Jun 18:00 pm to approximately 1200 ng m <sup>-3</sup> within less than 3
293	hours and then decreased to its beginning concentration (less than 100 ng m <sup>-3</sup> ) within
294	6 hours (2 samples) afterwards. The concentrations of OC, WSOC and EC also
295	showed obvious peaks during this event. However, based on analyses of back-
296	trajectories (Figure 1c) and wind conditions (Figure S1), we didn't findno significant
297	change of air mass origins was observed. Also, not all organic markers showeded
298	similar variations as to levoglucosan. For example, the concentrations of PAHs,
299	hopanes, and phthalate esters changed little during this event. Thus, it is plausible to
300	conclude that this variation was caused by emissions from biomass burning activities
301	nearby the sampling site. For this reason, the data of the 2 samples were excluded
302	from P2.
303	The two isomers of levoglucosan, galactosan and mannosan, are also produced
304	by the pyrolysis of cellulose/hemicelluloses (Simoneit, 2002), and thus also
305	considered as important markers of biomass burning. Similar to levoglucosan, the

concentrations of these two anhydrosugars in P1 are 5-6 times higher than those in P2.
The isomeric ratios of levoglucosan to other anhydrosugars are considered as good
indicators of biomass burning. Fabbri et al. (2009) compared the concentrations of the
three anhydrosaccharides in the smokes from different fuel types, and proposed that
levoglucan/(galactosan+mannosan) (L/G+M) and levoglucan/mannosan (L/M) values
range in 0.2-18 and 0.23-33 for various source tests for biomass burning as compared
to the average of 54 and 54 for lignites. As shown in Table 2, average ratios of L/G+M

and L/M during P1 (10.1 $\pm$ 3.41 and 6.77 $\pm$ 1.97, respectively) and P2 (29.7 $\pm$ 12.2 and

 $18.0 \pm 4.28$ ) suggest that biomass burning is always the dominated contributor for

these compounds in the rural area of NCP in summer.

## 316 **3.3.2** Aliphatic lipid composition

317	The average concentration of all the <i>n</i> -alkanes ( $C_{18}$ - $C_{36}$ ) measured in this study
318	is 207 $\pm$ 149 ng m <sup>-3</sup> with the most abundant individual compound being nonacosane
319	(C <sub>max</sub> =C <sub>29</sub> H <sub>60</sub> ) (Table S1). <i>n</i> -Alkanes derived from terrestrial plants are dominated by
320	high molecular weight species (HMW, carbon number >25) with an odd number
321	preference. In contrast, fossil fuel derived <i>n</i> -alkanes do not have odd/even number
322	preference (Rogge et al., 1993a;Simoneit et al., 2004b). In general, <i>n</i> -alkanes with a
323	carbon preference index (CPI, odd/even) more than 5 are considered as plant wax,
324	while those with a CPI nearly unity are mostly derived from fossil fuel combustion
325	(Rogge et al., 1993a, b). In this study, the mean value of CPI is $2.47 \pm 1.12$ , indicating
326	that both fossil fuel and plant wax contributed to <i>n</i> -alkanes in the rural areas of NCP
327	in summer. However, <i>n</i> -alkanes showed a stronger odd/even carbon number
328	predominance in P1 (CPI=2.85) than in P2 (CPI=1.64). In addition, all the low
329	molecular weight <i>n</i> -alkanes (LMW, carbon number <25) presented a higher
330	contribution to total <i>n</i> -alkanes in P2 than in P1 (Figure 7 a and d). These results
331	demonstrate that plant waxes from biomass burning emissions made a bigger
332	contribution to organic aerosols in the sampling region during P1.
333	A homologous series of 19 saturated fatty acids ( $C_{12:0}$ – $C_{32:0}$ ) and 3 unsaturated
334	fatty acids (C <sub>16:1</sub> , C <sub>18:1</sub> , and C <sub>18:2</sub> ) were detected in the samples (Table S1), and their
335	total concentration was-is $514 \pm 384$ ng m <sup>-3</sup> during the whole period. A strong even

1	
336	carbon number predominance was is observed with $C_{\text{max}}$ at $C_{28:0}$ and $C_{16:0}$ (Table S1).
337	Higher molecular weight (HMW) fatty acids ( $\geq C_{20}$ ) are derived from terrestrial plant
338	waxes, while LMW fatty acids ( $\leq C_{19}$ ) have multiple sources such as vascular plants,
339	microbes and marine phytoplankton as well as kitchen emissions (Rogge et al.,
340	1993b;Kawamura et al., 2003). The total concentrations of fatty acids presented
341	similar temporal variation to levoglucosan with a robust linear correlation (R <sup>2</sup> =0.72)
342	(Figure 8a), indicating that fatty acids are mostly affected by biomass burning
343	emission during the whole sampling period. Still, there are some evidences that
344	regional emission from wheat straw burning significantly affected the distribution of
345	fatty acids in the aerosols of Gucheng during P1. Firstly, the total concentrations of
346	fatty acids in P1 (900 $\pm$ 358 ng m <sup>-3</sup> ) are more than 6 times higher than those in P2
347	(145 $\pm$ 48 ng m <sup>-3</sup> ). Secondly, the concentrations and relative contributions of HMW
348	fatty acids $(C_{20:0}-C_{32:0})$ are much higher in P1 than in P2, similar to the results of <i>n</i> -
349	alkanes. In addition, the mean value of CPI of HMW fatty acids in P1 (4.21 $\pm$ 1.14) is
350	also higher than that in P2 $(3.50 \pm 1.64)$ .
351	Fatty alcohols in the range of $C_{22}$ – $C_{30}$ were detected for the PM <sub>2.5</sub> samples with a
352	mean concentration of 193 $\pm$ 187 ng m <sup>-3</sup> (Table 2 and S1) during the whole sampling
353	period. Their distributions are characterized by even carbon number predominance
354	with a maximum at $C_{28}$ (Figure 7c and f). The total concentration of fatty alcohols
355	strongly correlate <u>correlates</u> d-with levoglucosan (R <sup>2</sup> =0.73) (Figure 8b), suggesting

alcohols ( $\geq C_{20}$ ) abundantly present in higher plants and loess deposits (Wang and

356

that they can also  $\underline{be}$  emitted from biomass burning. It is reasonable since HMW fatty

Kawamura, 2005). Similar to fatty acids, nearly 10 times higher concentration of fatty alcohols <u>was-is</u> observed in P1 ( $322\pm151$  ng m<sup>-3</sup>) compared with those in P2 ( $34\pm23$ ng m<sup>-3</sup>).

361 3.3.3 Primary saccharides

362 In addition to the three anhydrosugars, 4 primary sugars (fructose, glucose, sucrose and trehalose) and 3 sugar alcohols (arabitol, mannitol and inositol) were detected in 363 the samples. Primary saccharides have been used as biomarkers for primary biota 364 365 emissions (Wang et al., 2011). Their mean concentrations ranged from 3.6 to 49 ng m<sup>-</sup> <sup>3</sup> during the whole sampling period. In this study, concentrations of fructose, sucrose 366 367 and trehalose in P1 were-are 7-10 times higher than those in P2 (Table S1). They well correlated with levoglucosan (R<sup>2</sup>=0.47-0.62, Figure S3) during P1, in contrast to P2, 368 369 during which no relationships were are found between them. These results indicated 370 that these primary sugars were are also affected by open-field emissions of biomass 371 burning during P1. Sugar alcohols, mainly arabitol and mannitol, are abundant in 372 airborne fungal spores (Graham et al., 2002). Some studies suggested that biomass 373 burning activities can enhance the emission of sugar alcohols at a certain level (Engling et al., 2009;Fu et al., 2012;Yang et al., 2012). However, no significant relationship 374  $(R^2 < 0.10)$  can be found between these sugar alcohols and levoglucosan even in P1, 375 376 indicating the negligible contribution of biomass burning to the tracers in this study.

#### 377 3.3.4 PAHs, Hopanes and Phthalates

As shown in Figure 5, the temporal variation of PAHs, hopanes, and phthalate esters were clearly different from those of the molecular tracers for biomass burning, 380 especially in P1. In contrast to the continuous increase of sugars, fatty acids, fatty 381 alcohols, and n-alkanes during P1, the concentrations of PAHs, hopanes, and phthalate esters showed obvious day-night variations, indicating that biomass burning activities 382 383 contributed little on these species. Phthalates are widely used as plasticizers in synthetic polymers or softeners in polyvinylchlorides (PVC) (Simoneit et al., 2004b) and can be 384 directly emitted from the matrix into the air as they are not chemically bonded with the 385 matrix. Six phthalate esters were detected in the sampling aerosols, i.e., dimethyl 386 (DMP), diethyl (DEP), diisobutyl (DiBP), butyl isobutyl (BiBP), di-n-butyl (DnBP), 387 and bis(2-ethylhexyl) (BEHP) phthalates (Table S1). The concentrations of total 388 detected phthalate esters in P1 ( $112\pm33$  ng m<sup>-3</sup>) are around 2 times only higher than 389 390 those in P2 ( $51 \pm 18$  ng m<sup>-3</sup>). Hopanes are abundant in coal and crude oils, and enriched in lubricant oil fraction (Oros and Simoneit, 2000;Kawamura et al., 1995).They can be 391 392 emitted to the atmosphere from coal burning and/or internal combustion of fuel in engines. Only two dominant hopanes,  $17\alpha(H)$ ,  $21\beta(H)$ -30-norhopane(C<sub>29 $\alpha\beta$ </sub>) and 393 394  $17\alpha(H), 21\beta(H)$ -hopane(C<sub>30\alpha\beta</sub>), were detected in all of the samples in this study. Their average concentration in P1 (4.40 $\pm$ 2.48 ng m<sup>-3</sup>) is ~ 2.5 times of that in P2 (1.81 $\pm$ 395 0.31 ng m<sup>-3</sup>). Considering the much higher concentrations of levoglucosan in P1 (on 396 397 average ~ 8 times higher than P2), these results again confirmed a limited influence of 398 biomass burning on concentrations of phthalate esters and hopanes in the aerosols in the rural region. Thus, there were no significant concentration changes of the two 399 species at 21st Jun 18:00-24:00 pm, when the air masses were highly affected by nearby 400 401 biomass burning activities.

402	PAHs are the products of incomplete combustion of carbon-containing materials
403	and are of high toxicity and carcinogenicity (Halek et al., 2008;Sultan et al., 2001).
404	Previous studies indicated that PAHs are mainly emitted from coal burning and vehicle
405	exhaust in most areas of China (Wang et al., 2006a). However, it has been reported that
406	combustion of biomass materials can also contribute to the PAHs in the atmosphere
407	(Simoneit, 2002;Ge et al., 2012;Young et al., 2016). In this study, PAH has no
408	significant correlation with levoglucosan during the whole sampling period. Yet the
409	concentrations of total PAHs in P1 (18.6 $\pm$ 11 ng m <sup>-3</sup> ) are nearly 8 times higher than
410	those in P2 ( $2.3 \pm 1.0$ ng m <sup>-3</sup> ). These results <u>mean suggest</u> that although the emission of
411	biomass burning is not the most important source for PAHs during the entire period, the
412	intensive regional burning of wheat straw in P1 can also enhance the PAHs
413	concentration in the atmosphere of Gucheng.

As shown in Figure 9, all the primary aerosol markers mentioned above showed lower concentrations in daytime with lowest concentrations at afternoon (12:00-18:00 pm), in consistent with the favorable dispersion conditions caused by high temperature and planetary boundary layer (PBL) height. However, the day-night variation of PAHs, hopanes, and phthalate esters are more obvious than other species, again confirming the lower contribution of biomass burning to these organic compositions.

### 420 3.3.5 Secondary organic aerosols (SOA) tracers

Eight compounds were identified as isoprene oxidation products in the PM<sub>2.5</sub> samples, including two methyltetrahydrofuran diols, three C<sub>5</sub>-alkene triols, two 2methyltetrols, and 2-methylglyceric acid (Table S1). Detailed information about

formation and contribution of these compositions were discussed in our previous paper (Li et al., 2018). The concentrations of total detected isoprene-derived products are 112  $\pm$  86 ng m<sup>-3</sup>, with much higher concentration in P1 (209 $\pm$ 105 ng m<sup>-3</sup>) than in P2 (57  $\pm$  29 ng m<sup>-3</sup>).

428 cis-Pinonic acid (PNA), pinic acid (PA), 3-hydroxyglutaric acid (HGA) and 3methyl-1,2,3-butanetricarboxylic acid (MBTCA) were detected as tracers for  $\alpha$ -/ $\beta$ -429 430 pinene oxidation in this study, and their concentrations are shown in Table S1. The concentration of total detected  $\alpha$ -/ $\beta$ -pinene oxidation tracers is 66±31 ng m<sup>-3</sup>, with 431 MBTCA  $(31 \pm 14 \text{ ng m}^{-3})$  being the major compound during the whole sampling period. 432 PNA and PA are considered as first-generation products of  $\alpha$ -/ $\beta$ -pinene oxidation. They 433 can be produced by further oxidation of carbonyl-substituted Criegee intermediates 434 435 formed by α-pinene ozonolysis (Jenkin et al., 2000;Ma et al., 2008), or by OH oxidation of α-pinene under NO<sub>x</sub> free conditions (Eddingsaas et al., 2012;Xuan et al., 2015). The 436 formation of 3-HGA is supposed to be based on a ring opening mechanism and may be 437 438 related to a heterogeneous reaction of these monoterpenes with irradiation in the presence of NOx (Jaoui et al., 2005; Claeys et al., 2007). As shown in Figure 6b-d, PNA, 439 440 PA and HGA present similar temporal variations and correlated well each other (R<sup>2</sup>=0.48-0.76, Figure S4). The formation of MBTCA is explained by further 441 photodegradation of cis-pinonic acid and pinic acid with OH radical (Müller et al., 442 443 2012;Szmigielski et al., 2007). As a later-generation oxidation products, MBTCA 444 showed an obviously different temporal variation profile than those of PNA and PA, 445 and had-has no significant increase during P1. In addition, the concertations of PNA,

446	PA and HGA in P1 are 2-8 times higher than those in P2. However, the concentrations
447	of MBTCA in the two periods are comparable. These results are consistent with the
448	longer time scales of formation pathway, lower volatility and longer lifetime of
449	MBTCA in the atmosphere compared to the first-generation products of $\alpha$ -/ $\beta$ -pinene
450	oxidation. $\beta$ -Caryophyllinic acid, formed either by ozonolysis or photo-oxidation of $\beta$ -
451	caryophyllene (a sesquiterpene) (Jaoui et al., 2007), was also determined in this study,
452	and its concentration ranged from 0.49 to 78 ng m <sup>-3</sup> (Ave. 17 $\pm$ 17 ng m <sup>-3</sup> ). The mean
453	concentration of $\beta$ -caryophyllinic acid in P1 is 35±21 ng m <sup>-3</sup> , being 5 times higher than
454	that in P2 (4.1±1.2 ng m <sup>-3</sup> ).

455 Undoubtedly, the combustion of biomass materials can release a large amount of 456 volatile organic compound VOCs, including isoprene and terpenoids (Andreae and Merlet, 2001). As shown in Figure 5 and 6, the total biogenic SOA tracers, the sum of 457 458 detected tracers of isoprene,  $\alpha$ -/ $\beta$ -pinene, and  $\beta$ -caryophyllene derived SOA, showed a 459 similar temporal variation pattern as levoglucosan with a moderate correlation (R<sup>2</sup>=0.56, 460 Figure S5a). Specifically, levoglucosan showsed strong linearly correlations with 3-461 hydroxyglutaric acid (R<sup>2</sup>=0.72) (Figure 8c) and β-caryophyllinic acid (R<sup>2</sup>=0.80) (Figure 8d), indicating a significant contribution of biomass burning emissions to the formation 462 463 of SOA derived from mono- and sesqui- terpene oxidation. In our previous paper (Li et 464 al., 2018), we discussed the different diurnal variations of isoprene-derived SOA tracers. In this study, the diurnal variations of other SOA tracers are shown in Figure 10. All the 465 466 SOA tracers presented weaker day-night variations compared to primary organic 467 aerosol markers, because of the competition between the enhanced daytime formation

by photoxidation and the nighttime accumulation associated with a low PBL. Yet, there
are some differences between these SOA tracers. For example, PNA and PA presented
lowest concentrations in the afternoon (12:00-18:00 pm) due to their relatively high
volatilities, which is unfavorable for gas-to-particle phase partitioning. However, the
later-generation product of PNA and PA, i.e., the less volatile MBTCA, continuously
increasesd during the daytime.

474 Two classes of aromatic SOA markers, phthalic acids and 2,3-dihydroxy-4oxopentanoic acid (DHOPA), were detected in the samples as well. Phthalic acids are 475 believed to be produced by the oxidation of naphthalene and other PAHs (Kawamura 476 et al., 2005;Kawamura and Ikushima, 1993;Kanakidou et al., 2005). The mean 477 478 concentrations of total phthalic acids in the whole sampling period ranged from 17 to 487 ng m<sup>-3</sup> with a mean value of 155±94 ng m<sup>-3</sup>. Their different temporal variation 479 patterns than levoglucosan suggest that biomass burning emission contributes little to 480 phthalic acids formation in the region. The DHOPA was considered to be a tracer 481 compound for toluene-derived SOA (Kleindienst et al., 2004). DHOPA presentsed a 482 similar temporal variation and <u>a</u> moderate correlation to-with levoglusoan (R<sup>2</sup>=0.51, 483 484 Figure S5b), indicating a certain contribution of biomass burning. Similar to MBTCA, 485 the volatility of DHOPA is quite low, and thus mainly exists in the particle phase at field 486 temperature (Ding et al., 2017). Thus, DHOPA showsed a similar diurnal variation to 487 MBTCA, with higher concentrations during daytime.

488 **3.4 Assessment of source contributions** 

489

In order to investigate the differences in organic aerosol sources between the two

representative periods, we classified all the measured organic compounds into seven 490 different sources: (a) "plant emission" represented by higher plant wax n-alkanes, 491 HMW fatty acids and fatty alcohols ( $\geq C_{20}$ ); (b) "fossil fuel combustion" mainly 492 493 represented by fossil fuel derived n-alkanes, hopanes, and PAHs; (c) "biomass burning" 494 represented by levoglucosan and its isomers; (d) "marine/microbial source" represented by LMW fatty acids (<C<sub>20</sub>); (e) "soil/fungal spore/pollen" represented by primary 495 saccharides and sugar alcohols; (f) "plastic emission" represented by phthalate esters; 496 and (g) "secondary oxidation" represented by biogenic SOA tracers, DHOPA, and 497 phthalic acids. The concentrations of individual classes and their contributions to OC 498 content during P1 and P2 are summarized in Figure 11. Plant emission-derived 499 compounds accounted for a larger fraction of PM2.5 OC during P1 than during P2 (mean 500 501 fractions of  $28.7 \pm 9.3\%$  in P1 vs.  $16.5 \pm 7.2\%$  in P2). The average faction of biomass burning-derived organics in P1 ( $6.0 \pm 3.9\%$ ) was is also higher in P1-than that in P2 502  $(6.0 \pm 3.9\% \text{ vs.} (4.6 \pm 2.1\%)$ , so do organics derived from soil/fungal spore/pollen. 503 504 However, organic molecules from the other 4 sources presented a higher contribution to OC in P2 than in P1. 505

Since organic compounds in the PM<sub>2.5</sub> samples cannot be completely determined and some of them are of different sources, thus the above classification based on the measured compounds could result in certain uncertainty in assessing source contributions (Simoneit et al., 2004b). Here, we further used a tracer-based source apportionment method to quantitatively estimate the contributions of primary and secondary sources to the fine particulate OC at the rural site. As described above, two samples collected at 21<sup>st</sup> Jun 18:00-24:00 pm were considered to be highly affected by the direct emission from biomass burning nearby the sampling site. Thus, the average OC/levoglucosan ratio in the smoke of biomass burning  $\left(\left(\frac{OC}{Levo}\right)_{BB}\right)$  can be estimated by using the following equation:

516 
$$\left(\frac{OC}{Levo}\right)_{BB} = \frac{OC_n - \frac{1}{2}(OC_{before} + OC_{after})}{Levo_n - \frac{1}{2}(Levo_{before} + Levo_{Levo})}$$
(E1)

where  $OC_n$  and  $Levo_n$  are the average concentrations of OC and levoglucosan in 517 the two  $PM_{2.5}$  samples affected by the nearby sources.  $OC_{before}$  and  $Levo_{before}$  are the 518 519 concentrations of OC and levoglucosan in the samples collected before the event, while OCafter and Levoafter are the concentrations of OC and levoglucosan in the samples 520 521 collected after the event. The mean values in the "before" and the "after" samples were subtracted to minimize the influence of local background contribution. The calculated 522  $\left(\frac{OC}{Levo}\right)_{BB}$  in this study is 18.7, which is somewhat higher than the average value of 12.1 523 measured in PM2.5 aerosols emitted from the burning of three types of cereal straws 524 (i.e., wheat, maize, and rice) in China (Zhang et al., 2007b). This difference can be 525 526 attributed to the differences of burning conditions. For other sources, the measured concentrations of mannitol were used to calculate the contributions of fungal spores to 527 528 OC (Bauer et al., 2008), and SOA tracers were used to estimate the SOC formed from the oxidation of isoprene,  $\alpha$ -/ $\beta$ -pinene,  $\beta$ -caryophyllene, and toluene (Kleindienst et al., 529 2007). Also, these tracer-based approaches tend to have large uncertainties, especially 530 for SOC estimation (Li et al., 2013a). However, our results are still meaningful to 531 understand the relative abundances of organic aerosols from these sources in different 532 533 periods.

534	As shown in Figure 12, biomass-burning derived OC, ranging from $0.11-27.5 \ \mu gC$
535	$m^{-3},$ is the dominant source, which accounts for 1.16-74.8% (ave. 22.6%) of OC in the
536	aerosols of the rural region during the whole sampling period. Fungal-spore derived OC
537	(0.003-5.12 $\mu gC~m^{-3})$ is a minor source, only accounting for 0.43% (0.003-5.12%) of
538	OC. The contribution of total SOC derived from oxidation of isoprene, $\alpha\text{-/}\beta\text{-pinene},\beta\text{-}$
539	caryophyllene, and toluene to OC ranged from 5.90-34.1% with an average at of 16.7%.
540	Among the four SOC precursors, toluene-derived products accounted for 7.78% (2.06-
541	21.7%) of OC, being the most important SOC contributor. The relative abundances of
542	these sources showed clear temporal variations during the whole sampling period
543	(Figure 12). The contribution of biomass burning derived OC to total OC in P1 (27.6%)
544	iswas 1.7 times of that in P2 (17.1%) (Figure 13), further indicating the strong regional
545	impact of open-field wheat straw burning on the molecular compositions of organic
546	aerosols in the rural area of NCP. The contribution of SOC from oxidation of the four
547	VOCs increased slightly from P1 (16.3%) to P2 (21.1%). It should be noted that
548	biomass burning can also release a large amount of VOCs, which may produce more
549	secondary organic aerosols during the long-range transport. Thus, the impact of
550	intensive biomass burning in the southern region of NCP on organic aerosols in the
551	Gucheng area is likely even stronger than the estimation presented above with
552	implications for regional climate.

- 553 4. Summary and Conclusion
- 554 During the entire sampling period, OC and WSOC showed strong positive
- correlations with levoglucosan, and the diurnal variation of WSOC/OC was similar to

577	Author Contributions
576	
575	regional scale.
574	composition of aerosols, and thus affect atmospheric chemistry and climate on a
573	burning of biomass fuels can significantly influence the concentration and
572	formation of SOA in the atmosphere. Our results confirmed that intensive field
571	be underestimated since BB can also release a large amount of VOCs enhancing the
570	when local emissions were more dominant (P2). However, this contribution may even
569	period influenced by regional open-field biomass burning (P1) compared to the period
568	the contribution of BB to organic aerosols increased by more than 50% during the
567	The source distribution results derived using a tracer-based method demonstrated that
566	emission, and thus show <u>ed</u> little changes in concentrations between the two periods.
565	products, e.g., MBTCA in this study, are-were unlikely affected directly by BB
564	enhancing effect of BB emission on BSOA formation. Later-generation SOA
563	temporal variation and a moderate correlation with levoglucosan, demonstrating the
562	to these organic compositions. The total biogenic SOA tracers showed a similar
561	and diurnal variations from to levoglucosan because of the lower contribution of BB
560	of organic aerosols. PAHs, hopanes, and phthalate esters presented different temporal
559	enhancing-enhanced effect of open-field biomass burning on molecular composition
558	and CPI values of HMW n-alkanes, fatty acids and fatty alcohols in P1 indicated an
557	area of NCP are highly affected by direct emission of BB. Higher relative abundances
556	that of levoglucosan/OC, suggesting that summertime organic aerosols in the rural

578 G.H. Wang designed the experiment. G.H. Wang, T. Zhu and L.M. Zeng arranged the

sample collection. J.J. Li. and G.H. Wang collected the samples. J.J. Li, G.H. Wang, J.

580 Li, C. Wu and W.Q. Jiang analyzed the samples. J.J. Li, and G.H. Wang performed the

- data interpretation. J.J. Li, G.H. Wang and Q. Zhang wrote the paper.
- 582
- 583

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- and/or READY website (<u>http://www.ready.noaa.gov</u>) used in this publication.
- 593
- 594

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C .	Whole period (N=117)			Period	Period 1 (N=28)			Period 2 (N=13)		
Component	Range	Mean	SD	Range	Mean	SD	Range	Mean	SD	
PM <sub>2.5</sub> (µg m <sup>-3</sup> )	21~395	159	89	133~347	231	59	21~62	43	14	
OC (µg m <sup>-3</sup> )	1.7~45.7	17.3	11.1	13.8~44.4	29.4	7.8	3.6~8.8	5.5	1.7	
EC (µg m <sup>-3</sup> )	0.2~22.3	6.5	4.9	5.3~22.3	12.1	4.0	0.9~2.6	1.5	0.5	
WSOC (µg m <sup>-3</sup> )	0.7~33.0	11.5	8.2	5.3~33.0	19.1	8.3	1.2~4.2	2.6	0.8	
WIOC (µg m <sup>-3</sup> )	0.3~28.1	6.4	5.1	4.5~28.1	10.3	4.4	1.2~5.5	3.0	1.3	
OC/EC	1.2~7.6	3.0	0.9	1.9~3.2	2.5	0.4	2.5~5.7	3.8	1.0	
WSOC/OC	0.07~0.95	0.63	0.18	0.30~0.85	0.62	0.16	0.18~0.67	0.48	0.12	
WIOC/OC	0.05~0.93	0.37	0.18	0.15~0.70	0.38	0.16	0.33~0.82	0.52	0.12	

Table 1 Concentrations of carbonaceous components in the time-resolved (3-h) PM<sub>2.5</sub> samples in
the rural site of NCP during the whole sampling period, Period 1 (P1) and Period 2 (P2).

#### 852 Table 2 Average concentrations of the organic compound classes (ng m<sup>-3</sup>) in the time-resolved (3-h)

853 PM<sub>2.5</sub> samples in the rural site of NCP during the whole study period, Period 1 (P1) and Period 2

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(P2).									
Compounds	Whole perio	od (N=1	17)	Period 1 (N=28)			Period 2 (N=13)		
Compounds	Range	Mean	SD	Range	Mean	SD	Range	Mean	SD
n-Alkanes	9.97~722.2	206.9	149.3	94.7~722.3	343.7	134.1	25.1~103.2	54.3	22.4
CPI (C <sub>18</sub> -C <sub>36</sub> ) <sup>a</sup>	$1.08 \sim 8.62$	2.47	1.12	1.38~4.67	2.85	0.87	1.08~3.5	1.64	0.59
Fatty acids	64.6~1777	514.4	384.3	206.7~1528	900.3	358.3	81.4~234.4	145.3	47.7
CPI (C <sub>21:0</sub> -C <sub>30:0</sub> ) <sup>b</sup>	2.26~9.15	4.24	1.14	3.49~6.11	4.21	0.64	2.26~8.57	3.50	1.64
Fatty alcohols	3.18~975.9	192.6	187.4	62.4~638.2	322.0	150.7	16.6~100.2	33.9	22.6
Sugar compounds	15.9~2228	432.8	428.9	151.9~1727	718.0	403.1	39.7~241.3	93.2	52.9
galactosan (G)	1.03~97.78	18.5	20.6	2.16~97.8	29.5	27.9	1.45~13.3	4.61	3.13
mannosan (M)	0.69~54.82	9.78	10.4	1.61~54.8	15.0	13.3	0.96~6.63	2.83	1.43
levoglucosan (L)	5.56~1447	240.1	287.8	29.3~1428	404.0	344.0	11.2~123	47.8	26.2
L/M ratio	4.03~71.8	22.8	8.85	13.9~71.8	29.7	12.2	11.3~23.1	18.0	4.28
L/(G+M) ratio	1.38~19.3	8.05	2.59	5.3~19.3	10.1	3.41	4.58~10.2	6.77	1.97
PAHs	1.11~48.5	12.0	11.0	4.21~37.7	18.6	11.0	1.25~5.01	2.33	0.98
Hopanes	0.66~10.81	3.46	2.38	0.86~9.97	4.40	2.48	1.14~2.28	1.81	0.31
Phthalate esters	17.7~219.9	84.9	41.3	68.8~183.1	111.5	32.7	31.5~100.8	51.1	18.1
Phthalic acids	17.1~487.2	154.5	93.9	91.3~388.6	211.0	87.1	17.1~81	46.3	17.1
Isoprene SOA tracers	11.1~404.1	111.9	85.8	48.3~404.1	208.5	104.9	34.8~127.5	57.0	29.4
Monoterpene SOA tracers	11.1~166.2	66.1	31.2	37.3~166.2	85.3	34.9	26.7~64.5	44.6	12.6
$\beta$ -Caryophyllinic acid <sup>c</sup>	$0.49 \sim 77.7$	17.4	17.1	4.6~77.8	34.7	20.8	2.44~6.28	4.08	1.21
DHOPA <sup>d</sup>	1.59~35.3	9.36	7.15	4.06~35.3	15.6	9.80	2.7~6.99	4.16	1.42
Total measured organics	176.9~6249	1806	1308	843.3~5499	2973	1219	334.2~913.7	537.9	151.1
Total organics_C/OC <sup>e</sup> (%)	3.19~16.0	6.99	1.97	3.43~8.86	6.43	1.36	3.77~8.61	6.41	1.27

855 a CPI (C<sub>18</sub>-C<sub>36</sub>): carbon preference index for *n*-alkanes,  $(C_{19}+C_{21}+C_{23}+C_{25}+C_{27}+C_{29}+C_{31}+C_{33}+C_{35})/(C_{18}+C$ 

 $\textbf{856} \qquad (C_{18} + C_{20} + C_{22} + C_{24} + C_{26} + C_{28} + C_{30} + C_{32} + C_{34}).$ 

 $857 \qquad ^{b} CPI \left( C_{21.0} - C_{30.0} \right): earbon \ preference \ index \ for \ fatty \ acids, \\ \left( C_{22.0} + C_{24.0} + C_{26.0} + C_{30.0} \right) / \left( C_{21.0} + C_{23.0} + C_{25.0} + C_{27.0} + C_{29.0} \right).$ 

858 <sup>c</sup> $\beta$ -Caryophyllinic acid: a tracer of  $\beta$ -caryophyllene-derived SOA.

859 <sup>d</sup> DHOPA: 2,3-dihydroxy-4-oxopentanoic acid, a tracer of toluene-derived SOA.

860 • All the quantified organic compounds were converted to their carbon contents to calculate the OC ratios.

# **Figure Captions**

862	Figure 1. Backward trajectories of air masses (a,c) (provided by NOAA HYSPLIT modeling system,
863	http://ready.arl.noaa.gov/HYSPLIT.php), and fire spots (b,d) (provided by Fire Information
864	for Resource Management System, FIRMS, https://firms.modaps.eosdis.nasa.gov/firemap/),
865	during Period 1 (P1) (Jun 13th 21:00-16th 15:00, 2013) and Period 2 (P2) (Jun 22nd 12:00-24th
866	06:00, 2013). Sampling site represented as purple star.
867	Figure 2. Temporal variations of PM <sub>2.5</sub> , OC, EC, and WSOC during the whole sampling period.
868	Shadows denote the two representative periods.
869	Figure 3. Linear correlations of OC with WSOC (a), levoglucosan with OC and WSOC(b).
870	Figure 4. Diurnal variation of OC/EC (a), WSOC/OC and levoglucosan/OC (b).
871	Figure 5. Temporal variations of ten organic compound classes detected in the summertime PM <sub>2.5</sub>
872	samples at the rural site of NCP.
873	Figure 6. Temporal variations of organic tracers for biomass burning (a), and secondary products
874	derived from $\alpha$ -/ $\beta$ -pinene (b-d), $\beta$ -caryophyllene (e), and toluene (f).
875	Figure 7. Molecular distributions of <i>n</i> -alkanes (a and d), fatty acids (b and e), and fatty alcohols (c and
876	f) in the $PM_{2.5}$ of the rural area.
877	Figure 8. Linear correlations of fatty acids (a), fatty alcohols (b), 3-hydroxyglutaric acid (c), and $\beta$ -
878	caryophyllinic acid (d) with levoglucosan.
879	Figure 9. Diurnal variation of the detected organic compound classes.
880	Figure 10. Diurnal variation of the SOA tracers derived from oxidation of $\alpha$ -/ $\beta$ -pinene (a-d), $\beta$ -
881	caryophyllene (e), and toluene (f).
882	Figure 11. A comparison of the average contributions of different sources-derived organics (converted
883	to carbon content) to OC during P1 and P2.
884	Figure 12. Contributions (above) of primary organic carbon from biomass burning (OC <sub>bb</sub> ) and fungal
885	spores (OC <sub><i>j</i></sub> ), and secondary organic carbon from isoprene (SOC <sub><i>i</i></sub> ), $\alpha$ -/ $\beta$ -pinene (SOC <sub><i>p</i></sub> ), $\beta$ -
886	<i>c</i> aryophyllene (SOC <sub><i>p</i></sub> ), and toluene (SOC <sub><i>t</i></sub> ) to OC in the time-resolved (3 h) rural aerosols,
887	and their relative abundances (down). All the contributions were estimated by tracer-based
888	method.
889	Figure 13. Average contributions of direct emissions from biomass burning (BB) and fungal spores
890	$(OC_{fp})$ , secondary oxidation from isoprene $(SOC_i)$ , $\alpha$ -/ $\beta$ -pinene $(SOC_p)$ , $\beta$ -caryophyllene
891	$(SOC_p)$ , and toluene $(SOC_t)$ to OC in P1 and P2. All the contributions were estimated by
892	tracer-based method.
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899 Figure 1. Backward trajectories of air masses (a,c) (provided by NOAA HYSPLIT modeling system,

- 900 <u>http://ready.arl.noaa.gov/HYSPLIT.php</u>), and fire spots (b,d) (provided by Fire Information for
- Resource Management System, FIRMS, <u>https://firms.modaps.eosdis.nasa.gov/firemap/</u>), during Period
  1 (P1) (Jun 13<sup>th</sup> 21:00-16<sup>th</sup> 15:00, 2013) and Period 2 (P2) (Jun 22<sup>nd</sup> 12:00-24<sup>th</sup> 06:00, 2013). Sampling
  site represented as purple star.







911 Shadows denote the two representative periods.





Figure 3. Linear correlations of OC with WSOC (a), levoglucosan with OC and WSOC(b). 







Figure 4. Diurnal variation of OC/EC (a), WSOC/OC and levoglucosan/OC (b). 



923 Figure 5. Temporal variations of ten organic compound classes detected in the summertime  $PM_{2.5}$ 

samples at the rural site of NCP.



928 Figure 6. Temporal variations of organic tracers for biomass burning (a), and secondary products

929 derived from  $\alpha$ -/ $\beta$ -pinene (b-d),  $\beta$ -caryophyllene (e), and toluene (f).



Figure 7. Molecular distributions of n-alkanes (a and d), fatty acids (b and e), and fatty alcohols (c and







Figure 8. Linear correlations of fatty acids (a), fatty alcohols (b), 3-hydroxyglutaric acid (c), and β-caryophyllinic acid (d) with levoglucosan.



940 Figure 9. Diurnal variation of the detected organic compound classes.



947 Figure 10. Diurnal variation of the SOA tracers derived from oxidation of  $\alpha$ -/ $\beta$ -pinene (a-d),  $\beta$ -

948 caryophyllene (e), and toluene (f).



Figure 11 A comparison of the average contributions of different sources-derived organics (converted
 to carbon content) to OC during P1 and P2.



**959** Figure 12. Contributions (above) of primary organic carbon from biomass burning (OC<sub>bb</sub>) and fungal

960 spores (OC<sub>*fp*</sub>), and secondary organic carbon from isoprene (SOC<sub>*i*</sub>),  $\alpha$ -/ $\beta$ -pinene (SOC<sub>*p*</sub>),  $\beta$ -

962 relative abundances (down). All the contributions were estimated by tracer-based method.

**<sup>961</sup>** *caryophyllene* (SOC $_p$ ), and toluene (SOC $_t$ ) to OC in the time-resolved (3 h) rural aerosols, and their









- toluene (SOC<sub>t</sub>) to OC in P1 and P2. All the contributions were estimated by tracer-based method.