Authors' Response to ACP MS No.: acp-2019-669

We thank the reviewer for their comments and suggestions. In the text below, we outline our responses in blue. Line numbers refer to the revised manuscript.

Anonymous Referee #1

Winton et al report observations of the nitrogen isotopic composition of nitrate in Antarctic (DML) in the atmosphere, snow "skin layer", and depth profiles in snow pits. They use these observations combined with a snow chemistry model to understand what controls the variability in the nitrogen isotopes and nitrate concentrations and to assess the utility of such observations in ice cores as a proxy for past surface UV radiation. They conclude that although photolytic driven recycling and loss of reactive nitrogen is what determines the nitrate isotopes and abundances at this location, that variability in parameters such as snow accumulation rate have a large enough impact on the nitrogen isotopes so as to preclude the influence of variability in surface UV in determining the nitrogen isotopic composition of snow nitrate.

We agree that variability in the snow accumulation rate precludes the use of using the δ^{15} N-NO₃⁻ composition as a surface UV proxy in the short-term. However, longer-term UV trends may be inferred from ice cores at the site taking into account or constraining factors masking the UV-impact such as e-folding depth and accumulation rate.

The observations and model-based interpretation are robust and important as it would be useful to have a paleo-UV proxy. As the authors state, it is important to assess the sensitivity of nitrogen isotopes at particular locations in order to determine its usefulness as a proxy for any given process. Unfortunately, the paper is frustrating to read because it is it so repetitive and spends so much time on introductory material throughout the manuscript that it is hard to find the actual interpretation of the results. It is as if the authors are afraid to state their interpretation. I suggest moving (and condensing) the introductory material that is spread throughout the manuscript (especially in section 4) to the introduction and making sure that the actual interpretation is presented up front instead of hidden. Because of this, it's sometimes hard to determine if the results support their conclusions. The paper as is reads as a first or second draft, not a final manuscript ready for submission for publication. This is particularly evident in the conclusions, which is not even written in paragraph form.

We agree with the reviewer that too much background information is spread throughout the discussion rather than the introduction. We have condensed and moved the most relevant material to the introduction. Overall, we have revised the manuscript to provide better clarity on our interpretation of the results and conclusions.

Additional comments:

I find the concept of the number of times nitrate is recycled difficult to wrap my head around. Does this mean that each molecule of nitrate is recycled on average three times before being archived? Is this averaged over the depth of the photic zone? I would imagine some molecules (like those that are close to the surface in summer) are recycled more than others, so that there is actually some distribution of recycling events on a molecule per molecule basis so as to average around the number 3. Is this interpretation correct? I suggest elaborating on this in the methods section.

The number of recylings is the "average number of recyclings" undergone by the archived nitrate, i.e., below the zone of active photochemistry. Recycling includes the following processes: the combination of NOx production from nitrate photolysis in snow, venting to the air above the snowpack and subsequent atmospheric re-oxidation of NO_x to form atmospheric nitrate, the deposition (dry and/or wet) of a fraction of the product, and the export of another fraction (Erbland et al., 2015). In TRANSITS, there is a tracer/counter called CYCL, which, in a given box (snow layer or atmosphere), represents the average number of recyclings undergone by nitrate in that box. The CYCL value for primary nitrate is set to 0, and CYCL variables in the boxes are incremented by 1 each time NO₂ molecules cross the air-snow interface. The average number of recyclings is calculated as a mass weighted average of the CYCL values of the 52 snow layers (representing one week of snowfall) which are archived below 1 m over the course of 1 year, in order to average out any seasonal variability.

The average number of recyclings undergone by the archived nitrate at our study site in Dronning Maud Land (DML) is 2 for the last layer before leaving the photic zone, which means that, on average, the archived nitrate at DML has undergone 2 recyclings (i.e., loss, local re-oxidation, deposition). Erbland et al. (2015) notes that this number of recyclings represents an average value for the archived nitrate. Considering individual ions in the archived nitrate, the number of recyclings could be variable as some ions may have travelled through the entire snowpack zone of active photochemistry without being recycled, while some underwent many recyclings.

We have expanded the explanation of the number of recycling's in the methodology (lines 337-345) as follows:

Modified text: "TRANSITS calculates the average number of recyclings undergone by the archived NO_3^- , i.e., below the zone of active photochemistry. In TRANSITS, the average number of recyclings undergone by NO_3^- in a given box (snow layer or atmosphere) is represented by a tracer (or counter) called CYCL. The CYCL value for primary NO_3^- is set to 0, and CYCL variables in the boxes are incremented by 1 each time NO_2 molecules cross the air-snow interface. The average number of recyclings is calculated as a mass weighted average of the CYCL values of the 52 snow layers (representing one week of snowfall) which are archived below 1 m over the course of 1 year, in order to average out any seasonal variability. Erbland et al. (2015) notes that the number of recyclings represents an average value for the archived NO_3^- , i.e., considering individual ions in the archived NO_3^- , the number of recyclings could be variable as some ions may have travelled through the entire snowpack zone of active photochemistry without being recycled, while some underwent many recyclings."

Second paragraph of the introduction: It seems strange to say that the primary sources of nitrate are transport of nitrate from the stratosphere and transport of alkyl nitrates in the troposphere without mentioning transport of inorganic nitrate in the troposphere. Where is the evidence that inorganic nitrate is not transported in the troposphere to Antarctica? A model study suggests that it is certainly possible (Lee et al., 2014). Also, soil NOx should be mentioned as a NOx source in the troposphere.

We have added the transport of inorganic nitrate (fossil fuel combustion, soil, and lightning) and referenced Lee et al. (2014) in lines 53-56 as follows:

Modified text: "Primary sources of reactive nitrogen species to the Antarctic lower atmosphere and snow pack include the sedimentation of polar stratospheric clouds (PSC) in late winter (Savarino et al., 2007) and, to a minor extent, advection of oceanic methyl nitrate (CH₃NO₃) and peroxyacyl nitrates (PAN) (Jacobi et al., 2000;Jones et al., 1999;Beyersdorf et al., 2010), in addition to tropospheric transport of inorganic NO₃⁻ from lightning, biomass burning and soil emissions (Lee et al., 2014)."

Page 2 lines 61 – 62: Reprase to: "Model results from Zatko et al. (2016) suggest that..."

Done

Page 4 line 116: Are you referring to skin layer nitrate here? Since you are measuring nitrate in three different locations, you should always be specific about which location you are talking about and not just say "nitrate".

No, we are referring to ice core nitrate as mentioned in the sentence. We have added "ice core nitrate" for clarity.

Page 4 line 128: Perhaps you should say that you are referring to the e-folding depth of the snow photic zone, UV radiation, or something less vague.

Done

Page 5 line 157: If you remove the word "While" from the beginning of this sentence it will be grammatically correct.

Done

Page 8 line 259: By "lower" do you mean "shallower"? Lower could also mean deeper, so it would be better to use words like deeper and shallower when referring to the depth in the snow beneath the surface.

"Lower" has been replaced with "shallower".

Section 2.7: It seems quite important to state what value you assume for the N-isotopic composition of primary nitrate, before it is impacted by photolysis.

The δ^{15} N-NO₃⁻ value of primary nitrate is set to 19 ‰ as estimated by Savarino et al. (2007). This value is reported in Table S3.

Page 12 line 371: "values" of what? Concentration? Isotopes? Both? Since you measured more than one thing you need to be specific.

We have added δ^{15} N-NO₃⁻ values to this sentence.

Page 13 Section 3.5: You should probably note that gas phase HNO3 and particulate nitrate have different dry deposition rates.

Agreed, we have noted this in lines 418-419 as follows:

Modified text: "Although gas phase HNO_3 and particulate NO_3^- have different dry deposition rates..."

Section 3.6: I think you should elaborate on the difference between the actual (from photolysis) and apparent fraction factor. The latter is lower than the actual due to recycling. Also, related

to this, can the difference between the actual and apparent fractionation factor be used to calculate fractional loss?

We have discussed the difference in lines 88-91. We found that the single loss Raleigh model does not work well at sites with an annual signal in δ^{15} N-NO₃⁻, and therefore suggest that at DML it is not useful for calculating fractional loss. The text was modified as follows:

Modified text: "As this approach may oversimplify the processes occurring at the air-snow interface, Erbland et al. (2013) referred to the quantity as an "apparent" fractionation constant ($^{15}\varepsilon_{app}$). Thus, the apparent fractionation constant represents the integrated isotopic effect of the processes involving NO₃⁻ in the surface of the snow pack and in the lower atmosphere."

Page 15 line 455: Is this at DML?

Yes, the Weller et al. (2004) study is at DML. We have added "at DML" for clarity.

Page 15 line 459: Which results? Concentration? Isotopes? Both?

Both. We have added "NO₃⁻ mass concentrations and δ^{15} N-NO₃⁻" to the sentence.

Page 15 lines 472 - 473: Does the model simulate the influence of hoar frost and diamond dust on nitrate deposition? The way this sentence is written it seems that it does, but this is surprising to me as I didn't think the model was a meteorological/snow model that calculated such things. It seems that you are saying that diamond dust and hoar frost explain the difference between the model and the observations, and that the model has these things but they didn't happen in reality.

No, the model does not simulate the influence of hoar frost and diamond dust on nitrate deposition. We suggest that the difference between the simulated and observed nitrate concentrations in the skin layer is due to a sampling artefact where we are diluting the high nitrate concentrations of diamond dust and hoar frost observed in new deposition. We have modified the sentence in lines 564-566 as follows:

Modified text: "Here, extremely high concentrations of NO_3^- from new deposition from diamond dust and hoar frost are also found. In summary, it is likely that we do not measure such high NO_3^- mass concentrations in hoar frost and diamond dust in the skin layer because of sampling artefacts or blowing snow, which can dilute or remove the diamond dust and hoar frost."

Section 4.2: This is a perfect example of the section 4 having lots of introductory material without any results. The entire first paragraph reads like an introduction except for the very last sentence. However, the very last sentence of the first paragraph is not specifically supported by your observations. By the time I finished the 5 pages of section 4.2, I have no idea what you learned or what you are concluding from your new observations.

Section 4.2 has been condensed to 2 pages and the interpretation of results clarified.

Section 4.1: Again, I have no idea what you are concluding in this section.

We conclude that overall our nitrate conclusion measurements agree well with the literature and that the simulated results from TRANISTS match our observations with the exception of the skin layer concentrations. We provide possible reasons for this difference, and have modified Section 4.1 for clarity.

Page 17 line 540: "is no exception" to what?

We have deleted "is no exception".

Page 18 lines 547 - 551: It seems that if you are collecting both gas and particulate phase nitrate that you cannot conclude that sea salt promotes conversion of HNO3 to nitrate. I'm sure that it does, however, I don't see how your observations provide evidence of this.

We agree and have removed this section.

Page 18 lines 558-559: I suggest rephrasing this as "...Dome C where the underlying snowpack is the dominant... skin layer via photolytic recycling and redeposition.

Done

First paragraph of section 4.3: Need a reference for this statement.

Section of text has been removed.

Page 20 line 617: It could also be transported away as nitrate, not just NO2.

We have added nitrate to the sentence.

Page 21 line 648: "The contribution to what? of natural sources"

The "contribution of nitrate from natural sources" has been added to the sentence.

Page 21 lines 659-660: If you say that denitrification causes O-isotopes to increase, then you need cite a paper that provides evidence of this. However, I would delete this sentence since this paper does not address O-isotopes.

Agreed, we have delete the sentence.

Page 22 line 690: How was the apparent fractionation factor calculate? There should be an equation for this in the methods section.

Please see equation 7 in the methods in section 3.6.

Page 22 line 706: are these FNO2 values a daily mean, a daily maximum?

The model estimates are mean values for the 1-14 January 2017 period as stated in line 465.

Page 24 lines 742 - 746: What is the difference between the grain size and impurity concentrations at these two locations? The N-isotopes may be sensitive to something, but if that something shows no significant difference between these two locations than it won't be able to explain the differences in Nisotopes.

As far as we are aware, at Dome C and DML, there are only published values of major ions in the snow pack in the top 30 cm which is the depth relevant for influencing the e-folding depth. For example, in the top 30 cm of the snow pack nitrate concentrations at Dome C are \sim 75 ppb (Frey et al., 2009) while they are \sim 55 ppb at DML (this study). Insoluble dust concentrations are higher at DML than Dome C in the Holocene (Delmonte et al., 2019), while no black carbon data is available at DML for comparison. There is considerate variability in the snow grain size

across Antarctica (Brucker et al., 2010). In particular, wind crust layers which occur in the snowpack have larger grain sizes and these have been observed at Dome C (France et al., 2011).

Based on the available data, we don't have a clear understanding of why the e-folding depth is lower at DML. Station pollution is less than at Dome C (Helmig et al., 2020), thus it is unlikely related to black carbon. However, other impurities are deposited in DML snow due to the closer proximity to marine sources. Snow grain sizes may be smaller, which will increase scattering, at DML than Dome C (Brucker et al., 2010). The larger e-folding depth at Dome C is in part due the presence of windcrust layers which comprise larger grain sizes. Sensitivity studies show that nitrate impurities have only a small contribution on the e-folding depth compared to scattering by snow grains which dominate (Chan et al., 2015;France et al., 2011;Zatko et al., 2013). Further work is required to determine why the e-folding depth is lower at DML. We have discussed this in lines 692-700 as follows:

Modified text: "The e-folding depth depends on the density and grain size of snow crystals, and the concentration of impurities. In terms of published values, impurity concentrations are generally higher at DML, for example dust and major ion concentrations (Delmonte et al., 2019;Legrand and Delmas, 1988), due to proximity to marine sources. Yet station pollution is greater at Dome C (Helmig et al., 2020), and thus the lower e-folding depth is unrelated to black carbon concentrations. Furthermore, there is considerate variability in snow grain size across Antarctica. The larger e-folding depth in windcrust layers at Dome C is due to larger grain sizes in those layers (France et al., 2011). Snow grain size may be smaller at DML, which will increase scattering (Brucker et al., 2010), but further work is required to confirm if this is the dominate factor influencing the lower e-folding depth at DML. Sensitivity studies show that NO_3^- impurities make a small contribution to the e-folding depth compared to scattering by snow grains which dominate (France et al., 2011;Chan et al., 2015;Zatko et al., 2013)."

Page 25 lines 786-787: The e-folding depth is 20 years yet the nitrate is archived after 5-6 years. This seems conflicting. Why is nitrate archived when it is still in the photic zone?

Weller et al. (2004) determine archived nitrate as the typical mean concentrations representative for the last 100 years. We have deleted the e-folding time of 20 years from sentence.

Page 25 lines 792 and 796: Are you referring to observed or modeled values here?

Observed values. We have added "observed" sentence.

Page 25 line 807: Snow optical properties are part of what determines the depth of the photic zone, so it seems weird to mention these two things separately here as if they aren't related.

We have deleted snow optical properties.

Page 26 Line 821: "lower" than what?

Lower than the base case profile. We have added "base case" to the sentence.

Page 28, last sentence of section 4.5.3: Explicitly state here why the sensitivity is greater at Dome C than at DML.

Done

Page 28 line 899: What would cause a change in the e-folding depth?

The e-folding depth could change due to a change in dust or black carbon concentrations (which are light absorbing impurities) or a change in the snow morphology in a particular snow layer. This has been added to lines 843-844 as follows:

Modified text: "The e-folding depth could change over time due to higher or lower dust or black carbon concentrations or a change in the snow grain size in a particular snow layer."

Page 29 line 919: What does "unpicking" mean in this context? Perhaps choose a better word.

Unpicking been replaced with disentangling.

Page 29 line 931: This is percent. Should it be permil?

Yes, percent symbols have been changed to permil symbols.

Figure 1 caption: State the difference between HiVol 2 and 3.

Done.

Figure 2: The boxes in this figure are totally unclear to me. What are the gray boxes trying to convey?

The box a) is in insert of the Kohnen Station and boxes b) and c) our inserts of our two sampling sites at the station. We have edited the caption to make this clearer.

Figure 4: Mark the seasons (e.g., shade winter) in c) and f). It's hard to see the seasonal cycles.

We have shaded the seasons in panel c) and updated the caption. The snow pit in Frey et al. (2009) was not dated and thus we cannot shade the seasons for Dome C.

Figure 4: Why does the x-axis scales to all the way to 1500 and 300 when the observations are much lower than this? The axes should be scaled by the range of the data.

We deliberately chose to keep the x-axis the same for each plot rather than scaling by the range of the data. This is so the reader can see the difference in concentration and isotopic values between the atmosphere, skin layer and snowpack profiles and between the two sites. In particular, we wanted to highlight the denitrification of the snowpack from enriched values in the snow pits to depleted values in the atmosphere.

Figure 5: Why is "recycling factor" listed as a process occurring below the snow photic zone? How is "tropospheric input" different from "net lateral advection"?

We have moved recycling to encompass the atmosphere and snow boxes. We have renamed net lateral advection to tropospheric input.

Figure 7: Why are these values for accumulation rates chosen as sensitivity studies in the model? What is the base case that you are changing each parameter around?

The justification for the range of accumulation rates used in the sensitivity study can be found in section 4.5.1.

The base case is explained in section 4.5 and in the caption, we refer the reader to that section.

Figure 8: This is showing the change in TCO around what value?

We used present day TCO values that were used in all our calculations. These values vary weekly and can be found the supplement (Table S3). For each week, a constant amount of ozone (e.g. 100 DU) was added or subtracted from these values. We have added this information to the caption and text in lines 816-817 as follows:

Modified text: "For each week, a constant amount of ozone (e.g. 100 DU) was added or subtracted from these present day values."

References

Brucker, L., Picard, G., and Fily, M.: Snow grain-size profiles deduced from microwave snow emissivities in Antarctica, Journal of Glaciology, 56, 514-526, 2010.

Chan, H., King, M., and Frey, M.: The impact of parameterising light penetration into snow on the photochemical production of NOx and OH radicals in snow, Atmos. Chem. Phys, 15, 7913-7927, 2015.

Delmonte, B., Winton, H., Baroni, M., Baccolo, G., Hansson, M., Andersson, P., Baroni, C., Salvatore, M. C., Lanci, L., and Maggi, V.: Holocene dust in East Antarctica: Provenance and variability in time and space, The Holocene, 0959683619875188, 2019.

Erbland, J., Savarino, J., Morin, S., France, J., Frey, M., and King, M.: Air–snow transfer of nitrate on the East Antarctic Plateau–Part 2: An isotopic model for the interpretation of deep ice-core records, Atmospheric Chemistry and Physics, 15, 12079-12113, 2015.

France, J., King, M., Frey, M., Erbland, J., Picard, G., Preunkert, S., MacArthur, A., and Savarino, J.: Snow optical properties at Dome C (Concordia), Antarctica; implications for snow emissions and snow chemistry of reactive nitrogen, Atmospheric Chemistry and Physics, 11, 9787-9801, 2011.

Frey, M. M., Savarino, J., Morin, S., Erbland, J., and Martins, J.: Photolysis imprint in the nitrate stable isotope signal in snow and atmosphere of East Antarctica and implications for reactive nitrogen cycling, Atmospheric Chemistry and Physics, 9, 8681-8696, 2009.

Helmig, D., Liptzin, D., Hueber, J., and Savarino, J.: Impact of exhaust emissions on chemical snowpack composition at Concordia Station, Antarctica, The Cryosphere, 14, 199-209, 2020. Legrand, M. R., and Delmas, R. J.: Soluble impurities in four Antarctic ice cores over the last 30 000 years, Annals of Glaciology, 10, 116-120, 1988.

Weller, R., Traufetter, F., Fischer, H., Oerter, H., Piel, C., and Miller, H.: Postdepositional losses of methane sulfonate, nitrate, and chloride at the European Project for Ice Coring in Antarctica deep-drilling site in Dronning Maud Land, Antarctica, Journal of Geophysical Research: Atmospheres, 109, 2004.

Zatko, M. C., Grenfell, T. C., Alexander, B., Doherty, S. J., Thomas, J. L., and Yang, X.: The influence of snow grain size and impurities on the vertical profiles of actinic flux and associated NO x emissions on the Antarctic and Greenland ice sheets, Atmospheric Chemistry and Physics, 13, 3547-3567, 2013.

Authors' Response to ACP MS No.: acp-2019-669

We thank the reviewer for the comments and suggestions. In the text below, we outline our responses in blue. Line numbers refer to the revised manuscript.

Anonymous Referee #2

This work present new observations of the isotopic composition of nitrate in atmospheric, skin layer, surface snow and snowpit samples from Dronning Maud Land, Antarctica. The goal of the work is to understand the primary driver(s) of post-depositional processing of nitrate in this environment, based upon a snowpack box model, and determine whether it would be possible to uncover a d15N-nitrate signal that is dependent upon total column ozone. A great deal of work has been done on interpreting the d15N-nitrate in surface snow and the atmosphere at Dome C, and this work seeks to expand the application of similar tools to another site with different environmental conditions (i.e. accumulation rate, snowfall timing, and e-folding depth).

While the data and methods in this work appear sound, and the results are interesting, the manuscript reads as a first draft. With so many authors on this paper, and several who have led work at Dome C, it is surprising how disorganized and filled with typos this work is. Overall, it is difficult to connect the results with the interpretation. Much of the discussion section reads as introduction, and the introduction itself is overly repetitive.

We agree with reviewer, and have reorganised and edited the entire manuscript with particular focus on the introduction and discussion. Please also see our response to referee #1's main comment.

Most critically on the interpretation front, is that the authors must first consider the efolding depth results THEN the results considering changes in the snow accumulation, timing, etc. The model (shown in Figure 7) is simply not at all good enough to draw the conclusions the authors are drawing UNTIL it is shown that with the reduced e-folding depth the model can actually reproduce the observations. This is done as a final step and negates all of the previous discussion that suggests that the model output is robust, and therefore negates the conclusions being drawn.

We agree with the reviewer that the e-folding depth results need to be accounted for before suggesting that the model output fits the observations. We reran TRANISTS with a 5 cm e-folding depth scenario as suggested by reviewer #2. First, we modified the methods section to include a 5 cm e-folding depth scenario (section 2.7 Air-snow transfer modelling). Second, we have added a section on the e-folding depth in the results section (section 3.8 Light attenuation through the snow pack) so the reader can see the observed results of a lower measured e-folding depth at DML upfront. We have also compared the new 5 cm e-folding depth case to the base case in section 3.9 Simulated nitrate mass concentrations and isotopic ratios from TRANSITS modelling. Third, we compare the lower e-folding depth to Dome C and discuss the impact of the lower e-folding depth on post-depositional processes (section 4.3.2.2 Nitrate recycling). Fourth, we have rerun the snow accumulation rate and snowfall timing sensitivity tests with an e-folding depth of 5 cm and modified Figs. 6, 7 and 8 and Table 2 with the new sensitivity test results. An observed e-folding depth of 5 cm was used as it has good fit with observations down to 30 cm depth. In light of the new sensitivity results from TRANSITS, we have reorganised section "4.5 Sensitivity of δ^{15} N-NO₃⁻ to deposition parameters and implications for interpreting

ice core records of δ^{15} N-NO₃⁻ at DML". Here, we discuss the TRANSITS modelling results by first showing that the base case scenario cannot reproduce the observations and that a reduced e-folding depth is required. Next, we discuss the sensitivity results of a variable snow accumulation rate and snowfall timing with an e-folding depth of 5 cm. With an e-folding depth of 5 cm we are able to reproduce the observations and thus our original conclusions that TRANSITS does a good job are valid.

In section 4.2.3, confidence is built in that TRANSITS can reproduce the seasonal cycle, not at the site here that we are comparing with, but at other sites in Antarctica? This makes no sense.

There are no year-round measurements of atmospheric or skin layer δ^{15} N-NO₃⁻ at DML to compare to the TRANSITS seasonality simulations. This section has been rewritten to show that of the available year-round observations and seasonality simulations of atmospheric and skin layer δ^{15} N-NO₃⁻ and nitrate mass concentrations in Antarctica, the seasonal pattern is the same at all Antarctic sites. The section has been renamed 4.2.2 Temporal variability of nitrate deposition.

The introduction should be rewritten to better frame where the paper is headed, after the discussion section is reorganized and edited.

Done.

It is well established that photolysis is a major driver of nitrate loss across East Antarctica, between the vast literature at Dome, Erbland's work and Shi's work. At this point, this should be a starting point, not something that is derived step-wise in the manuscript. Furthermore, it would help tighten up and shorten the manuscript. Finally, take a closer look at and include Shi et al., Investigation of post-depositional processing of nitrate in East Antarctic snow: isotopic constraints on photolytic loss, re-oxidation, and source inputs, Atmos. Chem. Phys., 15, 9435–9453, 2015, https://doi.org/10.5194/acp-15-9435-2015.

We agree, and this is reflected in the revised introduction and discussion. In addition, we have cited the Shi et al. (2015) reference in the appropriate places throughout the manuscript, and added the key findings of the paper in lines 137-140 as follows:

Modified text: "Erbland et al. (2013) suggest that NO_3^- loss at the coast reflects both photolysis and evaporation processes, while Shi et al. (2015) proposes that NO_3^- loss at the coast cannot be fully explained by local post-deposition processes and that seasonal cycles in the snowpack reflect stratospheric and troposphere NO_3^- sources during the cold and warm seasons respectively."

The evaporation (volatilization) of nitrate needs more discussion in the introduction. It is stated that that is negligible several times and then, finally, it is explained at lines 680-685 that this is temperature dependent process and THEREFORE not important AT THIS SITE. This should be detailed much earlier on.

As part of the revised introduction, the evaporation of nitrate is discussed in lines 86-98 as follows:

Modified text: "Fractionation constants, which assume a Rayleigh single loss and irreversible process of NO₃⁻ removal from the snow between phases during evaporation-condensation

processes, have been calculated to separate the isotopic signature of evaporation and photolysis processes. As this approach may oversimplify the processes occurring at the air-snow interface, Erbland et al. (2013) referred to the quantity as an "apparent" fractionation constant ($^{15}\varepsilon_{app}$). Thus, the apparent fractionation constant represents the integrated isotopic effect of the processes involving NO₃⁻ in the surface of the snow pack and in the lower atmosphere. Nitrate evaporation from the snow pack has a $^{15}\varepsilon_{app}$ of ~0 as determined by two independent studies (Erbland et al., 2013;Shi et al., 2019). This indicates that during NO₃⁻ evaporation, the air above the snow is not replenished and thus there is only a small NO₃⁻ mass loss. In comparison, fractionation constants associated with laboratory studies and field observations of NO₃⁻ photolysis are large: $^{15}\varepsilon_{app} = -34$ ‰ (Berhanu et al., 2014;Meusinger et al., 2014) and -54 < $^{15}\varepsilon_{app} < -60$ ‰ (Frey et al., 2009;Erbland et al., 2013), respectively. The negative fractionation constant obtained from photolysis implies that the remaining NO₃⁻ in the skin layer snow is enriched in δ^{15} N-NO₃⁻. In turn, the atmosphere is left with the source of NO_x that is highly depleted in δ^{15} N-NO₃⁻. It follows that evaporation of NO₃⁻ is negligible on high-elevation Antarctic sites (Erbland et al., 2013;Shi et al., 2019)."

More evidence should be provided that the collection method is robust for representing the isotopic composition of nitrate in the air. The authors state that is "assumed that the atmospheric NO3- collected on glass fibre filters represents the sum of atmospheric particulate NO3- and HNO3 (gas phase)" and then says this is described in Frey et al. (2009). Except Frey et al. makes this exact same ASSUMPTION without providing any evidence that this is the case. Later in this manuscript it is suggested that differences found from earlier work by Weller and Wagenbach may be because different filters were used (Teflon) – which filters are robust? Might the Weller and Wagenbach filters only represent one phase? and if so the authors should understand what the implications of this is for the data comparison.

The glass fiber filters used in this study were employed and tested previously at Dome C, i.e., Frey et al. (2009) do not state an assumption but report evidence from tests with second stage filters. Accordingly, the atmospheric nitrate collected on glass fiber filters represents the sum of atmospheric particulate nitrate ($p-NO_3^-$) and gaseous nitric acid (HNO_3). The bulk of HNO_3 present in the gas phase adsorbed most likely to aerosols on the filter. This is supported by the observation that second-stage filters (Whatman 41), known to trap HNO_3 quantitatively (Morin et al., 2007), showed either very low nitrate concentrations or none at all.

In section 3.5 and in figure 5, it is never explained what scenario 1 vs scenario 2 is, and where does the 296 for wet deposition come from?

The value for wet deposition (296 pg m⁻² s⁻¹) in scenario two is calculated using equations 3-5 (total deposition – dry deposition). Scenario one and scenario two are now described in the caption of Fig. 5 and lines 429-443 as follows:

Modified text: "Taking this simple mass balance approach, a schematic of NO_3^- mass fluxes for two scenarios are illustrated in Fig. 5. Scenario 1 is an average annual budget for DML (Fig. 5a). As the atmospheric campaign did not cover an entire annual cycle, we use estimates of atmospheric NO3- fluxes at DML reported by Pasteris et al. (2014) and Weller and Wagenbach (2007) of 43 and 45 pg m⁻² s⁻¹, respectively, as year round dry deposition fluxes. Due to the linear relationship of ice core NO_3^- mass concentrations with the inverse accumulation, the authors assume that the magnitude of the dry deposition flux is homogenous over the DML region. Mean annual mass concentrations of NO_3^- in our snow pits suggest a total NO₃⁻ deposition mass flux of 110 pg m⁻² s⁻¹ and therefore a wet deposition mass flux of 65 pg m⁻² s⁻¹.

However, at relatively low snow accumulation sites where photolysis drives the fractionation of NO_3^- from the surface snow to atmosphere (Frey et al., 2009), it is necessary to take into account the skin layer in the NO_3^- flux budget as this air-snow interface is where air-snow transfer of NO_3^- takes place. In scenario 2, we utilise the available NO_3^- mass concentrations measured in aerosol, skin layer, and snow pits from the ISOL-ICE campaign to estimate the mass flux budget for January 2017 (Fig. 5b). The dry deposition mass flux of atmospheric NO_3^- during January 2017 at DML averages 64 ± 38 pg m⁻² s⁻¹ (Table S5). The NO_3^- mass flux to the skin layer is 360 pg m⁻² s⁻¹, however only 110 pg m⁻² s⁻¹ of NO_3^- is archived. Considering the active skin layer, only 30 % of deposited NO_3^- is archived in the snow pack while 250 pg m⁻² s⁻¹ is re-emitted to the overlaying atmosphere."

The authors make a claim regarding d18O-NO3- data late in the manuscript (line 660) – this is inappropriate given that none of that data is shown. Further, the claim is that the d18O rises due to denitrification, but there is not previous validity to this statement in the literature.

We have deleted this sentence as this manuscript does not address O-isotopes.

It is stated that the poorly constrained "quantum yield of NO3- photolysis in natural snow" yields a flux of NOx that is 50 times too high. Can this not be tested in TRANSITS? And shouldn't this affect the TRANSITS results as well and not just the estimated calculations here?

We agree that it would be useful to further test the sensitivity of NO_x fluxes to quantum yield in TRANSITS given the large uncertainty of this quantity. However, this has been done previously and we therefore refer to the literature and clarify the statement in lines 471-487 as follows:

Modified text: "It should be borne in mind that the above simple model estimates (Eq. (8)) may significantly underestimate the real emission flux. Previous comparisons of F_{NO2} computed with Eq. (8) and F_{NOx} measured at Dome C showed that observations can exceed model predictions by up to a factor 50 (Frey et al., 2015;Frey et al., 2013). While NO₃⁻ mass concentrations in snow, the surface actinic flux and e-folding depth were measured at the DML field site, quantum yield of NO₃⁻ photolysis in surface snow (Φ NO₃⁻) was not, but introduces significant uncertainty in the model estimates. Previous lab measurements on natural snow samples collected at Dome C showed Φ NO₃⁻ to vary between 0.003 and 0.05 (Meusinger et al., 2014). As described above (section 2.6) JNO₃⁻ used in Eq. (8) was calculated with Φ NO₃⁻ at -30 °C (= 2 x 10⁻³) after Chu and Anastasio (2003), which is near the lower end of the observed range. Thus, up to half of the mismatch between Eq. (8) and Dome C observations can be explained by adjusting Φ NO₃⁻. Another factor contributing to larger fluxes and not included in Eq. (8) is forced ventilation.

In the more sophisticated TRANSITS model, Erbland et al. (2015) found that the photolytic quantum yield was one of the major controls on archived flux and primary input flux at Dome C. Erbland et al. (2015) initially used a quantum yield of 2.1 x 10^{-3} at 246 K (France et al., 2011) but it underestimated NO₃⁻ recycling and overestimated primary NO₃⁻ trapped in snow. Adjusting the quantum yield to 0.026, within the range observed in the lab (Meusinger et al., 2014), gave more realistic archived δ^{15} N-NO₃⁻ values. However, at Dome C TRANSITS simulated F_{NO2} fluxes were about a factor of 9 - 18 higher than observed F_{NOx}. Erbland et al.

(2015) suggested that the discrepancy could result from the simplifications made in the TRANSITS model regarding the fate of NO_3^- photolysis products."

Finally, consider better comparing with Zatko et al., 2016 throughout the discussion – the equations used here are very similar to that paper and that work does in fact quantify the recycling despite the line later that only two studies have done so and then Zatko's work is compared with.

What we meant to say is that there are only two methods in the literature to quantify the number of recyclings (Erbland et al., 2015;Davis et al., 2008). Zatko et al. (2016) uses the Davis approach. We have edited the section on recycling, stating there are in fact three studies and have included the Zatko et al. (2016) paper in our comparison in lines 671-672 as follows:

Modified text: "Only three studies have attempted to quantify the degree of NO₃⁻ recycling between the air and snow (Davis et al., 2008;Erbland et al., 2015;Zatko et al., 2016)."

Also Zatko's earlier work (2013) on impurities should be better considered. Later it is attributed to Geng et al. for some reason. Can impurities in the snow not help account for some of the difference in the photolysis results? i.e. difference in impurities in the snow at DML and Dome C could help to account for the significantly lower photolysis rates at DML.

The impact of impurities on e-folding depth is addressed in section 4.3.2.2. Please see response to referee #1 concerning the impact of grain size and impurities on e-folding depth.

The timing and rate of snowfall CANNOT explain the misalignment between the observations and model results (Line 845). The e-folding depth is critical to right FIRST, then test the sensitivity of these other parameters to determine how to make the fit better. Literally none of the model results presented in Figure 7 before the e-folding depth results come close to overlapping with the observations. Also, you should consider having the model results on a different x-axis so that the depth profile, especially for d15N, can be seen. At this point, the idea of seasonality and the possibility of interannual interpretation is difficult to see.

We agree and thank the reviewer for the valuable comment to improve the manuscript. Please see our response to the e-folding depth comment above. We carried out the TRANISTS runs as suggested by referee #1. An e-folding depth of 5 cm has a much better fit with the observations. With the new TRANSITS runs, our conclusions reinforce the importance of accounting for the e-folding depth measurements across Antarctica. Regarding the x-axis on Fig. 7, the new TRANSITS runs with an e-folding depth of 5 cm move the simulated δ^{15} N-NO₃⁻ values to more negative values. The better fit with the observations means it is much easier to see the interannual variability in the δ^{15} N-NO₃⁻. In addition, the seasonal variability is clearly visible in Fig. 4.

Technical comments:

I do my best to point out a lot of simple errors, but it would behoove the authors to take a closer read on the next version of the manuscript.

Line 10: d15N-NO3- should be (d15N-NO3-)

Done.

Line 20: photochemical processes cannot drive the archiving of nitrate; it drives the loss of nitrate or recycling of nitrate from the snow.

"Photochemical processes" has been replaced with "nitrate recycling".

Line 34: TOC should be TCO

Done.

Line 38: rephrase this line – it appears as if you are suggesting that NO3- is formed from oxidation of N2 $\,$

Done.

Line 57: J should be (J)

Done.

Line 64: add a comma after Greenland

Done.

Line 97: this is the first use of PSC, spell it out and explain their purpose here

Done.

Line 169-170: the additional skin layer samples for comparison should be represented in the data figures.

The samples representing spatial variability are already plotted in Fig. S6 and Fig 3. To increase visibility of these samples, we have added an error bar representing the spatial variability to Fig. 3. This is considerably lower than the instrumental variability (error bars are smaller than sample points).

Line 219: a references for the seawater ratio (I assume this means sea salt ratio) should be included.

Keene et al. (1986) reference added for the sea salt ratio.

Line 300: followed should be follows

Done.

Line 303: add the word in after changes

Done.

Line 370: remove and before archived

Done.

Line 398: as a year round does not make sense

Done.

Line 457: inter-decadal should be interannual

Done.

Line 475: Fig 5 should not be cited here

Changed to Fig. 7.

Line 490: remove are

This sentence has been removed following the main comments.

Line 530: up taken should be taken up

This sentence has been removed following the main comments.

Line 536: the idea that "NO3- peaks are substantially modified after burial" undermines so much of the current manuscript that suggests that NO3- is archived once buried. Rephrase.

This sentence has been removed following the main comments.

Line 566: rephrase "form of nitrogen to skin layer"

Done.

Line 612: back to THE skin later with a day, or IS transported away

Done.

Line 624-625: see comment above about better explaining the evaportation results; and Shi et al 2019 reflect field conditions, not laboratory conditions.

Section 4.3.1 Evaporation has been removed following the main comments of referees #1 and #2.

Line 643-653: I have serious issues here with the interpretation of previous literature. First and foremost, the assumption that anthropogenic emissions of NOx are positive in d15N and natural emissions are negative in d15N is false and not up to date. Recent work shows that vehicle emissions are, in fact, negative in d15N (Miller et al., 2017, Vehicle NOx emission plume isotopic signatures: Spatial variability across the eastern United States, J. Geophys. Res. Atmos., 122, doi:10.1002/2016JD025877)/ At least three works show that fertilized soil emissions (which are not considered a natural source in emission inventories) are very negatives in d15N (Yu & Elliott, 2017, Novel method for nitrogen isotopic analysis of soilemitted nitric oxide. Environmental Science & Technology, 51(11), 6268-6278, https://doi.org/10.1021/acs.est.7b00592; Miller et al., 2018, Isotopic composition of in situ soil NOx emissions in manure fertilized cropland, Geophysical Research Letters, 45(21), 12058-12066, https://doi.org/10.1029/2018GL079619.; Li & Wang, 2008, Nitrogen isotopic signature of soil-released nitric oxide (NO) after fertilizer application. Atmospheric Environment, 42(19), 4747–4754. https://doi.org/10.1016/j.atmosenv.2008.01.042.). Geng et al. make the exact same false interpretation that anthropogenic sources are positive and therefore are ruled out in interpretation of a Summit, Greenland core – do not cite this is evidence when it is simply an unwarranted assumption. Finally, please be more precise in the language here – emission sources emit NO (except diesel engines, which can also emit NO2) or you can call it NOx; they do not "have positive d15N-NO3-" values since they do not emit nitrate, it is secondarily formed and subject to partitioning in the atmosphere, which Geng et al. invoke as a major mechanism to explain Greenland ice results, and this is wholly ignored in the current work.

We thank the reviewer for providing additional references concerning the negative isotopic signature of anthropogenic emissions of NO_x. We have edited this section to include the recent work on vehicle NO_x emissions and fertilised soil emissions and provided the references suggested by the reviewer. We have removed the Geng et al. reference which discards anthropogenic nitrate as a potential source to Greenland snow. When referring to the isotopic signature of emission sources, we have replaced δ^{15} N-NO₃⁻ with δ^{15} N-NO_x. In light of the negative source signature of NO_x emissions, our interpretation that anthropogenic sources do not contribute to the atmospheric δ^{15} N-NO₃⁻ at DML remains unchanged based on i) the well-established literature in which photolysis is the major driver of atmospheric δ^{15} N-NO₃⁻ values over low accumulation sites in East Antarctica (e.g. Frey et al., 2009;Erbland et al., 2015;Erbland et al., 2013;Shi et al., 2015;Shi et al., 2018), and ii) modelling study by Lee et al. (2014) that shows fertilised soil NO_x emissions to Antarctica are minor. We modified the text as follows (lines 637-642):

Modified text: "The δ^{15} N-NO_x source signature of the main natural NO_x sources (biomass burning, lightning, soil emissions; δ^{15} N-NO_x <0 ‰) is lower than anthropogenic NO_x sources, which generally have positive δ^{15} N-NO_x values (-13 < δ^{15} N-NO_x < 13 ‰; e.g. (Hastings et al., 2013;Kendall et al., 2007;Hoering, 1957) except in the case of vehicle and fertilised soil NO_x emissions which have negative δ^{15} N-NO_x values (-60 < δ^{15} N-NO₃⁻ <12 ‰; Miller et al. (2017);Yu and Elliott (2017);Miller et al. (2018);Li and Wang (2008). However, a NO₃⁻ source contribution from fertilised soil NO_x emissions to Antarctica is thought to be minor (Lee et al., 2014)."

Line 690-695: a range of -19 per mil to +12 per mil does not at all "nicely match" with the expected -59 per mil to -16 per mil.

Replaced "nicely match" with "falls within the range".

Line 698: add an before annual

Done.

Section 4.3.3 – seems odd to switch to section i, ii, iii, etc here when earlier subsections are numbered in series (i.e. 4.3.3.1, 4.3.3.2, etc).

Replaced i, ii, iii with 1, 2, 3.

Line 780: Concluding that photolysis is an important driver is not an interesting result given the vast evidence for this throughout the EAIS. The other conclusions are still interesting but this should not be the primary focus. Furthermore, the fact that there is less photolysis than expected is really very interesting.

This section has been removed following the main comments of referee #1.

Line 850-851: Interestingly, Geng et al. 2015 ignores surface snow work at Summit, Greenland to invoke that post-depositional processing can explain everything (Fibiger et al. (2016), Analysis of nitrate in the snow and atmosphere at Summit, Greenland: Chemistry and transport, J. Geophys. Res. Atmos., 121, 5010–5030, doi:10.1002/2015JD024187; Fibiger et al. (2013), The preservation of atmospheric nitrate in snow at Summit, Greenland, Geophys. Res. Lett., 40, 3484–3489, doi:10.1002/grl.50659.). Here, you are showing that is critical to use surface

work to best determine how to interpret ice cores. I suggest you reconsider including comparison with Geng's work here.

Done.

Line 855: depending ON the timing

Done.

Line 902: due TO photolysis-driven

Done.

Line 905: This should reference Zatko et al. not Geng et al.

Done.

Line 913: I strongly disagree with the statement that "TRANSITS doe such a good job of simulating NO-3 recycling in Antarctica" unless you do the e-folding fit first and then explore sensitivities. (I do agree that it is an excellent recommendation to use TRANSITS to assess sites that would be useful for interpreting nitrate isotopic records!)

Please see our response to the e-folding depth comment above. Based on the improved fit of the new sensitivity tests using the e-folding depth of 5 cm, we have kept this statement in the manuscript.

Line 919: unpicking is a strange word here – distinguishing?

Conclusions were rewritten following the suggestion of referee #1. "Unpicking" is no longer used.

Line 922: resulting IN an enrichment

Done.

Line 929: this conclusion would make more sense if the e-folding depth model experiments were done first.

Done.

Line 945: THE ground

Done.

Figure 3: how is the data averaged here?

The RACMO precipitation data is published as daily values. See the data publication for further information <u>https://data.bas.ac.uk/full-record.php?id=GB/NERC/BAS/PDC/01137</u>. The wind data from the AWS is hourly. We have added this to the caption.

Figure 4: x-axis on right side is misspelled

Done.

Figure 5: what is scenario 1 vs 2, this is never explained

This is now explained in text (lines 429-443; see comment above) and in the caption of Fig. 5.

Table 2: Archived should be >30 cm not <30 cm, correct?

Yes, this has been corrected.

References

Berhanu, T. A., Meusinger, C., Erbland, J., Jost, R., Bhattacharya, S., Johnson, M. S., and Savarino, J.: Laboratory study of nitrate photolysis in Antarctic snow. II. Isotopic effects and wavelength dependence, The Journal of chemical physics, 140, 244306, 2014.

Chu, L., and Anastasio, C.: Quantum yields of hydroxyl radical and nitrogen dioxide from the photolysis of nitrate on ice, The Journal of Physical Chemistry A, 107, 9594-9602, 2003.

Davis, D. D., Seelig, J., Huey, G., Crawford, J., Chen, G., Wang, Y., Buhr, M., Helmig, D., Neff, W., and Blake, D.: A reassessment of Antarctic plateau reactive nitrogen based on ANTCI 2003 airborne and ground based measurements, Atmospheric Environment, 42, 2831-2848, 2008.

Erbland, J., Vicars, W., Savarino, J., Morin, S., Frey, M., Frosini, D., Vince, E., and Martins, J.: Air–snow transfer of nitrate on the East Antarctic Plateau–Part 1: Isotopic evidence for a photolytically driven dynamic equilibrium in summer, Atmospheric Chemistry and Physics, 13, 6403-6419, 2013.

Erbland, J., Savarino, J., Morin, S., France, J., Frey, M., and King, M.: Air–snow transfer of nitrate on the East Antarctic Plateau–Part 2: An isotopic model for the interpretation of deep ice-core records, Atmospheric Chemistry and Physics, 15, 12079-12113, 2015.

Frey, M., Roscoe, H., Kukui, A., Savarino, J., France, J., King, M., Legrand, M., and Preunkert, S.: Atmospheric nitrogen oxides (NO and NO 2) at Dome C, East Antarctica, during the OPALE campaign, Atmospheric Chemistry and Physics, 15, 7859-7875, 2015.

Frey, M. M., Savarino, J., Morin, S., Erbland, J., and Martins, J.: Photolysis imprint in the nitrate stable isotope signal in snow and atmosphere of East Antarctica and implications for reactive nitrogen cycling, Atmospheric Chemistry and Physics, 9, 8681-8696, 2009.

Frey, M. M., Brough, N., France, J. L., Anderson, P. S., Traulle, O., King, M. D., Jones, A. E., Wolff, E. W., and Savarino, J.: The diurnal variability of atmospheric nitrogen oxides (NO and NO₂) above the Antarctic Plateau driven by atmospheric stability and snow emissions, Atmos. Chem. Phys., 13, 3045-3062, 10.5194/acp-13-3045-2013, 2013.

Hastings, M. G., Casciotti, K. L., and Elliott, E. M.: Stable isotopes as tracers of anthropogenic nitrogen sources, deposition, and impacts, Elements, 9, 339-344, 2013.

Hoering, T.: The isotopic composition of the ammonia and the nitrate ion in rain, Geochimica et Cosmochimica Acta, 12, 97-102, 1957.

Keene, W. C., Pszenny, A. A., Galloway, J. N., and Hawley, M. E.: Sea-salt corrections and interpretation of constituent ratios in marine precipitation, Journal of Geophysical Research: Atmospheres, 91, 6647-6658, 1986.

Kendall, C., Elliott, E. M., and Wankel, S. D.: Tracing anthropogenic inputs of nitrogen to ecosystems, Stable isotopes in ecology and environmental science, 2, 375-449, 2007.

Lee, H.-M., Henze, D. K., Alexander, B., and Murray, L. T.: Investigating the sensitivity of surface-level nitrate seasonality in Antarctica to primary sources using a global model, Atmospheric Environment, 89, 757-767, 2014.

Li, D., and Wang, X.: Nitrogen isotopic signature of soil-released nitric oxide (NO) after fertilizer application, Atmospheric Environment, 42, 4747-4754, <u>https://doi.org/10.1016/j.atmosenv.2008.01.042</u>, 2008.

Meusinger, C., Berhanu, T. A., Erbland, J., Savarino, J., and Johnson, M. S.: Laboratory study of nitrate photolysis in Antarctic snow. I. Observed quantum yield, domain of photolysis, and secondary chemistry, The Journal of chemical physics, 140, 244305, 2014.

Miller, D. J., Wojtal, P. K., Clark, S. C., and Hastings, M. G.: Vehicle NOx emission plume isotopic signatures: Spatial variability across the eastern United States, Journal of Geophysical Research: Atmospheres, 122, 4698-4717, 2017.

Miller, D. J., Chai, J., Guo, F., Dell, C. J., Karsten, H., and Hastings, M. G.: Isotopic Composition of In Situ Soil NOx Emissions in Manure-Fertilized Cropland, Geophysical Research Letters, 45, 12,058-012,066, 2018.

Morin, S., Savarino, J., Bekki, S., Gong, S., and Bottenheim, J.: Signature of Arctic surface ozone depletion events in the isotope anomaly (Δ 17 O) of atmospheric nitrate, 2007.

Pasteris, D., McConnell, J. R., Edwards, R., Isaksson, E., and Albert, M. R.: Acidity decline in Antarctic ice cores during the Little Ice Age linked to changes in atmospheric nitrate and sea salt concentrations, Journal of Geophysical Research: Atmospheres, 119, 5640-5652, 2014.

Shi, G., Buffen, A., Hastings, M., Li, C., Ma, H., Li, Y., Sun, B., An, C., and Jiang, S.: Investigation of post-depositional processing of nitrate in East Antarctic snow: isotopic constraints on photolytic loss, re-oxidation, and source inputs, Atmospheric Chemistry and Physics, 15, 9435-9453, 2015.

Shi, G., Buffen, A., Ma, H., Hu, Z., Sun, B., Li, C., Yu, J., Ma, T., An, C., and Jiang, S.: Distinguishing summertime atmospheric production of nitrate across the East Antarctic Ice Sheet, Geochimica et Cosmochimica Acta, 231, 1-14, 2018.

Shi, G., Chai, J., Zhu, Z., Hu, Z., Chen, Z., Yu, J., Ma, T., Ma, H., An, C., and Jiang, S.: Isotope fractionation of nitrate during volatilization in snow: a field investigation in Antarctica, Geophysical Research Letters, 46, 3287-3297, 2019.

Weller, R., and Wagenbach, D.: Year-round chemical aerosol records in continental Antarctica obtained by automatic samplings, Tellus B: Chemical and Physical Meteorology, 59, 755-765, 2007.

Yu, Z., and Elliott, E. M.: Novel method for nitrogen isotopic analysis of soil-emitted nitric oxide, Environmental science & technology, 51, 6268-6278, 2017.

Zatko, M., Geng, L., Alexander, B., Sofen, E., and Klein, K.: The impact of snow nitrate photolysis on boundary layer chemistry and the recycling and redistribution of reactive nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem. Phys, 16, 2819-2842, 2016.

Deposition, recycling and archival of nitrate stable isotopes between the air-snow interface: comparison between Dronning Maud Land and Dome C, Antarctica

V. Holly L. Winton¹, Alison Ming¹, Nicolas Caillon², Lisa Hauge¹, Anna E. Jones¹, Joel Savarino², Xin 5 Yang¹, Markus M. Frey¹

¹British Antarctic Survey, Cambridge, CB3 0ET, UK ²University of Grenoble Alpes, CNRS, IRD, Grenoble INP, IGE, F-38000 Grenoble, France *Correspondence to*: V. Holly L. Winton (<u>viewin@bas.ae.ukholly.winton@vuw.ac.nz</u>)

Abstract

- 10 The nitrogen stable isotopic composition in nitrate (δ¹⁵N-NO₃⁻) measured in polar ice cores has the potential to provide constraints on past ultraviolet (UV) radiation and thereby total column ozone (TCO) due to the sensitivity of nitrate (NO₃⁻) photolysis to UV radiation. However, understanding the transfer of reactive nitrogen at the air-snow interface in Polar Regions is paramount for the interpretation of ice core records of δ¹⁵N-NO₃⁻ and NO₃⁻ mass concentrations. As NO₃⁻ undergoes a number of post-depositional processes before it is archived in ice cores, site-specific observations of δ¹⁵N-NO₃⁻ and air-snow transfer
- 15 modelling are necessary in order to understand and quantify the complex photochemical processes at play. As part of the Isotopic Constraints on Past Ozone Layer Thickness in Polar Ice (ISOL-ICE) project, we report new measurements of $NO_3^$ mass concentration and $\delta^{15}N$ -NO₃⁻ in the atmosphere, skin layer (operationally defined as the top 5 mm of the snow pack), and snow pit depth profiles at Kohnen Station, Dronning Maud Land (DML), Antarctica. We compare the results to previous studies and new data, presented here, from Dome C, East Antarctic Plateau. Additionally, we apply the conceptual one-
- 20 dimensional model of TRansfer of Atmospheric Nitrate Stable Isotopes To the Snow (TRANSITS) to assess the impact of photochemical processes NO₃⁻ recycling on that drive the archival of δ¹⁵N-NO₃⁻ and NO₃⁻ mass concentrations archived in the snow and firmpack. We find clear evidence of NO₃⁻ photolysis at DML, and confirmation of previous theoretical, field and laboratory studies that UV-photolysis is driving our hypothesis that UV-photolysis is driving.NO₃⁻ recycling and redistribution at DML. Firstly, strong denitrification of the snow pack is observed through the δ¹⁵N-NO₃⁻ signature which evolves from the
- 25 enriched snow pack (-3 to 100 ‰), to the skin layer (-20 to 3 ‰), to the depleted atmosphere (-50 to -20 ‰) corresponding to mass loss of NO₃⁻ from the snow pack. Secondly, constrained by field measurements of snow accumulation rate, light attenuation (e-folding depth) and atmospheric NO₃⁻ mass concentrations, the TRANSITS model is able to reproduce our <u>observed</u> δ¹⁵N-NO₃⁻ observations in depth profiles. We find that NO₃⁻ is recycled three-two times before it is archived (i.e., below the photic zone) in the snow pack below 15 cm and within 0.75 years (i.e., below the photic zone). Mean annual aArchived δ¹⁵N-NO₃⁻ and NO₃⁻ mass concentration values are 50 ‰ and 60 ng g⁻¹ arespectively, at the DML site. We report an

Formatted: Not Superscript/ Subscript
Formatted: Not Superscript/ Subscript
Formatted: Not Superscript/ Subscript



e-folding depth of 2 - 5 cm for the DML site which is considerably lower than Dome C. NO₃⁻ photolysis is weaker at DML than at Dome C, due primarily to the higher DML snow accumulation rate; this results in a more depleted δ^{15} N-NO₃⁻ signature at DML than at Dome C. A reduced photolytic loss of NO₃⁻ at DML results in less enrichment of δ^{15} N-NO₃⁻ than at Dome C mainly due to the smaller e-folding depth but also due to the higher accumulation rate based on TRANSITS modelled

- 35 sensitivities, Even at a relatively low_snow accumulation_rate of 6 cm yr_1 (water equivalent; w.e.), the accumulation rate at DML is great enough to preserve the seasonal cycle of NO₃⁺ mass concentration and δ^{15} N-NO₃⁺, in contrast to Dome C where the <u>depth</u> profiles are smoothed due to <u>lstronger photochemistryonger exposure of surface snow layers to incoming UV</u> <u>radiation before burial</u>. TRANSITS sensitivity analysis of δ^{15} N-NO₃⁺ at DML highlights that the dominant factors controlling the archived δ^{15} N-NO₃⁺ signature are the <u>e-folding depth and</u> snow accumulation rate and <u>e-folding depth</u>, with a smaller role
- 40 from changes in the snowfall timing and TCOOC. Mean TRANSITS model sensitivities of archived δ¹⁵N-NO₃, at the DML_site are 100 ‰ for an e-folding depth change of 8 cm; 110 ‰ for an annual snow accumulation rate change of 8.5 cm yr⁻¹ (w.e.); 10 ‰ for a change in the dominant snow deposition season between winter and summer, and 10 ‰ for a TCO change of 100 DU. Here we set the framework for the interpretation of a 1000-year ice core record of δ¹⁵N-NO₃⁻ from DML. Ice core δ¹⁵N-NO₃⁻ records at DML will be less sensitive to changes in UV than at Dome C, however the higher snow accumulation
 45 rate and more accurate dating at DML allows for higher resolution δ¹⁵N-NO₃⁻ records.

1 Introduction

Nitrate (NO₂⁻) is a naturally occurring ion formed by the oxidation of nitrogen, and plays a major role in the global nitrogen eycle. It is one of the most abundant ions in Antarctic snow and is commonly measured in ice cores (e.g. Wolff, 1995). Nitrate
in polar ice provides constraints on past solar activity (Traversi et al., 2012), NO₃⁻ sources and the oxidative capacity of the atmosphere (Geng et al., 2017;Mulvaney and Wolff, 1993;Hastings et al., 2009;Hastings et al., 2004;McCabe et al., 2007;Savarino et al., 2007;Morin et al., 2008). However, NO₃⁻ is a non-conservative ion in snow, and due to post-depositional

- processes (e.g. Mulvaney et al., 1998;Zatko et al., 2016), the interpretation of NO₂⁻ concentration records from ice core records is challenging (Erbland et al., 2015). The recent development of the analysis of nitrogen isotopic composition of NO₂⁻ (8¹⁵N NO₃⁻) in snow, ice and aerosol provides a powerful means to understand the sources and processes involved in NO₂⁻ post-depositional processes, i.e., NO₃⁻ recycling at the interface between air and snow _______
 Primary sources of reactive nitrogen species to the Antarctic lower atmosphere and snow pack include the sedimentation of polar stratospheric clouds (PSC) in late winter (Savarino et al., 2007) and to a minor extent advection of oceanic methyl nitrate (CH₃NO₃) and peroxyacyl nitrates (PAN) (Jacobi et al., 2000;Jones et al., 1999;Beyersdorf et al., 2010;Lee et al., 2014). In
- 60 the stratosphere, NO₂⁻ is produced through the stratospheric oxidation of nitrous oxide (N₂O) from extra-terrestrial fluxes of energetic particles and solar radiation, whereas in the troposphere lightning and biomass burning provide background tropospheric reactive nitrogen species to the snow pack (Savarino et al., 2007;Wolff, 1995;Wagenbach et al., 1998). A local

2

Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript

Formatted: Subscript

secondary source of reactive nitrogen (nitrous acid (HONO), nitrogen oxides (NO_{*})) originates from post-depositional processes driven by sunlight leading to re-emission from the snow pack and subsequent deposition (Savarino et al., 2007;Frey et al., 2009).

65

Local nitrogen dioxide (NO2) emissions in Polar Regions are produced from NO2⁻ photolysis in the snow pack under sunlit conditions (Jones et al., 2001;Honrath et al., 1999;Oncley et al., 2004). Nitrate photolysis occurs at wavelengths (*λ*) = 290-345 nm with a maximum at 320 nm. Once NO2⁻ is produced by NO2⁻ photolysis, it is expected to have a lifetime in the polar troposphere of <1 day before it is oxidised to nitric acid (HNO2) at Dome C and South Pole (Davis et al., 2004b), and can then
 be redeposited to the skin layer (e.g. Mulvaney et al., 1998).

Photolysis rate J depends on the adsorption cross section of NO₂⁻, the quantum yield and actinic flux within the snow pack. Photochemical production of NO₂ is dependent on the NO₃⁻ concentration in the snow pack, the snow pack properties, and the intensity of solar radiation within the snow pack. The latter is sensitive to solar zenith angle and snow optical properties i.e. scattering and adsorption coefficients, which depends on snow density and morphology, and the light absorbing impurity

- 75 content (France et al., 2011;Erbland et al., 2015;Zatko et al., 2013). Recently, Zatko et al. (2016) found that the range of modelled NO_x fluxes from the snow pack to the overlaying air are similar in both Polar Regions due to the opposing effects of higher concentrations of both photolabile NO_x⁻ and light absorbing impurities (e.g. dust and black carbon) in Antarctica and Greenland respectively. At Concordia Station on Dome C in East Antarctica, the light penetration depth (e-folding depth) is ~10 cm for wind pack layers and ~20 cm for hoar layers (France et al., 2011). Based on the propagation of light into the snow
- 80 pack, the snow pack can be divided into three layers. The first layer is known as the skin layer (a few mm thick) where direct solar radiation is converted into diffuse radiation. The second layer is called the active photic zone (below the skin layer layer), where solar radiation is effectively diffuse and the intensity of the radiation decays exponentially (Warren, 1982). The third layer is called the archived zone (below the active photic zone), where no photochemistry occurs.
- Previous research has focused predominantly on the high elevation polar plateau (Dome C). Here, the exponential decay of NO₃⁻ mass concentrations in the snow pack and thus post-depositional processing of NO₃⁻ were attributed to either evaporation or ultra violet (UV) photolysis (Röthlisberger et al., 2000;Röthlisberger et al., 2002). The open-debate of which post-depositional process controlled NO₃⁻ mass concentrations in the snow pack led to the use of a new isotopic tool, the stable isotopic composition of nitrate (δ¹⁵N NO₃⁻) Blunier et al. (2005). More recently, theoretical (Frey et al., 2009), laboratory (Meusinger et al., 2014;Erbland et al., 2013;Erbland et al., 2015;Shi et al., 2019;Perhanu et al., 2014), and field (Erbland et al., 2015;Shi et al., 2019;Perhanu et al., 2014).
- 90 al., 2013;Frey et al., 2009;Shi et al., 2015) evidence show that NO₃⁻mass loss from the surface snow to the overlying atmosphere and its associated isotopic fractionation is driven by photolysis. The physical release or evaporation of NO₃⁻ is negligible (Erbland et al., 2013;Shi et al., 2019). between phases during evaporation condensation processes, A _ _ _ _ _ _ _ _ _ _ _ _ At Dome C, the large redistribution and net mass loss of NO₃⁻ below the skin layer and the simultaneous isotopic fractionation of δ¹⁵N-NO₃⁻⁻ in the snow pack indicate that post-depositional processes significantly modify the original NO₃⁻⁻ concentration
- 95 and δ¹⁵N-NO₃⁻ composition (Frey et al., 2009). Skin layer observations of δ¹⁵N-NO₃⁻ in the surface snow at Dome C show strong enrichment compared to the atmospheric δ¹⁵N-NO₃⁻ signature. Furthermore, snow pit profiles show an exponential

3

Formatted: Font:

decrease of NO3-concentration and an enrichment in the 8⁴⁵N-NO3-composition with depth (Erbland et al., 2013). Once NO4 is produced by NO2-photolysis, it is expected to have a lifetime in the polar troposphere of <1 day before it is oxidised to nitrie acid (HNO2) at Dome C and South Pole (Davis et al., 2004b), and can then be redeposited to the skin layer (e.g. Mulvaney et al 1998)

100

This research at Dome C laid the foundation for Erbland et al. (2015) to derive a conceptual model of UV-photolysis induced post-depositional processes of NO3⁻ at the air-snow interface. TRANSITS is a conceptual multi-layer 1D model which aims to represent NO2-recycling at the air-snow interface including processes relevant for NO2-snow photochemistry (UV-photolysis of NO2, emission of NO2, local oxidation, deposition of HNO2) and explicitly calculates NO2 mass concentrations and 845N

- 105 NO3-in snow. "Nitrate recycling" is the combination of NO* production from NO3-photolysis in snow, the subsequent atmospheric processing and oxidation of NOx to form atmospheric HNOs, the deposition (dry and/or wet) of a fraction of the HNO3, and the export of another fraction. In NO2⁻ recycling, the skin layer is an active component of the atmosphere. This recycling can occur multiple times before NOr is eventually archived below the active photic zone in ice cores (Davis et al., 2008;Erbland et al., 2015;Zatko et al., 2016;Sofen et al., 2014). We refer to atmospheric NO2-as the combination (i.e., total)
- 110 of HNO2 (gas phase) and particulate NO2 We refer to atmospheric NO2 as the combination (i.e., total) of HNO2 (gas phase) and particulate NO2 ... The desorption of HNO2 from the snow crystal reduces the NO2 concentration in the snow in coastal Antarctica (Mulvaney et al., 1998). The evaporation of HNO2 is a two-step process, which involves the recombination of NO2 HNO1 followed by a phase change to HNO2 (gas phase). First, theoretical estimates indicated that evaporation of HNO3 should preferentially remove ⁴⁵N from the snow and release to the atmosphere leading to depletion in 8⁴⁵N-NO3⁻ in the
- 115 residual snow pack (Frey et al., 2009). Furthermore, recent laboratory experiments showed that evaporation imposes a negligible fractionation of 8¹⁵N-NOx-(!!! INVALID CITATION !!! (Erbland et al., 2013;Shi et al., 2019)). However, we find that the snow pack is enriched in δ⁴⁵N-NO₂⁻ relative to the atmosphere at DML (Figs. 3 and 6) and at Dome C (section 4.3.2). This fractionation observed in field studies cannot therefore be explained by evaporation, and must be attributed to different processes. It therefore follows that evaporation must be only a minor process in the redistribution of NO2⁻ between atmosphere
- 120 and the snow pack above the Antarctic plateau. Nitrate evaporation from the snow pack has a ¹⁵ came of -0 as determined by two independent studies (!!! INVALID CITATION !!! (Erbland et al., 2013;Shi et al., 2019)). This indicates that during NO₄⊤evaporation, the air above the snow is not replenished
 - and thus there is only a small NO₂-mass loss. The isotopic fractionation of NO₂-evaporation is negligible across most of Antarctica at cold temperatures of < 24 °C (Shi et al., 2019) which is the case for DML. However, evaporation of NOg at
- 125 warmer temperatures (4 °C) depletes the heavy isotopes of NO₂-remaining in the snow, and decreases the $\delta^{\pm S}$ NO₂-and the remaining snow by a few ‰ contrary to isotope effects of photolysis. In comparison, fractionation constants associated with laboratory studies and field observations of NO₄⁻photolysis are large: ⁴⁵Gapp = -34 ‰ (Berhanu et al., 2014;Meusinger et al., 2014) and 54 <-45 emp <- 60 % (Frey et al., 2009; Erbland et al., 2013), respectively. The negative fractionation constant obtained from photolysis implies that the remaining NO₂⁻ in the skin layer snow is enriched in δ^{45} N-NO₂⁻. In turn, the
- atmosphere is left with the source of NO₂ that is highly depleted in δ⁴⁵N-NO₂. This enrichment (depletion) is exactly what we 130

4

Formatted: Highlight Formatted: Highlight

Formatted: Highlight

observe in the snow pack (atmosphere) at DML (Figs. 4 and 6). The marked difference in values from the evaporation experiments and those observed in snow at Dome C allows us to separate out the isotopic signature of evaporation and photolysis processes:

Year round measurements of NO₄⁻⁻mass concentrations and ö⁴⁵N NO₄⁻⁻ in the skin layer and atmosphere at Dome C have provided insights into the annual NO₄⁻⁻ cycle in Antarctica (Fig. 1) Erbland et al. (2013). In the early winter, the stratosphere undergoes denitrification via formation of PSC. As PSC sediment slowly, there is a delay between the maximum stratospheric NO₄⁻⁻ concentration and the maximum NO₄⁻⁻ concentration deposited in the skin layer in late winter (Mulvaney and Wolff, 1993;Savarino et al., 2007). In spring, surface UV increases and initiates photolysis-driven post-depositional processes, which
redistribute NO₄⁻⁻ between the snow pack and overlying air throughout the sunlit summer season. This results in the 8⁴⁵N-NO₄⁻⁻ isotopic enrichment of the NO₄⁻⁻ skin layer reservoir, and maximum atmospheric NO₄⁻⁻ mass concentrations in October-

November. In summer, NO₂-resembles a strongly asymmetric distribution within the atmosphere-snow column with the bulk residing in the skin layer and only a small fraction in the atmospheric column above.

Over longer time scales, UV-driven post-depositional processing of NO₂⁻ is also driven by changes in the degree of postdepositional loss of NO₂⁻ with greater NO₂⁻ loss during the glacial period relative to the Holocene. The observed glacialinterglacial difference in post-depositional processing of NO₂⁻ is dominated by variations in snow accumulation rate (Geng et al., 2015).

Nitrate is not preserved in the snow pack at sites with very low snow accumulation rates (i.e., Dome C: 2.5-3 cm yr⁻¹) because snow layers remain close to the surface and in contact with the overlaying atmosphere for a relatively long time enhancing the

- 150 effect of post-depositional processes. At sites with low snow accumulation rates, the source signature of δ¹⁵N-NO₂⁻ is erased by post-depositional processe. Therefore, photolysis induced NO₃⁻ loss and δ⁴⁵N-NO₂⁻ fractionation is dependent on snow accumulation. Three distinct transects from coastal Antarctica to the East Antarctic Plateau show that NO₃⁻ fractionation is strongest with decreasing snow accumulation (Shi et al., 2018;Erbland et al., 2013;Noro et al., 2018). Skin layer NO₃⁻ mass concentrations are significantly higher at low snow accumulation sites, for example ~160 ng g⁺ (winter) to 1400 ng g⁺
- 155 (summer) at Dome C compared to 50 ng g⁺ (winter) to 300 ng g⁺ (summer) at Dumont d'Urville (DDU) on the Antarctice coast. (Shi et al., 2015;Erbland et al., 2013)Furthermore, the strong inverse linear relationship between NO₃⁻ concentration and accumulation rate was revealed in a composite of seven ice cores across Dronning Maud Land (DML) Pasteris et al. (2014).<u>Over longer time scales, UV driven post depositional processing of NO₃⁻ is also driven by changes in the degree of post depositional loss of NO₂⁻ with greater NO₃⁻ loss during the glacial period relative to the Holocene. The observed glacial interglacial difference in post depositional processing of NO₃⁻ is dominated by variations in snow accumulation rate (Geng et al., 2015).</u>

Yet, NO₃⁻ photolysis leaves its own process specific imprint in the snow pack (Shi et al., 2019;Erbland et al., 2015;Erbland et al., 2013), which opens up the possibility to use δ^{45} N-NO₃⁻ to infer past surface UV variability (Frey et al., 2009). However,

- 165 NO₃⁻ photolysis rates in snow depend on a number of site-specific factors as does the degree of photolytic isotopic fractionation of NO₃⁻ eventually preserved in ice cores (Erbland et al., 2013;Berhanu et al., 2014). These factors need to be quantitatively understood at a given ice core site to enable quantitative interpretation of ice core records. Here, we carry out a comprehensive study of the air-snow transfer of NO₃⁻ at Kohnen Station in DML, East Antarctica through δ¹⁵N-NO₃⁻ measurements in the atmosphere, skin layer and snow pits, and compare the observations to Dome C. Due to the previous research outlined above, we assume that the photolysis is the dominant driver of NO₃⁻-post depositional processes, and later assess the validity of this
- 170 we assume that the photolysis is the dominant driver of NO₄^T post_depositional processes, and later assess the validity of this_ this assumption (section 4.3).-We apply the Transfer of Atmospheric Nitrate Stable Isotopes To the Snow (TRANSITS) model (Erbland et al., 2015) to i) understand how NO₄^T mass concentrations and δ¹⁵N-NO₄^T are archived in deeper snow and ice layers, and ii) investigate the sensitivity of changes in the past snow accumulation rate, snowfall timing, e-folding depth, and TCO on the δ¹⁵N-NO₄^T signature. In order to interpret this novel UV proxy, it is paramount to understand the air-snow transfer processes
- 175 specific to an ice core site, and how δ¹⁵N-NO₂⁻ is archived in the deeper snow and ice layers (Geng et al., 2015;Morin et al., 2009;Erbland et al., 2015). Within the framework of the Isotopic Constraints on Past Ozone Layer Thickness in Polar Ice (ISOL-ICE) project, this study provides a basis for the interpretation of δ¹⁵N-NO₃⁻ from a 1000 year ice core recovered in 2016/17 at Kohnen Station₃

1 Introduction

- 180 Nitrate (NO₃⁻) is a naturally occurring ion, and plays a major role in the global nitrogen cycle. It is one of the most abundant ions in Antarctic snow and is commonly measured in ice cores (e.g. Wolff, 1995). Nitrate in polar ice provides constraints on past solar activity (Traversi et al., 2012), NO₃⁻ sources and the oxidative capacity of the atmosphere (Geng et al., 2017;Mulvaney and Wolff, 1993;Hastings et al., 2009;Hastings et al., 2004;McCabe et al., 2007;Savarino et al., 2007;Morin et al., 2008). However, NO₃⁻ is a non-conservative ion in snow, and due to post-depositional processes (e.g. Mulvaney et al., 2008).
- 185 1998;Zatko et al., 2016), the interpretation of NO₃⁻ mass concentration records from ice core records is challenging (Erbland et al., 2015). The recent development of the analysis of the nitrogen isotopic composition of NO₃⁻ (δ¹⁵N-NO₃⁻) in snow, ice and aerosol provides a powerful means to understand the sources and processes involved in NO₃⁻ post-depositional processes, i.e., NO₃⁻ recycling at the interface between air and snow.
- Primary sources of reactive nitrogen species to the Antarctic lower atmosphere and snow pack include the sedimentation of
 polar stratospheric clouds (PSC) in late winter (Savarino et al., 2007) and, to a minor extent, advection of oceanic methyl nitrate (CH₃NO₃) and peroxyacyl nitrates (PAN) (Jacobi et al., 2000;Jones et al., 1999;Beyersdorf et al., 2010), in addition to tropospheric transport of inorganic NO₃⁻ from lightning, biomass burning and soil emissions (Lee et al., 2014). In the stratosphere, NO₃⁻ is produced through the stratospheric oxidation of nitrous oxide (N₂O) from extra-terrestrial fluxes of energetic particles and solar radiation (Savarino et al., 2007;Wolff, 1995;Wagenbach et al., 1998). A local secondary source
- 195 of reactive nitrogen (nitrous acid (HONO), nitrogen oxides (NO_x)) originates from post-depositional processes driven by sunlight leading to re-emission from the snow pack and subsequent deposition (Savarino et al., 2007;Frey et al., 2009).

6

Formatted: Not Superscript/ Subscript

Formatted: Subscript

Local nitrogen dioxide (NO_2) emissions in Polar Regions are produced from NO_3^- photolysis in the snow pack under sunlit conditions (Jones et al., 2001;Honrath et al., 1999;Oncley et al., 2004). Once NO_8 is produced by NO_3^- photolysis, it is expected to have a lifetime in the polar troposphere of <1 day before it is oxidised to nitric acid (HNO₃) at Dome C and South Pole

- 200 (Davis et al., 2004b), and can then be redeposited to the skin layer (e.g. Mulvaney et al., 1998). Nitrate photolysis occurs at wavelengths (λ) = 290 - 345 nm with a maximum at 320 nm. Photolysis rate (J) depends on the adsorption cross section of NO₃⁻, the quantum yield, and actinic flux within the snow pack. Photochemical production of NO₂ is dependent on the NO₃⁻, mass concentration in the snow pack, the snow pack properties, and the intensity of solar radiation within the snow pack. The latter is sensitive to solar zenith angle and snow optical properties, i.e., scattering and adsorption coefficients, which depends
- 205 on snow density and morphology, and the light absorbing impurity content (e.g. dust and black carbon) (France et al., 2011;Erbland et al., 2015;Zatko et al., 2013). Model results from Zatko et al. (2016) suggest that the range of modelled NO_x fluxes from the snow pack to the overlaying air are similar in both Polar Regions due to the opposing effects of higher concentrations of both photolabile NO₃⁻ and light absorbing impurities in Antarctica and Greenland, respectively. At Concordia Station on Dome C in East Antarctica, the light penetration depth (e-folding depth) is ~10 cm for wind pack layers and ~20
- 210 cm for hoar layers (France et al., 2011). Based on the propagation of light into the snow pack, the snow pack can be divided into three layers. The first layer is known as the skin layer (5 mm thick) where direct solar radiation is converted into diffuse radiation. The second layer is called the active photic zone (below the skin layer layer), where solar radiation is effectively diffuse and the intensity of the radiation decays exponentially (Warren, 1982). The third layer is called the archived zone (below the active photic zone), where no photochemistry occurs.
- 215 Previous research has focused predominantly at Dome C on the high elevation polar plateau. Here, the exponential decay of NO₃⁻ mass concentrations in the snow pack, and thus post-depositional processing of NO₃⁻, were attributed to either evaporation or ultraviolet (UV)-photolysis (Röthlisberger et al., 2000;Röthlisberger et al., 2002). The open debate of which post-depositional process controlled NO₃⁻ mass concentrations in the snow pack led to the use of a new isotopic tool, the nitrogen isotopic composition of NO₃⁻ (δ¹⁵N-NO₃⁻; Blunier et al. (2005). More recently, theoretical (Frey et al., 2009), laboratory
- 220 (Meusinger et al., 2014;Erbland et al., 2013;Erbland et al., 2015;Shi et al., 2019;Berhanu et al., 2014), and field (Erbland et al., 2013;Frey et al., 2009;Shi et al., 2015) evidence show that NO₃⁺ mass loss from the surface snow to the overlying atmosphere and its associated isotopic fractionation is driven by photolysis. Fractionation constants, which assume a Rayleigh single loss and irreversible process of NO₃⁺ removal from the snow between phases during evaporation-condensation processes, have been calculated to separate the isotopic signature of evaporation and photolysis processes. As this approach may
- 225 oversimplify the processes occurring at the air-snow interface, Erbland et al. (2013) referred to the quantity as an "apparent" fractionation constant ($^{15}\varepsilon_{app}$). Thus, the apparent fractionation constant represents the integrated isotopic effect of the processes involving NO₃⁻ in the surface of the snow pack and in the lower atmosphere. Nitrate evaporation from the snow pack has a $^{15}\varepsilon_{app}$ of ~0 as determined by two independent studies (Erbland et al., 2013;Shi et al., 2019). This indicates that during NO₃⁻ evaporation, the air above the snow is not replenished and thus there is only a small NO₃⁻ mass loss. In comparison,
- 230 fractionation constants associated with laboratory studies and field observations of NO₃⁻ photolysis are large: ¹⁵ε_{app} = -34 ‰

7

Formatted: Not Superscript/ Subscript

(Berhanu et al., 2014; Meusinger et al., 2014) and -54<¹⁵_{Egpp} <-60 ‰ (Frey et al., 2009; Erbland et al., 2013), respectively. The negative fractionation constant obtained from photolysis implies that the remaining NO₃⁻ in the skin layer snow is enriched in δ^{15} N-NO₃⁻. In turn, the atmosphere is left with the source of NO₃ that is highly depleted in δ^{15} N-NO₃⁻. It follows that evaporation of NO₃⁻ is negligible on high-elevation Antarctic sites (Erbland et al., 2013; Shi et al., 2019).

- At Dome C, the large redistribution and net mass loss of NO₃⁺ below the skin layer and the simultaneous isotopic fractionation of δ¹⁵N-NO₃⁻ in the snow pack indicates that post-depositional processes significantly modify the original NO₃⁻ mass concentration and δ¹⁵N-NO₃⁻ composition (Frey et al., 2009). Skin layer observations of δ¹⁵N-NO₃⁻ in the surface snow at Dome C show strong enrichment compared to the atmospheric δ¹⁵N-NO₃⁻ signature. Furthermore, snow pit profiles display an exponential decrease of NO₃⁺ mass concentration and an enrichment in the δ¹⁵N-NO₃⁻ composition with depth (Erbland et al., 2013).
 - This research at Dome C laid the foundation for Erbland et al. (2015) to derive a conceptual model of UV-photolysis induced post-depositional processes of NO₃⁻ at the air-snow interface. Transfer of Atmospheric Nitrate Stable Isotopes To the Snow (TRANSITS) is a conceptual multi-layer 1D model which aims to represent NO₃⁻ recycling at the air-snow interface including processes relevant for NO₃⁻ snow photochemistry (UV-photolysis of NO₃⁻, emission of NO₃, local re-oxidation, deposition of
- 245 HNO₃) and explicitly calculates NO₃⁻ mass concentrations and δ¹⁵N-NO₃⁻ in snow. The term "NO₃⁻ recycling" refers to the following processes. Nitrate on the surface of a snow crystal can be lost from the snow pack (Dubowski et al., 2001), either by UV-photolysis or evaporation. UV-photolysis produces NO, NO₂ and HONO while only HNO₃ can evaporate. Both of these processes produce reactive nitrogen that can be released from snow crystal into the interstitial air and rapidly transported out of the snow pack to the overlaying air via wind pumping (Zatko et al., 2013;Jones et al., 2000;Honrath et al., 1999;Jones et al.,
- 2001). Here, NO₂ is either oxidised to HNO₃, which undergoes wet or dry deposition back to the skin layer within a day, or is transported away from the site (Davis et al., 2004a). If HNO₃ is re-deposited to the skin layer, it is available for NO₃⁻ photolysis and/or evaporation again. Any locally produced NO₂ and NO₃⁻ that is transported away from the site of emission represents a loss of NO₃⁻ from the snow pack. Nitrate recycling can occur multiple times before NO₃⁻ is eventually archived below the active photic zone in ice cores (Davis et al., 2008;Erbland et al., 2015;Zatko et al., 2016;Sofen et al., 2014).
- 255 Year round measurements of NO₃⁻ mass concentrations and δ¹⁵N-NO₃⁻ in the skin layer and atmosphere at Dome C have provided insights into the annual NO₃⁻ cycle in Antarctica (Fig. 1; Erbland et al. (2013). In the early winter, the stratosphere undergoes denitrification via formation of PSC. As PSC sediment slowly, there is a delay between the maximum stratospheric NO₃⁻ mass concentration and the maximum NO₃⁻ mass concentration deposited in the skin layer in late winter (Mulvaney and Wolff, 1993;Savarino et al., 2007). In spring, surface UV increases and initiates photolysis-driven post-depositional processes,
- 260 which redistribute NO₃⁻ between the snow pack and overlying air throughout the sunlit summer season. This results in the <u>δ¹⁵N-NO₃⁻</u> isotopic enrichment of the NO₃⁻ skin layer reservoir, and maximum atmospheric NO₃⁻ mass concentrations in <u>October-November</u>. In summer, NO₃⁻ resembles a strongly asymmetric distribution within the atmosphere-snow column with the bulk residing in the skin layer and only a small fraction in the atmospheric column above.

- Nitrate is not preserved in the snow pack at sites with very low snow accumulation rates (i.e., Dome C: 2.5 3 cm yr⁻¹) because
 snow layers remain close to the surface and in contact with the overlaying atmosphere for a relatively long time enhancing the effect of post-depositional processes. At sites with low snow accumulation rates, the source signature of δ¹⁵N-NO₃⁻ is erased by post-depositional processe. Therefore, photolysis induced NO₃⁻ loss and δ¹⁵N-NO₃⁻ fractionation is dependent on snow accumulation. Three distinct transects from coastal Antarctica to the East Antarctic Plateau show that NO₃⁻ fractionation is strongest with decreasing snow accumulation (Shi et al., 2018;Erbland et al., 2013;Noro et al., 2018). Skin layer NO₃⁻ mass concentrations are significantly higher at low snow accumulation sites, for example ~160 ng g⁻¹ (winter) to 1400 ng g⁻¹ (summer) at Dome C compared to 50 ng g⁻¹ (winter) to 300 ng g⁻¹ (summer) at Dumont d'Urville (DDU) on the Antarctic coast. In contrast to low snow accumulation sites, NO₃⁻ loss is less pronounced on the coast and seasonal cycles of NO₃⁻ mass concentration and δ¹⁵N-NO₃⁻ are preserved in the snowpack (Shi et al., 2015;Erbland et al., 2013). Erbland et al. (2013) suggest
- that NO₃⁻ loss at the coast reflects both photolysis and evaporation processes, while Shi et al. (2015) proposes that NO₃⁻ loss
 at the coast cannot be fully explained by local post-deposition processes and that seasonal cycles in the snowpack reflect stratospheric and troposphere NO₃⁻ sources during the cold and warm seasons respectively. Furthermore, the strong inverse linear relationship between ice core NO₃⁻ mass concentration and accumulation rate was revealed in a composite of seven ice cores across Dronning Maud Land (DML; Pasteris et al. (2014). Over longer time scales, UV-driven post-depositional processing of NO₃⁻ is also driven by changes in the degree of post-depositional loss of NO₃⁻ with greater NO₃⁻ loss during the
- 280 glacial period relative to the Holocene. The observed glacial-interglacial difference in post-depositional processing of NO₃⁻ is dominated by variations in snow accumulation rate (Geng et al., 2015). Yet, NO₃⁻ photolysis leaves its own process-specific imprint in the snow pack (Shi et al., 2019;Erbland et al., 2015;Erbland et

al., 2013), which opens up the possibility to use δ^{15} N-NO₃⁻ to infer past surface-UV variability (Frey et al., 2009). However, NO₃⁻ photolysis rates in snow depend on a number of site-specific factors as does the degree of photolytic isotopic fractionation

- 285 of NO_3^- eventually preserved in ice cores (Erbland et al., 2013;Berhanu et al., 2014). These factors need to be quantitatively understood at a given ice core site to enable quantitative interpretation of ice core records. Here, we carry out a comprehensive study of the air-snow transfer of NO_3^- at Kohnen Station in DML, East Antarctica through $\delta^{15}N-NO_3^-$ measurements in the atmosphere, skin layer and snow pits, and compare the observations to Dome C. Due to the previous research outlined above, we assume that the photolysis is the dominant driver of NO_3^- post-depositional processes, and later assess the validity of this
- 290 this assumption (section 4.3). We apply the TRANSITS model (Erbland et al., 2015) to i) understand how NO_3^- mass concentrations and $\delta^{15}N-NO_3^-$ values are archived in deeper snow and ice layers, and ii) investigate the sensitivity of changes in the past snow accumulation rate, snowfall timing, e-folding depth of the snow photic zone, and total column ozone (TCO) on the $\delta^{15}N-NO_3^-$ signature. In order to interpret this novel UV proxy, it is paramount to understand the air-snow transfer processes specific to an ice core site, and how $\delta^{15}N-NO_3^-$ is archived in the deeper snow and ice layers (Geng et al., 2015;Morin
- 295 et al., 2009;Erbland et al., 2015). Within the framework of the Isotopic Constraints on Past Ozone Layer Thickness in Polar Ice (ISOL-ICE) project, this study provides a basis for the interpretation of δ¹⁵N-NO₃⁻ from a 1000-year ice core recovered in 2016/17 at Kohnen Station.

2 Methods

The ISOL-ICE project aims to understand natural causes of past TCO variability by i) investigating thean air-snow exchange
processesstudy of NO₃⁺ to enable the interpretation of ice core records of NO₃⁺ and δ¹⁵N-NO₃⁻, ii) reconstructing a 1000-year record of UV using a new ice core proxy based on δ¹⁵N-NO₃⁻ (Ming et al., submitted; Winton et al., 2019a), and iii) numerical modelling of the natural causes of TCO variability. In the air snow-transfer study presented here, we report new atmospheric, skin layer and snow pit NO₃⁻ mass concentration and δ¹⁵N-NO₃⁻ observations from DML, and compare them to new and published (Erbland et al., 2015;Erbland et al., 2013;Frey et al., 2009) observations from Dome C. Published data from Dome C comprises year round atmospheric and skin layer measurements from 2009-2010 (Erbland et al., 2013), and multiple snow pit profiles (Erbland et al., 2013;Frey et al., 2009). New data from Dome C encompasses an We present a new extended time

series at Dome C of <u>year roundyear-round</u> atmospheric and skin layer NO₃⁻ mass and concentration, and δ^{15} N-NO₃⁻ from 2011 - 2015.

Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript

2.1 Study sites

310 The ISOL-ICE campaign was carried out at the summer only, continental Kohnen Station where the deep European Project for Ice Coring in Antarctica (EPICA) Dronning Maud Land (EDML; 75°00' S, 0°04' E; 2982 m a.s.l.; https://www.awi.de/en/expedition/stations/kohnen-station.html) ice core was recovered in 2001 - 2006 to a depth of ~2800 m (Wilhelms et al., 2017). As part of the ISOL-ICE campaign, a new ice core (ISOL-ICE; Winton et al. (2019a)) was drilled 1 km from the EDML borehole (Figs. 2a-b). In addition, the ISOL-ICE air-snow transfer study site was located ~200 m from 315 the EDML ice core site (Fig. 2c). Here we compare two ice core drilling sites in Antarctica: Kohnen Station (referred to as DML henceforth) and EPICA-Dome C (75°05'59" S, 123°19'56" E;)(Fig. 2). Both sites are similar in terms of the latitude and therefore in terms of radiative forcing at the top of the atmosphere (Table 1). Satellite images of TCO over Antarctica show that the lowest annual TCO values are centred over the South Pole region encompassing DML and usually Dome C although the spatial variability is significant from year to year (https://ozonewatch.gsfc.nasa.gov/). The sites are different in terms of 320 their location with respect to moisture source, elevation and precipitation regime. The DML site is situated ~550 km from the ice shelf edge, is subject to cyclonic activity and receives ~80 % of its precipitation from frontal clouds (Reijmer and Oerlemans, 2002). While Dome C is more remote (~1100 km from the coast) and diamond dust is the dominant form of precipitation. The annual snow accumulation rate also differs between the sites; while both sites have exceptionally low accumulation compared to the coast, DML (annual mean: 6 cm yr⁻¹ (water equivalent (+w.e.)); Hofstede et al. (2004); Sommer 325 et al. (2000)) receives more than double that of Dome C (annual mean: 2.5 cm yr⁻¹ (w.e.)); Le Meur et al. (2018). Throughout the study we refer to our sampling site as "DML".

2.2 Snow and aerosol sampling

Daily skin layer samples, (which we operationally defined as the top 5 mm of the snow pack following Erbland et al. (2013),) were collected from the DML site (Fig. 2c) in January 2017 during the ISOL-ICE ice core drilling and atmospherie monitoringair-snow transfer campaign. To prevent contamination from the nearby Kohnen Station, snow samples were collected from the "flux site" within the station's designated clean air sector (defined as 45° from both ends of the station building) located ~1 km from the station (Fig. 2c). The skin layer samples were collected daily between 31 December 2016 and 29 January 2017 from a designated sampling site each day-during the campaign (75°00.184' S, 000°04.527' E; Fig. 2c). To 335 determine the spatial variability of NO₃- in the skin layer at the flux site, an additional five skin layer samples were collected in a ~2500 m² area of the flux site (75°00.161' S - 000°04.441' E, 75°00.175' S - 000°04.518' E; Fig. 2c).

- Adjacent to the skin layer samples, snow was sampled from a 1.6 m snow pit at the flux site (snow pit B; Fig. 2c) and a 2 m snow pit at the "ice core" site (snow pit A; Fig. 2b). Two parallel profiles were sampled, i) for i) major ion mass concentrations (including NO₃⁻) collected in pre-washed 50 mL Corning® centrifuge tubes at 3 cm resolution by inserting the tube directly
 into the snow face, and ii) for stable NO₃⁻ isotope analysis collected in Whirl-pak® bags at 2 cm resolution using a custom-
- made stainless-steelstainless-steel tool. Exposure blanks (following the same method as the samples by opening the tube/ Whirl-pak® bag at the field site but not filling the sample container with snow) were also collected for both types of samples. Snow density and temperature were measured every 3 cm, and a visual log of snow pit stratigraphy was recorded.
- Daily aerosol filters were collected using high-volume aerosol samplers custom-built at the Institute of Environmental
 Geosciences (IGE), University of Grenoble-Alpes, France described previously (Frey et al., 2009; Erbland et al., 2013). The
 high-volume aerosol sampler collected atmospheric aerosol on glass fibre filters (Whatman GF/A filter sheets; 20.3 × 25.4 cm) at an average flow rate of 1.2 m³ min⁻¹ at standard temperature and pressure (STP; temperature: 273.15 K; pressure: 1 bar) to
 determine the emass concentration and isotopic composition of atmospheric NO₃⁻. It is assumed that the atmospheric NO₃⁻ collected on glass fibre filters represents the sum of atmospheric particulate NO₃⁻ and HNO₃ (gas phase). The bulk of HNO₃
- 350 present in the gas phase is most likely adsorbed to aerosols on the filter, as described previously (Frey et al., 2009). Following the terminology of Erbland et al. (2013), we refer to "atmospheric NO₃" as the combination (i.e., total) of HNO₃ (gas phase) and particulate NO₃" and is represented by the NO₃" mass concentrations measured on aerosol filters.

The high-volume <u>aerosol</u> sampler was located 1 m above the snow surface at the flux site at the DML site (Fig. 2c), where a total of 35 aerosol filters were sampled daily between 3 and 27 January 2017. In addition, we coordinated an intensive 4-hour sampling campaign in phase with Dome C, East Antarctica (Fig. 2) between 21 and 23 January 2017. At Dome C, <u>a</u> high-volume <u>aerosol</u> sampler <u>wais</u> located on the roof of the atmospheric shelter (6 m above the snow surface), where a total of 12 samples were collected. At DML, loading and changing of aerosol collection substrates was carried out in a designated clean area. Aerosol laden filters were transferred into individual double zip-lock plastic bags immediately after collection and stored frozen until analysis at the British Antarctic Survey (BAS; major ions) and IGE (NO₃⁻ isotopic composition). For the

360 atmospheric NO₃⁻ work, three types of filter blanks were carried out; i) laboratory filter blanks (n = 3; Whatman GF/A filters that underwent the laboratory procedures without going into the field), ii) procedural filter blanks (DML: n = 4; Dome C: n = 1; filters that had been treated as for normal samples but which were not otherwise used; once a week, during daily filter change-over, a procedural blank filter was mounted in the aerosol collector for 5 min without the collector pump in operation – this type of filter provides an indication of the operational blank associated with the sampling procedure), and iii) 24 h exposure filter blanks sampled at the beginning and end of the field campaign (DML: n = 2; Dome C: n = 1; filters treated like

a procedural blank but left in the collector for 24 h-without switching the collector on). All samples were kept frozen below - 20 °C during storage and transport prior to analysis.

In addition, skin layer and aerosol samples whave beenere sampled continuously at Dome C over the period 2009-2015 following Erbland et al. (2013);Frey et al. (2009). The sampling resolution for skin layer wais every 2-4 days, and weekly for aerosol samples. Data from 2009-2010 have previously been published by Erbland et al. (2013), and we report the 2011_2015 data here (Fig. 1).

2.3 Major ion mass concentrations in snow and aerosol

Aerosol Atmospheric NO₃⁻ and other major ions were extracted in 40 mL of ultra-pure water (resistivity of 18.2 MΩ; Milli-Q water) by centrifugation using Millipore Centricon® Plus-70 Filter Units (10 kD filters) in a class-100 clean room at the BAS.
Major ion mass concentrations in DML snow samples were determined in an aliquot of melted snow from skin layer and snow pit samples, and aerosol extracts by suppressed ion chromatography (IC) using a DionexTM ICS-4000 Integrated Capillary HPICTM System ion chromatograph. A suite of anions, including NO₃⁻, chloride (Cl⁻), methanesulfonic acid (MSA) and sulphate (SO₄²⁻), were determined using an AS11-HC column and a CES 500 suppressor. Cations, including sodium (Na⁺), were determined using a CS12A column and a CES 500 suppressor. During the course of the sample sequence, instrumental blank solutions and certified reference materials (CRM; ERM-CA616 groundwater standard and ERM-CA408 simulated

rainwater standard; Sigma-Aldrich) were measured regularly for quality control and yielded an accuracy of 97 % for NO₃⁻. Nitrate mass concentrations in Dome C samples were determined by colorimetry at IGE following the procedure described in Frey et al. (2009). Blank concentrations for exposure blank, procedural blank and laboratory blank and detection limits are reported in Table S1. The non-sea-salt sulphate (nss-SO₄²⁻) fraction of SO₄²⁻ was obtained by subtracting the contribution of sea-salt-derived SO₄²⁻ from the measured SO₄²⁻ mass concentrations (nss-SO₄²⁻ = SO₄²⁻ - 0.252 × Na⁺, where Na⁺ and SO₄²⁻ are

the measured concentrations in snow pit samples and 0.252 is the SO₄²⁻/Na⁺ ratio in bulk seawater (Keene et al., 1986).

2.4 Nitrate isotopic composition in snow and aerosol

390

Samples were shipped frozen to IGE where the NO₃⁻ isotope analysis was performed. The denitrifier method was used to determine the stable NO₃⁻ isotopic composition in samples at IGE following Morin et al. (2008). Briefly, samples were preconcentrated due to the low NO₃⁻ mass concentrations found in the atmosphere and snow over Antarctica. To obtain 100 nmol of NO₃⁻ required for NO₃⁻ isotope analysis, the meltwater of snow samples and aerosol extracts were were sorbed onto 0.3 mL

of anion exchange resin (AG1-X8 chloride form; Bio-Rad) and eluted with 5 x 2 mL of 1 M NaCl (high purity grade 99.0 %; American Chemical Society (ACS grade); AppliChem Panreac) following Silva et al. (2000). Recovery tests yielded 100 % recovery of NO₃⁻ (Frey et al., 2009;Erbland et al., 2013). Once pre-concentrated, NO₃⁻ wais converted to N₂O gas by

- 395 denitrifying bacteria, *Pseudomonas aureofaciens*. The N₂O wais split into O₂ and N₂ on a gold furnace heated to 900 °C followed by gas chromatographic separation and injection into the isotope ratio mass spectrometer (IRMS) for duel O and N analysis using a Thermo Finnigan[™] MAT 253 IRMS equipped with a GasBench II[™] and coupled to an in-house-built NO₃⁻ interface (Morin et al., 2009).
- Certified reference materials (IAEA USGS-32, USGS-34 and USGS-35; Böhlke et al., 1993; Böhlke et al., 2003) were
 prepared (matrix match 1 M NaCl in identical water isotopic composition as samples; ACS grade) and subject to the same analytical procedures as snow and aerosol samples. The nitrogen isotopic ratio was referenced against N₂-Air (Mariotti, 1983). We report ¹⁵N/¹⁴N of NO₃⁻ (δ¹⁵N-NO₃⁻) as δ-values following Eq. (1).

 δ^{15} N-NO₃⁻ = (¹⁵N/¹⁴N_{sample} / ¹⁵N/¹⁴N_{standard} - 1)

(1)

For each batch of 60 samples, the overall accuracy of the method wais estimated as the reduced standard deviation of the residuals from the linear regression between the measured reference materials (n = 16) and their expected values. For the snow (n = 118) and aerosol samples (n = 35), the average uncertainty values obtained for $\delta^{15}N$ was 0.5 ‰ for both datasets.

2.5 Light attenuation through the snow pack (e-folding depth)

Measurements of light attenuation through the snow pack were made at the two snow pit sites during the ISOL-ICE campaign following a similar approach of previous studies (France and King, 2012;France et al., 2011). Vertical profiles of down-welling
irradiance in the top 0.4 m of the snow pack were measured using a high-resolution spectrometer (HR4000; Ocean Optics) covering a spectral range of 280 to 710 nm. To do this, a fiber optic probe attached to the spectrometer and equipped with a cosine corrector with spectralon diffusing material (CC-3-UV-S; Ocean Optics) was inserted into the snow to make measurements at approximately 0.03 m depth intervals. The fiber optic probe was either inserted horizontally into pre-cored

- holes, at least 0.5 m in length to prevent stray light, into the side wall of a previously dug snow pit or pushed gradually into the undisturbed snow pack starting at the surface at a 45° angle, which was maintained by a metal frame. Most measurements with integration time ranging between 30 and 200 ms were carried out at noon to minimise changing sky conditions, and each vertical snow profile was completed within 0.5 hr. The spectrometer was calibrated against a known reference spectrum from a Mercury Argon calibration source (HG-1; Ocean Optics), dark spectra were recorded in the field by capping the fibre optic probe and spectral irradiance was then recorded at depth relative to that measured right above the snow surface.
- 420 The e-folding depth was then calculated according to the Beer-Bouguer Lambert law. Stratigraphy of the snow pack recorded at each site showed presence of several thin (10 mm) wind crust layers over the top 0.4 m of snow pack. However, calculating e-folding depths for each layer in between wind crusts yielded inconclusive results. Therefore, reported e-folding depths (Fig. S1, Table S2) are based on complete profiles integrating potential effects from wind crust layers. Resulting e-folding depths relevant for the photolysis of NO₃⁻ (Table S2) show significant standard deviations, and also considerable variability (0.9-4.0

13

Formatted: Not Superscript/ Subscript

 425 cm) between profiles, which reflect both systematic experimental errors as well as spatial variability of snow optical properties. They are lower than at Dome C but similar to previous model estimates for South Pole (France et al., 2011;Wolff et al., 2002). The origin of the reduced e-folding depth relative to Dome C is not known but is likely due to greater HUmic-LIke Substances (HULIS) content or different snow morphology (Libois et al., 2013;Zatko et al., 2013). We use e-folding depths observed in this study at DML and those reported previously at Dome C as guidance for our model sensitivity study to quantify the impact 430 of the variability of e-folding depth on archived δ¹⁵N-NO₃⁻ in snow.

2.6 Nitrate photolysis rate coefficient

Hemispheric or 2π spectral actinic flux from 270 to 700 nm was measured at 2.1 m above the snow surface using an actinic flux spectroradiometer (Meteorologieconsult GmbH; Hofzumahaus et al. (2004). 2π NO₃⁻ photolysis rate coefficients J(NO₃⁻) were then computed using the NO₃⁻ absorption cross section and quantum yield on ice estimated for -30 °C from Chu and Anastasio (2003). The mean 2π J-NO₃⁻ value at DML during January 2017 was 1.02 x 10⁻⁸ s⁻¹, and 0.98 x 10⁻⁸ s⁻¹ during the 1

- to 14 January 2017 period. The observed 2π J(NO₃⁻) at DML was a factor of three lower than Dome C (2.97 x 10⁻⁸ s⁻¹; 1 to 14 January 2012) which was previously measured using the same instrument make and model, and at the same latitude (Kukui et al., 2013). Only ~5 % of the apparent inter-site difference can be attributed to TCO being ~25 DU larger at DML (306 DU) than at Dome C (287 DU) during the comparison period. The remainder was possibly due to greater cloudiness at DML and differences in calibration. In this study, the observed 2π J(NO₃⁻) is used to estimate the snow emission flux of NO₂.

2.7 Air-snow transfer modelling

Table S3.

435

In order to evaluate the driving parameters of isotope air-snow transfer at DML we used the TRANSITS model (Erbland et al., 2015) to simulate snow depth profiles of NO₃⁻ mass_concentration and δ¹⁵N-NO₃⁻ and compare them to our observations. TRANSITS is a conceptual multi-layer 1D model which aims to represent NO₂⁻-recycling at the air snow interface including processes relevant for NO₂⁻-snow photochemistry (UV photolysis of NO₃⁻, mission of NO₃⁻ local oxidation, deposition of HNO₂) and explicitly calculates NO₂⁻ mass concentrations and δ⁴⁺N-NO₃⁻ in snow. Due to the reproducible depth profile of observed δ¹⁵N-NO₃⁻ within 1 km (section 3.3), we assume the δ¹⁵N-NO₃⁻ composition is spatially uniform at DML and thus a 1D model is appropriate for our-the site. The atmospheric boundary layer in the model is represented by a single box above the snow pack. The 1 m snow pack is divided into 1000 layers of 1 mm thickness. Below the photic zone of the snow pack, the NO₃⁻ mass concentrations and δ¹⁵N-NO₃⁻ values are assumed to be constant and thus archived during the model run. The model is run for 25 years (with a timestep of one week), which is sufficient to reach steady state. The input data is provided in

Photolysis rate coefficients of NO₃⁻ (J(NO₃⁻)) above and within the snowpack are used by the TRANSITS model runs as input for this study, and are modelled off-line using the tropospheric ultraviolet and visible (TUV)-snow radiative transfer model (Lee-Taylor and Madronich, 2002). The following assumptions were made: i) a clear aerosol-free sky, ii) extra-terrestrial

irradiance from Chance and Kurucz (2010), and iii) a constant Earth-Sun distance as that on 27 December 2010 (Erbland et al., 2015). The TUV-snow radiative transfer model was constrained by optical properties of the Dome C snow pack (France et al., 2011), notably an e-folding depth of i) 10 cm in the top 0.3 m, and ii) 20 cm below 0.3 m (Erbland et al., 2015), to compute
 J(^{14/15}NO₃) profiles as a function of solar zenith angle (SZA) and TCO (Erbland et al., 2015) (Fig. S2; dashed lines).

The set up used in this paper is similar to Erbland et al. (2015) except for the following modifications. We use the TCO from the NIWA Bodeker combined dataset version 3.3, at the location of the snow pit site, averaged from 2000 to 2016 (http://www.bodekerscientific.com/data/total-column-ozone). The year-round atmospheric NO3 mass_concentrations areis taken from Weller and Wagenbach (2007), and the meteorology data is taken from Utrecht University A-automatic weather 465 Weather Station (AWS) DML05/Kohnen (AWS9: at https://www.projects.science.uu.nl/iceclimate/aws/files_oper/oper_20632). The snow accumulation rate is set to 6 cm yr¹ (w.e.); Sommer et al. (2000), and we refer to this simulation as our "base case" scenario. We carried out a sensitivity analysis to evaluate the impact of variable accumulation rate, timing of snowfall, and e-folding depth on the snow profile of NO3⁻ amass 470 concentrations and δ^{15} N-NO₃. Our first set of sensitivity tests account for the new e-folding depth measurements at the DML site; the e-folding depth was varied within the range of observations from this study and previously at Dome C. The second set of sensitivity tests use an e-folding depth of 5 cm and The sensitivity tests were as followesd: the snow accumulation rate was varied between the bounds seen in the last 1000-years at DML; the snow accumulation rate was varied from year to year according to observations from our snow pit profile which ranged between 6.0 and 7.1 cm yr⁻¹ (w.e.);- and the timing of the 475 snow accumulation was varied throughout the year, and the e-folding depth was varied within the range of observations from

this study and previously at Dome C. We compare the second set of sensitivity tests to the 5 cm e-folding depth scenario and refer to this as our "5 cm EFD (e-folding depth) case" (section 4.5). To evaluate the sensitivity of archived δ^{15} N-NO₃" to e-folding depth, changes in the J(^{14/15}NO₃") profiles for Dome C (Erbland et al., 2015) were recalculated and used as TRANSITS input by scaling the surface value of J(^{14/15}NO₃") with a new e-folding depth (2, 5, 10, 20 cm). An example is shown in Fig.

480 S2a for SZA = 70°, TCO = 300 DU and an e-folding depth of 5 cm. The top 2 mm are retained from the Dome C base-case in Erbland et al. (2015) to account for non-linearities in snow radiative transfer in snow, which are strongest in the non-diffuse zone right below the snow surface (Fig. S2b). It is noted that TUV-snow model estimates of down-welling or $2\pi J(NO_3^-)$ above the snow surface at the latitude of Dome C or DML (75° S) compare well to observations at Dome C in January 2012, whereas they are a factor three higher than measurements at DML in January 2017 (Table S4 and section 2.6). This should not affect

485 the results of the sensitivity study, which aims to explore relative changes of archived δ¹⁵<u>N-NO3</u>⁻ due to a prescribed change in e-folding depth <u>TRANSITS calculates the average number of recyclings undergone by the archived NO3⁻, i.e., below the zone of active photochemistry. In TRANSITS, the average number of recyclings undergone by NO3⁻ in a given box (snow layer or atmosphere) is represented by a tracer (or counter) called CYCL. The CYCL value for primary NO3⁻ is set to 0, and CYCL</u>

490 variables in the boxes are incremented by 1 each time NO₂ molecules cross the air-snow interface. The average number of

15

Formatted: Font color: Auto

Formatted: Not Superscript/ Subscript

Formatted: English (United Kingdom)

Formatted: Subscript

recyclings is calculated as a mass weighted average of the CYCL values of the 52 snow layers (representing one week of snowfall) which are archived below 1 m over the course of 1 year, in order to average out any seasonal variability. Erbland et al. (2015) notes that the number of recyclings represents an average value for the archived NO₃, i.e., considering individual ions in the archived NO₃, the number of recyclings could be variable as some ions may have travelled through the entire snowpack zone of active photochemistry without being recycled, while some underwent many recyclings

Formatted: English (United Kingdom)

3 Results

495

3.1 Snow pit dating

Dating of the snow pits was based on the measured concentrations of Na⁺, MSA, and nss-SO4²⁻ following previous aerosol and ice core studies at DML (Göktas et al., 2002;Weller et al., 2018). Here, Na⁺ mass concentrations have a sharp, well-defined peak in the austral spring/late winter, while MSA and nss-SO4²⁻, primarily derived from the biogenic production of dimethylsulfide (DMS), record maximum concentrations in the austral autumn. Non-sea salt SO4²⁻ (nss-SO4²⁻) often displays a second peak corresponding to late austral spring/summer sometimes linked to MSA. Spring seasons were defined as 1 September and positioned at the Na⁺ peak, while autumn seasons were defined as 1 April and positioned where a MSA and nss-SO4²⁻ peak aligned (Fig. S3). Annual layer counting of Na⁺ layers shows that snow pit A spans 8 years from autumn 2009

505 to summer 2017 and snow pit B spans 9 years from summer 2008 to summer 2017 with an age uncertainty of ± 1 year at the base of the snow pit. The mean snow accumulation rate for the snow pits is estimated to be 6.3 ± 1.4 cm yr⁻¹ (w.e.), consistent with published accumulation rates of 6.0 - 7.1 cm yr⁻¹ (w.e.) from snow pits and ice cores from DML (Sommer et al., 2000;Hofstede et al., 2004;Oerter et al., 2000).

3.2 Nitrate mass concentrations

510 Atmospheric NO₃⁻ mass concentrations (C_{aerosol}) were estimated from high-volume aerosol filters by the ratio of total NO₃⁻ mass loading to the total volume of air pumped through the filter at STP conditions following Eq. (2), and assuming a uniform loading of the aerosol filter.

Caerosol = NO3⁻ mass loading / air volume (STP)

(2)

- Aerosol mass concentrations range from 0.5 to 19 ng m⁻³ and show a downward trend throughout January 2017 (R²=0.55; p= <0.001; Fig. 3). In contrast, NO₃⁻ mass concentrations in the skin layer increase during the month from 136 to 290 ng g⁻¹. Nitrate mass concentrations in both snow pits, which range from 23 to 142 ng g⁻¹, are substantially lower than those in the skin layer. Compared to Dome C, average annual atmospheric, skin layer and snow pit mass concentrations are lower at DML (Table 2), in agreement with higher NO₃⁻ mass concentrations found at lower snow accumulation sites (Erbland et al., 2013). The NO₃⁻ mass concentration profile in the upper 50 cm of the snow pack at Dome C shows an exponential decrease with
- 520 depth and becomes relatively constant at 35 ng g⁻¹ at 20 cm compared to 160_-_1400 ng g⁻¹ in the skin layer (Figs. 1 and 4; (Erbland et al., 2013;Frey et al., 2009). While the highest NO₃⁻ mass concentrations in the snow pack at DML are also found

in the skin layer, the <u>mass</u> concentration profile exhibits a different pattern. The sharp decrease in NO_3^- mass concentration occurs in the top ~5 mm at which point the snow pit records inter-annual variability in the NO_3^- mass concentration. Nitrate mass concentrations at DML exhibit a maximum in summer and <u>winter-minimum in winter</u>.

525 Although the Dome C depth profiles of NO₃⁻ mass concentration do not record seasonal variability, year-round measurements of skin layer and atmospheric NO₃⁻ mass concentrations exhibit sharp maximum during sunlit conditions in spring and summer and low mass concentrations in winter. This annual cycle is consistent both i) spatially across Antarctica (McCabe et al., 2007;Wolff et al., 2008;Erbland et al., 2013;Frey et al., 2009), and ii) temporally over last 7 years (Fig. 1) (Erbland et al., 2015;Erbland et al., 2013;Frey et al., 2009).

530 While the precision of the IC measurement of NO₃⁻ is better than 2 %, the spatial variability at DML of NO₃⁻ in the skin layer exceeds this. During the sampling campaign, five skin layer samples were taken from an area of ~2500 m² at the flux site (snow surface had sastrugi up to 10 cm) to understand how representative the snow pit mass concentrations are of the greater study area. We found that the spatial variability of NO₃⁻ mass concentrations and δ¹⁵N-NO₃⁻ at DML was 10 % and 17 % respectively. (Fig. 3c-d). At Dome C, the spatial variability of NO₃⁻ mass concentrations wais between 15 and 20 %. We note that this variability includes the natural spatial variability and the operator sampling technique.

3.3 Isotopic composition of nitrate

Atmospheric δ¹⁵N-NO₃⁻ ranges from -49 to -20 ‰ at DML and -9 to 8 ‰ at Dome C during the January campaign, and is depleted with respect to the skin layer, which ranges from -22 to 3 ‰ at DML (Fig. 3). Similar to the NO₃⁻ mass concentrations, the δ¹⁵N-NO₃⁻ in the depth profile at DML exhibits large variability between seasons (-3 to 99 ‰) with more enriched values in spring and summer with respect to winter (Fig. 4). The δ¹⁵N-NO₃⁻ values in both snow pits at DML show extremely good reproducibility with depth indicating there is little spatial variability within 1 km at the site (Fig. 4). The δ¹⁵N-NO₃⁻ in snow pits at Dome C does not preserve a seasonal cycle. However, in parallel with the exponential decay of NO₃⁻ mass concentrations with depth at Dome C, there is a strong increase in the δ¹⁵N-NO₃⁻ with depth. At Dome C, δ¹⁵N-NO₃⁻ increases up to 250 ‰ in the top 50 cm, this increase is weaker at DML (up to 8100 ‰ in the top 3040 cm at which point seasonal cycles are evident).
545 At Dome C, although no annual cycle is preserved in the snow pack, the year-round measurements of atmospheric δ¹⁵N-NO₃⁻ in the skin layer has a spring minimum that increases to a maximum at the end of summer (Fig. 1). Skin layer δ¹⁵N-NO₃⁻ is about 25 ‰ higher than atmospheric δ¹⁵N-NO₃⁻. Nitrate mass concentration and δ¹⁵N-NO₃⁻ composition data for aerosol, skin layer and

550 3.4 Archived nitrate mass concentration and isotopic composition

snow pit samples are available in Winton et al. (2019b).

We calculate archived values of NO₃⁻ mass concentration and δ^{15} N-NO₃⁻ which represent the archived mass fraction and isotopic composition reached below the photic zone. Archived values were calculated by averaging the NO₃⁻ mass concentration and δ^{15} N-NO₃⁻ values below the photic zone, i.e., <u>30–15</u> cm (section 4.44.4). The archived NO₃⁻ mass

Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript
concentration and δ¹⁵N-NO₃⁻ values for snow pit A were 60 ng g⁻¹ and 50 ‰,-the and the archived NO₃⁻ mass concentration for snow pit B was 50 ng g⁻¹. Note that <u>no δ¹⁵N-NO₃⁻ values</u> were measured below 30 cm in snow pit B. <u>ObservedThese measured δ¹⁵N-NO₃⁻ values are half of those expected for a site with a snow accumulation rate of 6 cm yr⁻¹ (w.e.) in the spatial survey from Erbland et al. (2013) (Table 2).</u>

3.5 Nitrate mass flux estimates

The total deposition flux (F) of NO_3^- is partitioned into wet and dry deposition fluxes (F_{wet} and F_{dry} respectively; Eq. (3)), and 560 can be estimated using the measured mass concentration of NO_3 in the snow pack (C_{snow}) and the local snow accumulation rate (A; Eq. (4)). Estimates of the dry deposition rate (Fdry) of NO3⁻ were calculated using Eq. (5) using the atmospheric mass concentrations of NO3⁻ (Caerosol) and a dry deposition velocity (Vdry deposition) of 0.8 cm s⁻¹, and are reported in Table S5. This deposition velocity is based on the dry deposition of HNO3 at South Pole (Huey et al., 2004) which has a similar snow accumulation rate (6.4 cm yr⁻¹ (w.e.); Mosley-Thompson et al. (1999)) to DML. Other estimates of dry deposition velocities include 0.05_-_0.5 cm s⁻¹ for HNO₃ over snow (Hauglustaine et al., 1994;Seinfeld and Pandis, 1998), 1.0 cm s⁻¹ for NO₃- over the open ocean (Duce et al., 1991), and an apparent deposition velocity of 0.15 cm s⁻¹ for summer HNO₃ at Dome C (Erbland et al., 2013). The estimated apparent NO3⁻ deposition velocity at Dome C is low because of the strong recycling of NO3⁻ on the polar plateau in summer, i.e., reactive nitrogen is re-emitted from the skin layer to the atmosphere. Although gas phase HNOg and particulate NO3 have different dry deposition rates Thus, the dry deposition velocity at DML is likely to lie between _ 570 0.15 and 0.8 cm s⁻¹. We assume that a constant deposition velocity throughout the campaign is appropriate for DML. $F = F_{wet} + F_{dry}$ (3)

$C_{snow} = F / A$	(4)
$F_{dry} = \mathbf{C}_{acrosol} \mathbf{V}_{drydeposition}$	(5)

575 Note that Eq. (4) does not take into account post-depositional processes of non-conservative ions, such as NO₃⁻. We follow the approach of Erbland et al. (2013) who use an archived NO₃⁻ mass flux (F_a) to represent the downward NO₃⁻ mass flux which escapes the photic zone towards deeper snow layers. Using simple mass balance, we can then estimate the mass flux of NO₃⁻ (F_{re-emit}), which is re-emitted from the snow pack to the overlaying atmosphere (Eq. (6)). $F_{re-emit} = F - F_a$ (6)

580 Taking this a simple mass balance approach, a schematic of NO₃⁻ mass fluxes for two scenarios are illustrated in Fig. 5a. Scenario 1 is an average annual budget for DML (Fig. 5a). As the atmospheric campaign did not cover an entire annual cycle, we use estimates of atmospheric NO₃⁻ mass fluxes at DML reported by Pasteris et al. (2014) and Weller and Wagenbach (2007) of 43 and 45 pg m⁻² s⁻¹, respectively, as a-year round dry deposition fluxes. Due to the linear relationship of ice core NO₃⁻ mass concentrations with the inverse accumulation, the authors assume that the magnitude of the dry deposition flux is homogenous over the DML region. Mean annual mass concentrations of NO₃⁻ in our snow pits suggest a total NO₃⁻ deposition mass flux of

585 over the DML region. Mean annual mass concentrations of NO_3^- in our snow pits suggest a total NO_3^- deposition mass flux of 110 pg m⁻² s⁻¹ and therefore a wet deposition mass flux of 65 pg m⁻² s⁻¹.

18

Formatted: Subscript

However, at relatively low snow accumulation sites where photolysis drives the fractionation of NO₃⁻ from the surface snow to atmosphere (Frey et al., 2009), it is necessary to take into account the skin layer in the NO₃⁻ mass_flux budget as this airsnow interface is where air-snow transfer of NO₃⁻ takes place. In scenario 2, wWe usutilisee the available NO₃⁻ mass concentrations measured in aerosol, skin layer, and snow pits from the ISOL-ICE campaign to estimate the mass flux budget for January 2017 (Fig. 5b). The dry deposition mass flux of atmospheric NO₃⁻ during January 2017 at DML averages 64 ± 38 pg m⁻² s⁻¹ (Table S5). The NO₃⁻ mass flux to the skin layer is 360 pg m⁻² s⁻¹, however only 110 pg m⁻² s⁻¹ of NO₃⁻ is archived. Considering the active skin layer, only 30 % of deposited NO₃⁻ is archived in the snow pack while 250 pg m⁻² s⁻¹ is re-emitted to the overlaying atmosphere.

595

590

3.6-6 Fractionation constants

Fractionation constants were calculated following the approach of Erbland et al. (2013), which assumes a Rayleigh single loss and irreversible process of NO₃⁻ removal from the snow. As this approach may oversimplify the processes occurring at the airsnow interface, Erbland et al. (2013) referred to the quantity as an "apparent" fractionation constant. Thus, the apparent fractionation constant represents the integrated isotopic effect of the processes involving NO₃⁻ in the surface of the snow pack and in the lower atmosphere. The apparent fractionation constant is denoted as ¹⁵ ε_{app} and calculated using Eq. (7). $\ln(\delta^{15}Nf + 1) = {}^{15}\varepsilon x \ln f + \ln(\delta^{15}N_0 + 1)$ (7)

where $\delta^{15}Nf$ and $\delta^{15}N_0$ are the δ -values in the initial and remaining NO₃⁺, and *f* is the remaining NO₃⁺ mass concentration. The ϵ values are related to the commonly used fractionation factor α by $\epsilon = \alpha - 1$. The ${}^{15}\epsilon_{app}$ derived for snow pits in the photic state.

3.7 Light attenuation through the snow pack (e-folding depth)

e-folding depths relevant for the photolysis of NO₃⁻ are reported in Table S2 and show significant standard deviations, and also considerable variability (0.9 - 4.0 cm) between profiles, which reflect both systematic experimental errors as well as spatial variability of snow optical properties. They are shallower than at Dome C but similar to previous model estimates for South
Pole (France et al., 2011;Wolff et al., 2002). The origin of the reduced e-folding depth relative to Dome C is not known but is likely due to greater HUmic-LIke Substances (HULIS) content or different snow morphology (section 4.3.2) (Libois et al., 2013;Zatko et al., 2013;Brucker et al., 2010).

3.8 Snow emission of NO₂

The potential snow emission flux of NO₂ (F_{NO2}) from NO₃⁻ photolysis in snow was estimated using Eq. (8). 615 $F_{NO2} = \int_{z=0}^{z=1} m [NO_3^-]_z J(NO_3^-) dz$

19

Formatted: Heading 2

Eq. (8)

where $J_2(NO_3^-)$ is the photolysis rate coefficient of reaction $NO_3^- +h\nu \rightarrow NO_2^- + O^-$ at depth, z, in the snowpack, and is derived by scaling surface measurements (section 2.6) with e-folding depth (2 - 10 cm), and $[NO_3^-]_z$ is the amount of NO_3^- per unit volume of snow at depth, z, in the snowpack. The calculated F_{NO2} is a potential emission flux assuming that all NO_3^- within the snow grain is photo-available, no cage effects are present and NO_2 is vented immediately after release from the snow grain

- 620 to the air above the snowpack without undergoing any secondary reactions. For the 1 to 14 January 2017 period, model estimates of F_{NO2} scaled approximately linearly with e-folding depth were 0.4, 1.0 and 1.9 x 10¹¹ molecule m² s⁻¹ for e-folding depths of 2, 5 and 10 cm, respectively. Spatial variability of NO₃⁻ in the top 30 cm of surface snow at DML based on snow pits A and B is on the order of 13 % inducing similar variability in the model estimates of F_{NO2} . Estimates of F_{NO2} at Dome C, based on the same model during 1 to 14 January 2012, were larger with 1.2 - 7.3 x 10¹¹ molecule m⁻² s⁻¹ (Frey et al., 2013),
- 625 mostly due to larger J(NO₃⁻) values observed above the surface (section 2.6) as well as a larger e-folding depth (10 cm near the surface). It should be borne in mind that the above simple model estimates (Eq. (8)) may significantly underestimate the real emission flux. Previous comparisons of F_{NO2} computed with Eq. (8) and F_{NOx} measured at Dome C showed that observations can exceed model predictions by up to a factor 50 (Frey et al., 2015;Frey et al., 2013). While NO₃⁻ mass concentrations in snow, the surface actinic flux and the e-folding depth were measured at the DML field site, the quantum
- 630 yield of NO₃⁻ photolysis in surface snow (ΦNO₃⁻) was not, but introduces significant uncertainty in the model estimates. Previous lab measurements on natural snow samples collected at Dome C showed ΦNO₃⁻ to vary between 0.003 and 0.05 (Meusinger et al., 2014). As described above (section 2.6) JNO₃⁻ used in Eq. (8) was calculated with ΦNO₃⁻ at -30 °C (= 2 x 10⁻³) after Chu and Anastasio (2003), which is near the lower end of the observed range. Thus, up to half of the mismatch between Eq. (8) and Dome C observations can be explained by adjusting ΦNO₃⁻. Another factor contributing to larger fluxes
- 635 and not included in Eq. (8) is forced ventilation. In the more sophisticated TRANSITS model, Erbland et al. (2015) found that the photolytic quantum yield was one of the major controls on archived flux and primary input flux at Dome C. Erbland et al. (2015) initially used a quantum yield of 2.1 x 10⁻³ at 246 K (France et al., 2011) but it underestimated NO₃⁻ recycling and overestimated primary NO₃⁻ trapped in snow. Adjusting the quantum yield to 0.026, within the range observed in the lab –(Meusinger et al., 2014), gave more realistic
- 640 archived δ^{15} N-NO₃⁻ values. However, at Dome C TRANSITS simulated F_{NO2} fluxes were about a factor of 9 18 higher than observed F_{NOx}. Erbland et al. (2015) suggested that the discrepancy could result from the simplifications made in the TRANSITS model regarding the fate of NO₃⁻ photolysis products.

3.97 Simulated nitrate mass concentrations and isotopic ratios from TRANSITS modelling

645 Simulated TRANSITS results for the <u>base case and 5 cm EFD case scenarios at the</u> air_snow interface are illustrated in Fig. 6 <u>along with TCO data (Fig. 6a)</u>. In the atmosphere, the TRANSITS model is forced with the smoothed profile of year-round atmospheric NO₃⁻ measurements from the DML site (Weller and Wagenbach, 2007) <u>wwhich hashere</u> the highest mass concentrations <u>are in</u> spring and summer with a maximum of 80 ng m⁻³ in November and a <u>winter-minimum of 2 ng m⁻³ in</u>

- winter (Fig. 6b). Overall, the simulated values in the base case scenario are higher than the 5 cm EFD case in summer and autumn, and converge to similar values in winter. TAlthough we only have measurements of $\delta^{15}N$ -NO₃-in January, the simulated atmospheric $\delta^{15}N$ -NO₃ values in the base case for January, are greater than the measurements available from this study, while the $\delta^{15}N$ -NO₃ values in the 5 cm EFD case fall within the range of observations. TThe annual cycle of simulated atmospheric $\delta^{15}N$ -NO₃ for the 5 cm EFD case shows a 40-50 ‰ dip in spring to -43-2 ‰ from winter values which coincides with the simulated atmospheric NO₃ mass concentration increase in spring (Fig. 6c). The highest <u>simulated</u> atmospheric $\delta^{15}N$ -
- NO₃⁻ values (7 ‰) occur in winter, for both scenarios. In the skin layer, the simulated NO₃⁻ mass concentrations are an order of magnitude greater than our observations in January and we outline possible reasons for this discrepancy in the discussion (section 4.1). The simulated annual cycle of NO₃⁻ mass concentrations in the skin layer steadily rise in spring and reach a peak in January when they begin to decline to the lowest mass concentration in winter (Fig. 6d). Simulated skin layer δ¹⁵N-NO₃⁻ values in January for the base case are ~10 ‰ higher than our highest observations for that month but the average January
- 660 value in the 5 cm EFD case (c7 ‰) falls in the range of observed values (-10 ‰) (Fig. 6e). For the 5 cm EFD case, tFhey begin to decrease by 24,30 ‰- in spring at the same time as atmospheric δ¹⁵N-NO₃⁻ values decrease. In October and November, the skin layer δ¹⁵N-NO₃⁻ values begin to rise up to c1144 ‰ in February in the 5 cm EFD case in February. The seasonality of simulated NO₃⁻ mass concentrations and δ¹⁵N-NO₃⁻ values in the atmosphere and skin layer at DML isis
- consistent with Dome C (Fig. 1). Similar to Dome C, <u>simulated</u> NO₃⁻ mass concentrations in the skin layer start to rise two 665 months earlier than atmospheric NO₃⁻ mass concentrations and the summer maximum is later. While the seasonality of δ^{15} N-
- NO₃⁻ in the skin layer and atmosphere co-vary, simulated skin layer δ¹⁵N-NO₃⁻ values are enriched relative to atmospheric values on average by <u>R(1⁻ss</sub></u>.
 The simulated NO₃⁻ mass concentrations and δ¹⁵N-NO₃⁻ values in the snow pitdepth profiles are illustrated in Fig. 7. <u>The e-folding depth sensitivity tests show that a deeper e-folding depth i) increases the δ¹⁵N-NO₃⁻ enrichment in the photic zone, and
 </u>
- 670 ii) increases in the mean annual archived δ¹⁵N-NO₃⁻ value (Fig. 7a). Out of the e-folding depths explored in the sensitivity analysis, an e-folding depth in the range of that observed at DML, i.e., 2 5 cm, has the closest mean annual δ¹⁵N-NO₃⁻ value to the observations (Fig. 7a). -BBoth the simulated-depth profile of simulated NO₃⁻ mass concentration and δ¹⁵N-NO₃⁻ in the base case for an accumulation rate of 6 cm yr⁻¹ (w.e.) show seasonal variability in the first year with a range of of 380 ng g⁻¹ and 20 ‰, which decreases with depth to a range of 95 ng g⁻¹ and 10 ‰ in the fourth year. In comparison, in the 5 cm EFD
- 675 case, the seasonality of δ¹⁵N-NO₃⁻ and NO₃ mass concentrations in the first year ranges from 290 ng g⁻¹ and 40 ‰ to 75 ng g⁻¹ and 20 ‰ in the fourth year (Fig. 7a). For the base case scenario, t the simulated archived (i.e., annual average of the first / year below 1 m) NO₃⁻ mass concentration, δ¹⁵N-NO₃⁻, and NO₃⁻ mass flux values are 120 ng g⁻¹, 130 ‰, and 210 pg m⁻² yr⁻¹, respectively. The simulated annual average ¹⁵ε_{app} is -19 ‰ for the top 30 cm (i.e., active photic zone with an e-folding depth of 10 cm). In comparison, in the 5 cm EFD case, the simulated archived NO₃⁻ mass concentration, δ¹⁵N-NO₃⁻, and NO₃⁻ mass /
- 680 flux values are 280 ng g⁻¹, 50 ‰, and 480 pg m⁻² yr⁻¹, respectively. The simulated annual average ¹⁵ ϵ_{app} is -11 ‰ for the top 30 / cm. The 5 cm EFD case falls within the range of observations for δ^{15} N-NO₃⁻ (Figs. 7a) but is significantly higher than the / observed NO₃⁻ mass concentrations (Fig. 7c). Also plotted in Figs. 7b-c are the simulated NO₃⁻ mass concentration and δ^{15} N-

21

Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight

Formatted: Not Highlight

Formatted: Not Highlight
Formatted: Not Highlight
Formatted: Not Highlight
Formatted: Not Highlight

Formatted: Pattern: Clear (Yellow)

Formatted: Not Highlight Formatted: Pattern: Clear Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight Formatted: Pattern: Clear Formatted: Not Highlight Formatted: Pattern: Clear Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Superscript/ Subscript NO_3^- depth profiles for accumulation rates of 2.5 cm yr⁻¹ (w.e.) and <u>100-11</u> cm yr⁻¹ (w.e.) for the 5 cm EFD case. As the accumulation rate increases, the annual layers of $\delta^{15}N$ -NO₃⁻ become thicker, the seasonal amplitude increases, the mean annual

- δ^{15} N-NO₃⁻ value decreases, and there is less δ^{15} N-NO₃⁻ enrichment in the photic values in the top 10 one decreases zone (Fig. 7b). At very low snow accumulation rates, the seasonal cycle is smoothed, as in the case of Dome C (Fig. 7b). A similar pattern is observed for the simulated NO₃⁻ mass concentrations with depth: seasonal cycles of NO₃⁻ mass concentrations are more pronounced at higher snow accumulation rates, while inter-annual variability is smoothed at very low accumulation rates such as Dome C (Fig. 7c). The relationship between the snow accumulation rate and δ^{15} N-NO₃⁻ is non-linear (Figs b-c).
- 690 Overall the TRANSITS modelling shows that the i) simulated values in the base case scenario are higher than the 5 cm EFD case, and ii) TRANSITS modelling simulations using the observed e-folding depth of 5 cm are good fit with observations. Differences between the simulated $\delta^{15}N-NO_3^-$ depth profiles for the two cases and observed $\delta^{15}N-NO_3^-$ could be due to uncertainties in a number of factors, for example: i) a shallower e-folding depth than modelled (section 4.5.1), ii) lower JNO₃⁻ values (NO₃⁻ photolysis rate), which are related to a lower e-folding depth, and would lead to less enrichment of $\delta^{15}N-NO_3^-$ in
- 695 the snow pack (section 4.3.2), iii) higher atmospheric NO₃⁻ input, however δ¹⁵N-NO₃⁻ values are not sensitive to variable atmospheric NO₃⁻ mass concentrations (Erbland et al., 2015), and/or iv) variable snow accumulation which would shift the oscillations to the correct depth and lower the mean δ¹⁵N-NO₃⁻ values below the photic zone (section 4.5.2). These differences are further addressed in section 4.5.
- The simulated archived (i.e., annual average of the first year below 1 m) NO₂=mass concentration, 8¹⁵N-NO₂; and NO₂=mass 700 flux values are 120 ng g⁺, 130 ‰, and 210 pg m⁻² yr⁻¹, respectively. The simulated annual average ¹⁵e_{app} is -19 ‰ for the top 30 cm (i.e., active photic zone with an e-folding depth of 10 cm).

4 Discussion

4.1 Validation of results

- Our January 2017-NO₃⁻ measurements at DML agree well with-values reported in the literature, and largely with largely with
 the simulated 5 cm EFD case results from the TRANSITS model with the exception of except for the skin layer NO₃⁻ mass concentrations. While we made the first measurements of atmospheric, skin layer and snow pit 8⁴⁵N-NO₃⁻, and skin layer NO₃⁻ mass concentrations at DML, there are published measurements of NO₃⁻ mass concentrations in snow pits and In particular, oeur NO₃⁻ concentrations observations in snow pits agree well with published measurements of ANO₃⁻ mass concentrations of NO₃⁻ mass concentrations in snow pits and In particular, oeur NO₃⁻ concentrations observations in snow pits agree well with published measurements of NO₃⁻ mass concentrations of atmospheric NO₃⁻ mass concentrations in snow pits at DMLthose (Weller et al., 2004). While our January 2017 observations of atmospheric NO₃⁻ mass concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass NO₃⁻ concentrations in January 2017 are lower than those observed in <u>-Our atmospherie mass</u></u></u></u></u></u></u></u></u></u></u>

22

Farmer Marked, Mark I Balak Balak	
Formatted: Not Highlight	
Formatted: Not Highlight	
Formatted: Highlight	
Formatted: Pattern: Clear	
Formatted: Pattern: Clear	

Formatted: Pattern: Clear

Formatted: Superscript

Formatted: Not Superscript/ Subscript Formatted: Not Superscript/ Subscript

	Overall, For the skin layer, the simulated NO3 mass concentrations results from TRANSITS are greater than our January
715	observations, particularly the skin layer NO2-mass concentrations (Fig. 6d). The discrepancy between the significantly higher
1	simulated NO3 ⁻ mass concentrations than observations in the skin layer was also found at Dome C. Erbland et al. (2015)
	suggested that this discrepancy could be related to either a sampling artefact, snow erosion or a modelled time response to
	changes in past primary inputs. We provide an alternative explanation for the extremely high simulated NO_3^- mass
	concentrations in the skin layer using daily measurements of NO3 ⁻ mass concentration in diamond dust and hoar frost collected
720	daily from Polyvinyl chloride (PVC) sheets at Dome C in summer 2007/08, i.e. new deposition. New deposition of diamond
I	dust had NO_3 mass concentrations up to 2000 ng g ⁻¹ , which is four times greater than that observed in natural snow from the
	skin layer at the same time (Fig. S4). Similarly, new deposition of hoar frost had NO3 ⁻ mass concentrations up to 900 ng g ⁻¹ ,
I	which is three times greater than the skin layer snow. The formation of surface hoar frost occurs by co-condensation, i.e. the
	simultaneous condensation of water vapour and NO3 ⁻ at the air-snow interface. Recent modelling suggests that co-condensation
725	is the most important process explaining NO3 ⁻ incorporation in snow undergoing temperature gradient metamorphism at Dome
	C (Bock et al., 2016). Diamond dust can also scavenge high concentrations of HNO3 at Dome C (Chan et al., 2018).
	Furthermore, the top layer of the snow pack is only 1 mm thick in the TRANSITS model, whereas our observations of the skin
	layer are 5 mm thick. Due to the photochemical loss of NO3- mass concentrations with depth, the highest NO3- mass
	concentrations are expected in the top 1 mm layer which is the layer best in equilibrium with the atmosphere. Here, extremely
730	high mass concentrations of NO3 ⁻ from new deposition from diamond dust and hoar frost are also found. In summary, In
	summary, it is likely that we do not measure such high NO3 mass concentrations in hoar frost and diamond dust values in the
	skin layer because of sampling artefacts or blowing snow, which can dilute or remove the diamond dust and hoar frost-which
	is where we would expect the highest concentrations due to the exponential decay of NO3 ⁻ with depth (Fig. S4). If indeed the
	higher simulated values in the skin layer can be explained by hoar frost and diamond dust, then we can have greater confidence
735	in the depth profile of NO3 ⁻ concentration. It is interesting to note that these higher simulated values in the skin layer do not
	impact the simulated depth profiles (Fig. 57). In summary, it is likely that we do not measure such high hoar frost and diamond
	dust values in the skin layer because of sampling artefacts or blowing snow, which can dilute or remove the diamond dust and

While not yet observed elsewhere on the Antarctic continent, over the short intensive sampling period at DML we 740 observe significant variability in NO₂⁻mass concentrations and δ¹⁵N NO₂⁻values that resembles a diurnal cycle. Over 4 hours, the skin layer NO₂⁻mass concentrations varied by 46 ng g⁺, the skin layer δ¹⁵N NO₂⁻by 21 ‰, and the atmospheric δ¹⁴N NO₂⁻ by 18 ‰. Other coastal studies have attributed daily variability to individual storm events

hoar frost.

Formatted: Heading 2

(Mulvaney et al., 1998;Weller et al., 1999). We note that the sampling duration is too short to confirm any diurnal patterns but it would be interesting to investigate this further in future work.

745 4.2 Nitrate deposition

Here we discuss the various processes in which NO₃⁻ can be deposited to the skin layer at DML. As we have just one month of atmospheric and skin layer data, our ability to look at the deposition on seasonal seales is limited, however we provide new insights into the austral summer deposition processes.

- 750 While it is common to measure nitrogen species in snow and aerosol samples as the NO₃⁻ ion using ion chromatography, nitrogen species can be deposited in various forms either by wet or dry deposition to the skin layer. We note that organic NO₃⁻ plays are little role in determining snow concentrations (Jones et al., 2007; Wolff et al., 2008), and as such we focus our discussion on inorganic NO₃⁻. The various nitrogen species include, i) a neutral salt (NO₃⁻ co-deposition with sea our discussion on literative (2009). NO 3⁻ is (ICDE) and the salt (NO₃⁻ co-deposition with sea our discussion of the sea out of the salt (NO₃⁻ co-deposition with sea out of the salt (NO₃⁻ co-deposition sea out of the salt (NO₃⁻ co-deposition with sea out out of the salt (NO₃⁻ co-deposition with sea out of the salt (NO₃⁻ co-deposition sea out of the salt (NO₃⁻ co-
- salt or mineral dust; Wolff et al. (2008)), ii) NO₃⁻ in air (HNO₃ in gas phase plus particulate NO₃⁻). Following the terminology of Erbland et al. (2013), this is referred to as "atmospheric NO₃⁻ and is represented by the NO₃⁻ mass concentrations measured on our aerosol filters. Atmospheric NO₃⁻ can either be deposited as dry deposition by adsorption to the snow surface as HNO₃ has a strong affinity for ice surfaces (Abbatt, 2003;Huthwelker et al., 2006) or seavenged by precipitation as wet deposition, and iii) co-condensation of HNO₃ and water vapour onto snow crystals (Thibert and Domine, 1998).
- 760 Depending on the deposition pathway, NO₃⁻ can either be predominantly incorporated into the bulk snow crystal or be adsorbed onto the surface of the snow crystal. Deposition pathways include co-condensation (formation of surface hoar frost), riming (deposition of supercooled fog droplets), and adsorption of HNO₃-onto the snow crystal surface (dry deposition) (Röthlisberger et al., 2002). Both co-condensation (Bock et al., 2016) and dry deposition of HNO₃, at very cold temperatures, can elevate NO₃⁻-mass concentrations in the skin layer. Furthermore, trace nitrogen impurities present in the interstitial air in the norous snow nack may be incorporated in snow crystal. While can engine of NO₃⁻
- 765 present in the interstitial air in the porous snow pack may be incorporated in snow crystals. While scavenging of NO₃⁻ by snow (wet deposition) occurs sporadically throughout the year, dry deposition of particulate NO₃⁻ or surface adsorption may take place continuously throughout the year. We see both of these deposition processes taking place during January 2017.

4.2.1 Wet and dry deposition

- 770 Here we discuss the various processes in which NO₃⁻ can be deposited to the skin layer at DML. Firstly, we first look at atmospheric NO₃⁻ deposition in relation to the source region of the air mass. The mean annual wind direction at the site is 65° (Figs. 3 and S5). There is an excursion from this predominant wind direction between 19 22 January, where the wind direction switches to the southwest, i.e., atmosphere transport from the plateau. We do not see elevated NO₃⁻ mass concentrations during this period nor do we see a marked difference in isotopic signature that is similar to Dome C at this time (Fig. 4). This, in line
- 775 with air mass back trajectories (not shown) suggests that transport of NO₃⁻ re-emitted from inland sites of the Antarctic, carrying a distinctively enriched δ¹⁵N-NO₃⁻ signature, did not influence DML during our campaign. We can also rule out any downwind contamination from the station.

Secondly, we use Precipitation at DML can occur either through sporadic cyclonic intrusions of marine air masses from the adjacent ocean associated with large amounts of precipitation, or clear sky diamond dust that contributes smaller amounts to

780 the total precipitation (Schlosser et al., 2010). Overall, extreme precipitation events dominate the total precipitation (Turner et

al., 2019). In austral summer, the transport of marine aerosol to DML is mediated by two synoptic situations, i) low-pressure systems from the eastern South Atlantic associated with high marine aerosol concentrations, and ii) persistent long-range transport that provides background aerosol deposition during clear sky conditions (Weller et al., 2018). Weller et al. (2018) suggest that dry deposition of marine aerosol is dominant over wet deposition at DML. In contrast, Dome C receives predominantly diamond dust, and thus aerosol deposition is different there.

- More specifically, precipitation during our sampling campaign in January 2017 was relatively low compared to previous years.
 Modelled daily precipitation at the nearest Regional Atmospheric Climate Model (RACMO2; Van Meijgaard et al. (2008))
 grid point (75.0014°S, 0.3278°W) is illustrated in Fig. 3a. The largest precipitation event of the month was on 1 January (0.27 mm) resulting from a low-pressure system in the South Atlantic (Fig. S5). For the rest of the month, half of the days had zero
 790 precipitation and the other half had very little precipitation (~0.05 mm per day).
- We use <u>mthe odelled daily precipitation at the nearest Regional Atmospheric Climate Model (RACMO2; Van Meijgaard et al.</u> (2008) grid point (75.0014°S, 0.3278°W; Fig. 3a) RACMO2 daily precipitation data to identify whether the influence of cyclonic intrusions of marine air masses provide to wet deposition of NO₃- to the site in January. We observe that .- In the skin layer, we observe that NO₂- mass concentrations and other sea salt ions co-vary (Fig. S6) suggesting similar deposition
- 795 pathways of these ions_Some peaks in the skin layer NO₃⁻ mass concentration are accompanied by fresh snow laden with relatively high sea salt aerosol mass concentrations and atmospheric NO₃⁻ mass concentrations, for example on 1, 13, and 18 January 2017 (Fig. S6). In the skin layer, we observe that NO₃⁻ mass concentrations and other sea salt ions co-vary (Fig. S6) suggesting similar deposition pathways of these ions. Such deposition events have also been observed on the Antarctic coast (Wolff et al., 2008). During the formation of precipitation, essentially all HNO₃ is removed from the gas phase due to its high
- 800 solubility in liquid clouds (Seinfeld and Pandis, 1998). Therefore, HNO₂ can be scavenged from the atmosphere and deposited as NO₂⁻ in the skin layer. The uptake of HNO₂ onto the snow and ice crystal surface during and after precipitation can also contribute further to the NO₂⁻ mass concentrations found in the skin layer. On someWhereas on other precipitation days, we observe lower atmospheric NO₃⁻ mass concentrations and higher skin layer NO₃⁻ mass concentrations that could be a result of HNO₃ scavenging. Mulvaney et al. (1998) observed higher skin layer concentrations in days when there was little snow
- 805 accumulation and concluded that NO₂⁻ is directly up taken onto the surface by dry deposition of particulate NO₂⁻ and surface adsorption of HNO₂ (gas-phase) (e.g. Mulvaney et al., 1998). With only one month of data it is difficult to see the impact of wet deposition on the NO₃⁻ mass concentration in the skin layer; i.e. whether fresh snowfall dilutes the NO₃⁻ mass concentration in the skin layer or whether it scavenges HNO₃ (gas-phase) resulting in higher mass concentrations of NO₃⁻ in the skin layer. Most likely both processes are occurring. We note that due to post-depositional processes (section 3) any short-term signals
- 810 observed in the skin layer are unlikely to be preserved. Even at the South Pole where the snow accumulation rate is slightly higher (8.5 cm yr⁺(w.e.); (Mosley-Thompson et al., 1999) than DML deposition, NO₂-peaks are substantially modified after burial (Dibb and Whitlow, 1996).

Thirdly, we

785

investigate daily changes in the atmospheric and skin layer NO₃⁺ mass concentrations and δ¹⁵N-NO₃⁻ over the campaign to see
 the influence of dry deposition, by adsorption of atmospheric NO₃⁺ to the snow surface, 4.2.2 Dry deposition <u>NO₃⁺ is directly</u> up taken onto the surface by dry deposition of particulate NO₃⁺ and surface adsorption of HNO₂ (gas phase) (e.g. Mulvaney et al., 1998)

up taken onto the surface by dry deposition of particulate NO₃ and surface adsorption of HNO₃ (gas phase) (e.g. Mulvaney et al., 1998) In order to investigate dry deposition of NO₃, we first look at atmospheric NO₃ in relation to the wind direction and air mass back trajectories. The mean annual wind direction at the site is 65° , and January 2017 is no exception (Figs. 3 and S7). There

820 is an excursion from the predominant wind direction between 19-22 January, where the wind direction switches to the southwest. Although there are no studies indicating fractionation of δ¹⁵N-NO₃⁻ in the atmosphere during atmospheric transport from the plateau to the coast, we do not see elevated NO₃⁻ mass concentrations during this period nor do we see a marked difference in isotopic signature that is similar to Dome C at this time (Fig. 4). This, in line with air mass back trajectories (not shown) suggests that long range transport of NO₃⁻ re-emitted from inland sites of the Antarctic did not reach DML during our campaign. We can also rule out any downwind contamination from the station.

High concentrations of sea salt and mineral dust can promote the conversion of HNO₃ (gas-phase) to aerosol, as well as trapping NO₃⁻ (gas phase) on salty snow surfaces. We see a relationship between sea salt aerosol and atmospheric NO₃⁻ (R²= 0.59; p=<0.001) suggesting that even 550 km inland from the coast sea salt could promote the conversion of HNO₃⁻ to atmospheric NO₃⁻, although we acknowledge that our filters capture both aerosol NO₃⁻ and HNO₃, and sea salt concentrations are much higher at Halley and coastal Antarctica where this mechanism sporadically occurs (Wolff et al., 2008).

- Seavenging of atmospheric NO₃⁻ is largely responsible for the<u>on</u> the high mass concentrations observed in the skin layer. <u>Temporal v</u>Variation in the <u>mass</u> concentration and isotopic signature of aerosol and surface snow at DML over January 2017 suggests atmospheric NO₃⁻ is the source of NO₃⁻ to the skin layer. Throughout the month, the increase in the skin layer <u>mass</u> concentration of summer NO₃⁻ appears to be closely related to the decrease in the atmospheric NO₃⁻ mass concentrations (Fig.
- 3). There is a lag between atmospheric and skin layer NO₃⁻ i.e. atmospheric NO₃⁻ mass concentrations precede skin layer NO₃⁻ mass concentrations by day or two, however a longer time series is required to confirm this. The lag suggests that atmospheric NO₃⁻ is a source of NO₃⁻ to the skin layer, in line with Dome C where the <u>underlying</u> snow pack is the dominant source of NO₃⁻ to the skin layer via<u>photolytic recycling and re-depositionthe-overlying air in summer</u>. Furthermore, as atmospheric NO₃⁻ is deposited to the snow surface, ¹⁵N is preferentially removed first leaving the air isotopically depleted relative to the isotopically enriched snow (Frey et al., 2009). Figs. <u>3-4</u> illustrates that the δ¹⁵N-NO₃⁻ in the atmosphere is depleted with respect
- to the δ¹⁵N-NO₃⁻ in the skin layer snow. In the short time series, there are some periods where the δ¹⁵N-NO₃⁻ in the snow and atmosphere are in phase, for example, 3₋-13 January 2017. During other periods, the δ¹⁵N-NO₃⁻ in the snow and atmosphere switch to being out of phase emphasising NO₃⁻ isotopic fractionation during those periods. Both HNO₃ and peroxynitric acid (HNO₄) can be adsorbed to the snow surface in tandem (Jones et al., 2014), and although we have no direct measurements of these during the campaign, based on previous studies we suggest that HNO₃ is the dominantly adsorbed most likely form of
 - nitrogen to the skin layer (Jones et al., 2007;Chan et al., 2018),

Formatted: Normal

Formatted: Not Superscript/ Subscript Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript

We conclude that HNO₃⁺ scavenging, adsorption and cyclonic intrusions of marine air masses deliver NO₃, to the skin layer at DML in summer. During the campaign, deposition is not influenced by the transport of airmasses from the polar plateau which carry a distinct atmospheric δ^{15} N-NO₃, signature. Furthermore, the adsorption of HNO₂ on ice surfaces is temperature

- 850 dependent with higher uptake at lower temperatures (Abbatt, 1997;Jones et al., 2014). However, there is only a relatively small temperature difference between Dome C and DML (summer mean temperature -30 °C and -25 °C respectively) which is not enough to drive a large difference in HNO, uptake (Jones et al., 2014). In addition, the uptake is not dependent on the HNO, eoneentration in the air (Abbatt, 1997). However, the seasonal temperature difference at an individual site (i.e., DML or Dome C) is far greater, which could allow a seasonal dependence on the uptake and loss of NO₂- in the skin layer, which results in the retention of a greater proportion of NO₂- in summer (Chan et al., 2018). We note that due to post-depositional processes
- (section 3) any short-term signals observed in the skin layer are unlikely to be preserved. Even at the South Pole where the snow accumulation rate is slightly higher (8.5 cm yr⁺(w.e.); (Mosley Thompson et al., 1999) than DML deposition, NO₂= peaks are substantially modified after burial (Dibb and Whitlow, 1996).

860 4.2.<u>2</u>³ Annual cycle <u>Temporal variability</u> of nitrate deposition

The seasonality of simulated skin layer and atmospheric NO_3^- mass concentrations at DML matches observations at other <u>Antarctic sites</u>. We use the simulated annual cycle of NO_3^- from TRANSITS model to describe the seasonal evolution of NO_3^- deposition to DML. While NO_3^- deposited to DML can be sourced from the sedimentation of polar stratospheric clouds in winter and we assume the atmospheric NO_3^- loading is uniform under the polar vortex, in spring and summer NO_3^- net

- 865 deposition is related to local photochemistry and subsequent post-depositional processing rather than primary NO₃⁻ sources. At this time, deposition of NO₃⁻ can be through the transport of re-emitted NO₃⁻ from the surface snow at low accumulation regions of the polar plateau, or NO₃⁻ produced *in situ* at DML in spring and summer. Year-round measurements of The annual cycle of atmospheric and/or skin layer NO₃^{*} mass deposition concentration have previously been made-observed at DML (Figs. 5 and 6; Weller and Wagenbach (2007), <u>Halley Station (Mulvaney et al., 1998; Jones et al., 2011), and Neumayer</u>
- 870 Stations (Wagenbach et al., 1998), and the low snow accumulation site at Dome C (Fig. 1). These measurements describe the seasonal evolution of NO₃⁻ deposition to the skin layer from the atmosphere.(Weller and Wagenbach, 2007) indicates how much NO₃⁻ is deposited to the skin layer from the atmosphere (Figs. 5 and 6). Year-round NO₃⁻ mass concentrations have been measured in surface snow at the coastal sites of Halley (Mulvaney et al., 1998;Jones et al., 2011) and Neumayer Stations (Wagenbach et al., 1998), and the low snow accumulation site at Dome C (Fig. 1). An agreement with our simulated results ₃.
- 875 at all Antarctic sites the highest atmospheric NO₂⁺ mass concentrations are found during summer when the solar radiation is close to its annual maximum and NO₂⁺ photolysis is strongest. The summer maximum at Dome C results from co-condensation of NO₂⁺ (Book et al., 2016). This intense uplake in the skin layer in summer is driven by the strong temperature gradient in the upper few centimetres of the snow pack, highlighting that both physical (deposition; Bock et al., 2016). Chan et al., 2018) and

Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript

-	Formatted: Highlight
1	Formatted: Highlight
+	Formatted: Highlight
+	Formatted: Highlight
1	Formatted: Highlight

shemical (NO, re-emission Fribland et al. (2015) processes explain the cycling of NO, between the air and snow. The lowest NO, mass concentrations in the skin layer are found in winter. Year round atmospheric NO, data at DML and Dome C shows atmospheric NO, is at a minimum in April to June and reaches a maximum in late November, slightly out of phase with skin layer NO, Wagenbach et al., 1998. Folland et al., 2013) (Figs. <u>1 and 6)</u> (Figs. 1 and 6). The fact that the seasonality of simulated skin layer and atmospheric NO, at DML matches observations at other sites in Antarctica gives confidence in our TRANSITS model results (Fig. 6).

- 885 We also observe variability on shorter timescales. While not yet observed elsewhere on the Antarctic continent, over the short intensive sampling period at DML we observe significant variability in NO₃⁻ mass concentrations and δ¹⁵N-NO₃⁻ values that resembles a diurnal cycle. Over 4 hours, the skin layer NO₃⁻ mass concentrations varied by 46 ng g⁻¹, the skin layer δ¹⁵N-NO₃⁻ by 21 ‰, and the atmospheric δ¹⁵N-NO₃⁻ by 18 ‰. Other coastal studies have attributed daily variability to individual storm events (Mulvaney et al., 1998; Weller et al., 1999). We note that Tthe sampling duration in this study is too short to confirm
- 890 <u>any diurnal patterns but it would be interesting to investigate this further in future work. We note that due to post-depositional processes (section 4.3) any short-term signals observed in the skin layer are unlikely to be preserved.</u>

4.2.43 Nitrate mass fluxes

Our two NO₃⁻ mass flux scenarios in Fig. 5 highlight the importance of the skin layer in the air-snow transfer of NO₃⁻. Like Bome C, the greatest deposition flux of NO₃⁻ is to the skin layer. The January dry deposition flux is greater than the annual mean flux estimated by Pasteris et al. (2014) and Weller and Wagenbach (2007) which is to be expected given the higher atmospheric NO₃⁻ mass concentrations in summer (Fig. 6). The wet deposition flux, calculated for the greater DML region by Pasteris et al. (2014), falls within our two scenarios. Furthermore, the simulated archived NO₃⁻ mass flux at DML of 210 pg m⁻² s⁻¹ for the base case and <u>480 pg m⁻² s⁻¹ for the 5 cm EFD case</u> over predict s-the observed NO₃⁻ archived mass flux of 110

900 pg m⁻² s⁻¹ due to the higher simulated archived NO₃⁻ mass concentrations.- Interestingly, the simulated archived mass flux at Dome C (88 pg m⁻² s⁻¹) is lower than DML, yet the NO₃⁻ deposition flux to the skin layer in January at Dome C is similar to DML. We continue our discussion focusing on the recycling and redistribution of NO₃⁻ that occurs in the active skin layer emphasising its importance.

4.3 Recycling and Post-depositional processes

905 <u>4.3.1 Nitrate redistribution</u>

In corroboration with earlier work on the East Antarctic plateau,

Post-depositional loss and redistribution of NO_3^{-1} is the dominant control on snow pack mass concentrations and $\delta^{15}N \cdot NO_3^{-1}$ isotopic signature on the Antarctic Plateau(Erbland et al., 2015). Recycling of NO_3^{-1} at the air snow interface comprises the following processes. Nitrate on the surface of a snow crystal can be lost from the snow pack (Dubowski et al., 2001), either by

28

Formatted: Highlight Formatted: Highlight Formatted: Highlight Formatted: Highlight Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight

Formatted: Not Highlight

- - Formatted: Heading 3

- 910 UV photolysis or evaporation. UV photolysis produces NO, NO₂ and HONO while only HNO₂ can evaporate. Both of these processes produce reactive nitrogen that can be released from snow crystal into the interstitial air and rapidly transported out of the snow pack to the overlaying air via wind pumping (Zatko et al., 2013;Jones et al., 2000;Honrath et al., 1999;Jones et al., 2001). Here, NO₂ is either oxidised to HNO₃, which undergoes wet or dry deposition back to skin layer within a day, or transported away from the site (Davis et al., 2004a). If HNO₃ is re-deposited on the snow skin layer, it is available for NO₃⁻ photolysis and/or evaporation again. Nitrate can be recycled multiple times between the boundary layer and the skin layer before it is buried in deeper layers of the snow pack. Photolysis and/or evaporation of NO₃⁻ and subsequent recycling between the air and snow alters the concentration and δ⁴⁵N-NO₃⁻ that is ultimately preserved in ice cores. Nitrate recycling therefore redistributes NO₃⁻ from the active snow pack column to the skin layer via the atmosphere. Any locally produced NO₂ that is transported away from the site of emission represents a loss of NO₃⁻ from the snow pack.
- 920 4.3.1 Evaporation

The desorption of HNO₂ from the snow crystal reduces the NO₂-concentration in the snow in coastal Antarctica (Mulvancy et a<mark>l., 1998), The evaporation of HNO₂ is a two-step process, which involves the recombination of NO₂-+H^{*}---> HNO₂ followed by a phase change to HNO₃ (gas-phase). First, theoretical estimates indicated that evaporation of HNO₂ should preferentially remove ¹⁹N from the snow and release to the atmosphere leading to depletion in 8⁴⁵N-NO₂⁻⁻in the residual snow pack (Frey et</mark>

- 925 al., 2009). Furthermore, recent laboratory experiments showed that evaporation imposes a negligible fractionation of 8⁴⁴N-NO₂^{*} (Erbland et al., 2013;Shi et al., 2019). However, we find that the snow pack is enriched in 8⁴⁵N-NO₂^{*} relative to the atmosphere at DML (Figs. 3 and 6) and at Dome C (section 4.3.2). This fractionation observed in field studies cannot therefore be explained by evaporation, and must be attributed to different processes. It therefore follows that evaporation must be only a minor process in the redistribution of NO₂^{*} between atmosphere and the snow pack above the Antaretic plateau.
- 930 4.3.2 Photolysis

We focus our discussion on photolysis, which is the dominant process responsible for NO₂⁻-loss⁻ and redistribution and associated δ¹⁵N-NO₂⁻-isotopic fractionation at low accumulation sites in Antarctica (Erbland et al., 2013;France et al., 2011). Nitrate photolysis occurs in the photoehemically active zone of the snow pack. Inown as the snow photic zone. Below this NO₂⁻ is buried. Nitrate photolysis in the active snow pack results in the production of NO₂-leading to a reduction in the NO₂⁻
 generation with depth in the snow pack (Fig. 4). In the photolysis induced fractionation of NO₂⁻, ^{ae}N is preferentially removed first resulting in an enrichment of δ¹⁵N-NO₂⁻ in the snow pack. An individual snow layers undergo strong δ¹⁶N-NO₂⁻ enrichment as they are exposed to UV near the surface for the longest; late summer and autumn layers experience less δ¹⁵N-NO₂⁻ enrichment as they are exposed for less time before sunlight disappears at the start of polar-winter, during which new

940 precipitation buries existing snowfall.

29

Formatted: Highlight	
Formatted: Highlight	
Formatted: Highlight	
-	
Formatted: Highlight	
Field Code Changed	

Formatted: Highlight

We provide five lines of evidence that photolysis is the dominant process for NO₃⁻ recycling and redistribution at DML.-we find clear evidence of NO₃⁻ redistribution via photolysis at DML, and confirmation of our hypothesis that UV-photolysis is driving NO₃⁻ recycling at DML. Firstly, the highly enriched δ^{15} N-NO₃⁻ values of snow at DML and other Antarctic sites(-3 to 99 ‰), and the highly depleted atmospheric δ^{15} N-NO₃⁻ values at DML (-20 to -49 ‰) are among the most extreme observed

- 945 on earth (Fig. <u>S8S7</u>;) Savarino et al. (2007), and cannot be explained by any known anthropogenic, marine or other natural sources. The δ¹⁵N-NO_x-NO_x⁻-source signature of the main natural NO_x sources (biomass burning, lightning, soil emissions-is lower; δ¹⁵N-NO_x³ < 0 ‰) is lower than anthropogenic NO_x sources, which generally have positive δ¹⁵N-NO_x⁻- values (-13< δ¹⁵N-NO_x³ < 13 ‰; e.g.) Hastings et al. (2013);Kendall et al. (2007);Hoering (1957) except in the case of vehicle and fertilised soil NO_x emissions which have negative δ¹⁵N-NO_x values (-60< δ¹⁵N-NO₃⁻ < 12 ‰; Miller et al. (2017);Yu and Elliott</p>
- 950 (2017);Miller et al. (2018);Li and Wang (2008). However, a NO₃⁻ source contribution from fertilised soil NO₈ emissions to Antarctica is thought to be minor (Lee et al., 2014). Such low atmospheric δ¹⁵N-NO₃⁻ values at DML show a marked difference to other mid-latitude tropospheric aerosol (-10< δ¹⁵N-NO₃⁻ <10 ‰; Freyer (1991). We acknowledge that stratospheric NO₃[±] contributes to NO₂⁻ mass concentrations in snow in Antarctica. Although its isotopic signature is uncertain, estimates of stratospheric δ¹⁵N-NO₃⁻ are 19 ± 3 ‰ (Savarino et al., 2007), and fall well outside of atmospheric observations at DML.
- 955 <u>T</u>Therefore, δ¹⁵N-NO₃⁻ observations of aerosol, skin layer and snow pit at DML (-49< δ¹⁵N-NO₃⁻<99 ‰) lie outside-of the range of natural and anthropogenic source end members (with the exception of anthropogenic emissions NO₃ from vehicle and fertilised soil which can be ignored as a source to Antarctica), and thus cannot be explained by a mixture of sources (Fig. S7) or attributed. Thus, our measurements at DML are unrelated to seasonal variations in mid-low latitude NO₃ sources e.g. increased springtime agricultural emissions, which has been observed in the mid-latitudes. The contribution of natural sources
- 960 to the Greenland snow pack δ¹⁴N NO₃⁻ signature has also been discarded (Geng et al., 2014;Geng et al., 2015). Furthermore, the negative atmospheric δ¹⁴N NO₃⁻ values at DML (-20 to -49 ‰) are extremely low. Such low atmospheric δ¹⁴N NO₃⁻ values have only been observed in Antarctica, and show marked difference to other mid-latitude tropospheric aerosol (-10< δ¹⁴N NO₃⁻ values) have only been observed in Antarctica, and show marked difference to other mid-latitude tropospheric aerosol (-10< δ¹⁴N NO₃⁻ values) have only been observed in Antarctica, and show marked difference to other mid-latitude tropospheric aerosol (-10< δ¹⁴N NO₃⁻ values) have only been observed in Antarctica, and show marked difference to other mid-latitude tropospheric aerosol (-10< δ¹⁴N NO₃⁻ contributes); Freyer (1991); Li and Wang (2008); Miller et al. (2018); Yu and Elliott (2017). We aeknowledge that stratospheric NO₃⁻ contributes to NO₃⁻ mass concentrations in snow in Antarctica. Although its isotopic signature is uncertain, estimates of stratospheric δ¹⁴N NO₃⁻ are 19 ± 3 ‰ (Savarino et al., 2007), and fall well outside of atmospheric observations at DML. The

unique <u>snow and aerosol</u> δ^{15} N-NO₃⁻ signature_<u>of low accumulation Antarctic snow and aerosol</u> is thus related to postdepositional processes specific to low accumulation sites in Antarctica.

Secondly, denitrification of the snow pack is seen through the δ¹⁵N-NO₃⁻ signature which evolves from the enriched snow pack (-3 to 99 ‰), to the skin layer (-22 to 3 ‰), to the depleted atmosphere (-49 to -20 ‰) corresponding to mass loss from the snow pack (Figs. 4 and S7). Denitrification causes the δ¹⁵N-NO₃⁻ of the residual snow pack NO₃⁻ to increase exponentially as

NO₃⁻ mass concentrations decrease. Thirdly, sensitivity analysis with TRANSITS, where photolysis is the driving process, is able to explain the observed snow pit

 δ^{15} N-NO₃ variability when the e-folding depth is taken into account (section 4.5).

30

Formatted: Not Highlight

	ThFourthly, enrichment of δ^{15} N-NO ₃ ⁻ is observed in the top 30 cm of the snowpack at DML indicating NO ₃ ⁻ photolytic		
975	redistribution at DML in the photic zone of the snow pack (Fig. 7).i		
	Nitrate isotope enrichment takes place in the top 25 cm, which is consistent with an e-folding depth of 10 cm used in the base		
	$ \underline{ case \ scenario.} \ \underline{In \ the \ photic \ zone} \\ \underline{Here}, \ the \ \delta^{15} N-NO_3^- \ observations \ closely \ match \ the \ simulated \ \delta^{15} N-NO_3^- \ values \ \underline{from} \ begin{tabular}{lllllllllllllllllllllllllllllllllll$		
	TRANSITS and show enrichment to this depth indicating NO3-photolytic redistribution at DML in the active photic zone of		
	the snow pack (Fig. 7). Below the photic zone, 8 ¹⁴ N-NO ₂ -values oscillate around a mean of -125 5. The mean values of the		Formatted: Highlight
980	845 N-NO3* observations are lower than the simulated values, which could be related to uncertainties in a number of factors, for		
	example: i) a shallower e-folding depth than modelled. During our field measurements, we derived a lower e-folding depth of		
	2-5 cm (Fig. S1) at DML which could explain the lower enrichment in δ ¹⁵ N-NO3 ⁺ (section 4.5.2), ii) lower JNO3 ⁺ values (NO3 ⁺		
	photolysis rate), which are related to a lower e-folding depth, and would lead to less enrichment of 8 ⁴⁵ N-NO ₂ ⁻ in the snow		
	pack, iii) higher atmospheric NO37-input, however-8 ⁴⁴ N-NO37 values are not sensitive to variable atmospheric NO37-mass		
985	concentrations (Frbland et al., 2015), and/or iv) variable accumulation which would shift the oscillations to the correct depth		Formatted: Highlight
	and lower the mean 8 ⁴⁵ N-NO ₂ -values below the photic zone (section 4.5.1). The difference between the simulated and snow		Formatted: Highlight
	pit values shows that DML site is less sensitive to photolysis than we expected from TRANSITS modelling of 8 ¹⁵ N-NO ₅ ⁺ along		
	an accumulation gradient (Erbland et al., 2015).		Formatted: Highlight
	-Additionally, although not the focus of the study, denitrification causes the δ^{18} O-NO ₃ values to increase in the residual NO ₃ -	5.7-	Formatted: Highlight
990	snow pack.		Formatted: Font: 10 pt, Font color: Auto
	Thirdly, the application of TRANSITS to DML observations show that our observed atmospheric, skin layer and snow depth		
	profiles of 814N-NOg- are similar to the simulated values where photolysis is the driving process (Figs. 6-7) Sensitivity analysis		
	with TRANSITS is able to explain the observed snow pit 8 ¹⁵ N-NO ₃ ⁻ variability (section 4.5). <u>the application of TRANSITS to</u>		
	DML observations show that our observed atmospheric, skin layer and snow depth profiles of 845N-NO2- are similar to the		
995	simulated values where photolysis is the driving process (Figs. 6-7). Nitrate isotope enrichment takes place in the top 25 cm,		
	which is consistent with an e-folding depth of 10 cm. Here, the δ^{15} N-NO ₃ ⁻ observations closely match the simulated δ^{15} N-NO ₃ ⁻		
	values and show enrichment to this depth indicating NO3-photolytic redistribution at DML in the active photic zone of the		
	snow pack (Fig. 7). Below the photic zone, δ^{15} N-NO ₃ ⁻ values oscillate around a mean of ~125 ‰. The mean values of the		
	δ^{15} N-NO ₃ ⁻ observations are lower than the simulated values, which could be related to uncertainties in a number of factors, for		
1000	example: i) a shallower e-folding depth than modelled. During our field measurements, we derived a lower e-folding depth of		
	$2-5$ cm (Fig. S1) at DML which could explain the lower enrichment in δ^{15} N-NO ₃ ⁻ (section 4.5.2), ii) lower JNO ₃ ⁻ values (NO ₃ ⁻)		
	photolysis rate), which are related to a lower e-folding depth, and would lead to less enrichment of 845N-NO3- in the snow		
	pack, iii) higher atmospheric NO3 ⁻ input, however-8 ¹⁴ N-NO3 ⁻ values are not sensitive to variable atmospheric NO3 ⁻ mass		
	concentrations (Erbland et al., 2015), and/or iv) variable accumulation which would shift the oscillations to the correct depth		
1005	and lower the mean 8 ⁴⁵ N-NO3 ⁻ values below the photic zone (section 4.5.1		
	Lastly,-we-calculated fractionation constants s- $(^{15}\epsilon_{app})$ using our simulated results from the TRANSITS model base case		
	[Nitrate evaporation from the snow pack has a ¹⁵ e _{upp} of -0 as determined by two independent studies [Erbland et al., 2013;Shi		Field Code Changed
I	21		

et al., 2019). This indicates that during NO $_2$ -evaporation, the air above the snow is not replenished and thus there is only ϵ small NO2 mass loss. The isotopic fractionation of NO2 evaporation is negligible across most of Antarctica at cold temperatures of <-24 °C (Shi et al., 2019) which is the case for DML. However, evaporation of NO₂ at warmer temperatures 010 (4 °C) depletes the heavy isotopes of NO2 remaining in the snow, and decreases the 845N-NO2 and the remaining snow by a few % contrary to isotope effects of photolysis. In comparison, fractionation constants associated with laboratory studies and -60 % (Frey et al., 2009; Erbland et al., 2013), respectively. The negative fractionation constant obtained from photolysis 015 implies that the remaining NO2- in the skin layer snow is enriched in 845N-NO2-. In turn, the atmosphere is left with the source of NO* that is highly depleted in 815N-NO2- This enrichment (depletion) is exactly what we observe in the snow pack (atmosphere) at DML (Figs. 4 and 6). The marked difference in values from the evaporation experiments and those observed in snow at Dome C allows us to separate out the isotopic signature of evaporation and photolysis processes at DML of 12 ‰ and 19 ‰ using the active photic zone section of the snow pack (top 30 cm), and using our simulated results from the 020 TRANSITS model respectively. The later nicely matches theall in the range of expected $^{15}\varepsilon_{app}$ values (-59< $^{15}\varepsilon_{app}$ < -16 ‰) within the "transition zone" characterised by snow accumulation rates typical of sites located between the Antarctic plateau and coast of (5 - 20 cm yr⁻¹ (w.e.,-);) modelled by Erbland et al. (2015). While the $15 \epsilon_{app}$ for the 5 cm EFD case is lower than predicted for a site with the same snow accumulation rate highlighting the sensitivity of e-folding depth on NO3 redistribution. , and implies a weaker photolytic loss of NO₄⁻ than Dome C (⁴⁵C_{app} < 59 ‰) (Erbland et al., 2013). The discrepancy between 025 our observed and simulated ⁴⁵Game is due to the higher snow accumulation rate, which preserves seasonality, and with a noisy signal, there is no pure separation of the loss processes assuming Rayleigh isotopic fractionation. The single-process Raleigh model does not work well at sites with annual signal in 8¹⁵N-NO₃: Erbland et al. (2013) noted that uncertainties in the ¹⁵E_{app} for snow pits in the transition zone were greater than coastal and plateau zones indicating that the assumed single loss Rayleigh model is not appropriate for transition zones. The discrepancy between our observed (12 ‰) and simulated (-19 and -11 ‰ 030 for the base case and 5 cm EFD case respectively) ¹⁵E_{app} is due to the higher snow accumulation rate, which preserves seasonality, and with a noisy signal, there is no pure separation of the loss processes assuming Rayleigh isotopic fractionation.

4.3.23 Nitrate recycling

Only two-three studies have attempted to quantify the degree of NO₃⁻ recycling between the air and snow (Davis et al., 2008;Erbland et al., 2015;Zatko et al., 2016). Erbland et al. (2015) used the TRANSITS model to estimate that NO₃⁻ is recycled 4 times on average before burial beneath the photic zone at Dome C, similar to the findings of Davis et al. (2008) for the same site. Using the approach of Erbland et al. (2015), we find that NO₃⁻ is recycled 3 times on average before it is archived at DML for the base case, and 2 times on average for the 5 cm EFD case. Thus, a shallower e-folding depth reduces the recycling strength. These findings are A lower recycling factor than Dome C is consistent with spatial patterns of NO₃⁻ recycling factors across Antarctica reported by Zatko et al. (2016). Erbland et al. (2015)As Dome C and DML lie on the same latitude (75° S),

incoming UV-radiation (except for cloud cover) should not impact the efficiency of photolysis and thus recycling at the two sites. While photolysis driven NO₃⁻ recycling can occur at all polar sites, the most intense enrichment of ³¹⁵N-NO₃⁻ in the depth profile is seen at Dome C and Vostok (Erbland, 2011). Below we provide some explanations for the weakened recycling at DMI.

1045 1. Higher snow accumulation rate

The TRANSITS modelling shows the influence of the snow accumulation rate on the depth profile of NO₃⁻ mass concentration and δ¹⁵N-NO₃⁻, including the preservation of a seasonal cycle at higher snow accumulation rates (Fig. 7). At low accumulation sites, i.e., Dome C, the annual layer thickness is thinner so that NO₃⁻ in the skin layer and thinnerthose layers snow layers is exposed to sunlight (and the actinic flux) and photochemical processes for longer allowing-resulting more photochemistry and thus a very active snow pack with strong NO₃⁻ recycling and δ¹⁵N-NO₃⁻ enrichment in the snowpackt. At DML, which has a higher snow accumulation rate than Dome C, the snskin-ow layers are-is buried more rapidly, leaving less time tforo adsorb additional HNO₃ to adsorb to the skin layer from the atmosphere and less time for photolysis to redistribute snow pack NO₃⁻ to the overlying air for re-adsorption to the skin layer. Therefore, photolysis-driven recycling of NO₃⁻ is largely dependent on the time that NO₃⁻ remains in the snow photic zone. Following photolysis at DML, the recycling of NO₃⁻ in snow remains intact as there is less redistribution and a lower loss of NO₃⁻ than at Dome C.

2. Shallower e-folding depth

Based on measurements we derived an e-folding depth for DML ranging between 2 and 5 cm (Fig. S1). This estimate is similar to a modelled value at South Pole (3.7 cm; Wolff et al. (2002) which has a similar accumulation rate, and Alert, Canada (5-6 em; King and Simpson, 2001). The e-folding depth at Dome C is considerably deeper, ranging between 10 cm to 20 cm 060 depending on the snow properties (France et al., 2011) (France et al., 2011). The e-folding depth depends on the density and grain size of snow crystals, and the concentration of impurities. In terms of published values, impurity concentrations are generally higher at DML, for example dust and major ion concentrations (Delmonte et al., 2019;Legrand and Delmas, 1988), due to proximity to marine sources. Yet station pollution is greater at Dome C (Helmig et al., 2020), and thus the lower e-065 folding depth is unrelated to black carbon concentrations. Furthermore, there is considerate variability in snow grain size isacross Antarctica. The larger e-folding depth in windcrust layers at Dome C is due to larger grain sizes in those layers (France et al., 2011). Snow grain size may be smaller at DML, which will increase scattering (Brucker et al., 2010), but further work is required to confirm if this is the dominate factor influencing the lower e-folding depth at DML. (France et al., 2011)Sensitivity studies show that NO3- impurities make a small contribution to the e-folding depth compared to scattering by 070 snow grains which dominate (France et al., 2011; Chan et al., 2015; Zatko et al., 2013). (France et al., 2011; Delmonte et al.,

2019;Legrand and Delmas, 1988)

3. as an (Frey et al., 2009) edto 'France et al. (2012)Using concentrations in the range of those observed at Dome C and DML we find that e hasTherefore, the France et al. (2011);Zatko et al. (2013)_Wthat The larger e folding depth at Dome C is due to the larger grain sizes (France et al., 2011;Gay et al., 2002) and low impurity content. The Formatted: Numbered + Level: 1 + Numbering Style: 1, 2, 3, ... + Start at: 1 + Alignment: Left + Aligned at: 1.27 cm + Indent at: 1.9 cm



1075	impact of impurities in the range of observed polar snow concentrations on e-folding depth is small compared to	
	the contribution from scattering by snow grains (France et al., 2011;Zatko et al., 2013). Lower photolysis rate	
	At DML, NO2 ⁻ photolysis produces a lower snow emission flux of NO2 produces a lower NO2 flux to the atmosphere and	Formatted: Normal, No bullets or numbering
	lower ¹⁵ C _{app} than at Dome C (section 3.8). This is due to i) the shallower e-folding depth compared to Dome C which implies	
	reduced emission flux of NO ₃ , and ii) the reduced UV exposure time of surface snow due to higher annual snow accumulation	
1080	compared to Dome C. Furthermore, the large ${}^{15}\varepsilon_{app}$ associated with NO ₃ ⁻ photolysis has been determined for snow at Dome C	
	(Berhanu et al., 2014;Frey et al., 2009;Erbland et al., 2013) and DML. At both sites, $\delta^{15}N-NO_3^{-1}$ is enriched in the remaining	
	skin layer. However, at DML, the ${}^{15}\varepsilon_{app}$ is lower which implies a weaker photolytic loss of NO ₃ ⁺ associated with a higher snow	
	$accumulation rate. The lower snow emission flux of NO_3 and lower {}^{15}\epsilon_{app} highlighting are evidence of a that the photolysis rate$	Formatted: Not Superscript/ Subscript
	is lower thus thereduced recycling strength at DML relative to Dome C. is reduced (section 4.3.23).	
1085	<u>Furthermore, the large ¹⁵ Gung associated with NO2 photolysis has been determined for snow at Dome C (!!!</u>	Formatted: List Paragraph, Numbered + Level: 1 +
	INVALID CITATION !!! (Berhanu et al., 2014;Frey et al., 2009;Erbland et al., 2013)) <u>and DML. At both sites.</u>	Numbering Style: 1, 2, 3, + Start at: 1 + Alignment: Left + Aligned at: 1.27 cm + Indent at: 1.9 cm
	δ^{15} N-NO ₂ ⁻ is enriched in the remaining skin layer snow. However, at DML, the ¹⁵ c _{mm} is lower due to less active	
	photochemistry associated with a higher snow accumulation rate. Our results are consistent with Zatko et al.	
	(2016) who suggest that the large fractionation constant associated with photolysis is greatest on the polar plateau	
1090	where strong winds are most efficient at exporting NO3 away from the site.	
	<u>i.4.</u> Lower nitrate uptake at warmer temperatures	
	Temperature can control skin layer NO3-uptake and loss. At colder snow temperatures, there is greater adsorption of HNO3 to	
	the skin layer (Abbatt, 1997; Jones et al., 2014). Although the difference in the mean annual temperature at Dome C compared	
1095	to DML (~5 °C) is not large enough to explain the significantly higher skin layer NO3 ⁻ mass concentrations there. Compounding	
	this, NO ₃ -loss by evaporation is also dependent on temperature with maximum NO ₃ -loss at higher temperatures, i.e., diffusion	
	of NO37 in ice is slower at colder temperatures (Thibert and Domine, 1998). A compilation of NO37 concentration data from	
	Greenland and Antarctic ice cores showed that at very low accumulation rates lower temperatures lead to higher NO3 ⁻ mass	
	concentrations preserved in the snow (Röthlisberger et al., 2000). Although the snow accumulation rate is closely linked to	
1100	temperature, photolysis is the dominant NO3-loss process at low snow accumulation sites in Antarctica. Therefore, any	
	differences in temperature between DML and Dome C could partly explain the greater uptake of HNO3 to the skin layer, higher	
	mass concentrations of NO3- in the skin layer, and stronger recycling at Dome C compared to DML TFurthermore, the	Formatted: Not Highlight
	adsorption of HNO3 on ice surfaces is temperature dependent with higher uptake at lower temperatures (Abbatt, 1997; Jones et	
	al., 2014), Nitrate loss by evaporation is also dependent on temperature with maximum NO3 ⁻ loss at higher temperatures	Formatted: Not Highlight
1105	(Thibert and Domine, 1998;Röthlisberger et al., 2000). The seasonal temperature difference at an individual site (i.e., DML or	
	Dome C) could allow a seasonal dependence on the uptake and loss of NO3 ⁻ in the skin layer, which results in the retention of	
	a greater proportion of NO3 ⁻ in summer (Chan et al., 2018). However, t _{However} , there is only a relatively small temperature	
	difference between Dome C and DML (summer mean temperature -30 °C and -25 °C respectively) (Table 1) which is not	Formatted: Not Highlight
I	34	

enough to drive a large difference in HNO₃ uptake (Jones et al., 2014). In addition, the uptake is not dependent on the HNO₃
 concentration in the air (Abbatt, 1997). However, the seasonal temperature difference at an individual site (i.e., DML or Dome
 <u>C</u>) is far greater, which could allow a seasonal dependence on the uptake and loss of NO₃⁻ in the skin layer, which results in the retention of a greater proportion of NO₂⁻ in summer (Chan et al., 2018).

ii.<u>1. Lower photolysis rate</u>

115 iii.<u>1.</u> At DML, NO₂⁻ photolysis produces a lower NO_{*} flux to the atmosphere and lower ⁴⁵e_{wpp}-highlighting that the photolysis rate is lower thus the recycling strength is reduced (section 4.3.2). Furthermore, the large ⁴⁵e_{wpp} associated with NO₂⁻ photolysis has been determined for snow at Dome C (Berhanu et al., 2014;Frey et al., 2009;Erbland et al., 2013) and DML. At both sites, δ⁴⁵N NO₂⁻ is enriched in the remaining skin layer snow. However, at DML, the ⁴⁵e_{wpp} is lower due to less active photochemistry associated with a higher snow accumulation rate. Our results are consistent with Zatko et al. (2016) who suggest that the large fractionation constant associated with photolysis is greatest on the polar plateau where strong winds are most efficient at exporting NO₂⁻ away from the site.

iv.5.Lower export of locally produced nitrate

The degree of NO₃⁻ recycling is also determined by atmospheric transport patterns across Antarctica. Export of locally produced NO_x on the Antarctic Plateau-leads to greater enrichment in the depth profiles of δ^{15} N-NO₃⁻ relative to the coast due to isotopic mass balance (Savarino et al., 2007;Zatko et al., 2016)-, Zatko et al. (2016) modelled the export of snow sourced NO_x away from the original site of NO₃⁻ photolysis, and found that the largest loss of NO₃⁻ occurs in central Antarctica where most NO_x⁻ is transported away by katabatic winds. At the coast, photolysis driven loss of NO₃⁻ from the snow is minimal due to high snow accumulation rates. Here, Oobservations of enriched atmospheric δ^{15} N-NO₃⁻ at the coast show-suggest that NO_x

130 has been transported sourced away from *in situ* production on the Antarctic Plateau the location of its production on the Antarctic Plateau to the coast (Savarino et al., 2007;Morin et al., 2009;Shi et al., 2018). <u>If there was less The greater export of NO₃⁻ from Dome C allows efficient removal of recycled NO₃⁻ from that site, resulting in a lower archived NO₃⁻ mass flux and enriched δ¹⁵N-NO₃⁻ signature in the surface snow. The enrichment of δ¹⁵N-NO₃⁻ is due to the isotopic mass balance rather than an increase for photolysis intensity. With less export of NO₃⁻ away from the DML site than Dome C, locally sourced NO₃⁻ models not be sourced NO₃⁻ and the depth profile of the δ¹⁵N-NO₃⁻ would beis redeposited back to the skin layer at the site of the emission and the depth profile of the δ¹⁵N-NO₃⁻ would is not be</u>

as dramatically impacted as sites where there is substantial loss of NO3-Dome C where there is substantial loss of NO3-.

Formatted: Numbered + Level: 1 + Numbering Style: 1, 2, 3, ... + Start at: 1 + Alignment: Left + Aligned at: 1.27 cm + Indent at: 1.9 cm

Field Code Changed

-	Formatted: Not Highlight
{	Formatted: Not Highlight
1	Formatted: Not Highlight
4	Formatted: Not Highlight

-{	Formatted: Not Highlight
1	Formatted: Not Highlight
1	Formatted: Not Highlight
1	Formatted: Font: Italic
1	Formatted: Not Highlight
1	Formatted: Not Highlight
-{	Formatted: Not Superscript/ Subscript
-{	Formatted: Not Highlight
-{	Formatted: Not Highlight

	4.3.3 Snow emission of NO2	
	We estimate the potential snow emission flux of NO ₂ (F _{NO2}) from NO ₃ ⁻ photolysis in snow using Eq. (8).	Formatted: Heading 3
	$F_{NO2} = \int_{z=0.m}^{z=1.m} [NO_3^-]_z J(NO_3^-) dz \qquad $	
140	where $J_{a}(NO_{3}^{-})$ is the photolysis rate coefficient of reaction $NO_{3}^{-} \pm hv \rightarrow NO_{2} \pm O^{-}$ at depth, z, in the snowpack, and is derived	
	by scaling surface measurements (section 2.6) with e-folding depth (= 2-10 cm), and [NO3-]# is the amount of NO3-per unit	
	volume of snow at depth, z, in the snowpack. The calculated F_{NO2} is a potential emission flux assuming that all NO ₂ ⁻ within	
	the snow grain is photo-available, no cage effects are present and NO2 is vented immediately after release from the snow grain	
	to the air above the snowpack without undergoing any secondary reactions. For the 1 to 14 January 2017 period, model	
145	estimates of F _{NO2} scaled approximately linearly with e-folding depth were 0.4, 1.0 and 1.9 x 10 ⁴⁴ molecule m ⁻² s ⁴ for e-folding	
	depths of 2, 5 and 10 cm, respectively. Spatial variability of NO37 in the top 30 cm of surface snow at DML based on snow pit	
	A and B is on the order of 13 % inducing similar variability in the model estimates of F_{NO2} . Estimates of F_{NO2} at Dome C,	
	based on the same model during 1 to 14 January 2012, were larger with 1.2-7.3 x 10^{44} molecule m ⁻² s ⁻⁴ (Frey et al., 2013),	
	mostly due to larger J(NO ₃ ⁻) values observed above the surface (section 2.6) as well as a larger e-folding depth (= 10 cm near	
150	the surface). It should be noted that the observed F_{NOx} was found to be up to 50 times larger than model estimates, which is	
	attributed to the poorly constrained quantum yield of NO3 ⁻ photolysis in natural snow	
	-(Frey et al., 2015;Frey et al., 2013). In summary, the weakened air-snow recycling at DML is due to i) the shallower e-folding	
	depth compared to Dome C which implies reduced emission flux of NO _x , and ii) the reduced UV exposure time of surface	
	snow due to higher annual accumulation compared to Dome C. We estimate that NO3 ⁻ has a mean lifetime in the skin layer of	
155	12 days to 3 years before it is photolysed back to atmosphere.	
	Based on field, laboratory and theory, we conclude that NO, photolysis is the dominant-post-depositional-process on the	Formatted: Highlight
	Antarctic plateau controlling NO ₂ -mass concentrations and δ^{14} N-NO ₂ -values in the snow and atmosphere. Nitrate photolysis	
	in snow causes 8 ³³ N-NO ₂ ⁺ fractionation of the magnitude needed to explain field and lab observations. The development of	
	TRANSITS allows us to model the archived of NO2 values taking into account all parameters in the air snow system.	

1160 4.4 Preservation and archival

We provide new constraints on the archival values and archival time of NO₃⁻ at DML. The photolysis driven recycling of NO₃⁻ is largely dependent on the time that NO₃⁻ remains in the snow photic zone. Post depositional loss of NO₃⁻ at DML was quantified in a number of firn cores and snow pits by Weller et al. (2004) who found that -26 % of the NO₃⁻ originally deposited to the snow pack was lost. The e-folding time for NO₃⁻ at the site was reported as -20 years, and NO₃⁻ was archived after 5 to 6 years of deposition (or 1.1 – 1.4 m depth) which is the time it takes for the NO₃⁻ mean concentration to become representative of the last 100 years. At this point, the authors considered post-depositional loss of NO₃⁻ to be negligible, and therefore archived. However, no skin layer measurements were made in the study and given how active the skin layer is NO₃⁻.

redistribution and recycling, we use our skin layer measurements to provide new constraints on the archival values and time of NO₄⁻ at DML.

- 170 Taking the high <u>observed</u> skin layer NO₃⁻ mass concentrations into account (average of 230 ng g⁻¹ in January for DML), we calculate a <u>post-depositional</u> NO₃⁻ loss of 60 ng g⁻¹ (or 75 %) and enrichment of 170 ‰ from the snow pack <u>at DML following</u> the <u>approach of</u> Weller et al. (2004). Fig. 7 shows a clear signal of δ^{15} N-NO₃⁻ enrichment in the top 30 cm of the snowpack where the simulated 5 cm EFD case depth profile parallels the observed depth profile indicating NO₃⁻ photolytic redistribution at DML in the photic zone of the snow pack. Assuming all NO₃⁻ is archived below the photic zone, i.e., an e-folding depth of
- 1175 5 cm, archival occurs below a depth of 15 cm, where NO₃⁻ has a residence time of 0.75 years in the photic zone corresponding to one summer. At this point, the amplitude of the annual cycle of <u>observed</u> δ¹⁵N-NO₃⁻ at DML does not vary. <u>Our observed</u> archived values of 50 ‰ and 60 ng g⁻¹ agree well with the mean values of the snow pit below the photic zone, and the archived δ¹⁵N-NO₃⁻-values of the 5 cm EFD case (50 ‰). The for the 5 cm EFD case are waxed and for the base case. The seasonal variability of the simulated δ¹⁵N-NO₃⁻⁵ cm EFD case depth profile for the 5 cm EFD case is constant between 30-80 ‰ below.
- 1180 the photic zone indicating that no further enrichment or NO₃ redistribution is taking place in the archived section of the snow pack. The DML site has a lower observed archived δ^{15} N-NO₃ value and is less sensitive to NO₃ recycling photolysis than expected from TRANSITS modelling of δ^{15} N-NO₃ along a snow accumulation gradient (Table 2; Erbland et al. (2015), and we suggest this is due to the lower observed e-folding depth than modelled. Differences between the values for the two scenarios and could be 12 What about the higher simulated mass concentrations than observed — why does a lower e-folding

185 depth increase the nitrate conc???

Despite the relatively high NO₃⁻ mass concentrations and enriched δ^{15} N-NO₃⁻ in the skin layer at DML, clear seasonal cycles remain in the depth profile in contrast to the lower snow accumulation site of Dome C where the depth profile is relatively constant below the photic zone (Fig. 4). At higher snow accumulation rates, the seasonality of atmospheric NO₃⁻ mass concentrations and δ^{15} N-NO₃⁻ is preserved due to faster burial. Even at 6 cm yr⁻¹ (w.e.), the snow layers remain in the active

190 photic zone for 0.75 years and the weaker recycling factor is low enough to conserve the seasonality. Whereas at Dome C, snow layers remain within the photic zone for longer (about 3 years or 3 summers), due to the deeper e-folding depth and NO₂ emission and redistribution continues until the seasonal cycle becomes smoothed (Fig. 4). At Dome C, archival of NO₃⁻ occurs below a depth of 30 cm. Compared to Dome C, the archived values at DML have a similar mass concentration (Dome C: 35 ng g⁻¹) but lower δ¹⁵N-NO₃⁻ value (Dome C: 300 ‰), due to the deeper photic zone, stronger redistribution and recycling there.

195

Our archived values of 50 ‰ and 60 ng g⁺ agree well with the mean values of the snow pit below the photic zone (30 cm), and are lower than the simulated archived values from TRANSITS (120 ng g⁺ and 130 ‰) due to the stronger photochemistry in the model. Due to the larger e-folding depth and hence larger photic zone at Dome C, NO₃- has a longer residence time of 3 years (3 summers) in the photic zone. Here, archival of NO₃- occurs below a depth of 30 cm.
 Compared to Dome C, the archived values at DML have a similar concentration (Dome C: 35 ng g⁺) but lower δ⁴⁵N-

NO3⁻ value (Dome C: 300 ‰), due to the thicker photic zone, stronger redistribution and recycling there.

37

Formatted: Not Superscript/ Subscript

-{	Formatted: Not Highlight
1	Formatted: Not Highlight
	~
-	Formatted: Not Superscript/ Subscript
4	Formatted: Not Superscript/ Subscript

Formatted: Not Superscript/ Subscript

Formatted: Subscript

4.5 Sensitivity of δ^{15} N-NO₃⁻ to deposition parameters and implications for interpreting ice core records of δ^{15} N-NO₃⁻ at DML

As first proposed by Frey et al. (2009) and later confirmed by field and lab studies (Erbland et al., 2015;Berhanu et al., 2014;Shi 1205 et al., 2019) it is UV-photolysis of NO₃⁻ that dominates post-depositional fractionation of δ^{15} N-NO₃⁻ in snow and firn. Yet the

extent of photolytic fractionation and the δ^{15} N-NO₃⁻ signature ultimately preserved in firm and ice depends on the UV-spectrum of down-welling irradiance and, on the time snow layers are exposed to incoming UV-radiation as well as on the snow optical properties. Previous studies showed that δ^{15} N-NO₃⁻ is sensitive to TCO but also to deposition parameters such as the annual snow accumulation rate (Shi et al., 2018;Noro et al., 2018;Erbland et al., 2013). Thus, if all deposition parameters remained

- constant or are well-constrained it should be theoretically possible to use δ^{15} N-NO_{3^{*}} as an ice core proxy for past surface UVradiation and stratospheric ozone. Understanding the depositional parameters and their impact on δ^{15} N-NO₃^{*} is paramount for the interpretation of δ^{15} N-NO₃^{*} signals preserved in ice cores. As the interpretation of δ^{15} N-NO₃^{*} is site-specific, we investigate the sensitivity of the δ^{15} N-NO₃^{*} signature at DML to snow accumulation rate, e-folding depth and TCO. <u>Throughout section</u> 4.5 we compare sensitivity results to a "base case" simulation which was simulated using the <u>As the</u> mean annual snow
- 215 accumulation rate at DML is of 6 cm (w.e.) yr¹ and an e-folding depth of 10 cm, we take this simulation as our base case. The base case simulation and snow pit δ¹⁵N-NO₃⁺ depth profiles parallel each other in the top 30 cm of the snow pack, but below the active photic zone, there is an offset between the depth profiles in terms of i) the amplitude of the summer and winter δ¹⁵N-NO₃⁺ values, and ii) the mean δ¹⁵N-NO₃⁺ value (Fig. 7).

4.5.1 Sensitivity of the ice core δ^{15} N-NO₃- signal to e-folding depth

- 220 We measured an e-folding depth at DML between 2 and 5 cm which is lower than that employed in the base case TRANSITS model simulation (10 cm). Furthermore, a range of e-folding depth values, between 3.7 and 20 cm, have been reported for Antarctica (Wolff et al., 2002;France et al., 2011). The positive bias of the TRANSITS base case simulation in archived δ^{15} N-NO₃⁻ at DML may be due to e-folding depth being smaller than at Dome C as indicated by direct observations. In order to test this assumption, the sensitivity of archived δ^{15} N-NO₃⁻ to the e-folding depth parameter needs to be quantified, which has not
- 225 been done before as far as we know. Zatko et al. (2016) modelled the e-folding depth over Antarctica and investigated the impact of snow-sourced NO₅ fluxes but not on δ^{15} N-NO₃^{*}. Fig. 7**a** shows that the e-folding depth has a large influence on the δ^{15} N-NO₃^{*} depth profile in terms of i) depth of the photic zone and thus depth of the δ^{15} N-NO₃^{*} enrichment, and ii) the mean archived δ^{15} N-NO₃^{*} value below the photic zone. A larger e-folding depth increases the δ^{15} N-NO₃^{*} enrichment in the photic zone and increases the archived mean δ^{15} N-NO₃^{*} value. For example, an e-folding depth of 10 cm at DML gives δ^{15} N-NO₃^{*}
- 230 enrichment down to 30 cm and an archived mean δ¹⁵N-NO₃⁻ value of 125 ‰ in the snow pack compared to an e-folding depth of 20 cm, which enriches the snow pack down to 45 cm and more than doubles the archived mean δ¹⁵N-NO₃⁻ value to 320 ‰. Meanwhile, an e-folding depth of 2 cm gives minimal enrichment and a low archived mean δ¹⁵N-NO₃⁻ value of 25 ‰. In comparison to the base case simulation, which has an e-folding depth of 10 cm, a lower e-folding depth of 5 cm decreases the archived mean δ¹⁵N-NO₃⁻ in the snow pack to ~50 ‰, closely matching our snow pit observations. Hence, a shallower e-

38

Formatted: Not Superscript/ Subscript

Formatted: Not Highlight

235 folding depth of 5 cm can explain the more depleted δ^{15} N-NO₃⁺ snow pit profile, relative to the base case simulation, as NO₃⁺ photolysis occurs in a shallower depth. Therefore, e-folding depth knowledge is required to understand the sensitivity of archived δ^{15} N-NO₃⁻ at specific sites. We continue our sensitivity analysis using an e-folding depth of 5 cm and observed accumulation rate and refer to this scenario as our "5 cm EFD case".

4.5.1-2 Sensitivity of the ice core δ^{15} N-NO₃⁻ signal to accumulation rate

- 240 The δ^{15} N-NO₃⁻ signal is <u>also indeed</u> sensitive to the snow accumulation rate at DML. Here, the accumulation rate varied between 2.5 and 11 cm yr+ (w.e.) over the last 1000 years (Sommer et al., 2000). Figs. <u>7b-cc-</u> shows the potential impact of their variability in the snow accumulation rate on the NO₃⁻ mass concentration and δ^{15} N-NO₃⁻ signature at DML calculated with the TRANSITS model using an e-folding depth of 5 cm. Even in the 5 cm EFD case, there is still an offset with the snow pit δ^{15} N-NO₃⁻ depth profile below the active photic zone. Considering that the actual snow accumulation rate varied between
- 245 3.5 and 7.1 cm yr⁻⁺(w.e.) in our snow pit, our δ¹⁵N-NO₂⁻ measurements fall within the simulated δ¹⁵N-NO₂⁻ depth profile for the accumulation rates over the past 1000 years. Below the active photic zone, there is an offset between the base case and snow pit ô¹⁵N NO2⁻ depth profile in terms of i) the amplitude of the summer and winter ô¹⁵N NO2 values, and ii) the mean $\frac{\delta^{15}N-NO_{3}}{\delta^{15}N-NO_{3}}$ -value (Fig. 7). To account for the is offset, we investigated how the timing of snow deposition altered the $\delta^{15}N$ -NO3⁻ depth profile. Rather than assuming a constant accumulation rate of 6 cm yr¹ (w.e.), as in the 5 cm EFD casebase case,
- 250 we find that a variable snow accumulation rate, based on our observations from the snow pit, alters the depth of the summer and winter δ^{15} N-NO₃⁻ peaks (Fig. 7e7b₇). Using the actual annual snow accumulation rates improves the model fit in the top 30 cm; Fig 7c). Furthermore, the timing of the snow accumulation throughout the year has a significant control on the amplitude of the seasonal 815N-NO3 cycle. Snowfall at DML has a bimodal distribution with higher accumulation in austral autumn and early austral summer (Fig. S89). In Fig. 7d, we modified the timing of the snow accumulation during the year by depositing
- 255 90 % of the annual snowfall in i) the first week of winter, and ii) the first week of summer, which represents the upper bound for snow accumulation in winter and summer respectively. The remaining 10 % of the annual snowfall is distributed evenly across the rest of the weeks of the year. Summer snow accumulation results in a higher δ^{15} N-NO₃⁻ enrichment compared to winter snow accumulation, as the exposure of summer layers to UV is longer and thus NO3 photolysis is stronger. Therefore, the timing and rate of snowfall can explain the misalignment between snow pit observations and 5 cm EFD case base case simulation, which shifts the depth and amplitude of the δ^{15} N-NO₃⁻ peaks in the depth profile.

260

265

Although the mean snow pit 845N-NO3- is ~50 ‰ lower, the snow pit depth profile parallels the base case profile for the top 30 cm. Here, there is a clear enrichment of δ^{15} N-NO₃⁻ in both the snow pit and base case profiles corresponding to the depth of the photic zone (30 cm), and demonstrating that NO₃⁻ photolysis is taking place in this section of the snow pack. Below the photic zone, the seasonal variability of the base case δ¹⁵N-NO3⁻ depth profile is constant between 100-153 ‰ indicating that no further enrichment or NO3⁻ redistribution is taking place in the archived section of the snow pack.

39

Formatted: Not Highlight Formatted: Not Highlight

Formatted: Not Highlight Formatted: Not Highlight

Formatted: Not Highlight Formatted: Not Highlight

Formatted: Not Highlight

Despite the relatively high NO₃⁻ mass concentrations and enriched δ¹⁵N-NO₃⁻ in the skin layer at DML, clear seasonal cycles remain in the depth profile in contrast to the lower snow accumulation site of Dome C where the depth profile is relatively constant below the photic zone. Figs. 7a-b indicate that at higher snow accumulation rates, the seasonality of atmospheric NO₃⁻ and δ¹⁵N-NO₃⁻ is preserved due to faster burial. Even at 6 cm yr⁻¹ (w.e.), the snow layers remain in the active photic zone for 0.75 years and the weaker recycling factor is low enough to conserve the seasonality. Whereas at Dome C, snow layers remain within the photic zone for longer (about 3 years), and NO₃⁻ loss and redistribution continues until the seasonal cycle becomes smoothed (Figs. 7a-b). Thus, NO₃⁻ recycling is strongest in the lowermost snow accumulation regions.

Below the active photic zone, there is an offset between the base case and snow pit 8¹⁵N-NO₂⁻ depth profile in terms of i) the amplitude of the summer and winter 815N-NO2 values, and ii) the mean 815N-NO2 value (Fig. 7). To account for this offset we 275 investigated how the timing of snow deposition altered the 815N NO2⁻ depth profile. Rather than assuming a constant cumulation rate of 6 cm yr⁺(w.c.), as in the base case, we find that a variable snow accumulation rate, based on our observations from the snow pit, alters the depth of the summer and winter of 15 N-NO2-peaks (Fig. 7b.). Using the actual annual accumulation rates improves the model fit (-10 em depth; Fig 7a). Furthermore, the timing of the snow accumulation 280 throughout the year has a significant control on the amplitude of the seasonal 8⁴⁵N-NO₂⁻eyele. Snowfall at DML has a bimodal distribution with higher accumulation in austral autumn and early austral summer (Fig. S9). In Fig. 7c. we modified the timing of the snow accumulation during the year by depositing 90 % of the annual snowfall in i) the first week of winter, and ii) the first week of summer, which represents the upper bound for snow accumulation in winter and summer respectively. The remaining 10 % of the annual snowfall is distributed evenly across the rest of the weeks of the year. Summer snow 285 accumulation results in a higher 815N-NO2- enrichment compared to winter snow accumulation, as the exposure of summer layers to UV is longer and thus NO2 photolysis is stronger. Therefore, the timing and rate of snowfall can explain the misalignment between snow pit observations and base case simulation, which shifts the depth and amplitude of the δ^{45} N-NO $_2$ -

peaks in the depth profile.

On centennial to millennial timescales, the snow accumulation rate has varied in regions of Antarctica (e.g. Thomas et al., 2017), which could potentially modify the degree of post-depositional processing and thus impact the archival and temporal variability of δ¹⁵N-NO₃⁻ in ice cores. For example, the snow accumulation rate varied between 2.5 and 11 cm yr⁻¹ (w.e.) over the last 1000 years at DML (Sommer et al., 2000). Interestingly, Geng et al. (2015) found that post-depositional loss of NO₃⁻ in Greenland could fully account for the large difference between the glacial and Holocene δ¹⁵N-NO₃⁻ signature. At DML, higher snow accumulation rates would result in lower-lower NO₃⁻ mass concentrations and more depleted δ¹⁵N-NO₃⁻ values in

1295 the skin layer, thus reducing the recycling strength and lowering the sensitivity of the UV proxy recorded in the ice over time, and vice versa. TRANSITS modelling predicts that the upper and lower bounds of δ^{15} N-NO₃⁻ values in a 1000-year ice core from DML that has an accumulation rate between 2.5 and 11 cm yr⁻¹ (w.e.) and e-folding depth of 5 cm to be <u>between 70,30</u> and -360,140 ‰. Furthermore, δ^{15} N-NO₃⁻ values could range between 9040 and -110,50 ‰ depending on the timing of snowfall and extreme precipitation events, which are known to play a dominant role in snowfall variability across Antarctica (Turner et

1300 al., 2019). At DML, snow pit observations suggest that the variation of δ^{15} N-NO₃⁻ between the polar day and polar night is 20

Formatted: Not Highlight	
Formatted: Not Highlight	

%. This seasonality is within the range of less than δ^{15} N-NO₃⁻ values expected for changes in snow accumulation rates over time (Fig. 7). Therefore, any seasonal-variation in snow accumulationice core δ^{15} N-NO₃⁻ will need to be accounted for in order to observe decadal, centennial and millennial scale trends in δ^{15} N-NO₃⁻.

4.5.2 Sensitivity of the ice core 815N-NO3- signal to e-folding depth

- 305 We measured an e-folding depth at DML (2-5 cm) which is lower than that employed in the TRANSITS model (10 cm). Furthermore, a range of e-folding depth values, between 3.7 and 20 cm, have been reported for Antarctica<u>This estimate is</u> <u>similar to a modelled value at South Pole (3.7 cm;</u> (Wolff et al., 2002;France et al., 2011) which has a similar accumulation <u>rate, and Alert, Canada (5-6 cm; King and Simpson, 2001)</u>. The positive bias of the TRANSITS simulation in archived δ¹⁴N-NO₄⁻ at DML may be due to e-folding depth being smaller than at Dome C as indicated by direct observations. In order to test
- 310 this assumption, the sensitivity of archived δ¹⁵N-NO₃⁻ to the parameter e-folding depth needs to be quantified, which has not been done before as far as we know. Zatko et al. (2016) modelled the e-folding depth over Antarctica and investigated the impact of snow-sourced NO_x fluxes but not on δ¹⁵N-NO₃⁻. The e-folding depth has a large influence on the δ¹⁵N-NO₃⁻ depth profile in terms of i) depth of the photic zone and thus depth of the δ¹⁵N-NO₃⁻ enrichment, and ii) the mean archived δ¹⁵N-NO₃⁻ value below the photic zone (Fig. 7d). A larger e-folding depth strengthens the δ¹⁵N-NO₃⁻ enrichment in the photic zone and
- 315 archived mean δ¹⁵N-NO₃⁻value. For example, an e-folding depth of 10 cm at DML gives δ¹⁵N-NO₃⁻ enrichment down to 25 cm and an archived mean δ¹⁵N-NO₃⁻ value of 125 ‰ in the snow pack compared to an e-folding depth of 20 cm, which enriches the snow pack to 45 cm and more than doubles the archived mean δ¹⁵N-NO₃⁻ value to 320 ‰. Meanwhile, an e-folding depth of 2 cm gives minimal enrichment and a low archived mean δ¹⁵N-NO₃⁻ value of 25 ‰. In comparison to the base case simulation, which has an e-folding depth of 10 cm, a lower e-folding depth of 5 cm decreases the archived mean δ¹⁵N-NO₃⁻ value of 25 ‰.
- 320 the snow pack to ~50 ‰, closely matching our snow pit observations. Hence, a shallower e-folding depth of 5 cm can explain the more depleted δ¹⁵N-NO₃⁻ snow pit profile, relative to the base case simulation, as NO₃⁻ photolysis occurs in a shallower depth. Therefore, e-folding depth knowledge is required to understand the sensitivity of archived δ¹⁵N-NO₃⁻ at specific sites. A lower e-folding depth and variable snowfall throughout the year can explain the misalignment between the snow pit observations and simulated δ¹⁵N-NO₃⁻ depth profiles.

1325 4.5.3 Sensitivity of ice core δ^{15} N-NO₃⁻ signal to TCO

Fig. 8 shows the strong sensitivity of δ¹⁵N-NO₃⁻ to variations in decreasing. TCO. For each week, a constant amount of ozone (e.g. 100 DU) was added or subtracted from these present day values. A decrease in TCO will increase UV radiation reaching the surface at an ice core site. As a result, stronger photolysis enhances NO₃⁻ loss, redistribution and recycling from the snow pack and ultimately decreases the archived NO₃⁻ mass_concentration. Furthermore, a decrease in TCO enriches the δ¹⁵N-NO₃⁻
330 signature as the snow is exposed to a greater UV dose. We expect predict that a change of 100 Dobson Units (DU), i.e. the amount that ozone now decreases each spring as a result of stratospheric ozone destruction processes, will result in a 22,10 ‰ change in δ¹⁵N-NO₃⁻ at DML. (Erbland et al., 2015) The variability in δ¹⁵N-NO₃⁻ induced by TCO is similar-less thanto the

Formatted: Not Highlight

seasonal variability of δ¹⁵N-NO₃⁻ recorded in the snow pit (20 ‰)₂ and less than the predicted variability of δ¹⁵N-NO₃⁻ due to variability changes in snow accumulation (340-110 ‰) or e-folding depth (100 ‰)₅. As the above sensitivities have been evaluated individually, TCO depletion over many years may still be recoverable from ice core δ¹⁵N-NO₃⁻ if the other factors

- are constrained. For example, the e-folding depth at the DML site appears stable over the 8 year snow pit: the modelled δ^{15} N-NO₃⁻ sensitivity of 100 ‰ represents an upper limit for changes in the e-folding depth ranging between 2 and 10 cm and if the e-folding depth had changed recently, in an irregular manor, a regular annual cycle in δ^{15} N-NO₃⁻ wouldn't be evident (Fig. 4). Although additional studies of e-folding depth are required to confirm the variability of e-folding depth. thus the development
- 340 of a large ozone hole is unlikely to be observed above the natural background δ¹⁵N-NO₃⁻ variability in the ice core at this site. The sensitivity of δ¹⁵N-NO₃⁻ to TCO is greater at Dome C than DML (Fig. 8) due to the longer duration of surface snow exposure to UV radiation, stronger recycling and greater enrichment of δ¹⁵N-NO₃⁻ in the photic zone. The sensitivity of δ¹⁵N-NO₃⁻ recycling at DML is lower than expected from TRANSITS modelling for the same snow accumulation rate by Erbland et al. (2015), namely due to a lower e-folding depth than modelled, and thus is the sensitivity of δ¹⁵N-NO₃⁻ as a UV proxy is also lower than expected (Fig. 8).

4.5.4 Implications for interpreting ice core $\delta^{15}N\text{-}NO_3\text{-}$

Site-specific air-snow transfer studies provide an understanding of the mechanisms that archive δ^{15} N-NO₃⁻ in ice cores, thus allowing for the interpretation of longer records of δ^{15} N-NO₃⁻ from the site. Ice core records of archived NO₃⁻ mass concentrations and δ^{15} N-NO₃⁻ at DML are a result of <u>three-two</u> uptake and loss cycles that occur in the top <u>30-15</u> cm during sunlit conditions. While we do not observe further redistribution of NO₃⁻ in layers deeper than the photic zone, we cannot rule out any further NO₃⁻ diffusion within the firm or ice sections of an ice core. This redistribution unlikely results in a loss of NO₃⁻ but could migrate NO₃⁻ to different layers, for example in acidic layers around volcanic horizons (Wolff, 1995).

- There are a number of factors that will control the variability of the archived δ^{15} N-NO₃⁻ signature in ice cores recovered from 1355 DML. The δ^{15} N-NO₃⁻ signature in the snow pack is most sensitive to changes in the snow accumulation rate and e-folding depth, with snowfall timing and TCO-also playing a smaller role. The e-folding depth could change over time due to higher or lower dust or black carbon concentrations or a change in the snow grain size in a particular snow layer. The snow accumulation rate and e-folding depth could influence the archived δ^{15} N-NO₃⁻ composition by up to <u>110 and 3100</u> % respectively, over the last 1000-years. This magnitude is comparable to modelled enrichment in ice_core δ^{15} N-NO₃⁻ (0 to 363 ‰) due to photolysis-
- driven loss of NO₃⁻ at low accumulation sites in Antarctica by Zatko et al. (2016). While the timing of snowfall and changes in TCO will have a smaller impact of <u>120</u> ‰ on archived δ¹⁵N-NO₃⁻. Ice core δ¹⁵N-NO₃⁻ records at DML will be less sensitive to changes in UV than those at Dome C (Fig. 8), however the higher snow accumulation rate and more accurate dating at DML allows for higher resolution ice core δ¹⁵N-NO₃⁻ records. We acknowledge that in addition, other factors such as light absorbing impurities (Zatko et al., 2013), local meteorology, source of emissions and transport of NO_x and NO₃⁻, atmospheric oxidant
- 1365 concentrations, and polar NO₃⁻ formation can influence the rate of recycling and export of snow sourced NO_x. We discussed

42

Formatted: Not Highlight Formatted: Not Highlight

Formatted: Not Superscript/ Subscript

Formatted: Not Highlight

Formatted: Not Highlight

above that atmospheric δ^{15} N-NO₃⁻ values are unlikely to be influenced or sourced from snow exported up wind from the polar plateau due to the local meteorology at DML at least for the duration of the campaign. Yet these factors may have changed over time. Given a variable snow accumulation rate and smaller shallower e-folding depth, which we provide evidence for at DML. t. the 1370 TRANSITS model is able to reproduce our snow pit observations, justifying our previous assumption that photolysis is the

main driver of NO3⁻ post-depositional processes at DML. In fact, TRANSITS does such a good job at simulating NO3⁻ recycling in Antarctica that we recommend that this tool is employed before the commencement of future ice core δ^{15} N-NO₃⁻ studies to understand the sensitivity of the signal to various factors. Taking changes in snow accumulation into account, it may be possible to reconstruct past UV and TCO on longer timescales from the δ^{15} N-NO₃ - signal in DML ice cores provided other factors such

1375

5 Conclusions

Nitrogen Our key findings are:

as the e-folding depth have remained the same.

	stable nitrate isotopes of NO3, are a powerful tool for unpicking disentangling post-depositional processes affecting
	ice core signals of NO3, at low accumulation sites in Antarctica.
138	0At DML, post-depositional loss of NO3 ⁻ is controlled predominantly by <u>NO3⁻</u> photoly <u>sistic loss</u> ;
	<u>P</u> Photolysis redistributes NO ₃ ⁻ between the snow pack and atmosphere resulting in an enrichment of δ^{15} N-NO ₃ ⁻ in the
	skin layer 🙀
	archived in the snow pack below 15 cm and within 0.75 years
1385	5Once archived, the seasonal variability of δ^{15} N-NO ₃ ⁻ values and NO ₃ ⁻ mass concentrations oscillate between -1 to 80
	‰ and 30 to 80 ng g ⁻¹ , respectively <u>the e-folding depth at DML ranges between 2 - 5 cm</u> , which is lower than previous
	observations at Dome C (10 and 20 cm). As constraints on e-folding depth are critical for calculating photolytic loss of snow
	pack NO ₃ ⁺ and for interpreting δ^{15} N-NO ₃ ⁺ preserved in ice cores, additional studies of e-folding depth across a range of
	Antarctic sites would help determine key factors influencing this parameter.
139	0TRANSITS, a photolysis driven model, can explain the observed snow depth profiles of δ^{15} N-NO ₃ ⁻ at DML
	constrained by an e-folding depth of 5 cm, the observed snow accumulation rate, and variable snowfall timing.
	TRANSITS sensitivity analysis showed that the δ^{15} N-NO ₃ ⁻ signature in the snow pack is most sensitive to changes in the e-
	folding depth (100 ‰ for an 8 cm change in e-folding depth) and the snow accumulation rate (up to 300,100 ‰ for an 8.5 cm
1	

yr¹ (w.e.) change in annual snow accumulation rate%) and e-folding depth (up to 300 %), with snowfall timing (1-20 ‰ for 395 smaller role. The NO3 recycling process at DML is weaker than Dome C, largely because of the higher snow accumulation rate and lower e-folding depth. :

Formatted: Font:

Formatted: Not Superscript/ Subscript Formatted: Normal, No bullets or numbering Formatted: Not Highlight Formatted: Not Superscript/ Subscript Formatted: Not Highlight Formatted: Not Superscript/ Subscript Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight

-{	Formatted: Not Highlight
-{	Formatted: Superscript
-{	Formatted: Not Highlight
-{	Formatted: Not Highlight
-	Formatted: Not Highlight



reconstruct past UV-radiation at ice core sites with very a low accumulation rate and low accumulation variability, as low	
accumulation variability will have little effect on δ^{15} N-NO ₃ ⁻ in comparison to the UV dose reaching the ground	
;	
- Constraints on e-folding depth are critical for calculating photolytic loss of snow pack NO3 ⁻ and for interpreting 8 ¹⁵ N-	
NO ₃ - preserved in ice cores;	
-Additional studies of e-folding depth across a range of Antarctic sites would help determine key factors influencing	
this parameter	-
- The NO2 ⁻ recycling process at DML is weaker than Dome C, largely because of the higher snow accumulation rate and	
	reconstruct past UV-radiation at ice core sites with very a low accumulation rate and low accumulation variability, as low accumulation variability will have little effect on δ ¹⁵ N-NO ₃ ⁻ in comparison to the UV dose reaching the ground. _A Constraints on e-folding depth are critical for calculating photolytic loss of snow pack NO ₃ ⁻ and for interpreting δ ¹⁵ N-NO ₃ ⁻ preserved in ice cores; Additional studies of e-folding depth across a range of Antarctic sites would help determine key factors influencing this parameter The NO ₃ ⁻ recycling process at DML- is weaker than Dome C, largely because of the higher snow accumulation rate and

410 TRANSITS has now been tested at two sites in Antarctica, namely DML and Dome C, and we recommend applying this model to new ice core sites to understand the sensitivity of the 8¹⁴N-NO₂⁻ signal before embarking on new ice core projects; By accounting for variability in the snow accumulation rate and assuming a constant e-folding depth, it may be possible to reconstruct past UV-radiation at ice core sites with very a low accumulation rate and low accumulation variability, as low accumulation variability will have little effect on 8¹⁴N-NO₂⁻ in comparison to the UV dose reaching ground.

1415 Acknowledgments

-folding denth:

This project was funded by a National Environment Research Council (NERC) Standard Grant (NE/N011813/1) to M.F. V.H.L.W would like to thank the University of Cambridge Doctoral Training Program (DTP) for funding a NERC Research Experience Project (REP) that contributed to this manuscriptresearch. We would like to thank British Antarctic Survey (BAS) and Alfred Wegener Institute (AWI) staff for their field and logistics support at Halley Station and Kohnen Station, 1420 respectively. Technical support for nitrate isotope analysis at the Institut des Géosciences de l'Environnement (IGE), Grenoble was provided by Joris Leglise, Ines Ollivier and Ilan Bourgeois. We thank Joseph Erbland for providing the TRANSITS model. Field samples collected at Dome C was possible through the program SUNITEDC/CAPOXI (grant 1011/1177) funded by the Institut Polaire Français IPEV. J.S and N.C thank the ANR (Investissements d'avenir ANR-15-IDEX-02 and EAIIST grant ANR-16-CE01-0011-01) and the INSU program LEFE-CHAT for supporting the stable isotope laboratory. This is 1425 publication 1 of PANDA platform on which isotope analysis were performed. PANDA was partially funded by the LabEx OSUG@2020 (ANR10 LABX56). All winter over personal who collected the year-round Dome C samples in extreme conditions, years after years, are deeply acknowledged. In addition, we thank Emily Ludlow, Shaun Miller, Catriona Sinclair, Rebecca Tuckwell, and Neil Brough for technical support at BAS. Thanks to James France for discussions around the e-folding depth measurements and interpretation, and to John Turner for discussions of the local meteorology. We acknowledge Utrecht

44

Formatted: Not Highlight Formatted: Not Highlight

Formatted: Highlight

AWS DML05/Kohnen 1430 University who supplied the data for AWS9 at (https://www.projects.science.uu.nl/iceclimate/aws/files_oper/oper_20632), and the precipitation data from the RACMO2 model (https://doi.org/10/c2pv). We would like to thank Bodeker Scientific, funded by the New Zealand Deep South National Science Challenge, for providing the combined NIWA-BS total column ozone database. Wind roses were plotted using the openair package in R. The data set for the DML nitrate isotopic ratios and nitrate mass concentrations in aerosol, skin layer and snow pits is available through the Polar Data Centre https://doi.org/10.5285/1467b446-54eb-45c1-8a31-f4af21e60e60, 1435 and supporting data are also included as figures and tables in the supplement.

Author contributions

V.H.L.W, J.S and M.F designed the research. V.H.L.W, M.F and J.S, N.C collected samples at DML and Dome C, respectively. V.H.L.W analysed the major ion data. V.H.L.W, L.H, and N.C analysed the nitrate isotope data. A.M and V.H.L.W designed

the TRANSITS experiments. A.M performed the TRANSITS experiments. M.F did e-folding depth and <u>snow emission</u> flux calculations. V.H.L.W prepared the manuscript with contributions from all co-authors.

References

, !!! INVALID CITATION !!! (Berhanu et al., 2014;Frey et al., 2009;Erbland et al., 2013).

- , !!! INVALID CITATION !!! (Erbland et al., 2013;Shi et al., 2019). Abbatt, J. P.: Interaction of HNO3 with water-ice surfaces at temperatures of the free troposphere, Geophysical Research Letters, 24, 1479-
- 1482, 1997.
 Abbatt, J. P.: Interactions of atmospheric trace gases with ice surfaces: Adsorption and reaction, Chemical reviews, 103, 4783-4800, 2003.
 Berhanu, T. A., Meusinger, C., Erbland, J., Jost, R., Bhattacharya, S., Johnson, M. S., and Savarino, J.: Laboratory study of nitrate photolysis
- in Antarctic snow. II. Isotopic effects and wavelength dependence, The Journal of chemical physics, 140, 244306, 2014.
 Beyersdorf, A. J., Blake, D. R., Swanson, A., Meinardi, S., Rowland, F., and Davis, D.: Abundances and variability of tropospheric volatile
- organic compounds at the South Pole and other Antarctic locations, Atmospheric Environment, 44, 4565-4574, 2010. Blunier, T., Floch, G. L., Jacobi, H. W., and Quansah, E.: Isotopic view on nitrate loss in Antarctic surface snow, Geophysical Research Letters, 32, 2005.
- Bock, J., Savarino, J., and Picard, G.: Air-snow exchange of nitrate: a modelling approach to investigate physicochemical processes in surface snow at Dome C, Antarctica, Atmospheric Chemistry and Physics, 16, 12531-12550, 2016.
 Brucker, L., Picard, G., and Fily, M.: Snow grain-size profiles deduced from microwave snow emissivities in Antarctica, Journal of Glaciology, 56, 514-526, 2010.

Chan, H., King, M., and Frey, M.: The impact of parameterising light penetration into snow on the photochemical production of NOx and OH radicals in snow, Atmos. Chem. Phys, 15, 7913-7927, 2015.

1460 Chan, H. G., Frey, M. M., and King, M. D.: Modelling the physical multiphase interactions of HNO 3 between snow and air on the Antarctic Plateau (Dome C) and coast (Halley), Atmospheric Chemistry and Physics, 18, 1507-1534, 2018.

Chance, K., and Kurucz, R. L.: An improved high-resolution solar reference spectrum for earth's atmosphere measurements in the ultraviolet, visible, and near infrared, Journal of quantitative spectroscopy and radiative transfer, 111, 1289-1295, 2010.
 Chu, L., and Anastasio, C.: Quantum yields of hydroxyl radical and nitrogen dioxide from the photolysis of nitrate on ice, The Journal of Physical Chemistry A, 107, 9594-9602, 2003.

- Davis, D., Chen, G., Buhr, M., Crawford, J., Lenschow, D., Lefer, B., Shetter, R., Eisele, F., Mauldin, L., and Hogan, A.: South Pole NOx chemistry: an assessment of factors controlling variability and absolute levels, Atmospheric Environment, 38, 5375-5388, 2004a. Davis, D., Eisele, F., Chen, G., Crawford, J., Huey, G., Tanner, D., Slusher, D., Mauldin, L., Oncley, S., and Lenschow, D.: An overview of ISCAT 2000. Atmospheric Environment, 38, 53573-2004b.
- ISCAT 2000, Atmospheric Environment, 38, 5363-5373, 2004b.
 Davis, D. D., Seelig, J., Huey, G., Crawford, J., Chen, G., Wang, Y., Buhr, M., Helmig, D., Neff, W., and Blake, D.: A reassessment of Antarctic plateau reactive nitrogen based on ANTCI 2003 airborne and ground based measurements, Atmospheric Environment, 42, 2831-2848, 2008.
- Delmonte, B., Winton, H., Baroni, M., Baccolo, G., Hansson, M., Andersson, P., Baroni, C., Salvatore, M. C., Lanci, L., and Maggi, V.: Holocene dust in East Antarctica: Provenance and variability in time and space, The Holocene, 0959683619875188, 2019.
 Dibb, J. E., and Whitlow, S. I.: Recent climate anomalies and their impact on snow chemistry at South Pole, 1987-1994, Geophysical

Research Letters, 23, 1115-1118, 1996. Dubowski, Y., Colussi, A., and Hoffmann, M.: Nitrogen dioxide release in the 302 nm band photolysis of spray-frozen aqueous nitrate solutions. Atmospheric implications, The Journal of Physical Chemistry A, 105, 4928-4932, 2001.

- Duce, R. A., Liss, P. S., Merrill, J. T., Atlas, E. L., Buat-Menard, P., Hicks, B. B., Miller, J. M., Prospero, J. M., Arimoto, R., Church, T.
 M., Ellis, W., Galloway, J. N., Hansen, L., Jickells, T. D., Knap, A. H., Reinhardt, K. H., Schneider, B., Soudine, A., Tokos, J. J., Tsunogai, S., Wollast, R., and Zhou, M.: The atmospheric input of trace species to the world ocean, Global Biogeochem. Cycles, 5, 193-259, 10.1029/91gb01778, 1991.
- Erbland, J.: Isotopic constraints on the interpretation of the nitrate record in the Vostok ice core, University of Grenoble, University of Grenoble, 299 pp., 2011.
- 1485 Erbland, J., Vicars, W., Savarino, J., Morin, S., Frey, M., Frosini, D., Vince, E., and Martins, J.: Air-snow transfer of nitrate on the East Antarctic Plateau–Part 1: Isotopic evidence for a photolytically driven dynamic equilibrium in summer, Atmospheric Chemistry and Physics, 13, 6403-6419, 2013.
- Erbland, J., Savarino, J., Morin, S., France, J., Frey, M., and King, M.: Air-snow transfer of nitrate on the East Antarctic Plateau-Part 2: An isotopic model for the interpretation of deep ice-core records, Atmospheric Chemistry and Physics, 15, 12079-12113, 2015.
 France, J., King, M., Frey, M., Erbland, J., Picard, G., Preunkert, S., MacArthur, A., and Savarino, J.: Snow optical properties at Dome C
- 1490 France, J., King, M., Frey, M., Erbland, J., Picard, G., Preunkert, S., MacArthur, A., and Savarino, J.: Snow optical properties at Dome C (Concordia), Antarctica; implications for snow emissions and snow chemistry of reactive nitrogen, Atmospheric Chemistry and Physics, 11, 9787-9801, 2011.

France, J., and King, M.: The effect of measurement geometry on recording solar radiation attenuation in snowpack (e-folding depth) using fibre-optic probes, Journal of Glaciology, 58, 417-418, 2012.

- 1495 France, J., Reay, H., King, M., Voisin, D., Jacobi, H., Domine, F., Beine, H., Anastasio, C., MacArthur, A., and Lee-Taylor, J.: Hydroxyl radical and NOx production rates, black carbon concentrations and light-absorbing impurities in snow from field measurements of light penetration and nadir reflectivity of onshore and offshore coastal Alaskan snow, Journal of Geophysical Research: Atmospheres, 117, 2012. Frey, M., Roscoe, H., Kukui, A., Savarino, J., France, J., King, M., Legrand, M., and Preunkert, S.: Atmospherein nitrogen oxides (NO and NO 2) at Dome C, East Antarctica, during the OPALE campaign, Atmospheric Chemistry and Physics, 15, 7859-7875, 2015.
- 1500 Frey, M. M., Savarino, J., Morin, S., Erbland, J., and Martins, J.: Photolysis imprint in the nitrate stable isotope signal in snow and atmosphere of East Antarctica and implications for reactive nitrogen cycling, Atmospheric Chemistry and Physics, 9, 8681-8696, 2009. Frey, M. M., Brough, N., France, J. L., Anderson, P. S., Traulle, O., King, M. D., Jones, A. E., Wolff, E. W., and Savarino, J.: The diurnal variability of atmospheric nitrogen oxides (NO and NO<sub>2</ub>) above the Antarctic Plateau driven by atmospheric stability and snow emissions, Atmos. Chem. Phys., 13, 3045-3062, 10.5194/acp-13-3045-2013, 2013.
- 1505 Freyer, H.: Seasonal variation of 15N/14N ratios in atmospheric nitrate species, Tellus B, 43, 30-44, 1991. Gay, M., Fily, M., Genthon, C., Frezzotti, M., Oerter, H., and Winther, J.-G.: Snow grain-size measurements in Antarctica, Journal of Glaciology, 48, 527-535, 2002.
- Geng, L., Alexander, B., Cole-Dai, J., Steig, E. J., Savarino, J., Sofen, E. D., and Schauer, A. J.: Nitrogen isotopes in ice core nitrate linked to anthropogenic atmospheric acidity change, Proceedings of the National Academy of Sciences, 111, 5808-5812, 2014.
 Geng, L., Zatko, M. C., Alexander, B., Fudge, T., Schauer, A. J., Murray, L. T., and Mickley, L. J.: Effects of postdepositional processing
- 1510 Geng, L., Zatko, M. C., Alexander, B., Fudge, I., Schauer, A. J., Murray, L. I., and Mickley, L. J.: Effects of postdepositional processing on nitrogen isotopes of nitrate in the Greenland Ice Sheet Project 2 ice core, Geophysical Research Letters, 42, 5346-5354, 2015. Geng, L., Murray, L. T., Mickley, L. J., Lin, P., Fu, Q., Schauer, A. J., and Alexander, B.: Isotopic evidence of multiple controls on atmospheric oxidants over climate transitions, Nature, 546, 133, 2017.

Göktas, F., Fischer, H., Oerter, H., Weller, R., Sommer, S., and Miller, H.: A glacio-chemical characterization of the new EPICA deep-1515 drilling site on Amundsenisen, Dronning Maud Land, Antarctica, Annals of Glaciology, 35, 347-354, 2002.

- 1515 drilling site on Amundsenisen, Dronning Maud Land, Antarctica, Annals of Glaciology, 35, 347-354, 2002. Hastings, M., Jarvis, J., and Steig, E.: Anthropogenic impacts on nitrogen isotopes of ice-core nitrate, Science, 324, 1288-1288, 2009. Hastings, M. G., Steig, E., and Sigman, D.: Seasonal variations in N and O isotopes of nitrate in snow at Summit, Greenland: Implications for the study of nitrate in snow and ice cores, Journal of Geophysical Research: Atmospheres, 109, 2004. Hastings, M. G., Casciotti, K. L., and Elliott, E. M.: Stable isotopes as tracers of anthropogenic nitrogen sources, deposition, and impacts,
- Hasturgs, M. G., Casciotti, K. L., and Elliott, E. M.: Stable isotopes as tracers of anthropogenic nitrogen sources, deposition, and impacts,
 Elements, 9, 339-344, 2013.
 Hauglustaine, D., Granier, C., Brasseur, G., and Megie, G.: The importance of atmospheric chemistry in the calculation of radiative forcing
- Haugustaine, D., Graner, C., Brasseur, G., and Megre, G.: The importance of atmospheric chemistry in the calculation of radiative forcing on the climate system, Journal of Geophysical Research: Atmospheres, 99, 1173-1186, 1994.
 Helmig, D., Liptzin, D., Hueber, J., and Savarino, J.: Impact of exhaust emissions on chemical snowpack composition at Concordia Station, Antarctica, The Cryosphere, 14, 199-209, 2020.
- 1525 Hoering, T.: The isotopic composition of the ammonia and the nitrate ion in rain, Geochimica et Cosmochimica Acta, 12, 97-102, 1957. Hofstede, C. M., van de Wal Roderik, S., Kaspers, K. A., Van Den Broeke, M. R., Karlöf, L., Winther, J.-G., Isaksson, E., Lappegard, G., Mulvaney, R., and Oerter, H.: Firn accumulation records for the past 1000 years on the basis of dielectric profiling of six cores from Dronning Maud Land, Antarctica, Journal of Glaciology, 50, 279-291, 2004.
- Hofzumahaus, A., Lefer, B., Monks, P., Hall, S., Kylling, A., Mayer, B., Shetter, R., Junkermann, W., Bais, A., and Calvert, J.: Photolysis
 frequency of O3 to O (1D): Measurements and modeling during the International Photolysis Frequency Measurement and Modeling Intercomparison (IPMMI), Journal of Geophysical Research: Atmospheres, 109, 2004.
 Hursth, P., Beitrer, M.C., Gras, S., Dick, J.E., Schurger, P., and Granchall, D. Bridmer, et Alloren, et al. (2014).

Honrath, R., Peterson, M. C., Guo, S., Dibb, J. E., Shepson, P., and Campbell, B.: Evidence of NOx production within or upon ice particles in the Greenland snowpack, Geophysical Research Letters, 26, 695-698, 1999.

- Huey, L. G., Tanner, D., Slusher, D., Dibb, J. E., Arimoto, R., Chen, G., Davis, D., Buhr, M., Nowak, J., and Mauldin Iii, R.: CIMS
 measurements of HNO3 and SO2 at the South Pole during ISCAT 2000, Atmospheric environment, 38, 5411-5421, 2004.
 Huthwelker, T., Ammann, M., and Peter, T.: The uptake of acidic gases on ice, Chemical reviews, 106, 1375-1444, 2006.
- Jacobi, H.-W., Weller, R., Jones, A., Anderson, P., and Schrems, O.: Peroxyacetyl nitrate (PAN) concentrations in the Antarctic troposphere measured during the photochemical experiment at Neumayer (PEAN'99), Atmospheric Environment, 34, 5235-5247, 2000.
- Jones, A., Weller, R., Minikin, A., Wolff, E., Sturges, W., McIntyre, H., Leonard, S., Schrems, O., and Bauguitte, S.: Oxidized nitrogen chemistry and speciation in the Antarctic troposphere, Journal of Geophysical Research: Atmospheres, 104, 21355-21366, 1999. Jones, A., Weller, R., Wolff, E., and Jacobi, H. W.: Speciation and rate of photochemical NO and NO2 production in Antarctic snow, Geophysical Research Letters, 27, 345-348, 2000.
- Jones, A., Weller, R., Anderson, P., Jacobi, H. W., Wolff, E., Schrems, O., and Miller, H.: Measurements of NOx emissions from the Antarctic snowpack, Geophysical Research Letters, 28, 1499-1502, 2001.
- 1545 Jones, A., Wolff, E., Ames, D., Bauguitte, S.-B., Clemitshaw, K., Fleming, Z., Mills, G., Saiz-Lopez, A., Salmon, R., and Sturges, W.: The multi-seasonal NO y budget in coastal Antarctica and its link with surface snow and ice core nitrate: results from the CHABLIS campaign, Atmospheric Chemistry and Physics Discussions, 7, 4127-4163, 2007.
 - Jones, A., Brough, N., Anderson, P., and Wolff, E. W.: HO 2 NO 2 and HNO 3 in the coastal Antarctic winter night: a" lab-in-the-field" experiment, Atmospheric Chemistry and Physics, 14, 11843-11851, 2014.

- 1550 Jones, A. E., Wolff, E. W., Ames, D., Bauguitte, S.-B., Clemitshaw, K., Fleming, Z., Mills, G., Saiz-Lopez, A., Salmon, R. A., and Sturges, W.: The multi-seasonal NO y budget in coastal Antarctica and its link with surface snow and ice core nitrate: results from the CHABLIS campaign, Atmospheric Chemistry and Physics, 11, 9271-9285, 2011.
- Keene, W. C., Pszenny, A. A., Galloway, J. N., and Hawley, M. E.: Sea-salt corrections and interpretation of constituent ratios in marine precipitation, Journal of Geophysical Research: Atmospheres, 91, 6647-6658, 1986.
- 1555 Kendall, C., Elliott, E. M., and Wankel, S. D.: Tracing anthropogenic inputs of nitrogen to ecosystems, Stable isotopes in ecology and environmental science, 2, 375-449, 2007.
- King, M. D., and Simpson, W. R.: Extinction of UV radiation in Arctic snow at Alert, Canada (82 N), Journal of Geophysical Research: Atmospheres, 106, 12499-12507, 2001.
- Kukui, A., Loisil, R., Kerbrat, M., Frey, M., Gil Roca, J., Jourdain, B., Ancellet, G., Bekki, S., Legrand, M., and Preunkert, S.: OH and RO2
 radicals at Dome C (East Antarctica): first observations and assessment of photochemical budget, EGU General Assembly Conference Abstracts, 2013.
- Le Meur, E., Magand, O., Arnaud, L., Fily, M., Frezzotti, M., Cavitte, M., Mulvaney, R., and Urbini, S.: Spatial and temporal distributions of surface mass balance between Concordia and Vostok stations, Antarctica, from combined radar and ice core data: first results and detailed error analysis, The Cryosphere, 2018.
- 1565 Lee-Taylor, J., and Madronich, S.: Calculation of actinic fluxes with a coupled atmosphere-snow radiative transfer model, Journal of Geophysical Research: Atmospheres, 107, 2002. Lee H.-M. Horze, D.K. Alexander, B. and Murray, L. T.: Investigating the sensitivity of surface-level nitrate seasonality in Antarctica to
- Lee, H.-M., Henze, D. K., Alexander, B., and Murray, L. T.: Investigating the sensitivity of surface-level nitrate seasonality in Antarctica to primary sources using a global model, Atmospheric Environment, 89, 757-767, 2014. Legrand, M. R., and Delmas, R. J.: Soluble impurities in four Antarctic ice cores over the last 30 000 years, Annals of Glaciology, 10, 116-
- 1570 120, 1988. Li, D., and Wang, X.: Nitrogen isotopic signature of soil-released nitric oxide (NO) after fertilizer application, Atmospheric Environment, 42, 4747-4754, <u>https://doi.org/10.1016/j.atmosenv.2008.01.042</u>, 2008. Libois, Q., Picard, G., France, J., Arnaud, L., Dumont, M., Carmagnola, C., and King, M.: Influence of grain shape on light penetration in
- Soors, Q., Picard, G., France, J., Arnaud, L., Dumont, M., Carmagnola, C., and King, M.: Influence of grain snape on light penetration in snow, The Cryosphere, 7, 1803-1818, 2013.
 Mariotti, A.: Atmospheric nitrogen is a reliable standard for natural 15N abundance measurements, Nature, 303, 685, 1983.
- Martota, A. Admosphere integration is a reliable standard for hardinal of natural individual enclassification, (adde, 505, 605, 1765). McCabe, J. R., Thiemens, M. H., and Savarino, J.: A record of zone variability in South Pole Antarctic snow: Role of nitrate oxygen isotopes, Journal of Geophysical Research: Atmospheres, 112, 2007. Meusinger, C., Berhanu, T. A., Erbland, J., Savarino, J., and Johnson, M. S.: Laboratory study of nitrate photolysis in Antarctic snow. I.
- Observed quantum yield, domain of photolysis, and secondary chemistry, The Journal of chemical physics, 140, 244305, 2014. 1580 Miller, D. J., Wojtal, P. K., Clark, S. C., and Hastings, M. G.: Vehicle NOx emission plume isotopic signatures: Spatial variability across
- 1900 hind, D. S., Woha, F. K., Chak, S. C., and Hashings, M. G.: Ventos emission prane cooper signatures. Spatial variability across the eastern United States, Journal of Geophysical Research: Atmospheres, 122, 4698-4717, 2017. Miller, D. J., Chai, J., Guo, F., Dell, C. J., Karsten, H., and Hastings, M. G.: Isotopic Composition of In Situ Soil NOX Emissions in Manure-
- Fertilized Cropland, Geophysical Research Letters, 45, 12,058-012,066, 2018. Morin, S., Savarino, J., Frey, M. M., Yan, N., Bekki, S., Bottenheim, J. W., and Martins, J. M.: Tracing the origin and fate of NOx in the
- Morin, S., Savarino, J., Frey, M. M., Yan, N., Becki, S., Bottenneim, J. W., and Martins, J. M.: Tracing the origin and rate of NOX in the
 Arctic atmosphere using stable isotopes in nitrate, Science, 322, 730-732, 2008.
 Morin, S., Savarino, J., Frey, M. M., Domine, F., Jacobi, H. W., Kaleschke, L., and Martins, J. M.: Comprehensive isotopic composition of
- Aronn, G. Gavanno, J., Tey, M. M., Dohme, F., Jacobi, H. W., Kateshira, E., and Wattins, J. M.: Completions to isotopic composition of atmospheric nitrate in the Atlantic Ocean boundary layer from 65 S to 79 N, Journal of Geophysical Research: Atmospheres, 114, 2009. Mosley-Thompson, E., Paskievitch, J. F., Gow, A. J., and Thompson, L. G.: Late 20th century increase in South Pole snow accumulation, Journal of Geophysical Research: Atmospheres, 104, 3877-3886, 1999.
- 1590 Mulvaney, R., and Wolff, E. W.: Evidence for winter/spring denitrification of the stratosphere in the nitrate record of Antarctic firm cores, Journal of Geophysical Research: Atmospheres, 98, 5213-5220, 1993.
- Mulvaney, R., Wagenbach, D., and Wolff, E.: Postdepositional change in snowpack nitrate from observation of year-round near-surface snow in coastal Antarctica, Journal of Geophysical Research: Atmospheres, 103, 11021-11031, 1998.
- Noro, K., Hattori, S., Uemura, R., Fukui, K., Hirabayashi, M., Kawamura, K., Motoyama, H., Takenaka, N., and Yoshida, N.: Spatial variation of isotopic compositions of snowpack nitrate related to post-depositional processes in eastern Dronning Maud Land, East Antarctica, GEOCHEMICAL JOURNAL, 52, e7-e14, 10.2343/geochemj.20519, 2018.
 Oerter, H., Wilhelms, F., Jung-Rothenhäusler, F., Göktas, F., Miller, H., Graf, W., and Sommer, S.: Accumulation rates in Dronning Maud
- Land, Antarctica, as revealed by dielectric-profiling measurements of shallow firm cores, Annals of Glaciology, 30, 27-34, 2000.
 Oncley, S., Buhr, M., Lenschow, D., Davis, D., and Semmer, S.: Observations of summertime NO fluxes and boundary-layer height at the
- 1600 South Pole during ISCAT 2000 using scalar similarity, Atmospheric Environment, 38, 5389-5398, 2004. Pasteris, D., McConnell, J. R., Edwards, R., Isaksson, E., and Albert, M. R.: Acidity decline in Antarctic ice cores during the Little Ice Age linked to changes in atmospheric nitrate and sea salt concentrations, Journal of Geophysical Research: Atmospheres, 119, 5640-5652, 2014. Reijmer, C., and Oerlemans, J.: Temporal and spatial variability of the surface energy balance in Dronning Maud Land, East Antarctica, Journal of Geophysical Research: Atmospheres, 107, ACL 9-1-ACL 9-12, 2002.

- Röthlisberger, R., Hutterli, M. A., Sommer, S., Wolff, E. W., and Mulvaney, R.: Factors controlling nitrate in ice cores: Evidence from the Dome C deep ice core, Journal of Geophysical Research: Atmospheres, 105, 20565-20572, 2000. 1605 Röthlisberger, R., Hutterli, M. A., Wolff, E. W., Mulvaney, R., Fischer, H., Bigler, M., Goto-Azuma, K., Hansson, M. E., Ruth, U., and Siggaard-Andersen, M.-L.: Nitrate in Greenland and Antarctic ice cores: a detailed description of post-depositional processes, Annals of Glaciology, 35, 209-216, 2002.
- Savarino, J., Kaiser, J., Morin, S., Sigman, D., and Thiemens, M.: Nitrogen and oxygen isotopic constraints on the origin of atmospheric nitrate in coastal Antarctica, Atmospheric Chemistry and Physics, 7, 1925-1945, 2007. 1610 Schlosser, E., Manning, K., Powers, J., Duda, M., Birnbaum, G., and Fujita, K.: Characteristics of high-precipitation events in Dronning
- Maud Land, Antarctica, Journal of Geophysical Research: Atmospheres, 115, 2010.
- Seinfeld, J. H., and Pandis, S. N.: Form air pollution to climate change, Atmospheric Chemistry and Physics, 1326, 1998. Shi, G., Buffen, A., Hastings, M., Li, C., Ma, H., Li, Y., Sun, B., An, C., and Jiang, S.: Investigation of post-depositional processing of 1615 nitrate in East Antarctic snow: isotopic constraints on photolytic loss, re-oxidation, and source inputs, Atmospheric Chemistry and Physics, 15, 9435-9453, 2015.
- Shi, G., Buffen, A., Ma, H., Hu, Z., Sun, B., Li, C., Yu, J., Ma, T., An, C., and Jiang, S.: Distinguishing summertime atmospheric production 1620
- of nitrate across the East Antarctic Ice Sheet, Geochimica et Cosmochimica Acta, 231, 1-14, 2018. Shi, G., Chai, J., Zhu, Z., Hu, Z., Chen, Z., Yu, J., Ma, T., Ma, H., An, C., and Jiang, S.: Isotope fractionation of nitrate during volatilization in snow: a field investigation in Antarctica, Geophysical Research Letters, 46, 3287-3297, 2019. Silva, S., Kendall, C., Wilkison, D., Ziegler, A., Chang, C. C., and Avanzino, R.: A new method for collection of nitrate from fresh water and the analysis of nitrogen and oxygen isotope ratios, Journal of Hydrology, 228, 22-36, 2000.
- Sofen, E., Alexander, B., Steig, E., Thiemens, M., Kunasek, S., Amos, H., Schauer, A., Hastings, M., Bautista, J., and Jackson, T.: WAIS 1625 Divide ice core suggests sustained changes in the atmospheric formation pathways of sulfate and nitrate since the 19th century in the extratropical Southern Hemisphere, Atmospheric Chemistry and Physics, 14, 5749-5769, 2014.
- Sommer, S., Appenzeller, C., Röthlisberger, R., Hutterli, M. A., Stauffer, B., Wagenbach, D., Oerter, H., Wilhelms, F., Miller, H., and Mulvaney, R.: Glacio-chemical study spanning the past 2 kyr on three ice cores from Dronning Maud Land, Antarctica: 1. Annually resolved accumulation rates, Journal of Geophysical Research: Atmospheres, 105, 29411-29421, 10.1029/2000jd900449, 2000.
- Thibert, E., and Domine, F.: Thermodynamics and kinetics of the solid solution of HNO3 in ice, The Journal of Physical Chemistry B, 102, 1630 4432-4439, 1998. Thomas, E. R., van Wessem, J. M., Roberts, J., Isaksson, E., Schlosser, E., Fudge, T., Vallelonga, P., Medley, B., Bertler, N., and van de
- Broeke, M.: Review of regional Antarctic snow accumulation over the past 1000 years, Climate of the Past Discussion, 2017. Traversi, R., Usoskin, I., Solanki, S., Becagli, S., Frezzotti, M., Severi, M., Stenni, B., and Udisti, R.: Nitrate in polar ice: a new tracer of solar variability, Solar Physics, 280, 237-254, 2012.
- 1635 Turner, J., Phillips, T., Thamban, M., Rahaman, W., Marshall, G. J., Wille, J. D., Favier, V., Winton, H., Thomas, E., and Wang, Z.: The Dominant Role of Extreme Precipitation Events in Antarctic Snowfall Variability, Geophysical Research Letters, 2019. Van Meijgaard, E., Van Ulft, L., Van de Berg, W., Bosveld, F., Van den Hurk, B., Lenderink, G., and Siebesma, A.: The KNMI regional atmospheric climate model RACMO version 2.1, Koninklijk Nederlands Meteorologisch Instituut, 43, 2008.
- Wagenbach, D., Legrand, M., Fischer, H., Pichlmayer, F., and Wolff, E.: Atmospheric near-surface nitrate at coastal Antarctic sites, Journal 1640 of Geophysical Research: Atmospheres, 103, 11007-11020, 1998.
- Warren, S. G.: Optical properties of snow, Reviews of Geophysics, 20, 67-89, 1982. Weller, R., Minikin, A., König-Langlo, G., Schrems, O., Jones, A., Wolff, E., and Anderson, P.: Investigating possible causes of the observed diurnal variability in Antarctic NOy, Geophysical research letters, 26, 2853-2856, 1999.
- Weller, R., Traufetter, F., Fischer, H., Oerter, H., Piel, C., and Miller, H.: Postdepositional losses of methane sulfonate, nitrate, and chloride 1645 at the European Project for Ice Coring in Antarctica deep-drilling site in Dronning Maud Land, Antarctica, Journal of Geophysical Research: Atmospheres, 109, 2004.
- Weller, R., and Wagenbach, D.: Year-round chemical aerosol records in continental Antarctica obtained by automatic samplings, Tellus B: Chemical and Physical Meteorology, 59, 755-765, 2007.
- 1650 Weller, R., Legrand, M., and Preunkert, S.: Size distribution and ionic composition of marine summer aerosol at the continental Antarctic site Kohnen, Atmospheric Chemistry and Physics, 18, 2413-2430, 2018. Wilhelms, F., Miller, H., Gerasimoff, M. D., Drücker, C., Frenzel, A., Fritzsche, D., Grobe, H., Hansen, S. B., Hilmarsson, S. Æ., Hoffmann, G., Hörnby, K., Jaeschke, A., Jakobsdóttir, S. S., Juckschat, P., Karsten, A., Karsten, L., Kaufmann, P. R., Karlin, T., Kohlberg, E., Kleffel, G., Lambrecht, A., Lambrecht, A., Lawer, G., Schärmeli, I., Schmitt, J., Sheldon, S. G., Takata, M., Trenke, M., Twarloh, B., Valero-
- 1655 Delgado, F., and Wilhelms-Dick, D.: The EPICA Dronning Maud Land deep drilling operation, Annals of Glaciology, 55, 355-366, 10.3189/2014AoG68A189, 2017. Winton, V. H. L. W., Caillon, N., Hauge, L., Mulvaney, R., Rix, J., Tuckwell, R., Savarino, J., and Frey, M.: Ice core chemistry, density,
- conductivity, dust, snow accumulation rate, and stable nitrate isotopic composition of the 120 m ISOL-ICE ice core, Dronning Maud Land, Antarctica (Version 1.0) [Data set], in, UK Polar Data Centre, Natural Environment Research Council, UK Research & Innovation. 1660 10.5285/1467b446-54eb-45c1-8a31-f4af21e60e60, 2019a.

Winton, V. H. L. W., Frey, M., Hauge, L., Caillon, N., and Savarino, J.: Major ion chemistry and stable nitrate isotopic composition of aerosol, skin layer snow and snow pits at Dronning Maud Land, Antarctica (Version 1.0) [Data set], in, UK Polar Data Centre, Natural Environment Research Council, UK Research & Innovation. 10,5285/1467b446-54eb-45c1-8a31-f4af21e60e60, 2019b.

Wolff, E., Jones, A. E., Bauguitte, S.-B., and Salmon, R. A.: The interpretation of spikes and trends in concentration of nitrate in polar ice cores, based on evidence from snow and atmospheric measurements, Atmospheric Chemistry and Physics, 8, 5627-5634, 2008.

Wolff, E. W.: Nitrate in polar ice, in: Ice core studies of global biogeochemical cycles, Springer, 195-224, 1995.
 Wolff, E. W., Jones, A. E., Martin, T. J., and Grenfell, T. C.: Modelling photochemical NOx production and nitrate loss in the upper snowpack of Antarctica, Geophysical Research Letters, 29, 5-1-5-4, 2002.

Yu, Z., and Elliott, E. M.: Novel method for nitrogen isotopic analysis of soil-emitted nitric oxide, Environmental science & technology, 51, 6268-6278, 2017.

Zatko, M., Geng, L., Alexander, B., Sofen, E., and Klein, K.: The impact of snow nitrate photolysis on boundary layer chemistry and the recycling and redistribution of reactive nitrogen across Antarctica and Greenland in a global chemical transport model, Atmos. Chem. Phys, 16, 2819-2842, 2016.

Zatko, M. C., Grenfell, T. C., Alexander, B., Doherty, S. J., Thomas, J. L., and Yang, X.: The influence of snow grain size and impurities
 n the vertical profiles of actinic flux and associated NO x emissions on the Antarctic and Greenland ice sheets, Atmospheric Chemistry and Physics, 13, 3547-3567, 2013.



Figure 1: <u>Year round Year-round</u> atmospheric and skin layer NO₃⁻ mass concentration and δ¹⁵N-NO₃⁻ at Dome C. <u>Two high-volume</u> aerosol samplers were used at Dome C (HiVol 2 and HiVol 3) over the campaign and showed good reproducibility. Data source:
 years 2009-2010: Erbland et al. (2013); 2011-2015: this study.





Figure 2: Map of ISOL-ICE ice core drilling and atmospheric campaign, and ice core sites and Antarctica stations mentioned in this study. a) Insert of Kohnen Station in Dronning Maud Land (DML) highlighting the predominate wind direction, deep EDML ice core site and the ISOL-ICE <u>"ice core" (b) and "flux"-and (c) "ice cores"</u> sites; b) ISOL-ICE "ice core site" showing ice core, firn core and snow pit A locations, and co ISOL-ICE "flux" site showing location of *in situ* atmospheric instruments, surface snow, snow pit and aerosol sampling locations and e-folding depth measurements.







Figure 3: January 2017 time series inat Dronning Maud Land (DML) of a) <u>daily</u> precipitation, b) <u>hourly</u> wind direction and wind speed, c) atmospheric and skin layer 8¹⁵N-NO₃⁻, and d) atmospheric and skin layer NO₃⁻ mass concentration. <u>Error bars in panels</u> c-d indicate the spatial variability of the site determined by multiple skin layer samples collected on 28/01/2017. The spatial variability exceeds the instrumental error which is smaller than the symbol size. Meteorological data source: University of Utrecht (AWS9; DML05/Kohnen; 75°00'S, 00°00' E/W; ~2900 m.a.s.l.). Precipitation data source: RACMO2 (https://doi.org/10/c2pv).






Figure 4: Comparison of NO₃ mass concentration and δ^{15} N-NO₃ at Dronning Maud Land (DML) and Dome C in January 2017. NO₃ mass concentration in a) atmosphere, b) skin layer, and c) depth profiles. Insert: Depth profile of NO₃ mass concentration highlighting seasonal variability. δ^{15} N-NO₃ in d) atmosphere, e) skin layer, and f) depth profiles. Grey bars indicate summer seasons for DML depth profiles.

Formatted: Not Superscript/ Subscript



Figure 5: Schematic of NO₃⁻ mass fluxes at Dronning Maud Land (DML) for a) annual mean scenario and b) January scenario.



Figure 6: ISOL-ICE observations and simulated annual cycle of skin layer and atmospheric NO₃⁻ mass_concentration and δ¹⁵N-NO₃⁻
 at Dronning Maud Land (DML) from the base case and 5 cm EFD case the TRANSITS model simulations for January 2017. a) Total column ozone: NIWA Bodeker combined dataset version 3.3 at DML averaged from 2000 to 2016 (http://www.bodekerscientific.com/data/total-column-ozone. b) Atmospheric NO₃⁻ mass_concentration-datas are observations from Kohnen Station (Weller and Wagenbach, 2007) that are used as input into the model. ISOL-ICE observations and TRANSIASTS simulations of c) atmospheric δ¹⁵N-NO₃⁻, d) skin layer NO₃⁻ mass_concentration and e) skin layer δ¹⁵N-NO₃⁻.







Figure 7: Snow pit depth profiles of observations and simulations from TRANSITS. <u>Sensitivity of a Sensitivity of b¹⁵N-NO₃: to the e-folding depth, b) Sensitivity of b¹⁵N-NO₃: and <u>cb</u>) sensitivity of NO₃: <u>mass</u> concentration to the upper and lower bounds of accumulation rates observed over the last thousand years at Dronning Maud Land (DML). Also shown are our snow pit observations, and the depth profiles of the simulated b¹⁵N-NO₃: values and NO₃: <u>mass</u> concentration using the observed accumulation rate in our snow pits, i.e., variable accumulation rate with depth (orange line). The observed snow accumulation rate from the snow pits varied between 3.5 and 7.1 cm vr⁻¹ (w.e.). Sensitivity of <u>d</u>(e) Sensitivity of b¹⁵N-NO₃: to the timing of snow accumulation, <u>d</u>) b⁴⁵N-NO₃: to the</u>

Formatted: Not Highlight	
Formatted: Not Highlight	
Formatted: Not Highlight	
Formatted: Not Highlight	

e-folding depth. In each panel, bBlue is the base case simulation, and green is the 5 cm EFD case simulation, which in which we refer to throughout the study. Note that panels a-b) have the same legend, and the nominal date refers to the base case simulation. Formatted: Not Highlight Formatted: Not Highlight Formatted: Not Highlight





righte 8: Expected response of archived 5"N-NO3 to changes in tota column ozone at Droming Waud Land (DWL) and Done C. Calculated sensitives represent an upper range as the real ozone hole lasts September to November before recovery, and not as modelled using the entire sunlit season. Archived DML δ¹⁵N-NO3 values were simulated using the observeda-fixed accumulation rate of 6 cm yr⁺(w.c.)_x and e-folding depth of 10-5 cm (5 cm (EFD case), and present day TCO values. These TCO values, that were used in all our calculations, vary weekly and can be found in Table S3. For each week, a constant amount of ozone (e.g. 100 DU) was added or subtracted from these present day values. Dome C data source: Erbland et al. (2015).

Formatted: Not Highlight

Table 1: Site characteristics of Dronning Maud Land (DML) and Dome C ice core sites.

	DML	Dome C	
Latitude (°S)	75	75	
Elevation (m a.s.l.)	2892	3233	
Distance from the coast (km)	550	900	
Mean snow accumulation (cm y ⁻¹ ; w.e.)	¹ 6	² 2.5	
Predominate wind direction (°)	45	180-200	
Mean summer temperature (°C)	<u>3-28</u>	4-30	
Annual mean temperature (° C)	<u>3-41</u>	<u>4-52</u>	
Maximum summer temperature (°C)	<u>3-9</u>	<u>4-17</u>	
Minimum winter temperature (°C)	<u>3-74</u>	<u>4-80</u>	
e-folding depth (cm)	4 3 2-5	⁴ 10 <u>610</u> -20	
Average January nitrate mass concentration in skin layer (ng g-1)	4 3 230	³ 600 ⁵ 600	
Average annual nitrate mass concentration in firn (ng g ⁻¹)	4 3 60	*50 750	
Average January nitrate mass concentration in atmosphere (ng m ⁻³)	⁴³ 10	³ 60 ⁵ 60	
¹ Sommer et al. (2000);Hofstede et al. (2004)			
² Le Meur et al. (2018)			
³ University of Utrecht (AWS9; DML05/Kohnen)			
⁴ Erbland et al. (2013)			
⁵ This study			

⁴France et al. (2011)⁶France et al. (2011)

1735 ⁵Frey et al. (2009)⁷Frey et al. (2009)

Table 2: Summary of observed and simulated archived, aerosol and skin layer NO₃⁻ mass concentrations, and δ^{15} N-NO₃⁻ composition and NO₃⁻ mass fluxes at Dronning Maud Land (DML) and Dome C. n.d.: no data. Base case refers to the TRANSITS simulation with a snow accumulation rate of 6 cm yr_x⁻¹ (w.e.) and an e-folding depth of 10 cm, while the 5 cm EFD case refers to a TRANSITS simulation with an observed snow accumulation rate that varied year to year between 6.0 and 7.1 cm yr⁻¹ (w.e.) and an e-folding depth of 5 cm.

740

Archived (2~30 cm)	NO ₃ ⁻ (ng g ⁻¹)	δ ¹⁵ N-NO ₃ ⁻ (‰)	Flux (pg m ⁻² s ⁻¹)	Reference]	
DML Pit A	60	50	110	This study		
DML Pit B	50	n.d.	120	This study		
DML TRANSITS (base case)	120	130	210	This study		Formatted: Not Highlight
DML TRANSITS (5 cm EFD case)	<u>280</u>	<u>50</u>	<u>480</u>	This study		Formatted: Not Highlight
*DML expected	100	100	140	Erbland et al. (2015);Erbland etal. (2013)		Field Code Changed
Dome C	50	280	<140	Erbland et al. (2013)		Field Code Changed
Aerosol (January mean)	NO ₃ - (ng m ⁻²)	δ ¹⁵ N-NO ₃ - (‰)	Flux (pg m ⁻² s ⁻¹)	Reference		
DML	10	-30	70	This study		
DML TRANSITS (base case)	30	-20	190	This study; Weller and Wagenbach (2007)		Field Code Changed
DML TRANSITS (5 cm EFD case)	30	40	<u>50</u>	This study; Weller and Wagenbach (2007)		Formatted: Not Highlight Formatted: Not Highlight
Dome C	60	-10	90	This study; Erbland et al. (2013)	1	
Skin layer (January mean)	NO ₃ ⁻ (ng g ⁻¹)	δ ¹⁵ N-NO ₃ ⁻ (‰)	Flux (pg m ⁻² s ⁻¹)	Reference		
DML	230	-10	360	This study		
DML TRANSITS (base case)	2800	10	4800	This study		Formatted: Not Highlight
DML TRANSITS (5 cm EFD case)	<u> </u>	<u>-10</u>	<u>2900</u>	This study		Formatted: Not Highlight
Dome C	590	10	470	This study; Erbland et al. (2013)		

Formatted: Not Superscript/ Subscript

Formatted: Superscript

*Expected values for a site with an accumulation rate of 6 cm yr¹ (w.e.) based on the spatial transect of Erbland et al. (2015).