

1 The characteristics of atmospheric brown carbon in Xi'an,
2 inland China: sources, size distributions and optical properties

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40 **Abstract:** To investigate the characteristic of atmospheric brown carbon (BrC) in the
41 semi-arid region of East Asia, PM_{2.5} and size-resolved particles in the urban atmosphere of
42 Xi'an, inland China during the winter and summer of 2017 were collected and analyzed for
43 optical properties and chemical compositions. Methanol extracts (MeOH-extracts) were more
44 light-absorbing than water extracts (H₂O- extracts) in the optical wavelength of 300-600 nm,
45 and well correlated with nitrophenols, polycyclic aromatic hydrocarbons (PAHs) and
46 oxygenated PAHs ($r > 0.78$). The light absorptions ($abs_{\lambda=365nm}$) of H₂O- extracts and
47 MeOH-extracts in winter were 28 ± 16 M/m and 49 ± 32 M/m, respectively, which are about 10
48 times higher than those in summer, mainly due to the enhanced emissions from biomass
49 burning for house heating. Water extracted BrC predominately occurred in the fine mode ($<$
50 $2.1 \mu m$) during winter and summer, accounting for 81% and 65% of the total absorption of
51 BrC, respectively. The light absorption and stable carbon isotope composition measurements
52 showed an increasing ratio of $abs_{\lambda=365nm}$ -MeOH to $abs_{\lambda=550nm}$ -EC along with an enrichment of
53 ¹³C in PM_{2.5} during the haze development, indicating an accumulation of secondarily formed
54 BrC (e.g., nitrophenols) in aerosol aging process. PMF analysis showed that biomass burning,
55 fossil fuel combustion, secondary formation, and fugitive dust are the major sources of BrC in
56 the city, accounting for 55%, 19%, 16%, and 10% of the total BrC of PM_{2.5}, respectively.

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58 **Key words:** Brown Carbon; Haze; Stable carbon isotope composition; Biomass burning;
59 Secondary formation.

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62 **1. Introduction**

63 Brown carbon (BrC) is a vital fraction of carbonaceous aerosols, and exhibits strong
64 absorption **ability** from near ultraviolet (UV) to visible light region. Thus, it has been given
65 extensive investigation in the recent decades (Laskin et al., 2015; Yan et al., 2018; Gustafsson
66 et al., 2009). BrC has significant impact on climate change directly by absorbing solar
67 radiation and indirectly by accelerating snowmelt and affecting the albedo (Qian et al.,
68 2015; Andreae and Ramanathan, 2013). Based on the remote sensing observations and
69 chemical transport models (Chung et al., 2012; Wang et al., 2014; Jo et al., 2016), a
70 non-negligible positive radiative forcing by BrC was found on a global scale with a range
71 from 0.1 to 0.6 W m⁻². Beyond that, BrC also influence the atmospheric chemistry and human
72 health. For example, BrC can shield polycyclic aromatic hydrocarbons (PAHs) from being
73 oxidized, and thus substantially elevate lung cancer risk from PAHs (Hsu et al., 2014; Yan et
74 al., 2018).

75 The sources of BrC are complicated, which can be primarily emitted from incomplete
76 combustion of carbon-containing materials (e.g., biomass, coal and petroleum products.) and
77 secondarily derived from aqueous-phase reaction (Sun et al., 2017; Gilardoni et al., 2016; Xie
78 et al., 2018; Nakayama et al., 2013). Biomass burning was found to be a **major** source of BrC
79 (Chen and Bond, 2010; Chakrabarty et al., 2010; Saleh et al., 2014), because lignin is of an
80 unsaturated benzene-like structure, which is a chromophore group. Field measurements and
81 laboratory studies found that BrC is also secondary sources by forming chromophores during
82 the atmosphere ageing process, e.g., high-NO_x photooxidation (Liu et al., 2016; Xie et al.,
83 2017), ozonolysis of aromatic precursors (Lee et al., 2014), and aqueous-phase photochemical

84 oxidation and polymerization (Smith et al., 2014; Flores et al., 2014; Bones et al., 2010). BrC
85 products account for very small weight fraction of organic aerosol (OA), but have a
86 significant effect on OA optical properties. For example, nitroaromatic compounds generated
87 by photooxidation of toluene under high NO_x conditions may account for 40-60% of the total
88 light absorption of toluene-SOA (Lin et al., 2015).

89 Multiple approaches have been developed to quantify the light absorption properties of
90 BrC (Moosmuller et al., 2009), and a common and sensitive approach is the direct
91 measurement of spectrophotometric properties of aerosol water or filter extracts by using
92 optical instrumentation. The advantage of this method can avert interference from insoluble
93 absorption material (e.g., black carbon) (Cheng et al., 2016; Shen et al., 2017), and supply
94 high-resolution spectrum over a wide wavelength coverage. Furthermore, it is favorable for
95 characterization of BrC light-absorbing components by combining with other analytical
96 techniques, such as mass spectrometry (MS) (Laskin et al., 2015; Corr et al., 2012; Satish et al.,
97 2017).

98 Many studies have been conducted on the BrC optical properties in China, but most of
99 those were based on PM_{2.5} and PM₁₀ sample collection and focused on the bulk aerosol
100 optical properties with no information on the size distributions (Shen et al., 2017; Huang et al.,
101 2018). Xi'an is a metropolitan city located in Guanzhong Basin of inland China, which is a
102 typical semiarid region in East Asia and have been suffering from serious particle pollution
103 due to the large emission of anthropogenic pollutants (Wu et al., 2018; Wang et al., 2016; Wu
104 et al., 2019), especially intensive coal combustion and biomass burning in winter for house
105 heating (Wang et al., 2017). In this study, both PM_{2.5} and size-segregated aerosol samples in

106 Xi'an were collected during the 2017 winter and summer and analyzed for the characteristics
107 of BrC. We firstly investigated the seasonal variations of chemical composition and
108 light-absorption of BrC in the city, then discussed the size distribution of BrC and the impact
109 of aerosol ageing process on BrC, and finally quantified its source contributions.

110 **2. Experimental section**

111 **2.1 Sample collection**

112 **Aerosol samples were collected on a day/night basis with each for 12-hrs by using a**
113 **high-volume ($\sim 1.13 \text{ m}^3 \text{ min}^{-1}$) air sampler** (Tisch Environmental, Inc., OH, USA) from
114 December 31, 2016 to January 22, 2017 (in winter) and from July 18 to August 6, 2017 (in
115 summer). The sampler was installed on the roof of a three-story building on the campus of the
116 Institute of Earth Environment, CAS (34.22°N, 108.88°E), which is located at the urban center
117 of Xi'an, inland China. Meanwhile, size-resolved aerosols with 9 size bins (cutoff points were
118 0.43, 0.65, 1.1, 2.1, 3.3, 4.7, 5.8, and 9.0 μm , respectively) were collected by using an
119 Anderson sampler at an airflow rate of 28.3 L min^{-1} for 24 hr. All samples were collected onto
120 the pre-baked (450°C for 6 hr) quartz filters and stored in a freezer (-18°C) prior to analysis.

121 **2.2 Chemical analysis**

122 A punch (0.526 cm^3) from each $\text{PM}_{2.5}$ filter sample was analyzed for organic carbon (OC)
123 and elemental carbon (EC) with a DRI Model 2001 Thermal/Optical Carbon Analyzer
124 (Atmoslytic Inc., Calabasas, CA, USA) following the IMPROVE-A protocol (Chow et al.,
125 2007). More details of the method including quality assurance and quality control (QA/QC)
126 can be found elsewhere (Wang et al., 2010).

127 Partial filters were cut into pieces, and then extracted three times under sonication with

128 15ml Milli-Q pure water (18.2 MΩ). Ten ions such as SO_4^{2-} , NO_3^- , Cl^- , NH_4^+ , and K^+ were
129 determined using ion chromatography (Dionex, ICS-1100). Similar extraction processes were
130 also applied to measure the water-soluble organic carbons (WSOC) of the samples, which was
131 determined by using Shimadzu TOC-5000 Carbon Analyzer. The detailed method has been
132 reported by Wang et al. (2013). In order to analyze the organic compounds in the samples
133 such as levoglucosan, PAHs, OPAH and nitrophenols, aliquot of the filter was extracted with a
134 mixture of methanol and DCM (1:5, v/v), derivatized with BSTFA and measured by using gas
135 chromatography (HP 7890A, Agilent Co., USA) coupled with mass spectroscopy detector
136 (GC/MS) (HP 5975, Agilent Co., USA). Details of sample extraction and derivatization were
137 documented elsewhere (Wang et al., 2009b; Ren et al., 2017). Stable carbon isotope
138 composition of total carbon ($\delta^{13}\text{C}_{\text{TC}}$) was determined by using an elemental analyzer (EA)
139 (Carlo Erba, NA 1500) coupled with an isotope ratio mass spectrometer (IRMS, Finnigan
140 MAT Delta Plus), more details of the method can be referred to elsewhere (Cao et al., 2016).

141 2.3 Light absorption measurements

142 Brown carbon (BrC) was extracted from a size of 6 cm^3 filter samples for 30min
143 ultrasonication with 20 ml Milli-Q pure water or methanol. All extracts were then filtered
144 through $0.45\ \mu\text{m}$ PTFE (for water) and $0.22\ \mu\text{m}$ PES (for methanol) pore syringe filter to
145 remove insoluble components and filter debris. The light-absorption spectra were analyzed
146 with a UV–visible spectrophotometer (AOE INSTRUMENTS, China) over a wavelength
147 range of 190–900 nm (Hecobian et al., 2010). The absorption coefficient of water or methanol
148 extracts (M m^{-1}) could be calculated as the following equation (Teich et al., 2017):

$$\text{abs}_\lambda = (A_\lambda - A_{700}) \frac{V_1}{V_a \times L} \times \ln(10) \quad (1)$$

149 Where A_λ and A_{700} were the light absorption of the extracts at the wavelength of λ and
 150 700 nm, respectively. V_1 represented the volume of the solvent extracting the filter sample,
 151 and V_a was referred to the volume of air corresponding to the filter punch. L was the
 152 absorbing path length (i.e., 1 cm for the currently used quartz cuvettes). The $\ln(10)$ was
 153 converted from base 10 (the form provided by the spectrophotometer) to natural logarithms.
 154 According to the previous studies, the absorption coefficient at 365nm was used as the brown
 155 carbon absorption in order to avoid disturbance of inorganic salts such as nitrate.

156 The bulk mass absorption coefficient (MAC, m^2/g) of the extracts at a given wavelength
 157 can be described by the following equation:

$$\text{MAC} = \frac{\text{abs}_\lambda}{C_{W(M)SOC}} \quad (2)$$

158 Where $C_{W(M)SOC}$ was the atmospheric concentration of the particulate water-soluble
 159 (WSOC) or methanol-soluble organic carbon (MSOC, $\mu\text{gC}/\text{m}^3$). In this study, we assumed
 160 that OC could be completely dissolved in methanol solvent and substituted the MSOC for the
 161 calculation. This hypothesis would possibly lead to somewhat underestimation of the MAC of
 162 the methanol extracts, although high extraction efficiency of methanol solvent had been
 163 reported by previous studies (Liu et al., 2013) .

164 The wavelength dependence of light-absorption with respect to the empirically defined
 165 power law relationship is described by the following equation (Laskin et al., 2015):

$$\text{MAC} = K\lambda^{-AAE} \quad (3)$$

166 Where K is a factor that includes aerosol mass concentrations, the AAE is termed as
167 absorption Angström exponent. In this study, the AAE value of the filter extracts was
168 determined by a linear regression of $\log(\text{abs}_\lambda)$ versus $\log(\lambda)$ over a wavelength range of
169 300-450nm.

170 **2.4 Positive Matrix Factorization (PMF) source apportionment**

171 PMF, as a receptor model, decomposes the sample matrix into two matrices (factor
172 contributions and factor profiles), and has been widely used for the source apportionment of
173 atmospheric pollutants. More details on PMF can be found on the EPA website
174 ([https://www.epa.gov/air-research/epa-positive-matrix-factorization-50-fundamentals-and-use-
175 r-guide](https://www.epa.gov/air-research/epa-positive-matrix-factorization-50-fundamentals-and-use-r-guide)). In the present work, the mass concentrations of major species (OC, EC, WSOC,
176 SO_4^{2-} , NO_3^- , NH_4^+ , Ca^{2+}), organic markers (benzo(b)fluoranthene (BbF), benzo(e)pyrene
177 (BeP), indeno(1,2,3-c,d)pyrene (IP), levoglucosan, and nitrophenols), and abs_λ of water extracts
178 have been used as the input data to perform the source apportionment for brown carbon with
179 the EPA PMF 5.0 version, similar reports have been found elsewhere (Hecobian et al., 2010).
180 The model was run numerous times with 3–7 factors and various combinations of the
181 concentration and absorption data set. Base on the Q value (Q_{true} and Q_{robust}) and r , which are
182 indicative of the agreement of the model fit, four factors were obtained as the optimal
183 solution.

184 **3. Results and discussion**

185 **3.1 Carbonaceous species in $\text{PM}_{2.5}$ during summer and winter**

186 Figure 1 shows the temporal variations in the concentrations of $\text{PM}_{2.5}$, WSOC, OC and
187 $\text{abs}_{\lambda=365\text{nm}}$ value during the two seasons. WSOC varied from 5.3 to 67 $\mu\text{gC}/\text{m}^3$ in winter with

188 an average of $23 \pm 13 \mu\text{gC}/\text{m}^3$ (Table 1), which was 4.0 times higher than that in summer. OC
189 exhibited a similar seasonal variation with WSOC with an average of $41 \pm 25 \mu\text{gC}/\text{m}^3$ in
190 winter and $8.4 \pm 2.4 \mu\text{gC}/\text{m}^3$ in summer, respectively. Whereas, WSOC/OC ratio was much
191 higher in summer (0.70 ± 0.12) than that in winter (0.58 ± 0.13), partly as a result of an
192 enhanced photochemical formation of WSOC under the intense sunlight conditions. Similar
193 phenomena were also found in Beijing (Ping et al., 2017), Shanghai (Zhao et al., 2015a),
194 Tokyo (Miyazaki et al., 2006) and Southeastern United States (Ding et al., 2008).

195 PAHs, OPAHs, and nitrophenols are ubiquitous in the atmosphere, which can be directed
196 emitted from incomplete combustion of carbon -containing fuels (e.g., coal, biomass) (Shen et
197 al., 2013; Zhang and Tao, 2009). In addition, OPAHs and nitrophenols can also be produced
198 from photochemical reactions (Cochran et al., 2016; Keyte et al., 2013; Yuan et al., 2016).
199 These compounds are the efficient light-absorbing species, because their molecular structures
200 consist of chromophores (Lin et al., 2017; Bluvshstein et al., 2017). Herein, 14 PAHs, 7 OPAHs,
201 and 7 nitrophenols were examined for investigating their effect on BrC absorption. As seen in
202 Figure S1, the temporal variations of PAHs, OPAHs, and nitrophenols were similar with
203 levoglucosan, which is the tracer of biomass burning emissions, indicating that biomass
204 burning is one of the major sources of these compounds. Concentrations of PAHs, OPAHs,
205 and nitrophenols during winter were $149 \pm 89 \text{ ng}/\text{m}^3$, $174 \pm 98 \text{ ng}/\text{m}^3$ and $17 \pm 12 \text{ ng}/\text{m}^3$
206 (Table 1), respectively, and were 10 - 43 times higher than those in summer, which can be
207 explained by an increasing emission from residential heating during winter in the city and its
208 surrounding regions.

209 As shown in Table S1, $\text{abs}_{\lambda=365\text{nm}}$ extracted by methanol displays showed well correlations

210 with PAHs, OPAHs, and nitrophenols, especially in winter ($r > 0.89$), which suggests that
211 those species are important light absorption contributors for BrC in Xi'an. Huang et al. (2018)
212 found that PAHs and OPAHs in Xi'an accounted for, on average, 1.7% of the overall
213 absorption of methanol-soluble BrC, but their mass fraction in OC was only 0.35%. A recent
214 study reported that biomass burning also emitted nitroaromatic compounds, particularly
215 nitrophenols, and accounted for 50-80% of the total visible light absorption (> 400 nm) (Lin
216 et al., 2017). The robust correlations of above compounds with the absorption at $\lambda=365$ nm
217 suggest that PAHs, OPAHs and nitrophenol are strong light-absorbing species.

218 **3.2 Light absorption of BrC in water and methanol extracts**

219 **3.2.1 Seasonal variations of light absorption by BrC**

220 As shown in Figure 2a and 2b, the marked feature of BrC in Xi'an is that the absorption
221 spectrum increased notably from the visible to the ultraviolet ranges, and the average
222 abs-MeOH at $\lambda=365$ nm was 1.5 - 1.7 times higher than $\text{abs-H}_2\text{O}$ in the two seasons,
223 indicating that MSOC provided a more comprehensive estimation for BrC. Due to enhanced
224 emission of BrC, average $\text{abs}_{\lambda=365\text{nm}}$ of BrC found in winter was 49 ± 32 M/m for MeOH and
225 28 ± 16 M/m for WSOC, which were 9.5- and 8.1-fold higher than that in summer. This
226 phenomenon was also observed in previous studies in Xi'an (Shen et al., 2017;Huang et al.,
227 2018) and other areas of China (Du et al., 2014;Chen et al., 2018). Compared with other
228 regions (Table 2), the absolute $\text{abs}_{\lambda=365\text{nm}}$ values in Xi'an were slightly lower than that in
229 Indo-Gangetic Plain, India (Satish et al., 2017;Bachi, 2016), but were considerably higher
230 than those in Beijing, China (Du et al., 2014), US (Zhang et al., 2011) and Korea (Kim et al.,
231 2016), suggesting a heavy pollution of light-absorbing aerosols in Xi'an. Furthermore,

232 enhanced $\text{abs}_{\lambda=365\text{nm}}$ loading in the nighttime was observed during the two seasons, which can
233 be ascribed to the shallower boundary layer height and the absence of photo-bleaching
234 processes at night (Saleh et al., 2013;Zhao et al., 2015b).

235 Linear regression slopes on the scatter plots of $\text{abs}_{\lambda=365\text{nm}}$ values versus WSOC or MSOC
236 represented the average of MAC at 365 nm (i.e., MAC_{WSOC} and MAC_{MSOC}). During winter,
237 there was a slight disparity between the MAC_{WSOC} and MAC_{MSOC} with the averages of $1.2 \pm$
238 0.06 and $1.3 \pm 0.03 \text{ m}^2/\text{g}$ (Figure 2e), respectively, which indicates that there are some similar
239 chromophores of BrC between the two fractions. As seen in Table 2, both MAC_{WSOC} and
240 MAC_{MSOC} in Xi'an during the two seasons are higher than those in US and Korea, suggesting
241 that BrC in the city was comprised of stronger light-absorbing compounds. $\text{abs}_{\lambda=365\text{nm}}$ showed
242 a strong linear correlation with levoglucosan ($r > 0.98$), suggesting that abundant BrC was
243 largely derived from biomass burning. As shown in Fig. S2, mass ratios of
244 levoglucosan/mannosan and levoglucosan/galacosan in the $\text{PM}_{2.5}$ samples are similar to
245 biomass types (i.e., woods, leaves, wheat straw), again reflecting that biomass burning
246 combustion in Xi'an and its surrounding regions are probably the major sources of BrC in the
247 city during winter. Compared to winter, the MAC in summer was slightly lower, which can be
248 in part attributed to the less abundant light-absorbing PAHs and OPAHs due to no biomass
249 burning for house heating. Moreover, with increasing photooxidation in summer,
250 fragmentation reactions would occur and thus decrease light absorption for BrC aerosols, as
251 reported by Sumlin et al. (2017), because higher levels of O_3 and OH radicals in summer
252 intensify the photooxidation and diminish the BrC aerosol light absorption by reducing the
253 size of conjugated molecular systems. Interestingly, we found that the MAC_{WSOC} (1.1 ± 0.2

254 m^2/g) in summer was significantly enhanced compared to MAC_{MSOC} ($0.8 \pm 0.1 \text{ m}^2/\text{g}$), which
255 can be ascribed to more amount of non-BrC in the methanol extracts such as phthalates, of
256 which the abundance relative to OC was about 10 time higher in summer than in winter. The
257 $\text{abs}_{\lambda=365\text{nm}}$ showed a poor correlation with levoglucosan (Table S1), further indicating that the
258 biomass burning was not the dominant source for BrC in summer.

259 Absorption Ångström exponents (AAE), which were derived from the filter methanol-
260 and water-extracted BrC (AAE_{WSOC} and AAE_{MSOC}) for wavelengths between 300 and 450 nm,
261 were 6.1 ± 9.7 and 5.3 ± 8.5 (Table 2) in winter, respectively, and resembled that in Beijing
262 (Cheng et al., 2016), Guangzhou (Liu et al., 2018) and Indo- Gangetic Plain (Bachi, 2016),
263 possibly indicating that the chemical compositions of BrC chromophores in these regions are
264 similar during winter. As seen in Table 2, unlike those of H_2O -extracts, the averaged values of
265 MAC and AAE of MeOH extracts were 40% and 10% higher in winter than in summer,
266 respectively, suggesting that chemical compositions of BrC are different between the two
267 seasons in the city and the winter BrC contained more non-polar compounds that are of
268 stronger light-absorbing ability.

269 **3.2.2 Aerosol size distribution of BrC**

270 Particles with different sizes are of different chemical compositions, and thus optical
271 properties of BrC in different size of particles are also different (Zhang et al., 2015;Zhai et al.,
272 2017). However, information on size distribution of BrC absorption is very limited. In this
273 study, we mainly focused on the water-extracted samples, because particles deposited on the
274 filter surface are unevenly distributed, making the quantifications of OC and EC in the
275 size-segregated samples not accurate enough. As shown in Figure S3, there was a good

276 relationship between the $\text{abs}_{\lambda=365\text{nm}}$ ($r > 0.98$) of the samples collected by Anderson sampler
277 and those collected by high-volume $\text{PM}_{2.5}$ sampler (Fig. S3), suggesting a good agreement
278 between the two sampling methods.

279 As show in Figure 3, $\text{abs}_{\lambda=365\text{nm}}$ presented a bimodal pattern during winter and summer,
280 dominating at the fine mode ($D_p < 2.1\mu\text{m}$) with relative contributions of 81% and 65% to the
281 total absorption in the two seasons, respectively. These proportions are similar to those
282 reported for a forest wildfire event, which showed that 93% of the total BrC absorption was in
283 the fine particles ($0.10 < D_p < 1.0\mu\text{m}$) (Lorenzo et al., 2018). Maximum absorptions were
284 observed at 1.02 and $0.71\mu\text{m}$ (D_{pg} - geometric mean diameters, Figure 3a and 3b) in winter
285 and summer, respectively, which is in agreement with the observations by Lei et al (2018),
286 who found that the major peaks for BrC absorption were in the rang from $0.5\mu\text{m}$ to $1.0\mu\text{m}$ in
287 urban and may shift toward smaller size ($< 0.4\mu\text{m}$) for particles released from burning
288 experiments (Lei et al., 2018). However, the size distribution pattern of MAC was different
289 from that of $\text{abs}_{\lambda=365\text{nm}}$ in Xi'an, which presented a monomodal distribution with a peak in the
290 fine mode ($< 2.1\mu\text{m}$) in winter and a bimodal distribution in summer with two peaks in the fine
291 ($< 2.1\mu\text{m}$) and coarse ($> 2.1\mu\text{m}$) modes, respectively (Figure 3c and 3d). As seen in Figure 3c
292 and 3d, the fine mode of MAC was around 50% larger in winter than that in summer,
293 suggesting that water-soluble fraction of winter fine particles was more light-absorbing
294 compared to that in summer, probably due to the summertime stronger bleaching effect.

295 **3.3 Underestimation of BrC absorption by solvent extraction methods**

296 A few studies pointed out that absorption properties of BrC extracted by bulk solution
297 may not entirely reflect the light absorption by ambient aerosols. Here, we further calculated

298 the light absorption of the samples using the Mie theory combined with an imaginary (k ,
299 responsible for absorption) refractive index with assumptions that particles were of spherical
300 morphology and externally mixed with other light-absorbing components. The imaginary
301 refractive index could be obtained from MAC using follow equation (Laskin et al., 2015):

$$k_{(\lambda)} = \frac{\rho \lambda \text{ abs}}{4\pi \times \text{WSOC}} = \frac{\rho \lambda \text{ MAC}}{4\pi} \quad (4)$$

302 Where ρ (g/cm^3) was particle density and assigned as 1.5, more details about Mie
303 calculations can be referred to the study by Liu et al. (2013).

304 As noted above, most BrC aerosols were in the fine mode ($<2.1\mu\text{m}$), thus, here we only
305 focused on this fraction for the Mie calculations. The values of imaginary refractive in winter
306 remains nearly constant (0.038-0.048) for different particle sizes at $\lambda=365$ nm (Table 3),
307 which was about two times smaller than that (0.093 ± 0.049) over Gangetic Plain, India
308 (Shamjad et al., 2017). Values of k in summer were slight smaller when compared to those in
309 winter, suggesting that the aerosols in summer were more aged. Sumlin et al. (2017) found
310 that k decreases along with the atmospheric aging from 0.029 ± 0.001 to 0.019 ± 0.001 at
311 $\lambda=375$ nm. However, k values in this study were 5.0 times (avg.) higher than those reported
312 from the United States (Liu et al., 2013). This is because that $\text{PM}_{2.5}$ particles in Xi'an, China
313 are enriched in BrC and the mass absorption coefficient was considerably higher than that in
314 US. Figure 4 compares the difference between $\text{abs}_{\lambda=365\text{nm}}$ predicted by Mie theory (abs-Mie)
315 and that extracted by the bulk solution (abs-Measure). Mie theory predicted $\text{abs}_{\lambda=365\text{nm}}$ was
316 1.5-fold higher than that measured by the bulk solution, suggesting that the solvent extraction
317 methods, which have commonly been used for atmospheric BrC measurements, could result

318 in an underestimation on optical absorption of aerosols. Hence, a factor of 1.5 is
319 recommended to convert the liquid-based data (at least for the water-soluble data) reported by
320 this work for estimating optical properties of atmospheric aerosols in Xi'an and its
321 surrounding regions in order to better quantify the BrC light-absorption.

322 3.4 The characteristic of BrC with the aerosol aging

323 During the ageing process, secondary organic aerosols (SOA) with strong chromophores
324 can be generated and efficiently absorb solar radiation (Lin et al., 2014; Lin et al., 2016). From
325 Figure 5, it can be found that air quality in Xi'an during the winter varied from the clean
326 ($PM_{2.5} < 75 \mu g/m^3$) to the polluted conditions ($PM_{2.5} > 75 \mu g/m^3$) from the period of 12th
327 January to 19th January. Such a case provides an opportunity to investigate the changes in
328 light-absorption by BrC during the aerosol ageing process.

329 As shown in Figure 5a and 5b, $ab_{S_{\lambda=365nm}}$ extracted by water and MeOH in Xi'an during
330 the campaign showed an increasing trend from 12th January to 19th January, which is similar
331 to $PM_{2.5}$ loadings but opposite to the visibility, indicating that BrC is one of the important
332 factors leading to the visibility deterioration. From Figure 5b, it can also be seen that light
333 absorption of water-extracts dominated over the total BrC absorption especially in daytime
334 and showed a variation pattern similar to the $PM_{2.5}$ (Figure 5a) and WSOC loadings (Figure
335 5c), indicating a continuous formation of secondary BrC during the aerosol ageing process. To
336 illustrate this point, the stable carbon isotopic composition ($\delta^{13}C_{TC}$) of total carbon (TC) in the
337 samples was measured. WSOC/OC showed a positive correlation with the $\delta^{13}C_{TC}$,
338 demonstrating an ageing process of aerosols during the haze development from 12th to 19th,
339 January, although it was weak ($r = 0.47$, $n = 17$). Similar conclusions were also reported by

340 Yang et al. (2004) and Pavuluri et al. (2015). From Figure 5c, increasing trends of OPAHs and
341 nitrophenols were observed during the haze development, suggesting that more SOAs with
342 chromophores were generated during such an aerosol ageing process, because these
343 compounds are also of secondary origins. To exclude the possible impact of the changes in
344 BrC source emissions, the values of PAHs/OC and levoglucosan/OC were applied in this study,
345 because PAHs and levoglucosan emission factors are different for different
346 sources (Nguyen-Duy and Chang, 2017). As shown in Fig. S4, both values indistinctively
347 changed during the aerosol ageing process, indicating that the increasing $abs_{\lambda=365nm}$ were not
348 caused by the changes in source emissions. Moreover, we found that MAC_{MSOC} values during
349 the age process also increased (Figure 5a), further suggesting that the bleaching effect on
350 light-absorbing BrC was reducing during the haze developing process.

351 EC is one of the major light-absorbing aerosols in the atmosphere (Collier et al.,
352 2018; Peng et al., 2016). To further discuss the changes of BrC during the aerosol ageing
353 process, we compared the mass absorption efficiency of EC at $\lambda=550\text{ nm}$ ($7.5 \pm 1.2\text{ m}^2/\text{g}$)
354 with BrC by using the method reported from Yan et al. (2015) and Kirillova et al. (2014). As
355 shown in Figure 5c, the concentrations of EC have a slight change in the haze period, so the
356 changes in light absorption of EC remained nearly constant. However, the ratio of
357 $abs_{\lambda=365nm}\text{-MeOH}/abs_{\lambda=550nm}\text{-EC}$ increasingly became larger along with the visibility
358 deterioration from January 12th to January 19th (Fig. 5b), while the mass ratios of PAHs/EC,
359 OPAHs/EC and nitrophenols /EC during the period showed a significant negative correlation
360 with visibility (Fig. S5), further suggesting that the impairment of BrC on the visibility was
361 getting more significant during the haze development process.

362 During the haze developing process, organic aerosols are usually getting more aged and
363 enriched in heavier ^{13}C due to the kinetic isotopic effect (KIE) (Wang et al., 2010). As shown
364 in Figure 6a and b, $\delta^{13}\text{C}$ of $\text{PM}_{2.5}$ samples presented a strong positive correlation with $\text{abs}_{\lambda=365}$
365 nm-MeOH ($r=0.82$) in the daytime, while there was no such a correlation in the nighttime
366 during the haze period of January 12th -19th, indicating a daytime formation of secondary BrC.
367 From Figure 6c and 6d, we also found that the correlation of $\text{abs}_{\lambda=365 \text{ nm-MeOH}} / \text{abs}_{\lambda=550 \text{ nm-EC}}$
368 ratio with nitrophenol was much stronger in daytime than in nighttime, which is opposite to
369 the correlation of $\text{abs}_{\lambda=365 \text{ nm-MeOH}} / \text{abs}_{\lambda=550 \text{ nm-EC}}$ ratio with PAHs. Nitrophenols can be
370 produced from secondary photooxidation of phenol with NO_x , while PAHs are produced
371 solely from direct emissions especially from coal and biomass burning for house heating. The
372 opposite diurnal correlations of $\text{abs}_{\lambda=365 \text{ nm-MeOH}} / \text{abs}_{\lambda=550 \text{ nm-EC}}$ ratio with nitrophenols and
373 PAHs again revealed an enhanced formation of secondary BrC during the aerosol ageing
374 process.

375 **3.5 Positive matrix factorization (PMF) analysis for BrC source apportionment**

376 In the current work, The EPA PMF 5.0 model was used for identifying the possible
377 sources of BrC. Because the number of the collected samples in each season was not large
378 enough, data from the two seasons were merged together to form a dataset of 80×12 (80
379 samples with 12 species) in order to obtain an accurate analysis according to the PMF user
380 guide. The resolved source profiles (factors) represented the sources that influenced
381 variability in the selected components throughout two seasons in Xi'an. Similar approach was
382 also reported by Zhang et al. (2010). With several iterative testes, a solution with four factors
383 was identified as the optimal solution. As shown in Table S2, the values of Q_{true} and Q_{robust}

384 were consistent, which indicates that the model fits the input data well. Furthermore, the
385 correlation coefficient between input and model values ranged from 0.82 to 0.99 with an
386 average 0.96, also implying that the model fit well. This assess method was widely used in
387 previous studies (Ren et al., 2017;Wang et al., 2009a).

388 Figure7 shows the factor profiles resolved by the model. Factor 01 was characterized by
389 high levels of BeF (52%), BeP (57%), and IP (67%), which were primarily derived from coal
390 combustion and vehicle exhausts (Kong et al., 2010;Ma et al., 2010;Harrison et al., 1996),
391 further, relatively high OC (29%) and EC (25%) associated with this factor was well known
392 tracers of exhaust emissions (Zong et al., 2016), so we identified Factor 01as the source from
393 fossil fuel combustion. Factor 02 (fugitive dust) shows high contribution of Ca²⁺ (69%) and a
394 moderate loading of EC (39%). Ca, as one of the most abundant crustal elements, is largely
395 from construction work, resuspended dust or soil sources (Chow et al., 2004;Han et al., 2007).
396 In addition, EC was a well-known tracer of vehicular emissions (Dorado et al., 2003), so this
397 factor can be attributed to the impact of vehicles passing with higher speeds, leading to
398 resuspend non-tailpipe particles. Moreover, the concentrations of Ca²⁺ in the night were
399 almost higher than that during the day time, with averages of 1.8 ± 1.56 and $1.43 \pm 0.85 \mu\text{g}/\text{m}^3$,
400 respectively. This is consistent with time for transporting the construction wastes by lorry.
401 Thus, factor 02 was identified as fugitive dust. Factor 03 was identified as secondary
402 formation, as it is associated with high loadings of NO₃⁻ (63%), SO₄²⁻ (73%), NH₄⁺ (69%) and
403 a moderate loading of OC and WSOC, indicating the presence of secondary inorganic and
404 organic aerosols. The factor 04 showed high loadings with nitrophenols, levoglucosan, and
405 abs-MeOH and was identified as biomass burning, because levoglucosan is the tracer for

406 biomass burning smoke, and nitrophenols can be produced in the aging process of biomass
407 burning plume.

408 Figure 8 shows the contributions of the above sources to the light absorption at $\lambda=365\text{nm}$,
409 which also represents the fraction of BrC for the factors. Biomass burning was the primary
410 source of the BrC, accounting for 55% of the total BrC in the city, which is coincided with the
411 results discussed in the section 3.2.1. A significant fraction (about 19%) of BrC was
412 associated with fossil fuel combustion. The fraction of secondary BrC was about 16%, which
413 was enhanced during the summer due to the efficient photochemical formation of secondary
414 chromophores. The AAE value of total BrC, closed to the aged SOA-AAE (4.7-5.3) (Bones et
415 al., 2010), can also verify it. The remaining fraction of BrC was derived from the fugitive dust
416 in the city. The results of BrC source apportionment for the Xi'an samples are in line with the
417 work by Shen et al. (2017) and also similar to the results obtained in Beijing by using
418 radiocarbon fingerprinting (Yan et al., 2017).

419 **4. Conclusions**

420 This study investigated the seasonality of the light-absorption characteristics of BrC in
421 Xi'an. Light absorption coefficient (MAC) of methanol-extracts at 365nm was 1.5-1.7 folds
422 higher than that of water-extracts in the two seasons, suggesting non-polar compounds in the
423 city are of stronger light-absorbing ability than that of polar compounds. The strong
424 correlation of levoglucosan with BrC and the diagnostic ratios of levoglucosan/mannosan and
425 levoglucosan/galacosan revealed that the wintertime abundant BrC ($\text{abs}_{\lambda=365\text{nm}}\text{-MeOH}$ of
426 $49.18 \pm 31.67 \text{ M/m}$) in Xi'an was mainly derived from the residential biofuel combustion for
427 house heating in the city and its surrounding region. Size distribution results showed that 81%

428 and 65% of BrC occurred in the fine mode ($< 2.1\mu\text{m}$) during winter and summer, respectively,
429 which is characterized by a monomodal size distribution with a peak in winter and a bimodal
430 size distribution in summer with two peaks in the fine and coarse modes, respectively. The
431 fine mode of MAC is 50% higher in winter than that in summer, suggesting that the
432 light-absorbing ability of wintertime fine particles is stronger, due to the abundant occurrence
433 of PAHs and other aromatic compounds in the fine mode.

434 The linear correlation between the ratio of $\text{abs}_{\lambda=365\text{nm}}\text{-MeOHO}/\text{abs}_{\lambda=550\text{nm}}\text{-EC}$ and the
435 enrichment of ^{13}C during the haze development indicated an accumulation of secondary BrC
436 in the aerosol ageing process. The daytime strong correlation of the ratio of
437 $\text{abs}_{\lambda=365\text{nm}}\text{-MeOHO}/\text{abs}_{\lambda=550\text{nm}}\text{-EC}$ with nitrophenols in the haze event further revealed that
438 such an enhanced production of secondary BrC is related to the photooxidation of aromatic
439 compounds with NO_x . Source apportionment by using PMF showed that 55% of the BrC was
440 associated with biomass burning in the city during the campaign, with 19 and 16% of BrC
441 derived from fossil fuel combustion and secondary formation, respectively.

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445 Author contributions. GW designed the experiment. CW, JianjunL, JinL and CC collected the
446 samples. CW and ZZ conducted the experiments. CW and GW performed the data
447 interpretation and wrote the paper. All authors contributed to the paper with useful scientific
448 discussions or comments.

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452 Competing interests. The authors declare that they have no conflict of interest.

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752 **Table List**

753 Table 1. Concentrations of organic carbon in PM_{2.5} and meteorological conditions during
754 winter and summer of 2017 in Xi'an, inland China.

755
756 Table 2. Comparison on light absorption ($abs_{\lambda=365nm}$), MAC, and AAE values of water-extracts of
757 PM_{2.5} in Xi'an, China with those in other cities.

758
759 Table 3. Complex refractive index (k) of brown carbon from samples extracted by water in
760 two seasons.

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763 **Figure caption**

764 Fig. 1 Temporal variations of WSOC, OC, PM_{2.5}, and $abs_{\lambda=365nm}$ of PM_{2.5} samples extracted by
765 water (H₂O extraction) and methanol (MeOH extraction) during winter (**a** and **c**) and summer
766 (**b** and **d**).

767
768 Fig. 2 Seasonal average values of $abs_{\lambda=365nm}$, AAE, and MAC extracted by MeOH and H₂O.
769 AAE is calculated by linear regression fit $\log(abs_{\lambda=365nm})$ versus $\log(\lambda)$ in the wavelength
770 range of 300–450 nm. (The shadows indicating the standard deviations)

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772 Fig. 3 Size distributions of $abs_{\lambda=365nm}$ and MAC of PM_{2.5} samples extracted by water during
773 the winter and summer of 2017 in Xi'an.

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775 Fig. 4 An orthogonal regression analysis for $abs_{\lambda=365nm}$ of samples between predicted by Mie
776 theory and extracted by water for different particle size ($D_p < 2.1\mu m$).

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778 Fig. 5 Temporal variations of PM_{2.5}, meteorological parameters, $abs_{\lambda=365nm}$ of W(M)SOC and
779 organic compounds in the period of January 10th–20th (The cyan shadow indicates a haze
780 period from January 12th to 19th with a daily PM_{2.5} > 75 $\mu g/m^3$).

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783 Fig.6 Linear fit regressions for the ratio of light absorption of methnoal-extracts to light
784 absorption of EC ($abs_{\lambda=365nm}\text{-MeOH}/abs_{\lambda=550nm}\text{-EC}$) with (**a** and **b**) $\delta^{13}C$ and (**c** and **d**) relative
785 abundance of nitrophenol to EC(Nitrophenol/EC) in the day- and night-PM_{2.5} samples
786 collected during the haze period of January 12th to19th (corresponding to the cyan shadow in
787 Figure 5) in Xi'an.

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789 Fig. 7 Factor profiles resolved by PMF mode during the winter and summer sampling period.
790 The bars represent the concentrations of species and the dots represent the contributions of
791 species appointed to the factors (the summer and winter samples were merged together for the

792 PMF analysis due to the limited number of samples).

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794 Fig. 8 Source apportionment for airborne fine particulate BrC in Xi'an during the campaign.

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800 Table 1. Concentrations of organic carbon in PM_{2.5} and meteorological conditions during
801 winter and summer of 2017 in Xi'an, inland China.

| | Winter | Summer |
|---|---------------|-----------------|
| I. Mass concentrations of organic matter in PM _{2.5} | | |
| WSOC ($\mu\text{gC}/\text{m}^3$) | 23 ± 13 | 5.8 ± 1.4 |
| OC ($\mu\text{gC}/\text{m}^3$) | 41 ± 25 | 8.4 ± 2.4 |
| PAHs (ng/m^3) | 149 ± 89 | 8.1 ± 6.5 |
| OPAHs (ng/m^3) | 174 ± 98 | 17 ± 8.7 |
| Nitrophenols (ng/m^3) | 17 ± 12 | 0.40 ± 0.27 |
| Levoglucosan (ng/m^3) | 739 ± 432 | 29 ± 22 |
| II. PM _{2.5} and meteorological parameters | | |
| PM _{2.5} ($\mu\text{g}/\text{m}^3$) | 194 ± 141 | 37 ± 16 |
| T ($^{\circ}\text{C}$) | 2.6 ± 2.9 | 31 ± 5.4 |
| RH (%) | 60 ± 20 | 58 ± 19 |
| Visibility (km) | 7.0 ± 7.0 | 21 ± 11 |

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822 Table 2. Comparison on light absorption ($abs_{\lambda=365nm}$), MAC, and AAE values of water-extracts of
 823 $PM_{2.5}$ in Xi'an, China with those in other cities.

| Location | Time | $abs_{\lambda=365nm}$ (M/m) | | MAC (m^2/g) | | AAE | | References |
|-------------------------------|-----------|-----------------------------|----------------------|-----------------------|------------------------------------|----------------------|----------------------|-------------------------|
| | | Winter | Summer | Winter | Summer | Winter | Summer | |
| Xi'an, China | 2016-2017 | 49±32 ^a | 5.2±2.1 ^a | 1.3±0.03 ^a | 0.8 ^a ±0.1 ^a | 6.1±9.7 ^a | 5.5±8.8 ^a | This study |
| | | 28±16 | 3.5±1.7 | 1.2±0.06 | 1.1±0.2 | 5.3±8.5 | 4.8±7.7 | |
| | 2008-2009 | 46±20 ^a | 8.3±2.3 ^a | 1.3 ^a | 0.7 ^a | 6.0 ^a | 6.0 ^a | Huang et al. (2018) |
| | | 25±12 | 5.0±1.3 | 1.7 | 1.0 | 5.7 | 5.7 | |
| Beijing, China | 2010-2011 | 10±8.6 | 3.7±3.8 | 1.3 | 0.5 | | | Du et al. (2014) |
| | 2011 | 10±6.9 | | 1.2 | | 7.3 | | Cheng et al. (2016) |
| | 2013 | 14±5.2 | 4.6±2.2 | 1.5 | 0.7 | 5.3 | 5.8 | Yan et al. (2015) |
| Nanjing, China | 2015-2016 | 9.4 ± 4.7 | 3.3±2.4 | 1.0 | 0.5 | 6.7 | 7.3 | Chen et al. (2018) |
| Guangzhou, China | 2012 | 3.6±1.3 | | 0.8 | | 5.3 | | Liu et al. (2018) |
| Delhi, India | 2010-2011 | | | 1.6 | | 5.1 | | Kirillova et al. (2014) |
| | 2015-2016 | 24±19 | | 1.2 | | | | Satish et al. (2017) |
| Indo- Gangetic Plain India | 2011 | 40 ± 18 ^b | | 1.3 ^b | | 5.1 ^b | | Bachi et al. (2016) |
| | | 52 ± 27 ^c | | 1.3 ^c | | 5.3 ^c | | |
| Seoul, Korea | 2013-2013 | 11 ^a | 5.8 ^a | 0.9 ^a | 1.5 ^a | 5.5 ^a | 4.1 ^a | Kim et al.(2016) |
| | | 7.3 | 0.9 | 1.0 | 0.3 | 5.8 | 8.7 | |
| Atlanta, US | 2010 | | 0.6±0.4 | | 1.2-0.2 | | 3.4 | Zhang et al. (2011) |
| Los Angeles Basin, US | 2010 | | 0.4-1.6 | | 0.7 | | 7.6 | Zhang et al. (2013) |

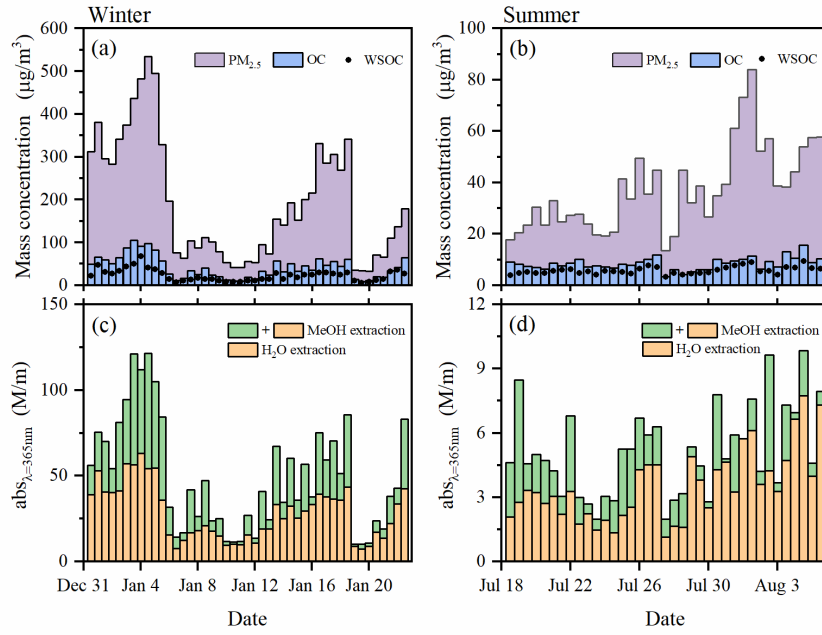
824 Notes: ^a solution extracted by MeOH; ^b samples collected at day time; ^c samples collected in the night

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831 Table 3. Complex refractive index (k) of brown carbon from samples extracted by water in
 832 two seasons.

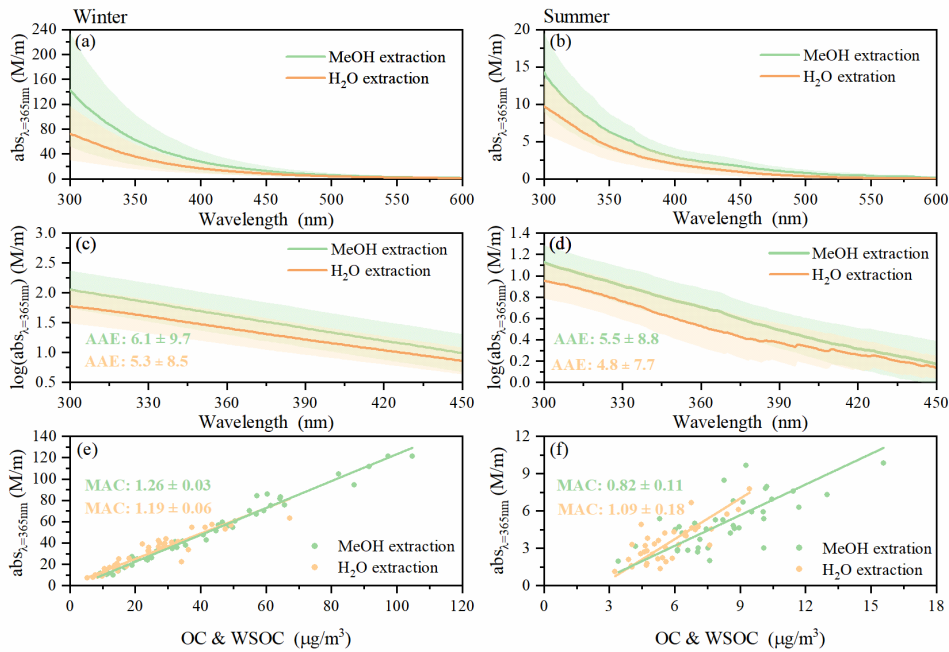
| Particle size (μm) | Winter | Summer |
|---------------------------|---------------|---------------|
| 1.31 | 0.047 ± 0.005 | 0.021 ± 0.010 |
| 0.73 | 0.048 ± 0.008 | 0.033 ± 0.010 |
| 0.45 | 0.048 ± 0.013 | 0.031 ± 0.009 |
| 0.18 | 0.038 ± 0.016 | 0.026 ± 0.008 |

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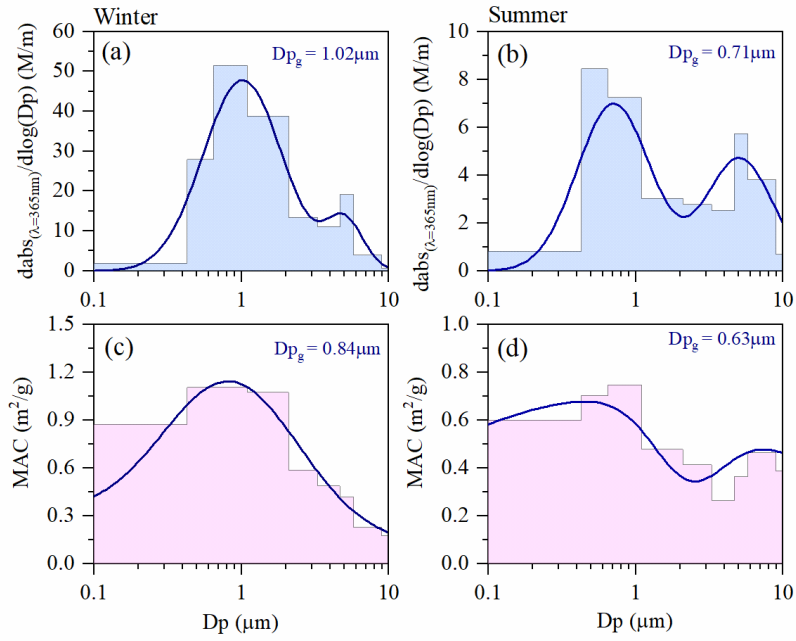
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Fig. 1 Temporal variations of WSOC, OC, $\text{PM}_{2.5}$, and $\text{abs}_{\lambda=365\text{nm}}$ of $\text{PM}_{2.5}$ samples extracted by water (H_2O -extraction) and methanol (MeOH-extraction) during winter (a and c) and summer (b and d).



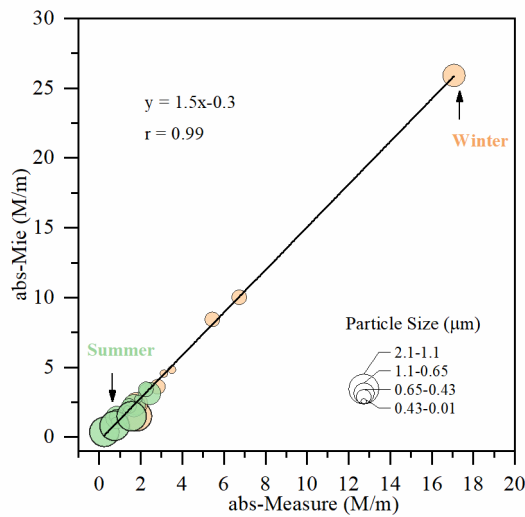
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Fig. 2 Seasonal average values of $\text{abs}_{\lambda=365\text{nm}}$, AAE, and MAC extracted by MeOH and H_2O . AAE is calculated by linear regression fit $\log(\text{abs}_{\lambda=365\text{nm}})$ versus $\log(\lambda)$ in the wavelength range of 300–450 nm. (The shadows indicating the standard deviations)



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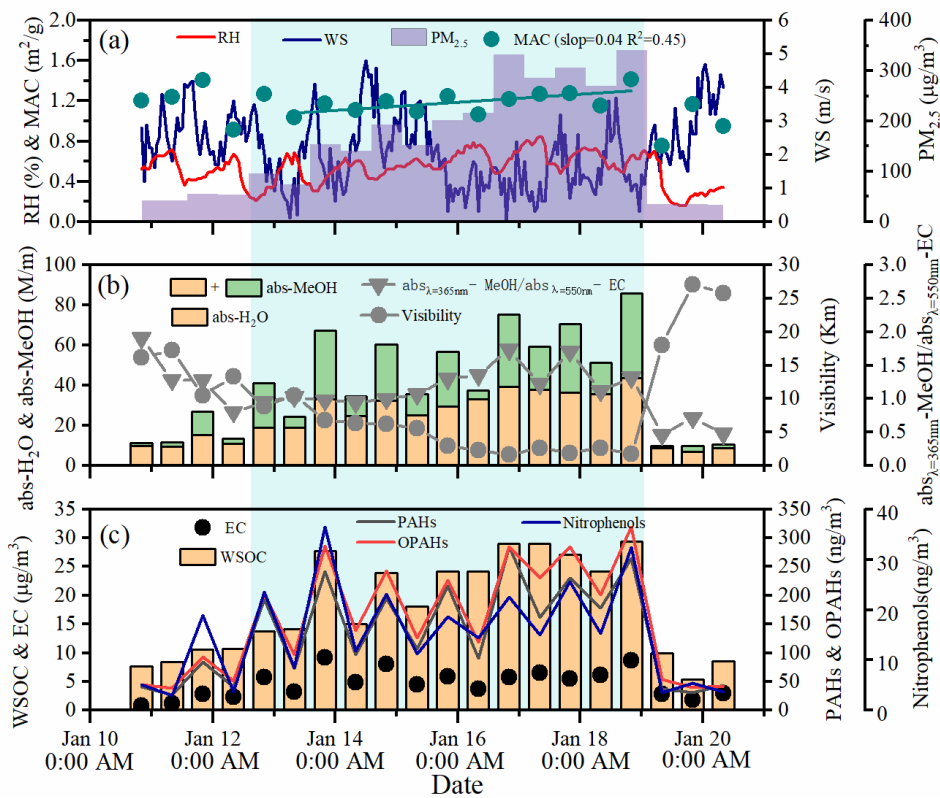
Fig. 3 Size distributions of $abs_{\lambda=365nm}$ and MAC of $PM_{2.5}$ samples extracted by water during the winter and summer of 2017 in Xi'an.



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Fig. 4 An orthogonal regression analysis for $abs_{\lambda=365nm}$ of samples between predicted by Mie theory and extracted by water for different particle size ($D_p < 2.1 \mu m$).

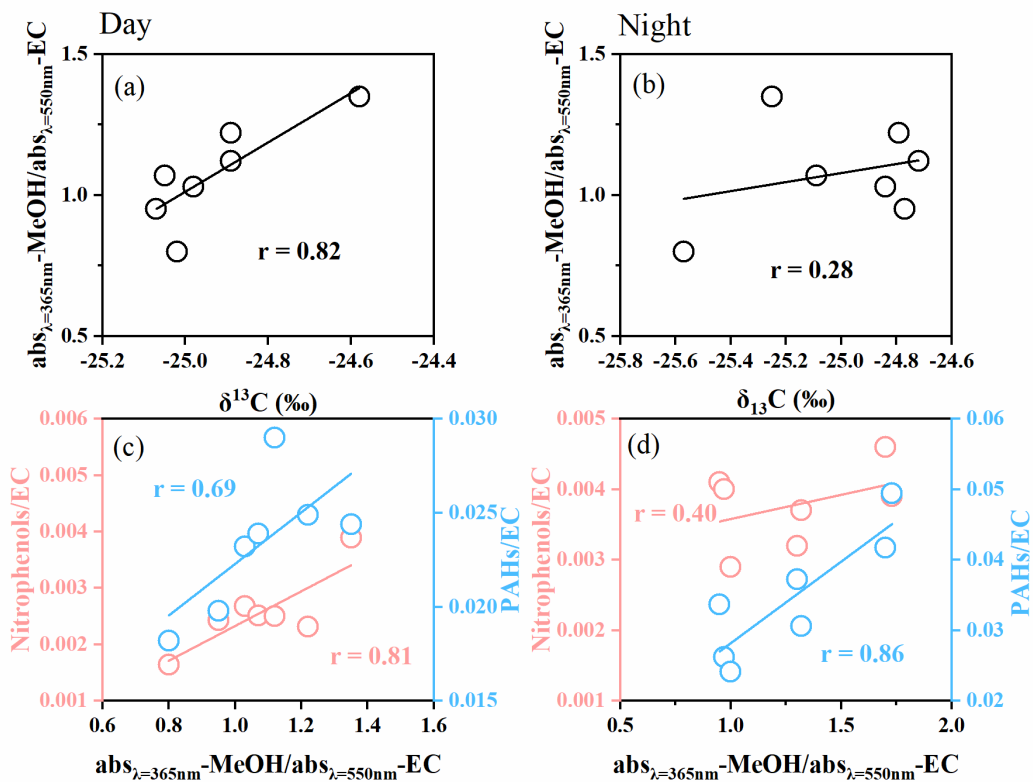
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 873 Fig. 5 Temporal variations of PM_{2.5}, meteorological parameters, $\text{abs}_{\lambda=365\text{nm}}$ of W(M)SOC and
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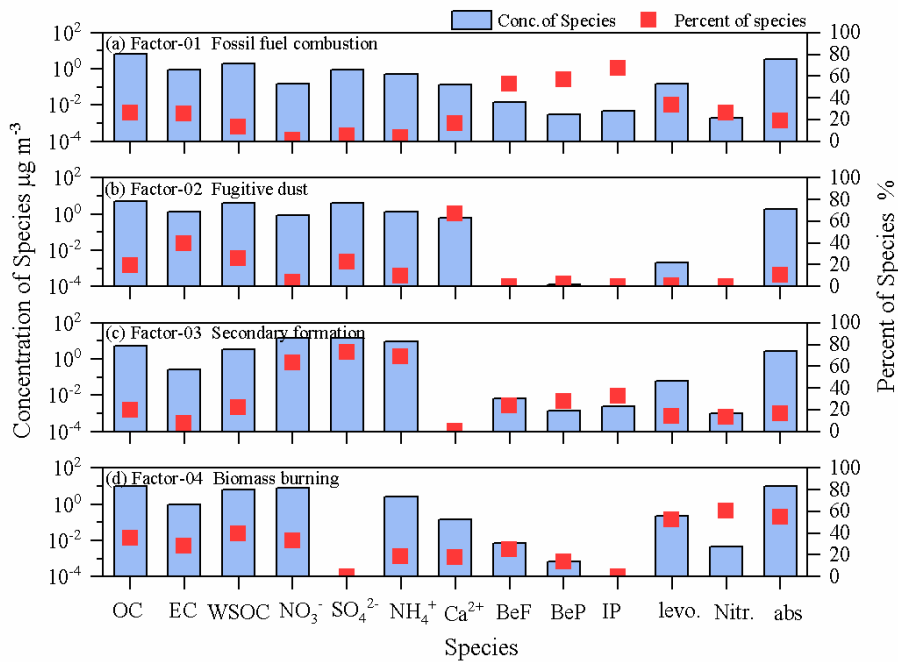
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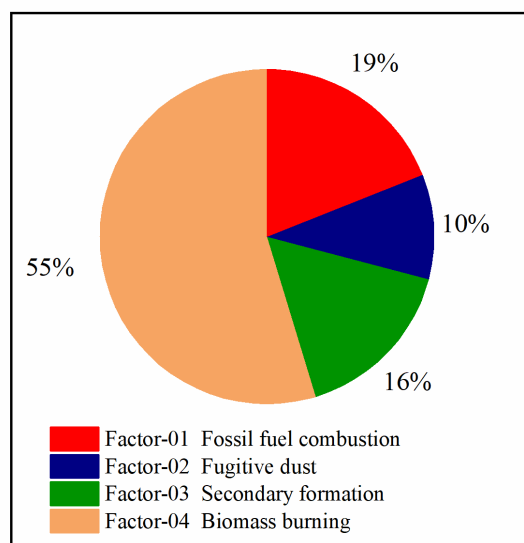
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Fig.6 Linear fit regressions for the ratio of light absorption of methanol-extracts to light absorption of EC ($\text{abs}_{\lambda=365\text{nm}}\text{-MeOH}/\text{abs}_{\lambda=550\text{nm}}\text{-EC}$) with (a and b) $\delta^{13}\text{C}$ and (c and d) relative abundance of nitrophenol to EC (Nitrophenol/EC) in the day- and night-PM_{2.5} samples collected during the haze period of January 12th to 19th (corresponding to the cyan shadow in Figure 5) in Xi'an.



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Fig. 8 Source apportionment for airborne fine particulate BrC in Xi'an during the campaign.