

# **Organic aerosol source apportionment in Zurich using an extractive electrospray ionization time-of-flight mass spectrometry (EESI-TOF): Part II, biomass burning influences in winter**

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## Abstract.

Real-time, in situ molecular composition measurements of the organic fraction of fine particulate matter (PM<sub>2.5</sub>) remain challenging, hindering a full understanding of the climate impacts and health effects of PM<sub>2.5</sub>. In particular, the thermal decomposition and ionization-induced fragmentation affecting current techniques has limited a detailed investigation of secondary organic aerosol (SOA), which typically dominates OA. Here we deploy a novel extractive electrospray ionization time-of-flight mass spectrometer (EESI-TOF-MS) during winter 2017 in downtown Zurich, Switzerland, which overcomes these limitations, together with an Aerodyne high resolution time of flight aerosol mass spectrometer (HR-TOF-AMS) and supporting instrumentation. Positive matrix factorization (PMF) implemented within the Multilinear Engine (ME-2) was applied to the EESI-TOF data to quantify the primary and secondary contributions to OA. An 11-factor solution was selected as the best representation of the data, including 5 primary and 6 secondary factors. Primary factors showed influence from cooking, cigarette smoke, biomass burning (2 factors) and a special local unknown event occurred only during two nights. Secondary factors were affected by biomass burning (3 factors, distinguished by temperature and/or wind direction), organonitrates, monoterpene oxidation, and undetermined regional processing, in particular the contributions of wood combustion. While the AMS attributed slightly over half the OA mass to SOA but did not identify its source, the EESI-TOF showed that most (> 70 %) of the SOA was derived from biomass burning. Together with significant contributions from less aged biomass burning factors identified by both AMS and EESI-TOF, this firmly establishes biomass burning as the single most important contributor to OA mass at this site during winter. High correlation was obtained between EESI-TOF and AMS PMF factors where specific analogues existed, as well as between total signal and POA/SOA apportionment. This suggests the EESI-TOF apportionment in the current study can be approximately taken at face value, despite ion-by-ion differences in relative sensitivity. The apportionment of specific ions measured by the EESI-TOF (e.g. levoglucosan, nitrocatechol, and selected organic acids), and utilization of a cluster analysis-based approach to identify key marker ions for the EESI-TOF factors are investigated. The interpretability of the EESI-TOF results and improved source separation relative to the AMS within this pilot campaign validate the EESI-TOF as a promising approach to source apportionment and atmospheric composition research.

## 25 1 Introduction

Organic aerosol (OA) is relevant due to its roles in several atmospheric processes including radiative forcing, visibility, heterogeneous reactions, and uncertain effects on human health (Nel, 2005; Docherty et al., 2008; Stocker, 2013). OA sources are typically classified as either directly emitted primary organic aerosol (POA) or secondary organic aerosol (SOA) formed from gas-to-particle conversion after chemical reactions. SOA is estimated to comprise approximately 20 % to 90 % of OA, depending on location and time of year (Jimenez et al., 2009; Hallquist et al., 2009). Many studies have successfully linked POA to specific sources, but the level of chemical characterization achieved by conventional online instrumentation has been

in most cases proven insufficient for quantitative resolution of SOA source contributions and/or formation pathways. Therefore, the effects of individual SOA sources on health and climate remain poorly constrained, hampering the design of efficient emissions control policies.

A range of methods to measure molecular composition of aerosol particles have so far mostly been conducted offline, using filter samples (Wang et al., 2009; Daellenbach et al., 2017; Wang et al., 2017). Compared to online methods, offline methods have low time resolution typically integrating aerosol over hours and introducing sampling/storage artifacts (Timkovsky et al., 2015). Moreover, offline measurement techniques like gas chromatography-mass spectrometry (GC-MS) or liquid chromatography-mass spectrometry (LC-MS), are chemically highly specific, but often struggle with the fraction of mass that can be characterized (typically < 20 % of the total OA), which hinders our understanding of the SOA.

Currently available online speciation techniques to measure aerosol particle composition often rely on some type of thermal desorption and/or hard ionization leading to thermal decomposition and/or ionization-induced fragmentation of the original molecules. For example, the Aerodyne aerosol mass spectrometer (AMS) vaporizes molecules at 600 C followed by electron ionization at 70 eV, facilitating quantification but yielding extensive decomposition and fragmentation (Jayne et al., 2000; Sasaki et al., 2001; Samy et al., 2011; Hayes et al., 2013). The chemical analysis of aerosol online-proton transfer reaction mass spectrometer (CHARON-PTR-MS) has no significant thermal decomposition but **the ionization scheme fragments typical SOA molecules** (Eichler et al., 2015; Muller et al., 2017). Several semi-continuous methods have also been developed, including the thermal desorption aerosol-GC (TAG-MS, **GC-family**, Williams et al., 2006) and gas and aerosols-chemical ionization time-of-flight mass spectrometer (FIGAERO-CIMS, Lopez-Hilfiker et al., 2014). However, these systems remain subject to some degree of thermal decomposition, as well as potential reaction on the collection substrate, and significantly lower time resolution. Above all, an online instrument able to detect the original OA and resolve its chemical composition at the molecular level with higher time resolution is needed. The Paul Scherrer Institute (PSI) has developed such an instrument, i.e., the extractive electrospray ionization time-of-flight mass spectrometer (EESI-TOF), measuring particles at molecule level with a time resolution of seconds while overcoming the usual limitations, e.g. thermal decomposition, ionization-induced fragmentation, semi-continuous operation (**Lopez-Hilfiker et al., 2019**).

Due to the lacking ability to apportion SOA to specific sources, a terminology based on properties rather than sources was previously introduced, such as the AMS-based discrimination into semi-volatile and low-volatility oxygenated organic aerosol (SV-OOA and LV-OOA, respectively). The current state-of-the art SOA source apportionment is to be improved based on large laboratory experiments which generate a “library” of species of the SOA products (Zhang et al., 2015; Bianchi et al., 2017; Nakao et al., 2011; Nah et al., 2016; Zhang et al., 2017a). An isoprene-OA source was identified based on fragments in AMS and ACSM (Aerosol Chemical Speciation Monitor) mass spectra that are consistent with those of laboratory-generated isoprene SOA (via reactive uptake of epoxydiols (IEPOX), Xu et al., 2015; Zhang et al., 2017b). Offline analysis identified winter OOA and summer OOA which to some extent appears to be linked to sources (Daellenbach et al., 2017; Daellenbach et al., 2016; Bozzetti et al., 2016), even though the corresponding sources cannot be retrieved (**Lu et al., in prep**). Zhang et al.,

(2018) combined the offline GC-MS method and online FIGAERO-CIMS measurements to better characterize summertime monoterpene SOA.

Domestic wood combustion has been identified as a major source of OA in central Europe (Lanz et al., 2010; Herich et al., 2014), as well as in Asia (Sun et al., 2013; Quan et al., 2014). Recent studies have been devoted to the chemical characterization of the gas and particle-phase emissions from biomass burning in the laboratory, to provide information for a better source apportionment of primary and secondary biomass burning OA (Iinuma et al., 2010; Nakao et al., 2011; Ofner et al., 2011; Chan et al., 2005; Bruns et al., 2017; Bertrand et al., 2018). Various tracer compounds for biomass burning were reported, including levoglucosan, which is a sugar anhydride compound produced from the pyrolysis of cellulose and hemicellulose (Fine et al., 2001), or methoxyphenols (e.g. guaiacol and syringol), derived from the pyrolysis of lignin (Coeur-Tourneur et al., 2009; Veres et al., 2010; He et al., 2018), and methyl-nitrocatechols, nitrated aromatic compounds from biomass burning (Iimuma et al., 2010a). Furthermore, biomass burning has been shown to produce significant SOA in laboratory measurements (Brunns et al., 2016; Nakao et al., 2011; Yee et al., 2013; Stefenelli et al., 2019), but this component has not yet been resolved in the field with the partial exception of winter OOA.

Here, we report on a study in Zurich, a mid-size city in Central Europe, utilizing the EESI-TOF, complemented with AMS source apportionment results for a winter case. Summer measurement and source apportionment are presented in the companion paper (Stefenelli et al., 2019). In both cases, due to the enhanced chemical resolution of the EESI-TOF we are able to resolve more POA and SOA sources than in previous studies at the same site.

## 2 Methodologies

### 2.1 Measurement Campaign

Measurements were performed from 25 January to 5 February, 2017 at the Swiss National Air Pollution Monitoring Network (NABEL) station at Zurich Kaserne, Switzerland (Richard et al., 2011). The station is located in the center of the metropolitan area of Zurich (1.3 million inhabitants). It is characterized as an urban background site, although several restaurants are nearby (Lanz et al., 2007). Long-term measurements at the site include ambient meteorological data such as temperature, relative humidity (RH), solar radiation, wind speed and direction, trace gas measurements comprising nitrogen oxides (NO<sub>x</sub>, Thermo Environmental Instruments 42i, Thermo Electro Corp., Waltham, MA) and ozone (O<sub>3</sub>, Thermo Environmental Instruments 49C, Thermo Electro Corp., Waltham, MA), and particle measurements which include size distributions (scanning mobility particle sizer, SMPS, TSI) and number concentration (condensation particle counter, CPC). Although the measurement period is relatively short (12 days), the similarity of the AMS results obtained in the current study to previous AMS and ACSM measurements at the same site (Lanz et al., 2007, Canonaco et al., 2013, Richard et al., 2011, Daellenbach et al., 2016) give us high confidence that the sampled aerosol is representative of typical wintertime conditions. Exceptions to this are resolved by the source apportionment into unique event-driven factors, as discussed in the results section.

For the intensive campaign, an EESI-TOF, an HR-TOF-AMS (Aerodyne Research Inc.) and an SMPS were additionally deployed. The sampling was performed in a mobile trailer installed outside the NABEL station. Ambient air was sampled through a PM<sub>2.5</sub> cyclone to remove coarse particles (~75 cm above the trailer roof and ~5 m above ground). The air passed through a stainless steel (~6 mm) tube into the AMS, EESI-TOF, and SMPS, installed on the same line and in close proximity.

## 5 2.2 Instrumentation

### 2.2.1 Extractive Electrospray Ionization Time-of-flight Mass Spectrometer (EESI-TOF)

The extractive electrospray ionization time-of-flight mass spectrometer (EESI-TOF) is a novel instrument for real-time measurement of organic aerosol without thermal decomposition or ionization-induced fragmentation. The instrument is discussed in detail elsewhere (Lopez-Hilfiker et al., 2019) and a brief overview is presented here. Ambient aerosol is continuously sampled at 900 cm<sup>3</sup> min<sup>-1</sup>, either directly or through a particle filter to yield a background measurement. In this study, 10 mins ambient air sampling was alternated with 2 mins through the filter with spectra recorded with 40 s time resolution. The flow then passes through a 5 cm long 6 mm outer diameter (OD) multi-channel extruded carbon denuder housed in a stainless steel tube, which removes most trace gas phase species. The denuder eliminates artifacts from semi-volatile species desorbing from the filter, and also improves detection limits by reducing the gas-phase background. The particle-laden flow then intersects a spray of charged droplets generated by a conventional electrospray capillary. Particles collide with the electrospray droplets and the soluble components are extracted, ionized by Coulomb explosion of the charged droplets, and detected by TOF-MS (resolution ~4000 at mass to charge ratio ( $m/z$ ) 185). The electrospray droplets are generated by a commercially available 360  $\mu$ m OD untreated fused silica capillary with an inner diameter of 50  $\mu$ m (BGB Analytic). The sample flow remains unheated until after extraction of aerosol material into the electrospray droplets, minimizing volatilization of labile particle phase components and thermal decomposition. The droplets then enter the mass spectrometer through a capillary heated to 250 C, however, the very short residence time in this capillary means that the effective temperature experienced by the analyze is much lower and no thermal decomposition is observed. The electrospray working solution is a 50/50 water/acetonitrile (> 99.9 %, Sigma-Aldrich) mixture, which has less background signal compared to the water/methanol mixture, with 100 ppm of sodium iodide (NaI) as a charge carrier. Spectra are recorded in positive ion mode, in nearly all cases as adducts with Na<sup>+</sup>. Depending on voltage settings in the ion transfer optics (i.e. collision energy), clusters with acetonitrile can potentially be detected, however these clusters were observed to be negligible during the current study. The recorded signals are linear with mass and free of detectable matrix effects, in part due to the suppression of ionization pathways other than Na<sup>+</sup> adduct formation (Lopez-Hilfiker et al., 2019). Here we report the signal measured by the EESI-TOF in terms of the mass flux of ions to the microchannel plate detector (attograms s<sup>-1</sup>, neglecting Na<sup>+</sup>), calculated as shown in Eq. 1.

$$30 \quad M_x = I_x \cdot (MW_x - MW_{cc}) \quad (1)$$

Here  $M_x$  is the mass flux of ions united in  $\text{ag s}^{-1}$ ,  $x$  represents the measured molecular composition.  $I_x$  is the recorded signal measured by EESI-TOF.  $MW_x$  and  $MW_{cc}$  represent the molecular weight of the ion and the charge carrier (e.g. H<sup>+</sup>, Na<sup>+</sup>),

respectively. Note that this measured mass flux can be related to ambient concentration by the instrument flow rate, EESI extraction/ionization efficiency, declustering probability, and ion transmission, where several of these parameters are ion-dependent (Lopez-Hilfiker et al., 2019). A comparison of the EESI-TOF mass flux to the AMS signal in terms of total signal or mass, bulk properties, and source apportionment results in section 3.5.

5 With the EESI-TOF, we almost continuously collected data from 25 January to 5 February, 2017 (84.6 %), missing a few time points due to instrumental calibration and issues such as cleaning the electrospray capillary due to lost or unstable signal. EESI-TOF stability and linearity with mass were confirmed by periodic measurement of nebulized levoglucosan aerosol with quantification of the mass concentration with an SMPS.

Data processing was executed using Tofware version 2.5.7 (Tofwerk AG, Thun, Switzerland). The total number of 1125 fitted ions (including 882 Na<sup>+</sup> adducts, one H<sup>+</sup> adduct, and 242 unknown ions) between  $m/z$  135 and 400 were identified. Negligible signal was detected below  $m/z$  135 due to the selected mass spectrometer transmission window. Data were pre-averaged to 1 min time resolution, and high resolution peak fitting was performed. Individual 1-min spectra were classified as either ambient measurements, background sampling (through the particle filter), or transitional measurements immediately after switching between ambient/background sampling. Transitional measurements were excluded from further analysis. Background spectra were averaged across each 2-min filter period, and these filter periods were interpolated to estimate the background spectrum during each ambient sampling period. The estimated backgrounds were subtracted from individual ambient spectra to yield the final ion time series of difference spectra. Ions with a mean signal-to-noise ratio (SNR) below 2 were removed from further analysis. No corrections for the relative sensitivity of individual ions or drift in instrument sensitivity were applied. For the Multilinear engine (ME-2) source apportionment analysis (Sect. 2.3), data were re-averaged to 2 mins. The corresponding error matrix  $\sigma_{ij}$ , which has the same dimensions as the data matrix, follows the model of Allan et al. (2003), which calculation includes the uncertainty deriving from electronic noise, ion-to-ion variability at the detector and ion counting statistics. The error estimates in this case incorporate the uncertainties related to both the ambient measurements ( $\delta_i$ ) (direct ambient sampling period) and the background ( $\beta_{ij}$ ) (filter blank measuring period, both are processed with Tofware), which are combined in quadrature according to Eq. 2:

$$25 \quad \sigma_{ij} = \sqrt{\delta_{ij}^2 + \beta_{ij}^2} \quad (2)$$

The final data matrix and error matrix has the size of 10165 (time series)  $\times$  892 (variables).

### 2.2.2 Aerosol Mass spectrometer (AMS)

30 An HR-TOF-AMS was deployed for online measurements of non-refractory (NR) PM<sub>2.5</sub> (with an inline PM<sub>2.5</sub> cyclone). A detailed description of the instrument can be found elsewhere (Jayne et al., 2000; DeCarlo et al., 2006). The AMS recorded data with 1 min time resolution, of which 30 s was spent recording the ensemble mass spectrum (mass spectrum (MS) mode) and 30 s recording size-resolved mass spectra (“particle time-of-flight (ePTof) mode”). A Nafion dryer was used to dry the

sampled air stream, which kept the relative humidity (RH) of air below 30 % within the AMS. Particles are continuously sampled ( $\sim 0.8 \text{ L min}^{-1}$ ) through a  $100 \mu\text{m}$  critical orifice and are focused by a recently developed  $\text{PM}_{2.5}$  aerodynamic lens (Williams et al., 2013). The particles impact on a heated tungsten surface (heated to  $600^\circ$ ) at  $\sim 10^{-7}$  Torr and the NR components are flash vaporized. The resulting gases are ionized by electron ionization (EI,  $\sim 70 \text{ eV}$ ) and the  $m/z$  values of the resulting fragments are determined by the TOF mass spectrometer. The AMS was calibrated for inlet flow, ionization efficiency (IE) at the beginning, middle and end of the campaign following standard protocols.

AMS data were analyzed in Igor Pro 6.36 using the SQUIRREL (version 1.57) and PIKA (1.16) analysis software (Donna Sueper, ToF-AMS High Resolution Analysis Software). The collection efficiency (CE) was estimated using a composition-dependent collection efficiency (CDCE) algorithm (Middlebrook et al., 2012) implemented in SQUIRREL. A CE=0.5 was assumed except in the case of strongly acidic aerosols, and high ammonium nitrate content where the approach by Middlebrook et al. (2012) was applied.

For ME-2 analysis, the input matrices consisted of the time series of fitted ions from high resolution mass spectral analysis, together with their corresponding uncertainties (Allan et al., 2003). According to the recommendations of Ulbrich et al. 2009, a minimum error value was added to the error matrix and ions were assessed and treated according to their signal-to-noise ratio (SNR) as follows: ions with an SNR less than 0.2 were excluded from ME-2 analysis, while those with an SNR between 0.2 and 2 were down-weighted by increasing their uncertainties by a factor of 2. Further, ions that were not independently fit but rather calculated from  $\text{CO}_2^+$  were removed to avoid overweighting  $\text{CO}_2^+$ . Likewise, isotopes were not included in the matrices to avoid overweighting the parent ions. The source apportionment input matrices consisted of 257 ions between  $m/z$  12 and 120.

### 2.3 Source apportionment techniques

Source apportionment was performed on the organic AMS and EESI-TOF data separately using PMF as implemented by the multilinear engine (ME-2) (Paatero, 1997) and with model configuration and analysis executed via the SoFi (Source Finder, version 6.39) interface (Canonaco et al., 2013), coded in Igor Pro (WaveMetrics 6.37). PMF represents the input data matrix as a linear combination of characteristic factor profiles and their time-dependent contributions, which can be expressed in matrix notation as:

$$\mathbf{X} = \mathbf{G} \times \mathbf{F} + \mathbf{E} \quad (3)$$

The measured  $\mathbf{X}$  is an  $m \times n$  matrix, representing  $m$  measurements of  $n$   $m/z$ .  $\mathbf{G}$  and  $\mathbf{F}$  are  $m \times p$  and  $p \times n$  matrices, respectively, where  $p$  is the number of factors contained in a given model solution and is selected by the user.

Equation (3) is solved using a least squares algorithm that iteratively minimizes the quantity  $Q$  (Eq. 4), defined as the sum of the squared residuals weighted by their respective uncertainties, where the uncertainty may contain the measurements and model uncertainty:

$$Q = \sum_i \sum_j \left( \frac{e_{ij}}{\sigma_{ij}} \right)^2 \quad (4)$$

Here,  $e_{ij}$  represents the residuals (elements of  $\mathbf{E}$ ), with  $i$  and  $j$  denoting respectively the time and  $m/z$  indices, and  $\sigma_{ij}$  is the corresponding measurement uncertainty. Rotations are explored by using the  $a$ -value approach, here implemented by constraining one or more output factor profiles to resemble a selected source, improving source separation (Crippa et al., 2014; Canonaco et al., 2013). The  $a$ -value (ranging from 0 to 1) determines how much the constrained factor ( $f_j$ , solution) is allowed to vary from its anchor ( $f_j$ ), as defined in Eq. (5).

$$f_{j,solution} = f_j \pm a \cdot f_j \quad (5)$$

Execution of PMF analysis on separated AMS and EESI-TOF datasets minimizes the complexity of the analysis, while maximizing the factor resolution ability of the EESI-TOF. Factor related to traffic was constrained for the AMS analysis, while a factor related to cigarette smoke was constrained for the EESI-TOF. Details are presented in Sect. 3.1 and 3.2. Different factors were constrained in the two datasets due to the fundamental differences between the AMS and EESI-TOF measurements. Specifically, the absence of fragmentation in the EESI-TOF allowed clear separation of cooking without the need for constraints, while separation of a cigarette smoke factor was only achieved for the EESI-TOF. In addition, constraining an AMS cigarette smoke factor was attempted but failed.

## 2.4 Wind regression analysis

Wind regression analysis has been developed as a means of using meteorological and pollutant data to estimate the percent of a given pollutant originating from a specific wind sector. This study utilizes the Sustained Wind Incidence Method (SWIM), a quantitative model that estimates the weighted pollutant concentrations and uncertainties from a given wind direction and speed (Henry et al., 2009; Olson et al., 2012). The expected concentration ( $E$ ) of a pollutant for each wind direction/wind speed pair ( $\theta, u$ ) is calculated as a weighted average of the concentration data in a window around ( $\theta, u$ ) represented by smoothing parameters  $\sigma$  and  $h$  using a weighting function  $\mathbf{K}(\theta, u, \sigma, h) = \mathbf{K}_1(\theta, \sigma) \mathbf{K}_2(u, h)$ , given by Eq (6):

$$E(C|\theta, u) = \frac{\sum_{i=1}^N K_1\left(\frac{\theta - W_i}{\sigma}\right) \cdot K_2\left(\frac{u - U_i}{h}\right) \cdot C_i}{\sum_{i=1}^N K_1\left(\frac{\theta - W_i}{\sigma}\right) \cdot K_2\left(\frac{u - U_i}{h}\right)} \quad (6)$$

$$K_1(x) = \frac{1}{\sqrt{2\pi}} \cdot e^{-0.5 \cdot x^2}, -\infty < x < \infty \quad (7)$$

$$K_2(x) = 0.75 \cdot (1 - x^2), -1 < x < 1 \quad (8)$$

$$W_i = \frac{C_i U_i}{\max(C_i U_i)} \cdot \frac{(\overline{\sigma \theta_i})}{\sigma \theta_i} \quad (9)$$

where  $C_i$ ,  $U_i$ , and  $W_i$  are the observed concentration of a particular pollutant, resultant wind speed and directional standard deviation, respectively,  $N$  is the total number of observations;  $\mathbf{K}_1$  (Eq (7)) and  $\mathbf{K}_2$  (Eq (8)) are smoothing Gaussian kernel and



the Epanechnikov kernel,  $\sigma$  and  $h$  are smoothing parameters for wind direction and wind speed, respectively. The conditional probability of pollutant concentration (Eq (6)) is then weighted by the frequency of the wind using a joint probability of wind speed and wind direction, resulting in the following expression for the mean value of the pollutant concentration associated with winds from the sector defined by the intervals  $U$  ( $U = [u_1, u_2]$ ) and  $\theta$  ( $\theta = [\theta_1, \theta_2]$ ).

$$S(\theta, U) = \int_{u_1}^{u_2} \int_{\theta_1}^{\theta_2} f(\theta, u) E(C|\theta, u) d\theta du \quad (10)$$

The joint probability of wind speed and wind direction ( $f$ ) is calculated by using a kernel density, estimated as:

$$f(\theta, u) = \frac{1}{N\sigma h} \sum_{i=1}^N K_1\left(\frac{(\theta - \theta_i)}{\sigma}\right) K_2\left(\frac{(u - u_i)}{h}\right) \quad (11)$$

Calculations have been performed on Igor Pro with ZeFir package (Petit et al., 2017).

## 2.5 Identification of source-specific ions

- 10 To determine ions characteristic of individual factors (or groups of related factors), agglomerative hierarchical clustering was conducted on the EESI-TOF matrix of PMF profiles and standardizing data along the ions, clustering first along the columns (producing row-clustered groups of factor), and then along the rows (producing the clustered ions to each group) in the matrix data. In hierarchical cluster analysis, a dendrogram, used to show relationships between members of a group, i.e., a family tree with the oldest common ancestor at the base and branches for various divisions of lineage was generated with the following
- 15 steps (Matlab R2017b): (1) Calculate the distance by using Euclidean distance to find the similarity or dissimilarity between every ion and every pair of factors in our data set. (2) Link pairs of ions and factors that are in close proximity using the average linkage function. (3) Use the cluster function to prune branches off the bottom of hierarchical tree, and assign all the objects below each cut to a single cluster. Here, the clustergram function transforms the standardized values so that the mean is 0.

## 20 3 Results and Discussion

- Results of AMS and EESI-TOF PMF analyses are presented in sections 3.1 and 3.2, respectively. Section 3.3 focuses on the EESI-TOF PMF results are then exploited to assess the apportionment of specific ions related to key marker compounds (section 3.3) and to identify groups of molecules uniquely characteristic of the retrieved factors (section 3.4). However, quantitative interpretation of the EESI-TOF PMF results is complicated by differences in the relative sensitivity of the EESI-
- 25 TOF to different molecules. Therefore section 3.5 presents a comparison of the EESI-TOF and AMS results in terms of total signal, bulk atomic composition, and relative apportionment to the different factors.

### 3.1 AMS source apportionment

The AMS PMF analysis yielded seven OA factors: hydrocarbon-like OA ( $\text{HOA}_{\text{AMS}}$ ), cooking-related OA ( $\text{COA}_{\text{AMS}}$ ), biomass burning OA ( $\text{BBOA}_{\text{AMS}}$ ), two oxygenated OA factors (less oxygenated ( $\text{LO-OOA}_{\text{AMS}}$ ) and more oxygenated ( $\text{MO-OOA}_{\text{AMS}}$ )), nitrogen-containing OA ( $\text{NOA}_{\text{AMS}}$ ), and a factor due to an isolated local event ( $\text{EVENT}_{\text{AMS}}$ ). Factor mass spectra are shown in Fig. 1, while Fig. S1 shows factor time series, together with selected external tracers, and diurnal cycles which may be less convinced due to the short period of the measurement. Salient characteristics of these factors are discussed below;  $\text{HOA}_{\text{AMS}}$ ,  $\text{COA}_{\text{AMS}}$ ,  $\text{BBOA}_{\text{AMS}}$ ,  $\text{LO-OOA}_{\text{AMS}}$ , and  $\text{MO-OOA}_{\text{AMS}}$  are similar to factors frequently observed in other studies (Crippa et al., 2013a; Zhang et al., 2011; Young et al., 2016).

$\text{HOA}_{\text{AMS}}$  was constrained using a factor mass spectrum from Paris (Crippa et al., 2013b) and an  $a$ -value of 0.1 (the  $a$ -value was selected according to the correlations between the time series of HOA with the traffic species  $\text{NO}_x$ ), yielding a factor with a low O:C ratio (0.04) and high H:C ratio (1.8), consistent with a dominant contribution from aliphatic hydrocarbons. Strong signals from  $\text{C}_x\text{H}_y^+$  ions are evident, especially  $\text{C}_3\text{H}_5^+$ ,  $\text{C}_3\text{H}_7^+$ ,  $\text{C}_4\text{H}_7^+$ , and  $\text{C}_4\text{H}_9^+$  ions. Consistent with previous studies, the HOA mass spectrum is similar to vehicle emission studies (Zhang et al., 2005; Sun et al., 2012; Young et al., 2016).

The  $\text{COA}_{\text{AMS}}$  mass spectrum is similar to primary cooking emissions (Crippa et al., 2013b) and exhibits a unique diurnal pattern peaking during lunch and dinner time. The  $\text{COA}_{\text{AMS}}$  spectrum is characterized by a high ratio of  $\text{C}_4\text{H}_7^+:\text{C}_4\text{H}_9^+$  ratio and a high fraction of  $\text{C}_3\text{H}_3\text{O}^+$  and  $\text{C}_4\text{H}_7^+$ , consistent with  $\text{COA}_{\text{AMS}}$  factors previously identified at urban locations (Crippa et al., 2013a; Ge et al., 2012; Mohr et al., 2012).

$\text{BBOA}_{\text{AMS}}$  has been identified as a significant source of aerosol in previous wintertime source apportionment studies in Switzerland and Central Europe (Lanz et al., 2008; Daellenbach et al., 2017). Similar to previous studies,  $\text{BBOA}_{\text{AMS}}$  shows a high fraction  $\text{C}_2\text{H}_4\text{O}_2^+$  at  $m/z$  60 and  $\text{C}_3\text{H}_5\text{O}_2^+$  at  $m/z$  73 and explains most of the variation of these ions (77 %, 65 %, respectively). A strong diurnal trend is evident, with concentrations peaking overnight and decreasing during the day.

$\text{LO-OOA}_{\text{AMS}}$  and  $\text{MO-OOA}_{\text{AMS}}$  mass spectra are characterized by dominant peaks at  $m/z$  28 ( $\text{CO}^+$ ), 44 ( $\text{CO}_2^+$ ), similar to  $\text{OOA}_{\text{AMS}}$  factors observed at other sites (Sun et al., 2011; Ng et al., 2010). The main difference between the  $\text{LO-OOA}_{\text{AMS}}$  and  $\text{MO-OOA}_{\text{AMS}}$  mass spectra is the relative contribution of  $\text{C}_2\text{H}_3\text{O}^+$  compared to  $\text{CO}_2^+$ , with  $\text{C}_2\text{H}_3\text{O}^+$  enhanced in  $\text{LO-OOA}_{\text{AMS}}$ . Also enhanced in  $\text{LO-OOA}$  are ions at  $m/z$  39 ( $\text{C}_3\text{H}_3^+$ ), 41 ( $\text{C}_3\text{H}_5^+$ ), 55 ( $\text{C}_4\text{H}_7^+$ ). Further insight into the  $\text{OOA}_{\text{AMS}}$  factors is obtained through the EESI analysis (Sect. 3.2).

$\text{NOA}_{\text{AMS}}$  exhibits a significantly higher N:C ratio (0.04) than the other factors and explains most of the organic nitrogen signal. This factor includes a strong signal from  $\text{C}_5\text{H}_{10}\text{N}^+$  signal ( $m/z$  84), which is consistent with N-methyl-pyrrolidine which has previously been identified in AMS spectra as a tracer for cigarette smoke (Struckmeier et al., 2016). This ion is also observed in the EI mass spectra of nicotine (NIST, <https://webbook.nist.gov/cgi/cbook.cgi?ID=C54115&Mask=200#Mass-Spec>). However, other spectral features (e.g. the high  $\text{CO}_2^+$  signal) are not typical of primary cigarette smoke and suggest a contribution from secondary formation processes. This interpretation is consistent with correlations of  $\text{NOA}_{\text{AMS}}$  with EESI-TOF factors, suggesting  $\text{NOA}_{\text{AMS}}$  to be a mixed factor, as discussed in Sect. 3.2 and 3.4.

EVENT<sub>AMS</sub> factor is a special case in our study as the mass spectrum is dominated by  $m/z$  15 ( $\text{CH}_3^+$ ), 27 ( $\text{C}_2\text{H}_3^+$ ), 31 ( $\text{CH}_3\text{O}^+$ ), and 43 ( $\text{C}_2\text{H}_3\text{O}^+$ ). The time series only contributes during two nights (28 January and 29 January) from 00:00 am to 07:00 am with the concentrations peaking at  $3.8 \mu\text{g m}^{-3}$  but being below  $0.2 \mu\text{g m}^{-3}$  for the rest of the study. No associations with any markers are evident.

## 5 3.2 EESI-TOF source apportionment

An 11-factor solution was selected as the best representation of the EESI-TOF data, with 5 factors attributed mostly to POA and 6 to SOA. The POA factors include cooking-related OA ( $\text{COA}_{\text{EESI}}$ ), two less aged biomass burning factors ( $\text{LABB1}_{\text{EESI}}$  and  $\text{LABB2}_{\text{EESI}}$ ) which are mostly dominated by primary organic aerosol compounds, cigarette smoke-influenced OA ( $\text{CS-OA}_{\text{EESI}}$ ), and a factor related to an isolated special event ( $\text{EVENT}_{\text{EESI}}$ ). The SOA factors consist of 3 more aged biomass burning factors dominated by secondary organic aerosol compounds and distinguished by mean daily temperature ( $\text{MABB\_LOW}_{\text{EESI}}$ ,  $\text{MABB\_HIGH}_{\text{EESI}}$ , and  $\text{MABB\_TRANS}_{\text{EESI}}$ , corresponding to low temperature, high temperature, and transition periods, respectively), two additional SOA factors lacking a clear attribution to biomass burning ( $\text{SOA1}_{\text{EESI}}$  and  $\text{SOA2}_{\text{EESI}}$ ), and nitrogen-containing SOA ( $\text{NSOA}_{\text{EESI}}$ ). This solution was obtained by constraining the CS-OA<sub>EESI</sub> factor with an  $\alpha$ -value of 0.1, and all other factors unconstrained. This constraining approach and the solution selection criteria are discussed in Sect. 3.2.1, while the POA and SOA factors are discussed in Sect. 3.2.2 and 3.2.3, respectively. A detailed investigation of the factor mass spectra is presented in Sect. 3.4.

### 3.2.1 Selection of PMF solution

In selecting the PMF solution that best represents the EESI-TOF dataset, we considered both mathematical diagnostics (e.g.  $Q/Q_{\text{exp}}$  and residuals) as a function of the number of factors, as well as the interpretability of the retrieved factors. Interpretability was judged according to the following criteria:

- i. Correlation of the time series and diurnal patterns between the AMS and EESI-factors.
- ii. Comparison of factor profiles with mass spectra retrieved from less and more aged biomass burning exhaust from simulation chamber experiments at PSI (Bertrand et al., submitted).
- iii. Similarities to EESI-TOF factor mass spectra retrieved from summer measurements at the same site in Zurich (Stefenelli et al., 2019)
- iv. Identification of key ions in the factor profiles, including ions contributing a major fraction of the total factor signal, ions apportioned predominantly to a certain factor or related to a set of factors, and ions established in the literature as known tracers for specific sources/processes.
- v. Interpretation of the temporal behavior in terms of meteorological data, including temperature, solar radiation, and wind speed/direction.

For the EESI-TOF source apportionment, we considered unconstrained solutions from 7 to 20 factors (see Fig. S2a). Of these solutions, a 10-factor solution was found to best explain the data at a preliminary stage. This was preferred to lower-order solutions because all factors were interpretable according to the above criteria. Solutions with more factors lead to additional factors related to more aged biomass burning without obvious additional information. In addition, the investigation of  $Q/Q_{\text{exp}}$  as a function of the number of factors (Fig. S2b) did not show any significant change with the increase of the  $a$ -value from 7 factors. Fig. S3 and Fig. S4 show the mass spectra and time series of the 8- to 11-factor solutions.

Nonetheless, the unconstrained 10-factor solution revealed evidence of factor mixing, as the cooking-related ( $\text{COA}_{\text{EESI}}$ ) factor mass spectrum had a strong contribution from  $m/z$  163 ( $\text{C}_{10}\text{H}_{15}\text{N}_2$ , nicotine), which should rather be associated with cigarette smoke (Fig. S5). This suggests that at least one more factor remains to be resolved. The difficulty in separating these factors, despite their expected chemical differences, is likely due to strong temporal correlation between cooking and cigarette-smoking emissions due to the proximity of local restaurants (Fig. S6, the diurnal patterns of nicotine and  $\text{COA}_{\text{EESI}}$  factors), where people gather outside to smoke during mealtimes. We therefore attempted to obtain a clean cigarette smoking signature from the dataset to serve as an anchor profile with which to constrain this source. For solutions with fifteen to twenty factors, a factor was retrieved with an MS dominated by nicotine and to which  $> 90\%$  of nicotine was apportioned. We therefore constructed a profile (average from 15-20 factors) for this nicotine-containing factor (apportioned to cigarette smoke, i.e.  $\text{CS-OA}_{\text{EESI}}$ ). This profile was then constrained in an 11-factor solution (based on the selection of a 10-factor unconstrained solution, as discussed above) using an  $a$ -value approach (from 0 to 1 with steps of 0.1, 0.1 was chosen finally). The main criterion of the constraint was the fraction of nicotine apportioned to the constrained factor. Also in our case, the  $R^2$  (Pearson) for the correlations between the time series of the solutions was constructed with the final 11-factor solution. Based on these considerations, we concluded that the source apportionment solution with eleven factors was the optimal solution.

### 3.2.2 EESI-TOF Factors: Primary Organic Aerosols (POA)

Figure 2a shows the time series of the five EESI-TOF factors attributed to primary organic aerosol:  $\text{COA}_{\text{EESI}}$ ,  $\text{LABB1}_{\text{EESI}}$ ,  $\text{LABB2}_{\text{EESI}}$ ,  $\text{CS-OA}_{\text{EESI}}$ , and  $\text{EVENT}_{\text{EESI}}$ . Also shown are relevant ancillary measurements, including AMS PMF factors and meteorological parameters. Figure 2b shows the corresponding factor mass spectra, colored by the number of nitrogen atoms. A discussion of each factor follows. Figure 3a shows the diurnal patterns of the  $\text{LABB}_{\text{EESI}}$  factors, as well as  $\text{COA}_{\text{EESI}}$  and  $\text{COA}_{\text{AMS}}$ .

#### *Less aged biomass burning ( $\text{LABB1}_{\text{EESI}}$ and $\text{LABB2}_{\text{EESI}}$ )*

The  $\text{LABB}$  factors are both enhanced at night, consistent with domestic heating activities. Considering the full campaign time series (Fig. 2a), this repeating pattern, opposed to solar radiation, is evident for  $\text{LABB1}_{\text{EESI}}$ , while the time series of  $\text{LABB2}_{\text{EESI}}$  is driven by intense events ( $\sim 6.5$  times higher than  $\text{LABB1}_{\text{EESI}}$ ) during two nights: from 18:00 on 27 January to 08:00 on 28 January, and from 18:00 on 28 January to 08:00 on 29 January. As shown in Fig. 2b, both factor profiles are dominated by  $\text{C}_6\text{H}_{10}\text{O}_5$  and  $\text{C}_8\text{H}_{12}\text{O}_6$ .  $\text{C}_6\text{H}_{10}\text{O}_5$  is attributed primarily to levoglucosan, which is a well-established tracer for biomass burning. The mass spectrum features of both factors are very similar to less aged biomass burning emissions measured directly from a

domestic biomass combustion appliance in the PSI smog chamber (Bertrand et al., submitted). Figures 4a and 4b show Van Krevelen plots (i.e., atomic ratios H:C as a function of O:C) for LABB1<sub>EESI</sub> and LABB2<sub>EESI</sub>, respectively, with points colored by the number of atoms and sized by the fraction of each ion apportioned to the respective factor. Both LABB1<sub>EESI</sub> and LABB2<sub>EESI</sub> are dominated by ions with low H:C (1.04) and low O:C (0.35, excluding the sugars C<sub>6</sub>H<sub>10</sub>O<sub>5</sub> and C<sub>8</sub>H<sub>12</sub>O<sub>6</sub>, which exhibit high variability, Table S1), suggesting a strong contribution from primary or slightly aged aromatics. The wind regression analysis of these two factors are shown in Fig. S7. LABB1<sub>EESI</sub> does not correspond to a specific wind direction, in contrast, LABB2<sub>EESI</sub> originates predominantly from a single wind direction, excluding the smaller source to the SE on the third day. Fig. S8 compares the BBOA<sub>AMS</sub> factor (Fig. 2a) with LABB1<sub>EESI</sub>, LABB2<sub>EESI</sub>, and the sum of LABB1<sub>EESI</sub> + LABB2<sub>EESI</sub>, with R 0.59, 0.79, and 0.82, respectively. The correlation is generally good except during the first part of the campaign (25 January to 27 January) which as discussed later relates to the complexity of wood burning classification between the EESI-TOF and AMS. The correlation of BBOA<sub>AMS</sub> with either LABB2<sub>EESI</sub> or LABB1<sub>EESI</sub>+LABB2<sub>EESI</sub> is rather high at night (R=0.59 to 0.82), while the LABB<sub>EESI</sub> factors are consistently lower than BBOA<sub>AMS</sub> during the day. We assign the high correlation of LABB2<sub>EESI</sub> with BBOA<sub>AMS</sub> to the high abundance of levoglucosan which drives the variation in *f*<sub>60</sub> in the AMS. Some specific features of BBOA<sub>AMS</sub> do not appear in any LABB factor because less aged and more aged biomass burning OA are not unambiguously separated in the AMS.

### ***Cooking-related OA (COA<sub>EESI</sub>)***

The COA<sub>EESI</sub> and COA<sub>AMS</sub> factor time series are strongly correlated (R=0.88), as shown in Fig. 2a. The diurnal variation of the COA<sub>EESI</sub> is also similar to COA<sub>AMS</sub>, with strong peaks at lunch and dinner time (Fig. 3a). In addition to this diurnal pattern, both COA<sub>EESI</sub> and COA<sub>AMS</sub> are significantly elevated during two periods: from 18:00 on 27 January to 01:00 on 28 January (Friday night), and from 18:00 on 28 January to 01:00 on 29 January (Saturday night). These periods occur on the same evening as the unknown special event giving rise to the EVENT<sub>AMS</sub> and EVENT<sub>EESI</sub> factors, but are slightly offset in time, with the COA factors peaking approximately four hours earlier. The distinct contribution from the COA<sub>EESI</sub> factor is due in part to the location of several restaurants within a 100-m radius, including one adjacent to the site.

As shown in Fig. 2b, the COA<sub>EESI</sub> mass spectrum is unique in having most of the mass at ions with higher *m/z*. Several of the dominant ions can be attributed to fatty acids and alcohols, which are associated with cooking emissions and oils. For example, C<sub>13</sub>H<sub>22</sub>O<sub>4</sub> (dibutyl itaconate), C<sub>16</sub>H<sub>30</sub>O<sub>3</sub> (2-oxo-tetradecanoic acid), and C<sub>18</sub>H<sub>34</sub>O<sub>3</sub> (ricinoleic acid), are prominent, and contributed 0.89 %, 1.7 %, and 2.0 %, respectively, of the total mass spectrum. Figure 3b shows a Van Krevelen plot of the COA<sub>EESI</sub> factor mass spectrum, with points sized by the fraction of each ion apportioned to COA<sub>EESI</sub> and colored by the number of carbon atoms. The dominant contribution of ions with higher carbon number (C13-C25) and high H:C ratio (greater than 1.5) but low O:C ratio (below 0.2) indicates that these ions are more consistent with fatty acids or alcohols rather than aromatic-derived ions.

### ***Special event (Event<sub>EESI</sub>)***

The time series of  $\text{EVENT}_{\text{EESI}}$  is highly correlated with  $\text{EVENT}_{\text{AMS}}$  ( $R=0.99$ , Fig. 2a). Both factors are near-zero except for two intense events beginning at approximately midnight and lasting till the early morning on 28 and 29 January, supporting the hypothesis of a unique event as opposed to variation in BBOA. The Zurich game festival was taking place at the weekend (the event is apparently held in a building on the SW side of the courtyard), though no human activities in the immediate vicinity of the sampling inlet were evident by inspection of the on-site camera. The EESI-TOF factor mass spectrum is dominated by an ion at  $m/z$  174.08, tentatively assigned to  $\text{C}_8\text{H}_{11}\text{N}_2\text{O}$ . However, the EESI-TOF does not provide structural information and to our knowledge no compound with this formula has been reported as a major constituent of an atmospheric emission source, preventing its use as a diagnostic tracer. Other significant ions are  $\text{C}_8\text{H}_{12}\text{O}_4$  and  $\text{C}_8\text{H}_{18}\text{O}_5$ . The  $\text{C}_8\text{H}_{12}\text{O}_4$  ion likely represents 1,2-cyclohexane dicarboxylic acid diisononyl ester, a plasticizer for the manufacture of food packaging, belonging to the group of aliphatic esters from a chemical point of view. This indicates that the source may be from food plastic burning in a nearby restaurant.

### ***Cigarette smoke-influenced OA ( $\text{CS-OA}_{\text{EESI}}$ )***

Cigarette smoke-influenced OA ( $\text{CS-OA}_{\text{EESI}}$ ) is a constrained factor, based on a reference profile retrieved from higher-order PMF solutions as described in Sect. 3.2.1. The mass spectrum of  $\text{CS-OA}_{\text{EESI}}$  is dominated by the  $\text{C}_{10}\text{H}_{14}\text{N}_2\text{H}^+$  ion (Fig. 2b). This ion is the only ion (out of 892 ions) that does not appear as an adduct with  $\text{Na}^+$ . Instead, the observed molecular formula corresponds to that of nicotine with an extra hydrogen. As a reduced nitrogen compound, nicotine likely forms a stable ion by abstracting a hydrogen from water, leading to the observed cation. However, the time series and the mass flux of this ion should be interpreted with caution, because it is formed by a different ionization pathway than the majority of the spectrum, its relative sensitivity may be significantly different from that of the other ions. Additionally, we have not characterized such non-Na-adducts in terms of ion suppression or matrix effects and cannot rule out a nonlinear response to mass. However, the comparison of the  $\text{CS-OA}_{\text{EESI}}$  factor with AMS PMF results and individual ions discussed below suggest that such nonlinear effects are not significant.

Oxidized organic nitrogen species such as  $\text{C}_x\text{H}_y\text{N}_1\text{O}_z$  (34.9 %) and  $\text{C}_x\text{H}_y\text{N}_2\text{O}_z$  (6.8 %) are also significant in the  $\text{CS-OA}_{\text{EESI}}$  factor, as shown in Fig. 2b and Fig. 5a.  $\text{CS-OA}_{\text{EESI}}$  is only slightly oxygenated ( $\text{O:C}=0.31$ ) and has an H:C ratio of approximately 1.51 (Table S1). The  $\text{CS-OA}_{\text{EESI}}$  time series exhibits two large evening peaks (27 Jan and 28 Jan). These peaks are likely associated with cigarette smoking outside the nearby restaurants. A high correlation is observed between the time series of  $\text{CS-OA}_{\text{EESI}}$  and the AMS  $\text{C}_5\text{H}_{10}\text{N}^+$  ion ( $R=0.91$ , Fig. 5b), which has been proposed as a tracer for nicotine (Struckmeier et al., 2016).

### **3.2.3 EESI-TOF Factors: Secondary Organic Aerosols**

Here we discuss the EESI-TOF SOA factors in three groups: (1) more aged wood-burning related ( $\text{MABB\_LOW}_{\text{EESI}}$ ,  $\text{MABB\_TRANS}_{\text{EESI}}$ , and  $\text{MABB\_HIGH}_{\text{EESI}}$ ); (2) non-source-specific SOA ( $\text{SOA1}_{\text{EESI}}$  and  $\text{SOA2}_{\text{EESI}}$ ); and (3) high nitrogen content ( $\text{NSOA}_{\text{EESI}}$ ). Factor mass spectra for these factors are shown in Fig. 6a, with the spectra colored by the number of N atoms and normalized such that the sum of the peaks in each spectrum is 1. Figure 6b shows a stacked time series of all 6

EESI-TOF SOA factors, such that the sum of the stacked plot represents the total EESI-TOF mass flux attributed to SOA. For comparison, the time series of the estimated AMS SOA is shown, calculated as  $LO\text{-}OOA_{AMS} + MO\text{-}OOA_{AMS}$ .  $NOA_{AMS}$  is excluded from this calculation due to the contribution from primary cigarette smoke discussed above. The total EESI-TOF SOA and AMS SOA estimates are in general well-correlated ( $R=0.90$ ), even though the EESI-TOF mass flux is proportionally lower during the first few days of the study.

***More aged biomass burning-related factors ( $MABB\_LOW_{EESI}$ ,  $MABB\_TRANS_{EESI}$ , and  $MABB\_HIGH_{EESI}$ )***

Three more aged biomass burning (MABB) factors are identified in this study:  $MABB\_LOW_{EESI}$ ,  $MABB\_TRANS_{EESI}$ , and  $MABB\_HIGH_{EESI}$ . Each  $MABB_{EESI}$  factor is enhanced relative to the others during a different part of the campaign, which correspond to both changes in the daily temperature cycle and wind direction. As shown in Fig. 6b, the coldest part of the study, period 1, occurs from 25 to 27 January (mean  $-5.4^\circ$ , min  $-6.4^\circ$ , max  $-2.2^\circ$ ). During this period,  $MABB\_LOW_{EESI}$  contributes 84 % of the total MABB ( $MABB\_LOW_{EESI}/(MABB\_LOW_{EESI} + MABB\_TRANS_{EESI} + MABB\_HIGH_{EESI})$ ). From 27 to 29 January, Period 2, temperature increases (mean  $1.4^\circ$ , min  $-2.2^\circ$ , max  $7.4^\circ$ ), and the  $MABB\_TRANS_{EESI}$  factor constitutes the dominant  $MABB_{EESI}$  fraction (65 %). Period 3, from 29 January to the campaign end on 4 February, corresponds to higher temperatures (mean  $5.7^\circ$ , min  $0.8^\circ$ , max  $8.7^\circ$ ), and the  $MABB_{EESI}$  fraction is dominated by  $MABB\_HIGH_{EESI}$  (90 %) until a substantial precipitation event beginning on 31 January, after which relatively clean air is observed for the remainder of the campaign. Figure 7 shows the source-specific wind sectors determined by SWIM (see Sect. 2.4) for the three MABB factors. This analysis assigns the three factors to distinct wind vectors: NNE for  $MABB\_LOW_{EESI}$ , NNW for  $MABB\_TRANS_{EESI}$ , and SE for  $MABB\_HIGH_{EESI}$ . Because each factor is predominantly observed during a single time period, it is difficult to assess the relative importance of temperature vs. source region for these three factors.

As shown in Fig. 6a, all three  $MABB_{EESI}$  factor mass spectra are qualitatively similar, with many of the same ions enhanced. These spectra are also similar to the mass spectrum of aged biomass burning emissions retrieved from smog chamber experiment (Bertrand et al., submitted). For both the  $MABB_{EESI}$  and chamber spectra, the major ions,  $C_7H_{10}O_5$ ,  $C_9H_{14}O_4$ ,  $C_8H_{12}O_6$ , are in common. The main difference between the EESI-TOF factors and the chamber mass spectrum is that the chamber data show a higher fraction of signal at lower  $m/z$ . This is likely due to the higher concentrations used during the chamber experiments, causing increased partitioning of semi-volatile compounds to the particle phase.  $MABB\_LOW_{EESI}$  also exhibits somehow enhanced intensities at lower  $m/z$  compared to the other  $MABB_{EESI}$  factors. As  $MABB\_LOW_{EESI}$  is dominant during the coldest period 1, the  $MABB\_LOW_{EESI}$  factor is possibly separated from the other  $MABB_{EESI}$  factors due to partitioning of semi-volatile material to the particle phase due to colder temperatures.

Further insight into the composition trends across the  $MABB_{EESI}$  factors is obtained through Fig. 8 which represents the three  $MABB_{EESI}$  mass spectra as the carbon oxidation state ( $OS_c$ ) (Kroll et al., 2011) of each ion as a function of the carbon number ( $n_c$ ). Data points are colored by the H:C ratio and sized by the fraction of each ion apportioned to the designated factor. The figure shows that  $MABB\_LOW_{EESI}$  is enhanced in ions with low  $n_c$ , consistent with condensation of semi-volatile OA ( $C_5H_6O_4$ ,  $C_8H_6O_4$ ,  $C_5H_8O_7$ ) at low temperature. Otherwise, all three  $MABB_{EESI}$  factors are rather similar. Figure 8 also shows the  $OS_c$  of non- $MABB_{EESI}$  (weighted average of  $SOA1_{EESI} + SOA2_{EESI}$ ) and  $LABB_{EESI}$  (weighted average of  $LABB1_{EESI}$  and

LABB<sub>2EESI</sub>) factors. Obviously, the non-MABB<sub>EESI</sub> and LABB<sub>EESI</sub> factors are less oxidized than the MABB factors, with lower OS<sub>c</sub>.

### **Other SOA factors (SOA<sub>1EESI</sub> and SOA<sub>2EESI</sub>)**

The mass spectra of SOA<sub>1EESI</sub> and SOA<sub>2EESI</sub> are qualitatively similar to factors retrieved from PMF analysis of EESI-TOF data from Zurich during summer, when monoterpenes are the dominant SOA precursors (Stefenelli et al, 2019, Fig. S9). Major ions include C<sub>8</sub>H<sub>12</sub>O<sub>4</sub>, C<sub>9</sub>H<sub>14</sub>O<sub>4</sub>, C<sub>10</sub>H<sub>16</sub>O<sub>4</sub>, C<sub>10</sub>H<sub>18</sub>O<sub>4</sub>, C<sub>10</sub>H<sub>16</sub>O<sub>5</sub> and C<sub>10</sub>H<sub>16</sub>O<sub>2</sub>, C<sub>10</sub>H<sub>16</sub>O<sub>3</sub>, C<sub>10</sub>H<sub>18</sub>O<sub>4</sub>, separately. In contrast to the MABB<sub>EESI</sub> factors, the SOA<sub>EESI</sub> factors have a negligible contribution from levoglucosan (C<sub>6</sub>H<sub>10</sub>O<sub>5</sub>). Approximately 57 % of the total C10 ion signal is apportioned to the SOA<sub>EESI</sub> factors. Figures 9a and 9b show the atomic ratio of H:C as a function of O:C for the two SOA<sub>EESI</sub> factors. These H:C ratios are higher than typically observed from the oxidation of aromatic emissions and are instead consistent with monoterpene oxidation. The Van Krevelen plots show clear differences between these two factors, SOA<sub>2EESI</sub> is less oxygenated than SOA<sub>1EESI</sub> with lower O:C ratio and lower H:C ratio. The time series of SOA<sub>1EESI</sub> shows a higher contribution during the period 1, 2, while SOA<sub>2EESI</sub> has a more regular cycle contribution during daytime (Fig. 6b). Since we have clear evidence that these EESI-retrieved factors are related to secondary organic aerosol we call them SOA<sub>EESI</sub>, in contrast to the OOA<sub>AMS</sub> factors, where this evidence is less clear. The more detailed comparison between EESI-TOF\_SOA factors and AMS\_OOA factors will be discussed in Sect. 3.5.

### **Nitrogen-containing SOA factor (NSOA<sub>EESI</sub>)**

As mentioned in Sect. 3.1, the EESI-TOF source apportionment also resolves a nitrogen-containing SOA factor (NSOA<sub>EESI</sub>). NSOA<sub>EESI</sub> is dominated by highly oxygenated organonitrate molecules, including C<sub>8</sub>H<sub>13</sub>NO<sub>5</sub>, C<sub>10</sub>H<sub>15</sub>NO<sub>6</sub>, C<sub>10</sub>H<sub>19</sub>NO<sub>8</sub>. Ions like C<sub>6</sub>H<sub>10</sub>O<sub>5</sub>, C<sub>10</sub>H<sub>16</sub>O<sub>2</sub> and C<sub>8</sub>H<sub>12</sub>O<sub>6</sub> are comprising another fraction of the NSOA<sub>EESI</sub> signal, but are not unique to the NSOA<sub>EESI</sub> factor and rather spread over many other factors. The significant contribution of organonitrates results in an N:C ratio (0.04) and suggests a secondary origin for this factor. Therefore, we call it NSOA<sub>EESI</sub>, in contrast to NOA<sub>AMS</sub> for which the primary/secondary origin is less certain. The time series of the factor is quite unique, shows maximum mass flux at the end of this campaign with the highest peak at night (3-4 February), and a smaller peak during the night of 28 to 29 February.

Figure S10 shows a comparison of the NSOA<sub>EESI</sub> time series and CS-OA<sub>EESI</sub> time series with the CHON ions from the EESI and CHN ions from the AMS, respectively. The group of EESI\_CHON ions shows the same temporal variation as the NSOA<sub>EESI</sub> factor (Fig. S10) while the AMS\_CHN group is more correlated to the primary organic group.

## **3.3 Analysis of marker ions**

Laboratory, as well as offline and semi-continuous field studies have identified a number of tracer molecules that are useful for the investigation of primary and secondary OA from various sources, including biomass burning. The real-time and in situ measurement of these compounds is a novel feature of the EESI-TOF, and their apportionment gives further insight into the nature of the factors described above. Here we investigate the apportionment of eight ions associated with compounds of interest: C<sub>6</sub>H<sub>10</sub>O<sub>5</sub> (approximately assigned to levoglucosan), C<sub>7</sub>H<sub>7</sub>NO<sub>4</sub> (methyl-nitrocatechol), C<sub>9</sub>H<sub>10</sub>O<sub>5</sub> (syringic acid), C<sub>8</sub>H<sub>8</sub>O<sub>4</sub> (vanillic acid), C<sub>8</sub>H<sub>6</sub>O<sub>4</sub> (phthalic acid), C<sub>5</sub>H<sub>6</sub>O<sub>4</sub> (glutaconic acid), C<sub>7</sub>H<sub>8</sub>O<sub>4</sub> (tetrahydroxy toluene) and C<sub>7</sub>H<sub>10</sub>O<sub>5</sub>



(pentahydroxy toluene). Note that because the EESI-TOF can provide only a molecular formula, we cannot establish for certain the identity of a compound or assess the relative isomeric abundances. For example,  $C_6H_{10}O_5$  is likely to consist not only of levoglucosan, but also other sugars such as mannosan and galactosan. The named compounds are thus provided for reference, but their identification should not be considered as conclusive and the ions cannot be assumed to be isomerically pure.

5 Nevertheless, as these assignments are based on molecular investigations of wood burning-related emissions they are likely to be qualitatively correct and provide a useful framework for interpreting molecular aspects of the source apportionment results.

Figure 10a shows a stacked time series of the mass flux of these compounds representing the contribution of each EESI-TOF PMF factor to the total mass flux (assuming no significant conformational isomers). Levoglucosan, which is derived from the pyrolysis of cellulose and hemicellulose, is commonly used as an indicator for the presence of primary aerosols originating from biomass combustion (Fine et al., 2001). Figure 10b shows that levoglucosan appears in both POA (total contribution, 62 %, mostly from  $LABB1_{EESI}$  (22 %) and  $LABB2_{EESI}$  (37 %), and minor contributions by  $COA_{EESI}$ ,  $CS-OA_{EESI}$ , and  $EVENT_{EESI}$ ) and SOA (total contribution, 38 %, of which 36 % related to the sum of  $MABB\_HIGH_{EESI}$ ,  $MABB\_TRANS_{EESI}$ ,  $MABB\_LOW_{EESI}$ , plus minor contributions from  $NSOA_{EESI}$ ). Due to the high biomass burning emission background and the lifetime of levoglucosan, it is inevitable to find a contribution of levoglucosan in the MABB factor, which is consistent with our aged biomass burning discussion above. In contrast, nitrocatechol ( $C_7H_7NO_4$ ) has been established as a secondary species originating from the oxidation of biomass burning (Inuma et al., 2010; Finewax et al., 2018). Here 86 % of nitrocatechol is apportioned to the less aged (49 %) and more aged (37 %) biomass burning factors. Syringic acid and vanillic acid are phenolic acids derived from the oxidation of lignin decomposition products (He et al., 2018), which in turn are a major component of biomass combustion emissions, and are apportioned primarily to the  $MABB_{EESI}$  factors (52 % for syringic acid and 66 % for vanillic acid).

20 Phthalic acid ( $C_8H_6O_4$ ) and glutaconic acid ( $C_5H_6O_4$ ) are apportioned to the SOA factors (91 % and 94 % in total, respectively), with main contributions from the  $MABB_{EESI}$  factors and in particular the  $MABB\_LOW_{EESI}$  factor (53 % and 59 %, respectively). These dicarboxylic acids are ubiquitous water-soluble organic compounds which have been detected in a variety of aerosol samples, and originate from the combustion of biomass burning and fossil fuels, as well as from biogenic emission and photo-oxidation of organic gases. For example, phthalic acid has been identified based on field measurements, as a tracer of naphthalene oxidation (Kleindienst et al., 2012) or oxidation products from PAHs (Chan et al., 2009), and is also consistently found in combustion products of lignin, which is likely to explain the contribution in the MABB factors (Fu et al., 2010; Wang et al., 2007).

25 Tetrahydroxy toluene ( $C_7H_8O_4$ ) and pentahydroxy toluene ( $C_7H_8O_5$ ) are apportioned mainly to secondary factors (85 % and 78 %, respectively). Tetrahydroxy toluene and pentahydroxy toluene have been detected as dominant products both in the particle phase and gas phase under low-NO oxidation of toluene (Nakao et al., 2012; Schwantes et al., 2017). The *o*-cresol oxidation mechanism for tetrahydroxy toluene and pentahydroxy toluene is found in MCM v3.3.1, based on Olariu et al. (2002). This formation indicates that these two low-volatility ions are indeed secondary organic compounds, consistent with our results shown in Fig. 10. In addition, the temporal variation of the pentahydroxy toluene contribution is consistent with the one of

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tetrahydroxy toluene except for the  $\text{EVENT}_{\text{EESI}}$  factor, which may indicate that during this night event an isomer of pentahydroxy toluene was present.

### 3.4 EESI-TOF cluster analysis

As evidenced from the previous section and Figs. 2 and 6, many of the dominant ions in the EESI-TOF PMF analysis are shared by multiple factors. Here, we utilize a cluster analysis to identify ions unique or nearly unique to a single factor or group of factors. As discussed in Sect. 2.5, hierarchical agglomerative clustering is performed separately on the set of all EESI-TOF ions and **all EESI-TOF factor time series**. Figure 11 shows the resulting dendrogram of the ions and factors along the vertical and horizontal axes, respectively; the ion dendrogram is colored subjectively to guide the eye. Comparison of the ions to the factors yields a matrix, also shown in Fig. 11, which is colored by the z-score, with brown colors denoting high correlation. In this representation, an ion unique to a given factor is brown for one and only one rectangle in the horizontal dimension.

The factor dendrogram identifies several groups of EESI-TOF PMF factors consistent with the interpretations provided above: (1) more aged biomass burning factors ( $\text{MABB\_LOW}_{\text{EESI}}$ ,  $\text{MABB\_TRANS}_{\text{EESI}}$  and  $\text{MABB\_HIGH}_{\text{EESI}}$ ), (2) less aged biomass burning factors ( $\text{LABB1}_{\text{EESI}}$  and  $\text{LABB2}_{\text{EESI}}$ ), and (3) the cooking-related OA and cigarette smoking OA factors. The more aged and less aged biomass burning factor groups are themselves likewise grouped. This clustering is consistent with our interpretation of these factors, as discussed in the previous section. Ions are clustered to different groups using the standardized values. In each factor, there are distinguished molecules (lists of the specific ions (standardized value above 1.5) for each factor is shown in Table S2). The other two resolved groups, one group including  $\text{SOA1}$  and  $\text{EVENT}$  factor, one group containing  $\text{SOA2}$  and  $\text{NSOA}$  factor, apparently don't retrieve the common ions, which make less sense for the current study.”

For several of the factors, the uniquely assigned ions exhibit systematic patterns contributing to the identification or deconvolution of the factors. Figure 12a shows the mass defect, defined as the exact  $m/z$  minus the nearest integer  $m/z$ , as a function of  $m/z$  for the uniquely assigned ions for the five  $\text{POA}_{\text{EESI}}$  factors. Figure 12b shows the equivalent plot for the three  $\text{MABB}_{\text{EESI}}$  factors and  $\text{SOA1}_{\text{EESI}}$  ( $\text{SOA2}_{\text{EESI}}$  and  $\text{NSOA}_{\text{EESI}}$  have a high degree of scatter and are omitted to avoid masking trends in the other secondary factors). The displayed factors exhibit linear correlations or tight clusters of points; all factors are shown independently in Fig. S11).  $\text{LABB1}_{\text{EESI}}$  and  $\text{LABB2}_{\text{EESI}}$  have a lower mass defect and shallower slope than  $\text{COA}_{\text{EESI}}$  and  $\text{CS-OA}_{\text{EESI}}$ , consistent with increased aromaticity. The slopes are  $(4.9 \pm 0.4) \cdot 10^{-4}$ ,  $(5.9 \pm 0.6) \cdot 10^{-4}$ ,  $(8 \pm 0.5) \cdot 10^{-4}$  and  $(8 \pm 0.3) \cdot 10^{-4}$  for  $\text{LABB1}_{\text{EESI}}$ ,  $\text{LABB2}_{\text{EESI}}$ ,  $\text{COA}_{\text{EESI}}$  and  $\text{CS-OA}_{\text{EESI}}$ , respectively. The slopes of the two LABB factors as well as those of  $\text{COA}_{\text{EESI}}$  and  $\text{CS-OA}_{\text{EESI}}$  are very similar to each other and have a high possibility to be consistent with CH addition for the former (i.e.  $\text{C}_{10+x}\text{H}_{14+x}\text{O}_{4-5}$ , theoretical slope  $6 \cdot 10^{-4}$ ), and  $\text{CH}_2$  addition for the latter (i.e.  $\text{C}_{10+x}\text{H}_{20+2x}\text{O}_{3-5}$  for  $\text{COA}_{\text{EESI}}$  and  $\text{C}_{10+y}\text{H}_{15+2y}\text{NO}_{3-5}$  for  $\text{CS-OA}_{\text{EESI}}$  as nearly every  $\text{CS-OA}$ -specific ion contains a single N atom, theoretical slope  $1.1 \cdot 10^{-3}$ ).

The MABB and LABB factors have similar slopes, despite different ion lists. The slopes of two MABB factors ( $0.9 \cdot 10^{-3}$ ), as shown in Fig. 12b, are consistent with the addition of CHO functionality (theoretical slope =  $0.1 \cdot 10^{-2}$ ). Due to the high

variability of the slopes of the MABB factors, it may also contain the other potential possibility for the added functionalities.

Both mass defect and slope are higher for MABB\_LOW<sub>EESI</sub> than for MABB\_HIGH<sub>EESI</sub>, which is consistent with our discussion in Sect. 3.2.3, assuming that the organics of the MABB\_LOW<sub>EESI</sub> factor are more oxidized than those of the MABB\_HIGH<sub>EESI</sub> factors. In addition, the MABB intercepts are more positive than those of LABB, consistent with the higher oxidation state shown above.

### 3.5 Comparison of AMS and EESI-TOF

Fig. 13a shows the sum of the mass flux of the ions measured by the EESI-TOF as a function of the OA concentration measured by the AMS, with the points colored by date and time. We apply no ion-dependent sensitivity corrections for the EESI-TOF, although ion-by-ion differences are known to exist (Lopez-Hilfiker et al., 2019). Note that the AMS signal includes the minor OA source, HOA<sub>AMS</sub>, which is mostly insoluble in the electrospray droplets and thus expected to be basically undetectable by the EESI-TOF. Nevertheless, the two instruments are well-correlated ( $R=0.94$ ). The strong correlation in Fig. 13a suggests that the overall EESI-TOF sensitivity to OA does not vary significantly throughout the study, and therefore it is unlikely that the major individual EESI-TOF PMF factors (which describe the compositional variability) have dramatically different response factors. We therefore interpret the EESI-TOF PMF results without correction of the data for factor-specific sensitivities. Several features are evident from dependence of the sensitivity on the mass flux of levoglucosan (Fig. 13b) which may explain the discrepancy in the first part of the campaign (period 1) vs. the rest of the campaign. An SOA-dominated period with low levoglucosan concentration (red line) toward the beginning of the campaign exhibits a lower sensitivity than during a period with higher levoglucosan concentrations (black line), which includes the events on 28.01.2017 and 29.01.2017 characteristic of EVENT<sub>EESI</sub> (Lopez-Hilfiker et al., 2019). Figures 13c and 13d respectively show the O:C and H:C atomic ratios for the EESI-TOF as a function of those for the AMS. Here again no ion-dependent sensitivity corrections are applied. The EESI-TOF and AMS O:C ratios are correlated ( $R=0.62$ ), however, the O:C ratios estimated by the EESI-TOF are systematically higher than those measured by the AMS. For H:C ratios, we do not observe a correlation. The EESI-TOF values are scattered around approximately 1.56, independent of the AMS H:C ratios which vary between 1.11 and 1.44. The cause for this discrepancy is not yet understood but may be related to differences in ion relative sensitivity (Bertrand et al., submitted). Fig. 14 shows the stacked time series of the EESI-TOF PMF factors (together with total AMS OA concentration) and of the AMS PMF factors. Also shown are pie charts denoting the mean OA PMF composition over the entire campaign from the EESI-TOF and AMS data. Despite uncertainties in the definition and resolution of primary vs. more aged biomass burning, the AMS and EESI-TOF are in relatively good agreement with respect to the total POA and SOA fractions. The SOA factors comprise 58.8 % of the mass flux for the EESI-TOF and 69.4 % of the mass for the AMS. The agreement may in fact be better than these values indicate: as noted above the NOA<sub>AMS</sub> factor, comprising 17.9 % of the mass and fully associated to SOA in our solution, is likely composed of both POA (derived from cigarette smoke, as resolved in CS-OA<sub>EESI</sub>) and SOA (from organonitrate-containing SOA, as resolved in NSOA<sub>EESI</sub>), resulting in a low total POA fraction in the AMS solution. Since both CS-OA<sub>EESI</sub> and NSOA<sub>EESI</sub> are enriched with the nitrogen-containing ions, we compare in Fig. 15, the O:C and N:C ratios

for these two factors, where the size of the colored star and circle corresponds to the H:C ratio. A distinct separation between CS-OA<sub>EESI</sub> and NSOA<sub>EESI</sub> is evident due to a significantly higher O:C ratio for a given N:C ratio, i.e., higher degree of oxygenation for the NSOA<sub>EESI</sub> factor and a higher abundance of organic-nitrate molecules in the NSOA<sub>EESI</sub> factor. Moreover, this separation was not possible for AMS PMF.

- 5 Both AMS and EESI-TOF factors stacked time series (Fig. 14) show clearly that biomass burning is dominated by secondary fractions early in the campaign, mixed fractions in the middle of the campaign, and **a primary fractions** late in the campaign. As discussed in Sect 3.2.2, BBOA<sub>AMS</sub> is a mixture of primary and secondary ions, and OOA<sub>AMS</sub> is a mixture of biomass burning fragments and background SOA fragments from photochemistry production ions. Although the fraction of OOA comprised more than 50 % percent of total OA (Fig. 14), it is hard to define how much of AMS OOA is **BB-related** as a function of time.
- 10 The EESI-TOF separates the biomass burning factors into LABB<sub>EESI</sub> and MABB<sub>EESI</sub> and splits the background SOA factors into separate factors, which provides evidence that biomass burning is the single most important contributor to the organic aerosol at the measurement site during winter.

#### 4 Conclusions

- 15 Real-time, near-molecular level measurements of OA composition were performed during winter in Zurich using a novel extractive electrospray ionization time-of-flight mass spectrometer (EESI-TOF). The lack of thermal decomposition or ionization-induced fragmentation in the EESI-TOF provides an improved description of SOA in particular, facilitating SOA source identification by PMF. We retrieve eleven factors, of which 5 are dominated by POA and 6 by SOA. The POA factors included cooking-influenced OA (COA<sub>EESI</sub>, which strongly correlates with an equivalent AMS factor), cigarette smoke-
- 20 influenced OA (CS-OA<sub>EESI</sub>, characterized by a strong contribution from nicotine) and a special event also captured by the AMS. Two less aged biomass burning factors are also resolved. Of the six SOA factors, three are clearly related to biomass burning and are distinguished by temperature and possibly wind direction. We also observe two SOA factors with no clear biomass burning signatures, one of which closely resembles monoterpene oxidation. Finally, we observe a minor factor with a high organonitrate fraction.
- 25 We performed cluster analysis of the EESI-TOF ions followed by correlation with the resolved factors, which identifies groups of ions characteristic of each factor. These characteristic ions represent potential tracers for future studies; they indicate strong aromatic influence in both **less aged** and more aged biomass burning, and support the primary / secondary assignment of biomass burning-influenced factors.

- The increased chemical specificity of the EESI-TOF allows for additional, meaningful factors to be resolved relative to the
- 30 AMS. Comparisons of bulk measurements, as well as of individual factors or groups of factors between the EESI-TOF and AMS indicate good agreement, **but with the differences in elemental ratios**. This suggests that, despite significant uncertainties in the relative response factors of individual ions measured by the EESI-TOF, responses at the level of the PMF factors are

relatively similar, with the main differences resulting from the high sensitivity to levoglucosan in the EESI. Furthermore, source apportionment of EESI-TOF provides more classification of SOA factors, separating EESI biomass burning factors as more/ less aged instead of primary / secondary, identifying organic nitrogen containing factors as primary-dominated nitrogen factor / organonitrate-containing secondary factor, which are not possible for AMS PMF. As a result, the EESI-TOF represents a promising new approach for source apportionment and atmospheric composition studies.

### **Author Contribution.**

LQ was the main author. LQ, GS, VP, YT, and CH conducted the field campaign. MD, XG, JS, AP, and UB were the supervisors. All contributed to the corrections of the paper.

### **Competing interests.**

The authors declare that they have no conflict of interest.

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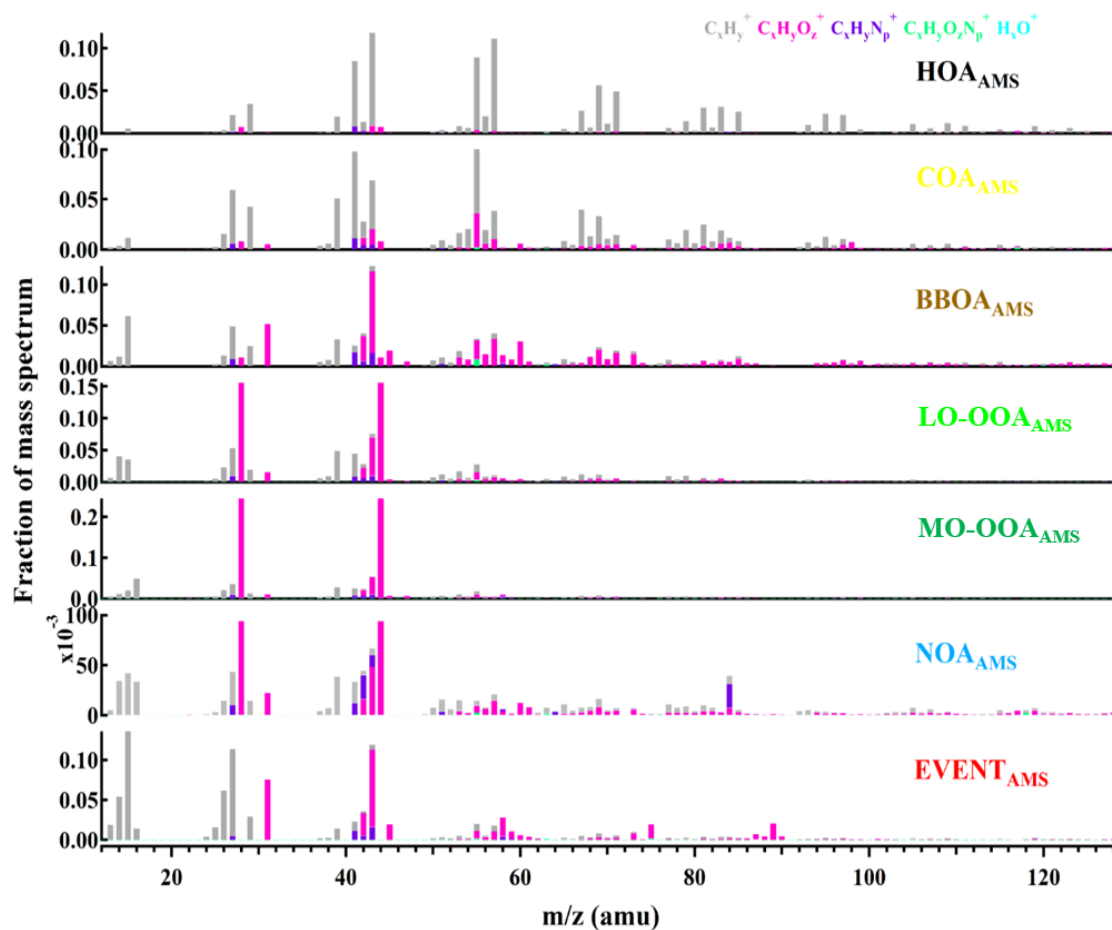
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## Figures



5 Fig. 1. Factor profiles for the 7-factor AMS PMF solution.  $\text{HOA}_{\text{AMS}}$  is constrained by  $a$ -value 0.1. The total signal of each factor is normalized to unity. ( $\text{HOA}_{\text{AMS}}$ : Hydrocarbon OA,  $\text{COA}_{\text{AMS}}$ : Cooking-related OA,  $\text{BBOA}_{\text{AMS}}$ : Biomass burning OA,  $\text{OOA}_{\text{AMS}}$ : Oxygenated OA,  $\text{NOA}_{\text{AMS}}$ : Nitrogen containing OA,  $\text{EVENT}_{\text{AMS}}$ , an isolated local event)

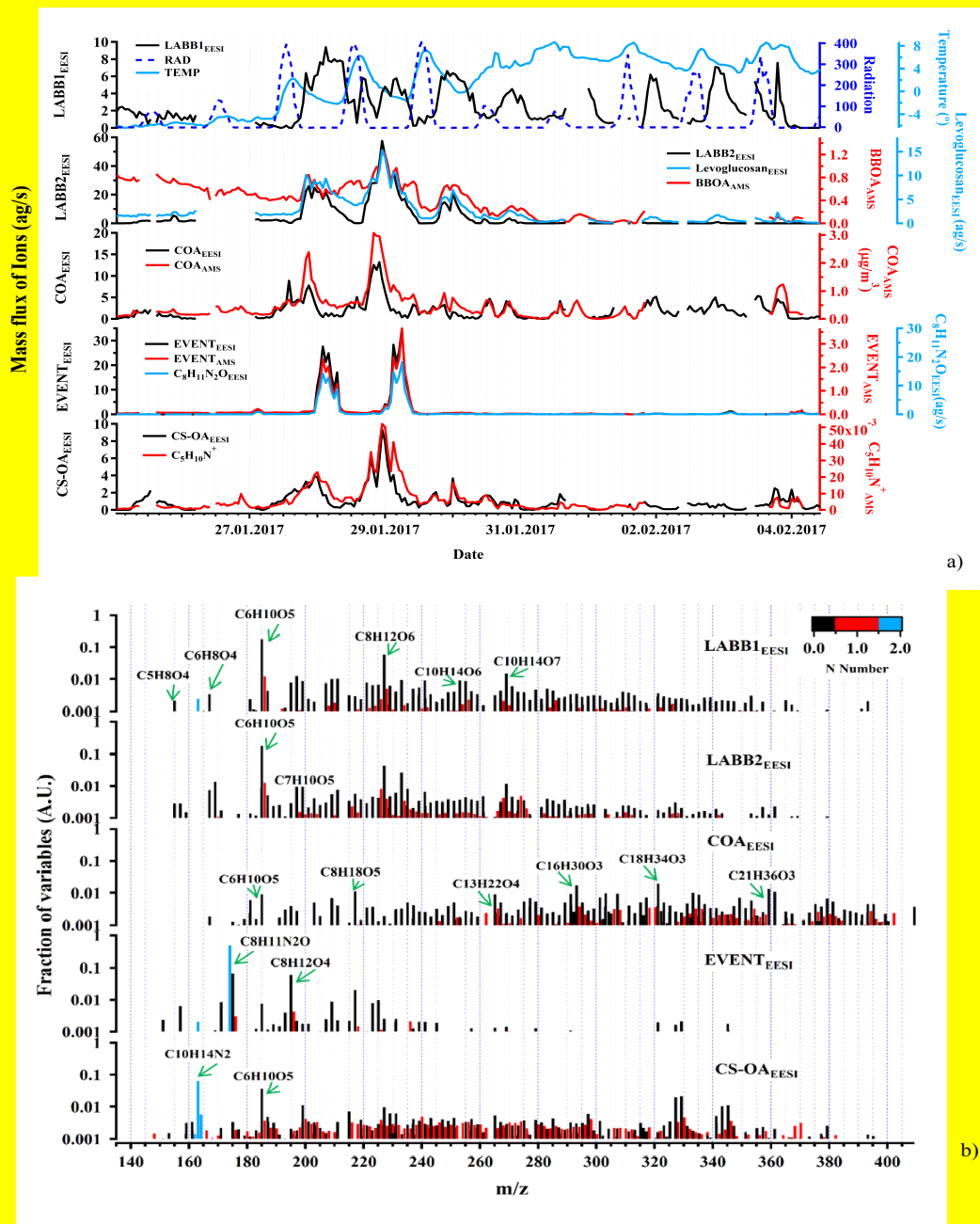
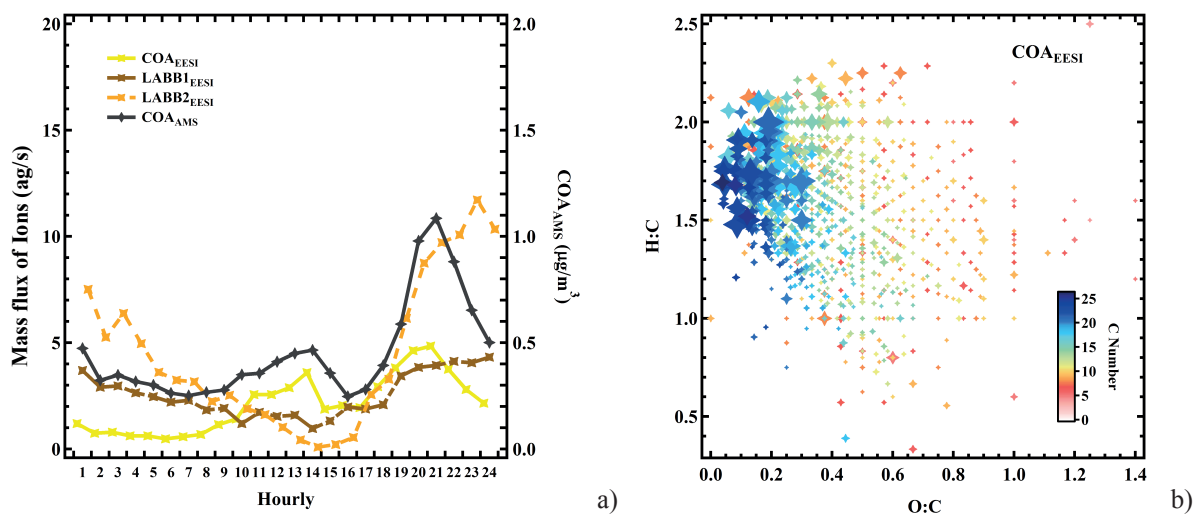
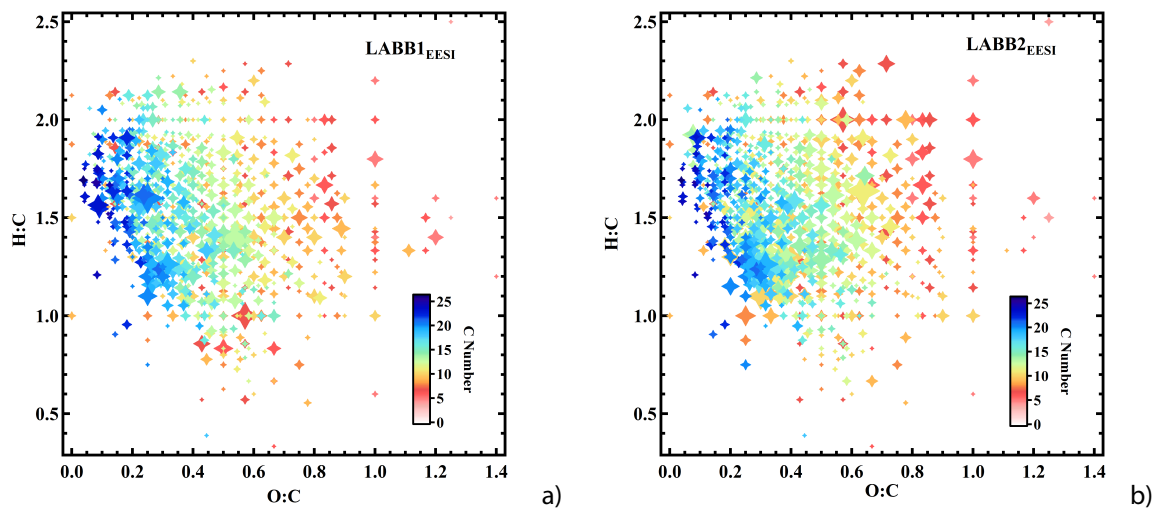


Fig. 2. Time series of the POA factors retrieved from EESI-TOF PMF analysis, along with ancillary data (a), and corresponding factor profiles (b). For all y-axes, EESI-TOF data are shown in mass flux (ag/s), AMS data are shown in  $\mu\text{g m}^{-3}$ , and other units are given. Factor profiles are molecular weighted and are normalized such that the sum of each profile is 1.

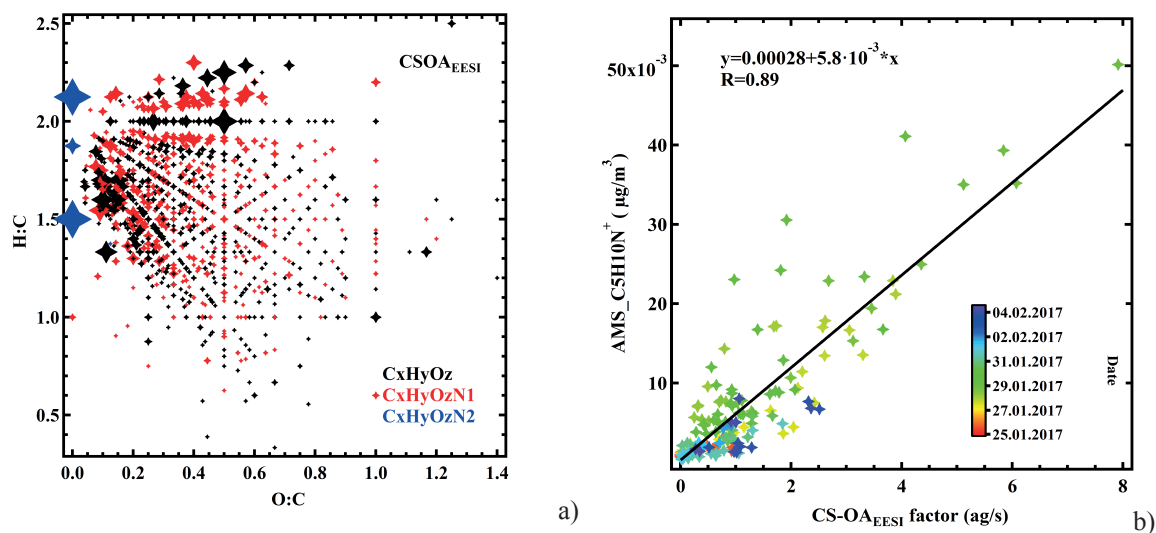


5 Fig. 3. a) Diurnal cycles of the EESI-TOF less aged biomass burning and cooking factors, together with AMS cooking. b) Van Krevelen plot (atomic H:C vs. O:C of each ion) for the COA<sub>EESI</sub> factor mass spectrum, with points sized by the fraction of each ion apportioned to COA<sub>EESI</sub> and colored by number of carbon atoms.



10 Fig. 4. a) b) Van Krevelen plot (atomic H:C vs. O:C ratio) of the LABB1<sub>EESI</sub> and LABB2<sub>EESI</sub> factor mass spectra. Points are sized by the fraction of each ion apportioned to LABB1<sub>EESI</sub> and LABB2<sub>EESI</sub> and colored by number of carbon atoms.





5 Fig. 5. a) Van Krevelen plot (atomic H:C vs. O:C ratio) of the cigarette smoking (CS-OA<sub>EESI</sub>) factor mass spectrum. Points are sized by the fraction of each ion apportioned to CS-OA<sub>EESI</sub>. Colors denote C<sub>x</sub>H<sub>y</sub>O<sub>z</sub>, C<sub>x</sub>H<sub>y</sub>N<sub>1</sub>O<sub>z</sub>, and C<sub>x</sub>H<sub>y</sub>N<sub>2</sub>O<sub>z</sub> groups. b) Comparison of CS-OA<sub>EESI</sub> and C5H10N<sup>+</sup><sub>AMS</sub>, colored by time.

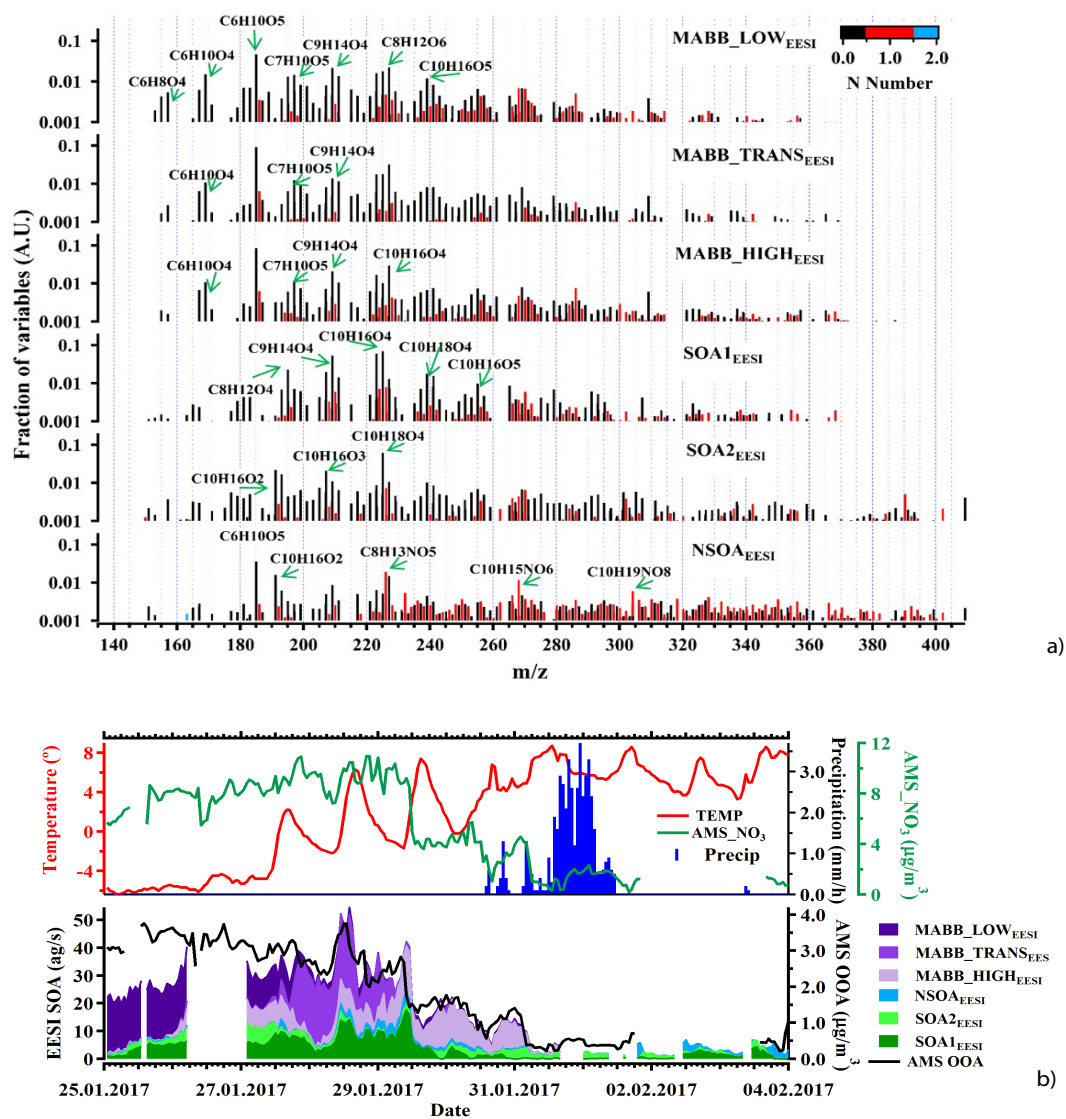
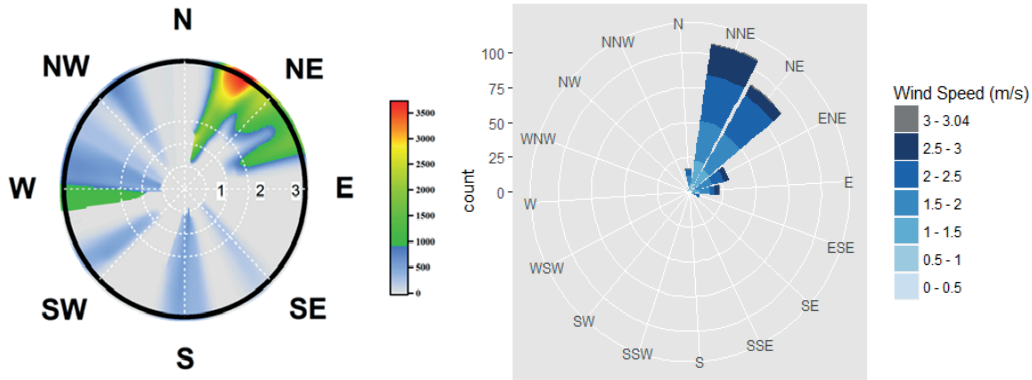
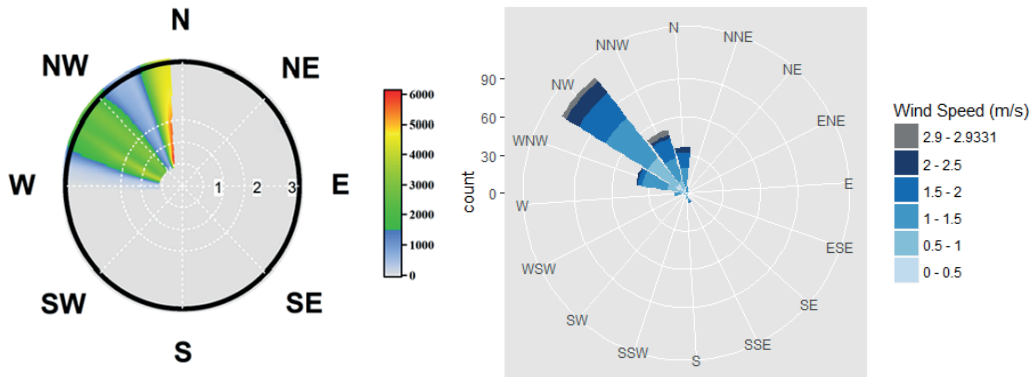


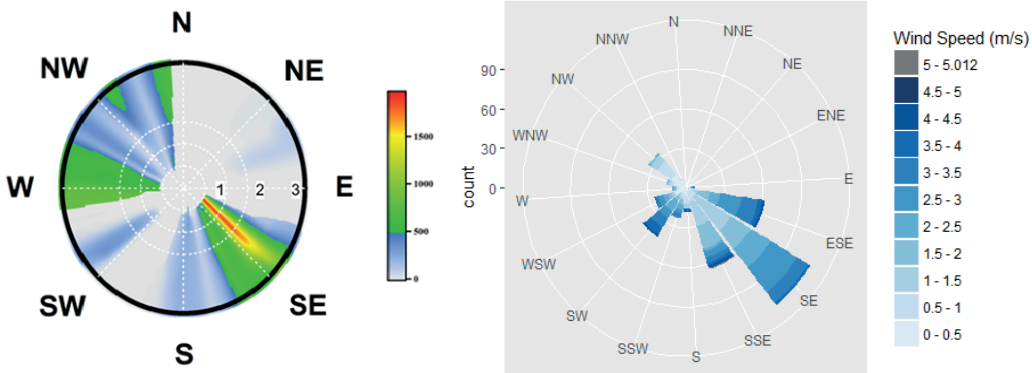
Fig. 6. Factor profiles (a) and stacked time series (b) of the 6 EESI-TOF SOA PMF factors, together with AMS OOA. The latter panel also shows meteorological data. All EESI-TOF data are plotted in mass flux (ag/s), AMS in  $\mu\text{g m}^{-3}$ , other units are included. Factor profiles (b) are molecular weighted and are normalized such that the sum of each profile is 1.



MABB\_LOW<sub>EESI</sub>

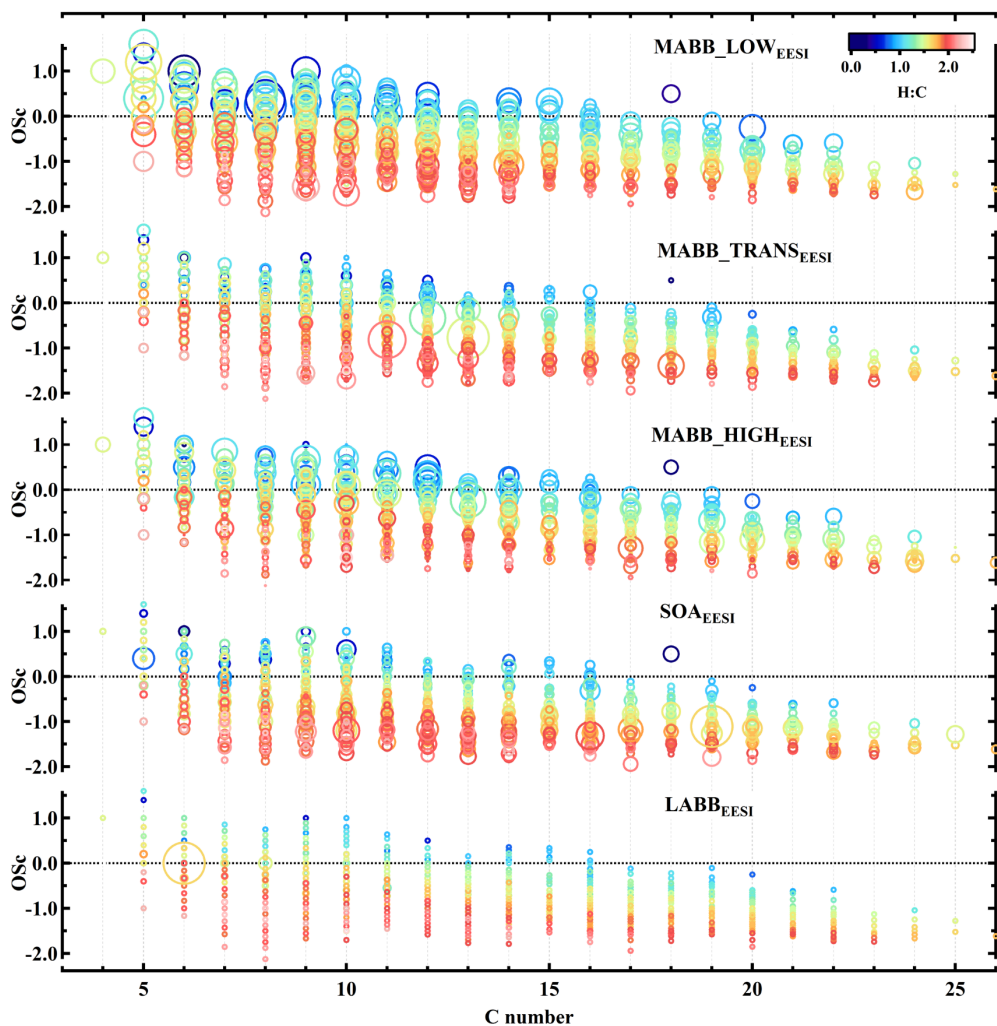


MABB\_TRANS<sub>EESI</sub>

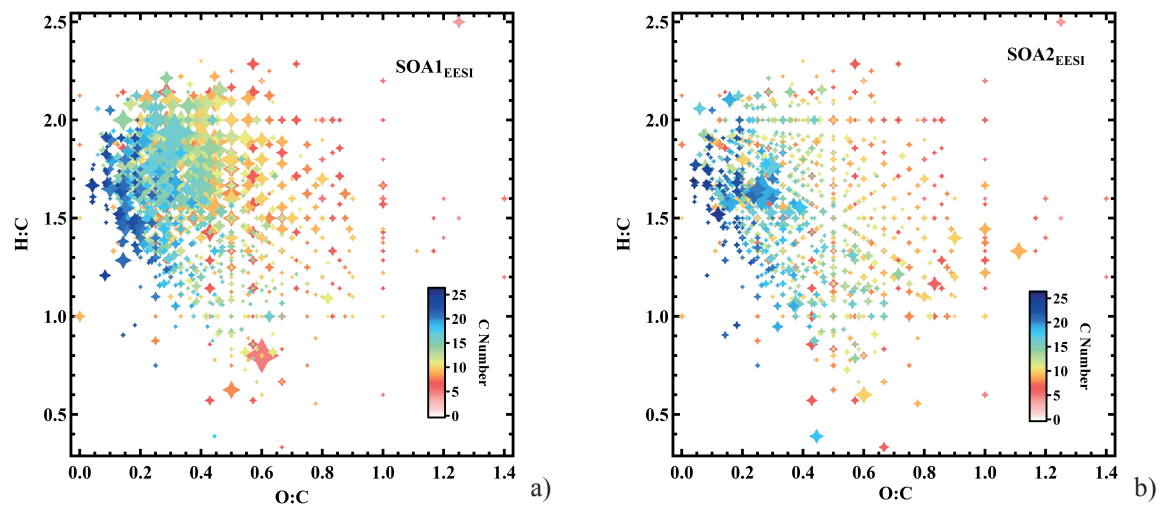


MABB\_HIGH<sub>EESI</sub>

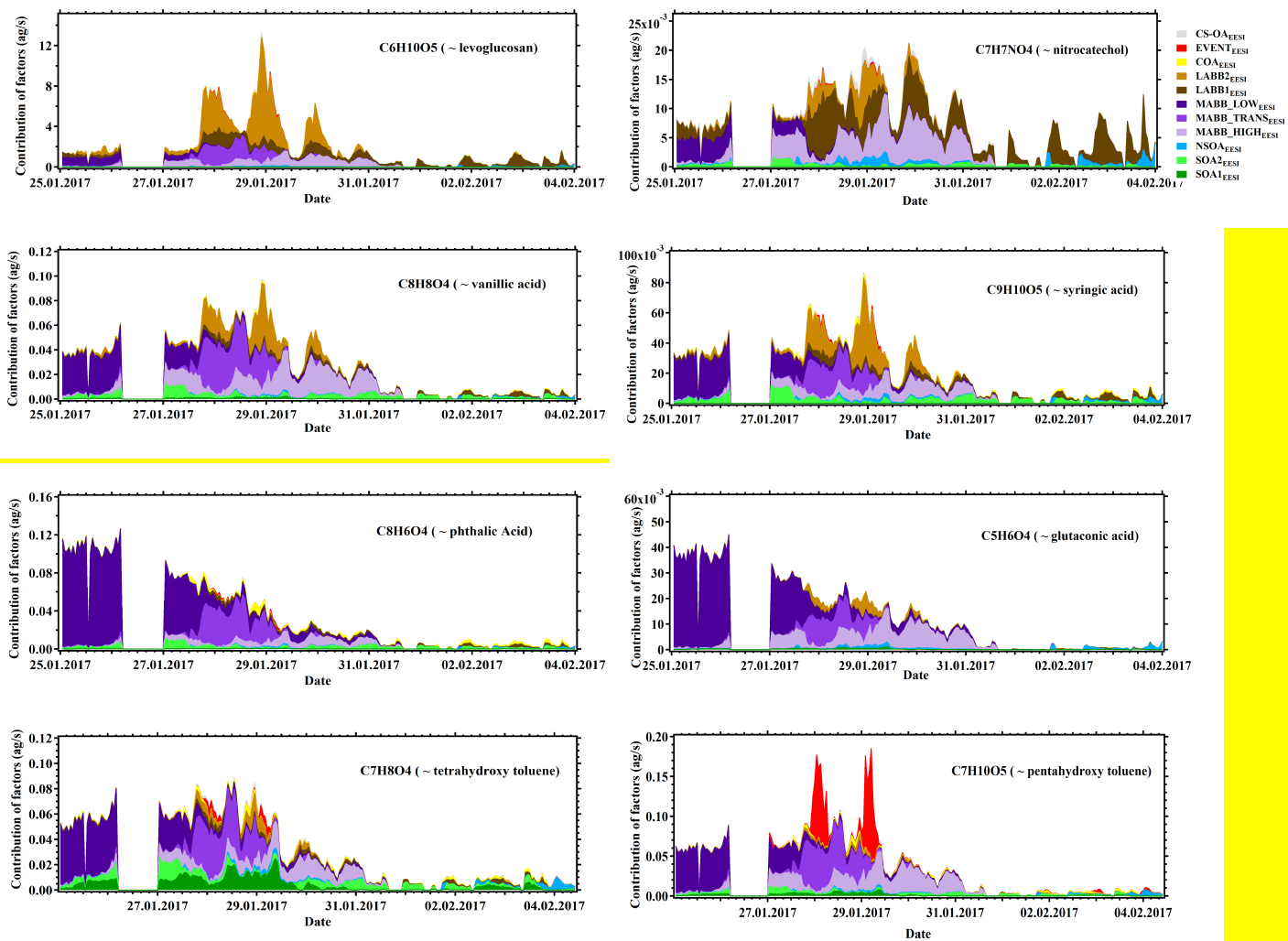
Fig. 7. Wind analysis results using the SWIM model on the concentrations of MABB\_LOW<sub>EESI</sub>, MABB\_TRANS<sub>EESI</sub>, MABB\_HIGH<sub>EESI</sub>. Left: wind direction combined with frequency, wind speed in m/s. Right: the wind speed and wind direction.



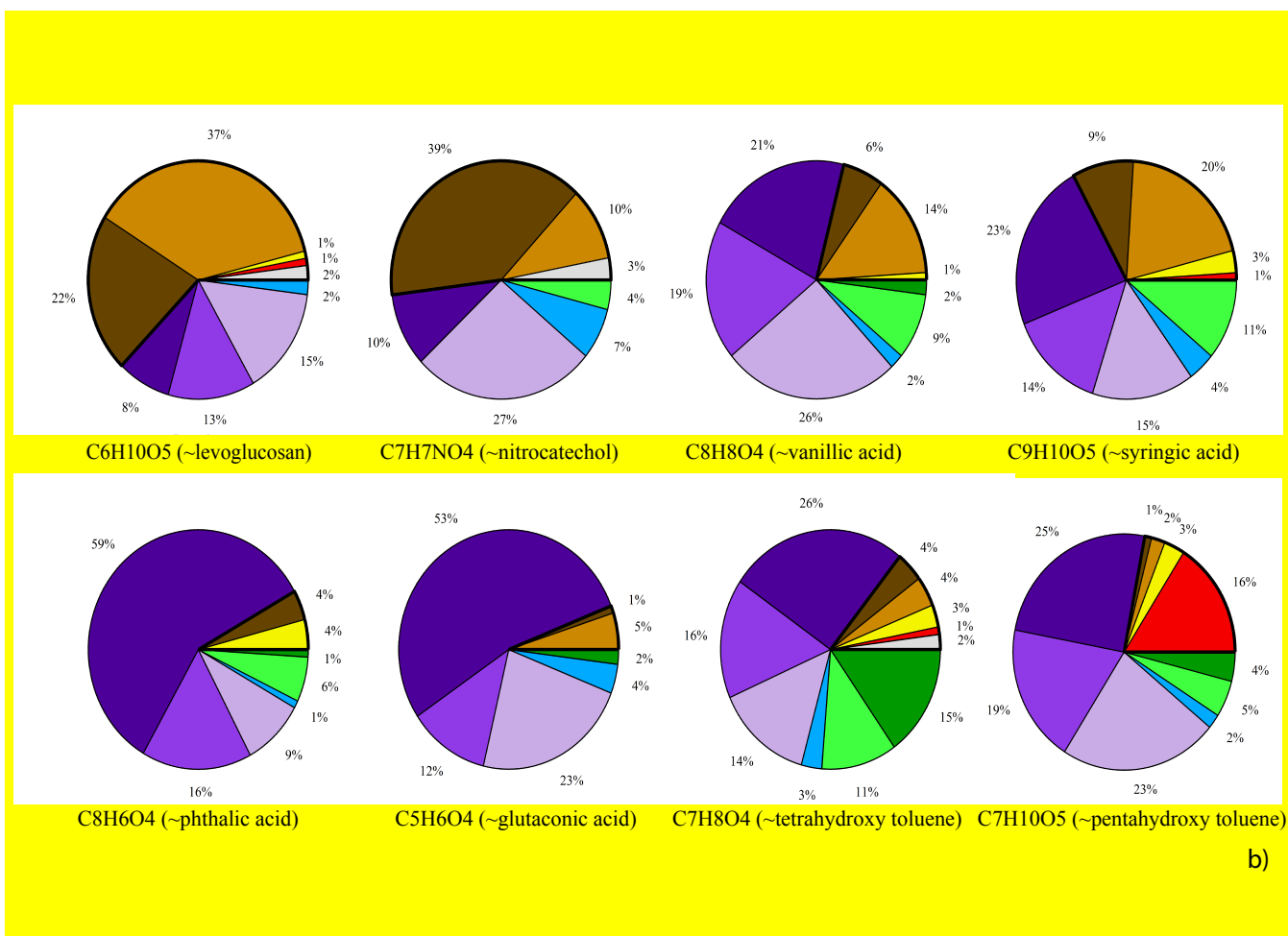
5 Fig. 8. Carbon oxidation state (Osc) as a function of number of carbons atoms for the factors, More Aged Biomass Burning\_Low temperature, More Aged Biomass Burning\_Transition, More Aged Biomass Burning\_High temperature, Secondary organic aerosol, Less Aged Biomass Burning. Points are colored by atomic H:C ratio and sized by the fraction of each ion apportioned to the designated factor.



**Fig. 9** Van Krevelen plots (atomic H:C vs. O:C) for the SOA1<sub>EESI</sub> and SOA2<sub>EESI</sub> factor mass spectra. The points are sized by the fraction of each ion apportioned to SOA1<sub>EESI</sub> and SOA2<sub>EESI</sub> and colored by the number of carbon atoms.



a)



**Fig. 10. Apportionment of selected ions by EESI-TOF PMF. (a) Time series of the mass flux ( $\text{ag s}^{-1}$ ) and (b) mean fraction apportioned to each factor. Each ion is associated with a compound of interest having this molecular formula, however, the relative isomeric abundance of this compound cannot be confirmed by the EESI-TOF.**

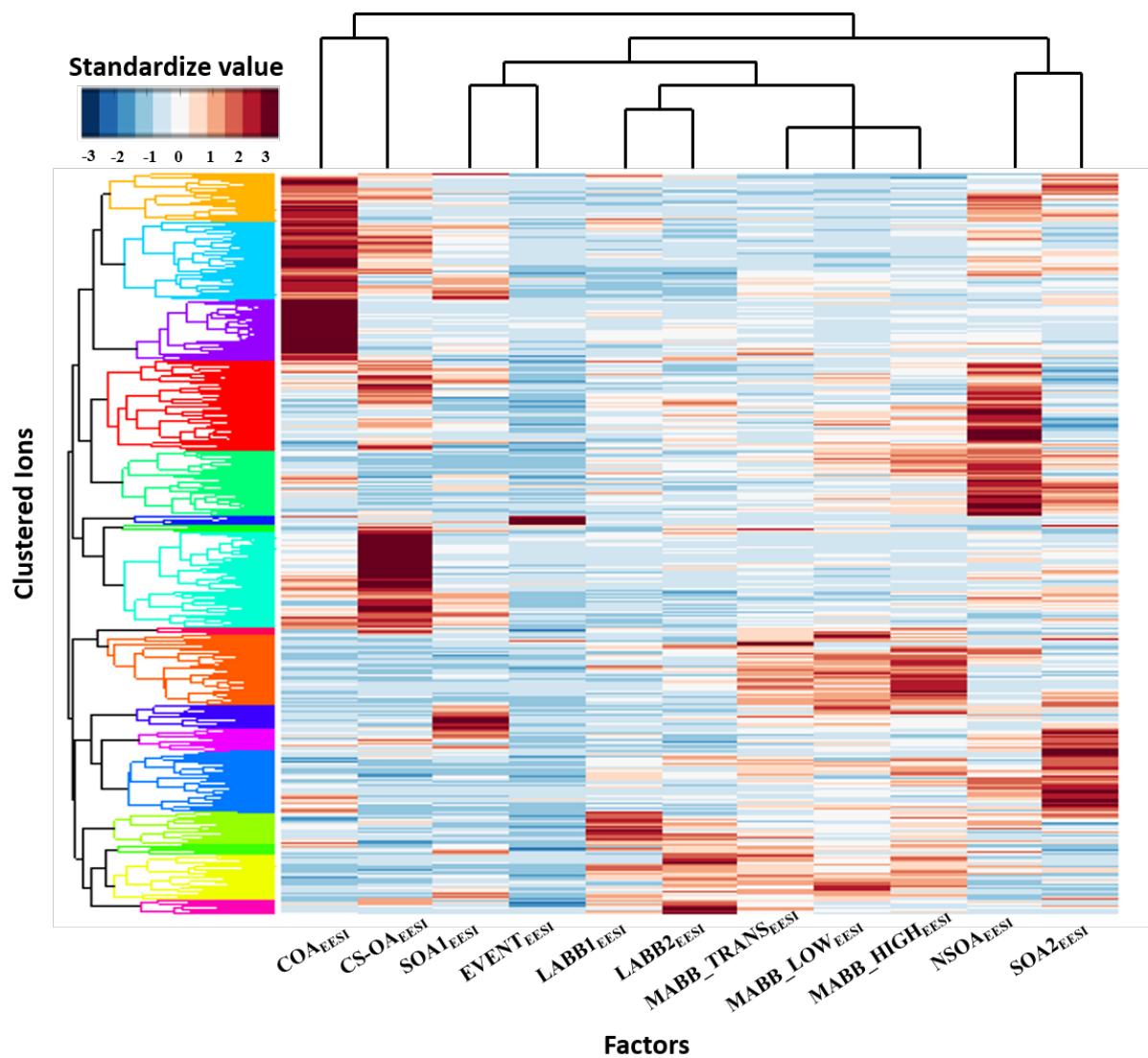


Fig. 11. Standardize matrix of individual EESI-TOF ions vs. EESI-TOF PMF factors. Ions and factors are sorted according to the results of their respective hierarchical clustering analysis; the resulting dendrograms are shown on the respective axes. The color of the compounds' groups in the dendrogram are chosen to make groupings convenient to read (color is random).



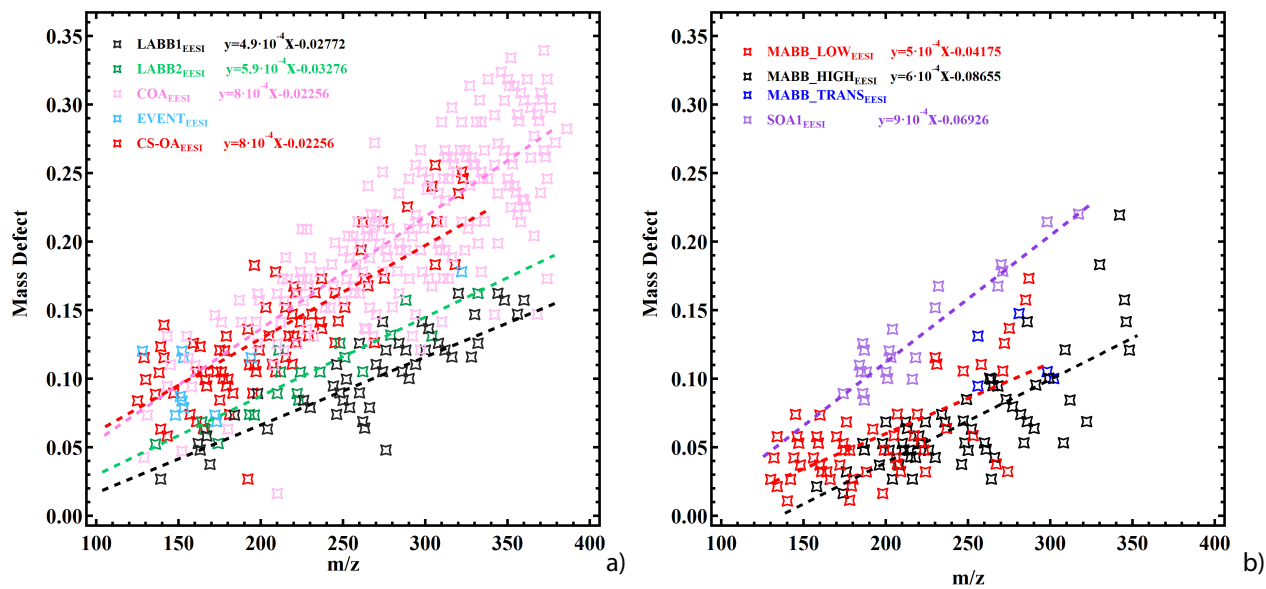
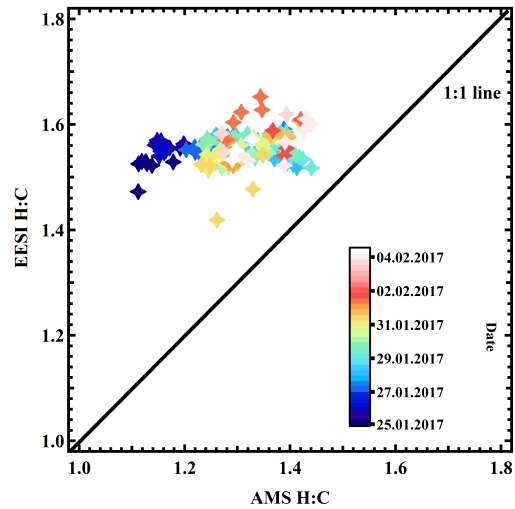
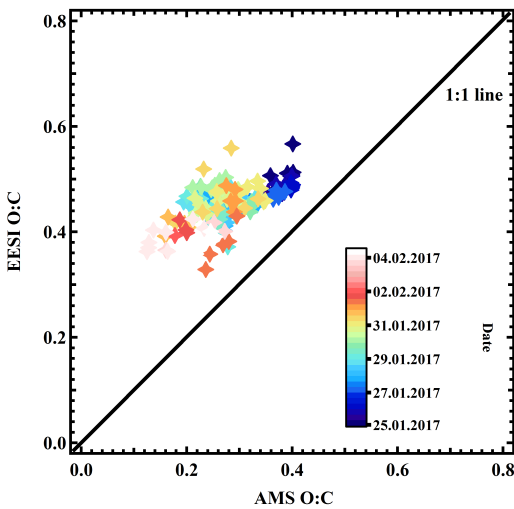
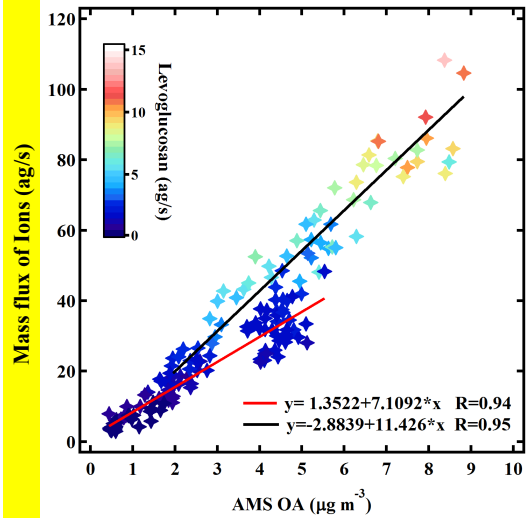
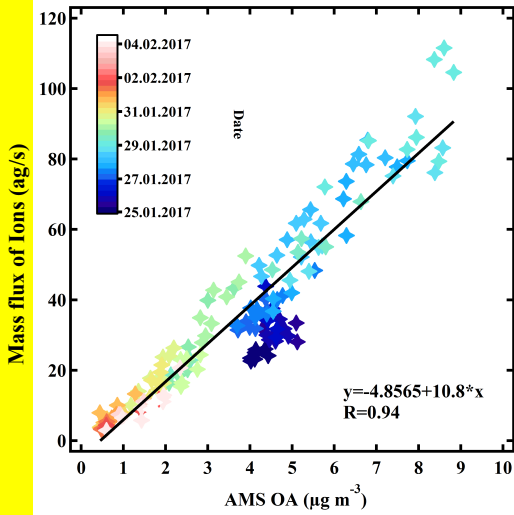


Fig. 12. Mass defect filtering plot of factor-specific ions (identified from the cluster analysis) for selected EESI-TOF POA (a) and SOA (b) factors.

5

10

15



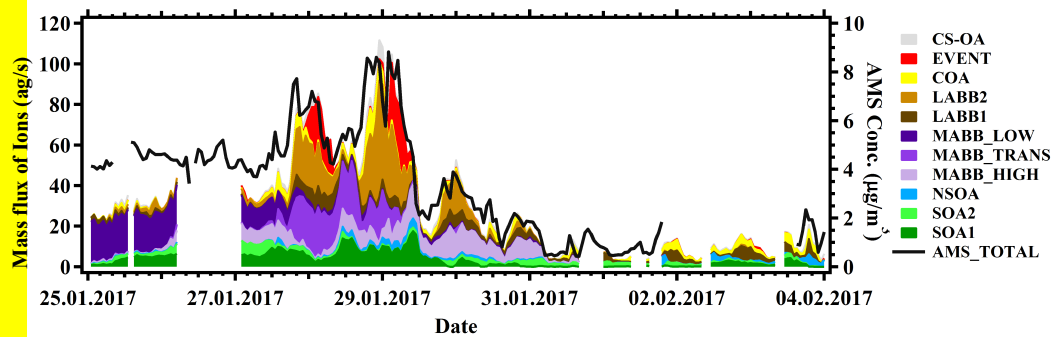
a)

b)

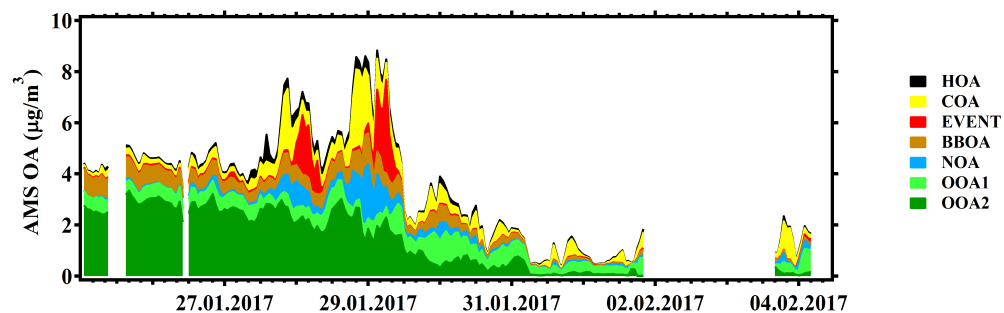
c)

d)

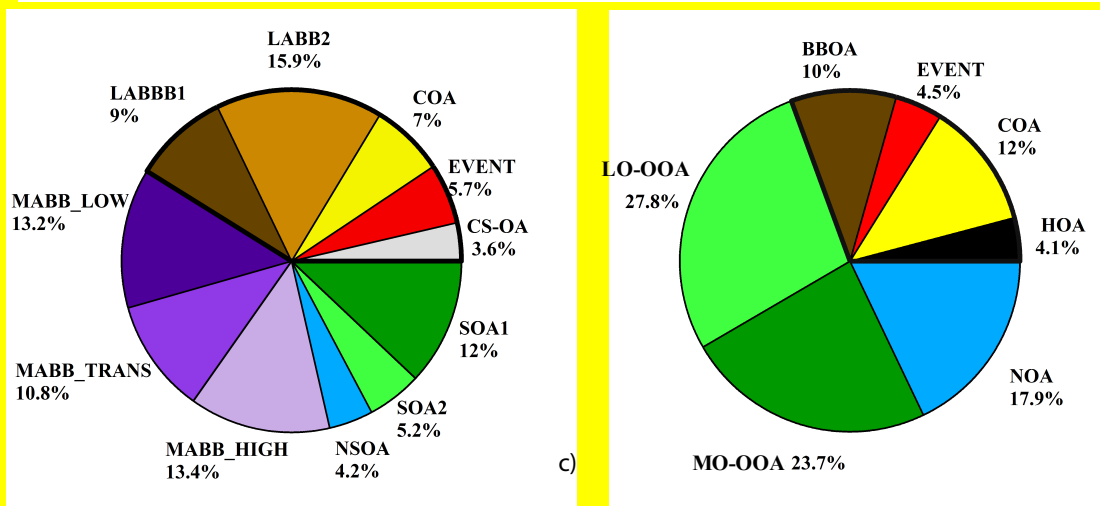
Fig. 13. Comparison of EESI-TOF and AMS. Total EESI-TOF mass flux (ag/s) as a function of AMS OA, points are colored by date (a) and the fraction of levoglucosan (b). The EESI-TOF and AMS comparison in terms of H:C (c) and O:C (d), points are colored by date.



a)



b)



c)

Fig. 14. Comparison between EESI factors and AMS factors: time series of the mass flux of each EESI PMF factors (a) and time series of concentrations of each AMS PMF factors (b). Pie charts of source apportionment results from the EESI (left) and AMS (right) (c). The thick block frame denotes the sum of the primary OA for both data sets.

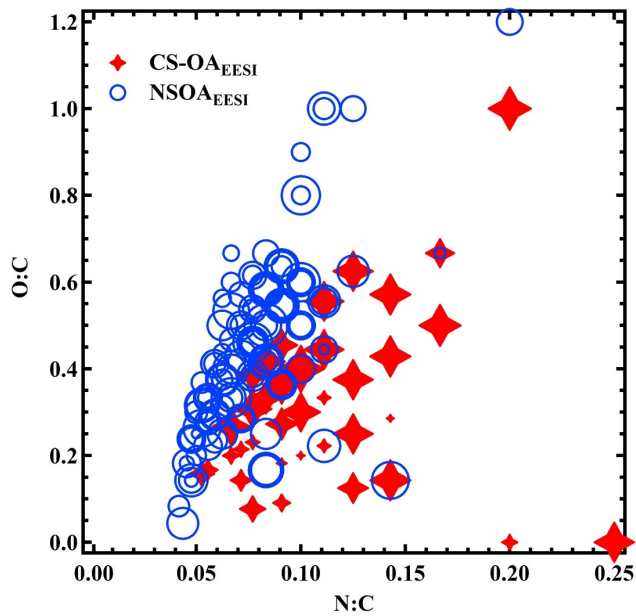


Fig. 15. The atomic O:C vs. N:C plot of the CS-OA<sub>EESI</sub> and NSOA<sub>EESI</sub> factors mass spectra. Points are sized by H:C value of each distinguished ion of the CS-OA<sub>EESI</sub> and NSOA<sub>EESI</sub>.

## Response to the comments of anonymous referee #1

We thank the referee for the valuable comments which have greatly helped us improve the manuscript. Please find below our responses (in black) after the referee comments (in blue). The changes in the revised manuscript are written in *italic*.

### General comments

The measurement campaign was very short (January 25 – February 5), so the authors should add a short discussion on the significance and representativeness of their results.

We agree with the comment that the period of the measurement campaign is short. However, the Zurich-Kaserne site has been extensively characterized in previous studies using AMS and ACSM. The general similarity of the AMS results obtained in the current study to previous publications gives us high confidence that the results obtained here are typical of wintertime conditions at the site (with the exception of the special events clearly separated by the  $EVENT_{EESI}$  and  $EVENT_{AMS}$  factors).

We add a short note in section 2.1 (P4, L27-31), “*Although the measurement period is relatively short (12 days), the similarity of the AMS results obtained in the current study to previous AMS and ACSM measurements at the same site (Lanz et al., 2007, Canonaco et al., 2013, Richard et al., 2011, Daellenbach et al., 2016) give us high confidence that the sampled aerosol is representative of typical wintertime conditions. Exceptions to this are resolved by the source apportionment into unique event-driven factors, as discussed in the results section.*”

Also a short note should be added on why PMF was done separately for AMS and EESI-TOF data, and if the authors expect results to differ for a combined approach (if possible at all).

A combined AMS/EESI-TOF PMF analysis is potentially of high interest and may facilitate quantitative interpretation of EESI-TOF data. However, such combined analyses are highly complex, requiring careful balancing of the explained variability within the two component datasets (Slowik et al., 2010; Crippa et al., 2013). In addition, such combined analyses tend to decrease the ability of the component datasets to retrieve factors resolvable by only a single instrument (such as  $HOA_{AMS}$ , or factors driven by chemical signatures observable only by the EESI-TOF). As a result, we focus here on exploring the ability of the novel EESI-TOF measurements to improve factor separation and design the source apportionment analysis to maximize this potential.

As requested, we address this in the manuscript (section 2.3) (P8, L8-9) as follows:

*“Execution of PMF analysis on separated AMS and EESI-TOF datasets minimizes the complexity of the analysis, while maximizing the factor resolution ability of the EESI-TOF.”*

Conclusions are a bit meager, and some effort could be taken in better describing the (atmospheric) implications of the results.

The following text has been added to the conclusion (P21, L1-4), “*Comparisons of bulk measurements, as well as of individual factors or groups of factors between the EESI-TOF and AMS indicate good agreement, but with the differences in elemental ratios. This suggests that, despite significant uncertainties in the relative response*

*factors of individual ions measured by the EESI-TOF, responses at the level of the PMF factors are relatively similar, with the main differences resulting from the high sensitivity to levoglucosan in the EESI. Furthermore, source apportionment of EESI-TOF provides more classification of SOA factors, separating EESI biomass burning factors as more/ less aged instead of primary / secondary, identifying organic nitrogen containing factors as primary-dominated nitrogen factor / organonitrate-containing secondary factor, which are not possible for AMS PMF.”*

A few important references are given as “in prep.” (also see specific comments below) – if these references are not available soon the authors should consider removing them and adding more information to the present manuscript.

We agree with the comment. In this manuscript, we presented four “in prep” references, and we modified as bellow:

1. “Lopez-Hilfiker et al., in prep”, now the paper is public on AMTD, so it is cited as “Lopez-Hilfiker et al., 2019”.
2. “Stefenelli et al. in prep”, the paper is now online in ACPD, so it is cited as “Stefenelli et al., 2019”.
3. “Bertrand et al., in prep”, the paper has been submitted, so here we change it to “submitted”.
4. “Lu et al., in prep”, status is unchanged.

#### Specific comments

P. 5, l. 9: What is “most”? Since the paper about the instrument is not available yet, this statement has to be made more quantitative/explicit. Which gas phase species are removed, based on what properties? The denuder only “reduces the gas phase background” – so what is left?

The referenced paper is now available from Atmospheric Measurement Techniques (<http://doi.org/10.5194/amt-2019-4>), and the reference has been updated. As a result, we have not otherwise modified the current paper but summarize here for reference.

The denuder has not been fully characterized on a compound-by-compound basis, but removes most gas-phase organics with high efficiency (e.g. pinonic acid, 99.6%).

Otherwise, the main source of background (i.e. non-particle-derived) signal is the working solution. This includes a variety of ions related to the NaI dopant and its clusters with acetonitrile and/or water. However, impurities in the working solution also generate detectable background signal. Finally, particles can pass through the denuder but then deposit on a surface rather than be extracted in the spray. Semi-volatile material desorbing from such deposited particles constitutes an additional source of background.

P. 5, l. 12: This might be discussed in the instrument paper, however, as this is not available, a short discussion should be included here: Are artefacts due to extraction to be expected, depending on solvent?

As noted in the previous comment, the instrument paper is now available and this issue is discussed in detail there. Briefly, the principal artifacts deriving from the use of the water, acetonitrile solvent are the potential for clusters of analyte ions with acetonitrile. These are weakly-bound clusters, and their prevalence depends strongly on voltage settings (i.e. collision energy) in the ion transfer optics. Formation of these clusters was found to be negligible at the settings used in the current study. We have clarified this as follows (P5, L25-26):

*“Depending on voltage settings in the ion transfer optics (i.e. collision energy), clusters with acetonitrile can potentially be detected, however these clusters were observed to be negligible during the current study.”*

P. 5, l. 15: This implies heating afterwards. Please clarify.

This is discussed in detail in the instrument paper and clarified in the manuscript as follows (P5, L20-22):

*“The droplets then enter the mass spectrometer through a capillary heated to 250 C, however, the very short residence time in this capillary means that the effective temperature experienced by the analyte is much lower and no thermal decomposition is observed.”*

P. 5, l. 26 – 32: Have the authors tried to relate the mass flux to ambient concentrations? Please discuss this. Do the authors expect a simple calibration with levoglucosan to be able to cover “instrument flow rate, EESI extraction/ionization efficiency, declustering probability, and ion transmission”?

The reviewer raises two issues here: (1) assessment of the EESI-TOF mass flux in terms of reference measurements, and (2) utility of the levoglucosan calibration. These points are discussed separately below.

The comparison of EESI-TOF mass flux and ambient OA concentrations is the subject of section 3.5 and Fig. 13. Fig. 13a presents the correlation 0.94 between the EESI-TOF mass flux and ambient concentrations. Figures 13c and 13d respectively show the O:C and H:C atomic ratios for the EESI-TOF as a function of those for the AMS. The EESI-TOF and AMS O:C ratios are correlated ( $R=0.62$ ), however, the O:C ratios estimated by the EESI-TOF are systematically higher than those measured by the AMS. For H:C ratios, we do not observe a correlation. The EESI-TOF values are scattered around approximately 1.56, independent of the AMS H:C ratios which vary between 1.11 and 1.44. The cause for this discrepancy is not yet understood but may be related to differences in ion relative sensitivity.

As this is already a major section of the manuscript, we assume that the reviewer’s question was triggered by the sequence of discussion in the original manuscript rather than a general inadequacy in section 3.5. As a result, we have added the following statement to the initial discussion of EESI-TOF mass flux identified by the reviewer:

*“A comparison of the EESI-TOF mass flux to the AMS signal in terms of total signal or mass, bulk properties, and source apportionment results in section 3.5.”*

Regarding the second point, here it is important to distinguish between factors affecting the EESI-TOF sensitivity that are ion-dependent, and those which act uniformly across all ions. We use the levoglucosan calibration only to assess the stability of the instrument with respect to the second category (e.g. flow rate, effects of geometric overlap between aerosol and spray droplets on extraction efficiency, effective primary ion concentrations). Ion-specific considerations (extraction/ionization efficiency and ion transmission) cannot be characterized through this simple calibration. However, these are expected to be fundamental properties of the detected ions (in combination with specific instrument settings which are unchanged throughout the study (e.g. voltages in the ion optics), and although unknown are thus assumed to remain constant.

The statement on levoglucosan calibration has been revised for clarity as follows (P6, L7-8):

*“EESI-TOF stability and linearity with mass were confirmed by periodic measurement of nebulized levoglucosan aerosol with quantification of the mass concentration with an SMPS.”*

P. 6, l. 11 -13: Please include (e.g in the supplementary) more details on error calculations (show data periods chosen, values etc.) This can be very useful for readers / future users.

The original error matrix includes mass spectra from both direct ambient sampling and filter blank processed with Tofware. Then filter periods were interpolated to yield an estimated background spectrum during ambient measurements. We describe and add more clear detail in the main text as following (P6, L19-24):

*“The corresponding error matrix  $\sigma_{ij}$ , which has the same dimensions as the data matrix, follows the model of Allan et al. (2003), which calculation includes the uncertainty deriving from electronic noise, ion-to-ion variability at the detector and ion counting statistics. The error estimates in this case incorporate the uncertainties related to both the ambient measurements ( $\delta_i$ ) (direct ambient sampling period) and the background ( $\beta_{ij}$ ) (filter blank measuring period, both are processed with Tofware), which are combined in quadrature according to Eq. 2:”.*

P. 8, l. 1-2: Why were exactly these factors constrained in the AMS PMF? Please clarify. Does that introduction of subjectivity distort your solution?

First, we correct a small mistake in the original manuscript, where it was stated that both traffic and cooking factors were constrained in the AMS PMF analysis whereas in fact only traffic was constrained.

With regards to the proposed introduction of subjectivity, we note that it is well-established that factor constraints select specific solutions (i.e., the selection environmentally reasonable subset) from a large set of solutions of approximately equal mathematical quality. These solutions may not be operationally accessible during analysis in the absence of factor constraints (or other rotational control allowing multidimensional exploration; note that rotation via the global  $f_{\text{peak}}$  parameter is insufficient). As a result, the solution returned by unconstrained PMF analysis is itself subject to distortion, as its selection by the model from among other solutions of similar quality is effectively arbitrary. Factor constraints address this problem and have been shown to significantly improve PMF model performance by minimizing such arbitrary distortions (Canonaco et al., 2013, Elser et al., 2016).



In the current study, the  $\alpha$ -value for HOA was selected according to the correlations between the time series of HOA with the traffic species NO<sub>x</sub> (P10, L9-10).

P. 11, l. 16 – 18: How do the diel patterns of the nicotine and COA factors compare? Could it be that they are similar due the influence of restaurant opening times, with people gathering outside the restaurants to smoke?

This is a good point and likely contributes to the necessity for constraining the CS-OA<sub>EESI</sub> to obtain a clear separation. We have added diurnal plots of the EESI-TOF factors to the supplement (Fig. S6). The following statement has been added to the manuscript (P12, L9-13):

*“The difficulty in separating these factors, despite their expected chemical differences, is likely due to strong temporal correlation between cooking and cigarette-smoking emissions due to the proximity of local restaurants (Fig. S6, the diurnal patterns of nicotine and COA<sub>EESI</sub> factors), where people gather outside to smoke during mealtimes. We therefore attempted to obtain a clean cigarette smoking signature from the dataset to serve as an anchor profile with which to constrain this source.”*

P. 12, l. 6-7: As Figure 2b shows, C<sub>8</sub>H<sub>12</sub>O<sub>6</sub> has a very prominent signal in the LABB spectra. The authors speculate that this ion represents hydroperoxides from the oxidation of phenolic compounds by OH radicals during daytime. Biomass burning seems to be mostly going on during evening/night times – how come daytime oxidation of compounds primarily emitted at night would have such a bit signal?

Here, we made a mistake. This was an early interpretation, but forgot to revise in the text. Oxidation typically leads to a large set chemically related compounds because all these reactive pathways branch in complex ways. In contrast, strong isolated peaks (e.g. levoglucosan) are more likely to result from a specific emissions source and/or process (e.g., because cellulose is a polymer, its pyrolysis leads to a relatively small number of discrete major products including levoglucosan). Although we are unsure of the compound(s) comprising C<sub>8</sub>H<sub>12</sub>O<sub>6</sub>, it is very likely to be primary and not an oxidation product, since it is observed as an isolated peak with high relative intensity.

The incorrect interpretation is deleted in the text.

P. 12, l. 14-15: Already mention here what this “different” thing is

This statement related to the lack of correlation observed between primary/less aged AMS vs. EESI-TOF wood burning factors at the start of campaign, and has been clarified as follows (P13, L8-11):

*“Fig. S8 compares the BBOA<sub>AMS</sub> factor (Fig. 2a) with LABB1<sub>EESI</sub>, LABB2<sub>EESI</sub>, and the sum of LABB1<sub>EESI</sub> + LABB2<sub>EESI</sub>, with R 0.59, 0.79, and 0.82, respectively. The correlation is generally good except during the first part of the campaign (25 January to 27 January) which as discussed later relates to the complexity of wood burning classification between the EESI-TOF and AMS.”*

P. 12, l. 23 – 25: January 27 – 29 was a weekend, and a quick google search revealed that the Zurich game festival (<http://www.ludicious.ch/ludicious-2017/>) was taking place then, which would mean a lot around Zurich Kaserne, eating, smoking: :Also LABB2 is high then, despite higher temperatures. How do the authors explain this?

Thanks for the significant information. The following statement is added to the manuscript (P14, L3-5):

*“The Zurich game festival was taking place at the weekend (the event is apparently held in a building on the SW side of the courtyard), though no human activities in the immediate vicinity of the sampling inlet were evident by inspection of the on-site camera.”*

LABB2 is the more event-driven WB and as such is likely not connected to regular (approximately temperature-driven) domestic heating but rather the activities of the large number of people nearby participating in this event. Probably some local wood burning is associated with this.

P. 15, l. 31 – 33: How sure can the authors be that the molecular formulae they measure correspond to the mentioned compounds? Please add a short discussion on this uncertainty.

We agree that this section requires clarification. In particular, the implication that the EESI-TOF identifies specific molecules is misleading, as the instrument can provide only a molecular formula. In some cases, e.g.  $C_6H_{10}O_5$ , we know that at a minimum several chemically similar isomers are present (i.e., not only levoglucosan but other sugars such as mannosan and galactosan). We have modified both the labels and caption of Fig. 10, as well as the accompanying text, to clarify this point.

Figure labels now highlight the molecular formula, for example: “ $C_6H_{10}O_5$ ” (~ levoglucosan).

The revised Fig. 10 caption is as follows:

*“Fig. 10. Apportionment of selected ions by EESI-TOF PMF. (a) Time series of the mass flux ( $ag\ s^{-1}$ ) and (b) mean fraction apportioned to each factor. Each ion is associated with a compound of interest having this molecular formula, however, the relative isomeric abundance of this compound cannot be confirmed by the EESI-TOF.”*

Revised discussion (P16, L31):

*“Here we investigate the apportionment of eight ions associated with compounds of interest:  $C_6H_{10}O_5$  (approximately assigned to levoglucosan),  $C_7H_7NO_4$  (methyl-nitrocatechol),  $C_9H_{10}O_5$  (syringic acid),  $C_8H_8O_4$  (vanillic acid),  $C_8H_6O_4$  (phthalic acid),  $C_5H_6O_4$  (glutaconic acid),  $C_7H_8O_4$  (tetrahydroxy toluene) and  $C_7H_{10}O_5$  (pentahydroxy toluene). Note that because the EESI-TOF can provide only a molecular formula, we cannot establish for certain the identity of a compound or assess the relative isomeric abundances. For example,  $C_6H_{10}O_5$  is likely to consist not only of levoglucosan, but also other sugars such as mannosan and galactosan. The named compounds are thus provided for reference, but their identification should not be considered as conclusive and the ions cannot be assumed to be isomerically pure. Nevertheless, as these assignments are based on molecular*

*investigations of wood burning-related emissions they are likely to be qualitatively correct and provide a useful framework for interpreting molecular aspects of the source apportionment results.”*

#### Technical corrections

##### P. 2, l. 19 – 21: Sentence structure

This sentence was modified in response to a comment by Reviewer #3, and now reads as follows (P2, L18-20):

*“This suggests the EESI-TOF apportionment **in the current study** can be approximately taken at face value, despite ion-by-ion differences in relative sensitivity.”*

##### P. 3, l. 16: Family?

Done. We change to **GC-family (P3, L16)**.

##### P. 3, l. 23: It would be beneficial if this paper was available once this manuscript is online

Done. The instrument paper “An Extractive Electrospray Ionization Time-of-Flight Mass Spectrometer (EESI-TOF) for online measurement of atmospheric aerosol particles” now is public on Atmospheric Measurement Techniques Discussions (**AMTD, <http://doi.org/10.5194/amt-2019-45>**).

##### P. 4, l. 24: I suggest removing this reference unless the paper is available at the time of publication of this one.

We believe there is small typo here, and the reviewer refers to a reference at P.4. l. 14. This paper is now available at ACPD, and the reference has been updated.

##### P. 6, l. 1: Number fitting?

Done (P6, L9-10). *“**The total number of 1125 fitted ions** (including 882 Na<sup>+</sup> adducts, one H<sup>+</sup> adduct, and 242 unknown ions) between m/z 135 and 400 were identified.”*

##### P. 6, l. 6: Servo and MS? Specify

The revised text reads (P6, L11-14):

*“Data were pre-averaged to 1 min time resolution, and high resolution peak fitting was performed. Individual 1-min spectra were classified as either ambient measurements, background sampling (through the particle filter), **or transitional measurements immediately after switching between ambient/background sampling. Transitional measurements were excluded from further analysis.**”*

##### P. 7, l. 20: Minimizes

Done (P7, L29). “Equation (3) is solved using a least squares algorithm that iteratively **minimizes** the quantity  $Q$  (Eq. 4),...”

P. 17, l. 7: [separately?](#)

We have revised the text for clarity (P18, L12-19):

*The factor dendrogram identifies several groups of EESI-TOF PMF factors consistent with the interpretations provided above: (1) more aged biomass burning factors (MABB\_LOWEESI, MABB\_TRANSEESI and MABB\_HIGHEESI), (2) less aged biomass burning factors (LABB1EESI and LABB2EESI), and (3) the cooking-related OA and cigarette smoking OA factors. The more aged and less aged biomass burning factor groups are themselves likewise grouped. This clustering is consistent with our interpretation of these factors, as discussed in the previous section. Ions are clustered to different groups using the standardized values. In each factor, there are distinguished molecules (lists of the specific ions (standardized value above 1.5) for each factor is shown in Table S2). The other two resolved groups, one group including SOA1 and EVENT factor, one group containing SOA2 and NSOA factor, apparently don't retrieve the common ions, which make less sense for the current study.”*

[Fig. 2, 6: Add arrows/lines to clarify the corresponding stick labels](#)

Done. The arrows are added to the figures.

[Fig. 10: The figure looks squished](#)

Done. The shape of the figure is changed.

## Response to the comments of anonymous referee #2

We thank the referee for the valuable comments which have greatly helped us improve the manuscript. Please find below our responses (in black) after the referee comments (in blue). The changes in the revised manuscript are written in *italic*.

### Specific comments:

a) In Figure 2a, time delay seems to exist between LABB1 and LABB2 and levoglucosan (C<sub>6</sub>H<sub>10</sub>O<sub>5</sub>) for peaks on 28-29 Jan. Also, in Figure 4a, LABB1 has a higher O:C and lower H:C compared to LABB2EESI. These might indicate LABB1 is more oxygenated. It could also originate from a different source from LABB2. How does the wind regression analysis for these factors show?

Concentrations of LABB2 are high only on three days, roughly corresponding to the EVENT<sub>EESI</sub> factor that we now associate with a local festival (the Zurich game festival, the event is held in a building on the SW side of the courtyard in which the instrument is deployed). As a result, the reviewer's suggestion of a different source is likely correct. This is further supported by wind regression analysis of these two factors, shown below and added to the supplement as Fig. S7. LABB1 does not correspond to a specific wind direction, consistent with local, widespread domestic wood combustion. In contrast, LABB2 originates predominantly from a single wind direction while the smaller source to the SE is on the third day.

The sentences are added to the manuscript (P13, L5-11): *“The wind regression analysis of these two factors are shown in Fig. S7. LABB1<sub>EESI</sub> does not correspond to a specific wind direction, in contrast, LABB2<sub>EESI</sub> originates predominantly from a single wind direction, excluding the smaller source to the SE on the third day.”*

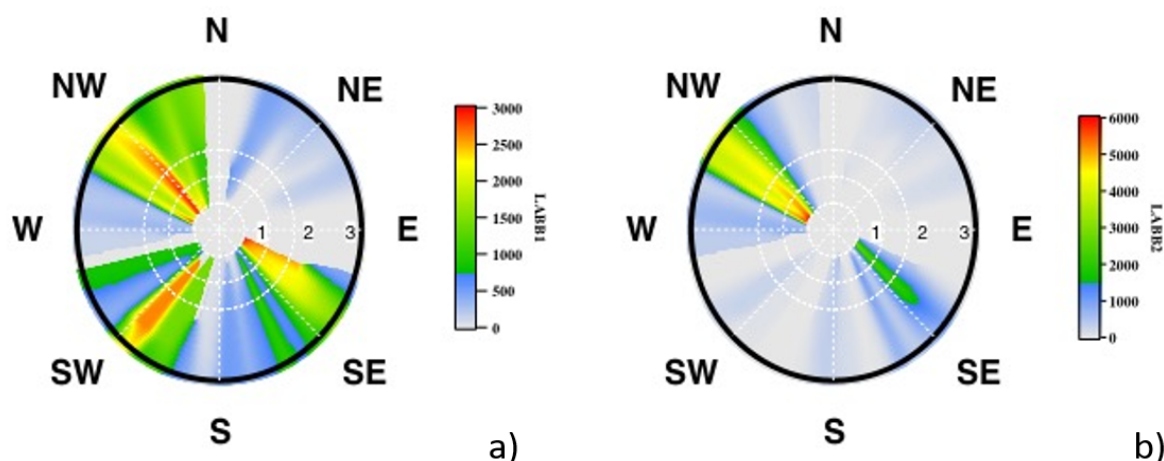


Fig. S7. Wind analysis results using the SWIM model on the concentrations of LABB1 (a) and LABB2 (b).

b) In the discussion of SOA factors (Pg. 15 Lns. 12-15), instead of a time series plot (Figure 6b), a diurnal plot will better show the daytime cycle of SOA<sub>2</sub>. Also, I am not sure what is the evidence of that both factors (SOA<sub>1</sub> and SOA<sub>2</sub>) are associated with SOA as opposed to the OOA-AMS factors. If it is due to similarities of the SOA

factors mass spectra to the monoterpene-related factors (Pg. 15 Lns. 5-6), I think it is more convincing to compare (plot correlation) of the mass spectra.

This is a good point, we added diurnal plots of the EESI-TOF factors to the supplement (Fig. S6), shown below. We have also added a comparison of the SOAEESI mass spectra with the monoterpene-related SOA factor from the Zurich summer campaign as Fig. S9.

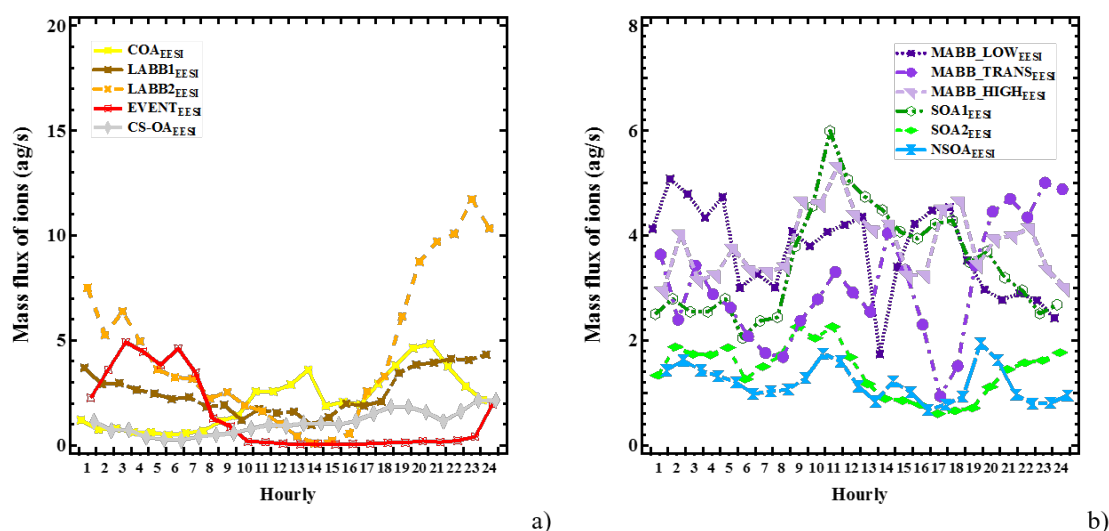
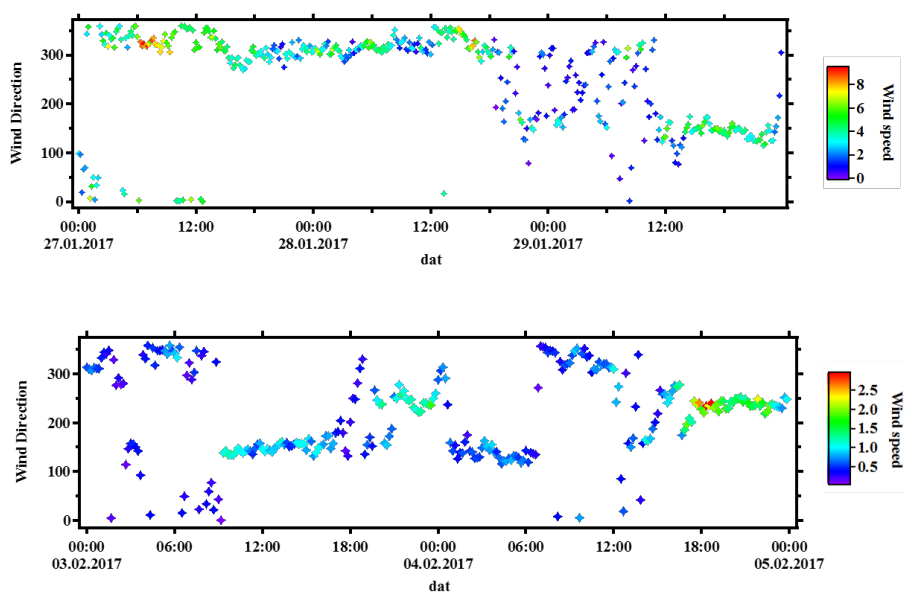


Fig. S6. The diurnal variation of EESI POA factors (a) and EESI SOA factors (b).

c) For the nitrogen-containing SOA factor, could the variability in time between the high peak at the night of 3-4 Feb and the small peak at the night of 28-29 Jan associated by a change in temperature, or was it caused by shifting of air masses? What the wind analysis or back-trajectory analysis suggest?

Yes, we agree with the comment, it is possible that the two nitrogen peaks are associated by a change in temperature. The temperature on 3-4 Feb was much higher than 28-29 Jan, but it was enhanced from 1 Feb. We assume the unique time series may also indicate other chemistry or emission process of the nitrogen-containing compounds, which we also plan to further study.

We plot the time series of wind speed and wind direction on 3-4 Feb and 27-29 Jan, shown below. The wind direction during these days are variable, especially on 3-4 Feb, which may not be a clear evidence.



d) The factor dendrogram seems to resolve five groups instead of three (Figure 11); SOA1 and EVENT factors are one group, and SOA2 and NSOA factors are another group. What do these groupings suggest in terms of characteristics? The discussion of factors dendrogram in Pg. 17 Lns. 6-10 could be expanded to include these groups.

We agree with the comment and have clarified the figure description. In the current study, the agglomerative hierarchical clustering is conducted based on the profiles from PMF. The dendrogram is generated with Euclidean distance metric and average linkage, showing relationships between each group and each factor. However, the dendrogram does not directly show which groups are “tight” (i.e. containing closely related factors or elements) and which are loose. With respect to the factors, the original text focused on three of the five groups where the grouping was consistent with factor definitions (implicitly assuming them to be tightly grouped) while not commenting on the other groups (assumed to be more loose).

We add a short illustration in the main text.

The revised text reads (P18, L12-19): *“The factor dendrogram identifies several groups of EESI-TOF PMF factors consistent with the interpretations provided above: (1) more aged biomass burning factors ( $MABB\_LOW_{EESI}$ ,  $MABB\_TRANS_{EESI}$  and  $MABB\_HIGH_{EESI}$ ), (2) less aged biomass burning factors ( $LABB1_{EESI}$  and  $LABB2_{EESI}$ ), and (3) the cooking-related OA and cigarette smoking OA factors. The more aged and less aged biomass burning factor groups are themselves likewise grouped. This clustering is consistent with our interpretation of these factors, as discussed in the previous section. Ions are clustered to different groups using the standardized values. In each factor, there are distinguished molecules (lists of the specific ions (standardized value above 1.5) for each factor is shown in Table S2). The other two resolved groups, one group including SOA1 and EVENT factor, one group containing SOA2 and NSOA factor, apparently don’t retrieve the common ions, which make less sense for the current study.”*

Technical comments:

a) Pg. 15 Ln. 21: Ratio of N:C of NSOA in Table S1 is 0.04.

Done (P16, L21). The value in **Table S1 (0.04)** is correct, and the value in the main text has been fixed.

b) Pg. 16 Lns. 13-14: Check the percentage contribution of syringic acid and vanillic acid that are apportioned to MABB factors. Based on Figure 10b, they are supposed to be 52% and 66%, respectively.

Done (P17, L19). “..., which in turn are a major component of biomass combustion emissions, and are apportioned primarily to the  $MABB_{ESI}$  factors **52 %** for syringic acid and **66 %** for vanillic acid).”

c) Pg. 16 Lns. 23-24: Be consistent with the decimal of percentage. The percentage can be off if the decimal is included.

Done (P17, L29). The main text is changed. “Tetrahydroxy toluene ( $C_7H_8O_4$ ) and pentahydroxy toluene ( $C_7H_8O_5$ ) are apportioned mainly to secondary factors **85 % and 78 %**, respectively).”

d) Table S2: LAWBI refers to LABB1? Check the acronyms of factors and make them consistent throughout the main text and supporting information.

Done. Yes, it should be **LABB1**. we correct the other names, **LABB2, MABB\_LOW, MABB\_TRANS, MABB\_HIGH, CS-OA**.



### Response to the comments of anonymous referee #3

We thank the referee for the valuable comments which have greatly helped us improve the manuscript. Please find below our responses (in black) after the referee comments (in blue). The changes in the revised manuscript are written in *italic*.

#### Technical comments:

P1, L19: “This suggests the EESI-TOF apportionment can be approximately taken at face value, despite ion-by-ion differences in relative sensitivity.” Perhaps this is too strong of a statement since this environment is under major influence of BB (and little HOA) and not a mix of very different sources as can be common in other environments, especially in the summer.

We have revised this statement to note that it applies specifically to the range of aerosol composition observed in the current study. We agree that similar analyses are required in different environments before a general conclusion can be drawn. However, even in the current study there is significant variation in the chemical composition (e.g., primary vs. secondary biomass burning, as well as SOA factors with terpene-like signatures that comprise nearly 20 % of the EESI-TOF signal).

The revised statement reads as follows (P2, L18-20):

*“This suggests that the EESI-TOF source apportionment in the current study can be approximately taken at face value, despite ion-by-ion differences in relative sensitivity.”*

P6, L3: why were the ions smaller than 135 amu small? Was tuning changed on purpose to increase transmission of larger ions? Or is there something different in the mass spec design compared to that of a typical CIMS/AMS?

For EESI-TOF, we aim to focus on the molecular compositions. The quads were operated such that the transmission decreases rapidly below approximately  $m/z$  150.

P6, L8: Just out of curiosity, how stable was the background?

In this campaign, the background measurement is in the same level (almost the same value) during the whole measurement period.

P8, L2: why were different factors constrained for AMS and EESI-ToF?

Factors may be constrained to overcome a variety of issues compromising an unconstrained PMF solution, including factors with strong temporal correlation, two or more factors with similar chemical signatures, or factors with chemical features that due to the overall chemical variability within the dataset are not clearly mathematically

resolved. In the first case (temporal correlation), the reviewer is correct that one would expect similar factors to be constrained between AMS and EESI-TOF (with the exception of HOA<sub>AMS</sub>, which is primarily composed of hydrocarbon species undetectable by the EESI-TOF). However, the remaining two cases depend on the nature of the chemical measurement by the AMS and EESI-TOF, which are fundamentally different. Therefore, the constrained factors are expected to vary between instruments. For example, in the current analysis, we note that the extensive fragmentation in the AMS results in both the long-chain hydrocarbons observed HOA<sub>AMS</sub> and the fatty acids in COA<sub>AMS</sub> yielding C<sub>x</sub>H<sub>y</sub><sup>+</sup> fragments, which makes these factors appear chemically similar (and requiring constraints). In contrast, the EESI-TOF directly detects cooking-derived fatty acids, making this factor appear chemically unique (and not requiring constraint). Meanwhile, the increased chemical specificity of the EESI-TOF allows identification of a cigarette-smoke related factor based largely on the distinctive C<sub>10</sub>H<sub>14</sub>N<sub>2</sub>H<sup>+</sup> ion, with the separation optimized by factor constraints, whereas the reduced chemical specificity necessitated the incorporation of cigarette smoke into a mixed factor.

We clarify this issue in the manuscript as follows (P8, L10-14):

*“Different factors were constrained in the two datasets due to the fundamental differences between the AMS and EESI-TOF measurements. Specifically, the absence of fragmentation in the EESI-TOF allowed clear separation of cooking without the need for constraints, while separation of a cigarette smoke factor was only achieved for the EESI-TOF. In addition, constraining an AMS cigarette smoke factor was attempted but failed.”*

P10, L1-2: If OOA1 has some of the C<sub>x</sub>H<sub>y</sub> fragments, shouldn't that be the more volatile OOA factor?

We agree with this comment.

The OOA1 factor is the less oxygenated factor (LO-OOA), while the current OOA2 factor is more oxygenated OOA (MO-OOA). We had fixed all the corresponding text and figures.

P15, L3: It's surprising that the SOAEESI factor is less oxidized than the oxidized BB factors. At least in the AMS-based PMF factors, the SOA (OOA) factors are quite more oxidized than the BBOA factor. Do the EESI data suggest that AMS OOA factors may contain the aged BB emissions too? Could the authors elaborate on this. After reading the paper, I realize the authors address this in Section 3.5. It will be good to include a sentence in P15 indicating that this surprising result is going to be further examined.

We agree with this suggestion.

Here we add a short sentence to guide readers (P16, L14-15), *“The more detailed comparison between EESI-TOF\_SOA factors and AMS\_OOA factors will be discussed in Sect. 3.5.”*

P17, L27-28: If the slopes of the mass defect vs. m/z points for both less and more aged BB is similar, how can one trust interpreting the slopes to understand types of functional groups added to the molecules included in the other factors?

We have clarified this discussion. The main point in these mass defect plots is that the BB-related factors (both LABB and MABB) exhibit slopes (and intercepts) that are significantly lower than those of the other investigated factors, suggesting increased aromaticity. Within the BB-related factors, LABB2<sub>EESI</sub> and MABB\_HIGH<sub>EESI</sub> have similar slopes, which are slightly higher than LABB1<sub>EESI</sub> and MABB\_LOW<sub>EESI</sub>.

Section 3.4 was revised as follows (P18-19):

*“LABB1<sub>EESI</sub> and LABB2<sub>EESI</sub> have a lower mass defect and shallower slope than COA<sub>EESI</sub> and CS-OA<sub>EESI</sub>, consistent with increased aromaticity. The slopes are  $(4.9\pm0.4)\cdot10^{-4}$ ,  $(5.9\pm0.6)\cdot10^{-4}$ ,  $(8\pm0.5)\cdot10^{-4}$  and  $(8\pm0.3)\cdot10^{-4}$  for LABB1<sub>EESI</sub>, LABB2<sub>EESI</sub>, COA<sub>EESI</sub> and CS-OA<sub>EESI</sub>, respectively. The slopes of the two LABB factors as well as those of COA<sub>EESI</sub> and CS-OA<sub>EESI</sub> are very similar to each other and have a high possibility to be consistent with CH addition for the former (i.e.  $C_{10+x}H_{14+x}O_{4-5}$ , theoretical slope  $6\cdot10^{-4}$ ), and CH<sub>2</sub> addition for the latter (i.e.  $C_{10+x}H_{20+2x}O_{3-5}$  for COA<sub>EESI</sub> and  $C_{10+y}H_{15+2y}NO_{3-5}$  for CS-OA<sub>EESI</sub> as nearly every CS-OA-specific ion contains a single N atom, theoretical slope  $1.1\cdot10^{-3}$ ).*

*The MABB and LABB factors have similar slopes, despite different ion lists. The slopes of two MABB factors ( $0.9\cdot10^{-3}$ ), as shown in Fig. 12b, are consistent with the addition of CHO functionality (theoretical slope =  $0.1\cdot10^{-2}$ ). Due to the high variability of the slopes of the MABB factors, it may also contain the other potential possibility for the added functionalities.”*

Section 3.5: I believe this section should be discussed earlier. As I was reading the paper, I kept thinking what if differences in sensitivity of the EESI to different molecules is playing a role in determining the identified factors, so it is reassuring if one reads this section before getting deep into the EESI-based PMF factors. Perhaps the first paragraph of this section that's not discussing the PMF factors yet can be moved to earlier parts of the paper.

We agree with the reviewer that molecule-dependent sensitivity of the EESI-TOF is an important point and that the reader should be aware from the start that an AMS/EESI-TOF comparison is presented. With that said, we prefer to retain the current order of sections because the bulk AMS/EESI-TOF comparisons cannot be well understood without also considering the factor-by-factor comparisons, and the factor comparisons in turn cannot be understood without first presenting the PMF analysis. However, we have added the following text to the beginning of section 3 to clarify the discussion structure (P9, L21-26):

*“Results of AMS and EESI-TOF PMF analyses are presented in sections 3.1 and 3.2, respectively. Section 3.3 focuses on the EESI-TOF PMF results are then exploited to assess the apportionment of specific ions related to key marker compounds (section 3.3) and to identify groups of molecules uniquely characteristic of the retrieved factors (section 3.4). However, quantitative interpretation of the EESI-TOF PMF results is complicated by differences in the relative sensitivity of the EESI-TOF to different molecules. Therefore section 3.5 presents a comparison of the EESI-TOF and AMS results in terms of total signal, bulk atomic composition, and relative apportionment to the different factors.”*

P18, L7-9: What's the explanation for a lower sensitivity of EESI for the times levoglucosan was low? Based on the AMS factors, the initial period is not dominated by HOA that EESI is blind to.

Laboratory measurements indicate that the EESI-TOF is likely more sensitive to levoglucosan than to typical SOA components (Lopez-Hilfiker et al., 2019). This has been clarified in the text as follows (P19, L16-19):

*“An SOA-dominated period with low levoglucosan concentrations (red line) toward the beginning of the campaign exhibits a lower sensitivity than during a period with higher levoglucosan concentrations (black line), which includes the events on 28.01.2017 and 29.01.2017 characteristic of  $EVENT_{EESI}$  (Lopez-Hilfiker et al., 2019).”*

Editorial comments: Some correlation coefficients were presented as R, some as R2. Please use one consistently.

Done. We change all the R2 to R. R2 is focus on the section 3.2.2, LABB factors.

Quality of some figures was not good when viewed at 100% and the legends/axis labels were blurry.

Done. Fig. 10 and Fig. 12 are changed to the high resolution.

P 1, L15: “: : was derived: : :”

Done. “While the AMS attributed slightly over half the OA mass to SOA but did not identify its source, the EESI-TOF showed that most (> 70 %) of the SOA was derived from biomass burning.” (P2, L15)

P1, L21: consider changing “..utilize a : : :” to “: : :utilization of a : : :”

Done. “The apportionment of specific ions measured by the EESI-TOF (e.g. levoglucosan, nitrocatechol, and selected organic acids), and utilization of a cluster analysis-based approach to identify key marker ions for the EESI-TOF factors are investigated.” (P2, L21)

P2, L14: delete “of” in “: : :fragments typical SOA molecules”.

Done. “The chemical analysis of aerosol online-proton transfer reaction mass spectrometer (CHARON-PTR-MS) has no significant thermal decomposition but the ionization scheme fragments typical SOA molecules.” (P3, L15-16)

P6, L2: change “The total number fitting of 1125 ions : : :” to “The total number of 1125 fitted ions : : :”.

Done. “The total number of 1125 fitted ions (including 882  $Na^+$  adducts, one  $H^+$  adduct, and 242 unknown ions) between  $m/z$  135 and 400 were identified.” (P6, L9-10)

P6, L9: delete “the” to read “: : :were removed from further analysis.”

Done. “Ions with a mean signal-to-noise ratio (SNR) below 2 were removed from further analysis.” (P6, L17-18)

P6, L19: “An HR-ToF-AMS: : :”

Done. “An HR-TOF-AMS was deployed for online measurements of non-refractory (NR)  $PM_{2.5}$ .” (P6, L28)

P6, L21: “spent”

Done. “The AMS recorded data with 1 min time resolution, of which 30 s was spent recording the ensemble mass spectrum (mass spectrum (MS) mode) and 30 s recording size-resolved mass spectra.” (P6, L29-31)

P14, L4/5: it should be NOAAMS (not NSOAAMS). Related to this, the discussion on NOAAMS and the EESI-based factor is provided after this sentence, not before.

Done. In the section 3.1, the AMS source apportionment, we suggest this factor is a mixed factor, so here we don't include it into the calculation of AMS OOA contribution. “NOA<sub>AMS</sub> is excluded from this calculation due to the contribution from primary cigarette smoke discussed above.” (P15, L2-3)

Fig. 6b- unit for precipitation is not included.

Done. The unit for precipitation has been added (mm/h).

P14, L12: “period 1” should be move to “: : :study, period 1, occurs : : :”

Done. “As shown in Fig. 6b, the coldest part of the study, period 1, occurs from 25 to 27 January.” (P15, L9-10)

P15, L25: I believe you mean EESI-CHON fragments correlate better NSOA\_EESI. I don't think AMS\_CHON fragments are displayed.

Done. “Figure S10 shows a comparison of the NSOA<sub>EESI</sub> time series and CS-OA<sub>EESI</sub> time series with the CHON ions from the EESI and CHN ions from the AMS, respectively. The group of EESI\_CHON ions shows the same temporal variation as the NSOA<sub>EESI</sub> factor (Fig. S10) while the AMS\_CHN group is more correlated to the primary organic group.” (P16, L24-26)

Fig. 12b. Green color for more aged secondary BB factor is too hard to see against the white background.

Done. We change to the black color.

P18, L10: consider replacing “big events” to some other phrase

Done. “An SOA-dominated period with low levoglucosan concentrations (red line) toward the beginning of the campaign exhibits a lower sensitivity than during a period with higher levoglucosan concentrations (black line), which includes the events on 28.01.2017 and 29.01.2017 characteristic of  $EVENT_{EESI}$ .” (P19, L16-19)

Fig. 13b: intercept of the black line appears to have two negative signs in front

Done. We delete one negative sign.

P18, L31: change “primarily” to “primary”

Done. “Both AMS and EESI-TOF factors stacked time series (Fig. 14) show clearly that biomass burning is dominated by secondary fractions early in the campaign, mixed fractions in the middle of the campaign, and primary fractions late in the campaign.” (P20, L6)

P18, L34: replace “WB-related” by “BB-related”?

Done. “..., it is hard to define how much of AMS OOA is BB-related as a function of time.” (P20, L9)

# Supplement Information

## **Organic aerosol source apportionment in Zurich using an extractive electrospray ionization time-of-flight mass spectrometer (EESI-TOF): Part II, biomass burning influences in winter**

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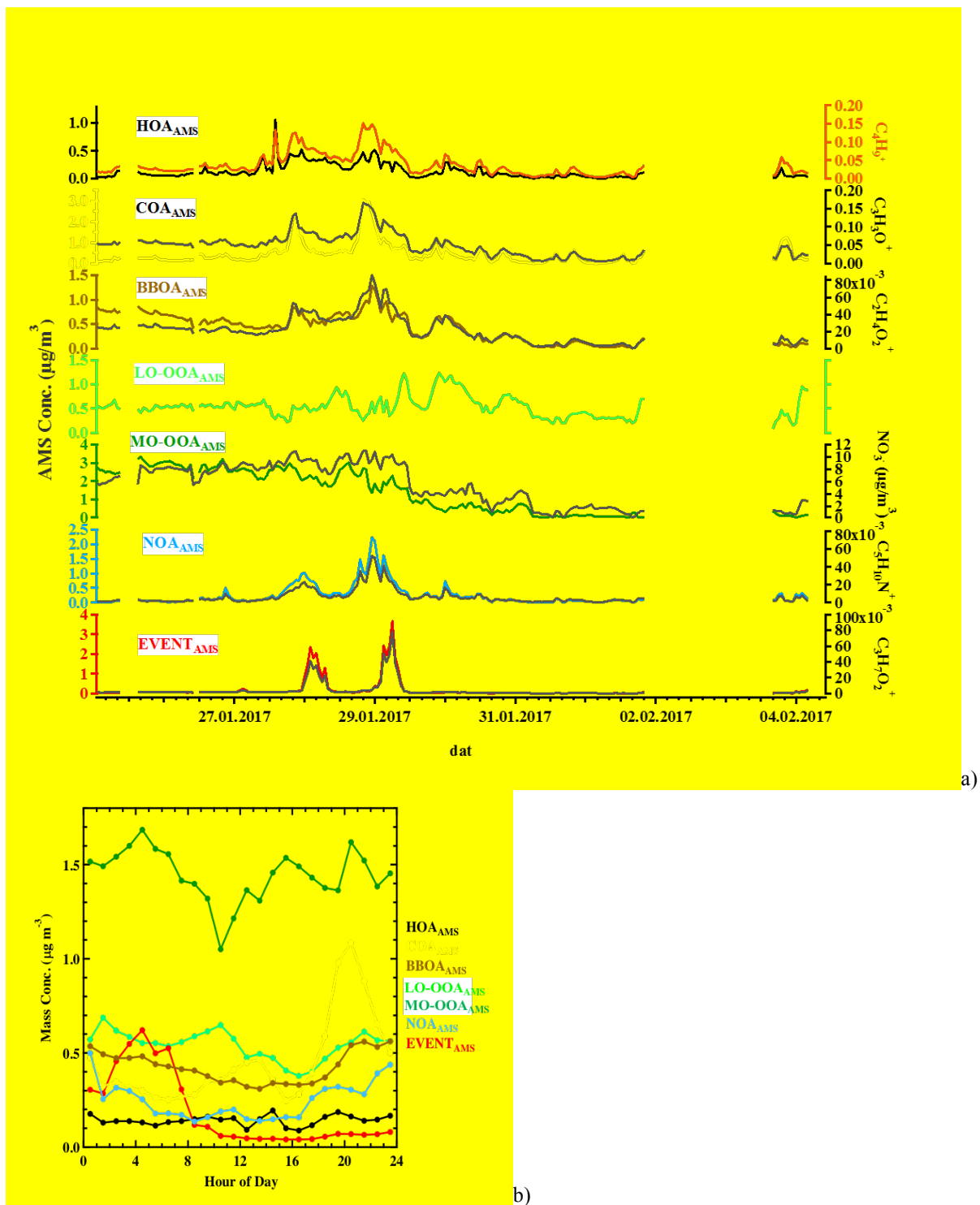
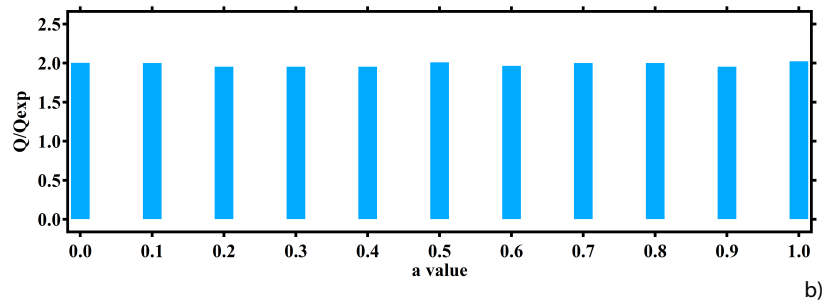
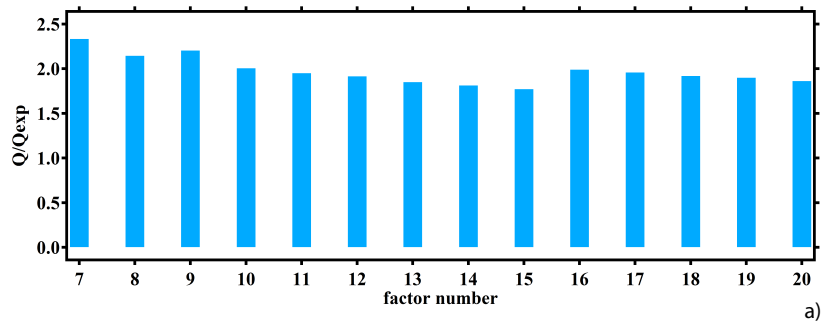
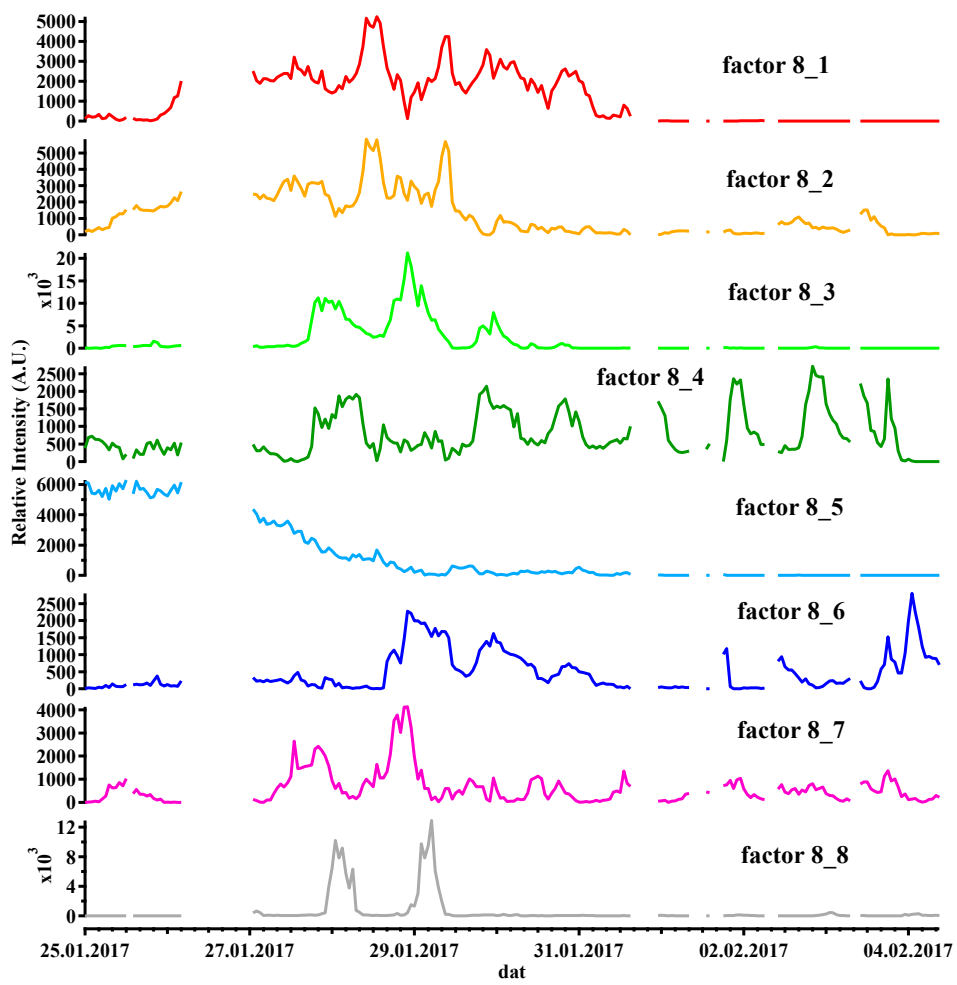


Fig. S1 a) Time-series of the AMS factors. b) Diurnal variations of the AMS factors.

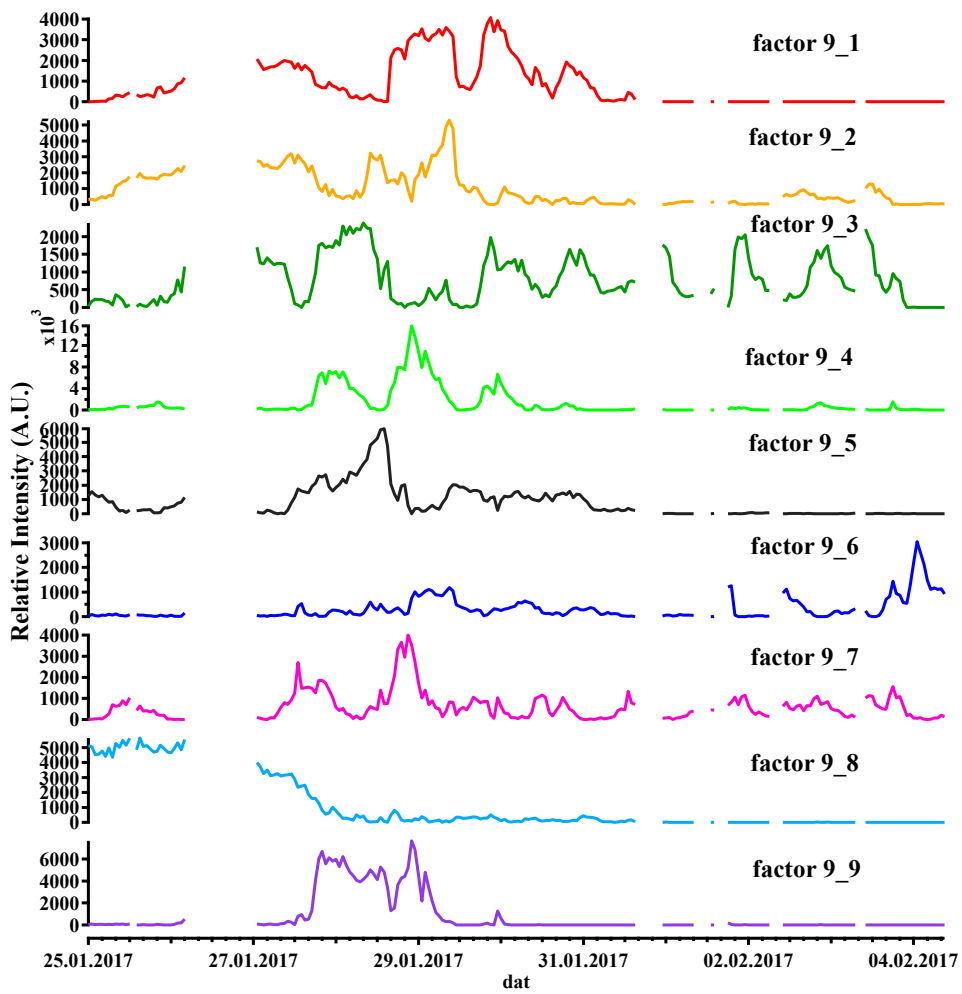




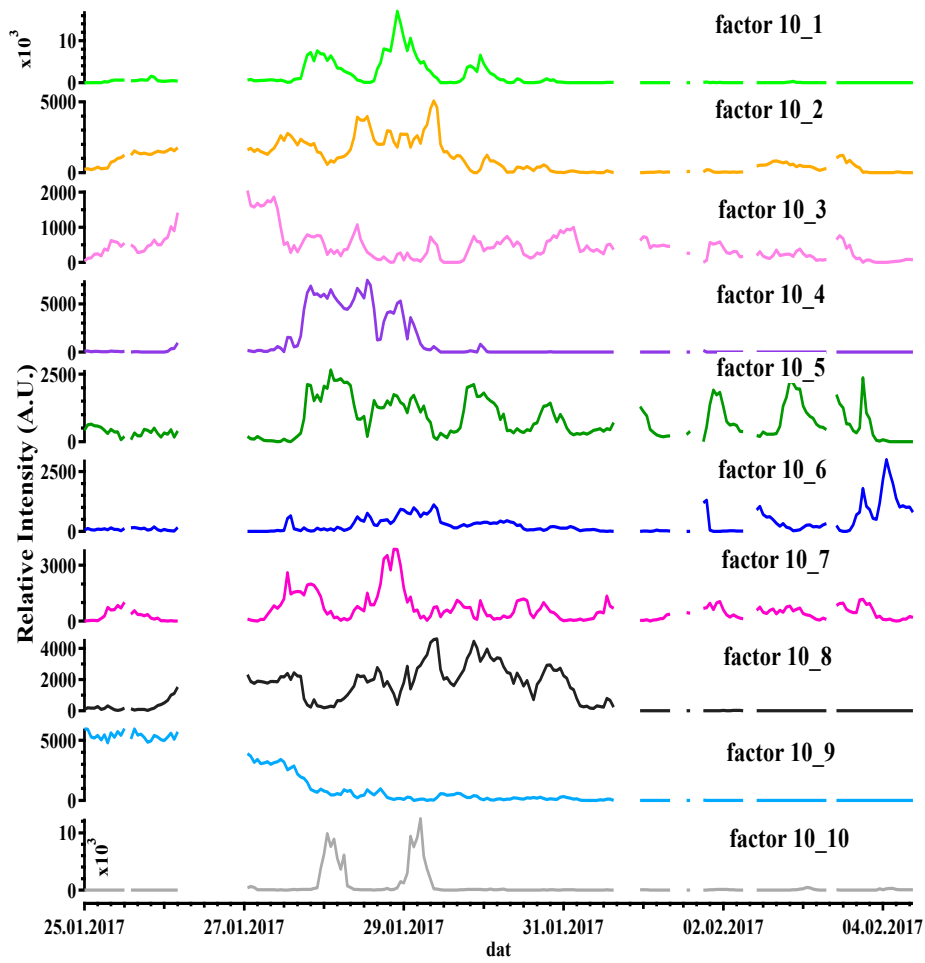
**Fig. S2 a)  $Q/Q_{exp}$  for the unconstrained solutions from 7 to 20 factors. b)  $Q/Q_{exp}$  for the set of final 11-factor solutions (with CSOA constrained).**



a)



b)



c)

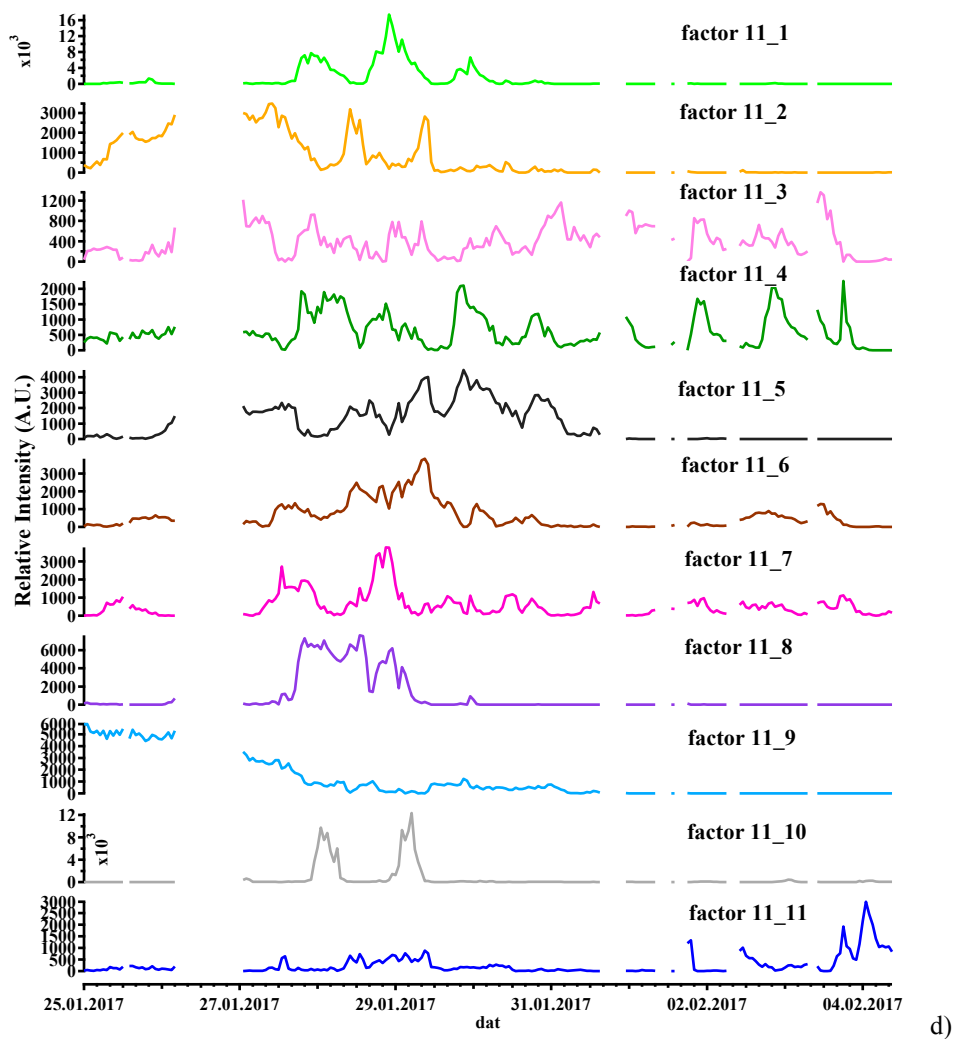
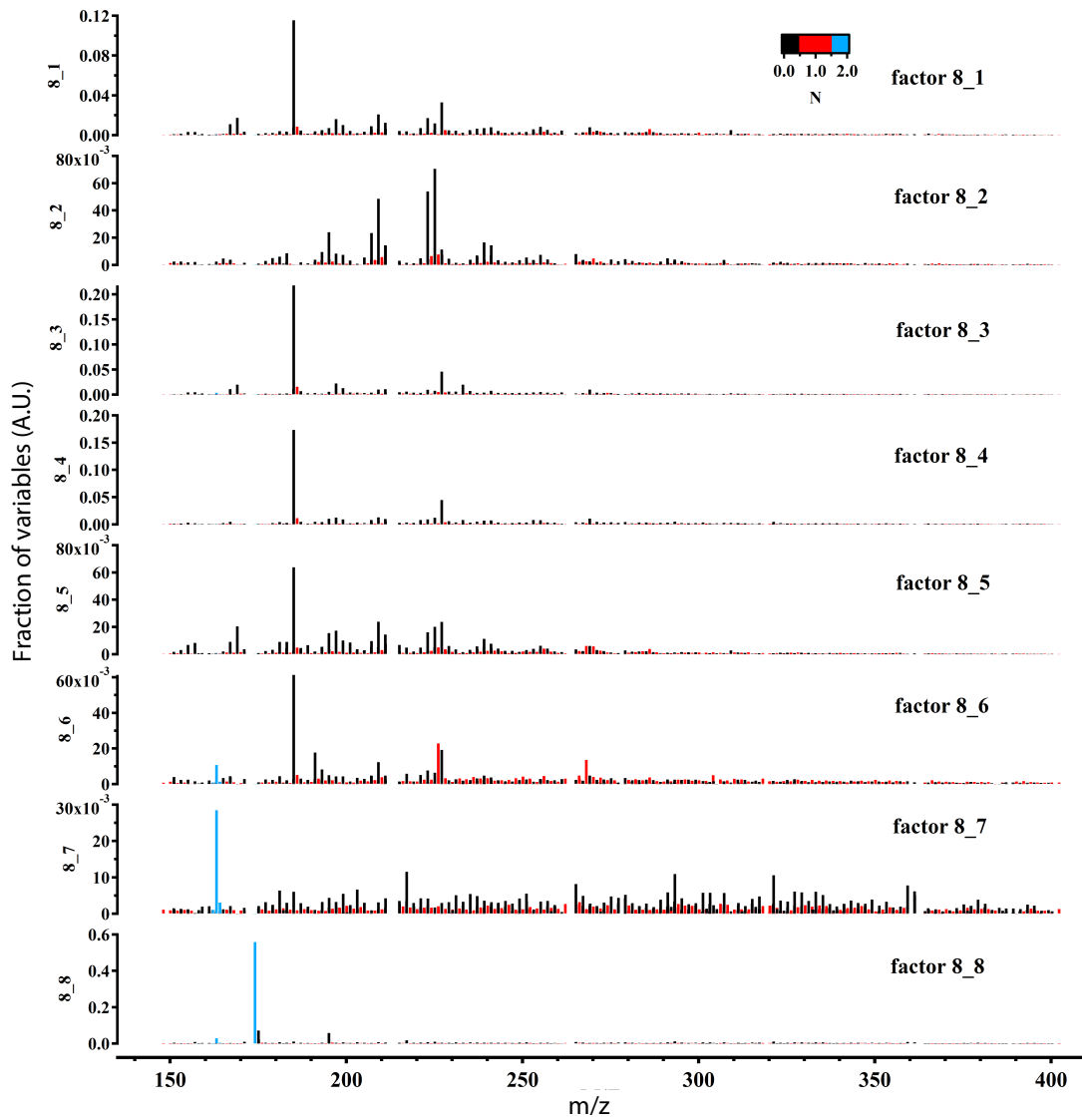
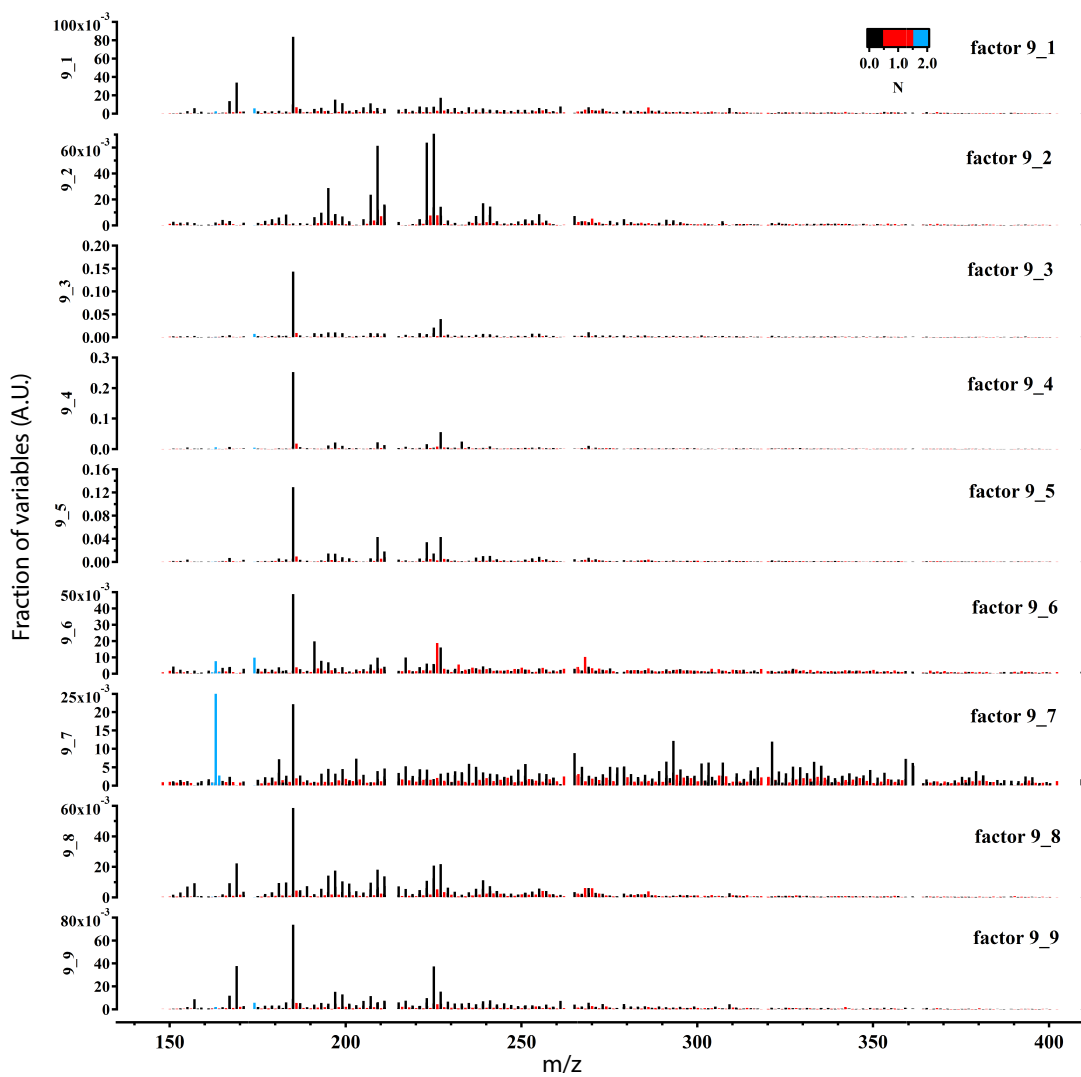


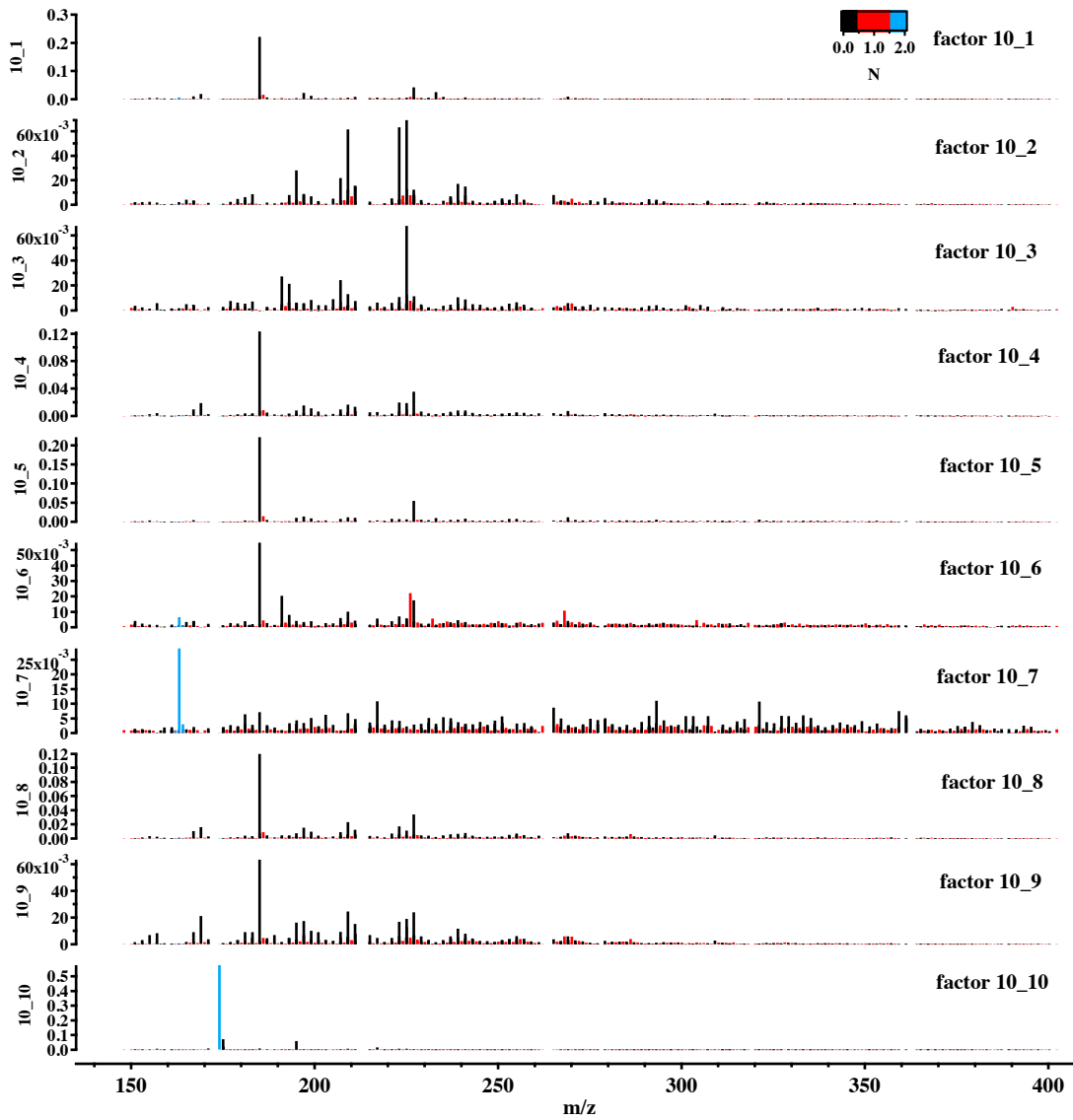
Fig. S3 Time series of solutions for 8 to 11 factors without constraining CSOA.



a)

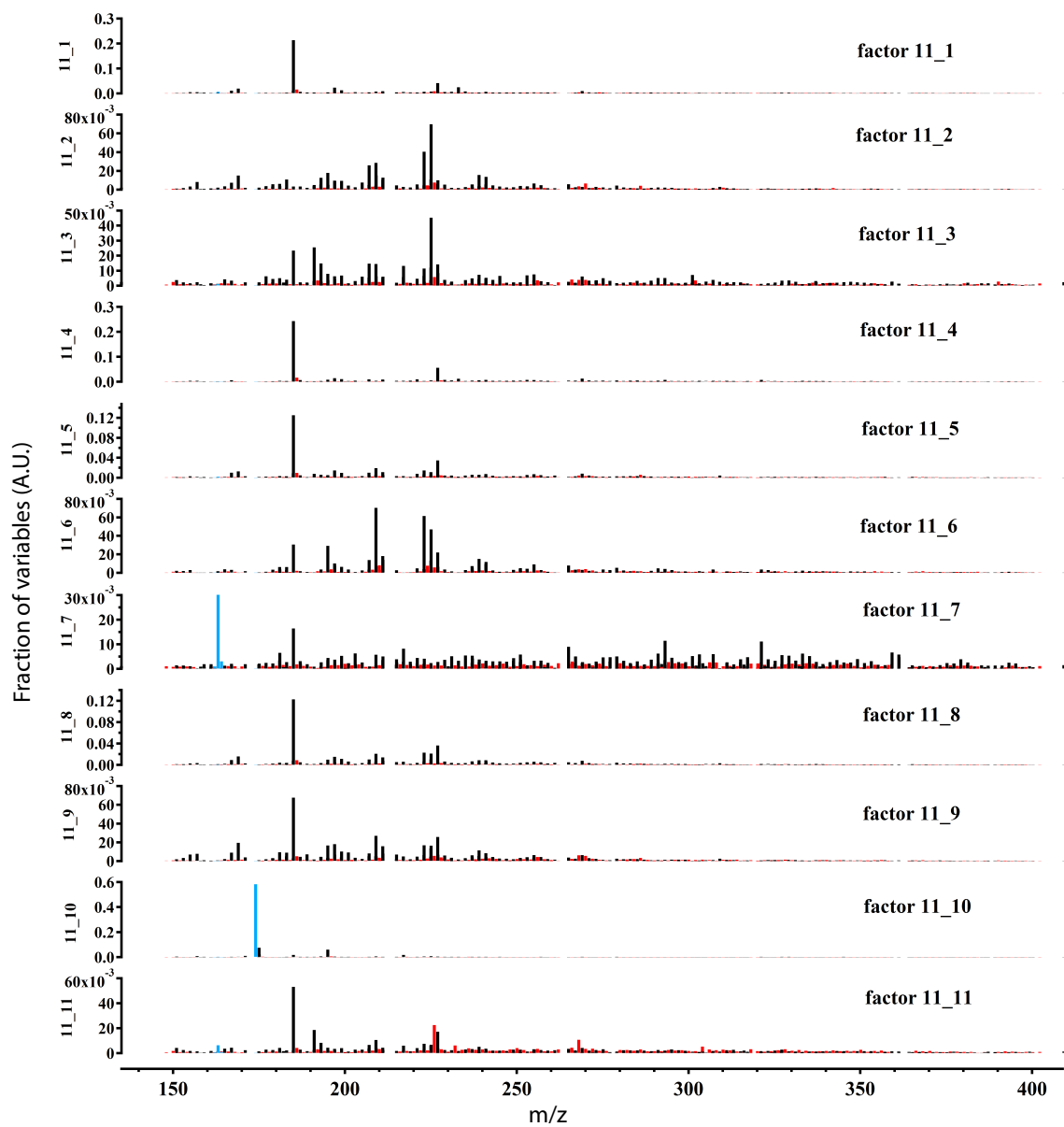


b)



c)





d)

Fig. S4 Profiles of solutions for 8 to 11 factors without constraining CS-OA.

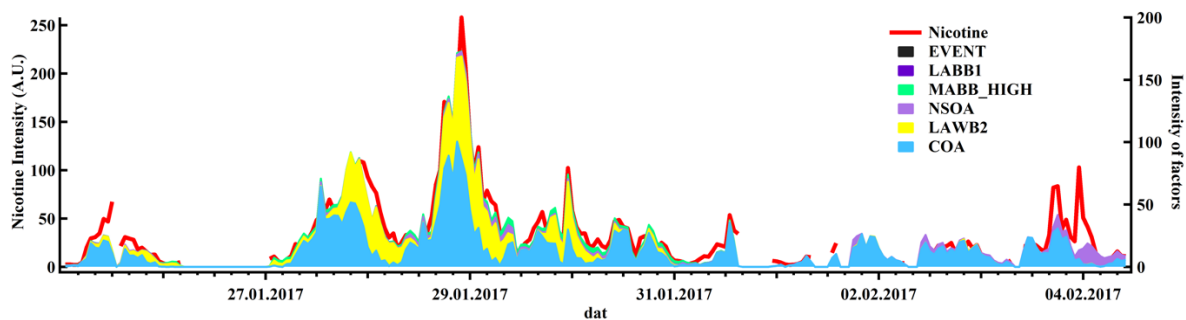


Fig. S5. Comparison between time series of the intensity of nicotine composition and the intensity of contributions of nicotine to each factor (10 factors' solution).

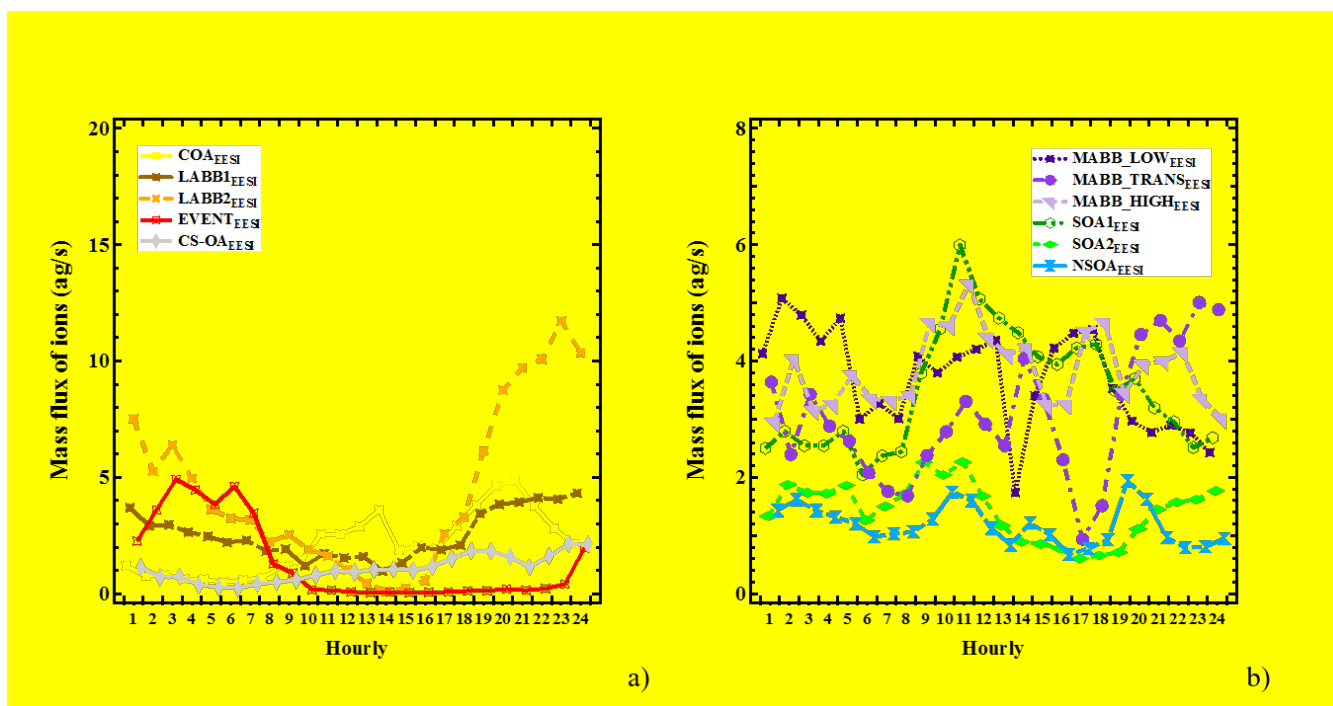


Fig. S6. The diurnal variation of EESI POA factors (a) and EESI SOA factors (b).

Table S1. The O:C, H:C, and N:C values for the EESI and AMS factors (excluding the C<sub>6</sub>H<sub>10</sub>O<sub>5</sub> and C<sub>8</sub>H<sub>12</sub>O<sub>6</sub> ions).

|                 | O:C  | H:C  | N:C  |
|-----------------|------|------|------|
| <b>EESI-TOF</b> |      |      |      |
| LABB1           | 0.35 | 1.04 | 0.02 |
| MABB_LOW        | 0.50 | 1.39 | 0.02 |
| SOA1            | 0.42 | 1.61 | 0.02 |
| COA             | 0.30 | 1.56 | 0.02 |
| MABB_HIGH       | 0.45 | 1.26 | 0.02 |
| EVENT           | 0.23 | 1.44 | 0.15 |
| LABB2           | 0.37 | 1.09 | 0.02 |
| MABB_TRANS      | 0.43 | 1.28 | 0.02 |
| SOA2            | 0.43 | 1.54 | 0.02 |
| NSOA            | 0.39 | 1.41 | 0.04 |
| CS-OA           | 0.31 | 1.51 | 0.05 |
| <b>AMS</b>      |      |      |      |
| HOA             | 0.04 | 1.79 | 0.01 |
| OOA1            | 0.58 | 1.01 | 0.02 |
| OOA2            | 0.84 | 0.84 | 0.02 |
| EVENT           | 0.27 | 1.89 | 0.03 |
| COA             | 0.10 | 1.55 | 0.02 |
| NOA             | 0.43 | 1.33 | 0.05 |
| BBOA            | 0.39 | 1.49 | 0.04 |

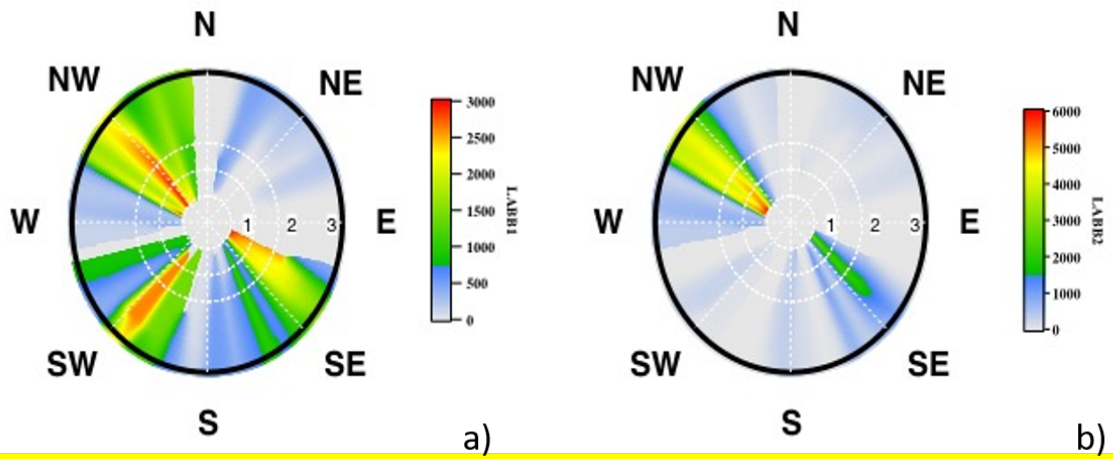


Fig.S7. Wind analysis results using the SWIM model on the concentrations of LABB1 (a) and LABB2 (b).

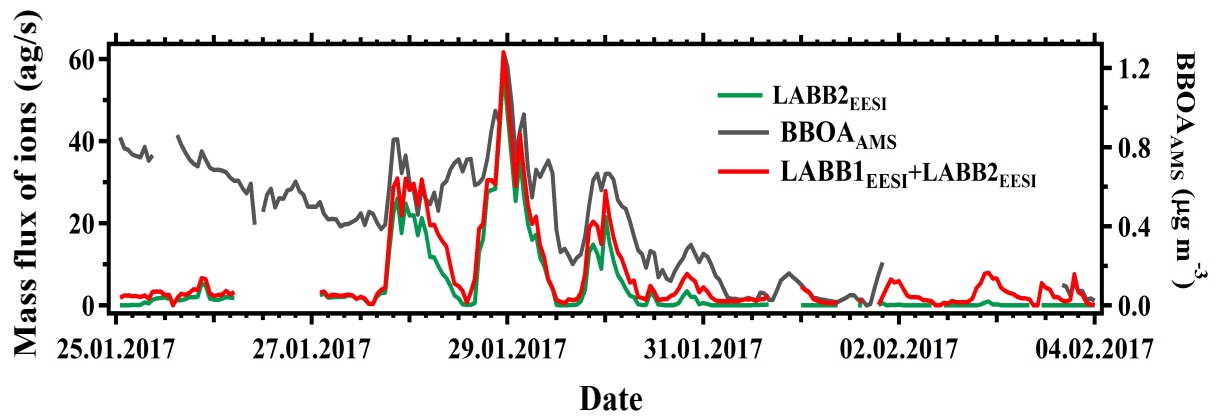


Fig.S8. Comparison between the LABB<sub>EESI</sub> factors and the WBOA<sub>AMS</sub> factor.

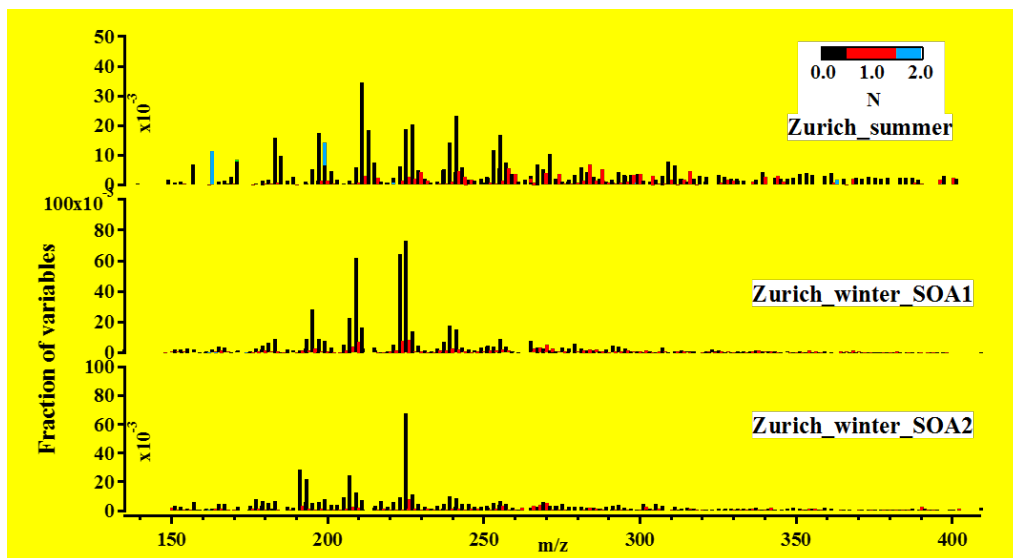
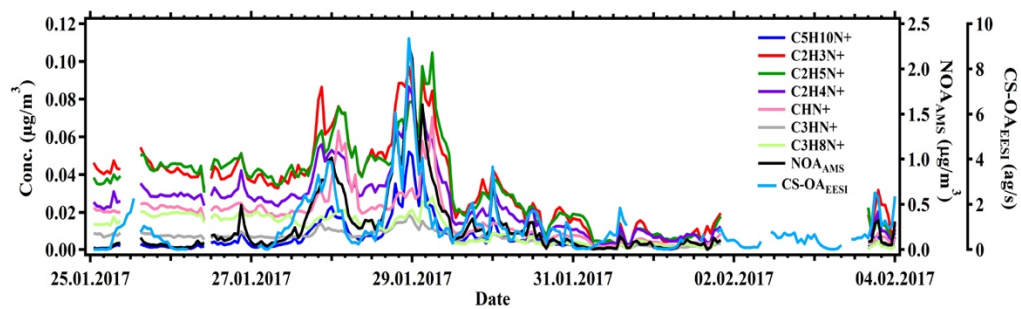


Fig. S9. Mass spectra comparison between SOA factors and reference mass spectra from Zurich summer source apportionment.



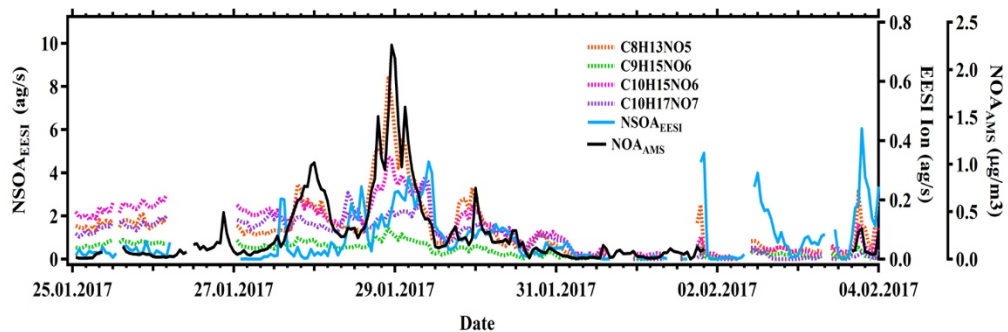
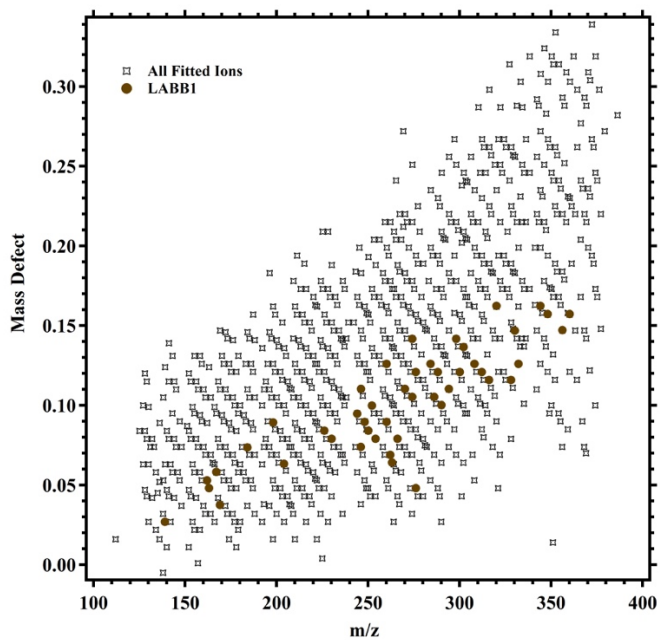
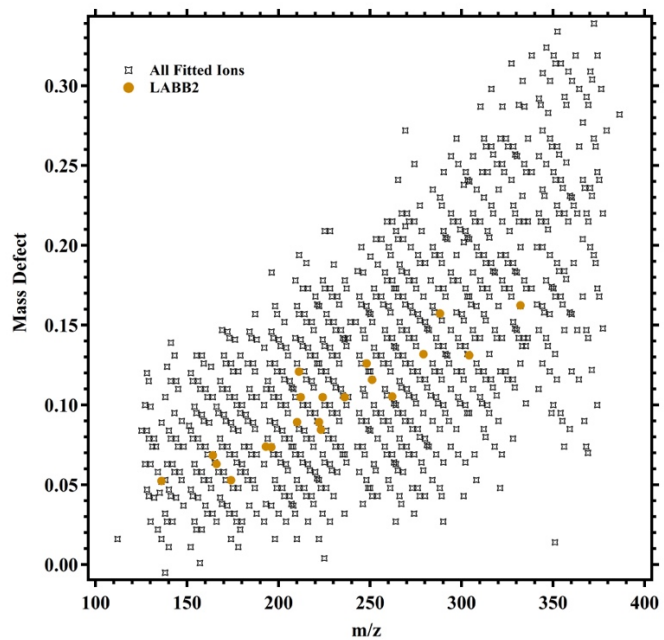
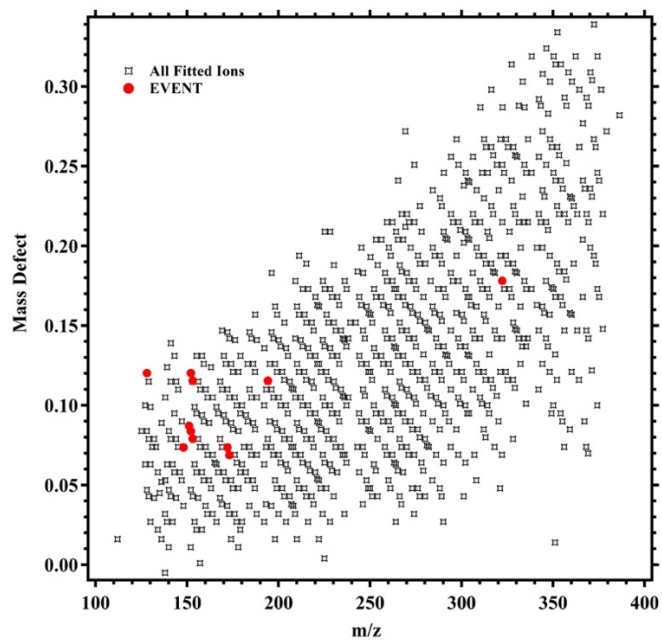
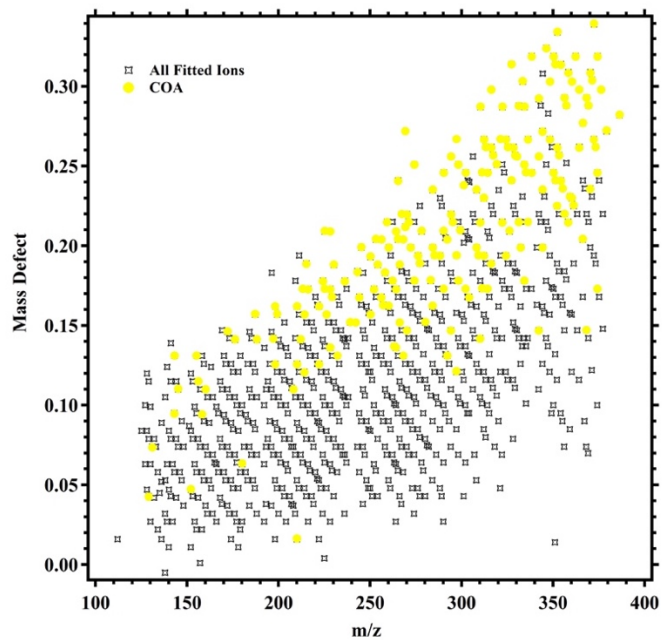


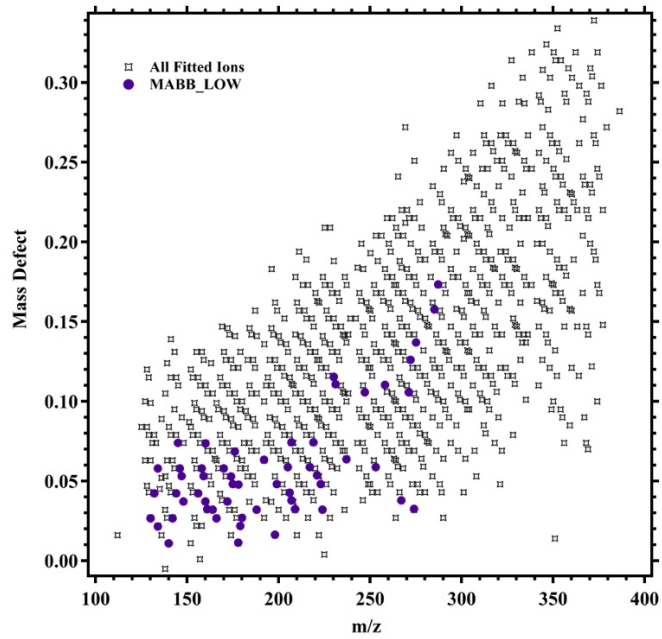
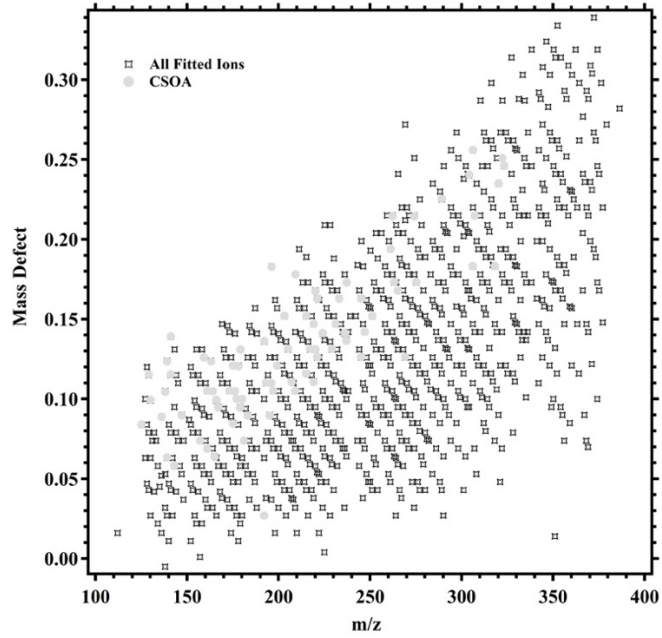
Fig. S10. Time series of fragments group of AMS CHON and AMS CHN compared to the time series of NSOA<sub>EESI</sub> factor and CS-OA<sub>EESI</sub> factor, respectively.

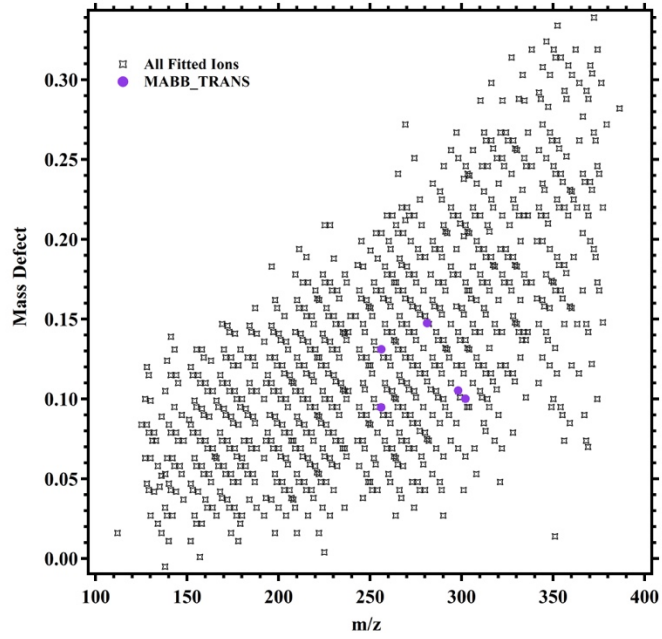


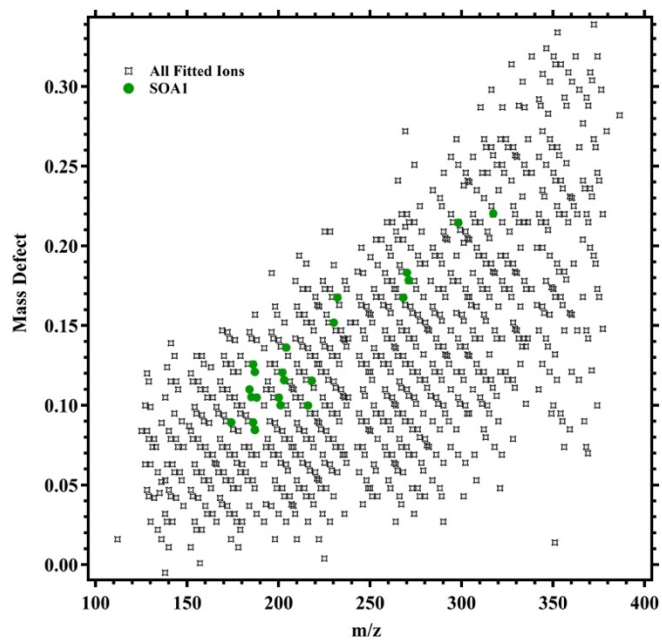
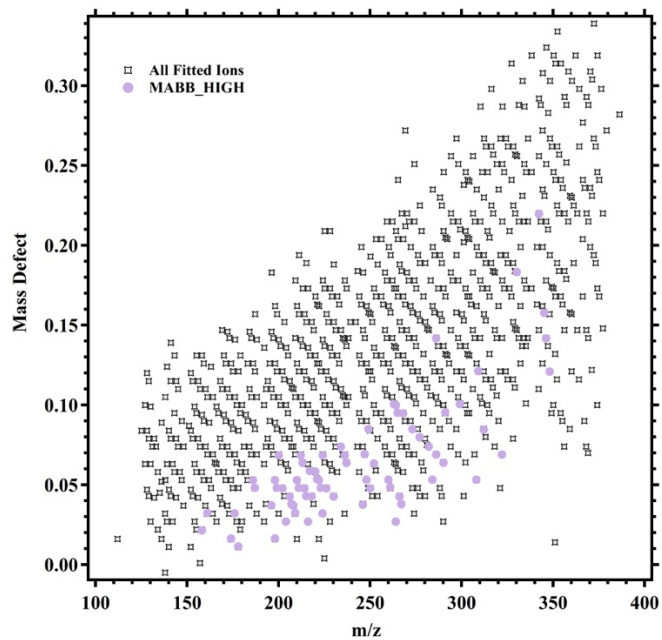


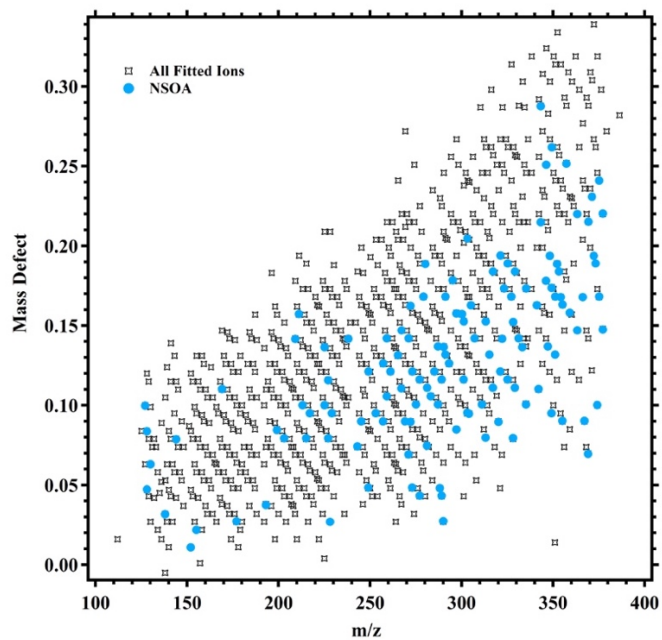
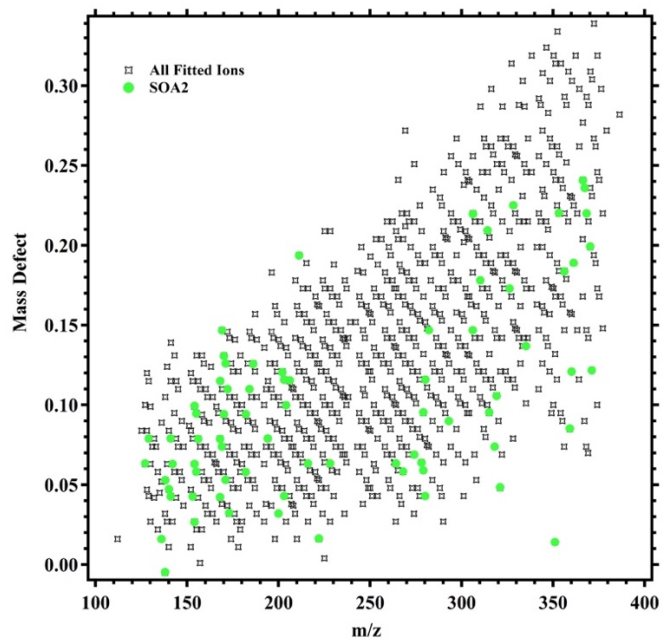












**Fig. S11. Mass defect plots for each factor, with the unique ions marked in colors.**

**Table S2. Tentative formula assignment of MS peaks with m/z values for cluster groups.**

| <b>Factor Name</b> | <b>Ion formula</b> | <b>m/z</b> |
|--------------------|--------------------|------------|
| <b>COA</b>         |                    |            |
|                    | C5H4O3             | 112.016    |
|                    | C5H7NO3            | 129.043    |
|                    | C9H9N              | 131.074    |
|                    | C7H13NO2           | 143.095    |
|                    | C8H17NO            | 143.131    |
|                    | C7H15NO2           | 145.11     |
|                    | C8H8O3             | 152.047    |
|                    | C9H17NO            | 155.131    |
|                    | C9H16O2            | 156.115    |
|                    | C8H14O3            | 158.094    |
|                    | C8H16O3            | 160.11     |
|                    | C10H20O2           | 172.146    |
|                    | C9H19NO2           | 173.142    |
|                    | C9H20O3            | 176.141    |
|                    | C6H12O6            | 180.063    |
|                    | C11H20O2           | 184.146    |
|                    | C10H19NO2          | 185.142    |
|                    | C10H21NO2          | 187.157    |
|                    | C10H20O3           | 188.141    |
|                    | C11H19NO2          | 197.142    |
|                    | C11H18O3           | 198.126    |
|                    | C12H22O2           | 198.162    |
|                    | C11H21NO2          | 199.157    |
|                    | C11H20O3           | 200.141    |
|                    | C12H16O3           | 208.11     |
|                    | C9H6O6             | 210.016    |
|                    | C12H18O3           | 210.126    |
|                    | C13H22O2           | 210.162    |
|                    | C12H20O3           | 212.141    |
|                    | C12H23NO2          | 213.173    |
|                    | C11H18O4           | 214.121    |
|                    | C12H22O3           | 214.157    |
| C12H25NO2          | 215.189            |            |
| C12H24O3           | 216.173            |            |
| C13H18O3           | 222.126            |            |
| C14H22O2           | 222.162            |            |
| C14H24O2           | 224.178            |            |

|           |         |
|-----------|---------|
| C13H23NO2 | 225.173 |
| C14H27NO  | 225.209 |
| C13H22O3  | 226.157 |
| C12H20O4  | 228.136 |
| C13H24O3  | 228.173 |
| C14H28O2  | 228.209 |
| C12H23NO3 | 229.168 |
| C13H26O3  | 230.188 |
| C11H20O5  | 232.131 |
| C15H24O2  | 236.178 |
| C13H21NO3 | 239.152 |
| C13H22O4  | 242.152 |
| C13H25NO3 | 243.184 |
| C13H24O4  | 244.168 |
| C13H27NO3 | 245.199 |
| C15H22O3  | 250.157 |
| C16H26O2  | 250.193 |
| C15H24O3  | 252.173 |
| C15H27NO2 | 253.204 |
| C15H26O3  | 254.188 |
| C14H24O4  | 256.168 |
| C15H28O3  | 256.204 |
| C13H23NO4 | 257.163 |
| C14H26O4  | 258.183 |
| C14H29NO3 | 259.215 |
| C13H24O5  | 260.162 |
| C14H28O4  | 260.199 |
| C13H26O5  | 262.178 |
| C11H21NO6 | 263.137 |
| C15H20O4  | 264.136 |
| C16H24O3  | 264.173 |
| C17H28O2  | 264.209 |
| C16H27NO2 | 265.204 |
| C17H31NO  | 265.241 |
| C15H22O4  | 266.152 |
| C16H26O3  | 266.188 |
| C16H29NO2 | 267.22  |
| C14H20O5  | 268.131 |
| C16H28O3  | 268.204 |
| C16H29O3  | 269.212 |
| C17H35NO  | 269.272 |

|           |         |
|-----------|---------|
| C14H22O5  | 270.147 |
| C16H30O3  | 270.22  |
| C15H29NO3 | 271.215 |
| C15H28O4  | 272.199 |
| C14H26O5  | 274.178 |
| C16H34O3  | 274.251 |
| C14H28O5  | 276.194 |
| C13H27NO5 | 277.189 |
| C14H30O5  | 278.209 |
| C12H24O7  | 280.152 |
| C15H25NO4 | 283.178 |
| C15H24O5  | 284.162 |
| C16H28O4  | 284.199 |
| C17H32O3  | 284.235 |
| C15H27NO4 | 285.194 |
| C15H28O5  | 288.194 |
| C14H26O6  | 290.173 |
| C15H30O5  | 290.209 |
| C16H34O4  | 290.246 |
| C16H20O5  | 292.131 |
| C18H28O3  | 292.204 |
| C16H22O5  | 294.147 |
| C18H30O3  | 294.22  |
| C19H34O2  | 294.256 |
| C17H29NO3 | 295.215 |
| C14H19NO6 | 297.121 |
| C16H27NO4 | 297.194 |
| C18H35NO2 | 297.267 |
| C16H26O5  | 298.178 |
| C18H34O3  | 298.251 |
| C16H29NO4 | 299.21  |
| C17H33O4  | 301.238 |
| C15H26O6  | 302.173 |
| C17H34O4  | 302.246 |
| C18H24O4  | 304.168 |
| C18H28O4  | 308.199 |
| C19H32O3  | 308.235 |
| C17H27NO4 | 309.194 |
| C16H22O6  | 310.142 |
| C18H30O4  | 310.215 |
| C20H38O2  | 310.287 |



|           |         |
|-----------|---------|
| C16H25NO5 | 311.173 |
| C18H33NO3 | 311.246 |
| C17H28O5  | 312.194 |
| C18H32O4  | 312.23  |
| C19H36O3  | 312.267 |
| C18H35NO3 | 313.262 |
| C16H26O6  | 314.173 |
| C18H34O4  | 314.246 |
| C16H28O6  | 316.189 |
| C18H36O4  | 316.262 |
| C19H40O3  | 316.298 |
| C17H35NO4 | 317.257 |
| C20H33NO2 | 319.251 |
| C20H35NO2 | 321.267 |
| C19H30O4  | 322.215 |
| C21H38O2  | 322.287 |
| C18H28O5  | 324.194 |
| C20H36O3  | 324.267 |
| C19H35NO3 | 325.262 |
| C18H30O5  | 326.209 |
| C18H33NO4 | 327.241 |
| C20H41NO2 | 327.314 |
| C19H36O4  | 328.262 |
| C18H35NO4 | 329.257 |
| C22H34O2  | 330.256 |
| C20H29NO3 | 331.215 |
| C22H37NO  | 331.288 |
| C20H28O4  | 332.199 |
| C22H39NO  | 333.303 |
| C19H26O5  | 334.178 |
| C21H34O3  | 334.251 |
| C22H38O2  | 334.287 |
| C20H33NO3 | 335.246 |
| C16H32O7  | 336.215 |
| C21H36O3  | 336.267 |
| C20H34O4  | 338.246 |
| C22H42O2  | 338.319 |
| C20H22O5  | 342.147 |
| C23H34O2  | 342.256 |
| C24H38O   | 342.292 |
| C21H28O4  | 344.199 |

|           |         |
|-----------|---------|
| C22H32O3  | 344.235 |
| C23H36O2  | 344.272 |
| C24H42O   | 346.324 |
| C22H36O3  | 348.267 |
| C23H40O2  | 348.303 |
| C21H34O4  | 350.246 |
| C23H42O2  | 350.319 |
| C20H33NO4 | 351.241 |
| C22H41NO2 | 351.314 |
| C20H32O5  | 352.225 |
| C21H36O4  | 352.262 |
| C23H44O2  | 352.334 |
| C20H35NO4 | 353.257 |
| C20H34O5  | 354.241 |
| C22H42O3  | 354.314 |
| C19H33NO5 | 355.236 |
| C19H32O6  | 356.22  |
| C21H40O4  | 356.293 |
| C20H39NO4 | 357.288 |
| C22H30O4  | 358.215 |
| C21H42O4  | 358.309 |
| C18H33NO6 | 359.231 |
| C22H32O4  | 360.23  |
| C24H40O2  | 360.303 |
| C21H31NO4 | 361.225 |
| C24H42O2  | 362.319 |
| C22H36O4  | 364.262 |
| C23H40O3  | 364.298 |
| C20H30O6  | 366.204 |
| C22H38O4  | 366.277 |
| C18H24O8  | 368.147 |
| C22H40O4  | 368.293 |
| C21H39NO4 | 369.288 |
| C20H34O6  | 370.236 |
| C22H42O4  | 370.309 |
| C21H41NO4 | 371.304 |
| C24H36O3  | 372.267 |
| C26H44O   | 372.339 |
| C23H35NO3 | 373.262 |
| C21H26O6  | 374.173 |
| C23H34O4  | 374.246 |

|              |           |         |
|--------------|-----------|---------|
|              | C25H42O2  | 374.319 |
|              | C24H40O3  | 376.298 |
|              | C22H37NO4 | 379.272 |
|              | C25H38O3  | 386.282 |
| <b>LABBI</b> |           |         |
|              | C6H5NO3   | 139.027 |
|              | C6H10O5   | 162.053 |
|              | C5H9NO5   | 163.048 |
|              | C8H9NO3   | 167.058 |
|              | C7H7NO4   | 169.038 |
|              | C9H12O4   | 184.074 |
|              | C10H14O4  | 198.089 |
|              | C8H12O6   | 204.063 |
|              | C10H14O6  | 230.079 |
|              | C13H14O5  | 250.084 |
|              | C13H16O5  | 252.1   |
|              | C10H14O8  | 262.069 |
|              | C9H13NO8  | 263.064 |
|              | C13H14O6  | 266.079 |
|              | C13H18O6  | 270.11  |
|              | C12H18O7  | 274.105 |
|              | C13H22O6  | 274.142 |
|              | C10H12O9  | 276.048 |
|              | C12H20O7  | 276.121 |
|              | C14H20O6  | 284.126 |
|              | C13H18O7  | 286.105 |
|              | C13H20O7  | 288.121 |
|              | C12H18O8  | 290.1   |
|              | C15H18O6  | 294.11  |
|              | C15H22O6  | 298.142 |
|              | C14H20O7  | 300.121 |
|              | C14H22O7  | 302.137 |
|              | C16H20O6  | 308.126 |
|              | C15H20O7  | 312.121 |
|              | C14H20O8  | 316.116 |
|              | C18H24O5  | 320.162 |
|              | C15H20O8  | 328.116 |
|              | C19H22O5  | 330.147 |
|              | C18H20O6  | 332.126 |
|              | C20H24O5  | 344.162 |
|              | C19H24O6  | 348.157 |

|              |           |         |
|--------------|-----------|---------|
|              | C17H24O8  | 356.147 |
|              | C20H24O6  | 360.157 |
| <b>LABB2</b> |           |         |
|              | C8H8O2    | 136.052 |
|              | C6H12O5   | 164.069 |
|              | C9H10O3   | 166.063 |
|              | C10H11NO3 | 193.074 |
|              | C10H12O4  | 196.074 |
|              | C11H14O4  | 210.089 |
|              | C11H17NO3 | 211.121 |
|              | C11H16O4  | 212.105 |
|              | C12H14O4  | 222.089 |
|              | C11H13NO4 | 223.085 |
|              | C12H16O4  | 224.105 |
|              | C13H16O4  | 236.105 |
|              | C11H20O6  | 248.126 |
|              | C13H17NO4 | 251.116 |
|              | C11H18O7  | 262.105 |
|              | C11H21NO7 | 279.132 |
|              | C14H24O6  | 288.157 |
|              | C17H20O5  | 304.131 |
|              | C19H24O5  | 332.162 |
| <b>EVENT</b> |           |         |
|              | C8H16O    | 128.12  |
|              | C6H12O4   | 148.074 |
|              | C8H11N2O  | 151.087 |
|              | C9H12O2   | 152.084 |
|              | C10H16O   | 152.12  |
|              | C8H11NO2  | 153.079 |
|              | C9H15NO   | 153.115 |
|              | C8H12O4   | 172.074 |
|              | C7H11NO4  | 173.069 |
|              | C8H18O5   | 194.115 |
|              | C11H17NO2 | 195.126 |
|              | C18H26O5  | 322.178 |
| <b>CS-OA</b> |           |         |
|              | C7H11NO   | 125.084 |
|              | C7H15NO   | 129.115 |
|              | C7H14O2   | 130.099 |
|              | C9H12O    | 136.089 |
|              | C9H14O    | 138.105 |

|           |         |
|-----------|---------|
| C7H9NO2   | 139.063 |
| C8H15N2   | 139.124 |
| C8H15NO   | 141.115 |
| C8H17N2   | 141.139 |
| C6H9NO3   | 143.058 |
| C6H13NO3  | 147.09  |
| C7H11NO3  | 157.074 |
| C8H17NO2  | 159.126 |
| C6H11NO4  | 161.069 |
| C7H15NO3  | 161.105 |
| C10H15N2  | 163.124 |
| C7H16O4   | 164.105 |
| C5H11NO5  | 165.064 |
| C10H14O2  | 166.099 |
| C9H13NO2  | 167.095 |
| C8H15NO3  | 173.105 |
| C7H13NO4  | 175.085 |
| C8H17NO3  | 175.121 |
| C8H16O4   | 176.105 |
| C7H15NO4  | 177.1   |
| C8H18O4   | 178.121 |
| C10H13NO2 | 179.095 |
| C11H17NO  | 179.131 |
| C7H16O5   | 180.1   |
| C9H11NO3  | 181.074 |
| C9H13NO3  | 183.09  |
| C6H8O7    | 192.027 |
| C9H20O4   | 192.136 |
| C11H15NO2 | 193.11  |
| C10H13NO3 | 195.09  |
| C11H16O3  | 196.11  |
| C13H24O   | 196.183 |
| C10H15NO3 | 197.105 |
| C10H17NO3 | 199.121 |
| C10H21NO3 | 203.152 |
| C9H19NO4  | 205.131 |
| C8H17NO5  | 207.111 |
| C11H15NO3 | 209.105 |
| C13H23NO  | 209.178 |
| C11H21NO3 | 215.152 |
| C10H19NO4 | 217.131 |

|  |                 |         |
|--|-----------------|---------|
|  | C9H17NO5        | 219.111 |
|  | C10H21NO4       | 219.147 |
|  | C10H20O5        | 220.131 |
|  | C11H24O4        | 220.168 |
|  | C9H19NO5        | 221.126 |
|  | C10H23NO4       | 221.163 |
|  | C13H20O3        | 224.141 |
|  | C11H21NO4       | 231.147 |
|  | C11H23NO4       | 233.163 |
|  | C14H20O3        | 236.141 |
|  | C13H19NO3       | 237.137 |
|  | C14H23NO2       | 237.173 |
|  | C12H23NO4       | 245.163 |
|  | C11H21NO5       | 247.142 |
|  | C14H21NO3       | 251.152 |
|  | C13H27NO4       | 261.194 |
|  | C14H30O4        | 262.215 |
|  | C12H25NO5       | 263.173 |
|  | C15H23NO3       | 265.168 |
|  | C13H19NO5       | 269.126 |
|  | C15H30O4        | 274.215 |
|  | C13H25NO5       | 275.173 |
|  | C15H31NO4       | 289.225 |
|  | C20H32O2        | 304.24  |
|  | C18H26O4        | 306.183 |
|  | C20H34O2        | 306.256 |
|  | C18H29NO3       | 307.215 |
|  | C19H26O4        | 318.183 |
|  | C20H32O3        | 320.235 |
|  | C20H34O3        | 322.251 |
|  | C19H33NO3       | 323.246 |
|  | <b>MABB_LOW</b> |         |
|  | C5H6O4          | 130.027 |
|  | C5H8O4          | 132.042 |
|  | C4H6O5          | 134.022 |
|  | C5H10O4         | 134.058 |
|  | C6H4O4          | 140.011 |
|  | C6H6O4          | 142.027 |
|  | C6H8O4          | 144.042 |
|  | C6H11NO3        | 145.074 |
|  | C6H10O4         | 146.058 |

|           |         |
|-----------|---------|
| C5H9NO4   | 147.053 |
| C5H8O5    | 148.037 |
| C7H8O4    | 156.042 |
| C7H10O4   | 158.058 |
| C6H9NO4   | 159.053 |
| C6H8O5    | 160.037 |
| C7H12O4   | 160.074 |
| C5H7NO5   | 161.032 |
| C5H8O6    | 164.032 |
| C8H6O4    | 166.027 |
| C8H10O4   | 170.058 |
| C7H8O5    | 172.037 |
| C7H10O5   | 174.053 |
| C6H9NO5   | 175.048 |
| C7H12O5   | 176.069 |
| C5H6O7    | 178.011 |
| C6H10O6   | 178.048 |
| C8H5NO4   | 179.022 |
| C5H8O7    | 180.027 |
| C7H8O6    | 188.032 |
| C7H12O6   | 192.063 |
| C8H6O6    | 198.016 |
| C8H9NO5   | 199.048 |
| C7H11NO6  | 205.059 |
| C7H10O7   | 206.043 |
| C6H9NO7   | 207.038 |
| C7H13NO6  | 207.074 |
| C9H7NO5   | 209.032 |
| C8H11NO6  | 217.059 |
| C8H13NO6  | 219.074 |
| C7H11NO7  | 221.054 |
| C10H9NO5  | 223.048 |
| C10H8O6   | 224.032 |
| C11H18O5  | 230.115 |
| C10H17NO5 | 231.111 |
| C11H11NO5 | 237.064 |
| C10H17NO6 | 247.106 |
| C11H11NO6 | 253.059 |
| C12H18O6  | 258.11  |
| C11H9NO7  | 267.038 |
| C12H17NO6 | 271.106 |

|                   |           |         |
|-------------------|-----------|---------|
|                   | C13H20O6  | 272.126 |
|                   | C10H10O9  | 274.032 |
|                   | C12H21NO6 | 275.137 |
|                   | C14H23NO5 | 285.158 |
|                   | C14H25NO5 | 287.173 |
| <b>MABB_TRANS</b> |           |         |
|                   | C12H16O6  | 256.095 |
|                   | C13H20O5  | 256.131 |
|                   | C11H23NO7 | 281.148 |
|                   | C14H18O7  | 298.105 |
|                   | C13H18O8  | 302.1   |
| <b>MABB_HIGH</b>  |           |         |
|                   | C6H6O5    | 158.022 |
|                   | C5H7NO5   | 161.032 |
|                   | C6H6O6    | 174.016 |
|                   | C6H8O6    | 176.032 |
|                   | C5H6O7    | 178.011 |
|                   | C8H10O5   | 186.053 |
|                   | C7H9NO5   | 187.048 |
|                   | C9H8O5    | 196.037 |
|                   | C8H6O6    | 198.016 |
|                   | C9H10O5   | 198.053 |
|                   | C8H9NO5   | 199.048 |
|                   | C9H12O5   | 200.069 |
|                   | C8H10O6   | 202.048 |
|                   | C7H8O7    | 204.027 |
|                   | C7H10O7   | 206.043 |
|                   | C6H9NO7   | 207.038 |
|                   | C10H8O5   | 208.037 |
|                   | C9H7NO5   | 209.032 |
|                   | C10H10O5  | 210.053 |
|                   | C9H9NO5   | 211.048 |
|                   | C10H12O5  | 212.069 |
|                   | C9H11NO5  | 213.064 |
|                   | C9H10O6   | 214.048 |
|                   | C8H9NO6   | 215.043 |
|                   | C8H8O7    | 216.027 |
|                   | C8H11NO6  | 217.059 |
|                   | C8H10O7   | 218.043 |
|                   | C8H12O7   | 220.058 |
|                   | C7H11NO7  | 221.054 |



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| C10H9NO5  | 223.048 |
| C10H8O6   | 224.032 |
| C11H12O5  | 224.069 |
| C10H10O6  | 226.048 |
| C9H10O7   | 230.043 |
| C9H14O7   | 234.074 |
| C12H12O5  | 236.069 |
| C11H11NO5 | 237.064 |
| C9H10O8   | 246.038 |
| C9H13NO7  | 247.069 |
| C9H12O8   | 248.053 |
| C9H15NO7  | 249.085 |
| C12H10O6  | 250.048 |
| C12H12O6  | 252.063 |
| C10H12O8  | 260.053 |
| C9H11NO8  | 261.048 |
| C10H17NO7 | 263.101 |
| C12H8O7   | 264.027 |
| C14H16O5  | 264.1   |
| C13H15NO5 | 265.095 |
| C12H10O7  | 266.043 |
| C11H9NO7  | 267.038 |
| C13H16O6  | 268.095 |
| C11H15NO7 | 273.085 |
| C10H15NO8 | 277.08  |
| C13H14O7  | 282.074 |
| C12H12O8  | 284.053 |
| C12H14O8  | 286.069 |
| C14H22O6  | 286.142 |
| C11H14O9  | 290.064 |
| C11H17NO8 | 291.095 |
| C13H17NO7 | 299.101 |
| C14H12O8  | 308.053 |
| C15H19NO6 | 309.121 |
| C14H16O8  | 312.085 |
| C15H14O8  | 322.069 |
| C20H26O4  | 330.183 |
| C22H30O3  | 342.22  |
| C19H23NO5 | 345.158 |
| C19H22O6  | 346.142 |

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|             | C18H20O7  | 348.121 |
| <b>SOA1</b> |           |         |
|             | C8H14O4   | 174.089 |
|             | C9H15NO3  | 185.105 |
|             | C9H14O4   | 186.089 |
|             | C8H13NO4  | 187.085 |
|             | C9H17NO3  | 187.121 |
|             | C9H16O4   | 188.105 |
|             | C10H16O4  | 200.105 |
|             | C9H15NO4  | 201.1   |
|             | C10H20O4  | 204.136 |
|             | C10H16O5  | 216.1   |
|             | C10H18O5  | 218.115 |
|             | C12H22O4  | 230.152 |
|             | C12H24O4  | 232.168 |
|             | C14H25NO4 | 271.178 |
|             | C17H30O4  | 298.215 |
|             | C16H31NO5 | 317.22  |
| <b>SOA2</b> |           |         |
|             | C6H9NO2   | 127.063 |
|             | C6H11NO2  | 129.079 |
|             | C7H4O3    | 136.016 |
|             | C6H2O4    | 137.995 |
|             | C4H10O5   | 138.053 |
|             | C7H8O3    | 140.047 |
|             | C6H7NO3   | 141.043 |
|             | C7H11NO2  | 141.079 |
|             | C7H10O3   | 142.063 |
|             | C7H7NO3   | 153.043 |
|             | C7H6O4    | 154.027 |
|             | C8H10O3   | 154.063 |
|             | C9H14O2   | 154.099 |
|             | C7H9NO3   | 155.058 |
|             | C8H13NO2  | 155.095 |
|             | C8H12O3   | 156.079 |
|             | C8H8O4    | 168.042 |
|             | C9H12O3   | 168.079 |
|             | C10H16O2  | 168.115 |
|             | C8H11NO3  | 169.074 |
|             | C10H19NO  | 169.147 |
|             | C9H14O3   | 170.094 |

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| C10H18O2  | 170.131 |
| C7H9NO4   | 171.053 |
| C9H17NO2  | 171.126 |
| C9H16O3   | 172.11  |
| C6H7NO5   | 173.032 |
| C9H10O4   | 182.058 |
| C10H14O3  | 182.094 |
| C8H8O5    | 184.037 |
| C10H16O3  | 184.11  |
| C10H18O3  | 186.126 |
| C7H14O6   | 194.079 |
| C8H8O6    | 200.032 |
| C10H18O4  | 202.121 |
| C7H9NO6   | 203.043 |
| C9H17NO4  | 203.116 |
| C9H16O5   | 204.1   |
| C9H18O5   | 206.115 |
| C13H25NO  | 211.194 |
| C9H12O6   | 216.063 |
| C10H6O6   | 222.016 |
| C9H5O7    | 225.004 |
| C10H12O6  | 228.063 |
| C9H11NO6  | 229.059 |
| C9H12O7   | 232.058 |
| C10H18O6  | 234.11  |
| C11H8O6   | 236.032 |
| C10H12O7  | 244.058 |
| C11H10O7  | 254.043 |
| C11H14O7  | 258.074 |
| C13H12O6  | 264.063 |
| C12H12O7  | 268.058 |
| C15H24O4  | 268.168 |
| C12H14O7  | 270.074 |
| C15H26O4  | 270.183 |
| C11H14O8  | 274.069 |
| C10H14O9  | 278.064 |
| C9H13NO9  | 279.059 |
| C10H17NO8 | 279.095 |
| C9H12O10  | 280.043 |
| C11H20O8  | 280.116 |
| C15H22O5  | 282.147 |

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|  | C13H16O7    | 284.09  |
|  | C14H15NO6   | 293.09  |
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|  | C17H22O5    | 306.147 |
|  | C19H30O3    | 306.22  |
|  | C15H18O7    | 310.105 |
|  | C17H26O5    | 310.178 |
|  | C16H24O6    | 312.157 |
|  | C14H18O8    | 314.1   |
|  | C17H30O5    | 314.209 |
|  | C13H17NO8   | 315.095 |
|  | C16H14O7    | 318.074 |
|  | C16H17NO6   | 319.106 |
|  | C14H11NO8   | 321.048 |
|  | C17H26O6    | 326.173 |
|  | C18H32O5    | 328.225 |
|  | C15H23NO7   | 329.148 |
|  | C17H21NO6   | 335.137 |
|  | C18H26O6    | 338.173 |
|  | C19H20O6    | 344.126 |
|  | C18H7O8     | 351.014 |
|  | C19H31NO5   | 353.22  |
|  | C16H18O9    | 354.095 |
|  | C15H16O10   | 356.074 |
|  | C18H28O7    | 356.184 |
|  | C14H17NO10  | 359.085 |
|  | C19H20O7    | 360.121 |
|  | C20H27NO5   | 361.189 |
|  | C21H34O5    | 366.241 |
|  | C20H33NO5   | 367.236 |
|  | C20H32O6    | 368.22  |
|  | C17H23NO8   | 369.142 |
|  | C19H30O7    | 370.199 |
|  | C16H21NO9   | 371.122 |
|  | <b>NSOA</b> |         |
|  | C7H13NO     | 127.1   |
|  | C6H8O3      | 128.047 |
|  | C7H12O2     | 128.084 |
|  | C6H10O3     | 130.063 |
|  | C7H6O3      | 138.032 |
|  | C7H12O3     | 144.079 |

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| C7H4O4    | 152.011 |
| C6H5NO4   | 155.022 |
| C9H15NO2  | 169.11  |
| C5H7NO6   | 177.027 |
| C8H9NO4   | 183.053 |
| C8H11NO4  | 185.069 |
| C9H7NO4   | 193.038 |
| C9H11NO4  | 197.069 |
| C9H13NO4  | 199.085 |
| C8H11NO5  | 201.064 |
| C8H13NO5  | 203.079 |
| C8H15NO5  | 205.095 |
| C12H19NO2 | 209.142 |
| C10H13NO4 | 211.085 |
| C12H21NO2 | 211.157 |
| C10H15NO4 | 213.1   |
| C11H19NO3 | 213.137 |
| C9H13NO5  | 215.079 |
| C10H17NO4 | 215.116 |
| C9H15NO5  | 217.095 |
| C7H9NO7   | 219.038 |
| C8H15NO6  | 221.09  |
| C12H17NO3 | 223.121 |
| C10H11NO5 | 225.064 |
| C11H15NO4 | 225.1   |
| C12H19NO3 | 225.137 |
| C12H18O4  | 226.121 |
| C10H13NO5 | 227.079 |
| C11H17NO4 | 227.116 |
| C9H8O7    | 228.027 |
| C10H15NO5 | 229.095 |
| C11H19NO4 | 229.131 |
| C9H15NO6  | 233.09  |
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| C10H21NO5 | 235.142 |
| C12H15NO4 | 237.1   |
| C10H22O6  | 238.142 |
| C10H13NO6 | 243.074 |
| C12H20O5  | 244.131 |
| C10H15NO6 | 245.09  |
| C11H19NO5 | 245.126 |

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| C8H11NO8  | 249.048 |
| C10H19NO6 | 249.121 |
| C14H20O4  | 252.136 |
| C12H15NO5 | 253.095 |
| C13H19NO4 | 253.131 |
| C11H15NO6 | 257.09  |
| C12H19NO5 | 257.126 |
| C13H22O5  | 258.147 |
| C10H13NO7 | 259.069 |
| C11H17NO6 | 259.106 |
| C12H21NO5 | 259.142 |
| C11H19NO6 | 261.121 |
| C12H23NO5 | 261.158 |
| C12H22O6  | 262.142 |
| C12H11NO6 | 265.059 |
| C14H19NO4 | 265.131 |
| C12H13NO6 | 267.074 |
| C13H17NO5 | 267.111 |
| C14H21NO4 | 267.147 |
| C12H15NO6 | 269.09  |
| C11H13NO7 | 271.069 |
| C13H21NO5 | 271.142 |
| C12H16O7  | 272.09  |
| C14H24O5  | 272.162 |
| C10H11NO8 | 273.048 |
| C12H19NO6 | 273.121 |
| C11H17NO7 | 275.101 |
| C9H11NO9  | 277.043 |
| C11H19NO7 | 277.116 |
| C14H31NO4 | 277.225 |
| C12H25NO6 | 279.168 |
| C13H28O6  | 280.189 |
| C9H15NO9  | 281.075 |
| C10H19NO8 | 281.111 |
| C13H17NO6 | 283.106 |
| C13H19NO6 | 285.121 |
| C12H17NO7 | 287.101 |
| C13H21NO6 | 287.137 |
| C11H12O9  | 288.048 |
| C10H11NO9 | 289.043 |
| C12H19NO7 | 289.116 |

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| C13H23NO6 | 289.153 |
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| C13H22O7  | 290.137 |
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| C13H25NO6 | 291.168 |
| C14H29NO5 | 291.205 |
| C15H19NO5 | 293.126 |
| C14H14O7  | 294.074 |
| C16H25NO4 | 295.178 |
| C13H15NO7 | 297.085 |
| C15H23NO5 | 297.158 |
| C14H21NO6 | 299.137 |
| C15H24O6  | 300.157 |
| C11H11NO9 | 301.043 |
| C13H19NO7 | 301.116 |
| C14H23NO6 | 301.153 |
| C16H29O5  | 301.202 |
| C12H14O9  | 302.064 |
| C12H17NO8 | 303.095 |
| C15H29NO5 | 303.205 |
| C16H16O6  | 304.095 |
| C17H23NO4 | 305.163 |
| C14H13NO7 | 307.069 |
| C16H21NO5 | 307.142 |
| C14H17NO7 | 311.101 |
| C13H15NO8 | 313.08  |
| C15H23NO6 | 313.153 |
| C16H27NO5 | 313.189 |
| C14H21NO7 | 315.132 |
| C16H29NO5 | 315.205 |
| C13H19NO8 | 317.111 |
| C15H27NO6 | 317.184 |
| C16H16O7  | 320.09  |
| C16H19NO6 | 321.121 |
| C18H27NO4 | 321.194 |
| C17H22O6  | 322.142 |
| C17H25NO5 | 323.173 |
| C15H19NO7 | 325.116 |
| C17H27NO5 | 325.189 |
| C16H25NO6 | 327.168 |
| C14H16O9  | 328.079 |

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| C16H24O7  | 328.152 |
| C14H19NO8 | 329.111 |
| C16H27NO6 | 329.184 |
| C18H21NO5 | 331.142 |
| C21H19NO3 | 333.137 |
| C20H31NO3 | 333.231 |
| C20H30O4  | 334.215 |
| C16H17NO7 | 335.101 |
| C18H25NO5 | 335.173 |
| C20H23NO4 | 341.163 |
| C19H18O6  | 342.11  |
| C21H29NO3 | 343.215 |
| C23H37NO  | 343.288 |
| C21H31NO3 | 345.231 |
| C20H26O5  | 346.178 |
| C22H34O3  | 346.251 |
| C18H21NO6 | 347.137 |
| C20H29NO4 | 347.21  |
| C22H37NO2 | 347.283 |
| C20H28O5  | 348.194 |
| C20H15NO5 | 349.095 |
| C15H27NO8 | 349.174 |
| C21H35NO3 | 349.262 |
| C17H18O8  | 350.1   |
| C19H26O6  | 350.173 |
| C17H21NO7 | 351.132 |
| C18H25NO6 | 351.168 |
| C17H20O8  | 352.116 |
| C19H28O6  | 352.189 |
| C18H27NO6 | 353.184 |
| C18H26O7  | 354.168 |
| C15H17NO9 | 355.09  |
| C17H25NO7 | 355.163 |
| C19H35NO5 | 357.252 |
| C16H25NO8 | 359.158 |
| C22H21NO4 | 363.147 |
| C24H29NO2 | 363.22  |
| C18H20O8  | 364.116 |
| C19H26O7  | 366.168 |
| C16H17NO9 | 367.09  |
| C16H16O10 | 368.074 |



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| C15H15NO10 | 369.07  |
| C19H31NO6  | 369.215 |
| C19H33NO6  | 371.231 |
| C22H28O5   | 372.194 |
| C21H27NO5  | 373.189 |
| C19H18O8   | 374.1   |
| C20H25NO6  | 375.168 |
| C22H33NO4  | 375.241 |
| C19H23NO7  | 377.148 |
| C21H31NO5  | 377.22  |