



1 **High resolution vertical distribution and sources of HONO and NO₂ in the**
2 **nocturnal boundary layer in urban Beijing, China**

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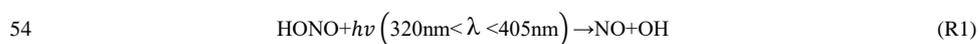
20 **Abstract.** The production of HONO on aerosol surfaces and ground surfaces in urban atmosphere is of
21 interests. However, ground surface measurement commonly in our society is not able to distinguish
22 these two parts. Here, for the first time, we reported high-resolution vertical profile measurements of
23 HONO and NO₂ in urban Beijing at night using an incoherent broadband cavity enhanced absorption
24 spectrometer (IBBCEAS) amounted on a movable container which attached to a meteorological tower
25 of 325 m high. The mixing ratios of HONO during one haze episode (E1), the clean episode (C2) and
26 another haze episode (E3) were 4.26 ± 2.08 , 0.83 ± 0.65 , and 3.54 ± 0.91 ppb, respectively.
27 High-resolution vertical profiles revealed that the vertical distribution of HONO is consistent with
28 stratification and layering in the nocturnal urban atmosphere below 250 m. Direct emissions from
29 combustion processes contributed 51.1% to ambient HONO concentration at night. The HONO
30 production from the heterogeneous conversion of NO₂ on the aerosol surfaces cannot explain HONO



31 vertical measurements at night, indicating that the heterogeneous reaction of NO₂ on ground surfaces
32 dominated the nocturnal HONO production. The nocturnal HONO in the boundary layer is primarily
33 derived from the heterogeneous conversion of NO₂ at ground level and direct emissions; it is then
34 transported throughout the column by vertical convection. $\phi_{NO_2 \rightarrow HONO}$, the HONO yield from
35 deposited NO₂, is used to evaluate HONO production from the heterogeneous conversion of NO₂ at
36 night. The derived $\phi_{NO_2 \rightarrow HONO}$ values on 9 (C2), 10 (C2) and 11 December (E3) were 0.10, 0.08, and
37 0.09, respectively, indicating a significant production of HONO from heterogeneous reaction of NO₂ at
38 ground level. The similar $\phi_{NO_2 \rightarrow HONO}$ values measured during clean and haze episodes suggest that
39 the heterogeneous conversion potential of NO₂ at ground level is consistent at night. Furthermore, the
40 dry deposition loss of HONO to the ground surface and vertical mixing effects associated with
41 convection reached a near steady state at midnight on 11-12 December, indicating that significant
42 quantities of HONO are deposited to the ground surface at night, and the ground surface is the source
43 and sink of HONO at night.

44 **1 Introduction**

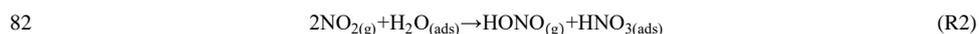
45 It is well known that the rapid photolysis of HONO (R1) after sunrise is the most important
46 hydroxyl radical (OH) source. 25-90% of daytime OH production was accounted by HONO photolysis
47 according to previous reports (Lu et al., 2012; Ma et al., 2017; Tong et al., 2016; Su et al., 2008b; Huang
48 et al., 2017; Hendrick et al., 2014). OH initiates daytime photochemistry and promotes the formation of
49 secondary products (including ozone (O₃), peroxyacetyl nitrate (PAN)) and secondary aerosols (Alicke
50 and Platt, 2002; Tang et al., 2015; Kleffmann, 2007; An et al., 2012). Besides, there is a growing concern
51 about the possible health effects of the formation of nitrosamines (Hanst et al., 1977; Pitts et al., 1978),
52 in which HONO acts as a nitrosating agent to form carcinogenic nitrosamines (Sleiman et al.,
53 2010; Bartolomei et al., 2015; Gómez Alvarez et al., 2014).



55 Despite the importance of HONO, the details of the formation processes of HONO in the
56 atmosphere remain unclear. Newly available instruments have observed much higher daytime HONO
57 concentrations than simulated values of atmospheric chemical models in both rural and urban areas,
58 implying missing HONO sources (Li et al., 2012; Wang et al., 2017; Oswald et al., 2015; Wong et al.,
59 2012; Li et al., 2014; Liu et al., 2019; Karamchandani et al., 2015; Kleffmann, 2007; Mendez et al.,



60 2017;Michoud et al., 2014;Michoud et al., 2015;Tang et al., 2015;Vogela et al., 2003;Sörgel et al.,
61 2011). Several homogeneous reaction mechanisms for HONO have been proposed but latter have been
62 considered as irrelevant in real atmospheric conditions, including photolysis of ortho-substituted
63 nitroaromatics (Bejan et al., 2006) and the reaction of photoexcited NO₂ with H₂O (Li et al., 2008). In
64 contrast to the homogeneous formation of HONO, the heterogeneous conversion from NO₂ to HONO
65 on humid surfaces (R2) is considered the most likely explanation for the observed HONO
66 concentrations. However, whether the conversion processes primarily occurs on the ground surface or
67 on aerosol surface remains controversial. (Kleffmann et al., 2003;Su et al., 2008a;Cui et al., 2018;R.
68 Bröske et al., 2003;Acker et al., 2006;VandenBoer et al., 2013;Bao et al., 2018;Liu et al., 2014;Meusel
69 et al., 2016;Reisinger, 2000;Tong et al., 2016;Ye et al., 2017). A lot of ground measurements have
70 found significantly positive correlations between HONO and aerosol surface area, suggesting that
71 aerosol surfaces play an important role in the heterogeneous conversion from NO₂ to HONO (Reisinger,
72 2000;Cui et al., 2018;Zhang et al., 2018;Hou et al., 2016;Tong et al., 2016;An et al., 2012). Various
73 chemical compositions both in the soil and on aerosol particle are reactive towards HONO production.
74 Recent laboratory studies have found that reaction of NO₂ and H₂O adsorbed on mineral dust, glass and
75 buildings can produce HONO (Finlayson-Pitts et al., 2003;Ma et al., 2017;Mendez et al., 2017). And
76 the redox reaction of NO₂ has been extensively investigated on the surfaces of soot, mineral dust, and
77 humic acid (Kleffmann et al., 1999;Aubin and Abbatt, 2007;Scharko et al., 2017;Ma et al., 2017). The
78 photosensitized conversion of NO₂ on organic surfaces and emission from biological processes have
79 also been suggested to be a potentially important source of HONO in forested and agricultural region
80 (Gómez Alvarez et al., 2014;Stemmler et al., 2006;Monge et al., 2010;Han et al., 2016;Han et al.,
81 2017;Su et al., 2011;Oswald et al., 2013;Tang et al., 2019;Laufs et al., 2017):



82
83 While vertical gradients provide direct evidence about the ground and *in situ* HONO formation.
84 The long-path differential optical absorption spectroscopy (LP-DOAS), instruments mounted on the
85 movable elevator of a tall tower or the fixed height of a building and aircraft measurements are the
86 primary HONO vertical gradient measurement methods (Kleffmann et al., 2003;Stutz,
87 2002;VandenBoer et al., 2013;Villena et al., 2011;Wong et al., 2011;Wong et al., 2012;Zhang et al.,
88 2009;Li X et al., 2017;Ye et al., 2018;Li et al., 2014;Oswald et al., 2015;Kleffmann, 2007). Wong et al.
89 (2011) measured the vertical gradients of HONO using LP-DOAS instrument in downtown Houston.



90 The observations showed that NO₂-to-HONO conversion on the ground was the dominant source of
91 HONO, and vertical transport was found to be the primary source of HONO aloft. Similar results were
92 found by Villena et al. (2011), who conducted HONO vertical gradient measurements on the 3rd and
93 21st floor of the high-rise building (6 and 53 m above ground level) in Santiago de Chile. However,
94 these that have been conducted are limited by measurement frequency and vertical resolution between
95 surface and planetary boundary layer (PBL). VandenBoer et al (2013) carried out measurements of high
96 resolution vertical profiles (vertical resolution ~10m over 250m) of HONO on a 300 m triangular
97 open-frame tower. They found that the ground surface HONO production was dominated by
98 heterogeneous uptake of NO₂, and nocturnally deposited HONO may form a conservative surface
99 reservoir which is released on the following day. However, the above HONO vertical gradient
100 observations were predominantly conducted in the lower PBL. In order to understand the important
101 role of HONO photochemistry in troposphere, Zhang et al. (2009) measured HONO vertical profiles
102 using aircraft in the PBL and the lower free troposphere (FT) over a forested region in northern
103 Michigan. The study also found that the ground surface was a major source of HONO in the lower PBL,
104 and the HONO emitted from ground surfaces accounted for 16-27% of the overall HONO in the PBL,
105 most of which was distributed in the lower PBL.

106 Based on the measurement of vertical gradient and HONO flux, most studies suggests the
107 dominate role of HONO formation on the ground surface (Kleffmann et al., 2003;Oswald et al.,
108 2015;VandenBoer et al., 2013;Villena et al., 2011;Wong et al., 2011;Wong et al., 2013;Stutz,
109 2002;Zhang et al., 2009;Meusel et al., 2016;Neuman et al., 2016;Harrison and Kitto, 1994;Harrison et
110 al., 1996), but Cui et al. (2018) also reported evidence for the important role of HONO formation on
111 aerosol. The aerosol mass loading and its chemical composition seem to be a key parameter influencing
112 the relative importance of HONO formation on aerosol surface.

113 Beijing, as the largest and the most densely populated city in China, has suffered from severe haze
114 pollution for several years due to rapid economic development and urbanization. The aerosol surface
115 density has been reported to be 2-3 orders of magnitude higher than typical background area (Cai et al.,
116 2017;Liu et al., 2012;Zhang et al., 2015). And several studies indicated that the increase of secondary
117 aerosols were caused by high levels of OH from photolysis of HONO, which resulted in frequent
118 occurrence of haze pollution (Fu et al., 2019;An et al., 2012;Huang et al., 2014). Several observations
119 of HONO have been conducted in urban and suburban areas of Beijing in recent years (Tong et al.,



120 2016;Zhang et al., 2018;Hou et al., 2016;Wang et al., 2017;Lu et al., 2012;Hendrick et al., 2014).
121 Higher levels of HONO have been observed (up to 9.71 ppb) in Beijing in winter (Spataro et al., 2013)
122 and the contribution of HONO photolysis to OH budget at noon can reach 90% in Beijing in winter
123 (Hendrick et al., 2014). However, most of these studies were conducted at ground level, while the
124 vertical measurements are very limited in Beijing, China.

125 Here, we report the first high-resolution vertical profile measurements of HONO and NO₂ in the
126 megacity of Beijing at different pollution levels (following the transition from the clean episode to haze
127 episode). The vertical profiles of HONO and NO₂ were measured at high vertical resolution (< 2.5 m
128 over 240 m height) between the surface, the nocturnal boundary layer and residual layer. Although the
129 vertical profile measurements are rather limited in scope, including only four nights in December 2016
130 with limited ancillary data, it is unique both because of its high vertical resolution and because of the
131 continuous vertical measurements of HONO and NO₂ at different stage of pollution. Based on the
132 dataset, the heterogeneous formation of HONO on the ground surfaces and aerosol surfaces is
133 investigated. Evidences also suggest HONO deposition at certain atmospheric conditions.

134 **2 Experimental Methods**

135 **2.1 Measurement site**

136 Vertical profile measurements were conducted from 7 to 12 December 2016 at the Tower Branch
137 of the Institute of Atmospheric Physics (IAP), Chinese Academy of Science (39°58'N, 116°23'E) as
138 part of the “In-depth study of air pollution sources and processes within Beijing and its surrounding
139 region (APHH-Beijing)” winter campaign. The site is a typical urban residential area between the 3rd
140 and 4th Ring Road in the north of Beijing. It is approximately 1 km from the 3rd Ring Road; 200 m
141 from the Beijing-Tibet Expressway and 50 m from the Beitucheng West Road (Fig. S1). The main
142 sampling platform is the Beijing 325-m meteorological tower (BMT), equipped with an external
143 container which can ascend and descend at a relatively constant rate of ~ 9 m min⁻¹. A single vertical
144 ascent or descent takes less than 30 min. After reaching the top, the container stopped and data were
145 measured continuously for 5-20 min of each cycle. For security reasons, the container reached a
146 maximum height limit of 260 m during the daytime and 240 m at night (Fig. 1). The container
147 instruments include: a global position system (GPS), an altimeter and an incoherent broadband cavity
148 enhanced absorption spectrometer (IBBCEAS) for measurements of HONO and NO₂. In addition,



149 another IBBCEAS was mounted in temperature-stabilized lab containers for the measurement of
150 HONO and NO₂ at ground level.

151 2.2 Instrumentation

152 HONO and NO₂ were simultaneously measured by a home-made IBBCEAS. Detailed description
153 of the IBBCEAS instrument can be found in Duan et al (2018); its application to the measurements
154 made during this study is described below. The HONO was sampled into an inlet tube (1.5 m length, 4
155 mm outside diameter (OD)) before entering the optical cavity (550 mm length, 25.4 mm OD), which
156 utilized PFA to minimize the HONO loss. The sampling gas flow rate was controlled at 6 SLPM
157 (Standard Liters per Minute) by a gas pump (KNF). The light emitted by the ultraviolet light-emitting
158 diode (UV-LED) was collimated by achromatic lens and coupled into the optical cavity. In the optical
159 cavity, light was reflected between the two highly reflective mirrors (R = 99.980% @368 nm, CRD
160 Optics, California, USA) to obtain a long optical absorption length. Then the light through the cavity
161 was coupled into an optical fiber by another achromatic lens before being received by a charge coupled
162 device (CCD) spectrometer (QE65000, Ocean Optics, Florida, USA). To protect the highly reflective
163 mirrors, pure N₂ was used to continuously purge the mirrors to prevent contact between the mirrors and
164 sample airflow. The purge flow rate was controlled at 0.1 SLPM by mass flow controllers (MFCs,
165 CS200A, Sevenstar, Beijing, China). The typical time resolution of the IBBCEAS instrument is 30 s,
166 and the 1 σ detection limits for HONO and NO₂ are 90 ppt and 170 ppt, respectively. In this study, the
167 IBBCEAS instrument was mounted in the movable container of the BMT for vertical profile
168 measurements and made measurements with a time resolution of 15 s (vertical resolution of 2.4 m).
169 Another IBBCEAS instrument was mounted in temperature-stabilized lab containers at ground level
170 and collected data with a time resolution of 30 s. The relative measurement error of the IBBCEAS
171 instrument was approximately 9%. Correction of the light intensity was performed every hour, and
172 mirror reflectivity was calibrated every day.

173 Meteorological parameters, including wind speed (WS), wind direction (WD), temperature (*T*),
174 and relative humidity (RH) were obtained using a 15-level meteorological gradient observation system
175 installed at fixed intervals along the meteorological tower (at heights of 8, 15, 32, 47, 65, 80, 100, 120,
176 140, 160, 180, 200, 240, 280 and 320 m). The gaseous species, including nitrogen monoxide (NO),
177 ozone (O₃), carbon monoxide (CO), and sulfur dioxide (SO₂), were measured using commercial gas



178 analyzer from Thermo Scientific. The 7-wavelength aethalometer (AE33, Magee Scientific Corp,
179 Berkeley, USA) was deployed to measure the black carbon (BC) mass concentration. CO, O₃, SO₂, and
180 BC were measured simultaneously at ground level and 260 m on the tower. The non-refractory
181 submicron aerosol (NR-PM₁) species were also measured simultaneously at ground level and 260 m on
182 the tower with an Aerodyne high-resolution time-of-flight aerosol mass spectrometer (AMS) and an
183 aerosol chemical speciation monitor (ACSM), respectively. The detailed sampling setup and calibration
184 of the AMS and ACMS as well as data analysis have been described in Xu et al. (2019) and Sun et al.
185 (2013).

186 2.3 Inter-comparison

187 In the present study, the measurements of HONO and NO₂ were conducted simultaneously in the
188 container and at ground level. Therefore, the calibration and inter-comparison of the two IBBCEAS
189 instruments were crucial. Comparison experiments were carried out in a temperature-stabilized
190 laboratory. The sampling unit and sampling flow rate of the two instruments were identical to minimize
191 the measured deviations. Figure. 2 shows significantly positive correlations between the two IBBCEAS
192 instruments (HONO: $R^2 = 0.99$, NO₂: $R^2 = 0.95$), with a slope of 0.99 (NO₂), 1.03 (HONO) and an
193 intercept of 750 ppt (NO₂) and 4 ppt (HONO). The difference was approximately 3%, within the
194 measurement error range of the instruments.

195 To verify the accuracy of the IBBCEAS instrument, an inter-comparison between the IBBCEAS
196 of this study and the IBBCEAS of Cambridge University was conducted. The HONO measurements
197 from the two different instruments were highly correlated ($R^2 = 0.95$, Fig. 2c). In addition, the
198 IBBCEAS instrument was also compared with the long optical path absorption photometer (LOPAP)
199 and the stripping coil ion chromatography (SC-IC) in our previous studies (Tang et al., 2019; Duan et al.,
200 2018), which also showed good correlations for HONO measurements (LOPAP: $R^2 = 0.894$, SC-IC: R^2
201 = 0.98).

202 3 Results and discussion

203 3.1 General observations and vertical measurements

204 The time-series of meteorological parameters, NR-PM₁ mass concentration, HONO, NO_x and
205 other relevant species are shown in Fig. 3. Based on the NR-PM₁ mass concentration level, three
206 different meteorological conditions were characterized during the measurement period (Table 1). The



207 first episode (E1) from 7 December to 10:00 on 8 December was a haze event. NR-PM₁ mass
208 concentration increased rapidly from 30 to ~150 $\mu\text{g}\cdot\text{m}^{-3}$ at ground level and 260 m on the tower due to
209 a low wind speed ($0.78 \pm 0.42 \text{ m}\cdot\text{s}^{-1}$) and high RH ($51\% \pm 13\%$).

210 The second episode (8-11 December, C2) was a clean event with low NR-PM₁ mass loading
211 (mean: $24 \pm 19 \mu\text{g}\cdot\text{m}^{-3}$), high wind speed ($> 5 \text{ m}\cdot\text{s}^{-1}$) dominantly from northwest. The third episode (E3)
212 from 11 December to 12 December was another haze event. During this period, characterized by
213 stagnant weather, lower wind speeds (an average of $0.77 \pm 0.4 \text{ m}\cdot\text{s}^{-1}$), and high RH ($55\% \pm 5\%$). The
214 mass concentration of NR-PM₁ gradually increased and then remained at relatively constant levels at
215 ground level and 260 m on the tower, and ranging from 69 to 218 $\mu\text{g}\cdot\text{m}^{-3}$ with the mean value of $154 \pm$
216 $35 \mu\text{g}\cdot\text{m}^{-3}$.

217 Throughout the entire measurement period, HONO concentrations ranged from 0.05 to 7.59 ppb.
218 The mean HONO mixing ratios during E1, C2, and E3 were 4.26 ± 2.08 , 0.83 ± 0.65 , and 3.54 ± 0.91
219 ppb, respectively. The maximum concentration of HONO was 7.59 ppb, which was observed during E1
220 (at 08:10 on 8 December). From 11 to 12 December, with stagnant weather, the pollutants continuously
221 increased. HONO concentrations remained a high level, and the daytime mean HONO mixing ratio
222 even reached 3.1 ± 0.92 ppb. Figure 3 also presents the time series of measured simultaneously other
223 relevant species. The mean NO₂ mixing ratios during E1, C2 and E3 were 51.98 ± 8.41 , 23.30 ± 11.91 ,
224 and 51.88 ± 5.97 ppb, respectively. Because NO and O₃ were not measured at ground level after 14:00
225 on 10 December, the mean concentrations of NO and O₃ during E1 and C2 were 90.99 ± 67.98 , $14.66 \pm$
226 21.79 ppb, and 4.04 ± 1.81 , 14.37 ± 10.65 ppb, respectively. After sunset, the concentration of O₃ at the
227 surface was rapidly titrated by elevated NO and increased with the increase of height. The mixing ratio
228 of O₃ below 260 m was less than 9 ppb during the vertical measurements.

229 Nocturnal stable surface layers of air are generally formed at low wind speed ($< 6 \text{ m}\cdot\text{s}^{-1}$)
230 (VandenBoer et al., 2013). Hence vertical profile data are adopted when wind speed was less than 6
231 $\text{m}\cdot\text{s}^{-1}$ except on 7 December (Fig. 4). Vertical measurements with low wind events were successfully
232 conducted in three occasions (9-10, 10-11, and 11-12 December) and would be discussed below. The
233 near-continuous vertical measurement avoids the observation bias from prolonged fixed sampling. The
234 date and time of measurement for each vertical profile is detailed in Table S1 in the Supplement.



235 **3.2 Nocturnal HONO vertical profiles**

236 **3.2.1 Vertical measurements after sunset**

237 Vertical measurements were conducted from ground level to 240 m after sunset. Figure 5 show the
238 temporal and spatial variations of HONO and NO₂ during the clean episode (C2) and the haze episode
239 (E3). The mixing ratios of HONO and NO₂ at ground level were consistent with those measured in the
240 container, indicating that HONO and NO₂ were relatively well mixed during these two episodes.

241 The vertical profiles of HONO, NO₂, and Δ HONO just after sunset are shown in Fig. 6. During
242 C2 and E3, the mixing ratios of HONO and NO₂ (Fig. 6a) showed nearly flat profiles throughout the
243 column. The vertical variations of Δ HONO (Fig. 6b), which is the difference of HONO
244 concentration between measured in the container and at ground level, centered around 0 ppb and
245 varied between -0.4 and 0.4 ppb, confirming the relatively uniform vertical distribution of both
246 HONO and NO₂. The vertical variations of *T* and RH during these three measurements were similar
247 (Fig. S2). While *T* decreased gradually as the increase with increasing height, RH increased gradually
248 with increasing height, and RH was relatively higher during the haze episode. Also, there were no *T*
249 inversion just after sunset, and the consistent variations of HONO and NO₂ at ground level and
250 vertical measurements supports a relatively well-mixed boundary layer, which explains the uniform
251 vertical distribution of HONO and NO₂.

252 **3.2.2 Nocturnal vertical profiles**

253 Nocturnal small-scale stratification and layering was determined according to the method of
254 Brown et al. (2012), who used the potential temperature profile as an indicator of atmospheric static
255 stability. According to the vertical variations in potential temperature, the stable layer was divided into
256 the “surface layer”, the “nocturnal boundary layer (NBL)”, the “top of the nocturnal boundary layer”
257 and the “residual layer (RL)”.

258 Figure 7 depicts the nocturnal vertical profiles of HONO, NO₂ and potential temperature during
259 the clean episode. On the night of 9 December (C2), negative profiles of both HONO and NO₂ were
260 clearly seen. When the container ascended during 22:42-23:06, the potential temperature profile
261 showed distinct stratification. The surface layer extended to 10-20 m and the NBL extended to ~140 m.
262 There was a significant negative gradient HONO within NBL. Above the NBL, a negative gradient of
263 HONO was also observed in the RL but was not consistently observed in other measurements (see
264 below). When the descent of the container during 23:15-23:40, the potential temperature profile



265 showed that a shallow T inversion was rapidly formed between 130 and 200 m. Within the shallow
266 inversion layer, the vertical convection and transport was inhibited, and the remarkable negative
267 gradient was observed there. Within the NBL, the negative gradient of HONO and NO_2 , however,
268 disappeared. This might due to the continuous vertical mixing from 23:06 to 23:40. By the way, the
269 surface source of HONO is obvious as seen from the apparent accumulation of HONO within NBL. In
270 addition, the obviously vertical variation in RH during 23:15-23:40 (Fig. S3) indicated the different
271 layers at different height, which was due to the influence of the shallow inversion layer.

272 The vertical profile of potential temperature on 10 December (C2) shows that shallow inversion
273 layer formed between the surface layer and the NBL. In the shallow inversion layer, the mixing ratios
274 of HONO decreased rapidly with increasing height, and a significant negative gradient was observed
275 within the shallow inversion layer and surface layer. With the attenuation of the shallow inversion layer
276 during the descent of the container from 23:01 to 23:25, the inhibition of vertical transport and mixing
277 gradually weakened, and the negative gradient of HONO disappeared below ~ 100 m, which indicated
278 the surface source of HONO or the interaction of different air masses due to the change of WD (Fig.
279 S3). And the attenuation event of shallow inversion layer may also be the result of the increase of wind
280 speed and interaction of different air masses that changed from the west to southeast between 15 and
281 100 m. Above 100 m height, the mixing ratio of HONO decreased with increasing height, and the
282 fluctuation of HONO likely due to the interaction of different air masses. In contrast, the vertical
283 profile of NO_2 shows that NO_2 rapidly decreased towards the ground; a significant positive gradient
284 was observed near the surface, which was caused by several factors. The nocturnal NO_2 is produced by
285 the reaction of O_3 with NO , which mainly occurs near the surface, resulting in a negative gradient in
286 NO_2 . However, this effect was counteracted by the dry deposition of NO_2 , which by itself would result
287 in a positive gradient (Stutz et al., 2004b). Meanwhile, the mixing ratio of NO_2 was also affected by
288 local traffic emission sources. All of these lead to the positive gradient of NO_2 near the surface.
289 Compared to the vertical profiles collected on 9 December, a near-surface shallow inversion layer was
290 formed on 10 December, resulting in the clearly positive gradient of near-surface NO_2 . Moreover, the
291 increase of the positive gradient of NO_2 indicates the formation of near-surface shallow inversion layer
292 likely caused the increase of dry deposition, which affects the vertical distribution of HONO and NO_2
293 at night.

294 Although the surface layer was a common feature in the potential temperature profiles, it was



295 absent on the night of 11-12 December (E3), and the NBL extended downward to the lowest
296 measurement height (8 m above the ground; Fig. 8). With the development of the boundary layer, the
297 negative gradient of HONO and NO₂ gradually disappeared during 22:35-23:29, which was similar to
298 the vertical profile on 9 December. As shown in Fig. 8, the vertical profile of HONO showed a
299 significant negative gradient as the container ascent during 22:35-23:00, and higher HONO mixing
300 ratios were observed at ground level. However, the mixing ratios of HONO and NO₂ approached a
301 steady-state plateau within NBL around midnight. A possible physical and chemical process, the loss of
302 HONO to the ground surface by dry deposition and vertical convection could account for approaching
303 a near-steady states in the HONO mixing ratio and HONO/NO₂ (Fig. S4). Similar vertical
304 measurements were reported by VandenBoer et al (2013) who also observed a near-steady state in the
305 HONO mixing ratio and HONO/NO₂. Dry deposition loss of HONO to the ground surface and vertical
306 mixing effects associated with vertical convection reached a near-steady state around midnight, which
307 indicated that significant quantities of HONO are deposited to the ground surface at night, and
308 deposition of HONO at ground surface is an important sink of HONO at night.

309 3.3 Direct emission

310 In the present study, the measurement site is surrounded by several main roads, and thus might be
311 affected by substantial vehicle emissions at night. To evaluate the influence of direct emissions,
312 emission ratios of HONO/NO_x was derived from our measurement. Since NO was not measured at
313 ground level after 10 December, the nocturnal measurement data of HONO and NO_x (18:00-6:00) from
314 9 November to 10 December were used to evaluate the local HONO emission factor. Considering the
315 differences in the type of vehicles, fuel compositions, etc., the reported emission factor of HONO/NO_x
316 might not be representative for Beijing region. In order to evaluate the influence of direct emission, the
317 local emission factor of HONO was estimated from the ambient measurements.

318 As an air mass becomes aged, the conversion of NO₂ to HONO will result in the increase of
319 HONO/NO_x; freshly emitted air masses are characterized by the lowest HONO/NO_x. Thus, the
320 observed minimum HONO/NO_x could be used as an upper limit for the emission factor (Li et al.,
321 2012; Su et al., 2008a). In order to capture as much of the freshly emitted air mass as possible, two
322 criteria were chosen to ensure that the air mass was dominated by fresh vehicle emission: (a) NO₂ ≥ 80
323 ppb and (b) NO/NO_x ≥ 80% (Xu et al., 2015). The derived HONO/NO_x of 1.41% is comparable to the



324 value of 1.24% reported by Liang et al. (2017), who measured HONO emission factor in a road tunnel
325 in Hong Kong. $HONO_{emis}$ was calculated by the following equation: $[HONO]_{emis} = 0.0141 \times [NO_x]$.
326 The frequency distribution of $HONO_{emission}/HONO$ during the measurement period is shown in Fig. S5.
327 The contribution of direct emissions to ambient HONO was estimated to be 51.1%, suggesting that
328 vehicle emissions is an important nocturnal HONO source in Beijing (Zhang et al., 2018).

329 3.4 Nocturnal HONO chemistry

330 3.4.1 Correlation studies

331 The heterogeneous conversion of NO_2 is an important pathway of HONO formation as many field
332 measurements have also found a strong correlation between HONO and NO_2 (Zhou et al., 2006; Su
333 et al., 2008a; Wang et al., 2013; Huang et al., 2017). However, the use of correlation analysis to interpret
334 the heterogeneous conversion of NO_2 should be carefully treated, as physical transport and source
335 emissions also contribute to the correlation. In this study, the correlation of vertical measurements
336 between HONO and NO_2 were analyzed. Vertical profile data without horizontal transport were used to
337 avoid the influence of physical transport. As shown in Fig. 9, HONO and NO_2 exhibited moderate but
338 significant correlations (C2: $R^2 = 0.72$, E3: $R^2 = 0.69$), indicating that NO_2 participated in the formation
339 of HONO.

340 The adsorbed water on the surface affects the heterogeneous formation of HONO (Stutz et al.,
341 2004a). The relationship between $HONO/NO_2$ and RH is illustrated in Fig. 10. Following the method
342 introduced by Stutz et al (2004a), we analyzed the average of the five highest $HONO/NO_2$ value in
343 each 10% RH interval to eliminate much of the influence of factors like the time of night, advection,
344 the surface density, etc. An increase in the $HONO/NO_2$ ratio along with RH was observed at each
345 height interval when RH was less than 70%. Previous observation at ground level also reported that the
346 $HONO/NO_2$ ratio increased with the increase of RH when RH was less than 70%, while the further
347 increase of RH would lead to decrease in the $HONO/NO_2$, which may be caused by an increase in the
348 number of water layers formed on the surface with the increase of RH, resulting in the more efficient
349 uptake of HONO (Li et al., 2012; Wang et al., 2013b; Yu et al., 2009). The $HONO/NO_2$ ratio decreased
350 with the increase of height at similar RH level for both C2 and E3, which implied that the ground
351 surface was the main heterogeneous reaction surface at night. It is important to note that the limited
352 vertical measurements result in the limited variation range of RH, which limits the analysis. Additional



353 efforts are needed to conduct more comprehensive vertical measurements to interpret the HONO/NO₂
354 ratios versus RH for different height in the future.

355 3.4.2 Influence of aerosol surface on nocturnal HONO production

356 The relative importance of aerosol surfaces and ground surfaces in nocturnal HONO production
357 has been widely investigated in the field (Kleffmann et al., 2003; Oswald et al., 2015; VandenBoer et al.,
358 2013; Stutz, 2002; Wong et al., 2011; Ye et al., 2018; Li et al., 2012; Su et al., 2008a). An estimate of the
359 contribution of nocturnal HONO production from aerosol surfaces, we can expect that the air mass in
360 the RL (above 200 m) is less affected by the ground-level processes (Brown et al., 2012; VandenBoer et
361 al., 2013). It can be approximately considered that the HONO production was primarily derived from
362 the heterogeneous conversion of NO₂ on aerosol surfaces, HONO production ($P(HONO)$) can then be
363 expressed using the following equation:

$$364 \frac{P(HONO)}{[NO_2]} = \frac{1}{4} \times \left[\frac{s}{v} \right] \times \sqrt{\frac{8RT}{\pi M}} \times \gamma \quad (1)$$

365 where γ is the uptake coefficient of NO₂, R is the gas constant, T is the absolute temperature (K), M is
366 the molecular mass of NO₂ ($M=4.6 \times 10^{-2}$ kg mol⁻¹), and $\left[\frac{s}{v} \right]$ is the aerosol surface area density (m⁻¹).

367 The NO₂-normalized HONO production over time, $\Delta \frac{[HONO]}{[NO_2]} / \Delta t$, can be calculated using Eq. (2):

$$368 \Delta \frac{[HONO]}{[NO_2]} / \Delta t \sim \frac{1}{4} \times \left[\frac{s}{v} \right] \times \sqrt{\frac{8RT}{\pi M}} \times \gamma \quad (2)$$

369 Assuming that the uptake coefficient of NO₂ is 1×10^{-5} , which fits NO₂ uptake coefficient value of
370 1×10^{-6} - 1×10^{-5} from those observed in relevant studies (J.Kleffmann et al., 1998; Kurtenbach et al.,
371 2001; George et al., 2005; Stemmler et al., 2007), with a $\left[\frac{s}{v} \right]$ value of $\sim 10^{-3}$ m⁻¹ (Wang et al., 2018). A
372 relative HONO formation rate of $\Delta \frac{[HONO]}{[NO_2]} / \Delta t$ is ~ 0.0032 h⁻¹, which is equivalent to the HONO
373 production of 0.17 ppb h⁻¹ when the upper limit of observed nocturnal NO₂ above 200 m was 52.9 ppb
374 during E3. The nocturnal accumulation of HONO (18:00-00:00) in the RL (above 200 m) is 1.02 ppb
375 during E3, which is much less than the HONO vertical measurements of 3.06 ppb. It is necessary to
376 elaborate that: (1) the typical $\left[\frac{s}{v} \right]$ is $\sim 10^{-3}$ m⁻¹ during the pollution period in winter in Beijing (NO₂=
377 45 ppb, temperature = 273 K); (2) the upper limit of NO₂ uptake coefficient and the upper limit of
378 observed nocturnal NO₂ in the RL are used to calculate the HONO production on aerosol. The HONO
379 production could be actually overestimated in our study. As described, HONO production from the



380 heterogeneous conversion of NO₂ solely on the aerosol surfaces cannot explain HONO vertical
381 measurements at night. In addition, the integrated column of HONO (HONO_{column}) and the mixing
382 ratios of HONO observed from ground level to 10 m height (HONO_{ground}) show a significant
383 correlation ($R^2 = 0.85$) (Fig. 11), which also indicates that the ground surface is the primary reaction
384 surface on which HONO is formed, and then transported throughout the column.

385 3.4.3 Nocturnal HONO production at ground level

386 The nocturnal HONO in the boundary layer is primarily derived from the heterogeneous
387 conversion from NO₂ to HONO on the ground surface, which is consistent with the results of other
388 studies in vertical gradient of HONO (Kleffmann et al., 2003; Wong et al., 2011; Zhang et al., 2009; Ye et
389 al., 2018). The conversion rate of NO₂ can be evaluated using the HONO conversion frequency.
390 However, real atmospheric components are affected by physical processes, chemical processes, and
391 direct emissions. To reduce the uncertainties associated with transport process and source emissions,
392 the HONO conversion frequency was calculated using the scaling method proposed by Su et al (2008)
393 and adopting NO₂ and BC as reference species. Before calculation, the HONO concentration was
394 corrected by direct emission (Eq. (3)). Because NO was not measured at ground level after 14:00 on 10
395 December, the data of NO_x was not available during the vertical measurements at night on 10 and 11
396 December. The monthly average HONO_{emis}/HONO ratio of 51.1% was calculated to correct the
397 observed HONO (HONO_{corr}). The HONO conversion frequency C_{HONO} can be expressed as:

$$398 \quad [HONO]_{corr} = [HONO] - [HONO]_{emis} = [HONO] - [HONO] \times 0.511 \quad (3)$$

$$399 \quad C_{HONO}^X = \frac{2 \left(\frac{[HONO_{corr}]_{t_2} \times [X] - [HONO_{corr}]_{t_1} \times [X]}{[X]_{t_2}} - \frac{[HONO_{corr}]_{t_1} \times [X]}{[X]_{t_1}} \right)}{(t_2 - t_1) \left(\frac{[NO_2]_{t_2} + [NO_2]_{t_1}}{[X]_{t_2}} + \frac{[NO_2]_{t_1}}{[X]_{t_1}} \right) [X]}$$
$$400 \quad = \frac{2 \left(\frac{[HONO_{corr}]_{t_2}}{[X]_{t_2}} - \frac{[HONO_{corr}]_{t_1}}{[X]_{t_1}} \right)}{(t_2 - t_1) \left(\frac{[NO_2]_{t_2} + [NO_2]_{t_1}}{[X]_{t_2}} + \frac{[NO_2]_{t_1}}{[X]_{t_1}} \right)} \quad (4)$$

$$401 \quad C_{HONO} = \frac{1}{3} (C_{HONO}^0 + C_{HONO}^{NO_2} + C_{HONO}^{BC}) \quad (5)$$

402 where $[HONO_{corr}]_t$, $[NO_2]_t$, and $[X]_t$ represent mixing ratios of HONO, NO₂, and the reference gas
403 at the sampling time t , C_{HONO} is average conversion frequency, C_{HONO}^X is the conversion frequency
404 scaled with species X , and C_{HONO}^0 is the conversion frequency which is not scaled. The average
405 conversion frequencies C_{HONO} on 9, 10 and 11 December were 0.0039 h⁻¹, 0.0026 h⁻¹, and 0.0039 h⁻¹,
406 respectively, which was comparable to the observations of Hou et al. (2016) in Beijing (the clean



407 episode: 0.0065 h⁻¹; the haze episode: 0.0039 h⁻¹) and Lammel et al. (1996) in Mainz and Milan (0.0041
408 h⁻¹ and 0.00491 h⁻¹), but much lower than observations made by Li et al. (2012) (0.024 h⁻¹) and Su et al.
409 (2008) (0.016 h⁻¹) at a rural site in southern China.

410 As illustrated above, HONO production from the heterogeneous conversion of NO₂ on the aerosol
411 surfaces cannot explain our HONO vertical measurements. The heterogeneous conversion from NO₂ to
412 HONO on ground surfaces at nighttime was further evaluated,

$$413 \quad C_{\text{HONO}} = \frac{P_{\text{HONO}}}{[\text{NO}_2]} = \frac{\phi_{\text{NO}_2 \rightarrow \text{HONO}} \times v_{\text{NO}_2}^{\text{ground}}}{H} - \frac{v_{\text{HONO}}^{\text{ground}}}{H} \times \frac{[\text{HONO}_{\text{corr}}]}{[\text{NO}_2]} \quad (6)$$

414 where $v_{\text{HONO}}^{\text{ground}}$ and $v_{\text{NO}_2}^{\text{ground}}$ are the dry deposition velocities of HONO and NO₂, which were
415 expected to be similar since the nocturnal controlling resistances were mainly the aerodynamic
416 resistance and the quasi-laminar layer resistance (Su et al., 2008a; Stutz et al., 2002). $\phi_{\text{NO}_2 \rightarrow \text{HONO}}$ is
417 the HONO yield from deposited NO₂, which varies between 0 and 1, indicating that the deposited NO₂
418 molecule is not necessarily converted to HONO. H is the mixing depth.

419 Previous studies reported the value of $v_{\text{HONO}}^{\text{ground}}$ between 0.077 and 3 cm s⁻¹ (Harrison and Kitto,
420 1994; Harrison et al., 1996; Spindler et al., 1998; Stutz, 2002; Coe and Gallagher, 1992). In the present
421 study, we used a mean $v_{\text{HONO}}^{\text{ground}}$ of 0.095 cm s⁻¹ (Su et al., 2008a; Stutz, 2002; Coe and Gallagher, 1992)
422 and assumed that boundary layer mixing depth H is 100 m high. The $\phi_{\text{NO}_2 \rightarrow \text{HONO}}$ values on 9 (C2), 10
423 (C2) and 11 December (E3) were 0.10, 0.08, and 0.09, respectively, which means that every 10-13
424 deposited NO₂ molecules will result in one HONO being released into atmosphere. The value of
425 deposited NO₂ molecules converted into gas phase HONO is within the range of published results
426 (3-33) (Stutz et al., 2002; Su et al., 2008a; Li et al., 2012). The derived $\phi_{\text{NO}_2 \rightarrow \text{HONO}}$ value is lower
427 than calculation by Su et al (2008a) (0.34), which is caused by the differences in surface environment
428 between urban and rural areas. Similar $\phi_{\text{NO}_2 \rightarrow \text{HONO}}$ values were found during C2 and E3, suggesting
429 that the potential of heterogeneous conversion from NO₂ to HONO on ground surface at night is
430 consistent.

431 **4 Conclusions**

432 High-resolution vertical profiles of HONO and NO₂ were measured using an IBBCEAS
433 instrument during the APHH-Beijing winter campaign. Although the data set of this study is rather
434 limited in scope, encompassing only four nights in December 2016 with a limited set of ancillary data,
435 it is unique because, to our knowledge, this is the first high-resolution vertical measurements of HONO



436 and NO_2 in urban areas of China. The observed HONO concentrations during E1, C2, and E3 were 4.26
437 ± 2.08 , 0.83 ± 0.65 , and 3.54 ± 0.91 ppb, respectively. The vertical distribution of HONO is consistent
438 with reduced mixing and stratification in the lower several hundred meters of the nocturnal urban
439 atmosphere. The $\text{HONO}_{\text{emis}}/\text{HONO}$ ratio was 51.1%, indicating that direct emissions from combustion
440 processes have a great deal of influence on ambient HONO concentrations and is an important
441 nocturnal HONO sources at the measurement site. In addition, high-resolution vertical profiles of
442 HONO, HONO production from the heterogeneous conversion of NO_2 on aerosol surfaces cannot
443 explain HONO vertical measurements and the correlation between the integrated column of HONO and
444 HONO measured from the surface to 10 m height, all suggesting that the heterogeneous conversion of
445 NO_2 on the ground surface dominated the HONO production at night, and then transported throughout
446 the column due to the vertical convection.

447 A relatively well-mixed boundary layer was observed after sunset, and the ΔHONO fluctuated
448 around zero. The small-scale stratification of the nocturnal urban atmosphere and the formation of a
449 shallow inversion layer affect the vertical distribution of HONO and NO_2 . The high-resolution vertical
450 profiles showed that dry deposition loss of HONO to the ground surface and vertical mixing effects
451 reached a near-steady state on the nighttime of 11-12 December, which revealed (1) the ground surface
452 is the dominant reaction surface on which HONO is formed from the heterogeneous conversion of NO_2 ;
453 (2) significant quantities of HONO are deposited to the ground surface at night, which is an important
454 nocturnal sink of HONO. The nocturnal HONO production from the heterogeneous reaction of NO_2 at
455 ground level was evaluated, and the $\phi_{\text{NO}_2 \rightarrow \text{HONO}}$ values on 9 (C2), 10 (C2) and 11 December (E3)
456 were 0.10, 0.08 and 0.09, respectively. The similar values of $\phi_{\text{NO}_2 \rightarrow \text{HONO}}$ indicate that the potential of
457 heterogeneous conversion from NO_2 to HONO at the ground level is consistent for both the clean and
458 the haze episodes. Because of the relatively few vertical measurement cases and the limited height of
459 the meteorological tower, future vertical measurements of HONO to a high height (e.g., using tethered
460 balloons) and more comprehensive vertical measurements in the megacities, are urgently needed for a
461 better understanding of the vertical distribution and the formation mechanisms of HONO in the PBL in
462 megacities.

463

464 *Data availability.* The data used in this study are available from the corresponding author upon request
465 (mqin@aiofm.ac.cn).



466

467 *Supplement.*

468

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471 instrument. JD and KT conducted the measurements. FM and MQ analyzed the data. FM wrote the paper. MQ, YS
472 and CY revised and commented on the paper.

473

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475

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479

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485 **References**

- 486 Acker, K., Febo, A., Trick, S., Perrino, C., Bruno, P., Wiesen, P., Möller, D., Wieprecht, W., Auel, R., Giusto, M.,
487 Geyer, A., Platt, U., and Allegrini, I.: Nitrous acid in the urban area of Rome, *Atmos. Environ.*, 40, 3123-3133,
488 <https://doi.org/10.1016/j.atmosenv.2006.01.028>, 2006.
- 489 Aliche, B., Platt, U., Stutz, J.: Impact of nitrous acid photolysis on the total hydroxyl radical budget during the
490 Limitation of Oxidant Production/Pianura Padana Produzione di Ozono study in Milan, *J. Geophys. Res.*, 107,
491 LOP 9-1-LOP 9-17, <https://doi.org/10.1029/2000jd000075>, 2002.
- 492 An, J., Li, Y., Chen, Y., Li, J., Qu, Y., and Tang, Y. J.: Enhancements of major aerosol components due to additional
493 HONO sources in the North China Plain and implications for visibility and haze, *Adv. Atmos. Sci.*, 30, 57-66,
494 <https://doi.org/10.1007/s00376-012-2016-9>, 2012.
- 495 Aubin, D. G., and Abbatt, J. P. D.: Interaction of NO₂ with Hydrocarbon Soot: Focus on HONO Yield, Surface
496 Modification, and Mechanism, *J. Phys. Chem. A.*, 111, 6263-6273, <https://doi.org/10.1021/jp068884h>, 2007.
- 497 Bao, F. X., Li, M., Zhang, Y., Chen, C. C., and Zhao, J. C.: Photochemical Aging of Beijing Urban PM_{2.5}: HONO
498 Production, *Environ. Sci. Technol.*, 52, 6309-6316, <https://doi.org/10.1021/acs.est.8b00538>, 2018.
- 499 Bartolomei, V., Alvarez, E. G., Wittmer, J., Thili, S., Strekowski, R., Temime-Roussel, B., Quivet, E., Wortham, H.,



- 500 Zetzsch, C., Kleffmann, J., and Gligorovski, S.: Combustion Processes as a Source of High Levels of Indoor
501 Hydroxyl Radicals through the Photolysis of Nitrous Acid, *Environ. Sci. Technol.*, 49, 6599-6607,
502 <https://doi.org/10.1021/acs.est.5b01905>, 2015.
- 503 Bejan, I., Abd-El-Aal, Y., Barnes, I., Benter, T., Bohn, B., Wiesen, P., and Kleffmann, J.: The photolysis of
504 *ortho*-nitrophenols: a new gas phase source of HONO, *Phys. Chem. Chem. Phys.*, 8, 2028-2035,
505 <https://doi.org/10.1039/b516590c>, 2006.
- 506 Brown, S. S., Dubé, W. P., Osthoff, H. D., Wolfe, D. E., Angevine, W. M., and Ravishankara, A. R.: High
507 resolution vertical distributions of NO₃ and N₂O₅ through the nocturnal boundary layer, *Atmos. Chem. Phys.*, 7,
508 139-149, <https://doi.org/10.5194/acp-7-139-2007>, 2007.
- 509 Cai, R. L., Yang, D. S., Fu, Y. Y., Wang, X., Li, X. X., Ma, Y., Hao, J. M., Zheng, J., and Jiang, J. K.: Aerosol
510 surface area concentration: a governing factor in new particle formation in Beijing, *Atmos. Chem. Phys.*, 17,
511 12327-12340, <https://doi.org/10.5194/acp-17-12327-2017>, 2017.
- 512 Coe, H., and Gallagher, M. W.: Measurements of Dry Deposition of NO₂ to A Dutch Heathland Using the
513 Eddy-Correlation Technique, *Q. J. Roy. Meteor. Soc.*, 118, 767-786, <https://doi.org/10.1002/qj.49711850608>, 1992.
- 514 Cui, L. L., Li, R., Zhang, Y. C., Meng, Y., Fu, H. B. and Chen, J. M.: An observational study of nitrous acid
515 (HONO) in Shanghai, China: The aerosol impact on HONO formation during the haze episodes, *Sci. Total
516 Environ.*, 630, 1057-1070, <https://doi.org/10.1016/j.scitotenv.2018.02.063>, 2018.
- 517 Duan, J., Qin, M., Ouyang, B., Fang, W., Li, X., Lu, K. D., Tang, K., Liang, S. X., Meng, F. H., Hu, Z. K., Xie, P.
518 H., Liu, W. Q., and Häslner, R.: Development of an incoherent broadband cavity-enhanced absorption spectrometer
519 for in situ measurements of HONO and NO₂, *Atmos. Meas. Tech.*, 11, 4531-4543,
520 <https://doi.org/10.5194/amt-11-4531-2018>, 2018.
- 521 Finlayson-Pitts, B. J., Wingen, L. M., Sumner, A. L., Syomin, D., and Ramazan, K. A.: The heterogeneous
522 hydrolysis of NO₂ in laboratory systems and in outdoor and indoor atmospheres: An integrated mechanism, *Phys.
523 Chem. Chem. Phys.*, 5, 223-242, <https://doi.org/10.1039/b208564j>, 2003.
- 524 Fu, X., Wang, T., Zhang, L., Li, Q. Y., Wang, Z., Xia, M., Yun, H., Wang, W. H., Yu, C., Yue, D. L., Zhou, Y.,
525 Zheng, J. Y., and Han, R.: The significant contribution of HONO to secondary pollutants during a severe winter
526 pollution event in southern China, *Atmos. Chem. Phys.*, 19, 1-14, <https://doi.org/10.5194/acp-19-1-2019>, 2019.
- 527 George, C., Strekowski, R. S., Kleffmann, J., Stemmler, K., and Ammann, M.: Photoenhanced uptake of gaseous
528 NO₂ on solid organic compounds: a photochemical source of HONO?, *Faraday Discuss.*, 130, 195-210,
529 <https://doi.org/10.1039/b417888m>, 2005.
- 530 Gómez Alvarez, E., Sörgel, M., Gligorovski, S., Bassil, S., Bartolomei, V., Coulomb, B., Zetzsch, C., and Wortham,
531 H.: Light-induced nitrous acid (HONO) production from NO₂ heterogeneous reactions on household chemicals,
532 *Atmos. Environ.*, 95, 391-399, <https://doi.org/10.1016/j.atmosenv.2014.06.034>, 2014.
- 533 Han, C., Yang, W. J., Wu, Q. Q., Yang, H., and Xue, X. X.: Heterogeneous Photochemical Conversion of NO₂ to
534 HONO on the Humic Acid Surface under Simulated Sunlight, *Environ. Sci. Technol.*, 50, 5017-5023,
535 <https://doi.org/10.1021/acs.est.5b05101>, 2016.
- 536 Han, C., Yang, W. J., Yang, H., and Xue, X. X.: Enhanced photochemical conversion of NO₂ to HONO on humic
537 acids in the presence of benzophenone, *Environ. Pollut.*, 231, 979-986,
538 <https://doi.org/10.1016/j.envpol.2017.08.107>, 2017.
- 539 Hanst, P. L., Spence, J. W., and Miller, M.: Atmospheric Chemistry of N-nitroso Dimethylamine, *Environ. Sci.
540 Technol.*, 11, 403-405, <https://doi.org/10.1021/es60127a007>, 1977.
- 541 Harrison, R. M., and Kitto, A. M. N.: Evidence for a surface source of atmospheric nitrous acid, *Atmos. Environ.*,
542 28, 1089-1094, [https://doi.org/10.1016/1352-2310\(94\)90286-0](https://doi.org/10.1016/1352-2310(94)90286-0), 1994.
- 543 Harrison, R. M., Peak, J. D., and Collins, G. M.: Tropospheric cycle of nitrous acid, *J. Geophys. Res.*, 101,



- 544 14429-14439, <https://doi.org/10.1029/96JD00341>, 1996.
- 545 Hendrick, F., Müller, J. F., Clémer, K., Wang, P., De Mazière, M., Fayt, C., Gielen, C., Hermans, C., Ma, J. Z.,
546 Pinardi, G., Stavrou, T., Vlemmix, T., and Van Roozendaal, M.: Four years of ground-based MAX-DOAS
547 observations of HONO and NO₂ in the Beijing area, *Atmos. Chem. Phys.*, 14, 765-781,
548 <https://doi.org/10.5194/acp-14-765-2014>, 2014.
- 549 Hou, S. Q., Tong, S. R., Ge, M. F., and An, J. L.: Comparison of atmospheric nitrous acid during severe haze and
550 clean periods in Beijing, China, *Atmos. Environ.*, 124, 199-206, <https://doi.org/10.1016/j.atmosenv.2015.06.023>,
551 2016.
- 552 Huang, R. J., Zhang, Y. L., Bozzetti, C., Ho, K. F., Cao, J. J., Han, Y. M., Daellenbach, K. R., Slowik, J. G., Platt, S.
553 M., Canonaco, F., Zotter, P., Wolf, R., Pieber, S. M., Bruns, E. A., Crippa, M., Ciarelli, G., Piazzalunga, A.,
554 Schwikowski, M., Abbaszade, G., Schnelle-Kreis, J., Zimmermann, R., An, Z. S., Szidat, S., Baltensperger, U., El
555 Haddad, I., and Prévôt, A. S. H.: High secondary aerosol contribution to particulate pollution during haze events in
556 China, *Nature*, 514, 218-222, <https://doi.org/10.1038/nature13774>, 2014.
- 557 Huang, R. J., Yang, L., Cao, J. J., Wang, Q. Y., Tie, X. X., Ho, K. F., Shen, Z. X., Zhang, R. J., Li, G. H., Zhu, C. S.,
558 Zhang, N. N., Dai, W. T., Zhou, J. M., Liu, S. X., Chen, Y., Chen, J., and O'Dowd, C. D.: Concentration and
559 sources of atmospheric nitrous acid (HONO) at an urban site in Western China, *Sci. Total Environ.*, 593-594,
560 165-172, <https://doi.org/10.1016/j.scitotenv.2017.02.166>, 2017.
- 561 Hao, N., Zhou, B., Chen, D., and Chen, L. M.: Observations of nitrous acid and its relative humidity dependence
562 in Shanghai, *J. Environ. Sci.*, 18, 910-915, [https://doi.org/10.1016/S1001-0742\(06\)60013-2](https://doi.org/10.1016/S1001-0742(06)60013-2), 2006.
- 563 Kleffmann, J., Beckera, K. H., and Wiesena, P.: Heterogeneous NO₂ conversion processes on acid surfaces:
564 possible atmospheric implications, *Atmos. Environ.*, 32, 2721-2729,
565 [https://doi.org/10.1016/S1352-2310\(98\)00065-X](https://doi.org/10.1016/S1352-2310(98)00065-X), 1998.
- 566 Karamchandani, P., Emery, C., Yarwood, G., Lefer, B., Stutz, J., Couzo, E., and Vizuete, W.: Implementation and
567 refinement of a surface model for heterogeneous HONO formation in a 3-D chemical transport model, *Atmos.*
568 *Environ.*, 112, 356-368, <https://doi.org/10.1016/j.atmosenv.2015.01.046>, 2015.
- 569 Kleffmann, J., Becker, K. H., Lackhoff, M., and Wiesen, P.: Heterogeneous conversion of NO₂ on carbonaceous
570 surfaces, *Phys. Chem. Chem. Phys.*, 1, 5443-5450, <https://doi.org/10.1039/A905545B>, 1999.
- 571 Kleffmann, J., Kurtenbach, R., Lörzer, J., Wiesen, P., Kalthoff, N., Vogel, B., and Vogel, H.: Measured and
572 simulated vertical profiles of nitrous acid—Part I: Field measurements, *Atmos. Environ.*, 37, 2949-2955,
573 [https://doi.org/10.1016/S1352-2310\(03\)00242-5](https://doi.org/10.1016/S1352-2310(03)00242-5), 2003.
- 574 Kleffmann, J.: Daytime Sources of Nitrous acid (HONO) in the Atmospheric Boundary Layer, *Chemphyschem*, 8,
575 1137-1144, <https://doi.org/10.1002/cphc.200700016>, 2007.
- 576 Kurtenbach, R., Becker, K. H., Gomes, J. A. G., Kleffmann, J., Lörzer, J., Spittler, M., Wiesen, P., Ackermann, R.,
577 Geyer, A., and Platt, U.: Investigations of emission and heterogeneous formation of HONO in a road traffic tunnel,
578 *Atmos. Environ.*, 35, 3385-3394, [https://doi.org/10.1016/S1352-2310\(01\)00138-8](https://doi.org/10.1016/S1352-2310(01)00138-8), 2001.
- 579 Laufs, S., Cazaunau, M., Stella, P., Kurtenbach, R., Cellier, P., Mellouki, A., Loubet, B., and Kleffmann, J.: Diurnal
580 fluxes of HONO above a crop rotation, *Atmos. Chem. Phys.*, 17, 6907-6923,
581 <https://doi.org/10.5194/acp-17-6907-2017>, 2017.
- 582 Li, S. P., Matthews, J., and Sinha, A.: Atmospheric Hydroxyl Radical Production from Electronically Excited NO₂
583 and H₂O, *Science*, 319, 1657-1660, <https://doi.org/10.1126/science.1151443>, 2008.
- 584 Li, X., Brauers, T., Häsel, R., Bohn, B., Fuchs, H., Hofzumahaus, A., Holland, F., Lou, S., Lu, K. D., Rohrer, F.,
585 Hu, M., Zeng, L. M., Zhang, Y. H., Garland, R. M., Su, H., Nowak, A., Wiedensohler, A., Takegawa, N., Shao, M.,
586 and Wahner, A.: Exploring the atmospheric chemistry of nitrous acid (HONO) at a rural site in Southern China,
587 *Atmos. Chem. Phys.*, 12, 1497-1513, <https://doi.org/10.5194/acp-12-1497-2012>, 2012.



- 588 Li, X., Rohrer, F., Hofzumahaus, A., Brauers, T., Häsel, R., Bohn, B., Broch, S., Fuchs, H., Gomm, H., Holland,
589 F., Jäger, J., Kaiser, J., Keutsch, F. N., Lohse, I., Lu, K. D., Tillmann, R., Wegener, R., Wolfe, G. M., Mentel, T. F.,
590 Kiendler-Scharr, A., Wahner, A.: Missing Gas-Phase Source of HONO Inferred from Zeppelin Measurement in the
591 Troposphere, *Science*, 334, 292-296, <https://doi.org/10.1126/science.1248999>, 2014.
- 592 Liang, Y. T., Zha, Q. Z., Wang, W. H., Cui, L., Lui, K. H., Ho, K. F., Wang, Z., Lee, S. C., and Wang, T.: Revisiting
593 nitrous acid (HONO) emission from on-road vehicles: A tunnel study with a mixed fleet, *J. Air Waste Manag.*, 67,
594 797-805, <https://doi.org/10.1080/10962247.2017.1293573>, 2017.
- 595 Liu, Y. H., Lu, K. D., Li, X., Dong, H. B., Tan, Z. F., Wang, H. C., Zou, Q., Wu, Y. S., Zeng, L. M., Hu, M., Min, K.
596 E., Kecorius, S., Wiedensohler, A., and Zhang, Y. H.: A Comprehensive Model Test of the HONO Sources
597 Constrained to Field Measurements at Rural North China Plain, *Environ. Sci. Technol.*, 53, 3517-3525,
598 <https://doi.org/10.1021/acs.est.8b06367>, 2019.
- 599 Liu, Z., Wang, Y. h., Costabile, F., Amoroso, A., Zhao, C., Huey, L. G., Stickel, R., Liao, J., and Zhu, T.: Evidence
600 of Aerosols as a Media for Rapid Daytime HONO Production over China, *Environ. Sci. Technol.*, 48, 14386-14391,
601 <https://doi.org/10.1021/es504163z>, 2014.
- 602 Liu, Z., Wang, Y., Gu, D., Zhao, C., Huey, L. G., Stickel, R., Liao, J., Shao, M., Zhu, T., Zeng, L., Amoroso, A.,
603 Costabile, F., Chang, C.-C., and Liu, S.-C.: Summertime photochemistry during CAREBeijing-2007: RO_x budgets
604 and O₃ formation, *Atmos. Chem. Phys.*, 12, 7737-7752, <https://doi.org/10.5194/acp-12-7737-2012>, 2012.
- 605 Lu, K. D., Rohrer, F., Holland, F., Fuchs, H., Bohn, B., Brauers, T., Chang, C. C., Häsel, R., Hu, M., Kita, K.,
606 Kondo, Y., Li, X., Lou, S. R., Nehr, S., Shao, M., Zeng, L. M., Wahner, A., Zhang, Y. H., and Hofzumahaus, A.:
607 Observation and modelling of OH and HO₂ concentrations in the Pearl River Delta 2006: a missing OH source in a
608 VOC rich atmosphere, *Atmos. Chem. Phys.*, 12, 1541-1569, <https://doi.org/10.5194/acp-12-1541-2012>, 2012.
- 609 Ma, Q. X., Wang, T., Liu, C., He, H., Wang, Z., Wang, W. H., and Liang, Y. T.: SO₂ Initiates the Efficient
610 Conversion of NO₂ to HONO on MgO Surface, *Environ. Sci. Technol.*, 51, 3767-3775,
611 <https://doi.org/10.1021/acs.est.6b05724>, 2017.
- 612 Mendez, M., Blond, N., Amedro, D., Hauglustaine, D. A., Blondeau, P., Afif, C., Fittschen, C., and Schoemaeker,
613 C.: Assessment of indoor HONO formation mechanisms based on in situ measurements and modeling, *Indoor Air*,
614 27, 443-451, <https://doi.org/10.1111/ina.12320>, 2017.
- 615 Meusel, H., Kuhn, U., Reiffs, A., Mallik, C., Harder, H., Martinez, M., Schuladen, J., Bohn, B., Parchatka, U.,
616 Crowley, J. N., Fischer, H., Tomsche, L., Novelli, A., Hoffmann, T., Janssen, R. H. H., Hartogensis, O., Pikridas,
617 M., Vrekoussis, M., Boursoukoudis, E., Weber, B., Lelieveld, J., Williams, J., Pöschl, U., Cheng, Y. F., and Su, H.:
618 Daytime formation of nitrous acid at a coastal remote site in Cyprus indicating a common ground source of
619 atmospheric HONO and NO, *Atmos. Chem. Phys.*, 16, 14475-14493, <https://doi.org/10.5194/acp-16-14475-2016>,
620 2016.
- 621 Michoud, V., Colomb, A., Borbon, A., Miet, K., Beekmann, M., Camredon, M., Aumont, B., Perrier, S., Zapf, P.,
622 Siour, G., Ait-Helal, W., Afif, C., Kukui, A., Furger, M., Dupont, J. C., Haefelin, M., and Doussin, J. F.: Study of
623 the unknown HONO daytime source at a European suburban site during the MEGAPOLI summer and winter field
624 campaigns, *Atmos. Chem. Phys.*, 14, 2805-2822, <https://doi.org/10.5194/acp-14-2805-2014>, 2014.
- 625 Michoud, V., Doussin, J.-F., Colomb, A., Afif, C., Borbon, A., Camredon, M., Aumont, B., Legrand, M., and
626 Beekmann, M.: Strong HONO formation in a suburban site during snowy days, *Atmos. Environ.*, 116, 155-158,
627 <https://doi.org/10.1016/j.atmosenv.2015.06.040>, 2015.
- 628 Monge, M. E., D'Anna, B., Mazri, L., Giroir-Fendler, A., Ammann, M., Donaldson, D. J., and George, C.: Light
629 changes the atmospheric reactivity of soot, *P. Natl. Acad. Sci. USA*, 107, 6605-6609,
630 <https://doi.org/10.1073/pnas.0908341107>, 2010.
- 631 Neuman, J. A., Trainer, M., Brown, S. S., Min, K. E., Nowak, J. B., Parrish, D. D., Peischl, J., Pollack, I. B.,



- 632 Roberts, J. M., Ryerson, T. B., and Veres, P. R.: HONO emission and production determined from airborne
633 measurements over the Southeast U.S. *J. Geophys. Res.-Atmos.*, 121, 9237-9250,
634 <https://doi.org/10.1002/2016JD025197>, 2016.
- 635 Oswald, R., Behrendt, T., Ermel, M., Wu, D., Su, H., Cheng, Y., Breuninger, C., Moravek, A., Mougou, E., Delon,
636 C., Loubet, B., Pommerening-Röser, A., Sörgel, M., Pöschl, U., Hoffmann, T., Andreae, M. O., Meixner, F. X. and
637 Trebs, I.: HONO Emissions from Soil Bacteria as a Major source of Atmospheric Reactive Nitrogen, *Science*, 341,
638 1233-1235, <https://doi.org/10.1126/science.1242266>, 2013.
- 639 Oswald, R., Ermel, M., Hens, K., Novelli, A., Ouwersloot, H. G., Paasonen, P., Petäjä, T., Sipilä, M., Keronen, P.,
640 Bäck, J., Königstedt, R., Hosaynali Beygi, Z., Fischer, H., Bohn, B., Kubistin, D., Harder, H., Martinez, M.,
641 Williams, J., Hoffmann, T., Trebs, I., and Sörgel, M.: A comparison of HONO budgets for two measurement
642 heights at a field station within the boreal forest in Finland, *Atmos. Chem. Phys.*, 15, 799-813,
643 <https://doi.org/10.5194/acp-15-799-2015>, 2015.
- 644 Pitts, J. N., Grosjean, D., Cauwenbergh, K. V., Schmid, J. P., and Fitz, D. R.: Photooxidation of aliphatic amines
645 under simulated atmospheric conditions: formation of nitrosamines, nitramines, amides, and photochemical
646 oxidant, *Environ. Sci. Technol.*, 12, 946-953, <https://doi.org/10.1021/es60144a009>, 1978.
- 647 Bröske, R., Kleffmann, J., and Wiesen, P.: Heterogeneous conversion of NO₂ on secondary organic aerosol
648 surfaces: A possible source of nitrous acid (HONO) in the atmosphere?, *Atmos. Chem. Phys.*, 3, 469-474,
649 <https://doi.org/10.5194/acp-3-469-2003>, 2003.
- 650 Reisinger, A. R.: Observations of HNO₂ in the polluted winter atmosphere: possible heterogeneous production on
651 aerosols, *Atmos. Environ.*, 34, 3865-3874, [https://doi.org/10.1016/S1352-2310\(00\)00179-5](https://doi.org/10.1016/S1352-2310(00)00179-5), 2000.
- 652 Scharko, N. K., Martin, E. T., Losovyj, Y., Peters, D. G., and Raff, J. D.: Evidence for Quinone Redox Chemistry
653 Mediating Daytime and Nighttime NO₂-to-HONO Conversion on Soil Surfaces, *Environ. Sci. Technol.*, 51,
654 9633-9643, <https://doi.org/10.1021/acs.est.7b01363>, 2017.
- 655 Sleiman, M., Gundel, L. A., Pankow, J. F., Jacob III, P., Singer, B. C., and Destailhats, H.: Formation of
656 carcinogens indoors by surface-mediated reactions of nicotine with nitrous acid, leading to potential thirdhand
657 smoke hazards, *P. Natl. Acad. Sci. USA*, 107, 6576-6581, <https://doi.org/10.1073/pnas.0912820107>, 2010.
- 658 Sörgel, M., Regelin, E., Bozem, H., Diesch, J. M., Drewnick, F., Fischer, H., Harder, H., Held, A.,
659 Hosaynali-Beygi, Z., Martinez, M., and Zetzsch, C.: Quantification of the unknown HONO daytime source and its
660 relation to NO₂, *Atmos. Chem. Phys.*, 11, 10433-10447, <https://doi.org/10.5194/acp-11-10433-2011>, 2011.
- 661 Spataro, F., Ianniello, A., Esposito, G., Allegrini, I., Zhu, T., and Hu, M.: Occurrence of atmospheric nitrous acid in
662 the urban area of Beijing (China), *Sci. Total Environ.*, 447, 210-224,
663 <https://doi.org/10.1016/j.scitotenv.2012.12.065>, 2013.
- 664 Spindler, G., Brüggemann, E., and Herrmann, H.: Nitrous acid (HNO₂) Concentration Measurements and
665 Estimation of Dry Deposition over Grassland in Eastern Germany, *Transactions on Ecology and Environment*, 28,
666 223-227, 1999.
- 667 Stemmler, K., Ammann, M., Donders, C., Kleffmann, J., and George, C.: Photosensitized reduction of nitrogen
668 dioxide on humic acid as a source of nitrous acid, *Nature*, 440, 195-198, <https://doi.org/10.1038/nature04603>,
669 2006.
- 670 Stemmler, K., Ndour, M., Elshorbany, Y., Kleffmann, J., D'Anna, B., George, C., Bohn, B., and Ammann, M.:
671 Light induced conversion of nitrogen dioxide into nitrous acid on submicron humic acid aerosol, *Atmos. Chem.*
672 *Phys.*, 7, 4237-4248, <https://doi.org/10.5194/acp-7-4237-2007>, 2007.
- 673 Stutz, J., Alicke, B., Neftel, A.: Nitrous acid formation in the urban atmosphere: Gradient measurements of NO₂
674 and HONO over grass in Milan, Italy, *J. Geophys. Res.*, 107, LOP 5-1-LOP 5-15,
675 <https://doi.org/10.1029/2001JD000390>, 2002.



- 676 Stutz, J., Alicke, B., Ackermann, R., Geyer, A., Wang, S. H., White, A. B., Williams, E. J., Spicer, C. W., and Fast,
677 J. D.: Relative humidity dependence of HONO chemistry in urban areas, *J. Geophys. Res.-Atmos.*, 109, D03307,
678 <https://doi.org/10.1029/2003JD004135>, 2004a.
- 679 Stutz, J., Alicke, B., Ackermann, R., Geyer, A., White, A., and Williams, E.: Vertical profiles of NO₃, N₂O₅, O₃,
680 and NO_x in the nocturnal boundary layer: 1. Observations during the Texas Air Quality Study 2000, *J. Geophys.*
681 *Res.-Atmos.*, 109, D12306, <https://doi.org/10.1029/2003JD004209>, 2004b.
- 682 Su, H., Cheng, Y. F., Cheng, P., Zhang, Y. H., Dong, S. F., Zeng, L. M., Wang, X. S., Slanina, J., Shao, M., and
683 Wiedensohler, A.: Observation of nighttime nitrous acid (HONO) formation at a non-urban site during
684 PRIDE-PRD2004 in China, *Atmos. Environ.*, 42, 6219-6232, <https://doi.org/10.1016/j.atmosenv.2008.04.006>,
685 2008a.
- 686 Su, H., Cheng, Y. F., Shao, M., Gao, D. F., Yu, Z. Y., Zeng, L. M., Slanina, J., Zhang, Y. H., and Wiedensohler, A.:
687 Nitrous acid (HONO) and its daytime sources at a rural site during the 2004 PRIDE-PRD experiment in China, *J.*
688 *Geophys. Res.*, 113, D14312, <https://doi.org/10.1029/2007JD009060>, 2008b.
- 689 Su, H., Cheng, Y. F., Oswald, R., Behrendt, T., Trebs, I., Meixner, F. X., Andreae, M. O., Cheng, P., Zhang, Y. H.,
690 and Pöschl, U.: Soil nitrite as a Source of Atmospheric HONO and OH Radicals, *Science*, 333, 1616-1618,
691 <https://doi.org/10.1126/science.1207687>, 2011.
- 692 Sun, Y. L., Wang, Z. F., Fu, P. Q., Yang, T., Jiang, Q., Dong, H. B., Li, J., and Jia, J. J.: Aerosol composition,
693 sources and processes during wintertime in Beijing, China, *Atmos. Chem. Phys.*, 13, 4577-4592,
694 <https://doi.org/10.5194/acp-13-4577-2013>, 2013.
- 695 Tang, K., Qin, M., Duan, J., Fang, W., Meng, F. H., Liang, S. X., Xie, P. H., Liu, J. G., Liu, W. Q., Xue, C. Y., and
696 Mu, Y. J.: A dual dynamic chamber system based on IBBCEAS for measuring fluxes of nitrous acid in agricultural
697 fields in the North China Plain, *Atmos. Environ.*, 196, 10-19, <https://doi.org/10.1016/j.atmosenv.2018.09.059>,
698 2019.
- 699 Tang, Y., An, J., Wang, F., Li, Y., Qu, Y., Chen, Y., and Lin, J.: Impacts of an unknown daytime HONO source on
700 the mixing ratio and budget of HONO, and hydroxyl, hydroperoxyl, and organic peroxy radicals, in the coastal
701 regions of China, *Atmos. Chem. Phys.*, 15, 9381-9398, <https://doi.org/10.5194/acp-15-9381-2015>, 2015.
- 702 Tong, S. R., Hou, S. Q., Zhang, Y., Chu, B. W., Liu, Y. C., He, H., Zhao, P. S., and Ge, M. F.: Exploring the nitrous
703 acid (HONO) formation mechanism in winter Beijing: direct emissions and heterogeneous production in urban and
704 suburban areas, *Faraday Discuss.*, 189, 213-230, <https://doi.org/10.1039/c5fd00163c>, 2016.
- 705 VanDenBoer, T. C., Brown, S. S., Murphy, J. G., Keene, W. C., Young, C. J., Pszenny, A. A. P., Kim, S., Warneke,
706 C., de Gouw, J. A., Maben, J. R., Wagner, N. L., Riedel, T. P., Thornton, J. A., Wolfe, D. E., Dubé, W. P., Öztürk, F.,
707 Brock, C. A., Grossberg, N., Lefer, B., Lerner, B., Middlebrook, A. M., and Roberts, J. M.: Understanding the role
708 of the ground surface in HONO vertical structure: High resolution vertical profiles during NACHTT-11, *J. Geophys.*
709 *Res.- Atmos.*, 118, 10155-110171, <https://doi.org/10.1002/jgrd.50721>, 2013.
- 710 Villena, G., Kleffmann, J., Kurtenbach, R., Wiesen, P., Lissi, E., Rubio, M. A., Croxatto, G., and Rappenglück, B.:
711 Vertical gradients of HONO, NO_x and O₃ in Santiago de Chile, *Atmos. Environ.*, 45, 3867-3873,
712 <https://doi.org/10.1016/j.atmosenv.2011.01.073>, 2011.
- 713 Vogel, B., Vogel, H., Kleffmann, J., and Kurtenbach, R.: Measured and simulated vertical profiles of nitrous
714 acid—Part II. Model simulations and indications for a photolytic source, *Atmos. Environ.*, 37, 2957-2966,
715 [https://doi.org/10.1016/S1352-2310\(03\)00243-7](https://doi.org/10.1016/S1352-2310(03)00243-7), 2003.
- 716 Wang, H. C., Lu, K. D., Chen, X. R., Zhu, Q. D., Wu, Z. J., Wu, Y. S., and Sun, K.: Fast particulate nitrate
717 formation via N₂O₅ uptake aloft in winter in Beijing, *Atmos. Chem. Phys.*, 18, 10483-10495,
718 <https://doi.org/10.5194/acp-18-10483-2018>, 2018.
- 719 Wang, J. Q., Zhang, X. S., Guo, J., Wang, Z. W., and Zhang, M. G.: Observation of nitrous acid (HONO) in Beijing,



- 720 China: Seasonal variation, nocturnal formation and daytime budget, *Sci. Total Environ.*, 587-588, 350-359,
721 <https://doi.org/10.1016/j.scitotenv.2017.02.159>, 2017.
- 722 Wang, S. S., Zhou, R., Zhao, H., Wang, Z. R., Chen, L. M., and Zhou, B.: Long-term observation of atmospheric
723 nitrous acid (HONO) and its implication to local NO₂ levels in Shanghai, China, *Atmos. Environ.*, 77, 718-724,
724 <https://doi.org/10.1016/j.atmosenv.2013.05.071>, 2013.
- 725 Wong, K. W., Oh, H. -J., Lefer, B. L., Rappenglück, B., and Stutz, J.: Vertical profiles of nitrous acid in the
726 nocturnal urban atmosphere of Houston, TX, *Atmos. Chem. Phys.*, 11, 3595-3609,
727 <https://doi.org/10.5194/acp-11-3595-2011>, 2011.
- 728 Wong, K. W., Tsai, C., Lefer, B., Haman, C., Grossberg, N., Brune, W. H., Ren, X., Luke, W., and Stutz, J.:
729 Daytime HONO vertical gradients during SHARP 2009 in Houston, TX, *Atmos. Chem. Phys.*, 12, 635-652,
730 <https://doi.org/10.5194/acp-12-635-2012>.
- 731 Wong, K. W., Tsai, C., Lefer, B., Grossberg, N., and Stutz, J.: Modeling of daytime HONO vertical gradients
732 during SHARP 2009, *Atmos. Chem. Phys.*, 13, 3587-3601, <https://doi.org/10.5194/acp-13-3587-2013>, 2013.
- 733 Xu, W. Q., Sun, Y. L., Wang, Q. Q., Zhao, J., Wang, J. F., Ge, X. L., Xie, C. H., Zhou, W., Du, W., Li, J., Fu, P. Q.,
734 Wang, Z. F., Worsnop, D. R., and Coe, H.: Changes in Aerosol Chemistry From 2014 to 2016 in Winter in Beijing:
735 Insights From High-Resolution Aerosol Mass Spectrometry, *J. Geophys. Res.- Atmos.*, 124, 1132-1147,
736 <https://doi.org/10.1029/2018JD029245>, 2019.
- 737 Xu, Z., Wang, T., Wu, J. Q., Xue, L. K., Chan, J., Zha, Q., Z., Zhou, S. Z., Louie, P. K. K., and Luk, C. W. Y.:
738 Nitrous acid (HONO) in a polluted subtropical atmosphere: Seasonal variability, direct vehicle emissions and
739 heterogeneous production at ground surface, *Atmos. Environ.*, 106, 100-109,
740 <https://doi.org/10.1016/j.atmosenv.2015.01.061>, 2015.
- 741 Ye, C. X., Zhang, N., Gao, H. L., and Zhou, X. L.: Photolysis of Particulate Nitrate as a Source of HONO and NO_x,
742 *Environ. Sci. Technol.*, 51, 6849-6856, <https://doi.org/10.1021/acs.est.7b00387>, 2017.
- 743 Ye, C. X., Zhou, X. L., Pu, D., Stutz, J., Festa, J., Spolaor, M., Tsai, C., Cantrell, C., Mauldin III, R. L.,
744 Weinheimer, A., Hornbrook, R. S., Apel, E. C., Guenther, A., Kaser, L., Yuan, B., Karl, T., Haggerty, J., Hall, S.,
745 Ullmann, K., Smith, J., and Ortega, J.: Tropospheric HONO distribution and chemistry in the southeastern US,
746 *Atmos. Chem. Phys.*, 18, 9107-9120, <https://doi.org/10.5194/acp-18-9107-2018>, 2018.
- 747 Yu, Y., Galle, B., Panday, A., Hodson, E., Prinn, R., and Wang, S.: Observations of high rates of NO₂-HONO
748 conversion in the nocturnal atmospheric boundary layer in Kathmandu, Nepal, *Atmos. Chem. Phys.*, 9, 6401-6415,
749 <https://doi.org/10.5194/acp-9-6401-2009>, 2009.
- 750 Zhang, N., Zhou, X. L., Shepson, P. B., Gao, H. L., Alaghmand, M., and Stirm, B.: Aircraft measurement of
751 HONO vertical profiles over a forested region, *Geophys. Res. Lett.*, 36, L15820,
752 <https://doi.org/10.1029/2009GL038999>, 2009.
- 753 Zhang, R., Wang, G., Guo, S., Zamora, M. L., Ying, Q., Lin, Y., Wang, W., Hu, M., and Wang, Y.: Formation of
754 Urban Fine Particulate Matter, *Chem. Rev.*, 115, 3303-3855, <https://doi.org/10.1021/acs/chemrev.5b00067>, 2015.
- 755 Zhang, W. Q., Tong, S. R., Ge, M. F., An, J. L., Shi, Z. B., Hou, S. Q., Xia, K. H., Qu, Y., Zhang, H. X., Chu, B. W.,
756 Sun, Y. L., and He, H.: Variations and sources of nitrous acid (HONO) during a severe pollution episode in Beijing
757 in winter 2016, *Sci. Total Environ.*, 648, 253-262, <https://doi.org/10.1016/j.scitotenv.2018.08.133>, 2018.



Table

Table 1 Classification of meteorological conditions and corresponding concentrations of NR-PM₁, NO₂ and HONO from 7 to 12 December

Time period	Weather condition	NR-PM ₁ (µg·m ⁻³)	HONO (ppb)	NO ₂ (ppb)	WS (m·s ⁻¹)	WD	T (°C)	RH (%)
7 Dec - 8 Dec (10:00)	Haze (E1)	30-184	1.49-7.59	24.91-65.48	0.03-1.95	NW-ESE ^a	1.6-9.3	36-82
8 Dec (10:00) - 11 Dec	Clean (C2)	3-97	0.05-3.75	3.33-47.84	0.01-6.24	NE-NW	-2.4-9.1	16-53
11 Dec - 12 Dec	Haze (E3)	69-217	1.54-5.51	38.58-66.57	0.02-1.81	NE-NW	-1.6-6.9	40-69

^a NE: Northeast; ESE: East-southeast; NW: Northwest;



Figures

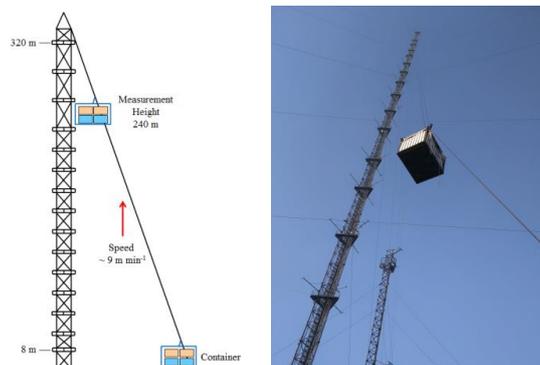


Figure 1. The Beijing 325-m meteorological tower (BMT) at the Institute of Atmospheric Physics (IAP).

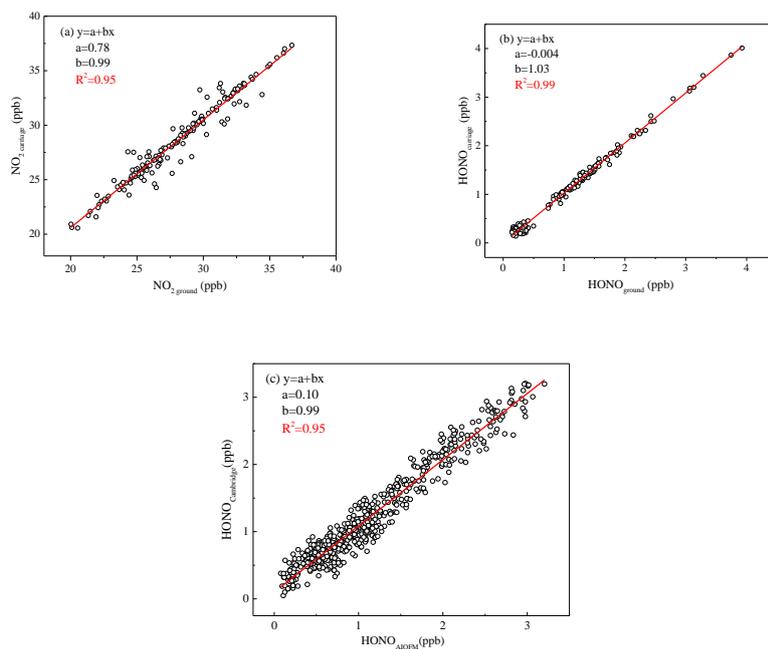


Figure 2. (a) Correlation of NO_2 concentration was measured by the two IBCEAS instruments; (b) correlation of HONO concentration was measured by the two IBCEAS instruments; (c) inter-comparison between the IBCEAS of Cambridge university and the IBCEAS of Anhui Institute of Optics and Fine Mechanics (AIOFM).

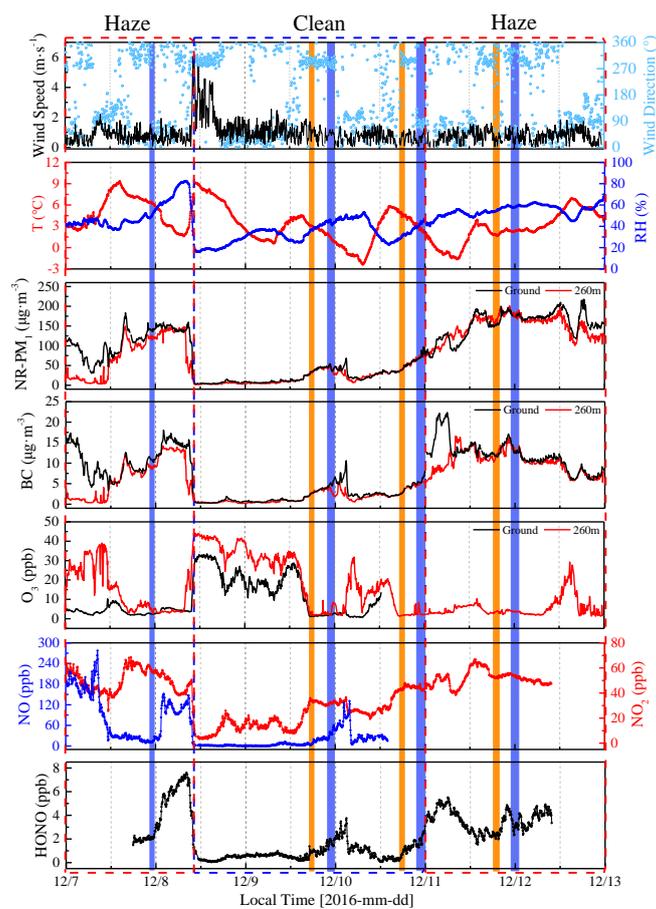


Figure 3. Time-series of wind speed (WS) and direction (WD), temperature (T), relative humidity (RH), NR-PM₁, BC, O₃, NO, NO₂, and HONO from 7 to 12 December 2016 at IAP-Tower Division in Beijing, China. The shaded region represents the eight vertical measurements (Table S1). The orange shaded region represents the vertical measurements after sunset, and the violet shaded region represents the vertical measurements at night and midnight.

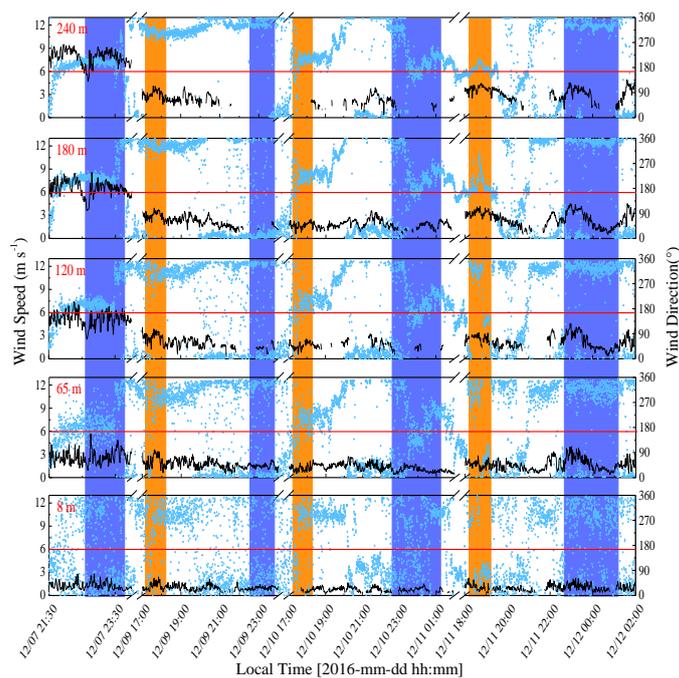


Figure 4. Temporal variations of the wind speed at fixed heights on the meteorological tower (8, 65, 120, 180, and 240 m) throughout the measurement period. Vertical measurements were conducted after sunset (orange shaded regions) and during nighttime (violet shaded regions). The red line denoted the wind speed is $6 \text{ m} \cdot \text{s}^{-1}$.

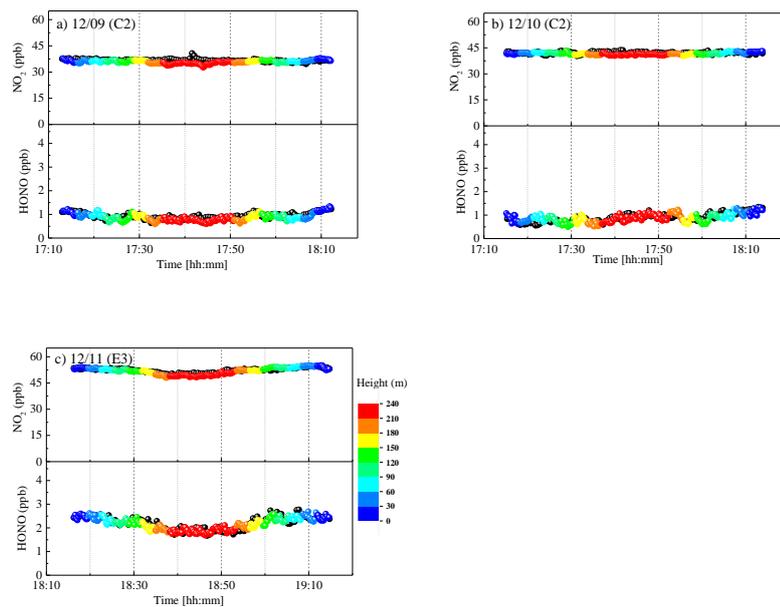


Figure 5. Temporal variations in the mixing ratios of HONO and NO_2 at ground level (black solid circles) and in the container (colored solid circles) during (a, b) the clean episode (C2) and (c) the haze episode (E3). The vertical measurements of HONO and NO_2 are color code by height.

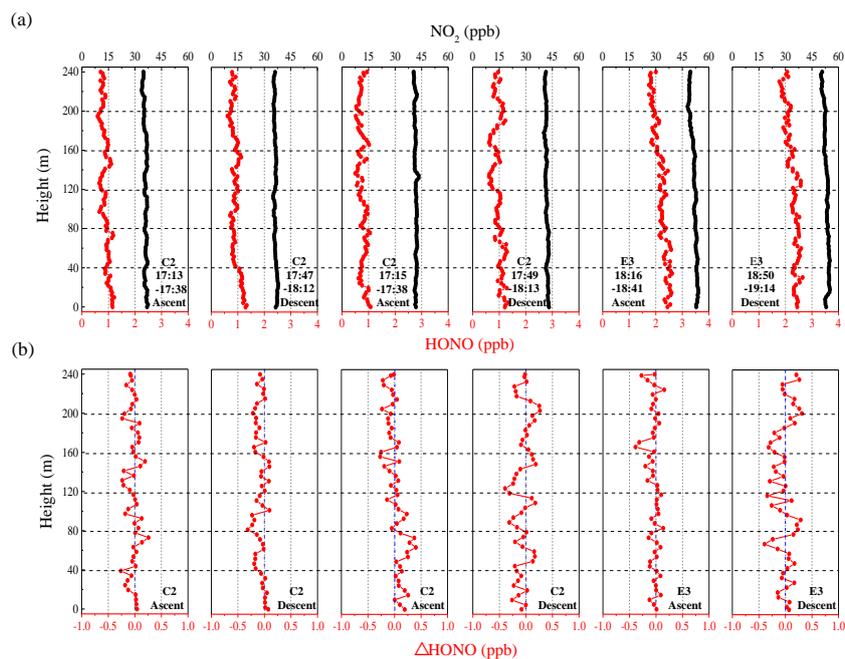


Figure 6. (a) Vertical profiles of HONO (red solid line) and NO_2 (black solid line) after sunset during the clean episode (C2) and the haze episode (E3). Each vertical measurement includes the ascent process and the descent process. The time on each plot corresponds to the measurement time of the vertical profile during the ascent or descent process. (b) Vertical profiles of ΔHONO (red dotted line) during C2 and E3 are also shown.

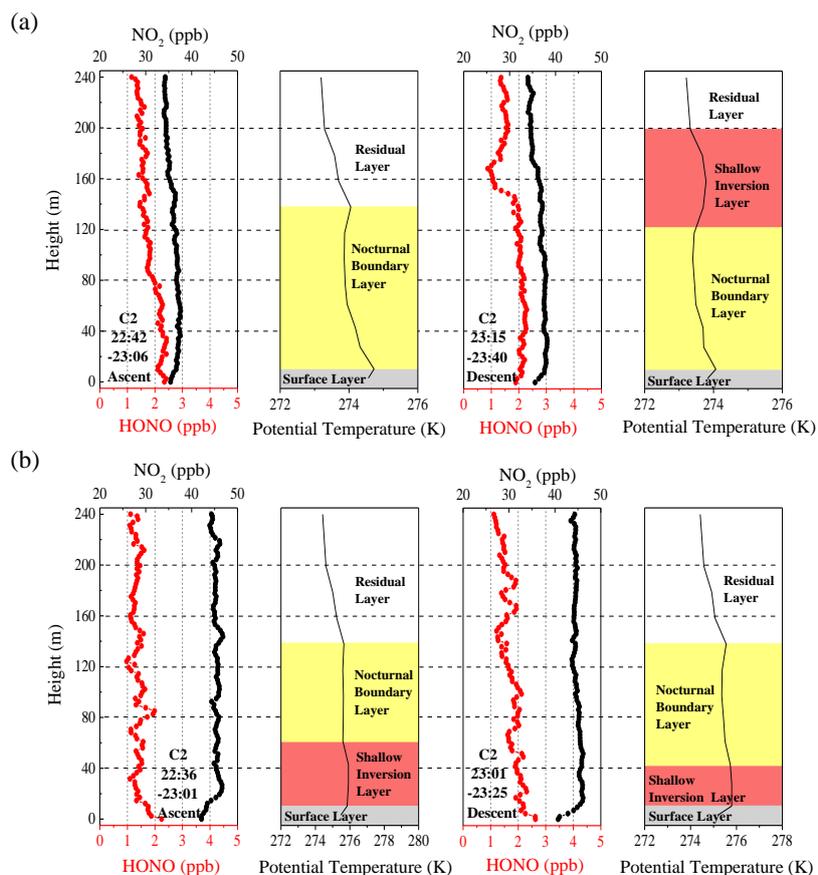


Figure 7. Nocturnal vertical profiles of HONO, NO₂ and potential temperature during the ascent and descent of the container on (a) 9 and (b) 10 December. The time in the figure corresponds to the measurement time of the vertical profile of HONO and NO₂. The different colors shaded region indicates the nocturnal small-scale stratification (surface layer, nocturnal boundary layer, shallow inversion layer and residual layer). The heights of the surface layer, shallow inversion layer, nocturnal boundary layer and residual layer are denoted by grey shaded regions, pink shaded regions, yellow shaded regions, and white shaded regions, respectively.

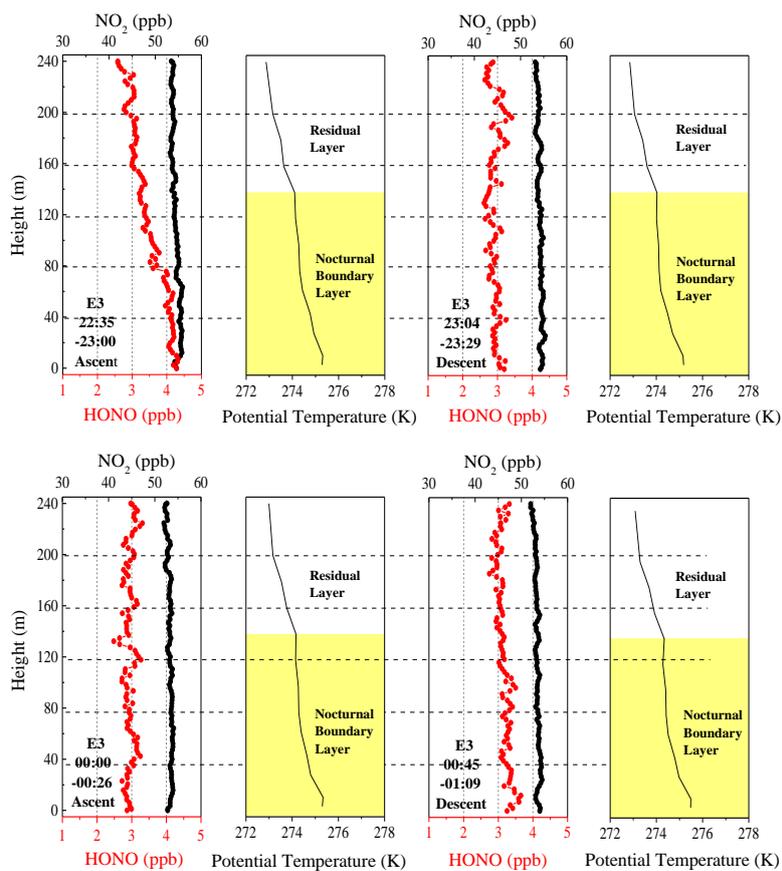


Figure 8. Vertical profiles of HONO and NO₂ on the night of 11 December and midnight of 12 December. The potential temperature profiles indicate nocturnal small-scale stratification (nocturnal boundary layer and residual layer). The height of the nocturnal boundary layer (NBL) is denoted by the yellow shaded region. The time in the figure corresponds to the measurement time of the vertical profile of HONO and NO₂.

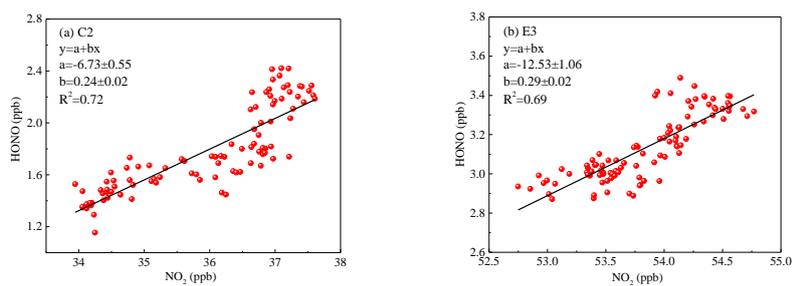


Figure 9. The correlation of the vertical profiles between HONO and NO_2 during (a) the clean episode (C2) and (b) the haze episode (E3).

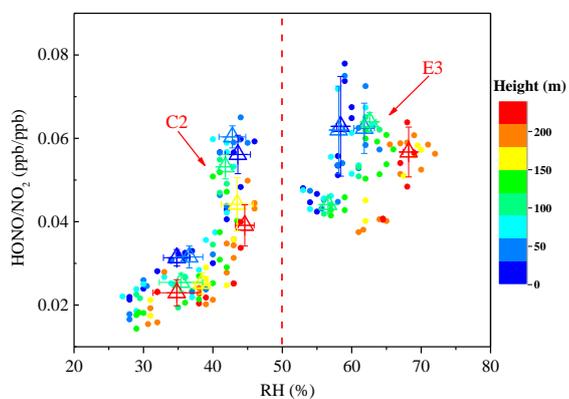


Figure 10. Scatter plot of HONO/NO_2 against RH of all vertical profiles during the clean episode (C2) and the haze episode (E3). The HONO/NO_2 ratio is color coded by the heights. Triangles are the average of the first five HONO/NO_2 value in each 10% RH interval at different height intervals (8-65 m, 65-120 m, 120-180 m and 180-240 m).

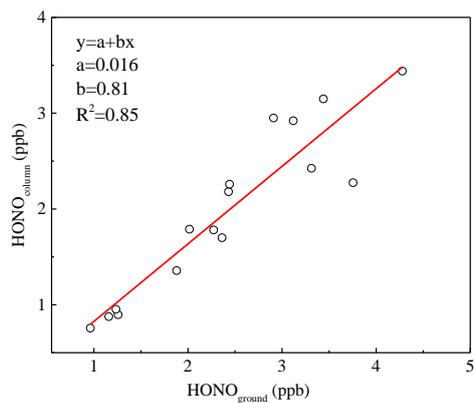


Figure 11. Correlation between the integrated column of HONO (10-240 m) and HONO measured from ground level to 10 m above ground level (AGL). Column values of vertical measurements were calculated for 9 to 12 December.