



# Composition and origin of PM<sub>2.5</sub> aerosol particles in the upper Rhine valley in summer

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**Abstract.** We conducted a six-week measurement campaign in summer 2016 at a rural site about 11 km north of the city of  
15 Karlsruhe in southwest Germany in order to study the chemical composition and origin of aerosols in the upper Rhine valley. In particular, we deployed a single particle mass spectrometer (LAAPTOF) and an aerosol mass spectrometer (AMS) to provide complementary chemical information of the aerosol particles smaller than 2.5 μm. For the entire measurement period, the total aerosol particle mass was dominated by sodium salts contributing on average (36 ± 27)% to the total single particles. The total particulate organic compounds, sulfate, nitrate, and ammonium contributed on average (58 ± 12)%, (22 ± 7)%, (10 ± 1)%, and  
20 (9 ± 3)% to the total non-refractory particle mass, respectively. The regional transport model COSMO-ART was applied for source apportionment and to achieve a better understanding of the impact of complex transport pattern on the field observations. Combining field observations and model simulations, we attributed high particle numbers and SO<sub>2</sub> concentrations observed at this rural site to industrial emissions from power plants and a refinery in Karlsruhe. In addition, two characteristic episodes with aerosol particle mass dominated by sodium salts particles comprising (70 ± 24)% of the total single particles and organic  
25 compounds comprising accounting for (77 ± 6)% of total non-refractory species, respectively, were investigated in detail. For the first episode, we identified relatively fresh and aged sea salt particles originating from the Atlantic Ocean more than 800 km away. These particles showed markers like m/z 129 C<sub>5</sub>H<sub>7</sub>NO<sub>3</sub><sup>+</sup> indicating the influence of anthropogenic emissions modifying their composition e.g., from chloride to nitrate salts during the long-range transport. For a 3-day episode including high organic mass concentrations, model simulations show that on average (74 ± 7)% of the particulate organics at this site were of biogenic  
30 origin. Detailed model analysis allowed us to find out that three subsequent peaks of high organic mass concentrations originated from different sources, including local emissions from the city and industrial area of Karlsruhe, regional transport from the city of Stuttgart (~64 km away), and potential local night-time formation and growths. Biogenic (forest) and anthropogenic (urban)



emissions were mixed during transport and contributed to the formation of organic particles. In addition, topography, temperature inversion, and stagnant meteorological conditions also played a role in the build-up of higher organic particle mass concentrations. Furthermore, the model was evaluated using the field observations and corresponding sensitivity tests. The model results show good agreement with trends and concentrations observed for several trace gases (e.g., O<sub>3</sub>, NO<sub>2</sub>, and SO<sub>2</sub>) and aerosol particle compounds (e.g., ammonium and nitrate). However, the model underestimates the number of particles by an order of magnitude and underestimates the mass of organic particles by a factor of 2.3. The discrepancy was expected for particle number since the model does not include all nucleation processes. The missing organic mass indicates either an underestimated regional background, missing sources, and/or mechanisms in the model like night-time chemistry. This study demonstrates the potential of combining comprehensive field observations with dedicated transport modelling to understand the chemical composition and complex origin of aerosols.

## 1 Introduction

Aerosols are ubiquitous in the atmosphere and play a crucial role in air quality, human health, ecosystem and climate change. In particular, aerosol particles modify atmospheric radiative fluxes and can serve as cloud condensation nuclei (CCN) and/or be activated as ice nuclei (IN), which influences the cloud formation, properties (albedo) and precipitation, thus has further impact on weather and climate change (Fuzzi et al., 2015; Pöschl, 2005; Seinfeld and Pandis, 2006). Such indirect impacts of aerosols on climate are still related to large uncertainties and low confidence levels (IPCC, 2013). There are various types of atmospheric aerosol particles, such as mineral dust, sea salt, soot, and organic aerosol particles, originating from biogenic and/or anthropogenic emission sources, as well as secondary formation from precursors within the atmosphere (Tsigaridis et al., 2006). Aerosol particles are not typically purely inorganic or organic; instead, they are often composed of different inorganic and organic species. The diverse origins and atmospheric aging cause the complexity of the aerosol particle mixtures.

Real-time mass spectrometry is widely used to measure size and chemical composition of aerosol particles with high temporal resolution. Two major instrument types are single particle mass spectrometers (single-particle MS) and aerosol mass spectrometers (AMS). The single-particle MS is sensitive to both non-refractory and refractory species and to the mixing state of individual particles, and thus identifies particle types (Murphy, 2007), while the AMS can quantify mass concentrations of non-refractory particulate compounds in bulk samples (Canagaratna et al., 2007). Complementary data from single-particle MS and AMS allows quantitative interpretation of the internal and external mixing state of the aerosol (Gemayel et al., 2017; Gansch et al., 2018; Healy et al., 2013; Jeong et al., 2011; Shen et al., 2019; Zhou et al., 2016). The mixing state of the aerosol particles can reflect their evolution e.g. caused by chemical reactions leading to changing compositions or coatings (Reitz et al., 2016). A good example is the aging of sea salt particles: chloride replacement by nitrate on sea-salt particles can be well identified by single-particle MS (Gard et al., 1998).



The large diversity of aerosols, containing hundreds of gaseous and particulate components, makes it difficult to trace their sources. Therefore, several source apportionment methods have been developed with the common aim of reducing the complexity of the aerosol particle mixtures and thus helping to identify the major particle types in the atmosphere. Clustering algorithms have been used commonly for the single-particle MS data analysis (Reitz et al., 2016), while positive matrix factorization (PMF; Paatero, 1997; Paatero and Tapper, 1994) is commonly applied for AMS data analysis (Canonaco et al., 2013; Canonaco et al., 2015; Crippa et al., 2014; Zhang et al., 2011). However, these methods can only provide indirect source information. The application of a transport model gives additional information. It can predict the state of the atmosphere based on detailed meteorology, emission data, as well as conversions of reactive species and thus allows studying the aerosol evolution from the emissions to a site of interest.

There exists a variety of so-called online coupled chemical transport models. A recent review is given by Baklanov et al. (2014). Such models are important tools to better understand aerosol particles' sources and aging processes, and their impact on particles' climate-relevant properties, such as cloud condensation nuclei (CCN) activity, optical properties, and hygroscopicity (Fierce et al., 2017). Models can be improved with a good process understanding, e.g., emissions, transport and deposition, and corresponding parameterisations obtained from dedicated laboratory studies (Burkholder et al., 2017). They are often used to predict the evolution of the atmospheric composition on different spatial scales, which can be evaluated by comparison with field measurements. Besides, comparisons between model simulations and field observations also help to validate the model or to identify weaknesses like poorly represented chemistry, missing sources or mechanisms.

For comparison with measurements at a single location, regional scale models with small grid sizes (down to a few kilometres) are required. In addition, phenomena such as mountain and valley winds, land-sea breezes or lee waves become important, which requires a non-hydrostatic formulation of the model equations. In this study we use the regional-scale model system COSMO-ART (Consortium for small-scale modelling - aerosols and reactive trace gases; Vogel et al., 2009). The part of the model describing aerosols and reactive trace gases was developed at the Karlsruhe Institute of Technology (<http://www.imk-tro.kit.edu/english/3509.php>). It extends the operational weather forecast model COSMO of the German Weather Service (DWD), and it allows studying the feedback processes between aerosol particles and the state of the atmosphere covering the continental to the regional scale.

Mobile particle measurements on a tram line intersecting the city of Karlsruhe in southwest Germany and in the upper Rhine valley, showed particle number concentrations in the rural area north of the city almost as high as in downtown (particle size > 4 nm; mean value =  $1.2 \times 10^4 \text{ cm}^{-3}$ , maximum values  $3 \times 10^4 \text{ cm}^{-3}$ ) (Hagemann et al., 2014). The Rhine valley provides a well-known wind channel of large-scale flow. This results in faster transport of pollutants along the valley axis, e.g.,  $\text{NO}_x$ ,  $\text{SO}_2$ , volatile organic compounds (VOCs) and particles (Gross and Wippermann, 1987; Hagemann et al., 2014; Kalthoff and Vogel, 1992; Vogel et al., 1986). According to data from the local air quality monitoring administration (LUBW, <https://www.lubw.baden-wuerttemberg.de/luft>) from 2016 to 2018, the annual averaged mass concentrations of  $\text{NO}_2$ ,  $\text{SO}_2$ ,  $\text{PM}_{10}$ , and soot are 20, 2, 16, and  $1.8 \mu\text{g m}^{-3}$ , respectively. They were measured at Eggenstein monitoring station (3.3 km south



of the measurement site used in this study and about 9 km north of the city of Karlsruhe). Specifically, NO<sub>2</sub> concentrations are about half of the values measured in downtown Karlsruhe, but they exhibit comparable peak values (around 50–80 μg m<sup>-3</sup>). SO<sub>2</sub> levels at Eggenstein are the second highest in the state of Baden-Württemberg (from the available data from 2016 to 2018), and maximum hourly values can reach up to ~50 μg m<sup>-3</sup>.

100 In order to understand the sources for high local particle concentrations as well as the chemical nature and mass concentrations of aerosol types prevailing in the upper Rhine valley in summer-time we combine comprehensive aerosol information from point measurements with a regional transport model. Besides a better understanding of the origin of the aerosol in that region, we want to validate the capabilities of our transport model to predict characteristic air quality parameters.

## 2 Methods

### 105 2.1 Measurement site and instrumentation

The measurement site (49°6'10.54" N, 8°24'26.07" E) is in the upper Rhine valley 175 m north of the village of Leopoldshafen and 10 m east of the tramline No S1/11 described by Hagemann et al., (2014) (marked with a star in Fig. 1). The Rhine valley often acts as a wind channel leading to wind directions mainly from southwest or northeast. As shown in Fig. 1, the measurement site is located ~11 km north of downtown Karlsruhe and ~5 to 15 km northeast of an industrial area, including a refinery with 110 15.5 Mt/yr capacity, a 1450 MW hard coal fired power plant, as well as a 365 MW gas fired power plant (Hagemann et al., 2014). Therefore, the measurement site is potentially subject to the urban and industrial emissions especially for southerly wind directions. Apart from that, there are several nearby biogenic emission sources, e.g., forests, sunflower fields, as well as other agricultural areas (Hagemann et al., 2014; Huang et al., 2019). Local traffic emissions from the minor road near the measurement site was identified by measurements with high time resolution. Its contributions to the aerosol load are small due to the generally 115 weak traffic.

The instrumentation for this campaign has been described in detail in previous publications (Huang et al., 2019; Shen et al., 2019). Briefly, the data we use here are from the 6-week (26 July to 31 August 2016) deployment of a laser ablation aerosol particles time-of-flight mass spectrometer (LAAPTOF; AeroMegt GmbH) and a high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS; Aerodyne Inc.; hereafter AMS). These instruments were used to provide real time information 120 on size and chemical composition for individual particles and bulk samples, respectively. The LAAPTOF is a commercially available single particle mass spectrometer and has been well described in recent publications (Ahern et al., 2016; Gemayel et al., 2016; Gemayel et al., 2017; Marsden et al., 2018; Marsden et al., 2016; Marsden et al., 2019; Ramisetty et al., 2018; Reitz et al., 2016; Shen et al., 2018; Shen et al., 2019). In brief, aerosol particles with a size range of 70 nm to 2.5 μm vacuum aerodynamic diameter ( $d_{va}$ ) are sampled with a flowrate of ~80 cm<sup>3</sup> min<sup>-1</sup>, focused and accelerated via an aerodynamic lens 125 (ADL). Afterwards, the particles of ~200 nm to 2.5 μm  $d_{va}$  can be detected with two laser diodes ( $\lambda = 405$  nm) mounted in the



particle time-of-flight (PTOF) chamber. Once a single particle is detected successively by both of the detection lasers, its  $d_{va}$  is calculated from the time of flight and an excimer laser ( $\lambda = 193$  nm) is triggered to desorb and ionize the refractory and non-refractory species of the particle. The resulting cations and anions are analysed with a bipolar ToF mass spectrometer. The AMS measures the particle size-resolved bulk chemical composition, e.g., organic compounds, ammonium, nitrate, sulfate, and chloride, in mass concentrations with high time resolution and sensitivity (DeCarlo et al., 2006). Similarly as the LAAPTOF, aerosols are sampled with a flowrate of  $\sim 84$  cm<sup>3</sup> min<sup>-1</sup> via a similar ADL in the size range 70 nm to 2.5  $\mu$ m  $d_{va}$  and then pass through a PTOF chamber. At the end of the PTOF chamber, particles encounter a 600 °C heater that vaporises the non-refractory species. The corresponding vapours are ionized by electron impact with high energy of 70 eV, generating positive ions that are analysed with a ToF mass spectrometer. To account for a particle bouncing effect, a collection efficiency (CE, the product of net particle transmission and detection efficiency) is used (Canagaratna et al., 2007). The CE depends on composition and phase of the particles (Bahreini et al., 2005). Therefore, composition dependent CEs were used to correct the AMS data for this measurement campaign.

All sampling inlets (PM<sub>10</sub>, PM<sub>2.5</sub>, TSP) were positioned at the top of a measurement container (3.7 m above ground level, a.g.l.). Particle number (size >2.5 and >7 nm, CPC3776 and CPC 3022A, TSI Inc.), particle mass (total suspended particulates, PM<sub>10</sub>, and PM<sub>2.5</sub>; FIDAS, PALAS GmbH), black carbon (AE33-7 Aethalometer, Magee Scientific Co.), and trace gases O<sub>3</sub> (O341M, Environment SA), SO<sub>2</sub> (AF22M, Environment SA), NO<sub>2</sub> (AS32M, Environment SA) CO<sub>2</sub> (NGA2000, Rosemont Inc.) were measured. Meteorological parameters temperature, relative humidity, solar radiation, wind direction and speed (WS700, Lufft GmbH) were measured 10 m a.g.l. and complemented by meteorological data from a 200 m tower located at the KIT campus north,  $\sim 2$  km east of the measurement site (49°5'43.0"N 8°25'46.0"E).

## 2.2 Regional transport model COSMO-ART

The regional transport model COSMO-ART is used in this study to simulate an episode with high mass concentration of organics (up to  $\sim 25$   $\mu$ g m<sup>-3</sup> measured by AMS). Detailed model descriptions can be found in previous studies (Bangert et al., 2011; Knote et al., 2011; Stanelle et al., 2010; Vogel et al., 2009). Therefore, we will only give a brief review here. COSMO-ART is a fully online coupled model system, which uses the same spatial and temporal resolution for all scalars, such as temperature, gas and aerosol particle concentrations. The meteorological module is based on the operational weather forecast model COSMO of the German weather service (Deutscher Wetterdienst, DWD). When atmospheric chemistry and aerosol dynamics are coupled to COSMO, it allows the online calculation of the transport of reactive trace gases and aerosol particles as well as their interaction with the state of the atmosphere. Physical processes, e.g., nucleation, coagulation, condensation and sedimentation are included in the model. The aerosol particles are represented by five interactive lognormal modes representing particles consisting of internally mixed sulfate, ammonium, nitrate and organic compounds, pure soot, as well as internal mixtures of them all. The hourly biogenic emissions are calculated online according to the land use data, the simulated temperatures, and radiative fluxes (Vogel et al., 1995). The anthropogenic emissions, including traffic emissions, large point and area sources such as industrial



and households' areas, were pre-calculated by J. Bieser at Helmholtz Zentrum Geesthacht (Backes et al., 2016; Bieser et al., 2011). In our study, the simulation domain is covered by a 350×330 grid with a horizontal resolution of 1×1 km<sup>2</sup> and 50 vertical atmospheric layers (nonlinear increase up to 22 km a.g.l.). Most of the simulation results shown in this paper correspond to the lowest atmospheric level (10 m a.g.l.) and we indicate if results are from higher levels. The hourly emissions of SO<sub>2</sub>, NO<sub>x</sub>, CO, NH<sub>3</sub>, anthropogenic volatile organic compounds (VOCs), and primary organic aerosols (POA) were pre-calculated. As shown in Fig S1, VOCs, NO<sub>x</sub> and POA are mainly emitted at a lower atmospheric level in urban areas, such as the cities of Karlsruhe, Stuttgart, Mannheim, and Strasburg, as well as the corresponding highways. SO<sub>2</sub> hot spots typically correspond to the stack heights of the power plants emitting, such as the one in the western part of Karlsruhe.

### 3 Results and Discussion

In the first section, we give an overview of the meteorological conditions, the trace gases, and particles measured during the whole measurement campaign and will then give an example for source identification using the transport model. In the following two sections, we discuss two selected episodes in more detail, which are dominated by aged sea salt particles and relatively high organic particle mass concentrations, respectively.

#### 3.1 Overview of the field observation

Table 1 and Fig. S2 give an overview of meteorological conditions, trace gases, and particle concentrations for the entire measurement campaign. In summer 2016, the major wind direction was southwest ( $174 \pm 75$ )° and an average wind speed was  $(1.3 \pm 0.9)$  m s<sup>-1</sup> at the measurement site in the upper Rhine valley. This wind speed allows trace gases and particles to be transported over a distance of 10 km within 2 hours e.g. from the city of Karlsruhe to the measurement site. During the measurement time the sunrise and sunset were at around 06:30 and 20:30, respectively. Temperature and relative humidity (RH) were showing relatively stable anti-correlated diurnal patterns. Given the frequent clear nights, night-time temperature inversions occurred for 32 out of 37 nights during the entire measurement period (Fig. S3). Among the trace gases, CO<sub>2</sub> and O<sub>3</sub> were showing clear diurnal patterns with ozone peaking in the afternoon. NO<sub>2</sub> concentrations were higher in the morning hours and in the late evenings and generally behaved anti-correlated to O<sub>3</sub>. NO<sub>2</sub> and particle number were showing spikes at similar times (more frequently during the day-time), mainly due to the local emissions from nearby road traffic. The averaged particle number concentration of  $(1.2 \pm 0.8) \times 10^4$  cm<sup>-3</sup> for particles larger than 2.5 nm measured in this study, is comparable with that observed at the same site in previous tram-based measurements (Hagemann et al., 2014). SO<sub>2</sub> concentrations showed peaks with low frequency (up to >10 ppb) but only when the wind was coming from the direction of the industrial area in the southwest. As shown in Fig. S2, these SO<sub>2</sub> peaks correspond to higher particle number concentrations at around noontime, when the wind speed is higher ( $\sim 1.7$  m s<sup>-1</sup>) and vertical mixing intensified. This is in agreement with downward mixing of stack emissions from industrial sources, which can be validated by model calculations using COSMO-ART (refer to section 3.2).





190 In the entire measurement period, the total organic compounds, sulfate, nitrate, and ammonium were analysed to contribute on average ( $58 \pm 12\%$ ), ( $22 \pm 7\%$ ), ( $10 \pm 1\%$ ), and ( $9 \pm 3\%$ ) to the total non-refractory  $PM_{2.5}$  mass measured by AMS, respectively. In a previous publication (Shen et al., 2019), we could identify seven major particle classes at the measurement site among  $\sim 3.7 \times 10^5$  single particles with sizes between 200 nm and  $2.5 \mu\text{m } d_{va}$ : Calcium-soil (class 1, C1); Aged soot (C2); Sodium salts (C3); Secondary inorganic–amine (C4); Biomass burning–soil (C5); Biomass burning–organosulfate (C6); and Mixed/aged dust (C7). These particle types were not always present in similar fractions but showed distinct periods with one or two types dominating the particle mass (Shen et al., 2019). Here, we will mainly discuss two particle classes that dominated the total particle mass, i.e., sodium salts and biomass burning–soil particles, contributing on average ( $36 \pm 27\%$ ) and ( $30 \pm 18\%$ ) to the  $PM_{2.5}$  mass measured by the LAAPTOF, respectively. We will also discuss one characteristic particle class, i.e., secondary inorganic–amine particles, which showed relatively intensive and distinct peaks three times during this campaign (Fig. S2) as well as several weaker peaks.

200 As shown in Fig. 2 (a), total particle mass concentrations have clear diurnal pattern, showing higher night-time values and a peak during the morning rush hour ( $\sim 9:00$  local time). Higher particle mass values mainly correspond to southern wind directions, with one exception of extremely high masses associated with northwesterly winds (Fig. 2c). Specifically, sodium salt containing particles have higher fractions during day-time and correspond to southwesterly wind directions (Fig. 2b). Such wind was observed mainly during day-time and with higher wind speeds than at night. Biomass burning–soil particles enrich during night-time and mainly correspond to southeast wind (Fig. 2d). Secondary inorganic–amine particles typically rise in the morning and correspond to south and southwest wind directions (Fig. 2). When we compare weekdays (Monday to Friday) and weekends (Saturday and Sunday) in Fig. S4, the sodium salt particles were always correlated with southwest wind directions. Obvious differences can be found for secondary inorganic–amine and biomass burning–soil, respectively. For secondary inorganic–amine type particles, they were mainly advected by south and southwest wind during weekdays. However, they were much less abundant during weekends and did not show a clear relationship with wind directions (only a small peak for northwest). For biomass burning–soil particles the observations are a bit more complex.

210 The biomass burning–soil particles measured during weekdays were correlated with south and southwest wind as well, indicating potential sources from the city of Karlsruhe and its industrial area. Please note, that the representative spectra of these particles show aromatic ion peaks which may arise from aromatic compounds such as toluene and xylene due to motor vehicle emissions. Besides, this type of particles contains also some metals such as barium and lead that are related to coal combustion from industry. These features correspond to urban and industrial sources. During the weekends, biomass burning–soil particles had higher mass concentrations than during weekdays and were correlated with wind directions from east, southeast and northwest, where are mainly forests and agricultural areas. Therefore, we can infer that there were various sources for biomass burning–soil particles and these sources differed between weekdays and weekends. This could be e.g., barbeque fires during weekends while the weekdays' source is likely to be associated with that for secondary inorganic–amine particles.



220 In order to refine this type of particles, we compared them with black carbon particles measured by an Aethalometer (AE33-  
7). The equivalent black carbon concentration was calculated by the mass absorption cross sections value of  $11.7 \text{ m}^2 \text{ g}^{-1}$  (Herich  
et al., 2011). They seem to show a similar trend in the time series with a triplet of peaks from August 25<sup>th</sup> to 28<sup>th</sup> (refer to Fig.  
S2 and Fig. S5). We found that biomass burning-soil and black carbon concentrations are well correlated only for the weekends,  
with Pearson's correlation coefficients  $\gamma \geq 0.6$  (Fig. S5 b), and poorly correlated for weekdays ( $\gamma < 0.3$ ). This points to different  
225 major sources for weekdays and weekends. In addition, the black carbon instrument measuring the light absorption at different  
wavelengths ( $\lambda$ ) can be used to determine the absorption Ångström exponent ( $\alpha$ ) (Drinovec et al., 2015). For the entire campaign,  
the average  $\alpha$  was  $0.96 \pm 0.24$ , calculated by using a pair of wavelengths and their corresponding mass absorption cross sections  
(MAC=11.7 and  $23.6 \text{ m}^2 \text{ g}^{-1}$  for  $\lambda=950$  and  $470 \text{ nm}$ , respectively, Herich et al., 2011). Herich et al. (2011) and Zotter et al. (2017)  
have assigned an  $\alpha = \sim 0.9$  to BC from road traffic emissions, while the  $\alpha$  values for wood burning BC were typically higher  
230  $\sim 1.68$  (Zotter et al., 2017). Our measurements indicate that traffic was the dominating source of BC. Considering the good  
correlation between BC and biomass burning-soil particles during the weekend, we could infer the corresponding traffic  
emissions that contributed to biomass burning-soil particles. All observations support the conclusion that this particle class has  
significant anthropogenic sources, which may vary between weekends and weekdays. However, it still needs some more  
analysis to identify the different sources contributing to this particle class.

235 As mentioned before, the dominating particle classes varied for different time periods (Shen et al., 2019). In the following  
we focus on an episode with particles rich in sodium salts, to find out if sea salt particles can significantly influence the aerosol  
particle composition at this central European location about 800 km inland. Furthermore, we discuss another episode with  
relatively high organic mass concentrations to find out how these mass loads could form and exist for a relatively long time  
period. Both episodes are easily recognized from the time series of particles measured by LAAPTOF and organic compounds  
240 measured by AMS (Fig. S2 and Fig. 3; see also Fig. 4 in Shen et al, 2019). Apart from the difference of dominating particle  
classes and particulate organics, the particle acidity, organonitrate fraction and organic carbon oxidation state, as well as  
meteorological conditions were quite different between these two episodes. For particle acidity, we used an acidity indicator,  
namely the ratio of  $\text{NH}_4^+$  measured/predicted from AMS measurement (Zhang et al., 2007). The predicted  $\text{NH}_4^+$  value is the  
mass concentration required to fully neutralize sulfate, nitrate, and chloride measured by AMS. The averaged value of  $\text{NH}_4^+$   
245 measured/predicted for the entire campaign is  $0.82 \pm 0.10$ . As shown in Fig. 3, this ratio reaches lowest value during sodium  
salts rich episode, corresponding to the highest acidity, while the value in the organic rich episode is at the averaged level. While  
the apparent high acidity for the sea salt may partially be due to sea salt sulfate included in the AMS measurement, the correlation  
between particle acidity and organics concentrations may be related to acid-catalysed organic particle formation. For the fraction  
of organonitrates, the AMS derived ratio of  $\text{NO}^+/\text{NO}_2^+$  is commonly used as an indicator (Farmer et al., 2010; Kiendler-Scharr  
250 et al., 2016). In the episode rich in sodium salts, particles contain the highest organonitrate fraction while the particles in the  
organic rich episode contain the lowest fraction (refer to Fig. S6). Both, higher acidity and higher fraction of organonitrates are  
indicators for a stronger influence of anthropogenic emissions on the sodium rich aerosol. The averaged carbon oxidation state





( $\overline{OS}_C$ ) of the organic compounds is a robust metric of its degree of oxidation or processing (Canagaratna et al., 2015; Kroll et al., 2011). It turns out that the organics were more oxidized in the organic rich episode (refer to Fig. S6). In addition, the major wind with higher speeds was from southwest in the episode rich in sodium salt particles, and we will show in section 3.3 that this is related to long-range transport of sea salt particles. For the organic rich episode, the wind speeds were lower for most of the time. The different features of these two episodes are summarized in Table 2. We will discuss the episodes in more detail in sections 3.3 and 3.4, but will first demonstrate the usefulness of the transport model to identify an industrial point source.

### 3.2 Identification of an industrial point source

As shown in Fig. 4, we have modelled  $O_3$ ,  $NO_2$ ,  $SO_2$ , and particle number for the organic rich episode with COSMO-ART. Model simulations show comparable results and/or trends with field observations. For  $SO_2$ , there is a modelled peak at 10:00 on August 26<sup>th</sup>. As mentioned above, in the upwind direction of our measurement site is a coal fired power plant, where  $SO_2$  is emitted from a stack at 230 m a.g.l.. Therefore, we compared  $SO_2$  concentrations in an area including the power plant and our measurement site (refer to Fig. 5 a1 and a2), as well as time series of their vertical profile at the measurement site (Fig. 5b). The model results show that  $SO_2$  emissions were starting at 02:00 from the stack of the power plant, continuously transported by southwest wind (from 06:00, not shown here). Around noon, the  $SO_2$  was transported downward due to enhanced vertical mixing. Similar to  $SO_2$ , we found corresponding vertical transport of  $NO_x$  and primary particles (refer to Fig. 5 c and d), as well as secondary inorganic particles which will be discussed in section 3.3.3. In addition to the power plant emissions, particle emissions from the Karlsruhe refinery (e.g., around 8:00 at 122 m a.g.l.) and downtown Karlsruhe (at 10 m a.g.l.) were observed in the model results, which are visualized via videos of the evolution of particle number at these two altitudes (refer to Data availability). Given these findings, we have verified industrial contributions to the particles measured at this rural site. However, the particle numbers calculated by the model are 10 times lower than the measured values. In this context, it should be noted that, the observations relate to particles larger than 2.5 nm and 7 nm, respectively. The modelled particles include Aitken and accumulation mode particles with median diameter at around 30–80 nm and 300–400 nm, respectively. The model does not represent the ultrafine particles to their full extend, especially e.g., the contribution of organic compounds or ammonia to nucleation.

### 3.3 Aged sea salt in central Europe

Sodium salt particles accounted on average for  $(35.6 \pm 26.7)\%$  of the particle mass measured by LAAPTOF during the whole campaign, and they dominated several time periods (refer to Fig. S2). However, not all the peak events were the same. We chose a unique episode when sodium chloride signatures, e.g.,  $m/z$  81  $Na_2Cl^+$ , were showing the strongest signal, as plotted in Fig. 6 (a). In this episode, the sodium salts accounted for  $(70 \pm 24)\%$  of total mass of the single particles. To study the origin of these particles, we used the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT; Stein et al., 2015) on the READY website (<http://www.ready.noaa.gov>; Rolph et al., 2017) for back-trajectory analysis. The corresponding result suggests that



285 these particles were coming from the Atlantic Ocean, which is more than 800 km west of our measurement site (refer to Fig. S7 a). Therefore, we can constrain the type of sodium salts particles as sea salts with an age of about 1.5 to 3 days.

In this episode, sea salt particles retained strong chloride signatures in spite of the long-range transport, due to relatively high wind speeds (e.g., the average wind speed was  $\sim 6 \text{ m s}^{-1}$  from the ocean to our measurement site at noontime on August 21<sup>st</sup> and 22<sup>nd</sup>, based on 72 h back-trajectory analysis). Therefore, we interpret them as relatively fresh sea salt, which is the 1<sup>st</sup> sub-class of sodium salts particles identified with the fuzzy clustering analysis of the single particle mass spectra. As shown in 290 Fig. 6 (a), there is an obvious decrease of  $m/z$  81  $\text{Na}_2\text{Cl}^+$  at the end of this episode, while the sodium nitrate signature at  $m/z$  108  $\text{Na}_2\text{NO}_3^+$  increases. Such trends can also be found in similar but much less prominent sodium chloride rich episodes (see the red guiding arrows in Fig. 6a). This indicates the aging process of sea salt particles, namely heterogeneous replacement of chloride by nitrate on sea salt particles (Gard et al., 1998). We assigned the 2<sup>nd</sup> sub-class of sodium salts to more aged sea salt particles, which are bigger than the relatively young sea salt (Fig. 6b). 72 h back-trajectory analysis suggests that these particles 295 were coming from the Atlantic Ocean and were carried with southwesterly wind of relatively lower speed ( $\sim 3.5 \text{ m s}^{-1}$ ) to reach our measurement site on August 23<sup>rd</sup>, thus they spent more time ( $\sim 3$  days) over the continent and are more aged. The representative spectra of sodium salt particles and the averaged spectra of their sub-classes can be found in Fig. S8. The signal difference between aged and young sea salts spectra (aged - young) is shown in Fig. 6 (c), highlighting their signatures. Similar relationships as for  $m/z$  81  $\text{Na}_2\text{Cl}^+$  and 108  $\text{Na}_2\text{NO}_3^+$  can also be found for the other signature peaks (refer to Fig. S9). It is worth 300 noting that there are also nitrate related peaks in young sea salt, such as  $m/z$  104  $\text{NaClNO}_2^-$ , verifying that these young sea salts were not fresh. In addition,  $m/z$  92  $\text{Na}_2\text{NO}_2^+$  and 131  $\text{NaNO}_2\text{NO}_3^-$  increase earlier than 108  $\text{Na}_2\text{NO}_3^+$  and 147  $\text{Na}(\text{NO}_3)_2^-$ . Such that, we can infer the aging degree from a series of signature peaks in LAAPTOF spectra:  $81 \text{ Na}_2\text{Cl}^+ \rightarrow 104 \text{ NaClNO}_2^- \rightarrow 92 \text{ Na}_2\text{NO}_2^+$  and  $131 \text{ NaNO}_2\text{NO}_3^- \rightarrow 108 \text{ Na}_2\text{NO}_3^+$  and  $147 \text{ Na}(\text{NO}_3)_2^-$  (from relatively fresh to more aged, corresponding to chloride to nitrite to nitrate salts). However, this requires further systematic studies to be validated.

305 For sodium salt particles, a weak but characteristic peak at  $m/z$  129  $\text{C}_5\text{H}_7\text{NO}_3^+$  was assigned to organonitrate in LAAPTOF spectra in our previous study (Shen et al., 2019). As shown in Fig. S9,  $m/z$  129  $\text{C}_5\text{H}_7\text{NO}_3^+$  has a similar trend as  $m/z$  131  $\text{NaNO}_2\text{NO}_3^-$  and it is more pronounced in the LAAPTOF spectra of aged sea salt. In the sodium chloride rich episode,  $m/z$  129  $\text{C}_5\text{H}_7\text{NO}_3^+$  has a similar trend as organonitrate mass concentrations estimated by AMS results. The method for organonitrate estimation can be referred to Farmer et al. (2010) and the parameters used here are the same as those used by Huang et al. (2019). 310 We have briefly described this method in the supplement after the caption for Fig. S9. AMS derived organonitrate mass concentrations are comparable with the results of molecular analysis using FIGAERO-HR-ToF-CIMS (stands for a filter inlet for gases and aerosols coupled to a high-resolution time-of-flight chemical ionization mass spectrometer) in a companion study by (Huang et al., 2019). Therefore, we can confirm that the  $m/z$  129  $\text{C}_5\text{H}_7\text{NO}_3^+$  signal is due to organonitrates, rather than the recombination of nitrate and organic fragments due to the laser desorption and ionization (LDI) matrix effect in single-particle 315 MS (Murphy, 2007; Reilly et al., 2000; Reinard and Johnston, 2008; Zenobi and Knochenmuss, 1998). In addition, our simulation chamber studies revealed that  $m/z$  129  $\text{C}_5\text{H}_7\text{NO}_3^+$  in LAAPTOF spectra could be a signature for anthropogenic



320 sources (refer to Fig. S10 and the corresponding detailed description in the supporting information). Therefore,  $m/z$  129  $C_5H_7NO_3^+$  on sea salt particles indicates interactions between the natural sea salt and anthropogenic emissions. It is worth noting that another weak but characteristic peak was observed at  $m/z$  165  $Na_3SO_4^+$ . This signature peak for sea salt sulfate can be found with stronger signal in young sea salt than in aged sea salt (Gard et al., 1998).

### 3.4 Origin of the aerosol particles during the organics rich episode

At the end of the measurement period, we observed three subsequent organic peaks (see Fig. 3, Fig. S2, or Fig. S5): one peak during Friday August 26<sup>th</sup> day-time and two peaks during following two nights. Organics measured in this episode exhibited the highest mass concentrations of  $(12.9 \pm 3.9) \mu\text{g m}^{-3}$ , which accounted for  $(77 \pm 6)\%$  of the total non-refractory particle mass and is almost 3 times more than organic masses measured in other periods of the entire AMS measurement period. Please note, some secondary inorganic components were also measured in this episode, showing some peaks at similar times as the high organic peaks. As for the sodium chloride rich episode, we have also done a HYSPLIT back-trajectory analysis for the organic rich episode. Fig. S7 (b) shows these results: the air mass was coming from the continent during this episode. In order to better interpret the measurements and to understand the impact of transport processes for the particles measured in the organics rich episode, we have done COSMO-ART simulations covering the time period from August 25<sup>th</sup> to 28<sup>th</sup>, 2016. Fig. 7 shows the particulate organic compounds, the secondary inorganics (ammonium, nitrate, and sulfate) measured by AMS, and the result calculated using COSMO-ART (panels d to g). The corresponding meteorological conditions measured are shown in panels (a) to (c) of Figure 7. In general, the trends and also the amplitudes of the model results compare quite well with the observations. In the following, we will combine field observation and transport model simulation results to address these questions: 1) were the organic rich aerosols due to local formation, regional transport, or vertical mixing; 2) were all three organic peaks caused by the same reasons; 3) were the secondary inorganics originating from the same sources as the organics. We will start the discussion with the inorganics, will then go to the organics, and will end up with a typical particle type in this episode.

#### 3.4.1 Secondary inorganic compounds

As shown in Fig. 7 (e) to (g), observations and model results for secondary inorganic compounds show comparable diurnal cycles and even the concentration levels match quite well. While ammonium and nitrate have a similar trend for most of the time, the trend for sulfate seems to be somewhat different. The major peaks of ammonium and nitrate appear at similar times as the high organic peaks. As will be shown in the following section, these compounds originated from anthropogenic sources in the urban and industrial area of Karlsruhe and from the city of Stuttgart. There is only one instant when all the measured secondary inorganics are showing peaks together: the inorganics are peaking at ~10:00 on August 26<sup>th</sup>, when also high  $SO_2$  concentrations were observed (c.f. Fig. 4) which were due to the emissions from a power plant located in upwind direction. To verify this, we can refer to the time series of vertical profiles for the simulated secondary inorganics (Fig. S11 b to d). Despite similar spatial distributions of nitrate, ammonium,  $NO_2$ , and  $SO_2$  from 02:00 to 12:00 (c.f. Fig. 5) the measured sulfate peak for



that time is missing in the model output. The reason for this is still unclear. It should be noted that, compared to the sulfate peak observed, the companion ammonium and nitrate peaks are much broader, attributed to additional anthropogenic sources from the city and refinery rather than only the power plant.

Some of the differences between model result and observation, such as that on August 25<sup>th</sup> or e.g., for sharper peaks, may be explained by missing (point) sources in the emission data and/or the limited spatial resolution. The general agreement is however quite well and also for some point sources like the largest power plant in Karlsruhe. Some of the differences in the trends of nitrate and sulfate may be caused by the different nature of the sources of their precursor gases SO<sub>2</sub> and NO<sub>x</sub>. While SO<sub>2</sub> is emitted in this area mainly from a few point sources, NO<sub>x</sub> is emitted from various sources over a larger area.

### 3.4.2 Organic compounds

For the organic compounds, the general temporal development is represented well by the model, showing similar trends as measured by AMS, but underestimates the mass concentrations by a factor of ~2.3; averaged model result is  $(5.5 \pm 3.3) \mu\text{g m}^{-3}$ , while observed were  $(12.9 \pm 3.9) \mu\text{g m}^{-3}$ . Potential reasons for this discrepancy will be discussed in the light of sensitivity tests at the end of this section.

As mentioned above, the organic particle fraction shows a peak on Friday August 26<sup>th</sup> during day-time (peak 1) and two more during the following nights of the weekend (peak 2 and 3) in the organics rich episode. Model result cover peak 1 and 2, as well as of the first part of peak 3. Analysis of the model results shows that the three peaks were caused by different reasons including meteorological conditions, regional transport, and local or in-situ new particle formation and growths (such as night-time new particle formation; Huang et al., 2019). As shown in Fig. 7a and d, peak 1 corresponds to south and southwest wind, indicating that the air masses came from the urban and industrial area of Karlsruhe. This is illustrated very well in a video showing the transmission of particulate organics leading to peak 1 (refer to Data availability). The most intensive peak in organic particle mass (peak 2), starts from August 26<sup>th</sup> 20:30 (Fig. 3 and Fig. S2) when the wind direction shifts from northwest to east (northeast, straight east, and southeast), the wind speed decreases to  $< 0.5 \text{ m s}^{-1}$ , and the night-time temperature inversion starts. Such stagnant conditions limit aerosol mixing and support build up aerosols and their growth. During the entire campaign, the strongest temperature inversion was observed during this night from August 26<sup>th</sup> to 27<sup>th</sup> (peak inversion at 02:40 on August 27; Fig. S3). In spite of the stagnant conditions and rather low wind speeds, regional transport still plays an important role. The model results for this time show transport of organic aerosol particles from the city of Stuttgart (southeast to the measurement site) over forested areas into the Rhine valley. This is clearly visible in a corresponding video of the model results for particulate organics and wind directions (refer to Data availability). Furthermore, we investigate the relative contribution of biogenic and anthropogenic emissions and their interactions for this episode by comparing the model results with and without biogenic emissions. It turns out that the corresponding air masses originating from anthropogenic emissions in the Stuttgart area were transported to the measurement site by south-easterly winds via forested hills. During this transport, the organic particle mass concentrations increased while being mixed with biogenic emissions from the forest. The resulting air mass finally reaching the



380 measurement site consisted therefore of anthropogenic and biogenic organics. For peak 3, we can also find impact of transport  
as illustrated in another video of particulate organic mass evolution (refer to data availability). However, the origin of this peak  
is more complex and is most likely more influenced by local particle formation and growths. After the night-time peak 2, from  
12:00 on August 27 the air mass were doing back and forth movements in west-east directions, showing a strong concentration  
gradient (west-east). Such a conditions lead to a high uncertainty of the model results for one point. Hence, although we observed  
385 quite similar particle compositions in these three subsequent peaks with high organic mass concentrations for similar  
meteorological conditions with low wind speeds, all three peaks were caused by air masses of different origin.

As mentioned above, the model underestimates the organic aerosol mass on average by a factor of 2.3. In order to better  
understand this discrepancy, we have done three sensitivity tests: with/without biogenic emissions (test 01), one with doubled  
biogenic emissions (test 02) and one with boundary values or regional background values (outside the model domain) of  
390 particulate organic compounds increased by a factor of 10 (test 03). The results of test 01 suggest that  $(80 \pm 4)\%$  of the organic  
mass calculated was resulting from biogenic sources for the peak 1 time period, corresponding to the air masses coming from  
Karlsruhe. Regarding the time period of peak 2, there was a sharp drop of the biogenic contribution from the 88% to 68%  
corresponding to August 26<sup>th</sup> 23:00 to August 27<sup>th</sup> 2:00 when the air mass was coming from direction of Stuttgart (see Fig. 8).  
Hence, we can infer that the anthropogenic contribution to the air mass that originated from Stuttgart was somewhat larger than  
395 that originated from Karlsruhe. Over the entire organic rich episode, on average  $(74 \pm 7)\%$  of the organics mass was contributed  
by biogenic sources. To test if missing emissions of biogenic precursors can explain the underestimated organic particulate  
masses we doubled this major (biogenic) source for organics (test 02). As shown in Fig. 9, doubling biogenic emissions makes  
some of the peaks of organic mass fit the observations better but the concentrations before and in between the peak values are  
still underestimated. Another potential reason for the discrepancy could be to low boundary values (regional background);  
400 therefore, we increased them by a factor of 10 (test 03). This elevates the overall simulated concentration levels, leading to an  
overestimation especially of the first peak. These tests demonstrate the potential role of precursor emissions and regional  
transport of organic aerosol particles on the model results. There seems to be evidence that the model underestimates the regional  
background of organic aerosol particles. However, we cannot rule out that also underestimated sources of VOC or weaknesses  
in the SOA formation mechanism may contribute to the underestimation of the organic aerosol particle mass. For example,  
405 night-time new particle formations mechanisms are not included in the current model version.

#### 4 Conclusions and atmospheric implication

High particle number concentrations have been observed at a rural site in the upper Rhine valley by measurements on a tramline  
(Hagemann et al., 2014). In order to study the nature of the aerosols in the same region, we conducted a 6-week stationary field  
measurement campaign at a rural site next to the tramline in summer 2016, and applied a regional transport model COSMO-



410 ART to analyse the origin of the aerosols. Comparing the field observations and model calculations, we have verified that industrial emissions contribute to the high particle number concentrations at this rural site.

For the whole measurement period, the particles that dominate the mass are sodium salts, which contributed  $(36 \pm 27)$  % to the  $PM_{2.5}$  mass measured by the LAAPTOF. According to the AMS measurements, organics, sulfate, nitrate, and ammonium contributed  $(58 \pm 12)\%$ ,  $(22 \pm 7)\%$ ,  $(10 \pm 1)\%$ , and  $(9 \pm 3)\%$  to the total non-refractory  $PM_{2.5}$  mass, respectively. Furthermore, 415 we have investigated two episodes, which were dominated by aged sea salt and organics, which show characteristic features such as higher particle acidities and carbon oxidation states, respectively. Both episodes selected, were dominated by particles formed from the interaction of natural/biogenic and anthropogenic emissions.

For the sodium chloride rich episode, sodium salt particles accounted for  $(70 \pm 24)\%$  of total mass of the single particles. We could show that the dominating particles were originally sea salt, from the Atlantic Ocean (~800 km away from the 420 measurement site). These particles consisted of two classes with different aging stages. After 1.5–3 days of transport, the original sea salt particles retained strong chloride signatures while after 3.5–4 days of transport over the continent aged sea salt with stronger nitrate signatures and larger size arrived at the measurement location. In addition, the organonitrate signature peaks at  $m/z$  129  $C_5H_7NO_3^+$  were stronger for the aged sea salt particles. According to chamber experiments on OH oxidation of  $\alpha$ -pinene and/ or toluene in the presence of  $NO_x$  on sodium nitrate seed particles, we have evidence that  $m/z$  129  $C_5H_7NO_3^+$  is a 425 signature peak for anthropogenic organics in single particle mass spectra. Hence, the  $m/z$  129  $C_5H_7NO_3^+$  observed on sea salt particles in this study is another indicator for the interaction of anthropogenic emissions with the sea salt particles. Besides, the observed signature peak at  $m/z$  165  $Na_3SO_4^+$  for sea salt sulfate may indirectly explain apparently higher particle acidity in this episode. Some other signature peaks such as 81  $Na_2Cl^+$ , 92  $Na_2NO_2^+$ , 108  $Na_2NO_3^+$ , 104  $NaClNO_2^-$ , 131  $NaNO_2NO_3^-$ , and 147  $Na(NO_3)_2^-$  have the potential to indicate degree of aging of sea salt particles.

430 For the organics rich episode, the particulate organic compounds accounted for  $(77 \pm 6)\%$  of total non-refractory particle mass. We identified potential sources of the elevated organic particle concentrations and of secondary inorganic compounds. Transport model results suggest that  $(74 \pm 7)\%$  of the particulate organic mass calculated was related to biogenic emissions. With the combination of mass spectrometry and COSMO-ART modelling, we can confirm that during Friday, August 26<sup>th</sup>, air masses came from an industrial area and the city of Karlsruhe, leading to high particle numbers (including all the particles with 435 size  $>2.5$  nm), high particulate organics and secondary inorganics, as well as high fractions of aged-biomass-burning and soil-dust like particles. During night-time on August 26<sup>th</sup>, south-easterly winds transported air masses from Stuttgart via forested areas to the measurement site, resulting in particles formed by the interaction of anthropogenic and biogenic precursors. Such air masses flowed into the Rhine valley and contributed also to the high organic aerosol mass concentrations during the night from August 27<sup>th</sup> to 28<sup>th</sup>. In addition, low wind speeds and temperature inversion limited mixing and supported new particle 440 formation and growth, partially contributing to higher organic particle masses.

Comparing comprehensive field observations with the COSMO-ART model results showed quite good agreement for trends and most concentration levels for several aerosol particle compounds and trace gases even if not all details could be reproduced





445 by the model. However, the simulated particle numbers are 10 times less than the observed values. This is mainly due to the fact that not all new particle formation processes (nucleation mode) are included in the model (e.g. night time chemistry as well as the role of organics or ammonia). Furthermore, the model underestimates particulate organics by a factor of 2.3. This discrepancy may be partly explained by an underestimated regional background, missing sources, or incomplete SOA formation and growths mechanisms, in the current model version.

450 As discussed above, we can conclude that factors related to topography, metrological conditions, local emissions, in-situ formation and growths, regional transport, as well as the interaction of biogenic and anthropogenic compounds need to be considered for a comprehensive understanding of aerosol composition and processing in this region. This study provides a good insight into the chemical nature and complex origin of aerosols in the upper Rhine valley, achieved by combining complimentary mass spectrometry and transport modelling.

#### **Data availability**

Five supporting videos (.gif files) are available at KIT open data (<https://doi.org/10.5445/IR/1000094401>).

#### **455 Author contributions**

460 X.S. operated LAAPTOF and AMS during the whole field campaign, did the LAAPTOF data analysis, did COSMO-ART model runs and model output data analysis, did HYSPLIT back-trajectory analysis, produced all figures, and wrote the manuscript. H.S. organized the campaign, provided trace gas, meteorology, and particle measurements as well as suggestions for the data analysis, interpretation, and discussion. H.V. and B.V. provided the input data for COSMO-ART model, provided suggestions for the model simulation, the model data analysis, interpretation, and discussion. W.H. operated AMS during the whole campaign and did AMS data analysis. C.M. helped to operate the instruments, did black carbon data analysis, and provided suggestions for the data analysis, interpretation, and discussion. R.R. helped to operate LAAPTOF. A.S.H.P. gave comments for interpretation of black carbon data. T.L. gave general advices and comments for this paper. All authors contributed to the final text.

#### **465 Competing interests**

The authors declare no conflict of interest.



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## References

- 475 Ahern, A. T., Subramanian, R., Saliba, G., Lipsky, E. M., Donahue, N. M., and Sullivan, R. C.: Effect of secondary organic aerosol coating thickness on the real-time detection and characterization of biomass-burning soot by two particle mass spectrometers, *Atmos Meas Tech*, 9, 6117–6137, 2016.
- Backes, A., Aulinger, A., Bieser, J., Matthias, V., and Quante, M.: Ammonia emissions in Europe, part I: Development of a dynamical ammonia emission inventory, *Atmos Environ*, 131, 55–66, 2016.
- 480 Bahreini, R., Keywood, M. D., Ng, N. L., Varutbangkul, V., Gao, S., Flagan, R. C., Seinfeld, J. H., Worsnop, D. R., and Jimenez, J. L.: Measurements of secondary organic aerosol from oxidation of cycloalkenes, terpenes, and m-xylene using an Aerodyne aerosol mass spectrometer, *Environ Sci Technol*, 39, 5674–5688, 2005.
- Baklanov, A., Schlunzen, K., Suppan, P., Baldasano, J., Brunner, D., Aksoyoglu, S., Carmichael, G., Douros, J., Flemming, J., Forkel, R., Galmarini, S., Gauss, M., Grell, G., Hirtl, M., Joffre, S., Jorba, O., Kaas, E., Kaasik, M., Kallos, G., Kong, X., Korsholm, U., Kurganskiy, A., Kushta, J., Lohmann, U., Mahura, A., Manders-Groot, A., Maurizi, A., Moussiopoulos, N., Rao, S. T., Savage, N., Seigneur, C., Sokhi, R. S., Solazzo, E., Solomos, S., Sorensen, B., Tsegas, G., Vignati, E., Vogel, B., and Zhang, Y.: Online coupled regional meteorology chemistry models in Europe: current status and prospects, *Atmos Chem Phys*, 14, 317–398, 2014.
- 485
- Bangert, M., Kottmeier, C., Vogel, B., and Vogel, H.: Regional scale effects of the aerosol cloud interaction simulated with an online coupled comprehensive chemistry model, *Atmos Chem Phys*, 11, 4411–4423, 2011.
- 490
- Bieser, J., Aulinger, A., Matthias, V., Quante, M., and Builtjes, P.: SMOKE for Europe - adaptation, modification and evaluation of a comprehensive emission model for Europe, *Geosci Model Dev*, 4, 47–68, 2011.
- Burkholder, J. B., Abbate, J. P. D., Barnes, I., Roberts, J. M., Melamed, M. L., Ammann, M., Bertram, A. K., Cappa, C. D., Carlton, A. G., Carpenter, L. J., Crowley, J. N., Dubowski, Y., Georges, C., Heard, D. E., Herrmann, H., Keutsch, F. N., Kroll, J. H., McNeill, V. F., Ng, N. L., Nizkorodov, S. A., Orlando, J. J., Percival, C. J., Picquet-Varrault, B., Rudich, Y., Seakins, P. W., Surratt, J. D., Tanimoto, H., Thornton, J. A., Tong, Z., Tyndall, G. S., Wahner, A., Weschler,
- 495



C. J., Wilson, K. R., and Ziemann, P. J.: The Essential Role for Laboratory Studies in Atmospheric Chemistry, *Environ Sci Technol*, 51, 2519–2528, 2017.

500 Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q., Onasch, T. B., Drewnick, F., Coe, H., Middlebrook, A., Delia, A., Williams, L. R., Trimborn, A. M., Northway, M. J., DeCarlo, P. F., Kolb, C. E., Davidovits, P., and Worsnop, D. R.: Chemical and microphysical characterization of ambient aerosols with the aerodyne aerosol mass spectrometer, *Mass Spectrom Rev*, 26, 185–222, 2007.

505 Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P., Hildebrandt Ruiz, L., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue, N. M., Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of organic compounds using aerosol mass spectrometry: characterization, improved calibration, and implications, *Atmos Chem Phys*, 15, 253–272, 2015.

Canonaco, F., Crippa, M., Slowik, J. G., Baltensperger, U., and Prevot, A. S. H.: SoFi, an IGOR-based interface for the efficient use of the generalized multilinear engine (ME-2) for the source apportionment: ME-2 application to aerosol mass spectrometer data, *Atmos Meas Tech*, 6, 3649–3661, 2013.

510 Canonaco, F., Slowik, J. G., Baltensperger, U., and Prevot, A. S. H.: Seasonal differences in oxygenated organic aerosol composition: implications for emissions sources and factor analysis, *Atmos Chem Phys*, 15, 6993–7002, 2015.

515 Crippa, M., Canonaco, F., Lanz, V. A., Aijala, M., Allan, J. D., Carbone, S., Capes, G., Ceburnis, D., Dall'Osto, M., Day, D. A., DeCarlo, P. F., Ehn, M., Eriksson, A., Freney, E., Hildebrandt Ruiz, L., Hillamo, R., Jimenez, J. L., Junninen, H., Kiendler-Scharr, A., Kortelainen, A. M., Kulmala, M., Laaksonen, A., Mensah, A., Mohr, C., Nemitz, E., O'Dowd, C., Ovadnevaite, J., Pandis, S. N., Petaja, T., Poulain, L., Saarikoski, S., Sellegri, K., Swietlicki, E., Tiitta, P., Worsnop, D. R., Baltensperger, U., and Prevot, A. S. H.: Organic aerosol components derived from 25 AMS data sets across Europe using a consistent ME-2 based source apportionment approach, *Atmos Chem Phys*, 14, 6159–6176, 2014.

520 DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C., Gonin, M., Fuhrer, K., Horvath, T., Docherty, K. S., Worsnop, D. R., and Jimenez, J. L.: Field-deployable, high-resolution, time-of-flight aerosol mass spectrometer, *Anal Chem*, 78, 8281–8289, 2006.

Drinovec, L., Mocnik, G., Zotter, P., Prevot, A. S. H., Ruckstuhl, C., Coz, E., Rupakheti, M., Sciare, J., Muller, T., Wiedensohler, A., and Hansen, A. D. A.: The "dual-spot" Aethalometer: an improved measurement of aerosol black carbon with real-time loading compensation, *Atmos Meas Tech*, 8, 1965–1979, 2015.

525 Farmer, D. K., Matsunaga, A., Docherty, K. S., Surratt, J. D., Seinfeld, J. H., Ziemann, P. J., and Jimenez, J. L.: Response of an aerosol mass spectrometer to organonitrates and organosulfates and implications for atmospheric chemistry, *P Natl Acad Sci USA*, 107, 6670–6675, 2010.



- Fierce, L., Riemer, N., and Bond, T. C.: Toward reduced representation of mixing state for simulating aerosol effects on climate, *B Am Meteorol Soc*, 98, 971–980, 2017.
- 530 Fuzzi, S., Baltensperger, U., Carslaw, K., Decesari, S., van Der Gon, H. D., Facchini, M. C., Fowler, D., Koren, I., Langford, B., Lohmann, U., Nemitz, E., Pandis, S., Riipinen, I., Rudich, Y., Schaap, M., Slowik, J. G., Spracklen, D. V., Vignati, E., Wild, M., Williams, M., and Gilardoni, S.: Particulate matter, air quality and climate: lessons learned and future needs, *Atmos Chem Phys*, 15, 8217–8299, 2015.
- 535 Gard, E. E., Kleeman, M. J., Gross, D. S., Hughes, L. S., Allen, J. O., Morrical, B. D., Fergenson, D. P., Dienes, T., Galli, M. E., Johnson, R. J., Cass, G. R., and Prather, K. A.: Direct observation of heterogeneous chemistry in the atmosphere, *Science*, 279, 1184–1187, 1998.
- Gemayel, R., Hellebust, S., Temime-Roussel, B., Hayeck, N., Van Elteren, J. T., Wortham, H., and Gligorovski, S.: The performance and the characterization of laser ablation aerosol particle time-of-flight mass spectrometry (LAAP-ToF-MS), *Atmos Meas Tech*, 9, 1947–1959, 2016.
- 540 Gemayel, R., Temime-Roussel, B., Hayeck, N., Gandolfo, A., Hellebust, S., Gligorovski, S., and Wortham, H.: Development of an analytical methodology for obtaining quantitative mass concentrations from LAAP-ToF-MS measurements, *Talanta*, 174, 715–724, 2017.
- Gross, G. and Wippermann, F.: Channeling and Countercurrent in the Upper Rhine Valley - Numerical Simulations, *J Clim Appl Meteorol*, 26, 1293–1304, 1987.
- 545 Gunsch, M. J., May, N. W., Wen, M., Bottenus, C. L. H., Gardner, D. J., VanReken, T. M., Bertman, S. B., Hopke, P. K., Ault, A. P., and Pratt, K. A.: Ubiquitous influence of wildfire emissions and secondary organic aerosol on summertime atmospheric aerosol in the forested Great Lakes region, *Atmos Chem Phys*, 18, 3701–3715, 2018.
- Hagemann, R., Corsmeier, U., Kottmeier, C., Rinke, R., Wieser, A., and Vogel, B.: Spatial variability of particle number concentrations and NO<sub>x</sub> in the Karlsruhe (Germany) area obtained with the mobile laboratory 'AERO-TRAM', *Atmos Environ*, 94, 341–352, 2014.
- 550 Healy, R. M., Sciare, J., Poulain, L., Crippa, M., Wiedensohler, A., Prevot, A. S. H., Baltensperger, U., Sarda-Estève, R., McGuire, M. L., Jeong, C. H., McGillicuddy, E., O'Connor, I. P., Sodeau, J. R., Evans, G. J., and Wenger, J. C.: Quantitative determination of carbonaceous particle mixing state in Paris using single-particle mass spectrometer and aerosol mass spectrometer measurements, *Atmos Chem Phys*, 13, 9479–9496, 2013.
- 555 Herich, H., Hueglin, C., and Buchmann, B.: A 2.5 year's source apportionment study of black carbon from wood burning and fossil fuel combustion at urban and rural sites in Switzerland, *Atmos Meas Tech*, 4, 1409–1420, 2011.



- Huang, W., Saathoff, H., Shen, X. L., Ramakrishna, R., Leisner, T., and Mohr, C.: Chemical characterization of highly functionalized organonitrates contributing to high night-time organic aerosol mass loadings and particle growth, *Environ Sci Technol*, 53, 1165–1174, 2019.
- 560 IPCC: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change. Cambridge Univ. Press.
- Jeong, C. H., McGuire, M. L., Godri, K. J., Slowik, J. G., Rehbein, P. J. G., and Evans, G. J.: Quantification of aerosol chemical composition using continuous single particle measurements, *Atmos Chem Phys*, 11, 7027–7044, 2011.
- Kalthoff, N. and Vogel, B.: Countercurrent and Channeling Effect under Stable Stratification in the Area of Karlsruhe, *Theor Appl Climatol*, 45, 113–126, 1992.
- 565 Kiendler-Scharr, A., Mensah, A. A., Friese, E., Topping, D., Nemitz, E., Prevot, A. S. H., Aijala, M., Allan, J., Canonaco, F., Canagaratna, M., Carbone, S., Crippa, M., Dall'Osto, M., Day, D. A., De Carlo, P., Di Marco, C. F., Elbern, H., Eriksson, A., Freney, E., Hao, L., Herrmann, H., Hildebrandt, L., Hillamo, R., Jimenez, J. L., Laaksonen, A., McFiggans, G., Mohr, C., O'Dowd, C., Otjes, R., Ovadnevaite, J., Pandis, S. N., Poulain, L., Schlag, P., Sellegri, K., Swietlicki, E., Tiitta, P., Vermeulen, A., Wahner, A., Worsnop, D., and Wu, H. C.: Ubiquity of organic nitrates from nighttime chemistry in the European submicron aerosol, *Geophys Res Lett*, 43, 7735–7744, 2016.
- 570 Knote, C., Brunner, D., Vogel, H., Allan, J., Asmi, A., Aijala, M., Carbone, S., van der Gon, H. D., Jimenez, J. L., Kiendler-Scharr, A., Mohr, C., Poulain, L., Prevot, A. S. H., Swietlicki, E., and Vogel, B.: Towards an online-coupled chemistry-climate model: evaluation of trace gases and aerosols in COSMO-ART, *Geosci Model Dev*, 4, 1077–1102, 2011.
- Kroll, J. H., Donahue, N. M., Jimenez, J. L., Kessler, S. H., Canagaratna, M. R., Wilson, K. R., Altieri, K. E., Mazzoleni, L. R., 575 Wozniak, A. S., Bluhm, H., Mysak, E. R., Smith, J. D., Kolb, C. E., and Worsnop, D. R.: Carbon oxidation state as a metric for describing the chemistry of atmospheric organic aerosol, *Nat Chem*, 3, 133–139, 2011.
- Marsden, N. A., Flynn, M. J., Allan, J. D., and Coe, H.: Online differentiation of mineral phase in aerosol particles by ion formation mechanism using a LAAP-TOF single-particle mass spectrometer, *Atmospheric Measurement Techniques Discussions*, 11, 195–213, 2018.
- 580 Marsden, N. A., Flynn, M. J., Taylor, J. W., Allan, J. D., and Coe, H.: Evaluating the influence of laser wavelength and detection stage geometry on optical detection efficiency in a single-particle mass spectrometer, *Atmos Meas Tech*, 9, 6051–6068, 2016.
- Marsden, N. A., Ullrich, R., Möhler, O., Hammer, S. E., Kandler, K., Cui, Z., Williams, P. I., Flynn, M. J., Liu, D., Allan, J. D., and Coe, H.: Mineralogy and mixing state of north African mineral dust by online single-particle mass spectrometry, 585 *Atmos Chem Phys*, 19, 2259–2281, 2019.



- Murphy, D. M.: The design of single particle laser mass spectrometers, *Mass Spectrom Rev*, 26, 150–165, 2007.
- Paatero, P.: Least squares formulation of robust non-negative factor analysis, *Chemometr Intell Lab*, 37, 23–35, 1997.
- Paatero, P. and Tapper, U.: Positive Matrix Factorization - a Nonnegative Factor Model with Optimal Utilization of Error-  
Estimates of Data Values, *Environmetrics*, 5, 111–126, 1994.
- 590 Pöschl, U.: Atmospheric aerosols: Composition, transformation, climate and health effects, *Angew Chem Int Edit*, 44, 7520–  
7540, 2005.
- Ramisetty, R., Abdelmonem, A., Shen, X. L., Saathoff, H., Leisner, T., and Mohr, C.: Exploring femtosecond laser ablation in  
single-particle aerosol mass spectrometry, *Atmos Meas Tech*, 11, 4345–4360, 2018.
- Reitz, P., Zorn, S. R., Trimborn, S. H., and Trimborn, A. M.: A new, powerful technique to analyze single particle aerosol mass  
595 spectra using a combination of OPTICS and the fuzzy c-means algorithm, *J Aerosol Sci*, 98, 1–14, 2016.
- Rolph, G., Stein, A., and Stunder, B.: Real-time Environmental Applications and Display sYstem: READY, *Environ Modell  
Softw*, 95, 210–228, 2017.
- Seinfeld, J. H. and Pandis, S. N.: Atmospheric chemistry and physics: From air pollution to climate change, Wiley, J & Sons,  
Inc., Hoboken, New Jersey, 2006.
- 600 Shen, X. L., Ramisetty, R., Mohr, C., Huang, W., Leisner, T., and Saathoff, H.: Laser ablation aerosol particle time-of-flight  
mass spectrometer (LAAPTOF): performance, reference spectra and classification of atmospheric samples, *Atmos  
Meas Tech*, 11, 2325–2343, 2018.
- Shen, X. L., Saathoff, H., Huang, W., Mohr, C., Ramisetty, R., and Leisner, T.: Understanding of atmospheric aerosol particles  
with improved particle identification and quantification by single particle mass spectrometry, *Atmos. Meas. Tech.*, 12,  
605 2219–2240, 2019.
- Stanelle, T., Vogel, B., Vogel, H., Baumer, D., and Kottmeier, C.: Feedback between dust particles and atmospheric processes  
over West Africa during dust episodes in March 2006 and June 2007, *Atmos Chem Phys*, 10, 10771–10788, 2010.
- Stein, A. F., Draxler, R. R., Rolph, G. D., Stunder, B. J. B., Cohen, M. D., and Ngan, F.: Noaa's Hysplit Atmospheric Transport  
and Dispersion Modeling System, *B Am Meteorol Soc*, 96, 2059–2077, 2015.
- 610 Tsigaridis, K., Krol, M., Dentener, F. J., Balkanski, Y., Lathière, J., Metzger, S., Hauglustaine, D. A., and Kanakidou, M.:  
Change in global aerosol composition since preindustrial times, *Atmos Chem Phys*, 6, 5143–5162, 2006.





- Vogel, B., Fiedler, F., and Vogel, H.: Influence of Topography and Biogenic Volatile Organic-Compounds Emission in the State of Baden-Württemberg on Ozone Concentrations during Episodes of High Air Temperatures, *J Geophys Res-Atmos*, 100, 22907–22928, 1995.
- 615 Vogel, B., Gross, B., and Wippermann, F.: Mesoklip (1st Special Observation Period) - Observations and Numerical-Simulation - a Comparison, *Bound-Lay Meteorol*, 35, 83–102, 1986.
- Vogel, B., Vogel, H., Baumer, D., Bangert, M., Lundgren, K., Rinke, R., and Stanelle, T.: The comprehensive model system COSMO-ART - Radiative impact of aerosol on the state of the atmosphere on the regional scale, *Atmos Chem Phys*, 9, 8661–8680, 2009.
- 620 Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Ulbrich, I. M., Ng, N. L., Worsnop, D. R., and Sun, Y. L.: Understanding atmospheric organic aerosols via factor analysis of aerosol mass spectrometry: a review, *Anal Bioanal Chem*, 401, 3045–3067, 2011.
- Zhang, Q., Jimenez, J. L., Worsnop, D. R., and Canagaratna, M.: A case study of urban particle acidity and its influence on secondary organic aerosol, *Environ Sci Technol*, 41, 3213–3219, 2007.
- 625 Zhou, Y., Huang, X. H. H., Griffith, S. M., Li, M., Li, L., Zhou, Z., Wu, C., Meng, J. W., Chan, C. K., Louie, P. K. K., and Yu, J. Z.: A field measurement based scaling approach for quantification of major ions, organic carbon, and elemental carbon using a single particle aerosol mass spectrometer, *Atmos Environ*, 143, 300–312, 2016.
- Zotter, P., Herich, H., Gysel, M., El-Haddad, I., Zhang, Y. L., Mocnik, G., Hüglin, C., Baltensperger, U., Szidat, S., and Prevot, A. H.: Evaluation of the absorption angstrom ngstrom exponents for traffic and wood burning in the Aethalometer-based source apportionment using radiocarbon measurements of ambient aerosol, *Atmos Chem Phys*, 17, 4229–4249, 630 2017.



**Table 1: Mean values ( $\pm$ sdev) of characteristic parameters for the entire measurement period.**

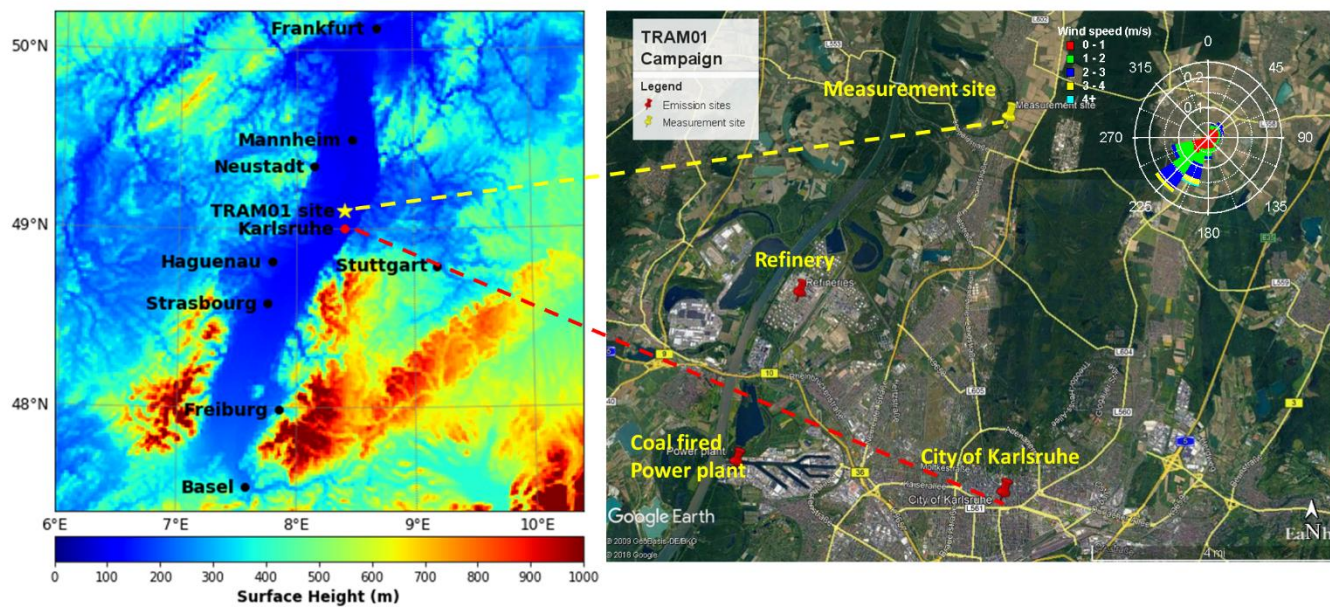
| Parameter   | Mean value $\pm$ sdev | Parameter  | Mean value $\pm$ sdev                             |   |
|---|-----------------------|--|---|---|
| Wind Direction ( $^{\circ}$ )                                     | 174.0 $\pm$ 75.0      | <b>Particulate compounds measured by AMS</b>               |   |   |
| Wind Speed ( $\text{m s}^{-1}$ )                                  | 1.3 $\pm$ 0.9         |  | Organics ( $\mu\text{g m}^{-3}$ )                 | 5.1 $\pm$ 3.3                             |
| Temperature (K)   | 294.4 $\pm$ 4.5       |  | Ammonium ( $\mu\text{g m}^{-3}$ )                 | 0.7 $\pm$ 0.4                             |
| Relative Humidity (%)   | 70.0 $\pm$ 17.0       |  | Nitrate ( $\mu\text{g m}^{-3}$ )                  | 0.8 $\pm$ 0.8                             |
| O <sub>3</sub> (ppb)  | 26.5 $\pm$ 15.8       |  | Sulfate ( $\mu\text{g m}^{-3}$ )                  | 1.7 $\pm$ 0.8                             |
| NO <sub>2</sub> (ppb)   | 11.4 $\pm$ 6.9        |  | <b>Seven particle classes measured by LAAPTOF</b> |   |
| SO <sub>2</sub> (ppb)   | 1.2 $\pm$ 2.7         |  |   | C1: Calcium-soil ( $\mu\text{g m}^{-3}$ ) |
| CO <sub>2</sub> (ppm)   | 446.0 $\pm$ 32.0      | C2: Aged soot ( $\mu\text{g m}^{-3}$ )                     |   | 0.1 $\pm$ 0.4                             |
| Particle Number ( $10^4 \text{ cm}^{-3}$ )<br>(size >2.5 nm)      | 1.2 $\pm$ 0.8         | C3: Sodium salts ( $\mu\text{g m}^{-3}$ )                  |   | 1.2 $\pm$ 1.2                             |
| Particle Number ( $10^3 \text{ cm}^{-3}$ )<br>(size: 2.5 to 7 nm) | 4.7 $\pm$ 4.9         | C4: Secondary inorganics-amine ( $\mu\text{g m}^{-3}$ )    |   | 0.1 $\pm$ 0.6                             |
| BC ( $\mu\text{g m}^{-3}$ )                                       | 0.8 $\pm$ 0.7         | C5: Biomass burning–soil ( $\mu\text{g m}^{-3}$ )          |   | 1.2 $\pm$ 1.3                             |
| PM <sub>2.5</sub> ( $\mu\text{g m}^{-3}$ )                        | 6.3 $\pm$ 3.2         | C6: Biomass burning–organosulfate ( $\mu\text{g m}^{-3}$ ) |   | 0.4 $\pm$ 0.6                             |
| PM <sub>10</sub> ( $\mu\text{g m}^{-3}$ )                         | 8.6 $\pm$ 4.2         | C7: Mixed/aged dust ( $\mu\text{g m}^{-3}$ )               | 0.5 $\pm$ 0.7                                     |   |

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Note that the full names of seven particle classes are C1: calcium-rich and soil-dust-like particles; C2: aged-soot-like particles; C3: sodium-salt-like particles; C4: secondary- inorganics-rich and amine containing particles; C5: aged-biomass-burning and soil-dust like particles; C6: aged-biomass-burning and organosulfate-containing particles; C7: mixed/aged and dust-like particles

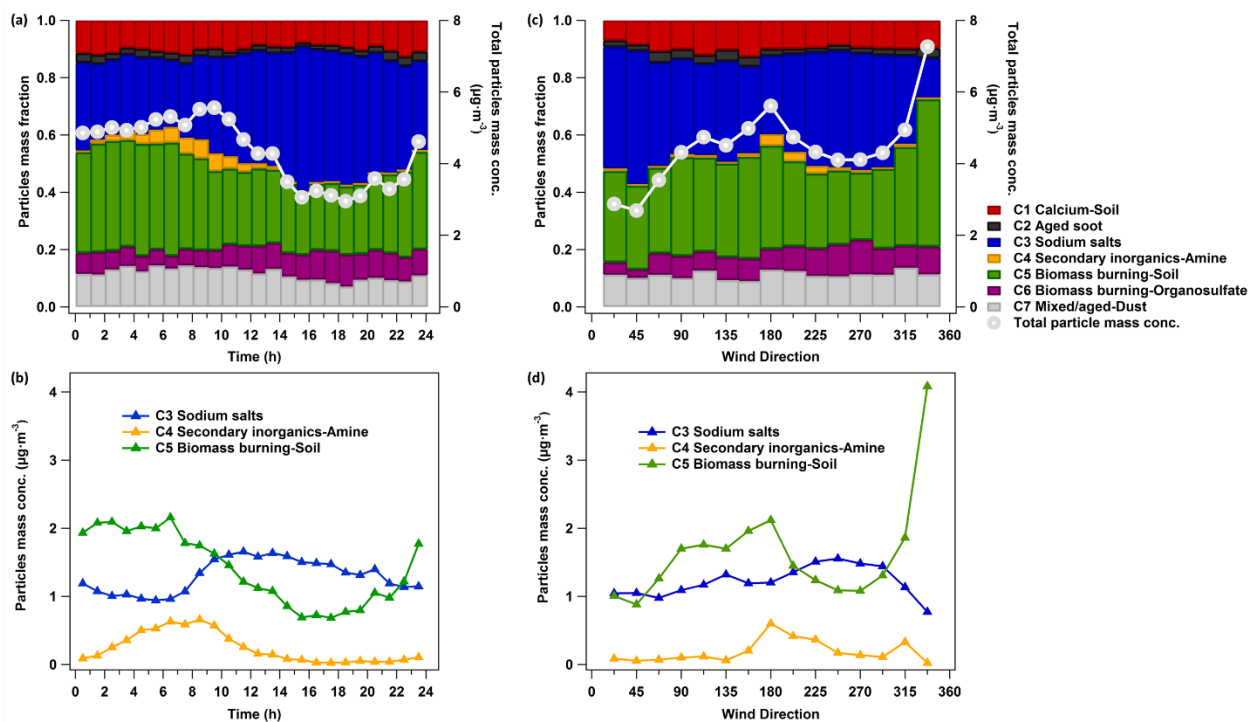
**Table 2: Features and mean values ( $\pm$ sdev) of characteristic parameters for the two selected episodes**

|   | Sodium chloride rich episode                               | Organics rich episode  | Instrument          |
|---|--|--|---------------------|
| Date and time   | 0820 noon to 0823 noon                                     | 0825 midnight to 0828 midnight                                 |                     |
| Sodium salts ( $\mu\text{g m}^{-3}$ )   | 2.2 $\pm$ 1.0<br>Higher                                    | 0.6 $\pm$ 0.4<br>Lower   | LAAPTOF             |
| Biomass burning–soil ( $\mu\text{g m}^{-3}$ )                                     | 0.7 $\pm$ 0.7<br>Lower                                     | 1.7 $\pm$ 1.1<br>Higher  | LAAPTOF             |
| Organics ( $\mu\text{g m}^{-3}$ )   | 2.0 $\pm$ 1.2<br>Lowest                                    | 12.9 $\pm$ 3.9<br>Highest                                      | HR-ToF-AMS          |
| Particle acidity indicator<br>NH <sub>4</sub> <sup>+</sup> measured/predicted     | 0.67 $\pm$ 0.10<br>Highest acidity                         | 0.83 $\pm$ 0.06<br>Average acidity<br>Anti-correlated with org | HR-ToF-AMS          |
| Organonitrate fraction indicator<br>NO <sup>+</sup> /NO <sub>2</sub> <sup>+</sup> | 11.0 $\pm$ 6.1<br>Highest                                  | 6.1 $\pm$ 0.9<br>Lowest  | HR-ToF-AMS          |
| Oxidation state (Osc)<br>(Anti-correlated with org)                               | -0.24 $\pm$ 0.34<br>Less oxidized<br>Strongest variability | -0.10 $\pm$ 0.20<br>More oxidized<br>Least variability         | HR-ToF-AMS          |
| Wind Direction ( $^{\circ}$ )   | Southwest  | Varying  | Meteorology sensor  |
| Wind Speed ( $\text{m s}^{-1}$ )  | 1.6 $\pm$ 1.0  | 1.2 $\pm$ 1.0  | (WS700, Lufft GmbH) |



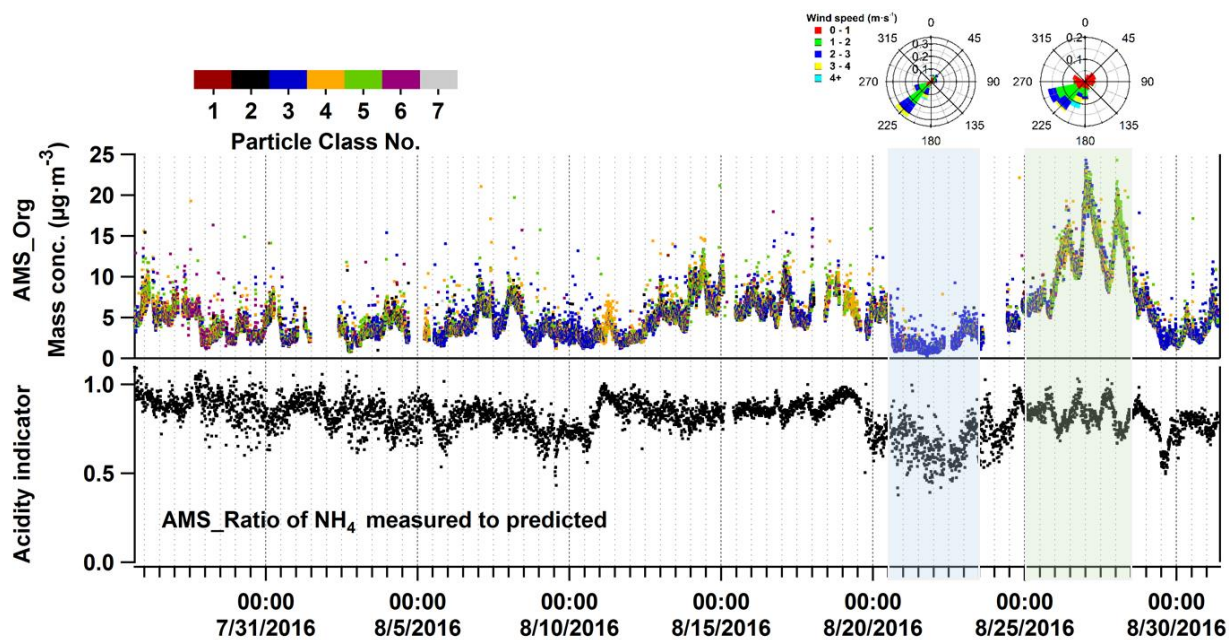
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Figure 1: Measurement location and topography (left), nearby peak emission sites and wind rose for the complete measurement campaign in summer 2016 (right).



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Figure 2: Diurnal pattern (a and b) and wind direction resolved particle class masses (c and d) measured by LAAPTOF. Please note that the uncertainties of the particle class mass concentrations are relatively large (Shen et al., 2019).



650 Figure 3: Time series of particulate organic mass measured by the AMS (AMS org in the upper panel), the particle acidity indicator (the lower the value the more acidic, lower panel), and the particle class number classified by LAAPTOF (colour code, upper panel). The seven classes are class 1: calcium-Soil; class 2: aged soot; class3: sodium salts; class 4: secondary inorganics-amine; class 5: biomass burning-soil; class 6: biomass burning-organosulfate; and class 7: mixed/aged-dust. The first (blue) and second (green) shaded areas mark the sodium chloride rich and organic rich episodes, respectively.

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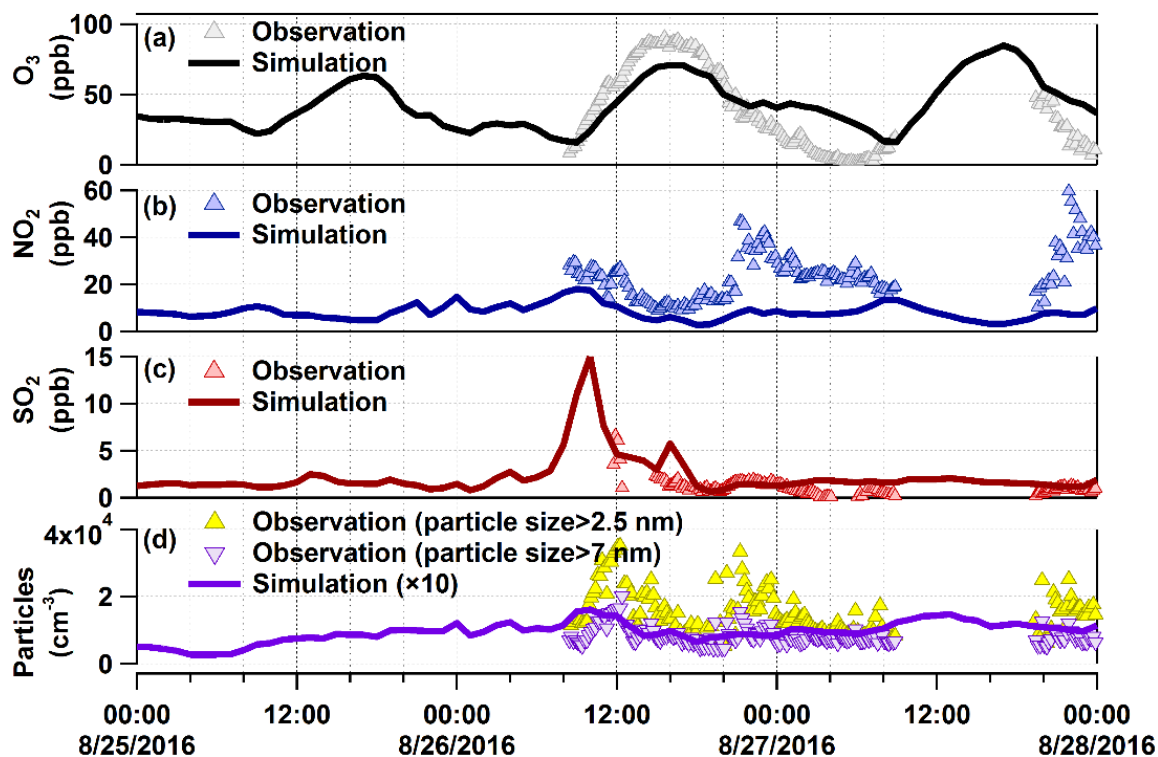
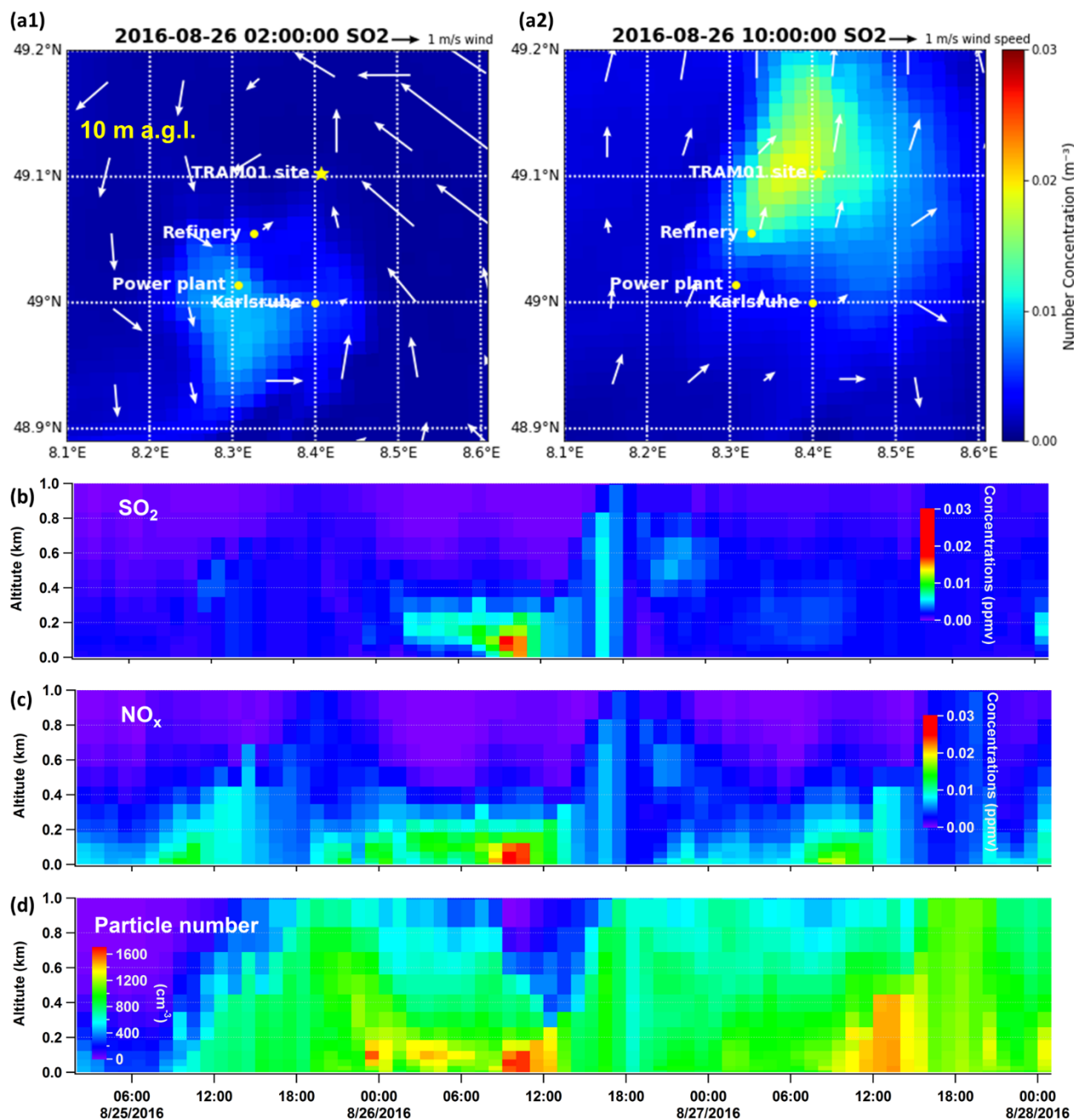


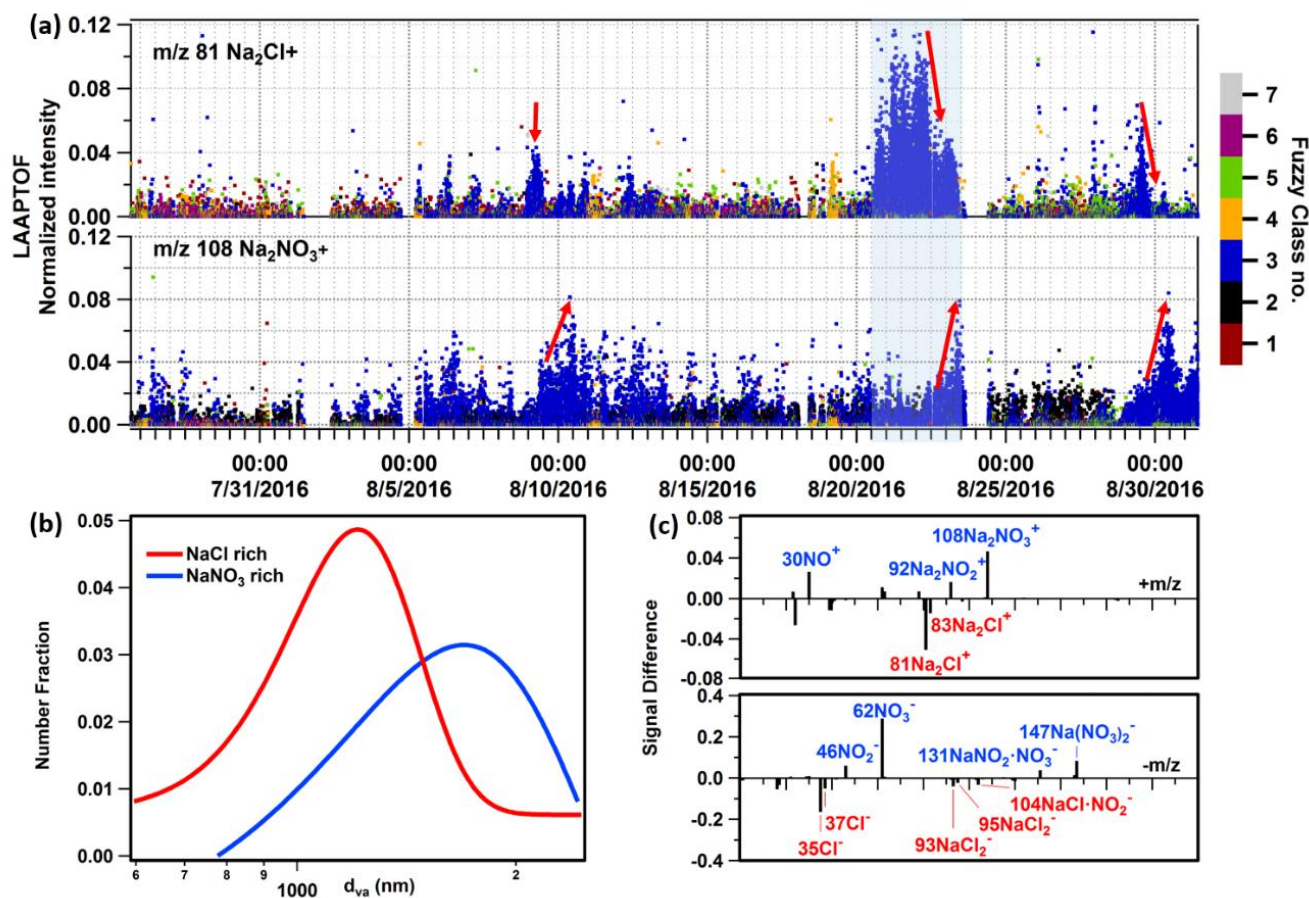
Figure 4: Comparison between observations and model results for  $O_3$ ,  $NO_2$ ,  $SO_2$ , and particle number in organic-rich episode. The modelled particles do not represent the ultrafine particles to their full extend, especially for the particles in the nucleation mode.





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Figure 5: Modelled SO<sub>2</sub> concentrations at 2:00 (a1) and 10:00 (a2) on August 26<sup>th</sup> at 10 m a.g.l.. Panels (b) to (d) are time series of vertical profiles above the measurement site for simulated SO<sub>2</sub>, NO<sub>x</sub> and total particle number.



665 **Figure 6:** (a) Time series of sodium chloride and nitrate signatures in seven particle classes. Y-axis is the normalized ion intensity  
 (each ion peak intensity is normalized to the sum of all ion signals; positive and negative ions were analysed separately). The seven  
 classes are class 1: calcium-Soil; class 2: aged soot; class3: sodium salts; class 4: secondary inorganics-amine; class 5: biomass burning-  
 soil; class 6: biomass burning-organosulfate; and class 7: mixed/aged-dust. Obvious sodium chloride and nitrate signatures are mainly  
 found in class 3 labelled in blue colour. (b) Size distributions of NaCl rich (young sea salt) and  $\text{NaNO}_3$  rich (aged sea salt) particles.  
 Gaussian fit  $\text{NaCl}$   $d_{va} (\pm\sigma) = 1211(\pm 352)$  nm;  $\text{NaNO}_3$   $1697(\pm 946)$  nm. (c) Spectra difference between aged and young sea salt particles  
 670 (aged-young). The blue band marks the sodium chloride rich episode.

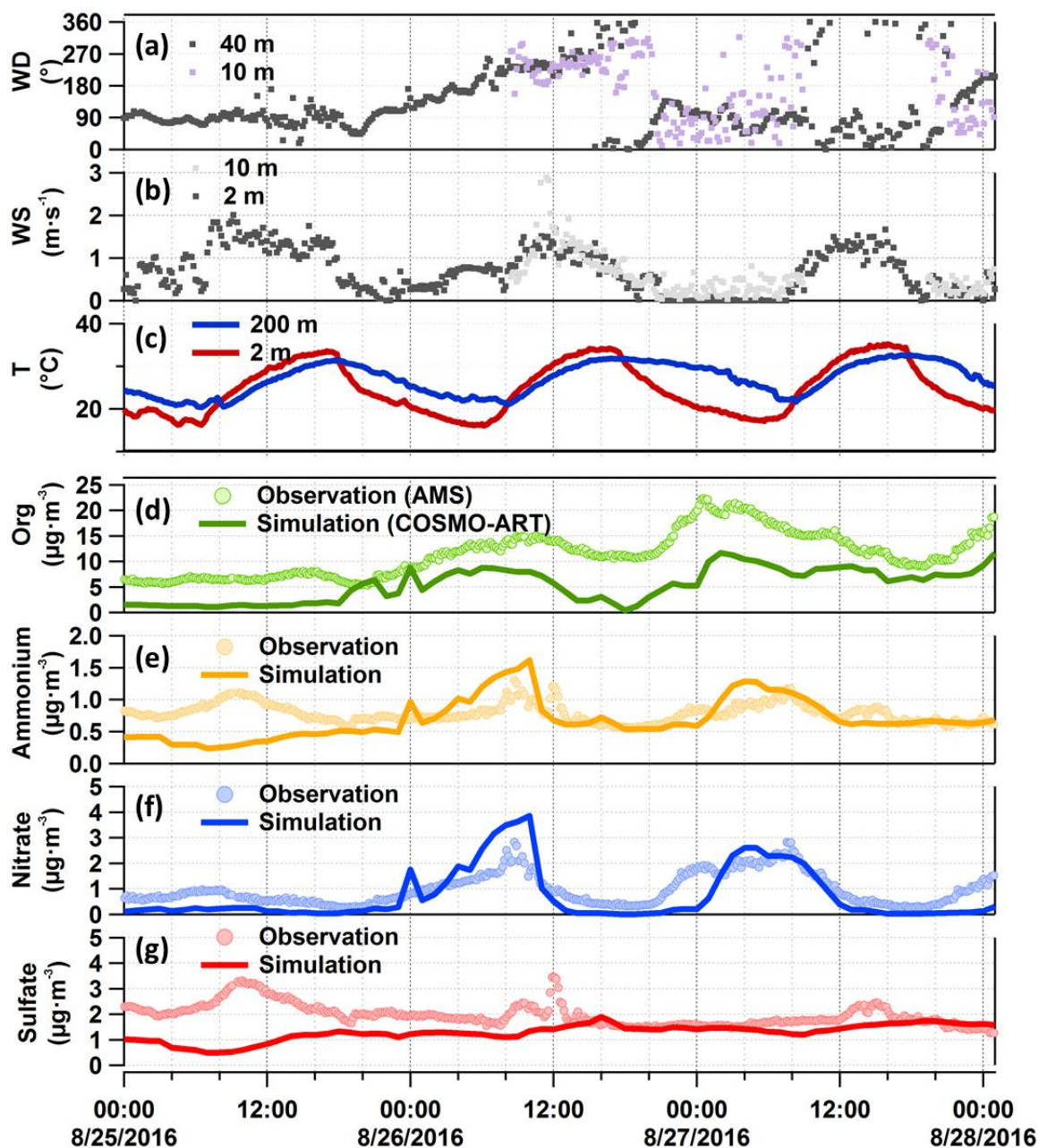


Figure 7: Comparison between observations and model simulations for organic-rich episode. Panel (a) and (b) are wind conditions measured 10 m a.g.l. at our measurement site (purple and light grey dots for wind direction and speed, respectively), as well as the complementary KIT-tower data (dark grey dots). Panel (c) is the temperature data from the KIT-tower. Panel (d) to (g) are the particulate organic compounds, ammonium, nitrate, and sulfate measured by AMS and simulated by COSMO-ART. The simulation results are at the lowest atmospheric level in the model, namely 10 m a.g.l.

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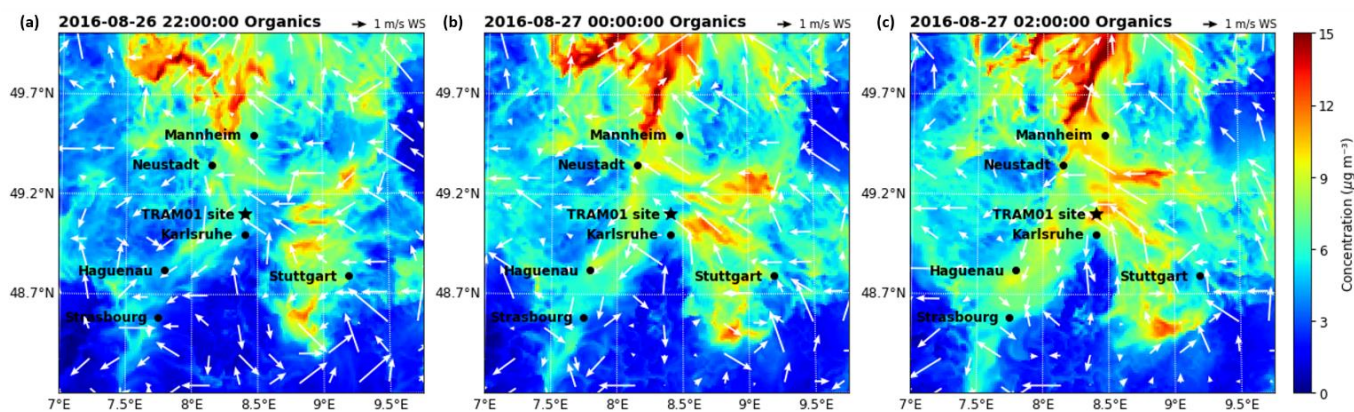
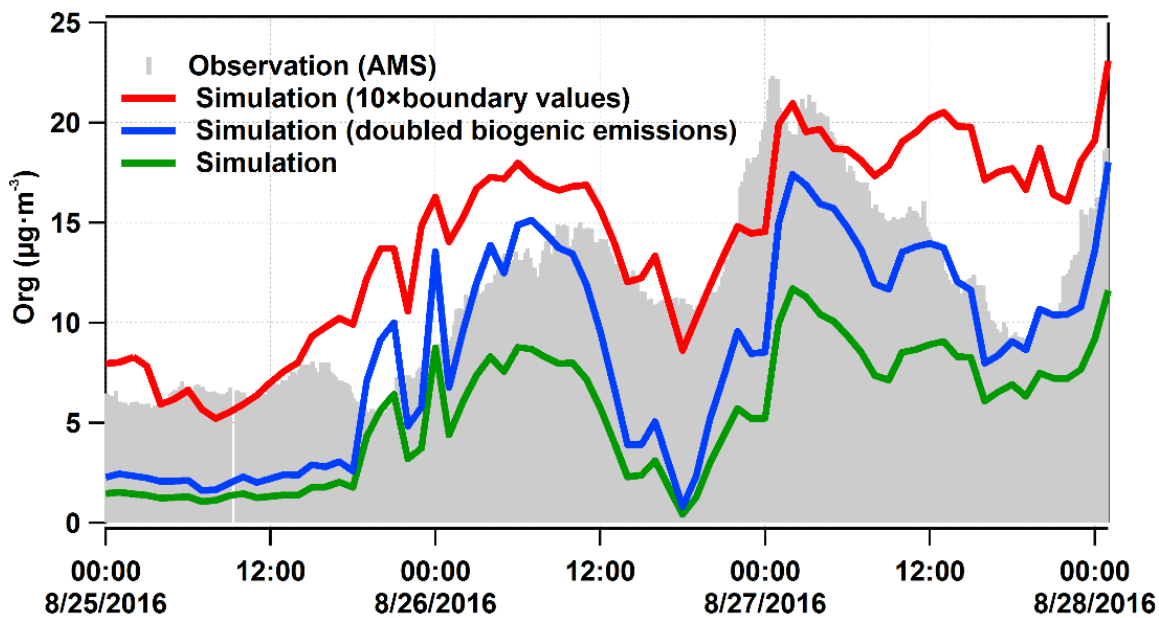


Figure 8: Particulate organics concentrations calculated by COSMO-ART for three different time points (a) to (c) during the night from August 26<sup>th</sup> to 27<sup>th</sup> for 10 m a.g.l..



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Figure 9: Modelled organic mass concentrations (lines) compared to measured values (grey area). The green line is the same as in Figure 7, the red line was calculated for a tenfold-enhanced regional background in organics, and the blue was calculated for doubles biogenic emissions in COSMO-ART.