

Reply to Co-Editor and Referees

We gratefully thank the co-editor and referees for their helpful comments to improve the quality of our manuscript. The comments are in black. Our point-to-point replies marked by “R” are in blue. Changes to the manuscript text are in green.

Co-editor’s comments:

1. Line 136: The use of the PM_{2.5} lens on the AMS needs to be explicitly stated and perhaps cited (Peck et al., 2016) since it is not standard yet.

R: We have cited this reference in the revised manuscript.

2. Line 151: The software is written for Igor, it is not IGOR software.

R: We have revised this to “standard software written for Igor”

3. Lines 244-259: (weekday/weekend analysis) Can you please add details about weekday/weekend statistics (supplement would be fine). In particular, some indication of the observed variability in the particle class concentrations in Figs. 2 and S5 would help the reader judge the significance of the weekday/weekend claims. In particular, the Biomass burning-soil effect looks like it may be mostly driven by 1 weekend event in mid-August.

R: We have added details about weekday/weekend statistics in the caption to Fig S5 and indicated the variability in the caption to Fig 2, as follows:

Fig S5 caption:

The data comprises 4842 time points of 10 min averaged data, including 3702 for weekdays and 1140 for weekends, including Saturday and Sunday. The total particle mass concentration for each wind direction is on average $3.8 (\pm 2.3, \text{one } \sigma) \mu\text{g m}^{-3}$ and $5.2 (\pm 3.0) \mu\text{g m}^{-3}$ during weekdays and the weekends, respectively. For the dominating particle classes the average mass concentrations are $1.3 (\pm 1.2) \mu\text{g m}^{-3}$ and $0.8 (\pm 0.5) \mu\text{g m}^{-3}$ for class 3, $0.2 (\pm 0.5) \mu\text{g m}^{-3}$ and $0.1 (\pm 0.2) \mu\text{g m}^{-3}$ for class 4, $1.1 (\pm 1.0) \mu\text{g m}^{-3}$ and $2.3 (\pm 1.7) \mu\text{g m}^{-3}$ for class 5, for weekdays and weekends respectively.

Fig 2 caption:

In total 4842 time points of 10 min averaged data were analysed for these plots. The total particle mass concentration for each hour of the day is on average $4.3 (\pm 2.6, \text{one } \sigma) \mu\text{g m}^{-3}$ and for each wind direction is on average $4.5 (\pm 2.9) \mu\text{g m}^{-3}$. For the dominating classes the average mass concentrations are $1.3 (\pm 1.1) \mu\text{g m}^{-3}$ and $1.2 (\pm 1.0) \mu\text{g m}^{-3}$ for class 3, $0.2 (\pm 0.5) \mu\text{g m}^{-3}$ and $0.2 (\pm 0.4) \mu\text{g m}^{-3}$ for class 4, $1.4 (\pm 1.0) \mu\text{g m}^{-3}$ and $1.6 (\pm 1.5) \mu\text{g m}^{-3}$ for class 5, for (b) and (d) respectively.

4. Line 273-275: Please clarify that the PMF analysis is for the AMS data.

R: We have clarify this as “AMS-PMF”

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5. Lines 260-278: I find this section confusing. It seems that the conclusion at the end is that biomass burning is not a source of the biomass burning-soil particles. Is this correct? If so, would it be better to give the class a different name?

R: The names for particle classes were only based on the LAAPTOF data analysis (different classes were directly derived from clustering for single particle data). This is discussed in detail in our previous study (Shen et al., 2019). The representative mass spectra of class 5 particles most likely feature aged biomass-burning and soil dust (Shen et al., 2019). In this study, we tried to refine the type for class 5, however, we did not achieve a consistent identification for this class even with the additional information obtained from other instruments, i.e., AMS and Aethalometer. Therefore, we would like to keep this name that is used in our previous publication (Shen et al., 2019) even though we discuss that also other particle types may contribute to this class.

6. Lines 287-293: This acidity indicator is highly uncertain (e.g., Hennigan et al., (2015)). Given the presence of Na⁺, it is even more likely to be erroneous under these conditions. Please add a discussion regarding the limitations of this acidity indicator and its impacts on your conclusions.

R: We have revised the corresponding contents in the last paragraph of section 3.1, as follows:

“For particle acidity, the molar ratio of NH₄⁺ measured/predicted from AMS measurement, is commonly used as an indicator (Zhang et al., 2007). The predicted NH₄⁺ value is the mass concentration required to fully neutralize sulfate, nitrate, and chloride measured by AMS. Lower ratio value represents higher acidity under certain conditions, i.e., inorganic species are dominating the particle mass and crustal elements have relatively low concentrations (Zhang et al., 2007). It should be noted that the organic acids and crustal elements such as sodium can bias the results (Hennigan et al., 2015, and the reference therein). Despite of the large uncertainty for indicating acidity, we still can use such ratio to indicate special events, e.g., sodium nitrate rich particles will provide nitrate, leading to lower values. As shown in Fig. 3, this ratio reaches lowest value during the sodium salts rich episode, while the value in the organic rich episode is at the average level. If we assume the organic acids only have minor influence on the particle acidity, the anti-correlation between acidity indicator value and organic concentrations may be related to acid-catalysed organic particle formation.”

In addition, we have changed “highest acidity” to “lowest value” (for acidity indicator) in table 2

7. Lines 294-296: NaNO₃ gives a different NO⁺/NO₂⁺ ratio than does ammonium nitrate. Please comment on how this could impact the interpretation of these results.

R: We have added the comments as follows:

“It should be noted that NaNO₃ can also result in higher NO⁺/NO₂⁺ ratios (5 to 7, Allan et al., 2004) than NH₄NO₃ (2.3 in our calibration). However, the NO⁺/NO₂⁺ values here are much higher (c.f. Fig. S7) and thus we can expect a major contribution by organonitrates.”

8. Line 299: The average oxidation state does not look all that much different to me during the organic rich episode. Perhaps there is slightly less diurnal variability which results in an average state that is more oxidized? Please discuss further.

R: Yes, the time series of oxidation state (Fig. S7) does not show much difference regarding the average, however, the statistic calculation results are -0.19 ± 0.23 for the entire campaign, -0.10 ± 0.20 for the organic rich episode,

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and -0.24 ± 0.34 for sodium chloride rich episode, respectively. The latter two values are listed in Table 2 with the variability description. We have added some statement, as follows:

“It turns out that the organics were more oxidized in the organic rich episode with relatively higher \overline{OS}_c value and less variability (-0.10 ± 0.20) than that for the entire campaign (-0.19 ± 0.23) (refer to Fig. S7 and Table 2).”

9. Lines 350-360: Please see previous comment regarding NO⁺/NO₂⁺ ratio. It is very difficult to judge correlation between ions from Figure S10 due to the amount of data included. Please consider showing only the time period in question.

R: We have revised Figure S10 to show the sodium salts rich and organic rich episodes. In addition, we have added FIGAERO-HR-ToF-CIMS (stands for a filter inlet for gases and aerosols coupled to a high-resolution time-of-flight chemical ionization mass spectrometer, hereafter CIMS) results on organonitrates and modified the caption, as follows:

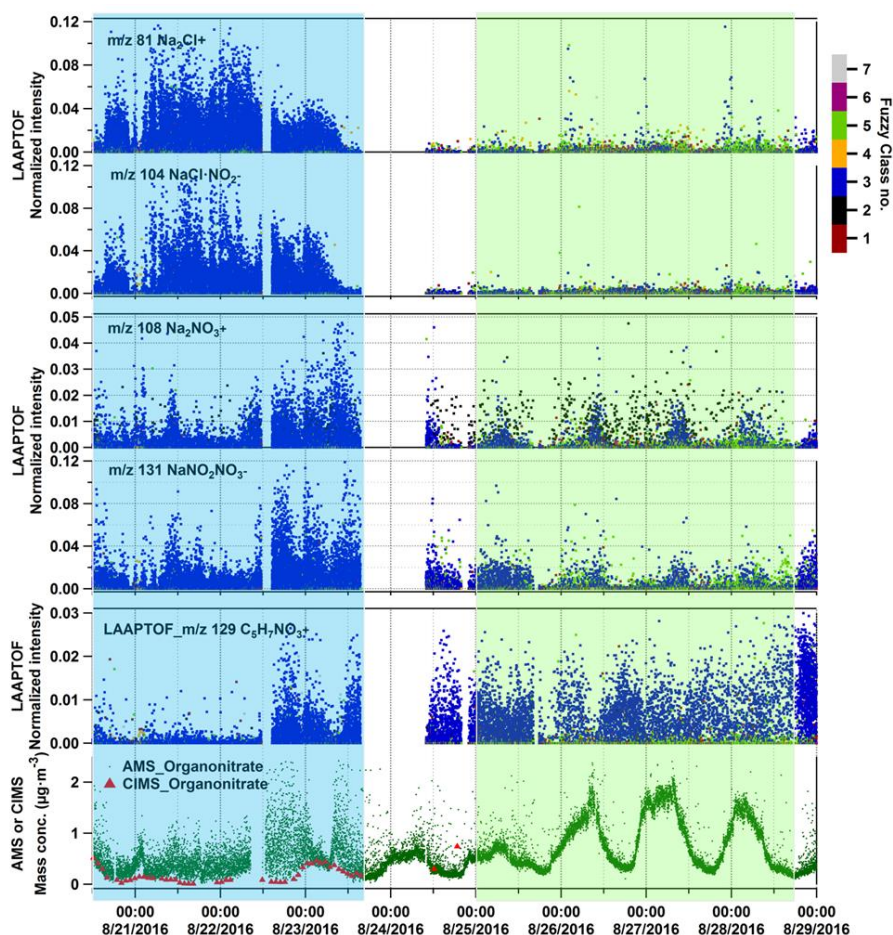


Figure S10: Time series of sodium chloride and nitrate, and organonitrate signatures in LAAPTOF mass spectra, as well as organonitrate mass concentration derived from AMS and FIGAERO-HR-ToF-CIMS (stands for a filter inlet for gases and aerosols coupled to a high-resolution time-of-flight chemical ionization mass spectrometer, hereafter CIMS) measurement. For LAAPTOF results, Y-axis is the normalized ion intensity (each ion peak intensity is normalized to the sum of all ion signals; positive and negative ions were analysed separately). The seven classes are class 1: calcium-Soil; class 2: aged soot; class3: sodium salts; class 4: secondary inorganics-amine; class 5: biomass burning-soil; class 6: biomass burning-organosulfate; and class 7: mixed/aged-dust. Obvious sodium chloride and nitrate signatures are mainly found in class 3 labelled in blue colour. AMS and CIMS results (Pearson’s correlation coefficient $\gamma=0.52$) are adapted from our companion study by Huang et al (2019). It should be noted that during the sodium chloride rich episode (first band in blue) the fraction of organonitrates is higher (Fig. S7), although the organonitrate mass concentration is lower than that during the organic rich episode (second band in green).

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10. Lines 360-363: It seems to me premature to “confirm” that the $C_5H_7NO_3^+$ signal is due to organonitrates in this data given the possible issues with AMS NO/NO₂.

R: Our AIDA simulation chamber study has shown evidence that this LAAPTOF signal m/z 129 $C_5H_7NO_3^+$ can arise from the recombination of inorganic nitrate and organics due to the laser ionization matrix effect and even stronger when arising from organonitrates formed under NO_x rich conditions (refer to Fig. S11). In the field measurement, we have observed quite strong $C_5H_7NO_3^+$ signals at the end of sodium salts rich episode, when a CIMS was also deployed. In the companion study, Huang et al. (2019) showed that AMS derived organonitrate mass concentrations were comparable with the results of molecular analysis using CIMS with Pearson's correlation coefficient $\gamma = 0.52$ (cf. Fig. S1 in Huang et al., 2019). This validates to some extent that the method for organonitrate estimation based on the AMS NO₂/NO₊ ratio (Farmer et al., 2010) is applicable in our study. In the revised supplement here, we have added the AMS and CIMS results for comparison. Taken together, we can conclude that m/z 129 $C_5H_7NO_3^+$ was mainly resulting from organonitrates, but we cannot rule out the contribution from the recombination of inorganic nitrate and organics due to the laser ionization matrix effect.

We have revised the corresponding sentences, as follows:

“For sodium salt particles, a weak but characteristic peak at m/z 129 $C_5H_7NO_3^+$ is likely to be an organonitrate signature (Shen et al., 2019). ... In the sodium chloride rich episode, in general m/z 129 $C_5H_7NO_3^+$ has a similar trend as organonitrate mass concentrations estimated by AMS results and the results of a molecular analysis using FIGAERO-HR-ToF-CIMS (stands for a filter inlet for gases and aerosols coupled to a high-resolution time-of-flight chemical ionization mass spectrometer) in a companion study by (Huang et al., 2019). ... In addition, our simulation chamber study has shown that m/z 129 $C_5H_7NO_3^+$ arises from the recombination of inorganic nitrate and organics due to the laser desorption and ionization (LDI) matrix effect in single-particle MS (Murphy, 2007; Reilly et al., 2000; Reinard and Johnston, 2008; Zenobi and Knochenmuss, 1998) and even stronger when it is arising from organonitrates formed under NO_x rich conditions (refer to Fig. S11). Taken together, we can conclude that the m/z 129 $C_5H_7NO_3^+$ signal is mainly resulting from organonitrates, but we still cannot rule out the LDI effect.”

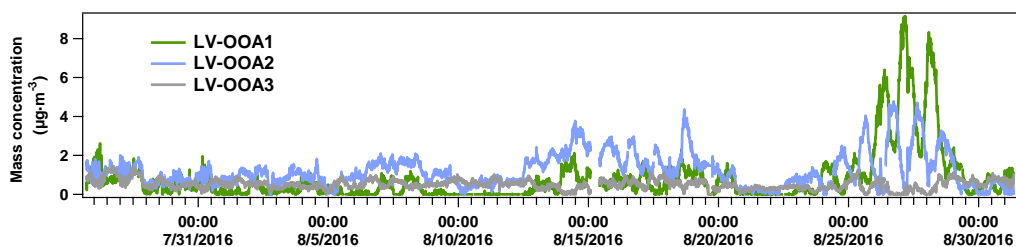
11. Lines 411-420: The diurnal cycle of LV-OOA1 is difficult to see with the y-axis range given in Fig. S4. I cannot see that there are 2 peaks.

R: We have change the y-axis range in Fig. S4 (c) and added arrows to point out the 2 peaks in the inserted figure in panel (d). Please refer to our reply to the 1st Referee comment #6.

12. Sect 3.4: More information needs to be provided on the AMS PMF results (supplement is ok). Specifically, why were 5 factors and not 4 chosen? There is significant similarity in the three LV-OOA signals and LV-OOA3 doesn't seem to have many distinguishing features.

R: The criteria to choose factors have been given in method section 2.2 of the main manuscript. The reasons to choose 5 factors rather than 4 are: 1) residuals (unexplained data) resulting from factor 4 is more than factor 5; 2) in spite of some similarity, distinguishable features can still be found for three LV-OOA signals, according to their time series and diurnal patterns. In particular during the organic rich episode, LV-OOA1 and LV-OOA2 show negative correlation, the LV-OOA3 signal is much weaker than the other two LV-OOA during the organic rich episode. Obvious differences can be seen in the time series shown in the figure below:

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We have added more information in the supplement, as follows:

“The criteria for choosing factor number are given in section 2.2 of main manuscript. As shown in the Fig. S4 (a) HOA, SV-OOA, and LV-OOA have distinguishable spectra. For three LV-OOA, in spite of some similarity, distinguishable features can still be found, according to their time series (b) and diurnal patterns (c). In particular for the organic rich episode, LV-OOA1 and LV-OOA2 show anti-correlation, the LV-OOA3 signal is much weaker than the other two LV-OOA. Therefore, a five-factor solution was chosen rather than four or three-factor solution.”

13. Line 453: I don't understand what is meant by “with/without biogenic emissions” for test 1. Are the biogenic emissions included or are they not?

R: “With/without biogenic emission test” means the comparison between the results with and without biogenic emissions. The basic model setting is with biogenic emissions, the sensitivity test is without them. Thank you for pointing out this confusing expression. We have revised it as follows:

“one without biogenic emissions (test 01)”

14. Figure 2 caption: Can you expand on what is meant by “relatively large” and what the implications of this uncertainty is for the data interpretation.

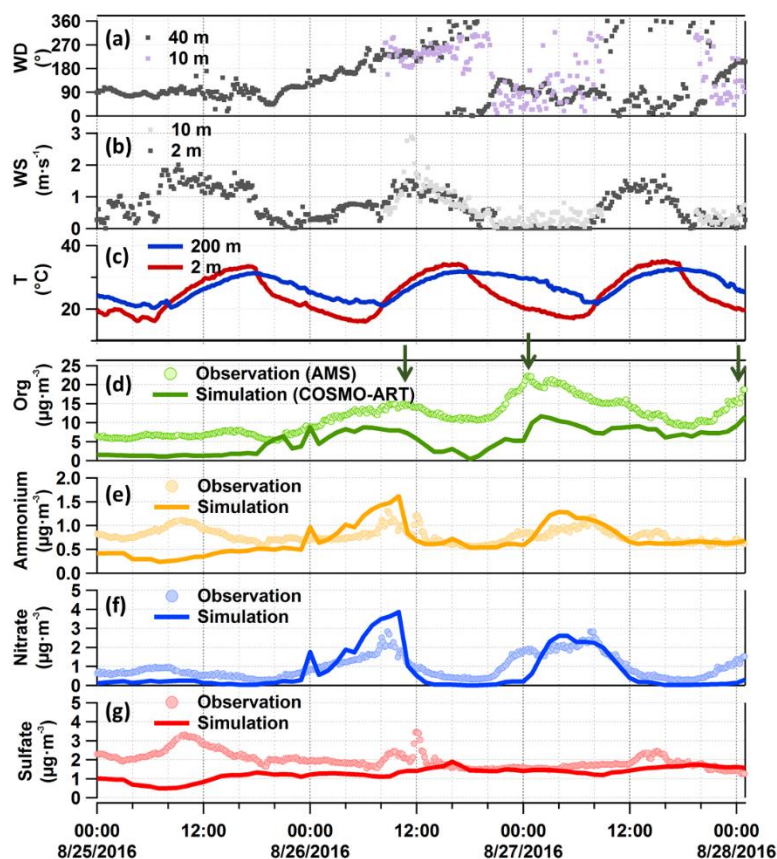
R: Yes. We have added more explanation in the caption as follows:

“It is worthy to note that mass quantification is one of the most challenging issues in single particle mass spectrometry (Shen et al., 2019 and the references therein). There are several assumptions mainly regarding particle shape and density in our particle mass quantification method, leading to potential uncertainties of up to ~300 %. Despite such large uncertainty, the estimated total particle mass has good correlation with the total mass measured by the AMS, covering ~85% of the measurement time (Shen et al., 2019).”

15. Figure 7: Please consider using shading to highlight the organic peaks of interest. It will make it easier for the reader to move between the text and the figure.

R: Thank you for your suggestion. Considering the cleanliness, we have used three green arrows to point out the peaks, as shown below:

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References:

- Allan, J. D., Bower, K. N., Coe, H., Boudries, H., Jayne, J. T., Canagaratna, M. R., Millet, D. B., Goldstein, A. H., Quinn, P. K., Weber, R. J., and Worsnop, D. R.: Submicron aerosol composition at Trinidad Head, California, during ITCT 2K2: Its relationship with gas phase volatile organic carbon and assessment of instrument performance, *J Geophys Res-Atmos*, 109, 2004.
- Hennigan, C. J., Izumi, J., Sullivan, A. P., Weber, R. J., and Nenes, A.: A critical evaluation of proxy methods used to estimate the acidity of atmospheric particles, *Atmospheric Chemistry and Physics*, 15, 2775–2790, 2015.
- Huang, W., Saathoff, H., Shen, X. L., Ramakrishna, R., Leisner, T., and Mohr, C.: Chemical characterization of highly functionalized organonitrates contributing to high night-time organic aerosol mass loadings and particle growth, *Environ Sci Technol*, 53, 1165–1174, 2019.
- Zhang, Q., Jimenez, J. L., Worsnop, D. R., and Canagaratna, M.: A case study of urban particle acidity and its influence on secondary organic aerosol, *Environ Sci Technol*, 41, 3213–3219, 2007.

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Referee #1 comments:

General comments:

The present work conducted a 6 week measurement of particle chemical composition combining LAAPToF and AMS technique at a rural site of southwest Germany. Also the regional transport model COSMO-ART was used to investigate particle sources as well as their transport pattern. The combination of measurement and model can better understand the particle sources and regional transport. The results are straightforward and useful, but some clarification and minor changes are still needed for this manuscript. I suggest a minor revision for this manuscript before publication.

Specific comments:

1. Line 18: It's a little confusing with the statement: sodium salts accounted for 36%, but organics, sulfate, nitrate, and ammonium accounted for 58%, 22%, 10%, and 9% (sum up 99%). I believe that the total mass the authors used for AMS is the sum of org and SNA. While LAAPToF measures not only non-refractory particles. It seems these results cannot be closure, especially for some readers who are not very familiar with these two instruments. Anyway, some clarification is needed to avoid misunderstanding these results.

R: Yes, the total mass used for AMS is the sum of organics and SNA, as well as chloride. In contrast, the total mass determined by LAAPTOF is the sum of 7 particle types. We have revised this sentence for clarification, as follows:

“For the entire measurement period, the total aerosol particle mass was dominated by sodium salts contributing on average (36 ± 27) % to the total single particle mass measured by the LAAPTOF. The total particulate organic compounds, sulfate, nitrate, and ammonium contributed on average (58 ± 12) %, (22 ± 7) %, (10 ± 1) %, and (9 ± 3) % to the total non-refractory particle mass measured by the AMS.”

2. Line 123: Do AMS use PM2.5 lenses?

R: AMS and LAAPTOF are both are equipped with PM2.5 aerodynamic inlet lenses.

3. Line 121: It's no need for this too many references here.

R: We have removed some references here. The revised sentence is as follows:

“The LAAPTOF is a commercially available single particle mass spectrometer and has been well described in recent publications (Gemayel et al., 2016; Marsden et al., 2016; Ramisetty et al., 2018; Shen et al., 2018).”

4. Line195: Since the MS discussed long-range transport, I suggest discussing more about secondary aerosols as well.

R: The long-range transport we discuss is related to a unique episode where the sodium salts accounted for (70 ± 24) % of the total mass of the single particles measured by the LAAPTOF. In addition, the mass concentrations of organics and secondary inorganics (ammonium, sulfate, and nitrate) measured by the AMS were at the lowest level compared to the other periods during the entire measurement campaign. During the long-range transport episode secondary species mainly formed on the sodium salt particles e.g. NaNO_3 . This is discussed regarding sub-classes and the sea salt aging process in section 3.3 “Aged sea salt in central Europe”.

5. Line 188: I suggest to explain a little bit about the particle classes, not only show the names, because not all the readers have read the paper (Shen et al. 2019)

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R: We have added more explanations, as follows:

“In a previous publication (Shen et al., 2019), we could identify seven major particle classes at the measurement site among $\sim 3.7 \times 10^5$ single particles with sizes between 200 nm and 2.5 μm d_{va} . In brief, Class 1 are calcium-rich and soil-dust-like particles (calcium-soil) with marker peaks at m/z 40 Ca^+ , 56 CaO^+ , 57 CaOH^+ , 75 CaCl^+ , 96 Ca_2O^+ , and 112 $(\text{CaO})_2^+$. This particle class shows a good correlation (Pearson’s correlation coefficient $\gamma \geq 0.6$) with the laboratory derived reference spectra of soil dust particles. Class 2 are aged-soot-like particles (aged soot) with characteristic C_n^\pm patterns and are mixed with sulfate and nitrate. Class 3 are sodium-salt-like particles (sodium salts) with marker peaks at m/z 23 Na^+ , 81/83 Na_2Cl^+ , 139/141 Na_3Cl_2^+ , 165 Na_3SO_4^+ , 35/37 Cl^- , 93/95 NaCl_2^- , 115 $\text{Na}(\text{NO}_2)_2^-$, 131 $\text{NaNO}_2\text{NO}_3^-$, and 147 $\text{Na}(\text{NO}_3)_2^-$. Class 4 are “Secondary inorganic–amine” particles with prominent ammonium, nitrate and sulfate markers, as well as amine marker peaks at m/z 58 $\text{C}_2\text{H}_5\text{NHCH}_2^+$, 59 $(\text{CH}_3)_3\text{N}^+$, 86 $(\text{C}_2\text{H}_5)_2\text{NCH}_2^+$, 88 $(\text{C}_2\text{H}_5)_2\text{NO}/\text{C}_3\text{H}_6\text{NO}_2^+$, and 118 $(\text{C}_2\text{H}_5)_2\text{NCH}_2^+$. Class 5, aged-biomass-burning and soil-dust like particles (biomass burning–soil), has aged biomass burning marker peaks at m/z 39 K^+ , 213 K_3SO_4^+ , 46 NO_2^- , 62 NO_3^- , 97 HSO_4^- , and peaks related to aromatic compounds (50 C_4H_2^+ , 63 C_5H_3^+ , 77 C_6H_5^+ , 85 C_7H^+ , 91 C_7H_7^+ , 95 $\text{C}_7\text{H}_{11}^+$, 104 C_8H_8^+ , 115 C_9H_7^+). Furthermore, class 5 shows a strong correlation with reference spectra of soil dust particles. The representative spectra of class 6, aged-biomass-burning and organosulfate-containing particles (biomass burning–organosulfate), also features aged biomass burning particles and has organosulfates marker peaks at m/z 141 $\text{C}_2\text{H}_5\text{OSO}_4^-$, 155 $\text{C}_2\text{H}_3\text{O}_2\text{SO}_4^-$, and 215 $\text{C}_5\text{H}_{11}\text{OSO}_4^-$. Class 7 particles contain all the marker peaks from the other six classes and show a good correlation with the reference spectra of mineral and soil dust particles, and are consequently labelled as “mixed/aged dust” (short for mixed/aged and dust-like particles) ...”

6. Session 3.4.1 and 3.4.2: Because SOA is also an important component of fine particles, I suggest discussing more about SOA. Besides, AMS-PMF can provide OOA (SOA). This could be considered in the discussion.

R: Thank you for your suggestion. We have done a PMF analysis, added more explanation in the introduction, added one subsection in method part, added more discussion about SOA in section 3.1 and 3.4, put the corresponding results in the supplement, and modified the conclusions accordingly. The corresponding changes are listed as follows:

Abstract

“...Positive matrix factorization (PMF) analysis for the AMS data suggests that the total organic aerosol (OA) consisted of five components, including (9 ± 7) % hydrocarbon-like OA (HOA), (16 ± 11) % semi-volatile oxygenated OA (SV-OOA), and (75 ± 15) % low-volatility oxygenated OA (LV-OOA) ...”

Section 1 Introduction (3rd paragraph)

“...According to PMF analysis, the organic aerosol (OA) can be separated into e.g., hydrocarbon-like OA (HOA), cooking-related OA (COA), nitrogen-enriched OA (NOA), biomass burning OA (BBOA), and semi-volatile oxygenated OA (SV-OOA), and low-volatility oxygenated OA (LV-OOA). This information hints to OA aging and potential sources...”

Section 2.2 Mass spectrometer data analysis

“Single particle data were analysed by using the Igor LAAPTOF data analysis software (version 1.0.2, AeroMegt

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GmbH). The corresponding basic procedures (e.g., mass calibration and fuzzy c-means clustering) and particle mass quantification method have been described in detail in our previous studies (Shen et al., 2018 and 2019). The AMS data analysis was done with the standard software written for Igor, including SQUIRREL (version 1.60C) and PIKA (version 1.20C). To account for a particle bouncing effect, we applied a composition-dependent collection efficiency (CE, the product of net particle transmission and detection efficiency) to the entire AMS dataset (Canagaratna et al., 2007; Middlebrook et al., 2012). In order to investigate the potential sources for organic aerosols, we have done positive matrix factorization (PMF) analysis for AMS high-resolution (HR) spectra of organic compounds, by using the PMF Evaluation Tool (PET version 3.00D) (Ulbrich et al., 2009). Detailed protocols were reported by Ulbrich et al. (2009) and (DeCarlo et al., 2010). In brief, the HR input matrices without isotopes were created using PIKA. A five-factor-PMF solution with FPEAK=0 was chosen according to the following criteria, including least residuals (unexplained data), $Q/Q_{\text{exp}} \approx 1$ (Q is the sum of the squares of the scaled residuals; Q_{exp} is the expected Q), and distinguishable features (e.g., spectra and diurnal patterns) of the components. It should be noted again that although fuzzy clustering and PMF analyses are commonly used methods in source apportionment, they cannot provide the information of aerosols origin and the interaction between aerosol transport and transformation processes, which can be studied with aerosol transport models.”

Section 3.1 Overview of the field observation (2nd and last paragraph)

“...Given the AMS-PMF results shown in Fig. S4, five OA factors (components) were determined with characteristic mass spectra and corresponding elemental ratios (O:C and H:C), and distinct temporal variations. There were (9 ± 7) % HOA, (16 ± 11) % SV-OOA, and (75 ± 15) % LV-OOA, i.e., (16 ± 15) % LV-OOA1, (37 ± 17) % LV-OOA2, and (21 ± 13) % LV-OOA3, contributing to the OA mass. The HOA and OOA are linked to primary and secondary organic aerosol (POA and SOA), respectively (Ulbrich et al., 2009). Thus, the SOA dominated the OA mass at this measurement site. Further descriptions of PMF results can be found in the supplement (Fig. S4) and more discussions are given in section 3.4.2 especially regarding the organic compounds.
...

... Furthermore, we discuss another episode with relatively high organic mass concentrations to find out how these mass loads could form and exist for a relatively long time period. Both episodes are easily recognized from the time series of particles measured by LAAPTOF and the organic compounds measured by AMS (Fig. S2 and Fig. 3; see also Fig. 4 in Shen et al., 2019), as well as the time series of OOA, in particular for SV-OOA, LV-OOA1 and LV-OOA2 (Fig. S4) ...”

Section 3.4.2 Organic compounds (1st paragraph)

“The dominating OA components in this episode were LV-OOA1 (42 ± 15) % and LV-OOA2 (32 ± 21) %, which are surrogates for more aged SOA (Zhang et al., 2011). It should be noted that LV-OOA1 has strong correlation with SV-OOA ($\gamma = 0.8$), which is a surrogate for local (less-aged) SOA (Zhang et al., 2011). As shown in Fig. S4 (the insert in panel d), they are both peaking at a similar time during the night with broader peaks for LV-OOA1. This indicates that the LV-OOA1 might have a similar origin as the SV-OOA (more related with local biogenic sources). However, it can also be influenced by regional transport. For example, the first LV-OOA1 peak has two

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sub-peaks: the first one appears during the night, e.g., related to NO₃ oxidation (strong correlation with $\gamma = 0.9$ between LV-OOA1 and CHO1N family has been found here, refer to their time series in Fig. S4 b); and the second sub-peak appears in the morning rush hour and corresponds to south and southwest wind directions, indicating the potential transport from the urban and industrial area of Karlsruhe. The more oxidized LV-OOA2 is peaking during day-time and shows a negative correlation with SV-OOA ($\gamma = -0.5$), possibly due to photochemical aging of SV-OOA. Therefore, we conclude that the SOA measured in this study was likely due to both local formation and regional transport.”

Section 4 Conclusion (2nd and 4th paragraphs)

“...AMS-PMF results suggest the organic aerosols (OA) consisted of (9 ± 7) % HOA, (16 ± 11) % SV-OOA, (75 ± 15) % LV-OOA with three sub-types (LV-OOA1, LV-OOA2, and LV-OOA3).

...

For the organics rich episode, the particulate organic compounds accounted for (77 ± 6) % of total non-refractory particle mass measured by the AMS. AMS-PMF analysis suggests the dominating OA components were LV-OOA1 and LV-OOA2, contributing (42 ± 15) % and (32 ± 21) % to the total OA mass in this episode. LV-OOA1 and LV-OOA2 had positive ($\gamma = 0.9$) and negative ($\gamma = -0.5$) correlation with SV-OOA, respectively, indicating their local formation. In order to investigate origin of the aerosols and their transport and transformation processes, we applied a regional transport model, COSMO-ART. ...”

Supplement

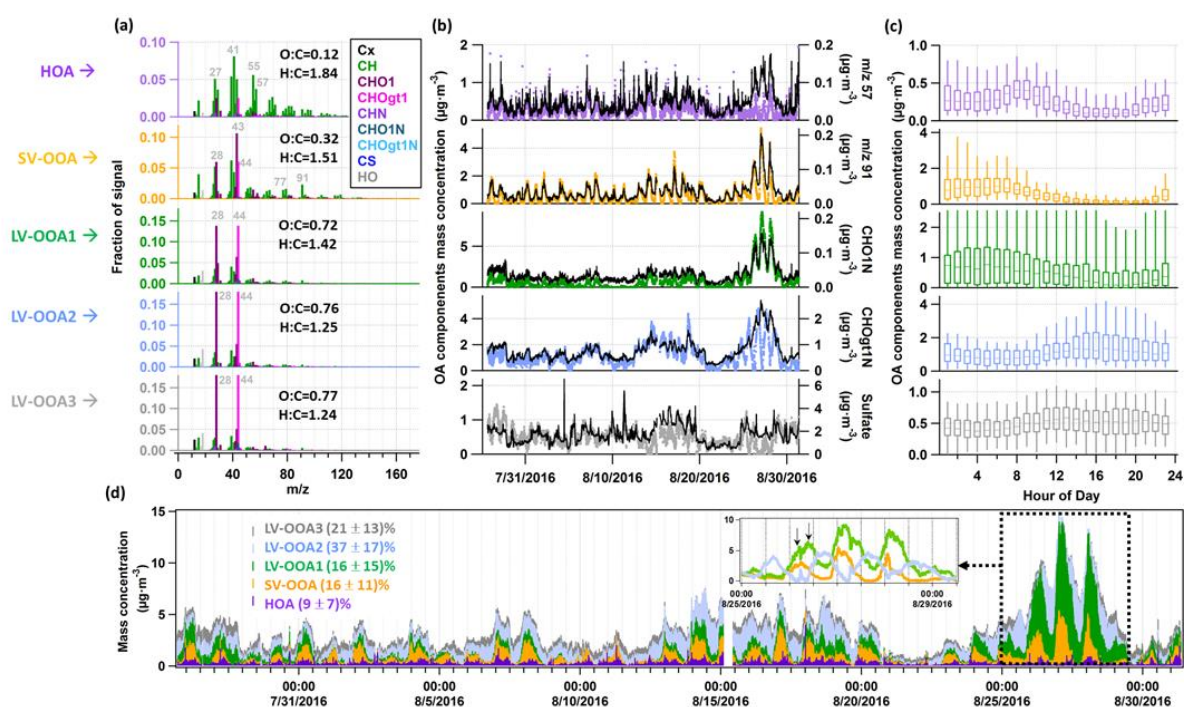


Figure S4: Five-factor-PMF solution with FPEAK=0 for organic compounds measured by AMS during the whole measurement time. (a) Mass spectra of the five components, i.e., HOA, SV-OOA, LV-OOA1, LV-OOA2, and LV-OOA3. (b) Time series of the five components mass concentrations (left Y-axis) and some related species (right Y-axis). (c) Diurnal patterns of the five components. (d) Time series of the components with the same data as panel but stacked (b). The percentages are their contributions to OA. Note that the total OA value showed here is lower than the input OA for the PMF analysis. This is because the components identification might be interfered by extraneous variability, arising from some causes such as instrumental issues. Such extraneous variability will lead to disproportionate effects on the fitting outcome. Therefore, we need to downweight the corresponding variables in a proper way or even remove them, resulting the lower output total OA value reconstructed by the components.

The criteria for choosing factor number are given in section 2.2 of main manuscript. As shown in the Fig. S4 (a) HOA, SV-OOA, and LV-OOA have distinguishable spectra. For three LV-OOA, in spite of some similarity, distinguishable features can still be found, according to their time series (b) and diurnal patterns (c). In particular for the organic rich episode, LV-OOA1 and LV-OOA2 show anti-correlation, the LV-OOA3 signal is much weaker than the other two LV-OOA. Therefore, a five-factor solution was chosen rather than four or three-factor solution.

HOA is associated with POA, e.g., from urban emissions. In this study, the HOA (high H:C = 1.84 and low O:C = 0.12, herein) has a similar trend as m/z 57 (major contribution from C₄H₉) and NO₂ (not shown), with Pearson's correlation coefficients $\gamma = 0.5$ and 0.4 , respectively. SV-OOA (O:C = 0.32) shows a very strong correlation with m/z 91 (mainly from C₇H₇) ($\gamma = 0.9$) and CH family ($\gamma = 0.8$). Three LV-OOA have prominent marker peaks at m/z 28 and 44, and all have high O:C ratio (> 0.7), which indicates highly oxidized or aged OA. LV-OOA1 shows strong correlations ($\gamma = 0.8$ to 1) with CH, CHO1, CHOgt1, CHN, CHO1N, and CHOgt1N. LV-OOA2 shows good correlations with CHOgt1N ($\gamma = 0.6$). LV-OOA3 shows no good correlation with any ion families, but a similar trend as sulfate ($\gamma = 0.5$).

Similar elemental ratios for HOA, SV-OOA, and LV-OOA and their relationships with tracers such as HOA with NO_x and LV-OOA with secondary inorganics, have also been reported in previous AMS-PMF studies. Readers may refer to a review paper by Zhang et al. (2011).

Reply to Co-Editor and Referees

7. Session 4 conclusion: More implication should be provided, because this work presents a method of combination of field measurement and model. Can this method be expanded to other places and time? Otherwise this work is only a very local study.

R: Yes, this combination method can be expanded to other places and time. We have added more implications in ours conclusion, as follows:

2nd last paragraph

“...Such field-model comparisons and corresponding sensitivity tests are useful for targeting missing sources and mechanisms in the current model and thus help its development. Such that model can be improved to become better predictive, which is useful for e.g., legislation and implementation of policies against air pollution. Consequently, model systems suitable for a reasonable environmental policy should be validated by comprehensive field studies with state-of-the-art aerosol analytics and meteorological parameters.”

Last paragraph

“...Our results show how close the physical and chemical nature of aerosols is related to emission sources, transformation processes, and complex transport pattern. The method of combination of filed measurement and model calculation in this study can be widely used in other places in the world, in order to investigate the origin of aerosols and their interaction between transport and transformation. For example, in some pollution hot spots like Po Valley, and some other places which are subject to both biogenic and anthropogenic emissions. Besides, it is also not limited to the summer time that we studied...”

Reply to Co-Editor and Referees

Referee #2 comments:

This paper describes a six-week ground-based study of aerosol chemistry in the Rhine Valley, near Karlsruhe. Both LAAPTOF and AMS were deployed and the field data is interpreted in conjunction with COSMO-ART model. This paper is well written, and it is a valuable contribution to ACP. While the paper is somewhat closely related to the author's previous paper, I think that there is enough new material to merit publication. I have some concerns about how AMS data is used, so I recommend some major revisions before publication.

Major comments:

1. There are some outstanding questions about source apportionment (for example, the biomass burning class identified by LAAPTOF). I think running PMF analysis on AMS data would help answer those questions and make the analysis more complete. There is certainly enough data to do so and it is an odd choice not to include such analysis.

R: Thank you for your suggestion. We have done PMF analysis for organics data measured by the AMS, added more explanation and discussion throughout the whole manuscript, and modified the conclusion as well. Please refer to our reply to the 1st Referee comment #6.

In addition, we couldn't identify biomass burning organic aerosols (BBOA) for the entire measurement period, based on AMS-PMF results. We have added this point in the last second paragraph of section 3.1, as follows:

“... It should be noted that, there were no significant biomass burning organic aerosols (BBOA) identified by PMF for the entire measurement period. The marker f_{60} (the mass fraction of m/z 60 to the total organics) of BBOA being very close to the background fraction of 0.003 (Crippa et al., 2014; Cubison et al., 2011). In addition to biomass burning, also other potential sources for potassium and sulfate rich particles detected by LAAPTOF must be taken into consideration, such as coal combustion (Xu et al., 2018). Therefore, further analysis is required to identify the different sources contributing to this particle class.”

2. To add on to the above, a unique feature of this study is that SPMS and AMS are deployed together. It would be nice to see a bit more synergy in data analysis as well. Currently, it seems like the details focus on SPMS. For example, for marine-influenced air masses, did high resolution AMS spectra show typical AMS marine markers, such as MSA?

R: Thank you for your suggestion. We have added more synergy in data analysis regarding the marine influenced air masses. In high resolution AMS spectra we have seen the peak at 78.99 CH_3SO_2^+ , which is reported to be the marker for methanesulfonic acid (MSA) (Huang et al., 2015). Based on the method developed by Huang et al. (2015), we have also estimated the MSA fraction and mass concentrations for the entire campaign and compare them with sodium salts measured by the LAAPTOF. We have added the corresponding discussion and figure are as follows:

Section 3.3 Aged sea salt in central Europe (last paragraph)

“Apart from that, in high resolution AMS spectra we have observed the peak at 78.99 CH_3SO_2^+ , which is the marker for methanesulfonic acid (MSA) from marine or coastal areas (Huang et al., 2015) although other sources cannot be excluded (Daellenbach et al., 2017). Based on the method developed by Huang et al. (2015), we have also estimated the MSA fraction (f_{MSA} , signal fraction of MSA to total org and sulfate) and mass concentrations

for the entire campaign (refer to Fig. S12). It turns out that during the sodium chloride rich episode, the f_{MSA} is enhanced compared to the other low sodium salts time period, such as organic rich episode. This also supports that sodium salts particles measured by the LAAPTOF had sea salt origin.”

Supplement

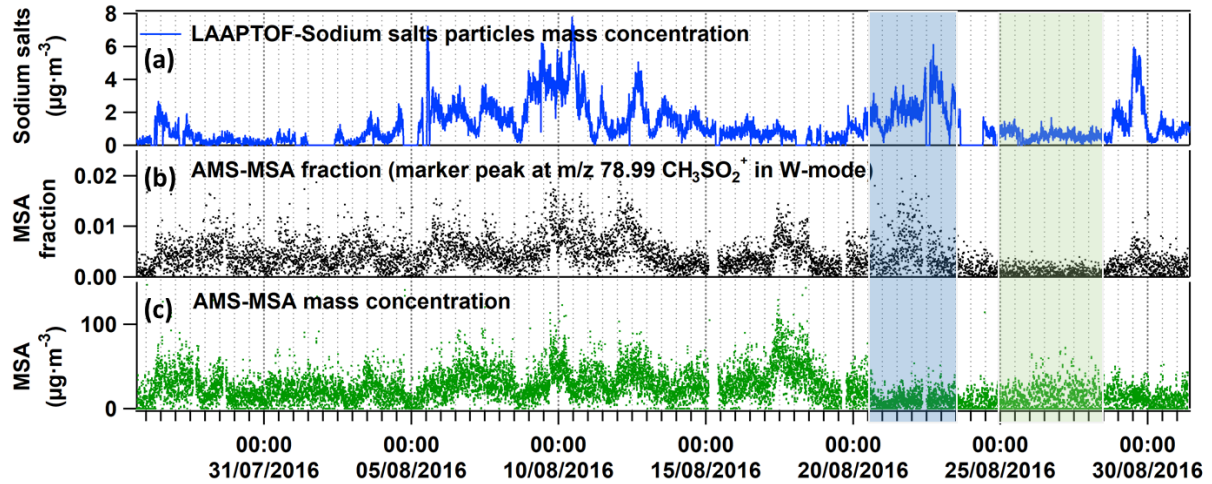


Figure S12: Time series of the estimated mass concentration of sodium salts particles measured by the LAAPTOF (a), methanesulfonic acid (MSA) fraction to total organics and sulfate measured by the AMS (b) and MSA mass concentration (c). The first (blue) and second (green) shaded areas mark the sodium chloride rich and organic rich episodes, respectively.

The method used to estimate MSA fraction (f_{MSA} , signal fraction of MSA to total org and sulfate) and its mass concentration can be referred to Huang et al. (2015). In brief, W-mode f_{MSA} is calculated and then used to multiply V-mode mass concentration of organics and sulfate, resulting estimated MSA mass concentration. The corresponding equations are as follows:

$$\sum I_{\text{MSA}} = \frac{I_{\text{CH}_3\text{SO}_2}}{9.7\%} \quad (\text{S2})$$

$$f_{\text{MSA}} = \frac{\frac{\sum I_{\text{MSA}}}{\text{RIE}_{\text{MSA}}}}{\frac{\sum I_{\text{Org}}}{\text{RIE}_{\text{Org}}} + \frac{\sum I_{\text{SO}_4}}{\text{RIE}_{\text{SO}_4}}} \quad (\text{S3})$$

$$C_{\text{MSA}} = f_{\text{MSA}} \times (C_{\text{Org}} + C_{\text{SO}_4}) \quad (\text{S4})$$

where $\sum I_{\text{MSA}}$ is the total signal intensity of all MSA fragments measured in W-mode, which can be estimated based on the fragmentation patterns determined from the laboratory experiments (Huang et al., 2015). It was reported that the signal intensity of marker peak at m/z 78.99 (CH_3SO_2^+) accounted for $(9.7 \pm 1.6) \%$ of the total signal intensity of all MSA fragments. $\sum I_{\text{Org}}$ and $\sum I_{\text{SO}_4}$ are total signal intensity of organics and sulfate fragments measured in W-mode, respectively. Relative ion efficiency of MSA ($\text{RIE}_{\text{MSA}}=1.3$) is the averaged value from $\text{RIE}_{\text{Org}} (=1.4)$ and $\text{RIE}_{\text{SO}_4} (=1.2)$. C_{Org} and C_{SO_4} are mass concentrations calculated from V-mode data. C_{MSA} is the estimated MSA mass concentration.

Reply to Co-Editor and Referees

Minor comments:

3. Page 4, line 118: “: : deployment of a laser ablation aerosol particles time-of-flight mass spectrometer (LAAPTOF; AeroMegt GmbH)” Pretty sure this should be “particle” instead of “particles”

R: Yes, we have revised it.

4. Page 5, line 138: Acronym TSP shows up for the first time and has not been defined.

R: We have replace Acronym TSP with the full name “total suspended particulates”

5. Page 6, line 180: “behaved anti-correlated” is an odd turn of the phrase. Would revise to “were anti-correlated”

R: We have revised it to “were anti-correlated”

6. Page 6, line 182: should this read “: : for particles larger than 2.5 μm measured in this study: :” instead of 2.5 nm?

R: This particle number concentration was measured by CPC3776 (TSI Inc.) which has the cut off at 2.5 nm rather than 2.5 μm . We have added this point to make it clear, as follows:

“The averaged particle number concentration of $(1.2 \pm 0.8) \times 10^4 \text{ cm}^{-3}$ for particles larger than 2.5 nm measured by CPC3776 in this study”

7. For Figure 2 and associated discussion, are these averages over the entire campaign?

R: Yes. We have added “the data is averaged over the entire campaign” in the figure caption and discussion part (the third paragraph of section 3.1).

8. Page 8, line 246: averaged -> average

R: We have revised it.

9. Page 8, line 248: organics -> organic

R: We have revised it.

10. Figure 4 caption: extend -> extent

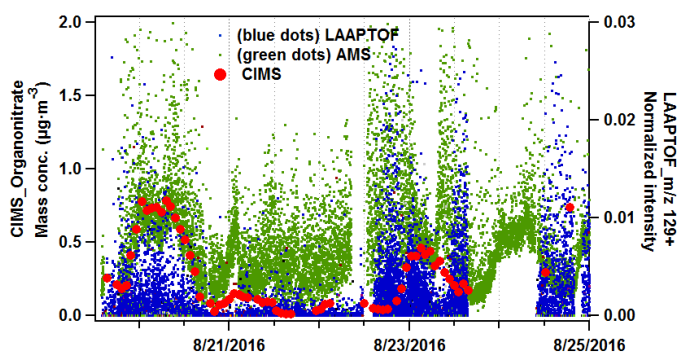
R: We have revised it

11. In Figure S9, it is hard to tell visually that LAAPTOF m/z 129 and AMS organonitrate are correlated. There are quite a few instances where they are not. Can you plot these against each other and give the R2?

R: We agree that it is hard to tell they are correlated. Due to complex laser desorption and ionization (LDI) mechanisms, the resulting ion fragments usually do not have good correlation or direct clear relationship with the results from quantitative instruments such as AMS. This is also the case for LAAPTOF m/z 129 and AMS organonitrate. Specific relationships of LAAPTOF ion intensities and AMS mass concentrations for non-refractory compounds were found based on specific calculation, such as the fraction of org/(org+nitrate), reported by our previous study (Shen et al., 2019). However, in this study, we aimed to confirm that the m/z 129 $\text{C}_5\text{H}_7\text{NO}_3^+$ signal is mainly due to organonitrates rather than the recombination of nitrate and organic fragments due to the LDI matrix effect in single-particle MS. First, we found that during sodium chloride rich episode, organic mass

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concentration measured by the AMS was relatively low. However, at the end of this episode m/z 129 $C_5H_7NO_3^+$ relative intensity was obviously strong, which was comparable with that observed in high organic episode. Furthermore, we referred to AMS results as well as CIMS results (Huang et al., 2019), both of which show high organonitrate mass concentration at the similar time period as LAAPTOF_ m/z 129 shows high relative intensity (as shown in the figure below). Therefore, we can conclude that m/z 129 $C_5H_7NO_3^+$ was mainly contributed by organonitrate.



We have added more explanation in the last paragraph of section 3.3. Please refer to our reply to the co-editor's comment #9

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Best regards,

Xiaoli Shen and co-authors

Composition and origin of PM_{2.5} aerosol particles in the upper Rhine valley in summer

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Abstract. We conducted a six-week measurement campaign in summer 2016 at a rural site about 11 km north of the city of
15 Karlsruhe in southwest Germany in order to study the chemical composition and origin of aerosols in the upper Rhine valley. In particular, we deployed a single particle mass spectrometer (LAAPTOF) and an aerosol mass spectrometer (AMS) to provide complementary chemical information of the aerosol particles smaller than 2.5 µm. For the entire measurement period, the total aerosol particle mass was dominated by sodium salts contributing on average $(36 \pm 27\%)$ % to the total single particles
20 measured by the LAAPTOF. The total particulate organic compounds, sulfate, nitrate, and ammonium contributed on average $(58 \pm 12\%)$ %, $(22 \pm 7\%)$ %, $(10 \pm 1\%)$ %, and $(9 \pm 3\%)$ % to the total non-refractory particle mass, ~~respectively.~~ measured by the AMS. Positive matrix factorization (PMF) analysis for the AMS data suggests that the total organic aerosol (OA) consisted of five components, including (9 ± 7) % hydrocarbon-like OA (HOA), (16 ± 11) % semi-volatile oxygenated OA (SV-OOA), and (75 ± 15) % low-volatility oxygenated OA (LV-OOA). The regional transport model COSMO-ART was applied for source apportionment and to achieve a better understanding of the impact of complex transport pattern on the field
25 observations. Combining field observations and model simulations, we attributed high particle numbers and SO₂ concentrations observed at this rural site to industrial emissions from power plants and a refinery in Karlsruhe. In addition, two characteristic episodes with aerosol particle mass dominated by sodium salts particles comprising $(70 \pm 24\%)$ % of the total single particles and organic compounds comprising accounting for $(77 \pm 6\%)$ % of total non-refractory species, respectively, were investigated in detail. For the first episode, we identified relatively fresh and aged sea salt particles originating from the Atlantic Ocean more
30 than 800 km away. These particles showed markers like m/z 129 C₅H₇NO₃⁺ indicating the influence of anthropogenic emissions modifying their composition e.g., from chloride to nitrate salts during the long-range transport. For a 3-day episode including high organic mass concentrations, model simulations show that on average $(74 \pm 7\%)$ % of the particulate organics at this site were of biogenic origin. Detailed model analysis allowed us to find out that three subsequent peaks of high organic mass

35 concentrations originated from different sources, including local emissions from the city and industrial area of Karlsruhe, regional transport from the city of Stuttgart (~64 km away), and potential local night-time formation and growths. Biogenic (forest) and anthropogenic (urban) emissions were mixed during transport and contributed to the formation of organic particles. In addition, topography, temperature inversion, and stagnant meteorological conditions also played a role in the build-up of higher organic particle mass concentrations. Furthermore, the model was evaluated using the field observations and corresponding sensitivity tests. The model results show good agreement with trends and concentrations observed for several 40 trace gases (e.g., O₃, NO₂, and SO₂) and aerosol particle compounds (e.g., ammonium and nitrate). However, the model underestimates the number of particles by an order of magnitude and underestimates the mass of organic particles by a factor of 2.3. The discrepancy was expected for particle number since the model does not include all nucleation processes. The missing organic mass indicates either an underestimated regional background, missing sources, and/or mechanisms in the model like night-time chemistry. This study demonstrates the potential of combining comprehensive field observations with dedicated 45 transport modelling to understand the chemical composition and complex origin of aerosols.

1 Introduction

Aerosols are ubiquitous in the atmosphere and play a crucial role in air quality, human health, ecosystem and climate change. In particular, aerosol particles modify atmospheric radiative fluxes and can serve as cloud condensation nuclei (CCN) and/or be activated as ice nuclei (IN), which influences the cloud formation, properties (albedo) and precipitation, thus has further 50 impact on weather and climate change (Fuzzi et al., 2015; Pöschl, 2005; Seinfeld and Pandis, 2006). Such indirect impacts of aerosols on climate are still related to large uncertainties and low confidence levels (IPCC, 2013). There are various types of atmospheric aerosol particles, such as mineral dust, sea salt, soot, and organic aerosol particles, originating from biogenic and/or anthropogenic emission sources, as well as secondary formation from precursors within the atmosphere (Tsigaridis et al., 2006). Aerosol particles are not typically purely inorganic or organic; instead, they are often composed of different inorganic and 55 organic species. The diverse origins and atmospheric aging cause the complexity of the aerosol particle mixtures.

Real-time mass spectrometry is widely used to measure size and chemical composition of aerosol particles with high temporal resolution. Two major instrument types are single particle mass spectrometers (single-particle MS) and aerosol mass spectrometers (AMS). The single-particle MS is sensitive to both non-refractory and refractory species and to the mixing state of individual particles, and thus identifies particle types (Murphy, 2007), while the AMS can quantify mass concentrations of 60 non-refractory particulate compounds in bulk samples (Canagaratna et al., 2007). Complementary data from single-particle MS and AMS allows quantitative interpretation of the internal and external mixing state of the aerosol (Gemayel et al., 2017; Gansch et al., 2018; Healy et al., 2013; Jeong et al., 2011; Shen et al., 2019; Zhou et al., 2016). The mixing state of the aerosol particles can reflect their evolution e.g. caused by chemical reactions leading to changing compositions or coatings (Reitz et al., 2016). A good example is the aging of sea salt particles: chloride replacement by nitrate on sea-salt particles can be well identified by 65 single-particle MS (Gard et al., 1998).

The large diversity of aerosols, containing hundreds of gaseous and particulate components, makes it difficult to trace their sources. Therefore, several source apportionment methods have been developed with the common aim of reducing the complexity of the aerosol particle mixtures and thus helping to identify the major particle types in the atmosphere. Clustering algorithms have been used commonly for the single-particle MS data analysis (Reitz et al., 2016), while positive matrix factorization (PMF; Paatero, 1997; Paatero and Tapper, 1994) is commonly applied for AMS data analysis (Canonaco et al., 2013; Canonaco et al., 2015; Crippa et al., 2014; [DeCarlo et al., 2010](#); [Mohr et al., 2012](#); [Ulbrich et al., 2009](#); Zhang et al., 2011). According to PMF analysis, the organic aerosol (OA) can be separated into e.g., hydrocarbon-like OA (HOA), cooking-related OA (COA), nitrogen-enriched OA (NOA), biomass burning OA (BBOA), and semi-volatile oxygenated OA (SV-OOA), and low-volatility oxygenated OA (LV-OOA). This information can give valuable hints to OA aging and potential sources.

However, these methods can only provide indirect source information. The application of a transport model gives additional information. It can predict the state of the atmosphere based on detailed meteorology, emission data, as well as conversions of reactive species and thus allows studying the aerosol evolution from the emissions to a site of interest.

There exists a variety of so-called online coupled chemical transport models. A recent review is given by [\(Baklanov et al., 2014\)](#). Such models are important tools to better understand aerosol particles' sources and aging processes, and their impact on particles' climate-relevant properties, such as cloud condensation nuclei (CCN) activity, optical properties, and hygroscopicity (Fierce et al., 2017). Models can be improved with a good process understanding, e.g., emissions, transport and deposition, and corresponding parameterisations obtained from dedicated laboratory studies (Burkholder et al., 2017). They are often used to predict the evolution of the atmospheric composition on different spatial scales, which can be evaluated by comparison with field measurements. Besides, comparisons between model simulations and field observations also help to validate the model or to identify weaknesses like poorly represented chemistry, missing sources or mechanisms.

For comparison with measurements at a single location, regional scale models with small grid sizes (down to a few kilometres) are required. In addition, phenomena such as mountain and valley winds, land-sea breezes or lee waves become important, which requires a non-hydrostatic formulation of the model equations. In this study we use the regional-scale model system COSMO-ART (Consortium for small-scale modelling - aerosols and reactive trace gases; Vogel et al., 2009). The part of the model describing aerosols and reactive trace gases was developed at the Karlsruhe Institute of Technology (<http://www.imk-tro.kit.edu/english/3509.php>). It extends the operational weather forecast model COSMO of the German Weather Service (DWD), and it allows studying the feedback processes between aerosol particles and the state of the atmosphere covering the continental to the regional scale.

Mobile particle measurements on a tram line intersecting the city of Karlsruhe in southwest Germany and in the upper Rhine valley, showed particle number concentrations in the rural area north of the city almost as high as in downtown (particle size > 4 nm; mean value = $1.2 \times 10^4 \text{ cm}^{-3}$, maximum values $3 \times 10^4 \text{ cm}^{-3}$) (Hagemann et al., 2014). The Rhine valley provides a well-known wind channel of large-scale flow. This results in faster transport of pollutants along the valley axis, e.g., NO_x , SO_2 , volatile organic compounds (VOCs) and particles (Gross and Wippermann, 1987; Hagemann et al., 2014; Kalthoff and Vogel, 1992; Vogel et al., 1986). According to data from the local air quality monitoring administration (LUBW,

100 <https://www.lubw.baden-wuerttemberg.de/luft>) from 2016 to 2018, the annual averaged mass concentrations of NO₂, SO₂, PM₁₀, and soot are 20, 2, 16, and 1.8 μg m⁻³, respectively. They were measured at Eggenstein monitoring station (3.3 km south of the measurement site used in this study and about 9 km north of the city of Karlsruhe). Specifically, NO₂ concentrations are about half of the values measured in downtown Karlsruhe, but they exhibit comparable peak values (around 50–80 μg m⁻³). SO₂ levels at Eggenstein are the second highest in the state of Baden-Württemberg (from the available data from 2016 to 2018), and maximum hourly values can reach up to ~50 μg m⁻³.

In order to understand the sources for high local particle concentrations as well as the chemical nature and mass concentrations of aerosol types prevailing in the upper Rhine valley in summer-time we combine comprehensive aerosol information from point measurements with a regional transport model. Besides a better understanding of the origin of the aerosol in that region, we want to validate the capabilities of our transport model to predict characteristic air quality parameters.

110 2 Methods

2.1 Measurement site and instrumentation

The measurement site (49°6'10.54" N, 8°24'26.07" E) is in the upper Rhine valley 175 m north of the village of Leopoldshafen and 10 m east of the tramline No S1/11 described by Hagemann et al., (2014) (marked with a star in Fig. 1). The Rhine valley often acts as a wind channel leading to wind directions mainly from southwest or northeast. As shown in Fig. 1, the measurement site is located ~11 km north of downtown Karlsruhe and ~5 to 15 km northeast of an industrial area, including a refinery with 15.5 Mt/yr capacity, a 1450 MW hard coal fired power plant, as well as a 365 MW gas fired power plant (Hagemann et al., 2014). Therefore, the measurement site is potentially subject to the urban and industrial emissions especially for southerly wind directions. Apart from that, there are several nearby biogenic emission sources, e.g., forests, sunflower fields, as well as other agricultural areas (Hagemann et al., 2014; Huang et al., 2019). Local traffic emissions from the minor road near the measurement site was identified by measurements with high time resolution. Its contributions to the aerosol load are small due to the generally weak traffic.

The instrumentation for this campaign has been described in detail in previous publications (Huang et al., 2019; Shen et al., 2019). Briefly, the data we use here are from the 6-week (26 July to 31 August 2016) deployment of a laser ablation aerosol ~~partielesparticle~~ time-of-flight mass spectrometer (LAAPTOF; AeroMegt GmbH) and a high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS; Aerodyne Inc.; hereafter AMS). These instruments were used to provide real time information on size and chemical composition for individual particles and bulk samples, respectively. The LAAPTOF is a commercially available single particle mass spectrometer and has been well described in recent publications (~~Ahern et al., 2016;~~ Gemayel et al., 2016; ~~Gemayel et al., 2017;~~ Marsden et al., 2018; ~~Marsden et al., 2016;~~ ~~Marsden et al., 2019;~~ Marsden et al., 2016; Ramisetty et al., 2018; ~~Reitz et al., 2016;~~ Shen et al., 2018; ~~Shen et al., 2019~~). In brief, aerosol particles with a size range of 70 nm to 2.5 μm vacuum aerodynamic diameter (d_{va}) are sampled with a flowrate of ~80 cm³ min⁻¹, focused and accelerated via an

aerodynamic lens (ADL). Afterwards, the particles of ~ 200 nm to $2.5 \mu\text{m}$ d_{va} can be detected with two laser diodes ($\lambda = 405$ nm) mounted in the particle time-of-flight (PTOF) chamber. Once a single particle is detected successively by both of the detection lasers, its d_{va} is calculated from the time of flight and an excimer laser ($\lambda = 193$ nm) is triggered to desorb and ionize the refractory and non-refractory species of the particle. The resulting cations and anions are analysed with a bipolar ToF mass spectrometer. The AMS measures the particle size-resolved bulk chemical composition, e.g., organic compounds, ammonium, nitrate, sulfate, and chloride, in mass concentrations with high time resolution and sensitivity (DeCarlo et al., 2006). Similarly as the LAAPTOF, aerosols are sampled with a flowrate of $\sim 84 \text{ cm}^3 \text{ min}^{-1}$ via a similar ADL (Peck et al., 2016) in the size range 70 nm to $2.5 \mu\text{m}$ d_{va} and then pass through a PTOF chamber. At the end of the PTOF chamber, particles encounter a 600°C heater that vaporises the non-refractory species. The corresponding vapours are ionized by electron impact with high energy of 70 eV, generating positive ions that are analysed with a ToF mass spectrometer. ~~To account for a particle bouncing effect, a collection efficiency (CE, the product of net particle transmission and detection efficiency) is used (Canagaratna et al., 2007). The CE depends on composition and phase of the particles (Bahreini et al., 2005). Therefore, composition dependent CEs were used to correct the AMS data for this measurement campaign.~~

All sampling inlets (~~PM₁₀~~, ~~PM_{2.5}~~, ~~TSPPM₁₀~~, and total suspended particulates) were positioned at the top of a measurement container (3.7 m above ground level, a.g.l.). Particle number (size >2.5 and >7 nm, CPC3776 and CPC 3022A, TSI Inc.), particle mass (~~PM_{2.5}~~, ~~PM₁₀~~, and total suspended particulates, ~~PM₁₀~~, and ~~PM_{2.5}~~; FIDAS, PALAS GmbH), black carbon (AE33-7 Aethalometer, Magee Scientific Co.), and trace gases O₃ (O341M, Environment SA), SO₂ (AF22M, Environment SA), NO₂ (AS32M, Environment SA) CO₂ (NGA2000, Rosemont Inc.) were measured. Meteorological parameters temperature, relative humidity, solar radiation, wind direction and speed (WS700, LufftLuft GmbH) were measured 10 m a.g.l. and complemented by meteorological data from a 200 m tower located at the KIT campus north, ~ 2 km east of the measurement site ($49^\circ 5' 43.0''\text{N}$ $8^\circ 25' 46.0''\text{E}$).

2.22.2 Mass spectrometer data analysis

~~Single particle data were analysed by using the Igor LAAPTOF data analysis software (version 1.0.2, AeroMegt GmbH). The corresponding basic procedures (e.g., mass calibration and fuzzy c-means clustering) and particle mass quantification method have been described in detail in our previous studies (Shen et al., 2018 and 2019). The AMS data analysis was done with the standard software written for Igor, including SQUIRREL (version 1.60C) and PIKA (version 1.20C). To account for a particle bouncing effect, we applied a composition-dependent collection efficiency (CE, the product of net particle transmission and detection efficiency) to the entire AMS dataset (Canagaratna et al., 2007; Middlebrook et al., 2012). In order to investigate the potential sources for organic aerosols, we have done positive matrix factorization (PMF) analysis for AMS high-resolution (HR) spectra of organic compounds, by using the PMF Evaluation Tool (PET version 3.00D) (Ulbrich et al., 2009). Detailed protocols were reported by Ulbrich et al. (2009) and (DeCarlo et al., 2010). In brief, the HR input matrices without isotopes were created using PIKA. A five-factor-PMF solution with FPEAK=0 was chosen according to the following criteria, including least residuals (unexplained data), $Q/Q_{\text{exp}} \approx 1$ (quality-of-fit parameter Q is the sum of the squares of the scaled residuals; Q_{exp} is the expected~~

165 Q), and distinguishable features (e.g., spectra and diurnal patterns) of the components. It should be noted again that although fuzzy clustering and PMF analyses are commonly used methods in source apportionment, they cannot provide the information of aerosols origin and the interaction between aerosol transport and transformation processes, which can be studied with aerosol transport models.

2.3 Regional transport model COSMO-ART

170 The regional transport model COSMO-ART is used in this study to simulate an episode with high mass concentration of organics (up to $\sim 25 \mu\text{g m}^{-3}$ measured by AMS). Detailed model descriptions can be found in previous studies (Bangert et al., 2011; Knotte et al., 2011; Stanelle et al., 2010; Vogel et al., 2009). Therefore, we will only give a brief review here. COSMO-ART is a fully online coupled model system, which uses the same spatial and temporal resolution for all scalars, such as temperature, gas and aerosol particle concentrations. The meteorological module is based on the operational weather forecast model COSMO of the German weather service (Deutscher Wetterdienst, DWD). When atmospheric chemistry and aerosol dynamics are coupled to
175 COSMO, it allows the online calculation of the transport of reactive trace gases and aerosol particles as well as their interaction with the state of the atmosphere. Physical processes, e.g., nucleation, coagulation, condensation and sedimentation are included in the model. The aerosol particles are represented by five interactive lognormal modes representing particles consisting of internally mixed sulfate, ammonium, nitrate and organic compounds, pure soot, as well as internal mixtures of them all. The hourly biogenic emissions are calculated online according to the land use data, the simulated temperatures, and radiative fluxes
180 (Vogel et al., 1995). The anthropogenic emissions, including traffic emissions, large point and area sources such as industrial and households' areas, were pre-calculated by J. Bieser at Helmholtz Zentrum Geesthacht (Backes et al., 2016; Bieser et al., 2011). In our study, the simulation domain is covered by a 350×330 grid with a horizontal resolution of $1 \times 1 \text{ km}^2$ and 50 vertical atmospheric layers (nonlinear increase up to 22 km a.g.l.). Most of the simulation results shown in this paper correspond to the lowest atmospheric level (10 m a.g.l.) and we indicate if results are from higher levels. The hourly emissions of SO_2 , NO_x , CO, NH_3 , anthropogenic volatile organic compounds (VOCs), and primary organic aerosols (POA) were pre-calculated. As shown in Fig S1, VOCs, NO_x and POA are mainly emitted at a lower atmospheric level in urban areas, such as the cities of Karlsruhe, Stuttgart, Mannheim, and Strasburg, as well as the corresponding highways. SO_2 hot spots typically correspond to the stack heights of the power plants emitting, such as the one in the western part of Karlsruhe.

3 Results and Discussion

190 In the first section- (3.1), we give an overview of the meteorological conditions, the trace gases, and particles measured during the whole measurement campaign and will then give an example for source identification using the transport model- in section 3.2. In the following two sections- (3.3 and 3.4), we discuss two selected episodes in more detail, which are dominated by aged sea salt particles and relatively high organic particle mass concentrations, respectively.

3.1 Overview of the field observation

195 Table 1 and Fig. S2 give an overview of meteorological conditions, trace gases, and particle concentrations for the entire measurement campaign. In summer 2016, the major wind direction was southwest (174 ± 75)° and an average wind speed was (1.3 ± 0.9) m s⁻¹ at the measurement site in the upper Rhine valley. This wind speed allows trace gases and particles to be transported over a distance of 10 km within 2 hours e.g. from the city of Karlsruhe to the measurement site. During the measurement time the sunrise and sunset were at around 06:30 and 20:30, respectively. Temperature and relative humidity (RH) were showing relatively stable anti-correlated diurnal patterns. Given the frequent clear nights, night-time temperature inversions occurred for 32 out of 37 nights during the entire measurement period (Fig. S3). Among the trace gases, CO₂ and O₃ were showing clear diurnal patterns with ozone peaking in the afternoon. NO₂ concentrations were higher in the morning hours and in the late evenings and generally behaved were anti-correlated to O₃. NO₂ and particle number were showing spikes at similar times (more frequently during the day-time), mainly due to the local emissions from nearby road traffic. The averaged particle number concentration of $(1.2 \pm 0.8) \times 10^4$ cm⁻³ for particles larger than 2.5 nm measured by CPC3776 in this study, is comparable with that observed at the same site in previous tram-based measurements (Hagemann et al., 2014). SO₂ concentrations showed peaks with low frequency (up to >10 ppb) but only when the wind was coming from the direction of the industrial area in the southwest. As shown in Fig. S2, these SO₂ peaks correspond to higher particle number concentrations at around noontime, when the wind speed is higher (~ 1.7 m s⁻¹) and vertical mixing intensified. This is in agreement with downward mixing of stack emissions from industrial sources, which can be validated by model calculations using COSMO-ART (refer to section 3.2).

200 In the entire measurement period, the total organic compounds, sulfate, nitrate, and ammonium were analysed to contribute on average $(58 \pm 12)\%$, $(22 \pm 7)\%$, $(10 \pm 1)\%$, and $(9 \pm 3)\%$ to the total non-refractory PM_{2.5} mass measured by AMS, respectively. Given the AMS-PMF results shown in Fig. S4, five OA factors (components) were determined with characteristic mass spectra and corresponding elemental ratios (O:C and H:C), and distinct temporal variations. There were $(9 \pm 7)\%$ HOA, $(16 \pm 11)\%$ SV-OOA, and $(75 \pm 15)\%$ LV-OOA, i.e., $(16 \pm 15)\%$ LV-OOA1, $(37 \pm 17)\%$ LV-OOA2, and $(21 \pm 13)\%$ LV-OOA3, contributing to the OA mass. The HOA and OOA are linked to primary and secondary organic aerosol (POA and SOA), respectively (Ulbrich et al., 2009). Thus, the SOA dominated the OA mass at this measurement site. Further descriptions of PMF results can be found in the supplement (Fig. S4) and more discussions are given in section 3.4.2 especially regarding the organic compounds.

215 In a previous publication (Shen et al., 2019), we could identify seven major particle classes at the measurement site among $\sim 3.7 \times 10^5$ single particles with sizes between 200 nm and 2.5 μm d_{va}: Calcium soil (class 1, C1); Aged soot (C2); Sodium salts (C3); Secondary inorganic amine (C4); Biomass burning soil (C5); Biomass burning organosulfate (C6); and Mixed/aged dust (C7). In brief, Class 1 are calcium-rich and soil-dust-like particles (calcium-soil) with marker peaks at m/z 40 Ca⁺, 56 CaO⁺, 57 CaOH⁺, 75 CaCl⁺, 96 Ca₂O⁺, and 112 (CaO)₂⁺. This particle class shows a good correlation (Pearson's correlation coefficient

$\gamma \geq 0.6$) with the laboratory derived reference spectra of soil dust particles. Class 2 are aged-soot-like particles (aged soot) with characteristic C_n^+ patterns and are mixed with sulfate and nitrate. Class 3 are sodium-salt-like particles (sodium salts) with marker peaks at m/z 23 Na^+ , 81/83 Na_2Cl^+ , 139/141 $Na_3Cl_2^+$, 165 $Na_3SO_4^+$, 35/37 Cl^- , 93/95 $NaCl_2^-$, 115 $Na(NO_2)_2^-$, 131 $NaNO_2NO_3^-$, and 147 $Na(NO_3)_2^-$. Class 4 are secondary-inorganics-rich and amine-containing particles (Secondary inorganic-amine) with prominent ammonium, nitrate and sulfate markers, as well as amine marker peaks at m/z 58 $C_2H_5NHCH_2^+$, 59 $(CH_3)_3N^+$, 86 $(C_2H_5)_2NCH_2^+$, 88 $(C_2H_5)_2NO/C_3H_6NO_2^+$, and 118 $(C_2H_5)_2NCH_2^+$. Class 5, aged-biomass-burning and soil-dust like particles (biomass burning-soil), has aged biomass burning marker peaks at m/z 39 K^+ , 213 $K_3SO_4^+$, 46 NO_2^- , 62 NO_3^- , 97 HSO_4^- , and peaks related to aromatic compounds (50 $C_4H_2^+$, 63 $C_5H_3^+$, 77 $C_6H_5^+$, 85 C_7H^+ , 91 $C_7H_7^+$, 95 $C_7H_{11}^+$, 104 $C_8H_8^+$, 115 $C_9H_7^+$). Furthermore, class 5 shows a strong correlation with reference spectra of soil dust particles. The representative spectra of class 6, aged-biomass-burning and organosulfate-containing particles (biomass burning-organosulfate), also features aged biomass burning particles and has organosulfates marker peaks at m/z 141 $C_2H_5OSO_4^-$, 155 $C_2H_3O_2SO_4^-$, and 215 $C_5H_{11}OSO_4^-$. Class 7 particles contain all the marker peaks from the other six classes and show a good correlation with the reference spectra of mineral and soil dust particles, and are consequently labelled as “mixed/aged dust” (short for mixed/aged and dust-like particles). These particle types were not always present in similar fractions but showed distinct periods with one or two types dominating the particle mass (Shen et al., 2019). Here, we will mainly discuss two particle classes that dominated the total particle mass, i.e., sodium salts and biomass burning-soil particles, contributing on average $(36 \pm 27\%)$ % and $(30 \pm 18\%)$ % to the $PM_{2.5}$ mass measured by the LAAPTOF, respectively. We will also discuss one characteristic particle class, i.e., secondary inorganic-amine particles, which showed relatively intensive and distinct peaks three times during this campaign (Fig. S2) as well as several weaker peaks.

As shown in Fig. 2 (a), total particle mass concentrations have clear diurnal pattern, (the data is averaged over the entire campaign), showing higher night-time values and a peak during the morning rush hour (~9:00 local time). Higher particle mass values mainly correspond to southern wind directions, with one exception of extremely high masses associated with northwesterly/north-westerly winds (Fig. 2c). Specifically, sodium salt containing particles have higher fractions during day-time and correspond to southwesterly/south-westerly wind directions (Fig. 2b). Such wind was observed mainly during day-time and with higher wind speeds than at night. Biomass burning-soil particles enrich during night-time and mainly correspond to southeast wind (Fig. 2d). Secondary inorganic-amine particles typically rise in the morning and correspond to south and southwest wind directions (Fig. 2). When we compare weekdays (Monday to Friday) and weekends (Saturday and Sunday) in Fig. S4S5, the sodium salt particles were always correlated with southwest wind directions. Obvious differences can be found for secondary inorganic-amine and biomass burning-soil, respectively. For secondary inorganic-amine type particles, they were mainly advected by south and southwest wind during weekdays. However, they were much less abundant during weekends and did not show a clear relationship with wind directions (only a small peak for northwest). For biomass burning-soil particles the observations are a bit more complex.

The biomass burning-soil particles measured during weekdays were correlated with south and southwest wind as well, indicating potential sources from the city of Karlsruhe and its industrial area. Please note, that the representative spectra of these

particles show aromatic ion peaks which may arise from aromatic compounds such as toluene and xylene due to motor vehicle emissions. Besides, this type of particles contains also some metals such as barium and lead that are related to coal combustion from industry- (Hudson et al., 2004; Xu et al., 2018). These features correspond to urban and industrial sources. During the weekends, biomass burning–soil particles had higher mass concentrations than during weekdays and were correlated with wind directions from east, southeast and northwest, where are mainly forests and agricultural areas. Therefore, we can infer that there were various sources for biomass burning–soil particles and these sources differed between weekdays and weekends. This could be e.g., barbeque fires during weekends while the weekdays’ source is likely to be associated with that for secondary inorganic–amine particles.

In order to refine this type of particles, we compared them with black carbon particles measured by an Aethalometer (AE33-7). The equivalent black carbon concentration was calculated by the mass absorption cross sections value of $11.7 \text{ m}^2 \text{ g}^{-1}$ (Herich et al., 2011). They seem to show a similar trend in the time series with a triplet of peaks from August 25th to 28th (refer to Fig. S2 and Fig. S5S6). We found that biomass burning-soil and black carbon concentrations are well correlated only for the weekends, with Pearson’s correlation coefficients $\gamma \geq 0.6$ (Fig. S5S6 b), and poorly correlated for weekdays ($\gamma < 0.3$). This points to different major sources for weekdays and weekends. In addition, the black carbon instrument measuring the light absorption at different wavelengths (λ) can be used to determine the absorption Ångström exponent (α) (Drinovec et al., 2015). For the entire campaign, the average α was 0.96 ± 0.24 , calculated by using a pair of wavelengths and their corresponding mass absorption cross sections ($\text{MAC} = 11.7$ and $23.6 \text{ m}^2 \text{ g}^{-1}$ for $\lambda = 950$ and 470 nm , respectively, Herich et al., 2011). Herich et al. (2011) and Zotter et al. (2017) have assigned an $\alpha = \sim 0.9$ to BC from road traffic emissions, while the α values for wood burning BC were typically higher ~ 1.68 (Zotter et al., 2017). Our measurements indicate that traffic was the dominating source of BC. Considering the good correlation between BC and biomass burning–soil particles during the weekend, we could infer the corresponding traffic emissions that contributed to biomass burning–soil particles. All observations support the conclusion that this particle class has significant anthropogenic sources, which may vary between weekends and weekdays. However, it still needs some more analysis. It should be noted that, there were no significant biomass burning organic aerosols (BBOA) identified by AMS-PMF for the entire measurement period. The marker f60 (the mass fraction of m/z 60 to the total organics) of BBOA being very close to the background fraction of 0.003 (Crippa et al., 2014; Cubison et al., 2011). In addition to biomass burning, also other potential sources for potassium and sulfate rich particles detected by LAAPTOF must be taken into consideration, such as coal combustion (Xu et al., 2018). Therefore, further analysis is required to identify the different sources contributing to this particle class.

As mentioned before, the dominating particle classes varied for different time periods (Shen et al., 2019). In the following we focus on an episode with particles rich in sodium salts, to find out if sea salt particles can significantly influence the aerosol particle composition at this central European location about 800 km inland. Furthermore, we discuss another episode with relatively high organic mass concentrations to find out how these mass loads could form and exist for a relatively long time period. Both episodes are easily recognized from the time series of particles measured by LAAPTOF and the organic compounds measured by AMS (Fig. S2 and Fig. 3; see also Fig. 4 in Shen et al, 2019), as well as the time series of OOA, in particular for

295 SV-OOA, LV-OOA1 and LV-OOA2 (Fig. S4). Apart from the difference of dominating particle classes and particulate organics, the particle acidity indicator value, organonitrate fraction and organic carbon oxidation state, as well as meteorological conditions were quite different between these two episodes. For particle acidity, ~~we used an acidity indicator, namely the~~ molar ratio of NH_4^+ measured/predicted from AMS measurement, is commonly used as an indicator (Zhang et al., 2007). The predicted NH_4^+ value is the mass concentration required to fully neutralize sulfate, nitrate, and chloride measured by AMS. ~~The~~ averaged Lower ratio value represents higher acidity under certain conditions, i.e., inorganic species are dominating the particle mass and crustal elements have relatively low concentrations (Zhang et al., 2007). It should be noted that the organic acids and crustal elements such as sodium can bias the results (Hennigan et al., 2015, and the reference therein). Despite of NH_4^+ ~~measured/predicted~~ the large uncertainty for the entire campaign is 0.82 ± 0.10 , indicating acidity, we still can use such ratio to indicate special events, e.g., sodium nitrate rich particles will provide nitrate, leading to lower values. As shown in Fig. 3, this ratio reaches lowest value during the sodium salts rich episode, corresponding to the highest acidity, while the value in the organic rich episode is at the ~~averaged~~ average level. ~~While the apparent high acidity for the sea salt may partially be due to sea salt sulfate included in the AMS measurement, the~~ If we assume the organic acids only have minor influence on the particle acidity, the anti-correlation between ~~partiele~~ particle acidity indicator value and ~~organies~~ organic concentrations may be related to acid-catalysed organic particle formation. For the fraction of organonitrates, the AMS derived ratio of $\text{NO}^+/\text{NO}_2^+$ is commonly used as an indicator (Farmer et al., 2010; Kiendler-Scharr et al., 2016). In the episode rich in sodium salts, particles contain the highest organonitrate fraction while the particles in the organic rich episode contain the lowest fraction (refer to Fig. ~~S6~~). ~~Both, higher acidity and higher fraction of organonitrates are indicators for~~ S7. It should be noted that NaNO_3 can also result in higher $\text{NO}^+/\text{NO}_2^+$ ratios (5 to 7, Allan et al., 2004) than NH_4NO_3 (2.3 in our calibration). However, the $\text{NO}^+/\text{NO}_2^+$ values here are much higher (c.f. Fig. S7) and thus we can expect a major contribution by organonitrates. Higher fraction of organonitrates suggest a stronger influence of anthropogenic emissions on the sodium rich aerosol. The averaged carbon oxidation state ($\overline{\text{OS}}_{\text{C}}$) of the organic compounds is a robust metric of its degree of oxidation or processing (Canagaratna et al., 2015; Kroll et al., 2011). It turns out that the organics were more oxidized in the organic rich episode (~~refer to Fig. S6~~) with relatively higher $\overline{\text{OS}}_{\text{C}}$ value and less variability (-0.10 ± 0.20) than that for the entire campaign (-0.19 ± 0.23) (refer to Fig. S7 and Table 2). In addition, the major wind with higher speeds was from southwest in the episode rich in sodium salt particles, and we will show in section 3.3 that this is related to long-range transport of sea salt particles. For the organic rich episode, the wind speeds were lower for most of the time. The different features of these two episodes are summarized in Table 2. We will discuss the episodes in more detail in sections 3.3 and 3.4, but will first demonstrate the usefulness of the transport model to identify an industrial point source.

3.2 Identification of an industrial point source

As shown in Fig. 4, we have modelled O_3 , NO_2 , SO_2 , and particle number for the organic rich episode with COSMO-ART. Model simulations show comparable results and/or trends with field observations. For SO_2 , there is a modelled peak at 10:00 on August 26th. As mentioned above, in the upwind direction of our measurement site is a coal fired power plant, where SO_2 is emitted from a stack at 230 m a.g.l.. Therefore, we compared SO_2 concentrations in an area including the power plant and our

measurement site (refer to Fig. 5 a1 and a2), as well as time series of their vertical profile at the measurement site (Fig. 5b). The model results show that SO₂ emissions were starting at 02:00 from the stack of the power plant, continuously transported by southwest wind (from 06:00, not shown here). Around noon, the SO₂ was transported downward due to enhanced vertical mixing. Similar to SO₂, we found corresponding vertical transport of NO_x and primary particles (refer to Fig. 5 c and d), as well as secondary inorganic particles which will be discussed in section 3.3.3. In addition to the power plant emissions, particle emissions from the Karlsruhe refinery (e.g., around 8:00 at 122 m a.g.l.) and downtown Karlsruhe (at 10 m a.g.l.) were observed in the model results, which are visualized via videos of the evolution of particle number at these two altitudes (refer to Data availability). Given these findings, we have verified industrial contributions to the particles measured at this rural site. However, the particle numbers calculated by the model are 10 times lower than the measured values. In this context, it should be noted that, the observations relate to particles larger than 2.5 nm and 7 nm, respectively. The modelled particles include Aitken and accumulation mode particles with median diameter at around 30–80 nm and 300–400 nm, respectively. The model does not represent the ultrafine particles to their full extend, especially e.g., the contribution of organic compounds or ammonia to nucleation.

3.3 Aged sea salt in central Europe

Sodium salt particles accounted on average for $(35.6 \pm 26.7\%)$ % of the particle mass measured by LAAPTOF during the whole campaign, and they dominated several time periods (refer to Fig. S2). However, not all the peak events were the same. We chose a unique episode when sodium chloride signatures, e.g., m/z 81 Na₂Cl⁺, were showing the strongest signal, as plotted in Fig. 6 (a). In this episode, the sodium salts accounted for $(70 \pm 24\%)$ % of total mass of the single particles. To study the origin of these particles, we used the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT; Stein et al., 2015) on the READY website (<http://www.ready.noaa.gov>; Rolph et al., 2017) for back-trajectory analysis. The corresponding result suggests that these particles were coming from the Atlantic Ocean, which is more than 800 km west of our measurement site (refer to Fig. S7S8 a). Therefore, we can constrain the type of sodium salts particles as sea salts with an age of about 1.5 to 3 days.

In this episode, sea salt particles retained strong chloride signatures in spite of the long-range transport, due to relatively high wind speeds (e.g., the average wind speed was $\sim 6 \text{ m s}^{-1}$ from the ocean to our measurement site at noontime on August 21st and 22nd, based on 72 h back-trajectory analysis). Therefore, we interpret them as relatively fresh sea salt, which is the 1st sub-class of sodium salts particles identified with the fuzzy clustering analysis of the single particle mass spectra. As shown in Fig. 6 (a), there is an obvious decrease of m/z 81 Na₂Cl⁺ at the end of this episode, while the sodium nitrate signature at m/z 108 Na₂NO₃⁺ increases. Such trends can also be found in similar but much less prominent sodium chloride rich episodes (see the red guiding arrows in Fig. 6a). This indicates the aging process of sea salt particles, namely heterogeneous replacement of chloride by nitrate on sea salt particles (Gard et al., 1998). We assigned the 2nd sub-class of sodium salts to more aged sea salt particles, which are bigger than the relatively young sea salt (Fig. 6b). 72 h back-trajectory analysis suggests that these particles were coming from the Atlantic Ocean and were carried with southwesterly-south-westerly wind of relatively lower speed (~ 3.5

m s⁻¹) to reach our measurement site on August 23rd, thus they spent more time (~3 days) over the continent and are more aged. The representative spectra of sodium salt particles and the averaged spectra of their sub-classes can be found in Fig. S8S9. The signal difference between aged and young sea salts spectra (aged - young) is shown in Fig. 6 (c), highlighting their signatures. Similar relationships as for m/z 81 Na₂Cl⁺ and 108 Na₂NO₃⁺ can also be found for the other signature peaks (refer to Fig. S9S10).
365 It is worth noting that there are also nitrate related peaks in young sea salt, such as m/z 104 NaClNO₂⁻, verifying that these young sea salts were not fresh. In addition, m/z 92 Na₂NO₂⁺ and 131 NaNO₂NO₃⁻ increase earlier than 108 Na₂NO₃⁺ and 147 Na(NO₃)₂⁻. Such that, we can infer the aging degree from a series of signature peaks in LAAPTOF spectra: 81 Na₂Cl⁺ → 104 NaClNO₂⁻ → 92 Na₂NO₂⁺ and 131 NaNO₂NO₃⁻ → 108 Na₂NO₃⁺ and 147 Na(NO₃)₂⁻ (from relatively fresh to more aged, corresponding to chloride to nitrite to nitrate salts). However, this requires further systematic studies to be validated.

370 For sodium salt particles, a weak but characteristic peak at m/z 129 C₅H₇NO₃⁺ ~~was assigned to organonitrate in LAAPTOF spectra in our previous study (Shen et al., 2019). As shown in Fig. S9 is likely to be an organonitrate signature (Shen et al., 2019). As shown in Fig. S10, m/z 129 C₅H₇NO₃⁺ has a similar trend as m/z 131 NaNO₂NO₃⁻ and it is more pronounced in the LAAPTOF spectra of aged sea salt. In the sodium chloride rich episode, in general m/z 129 C₅H₇NO₃⁺ has a similar trend as organonitrate mass concentrations estimated by AMS results and the results of a molecular analysis using FIGAERO-HR-ToF-CIMS (stands for a filter inlet for gases and aerosols coupled to a high-resolution time-of-flight chemical ionization mass spectrometer) in a companion study by (Huang et al., 2019). m/z 129 C₅H₇NO₃⁺ has a similar trend as organonitrate mass concentrations estimated by AMS results.~~ The method for organonitrate estimation can be referred to Farmer et al. (2010) and the parameters used here are the same as those used by Huang et al. (2019). We have briefly described this method in the supplement after the caption for Fig. S9. ~~AMS derived organonitrate mass concentrations are comparable with the results of S10. In addition, our simulation chamber study has shown that this m/z 129 C₅H₇NO₃⁺ arises from the recombination of inorganic nitrate and organics~~ molecular analysis using FIGAERO-HR-ToF-CIMS (stands for a filter inlet for gases and aerosols coupled to a high-resolution time-of-flight chemical ionization mass spectrometer) in a companion study by (Huang et al., 2019). ~~Therefore, we can confirm that the m/z 129 C₅H₇NO₃⁺ signal is due to organonitrates, rather than the recombination of nitrate and organic fragments~~ due to the laser desorption and ionization (LDI) matrix effect in single-particle MS (Murphy, 2007; Reilly et al., 2000; Reinard and Johnston, 2008; Zenobi and Knochenmuss, 1998). ~~In addition, our simulation chamber studies) and even stronger when it is arising from organonitrates formed under NO_x rich conditions (refer to Fig. S11). Taken together, we can conclude that the m/z 129 C₅H₇NO₃⁺ signal is mainly resulting from organonitrates, but we still cannot rule out the LDI effect. Furthermore, our simulation chamber studies also~~ revealed that m/z 129 C₅H₇NO₃⁺ in LAAPTOF spectra could be a signature for anthropogenic sources (refer to Fig. S4S11 and the corresponding detailed description in the supporting information). Therefore, m/z 129 C₅H₇NO₃⁺ on sea salt particles indicates interactions between the natural sea salt and anthropogenic emissions. It is worth noting that another weak but characteristic peak was observed at m/z 165 Na₃SO₄⁺. This signature peak for sea salt sulfate can be found with stronger signal in young sea salt than in aged sea salt (Gard et al., 1998).

390 Apart from that, in high resolution AMS spectra we have observed the peak at 78.99 CH₃SO₂⁺, which is the marker for methanesulfonic acid (MSA) from marine or coastal areas (Huang et al., 2015), although other sources cannot be excluded

395 [\(Daellenbach et al., 2017\)](#). Based on the method developed by [Huang et al. \(2015\)](#), we have also estimated the MSA fraction (f_{MSA} , signal fraction of MSA to total org and sulfate) and mass concentrations for the entire campaign (refer to Fig. S12). It turns out that during the sodium chloride rich episode, the f_{MSA} is enhanced compared to the other low sodium salts time period, such as organic rich episode. This also supports that sodium salts particles measured by the LAAPTOF had sea salt origin.

3.4 Origin of the aerosol particles during the organics rich episode

400 At the end of the measurement period, we observed three subsequent organic peaks (see Fig. 3, Fig. S2, or Fig. [S5S6](#)): one peak during Friday August 26th day-time and two peaks during following two nights. Organics measured in this episode exhibited the highest mass concentrations of $(12.9 \pm 3.9) \mu\text{g m}^{-3}$, which accounted for $(77 \pm 6\%)$ % of the total non-refractory particle mass and is almost 3 times more than organic masses measured in other periods of the entire AMS measurement period. Please note, some secondary inorganic components were also measured in this episode, showing some peaks at similar times as the high organic peaks. As for the sodium chloride rich episode, we have also done a HYSPLIT back-trajectory analysis for the organic rich episode. Fig. [S7S8](#) (b) shows these results: the air mass was coming from the continent during this episode.

405 In order to better interpret the measurements and to understand the impact of transport processes for the particles measured in the organics rich episode, we have done COSMO-ART simulations covering the time period from August 25th to 28th, 2016. Fig. 7 shows the particulate organic compounds, the secondary inorganics (ammonium, nitrate, and sulfate) measured by AMS, and the result calculated using COSMO-ART (panels d to g). The corresponding meteorological conditions measured are shown in panels (a) to (c) of [FigureFig. 7](#). In general, the trends and also the amplitudes of the model results compare quite well with the observations. In the following, we will combine field observation and transport model simulation results to address these questions: 1) were the organic rich aerosols due to local formation, regional transport, or vertical mixing; 2) were all three organic peaks caused by the same reasons; 3) were the secondary inorganics originating from the same sources as the organics. We will start the discussion with the inorganics, will then go to the organics, and will end up with a typical particle type in this episode.

3.4.1 Secondary inorganic compounds

420 As shown in Fig. 7 (e) to (g), observations and model results for secondary inorganic compounds show comparable diurnal cycles and even the concentration levels match quite well. While ammonium and nitrate have a similar trend for most of the time, the trend for sulfate seems to be somewhat different. The major peaks of ammonium and nitrate appear at similar times as the high organic peaks. As will be shown in the following section, these compounds originated from anthropogenic sources in the urban and industrial area of Karlsruhe and from the city of Stuttgart. There is only one instant when all the measured secondary inorganics are showing peaks together: the inorganics are peaking at ~10:00 on August 26th, when also high SO₂ concentrations were observed (c.f. Fig. 4) which were due to the emissions from a power plant located in upwind direction. To verify this, we can refer to the time series of vertical profiles for the simulated secondary inorganics (Fig. [S4+S13](#) b to d). Despite similar spatial distributions of nitrate, ammonium, NO₂, and SO₂ from 02:00 to 12:00 (c.f. Fig. 5) the measured sulfate

peak for that time is missing in the model output. The reason for this is still unclear. It should be noted that, compared to the sulfate peak observed, the companion ammonium and nitrate peaks are much broader, attributed to additional anthropogenic sources from the city and refinery rather than only the power plant.

430 Some of the differences between model result and observation, such as that on August 25th or e.g., for sharper peaks, may be explained by to missing (point) sources in the emission data and/or the limited spatial resolution. The general agreement is however quite well and also for some point sources like the largest power plant in Karlsruhe. Some of the differences in the trends of nitrate and sulfate may be caused by the different nature of the sources of their precursor gases SO₂ and NO_x. While SO₂ is emitted in this area mainly from a few point sources, NO_x is emitted from various sources over a larger area.

435 3.4.2 Organic compounds

For the organic compounds, the general temporal developmentThe dominating OA components in this episode were LV-OOA1 (42 ± 15) % and LV-OOA2 (32 ± 21) %, which are surrogates for more aged SOA (Zhang et al., 2011). It should be noted that LV-OOA1 has strong correlation with SV-OOA ($\gamma = 0.8$), which is a surrogate for local (less-aged) SOA (Zhang et al., 2011). As shown in Fig. S4 (the insert in panel d), they are both peaking at a similar time during the night with broader peaks for LV-OOA1. This indicates that the LV-OOA1 might have a similar origin as the SV-OOA (more related with local biogenic sources). However, it can also be influenced by regional transport. For example, the first LV-OOA1 peak has two sub-peaks: the first one appears during the night, e.g., related to NO₃ oxidation (strong correlation with $\gamma = 0.9$ between LV-OOA1 and CHO1N family has been found here, refer to their time series in Fig. S4 b); and the second sub-peak appears in the morning rush hour and corresponds to south and southwest wind directions, indicating the potential transport from the urban and industrial area of Karlsruhe. The more oxidized LV-OOA2 is peaking during day-time and shows a negative correlation with SV-OOA ($\gamma = -0.5$), possibly due to photochemical aging of SV-OOA. Therefore, we conclude that the SOA measured in this study was likely due to both local formation and regional transport.

For the field and model comparison, the general temporal development of organics is represented well by the model, showing similar trends as measured by AMS, but underestimates the mass concentrations by a factor of ~2.3; averaged model result is (5.5 ± 3.3) μg m⁻³, while observed were (12.9 ± 3.9) μg m⁻³. Potential reasons for this discrepancy will be discussed in the light of sensitivity tests at the end of this section.

As mentioned above, the ~~organic particle fraction~~organics shows a peak on Friday August 26th during day-time (peak 1) and two more during the following nights of the weekend (peak 2 and 3) in the organics rich episode. Model result cover peak 1 and 2, as well as of the first part of peak 3. Analysis of the model results shows that the three peaks were caused by different reasons including meteorological conditions, regional transport, and local or in-situ new particle formation and growths (such as night-time new particle formation; Huang et al., 2019). As shown in Fig. 7a and d, peak 1 corresponds to south and southwest wind, indicating that the air masses came from the urban and industrial area of Karlsruhe. This is illustrated very well in a video showing the transmission of particulate organics leading to peak 1 (refer to Data availability). The most intensive peak in organic particle mass (peak 2), starts from August 26th 20:30 (Fig. 3 and Fig. S2) when the wind direction shifts from northwest to east

460 (northeast, straight east, and southeast), the wind speed decreases to $< 0.5 \text{ m s}^{-1}$, and the night-time temperature inversion starts. Such stagnant conditions limit aerosol mixing and support build up aerosols and their growth. During the entire campaign, the strongest temperature inversion was observed during this night from August 26th to 27th (peak inversion at 02:40 on August 27; Fig. S3). In spite of the stagnant conditions and rather low wind speeds, regional transport still plays an important role. The model results for this time show transport of organic aerosol particles from the city of Stuttgart (southeast to the measurement site) over forested areas into the Rhine valley. This is clearly visible in a corresponding video of the model results for particulate organics and wind directions (refer to Data availability). Furthermore, we investigate the relative contribution of biogenic and anthropogenic emissions and their interactions for this episode by comparing the model results with and without biogenic emissions. It turns out that the corresponding air masses originating from anthropogenic emissions in the Stuttgart area were transported to the measurement site by south-easterly winds via forested hills. During this transport, the organic particle mass concentrations increased while being mixed with biogenic emissions from the forest. The resulting air mass finally reaching the measurement site consisted therefore of anthropogenic and biogenic organics. For peak 3, we can also find impact of transport as illustrated in another video of particulate organic mass evolution (refer to data availability). However, the origin of this peak is more complex and is most likely more influenced by local particle formation and growths. After the night-time peak 2, from 12:00 on August 27 the air mass were doing back and forth movements in west-east directions, showing a strong concentration gradient (west-east). Such conditions lead to a high uncertainty of the model results for one point. Hence, although we observed quite similar particle compositions in these three subsequent peaks with high organic mass concentrations for similar meteorological conditions with low wind speeds, all three peaks were caused by air masses of different origin.

475 As mentioned above, the model underestimates the organic aerosol mass on average by a factor of 2.3. In order to better understand this discrepancy, we have done three sensitivity tests: ~~with one~~ without biogenic emissions (test 01), one with doubled biogenic emissions (test 02) and one with boundary values or regional background values (outside the model domain) of particulate organic compounds increased by a factor of 10 (test 03). The results of test 01 suggest that $(80 \pm 4)\%$ of the organic mass calculated was resulting from biogenic sources for the peak 1 time period, corresponding to the air masses coming from Karlsruhe. Regarding the time period of peak 2, there was a sharp drop of the biogenic contribution from the 88% to 68% corresponding to August 26th 23:00 to August 27th 2:00 when the air mass was coming from direction of Stuttgart (see Fig. 8). Hence, we can infer that the anthropogenic contribution to the air mass that originated from Stuttgart was somewhat larger than that originated from Karlsruhe. Over the entire organic rich episode, on average $(74 \pm 7)\%$ of the organics mass was contributed by biogenic sources. To test if missing emissions of biogenic precursors can explain the underestimated organic particulate masses we doubled this major (biogenic) source for organics (test 02). As shown in Fig. 9, doubling biogenic emissions makes some of the peaks of organic mass fit the observations better but the concentrations before and in between the peak values are still underestimated. Another potential reason for the discrepancy could be to low boundary values (regional background); therefore, we increased them by a factor of 10 (test 03). This elevates the overall simulated concentration levels, leading to an overestimation especially of the first peak. These tests demonstrate the potential role of precursor emissions and regional transport of organic aerosol particles on the model results. There seems to be evidence that the model underestimates

the regional background of organic aerosol particles. However, we cannot rule out that also underestimated sources of VOC or weaknesses in the SOA formation mechanism may contribute to the underestimation of the organic aerosol particle mass. For example, night-time new particle formations mechanisms are not included in the current model version.

4 Conclusions and atmospheric implication

High particle number concentrations have been observed at a rural site in the upper Rhine valley by measurements on a tramline (Hagemann et al., 2014). In order to study the nature of the aerosols in the same region, we conducted a 6-week stationary field measurement campaign at a rural site next to the tramline in summer 2016, and applied a regional transport model COSMO-ART to analyse the origin of the aerosols. Comparing the field observations and model calculations, we have verified that industrial emissions contribute to the high particle number concentrations at this rural site.

For the whole measurement period, the particles that dominate the mass are sodium salts, which contributed $(36 \pm 27) \%$ to the $PM_{2.5}$ mass measured by the LAAPTOF. According to the AMS measurements, organics, sulfate, nitrate, and ammonium contributed $(58 \pm 12) \%$, $(22 \pm 7) \%$, $(10 \pm 1) \%$, and $(9 \pm 3) \%$ to the total non-refractory $PM_{2.5}$ mass, respectively. AMS-PMF results suggest the organic aerosols (OA) consisted of $(9 \pm 7) \%$ HOA, $(16 \pm 11) \%$ SV-OOA, and $(75 \pm 15) \%$ LV-OOA with three sub-types (LV-OOA1, LV-OOA2, and LV-OOA3). Furthermore, we have investigated two episodes, which were dominated by aged sea salt and organics, which show characteristic features such as higher particle acidities and carbon oxidation states, respectively. Both episodes selected, were dominated by particles formed from the interaction of natural/biogenic and anthropogenic emissions.

For the sodium chloride rich episode, sodium salt particles accounted for $(70 \pm 24) \%$ of total mass of the single particles. We could show that the dominating particles were originally sea salt, from the Atlantic Ocean (~800 km away from the measurement site). These particles consisted of two classes with different aging stages. After 1.5–3 days of transport, the original sea salt particles retained strong chloride signatures while after 3.5–4 days of transport over the continent aged sea salt with stronger nitrate signatures and larger size arrived at the measurement location. In addition, the organonitrate signature peaks at m/z 129 $C_5H_7NO_3^+$ were stronger for the aged sea salt particles. According to chamber experiments on OH oxidation of α -pinene and/ or toluene in the presence of NO_x on sodium nitrate seed particles, we have evidence that m/z 129 $C_5H_7NO_3^+$ is a signature peak for anthropogenic organics in single particle mass spectra. Hence, the m/z 129 $C_5H_7NO_3^+$ observed on sea salt particles in this study is another indicator for the interaction of anthropogenic emissions with the sea salt particles. ~~Besides, the observed signature peak at m/z 165 $Na_3SO_4^+$ for sea salt sulfate may indirectly explain apparently higher particle acidity in this episode. Some~~ Besides, some other signature peaks such as 81 Na_2Cl^+ , 92 $Na_2NO_2^+$, 108 $Na_2NO_3^+$, 104 $NaClNO_2^-$, 131 $NaNO_2NO_3^-$, and 147 $Na(NO_3)_2^-$ have the potential to indicate degree of aging of sea salt particles.

For the organics rich episode, the particulate organic compounds accounted for $(77 \pm 6) \%$ of total non-refractory particle mass. ~~We identified potential sources measured by the AMS. AMS-PMF analysis suggests the dominating OA components were LV-OOA1 and LV-OOA2, contributing $(42 \pm 15) \%$ and $(32 \pm 21) \%$ to the total OA mass in this episode. LV-OOA1 and~~

LV-OOA2 had positive ($\gamma = 0.9$) and negative ($\gamma = -0.5$) correlation with SV-OOA, respectively, indicating their local formation. In order to investigate origin of the elevated organic particle concentrations and of secondary inorganic compounds aerosols and their transport and transformation processes, we applied a regional transport model, COSMO-ART. Transport model results suggest that $(74 \pm 7)\%$ of the particulate organic mass calculated was related to biogenic emissions. With the combination of mass spectrometry and COSMO-ART modelling, we can confirm that during Friday, August 26th, air masses came from an industrial area and the city of Karlsruhe, leading to high particle numbers (including all the particles with size >2.5 nm), high particulate organics and secondary inorganics, as well as high fractions of aged-biomass-burning and soil-dust like particles. During night-time on August 26th, south-easterly winds transported air masses from Stuttgart via forested areas to the measurement site, resulting in particles formed by the interaction of anthropogenic and biogenic precursors. Such air masses flowed into the Rhine valley and contributed also to the high organic aerosol mass concentrations during the night from August 27th to 28th. In addition, low wind speeds and temperature inversion limited mixing and supported new particle formation and growth, partially contributing to higher organic particle masses.

Comparing comprehensive field observations with the COSMO-ART model results showed quite good agreement for trends and most concentration levels for several aerosol particle compounds and trace gases even if not all details could be reproduced by the model. However, the simulated particle numbers are 10 times less than the observed values. This is mainly due to the fact that not all new particle formation processes (nucleation mode) are included in the model (e.g. night time chemistry as well as the role of organics or ammonia). Furthermore, the model underestimates particulate organics by a factor of 2.3. This discrepancy may be partly explained by an underestimated regional background, missing sources, or incomplete SOA formation and growths mechanisms, in the current model version. Such field-model comparisons and corresponding sensitivity tests are useful for targeting missing sources and mechanisms in the current model and thus help its development. Such that model can be improved to become better predictive, which is useful for e.g., legislation and implementation of policies against air pollution. Consequently, model systems suitable for a reasonable environmental policy should be validated by comprehensive field studies with state-of-the-art aerosol analytics and meteorological parameters.

As discussed above, we can conclude that factors related to topography, metrological conditions, local emissions, in-situ formation and growths, regional transport, as well as the interaction of biogenic and anthropogenic compounds need to be considered for a comprehensive understanding of aerosol composition and processing in this region. This Our results show how close the physical and chemical nature of aerosols is related to emission sources, transformation processes, and complex transport pattern. The method of combination of field measurement and model calculation in this study can be widely used in other places in the world, in order to investigate the origin of aerosols and their interaction between transport and transformation. For example, in some pollution hot spots like Po Valley, and some other places which are subject to both biogenic and anthropogenic emissions. Besides, it is also not limited to the summer time that we studied. Taken together, this study provides a good insight into the chemical nature and complex origin of aerosols in the upper Rhine valley, achieved by combining complimentary mass spectrometry and transport modelling.

Data availability

560 Five supporting videos (.gif files) are available at KIT open data (<https://doi.org/10.5445/IR/1000094401>).

Author contributions

X.S. operated LAAPTOF and AMS during the whole field campaign, did the LAAPTOF data analysis, did COSMO-ART model runs and model output data analysis, did HYSPLIT back-trajectory analysis, [did further analysis for AMS data](#), produced all figures, and wrote the manuscript. H.S. organized the campaign, provided trace gas, meteorology, and particle measurements as well as suggestions for the data analysis, interpretation, and discussion. H.V. and B.V. provided the input data for COSMO-ART model, provided suggestions for the model simulation, the model data analysis, interpretation, and discussion. W.H. operated AMS during the whole campaign and did [basic AMS data analysis, as well as PMF analysis](#). C.M. helped to operate the instruments, did black carbon data analysis, and provided suggestions for the data analysis, interpretation, and discussion. R.R. helped to operate LAAPTOF. A.S.H.P. gave comments for interpretation of black carbon data. T.L. gave general advices and comments for this paper. All authors contributed to the final text.

Competing interests

The authors declare no conflict of interest.

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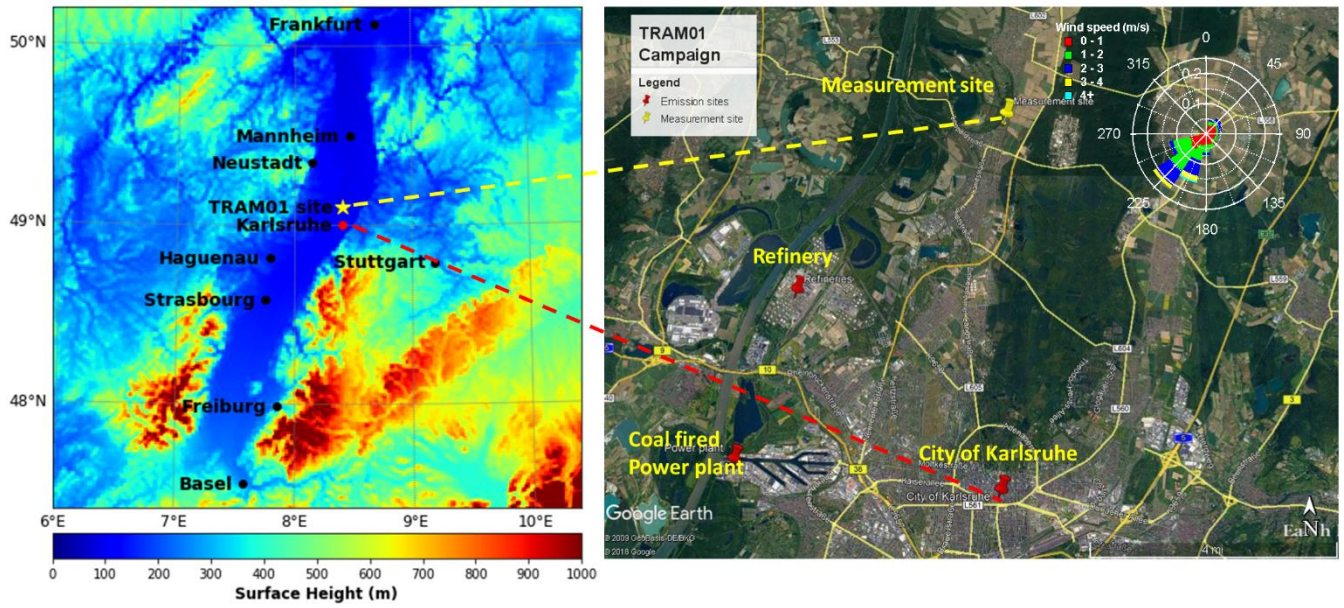
Table 1: Mean values (\pm sdev) of characteristic parameters for the entire measurement period.

Parameter	Mean value \pm sdev	Parameter	Mean value \pm sdev	
Wind Direction ($^{\circ}$)	174.0 \pm 75.0	Particulate compounds measured by AMS		
Wind Speed (m s ⁻¹)	1.3 \pm 0.9		Organics ($\mu\text{g m}^{-3}$)	5.1 \pm 3.3
Temperature (K)	294.4 \pm 4.5		Ammonium ($\mu\text{g m}^{-3}$)	0.7 \pm 0.4
Relative Humidity (%)	70.0 \pm 17.0		Nitrate ($\mu\text{g m}^{-3}$)	0.8 \pm 0.8
O ₃ (ppb)	26.5 \pm 15.8	Sulfate ($\mu\text{g m}^{-3}$)	1.7 \pm 0.8	
NO ₂ (ppb)	11.4 \pm 6.9	Seven particle classes measured by LAAPTOF		
SO ₂ (ppb)	1.2 \pm 2.7		C1: Calcium-soil ($\mu\text{g m}^{-3}$)	0.4 \pm 0.5
CO ₂ (ppm)	446.0 \pm 32.0		C2: Aged soot ($\mu\text{g m}^{-3}$)	0.1 \pm 0.4
Particle Number (10 ⁴ cm ⁻³) (size >2.5 nm)	1.2 \pm 0.8		C3: Sodium salts ($\mu\text{g m}^{-3}$)	1.2 \pm 1.2
Particle Number (10 ³ cm ⁻³) (size: 2.5 to 7 to 7 nm)	4.7 \pm 4.9		C4: Secondary inorganics-amine ($\mu\text{g m}^{-3}$)	0.1 \pm 0.6
BC ($\mu\text{g m}^{-3}$)	0.8 \pm 0.7		C5: Biomass burning-soil ($\mu\text{g m}^{-3}$)	1.2 \pm 1.3
PM _{2.5} ($\mu\text{g m}^{-3}$)	6.3 \pm 3.2		C6: Biomass burning-organosulfate ($\mu\text{g m}^{-3}$)	0.4 \pm 0.6
PM ₁₀ ($\mu\text{g m}^{-3}$)	8.6 \pm 4.2	C7: Mixed/aged dust ($\mu\text{g m}^{-3}$)	0.5 \pm 0.7	

775 Note that the full names of seven particle classes are C1: calcium-rich and soil-dust-like particles; C2: aged-soot-like particles; C3: sodium-salt-like particles; C4: secondary- inorganics-rich and amine containing particles; C5: aged-biomass-burning and soil-dust like particles; C6: aged-biomass-burning and organosulfate-containing particles; C7: mixed/aged and dust-like particles

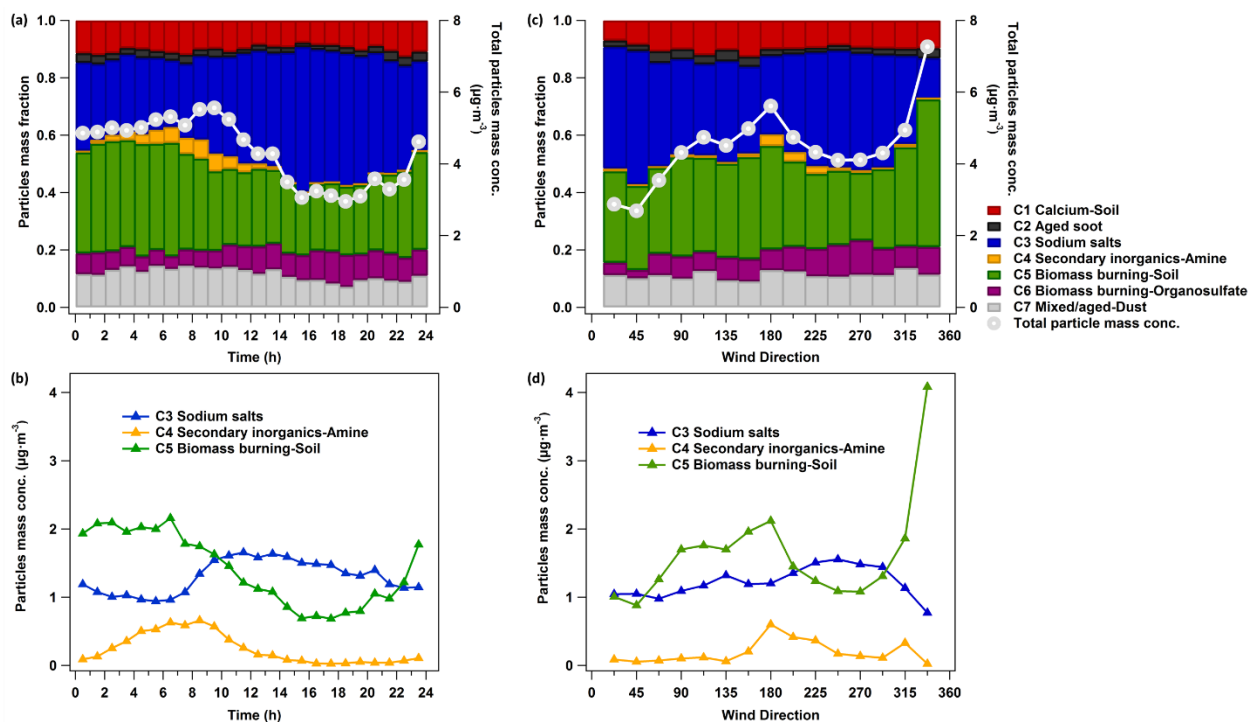
Table 2: Features and mean values (\pm sdev) of characteristic parameters for the two selected episodes

-	Sodium chloride rich episode	Organics rich episode	Instrument
Date and time	0820 noon to 0823 noon	0825 midnight to 0828 midnight	
Sodium salts ($\mu\text{g m}^{-3}$)	2.2 \pm 1.0 Higher	0.6 \pm 0.4 Lower	LAAPTOF
Biomass burning-soil ($\mu\text{g m}^{-3}$)	0.7 \pm 0.7 Lower	1.7 \pm 1.1 Higher	LAAPTOF
Organics ($\mu\text{g m}^{-3}$)	2.0 \pm \pm 1.2 Lowest	12.9 \pm \pm 3.9 Highest	HR-ToF-AMS
Particle acidity indicator NH ₄ ⁺ measured/predicted	0.67 \pm \pm 0.10 Highest acidity Lowest value	0.83 \pm 0.06 Average acidity value Anti-correlated with org	HR-ToF-AMS
Organonitrate fraction indicator NO ⁺ /NO ₂ ⁺	11.0 \pm 6.1 Highest	6.1 \pm 0.9 Lowest	HR-ToF-AMS
Oxidation state (Osc) (Anti-correlated with org)	-0.24 \pm 0.34 Less oxidized Strongest variability	-0.10 \pm 0.20 More oxidized Least variability	HR-ToF-AMS
Wind Direction ($^{\circ}$)	Southwest	Varying	Meteorology sensor
Wind Speed (m s ⁻¹)	1.6 \pm 1.0	1.2 \pm 1.0	(WS700, Lufft Lufft GmbH)



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Figure 1: Measurement location and topography (left), nearby peak emission sites and wind rose for the complete measurement campaign in summer 2016 (right).



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Figure 2: Diurnal pattern (a and b) and wind direction resolved particle class masses (c and d) measured by LAAPTOF. Please note that the uncertainties of the particle class mass concentrations are relatively large (Shen et al., 2019). The data is averaged over the entire campaign. In total 4842 time points of 10 min averaged data were analysed for these plots. The total particle mass concentration for each hour of the day is on average $4.3 (\pm 2.6, \text{one } \sigma) \mu\text{g m}^{-3}$ and for each wind direction is on average $4.5 (\pm 2.9) \mu\text{g m}^{-3}$. For the dominating classes the average mass concentrations are $1.3 (\pm 1.1) \mu\text{g m}^{-3}$ and $1.2 (\pm 1.0) \mu\text{g m}^{-3}$ for class 3, $0.2 (\pm 0.5) \mu\text{g m}^{-3}$ and $0.2 (\pm 0.4) \mu\text{g m}^{-3}$ for class 4, $1.4 (\pm 1.0) \mu\text{g m}^{-3}$ and $1.6 (\pm 1.5) \mu\text{g m}^{-3}$ for class 5, for (b) and (d) respectively. It is worthy to note that particle mass quantification is one of the most challenging issues in single particle mass spectrometry (Shen et al., 2019, and the references therein). There are several assumptions mainly regarding particle shape and density in our particle mass quantification method, leading to potential uncertainties of up to $\sim 300\%$. Despite such large uncertainty, the estimated total particle mass has good correlation with the total mass measured by the AMS, covering $\sim 85\%$ of the measurement time (Shen et al., 2019).

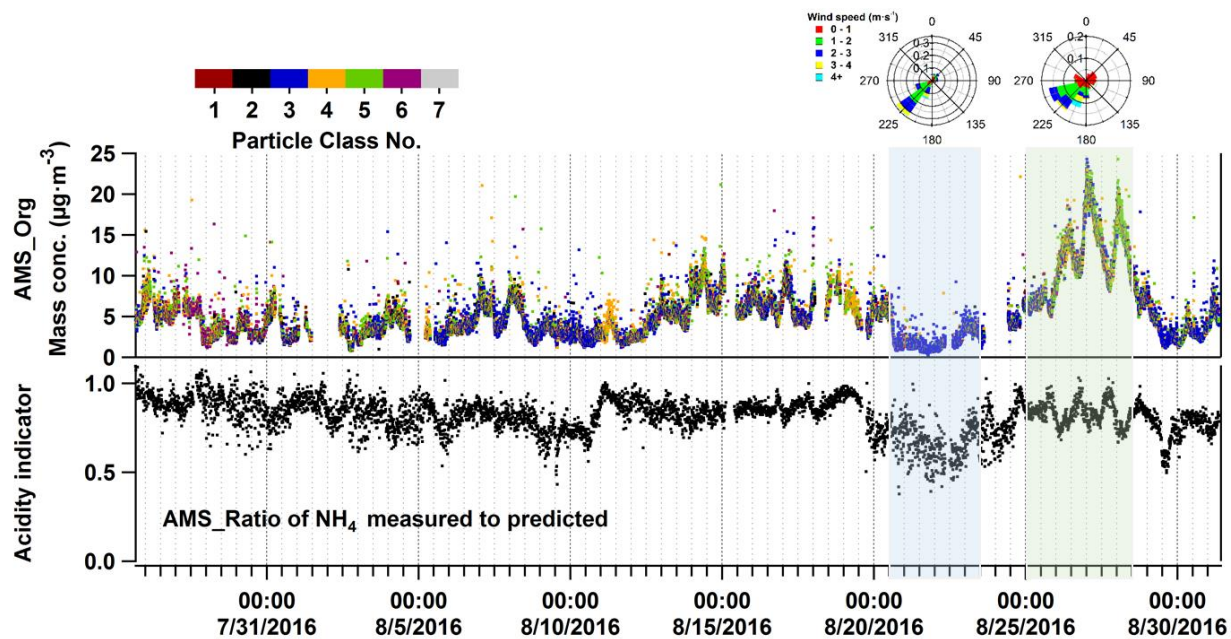


Figure 3: Time series of particulate organic mass measured by the AMS (AMS org in the upper panel), the particle acidity indicator (the lower the value the more acidic, lower panel), and the particle class number classified by LAAPTOF (colour code, upper panel). The seven classes are class 1: calcium-Soil; class 2: aged soot; class3: sodium salts; class 4: secondary inorganics-amine; class 5: biomass burning-soil; class 6: biomass burning-organosulfate; and class 7: mixed/aged-dust. The first (blue) and second (green) shaded areas mark the sodium chloride rich and organic rich episodes, respectively.

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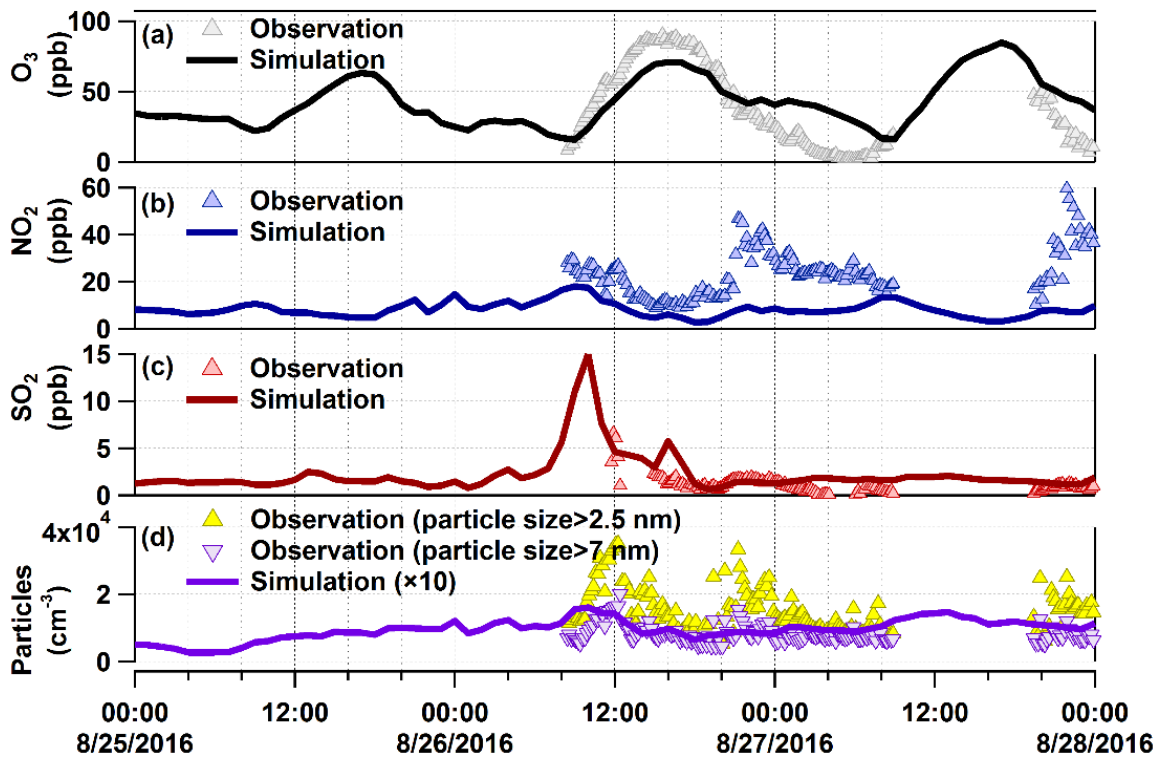
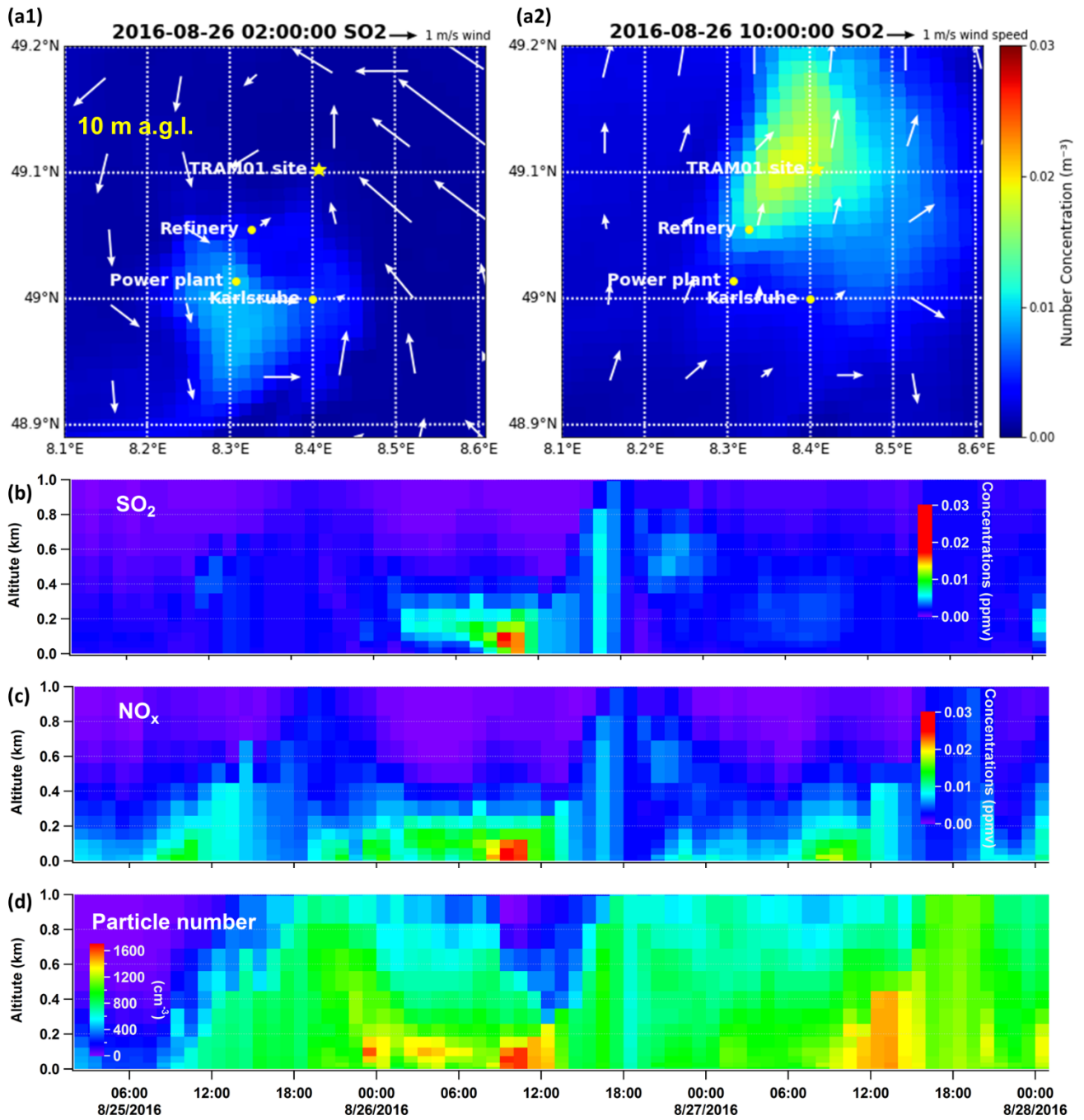


Figure 4: Comparison between observations and model results for O₃, NO₂, SO₂, and particle number in organic-rich episode. -The modelled particles do not represent the ultrafine particles to their full extendextent, especially for the particles in the nucleation mode.

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810 Figure 5: Modelled SO₂ concentrations at 2:00 (a1) and 10:00 (a2) on August 26th at 10 m a.g.l.. Panels (b) to (d) are time series of vertical profiles above the measurement site for simulated SO₂, NO_x and total particle number.

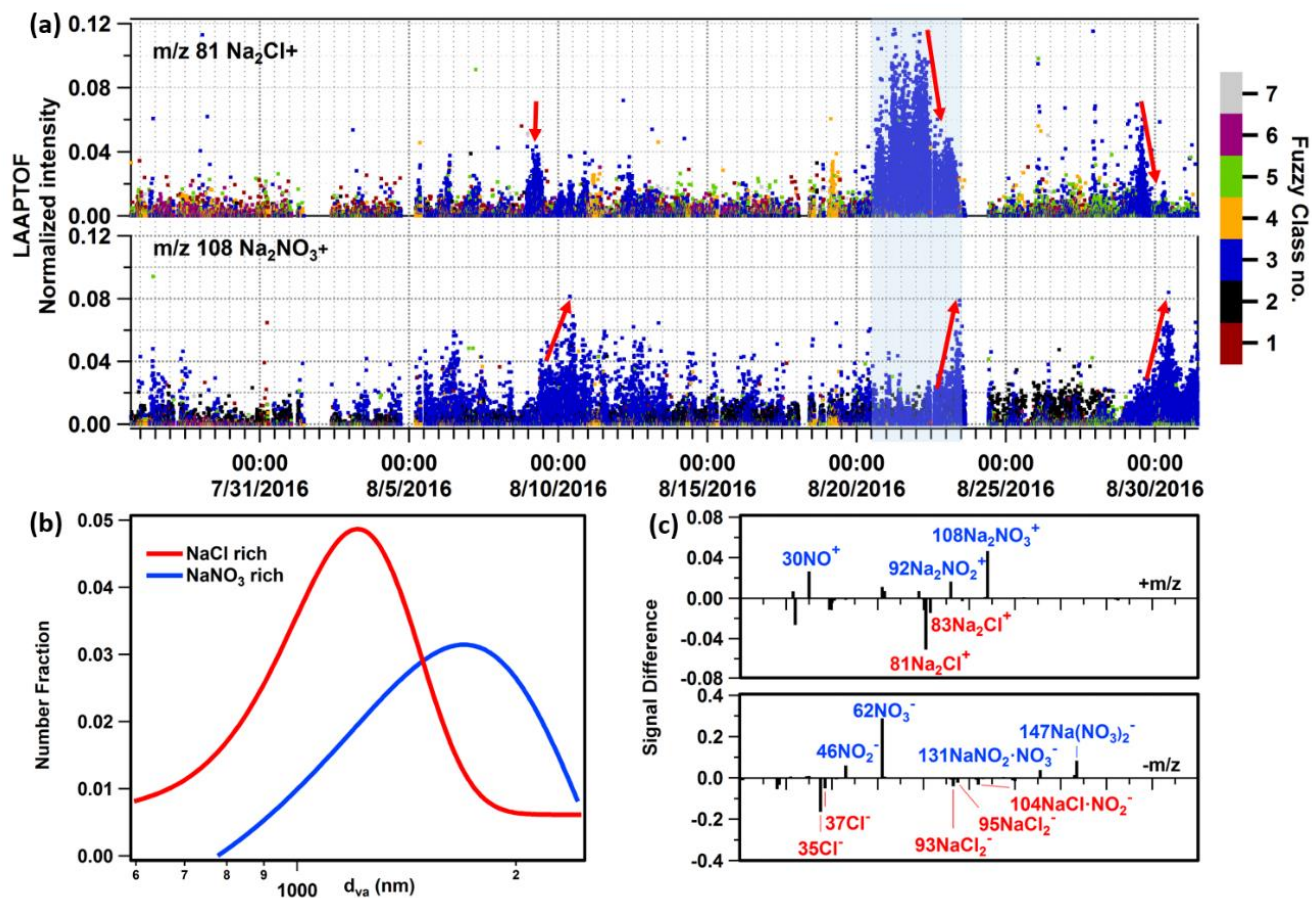
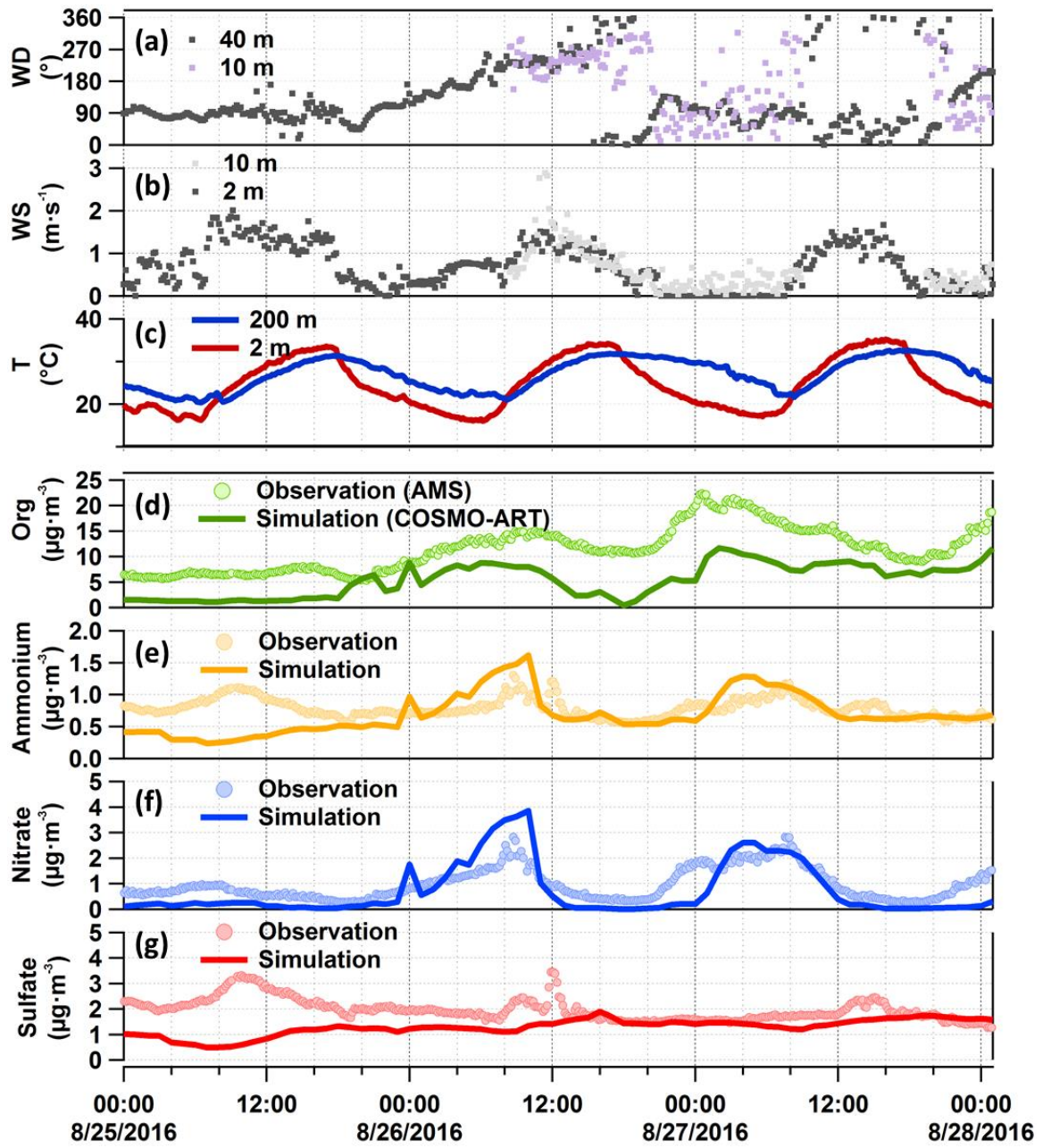


Figure 6: (a) Time series of sodium chloride and nitrate signatures in seven particle classes. Y-axis is the normalized ion intensity (each ion peak intensity is normalized to the sum of all ion signals; positive and negative ions were analysed separately). The seven classes are class 1: calcium-Soil; class 2: aged soot; class3: sodium salts; class 4: secondary inorganics-amine; class 5: biomass burning-soil; class 6: biomass burning-organosulfate; and class 7: mixed/aged-dust. Obvious sodium chloride and nitrate signatures are mainly found in class 3 labelled in blue colour. (b) Size distributions of NaCl rich (young sea salt) and NaNO_3 rich (aged sea salt) particles. Gaussian fit NaCl $d_{va} (\pm\sigma) = 1211(\pm 352)$ nm; NaNO_3 $1697(\pm 946)$ nm. (c) Spectra difference between aged and young sea salt particles (aged-young). The blue band marks the sodium chloride rich episode.

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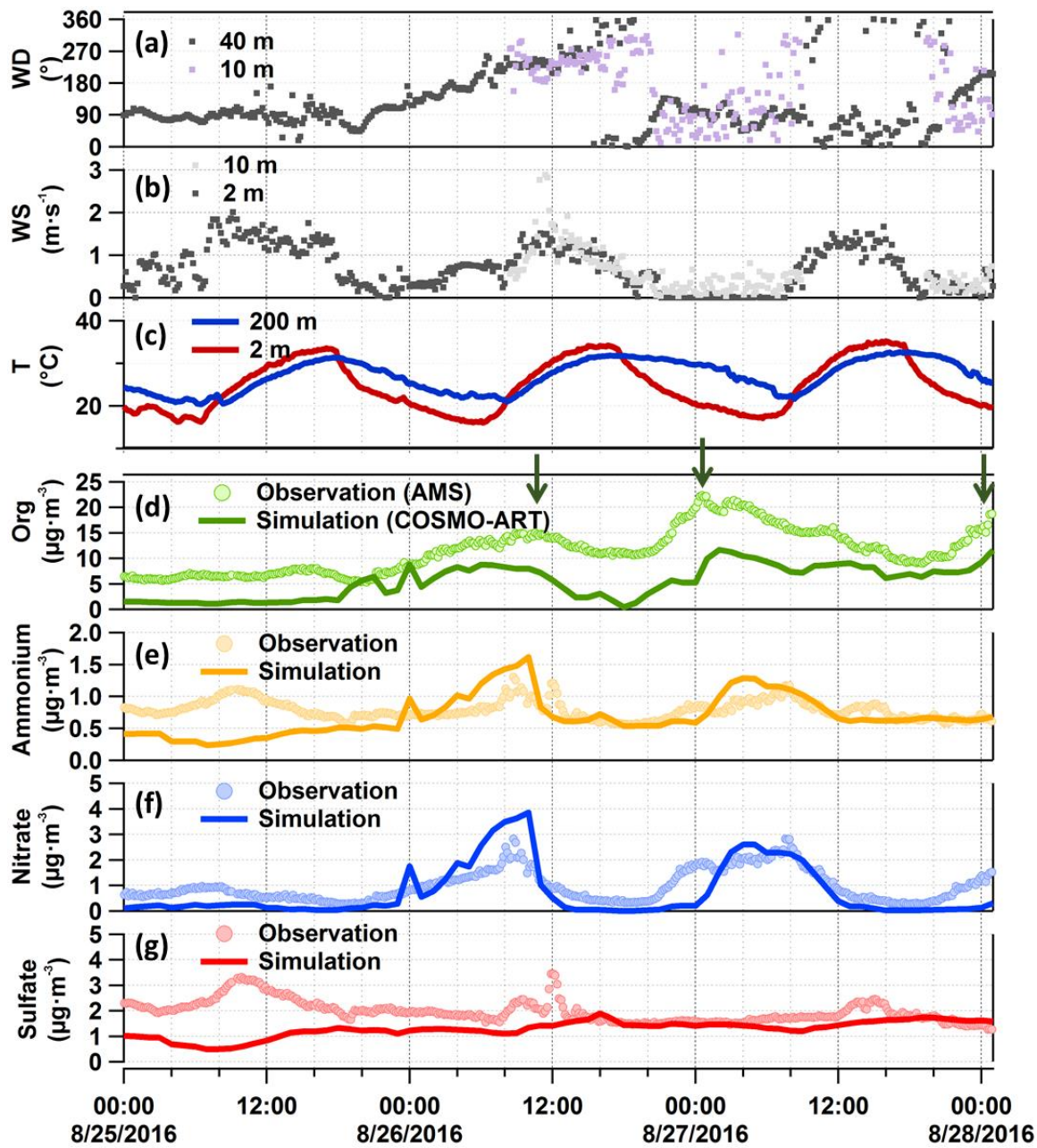


Figure 7: Comparison between observations and model simulations for organic-rich episode. Panel (a) and (b) are wind conditions measured 10 m a.g.l. at our measurement site (purple and light grey dots for wind direction and speed, respectively), as well as the complementary KIT-tower data (dark grey dots). Panel (c) is the temperature data from the KIT-tower. Panel (d) to (g) are the particulate organic compounds, ammonium, nitrate, and sulfate measured by AMS and simulated by COSMO-ART. The simulation results are at the lowest atmospheric level in the model, namely 10 m a.g.l.

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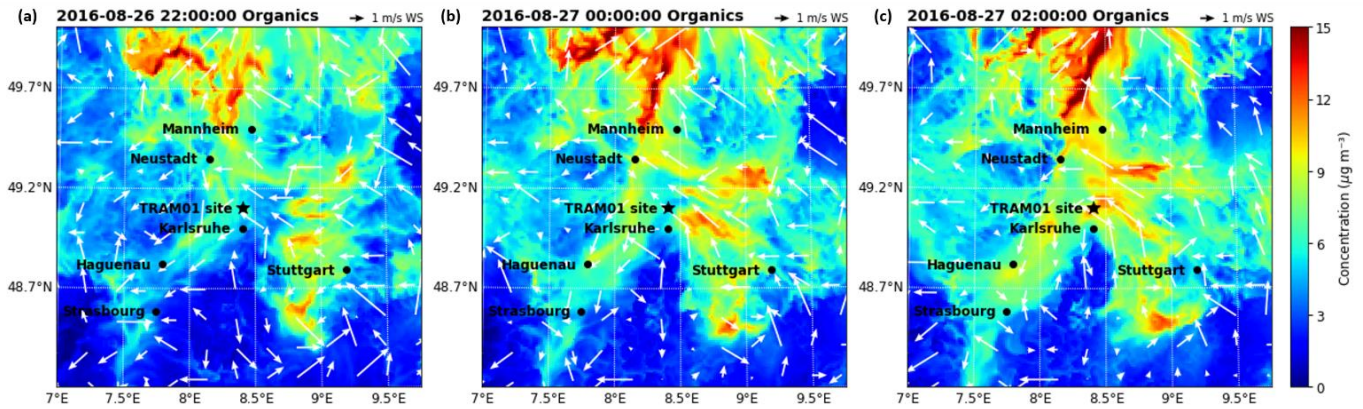


Figure 8: Particulate organics concentrations calculated by COSMO-ART for three different time points (a) to (c) during the night from August 26th to 27th for 10 m a.g.l..

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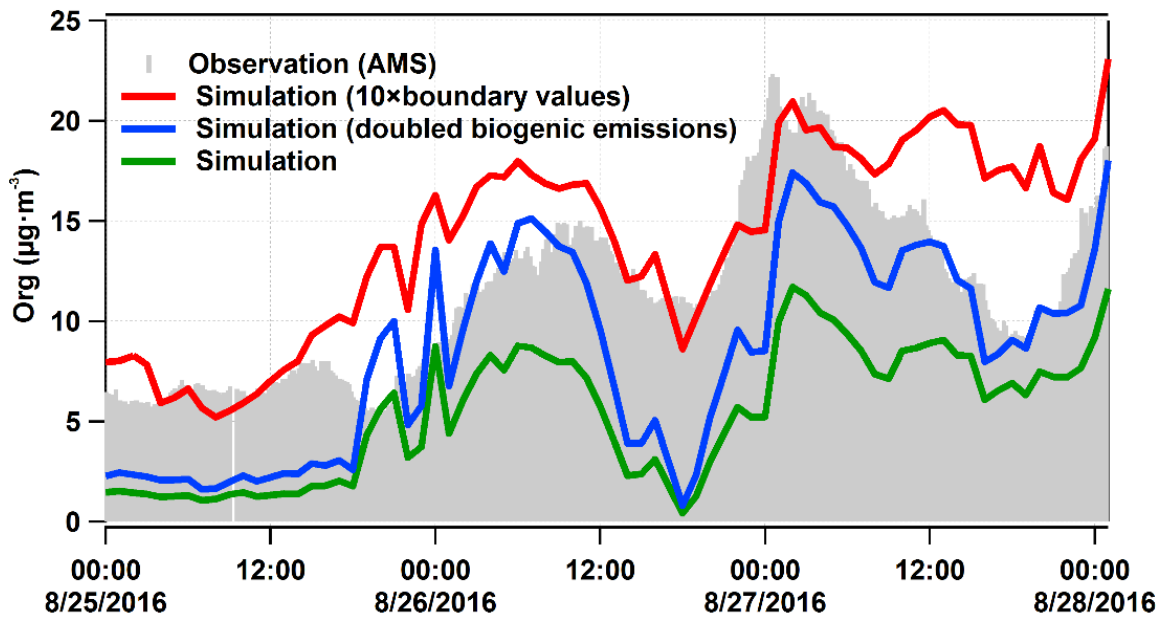
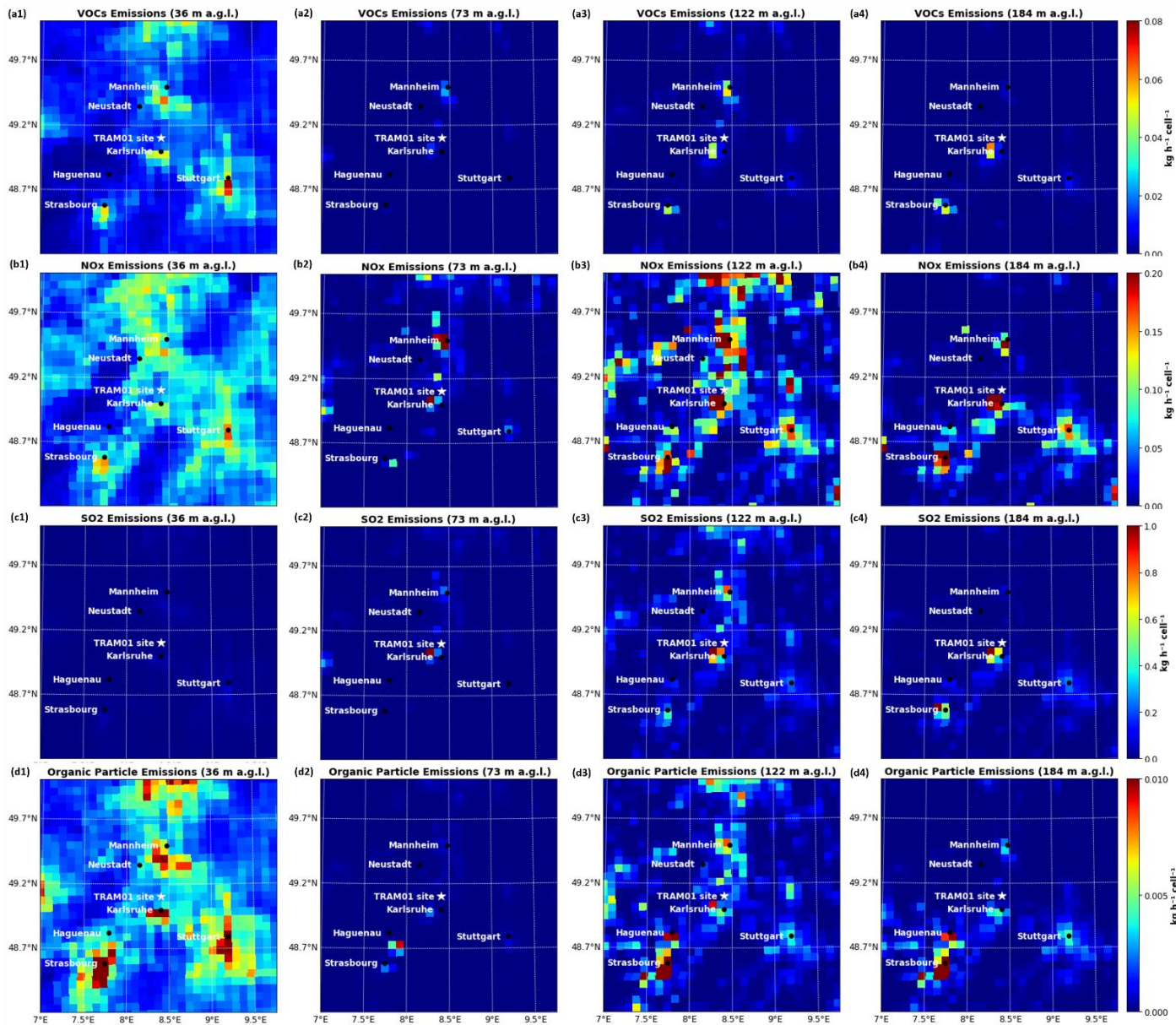


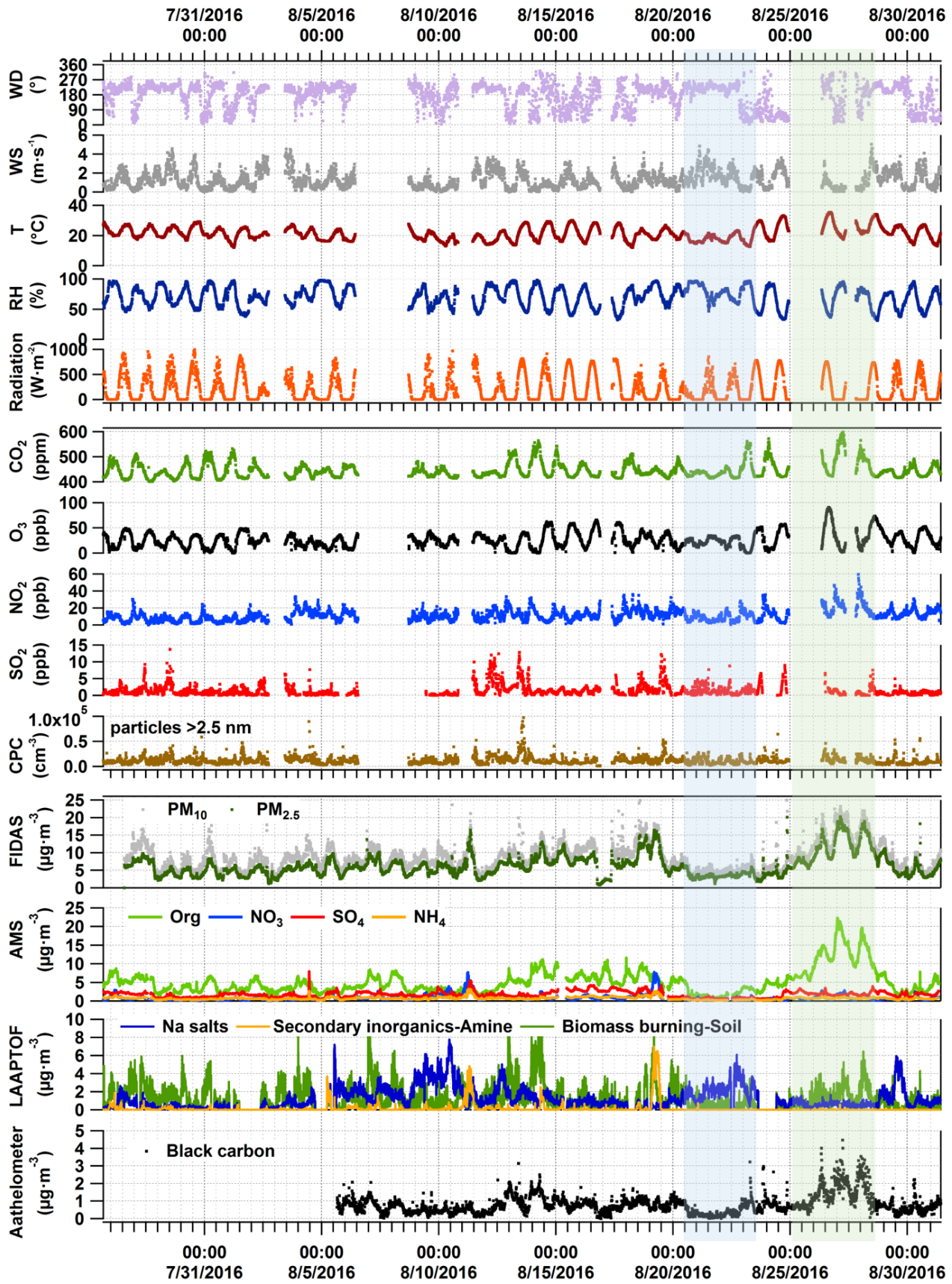
Figure 9: Modelled organic mass concentrations (lines) compared to measured values (grey area). The green line is the same as in Figure 7, the red line was calculated for a tenfold-enhanced regional background in organics, and the blue was calculated for doubles biogenic emissions in COSMO-ART.

In order to study the nature and origin of aerosols at a rural site in the upper Rhine valley, we conducted a stationary measurement campaign (TRAM01) in summer 2016, and applied a transport model COSMO-ART to help source apportionment and to achieve a better understanding of the impact of complex transport pattern on the field observations. In the main manuscript, we have mainly discussed two selected episodes, one is rich in sodium chloride and another is rich in organics. Here is the supporting information, including figures, as well as some detailed descriptions.

5



10 Figure S1: Emission maps of VOCs (a1 to a4), NO_x (b1 to b4), SO₂ (c1 to c4), and primary organic particles (d1 to d4) at different altitudes, i.e., 36, 74, 122, and 184 m (from left to right) above ground level (a.g.l.).



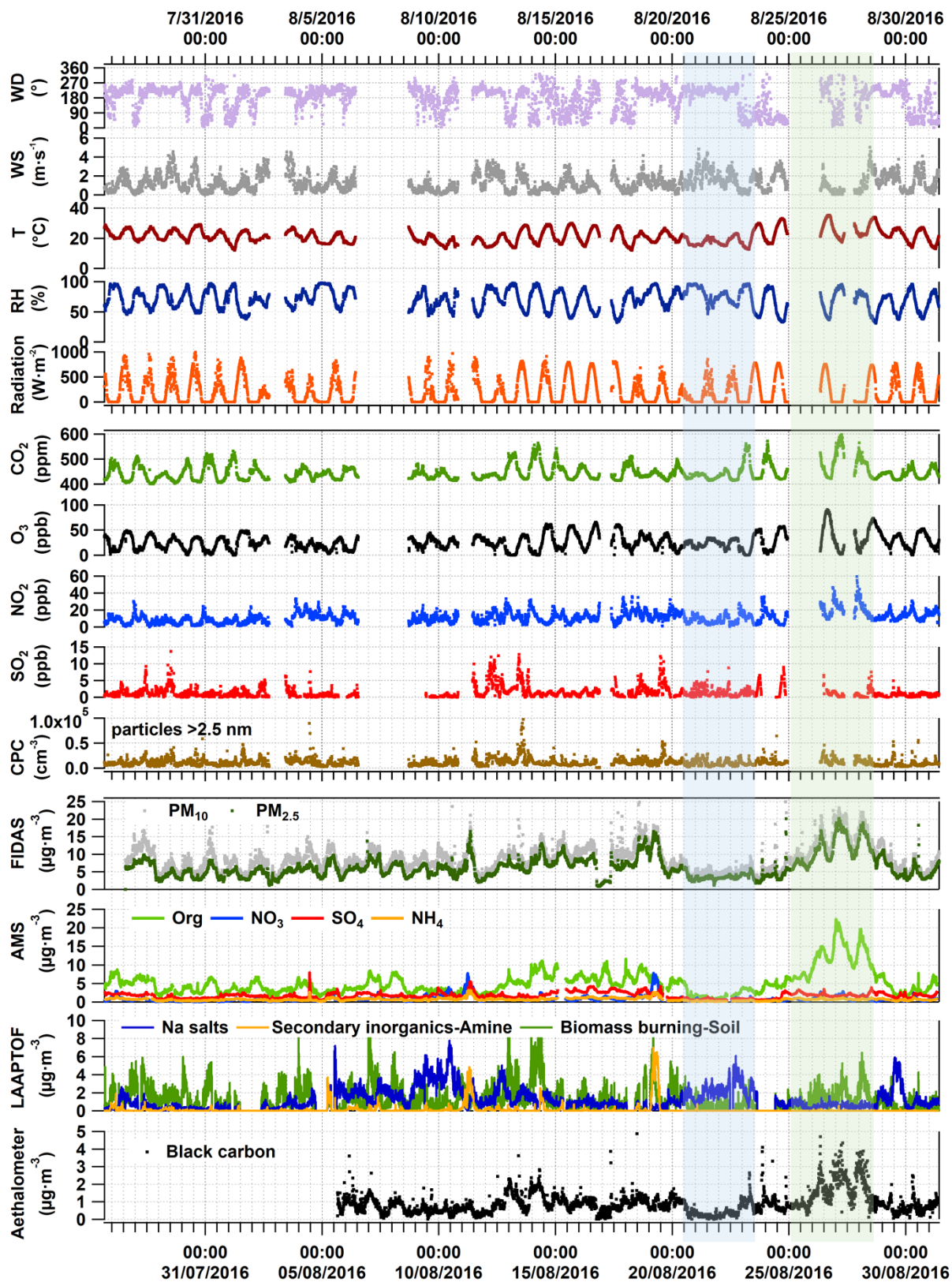


Figure S2: Overview of the whole measurement campaign including meteorology parameters, trace gases, and aerosol particle mass and composition. The first (blue) and second (green) shaded areas mark the sodium chloride rich and the organic rich episodes, respectively.

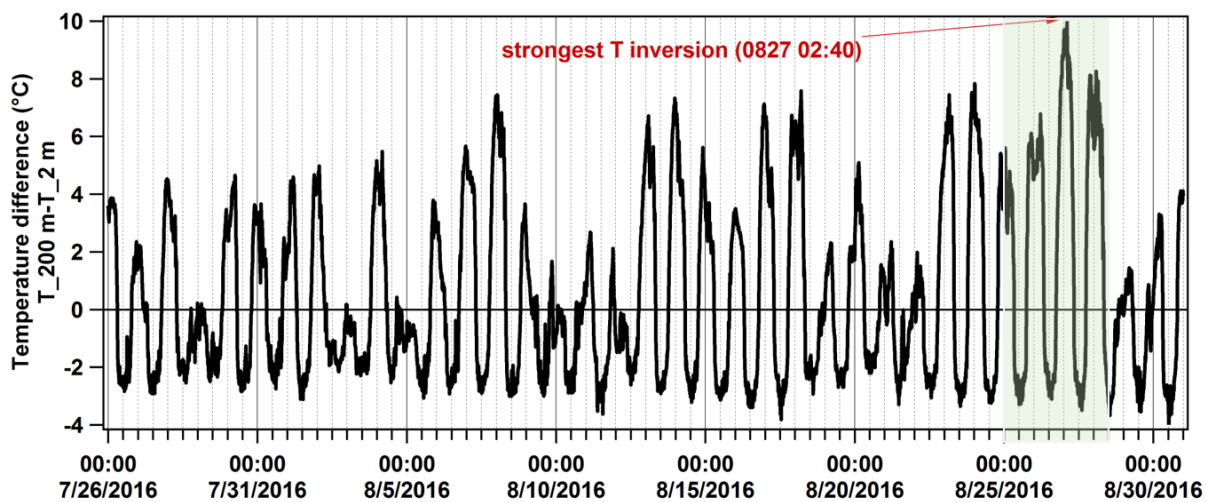


Figure S3: Time series of temperature difference between 200 m and 2 m above ground level (a.g.l.), measured at the KIT-tower. The bigger difference denotes the stronger temperature inversion. Temperature inversions are stronger during the organics rich episode marked with a shaded background. The strongest temperature inversion was measured at 02:40 on August 27.

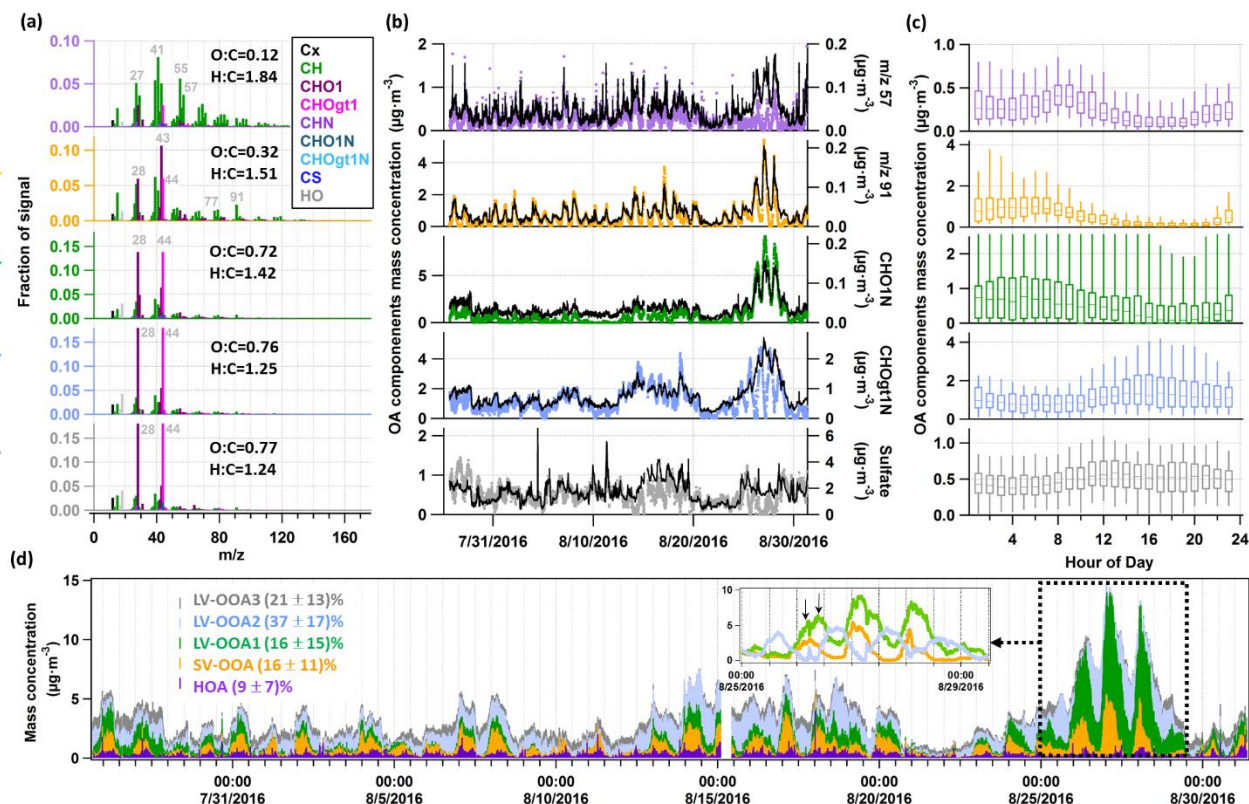


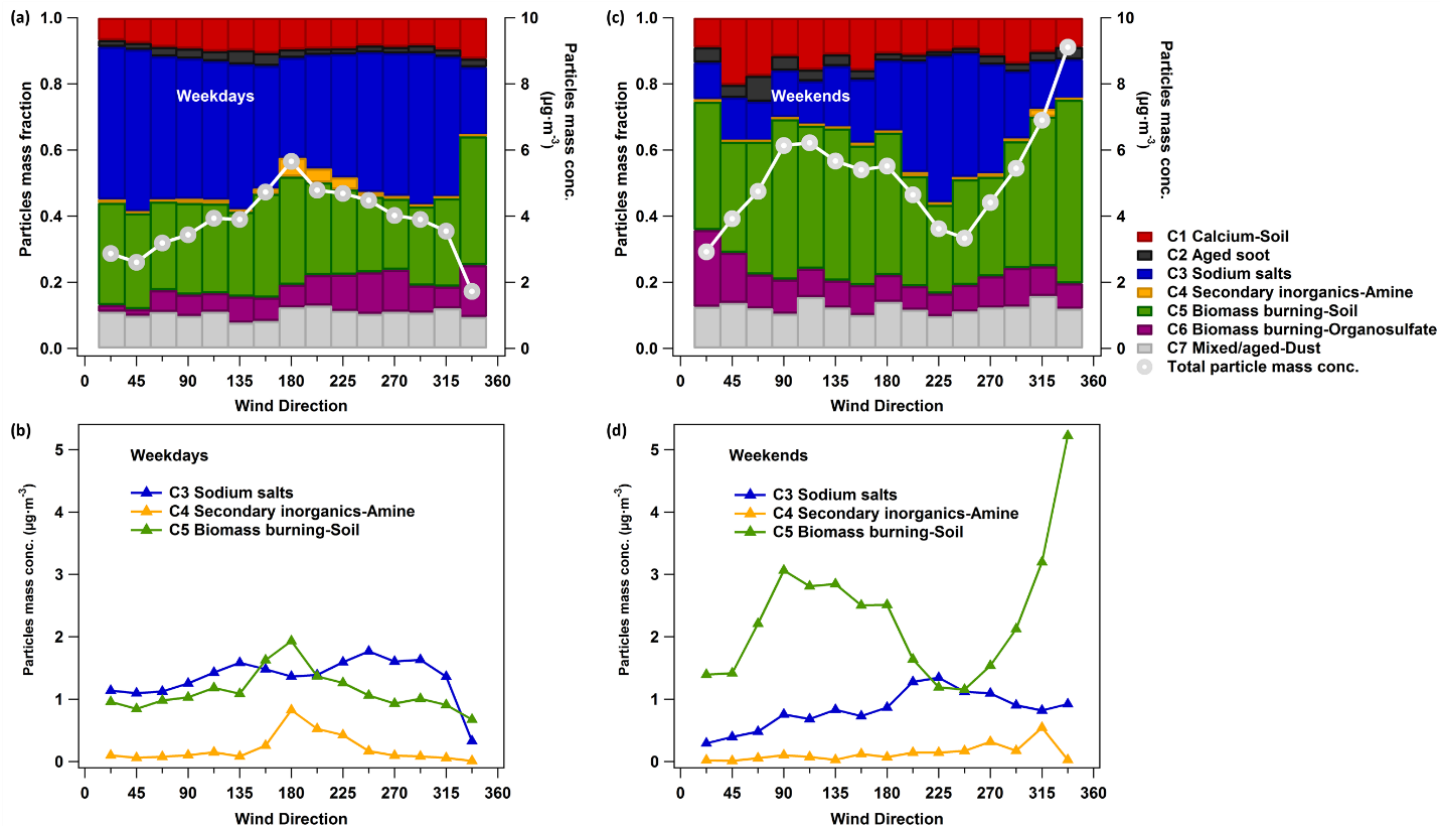
Figure S4: Five-factor-PMF solution with FPEAK=0 for organic compounds measured by AMS during the whole measurement time. (a) Mass spectra of the five components, i.e., HOA, SV-OOA, LV-OOA1, LV-OOA2, and LV-OOA3. Peaks are coloured with respect to different ion families. (b) Time series of the five components mass concentrations (left Y-axis) and their tracers or related ion family (right Y-axis). (c) Diurnal patterns of the five components. (d) Time series of the stacked components with the same data as panel (b), but the contribution of the five components can be visualized clearly. The percentages are their contributions to OA. The insert shows time series of SV-OOA, LV-OOA1 and LV-OOA2 during organic rich episode, and the two back arrows point out night-time and day time sub-peaks of the first LV-OOA1 peak. Note that the total OA value showed here is lower than the input OA for the PMF analysis. This is because the components identification might be interfered by extraneous variability, arising from some causes such as instrumental issues. Such extraneous variability will lead to disproportionate effects on the fitting outcome. Therefore, we need to downweight the corresponding variables in a proper way or even remove them, resulting the lower output total OA value reconstructed by the components.

The criteria for choosing factor number are given in section 2.2 of main manuscript. As shown in the Fig. S4 (a) HOA, SV-OOA, and LV-OOA have distinguishable spectra. For three LV-OOA, in spite of some similarity, distinguishable features can still be found, according to their time series (b) and diurnal patterns (c). In particular for the organic rich episode, LV-OOA1 and LV-OOA2 show anti-correlation, the LV-OOA3 signal is much weaker than the other two LV-OOA. Therefore, a five-factor solution was chosen rather than four or three-factor solution.

HOA is associated with POA, e.g., from urban emissions. In this study, the HOA (high H:C = 1.84 and low O:C = 0.12, herein) has a similar trend as m/z 57 (major contribution from C₄H₉) and NO₂ (not shown), with Pearson's correlation coefficients $\gamma = 0.5$ and 0.4, respectively. SV-OOA (O:C = 0.32) shows a very strong correlation with m/z 91 (mainly from C₇H₇) ($\gamma = 0.9$) and CH family ($\gamma = 0.8$). Three LV-OOA have prominent marker peaks at m/z 28 and 44, and all have high O:C ratio (> 0.7), which indicates highly oxidized or aged OA. LV-OOA1 shows strong correlations ($\gamma = 0.8$ to 1) with CH, CHO1, CHOgt1, CHN, CHO1N, and CHOgt1N. LV-OOA2 shows good correlations with CHOgt1N ($\gamma = 0.6$). LV-OOA3 shows no good correlation with any ion families, but a similar trend as sulfate ($\gamma = 0.5$).

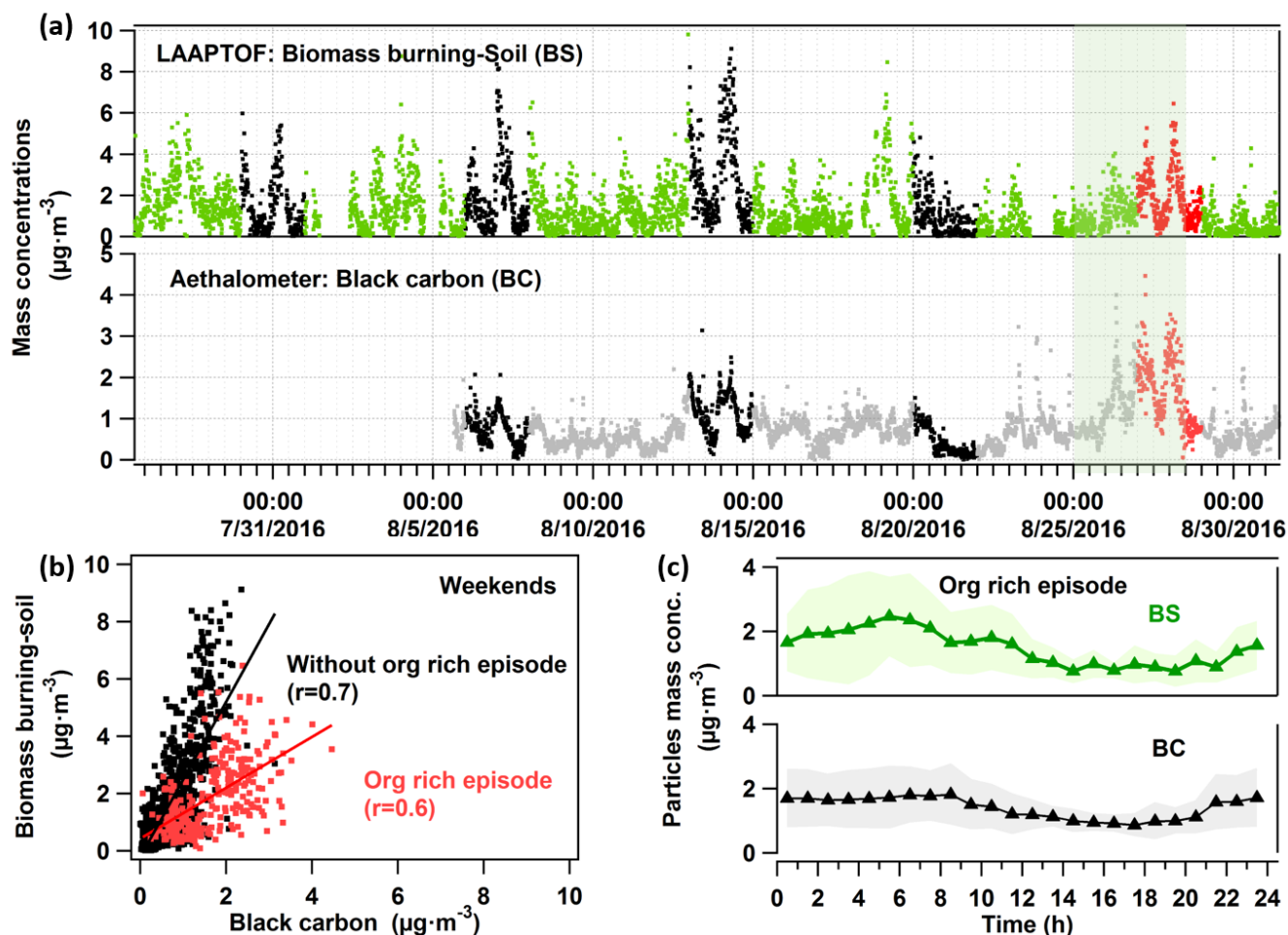
Similar elemental ratios for HOA, SV-OOA, and LV-OOA and their relationships with tracers such as HOA with NO_x and LV-OOA with secondary inorganics, have also been reported in previous AMS-PMF studies. Readers may refer to a review paper by Zhang et al. (2011).

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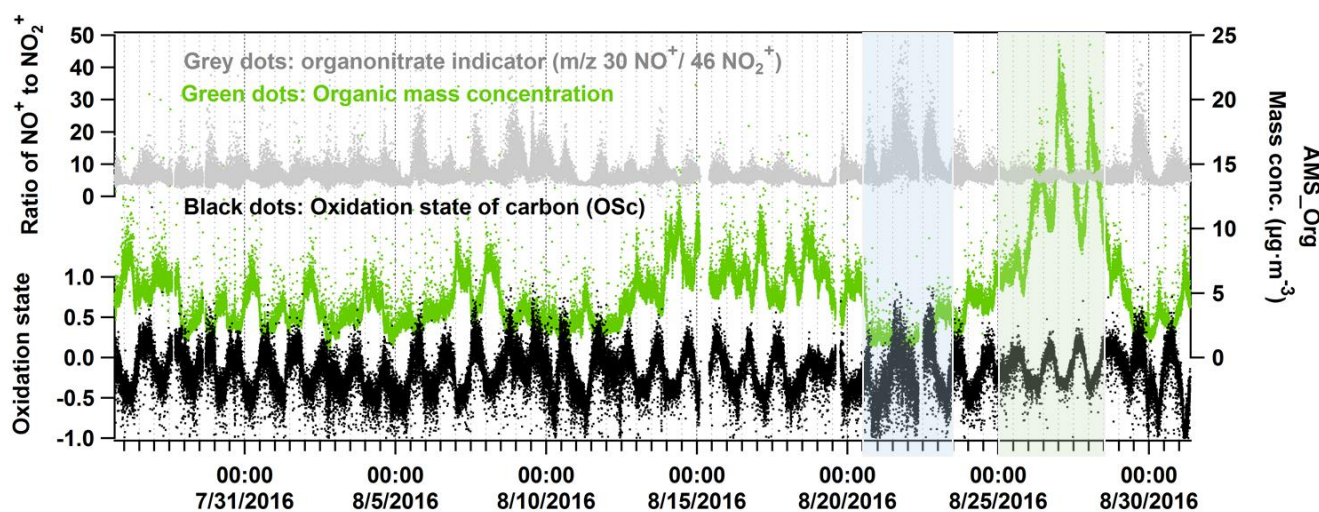


10 **Figure S5: Comparison of particle classes measured by LAAPTOF as a function of wind direction for weekdays (left) and weekends**
(right). The data is averaged over the entire campaign. The data comprises 4842 time points of 10 min averaged data, including 3702 for
weekdays and 1140 for weekends, including Saturday and Sunday. The total particle mass concentration for each wind direction is on
average $3.8 (\pm 2.3, \text{one } \sigma) \mu\text{g m}^{-3}$ and $5.2 (\pm 3.0) \mu\text{g m}^{-3}$ during weekdays and the weekends, respectively. For the dominating particle
classes the average mass concentrations are $1.3 (\pm 1.2) \mu\text{g m}^{-3}$ and $0.8 (\pm 0.5) \mu\text{g m}^{-3}$ for class 3, $0.2 (\pm 0.5) \mu\text{g m}^{-3}$ and $0.1 (\pm 0.2) \mu\text{g m}^{-3}$
for class 4, $1.1 (\pm 1.0) \mu\text{g m}^{-3}$ and $2.3 (\pm 1.7) \mu\text{g m}^{-3}$ for class 5, for weekdays and weekends respectively.

15



5 Figure S5S6: (a) Time series of aged-biomass-burning and soil-dust like particles and black carbon (black and red dots are weekend data); red dots are in organics rich episode. The shaded area marks the organics rich episode. (b) Comparison of mass concentration between biomass burning-soil and black carbon measured during the weekends. Black and red dots are the data without and with organic rich episode, respectively. (c) Diurnal pattern of particle mass concentration of biomass burning-soil (BS) and black carbon (BC) during organics rich episode, respectively.



10 Figure S6S7: Time series of organonitrate fraction indicator (grey dots) and oxidation state (black dots). The first (blue) and second (green) shaded areas mark the sodium chloride rich and organic rich episodes, respectively.

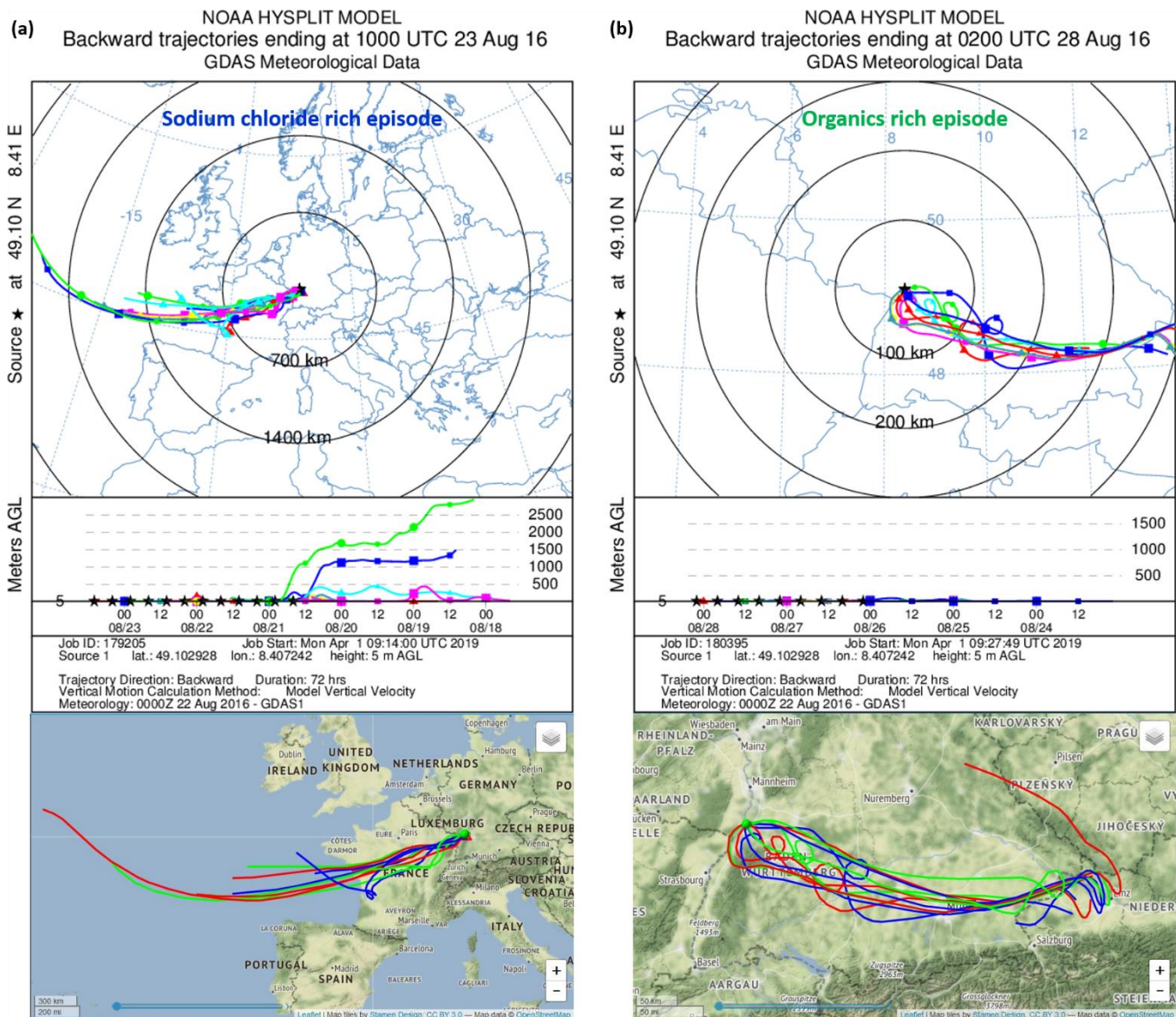


Figure S7S8: Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) back-trajectory analysis for 72 hours for sodium chloride rich (a) and organics rich (b) episodes, respectively.

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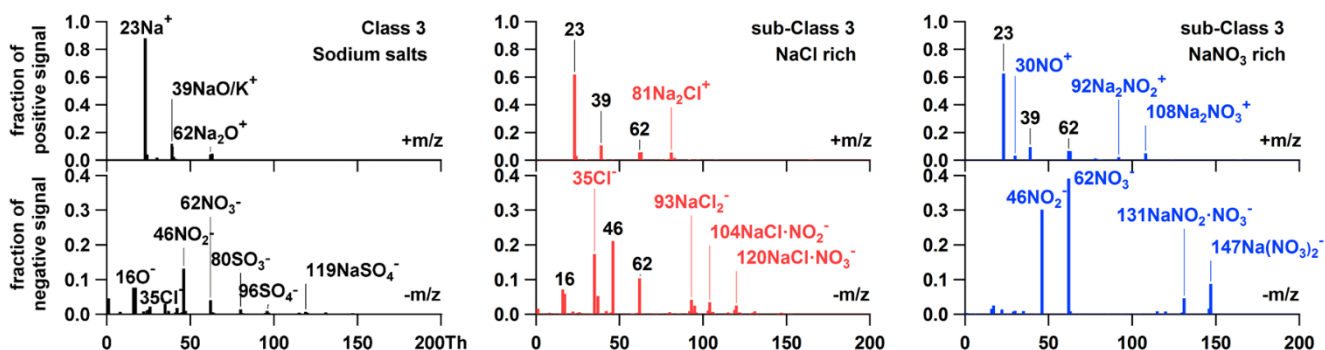
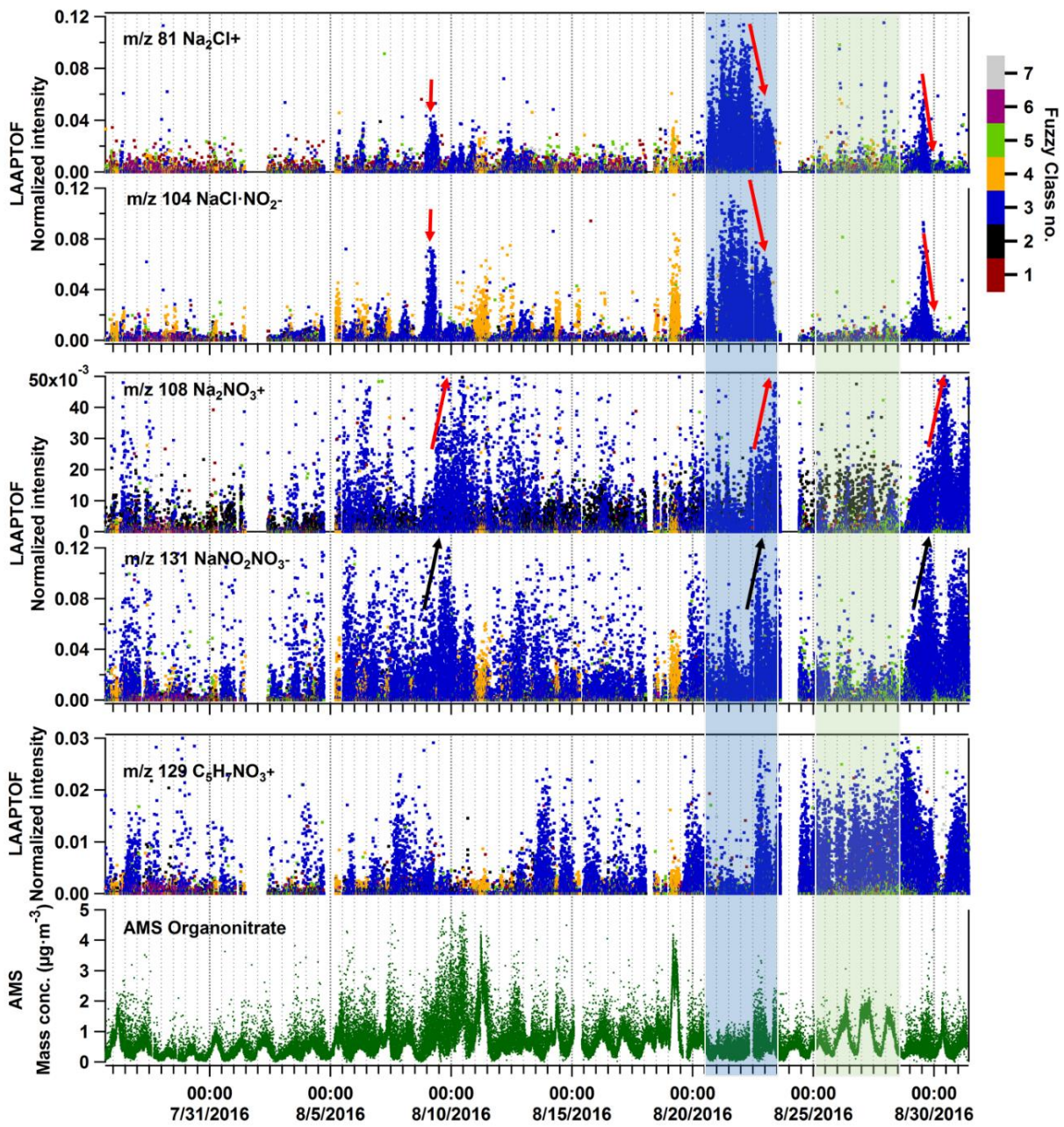


Figure S8S9: Spectra of the sodium salts class (Shen et al., 2019) and its sub-classes: NaCl rich (averaged from 1487 particles) and NaNO₃ rich (averaged from 300 particles).



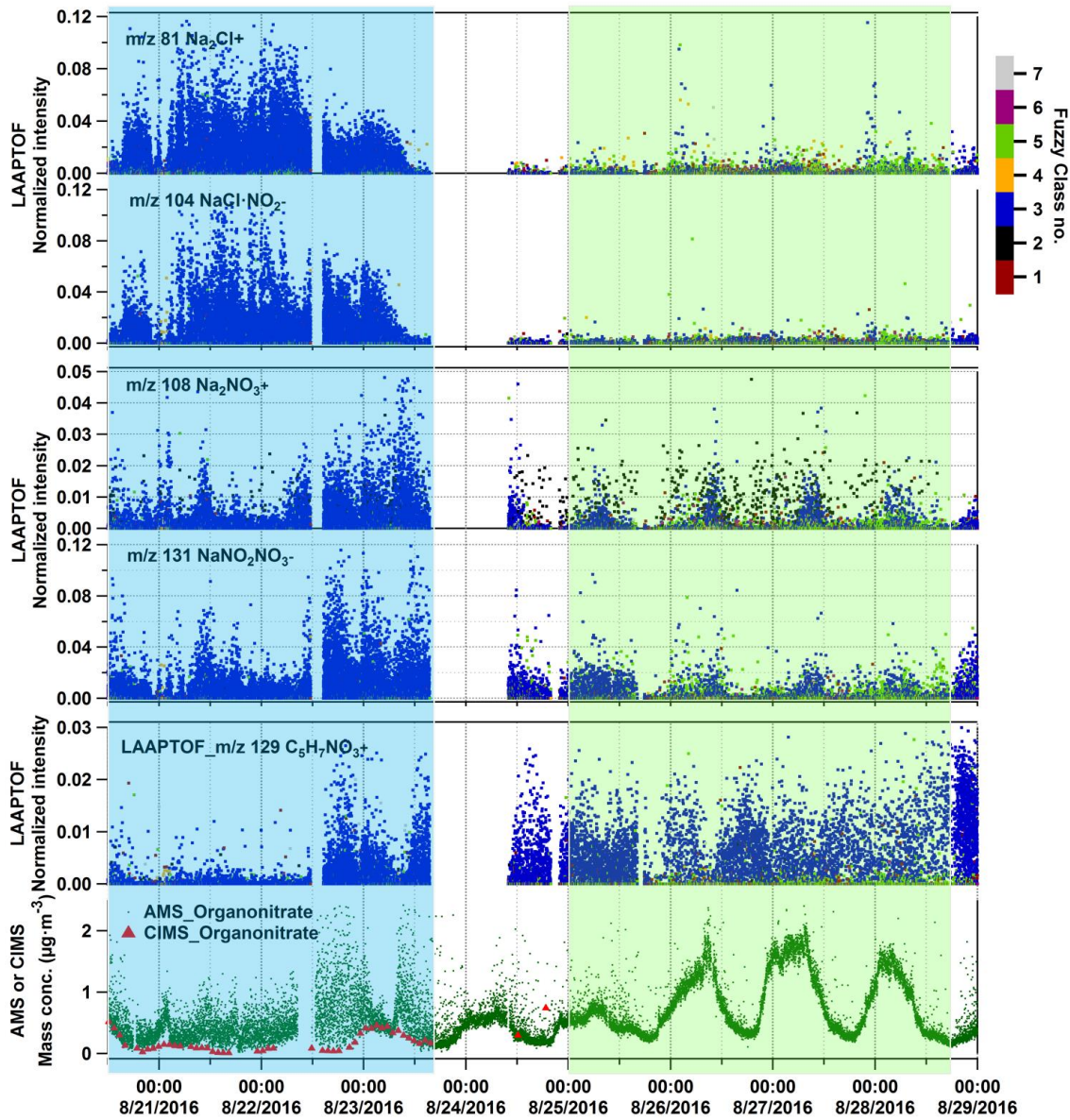


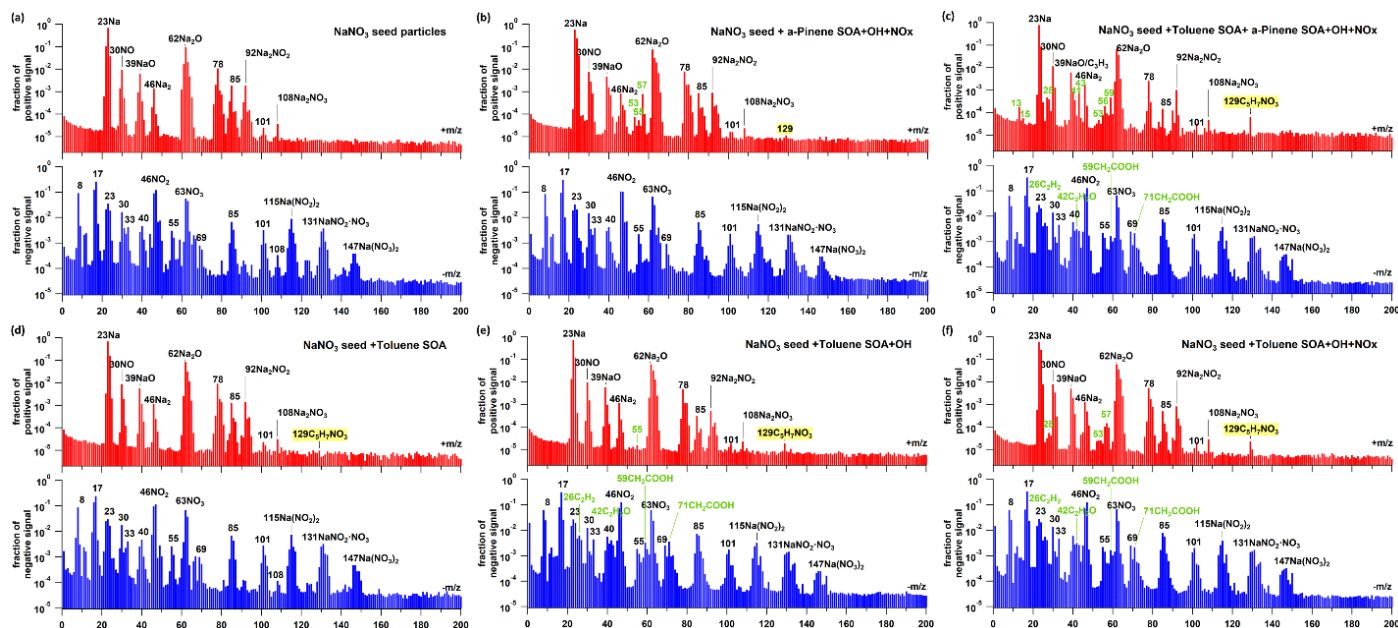
Figure S9S10: Time series of sodium chloride, and nitrate, and organonitrate signatures in LAAPTOF mass spectra, as well as organonitrate mass concentration derived from AMS and FIGAERO-HR-ToF-CIMS (stands for a filter inlet for gases and aerosols coupled to a high-resolution time-of-flight chemical ionization mass spectrometer, hereafter CIMS) measurement. For LAAPTOF results, Y-axis is the normalized ion intensity (each ion peak intensity is normalized to the sum of all ion signals; positive and negative ions were analysed separately). The seven classes are class 1: calcium-Soil; class 2: aged soot; class3: sodium salts; class 4: secondary inorganics-amine; class 5: biomass burning-soil; class 6: biomass burning-organosulfate; and class 7: mixed/aged-dust. Obvious sodium chloride and nitrate signatures are mainly found in class 3 labelled in blue colour. The first (blue) and second (green) bands mark the sodium chloride rich and organic rich episodes, respectively. AMS and CIMS results (Pearson's correlation coefficient $\gamma=0.52$) are adapted from our companion study by Huang et al (2019). It should be noted that during the sodium chloride rich episode (first band in blue) the fraction of organonitrates is higher (Fig. S7), although the organonitrate mass concentration is lower than that during the organic rich episode (second band in green).

The method to estimate organonitrate based on AMS measurement can be referred to Farmer et al. (2010) and the parameters used here are the same as that used by Huang et al. (2019). The result in Fig. S8S10 is based on our AMS measurements and the corresponding equation suggested by Farmer et al. (2010), as follows:

$$y = \frac{[(\text{Robs}-\text{Ran})(1+\text{Ron})]}{[(\text{Ron}-\text{Ran})(1+\text{Robs})]} \quad (S1)$$

where y is the estimated organonitrate mass concentration ($\mu\text{g}\cdot\text{m}^{-3}$), fraction of organic nitrate in the total nitrate measured by the AMS. Robs is the observed ambient ratio of $\text{NO}_2^+/\text{NO}^+$; Ran and Ron are the $\text{NO}_2^+/\text{NO}^+$ ratio for ammonium nitrate and

organonitrate, respectively. In our study $R_{an}=0.43$ based on the measurement and $R_{on}=0.1$ based on literature data. More details can be found by Huang et al. (2019).



5 **Figure S10S11:** Laboratory evidence for anthropogenic organonitrate (LAAPTOF signature m/z 129 $C_5H_7NO_3^+$). (a) to (f) are the averaged spectra from several thousand single particles, respectively. Y-axis is the normalized ion intensity (each ion peak intensity is normalized to the sum of all ion signals; positive and negative ions were analysed separately).

Regarding organonitrate, we have found some laboratory evidence for its potential origin. AIDA (aerosol interaction and dynamics in the atmosphere) simulation chamber studies have been done with α -pinene and/or toluene OH oxidations on $NaNO_3$ seeds in the presence/absence of NO_x . The corresponding single particle mass-spectra are shown in Fig. S9S11. It turns out that m/z 129+ is quite weak in α -pinene system even with NO_x (panel b), but it is much stronger after toluene is added (panel c). In the toluene system (refer to panel d to f), m/z 129+ is already present when $NaNO_3$ seeds are coated with toluene derived secondary organic aerosol (SOA) (panel d), likely due to the recombined fragments from toluene and nitrate. After adding extra OH, m/z 129+ is becoming stronger, likely due to the more aged toluene SOA- (panel e). When NO_x is added to toluene system, m/z 129+ becomes much stronger (panel ef), likely due to the organonitrate formation. Hence, we can conclude that m/z 129 $C_5H_7NO_3^+$ is more related to toluene rather than α -pinene, namely m/z 129 $C_5H_7NO_3^+$ could be a signature peak for anthropogenic sources in LAAPTOF spectra.

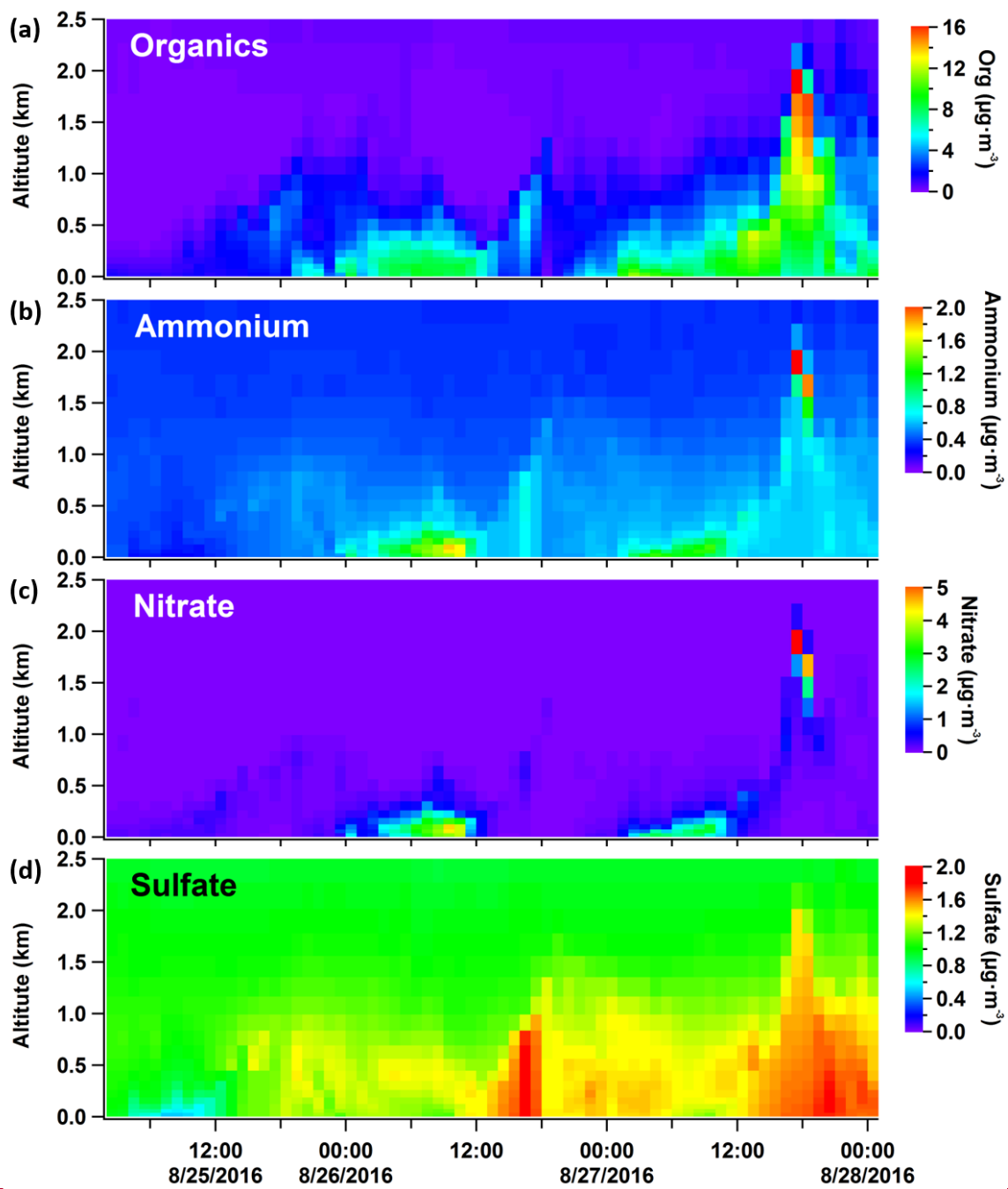


Figure S11

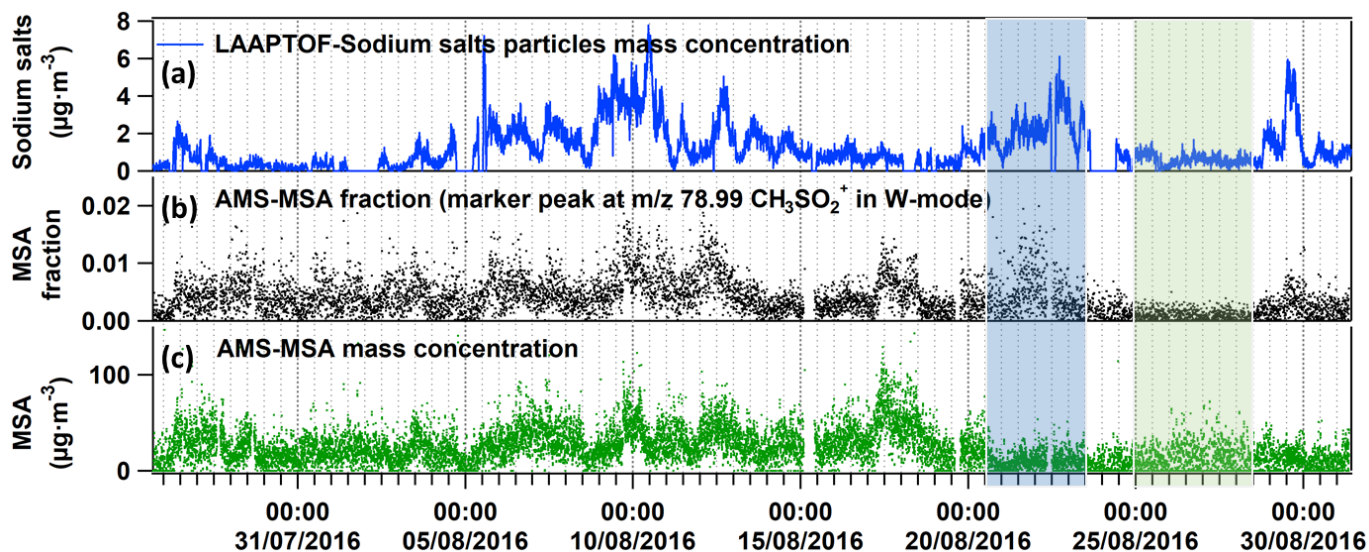


Figure S12: Time series of the estimated mass concentration of sodium salts particles measured by the LAAPTOF (a), methanesulfonic acid (MSA) fraction to total organics and sulfate measured by the AMS (b) and MSA mass concentration (c). The first (blue) and second (green) shaded areas mark the sodium chloride rich and organic rich episodes, respectively.

5

The method used to estimate MSA fraction (f_{MSA} , signal fraction of MSA to total org and sulfate) and its mass concentration can be referred to Huang et al. (2015). In brief, W-mode f_{MSA} is calculated and then used to multiply V-mode mass concentration of organics and sulfate, resulting estimated MSA mass concentration. The corresponding equations are as follows:

$$\sum I_{MSA} = \frac{I_{CH_3SO_2}}{9.7\%} \quad (S2)$$

$$f_{MSA} = \frac{\frac{\sum I_{MSA}}{RIE_{MSA}}}{\frac{\sum I_{Org}}{RIE_{Org}} + \frac{\sum I_{SO_4}}{RIE_{SO_4}}} \quad (S3)$$

$$C_{MSA} = f_{MSA} \times (C_{Org} + C_{SO_4}) \quad (S4)$$

10

where $\sum I_{MSA}$ is the total signal intensity of all MSA fragments measured in W-mode, which can be estimated based on the fragmentation patterns determined from the laboratory experiments (Huang et al., 2015). It was reported that the signal intensity of marker peak at m/z 78.99 ($CH_3SO_2^+$) accounted for $(9.7 \pm 1.6) \%$ of the total signal intensity of all MSA fragments. $\sum I_{Org}$ and $\sum I_{SO_4}$ are total signal intensity of organics and sulfate fragments measured in W-mode, respectively. Relative ion efficiency of MSA ($RIE_{MSA}=1.3$) is the averaged value from $RIE_{Org} (=1.4)$ and $RIE_{SO_4} (=1.2)$. C_{Org} and C_{SO_4} are mass concentrations calculated from V-mode data. C_{MSA} is the estimated MSA mass concentration.

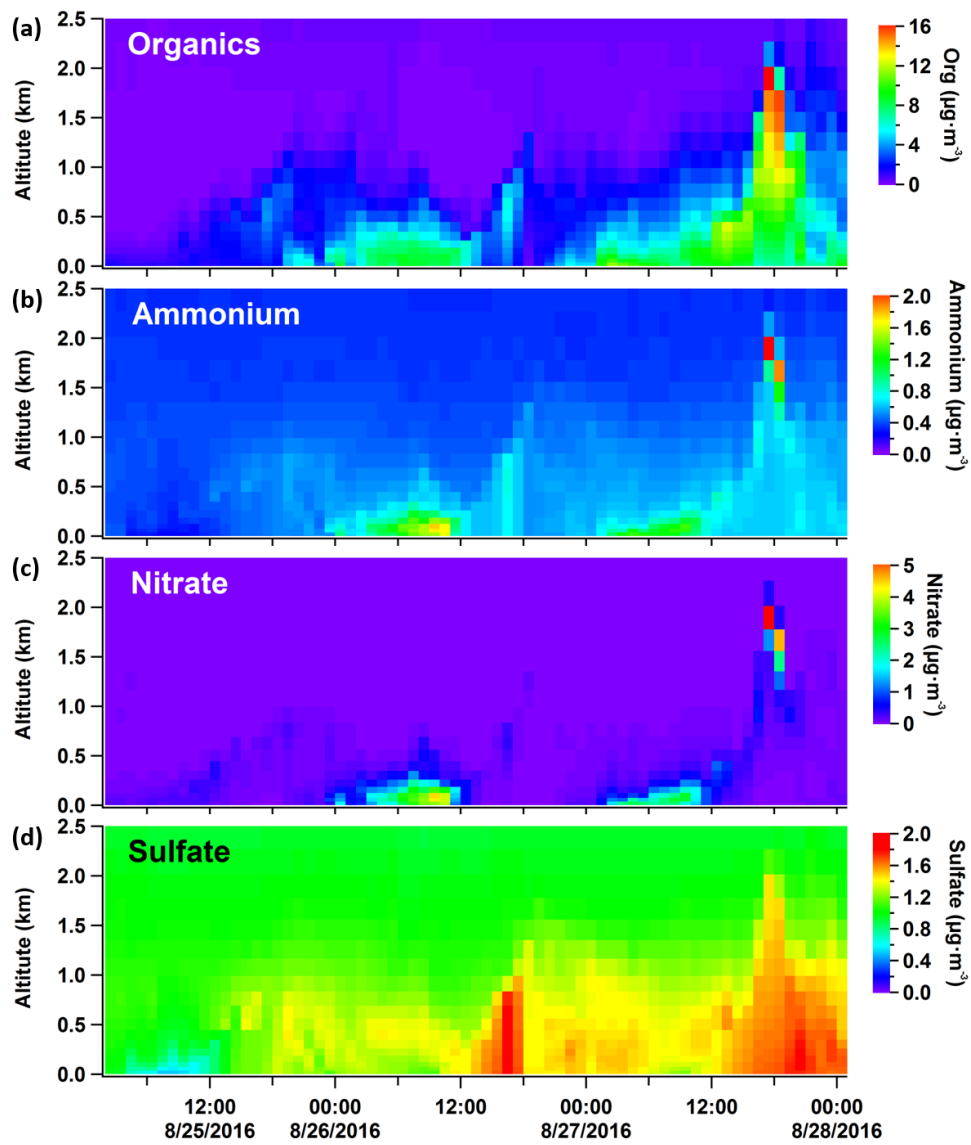


Figure S13: Time series of vertical profiles for organics, ammonium, nitrate, and sulfate as calculated by COSMO-ART.

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- 5 Huang, [D. D., Li, Y. J., Lee, B. P., and Chan, C. K.: Analysis of Organic Sulfur Compounds in Atmospheric Aerosols at the HKUST Supersite in Hong Kong Using HR-ToF-AMS, *Environ Sci Technol*, 49, 3672–3679, 2015.](#)
- [Huang, W., Saathoff, H., Shen, X. L., Ramakrishna, R., Leisner, T., and Mohr, C.: Chemical characterization of highly functionalized organonitrates contributing to high night-time organic aerosol mass loadings and particle growth, *Environ Sci Technol*, 53, 1165–1174, 2019.](#)
- 10 [Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Ulbrich, I. M., Ng, N. L., Worsnop, D. R., and Sun, Y. L.: Understanding atmospheric organic aerosols via factor analysis of aerosol mass spectrometry: a review, *Anal Bioanal Chem*, 401, 3045–3067, 2011.](#)
- 15 **Five supporting videos (.gif files) are available at KIT open data (<https://doi.org/10.5445/IR/1000094401>).**
- 1. SI_PN_10m.gif**
- Particle number concentrations and wind field calculated by COSMO-ART during high SO₂ period (from 20160826 0:00 to 13:00 (local time) for 10 m a.g.l.
- 2. SI_PN_122m.gif**
- 20 Particle number concentrations and wind field calculated by COSMO-ART during high SO₂ period from 20160826 0:00 to 13:00 (local time) for 122 m a.g.l.
- 3. SI_Org1_10m.gif**
- Particulate organic mass concentrations and wind field calculated by COSMO-ART from 20160825 17:00 to 20160826 18:00 (local time) for 10 m a.g.l.
- 25 **4. SI_Org2_10m.gif**
- Particulate organic mass concentrations and wind field calculated by COSMO-ART from 20160826 18:00 to 20160827 12:00 (local time) for 10 m a.g.l.
- 5. SI_Org3_10m.gif**
- 30 Particulate organic mass concentrations and wind field calculated by COSMO-ART from 20160827 12:00 to 20160828 01:00 (local time) for 10 m a.g.l.