

# Supporting Information

## Enhanced Heterogeneous Uptake of Sulfur Dioxide on Mineral Particles through Modification of Iron Speciation during Simulated Cloud Processing

Zhenzhen Wang<sup>1</sup>, Tao Wang<sup>1</sup>, Hongbo Fu<sup>1, 2, 3</sup>, Liwu Zhang<sup>1</sup>, Mingjin Tang<sup>4</sup>, Christian George<sup>5</sup>, Vicki H. Grassian<sup>6</sup>, and Jianmin Chen<sup>1</sup>

<sup>1</sup>Shanghai Key Laboratory of Atmospheric Particle Pollution and Prevention, Department of Environmental Science & Engineering, Institute of Atmospheric Sciences, Fudan University, Shanghai, 200433, China

<sup>2</sup>Shanghai Institute of Pollution Control and Ecological Security, Shanghai 200092, China

<sup>3</sup>Collaborative Innovation Center of Atmospheric Environment and Equipment Technology (CICAET), Nanjing University of Information Science and Technology, Nanjing 210044, China

<sup>4</sup>State Key Laboratory of Organic Geochemistry and Guangdong Key Laboratory of Environmental Protection and Resources Utilization, Guangzhou Institute of Geochemistry, Chinese Academy of Sciences, Guangzhou 510640, China

<sup>5</sup>University of Lyon, Université Claude Bernard Lyon 1, CNRS, IRCELYON, F-69626, Villeurbanne, France

<sup>6</sup>Departments of Chemistry and Biochemistry, University of California, San Diego, La Jolla, California 92093, United States

*Correspondence to:* Hongbo Fu (fuhb@fudan.edu.cn); Jianmin Chen ([jmchen@fudan.edu.cn](mailto:jmchen@fudan.edu.cn))

Page: 4.

Figure: 0.

Table 3.

**Table S1.** Properties of the standard clay and ATD samples

	IMt-2	NAu-2	SWy-2	ATD
Total Iron Content (% mass from Source Clays Repository Website)	5.55	26.19	2.58	-
Total Iron Content (% mass from XRF)	5.57	27.32	3.88	1.80
Total Iron Content (% mass from ICP-AES)	$5.45 \pm 0.04$	$26.30 \pm 0.57$	$2.36 \pm 0.56$	$1.48 \pm 0.56$
Specific Surface Area ( $\text{m}^2/\text{g}$ from $N_2$ -BET)	$20.1 \pm 1.5$	$19.8 \pm 1.3$	$22.6 \pm 2.3$	$4.3 \pm 0.3$

**Table S2.** Chemical composition of the standard clay and ATD samples

<b>Sample</b>	<b>SiO<sub>2</sub></b>	<b>Al<sub>2</sub>O<sub>3</sub></b>	<b>Fe<sub>2</sub>O<sub>3</sub></b>	<b>FeO</b>	<b>MgO</b>	<b>CaO</b>	<b>Na<sub>2</sub>O</b>	<b>K<sub>2</sub>O</b>	<b>TiO<sub>2</sub></b>	<b>P<sub>2</sub>O<sub>5</sub></b>	<b>Total</b>	<b>Total Fe</b>
<b>IMt-2</b>	59.57	19.47	7.95	0.05	2.42	0.37	0.08	8.72	0.99	0.07	99.75	5.57
<b>NAu-2</b>	54.99	2.81	39.03	n.d.	0.95	1.45	0.19	0.14	0.23		99.79	27.32
<b>SWy-2</b>	67.61	19.27	5.55	0.32	3.08	1.77	1.30	0.77	0.21	0.05	100.12	3.88
<b>ATD</b>	78.11	7.19	2.57		1.22	3.03	1.39	2.06	0.46		99.21	1.80

**Table S3.** Parameters of the Mössbauer spectra for the samples

		IS (mm/s)	QS (mm/s)	H <sub>f</sub> (T)	LW (mm/s)	A (%)	Fe species
IMt-2	D <sub>1</sub>	0.24	0.85	-	0.56	66.03	Al/Si, poor oct-Fe <sup>3+</sup>
	D <sub>2</sub>	1.30	2.40	-	0.44	33.97	Al/Si -Fe <sup>2+</sup>
IMt-2 after CP	D <sub>1</sub>	0.36	0.63	-	0.56	68.55	Al/Si, poor oct-Fe <sup>3+</sup>
	D <sub>2</sub>	1.13	2.78	-	0.46	31.45	Al/Si -Fe <sup>2+</sup>
NAu-2	D <sub>1</sub>	0.36	0.26	-	0.54	87.15	Al/Si, oct-Fe <sup>3+</sup>
	D <sub>2</sub>	0.41	1.22	-	0.21	12.85	Al/Si, X-Fe <sub>x</sub> O
NAu-2 after CP	D <sub>1</sub>	0.36	0.29	-	0.53	88.44	Al/Si, oct-Fe <sup>3+</sup>
	D <sub>2</sub>	0.41	1.20	-	0.26	11.56	Al/Si, X-Fe <sub>x</sub> O
SWy-2	D <sub>1</sub>	1.05	2.46	-	0.74	18.34	Al/Si-Fe <sup>2+</sup>
	D <sub>2</sub>	0.39	0.70	-	0.57	81.66	Al/Si-Fe <sup>3+</sup>
SWy-2 after CP	D <sub>1</sub>	0.88	2.34	-	0.82	17.09	Al/Si-Fe <sup>2+</sup>
	D <sub>2</sub>	0.36	0.61	-	0.58	82.91	Al/Si-Fe <sup>3+</sup>
ATD	D <sub>1</sub>	0.36	0.65	-	0.51	38.96	Al/Si-Fe <sup>3+</sup>
	D <sub>2</sub>	0.99	2.25	-	1.21	29.20	Al/Si-Fe <sup>2+</sup>
	S <sub>1</sub>	0.39	-0.13	51.1	0.47	31.84	α-Fe <sub>2</sub> O <sub>3</sub>
ATD after CP	D <sub>1</sub>	1.06	2.54	-	0.50	10.91	Al/Si -Fe <sup>2+</sup>
	D <sub>2</sub>	0.37	0.61	-	0.44	32.95	Al/Si-Fe <sup>3+</sup>
	D <sub>3</sub>	0.67	1.21	-	1.23	23.81	Fe <sup>3+</sup>
	S <sub>1</sub>	0.37	-0.21	51.0	0.52	32.33	α-Fe <sub>2</sub> O <sub>3</sub>