Dear Dr. Tsigaridis,

Thank you very much for the efficient handling of the reviewing process for our manuscript. We have carefully addressed all of the referees' comments. The corresponding changes in the manuscript are described below in our point-by-point responses to the referees' comments. We have also corrected some minor textual and stylistic inaccuracies, which were not indicated by the referees. All the changes and corrections are highlighted in the revised manuscript provided below. We hope that you will find the revisions satisfactory and sufficient.

Respectfully,

Igor Konovalov

on behalf of all the authors

Authors' response to the comments of the Anonymous Referee # 1

We thank the Referee for the positive evaluation of our manuscript. The useful comments and suggestions made by the Referee are highly appreciated and carefully addressed in the revised manuscript. Our point-by-point responses to the Referee's comments are described below.

Referee's comment: *The use of chemistry transport models to simulate a single plume seems to be off-scale. Would the authors comment on the uncertainties associated with this disadvantage?*

We agree with the Referee that chemistry transport models (CTMs) are not designed to simulate a single BB plume and should not normally be used for this purpose. However, in practice, CTMs are generally applied to situations where the actual spatial inhomogeneity of BB OA emissions is not resolved in the simulations. This is what we meant in our remark about simulations of BB OA evolution with CTMs in the introduction ("While three-dimensional chemistry transport models are intended to provide the best possible quantitative representation of the evolution of OA and its gaseous precursors…"). To make our point clearer and to address the Referee's comment, the following sentence has been included in the Introduction of the revised manuscript: "Note that while the spatial scales representative of isolated BB plumes are typically not resolved by chemistry transport models, simulations of a single BB plume with a box model can provide useful insights into possible uncertainties introduced by neglecting the spatial inhomogeneity of BB OA emissions in chemistry transport models at the sub-grid scales." The nature of these possible uncertainties is further discussed in Sect. 4.

Referee's comment: *Page 10: Would the authors comment on why they did not include NVSOA formation in MDMOA? Or is it the conventional OA scheme that was mentioned on page 8 line 6?*

The main reason why we disregarded condensed-phase processes and NVSOA formation in this study was briefly explained in the paragraph devoted to the description of the K15 scheme ("In view of the lack of robust knowledge about the condensed-phase processes (see also Section 4) and for consistency with the other numerical experiments performed in the present study, the transformation of SOA into NVSOA has been disregarded in our simulations."). In the revised manuscript, we provide a similar explanation but concerning the S15 scheme. Our choice of the FragSVSOA configuration for our experiments is further justified in the revised manuscript (specifically, in the description of the S15 scheme). In particular, we note that the FragSVSOA configuration (where NVSOA formation is disregarded) enables better consistency of the S15 scheme with the other VBS schemes considered in our study, and thus any differences between the simulations performed with the S15 scheme and the other schemes are easier to interpret. We also note that possible formation of NVSOA due to particle-phase reactions is among the factors (discussed more in detail in Sect. 4) that can affect the real BB OA evolution, but which were not analyzed in our study, as it is focused on identification of major qualitative nonlinear effects in the BB OA behavior due to gas-phase oxidation reactions in BB plumes.

Referee's comment: *Page 13 line 24-26: Would the authors specify in the text each of the size bin's range? Also, which 3 size bins were used for T18 and T18f?*

In the revised manuscript, we specified that in the experiments with the C17, K15, S15, and LIN schemes, the aerosol size distribution included 9 size bins covering the range from 20 nm to 10 µm and following a *geometric* progression with the common ratio of $500^{1/9}$ (~2.0), while the experiments with the T18 and T18f schemes were conducted using only 3 size bins that were defined to cover the same range (from 20 nm to 10 µm) using a *geometric* progression with the common ratio of $500^{1/3}$ (~7.9). We believe that given this information, a reader can easily evaluate each of the size bin's ranges.

Referee's comment: *Page 14 lines 3-7: "The concentration of OH .... based on the ambient measurements by Akagi et al. (2012), its value was set to 5×10^6 cm^{-3} in all our simulations. We also assumed a constant temperature of 298 K... " Is a plume environment equivalent to ambient conditions? e.g. is it valid to assume a constant temperature of 298K and ambient OH concentration in a plume? Would limited photochemistry within a plume reduce OH concentration?*

We are sorry for a somewhat misleading word "ambient" which we used to characterize the OH concentration measurements by Akagi et al. (2012). Actually, the OH concentration was indirectly measured by Akagi et al. inside of a BB plume. And indeed, the limited photochemistry within a plume is likely to reduce OH concentration. In the revised manuscript, our assumptions concerning the OH concentration and temperature and corresponding limitations of our study are explained more clearly. In particular, we note that the OH concentration within a plume can be affected by many factors (such as, e.g., the UV flux, the concentrations of nitrogen oxides and VOCs within the plume) which can cause variability of the OH concentration level across different plumes as well as temporal and spatial fluctuations of OH concentration within a given plume. We note further that temperature is also likely to vary, both spatially and temporally, within real-world BB plumes: in particular, it is likely to be lower in the upper part of a plume than near the surface. Finally, we argue that although all possible variability *and inhomogeneities* of the OH concentration and temperature were disregarded in our simulations, this limitation allowed us to isolate and investigate the internal dynamics of the BB OA system under fixed pre-defined conditions.

Referee's comment: *Page 14 lines 8: "Along with aerosol species, MDMOA has been configured to simulate the evolution of an inert tracer." Would the authors please clarify the concept and use of an inert tracer (it was initially mentioned in the abstract)? What is its composition and properties?*

The requested clarification concerning the use and properties of an inert tracer is provided in the revised manuscript (Sect. 2.3). In particular, we explain that the tracer is intended to represent the evolution of the BB OA mass concentration in a hypothetical situation where BB aerosol is composed of chemically inert and non-volatile components, and so the tracer was introduced in our model as a chemically inert species which can be affected only by the dilution process (since the dry and wet deposition processes were not considered in our simulations); for definiteness, the molecular weight of the tracer has been set to be the same as of carbon monoxide (CO). We note that the concept of analyzing the evolution of BB aerosol versus the evolution of an inert tracer (usually represented by CO) has been fruitfully exploited in many previous experimental and modeling studies of BB aerosol. The corresponding references are provided in the revised manuscript.

Referee's comment: *Page 19, Figure 4, perhaps setting both figures with the same y-axis scale would be helpful. Also, in the text, it's unclear why only T18 and T18f schemes are shown, but not other schemes.*

Figure 4 is redrawn with the same y-axis scale, as suggested by the Referee. We also provided an additional plot (in the same figure) showing the simulations with the C17 scheme. At the end of Sect. 2.3 of the revised manuscript, we explain that the hygroscopicity parameter was calculated only with the C17, T18 and T18f schemes because the other oxidation schemes (K15, S17 and LIN) considered in our study are not designed to evaluate the O:C ratio.

Referee's comment: *Page 24 "This observation indicates that the mass concentration of aged BB OA is likely to be much more strongly affected in the simulations by uncertainties in available representations of the BB OA evolution than the mass concentration of relatively fresh BB OA. One of the reasons is that fragmentation reactions become increasingly important with time when the SOG oxidation level increases, and then the competition between functionalization and fragmentation creates the more complex dependence on the plume parameters." Would the authors please elaborate on or quantify the competition between functionalization and fragmentation? e.g. the branching ratios?*

Indeed, the competition between functionalization and fragmentation can be quantified using the fragmentation branching ratio. To address the Referee's comment, we have modified one of the sentences cited by the Referee and included an additional sentence: "As a result of this competition, the outcome of the BB OA evolution becomes strongly dependent on the fragmentation branching ratio associated with a given OA scheme."

Referee's comment: *Page 32 lines 28-29: The authors mentioned that such differences among VBS are under the "typical conditions in summer mid-latitudes", it would certainly be intriguing for future studies to examine how these behaviors vary in different environments.*

We agree with the Referee and hope that future studies can use a similar approach to examine BB OA evolution under a wider range of environmental conditions. A corresponding remark is included in Conclusions.

Referee's comment: *Figures comments: It's unclear what the shaded greys represent in the figures, and they are quite distracting, I suggest that the authors justify them in the caption or remove them.*

The meaning of the shaded grey bands was explained in the caption for Figure 2 ("Shaded bands depict nighttime periods when no oxidation reactions were allowed to occur."). The same explanation is included in the captions for Figures 3, 4, and 10.

Referee's comment: *It may be already sufficient that figures have different colored lines. Adding different symbols are just adding noise to the figures (just as with the grey shades). But it could be a personal preference, just a suggestion.*

We appreciate the Referee's opinion, but we presume that some potential readers of our paper will prefer to print it out on a black and white printer that makes different colored lines indistinguishable. For such readers, we marked the curves with different symbols.

Referee's comment: *Again, it would be helpful to compare different schemes if the axis scales are the same when possible (in the same magnitude range).*

To address the referee's comment, we modified Figures 4, 6 and 7 (except for the panel b) accordingly. Note that not all of the panels have the same scale in other figures because our main intention was to clearly demonstrate the distinctive qualitative (rather than quantitative) features of the simulated BB OA evolution under different values of parameters.

Referee's comment: *For consistency among figures, I would suggest that Figure 4 follow other figures (Fig 3 etc) to include legends in the same locations.*

We thank the Referee for this reasonable suggestion. We have placed one of the figure legends (indicating a scheme) in Figure 4 into the left upper corner, similar to the legends in Figures 2, 3, and 5. Given the large size of the other legend in Figure 4, we preferred to show this legend in a separate panel. Most of the other figures have the legends placed uniformly into the right upper corner, as in many instances the left upper corner is filled with curves. We found that using the same layout for all the figures would not ensure the best visibility for our findings.

Referee's comment: *Figure 6's (and Figure 7's) caption mentions "dashed lines," although at first read (with-out looking at the legend's C_{tot}/C_0), it's unclear if it's the dashed lines with dots and dashes, or dashes alone (which should be C_{tot}/C_0). There two types of dashes, it'd be helpful to distinguish the two.*

We have made the dashed lines for C_{tot}/C_0 and dash-dot lines for "neutral" enhancements more distinguishable in the revised manuscript by increasing and decreasing the lengths of the dashes for the dash-dot and dashed lines, respectively.

Reference

Akagi, S. K., Craven, J. S., Taylor, J. W., McMeeking, G. R., Yokelson, R. J., Burling, I. R., Urbanski, S. P., Wold, C. E., Seinfeld, J. H., Coe, H., Alvarado, M. J., and Weise, D. R.: Evolution of trace gases and particles emitted by a chaparral fire in California, Atmos. Chem. Phys., 12, 1397–1421, https://doi.org/10.5194/acp-12-1397-2012, 2012.

Authors' response to the comments of the Anonymous Referee # 2

We are very grateful to the Referee for the positive evaluation of our manuscript and for the useful comments and suggestions. Our point-by-point responses to the Referee's comments are described below.

Referee's comment: *Part of the basis of this study is to build upon the Bian et al. (2017) model. Hodshire et al. (2019) also built upon the Bian et al. (2017) model to test in a similarly theoretical model the effects of fire (plume) size and background aerosol concentration on the near-field aging of aerosol size distributions. This paper may be worth mentioning within the intro and may provide as an appropriate citation for portions of the near-field discussions of EnR.*

We thank the referee for pointing out this omission: unfortunately, we were not aware of the very recent paper by Hodshire et al. prior to completion of our study and submission of our manuscript to ACP. The corresponding reference is provided in the revised manuscript, and the Hodshire et al. (2019) paper is mentioned there several times in various contexts. In particular, we note (in the Introduction) that Hodshire et al. (2019) pointed out a significant impact of background aerosol on near-field BB OA aging processes, and (in Sect. 4) that our findings concerning the impact of the plume size on EnR after a few initial hours of aging are qualitatively consistent with the results of numerical experiments conducted by Bian et al. (2017) and Hodshire et al. (2019). We also tried to make it clear (in Sect 2.3) that the configuration of the numerical experiments in our study is largely similar to that in both Bian et al. (2017) and Hodshire et al. (2019).

Referee's comment: *Page 9: In Table 1 for the S15 scheme, the authors reference '"fresh" SOG (β_{frag} = 0) and reactions involving "aged" SOG (β_{frag} = 0.85)': it's clear where β_{frag} = 0.85 come in from the text on pg 9 and equation 11. However, it's not clear where β_{frag} = 0 would be in equations 9 or 10, if one to assume that β_{frag}= 0 for the fresh SOG. Can the authors include this information within the relevant equations?*

We are sorry for this minor textual inconsistency. To address it, the description of the S15 scheme in Table 1 has been revised: we tried to make clear that a specific value of the fragmentation branching ratio is applicable only to oxidation reactions involving "aged" SOGs, while POGs and "fresh" SOGs are assumed to be not affected by fragmentation at all.

Referee's comment: *Page 10: Can the authors comment in the text on why they chose to only use the FragSVSOA configuration?*

A corresponding comment is provided in the revised manuscript. In particular, we note that the FragSVSOA configuration enables better consistency of the S15 scheme with the other VBS schemes considered in our study, and thus any differences between the simulations performed with the S15 scheme and the other schemes are easier to interpret. We also note that possible formation of NVSOA due to particle-phase reactions is among the factors (discussed more in detail in Sect. 4) that can affect the real BB OA evolution but that were not analyzed in our study, as it is focused on identification of major qualitative nonlinear effects in the BB OA behavior due to gas-phase oxidation reactions in BB plumes.

Referee's comment: *Page 13: "This period is representative of the typical lifetime of BB aerosol in Siberia under conditions without precipitation (Paris et al., 2009)": can the authors comment on the relative lifetime of BB aerosol in other important fire environments, such as the Amazon, Africa, etc, in order to place the 120 hour designation into a broader context? Also, as the authors are choosing variables representative of Siberia (e.g. diurnal cycle) chose to use 5 ug m^{-3} as their background aerosol concentration, can they comment on how well this is anticipated to represent the Siberian natural background during the fire season? The authors may also consider pointing out that only considering a relatively clean background is a limitation of the study, as entrainment of more polluted backgrounds will change the partitioning and evaporation rates of the plume particles (see e.g. Hodshire et al., 2019).*

In response to the Referee's comments, we have revised the second sentence of Sect. 2.3. We point out that the period of 120 hours has been chosen to be within the range of typical atmospheric lifetimes of submicron aerosol particles emitted from open vegetation fires in the major BB regions worldwide, as indicated, e.g., by a measurement-based estimate (5.1 days) of the lifetime of black carbon (BC) in Siberia (Paris et al., 2009) and global-model estimates (Wang et al., 2016) of the BC lifetimes for open fires in northern Africa (5.6 days) and northern South America (3.1 days).

In respect to our choice of the background OA concentration of *5* μg m^{-3}, we note that the same value was specified in the box model simulations performed by Bian et al. (2017) and that, for comparison, particulate matter (PM$_{10}$) in a boreal environment of central Siberia under background conditions (that is, without the detectable influence of local or regional pollution sources, including fires) was found by Mikhailov et al. (2017) to have concentrations ranging from about 2 to 10 μg m^{-3} in summer, being composed mostly of organic material. Finally, we also note that specifying a much larger or much smaller value of the background OA concentration would likely result in noticeable quantitative changes of the simulated BB OA behavior, since entrainment of background aerosol affects evaporation rates and gas-particle partitioning in a BB plume (Hodshire et al., 2019).

Referee's comment: *Page 15: The authors could consider using only the 'EnR' or the 'γ_a' notation throughout, as having 2 different variable names for the same thing is a little confusing.*

We have carefully considered the Referee's suggestion. We would like to note that as explained in the reviewed manuscript before Eq. (19), "EnR" is introduced as an abbreviation (rather than a notation) for the "enhancement ratio". A corresponding notation (γ_a) was introduced in Eq. (19) to allow us to present our quantitative results in a concise way. We hope that using both an abbreviation and a mathematical notation for the same physical characteristic is consistent with the standards of ACP. Nonetheless, to enhance the readability of the text, we tried to avoid (or at least, to minimize) using both γ_a and EnR in the same section of the revised manuscript. More specifically, the use of "γ_a" is predominately reserved for presentation of quantitative results of our simulations in Sect. 3, while "EnR" is mostly used to discuss qualitative implications of our findings in Sects. 4 and 5. We hope that in this way a possible confusion between "EnR" and "γ_a" has generally been avoided.

Referee's comment: *Section 3.1: Can the authors briefly justify in the text why they chose their given fixed value of initial mass loading for the analysis in Fig 2 and fixed value of plume size for the analysis in Fig. 3?*

A corresponding brief explanation in introduced in the first paragraph of Sect. 3.1 of the revised manuscript. Specifically, we note that the fixed values of C_0 (10^3 µg m^3) and S_p (5 km) in the simulations shown in, respectively, Fig. 2 and Fig. 3 are chosen to approximately represent mid-range values of the corresponding parameters (on a logarithmic scale).

Referee's comment: *Also, why does Figure 4 (and Figure 8) only show results for the T18 and T18f schemes? Perhaps along with a brief justification for this choice the authors could also consider including the results of the other VBS schemes in a supplemental figure. I see that the authors note on lines 9-10 of page 20 "Note that only simulations for the "extreme" values of C0 and Sp (among those considered in this study) are shown in Fig 4. Simulations with other (intermediate) parameter values would fall between the brown and blue curves." But does this include all other schemes, that is, that all of the VBS schemes used fall between the brown and blue curves? This should be made clear in the text.*

Following the Referee's suggestion, results of our simulations with the C17 scheme have also been included in Figures 4 and 8 of the revised manuscript. In addition, we explain (both in Sect. 2.3 and Sect. 3.1) that the hygroscopicity parameter was calculated only with the C17, T18 and T18f schemes because the other oxidation schemes (K15, S17 and LIN) considered in our study are not designed to evaluate the O:C ratio. We also tried to make clear that in the text fragment cited by the Referee, we mean the simulations performed using a particular scheme corresponding to each plot in Fig. 4.

Referee's comment: *The authors could help the reader by being more explicit as to why smaller fires have higher hygroscopicity parameter values than larger fires. A brief sentence or 2 would go well on pg 20 (first paragraph) that clearly points out again that smaller fires can undergo more oxidation reactions (more gas phase material available both from initial partitioning and initial evaporation by dilution). Plume-size dependency on e.g. oxidation and partitioning is a complex subject that could aid from simple explanations and reminders such as this throughout.*

Following the Referee's suggestion, a corresponding explanation is introduced in Sect. 3.1 of the revised manuscript.

Referee's comment: *Page 26: Can the authors comment in the text on whether there is information to say how realistic each β_{frag} value is?*

Unfortunately, we are not aware of any strong experimental or theoretical constraints to the β_{frag} values involved in the BB OA oxidation schemes. To address the Referee's comment, we note in the revised manuscript that simulations with the extreme values of β_{frag} (0 and 1) can hardly correspond to any realistic situation and are provided only for reference purposes, but taking into account that the fragmentation branching ratio is a highly uncertain parameter of VBS schemes, neither of the simulations with the other values of β_{frag} is intended to be more or less realistic.

Referee's comment: *It is worth pointing out somewhere within the text (possibly in the methods?) that a limitation of this study is that the largest fires may be have limited oxidation reactions and thus limited SOA formation and/or fragmentation occurring within the dense initial plumes if the plume is dense enough to limit photochemistry, and that this study doesn't try to account for that effect.*

The suggested comment (with an appropriate reference to Konovalov et al. (2016) where the effect mentioned by the Referee was addressed using a 3D model) is included in Sect. 2.3 of the revised manuscript.

Referee's comment: *For each figure that has a horizontal dashed line at EnR=1, the caption should state that that line is there to indicate where 'no mass enhancement' occurs.*

The suggested explanation is included in figure captions in the revised manuscript.

Referee's comment: *Figure 2: What are the shaded bands on panel d? Presumably this is not a designation of 'nighttime'. They may be present to guide the eye but are confusing and should be explained. I recommend either removing them or making them distinct from the nighttime bands in panels a-c.*

There should have been no shaded bands in panel d. We are sorry for this technical error that has been corrected in the revised manuscript.

Referee's comment: *Figure 6 and 7: It's difficult to see the dashed line for C_{tot}/C_o within some of the panels (e.g. Fig. 6 panel a); can this line be made more distinctive?*

Visibility of the dashed lines for C_{tot}/C_0 is improved in the revised manuscript by increasing their width.

Referee's comment: *Page 5 line 11: Bian et al., 2017 simulated 4 hours of aging (not 5).*

We thank the Referee for this correction. In the same sentence in the revised manuscript, we mention numerical experiments by Hodshire et al. (2019) along with those by Bian et al. (2017) and refer to the "first few hours" of BB OA evolution.

Referee's comment: *Page 8 line 17: are the SVOCs evenly distributed across the 5 bins? Please clarify.*

Actually, the SVOCs were not distributed evenly across the different volatility bins. To avoid possible confusion, the corresponding sentence has been revised. Instead of saying that the SVOCs were distributed across the 5 bins, we say that all SVOCs are represented using five volatility classes. The volatility distributions used in our experiments are specified in Sect. 2.3.

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Nonlinear behavior of organic aerosol in biomass burning plumes: a microphysical model analysis

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Abstract. Organic aerosol (OA) is a major component of smoke plumes from open biomass burning (BB). Therefore, adequate representation of the atmospheric transformations of BB OA in chemistry-transport and climate models is an important prerequisite for accurate estimates of the impact of BB emissions on air quality and climate. However, field and laboratory studies of atmospheric transformations (aging) of BB OA have yielded a wide diversity of observed effects. This diversity is still not sufficiently understood and thus not addressed in models. As OA evolution is governed by complex nonlinear processes, it is likely that at least a part of the observed variability of the BB OA aging effects is due to the factors associated with the intrinsic nonlinearity of the OA system. In this study, we performed a numerical analysis in order to gain a deeper understanding of ~~such~~ these factors. We employ a microphysical dynamic model that represents gas-particle partitioning and OA oxidation chemistry within the volatility basis set (VBS) framework and includes a schematic parameterization of BB OA dilution due to dispersion of an isolated smoke plume. Several VBS schemes of different complexity, which have been suggested in the literature to represent BB OA aging in regional and global chemistry-transport models, are applied to simulate BB OA evolution over a five-day period representative of ~~a~~ the BB aerosol lifetime in the dry atmosphere. We consider the BB OA mass enhancement ratio (EnR), which is defined as the ratio of the mass concentration of BB OA to that of an inert tracer and allows us to eliminate the linear part of the dilution effects. We also analyze the behavior of the hygroscopicity parameter, κ, that was simulated in a part of our numerical experiments. As a result, five qualitatively different regimes of OA evolution are identified, which comprise (1) a monotonic saturating increase of EnR, (2) an increase of EnR followed by a decrease, (3) an initial rapid decrease of EnR followed by a gradual increase, (4) an EnR increase between two intermittent stages of its decrease, or (5) a gradual decrease of EnR. We find that the EnR for BB aerosol aged from a few hours to a few tens of hours typically increases for larger initial sizes of the smoke plume (and therefore, smaller dilution rates) or for lower initial OA concentrations (and thus more organic gases available to form secondary OA). However, these dependencies can be weakened or even reversed, depending on the BB OA age and on the ratio between the fragmentation and functionalization oxidation pathways. Nonlinear behavior of BB OA is also exhibited in the dependencies of κ on the parameters of the plume. Application of the different VBS schemes results in large quantitative and qualitative differences between the simulations, although our analysis suggests also that the main qualitative features of OA evolution simulated with a complex two-dimensional VBS scheme can also be reproduced with a much simpler scheme. Overall, this study indicates that the BB aerosol evolution may strongly depend on parameters of the individual BB smoke plumes (such as the initial organic aerosol concentration and plume size) that are typically not resolved in chemistry transport models.

1 Introduction

Atmospheric aerosol is known to play an important role as a climate driver on global and regional scales and to adversely affect human health. A large fraction of the aerosol mass is typically represented by organic components forming liquid, amorphous, or glassy particulate matter, which here is referred to as organic aerosol (OA). As a climate forcer, OA scatters solar radiation and provides cloud condensation nuclei, thus directly and indirectly contributing to cooling of the atmosphere on the global scale (IPCC, 2013; Lelieveld et al., 2019), although part of it, so called brown carbon, can absorb sunlight, thus contributing to warming (see, e.g., Andreae, and Gelencsér, 2006; Feng et al., 2013; Jo et al., 2016). On a regional scale, of particular significance is the cooling effect of OA on climate in the Arctic (Sand et al., 2015), opposing the rapid increase of surface temperature that has been observed in recent decades (Bekryaev et al., 2010). As an agent of air pollution, OA constitutes a considerable fraction of fine particulates ($PM_{2.5}$) (Jimenez et al., 2009) that causes human health disorders and premature deaths (Pope et al., 2009; Burnett et al., 2018; Lelieveld et al., 2019). However, as evidenced by the large differences between the OA atmospheric budgets evaluated with different models and also by considerable discrepancies between simulations and observations of OA (see, e.g., Tsigaridis et al., 2014; Bessagnet et al., 2016; Tsigaridis and Kanakidou, 2018), the current knowledge of the OA sources and its atmospheric transformations of OA is still deficient, and corresponding modeling representations are very imperfect.

Open biomass burning (BB), i.e., vegetation fires and agricultural waste burning, provides a major source of OA on the global scale. Specifically, it has been estimated that BB emissions of primary OA (POA), which typically constitutes the predominant fraction of BB aerosol, contribute about 70 % of total POA emissions (Bond et al., 2013). In recent years, numerous studies have been aimed at investigating and modeling sources (e.g., May et al., 2013; Jathar et al., 2014; Konovalov et al., 2015; van der Werf et al., 2017), radiative effects (e.g., Saleh et al., 2013; 2015; Archer-Nicholls et al., 2016; Pokhrel et al., 2017; Yao et al., 2017), and atmospheric transformations (e.g., Cubison et al., 2011; Jolleys et al., 2012; Forrister et al., 2015; Shrivastava et al., 2015; Konovalov et al., 2015; 2017; Tsimpidi et al., 2018; Theodoritsi and Pandis, 2019) of BB OA and its components.

Both laboratory experiments and ambient observations suggest that the mass concentration of BB OA may undergo major, yet highly diverse, changes as a result of its aging under typical atmospheric conditions. These changes are commonly evaluated by means of the BB OA mass enhancement ratio (EnR), which is usually defined as the normalized ratio of the BB OA mass concentration to the concentration of an inert BB tracer. This normalization makes the EnR for freshly emitted aerosol equal to unity. In particular, considerable increases (in many cases exceeding a factor of 2) of EnR were found in smog chamber experiments after a few hours of photochemical aging of smoke from wood or grass burning (e.g., Grieshop et al., 2009; Hennigan et al., 2011; Tiitta et al., 2016; Ciarelli et al., 2017a; Fang et al., 2017; Ahern et al., 2019), although there has been a large diversity between results of individual chamber experiments. As a result of aircraft experiments conducted in North America around Mexico City and on the Yucatan Peninsula, significant increases in EnR have been reported by DeCarlo et al. (2008) and Yokelson et al. (2009) for aging BB plumes (from about 30 % up to a factor of 2). Konovalov et al. (2015) identified a substantial increase (by a factor of 2) in the enhancement ratio for mass concentration of particulate matter in smoke plumes after 1-2 days of transport over regions of Eastern Europe. A similar major increase in the enhancement ratio for BB aerosol mass concentration, but over about 15 hours, of photochemical oxidation of BB plumes, was deduced from an analysis of satellite measurements of aerosol optical depth (AOD) over Siberia (Konovalov et al., 2017). Based on several years of continuous measurements of BB OA in an African savannah, Vakkari et al. (2018) found that EnR more than doubles on average after three hours of daytime aging. However, there is also evidence that EnR may decrease or remain nearly constant in aging smoke plumes. For example, based on aircraft measurements, Akagi et al. (2012) identified a sharp decrease of EnR during the first hour after emissions. Using data from several field campaigns conducted in Australia, North America, and western Africa, Jolleys et al. (2012; 2015) found that the BB OA enhancement

ratios in highly aged BB plumes (typically transported between three and six days before the measurements were taken) were consistently smaller than those in the fresh plumes. The aforementioned analysis of satellite data (Konovalov et al., 2017) suggested evidence for a gradual decrease of EnR after its initial strong increase. At the same time, several observational studies (e.g., Capes et al., 2008; Brito et al., 2014; Sakamoto et al., 2015; May et al., 2015; Zhou et al., 2017) did not reveal any significant net changes of EnR in aged BB plumes.

Numerous studies reported major changes in the chemical composition of BB OA due to its aging (e.g., DeCarlo et al., 2008; Cubison et al., 2011; Pratt et al., 2011; Jolleys et al., 2012; 2015; Brito et al., 2014; May et al., 2015; Bertrand et al., 2018; Lim et al., 2019) regardless of whether or not significant net changes were detected in the BB OA mass concentration. On the one hand, BB OA aging has been found to be typically associated with a rapid decay (over a period of a few hours under typical atmospheric conditions) of some key chemical compounds contributing to POA (such as, e.g., levoglucosan): using aerosol mass spectrometry, such a decay can be inferred from a decrease of the mass fragment signatures at m/z 60 (e.g., May et al., 2015) as well as from a more comprehensive analysis at the molecular level (Bertrand et al., 2018). On the other hand, atmospheric processing of BB OA has been reported to result in strong enhancements of the oxidation state of the organic matter. The increases of the O:C ratio (due to addition of, e.g., alcohol and carbonyl groups) are usually inferred from increases in the mass spectrometric signal at m/z 44 (e.g., Brito et al., 2014; May et al., 2015; Fang et al., 2017) and can be indicative of secondary organic aerosol (SOA) formation. Note that changes in the chemical composition and oxidation state of OA particles can affect their hygroscopic and optical properties (e.g., Lambe et al., 2011; Adler et al., 2011; Akagi et al., 2012; Fan et al., 2019), which need to be adequately specified in chemistry-transport and climate models.

Useful insights into the possible reasons behind the large variability of the EnR ~~values~~ trends reported earlier for aging BB aerosol have been provided by recent smog chamber experiments (Ahern et al., 2019; Lim et al., 2019) that revealed a strong dependence of SOA formation on variable initial concentrations of organic gases. These experiments, however, do not rule out the possibility that there are some other factors contributing significantly to the observed diversity of changes in EnR during the atmospheric lifetime of BB aerosol. In view of significant nonlinear interactions of the processes affecting properties, formation and evolution of SOA (Shrivastava et al., 2017), it seems reasonable to expect that the diversity of observational findings concerning BB aerosol atmospheric aging can in part be due to nonlinear behavior of OA transformations. By ~~the~~ nonlinear behavior, we mean here any manifestations of a dependence of the relative rate of change of OA mass concentration at a given moment of time on the mass concentration of OA itself at this or previous moments.

In this study, we investigate qualitative nonlinear features of the behavior of OA within an isolated BB plume and attempt to reconcile some of the diverse observational findings concerning BB aerosol aging effects from a theoretical viewpoint. To this end, using some routines and interfaces of the CHIMERE chemistry transport model (Menut et al., 2013), we developed and employed a microphysical dynamic (box) model of organic aerosol (MDMOA). While three-dimensional chemistry transport models are intended to provide the best possible quantitative representation of the evolution of OA and its gaseous precursors from various anthropogenic and natural sources, the principal purpose of MDMOA is to isolate and simulate, under fixed ambient conditions, the effects of key processes responsible for chemical and physical transformations of OA from other complex processes affecting evolution of OA in the real atmosphere (such as mixing with aerosols and their gaseous precursors from multiple sources, vertical advection, dry and wet deposition, in-cloud processing, etc.). In this sense, our study is similar to several previous studies employing box models to study OA processes (e.g., Camredon et al., 2007; Lee-Taylor et al., 2011; 2015; Lannuque et al., 2018). Note that ~~since~~ while the spatial scales representative of isolated BB plumes are typically not resolved by chemistry transport models, simulations of a single BB plume with a box model can provide useful insights into possible uncertainties introduced by neglecting the spatial inhomogeneity of BB OA emissions in chemistry transport models at the sub-grid scales. ~~Note that~~ Furthermore, compared to smog chamber and dedicated field studies, a box model analysis enables a much more comprehensive examination of the parameter space of the BB OA system.

It has been proposed that complex atmospheric transformations of OA (regardless of its origin), including SOA formation, can be adequately represented in chemistry transport models within the volatility basis set (VBS) modeling framework (Donahue et al., 2006; 2011; 2012a; Robinson et al., 2007). This framework has been implemented in MDMOA. The VBS method involves splitting semi-volatile organic compounds (SVOCs) and the more volatile intermediate-volatility organic compounds (IVOCS), into several classes with respect to volatility and applying the absorptive partitioning theory (Pankow et al., 1994) to distribute the organic compounds between gas phase and particles. The SVOCs and IVOCs can also be distributed between several model types, depending, e.g., on their oxidation state (O:C ratio), origin (e.g., primary or secondary, anthropogenic or biogenic, etc.), and photochemical age (Donahue et al., 2012a,b; Shrivastava et al., 2013; Tsimpidi et al., 2018). Representing the processes involving SVOCs and IVOCs within the VBS framework has been shown to allow improving the performance of simulations of OA from vegetation fires with respect to simulations using the "conventional" OA modeling framework, in which these processes are basically disregarded and only specific volatile organic compounds (VOCs) are considered as precursors of SOA (Hodzic et al., 2010; Shrivastava et al., 2015; Konovalov et al., 2015; 2017). Based on simulations using the VBS method, it has also been argued (Konovalov et al., 2015; 2017) that disregard for the BB OA aging processes might be one of the main reasons for a strong underestimation of aerosol optical depth in BB plumes by chemistry transport models using the conventional representation of OA evolution (e.g., Tosca et al., 2013; Konovalov et al., 2014; 2018; Reddington et al., 2016; Petrenko et al., 2017). It should be noted, however, that the representation of the BB OA evolution within the VBS framework in chemistry transport models is still associated with major uncertainties: while a variety of VBS schemes of different complexities have been suggested for BB OA modeling (e.g., Grieshop et al., 2009; Koo et al., 2014; Shrivastava et al., 2015; Ciarelli et al., 2017a; Tsimpidi et al., 2018), any of these schemes have only partially been constrained by laboratory or ambient measurements. In view of these uncertainties, we performed our analysis using several different available VBS schemes.

Based on simulations of the first ~~five~~few hours of BB OA evolution with a similar microphysical box model, Bian et al. (2017) and Hodshire et al. (2019) showed that apart from oxidation, evaporation and condensation of SVOCs, BB OA dynamics is strongly affected by the dilution process. Hodshire et al. (2019) also pointed out a significant impact of the background aerosol on near-field BB OA aging processes. Accordingly, both dilution and entrainment of background aerosol are ~~is~~ also taken into account in our model (although investigating the role of the latter process is not the focus of this study). Following Bian et al. (2017) and Hodshire et al. (2019), we approximate the dilution rate as a function of the initial plume size by using the formulations of the stationary Gaussian dispersion model and analyze the dependence of the BB OA mass enhancement ratio on the initial plume size, which controls the dilution rate. However, we considerably extend the period of analysis (up to five days), and instead of the simple single-step oxidation scheme that was used by Bian et al. (2017) to analyze atmospheric implications of short-run smoke chamber experiments and by Hodshire et al. (2019) to investigate the near-source relationships between parameters of diluting BB plumes and BB OA physical properties~~study the atmospheric implications of their results~~, we use several multi-step oxidation schemes that have been suggested for modeling of BB OA specifically within chemistry transport models. Along with the dynamics of EnR, we consider the evolution of the hygroscopicity parameter, κ, which is commonly used to characterize the uptake of water by aerosol particles and their cloud condensation nucleus (CCN) activity (e.g., Petters and Kreidenweis, 2007; Chang et al., 2010; Mikhailov et al., 2015). The evolution of the hygroscopicity parameter, however, is not the main focus of this study.

By using MDMOA, we address the following questions. What are the manifestations of the OA system's nonlinearity in the dependencies of EnR and κ on the initial size and initial density of a smoke plume? Can variability of the parameters of the plume lead to qualitatively different types of BB OA evolution? Can differences between available VBS schemes be associated with qualitatively different responses of EnR to variations in the plume's parameters? It should be emphasized that our simple model and its application in this study are not intended to reproduce any realistic scenarios of atmospheric evolution of BB OA in a quantitatively accurate way. Instead, we focus our analysis on identification of possible qualitative

features of the BB OA behavior, which may have a sufficiently general character. We believe that the results of this kind of analysis can be useful as theoretical guidance for ~~corresponding~~ future experimental studies and for improving parameterizations of BB OA processes in chemistry transport models.

2 Model and method description

2.1 Microphysical dynamic model of organic aerosol (MDMOA): dynamic equations

The CHIMERE based box model MDMOA is intended to represent the following processes: (1) growth and evaporation of multi-disperse particles of OA due to partitioning of SVOCs between gas phase and particles, (2) gas-phase oxidation of VOCs, IVOCs and SVOCs, and (3) atmospheric dilution of OA. The model also includes a representation of coagulation, but this process has not been taken into account in the present study. MDMOA has been developed by adopting and modifying several modules of the CHIMERE chemistry transport model (Menut et al., 2013), including the routines implementing the Gauss-Seidel iteration scheme (Verwer et al., 1994) to solve a set of dynamic equations, a sectional representation of the OA mass absorption and evaporation processes (Gelbard and Seinfeld, 1980), and some model interfaces facilitating modifications of ~~a~~ the simulation configuration as well as providing simulation outputs in a convenient NetCDF format. Dynamic mass transfer equations for a semi-volatile species, s, in a particle size section, l, are formulated as follows:

$$\frac{dC_s^l}{dt} = \frac{2}{3}\pi d_p^l c\lambda F N_p^l \left(C_s^g - \mathrm{K}C_s^{eq}\right) + I_{l-1}^l + I_l^{l+1},$$ (1)

where C_s^l is the condensed-phase mass concentration, d_p^l is the particle diameter, c and λ are the mean velocity and free path of the organic molecule in the air, F is the Fuchs-Sutugin correction factor, N_p^l is the number of particles in the size bin l, C_s^g is the instantaneous gas-phase concentration, K is the Kelvin effect factor, C_s^{eq} is the equilibrium gas-phase concentration, and I_{l-1}^l and I_l^{l+1} are the intersectional fluxes between the bins l-1 and l and between the bins l and l+1.

The molecular mean free path, the Fuchs-Sutugin correction factor and the Kelvin effect factor are evaluated using the conventional formulations (Seinfeld and Pandis, 2016):

$$\lambda = 3Dc^{-1},$$ (2)

$$F = \frac{1+Kn}{1+0.3773Kn+1.33Kn\left(\frac{1+Kn}{\alpha}\right)},$$ (3)

$$\mathrm{K} = \exp\left(\frac{4\sigma_p MW_s}{RT\rho_p d_p^l}\right),$$ (4)

where D is the molecular diffusion coefficient, Kn is the Knudsen number (Kn=$2\lambda/d_p^l$), α is the mass accommodation coefficient (~~that~~ which is assumed to be unity in all our simulations), σ_p and ρ_p are the surface tension and density of the particle material, MW_s is the molecular weight, R is the ideal gas constant, and T is temperature. Following the basic formulations for the VBS framework, the gas-phase equilibrium concentration is expressed through the total mass concentration of SVOCs, C^{tot}, the mass fraction of a given species, f_s, the total mass concentration of OA particles, C_{OA}, and the saturation concentration C_s^*:

$$C_s^{eq} = C_s^* \frac{f_s C^{tot}}{C_{OA}}\left(1 + \frac{C_s^*}{C_{OA}}\right)^{-1}.$$ (5)

Note that the formulation of the mass flux term for transfer of SVOCs from and into the gas phase in Eq. (1) is essentially the same as that in the kinetic model used by May et al. (2013) to derive the volatility distributions for BB POA. Following May et al. (2013), we also assumed, for definiteness, that the diffusion coefficient, surface tension, and the particle bulk density are equal to 5×10^{-6} m^2 s^{-1}, 0.05 N m^{-1} and 1.2×10^3 kg m^{-3}, respectively.

The intersectional fluxes are calculated according to Gelbard and Seinfeld (1980) as a combination of the weighed mass fluxes between the gas and particle phases for the bins l-1, l, and l+1. A concrete representation of the intersectional fluxes is not of significance in this study since they cannot, by definition, contribute to the mass balance (their sum over the all particle size bins equals zero), and we do not consider here the evolution of the particle size distribution. Furthermore, as argued below, the equilibration time scales determined by Eq. (1) are typically much smaller than the time scales associated with oxidation of SVOCs and so our simulations are not sensitive to the particle size distribution.

The dynamics of the total concentration (both in the gas phase and in particles), C_s^{tot}, of a given SVOCs species is driven by the following mass balance equations:

$$\frac{dC_s^{tot}}{dt} = -\frac{1}{V_P}\frac{dV_P}{dt}C_s^{tot} - k_{OH}^s\,[OH]C_s^g + P_s, \tag{6}$$

where k_{OH}^s is the oxidation reaction rate, [OH] is concentration of hydroxyl radical, V_P is the volume of a BB plume, and P_s is the chemical production rate of s. The reaction rates and chemical processes specified in the model are described in the next section (Sect. 2.2). Note that Eqs. (6) for several different species compose an essentially nonlinear system. In particular, not only does C_s^g in thermodynamic equilibrium depends nonlinearly on the total aerosol concentration, C_{OA}, in accordance with Eq. (5), but C_{OA} itself also depends in a complex nonlinear manner on the total concentrations, C_s^{tot}, of all SVOCs. Furthermore, P_s is determined by the gas phase concentrations of SVOCs, too, and therefore depends nonlinearly on both C_s^{tot} and C_{OA}.

Representation of the dilution process (described in Eq. (6) by the term proportional to $\frac{dV_P}{dt}$) in our simulations largely follows Bian et al. (2017). Specifically, we assume that all the species considered are uniformly distributed within a box with a half-width of $2\sigma_y$ across the wind direction and a half-thickness of $2\sigma_z$ in the vertical. The thickness of the box in the wind direction does not need to be explicitly specified in our simulations (but just for definiteness, it can be assumed to be equal to 1 m). The evolution of σ_y and σ_z is represented by the power law expressions according to Klug et al. (1969) (see also Seinfeld and Pandis, 2016) for the neutral (D) Pasquill atmospheric stability class. The plume is assumed to be transported along the wind direction with a constant speed of 5 m s⁻¹. The initial width of the plume ($4\sigma_y$) is considered as a control parameter, S_p, in our simulations. The initial plume width, S_p, can also be interpreted as the across-wind width of the area affected by the fire. The initial value of σ_z is expressed as a function of σ_y. It is assumed that the plume's thickness in the vertical direction ($4\sigma_z$) cannot exceed the mixed layer height, which is fixed at 2500 m: that is, once σ_z calculated according to Klug et al. (1969) reaches 625 m, the plume is allowed to disperse only in the horizontal direction. Such a simple representation of the plume's evolution is by no means intended to be quantitatively accurate under any real conditions but is used mainly to roughly characterize a dependence of the temporal scale of the dilution process on the horizontal spatial scale of a BB plume, especially during the first few hours of evolution.

2.2 Representations of BB OA oxidation processes and gas-particle partitioning in MDMOA

To take into account the existing ambiguity associated with the representation of the oxidation of organic matter within the VBS framework, we performed our simulations using several VBS schemes of varying complexity. A summary of the main features of the schemes and their reference codes is provided in Table 1. We also used a "conventional" OA scheme that assumes that POA is composed of non-volatile species. Below we describe the schemes in more detail.

The scheme "C17" has been described and evaluated by Ciarelli et al. (2017a, b). It is a relatively simple scheme which has been referred to as a hybrid 1.5-dimensional (1.5D-VBS) and is based on a similar scheme proposed by Koo et al. (2014). The idea behind this scheme is to characterize sources, volatilities, chemical transformation, and oxidation state of a complex mixture of SVOCs by considering several surrogate species which are given average molecular compositions and molecular weights. The scheme distinguishes between three sets of the surrogate species, i.e., the POA set (set 1), the set containing

oxidation products from reactions of hydroxyl radical with semi-volatile gases from POA (set 2), and the SOA set (set 3) representing products of reactions of OH with any VOCs (and, implicitly, also IVOCs). Semi-volatile gases from sets 2 and 3 are also allowed to react with OH, with the volatility of the product being an order of magnitude lower than that of the reactant. The different reaction products have different molecular weights and are assumed to represent the net effects of the actual functionalization and fragmentation reactions (neither of which are specified explicitly). All SVOCs are represented using five volatility classes distributed among 5 bins covering the volatility range from 10^{-1} to 10^3 μg m^{-3}. The reaction rate (k_{OH}) is fixed at 4.0×10^{-11} cm^3 molec^{-1} s^{-1} for all the oxidation reactions. Some parameters of the scheme have been optimized by fitting box-model simulations to the data from several aging experiments with BB aerosol from stove wood combustion (Ciarelli et al., 2017a). Based on the optimization results, the average ratio of the initial total mass concentrations of VOCs and SVOCs was set at 4.75. The scheme has been then implemented into a chemistry transport model and successfully evaluated against ambient measurements performed with an aerosol mass spectrometer (AMS) across Europe, specifically in situations where a considerable part of OA originated from residential wood burning (Ciarelli et al., 2017b). Note, however, that the composition and conditions of atmospheric aging of OA from residential wood burning are not necessary representative of those of OA from vegetation fires.

The scheme "K15" is a 1-dimensional (1D) scheme that has been introduced by Konovalov et al. (2015) in an air pollution case study to represent atmospheric aging of BB OA from the 2010 Russian fires. It combines one of the simplest 1D schemes (Grieshop et al., 2009), in which only functionalization reactions have been taken into account, and a more complex scheme in which both functionalization and fragmentation processes are taken into account (albeit in a very simplified manner) and which is referred below as the scheme "S15" (Shrivastava et al., 2013; 2015). The oxidation processes are described using a volatility grid that includes 7 bins ($10^{-2} \le C^* \le 10^4$ μg m^{-3}). The scheme distinguishes between oxidation of primary organic gases (POG), which is assumed to result only in functionalization, and oxidation of secondary organic gases (SOG), which is assumed to include both functionalization and fragmentation branches. The products of the functionalization branch get their mass increased by 40 % and the volatility reduced by two orders magnitude with respect to those of the reactants. Specifically, oxidation of POG and SOG from volatility bin i is represented as follows:

$$POG_{i>2} + OH \rightarrow 1.4\ SOG_{i-2}, \tag{7}$$

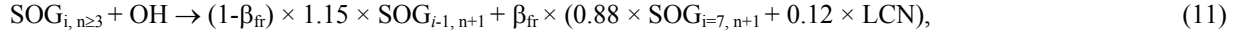$$SOG_{i>2} + OH \rightarrow 0.5 \times 1.4\ SOG_{i-2} + 0.4\ SOG_{i=7} + 0.1\ LCN, \tag{8}$$

where LCN denotes the highly volatile low-carbon-number species that are the products of the fragmentation branch, and all the species are assumed to have the same molecular weight (250 g mole^{-1}). Along with LCN, the fragmentation branch yields SOG in the highest volatility bin. While LCN species are not allowed to participate in any reactions, SOG species can be reprocessed according to Eq. (8). Note that oxidation of SOGs results in a net increase of the organic mass, although Eq. (8) formally corresponds to a fragmentation branching ratio (Jimenez et al., 2009) of 0.5. Note also that the simulations reported by Konovalov et al. (2015) included the transformation of condensed-phase SOA into non-volatile SOA (NVSOA) and indicated that this process had only a small impact on the simulated evolution of BB aerosol in the case considered. In view of a the lack of robust knowledge about the condensed-phase processes (see also Section 4) and for consistency with the other numerical experiments performed in the present study, the transformation of SOA into NVSOA has been disregarded in our simulations.

The scheme "S15" is a slightly modified version of the VBS scheme that was proposed by Shrivastava et al. (2013) and adopted, as a part of a global chemistry transport model, for BB aerosol modeling in a subsequent study (Shrivastava et al., 2015). Unlike the original VBS scheme by Shrivastava et al. (2013, 2015), where only five volatility classes are used for computational reasons, the volatility basis set in the S15 scheme is specified using the same seven volatility bins as in the K15 scheme for the sake of easier interpretation of differences between the respective simulation results. The S15 scheme can be regarded as a quasi-2-dimensional scheme as it realizes a computationally efficient way to account for the increasing

probability of fragmentation reactions with BB OA aging (and, implicitly, with increasing oxidation state) of BB OA by distinguishing between different generations, n, of SOA precursors. Specifically, while POGs and the first two generations of SOGs are assumed to undergo only functionalization reactions:

$$POG_{i>1} + OH \rightarrow 1.15 \ SOG_{i-1, \, n=1}, \tag{9}$$

$$SOG_{i>1, \, n\leq 2} + OH \rightarrow 1.15 \ SOG_{i-1, \, n+1}, \tag{10}$$

the third and further generations undergo both functionalization and fragmentation reactions:

$$SOG_{i, \, n\geq 3} + OH \rightarrow (1-\beta_{fr}) \times 1.15 \times SOG_{i-1, \, n+1} + \beta_{fr} \times (0.88 \times SOG_{i=7, \, n+1} + 0.12 \times LCN), \tag{11}$$

where β_{fr} is the fragmentation branching ratio which in this case is assumed to be equal, in this case, to 0.85.

According to Eqs. (9)-(11), the functionalization reactions of both POGs and SOGs yield SOG species in the next lower volatility bin and result in an increase of the molecular weight by 15 %. Similar to the K15 scheme, all the VBS species are assumed to have the same molar mass of 250 g mol^{-1}. On the basis of Note that using the OA oxidation scheme described above, Shrivastava et al. (2015) defined the two modeling configurations, FragSVSOA and FragNVSOA, with SOA treated as semi-volatile or non-volatile, respectively. The simulations performed in this study with the S15 scheme in this study involved correspond only to the FragSVSOA configuration. This configuration enables better consistency of the S15 scheme with the other VBS schemes considered here, and thus any differences between the simulations performed with the S15 scheme and the other schemes are easier to interpret. WhileWhen choosing the FragSVSOA configuration, we also took into account that global model simulations involving this configuration were found by Shrivastava et al. (2015) to agree better with both surface observations of OA at a South Africa measurement site positioned to investigate BB aerosol and with satellite observations of aerosol optical depth on the global scale than the simulations with the FragNVSOA configuration. Possible formation of NVSOA due to particle-phase reactions (e.g., Barsantie et al., 2014; Jang et al., 2002; Shiraiwa et al., 2013b) is one of the factors (see Sect. 4) that can affect the real BB OA evolution, but were not analyzed in this study focused on identification of major qualitative nonlinear effects in the BB OA behavior due to gas-phase oxidation reactions in BB plumes. (that is, the NVSOA formation has not been included in MDMOA).

Scheme "T18" is a 2-dimensional (2D) VBS scheme that is adopted (with minor modifications) from Tsimpidi et al. (2018), where it has been introduced as a-part of the ORACLE v2.0 aerosol module of the ECHAM/MESSy Atmospheric Chemistry (EMAC) global model. The scheme represents the oxidation of BB OA on the 2D VBS grid constructed in the space of the volatility and the oxygen content (O:C ratio). The volatility dimension is discretized into 4 bins: $C^* = \{10^{-2}; 10^0; 10^2; 10^4\}$ μg m^{-3} and the oxygen content dimension is divided into 11 O:C bins covering the O:C values from 0.2 to 1.2 with a constant step of 0.1. The smallest value of the O:C ratio is assumed to be representative of fresh BB emissions and is attributed to POA. Each species is identified with a representative number of carbon atoms per molecule, n_c, and with a molecular weight, MW, evaluated using structure–activity relationships (Pankow and Asher, 2008; Donahue et al., 2011) and an approximation of the hydrogen to carbon atomic ratio (Heald et al., 2010) as follows:

$$n_c = \frac{11.875 - \log_{10} C^*}{0.475 + 2.3(O:C) - 0.6(O:C)\left(1+(O:C)\right)^{-1}}, \tag{12}$$

$$MW = (15(O:C) + 14)n_c \tag{13}$$

Oxidation of POG (or SOG) species is assumed to result in addition of two or three oxygen atoms to their molecules with an equal probability. The O:C ratio of the products is therefore evaluated as follows:

$$(O:C)_{product} = (O:C)_{reactant} + \frac{\Delta O}{(n_c)_{reactant}}, \tag{14}$$

where ΔO (equal to 2 or 3) is an assumed increment of the oxygen atomic content. The product is assumed to belong to the next lower volatility class with respect to the volatility class of the reactant. Similar oxidation reactions also apply to POG

and SOG species from the lowest volatility class, except that the products of these reactions keep the volatility of the reactants. Fragmentation reactions are not explicitly taken into account. As noted by Tsimpidi et al. (2018), neglecting fragmentation may result in overestimation of OA concentration at long aging timescales. However, it should also be noted that since n_c can decrease as a result of an oxidation step in accordance with Eq. (12), the fragmentation pathway is, to some extent, taken into account in the T18 scheme implicitly.

Along with similar 2D-VBS schemes for anthropogenic and biogenic OA, the T18 scheme described above has been used for multi-year simulations of OA with the EMAC model, and the simulation results were compared against AMS measurements at urban downwind and rural environments in the Northern Hemisphere (Tsimpidi et al., 2018). However, the comparison results did not provide enough information about the performance of the T18 scheme in simulations of OA specifically from BB sources.

The last VBS scheme that we used, "T18f", is our modification of the original T18 scheme. It has been obtained ~~in~~ for this study by adding explicit fragmentation pathways to the original T18 scheme, so that any POG or SOG species considered in the T18 scheme is assumed to participate in both functionalization and fragmentation reactions. The reactions originally included in the T18 scheme are assumed (for simplicity) to represent only the functionalization pathways. The probability of a given pathway is controlled by the fragmentation branching ratio, β_{frag}, which is parameterized as follows (Jimenez et al., 2009; Donahue at al., 2012b):

$$\beta_{frag} = (O{:}C)^{1/4}. \tag{15}$$

Following Murphy et al. (2012), we assume that splitting of an organic molecule as a result of fragmentation reactions occurs with a uniform probability at any site throughout its carbon backbone. Accordingly, a fragmentation reaction of any species containing (according to Eq. 12) n_c carbon atoms can potentially yield $(n_c-1)/2+1$ (if n_c is an odd number) or $n_c/2$ (if n_c is an even number) different (with respect to the atomic carbon content) products. Furthermore, consistently with the assumptions underlying the original T18 scheme, we assume that as a result of any oxidation reaction, one of the two fragments receives two or three additional oxygen atoms, and so its O:C ratio increases in accordance with Eq. (14), except that the atomic carbon number of the reactant in the right-hand part of the equation should be substituted for that of the product. The O:C ratio of the other fragment is kept the same as that of the reactant. If the calculated O:C ratio of a product exceeds 1.2 (that is, the maximum value covered by the O:C grid considered), this product is assumed to be irreversibly lost into the gas phase. All possible pairs of fragmentation products described above are introduced in the corresponding mass balance equations (see Eq. 6). The stoichiometric coefficients for the products in the functionalization, k_s^{fn}, and fragmentation, k_s^{fr}, pathways are evaluated as follows:

$$k_s^{fn} = (1 - \beta_{frag})/N_{ps}^{fn}, \tag{16}$$

$$k_s^{fr} = \beta_{frag}/N_{ps}^{fr}, \tag{17}$$

where N_{ps}^{fn} and N_{ps}^{fr} are the total numbers of possible products in the functionalization and fragmentation pathways, respectively.

Finally, we also consider a linear analogue to the above nonlinear representations of BB OA evolution. The scheme "LIN" is based on a simple oxidation scheme (Pun et al., 2006; Bessagnet et al., 2008) designed in the framework of the conventional approach to representation of OA evolution and SOA formation and implemented in the CHIMERE model. The key assumptions underlying this scheme are that POA is composed of nonvolatile species and that SOA is formed from oxidation of several specific ("traditional") volatile precursors that were identified earlier in smog chamber experiments (Odum et al., 1997). In this study, the original scheme described in detail by Menut et al. (2013) was simplified (linearized) by assuming that all oxidation products are nonvolatile. The emission factors from Andreae (2019) for the boreal forest were used to specify initial concentrations of POA precursors as a function of the initial concentration of BB OA. Note that the original

SOA formation scheme from the CHIMERE model was earlier used in the simulations of evolution of BB aerosol from Russian fires (Konovalov et al., 2015; 2017; 2018) and was found to produce rather negligible amounts of SOA in BB plumes.

Table 1: Reference codes and main features of the BB OA modeling schemes used in the simulations. SVOC: semi-volatile organic compound; VOC: volatile organic compound; POG: primary organic gas; SOG: secondary organic gas; POA: primary organic aerosol; SOA: secondary organic aerosol; C*: saturation mass concentration; β_{frag}: fragmentation branching ratio.

Oxidation scheme	Key features	References
C17 (a hybrid 1.5D-VBS scheme)	Three sets of SVOCs to model oxidation of organics; five volatility classes ($0.1\ \mu g\ m^{-3} \leq C* \leq 10^3\ \mu g\ m^{-3}$) to model gas-particle partitioning; an implicit representation of functionalization and fragmentation reaction pathways; one-bin shift in volatility of a product of oxidation reactions of POG or SOG with OH ($k_{OH}=4\cdot10^{-11}\ cm^3\ s^{-1}$) with respect to those of a reactant	Ciarelli et al. (2017a,b)
K15 (a 1D-VBS scheme)	An explicit representation of the functionalization and fragmentation branches ($\beta_{frag} = 0.5$) of each reaction of POG or SOG with OH ($k_{OH} = 2\cdot10^{-11}\ cm^3\ s^{-1}$); a two-bin shift in volatility and 40 % increase of the molecular weight for a product of the functionalization reaction pathway; seven volatility classes ($0.01\ \mu g\ m^{-3} \leq C* \leq 10^4\ \mu g\ m^{-3}$) for both primary and secondary SVOCs	Konovalov et al. (2015)
S15 (a quasi 2D-VBS scheme)	An explicit representation of the functionalization and fragmentation branches with distinction between ~~reactions involving~~oxidation of POGs and "fresh" SOGs ($\beta_{frag} = 0$which undergo only functionalization reactions) and ~~reactions involving~~ "aged" SOGs (which undergo both functionalization and fragmentation reactions, $\beta_{frag} = 0.85$); a one-bin shift in volatility and 15 % increase of the molecular weight for a product of the functionalization pathway of each reaction of POG or SOG with OH ($k_{OH} = 4\cdot10^{-11}\ cm^3\ s^{-1}$); seven[(1)] volatility classes ($10^{-2}\ \mu g\ m^{-3} \leq C* \leq 10^4\ \mu g\ m^{-3}$) for both primary and secondary SVOCs	Shrivastava et al. (2013; 2015)
T18 (a 2D-VBS scheme without fragmentation)	A representation of SVOCs on a 2D grid space covering four volatility classes ($C*=\{10^{-2};\ 10^0;\ 10^2;\ 10^4\}\ \mu g\ m^{-3}$) and 11 linearly spaced oxygen content bins (O:C=$\{0.2, 0.3, … 1.2\}$); the SVOC molecular mass defined as function of $C*$ and O:C; no explicit representation of fragmentation; a one-bin shift in volatility for a product of any oxidation reaction ($k_{OH} = 2\cdot10^{-11}\ cm^3\ s^{-1}$)	Tsimpidi et al. (2018)
T18f (a modified T18 scheme with fragmentation)	The same as the T18 scheme but with fragmentation reactions ($\beta_{frag} = $(O:C)$^{1/4}$, a uniform probability of fragmentation across the backbone of an organic molecule)	Tsimpidi et al. (2018); Donahue at al. (2012b); Murphy et al. (2012)
LIN (a "linear" OA scheme)	POA and SOA composed of nonvolatile species; SOA formation from oxidation of several specific VOCs	Pun et al. (2006); Bessagnet et al. (2008); Menut et al. (2013)

[(1)] Note that the original VBS scheme (Shrivastava et al., 2015) involves only five volatility classes

It should be stressed that the different VBS schemes outlined above certainly do not comprise all known mechanisms and pathways of oxidation of organic compounds composing BB OA. Nonetheless, consideration of even a limited spectrum of available representations of the BB OA aging processes allows us to get a useful insight into the uncertainty associated with simulations of BB OA aerosol evolution using the VBS framework. Some processes which could further enhance the diversity of our simulations of BB OA evolution are briefly discussed in Sect. 4.

2.3 Configuration of the numerical experiments and processing of output data

MDMOA was run using each VBS scheme described above for a period of 120 hours. This period ~~is~~ has been chosen to be within the ~~representative of the~~ range of typical atmospheric lifetimes of submicron ~~BB~~ aerosol particles emitted from open vegetation fires in the major BB regions worldwide, as indicated, e.g., by a measurement-based estimate (5.1 days) of the

lifetime of black carbon (BC) in Siberia ~~under conditions without precipitation~~ (Paris et al., 2009) and global-model estimates of the BC lifetimes for open fires in northern Africa (5.6 days) and northern South America (3.1 days) (Wang et al., 2016). A part of the simulation period (7 hours per day) was assumed to correspond to nighttime conditions when any oxidation processes were disabled (OH concentration in Eq. (6) was set to be zero); such a nighttime duration is typical, for instance, for central Siberia in summer.

The initial conditions for SVOCs in particles and in the gas phase correspond to the gas-particle equilibrium determined in accordance with the partitioning theory. Specifically, the gas-phase initial concentration for a species s was calculated using Eq. (5) where the OA mass concentration was assumed to include, along with the BB fraction, a background OA concentration, C_{bg}, of 5 µg m^{-3}. The same background OA concentration ~~was~~ had been specified in the box model simulations performed by Bian et al. (2017). For comparison, particulate matter (PM$_{10}$) in a boreal environment of central Siberia under background conditions (that is, without the detectable influence of local or regional pollution sources, including fires) was found by Mikhailov et al. (2017) to have concentrations ranging from about 2 to 10 µg m^{-3} in summer, being composed mostly of organic material. Therefore, the background OA concentration in our simulations can be assumed to be representative of typical background conditions in Siberian boreal forest in summer. Note that specifying a much larger or much smaller value of C_{bg} would likely result in noticeable quantitative changes of the simulated BB OA behavior, since entrainment of background aerosol affects evaporation rates and gas-particle partitioning in a BB plume (Hodshire et al., 2019). The total mass concentration of SVOCs, C^{tot}, that is involved in Eq. (5) was evaluated as follows:

$$C^{tot} = C_0 \Bigg/ \sum_i \frac{f_i}{\left(1+\frac{c_i^*}{C_0+C_{bg}}\right)}, \qquad (18)$$

where f_i defines the mass fraction of all species in ~~the~~ bin i of the volatility distribution, and C_0 is the initial BB OA concentration. C_0 is considered – along with the initial plume size, S_p – as a control parameter in our simulations. Test experiments (see Sect. 3.1) have shown that the characteristic time scales for the adjustment to the "local" thermodynamic equilibrium are short (seconds or minutes) compared to the time scales associated with the oxidation processes (hours). The background aerosol concentration was not affected by any process except for the intersectional fluxes and so was basically kept constant in all the simulations.

The volatility distribution for all our experiments with the K15 and S15 schemes (in which the volatility grid includes seven bins) was adopted from the study by Konovalov et al. (2015): f = {0.1; 0; 0.05; 0.05; 0.2; 0.15; 0.45} at a temperature of 298 K. This distribution is consistent (within the range of uncertainties) with the data from thermodenuder measurements of BB emissions (May et al., 2013). The volatility distributions for the C17, T18 and T18f schemes were obtained from the same distribution by disregarding or aggregating the corresponding volatility bins.

In the experiments with the C17, K15~~, and~~ S15, and LIN schemes, the aerosol size distribution was modeled using 9 size bins covering the range from 20 nm to 10 µm and following a geometric progression with the common ratio of 500$^{1/9}$ ($\approx$ 2.0). To limit the computational time, the experiments with the T18 and T18f schemes were conducted using only 3 size bins that were defined to cover the same range (from 20 nm to 10 µm) using a geometric progression with the common ratio of 500$^{1/3}$ ($\approx$ 7.9). In all cases, we used a lognormal distribution with a mass mean diameter of 0.3 µm and a geometric standard deviation of 1.6 (Reid et al., 2005). While the size distribution can affect the time scales for evaporation and growth of particles, these time scales, as noted above, are much smaller than those associated with chemical aging. Accordingly, in all our simulations, the results of the simulations have been practically independent of the assumed particle size distribution (except for a very minor influence of the Kelvin effect). Note that the representation of non-equilibrium processes in accordance with Eq. (1) has been included in MDMOA mainly to enable simulations of the evolution of BB aerosol optical properties in prospective studies.
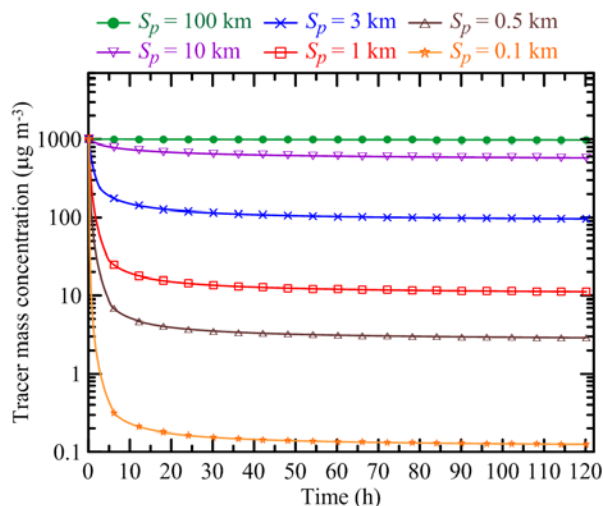
Figure 1: Simulated evolution of the mass concentration of an inert tracer for different values of the initial plume size, S_p. The simulations were done with an initial tracer concentration of 1000 μg m³.

In accordance with Bian et al. (2017) and Hodshire et al. (2019), we assume that the OH concentration does not depend on the initial parameters of the plume and does not change as the plume evolves. Based on the in-plume measurements by Akagi et al. (2012), its value was set to 5×10^6 cm^{-3} in all our simulations, which is larger than the value (1.08×10^6 cm^{-3}) specified in the simulations by Bian et al. (2017) and Hodshire et al. (2019). Assuming a lower or higher constant value of the OH concentration would slow down or speed up chemical evolution of BB OA in our simulations but is not expected to alter major qualitative nonlinear features of the BB OA behavior, which are focus of this study. Like Bian et al. (2017) and Hodshire et al. (2019), we assumed a constant temperature of 298 K within the plume, thus making unnecessary any assumptions regarding enthalpies of vaporization of SVOCs. The concentration of OH within real BB plumes ~~is not directly determined by either the density or the age of the plume~~ can be affected by many factors (such as, e.g., the UV flux, the concentrations of nitrogen oxides and VOCs within the plume), which cannot be unambiguously simulated within our box model. Variability of these factors can cause variability in the OH concentration levels across different plumes as well as temporal and spatial fluctuations of the OH concentration within a given plume. In particular, attenuation of the downwelling UV flux within dense plumes (Hobbs et al., 2003) can suppress both the OH levels and SOA formation rates (Konovalov et al., 2016). Temperature is also likely to vary, both spatially and temporally, within real-world BB plumes. In particular, it is likely to be lower in the upper part of a plume than near the surface (Hodshire et al., 2019). All possible variability and inhomogeneities of the OH concentration and temperature are disregarded in our simulations. Out study is therefore limited in this respect, but this limitation allowed us to isolate and investigate the internal dynamics of the BB OA system under fixed pre-defined conditions.

~~and is not directly determined by either the density or the age of the plume. For definiteness and simplicity, we assume that it does not depend on the initial parameters of the plume and does not change as the plume evolves; based on the ambient measurements by Akagi et al. (2012), its value was set to 5×10⁶ cm⁻³ in all our simulations.~~

~~We also assumed a constant temperature of 298 K, thus making unnecessary any assumptions regarding enthalpies of vaporization of SVOCs.~~

Along with aerosol species, MDMOA has been configured to simulate the evolution of a special tracer which is intended to represent the evolution of the BB OA mass concentration in a hypothetical situation where BB aerosol is composed of chemically inert and non-volatile components. Accordingly, the tracer ~~n~~ has been introduced into our model as a chemically inert species ~~which~~ that can be affected only by ~~the~~ dilution ~~process~~ (since ~~the~~ dry and wet deposition processes were not

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considered in our simulations), and ~~tracer. The tracer's initial mass concentration was the same as that of BB OA (that is, C_a).~~ Tthe behavior of the tracer's mass concentration ~~is affected only by dilution and~~ was controlled only by the parameter S_p (initial plume size). The tracer's initial mass concentration in all our simulations was the same as that of BB OA (that is, C_0). The molecular weight of the tracer has been set to be the same as of carbon monoxide (CO). Note that the concept of analyzing the evolution of BB aerosol versus the evolution of an inert tracer (usually represented by CO) has been fruitfully exploited in many previous experimental and modeling studies of BB aerosol (e.g., Akagi et al., 2012; Konovalov et al., 2015; Bian et al., 2017, Vakkari et al., 2018). Figure 1 shows the dynamics of the mass concentration of the inert tracer for several different values of S_p according to our simulations. Consistent with the tracer dynamics demonstrated by Bian et al. (2017), our simulations indicate that the initial plume size may have a major impact on the subsequent plume evolution (and thus on the BB OA mass concentration). In particular, whereas the tracer concentration drops by four orders of magnitudes during the evolution of a plume with the smallest value of S_p (100 m), it keeps a practically constant value (decreasing by less than 3 %) within the largest plume considered (with S_p of 100 km). Note that for the plumes with S_p smaller than 3 km, the largest changes of tracer concentration occur during the first 5 hours of evolution; afterwards, the changes are relatively small and are not of significance for the results of this study. It should be emphasized that our simulations of the BB plume dispersion were not intended to be fully realistic and quantitatively accurate. Rather, we used a simple plume dispersion model (which is formally not applicable at time scales exceeding a few hours and has many other limitations) just to specify several definite scenarios for the dilution process, with strongly different dilution rates during the initial stage of the BB plume evolution.

Using the simulated tracer mass concentration (*Tr*), we evaluated the BB OA mass enhancement ratio (abbreviated as EnR throughout this paper and also denoted as γ_a below) at a given time *t* as follows:

$$\gamma_a(t) = \frac{C_{OA}(t) - C_{bg}}{Tr(t)}, \tag{19}$$

where $C_{OA}(t)$ is the total OA mass concentration. Note that γ_a is analogous to the "inert OA mass enhancement ratio" introduced by Bian et al. (2017). Since the initial concentration of the tracer was set to be the same as the initial concentration of BB OA, the initial value of γ_a in our simulations was always equal to one without using any additional normalization (~~that~~ which is usually involved in similar definitions of EnR used in chamber and field studies). The simulations were performed with a sufficiently small nominal time step of 1 s. Based on some test simulations performed with smaller time steps, we estimate that the numerical error in γ_a does not exceed 10 % in any case considered (but is typically much smaller).

Analysis of γ_a allows us to identify changes of BB OA concentration due to the combined effects of oxidation and gas-particle partitioning processes. Furthermore, any kind of dependence of γ_a on the initial BB plume size (S_p) or the initial concentration of BB OA (C_0) can be considered as a manifestation of a nonlinear behavior of the BB OA mass concentration. Indeed, it is easy to show using Eq. (6) that if the gas-phase concentration of each SVOC, C_s^g, were a linear function of the total concentration, C_s^{tot}, of the same species, then the dynamics of γ_a would not depend on C_0 and could not be affected by dilution (and accordingly would not depend on S_p, either).

Interaction of SVOCs with water is not taken into account in the OA schemes described above; thus any known effects of humidity on evolution of BB OA, such as, e.g., formation of SOA from oxidation of water--soluble organic compounds in the liquid phase (Brege et al., 2018), have been disregarded. However, as suggested by Tsimpidi et al. (2018), we used the calculations of the O:C ratio ~~within the T18 and T18f schemes~~ to evaluate the hygroscopicity parameter κ (Petters and Kreidenweis, 2007), which characterizes hygroscopicity and cloud condensation nucleus (CCN) activity of BB OA ~~by means of the hygroscopicity parameter κ (Petters and Kreidenweis, 2007)~~. Specifically, we expressed the hygroscopicity parameter

for any organic species s, κ_s, as a linear function of the O:C ratio by using a parameterization proposed by Lambe et al. (2011):

$$\kappa_s = 0{:}18\,(O{:}C) + 0.03. \tag{20}$$

The overall value of κ for BB OA, κ_{org}, was obtained using a mixing rule (Petters and Kreidenweis, 2007):

$$\kappa_{org} = \sum_s \varepsilon_s \kappa_s, \tag{21}$$

where ε_s is a volume fraction of a given species. The volume fractions were estimated by assuming a constant volumetric mass densityies of 1.2×10^3 kg m^{-3} for any OA species considered. Note that the hygroscopicity parameter was calculated only with the C17, T18, and T18f schemes, as the other oxidation schemes (K15, S17, and LIN) considered in this study are not designed to evaluate the O:C ratio.

3 Results

3.1 Dynamical regimes of the BB OA evolution

The evolution of EnR (γ_a) according to our simulations performed with the different OA schemes is presented in Figs. 2 and 3. The simulations were done with several different values of the initial plume size, S_p, (see Fig. 2) and initial BB OA concentration, C_0 (see Fig. 3). The parameter range considered in our simulations is intended to represent the highly variable characteristics of typical smoke plumes from any kind of vegetation fires. The fixed values of C_0 (10^3 µg m^3) and S_p (5 km) in the simulations shown in, respectively, Fig. 2 and Fig. 3 are chosen to approximately represent mid-range values of the corresponding parameters (on a logarithmic scale). The simulation results allow us to identify five distinctive dynamical regimes of the BB OA evolution. The simulations corresponding to the specific regimes are marked in the figures by the numbers from 1 to 5. The first regime ("1") corresponds to a monotonic saturating increase of γ_a. This regime is found in the simulations with the C17, K15, and LIN schemes. An increase of γ_a is followed by its decrease in the second regime ("2"), that which is typical for the S15, T18, and T18f schemes. The third regime ("3") features a sharp initial decrease of γ_a, followed by a slow monotonic decrease. This regime is found with the C17, K15, and T18 schemes but only when S_p is relatively small (in particular, when S_p equals to 100 m, see Fig. 2c, d). The most complex behavior of γ_a corresponds to the fourth regime ("4") and is found with the S15, T18, and T18f schemes (see Figs. 2c, d, and Fig. 3d) for specific C_0 and S_p values. In this regime, γ_a first decreases, then increases and finally decreases again. Finally, the fifth regime ("5") corresponds to the monotonic decrease of γ_a. It is found only in a simulation with the T18f scheme (see Fig. 3d).

The results shown in Figs. 2 and 3 demonstrate, on the one hand, that the differences in the considered representations of the OA evolution are associated with both major quantitative and qualitative differences in the BB OA behavior even when the control parameters are the same. For example, if $S_p = 5$ km and $C_0 = 10^3$ µg m^3 (see Fig. 2b), the simulation with the K15 scheme predicts a strong monotonic increase of γ_aEnR up to a factor of 3 after a 120-hour evolution. At the same time, the T18f scheme predicts a slight net decrease of γ_aEnR, even though it also predicts an increase of γ_aEnR at the initial stage of the evolution. Even larger differences (exceeding a factor of 8) between the simulations are evident if S_p equals to 0.1 km.

On the other hand, changing parameter values in the simulations with the same VBS scheme can result in "switching" between different types of the BB OA evolution. In particular, depending on S_p, the simulations with the C17 and K15 schemes can demonstrate both the regimes "1" (Figs. 2a, b and Fig. 3) and the regime "3" (Fig. 2c, d). Both the regimes "2" (e.g., Fig. 2a, b) and "4" (e.g., Fig. 2c) are found in the simulations performed with the S15 and T18f schemes. The simulations with the T18 scheme are found to manifest four regimes, from "1" to "4" (see Figs. 3a, c, d and Fig. 2c). In contrast to the simulations using the VBS framework, the simulations with the LIN scheme manifest a single dynamic

regime (regime "1"). Note that the simulation results shown in Figs. 2 and 3 are not meant to provide an exhaustive analysis of the parameter space with regard to possible dynamical regimes, but rather are intended to illustrate the diversity of the BB OA behavior simulated with different OA schemes and different parameter values.
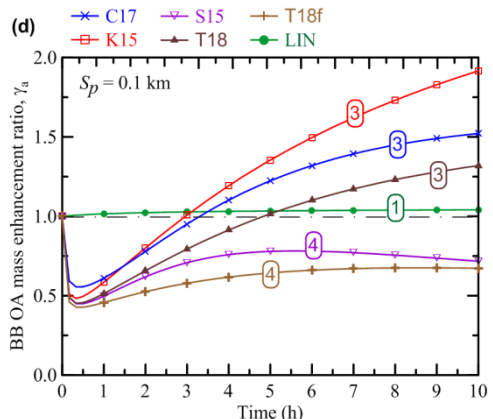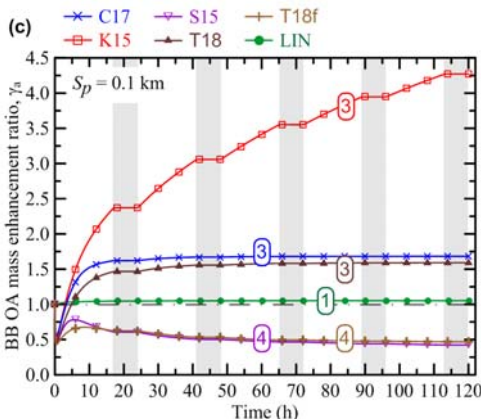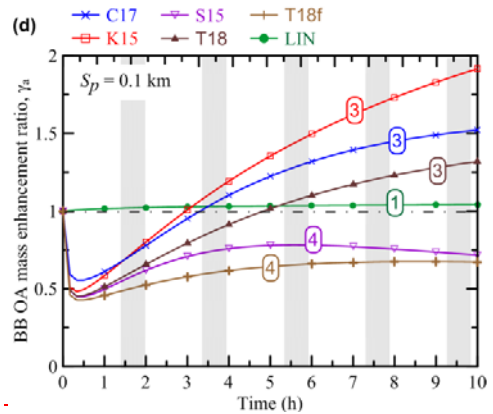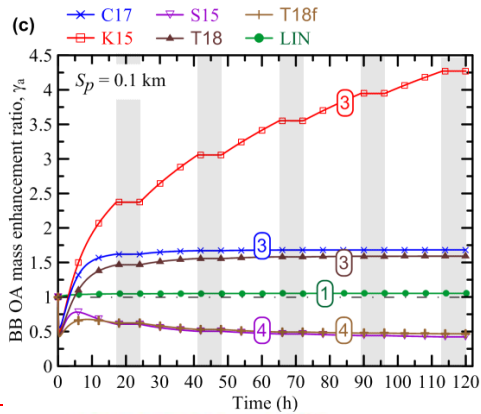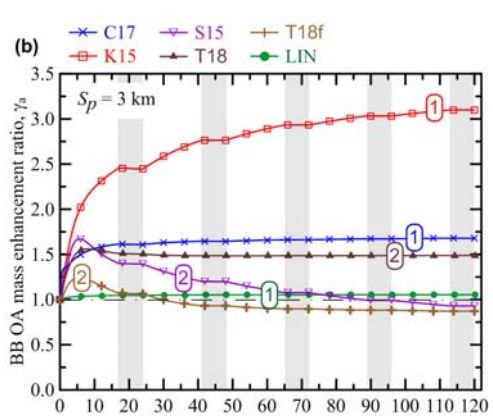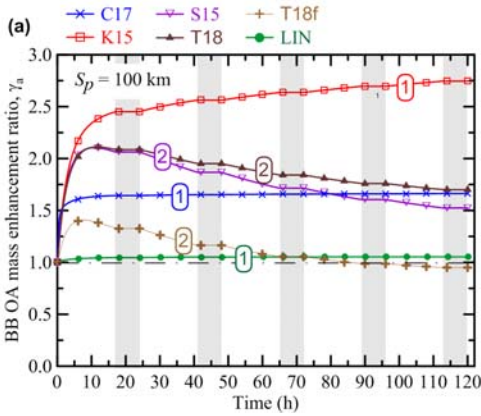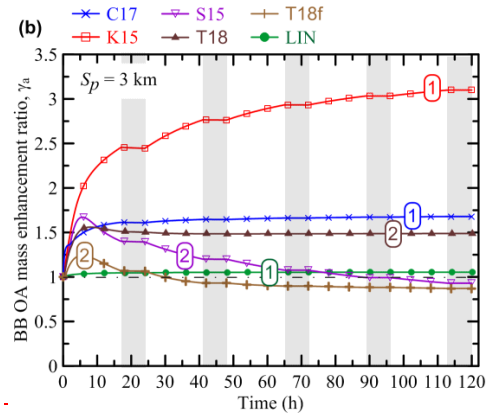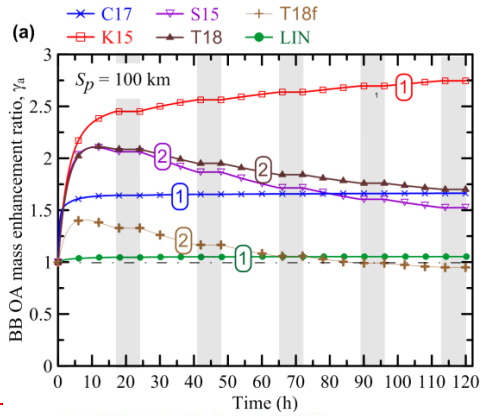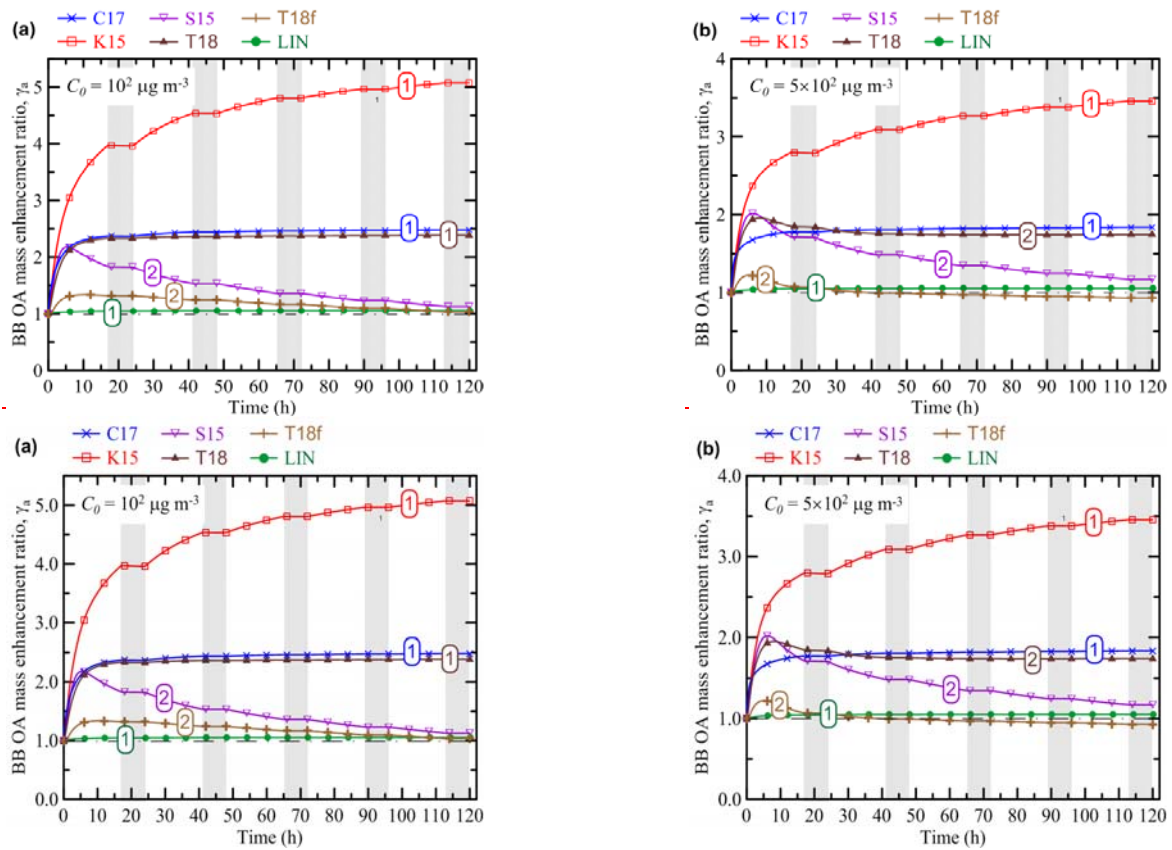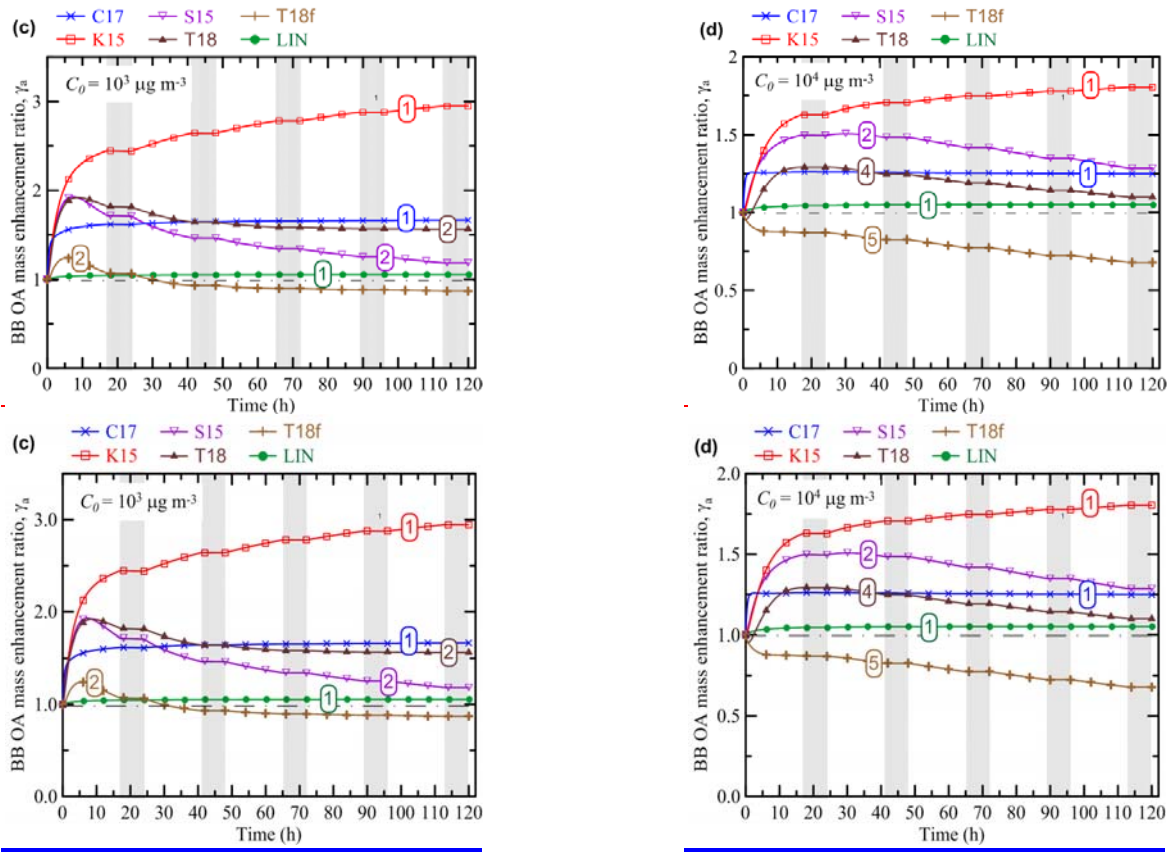
Figure 2: Evolution of the BB OA mass enhancement ratio (γ_a) according to the simulations performed with the different OA schemes and with different values of the initial plume size: (a) $S_p = 100$ km, (b) $S_p = 3$ km, (c, d) $S_p = 0.1$ km. The initial BB OA concentration, C_0, was fixed at 10^3 µg m^3 in all the simulations. ~~Note that the plotp~~pPanel (d) shows a zoomed fragment of the simulations presented in ~~the plot~~panel (c). Shaded bands depict nighttime periods when no oxidation reactions were allowed to occur. The numbers on the curves denote the different dynamical regimes of BB OA evolution according to the definitions in Sect. 3.1. A horizontal dash-dotted line in each panel indicates the situation where no BB OA mass enhancement occurs ($\gamma_a = 1$).

Figures 2 and 3 also show that using the same OA scheme with different values of S_p and C_0 can be associated with not only qualitative differences in the EnR behavior (as indicated above) but also with considerable quantitative differences between the simulated EnR values. For example, γ_a for the BB OA at an age of 120 hours ~~that is obtained from the simulations~~ with the K15 scheme and $C_0 = 1000$ µg m^3 increases from about 2.7 to 4.2 as S_p decreases from 100 to 0.1 km (cf. Figs. 2a and 2c). In contrast, the same change of S_p in the simulation with the S15 scheme results in a dramatic drop of γ_a from about 1.5 to 0.5. The simulations with the C17 and T18 are relatively insensitive to changes of S_p. An increase of C_0 is typically associated with a decrease of γ_a. The sensitivity of γ_a to C_0 is very significant in the simulations with the K15 scheme, but is weak in the simulations with the C15 and T18f schemes.

Figure 3: Evolution of the BB OA mass enhancement ratio (γ_a) according to the simulations preformed with the different OA schemes and with different values of the initial mass concentration of BB OA: (a) $C_0 = 10^2$ µg m^3, (b) $C_0 = 5 \times 10^2$ µg m^3, (c) $C_0 = 10^3$ µg m^3, (d) $C_0 = 10^4$ µg m^3. The initial BB plume size, S_p, was fixed at 5 km in all the simulations shown. <u>Shaded bands depict nighttime periods when no oxidation reactions were allowed to occur. The numbers on the curves denote the different dynamical regimes of BB OA evolution according to the definitions in Sect. 3.1. A horizontal dash-dotted line in each panel indicates the situation where no BB OA mass enhancement occurs ($\gamma_a = 1$).</u>

As noted in Sect. 2.3, any effects of changes in the parameters S_p and C_0 on γ_a~~EnR~~ can be considered as a manifestation of a nonlinear behavior of BB OA. In contrast to the nonlinear behavior demonstrated by BB OA in the simulations with the VBS schemes, the simulations with the LIN scheme manifest a simple "linear" behavior, being quite insensitive to changes of S_p and C_0: γ_a monotonically increases by ~5 % after 120 h irrespective of the parameter values. This insensitivity is a result of the assumptions that POA is not affected by any processes except for dilution and that the initial concentration of SOA precursors is proportional to C_0. The dependencies of γ_a on the control parameters of our simulations are analyzed in more detail in the next section (Sect. 3.2).

(a)   (b)
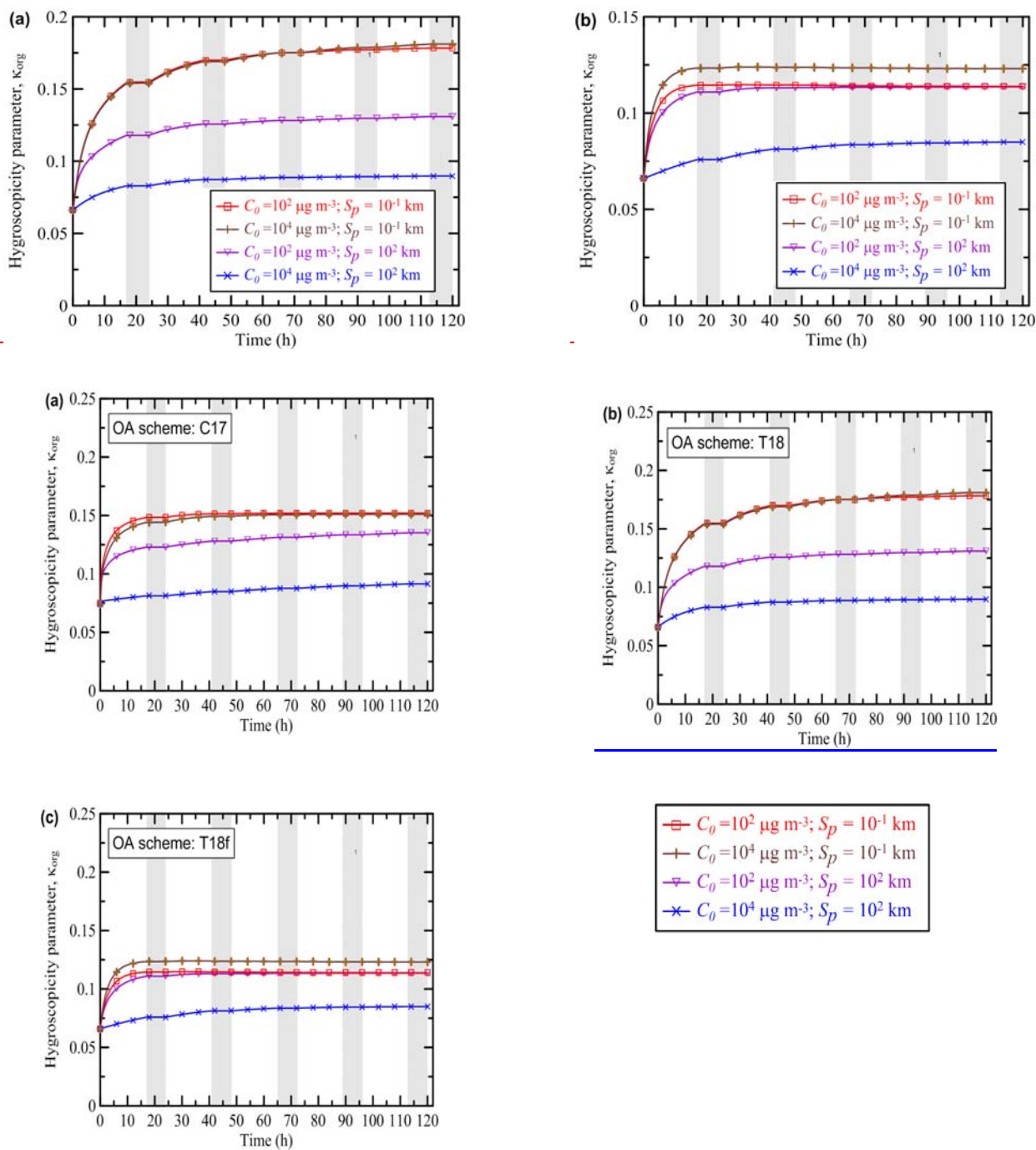
Figure 4: Evolution of the hygroscopicity parameter κ_{org} simulated as a function of time with different values of the initial BB OA mass concentration, C_0 and the initial BB plume size, S_p, using the (a) C17, (b) T18 and (c) T18f schemes. Shaded bands depict nighttime periods when no oxidation reactions were allowed to occur.

The existence of the different dynamical regimes of BB OA evolution can be interpreted by considering the interplay between several competing processes, such as functionalization, fragmentation, and evaporation with dilution. In particular, regime "1" featuring a saturating enhancement of γ_a can be explained by SOA formation dominated by functionalization reactions associated with a decrease of volatility and an increase of the molecular weight of SVOCs. A steady state is reached in this regime when all SVOCs belong to the two lowest-volatility bins. Rapid fragmentation reactions in the case of S15 and T18f schemes or the decrease of the molecular weight of the products of the oxidation reactions (effectively combining the effects of the functionalization and fragmentation oxidation pathways) in the case of the T18 scheme can cause a depletion of the SOA amounts that have initially been formed from oxidation of POGs, leading to regime "2". A rapid POA evaporation caused by dilution at the initial stage of the BB plume evolution can result in a depletion of the bulk

OA amounts and give rise to the regimes "3", "4" or "5", depending on the interplay between fragmentation and functionalization after the fast dilution.

Unlike the BB OA enhancement ratio, the hygroscopicity parameter κ_{org} exhibits, ~~according to the simulations with the T18 and T18f schemes,~~ a rather simple behavior (~~see~~ Fig. 4), similar to that of γ_a in the regime 1. The monotonic growth of κ_{org} over time is hardly surprising, since each reaction is assumed to result in an increase of the O:C ratio (according to, e.g., Eq. 14) for at least one of the products with respect to that of the reactant (and the O:C ratio of the other products is assumed to remain the same as that of the reactant). The initial rapid increase of κ_{org} is eventually slowing down because the semi-volatile POGs and SOGs participating in the oxidation processes are eventually transformed either into low-volatile products (as in the simulations with the C17 and T18 schemes) ~~scheme~~ or, on the contrary, into volatile gases (as in the simulations with the T18f scheme). Note again that the hygroscopicity parameter was calculated using only the C17, T18 and T18f schemes, since the other oxidation schemes (K15, S17 and LIN) considered in this study are not designed to track changes of the O:C ratio.
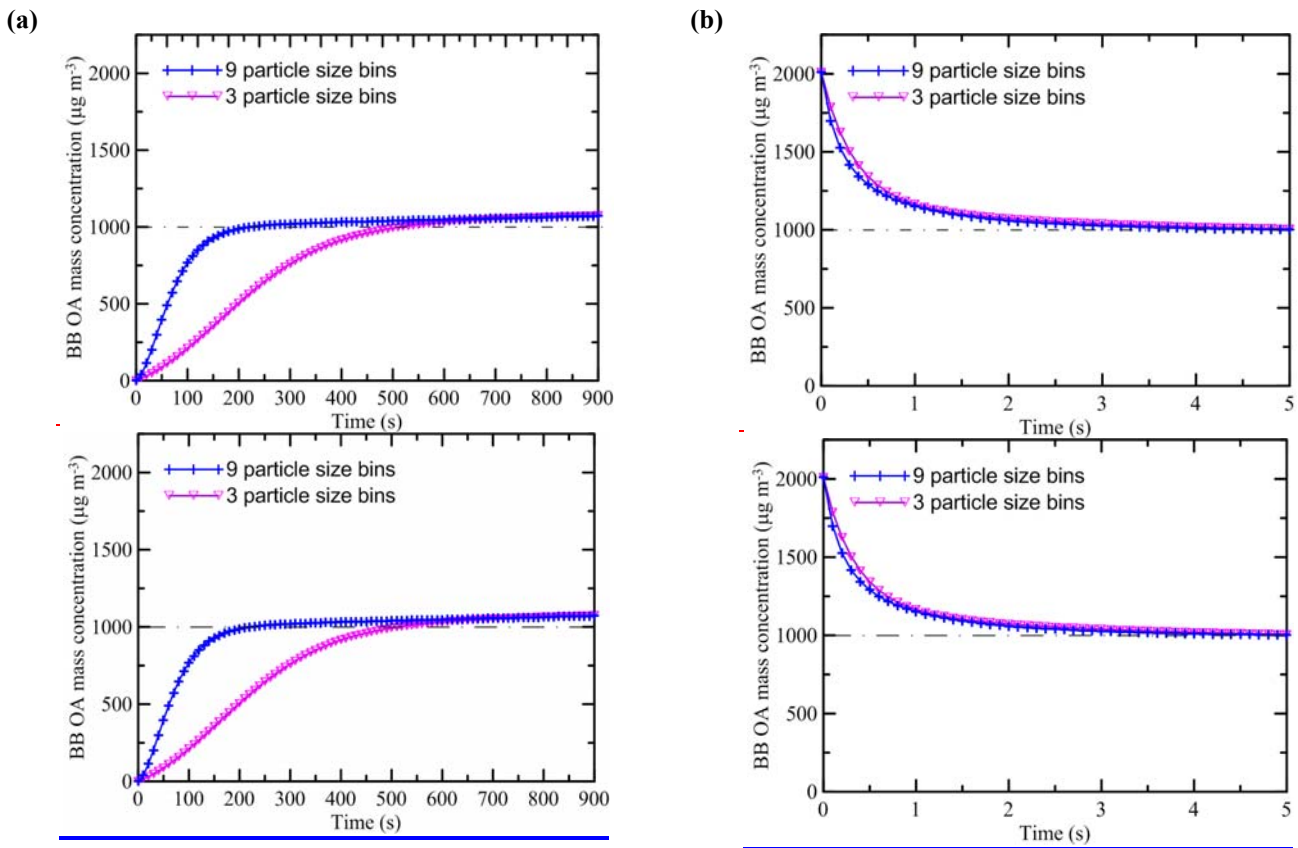
Figure 5: Adjustment of mass concentration of fresh BB OA to thermodynamic equilibrium according to test simulations performed with the S15 scheme without dilution. The simulations were initiated from different initial conditions: (a) $C_0 = 0$ μg m^3; (b) $C_0 = 2 \times 10^2$ μg m^3. A horizontal dash-dotted line in each panel indicates ~~T~~the BB OA concentration (10^3 μg m^3) at thermodynamic equilibrium ~~in the both cases corresponds to a BB OA concentration of 10^3 μg m^3~~. The time step in the test simulations was set at 0.1 s.

Nonetheless, in spite of a relatively simple behavior, the simulations of κ_{org} are still rather sensitive to the parameters of the BB plume. In particular, the value of κ_{org} after the 120-hour evolution is found, according to the simulation with the T18 scheme, to be about a factor of two bigger in the cases with a small plume ($S_p = 0.1$ km), irrespective of the initial aerosol concentration, than in the cases with a large and dense plume ($S_p = 100$ km; $C_0 = 10^4$ μg m^3). Values of κ_{org} are higher for smaller plumes, because a larger fraction of POA species evaporates due to ~~upon~~ more rapid dilution of smaller plumes, followed by ~~being replaced by~~ gas phase formation of highly oxidized SOA species. The sensitivity of the simulations of κ_{org} with the T18f scheme to the control parameter values is smaller but is still considerable. The fact that κ_{org} is sensitive to the parameters of a BB plume can be regarded as another manifestation of the nonlinear behavior of BB OA~~: it is easy to show that in the absence of transfer of organic species between the gaseous and condensed phases of organic species, the oxidation rate (and, therefore, κ_{org}) would depend neither on C_a nor S_p (similar to the case of the simulations of γ_a with the LIN scheme)~~. Note that only simulations for the "extreme" values of C_0 and S_p (among those considered in this study) are shown in Fig. 4 for all of the three oxidation schemes allowing an evaluation of the hygroscopicity parameter. Simulations performed using a given scheme with other (intermediate) parameter values would fall between the brown and blue curves in a corresponding plot. The dependencies of κ_{org} on C_0 and S_p are analyzed in some more detail in the next section. Note also that quantitatively, the hygroscopicity parameter values ranging from 0.08 to 0.18 for aged aerosol in our simulations look rather reasonable. For example, Mikhailov et al. (2015) reported a volume-based hygroscopicity parameter value of about 0.1 (on average) for the accumulation mode of aerosol in Siberia during the period of intense fires.

As noted above (Sect. 2.3), the initial conditions for our BB OA simulations correspond to ~~the~~ instantaneous thermodynamic equilibrium. Additional test simulations performed using the S15 scheme with perturbed initial conditions and two different numbers of size bins (nine, as in the simulations with the C17, K15 and S15 schemes, or three, as in the simulations with the T18 and T18f schemes) confirm (~~see~~ Fig. 5) that thermodynamic equilibrium between POGs and particles is typically reached before any noticeable changes occur due to chemical processes. The equilibration time scales are found to range from just a few seconds (when the balance between gas and particle phases is shifted toward the particle phase) to about ten minutes (when all the BB emissions are assumed to be in the gas phase and the bin number equals three). In accordance with these results, the BB OA mass concentration after the five-day evolution was not found to be sensitive in our additional test simulations (which are not presented here) to either the initial perturbations of the gas-particle equilibrium or to the change in the number of bins in the particle size distribution in the range of the considered values of the plume parameters, even though aged as well as strongly diluted OA can feature much longer equilibration time scales (Riipenen et al., 2010).

3.2 Dependencies of the BB OA enhancement ratio and hygroscopicity parameter on the parameters of the smoke plume

Figures 6 and 7 present the simulated dependencies of EnR on the initial BB OA concentration, C_0, after 5 and 120 hours of evolution, $\gamma_a(5h)$ and $\gamma_a(120h)$, respectively. These dependencies have been calculated using the different OA schemes with varying values of the initial plume size, S_p. Figure 6 also shows (by dashed lines) the initial ratio of the total SVOCs mass concentration (C^{tot}) to the total BB OA mass concentration (C_0) at thermodynamic equilibrium according to Eq. (18). The C^{tot}/C_0 ratio depends on the parameters of the assumed volatility distribution, and for this reason is different for different schemes. Our simulations show that $\gamma_a(5h)$ follows, in most cases, a slow and monotonic inverse dependence on C_0. This dependence is qualitatively (and in some cases, but not always, even quantitatively) rather similar to that of C^{tot}/C_0: evidently, the decreasing character of the dependence of γ_a on C_0 can be due to the fact that larger values of C_0 correspond to a smaller SVOC gas-phase fraction available for SOA formation. Larger values of S_p typically correspond to larger values of $\gamma_a(5h)$; such a dependence is evidently due to evaporation of POA with dilution. However, the dependences of $\gamma_a(5h)$ on C_0 and S_p are not always monotonic. In particular, the dependence of γ_a on C_0 has a "weak" minimum in the simulations with the T18 scheme and S_p of 1 km when C_0 is around 5×10^3 µg m^3. A similar minimum (but around 2×10^3 µg m^3 ~~)~~ can be seen in the simulations with the T18f scheme~~)~~. When C_0 equals ~~to~~ 5×10^3 or 10^4 µg m^3 in the simulations with the T18 scheme, γ_a calculated with $S_p = 5$ km is larger than that calculated with $S_p = 100$ km, although an inverse relation ~~takes place~~exists between the corresponding values of γ_a simulated with $S_p = 0.1$ km and $S_p = 0.5$ km. These irregularities cannot be easily explained but are certainly not due to numerical errors (as has been confirmed in additional test simulations with a reduced time step), emphasizing the essentially nonlinear character of the BB OA behavior, especially when it is modeled using such complex VBS schemes as T18 or T18f.

~~A~~ The value of $\gamma_a(120h)$ also tends to decrease as C_0 increases in most cases, but with differences among the schemes~~-~~: C17, K15 and T18 schemes with a weaker degree of fragmentation show a pronounced decrease, while schemes with a stronger degree of fragmentation are ~~more~~ less insensitive to ~~on~~ C_0 (~~see~~ Fig. 7). This is presumably due to the fact that if functionalization is more important than fragmentation, the~~an~~ availability of more SOGs at lower C_0 leads to enhances in SOA formation.
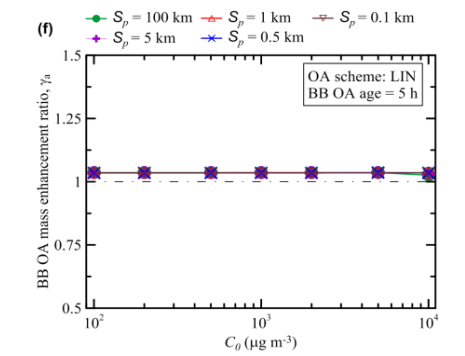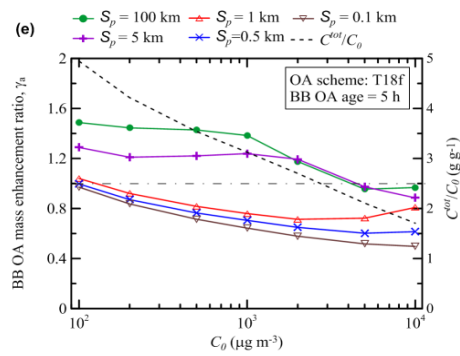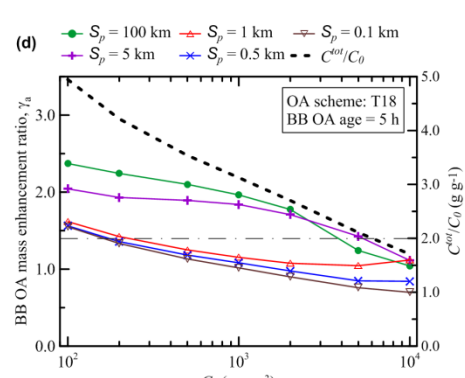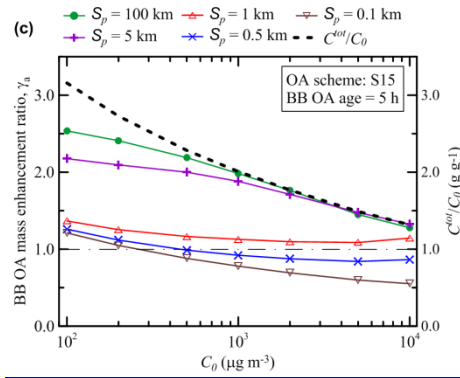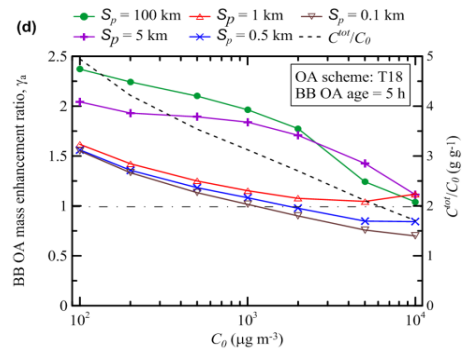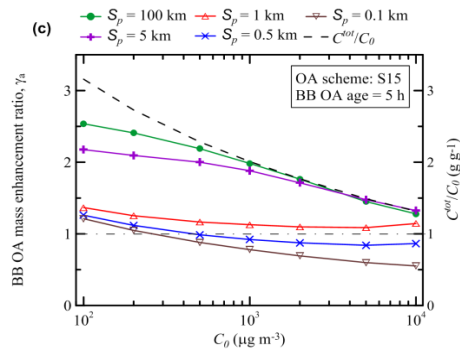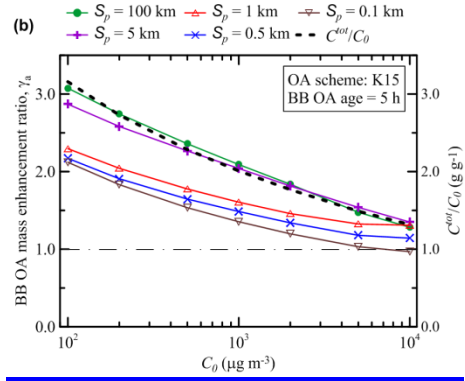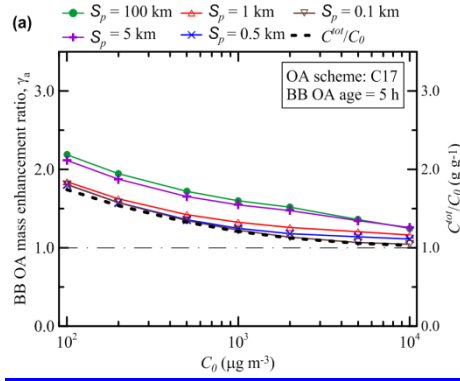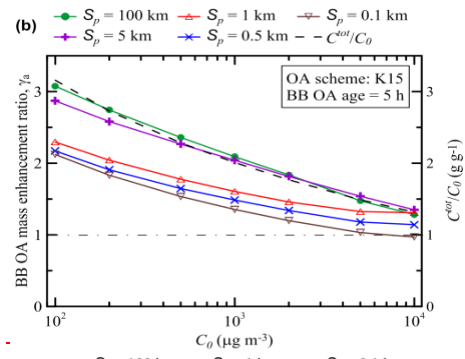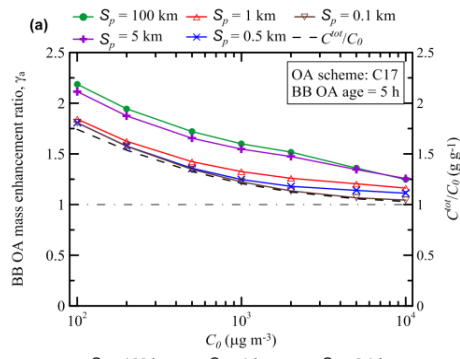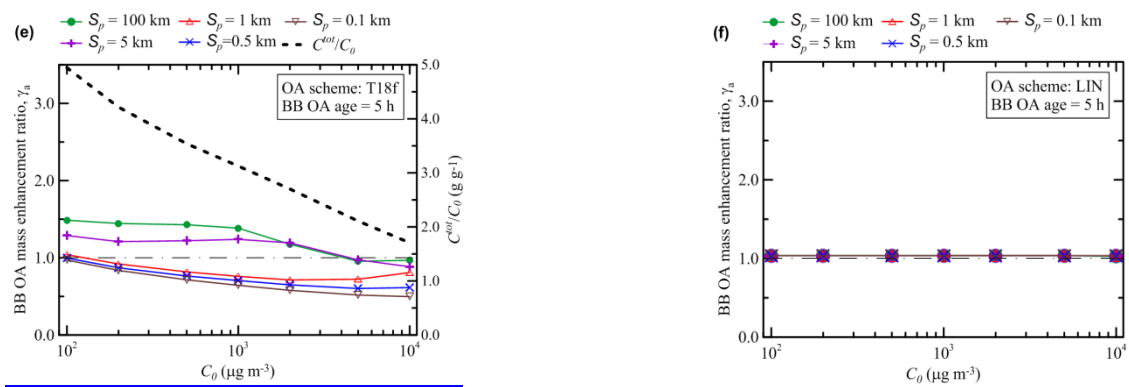
Figure 6: **Simulated dependencies of the BB OA enhancement ratio after 5 hours of evolution on the initial BB OA mass concentration, C_0. The dependencies are obtained with different OA schemes (~~see~~ Table 1) and with different values of the initial plume size, S_p. Dashed lines show the ratio of the total initial mass concentration of SVOCs (C^{tot}) to the initial BB OA concentration (C_0) at thermodynamic equilibrium according to Eq. (18). A horizontal dash-dotted line in each panel indicates the situation where no BB OA mass enhancement occurs ($\gamma_a = 1$).**

However, different schemes features more complex and diverse dependencies on S_p (~~see~~ Fig. 7). For example, while the simulations with the C17 and T18 schemes, where the fragmentation and functionalization processes are treated implicitly, do not reveal any strong sensitivity of $\gamma_a(120)$ on S_p, the scheme K15, where functionalization strongly dominates over fragmentation, yields a rather strong but inverse dependence of the same parameter (that is, unlike $\gamma_a(5h)$ calculated with the same scheme, $\gamma_a(120h)$ decreases with increasing S_p). Presumably, evaporation of POA due to dilution provides more "fuel"
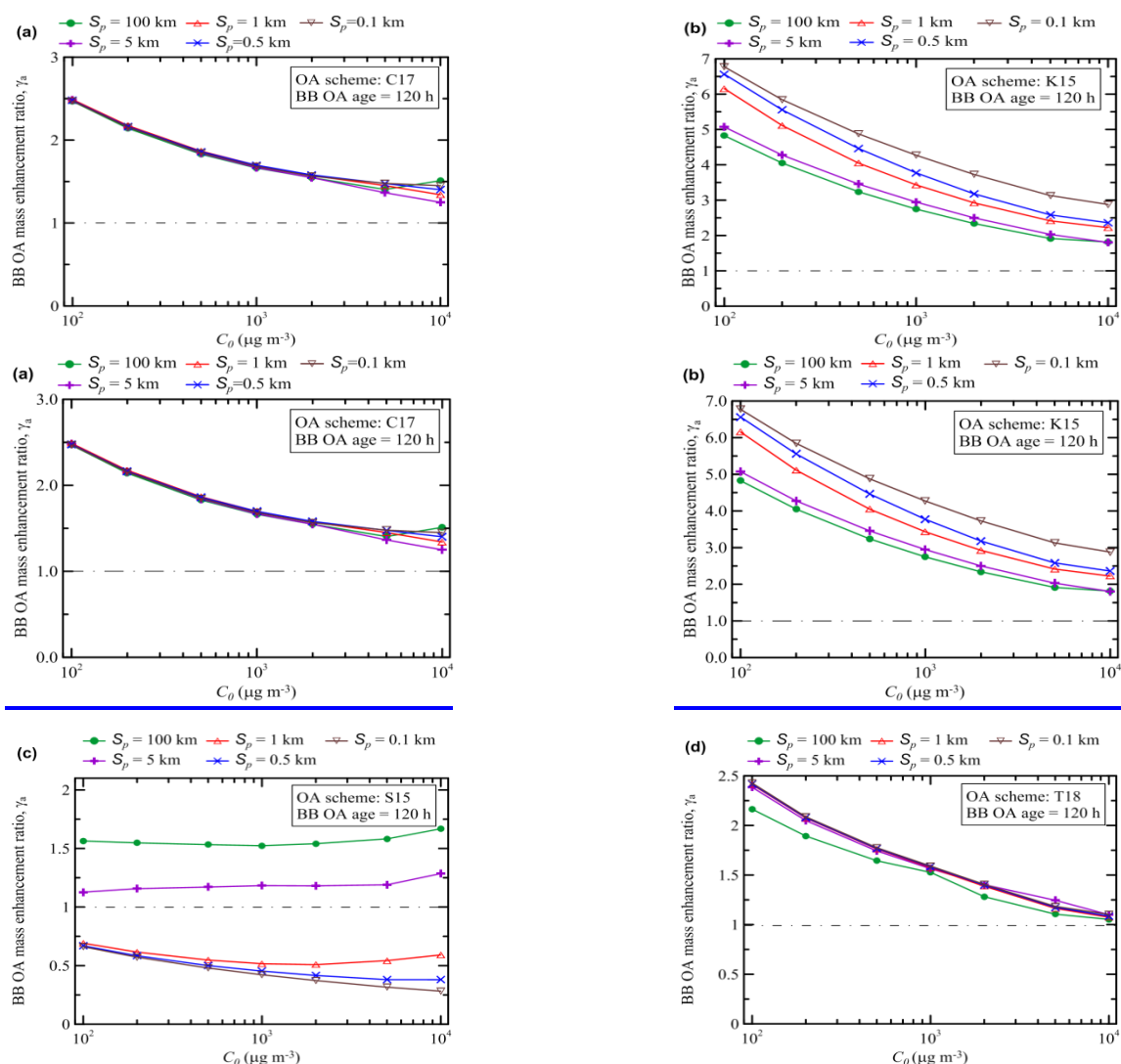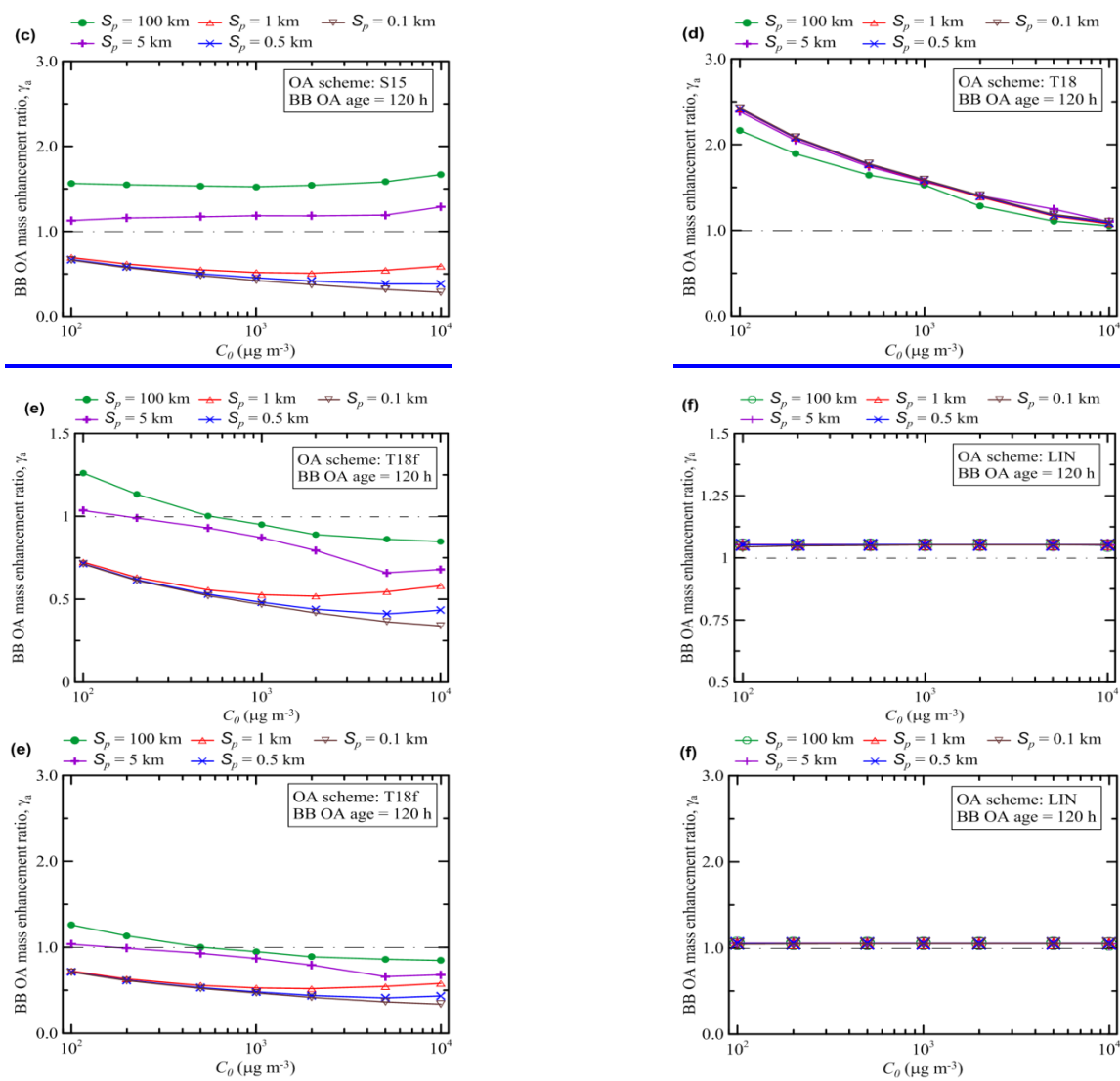
Figure 7: The same as in Fig. 6 but for the BB OA enhancement ratio after 120 hours of evolution and except that the ratio of C^{tot} to C_0 (which is not supposed to characterize the state of BB OA after five-day aging) is not shown. Note that the y-axis scale in the panel (b) is not the same as in other panels. A horizontal dash-dotted line in each panel indicates the situation where no BB OA mass enhancement occurs ($\gamma_a = 1$).

for functionalization reactions, leading to a larger increase of γ_a when S_p is smaller (as found in the simulations with K15 scheme). In contrast, the schemes with strong fragmentation reactions, S15 and T18f, reveal a strong increasing dependence of $\gamma_a(120h)$ on S_p, while they are less sensitive on C_0 relatively to the other schemes. This dependence can mainly be due to the fact that the evaporated organic compounds are exposed to fragmentation reactions: accordingly, a decrease in S_p results in larger losses of SVOCs and smaller SOA concentrations.

A comparison of the results shown in Figs. 6 and ~~Fig.~~ 7 reveals that the differences between the different VBS schemes are associated with mostly only quantitative differences in the dependencies of $\gamma_a(5h)$ on C_0 and S_p~~-~~, but with both quantitative and some qualitative differences in the corresponding dependencies of $\gamma_a(120h)$. This observation indicates that the mass concentration of aged BB OA is likely to be much more strongly affected in the simulations by uncertainties in available representations of the BB OA evolution than the mass concentration of relatively fresh BB OA. One of the reasons is that, as the amount of SOGs susceptible to fragmentation reactions increases with time, the competition between functionalization and fragmentation reactions ~~a~~ plays an ~~become~~ increasingly important role, ~~with time when the SOG oxidation level increases, and then the competition between functionalization and fragmentation~~ creat~~es~~ing ~~the~~ a more complex dependence on the plume parameters. As a result of this competition, the outcome of the BB OA evolution becomes strongly dependent

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on the fragmentation branching ratio associated with a given OA scheme. One of the most significant findings of our analysis in this respect is that the sensitivity of γ_a to the initial plume size (or, similarly, to the dilution rate) may be positive or negative, depending on a the VBS scheme and the BB OA age. Factors controlling the dependence of γ_a on S_p are examined more in detail in the next section (Sect. 3.3).
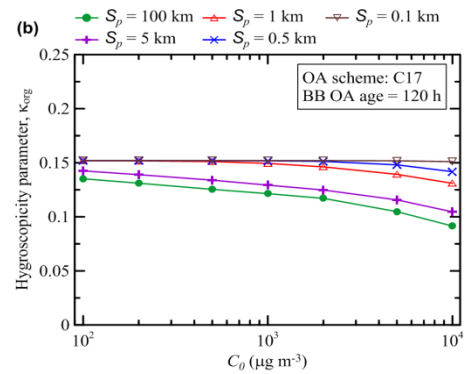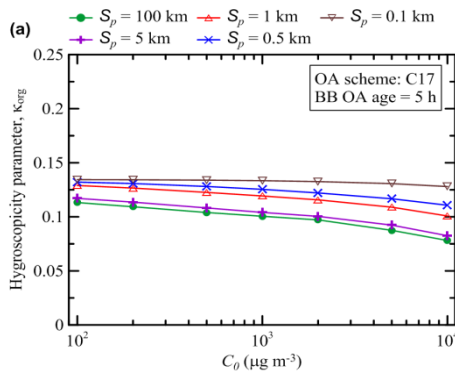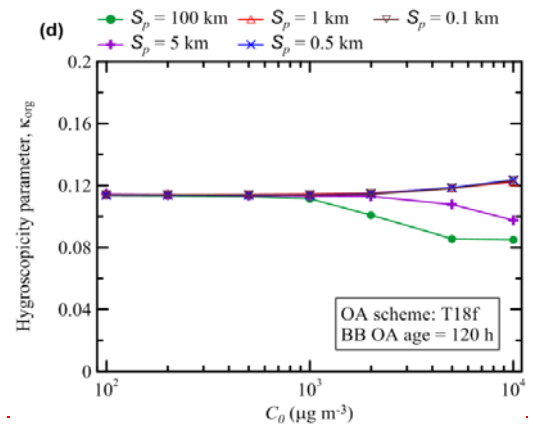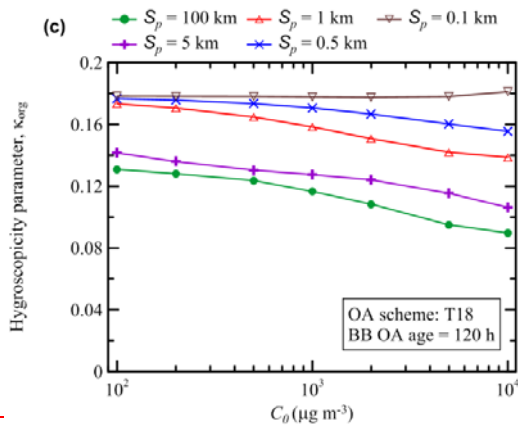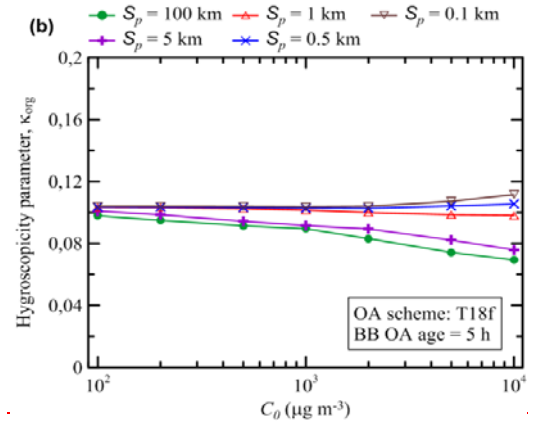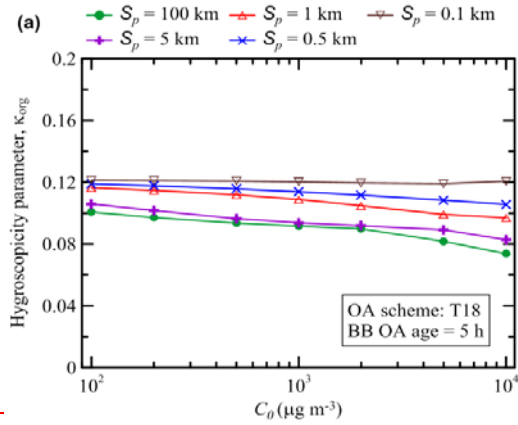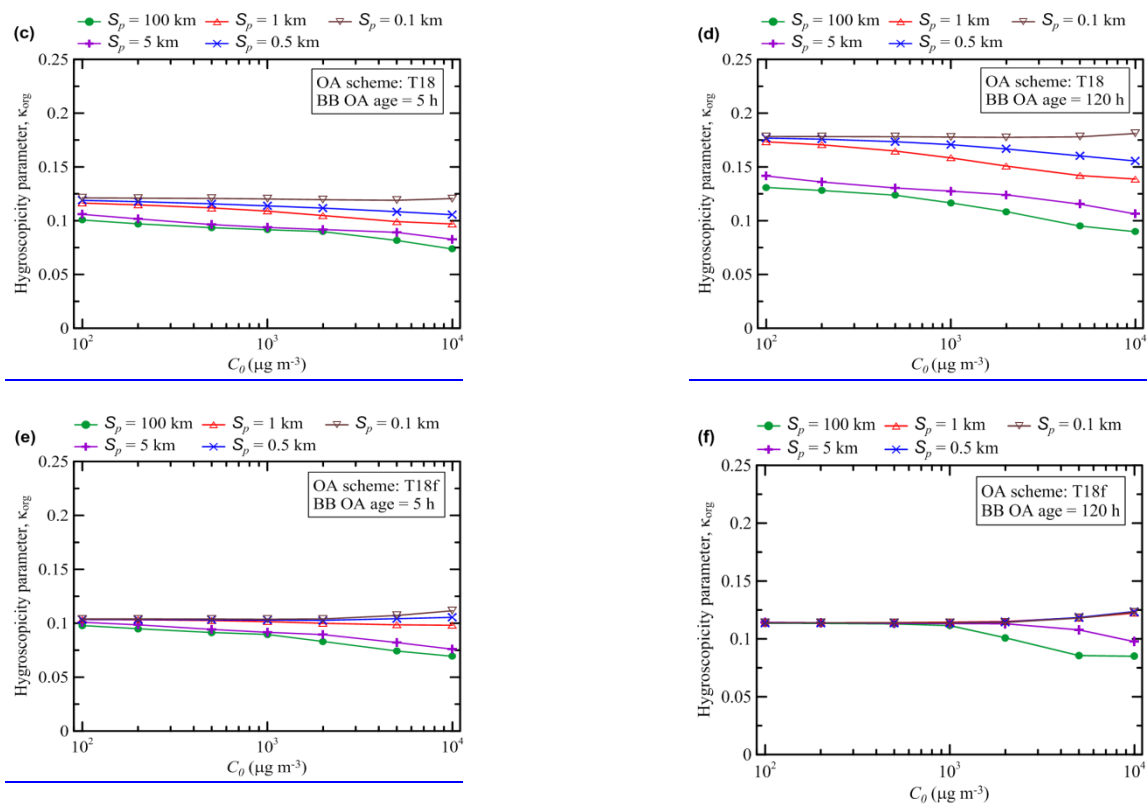
Figure 8: Dependencies of the hygroscopicity parameter, κ_org, after (a, c, be) 5 hours and (b, de, df) 120 hours of BB OA evolution simulated using the (a, eb) C17, (c, d) T18 and (be, df) T18f schemes.
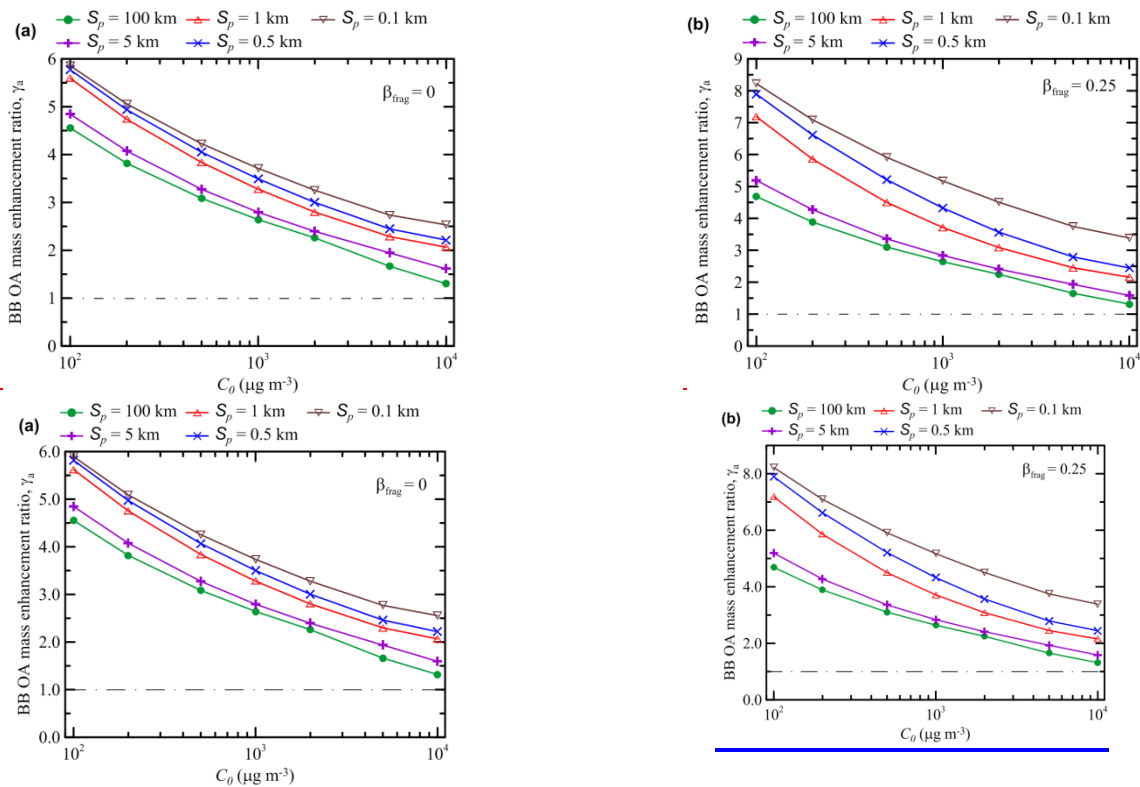
The simulated dependencies of the values of the hygroscopicity parameter after 5 and 120 hours of evolution, $\kappa_{org}(5h)$ and $\kappa_{org}(120h)$, respectively, on the initial BB OA concentration, C_0, and the plume size, S_p, are presented in Fig. 8. Evidently, similar to $\gamma_a(5h)$ and $\gamma_a(120h)$, both $\kappa_{org}(5h)$ and $\kappa_{org}(120h)$ values are also affected by changes of the control parameters in a nonlinear way, although the manifestations of the nonlinearities are not as spectacular pronounced as in the case of the corresponding dependencies for EnR. In particular, while the values of $\kappa_{org}(5h)$ and $\kappa_{org}(120h)$ are decreasing with increasing C_0 for the largest plumes ($S_p = 100$ km), they are nearly constant and even slightly increasing with increasing C_0 for the smallest plumes ($S_p = 0.1$ km). According to the simulations with both all three of the schemes enabling simulations of the hygroscopicity parameter, both $\kappa_{org}(5h)$ and $\kappa_{org}(120h)$ are more sensitive to the changes of S_p than to the changes of C_0. Furthermore, the sensitivity of $\kappa_{org}(5h)$ and $\kappa_{org}(120h)$ to S_p increases with increasing C_0. Note that as C_0 increases, a larger fraction of the total POM concentration, C^{tot}, belongs, under thermodynamic equilibrium, to the condensed phase. This fraction is oxidized (due to as a result of mass transfer between the particles and gas phase) slowly relative compared to the gas-phase fraction, which is why both $\kappa_{org}(5h)$ and $\kappa_{org}(120h)$ mostly tend to decrease with C_0. However, the particulate fraction of C^{tot} can readily evaporate and become accessible for fast gas-phase oxidation as a result of evaporation in the rapidly diluted small plumes: this may explain why $\kappa_{org}(5h)$ and $\kappa_{org}(120h)$ are nearly insensitive to C_0 in the simulations with $S_p = 0.1$ km.

The simulations of κ_{org} performed with the T18 and T18f different schemes are noticeably different both quantitatively and qualitatively. Quantitatively, the C17 and T18 schemes typically yields larger values of κ_{org} than the T18f scheme, and, for the smaller plumes (with $S_p \leq 1$ km), the T18 scheme predicts bigger values of $\kappa_{org}(120h)$ than the C17 scheme. Qualitatively, while aerosol aging is associated with increasing sensitivity of κ_{org} to S_p in the simulations with the T18 scheme (cf. Figs. 8ac and Fig. 8ed), the sensitivity of κ_{org} simulated with the T18f scheme either diminishes or does not change significantly

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(depending on C_0) ~~or diminishes~~ as the aerosol ages in the simulations with the C17 (cf. Figs. 8a and ~~Fig.~~ 8b) and T18f schemes (cf. Figs. ~~8b~~ 8e and ~~Fig.~~ 8~~d~~f)~~)~~. The likely reasons for the differences between the simulations of κ_{org} with, on the one hand, ~~the two~~ the C17 and T18 schemes, and, on the other hand, the T18f scheme, have already been mentioned above (see Sect. 3.~~2~~1). That is, on the one hand, oxidation re~~du~~actions in the C17 and T18 scheme~~s~~ yield less-~~-~~volatile products, which eventually accumulate in particles and increase the overall O:C ratio for BB OA. On the other hand, fragmentation reactions in the T18f scheme result in irreversible loss of the condensable organic matter into the gas phase, thus limiting an increase of the O:C ratio for the condensed phase. The significant differences between the $\kappa_{org}(120h)$ values simulated with the C17 and T18 schemes for the smaller plumes (with $S_p \leq 1$ km) are likely due to the fact that semi-volatile products from oxidation of OA can achieve a higher degree of oxidation in the T18 scheme than in the C17 scheme.

3.3 The effects of fragmentation reactions

As noted in the previous section, our simulations performed with the S15 (or T18f) and K15 schemes reveal qualitatively different dependencies of the BB OA enhancement ratio after 120 hours of evolution, $\gamma_a(120h)$, on the initial concentration C_0, but even more on the initial plume size, S_p. Specifically, while the S15 and T18f schemes predict that $\gamma_a(120h)$ increases with an increase of S_p, the K15 scheme yields an inverse dependence. Meanwhile, the main difference between the S15 and K15 schemes is that the fragmentation pathway of the oxidation reactions has a larger weight ~~with respect~~relative to the functionalization pathway in the S15 scheme~~, compared to~~ than in the K15 scheme. The effect of fragmentation reactions is also well visible in simulations with the T18f scheme~~, when~~ compared to the simulations with the T18 scheme. Taking these considerations into account, we performed additional simulations in which we varied the fragmentation branching ratio in the same scheme. For definiteness, we used the S15 scheme which is rather "transparent" and flexible but yet not oversimplified.
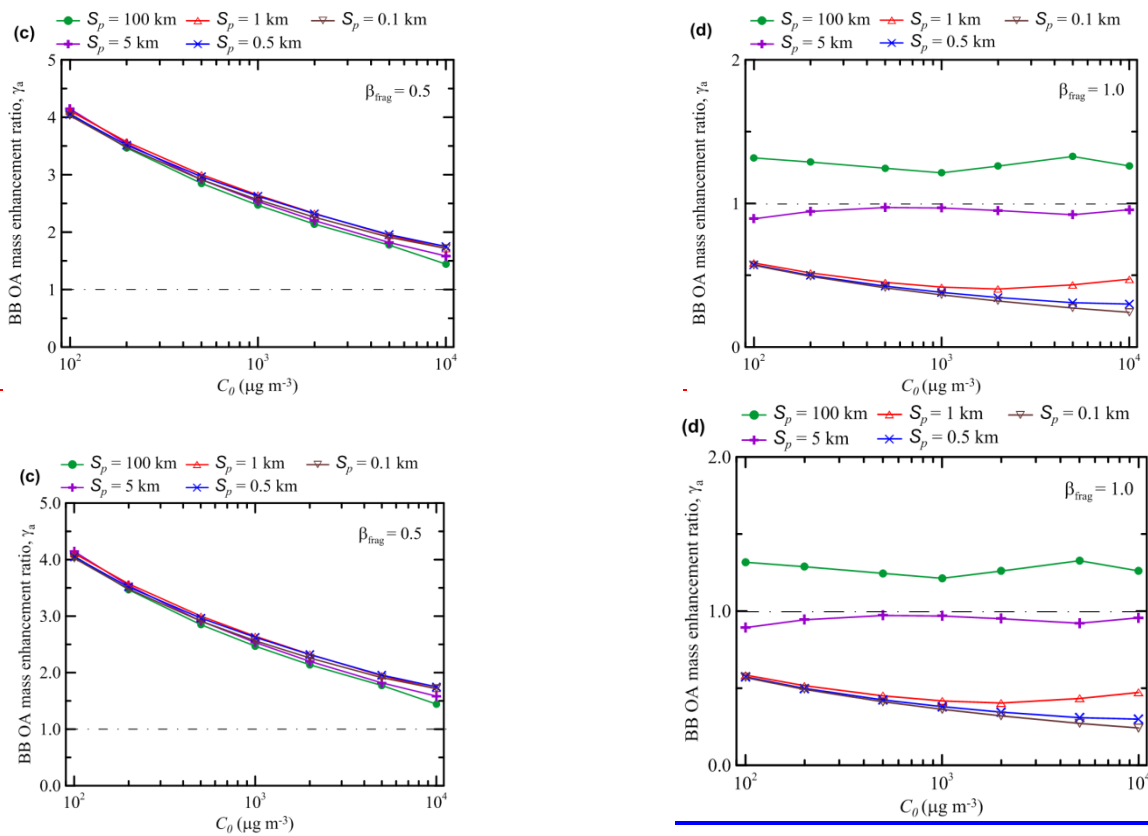
Figure 9: Dependencies of the BB OA enhancement ratio after 120 hours of evolution on the initial BB OA mass concentration, C_0, according to the simulations using the S15 scheme with different values of S_p and varying values of the fragmentation branching ratio: (a) β_{frag} = 0, (b) β_{frag} = 0.25, (c) β_{frag} = 0.5 and (d) β_{frag} = 1.0. A horizontal dash-dotted line in each panel indicates the situation where no BB OA mass enhancement occurs (γ_a = 1).

Figure 9 shows the dependencies of γ_a(120h) on both C_0 and S_p according to our simulations performed with four different values of the fragmentation branching ratio, β_{frag}, for the SOA oxidation reactions (see Eq. 11) : 0, 0.25, 0.5, and 1.0. Figure 7c shows similar calculations ~~corresponding to~~for the case with β_{frag} = 0.85. The simulations with the extreme values of β_{frag} (0 and 1) do not correspond to any realistic situation and are provided only for reference purposes, but taking into account that the fragmentation branching ratio is a highly uncertain parameter of VBS schemes (Donahue et al., 2012b; Shrivastava et al., 2015), neither of the simulations with the other values of β_{frag} is intended to be more or less realistic. It is noticeable that the dependencies obtained using the S15 scheme with β_{frag}=0 (see Fig. 9a) are very similar to those obtained with the K15 scheme (~~see~~ Fig. 7b). Specifically, according to the simulation with both schemes, γ_a(120h) is inversely dependent on S_p. An increase of β_{frag} (~~see~~ Fig. 9b) to 0.25 results in ~~the~~ a stronger sensitivity of γ_a(120h) to S_p and in an increase of maximum values of γ_a(120h) (reached with S_p=0.1 km): apparently, this is due to the fact that the effect of fragmentation of SOGs is counteracted by oxidation of their products which are reprocessed according to Eq. (11). However, as β_{frag} increases further, the reprocessing of SOGs cannot compensate for a loss of SVOCs in the fragmentation branch: the values of γ_a(120h) calculated with β_{frag}=0.5 (~~see~~ Fig. 9c) are typically lower (and sometimes very considerably) than those calculated with β_{frag} =0.25. Furthermore, when β_{frag} equals 0.5, the dependence of γ_a(120h) on S_p "collapses". That is, an approximate balance between the fragmentation and functionalization branches makes the BB OA enhancement ratio after 5 days of evolution nearly insensitive to dilution. Note that there is an obvious similarity between the dependences obtained using, on the one hand, the S15 scheme with the branching ratio of 0.5 (Fig. 9c), and, on the other hand, the C17 and T18 schemes. This similarity implies that the fragmentation and functionalization processes are, effectively, nearly balanced in the C17 and

T18 schemes. The dominance of fragmentation over functionalization is again associated with a rather strong sensitivity of $\gamma_a(120h)$ to S_p (~~see~~ Fig. 7c and Fig. 9d). However, unlike the cases with $\beta_{frag} < 0.5$, $\gamma_a(120h)$ increases with increasing S_p. Another noteworthy feature of the simulations with $\beta_{frag} = 0.85$ or $\beta_{frag} = 1.0$ is a rather weak sensitivity of $\gamma_a(120h)$ to C_0. This is likely because the oxidation of POGs, which are available in larger amounts when C_0 is smaller, yields much less SOA in the cases with dominating fragmentation, compared to the cases with dominating functionalization.

The effects of fragmentation and functionalization reactions on the composition of BB OA in the simulation performed with the S15 scheme on the composition of BB OA are further visualized in Fig. 10, which illustrates the BB OA evolution by distinguishing primary and secondary fractions in particles (POA and SOA) and in the gas phase (POG and SOG), as well as between "fresh" ($n \leq 2$, see Sect. 2.2) and "aged" ($n > 2$) fractions of SOA (SOA-f and SOA-a, respectively). The simulations shown in Fig. 10 were performed for the four cases with different values of C_0 (10^2 and 10^3 µg m^{-3}) and β_{frag} (0 and 1). The initial plume size was taken to be 100 km in all these simulations; that is, the BB OA evolution was practically not affected by dilution.

It is evident that after about 5-10 hours of evolution, SOA already provides a major contribution to the OA mass concentration in all of the cases considered. Not surprisingly, the SOA fraction is larger in the simulations with $\beta_{frag} = 0$ (in the "functionalization" case) than in the simulations with the $\beta_{frag} = 1$ (in the "fragmentation" case). In addition, the aged SOA fraction is larger for the non-fragmentation case. Consistent with the dependence of γ_a on C_0 in Fig. 9a, the SOA fraction calculated with $\beta_{frag} = 0$ is also larger in the simulations with a smaller value of C_0 (cf. Figs. 10a and 10b). A smaller C_0 value is also associated with a larger contribution of SOA-a (and correspondingly, with a smaller contribution of SOA-f) to the total concentration of SOA. It is worth noting that compared with the simulations performed with $C_0 = 10^3$ µg m^{-3}, the simulations with $C_0 = 10^2$ µg m^{-3} feature a faster decrease of SOA-f. This is probably due to an initially larger fraction of SOGs in the simulations with $C_0 = 10^2$ µg m^{-3}: rapid oxidation of "fresh" SOGs causes evaporation of corresponding species from particles, thus depleting SOA-f. A major qualitative difference between the simulations performed for the "functionalization" and "fragmentation" cases is that the OA, SOA, and SOA-a concentrations monotonically increase with time in the former case (regime 1), but ~~demonstrate~~ show a "humped" dependence on time in the latter case (regime 2). That is, the simulations shown in Fig. 10 confirm our initial suggestion (see Sect. 3.1) that realization of regimes "1" and "2" depends on the ratio between the fragmentation and functionalization branches of the oxidation reactions.
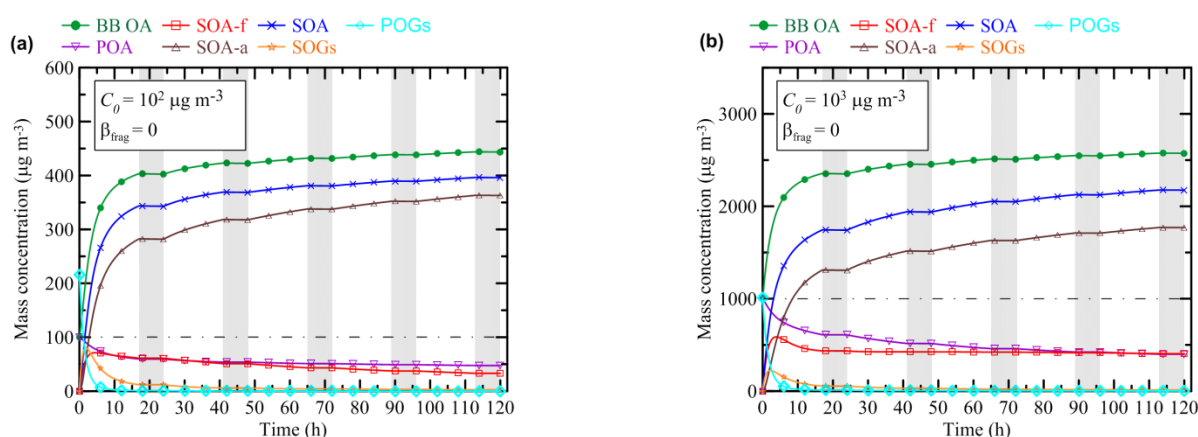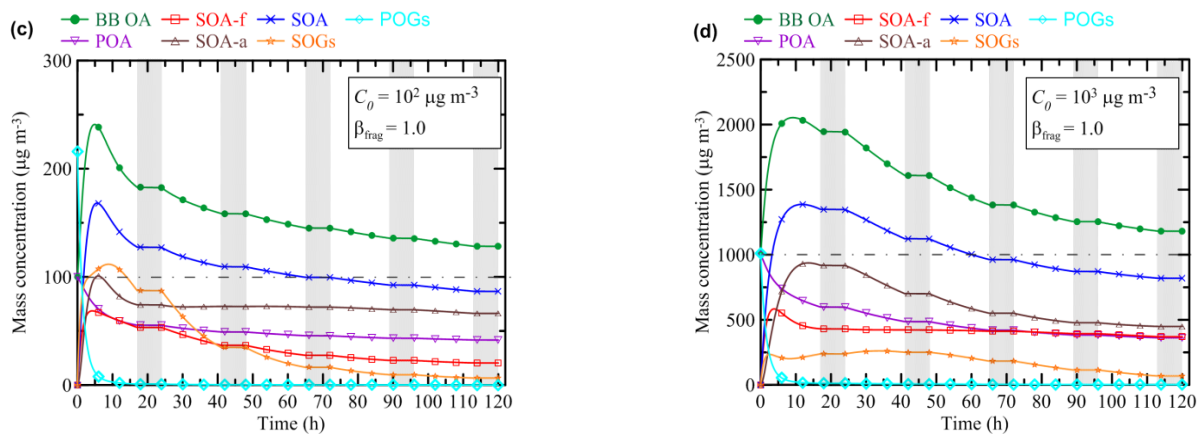
Figure 10: Evolution of the mass concentration of BB OA and its components, including POA, "fresh" SOA (SOA-f), and "aged" SOA (SOA-a) according to the simulations performed with the S15 scheme. Also shown are the total mass concentration of SOA and the mass concentrations of POGs and SOGs. The simulations were done with two different values of β_{frag}: (a, b) $\beta_{frag} = 0$, (c, d) $\beta_{frag} = 1$ and two different values of C_0: (a, c) $C_0 = 10^2$ µg m³, (b, d) $C_0 = 10^3$ µg m³. The initial plume size was set at 100 km in all the simulations. A horizontal dash-dotted line in each panel indicates the initial concentration of BB OA aerosol. Shaded bands depict nighttime periods when no oxidation reactions were allowed to occur.

4 Discussion

In this section, we briefly discuss possible implications and limitations of our findings presented in Sect. 3. First of all, our results indicate that uncertainties associated with the representation of BB OA evolution in chemistry transport models (CTMs) may have a major impact on the simulated behavior of BB aerosol. Indeed, our simulations show that differences between available OA schemes may entail differences in the calculated EnR, (γ_a,) as large as a factor of five (see Fig. 3a) or even bigger greater (see Fig. 2c) after five days of evolution. Furthermore, there may be not only quantitative but also qualitative differences in the BB OA simulations, such as a difference between a gradually increasing γ_a EnR (in the regime "1") and a "humped" temporal dependence of EnRγ_a (in the regime "2") (see, e.g., Fig. 2a). Typically, OA VBS schemes are constrained (at least to some extent) with data from aerosol chamber experiments (e.g., Hennigan et al., 2011) which are, however, rarely representative of more than a few hours of BB aerosol aging under atmospheric conditions. Taking into account that major quantitative and qualitative differences between our simulations with different OA schemes are manifested at longer timescales, our results therefore indicate the particular importance of observational constraints at the time scales ranging from tens of hours to several days. Potentially, such constraints can be provided by in situ aerosol measurements (e.g., Jolleys et al., 2012; Shrivastava et al., 2015) or remote (e.g., satellite) observations (Konovalov et al., 2017).

Even if one of the VBS schemes considered in this study were perfectly adequate, its application in the framework of a typical CTM would still likely entail considerable model biases due to several factors. First, the bias may be due to a dependence of BB OA evolution on its initial concentration. Indeed, smearing of BB emissions from an intense but small (relative to the grid cell size) fire within a model grid cell would result in smaller initial concentration (C_0) of BB OA (C_0) in the model than in the real atmosphere. For example, if the model had a horizontal resolution of 50×50 km² but the fire size was 5×5 km², the initial concentration of BB OA in the model would be two orders of magnitude lower than within the actual plume (e.g., 10^2 instead of 10^4 µg m³). As indicated by our analysis for several cases (see Figs. 6a, b, d and Fig. 7a, b, d), the EnR simulated for such a situation can be strongly overestimated due to its inverse dependence on C_0. Second, a model bias in EnR can be caused by a dependence of EnRγ_a on the initial plume size, S_p. For example, due to the effect of

rapid dilution of BB OA plumes originating from relatively small (with a typical size of about 1 km or less) but multiple fires, a model that does not resolve the plume scales will likely overestimate EnR and corresponding changes of the BB OA mass concentration ~~as a result of~~ after its aging during several days, if the evolution of BB OA is strongly affected by fragmentation (~~see~~ Fig. 7c, e)~~.,~~ ~~but~~ However, it is also possible that a CTM will underestimate EnR, if functionalization dominates over fragmentation (~~see~~ Fig. 7b). ~~–~~ If the BB OA evolution is reproduced adequately with the C17 and T18 scheme~~s~~, then the same situation (where BB OA plumes originate~~ing~~ from relatively small fires) may be associated with a negative bias in the hygroscopicity parameter κ_{org} (~~see~~ Fig. 8~~a~~b, d) rather than with any significant bias in EnR~~γ_a~~ (Fig. 7a, d). Over shorter time scales (a few hours), a CTM is likely to overestimate EnR and underestimate κ_{org} in the case of small fires, irrespective of the relative importance of functionalization and fragmentation reactions (~~see~~ Figs. 6a-e and ~~Fig.~~ 8a, c, e). Note that our findings concerning the impact of the plume size on EnR after a few initial hours of aging are qualitatively consistent with the results of numerical experiments conducted by Bian et al. (2017) and Hodshire et al. (2019). ~~Net~~ The net effects of ~~a~~ the limited horizontal resolution of a CTM on the simulated BB OA evolution are likely very variable and are hardly possible to ~~accurately~~ predict accurately with a box model, as they should be dependent on the spatial structure and intensity of fire emissions as well as on meteorological condition~~s~~. Third, the simulated values of EnR~~γ_a~~ (and to a less~~er~~ extent, those of the hygroscopicity parameter~~κ_{org}~~) may be biased as a result of model errors in the vertical injection profile of fire emissions. Indeed, available parameterizations can strongly underestimate or overestimate the smoke injection height, sometimes by an order of magnitude (Sofiev et al., 2012); these errors can cause biases in EnR as a result of the sensitivity of the simulations of BB OA to uncertainties in both the plume's density and dilution rate, and also to its temperature, affecting thermodynamic ~~equilibrium~~equilibria. Similar effects of chemical nonlinearities on simulations of the consequences of atmospheric processing of BB emissions were first pointed out by Chatfield and Delany (1990).

The results of our analysis may be helpful for understanding the reasons for the observational diversity of BB OA aging effects. Indeed, our findings suggest that the occurrence of differences in the observed effects of BB OA aging, such as an increase (~~see,~~ e.g., Yokelson et al. 2009; Konovalov et al., 2015; Vakkari et al., 2018) or a decrease (e.g., Jolleys et al., 2012; 2015) of EnR can be driven by differences in the initial parameters of the BB plumes, as well as by the BB aerosol's photochemical age and the ratio between the fragmentation and functionalization reaction pathways. In particular, our simulations predict that aging of BB aerosol in a large plume is more likely to result initially in an increase of EnR~~γ_a~~, whereas the initial evolution of a small plume is likely to be associated with an initial decrease of EnR~~γ_a~~ (~~cf., e.g.,~~ Fig~~s~~. 2a and ~~Fig.~~ 2c). Consistent with these predictions, there is observational evidence for a growth of EnR~~γ_a~~ in the situations with large-scale fires (e.g., Konovalov et al., 2015; 2017; Vakkari et al., 2018) as well as evidence for an initial drop of γ_a in a relatively small isolated plume (Akagi et al., 2012). It is noteworthy that in agreement with our simulations corresponding to the regimes~~-~~ "3" and "4", ~~-~~the observations by Akagi et al. (2012) show an increase of EnR~~γ_a~~ after the initial decrease. Our analysis also indicates that if fragmentation dominates over functionalization, an initial growth of EnR~~γ_a~~ in situations with large fires is likely to be followed by a decreasing stage (see the simulations for regime "2" in Fig~~s~~. 2a, b and ~~Fig.~~ 3). A similar "humped" dependence of EnR~~γ_a~~ on the BB aerosol photochemical age was identified earlier in the analysis of satellite observations of BB aerosol in Siberia (Konovalov et al., 2017). The fact that, according to aircraft observations, EnR can be around or smaller than unity even in relatively large plumes transported during a few days (Jolleys et al., 2012; 2015) is consistent with our simulations corresponding to ~~the~~ regimes "2", "4" or "5" in Fig. 3.

Our analysis indicates that BB OA evolution is strongly dependent on the fragmentation branching ratio (~~see~~ Fig. 9) which is known to be a growing function of the oxidation state of OA (Jimenez et al., 2009). It seems reasonable to suggest that the oxidation state of BB OA (and thus the effective fragmentation branching ratio) within a BB plume at any given moment is dependent on the initial chemical composition (and therefore oxidation state) of fresh BB OA. There is evidence (e.g., Grieshop et al., 2009; May et al., 2015; Tiita et al., 2016; Ahern et al., 2019; Lim et al., 2019) that the oxidation state of fresh

BB OA is strongly variable, depending on conditions of burning and fuel type. Variable effects of fragmentation reactions have been suggested to be a possible reason for the diversity of values of the BB OA EnR (ranging from about 0.8 to almost 3) in ~~the~~ BB aerosol aging experiments (Hennigan et al., 2011). Therefore, the differences between our simulations performed with different oxidation schemes may reflect, to some extent, the diversity of the observed aging effects of BB OA due to variability of its initial chemical composition.

It should be especially noted that the VBS schemes employed in our simulations do not comprise the whole range of uncertainties associated with the representation of BB OA aging in models. In particular, all of the considered schemes share the common assumption that OA particles are composed of well-mixed liquids, which is also used in the majority of other modeling studies using the VBS framework. However, based on findings from some laboratory experiments, Shrivastava et al. (2013; 2015) argued that SOA should rather be represented as non-volatile glassy semi-solid mass. Available estimates of the annual-mean SOA glass transition temperature (Shiraiwa et al., 2017) indicate that in major biomass burning regions (such as the Amazon basin and Siberia), SOA is typically in the liquid state near the ground but transits to the semi-glassy state towards the top of the boundary layer. Taking these estimates into account, it seems reasonable to expect that the SOA phase state in ambient BB aerosol in these regions is quite variable, depending on relative humidity and ambient temperature. When the equilibrium state of SOA is semi-solid, the effects of gas-phase fragmentation reactions are expected to diminish (since SOA, once formed, cannot evaporate), depending on the highly uncertain time scale of transformation of semi-volatile SOA into a nonvolatile state (Shrivastava et al., 2015). Another uncertain factor that affects the evolution of the ambient BB aerosol but is not taken into account in our simulations is heterogeneous oxidation of OA particles. Similar to gas-phase fragmentation reactions, heterogeneous oxidation increases the oxidation state of particulate carbon, resulting in formation of volatile products escaping irreversibly to the gas phase. Kroll et al. (2015) estimated carbon loss from the particles after one week of OA aging under typical atmospheric conditions to be in the range from 3 to 13%. Heterogeneous reactions would be slowed down further if the particles are in a semi-solid state (Shiraiwa et al., 2013a). These estimates mean that heterogeneous oxidation is unlikely to affect BB OA behavior significantly at the time scales addressed in this study, except that including even a slow heterogeneous oxidation into our model would inevitably transform regime "1" (a monotonic saturating increase of EnR_{γ_a}, leading to a steady state) into regime "2" (an increase of EnR_{γ_a}, followed by its decrease), as heterogeneous oxidation would eventually convert all particulate carbon into CO_2 (Donahue et al., 2013). Finally, it should again be noted that none of the VBS schemes considered takes into account interaction of SVOCs with water. That is, we essentially assume that both POA and SOA are formed of hydrophobic species. This assumption is not expected to have a strong effect on BB aerosol evolution in a relatively dry atmosphere (with the relative humidity below about 60 %) where the water uptake by BB aerosol is small (Hand et al., 2010). Significant uptake of water can result in important contributions of aqueous phase oxidation reactions to transformations of BB OA chemical composition and to formation of SOA (Brege et al., 2018).

As argued above, the nonlinear behavior of BB OA in our model experiments stems merely from the well-established semi-volatile nature of organic compounds composing it. This fact indicates that the nonlinear features of BB OA evolution, which have been revealed in our simulations, are not a consequence of any simplifications involved in our model but rather an inherent property of real BB aerosol. However, the great diversity of the simulations using different VBS schemes indicates that an accurate quantitative representation of these features in the presently available atmospheric models is yet hardly feasible. Taking into account that regional and global CTMs are not designed to address the scales associated with individual plumes, the results of our study indicate the need for robust subgrid parameterizations of BB OA evolution. Instead of explicitly assuming variable properties of individual smoke plumes and the BB aerosol they contain, such parameterizations might rely on some external observable characteristics of fire emissions, such as the density and size distribution of fires spots and the fire radiative power, as well as on common meteorological parameters (e.g., temperature

and relative humidity). Useful observational constraints for robust representations of atmospheric aging of BB aerosol in CTMs can hopefully be inferred from satellite measurements of BB aerosol (Konovalov et al., 2017).

5 Conclusions

In this paper, we analyzed the role of the intrinsic nonlinearity of the processes driving gas-particle partitioning and oxidation of SVOCs during the atmospheric evolution of BB organic aerosol. We performed simulations of BB OA evolution during a five-day period, using a microphysical box model in which BB OA chemical transformations and SOA formation were represented within the VBS framework. A simple parameterization based on a Gaussian dispersion model was used to specify several scenarios for dilution of a BB plume. The model was run with several VBS schemes of varying ~~complexities~~complexity, including 1D-, 1.5D-, quasi-2D- and 2D--VBS schemes that had been proposed in the literature to represent BB OA evolution specifically in the framework of regional and global chemistry transport models (CTMs). We analyzed the BB OA evolution by considering the BB OA mass enhancement ratio (EnR)~~, γa,~~ as a function of two control parameters, i.e., the initial horizontal plume size, S_p, across the wind direction and the initial BB OA concentration, C_0, corresponding to thermodynamic equilibrium. For a part of our simulations, we also considered the dependence of the hygroscopicity parameter, κ_{org}, on the same parameters. The initial plume size controls the dilution rate (the larger S_p, the smaller the dilution rate), and the initial BB OA concentration determines the partitioning of SVOCs between the gas phase and the particles (the larger C_0, the smaller the gas-phase fraction of SVOCs).

The simulation results allowed us to identify several qualitatively different regimes of BB OA evolution, which feature a monotonic saturating increase of EnR~~γa~~ (regime "1"), increasing and then decreasing EnR~~γa~~ (regime "2"), an increase of EnR~~γa~~ after its initial rapid decrease (regime "3"), a stage with increasing EnR~~γa~~ between two intermittent stages of its decrease (regime "4"), or monotonically decreasing EnR~~γa~~ (regime "5"). The manifestations of nonlinear behavior of BB OA are found to include pronounced dependencies of EnR~~γa~~ on both S_p and C_0. For relatively fresh BB aerosol (with the age ranging from a few hours to several tens of hours), EnR increases as S_p increases or C_0 decreases. However, these kinds of dependencies can be strongly suppressed or even reversed, depending on the VBS scheme used and the aerosol age. Another interesting manifestation of nonlinear behavior of BB OA are possible shifts between the regimes as a result of a change in S_p: for example, the regimes "1" and "2" for slowly diluting large smoke plumes can transform into the regimes "3" and "4", respectively, for small plumes with fast initial dilution. Evolution of κ_{org} is also found to be affected by changes of the control parameters in a nonlinear way, although the manifestations of the nonlinearities are not as spectacular as in the case of the corresponding dependencies for EnR.

Differences between the VBS schemes can result in large quantitative differences between the simulations; they increase with the aerosol age and can almost be as large as a factor of 10 in the EnR values after a five-day evolution under typical conditions in summer mid-latitudes. Such large quantitative differences are usually associated with qualitative differences between the simulations: specifically, the simulations resulting in the largest values of EnR correspond to regime "1" of BB OA evolution, while those yielding relatively small EnR values typically correspond to regimes "2", "4" or "5". Our analysis indicates that one of the major factors behind the quantitative and qualitative differences between the simulations with the different VBS schemes is the ratio between fragmentation and functionalization. Specifically, prevalence of fragmentation over functionalization (when the effective fragmentation branching ratio exceeds 0.5) gives rise to regimes "2" and "4" and is associated with an eventual decrease of EnR~~γa~~, while the dominance of functionalization over fragmentation is associated with regimes "1" and "3", ~~that~~ which correspond to a saturating increase of EnR~~γa~~. A change of the fragmentation branching ratio also can eventually cause a reversal of the dependence of EnR~~γa~~ calculated after a five-day evolution on S_p: that is, EnR~~γa~~ increases with increasing S_p if fragmentation is prevailing over functionalization and decreases otherwise. Future

studies involving a microphysical model with different OA oxidation schemes are needed to examine how the BB OA behavior depends on varying environmental conditions and on several factors (such as, e.g., a possibility of SOA being in semi-glassy state, heterogeneous and aqueous phase oxidation reactions) that have not been taken into account in our simulations.

We argue that in spite of the inevitable limitations of our study, its the results of our study have important implications for modeling of BB OA in the framework of CTMs. First, our analysis allows us to point out nonlinear behavior of the OA system as a possible reason for the observed diversity of effects of aging of ambient BB aerosol (see, e.g., Cubison et al., 2011). A better understanding of the factors behind the diversity of BB OA aging effects is essential for ensuring the efficiency of in situ ambient observations of BB OA as observational constraints to representations of BB OA processes in CTMs. Second, our findings suggest that uncertainties associated with the representation of BB OA in CTMs may have a major impact on the simulated behavior of BB aerosol at the scales of its typical lifetime in the boundary layer with respect to dry deposition. Note that these uncertainties can be especially important in the context of modeling rapid climate change in the Arctic, where BB OA provides a considerable contribution to the radiative balance (Sand et al., 2015). Third, a rather strong sensitivity of EnR evolution to the parameters of a BB plume (such as C_0 and S_p) indicates that application of even a perfect VBS scheme in the framework of available regional or global models would likely entail considerable model biases in simulations of atmospheric transformations of BB aerosol due to the fact that the evolution of individual BB plumes with varying parameters cannot be accurately represented on a typical model grid with a horizontal resolution ranging from tens to a few hundreds of kilometers. Overall, our findings call for the development of subgrid parameterizations of the BB OA evolution, which could be constrained with available in situ and satellite measurements but, at the same time, would be sufficiently robust with respect to nonlinear effects that cannot be properly addressed in typical CTMs.

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