

Responses to Reviewers:

Reviewer 1:

We thank the reviewer for the careful reading of the manuscript and helpful comments. We have revised the manuscript following their suggestions as is described below.

This study could be a very meaningful work. The paper addressed the relevant scientific questions within the scope of ACP. This manuscript studied the possible reasons enhancing the ozone formation under high PM_{2.5} concentrations. It is not a very novel concept since some previous studies already reported the positive correlation between PM_{2.5} and ozone, and analyzed the underestimated HONO sources in China and other areas in the world. However, better understanding the mechanisms in different locations is scientifically significant in modeling studies. In addition, as the authors mentioned, the results bring important insights for control strategy of air pollution, because both PM_{2.5} and ozone are significant air pollutants in China.

There sever major concerns as follows:

- (1) Both cloud and aerosol can affect the solar radiation. In order to separate these two factors, especially for case studies, people usually will analyze the meteorological conditions during the measurement period, or only analyze the data under the cloud-free conditions. However, the authors of this manuscript never mentioned the cloud factor.

Thanks for the valuable comments of the reviewer. We have checked the meteorological condition (especially cloud condition) during the period of the case study (between Oct 5 and 6, 2015) in the Beijing region. It shows that there was close to the cloud free condition (see attached Fig-A1. Now it is Fig. 5 in the revised paper). In order to evaluate the effect of cloud, we made additional model runs (with thin and thick cloud conditions). The results show that clouds have important impact on the result of this study, and this study is more suitable for the cloud free conditions (see attached Fig-A2. Now it is Fig. 12 in the revised paper). The results show that the thin cloud (cloud cover in 2 km, with cloud water of 10 g/m³), could reduce the photolysis rate of HONO by about 40%, but the HONO could still remain important effects. However, with dense cloud condition (cloud covers at 2 and 3 km, with cloud water of 50 10 g/m³), the photolysis rate of HONO could reduce by 9-10 times by the cloud. In this case, adding photolysis rate of HONO cannot produce important effect on OH radicals and the production of O₃. The above statements have been added in the revised manuscript.

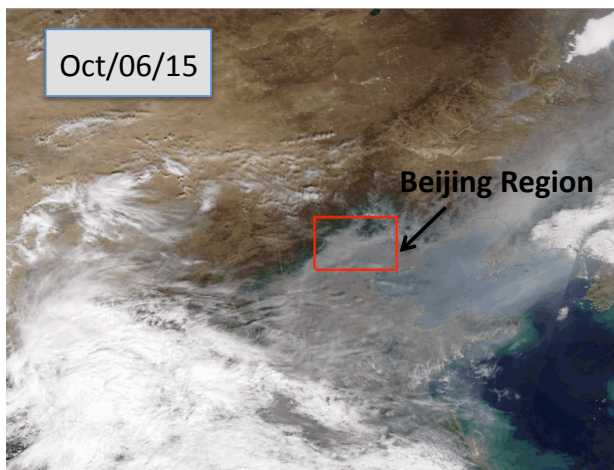
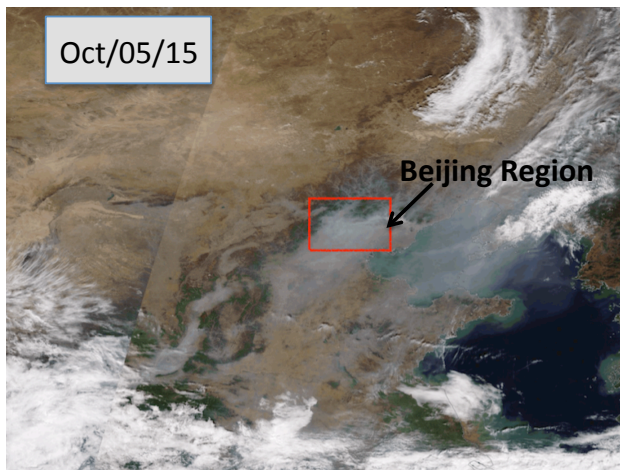


Fig-A1. The cloud condition during the period of the case study (between Oct 5 and 6, 2015 in the Beijing region. The bright white color shows the cloud covers, and the grey white shows the haze covers. The Beijing region is under the heavy haze conditions during the period.

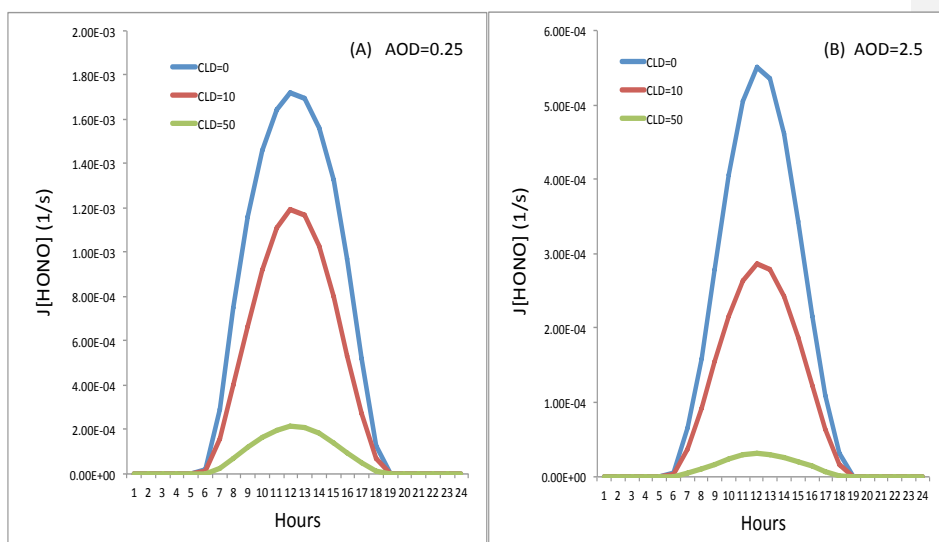


Fig-A2. The effect of cloud cover on the photolysis rate of HONO ($J[\text{HONO}]$). The blue, red, and green lines represent the cloud water vapor of 0 (cloud-free), $10 \text{ (g/m}^3 \text{ - thin cloud)}$, and $50 \text{ (g/m}^3 \text{ - thick cloud)}$, respectively. The left panel (A) represents the light aerosol condition, with AOD of 0.25, and the right panel (B) represents the heavy aerosol condition, with AOD of 2.5.

(2) Several important previous studies should be mentioned so that some conclusions from this manuscript can be more solid. For example (but not limited to), Zhang et al. (2016) already parameterized up-to-date HONO sources into WRF-Chem model such as the heterogeneous reactions on ground and aerosol surfaces, direct vehicle and vessel emissions, conversion of NO_2 at the ocean surface, and emissions from soil bacteria. The modified WRF-Chem substantially reproduced the observed HONO levels, and greatly improved the ozone simulations. However, in this manuscript, the calculated HONO level was still very low in Fig. 6. More information about the WRF-Chem setup is needed. In addition, some other studies (e.g., Shi et al., 2015) already reported the positive correlation between aerosol and ozone. The ozone formation is also strongly dependent on the aerosol size and composition. The process might be a complex interaction between aerosols and photochemical reactions. For example, the scattering aerosol could considerably diffuse the solar radiation and enhance the flux density inside the boundary layer (He and Carmichael, 1999). Thus, the scattering aerosols may favor the ozone formation through increasing solar flux in the boundary layer (Shi et al., 2015). More discussions are needed in the manuscript.

Thanks for the valuable comments of the reviewer. We think that adding these previous studies will enhance the understanding of the highlights of our paper. The reviewer points that some recent versions of the WRF-Chem model add some missing HONO sources (surface emissions, conversion of NO_2 at the ocean surface, etc.) can improve the HONO calculations (Shi et al., 2015). In our calculation, we only use the classical gas-phase

chemistry to illustrate that the importance of these missing sources for the production of OH radicals. Adding these missing sources (there are not fully understand and remain a large uncertainty) could be a future work. In the revised paper, we add the above clarifications. We also add the reference of He and Carmichael (1999) to add their point that there maybe another factor that the ratio of the scattering and absorbing aerosols could be another factor to affect the relationship between aerosols and ozone. All the valuable references are included in the revised paper.

Zhang, L., Wang, T., Zhang, Q., Zheng, J., Xu, Z., & Lv, M. (2016). Potential sources of nitrous acid (HONO) and their impacts on ozone: A WRF/Chem study in a polluted subtropical region. *Journal of Geophysical Research: Atmospheres*, 121(7), 3645- 3662.

Shi, C., Wang, S., Liu, R., Zhou, R., Li, D., Wang, W., ... & Zhou, B. (2015). A study of aerosol optical properties during ozone pollution episodes in 2013 over Shanghai, China. *Atmospheric Research*, 153, 235-249.

He, S., & Carmichael, G. R. (1999). Sensitivity of photolysis rates and ozone production in the troposphere to aerosol properties. *Journal of Geophysical Research: Atmospheres*, 104(D21), 26307-26324.

Generally, this manuscript presents a significant study; however, the analysis should be in more depth. The authors should give proper credit to related work, and clearly indicate this manuscript's original contribution. I would not recommend using a vague word (such as "low solar radiation") in the title.

Thanks. We change title from "Ozone formation under low solar radiation in eastern China" to "Ozone enhancement due to photo-disassociation of nitrous acid in eastern China"

Responses to Reviewers:

Reviewer 2:

We thank the reviewer for the careful reading of the manuscript and helpful comments. We have revised the manuscript following their suggestions as is described below.

This work tried to explain the measured co-occurrence of high PM_{2.5} and O₃ concentrations. The authors report that the high daytime HONO concentrations could be photo-dissociated to be OH radicals, which enhance the photochemical production of O₃, although depressed solar radiation under heavy PM_{2.5} pollutions. It is an interesting scientific issues. However, the data and method in the manuscript do not support such a conclusion very well at this stage.

My major concerns are listed as follows:

- (1) The authors mixed observations from Shanghai and Beijing to create an illusion. There are no observations to show high PM_{2.5}-O₃-HONO concentrations both at Shanghai and at Beijing. I just see high PM_{2.5}-O₃ during Oct.5-6, 2015 in Beijing and high PM_{2.5}-HONO during September, 2009 in Shanghai.

Thanks for pointing out this issue. The reason we chose the data by the following reasons.

(1) Because the co-occurrence between O₃ and PM_{2.5} are not always happened, it happens only in some episodes, especially in spring and fall. In winter, O₃ and PM_{2.5} are actually anti-correlated due to low solar radiation (This also can see in Fig. 2 of the paper). It occurs under the following condition, (a) under cloud-free condition, (b) solar radiation is not too low, (c) during heavy aerosol pollutions in large cities in eastern China. Due to these limitations, it requires continuously measurements of O₃ and PM_{2.5}, and HONO concentrations. Recently, there are some continuously measurements of PM_{2.5}, and O₃ concentrations released by EPA of China. However, HONO measurements are not continuously measured, and we cannot find the HONO data with the period of co-occurrence between O₃ and PM_{2.5}. However, we do find some HONO measurements, which all shows that in all major Chinese cities in either fall or winter (Shanghai, Beijing, and Xian), the HONO concentrations were significant higher than other regions (see attached Fig-A1; Now in Fig. 8 of revised paper). For example, HONO concentrations reached highest in night, ranging from 1 to 2.5 ppbv in the morning at 6-9am. In daytime, the concentrations were lowest (ranging from 0.3 to 1.0 ppbv at 12-18pm), but the concentrations were still significant higher than other regions, which could have significant effect on the production of OH radicals in daytime. As a result, we think that the high HONO is a common event in large cities in eastern China, especially in daytime. This high daytime high HONO is supported by the measurements in previous studies (Zhang et al. 2016; Huang et al. 2017). In this study, we make an assumption that the co-occurrence between O₃ and PM_{2.5} occurred under high HONO concentrations. From Fig.-A1, we also note that using this assumption may result in some uncertainties in estimating the effect of HONO on OH. For example, using the measured HONO in Xi'an

and Beijing could produce 1-2 times higher OH production by photolysis of HONO than the result by using the measured data from Shanghai. In this case, we use the measured HONO from Shanghai to avoid the over estimate of the HONO effect, which can be considered as a low-limit estimation. The above statements are added in the revised paper.

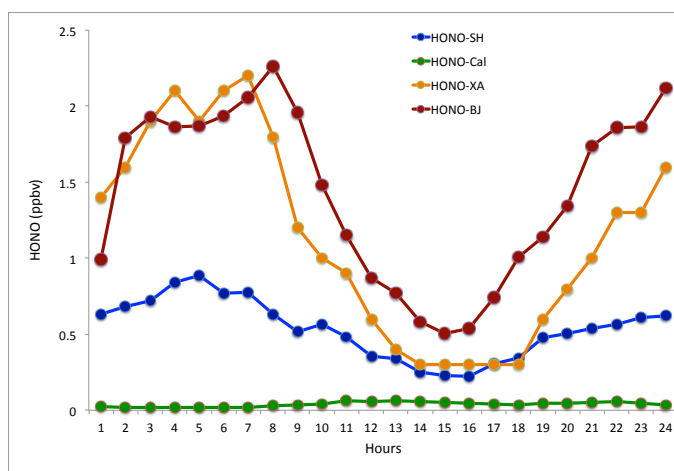


Fig. A1. The measured HONO concentrations (ppbv) in three large cities in China. The red line was measured in Xi'An from 24 July to August 6, 2015. The blue line was measured in Shanghai from 9 to 18 September, 2009. The dark-red line was measured in Beijing from 1 to 27 January, 2014. The green line is calculated by the WRF-Chem model. The measurement in fall of Shanghai is applied to the calculation for the OH production of HONO.

(2) Is the observed co-occurrence of high PM2.5 and O3 concentrations of statistical significance? Are the authors sure it's (measurements during Oct.5-6) not a special case?

The co-occurrence of high PM2.5 and O3 concentrations was occurred in several cases in the past years. The attached Fig.-A2 shows some examples. Because it happened under some special conditions (see the reply in question 1), it most occurred in spring and fall seasons.

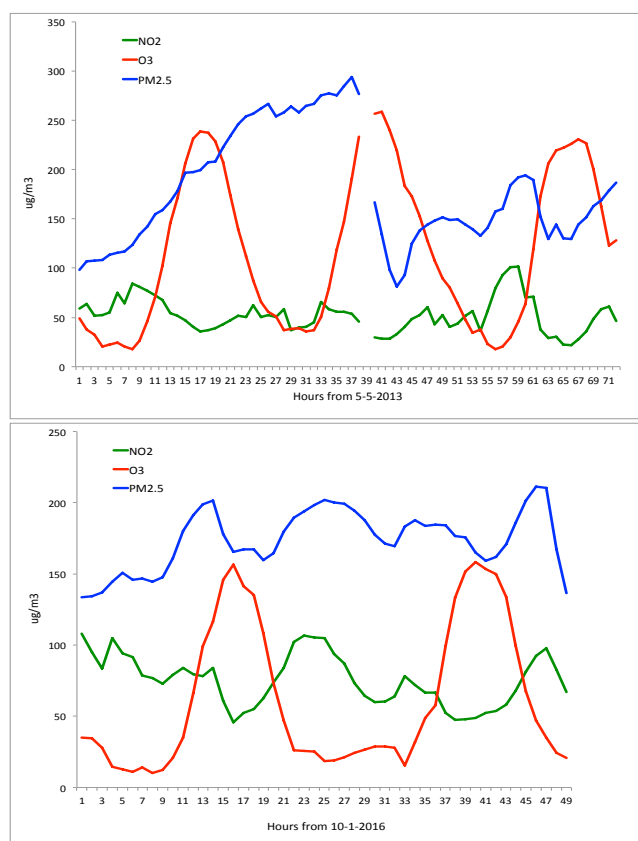


Fig.-A2. The diurnal variations of PM_{2.5} (blue line) and O₃ (red line), and NO₂ (green line) during the periods (from May 5 to May 7, 2013 (upper panel) and from Oct. 1 to Oct. 2, 2016 (lower panel)).

(3) Could the authors make an effort to exclude the effects of precursor emissions (e.g., being sure that the VOCs/NO_x ratios are not more beneficial for ozone production during Oct.5-6 than other days) and meteorological conditions (e.g., temperature and relative humidity; under low humidity, although the PM_{2.5} concentration is high, the solar radiation would not be depressed much)? Moreover, there are no observations show the solar radiation are exactly depressed during Oct.5-6 in Beijing or September in Shanghai?

Thanks for the valuable comments. We tried to find the available data, which is available during the period of Oct. 5 to 6, 2015. We do find some interesting data, which could answer the some comments of the reviewer. The additional data also helps to improve the quality of the paper. Fig.-A3 (now Fig. 5 in the revised paper) shows the cloud conditions in Beijing. During the period, there was close to the cloud free condition, but there was a very heavy aerosol layer. Fig.-A4 shows the relative humidity (RH) conditions. It shows that the RH (%) was generally higher than

60%, with a maximum of 95% during the period. As a result, the high aerosol concentrations accompanied by high RH produced important effects on solar radiation. As shown in Fig.-A5 (now Fig. 6 in revised paper), the daytime averaged solar radiation was significantly reduced (about 40% reduction in Oct. 5-6 compared with the value of Oct. 8). We thanks the comments by the reviewer, these addition (figures and text) can significant enhance the quality of the paper.

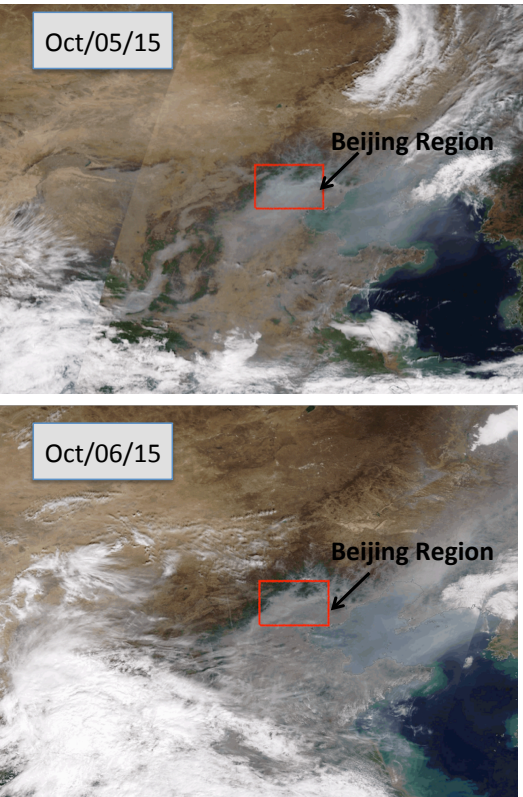


Fig-A3. The cloud condition during the period of the case study (between Oct 5 and 6, 2015 in the Beijing region. The bright white color shows the cloud covers, and the grey white shows the haze covers. The Beijing region is under the heavy haze conditions during the period.

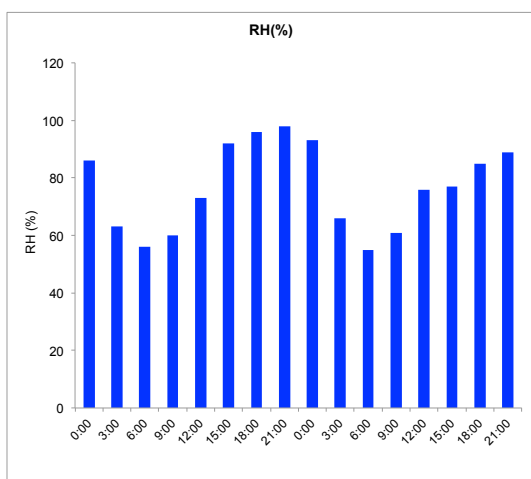


Fig-A4. The measured relative humidity (RH) conditions between Oct. 5 and Oct.6, 2015.

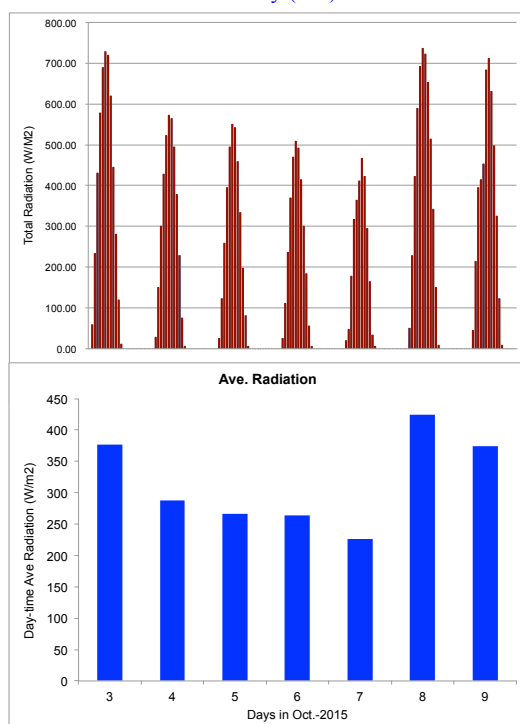


Fig-A5. The measured solar radiation (W/m^2) from Oct. 3 to Oct. 9, 2015 in Beijing. The upper panel shows hourly values, and the lower panel shows the daytime averaged values.

(4) If the authors insist the high PM2.5-O3-HONO mechanism, could this possible new mechanism be added to the WRF-Chem model for verification?

Adding the high PM2.5-O3-HONO mechanism is a very challenge work, and could be another

207 scientific work in the future. The major difficulty is that the causes (surface emissions or chemical
 208 transformations?) for the high HONO concentrations in large Chinese cities are not clearly
 209 understood. This could be a very interesting work in the future.
 210

211 (5) Discussion in sect.3.3: the conclusion (solar radiation in winter reaches a threshold level to
 212 prevent the OH chemical production, even by including the HONO production term) came too
 213 hastily without no direct evidence.
 214

215 Thanks for the comment. We agree with the reviewer that this conclusion is not very certain, and
 216 we re-write these sentences to soft the tone of this conclusion. In the revised paper, we change
 217 “When the solar radiation is in a very low level in winter, it reaches the threshold level to prevent
 218 the OH chemical production, even by including the HONO production of OH.” to “Because the
 219 solar radiation is in a very low level in winter, adding the photolysis of HONO has smaller effect
 220 in winter than in fall, and OH remains low values by including the HONO production term.”
 221

222 Specific comments:
 223

224 (1) L167-169: there are no data to show the solar radiation are reduced
 225 We add a new figure and text to show the solar reduction.
 226

227 (2) L185: same above
 228 Answered in the above.
 229

230 (3) L188-190: same above
 231 Answered in the above.
 232

233 (4) L199: "Chine" should be "China"
 234 Corrected.
 235

236 (5) L201: removed "OH"
 237 Corrected.
 238

239 (6) L218: what is "am" in $\text{O1D} + \text{am} \rightarrow \text{O3P}$
 240 am represents air mass in chemical reaction equations.
 241

242 (7) L222: "Madronich and Flocke (1999)" should be "(Madronich and Flocke, 1999)"
 243 Corrected.
 244

245 (8) L295-296: one of "P1" should be "P2"?
 246 Corrected.
 247

248 (9) L298-299: one of "P1" should be "P2" ? (10) L241: What are possible sources of HONO?
 249 Corrected. The possible sources of HONO could be surface sources or heterogeneous chemical
 250 reactions (but they are not fully understood at present).

251

252 **Reference:**

253

254 Zhang, L., Wang, T., Zhang, Q., Zheng, J., Xu, Z., & Lv, M. (2016). Potential sources of nitrous
255 acid (HONO) and their impacts on ozone: A WRF/Chem study in a polluted subtropical region.
256 Journal of Geophysical Research: Atmospheres, 121(7), 3645- 3662.

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258 Huang, R. J., L. Yang, JJ Cao, QY Wang, X. Tie, et al., Concentration and sources of atmospheric
259 nitrous acid (HONO) at an urban site in Western China. *Sci. of Total Environ.*, 593-594, 165-172,
260 doi.org/10.1016/j.scitotenv.2017.02.166, 2017.

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Ozone enhancement due to photo-disassociation of nitrous acid in eastern China

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Abstract

PM_{2.5}, a particulate matter with a diameter of 2.5 micrometers or less, is one of the major components of the air pollution in eastern China. In the past few years, China's government made strong efforts to reduce the PM_{2.5} pollutions. However, another important pollutant (ozone) becomes an important problem in eastern China. Ozone (O₃) is produced by photochemistry, which requires solar radiation for the formation of O₃. Under heavy PM_{2.5} pollution, the solar radiation is often depressed, and the photochemical production of O₃ is prohibited. This study shows that during fall in eastern China, under heavy PM_{2.5} pollutions, there were often strong O₃ photochemical productions, causing a co-occurrence of high PM_{2.5} and O₃ concentrations. This co-occurrence of high PM_{2.5} and O₃ is un-usual and is the main focus of this study. Recent measurements show that there were often high HONO surface concentrations in major Chinese mega cities, especially during daytime, with maximum concentrations ranging from 0.5 to 2 ppbv. It is also interesting to note that the high HONO concentrations were occurred during high aerosol concentration periods, suggesting that there were additional HONO surface sources in eastern China. Under the high daytime HONO concentrations, HONO can be photo-dissociated to be OH radicals, which enhance the photochemical production of O₃. In order to study the above scientific issues, a radiative transfer model (TUV; Tropospheric Ultraviolet-Visible) is used in this study, and a chemical steady state model is established to calculate OH radical concentrations. The calculations show that by including the OH production of the photo-dissociated of HONO, the calculated OH concentrations are significantly higher than the values without including this production. For example, by including HONO production, the maximum of OH concentration under the high aerosol condition (AOD=2.5) is similar to the value under low aerosol condition (AOD=0.25) in the no-HONO case. This result suggests that even under the high aerosol condition, the chemical oxidizing process for O₃ production can occurred, which explain the co-occurrence of high PM_{2.5} and high O₃ in [spring and fall seasons](#) in eastern China. However, the O₃ concentrations were not

322 significantly affected by the appearance of HONO in winter. This study shows that
323 the seasonal variation of solar radiation plays important roles for controlling the OH
324 production in winter. Because the solar radiation is in a very low level in winter,
325 adding the photolysis of HONO has smaller effect in winter than in fall, and OH
326 remains low values by including the HONO production term. This study provides
327 some important scientific highlights to better understand the O₃ pollutions in eastern
328 China.

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330 **Keywords; High PM_{2.5} and O₃, eastern China, HONO photolysis**
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1. Introduction

Currently, China is undergoing a rapid economic development, resulting in a higher demand for energy and greater use of fossil fuels. As a result, the high emissions of pollutants produce heavy pollutions in mega cities of eastern China, such as Beijing and Shanghai. For example, in the city of Shanghai (a largest mega city in China), the urban and economical developments of the city are very rapid. During 1990 to 2015, the population increased from 13.3 to 24.1 million. The number of automobiles increased from 0.2 million (1993) to 2.0 million (2011). The rapid growing population and energy usage caused a rapid increase in the emissions of pollutants, leading to severe air pollution problems in these mega cities (Zhang et al., 2006; Geng et al., 2007; Deng et al., 2008).

Measurements, such as satellite observations have revealed much higher aerosol pollution in eastern China than in eastern US (Tie et al., 2006). The high aerosol pollution causes a wide range of environmental consequences. According to a study by Tie et al. (2009a), exposure to extremely high particle concentrations leads to a great increase of lung cancer cases. High PM (particulate matter) concentrations also significantly reduce the range of visibility in China's mega cities (Deng et al., 2008). According to a recent study, the high aerosol pollution causes important effects on the crop (rice and wheat) production in eastern China (Tie et al., 2016).

In the troposphere, ozone formation is resulted from a complicated chemical process, and requires ozone precursors, such as VOCs (volatile organic carbons) and $\text{NO}_x = \text{NO} + \text{NO}_2$ (nitrogen oxides) (Sillman, 1995). As the increase in industrial activity and number of automobiles, the precursors of ozone (O_3) and the global budget of oxidization are also significantly increased (Huang et al., 2017; Huang et al., 2018). As a result, O_3 pollutions are becomes a serous pollution problem in Shanghai and other Chinese mega cities (Geng et al., 2010; Tie 2009b; Tie et al., 2015). The effects on O_3 production rate can be characterized as either NO_x -sensitive or VOC-sensitive conditions (Sillman, 1995; Zhang et al., 2003; Lei et al., 2004; Tie et al., 2013). Thus, better understanding the trends of O_3 precursors (VOCs, NO_x) is important to

374 determine the O₃ trends in Shanghai (as well as many large cities in China).

375 In the past few years, China's government made strong efforts to reduce the PM_{2.5}
376 pollutions. However, another important pollutant (O₃) becomes an important problem
377 in eastern China. Several studies regarding the O₃ formation are previously studied in
378 Shanghai. For example, Geng et al. (2007; 2008) study the relationship between O₃
379 precursors (NO_x and VOCs) for the ozone formation in Shanghai. Tie et al. (2009)
380 study the short-term variability of O₃ in Shanghai. Their study suggested that in
381 addition to the ozone precursors, meteorological conditions, such as regional transport,
382 have also strong impacts on the ozone concentrations. During September 2009, a
383 major field experiment (the MIRAGE-Shanghai) was conducted in Shanghai, and
384 multiply chemical species were measured during the experiment. The summary of the
385 measurements by Tie et al (2013) suggests that the ozone formation in Shanghai is
386 under VOC-sensitive condition. However, if the emission ration of NO_x/VOCs
387 reduces to a lower value (0.1-0.2), the ozone formation in Shanghai will switch from
388 VOC-sensitive condition to NO_x-sensitive condition.

389 Despite of some progresses have been made for the ozone formation in mega cities in
390 China, it is still lack of study of ozone development in large cities of China. For
391 example, this study shows that during fall in eastern China, under heavy PM_{2.5}
392 pollutions, there were often strong O₃ chemical productions, causing the
393 co-occurrence of high PM_{2.5} and O₃ concentrations. Under heavy aerosol condition,
394 the solar radiation is depressed, significantly reducing the photochemical production
395 of O₃. This co-occurrence of high PM_{2.5} and O₃ is an unusual and is the focus of this
396 study. [He and Carmichael \(1999\) suggest that aerosol particles can enhance the](#)
397 [scattering of solar radiation, enhancing the flux density inside the boundary layer.](#)
398 Recent measurements [also](#) show that there were often high HONO concentrations in
399 major Chinese mega cities, especially during daytime, with maximum concentrations
400 ranging from 0.5 to 2 ppbv (Huang et al., 2017). [Shi et al. \(2015\) suggest that there](#)
401 [are several potential HONO sources, including surface emissions, conversion of NO₂](#)
402 [at the ocean surface, etc., and adding these sources can improve the calculated HONO](#)

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[concentrations](#). It is also interesting to note that the high HONO surface concentrations were occurred during high aerosol concentration periods, suggesting that there are additional HONO surface sources in eastern China. Under the high daytime HONO concentrations, HONO can be photo-dissociated to be OH radicals, which enhance the photochemical production of O₃.

The paper is organized as follows: in Section 2, we describe the measurement of O₃ and PM_{2.5}. In Section 3, we describe the calculation of photo-dissociated rate of HONO and a steady state model for the calculation of OH, and the causes of high O₃ production under the heavy aerosol condition. Section 4 shows a brief conclusion of the results.

2. Measurements of O₃ and PM_{2.5}

There are long-term measurements in Eastern China by Chinese Environment Protection Agency (CEPA) for monitoring the air quality in China. In eastern China, especially in the capital city of China (Beijing), there are often heavy air pollutions, especially for fine particular matter (PM_{2.5} – the radius of particle being less than 2.5 μm). Figure 1 shows the measurement sites in Beijing, in which the measured concentrations of PM_{2.5} and O₃ are used to the analysis. In the region, the air pollutions were very heavy, especially in winter (Long et al., 2016; Tie et al., 2017). The previous studies suggested that the both aerosol and O₃ pollutions became the major pollutants in the region (Li et al., 2017).

Figure 2 shows the daily averaged concentrations of PM_{2.5} and O₃ in the Beijing region in 2015. The daily averaged concentrations show that there were strong daily and seasonal variations for both the concentrations of PM_{2.5} and O₃. Despite the daily variation, the concentrations of PM_{2.5} existed a strong seasonal variation. For example,

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there were very high concentrations during winter, with maximum of $\sim 300 \mu\text{g}/\text{m}^3$. While in summer, the maximum concentrations reduced to $\sim 150 \mu\text{g}/\text{m}^3$. The seasonal variability of O_3 concentrations were opposite with the $\text{PM}_{2.5}$ concentrations, with lower concentrations in winter ($< 50 \mu\text{g}/\text{m}^3$) and higher concentrations in summer ($> 150 \mu\text{g}/\text{m}^3$). These seasonal variations of $\text{PM}_{2.5}$ and O_3 have been studied by previous studies (Tie and Cao, 2017; Li et al., 2017). Their results suggest that the winter high $\text{PM}_{2.5}$ concentrations were resulted from the combination of both the high emissions (heating season in the Beijing region), and poor meteorological ventilation conditions, such as lower PBL (Planetary Boundary Layer) height (Quan et al., 2013; Tie et al. 2015). According to the photochemical theory of O_3 formation, the summer high and winter low O_3 concentrations are mainly due to seasonal variation of the solar radiation (Seinfeld, J. H. and Pandis, 2006).

In addition to the seasonal variation of solar radiation, the heavy aerosol concentrations play important roles to reduce solar radiation, causing the reduction of solar radiation and O_3 formation (Bian et al., 2007). As we show in Fig. 3a, during wintertime, the O_3 concentrations were strong anti-correlated with the $\text{PM}_{2.5}$ concentrations, suggesting that the reduction of solar radiation by aerosol particles have important impact on the reduction of O_3 concentrations. Figure 3a also shows that the relationship between O_3 and $\text{PM}_{2.5}$ was not linearly related. For example, when the concentrations of $\text{PM}_{2.5}$ were less than $100 \mu\text{g}/\text{m}^3$, O_3 concentrations rapidly decreased with the increase of $\text{PM}_{2.5}$ concentrations. In contrast, when the concentrations of $\text{PM}_{2.5}$ were greater than $100 \mu\text{g}/\text{m}^3$, O_3 concentrations slowly decreased with the increase of $\text{PM}_{2.5}$ concentrations. This is consistent with the result of Bian et al (2007).

It is interesting to note that during late spring, summer, and early fall periods, the correlation between $\text{PM}_{2.5}$ and O_3 concentrations was positive relationship compared to the negative relationship in winter (see Fig. 3b). This result suggests that O_3 production was high during the heavy haze period, despite the solar radiation was

greatly depressed. In order to clearly display this unusual event, we illustrate diurnal variations of $\text{PM}_{2.5}$ and O_3 , and NO_2 during a fall period (from Oct.5 to Oct. 6, 2015). Figure 4 shows that during this period (as a case study), the $\text{PM}_{2.5}$ concentrations were very high, ranging from 150 to 320 $\mu\text{g}/\text{m}^3$. Under such high aerosol condition, the solar radiation should be significantly reduced, and O_3 photochemical production would be reduced. However, the diurnal variation of O_3 was unexpectedly strong, with high noontime concentration of $>220 \mu\text{g}/\text{m}^3$ and very low nighttime concentration of $\sim 25 \mu\text{g}/\text{m}^3$. This strong diurnal variation was due to the photochemical activity, which suggested that during relatively low solar conditions, the photochemical activities of O_3 production was high. According to the theory of the O_3 chemical production, the high O_3 production is related to high oxidant of OH (Seinfeld and Pandis, 2006), which should not be occurred during lower solar radiation. This result brings important issue for air pollution control strategy, because the both air pollutants (high $\text{PM}_{2.5}$ and O_3) were important air pollution problems in eastern China.

To clearly understand the effect of the high aerosol concentrations on solar radiation, we investigate the meteorological conditions, such as cloud covers, relative humidity (RH), and solar radiation during the period of the case study (see Figs. 5 and 6). Figure 5 shows that the cloud condition was close to the cloud free condition, but there was a very heavy aerosol layer in the Beijing region, suggesting that cloud cover played a minor role in the reduction of the solar radiation. The measured RH values (not shown) were generally higher than 60%, with a maximum of 95% during the period. As a result, the high aerosol concentrations accompanied by high RH produced important effects on solar radiation. As shown in Fig. 6, the daytime averaged solar radiation was significantly reduced (about 40% reduction in Oct. 5-6 period compared with the value of Oct. 8).

2. Method

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491 In order to better understand the O₃ chemical production occurred in heavy aerosol
492 condition in eastern China, the possible O₃ production in such condition is discussed.

493 Ozone photochemical production (P[O₃]) is strongly related to the amount of OH
494 radicals (Chameides et al., 1999). According to the traditional theory, the amount of
495 surface OH radicals is proportional to the surface of solar radiation, which is
496 represented by

497

$$498 \quad [OH] = P[HOx]/L[HOx]^* \quad (R-1)$$

499

500 Where [OH] represents the concentration of hydroxyl radicals (#/cm³); HOx
501 represents the concentration of HO₂ + OH (#/cm³); P[HOx] represents the
502 photochemical production of HOx (#/cm³/s); and L[HOx]* (1/s) represents the
503 photochemical destruction of HOx, which is normalized by the concentrations of OH.

504

505 The major process for the photochemical production of P[HOx] is through the O₃
506 photolysis and follows by the reaction with atmospheric water vapor. It can express
507 by

$$508 \quad P[HOx] = J_1[O_3]/(k_1 \times am) \times 2.0 \times k_2[H_2O] = P_1[HOx] \quad (R-2)$$

509

510 Where J₁ represents the photolysis of O₃ + hν → O¹D; k₁ represents the reaction rate
511 of O¹D + am → O³P; and k₂ represents the reaction rate of O¹D + H₂O → 2OH. As
512 we can see, this HOx production is proportional to the magnitude of solar radiation
513 (J₁), and J₁ is the O₃ photolysis with the solar radiation. Figure 7 shows the
514 relationship between the values of J₁ and aerosol concentrations in October at
515 middle-latitude calculated by the TUV model (Madronich and Flocke, 1999). This
516 result suggests that under the high aerosol concentrations (AOD = 2.5), the J₁ value is
517 strongly depressed, resulting in significant reduction of OH concentrations and O₃
518 production. For example, the maximum J₁ value is about 2.7×10⁻⁵ (1/s) with lower
519 aerosol values (AOD = 0.25). According to the previous study, the surface PM_{2.5}

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524 concentrations were generally smaller than $50 \mu\text{g}/\text{m}^3$ with this AOD value (Tie et al.,
525 2017). However, when the AOD value increase to 2.5 (the $\text{PM}_{2.5}$ concentrations are
526 generally $>100 \mu\text{g}/\text{m}^3$), the maximum J_1 value rapidly decreases to about 6×10^{-6} (1/s),
527 which is about 450% reduction compared to the value with AOD=0.25. This study
528 suggests that under high $\text{PM}_{2.5}$ concentrations ($>100 \mu\text{g}/\text{m}^3$), the photochemical
529 production of OH (P[HOx]) is rapidly decreased, leading to low OH concentrations,
530 which cannot initiate the high oxidation of O_3 production. As a result, the high O_3
531 production shown in Fig. 4 cannot be explained. Other sources for O_3 oxidation are
532 needed to explain this result.

533

534 Recent studies show that the HONO concentrations are high in eastern China (Huang
535 et al., 2017). Because under high solar radiation, the photolysis rate of HONO is very
536 high, resulting in very low HONO concentrations in daytime (Seinfeld and Pandis,
537 2006). These measured high HONO concentrations are explained by their studies.
538 One of the explanations is that there are high surface HONO sources during daytime,
539 which produces high HONO concentrations (Huang et al., 2017). [Shi et al. \(2015\)](#)
540 [suggest that there are several potential HONO sources, including surface emissions,](#)
541 [conversion of \$\text{NO}_2\$ at the ocean surface, etc.](#) [Zhang et al. \(2016\)](#) parameterized these
542 [potential HONO sources in the WRF-Chem model, and the calculated HONO](#)
543 [concentrations are increased in the WRF-Chem model. In our calculation, we only use](#)
544 [the classical gas-phase chemistry to calculate HONO concentrations, and to illustrate](#)
545 [that the importance of these missing sources for the production of OH radicals.](#)
546 [Adding these missing sources \(there are not fully understand and remain a large](#)
547 [uncertainty\) could be a very important future work.](#)

548

549 Figure 8 shows the measured HONO concentrations in [three](#) large cities in China
550 (Shanghai, [Xi'an](#), [and Beijing](#)) during fall [and winter](#). It shows that the measured
551 HONO concentrations were high, with a maximum concentration of 2.3 ppbv during
552 morning, and about 0.5-1.0 ppbv in daytime. [As a result, we think that the high](#)
553 [HONO is a common event in large cities in eastern China, especially in daytime. This](#)

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high HONO is also measured by previous studies (Zhang et al. 2016; Huang et al. 2017). In this study, we make an assumption that the co-occurrence between O_3 and $PM_{2.5}$ occurred under high HONO concentrations. We note that using this assumption may result in some uncertainties in estimating the effect of HONO on OH. For example, using the measured HONO in Xi'an and Beijing could produce 1-2 times higher OH production by photolysis of HONO than the result by using the data from Shanghai. In this case, we use the measured HONO from Shanghai to avoid the over estimate of the HONO effect, which can be considered as a low-limit estimation.

It is also interesting to note that the high HONO concentrations were occurred during high aerosol concentration periods. Figure 9 illustrates that when the $PM_{2.5}$ concentrations increased to $70-80 \mu g/m^3$, and the HONO concentrations enhanced to 1.4-18 ppbv during September in Shanghai. This measured high HONO concentrations were significantly higher than the calculated concentrations (shown in Fig. 8), suggesting that some additional sources of HONO are needed. This result is consistent with the HONO measurements in other Chinese cities (Huang et al. 2017).

Under the high HONO concentrations in daytime, HONO can be photolyzed to be OH, and become another important process to produce OH. As a result, the OH production rate ($P[HOx]$) can be written to the following reactions.

$$P_2[HOx] = J_2 \times [HONO] \quad (R-3)$$

$$P[HOx] = P_1[HOx] + P_2[HOx]$$

$$= J_1[O_3]/(k_1 \times am) \times 2.0 \times k_2[H_2O] + J_2 \times [HONO] \quad (R-4)$$

Because the chemical lifetime of OH is less than second, OH concentrations can be calculated according to equilibrium of chemical production and chemical loss. With the both OH chemical production processes, the OH concentrations can be calculated by the following equation (Seinfeld and Pandis, 2006).

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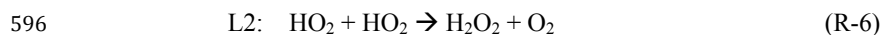
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$$P1 + P2 = L1 + L2$$

591

592 Where P1 and P2 are the major chemical productions, expressed in R-4, and L1 and
593 L2 are the major chemical loss of OH, and represent by

594



597

598 Under high NO_x condition, such as in the Shanghai region, NO_x concentrations were
599 often higher to 50 ppbv (shown in Fig. 3), the L1 term is larger than L2. The OH
600 concentrations can be approximately expressed by

601

$$[\text{HO}] = \{J_1[\text{O}_3]/(k_1 \times \text{am}) \times 2.0 \times k_2[\text{H}_2\text{O}] + J_2 \times [\text{HONO}]\} / k_3[\text{NO}_2] \quad (\text{R-5})$$

604

605 Where k_3 is the reaction coefficient of $\text{OH} + \text{NO}_2 \rightarrow \text{HNO}_3$.

606

607 3. Result and analysis

608

609 3.1. OH productions in different HONO conditions

610

611 In order to quantify the individual effects of these two OH production terms (P1 and
612 P2) on the OH concentrations, the P1 and P2 are calculated under different daytime
613 HONO conditions (calculated low HONO and measured high HONO concentrations).

614 Figure 10 shows that under the low HONO condition, the P1 is significantly higher
615 than P2, and P2 has only minor contribution to the OH values. For example, the
616 maximum of P1 occurred at 13 pm, with a value of $65 \times 10^6 \text{ \#}/\text{cm}^3/\text{s}$. In contrast, the
617 maximum of P2 occurred at 10 am, with a value of $15 \times 10^6 \text{ \#}/\text{cm}^3/\text{s}$. However, under
618 high HONO condition, the P2 plays very important roles for the OH production. The
619 maximum of P1 occurred at 11 am, with a value of $350 \times 10^6 \text{ \#}/\text{cm}^3/\text{s}$, which is about

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500% higher than the P1 value. It is important to note that this calculation is based on the high aerosol condition (AOD = 2.5) in September. This result can explain the high O₃ chemical production in Fig. 4.

3.2. OH in different aerosol conditions

In order to understand the effect of aerosol conditions, especially high aerosol conditions, on the OH concentrations. Figure 11 shows the OH concentrations with and without HONO production of OH. With including the HONO production (i.e., including P1 and P2), the calculated OH concentrations are significantly higher than without including this production (i.e., only including P1). The both calculated OH concentrations are rapidly changed with different levels of aerosol conditions. For example, without HONO production, the maximum OH concentration is about $7.5 \times 10^5 \text{ \#/cm}^3$ under low aerosol condition (AOD=0.25). In contrast, the maximum OH concentration rapidly reduced to $1.5 \times 10^5 \text{ \#/cm}^3$ under high aerosol condition (AOD=2.5), and further decreased to $1.0 \times 10^5 \text{ \#/cm}^3$ with the AOD value of 3.5. In contrast, with including HONO production, the OH concentrations significantly increased. Under higher aerosol condition (AOD=2.5), the maximum of OH concentration is about $7.5 \times 10^5 \text{ \#/cm}^3$, which is the same value under low aerosol condition in the no-HONO case. This result suggests that the measured high O₃ production occurred in the high aerosol condition is likely due to the high HONO concentrations in Shanghai.

3.3. Effects of clouds

Cloud cover can have very important impacts on the photolysis of HONO, which can affect the effect of HONO on the OH radicals. The above calculations are based on the cloud-free condition, with heavy aerosol concentration in the Beijing region. As shown in Fig. 5, during the case study period (Oct 5 to 6, 2015) (see Fig. 4), the weather map shows that the cloud-free condition, with heavy aerosol condition.

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In order to understand the effects of cloud on the photolysis of HONO, we include different cloud covers in the TUV model. The calculated results show in Fig. 12. The results show that the thin cloud (with cloud cover in 2 km and cloud water of 10 g/m³), could reduce the photolysis rate of HONO by about 40%, but the HONO could still remain important effects. However, with dense cloud condition (with cloud covers at 2 and 3 km and cloud water of 50 10 g/m³), the photolysis rate of HONO could reduce by 9-10 times by the cloud. In this case, adding photolysis rate of HONO cannot produce important effect on OH radicals and the production of O₃.

3.3. OH in winter

The measurement of O₃ also shows that the concentrations in winter were always low (see Fig. 2), suggesting that the O₃ concentrations were not significantly affected by the appearance of HONO. Figure 10 shows the OH concentrations in September and December. It shows that under different aerosol conditions, OH concentrations in December were very low compared with the values in September. Both the calculated OH concentrations include the HONO production term. For example, under the condition of AOD=2.5, the maximum OH is about 7.5×10⁵ #/cm³ in September, while it rapidly reduces to 1.5×10⁵ #/cm³ in December. Under the condition of AOD=3.5, the maximum OH is still maintaining to a relative high level (4.5×10⁵ #/cm³) in September. However, the maximum OH values are extremely low in December, with maximum value of 0.5×10⁵ #/cm³ in December. Because both the OH chemical productions (P1 and P2) are strongly dependent upon solar radiation (see equation R-4), the seasonal variation of solar radiation plays important roles for controlling the OH production in winter (see Fig. 13). Because the solar radiation is in a very low level in winter, adding the photolysis of HONO has smaller effect in winter than in fall, and OH remains low values by including the HONO production term.

Summary

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Currently, China is undergoing a rapid economic development, resulting in a high demand for energy, greater use of fossil fuels. As a result, the high emissions of pollutants produce heavy aerosol pollutions ($PM_{2.5}$) in eastern China, such as in the mega city of Beijing. The long-term measurements show that in addition to the heavy aerosol pollution, the O_3 pollution becomes another major pollutants in the Beijing region. The measured results show that there were very strong seasonal variation in the concentrations of both $PM_{2.5}$ and O_3 in the region. During winter, the seasonal variability of O_3 concentrations were anti-correlated with the $PM_{2.5}$ concentrations. However, during late spring and fall periods, the correlation between $PM_{2.5}$ and O_3 concentrations was positive compared to the negative in winter. This result suggests that during heavy aerosol condition (the solar radiation was depressed), the O_3 chemical production was still high, appearing a double peak of $PM_{2.5}$ and O_3 during fall period. This co-occurrence of high $PM_{2.5}$ and O_3 is the focus of this study. The results are highlighted as follows;

- (1) There are high daytime HONO concentrations in major Chinese mega cities, such as in Beijing and Shanghai. It is also interesting to note that the high HONO concentrations were occurred during high aerosol concentration periods. Under the high daytime HONO concentrations, HONO can be photo-dissociated to be OH radicals, and becomes an important process to produce OH.
- (2) With including the OH production of measured HONO concentrations, the calculated OH concentrations are significantly higher than without including this production. For example, without HONO production, the maximum OH concentration is about $7.5 \times 10^5 \text{ \#}/\text{cm}^3$ under low aerosol condition ($AOD=0.25$), and rapidly reduced to $1.5 \times 10^5 \text{ \#}/\text{cm}^3$ under high aerosol condition ($AOD=2.5$) in September. In contrast, by including HONO production, the OH concentrations significantly increased. For example, under higher aerosol condition ($AOD=2.5$), the maximum of OH concentration is about $7.5 \times 10^5 \text{ \#}/\text{cm}^3$, which is similar to the value under low aerosol condition in the no-HONO case. This result suggests that even under the high aerosol conditions, the chemical oxidizing process for O_3

719 production can be active. This result is likely for explaining the co-occurrence of
720 high PM_{2.5} and high O₃ in fall season in eastern China.

721 (3) The measurement of O₃ also shows that the concentrations in winter were always
722 low, suggesting that the O₃ concentrations were not significantly affected by the
723 appearance of HONO. The calculated result shows that the seasonal variation of
724 solar radiation plays important roles for controlling the OH production in winter.
725 Because the solar radiation is in a very low level in winter, adding the photolysis
726 of HONO has smaller effect in winter than in fall, and OH remains low values by
727 including the HONO production term. ▼

728 Because in recent years, the PM_{2.5} pollutions are reduced due to the large control
729 efforts by the Chinese government, the O₃ pollutions become another severe pollution
730 problem in eastern China. This study is important, because it provides some important
731 scientific highlights to better understand the O₃ pollutions in eastern China.

732

733 **Author contributions.** XT came up with the original idea of investigating the
734 scientific issue. XT and JX designed the analysis method. XL, GL and SZ provided
735 the observational data and helped in discussion. XT prepared the manuscript with
736 contributions from all co-authors.

737

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742 Environment, Chinese Academy of Sciences. ▼

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Figure Caption

Fig. 1. The geographic locations of the measurement sites in Beijing, in which the measured concentrations of PM_{2.5} and O₃ are used to the analysis.

Fig. 2. The daily averaged concentrations of PM_{2.5} and O₃ in the Beijing region in 2015. The concentrations are averaged over all sites shown in Fig. 1. The blue lines represent the PM_{2.5} concentrations (μg/m³), and the red bars represent the O₃ concentrations (μg/m³). The rectangles show some typical events during winter (green), spring and fall (orange), and summer (red).

Fig. 3. The correlation between O₃ and PM_{2.5} concentrations during winter (upper panel) and during late spring and fall (lower panel). During winter, O₃ concentrations were strong anti-correlated with the PM_{2.5} concentrations. During late spring and fall, O₃ concentrations were correlated with the PM_{2.5} concentrations.

Fig. 4. The diurnal variations of PM_{2.5} (blue line) and O₃ (red line), and NO₂ (green line) during a fall period (from Oct.5 to Oc. 6, 2015). It shows that with high PM_{2.5} condition, there was a strong O₃ diurnal variation.

Fig. 5. The cloud condition during the period of the case study (between Oct 5 and 6, 2015) in the Beijing region. The bright white color shows the cloud covers, and the grey white shows the haze covers. The Beijing region was under the heavy haze conditions during the period.

Fig. 6. The measured solar radiation (W/m²) from Oct. 3 to Oct. 9, 2015 in Beijing. The upper panel shows hourly values, and the lower panel shows the daytime averaged values.

Fig. 7. The effect of aerosol levels with AOD = 0.25 (black line), AOD = 2.5 (red line), AOD = 3.5 (blue line), and AOD = 4.0 (green line) on the O₃ photolysis calculated by the TUV model in October at middle-latitude.

Fig. 8. The measured HONO concentrations (ppbv) in three large cities in China. The red line was measured in Xi'An from 24 July to August 6, 2015. The blue line was measured in Shanghai from 9 to 18 September, 2009. The dark-red line was measured in Beijing from 1 to 27 January, 2014. The green line is calculated by the WRF-Chem model. The measurement in fall of Shanghai is applied to the calculation for the OH production of HONO.

Fig. 9. The measured HONO (upper panel) and PM_{2.5} concentrations (lower panel) in fall in Shanghai. It illustrates that the high HONO concentrations were corresponded with high PM_{2.5} concentrations.

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900 | **Fig. 10.** The calculated OH production $P(\text{HOx})$ ($\#/\text{cm}^3/\text{s}$) by using the model
901 | calculated HONO (low concentrations) (in the upper panel) and by using the
902 | measured HONO (high concentrations) (in the lower panel). The red bars represent
903 | the calculation of the P1 term, and the red bars represent the calculation of the P2
904 | term (OH production from HONO).

905

906 | **Fig. 11.** The calculated OH concentrations ($\#/\text{cm}^3$) with (upper panel) and without
907 | (lower panel) HONO production of OH, under different aerosol levels. Dark red
908 | ($\text{AOD}=0.25$), red ($\text{AOD}=2.5$), red ($\text{AOD}=3.5$), and red ($\text{AOD}=4.0$).

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910 | 2015 in the Beijing region. The bright white color shows the cloud covers, and the
911 | grey white shows the haze covers. The Beijing region is under the heavy haze
912 | conditions during the period.

913

914 | **Fig. 12.** The effect of cloud cover on the photolysis rate of HONO ($J[\text{HONO}]$). The
915 | blue, red, and green lines represent the cloud water vapor of 0 (cloud-free), 10 (g/m^3 –
916 | thin cloud), and 50 (g/m^3 – thick cloud), respectively. The left panel (A) represents
917 | the light aerosol condition, with AOD of 0.25, and the right panel (B) represents the
918 | heavy aerosol condition, with AOD of 2.5.

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920 | **Fig. 13.** The calculated OH concentrations in September (blue bars) and December
921 | (dark red bars), under different aerosol levels.

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Figures

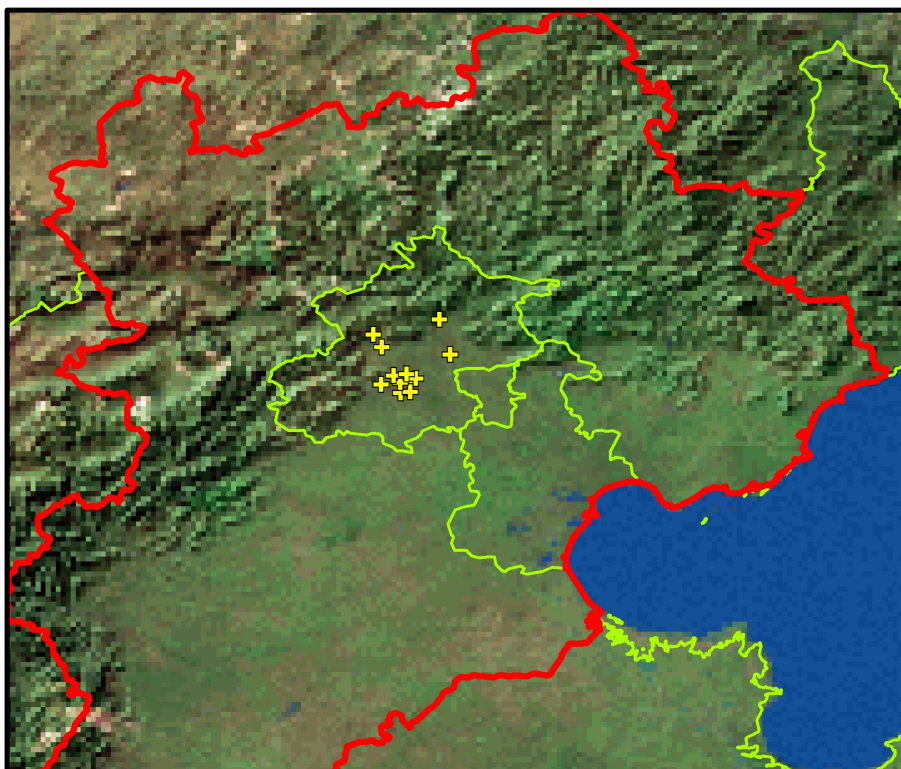


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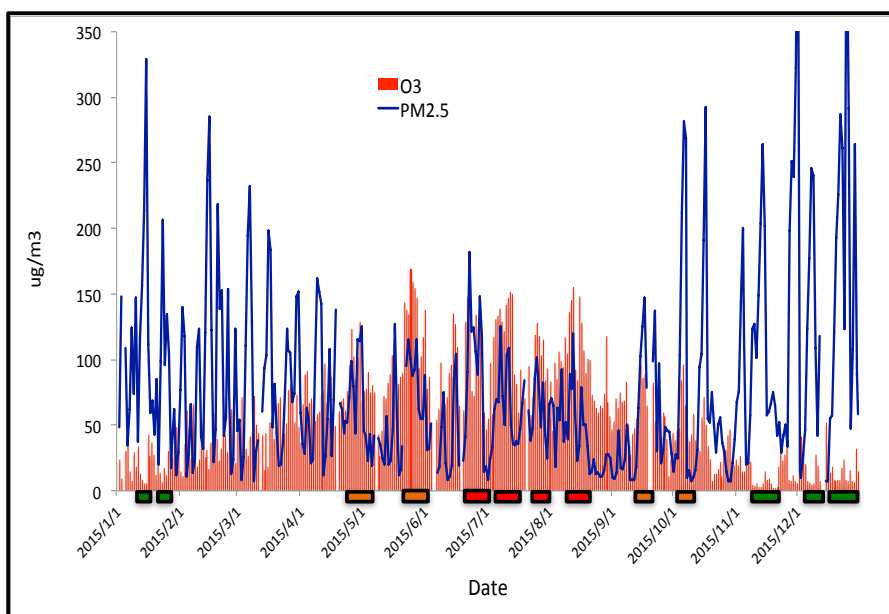


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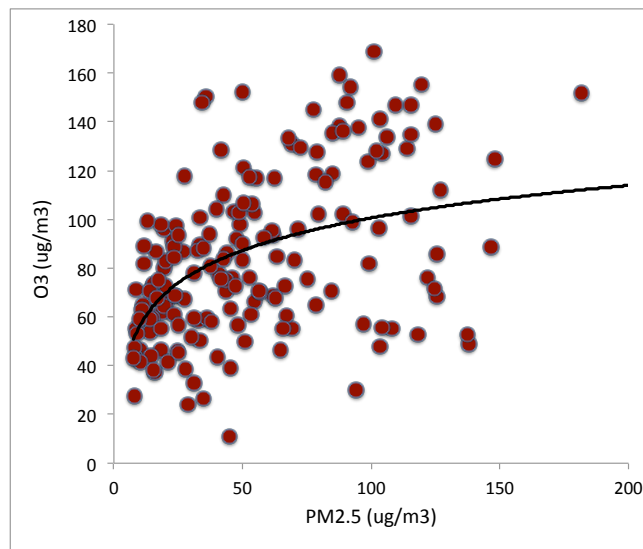
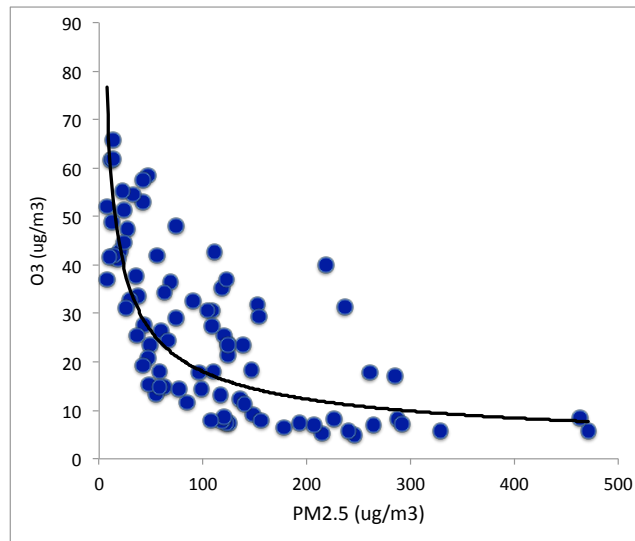


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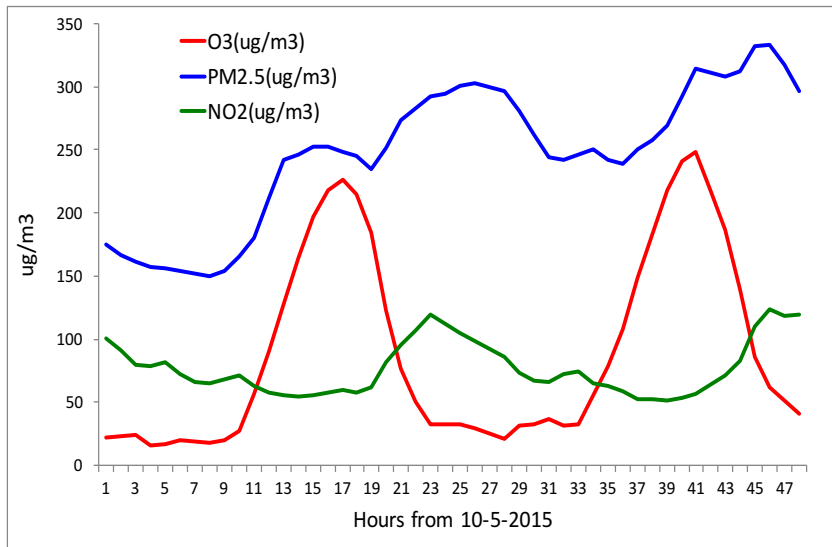


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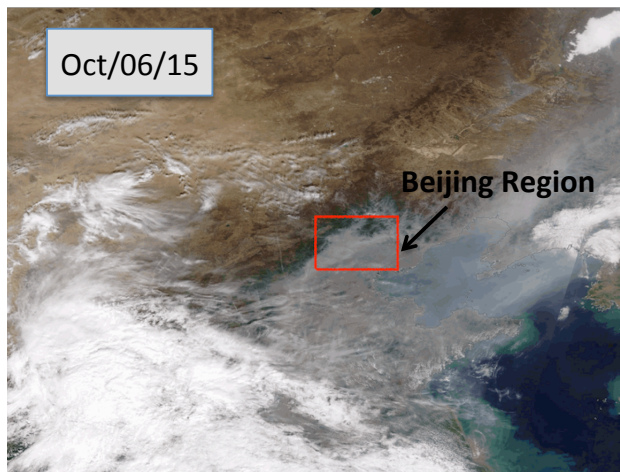
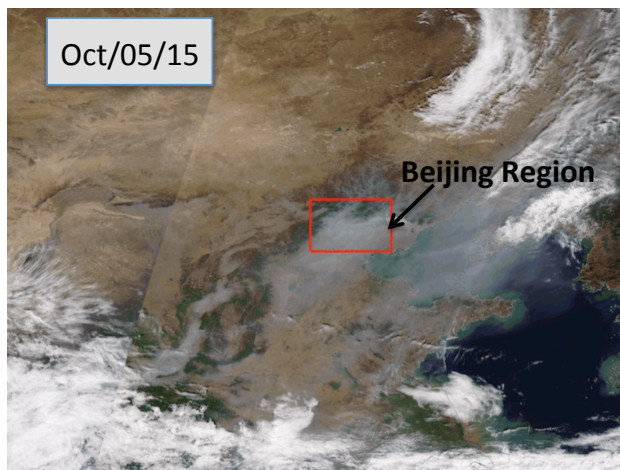


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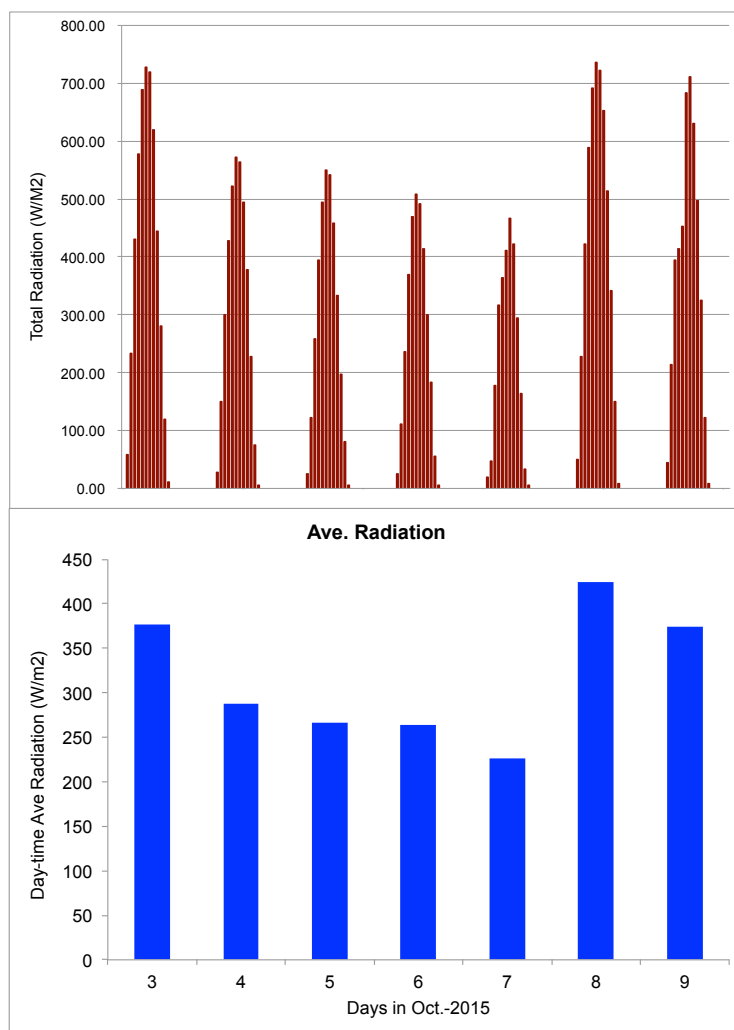


Fig. 6. The measured solar radiation (W/m^2) from Oct. 3 to Oct. 9, 2015 in Beijing. The upper panel shows hourly values, and the lower panel shows the daytime averaged values.

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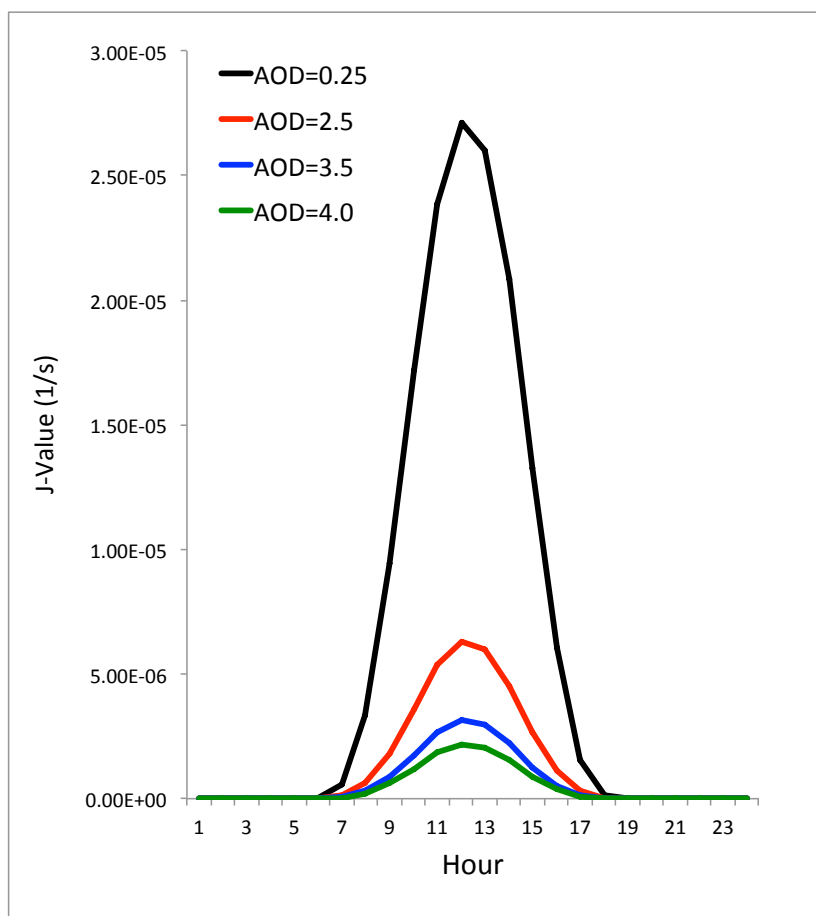


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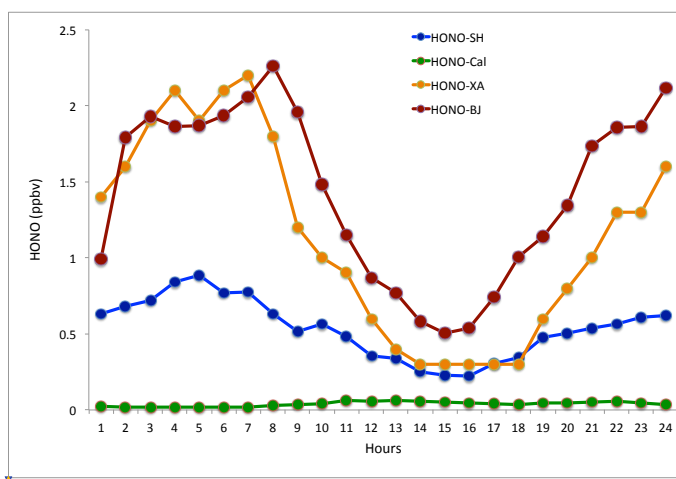
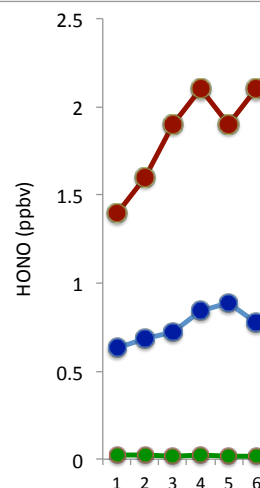


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已删除: The measured HONO concentrations (ppbv) in two large cities in China. The red line was measured in Xi'An from 24 July to August 6, 2015. The blue line was measure in Shanghai from 9 to 18 September, 2009. The green line is calculated by the WRF-Chem model. The measurement in fall of Shanghai is applied to the calculation for the OH production of HONO.

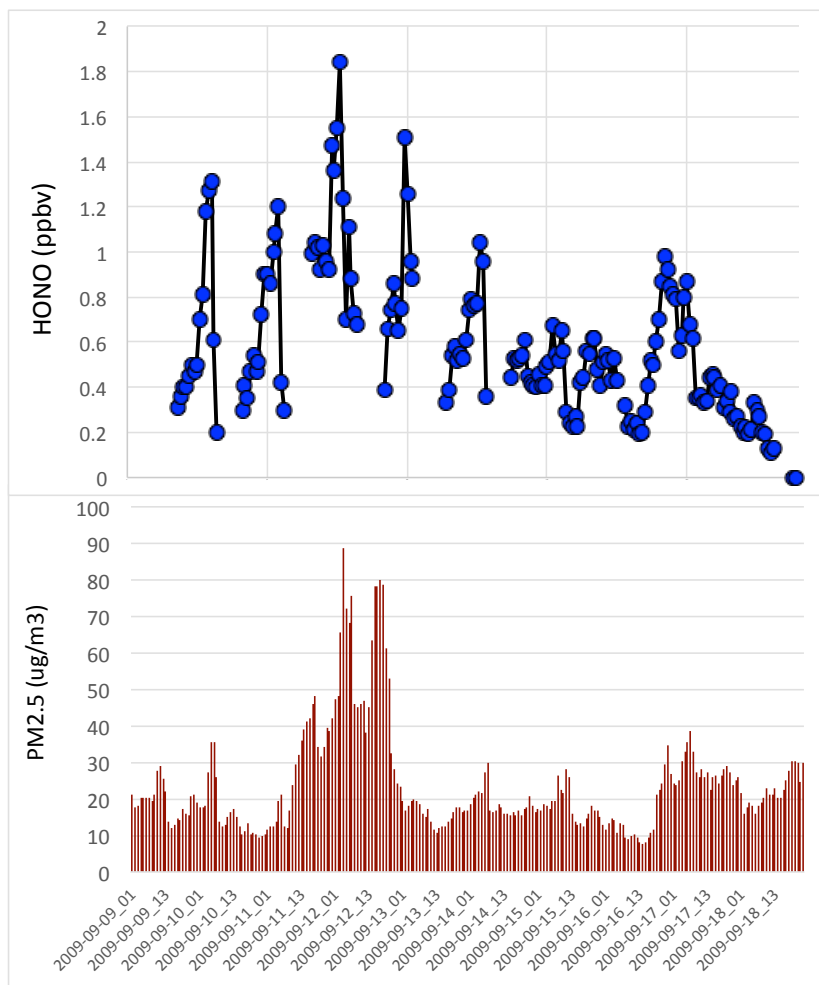


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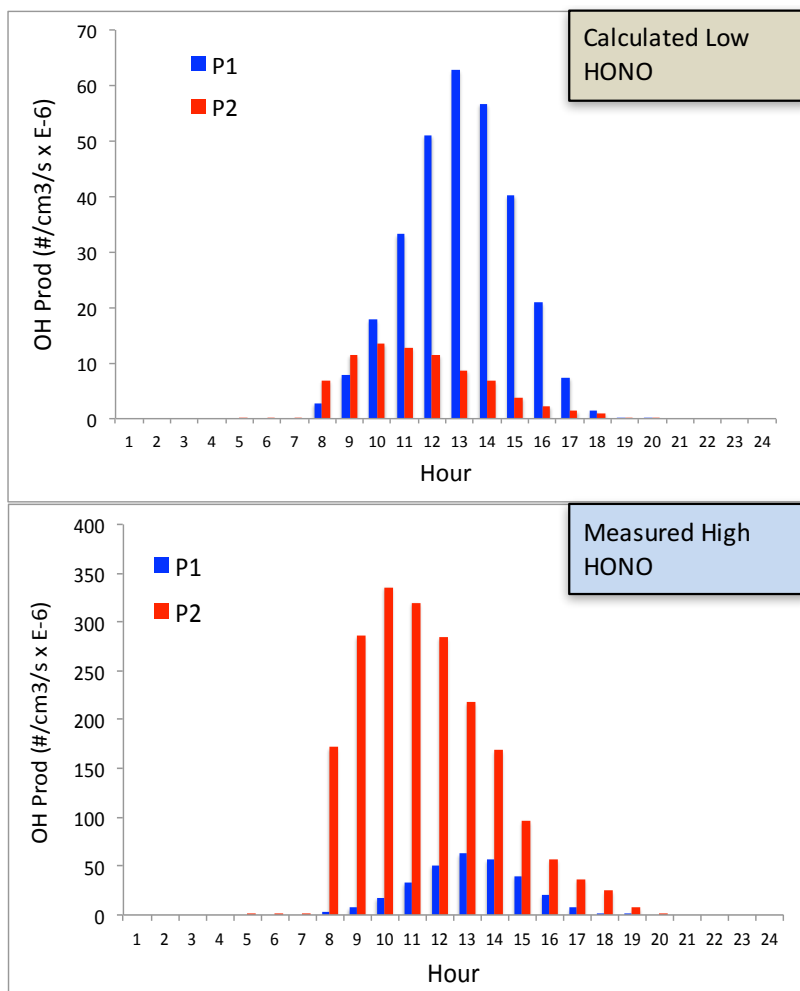


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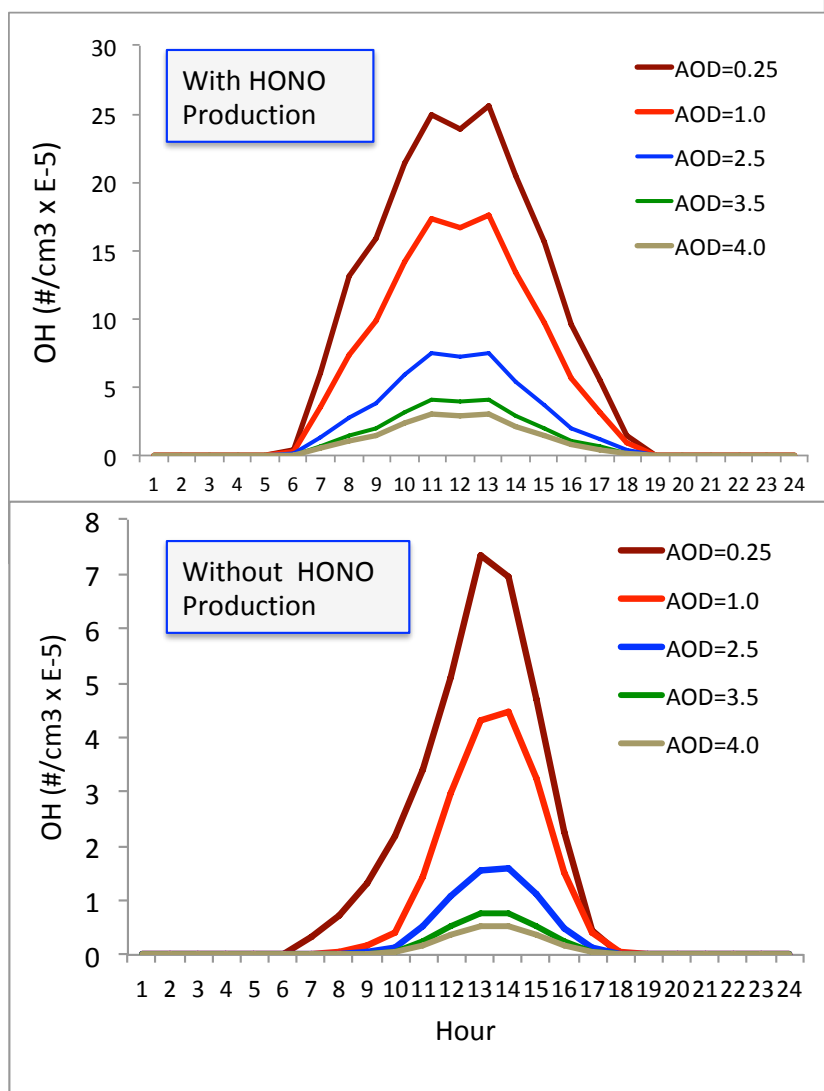


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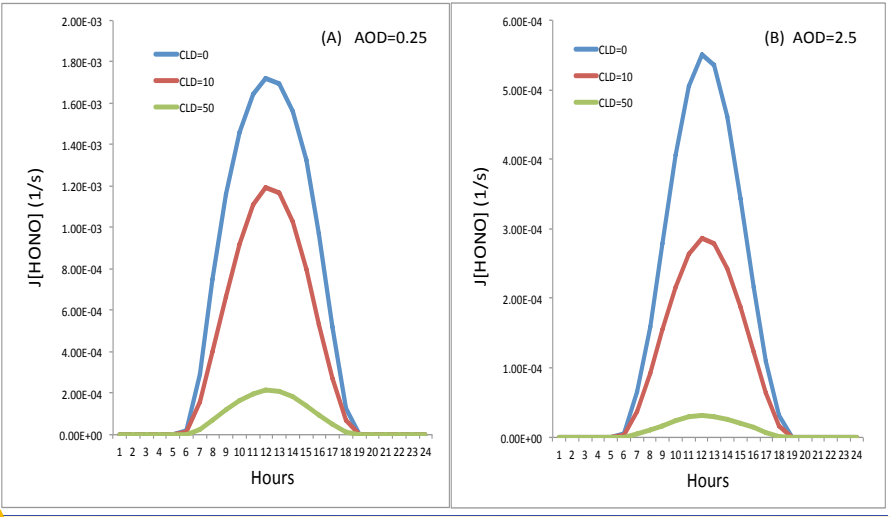


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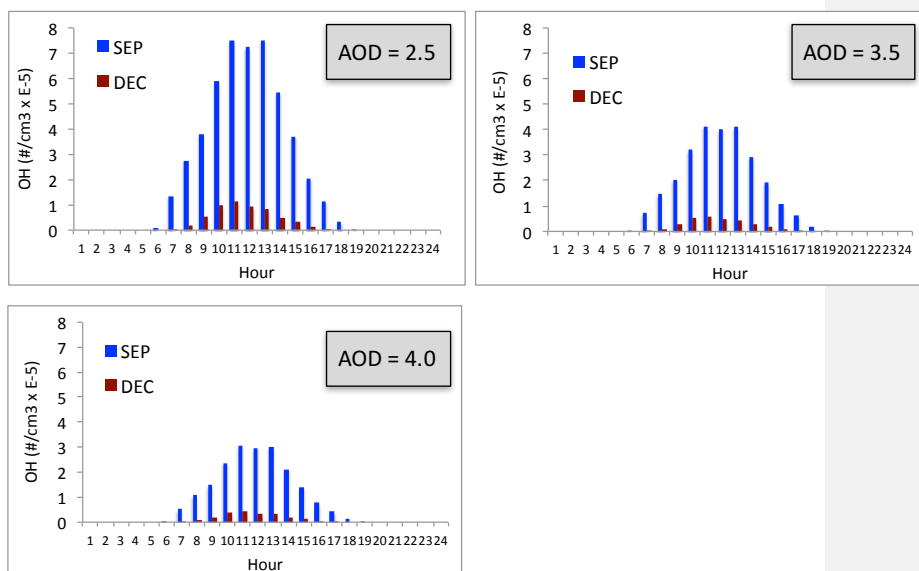


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