Size-resolved Composition and Morphology of Particulate Matter During the Southwest Monsoon in Metro Manila,

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21 Abstract

22 This paper presents novel results from size-resolved particulate matter (PM) mass, composition, 23 and morphology measurements conducted during the 2018 Southwest Monsoon (SWM) season in 24 Metro Manila, Philippines. Micro-Orifice Uniform Deposit Impactors (MOUDIs) were used to 25 collect PM sample sets composed of size-resolved measurements at the following aerodynamic 26 cutpoint diameters (D_p): 18, 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10, 0.056 µm. Each sample 27 set was analyzed for composition of the water-soluble fraction. Analysis for mass were done on 28 two sample sets while black carbon (BC) and morphology analysis were done on a single sample 29 set. The bulk of the PM mass was between 0.18–1.0 µm with a dominant mode between 0.32– 30 $0.56 \,\mu\text{m}$. Similarly, most of the black carbon (BC) mass was found between $0.10-1.0 \,\mu\text{m}$, peaking 31 between 0.18–0.32 µm. These peaks are located in the Greenfield Gap or the size range between 32 0.10–1.0 μ m, where wet scavenging by rain is relatively inefficient. In the range of 0.10 – 0.18 33 µm, BC constituted 78.1% of the measured mass. Comparable contributions of BC (26.9%) and 34 the water-soluble fraction (31.3%) to total PM were observed and most of the unresolved mass, 35 which in total amounted to 41.8%, was for diameters exceeding 0.32 µm. The water-soluble ions and elements exhibited an average combined concentration of 8.53 μ g m⁻³, with SO₄²⁻, NH₄⁺, NO₃⁻ 36 37 , Na⁺, and Cl⁻ as the major contributors. Positive Matrix Factorization (PMF) was applied to 38 identify the possible aerosol sources and estimate their contribution to the water-soluble fraction 39 of collected PM. The factor with the highest contribution was attributed to "Aged Aerosol" 40 (48.0%) while "Sea Salt" (22.5%) and "Combustion" emissions (18.7%) had comparable 41 contributions. "Vehicular/Resuspended Dust" (5.6%) as well as "Waste Processing" emissions 42 (5.1%) were also identified. Microscopy analysis highlighted the ubiquity of non-spherical 43 particles regardless of size, which is significant when considering calculations of parameters such 44 as single scattering albedo, asymmetry parameter, and extinction efficiency.

The significant influence from Aged aerosol to Metro Manila during the SWM season indicates that local sources in this megacity do not fully govern this coastal area's aerosol properties. That the majority of the regional aerosol mass burden is accounted for by BC and other insoluble components has important downstream effects on the aerosol hygroscopic properties, which depend on composition. The results are relevant for understanding the impacts of monsoonal features on size-resolved aerosol properties, notably aqueous processing and wet scavenging. Finally, the results of this work provide contextual data for future sampling campaigns in Southeast

- 52 Asia such as the airborne component of the Cloud, Aerosol, and Monsoon Processes Philippines
- 53 Experiment (CAMP²Ex) planned for the SWM season in 2019.

55 1. Introduction

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Ambient atmospheric aerosol particles impact human health, visibility, climate, and the 57 58 hydrological cycle. Major factors governing these behaviors, such as deposition fraction in the 59 respiratory system and activation into cloud condensation nuclei (CCN), include size and chemical 60 composition. Therefore, size-resolved measurements of ambient aerosol particles can lend 61 additional insights to the behavior and implications of particulate matter (PM) in the atmosphere. 62 One region of interest for characterization of PM is Southeast Asia due to increasing urbanization 63 and the exposure of the population to a variety of aerosol sources, both natural and anthropogenic 64 (Hopke et al., 2008). However, use of space-borne remote-sensing instrumentation presents a 65 challenge for characterization of aerosol in this region, due to issues such as varying terrain and 66 cloud cover (Reid et al., 2013).

67 The Philippines represents a country in Southeast Asia with a developing economy, rapid 68 urbanization, old vehicular technology, and less stringent air quality regulations (e.g., Alas et al., 69 2017). It is also highly sensitive to the effects of climate change including prolonged dry periods 70 and reductions in southwest monsoon (SWM) rainfall in recent decades (e.g., Cruz et al., 2013). 71 Metro Manila is the country's capital and center of political and economic activities. Also referred 72 to as the National Capital Region, Metro Manila is composed of 16 cities and a municipality that 73 collectively occupy a land area of $\sim 619 \text{ km}^2$. As of 2015, Metro Manila had a population of 74 approximately 12.88 million (Philippine Statistics Authority, 2015). Of the cities comprising the 75 Metro Manila area, the one that is the focus of this study, Ouezon City, is the most populated (2.94) million people) with a population density of ~17,000 km⁻² as of 2015 (Philippine Statistics 76 77 Authority, 2015).

78 The rainfall pattern in Southeast Asia is governed by topographic effects and the prevailing 79 surface winds brought by the monsoons. Mountain ranges in the Philippines are generally oriented 80 north to south in the eastern and western coasts. As such, northeasterly winds during the East Asian winter monsoon that starts in November brings wetness (dryness) on the eastern (western) coasts 81 82 of the country. In contrast, the rainy season starts in May when the Western North Pacific 83 subtropical high moves northeast and the Asian summer monsoon enables the propagation of 84 southwesterly wind through the Philippines (Villafuerte et al., 2014). Metro Manila, located on 85 the western side of the Philippines, therefore experiences wet (May-October) and dry (November-April) seasons. The large seasonal shift in prevailing wind directions can cause changes in the 86

87 source locations of aerosol transported to the Philippines and the subsequent direction in which 88 emissions from the Philippines are transported, such as to the northwest (e.g., Chuang et al., 2013) 89 or southwest (e.g., Farren et al., 2019). However, one interesting feature of Metro Manila is the consistency of $PM_{2.5}/PM_{10}$ mass concentrations during both the dry (44/54 µg m⁻³) and wet seasons 90 (43/55 µg m⁻³) (Kim Oanh et al., 2006), which stands in contrast to typical assumptions that 91 92 increased wet scavenging during rainy seasons would lead to decreases in measured PM (e.g., Liao 93 et al., 2006). While similar results are observed in Chennai, India, this behavior is different than 94 other cities in Asia, including Bandung City (Indonesia), Bangkok (Thailand), Beijing (China), 95 and Hanoi City (Vietnam), which exhibit reduced PM_{2.5} levels during the wet season as compared 96 to the dry season (Kim Oanh et al., 2006). While the total PM levels may stay constant across the 97 wet and dry seasons, seasonally-resolved analyses will provide additional insights into how the 98 composition, morphology, and sources (transported vs. local emissions) change on a seasonal 99 basis.

100 Metro Manila has been drawing growing interest for PM research owing to the significant 101 levels of black carbon (BC). A large fraction of PM in Metro Manila can be attributed to BC (e.g., ~50% of $PM_{2.5}$; Kim Oanh et al., 2006), with previously measured average values of BC at the 102 Manila Observatory (MO) reaching ~10 μ g m⁻³ for PM_{2.5} (Simpas et al., 2014). The impacts of the 103 104 high levels of BC present on human health have also received attention (Kecorius et al., 2019). 105 Identified major sources of BC include vehicular, industrial, and cooking emissions (Bautista et 106 al., 2014; Kecorius et al., 2017). Vehicular emissions, especially along roadways where personal 107 cars and motorcycles, commercial trucks, and motorized public transportation, including powered 108 tricycles and *jeepneys*, are plentiful. For instance, measurements of PM_{2.5} at the National Printing 109 Office (NPO) located alongside the major thoroughfare Epifanio de los Santos Avenue (EDSA) were on average 72 μ g m⁻³; this value is twice the average concentration at MO, an urban mixed 110 111 site located approximately 5 km from NPO (Simpas et al., 2014). In addition to local emissions, 112 long-range transport of pollution, such as biomass burning, can also impact the study region (e.g., 113 Xian et al., 2013; Reid et al., 2016a/b). However, most past work referenced above has focused on 114 either total PM_{2.5} or PM₁₀ composition, and therefore, detailed size-resolved composition 115 information has been lacking in this region. Like other monsoonal regions (Crosbie et al., 2015; 116 Qu et al., 2015), it is of interest for instance to know if products of aqueous processing (e.g., sulfate, 117 organic acids) during the monsoonal period, promoted by the high humidity, become more

118 prominent in certain size ranges to ultimately enhance hygroscopicity, which is otherwise 119 suppressed with higher BC influence.

120 A year-long sampling campaign (Cloud, Aerosol, and Monsoon Processes Philippines 121 Experiment (CAMP²Ex) weatHEr and CompoSition Monitoring (CHECSM) study) was 122 established in July 2018 to collect size-resolved aerosol measurements in Metro Manila. The aim 123 of this study is to report size-resolved PM measurements taken over the course of the SWM (July-124 October) of 2018 in Quezon City, Metro Manila, Philippines as part of CHECSM. The results of 125 this study are important for the following reasons: (i) they provide size-resolved analysis of BC in 126 an area previously characterized as having one of the highest BC mass percentages in the whole 127 world; (ii) they provide a basis for better understanding the unusual phenomenon of having similar 128 PM levels during a wet and dry season; (iii) they provide contextual data for contrasting with both 129 other coastal megacities and also other monsoonal regions; and (iv) they can lend insights into the 130 characteristics of aerosol transported both into and out of Metro Manila and how important local 131 sources are in Metro Manila relative to transported pollution.

132 Outcomes of this study include (i) the first size-resolved characterization of both aerosol 133 composition and morphology in Metro Manila for the SWM, with implications in terms of PM 134 effects on climate, visibility, the hydrological cycle, and public health owing to the dependence of 135 these impacts on particle size; (ii) archival data that contributes to the timeline of aerosol research 136 in Metro Manila, and more broadly Southeast Asia, where there is considerable concern over air 137 pollution; and (iii) baseline data for aerosol composition to be used to inform and assist research 138 to be conducted during future field campaigns in Southeast Asia including the same seasonal 139 period (i.e., SWM) in 2019 as part of CAMP²Ex, which will involve both surface and airborne 140 measurements.

141 **2. Experimental Methods**

142 **2.1 Sample Site**

Sampling was performed at MO in Quezon City, Philippines (14.64° N, 121.08° E). Two MOUDIs were placed inside an unoccupied room on the 3rd floor of the MO administration building (~87 m above sea level). The inlet, located just outside the window, consists of a 2 m long stainless steel tube and a reducer that is connected directly to the MOUDI inlet. Figure 1 visually shows the sampling location and potential surrounding aerosol sources. Past work focused on PM_{2.5} suggested that the study location is impacted locally mostly by traffic, various forms of industrial activity, meat cooking from local eateries, and, based on the season, biomass burning (Cohen et al., 2009). This is consistent with another source apportionment study which reported that potential sources in six sites across Metro Manila include traffic, secondary particles, and biomass burning (Kim Oanh et al., 2013).

153 Meteorological data were collected using a Davis Vantage Pro 2 Plus weather station 154 located on the roof (~90 m above sea level, ~15 m above ground level) above where the MOUDIs 155 were located. Except for precipitation, which is reported here as accumulated rainfall, reported 156 values for each meteorological parameter represent averages for the sampling duration of each 157 aerosol measurement. The mean temperature during the periods of MOUDI sample collection 158 ranged from 24.9 to 28.1° C, with accumulated rainfall ranging widely from no rain to up to 78.4 159 mm. To identify sources impacting PM via long-range transport to the Metro Manila region, Figure 160 1a summarizes the five-day back-trajectories for air masses arriving at MO on the days when 161 samples were being collected, calculated using the NOAA Hybrid Single-Particle Lagrangian 162 Integrated Trajectory (HYSPLIT) model (Stein et al., 2015; Rolph, 2016). Trajectory calculations 163 were started at 00, 06, 12, and 18 hours in MO using a model run height of 12 m above ground 164 level and meteorological files from the NCEP/NCAR Reanalysis dataset. Trajectory cluster 165 analysis was conducted using TrajStat (Wang et al., 2009). The back-trajectories in Figure 1a show 166 that indeed 66% of the wind came from the southwest during the sampling periods.

167 2.2 MOUDI Sample Sets

168 PM was collected on Teflon substrates (PTFE membrane, 2 µm pore, 46.2 mm, Whatman) 169 in Micro-Orifice Uniform Deposit Impactors (MOUDI, MSP Corporation, Marple et al., 2014). 170 Size-resolved measurements were taken at the following aerodynamic cutpoint diameters (D_p) : 18, 171 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10, 0.056 µm. Fourteen sample sets were collected during 172 the SWM season (July-October 2018), with details about the operational and meteorological 173 conditions during each sample set shown in Table 1. To determine the optimum sampling time that 174 will collect enough sample for subsequent analyses, collection time for the first four samples 175 ranged from 24 to 119 hours. Subsequent sampling were then fixed to 48 hours with one sample 176 set collected every week. The sampling collection was designed to include samples from each day 177 of the week so the collection cycled between Monday - Wednesday, Tuesday - Thursday,

Wednesday – Friday, and Saturday – Monday, starting at 1400 (local time) for the weekday samples and 0500 for the weekend samples. The Teflon substrates were pretreated by washing with deionized water and air drying in a covered box. Substrates were placed and retrieved from the cascade impactor inside the laboratory in an adjacent building and transported to and from the sampling site using an impactor holder (Csavina et al., 2011). Samples are immediately placed in the freezer upon retrieval.

184 On two occasions, two pairs of MOUDI sets (Sets MO3/MO4 and MO13/MO14) were 185 collected simultaneously such that one set in each pair could undergo different types of analyses. 186 Sets 3 and 13 underwent gravimetric analysis using a Sartorius ME5-F microbalance. Substrates 187 were conditioned for at least 24 h at a mean temperature of 20-23 °C and a mean relative humidity 188 of 30-40% before pre- and post-weighing (U.S. Environmental Protection Agency, 2016). 189 MOUDI set 13 was additionally examined with a Multi-wavelength Absorption Black Carbon 190 Instrument (MABI; Australian Nuclear Science and Technology Organisation). This optically-191 based instrument quantifies absorption and mass concentrations at seven wavelengths between 405 192 and 1050 nm; however, results are reported only for 870 nm to be consistent with other studies, as 193 BC is the predominant absorber at that wavelength (e.g., Ramachandran and Rajesh, 2007; Ran et 194 al., 2016). One additional sample set for microscopy analysis was collected for one hour on August 195 1 using aluminum substrates.

196 **2.3 Chemical Composition Analysis**

197 Twelve sample sets, composed of 11 samples each, were analyzed for water-soluble ions 198 and elements (Table 2). In order to preserve samples for additional analysis, each Teflon substrate 199 was cut in half. A half of each substrate was extracted in 8 mL of Milli-Q water (18.2 M Ω -cm) 200 through sonication for 30 min in a sealed polypropylene vial. A blank substrate was processed in 201 the same method to serve as a background control sample. Subsequent chemical analysis of the 202 water-soluble components in the aqueous extracts were performed using ion chromatography (IC; 203 Thermo Scientific Dionex ICS - 2100 system) for the following species: cations = Na^+ , NH_4^+ , 204 Mg^{2+} , Ca^{2+} , dimethylamine (DMA), trimethylamine (TMA), diethylamine (DEA); anions =, 205 methanesulfonate (MSA), pyruvate, adipate, succinate, maleate, oxalate, phthalate, Cl⁻, NO₃⁻, 206 SO₄²⁻. Owing to co-elution of TMA and DEA in the IC system, a cumulative sum of the two is 207 reported here, which represents an underestimate of their total mass concentration owing to overlap in parts of their peaks. Limits of detection (LOD) were calculated for each species based on their
respective calibration curve (Table S1), with LOD being three times the standard deviation of the
residuals (predicted signal minus measured signal) divided by the slope of the calibration curve
(Miller and Miller, 2018).

The aqueous extracts were simultaneously characterized for elemental composition using triple quadrupole inductively coupled plasma mass spectrometry (ICP-QQQ; Agilent 8800 Series) for the following species: K, Al, Fe, Mn, Ti, Ba, Zn, Cu, V, Ni, P, Cr, Co, As, Se, Rb, Sr, Y, Zr, Nb, Mo, Ag, Cd, Sn, Cs, Hf, Tl, Pb. Limits of detection of the examined elements were calculated automatically by the ICP-QQQ instrument and were in the ppt range (Table S1). The sample concentrations represent an average of three separate measurements with a standard deviation of 3% or less.

Note that some species were detected by both IC and ICP-QQQ (i.e., Na⁺, K⁺, Mg²⁺, Ca²⁺), and that the IC concentrations are used here for all repeated species with the exception of K⁺ owing to better data quality from ICP-QQQ. All IC and ICP-QQQ species concentrations for samples have been corrected by subtracting concentrations from background control samples. For more examples of the application of these methods used for substrate collection and IC/ICP analysis, the reader is referred to other recent work (Braun et al., 2017; Ma et al., 2019; Schlosser et al., 2017).

226 2.4 Microscopy Analysis

227 As already noted, one MOUDI set on August 1 was devoted to microscopy analysis. 228 Morphology and additional elemental composition analysis was carried out on this set of aluminum 229 substrates using scanning electron microscopy equipped with energy dispersive X-ray 230 spectroscopy (SEM-EDX) in the Kuiper Imaging cores at the University of Arizona. Secondary 231 electron (SE) imaging and EDX elemental analysis were performed using a Hitachi S-4800 high 232 resolution SEM coupled to a Noran system Six X-ray Microanalysis System by Thermo Fisher 233 Scientific. EDX analysis on individual particles was performed with 30 kV accelerating voltage to 234 obtain weight percentages of individual elements. SEM-EDX results showed that the background 235 control aluminum substrate was dominated by Al (88.27%), with minor contributions from Ag 236 (5.34%), C (4.87%), O (0.79%), Fe (0.67%), and Co (0.05%). Such contributions were manually subtracted from spectra of individual particles on sample substrates, with the remaining elements 237

scaled up to hundred percent. Image processing was conducted with Image J software to measure
particle dimensions and adjust the contrast and brightness of images to provide better visualization.

240 **2.5 Computational Analysis**

241 This study reports basic descriptive statistics for chemical concentrations and correlations 242 between different variables. Statistical significance hereafter corresponds to 95% significance 243 based on a two-tailed Student's t-test. To complement correlative analysis for identifying sources 244 of species, positive matrix factorization (PMF) modeling was carried out using the United States 245 Environmental Protection Agency's (US EPA) PMF version 5. A total of 132 samples from the 12 246 sets analyzed for water-soluble ions and elements were used in the PMF analysis. Species 247 concentrations were examined before being inputted to PMF. Species considered as "strong" based on high signal-to-noise ratios (S/N > 1) and those with at least 50% of the concentrations above 248 249 the LOD were used in the PMF modeling (Norris et al., 2014). This resulted in a 132 (samples) × 250 30 (species) data matrix that was inputted to PMF. Data points with concentrations exceeding the 251 LOD had uncertainty quantified as:

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$$253 \qquad \sigma_{ij} = 0.05 \cdot X_{ij} + LOD_{ij},$$

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255 where $\sigma_{i,i}$, X_{ij} , and LOD_{ij} are the uncertainty, concentration, and LOD, respectively, of the j^{th} species in the i^{th} sample (Reff et al., 2007). When concentration data were not available for a 256 257 particular stage of a MOUDI set for a species, the geometric mean of the concentrations for that 258 MOUDI stage and species was applied with uncertainty counted as four times the geometric mean 259 value (Polissar et al., 1998; Huang et al., 1999). A 25% extra modeling uncertainty was applied 260 to account for other sources of errors, such as changes in the source profiles and chemical 261 transformations (Dumanoglu et al., 2014; Norris et al., 2014). The model was run 20 times with a 262 randomly chosen starting point for each run.

(Equation 1)

263 **3. Results**

264 **3.1 Total Mass Concentrations and Charge Balance**

265 The average total mass concentration (\pm standard deviation) of water-soluble species across 266 all MOUDI stages (Table 1) during the study period was $8.53 \pm 4.48 \ \mu g \ m^{-3}$ (range = 2.7–16.6 μg 267 m^{-3}). The species contributing the most to the total water-soluble mass concentration during the SWM included SO_4^{2-} (44% ± 6%), NH₄⁺ (18% ± 5%), NO₃⁻ (10 ± 3%), Na⁺ (8 ± 3%), and Cl⁻ (6% 268 269 \pm 3%). The meteorological parameters from Table 1 best correlated to total water-soluble mass 270 concentrations were temperature (r = 0.64) and rainfall (r = -0.49). The highest total mass concentration (set MO13/14 = 16.6 μ g m⁻³) occurred during the period with one of the highest 271 272 average temperatures (27.8 °C) and second least total rainfall (0.8 mm). Other sampling periods 273 with high mass concentrations (sets MO7, MO8, and MO12) coincided with the highest 274 temperature and lowest rainfall observations. High temperatures, and thus more incident solar 275 radiation, presumably enhanced production of secondary aerosol species via photochemical 276 reactions as has also been observed in other regions for their respective monsoon season (Youn et 277 al., 2013).

278 Low rainfall is thought to have been coincident with reduced wet scavenging of aerosol at 279 the study site as has been demonstrated for other regions such as North America (Tai et al., 2010) 280 and megacities such as Tehran (Crosbie et al., 2014). However, set MO11 exhibited a very low 281 concentration even with high temperature and lack of rainfall, which may be due to changes in the 282 source and transport of aerosol since this sample set coincided with a significant change in average wind direction (290.2° for MO11 vs. $90.1^{\circ} - 127.5^{\circ}$ for all other MOUDI sets). While the reported 283 284 rainfall measurements were taken at MO, inhomogeneous rainfall patterns in the regions 285 surrounding the Philippines could also contribute to the wet scavenging of PM, thereby lowering 286 the quantity of transported particles reaching the sample site. Future work will address the 287 influence of spatiotemporal patterns of precipitation on PM loadings in the Philippines as a point 288 measurement at an aerosol observing site may be misleading.

289 On two occasions, two simultaneous MOUDI sets (Sets MO3/MO4 and MO13/MO14) 290 were collected for the potential to compare different properties that require separate substrates. 291 The total mass concentrations based on gravimetric analysis of sets MO3 and MO13 were 18.6 µg m⁻³ and 53.0 µg m⁻³, respectively (Figure 2). Both sets exhibited a dominant concentration mode 292 293 between 0.32–0.56 µm and the MO3 set was different in that it exhibited bimodal behavior with a 294 second peak between 1.8–3.2 µm. The sum of speciated water-soluble species accounted for only 295 27.8% and 31.3% of the total gravimetric mass of sets MO3 and MO13, respectively, indicative 296 of significant amounts of water-insoluble species undetected by IC and ICP-QQQ. When adding the total mass of BC (14.3 μ g m⁻³) to the other resolved species from set MO13 (the one time BC 297

298 was measured), there was still 22.1 μ g m⁻³ of unresolved mass (41.8% of total PM). Most of the 299 unaccounted mass was for D_p > 0.32 μ m.

The observation of BC accounting for 26.9% of total PM (14.3 μ g m⁻³) is consistent with 300 301 past work highlighting the significant fraction of BC in the ambient aerosol of Manila (Kim Oanh 302 et al., 2006; Bautista et al., 2014; Simpas et al., 2014; Kecorius et al., 2017). However, this fraction 303 of BC is very high compared to measurements during the monsoon season in other parts of the 304 world. The mass fraction of BC in total suspended PM (TSPM) was 1.6%/2.2% for the monsoon 305 season in 2013/2014 in Kadapa in southern India, even though the TSPM measured was 306 comparable to that in Manila (64.9 and 49.9 µg m⁻³, for 2013 and 2014 in Kadapa, respectively) 307 (Begam et al., 2017). Multiple studies during the monsoon season in a coastal region in southwest 308 India showed BC mass contributions of 1.9 - 5% (Aswini et al., 2019 and references therein). 309 Airborne measurements around North America and in Asian outflow revealed that BC accounted 310 for only ~1-2% of PM_{1.0} (Shingler et al., 2016) and ~5-15% of accumulation mode aerosol mass 311 (Clarke et al., 2004), respectively.

312 To investigate further about the missing species, a charge balance was carried out for all 313 MOUDI sets (Table 2) to compare the sum of charges for cations versus anions based on IC 314 analysis including K from ICP-QQQ analysis (species listed in Section 2.3). The slope of the 315 charge balances (cations on y-axis) for the cumulative dataset was 1.33 and ranged from 0.89 to 316 1.41 for the 12 individual MOUDI sets that had IC and ICP-QQQ analysis conducted on them. 317 Eleven of the 12 sets exhibited slopes above unity indicating that there was a deficit in the amount 318 of anions detected, which presumably included species such as carbonate and various organics. To 319 further determine if there were especially large anion or cation deficits in specific size ranges, 320 slopes are also reported for $0.056-1 \,\mu\text{m}$ and $> 1 \,\mu\text{m}$. There were no obvious differences other than 321 two MOUDI sets exhibited slopes below 1.0 for the smaller diameter range (0.056-1 µm) while 322 all slopes exceeded unity for $> 1 \mu m$.

323 **3.2 Mass Size Distributions and Morphology**

324 3.2.1 Black Carbon

325 The size-resolved nature of BC has not been characterized in Manila and MOUDI set 326 MO13 offered a view into its mass size distribution (Figure 3a). There was a pronounced peak 327 between 0.18–0.32 μ m (5.0 μ g m⁻³), which is evident visually in the substrate's color when compared to all other stages of that MOUDI set (Figure 3b). This observed peak in the mass size
distribution of BC is similar to previous studies of the outflow of East Asian countries (Shiraiwa
et al., 2008), biomass burning and urban emissions in Texas (Schwarz et al., 2008), measurements
in the Finnish Arctic (Raatikainen et al., 2015), and airborne measurements over Europe
(Reddington et al., 2013). In contrast, measurements in Uji, Japan showed a bimodal size
distribution for the mass concentration of BC in the submicrometer range (Hitzenberger and
Tohno, 2001).

335 In the present study, there were significant amounts of BC extending to as low as the 0.056-0.1 µm MOUDI stage (0.28 µg m⁻³) and extending up in the supermicrometer range, with up to 336 0.25 µg m⁻³ measured between 1.8–3.2 µm. Remarkably, BC accounted for approximately 78.1% 337 338 (51.8%) by mass of the total PM in the range of $0.10 - 0.18 \mu m (0.18 - 0.32 \mu m)$. For comparison, 339 the mass percent contribution of BC measured in the megacity of Nanjing, China was 3.3% (1.6%) 340 at 0.12 (0.08) µm (Ma et al., 2017). Based on visual inspection of color on all MOUDI sets, MO13 341 appears to be representative of the other sets based on the relative intensity of the color black on 342 substrates with different cutpoint diameters (Figure 3b); the $0.18-0.32 \mu m$ substrate always was 343 the most black, with varying degrees of blackness extending consistently into the supermicrometer 344 stages.

345 Microscopy analysis revealed evidence of non-spherical particles in each MOUDI stage 346 below 1 µm (Figure 4), which is significant as the common assumption theoretically is that 347 submicrometer particles are typically spherical (e.g., Mielonen et al., 2011). Errors in this 348 assumption impact numerical modeling results and interpretation of remote sensing data for 349 aerosol particles (e.g., Kahnert et al., 2005), owing to incorrect calculations of parameters such as 350 single scattering albedo, asymmetry parameter, and extinction efficiency (e.g., Mishra et al., 2015). 351 Some studies have noted that submicrometer particles could be composed of an agglomeration of 352 small spherical particles originally formed through gas-to-particle conversion processes (Almeida 353 et al., 2019), which could potentially explain the appearance for some of the observed particles in 354 Figure 4. Since only single particles were examined that may not be fully representative of all 355 particles on a particular MOUDI substrate, it is noteworthy that all five particles shown between 356 $0.056 - 1 \mu m$ were irregularly shaped with signs of both multi-layering and constituents adhered 357 to one another.

358 The images show that a potentially important source of BC in the area could be soot 359 aggregates, which are formed by a vaporization-condensation process during combustion often 360 associated with vehicular exhaust (e.g., Chen et al., 2006; Chithra and Nagendra, 2013; Wu et al., 361 2017). Kecorius et al. (2017) projected that 94% of total roadside refractory PM with number 362 concentration modes at 20 and 80 nm was linked to *jeepneys*, the most popular and inexpensive 363 mode of public transport in Metro Manila. They associated the larger mode with soot 364 agglomerates, which is consistent with the smallest MOUDI size range examined here (0.056-0.1 365 µm; Figure 4b) exhibiting signs of agglomeration.

366 The total BC mass concentration integrated across all stages of MOUDI set MO13 (14.3 367 μ g m⁻³) was remarkably high in contrast to BC levels measured via either filters, aethalometers, or 368 single particle soot photometers in most other urban regions of the world (Metcalf et al., 2012 and references therein): Los Angeles Basin (airborne: 0.002–0.53 µg m⁻³), Atlanta, Georgia (ground: 369 $0.5-3.0 \ \mu g \ m^{-3}$), Mexico City (airborne: 0.276–1.1 $\mu g \ m^{-3}$), Sapporo, Japan (ground: 2.3–8.0 μg 370 m⁻³), Beijing, China (ground: 6.3–11.1 µg m⁻³), Bangalor, India (ground: 0.4–10.2 µg m⁻³), Paris, 371 372 France (ground: 7.9 µg m⁻³), Dushanbe, Russia (ground: 4–20 µg m⁻³), Po Valley, Italy (ground: 0.5–1.5 µg m⁻³), Thessaloniki, Greece (ground: 3.3–8.9 µg m⁻³). This is intriguing in light of 373 374 extensive precipitation, and thus wet scavenging of PM, during the study period, which is offset 375 by enormous anthropogenic emissions in the region, such as by powered vehicles like the *jeepneys* 376 that are notorious for BC exhaust (Kecorius et al., 2017).

377 A possible explanation for the large contribution of BC to PM, and the persistence of PM 378 after rain events (Kim Oanh et al., 2006), is that the BC is not efficiently scavenged by precipitating 379 rain drops. Small particles enter rain drops via diffusion whereas large particles enter via 380 impaction. However, particles with a diameter in the range of 0.1–1 µm (known as the Greenfield 381 gap) are too large to diffuse efficiently and too small to impact, and are therefore not efficiently 382 scavenged (Seinfeld and Pandis, 2016). Absorption spectroscopy of set MO13 (Figure 2b) reveals 383 that 95% of the BC mass is concentrated in the Greenfield gap, and thus the removal of BC due to 384 precipitation is inefficient. The Greenfield gap contains $62 \pm 11\%$ of the total mass (calculated for 385 MO3/MO13) and $65 \pm 10\%$ of the water-soluble mass (calculated for the other 12 MO sets). As 386 noted earlier, BC observations discussed in this paper were based only on a single MOUDI set and 387 the effect of inefficient scavenging in the Greenfield Gap could just be one of the many potential

388 processes affecting the BC mass size distribution. Subsequent work that will include BC 389 measurements in the dry season will further investigate this hypothesis.

390 3.2.2 Water-Soluble Ions

391 There were two characteristic mass size distribution profiles for the water-soluble ions 392 speciated by IC, depending on whether the species were secondarily produced via gas-to-particle 393 conversion or associated with primarily emitted supermicrometer particles. The average IC species 394 mass concentration profile across all MOUDI sets is shown in Figure 5. Secondarily-produced 395 species exhibited a mass concentration mode between 0.32–0.56 µm, including common inorganic species (SO₄²⁻, NH₄⁺), MSA, amines (DMA, TMA+DEA), and a suite of organic acids, such as 396 397 oxalate, phthalate, succinate, and adipate, produced via precursor volatile organic compounds 398 (VOCs). Two organic acids with peaks in other size ranges included maleate (0.56–1 µm) and pyruvate (0.1–0.18 μ m). Sources of the inorganics are well documented with SO₄²⁻ and NH₄⁺ 399 produced by precursor vapors SO₂ and NH₃, respectively, with ocean-emitted dimethylsulfide 400 401 (DMS) as an additional precursor to SO_4^{2-} and the primary precursor to MSA.

402 Precursors leading to secondarily-produced alkyl amines such as DMA, TMA, and DEA 403 likely originated from a combination of industrial activity, marine emissions, biomass burning, 404 vehicular activity, sewage treatment, waste incineration, and the food industry (e.g., Facchini et 405 al., 2008; Sorooshian et al., 2009; Ge et al., 2011; VandenBoer et al., 2011); another key source of 406 these species, animal husbandry (Mosier et al., 1973; Schade and Crutzen, 1995; Sorooshian et al., 407 2008), was ruled out owing to a scarcity of such activity in the study region. Secondarily-produced amine salts likely were formed with SO4²⁻ as the chief anion owing to its much higher 408 409 concentrations relative to NO₃⁻ or organic acids.

410 Dimethylamine was the most abundant amine similar to other marine (Muller et al., 2009) and urban regions (Youn et al., 2015); the average concentration of DMA integrated over all 411 MOUDI stages for all sample sets was 62.2 ng m⁻³ in contrast to 29.8 ng m⁻³ for TMA+DEA. For 412 413 reference, the other key cation (NH₄⁺) participating in salt formation with acids such as H₂SO₄ and 414 HNO₃ was expectedly much more abundant (1.64 μ g m⁻³). With regard to the competitive uptake 415 of DMA versus NH₃ in particles, the molar ratio of DMA:NH₄⁺ exhibited a unimodal profile 416 between 0.1–1.8 µm with a peak of 0.022 between 0.32–0.56 µm and the lowest values at the tails (0.004 between 0.1–0.18 and 1–1.8 μ m); DMA was not above detection limits for either D_p < 0.1 417

418 μ m or D_p > 1.8 μ m. The molar ratios observed were consistent with values measured in urban air 419 of Tucson, Arizona and coastal air in Marina, California (0–0.04; Youn et al., 2015) and near the 420 lower end of the range measured in rural and urban air masses sampled near Toronto (0.005–0.2: 421 VandenBoer et al., 2011).

The most abundant organic acid was oxalate ($195 \pm 144 \text{ ng m}^{-3}$), followed by succinate (21 422 \pm 41 ng m⁻³), phthalate (19 \pm 25 ng m⁻³), maleate (17 \pm 15 ng m⁻³), and adipate (5 \pm 8 ng m⁻³). The 423 424 observation of mass concentrations increasing with decreasing carbon number for dicarboxylic 425 acids (i.e., oxalate > succinate > adipate) is consistent with many past studies for other regions as larger chain acids undergo oxidative decay to eventually form oxalate (e.g., Kawamura and 426 Ikushima, 1993; Kawamura and Sakaguchi, 1999; Sorooshian et al., 2007). Maleate is an 427 428 unsaturated dicarboxylic acid emitted from gas and diesel engines (Rogge et al., 1993) and a 429 product from the photo-oxidation of benzene (Kawamura and Ikushima, 1993). The aromatic 430 dicarboxylic acid phthalate is a known photo-oxidation product of naphthalene and stems largely 431 from plastic processing and fuel combustion (Fraser et al., 2003; Kautzman et al., 2010; Fu et al., 432 2012; Kleindienst et al., 2012). The oxidation product (MSA) of ocean-derived DMS exhibited an overall average concentration of 11 ± 7 ng m⁻³, which is near the lower end of the range of levels 433 reported in other coastal and marine environments (from undetected up to ~200 ng m⁻³) (e.g., 434 435 Saltzman et al., 1983, 1986; Berresheim 1987; Watts et al., 1987; Burgermeister and Georgii, 436 1991; Sorooshian et al., 2015; Xu and Gao, 2015).

Water-soluble species exhibiting a peak in the supermicrometer range, usually between 437 1.8–5.6 μ m, include those with known affiliations with sea salt (Na⁺, Cl⁻, K+, Mg²⁺) and crustal 438 materials such as dust (Ca^{2+}). Nitrate peaked between 1.8-3.2 µm, and was best correlated with 439 Na⁺ and Mg²⁺, suggestive of HNO₃ partitioning to sea salt as has been observed in other coastal 440 441 regions (e.g., Prabhakar et al., 2014a). There was very little NO₃⁻ in the submicrometer range (0.05 \pm 0.04 µg m⁻³) in contrast to supermicrometer sizes (0.78 \pm 0.47 µg m⁻³). More submicrometer 442 443 NO₃⁻ in the form of NH₄NO₃ would be expected if there was an excess of NH₃ after neutralizing SO_4^{2-} . The mean ammonium-to-sulfate molar ratio for submicrometer sizes was 2.32 ± 0.52 (range: 444 1.11 - 2.78), with full neutralization of SO₄²⁻ in 10 of 12 MOUDI sets. Thus, there was a non-445 negligible excess in NH₃ that presumably participated in salt formation with HNO₃ and organic 446 447 species. The significant levels of NO_3^- in the same mode as Na^+ and Cl^- contributed to the significant Cl⁻ depletion observed, as the mean Cl⁻:Na⁺ mass ratio between 1-10 µm (i.e., range of 448

449 peak sea salt influence) was 0.81 ± 0.28 , which is much lower than the ratio for pure sea salt (1.81) 450 (Martens et al., 1973). The subject of Cl⁻ depletion in this region will be investigated more 451 thoroughly in subsequent work.

452 Figure 6 shows SEM images of representative single particles in each supermicrometer 453 stage. As would be expected for sea salt and crustal material, most of the particles shown are not 454 spherical. Interestingly, only the particle shown between 1–1.8 µm was close to being spherical. 455 Its composition based on EDX analysis was accounted for mostly by carbon (93.7%) with lower 456 amounts of oxygen (5.8%) and Fe (0.5%). Sea salt particles were found in the next two stages 457 owing to the highest combined weight percentages of Na⁺ and Cl⁻ based on EDX analysis: 1.8–3.2 458 μ m = 36.9%; 3.2–5.6 μ m = 46.9%. The salt particles are not necessarily cubical but more rounded 459 with signs of agglomeration. These two particles were the only ones among the 11 MOUDI stages exhibiting an EDX signal for S, with contributions amounting to $\sim 2\%$ in each particle. This may 460 be linked to natural SO_4^{2-} existing in sea salt particles. Also, the particle between 3.2–5.6 µm 461 462 contained a trace amount of Sc (1%). The largest three particles ($\geq 5.6 \mu m$) were expectedly 463 irregularly shaped with both sharp and rounded edges, comprised mostly of oxygen, Al, Fe, and 464 Ca based on EDX analysis.

465 **3.2.3 Water-Soluble Elements**

466 Averaged data across all MOUDI sets reveal that ICP-QQQ elements exhibited a variety 467 of mass concentration profiles ranging from a distinct mode in either the sub- or supermicrometer 468 range to having multiple modes below and above 1 µm (averages across all MOUDI sets shown 469 in Figure 7). There were several elements with only one distinct peak, being in one of the two 470 stages between 0.18-1.0 µm, including As, Cd, Co, Cr, Cs, Cu, Hf, Mn, Mo, Ni, Rb, Se, Sn, Tl, V, 471 Pb, and Zn. In contrast, the following elements exhibited only one distinct peak in the 472 supermicrometer range: Al, Ba, P, Sr, Ti, Y, and Zr. The rest of the elements exhibited more 473 complex behavior with two distinct peaks in the sub- and supermicrometer range (Ag, Fe, Nb). 474 The following section discusses relationships between all of the ions and elements with a view 475 towards identifying characteristic sources.

476 **3.3 Characteristic Sources and Species Relationships**

477 A combination of PMF and correlation analysis helped identify clusters of closely-related 478 species stemming from distinct sources. The PMF solution with five factors (Figure 8) was chosen 479 because it passed the criteria of physical meaningfulness and it had a calculated ratio of 480 Qtrue: Qexpected (1.2) that was very close to the theoretical value of 1.0. There was a high coefficient 481 of determination between measured and predicted mass concentration when summing up all species for each MOUDI stage ($r^2 = 0.79$; sample size, n = 132), which added confidence in relying 482 483 on the PMF model for source apportionment of PM. The five distinct clusters were named for their 484 most plausible sources based on the species included in the groupings, with their overall 485 contributions to the total mass based on PMF analysis shown in parenthesis (Table 3): Aged 486 (48.0%), Sea Salt (22.5%), Combustion (18.7%), Vehicular/Resuspended Dust (5.6%), and Waste 487 Processing (5.1%). For reference, a previous study near the northwestern edge of the Philippines 488 identified six source factors for $PM_{2.5}$ that are fairly similar to those here (Bagtasa et al., 2018): 489 sea salt, resuspended fine dust, local solid waste burning, and long range transport of (i) industrial 490 emissions, (ii) solid waste burning, and (iii) secondary sulfate. Each of our five groupings will be 491 discussed in detail below in decreasing order of contribution to total measured mass 492 concentrations.

493 **3.3.1 Aged Aerosol**

494 Although not due to one individual source, there was a distinct PMF factor that included species commonly produced via gas-to-particle conversion processes (NH₄⁺, SO₄²⁻, MSA, 495 496 oxalate). Correlation analysis (Table 4) also pointed to a large cluster of species significantly 497 related to each other, including the aforementioned ions and a suite of other organic acids 498 (phthalate, succinate, adipate), MSA, and DMA. The latter three inorganic and organic acid ions 499 exhibited significant correlations with each other (r ≥ 0.68), but also with several elements (r \ge 500 0.36: K, V, Rb, Cs, Sn), which were likely co-emitted with the precursor vapors of the secondarily 501 produced ions. Although BC concentrations were quantified from set MO13 only, the results 502 showed that BC was significantly correlated (r: 0.61-0.92) with 15 species, including those 503 mentioned above (owing to co-emission) and also a few elements that were found via PMF to be 504 stronger contributors to the Combustion source discussed in Section 3.3.3 (Ni, Cu, As, Se, Cd, Tl, 505 Pb).

506 This PMF source factor is referred to as Aged Aerosol owing to its characteristic species 507 being linked to secondary particle formation from emissions of regional and distant sources. The presence of NH4⁺ and SO4²⁻ could be attributed to precursors from various local and regional 508 509 combustion sources, while MSA and DMA are secondarily produced from ocean-derived gaseous 510 emissions (e.g., Sorooshian et al., 2009). Biomass burning emissions from distant upwind regions 511 such as Sumatra and Borneo (Xian et al., 2013) are likely sources of K. Previous studies (Reid et 512 al., 2012; Wang et al., 2013) have shown that phenomena such as SWM and El-Nino events not 513 only influence biomass burning activities in the Malay Peninsula but also impact the transport and 514 distribution of emissions in the study region. For instance, Reid et al. (2016b) showed that 515 enhancement in monsoonal flow facilitates the advection of biomass burning and anthropogenic 516 emissions to the Philippines from Sumatra and Borneo. Subsequent work will investigate more 517 deeply the impact of biomass burning from those upwind regions on the sample site during the SWM. 518

While NH_4^+ and SO_4^{2-} require time for production owing to being secondarily-produced 519 520 from precursor vapors (i.e., SO₂, NH₃), oxalate is the smallest dicarboxylic acid and requires 521 lengthier chemistry pathways for its production and thus is more likely produced in instances of 522 aerosol transport and aging (e.g., Wonaschuetz et al., 2012; Ervens et al., 2018). The various 523 elements associated with this cluster are co-emitted with the precursors to the aforementioned ions 524 and are linked to a variety of sources: metallurgical processes (Anderson et al., 1988; Csavina et 525 al., 2011; Youn et al., 2016), fuel combustion (Nriagu, 1989; Allen et al., 2001; Shafer et al., 2012; 526 Rocha and Correa, 2018), residual oil combustion (Watson et al., 2004), biomass burning (Maudlin 527 et al., 2015), marine and terrestrial biogenic emissions (Sorooshian et al., 2015), and plastics 528 processing (Fraser et al., 2003). In addition, there is extensive ship traffic in the general study region, which is a major source of species in this cluster of species, particularly V and SO_4^{2-} (e.g., 529 530 Murphy et al., 2009; Coggon et al., 2012).

531 PMF analysis suggested that the Aged Aerosol factor contributed 48.0% to the total water-532 soluble mass budget during the study period. Most of the contribution resided in the submicrometer 533 range (68.9%) unlike the supermicrometer range (18.6%), which is consistent with the overall 534 mass size distribution of total PM peaking in the submicrometer range (Figure 2). The 535 reconstructed mass size distribution for this PMF source factor shows the dominance of the mass 536 in the submicrometer range with a peak between 0.32–0.56 μm (Figure 9). The correlation matrices for the sub- and supermicrometer size ranges also show that the correlations between the species most prominent in the Aged Aerosol category are stronger for the former size range (Tables S2-S3). The contribution of this PMF factor to the supermicrometer range is likely associated with species secondarily produced on coarse aerosol such as dust and sea salt. This is evident in the individual species mass size distributions where there is a dominant submicrometer mode but also non-negligible mass above 1 μ m.

543 Even though the PM in a heavily populated urban region, such as Metro Manila, is typically 544 thought to be dominated by local sources of aerosol, the current PMF results show that contribution 545 from long range transport is still discernible. This finding is contrary to the expectation that the 546 signal of transported aerosol would be lost in the noise of locally-produced aerosol.

547 3.3.2 Sea Salt

548 As the MO sampling site is approximately 13 km from the nearest shoreline (Figure 1a) 549 and downwind of Manila Bay in the SWM season, there was a great potential for marine emissions 550 to impact the samples. There were several species with similar mass size distributions (mode: 1.8-551 5.6 µm) and highly correlated total mass concentrations ($r \ge 0.51$) that are linked to sea salt: Cl⁻, Na^+ , Ca^{2+} , Mg^{2+} , Ba, and Sr. The correlations between these species were stronger when examining 552 553 just the supermicrometer range as compared to the submicrometer range (Tables S2-S3). The 554 majority of these species was used in PMF analysis and formed a distinct cluster amounting to 555 22.0% of the total study period's mass budget. This source contributed only 0.6% to the 556 submicrometer mass concentration but 53.5% for the supermicrometer size range. The 557 reconstructed mass size distribution for this source factor is shifted farthest to the larger diameters 558 as compared to the other four sources with a peak between 1.8-3.2 μ m (Figure 9).

559 It is noteworthy that this factor has the highest share of NO_3^- among all identified sources. 560 This result is consistent with mass size distributions shown in Figure 5 in which NO_3^- peaks in the 561 supermicrometer range similar to sea salt constituents (e.g., Na⁺ and Cl⁻). Although sea salt particles naturally contain NO₃⁻ (Seinfeld and Pandis, 2016) (mass ratio of NO₃⁻:Na⁺ = 9.8×10^{-8} 562 -6.5×10^{-5}), the extremely high ratio of NO₃⁻:Na⁺ (mass ratio ~1.8) suggests that only a negligible 563 564 portion of NO_3^- in this factor originated from primary sea salt particles. Thus, the majority of NO_3^- 565 is most likely due to HNO₃ partitioning to existing sea salt particles (e.g., Fitzgerald, 1991; Allen 566 et al., 1996; Dasgupta et al., 2007; Maudlin et al., 2015). In addition, the Cl⁻:Na⁺ mass ratio in this profile (0.65) is smaller than that in sea salt particles (1.81), indicating high Cl⁻ depletion mainly due to reactions of HNO₃ with NaCl (Ro et al., 2001; Yao et al., 2003; Braun et al., 2017). Moreover, elevated loadings of trace elements (e.g., Ba, Cu, Zn, and Co) could be linked to mixing of marine emissions with urban sources (e.g., vehicle and industrial emissions) during their transport inland to the sampling site (Roth and Okada, 1998). This process of aging is consistent with the observed morphology of the sea salt particles in this study, revealing non-cubical shapes that are rounded owing to the likely addition of acidic species such as HNO₃ (Figure 6).

3.3.3 Combustion

575 There are numerous sources of combustion in the study region, including a variety of 576 mobile sources (e.g., cars, utility vehicles, trucks, buses, motorcycles) and stationary sources (e.g., 577 power stations, cement works, oil refineries, boiler stations, utility boilers). Consequently, the next 578 highest contributor to total mass during the study period according to PMF (18.7%) was the cluster 579 of species including Ni, As, Co, P, Mo, and Cr, which is defined as the Combustion factor. These 580 species have been reported to be rich in particles emitted from combustion of fossil fuel and 581 residual oil (Linak and Miller, 2000; Allen et al., 2001; Wasson et al., 2005; Mahowald et al., 582 2008; Mooibroek et al., 2011; Prabhakar et al., 2014b). Although not included in PMF analysis, 583 other species significantly correlated with the previous ones include maleate and Ag, which also 584 stem from fuel combustion (Kawamura and Kaplan, 1987; Lin et al., 2005; Sorooshian et al., 585 2007). Ag specifically is an element in waste incinerator fly ash (Buchholz and Landsberger, 1993; 586 Tsakalou et al., 2018) and its strong correlation with Co (r = 0.85) and Mo (r = 0.64) provides 587 support for this source factor being linked to combustion processes. Maleate is commonly found 588 in engine exhaust (Kawamura and Kaplan, 1987), while Cr is a tracer for power plant emissions 589 (Singh et al., 2002; Behera et al., 2015). Of all species examined in this study, BC was best 590 correlated with As (r = 0.92), while its correlation with Ni (r = 0.85) was among the highest.

591 As the elements in this cluster peaked in concentration in the submicrometer mode, the 592 weight percentage of this factor is more than double below 1 μ m (23.9%) as compared to above 1 593 μ m (11.3%). The reconstructed mass size distribution for this source factor peaks between 0.18– 594 0.32 μ m, which is smaller than the modal diameter range for the Aged source factor (0.32–0.56 595 μ m) likely owing to closer sources and thus less time for growth to occur via condensation and 596 coagulation.

597 3.3.4 Vehicular/Resuspended Dust

598 The next PMF source factor contains chemical signatures of dust because of high 599 contributions to Al, Ti, Ca, and Fe. These crustal elements are strongly related to resuspension of 600 dust by traffic and construction activities (Singh et al., 2002; Harrison et al., 2011). Other elements 601 that were prominent in this factor included Zr, Y, Mn, Cr, and Ba, which are associated with tire 602 and brake wear (Adachi and Tainosho, 2004; Gietl et al., 2010; Song and Gao, 2011; Harrison et 603 al., 2012; Vossler et al., 2016), although some of them can be linked to exhaust as well (e.g., Lin 604 et al., 2005; Song and Gao, 2011). This source is named Vehicular/Resuspended Dust and 605 contributed 5.6% to the total study period's mass concentrations.

606 The weight percentage contribution of this factor was much higher for the supermicrometer 607 range (11.3%) as compared to the submicrometer range (1.5%), which is consistent with the Sea 608 Salt source factor owing to similar mass size distributions of the individual species associated with 609 the two source categories (Figures 5 and 7). Additional species correlated significantly with the 610 crustal species included Hf and Nb, which also exhibited mass peaks between $1.8-3.2 \mu m$. The 611 reconstructed mass size distribution for this source factor is similar to that of Sea Salt in that there 612 is a peak between $1.8-3.2 \,\mu\text{m}$, but there is less of a unimodal profile owing to what appears to be 613 a secondary mode between 0.56–1.0 µm (Figure 9), which could be linked to some of the non-dust 614 components of vehicular emissions.

615 3.3.5 Waste Processing

616 The final PMF source factor, contributing the least overall to total mass (5.1%), featured 617 Zn, Cd, Pb, Mn, and Cu as its main components. These species are linked to waste processing, 618 including especially electronic waste (e-waste) and battery burning and recycling (Gullett et al., 619 2007; Iijima et al., 2007), which was previously reported for Manila (Pabroa et al., 2011). The 620 latter study reported that although there are a few licensed operations for battery recycling, there 621 are numerous unregulated cottage melters across Manila that regularly melt metal from batteries 622 and discard the waste freely. Fujimori et al. (2012) additionally showed that e-waste recycling led 623 to emissions of the following elements (in agreement with this PMF cluster) around Metro Manila: 624 Ni, Cu, Pb, Zn, Cd, Ag, in, As, Co, Fe, and Mn.

This was the only PMF factor exhibiting comparable weight percentages both below (5.1%) and above 1 μ m (5.3%). This is reflected in the mass size distributions of the species

included in this cluster being fairly uniformly distributed below and above 1 μ m. This is also demonstrated in the reconstructed mass size distribution of this source factor as it clearly exhibits a mode between the other four sources (0.56–1.0 μ m) and is the broadest mode (Figure 9). The explanation for this is likely rooted in the diversity of sources contained within this source profile that lead to different sizes of particles. Examples of such sources include processing of different types of waste at varying temperatures and through various processes (e.g., burning, melting, grinding) (Keshtkar and Ashbaugh, 2007),

634 **4. Conclusions**

This study used various analytical techniques (gravimetry, ion chromatography, triple quadrupole inductively coupled plasma mass spectrometry, black carbon spectroscopy, and microscopy), meteorological data, and a source apportionment model (Positive Matrix Factorization) to characterize the sources, chemical composition, and morphology of size-resolved ambient particulate matter (PM) collected using Micro-Orifice Uniform Deposit Impactors (MOUDIs) in Metro Manila, Philippines during the southwest monsoon season (SWM) season of 2018. The main results of this study include the following:

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The total mass concentrations were measured on two occasions and were 18.6 µg m⁻³ and 53.0 µg m⁻³. Water-soluble mass concentrations were measured on 12 occasions and were on average 8.53 ± 4.48 µg m⁻³ (range = 2.7–16.6 µg m⁻³). Simultaneous measurements of total, water-soluble, and black carbon (BC) mass revealed a composition of 26.9% BC, 31.3% water-soluble components, and 41.8% unaccounted mass.

Size-resolved BC mass concentration was measured on one occasion, with the mass sum of all MOUDI stages reaching 14.3 μg m⁻³. Most of the BC mass (95%) was contained in the 0.1–1 μm range (i.e., the Greenfield gap) where wet scavenging by rain is relatively inefficient. The measured BC peaked in the size range of 0.18 – 0.32 μm and accounted for 51.8% of the measured PM for that stage. In the range of 0.10 – 0.18 μm, the mass percent contribution of BC to the measured PM was 78.1%.

• Most of the total mass resided in the submicrometer mode $(0.32-0.56 \ \mu m)$; however, one 655 MOUDI set revealed an additional supermicrometer mode $(1.8-3.2 \ \mu m)$. Water-soluble 656 species that peaked in the submicrometer mode were associated with secondarily produced 657 species, including inorganic acids, amines, Methanesulfonate (MSA), and organic acids. 658 Water-soluble species that peaked in the supermicrometer mode were associated with sea salt 659 and crustal material. Most of the unaccounted mass was for $D_p > 0.32$ μm.

• The most abundant water-soluble species was SO_4^{2-} (44% ± 6%), followed by NH₄⁺ (18% ± 5%), NO₃⁻ (10 ± 3%), Na⁺ (8 ± 3%), and Cl⁻ (6% ± 3%). Correlation analysis revealed that total water-soluble mass was most correlated with temperature (r = 0.64) and rainfall accumulation (r = -0.49) among meteorological factors considered, although other factors were likely influential such as wind direction and speed.

Regardless of particle size, the majority of single particles examined with energy dispersive
 X-ray spectroscopy (SEM-EDX) were non-spherical with evidence of agglomeration.

667 PMF analysis suggested that there were five factors influencing the water-soluble fraction of 668 PM collected at the sampling site. These factors, their contribution to total water-soluble mass, 669 and the main species that permit them to be linked to a physical source are as follows: Aged Aerosol (48.0%; NH4⁺, SO4²⁻, MSA, oxalate), Sea Salt (22.5%; Cl⁻, NO3⁻, Ca²⁺, Na⁺, Mg²⁺, Ba, 670 Sr), Combustion (18.7%; Ni, As, Co, P, Mo, Cr), Vehicular/Resuspended Dust (5.6%; Al, Ti, 671 672 Fe), and Waste Processing (5.1%; Zn, Cd, Pb, Mn, Cu). The dominant contribution of Aged 673 Aerosol to water-soluble mass contradicts the expectation that locally-produced sources in 674 polluted cities should drown out the signal of transported aerosol from distant upwind areas.

675

676 Although the current study focuses exclusively on the SWM season in Metro Manila, 677 results of this study are applicable to the study of aerosol impacts on Southeast Asia and other 678 regions. First, the detection of Aged Aerosol not only from regional but also distant sources 679 confirms previous studies that PM in the region has the ability to travel long distances during the 680 SWM season. Characterization of aerosol in Metro Manila is therefore important for better 681 understanding the impacts that local emissions will have on locations downwind of Metro Manila, 682 including other populated cities in Southeast and East Asia. Transport of pollution and decreased 683 wet scavenging during the SWM season may become increasingly important as studies have shown 684 a decrease in SWM rainfall and increase in the number of no-rain days during the SWM season in 685 the western Philippines in recent decades (e.g., Cruz et al., 2013).

686 Second, Southeast Asia has been named "one of the most hostile environments on the 687 planet for aerosol remote sensing" (Reid et al., 2013) because of high cloud occurrence. Therefore, 688 space-based remote sensing of aerosol characteristics, such as retrievals of aerosol optical depth 689 (AOD), in this region are difficult. In situ measurements are critical for characterization of PM in 690 this region, especially during seasons such as the SWM when clouds are especially prevalent and 691 remote-sensing retrievals dependent on clear-sky conditions are lacking.

Third, this study provides a valuable dataset to compare to other regions impacted by monsoons where the impacts of enhanced moisture and rainfall on size-resolved composition are not well understood. As aqueous processing results in enhanced production of water-soluble species (e.g., sulfate, organic acids), it is noteworthy for this monsoonal region that the watersoluble fraction remains low relative to BC and other insoluble components. This has major implications for the hygroscopicity of the regional PM.

Finally, the results of this study will be used to inform future sampling campaigns in this region, including CAMP²Ex planned for the SWM season of 2019 based in the Philippines. As the current MOUDI sampling campaign at the Manila Observatory is expected to extend for a full year, future work will focus on changes in aerosol characteristics and sources on a seasonal basis as well as scavenging processes upwind of the measurement site.

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704 *Data availability:* All data used in this work are available upon request.

705

Author Contribution: MTC, MOC, JBS, ABM, CS, and AS designed the experiments and all co authors carried out some aspect of the data collection. MTC, RAB, CS, LM, HD, and AS conducted
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 co-authors.

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711 *Competing interests:* The authors declare that they have no conflict of interest.

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720 **References**

- Adachi, K., and Tainosho, Y.: Characterization of heavy metal particles embedded in tire dust, Environ Int, 30, 1009-1017, 10.1016/j.envint.2004.04.004, 2004.
- 723

724 Alas, H. D., Müller, T., Birmili, W., Kecorius, S., Cambaliza, M.O., Simpas, J. B., Cayetano, M., 725 Weinhold, K., Vallar, E., Galvez, M.C., and Wiedensohler, A.: Spatial Characterization of Black 726 Carbon Mass Concentration in the Atmosphere of a Southeast Asian Megacity: An Air Quality 727 Manila, Philippines. Case Study for Metro Aerosol Air Oual Res. 728 doi.org/10.4209/aaqr.2017.08.0281, 2017.

729

Allen, H. C., Laux, J. M., Vogt, R., FinlaysonPitts, B. J., and Hemminger, J. C.: Water-induced
reorganization of ultrathin nitrate films on NaCl: Implications for the tropospheric chemistry of
sea salt particles, J Phys Chem-Us, 100, 6371-6375, DOI 10.1021/jp953675a, 1996.

733

Allen, A. G., Nemitz, E., Shi, J. P., Harrison, R. M., and Greenwood, J. C.: Size distributions of
trace metals in atmospheric aerosols in the United Kingdom, Atmos Environ, 35, 4581-4591, Doi
10.1016/S1352-2310(01)00190-X, 2001.

737

Almeida, G. P., Bittencourt, A. T., Evangelista, M. S., Vieira-Filho, M. S., and Fornaro, A.:
Characterization of aerosol chemical composition from urban pollution in Brazil and its possible
impacts on the aerosol hygroscopicity and size distribution, Atmos Environ, 202, 149-159,
10.1016/j.atmosenv.2019.01.024, 2019.

742

Anderson, J. R., Aggett, F. J., Buseck, P. R., Germani, M. S., and Shattuck, T. W.: Chemistry of
Individual Aerosol-Particles from Chandler, Arizona, an Arid Urban-Environment, Environ Sci
Technol, 22, 811-818, DOI 10.1021/es00172a011, 1988.

746

Aswini, A. R., Hegde, P., Nair, P. R., and Aryasree, S.: Seasonal changes in carbonaceous

- aerosols over a tropical coastal location in response to meteorological processes, Science of The
 Total Environment, 656, 1261-1279, https://doi.org/10.1016/j.scitotenv.2018.11.366, 2019.
- 750

Bagtasa, G., Cayetano, M. G., and Yuan, C. S.: Seasonal variation and chemical characterization
of PM_{2.5} in northwestern Philippines, Atmos Chem Phys, 18, 4965-4980, 10.5194/acp-18-49652018, 2018.

754

Bautista, A. T., Pabroa, P. C. B., Santos, F. L., Racho, J. M. D., and Quirit, L. L.: Carbonaceous
particulate matter characterization in an urban and a rural site in the Philippines, Atmos Pollut Res,
5, 245-252, 10.5094/Apr.2014.030, 2014.

758

759 Begam, G. R., Vachaspati, C. V., Ahammed, Y. N., Kumar, K. R., Reddy, R. R., Sharma, S. K.,

760 Saxena, M., and Mandal, T. K.: Seasonal characteristics of water-soluble inorganic ions and

carbonaceous aerosols in total suspended particulate matter at a rural semi-arid site, Kadapa

762 (India), Environmental Science and Pollution Research, 24, 1719-1734, 10.1007/s11356-016763 7917-1, 2017.

- 765 Behera, S. N., Betha, R., Huang, X., and Balasubramanian, R.: Characterization and estimation of 766 human airway deposition of size-resolved particulate-bound trace elements during a recent haze
- robin an way deposition of size-resolved particulate-bound trace clements during a recent haze repisode in Southeast Asia, Environ Sci Pollut R, 22, 4265-4280, 10.1007/s11356-014-3645-6,
- 768

2015.

- 769
 770 Berresheim, H.: Biogenic Sulfur Emissions from the Sub-Antarctic and Antarctic Oceans, J
 771 Geophys Res-Atmos, 92, 13245-13262, 10.1029/JD092iD11p13245, 1987.
- 772
- 773 Braun, R. A., Dadashazar, H., MacDonald, A. B., Aldhaif, A. M., Maudlin, L. C., Crosbie, E.,
- Aghdam, M. A., Mardi, A. H., and Sorooshian, A.: Impact of Wildfire Emissions on Chloride
- and Bromide Depletion in Marine Aerosol Particles, Environ Sci Technol, 51, 9013-9021,
 10.1021/acs.est.7b02039, 2017.
- 777
- Buchholz, B. A., and Landsberger, S.: Trace-Metal Analysis of Size-Fractioned Municipal SolidWaste Incinerator Fly-Ash and Its Leachates, J Environ Sci Heal A, 28, 423-441, Doi
 10.1080/10934529309375887, 1993.
- 781
- Burgermeister, S., and Georgii, H. W.: Distribution of Methanesulfonate, Nss Sulfate and
 Dimethylsulfide over the Atlantic and the North-Sea, Atmos Environ a-Gen, 25, 587-595, Doi
 10.1016/0960-1686(91)90056-D, 1991.
- 786 Chen, Y. Z., Shah, N., Huggins, F. E., and Huffman, G. P.: Microanalysis of ambient particles 787 Lexington, KY. electron microscopy, Atmos Environ, 40, 651-663, from by 788 10.1016/j.atmosenv.2005.09.036, 2006. 789
- Chithra, V. S., and Nagendra, S. M. S.: Chemical and morphological characteristics of indoor and
 outdoor particulate matter in an urban environment, Atmos Environ, 77, 579-587,
 10.1016/j.atmosenv.2013.05.044, 2013.
- 793
- Chuang, M.-T., Chang, S.-C., Lin, N.-H., Wang, J.-L., Sheu, G.-R., Chang, Y.-J., and Lee, C.-T.:
 Aerosol chemical properties and related pollutants measured in Dongsha Island in the northern
 South China Sea during 7-SEAS/Dongsha Experiment, Atmospheric Environment, 78, 82-92,
 https://doi.org/10.1016/j.atmosenv.2012.05.014, 2013.
- 798
- 799 Clarke, A. D., Shinozuka, Y., Kapustin, V. N., Howell, S., Huebert, B., Doherty, S., Anderson,
- 800 T., Covert, D., Anderson, J., Hua, X., Moore, K. G., McNaughton, C., Carmichael, G., and
- 801 Weber, R.: Size distributions and mixtures of dust and black carbon aerosol in Asian outflow:
- 802 Physiochemistry and optical properties, J Geophys Res-Atmos, 109, Artn D15s09
- 803 10.1029/2003jd004378, 2004.
- 804
- Coggon, M. M., Sorooshian, A., Wang, Z., Metcalf, A. R., Frossard, A. A., Lin, J. J., Craven, J.
 S., Nenes, A., Jonsson, H. H., Russell, L. M., Flagan, R. C., and Seinfeld, J. H.: Ship impacts on
- the marine atmosphere: insights into the contribution of shipping emissions to the properties of
- marine aerosol and clouds, Atmos Chem Phys, 12, 8439-8458, 10.5194/acp-12-8439-2012, 2012.
- 809

- 810 Cohen, D. D., Stelcer, E., Santos, F. L., Prior, M., Thompson, C., and Pabroa, P. C. B.:
- Fingerprinting and source apportionment of fine particle pollution in Manila by IBA and PMF 811 812 techniques: A 7-year study, X-Ray Spectrom, 38, 18-25, 10.1002/xrs.1112, 2009.
- 813
- 814 Crosbie, E., Sorooshian, A., Monfared, N. A., Shingler, T., and Esmaili, O.: A Multi-Year Aerosol 815 Characterization for the Greater Tehran Area Using Satellite, Surface, and Modeling Data, 816 Atmosphere-Basel, 5, 178-197, 10.3390/atmos5020178, 2014.
- 817
- 818 Crosbie, E., Youn, J. S., Balch, B., Wonaschutz, A., Shingler, T., Wang, Z., Conant, W. C., 819 Betterton, E. A., and Sorooshian, A.: On the competition among aerosol number, size and 820 composition in predicting CCN variability: a multi-annual field study in an urbanized desert, 821 Atmos Chem Phys, 15, 6943-6958, 10.5194/acp-15-6943-2015, 2015.
- 822

823 Cruz, F. T., Narisma, G. T., Villafuerte, M. Q., Chua, K. U. C., and Olaguera, L. M.: A 824 climatological analysis of the southwest monsoon rainfall in the Philippines, Atmos Res, 122, 609-825 616, 10.1016/j.atmosres.2012.06.010, 2013.

- 826
- 827 Csavina, J., Landazuri, A., Wonaschutz, A., Rine, K., Rheinheimer, P., Barbaris, B., Conant, W.,
- 828 Saez, A. E., and Betterton, E. A.: Metal and Metalloid Contaminants in Atmospheric Aerosols 829 from Mining Operations, Water Air Soil Poll, 221, 145-157, 10.1007/s11270-011-0777-x, 2011.
- 830
- 831 Dasgupta, P. K., Campbell, S. W., Al-Horr, R. S., Ullah, S. M. R., Li, J. Z., Amalfitano, C., and
- 832 Poor, N. D.: Conversion of sea salt aerosol to NaNO(3) and the production of HCl: Analysis of
- 833 temporal behavior of aerosol chloride/nitrate and gaseous HCl/HNO(3) concentrations with
- 834 AIM, Atmos Environ, 41, 4242-4257, 10.1016/j.atmosenv.2006.09.054, 2007.
- 835
- 836 Dumanoglu, Y., Kara, M., Altiok, H., Odabasi, M., Elbir, T., and Bayram, A.: Spatial and seasonal 837 variation and source apportionment of volatile organic compounds (VOCs) in a heavily 838 industrialized region, Atmos Environ, 98, 168-178, 10.1016/j.atmosenv.2014.08.048, 2014.
- 839
- 840 Ervens, B., Sorooshian, A., Aldhaif, A. M., Shingler, T., Crosbie, E., Ziemba, L., Campuzano-
- 841 Jost, P., Jimenez, J. L., and Wisthaler, A.: Is there an aerosol signature of chemical cloud
- 842 processing?, Atmos Chem Phys, 18, 16099-16119, 10.5194/acp-18-16099-2018, 2018.
- 843
- 844 Facchini, M. C., Decesari, S., Rinaldi, M., Carbone, C., Finessi, E., Mircea, M., Fuzzi, S.,
- 845 Moretti, F., Tagliavini, E., Ceburnis, D., and O'Dowd, C. D.: Important Source of Marine
- 846 Secondary Organic Aerosol from Biogenic Amines, Environ Sci Technol, 42, 9116-9121, 10.1021/es8018385, 2008.
- 847 848
- Farren, N. J., Dunmore, R. E., Mead, M. I., Mohd Nadzir, M. S., Samah, A. A., Phang, S. M., 849 850 Bandy, B. J., Sturges, W. T., and Hamilton, J. F.: Chemical characterisation of water-soluble 851 ions in atmospheric particulate matter on the east coast of Peninsular Malaysia, Atmos. Chem. 852 Phys., 19, 1537-1553, 10.5194/acp-19-1537-2019, 2019.
- 853

- 855 10.1016/0960-1686(91)90050-H, 1991.
- 856

⁸⁵⁴ Fitzgerald, J. W.: Marine Aerosols - a Review, Atmos Environ a-Gen, 25, 533-545, Doi

- Fraser, M. P., Cass, G. R., and Simoneit, B. R. T.: Air quality model evaluation data for organics.
- 858 6. C-3-C-24 organic acids, Environ Sci Technol, 37, 446-453, 10.1021/es0209262, 2003.
- 859
- Fu, P. Q., Kawamura, K., Chen, J., Li, J., Sun, Y. L., Liu, Y., Tachibana, E., Aggarwal, S. G.,
- 861 Okuzawa, K., Tanimoto, H., Kanaya, Y., and Wang, Z. F.: Diurnal variations of organic
- 862 molecular tracers and stable carbon isotopic composition in atmospheric aerosols over Mt. Tai in
- the North China Plain: an influence of biomass burning, Atmos Chem Phys, 12, 8359-8375, 10,5194/acp.12,8359-2012, 2012
- 864 10.5194/acp-12-8359-2012, 2012.
- 865
- Fujimori, T., Takigami, H., Agusa, T., Eguchi, A., Bekki, K., Yoshida, A., Terazono, A., and
 Ballesteros, F. C.: Impact of metals in surface matrices from formal and informal electronicwaste recycling around Metro Manila, the Philippines, and intra-Asian comparison, J Hazard
 Mater, 221-222, 139-146, https://doi.org/10.1016/j.jhazmat.2012.04.019, 2012.
- 870
- Ge, X. L., Wexler, A. S., and Clegg, S. L.: Atmospheric amines Part I. A review, Atmos Environ,
 45, 524-546, 10.1016/j.atmosenv.2010.10.012, 2011.
- 873
- Gietl, J. K., Lawrence, R., Thorpe, A. J., and Harrison, R. M.: Identification of brake wear particles
 and derivation of a quantitative tracer for brake dust at a major road, Atmos Environ, 44, 141-146,
 10.1016/j.atmosenv.2009.10.016, 2010.
- Gullett, B. K., Linak, W. P., Touati, A., Wasson, S. J., Gatica, S., and King, C. J.: Characterization
 of air emissions and residual ash from open burning of electronic wastes during simulated
 rudimentary recycling operations, J Mater Cycles Waste, 9, 69-79, 10.1007/s10163-006-0161-x,
 2007.
- 882
- 883 Harrison, R. M., Beddows, D. C. S., and Dall'Osto, M.: PMF Analysis of Wide-Range Particle
- 884 Size Spectra Collected on a Major Highway, Environ Sci Technol, 45, 5522-5528,
- 885 10.1021/es2006622, 2011. 886
- Harrison, R. M., Jones, A. M., Gietl, J., Yin, J. X., and Green, D. C.: Estimation of the
- 888 Contributions of Brake Dust, Tire Wear, and Resuspension to Nonexhaust Traffic Particles
- 889 Derived from Atmospheric Measurements, Environ Sci Technol, 46, 6523-6529,
- 890 10.1021/es300894r, 2012.
- 891
- 892 Hitzenberger, R., and Tohno, S.: Comparison of black carbon (BC) aerosols in two urban areas –
- 893 concentrations and size distributions, Atmospheric Environment, 35, 2153-2167,
- 894 https://doi.org/10.1016/S1352-2310(00)00480-5, 2001.
- 895
- Hopke, P. K., Cohen, D. D., Begum, B. A., Biswas, S. K., Ni, B., Pandit, G. G., Santoso, M.,
- 897 Chung, Y. S., Davy, P., Markwitz, A., Waheed, S., Siddique, N., Santos, F. L., Pabroa, P. C. B.,
- 898 Seneviratne, M. C. S., Wimolwattanapun, W., Bunprapob, S., Vuong, T. B., Duy Hien, P. and
- 899 Markowicz, A.: Urban air quality in the Asian region, Sci. Total Environ., 404(1), 103–112,
- 900 doi:10.1016/j.scitotenv.2008.05.039, 2008.
- 901

Huang, S. L., Rahn, K. A., and Arimoto, R.: Testing and optimizing two factor-analysis techniques
on aerosol at Narragansett, Rhode Island, Atmos Environ, 33, 2169-2185, Doi 10.1016/S13522310(98)00324-0, 1999.

905

Iijima, A., Sato, K., Yano, K., Tago, H., Kato, M., Kimura, H., and Furuta, N.: Particle size and
composition distribution analysis of automotive brake abrasion dusts for the evaluation of
antimony sources of airborne particulate matter, Atmos Environ, 41, 4908-4919,
10.1016/j.atmosenv.2007.02.005, 2007.

910

Kahnert, M., Nousiainen, T., and Veihelmann, B.: Spherical and spheroidal model particles as an
error source in aerosol climate forcing and radiance computations: A case study for feldspar
aerosols, J Geophys Res-Atmos, 110, Artn D18s13, 10.1029/2004jd005558, 2005.

914

Kautzman, K. E., Surratt, J. D., Chan, M. N., Chan, A. W. H., Hersey, S. P., Chhabra, P. S.,
Dalleska, N. F., Wennberg, P. O., Flagan, R. C., and Seinfeld, J. H.: Chemical Composition of
Gas- and Aerosol-Phase Products from the Photooxidation of Naphthalene, J Phys Chem A, 114,

- 918 913-934, 10.1021/jp908530s, 2010.
- 919

Kawamura, K., and Ikushima, K.: Seasonal-Changes in the Distribution of Dicarboxylic-Acids in
the Urban Atmosphere, Environ Sci Technol, 27, 2227-2235, DOI 10.1021/es00047a033, 1993.

922

Kawamura, K., and Kaplan, I. R.: Motor Exhaust Emissions as a Primary Source for DicarboxylicAcids in Los-Angeles Ambient Air, Environ Sci Technol, 21, 105-110, DOI
10.1021/es00155a014, 1987.

926

Kawamura, K., and Sakaguchi, F.: Molecular distributions of water soluble dicarboxylic acids in
marine aerosols over the Pacific Ocean including tropics, J Geophys Res-Atmos, 104, 3501-3509,
Doi 10.1029/1998jd100041, 1999.

930

931 Kecorius, S., Madueño, L., Löndahl, J., Vallar, E., Galvez, M. C., Idolor, L. F., Gonzaga-

- Cayetano, M., Müller, T., Birmili, W., and Wiedensohler, A.: Respiratory tract deposition of
- 933 inhaled roadside ultrafine refractory particles in a polluted megacity of South-East Asia, Science
- 934 of The Total Environment, 663, 265-274, https://doi.org/10.1016/j.scitotenv.2019.01.338, 2019.
 935
- Kecorius, S., Madueno, L., Vallar, E., Alas, H., Betito, G., Birmili, W., Cambaliza, M. O., Catipay,
 G., Gonzaga-Cayetano, M., Galvez, M. C., Lorenzo, G., Muller, T., Simpas, J. B., Tamayo, E. G.,
 and Wiedensohler, A.: Aerosol particle mixing state, refractory particle number size distributions
 and emission factors in a polluted urban environment: Case study of Metro Manila, Philippines,
- 940 Atmos Environ, 170, 169-183, 10.1016/j.atmosenv.2017.09.037, 2017.
- 941
- Keshtkar, H., and Ashbaugh, L. L.: Size distribution of polycyclic aromatic hydrocarbon
 particulate emission factors from agricultural burning, Atmos Environ, 41, 2729-2739,
 10.1016/j.atmosenv.2006.11.043, 2007.
- Kim Oanh, N. T., Upadhyay, N., Zhuang, Y. H., Hao, Z. P., Murthy, D. V. S., Lestari, P., Villarin,
 J. T., Chengchua, K., Co, H. X., Dung, N. T. and Lindgren, E. S.: Particulate air pollution in six

- Asian cities: Spatial and temporal distributions, and associated sources, Atmos. Environ., 40(18),
- 949 3367–3380, doi:10.1016/j.atmosenv.2006.01.050, 2006.
- 950
- 951 Kim Oanh, N. T., Pongkiatkul, P., Cruz, M. T., Trung Dung, N., Phillip, L., Zhang, G., and Lestari,
- 952 P.: Monitoring and Source Apportionment for Particulate Matter Pollution in Six Asian Cities, in:
- Integrated Air Quality Management: Asian Case Studies, Kim Oanh, N. T. (Ed.), CRC Press,
 Taylor & Francis Group, USA, 97-124, 2013.
- 955 Kleindienst, T. E., Jaoui, M., Lewandowski, M., Offenberg, J. H., and Docherty, K. S.: The
- 956 formation of SOA and chemical tracer compounds from the photooxidation of naphthalene and its
- 957 methyl analogs in the presence and absence of nitrogen oxides, Atmos Chem Phys, 12, 8711-8726,
- 958 10.5194/acp-12-8711-2012, 2012.
- 959
- Liao, H., Chen, W. T., and Seinfeld, J. H.: Role of climate change in global predictions of future tropospheric ozone and aerosols, J Geophys Res-Atmos, 111, Artn D12304,
- 962 10.1029/2005jd006852, 2006.
- 963
- Lin, C. C., Chen, S. J., Huang, K. L., Hwang, W. I., Chang-Chien, G. P., and Lin, W. Y.:
 Characteristics of metals in nano/ultrafine/fine/coarse particles collected beside a heavily
 trafficked road, Environ Sci Technol, 39, 8113-8122, 10.1021/es048182a, 2005.
- Linak, W. P., and Miller, C. A.: Comparison of particle size distributions and elemental
 partitioning from the combustion of pulverized coal and residual fuel oil, J Air Waste Manage, 50,
 1532-1544, Doi 10.1080/10473289.2000.10464171, 2000.
- Ma, Y., Li, S., Zheng, J., Khalizov, A., Wang, X., Wang, Z., and Zhou, Y.: Size-resolved
 measurements of mixing state and cloud-nucleating ability of aerosols in Nanjing, China, Journal
 of Geophysical Research: Atmospheres, 122, 9430-9450, 10.1002/2017jd026583, 2017.
- 975

Ma, L., Dadashazar, D., Braun, R. A., MacDonald, A. B., Aghdam, M. A., Maudlin, L. C., and
Sorooshian, A.: Size-resolved characteristics of water-soluble particulate elements in a coastal
area: Source identification, influence of wildfires, and diurnal variability, Atmos. Environ., 206,
72-84, <u>https://doi.org/10.1016/j.atmosenv.2019.02.045</u>, 2019.

- 980
- 981 Mahowald, N., Jickells, T. D., Baker, A. R., Artaxo, P., Benitez-Nelson, C. R., Bergametti, G.,
- Bond, T. C., Chen, Y., Cohen, D. D., Herut, B., Kubilay, N., Losno, R., Luo, C., Maenhaut, W.,
- 983 McGee, K. A., Okin, G. S., Siefert, R. L., and Tsukuda, S.: Global distribution of atmospheric
- 984 phosphorus sources, concentrations and deposition rates, and anthropogenic impacts, Global
 985 Biogeochem Cy, 22, 10.1029/2008gb003240, 2008.
- 986
- Marple, V., Olson, B., Romay, F., Hudak, G., Geerts, S. M. and Lundgren, D.: Second generation
 micro-orifice uniform deposit impactor, 120 MOUDI-II: Design, Evaluation, and application to
 long-term ambient sampling, Aerosol Sci. Technol., 48(4), 427–433,
 doi:10.1080/02786826.2014.884274, 2014.
- 991

- 992 Martens, C. S., Wesolowski, J. J., Harriss, R. C., and Kaifer, R.: Chlorine Loss from Puerto-Rican
- and San-Francisco-Bay Area Marine Aerosols, J Geophys Res, 78, 8778-8792, DOI
 10.1029/JC078i036p08778, 1973.
- 995
- 996 Maudlin, L. C., Wang, Z., Jonsson, H. H., and Sorooshian, A.: Impact of wildfires on size-
- resolved aerosol composition at a coastal California site, Atmos Environ, 119, 59-68,
 10.1016/j.atmosenv.2015.08.039, 2015.
- 999
- 1000 Metcalf, A. R., Craven, J. S., Ensberg, J. J., Brioude, J., Angevine, W., Sorooshian, A., Duong,
- 1001 H. T., Jonsson, H. H., Flagan, R. C., and Seinfeld, J. H.: Black carbon aerosol over the Los
- 1002 Angeles Basin during CalNex, J Geophys Res-Atmos, 117, 10.1029/2011jd017255, 2012.
- 1003
- Mielonen, T., Levy, R. C., Aaltonen, V., Komppula, M., de Leeuw, G., Huttunen, J., Lihavainen,
 H., Kolmonen, P., Lehtinen, K. E. J., and Arola, A.: Evaluating the assumptions of surface
 reflectance and aerosol type selection within the MODIS aerosol retrieval over land: the problem
 of dust type selection, Atmos Meas Tech, 4, 201-214, 10.5194/amt-4-201-2011, 2011.
- 1009 Miller, J., and Miller, J.C.: Statistics and chemometrics for analytical chemistry. Pearson 1010 Education, 2018.
- 1011
- 1012 Mishra, S. K., Agnihotri, R., Yadav, P. K., Singh, S., Prasad, M. V. S. N., Praveen, P. S., Tawale,
- 1013 J. S., Rashmi, Mishra, N. D., Arya, B. C., and Sharma, C.: Morphology of Atmospheric Particles
- 1014 over Semi-Arid Region (Jaipur, Rajasthan) of India: Implications for Optical Properties, Aerosol
 1015 Air Qual Res, 15, 974-+, 10.4209/aaqr.2014.10.0244, 2015.
- 1015
- 1017 Mooibroek, D., Schaap, M., Weijers, E. P., and Hoogerbrugge, R.: Source apportionment and 1018 spatial variability of $PM_{2.5}$ using measurements at five sites in the Netherlands, Atmospheric 1019 Environment, 45, 4180-4191, 10.1016/j.atmosenv.2011.05.017, 2011.
- 1020
- Mosier, A. R., Andre, C. E., and Viets, F. G.: Identification of Aliphatic-Amines Volatilized from
 Cattle Feedyard, Environ Sci Technol, 7, 642-644, DOI 10.1021/es60079a009, 1973.
- 1023
- Muller, C., Iinuma, Y., Karstensen, J., van Pinxteren, D., Lehmann, S., Gnauk, T., and Herrmann,
 H.: Seasonal variation of aliphatic amines in marine sub-micrometer particles at the Cape Verde
 islands, Atmos Chem Phys, 9, 9587-9597, 2009.
- 1027
- 1028 Murphy, S. M., Agrawal, H., Sorooshian, A., Padro, L. T., Gates, H., Hersey, S., Welch, W. A.,
- Jung, H., Miller, J. W., Cocker, D. R., Nenes, A., Jonsson, H. H., Flagan, R. C., and Seinfeld, J.
 H.: Comprehensive Simultaneous Shipboard and Airborne Characterization of Exhaust from a
- Modern Container Ship at Sea, Environ Sci Technol, 43, 4626-4640, 10.1021/es802413j, 2009.
- 1033 Norris, G., Duvall, R., Brown, S., and Bai, S.: EPA Positive Matrix Factorization (PMF) 5.0
 1034 fundamentals and User Guide Prepared for the US Environmental Protection Agency Office of
 1035 Research and Development, Washington, DC. Inc., Petaluma, 2014.
- 1036
- 1037 Nriagu, J. O.: A Global Assessment of Natural Sources of Atmospheric Trace-Metals, Nature, 338,
 1038 47-49, DOI 10.1038/338047a0, 1989.

- 1039
- Pabroa, P. C. B., Santos, F. L., Morco, R. P., Racho, J. M. D., Bautista, A. T., and Bucal, C. G. D.:
 Receptor modeling studies for the characterization of air particulate lead pollution sources in
 Valenzuela sampling site (Philippines), Atmos Pollut Res, 2, 213-218, 10.5094/Apr.2011.027,
 2011.
- 1043 1044
- 1045 Philippine Statistics Authority: https://psa.gov.ph/, Accessed 28 August 2018.
- Polissar, A., Hopke, P., Paatero, P., Malm, W., and Sisler, J.: Atmospheric aerosol over Alaska 2.
 Elemental composition and sources. Journal of Geophysical Research 103, 19045-19057, 1998.
- 1050 Prabhakar, G., Ervens, B., Wang, Z., Maudlin, L. C., Coggon, M. M., Jonsson, H. H., Seinfeld, J.
- 1051 H., and Sorooshian, A.: Sources of nitrate in stratocumulus cloud water: Airborne measurements
- during the 2011 E-PEACE and 2013 NiCE studies, Atmos Environ, 97, 166-173,
- 1053 10.1016/j.atmosenv.2014.08.019, 2014a.
- 1054
- Prabhakar, G., Sorooshian, A., Toffol, E., Arellano, A. F., and Betterton, E. A.: Spatiotemporal
 distribution of airborne particulate metals and metalloids in a populated arid region, Atmos
 Environ, 92, 339-347, 10.1016/j.atmosenv.2014.04.044, 2014b.
- 1058
- Qu, W. J., Wang, J., Zhang, X. Y., Wang, D., and Sheng, L. F.: Influence of relative humidity on
 aerosol composition: Impacts on light extinction and visibility impairment at two sites in coastal
 area of China, Atmos Res, 153, 500-511, 10.1016/j.atmosres.2014.10.009, 2015.
- Raatikainen, T., Brus, D., Hyvärinen, A. P., Svensson, J., Asmi, E., and Lihavainen, H.: Black
 carbon concentrations and mixing state in the Finnish Arctic, Atmos. Chem. Phys., 15, 1005710057, 10070, 10.5194/acp-15-10057-2015, 2015.
- 1066
- 1067 Ramachandran, S., and Rajesh, T. A.: Black carbon aerosol mass concentrations over
 1068 Ahmedabad, an urban location in western India: Comparison with urban sites in Asia, Europe,
 1069 Canada, and the United States, J Geophys Res-Atmos, 112, 10.1029/2006jd007488, 2007.
- 1070
- Ran, L., Deng, Z. Z., Wang, P. C., and Xia, X. A.: Black carbon and wavelength-dependent aerosol
 absorption in the North China Plain based on two-year aethalometer measurements, Atmos
 Environ, 142, 132-144, 10.1016/j.atmosenv.2016.07.014, 2016.
- 1074
- Reddington, C. L., McMeeking, G., Mann, G. W., Coe, H., Frontoso, M. G., Liu, D., Flynn, M.,
 Spracklen, D. V., and Carslaw, K. S.: The mass and number size distributions of black carbon
 aerosol over Europe, Atmos. Chem. Phys., 13, 4917-4939, 10.5194/acp-13-4917-2013, 2013.
- 1078
- 1079 Reff, A., Eberly, S.I., and Bhave, P.V.: Receptor modeling of ambient particulate matter data using
 1080 positive matrix factorization: Review of existing methods. J Air Waste Manage 57, 146-154, 2007.
 1081
- 1082 Reid, J. S., Xian, P., Hyer, E. J., Flatau, M. K., Ramirez, E. M., Turk, F. J., Sampson, C. R., Zhang,
- 1083 C., Fukada, E. M., and Maloney, E. D.: Multi-scale meteorological conceptual analysis of observed
 1084 active fire hotspot activity and smoke optical depth in the Maritime Continent, Atmos Chem Phys,
 1085 12, 2117-2147, 10.5194/acp-12-2117-2012, 2012.

1087 Reid, J. S., Hyer, E. J., Johnson, R. S., Holben, B. N., Yokelson, R. J., Zhang, J. L., Campbell, J. 1088 R., Christopher, S. A., Di Girolamo, L., Giglio, L., Holz, R. E., Kearney, C., Miettinen, J., Reid, 1089 E. A., Turk, F. J., Wang, J., Xian, P., Zhao, G. Y., Balasubramanian, R., Chew, B. N., Janjai, S., 1090 Lagrosas, N., Lestari, P., Lin, N. H., Mahmud, M., Nguyen, A. X., Norris, B., Oanh, N. T. K., Oo, 1091 M., Salinas, S. V., Welton, E. J., and Liew, S. C.: Observing and understanding the Southeast 1092 Asian aerosol system by remote sensing: An initial review and analysis for the Seven Southeast 1093 Asian Studies (7SEAS) program, Atmos Res, 122, 403-468, 10.1016/j.atmosres.2012.06.005, 1094 2013.

1095

Reid, J. S., Xian, P., Holben, B. N., Hyer, E. J., Reid, E. A., Salinas, S. V., Zhang, J. L., Campbell,
J. R., Chew, B. N., Holz, R. E., Kuciauskas, A. P., Lagrosas, N., Posselt, D. J., Sampson, C. R.,
Walker, A. L., Welton, E. J., and Zhang, C. D.: Aerosol meteorology of the Maritime Continent
for the 2012 7SEAS southwest monsoon intensive study - Part 1: regional-scale phenomena,
Atmos Chem Phys, 16, 14041-14056, 10.5194/acp-16-14041-2016, 2016a.

1101

1102 Reid, J. S., Lagrosas, N. D., Jonsson, H. H., Reid, E. A., Atwood, S. A., Boyd, T. J., Ghate, V. P., 1103 Xian, P., Posselt, D. J., Simpas, J. B., Uy, S. N., Zaiger, K., Blake, D. R., Bucholtz, A., Campbell, 1104 J. R., Chew, B. N., Cliff, S. S., Holben, B. N., Holz, R. E., Hyer, E. J., Kreidenweis, S. M., 1105 Kuciauskas, A. P., Lolli, S., Oo, M., Perry, K. D., Salinas, S. V., Sessions, W. R., Smirnov, A., 1106 Walker, A. L., Wang, Q., Yu, L. Y., Zhang, J. L., and Zhao, Y. J.: Aerosol meteorology of 1107 Maritime Continent for the 2012 7SEAS southwest monsoon intensive study - Part 2: Philippine 1108 receptor observations of fine-scale aerosol behavior, Atmos Chem Phys, 16, 14057-14078, 1109 10.5194/acp-16-14057-2016, 2016b.

1110

Rocha, L. D. S., and Correa, S. M.: Determination of size-segregated elements in diesel-biodiesel
blend exhaust emissions, Environ Sci Pollut R, 25, 18121-18129, 10.1007/s11356-018-1980-8,
2018.

1114

Rogge, W. F., Mazurek, M. A., Hildemann, L. M., Cass, G. R., and Simoneit, B. R. T.:
Quantification of Urban Organic Aerosols at a Molecular-Level - Identification, Abundance and
Seasonal-Variation, Atmos Environ a-Gen, 27, 1309-1330, Doi 10.1016/0960-1686(93)90257-Y,
1118 1993.

1119

1120 Ro, C. U., Oh, K. Y., Kim, H., Kim, Y. P., Lee, C. B., Kim, K. H., Kang, C. H., Osan, J., De

1121 Hoog, J., Worobiec, A., and Van Grieken, R.: Single-particle analysis of aerosols at Cheju

- 1122 Island, Korea, using low-Z electron probe X-ray microanalysis: A direct proof of nitrate
- 1123 formation from sea salts, Environ Sci Technol, 35, 4487-4494, 10.1021/es0155231, 2001.
- 1124
- 1125 Rolph, G.D.: Real-time Environmental Applications and Display sYstem (READY) website
- 1126 (http://ready.Arl.NOAA.Gov), NOAA Air Resour. Lab., Silver Spring, Md., 2016.
- 1127
- 1128 Roth, B., and Okada, K.: On the modification of sea-salt particles in the coastal atmosphere,
- 1129 Atmos Environ, 32, 1555-1569, Doi 10.1016/S1352-2310(97)00378-6, 1998.
- 1130

- 1131 Saltzman, E. S., Savoie, D. L., Zika, R. G., and Prospero, J. M.: Methane Sulfonic-Acid in the
- 1132 Marine Atmosphere, J Geophys Res-Oceans, 88, 897-902, DOI 10.1029/JC088iC15p10897, 1983.
- 1133
- 1134 Saltzman, E. S., Savoie, D. L., Prospero, J. M., and Zika, R. G.: Methanesulfonic-Acid and Non-
- 1135 Sea-Salt Sulfate in Pacific Air Regional and Seasonal-Variations, J Atmos Chem, 4, 227-240,
- 1136 Doi 10.1007/Bf00052002, 1986.
- 1137
- 1138 Schade, G. W., and Crutzen, P. J.: Emission of Aliphatic-Amines from Animal Husbandry and
- 1139 Their Reactions Potential Source of N₂O and HCN, J Atmos Chem, 22, 319-346, Doi
- 1140 10.1007/Bf00696641, 1995.
- 1141
- 1142 Schlosser, J. S., Braun, R. A., Bradley, T., Dadashazar, H., MacDonald, A. B., Aldhaif, A. M.,
- 1143 Aghdam, M. A., Mardi, A. H., Xian, P., and Sorooshian, A.: Analysis of Aerosol Composition
- 1144 Data for Western United States Wildfires Between 2005-2015: Dust Emissions, Chloride
- 1145 Depletion, and Most Enhanced Aerosol Constituents, J. Geophys. Res., 122, 1146 doi:10.1002/2017JD026547, 2017.
- 1147 Schwarz, J. P., Gao, R. S., Spackman, J. R., Watts, L. A., Thomson, D. S., Fahey, D. W., Ryerson,
- 1148 T. B., Peischl, J., Holloway, J. S., Trainer, M., Frost, G. J., Baynard, T., Lack, D. A., de Gouw, J.
- 1149 A., Warneke, C., and Del Negro, L. A.: Measurement of the mixing state, mass, and optical size
- 1150 of individual black carbon particles in urban and biomass burning emissions, Geophysical
- 1151 Research Letters, 35, 10.1029/2008gl033968, 2008.
- Seinfeld, J. H., and Pandis, S. N.: Atmospheric chemistry and physics (3rd ed.). New York: Wiley-Interscience, 2016.
- 1154

Shafer, M. M., Toner, B. M., Oyerdier, J. T., Schauer, J. J., Fakra, S. C., Hu, S. H., Herner, J. D.,
and Ayala, A.: Chemical Speciation of Vanadium in Particulate Matter Emitted from Diesel
Vehicles and Urban Atmospheric Aerosols, Environ Sci Technol, 46, 189-195,
10.1021/es200463c, 2012.

- 1159
- 1160 Shingler, T., Sorooshian, A., Ortega, A., Crosbie, E., Wonaschutz, A., Perring, A. E.,
- 1161 Beyersdorf, A., Ziemba, L., Jimenez, J. L., Campuzano-Jost, P., Mikoviny, T., Wisthaler, A., and
- 1162 Russell, L. M.: Ambient observations of hygroscopic growth factor and f(RH) below 1: Case
- studies from surface and airborne measurements, J Geophys Res-Atmos, 121, 13661-13677,
- 1164 10.1002/2016jd025471, 2016.
- 1165

Shiraiwa, M., Kondo, Y., Moteki, N., Takegawa, N., Sahu, L. K., Takami, A., Hatakeyama, S.,
Yonemura, S., and Blake, D. R.: Radiative impact of mixing state of black carbon aerosol in Asian

- 1168 outflow, Journal of Geophysical Research: Atmospheres, 113, 10.1029/2008jd010546, 2008.
- 1169
- 1170 Simpas, J., Lorenzo, G., and Cruz, M. T.: Monitoring Particulate Matter Levels and Composition
- 1171 for Source Apportionment Study in Metro Manila, Philippines, in: Improving Air Quality in Asian
- 1172 Developing Countries: Compilation of Research Findings, Kim Oanh, N. T. (Ed.), NARENCA,
- 1173 Vietnam Publishing House of Natural Resources, Environment and Cartography, Vietnam, 239-
- 1174 261, 2014.
- 1175

- 1176 Singh, M., Jaques, P. A., and Sioutas, C.: Size distribution and diurnal characteristics of particle-
- bound metals in source and receptor sites of the Los Angeles Basin, Atmos Environ, 36, 1675-
- 1178 1689, Pii S1352-2310(02)00166-8, Doi 10.1016/S1352-2310(02)00166-8, 2002.
- 1179
- Song, F., and Gao, Y.: Size distributions of trace elements associated with ambient particular
 matter in the affinity of a major highway in the New Jersey-New York metropolitan area, Atmos
 Environ, 45, 6714-6723, 10.1016/j.atmosenv.2011.08.031, 2011.
- 1183
- 1184 Sorooshian, A., Ng, N. L., Chan, A. W. H., Feingold, G., Flagan, R. C., and Seinfeld, J. H.:
- 1185 Particulate organic acids and overall water-soluble aerosol composition measurements from the
- 1186 2006 Gulf of Mexico Atmospheric Composition and Climate Study (GoMACCS), J Geophys
- 1187 Res-Atmos, 112, 10.1029/2007jd008537, 2007.
- 1188
- 1189 Sorooshian, A., Murphy, S. N., Hersey, S., Gates, H., Padro, L. T., Nenes, A., Brechtel, F. J.,
- 1190 Jonsson, H., Flagan, R. C., and Seinfeld, J. H.: Comprehensive airborne characterization of aerosol
- 1191 from a major bovine source, Atmos Chem Phys, 8, 5489-5520, DOI 10.5194/acp-8-5489-2008,
- 1192 2008. 1193
- 1194 Sorooshian, A., Padro, L. T., Nenes, A., Feingold, G., McComiskey, A., Hersey, S. P., Gates, H.,
- Jonsson, H. H., Miller, S. D., Stephens, G. L., Flagan, R. C., and Seinfeld, J. H.: On the link
- between ocean biota emissions, aerosol, and maritime clouds: Airborne, ground, and satellite
- 1197 measurements off the coast of California, Global Biogeochem Cy, 23,
- 1198 10.1029/2009gb003464, 2009.
- 1199
- Sorooshian, A., Crosbie, E., Maudlin, L. C., Youn, J. S., Wang, Z., Shingler, T., Ortega, A. M.,
 Hersey, S., and Woods, R. K.: Surface and airborne measurements of organosulfur and
 methanesulfonate over the western United States and coastal areas, J Geophys Res-Atmos, 120,
 8535-8548, 10.1002/2015jd023822, 2015.
- 1204
- Stein, A. F., Draxler, R. R., Rolph, G. D., Stunder, B. J. B., Cohen, M. D., and Ngan, F.: NOAA's
 Hysplit Atmospheric Transport and Dispersion Modeling System, B Am Meteorol Soc, 96, 20592077, 10.1175/Bams-D-14-00110.1, 2015.
- 1208
- Tai, A. P. K., Mickley, L. J., and Jacob, D. J.: Correlations between fine particulate matter (PM2.5)
 and meteorological variables in the United States: Implications for the sensitivity of PM2.5 to
 climate change, Atmos Environ, 44, 3976-3984, 10.1016/j.atmosenv.2010.06.060, 2010.
- 1212
- 1213 Tsakalou, C., Papamarkou, S., Tsakiridis, P. E., Bartzas, G., and Tsakalakis, K.: Characterization
- 1214 and leachability evaluation of medical wastes incineration fly and bottom ashes and their
- 1215 vitrification outgrowths, J Environ Chem Eng, 6, 367-376, 10.1016/j.jece.2017.12.012, 2018.
- 1216
- 1217 U.S. Environmental Protection Agency: Monitoring PM2.5 in Ambient Air
- 1218 Using Designated Reference or Class I Equivalent Methods. Report No. EPA-454/B-16-001. US
- 1219 Environmental Protection Agency, Research Triangle Park, NC., 2016.
- 1220

- 1221 VandenBoer, T. C., Petroff, A., Markovic, M. Z., and Murphy, J. G.: Size distribution of alkyl
- amines in continental particulate matter and their online detection in the gas and particle phase, Atmos Chem Phys, 11, 4319-4332, 10.5194/acp-11-4319-2011, 2011.
- 1224
- Villafuerte, M. Q., Matsumoto, J., Akasaka, I., Takahashi, H. G., Kubota, H., and Cinco, T. A.:
 Long-term trends and variability of rainfall extremes in the Philippines, Atmos Res, 137, 1-13,
 10.1016/j.atmosres.2013.09.021, 2014.
- 1228
- 1229 Vossler, T., Cernikovsky, L., Novak, J., and Williams, R.: Source apportionment with
- uncertainty estimates of fine particulate matter in Ostrava, Czech Republic using Positive Matrix
 Factorization, Atmos Pollut Res, 7, 503-512, 10.1016/j.apr.2015.12.004, 2016.
- 1232
- Wang, J., Ge, C., Yang, Z. F., Hyer, E. J., Reid, J. S., Chew, B. N., Mahmud, M., Zhang, Y. X.,
 and Zhang, M. G.: Mesoscale modeling of smoke transport over the Southeast Asian Maritime
 Continent: Interplay of sea breeze, trade wind, typhoon, and topography, Atmos Res, 122, 486-
- 1236 503, 10.1016/j.atmosres.2012.05.009, 2013.
- 1237
- Wang, Y. Q., Zhang, X. Y., and Draxler, R. R.: TrajStat: GIS-based software that uses various
 trajectory statistical analysis methods to identify potential sources from long-term air pollution
 measurement data, Environ Modell Softw, 24, 938-939, 10.1016/j.envsoft.2009.01.004, 2009.
- Wasson, S. J., Linak, W. P., Gullett, B. K., King, C. J., Touati, A., Huggins, F. E., Chen, Y. Z.,
 Shah, N., and Huffman, G. P.: Emissions of chromium, copper, arsenic, and PCDDs/Fs from open
 burning of CCA-treated wood, Environ Sci Technol, 39, 8865-8876, 10.1021/es050891g, 2005.
- Watson, J. G.: Protocol for Applying and Validating the CMB Model for PM_{2.5} and VOC. Report
 No. EPA-451/R-04-001. US Environmental Protection Agency, Research Triangle Park, NC.,
 2004.
- Watts, S. F., Watson, A., and Brimblecombe, P.: Measurements of the Aerosol Concentrations of
 Methanesulfonic Acid, Dimethyl-Sulfoxide and Dimethyl Sulfone in the Marine Atmosphere of
 the British-Isles, Atmos Environ, 21, 2667-2672, Doi 10.1016/0004-6981(87)90198-3, 1987.
- 1253
- Wonaschuetz, A., Sorooshian, A., Ervens, B., Chuang, P. Y., Feingold, G., Murphy, S. M., de
 Gouw, J., Warneke, C., and Jonsson, H. H.: Aerosol and gas re-distribution by shallow cumulus
 clouds: An investigation using airborne measurements, J Geophys Res-Atmos, 117,
- 1257 10.1029/2012jd018089, 2012. 1258
- Wu, D., Zhang, F., Lou, W. H., Li, D., and Chen, J. M.: Chemical characterization and toxicity
 assessment of fine particulate matters emitted from the combustion of petrol and diesel fuels, Sci
 Total Environ, 605, 172-179, 10.1016/j.scitotenv.2017.06.058, 2017.
- 1262
- Xian, P., Reid, J. S., Atwood, S. A., Johnson, R. S., Hyer, E. J., Westphal, D. L., and Sessions, W.:
 Smoke aerosol transport patterns over the Maritime Continent, Atmos Res, 122, 469-485,
 10.1016/j.atmosres.2012.05.006, 2013.
- 1266

- 1267 Xu, G. J., and Gao, Y.: Characterization of marine aerosols and precipitation through shipboard
 1268 observations on the transect between 31 degrees N-32 degrees S in the West Pacific, Atmos Pollut
 1269 Res, 6, 154-161, 10.5094/Apr.2015.018, 2015.
- 1270
- Yao, X. H., Fang, M., and Chan, C. K.: The size dependence of chloride depletion in fine and
 coarse sea-salt particles, Atmos Environ, 37, 743-751, 10.1016/S1352-2310(02)00955-X, 2003.
- 1273
- 1274 Youn, J. S., Wang, Z., Wonaschutz, A., Arellano, A., Betterton, E. A., and Sorooshian, A.:
- 1275 Evidence of aqueous secondary organic aerosol formation from biogenic emissions in the North
- 1276 American Sonoran Desert, Geophys Res Lett, 40, 3468-3472, 10.1002/grl.50644, 2013.
- 1277
- 1278 Youn, J. S., Crosbie, E., Maudlin, L. C., Wang, Z., and Sorooshian, A.: Dimethylamine as a major
- 1279 alkyl amine species in particles and cloud water: Observations in semi-arid and coastal regions,
- 1280 Atmos Environ, 122, 250-258, 10.1016/j.atmosenv.2015.09.061, 2015.
- 1281
- 1282 Youn, J. S., Csavina, J., Rine, K. P., Shingler, T., Taylor, M. P., Saez, A. E., Betterton, E. A., and
- 1283 Sorooshian, A.: Hygroscopic Properties and Respiratory System Deposition Behavior of
- 1284 Particulate Matter Emitted By Mining and Smelting Operations, Environ Sci Technol, 50, 11706-
- 1285 11713, 10.1021/acs.est.6b03621, 2016.
- 1286

- **Table 1.** Summary of average operating parameters, meteorological conditions, and total resolved water-soluble mass concentration for each MOUDI sample set collected at Manila Observatory (MO) during the 2018 Southwest Monsoon period. On two occasions, simultaneous MOUDI sets were collected for one set to undergo gravimetric analysis (MO3 and MO13) to compare with mass resolved from chemical speciation of the water-soluble fraction (MO4 and MO14). One additional MOUDI set devoted to microscopy analysis was collected using aluminum substrates for one hour
- 1293 on August 1 at 30 LPM.
- 1294

Sample	Dates	Durat	Flow	Wind	Wind	Т	Rain	Water-
set name		ion	rate	speed	direction	(°C)	(mm)	soluble mass
		(hrs)	(LPM)	(m/s)	(°)			(µg m ⁻³)
MO1	Jul 19-20	24	30	3.3	90.1	24.9	47	4.6
MO2	Jul 23-25	54	30	1.3	95.8	26.7	7.8	6.5
MO3/4	Jul 25-30	119	28/30	1.2	111.8	26.7	49.6	5.2
MO5	Jul 30-Aug 1	42	29	2.6	98.1	27.5	52.8	9.2
MO6	Aug 6-8	48	27	0.9	127.5	26.1	30.4	5.1
MO7	Aug 14-16	48	28	3.0	107.8	27.8	2.8	13.7
MO8	Aug 22-24	48	29	3.5	108.7	28.1	1	12.8
MO9	Sep 1-3	48	27	0.7	98.6	26.6	51.6	6.2
MO10	Sep 10–12	48	29	1.0	94.7	26.2	78.4	6.4
MO11	Sep 18–20	48	27	0.5	290.2	27.8	0	2.7
MO12	Sep 26-28	48	27	1.2	96.3	27.8	6.8	13.5
MO13/14	Oct 6-8	48	28/26	0.6	108.2	27.8	0.8	16.6

1297	Table 2. Charge balance slopes (cations on y-axis; anions on x-axis) for the MOUDI sets shown
1298	including the averages of all sets (All) for three size ranges: submicrometer stages spanning 0.056
1299	$-1.0 \mu\text{m}$; supermicrometer stages (> 1.0 μ m); and all stages (> 0.056 μ m). The species used in
1300	the charge balance analysis include those speciated with the IC (listed in Section 2.3) plus K from
1301	ICP-QQQ analysis.

Sample set	0.056 – 1.0 μm	> 1 µm	> 0.056 μm
MO1	0.87	1.37	0.89
MO2	1.46	1.26	1.41
MO4	1.25	1.17	1.21
MO5	1.35	1.43	1.41
MO6	1.29	1.45	1.31
MO7	1.40	1.23	1.36
MO8	1.35	1.33	1.36
MO9	1.28	1.55	1.26
MO10	1.37	1.36	1.35
MO11	0.97	1.60	1.27
MO12	1.37	1.19	1.33
MO14	1.31	1.28	1.29
All	1.35	1.24	1.33

Table 3. Contributions (in weight percentage) of each PMF source factor to the total mass indifferent diameter ranges.

Diameter	Aged	Sea Salt	Combustion	Vehicular/	Waste Processing
Range (µm)	Aerosol			Resuspended	
				Dust	
> 0.056	48.0%	22.5%	18.7%	5.6%	5.1%
0.056 - 1.0	68.9%	0.6%	23.9%	1.5%	5.1%
> 1.0	18.6%	53.5%	11.3%	11.3%	5.3%

- 1309 Table 4. Correlation matrix (r values) between water-soluble species based on total MOUDI-
- 1310 integrated mass concentrations (> $0.056 \mu m$). Blank cells represent statistically insignificant
- 1311 values. Results for the sub- and supermicrometer ranges are in Tables S2-S3. Panels A-E
- represent important species from each of the source profiles identified in Section 3.3: A = Aged
- 1313 Aerosol, B = Sea Salt, C = Combustion, D = Vehicular/Resuspended Dust, E = Waste
- 1314 Processing. DMA Dimethylamine, MSA Methanesulfonate, PH Phthalate, OX Oxalate,
- 1315 MA Maleate, SU Succinate, AD Adipate.
- 1316

A)															
OX	1.00														
SO ₄	0.74	1.00													
NH ₄	0.68	0.99	1.00												
Sn	0.71	0.87	0.85	1.00		_									
Rb	0.73	0.74	0.73	0.69	1.00										
K	0.76	0.71	0.69	0.69	0.97	1.00									
Cs	0.72	0.82	0.81	0.74	0.96	0.91	1.00		_						
V	0.36	0.64	0.63	0.48	0.53	0.51	0.57	1.00		_					
DMA		0.35		0.38	0.45	0.37	0.45		1.00						
MSA	0.71	0.89	0.89	0.79	0.90	0.85	0.92	0.51	0.47	1.00					
PH	0.68	0.67	0.68	0.73	0.82	0.76	0.80		0.38	0.88	1.00				
SU	0.63	0.56	0.59	0.44	0.87	0.81	0.82		0.68	0.78	0.84	1.00			
AD	0.40	0.66	0.70	0.62	0.70	0.70	0.77		0.84	0.74	0.75	0.90	1.00		
Se	0.75	0.75	0.73	0.66	0.80	0.78	0.79	0.32	0.34	0.78	0.80	0.88	0.88	1.00	
Tl	0.75	0.87	0.86	0.80	0.89	0.85	0.94	0.74	0.65	0.80	0.52	0.70		0.43	1.00
	OX	SO ₄	NH ₄	Sn	Rb	K	Cs	V	DMA	MSA	PH	SU	AD	Se	Π

	Cl	NO ₃	Ba	Sr	Ca	Na	Mg	Hf
Hf					0.57			1.00
Mg	0.91	0.87	0.77	0.87	0.66	0.99	1.00	
Na	0.93	0.87	0.75	0.85	0.63	1.00		
Ca	0.58	0.79	0.75	0.78	1.00			
Sr	0.78	0.87	0.91	1.00				
Ba	0.66	0.80	1.00					
NO ₃	0.76	1.00						
Cl	1.00							
B)								

C)								
As	1.00							
Ni	0.58	1.00						
Со			1.00					
Р		0.33	0.34	1.00				
Mo					1.00			
Cr	0.62	0.49		0.20		1.00		
MA			0.67		-0.42		1.00	
Ag			0.85		0.64			1.00
	As	Ni	Co	Р	Mo	Cr	Mal	Ag

D)						
Zr	1.00					
Y	0.75	1.00				
Al	0.88	0.76	1.00			
Fe	0.33	0.61	0.25	1.00		
Ti	0.84	0.66	0.82	0.41	1.00	
Nb	0.70	0.50	0.59	0.59	0.70	1.00
	Zr	Y	Al	Fe	Ti	Nb

	Cd	Zn	Cu	Mn	Pb	
Pb	0.78	0.58	0.38	0.27	1.00	
Mn	0.28	0.61	0.22	1.00		
Cu	0.21	0.27	1.00			
Zn	0.60	1.00				
Cd	1.00					
E)						

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Figure 1. (a) Location of Metro Manila, Philippines relative to Southeast Asia. Also shown are 5-day backward trajectory frequencies during the sampling duration based on HYSPLIT cluster analysis; note that 15% correspond to trajectories within the black square. (b) Close-up view of Metro Manila showing the location of the Manila Observatory sampling site with a black rectangle. The base map shows roads, commercial centers, and major transit lines in the city. (c) Land use classification in the vicinity of the sampling site. (Sources: GADM, Snazzy Maps, OpenStreetMap, NOAA HYSPLIT, & TrajSat)

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Culpoint Diameter (µm)

Figure 2. Mass size distributions of total PM (blue markers) and resolved chemical species

- 1337 (colored bars) for MOUDI sets (a) MO3/4 and (b) MO13/14. Note that set MO13 was the single
 1338 MOUDI set where BC was quantified. ICP-QQQ = sum of water-soluble elements except K;
- 1338 MOUDI set where DC was quantified. ICF-QQQ = sum of water-solutie elements except 1
- amines = sum of DMA, TMA, DEA; organic acids = sum of oxalate, succinate, adipate,
- 1340 pyruvate, phthalate, maleate.
- 1341





Figure 3. (a) Mass size distribution of BC retrieved from the MABI optical measurement at 870

- nm for set MO13. Missing values were below detection limits. (b) Photographs of each stage of
 set MO13 with numbers below each image representing the aerodynamic diameter ranges in units
 of µm.





1350 Figure 4. SEM image of a (a) blank filter and (b-f) individual particles in different sub-

- micrometer aerodynamic diameter ranges sampled by the MOUDI: (b) 0.056–0.1 µm, (c) 0.1–
- 0.18 μm, (d) 0.18–0.32 μm, (e) 0.32–0.56 μm, (f) 0.56–1.0 μm.



Figure 5. Average mass size distribution of water-soluble ions speciated via IC in addition to
 potassium from ICP-QQQ analysis.





1360 **Figure 6.** Same as Figure 4, but for different supermicrometer aerodynamic diameter ranges sampled by the MOUDI: (a) $1.0-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $10-1.8 \mu m$, (c) $10-1.8 \mu m$; (c) $10-1.8 \mu m$, (c) $10-1.8 \mu m$; (c) $10-1.8 \mu m$;

- $18 \,\mu\text{m}, (f) > 18 \,\mu\text{m}.$



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1366 Figure 7. Average mass size distribution of water-soluble elements speciated via ICP-QQQ.
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1369 Figure 8. Overview of the PMF five factor solution with blue bars representing mass

- concentrations and red squares signifying the percentage of mass concentration contributed to
- constituents by each source factor.



Figure 9. Reconstructed mass size distributions using PMF for the five major source profiles.