



Effects of organic coating on the nitrate formation by suppressing the N₂O₅ heterogeneous
 hydrolysis: A case study during wintertime in Beijing-Tianjin-Hebei (BTH)

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Abstract: Although stringent emission mitigation strategies have been carried out since 2013 16 17 in Beijing-Tianjin-Hebei (BTH), China, heavy haze with high levels of fine particulate matters (PM_{2.5}) still frequently engulfs the region during wintertime and the nitrate 18 contribution to PM_{2.5} mass has progressively increased. The N₂O₅ heterogeneous hydrolysis 19 is referred to as the most important pathway of the nitrate formation at nighttime. In the 20 present study, the WRF-CHEM model is applied to simulate a heavy haze episode from 10 to 21 22 27 February 2014 in BTH to evaluate the contribution of the N_2O_5 heterogeneous hydrolysis and the effect of organic coating to the nitrate formation. The model generally performs 23 reasonably well in simulating meteorological parameters, air pollutants and aerosol species 24 against observations in BTH. The N₂O₅ heterogeneous hydrolysis with all the secondary 25 organic aerosol assumed to be involved in coating considerably improves the nitrate 26 simulations compared to the measurements in Beijing. On average, organic coating decreases 27 nitrate concentrations by 8.4% in BTH during the episode, and the N_2O_5 heterogeneous 28 hydrolysis with organic coating contributes about 30.1% of nitrate concentrations. 29 Additionally, the reaction also plays a considerable role in the heavy haze formation, with a 30 PM2.5 contribution of about 11.6% in BTH. Sensitivity studies also reveal that future studies 31 need to be conducted to predict the organic aerosol hygroscopicity for accurately representing 32 33 the organic coating effect on the N₂O₅ heterogeneous hydrolysis.

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39 1 Introduction

Within recent decades, China has been suffering from pervasive and persistent haze 40 pollution caused by elevated levels of fine particulate matters (PM_{2.5}), particularly in 41 Beijing-Tianjin-Hebei (BTH) (Guo et al., 2014; Gao et al., 2016; Wang et al., 2016). 42 Numerous studies have revealed that the inorganic aerosols, including nitrate, sulfate and 43 44 ammonium, are the most abundant component of PM2.5 during haze pollution episodes in 45 BTH, and that the evolution of the haze pollution is characterized by the formation of substantial amounts of sulfate and nitrate (Sun et al., 2013; Zhang et al., 2013; Zhao et al., 46 47 2013; Sun et al., 2015). Since 2013, several aggressive emission control strategies have been implemented in China, including desulfurization and dedusting for coal combustion, vehicle 48 restriction and executing stringent emission standards in the key industries (Tao et al., 2017). 49 However, the control of emissions of nitrate gaseous precursors does not seem to be effective, 50 since many observations have shown that the nitrate aerosol concentration has progressively 51 increased in recent several years (Zhang et al., 2012; Sun et al., 2015; Zhang et al., 2015; Tao 52 et al., 2017). 53

In the atmosphere, nitrate aerosol is formed via nitrous acid (HNO₃) to balance the 54 inorganic cations in the aerosol phase. HNO₃ is produced through four pathways (Kim et al., 55 2014): (1) the reaction of OH and NO_2 (main gas phase pathway and usually considered as 56 the daytime pathway because the OH radical is severely limited at night due to lack of O_3 and 57 peroxide photolysis), (2) NO_3 radical reaction with hydrocarbons, (3) aqueous reaction of 58 NO₃ radical to form HNO₃, and (4) NO₃ conversion to N₂O₅ with subsequently 59 heterogeneous chemical conversion to form HNO₃. The last pathway is referred to as the 60 most important pathway during nighttime, since both NO₃ and N_2O_5 are photolytically liable, 61 or even under heavy haze situation with weak sunlight and high relative humidity (RH) 62 (Brown et al., 2016). 63





The heterogeneous hydrolysis of N2O5 on the surface of deliquescent aerosols to form 64 HNO₃ is quantified by the reaction probability ($\gamma_{N_2O_5}$) (Bertram and Thornton, 2009; Chen et 65 al., 2018; Davis et al., 2008; Riemer et al., 2003). $\gamma_{N_2O_5}$ has been measured by previous 66 laboratory experiments, dependent on particulate chemical composition, RH, temperature, 67 aerosol surface area and water content (Chang et al., 2011), and in the order of 10⁻² (Zheng et 68 al., 2015). In modeling studies, various parameterizations of $\gamma_{N_2O_5}$ are used to simulate the 69 nitrate formation. Dentener and Crutzen (1993) have first used 0.1 as the representative 70 $\gamma_{N_2O_5}$ in a three-dimensional global model. Riemer et al. (2003) have developed a $\gamma_{N_2O_5}$ 71 parameterization on the surface of aerosols containing sulfate and nitrate (hereafter referred 72 as to Riemer03), which has widely been used and further improved in air quality models. 73 Davis et al. (2008) have implemented a $\gamma_{N_2O_5}$ parameterization on the surface of particles 74 containing sulfate, nitrate, ammonium, as a function of RH, temperature and phase state, to 75 76 improve simulations of N₂O₅ hydrolysis. Bertram and Thornton (2009) have developed a 77 parameterization to consider the influence of chloride salts on $\gamma_{N_2O_5}$ as a function of RH.

78 The coating of particles by organic materials has been reported to inhibit N_2O_5 uptake (Anttila et al., 2006), and suggested as a possible explanation for field observations of 79 suppressed N₂O₅ uptake (Brown et al., 2006). Evans and Jacob (2005) have incorporated a 80 $\gamma_{N_2O_5}$ parameterization on surfaces of sulfate particles as a function of RH and temperature 81 into the GEOS-CHEM model, including the effects of dust, sea salt, sulfate, elemental carbon 82 and organic carbon but ignoring the nitrogen-containing species. Riemer et al. (2009) have 83 developed a N_2O_5 uptake parameterization (Riemer09) based on the laboratory results of 84 Anttila et al. (2006), which combines the nitrogen-containing and organic effects on N_2O_5 85 hydrolysis. The parameterization has been used to estimate the maximum effect of organic 86 87 coating by assuming that all available secondary organic compounds (SOC) contribute to the 88 coating in a 3-D model. The results show that SOC could suppress N_2O_5 uptake significantly,





reducing particulate nitrate concentrations by up to 90%. Lowe et al. (2015) have further 89 combined the organic coating and chloride salts effects on $\gamma_{N_2O_5}$ in the WRF-CHEM model. 90 Most recently, Chen et al. (2018) have developed a new $\gamma_{N_2O_5}$ parameterization with respect 91 to RH, temperature, and aerosol composition, showing that organic coating effect on $\gamma_{N_2O_{\rm F}}$ 92 is not as important as expected over western and central Europe. However, there is still a lack 93 of modeling studies focused on the effect of organic coating on $\gamma_{N_2O_5}$ and particulate nitrate 94 formation in China. Wang et al. (2017) have evaluated the potential particulate nitrate 95 formation through the N_2O_5 hydrolysis reaction without considering the organic coating 96 effect during a haze pollution episode in Beijing, and found that the observed nitrate 97 concentration (20.6 μ g m⁻³ on average) is lower than the assessment (57.0 μ g m⁻³ on average). 98 Considering the high organic aerosol concentration and increasing trend of particulate nitrate 99 during haze days in BTH, it is imperative to assess the effect of organic coating on N_2O_5 100 hydrolysis and its consequent contribution to the nitrate formation. 101

In the present study, based on Riemer09 parameterization, the contribution of the organic coating effect on N₂O₅ hydrolysis to the nitrate formation is investigated using the WRF-CHEM model. The model configuration and methodology are described in Section 2. Results and sensitivity studies are presented in Section 3. Discussion and summary are given in section 4.

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108 2 Model and Methodology

109 2.1 WRF-CHEM model and configuration

A modified version of the WRF-CHEM model (Grell et al., 2005) is used in this study, which is developed by Li et al. (2010; 2011a; 2011b; 2012) at the Molina Center for Energy and the Environment. The version includes a new flexible gas-phase chemical module and the Community Multi-scale Air Quality (CMAQ) aerosol module developed by the US EPA





(Binkowski and Roselle, 2003). Briefly, the wet deposition uses the method in the CMAQ 114 module and the dry deposition of chemical species is parameterized following Wesely (1989). 115 The photolysis rates are calculated using the Fast Tropospheric Ultraviolet and Visible 116 117 Radiation Model (FTUV; Tie et al., 2003; Li et al., 2005), with the aerosol and cloud effects on the photochemistry (Li et al., 2011a). The ISORROPIA Version 1.7 is applied to calculate 118 119 the inorganic components (Nenes et al., 1998). The secondary organic aerosol (SOA) is 120 simulated using a non-traditional module, including the volatility basis-set (VBS) modeling 121 approach and SOA contributions from glyoxal and methylglyoxal.

A heavy haze episode from 10 to 27 February 2014 in BTH is simulated in association with the field observation of air pollutants and secondary inorganic aerosols. Detailed model configuration can be found in Table 1 and the simulation domain is presented in Figure 1.

125 2.2 Parameterization of the heterogeneous hydrolysis of N₂O₅

The reaction of N_2O_5 heterogeneous hydrolysis on the surface of deliquescent aerosols to form HNO₃ can be represented as:

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$$N_2O_5 + H_2O \rightarrow 2 \cdot HNO_3$$
 (Eq. 1)

129 This reaction is usually implemented into air quality transport models as a first order loss:

130
$$\frac{\partial [N_2 O_5]}{\partial t} = -k_{N_2 O_5} \cdot [N_2 O_5]$$
(Eq. 2)

131 $[N_2O_5]$ represents the N₂O₅ concentration in the atmosphere. The loss rate constant, $k_{N_2O_5}$, is 132 parameterized in the following way:

133
$$k_{N_2O_5} = \frac{1}{4} \cdot c_{N_2O_5} \cdot S \cdot \gamma_{N_2O_5}$$
 (Eq. 3)

Where $c_{N_2O_5}$ is the average molecular velocity of N₂O₅, and *S* is the available aerosol surface area density. In the Riemer03 parameterization, $\gamma_{N_2O_5}$ is defined as:

136
$$\gamma_{N_2O_5} = f \cdot \gamma_1 + (1 - f) \cdot$$
 (Eq. 4)

137 with $\gamma_1 = 0.02$ and $\gamma_2 = 0.002$, and *f* is defined as:





138
$$f = \frac{m_{SO_4^2}}{m_{SO_4^2} + m_{NO_3^-}}$$
(Eq. 5)

139 $m_{SO_4^{2-}}$ and $m_{NO_3^{-}}$ are the aerosol mass concentrations of sulfate and nitrate. In the Riemer09 140 parameterization, an unreactive organic shell is considered for the suppression of N₂O₅ 141 hydrolysis by covering the reactive inorganic core. The resistor scheme to calculate $\gamma_{N_2O_5}$ is 142 parameterized as follows:

143
$$\frac{1}{\gamma_{N_2O_5}} = \frac{1}{\gamma_{N_2O_5,core}} + \frac{1}{\gamma_{N_2O_5,coat}}$$
(Eq. 6)

where $\gamma_{N_2O_5,core}$ is the reaction probability of the inorganic core which is calculated using Eq. 3, and $\gamma_{N_2O_5,coat}$ is the pseudo-reaction probability of the organic coating calculated by the following formulation:

147
$$\gamma_{N_2O_5,coat} = \frac{4 \cdot R \cdot T \cdot H_{org} \cdot D_{org} \cdot R_c}{c_{N_2O_5} \cdot \ell \cdot R_p}$$
(Eq. 7)

Where R is the universal gas constant, T is temperature, H_{org} is the Henry's law constant for 148 N_2O_5 in the organic coating, and D_{org} is the diffusion coefficient for N_2O_5 in the organic 149 coating. H_{org} and D_{org} depend on the physicochemical properties of the compounds 150 151 comprising the organic coating. In the Riemer09 scheme, $H_{org} \cdot D_{org}$ is defined as $0.03 \cdot H_{aq} \cdot D_{aq}$. H_{aq} is the Henry's law constant of N₂O₅ for the aqueous phase ($H_{aq} = 5000$ 152 M atm⁻¹) and D_{aq} is the diffusion coefficient of N₂O₅ in the aqueous phase ($D_{aq} = 10^{-9} \text{ m}^2$ 153 s⁻¹). R_p , R_c , and ℓ are the radius of the particle, radius of the inorganic core, and thickness 154 of the coating, respectively. R_p , R_c , and ℓ are calculated as follows: 155

156
$$R_p = R_c + \ell \tag{Eq. 8}$$

157
$$\ell = R_p \cdot (1 - \beta^{\frac{1}{3}})$$
 (Eq. 9)

158
$$\beta = \frac{V_{inorg}}{V_{inorg} + V_{org}}$$
(Eq. 10)

159 Where V_{inorg} and V_{org} are the volume of inorganic and organic materials, respectively.





160 2.3 Statistical methods for model evaluation

- 161 In this study, the mean bias (MB), root mean square error (RMSE) and the index of
- agreement (IOA) are used to evaluate the model performance in simulating air pollutants.

163
$$MB = \frac{1}{N} \sum_{i=1}^{N} (P_i - O_i)$$
(Eq. 11)

164
$$RMSE = \left[\frac{1}{N}\sum_{i=1}^{N}(P_i - O_i)^2\right]^{\frac{1}{2}}$$
 (Eq. 12)

165
$$IOA = 1 - \frac{\sum_{i=1}^{N} (P_i - O_i)^2}{\sum_{i=1}^{N} (|P_i - \bar{O}| + |O_i - \bar{O}|)^2}$$
 (Eq. 13)

Where P_i and O_i are the simulated and observed variables, respectively. N is the total number of the simulations for comparisons, and \overline{O} donates the average of the observation. The IOA ranges from 0 to 1, with 1 showing a perfect agreement of the simulation with the observation.

170 2.4 Air pollutants observations

Simulations are compared to available meteorological and air pollutants observations to evaluate the model performance. The meteorological parameters including temperature, RH, wind speed and direction are obtained from the website <u>http://www.meteomanz.com</u>. The hourly observations of PM_{2.5}, O₃, SO₂, NO₂, and CO concentrations are released by China National Environmental Monitoring Center and can be downloaded from the website http://106.37.208.233:20035.

Additionally, hourly OC and EC concentrations are measured using a thermal/optical reflectance carbon analyzer (OCEC RT-4, Sunset Lab, USA) at Chinese Research Academy of Environmental Sciences (CRAES, 40.04°N, 116.40°E) in Beijing (Wei et al., 2014; Liu et al., 2018). Hourly sulfate, nitrate, ammonium, and other inorganic ions are sampled and analyzed by ion chromatography (URG 9000S, Thermo Fisher Scientific, USA).

The OC/EC ratio approach is used to derive the SOA mass concentrations from EC and
OC filter measurements as follows (Strader, 1999; Cao et al., 2004):





184
$$POC = EC \times \left(\frac{POC}{EC}\right)$$
 (Eq. 14)

$$SOC = OC - POC$$
(Eq. 15)

186
$$SOA = SOC \times \left(\frac{SOA}{SOC}\right)$$
 (Eq. 16)

187 Where POC and SOC are the primary OC and secondary OC, respectively. In the present 188 study, $\frac{POC}{EC}$ and $\frac{SOA}{SOC}$ are assumed to be 2.4 and 1.6, respectively, based on the previous 189 studies (Cao et al., 2007; Aiken et al., 2008; Yu et al., 2009) and detailed information about 190 the approach can be found in Feng et al. (2016). It is worth noting that those assumed $\frac{POC}{EC}$ 191 and $\frac{SOA}{SOC}$ could potentially affect the model-measurement comparisons.

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193 **3 Results and discussion**

194 **3.1** Synoptic conditions during the wintertime of 2014

195 Based on the NCEP FNL reanalysis data (https://rda.ucar.edu/datasets/ds083.2), we have initially performed the analysis of synoptic conditions using the wind, temperature, relative 196 humidity, and geopotential height fields at 500hPa and 850hPa averaged from 10 to 27 197 February 2014 over China, respectively (Figure 2). At 500hPa, flat westerly winds prevail 198 over BTH and its surrounding area, indicating a stagnant atmospheric circulation conditions 199 (Figure 2a). Moreover, the flat isotherm distribution is similar to that of the isobar at 500hpa, 200 201 showing that there is no obvious exchange of clod and warm air masses, which together with the flat westerly leads to the weak turbulent mixing in the vertical direction and the stable 202 weather condition (Figure 2b). At 850hPa, the southeast coastal areas of China are controlled 203 204 by the anti-cyclone whose center locates over the South China Sea (Figure 2c). In the eastern China influenced by the anti-cyclone, the weak southerly wind prevails over the BTH and its 205 surrounding regions, providing a favorable condition for stagnant weather conditions and 206 207 further the formation of air pollution. With the prevailing southerly wind, the warm and





208 humid air flow and the polluted air mass are subject to being transported from south to north,

- aggravating the air pollution in BTH. In addition, high relative humidity conditions facilitate
- the heterogeneous reactions for the secondary aerosol formation (Figure 2d).

211 **3.2 Model performance**

In order to quantify effects of the N_2O_5 heterogeneous hydrolysis and organic coating on 212 213 the nitrate formation, three experiments have been performed in the study. In the base case, 214 Riemer09 parameterization is used to take into consideration the organic coating effect on the 215 N₂O₅ heterogeneous hydrolysis by assuming that all the SOA is involved in coating (hereafter 216 referred to as B-case). In the first sensitivity case, the contribution of N_2O_5 heterogeneous hydrolysis to the nitrate formation is not considered (hereafter referred to as H0-case); In the 217 second sensitivity case, the organic coating effect is not considered in Riemer09 218 parameterization (hereafter referred to as C0-case). The simulation results in the B-case are 219 220 compared to observations in BTH.

221 3.2.1 Meteorological parameters simulations in Beijing

Considering that the meteorological conditions play a crucial role in air pollution 222 simulations, which determine accumulation or dispersion of pollutants, verifications are first 223 performed for the simulations of meteorological fields. Figure 3 presents the temporal profile 224 of the simulated and observed temperature, RH, wind speed, and wind direction averaged 225 over 12 meteorological sites in Beijing from 10 to 27 February 2014. The WRF-CHEM 226 model reproduces well the temporal variation of the surface temperature during the whole 227 episode. The MB and RMSE is -0.2 and 1.7°C, and the IOA reaches 0.94, indicating good 228 agreement of the simulations with observations. The simulated temporal RH variations are 229 also well consistent with observations, with the MB, RMSE and IOA of 2.6%, 10.9% and 230 0.89, respectively. In addition, the model reasonably well tracks the temporal variations of the 231 232 surface wind, with IOAs of 0.73 and 0.66 for the wind speed and direction, respectively.





233 3.2.2 Air pollutants simulations in BTH

234	Figure 4 shows the relationship between observed and simulated mass concentrations of
235	PM _{2.5} , O ₃ , SO ₂ , NO ₂ , and CO in Beijing, Tianjin, and Hebei from 10 to 27 February 2014.
236	The correlation coefficient (R) of $PM_{2.5}$ mass concentrations between observations and
237	simulations in Beijing, Tianjin, and Hebei is 0.83, 0.80, and 0.90, respectively, indicating a
238	good performance of the WRF-CHEM model in simulating the PM _{2.5} concentration in BTH.
239	The correlation of O_3 and NO_2 mass concentrations between observations and simulations is
240	not as good as that of $PM_{2.5}$ concentrations in BTH, with the R between 0.6 and 0.8.
241	Apparently, the R of SO ₂ simulations with observations show that the WRF-CHEM model
242	still has difficulties in well simulating SO ₂ concentrations in BTH, particularly in Hebei.
243	Except uncertainties from SO ₂ emissions, such as source intensities and distributions, diurnal
244	profiles, et al., the bias of simulated wind fields also substantially influences the SO ₂
245	simulation (Bei et al., 2017). Particularly, SO ₂ is principally emitted by the point source,
246	including the power plants and agglomerated industrial zones, so the SO ₂ simulations is more
247	sensitive to the wind field simulation uncertainties. In terms of R, the SO_2 simulations in
248	Beijing and Tianjin is better than those in Hebei, indicating that the SO ₂ emissions in Beijing
249	and Tianjin are generally determined by area sources, i.e., the residential living, but the point
250	source dominates the SO ₂ concentration in Hebei. Considering the long-life time of CO in the
251	atmosphere, the CO simulation is decided by its emission and the meteorological fields. The
252	R of CO simulations with observations in BTH ranges from around 0.6 to 0.7, showing that
253	the CO emissions used in the study and simulated meteorological fields are generally
254	reasonable.

Figure 5 presents the diurnal profiles of simulated and observed PM_{2.5}, O₃, NO₂, SO₂, and CO mass concentrations averaged over all ambient monitoring stations in BTH during the simulated episode. The WRF-CHEM model well reproduces the diurnal variations of the





PM2.5 mass concentrations against observations in BTH. The MB and RMSE is -6.3 and 258 27.6µg m⁻³, respectively, and the IOA is 0.96. The model generally well replicates the haze 259 developing stage, but fails to capture the observed spikes of PM2.5 mass concentrations, 260 which might be caused by the uncertainty of the simulated meteorological fields or irregular 261 air pollutants emissions (Bei et al., 2017). The simulated O_3 diurnal variations are in good 262 agreement with observations, with the MB and IOA of 1.4 µg m⁻³ and 0.91, respectively. The 263 264 model tracks well the observed diurnal variations of NO2 mass concentrations with an IOA of 265 0.92, but it slightly overestimates NO₂ concentrations compared to observations with a MB of 6.6 µg m⁻³. However, during nighttime, the model overestimation is considerable, which is 266 perhaps due to the model biases in modeling nighttime PBL. Although the model reasonably 267 yields the variation trend of the observed SO_2 concentration, with an IOA of 0.85, the 268 dispersion of the simulated SO₂ concentration is rather large, with a RMSE of 27.8 μ g m⁻³. In 269 addition, the model overestimates the SO₂ concentration compared to observations, and the 270 MB is 7.6 μ g m⁻³, which might be mainly caused by the emission inventory that has 271 undergone noticeable changes since implementation of emission control strategies in 2013 in 272 BTH. The model performs well in simulating CO diurnal variations against observations, 273 with the MB and IOA of 0.2 μ g m⁻³ and 0.90, respectively. 274

Figure 6 presents the distributions of simulated and observed near-surface mass 275 concentrations of PM_{2.5}, O₃, NO₂, and SO₂ along with the predicted wind fields averaged 276 during the episode. Generally, the simulated wind in BTH is weak during the episode and the 277 southerly wind prevails, well corresponding to the synoptic situation at 850hPa and 500hPa, 278 which is favorable for the accumulation of air pollutants. The observed PM2.5 concentrations 279 are more than 115 µg m⁻³ on average, showing that BTH suffers from heavy haze pollution 280 (Figure 6a). The model generally well reproduces the spatial distribution of PM_{2.5} 281 concentrations against observations, with the $PM_{2.5}$ concentration exceeding 150 µg m⁻³ in 282





the plain area of BTH. The simulated and observed O₃ mass concentrations are less than 50 283 µg m⁻³ in the plain area of BTH, and in several megacities, including Beijing, Tianjin, 284 Baoding, and Shijiazhuang, the O_3 concentrations are less than 30 µg m⁻³ (Figure 6b). The 285 low O₃ concentrations during the episode are generally caused by the weak insolation during 286 wintertime, which is unfavorable for photochemical reactions, and the titration due to high 287 288 NO_x emissions in BTH (Figure 6c). The simulated NO₂ concentrations are generally more than 40 µg m⁻³, consistent with the observations at monitoring sites in BTH. The simulated 289 290 and observed SO_2 mass concentrations in cities or their surrounding areas are still rather high, exceeding 50 μ g m⁻³ (Figure 6d). Elevated SO₂ concentrations in BTH during wintertime are 291 to some degree contributed by the residential coal combustion (Li et al., 2018). High levels of 292 NO₂ and SO₂ show that stringent emission mitigation strategies still need to be implemented 293 in BTH. 294

295 3.2.3 Sulfate, ammonium and SOA simulations in Beijing

The SOA and sulfate concentration directly influences the N_2O_5 heterogeneous 296 hydrolysis in the Riemer09 parameterization, and the ammonium aerosol concentration 297 substantially affects the nitrate aerosol formation. Therefore, Figure 7 presents the temporal 298 profiles of observed and calculated SOA, sulfate, and ammonium mass concentrations at 299 CRAES site in Beijing 10 to 27 February 2014. The model reasonably tracks the diurnal 300 variation of the SOA concentration compared to observations, with the MB and IOA of -1.2 301 μg m⁻³ and 0.83, respectively. The observed SOA concentration exhibits rather large 302 fluctuations, which is not well reproduced by the model. The simulated sulfate trend is 303 generally in agreement with observations with an IOA of 0.88, but there are considerable 304 model biases. During the first pollution event, the model reasonably reproduces the sulfate 305 increase during the haze developing stage, but the early falloff of sulfate concentrations 306 307 during the dissipation stage causes the substantial underestimation. However, during the





second pollution event, the model considerably overestimates the sulfate concentration
against the measurement from 22 to 26 February 2014. The ammonium simulation is slightly
better than that of sulfate, with an IOA of 0.90.

In summary, the WRF-CHEM model performs reasonably well in simulating air pollutants and aerosol species, showing that the simulated meteorological fields and emissions used in the study are generally reasonable.

314 **3.3** Contributions of the N₂O₅ heterogeneous hydrolysis and organic coating to the 315 nitrate formation

316 Figure 8 provides the nitrate temporal variations in the three cases against observations at CRAES site in Beijing from 10 to 27 February 2014. When the N_2O_5 heterogeneous 317 hydrolysis is not considered in the H0-case, although the model well tracks the observed 318 nitrate variations with an IOA of 0.91, it considerably underestimates nitrate concentration 319 against the measurement, with a MB of -17.0 μ g m⁻³. When the N₂O₅ heterogeneous 320 hydrolysis is taken into consideration based on the Riemer09 parameterization without 321 organic coating in the C0-case, the nitrate simulation is improved compared to that in the 322 H0-case, with an IOA of 0.95. However, the model commences to overestimate the nitrate 323 concentration compared to the measurement, with a MB of 5.4 µg m⁻³. In the B-case, when 324 all the SOA is assumed to be involved in coating to suppress the N2O5 heterogeneous uptake 325 on surfaces of deliquescent aerosols, the model performs best in simulating the nitrate 326 variation compared to the measurement, with the MB and IOA of 0.1 µg m⁻³ and 0.96, 327 respectively. The remarkable consistency of the simulated nitrate in the B-case with the 328 measurement indicates that the organic coating plays an important role in improving the 329 nitrate simulation. 330

Figure 9a presents the distribution of contributions of the N_2O_5 heterogeneous hydrolysis to the nitrate formation averaged during the episode by differentiating simulations





in the B-case and H0-case. The contribution of the N2O5 heterogeneous hydrolysis to the 333 nitrate formation is substantial in BTH, exceeding 15 μ g m⁻³ in the plain area. Although the 334 O₃ concentration is fairly low in BTH during the episode (Figure 6b), particularly during 335 nighttime (Figure 5b), the elevated NO₂ level still facilitates the N₂O₅ formation to warrant 336 occurrence of the N_2O_5 heterogeneous hydrolysis. Previous studies have revealed that N_2O_5 337 338 heterogeneous hydrolysis is vital in nitrate formation. For example, Wang et al. (2017) have 339 calculated the daily average nitrate formation potential from the N₂O₅ heterogeneous hydrolysis in Beijing, showing that the reaction accounts for 52% of the total nitrate 340 341 formation. Su et al. (2017) have investigated the contribution of N_2O_5 heterogeneous hydrolysis to the nitrate formation in Beijing during autumn in 2015 and found that the 342 reaction causes a 21.0% enhancement of nitrate concentrations. In the present study, the 343 nitrate contribution of the N₂O₅ heterogeneous hydrolysis is 29.4% in Beijing during the 344 episode on average, which is close to the result in Su et al. (2017) but much lower than that in 345 Wang et al. (2017). The average nitrate contribution of the reaction in BTH is about 30.1%, 346 showing that the reaction constitutes an important nitrate source during the haze pollution 347 episode. Additionally, the N_2O_5 heterogeneous hydrolysis contributes 11.6% of the PM_{2.5} 348 concentration on average, playing a considerable role in the haze formation in BTH. 349

However, it is worth noting that the brute force method (BFM) is used to quantify the 350 contribution of the N_2O_5 heterogeneous hydrolysis to the nitrate formation (Dunker et al., 351 1996). The BFM is generally used to assess the importance of some source, but it lacks 352 consideration of interactions of the complicated physical and chemical processes in the 353 atmosphere (Zhang and Ying, 2011). Therefore, in the study, the contribution of the N_2O_5 354 heterogeneous hydrolysis to the nitrate formation might be underestimated, considering the 355 competition of inorganic cations from HNO₃ formed through gas-phase reactions and sulfate 356 357 aerosols in the atmosphere. It is imperative to use the source-oriented base module to





358 evaluate the nitrate contribution of the reaction.

Figure 9b shows the distribution of the average decrease of nitrate concentrations due to 359 suppression of organic coating during the episode by differentiating simulations in the B-case 360 and C0-case. The organic coating reduces the nitrate concentration by more than 5 µg m⁻³ in 361 the plain area of BTH, and on average, the decrease of nitrate aerosols is 4.7 μ g m⁻³ or 8.4% 362 363 in BTH during the episode. Riemer et al. (2009) have shown that when the nitrate levels are high (above 15 µg m⁻³), the organic coating decreases nitrate concentrations by 10-15% over 364 Europe. However, Cheng et al. (2018) have demonstrated that the suppression of organic 365 366 coating is negligible over western and central Europe, with an influence on nitrate concentrations of less than 2% on average and 20% at the most significant moment. 367 Apparently, except N_2O_5 and water soluble OA in the atmosphere, the effect of organic 368 coating is also dependent on NH₃, RH, and temperature. Hence, the inconsistency between 369 the model results about the organic coating effect can be attributed to the variation in 370 simulation conditions. For example, in order to obtain substantial effects of the organic 371 coating, N₂O₅, SOA, and NH₃ need to be present when RH is high and temperature is low. 372 However, those conditions are rarely fulfilled simultaneously over western and central 373 Europe, causing a negligible effect of organic coating (Chen et al., 2018). Additionally, Wang 374 et al. (2017) have indicated that the evaluated nitrate level with the N_2O_5 heterogeneous 375 hydrolysis in Beijing is much higher than the observation, which they have attributed to 376 atmospheric dilution and deposition. It is worth noting that the organic coating effect might 377 constitute one of the most possible reasons for the overestimation of nitrate concentrations, 378 considering the elevated SOA level in Beijing which suppresses the N_2O_5 heterogeneous 379 hydrolysis and results in high observed N₂O₅ concentrations (Wu et al., 2017). 380

381 3.4 Sensitivity studies of organic aerosol hygroscopicity to the nitrate formation

In section 3.3, the WRF-CHEM model considerably improves nitrate simulations when





considering the N₂O₅ heterogeneous hydrolysis and organic coating effects. Organic aerosols (OA) are broadly classified as primary OA (POA) directly emitted and SOA formed in the atmosphere, some of which are water soluble. In order to explore the effects of different OA coating on the nitrate formation, additional four sensitivity studies are conducted, in which half of SOA (C1-case), all SOA (C2-case), all SOA and half of POA (C3-case), and all SOA and POA (C4-case) are involved in coating, respectively.

389 Figure 10 shows the Taylor diagram (Taylor, 2001) to present the variance, bias and correlation of the observed and simulated nitrate concentrations in the four sensitivity cases at 390 391 CRAES site during the episode. In the C1-case, when half of SOA is considered to involve in coating, the simulated nitrate concentration is the best consistent with the observation, with a 392 correlation coefficient of 0.96. In the C2-case with all SOA assumed to engage in coating, the 393 correlation coefficient decreases to be 0.95. The normalized standardized deviation (NSD) is 394 1.02 for the C1-case and C2-case, showing the model overestimation in these two cases. With 395 half of POA involved in coating, the NSD is very close to 1.0 (0.99), indicating the simulated 396 nitrate concentration in the C3-case is almost the same as the observation on average, but the 397 correlation coefficient of 0.94 is less than those in the C1-case and C2-case. When all of OA 398 is assumed as the coating, the bias between simulated and observed nitrate concentrations is 399 the largest, and the effect of POA on suppressing the N_2O_5 heterogeneous hydrolysis might be 400 overestimated in the C4-case. 401

Sensitivity results show that the effects of different organic compounds on suppressing the N_2O_5 heterogeneous hydrolysis to form nitrate varies, depending on the content of water soluble OA. Laboratory and field measurements have revealed that OA becomes progressively oxidized and more hygroscopic during the aging process in the atmosphere (Jimenez et al., 2009). The OA hygroscopicity constitutes a necessary prerequisite for accurately representing the organic coating effect on the N_2O_5 heterogeneous hydrolysis.





According to the simulations in the present study, in BTH, not all of SOA can serve as the coating to suppress the nitrate formation, and the effect of POA on coating might be neglected. Xing et al. (2018) have shown that in BTH, the heterogeneous SOA formed by irreversible uptake of glyoxal and methylglyoxal on wet aerosol surfaces contributes about 30% of the SOA mass during haze days. Considering the possible heterogeneous SOA contribution of other carbonyl compounds and the atmospheric aging of OA, about half of SOA should likely be hygroscopic and involved in coating.

415

416 **4** Conclusion

Nitrate aerosol has constituted a main component of PM2.5 with implementation of 417 aggressive emission control strategies since 2013 in BTH. In the study, the Riemer09 418 parameterization is implemented into the WRF-CHEM model to simulate the nitrate 419 420 formation from the N_2O_5 heterogeneous hydrolysis referred to as the most important pathway of the nitrate formation at nighttime. A heavy haze episode from 10 to 27 February 2014 in 421 BTH is simulated using the WRF-CHEM model to verify the effect of organic coating on the 422 N₂O₅ heterogeneous hydrolysis and its consequent contribution to the nitrate formation. 423 Analyses of synoptic fields show a stagnant weather condition with the prevailing southerly 424 wind in the low-level atmosphere in BTH and surrounding areas during the episode, 425 facilitating accumulation of air pollutants and heavy haze formation. 426

The WRF-CHEM model performs reasonably in predicting the temporal variations of the meteorological parameters compared to observations in Beijing. The model generally reproduces well the temporal variations and spatial distributions of air pollutants against observations at monitoring sites in BTH. In addition, the simulated diurnal profiles of sulfate, ammonium and SOA are also in good agreements with the measurements at CRAES site in Beijing.





The Riemer09 parameterization with all the SOA assumed to be involved in coating 433 considerably improves the nitrate simulations compared to the measurements at CRAES site 434 435 in Beijing. When organic coating is not considered in the Riemer09 parameterization, the model overestimates the nitrate concentration against the measurements. On average, organic 436 coating decreases nitrate concentrations by 4.7 μ g m⁻³ or 8.4% in BTH during the episode. 437 438 Furthermore, the N₂O₅ heterogeneous hydrolysis with organic coating contributes about 30.1% 439 of nitrate concentrations, and 11.6% of the PM_{2.5} concentration in BTH, playing a considerable role in the haze formation. 440

Sensitivity studies reveal that the OA hygroscopicity is a necessary prerequisite for accurately evaluating the organic coating effect on the N_2O_5 heterogeneous hydrolysis. In the present study, POA might not serve as coating and about half of SOA should be involved in coating to suppress the nitrate formation. Future studies still need to be conducted to further predict the OA hygroscopicity, in order to more precisely represent the organic coating effect on the N_2O_5 heterogeneous hydrolysis in chemical transport models.

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Author contribution. Guohui Li, as the contact author, provided the ideas and financial 449 support, verified the conclusions, and revised the paper. Lang Liu conducted a research, 450 designed the experiments, carried the methodology out, performed the simulation, processed 451 the data, prepared the data visualization, and prepared the manuscript with contributions from 452 all authors. Jiarui Wu and Xia Li provided the treatment of meteorological data, analyzed the 453 study data, validated the model performance, and reviewed the manuscript. Suixin Liu, Yang 454 Qian, Tian Feng, and Jiamao Zhou provided the observation data used in the study, 455 synthesized the observation, and reviewed the paper. Xuexi Tie and Junji Cao provided 456 critical reviews pre-publication stage. 457





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707 Table 1 WRF-CHEM model configurations.

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Regions	Beijing-Tianjin-Hebei (BTH)
Simulation period	February 10 to 27, 2014
Domain size	200×200
Domain center	38.0°N, 116.0°E
Horizontal resolution	6km × 6 km
	35 vertical levels with a stretched vertical grid with spacing
Vertical resolution	ranging from 30m near the surface, to 500m at 2.5km and
	1km above 14km
Microphysics scheme	WSM 6-class graupel scheme (Hong and Lim, 2006)
Boundary layer scheme	MYJ TKE scheme (Janjić, 2002)
Surface layer scheme	MYJ surface scheme (Janjić, 2002)
Land-surface scheme	Unified Noah land-surface model (Chen and Dudhia, 2001)
Long-wave radiation scheme	Goddard longwave scheme (Chou and Suarez, 2001)
Short-wave radiation scheme	Goddard shortwave scheme (Chou and Suarez, 1999)
Meteorological boundary and	NCED 1°×1° regulation data
initial conditions	
Chemical initial and boundary	MOZART 6-hour output (Horowitz et al. 2003)
conditions	MOZART 0-hour output (Horowitz et al., 2005)
Anthropogenic emission	SAPRC-99 chemical mechanism emissions (Zhang et al.,
inventory	2009)
Biogenic emission inventory	MEGAN model developed by Guenther et al. (2006)
Model spin-up time	24 hours

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714	Figure Captions
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716 717 718 719	Figure 1 WRF-CHEM simulation domain with topography. The red filled circles show the locations of the cities with ambient air quality monitoring sites, and the size of the circles represents the number of sites in each city. The blue filled rectangle denotes the CRAES observation site in Beijing.
720 721 722	Figure 2 Distributions of average winds (black flag vectors), geopotential heights (blue lines), temperature (red lines), and relative humidity (contour fill) at (a) and (b) 500hPa and (c) and (d) 850hPafrom 10 to 27 February 2014, respectively.
723 724 725 726	Figure 3 Temporal variations of simulated (red line) and observed (black dots) meteorological parameters of near-surface (a) temperature, (b) relative humidity, (c) wind speed, and (d) wind direction averaged at 12 meteorological sites in Beijing from 10 to 27 February 2014.
727 728 729 730	Figure 4 Relationships between observed and simulated mass concentrations of PM _{2.5} , O ₃ , NO ₂ , SO ₂ , and CO in Beijing, Tianjin, and Hebei from 10 to 27 February 2014. The red line is the linear regression between observations and simulations, and the black dashed line presents the 1:1 line.
731 732 733	Figure 5 Comparison of observed (black dots) and simulated (red line) diurnal profiles of near surface hourly (a) PM _{2.5} , (b) O ₃ , (c) NO ₂ , (d) SO ₂ , and (e) CO averaged over all ambient monitoring stations in BTH from 10 to 27 February 2014.
734 735 736	Figure 6 Spatial distributions of average (a) PM _{2.5} , (b) O ₃ , (c) NO ₂ , and (d) SO ₂ mass concentrations from 10 to 27 February 2014. Colored dots, colored contour, and black arrows are observations, simulations, and simulated surface winds, respectively.
737 738 739	Figure 7 Comparison of observed (black dots) and simulated (red line) diurnal profiles of hourly (a) SOA, (b) sulfate, and (c) ammonium concentrations at CRAES site in Beijing from 10 to 27 February 2014.
740 741 742	Figure 8 Temporal variations of observed (black dot) and the simulated (Green line: H0-case; Blue line: C0-case; Red line: B-case) nitrate concentrations at CRAES site in Beijing from 10 to 27 February 2014.
743 744	Figure 9 Spatial distributions of average nitrate contributions of (a) the N ₂ O ₅ heterogeneous hydrolysis and (b) organic coating in BTH from 10 to 27 February 2014.
745 746 747	Figure 10 Taylor diagram (Taylor, 2001) to present the variance, bias and correlation of the observed and simulated nitrate concentrations at CRAES site in Beijing from 10 to 27 February 2014.
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Figure 1 WRF-CHEM simulation domain with topography. The red filled circles show the
locations of the cities with ambient air quality monitoring sites, and the size of the circles
represents the number of sites in each city. The blue filled rectangle denotes the CRAES
observation site in Beijing.

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Figure 2 Distributions of average winds (black flag vectors), geopotential heights (blue lines),
temperature (red lines), and relative humidity (contour fill) at (a) and (b) 500hPa and (c) and
(d) 850hPafrom 10 to 27 February 2014, respectively.







Figure 3 Temporal variations of simulated (red line) and observed (black dots) meteorological

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Figure 4 Relationships between observed and simulated mass concentrations of PM_{2.5}, O₃,
NO₂, SO₂, and CO in Beijing, Tianjin, and Hebei from 10 to 27 February 2014. The red line
is the linear regression between observations and simulations, and the black dashed line
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Figure 5 Comparison of observed (black dots) and simulated (red line) diurnal profiles of
 near surface hourly (a) PM_{2.5}, (b) O₃, (c) NO₂, (d) SO₂, and (e) CO averaged over all ambient
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Figure 6 Spatial distributions of average (a) PM_{2.5}, (b) O₃, (c) NO₂, and (d) SO₂ mass
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Blue line: C0-case; Red line: B-case) nitrate concentrations at CRAES site in Beijing from 10
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Figure 9 Spatial distributions of average nitrate contributions of (a) the N₂O₅ heterogeneous
hydrolysis and (b) organic coating in BTH from 10 to 27 February 2014.







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Figure 10 Taylor diagram (Taylor, 2001) to present the variance, bias and correlation of the
observed and simulated nitrate concentrations at CRAES site in Beijing from 10 to 27

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