1 Impacts of water partitioning and polarity of organic compounds on

2 secondary organic aerosol over Eastern China

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30 Abstract

Secondary organic aerosol (SOA) is an important component of fine particular matter 31 32 (PM_{2.5}). Most air quality models use an equilibrium partitioning method along with the saturation vapor pressure (SVP) of semi-volatile organic compounds (SVOCs) to predict 33 SOA formation. However, the models typically assume that the organic particulate matter 34 (OPM) is an ideal mixture and ignore the partitioning of water vapor to OPM. In this study, 35 36 the Community Multi-scale Air Quality model (CMAQ) is updated to investigate the 37 impacts of water vapor partitioning and non-ideality of the organic-water mixture on SOA formation during winter (January) and summer (July) of 2013 over eastern China. The 38 updated model treats the partitioning of water vapor molecules into OPM and uses the 39 40 UNIFAC model to estimate the activity coefficients of species in the organic-water mixture. The modified model can generally capture the observed surface organic carbon (OC) with 41 a correlation coefficient R of 0.7, and the surface OA with the mean fractional bias (MFB) 42 and mean fractional error (MFE) of -0.28 and 0.54, respectively. SOA concentration shows 43 44 significant seasonal and spatial variations, with high concentrations in the North China Plain (NCP), Central China and Sichuan basin (SCB) regions during winter (up to 25 µg 45 m^{-3}) and in the Yangtze River Delta (YRD) during summer (up to 16 μ g m⁻³). In winter, 46 SOA decreases slightly in the updated model, with the monthly-averaged relative change 47 of 10-20% in the highly concentrated areas, mainly due to organic-water interactions. The 48 monthly-averaged concentration of SOA increases greatly in summer, by 20-50% at the 49 50 surface and 30-60% in the whole column. The increase of SOA is mainly due to the increase in biogenic SOA in inland areas and anthropogenic SOA in coastal areas. As a result, the 51 averaged aerosol optical depth (AOD) is increased by up to 10% and the cooling effect of 52 aerosol radiative forcing (ARF) is enhanced by up to 15% over YRD in summer. The 53 aerosol liquid water content associated with OPM (ALW_{org}) at the surface is relatively high 54 55 in inland areas in winter and over the ocean in summer, with the monthly-averaged concentration of 0.5-3.0 and 5-7 μ g m⁻³, respectively. The hygroscopicity parameter (κ) of 56

57 OA based on the κ -Köhler theory is determined using the modeled ALW_{org}. The correlation of κ with O: C ratio varies significantly across different cities and seasons. 58 59 Analysis of two representative cities, Jinan (in NCP) and Nanjing (in YRD), shows that the impacts of water partitioning and non-ideality of the organic-water mixture on SOA are 60 sensitive to temperature, relative humidity (RH), and the SVP of SVOCs. The two 61 processes exhibit opposite impacts on SOA in eastern China. Water uptake increases SOA 62 63 by up to 80% in the organic phase, while including non-unity activity coefficients decreases 64 SOA by up to 50%. Our results indicate that both water partitioning into OPM and the activity coefficients of the condensed organics should be considered in simulating SOA 65 formation from gas-particle partitioning, especially in hot and humid environments. 66

67 **Keywords**: SOA, non-ideality, water partitioning, hygroscopicity

68 1 Introduction

Secondary organic aerosol (SOA) is formed via a complex interaction of volatile organic 69 compounds (VOCs) with oxidants and primary particles emitted from anthropogenic and 70 71 biogenic sources in the atmosphere. As an important component of fine particular matter 72 (PM_{2.5}), SOA can cause severe air pollution in urban and suburban areas (Huang et al., 73 2014) and exhibit adverse health effects (Atkinson et al., 2014). SOA also plays an important role in new particle formation and particle growth that further contribute to the 74 enhancement of cloud condensation nuclei (CCN) (Wiedensohler et al., 2009; Ehn et al., 75 2014). This will, in turn, impact the atmospheric aerosol burden, precipitation and water 76 77 circulation, solar radiation budget, and climate (Ramanathan et al., 2001). However, the extents of those influences are not well understood so far, due to the high uncertainties 78 79 associated with the formation and physical and chemical properties of SOA (Shrivastava et al., 2017). Large gaps still exist in SOA mass loading and properties between models 80 and observations (Gentner et al., 2017; Ervens et al., 2011; Hayes et al., 2015). Therefore, 81 82 it is crucial to explore and resolve this issue to improve our knowledge of the roles of SOA 83 in the environment, human health, and climate.

Gas-particle partitioning of semi-volatile and low-volatile organic compounds 84 (SVOCs and LVOCs) generated from VOC oxidation is an important pathway of SOA 85 formation. In most current chemical transport models (CTMs), this process is treated as an 86 equilibrium partitioning process that depends on the mass concentration of the organic 87 particulate matter (OPM), ambient temperature (T), the mean molecular weight of OPM, 88 and the volatility of condensed organics (Pankow, 1994). The formation of condensed 89 90 organic products is commonly represented by lumped surrogate SVOCs in a 2-product 91 model with volatilities and SVOC yields fitted to chamber experiments (Odum et al., 1996). 92 To better represent the volatility of primary organic aerosol (POA) and the multi-generation oxidation of SVOCs to a wider range, Donahue et al. (2006) proposed the volatility basis 93 set (VBS) model in which the mass yields of SVOCs are fitted to a fixed number of 94 volatility bins (usually $0.01-10^5 \ \mu g \ m^{-3}$). The VBS model has been adopted by several 95 CTMs (such as WRF-Chem, GEOS-Chem, etc.). 96

Although the models can capture the general trend of SOA evolution and mass 97 98 concentration to some extent (Li et al., 2017a; Bergström et al., 2012; Woody et al., 2016), two key factors currently neglected in models may lead to biases: 1) the molecular 99 100 structures and interactions of functional groups (-OH, -C=O, -COOH, etc.) of condensed organics (non-ideality); 2) partitioning of water vapor, the most abundant atmospheric 101 constituent besides O₂ and N₂, to OPM. The non-ideality alters the volatility of condensed 102 organics, and thus their contributions to the total SOA mass loading (Cappa et al., 2008). 103 104 Water partitioning into OPM can reduce the partial pressure of organics due to Raoult's Law effect (Prisle et al., 2010) and lead to increases in SOA mass. The amount of aerosol 105 liquid water associated with organics (ALW_{org}) may vary for different precursors (Healy et 106 al., 2009; Prisle et al., 2010). The above two aspects will not only affect the chemical 107 composition of SOA but also the inorganic portion (Ansari and Pandis, 2000) and optical 108 109 properties (Denjean et al., 2015) of aerosols.

Laboratory and field studies have observed water absorbed by SOA from a variety of 110 precursor VOCs (Lambe et al., 2011; Zhao et al., 2016b; Asa-Awuku et al., 2010; 111 Varutbangkul et al., 2006). The hygroscopicity of SOA, quantitatively described by the 112 hygroscopicity parameter, κ , is correlated with the oxygen-to-carbon ratio (O:C) and 113 increases with more oxidized SOA during photochemical aging (Lambe et al., 2011;Zhao 114 et al., 2016b). The OPM-associated water can be estimated using the κ-Köhler theory under 115 116 the Zdanovskii-Stokes-Robinson (ZSR) assumption of no interactions between any 117 constitutes in aerosols (Petters and Kreidenweis, 2007). The total water content is the summation of water associated with each solute at the same water activity. Guo et al. (2015) 118 119 found that this simplified method, along with the ISORROPIA model which is used to 120 predict aerosol liquid water (ALW) associated with the inorganic portion of aerosols, could reproduce the observed total ALW in the ambient environment. Pye et al. (2017) found that 121 the modeled organic aerosol (OA) improved significantly but biased high at nighttime 122 when ALWorg is included in the calculation. However, as the interaction among organic 123 124 species and between organics and water in the organic-water mixture has been shown to 125 play an important role in SOA formation and water partitioning to OPM (Kim et al., 2019), 126 the ALW_{org} estimated by the κ-Köhler theory and its impact on SOA might not be accurate. Using UNIversal Functional Activity Coefficient (UNIFAC) method (Fredenslund et al., 127 1975) for calculating activity coefficients of the organic-water mixture, it was found that 128 in the eastern U.S., where biogenic SOA dominated the OA, considering ALWorg leads to 129 130 a significant increase in predicted SOA (Pankow et al., 2015; Jathar et al., 2016). China has been suffering from severe PM_{2.5} pollution especially in the eastern region 131

with fast urbanization and economic development (Fu and Chen, 2017). SOA is a very
important component of PM_{2.5} in China that contributes about 20-50% (Li et al., 2017b).
The fraction of SOA in OA increases during haze events (Huang et al., 2014; Sun et al.,
2019). Previous modeling studies in China indicate that SOA was underpredicted (Lin et

al., 2016; Jiang et al., 2012) and the impacts of the non-ideality and water-OPM partitioning
on modeled SOA have not been evaluated.

In this study, regional simulations of SOA during January and July of 2013 over 138 eastern China were conducted to investigate the seasonal variation of SOA due to water 139 partitioning into OPM. The model performance was evaluated against observed 140 meteorological parameters (temperature and relative humidity, RH) as well as PM_{2.5}, 141 142 organic carbon (OC), and OA at ground monitoring sites. The regional and seasonal 143 impacts on SOA, ALW_{org}, and properties of aerosols were quantified. Lastly, influences of the results by several factors including meteorological parameters, estimations of saturation 144 vapor pressures (SVP) of condensed organics, and the individual impacts of ALW_{org} and 145 146 non-ideality of the organic-water mixture on SOA prediction were analyzed.

147 **2 Methodology**

148 **2.1 Model description**

The Community Multi-scale Air Quality model (CMAQ v5.0.1) coupled with a modified 149 150 SAPRC-11 was used in this study. Model configurations were largely based on that used 151 by Hu et al. (2016) as summarized below. Firstly, SAPRC-11 was expanded for a more 152 detailed treatment of isoprene oxidation and tracking dicarbonyl products (glyoxal and methylglyoxal) from different groups of major precursors (Ying et al., 2015). Secondly, 153 SOA from isoprene epoxydiols (IEPOX), methacrylic acid epoxide (MAE) and dicarbonyls 154 through surface-controlled irreversible reactive uptake were added (Hu et al., 2017; Li et 155 156 al., 2015; Liu et al., 2020; Ying et al., 2015). Thirdly, the heterogeneous formation of secondary nitrate and sulfate from NO2 and SO2 reaction on the particle surfaces (Ying et 157 158 al., 2014) were added, which is an important source of secondary inorganic aerosols (Zheng et al., 2015) and improves model estimates of nitrate and sulfate (Qiao et al., 2018; Shi et 159 al., 2017). Fourthly, SOA yields were corrected for vapor wall loss (Zhang et al., 2014). 160 Impacts of the above updates on model performances have been extensively discussed in 161 the cited work and will not be further investigated in the current study. 162

163 The SOA module mostly follows Pankow et al. (2015). Two types of SOA as traditionally treated in CMAQ were considered, "semi-volatile" (SV) portion that formed 164 via equilibrium absorption-partitioning of SVOCs, and "non-volatile" (NV) portion that 165 includes the oligomers and SOA formed via direct oxidation of aromatics at low-NOx. SOA 166 from dicarbonyls, IEPOX, and MAE were formed by irreversible reactive uptake and 167 categorized as NV-SOA in the current model as well. Some studies investigated SOA from 168 169 glyoxal, methylglyoxal, and IEPOX using detailed reactions and reversible pathways in 170 models or observed as reversible processes in chamber experiments, leading to a relatively lower SOA yield compared to the surface-controlled irreversible uptake (Lim et al., 2013; 171 Knote et al., 2014; Galloway et al., 2009; El-Sayed et al., 2018; Budisulistiorini et al., 2017). 172 173 The non-volatile assumption used in this paper allows an upper-limit estimation of the importance of these additional SOA formation pathways. POA was treated as non-volatile 174 and non-reactive. The mass distribution of SVOCs between the gas-phase and particle-175 phase follows the equation: 176

$$K_{p,i} = \frac{F_i}{M \cdot A_i} \tag{Eq1}$$

177 where $K_{p,i}$ $(m^3 \mu g^{-1})$ is the gas-particle partitioning constant for compound i, $F_i(\mu g m^{-3})$ 178 is the concentration of species i in the particle-phase, A_i $(\mu g m^{-3})$ is the concentration of 179 species i in the gas-phase, and M $(\mu g m^{-3})$ is the total mass concentration of the absorbing 180 organic phase (i.e. OPM). The gas-particle partitioning constant $K_{p,i}$ is dependent on the 181 chemical composition of OPM. Pankow et al. (1994) derived $K_{p,i}$ for SVOCs partitioning 182 into an absorbing organic phase as:

$$K_{p,i} = \frac{RT}{10^6 \overline{MW} \xi_i p_{L,i}^o} \tag{Eq2}$$

183 where $p_{L,i}^{0}(\text{atm})$ is the SVP of the pure compound i at temperature T(K), ξ_i is the activity 184 coefficient of species i in the absorbing organic phase, $\overline{MW}(\text{g mol}^{-1})$ is the average 185 molecular weight of OPM, R (8.314 J mol⁻¹ K⁻¹) is the gas constant, and 10⁶ is used to 186 convert the units to $m^3 \mu g^{-1}$. 187 There are 12 lumped SVOCs generated by oxidation of alkanes, alkenes, and aromatics under different NO_x conditions and 8 NV organic products as listed in Table S1 188 189 and Table S2. More details about the lumped precursors such as formation conditions 190 ("high" and "low" NO_x), lumping species and method, and yields from parent VOCs can be found in Carlton et al. (2010) and summarized in SI. Activity coefficients of SVOCs 191 192 were calculated based on the composition of OPM using the UNIFAC method, with assigned carbon number (nc), functional groups and energy interaction parameters to both 193 194 SV and NV compounds (Pankow et al., 2015). The UNIFAC model is one of the commonly used models that activity coefficients of condensed organics and their interactions with 195 196 water can be estimated. This method has been adopted to investigate the impacts of non-197 ideality and water partitioning into OPM on SOA for different precursors in box models (Seinfeld et al., 2001; Bowman and Melton, 2004) and CTMs (Pankow et al., 2015; Kim 198 et al., 2019). In the current model, POA was assumed to have a bulk composition of ten 199 200 categories of surrogate species (Table S3), as used by Li et al. (2015). POA is also involved 201 in the calculation of activity coefficients for the organic-water mixture. Detailed 202 information about the surrogate species including their structures and properties can be 203 found in Li et al. (2015) and references therein.

In addition to organic compounds, water partitioning into OPM is enabled according 204 205 to Eq 1 and Eq 2. In such a case, the absorbing phase in Eq 1 includes both organic aerosols and water associated with OPM. As water condenses in the absorbing organic phase, it will 206 207 further alter the molar fraction of each composition, the activity coefficient of SVOCs and the SV-SOA mass concentrations as a result. In the current model, we assumed no 208 209 interactions between the inorganic and organic phases.

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2.2 Estimating the hygroscopicity of OA

Based on the κ -Köhler theory with linearly additive hygroscopic behavior of each 211 212 component of the mixed particle, ALWorg is related to the hygroscopicity parameter for the 213 organic mixture (κ_{org}) by Eq3 (Petters and Kreidenweis, 2007):

$$ALW_{org} = \rho_w V_{org} \kappa_{org} \frac{a_w}{1 - a_w}$$
(Eq3)

where ρ_w is the density of water (assumed to be 1 g cm⁻¹), V_{org} is the volume concentration of organics, and a_w is the water activity (assumed to be the same as RH). Since ALW_{org} in this study is calculated mechanistically using the partitioning theory, κ_{org} can be estimated by rearranging Eq3:

$$\kappa_{org} = \frac{ALW_{org}}{\rho_w V_{org}} \times \frac{1 - a_w}{a_w}$$
(Eq4)

218 V_{org} can be estimated from the modeled mass concentration of OA, assuming the density 219 of OA to be 1.2 g cm⁻³ (Li et al., 2019).

In many studies, κ_{org} is assumed to increase linearly with the oxidation state of OA, expressed as the O:C ratio (Massoli et al., 2010;Duplissy et al., 2011;Lambe et al., 2011). The correlation of κ_{org} and O:C ratio at 9 representative cities was evaluated during January and July of 2013, with the reduced major axis regression method (Ayers, 2001). The O:C ratio was calculated using Eq5:

$$0:C = \sum_{i=1}^{n} f_i (0:C)_i$$
 (Eq5)

where f_i and $(O:C)_i$ are the molar fraction and O:C ratio of organic aerosol component i. For POA, a fixed molar fraction and composition were assumed following Li et al. (2015). For SOA, the O:C ratio was calculated by using their organic matter to organic carbon ratio (OM:OC) following Simon and Bhave (2012):

$$O:C = \frac{12}{15} (OM:OC) - \frac{14}{15}$$
(Eq6)

The OM:OC ratio of each SOA component follows Pankow et al. (2015) as shown in TableS1-S2.

231 2.3 Model application

The simulation domain has a horizontal resolution of $36 \text{ km} \times 36 \text{ km} (100 \times 100 \text{ grids})$ and a vertical structure of 18 layers up to 21 km, which covers eastern China as shown in Figure S1. Anthropogenic emissions were generated from the Multi-resolution Emission

Inventory for China (MEIC) (Zhang et al., 2009) v1.0 with a $0.25^{\circ} \times 0.25^{\circ}$ resolution 235 (http://www.meicmodel.org) for China, and the Regional Emission inventory in Asia 236 version 2 (REAS2) (Kurokawa et al., 2013) with a $0.25^{\circ} \times 0.25^{\circ}$ resolution 237 (http://www.nies.go.jp/REAS/) for the rest of the domain. Biogenic emissions were 238 generated by the Model for Emissions of Gases and Aerosols from Nature (MEGAN) v2.1, 239 with the leaf area index (LAI) from the 8-day Moderate Resolution Imaging 240 Spectroradiometer (MODIS) LAI product (MOD15A2) and the plant function types (PFTs) 241 242 from the Global Community Land Model (CLM 3.0). Open biomass burning emissions were generated from the Fire INvnetory from NCAR (FINN) (Wiedinmyer et al., 2011). 243 Dust and sea salt emissions were generated online during CMAQ simulations. The total 244 245 emissions of major SOA precursors and their spatial distributions are shown in Table S4 and Figure S2. Meteorological fields were generated using the Weather Research and 246 Forecasting (WRF) model v3.6.1 with initial and boundary conditions from the NCEP FNL 247 248 Operational Model Global Tropospheric Analyses dataset. More details about the model 249 application can be found in Hu et al. (2016).

250 Four scenarios are investigated in this study. The base case (BS) applies the default 251 secondary organic aerosol module of CMAQ v5.0.1. In this case, no water partitioning into OPM is considered. Lumped semi-volatile products from the oxidation of various 252 precursors partition into a single organic phase, which is considered as an ideal mixture of 253 254 POA and SOA with $\gamma_{ora}=1$. The water case (S1) includes water partitioning into OPM, which is again considered as an ideal solution ($\gamma_{org}=1$ and $\gamma_{H2o}=1$). The UNIFAC case 255 (S2) considers the interaction between organic constituents with UNIFAC calculated 256 activity coefficients ($\gamma_{org} \neq 1$) but does not allow water partitioning into OPM. The 257 combined case (S3) allows both water partitioning and interactions between all constituents 258 (including water and organics) using UNIFAC calculated activity coefficients ($\gamma_{org} \neq 1$ 259 and $\gamma_{H2o} \neq 1$). The results of BS and S3 are used to examine the overall impacts of water 260 261 partitioning into OPM and polarity of organics on SOA and ALWorg, as shown in Section

3.1-3.4. The separate influences of those two processes on SOA from S1 and S2 arediscussed in Section 3.5.

264 3 Results and discussion

265 3.1 Model evaluation

The meteorological inputs and emissions have been used in several previous publications. 266 267 Model performance on meteorological parameters (temperature and RH), gaseous species 268 and gas and aerosol concentrations have been extensively evaluated (Hu et al., 2016; Hu et 269 al., 2017; Qiao et al., 2018; Shi et al., 2017). A summary of the model performance related to this study is provided below. Observed meteorological data were obtained from the 270 271 National Climatic Data Center (ftp://ftp.ncdc.noaa.gov/pub/data/noaa). Observations of 272 OC at two urban locations, Beijing (Cao et al., 2014; Wang et al., 2015) and Guangzhou (Lai et al., 2016) and OA in Beijing (Sun et al., 2014) during January of 2013 as well as 273 surface PM_{2.5} at several monitoring sites during July of 2013 from China National 274 Environmental Monitoring Center (http://113.108.142.147:20035/emcpublish/) were used 275 276 to evaluate model estimates of aerosols. Details of measurement methodology and 277 uncertainties of observations are listed in the corresponding references.

278 Temperature and RH are the two meteorological factors that affect SOA formation. Table 1 lists model statistics of mean observation (OBS), mean prediction (PRE), mean 279 bias (MB), gross error (GE) and correlation coefficient (R) based on WRF and observations 280 at monitoring sites located in 8 sub-regions of the domain (Figure S1) during January and 281 282 July of 2013. The benchmarks for the MM5 model (another meteorology model) of 4-12km horizontal resolution suggested by Emery et al. (2001) are also listed in the table. Details 283 284 of monitoring sites in the 8 sub-regions are listed in Table S5. Overall, WRF tends to underestimate both temperature and RH. The model shows better agreement with observed 285 temperature as R is higher than that of RH. Both temperature and RH are well captured by 286 287 the model in YRD, the Pearl River Delta (PRD), and the central regions of China (the major regions of eastern China). In these regions, MB and GE of temperature are -1.2~0.7 K and
1.8~2.6 K, respectively, which are -11.8~5.6% and 9.2~16.8% for RH, respectively.

290 Model estimates of daily organic carbon (OC) from case S3 are compared with 291 measurements at monitoring sites in Beijing and Guangzhou in January of 2013 (Figure 1a). The factors used to convert SOA to OC (SOC) are listed in Table S1-S2. OC from 292 293 POA (POC) is directly predicted by the model. Generally, the ratio between modeled and 294 observed OC concentration falls in the range of 1:2 to 2:1, with an R-value of 0.7. The 295 model tends to underestimate OC on high concentration days. Overall, the mean fractional 296 bias (MFB) and mean fractional error (MFE) of OC are -0.20 and 0.27, within the criteria $(MFB \le \pm 0.6; MFE \le 0.75)$ suggested by EPA (2007). The bias in OC might be due to 297 298 underestimated POA emissions and underpredicted SOA in CMAQ from missing precursors (Hu et al., 2017; Zhao et al., 2016a). No significant differences in OC are 299 300 observed in S3 compared to BS (not shown), likely due to the biased-low SOA predicted in the current model that limits the impact of ALW_{org} on SOA formation. 301

302 The underestimate of SOA can be seen from Figure 1b as well. CMAQ can well capture the observed diurnal variation of OA in Beijing during wintertime, except for the 303 304 underestimates of peak values. The correlation coefficient of modeled to observed OA is 0.55. We find a 25% underestimate of OA on average. Better agreement between the model 305 and observations is shown on non-polluted days (daily-averaged concentration less than 75 306 $\mu g m^{-3}$). The mean fractional bias (MFB) and mean fractional error (MFE) of polluted days 307 308 are -0.38 and 0.64, respectively, which are worse than that of the non-polluted days (-0.26 for MFB and 0.52 for MFE). The overall MFB and MFE of OA during January are -0.28 309 310 and 0.54, within the criteria (MFB $\leq \pm 0.6$; MFE ≤ 0.75) suggested by EPA (2007). Again, no apparent changes of SOA nor OA are observed between case S3 and BS (not shown), 311 since POA is predicted to be the primary contributor to OA at Beijing in winter in the 312 current model, with an averaged SOA/POA ratio of 0.12. This ratio is much lower than the 313 field observation of about 0.45-1.94 (Zhao et al., 2019; Sun et al., 2013; Sun et al., 2016). 314

The bias might be due to the missing SOA converted by partitioning and aging of semivolatile POA as well as oxidation from intermediate volatile organic compounds (IVOCs) and VOC oxidation products. Those pathways are shown to be important for SOA formation by modeling, field and chamber studies (Hodzic et al., 2010; Jimenez et al., 2009; Murphy et al., 2017; Robinson et al., 2007; Shrivastava et al., 2008; Tkacik et al., 2012; Zhao et al., 2014; Zhao et al., 2016a).

A sensitivity test was performed by using the newest CMAQ model version 5.3.1 that includes all the above processes in the aerosol module. The SOA/POA ratio in Beijing is improved greatly to be 0.83 in winter. However, high uncertainties still exist in the emissions of the involved precursors and characterization of SOA formation through these processes, needing further constrains by observations. Their influences on water partitioning into OPM and non-ideality of the organic-water mixture on SOA will be evaluated in a future study.

Due to the lack of observed OC and OA in July of 2013, as an alternative, model 328 329 performances are evaluated by comparing predicted and observed PM_{2.5} at ground sites (Figure S1) as shown in Figure S3. Generally, the model can well reproduce the diurnal 330 331 variation of PM_{2.5} in most regions. Predicted PM_{2.5} on highly concentrated days is biased low, especially in the North China Plain (NCP). NCP has the highest PM2.5 ranging from 332 60 μ g m⁻³ to 300 μ g m⁻³. The bias in modeled PM_{2.5} is significant in cities in Northwest. 333 This might be due to missing dust emissions in the current inventory (Hu et al., 2016). To 334 335 further evaluate the model performance, averaged MFB and MFE of modeled PM_{2.5} are plotted against observations of each site as shown in Figure S4. The criteria of MFB and 336 337 MFE followed recommendations by Boylan and Russell (2006). Our model performs well since most of the predictions meet the criteria and a large fraction (>58%) meet the goal. 338 The averaged MFB and MFE across all the sites are -0.28 and 0.39, indicating slightly 339 340 underestimate of PM_{2.5} by the model.

341 3.2 Impacts on SOA and ALW_{org}

The spatial distribution of SOA varies greatly in the two seasons. In winter, SOA is 342 relatively high in eastern SCB and the central and eastern provinces of Shandong, Henan, 343 Anhui, and Hubei (Figure 2 and Figure S5). Monthly-averaged SOA concentrations in the 344 above areas are up to 25 and 15-20 µg m⁻³, respectively. Anthropogenic emissions are the 345 major sources of SOA (Figure S6), such as dicarbonyl products from the oxidation of 346 xylene and toluene (Hu et al., 2017). In summer, surface SOA is high in Northeast, NCP, 347 348 and YRD. The highest SOA occurs in Shanghai and Jiangsu provinces as well as the coastal area of the Yellow Sea, with the value of ~9-16 μ g m⁻³ at the surface and ~20-25 mg m⁻² in 349 the column (col-SOA) of the atmosphere below 21 km (Figure S5). Different from winter 350 351 SOA, a significant fraction of summer SOA is originated from biogenic emissions (Figure S7). Anthropogenic SOA is highly concentrated in the coastal areas of the Yellow Sea and 352 Bohai Bay. 353

Combined effects of water partitioning into OPM and non-ideality on SOA formation 354 also exhibit strong seasonal variation. In winter, SOA is slightly decreased by $1.5 \ \mu g \ m^{-3}$ 355 (10-20%) on average at the surface (Figure 2) and less than $\sim 1 \text{ mg m}^{-2}$ (20%) in the column 356 (Figure S5) over high SOA regions where anthropogenic sources dominate. We show later 357 that the decrease of SOA is mainly due to the large activity coefficients of SVOCs which 358 decrease K_{ni} . In summer, higher temperature and RH promote water partitioning and SOA 359 formation so that SOA increases apparently over the entire domain, with the highest 360 enhancement of 2-4 μ g m⁻³ (20-50%) at the surface (Figure 2) and 4-6 mg m⁻² (30-60%) in 361 the column (Figure S5) over YRD and the coastal area of Yellow Sea. Anthropogenic SOA 362 dominates the total change in winter as shown in Figure S6. In summer, the increase of 363 SOA is attributed to biogenic sources in inland areas and anthropogenic sources over the 364 ocean (Figure S7). 365

Regional distribution of ALW_{org} is similar to the change of SOA as shown in Figure 367 3. In winter, a maximum averaged concentration of 3.0 µg m⁻³ for ALW_{org} occurs in the 368 high SOA region, where significant changes of SOA also occur. In other areas, the averaged

concentration of ALW_{org} is about 0.5-1.5 µg m⁻³. Overall, the average ratio of ALW_{org} to 369 SOA is about 0.1-0.3 in winter. In summer, water partitioning into OPM mostly occurs in 370 the east coastal area at the surface where a significant increase of anthropogenic SOA (such 371 as those from toluene and xylenes) is observed. This might be due to the high polarity of 372 anthropogenic SVOCs (having more -COOH groups) that absorb more water. In the coastal 373 area, the averaged concentration of ALW_{org} is about 5-7 μ g m⁻³, with the ALW_{org}/SOA 374 375 ratio of 0.5-1.0. Over the land, the averaged concentration of ALW_{org} is about 1-3 μ g m⁻³ 376 (ALW_{org}/SOA ratio of 0.2-0.5) in Northeast and East China. Water partitioning is mostly associated with biogenic SOA originated from isoprene and monoterpenes oxidation that 377 378 produces SVOCs with abundant OH group.

379 Based on the column concentrations of ALWorg and ALWorg/SOA ratio (Figure S8), 380 in winter, more ALW_{org} must have occurred in the south and southwest regions at higher levels where a significant increase of col-SOA occurs (Figure S5). The averaged col-381 ALW_{org}/col-SOA ratio in the high SOA area is 0.1-0.3. In summer, the ALW_{org} must be 382 383 high at higher altitudes over the central regions in China. The maximum col-ALW_{org} is about 7 mg m⁻² over YRD, with the col-ALW_{org}/col-SOA ratio of about 0.3. 384

385

3.3 Impacts on aerosol properties

Since ALW_{org} is determined in S3, the values of κ_{org} can be estimated from the modeled 386 ALW_{org}, OA and RH using Eq 4. 9 representative cities were selected to investigate the 387 388 relationship of κ_{org} vs. O:C and its seasonal variation as shown in Figure S9 and S10. 389 The results of all the cities in winter and summer are merged and analyzed as shown in 390 Figure 4. Pairs of κ_{org} and O:C data are grouped into 10 O:C bins and the averaged κ_{org} 391 in each bin are then calculated. Overall, the estimated O:C ratio is within the range of 0.2-392 0.8. The averaged κ_{org} in each O:C bin is less than 0.1 in winter, with the highest value in Guangzhou. As more ALW_{org} is formed in summer, the averaged κ_{org} also increases 393 394 greatly with the highest value of 0.35 in Beijing. The linear correlation between κ_{org} and O:C shows significant spatial and seasonal variations. For example, the slope of κ_{org} -O:C 395

396 is 70-90% smaller in winter than in summer in the Northern cities such as Shenyang, Beijing, Zhengzhou, and Xi'an. However, in Guangzhou, the slope is 83% higher in winter 397 than in summer. In Chengdu, the slope is quite similar in both seasons. Overall, the slope 398 of κ_{org} vs. O:C in the 9 cities is 0.16 in winter and 0.40 in summer. Most of the fitted 399 linear correlations of the individual city fall outside of the range of 0.18 to 0.37 suggested 400 in previous studies (Duplissy et al., 2011;Lambe et al., 2011;Massoli et al., 2010;Chang et 401 402 al., 2010), indicating that the hygroscopicity of organic aerosols cannot be simply 403 represented by a single parameter such as the O:C ratio (Rickards et al., 2013). In both seasons, κ_{ora} approaches zero and negative values as O:C decreases, which might be due 404 405 to the linear regression of κ_{org} and O:C. To avoid this, an exponential fitting of the two variables is performed so that κ_{ora} falls in the range of (0,1) and is positively correlated 406 with O:C. In this case, the fitted correlations are $\kappa_{org} = 1 - \exp(-(O:C/1.88)^{2.29})$ and $\kappa_{org} = 1 - \exp(-(O:C/1.88)^{2.29})$ 407 exp(-(O:C/1.06)^{4.50}) for January and July of 2013, respectively. 408

The impacts on aerosol optical depth (AOD) and aerosol radiative forcing (ARF) are further investigated. Figure 5 shows the monthly-averaged AOD at 550 nm in January and July of 2013. It is calculated by summarizing the product of model estimated extinction coefficient of fine particles ($b_{ext,i}$) multiplied by the thickness (HL_i) in each layer:

$$AOD = \sum_{i=1}^{N} b_{ext,i} \times HL_i$$
 (Eq6)

where N is the number of layers. There are two methods to estimate the aerosol extinction coefficient in CMAQv5.0.1. One is based on the Mie theory and the predicted aerosol component concentrations ($b_{ext,m}$), and the other is based on correlation from the IMPROVE monitoring network that considers the impacts of hygroscopicity of different aerosol components ($b_{ext,r}$) (Malm et al., 1994). AOD calculated with the two types of extinction coefficient are denoted as AOD_m and AOD_r, respectively.

In Figure 5, a clear pattern of high AOD_r in SCB and NCP and low AOD_r in west China is observed in both winter and summer, consistent with previous studies (He et al.,

2019; He et al., 2016; Luo et al., 2014). An identical trend in AOD_m is shown in Figure 421 S11. The monthly-averaged AOD_r ranges from 1.0 to 3.2 in January and from 0.3 to 0.9 in 422 423 July. AOD_m is lower than AOD_r, falling in 0.7-2.2 in January and 0.3-0.6 in July. The model significantly overestimates AOD in January but agrees better with observations from 424 MODIS where AOD is high in July (Figure S12). The bias in the predicted AOD might be 425 partially due to the empirical equation applied in the calculation of AOD in CMAQ (Wang 426 427 et al., 2009; Liu et al., 2010), and partially due to the uncertainties of fine AOD overland 428 from MODIS data (Wang et al., 2009; Levy et al., 2010). The increase of AOD due to ALWorg shows a strong spatial and seasonal pattern. In winter, there are no significant 429 changes in AODr across the whole domain, due to insignificant changes of SOA. In summer, 430 431 AOD_r increases significantly in YRD and the adjacent areas by up to 10%.

ARF represents the change in the radiative flux at the top of the atmosphere due to 432 aerosols. An offline version of the Shortwave Radiative Transfer Model For GCMs 433 (RRTMG_SW) was used to calculate the direct radiative effect of aerosols on shortwave 434 435 radiation (Iacono et al., 2008). Generally, fine aerosols exhibit cooling effects on the shortwave radiation in both winter and summer over the entire domain as shown in Figure 436 437 6. This impact is much stronger in the areas where AOD is high (Figure 5). The ARF is highest in Shandong in winter and in the coastal area near Jiangsu province in summer, 438 which are about -5 W m⁻² and -6 W m⁻², respectively. In winter, no significant changes of 439 ARF are observed in eastern China (Figure 6b). This is likely attributed to an insignificant 440 441 contribution of SOA to PM_{2.5} in winter compared to other components with cooling effects, such as sulfate. In summer, SOA is an important component of PM2.5 (20-60%), and the 442 effects of water partitioning on shortwave radiation are relatively stronger. An 443 enhancement of up to 15% in the cooling effects of ARF occurs near the YRD region where 444 AOD significantly changes as well. 445

446 **3.4 Sensitivity to T, RH, and SVP**

Meteorological conditions and SOA precursors affect the impacts of water partitioning on 447 SOA. To better illustrate the dependency of SOA on temperature, RH, and SVP of SVOCs, 448 an offline calculation of SOA formation was performed at two representative cities (Jinan 449 450 in NCP during winter and Nanjing in YRD during summer) when the daily maximum SOA increases occurred. We assume temperature (T) and water vapor mixing ratio (QV) to be 451 within the range of $\overline{X} \pm \sigma$, where \overline{X} and σ are the mean and standard deviation 452 453 calculated based on WRF predictions at each location. We choose 10 evenly distributed values for T and QV within the range of $\overline{X} \pm \sigma$. The temperature dependence parameter of 454 SVP ($\Delta H/R$) is also scaled by 0.2, 0,8, 1.4, and 2.0 separately for all the SVOCs. As shown 455 456 in Figure 7, SOA exhibits a negative correlation with T and a positive correlation with QV 457 in both cities. SOA is more sensitive to QV under cold conditions (Jinan) and to temperature under hot conditions (Nanjing). When the temperature is fixed, the sensitivity 458 of SOA to $\Delta H/R$ is different in the two cities. We find more changes in SOA across $\Delta H/R$ 459 in Jinan. This is attributed to the temperature correction factor (ζ_{corr}) of K_p in CMAQ as 460 defined below: 461

$$\zeta_{corr} = \frac{K_{p,T_{ref}}}{K_{p,T}} = \frac{T_{ref}}{T} \exp\left[\frac{\Delta H}{R} \left(\frac{1}{T_{ref}} - \frac{1}{T}\right)\right]$$
(Eq8)

where T_{ref} is the reference temperature (298K) and *T* is temperature. According to Figure 7, the range of T is 265-274K in Jinan and 300-307K in Nanjing. The deviation of temperature from the reference value (298K) is greater in Jinan than in Nanjing. Therefore, the unit change of $\Delta H/R$ causes greater variations of ζ_{corr} and thus K_p in Jinan. As a result, SOA is more sensitive to $\Delta H/R$. The impacts of SVP estimation on SOA are more significant in winter than in summer.

468 **3.5 Separate impacts of water partitioning/polarity of condensed organics**

Impacts of water partitioning into OPM and non-ideality of the organic-water mixture on SOA are in opposite directions. Water partitioning increases SOA by ~10-20% in winter and ~30-80% in summer in most areas of the domain (Figure 8). This is because the 472 molecular weight of water is quite small and will reduce the molar averaged weight of OPM (\overline{MW}) in Eq2 (Pankow et al., 2015). The reduced \overline{MW} further increases K_p and 473 promotes mass transfer of SVOCs from the gas-phase to OPM. On the other hand, by 474 475 considering non-ideality of the organic-water mixture, activity coefficients of SVOCs are 476 usually greater than 1.0 in this study, leading to a decrease in K_p . As a result, the total SOA 477 concentration is reduced by ~10-20% in winter and ~10-50% in summer (Figure 9). Overall, 478 the final impacts are the combined consequences of the two processes. In winter, the increase of SOA caused by water partitioning is offset by the decrease of SOA due to the 479 polarity of SVOCs in most areas of the domain, resulting in slight decreases of SOA. In 480 summer, the effect of water partitioning overcomes that of SVOC polarity so the total SOA 481 482 loading increases. This further leads to an enhanced attenuation of shortwave solar radiation and cooling of the atmosphere. 483

484 **4 Conclusions**

485 The WRF/CMAQ model was used to investigate the impacts of water partitioning into 486 OPM and non-ideality of the organic-water mixture on SOA formation over eastern China 487 during January and July of 2013. SOA is greatly enhanced in summer especially in YRD and over the Yellow Sea by up to 50% and 60% at the surface and in the whole column, 488 respectively. No significant impacts on SOA are observed in winter. This might be due to 489 490 the underestimation of SOA in the current model. ALW_{org} is highly correlated with the change of SOA, with the ratio of ALWorg to SOA of 0.1-0.3 and 0.2-1.0 at the surface 491 492 where significant changes of SOA occur in winter and summer, respectively. By using the 493 modeled ALW_{org}, correlations between κ_{org} and O:C are examined in 9 representative 494 cities, showing significant spatial and seasonal variations. The increases in SOA lead to an 495 enhancement in the averaged AOD and the cooling effects of aerosols, by up to 10% and 15% respectively in summer. The model predicted SOA is sensitive to temperature and QV 496 497 in both seasons, with higher sensitivity to QV during winter and temperature during 498 summer. Estimation of SVP also affects modeled SOA, especially in a cold environment.

- The effects of water partitioning into OPM and non-ideality of the organic-water mixture 499 on SOA are the opposite. Since the activity coefficients of SVOCs are mostly greater than 500 1.0 during the simulated episode, SOA concentration decreases when the non-ideality 501 effect is considered. Averaged SOA concentration decreases by up to 20% in winter and 502 50% in summer. Water partitioning alone increases SOA by 10-20% in winter and 30-80% 503 in summer. It should be noticed that the results shown in this study are the lower limit as 504 505 the current model tends to underestimate SOA. It is crucial to consider both effects in 506 simulating SOA formation under hot and humid conditions in CTMs.
- 507

508 Data availability

509 Data used in this manuscript can be provided upon request by e-mail to the corresponding 510 authors, Qi Ying (qying@civil.tamu.edu) and Jianlin Hu (jianlinhu@nuist.edu.cn).

511

512 **Competing interests**

513 The authors declare that they have no conflict of interest.

514

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			Nort	heast	Z	GP	North	nwest	ΥF	Ð	Cen	tral	SC	B	PF	Ð	Sout	hwest	Benchmark
OBS 256.2 296.3 297.4 266.9 293.0 277.7 303.2 275.7 301.1 276.1 295.7 301.1 288.9 301.1 282.2 295.1 PRe 251.6 296.3 261.2 298.8 267.0 293.5 278.4 302.0 276.2 301.0 278.8 301.0 278.8 291.1 289.4 301.1 289.3 300.0 278.8 291.8 MB -4.6 0.1 -2.7 1.4 0.1 0.5 0.7 -1.2 0.5 -0.1 -2.6 -0.5 -1.0 -3.4 -3.4 -40.5 MB -4.6 0.1 -2.7 1.4 0.1 0.5 0.7 -1.2 0.5 -0.6 -1.0 -3.4 -3.4 -40.5 MB -7.5 80.3 74.5 72.1 60.2 69.1 79.3 71.8 6.8 74.4 77.0 80.6 79.3 77.9 MB 7.5 <td< th=""><th></th><th></th><th>Jan</th><th>Jul</th><th>Jan</th><th>Jul</th><th>Jan</th><th>Jul</th><th>Jan</th><th>Jul</th><th>Jan</th><th>Jul</th><th>Jan</th><th>Jul</th><th>Jan</th><th>Jul</th><th>Jan</th><th>Jul</th><th></th></td<>			Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	
PRE 251.6 296.3 261.2 298.8 267.0 293.5 278.4 302.0 273.5 293.1 288.9 300.0 278.8 291.8 MB -4.6 0.1 -2.7 1.4 0.1 0.5 -0.1 -2.6 -2.6 -0.5 -1.0 -3.4		OBS	256.2	296.2	263.9	297.4	266.9	293.0	277.7	303.2	275.7	301.1	276.1	295.7	289.4	301.1	282.2	295.1	
temperature (K) MB -4.6 0.1 -2.7 1.4 0.1 0.5 0.7 -1.2 0.5 -0.1 -2.6 -0.5 -1.0 -3.4	L. L	PRE	251.6	296.3	261.2	298.8	267.0	293.5	278.4	302.0	276.2	301.0	273.5	293.1	288.9	300.0	278.8	291.8	
VV GE 5.5 1.8 4.0 2.3 3.6 3.2 2.0 2.3 2.6 2.4 5.1 4.2 1.9 1.8 4.1 3.5 \leq 2.0 R 0.9 0.9 0.8 0.8 0.9 0.8 0.9 0.9 0.9 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.8 0.8 0.8 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.9 0.8 0.9 0.9 0.9 0.9 0.9 0.9 0.9 0.9 0.9 0.9	temperature (12)	MB	-4.6	0.1	-2.7	1.4	0.1	0.5	0.7	-1.2	0.5	-0.1	-2.6	-2.6	-0.5	-1.0	-3.4	-3.4	≤±0.5
R 0.9 0.9 0.8 0.8 0.9 0.8 0.9 0.6 0.9 0.8 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.9 0.8 0.19 0.19 0.19 11.0 11.6 16.8 11.7 22.8 11.6 10.4 10.4 Humidity R 0.5 0.8 0.4 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5		GE	5.5	1.8	4.0	2.3	3.6	3.2	2.0	2.3	2.6	2.4	5.1	4.2	1.9	1.8	4.1	3.5	≤2.0
OBS 77.5 80.3 74.5 72.1 60.2 69.1 79.3 71.3 76.3 71.8 68.5 74.4 77.0 80.6 70.3 77.9 Relative PKE 85.0 73.2 78.6 57.5 49.2 54.2 75.1 76.9 64.5 70.0 57.7 78.4 75.3 84.8 75.8 85.1 Humidity MB 7.5 -7.1 4.1 -14.6 -11.0 -15.0 -4.2 5.6 -11.8 -1.8 -10.8 4.0 -1.7 4.2 5.5 7.2 Humidity GE 11.3 14.6 16.6 19.8 19.1 11.0 11.6 16.8 11.7 22.8 11.6 9.5 9.2 75.4 10.4 N 0.5 0.8 0.4 0.8 0.9 0.6<		R	0.9	0.9	0.9	0.8	0.8	0.8	0.9	0.8	0.9	0.8	0.8	0.9	0.9	0.6	0.9	0.8	
Relative PRE 85.0 73.2 78.6 57.5 49.2 54.2 75.1 76.9 64.5 70.0 57.7 78.4 75.3 84.8 75.8 85.1 Humidity MB 7.5 -7.1 4.1 -14.6 -11.0 -15.0 -4.2 5.6 -11.8 -1.8 -10.8 4.0 -1.7 4.2 5.5 7.2 (%) GE 11.3 14.6 16.6 19.8 19.1 11.6 16.8 11.7 22.8 11.6 9.5 9.2 15.4 10.4 R 0.5 0.8 0.4 0.8 0.7 0.6 0.6 0.6 0.6 0.6 0.6 0.5 0.7 0.7 0.7		OBS	77.5	80.3	74.5	72.1	60.2	69.1	79.3	71.3	76.3	71.8	68.5	74.4	77.0	80.6	70.3	<i>9.17</i> .9	
Humidity MB 7.5 -7.1 4.1 -14.6 -11.0 -15.0 -4.2 5.6 -11.8 -1.8 -1.7 4.2 5.5 7.2 (%) GE 11.3 11.3 14.6 16.6 19.1 11.0 11.6 16.8 11.7 22.8 11.6 9.5 9.2 15.4 10.4 R 0.5 0.8 0.4 0.8 0.7 0.6 0.6 0.6 0.6 0.6 0.5 0.5 0.5 0.7 0.7 *OBS is mean observation, PRE is mean prediction, MB is mean bias, GE is gross error, R is the correlation coefficient of predictions vs. observations.	Relative	PRE	85.0	73.2	78.6	57.5	49.2	54.2	75.1	76.9	64.5	70.0	57.7	78.4	75.3	84.8	75.8	85.1	
(%) GE 11.3 14.6 16.6 19.8 19.1 11.0 11.6 16.8 11.7 22.8 11.6 9.5 9.2 15.4 10.4 R 0.5 0.8 0.4 0.8 0.7 0.6 0.6 0.6 0.6 0.6 0.5 0.5 0.7 0.7 *OBS is mean observation, PRE is mean prediction, MB is mean bias, GE is gross error, R is the correlation coefficient of predictions vs. observations.	Humidity	MB	7.5	-7.1	4.1	-14.6	-11.0	-15.0	-4.2	5.6	-11.8	-1.8	-10.8	4.0	-1.7	4.2	5.5	7.2	
R 0.5 0.8 0.4 0.8 0.7 0.6 0.6 0.6 0.2 0.6 0.5 0.7 0.7 *OBS is mean observation, PRE is mean prediction, MB is mean bias, GE is gross error, R is the correlation coefficient of predictions vs. observations. 0.5 0.5 0.7	(%)	GE	11.3	11.3	14.6	16.6	19.8	19.1	11.0	11.6	16.8	11.7	22.8	11.6	9.5	9.2	15.4	10.4	
*OBS is mean observation, PRE is mean prediction, MB is mean bias, GE is gross error, R is the correlation coefficient of predictions vs. observations.		R	0.5	0.8	0.4	0.8	0.3	0.7	0.6	0.6	0.6	0.6	0.2	0.6	0.6	0.5	0.5	0.7	
	*OBS is mean	i observ	ation, PI	RE is me	an predi	ction, M	B is mea	an bias, (GE is gr	oss error	, R is the	correla	tion coe	fficient o	of predic	tions vs	observ:	ations.	

Table 1. Statistical analysis of modeled temperature (K) and relative humidity (%) of January and July of 2013 at the monitoring sites in different geographical areas

**Northeast is Northeast China, NCP is North China Plain, Northwest is Northwest China, YRD is Yangtze River Delta, Central is Central China, SCB is Sichuan

***Benchmarks of MB and GE are for the MM5 model of 4-12 km horizontal resolution by Emery et al. (2001) Basin, PRD is Pearl River Delta, Southwest is Southwest China.



Figure 1. Comparison of (a) observed and modeled organic carbon concentration at University of Beihang (BH), Tsinghua University (TH) and Guangzhou (GZ); (b) observed organic aerosol (Obs-OA) at Beijing and predictions of total OA (pOA) and SOA (pSOA), unit is μ g m⁻³. Locations of monitoring sites are shown in Figure S1.



Figure 2. Monthly-averaged SOA from S3 and changes of SOA due to water partitioning into OPM and non-ideality of the organic-water mixture. "Abs. Diff." represents absolute differences (S3-BS); "Rel. Diff." represents relative differences ((S3-BS)/BS, %).



Figure 3. Monthly-averaged water partitioning into the organic-phase (ALW_{org}, μ g m⁻³) and the ratio to SOA (ALW_{org}/SOA) during January and July of 2013.



Figure 4. The correlation of hygroscopicity of organic aerosol (κ_{org}) and O:C ratio in 9 representative cities including Shenyang (SS), Beijing (BJ), Jinan (JN), Zhengzhou (ZZ), Xi'an (XA), Nanjing (NJ), Shanghai (SH), Chengdu (CD), and Guangzhou (GZ) in January (a) and July (b) of 2013. O:C ratios are categorized into 10 bins. In each bin, the ranges of O:C and κ_{org} are represented by bars. The mean values of O:C and κ_{org} are represented by bars. The mean values of O:C and κ_{org} are represented by the averaged RH of each bin. The relationship between κ_{org} and O:C is fitted by a linear function with reduced major axis regression (blue lines) and an exponential function (red lines), respectively. κ_{01} and κ_{07} represent the fitted correlation for January and July, respectively.



0 0.7 1.4 2.1 2.8 3.5 -0.04 -0.02 0 0.02 0.04 -9 -6 -3 0 3 6 9 **Figure 5**. Monthly-averaged AOD_r at 550 nm and changes of AOD_r due to water partitioning into OPM and non-ideality of the organic-water mixture. "Abs. Diff." represents absolute differences (S3-BS); "Rel. Diff." represents relative differences ((S3-BS)/BS, %).



Figure 6. Monthly-averaged shortwave direct aerosol radiative forcing at the top of the atmosphere from S3 and the relative changes due to water partitioning into OPM and non-ideality of the organic-water mixture during January and July of 2013. "Rel. Diff." represents relative differences ((S3-BS)/BS, %).



Figure 7. The sensitivity of SOA formation to temperature (TEMP), water mixing ratio, and the temperature dependence parameter of SVP ($\Delta H/R$) at Jinan (JN, first column) and Nanjing (NJ, second column).



Figure 8. Monthly-averaged impacts of water partitioning into OPM on SOA. "Abs. Diff." represents absolute differences (S3-S2); "Rel. Diff." represents relative differences ((S3-S2)/S2, %).



Figure 9. Monthly-averaged impacts of non-ideality of the organics-water mixture on SOA. "Abs. Diff." represents absolute differences (S3-S1); "Rel. Diff." represents relative differences ((S3-S1)/S1, %).