

1 **Impacts of water partitioning and polarity of organic compounds on**  
2 **secondary organic aerosol over Eastern China**

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28

29 **Abstract**

30 Secondary organic aerosol (SOA) is an important component of fine particular matter  
31 (PM<sub>2.5</sub>). Most air quality models use an equilibrium partitioning method along with the  
32 saturation vapor pressure (SVP) of semi-volatile organic compounds (SVOCs) to predict  
33 SOA formation. However, the models typically assume that the organic particulate matter  
34 (OPM) is an ideal mixture and ignore the partitioning of water vapor to OPM. In this study,  
35 the Community Multi-scale Air Quality model (CMAQ) is updated to investigate the  
36 impacts of water vapor partitioning and non-ideality of the organic-water mixture on SOA  
37 formation during winter (January) and summer (July) of 2013 over eastern China. The  
38 updated model treats the partitioning of water vapor molecules into OPM and uses the  
39 UNIFAC model to estimate the activity coefficients of species in the organic-water mixture.  
40 The modified model can generally capture the observed surface organic carbon (OC) with  
41 a correlation coefficient R of 0.7, and the surface OA with the mean fractional bias (MFB)  
42 and mean fractional error (MFE) of -0.28 and 0.54, respectively. SOA concentration shows  
43 significant seasonal and spatial variations, with high concentrations in the North China  
44 Plain (NCP), Central China and Sichuan basin (SCB) regions during winter (up to 25  $\mu\text{g m}^{-3}$ )  
45 and in the Yangtze River Delta (YRD) during summer (up to 16  $\mu\text{g m}^{-3}$ ). In winter,  
46 SOA decreases slightly in the updated model, with the monthly-averaged relative change  
47 of 10-20% in the highly concentrated areas, mainly due to organic-water interactions. The  
48 monthly-averaged concentration of SOA increases greatly in summer, by 20-50% at the  
49 surface and 30-60% in the whole column. The increase of SOA is mainly due to the increase  
50 in biogenic SOA in inland areas and anthropogenic SOA in coastal areas. As a result, the  
51 averaged aerosol optical depth (AOD) is increased by up to 10% and the cooling effect of  
52 aerosol radiative forcing (ARF) is enhanced by up to 15% over YRD in summer. The  
53 aerosol liquid water content associated with OPM ( $ALW_{\text{org}}$ ) at the surface is relatively high  
54 in inland areas in winter and over the ocean in summer, with the monthly-averaged  
55 concentration of 0.5-3.0 and 5-7  $\mu\text{g m}^{-3}$ , respectively. The hygroscopicity parameter ( $\kappa$ ) of

56 OA based on the  $\kappa$ -Köhler theory is determined using the modeled  $ALW_{org}$ . The  
57 correlation of  $\kappa$  with O/C ratio varies significantly across different cities and seasons.  
58 Analysis of two representative cities, Jinan (in NCP) and Nanjing (in YRD), shows that the  
59 impacts of water partitioning and non-ideality of the organic-water mixture on SOA are  
60 sensitive to temperature, relative humidity (RH), and the SVP of SVOCs. The two  
61 processes exhibit opposite impacts on SOA in eastern China. Water uptake increases SOA  
62 by up to 80% in the organic phase, while including non-unity activity coefficients decreases  
63 SOA by up to 50%. Our results indicate that both water partitioning into OPM and the  
64 activity coefficients of the condensed organics should be considered in simulating SOA  
65 formation from gas-particle partitioning, especially in hot and humid environments.

66 **Keywords:** SOA, non-ideality, water partitioning, hygroscopicity

## 67 **1 Introduction**

68 Secondary organic aerosol (SOA) is formed via a complex interaction of volatile organic  
69 compounds (VOCs) with oxidants and primary particles emitted from anthropogenic and  
70 biogenic sources in the atmosphere. As an important component of fine particulate matter  
71 ( $PM_{2.5}$ ), SOA can cause severe air pollution in urban and suburban areas (Huang et al.,  
72 2014) and exhibit adverse health effects (Atkinson et al., 2014). SOA also plays an  
73 important role in new particle formation and particle growth that further contribute to the  
74 enhancement of cloud condensation nuclei (CCN) (Wiedensohler et al., 2009; Ehn et al.,  
75 2014). This will, in turn, impact the atmospheric aerosol burden, precipitation and water  
76 circulation, solar radiation budget, and climate (Ramanathan et al., 2001). However, the  
77 extents of those influences are not well understood so far, due to the high uncertainties  
78 associated with the formation and physical and chemical properties of SOA (Shrivastava  
79 et al., 2017). Large gaps still exist in SOA mass loading and properties between models  
80 and observations (Gentner et al., 2017; Ervens et al., 2011; Hayes et al., 2015). Therefore,  
81 it is crucial to explore and resolve this issue to improve our knowledge of the roles of SOA  
82 in the environment, human health, and climate.

83 Gas-particle partitioning of semi-volatile and low-volatile organic compounds  
84 (SVOCs and LVOCs) generated from VOC oxidation is an important pathway of SOA  
85 formation. In most current chemical transport models (CTMs), this process is treated as an  
86 equilibrium partitioning process that depends on the mass concentration of the organic  
87 particulate matter (OPM), ambient temperature (T), the mean molecular weight of OPM,  
88 and the volatility of condensed organics (Pankow, 1994). The formation of condensed  
89 organic products is commonly represented by lumped surrogate SVOCs in a 2-product  
90 model with volatilities and SVOC yields fitted to chamber experiments (Odum et al., 1996).  
91 To better represent the volatility of primary organic aerosol (POA) and the multi-generation  
92 oxidation of SVOCs to a wider range, Donahue et al. (2006) proposed the volatility basis  
93 set (VBS) model in which the mass yields of SVOCs are fitted to a fixed number of  
94 volatility bins (usually  $0.01-10^5 \mu\text{g m}^{-3}$ ). The VBS model has been adopted by several  
95 CTMs (such as WRF-Chem, GEOS-Chem, etc.).

96 Although the models can capture the general trend of SOA evolution and mass  
97 concentration to some extent (Li et al., 2017a; Bergström et al., 2012; Woody et al., 2016),  
98 two key factors currently neglected in models may lead to biases: 1) the molecular  
99 structures and interactions of functional groups (-OH, -C=O, -COOH, etc.) of condensed  
100 organics (non-ideality); 2) partitioning of water vapor, the most abundant atmospheric  
101 constituent besides O<sub>2</sub> and N<sub>2</sub>, to OPM. The non-ideality alters the volatility of condensed  
102 organics, and thus their contributions to the total SOA mass loading (Cappa et al., 2008).  
103 Water partitioning into OPM can reduce the partial pressure of organics due to Raoult's  
104 Law effect (Prisle et al., 2010) and lead to increases in SOA mass. The amount of aerosol  
105 liquid water associated with organics ( $ALW_{\text{org}}$ ) may vary for different precursors (Healy et  
106 al., 2009; Prisle et al., 2010). The above two aspects will not only affect the chemical  
107 composition of SOA but also the inorganic portion (Ansari and Pandis, 2000) and optical  
108 properties (Denjean et al., 2015) of aerosols.

109 Laboratory and field studies have observed water absorbed by SOA from a variety of  
110 precursor VOCs (Lambe et al., 2011; Zhao et al., 2016b; Asa-Awuku et al., 2010;  
111 Varutbangkul et al., 2006). The hygroscopicity of SOA, quantitatively described by the  
112 hygroscopicity parameter,  $\kappa$ , is correlated with the oxygen-to-carbon ratio (O:C) and  
113 increases with more oxidized SOA during photochemical aging (Lambe et al., 2011; Zhao  
114 et al., 2016b). The OPM-associated water can be estimated using the  $\kappa$ -Köhler theory under  
115 the Zdanovskii-Stokes-Robinson (ZSR) assumption of no interactions between any  
116 constituents in aerosols (Petters and Kreidenweis, 2007). The total water content is the  
117 summation of water associated with each solute at the same water activity. Guo et al. (2015)  
118 found that this simplified method, along with the ISORROPIA model which is used to  
119 predict aerosol liquid water (ALW) associated with the inorganic portion of aerosols, could  
120 reproduce the observed total ALW in the ambient environment. Pye et al. (2017) found that  
121 the modeled organic aerosol (OA) improved significantly but biased high at nighttime  
122 when  $ALW_{org}$  is included in the calculation. However, as the interaction among organic  
123 species and between organics and water in the organic-water mixture has been shown to  
124 play an important role in SOA formation and water partitioning to OPM (Kim et al., 2019),  
125 the  $ALW_{org}$  estimated by the  $\kappa$ -Köhler theory and its impact on SOA might not be accurate.  
126 Using UNiVersal Functional Activity Coefficient (UNIFAC) method (Fredenslund et al.,  
127 1975) for calculating activity coefficients of the organic-water mixture, it was found that  
128 in the eastern U.S., where biogenic SOA dominated the OA, considering  $ALW_{org}$  leads to  
129 a significant increase in predicted SOA (Pankow et al., 2015; Jathar et al., 2016).

130 China has been suffering from severe  $PM_{2.5}$  pollution especially in the eastern region  
131 with fast urbanization and economic development (Fu and Chen, 2017). SOA is a very  
132 important component of  $PM_{2.5}$  in China that contributes about 20-50% (Li et al., 2017b).  
133 The fraction of SOA in OA increases during haze events (Huang et al., 2014; Sun et al.,  
134 2019). Previous modeling studies in China indicate that SOA was underpredicted (Lin et

135 al., 2016; Jiang et al., 2012) and the impacts of the non-ideality and water-OPM partitioning  
136 on modeled SOA have not been evaluated.

137 In this study, regional simulations of SOA during January and July of 2013 over  
138 eastern China were conducted to investigate the seasonal variation of SOA due to water  
139 partitioning into OPM. The model performance was evaluated against observed  
140 meteorological parameters (temperature and relative humidity, RH) as well as PM<sub>2.5</sub>,  
141 organic carbon (OC), and OA at ground monitoring sites. The regional and seasonal  
142 impacts on SOA, ALW<sub>org</sub>, and properties of aerosols were quantified. Lastly, influences of  
143 the results by several factors including meteorological parameters, estimations of saturation  
144 vapor pressures (SVP) of condensed organics, and the individual impacts of ALW<sub>org</sub> and  
145 non-ideality of the organic-water mixture on SOA prediction were analyzed.

## 146 **2 Methodology**

### 147 **2.1 Model description**

148 The Community Multi-scale Air Quality model (CMAQ v5.0.1) coupled with a modified  
149 SAPRC-11 was used in this study. Model configurations were largely based on that used  
150 by Hu et al. (2016) as summarized below. Firstly, SAPRC-11 was expanded for a more  
151 detailed treatment of isoprene oxidation and tracking dicarbonyl products (glyoxal and  
152 methylglyoxal) from different groups of major precursors (Ying et al., 2015). Secondly,  
153 SOA from isoprene epoxydiols (IEPOX), methacrylic acid epoxide (MAE) and dicarbonyls  
154 through surface-controlled irreversible reactive uptake were added (Hu et al., 2017; Li et  
155 al., 2015; Liu et al., 2020; Ying et al., 2015). Thirdly, the heterogeneous formation of  
156 secondary nitrate and sulfate from NO<sub>2</sub> and SO<sub>2</sub> reaction on the particle surfaces (Ying et  
157 al., 2014) were added, which is an important source of secondary inorganic aerosols (Zheng  
158 et al., 2015) and improves model estimates of nitrate and sulfate (Qiao et al., 2018; Shi et  
159 al., 2017). Fourthly, SOA yields were corrected for vapor wall loss (Zhang et al., 2014).  
160 Impacts of the above updates on model performances have been extensively discussed in  
161 the cited work and will not be further investigated in the current study.

162 The SOA module mostly follows Pankow et al. (2015). Two types of SOA as  
 163 traditionally treated in CMAQ were considered, “semi-volatile” (SV) portion that formed  
 164 via equilibrium absorption-partitioning of SVOCs, and “non-volatile” (NV) portion that  
 165 includes the oligomers and SOA formed via direct oxidation of aromatics at low-NO<sub>x</sub>. SOA  
 166 from dicarbonyls, IEPOX, and MAE were formed by irreversible reactive uptake and  
 167 categorized as NV-SOA in the current model as well. Some studies investigated SOA from  
 168 glyoxal, methylglyoxal, and IEPOX using detailed reactions and reversible pathways in  
 169 models or observed as reversible processes in chamber experiments, leading to a relatively  
 170 lower SOA yield compared to the surface-controlled irreversible uptake (Lim et al., 2013;  
 171 Knote et al., 2014; Galloway et al., 2009; El-Sayed et al., 2018; Budisulistiorini et al., 2017).  
 172 The non-volatile assumption used in this paper allows an upper-limit estimation of the  
 173 importance of these additional SOA formation pathways. POA was treated as non-volatile  
 174 and non-reactive. The mass distribution of SVOCs between the gas-phase and particle-  
 175 phase follows the equation:

$$K_{p,i} = \frac{F_i}{M \cdot A_i} \quad (\text{Eq 1})$$

176 where  $K_{p,i}$  ( $m^3 \mu g^{-1}$ ) is the gas-particle partitioning constant for compound  $i$ ,  $F_i$  ( $\mu g m^{-3}$ )  
 177 is the concentration of species  $i$  in the particle-phase,  $A_i$  ( $\mu g m^{-3}$ ) is the concentration of  
 178 species  $i$  in the gas-phase, and  $M$  ( $\mu g m^{-3}$ ) is the total mass concentration of the absorbing  
 179 organic phase (i.e. OPM). The gas-particle partitioning constant  $K_{p,i}$  is dependent on the  
 180 chemical composition of OPM. Pankow et al. (1994) derived  $K_{p,i}$  for SVOCs partitioning  
 181 into an absorbing organic phase as:

$$K_{p,i} = \frac{RT}{10^6 \overline{MW} \xi_i p_{L,i}^o} \quad (\text{Eq 2})$$

182 where  $p_{L,i}^o$  (atm) is the SVP of the pure compound  $i$  at temperature  $T$ (K),  $\xi_i$  is the activity  
 183 coefficient of species  $i$  in the absorbing organic phase,  $\overline{MW}$  ( $g mol^{-1}$ ) is the average  
 184 molecular weight of OPM,  $R$  ( $8.314 J mol^{-1} K^{-1}$ ) is the gas constant, and  $10^6$  is used to  
 185 convert the units to  $m^3 \mu g^{-1}$ .

186 There are 12 lumped SVOCs generated by oxidation of alkanes, alkenes, and  
187 aromatics under different NO<sub>x</sub> conditions and 8 NV organic products as listed in Table S1  
188 and Table S2. More details about the lumped precursors such as formation conditions  
189 (“high” and “low” NO<sub>x</sub>), lumping species and method, and yields from parent VOCs can  
190 be found in Carlton et al. (2010) and summarized in SI. Activity coefficients of SVOCs  
191 were calculated based on the composition of OPM using the UNIFAC method, with  
192 assigned carbon number ( $n_c$ ), functional groups and energy interaction parameters to both  
193 SV and NV compounds (Pankow et al., 2015). The UNIFAC model is one of the commonly  
194 used models that activity coefficients of condensed organics and their interactions with  
195 water can be estimated. This method has been adopted to investigate the impacts of non-  
196 ideality and water partitioning into OPM on SOA for different precursors in box models  
197 (Seinfeld et al., 2001; Bowman and Melton, 2004) and CTMs (Pankow et al., 2015; Kim  
198 et al., 2019). In the current model, POA was assumed to have a bulk composition of ten  
199 categories of surrogate species (Table S3), as used by Li et al. (2015). POA is also involved  
200 in the calculation of activity coefficients for the organic-water mixture. Detailed  
201 information about the surrogate species including their structures and properties can be  
202 found in Li et al. (2015) and references therein.

203 In addition to organic compounds, water partitioning into OPM is enabled according  
204 to Eq 1 and Eq 2. In such a case, the absorbing phase in Eq 1 includes both organic aerosols  
205 and water associated with OPM. As water condenses in the absorbing organic phase, it will  
206 further alter the molar fraction of each composition, the activity coefficient of SVOCs and  
207 the SV-SOA mass concentrations as a result. In the current model, we assumed no  
208 interactions between the inorganic and organic phases.

## 209 **2.2 Estimating the hygroscopicity of OA**

210 Based on the  $\kappa$ -Köhler theory with linearly additive hygroscopic behavior of each  
211 component of the mixed particle,  $ALW_{org}$  is related to the hygroscopicity parameter for the  
212 organic mixture ( $\kappa_{org}$ ) by Eq3 (Petters and Kreidenweis, 2007):

$$ALW_{org} = \rho_w V_{org} \kappa_{org} \frac{a_w}{1 - a_w} \quad (\text{Eq3})$$

213 where  $\rho_w$  is the density of water (assumed to be  $1 \text{ g cm}^{-3}$ ),  $V_{org}$  is the volume concentration  
 214 of organics, and  $a_w$  is the water activity (assumed to be the same as RH). Since  $ALW_{org}$   
 215 in this study is calculated mechanistically using the partitioning theory,  $\kappa_{org}$  can be  
 216 estimated by rearranging Eq3:

$$\kappa_{org} = \frac{ALW_{org}}{\rho_w V_{org}} \times \frac{1 - a_w}{a_w} \quad (\text{Eq4})$$

217  $V_{org}$  can be estimated from the modeled mass concentration of OA, assuming the density  
 218 of OA to be  $1.2 \text{ g cm}^{-3}$  (Li et al., 2019).

219 In many studies,  $\kappa_{org}$  is assumed to increase linearly with the oxidation state of OA,  
 220 expressed as the O:C ratio (Massoli et al., 2010; Duplissy et al., 2011; Lambe et al., 2011).  
 221 The correlation of  $\kappa_{org}$  and O:C ratio at 9 representative cities was evaluated during  
 222 January and July of 2013, with the reduced major axis regression method (Ayers, 2001).  
 223 The O:C ratio was calculated using Eq5:

$$O:C = \sum_{i=1}^n f_i (O:C)_i \quad (\text{Eq5})$$

224 where  $f_i$  and  $(O:C)_i$  are the molar fraction and O:C ratio of organic aerosol component  
 225 i. For POA, a fixed molar fraction and composition were assumed following Li et al. (2015).  
 226 For SOA, the O:C ratio was calculated by using their organic matter to organic carbon ratio  
 227 (OM:OC) following Simon and Bhawe (2012):

$$O:C = \frac{12}{15} (\text{OM:OC}) - \frac{14}{15} \quad (\text{Eq6})$$

228 The OM:OC ratio of each SOA component follows Pankow et al. (2015) as shown in Table  
 229 S1-S2.

### 230 **2.3 Model application**

231 The simulation domain has a horizontal resolution of  $36 \text{ km} \times 36 \text{ km}$  ( $100 \times 100$  grids)  
 232 and a vertical structure of 18 layers up to 21 km, which covers eastern China as shown in  
 233 Figure S1. Anthropogenic emissions were generated from the Multi-resolution Emission

234 Inventory for China (MEIC) (Zhang et al., 2009) v1.0 with a  $0.25^\circ \times 0.25^\circ$  resolution  
235 (<http://www.meicmodel.org>) for China, and the Regional Emission inventory in Asia  
236 version 2 (REAS2) (Kurokawa et al., 2013) with a  $0.25^\circ \times 0.25^\circ$  resolution  
237 (<http://www.nies.go.jp/REAS/>) for the rest of the domain. Biogenic emissions were  
238 generated by the Model for Emissions of Gases and Aerosols from Nature (MEGAN) v2.1,  
239 with the leaf area index (LAI) from the 8-day Moderate Resolution Imaging  
240 Spectroradiometer (MODIS) LAI product (MOD15A2) and the plant function types (PFTs)  
241 from the Global Community Land Model (CLM 3.0). Open biomass burning emissions  
242 were generated from the Fire INventory from NCAR (FINN) (Wiedinmyer et al., 2011).  
243 Dust and sea salt emissions were generated online during CMAQ simulations. The total  
244 emissions of major SOA precursors and their spatial distributions are shown in Table S4  
245 and Figure S2. Meteorological fields were generated using the Weather Research and  
246 Forecasting (WRF) model v3.6.1 with initial and boundary conditions from the NCEP FNL  
247 Operational Model Global Tropospheric Analyses dataset. More details about the model  
248 application can be found in Hu et al. (2016)

249 Four scenarios are investigated in this study. The base case (BS) applies the default  
250 secondary organic aerosol module of CMAQ v5.0.1. In this case, no water partitioning into  
251 OPM is considered. Lumped semi-volatile products from the oxidation of various  
252 precursors partition into a single organic phase, which is considered as an ideal mixture of  
253 POA and SOA with  $\gamma_{org}=1$ . The water case (S1) includes water partitioning into OPM,  
254 which is again considered as an ideal solution ( $\gamma_{org}=1$  and  $\gamma_{H_2O} = 1$ ). The UNIFAC case  
255 (S2) considers the interaction between organic constituents with UNIFAC calculated  
256 activity coefficients ( $\gamma_{org} \neq 1$ ) but does not allow water partitioning into OPM. The  
257 combined case (S3) allows both water partitioning and interactions between all constituents  
258 (including water and organics) using UNIFAC calculated activity coefficients ( $\gamma_{org} \neq 1$   
259 and  $\gamma_{H_2O} \neq 1$ ). The results of BS and S3 are used to examine the overall impacts of water  
260 partitioning into OPM and polarity of organics on SOA and  $ALW_{org}$ , as shown in Section

261 3.1-3.4. The separate influences of those two processes on SOA from S1 and S2 are  
262 discussed in Section 3.5.

### 263 **3 Results and discussion**

#### 264 **3.1 Model evaluation**

265 The meteorological inputs and emissions have been used in several previous publications.  
266 Model performance on meteorological parameters (temperature and RH), gaseous species  
267 and gas and aerosol concentrations have been extensively evaluated (Hu et al., 2016; Hu et  
268 al., 2017; Qiao et al., 2018; Shi et al., 2017). A summary of the model performance related  
269 to this study is provided below. Observed meteorological data were obtained from the  
270 National Climatic Data Center (<ftp://ftp.ncdc.noaa.gov/pub/data/noaa>). Observations of  
271 OC at two urban locations, Beijing (Cao et al., 2014; Wang et al., 2015) and Guangzhou  
272 (Lai et al., 2016) and OA in Beijing (Sun et al., 2014) during January of 2013 as well as  
273 surface PM<sub>2.5</sub> at several monitoring sites during July of 2013 from China National  
274 Environmental Monitoring Center (<http://113.108.142.147:20035/emcpublish/>) were used  
275 to evaluate model estimates of aerosols. Details of measurement methodology and  
276 uncertainties of observations are listed in the corresponding references.

277 Temperature and RH are the two meteorological factors that affect SOA formation.  
278 Table 1 lists model statistics of mean observation (OBS), mean prediction (PRE), mean  
279 bias (MB), gross error (GE) and correlation coefficient (R) based on WRF and observations  
280 at monitoring sites located in 8 sub-regions of the domain (Figure S1) during January and  
281 July of 2013. The benchmarks for the MM5 model (another meteorology model) of 4-12km  
282 horizontal resolution suggested by Emery et al. (2001) are also listed in the table. Details  
283 of monitoring sites in the 8 sub-regions are listed in Table S5. Overall, WRF tends to  
284 underestimate both temperature and RH. The model shows better agreement with observed  
285 temperature as R is higher than that of RH. Both temperature and RH are well captured by  
286 the model in YRD, the Pearl River Delta (PRD), and the central regions of China (the major

287 regions of eastern China). In these regions, MB and GE of temperature are -1.2~0.7 K and  
288 1.8~2.6 K, respectively, which are -11.8~5.6% and 9.2~16.8% for RH, respectively.

289 Model estimates of daily organic carbon (OC) from case S3 are compared with  
290 measurements at monitoring sites in Beijing and Guangzhou in January of 2013 (Figure  
291 1(a)). The factors used to convert SOA to OC (SOC) are listed in Table S1-S2. OC from  
292 POA (POC) is directly predicted by the model. Generally, the ratio between modeled and  
293 observed OC concentration falls in the range of 1:2 to 2:1, with an R-value of 0.7. The  
294 model tends to underestimate OC on high concentration days. Overall, the mean fractional  
295 bias (MFB) and mean fractional error (MFE) of OC are -0.20 and 0.27, within the criteria  
296 ( $MFB \leq \pm 0.6$ ;  $MFE \leq 0.75$ ) suggested by EPA (2007). The bias in OC might be due to  
297 underestimated POA emissions and underpredicted SOA in CMAQ from missing  
298 precursors (Hu et al., 2017; Zhao et al., 2016a). No significant differences in OC are  
299 observed in S3 compared to BS (not shown), likely due to the biased-low SOA predicted  
300 in the current model that limits the impact of  $ALW_{org}$  on SOA formation.

301 The underestimate of SOA can be seen from Figure 1(b) as well. CMAQ can well  
302 capture the observed diurnal variation of OA in Beijing during wintertime, except for the  
303 underestimates of peak values. The correlation coefficient of modeled to observed OA is  
304 0.55. We find a 25% underestimate of OA on average. Better agreement between the model  
305 and observations is shown on non-polluted days (daily-averaged concentration less than 75  
306  $\mu\text{g m}^{-3}$ ). The mean fractional bias (MFB) and mean fractional error (MFE) of polluted days  
307 are -0.38 and 0.64, respectively, which are worse than that of the non-polluted days (-0.26  
308 for MFB and 0.52 for MFE). The overall MFB and MFE of OA during January are -0.28  
309 and 0.54, within the criteria ( $MFB \leq \pm 0.6$ ;  $MFE \leq 0.75$ ) suggested by EPA (2007). Again,  
310 no apparent changes of SOA nor OA are observed between case S3 and BS (not shown),  
311 since POA is predicted to be the primary contributor to OA at Beijing in winter in the  
312 current model, with an averaged SOA/POA ratio of 0.12. This ratio is much lower than the  
313 field observation of about 0.45-1.94 (Zhao et al., 2019; Sun et al., 2013; Sun et al., 2016).

314 The bias might be due to the missing SOA converted by partitioning and aging of semi-  
315 volatile POA as well as oxidation from intermediate volatile organic compounds (IVOCs)  
316 and VOC oxidation products. Those pathways are shown to be important for SOA  
317 formation by modeling, field and chamber studies (Hodzic et al., 2010; Jimenez et al., 2009;  
318 Murphy et al., 2017; Robinson et al., 2007; Shrivastava et al., 2008; Tkacik et al., 2012;  
319 Zhao et al., 2014; Zhao et al., 2016a).

320 A sensitivity test was performed by using the newest CMAQ model version 5.3.1 that  
321 includes all the above processes in the aerosol module. The SOA/POA ratio in Beijing is  
322 improved greatly to be 0.83 in winter. However, high uncertainties still exist in the  
323 emissions of the involved precursors and characterization of SOA formation through these  
324 processes, needing further constrains by observations. Their influences on water  
325 partitioning into OPM and non-ideality of the organic-water mixture on SOA will be  
326 evaluated in a future study.

327 Due to the lack of observed OC and OA in July of 2013, as an alternative, model  
328 performances are evaluated by comparing predicted and observed PM<sub>2.5</sub> at ground sites  
329 (Figure S1) as shown in Figure S3. Generally, the model can well reproduce the diurnal  
330 variation of PM<sub>2.5</sub> in most regions. Predicted PM<sub>2.5</sub> on highly concentrated days is biased  
331 low, especially in the North China Plain (NCP). NCP has the highest PM<sub>2.5</sub> ranging from  
332 60  $\mu\text{g m}^{-3}$  to 300  $\mu\text{g m}^{-3}$ . The bias in modeled PM<sub>2.5</sub> is significant in cities in Northwest.  
333 This might be due to missing dust emissions in the current inventory (Hu et al., 2016). To  
334 further evaluate the model performance, averaged MFB and MFE of modeled PM<sub>2.5</sub> are  
335 plotted against observations of each site as shown in Figure S4. The criteria of MFB and  
336 MFE followed recommendations by Boylan and Russell (2006). Our model performs well  
337 since most of the predictions meet the criteria and a large fraction (>58%) meet the goal.  
338 The averaged MFB and MFE across all the sites are -0.28 and 0.39, indicating slightly  
339 underestimate of PM<sub>2.5</sub> by the model.

### 340 **3.2 Impacts on SOA and ALW<sub>org</sub>**

341 The spatial distribution of SOA varies greatly in the two seasons. In winter, SOA is  
342 relatively high in eastern SCB and the central and eastern provinces of Shandong, Henan,  
343 Anhui, and Hubei (Figure 2 and Figure S5). Monthly-averaged SOA concentrations in the  
344 above areas are up to 25 and 15-20  $\mu\text{g m}^{-3}$ , respectively. Anthropogenic emissions are the  
345 major sources of SOA (Figure S6), such as dicarbonyl products from the oxidation of  
346 xylene and toluene (Hu et al., 2017). In summer, surface SOA is high in Northeast, NCP,  
347 and YRD. The highest SOA occurs in Shanghai and Jiangsu provinces as well as the coastal  
348 area of the Yellow Sea, with the value of  $\sim 9\text{-}16 \mu\text{g m}^{-3}$  at the surface and  $\sim 20\text{-}25 \text{ mg m}^{-2}$  in  
349 the column (col-SOA) of the atmosphere below 21 km (Figure S5). Different from winter  
350 SOA, a significant fraction of summer SOA is originated from biogenic emissions (Figure  
351 S7). Anthropogenic SOA is highly concentrated in the coastal areas of the Yellow Sea and  
352 Bohai Bay.

353 Combined effects of water partitioning into OPM and non-ideality on SOA formation  
354 also exhibit strong seasonal variation. In winter, SOA is slightly decreased by  $1.5 \mu\text{g m}^{-3}$   
355 (10-20%) on average at the surface (Figure 2) and less than  $\sim 1 \text{ mg m}^{-2}$  (20%) in the column  
356 (Figure S5) over high SOA regions where anthropogenic sources dominate. We show later  
357 that the decrease of SOA is mainly due to the large activity coefficients of SVOCs which  
358 decrease  $K_{p,i}$ . In summer, higher temperature and RH promote water partitioning and SOA  
359 formation so that SOA increases apparently over the entire domain, with the highest  
360 enhancement of  $2\text{-}4 \mu\text{g m}^{-3}$  (20-50%) at the surface (Figure 2) and  $4\text{-}6 \text{ mg m}^{-2}$  (30-60%) in  
361 the column (Figure S5) over YRD and the coastal area of Yellow Sea. Anthropogenic SOA  
362 dominates the total change in winter as shown in Figure S6. In summer, the increase of  
363 SOA is attributed to biogenic sources in inland areas and anthropogenic sources over the  
364 ocean (Figure S7).

365 Regional distribution of  $\text{ALW}_{\text{org}}$  is similar to the change of SOA as shown in Figure  
366 3. In winter, a maximum averaged concentration of  $3.0 \mu\text{g m}^{-3}$  for  $\text{ALW}_{\text{org}}$  occurs in the  
367 high SOA region, where significant changes of SOA also occur. In other areas, the averaged

368 concentration of  $ALW_{org}$  is about 0.5-1.5  $\mu\text{g m}^{-3}$ . Overall, the average ratio of  $ALW_{org}$  to  
369 SOA is about 0.1-0.3 in winter. In summer, water partitioning into OPM mostly occurs in  
370 the east coastal area at the surface where a significant increase of anthropogenic SOA (such  
371 as those from toluene and xylenes) is observed. This might be due to the high polarity of  
372 anthropogenic SVOCs (having more -COOH groups) that absorb more water. In the coastal  
373 area, the averaged concentration of  $ALW_{org}$  is about 5-7  $\mu\text{g m}^{-3}$ , with the  $ALW_{org}/SOA$   
374 ratio of 0.5-1.0. Over the land, the averaged concentration of  $ALW_{org}$  is about 1-3  $\mu\text{g m}^{-3}$   
375 ( $ALW_{org}/SOA$  ratio of 0.2-0.5) in Northeast and East China. Water partitioning is mostly  
376 associated with BSOA originated from isoprene and monoterpenes oxidation that produces  
377 SVOCs with abundant OH group.

378 Based on the column concentrations of  $ALW_{org}$  and  $ALW_{org}/SOA$  ratio (Figure S8),  
379 in winter, more  $ALW_{org}$  must have occurred in the south and southwest regions at higher  
380 levels where a significant increase of col-SOA occurs (Figure S5). The averaged col-  
381  $ALW_{org}/col\text{-SOA}$  ratio in the high SOA area is 0.1-0.3. In summer, the  $ALW_{org}$  must be  
382 high at higher altitudes over the central regions in China. The maximum col- $ALW_{org}$  is  
383 about 7  $\text{mg m}^{-2}$  over YRD, with the col- $ALW_{org}/col\text{-SOA}$  ratio of about 0.3.

### 384 **3.3 Impacts on aerosol properties**

385 Since  $ALW_{org}$  is determined in S3, the  $\kappa_{org}$  values can be estimated from the modeled  
386  $ALW_{org}$ , OA and RH using Eq 4. 9 representative cities were selected to investigate  $\kappa_{org}$   
387 vs. O:C relationship and its seasonal variation as shown in Figure S9 and S10. The results  
388 of all the cities in winter and summer are merged and analyzed as shown in Figure 4. Pairs  
389 of  $\kappa_{org}$  and O:C data are grouped into 10 O:C bins and the average  $\kappa_{org}$  in each bin are  
390 then calculated. Overall, the estimated O:C ratio is within the range of 0.2-0.8. The  
391 averaged  $\kappa_{org}$  in each O:C bin is less than 0.1 in winter, with the highest value in  
392 Guangzhou. As more  $ALW_{org}$  is formed in summer, the averaged  $\kappa_{org}$  also increases  
393 greatly with the highest value of 0.35 in Beijing. The linear correlation between  $\kappa_{org}$  and  
394 O:C shows significant spatial and seasonal variations. For example, the slope of  $\kappa_{org}$ -O:C

395 is 70-90% smaller in winter than in summer in the Northern cities such as Shenyang,  
 396 Beijing, Zhengzhou, and Xi'an. However, in Guangzhou, the slope is 83% higher in winter  
 397 than in summer. In Chengdu, the slope is quite similar in both seasons. Overall, the slope  
 398 of  $\kappa_{org}$  vs. O:C at the 9 cities is 0.16 in winter and 0.40 in summer. Most of the fitted  
 399 linear correlations of the individual city fall outside of the range of 0.18 to 0.37 suggested  
 400 in previous studies (Duplissy et al., 2011; Lambe et al., 2011; Massoli et al., 2010; Chang et  
 401 al., 2010), indicating that the hygroscopicity of organic aerosols cannot be simply be  
 402 represented by a single parameter such as the O:C ratio (Rickards et al., 2013). In both  
 403 seasons,  $\kappa_{org}$  approaches zero and negative values as O:C decreases, which might be due  
 404 to the linear regression of  $\kappa_{org}$  and O:C. To avoid this, an exponential fitting of the two  
 405 variables is performed so that  $\kappa_{org}$  falls in the range of (0,1) and is positively correlated  
 406 with O:C. In this case, the fitted correlations are  $\kappa_{org}=1-\exp(-(O:C/1.88)^{2.29})$  and  $\kappa_{org}=1-$   
 407  $\exp(-(O:C/1.06)^{4.50})$  for January and July of 2013, respectively..

408 The impacts on aerosol optical depth (AOD) and aerosol radiative forcing (ARF) are  
 409 further investigated. Figure 5 shows the monthly-averaged AOD at 550 nm in January and  
 410 July of 2013. It is calculated by summarizing the product of model estimated extinction  
 411 coefficient of fine particles ( $b_{ext,i}$ ) multiplied by the thickness ( $HL_i$ ) in each layer:

$$AOD = \sum_{i=1}^N b_{ext,i} \times HL_i \quad (\text{Eq6})$$

412 where N is the number of layers. There are two methods to estimate the aerosol extinction  
 413 coefficient in CMAQv5.0.1. One is based on the Mie theory and the predicted aerosol  
 414 component concentrations ( $b_{ext,m}$ ), and the other is based on correlation from the IMPROVE  
 415 monitoring network that considers the impacts of hygroscopicity of different aerosol  
 416 components ( $b_{ext,r}$ ) (Malm et al., 1994). AOD calculated with the two types of extinction  
 417 coefficient are denoted as AOD<sub>m</sub> and AOD<sub>r</sub>, respectively.

418 In Figure 5, a clear pattern of high AOD<sub>r</sub> in SCB and NCP and low AOD<sub>r</sub> in west  
 419 China is observed in both winter and summer, consistent with previous studies (He et al.,

420 2019; He et al., 2016; Luo et al., 2014). An identical trend in AOD<sub>m</sub> is shown in Figure  
421 S11. The monthly-averaged AOD<sub>r</sub> ranges from 1.0 to 3.2 in January and from 0.3 to 0.9 in  
422 July. AOD<sub>m</sub> is lower than AOD<sub>r</sub>, falling in 0.7-2.2 in January and 0.3-0.6 in July. The  
423 model significantly overestimates AOD in January but agrees better with observations from  
424 MODIS where AOD is high in July (Figure S12). The bias in the predicted AOD might be  
425 partially due to the empirical equation applied in the calculation of AOD in CMAQ (Wang  
426 et al., 2009; Liu et al., 2010), and partially due to the uncertainties of fine AOD overland  
427 from MODIS data (Wang et al., 2009; Levy et al., 2010). The increase of AOD due to  
428 ALW<sub>org</sub> shows a strong spatial and seasonal pattern. In winter, there are no significant  
429 changes in AOD<sub>r</sub> across the whole domain, due to insignificant changes of SOA. In summer,  
430 AOD<sub>r</sub> increases significantly in YRD and the adjacent areas by up to 10%.

431 ARF represents the change in the radiative flux at the top of the atmosphere due to  
432 aerosols. An offline version of the Shortwave Radiative Transfer Model For GCMs  
433 (RRTMG\_SW) was used to calculate the direct radiative effect of aerosols on shortwave  
434 radiation (Iacono et al., 2008). Generally, fine aerosols exhibit cooling effects on the  
435 shortwave radiation in both winter and summer over the entire domain as shown in Figure  
436 6. This impact is much stronger in the areas where AOD is high (Figure 5). The ARF is  
437 highest in Shandong in winter and in the coastal area near Jiangsu province in summer,  
438 which are about  $-5 \text{ W m}^{-2}$  and  $-6 \text{ W m}^{-2}$ , respectively. In winter, no significant changes of  
439 ARF are observed in eastern China (Figure 6b and d). This is likely attributed to an  
440 insignificant contribution of SOA to PM<sub>2.5</sub> in winter compared to other components with  
441 cooling effects, such as sulfate. In summer, SOA is an important component of PM<sub>2.5</sub> (20-  
442 60%), and the effects of water partitioning on shortwave radiation are relatively stronger.  
443 An enhancement of up to 15% in the cooling effects of ARF occurs near the YRD region  
444 where AOD significantly changes as well.

#### 445 **3.4 Sensitivity to T, RH, and SVP**

446 Meteorological conditions and SOA precursors affect the impacts of water partitioning on  
 447 SOA. To better illustrate the dependency of SOA on temperature, relative humidity, and  
 448 SVP of SVOCs, an offline calculation of SOA formation was performed at two  
 449 representative cities (Jinan in NCP during winter and Nanjing in YRD during summer)  
 450 when the daily maximum SOA increases occurred. We assume temperature (T) and water  
 451 vapor mixing ratio (QV) to be within the range of  $\bar{X} \pm \sigma$ , where  $\bar{X}$  and  $\sigma$  are the mean  
 452 and standard deviation calculated based on WRF predictions at each location. We choose  
 453 10 evenly distributed values for T and QV within the range of  $\bar{X} \pm \sigma$ . The temperature  
 454 dependence parameter of SVP ( $\Delta H/R$ ) is also scaled by 0.2, 0.8, 1.4 and 2.0 separately for  
 455 all the SVOCs. As shown in Figure 7, SOA exhibits a negative correlation with T and a  
 456 positive correlation with QV in both cities. SOA is more sensitive to QV under cold  
 457 conditions (Jinan) and to temperature under hot conditions (Nanjing). When the  
 458 temperature is fixed, the sensitivity of SOA to  $\Delta H/R$  is different in the two cities. We find  
 459 more changes in SOA across  $\Delta H/R$  in Jinan. This is attributed to the temperature correction  
 460 factor ( $\zeta_{corr}$ ) of  $K_p$  in CMAQ as defined below:

$$\zeta_{corr} = \frac{K_{p,T_{ref}}}{K_{p,T}} = \frac{T_{ref}}{T} \exp \left[ \frac{\Delta H}{R} \left( \frac{1}{T_{ref}} - \frac{1}{T} \right) \right] \quad (\text{Eq8})$$

461 where  $T_{ref}$  is the reference temperature (298K) and  $T$  is temperature. According to  
 462 Figure 7, the range of T is 265-274K in Jinan and 300-307K in Nanjing. The deviation of  
 463 temperature from the reference value (298K) is greater in Jinan than in Nanjing. Therefore,  
 464 the unit change of  $\Delta H/R$  causes greater variations of  $\zeta_{corr}$  and thus  $K_p$  in Jinan. As a  
 465 result, SOA is more sensitive to  $\Delta H/R$ . The impacts of SVP estimation on SOA are more  
 466 significant in winter than in summer.

### 467 **3.5 Separate impacts of water partitioning/polarity of condensed organics**

468 Impacts of water partitioning into OPM and non-ideality of the organic-water mixture on  
 469 SOA are in opposite directions. Water partitioning increases SOA by ~10-20% in winter  
 470 and ~30-80% in summer in most areas of the domain (Figure 8). This is because the

471 molecular weight of water is quite small and will reduce the molar averaged weight of  
472 OPM ( $\overline{MW}$ ) in Eq 2 (Pankow et al., 2015). The reduced  $\overline{MW}$  further increases  $K_p$  and  
473 promotes mass transfer of SVOCs from the gas-phase to OPM. On the other hand, by  
474 considering non-ideality of the organic-water mixture, activity coefficients of SVOCs are  
475 usually greater than 1.0 in this study, leading to a decrease in  $K_p$ . As a result, the total SOA  
476 concentration is reduced by ~10-20% in winter and ~10-50% in summer (Figure 9). Overall,  
477 the final impacts are the combined consequences of the two processes. In winter, the  
478 increase of SOA caused by water partitioning is offset by the decrease of SOA due to the  
479 polarity of SVOCs in most areas of the domain, resulting in slight decreases of SOA. In  
480 summer, the effect of water partitioning overcomes that of SVOC polarity so the total SOA  
481 loading increases. This further leads to an enhanced attenuation of shortwave solar  
482 radiation and cooling of the atmosphere.

#### 483 **4 Conclusions**

484 The WRF/CMAQ model was used to investigate the impacts of water partitioning into  
485 OPM and non-ideality of the organic-water mixture on SOA formation over eastern China  
486 during January and July of 2013. SOA is greatly enhanced in summer especially in YRD  
487 and over the Yellow Sea by up to 50% and 60% at the surface and in the whole column,  
488 respectively. No significant impacts on SOA are observed in winter. This might be due to  
489 the underestimation of SOA in the current model.  $ALW_{org}$  is highly correlated with the  
490 change of SOA, with the ratio of  $ALW_{org}$  to SOA of 0.1-0.3 and 0.2-1.0 at the surface  
491 where significant changes of SOA occur in winter and summer, respectively. By using the  
492 modeled  $ALW_{org}$ , correlations between  $\kappa_{org}$  and O:C are examined in 9 representative  
493 cities, showing significant spatial and seasonal variations. The increases in SOA lead to an  
494 enhancement in the averaged AOD and the cooling effects of aerosols, by up to 10% and  
495 15% respectively in summer. The model predicted SOA is sensitive to temperature and QV  
496 in both seasons, with higher sensitivity to QV during winter and temperature during  
497 summer. Estimation of SVP also affects modeled SOA, especially in a cold environment.

498 The effects of water partitioning into OPM and non-ideality of the organic-water mixture  
499 on SOA are the opposite. Since the activity coefficients of SVOCs are mostly greater than  
500 1.0 during the simulated episode, SOA concentration decreases when the non-ideality  
501 effect is considered. Averaged SOA concentration decreases by up to 20% in winter and  
502 50% in summer. Water partitioning alone increases SOA by 10-20% in winter and 30-80%  
503 in summer. It should be noticed that the results shown in this study are the lower limit as  
504 the current model tends to underestimate SOA. It is crucial to consider both effects in  
505 simulating SOA formation under hot and humid conditions in CTMs.

506

#### 507 **Data availability**

508 Data used in this manuscript can be provided upon request by e-mail to the corresponding  
509 authors, Qi Ying (qying@civil.tamu.edu) and Jianlin Hu ([jianlinhu@nuist.edu.cn](mailto:jianlinhu@nuist.edu.cn)).

510

#### 511 **Competing interests**

512 The authors declare that they have no conflict of interest.

513

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861 North China, *Atmos. Chem. Phys.*, 15, 2031-2049, 10.5194/acp-15-2031-2015, 2015.  
862

**Table 1.** Statistical analysis of modeled temperature (K) and relative humidity (%) of January and July of 2013 at the monitoring sites in different geographical areas as shown in Figure S1 and listed cities in Table S5.

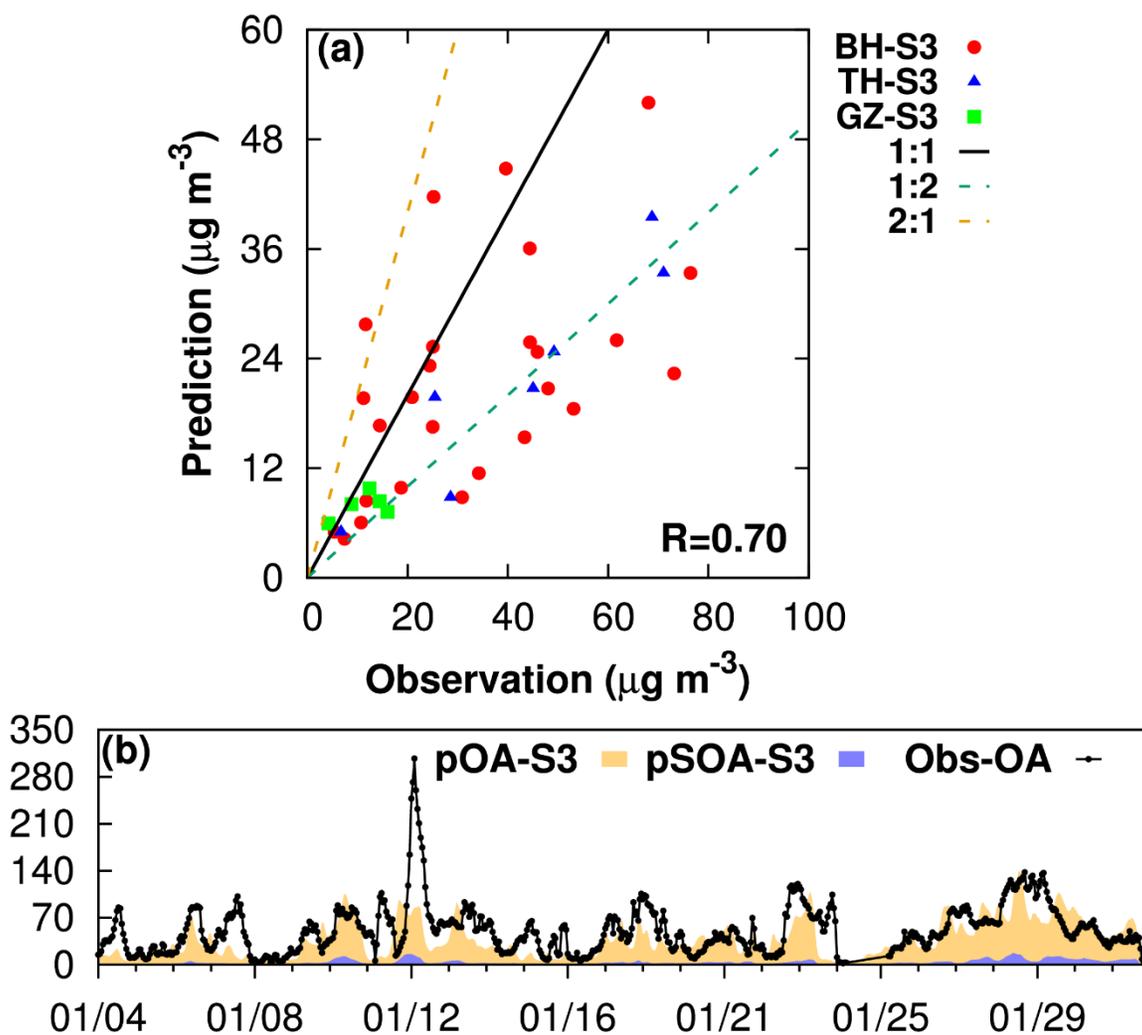
	Northeast		NCP		Northwest		YRD		Central		SCB		PRD		Southwest		Benchmark
	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	Jan	Jul	
<b>Temperature (K)</b>																	
<b>OBS</b>	256.2	296.2	263.9	297.4	266.9	293.0	277.7	303.2	275.7	301.1	276.1	295.7	289.4	301.1	282.2	295.1	
<b>PRE</b>	251.6	296.3	261.2	298.8	267.0	293.5	278.4	302.0	276.2	301.0	273.5	293.1	288.9	300.0	278.8	291.8	
<b>MB</b>	-4.6	0.1	-2.7	1.4	0.1	0.5	0.7	-1.2	0.5	-0.1	-2.6	-2.6	-0.5	-1.0	-3.4	-3.4	$\leq \pm 0.5$
<b>GE</b>	5.5	1.8	4.0	2.3	3.6	3.2	2.0	2.3	2.6	2.4	5.1	4.2	1.9	1.8	4.1	3.5	$\leq 2.0$
<b>R</b>	0.9	0.9	0.9	0.8	0.8	0.8	0.9	0.8	0.9	0.8	0.8	0.9	0.9	0.6	0.9	0.8	
<b>Relative Humidity (%)</b>																	
<b>OBS</b>	77.5	80.3	74.5	72.1	60.2	69.1	79.3	71.3	76.3	71.8	68.5	74.4	77.0	80.6	70.3	77.9	
<b>PRE</b>	85.0	73.2	78.6	57.5	49.2	54.2	75.1	76.9	64.5	70.0	57.7	78.4	75.3	84.8	75.8	85.1	
<b>MB</b>	7.5	-7.1	4.1	-14.6	-11.0	-15.0	-4.2	5.6	-11.8	-1.8	-10.8	4.0	-1.7	4.2	5.5	7.2	
<b>GE</b>	11.3	11.3	14.6	16.6	19.8	19.1	11.0	11.6	16.8	11.7	22.8	11.6	9.5	9.2	15.4	10.4	
<b>R</b>	0.5	0.8	0.4	0.8	0.3	0.7	0.6	0.6	0.6	0.6	0.2	0.6	0.6	0.5	0.5	0.7	

\* OBS is mean observation, PRE is mean prediction, MB is mean bias, GE is gross error, R is the correlation coefficient of predictions vs. observations.

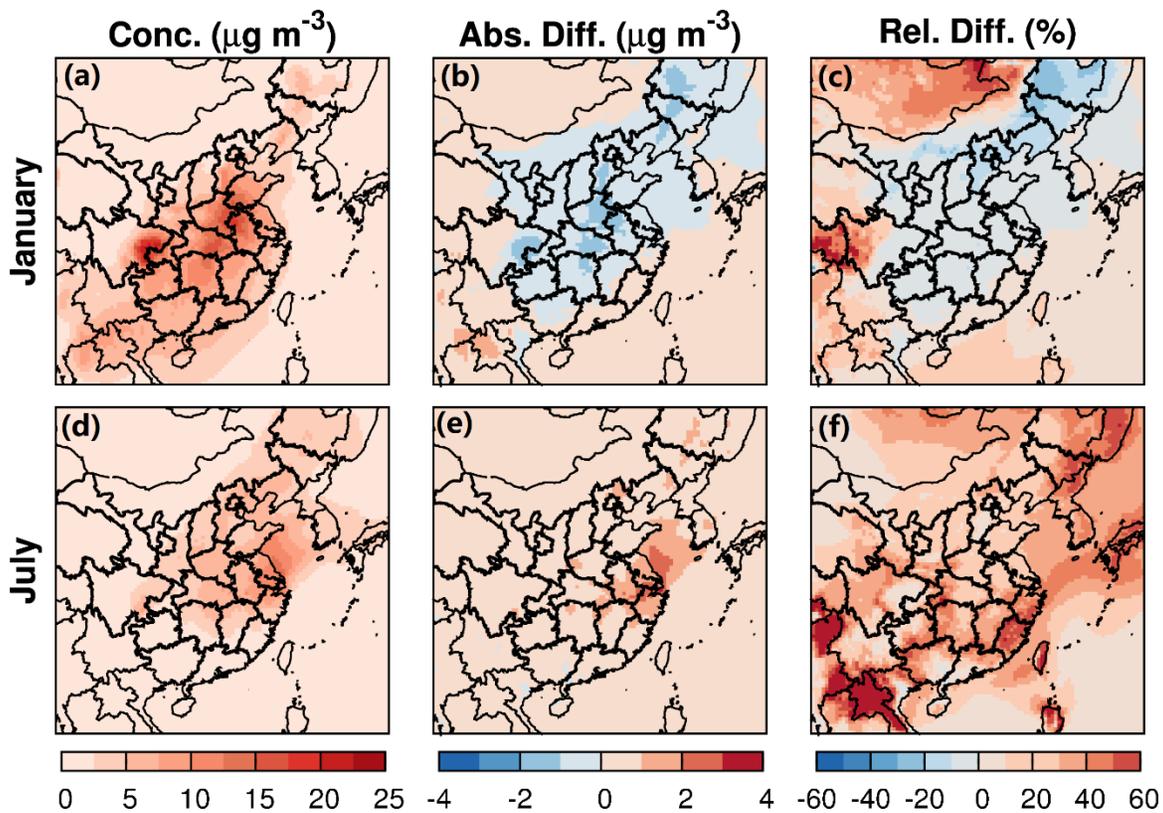
$MB = \sum_{i=1}^N (C_p - C_o) / N$ ;  $GE = \sum_{i=1}^N |C_p - C_o| / N$ , where  $C_p$  and  $C_o$  are the prediction and observation data, N is the total number of data.

\*\* Northeast is Northeast China, NCP is North China Plain, Northwest is Northwest China, YRD is Yangtze River Delta, Central is Central China, SCB is Sichuan Basin, PRD is Pearl River Delta, Southwest is Southwest China.

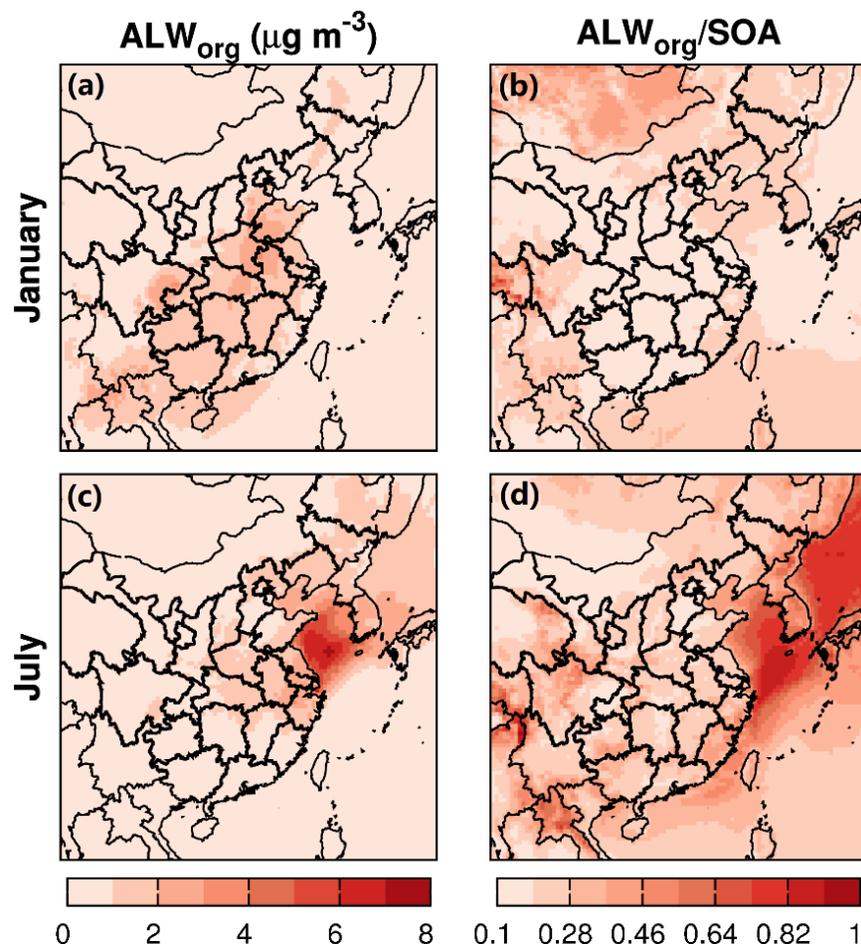
\*\*\* Benchmarks of MB and GE are for the MM5 model of 4-12 km horizontal resolution by Emery et al. (2001)



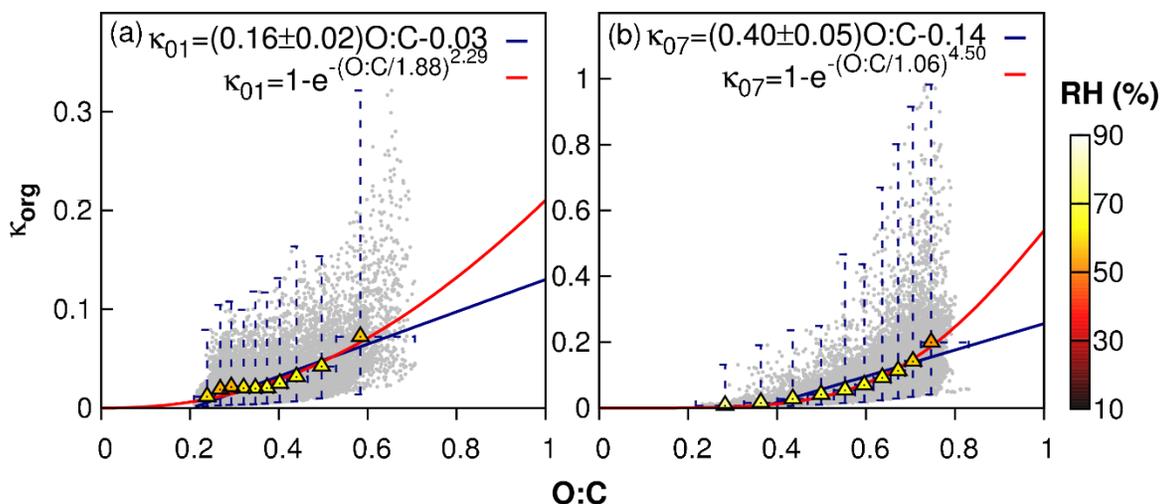
**Figure 1.** Comparison of (a) observed and modeled organic carbon concentration at University of Beihang (BH), Tsinghua University (TH) and Guangzhou (GZ); (b) observed organic aerosol (Obs-OA) at Beijing and predictions of total OA (pOA) and SOA (pSOA), unit is  $\mu\text{g m}^{-3}$ . Locations of monitoring sites are shown in Figure S1.



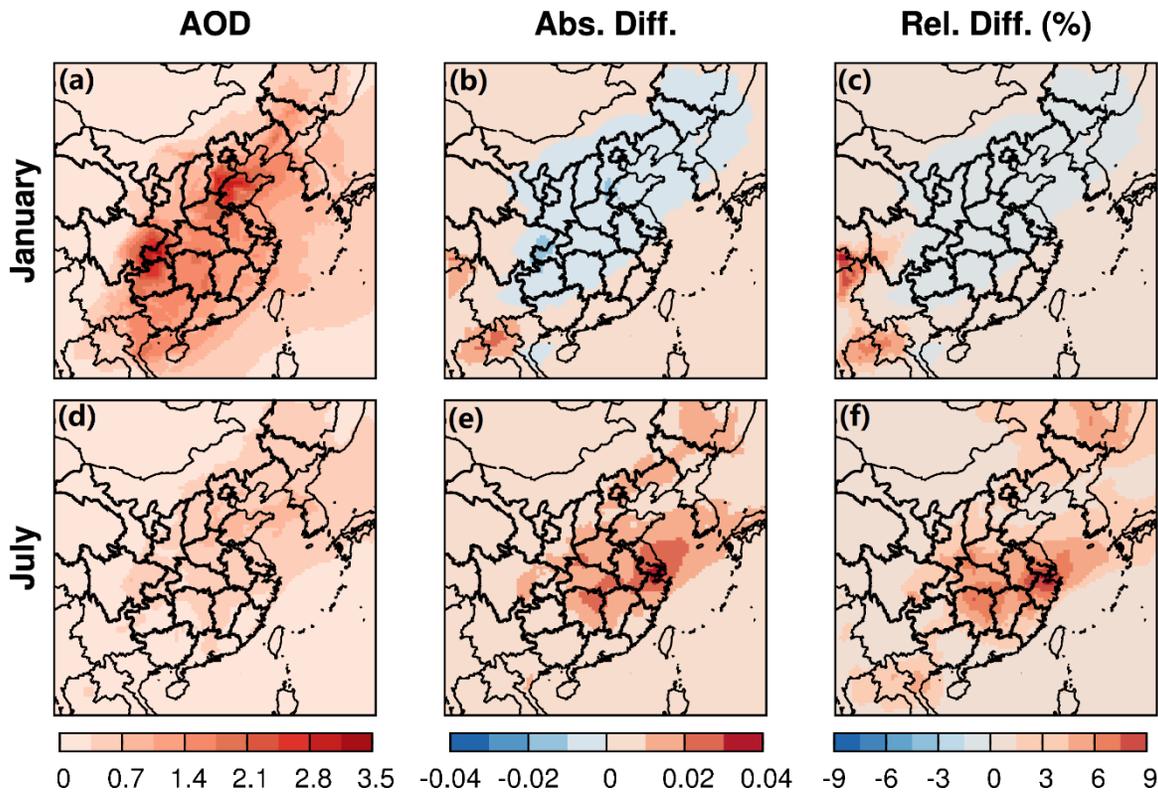
**Figure 2.** Monthly-averaged SOA from S3 and changes of SOA due to water partitioning into OPM and non-ideality of the organic-water mixture. “Abs. Diff.” represents absolute differences (S3-BS); “Rel. Diff.” represents relative differences ((S3-BS)/BS, %).



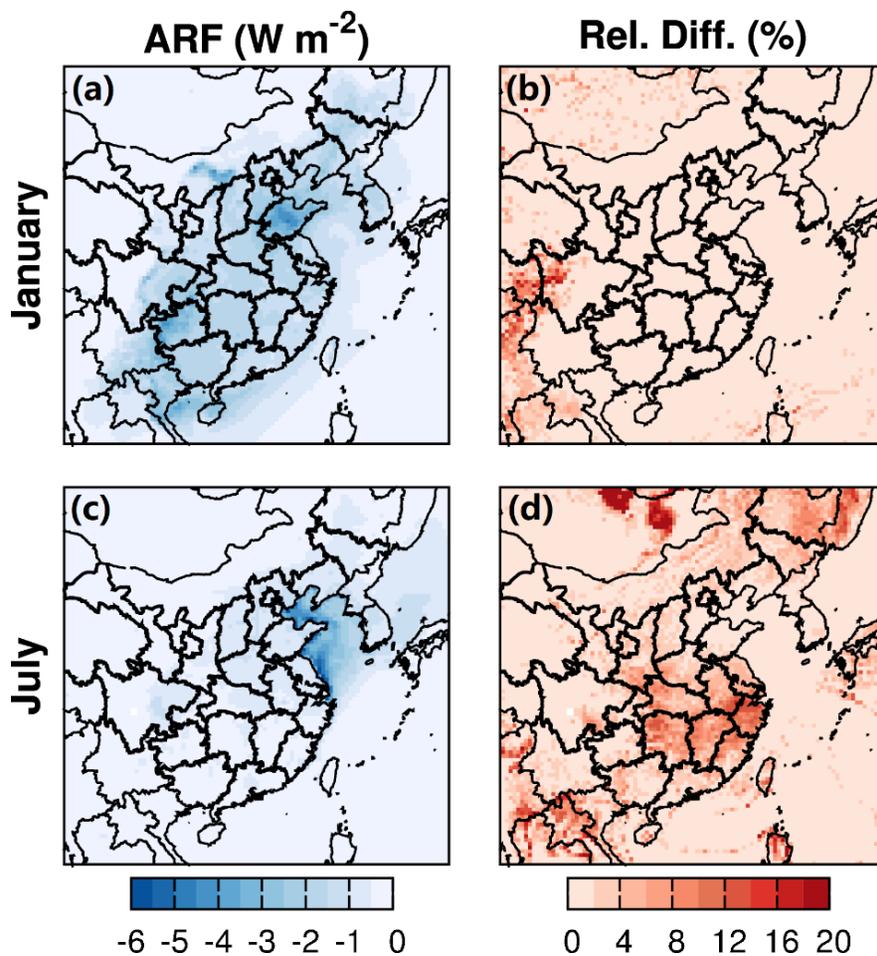
**Figure 3.** Monthly-averaged water partitioning into the organic-phase ( $ALW_{org}$ ,  $\mu\text{g m}^{-3}$ ) and the ratio to SOA ( $ALW_{org}/SOA$ ) during January and July of 2013.



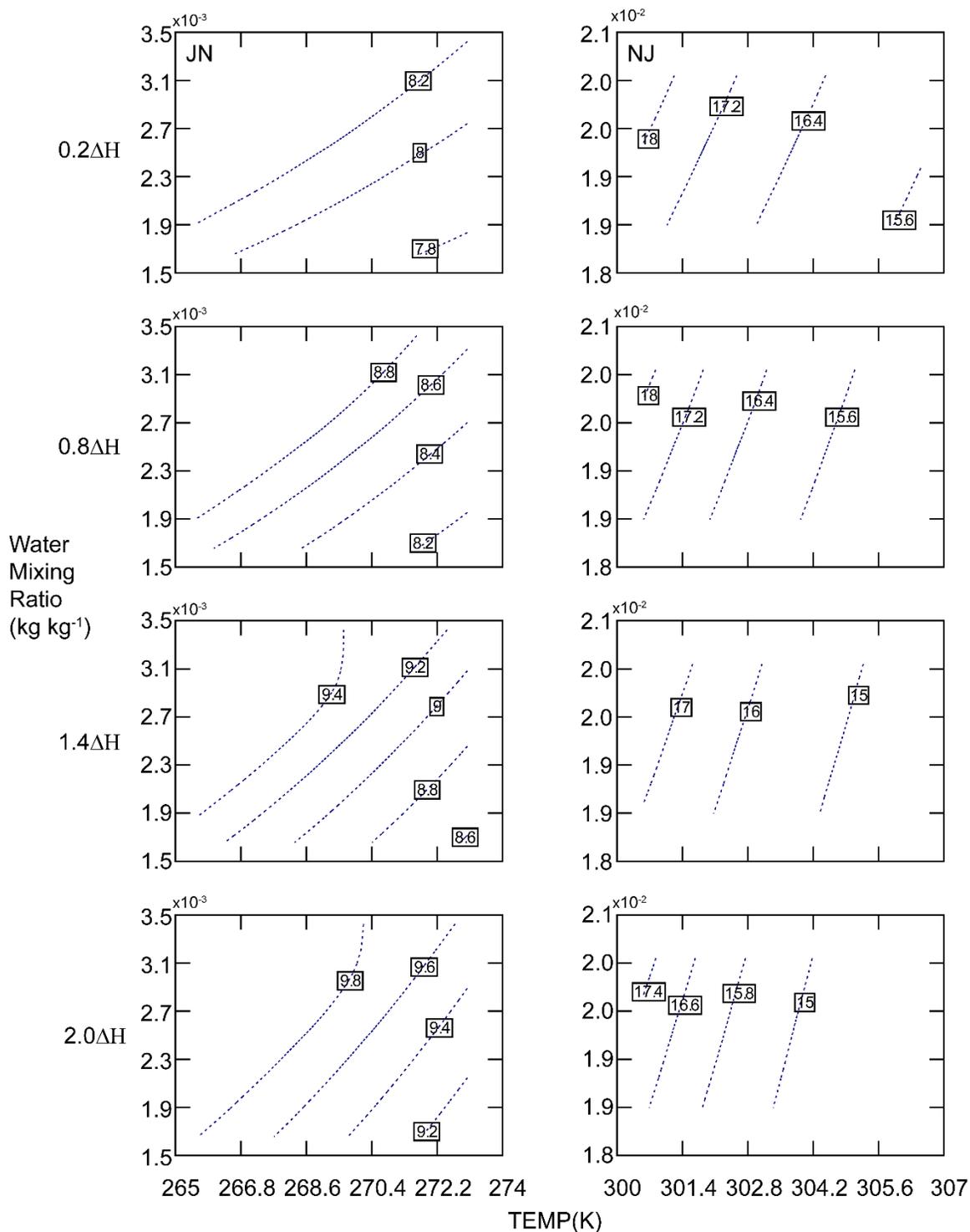
**Figure 4.** The correlation of hygroscopicity of organic aerosol ( $\kappa_{org}$ ) and O:C ratio at 9 representative cities including Shenyang (SS), Beijing (BJ), Jinan (JN), Zhengzhou (ZZ), Xi'an (XA), Nanjing (NJ), Shanghai (SH), Chengdu (CD), and Guangzhou (GZ) in January (a) and July (b) of 2013. O:C ratios are categorized into 10 bins. In each bin, the ranges of O:C and  $\kappa_{org}$  are represented by bars. The mean values of O:C and  $\kappa_{org}$  are represented by triangles colored by the averaged RH of each bin. The relationship between  $\kappa_{org}$  and O:C is fitted by a linear function with reduced major axis regression (blue lines) and an exponential function (red lines), respectively.  $\kappa_{01}$  and  $\kappa_{07}$  represent the fitted correlation for January and July, respectively.



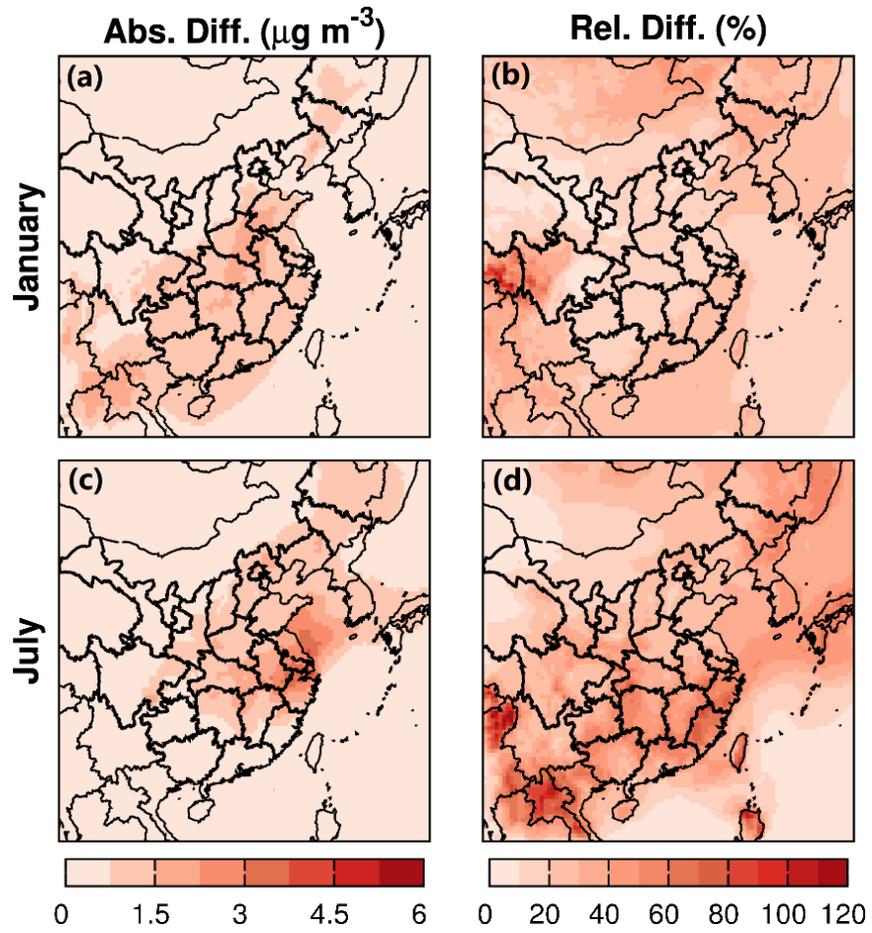
**Figure 5.** Monthly-averaged AOD<sub>r</sub> at 550 nm and changes of AOD<sub>r</sub> due to water partitioning into OPM and non-ideality of the organic-water mixture. “Abs. Diff.” represents absolute differences (S3-BS); “Rel. Diff.” represents relative differences ((S3-BS)/BS, %).



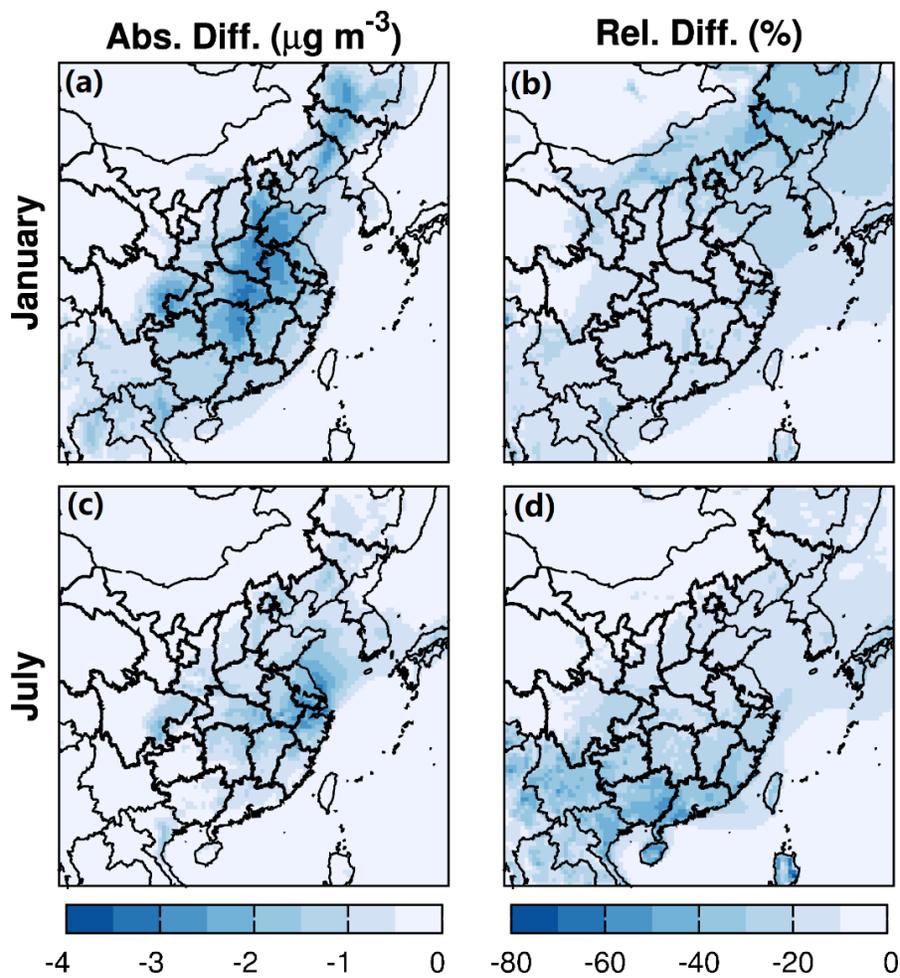
**Figure 6.** Monthly-averaged shortwave direct aerosol radiative forcing at the top of the atmosphere from S3 and the relative changes due to water partitioning into OPM and non-ideality of the organic-water mixture during January and July of 2013. “Rel. Diff.” represents relative differences  $((S3-BS)/BS, \%)$ .



**Figure 7.** The sensitivity of SOA formation to temperature (TEMP), water mixing ratio, and the temperature dependence parameter of SVP ( $\Delta H/R$ ) at Jinan (JN, first column) and Nanjing (NJ, second column).



**Figure 8.** Monthly-averaged impacts of water partitioning into OPM on SOA. “Abs. Diff.” represents absolute differences ( $S3-S2$ ); “Rel. Diff.” represents relative differences ( $(S3-S2)/S2$ , %).



**Figure 9.** Monthly-averaged impacts of non-ideality of the organics-water mixture on SOA. “Abs. Diff.” represents absolute differences ( $S3-S1$ ); “Rel. Diff.” represents relative differences ( $(S3-S1)/S1, \%$ ).