

1 **Elucidating the pollution characteristics of nitrate, sulfate and ammonium in**
2 **PM_{2.5} in Chengdu, southwest China, based on three-year measurements**

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17 **Abstract**

18 Nitrate, sulfate and ammonium (NSA) are the main secondary inorganic aerosols of
19 PM_{2.5} and play an important role in air pollution. In this study, a three-year
20 observational experiment was conducted from January 1, 2015, to December 31, 2017,
21 in Chengdu, southwest China. NSA pollution characteristics, chemical conversion
22 generation, emission reduction control sensitivity and pollutant regional transport
23 characteristics were analysed. NSA are the most important chemical compositions of
24 particles with aerodynamic equivalent diameter $\leq 2.5 \mu\text{m}$ in ambient air (PM_{2.5}), and
25 the contribution of nitrate to the accumulation of PM_{2.5} concentration is greater than
26 that of sulfate and ammonium. NSA also have obvious characteristics of annual,
27 monthly, seasonal, diurnal and weekly variations. Through observation data and model
28 simulation, it was also found that the existence of an aerosol aqueous environment plays

29 an important role in the formation and existence of NSA. Sensitivity analysis between
30 NSAs found that controlling NO_3^- and SO_4^{2-} play an important role in reducing the
31 contribution of NSA to $\text{PM}_{2.5}$, which also implies that the current control of NO_x and
32 SO_2 is important for improving air pollution. Combined with meteorological conditions
33 and potential source contribution function (PSCF) analysis, local emissions and
34 regional emissions of pollutants are found to have important impacts on Chengdu's
35 atmospheric environment. This research result not only provides an assessment of the
36 current atmospheric emission reduction effect but also provides an important reference
37 for atmospheric pollution control.

38 **Keywords:** Secondary inorganic aerosols; Three-year measurements; Pollution
39 characteristics; Chemical conversions; Regional transport; Chengdu

40 **1 Introduction**

41 In recent years, with the rapid development of China's domestic economy and
42 acceleration of urbanization, energy consumption and pollutant emissions have also
43 increased, which increases the burden on the atmospheric environment, and severe air
44 pollution has become a focus of social concern (Liu et al., 2013a; An et al., 2019; Fu et
45 al., 2014; Zhao et al., 2017). When air pollution forms, mass concentrations of particles
46 with aerodynamic equivalent diameter $\leq 2.5 \mu\text{m}$ in ambient air ($\text{PM}_{2.5}$, also known as
47 fine particles) can reach a higher pollution level, which not only reduces atmospheric
48 visibility but also carries a large number of toxic species into the human lungs,
49 increasing the risks of cardiovascular and cerebrovascular diseases (Chang et al.,
50 2018; Tie et al., 2009; Kong et al., 2019; Zhao et al., 2018; Yang et al., 2015a). Nitrate,
51 sulfate, ammonium, organic matter and elemental carbon are the main compositions of
52 $\text{PM}_{2.5}$, among which nitrate, sulfate and ammonium (NSA) are the main secondary
53 inorganic aerosols in $\text{PM}_{2.5}$ (Ji et al., 2019; Zheng et al., 2016). NSA mainly originates
54 from the secondary aerosols produced by complex chemical reactions of NO_x , SO_2 and
55 NH_3 from coal combustion, vehicle exhaust emissions and agricultural sources (Liu et
56 al., 2013b; Wang et al., 2016; Tian et al., 2017).

57 Because China's current main energy resource is still fossil fuels, which are widely used
58 in industry, for vehicles and residentially, the emission reduction space of NSA is still
59 restricted by a large number of gaseous precursors of NSA (Zhao et al., 2018; Tong et
60 al., 2019). In addition, the chemical conversion of NO₂, SO₂ and NH₃ to form NSA is
61 still very complex, and both homogeneous and heterogeneous reactions involve the
62 chemical conversion of secondary inorganic aerosols, such as photochemical reactions,
63 aqueous phase oxidation environments of aerosols and catalysis of mineral dust (Cheng
64 et al., 2016; Sun et al., 2014; Wang et al., 2016; Ohta and Okita, 1990; He et al., 2014).
65 The formation of sulfate can increase the acidity of aerosols (Sun et al., 2014). In
66 contrast, the presence of NH₃ can play a role in neutralization and maintain the acid-
67 base balance of aerosols (Wang et al., 2016). If improper control measures are taken in
68 pollution reduction control, such as further ammonia emission reduction, the
69 acidification of aerosols and environmental problems of acid rain may be aggravated
70 (Liu et al., 2019c). In addition to the air pollution caused by the local emission of
71 pollutants, the regional transportation of pollutants from its surrounding cities also has
72 an important impact on the urban air quality. Determination of regional transport
73 sources of pollutants, taking regional joint prevention and control measures, and jointly
74 reducing the emissions of pollutants will enable better air control effects, particularly
75 in the Beijing-Tianjin-Hebei region of northern China (Chen et al., 2019a).

76 Higher concentrations of NSA in PM_{2.5} were also found in regions with more serious
77 air pollution in China, such as Beijing-Tianjin-Hebei, the Yangtze River Delta, the Pearl
78 River Delta, the Fenwei Plain, and the Chengdu-Chongqing region (An et al., 2019; Li
79 et al., 2017; Liu et al., 2019d). In response to this situation, the Chinese government
80 issued an Air Pollution Prevention and Control Action Plan (2013-2017) in 2013 to
81 reduce pollutant emissions and improve air quality (the State Council, 2013, last access:
82 June 17, 2020). A large number of treatment measures have been taken regarding coal
83 combustion, motor vehicle emissions and outdated industrial capacities, and by 2017,
84 China's ambient air quality control measures had achieved good results (Liu et al.,

85 2019a;Chen et al., 2019b;Cheng et al., 2019;Li et al., 2019a). In Beijing, PM_{2.5}, NO₂
86 and SO₂ decreased by 35.2%, 17.9% and 69.8%, respectively, in 2017 compared with
87 2013 (Beijing Municipal Ecology and Environment Bureau, 2018, last access: June 17,
88 2020). In Chengdu, PM_{2.5}, NO₂ and SO₂ decreased by 42.3%, 15.9% and 64.5%,
89 respectively, in 2017 compared with 2013 (Chengdu Municipal Ecology and
90 Environment Bureau, 2018, last access: June 17, 2020). To continue to promote air
91 quality improvement, the Chinese government launched the "Three-Year Action Plan
92 for Winning the Blue Sky Defense Battle" in 2018, which puts forward stricter
93 requirements on how to further promote the implementation of emission reduction
94 plans (the State Council, 2018, last access: June 17, 2020). Through long-term
95 observations, a comprehensive analysis of PM_{2.5} chemical compositions and source
96 characteristics is carried out to verify the current implementation effects of emission
97 reduction, and in-depth analyses of pollution reduction control characteristics are of
98 great significance for the next step in air pollution control. However, these analyses
99 may be affected by the experimental equipment, observation stations and other
100 conditions, and the time span of these atmospheric observations usually includes
101 several pollution processes or lasts for weeks or months. Thus, it is difficult to analyse
102 the long-term variations in characteristics of air pollution through comprehensive
103 observation. In particular, there are few high-time-resolution (1 hour) observation
104 experiments carried out with online automatic observation systems (Sun et al., 2013;Tie
105 et al., 2017;Guo et al., 2014). Especially in the Sichuan Basin of southwest China, there
106 are few long-term observational experiments on NSA, which are the main chemical
107 compositions of PM_{2.5}.

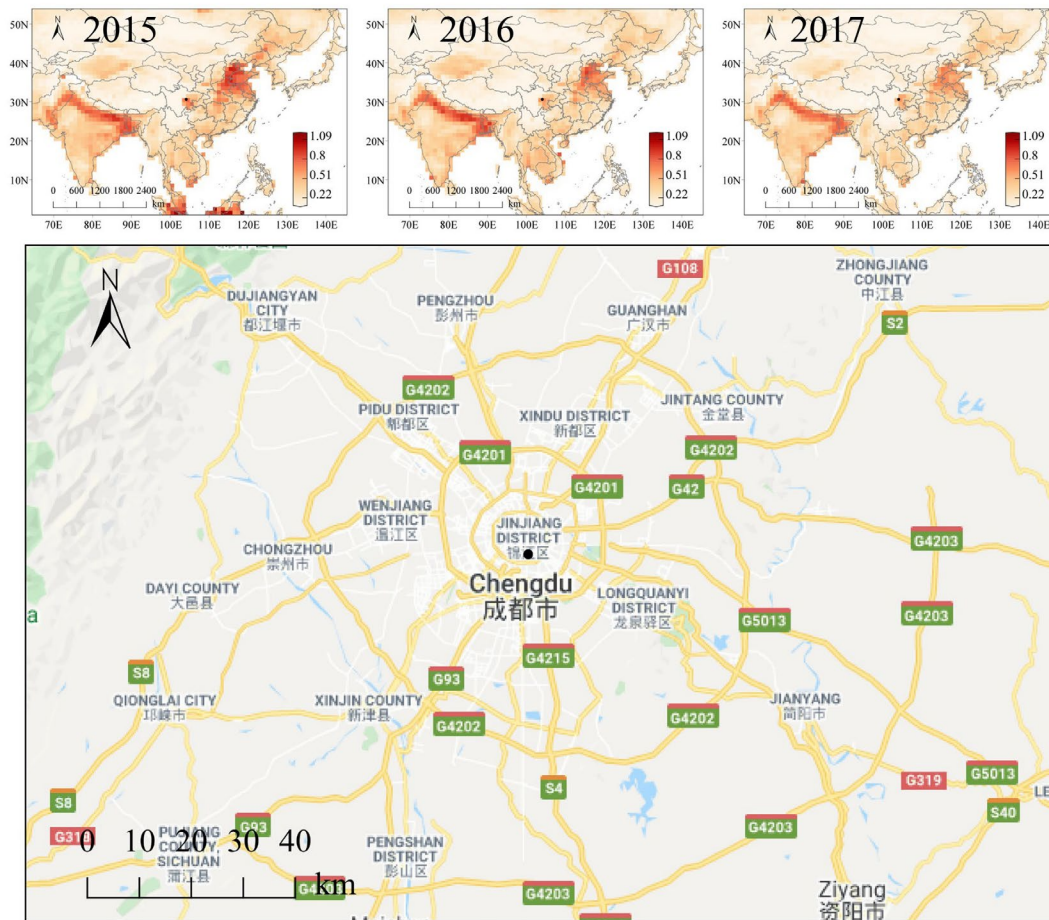
108 The Sichuan Basin is among the most important areas of air pollution in China (Qiao
109 et al., 2019;Gui et al., 2019;Zhong et al., 2019). Although there are many studies in this
110 area, there are few long-term studies of the hourly concentration data resolution of
111 PM_{2.5} chemical compositions. In this study, through three years of observations (from
112 January 1, 2015, to December 31, 2017), we analysed the pollution characteristics of

113 NSA, as well as their formation mechanism and pollution control sensitivity. Finally,
114 combined with local emissions and regional transport characteristics, we analysed the
115 air pollution transport characteristics of Chengdu air pollution.

116 **2 Experiment and methods**

117 **2.1 Observation site**

118 Comprehensive observations were carried out at the Chengdu comprehensive
119 observation station of atmospheric combined pollution (30.63°N, 104.08°E). The
120 observation equipment was placed on the top of a building, approximately 25 m from
121 the ground, and there was no obvious pollution source within approximately 200 m.
122 The site is located in south section 1 of Yihuan Road, Wuhou District, Chengdu (Fig.
123 1), and traffic emission sources may be the main pollution emission source around the
124 observation station. This is a typical residential, traffic and commercial mixed area that
125 represents the characteristics of the urban atmospheric environment. Chengdu is also a
126 megacity in the Sichuan Basin of southwest China, as well as an important part of the
127 Chengdu-Chongqing region, which is among the regions with serious air pollution in
128 China, and as shown in Fig. 1, the Sichuan Basin also has high aerosol optical depth
129 (AOD).



130

131 Fig. 1. Observation site in Chengdu. The image on the top shows the aerosol optical
 132 depth (AOD, 550 nm) from 2015 to 2017 (National Aeronautics and Space
 133 Administration, 2019, last access: June 17, 2020). The black dot in the image on the
 134 bottom shows the location of the observation site in Chengdu (background map from
 135 Google Maps, last access: June 17, 2020).

136 2.2 Instruments

137 During the research period, online experimental monitoring instruments were used to
 138 obtain the observation data with an hourly resolution (1 hour). The equipment list is
 139 shown in Table 1. Data quality control and assurance are an important part of the
 140 atmospheric comprehensive observation experiment, and this result is described in
 141 detail in the supplementary materials (Fig. S1-4).

142 Table 1. The experimental instruments used in this study

Instruments	Parameters	Manufacturer/Country
URG-9000	NO ₃ ⁻ /SO ₄ ²⁻ /NH ₄ ⁺ /Na ⁺ /Mg ²⁺ /Ca ²⁺ /Cl ⁻ /K ⁺	Thermo Fisher Scientific/USA
SHARP 5030	PM _{2.5}	Thermo Fisher Scientific/USA
RT-4	OC/EC	Sunset Laboratory/USA
Xact-625	Metal elements	Cooper Environmental Services /USA
17i/450i/48i/49i	NO _x /NO ₂ /NO/NH ₃ /SO ₂ /CO/O ₃	Thermo Fisher Scientific/USA
WXT520	Meteorological parameters	VAISALA/Germany

OC: organic carbon; EC: element carbon

143 2.3 Chemical conversions and model methods

144 To clarify the conversion of gaseous pollutants to secondary aerosols, the nitrogen
145 oxidation ratio (NOR) and sulfur oxidation ratio (SOR) were used to reflect the
146 conversions of NO₂ and SO₂ to NO₃⁻ and SO₄²⁻, respectively (Sun et al., 2014; Yang et
147 al., 2015a). These ratios can be calculated using Eq. (1) and Eq. (2):

$$148 \text{ NOR} = \frac{n\text{NO}_3^-}{(n\text{NO}_3^- + n\text{NO}_2)} \quad (1)$$

$$149 \text{ SOR} = \frac{n\text{SO}_4^{2-}}{(n\text{SO}_4^{2-} + n\text{SO}_2)} \quad (2)$$

150 where n is the molar concentration.

151 The ISORROPIA-II thermodynamic model was used to analyse the variation in
152 interaction characteristics among aerosol chemical compositions (Fountoukis and
153 Nenes, 2007;Guo et al., 2017a;Ding et al., 2019). Temperature (T), relative humidity
154 (RH) and the concentrations of Na⁺, SO₄²⁻, NH₃, NO₃⁻, Cl⁻, Ca²⁺, K⁺ and Mg²⁺ were
155 input into the ISORROPIA-II thermodynamic equilibrium model. In this study, we used
156 the “forward problems” mode to run the model, assuming that the aerosols were in a
157 “metastable” state (salts do not precipitate under supersaturated conditions). At the time
158 of data input, NH₃ data were the sum of NH₃ and NH₄⁺. Previous studies had shown
159 that the model has better performance when the RH is greater than 30%, and some
160 studies also believe that the model performance is greater than 40%, so this study
161 maintains the RH at higher than 40% when data are input (Ding et al., 2019;Guo et al.,
162 2016). The simulated data and observed data were compared and analysed, and the

163 observation data of NH₃ were consistent with the input data of the model. The linear
164 regression fitting slope of NH₃ was 0.96 (R²=0.98), which showed that the run result of
165 the model had good reliability and performance (Ding et al., 2019). Simultaneously, the
166 aerosol water content (AWC) was calculated, and the sensitivity of the interaction
167 between aerosol chemical compositions (NSA) and the pH of aerosols was analysed
168 (Ding et al., 2019; Fountoukis et al., 2009). The pH was calculated using Eq. (3):

$$169 \text{ pH} = -\log_{10} H_{\text{aq}}^+ \cong -\log_{10} \frac{1000 H_{\text{air}}^+}{\text{AWC}} \quad (3)$$

170 where H_{aq}⁺ (mol/L) is the concentration of hydronium ions in liquid water of
171 atmospheric particulate matter, which can be calculated by the H_{air}⁺ and AWC (μg/m³)
172 outputs from the ISORROPIA-II thermodynamic equilibrium model (Ding et al.,
173 2019; Guo et al., 2017a).

174 **2.4 CPF and PSCF methods**

175 We used the conditional probability function (CPF) to analyse the characteristics of
176 pollutants under the influence of wind direction (WD) and wind speed (WS). The
177 analysis results using CPF were obtained using the R programming language, named
178 openair. This function can be defined as CPF = m_{θ,j}/n_{θ,j}, where m_{θ,j} is the number of
179 samples in the WD interval θ and WS interval j with mixing ratios greater than some
180 ‘high’ pollution concentration, and n_{θ,j} is the total number of samples in the same WD-
181 WS interval (Uria-Tellaetxe and Carslaw, 2014). Usually, a higher given ‘high’
182 pollution concentration (percentile) is chosen, such as the 90th percentile, which will
183 mask the lower percentile pollution concentration source contributions. In this work, to
184 obtain a more complete contribution of pollution sources, a range of percentile values,
185 0-25, 25-50, 50-75 and 75-100 were selected for the CPF calculation.

186 The potential source contribution function (PSCF) is based on an analysis of pollution
187 sources given the air mass backward trajectory and can be used to judge the long-
188 distance regional transport of pollutants (Ji et al., 2019). In this study, MeteoInfoMap
189 and TrajStat (Wang et al., 2009) were used, and the model simulation data input model
190 was provided by the National Oceanic and Atmospheric Administration (National

191 Oceanic and Atmospheric Administration, 2019, last access: June 17, 2020); these data
 192 were calculated to the 24-hour backward trajectories at the observation site at a height
 193 of 500 m every 1 hour from January 1, 2015, to December 31, 2017 (UTC+8). The
 194 calculated domain for PSCF was a range of 20-50° N, 75-115° E, and a grid cell with
 195 a resolution of 0.5°×0.5° was divided. The PSCF could be defined using Eq. (4):

$$196 \quad \text{PSCF}_{ij} = \frac{M_{ij}}{N_{ij}} W_{ij} \quad (4)$$

$$197 \quad W_{ij} = \begin{cases} 1.0 & (N_{ij} \geq 3N_{ave}) \\ 0.7 & (3N_{ave} > N_{ij} \geq 1.5N_{ave}) \\ 0.4 & (1.5N_{ave} > N_{ij} \geq N_{ave}) \\ 0.2 & (N_{ave} > N_{ij}) \end{cases} \quad (5)$$

198 where PSCF_{ij} is the value for the ij th grid cell and M_{ij} is the total number of endpoints
 199 in the ij th grid cell, with pollution concentrations at the observation site (30.63°N,
 200 104.08°E) that are greater than a given threshold value (the 75th percentile was selected
 201 for gaseous pollutants). N_{ij} is the number of backward trajectory endpoints in the ij th
 202 grid cell (0.5°×0.5°) during the simulation period. Therefore, the PSCF reflects the two-
 203 dimensional planar position distribution characteristics of potential sources, not the
 204 three-dimensional characteristics that reflect the transmission of pollution. To reduce
 205 the uncertainty in N_{ij} , the empirical weight function W_{ij} was introduced in Eq. (5),
 206 where N_{ave} is the average of N_{ij} during the simulation period (Ji et al., 2019; Zhang et
 207 al., 2017; Wang et al., 2009).

208 **3 Results and discussion**

209 **3.1 Pollution characteristics of the interannual and entire observation periods**

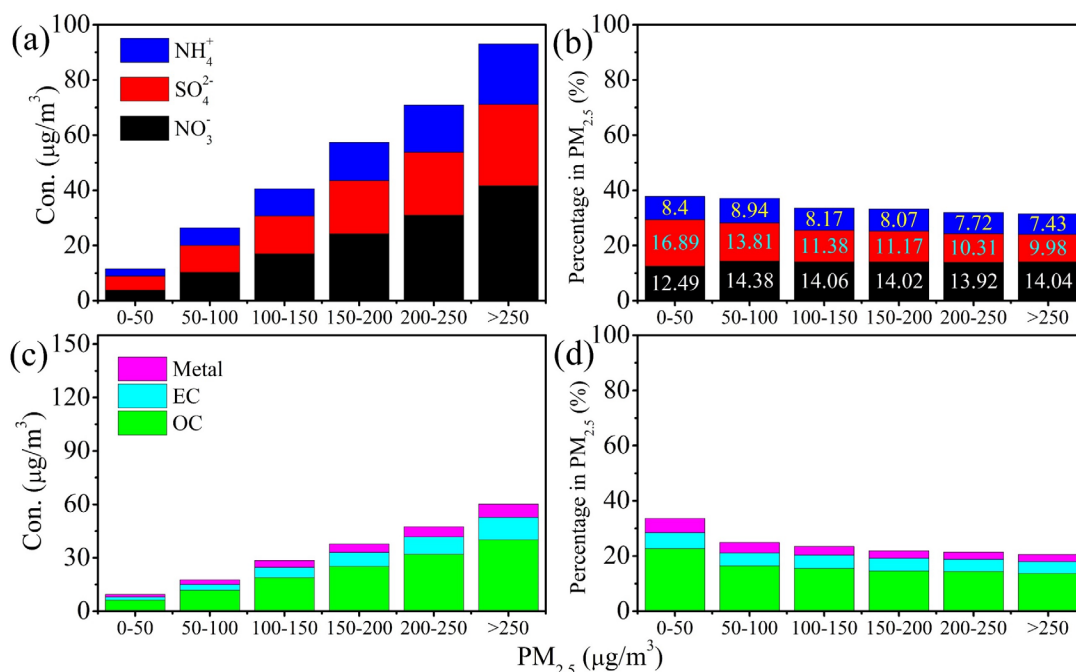
210 The annual average mass concentration of NSA and its proportion in $\text{PM}_{2.5}$ are shown
 211 in Table 2. The annual averages of $\text{PM}_{2.5}$ were 67.78, 71.88 and 59.68 $\mu\text{g}/\text{m}^3$,
 212 corresponding to 2015, 2016 and 2017, respectively. However, the pollution of $\text{PM}_{2.5}$
 213 in Chengdu was much higher than the annual secondary guideline value (35 $\mu\text{g}/\text{m}^3$,
 214 Ambient air quality standards/GB3095-2012) and the World Health Organization
 215 annual guideline value (10 $\mu\text{g}/\text{m}^3$). The same $\text{PM}_{2.5}$ pollution problem was also a
 216 serious problem in Beijing and Nanjing (Ji et al., 2019; Zheng et al., 2019). The annual

217 average mass concentration of NSA also changed significantly, and the difference was
218 large. The Mann-Whitney U test showed that the variation in NO_3^- was nonsignificant
219 ($p > 0.05$), and SO_4^{2-} and NH_4^+ had obvious significance from 2015 to 2017 ($p < 0.05$),
220 indicating that NO_3^- had not decreased significantly, and there was an increase in 2017
221 compared to 2015. SO_4^{2-} continued to decline, and NH_4^+ was also lower in 2017 than
222 in 2016. Notably, SO_4^{2-} and NH_4^+ decreased significantly in 2017 compared with 2015,
223 but the variation in NO_3^- was nonsignificant. Meanwhile, the annual averages of NO_3^-
224 / SO_4^{2-} were 0.95, 1.02 and 1.45 for 2015, 2016 and 2017, respectively, indicating that
225 the contribution of NOx emissions sources to $\text{PM}_{2.5}$ was increased compared with that
226 of SO_2 emissions sources (Li et al., 2017; Wang et al., 2015). As shown in Table S1,
227 from 2013 to 2017, the emissions of NO_2 in Chengdu were obviously higher than those
228 of SO_2 , but $\text{PM}_{2.5}$, NO_2 and SO_2 all decreased due to the implementation of the Air
229 Pollution Prevention and Control Action Plan launched by the Chinese government and
230 a more detailed pollution control plan launched by Sichuan Province. From 2015 to
231 2017, the measures taken by Sichuan Province in the coordinated reduction of multiple
232 pollutants have been continuously strengthened, and the scope of management and
233 control has been continuously expanded, for example, in the improvement of
234 desulfurization, denitrification and dust removal technologies in key industries, from
235 accelerated improvement in 2015 to deeper improvement in 2017. The process of
236 eliminating small coal-fired boilers began in 2015 and was completed in 2017, when
237 the ultra-low-emission coal-fired power plant transformation was promoted. In terms
238 of vehicle emission control, we accelerated the elimination of "yellow label" vehicles
239 (general term for gasoline vehicles with emission levels lower than the national I
240 emission standard and diesel vehicles with emission levels lower than the national III
241 emission standard when new vehicles are finalized) and "old vehicles" (the emission
242 level does not meet the national stage IV emission standard) in 2015 and basically
243 completed the elimination of "yellow label" vehicles in 2017. The quality supervision
244 of oil products has also been improved, and non-road mobile machinery pollution

245 control requirements were proposed in the 2017 plan (The People's Government of
 246 Sichuan Province, 2015, 2016, 2017, last access: June 17, 2020). Compared with 2015,
 247 NO_x and SO₂ decreased by 5.98% and 32.35%, respectively, in 2017, which shows that
 248 the treatment of NO_x and SO₂ emissions has achieved remarkable results, of which the
 249 SO₂ emission reduction effect is the best, followed by that of NO_x. The effect of this
 250 emission reduction is due to air pollution prevention measures, especially measures of
 251 "electricity instead of coal" and "natural gas instead of coal" (refers to increased use of
 252 electricity and natural gas in the residential sector to reduce coal combustion).
 253 Table 2. Comparison of annual mass averages ($\mu\text{g}/\text{m}^3$) and proportions (%) for NSA
 254 (nitrate, sulfate and ammonium) from 2015 to 2017.

	PM _{2.5}	SO ₄ ²⁻	NH ₄ ⁺	NO ₃ ⁻	NO ₃ ⁻ /PM _{2.5}	SO ₄ ²⁻ /PM _{2.5}	NH ₄ ⁺ /PM _{2.5}
2015	67.78	10.37	6.14	9.13	0.129	0.165	0.088
2016	71.88	8.53	6.16	9.27	0.123	0.133	0.089
2017	59.68	6.88	5.01	9.17	0.141	0.132	0.079

255



256

257 Fig. 2. Variation characteristics of the NSA (nitrate, sulfate and ammonium) and other
 258 chemical compositions with different concentrations of PM_{2.5}. (a) NSA mass

259 concentration. (b) Percentage of NSA in PM_{2.5}. (c) Chemical compositions of organic
260 carbon (OC), elemental carbon (EC), and metal elements. (d) Percentage of OC, EC
261 and metal elements in PM_{2.5}.

262 The chemical compositions of PM_{2.5} from 2015 to 2017 varies with concentration, as
263 shown in Fig. 2. With the accumulation of PM_{2.5} in the atmosphere, the concentration
264 of NSA also increased significantly, but the proportion of NSA in PM_{2.5} decreased (Fig.
265 2a and b). The variation trend of OC, EC and metal elements with increasing PM_{2.5}
266 concentration is similar to that of NSA (Fig. 2c and d), and this variation trend of OC
267 and EC is consistent with the results of long-term observation research carried out in
268 Beijing (Ji et al., 2019). With the accumulation of PM_{2.5} concentration, NSA, OC, EC
269 and metal element concentrations have an increasing trend, but their ratio with PM_{2.5}
270 gradually decreases, indicating that other compositions have a higher contribution. This
271 result, on the one hand, maybe since some chemical compositions such as ions and dust
272 have not included in the statistics, on the other hand, the unknown component may also
273 have a high contribution characteristic to PM_{2.5} (Zhang et al., 2015;Huang et al., 2014).
274 For the contribution characteristics of unknown components of PM_{2.5}, studies in some
275 regions of China show that the contribution of higher PM_{2.5} concentration is higher than
276 that of lower PM_{2.5} concentration (Huang et al., 2014;Li et al., 2017;Geng et al., 2019).
277 When PM_{2.5} was less than 50 $\mu\text{g}/\text{m}^3$ and greater than 250 $\mu\text{g}/\text{m}^3$, the mass
278 concentrations of NSA were 11.57 and 90.06 $\mu\text{g}/\text{m}^3$, respectively, and the proportions
279 were 37.78 and 31.45%, respectively. Comparing Fig. 2b and d, it was found that NSA
280 was always the main contributor in the entire process of PM_{2.5} accumulation, which
281 was significantly higher than the proportions of OC and EC (Ji et al., 2019;Li et al.,
282 2019b). In the accumulation process of PM_{2.5} concentrations greater than 50 $\mu\text{g}/\text{m}^3$,
283 NO₃⁻ accounts for a high proportion in NSA and is stable at approximately 14%, and
284 the proportion of SO₄²⁻ and NH₄⁺ continues to decrease (Li et al., 2019b;Wang et al.,
285 2016). When the PM_{2.5} concentration was less than 50 $\mu\text{g}/\text{m}^3$, the concentration of SO₄²⁻
286 was higher than that of NO₃⁻, and the concentration of NH₄⁺ was lower than the NH₄⁺

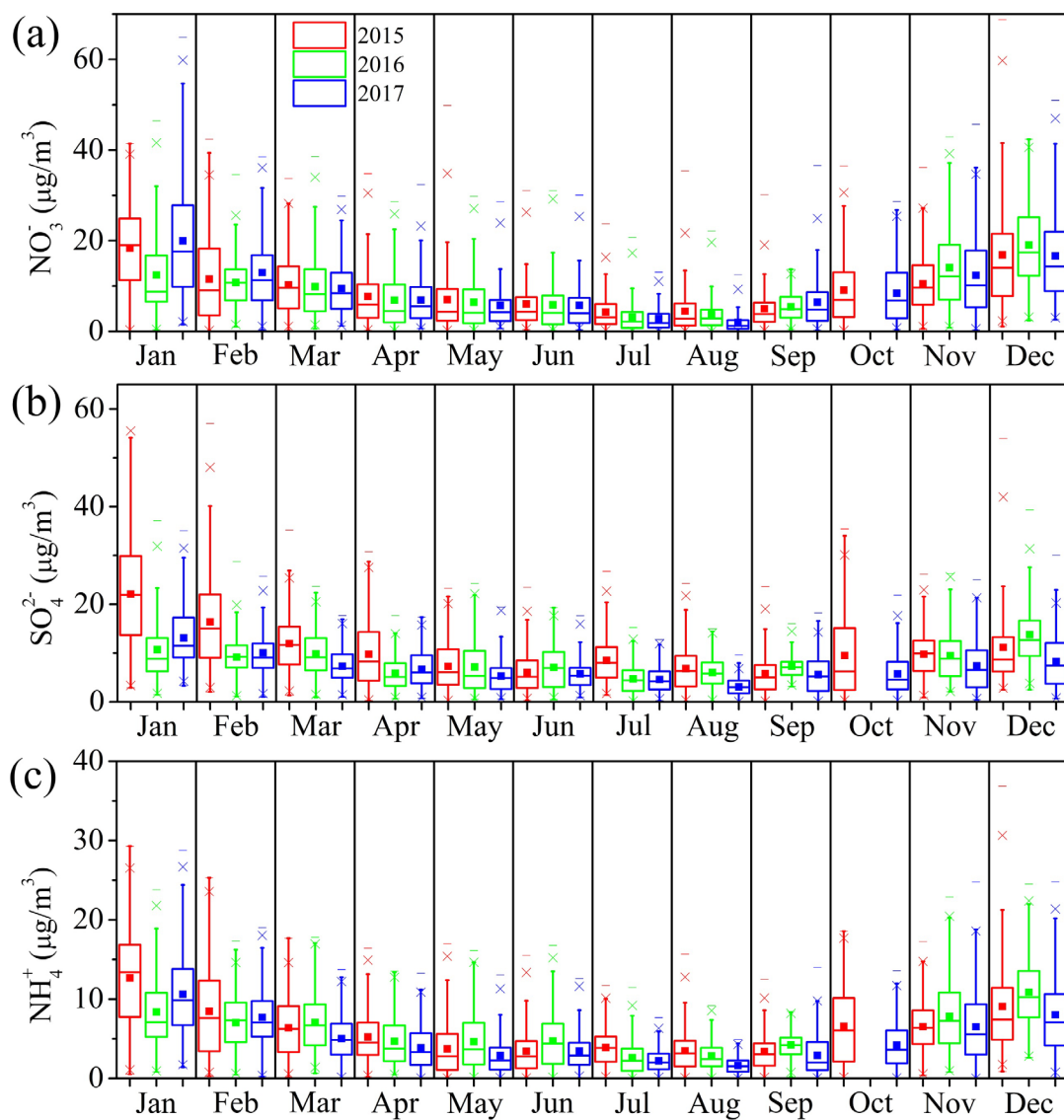
287 concentration of PM_{2.5} at 50 to 100 μg/m³, possibly due to SO₄²⁻ concentration being
288 higher than the NO₃⁻ concentration, forming more chemically stable (NH₄)₂SO₄ (Guo
289 et al., 2017a). In addition, when the PM_{2.5} was less than 50 μg/m³, low RH and strong
290 solar radiation were also important ways to generate sulfate (Yao et al., 2018).

291 **3.2 Monthly and seasonal variations**

292 The monthly variation characteristics of NSA from 2015 to 2017 are shown in Fig. 3.
293 At the beginning and end of each year, the pollutant concentration is relatively high and
294 relatively low in the middle of each year. The meteorological conditions also have
295 obvious monthly variation characteristics (Fig. S5 a and b); from April to August, they
296 have higher WS and lower RH, which is not only conducive to the dilution and diffusion
297 of pollutants but also reduces the chemical conversions of pollutants by aqueous phase
298 and influences the formation of secondary inorganic aerosols (Wang et al., 2016; Ji et
299 al., 2019). Overall, the concentrations are higher in January and December and lower
300 in July and August. The highest monthly average NO₃⁻ reached 19.98 μg/m³ in January
301 2017, and the highest monthly average SO₄²⁻ and NH₄⁺ were 22.08 μg/m³ and 12.66
302 μg/m³ in January 2015, respectively. The lowest concentrations of NSA appeared in
303 August 2017, which were 1.96, 3.07 and 1.62 μg/m³. The gaseous precursors of NSA
304 also have obvious monthly variations, and the NO_x and SO₂ trends were similar to those
305 of NO₃⁻ and SO₄²⁻ (Fig. 3 and 4). NH₃ emissions were significantly different, with
306 increases in warmer months (April-July) and colder months (September-December).
307 On the one hand, NH₃ volatilization was promoted by relatively high Ts (Fig. S5c); on
308 the other hand, the use of agricultural fertilizers and livestock farming were also
309 important sources of NH₃ in China. Second, from urban region, fossil fuel combustion
310 and motor vehicle emissions also contribute significantly (Liu et al., 2013b; Pan et al.,
311 2016). Notably, NH₃ increased significantly from April to December 2017 compared
312 with 2015 and 2016, especially during low-T months (Fig. 4c). The results of an
313 analysis of the monthly concentration variation of pollutants indicate that the
314 implementation of pollution reduction and control measures should be strengthened at

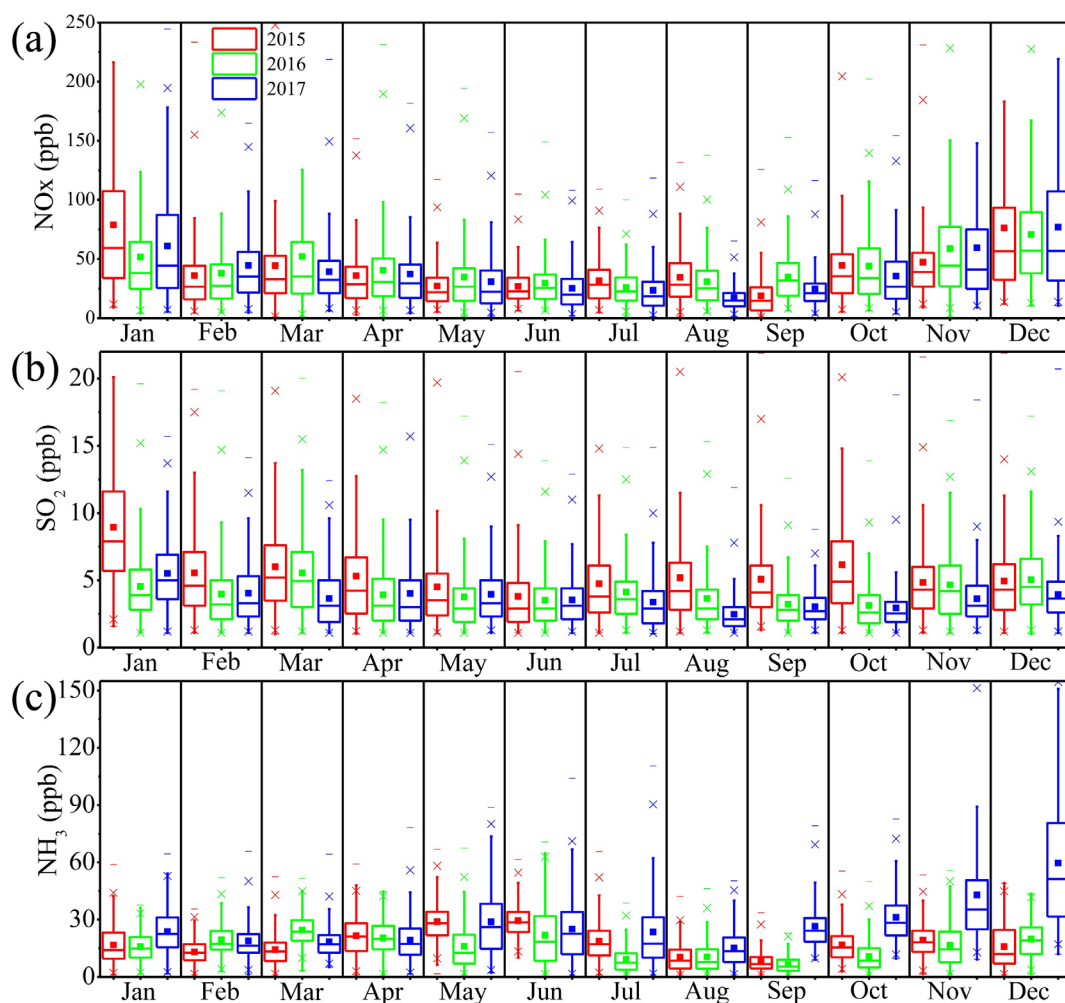
315 the beginning of each year (January to March) and the end of the year (October to
316 December).

317 The seasonal variation in NSA is shown in Fig. S6, and the concentration in winter was
318 much higher than that in summer. NO_3^- only declined in spring and summer from 2015
319 to 2017, with an increase in autumn and winter (Fig. S6a). Seasonal variations in NH_4^+
320 were similar to those of NO_3^- , with higher concentrations in winter and the lowest in
321 summer (Fig. S6c). This may be because higher Ts and WSs can not only promote the
322 decomposition of NH_4NO_3 in summer but also promote the dilution and diffusion of
323 pollutant concentrations (Guo et al., 2017a; An et al., 2019). There is a significant
324 downward trend in SO_4^{2-} , which continues to decrease in spring, summer and winter
325 from 2015 to 2017 (Fig. S6b). In autumn, the concentration was the highest in 2016,
326 and it was significantly lower in 2017 than in 2015 and 2016. The variation amplitude
327 of NSA and gaseous pollutants in cold months was significantly higher than that in
328 warm months (Figs. 3, 4 and S6). This higher variation amplitude may be due to the
329 differences in pollutant accumulation and scavenging processes. This finding also
330 indicates that the instability of local pollutant emissions and regional transport during
331 cold months was affected by meteorological conditions (Li et al., 2017; Ji et al., 2018).
332 The large variation amplitude of pollutants in different months, similar to the changes
333 in the Beijing-Tianjin-Hebei region of northern China and Chengdu, are due to the
334 accumulation and removal of pollution by meteorological conditions and pollutant
335 emissions (Ji et al., 2019; Qin et al., 2019; Zhang et al., 2019a).



336

337 Fig. 3. Monthly variations in NSA (nitrate, sulfate and ammonium) concentrations from
 338 2015 to 2017. (a) NO_3^- . (b) SO_4^{2-} . (c) NH_4^+ .



339

340 Fig. 4. Monthly variations in NO_x, SO₂ and NH₃ concentrations from 2015 to 2017. (a)

341 NO_x. (b) SO₂. (c) NH₃.

342 3.3 Diurnal and weekly variations

343 From 2015 to 2017, the concentration of NSA was higher in the daytime than in the

344 evening (Fig. 5a), and similar results were found in different seasons (Fig. 5b), which

345 may be due to the combination of pollutant emissions and meteorological conditions.

346 As shown in Fig. S7, from 9:00 to 11:00 a.m., the concentrations of SO₂, NO_x, NH₃

347 and CO increased significantly, indicating that the primary emission of pollutants was

348 relatively strong. At this time, although RH is in a declining stage, it still has a relatively

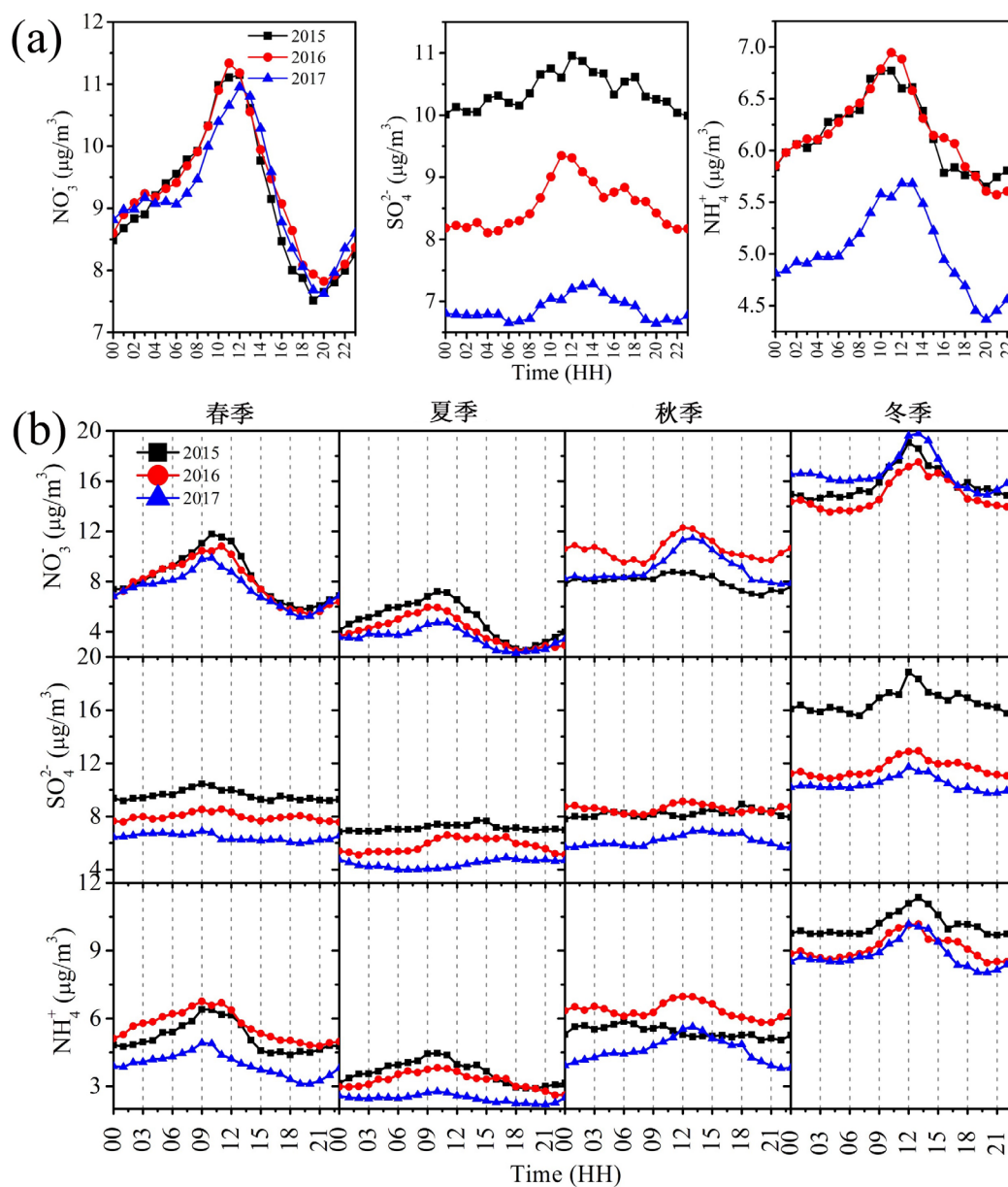
349 high atmospheric humidity (approximately 65%), and O₃ and NO₂/NO also

350 occasionally show an increasing trend, indicating that the atmospheric oxidizability has

351 also increased (Figs. S7 and S8). This situation also provides favourable conditions for

352 the formation of secondary aerosols and promotes the accumulation of NSA (Cheng et
353 al., 2016; Wang et al., 2016; Sun et al., 2014). In addition, before 10 o'clock, relatively
354 low WS will enable easy pollutant concentration accumulation. In contrast, the higher
355 WS in the afternoon may be the main factor for the decrease in pollutant concentration
356 (Figs. 5 and S8). Photochemical reactions may also be one of the factors in the
357 formation of NSA, and the concentration of O₃ peaks at approximately 15:00, which
358 may be affected by the free radicals generated by photochemistry. At approximately
359 19:00, the ratio of NO₂/NO reached its highest value, and the concentration of NO₂ also
360 increased significantly (Song et al., 2018; Zhu et al., 2019). At night, with the increase
361 in RH (Fig. S8), dissolved ozone, free radicals, hydrogen peroxide and NO₂ can catalyse
362 SO₂ to form secondary aerosols through an aqueous phase reaction (Zhang et al.,
363 2015; An et al., 2019). The seasonal diurnal variation in NSA is shown in Fig. 5b. The
364 concentration of NSA in winter was obviously higher than that in summer, and the
365 diurnal variation range was larger. The concentrations in spring and autumn were closer,
366 but the diurnal variation in spring was larger than that in autumn. The larger diurnal
367 variation range not only indicates serious pollution but also indicates the importance of
368 other factors affecting air quality, such as meteorological conditions and secondary
369 aerosol conversion conditions (Ji et al., 2019; Yang et al., 2015b). The peak value of the
370 NSA seasonal diurnal variation also varies in different seasons. The peak value appears
371 at approximately 13:00 in winter, approximately 10:00 in spring and summer, and
372 approximately 12:00 in autumn, possibly due to the influence of meteorological
373 conditions. In previous studies in Beijing-Tianjin-Hebei and the Pearl River Delta, the
374 concentration of pollutants was affected by meteorological factors, and it was usually
375 lower in the daytime than at night. In the Yangtze River Delta, the peak usually occurs
376 in the morning, but in our study, the concentration was higher in the daytime than at
377 night (Peng et al., 2011; Wang et al., 2018; Guo et al., 2017b). In addition to the diurnal
378 variations in WS and atmospheric humidity, some studies have shown that due to the
379 unique topographical structure of the Sichuan Basin, the atmospheric circulation

380 between the Qinghai-Tibet Plateau, Yunnan-Guizhou Plateau and Sichuan Basin and
 381 the meteorological conditions of the Chengdu region are affected, such as the
 382 characteristics of air mass transport and typical “night rain” (more precipitation at night
 383 than in the day) under the influence of atmospheric circulation (Zhang et al.,
 384 2019c;Zhang et al., 2019b).



385
 386 Fig. 5. Diurnal variations in NSA (nitrate, sulfate and ammonium) from 2015 to 2017.
 387 (a) Annual average. (b) Seasonal average.

388 The weekly variation in NSA is shown in Fig. S9. During the overall observational
 389 period, workdays (Monday to Friday) showed higher variations than weekends

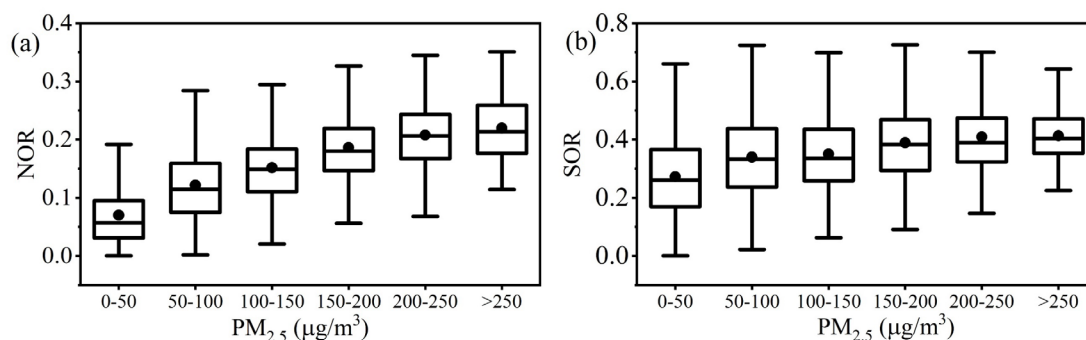
390 (Saturday and Sunday), with the highest variation being on Tuesday and the lowest
391 being on Sunday. Despite the difference in mean values between Tuesday and Sunday,
392 nonparametric tests show that the difference in mean values was nonsignificant (Mann-
393 Whitney U test, $P > 0.05$). As shown in Fig. S9, the average trends of NO_3^- and NH_4^+
394 were consistent from Monday to Sunday. The correlation coefficient was 0.94 ($P < 0.01$)
395 from 2015 to 2017, which indicates that they have a common source and that vehicle
396 emissions also have an important contribution to NH_4^+ (Pan et al., 2016). The average
397 NO_3^- , SO_4^{2-} and NH_4^+ concentrations from 2015 to 2017 were 9.21, 8.64 and 5.64 $\mu\text{g}/\text{m}^3$
398 on workdays and 8.56, 8.33 and 5.29 $\mu\text{g}/\text{m}^3$ on weekends, respectively. The average
399 values of NO_x , SO_2 and NH_3 were 42.43, 4.35 and 20.39 ppb on weekdays and 39.60,
400 4.34 and 19.67 ppb on weekends, respectively. Similarly, the mean difference between
401 NSA and gaseous precursors (NO_x , SO_2 and NH_3) was not significant by the Mann-
402 Whitney U test on weekdays and weekends. Population standard deviation comparisons
403 of NO_3^- , SO_4^{2-} and NH_4^+ showed that workdays had higher standard deviations than did
404 weekends, with 7.96, 6.04 and 4.35 on weekdays and 6.76, 5.69 and 3.88 on weekends,
405 respectively, and it could also be seen from the box chart of NSA weekly variation that
406 the concentration range on working days was slightly larger than that on weekends (Fig.
407 S10). Analysis of the diurnal variation in NSA gaseous precursors on weekdays and
408 weekends shows that the variation trend is relatively consistent (Fig. S11), and the
409 concentration of NO_x on weekdays will be slightly higher at the peak of 9:00 to 10:00
410 than on weekends, which may be affected by the morning rush hour of vehicles. In this
411 study, NSA and gaseous precursors are also slightly higher on weekdays than on
412 weekends, which indicates that in Chengdu's air pollution prevention and control
413 actions, the management of relevant industries and departments should be strengthened
414 on weekdays.

415 **3.4 Chemical characteristics of NSA**

416 **3.4.1 Chemical conversion characteristics of NSA**

417 Fig. 6 shows the abilities of NO_2 and SO_2 to chemically convert to NO_3^- and SO_4^{2-} at

418 different $PM_{2.5}$ concentrations. With the increase in $PM_{2.5}$ concentration, NOR and SOR
 419 gradually increased, indicating that the formation ability of NO_3^- and SO_4^{2-} increased
 420 during the formation of air pollution. In this study, when the $PM_{2.5}$ concentration was \leq
 421 $50 \mu\text{g}/\text{m}^3$, the average NOR and SOR were 0.07 and 0.27, respectively, and when the
 422 $PM_{2.5}$ concentration was greater than $250 \mu\text{g}/\text{m}^3$, the average NOR and SOR increased
 423 to 0.22 and 0.41, respectively, indicating that the chemical conversion and formation
 424 ability of secondary inorganic aerosols was obviously enhanced when air pollution was
 425 aggravated. Previous studies suggested that when NOR and SOR were greater than 0.1
 426 and 0.2, respectively, it has intense conversions and forms secondary inorganic aerosols
 427 (Yang et al., 2015a).

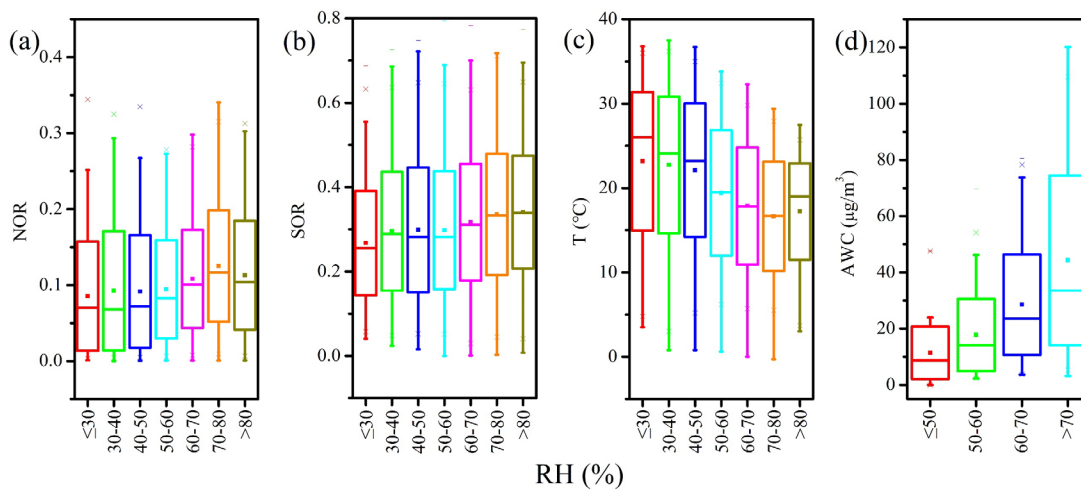


428

429 Fig. 6. Analysis of atmospheric chemical conversion ability at different $PM_{2.5}$
 430 concentrations. (a) NOR (nitrogen oxidation ratio). (b) SOR (sulfur oxidation ratio).

431 Fig. 7 shows the variation characteristics of NSA chemical conversions with increasing
 432 RH. NOR and SOR increased with increasing RH, suggesting that NO_2 and SO_2 were
 433 more likely to produce NO_3^- and SO_4^{2-} under higher RH conditions. Previous studies
 434 have shown that the presence of NH_3 and NO_2 can promote the chemical conversion of
 435 SO_2 to SO_4^{2-} in the aqueous phase (Wang et al., 2016). In an aerosol aqueous phase
 436 environment, alkaline aerosol (NH_3) components can promote the dissolution of SO_2
 437 and the formation of SO_4^{2-} under the oxidation of NO_2 (Cheng et al., 2016). Especially
 438 when the atmosphere is polluted, the formation of SO_4^{2-} by SO_2 through the aqueous
 439 phase environment can contribute most of the SO_4^{2-} (Sun et al., 2013). When the RH is
 440 greater than 80%, the NOR appears to decline, possibly because HNO_3 is semivolatile,

441 and the T increases at this time (Fig. 7 c), which is not conducive to the condensation
 442 of gaseous HNO₃ to particulate matter, which affects the amount of NO₃⁻ in PM_{2.5} (Guo
 443 et al., 2017a). According to the ISORROPIA-II thermodynamic equilibrium model
 444 simulation, AWC also increases with RH (Fig. 7 d), and the increase in AWC can
 445 provide a liquid environment for aerosols, which is conducive to the dissolution and
 446 conversion of gaseous precursors of NO₂, SO₂ and NH₃ and promotes the formation of
 447 more NSA. The Pearson's correlation coefficients of RH and NOR and SOR were 0.12
 448 and 0.16 (p<0.01), and the AWC and NOR and SOR were 0.73 and 0.37 (p<0.01),
 449 respectively, showing a significant positive correlation, indicating that the increase in
 450 AWC may be beneficial to the conversion of NO₂ and SO₂ to NO₃⁻ and SO₄²⁻. As shown
 451 in Fig. S12, the simulated values of NSA (metastable state, liquid phase components)
 452 are compared with the observed data. The linear regression fitting slope is
 453 approximately 1 (p<0.01), indicating that the effect of the liquid phase environment in
 454 PM_{2.5} is obvious; in addition, stable state simulation is also performed, and the linear
 455 regression fitting slopes of the NSA liquid phase state data output from the model and
 456 the observation data are 0.73, 0.63 and 0.74, and the Pearson's correlations are 0.82,
 457 0.71 and 0.80 (p<0.01), indicating that they are more often combined with AWC in the
 458 aerosol aqueous phase environment at a stable state. Previous studies have also
 459 confirmed that the aqueous phase environment of aerosols plays an important role in
 460 the formation of secondary inorganic aerosols (Wang et al., 2016; Cheng et al., 2016).



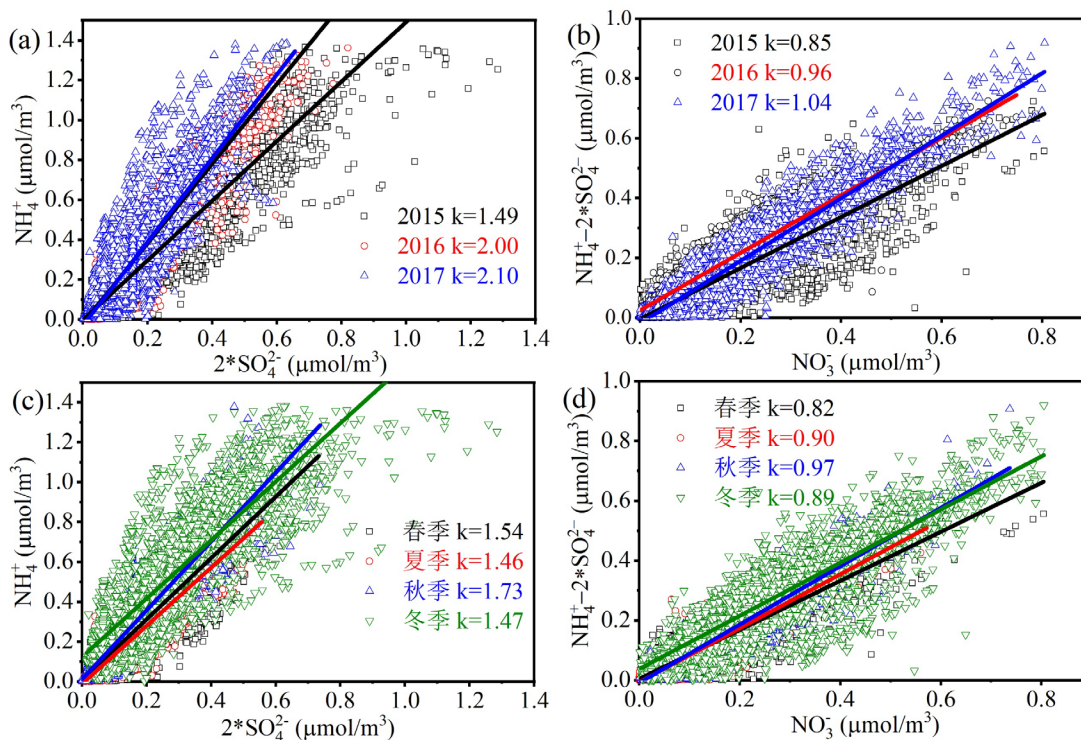
461

RH (%)

462 Fig. 7. Effects of RH on the chemical conversion of NSA (nitrate, sulfate and
463 ammonium). (a) NOR (nitrogen oxidation ratio). (b) SOR (sulfur oxidation ratio). (c)
464 Temperature (T). (d) AWC (aerosol water content).

465 **3.4.2 Sensitivity analysis**

466 The molar ratio analysis of NSA shown in Fig. 8 was used to analyse the chemical
467 relationships among NSA. $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 are mainly composed of NH_4^+ ,
468 SO_4^{2-} and NO_3^- in particulate matter (Malm and Hand, 2007; Meier et al., 2009).
469 Because $(\text{NH}_4)_2\text{SO}_4$ has better stability than NH_4NO_3 , NH_4^+ will first combine with
470 SO_4^{2-} and then with NO_3^- (Liu et al., 2012). The annual average molar ratio of NH_4^+ to
471 $2*\text{SO}_4^{2-}$ was more than 1, which indicates that SO_4^{2-} can be completely neutralized by
472 NH_4^+ (Fig. 8a). The molar ratios of residual NH_4^+ ($\text{NH}_4^+ - 2*\text{SO}_4^{2-}$) to NO_3^- were 0.85,
473 0.96 and 1.04 in 2015, 2016 and 2017, respectively. As shown in Fig. 8a and b, the
474 gradual increase in the ratio (slope k) from 2015 to 2017 indicates that there is an
475 increase in NH_4^+ in aerosol compared with SO_4^{2-} and NO_3^- , especially in 2017, with a
476 ratio of 1.04, indicating the presence of other forms of NH_4^+ , such as NH_4Cl and
477 $(\text{NH}_4)_2\text{C}_2\text{O}_4$ (Sun et al., 2006). Seasonal variations in NH_4^+ , SO_4^{2-} and NO_3^- are shown
478 in Fig. 8c and d. The higher molar ratio in autumn indicates that the intensity of
479 ammonia emission in autumn was higher than that in other seasons. This result also
480 shows that the proportion of NH_4^+ relative to NO_3^- and SO_4^{2-} in $\text{PM}_{2.5}$ has increased.
481 Therefore, while currently controlling NO_x and SO_2 emissions, it is also necessary to
482 strengthen NH_3 emissions control.



483

484 Fig. 8. Molar ratio analysis of NSA (nitrate, sulfate and ammonium). (a) Interannual
 485 variation in the molar ratio of SO_4^{2-} and NH_4^+ . (b) Interannual variation in the molar
 486 ratio of NO_3^- and NH_4^+ . (c) Seasonal variation in the molar ratio of SO_4^{2-} and NH_4^+ . (d)
 487 Seasonal variation in the molar ratio of NO_3^- and NH_4^+ . k: Fitting slope of linear
 488 regression.

489 Table 3 shows the sensitivity analysis of the concentration variations in SO_4^{2-} , NO_3^- and
 490 NH_4^+ . ISORROPIA-II thermodynamic equilibrium model sensitivity analysis is
 491 described in detail in the Supplementary Materials. The coefficient of variance
 492 represents the response of the species to variations in other components. The
 493 coefficients of variance for NH_4^+ and NO_3^- produced by SO_4^{2-} changes were 52.22 and
 494 1.70, respectively. Similarly, the coefficients of variance for NH_4^+ and SO_4^{2-} produced
 495 by NO_3^- changes were 51.42 and 0.0005, respectively. The large coefficient of variance
 496 for NH_4^+ indicates that the changes in NO_3^- and SO_4^{2-} can affect the presence of NH_4^+ ,
 497 which also indicates that NH_4NO_3 and $(\text{NH}_4)_2\text{SO}_4$ were the main states of NH_4^+ (Liu et
 498 al., 2012). The coefficients of variance for SO_4^{2-} and NO_3^- produced by TNH_3
 499 ($\text{NH}_3+\text{NH}_4^+$) changes were 0.47 and 15.76, respectively, and the effect of TNH_3 on

500 SO_4^{2-} was less than that of NO_3^- , which indicates that NH_4^+ was excessive to SO_4^{2-} and
 501 that NH_4^+ first combines with SO_4^{2-} to form stable $(\text{NH}_4)_2\text{SO}_4$, and the remaining NH_4^+
 502 and NO_3^- will combine to form NH_4NO_3 .

503 Table 3. Sensitivity analysis of NSA (nitrate, sulfate and ammonium) concentration
 504 variations during the different observation periods.

Period	Variation	Coefficients of variance		
		NO_3^-	NH_4^+	SO_4^{2-}
2015-2017	NO_3^-		51.42	0.0005
	TNH_3	15.76		0.47
	SO_4^{2-}	1.70	52.22	

Coefficients of variance: standard deviation/average *100%;

Variation TNH_3 : $\text{NH}_3+\text{NH}_4^+$ ($\mu\text{g}/\text{m}^3$);

Variation SO_4^{2-} and NO_3^- units: $\mu\text{g}/\text{m}^3$

505 Through the implementation of the Air Pollution Prevention and Control Action Plan,
 506 the reduction in SO_4^{2-} in $\text{PM}_{2.5}$ has achieved good results. Therefore, while continuing
 507 to promote "electricity instead of coal" and "natural gas instead of coal" to reduce coal
 508 combustion pollution, more stringent control measures should be added for NO_3^- and
 509 NH_4^+ emissions. To further improve air quality, the Chinese government launched a
 510 "Three-Year Action Plan for Winning the Blue Sky Defense Battle" in 2018 and
 511 proposed emission reduction targets for NO_x and SO_2 emissions, which will be 15%
 512 lower in 2020 than in 2015 (the State Council, 2018, last access: June 17, 2020). The
 513 results of using the ISORROPIA-II thermodynamic equilibrium model to simulate NO_3^- ,
 514 SO_4^{2-} and TNH_3 emission reduction control effects of 5%, 10%, 15% and 20%,
 515 respectively, are shown in Table S3, showing that controlling the concentration of NO_3^-
 516 and SO_4^{2-} is also helpful to reduce the concentration of NH_4^+ and indicating that
 517 controlling its precursor NO_x and SO_2 is of great significance to reduce the secondary
 518 inorganic aerosol in $\text{PM}_{2.5}$ (the detailed results are described in the supplementary
 519 materials). Previous studies have also shown that the conversion of SO_2 to SO_4^{2-} in the

520 aqueous phase not only increases the conversion of SO_4^{2-} but also enhances the
521 formation of NO_3^- in the aqueous phase (Wang et al., 2016). Therefore, SO_2 emission
522 reduction may play a key role in the process of controlling emission reduction in NSA
523 pollution, as it not only reduces the presence of NH_4^+ ($(\text{NH}_4)_2\text{SO}_4$) in particulate matter
524 but also affects the formation of NH_4NO_3 by influencing the formation of NO_3^- . NO_2
525 and NH_3 can also promote the conversion of SO_2 to SO_4^{2-} through an aqueous phase
526 environment (Wang et al., 2016). Therefore, priority control of NO_x and SO_2 emissions
527 is an important way to reduce NSA in particulate matter.

528 The increase in NSA can increase the hygroscopicity properties of aerosols, and more
529 AWC can increase the pH by diluting the hydrogen ion concentration (Kong et al.,
530 2020;Ding et al., 2019). Previous studies have also shown that SO_4^{2-} formation reduces
531 aerosol pH (Sun et al., 2014). The effects of NO_3^- , SO_4^{2-} and TNH_3 on pH when using
532 the ISORROPIA-II thermodynamic equilibrium model to simulate pollutant
533 concentration reduction are shown in Table S3. With the decrease in NO_3^- and SO_4^{2-} ,
534 the pH value increases, but NO_3^- has no obvious effect on the pH value, SO_4^{2-} has an
535 obvious effect on the pH value, which indicates that the formation of SO_4^{2-} in the
536 aerosol can increase the acidity of the aerosol (Sun et al., 2014). The greater the
537 reduction of TNH_3 , the lower the pH value is, which shows that the presence of NH_3 as
538 an alkaline gas can alleviate some of the acidity produced by SO_4^{2-} (Cheng et al., 2016).
539 When the synergistic control of pollutants is reduced, it also has a certain effect on pH,
540 increasing from 4.07 to 4.16. Some studies believe that if ammonia emissions are
541 reduced significantly, the risk of acid rain may increase (Liu et al., 2019c). As shown
542 in Fig. S13, the acid rain problem in China is mainly concentrated in southern China,
543 especially in southwestern China, southern China, and the Yangtze River Delta in
544 eastern China. Therefore, how to adjust the emission reduction ratio in combination
545 with the characteristics of regional air pollution and energy consumption and thus help
546 reduce the problem of aerosol acidity changes caused by air pollution reduction is a
547 problem worthy of in-depth study. Therefore, when controlling NO_x , SO_2 and NH_3

548 emissions, it is necessary to consider the aerosol acid and alkali changes caused by
549 emission reduction.

550 **3.5 Characteristics of local emissions and regional transport**

551 **3.5.1 Local emissions**

552 The concentration of pollutants is obviously affected by meteorological conditions; for
553 example, WS and WD can affect the accumulation and removal of pollutants (Li et al.,
554 2016). Figs. S14-16 show the annual variation characteristics of NSA and gas
555 precursors affected by the WS and WD using CPF. Overall, the higher WS was
556 accompanied by a lower pollutant concentration. As the WS decreased, the pollution
557 became serious, and the pollution hot spots were gradually concentrated. On the whole,
558 when the WS was usually greater than 2 m/s, the pollution was light (pollutant
559 concentration percentile was between 0-25). When WS was usually less than 1 m/s, the
560 pollution was heavy (pollutant concentration percentile was between 75-100), which
561 also reflects the distance and orientation between the emission source and the
562 observation station, indicating that when the pollution was serious, the contribution of
563 local source emissions was more prominent.

564 NO_3^- and NO_x have similar distributions of pollution hot spots in the polar plot diagram
565 (Fig. S14), and when the concentration percentile was 0-25, the hot spots were
566 concentrated in the northeast and southeast directions and widely distributed. When the
567 concentration percentile was 25-75, the sources of NO_3^- and NO_x were distributed west,
568 southwest and northeast of the observation site, and there were important contribution
569 sources in the northwest direction (WS was approximately 3-4 m/s) in 2017. When the
570 WS was approximately 1-2 m/s and the concentration percentile was 50-75, the
571 important NO_x source was in the northwest direction. When the accumulation of
572 pollution concentration was high (concentration percentile was 75-100), the NO_3^-
573 source was mainly concentrated in the east and southeast of the observation site, and
574 NO_x was distributed in the south and southeast, with WSs of less than 1 m/s.
575 Additionally, the distribution of pollution hot spots was relatively wide in 2016 (the

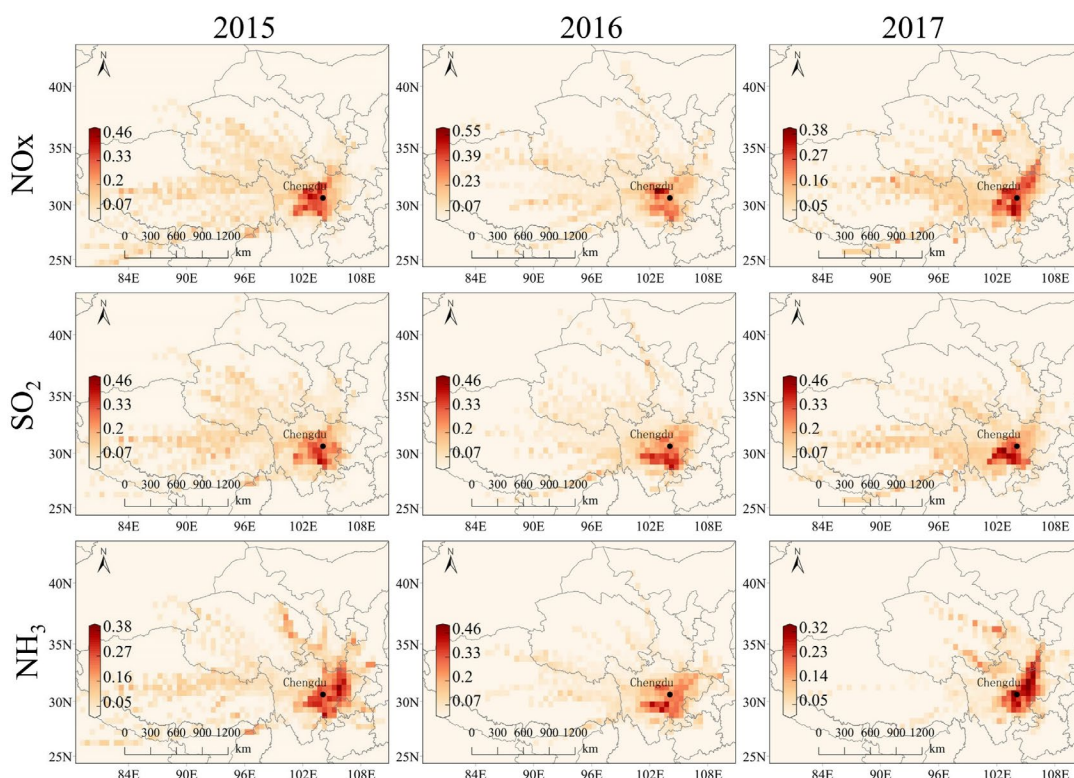
576 annual mean values of NO_x were 42.15, 43.99 and 39.63 ppb in 2015, 2016 and 2017,
577 respectively), indicating that the source was relatively wide, which may be one of the
578 reasons for the relatively high concentration. The SO₄²⁻ and SO₂ pollution sources
579 affected by meteorological conditions also have similar distribution characteristics (Fig.
580 S15). At a higher concentration of pollutants, the pollution hot spots of SO₄²⁻ were
581 distributed in the east and southeast of the observation site, and SO₂ was distributed in
582 the northeast, southeast and west. Compared with 2017 and 2016, the distribution of
583 SO₂ pollution sources in 2016 was also more extensive, mainly in the west and northeast.
584 The NH₃ emissions were slightly different from those of NO_x and SO₂ (Fig. S16).
585 Under conditions of high pollution concentration (concentration percentile was 75-100),
586 the pollution hot spots were distributed in the west in 2015 (WS was approximately 2-
587 3 m/s), in the north in 2016 (WS was approximately 1 and 3 m/s), and in the near
588 distance in 2017 (WS was approximately 0.5 m/s). The higher pollution concentration
589 was accompanied by a relatively higher WS (2015 and 2016), which indicates that the
590 NH₃ emission transport in the surrounding area was more obvious, which may come
591 from the surrounding agricultural source distribution area (Liu et al., 2019b; Liu et al.,
592 2013b). The annual mean value of NH₃ emissions in 2017 was 27.91 ppb, which is
593 significantly higher than those in 2015 and 2016 at 17.93 ppb and 16.55 ppb,
594 respectively. During the 25-50 concentration percentile period of NH₃, there was a WS
595 of approximately 2 m/s east of the observation site, and during the 50-75 concentration
596 percentile period, there was an obvious source northwest of the observation site, with a
597 WS of approximately 4 m/s. During the 75-100 concentration percentile periods, the
598 pollution sources were mainly local. This shows that in 2017, in addition to the pollution
599 sources being distributed in the east and northwest, the higher NH₃ emissions were also
600 contributed by the surrounding emission sources northwest of Chengdu.

601 **3.5.2 Gaseous precursors of NSA regional transport**

602 The PSCF is used to analyse the potential source distribution of pollutants to determine
603 the regional transport characteristics of pollutants (Ji et al., 2019). In addition,

604 considering the aerosol lifetime, SO₂ (approximately 9.6 d) and NO_x (approximately 1
605 d) are also very different (Guo et al., 2014), and the research also shows that NH₃ is
606 significantly contributed by local source emissions (Walker et al., 2004). Therefore, we
607 comprehensively consider selecting a 24-hour backward trajectory to carry out PSCF
608 simulation in the Chengdu region. Fig. 9 shows the PSCF analysis of NO_x, SO₂ and
609 NH₃, with significant differences in their potential source distributions. The higher
610 PSCF value of NO_x was mainly distributed west, northwest and southwest of Chengdu
611 in 2015, northwest and south of Chengdu in 2016, and south, west and northeast of
612 Chengdu in 2017. The PSCF of NO₂ and NO (Fig. S17) also reflects that their potential
613 sources are mainly influenced by the interior of Sichuan Province, especially in the
614 cities around Chengdu. Chengdu is located along the western margin of the Sichuan
615 Basin, and it was also observed through satellite remote sensing data that the higher
616 NO₂ emissions in the Sichuan Basin were distributed in Chengdu and Chongqing (Fig.
617 S18a). The SO₂ emissions were widely distributed, mainly in the Sichuan Basin. Among
618 them, Leshan city and Meishan city south of Chengdu had higher SO₂ emissions, and
619 another higher emission source was distributed in Chongqing (Fig. S18b). The PSCF
620 analysis of SO₂ showed that the higher PSCF values were distributed in southern,
621 western and southwestern Chengdu. Therefore, a comparison of Figs. 9 and S18b shows
622 that the main source of SO₂ may be distributed in the southern, western and
623 southwestern margin region of the Sichuan Basin. In particular, Leshan, Ya'an and
624 Meishan were important potential sources. As shown in the PSCF analysis of NH₃ in
625 Fig. 9, the higher PSCF was also concentrated in the inner Sichuan Basin, especially in
626 the urban agglomeration around Chengdu. In 2015, the potential source of NH₃ was
627 mainly distributed in the southwest and northeast of Chengdu, with higher PSCF in
628 Nanchong and other regions. In the southwest, it was concentrated in Ya'an, Meishan
629 and Leshan. In 2016, potential sources were mainly distributed in the southwest of Ya'an,
630 Meishan, Leshan and the southern part of the Ganzi Tibetan Autonomous Prefecture.
631 There were two characteristics of potential sources in 2017. A relatively light source

632 was relatively close to Chengdu, and the high PSCF was in Chengdu, which can be
633 considered the contribution of local emissions. The other contribution is obvious as a
634 long-distance potential source contribution, mainly in some cities in the northeast,
635 Nanchong, Guangyuan and Mianyang, and to a certain extent at the junction of Shaanxi,
636 Gansu and Sichuan. In 2017, in addition to the contribution of local emissions, the
637 contribution of regional transport in the northeast may also be an important reason for
638 the higher NH₃ concentration. Fig. S19 shows satellite remote sensing data of NH₃.
639 Overall, the higher NH₃ column concentration is distributed in the Sichuan Basin,
640 mainly concentrated in the region near Chengdu, showing that NH₃ is more discharged
641 in the Sichuan Basin, especially in the surrounding areas of Chengdu. In addition,
642 through the analysis of the Multiresolution Emission Inventory for China (MEIC), it is
643 also found that the higher NO_x, SO₂ and NH₃ emissions in Sichuan Basin are mainly
644 concentrated in Sichuan Basin, as shown in Fig. S20. It can also be seen that NO_x is
645 mainly concentrated in more-developed Chengdu and Chongqing, SO₂ emissions are
646 obvious in Chengdu and Western Chongqing, NH₃ emissions are widely distributed,
647 and there are higher emission characteristics in Chengdu and its surrounding areas.
648 Therefore, according to the analysis of pollution emissions and PSCF in Chengdu, it is
649 necessary to strengthen regional air pollution control and take regional joint prevention
650 and control measures to reduce the impact of air pollutant regional transport.



651

652 Fig. 9. PSCF (potential source contribution function) of NO_x, SO₂ and NH₃ in Chengdu
 653 from 2015 to 2017.

654 4 Conclusions

655 The three-year observation experiment with hourly resolution of NSA from January 1,
 656 2015 to December 31, 2017 was carried out in Chengdu in southwest China, which is
 657 in the Sichuan Basin. The pollution characteristics of NSA's annual, monthly, seasonal,
 658 diurnal and weekly variations were demonstrated. The characteristics of chemical
 659 conversion and the sensitivity of emission reduction control were analysed. Finally,
 660 combined with meteorological parameters and PSCF simulation, the local emission and
 661 regional transport characteristics of NSA gaseous precursors were also illustrated. The
 662 main conclusions were as follows:

663 (1) With the increase in PM_{2.5} concentration, the NSA mass concentration increased,
 664 accounting for 31.45-37.78% of PM_{2.5}, and the contribution of NSA was higher than
 665 that of carbon aerosol (OM and EC). From 2015 to 2017, the contribution of NO₃⁻ to
 666 PM_{2.5} increased, and in 2017, it became the main contribution component of NSA, and

667 it plays an important role in the concentration accumulation of PM_{2.5}. Higher and lower
668 NSA concentrations were seen in winter and summer, respectively, and higher
669 concentrations were seen more during the day than at night. Although the NSA
670 concentration on weekdays was slightly higher than that on weekends, the mean
671 difference between them was nonsignificant.

672 (2) With the increase in PM_{2.5} concentration, there is an increasing trend of NOR and
673 SOR, which indicates that the formation of NO₃⁻ and SO₄²⁻ increases obviously, and the
674 increase in RH will promote the formation of NO₃⁻ and SO₄²⁻. Using the ISORROPIA-
675 II thermodynamic equilibrium model, it is found that NSA in aerosols is more likely to
676 combine with AWC, which indicates that the aqueous environment of aerosols plays an
677 important role in promoting the formation of NSA. The analysis of the interaction
678 between NSA also confirmed that NH₄⁺ will first combine with SO₄²⁻ to form
679 (NH₄)₂SO₄, and the remaining NH₄⁺ will combine with NO₃⁻ to form NH₄NO₃. The
680 sensitivity analysis of NSA concentration shows that reducing NO_x and SO₂ is
681 beneficial to reducing NSA contribution in PM_{2.5}, but their changes also have an
682 important impact on the pH of aerosols.

683 (3) Local emissions and regional transport of NSA gaseous precursors have an
684 important impact on air pollution in Chengdu. When pollution is aggravated, the
685 contributions of NO_x and SO₂ to local emissions are relatively obvious. In addition to
686 the local emission of NH₃, the contribution of pollution sources around Chengdu is also
687 relatively obvious. PSCF analysis shows that the potential sources of pollution
688 transmission in Chengdu are mainly distributed in Sichuan Province, and the most
689 prominent contribution is made in Sichuan Basin, especially among the cities around
690 Chengdu. The analysis of local emissions and regional transport shows that it is
691 necessary to implement joint prevention and control of air pollution in the Sichuan
692 Basin.

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697 **Data availability**

698 The data are available on request to the corresponding author.

699 **Author contribution**

700 XL, QT and LK designed and led this study. QT and MF were responsible for the
701 observations. LK, MF, YL, YZ, CZ, and CL analysed the data. LK, YQ, JA, NC, YD,
702 RZ and ZW discussed the results. LK and XL wrote the paper. All authors commented
703 on the paper.

704 **Competing interests**

705 The authors declare that they have no conflicts of interest.

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