

Response to reviews on manuscript acp-2019-1094 “Evaluating the impact of blowing snow sea salt aerosol on springtime BrO and O₃ in the Arctic”

We thank both reviewers for their insightful and constructive comments. We list our responses in *blue* below.

Responses to Referee #1

Major Points

- The authors discussion of improving the model’s ability to reproduce ODEs suggest 4 potential sources of difficulty in line 518, but then the paper’s abstract and the rest of the conclusion only focus on the inclusion of a snowpack molecular halogen production mechanism.

While I agree that this important mechanism should be included in models, the authors could do a better job explaining how the other three sources mentioned might impact the modeled ODEs and why they chose to focus the bulk of the discussion on this mechanism in particular.

We have modified the abstract to include these 4 potential sources of difficulty, instead of just focusing on snowpack chemistry (also see our response to Referee #2). We have also modified the conclusions to include more discussion of these other sources of error.

- One other potential explanation for the model doing a better job with BrO than O₃ is the evaluation via column measurements rather than concentrations. The ozone loss rate is dependent on the BrO concentration (among other things, [Thompson et al., 2017]), but the measured BrO VCD reflects both concentration and vertical profile of BrO [Sihler et al., 2012]. Thus, the same VCD could have very different implications for ozone depletion near the surface depending on the vertical profile of BrO. This issue wouldn’t necessarily be resolved by adding another Br₂ source. While satellite-based measurements are a needed tool to evaluate performance over large spatial scales, it would also be good, in the future, to evaluate the model using ground-based BrO concentration measurements as the authors did with ozone.

Yes, indeed, evaluation of this simulation against ground-based BrO is another useful approach and is one that our colleagues at the University of Alaska and University of Florida are examining (Swanson et al., 2019).

Swanson, W., Confer, K., Holmes, C.D., Huang, J., Jaeglé, L., and Simpson, W.R., Evaluating GEOS-Chem with snow aerosol source against ground observations of Arctic reactive bromine events, Abstract A51H-2758 presented at 2019 Fall Meeting, AGU, San Francisco, CA, 9-13 Dec.

- Section 4.1: I found this section overly qualitative, particularly when compared to the rest of the paper. This issue also pops up in line 33 of the abstract. I would encourage the authors to come up with a more quantitative description of the fraction of ODEs observed at Arctic sites captured by the model, which would strengthen this section of the paper, and allow the authors to avoid phrases like "captures a few" and "misses some" which seem a bit out of place in a scientific publication.

Point well taken. We have revised the comparison between observed and modeled surface O₃ to be more quantitative:

“Figure 8 shows that FYI+MYI Snow simulation reproduces only 25-30% of the ODEs at

Utqiagvik and Alert, such as the 8 – 18 April and 26 April – 10 May depletion events at Utqiagvik as well as the April events at Alert. The model predicted magnitude of O₃ depletion at those sites is a factor of two lower than observed. The FYI+MYI Snow simulation performs somewhat better at Zeppelin, where it captures the timing and magnitude of 40% of the observed events, in particular between late March and late April. Our simulation thus misses 60-75% of observed ODEs at those three Arctic sites, in particular several March and late May events at Utqiagvik, as well as the sustained May events at Alert, and the late May ODEs at Zeppelin.

We have also modified the abstract and conclusions accordingly.

- Section 4.2: This section would potentially be improved by a discussion of the spatial extent of the modeled ODEs and comparison to prior studies. As an example, Halfacre et al. [2014] used buoy-based observations to suggest ODEs can have a spatial extent on the order of 100s of km, is this finding reflected by the model?

We have added the following sentences to this section to address this:

“The median aerial extent of ODEs ($\Delta O_3 > 20$ ppbv) simulated in the FYI+MYI Snow simulation is 0.35 million km² (horizontal extent of ~330 km) with some of the largest ODEs extending over areas of 1.5-7 million km² (~700-1500 km horizontal extent). Our results are consistent with the 282 km median size of major ODEs inferred by Halfacre et al. (2014) using ice-tethered buoys combined with back-trajectories.”

Minor points

- The units for vertical column density are molecules per cm². To my knowledge, the omission of molecules is not an appropriate convention.

We have added molecules to the units in the text and figures.

- Regarding Fig. 3a comparing the GOME and OMI measurements, is the slope calculation from a typical linear regression or an orthogonal distance regression? Since the measurements have comparable uncertainties, the residuals in both x and y should be minimized when determining the line of best fit.

We use the reduced-major-axis regression line to calculate the slope (as noted in the Figure caption, line 912). This is a symmetrical regression method assuming both variables are dependent and have errors. For Fig. 3a, the reduced major axis regression yields the same results as an orthogonal reduced major Axis regression.

- Line 331: Define what you mean by high wind speed $>10-12 \text{ m s}^{-1}$. *This has been clarified in the revised text.*

- Line 371: Missing reference for Cl acid displacement enhancing ODEs *We have added a reference to Keene et al. (2007).*

- Line 477: This propagation of bromine inland has been observed [Peterson et al., 2018]. *We have added a reference to Peterson et al. (2018) in the revised manuscript.*

- Fig. 4,8,10,12 Readability would be improved though the use of variable line styles as well as color. Figure 8 in particular is not readable.

We have worked to improve the readability of these figures by increasing the line thickness and changing the colors.

- Fig. 8 Perhaps move the full timeseries to a supplement and change the x axis to only show Mar 15th through April 15th to encompass the green shaded regions at all sites.

We have chosen to keep the full timeseries for this figure.

Responses to Referee #2

- Abstract (lines 34-36): I doubt your suggestion of inclusion of snowpack Br activation will definitely fix the model difficulty in reproducing local ODEs. I cannot see how you derive this conclusion, as in your manuscript (lines 366-374 and lines 518-520), you pointed out several potential factors that could cause model failure in reproducing local ODEs, which are (1) coarse resolution, (2) lack of chlorine chemistry, (3) difficulty in modelling boundary layer processes and (4) lack of snowpack emission. The model horizontal resolution used in this study is 2X2.5 degree, which is quite coarse. We know that model cannot capture any sub-grid scale events. If we assume a mean wind speed of 20 km per hour, then any transported event with a lifetime of around or less than ~10 hrs will be smoothed and cannot be reproduced by the model. In which case, it demands a high resolution model simulation. In addition, the poor model ability in reproducing shallow boundary layer, especially in polar regions (which is a common issue in most current models) is another killer. As the air exchange rate at the top of the boundary layer largely determines near surface ozone level. Therefore, if the model cannot reproduce well the dynamical process of the boundary layer, it will not reproduce well near surface ozone. For these reasons, I recommend the authors to discuss model limitations, rather than pointing to other possibilities.

Point well taken. We have modified the abstract to address this and now have a more complete description of other possibilities: “This difficulty in reproducing observed surface ODEs could be related to the coarse horizontal resolution of the model, the known biases in simulating Arctic boundary layer exchange processes, the lack of detailed chlorine chemistry, and/or the fact that we did not include direct halogen activation by snowpack chemistry.” We have also expanded the discussion of these other potential sources of error in the conclusions.

- Line 67: ‘Four’ or ‘three classes’?

Yes, we have corrected this in the revised manuscript.

- Line 333-334: A recent modelling work (by Rhodes et al., 2017) focusing on SSA simulation also showed a similar conclusion that when multiyear sea ice is considered as a source of SSA, modelled SSA has the lowest mean standard deviation across the Arctic sites comparing to other experiments.

We have added a reference to Rhodes et al. (2017) in the revised manuscript.

- Lines 526-628, you mentioned that ‘... in the shallow boundary layer (~50-250m) over the springtime Arctic direct snowpack halogen activation dominates Br release and is responsible for the most severe ODEs’, do you have any in situ data to support this

statement? In my point of view, the snowpack mechanism (as a dominant source of bromine in ODEs) is still a hypothesis without direct in situ data to support. If ODEs are long-distance transported in association with synoptic systems, why cannot the observed events in the shallow boundary layer be transport-related?

We recognize that our statement was too strong and have modified it as follows: "... in the shallow boundary layer (~50-250 m) over the springtime Arctic direct snowpack halogen activation could contribute to Br release and potentially be responsible for the most severe ODEs." The support for this statement is in the sentence following the sentence highlighted by the reviewer, in which we cite three studies: "Indeed, some of the strongest ODEs observed at the surface seem to occur when the stable Arctic boundary layer is decoupled from convective exchange with the free troposphere (Moore et al., 2014; Seabrook et al., 2011; Wang et al., 2019a)."

Evaluating the impact of blowing snow sea salt aerosol on springtime BrO and O₃ in the Arctic

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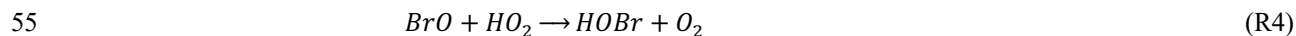
Abstract. We use the GEOS-Chem chemical transport model to examine the influence of bromine release from blowing snow sea salt aerosol (SSA) on springtime bromine activation and O₃ depletion events (ODEs) in the Arctic lower troposphere. We evaluate our simulation against observations of tropospheric BrO vertical column densities (VCD_{tropo}) from the GOME-2 and OMI spaceborne instruments for three years (2007-2009), as well as against surface observations of O₃. We conduct a simulation with blowing snow SSA emissions from first-year sea ice (FYI, with a surface snow salinity of 0.1 psu) and multi-year sea ice (MYI, with a surface snow salinity of 0.05 psu), assuming a factor of 5 bromide enrichment of surface snow relative to seawater. This simulation captures the magnitude of observed March-April GOME-2 and OMI VCD_{tropo} to within 17%, as well as their spatiotemporal variability ($r=0.76-0.85$). Many of the large-scale bromine explosions are successfully reproduced, with the exception of events in May, which are absent or systematically underpredicted in the model. If we assume a lower salinity on MYI (0.01 psu) some of the bromine explosions events observed over MYI are not captured, suggesting that blowing snow over MYI is an important source of bromine activation. We find that the modeled atmospheric deposition onto snow-covered sea ice becomes highly enriched in bromide, increasing from enrichment factors of ~ 5 in September-February to 10-60 in May, consistent with freshly fallen snow composition observations. We propose that this progressive enrichment in deposition could enable blowing snow-induced halogen activation to propagate into May and might explain our late-spring underestimate in VCD_{tropo}. We estimate that atmospheric deposition of SSA could increase snow salinity by up to 0.04 psu between February and April, which could be an important source of salinity for surface snow on MYI as well as FYI covered by deep snowpack. Inclusion of halogen release from blowing snow SSA in our simulations decreases monthly mean Arctic surface O₃ by 4-8 ppbv (15-30%) in March and 8-14 ppbv (30-40%) in April. We reproduce a transport event of depleted O₃ Arctic air down to 40° N observed at many sub-Arctic surface sites in early April 2007. While our simulation captures 25-40% of the ODEs observed at coastal Arctic surface sites, it underestimates the magnitude of many of these events and entirely misses 60-75% of ODEs. This difficulty in reproducing observed surface ODEs could be related to the coarse horizontal

35 resolution of the model, the known biases in simulating Arctic boundary layer exchange processes, the lack of detailed chlorine chemistry, and/or the fact that we did not include direct halogen activation by snowpack chemistry.

1 Introduction

Ozone depletion events (ODEs) are often observed in the springtime Arctic boundary layer (Barrie et al., 1988; Bottenheim et al., 2009, 1986; Bottenheim and Chan, 2006; Halfacre et al., 2014; Koo et al., 2012; Oltmans et al., 2012; Oltmans and Komhyr, 40 1986). While ODEs occur episodically at coastal Arctic sites, lasting 1-3 days, they are more persistent and widespread over the frozen ocean (e.g. Simpson et al., 2007 and references therein). These ODEs have been linked to the release of significant levels of bromine radicals, in a phenomenon called bromine explosion (Abbatt et al., 2012; Simpson et al., 2007b). Both ground-based and satellite instruments have reported elevated BrO columns over the Arctic during March and April (Choi et al., 2012, 2018; Hönninger and Platt, 2002; Jones et al., 2009; Liao et al., 2012; Neuman et al., 2010; Peterson et al., 2017; 45 Salawitch et al., 2010; Simpson et al., 2017; Theys et al., 2011; Wagner et al., 2001). In addition to ODEs in surface air during spring, the elevated bromine radicals result in the rapid oxidation of mercury and its deposition to snowpack (Ebinghaus et al., 2002; Schroeder et al., 1998; Steffen et al., 2008), which can have significant impacts on the health of people and wildlife in polar regions (AMAP, 2011).

An autocatalytic cycle involving the heterogeneous release of sea salt bromide (Br^-) via uptake of HOBr was suggested to be 50 the primary cause of bromine explosions and ODEs (Abbatt et al., 2012; Fan and Jacob, 1992; Hausmann and Platt, 1994; Simpson et al., 2007b):



Reaction (R1) is a multiphase reaction, which takes place on acidic aerosols, cloud droplets, and snow grains. This reaction is required for the initial release of bromine to the atmosphere, but also maintains high levels of BrO by allowing fast recycling 60 of Br^- to reactive bromine (Fan and Jacob, 1992; Lehrer et al., 2004). In addition, the heterogeneous reaction of HOBr with sea salt chloride (Cl^-) can release BrCl from the condensed phase (Abbatt et al., 2012).

In the global troposphere, inorganic bromine (Br_y) has three major sources: debromination of sea salt aerosol (SSA) produced by breaking waves in the open ocean, photolysis and oxidation of bromocarbons, and transport of Br_y from the stratosphere. Release of Br^- from oceanic SSA is estimated to be the largest global source of tropospheric Br_y (Sander et al., 2003; Zhu et al., 2019). Current global models that include these three sources of tropospheric bromine as well as multiphase reactions such as (R1), cannot reproduce the observed elevated levels of tropospheric BrO over polar regions during spring (Parrella et al., 2012; Schmidt et al., 2016). Four classes of substrates specific to polar regions have been proposed as a source of Br^- in reaction (R1): salty snowpack on sea ice and coastal regions (McConnell et al., 1992; Simpson et al., 2005; Toyota et al., 2011), first-year sea ice (Frieß et al., 2004; Nghiem et al., 2012; Wagner et al., 2007), SSA produced from frost flowers (Kaleschke et al., 2004; Rankin et al., 2002), and SSA produced from blowing salty snow (Jones et al., 2009; Yang et al., 2008).

Both laboratory and outdoor chamber experiments have detected Br_2 production when acidified surface saline snow was exposed to sunlight (Pratt et al., 2013; Wren et al., 2013). However, no Br_2 production was detected over sea ice and brine icicles, suggesting that brines and frost flowers on new sea ice surfaces are not a bromine activation source (Pratt et al., 2013). The proposed role of frost flowers as a direct source of SSA is disputed by several field and laboratory experiments showing that frost flowers are difficult to break even under strong wind conditions (e.g., Alvarez-Aviles et al., 2008; Roscoe et al., 2011; Yang et al., 2017).

A number of studies have focused on the role of aerosols, in particular SSA from blowing snow, as a key mechanism to initiate and sustain bromine activation aloft. Yang et al. (2008) proposed blowing snow SSA, produced after sublimation of wind-lofted salty snow on sea ice, as a source of BrO . This source is consistent with satellite observations of large-scale BrO column enhancements over Arctic and Antarctic sea ice, which are often associated with high surface wind speeds that can generate blowing snow events (Begoïn et al., 2010; Blechschmidt et al., 2016; Choi et al., 2018; Jones et al., 2009). Using ground-based measurements in coastal Alaska, Frieß et al. (2004) found that periods of enhanced BrO were coincident with an increase in aerosol extinction at higher wind speeds ($> 5 \text{ m s}^{-1}$), suggesting that halogen activation and/or recycling takes place in situ on SSA aerosol produced by sublimation of dispersed snow grains or frost flowers. Peterson et al. (2017) and Simpson et al. (2017) reported aircraft observations of the vertical distribution of BrO and aerosol extinction consistent with initial activation on snowpack followed by transport aloft (500-1000 m), where high BrO was sustained by recycling on aerosols. More recently, Frey et al. (2019) and Giordano et al. (2018) reported direct observations of SSA production from blowing snow above sea ice. In particular, Frey et al. (2019) found that the Br^-/Na^+ ratio of blowing snow SSA observed at 29 m above the ground decreased by a factor of 2-3 relative to observations at 2 m, suggesting rapid Br^- release via (R1) from blowing snow SSA.

There are only a few modeling studies which have examined the link between blowing snow SSA and bromine explosions. In their pioneering study, Yang et al. (2010) implemented the Yang et al. (2008) blowing snow parameterization in the p-TOMCAT model and presented a comparison to satellite retrievals of BrO columns from the Global Ozone Monitoring

Experiment (GOME) for 2 months. Their evaluation showed qualitative agreement with observations over polar regions for those months. They et al. (2011) describe two 3-day case studies over the Arctic and Antarctic comparing the p-TOMCAT simulation to tropospheric BrO columns from the second Global Ozone Monitoring Experiment (GOME-2). There was qualitative agreement with the location and timing of the two BrO explosion events. However, They et al. (2011) note that beyond these case studies there were large discrepancies between model and observations, reflecting uncertainties in the p-TOMCAT Br⁻ emissions, such as assumptions about snow salinity and fraction of Br⁻ in SSA released to the gas phase. Zhao et al. (2016) used the UM-UKCA chemistry-climate model to simulate a bromine explosion case study initiated by a blowing snow event and transported to the Canadian high Arctic. While the model reproduced the vertical extent of the BrO plume, it was not as successful at capturing the timing or magnitude of the observed BrO. Choi et al. (2018) demonstrated a strong spatial and temporal correlation between bromine explosions observed by OMI and blowing snow SSA as simulated in the GEOS-5 modeling system. However, their study did not include a simulation of bromine photochemistry.

Here, we use the GEOS-Chem global chemical transport model to further quantify the role played by blowing snow SSA in springtime Arctic bromine explosions and examine the associated tropospheric O₃ depletion. We systematically evaluate a 3-year GEOS-Chem simulation (2007-2009) via comparisons to satellite retrievals of tropospheric BrO vertical column densities (VCD_{tropo}) from GOME-2 and the Ozone Monitoring Instrument (OMI). We also compare the model to in situ surface observations of O₃ at several Arctic and sub-Arctic sites. Section 2 describes the GEOS-Chem simulations and the observations used in this study. We evaluate the model's capability to reproduce the timing, magnitude, and spatial extent of bromine explosion events observed by GOME-2 and OMI in Section 3. In section 4, we examine the impact of blowing snow SSA on surface O₃. We assess the contribution of atmospheric deposition to surface snow salinity and bromide content in Section 5. Conclusions are presented in Section 6.

2 Observations and model simulations

2.1 Satellite observations of tropospheric BrO vertical column densities

The Second Global Ozone Monitoring Experiment (GOME-2) is a nadir-scanning UV/visible spectrometer on the METOP-A satellite, which was launched on 19 October 2006 in a sun-synchronous polar orbit with an equator crossing time of 9:30 LT (Munro et al., 2006). The GOME-2 spectrometer covers the 240-790 nm wavelength region, with a spectral resolution between 0.26 and 0.51 nm. It has a ground pixel size of 80 km × 40 km and a scanning swath of 1920 km. In this study, we use the daily 2007-2009 GOME-2 VCD_{tropo} retrieved by They et al. (2011). Briefly, the VCD_{tropo} are derived from a residual technique that combines measured slant columns and calculated stratospheric columns, accounting for the impact of clouds, surface reflectivity and viewing geometry on the measurement sensitivity. The stratospheric BrO contribution is removed using the dynamic climatology based on the BASCOE CTM as described in They et al. (2009, 2011). The tropospheric slant columns are converted to vertical columns with a tropospheric air mass factor (AMF) assuming different BrO profile shapes depending on surface albedo. For low surface albedo (<0.5), a Gaussian shape BrO profile with a maximum at 6 km is used. For high

surface albedo (>0.5), the assumed tropospheric BrO concentration profile is constant in the first km above the Lambertian surface reflector. For this second case, which characterizes most of the ice- and snow-covered Arctic, there is a high sensitivity to BrO close to the surface and a weak dependence of the tropospheric AMF to the shape of the profile. As in Theys et al. (2011), we only consider retrievals for solar zenith angles (SZA) less than 80° and cloud fractions below 0.4 for which the pressure difference between the surface and the top of the cloud less than 400 mbar.

We also use the BrO VCD_{tropo} from the Ozone Monitoring Instrument (OMI) onboard the Aura satellite, which was launched on 15 July 2004 in a sun-synchronous polar orbit with a 13:30 LT equator overpass time (Levelt et al., 2006). OMI is a nadir solar backscatter spectrometer that measures ultraviolet–visible wavelengths (270–500 nm), and has a horizontal resolution of $13 \text{ km} \times 14 \text{ km}$ and swath width of 2600 km. Since 2008, the OMI swath coverage has been reduced due to an external obstruction. We use here the daily 2007–2009 VCD_{tropo} , which were retrieved by Choi et al. (2018) for the months of March and April poleward of 60°N . The total BrO slant column densities are derived from the OMI total BrO (OMBRO) product, by fitting a model function to OMI UV backscattered radiance at 319–347.5 nm (Choi et al., 2018). The VCD_{tropo} are retrieved using the same residual technique and dynamic climatology stratospheric BrO columns as in Theys et al. (2011). In the OMI tropospheric AMF calculation, the assumed BrO shape profile is based on a composite of aircraft measurements obtained during the NASA Arctic Research of the Composition of the Troposphere from Aircraft and Satellites (ARCTAS) campaign (Choi et al., 2012). We use the same selection criteria as in Choi et al. (2018): $\text{SZA} < 80^\circ$, surface reflectivity > 0.6 , and only retain observations with low cloud contamination (difference between OMI rotational Raman cloud pressure and terrain pressure $< 100 \text{ hPa}$). The uncertainty of OMI and GOME-2 VCD_{tropo} retrieved over the highly reflective surfaces of polar regions is on the order of 30–50%.

2.2 Surface O_3 measurements

We use hourly in situ O_3 measurements at three Arctic surface sites: Utqiagvik, Alaska (also known as Barrow, 71.3°N , 156.6°W , 8 m above sea level) from the NOAA Earth System Research Laboratory, Alert, Nunavut (82.5°N , 62.5°W , 187 m) from the Canadian Air and Precipitation Monitoring Network (CAPMoN), and Zeppelin, Spitzbergen (78.9°N , 11.8°E , 474 m) from the Norwegian Institute for Air Research. We also use hourly surface O_3 measurements at several sub-arctic sites from CAPMoN as well as from the Canadian National Air Pollution Surveillance (NAPS) network and from the United States Clean Air Status and Trends Network (CASTNET). For the NAPS sites, we only consider sites sampling background air in categories “Forest” and “Undeveloped Rural”, and for CASTNET we exclude sites located in “Urban/Agricultural” areas. This selection allows us to avoid more polluted sites, which can be subject to low winter-spring O_3 concentrations associated with NO titration of O_3 . The subarctic sites include Bonner Lake (49.4°N , 82.1°W , 242 m), Algoma (47.0°N , 84.38°W , 411 m), and Egbert (44.2°N , 79.8°W , 206 m) from the CAPMoN network, as well as Elk Island (53.68°N , 112.87°W , 714 m) from NAPS and Woodstock (43.94°N , 71.70°W , 255 m) from CASTNET.

2.3 The GEOS-Chem chemical transport model

GEOS-Chem is a 3-D global chemical transport model (Bey et al., 2001). In this work, we use GEOS-Chem v11-02d (<http://www.geos-chem.org>, last access 19 August 2019), driven by the Modern-Era Retrospective analysis for Research and Applications, version 2 (MERRA-2) assimilated meteorological fields (Gelaro et al., 2017), which have a native horizontal resolution of 0.5° latitude by 0.625° longitude with 72 vertical levels. We regrid the MERRA-2 fields to a 2°× 2.5° horizontal resolution and 47 vertical levels with merged levels above 80 hPa for computational expediency. For the time-period of the simulations conducted here (2007-2009), the daily boundary conditions for sea ice concentrations in MERRA-2 are from the high-resolution (1/20°) Operational SST and Sea Ice Analysis (OSTIA), which uses daily sea ice concentration products from multiple Special Sensor Microwave Imager (SSM/I) satellites (Donlon et al., 2012).

Global anthropogenic emissions are from EDGAR v4.2 (Olivier and Berdowski, 2001) for 1970-2008. Biomass burning emissions are from the Global Fire Emissions Database version 4 (GFEDv4) emission inventory (van der Werf et al., 2017). Biogenic emissions of volatile organic compounds (VOCs) are from the Model of Emissions of Gases and Aerosols from Nature version 2.1 (MEGAN 2.1) (Guenther et al., 2012).

GEOS-Chem simulates detailed HO_x-NO_x-VOC-O₃-halogen-aerosol tropospheric chemistry. The chemical mechanism in GEOS-Chem v11-2d was updated to the most recent JPL/IUPAC recommendations, as described in (Fischer et al., 2014; Fisher et al., 2016; Mao et al., 2013; Travis et al., 2016). This version of GEOS-Chem includes bromine-chlorine-iodine halogen chemistry. The bromine chemistry mechanism was first described in Parrella et al. (2012). The mechanism was then updated by Schmidt et al. (2016) to include extensive multiphase chemistry as well as coupling to tropospheric chlorine chemistry, which provides an important pathway to recycle bromine radicals. The uptake coefficients for the heterogeneous reactions of HOBr, ClNO₃ and O₃ with Br⁻ in aerosols, as well as for the reaction of HOBr with Cl⁻ in aerosols follow Ammann et al. (2013). The uptake coefficient between O₃ and Br⁻ considers both bulk and surface reactions. Sherwen et al. (2016a, 2016b) implemented iodine chemistry and Cl-Br-I interactions. Chen et al. (2017) added the in-cloud oxidation of dissolved SO₂ (S(IV)) by HOBr in GEOS-Chem, which decreased the global Br_y burden by 50% and resulted in improved agreement with GOME-2 VCD_{tropo} between 60° N and 60° S. Fifteen bromine tracers are transported (Br₂, Br, BrO, HOBr, HBr, BrNO₂, BrCl, BrONO₂, CHBr₃, CH₂Br₂, CH₃Br, IBr, CH₂IBr, Br⁻ on accumulation mode SSA and Br⁻ on coarse mode SSA), with sources from photolysis of CHBr₃, oxidation of CHBr₃, CH₂Br₂ and CH₃Br by OH radicals, transport of reactive bromine from the stratosphere, and SSA debromination driven by explicit heterogeneous reactions of SSA Br⁻ with HOBr, ClNO₃, and O₃.

Open ocean emissions of SSA are a function of wind speed and sea surface temperature (SST) as described in Jaeglé et al. (2011), with updates from Huang and Jaeglé (2017) for cold ocean waters (SST < 5°C). Two separate SSA tracers are transported: accumulation mode SSA ($r_{dry} = 0.01-0.5 \mu\text{m}$) and coarse mode SSA ($r_{dry} = 0.5-8 \mu\text{m}$). Sea salt Br⁻ is emitted

assuming a ratio of 2.11×10^{-3} kg Br per kg of dry SSA, based on the composition of sea water (Lewis and Schwartz, 2004; Sander et al., 2003). Sea salt Br^- is transported in two tracers as part of accumulation and coarse mode SSA.

2.3.1 The blowing snow SSA simulation in GEOS-Chem

190 The blowing snow SSA simulation in GEOS-Chem is described in Huang and Jaeglé (2017) and Huang et al. (2018). Blowing
snow SSA emissions are a function of relative humidity, temperature, age of snow, surface snow salinity and wind speed
following the parameterization of Yang et al. (2008, 2010). As in Huang and Jaeglé (2017), we assume $N=5$ for the number of
SSA particles produced per snowflake, and a mean snow age of 3 days over the Arctic. Two key parameters controlling the
magnitude of blowing snow SSA production and subsequent heterogeneous Br^- release are the surface snow salinity and Br^-
195 content, both of which have very few observational constraints. Sea salt and Br^- in surface snow originates from upward
migration of brine from sea ice, atmospheric deposition of SSA as well as gas-phase reactive bromine species, and
contamination of snow by frost flowers (Abbatt et al., 2012). Domine et al. (2004) estimates that frost flowers could account
for 10% of the observed surface snow salinity. Upward migration of brine is expected to be the dominant source of salinity for
thin snow over sea ice (<10-17 cm) (Domine et al., 2004; Peterson et al., 2019). As the snow depth gets thicker, the salinity of
200 surface snow decreases and the influence of atmospheric deposition likely becomes more important (Krnavek et al., 2012;
Nandan et al., 2017). Snow over first-year sea ice (FYI) is typically more saline than over multi-year sea ice (MYI), as MYI
is desalinated by flushing and gravity drainage during repeated summer melt cycles.

Krnavek et al. (2012) sampled the chemical composition of surface snow on land-fast sea ice near Utqiagvik, Alaska. They
reported a median salinity of 0.7 practical salinity unit (psu) for 2-3 weeks old FYI, 0.1 psu for thicker FYI, and 0.01 psu for
205 MYI. Domine et al. (2004) reported a salinity of 0.02 psu for MYI near Alert. Peterson et al. (2019) measured surface snow in
FYI and MYI regions near Greenland, Alaska, as well as in the central Arctic, finding higher mean salinities for snowpacks
less than 17 cm deep (0.15 psu) compared to deeper snowpacks (0.02 psu). The observed Br^- concentrations in surface snow
over Arctic sea ice is highly variable, ranging from 10^{-2} to 10^3 μM (Domine et al., 2004; Krnavek et al., 2012; Peterson et al.,
2019). This high variability is accompanied with either depletion or enhancement in the bromide to sodium (Br^-/Na^+) ratio
210 relative to sea water composition. Depletion in Br^- relative to seawater indicates loss to the atmosphere via heterogeneous
reactions. Enhancements can be the result of precipitation of hydrohalite ($\text{NaCl} \cdot 2\text{H}_2\text{O}$) below 251K in brine (Koop et al., 2000;
Morin et al., 2008). Indeed, the laboratory measurements of aqueous NaCl and sea-salt solution droplets of Koop et al. (2000)
show a factor of 12 increase in the Br^-/Na^+ ratio at 240K relative to 273K. Enhancements in Br^- could also be caused by the
deposition of atmospheric aerosol and HBr produced during bromine explosions onto surface snow (Simpson et al., 2005,
215 2007b). Domine et al. (2004) report measurements of surface snow on sea ice near Alert, with a factor of 5 enrichment in Br^-
relative to seawater, while surface snow over an Arctic Ocean site further north displayed a Br^- enrichment of 25. Samples of
surface snow on MYI appear to be enhanced in Br^- more often than on FYI (Krnavek et al., 2012), which suggests that MYI,
in addition to FYI, could play an active role in Arctic boundary layer bromine and chlorine chemistry (Peterson et al., 2019).

We conduct three GEOS-Chem simulations as part of this work: a standard simulation (referred to as “STD”) in which the
220 only source of SSA is from the open ocean, and two simulations in which we add a blowing snow source of SSA. In the first
blowing snow simulation (referred to as “FYI Snow”), we assume an Arctic surface snow salinity of 0.1 psu on FYI and 0.01
psu for MYI. In the second blowing snow simulation (“FYI+MYI Snow”), we use a salinity of 0.1 psu on FYI and 0.05 psu
225 on MYI. These two salinities assumed for MYI are used to examine the role of bromine activation on MYI. For both blowing
snow simulations, we assume a surface snow Br^- enrichment factor of 5 relative to sea water (sea salt Br^- is emitted as assuming
a ratio of 10.55×10^{-3} kg Br per kg of dry blowing snow SSA emitted). As sea ice age is not tracked in MERRA-2, for each
year we identify the location of Arctic MYI from the preceding September minimum sea ice extent in MERRA-2. The FYI
extent is calculated by subtracting the MYI extent from the total sea ice extent. Note that this very simple approach does not
account for advection and melting of MYI between one September to the next. All simulations are initialized with a 6-month
spin-up in 2006 and then followed by a 3-year simulation for 2007-2009.

230 In previous work (Huang et al., 2018; Huang and Jaeglé, 2017), we found that the GEOS-Chem blowing snow simulation
(using 0.1 psu on FYI and 0.01-0.1 psu on MYI) reproduced the seasonal cycle and magnitude of SSA mass concentrations
observed at Utqiagvik, Alert, and Zeppelin, but that the STD simulation underestimates observations by a factor of 2-10 during
winter and spring. The blowing snow simulations also reproduced the seasonal cycle of aerosol extinction coefficients
observed in the lower troposphere (0-2 km) over Arctic sea ice by the Cloud-Aerosol Lidar with Orthogonal Polarization
235 (CALIOP) on-board the CALIPSO satellite (Huang et al., 2018).

For comparison with satellite retrievals of $\text{VCD}_{\text{tropo}}$, we sample the model at 9:00-11:00 LT, which corresponds to the overpass
times of GOME-2 in the Northern Hemisphere. We found that sampling GEOS-Chem at 13:00-15:00 LT, matching the OMI
overpass time, results in less than a 1% difference in $\text{VCD}_{\text{tropo}}$. We thus only show the model results for 9:00-11:00 LT. We
regrid daily OMI and GOME-2 $\text{VCD}_{\text{tropo}}$ to a horizontal resolution of $2^\circ \times 2.5^\circ$ for comparison to GEOS-Chem. The model is
240 sampled on days and locations of GOME-2 and OMI overpasses with available retrievals.

3 Evaluation of the impact of blowing snow SSA on BrO $\text{VCD}_{\text{tropo}}$

3.1 Seasonal cycle of $\text{VCD}_{\text{tropo}}$ in the Northern Hemisphere

Several previous GEOS-Chem model versions have been evaluated against GOME-2 BrO $\text{VCD}_{\text{tropo}}$. Schmidt et al. (2016)
showed that the inclusion of SSA debromination resulted in a 50-100% overestimate in BrO in the northern hemisphere.
245 Sherwen et al. (2016a, 2016b) added iodine chemistry and disabled SSA debromination, finding relatively good agreement
with GOME-2 observations over the Arctic and the summertime low- and mid- latitudes. Chen et al. (2017) enabled SSA
debromination and added in-cloud oxidation of dissolved S(IV) with HOBr to the model version of Schmidt et al. (2016)
without iodine chemistry. They found improved agreement with GOME-2 $\text{VCD}_{\text{tropo}}$ over low- and mid- latitudes, but factors

of 3-10 underestimate over high latitudes. Our STD simulation includes SSA debromination, in-cloud oxidation of dissolved
250 S(IV) by HOBr, as well as iodine chemistry, and chlorine chemistry.

Fig. 1 (panels a and b) shows that with these four components, the STD simulation agrees well with the GOME-2 VCD_{tropo} at
0-30° N and 30-60° N averaged for 2007-2009. Over the Arctic (>60° N), however, the STD simulation underestimates
GOME-2 and OMI observations by up to 50% during spring (Fig. 1c). Note that we do not show monthly mean GOME-2
 VCD_{tropo} at 60-90° N for November-February, as less than 70% of the polar region has valid data (SZA is generally greater
255 than 80° for these months).

The 2007-2009 GOME-2 and OMI VCD_{tropo} display a March-April maximum of $\sim 3\text{-}3.5 \times 10^{13}$ molecules cm^{-2} poleward of 60°
N (Fig. 1c). The inclusion of blowing snow SSA emissions in the GEOS-Chem FYI Snow simulation increases the modeled
springtime VCD_{tropo} by 42% in March and 52% in April, improving the agreement with satellite retrievals (Fig. 1c). The
FYI+MYI Snow simulation increases the modeled VCD_{tropo} by another 10-20% in March and April. Both blowing snow
260 simulations are within 5-20% of observed VCD_{tropo} in March and April, however they predict too rapid a decrease in VCD_{tropo}
in May.

In late summer and fall, when sea ice extent is at its minimum and blowing snow SSA emissions are negligible, GEOS-Chem
underestimates GOME-2 VCD_{tropo} by 30-40%. Over the cloudy summer Arctic, the reaction $\text{HOBr} + \text{S(IV)}$ in cloud water
provides a sink for HOBr and decreases the Br_y abundance by about 70–90% (Chen et al., 2017). We hypothesize that including
265 acid displacement HCl from SSA would lead to an increase in HOBr because of the competition between $\text{HOBr} + \text{Cl}^-$ (which
recycles HOBr by producing BrCl) and $\text{HOBr} + \text{S(IV)}$ (which is a sink for HOBr) in cloud droplets. Indeed, in a subsequent
version of GEOS-Chem, Wang et al. (2019b) added this source of HCl, finding an increase in BrO, especially in cloudy high
latitudes.

3.1 Spatial distribution of springtime Arctic BrO VCD_{tropo}

270 Fig. 2 shows the spatial distribution of monthly mean VCD_{tropo} during March and April 2007-2009. In March, both GOME-2
and OMI exhibit enhanced VCD_{tropo} ($3.5\text{-}5 \times 10^{13}$ molecules cm^{-2}) in an arc between Baffin Bay - along the west coast of
Greenland - and the Laptev Sea, off the Northern coast of Siberia. In April, the observed VCD_{tropo} reach values $> 3.5\text{-}4 \times 10^{13}$
molecules cm^{-2} over most of the sea ice covered Arctic Ocean. Fig. 3a compares the GOME-2 and OMI VCD_{tropo} poleward of
60°N for March and April 2007-2009, showing that they are highly correlated ($r=0.92$, slope=0.99), with OMI being 13%
275 higher than GOME-2.

As shown in Figs. 2 and 3d, the GEOS-Chem STD simulation predicts VCD_{tropo} lower than $2.5\text{-}3 \times 10^{13}$ molecules cm^{-2} in
March-April, with little spatial variability compared to observations ($r=0.22\text{-}0.36$). The normalized mean bias
($\text{NMB} = 100 \times \sum(M_i - O_i) / \sum O_i$, with observations O_i and model M_i summed poleward of 60° N) is -41% compared to OMI and

-34% compared to GOME-2. The FYI Snow simulation has more success at reproducing the magnitude of the observed
280 VCD_{tropo} in March and April (NMB=-12% relative to OMI and NMB=+1% relative to GOME-2), as well as their spatial
distribution ($r=0.70-0.76$). The VCD_{tropo} predicted by the FYI+MYI Snow simulation agrees better with OMI (NMB=+2% and
 $r=0.76$), but overestimates GOME-2 (NMB=+17% and $r=0.79$). Similar to the satellite retrievals in March, the FYI and
FYI+MYI snow simulations displays the largest VCD_{tropo} over the Canadian Arctic Archipelago (Fig. 2), with secondary
maxima over Baffin Bay, the Greenland Sea, and the east Siberian Sea. In April, the VCD_{tropo} in the FYI snow simulation
285 underestimates GOME-2 and OMI throughout the Arctic, but the FYI+MYI snow simulation overestimates satellite
observations in the high Arctic ($>80^\circ$ N), while underestimating observations at $70-80^\circ$ N. This suggests spatial variability in
salinity and/or Br^- enhancement for snow that is not captured in our assumptions of spatially uniform values for FYI and MYI.
We also find that the simulated VCD_{tropo} underestimate observations over the coastal Arctic, especially in April. This could be
due halogen activation from snow over land (Simpson et al., 2005), which is not considered in our simulation as we assume
290 zero salinity and Br^- content in continental snow.

3.2 Daily variations in pan-Arctic BrO VCD_{tropo}

Fig. 4a shows that GOME-2 daily mean VCD_{tropo} poleward of 60° N is highest in March-April, and then decreases in May.
The OMI VCD_{tropo} in March and April displays a similar day-to-day variability as GOME-2. Overall the FYI+MYI Snow
simulation shows good agreement with the magnitude and daily variability of the combined mean GOME-2 and OMI VCD_{tropo}
295 (NMB=-8.3%, $r=0.85$). While the FYI+MYI Snow simulation reproduces the March-April VCD_{tropo} , it predicts too rapid a
decrease in May for all three years (Fig 4a). The difference between the FYI Snow and MYI+FYI Snow simulations is the
largest in April as polar sunrise reaches the high Arctic, where the majority of MYI is located. Most of the BrO enhancements
in the two blowing snow simulations occur below 1 km altitude, occasionally extending to 2 km (see Fig. S1ab).

As discussed in previous studies (Choi et al., 2018; Nghiem et al., 2012; Richter et al., 1998; Salawitch et al., 2010; Theys et
300 al., 2011) satellite observations show the frequent occurrence of “BrO hotspots” over Arctic sea ice during spring. For this
study, we define a BrO hotspot when local VCD_{tropo} exceed 4.5×10^{13} molecules cm^{-2} , which is the 90th percentile of GOME-2
 VCD_{tropo} poleward of 60° N in March-April. For OMI, the corresponding 90th percentile is 5.1×10^{13} molecules cm^{-2} . We use
these respective criteria to calculate the daily area covered by BrO hotspots as observed by GOME-2 and OMI. The episodic
nature of GOME-2 and OMI BrO hotspots is apparent in Fig. 4b, which shows a large degree of daily variability, with events
305 lasting between a few days to 2 weeks. The light orange shading in Fig. 4 identifies multi-day events (>5 days) when the
GOME-2 hotspot area exceeds 2 million km^2 . There are 2 such events in both 2007 and 2008, and 3 events in 2009 (Fig. 4b).
These events reach maximum extents of 4-6 million km^2 . For comparison, the mean sea-ice extent during March-April
poleward of 60° N is ~ 10 million km^2 (with a $\sim 65\%/35\%$ split between FYI and MYI).

We apply the OMI threshold ($VCD_{\text{tropo}} > 5.1 \times 10^{13}$ [molecules cm⁻²](#)) to the GEOS-Chem simulations. On average, the FYI+MYI Snow simulation reproduces the observed spatial extent of BrO hotspots to within 4% and captures the timing of the 2 large-scale episodes in 2007 (26 March – 4 May, 10–20 May 2007), the two episodes in 2008 (7–17 March and 1–8 April 2008). For 2009, the FYI+MYI Snow simulation predicts two episodes (5–15 March and 11–23 April), but the aerial extent of these episodes is overestimated, and the simulation misses the third episode on 6–19 May 2009 (Fig. 4b). In the model, the variability in BrO hotspots is driven by the temporal variations of the blowing snow SSA burden (Fig. 4c).

3.3 Two case studies of large BrO explosion events

Fig. 5 shows the daily spatial distribution of VCD_{tropo} on 25–30 March 2007, corresponding the largest BrO hotspot episode observed by GOME-2 and OMI during our study period (Fig. 4ab). The GOME-2 and OMI VCD_{tropo} exceed 7×10^{13} [molecules cm⁻²](#) (up to 15×10^{13} [molecules cm⁻²](#)) over the sea-ice covered region in the high Arctic region ($>70^\circ\text{N}$). The enhanced VCD_{tropo} start on 25 March over the East Siberian Sea and then rotate counterclockwise, reaching the Beaufort Sea on 27 March and the Canadian Archipelago on 28–30 March 2007. This event was previously discussed in Begoin et al. (2010) and Choi et al. (2018). Begoin et al. (2010) used the FLEXPART particle dispersion model to link the enhanced GOME-2 VCD_{tropo} to a cyclone with very high surface wind speeds, favorable for generating blowing snow SSA. Choi et al. (2018) found that the spatial pattern of OMI VCD_{tropo} was consistent with that of the GEOS-5 simulated blowing snow SSA burden on both FYI and MYI, while restricting the blowing snow emissions to FYI only did not agree as well with the observed VCD_{tropo} . Our FYI+MYI Snow simulation reproduces the evolution of this BrO hotspot, although the magnitude of the modeled VCD_{tropo} overestimates observations after 28 March 2007. The main difference between the FYI Snow and FYI+MYI Snow simulations is on 25–26 March 2007, when the high winds are located over the North Pole where MYI is (Figs. 5 and S2). As there are no valid GOME-2 and OMI retrievals at these high latitudes ($\text{SZA} > 80^\circ$), we cannot assess which simulation agrees better with observations for this case study.

Fig. 6 shows the daily distribution of VCD_{tropo} on 14–19 April 2007. During this event, there is a factor of 1.5 difference in pan-Arctic VCD_{tropo} calculated with the FYI Snow and FYI+MYI Snow simulations (Fig. 4a). The GOME-2 and OMI VCD_{tropo} show high values ($> 5 \times 10^{13}$ [molecules cm⁻²](#)) being maintained over the North Pole (Fig. 6). The MERRA-2 meteorological fields predict high surface winds ($> 10\text{--}12 \text{ m s}^{-1}$) in that area, which is covered by MYI (Fig. 7, bottom row). The VCD_{tropo} calculated by the FYI Snow simulation show little enhancement over the North Pole (Fig. 7), while the FYI+MYI Snow simulation better captures the magnitude and shape of the observed enhancement (Fig. 6), suggesting that blowing snow SSA emissions over MYI can be a significant source of BrO.

Overall, we find that the FYI+MYI Snow simulation captures reasonably well the spatial and temporal distribution of the observed VCD_{tropo} , as well as the spatial extent and frequency of BrO hotspots in March and April. In our simulation, blowing snow SSA emissions on MYI account for 20–30% of the VCD_{tropo} enhancement and could thus represent an important source

340 of bromine activation. This is consistent with previous studies, which have suggested that, in addition to FYI, MYI could play a significant role in halogen activation. Gilman et al. (2010) used backtrajectories to calculate sea ice exposure of air masses observed during the International Chemistry Experiment in the Arctic Lower Troposphere (ICEALOT) ship-based study, finding that exposure to both FYI and MYI was the best predictor of reduced O₃ levels. Choi et al. (2018) reported that the frequency of springtime BrO explosion events observed by OMI for 2005-2015 was strongly correlated with blowing snow
345 SSA emissions over all sea ice, but they found little to no correlation when only FYI was considered as a source of blowing snow SSA. Rhodes et al. (2017) found that simulations including blowing snow SSA emissions on MYI had better skill at reproducing surface observations of SSA in the Arctic compared to simulations without SSA emissions on MYI. More generally, low O₃ and/or high BrO VCD_{tropo} have been observed over and downwind of regions with a high fraction of MYI such as the Canadian Archipelago and the eastern Beaufort Sea (Bottenheim and Chan, 2006; Choi et al., 2012; Halfacre et al.,
350 2014; Koo et al., 2012; Richter et al., 1998; Salawitch et al., 2010). In contrast to the generally good agreement in March and April, during May our simulation is systematically too low compared to GOME-2 VCD_{tropo} (Figs. 1c and 4ab). In Section 5 we will discuss how the recycling of bromine deposition to snowpack in late spring could propagate blowing snow-induced halogen activation into May.

4 Impact of blowing snow SSA on tropospheric O₃ over the Arctic

355 4.1 Comparison to hourly surface O₃ observations

We evaluate the ability of our simulations to capture ODEs via comparisons to hourly O₃ observations at several Arctic and sub-Arctic sites in 1 March - 31 May 2007 (Fig. 8). Observations at Utqiaġvik and Alert show frequent occurrences of low O₃ mixing ratios (<10 ppbv), maintained for 1-7 days. These events are somewhat less frequent at Zeppelin. Figure 8 shows that FYI+MYI Snow simulation reproduces only 25-30% of the ODEs at Utqiaġvik and Alert, such as the 8 – 18 April and 26 April
360 – 10 May depletion events at Utqiaġvik as well as the April events at Alert. The model-predicted magnitude of O₃ depletion at those sites is a factor of two lower than observed. The FYI+MYI Snow simulation performs somewhat better at Zeppelin, where it captures the timing and magnitude of 40% of the observed events, in particular between late March and late April. Our simulation thus misses 60-75% of observed ODEs at those three Arctic sites, including several March and late May events at Utqiaġvik, as well as the sustained May events at Alert, and the late May ODEs at Zeppelin.

365 Observed surface O₃ depletion events can be caused by advection of upwind O₃ poor air and/or local depletion (Bottenheim and Chan, 2006; Halfacre et al., 2014; Hopper et al., 1998; Jacobi et al., 2006; Simpson et al., 2007b, 2017). The rapid recovery of surface O₃ after a depletion event is often due to turbulent vertical mixing of O₃-rich free tropospheric air down to the surface or to advection of O₃-rich continental air (Bottenheim et al., 2009; Gong et al., 1997; Hopper et al., 1998; Jacobi et al., 2010; Moore et al., 2014; Morin et al., 2005). Furthermore, some ODEs can be highly localized. Using ice-tethered buoys over
370 coastal regions and over the Arctic Ocean, Halfacre et al. (2014) found that large areas of the Arctic Ocean are partially depleted in O₃ during spring with local imbedded areas (~200-300 km) that are more depleted. Morin et al. (2005) report that

dynamic conditions of the boundary layer near the shore can lead to very different surface O₃ behaviors observed at Alert compared to at a site over the sea ice 10 km away. Similar results were found by Jacobi et al. (2006), when comparing O₃ observations at Zeppelin Station and nearby (~100-200 km) ship-based measurements. The coarse resolution of our simulation (2°×2.5°) and the general difficulty of models in reproducing boundary layer depth and vertical mixing processes could thus be one explanation for our poor representation of ODEs at some of these surface sites. Another explanation is the lack of detailed chlorine chemistry in this version of GEOS-Chem. In particular, we do not consider acid displacement of Cl⁻ in SSA (Keene et al., 2007) which, together with bromine chemistry, could act to enhance ODEs. Finally, our simulation does not include local snowpack Br activation, which has been shown to lead to surface ODEs when the stable Arctic boundary layer is decoupled from convective exchange with the free troposphere (Custard et al., 2017; Peterson et al., 2015, 2017; Pratt et al., 2013; Wang et al., 2019a).

4.2 Spatiotemporal distribution of O₃ depletion in GEOS-Chem

Fig. 9 shows the spatial distribution of monthly mean surface O₃ mixing ratios calculated in the FYI+MYI Snow simulation as well as the decrease in surface O₃ (ΔO_3) relative to the STD simulation for March and April 2007-2009. The lowest monthly mean O₃ mixing ratios are ~25 ppbv in March over the Canadian Archipelago and ~20-25 ppbv in April over the North Pole (Fig. 9a), with a spatial distribution corresponding to that of the simulated VCD_{tropo} (Fig. 2). Note that the low O₃ values simulated over Europe are due to NO_x titration. The monthly mean values from the FYI+MYI Snow simulation are within better than 8 ppbv of surface observations at Utqiagvik (model/obs: March 28/22 ppbv, April 28/20 ppbv), Alert (model/obs: March 29/33 ppbv, April 27/27 ppbv), Zeppelin (model/obs: March 32/39 ppbv, April 27/31 ppbv), and Elk Island (model/obs: March 28/36 ppbv, April 35/38 ppbv). The mean values of surface ΔO_3 increase from 4-8 ppbv (15-30% depletion relative to the STD simulation) in March, to 8-14 ppbv (30-40%) in April, when sufficient sunlight is available to drive photochemistry at high latitudes (Fig. 9b). Poleward of 60° N, the FYI+MYI Snow simulation displays a pan-Arctic O₃ decrease of 3.7 ppbv (11%) in March and 8.3 ppbv (23%) relative to the STD simulation.

We define the frequency of ODEs as the percent of time when more than 20 ppbv O₃ is lost due to blowing snow ($\Delta O_3 > 20$ ppbv, Fig. 9c). Applying this definition to the FYI+MYI Snow simulation, we find that ODEs occur up to 1-5% of the time in March, increasing to up to 15-25% of the time in April as sunlight extends to higher latitudes (Fig. 9c). Using an ODE definition including more moderate events ($\Delta O_3 > 10$ ppbv), we find that poleward of 70° N ODEs occur 20-60% of the time in April (Fig. 9d). The median aerial extent of ODEs ($\Delta O_3 > 20$ ppbv) simulated in the FYI+MYI Snow simulation is 0.35 million km² (horizontal extent of ~330 km) with some of the largest ODEs extending over areas of 1.5-7 million km² (~700-1500 km horizontal extent). Our results are consistent with the 282 km median size of major ODEs inferred by Halfacre et al. (2014) using ice-tethered buoys combined with back-trajectories.

The decrease in surface O₃ due to blowing snow SSA is 60% larger in the FYI+MYI Snow simulation compared to the FYI Snow simulation (Fig. 10a). The timeseries of pan-Arctic daily mean ΔO₃ for the FYI+MYI Snow simulation at different altitudes shows that ΔO₃ is uniform in the bottom 500 m altitude and extends to 1000 m altitude, where the model-calculated
405 ΔO₃ is about half of the surface ΔO₃. At 2000 m, ΔO₃ is generally below 4 ppbv (Fig. 10b). This is consistent with ozonesonde profiles, which indicate that depletion events are confined to the lowest 1000 m with an average height of the top of the layer at 500 m (Hopper et al., 1998; Oltmans et al., 2012; Tarasick and Bottenheim, 2002). We find that the Arctic tropospheric O₃ burden (>60° N) decreases by 3% in March and 6% in April in the FYI+MYI simulation relative to the STD simulation.

In our simulation, the timing of the maximum ΔO₃ takes place 4-5 days after the maximum in the blowing snow SSA burden
410 (Figs. 4c and 10b), reflecting increasing O₃ loss as BrO concentrations increase. This is illustrated during the March 25-30 2007 event: the pan-Arctic blowing snow SSA burden and VCD_{tr_op_o} reach their maximum value on 27 March, while the lowest values for surface O₃ occur 4 days later on 31 March (Fig. 5, bottom row). By 29-30 March the blowing snow SSA burden is back to low values over the Arctic, but the VCD_{tr_op_o} are still elevated and low surface O₃ mixing ratios are predicted throughout the Arctic. This can be explained by the different lifetimes of these species over the springtime Arctic: ~1 day for SSA, 4-7
415 days for the Br_y family (consisting of all the gas-phase inorganic bromine species), and 30-40 days for O₃.

4.3 Transport of O₃-depleted Arctic air to lower latitudes

In some instances, large-scale blowing snow induced bromine activation can influence surface O₃ in sub-Arctic regions. The large BrO explosion and associated O₃ depletion from late March to early April 2007 (Figs. 4ab) illustrates this. The [light orange](#) shading in Fig. 8 corresponds to the times when this BrO explosion is transported to the three Arctic sites and several
420 sub-Arctic sites. The GEOS-Chem FYI+MYI Snow simulation shows transport of O₃-depleted air to Utqiagvik on 25-27 March 2007, with observations showing much stronger depletion than the model (Fig. 8a). This air mass is observed at Alert on 31 March-4 April and the eastern edge of that O₃-depleted air reaches Zeppelin around the same time (Figs. 8bc). The O₃-depleted air is then transported southward over Hudson Bay and towards the northeastern United States during 4-9 April (Figs. 5 and 11), mostly below 1,000 m altitude. At the synoptic level, a surface low was sweeping across the United States from the central Rockies to the Great Lakes in early April, bringing sub-freezing Arctic air behind its cold front with record-breaking
425 cold temperatures being measured over the Central Plains and much of the Southeast (NOAA/USDA, 2008).

Between 2 April and 7 April, observed O₃ mixing ratios at 5 sub-Arctic sites from Elk Island (53.7° N) to Woodstock (43.9° N), decreased from background levels of 40 ppbv down to 10-20 ppbv (Fig. 8). This decrease is reproduced by both the FYI and FYI+MYI Snow simulations, which show that enhanced VCD_{tr_op_o} and blowing snow SSA were transported over Hudson
430 Bay towards the Great Lakes region (Fig. 11a-c). By the time the polar air mass reached the Northeast United States, the BrO VCD_{tr_op_o} were back to normal levels, but because of its longer lifetime, O₃ was 5-10 ppbv lower than background mixing ratios

as shown by the FYI+MYI Snow simulation (Fig. 11d). This decrease in surface O₃ was observed at NAPS and CASTNET surface sites throughout Southeast Canada and the Northeast United States (Fig. 11d).

Ridley et al. (2007) discuss a similar transport event of O₃-depleted air from the Canadian Archipelago and nearby Arctic Ocean to Hudson Bay observed in April 2000 during the Tropospheric Ozone Production about the Spring Equinox (TOPSE) aircraft campaign. During that event, O₃ levels were reduced from 30-40 ppbv down to values as low as 0.5 ppbv in a large area below 500 m altitude. Over the 3-year period that we examined, our simulations suggest that transport of O₃-depleted air ($\Delta\text{O}_3 > 20$ ppbv) to Hudson Bay occurs 1-10% of the time (0.3-3 days/month) in March and in April (Fig. 9c), with transport of more moderate O₃ depletion ($\Delta\text{O}_3 > 10$ ppbv) taking place 10-20% of the time (3-6 days/month). In March-April 2008, measurements in the ice-free North Atlantic showed transport of O₃-poor air from the Arctic basin down to 52° N during the ICEALOT cruise (Gilman et al., 2010): a 13 ppbv decrease in O₃ was accompanied by a simultaneous decrease in the acetylene to benzene ratio indicating exposure to halogen oxidation. In our simulation, we find that transport of air mass with $\Delta\text{O}_3 > 10$ ppbv air to sub-Arctic latitudes occurs with a frequency of 5-10% (1.5-3 days/month) down to 50° N and 1-5% (< 1.5 days/month) down to 40° N (Fig. 9d) and that this transport appears to be favored over Hudson Bay extending to the Northeast United States and the Western Atlantic off Nova Scotia (Fig. 9d). At mid-latitudes (30°-60° N), the FYI+MYI Snow simulation results in a mean surface O₃ decrease of 1.2 ppbv (3.3%) in March-April, thus a small but non-negligible contribution.

5 Atmospheric deposition on snow as a source of salinity and bromide

We use the results of the FYI Snow simulation to examine the potential role of atmospheric deposition of SSA and gas-phase Br_y as a source of salinity and bromide to snow on sea ice. Snow composition profiles indicate that atmospheric deposition can sometimes be a more important source of salinity and Br⁻ than upward brine migration, especially over MYI and in deep snowpack (>10-17 cm) over FYI (Domine et al., 2004; Krnavck et al., 2012; Peterson et al., 2019). Fig. 12a shows the evolution of deposition on sea ice (poleward of 60° N) for Na⁺ and for total bromine (sum of particulate Br⁻ and gas-phase Br_y) in the GEOS-Chem FYI Snow simulation. Between September and May, deposition of Na⁺ on snow-covered sea ice has an average daily value of 2.2×10^6 kg/day. For total bromine (SSA Br⁻ and gas-phase Br_y), the average daily value is 6.7×10^4 kg/day. Open ocean SSA and blowing snow SSA account nearly equally to these deposition fluxes.

From October to March the deposition of total bromine tracks that of Na⁺ as both are derived from SSA (Fig. 12a). Starting in late March, deposition of Na⁺ decreases rapidly following the decrease in blowing snow SSA emissions (Fig. 4c) due to slower wind speeds and warmer temperatures (Huang and Jaeglé, 2017). Deposition of bromine remains high for another month, however, as once bromine activation starts in early spring a large fraction of Br deposition is in the form of Br_y, in particular HBr which is the end product of reactive bromine chemistry. In April-May, deposition of Br_y accounts for more than half of total bromine deposition (Fig. 12a). As Br_y has a longer lifetime against deposition than particulate Br⁻ and Na⁺, Br_y shows a more gradual decrease in deposition in April and May. This behavior is consistent with observations of freshly fallen snow at

Alert during the Network on Climate and Aerosols Research (NETCARE) campaign, showing a broad spring peak in Br⁻ deposition between late March and late May (Macdonald et al., 2017). A similar increase of Br⁻ in snowfall samples after polar sunrise was reported at Alert by Toom-Sauntry & Barrie (2002). Furthermore, Spolaor et al. (2013, 2014) found a strong seasonal variability of Br⁻ in polar firn at both Arctic and Antarctic sites, with greater Br⁻ values in spring/summer compared to winter.

We derive the Br⁻/Na⁺ enrichment factor (EF) of deposition relative to seawater composition (which has a Br⁻/Na⁺ mass ratio of 0.00625 g/g):

$$EF = \left(\frac{[Br^- + Br_y]}{[Na^+]} \right)_{deposition} / \left(\frac{[Br^-]}{[Na^+]} \right)_{seawater}$$

EF is calculated for each day and each model grid box and then surface-area weighted over sea-ice to obtain the daily timeseries of pan-Arctic EF shown in Fig. 12b. The model-calculated EF is fairly constant between September and February, with values of ~3-10. Starting in late March, the divergence between bromine and Na⁺ deposition leads to an increase in EF to a value of 14 by the end of April, followed by a more rapid increase in May up to a value of ~60. In May, the mean model-calculated EF of deposition is 30. These values are remarkably consistent with the EF measured during the NETCARE campaign at Alert (Macdonald et al., 2017), as shown in Fig. 12b, with mean observed values of EF~7 between September and February, 15 in April, and 25 in May. Similarly, Toom-Sauntry & Barrie (2002) observed an increase in EF from low enrichment in the dark winter months (median 1.5-5) to a large enrichment (median 20-72) after polar sunrise. In our simulation of the blowing snow SSA, we assumed a constant value of EF=5, which appears to underestimate the EF of snow in May by a factor of 2-3. This springtime increase in the enrichment of Br⁻ in snow over sea ice could thus act to propagate bromine explosions and ODEs into May, and could explain our systematic underestimate in observed VCD_{tropo} in May. Using a one-dimensional model, Piot & von Glasow (2008) compared modeled and observed deposition on snow at Utqiagvik, demonstrating that 75% of deposited bromine may be re-emitted into the gas phase as Br₂ or BrCl. They proposed that cycles of deposition and re-emissions of bromine from snow could thus result in a “leap-frogging” process and explain observations of progressive Br⁻ enrichment of snow in coastal Arctic regions with distance inland (Simpson et al., 2005). [Indeed, Peterson et al. \(2018\) observed enhanced BrO up to 200 km inland over snow-covered tundra near Utqiagvik.](#) Our simulations suggest that a similar mechanism could occur in snow over sea ice via repeated cycles of blowing snow SSA sublimation and deposition of reactive bromine. These cycles could thus be important in propagating bromine explosions in space (onto coastal snow-covered land) as well as in time on sea ice (into late spring).

We now use the model-simulated Na⁺ deposition (FYI Snow simulations) on the sea ice covered Arctic ocean to estimate the salinity of surface snow due to SSA deposition. In our simulation, wet and dry deposition contribute nearly equally to SSA deposition. Over the 3-month period between February 1 and April 31, the cumulative SSA deposition on Arctic sea ice is

10.3×10⁸ kg SSA in the FYI snow simulation. Snow depth on sea ice is not calculated in MERRA-2 and reanalyses tend to have highly uncertain solid precipitation in polar regions (Lindsay et al., 2014; Uotila et al., 2019), so we rely on the limited observations available. The Warren et al. (1999) climatology, which is based on 1954-1991 Soviet drifting stations measurements on Arctic MYI, shows that snow depth increases from 29.7 cm to 34.4 cm during these three months, resulting in 4.7 cm snow accumulation. This is likely an overestimate of the pan-Arctic snow depth because MYI tends to have thicker snow and these measurements represent conditions from past decades (Kwok and Cunningham, 2008). Indeed, the 2003-2008 AMSR-E snow depth retrievals over FYI compiled by Zygmontowska et al. (2014) show much lower snow depths, increasing from 18 cm in February to 19 cm in in May, thus a 1 cm snow accumulation. These two estimates in snow accumulation result in pan-Arctic snow accumulations of 10¹⁴ kg to 2.2 10¹³ kg (sea ice extent is 11 million km² and fresh snow has a density of 200 kg/m³), which combined with our calculated SSA deposition results in a mean salinity of 0.01 psu - 0.04 psu of surface snow due to deposition. Fig. 12c shows the spatial distribution of snow salinity due to SSA deposition assuming a 1 cm snow accumulation. High salinities (>0.05 psu) are calculated in sea-ice regions near the ice-free open ocean in the North Atlantic and Bering Sea, with lower values on FYI (0.02-0.05 psu) and MYI (0.01-0.03 psu). Deposition could thus represent a significant source of salinity of surface snow on MYI by the time sunrise reaches high latitudes, and it could also be important for FYI as the snow gets deeper in late spring (Petty et al., 2018; Warren et al., 1999).

6 Conclusions

We used the GEOS-Chem model to examine the impact of blowing snow SSA on bromine chemistry and O₃ during Arctic spring. We conducted two blowing snow simulations assuming a 0.1 psu surface snow salinity on FYI and different salinities on MYI (0.01 psu for the FYI Snow simulation and 0.05 psu for the FYI+MYI Snow simulation). We evaluated these simulations against satellite observations of VCD_{tropo} from GOME-2 and OMI for 2007-2009. We found that our simulations reproduce the spatio-temporal distribution of the observed VCD_{tropo} to within -12% to +17% in March and April and capture the spatial extent as well as the frequency of most large BrO hotspots in March and April. However, our simulations predicted too rapid a decrease in VCD_{tropo} in May, while observations showed a more gradual decrease. The FYI+MYI Snow simulation captured observed BrO hotspots over the North Pole better than the FYI Snow simulation. In our FYI+MYI simulation, blowing snow SSA emissions on MYI account for 20-30% of the VCD_{tropo} enhancement and thus represent a significant source of bromine activation. In the past, the search for halogen-containing substrates in polar regions has often focused on the role of frost flowers, newly formed sea ice, and FYI snowpack on polar halogen activation, because of the higher salinity of these surfaces and of their snowpack relative to MYI (e.g., Abbatt et al., 2012; Jacobi et al., 2006; Kaleschke et al., 2004; Simpson et al., 2007a; Yang et al., 2010). Our simulations suggest that the role of MYI in halogen activation is likely also important, consistent with results of several studies (Choi et al., 2018; Gilman et al., 2010; Halfacre et al., 2014; Peterson et al., 2019).

Inclusion of blowing snow SSA in our FYI+MYI Snow simulation results in a pan-Arctic decrease in surface O₃ of 3.7 ppbv (11%) in March and 8.3 ppbv (23%) in April. As most of this decrease is confined to altitudes below 1000 m, the Arctic

525 tropospheric O₃ burden decreases by only 3-6% in March-April. Compared to surface O₃ observations at coastal Arctic sites,
we find that our simulation captures only 25-40% of observed ODEs and often underestimates their magnitude. The FYI+MYI
simulation misses 60-75% of observed ODEs, in particular the early March events at Utqiagvik and most of the May events at
Utqiagvik, Alert, and Zeppelin. The simulation does reproduce an event of low O₃ Arctic air transport down to 40° N, which
530 frequency of 1.5-3 days/month down to 50° N and < 1.5 days/month down to 40° N.

The mixed success of our simulation at capturing the rapid O₃ variations observed at coastal Arctic sites could be related to
the coarse resolution of the model, the difficulty in simulating Arctic boundary layer exchange processes, the lack of detailed
chlorine chemistry, or the fact that we did not include direct halogen activation by snowpack chemistry. The locations of the
three coastal Arctic sites with surface O₃ observations correspond to GEOS-Chem grid box sizes of ~40-90 km (latitude) by
535 220 km (longitude). It is thus not surprising that we are not able to capture some surface O₃ events which have been shown to
have large variability on scales of 10-100 km (Morin et al., 2005; Jacobi et al., 2006). Most reanalyses, including MERRA-2,
tend to simulate too weak Arctic surface temperature inversions, likely due to their low skill simulating turbulent heat fluxes
over sea ice (Graham et al., 2019). Excessive exchange between the boundary layer and free troposphere would transport O₃-
rich free tropospheric air down to the surface, which might explain our simulations of too weak ODEs compared to
540 observations. The influence of snowpack bromine chemistry was examined in two previous modeling studies (Falk and
Sinnhuber, 2018; Toyota et al., 2011). These studies displayed remarkable agreement with surface O₃ observations, in
particular at Alert and Utqiagvik, and reproduced the synoptic variability in GOME VCD. However, both studies
systematically underestimated the magnitude of observed BrO VCDs by at least a factor of 2. These results together with our
GEOS-Chem blowing snow simulations suggest that most of the VCD_{tropo} enhancements observed by satellites can be
545 explained by blowing snow SSA leading to BrO enhancements over larger vertical (0-2 km) and horizontal scales (pan-Arctic)
than snowpack chemistry, but that in the shallow boundary layer (~50-250 m) over the springtime Arctic direct snowpack
halogen activation could contribute to Br release and potentially be responsible for the most severe ODEs. Indeed, some of the
strongest ODEs observed at the surface seem to occur when the stable Arctic boundary layer is decoupled from convective
exchange with the free troposphere (Moore et al., 2014; Seabrook et al., 2011; Wang et al., 2019a). Furthermore, as boundary
550 layer O₃ reaches very low levels (<4 ppbv), BrO production via R3 is suppressed and atomic Br becomes much more abundant
than BrO (Neuman et al., 2010; Wang et al., 2019a), such that satellites would not necessarily observe co-located O₃ depletion
and enhanced BrO VCD_{tropo}.

Based on our analysis of the seasonal variation in modeled deposition of Na⁺ and bromine on snow, we propose that the
progressive enrichment of bromine in deposition onto sea ice could help propagate blowing snow SSA bromine activation into
555 May, even as the magnitude of blowing snow SSA emissions starts to decrease. The increase of our calculated springtime
enrichment of Arctic snow Br⁻ is similar to observations showing a peak after polar sunrise (Macdonald et al., 2017; Spolaor

et al., 2013; Toom-Sauntry and Barrie, 2002). Recycling of deposited bromine was previously proposed in the context of frost flower/snowpack activation followed by inland transport over coastal regions (Domine et al., 2004; Piot and Glasow, 2008; Simpson et al., 2005, 2007b). We propose that a similar mechanism takes place over sea ice and snow-covered coastal regions with blowing snow SSA, enabling BrO explosions to propagate spatially and to last longer into late spring. We also show that SSA deposition to surface snow in winter-spring, when snow accumulation on sea ice is at its minimum, could account for 0.01-0.03 psu on MYI, 0.02-0.05 psu on FYI, and >0.1 psu on sea ice areas close to the open ocean. While upward migration of brine from sea ice is the main source of salinity in surface snow over FYI with shallow snowpack, atmospheric deposition could thus be the dominant source in surface snow over the less saline MYI sea ice and could play an important role in late spring as the snowpack deepens over FYI (Domine et al., 2004; Krnavek et al., 2012; Peterson et al., 2019).

Our simulations did not include 2-way coupling between snowpack composition and atmospheric deposition, did not consider blowing snow on coastal snow enriched in Br⁻, and did not incorporate direct snowpack halogen activation. Incorporating these coupled processes together with blowing snow SSA promises to yield further insights into the mechanisms leading to polar bromine activation and ODEs. In addition, further examination of how horizontal resolution and turbulent mixing parameterizations influence simulations of ODEs would help improve our understanding of the interplay between meteorology and chemistry in inducing rapid variations in surface O₃ over the Arctic.

Data availability. The GOME-2 VCD_{tropo} data is available upon request from Nicolas Theys (Nicolas.Theys@aeronomie.be). The OMI VCD_{tropo} data is available upon request from Sungyeon Choi (sungyeon.choi@nasa.gov). The surface O₃ observations are available from the Observations from these surface monitoring sites are obtained through the World Data Centre for Reactive Gases (WDCRG, <http://ebas.nilu.no>), CAPMoN (<http://donnees.ec.gc.ca/data/air/monitor/monitoring-of-atmospheric-gases/ground-level-ozone/>), NAPS (<https://www.canada.ca/en/environment-climate-change/services/air-pollution.html>), and CASTNET (<https://www.epa.gov/castnet/castnet-ozone-monitoring>) data repositories. The GEOS-Chem simulations are available upon request from the corresponding author (jaegle@uw.edu).

Author contributions. JH and LJ designed the study, conducted the GEOS-Chem simulations, and analysed the results. QC, BA, TS, ME contributed to model development. NT and SC provided the tropospheric BrO columns for GOME-2 and OMI. JH wrote the paper with significant contributions from LJ. All the authors contributed to editing the paper.

Competing interests. The authors declare that they have no conflict of interest.

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References

- 590 Abbatt, J. P. D., Thomas, J. L., Abrahamsson, K., Boxe, C., Granfors, A., Jones, A. E., King, M. D., Saiz-Lopez, A., Shepson, P. B., Sodeau, J., Toohey, D. W., Toubin, C., von Glasow, R., Wren, S. N. and Yang, X.: Halogen activation via interactions with environmental ice and snow in the polar lower troposphere and other regions, *Atmospheric Chem. Phys.*, 12(14), 6237–6271, doi:10.5194/acp-12-6237-2012, 2012.
- Alvarez-Aviles, L., Simpson, W. R., Douglas, T. A., Sturm, M., Perovich, D. and Domine, F.: Frost flower chemical composition during growth and its implications for aerosol production and bromine activation, *J. Geophys. Res. Atmospheres*, 113(D21), doi:10.1029/2008JD010277, 2008.
- 595 Ammann, M., Cox, R. A., Crowley, J. N., Jenkin, M. E., Mellouki, A., Rossi, M. J., Troe, J. and Wallington, T. J.: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume VI – heterogeneous reactions with liquid substrates, *Atmospheric Chem. Phys.*, 13(16), 8045–8228, doi:https://doi.org/10.5194/acp-13-8045-2013, 2013.
- Anon: AMAP Assessment 2011: Mercury in the Arctic | AMAP, [online] Available from: <https://www.amap.no/documents/doc/amap-assessment-2011-mercury-in-the-arctic/90> (Accessed 13 July 2018), 2011.
- 600 Barrie, L. A., Bottenheim, J. W., Schnell, R. C., Crutzen, P. J. and Rasmussen, R. A.: Ozone destruction and photochemical reactions at polar sunrise in the lower Arctic atmosphere, *Nature*, 334(6178), 138–141, doi:10.1038/334138a0, 1988.
- Begoin, M., Richter, A., Weber, M., Kaleschke, L., Tian-Kunze, X., Stohl, A., Theys, N. and Burrows, J. P.: Satellite observations of long range transport of a large BrO plume in the Arctic, *Atmospheric Chem. Phys.*, 10(14), 6515–6526, doi:10.5194/acp-10-6515-2010, 2010.
- 605 Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q. B., Liu, H. G. Y., Mickley, L. J. and Schultz, M. G.: Global modeling of tropospheric chemistry with assimilated meteorology: Model description and evaluation, *J. Geophys. Res.-Atmospheres*, 106(D19), 23073–23095, doi:10.1029/2001JD000807, 2001.
- Blechschmidt, A.-M., Richter, A., Burrows, J. P., Kaleschke, L., Strong, K., Theys, N., Weber, M., Zhao, X. and Zien, A.: An exemplary case of a bromine explosion event linked to cyclone development in the Arctic, *Atmospheric Chem. Phys.*, 16(3), 1773–1788, doi:10.5194/acp-16-1773-2016, 2016.
- 610 Bottenheim, J. W. and Chan, E.: A trajectory study into the origin of spring time Arctic boundary layer ozone depletion, *J. Geophys. Res.-Atmospheres*, 111(D19), D19301, doi:10.1029/2006JD007055, 2006.
- Bottenheim, J. W., Natcheva, S., Morin, S. and Nghiem, S. V.: Ozone in the boundary layer air over the Arctic Ocean: measurements during the TARA transpolar drift 2006–2008, *Atmospheric Chem. Phys.*, 9(14), 4545–4557, doi:https://doi.org/10.5194/acp-9-4545-2009, 2009.
- 615 Bottenheim, Jan. W., Gallant, A. G. and Brice, K. A.: Measurements of NO_y species and O₃ at 82° N latitude, *Geophys. Res. Lett.*, 13(2), 113–116, doi:10.1029/GL013i002p00113, 1986.

- 620 Chen, Q., Schmidt, J. A., Shah, V., Jaeglé, L., Sherwen, T. and Alexander, B.: Sulfate production by reactive bromine: Implications for the global sulfur and reactive bromine budgets, *Geophys. Res. Lett.*, 44(13), 7069–7078, doi:10.1002/2017GL073812, 2017.
- 625 Choi, S., Wang, Y., Salawitch, R. J., Canty, T., Joiner, J., Zeng, T., Kurosu, T. P., Chance, K., Richter, A., Huey, L. G., Liao, J., Neuman, J. A., Nowak, J. B., Dibb, J. E., Weinheimer, A. J., Diskin, G., Ryerson, T. B., da Silva, A., Curry, J., Kinnison, D., Tilmes, S. and Levelt, P. F.: Analysis of satellite-derived Arctic tropospheric BrO columns in conjunction with aircraft measurements during ARCTAS and ARCPAC, *Atmospheric Chem. Phys.*, 12(3), 1255–1285, doi:10.5194/acp-12-1255-2012, 2012.
- 630 Choi, S., Theys, N., Salawitch, R. J., Wales, P. A., Joiner, J., Canty, T. P., Chance, K., Suleiman, R., Palm, S. P., Cullather, R. I., Darmenov, A. S., da Silva, A., Kurosu, T. P., Hendrick, F. and Van Roozendaal, M.: Link between Arctic tropospheric BrO explosion observed from space and sea salt aerosols from blowing snow investigated using Ozone Monitoring Instrument (OMI) BrO data and GEOS-5 data assimilation system, *J. Geophys. Res. Atmospheres*, 0(ja), doi:10.1029/2017JD026889, 2018.
- Custard, K. D., Raso, A. R. W., Shepson, P. B., Staebler, R. M. and Pratt, K. A.: Production and Release of Molecular Bromine and Chlorine from the Arctic Coastal Snowpack, *Acs Earth Space Chem.*, 1(3), 142–151, doi:10.1021/acsearthspacechem.7b00014, 2017.
- 635 Domine, F., Sparapani, R., Ianniello, A. and Beine, H. J.: The origin of sea salt in snow on Arctic sea ice and in coastal regions, *Atmospheric Chem. Phys.*, 4(9/10), 2259–2271, doi:https://doi.org/10.5194/acp-4-2259-2004, 2004.
- Donlon, C. J., Martin, M., Stark, J., Roberts-Jones, J., Fiedler, E. and Wimmer, W.: The Operational Sea Surface Temperature and Sea Ice Analysis (OSTIA) system, *Remote Sens. Environ.*, 116, 140–158, doi:10.1016/j.rse.2010.10.017, 2012.
- Ebinghaus, R., Kock, H. H., Temme, C., Einax, J. W., Lowe, A. G., Richter, A., Burrows, J. P. and Schroeder, W. H.: Antarctic springtime depletion of atmospheric mercury, *Environ. Sci. Technol.*, 36(6), 1238–1244, doi:10.1021/es015710z, 2002.
- 640 Falk, S. and Sinnhuber, B.-M.: Polar boundary layer bromine explosion and ozone depletion events in the chemistry-climate model EMAC v2.52: implementation and evaluation of AirSnow algorithm, *Geosci. Model Dev.*, 11(3), 1115–1131, doi:10.5194/gmd-11-1115-2018, 2018.
- Fan, S.-M. and Jacob, D. J.: Surface ozone depletion in Arctic spring sustained by bromine reactions on aerosols, *Nature*, 359(6395), 522–524, doi:10.1038/359522a0, 1992.
- 645 Fischer, E. V., Jacob, D. J., Yantosca, R. M., Sulprizio, M. P., Millet, D. B., Mao, J., Paulot, F., Singh, H. B., Roiger, A., Ries, L., Talbot, R. W., Dzepina, K. and Pandey Deolal, S.: Atmospheric peroxyacetyl nitrate (PAN): a global budget and source attribution, *Atmospheric Chem. Phys.*, 14(5), 2679–2698, doi:https://doi.org/10.5194/acp-14-2679-2014, 2014.
- 650 Fisher, J. A., Jacob, D. J., Travis, K. R., Kim, P. S., Marais, E. A., Chan Miller, C., Yu, K., Zhu, L., Yantosca, R. M., Sulprizio, M. P., Mao, J., Wennberg, P. O., Crouse, J. D., Teng, A. P., Nguyen, T. B., Clair, J. M. S., Cohen, R. C., Romer, P., Nault, B. A., Wooldridge, P. J., Jimenez, J. L., Campuzano-Jost, P., Day, D. A., Hu, W., Shepson, P. B., Xiong, F., Blake, D. R., Goldstein, A. H., Misztal, P. K., Hanisco, T. F., Wolfe, G. M., Ryerson, T. B., Wisthaler, A. and Mikoviny, T.: Organic nitrate chemistry and its implications for nitrogen budgets in an isoprene- and monoterpene-rich atmosphere: constraints from aircraft (SEAC⁴RS) and ground-based (SOAS) observations in the Southeast US, *Atmospheric Chem. Phys.*, 16(9), 5969–5991, doi:https://doi.org/10.5194/acp-16-5969-2016, 2016.

- 655 Frey, M. M., Norris, S. J., Brooks, I. M., Anderson, P. S., Nishimura, K., Yang, X., Jones, A. E., Nerentorp Mastromonaco, M. G., Jones, D. H. and Wolff, E. W.: First direct observation of sea salt aerosol production from blowing snow above sea ice, *Atmospheric Chem. Phys. Discuss.*, 1–53, doi:<https://doi.org/10.5194/acp-2019-259>, 2019.
- Frieß, U., Hollwedel, J., König-Langlo, G., Wagner, T. and Platt, U.: Dynamics and chemistry of tropospheric bromine explosion events in the Antarctic coastal region, *J. Geophys. Res.-Atmospheres*, 109(D6), D06305, doi:[10.1029/2003JD004133](https://doi.org/10.1029/2003JD004133), 2004.
- 660 Gelaro, R., McCarty, W., Suarez, M. J., Todling, R., Molod, A., Takacs, L., Randles, C. A., Darmenov, A., Bosilovich, M. G., Reichle, R., Wargan, K., Coy, L., Cullather, R., Draper, C., Akella, S., Buchard, V., Conaty, A., da Silva, A. M., Gu, W., Kim, G.-K., Koster, R., Lucchesi, R., Merkova, D., Nielsen, J. E., Partyka, G., Pawson, S., Putman, W., Rienecker, M., Schubert, S. D., Sienkiewicz, M. and Zhao, B.: The Modern-Era Retrospective Analysis for Research and Applications, Version 2 (MERRA-2), *J. Clim.*, 30(14), 5419–5454, doi:[10.1175/JCLI-D-16-0758.1](https://doi.org/10.1175/JCLI-D-16-0758.1), 2017.
- 665 Gilman, J. B., Burkhart, J. F., Lerner, B. M., Williams, E. J., Kuster, W. C., Goldan, P. D., Murphy, P. C., Warneke, C., Fowler, C., Montzka, S. A., Miller, B. R., Miller, L., Oltmans, S. J., Ryerson, T. B., Cooper, O. R., Stohl, A. and Gouw, J. A. de: Ozone variability and halogen oxidation within the Arctic and sub-Arctic springtime boundary layer, *Atmospheric Chem. Phys.*, 10(21), 10223–10236, doi:<https://doi.org/10.5194/acp-10-10223-2010>, 2010.
- 670 Giordano, M. R., Kalnajs, L. E., Goetz, J. D., Avery, A. M., Katz, E., May, N. W., Leemon, A., Mattson, C., Pratt, K. A. and DeCarlo, P. F.: The importance of blowing snow to halogen-containing aerosol in coastal Antarctica: influence of source region versus wind speed, *Atmospheric Chem. Phys.*, 18(22), 16689–16711, doi:<https://doi.org/10.5194/acp-18-16689-2018>, 2018.
- 675 Gong, S. L., Walmsley, J. L., Barrie, L. A. and Hopper, J. F.: Mechanisms for surface ozone depletion and recovery during polar sunrise, *Atmos. Environ.*, 31(7), 969–981, doi:[10.1016/S1352-2310\(96\)00264-6](https://doi.org/10.1016/S1352-2310(96)00264-6), 1997.
- [Graham, R. M., Cohen, L., Ritzhaupt, N., Segger, B., Graversen, R. G., Rinke, A., Walden, V. P., Granskog, M. A. and Hudson, S. R.: Evaluation of Six Atmospheric Reanalyses over Arctic Sea Ice from Winter to Early Summer, *J. Clim.*, 32\(14\), 4121–4143, doi:\[10.1175/JCLI-D-18-0643.1\]\(https://doi.org/10.1175/JCLI-D-18-0643.1\), 2019.](#)
- 680 Guenther, A. B., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L. K. and Wang, X.: The Model of Emissions of Gases and Aerosols from Nature version 2.1 (MEGAN2.1): an extended and updated framework for modeling biogenic emissions, *Geosci. Model Dev.*, 5(6), 1471–1492, 2012.
- 685 Halfacre, J. W., Knepp, T. N., Shepson, P. B., Thompson, C. R., Pratt, K. A., Li, B., Peterson, P. K., Walsh, S. J., Simpson, W. R., Matrai, P. A., Bottenheim, J. W., Netcheva, S., Perovich, D. K. and Richter, A.: Temporal and spatial characteristics of ozone depletion events from measurements in the Arctic, *Atmospheric Chem. Phys.*, 14(10), 4875–4894, doi:<https://doi.org/10.5194/acp-14-4875-2014>, 2014.
- Hausmann, M. and Platt, U.: Spectroscopic measurement of bromine oxide and ozone in the high Arctic during Polar Sunrise Experiment 1992, *J. Geophys. Res. Atmospheres*, 99(D12), 25399–25413, doi:[10.1029/94JD01314](https://doi.org/10.1029/94JD01314), 1994.
- Hönninger, G. and Platt, U.: Observations of BrO and its vertical distribution during surface ozone depletion at Alert, *Atmos. Environ.*, 36(15), 2481–2489, doi:[10.1016/S1352-2310\(02\)00104-8](https://doi.org/10.1016/S1352-2310(02)00104-8), 2002.
- 690 Hopper, J. F., Barrie, L. A., Silis, A., Hart, W., Gallant, A. J. and Dryfhout, H.: Ozone and meteorology during the 1994 Polar Sunrise Experiment, *J. Geophys. Res. Atmospheres*, 103(D1), 1481–1492, doi:[10.1029/97JD02888](https://doi.org/10.1029/97JD02888), 1998.

- Huang, J. and Jaeglé, L.: Wintertime enhancements of sea salt aerosol in polar regions consistent with a sea ice source from blowing snow, *Atmospheric Chem. Phys.*, 17(5), 3699–3712, doi:10.5194/acp-17-3699-2017, 2017.
- 695 Huang, J., Jaeglé, L. and Shah, V.: Using CALIOP to constrain blowing snow emissions of sea salt aerosols over Arctic and Antarctic sea ice, *Atmospheric Chem. Phys.*, 18(22), 16253–16269, doi:https://doi.org/10.5194/acp-18-16253-2018, 2018.
- Jacobi, H.-W., Kaleschke, L., Richter, A., Rozanov, A. and Burrows, J. P.: Observation of a fast ozone loss in the marginal ice zone of the Arctic Ocean, *J. Geophys. Res.-Atmospheres*, 111(D15), D15309, doi:10.1029/2005JD006715, 2006.
- 700 Jacobi, H.-W., Morin, S. and Bottenheim, J. W.: Observation of widespread depletion of ozone in the springtime boundary layer of the central Arctic linked to mesoscale synoptic conditions, *J. Geophys. Res. Atmospheres*, 115(D17), doi:10.1029/2010JD013940, 2010.
- Jaeglé, L., Quinn, P. K., Bates, T. S., Alexander, B. and Lin, J.-T.: Global distribution of sea salt aerosols: new constraints from in situ and remote sensing observations, *Atmospheric Chem. Phys.*, 11(7), 3137–3157, doi:10.5194/acp-11-3137-2011, 2011.
- 705 Jones, A. E., Anderson, P. S., Begoin, M., Brough, N., Hutterli, M. A., Marshall, G. J., Richter, A., Roscoe, H. K. and Wolff, E. W.: BrO, blizzards, and drivers of polar tropospheric ozone depletion events, *Atmos Chem Phys*, 9(14), 4639–4652, doi:10.5194/acp-9-4639-2009, 2009.
- Kaleschke, L., Richter, A., Burrows, J., Afe, O., Heygster, G., Notholt, J., Rankin, A. M., Roscoe, H. K., Hollwedel, J., Wagner, T. and Jacobi, H. W.: Frost flowers on sea ice as a source of sea salt and their influence on tropospheric halogen chemistry, *Geophys. Res. Lett.*, 31(16), L16114, doi:10.1029/2004GL020655, 2004.
- 710 Keene, W. C., Stutz, J., Pszenny, A. A. P., Maben, J. R., Fischer, E. V., Smith, A. M., von Glasow, R., Pechtl, S., Sive, B. C. and Varner, R. K.: Inorganic chlorine and bromine in coastal New England air during summer, *J. Geophys. Res. Atmospheres*, 112(D10), doi:10.1029/2006JD007689, 2007.
- 715 Koo, J.-H., Wang, Y., Kurosu, T. P., Chance, K., Rozanov, A., Richter, A., Oltmans, S. J., Thompson, A. M., Hair, J. W., Fenn, M. A., Weinheimer, A. J., Ryerson, T. B., Solberg, S., Huey, L. G., Liao, J., Dibb, J. E., Neuman, J. A., Nowak, J. B., Pierce, R. B., Natarajan, M. and Al-Saadi, J.: Characteristics of tropospheric ozone depletion events in the Arctic spring: analysis of the ARCTAS, ARCPAC, and ARCIONS measurements and satellite BrO observations, *Atmospheric Chem. Phys.*, 12(20), 9909–9922, doi:10.5194/acp-12-9909-2012, 2012.
- 720 Koop, T., Kapilashrami, A., Molina, L. T. and Molina, M. J.: Phase transitions of sea-salt/water mixtures at low temperatures: Implications for ozone chemistry in the polar marine boundary layer, *J. Geophys. Res. Atmospheres*, 105(D21), 26393–26402, doi:10.1029/2000JD900413, 2000.
- Krnavek, L., Simpson, W. R., Carlson, D., Domine, F., Douglas, T. A. and Sturm, M.: The chemical composition of surface snow in the Arctic: Examining marine, terrestrial, and atmospheric influences, *Atmos. Environ.*, 50, 349–359, doi:10.1016/j.atmosenv.2011.11.033, 2012.
- 725 Kwok, R. and Cunningham, G. F.: ICESat over Arctic sea ice: Estimation of snow depth and ice thickness, *J. Geophys. Res. Oceans*, 113(C8), doi:10.1029/2008JC004753, 2008.
- Lehrer, E., Honninger, G. and Platt, U.: A one dimensional model study of the mechanism of halogen liberation and vertical transport in the polar troposphere, *Atmospheric Chem. Phys.*, 4, 2427–2440, doi:10.5194/acp-4-2427-2004, 2004.

- Levelt, P. F., Oord, G. H. J. van den, Dobber, M. R., Malkki, A., Visser, H., Vries, J. de, Stammes, P., Lundell, J. O. V. and Saari, H.: The ozone monitoring instrument, *IEEE Trans. Geosci. Remote Sens.*, 44(5), 1093–1101, doi:10.1109/TGRS.2006.872333, 2006.
- Lewis, E. R. and Schwartz, S. E.: Fundamentals, in *Sea Salt Aerosol Production: Mechanisms, Methods, Measurements and Models*, pp. 9–99, American Geophysical Union (AGU), 2004.
- Liao, J., Huey, L. G., Scheuer, E., Dibb, J. E., Stickel, R. E., Tanner, D. J., Neuman, J. A., Nowak, J. B., Choi, S., Wang, Y., Salawitch, R. J., Canty, T., Chance, K., Kurosu, T., Suleiman, R., Weinheimer, A. J., Shetter, R. E., Fried, A., Brune, W., Anderson, B., Zhang, X., Chen, G., Crawford, J., Hecobian, A. and Ingall, E. D.: Characterization of soluble bromide measurements and a case study of BrO observations during ARCTAS, *Atmospheric Chem. Phys.*, 12(3), 1327–1338, doi:10.5194/acp-12-1327-2012, 2012.
- Lindsay, R., Wensnahan, M., Schweiger, A. and Zhang, J.: Evaluation of Seven Different Atmospheric Reanalysis Products in the Arctic, *J. Clim.*, 27(7), 2588–2606, doi:10.1175/JCLI-D-13-00014.1, 2014.
- Macdonald, K. M., Sharma, S., Toom, D., Chivulescu, A., Hanna, S., Bertram, A. K., Platt, A., Elsasser, M., Huang, L., Tarasick, D., Chellman, N., McConnell, J. R., Bozem, H., Kunkel, D., Lei, Y. D., Evans, G. J. and Abbatt, J. P. D.: Observations of atmospheric chemical deposition to high Arctic snow, *Atmospheric Chem. Phys.*, 17(9), 5775–5788, doi:https://doi.org/10.5194/acp-17-5775-2017, 2017.
- Mao, J., Paulot Fabien, Jacob Daniel J., Cohen Ronald C., Crouse John D., Wennberg Paul O., Keller Christoph A., Hudman Rynda C., Barkley Michael P. and Horowitz Larry W.: Ozone and organic nitrates over the eastern United States: Sensitivity to isoprene chemistry, *J. Geophys. Res. Atmospheres*, 118(19), 11,256–11,268, doi:10.1002/jgrd.50817, 2013.
- McConnell, J. C., Henderson, G. S., Barrie, L., Bottenheim, J., Niki, H., Langford, C. H. and Templeton, E. M. J.: Photochemical bromine production implicated in Arctic boundary-layer ozone depletion, *Nature*, 355(6356), 150–152, doi:10.1038/355150a0, 1992.
- Moore, C. W., Obrist, D., Steffen, A., Staebler, R. M., Douglas, T. A., Richter, A. and Nghiem, S. V.: Convective forcing of mercury and ozone in the Arctic boundary layer induced by leads in sea ice, *Nature*, 506(7486), 81–+, doi:10.1038/nature12924, 2014.
- Morin, S., Honninger, G. H., Staebler, R. M. and Bottenheim, J. W.: A high time resolution study of boundary layer ozone chemistry and dynamics over the Arctic Ocean near Alert, Nunavut, *Geophys. Res. Lett.*, 32(8), L08809, doi:10.1029/2004GL022098, 2005.
- Morin, S., Marion, G. M., Glasow, R. von, Voisin, D., Bouchez, J. and Savarino, J.: Precipitation of salts in freezing seawater and ozone depletion events: a status report, *Atmospheric Chem. Phys.*, 8(23), 7317–7324, doi:https://doi.org/10.5194/acp-8-7317-2008, 2008.
- Munro, M., Eisinger, M., Anderson, C., Callies, J., Corpaccioli, E., Lang, R., Lefebvre, A., Livschitz, Y. and Albinana, A. P.: GOME-2 on MetOp, vol. 1216, p. 48, Helsinki, Finland., 2006.
- Nandan, V., Geldsetzer, T., Yackel, J., Mahmud, M., Scharien, R., Howell, S., King, J., Ricker, R. and Else, B.: Effect of Snow Salinity on CryoSat-2 Arctic First-Year Sea Ice Freeboard Measurements, *Geophys. Res. Lett.*, 44(20), 10,419–10,426, doi:10.1002/2017GL074506, 2017.

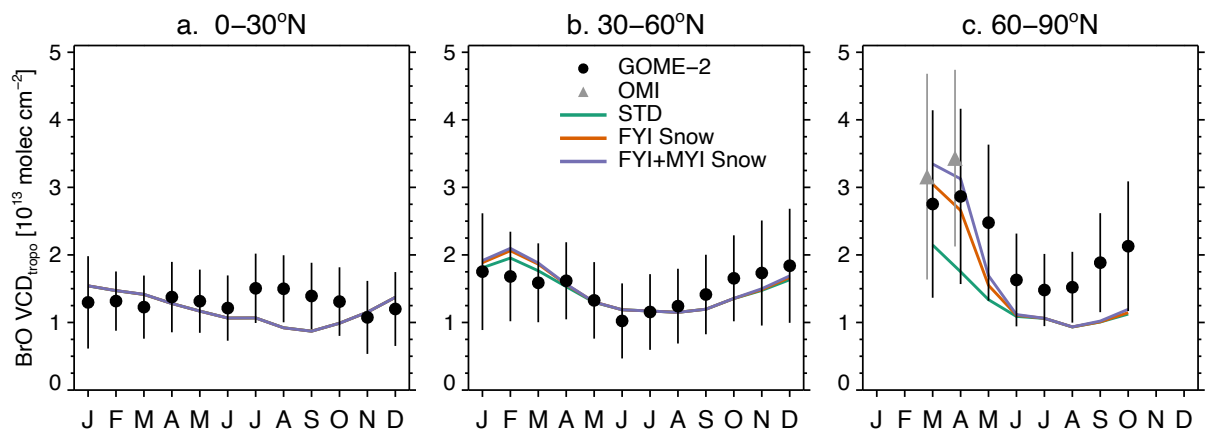
- 765 Neuman, J. A., Nowak, J. B., Huey, L. G., Burkholder, J. B., Dibb, J. E., Holloway, J. S., Liao, J., Peischl, J., Roberts, J. M., Ryerson, T. B., Scheuer, E., Stark, H., Stickel, R. E., Tanner, D. J. and Weinheimer, A.: Bromine measurements in ozone depleted air over the Arctic Ocean, *Atmospheric Chem. Phys.*, 10(14), 6503–6514, doi:10.5194/acp-10-6503-2010, 2010.
- 770 Nghiem, S. V., Rigor, I. G., Richter, A., Burrows, J. P., Shepson, P. B., Bottenheim, J., Barber, D. G., Steffen, A., Latonas, J., Wang, F., Stern, G., Clemente-Colon, P., Martin, S., Hall, D. K., Kaleschke, L., Tackett, P., Neumann, G. and Asplin, M. G.: Field and satellite observations of the formation and distribution of Arctic atmospheric bromine above a rejuvenated sea ice cover, *J. Geophys. Res.-Atmospheres*, 117, D00S05, doi:10.1029/2011JD016268, 2012.
- NOAA/USDA: The Easter Freeze of April 2007. [online] Available from: <https://www1.ncdc.noaa.gov/pub/data/techrpts/tr200801/tech-report-200801.pdf> (Accessed 31 August 2019), 2008.
- 775 Oltmans, S. J. and Komhyr, W. D.: Surface ozone distributions and variations from 1973–1984: Measurements at the NOAA Geophysical Monitoring for Climatic Change Baseline Observatories, *J. Geophys. Res. Atmospheres*, 91(D4), 5229–5236, doi:10.1029/JD091iD04p05229, 1986.
- Oltmans, S. J., Johnson, B. J. and Harris, J. M.: Springtime boundary layer ozone depletion at Barrow, Alaska: Meteorological influence, year-to-year variation, and long-term change, *J. Geophys. Res.-Atmospheres*, 117, D00R18, doi:10.1029/2011JD016889, 2012.
- 780 Parrella, J. P., Jacob, D. J., Liang, Q., Zhang, Y., Mickley, L. J., Miller, B., Evans, M. J., Yang, X., Pyle, J. A., Theys, N. and Van Roozendaal, M.: Tropospheric bromine chemistry: implications for present and pre-industrial ozone and mercury, *Atmos Chem Phys*, 12(15), 6723–6740, doi:10.5194/acp-12-6723-2012, 2012.
- Peterson, P. K., Simpson, W. R., Pratt, K. A., Shepson, P. B., Friess, U., Zielcke, J., Platt, U., Walsh, S. J. and Nghiem, S. V.: Dependence of the vertical distribution of bromine monoxide in the lower troposphere on meteorological factors such as wind speed and stability, *Atmospheric Chem. Phys.*, 15(4), 2119–2137, doi:10.5194/acp-15-2119-2015, 2015.
- 785 Peterson, P. K., Poehler, D., Sihler, H., Zielcke, J., General, S., Friess, U., Platt, U., Simpson, W. R., Nghiem, S. V., Shepson, P. B., Stirm, B. H., Dhaniyala, S., Wagner, T., Caulton, D. R., Fuentes, J. D. and Pratt, K. A.: Observations of bromine monoxide transport in the Arctic sustained on aerosol particles, *Atmospheric Chem. Phys.*, 17(12), 7567–7579, doi:10.5194/acp-17-7567-2017, 2017.
- 790 [Peterson, P. K., Pöhler, D., Zielcke, J., General, S., Frieß, U., Platt, U., Simpson, W. R., Nghiem, S. V., Shepson, P. B., Stirm, B. H. and Pratt, K. A.: Springtime Bromine Activation over Coastal and Inland Arctic Snowpacks, *ACS Earth Space Chem.*, 2\(10\), 1075–1086, doi:10.1021/acsearthspacechem.8b00083, 2018.](#)
- Peterson, P. K., Hartwig, M., May, N. W., Schwartz, E., Rigor, I., Ermold, W., Steele, M., Morison, J. H., Nghiem, S. V. and Pratt, K. A.: Snowpack measurements suggest role for multi-year sea ice regions in Arctic atmospheric bromine and chlorine chemistry, *Elem Sci Anth*, 7(1), 14, doi:10.1525/elementa.352, 2019.
- 795 Petty, A. A., Webster, M., Boisvert, L. and Markus, T.: The NASA Eulerian Snow on Sea Ice Model (NESOSIM) v1.0: initial model development and analysis, *Geosci. Model Dev.*, 11(11), 4577–4602, doi:https://doi.org/10.5194/gmd-11-4577-2018, 2018.
- Piot, M. and Glasow, R. von: The potential importance of frost flowers, recycling on snow, and open leads for ozone depletion events, *Atmospheric Chem. Phys.*, 8(9), 2437–2467, doi:https://doi.org/10.5194/acp-8-2437-2008, 2008.

- 800 Pratt, K. A., Custard, K. D., Shepson, P. B., Douglas, T. A., Poehler, D., General, S., Zielcke, J., Simpson, W. R., Platt, U., Tanner, D. J., Huey, L. G., Carlsen, M. and Stirm, B. H.: Photochemical production of molecular bromine in Arctic surface snowpacks, *Nat. Geosci.*, 6(5), 351–356, doi:10.1038/NGEO1779, 2013.
- Rankin, A. M., Wolff, E. W. and Martin, S.: Frost flowers: Implications for tropospheric chemistry and ice core interpretation, *J. Geophys. Res. Atmospheres*, 107(D23), AAC 4-1-AAC 4-15, doi:10.1029/2002JD002492, 2002.
- 805 [Rhodes, R. H., Yang, X., Wolff, E. W., McConnell, J. R. and Frey, M. M.: Sea ice as a source of sea salt aerosol to Greenland ice cores: a model-based study, *Atmospheric Chem. Phys.*, 17\(15\), 9417–9433, doi:https://doi.org/10.5194/acp-17-9417-2017, 2017.](https://doi.org/10.5194/acp-17-9417-2017)
- Richter, A., Wittrock, F., Eisinger, M. and Burrows, J. P.: GOME observations of tropospheric BrO in northern hemispheric spring and summer 1997, *Geophys. Res. Lett.*, 25(14), 2683–2686, doi:10.1029/98GL52016, 1998.
- 810 Ridley, B. A., Zeng, T., Wang, Y., Atlas, E. L., Browell, E. V., Hess, P. G., Orlando, J. J., Chance, K. and Richter, A.: An ozone depletion event in the sub-arctic surface layer over Hudson Bay, Canada, *J. Atmospheric Chem.*, 57(3), 255–280, doi:10.1007/s10874-007-9072-z, 2007.
- Roscoe, H. K., Brooks, B., Jackson, A. V., Smith, M. H., Walker, S. J., Obbard, R. W. and Wolff, E. W.: Frost flowers in the laboratory: Growth, characteristics, aerosol, and the underlying sea ice, *J. Geophys. Res. Atmospheres*, 116(D12),
815 doi:10.1029/2010JD015144, 2011.
- Salawitch, R. J., Canty, T., Kurosu, T., Chance, K., Liang, Q., da Silva, A., Pawson, S., Nielsen, J. E., Rodriguez, J. M., Bhartia, P. K., Liu, X., Huey, L. G., Liao, J., Stickel, R. E., Tanner, D. J., Dibb, J. E., Simpson, W. R., Donohue, D., Weinheimer, A., Flocke, F., Knapp, D., Montzka, D., Neuman, J. A., Nowak, J. B., Ryerson, T. B., Oltmans, S., Blake, D. R., Atlas, E. L., Kinnison, D. E., Tilmes, S., Pan, L. L., Hendrick, F., Van Roozendaal, M., Kreher, K., Johnston, P. V., Gao, R.
820 S., Johnson, B., Bui, T. P., Chen, G., Pierce, R. B., Crawford, J. H. and Jacob, D. J.: A new interpretation of total column BrO during Arctic spring, *Geophys. Res. Lett.*, 37, L21805, doi:10.1029/2010GL043798, 2010.
- Sander, R., Keene, W. C., Pszenny, A. a. P., Arimoto, R., Ayers, G. P., Baboukas, E., Caine, J. M., Crutzen, P. J., Duce, R. A., Hönninger, G., Huebert, B. J., Maenhaut, W., Mihalopoulos, N., Turekian, V. C. and Dingenen, R. V.: Inorganic bromine in the marine boundary layer: a critical review, *Atmospheric Chem. Phys.*, 3(5), 1301–1336, doi:https://doi.org/10.5194/acp-3-1301-2003, 2003.
825
- Schmidt, J. A., Jacob D. J., Horowitz H. M., Hu L., Sherwen T., Evans M. J., Liang Q., Suleiman R. M., Oram D. E., Le Breton M., Percival C. J., Wang S., Dix B. and Volkamer R.: Modeling the observed tropospheric BrO background: Importance of multiphase chemistry and implications for ozone, OH, and mercury, *J. Geophys. Res. Atmospheres*, 121(19), 11,819–11,835, doi:10.1002/2015JD024229, 2016.
- 830 Schroeder, W. H., Anlauf, K. G., Barrie, L. A., Lu, J. Y., Steffen, A., Schneeberger, D. R. and Berg, T.: Arctic springtime depletion of mercury, *Nature*, 394(6691), 331–332, doi:10.1038/28530, 1998.
- Seabrook, J. A., Whiteway, J., Staebler, R. M., Bottenheim, J. W., Komguem, L., Gray, L. H., Barber, D. and Asplin, M.: LIDAR measurements of Arctic boundary layer ozone depletion events over the frozen Arctic Ocean, *J. Geophys. Res. Atmospheres*, 116(D17), doi:10.1029/2011JD016335, 2011.
- 835 Sherwen, T., Schmidt, J. A., Evans, M. J., Carpenter, L. J., Großmann, K., Eastham, S. D., Jacob, D. J., Dix, B., Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A., Prados-Roman, C., Mahajan, A. S. and Ordóñez, C.: Global impacts of tropospheric halogens (Cl, Br, I) on oxidants and composition in GEOS-Chem, *Atmos Chem Phys*, 16(18), 12239–12271, doi:10.5194/acp-16-12239-2016, 2016a.

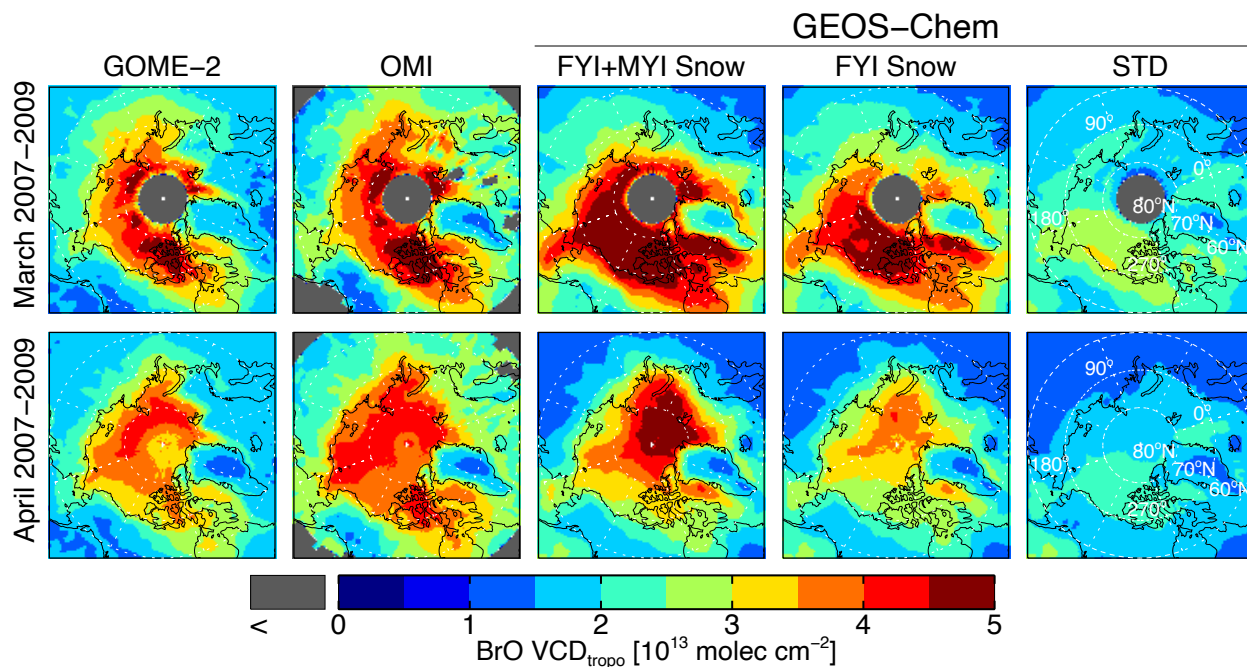
- 840 Sherwen, T., Evans, M. J., Carpenter, L. J., Andrews, S. J., Lidster, R. T., Dix, B., Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A., Prados-Roman, C., Mahajan, A. S. and Ordóñez, C.: Iodine's impact on tropospheric oxidants: a global model study in GEOS-Chem, *Atmos Chem Phys*, 16(2), 1161–1186, doi:10.5194/acp-16-1161-2016, 2016b.
- Simpson, W. R., Alvarez-Aviles, L., Douglas, T. A., Sturm, M. and Domine, F.: Halogens in the coastal snow pack near Barrow, Alaska: Evidence for active bromine air-snow chemistry during springtime, *Geophys. Res. Lett.*, 32(4), doi:10.1029/2004GL021748, 2005.
- 845 Simpson, W. R., Carlson, D., Hönninger, G., Douglas, T. A., Sturm, M., Perovich, D. and Platt, U.: First-year sea-ice contact predicts bromine monoxide (BrO) levels at Barrow, Alaska better than potential frost flower contact, *Atmospheric Chem. Phys.*, 7(3), 621–627, doi:https://doi.org/10.5194/acp-7-621-2007, 2007a.
- 850 Simpson, W. R., von Glasow, R., Riedel, K., Anderson, P., Ariya, P., Bottenheim, J., Burrows, J., Carpenter, L. J., Frieß, U., Goodsite, M. E., Heard, D., Hutterli, M., Jacobi, H.-W., Kaleschke, L., Neff, B., Plane, J., Platt, U., Richter, A., Roscoe, H., Sander, R., Shepson, P., Sodeau, J., Steffen, A., Wagner, T. and Wolff, E.: Halogens and their role in polar boundary-layer ozone depletion, *Atmospheric Chem. Phys.*, 7(16), 4375–4418, doi:10.5194/acp-7-4375-2007, 2007b.
- Simpson, W. R., Peterson, P. K., Frieß, U., Sihler, H., Lampel, J., Platt, U., Moore, C., Pratt, K., Shepson, P., Halfacre, J. and Nghiem, S. V.: Horizontal and vertical structure of reactive bromine events probed by bromine monoxide MAX-DOAS, *Atmos Chem Phys*, 17(15), 9291–9309, doi:10.5194/acp-17-9291-2017, 2017.
- 855 Spolaor, A., Gabrieli, J., Martma, T., Kohler, J., Björkman, M. B., Isaksson, E., Varin, C., Vallelonga, P., Plane, J. M. C. and Barbante, C.: Sea ice dynamics influence halogen deposition to Svalbard, *The Cryosphere*, 7(5), 1645–1658, doi:https://doi.org/10.5194/tc-7-1645-2013, 2013.
- 860 Spolaor, A., Vallelonga, P., Gabrieli, J., Martma, T., Björkman, M. P., Isaksson, E., Cozzi, G., Turetta, C., Kjær, H. A., Curran, M. a. J., Moy, A. D., Schönhardt, A., Blechschmidt, A.-M., Burrows, J. P., Plane, J. M. C. and Barbante, C.: Seasonality of halogen deposition in polar snow and ice, *Atmospheric Chem. Phys.*, 14(18), 9613–9622, doi:https://doi.org/10.5194/acp-14-9613-2014, 2014.
- 865 Steffen, A., Douglas, T., Amyot, M., Ariya, P., Aspö, K., Berg, T., Bottenheim, J., Brooks, S., Cobbett, F., Dastoor, A., Dommergue, A., Ebinghaus, R., Ferrari, C., Gardfeldt, K., Goodsite, M. E., Lean, D., Poulain, A. J., Scherz, C., Skov, H., Sommar, J. and Temme, C.: A synthesis of atmospheric mercury depletion event chemistry in the atmosphere and snow, *Atmospheric Chem. Phys.*, 8(6), 1445–1482, doi:10.5194/acp-8-1445-2008, 2008.
- Tarasick, D. W. and Bottenheim, J. W.: Surface ozone depletion episodes in the Arctic and Antarctic from historical ozonesonde records, *Atmospheric Chem. Phys.*, 2(3), 197–205, doi:https://doi.org/10.5194/acp-2-197-2002, 2002.
- 870 Theys, N., Roozendaal, M. V., Errera, Q., Hendrick, F., Daerden, F., Chabrilat, S., Dorf, M., Pfeilsticker, K., Rozanov, A., Lotz, W., Burrows, J. P., Lambert, J.-C., Goutail, F., Roscoe, H. K. and Mazière, M. D.: A global stratospheric bromine monoxide climatology based on the BASCOE chemical transport model, *Atmospheric Chem. Phys.*, 9(3), 831–848, doi:https://doi.org/10.5194/acp-9-831-2009, 2009.
- Theys, N., Van Roozendaal, M., Hendrick, F., Yang, X., De Smedt, I., Richter, A., Begoin, M., Errera, Q., Johnston, P. V., Kreher, K. and De Mazière, M.: Global observations of tropospheric BrO columns using GOME-2 satellite data, *Atmos Chem Phys*, 11(4), 1791–1811, doi:10.5194/acp-11-1791-2011, 2011.
- 875 Toom-Saunty, D. and Barrie, L. A.: Chemical composition of snowfall in the high Arctic: 1990–1994, *Atmos. Environ.*, 36(15), 2683–2693, doi:10.1016/S1352-2310(02)00115-2, 2002.

- 880 Toyota, K., McConnell, J. C., Lupu, A., Neary, L., McLinden, C. A., Richter, A., Kwok, R., Semeniuk, K., Kaminski, J. W., Gong, S.-L., Jarosz, J., Chipperfield, M. P. and Sioris, C. E.: Analysis of reactive bromine production and ozone depletion in the Arctic boundary layer using 3-D simulations with GEM-AQ: inference from synoptic-scale patterns, *Atmospheric Chem. Phys.*, 11(8), 3949–3979, doi:10.5194/acp-11-3949-2011, 2011.
- Travis, K. R., Jacob, D. J., Fisher, J. A., Kim, P. S., Marais, E. A., Zhu, L., Yu, K., Miller, C. C., Yantosca, R. M., Sulprizio, M. P., Thompson, A. M., Wennberg, P. O., Crounse, J. D., Clair, J. M. S., Cohen, R. C., Lauher, J. L., Dibb, J. E., Hall, S. R., Ullman, K., Wolfe, G. M., Pollack, I. B., Peischl, J., Neuman, J. A. and Zhou, X.: Why do models overestimate surface ozone in the Southeast United States?, *Atmos Chem Phys*, 16, 13–561, doi:10.5194/acp-16-13561-2016, 2016.
- 885 Uotila, P., Goosse, H., Haines, K., Chevallier, M., Barthélemy, A., Bricaud, C., Carton, J., Fučkar, N., Garric, G., Iovino, D., Kauker, F., Korhonen, M., Lien, V. S., Marnela, M., Massonnet, F., Mignac, D., Peterson, K. A., Sadikni, R., Shi, L., Tietsche, S., Toyoda, T., Xie, J. and Zhang, Z.: An assessment of ten ocean reanalyses in the polar regions, *Clim. Dyn.*, 52(3), 1613–1650, doi:10.1007/s00382-018-4242-z, 2019.
- 890 Wagner, T., Leue, C., Wenig, M., Pfeilsticker, K. and Platt, U.: Spatial and temporal distribution of enhanced boundary layer BrO concentrations measured by the GOME instrument aboard ERS-2, *J. Geophys. Res. Atmospheres*, 106(D20), 24225–24235, doi:10.1029/2000JD000201, 2001.
- Wagner, T., Ibrahim, O., Sinreich, R., Frieß, U., von Glasow, R. and Platt, U.: Enhanced tropospheric BrO over Antarctic sea ice in mid winter observed by MAX-DOAS on board the research vessel Polarstern, *Atmospheric Chem. Phys.*, 7(12), 3129–3142, doi:10.5194/acp-7-3129-2007, 2007.
- 895 Wang, S., McNamara, S. M., Moore, C. W., Obrist, D., Steffen, A., Shepson, P. B., Staebler, R. M., Raso, A. R. W. and Pratt, K. A.: Direct detection of atmospheric atomic bromine leading to mercury and ozone depletion, *Proc. Natl. Acad. Sci.*, 116(29), 14479–14484, doi:10.1073/pnas.1900613116, 2019a.
- 900 [Wang, X., Jacob, D. J., Eastham, S. D., Sulprizio, M. P., Zhu, L., Chen, Q., Alexander, B., Sherwen, T., Evans, M. J., Lee, B. H., Haskins, J. D., Lopez-Hilfiker, F. D., Thornton, J. A., Huey, G. L. and Liao, H.: The role of chlorine in global tropospheric chemistry, *Atmospheric Chem. Phys.*, 19\(6\), 3981–4003, doi:https://doi.org/10.5194/acp-19-3981-2019, 2019b.](https://doi.org/10.5194/acp-19-3981-2019)
- Warren, S. G., Rigor, I. G., Untersteiner, N., Radionov, V. F., Bryazgin, N. N., Aleksandrov, Y. I. and Colony, R.: Snow Depth on Arctic Sea Ice, *J. Clim.*, 12(6), 1814–1829, doi:10.1175/1520-0442(1999)012<1814:SDOASI>2.0.CO;2, 1999.
- van der Werf, G. R., Randerson, J. T., Giglio, L., van Leeuwen, T. T., Chen, Y., Rogers, B. M., Mu, M., van Marle, M. J. E., Morton, D. C., Collatz, G. J., Yokelson, R. J. and Kasibhatla, P. S.: Global fire emissions estimates during 1997–2016, *Earth Syst. Sci. Data*, 9(2), 697–720, 2017.
- 905 Wren, S. N., Donaldson, D. J. and Abbatt, J. P. D.: Photochemical chlorine and bromine activation from artificial saline snow, *Atmospheric Chem. Phys.*, 13(19), 9789–9800, doi:https://doi.org/10.5194/acp-13-9789-2013, 2013.
- Yang, X., Pyle, J. A. and Cox, R. A.: Sea salt aerosol production and bromine release: Role of snow on sea ice, *Geophys. Res. Lett.*, 35(16), doi:10.1029/2008GL034536, 2008.
- 910 Yang, X., Pyle, J. A., Cox, R. A., Theys, N. and Van Roozendaal, M.: Snow-sourced bromine and its implications for polar tropospheric ozone, *Atmospheric Chem. Phys.*, 10(16), 7763–7773, doi:10.5194/acp-10-7763-2010, 2010.
- Yang, X., Neděla, V., Runštuk, J., Ondrušková, G., Krausko, J., Vetráková, L. and Heger, D.: Evaporating brine from frost flowers with electron microscopy and implications for atmospheric chemistry and sea-salt aerosol formation, *Atmospheric Chem. Phys.*, 17(10), 6291–6303, doi:https://doi.org/10.5194/acp-17-6291-2017, 2017.

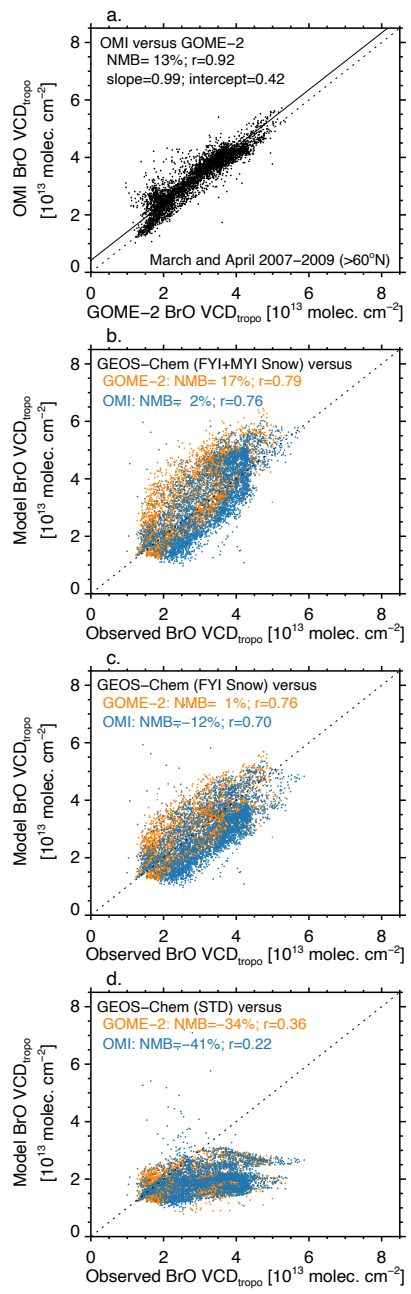
- 915 Zhao, X., Strong, K., Adams, C., Schofield, R., Yang, X., Richter, A., Friess, U., Blechschmidt, A.-M. and Koo, J.-H.: A case study of a transported bromine explosion event in the Canadian high arctic, *J. Geophys. Res.-Atmospheres*, 121(1), 457–477, doi:10.1002/2015JD023711, 2016.
- Zhu, L., Jacob, D. J., Eastham, S. D., Sulprizio, M. P., Wang, X., Sherwen, T., Evans, M. J., Chen, Q., Alexander, B., Koenig, T. K., Volkamer, R., Huey, L. G., Breton, M. L., Bannan, T. J. and Percival, C. J.: Effect of sea salt aerosol on tropospheric bromine chemistry, *Atmospheric Chem. Phys.*, 19(9), 6497–6507, doi:<https://doi.org/10.5194/acp-19-6497-2019>, 2019.
- 920 Zygmuntowska, M., Rampal, P., Ivanova, N. and Smedsrud, L. H.: Uncertainties in Arctic sea ice thickness and volume: new estimates and implications for trends, *The Cryosphere*, 8(2), 705–720, doi:<https://doi.org/10.5194/tc-8-705-2014>, 2014.



930 **Figure 1:** Seasonal variation in 2007-2009 monthly mean tropospheric BrO vertical column densities (VCD_{tropo} in units of 10^{13} molecules cm^{-2}) observed by GOME-2 (black circles) and OMI (gray triangles in panel c for April and May), and simulated with GEOS-Chem over (a) $0-30^\circ\text{N}$, (b) $30-60^\circ\text{N}$ and (c) $60-90^\circ\text{N}$. The three GEOS-Chem simulations shown are the standard simulation (STD, **green** line) and two blowing snow simulations (FYI Snow, **orange** line and FYI+MYI Snow, **purple** line). The black and gray error bars represent 1 standard deviation about the monthly mean GOME-2 and OMI VCD_{tropo} . For each latitude bin, we only show means for months where at least 70% of the surface area has valid satellite observations.



935 **Figure 2:** Spatial distribution of VCD_{tropo} in March (top row) and April (bottom row) 2007-2009. Satellite retrievals from GOME-2 and OMI are compared to the GEOS-Chem FYI+MYI Snow, FYI Snow, and STD simulations.



940 **Figure 3:** (a) Scatterplot of GOME-2 and OMI monthly mean VCD_{tropo} for March and April 2007-2009 poleward of 60° N. (b) Scatterplot of VCD_{tropo} calculated with the FYI+MYI Snow simulation and retrieved by GOME-2 (orange circles) and OMI (blue circles). (c) Same as (b) but for the FYI Snow simulation. (d) Same as (b) but for the STD simulation. The Normalized Mean Bias (NMB, see definition in section 3.1), Pearson correlation coefficient, and slope are shown in each panel. For panel (b), the slope and intercept of the reduced-major-axis regression line (solid line) are also indicated. The dashed line corresponds to the 1:1 line.

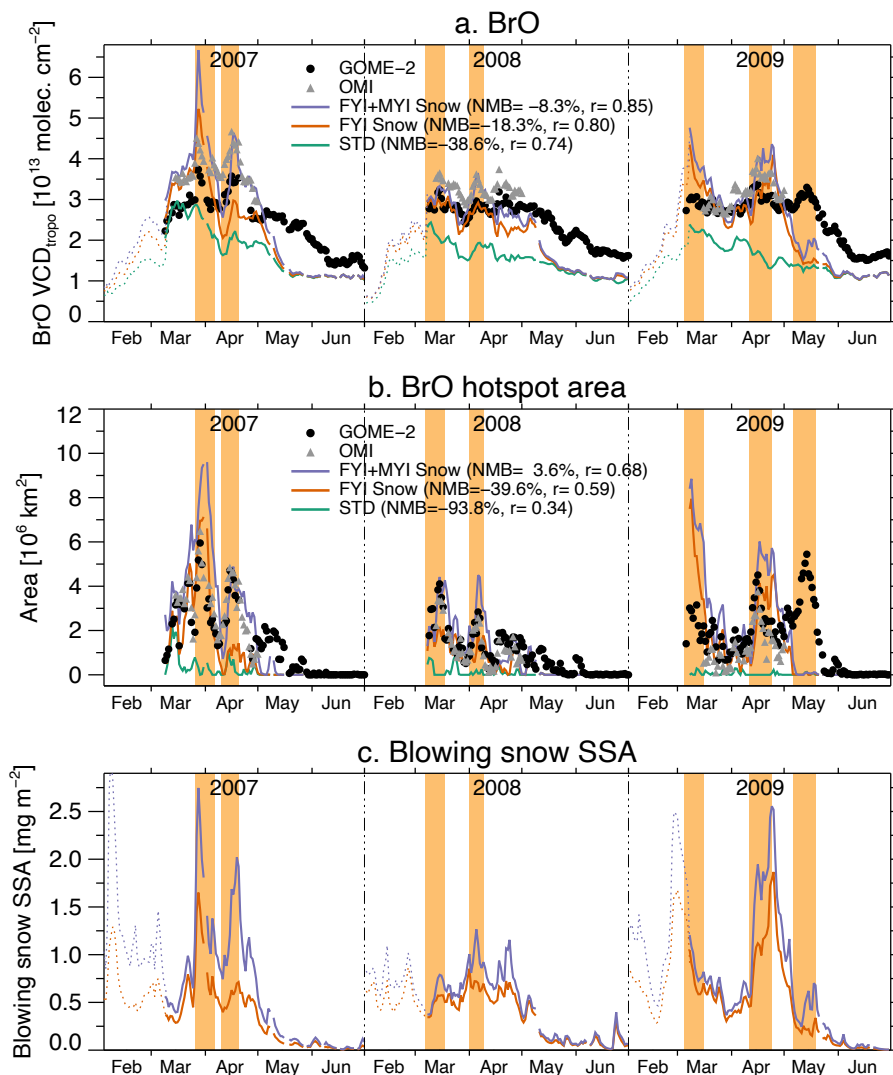


Figure 4: Timeseries of daily mean BrO VCD_{tropo}, tropospheric BrO hotspots area, and blowing snow SSA burden over high latitudes (>60° N) between February and June for 2007–2009. (a) Timeseries of daily VCD_{tropo} (in 10^{13} molecules cm^{-2}) averaged poleward of 60° N for GOME-2 (black circles), OMI (gray triangles) and simulated with GEOS-Chem (FYI+MYI Snow: purple, FYI Snow: orange, STD: green). (b) The daily area extent of BrO hotspots (in units of 10^6 km^2) for GOME-2, OMI, and GEOS-Chem. See section 3.2 for the definition of BrO hotspot areas. Note that the total area poleward of 60° N is $34.2 \cdot 10^6$ km^2 . (c) Timeseries of daily mean blowing snow SSA burden (mg m^{-2}) for the FYI+MYI Snow and FYI Snow simulations. The blowing snow SSA burden is obtained as the difference between the blowing snow and STD simulations. The normalized mean bias (NMB) and Pearson correlation coefficients displayed in panels (a) and (b) are relative to the combined GOME-2 and OMI timeseries. The events highlighted in light orange are defined as periods when GOME-2 or OMI BrO hotspots cover more than $2 \cdot 10^6$ km^2 for longer than 5 days. We only show days where valid satellite observations are available over at least 70% of the surface area poleward of 60° N. The dashed lines in (a) and (c) correspond to the GEOS-Chem daily means poleward of 60° N for days with less than 70% of valid observations.

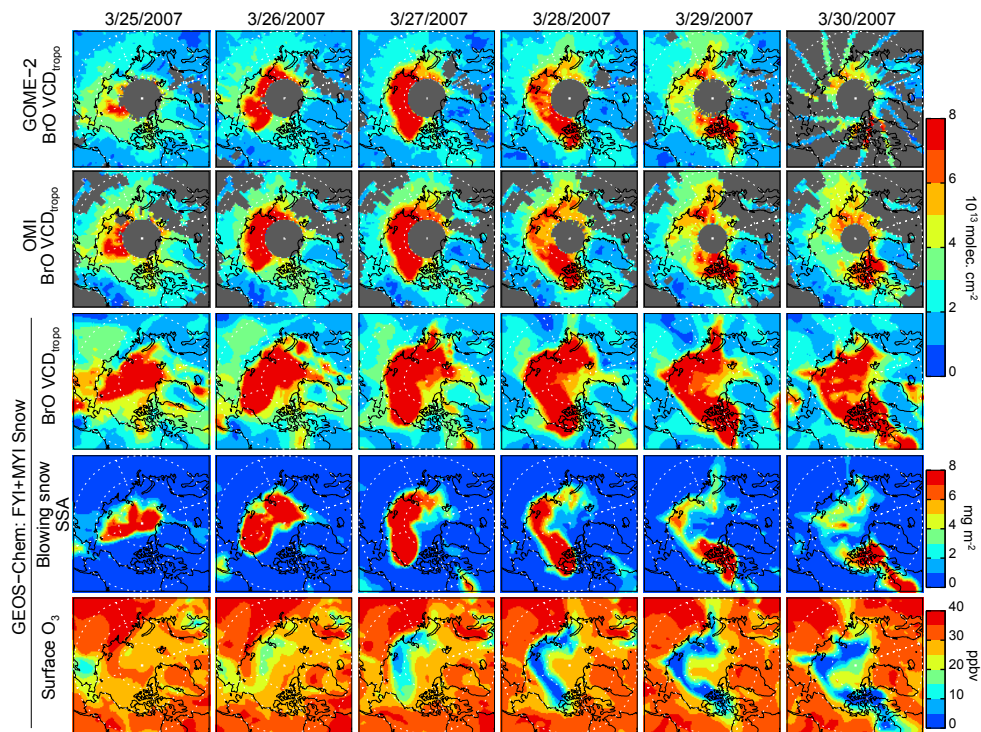
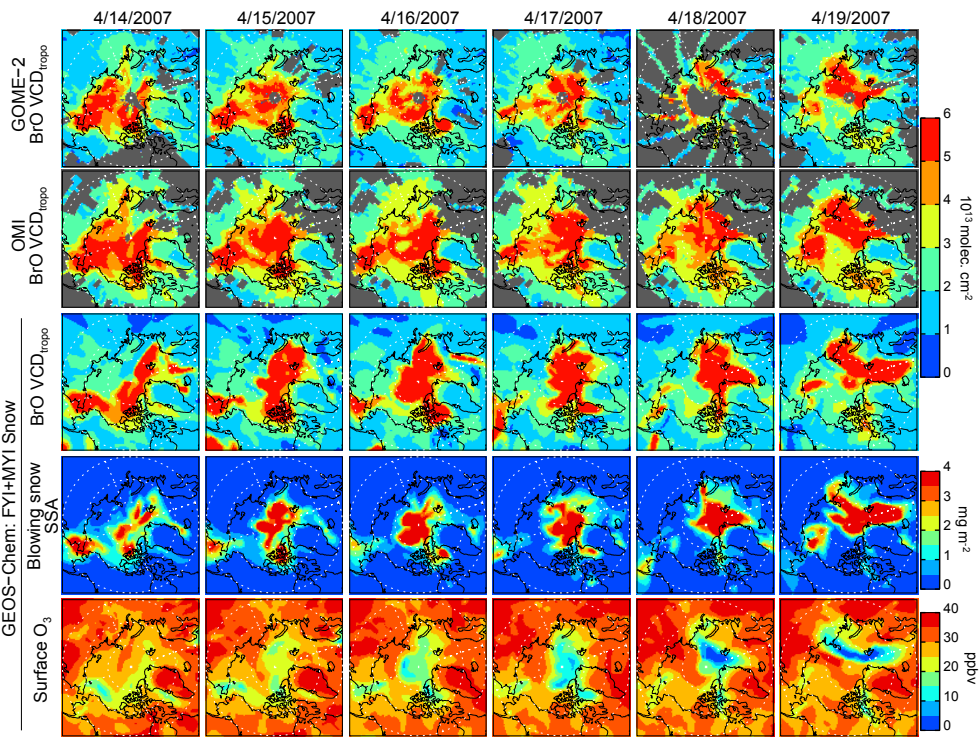
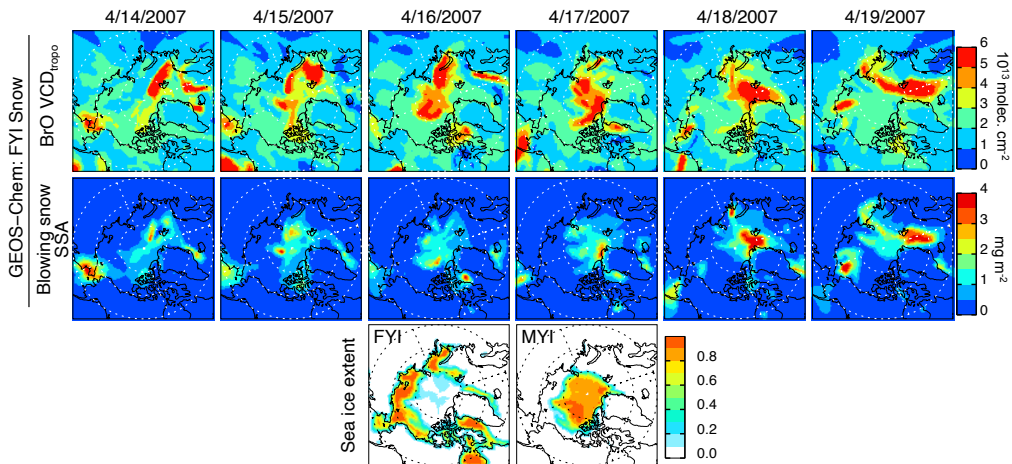


Figure 5: Daily 25-30 March 2007 distribution of BrO VCD_{tropo} for GOME-2, OMI and GEOS-Chem (FYI+MYI Snow simulation). Also shown are the blowing snow SSA mass burdens (mg m⁻²) and surface O₃ mixing ratios (ppbv) for the FYI+MYI Snow simulation. The gray areas in the first two rows correspond to regions with no BrO retrievals. Note that polar sunrise occurred equatorward of 80° N.

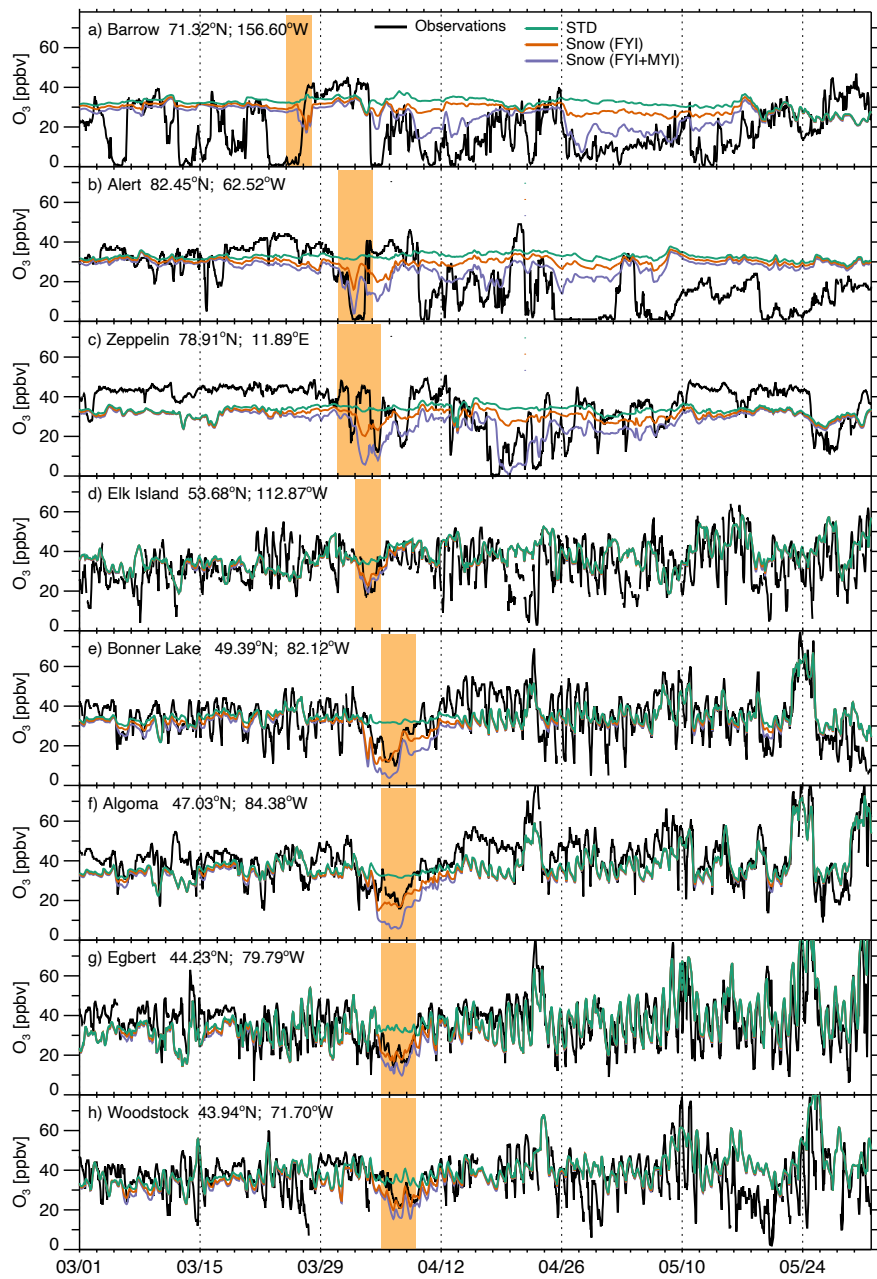
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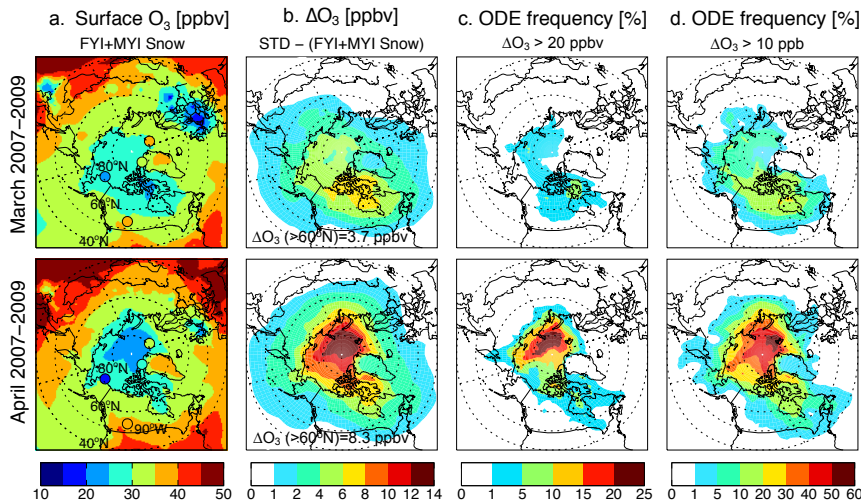
970 **Figure 6: Daily 14-19 April 2007 distribution of BrO VCD_{tropo} from GOME-2, OMI and GEOS-Chem (FYI+MYI Snow simulation). Also shown are the blowing snow SSA burdens (mg m⁻²) and surface O₃ mixing ratios (ppbv) in the FYI+MYI Snow simulation.**



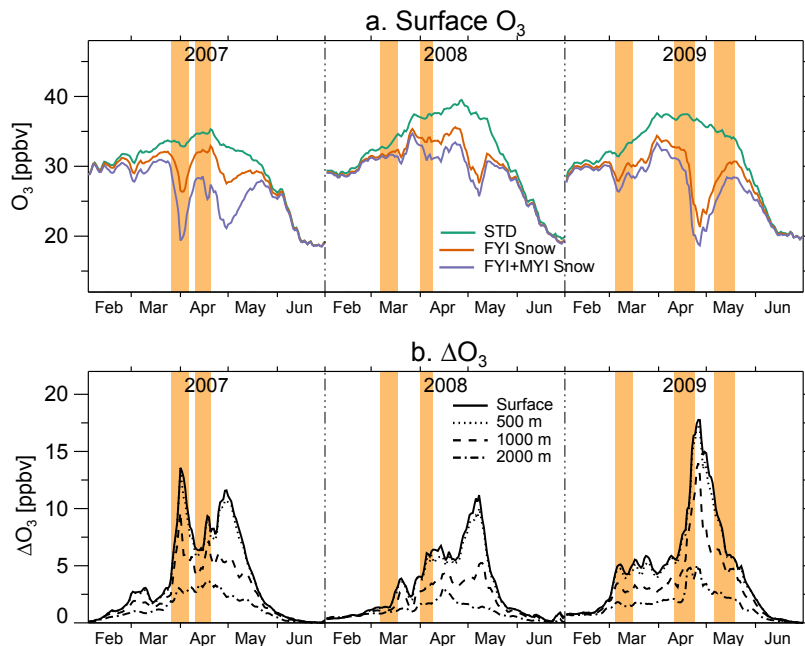
975 **Figure 7: Daily distribution of BrO VCD_{tropo} and blowing snow SSA burden from the GEOS-Chem FYI Snow simulation for 14-19 April 2007. The bottom row shows the distribution of sea ice extent for FYI and MYI.**



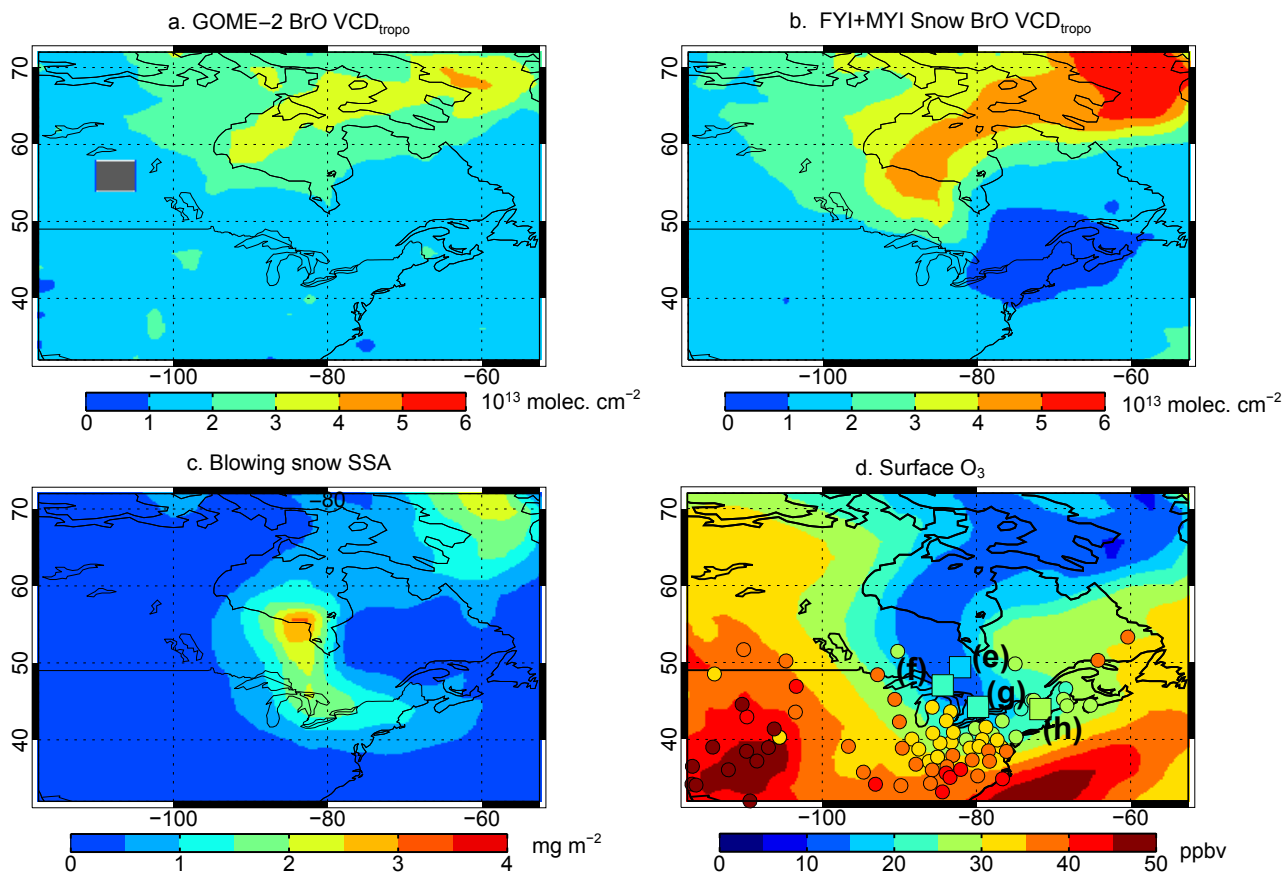
980 **Figure 8: Timeseries of hourly surface O₃ (ppbv) at three Arctic sites (a,b,c) and 5 sub-arctic sites (d-h) for March 1-May 31 2007:**
a) Utqiagvik (Barrow), Alaska, USA, b) Alert, Nunavut, Canada, c) Zeppelin, Spitsbergen, Norway, d) Elk Island, Alberta,
Canada, e) Bonner Lake, Ontario, Canada, f) Algoma, Ontario, Canada, g) Egbert, Ontario, Canada h) Woodstock, New
Hampshire, USA. In situ observations are shown with the black lines, while the GEOS-Chem STD, FYI Snow, and FYI+MYI
Snow simulations are shown in green, orange, and purple, respectively. The shaded orange areas corresponding to the times when the large BrO hotspot from 26 March - 4 April 2007 is predicted to be observed at the 3 Arctic sites and then transported to the
985 **sub-arctic sites.**



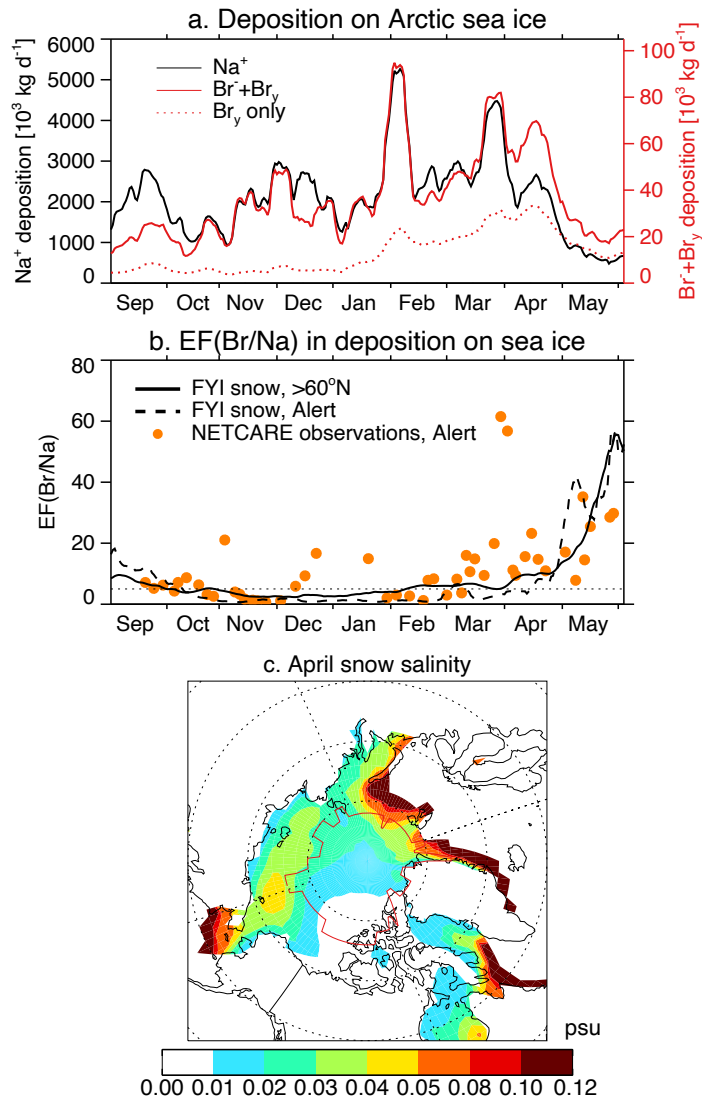
990 **Figure 9:** Monthly mean distribution of surface O_3 simulated with GEOS-Chem for March (top row) and April (bottom row) 2007-
 2009. Column a) Surface O_3 mixing ratios in the GEOS-Chem FYI+MYI Snow simulation. The circles are color-coded by monthly
 mean surface O_3 observed at 3 Arctic (Utqiagvik, Alert, Zeppelin) and 1 sub-Arctic site (Elk Island). Column b) Decrease in
 surface O_3 (ΔO_3 , ppbv) due to blowing snow obtained as the difference between the STD and FYI+MYI Snow simulations. Column
 995 c): Occurrence frequency of ODEs (in %), defined as the percent of time that more than 20 ppbv of O_3 is lost due to blowing snow
 ($\Delta O_3 > 20$ ppbv). Column d): Occurrence frequency of ODEs, defined as $\Delta O_3 > 10$ ppbv.



1000 **Figure 10:** Timeseries of daily mean surface O_3 and decrease in surface O_3 (ΔO_3) due to blowing snow at high latitudes ($>60^\circ N$)
 between February and June 2007-2009. (a) Daily mean surface O_3 mixing ratios averaged poleward of $60^\circ N$ for the STD (blue
 line), FYI Snow (purple) and FYI+MYI Snow (red line) GEOS-Chem simulations. (b) Daily mean ΔO_3 , obtained as the difference
 between the FYI+MYI Snow and STD simulations at the surface (solid line), 500 m altitude (dotted line), 1000 m altitude (dashed
 line), and 2000 m altitude (dashed-dotted line). The events highlighted in light orange are defined as periods when GOME-2 or
 OMI BrO hotspots cover more than $2 \cdot 10^6$ km² for longer than 5 days.



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Figure 11: Mean 5-9 April 2007 VCD_{tropo} for (a) GOME-2 and (b) the GEOS-Chem FYI+MYI Snow simulation. Mean 5-9 April 2007 (c) blowing snow SSA burden and (d) surface O₃ mixing ratios for April 5-9 2007 from the GEOS-Chem FYI+MYI Snow simulation. Surface O₃ observations are shown as color-coded symbols in panel (d), with the squares correspond to the sub-Arctic sites from Fig. 8. The letter numbering (e-h) corresponding to the panel numbers in Fig. 8.



015 **Figure 12: Modeled evolution of deposition and Br/Na enrichment factor on sea ice (>60° N) for 2007 calculated with the FYI Snow simulation. (a) Total deposition on sea ice for Na^+ (black line, left axis), $\text{Br}^- + \text{Br}_y$ (red line, right axis), and Br_y only (dashed red line, right axis) in units of 10^3 kg/day . (b) Modeled Br/Na enrichment factor EF (black line) relative to seawater in deposition over sea ice, see section 5 for the definition of EF. The orange circles show the enrichment factors observed during the NETCARE campaign at Alert, Nunavut (Macdonald et al., 2017). The dashed black line is the modeled EF sampled at Alert. A 10-day boxcar smoothing has been applied to all the modeled time series between September and May. The horizontal dotted line represents the value EF=5, assumed for the emitted blowing snow SSA. (c) Surface snow salinity due to the cumulative SSA deposition on sea ice for February 1-April 30 assuming a 1cm snow accumulation rate for that period (see text). The red line shows the location of MYI.**

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