

Seasonal impact of biogenic very short lived bromine on lowermost stratospheric ozone between 60° N and 60° S during the 21st century

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Received: 26 November 2019 – Discussion started: 24 January 2020 Revised: 7 May 2020 – Accepted: 13 May 2020 – Published:

Abstract. ISSIBiogenic very short lived bromine (VSL^{Br}) currently represents ~ 25% of the total stratospheric bromine loading. Owing to its much shorter lifetime compared to anthropogenic long-lived bromine (e.g. halons) and 5 chlorine (e.g. chlorofluorocarbons), the impact of VSL^{Br} on ozone peaks in the lowermost stratosphere, which is a key climatic and radiative atmospheric region. Here we present a modelling study of the evolution of stratospheric ozone and its chemical loss within the tropics and at-mid-latitudes dur-¹⁰ ing the 21st century. Two different experiments are explored: considering and neglecting the additional stratospheric in-

jection of 5 ppt biogenic VSL^{Br} naturally released from the ocean. Our analysis shows that the inclusion of VSL^{Br} results in a realistic stratospheric bromine loading and improves the ¹⁵ agreement between the model and satellite observations of the total ozone column (TOC) for the 1980–2015 period at mid-latitudes. We show that the overall ozone response to VSL^{Br} at mid-latitudes follows the stratospheric evolution of long-lived inorganic chlorine and bromine through-²⁰ out the 21st century. Additional ozone loss due to VSL^{Br} is maximized during the present-day period (1990–2010), with TOC differences of -8 DU (-3%) and -5.5 DU (-2%) for the Southern Hemisphere (SH) and Northern Hemisphere (NH) mid-latitudes (MLs^{CDI}), respectively. Moreover, the ²⁵ projected TOC differences at the end of the 21st century are

 $\sim 50~\%$ lower than the values found for the present-day period.

We find that seasonal VSLBr impacts on lowermost midlatitude stratospheric ozone are influenced by the seasonality of heterogeneous inorganic chlorine reactivation pro- 30 cesses in ice crystals. Indeed, due to the more efficient reactivation of chlorine reservoirs (mainly ClONO₂ and HCl) within the colder lowermost stratosphere at SH MLs, the seasonal VSL^{Br} impact shows a small but persistent hemispheric asymmetry through the whole modelled period. Our results 35 indicate that, although the overall VSLBr-driven ozone destruction is greatest during spring, the halogen-mediated (Halog_{x-Loss}) ozone loss cycle in the lowermost mid-latitude stratosphere, during winter is comparatively more efficient than the HO_x cycle with respect to other seasons. Indeed, $_{40}$ when VSL^{Br} is considered, $Halog_{x-Loss}$ dominates wintertime lowermost stratospheric ozone loss at SH MLs between 1985 and 2020, with a contribution of inter-halogen ClO_x -BrO_x cycles to Halog_{x-Loss} of ~ 50 %.

Within the tropics, a small (< -2.5 DU) and relatively ⁴⁵ constant (~ -1 %) ozone depletion mediated by VSL^{Br} is closely related to its fixed emissions throughout the modelled period. By including the VSL^{Br} sources, the seasonal Halog_{*x*-Loss} contribution to lowermost stratospheric ozone loss is practically dominated by the BrO_{*x*} cycle, reflecting the ⁵⁰ low sensitivity of VSL^{Br} to background halogen abundances,

which allows it to drive tropical stratospheric ozone depletion, We conclude that, considering the coupling between biogenic bromine sources and seasonal changes of the chlorine, heterogeneous reactivation is a key feature for future 5 projections of lowermost mid latitude stratospheric ozone

1 Introduction

during the 21st century.

The role of bromine in stratospheric ozone depletion has been discussed in several studies (Wofsy et al., 1975; Prather ¹⁰ and Watson, 1990; Daniel et al., 1999; Dvortsov et al., 1999; Solomon, 1999 **1529**). Although bromine is much less abundant than chlorine in the atmosphere, it is known to deplete stratospheric ozone 45 to 69 times more efficiently on a per atom basis (Daniel et al., 1999; Sinnhuber et ¹⁵ al., 2009). Moreover, the ozone depletion efficiency of ac-

- tive bromine (Br and BrO) is strongly related to the available amount of activated chlorine (mainly Cl and ClO radicals) in the atmosphere (McElroy et al., 1986; Solomon et al., 1999; and references (therein(1922)) as well as to en-20 hanced sulfate aerosol loading via heterogeneous reactions
- (Salawitch et al., 2005). Consequently, even low concentrations of bromine have a relatively large impact on stratospheric ozone. In addition to anthropogenic long-lived chlorine (LL^{Cl}) and bromine (LL^{Br}) substances such as chloroflu-
- ²⁵ orocarbons (CFCs), chlorocarbons, methyl bromide, and bromofluorocarbons (halon fire suppressants), other substances with photochemical lifetimes shorter than 6 months, often referred to as very short lived (VSL) substances, have the potential to transport a significant amount of reactive halogens
- ³⁰ into the stratosphere. Owing to their short lifetimes, the impact of VSL substances on stratospheric ozone peaks in the lowermost stratosphere (Salawitch et al., 2005; Feng et al., 2007; Sinnhuber et al., 2009; Sinnhuber and Meul, 2015 [IS3; Yang et al., 2014; Hossaini et al., 2015a; Falk et al., 2017),
 ³⁵ which is an important atmospheric region because surface
- temperature and climate are most sensitive to ozone perturbations (Riese et al., 2012; Hossaini et al., 2015b; Iglesias-Suarez et al., 2018). In fact, Hossaini et al. (2015b) found that very short lived bromine (VSL^{Br}) exerts an ozone radiative effect (normalized by halogen content) that is 3.6 times larger than that arising from long-lived substances.

VSL^{Br} is produced in the ocean via the metabolism of marine organisms such as phytoplankton and macroalgae (e.g. Carpenter et al., 2007; Quack et al., 2007; Leed-⁴⁵ ham Elvidge et al., 2015), even in sea ice (e.g. Abrahamsson et al., 2018). Due to its high volatility, it is transferred into the marine boundary layer through airsea exchange (Carpenter and Liss, 2000; Quack and Wallace, 2003). The most abundant VSL^{Br} compounds re-⁵⁰ leased to the atmosphere are bromoform (CHBr₃) and dibromomethane (methylene dibromide, CH₂Br₂), followed by a minor (but non-negligible) contribution of interhalogen species (bromochloromethane, CH_2BrCl ; dibromochloromethane, $CHBr_2Cl$; and bromodichloromethane, $CHBrCl_2$). Altogether, VSL^{Br} contributes ~ 5 (3–7) ppt to stratospheric bromine, which accounted for about 25 % of total stratospheric bromine in 2016 (WMO, 2018). Moreover, this additional input of bromine is required to reconcile current stratospheric bromine trends (Salawitch et al., 2010; WMO, 2018).

Production of anthropogenic LL^{Cl} and LL^{Br} substances has been restricted by the Montreal Protocol and its subsequent amendments and adjustments (Solomon, 1999 184; WMO, 2018). After application, the stratospheric LL^{Cl} and LL^{Br} load peaked at the end of the 20th century and has 65 been decreasing at a rate that depends on their respective atmospheric lifetimes. This implies, for example, that hydrogen chloride (HCl) shows a long-term decrease at a rate of about $0.5 \% \text{ yr}^{-1}$ in the middle stratosphere between 60° N and 60° S, while total stratospheric bromine decreased at a 70 rate of about $0.75 \% \text{ yr}^{-1}$ from 2004 to 2014 (WMO, 2018). Accordingly, stratospheric ozone is expected to recover from the effects of anthropogenic halogen-induced loss on a similar timescale, which is already detectable in the Antarctic and the upper stratosphere (e.g. Solomon et al., 2016a; 75 Chipperfield et al., 2017; Dhomse et al., 2018; Strahan and Douglass, 2008; and references therein). Hossaini et al. (2015a) quantified the stratospheric injection of organic and inorganic chlorine from anthropogenic VSL^{C1} sources such as chloroform (CHCl₃), dichloromethane (CH₂Cl₂), tetra- 80 chloroethene (C₂Cl₄), trichloroethene (C₂HCl₃), and 1,2dichloroethane (CH₂ClCH₂Cl), which are not controlled by the Montreal Protocol. From their results, the total stratospheric chlorine load from VSL^{Cl} inferred for 2013 is 123 ppt, with a stratospheric injection dominated by source 85 gases ($\sim 83\%$). Moreover, the stratospheric VSL^{CL} (organic and inorganic) injection increased by $\sim 52\%$ between 2005 and 2013, mainly due to recent and ongoing growth in anthropogenic CH₂Cl₂ emissions. In fact, Hossaini et al. (2017) showed that the impact of CH_2Cl_2 on stratospheric ozone has 90 increased markedly in recent years, and if these increases continue into the future, the recovery of the earth's ozone layer could be delayed even further, offsetting some of the gains achieved by the Montreal Protocol.

The total stratospheric chlorine and bromine budgets derived for 2016 were 3.29 ppb Cl and 19.60 ppt Br, respectively (WMO, 2018), and are expected to return to their 1980 values, an arbitrary reference date before the discovery of the Antarctic ozone hole, around the middle of this century (Dhomse et al., 2018). Although we still lack scientific consensus with respect to the future evolution of VSL^{Br} oceanic source strength and stratospheric injection (WMO, 2014; Lennartz et al., 2015; Ziska et al., 2017), the continuous decrease **in** LL^{Br} leads to an increase in the relative VSL^{Br} contribution to total stratospheric bromine into the fu-105 ture. Thus, understanding the role of natural VSL^{Br} sources is a key issue for chemistry–climate projections.

The impact of additional bromine is closely related to the heterogeneous chemistry of chlorine (Solomon, 1999

- $_5$ Salawitch et al., 2005; Müller, 2012). The influence of temperature and water vapour in the heterogeneous conversion processes of inorganic-chlorine reservoirs (mostly HCl and ClONO₂) into active radicals on sulfate aerosols has been reviewed elsewhere (e.g. Anderson et al., 2012, 2017; Drdla
- ¹⁰ and Müller, 2012; Anderson and Clapp, 2018). In particular, Drdla and Müller (2012) highlighted that an increase in water vapour above background values would allow chlorine activation at higher temperatures than those observed in polar regions, which led to the hypothesis that chlorine reacti-
- ¹⁵ vation and subsequent ozone loss could occur in the lower stratosphere at mid-latitudes during summer (Robrecht et al., 2019). Indeed, the spatial and seasonal distributions of chlorine monoxide and chlorine nitrate in the monsoon regions provide indicators of heterogeneous chlorine processing in
- ²⁰ the tropical and subtropical lowermost stratosphere of the Northern Hemisphere (Solomon et al., 2016b). Moreover, heterogeneous chorine reactivation has also been observed in ice particles in cirrus clouds located near the tropopause (Borrman et al., 1996, 1997; Solomon et al., 1997; Bregman
- es et al., 2002; Thornton et al., 2003). The occurrence of cirrus clouds was reported in the lowermost stratosphere of northern high-mid-latitudes (40–75° N; Spang et al., 2015), with summertime clouds located up to ~ 3 km (or 40–50 K of potential temperature) above the tropopause (Dessler, 2009).
- ³⁰ Furthermore, von Hobe et al. (2011) suggested that, at low temperatures, cirrus ice particles may promote a significant activation of heterogeneous chlorine in the tropical upper troposphere and lower stratosphere. Finally, satellite observations indicate that chlorine activation also occurs in the
- ³⁵ Antarctic and Arctic sub-vortex regions, where processed air dispersing from the decaying vortex in spring induces rapid changes in extra-vortex trace gas abundances (Santee et al., 2011).

Previous chemistry-climate modelling studies consider-⁴⁰ ing VSL^{Br} chemistry have mainly focused on improving the model vs. observed mid-latitude ozone trends at the end of the 20th century (Feng et al., 2007; Sinnhuber et al., 2009). More recently, Sinnhuber and Meul (2015) have shown much greater ozone depletion in the lowermost stratosphere during 45 the 1979 1995 period and a greater ozone increase during the 1996-2005 period, which is in better agreement with observations due to the inclusion of VSLBr sources, Moreover, Falk et al. (2017) have reported that the projected additional VSL^{Br} impact on ozone at the end of the 21st century is re-50 duced compared to present day. These results are in agreement with those of Yang et al. (2014), who performed time slice simulations to address the sensitivity of stratospheric ozone to a speculative doubling of VSLBr sources under different LL^{Cl} scenarios. However, neither of the previous stud-⁵⁵ ies had distinguished the contribution of ozone depletion – including its seasonal variability – that arises from VSL^{Br} with respect to LL^{Cl} and LL^{Br} throughout the 21st century.

In this work, using the CAM-chem model, we present a coupled chemistry-climate modelling study from 1960 to 2100 with and without the contribution from VSL^{Br} sources. We focus on natural VSL^{Br}-driven impacts on the temporal evolution of stratospheric ozone and the total ozone column (TOC) in the tropics and at mid-latitudes, distinguishing both the long-term seasonal change and the resulting hemispheric asymmetries. Additionally, we present 65 a timeline assessment of the individual contribution from anthropogenic (LL^{Cl} and LL^{Br}) and biogenic (VSL^{Br}) sources to stratospheric halogen loading throughout the 21st century while also recognizing the VSL^{Br} contribution to overall halogen-catalysed ozone loss in the lowermost stratosphere. 70 The layout of the paper is as follows: Sect. 2 describes the main characteristics and configuration of the CAM-chem model as well as the observation data sets that are used for the evaluation of the CAM-chem performance. Section 3 presents a quantitative comparison between the model and 75 observations of stratospheric chlorine and bromine loading (Sect. 3.1) and the TOC (Sect. 3.2) as well as discusses long-term seasonal impacts mediated by VSLBr on the TOC (Sect. 3.2) and the lowermost stratospheric ozone budget (Sect. 3.3). Changes in the seasonal evolution of halogen-80 driven ozone loss rates due to VSL^{Br} are shown in Sect. 3.4. The final concluding remarks are summarized in Sect. 4.

2 Methods: model description

The Community Earth System Model (CESM1) with the Community Atmospheric Model including interactive chem- 85 istry (CAM4-chem version; Lamarque et al., 2012; Tilmes et al., 2016) has been used to explore VSL^{Br}-driven stratospheric ozone loss in three latitude bands: Southern Hemisphere mid-latitudes (SH MLs; 60-35° S), tropics (25° N-25° S), and Northern Hemisphere mid-latitudes (NH MLs; 90 35-60° N). The model set-up follows the Chemistry-Climate Model Initiative (CCMI) REFC2 configuration described in detail by Tilmes et al. (2016), with the exception that in this work we consider a full halogen chemistry mechanism and oceanic emission - seasonally dependent and geograph- 95 ically distributed – of six bromocarbons (VSL^{Br} = CHBr₃, CH₂Br₂, CH₂BrCl, CHBrCl₂, CHBr₂Cl, and CH₂IBr) from the Ordóñez et al. (2012) emission inventory. This emission inventory is based on a monthly varying satellite $\frac{chl}{a}$ climatology, which allows the introduction of a complete 100 seasonal cycle to the emission strength and spatial distribution of oceanic bromocarbons. A full description of the VSL^{Br} mechanism implemented in CAM-chem - including both biogenic and anthropogenic sources, heterogeneous recycling reactions, dry and wet deposition, convective uplift, 105 and large-scale transport - has been given elsewhere (Ordóñez et al., 2012; Fernandez et al., 2014). Monthly and seasonally varying lower boundary conditions were considered for long-lived chlorine (LL^{Cl} = CH₃Cl, CH₃CCl₃, CCl₄, CFC-11, CFC-12, CFC-113, HCFC-22, CFC-114, CFC-115, HCFC-141b, and HCFC-142b) and long-lived bromine 5 (LL^{Br} = CH₃Br, H-1301, H-1211, H-1202, and H-2402) following the A1 halogenated ozone-depleting substance emis-

sions scenario from the WMO (2011) Ozone Assessment Report. The surface concentrations of CO₂, CH₄, H₂, and N₂O were specified following the moderate Representation Con-¹⁰ centration Pathway 6.0 (RCP6.0) scenario (Meinshausen et

al., 2011; Eyring et al., 2013). The CAM chem configuration used in this work extends

from the surface to approximately 40 km (3.5 hPa in the upper stratosphere), with 26 vertical levels, and includes a hor-

- ¹⁵ izontal resolution of 1.9° latitude by 2.5° longitude. The number of stratospheric levels changes depending on the latitudinal region: within the tropics, there are eight levels above the tropopause (~ 100 hPa), with a mean thickness of 1.25 km (15.5 hPa) for the lower stratospheric levels and
- $_{20}$ 5.2 km (3.8 hPa) between the two highest levels, while for the mid-latitudes, the tropopause is located at approximately ~ 200 hPa, and the stratosphere contains up to 12 levels. To ensure a consistent dynamical description of the stratosphere within the relatively low model top of CAM-chem_k(i.e. grav-
- ²⁵ ity wave dragging and the Brewer–Dobson circulation), the integrated momentum that would have been deposited above the model top is specified by an upper boundary condition (Lamarque et al., 2008). A similar procedure is applied to the altitude-dependent photolysis rate computations, which
- ³⁰ include an upper boundary condition that considers the ozone column fraction prevailing above the model top. The CAM4chem version used in this work includes a non-orographic gravity wave scheme based on the inertia–gravity wave parameterization and an observation-based implementation of
- ³⁵ stratospheric aerosol and surface area density. The quasibiennial oscillation is imposed in the model by relaxing equatorial zonal winds between 90 and 3 hPa to the observed interannual variability (see Tilmes et al., 2016, for more details).
- Equivalent CAM-chem configurations, such as the one implemented in this work, have been used for the CCMVal-2 (Chemistry–Climate Model Validation 2) and CMIP5 (Coupled Model Intercomparison Project, Phase 5) activities in order to represent trends in the ozone evolution and to estimate
- ⁴⁵ the return dates, which lie in the middle of the multi-model range (Eyring et al., 2010, 2013). Moreover, our model configuration uses a fully coupled earth system model approach, i.e. the ocean and sea ice are explicitly computed (Neale et al., 2013). This implies that instead of isolating the chem-⁵⁰ ical impact mediated by VSL^{Br} sources on the climatological ozone budget (as would be the case using a specified dynamic approach), the chemical interaction between VSL^{Br} and ozone in the lowermost stratosphere is affected by dynamic feedbacks (i.e. temperature, radiation, etc.).

Even though the interactive ocean coupling would have allowed us to compute the evolution of VSL^{Br} ocean source emissions throughout the modelled period, we still lack a considerable understanding of the seasonal processes that dominate VSL^{Br} ocean emissions in both the present and the future, which is combined with a limited observation data set of VSL bromocarbons (WMO, 2014, 2018). Therefore, in this work, we forced the model with annually cycled VSL^{Br} fluxes, replicating the Ordoñez et al. emission inventory for all years between 1950 and 2100. This procedure aims to reduce the uncertainties associated with the unknown evolution of both biogeochemical production and oceanic emissions of VSL^{Br} into the future (Lennartz et al., 2015; Ziska et al., 2017).

Two ensembles of independent modelled experiments were performed, each with three individual simulations 70 from 1960 to 2100, considering approximately 10 years of spin-up to allow stabilization of the stratospheric circulation. The individual simulations 001, 002, and 003 started in 1950, 1951, and 1952, respectively, with initial conditions taken from the equivalent years of the CESM1 WACCM (Whole 75 Atmosphere Community Climate Model) 20th-century ensemble for CMIP5 (Marsh et al., 2013). The baseline simulation set-up of the first experiment considered only the halogen LL^{Cl} and LL^{Br} contribution from anthropogenic chlorofluorocarbons, hydrochlorofluorocarbons, halons, methyl 80 chloride, and methyl bromide, while the simulation set-up of the second experiment included the background biogenic contribution from VSLBr oceanic sources in addition to the anthropogenic sources above. The mean ensemble of each experiment (i.e. with and without the VSL^{Br} sources) was 85 calculated as the average of the three individual simulations. Differences between these two mean ensembles allow quantification of the overall impact of VSLBr sources on stratospheric ozone, and they allow us to determine their relative contributions to the halogen-mediated catalytic ozone- 90 depleting families. In addition, from this analysis it is also possible to distinguish the contributions of both LL^{Br} and VSL^{Br} to the inorganic fraction of stratospheric bromine $(Br_v^{LL+VSL} = Br_v^{LL} + Br_v^{VSL} = Br + BrO + HBr + BrONO_2 +$ $BrCl + HOBr + 2Br_2 + BrNO_2$), whereas the inorganic frac- 95 tion of chlorine ($Cl_{\nu}^{LL} = ClO + Cl + HOCl + 2Cl_2O_2 + 2Cl_2 +$ $OClO + HCl + ClONO_2 + BrCl + ClONO_2)$ arises only from the degradation of LL^{C1}, which is identical for both experiments.

For the case of vertical distributions and latitudinal variations, the zonal mean of each mean ensemble was computed from the monthly output before processing the data, while a locally weighted scatter plot smoothing (LOWESS) and filter with a 0.2 fraction was applied to all long-term time series to smooth the data. Most of the figures and values in the Results section include geographically averaged quantities for the present day (defined here as the 1990–2010 mean, nominally year 2000) and the end of the 21st century (defined here as the 2080–2100 mean, nominally year 2100) over the latitudinal bands defined above. To highlight seasonal changes, we computed the average for the months of December–January– February (DJF), March–April–May (MAM), June–July– August (JJA), and September–October–November (SON).

⁵ Ozone loss trends due to VSL^{Br} between 2000 and 2100 were calculated on the basis of the least-squares linear trends, while the trend errors were estimated as twice the statistical deviation stemming from the least-squares fit.

To provide a basic evaluation of the model performance, ¹⁰ the modelled results of the TOC and the inorganic-halogen abundances were compared with the following selected observation data sets:

- i. the Microwave Limb Sounder (MLS) observations of the annual mean volume mixing ratio of HCl+ClO (Waters et al., 2006; Livesey et al., 2018), which provides a lower limit of the Cl_y abundance in the stratosphere from 2005 to 2015;
- ii. the Bry trend reported in the latest WMO Ozone Assessment Reports (WMO, 2014, 2018), which shows changes in total Bry between 1991 and 2012 derived from stratospheric BrO observations by balloonborne (Dorf et al., 2006) and ground-based UV-visible (at Harestua, 60° N, and Lauder, 45° S) measurements (Hendrick et al., 2007, 2008);
- ²⁵ iii. the Solar Backscatter Ultraviolet (SBUV) merged total ozone column data set (MOD; Frith et al., 2014, 2017) from 1980 to 2015.

An important point to highlight about the SBUV MOD (and its uncertainties) is that it was constructed from ozone pro-³⁰ files measured by individual SBUV instruments and retrieved using the Version 8.6 algorithm (Bhartia et al., 2013; Frith et al., 2014, 2017). To maintain consistency over the entire time series, the individual instrument records were analysed with respect to each other, and absolute calibration ad-³⁵ justments were applied as needed based on comparison of radiance measurements during periods of instrument over-

lap (DeLand et al., 2012; Weber et al., 2018 **TSG**). During the overlap periods, the records of the involved SBUV instruments were combined using a simple average. Frith et

- ⁴⁰ al. (2014) have estimated the potential errors in the SBUV MOD through Monte Carlo model simulations, taking into account the range of offsets and drifts observed during the overlap periods between individual SBUV instruments and between SBUV and Dobson ground-based measurements.
- ⁴⁵ The authors also assessed the extent to which these potential errors might affect the long-term variability of ozone in a multiple-regression model. See Frith et al. (2014) for a detailed description of the data used and their associated uncertainties in the SBUV MOD.

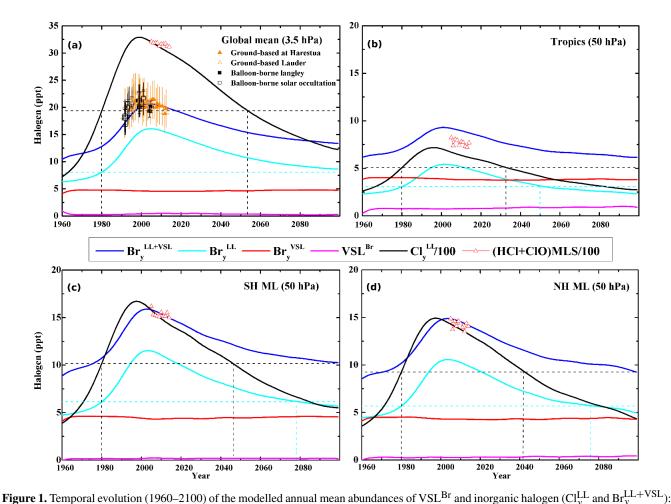
3 Results and discussion

3.1 Contribution of LL^{Cl}, LL^{Br}, and VSL^{Br} to stratospheric inorganic-halogen loading

The evolution (1960-2100) of the modelled annual mean abundances of inorganic chlorine (Cl_v) and bromine (Br_v) as well as the VSL^{Br} sources in the global upper stratosphere 55 (3.5 hPa) and lower stratosphere (50 hPa) at mid-latitudes and in the tropics are shown in Fig. 1. In addition, Fig. 1 includes global Br_x trends reported in recent WMO reports (WMO, 2014, 2018) and the MLS observations (HCl + ClO). The values of bias, normalized mean bias (NMB), and nor- 60 malized mean error (NME) used to evaluate the agreement between the modelled Cl_{ν} and observations for each of the regions under study are shown in Table S1 in the Supplement. Clearly, the temporal evolution of Cl_{ν}^{LL} and Br_{ν}^{LL} shows a pronounced peak at the end of the 20th century and beginning 65 of the 21st century, respectively, after which the abundance of both halogens declines. Within the A1 halogen emission scenario considered in this work, the Cl_{ν}^{LL} abundance in our model returns to its past 1980 levels just before ~ 2060 in the global upper stratosphere, whereas the prevailing Br_{y}^{LL} 70 abundance for the year 1980 is not recovered even by the end of the 21st century (see Fig. 1a). In comparison, the evolution of Br_{ν}^{VSL} abundances remains constant over time, resulting in an additional and time-independent fixed contribution of \sim 5 ppt total bromine injection, which leads to maxi- $_{75}$ mum global upper-stratosphere Br_{y}^{LL+VSL} abundances of up to ~ 20.5 ppt at the beginning of the 21st century. This is in agreement with previous studies performed only for presentday conditions (Fernandez et al., 2014). Furthermore, note that the modelled global Br_{y}^{LL+VSL} abundances are in good ⁸⁰ agreement with the observations within reported errors, highlighting the importance of considering the additional contribution from VSL^{Br} to determine the overall evolution of stratospheric ozone during the 21st century.

In the lower stratosphere, at 50 hPa, Br^{LL}_v abundances return to 1980 levels before \sim 2080 and \sim 2050 for the midlatitudes and tropics, respectively, while the CI_{ν}^{LL} return date remains unaltered (see Fig. 1b-d). Furthermore, the contribution from VSL^{Br} to mid-latitude total inorganic bromine also reaches \sim 5 ppt throughout the modelled period, while $_{90}$ in the tropics there prevails a small carbon-bonded organic fraction that has not been converted to the reactive inorganic form. Falk et al. (2017) showed an increase in VSL^{Br} of about 0.5 ppt in the tropical lowermost stratosphere by the end of the century compared to present day under the 95 RCP6.0 scenario and assuming constant VSL^{Br} fluxes, following scenario five of Warwick et al. (2006). This increase in the stratospheric VSLBr attributed to enhanced vertical transport in the tropics is counteracted by a decrease in the Br_{ν}^{VSL} injected into the stratosphere so that the total strato- 100 spheric $\frac{VSL^{Br}}{VSL}$ remains unchanged between the two periods. In this work, the stratospheric injection of VSL^{Br} and Br_{v}^{VSL}

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(a) global upper stratosphere (3.5 hPa) and lower stratosphere (50 hPa) (b) in the tropies and at (c) SH MLs and (d) NH MLs. The Br_y^{LL+VSL} abundance was split into long-lived (Br_y^{LL}) and very short lived (Br_y^{VSL}) contributions. The black squares (filled and open) and orange triangles (filled and open) of (a) show the total Br_y evolution (1991–2012) reported in the latest WMO Ozone Assessment Reports (2014, 2018). The red triangles show the mean annual mixing ratios (HCl + ClO) from the Microwave Limb Sounder (MLS) data (Waters et al., 2006; Livesey et al., 2018) from 2005 to 2015. The dashed horizontal lines indicate the modelled Cl_y^{LL} and Br_y^{LL} abundances for 1980. Note that both Cl_y^{LL} values and (HCl + ClO) MLS data were divided by 100.

between 2000 and 2100 remains unchanged. Based on the limited observational data, it is uncertain whether any trend in the stratospheric injection of organic and inorganic VSL^{Br} would be expected, and thus no conclusive statement can be ⁵ made CSL. Nevertheless, our present-day stratospheric injection of VSL^{Br} (~ 2 ppt) and Br_y^{VSL} (~ 3 ppt) is in perfect accordance with that reported in the latest WMO Ozone Assessment Report CCO(2018). Further studies are needed to evaluate the uncertain evolution of the stratospheric injection of organic and inorganic species from VSL^{Br} sources throughout the 21st century.

The modelled inorganic-chlorine abundance shows a good agreement with the (HCl + ClO) MLS observations for the 2005–2015 period mainly in the upper stratosphere, where ¹⁵ most chlorine has already been photochemically converted to Cl_v^{LL} . For example, the normal mean error of the compar-

ison between modelled Cl_{ν}^{LL} abundances and MLS observations is about 3.5% and 19% at 50 hPa for the mid-latitudes and tropics, respectively, while the NME is about 2% for the global mean at 3.5 hPa (see Table S1). This relatively 20 good agreement between the model and MLS observation occurs even without consideration of the recent contribution of anthropogenic VSL^{Cl} sources (i.e. Hossaini et al., 2015b, 2017). Moreover, the inter-annual variation in MLS observations at 50 hPa occurs mainly due to heterogeneous chlorine 25 reactivation, which can be influenced by the polar vortex dynamics at mid-latitudes (Dhomse et al., 2018). Note that although we processed output at 120 hPa in the lowermost midlatitude stratosphere in Sect. 3.4, we did not compare Cl_{y}^{LL} at this altitude level. This is because below 50 hPa, the frac-30 tional conversion of organic chlorine to Cl_{ν}^{LL} is very small, and thus the inferred HCl + ClO MLS data have a very large

uncertainty that prevents a reliable model observation intercomparison (Santee et al., 2011).

The stratospheric Br^{VSL} injection remains constant during the whole modelled period, representing $\sim 25\,\%$ of $_{5}$ Br_v^{LL+VSL} in the global upper stratosphere for the year 2000 and increasing up to ~ 40 % by the end of the 21st century. These results are in agreement with those reported by Fernandez et al. (2017). Due to the much shorter lifetimes of VSL^{Br} with respect to LL^{Br}, the contribution percentage is ¹⁰ larger in the lower stratosphere. For example, the Br_{ν}^{VSL} relative contribution to Br_{y}^{LL+VSL} for the year 2000 at 50 hPa represents ~ 30 % and ~ 45 % for the mid-latitudes and tropics, respectively. Furthermore, at mid-latitudes, Br_{ν}^{VSL} represents up to 45 % of Br_v^{LL+VSL} both at the end of the 21st century ¹⁵ and in the years prior to 1980, while in the tropics Br_{y}^{VSL} represents up to 65 % for these time periods. Consequently, these changes in the relative contribution of VSL^{Br} to total inorganic bromine with respect to time, region, and altitude in the presence of different inorganic-chlorine abundances

²⁰ lead to relatively different ozone loss within those regions and altitude regimes, as described in detail below.

Figure 2 shows the annual zonal mean distribution of inorganic-chlorine and CDD bromine abundances as well as the corresponding ClO_x/Cl_y and BrO_x/Br_y percentage ra-²⁵ tios (black contour lines) for the present day. Note that the ClO_x/Cl_y and BrO_x/Br_y ratios reflect the changes in

the contribution of the active-chlorine $(ClO_x = ClO + Cl + HOCl + 2Cl_2O_2 + OClO + 2Cl_2)$ and Expression (BrO_x = BrO + Br) fractions relative to their total inorganic abun-

- ³⁰ dances. Equivalent results are presented for the end of the 21st century in the Supplement (see Fig. S1). In the low-ermost stratosphere, the available inorganic fraction of both chlorine and bromine rapidly increases with altitude as halogen atoms are released by the photolysis and oxidation from ³⁵ all LL^{C1}, LL^{Br}, and VSL^{Br} species. In contrast, within the
- upper stratosphere no major changes in inorganic-halogen abundances are observed since the conversion from their organic sources is nearly complete. Although our model shows a symmetric hemispheric distribution in stratospheric Cl_y 40 and Br_y abundances, there is a marked difference in the

 ClO_x/Cl_y ratio. In fact, the experiments capture a local increase in the ClO_x/Cl_y ratio at 15° N and ~ 100 hPa as a result of a large increase in Cl_y species' heterogeneous reactivation due to enhanced vertical transport during the Indian 45 summer monsoon season (Solomon et al., 2016b). Along the

same lines, the higher ClO_x/Cl_y values modelled in the lowermost stratosphere at <u>SH MLs</u> compared to <u>NH MLs</u> are mainly due to the enhancement in the heterogeneous chlorine reactivation processes in ice crystals (i.e. HCl and ClONO₂

⁵⁰ reactivation; Santee et al., 2011; Solomon, 1999 **157**; and references therein), influenced mainly by the proximity of the Antarctic polar vortex edge and the lower temperatures prevailing at high and mid-latitudes of the Southern Hemisphere (see Fig. S2). Additionally, the favourable conditions for

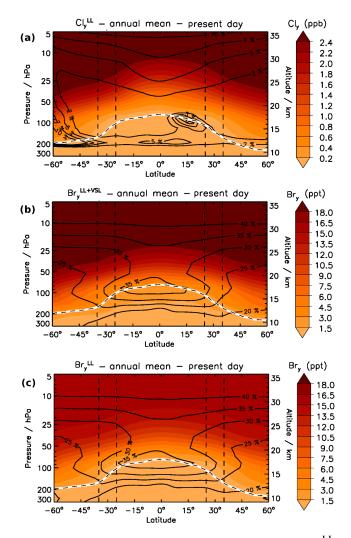


Figure 2. Annual zonal mean distribution of (a) Cl_y^{LL} , (b) $\text{Br}_y^{\text{LL}+\text{VSL}}$, and (c) Br_y^{LL} during the present-day period. The colour scale represents volume mixing ratios (ppb or ppt), while black contour lines show the contribution percentage of ClO_x to Cl_y and BrO_x to Br_y . The lower solid white line indicates the location of the tropopause (chemical definition of 150 ppb ozone level from the experiment without VSL^{Br} sources).

isentropic exchange of young subtropical air with extremely ⁵⁵ old winter polar vortex air masses at <u>SH MLs</u> (Spang et al., 2015; Rolf et al., 2015) could drive additional heterogeneous chlorine reactivation and thus enhance hemispheric asymmetry. Moreover, the increase in the ClO_x abundance and ClO_x/Cl_y ratio is highest during winter and spring for both ⁶⁰ <u>SH MLs and NH MLs</u>, which is consistent with the seasonal changes of the lowermost stratospheric temperatures that affect chlorine reactivation in ice crystals (see Fig. S3). In contrast, a symmetric hemispheric distribution of the BrO_x/Br_y ratio is modelled in the lowermost mid-latitude stratosphere and is associated with heterogeneous bromine reactivation in sulfate aerosols instead of ice crystals (i.e. HBr and BrONO₂

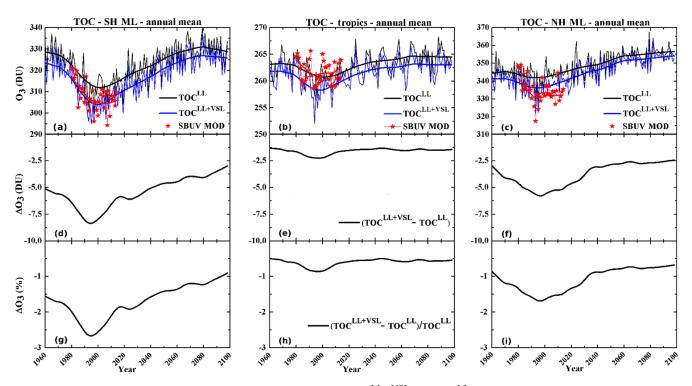


Figure 3. Temporal evolution of the annual mean total ozone column (TOC^{LL+VSL} and TOC^{LL}) (**a**) at SH-MLs (**b**) in the tropics, and (**c**) at NH-MLs as well as the corresponding absolute (DU; **d**–**f**) and relative (%; **g**–**i**) TOC difference (Δ TOC). TOC values of the mean ensemble (thin lines) and the smoothed time series via LOWESS filtering (0.2 fractions; thick lines) are shown in blue for TOC^{LL+VSL} and black for TOC^{LL}. The red lines and symbols show the observations from the Solar Backscatter Ultraviolet (SBUV) merged total ozone column data set (Frith et al., 2014, 2017) within the same spatial and temporal mask as the model output.

reactivation; Solomon, 1999, \blacksquare and references therein), which show neither any hemispheric asymmetry nor seasonal changes in the reactivation processes (see Fig. S4). Thus, even when the inclusion of VSL^{Br} sources increases the total stratospheric Br_y abundance and therefore its ac-

tive fraction (BrO_x) , the BrO_x/Br_y ratio remains unchanged in the stratosphere during the whole modelled period (see Figs. 2b, c and S1b, c), highlighting that this ratio is nearly independent of the total inorganic bromine abundance.

¹⁰ 3.2 Impact of VSL^{Br} on the seasonal evolution of the total ozone column (TOC)

The temporal evolution (1960–2100) of the modelled annual mean TOC at mid-latitudes and in the tropies, along with SBUV MOD TOC observations, is illustrated in Fig. 3.

- ¹⁵ Equivalent results for the temporal evolution of the individual simulations of each experiment are shown in Fig. S5. Table 1 shows the annual and seasonal mean values of the absolute and relative TOC differences (Δ TOC) between the experiments for the present day and the end of the 21st century as
- ²⁰ well as the Δ TOC trends (% per decade) over the century. In addition, Table S2 shows the values of bias, normalized mean bias (NMB), and normalized mean error (NME) to evaluate the agreement between the modelled TOC and observations

for each of the regions under study. Based on the comparison of the TOC results over the different latitudinal bands, the ²⁵ following features can be described:

- i. The constant emission of biogenic VSL^{Br} sources introduces a continuous reduction in the TOC that exceeds the mean ensemble variability between the experiments (see Fig. S5). Moreover, the inclusion of VSL^{Br} sources ³⁰ improves the overall agreement between the model and observations, with a reduction of 1 % and 0.9 % in NME for SH MLs and NH MLs respectively (see Table S2). However, the inclusion of VSL^{Br} sources leads to a slight increase of 0.2 % in NME for the tropics. Note that the statistical analysis presented in Table S2 does not take into account the uncertainties that may arise from the merging of the individual SBUV instruments in the SBUV MOD (see Sect. 2).
- ii. At mid-latitudes, the additional ozone loss due to VSL^{Br} 40 peaks during the present-day period, coinciding with the temporal location of the minimum TOC for both experiments and reaching a ΔTOC^{2000} of approximately -8 DU ($\sim -2.5 \%$) and -5.5 DU ($\sim -1.6 \%$) for SH MLs and NH MLs, respectively (see Table 1). Moreover, VSL^{Br} impacts on the TOC by the end of the 21st century are $\sim 50 \%$ lower than the values found

$ \begin{tabular}{ c c c c c c c } & DU (\%) & (\% \mbox{ per decade}) \\ \hline \begin{tabular}{ c c c c c c c } & DU (\%) & (\% \mbox{ per decade}) \\ & (\% \mbox{ per decade}) & (-1.6 \pm 0.6) & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & (-0.8 \pm 0.2) & (-1.2 \pm 0.3) & (-1.4 \pm 0.1) & (-0.2 \pm 0.3) & (-1.3 \pm 0.2) & (-0.7 \pm 0.1) & (-0.2 \pm 0.3) & (-1.4 \pm 0.1) & (-0.2 \pm 0.3) & (-1.3 \pm 0.2) & (-0.6 \pm 0.1) & (-0.2 \pm 0.3) & (-1.4 \pm 0.3) & (-1.4 \pm 0.3) & (-1.4 \pm 0.3) & ($	Season	Region	Δ τ ος ^{2000,a}	ΔTOC ^{2100,a}	ΔTOC trends ^b
$ \begin{array}{cccccc} (-1.6\pm0.6) & (-0.8\pm0.2) \\ -2.1\pm0.3 & -1.5\pm0.1 \\ (-0.8\pm0.2) & (-0.6\pm0.1) \\ SH MLs & -8.0\pm0.8 & -3.9\pm0.4 \\ (-2.5\pm0.3) & (-1.2\pm0.3) \end{array} & 0.15\pm0.04 \\ (-2.5\pm0.3) & (-1.2\pm0.3) \end{array} \\ \begin{array}{c} \text{DJF} & \begin{array}{c} \text{NH MLs} & -6.2\pm0.1 & -3.2\pm0.3 \\ (-1.7\pm0.6) & (-0.8\pm0.2) \\ \text{Trop} & -2.3\pm0.3 & -1.8\pm0.1 \\ (-0.9\pm0.1) & (-0.7\pm0.1) \\ SH MLs & -6.9\pm0.6 & -4.5\pm0.2 \\ (-2.4\pm0.4) & (-1.5\pm0.1) \end{array} & 0.02\pm0.01 \\ (-2.0\pm0.6) & (-0.9\pm0.2) \\ (-2.4\pm0.4) & (-1.5\pm0.1) \end{array} \\ \begin{array}{c} \text{MAM} & \begin{array}{c} \text{NH MLs} & -7.7\pm0.9 & -3.7\pm0.3 \\ (-2.4\pm0.4) & (-1.5\pm0.1) \end{array} & 0.14\pm0.04 \\ (-2.0\pm0.6) & (-0.9\pm0.2) \\ (-0.8\pm0.2) & (-0.6\pm0.1) \\ \text{SH MLs} & -6.2\pm0.5 & -4.2\pm0.2 \\ (-0.6\pm0.1) \\ \text{SH MLs} & -6.2\pm0.5 & -4.2\pm0.2 \\ (-1.4\pm0.1) \end{array} & 0.02\pm0.01 \\ (-0.8\pm0.2) & (-0.7\pm0.1) \\ \text{SH MLs} & -4.1\pm0.6 & -2.3\pm0.2 \\ (-2.2\pm0.3) & (-1.4\pm0.1) \\ \end{array} \\ \begin{array}{c} \text{JJA} & \begin{array}{c} \text{NH MLs} & -4.1\pm0.6 & -2.3\pm0.2 \\ (-2.2\pm0.3) & (-1.4\pm0.1) \\ \text{SH MLs} & -8.0\pm0.9 & -3.2\pm0.4 \\ (-2.5\pm0.2) & (-0.6\pm0.1) \\ \text{SH MLs} & -8.0\pm0.9 & -3.2\pm0.4 \\ (-2.5\pm0.2) & (-0.9\pm0.1) \\ \end{array} \\ \begin{array}{c} \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-2.5\pm0.2) & (-0.9\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ \text{SH MLs} & -3.8\pm0.4 \\ \text{SH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ \text{SH MLs} & -3.8\pm0.4 \\ \text{SH MLs} & -3.8\pm0.4 \\ \text{SH MLs} & -3.8\pm0.4$			()	DU (%)	(% per decade)
$ \begin{array}{c cccc} \mbox{Annual mean} & \begin{tabular}{lllllllllllllllllllllllllllllllllll$	Annual mean	NH MLs	-5.5 ± 0.6	-2.7 ± 0.2	0.10 ± 0.03
Annual mean (-0.8 ± 0.2) (-0.6 ± 0.1) -8.0 ± 0.8 (-3.9 ± 0.4) 0.15 ± 0.04 (-2.5 ± 0.3) BH MLs -6.2 ± 0.1 -3.2 ± 0.3 0.11 ± 0.03 (-1.7 ± 0.6) (-1.2 ± 0.3) DJFTrop -2.3 ± 0.3 -1.8 ± 0.1 (-0.9 ± 0.1) 0.02 ± 0.01 (-0.7 ± 0.1) BH MLs -6.9 ± 0.6 -4.5 ± 0.2 (-2.4 ± 0.4) 0.10 ± 0.03 (-1.5 ± 0.1) MAM -7.7 ± 0.9 -3.7 ± 0.3 (-2.4 ± 0.4) 0.14 ± 0.04 (-2.0 ± 0.6) MAM -7.7 ± 0.9 (-2.0 ± 0.6) -3.7 ± 0.3 (-2.4 ± 0.4) 0.14 ± 0.04 (-2.0 ± 0.6) MAM -7.7 ± 0.9 (-2.0 ± 0.6) -3.7 ± 0.3 (-0.6 ± 0.1) 0.14 ± 0.04 (-2.2 ± 0.3) MAM -6.2 ± 0.5 (-2.2 ± 0.3) -4.2 ± 0.2 (-1.4 ± 0.1) 0.02 ± 0.01 (-0.8 ± 0.2) JJANH MLs (-0.8 ± 0.2) (-0.6 ± 0.1) 0.09 ± 0.03 (-1.3 ± 0.2) (-0.6 ± 0.1) (-0.6 ± 0.1) JJANH MLs (-2.5 ± 0.2) (-2.5 ± 0.2) (-0.6 ± 0.1) (-0.6 ± 0.1) 0.19 ± 0.06 (-2.5 ± 0.2) SONNH MLs (-1.2 ± 0.1) (-0.7 ± 0.1) (-0.6 ± 0.1) (-0.7 ± 0.1) 0.01 ± 0.01 (-0.7 ± 0.1) SONTrop (-1.9 ± 0.2) -1.5 ± 0.1 (-0.7 ± 0.1) 0.01 ± 0.01 (-0.7 ± 0.1) SH MLs -10.0 ± 1.2 (-0.7 ± 0.1) 0.20 ± 0.06		-	(-1.6 ± 0.6)	(-0.8 ± 0.2)	
$ \begin{array}{c} \text{MAM} \\ \text{Max} \\ \text{Max} \\ \text{Max} \\ \text{Max} \\ \begin{array}{c} -0.8 \pm 0.2) \\ -8.0 \pm 0.8 \\ (-2.5 \pm 0.3) \\ (-1.2 \pm 0.4) \\ (-1.5 \pm 0.1) \\ \end{array} \\ \begin{array}{c} \text{MB} \text{MLs} \\ \text{MB} \\ \text{MB} \\ \begin{array}{c} -6.2 \pm 0.1 \\ (-0.9 \pm 0.1) \\ (-0.7 \pm 0.1) \\ (-0.9 \pm 0.1) \\ (-1.5 \pm 0.1) \\ \end{array} \\ \begin{array}{c} \text{MAM} \\ \text{MAM} \\ \begin{array}{c} \text{MB} \text{MLs} \\ -7.7 \pm 0.9 \\ (-2.4 \pm 0.4) \\ (-1.5 \pm 0.1) \\ (-0.9 \pm 0.2) \\ (-0.6 \pm 0.1) \\ (-0.8 \pm 0.2) \\ (-0.7 \pm 0.1) \\ \end{array} \\ \begin{array}{c} \text{MAM} \\ \text{MB} \\ \begin{array}{c} -4.1 \pm 0.6 \\ (-2.2 \pm 0.3) \\ (-1.4 \pm 0.1) \\ (-0.8 \pm 0.2) \\ (-0.6 \pm 0.1) \\ (-0.8 \pm 0.2) \\ (-0.6 \pm 0.1) \\ (-0.8 \pm 0.2) \\ (-0.9 \pm 0.1) \\ \end{array} \\ \begin{array}{c} \text{MB} \text{MLs} \\ \begin{array}{c} -8.0 \pm 0.9 \\ (-2.5 \pm 0.2) \\ (-0.9 \pm 0.1) \\ (-0.9 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.7 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ \end{array} \\ \begin{array}{c} \text{MB} \text{MLs} \\ \begin{array}{c} -3.8 \pm 0.4 \\ (-1.6 \pm 0.1 \\ (-0.7 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.01 \\ (-0.7 \pm 0.1) \\ (-0.6 \pm 0.1) \\ (-0.2 \pm 0.01 \\ (-0.7 \pm 0.1) \\ (-0.6 \pm 0.1) \\ (-0.2 \pm 0.06 \\ (-2.5 \pm 0.2) \\ (-0.9 \pm 0.1) \\ \end{array} \\ \begin{array}{c} \text{MB} \text{MLs} \\ \begin{array}{c} -3.8 \pm 0.4 \\ (-1.5 \pm 0.1 \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.1 \pm 0.03 \\ (-1.2 \pm 0.1) \\ (-0.5 \pm 0.$		Trop	-2.1 ± 0.3	-1.5 ± 0.1	0.03 ± 0.01
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$			(-0.8 ± 0.2)	(-0.6 ± 0.1)	
$ \begin{tabular}{ c c c c c c c } \hline NH MLs & -6.2 \pm 0.1 & -3.2 \pm 0.3 & 0.11 \pm 0.03 \\ & (-1.7 \pm 0.6) & (-0.8 \pm 0.2) & 0.02 \pm 0.01 \\ & (-0.9 \pm 0.1) & (-0.7 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.9 \pm 0.1) & (-0.7 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.9 \pm 0.4) & (-1.5 \pm 0.1) & 0.10 \pm 0.03 \\ & (-2.4 \pm 0.4) & (-1.5 \pm 0.1) & 0.10 \pm 0.03 \\ & (-2.4 \pm 0.4) & (-1.5 \pm 0.1) & 0.14 \pm 0.04 \\ & (-2.0 \pm 0.6) & (-0.9 \pm 0.2) & 0.14 \pm 0.04 \\ & (-2.0 \pm 0.6) & (-0.9 \pm 0.2) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.01 \pm 0.03 \\ & (-1.3 \pm 0.2) & (-0.7 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.02 \pm 0.01 \\ & (-2.5 \pm 0.2) & (-0.9 \pm 0.1) & 0.10 \pm 0.03 \\ & (-1.2 \pm 0.1) & (-0.5 \pm 0.1) & 0.10 \pm 0.03 \\ & (-1.2 \pm 0.1) & (-0.5 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.5 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.5 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.7 \pm 0.1) & 0.02 \pm 0.06 \\ & (-0.7 \pm 0.1) & (-0.7 \pm 0.$		SH MLs	-8.0 ± 0.8	-3.9 ± 0.4	0.15 ± 0.04
$ \begin{split} \text{DJF} & \begin{bmatrix} (-1.7\pm0.6) & (-0.8\pm0.2) \\ -2.3\pm0.3 & -1.8\pm0.1 \\ (-0.9\pm0.1) & (-0.7\pm0.1) \\ \text{SH MLs} & -6.9\pm0.6 & -4.5\pm0.2 \\ (-2.4\pm0.4) & (-1.5\pm0.1) \\ \end{bmatrix} \\ \text{MAM} & \begin{bmatrix} \text{NH MLs} & -7.7\pm0.9 & -3.7\pm0.3 \\ (-2.4\pm0.4) & (-1.5\pm0.1) \\ \text{NH MLs} & -7.7\pm0.9 & -3.7\pm0.3 \\ (-2.0\pm0.6) & (-0.9\pm0.2) \\ \end{bmatrix} \\ \text{MAM} & \begin{bmatrix} \text{Trop} & -2.1\pm0.2 & -1.5\pm0.1 \\ (-0.8\pm0.2) & (-0.6\pm0.1) \\ \text{SH MLs} & -6.2\pm0.5 & -4.2\pm0.2 \\ (-2.2\pm0.3) & (-1.4\pm0.1) \\ \end{bmatrix} \\ \end{bmatrix} \\ \text{MHMLs} & \begin{bmatrix} -4.1\pm0.6 & -2.3\pm0.2 \\ (-2.2\pm0.3) & (-1.4\pm0.1) \\ \end{bmatrix} \\ \text{NH MLs} & \begin{bmatrix} -4.1\pm0.6 & -2.3\pm0.2 \\ (-2.2\pm0.3) & (-1.4\pm0.1) \\ \end{bmatrix} \\ \text{NH MLs} & \begin{bmatrix} -4.1\pm0.6 & -2.3\pm0.2 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ (-0.8\pm0.2) & (-0.6\pm0.1) \\ (-0.8\pm0.2) & (-0.6\pm0.1) \\ \text{SH MLs} & -8.0\pm0.9 & -3.2\pm0.4 \\ (-2.5\pm0.2) & (-0.9\pm0.1) \\ \end{bmatrix} \\ \text{SON} & \begin{bmatrix} \text{NH MLs} & -3.8\pm0.4 & -1.6\pm0.1 \\ (-0.7\pm0.1) & (-0.5\pm0.1) \\ \text{Trop} & -1.9\pm0.2 & -1.5\pm0.1 \\ (-0.7\pm0.1) & (-0.6\pm0.1) \\ \text{SH MLs} & -10.0\pm1.2 & -5.7\pm0.4 \\ \end{bmatrix} $			(-2.5 ± 0.3)	(-1.2 ± 0.3)	
$ \begin{split} \text{DJF} & \begin{array}{c} \text{Trop} & -2.3 \pm 0.3 & -1.8 \pm 0.1 \\ (-0.9 \pm 0.1) & (-0.7 \pm 0.1) \\ \text{SH MLs} & -6.9 \pm 0.6 & -4.5 \pm 0.2 \\ (-2.4 \pm 0.4) & (-1.5 \pm 0.1) \end{array} & 0.10 \pm 0.03 \\ (-2.4 \pm 0.4) & (-1.5 \pm 0.1) \end{array} \\ & \begin{array}{c} \text{MAM} & \begin{array}{c} \text{NH MLs} & -7.7 \pm 0.9 & -3.7 \pm 0.3 \\ (-2.0 \pm 0.6) & (-0.9 \pm 0.2) \\ \text{Trop} & -2.1 \pm 0.2 & -1.5 \pm 0.1 \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ \text{SH MLs} & -6.2 \pm 0.5 & -4.2 \pm 0.2 \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \end{array} & 0.02 \pm 0.01 \\ & \begin{array}{c} (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \end{array} & 0.02 \pm 0.03 \\ (-1.3 \pm 0.2) & (-0.7 \pm 0.1) \\ \text{MH MLs} & -4.1 \pm 0.6 & -2.3 \pm 0.2 \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \end{array} & 0.02 \pm 0.01 \\ & \begin{array}{c} (-0.8 \pm 0.2) & (-0.7 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ \text{SH MLs} & -8.0 \pm 0.9 & -3.2 \pm 0.4 \\ (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{array} & 0.10 \pm 0.03 \\ & \begin{array}{c} (-1.2 \pm 0.1) & (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) & (-0.5 \pm 0.1) \\ \text{SON} & \begin{array}{c} \text{NH MLs} & -3.8 \pm 0.4 & -1.6 \pm 0.1 \\ (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ \text{SH MLs} & -10.0 \pm 1.2 & -5.7 \pm 0.4 \end{array} & 0.20 \pm 0.06 \\ \end{array} $	DJF	NH MLs	-6.2 ± 0.1	-3.2 ± 0.3	0.11 ± 0.03
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			· · · · · · · · · · · · · · · · · · ·	()	
$MAM = \begin{bmatrix} (-0.9 \pm 0.1) & (-0.7 \pm 0.1) \\ -6.9 \pm 0.6 & -4.5 \pm 0.2 \\ (-2.4 \pm 0.4) & (-1.5 \pm 0.1) \end{bmatrix} $ $MHMLs = \begin{bmatrix} -7.7 \pm 0.9 & -3.7 \pm 0.3 \\ (-2.0 \pm 0.6) & (-0.9 \pm 0.2) \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ SH MLs = \begin{bmatrix} -6.2 \pm 0.5 & -4.2 \pm 0.2 \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \end{bmatrix} $ $MHMLs = \begin{bmatrix} -4.1 \pm 0.6 & -2.3 \pm 0.2 \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.7 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.7 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ SH MLs = \begin{bmatrix} -8.0 \pm 0.9 & -3.2 \pm 0.4 \\ (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.7 \pm 0.1) \\ (-0.7 \pm 0.1) \\ (-0.6 \pm 0.1) \\ SH MLs = \begin{bmatrix} -3.8 \pm 0.4 & -1.6 \pm 0.1 \\ (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) \\ (-0.7 \pm 0.1) \\ (-0.6 \pm 0.1) \\ (-0.7 \pm 0.1) \\ (-0.6 \pm 0.1) \\ SH MLs \\ -10.0 \pm 1.2 \\ -5.7 \pm 0.4 \\ 0.20 \pm 0.06 \end{bmatrix}$		Trop			0.02 ± 0.01
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$					
$ MAM = \begin{bmatrix} NH MLs & -7.7 \pm 0.9 & -3.7 \pm 0.3 \\ (-2.0 \pm 0.6) & (-0.9 \pm 0.2) \\ Trop & -2.1 \pm 0.2 & -1.5 \pm 0.1 \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ SH MLs & -6.2 \pm 0.5 & -4.2 \pm 0.2 \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \\ \end{bmatrix} $		SH MLs			0.10 ± 0.03
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			(-2.4 ± 0.4)	(-1.5 ± 0.1)	
$\begin{array}{cccccccc} {\rm MAM} & {\rm Trop} & -2.1\pm0.2 & -1.5\pm0.1 & 0.02\pm0.01 \\ & (-0.8\pm0.2) & (-0.6\pm0.1) \\ {\rm SH MLs} & -6.2\pm0.5 & -4.2\pm0.2 & 0.11\pm0.03 \\ & (-2.2\pm0.3) & (-1.4\pm0.1) & \\ \end{array} \\ \\ {\rm JJA} & {\rm NH MLs} & -4.1\pm0.6 & -2.3\pm0.2 & 0.09\pm0.03 \\ & (-1.3\pm0.2) & (-0.7\pm0.1) & \\ {\rm Trop} & -2.0\pm0.3 & -1.6\pm0.1 & 0.02\pm0.01 \\ & (-0.8\pm0.2) & (-0.6\pm0.1) & \\ {\rm SH MLs} & -8.0\pm0.9 & -3.2\pm0.4 & 0.19\pm0.06 \\ & (-2.5\pm0.2) & (-0.9\pm0.1) & \\ \end{array} \\ \\ {\rm SON} & {\rm NH MLs} & -3.8\pm0.4 & -1.6\pm0.1 & 0.10\pm0.03 \\ & (-1.2\pm0.1) & (-0.5\pm0.1) & \\ {\rm Trop} & -1.9\pm0.2 & -1.5\pm0.1 & 0.01\pm0.01 \\ & (-0.7\pm0.1) & (-0.6\pm0.1) & \\ {\rm SH MLs} & -10.0\pm1.2 & -5.7\pm0.4 & 0.20\pm0.06 \end{array}$	MAM	NH MLs		-3.7 ± 0.3	0.14 ± 0.04
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					
$SON = \begin{bmatrix} (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ SH MLs & -6.2 \pm 0.5 & -4.2 \pm 0.2 \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \end{bmatrix} = \begin{bmatrix} (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ (-2.2 \pm 0.3) & (-1.4 \pm 0.1) \end{bmatrix}$ $IJA = \begin{bmatrix} NH MLs & -4.1 \pm 0.6 & -2.3 \pm 0.2 \\ (-1.3 \pm 0.2) & (-0.7 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.7 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ SH MLs & -8.0 \pm 0.9 & -3.2 \pm 0.4 \\ (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{bmatrix}$ $ITrop = \begin{bmatrix} -3.8 \pm 0.4 & -1.6 \pm 0.1 \\ (-0.5 \pm 0.1) & (-0.5 \pm 0.1) \\ (-0.7 \pm 0.1) & (-0.5 \pm 0.1) \\ (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ SH MLs & -10.0 \pm 1.2 & -5.7 \pm 0.4 \end{bmatrix}$		Trop			0.02 ± 0.01
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			(()	
$JJA \qquad \begin{array}{cccc} NH MLs & -4.1 \pm 0.6 & -2.3 \pm 0.2 & 0.09 \pm 0.03 \\ & (-1.3 \pm 0.2) & (-0.7 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.02 \pm 0.01 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) & 0.02 \pm 0.01 \\ & (-2.5 \pm 0.2) & (-0.9 \pm 0.1) & 0.19 \pm 0.06 \\ & (-2.5 \pm 0.2) & (-0.9 \pm 0.1) & 0.10 \pm 0.03 \\ & (-1.2 \pm 0.1) & (-0.5 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) & 0.20 \pm 0.06 \end{array}$		SH MLs			0.11 ± 0.03
$JJA \qquad \begin{array}{ccccc} (-1.3 \pm 0.2) & (-0.7 \pm 0.1) \\ Trop & -2.0 \pm 0.3 & -1.6 \pm 0.1 \\ & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ SH MLs & -8.0 \pm 0.9 & -3.2 \pm 0.4 \\ & (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{array} \begin{array}{c} 0.19 \pm 0.06 \\ (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{array}$			(-2.2 ± 0.3)	(-1.4 ± 0.1)	
$JJA \qquad \begin{array}{cccc} Trop & -2.0 \pm 0.3 & -1.6 \pm 0.1 & 0.02 \pm 0.01 \\ (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ SH MLs & -8.0 \pm 0.9 & -3.2 \pm 0.4 \\ (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{array} \qquad \begin{array}{c} 0.19 \pm 0.06 \\ 0.19 \pm 0.06 \\ (-2.5 \pm 0.2) \\ (-0.9 \pm 0.1) \end{array}$ $SON \qquad \begin{array}{c} NH MLs & -3.8 \pm 0.4 & -1.6 \pm 0.1 \\ (-1.2 \pm 0.1) & (-0.5 \pm 0.1) \\ Trop & -1.9 \pm 0.2 & -1.5 \pm 0.1 \\ (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ SH MLs & -10.0 \pm 1.2 & -5.7 \pm 0.4 \\ \end{array} \qquad \begin{array}{c} 0.20 \pm 0.06 \\ 0.20 \pm 0.06 \end{array}$	JJA	NH MLs			0.09 ± 0.03
$\begin{array}{cccccc} \text{JJA} & & (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ & \text{SH MLs} & -8.0 \pm 0.9 & -3.2 \pm 0.4 \\ & (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{array} & 0.19 \pm 0.06 \\ & (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{array}$ $\begin{array}{c} \text{NH MLs} & -3.8 \pm 0.4 & -1.6 \pm 0.1 & 0.10 \pm 0.03 \\ & (-1.2 \pm 0.1) & (-0.5 \pm 0.1) \end{array}$ $\begin{array}{c} \text{SON} & \text{Trop} & -1.9 \pm 0.2 & -1.5 \pm 0.1 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \end{array}$ $\begin{array}{c} \text{SH MLs} & -10.0 \pm 1.2 & -5.7 \pm 0.4 & 0.20 \pm 0.06 \end{array}$			· · · ·	· ,	
$SON = \begin{bmatrix} (-0.8 \pm 0.2) & (-0.6 \pm 0.1) \\ -8.0 \pm 0.9 & -3.2 \pm 0.4 \\ (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \end{bmatrix}$ $NH MLs = \begin{bmatrix} -3.8 \pm 0.4 & -1.6 \pm 0.1 \\ (-0.5 \pm 0.1) & (-0.5 \pm 0.1) \\ (-0.5 \pm 0.1) & (-0.5 \pm 0.1) \end{bmatrix}$ $SON = \begin{bmatrix} Trop & -1.9 \pm 0.2 & -1.5 \pm 0.1 \\ (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ SH MLs & -10.0 \pm 1.2 & -5.7 \pm 0.4 \end{bmatrix} = 0.20 \pm 0.06$		Trop			0.02 ± 0.01
$SON \qquad \begin{array}{c} (-2.5 \pm 0.2) & (-0.9 \pm 0.1) \\ NH MLs & -3.8 \pm 0.4 & -1.6 \pm 0.1 & 0.10 \pm 0.03 \\ & (-1.2 \pm 0.1) & (-0.5 \pm 0.1) \\ Trop & -1.9 \pm 0.2 & -1.5 \pm 0.1 & 0.01 \pm 0.01 \\ & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ SH MLs & -10.0 \pm 1.2 & -5.7 \pm 0.4 & 0.20 \pm 0.06 \end{array}$					
SON $\begin{array}{c cccc} NH MLs & -3.8 \pm 0.4 & -1.6 \pm 0.1 & 0.10 \pm 0.03 \\ & & (-1.2 \pm 0.1) & (-0.5 \pm 0.1) \\ Trop & -1.9 \pm 0.2 & -1.5 \pm 0.1 & 0.01 \pm 0.01 \\ & & (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ SH MLs & -10.0 \pm 1.2 & -5.7 \pm 0.4 & 0.20 \pm 0.06 \end{array}$		SH MLs			0.19 ± 0.06
SON $\begin{array}{cccc} (-1.2 \pm 0.1) & (-0.5 \pm 0.1) \\ \text{Trop} & -1.9 \pm 0.2 & -1.5 \pm 0.1 \\ (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ \text{SH MLs} & -10.0 \pm 1.2 & -5.7 \pm 0.4 \\ \end{array} 0.20 \pm 0.06 \end{array}$			(-2.5 ± 0.2)	(-0.9 ± 0.1)	
SON Trop -1.9 ± 0.2 -1.5 ± 0.1 0.01 ± 0.01 (-0.7 ± 0.1) (-0.6 ± 0.1) SH MLs -10.0 ± 1.2 -5.7 ± 0.4 0.20 ± 0.06	SON	NH MLs			0.10 ± 0.03
SON (-0.7 ± 0.1) (-0.6 ± 0.1) SH MLs -10.0 ± 1.2 -5.7 ± 0.4 0.20 ± 0.06				(-0.5 ± 0.1)	
$\begin{array}{c} (-0.7 \pm 0.1) & (-0.6 \pm 0.1) \\ \text{SH MLs} & -10.0 \pm 1.2 & -5.7 \pm 0.4 & 0.20 \pm 0.06 \end{array}$		Trop			0.01 ± 0.01
(-3.0 ± 0.8) (-1.6 ± 0.8)		SH MLs			0.20 ± 0.06
			(-3.0 ± 0.8)	(-1.6 ± 0.8)	

Table 1. Annual and seasonal TOC changes (Δ TOC) mediated by VSL^{Br} at mid-latitudes and in the tropics during the present-day period and the end of the 21st century as well as the Δ TOC trends (% per decade) over the century.

^a Absolute (DU) and relative (%; in brackets) TOC changes during the present-day

period (ΔTOC^{2000}) and the end of the 21st century (ΔTOC^{2100}) were computed considering the mean ensemble of each experiment at mid-latitudes (SH MLs and NH MLs) and in the tropics (Trop). Annual and seasonal ΔTOC errors were estimated from standard mean errors. ^b Trend errors were

estimated as twice the statistical deviation stemming from the least-squares fit.

for the present day, which is in line with the projected Δ TOC trends of 0.15 % and 0.10 % per decade for SH MLs and NH MLs, respectively.

iii. Within the tropics, the inclusion of VSL^{Br} leads to small impacts on the TOC compared to impacts observed at mid-latitudes. The TOC differences due to VSL^{Br} are close to -1.5 DU (< -1%) during practically the entire modelled period, with a maximum ΔTOC²⁰⁰⁰ reaching -2.1 DU (< -1%). This is in line with a projected ΔTOC trend of 0.03% per decade. Moreover, even though VSL^{Br} slightly worsens the agreement between the modelled TOC^{LL} and observations, the minimum TOC^{LL+VSL} is shifted to ~5 years earlier compared to

TOC^{LL} (i.e. towards where the inorganic-chlorine peak is located in the year 1995), which is in agreement with ¹⁵ observations.

iv. Overall, our model results show much higher ozone depletion due to VSL^{Br} at southern and northern midlatitudes compared to the tropics.

Our results of the mid-latitude Δ TOC due to VSL^{Br} lie ²⁰ within the lower range of previous modelling studies over the 1960–2005 (Sinnhuber and Meul, 2015) and 1980–2005 (Feng et al., 2007) periods. We expect to see the lower impacts modelled in this recent work, as for example the detailed treatment of VSL^{Br} sources in the results of Sinnhuber and Meul (2015) for an additional Br_{Ψ}^{VSL} injection of

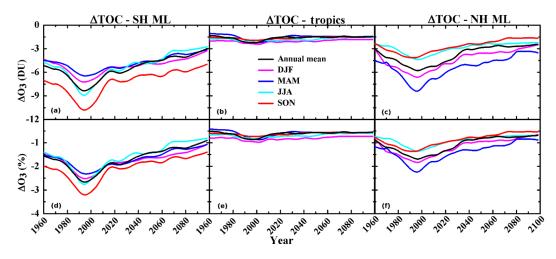


Figure 4. Temporal evolution of total ozone column difference (Δ TOC) between the experiments. Panels (**a–c**) show the annual and seasonal mean absolute Δ TOC (DU) at <u>SH–MLs</u> in the tropics, and at <u>NH–MLs</u> respectively, while panels (**d–f**) show the corresponding relative Δ TOC (%).

 $6 \text{ ppt } Br_y$. The model configurations in Feng et al. (2007) included the direct supply of tropospheric Br_y as a lower condition in addition to a VSL substance tracer. Furthermore, our results are in accord with the future projection trends in

5 Falk et al. (2017), although their analysis on annual mean ozone loss due to VSL^{Br} only focused on the late 21st century (2075–2100).

Figure 4 shows the temporal evolution (1960–2100) of the annual and seasonal Δ TOC between the experiments for SH

- ¹⁰ MLs, the tropics, and NH MLs. In addition, an analysis of Δ TOC as a function of latitude for the present day and the end of the 21st century is shown in Fig. 5. The inclusion of VSL^{Br} leads to a continuous reduction in the TOC at all latitudes, with a larger ozone destruction efficiency moving ¹⁵ from the tropics to the high latitudes. Moreover, note that when comparing the seasonal relative Δ TOC between both periods, a statistically significant difference (here defined as p < 0.05) is only observed at mid-latitudes, as shown by the horizontal lines at the bottom of Fig. 5d. This highlights that
- ²⁰ the VSL^{Br}-driven ozone depletion efficiency changes significantly over the century even though the contribution from VSL^{Br} sources to bromine stratospheric injection remains constant. Since gas-phase and heterogeneous inter-halogen reactions involving both bromine and chlorine play a funda-
- ²⁵ mental role in stratospheric ozone loss (McElroy et al., 1986; Solomon, 1999 ISE); Salawitch et al., 2005), the VSL^{Br} efficiency in ozone depletion is primarily linked to the background inorganic-halogen abundance, which shows a continuous decline over the course of the 21st century, which is
- ³⁰ contemplated in the A1 halogen emission scenario considered in this work. This is in line with the findings of Yang et al. (2014). Therefore, an additional stratospheric chlorine and bromine load from natural or anthropogenic substances not regulated by the Montreal Protocol will induce an increase in

the VSL^{Br} impacts modelled in this work on projected ozone ³⁵ loss trends throughout the current century.

The largest mid-latitude TOC differences between experiments occur during spring, when the maximum seasonal TOC levels are found, followed by the winter. The maximum springtime ΔTOC^{2000} reaches $-10 DU (\sim -3 \%)$ and $_{40}$ $-7.7 \text{ DU} (\sim -2\%)$ for SH MLs and NH MLs, respectively (see Table 1 and Fig. 4). In contrast, during summer and autumn, the maximum ΔTOC^{2000} remains below -6.8 (~ -2.4%) and -4 DU ($\sim -4.1\%$) for SH MLs and NH MLs. respectively. This is in line with the ozone loss mediated by 45 VSL^{Br} within the southern polar cap reported by Fernandez et al. (2017), who showed a maximum ozone depletion of up to -15 DU ($\sim -14 \%$ of TOC) during October at the beginning of the century. Moreover, as mentioned above, the projected seasonal VSL^{Br} impact on the TOC significantly 50 decreases towards the end of the century, with springtime Δ TOC trends of up to 0.20 % and 0.14 % per decade for SH MLs and NH MLs, respectively (see Table 1). In comparison, the projected ΔTOC trends for summer and autumn reach approximately 0.10 % per decade at mid-latitudes. Note also 55 that the changes in the magnitudes of these seasonal ΔTOC trends are mainly attributed to a marked seasonal difference of ozone loss mediated by VSL^{Br} during the period of highest background halogen abundance (i.e. present day). These seasonal differences are reduced towards the end of the 21st cen-60 tury due to the decline in long-lived inorganic-halogen abundance.

Within the tropics, no significant changes in the TOC loss due to VSL^{Br} are projected, though the model captures a slightly major TOC depletion during the 1990s. Therefore, ⁶⁵ VSL^{Br}-driven ozone depletion is less sensitive to background halogen abundances, introducing an almost constant impact

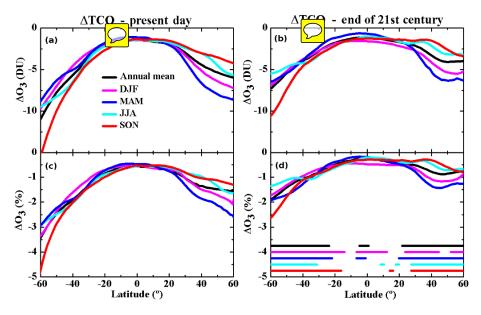


Figure 5. Latitude distributions of total ozone column difference (ΔTOC) between the experiments. Panels (**a**) and (**b**) show the annual and seasonal mean absolute ΔTOC (DU) for the present day and the end of 21st century, respectively, while panels (**c**) and (**d**) show the corresponding relative ΔTOC (%). The corresponding horizontal lines shown in panel (**d**) indicate where the relative ΔTOC between the present day and the end of the 21st century is statistically significant with a 95 % confidence interval using a two-tailed Student's *t* test.

that is in line with its fixed emissions considered throughout the whole modelled period.

3.3 VSL^{Br} impact on vertical ozone distributions

Timeline analysis of the annual mean stratospheric ozone ⁵ difference between the experiments as a function of altitude at mid latitudes and in the tropics is presented in Fig. 6. The inclusion of VSL^{Br} produces a reduction in ozone concentrations (i.e. $O_3(z)$ number densities) throughout most of the stratosphere during the whole ¹⁰ modelled period. The largest $O_3(z)$ reductions occur during the present-day period in the lowermost mid latitude stratosphere, reaching ozone differences ($\Delta O_3(z)$) of up to -8% (-0.25×10^{12} molecules cm⁻³) and -5% (-0.14×10^{12} molecules cm⁻³) for SH MLs and NH MLs, respec-¹⁵ tively. Above 50 hPa (~ 20 km), $\Delta O_3(z)$ is practically con-

stant at around -1% for all latitudes, which is linked to the lesser role played by $\frac{\text{VSL}^{\text{Br}}}{\text{Compared to catalytic ozone}}$ loss by hydrogen (HO_x = OH + HO₂) and nitrogen (NO_x = NO + NO₂) oxides as well as by long-lived reactive halogens ²⁰ (Brasseur and Solomon, 2005; Salawitch et al., 2005; Müller, 2012).

Our modelled $\Delta O_3(z)$ latitudinal distribution is in agreement with those reported by Sinnhuber and Meul et al. (2015) during the 1980–2005 period, with a larger percentage of ²⁵ ozone loss deepening around the mid-latitude tropopause. In addition, $O_3(z)$ reduction due to VSL^{Br} in the lowermost mid-latitude stratosphere is about 50 % smaller by the end of the 21st century compared to the present-day period (see Fig. 6d and e). This is in line with projected

local $\Delta O_3(z)$ trends of up to 1% and 0.5% per decade 30 at SH MLs and NH MLs, respectively (see Fig. 6f). The largest $O_3(z)$ reductions are located precisely at the same altitudes and latitudes where the ClO_x/Cl_y ratio peak is modelled within the lowermost stratosphere at SH MLs (see Fig. 2), which explains the hemispheric asymmetry in the 35 VSL^{Br} impacts. This highlights that the enhancement in local ozone depletion mediated by VSL^{Br} is largest in the context of maximum background active-chlorine abundance since it provides an additional partner for chlorine species involved in the ozone-depleting inter-halogen chemical reactions (e.g. ClO_x -BrO_{x-Loss}; see Sect. 3.4). These results are in agreement with Yang et al. (2014), who determined that VSL^{Br} has a significantly larger ozone impact in the presence of a high stratospheric chlorine background than a low chlorine background using a set of time slice sensitivity sim- 45 ulations with a variable and speculative VSL^{Br} contribution on different background stratospheric Cl_v abundances.

Figure 7 shows the seasonal zonal mean distribution of the $\Delta O_3(z)$ during the present-day period. The corresponding $\Delta O_3(z)$ trends (% per decade) over the century are shown in ⁵⁰ Fig. S6. The largest contributions to the annual mean ozone loss in the lowermost mid-latitude stratosphere correspond to spring, followed by the preceding winter and the posterior summer in each hemisphere, with springtime $\Delta O_3(z)$ reaching up to -10 % and -7 % for SH MLs and NH MLs, respectively (see Fig. 7b and d). This seasonal enhancement in VSL^{Br}-mediated local ozone depletion is influenced by the seasonal changes of the heterogeneous chlorine reactivation processes in lowermost stratospheric ice crystals since the

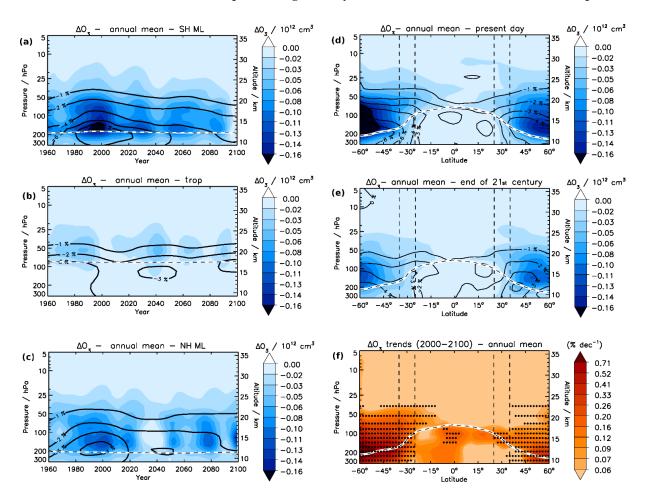


Figure 6. Temporal evolution of the annual mean ozone difference (ΔO_3) between experiments as a function of altitude (a) at SH-MLs, (b) in the tropics, and (c) at NH-MLs. The right panels show the zonal mean ΔO_3 distributions for the (d) present day and (e) the end of the 21st century as well as the (f) ΔO_3 trends (% per decade) over the century. The absolute (colour scale) and relative (contour line) ozone differences were calculated considering the ozone number densities (i.e. molecules cm⁻³) of each experiment. The masked regions in panel (f) indicate where the relative $\Delta O_3(z)$ between the present day and the end of the 21st century is statistically significant with a 95 % confidence interval using a two-tailed Student's *t* test.

main heterogeneous bromine reactivation processes occur in sulfate aerosols and therefore lack seasonal changes (see Figs. S3 and S4). Consequently, since the seasonal changes in ozone loss are maximized during the present-day period, the

 $_{5}$ largest local $\Delta O_{3}(z)$ trends are also projected for spring and winter, with springtime $\Delta O_{3}(z)$ trends reaching up to 1.6% and 0.7% per decade for SH MLs and NH MLs, respectively (see Fig. S6b and d).

Compared to the annual mean, springtime absolute $O_3(z)$ ¹⁰ reductions along mid-latitudes are deepened over the first ~ 5 km above the tropopause during the present-day period, whereas these reductions are located at higher altitudes (be-

tween 2 and 5 km above the tropopause) and over a narrower geographic area, mainly at <u>NH MLs</u>, during summer ¹⁵ and autumn. Therefore, VSL^{Br} introduces seasonal changes

not only in overall $O_3(z)$ reduction but also in vertical $O_3(z)$ distribution. Moreover, since seasonal VSL^{Br} impacts persist

until the end of the century, they could lead to significantly different seasonal perturbations of the radiative effects mediated by ozone. Future studies are needed to explore the po-²⁰ tential VSL^{Br}-mediated seasonal effects on the atmosphere's radiative balance.

Near the tropical tropopause, the $O_3(z)$ reductions due to VSL^{Br} are practically constant during the whole modelled period, with $\Delta O_3(z)$ of approximately -2% (-0.06×25 10^{12} molecules cm⁻³) at 50 hPa (see Fig. 6b). Even with an increase in the contribution of Br^{VSL}_y to Br^{LL+VSL}_y of up to 65% by the end of century – due to the implementation of the Montreal Protocol (WMO, 2014, 2018) and the fixed stratospheric Br^{VSL}_y injection – $\Delta O_3(z)$ changes in the tropical lowermost stratosphere are not significant over the century (see Fig. 6f). This is in line with the modelled impacts on the TOC. In contrast, at mid-latitudes, the seasonal VSL^{Br} impacts on ozone significantly decrease over the cen-

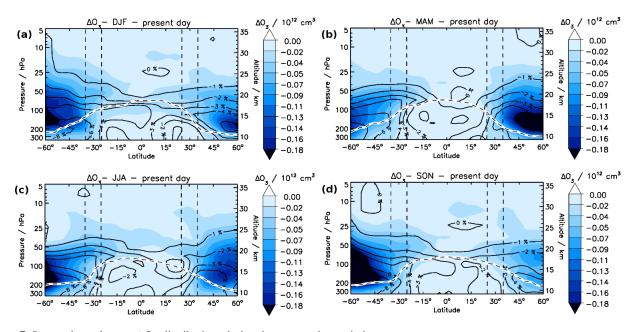


Figure 7. Seasonal zonal mean ΔO_3 distributions during the present-day period.

tury between 50 and 300 hPa (see Figs. 6f and S6a–d) even though the contribution percentage from Br_y^{VSL} to Br_y^{LL+VSL} increases towards the end of the century. Moreover, the projected seasonal and annual $\Delta O_3(z)$ trends peak precisely at

- ⁵ the same altitudes where the modelled relative $\Delta O_3(z)$ is significant (see Figs. 6f and S6a–d). These results highlight the clear dependence of VSL^{Br} impacts on the temporal evolution of background halogen abundances and the halogens' heterogeneous reactivation processes, which in turn depend
- ¹⁰ on temperature as well as the formation and distribution of ice crystals and stratospheric sulfate <u>aerosol surfaces</u>. In fact, the projected impacts on <u>vertical ozone distribution</u> at the end of the century are similar to those prevailing before 1980 for both <u>SH MLs and NH MLs</u> since the background halo-
- ¹⁵ gen abundances are similar between these periods. Thus, potential future changes in the lowermost stratospheric temperature (therefore influencing ice crystal formation), geoengineering of climate via injection of stratospheric sulfate, and the formation and distribution of stratospheric aerosol
- ²⁰ (e.g. future volcanic eruptions) will directly influence the processes of heterogeneous halogen reactivation and consequently VSL^{Br}-driven ozone depletion efficiency (Tilmes et al., 2008, 2012; Banerjee et al., 2016; Klobas et al., 2017; Heckendorn et al., 2009).

25 3.4 Changes in the long-term seasonal evolution of halogen-driven ozone loss rate due to VSL^{Br}

In order to determine the drivers controlling the VSL^{Br}driven seasonal impact on ozone in the lowermost stratosphere at mid-latitudes and in the tropics, we compute ³⁰ the odd-oxygen chemical loss from each ozone-depleting family considering the independent contributions of oxygen (O_{x-Loss}), hydrogen (HO_{x-Loss}), nitrogen (NO_{x-Loss}), and halogen ($Halog_{x-Loss}$; Brasseur and Solomon, 2005; Saiz-Lopez et al., 2014), Moreover, we discriminate between the individual contributions of pure bromine (BrO_{x-Loss}) and ³⁵ chlorine (CIO_{x-Loss}) cycles and the inter-halogen (CIO_{x} -BrO_{x-Loss}) cycle to the halogen family ($Halog_{x-Loss} =$ BrO_{x-Loss} + CIO_{x} -BrO_{x-Loss}). The odd-oxygen loss rate equations for the ozone-depleting families considered in this work are presented in Table S3. ⁴⁰

Figure 8 shows the temporal evolution of the annual contribution percentage of each ozone-depleting family to the total ozone loss rate in the lowermost stratosphere at SH MLs (~ 120 hPa) for both experiments as well as the seasonal evolution of HO_{x-Loss} and Halog_{x-Loss} contributions 45 in each experiment. The corresponding figures for the lowermost stratosphere at NH MLs (~ 120 hPa) and in the tropics (~ 50 hPa) are shown in Figs. S7 and S8, respectively. Halog_{x-Loss} (1) represents the second most important contribution to the total ozone loss rate after HO_{x-Loss} 50 $(\sim 80\%)$ at mid-latitudes in both experiments (see Figs. 8a and S7a), while within the tropics it represents the third family after HO_{x-Loss} and NO_{x-Loss} (see Fig. S8a). This partially explains the smaller modelled $\frac{\text{VSL}^{\text{Br}}}{\text{VSL}}$ impact on ozone within the tropics as $\text{Halog}_{x-\text{Loss}}^{\text{LL}+\text{VSL}}$ represents at most 10% 55 of the total ozone destruction over this region. Moreover, the enhancement in the $\text{Halog}_{x-\text{Loss}}^{\text{LL}+\text{VSL}}$ contribution with respect to $Halog_{x-Loss}^{LL}$ by including VSL^{Br} is mostly compensated by a decrease in the HO_{*x*-Loss} contribution at mid-*x*-Loss U + VSLlatitudes. For example, for the present day, Halog_{x-Loss} latitudes. For example, for the present day, $\text{Halog}_{x-\text{Loss}}^{\text{LL}}$ shows an increase of ~ 7 % with respect to $\text{Halog}_{x-\text{Loss}}^{\text{LL}}$ at SH MLs, which is compensated by a reduction of $\sim 6\%$ in

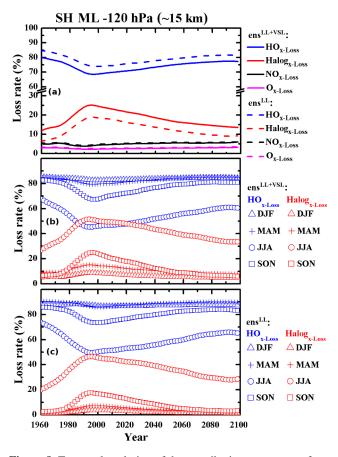


Figure 8. Temporal evolution of the contribution percentages from different ozone-depleting families (O_{x-Loss} , NO_{x-Loss} , HO_{x-Loss} , Halog_{*x*-Loss}) to the total odd-oxygen loss rate in the lowermost stratosphere at SH MLs₁(120 hPa). Panel (**a**) shows the annual mean contribution of each ozone-depleting family for the experiments with (solid line; ens^{LL+VSL}) and without (dashed line; ens^{LL}) VSL^{Br} sources. Panels (**b**) and (**c**) show the seasonal mean contributions of both HO_{*x*-Loss} and Halog_{*x*-Loss} for the experiments with and without VSL^{Br} sources, respectively.

 HO_{x-Loss}^{LL+VSL} . In particular, the interaction between different catalytic ozone-depleting cycles involving OH and halogen radicals as well as the influence of reactive halogens in altering the OH/HO₂ ratio has been described elsewhere (Bloss 5 et al., 2005; Saiz-Lopez and von Glasow, 2012).

Even though the additional seasonal ozone depletion due to VSL^{Br} peaks during spring followed by winter at midlatitudes, the largest Halog_{x-Loss} contribution to the total ozone loss rate occurs during winter in both experiments, ¹⁰ presenting a much smaller contribution during the rest of the

- year (see Figs. 8b, c and S7b, c). Indeed, this marked seasonal behaviour prevails both during boreal winter at NH MLs and during the austral winter at SH MLs. This is explained by considering the strong seasonal increase in active-chlorine
- ¹⁵ abundance (driven by heterogeneous reactivation in ice crystals) as well as the winter decline in HO_x abundance (driven by the decrease in solar radiation with respect to summer-

time). In fact, the enhancement in the $\text{Halog}_{x-\text{Loss}}^{\text{LL}+\text{VSL}}$ contribution due to VSL^{Br} makes it the dominant ozone-depleting family during winter between 1985 and 2020 at SH-MLS_{20} surpassing in importance the role played by the otherwise dominant $\text{HO}_{x-\text{Loss}}$. In contrast, no seasonal variability of $\text{Halog}_{x-\text{Loss}}$ contribution is modelled within the tropics in either experiment (see Fig. S8b and c), and thus the seasonal VSL^{Br} impact on ozone remains approximately constant during the whole modelled period around the annual mean.

Figure 9 shows the long-term evolution of the contribution percentage of BrO_{x-Loss} , ClO_{x-Loss} , and $ClO_{x}-BrO_{x-Loss}$ to the Halog_{x-Loss} as a function of months and years in the lowermost stratosphere at SH MLs (120 hPa) for both experi- 30 ments. Equivalent results are presented for NH MLs and the tropics in Figs. S9 and S10, respectively. For any fixed year in both experiments, the BrO_{x-Loss} contribution peaks during late summer and early autumn, while its minimum contribution occurs during the austral winter. In contrast, both the 35 ClO_{x-Loss} and ClO_{x} -BrO_{x-Loss} reach their maximum contribution during winter and early spring. Consequently, the inclusion of VSL^{Br} increases the BrO_{*x*-Loss}^{LL+VSL} contribution in all seasons while also increasing the ClO_x -Br O_{x-Loss}^{LL+VSL} contribution mainly in winter. This is at the expense of a de-40 crease in the $\text{ClO}_{x-\text{Loss}}^{\text{LL+VSL}}$ contribution. In addition, the wintertime $\text{ClO}_{x}^{\text{-}-\text{BrO}_{x-\text{Loss}}}$ contribution reaches up to 50 % between 1990 and 2020, decreasing towards the end of the century as Cl_v^{LL} and Br_v^{LL} abundances decrease within the A1 emission scenario. The decrease in both ClO_{x-Loss}^{LL+VSL} 45 and ClO_{χ} -BrO_{χ -Loss} over time is offset by an increase in the BrO_{x-Loss}^{LL+VSL} contribution. Accordingly, the inclusion of VSL^{Br} produces similar changes in the evolution of the contributions of each of the cycles that compose Halog_{x-Loss} at NH MLs, although the seasonal BrO_{x-Loss} contribution is 50 greater in both experiments compared to SH MLs during the entire modelled period (see Fig. S9).

In summary, the $Halog_{x-Loss}^{LL+VSL}$ contribution to the total ozone loss rate in summer and autumn is almost entirely dominated by $\text{BrO}_{x-\text{Loss}}^{\text{LL+VSL}}$, while in the winter months 55 when the seasonal $\text{Halog}_{x-\text{Loss}}^{\text{LL+VSL}}$ contribution to the total ozone loss rate is comparatively higher, a transition from the $\text{ClO}_x \text{BrO}_{x-\text{Loss}}^{\text{LL+VSL}}$ contribution (~ 50%) at the beginning of the century towards BrO_{x-Loss}^{LL+VSL} at the end of the century is modelled. Thus, the halogen-driven ozone loss projected towards the end of the 21st century is mostly dominated by BrO_{x-Loss}, leading to a less marked seasonal VSL^{Br} impact on both vertical ozone distribution and the TOC, in accordance with the modelled results. On the other hand, within the tropics, $\text{BrO}_{x-\text{Loss}}^{\text{LL+VSL}}$ dominates the seasonal Halog_{*x*-Loss}^{LL+VSL} contribution for practically the entire modelled period (see Fig. S10), while ClO_{x-Loss}^{LL+VSL} represents a maximum contribution of up to $\sim 35 \%$ during the end of the 20th century. The seasonal increase from BrO_{x-Loss}^{LL} to BrO_{x-Loss}^{LL+VSL} by including VSL^{Br} is well marked and is prac-

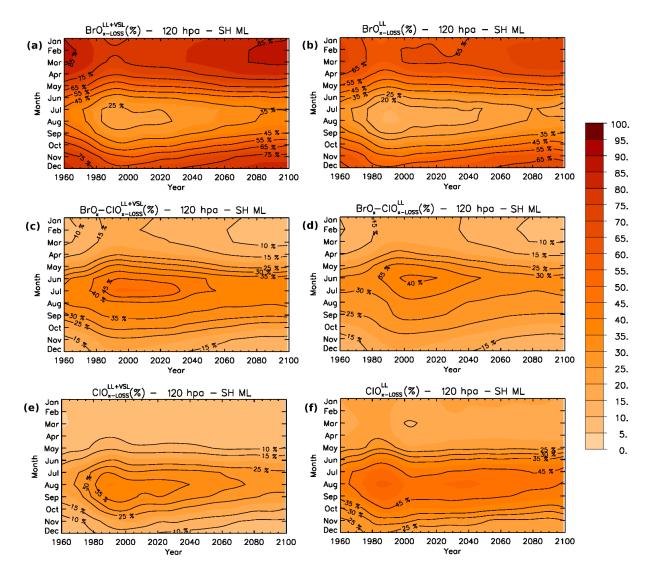


Figure 9. Evolution of the (**a**, **b**) $\text{BrO}_{x-\text{Loss}}$, (**c**, **d**) $\text{ClO}_x-\text{BrO}_{x-\text{Loss}}$, and (**e**, **f**) $\text{ClO}_{x-\text{Loss}}$ contribution percentages to $\text{Halog}_{x-\text{Loss}}$ as a function of the year and month in the lowermost stratosphere at <u>SH-MLs</u> (120 hPa) for the experiments (**a**, **c**, **e**) with and (**b**, **d**, **f**) without VSL^{Br} sources.

tically offset by a reduction in the $\text{ClO}_{x-\text{Loss}}^{\text{LL+VSL}}$ contribution. Furthermore, unlike mid-latitudes, $\text{ClO}_x - \text{BrO}_{x-\text{Loss}}^{\text{LL+VSL}}$ has a lower contribution to the tropical $\text{Halog}_{x-\text{Loss}}^{\text{LL+VSL}}$, showing that VSL^{Br} -driven topical ozone loss is independent of changes in the background halogen abundance throughout the modelled period.

4 Conclusions

This study has explored the impact of VSL^{Br} substances on the temporal evolution of stratospheric ozone both in the late 20th century and in the course of the 21st century at mid-latitudes and in the tropies. The analysis compares two independent experiments, one of which considers only the anthropogenic LL^{Cl} and LL^{Br} source contributions, while the other considers - in addition to long-lived sources the contributions of VSL^{Br} oceanic sources. We have evaluated annual and seasonal mean changes in vertical stratospheric ozone distribution and the TOC, and we have examined the projected ozone depletion (ΔO_3 and ΔTOC) trends over the century under a scenario in which both the relative contribution of VSL^{Br} to total inorganic bromine CE10 20 and the background inorganic-chlorine abundances shift towards the future. Finally, we have also assessed the long-term seasonal contribution changes of halogen-mediated ozone loss (Halog_{x-Loss}) due to $\frac{VSL^{Br}}{VSL^{Br}}$ focusing on the lowermost stratosphere and distinguishing the role of pure (BrO_{x-Loss} , ClO_{x-Loss}) and inter-halogen (ClO_x-BrO_{x-Loss}) cycles that compose Halog_{x-Loss}. The results presented here highlight the importance of considering natural sources of halocarbons

to determine the overall evolution of stratospheric ozone in the tropics and at mid-latitudes during the 21st century.

Our analysis shows that the inclusion of VSL^{Br} results in a realistic stratospheric bromine loading that improves the

- ⁵ agreement between the model and observations of the TOC at mid-latitudes, highlighting the need to consider these additional natural emissions to determine the overall evolution of ozone during the 21st century. Our results are in accordance with previous modelling studies that included natural
- ¹⁰ bromocarbons and explored stratospheric ozone in the recent past (Salawitch et al., 2005; Feng et al., 2007; Sinnhuber et al., 2015) as well as by the end of the 21st century (Falk et al., 2017). However, unlike these previous works, we have explored seasonal ozone depletion arising from VSL^{Br} with ¹⁵ respect to the background stratospheric LL^{Cl} and LL^{Br} abun-

dances between 1960 and 2100.

At mid-latitudes, VSL^{Br} introduces a continuous reduction in the TOC during the entire modelled period, with the largest TOC depletion occurring during the period with the high-²⁰ est background halogen abundance ($\Delta TOC^{2000} = -8$ DU,

- -2.5%, for SH MLs and -5.5 DU, -1.6%, for NH MLs). In turn, a significant decrease in the impact towards the end of the 21st century at mid-latitudes is projected, with Δ TOC trends of 0.15% and 0.10% per decade for SH MLs and NH
- ²⁶ MLs, respectively. Even though the contribution of VSL^{Br} to total stratospheric bromine reaches almost 50 % by the end of the 21st century, our simulations show a statistically significant smaller VSL^{Br} impact on lowermost stratospheric ozone as we move into the future. This result highlights that VSL^{Br}-
- ³⁰ driven ozone loss is closely linked to the temporal evolution of LL^{Cl} and LL^{Br} abundances regulated by the Montreal Protocol, as suggested by Yang et al. (2014). Along these same lines, Hossaini et al. (2015a) demonstrated that the chlorine load in the lowermost stratosphere has increased as a conse-
- ³⁵ quence of a recent and ongoing growth in emissions from anthropogenic VSL^{Cl} sources (mainly CH₂Cl₂) not controlled by the Montreal Protocol. If this VSL^{Cl} emissions trend continues into the future, additional studies will be required to quantify the increase in VSL^{Br} impact modelled in this work
 ⁴⁰ on the projected ozone loss trends over the century considering the additional inorganic chlorine from VSL^{Cl}.

The largest TOC depletion associated with VSL^{Br} corresponds to the spring months followed by winter, with maximum springtime ΔTOC^{2000} reaching -10 DU ($\sim -3 \%$) ⁴⁵ and -7.7 DU ($\sim -2 \%$) for SH MLs and NH MLs, respectively. We find that the inclusion of VSL^{Br} leads to seasonal changes in both local O₃(z) reduction and its vertical $\Delta O_3(z)$ distribution within the lowermost stratosphere, which could result in different seasonal perturbations of the ⁵⁰ radiative effects mediated by ozone. This seasonal enhancement in VSL^{Br}-mediated local ozone loss is strongly influenced by the seasonal changes in heterogeneous chlorine reactivation processes in icc crystals (mainly by ClONO₂ and HCl recycling). In fact, the modelled additional increase in ⁵⁵ heterogeneous chlorine reactivation efficiency in the lowermost stratosphere at SH MLs compared to NH MLs leads to hemispheric asymmetry in the VSL^{Br}-driven ozone reductions.

Even though the additional seasonal ozone depletion due to VSL^{Br} peaks during spring followed by winter, the largest ⁶⁰ Halog_{*x*-Loss}^{LL+VSL} contribution to the total ozone loss rate occurs during winter, with a much smaller contribution during all other seasons. In fact, the wintertime Halog_{*x*-Loss}^{LL+VSL} contribution dominates the total ozone loss rate between 1985 and 2020 in the lowermost stratosphere at <u>SH-MLs</u>, with a ClO_{*x*}-BrO^{LL+VSL}_{*x*-Loss} contribution to Halog_{*x*-Loss}^{LL+VSL} reaching up to 50%. Due to the expected decrease in the LL^{Cl} and LL^{Br} abundances, a transition from ClO_{*x*}-BrO^{LL+VSL}_{*x*-Loss} to BrO_{*x*-Loss} during the course of the 21st century is projected, leading to a less marked seasonal VSL^{Br} impact towards the ⁷⁰ end of the century.

Within the tropics, although our model captures a slightly greater TOC depletion associated with VSL^{Br} during the 1990s, no significant change is projected by the end of the 21st century. Halog_{x-Loss} **GEII** represents the third family ⁷⁵ in the contribution to the total ozone loss rate after HO_{x-Loss} and NO_{x-Loss} in the tropical lowermost stratosphere, with an almost constant seasonal contribution throughout the modelled period. BrO^{LL+VSL}_{x-Loss} practically dominates the contribution to Halog^{LL+VSL}_{x-Loss}, with a maximum ClO_x–BrO^{LL+VSL}_{x-Loss} contribution of up to ~ 30 % by the end of the 20th century, reflecting the low sensitivity of VSL^{Br} to background halogen abundances, which allows it to drive stratospheric ozone depletion.

Data availability. Computing resources, support, and data storage ⁸⁵ are provided and maintained by the Computational and Information System Laboratory from the National Center of Atmospheric Research (CISL, 2017). The code of the CAM-chem model can be downloaded from the website (https://www2.acom.ucar.edu/gcm/cam-chem). Data that support the findings of this study can ⁹⁰ be downloaded from the AC2 web page (https://ac2.iqfr.csic.es/en/publications).

Supplement. The supplement related to this article is available online at: https://doi.org/10.5194/acp-20-1-2020-supplement.

Author contributions. RPF and ASL designed the experiments. ⁹⁵ RPF and JAB configured and ran all the simulations. JAB processed the results of all simulations, led the manuscript, and prepared the figures. RPF, ASL, FIS, and CAC provided additional feedback on geophysical processing and representation. JAB, RPF, FIS, CAC, JFL, and ASLCEPP were involved in the discussion and iterations ¹⁰⁰ of the manuscript.

Competing interests. The authors declare that they have no conflict of interest.

Acknowledgements. This study has been funded by the Agencia Nacional de Promoción Científica y Técnica (ANPCYT PICT-2016-0714, Argentina) and the European Research Council Executive Agency under the European Union's Horizon 2020

- 5 Research and Innovation programme (project "ERC-2016 COG 726349 CLIMAHAL"). Rafael Pedro Fernandez and Javier Alejandro Barrera would like to thank CONICET, UNCuyo (SeC-TyP M032/3853), and UTN (PID 4920-194/2018) for additional financial support. Alfonso Saiz-Lopez, Carlos Alberto Cuevas, and
- ¹⁰ Fernando Iglesias-Suarez are supported by the Consejo Superior de Investigaciones Científicas (CSIC) of Spain. We are very thankful to $D_{\underline{\lambda}}$ Kinnison **TSID** for very helpful discussions and suggestions. We thank the two anonymous reviewers for their helpful comments that improved this paper.
- ¹⁵ Financial support. This research has been supported by the European Research Council Executive Agency under the European Union's Horizon 2020 Research and Innovation programme (grant no. ERC-2016-COG 726349 CLIMAHAL) and the Agencia Nacional de Promoción Científica y Técnica (grant no. AN-²⁰ PCYT PICT-2016-0714). **TSIS**

Review statement. This paper was edited by Marc von Hobe and reviewed by two anonymous referees.

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note the remarks at the end of the manuscript.

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