Response to Reviewer 1

The authors would like to thank the reviewer for their detailed comments and suggestions to improve the manuscript. Please find below our responses and details of corrections made to the manuscript.

Introduction, page 1, line 29:

References:

.....Villena et al., 2011 change to Villena et al., 2012, (please see Reference, page 22, line 13)

There are two different publications by Villena et al. cited in our manuscript. The paper from 2011 is about HO_x production from HONO photolysis and the 2012 paper is about interferences in NO_2 chemiluminescence instruments. We therefore believe that these are both cited correctly.

Introduction, page 3, line 11, 20....: General a range of numbers should be specified as "a to b" and not "a - b". And add for all given physical quantities one space between number and unit (please see guidelines for authors, mathematical notation and terminology). E.g. please change.....ranging from 0.16-1.00%.... toranging from 0.16 to 1.00 %....

We thank the reviewer for noting these errors. The manuscript has now been revised to change these instances.

Introduction, page 3, line 34: Data taken at site in North Kensington, London during 2012 as part of Clean Air for London (ClenarLo)...., please add the reference Bohnenstengel et al., 2015

Reference has been added.

Introduction, page 4, line 11: Supplementary table S1: Why are for the motorcycles and in particular buses in comparison with cars and light and heavy goods only the total number of licensed vehicles are listed? Buses and coaches have often a diesel engine and the contribution of diesel fuelled vehicles to the total number of vehicles will be higher than 44%.

The reviewer is correct in that if buses and coaches are included the contribution of total diesel vehicles if would be higher than 44%. The statistics from Driving and Vehicle Licensing Agency (DVLA) and Department for Transport (DfT), however, only provide numbers of vehicles by body type for buses and coaches (see: <u>https://www.gov.uk/government/statistical-data-sets/veh06-licensed-buses-and-coaches</u>) and not fuel type, therefore we are unable to provide an accurate number for the total licensed diesel buses and coaches. The total number of buses and coaches will include some minibuses which may be petrol fuelled and ultra-low emission buses.

Experimental, page 4, line 27: Please used SI units, e.g. please change 52 kph to 52 km h-1 and please check also all other units in text and in figures!

(please see guidelines for authors, mathematical notation and terminology) Experimental, page 5, line 22: Please change ...0.9 L/min to 0.9 L min-1 (please see guidelines for authors, mathematical notation and terminology).

We thank the reviewer for bringing these errors to our attention. Units have now been changed to the appropriate format in the revised manuscript. We have also added vehicle speed values in kilometres per hour (km h^{-1}) in addition to miles per hour (commonly used in the UK) in parentheses on page 4 line 27. The following text has also been added to the caption for Figure S2 in the supplementary material to provide further information on vehicle speeds in the tunnel:

"The speed limit in the tunnel is 30 miles per hour (48 km h^{-1}), therefore, most vehicles drive slightly above the speed limit, except when the tunnel is congested."

Experimental, page 6, line 6: Statistical errors are given in \pm range, please add for all statistical errors \pm , e.g. changewere 0.8 ppbv.....to were \pm 0.8 ppbv.....(see also line 8, 9, 12...) (please see guidelines for authors, mathematical notation and terminology).

Changes have been made to the final manuscript to include ± for all statistical errors.

We also have adjusted the text on page 6, lines 14 & 15, to emphasise the origin of the systematic errors in the BBCEAS measurements of HONO and NO₂: "…were dominated by the systematic uncertainties in the cross sections and the mirror reflectivity." And we have added the bracketed text "(from combining the statistical and systematic errors)" on page 6, line 16, to make it clear that the total measurement uncertainty includes both these two types of error.

Result and discussion, page 7, line 2 In figure 2 also the diurnal cycle of NO is shown, but not mention or discuss in the paper. In contrast to the shown diurnal cycle of NO the NOx diurnal cycle is discuss but not shown in figure 2, why?

When NO_x is discussed in this section the authors were referring to the NO and NO_2 cycles together. As the reviewer has rightly demonstrated however this is not clear, therefore the text in this section has been changed to the following:

"Figure 4 presents 15 minute averaged time series for gases measured in the tunnel. NO, NO₂, HONO and CO₂ follow the same diurnal cycle as each other indicating that they have a similar source. There is a clear difference in weekday and weekend diurnal cycles with peaks associated with rush hour traffic observed in NO, NO₂, HONO and CO₂ during the weekdays, which are not present at the weekend. The measured concentrations are also lower at the weekend, on average."

Result and discussion, page 10, line 13 How was the uncertainty of the wind speed measurements (see also table 1) ? Better than \pm 0.3 % ?, if not please change the given wind speed from3.89 m s-1 to 3.9 m s-1.

The reviewer is correct; we quoted too many decimal places, and the last decimal is not significant. The wind speed has now been changed from 3.89 m s^{-1} to 3.9 m s^{-1} . This change brings the text into alignment with the manufacturer's specification for the Kestrel anemometer that we used to record wind speed: "accuracy = ± 3 % or the last digit of the reading", which in our case means $\pm 0.1 \text{ m s}^{-1}$.

Result and discussion, page 14, line 25 Please change... Liang et al. estimated.... to Liang et al. (2017) estimated....

Change has been completed.

Result and discussion, page 15, line 4, 5, 9... References Carslaw et al., 2016 Matthaios et al., 2108 and Grange et al., 2017 are mention in the text but not listed in the references list, please add these references to the list and check all others!!!

We thank the reviewer for spotting this error. The references have been checked and updated to ensure they are all correct.

References The reference list contains some typos, some reference are unlisted and should be carefully checked by the authors. For example: References, page 20, line 41 Rappengluck, B. change to Rappenglück, B.....

As above, all references have been checked and corrected where needed.

Figures and Figure captions: Some displayed measurement data have no error bars e.g. figure 6, S1 and S7, why? Are the errors of the measured data are taken into account by the calculation of the regression parameters (e.g. slope and intercept) Please check the mathematical notation and terminology. e.g. figure 3a, y-axis, please change number of vehicles/hour....to.... number of vehicles h-1

Mathematical notation and terminology have been changed in the figures for the revised manuscript.

Error bars representing 1- σ standard deviation of the mean have now been added to Figures 7b and 7c in the main paper and Figure S7 in the supplementary material. In the original manuscript the diesel and petrol fractions in Figure 7 and Figure S7 were determined for each 1-hour bin from the sum of the number of vehicles over the four days of ANPR measurements. In the revised version the mean and 1- σ standard deviation for each 1-hour bin has been calculated to 2 decimal places. Additionally, the Δ HONO/ Δ NO_x data plotted in Figure S7 has also been rounded to 2 decimal places. These revisions have resulted in a small change in the slope and intercept from the previous version. The revised data also changed the calculated values of ER_{DV} and ER_{NDV} in section 3.2.3, however the final conclusions remain the same.

Additional text given below has also been added to the main paper on page 12, lines 31 & 32, to provide a calculated value of the emission ratio for diesel vehicles, and associated error.

"Extrapolating the best fit line in Figure S7 to x = 1 gives a plausible emission ratio for diesel vehicles of $ER_{DV} = 1.35 \pm 0.50$ %."

Error bars have not been included on the regression plots for Figure 6 and Figure S1 because this makes the figures look crowded and the regression lines are less clear. The regression was not weighted by the errors in the measurements. We have instead used a non-weighted reduced-major-axis method when calculating the regression parameters and quoted the 95% confidence interval on the slope to provide information on the uncertainty in the calculation of $\Delta HONO/\Delta NO_x$.

Response to Reviewer 2

We would like to thank the reviewer for their kind comments and suggestions. Please find below our responses and details of corrections made to the manuscript.

1. 2.3 Instrumental techniques. There is no information for the components of the BBCEAS. Because the BBCEAS system for HONO have been originally built up by the authors, the authors should show the company name and products number for the parts of the BBCEAS, especially LED, mirror and detector.

We agree that more technical information should have been provided about the BBCEAS instrument built by Dr Ball and his research group. The following text has been added to section 2.3:

* LED light source: "LEDengin LZ1-10UV00, nominal peak wavelength = 365 nm".

* Mirrors: "a cavity formed by two highly reflective mirrors (Layertec GmbH, 25 mm diameter, 0.5 m radius of curvature, high reflectivity 370 to 395 nm)". And a few lines later: "their measured reflectivity peaked at 99.940 % at 387 nm."

* Spectrometer: "BBCEAS spectra were integrated... using an OceanOptics QEPro spectrometer (instrument line shape = 0.20 nm HWHM)."

2. 3.2.3 HONO emission ratios The definition of "delta"HONO seems to be ambiguous. Does "delta_HONO" mean the background corrected HONO (delta_HONO=HONO_tunnel - HONO_bg), or the background and secondary formation corrected HONO(delta_HONO=HONO_tunnel - HONO_bg -HONO_secondary formation)? The authors should make clear.

We thank the reviewer for bringing this to our attention. We can see how the of use of Δ HONO could be ambiguous here. Δ HONO in this case represents the HONO corrected for background and secondary HONO formation.

We have added the following text at end of section 3.2.2 to clarify this point.

"In the following sections the final HONO dataset (Δ HONO) is corrected for both background HONO levels and the heterogeneous HONO contribution."

Nitrous acid (HONO) emissions under real-world driving conditions from vehicles in a UK road tunnel.

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Abstract.

Measurements of atmospheric boundary layer nitrous acid (HONO) and nitrogen oxides (NO_x) were performed in summer 2016 inside a city centre road tunnel in Birmingham, United Kingdom. HONO and NO_x mixing ratios were strongly correlated with traffic density, with peak levels observed during the early evening rush hour as a result of traffic congestion in the tunnel.

- 15 A daytime ΔHONO/ΔNO_x ratio of 0.85_% (0.72<u>to</u>-1.01%, 95_% Clconfidence interval) was calculated using reduced major axis regression as for the overall fleet-average (comprising 59_% diesel-fuelled vehicles). A comparison with previous tunnel studies and analysis on <u>the</u> composition of the fleet suggest that goods-vehicles have a large impact on the overall HONO vehicle emissions; however, new technologies aimed at reducing exhaust emissions, particularly for diesel vehicles, may have reduced the overall direct HONO emission in the UK. This result suggests that in order to accurately represent urban
- 20 atmospheric emissions and <u>the</u>OH radical budget, fleet-weighted HONO/NO_x ratios may better quantify HONO vehicle emissions in models, compared with use of a single emissions ratio for all vehicles. The contribution of the direct vehicular source of HONO to total ambient HONO concentrations is also investigated and results show that, in areas with high traffic density, vehicle exhaust emissions are likely to be the dominant HONO source to the boundary layer.

1 Introduction

Nitrous acid (HONO) is an important atmospheric constituent in the boundary layer, as its photolysis leads to the formation of OH radicals (R1), which drive atmospheric oxidation reactions, pollutant removal and the formation of secondary species. This is particularly important in urban areas where the measured HONO mixing ratios can reach up to parts per billion, and HONO photolysis can be the dominant HO_x source (Alicke et al., 2002; Elshorbany et al., 2009; Kleffmann, 2007; Lee et al., 2016; Michoud et al., 2012; Villena et al., 2011).

$HONO + hv \xrightarrow{\lambda < 400 nm} OH + NO$

(R1)

The sources of HONO in the atmosphere can be primary or secondary (Figure 1). Primary sources include; direct emissions

- 5 from combustion processes such as vehicle emissions (Kirchstetter et al., 1996; Liu et al., 2017; Rappenglück et al., 2013; Trinh et al., 2017; Xu et al., 2015), soil microbial activity (Laufs et al., 2017; Maljanen et al., 2013; Meusel et al., 2018; Oswald et al., 2013; Su et al., 2011) and biocrusts (Maier et al., 2018; Meusel et al., 2018; Weber et al., 2015). Secondary sources were thought to be dominated by <u>the homogeneous gas phase reaction of NO and OH during the day (resulting in a null cycle with R1)</u>, and the heterogeneous production of HONO from NO₂ on surfaces at night (Calvert et al., 1994; Finlayson-Pitts et al., 2017).
- 10 2003; Jenkin et al., 1988; Kleffmann et al., 1998; Stutz et al., 2002). More recently, studies have shown that heterogeneous reactions are also important sources of HONO during the day and include photo-enhanced reduction of NO₂ on organic substrates (George et al., 2005; Monge et al., 2010; Stemmler et al., 2006) and nitrate photolysis (Reed et al., 2017; Ye et al., 2017; Zhou et al., 2011). Despite numerous studies over the last few decades, substantial uncertainties remain regarding the relative magnitude of these sources, with models frequently unable to account for total measured HONO concentrations
- 15 without the inclusion of unknown sources, typically driven by photolysis (Huang et al., 2017; Lee et al., 2016; Michoud et al., 2014; Vandenboer et al., 2014; Vogel et al., 2003; Wang et al., 2017). A photo-stationary steady state method based on measurements of HONO, OH and NO_x has previously been used to infer missing HONO sources; however, in areas of high spatial heterogeneity this method breaks down because the lifetime of HONO is much longer than that of OH, or of NO (with respect to NO_x-O₃ equilibrium) and NO_x (Crilley et al., 2016; Lee et al., 2013).

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In areas with high traffic density, HONO emitted directly from vehicle exhausts is an important source, as indicated by large peaks in ambient HONO concentrations observed during rush hour periods (e.g. Qin et al., 2009; Rappenglück et al., 2013; Stutz et al., 2004; Tong et al., 2016; Wang et al., 2017; Xu et al., 2015). Tong et al. (2016) estimated that direct emissions from vehicles at an urban site in Beijing contributed 48.8_% of the total measured HONO, almost 5 times higher than the contribution at a suburban site (10.3_%) located 50 km northeast of Beijing city centre.

A HONO/NO_x emission ratio is often used to parameterise the HONO contribution from vehicles. Various studies have determined this ratio either directly via chassis dynamometers (Calvert et al., 1994; Liu et al., 2017; Nakashima and Kajii, 2017; Pitts et al., 1984; Trinh et al., 2017) or from ambient roadside/tunnel measurements (Kirchstetter et al., 1996; Kurtenbach et al., 2001; Liang et al., 2017; Rappenglück et al., 2013; Yang et al., 2014). Chassis dynamometer studies benefit by the direct quantification of HONO from vehicle exhausts under different driving cycles; however, the number of vehicles included tested in these studies is very limited and may not be representative of the wider fleet or real world operational conditions. Both tunnel and ambient roadside measurements on the other hand allow for analysis of HONO emissions for from a larger vehicle

fleet, under real-world driving conditions. Tunnel studies have the additional benefits of eliminating photolytic loss and photoenhanced surface sources of HONO and of reducing dispersion. However, care must be taken to accurately correct for background levels of HONO at the tunnel entrance and for other HONO sources, such as heterogeneous reactions on tunnel walls and emitted particles.

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Reports of the HONO/NO_x emission ratio from chassis dynamometer and ambient/tunnel studies are highly variable, ranging between 0.03 and 2.1 %, depending on fuel type and implementation of technologies aimed at reducing emissions, such as diesel oxidation catalysts (DOCs) and <u>diesel</u> particulate filters (DPFs). An emission ratio of 0.8_% is widely used in modelling studies, which is based on a tunnel study with a fleet comprising primarily gasoline vehicles (~75 %) conducted in 1997/98 (Kurtenbach et al., 2001). However, a recent chassis dynamometer study by Trinh et al. (2017), showed that HONO/NO_x ratios were higher from diesel vehicles (ranging from 0.16 to -1.00 %) compared to petrol vehicles (0 to -0.95 %), under most driving

conditions. The higher HONO/NO_x ratio observed from diesel vehicle exhausts is thought to be due to a reduction-oxidation reaction as proposed by Ammann et al. (1998) and Gerecke et al. (1998) where NO_2 can be converted heterogeneously to HONO on soot via R2.

$$NO_2 + [C - H]_{red} \longrightarrow HONO + [C]_{ox}$$
 (R2)

where, $[C - H]_{red}$ is a surface site on the soot particle.

- A number of laboratory studies have supported this reaction, with HONO yields varying between 20<u>and</u>-100 % depending on the fuel type used to generate the soot, initial NO₂ concentrations and soot coatings (Arens et al., 2001; Aubin and Abbatt, 2007; Gerecke et al., 1998; Guan et al., 2017; Khalizov et al., 2010; Kleffmann et al., 1999; Lelièvre et al., 2004; Romanias et al., 2013; Stadler and Rossi, 2000). An observation that is consistent across many laboratory experiments is that the uptake of NO₂ decreases over time, which has previously been attributed to the deactivation of soot surface receptor sites (Kalberer et
- 25 al., 1999). Lelièvre et al. (2004) observed complete deactivation after the soot was exposed to ambient outdoor conditions for approximately 20 hours. As a result of the soot surface deactivation, reaction R2 is not expected to have a large impact on atmospheric HONO levels once soot has been exhausted from the vehicle. However, Khalizov et al. (2010) found that if soot was pre-heated up to 300 °C, the HONO yield increases, as a result of the removal of products from incomplete combustion, allowing for a greater number of reactive sites to reduce NO₂ to HONO. Reactions within the vehicle exhaust pipe, on fresh
- 30 soot, may still be a potential source of HONO from vehicles. Determining the magnitude of this source under real-world conditions, however, is challenging. Two recent studies in Hong Kong investigated the relationship between soot and HONO directly emitted from vehicles, with contrasting results. Xu et al. (2015) found a strong positive correlation ($R^2 = 0.83$) between HONO/NO_x and black carbon (BC) in fresh traffic plumes sampled at an ambient air monitoring site, but no correlation was ascertained in a separate road tunnel experiment by Liang et al. (2017). Data taken at a site in North Kensington, London,

during 2012 as part of the Clean Air for London (ClearfLo) project (Bohnenstengel et al., 2015)_show a <u>modest</u> correlation between HONO and BC ($R^2=0.71$); however, there is no clear correlation between HONO/NO_x and BC when sampling fresh pollution plumes (i.e. when NO/NO_x ratios are large) (see Figure S1 in the supplementary material). This agrees with Liang et al. (2017) who suggested that HONO formed via NO₂ conversion on BC may be insignificant even in a road tunnel scenario shortly after emission.

The primary aim of this study is to determine HONO/NO_x emission ratios under real-world driving conditions, with a fleet containing a high proportion of diesel vehicles. In the UK, at the end of 2016, there were over 38 million vehicles licensed for use on the roads (DfT, 2017). The majority of licensed-vehicles were passenger cars (~83_%) and goods vehicles (11.4_%),
with the remainder consisting of motorcycles, buses/coaches and other vehicles (e.g. agricultural machines, ambulances). Diesel fuelled cars and goods vehicles, in total, account for 44_% of the vehicles on the road in the UK in 2016 (data on the statistics discussed here can be found in Table S1 in the supplementary material). To our knowledge this is the first study of direct HONO exhaust emissions performed in the UK and the first study in any country that has a diesel composition greater than 40_% of the vehicle fleet. Using co-located HONO, NO_x and CO₂ data, we investigated the HONO/NO_x ratio using measurements taken in a road-tunnel in the city centre of Birmingham and considered the impact of new technologies implemented in the European emission standards on the measured HONO emissions. Finally, the contribution of direct HONO emissions to the total HONO measured in an urban area was determined.

2 Experimental

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20 2.1 Queensway Tunnel

Measurements took place in the southbound bore of the Queensway Tunnel (Figure 2), a two-bore, twin lane tunnel, 548 m in length located in the centre of Birmingham, UK (52° 28' 46" N, 1° 54' 20" W). The tunnel forms part of the A38 roadway which is a major route in to and out of Birmingham City Centre, and links to the M6 motorway north of the city. The instruments were located at the distant end of a maintenance area approximately 435 m from the entrance to the southbound tunnel. The two bores of the tunnel are separated by a solid wall eliminating any influence in the measurements from vehicles travelling in the northbound bore. The tunnel was not mechanically ventilated during the measurement period. Airflow through the tunnel therefore is via natural wind flow or induced by vehicle movement (piston effect). The average speed of vehicles through the tunnel during the daytime (06:00 to -19:59) is 52 km h⁻¹kph (33 miles per hour (mph)mph) which drops to 33 km h⁻¹kph (21 mphmph) at 17:00 during peak evening rush hour as a result of congestion (see Figure S2).

2.2 Fleet information

Data from an Automatic Number Plate Recognition (ANPR) camera commissioned by Birmingham City Council were used to obtain information on the vehicle fleet passing through the tunnel. ANPR data were taken from 8 to -11 November 2016 and included vehicle type, euro-Euro classification and capture rate of the vehicle number plates, on a 15 minute time scale.

- 5 The ANPR capture rate is based on a comparison of manual classified counts (MCC) with the ANPR data. The capture rate was typically around 90 % of MCC, with counts missing from ANPR usually as a result of the number plate being obscured (Rhead et al., 2012). We assumed the same fleet proportions for the missing vehicle counts and adjusted the final data accordingly. Figure 3 shows the mean hourly number of vehicles travelling through the tunnel during the weekdays and a breakdown of vehicles by type. The ANPR data were also compared to manual traffic counts performed at the exit of the 10 southbound tunnel on selected days during the measurement period with good agreement. Approximately 37,700 vehicles travel through the southbound bore of the Queensway tunnel during an average weekday. The majority of these vehicles are petrol (40 %) and diesel (44 %) fuelled passenger cars, with the remaining fleet comprised primarily of diesel fuelled light-

15 2.3 Instrumental techniques

(2.5%) and passenger vehicles (taxis and buses) (1.3%).

The instruments were installed in the tunnel in close proximity (1.5 m) to the roadside and made measurements from 29th July until 8th August 2016. Access to the instruments was only possible when the tunnel was closed to traffic during maintenance periods, which occurred approximately every 2 weeks. Table 1 provides an overview of the instrumentation deployed in the tunnel during the campaign.

goods vehicles (LGVs, e.g. vans, small pick-ups) (10.4 %), ordinary goods vehicles (OGVs, e.g. trucks, articulated vehicles)

- 20 Direct spectroscopic measurements of HONO and NO₂ were made by broadband cavity enhanced absorption spectroscopy (BBCEAS) (Langridge et al., 2009; Thalman et al., 2015). The BBCEAS instrument operated in the wavelength range 363 to -388 nm, which includes a highly structured part of the NO₂ spectrum and two HONO absorption bands at 368 and 385 nm. Ultraviolet light from an LED light source (LEDengin LZ1-10UV00, nominal peak wavelength = 365 nm) was directed through a cavity formed by two highly reflective mirrors (Layertec GmbH, 25 mm diameter, 0.5 m radius of curvature, high
- 25 reflectivity 370 to 395 nm99.94%) separated by 80 cm, giving an effective path length of approximately 1.4 km. The mirrors were housed in bespoke mirror mounts (length = 11 cm) which were purged with nitrogen (0.9 L min⁻¹ L/min⁻¹ divided between the two mounts), and thus BBCEAS data were corrected for the cavity's length factor (LF = 1.37). The cavity was operated "open path", i.e. the portion of the cavity between the mirror mounts was open to the ambient atmosphere. This open path configuration has the advantage that there were no wall losses or heterogeneous production of HONO within the instrument.

In the field, $a\underline{A}$ Teflon tube was inserted between the mirror mounts in order to measure the reference spectrum of light transmitted when the cavity was purged with nitrogen. The reflectivity of the cavity mirrors (as a function of wavelength) was characterised in the laboratory before the campaign; their measured reflectivity peaked at 99.940 % at 387 nm. The reflectivity was verified in the field by measuring the 380 nm absorption band of O₄ when purging the cavity with pure oxygen, and measurements taken at the start and end of the campaign agreed to within 4.5 %.

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BBCEAS spectra were integrated for 20 s (the average of two 10 s acquisitions) using an OceanOptics QEPro spectrometer (instrument line shape = 0.20 nm HWHM). Spectra were fitted with reference absorption cross sections for HONO (Stutz et al., 2000), NO₂ (Vandaele et al., 1998) and O₄ (Hermans as given in the HITRAN database, Richard et al., 2012), and any remaining unfitted broadband absorption was attributed to extinction by ambient aerosol particles. Typical statistical errors for retrieving HONO and NO₂ concentrations from the spectral structure were ± 0.8 ppbv and ± 0.9 ppbv, (1σ in 20 s) respectively. The BBCEAS measurements are also affected by systematic uncertainties in the molecules' reference absorption cross sections (typically ± 3 %) and for determining the mirror reflectivity in the field (± 4.5 %). Extinction by ambient aerosol also reduced the effective path length of the BBCEAS measurement. HONO retrievals tended to be dominated by statistical (spectral fitting) errors, whereas retrievals of the much high ambient NO₂ concentrations were dominated by the systematic uncertainties in the cross sections and the mirror reflectivity. For the mean HONO (3 ppbv) and NO₂ (75 ppbv) amounts recorded during the

campaign, the total measurement uncertainties (from combining the statistical and systematic errors) were ± 1.2 ppbv for HONO and ± 5 ppbv for NO₂.

NO was measured by a commercial chemiluminescence NO_x analyser (Thermo Environmental Instruments Inc: Model 42C). The detection limit of the 42C was determined from 3 times the standard deviation of the measurement in zero air and was 20 calculated to be approximately 0.2 pbby for a 1 minute averaging period. The uncertainty of the instrument was estimated to be \pm 10 % from calibrations. The analyser is capable of measuring NO₂ and NO_x, however the instrument utilises a molybdenum converter to covert NO_2 to NO, an approach which is known to be influenced by positive interference from other NO_v species such as HONO, nitric acid (HNO₃), peroxyacetyl nitrate (PAN), and alkyl nitrates (e.g. Dunlea et al., 2007; Villena et al., 2012); therefore the final NO_x used in this study is calculated from the sum of the NO measured by the 42C and NO₂ by 25 the BBCEAS. The inlet for the 42C instrument was shared with a non-dispersive infrared instrument (LI-COR: Model LI-820) to measure CO₂. The LI-COR was calibrated with pure N_2 (0 ppmv CO₂) and 1500 ppmv CO₂ to determine the relative uncertainty (\pm 5 %) and precision (1.2 %, 2- σ). The LI-COR and 42C analysers were evaluated for drift from zero measurements taken before and after the measurement campaign. The drift for CO₂ and NO_x were determined to be 1.45 ppmv and 0.42 ppby, respectively. The drift for both instruments was less than 2 % of the minimum measured values in the tunnel 30 $(NO = 1.27 \%, CO_2 = 0.35 \%)$, therefore no correction was deemed necessary here. A mini-met station/anemometer (Kestrel 4500) was deployed near the top of the tunnel (3m above ground level and 1m from the road-sideroadside) to measure the

temperature, relative humidity and air flow in the tunnel.

3 Results and discussion

3.1. Data Overview

Figure 4 presents 15_-minute averaged time series for gases measured in the tunnel. <u>NO</u>, NO₂, <u>HONO</u> and CO₂ follow the same diurnal cycle <u>as each other</u> indicating that they have a similar source. There is a clear difference in weekday and weekend

5 diurnal cycles with peaks associated with rush hour traffic observed in NO_x, <u>NO₂</u>. HONO and CO₂ during the weekdays, which are not present at the weekend. <u>The measured NO_x</u>, <u>HONO and CO₂</u> concentrations are also lower at the weekend, <u>on average</u>. <u>On weekdays</u>, <u>T</u>the mixing ratios of all gas species <u>during weekdays</u> increase from around 06:00 (start of morning rush hour) and drop away in the afternoon. A large spike in the evening around 17:30 to – 18:00 is observed, which coincides with congestion in the southbound tunnel during peak evening rush hour. <u>The congestion is also evident by the lower mean vehicle</u>

10 speeds at 17:00 to 18:00 (Figure S2).

The mean weekday HONO mixing ratio during the daytime (06:00 to 19:59, calculated from the 15 minute averages) is 3.4 ± 1.0 ppbv (1- σ), which decreased to 2.4 ± 0.7 ppbv overnight. The maximum observed 15 minute average HONO level occurred during the evening rush hour, reaching 9.7 ppbv on 3rd August. Observed HONO levels in the current work are typically lower than in previous tunnel studies. Liang et al. (2017) measured a mean HONO level of 15.7 ± 4.2 ppbv during their study in

- 15 Hong Kong, whereas Kurtenbach et al. (2001) observed peak HONO levels of 45 ppbv in the daytime. Kirchstetter et al. (1999) only performed measurements between 16:00 to 18:00 in the Caldecott Tunnel, California and observed a mean HONO level of 6.9 ± 1.4 ppbv, higher than the mean level of 4.1 ± 1.4 ppbv for the same hours in this study. The lower levels measured in the current study may be the result of differences in vehicle fleet between studies as discussed further in Section 3.3, in combination with shorter tunnel length (Table 2) and shorter distance of the our sampling point into the tunnel.
- 20 The persistence of HONO overnight is unlikely to be due to vehicle exhaust emissions as traffic <u>flow</u> is low during this period, but rather from background ambient HONO entering at the tunnel's entrance and <u>from</u> heterogeneous formation of HONO on <u>particles deposited onto</u> the walls of the tunnel (Kurtenbach et al., 2001). This <u>also</u> suggests that vehicles were not the only source of HONO during the day, and the impact of heterogeneous HONO formation is considered in emission ratio calculations in Section 3.2.1.
- From late evening on 1st August until mid-morning on 2nd August the HONO mixing ratios are lower than otherwise observed for a weekday. Precipitation data from a nearby weather station (Figure S3) revealed that it rained continuously from 17:00 on 1st August to 16:00 on 2nd August with corresponding high relative humidity recorded inside the tunnel. It is likely that wet surfaces inside the tunnel, as a result of spray from tyres, resulted in a loss in HONO. The Henry's law constant (in water) of HONO ($k_H = 4.8 \times 10^{-1}$ mol m⁻³ Pa⁻¹) is approximately 4800 times greater than that of NO₂ ($k_H = 9.9 \times 10^{-5}$ mol m⁻³ Pa⁻¹)

(Sander, 2015), therefore HONO is likely to be washed out more rapidly than NO_2 on wet surfaces, resulting in a <u>deviation</u> reduction in the HONO/NO_x ratios. As a result, this precipitation event was excluded from the final dataset for calculation of emission ratios.

3.2 Relative HONO emission ratios

5 To determine a HONO/NO_x emission ratio representative of direct exhaust emissions, corrections for both heterogeneous formation of HONO from NO₂ and background HONO levels were considered.

3.2.1 Heterogeneous HONO formation from NO₂

A number of heterogeneous reactions resulting in the formation of HONO have been proposed in the literature, however, the formation of HONO from NO_2 on humid surfaces (R3) is thought to be the main pathway (Spataro and Ianniello, 2014 and references therein).

$$2 \operatorname{NO}_{2(g)} + \operatorname{H}_2O_{(ads)} \rightarrow \operatorname{HONO}_{(g)} + \operatorname{HNO}_{3(g)}$$
(R3)

The rate of reaction coefficient for R3 (k_{het}) is dependent on the geometric uptake coefficient of NO₂ on the tunnel walls (γ_{geo}) and <u>is</u> given by:

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$$k_{het} = \frac{1}{8} \bar{\nu}_{NO2} \frac{s}{v} \gamma_{geo}, \tag{Eq. 1}$$

where $\bar{\nu}_{NO2}$ is the mean molecular velocity of NO₂ and *S/V* is the surface-to-volume ratio of the tunnel (Kurtenbach et al., 2001). Kurtenbach et al. (2001) performed laboratory experiments to measure HONO generated on tunnel wall residue to directly calculate γ_{geo} and determined $k_{het} = 3 \times 10^{-3} \text{ min}^{-1}$.

The rate of formation of HONO from NO₂ can then be calculated via Eq. 2.

$$\frac{d[HONO]}{dt} = -\frac{1}{2} \frac{d[NO_2]}{dt} = k_{het} [NO_2]$$
(Eq. 2)

where *dt* is the residence time of the gases in the tunnel from the entrance to the sampling point.

Kurtenbach et al. (2001) calculated that the heterogeneous conversion on the tunnel wall contributed to approximately $13_{\%}$ of the measured HONO during the day and up to $80_{\%}$ at night (for a maximum NO₂ mixing ratio of 250 ppbv). This result is in contrast to measurements obtained in the Caldecott Tunnel by Kirchstetter et al. (1996) who showed that the HONO/NO_x ratio did not change between the middle of the tunnel and the tunnel exit, suggesting that there was no significant formation of HONO (or deposition of HONO) on the tunnel walls in their study.

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Liang et al. (2017) determined $k_{het} = 1.31 \times 10^{-3} \text{ min}^{-1}$ in the Shing Mun tunnel, Hong Kong from using an upper limit of $\gamma_{geo} = 10^{-6}$ as calculated by Kurtenbach et al. (2001). Using wind speed data in the tunnel to determine the residence time, Liang et al. found on average that the contribution of HONO/NO_x from heterogeneous reactions on the tunnel walls alone was 0.04_% on average. Since this value was less than the error when calculating HONO/NO_x from direct emissions, the authors did not perform any corrections to the their final HONO/NO_x ratio. As there is no consensus in the literature to about the relative contribution from tunnel walls to measured HONO, we calculated the contribution from heterogeneous HONO formation in the Queensway tunnel during the measurement period. Using Eq. 1, a geometric uptake coefficient for NO₂ of $\gamma_{geo} = 10^{-6}$ and the surface-to-volume ratio of the Queensway tunnel (0.64 m⁻¹), we calculated k_{het} to be 1.9×10^{-3} min⁻¹.

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The Kestrel anemometer used in this study logged data every 20 minutes, and as a result, the wind speed measurements inside the tunnel only provide a "snap-shot" of the data and are dependent on the traffic flow. Care should be taken when using wind speed measured by an anemometer to calculate the air²-s residence time in a tunnel. Rogak et al. (1998) compared residence times in a tunnel calculated using an SF₆ tracer and anemometer data and found that the wind speed in the tunnel measured by the anemometer overestimates volumetric flow during high wind speeds and underestimates when wind speed is low, requiring

20 the anemometer overestimates volumetric flow during high wind speeds and underestimates when wind speed is low, requiring a correction factor for the determination of emission factors in the tunnel. Tracer measurements were not performed during our study, consequently a modelling approach was used to correct the wind speed data from the anemometer. As outlined below the wall source contribution to the observed HONO was small (~5_% on average) during the daytime periods used to derive our final results.

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"True" wind speeds were inferred from computational fluid dynamics (CFD) modelling of CO_2 profiles measured along the tunnel. After the campaign, the LI-COR CO_2 instrument was installed into a van and driven repeatedly through the southbound bore of the tunnel at four different times of the day (late morning, late afternoon, evening and at night). Data recorded when the van was inside the tunnel were extracted from the CO_2 time series. The timestamps of these data points were converted into distance along the tunnel using the speed of the van and the time that it entered the tunnel, as recorded by a camera mounted on the van's dashboard. The CO_2 data were then averaged into 50 m bins along the tunnel to produce a profile of the CO_2 concentration increasing with distance into the tunnel. The CO_2 profiles were simulated by a CFD model which took as its inputs the tunnel's physical dimensions, the anemometer wind speed, the ANPR traffic counts and traffic composition, and emission factors from the DEFRA Emissions Factor Toolkit version (EFT v8.0.1) (DEFRA, 2017). Further information on the DEFRA EFT is provided in the supplementary material. The wind speeds that produced the optimum fits between modelled and measured CO₂ profiles are shown as the crosses in Figure S4. These wind speeds match closely with the anemometer wind speeds increased by a factor of 3.0 and they correctly capture the day and night-time differences. The optimum CFD-inferred wind speeds at 10:00, 16:00 and 20:00 (3.4 to 3.9 m s⁻¹) are also in agreement with the 3.6 m s⁻¹_wind speed (06:00 to 23:59) in the tunnel study by Liang et al. (2017); and consistent with the empirical correction factor produced by Rogak et al. (1998) of ×3.5 for an anemometer-measured wind speed of 1.5 m s⁻¹ applicable during much of the daytime. The results of CFD modelling of CO₂, NO₂ and HONO profiles measured in the Queensway tunnel will be published in a

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separate paper.

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During the night, the measured wind speed was often below the minimum speed required to turn the anemometer's impeller (< 0.4 m s⁻¹) so the residence time of the air parcel overnight is difficult to determine directly. As a result, we focus here on the heterogeneous formation of HONO in the tunnel during the daytime (see comments below, in Section 3.2.3, regarding use of daytime only data to infer the fleet average emission ratios). Between 06:00 and 19:59 the mean wind speed in the tunnel 15 was 3.89 m s⁻¹, giving a residence time of air in the tunnel up to it reaching the sampling point of 1.86 minutes. If we assume the mean residence time of NO₂ emissions in the tunnel is half of this residence time (Liang et al., 2017; Pierson et al., 1978), the HONO produced from heterogeneous formation on the tunnel walls is 0.18 ppbv for a mean daily weekday NO₂ mixing ratio of 98 ppbv, which is approximately 5_% of the mean measured HONO in-during the day. Although the mean heterogeneous contribution to HONO during the daytime is small, the heterogeneous contribution is higher (8_%) when the 20 tunnel becomes congested and the residence time of the air in the tunnel increases. Therefore, the final HONO data used in this study was-were corrected for the heterogeneous HONO contribution using measured NO₂ and the modelled wind speed, to better represent the direct HONO emissions in the tunnel.

25 3.2.2 Background corrections for HONO, NO_x and CO₂

As vehicles travel through the tunnel, cleaner air from outside is drawn into the tunnel diluting the vehicle emissions, known as the piston effect. To obtain direct vehicle emission measurements it is necessary to correct for the this dilution by subtracting ambient background levels from the concentrations measured in inside the tunnel. As no concurrent measurements were available at the tunnel entrance during this campaign, ambient NO_x , HONO and CO_2 data from nearby <u>air monitoring</u> stations

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were used to correct for the background levels. Hourly averaged background NO_x mixing ratios during the measurement period were taken from Acocks Green (AG), an Automatic Urban Rural Network (AURN) site located 6.9 km south east of the

Queensway Tunnel (http://uk-air. defra.gov.uk/). The AURN station does not measure HONO or CO₂, so data previously taken at the Birmingham Air Quality Supersite (BAQS) on the University of Birmingham campus ($52^{\circ} 27' 1"$ N, $1^{\circ} 55' 30"$ W) 3 km south west of the tunnel was used in this study. As CO₂ is well mixed within the troposphere, variability in background CO₂ mixing ratios levels-is primarily the result of a changing boundary layer height. Therefore, in this study, an average diurnal CO₂ cycle measured at BAQS was used to correct for background <u>CO₂</u>. The overnight background corrected CO₂ (Δ CO₂ = CO_{2.tunnel} – CO_{2.bkg}) levels dropped to near zero (see panel C of Figure 5) suggesting that the procedure was suitable. HONO on the other hand has a much shorter lifetime than CO₂ and HONO mixing ratios vary depending on local sources and <u>the</u> availability of NO₂ (Crilley et al., 2016). An average diurnal HONO/NO_x cycle (Figure S5) was calculated from measurements taken at BAQS between 18th March and 1st April, 2015 (Singh, 2017) and applied to the background NO_x from Acocks Green to determine hourly background HONO mixing ratios (Eq 3).

$$[HONO]_{bkg} = [NOx]_{bkg(AG)} \times \left(\frac{[HONO]}{[NOx]}\right)_{BAQS}$$
(Eq. 3)

Figure 5 shows the hourly time series of NO_x, HONO and CO₂ measured in the tunnel and the corresponding data corrected for background mixing ratios. The results show that mixing ratios measured in the tunnel were, on average, 73 and 44 times higher than the background NO_x and HONO, respectively, suggesting that direct emissions from vehicles were indeed the dominant source of these gases. Consequently, any uncertainties in the background HONO mixing ratios that were subtracted from the measured HONO have little impact on the final results. In the following sections, the final HONO dataset (Δ HONO) is corrected for both background HONO levels and the heterogeneous HONO contribution.

20 3.2.3 HONO emission ratios

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To determine the HONO to NO_x emission ratio, a reduced-major-axis (RMA) regression (Figure 6) was performed using the weekday hourly averaged data from 06:00 to 19:00, i.e. when there was minimal contribution from heterogeneous HONO sources and the traffic flow was high (> 1500 vehicles per hour). The slope of the regression gives an average emission ratio of Δ HONO/ Δ NO_x = 0.85-% (95% confidence interval = 0.72 to -1.01-%) for this time period.

Figure 7a shows that the Δ HONO/ Δ NO_x emission ratio varies during the day, from 0.66^w at 07:00 to 1.10^w at 19:00. In this section, we explore the relationship between Δ HONO/ Δ NO_x and changes in vehicle fleet composition using information on fuel and vehicle type from the ANPR dataset. Figure 7c shows the fraction of diesel and non-diesel (petrol, LPG and electric) vehicles travelling through the Queensway tunnel during an average weekday. Between 06:00 and 17:00, the Δ HONO/ Δ NO_x ratio appears to vary closely with changes in the fraction of diesel vehicles, with both variables peaking at 10:00 when the

 Δ HONO/ Δ NO_x emission ratio reaches a value of 0.91_% and the percentage of diesel fuelled vehicles in the fleet reaches a maximum of 66_%.

The Δ HONO/ Δ CO₂ ratio (Figure 7b and Table S2) also follows the change in fleet composition, with a higher ratio observed at 10:00 (3.3_%) compared to 17:00 (1.5_%). However this may also be the result of higher CO₂ emissions from petrol fuelled vehicles compared to diesel vehicles; for example, a study of 149 passengers cars showed that CO₂ emissions were 13<u>to</u>– 66_% higher for petrol vehicles than diesel (O'Driscoll et al., 2018). Modelled emission factors from the DEFRA Emissions Factors Toolkit (EFT v8.0.1) indicate a maximum in the CO₂ emissions (mg veh⁻¹ km⁻¹) at 17:00 (Figure S6), in line with the 46_% maximum in the non-diesel vehicle fraction at evening rush hour.

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From 17:00 to 19:00 the relationship between the Δ HONO ratios and fuel type is less clear. During this period the fraction of diesel vehicles remains almost constant, however, the Δ HONO/ Δ NO_x and Δ HONO/ Δ CO₂ emission ratios both sharply increase. Analysis of the modelled NO_x and CO₂ emission factors from the EFT (Figure S6) indicate a decrease in both NO_x and CO₂ emission factors from 17:00 to 19:00, which is likely to be related to the total vehicle flow and <u>lower</u> vehicle speed

15 <u>at times of peak congestion</u> (because the fleet composition remains very similar). The EFT does not have the capability to determine a HONO emission factor, therefore information on HONO emission variability with speed and vehicle flow is not available <u>*a priori*</u>. However, assuming the HONO emissions do not reduce by the same percentage after 17:00, this would lead to the increase in the observed $\Delta HONO/\Delta NO_x$ and $\Delta HONO/\Delta CO_2$ ratios. As no conclusion can be made at this point, without additional information, we focus the analysis in the following section on the data between 06:00 and 17:00 <u>only</u>.

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Ho et al. (2007) used equation 4 to determine emission factors (mg veh⁻¹ km⁻¹) for diesel and non-diesel fuelled vehicles based on a method described by Pierson et al. (Pierson et al., 1996).

$$EF = (EF_{DV} - EF_{NDV})x + EF_{NDV}$$
(Eq. 4)

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where, *x* is the fraction of diesel-fuelled vehicles, EF_{DV} is the emission factor for diesel vehicles, EF_{NDV} the emission factor for non-diesel vehicles and EF is the emission factor for the mixed fleet. In this format, a linear regression of *x* versus EF, based on equation 4, gives EF_{DV} at x = 1 and EF_{NDV} at x = 0.

30 In this study, instead of calculating emission factors, we investigated <u>application-the applicability</u> of equation 4 to determine Δ HONO/ Δ NO_x emission ratios for diesel (ER_{DV}) and non-diesel (ER_{NDV}) vehicles (see Figure S7). <u>Extrapolating the best fit</u> line in Figure S7 to *x* = 1 gives an plausible emission ratio for diesel vehicles of ER_{DV} = 1.35 ± 0.50 %. <u>However</u>Unfortunately, this method resulted in an unrealistic small negative emission ratio for non-diesel vehicles (ER_{NDV} = -0.0404 ± 0.26 %). This

method may not be appropriate for the current work because the range in the fuel fraction of vehicles is small, thus extrapolating to x = 0 and x = 1 resulted in very large uncertainties.

An alternative approach is to determine emission ratios using a pair of simultaneous equations based on the fraction of diesel 5 vehicles and $\Delta HONO/\Delta NO_x$ values for different hours of the day. For example, using the data as given in Table S2, the equation at 17:00 is given by:

$$\frac{\Delta HONO}{\Delta NOx} = 0.73 \% = 0.54 \times ER_{DV} + 0.46 \times ER_{NDV}$$
(Eq. 5)

- 10 Using the another equation for a different hour (i.e. different set of fractions), the values for ER_{DV} and ER_{NDV} can be determined. Care must be taken when interpreting these results, however, as the calculated ERs depend on the selection of pairs of measurement points and some pairs from our data resulted in negative ER values. Thus average ERs were determined using many pairs of simultaneous equations. Here we calculated the average of ten ERs, using one set of fractions at 17:00 and a second set for each hour from 06:00 to 15:00. The data from 16:00 was not used here as there was no change in the diesel
- 15 fraction between 16:00 and 17:00. -The mean $(\pm 1\sigma)$ for ER_{DV} and ER_{NDV} are $(1.66-04 \pm 20.247)$ -% and $(0.4-37 \pm 0.55)$ -%, respectively, suggesting that diesel fuelled vehicles do have higher Δ HONO/ Δ NO_x ratios.

It has previously been suggested that higher HONO/NO_x ratios are observed when the tunnel fleet contains a greater number of heavy duty vehicles (Trinh et al., 2017). A similar calculation to that outlined in the previous paragraph was performed <u>here</u> to determine the average emission ratio for cars and for heavy duty vehicles (goods vehicles and buses) by apportioning the observed Δ HONO/ Δ NO_x ratios between the car and heavy duty vehicle numbers given in Table S2. The Δ HONO/ Δ NO_x ratio emission ratio for heavy duty vehicles (ER_{HD}) is estimated to be 1.<u>328</u> ± 0.6<u>34</u>_%, approximately 1.9 times higher than the ratio determined for cars (ER_{CAR} = 0.69 ± 0.05_%). The impact of heavy duty vehicles on the HONO/NO_x emission ratio is discussed further in <u>section-Section</u> 3.3.

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It should be noted that the methods described above can only provide an estimate of the emission ratios for different fuel and vehicle types. In our study, the variability in the fraction of diesel vehicles and heavy duty vehicles across the day was small, therefore extracting emission ratios for individual vehicle types is challenging. To obtain more precise emission ratios for different engine types, a larger dataset (i.e. longer time series) and <u>fully</u>-contemporaneous ANPR data would be required.

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3.3 Comparison of HONO/NOx emission ratios across tunnel studies

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Table 2 shows a comparison of measured HONO/NO_x emission ratios <u>for from</u> studies performed in road tunnels, along with summaries of their reported vehicle fleets. The lowest HONO/NO_x ratio ($0.29_{\%}$) was measured in the Caldecott tunnel, California (Kirchstetter et al., 1996). As 99_% of the fleet was comprised of petrol fuelled vehicles, the low HONO/NO_x ratio

- 5 is expected because the analysis in <u>section_Section_3.2.3 above</u> and <u>previous_dynamometer studies have <u>both_demonstrated</u> that petrol vehicles typically emit less HONO than diesel. On the other hand, Liang et al. (2017) observed a HONO/NO_x emission ratio of 1.24 ± 0.35 -% in the Shing Mun Tunnel, Hong Kong in 2015, approximately 1.4 times higher than in the current work despite the Queensway Tunnel having a higher proportion of diesel vehicles compared to the Shing Mun Tunnel (59_% and 38_% respectively). The differences observed in the HONO/NO_x emission ratio between the Shing Mun and</u>
- Queensway tunnel studies may <u>therefore</u> be related to 1) the percentage of heavy duty & goods vehicles within the fleet, and
 2) <u>exhaust</u> after_-treatment technologies that affect NO₂ emitted directly from vehicle exhausts (known as primary NO₂).

In addition to these two points, which are discussed further below, it should also be noted that the fleet in the Shing Mun Tunnel consisted of 15_% liquefied petroleum gas (LPG) fuelled vehicles. On road sampling of emissions from buses fuelled by diesel and LPG in Hong Kong have indicated that LPG vehicles have lower NO_x emissions when compared to diesel (Ning et al., 2012). However, as far as we are aware there have been no published studies investigating HONO emissions from LPG fuelled vehicles, and therefore no conclusions can be made at this point regarding the impact of the LPG fleet on the observed differences in HONO/NO_x ratios.

- Liang et al. (2017) investigated the impact of diesel particle filters (DPFs) on HONO vehicle emissions and found that HONO emissions ratios were higher in vehicles equipped with DPFs (based on a higher ΔNO₂/ΔNO_x ratio) and suggest that this may be due to HONO formation from heterogeneous reactions involving NO₂ and black carbon within the filter. The diesel oxidation catalysts (DOCs) oxidise hydrocarbons and CO in excess oxygen over platinum/palladium catalysts; howeverhowever, NO can also be oxidized to NO₂ during this process. DPFs were mandatory for all new diesel vehicles from 2009 (Euro Class V) and operate by trapping soot in the exhaust where it is then oxidized at high temperatures. To ensure the filters do not become blocked from engines that run at lower temperatures, many DPFs use catalysts to convert NO to NO₂ to periodically oxidise the soot (He et al., 2015; Kim et al., 2010). Consequently, DOCs and DPFs both result in an increase in NO₂ emitted from exhausts, which may also lead to an increase in HONO emissions.
- 30 Liang et al. (2017) estimated that half of the diesel fleet in their study were buses and goods vehicles (Euro Class IV and above) equipped with diesel particulate filters. In the current work less than a quarter of diesel fuelled vehicles were goods vehicles or buses, and of those only 58_% were Euro Class V and above (i.e. with DPFs fitted). As a result, the percentage of diesel vehicles emitting high NO_x and HONO levels is expected to be much lower in the Queensway tunnel than the Shing

Mun tunnel, and this likely accounts for the lower HONO/NO_x emission ratio we observed. In contrast in the Kiesbergtunnel (Kurtenbach et al., 2001), medium and heavy-duty vehicles represented 12_% of the fleet (6_% heavy-duty trucks, 6_% commercial vans), similar to the fleet in the Queensway tunnel, which may explain why the observed HONO/NO_x emission ratios were so similar (0.85_% and 0.8_%, for the Queensway tunnel and Kiesbergtunnel studies, respectively). It should be noted that the Kiesbergtunnel fleet were all pre-Euro III standard, so higher emissions of NO_x are expected compared to later Euro standard vehicle classes. However, few of these vehicles in the Kiesbergtunnel would have been fitted with DOCs or

Recent trend analysis studies have shown that primary NO₂ has been decreasing over the last decade in the UK (Carslaw et al., 2016; Matthaios et al., 2019) and across Europe (Grange et al., 2017). For example, the mean NO₂/NO_x emission ratio 10 measured from ambient roadside monitoring sites in inner-London has decreased from a peak value of 25 % in 2009 to 15 % in 2014 (Carslaw et al., 2016). Although the exact cause of the NO₂ reduction is not clear, it is thought that the introduction of after-treatment technologies, such as selective catalytic reduction (SCR) installed on buses and heavy duty vehicles, and a reduction in the use of platinum group metals in catalysts may have contribute contributed to the observed decrease (Carslaw et al., 2016; Grange et al., 2017). With a reduction in NO_2 emissions from diesel vehicles overall, a decrease in HONO may also be expected in the UK. In contrast, Wang et al., (2018) reported an increase in the NO₂/NO_x ratio at the outlet of the Shing 15 Mun tunnel, from 9.5 % in 2003 to 16.3 % in 2015. Therefore, the higher HONO/NO_x ratio observed in Hong Kong (Liang et al., 2017) compared to our current work may also be related to differences in primary NO_2 emissions. It should be noted that in the study by Wang et al., NO_2 was measured using a standard chemiluminescence monitor, which typically use a molybdenum converter. Molybdenum converters also convert other NO_v species, such as HONO, resulting in an 20 overestimation of the reported NO₂. However, as the HONO/NO_x ratio was only 1.24_.%, it is unlikely that HONO has contributed to the factor of 1.7 increase in primary NO₂ recorded near the Shin Mung Tunnel.

4 Impact of HONO emissions from vehicles on total HONO measured in Birmingham.

DPFs, so this may have offset the amount of NO₂ (and thus HONO) emitted.

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Previous work has shown that ambient HONO has large heterogeneity when sampling sampled close to roads (Crilley et al., 2016). Here we use the mean HONO/NO_x emission factor of 0.85_%, determined from the tunnel measurements in section
Section 3.2.3, to investigate the contribution of direct HONO vehicle emissions to ambient HONO levels in urban and suburban areas. Measurements of HONO and NO_x were taken in a mobile laboratory on a return journey between Birmingham and Leicester, two cities in the UK Midlands separated by 55 km (straight line measurement from city centres). NO and NO₂ were measured every 60 s by a our chemiluminescence analyser fitted with a molybdenum NO₂ converter (Thermo 42i-TL). As discussed in section 2.3, molybdenum converters are known to result in interferences from NO_y species, therefore the NO_x measurements presented here represent an upper limit on NO + NO₂ concentrations. HONO was measured every 5

minutes with a Long-Path Absorption Photometer (LOPAP, Model 03, QUMA). Measurement uncertainties are primarily the

result of uncertainties in the calibrations and are estimated to be $5_\%$ and $10_\%$, for NO_x and HONO, respectively. Further details regarding the measurement techniques and the driving route can be found in Crilley et al. (2016).

Figure 8 shows measured HONO and NO_x mixing ratios measured during the transect from Birmingham to Leicester and the return journey; on 23 October 2015. Also shown is the directly emitted vehicular HONO (HONO_{veh}) calculated from the mean ΔHONO/ΔNO_x emission ratio (0.85_%). For the two transects, HONO_{veh} contributed on average 66_% to the total measured HONO. The highest contribution was observed on the M6 motorway, where HONO_{veh} typically accounted for 86_% of the measured HONO. During the motorway segments of the journeys, the mobile laboratory was operating in "chase mode", i.e. deliberately sampling plumes from a single vehicle ahead. Therefore, it was likely the high HONO_{veh}/HONO ratios observed

- 10 on the motorway were the result of predominantly sampling vehicle emission plumes with low background contribution. The lowest HONO_{veh} contribution was observed around the Birmingham University campus (24_%), which is expected as traffic density was low in this area. Whilst travelling through Birmingham city centre, HONO_{veh}/HONO increased to 70_%, indicating that vehicle exhaust emissions were the dominant source of HONO in this area. This value is higher than determined by Tong et al. (2016) in Beijing (48 %); however, in our study we were sampling directly on the road, whereas the site in Beijing was
- 15 located in a building at the Institute of Chemistry, Chinese Academy of Sciences, therefore significantly less influenced by direct traffic sources. This estimate neglects the differing atmospheric lifetimes of HONO and NO_x; the longer lifetime of the latter implies that HONO/NO_x ratios will fall as the <u>airmass-air mass</u> ages. Accordingly, estimates of the vehicular contribution derived from measurements made as close as possible to the emission source, i.e. on the roadway, as described here, will give the <u>lowest-highest</u> estimate of the relative importance of vehicle emissions.

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Overall direct vehicle emissions can be an important source of HONO in urban areas, particularly near roads with very high traffic density. Further investigation in these areas is paramount to fully understand the impact of OH produced from HONO photolysis on chemical processing of gases in the overlying atmosphere.

5 Summary

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Measurements of HONO, NO_x and CO₂ were performed in a city centre road tunnel in Birmingham, UK₂ for two weeks in July<u>and</u> August 2016; to investigate direct HONO emissions from vehicle exhausts under real-world driving conditions. HONO mixing ratios peaked when NO_x and CO₂ peaked during traffic congestion in the weekday evening rush hours (17:30 to 18:00), confirming that vehicle exhausts were the dominant source of HONO in the tunnel.

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A HONO/NO_x emission ratio of 0.85_% (0.72-<u>to</u> 1.01_% <u>at</u>, 95_% Clconfidence interval) was determined for an average weekday fleet comprising 59_% diesel fuelled vehicles. Our value is similar to the ratio of 0.8_% which is typically used in

modelling studies and which was determined for a predominately $petrol_-fuelled$ fleet <u>over</u> 20 years ago in Germany. The results show that despite an increase in the <u>number-proportion</u> of diesel fuelled vehicles over the past two decades in Europe, and updated emissions control technologies, the HONO/NO_x emission ratio has not varied significantly. A comparison with a tunnel study in Hong Kong suggested that the HONO/NO_x ratio may be less dependent on the percentage of diesel vehicles

5 but rather the percentage of large goods vehicles within the fleet, and the after-treatment technologies implemented on those vehicles.

The HONO/NO_x emission ratio determined in this study was used to investigate the contribution of vehicle exhaust HONO to ambient HONO in Birmingham. The results show that direct vehicles emissions contribute up to 70_% of the total measured
HONO in the city centre. As direct HONO emissions were found to be the dominant source where traffic density is high, it is important to obtain fuel-based emission ratios which also take into account after-treatment technologies, to ensure models can accurately simulate daytime OH radical production rates.

In this study the focus has been primarily on <u>HONO emissions from</u> diesel and petrol_-fuelled vehicle <u>HONO emissionss</u>.
However, in countries where alternative fuels such as liquefied petroleum gas (LPG) and compressed natural gas (CNG) are becoming more prevalent, in particular in the public transport sector, further investigation into HONO emissions from these fuel types is needed.

Author contribution

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Study conceived by WB, FP and SB. Measurements were performed by LK, LC, TA, SB and FP. Formal analysis performed by LK, TA and SB. LK prepared the manuscript with contributions from all co-authors.

Competing Interests

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The authors declare that they have no conflict of interest.

Data Availability

Hourly averaged fuel type, vehicle type and emission ratios are available in the Supplementary Information. The 15 minute 30 dataset from the tunnel measurements is available from the authors on request.

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5 References

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Alicke, B., Platt, U. and Stutz, J.: Impact of nitrous acid photolysis on the total hydroxyl radical budget during the Limitation of Oxidant Production/Pianura Padana Produzione di Ozono study in Milan, J. Geophys. Res. Atmos., 107(22), doi:10.1029/2000JD000075, 2002.

Ammann, M., Kalberer, M., Jost, D. T., Tobler, L., Rössler, E., Piguet, D., Gäggeler, H. W. and Baltensperger, U.: Heterogeneous production of nitrous acid on soot in polluted air masses, Nature, 395(6698), 157–160, doi:10.1038/25965, 1998.

10 Arens, F., Gutzwiller, L., Baltensperger, U., Gäggeler, H. W. and Ammann, M.: Heterogeneous Reaction of NO₂ Diesel Soot Particles, Environ. Sci. Technol., 35(11), 2191–2199, doi:10.1021/es000207s, 2001.

Aubin, D. G. and Abbatt, J. P. D.: Interaction of NO₂ with hydrocarbon soot: Focus on HONO yield, surface modification, and mechanism, J. Phys. Chem. A, 111(28), 6263–6273, doi:10.1021/jp068884h, 2007.

Bohnenstengel, S. I., Belcher, S. E., Aiken, A., Allan, J. D., Allen, G., Bacak, A., Bannan, T. J., Barlow, J. F., Beddows, D. C. S., Bloss, W.

- 15 J., Booth, A. M., Chemel, C., Coceal, O., Di Marco, C. F., Dubey, M. K., Faloon, K. H., Flemming, Z. L., Furger, M., Gietl, J. K., Graves, R. R., Green, D. C., Grimmond, C. S. B., Halios, C. H., Hamilton, J. F., Harrison, R. M., Heal, M. R., Heard, D. E., Helfter, C., Herndon, S. C., Holmes, R. E., Hopkins, J. R., Jones, A. M., Kelly, F. J., Kotthaus, S., Langford, B., Lee, J. D., Leigh, R. J., Lewis, A. C., Lidster, R. T., Lopez-Hilfiker, F. D., McQuaid, J. B., Mohr, C., Monks, P. S., Nemitz, E., Ng, N. L., Percival, C. J., Prévôt, A. S. H., Ricketts, H. M. A., Sokhi, R., Stone, D., Thornton, J. A., Tremper, A. H., Valach, A. C., Visser, S., Whalley, L. K., Williams, L. R., Xu, L., Young, D. E. and Zotter, P.: Meteorology, air quality, and health in London: The ClearfLo project, Bull. Am. Meteorol. Soc., 96(5), 779–804.
- 20 Zotter, P.: Meteorology, air quality, and health in London: The ClearfLo project, Bull. Am. Meteorol. Soc., 96(5), 7/9–804, doi:10.1175/BAMS-D-12-00245.1, 2015.

Calvert, J. G., Yarwood, G. and Dunker, A. M.: An evaluation of the mechanism of nitrous acid formation in the urban atmosphere, Res. Chem. Intermed., 20(3–5), 463–502, doi:10.1163/156856794X00423, 1994.

Carslaw, D. C., Murrells, T. P., Andersson, J. and Keenan, M.: Have vehicle emissions of primary NO₂ peaked?, Faraday Discuss., 189(0), 439–454, doi:10.1039/C5FD00162E, 2016.

Crilley, L. R., Kramer, L., Pope, F. D., Whalley, L. K., Cryer, D. R., Heard, D. E., Lee, J. D., Reed, C. and Bloss, W. J.: On the interpretation of in situ HONO observations via photochemical steady state, Faraday Discuss., 189, doi:10.1039/c5fd00224a, 2016.

DEFRA: DEFRA, 2017: Emission Factors Toolkit for Vehicle Emissions., [online] Available from: https://laqm.defra.gov.uk/review-and-assessment/tools/emissions-factors-toolkit.html (Accessed 14 July 2019), 2017.

30 DfT: Vehicle licensing statistics: 2016, [online] Available from: https://www.gov.uk/government/statistics/vehicle-licensing-statistics-2016 (Accessed 26 April 2019), 2017.

Dunlea, E. J., Herndon, S. C., Nelson, D. D., Volkamer, R. M., San Martini, F., Sheehy, P. M., Zahniser, M. S., Shorter, J. H., Wormhoudt, J. C., Lamb, B. K., Allwine, E. J., Gaffney, J. S., Marley, N. A., Grutter, M., Marquez, C., Blanco, S., Cardenas, B., Retama, A., Yillegas, C. R. R., Kolb, C. E., Molina, L. T. and Molina, M. J.: Evaluation of nitrogen dioxide chemiluminescence monitors in a polluted urban environment, Atmos. Chem. Phys., 7(10), 2691–2704, doi:10.5194/acp-7-2691-2007, 2007.

Elshorbany, Y. F., Kurtenbach, R., Wiesen, P., Lissi, E., Rubio, M., Villena, G., Gramsch, E., Rickard, A. R., Pilling, M. J. and Kleffmann, J.: Oxidation capacity of the city air of Santiago, Chile, Atmos. Chem. Phys., 9(6), 2257–2273, doi:10.5194/acp-9-2257-2009, 2009.

Finlayson-Pitts, B. J., Wingen, L. M., Sumner, A. L., Syomin, D. and Ramazan, K. A.: The heterogeneous hydrolysis of NO2 in laboratory

systems and in outdoor and indoor atmospheres: An integrated mechanism, Phys. Chem. Chem. Phys., 5(2), 223–242, doi:10.1039/b208564j, 2003.

George, C., Strekowski, R. S., Kleffmann, J., Stemmler, K. and Ammann, M.: Photoenhanced uptake of gaseous NO₂ on solid organic compounds: A photochemical source of HONO?, Faraday Discuss., 130(2), 195–210, doi:10.1039/b417888m, 2005.

5 Gerecke, A., Thielmann, A., Gutzwiller, L. and Rossi, M. J.: The chemical kinetics of HONO formation resulting from heterogeneous interaction of NO₂ with flame soot, Geophys. Res. Lett., 25(13), 2453–2456, doi:10.1029/98GL01796, 1998.

Grange, S. K., Lewis, A. C., Moller, S. J. and Carslaw, D. C.: Lower vehicular primary emissions of NO₂ in Europe than assumed in policy projections, Nat. Geosci., 10(12), 914–918, doi:10.1038/s41561-017-0009-0, 2017.

Guan, C., Li, X., Zhang, W. and Huang, Z.: Identification of nitration products during heterogeneous reaction of NO₂ on soot in the dark and under simulated sunlight, J. Phys. Chem. A, 121(2), 482–492, doi:10.1021/acs.jpca.6b08982, 2017.

He, C., Li, J., Ma, Z., Tan, J. and Zhao, L.: High NO₂/NO_x emissions downstream of the catalytic diesel particulate filter: An influencing factor study, J. Environ. Sci. (China), 35(2), 55-61, doi:10.1016/j.jes.2015.02.009, 2015.

Ho, K. F., Sai Hang Ho, S., Cheng, Y., Lee, S. C. and Zhen Yu, J.: Real-world emission factors of fifteen carbonyl compounds measured in a Hong Kong tunnel, Atmos. Environ., 41(8), 1747–1758, doi:10.1016/j.atmosenv.2006.10.027, 2007.

15 Huang, R. J., Yang, L., Cao, J., Wang, Q., Tie, X., Ho, K. F., Shen, Z., Zhang, R., Li, G., Zhu, C., Zhang, N., Dai, W., Zhou, J., Liu, S., Chen, Y., Chen, J. and O'Dowd, C. D.: Concentration and sources of atmospheric nitrous acid (HONO) at an urban site in Western China, Sci. Total Environ., 593–594, 165–172, doi:10.1016/j.scitotenv.2017.02.166, 2017.

Jenkin, M. E., Cox, R. A. and Williams, D. J.: Laboratory studies of the kinetics of formation of nitrous acid from the thermal reaction of nitrogen dioxide and water vapour, Atmos. Environ., 22(3), 487–498, doi:10.1016/0004-6981(88)90194-1, 1988.

20 Kalberer, M., Ammann, M., Arens, F., Gäggeler, H. W. and Baltensperger, U.: Heterogeneous formation of nitrous acid (HONO) on soot aerosol particles, J. Geophys. Res. Atmos., 104(D11), 13825–13832, doi:10.1029/1999JD900141, 1999.

Khalizov, A. F., Cruz-Quinones, M. and Zhang, R.: Heterogeneous reaction of NO₂ on fresh and coated soot surfaces, J. Phys. Chem. A, 114(28), 7516–7524, doi:10.1021/jp1021938, 2010.

Kim, J. H., Kim, M. Y. and Kim, H. G.: NO₂-assisted sort regeneration behavior in a diesel particulate filter with heavy-duty diesel exhaust gases, Numer. Heat Transf. Part A Appl., 58(9), 725–739, doi:10.1080/10407782.2010.523293, 2010.

Kirchstetter, T. W., Harley, R. A. and Littlejohn, D.: Measurement of Nitrous Acid in Motor Vehicle Exhaust, Environ. Sci. Technol., 30(9), 2843–2849, doi:10.1021/es960135y, 1996.

Kirchstetter, T. W., Harley, R. A., Kreisberg, N. M., Stolzenburg, M. R. and Hering, S. V.: On-road measurement of fine particle and nitrogen oxide emissions from light- and heavy-duty motor vehicles, Atmos. Environ., 33(18), 2955–2968, doi:10.1016/S1352-2310(99)00089-8, 1999.

30

Kleffmann, J.: Daytime sources of nitrous acid (HONO) in the atmospheric boundary layer, ChemPhysChem, 8(8), 1137–1144, doi:10.1002/cphc.200700016, 2007.

Kleffmann, J., Becker, K. H. and Wiesen, P.: Heterogeneous NO₂ conversion processes on acid surfaces: Possible atmospheric implications, Atmos. Environ., 32(16), 2721–2729, doi:10.1016/S1352-2310(98)00065-X, 1998.

35 Kleffmann, J., Becker, K. H., Lackhoff, M. and Wiesen, P.: Heterogeneous conversion of NO₂ on carbonaceous surfaces, Phys. Chem. Chem. Phys., 1(24), 5443–5450, doi:10.1039/a905545b, 1999.

Kurtenbach, R., Becker, K. H., Gomes, J. A. G., Kleffmann, J., Lörzer, J. C., Spittler, M., Wiesen, P., Ackermann, R., Geyer, A. and Platt, U.: Investigations of emissions and heterogeneous formation of HONO in a road traffic tunnel, Atmos. Environ., 35(20), 3385–3394, doi:10.1016/S1352-2310(01)00138-8, 2001.

Langridge, J. M., Gustafsson, R. J., Griffiths, P. T., Cox, R. A., Lambert, R. M. and Jones, R. L.: Solar driven nitrous acid formation on building material surfaces containing titanium dioxide: A concern for air quality in urban areas?, Atmos. Environ., 43(32), 5128–5131, doi:10.1016/i.atmosenv.2009.06.046, 2009.

Laufs, S., Cazaunau, M., Stella, P., Kurtenbach, R., Cellier, P., Mellouki, A., Loubet, B. and Kleffmann, J.: Diurnal fluxes of HONO above a crop rotation, Atmos. Chem. Phys., 17(11), 6907–6923, doi:10.5194/acp-17-6907-2017, 2017.

Lee, B. H., Wood, E. C., Herndon, S. C., Lefer, B. L., Luke, W. T., Brune, W. H., Nelson, D. D., Zahniser, M. S. and Munger, J. W.: Urban measurements of atmospheric nitrous acid: A caveat on the interpretation of the HONO photostationary state, J. Geophys, Res. Atmos., 118(21), 12274–12281, doi:10.1002/2013JD020341, 2013.

Lee, J. D., Whalley, L. K., Heard, D. E., Stone, D., Dunmore, R. E., Hamilton, J. F., Young, D. E., Allan, J. D., Laufs, S. and Kleffmann, J.: 10 Detailed budget analysis of HONO in central London reveals a missing daytime source. Atmos. Chem. Phys., 16(5), 2747–2764, doi:10.5194/acp-16-2747-2016, 2016.

Lelièvre, S., Bedianian, Y., Laverdet, G. and Le Bras, G.: Heterogeneous reaction of NO₂ with hydrocarbon flame soot, J. Phys. Chem. A. 108(49), 10807-10817, doi:10.1021/jp0469970, 2004.

Liang, Y., Zha, Q., Wang, W., Cui, L., Lui, K. H., Ho, K. F., Wang, Z., Lee, S. C. and Wang, T.: Revisiting nitrous acid (HONO) emission 15 from on-road vehicles: A tunnel study with a mixed fleet, J. Air Waste Manag. Assoc., 67(7), 797-805, doi:10.1080/10962247.2017.1293573, 2017.

Liu, Y., Lu, K., Ma, Y., Yang, X., Zhang, W., Wu, Y., Peng, J., Shuai, S., Hu, M. and Zhang, Y.: Direct emission of nitrous acid (HONO) from gasoline cars in China determined by vehicle chassis dynamometer experiments, Atmos. Environ., 169, 89-96, doi:10.1016/j.atmosenv.2017.07.019, 2017.

20 Maier, S., Tamm, A., Wu, D., Caesar, J., Grube, M. and Weber, B.: Photoautotrophic organisms control microbial abundance, diversity, and physiology in different types of biological soil crusts, ISME J., 12(4), 1032–1046, doi:10.1038/s41396-018-0062-8, 2018.

Maljanen, M., Yli-Pirilä, P., Hytönen, J., Joutsensaari, J. and Martikainen, P. J.: Acidic northern soils as sources of atmospheric nitrous acid (HONO), Soil Biol. Biochem., 67(2), 94–97, doi:10.1016/j.soilbio.2013.08.013, 2013.

Matthaios, V. N., Kramer, L. J., Sommariva, R., Pope, F. D. and Bloss, W. J.: Investigation of vehicle cold start primary NO₂ emissions 25 inferred from ambient monitoring data in the UK and their implications for urban air quality, Atmos, Environ., 199(2), 402-414, doi:10.1016/j.atmosenv.2018.11.031, 2019.

Meusel, H., Tamm, A., Kuhn, U., Wu, D., Lena Leifke, A., Fiedler, S., Ruckteschler, N., Yordanova, P., Lang-Yona, N., Pöhlker, M., Lelieveld, J., Hoffmann, T., Pöschl, U., Su, H., Weber, B. and Cheng, Y.: Emission of nitrous acid from soil and biological soil crusts represents an important source of HONO in the remote atmosphere in Cyprus, Atmos. Chem. Phys., 18(2), 799-813, doi:10.5194/acp-18-799-2018, 2018.

30

5

Michoud, V., Kukui, A., Camredon, M., Colomb, A., Borbon, A., Miet, K., Aumont, B., Beekmann, M., Durand-Jolibois, R., Perrier, S., Zapf, P., Siour, G., Ait-Helal, W., Locoge, N., Sauvage, S., Afif, C., Gros, V., Furger, M., Ancellet, G. and Doussin, J. F.: Radical budget analysis in a suburban European site during the MEGAPOLI summer field campaign, Atmos. Chem. Phys., 12(24), 11951-11974, doi:10.5194/acp-12-11951-2012, 2012.

35 Michoud, V., Colomb, A., Borbon, A., Miet, K., Beekmann, M., Camredon, M., Aumont, B., Perrier, S., Zapf, P., Siour, G., Ait-Helal, W., Afif, C., Kukui, A., Furger, M., Dupont, J. C., Haeffelin, M. and Doussin, J. F.: Study of the unknown HONO daytime source at a European suburban site during the MEGAPOLI summer and winter field campaigns, Atmos. Chem. Phys., 14(6), 2805-2822, doi:10.5194/acp-14-2805-2014, 2014.

Monge, M. E., D'Anna, B., Mazri, L., Giroir-Fendler, A., Ammann, M., Donaldson, D. J. and George, C.: Light changes the atmospheric 40 reactivity of soot, Proc. Natl. Acad. Sci., 107(15), 6605-6609, doi:10.1073/pnas.0908341107, 2010.

Nakashima, Y. and Kajii, Y.: Determination of nitrous acid emission factors from a gasoline vehicle using a chassis dynamometer combined with incoherent broadband cavity-enhanced absorption spectroscopy, Sci. Total Environ., 575, 287-293, doi:10.1016/j.scitotenv.2016.10.050, 2017.

Ning, Z., Wubulihairen, M. and Yang, F.: PM, NO_x and butane emissions from on-road vehicle fleets in Hong Kong and their implications on emission control policy, Atmos. Environ., 61(2), 265-274, doi:10.1016/j.atmosenv.2012.07.047, 2012.

O'Driscoll, R., Stettler, M. E. J., Molden, N., Oxley, T. and ApSimon, H. M.: Real world CO₂ and NO₂ emissions from 149 Euro 5 and 6 diesel, gasoline and hybrid passenger cars, Sci. Total Environ., 621(x), 282–290, doi:10.1016/j.scitotenv.2017.11.271, 2018.

5 Oswald, R., Behrendt, T., Ermel, M., Wu, D., Su, H., Cheng, Y., Breuninger, C., Moravek, A., Mougin, E., Delon, C., Loubet, B., Pommerening-Röser, A., Sörgel, M., Pöschl, U., Hoffmann, T., Andreae, M. O., Meixner, F. X. and Trebs, I.: HONO emissions from soil bacteria as a major source of atmospheric reactive nitrogen., Science, 341(6151), 1233–5, doi:10.1126/science.1242266, 2013.

Pierson, W. R., Brachaczek, W. W., Hammerle, R. H., McKee, D. E. and Butler, J. W.: Sulfate Emissions from Vehicles on the Road, J. Air Pollut. Control Assoc., 28(2), 123–132, doi:10.1080/00022470.1978.10470579, 1978.

10 Pierson, W. R., Gertler, A. W., Robinson, N. F., Sagebiel, J. C., Zielinska, B., Bishop, G. A., Stedman, D. H., Zweidinger, R. B. and Ray, W. D.: Real-world automotive emissions - summary of studies in the Fort McHenry and Tuscarora Mountain Tunnels, Atmos. Environ., 30(12), 2233–2256, doi:10.1016/1352-2310(95)00276-6, 1996.

Pitts, J. N., Biermann, H. W., Winer, A. M. and Tuazon, E. C.: Spectroscopic identification and measurement of gaseous nitrous acid in dilute auto exhaust, Atmos. Environ., 18(4), 847–854, doi:10.1016/0004-6981(84)90270-1, 1984.

15 Qin, M., Xie, P., Su, H., Gu, J., Peng, F., Li, S., Zeng, L., Liu, J., Liu, W. and Zhang, Y.: An observational study of the HONO-NO₂ coupling at an urban site in Guangzhou City, South China, Atmos. Environ., 43(36), 5731–5742, doi:10.1016/j.atmosenv.2009.08.017, 2009.

Rappenglück, B., Lubertino, G., Alvarez, S., Golovko, J., Czader, B. and Ackermann, L.: Radical precursors and related species from traffic as observed and modeled at an urban highway junction, J Air Waste Manag Assoc, 63(11), 1270–1286, doi:10.1080/10962247.2013.822438, 2013.

20 Reed, C., Evans, M. J., Crilley, L. R., Bloss, W. J., Sherwen, T., Read, K. A., Lee, J. D. and Carpenter, L. J.: Evidence for renoxification in the tropical marine boundary layer, Atmos. Chem. Phys. Discuss., 1–29, doi:10.5194/acp-2016-1111, 2017.

Rhead, M., Gurney, R., Ramalingam, S. and Cohen, N.: Accuracy of automatic number plate recognition (ANPR) and real world UK number plate problems, Proc. - Int. Carnahan Conf. Secur. Technol., 286–291, doi:10.1109/CCST.2012.6393574, 2012.

Richard, C., Gordon, I. E., Rothman, L. S., Abel, M., Frommhold, L., Gustafsson, M., Hartmann, J. M., Hermans, C., Lafferty, W. J., Orton,
G. S., Smith, K. M. and Tran, H.: New section of the HITRAN database: Collision-induced absorption (CIA), J. Quant. Spectrosc. Radiat. Transf., 113(11), 1276–1285, doi:10.1016/j.jqsrt.2011.11.004, 2012.

Rogak, S. N., Green, S. I. and Pott, U.: Use of tracer gas for direct calibration of emission-factor measurements in a traffic tunnel, J. Air Waste Manag. Assoc., 48(6), 545–552, doi:10.1080/10473289.1998.10463707, 1998.

Romanias, M. N., Bedjanian, Y., Zaras, A. M., Andrade-Eiroa, A., Shahla, R., Dagaut, P. and Philippidis, A.: Mineral oxides change the atmospheric reactivity of soot: NO₂ uptake under dark and UV irradiation conditions, J. Phys. Chem. A, 117(48), 12897–12911, doi:10.1021/jp407914f, 2013.

Sander, R.: Compilation of Henry's law constants (version 4.0) for water as solvent, Atmos. Chem. Phys., 15(8), 4399–4981, doi:10.5194/acp-15-4399-2015, 2015.

Singh, A.: Quantifying the effect of atmospheric pollution and meteorology on visibility and tropospheric chemistry, University of Birmingham. [online] Available from: https://etheses.bham.ac.uk//id/eprint/7828/, 2017.

Spataro, F. and Ianniello, A.: Sources of atmospheric nitrous acid: State of the science, current research needs, and future prospects, J. Air Waste Manage. Assoc., 64(11), 1232–1250, doi:10.1080/10962247.2014.952846, 2014.

Stadler, D. and Rossi, M. J.: The reactivity of NO_2 and HONO on flame soot at ambient temperature: The influence of combustion conditions, Phys. Chem. Phys., 2(23), 5420–5429, doi:10.1039/b0056800, 2000.

40 Stemmler, K., Ammann, M., Donders, C., Kleffmann, J. and George, C.: Photosensitized reduction of nitrogen dioxide on humic acid as a

source of nitrous acid., Nature, 440(7081), 195-198, doi:10.1038/nature04603, 2006.

Stutz, J., Kim, E. S., Platt, U., Bruno, P., Perrino, C. and Febo, A.: UV-visible absorption cross sections of nitrous acid, J. Geophys. Res. Atmos., 105(D11), 14585–14592, doi:10.1029/2000JD900003, 2000.

Stutz, J., Alicke, B. and Neftel, A.: Nitrous acid formation in the urban atmosphere: Gradient measurements of NO₂ and HONO over grass 5 in Milan, Italy, J. Geophys. Res., 107(D22), 8192, doi:10.1029/2001JD000390, 2002.

Stutz, J., Alicke, B., Ackermann, R., Geyer, A., Wang, S., White, A. B., Williams, E. J., Spicer, C. W. and Fast, J. D.: Relative humidity dependence of HONO chemistry in urban areas, J. Geophys. Res., 109(D3), D03307, doi:10.1029/2003JD004135, 2004.

Su, H., Cheng, Y., Oswald, R., Behrendt, T., Trebs, I., Meixner, F. X., Andreae, M. O., Cheng, P., Zhang, Y. and Pöschl, U.: Soil nitrite as a source of atmospheric HONO and OH radicals., Science, 333(6049), 1616–8, doi:10.1126/science.1207687, 2011.

- 10 Thalman, R., Baeza-Romero, M. T., Ball, S. M., Borrás, E., Daniels, M. J. S., Goodall, I. C. A., Henry, S. B., Karl, T., Keutsch, F. N., Kim, S., Mak, J., Monks, P. S., Muñoz, A., Orlando, J., Peppe, S., Rickard, A. R., Ródenas, M., Sánchez, P., Seco, R., Su, L., Tyndall, G., Vázquez, M., Vera, T., Waxman, E. and Volkamer, R.: Instrument intercomparison of glyoxal, methyl glyoxal and NO₂ under simulated atmospheric conditions, Atmos. Meas. Tech., 8(4), 1835–1862, doi:10.5194/amt-8-1835-2015, 2015.
- Tong, S., Hou, S., Zhang, Y., Chu, B., Liu, Y., He, H., Zhao, P. and Ge, M.: Exploring the nitrous acid (HONO) formation mechanism in
 winter Beijing: Direct emissions and heterogeneous production in urban and suburban areas, Faraday Discuss., 189, 213–230, doi:10.1039/c5fd00163c, 2016.

Trinh, H. T., Imanishi, K., Morikawa, T., Hagino, H. and Takenaka, N.: Gaseous nitrous acid (HONO) and nitrogen oxides (NO_x) emission from gasoline and diesel vehicles under real-world driving test cycles, J. Air Waste Manag. Assoc., 67(4), 412–420, doi:10.1080/10962247.2016.1240726, 2017.

20 Vandaele, A. C., Hermans, C., Simon, P. C., Carleer, M., Colin, R. and Coquartii, B.: Measurements of the NO2 absorption cross-section from 42 000 cm⁻¹ to 10 000 cm⁻¹ (238–1000 nm) at 220 K and 294 K, J. Quant. Spectrosc. Radiat. Transf., 59(3), 171–184, 1998.

Vandenboer, T. C., Markovic, M. Z., Sanders, J. E., Ren, X., Pusede, S. E., Browne, E. C., Cohen, R. C., Zhang, L., Thomas, J., Brune, W. H. and Murphy, J. G.: Evidence for a nitrous acid (HONO) reservoir at the ground surface in Bakersfield, CA, during CalNex 2010, Journal Geophys. Res., 119, 9093–9106, doi:10.1002/2013JD020971.Received, 2014.

Villena, G., Wiesen, P., Cantrell, C. A., Flocke, F., Fried, A., Hall, S. R., Hornbrook, R. S., Knapp, D., Kosciuch, E., Mauldin, R. L., McGrath, J. A., Montzka, D., Richter, D., Ullmann, K., Walega, J., Weibring, P., Weinheimer, A., Staebler, R. M., Liao, J., Huey, L. G. and Kleffmann, J.: Nitrous acid (HONO) during polar spring in Barrow, Alaska: A net source of OH radicals?, J. Geophys. Res. Atmos., 116(24), 1–12, doi:10.1029/2011JD016643, 2011.

Villena, G., Bejan, I., Kurtenbach, R., Wiesen, P. and Kleffmann, J.: Interferences of commercial NO₂ instruments in the urban atmosphere and in a smog chamber, Atmos. Meas. Tech., 5(1), 149–159, doi:10.5194/amt-5-149-2012, 2012.

Vogel, B., Vogel, H., Kleffmann, J. and Kurtenbach, R.: Measured and simulated vertical profiles of nitrous acid - Part II. Model simulations and indications for a photolytic source, Atmos. Environ., 37(21), 2957–2966, doi:10.1016/S1352-2310(03)00243-7, 2003.

Wang, J., Zhang, X., Guo, J., Wang, Z. and Zhang, M.: Observation of nitrous acid (HONO) in Beijing, China: Seasonal variation, nocturnal formation and daytime budget, Sci. Total Environ., doi:10.1016/j.scitotenv.2017.02.159, 2017.

35 Wang, X., Ho, K. F., Chow, J. C., Kohl, S. D., Chan, C. S., Cui, L., Lee, S. cheng F., Chen, L. W. A., Ho, S. S. H., Cheng, Y. and Watson, J. G.: Hong Kong vehicle emission changes from 2003 to 2015 in the Shing Mun Tunnel, Aerosol Sci. Technol., 52(10), 1085–1098, doi:10.1080/02786826.2018.1456650, 2018.

Weber, B., Wu, D., Tamm, A., Ruckteschler, N., Rodríguez-Caballero, E., Steinkamp, J., Meusel, H., Elbert, W., Behrendt, T., Sörgel, M., Cheng, Y., Crutzen, P. J., Su, H. and Pöschl, U.: Biological soil crusts accelerate the nitrogen cycle through large NO and HONO emissions in drylands., Proc. Natl. Acad. Sci. U. S. A., 112(50), 15384–9, doi:10.1073/pnas.1515818112, 2015.

40

Xu, Z., Wang, T., Wu, J., Xue, L., Chan, J., Zha, Q., Zhou, S., Louie, P. K. K. and Luk, C. W. Y.: Nitrous acid (HONO) in a polluted

subtropical atmosphere: Seasonal variability, direct vehicle emissions and heterogeneous production at ground surface, Atmos. Environ., 106, 100–109, doi:10.1016/j.atmosenv.2015.01.061, 2015.

Yang, Q., Su, H., Li, X., Cheng, Y., Lu, K., Cheng, P., Gu, J., Guo, S., Hu, M., Zeng, L., Zhu, T. and Zhang, Y.: Daytime HONO formation in the suburban area of the megacity Beijing, China, Sci. China Chem., 57(7), 1032–1042, doi:10.1007/s11426-013-5044-0, 2014.

5 Ye, C., Zhang, N., Gao, H. and Zhou, X.: Photolysis of Particulate Nitrate as a Source of HONO and NO_x, Environ. Sci. Technol., 51(12), 6849–6856, doi:10.1021/acs.est.7b00387, 2017.

Zhou, X., Zhang, N., Teravest, M., Tang, D., Hou, J., Bertman, S. and Stevens, P. S.: Nitric acid photolysis on forest canopy surface as a source for tropospheric nitrous acid, Nat. Geosci., 4(7), 440–443, doi:10.1038/ngeo1164, 2011.

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Figure 1: Sources and sinks of HONO in the troposphere, showing direct emission (red arrows), secondary sources (blue arrows) and HONO sinks (green arrows), dashed arrows represent solar driven reactions.



Figure 2: Map of Birmingham City Centre showing the M6 motorway (red dashed line) and schematic of the Queensway Tunnel (highlighted in blue on the map). © OpenStreetMap contributors 2019. Distributed under a Creative Commons BY-SA License.



Figure 3: a) Average hourly vehicle fleet in the southbound tunnel from the corrected ANPR counts for weekdays (LGV = light goods vehicles, HD = heavy duty vehicles e.g. trucks, lorries and buses), orange points represent the counts estimated from video footage <u>recorded</u> during the measurement period, b) percentage of vehicle types, determined from all weekday ANPR data in November 2016.



Figure 4: 15 minute averaged measurements of NO, NO₂, HONO and CO₂ from the Queensway tunnel between 29th July and 8th August 2016. Shaded areas indicate the weekends. Spikes in the measured mixing ratios on weekdays are the result of traffic congestion during morning and evening rush hour periods.



Figure 5: Hourly averaged time series a) NO_x in the tunnel and from Acocks Green urban background station, b) HONO in the tunnel and estimated background HONO, c) CO₂ in the tunnel and background CO₂ measured at BAQS.



Figure 6: RMA regression for hourly averaged ΔNO_x versus $\Delta HONO$, for weekdays between 06:00 and 19:00. The blue line represents the RMA slope and the grey lines the 95 % confidence intervals of the regression.



Figure 7: a) Hourly averaged Δ HONO/ Δ NO_x b) hourly averaged Δ HONO/ Δ CO₂ and c) fraction of diesel and non-diesel (petrol, biofuel, electric) vehicles calculated from the total petrol and diesel vehicles, for the period between 06:00 and 19:00 during the weekday. Error bars represent 1- σ standard deviation of the mean across the different days of this study.



Figure 8: Top: NO_x, HONO and inferred vehicle exhaust HONO from mobile measurements taken on 23rd October 2015 along a
 return journey transect-between Birmingham and Leicester (Crilley et al. 2016). Bottom: ratio of <u>vehicle-produced</u> HONO_{veh} to measured /HONO (%) for the same period. Grey shaded areas represent non-driving periods, i.e. when the mobile laboratory was parked either at the University of Birmingham or University of Leicester campus. Other colours represent: measurements around the University of Birmingham campus (light blue), A38 road through Birmingham city centre (dark blue), M6 and M/69 motorways (red), Leicester Ring Road (purple), and measurements around the University of Leicester campus (or ange). HONO_{veh}/HONO above

10 100_% are the result of uncertainties in the HONO and NO_x measurements and the HONO/NO_x <u>emission</u> ratio.

Tables

Cable 1: Overview of instruments	s deployed in the	Queensway	Tunnel
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Instrument	<u>Measurement</u>	Method	Sampling time (s)	<u>Uncertainty</u>
TEI: 42c	NO (and NO _y)	Chemiluminescence (Mo <u>converter</u> Conv.)	60	±_10%
BBCEAS (open-path)	HONO and NO ₂	Broadband cavity enhanced absorption spectroscopy	20	
LICOR: LI-820	CO ₂	non-dispersive infrared	1	± 5 %
Kestrel 4500 : Mini- Met anemometer	Wind speed (WS), Temperature (T), Relative Humidity (RH)	- <u>Impeller (WS)</u> Solid state sensor (T, RH)	600	$\frac{\text{The larger of } \pm 3}{\text{\% or } 0.1 \text{ m s}^{-1}}$ $\frac{\text{(WS)}}{\pm 0.5 \text{ °C}}$ $\frac{\pm 3 \text{ \% RH units}}{-}$

5 Table 2: Comparison of HONO/NO_x ratios from different tunnel studies.

Location	Year	Tunnel	Avg.	Diesel	Petrol	Other	HONO/NOx	Reference
	of	Length	Vehicles	Vehicles	Vehicles	(%)	(%)	
	Study	(m)	/Day	(%)	(%)			
Queensway	2016	548	37,700	59	40	<1	0.85	This study
Tunnel							(CI:0.72-	
Birmingham,							1.01)	
U.K.								
Shing Mun	2015	1600	26,970	38	47	15 ^a	1.24 ± 0.35	Liang et al.,
Tunnel								2017
Hong Kong								
Kiesbergtunnel,	1997/	1100	22,000	24.3 ^b	74.7	1	0.80 ± 0.10	Kurtenbach
Wuppertal,	1998							et al., 2001
Germany								
Caldecott	1995	1100	-	<0.2	99	-	0.29 ± 0.05	Kirchstetter
Tunnel								et al. 1996

California, U.S.

^a LPG fuelled

^b Includes LGVs and OGVs (assumption made here that these are diesel fuelled)

Supplementary Material

DEFRA Emission Factor Toolkit

The DEFRA Emissions Factors Toolkit (EFT v8.0.1) (DEFRA, 2017) is used by local authorities in the United Kingdom (UK) to determine vehicle emissions. The EFT includes updated PM and NO_x emission equations and Euro 6 standards from the European Environment Agency (EEA) v5 COPERT (COmputer Program to calculate Emissions from Road Transport) emission factor model. To determine CO₂ and NO_x emission factors from the EFT, detailed hourly fleet composition from the ANPR data was inputted along with information on the region (not London), road type (A road), year (2016), traffic flow (corrected ANPR counts) and average <u>vehicle</u> speeds (Figure S2). Euro class proportions for 2016 are automatically selected in the EFT and are based on the Department for Transport vehicle fleet composition projections (Base 2016) (DfT, 2016)

Supplementary Tables

Vehicle Type	Petrol	Diesel	Hybrid Electric	Gas	Electric	Other	Total	% of total
Cars	18,825.2	12,574.3	322.3	36.0	34.1	0.4	31,792.3	82.8
Motorcycles							1,270.2	3.3
Light Goods Vehicles	130.5	3,745.3		7.8	5.3	0.8	3,889.7	10.1
Heavy Goods Vehicles		517.1					517.1	1.3
Buses and coaches							167.1	0.4
Other vehicles ¹							751.9	2.0
Total							38,388.2	
1. Includes rear diggers, lift trucks, rollers, ambulances, Hackney Carriages (i.e. taxis), three wheelers, tricycles and agricultural vehicles.								

Table S1: Number of licensed vehicles in the UK (units of 1000) at the end of 2016 (Source: DVLA/DfT https://www.gov.uk/government/collections/vehicles-statistics)

Table S2: Fraction of fuel type and vehicle type for the vehicles travelling through the Queensway tunnel from 06:00 to 19:59 (timestamp is the start of the hourly average). The last two columns show hourly mean $\Delta HONO/\Delta NO_x$ and $\Delta HONO/\Delta CO_2$ emission ratios measured at the sampling site inside the tunnel.

Start Hour	Diesel	Non-Diesel fraction ^a	Car fraction ^b	LGV+HD fraction ^c	$\Delta HONO/\Delta NO_x$	$\Delta HONO/\Delta CO_2$		
	inaction	inaction		naction	(ppu/ppu)	(ppo/ppin)		
06:00	0.60	0.40	0.81	0.19	0.0071	0.033		
07:00	0.58	0.42	0.82	0.18	0.0066	0.026		
08:00	0.57	0.43	0.85	0.15	0.0075	0.027		
09:00	0.62	0.38	0.83	0.17	0.0078	0.028		
10:00	0.66	0.34	0.81	0.19	0.0091	0.033		
11:00	0.63	0.37	0.81	0.19	0.0086	0.031		
12:00	0.64	0.36	0.82	0.18	0.0088	0.029		
13:00	0.61	0.39	0.84	0.16	0.0083	0.028		
14:00	0.61	0.39	0.84	0.16	0.0079	0.026		
15:00	0.58	0.42	0.87	0.13	0.0076	0.023		
16:00	0.54	0.46	0.89	0.11	0.0078	0.021		
17:00	0.54	0.46	0.93	0.07	0.0073	0.015		
18:00	0.54	0.46	0.94	0.06	0.0095	0.021		
19:00	0.56	0.44	0.94	0.06	0.0111	0.031		
^a petrol biofuel and electric vehicles ^b cars and other small vehicles (e.g. motor cycles)								

^a petrol, biofuel and electric vehicles, ^b cars and other small vehicles (e.g. motor cycles), ^c light goods vehicles, heavy duty goods vehicles and buses

Supplementary Figures



Figure S1: a) Hourly averaged BC vs HONO data taken at the North Kensington site in London in 2012 during the ClearfLo project (Bohnenstengel et al., 2015), b) BC vs HONO/NO_x coloured by the NO/NO_x ratio for the same period. High NO/NO_x ratios (yellow/red coloured points) are indicative of fresh pollution plumes and low chemical processing. [BSM(1] There is no clear correlation observed here between BC and HONO/NO_x for fresh pollution plumes. HONO formation via NO₂ conversion on BC, therefore, may not be a large source of HONO after emission from the vehicle exhaust.



Figure S2: Hourly mean vehicle speed ($\underline{km h^{-1}}$) measured in the Southbound bore of the Queensway tunnel over 2015.



Figure S3: <u>An extract of</u> 15 minute averaged HONO data sampled inside the tunnel (green) and hourly averaged precipitation data (blue) from the Winterbourne weather station from <u>00:00 on 31 July to 00:00 on</u> <u>05</u> Aug 2016. The shaded area represents the <u>rain-affected</u> period that was <u>excluded</u> from the final dataset (see main text for more details).



Figure S4: Diurnal profiles of wind speeds measured by the Kestrel anemometer (red time series) and wind speeds inferred from CFD modelling of CO₂ concentration profiles measured in the tunnel (black crosses on yellow background). The model wind speeds agree well with the anemometer wind speeds multiplied by a factor of 3.0 (blue time series). Note also the drop in wind speeds around 17:00 to 19:00 when traffic flow is congested inside the tunnel.



Figure S5: Average diurnal weekday HONO/NO_x cycle calculated from measurements taken at BAQS between <u>18</u> March and <u>1</u> April, 2015. Error bars represent 1-σ standard deviation of the mean.



Figure S6: Hourly averaged CO₂ and NO_x emission factors calculated using the DEFRA Emissions Factors Toolkit (EFT v8.0.1) (DEFRA, 2017) for traffic flows in Queensway tunnel on weekdays from 06:00 to 19:00.



Figure S7: Linear regression analysis of hourly average diesel fraction vs Δ HONO/ Δ NO_x from 06:00 to 17:00 based on the data from Table S2. Error bars represent 1- σ standard deviation of the mean.

References

Bohnenstengel, S.I., S.E. Belcher, A. Aiken, J.D. Allan, G. Allen, A. Bacak, T.J. Bannan, J.F. Barlow, D.C.
Beddows, W.J. Bloss, A.M. Booth, C. Chemel, O. Coceal, C.F. Di Marco, M.K. Dubey, K.H. Faloon, Z.L. Fleming, M. Furger, J.K. Gietl, R.R. Graves, D.C. Green, C.S. Grimmond, C.H. Halios, J.F. Hamilton, R.M. Harrison, M.R.
Heal, D.E. Heard, C. Helfter, S.C. Herndon, R.E. Holmes, J.R. Hopkins, A.M. Jones, F.J. Kelly, S. Kotthaus, B.
Langford, J.D. Lee, R.J. Leigh, A.C. Lewis, R.T. Lidster, F.D. Lopez-Hilfiker, J.B. McQuaid, C. Mohr, P.S. Monks, E. Nemitz, N.L. Ng, C.J. Percival, A.S. Prévôt, H.M. Ricketts, R. Sokhi, D. Stone, J.A. Thornton, A.H. Tremper, A.C. Valach, S. Visser, L.K. Whalley, L.R. Williams, L. Xu, D.E. Young, and P. Zotter, 2015: Meteorology, Air Quality, and Health in London: The ClearfLo Project. *Bull. Amer. Meteor. Soc.*, 96, 779–804, https://doi.org/10.1175/BAMS-D-12-00245.1

DEFRA, 2017. DEFRA, 2017: Emission Factors Toolkit for Vehicle Emissions. [WWW Document]. URL https://laqm.defra.gov.uk/review-and-assessment/tools/emissions-factors-toolkit.html (accessed 7.14.19).

DfT, 2016. Vehicle fleet composition projections (Base 2016) [WWW Document]. URL http://naei.beis.gov.uk/data/ef-transport.