

Characteristics and sources of aerosol aminiums over the eastern coast of China: Insights from the integrated observations in a coastal city, adjacent island and the marginal seas

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Abstract. An integrated observation on aerosol aminiums was conducted in a coastal city (Shanghai) of eastern China, a nearby island (Huaniao Island) and over the Yellow Sea and East China Sea (YECS). Triethylammonium (TEAH⁺) was abundant over Shanghai but not detected over the island and the open seas, suggesting its predominantly terrestrial origin. By contrast, relatively high concentrations of dimethylammonium (DMAH⁺) and trimethylammonium+diethylammonium (TMDEAH⁺) were measured over the ocean sites, indicating the significant marine source contribution. Environmental factors, including boundary layer height (BLH), temperature, atmospheric oxidizing capacity and relative humidity, were found to be related to aminium concentrations. All the detected aminiums demonstrated the highest levels in winter in Shanghai, consistent with the lowest BLH and temperature in this season. Aminiums mainly existed in fine particles and showed a bimodal distribution with two peaks at 0.18–0.32 μm and 0.56–1.0 μm , indicating that condensation and cloud processing were main formation pathways for aminiums in analogy with NH_4^+ and non-sea-salt SO_4^{2-} (nss- SO_4^{2-}). Nonetheless, a unimodal distribution for aerosol aminiums was usually measured over the YECS or over the Huaniao Island when influenced mainly by the marine air-mass, which suggested that aminiums in marine aerosols may undergo different formation pathways from those on the land. Terrestrial anthropogenic sources and marine biogenic sources were both important contributors for DMAH⁺ and TMDEAH⁺, and the latter exhibited a significantly higher TMDEAH⁺ to DMAH⁺ ratio. By using the mass ratio of methanesulfonate (MSA) to nss- SO_4^{2-} as an indicator of marine biogenic source, we estimated that marine biogenic source contributed to 26–31% and 53–78% of aerosol aminiums over Huaniao Island in the autumn of 2016 and summer of 2017, respectively. Due to the important role of atmospheric amines in new particle formation, the estimation results highlighted the importance of marine biogenic emission of amines in the eastern coast of China, especially in summer.

1 Introduction

Low molecular weight amines are commonly found in the atmosphere in both gas and particle phases (Ge et al., 2011b, a). Base on present theoretical calculations (Kurten et al., 2008; Loukonen et al., 2010; Paasonen et al., 2012; Olenius et al., 2017), laboratory simulations (Wang et al., 2010a; Wang et al., 2010b; Kurten et al., 2014; Erupe et al., 2011; Almeida et al., 2013; Yu et al., 2012) and field observations (Smith et al., 2010; Kürten et al., 2016; Tao et al., 2016), amines in the atmosphere have been proved to play an important role in new particle formation and subsequent particle growth, and thus affect both the number concentrations of aerosols and cloud condensation nuclei which are closely relevant to regional climate (Tang et al., 2014; Yao et al., 2018). For example, dimethylamine (DMA) was found to be a key species involved in new particle formation events in the urban area of Shanghai, and the nucleation mechanism was likely to be H_2SO_4 -DMA- H_2O ternary nucleation

40 (Yao et al., 2018). Gaseous amines in the atmosphere can react with oxidants such as $\cdot\text{OH}$ and O_3 to form secondary organic
41 aerosols (SOA) (Murphy et al., 2007) or gaseous oxidation products such as imines, formamides, nitrosamines and nitramines
42 (Nielsen et al., 2012). In aerosols, amines are mainly in the form of protonated cations, namely aminiums (Ge et al., 2011a),
43 and the formation of aminium salts from acid-base reaction or heterogeneous reaction, such as replacing the NH_4^+ in particles,
44 is another important pathway for amines to form SOA in the atmosphere (Pankow, 2015; Kupiainen et al., 2012; Liu et al.,
45 2012; Chan and Chan, 2013).

46 Amines originate from a wide range of sources, including anthropogenic sources such as animal husbandry and industrial
47 emissions, as well as natural sources such as marine source, vegetation emission, and soil processing (Ge et al., 2011b;
48 Hemmilä et al., 2018). Dawson et al. (2014) measured the concentrations of trimethylamine (TMA, 1.3–6.8 ppt) near a cattle
49 farm which were 2–3 orders of magnitude higher than those in ambient environment. Shen et al. (2017) demonstrated that coal
50 combustion could emit abundant methylaminium (MMAH^+), ethylaminium (MEA^+) and diethylaminium (DEAH^+) through
51 combustion experiments, and the corresponding emission factors were 18.0 ± 16.4 , 30.1 ± 25.6 and 14.6 ± 10.1 mg (kg coal) $^{-1}$,
52 respectively. In marine boundary layer, marine source is an important contributor for amines and it was found to be closely
53 related to the biological activities in ocean surface. In the North Atlantic, the concentrations of dimethylaminium (DMAH^+)
54 and DEAH^+ were significantly higher during the periods with high biological activity and clean air masses than those with low
55 biological activity or polluted air masses advecting to the sampling site, and the contributions of these two aminiums to SOA
56 and water soluble organic nitrogen (WSO_N) reached 11% and 35%, respectively (Facchini et al., 2008). The observation in
57 Cape Verde also showed that the concentrations of aminiums were higher during the occurrence of algal blooms (Müller et al.,
58 2009). In addition to gas-to-particle conversion which has been generally considered to be the major formation pathways
59 (Facchini et al., 2008; Rinaldi et al., 2010), aminiums in marine boundary layer may also be generated with primary marine
60 aerosols. For example, Fourier Transform Infrared (FTIR) spectroscopy measurements demonstrated that the submicron
61 organic carbon was composed of 50% hydroxyl, 33% alkane and 14% amine in nascent sea spray aerosols artificially generated
62 off the California coast (Bates et al., 2012) and of 55% hydroxyl, 32% alkane and 13% amine over the open ocean (Frossard
63 et al., 2014). Aerosol Time-of-Flight Mass Spectrometry (ATOFMS) analyses of ambient aerosols in the Antarctic sympagic
64 environment also indicated that 11–25% of aminiums were contributed by primary marine source (Dall'Osto et al., 2019).

65 Given the potentially important roles of amines in the atmosphere and the complexity of their sources, it is important to conduct
66 a systematic analysis on their concentrations, affecting factors, formation pathways and source contributions. The eastern
67 China is a densely populated region with strong human activities and large emissions of atmospheric pollutants. Under the
68 influence of the summer monsoon, marine source components can be vital to the atmospheric composition of the coastal area.
69 Although the lifetime of gaseous amines in the atmosphere is only a few hours, it can be prolonged after amines partition into
70 the particle phase, and thus, they may be transported over a long range (Nielsen et al., 2012). Many studies have been done on
71 the gas and/or particle phases of amines over eastern China and adjacent seas (Huang et al., 2012; Hu et al., 2015; Zheng et
72 al., 2015; Huang et al., 2016; Tao et al., 2016; Yu et al., 2016; Shen et al., 2017; Xie et al., 2018; Yao et al., 2018; Yao et al.,
73 2016). For example, C1- to C6-amines over Shanghai were measured during the summer of 2015, of which C1-, C2- and C4-
74 amines were the dominant species with the average concentrations of 15.7, 40.0 and 15.4 ppt, respectively (Yao et al., 2016).
75 Zheng et al. (2015) measured an average concentration 7.2 ppt of total amines in a suburban site of Nanjing during the summer
76 of 2012, derived mainly from industrial emissions in adjacent areas. The aminiums in fine particles over Shanghai in the
77 summer of 2013 were found to exhibited a high concentration (mean 86.4 ng m^{-3}) and played an important role in the new
78 particle formation events (Tao et al., 2016). Previous studies on aminiums over the marginal seas of China indicated that
79 DMAH^+ and trimethylaminium (TMAH^+) were overwhelmingly from marine sources (Hu et al., 2015; Yu et al., 2016; Xie et
80 al., 2018). In May 2012, the concentrations of DMAH^+ and TMAH^+ over the Yellow Sea (YS) and Bohai Sea even reached
81 4.4 and 7.2 nmol m^{-3} , which was 1–3 orders of magnitude higher than those reported in other oceanic regions (Hu et al., 2015).
82 These extremely high concentrations were thought to be associated with high biological activities. In spite of these field studies,

83 the long-term observation of aminiums over the coastal sea and quantitative estimate of the contribution of marine biogenic
84 source to aerosol aminiums are still lacking.

85 In this study, the aminiums over a coastal megacity (Shanghai), a nearby island (Huaniao Island) and marginal seas (the Yellow
86 Sea and East China Sea, YECS) were measured. The relationships between aminium concentrations and environmental factors
87 were systematically analyzed. The size distributions of aminiums were investigated with the speculation of main formation
88 pathways. Besides, the dominant sources determining the concentrations and ratios between aminium species were elucidated,
89 and the contributions of terrestrial anthropogenic and marine biogenic sources to aminiums were quantitatively estimated. Our
90 results will be a great help for understanding the chemical properties, reaction pathways and sources of aerosol aminiums over
91 the coastal area and the ocean.

92 **2 Sampling and Analysis**

93 **2.1 Aerosol sampling**

94 The sampling site in Shanghai was located on top of the No.4 teaching building of Fudan University (31.30° N, 121.50° E)
95 (Fig. 1). This site is affected by the school, residential, commercial and traffic activities and can be representative of coastal
96 cities. Particulate matters with an aerodynamic diameter less than 2.5 μm ($\text{PM}_{2.5}$) were simultaneously collected by two
97 medium-flow samplers (100 L min^{-1} , HY-120B, Hengyuan) using a 90 mm pre-combusted quartz filter (Whatman) and a
98 cellulose filter (Grade 41, Whatman), respectively. A total of 131 samples were collected within four seasons with the sampling
99 duration around 24 hours (spring: 25 Mar. – 26 Apr., 2013; summer: 16 Jul. – 17 Aug., 2013; autumn: 30 Oct. – 30 Nov., 2013;
100 winter: 1 Dec., 2013 – 23 Jan., 2014) (Table 1).

101 Aerosols were also collected at Huaniao Island (HNI, 30.86° N, 121.67° E) which was about 80 km away from Shanghai in
102 the East China Sea (ECS) (Fig. 1). The locally anthropogenic emissions were negligible, but the site was affected by the
103 terrestrial transport and the ship emission from nearby container ports (Wang et al., 2016; Wang et al., 2018). $\text{PM}_{2.5}$ samples
104 were collected in the summer of 2016 (4 Aug. – 18 Aug.) and size-segregated samples were obtained using a 10-stage Micro-
105 Orifice Uniform Deposit Impactor (30 L min^{-1} , MOUDI, MSP Model 110-NR) and 47 mm PTFE filters (Zeflour, PALL) in
106 the autumn of 2016 (12 Nov. – 3 Dec.), the spring of 2017 (11 Mar. – 19 Mar.) and the early and late summer of 2017 (22 Jun.
107 – 9 Jul. and 27 Aug. – 12 Sep., respectively) (Table 1). The 50% cutoff diameters for 10 stages were 18, 10, 5.6, 3.2, 1.8, 1.0,
108 0.56, 0.32, 0.18, 0.10 and 0.056 μm , and the sampling durations were 24-48 hours.

109 The size-segregated samples were also collected over the YECS onboard research vessel (R/V) *Dong Fang Hong II* in the
110 spring of 2017. The cruise started from Qingdao on March 27 and returned on April 15 (Fig. 1), and a total of 9 sets of samples
111 were obtained.

112 **2.2 Chemical analysis**

113 One fourth of $\text{PM}_{2.5}$ and half of MOUDI sample filters were cut and placed into a polypropylene jar (Nelgene) with 15 mL and
114 20 mL of ultrapure water (18.25 $\text{M}\Omega\ \text{cm}^{-1}$) respectively for 40 min ultrasonic extraction. The extract was filtered through a
115 0.45 μm PTFE filter (Jinteng) and stored at 4 °C for ion measurement. Ion Chromatograph (DIONEX ICS-3000, Thermo-
116 Fisher) assembled with AG11-HC and AS11-HC was used to determine anions including Cl^- , NO_3^- , SO_4^{2-} , HCOO^- ,
117 methanesulfonate (MSA), malonate, succinate, glutarate, maleate and $\text{C}_2\text{O}_4^{2-}$. The columns CG17 and CS17 were used to
118 measure inorganic cations including Na^+ , NH_4^+ , K^+ , Mg^{2+} and Ca^{2+} and aminiums. Six aminiums including DMAH^+ ,
119 TMAH^+ + DEAH^+ , propylaminium (MPAH^+), triethylaminium (TEAH^+), ethanolaminium (MEOAH^+) and triethanolaminium
120 (TEOAH^+) could be effectively separated and measured using the IC method. The MMAH^+ and MEAH^+ in the aerosols could
121 not be quantified because their peaks were obscured by the wide and distorted peak of NH_4^+ . It should be noted that TMAH^+
122 and DEAH^+ could not be completely separated using the IC system (VandenBoer et al., 2012; VandenBoer et al., 2011; Zhou

123 et al., 2018; Huang et al., 2014). Therefore, the sum of TMAH⁺ and DEAH⁺ concentrations (named as TMDEAH⁺) were
124 quantified using the calibration curve of TMAH⁺ with errors less than 3% (Zhou et al., 2018). With the sampling volumes of
125 144 and 86 m³ for PM_{2.5} and MOUDI samples respectively, the detection limits of DMAH⁺, TMDEAH⁺, TEAH⁺, MPAH⁺,
126 MEOAH⁺ and TEOAH⁺ were 0.55, 0.78, 1.93, 2.59, 1.94 and 4.96 ng m⁻³ for PM_{2.5} samples and 0.20, 0.29, 0.71, 0.95, 0.56
127 and 1.82 ng m⁻³ for samples collected in each MOUDI stage. MPAH⁺, MEOAH⁺ and TEOAH⁺ were rarely detected in the
128 aerosol samples (<10%) and thereby not reported in this study. The detailed information about analysis of aminiums were
129 given in Zhou et al. (2018).

130 One fourth of PM_{2.5} cellulose sample filter was cut and digested with 7 mL of HNO₃ and 1 mL of HF (both acids were purified
131 from GR using a sub-boiling system) at 185 °C for 30 min in a microwave digestion system (MARS5 Xpress, CEM). An
132 Inductively Coupled Plasma Optical Emission Spectroscopy (ICP-OES, SPECTRO) was used for determining elements Al,
133 Ca, Fe, Na, P, S, Cu, K, Mg, Mn, Zn, As, Ba, Cd, Ce, Co, Cr, Mo, Ni, Pb, Ti, and V. The detailed procedures refer to Wang et
134 al. (2016).

135 2.3 Auxiliary data

136 The 3-hour resolution meteorological data of Baoshan station in Shanghai (WMO index: 58362) were obtained from the
137 National Climatic Data Center (NCDC, <https://www.ncdc.noaa.gov/isd>). The 10-second resolution meteorological data were
138 recorded by a shipborne meteorological station during the cruise. The planetary boundary layer height (BLH) and 6-hour
139 accumulated precipitation (TPP6) were extracted from NCEP's Global Data Assimilation System Data (GDAS). The daily
140 concentrations of gaseous pollutants (SO₂, CO, NO₂ and O₃) averaged from 9 real-time monitoring stations of Shanghai were
141 obtained from the Shanghai Environmental Monitoring Center (<http://www.semcc.gov.cn/aqi/home/DayData.aspx>).

142 Three-day air mass backward trajectories were calculated using a Hybrid Single-Particle Lagrangian Integrated Trajectory
143 (HYSPPLIT) model (<http://ready.arl.noaa.gov/HYSPLIT.php>) with the starting height at 100 meters.

144 3 Results and discussion

145 3.1 Seasonal and spatial variations of aminium concentrations

146 The mean concentrations of NH₄⁺ and aminiums in each campaign of this study and reported in literatures were listed in Table
147 2. It should be noted that TEAH⁺ concentrations over Huaniao Island and the YECS were mostly below the detection limits
148 (<DL). For other aminiums and TEAH⁺ over Shanghai, the number of samples below detection limits were generally less than
149 30%. These undetectable concentrations were considered to be zero for the calculation of means and standard deviations. Three
150 aminiums, DMAH⁺, TMDEAH⁺ and TEAH⁺, were commonly detected in the aerosol samples collected from Shanghai. The
151 most abundant aminiums were DMAH⁺ and TEAH⁺ with their annual means of 15.6 and 16.0 ng m⁻³, respectively. By
152 comparison, the average TMDEAH⁺ concentration (4.4 ng m⁻³) was significantly lower. All three aminiums showed the highest
153 concentrations in winter and the lowest levels in spring (DMAH⁺) and summer (TMDEAH⁺ and TEAH⁺), which generally
154 agreed with the seasonal trends of PM_{2.5} and NH₄⁺ concentrations in Shanghai (Figure 2). Specifically, the average TEAH⁺
155 reached 35.2 ng m⁻³ in winter in Shanghai, about 40 times as much as that in summer. TEAH⁺ was mostly below the detection
156 limit in the aerosols collected over Huaniao Island and the YECS, suggesting its dominant land sources and negligible marine
157 contribution. By contrast, the average DMAH⁺ and TMDEAH⁺ concentrations (14.0 and 13.2 ng m⁻³) over Huaniao Island
158 were close to and significantly higher than those over Shanghai, respectively. Similarly high concentrations of DMAH⁺ and
159 TMDEAH⁺ (11.9 and 14.6 ng m⁻³) were also observed over the YECS (Fig. 2 and Table 2), suggesting that the two aminiums
160 might have notable marine sources. Accordingly, both species reached the highest levels during the summer campaigns in 2017
161 at Huaniao Island, consistent with the highest primary productivity in the coastal ECS and prevailing winds from the ocean in
162 summer. As a major component of fine particles over eastern China with similar chemical properties to aminiums, NH₄⁺ was

163 mainly from terrestrial sources and its concentrations over Huaniao Island and YECS were much lower than those over
164 Shanghai (Fig. 2).

165 Our measurement of DMAH⁺ in Shanghai was comparable to those previously reported from the urban sites, but generally
166 higher than those measured in the forest areas of Toronto (VandenBoer et al., 2012), Hyytiälä (Hemmilä et al., 2018) and
167 Guangdong (Liu et al., 2018a). This implies that anthropogenic activities may be crucial sources of DMAH⁺ in the urban
168 atmosphere. The TMDEAH⁺ concentrations in our study were much lower than those reported by Tao et al. (2016) in Shanghai.
169 Their sampling location was close to the residential areas and could be influenced by the local sources such as human excreta
170 emission (Zhou et al., 2018). The aerosol TEAH⁺ concentrations in China were reported in our study for the first time and
171 could not be compared to previous work. According to previous measurement results for gaseous amines in the same site from
172 July 25 to August 25 in 2015, the average mass concentrations of C2-, C3- plus C4-, and C6-amines were 80.4, 53.1 and 15.8
173 ng m⁻³, respectively (Yao et al., 2016). The order of concentrations was consistent with that of the corresponding aerosol
174 aminiums in the summer of 2013 which was DMAH⁺ > TMDEAH⁺ > TEAH⁺ (9.1, 1.7 and 0.9 ng m⁻³, respectively) in this
175 study. Based on these concentrations, the ratios of each amine vs. aminium were roughly calculated and C2-amines/DMAH⁺,
176 (C3- plus C4-amines)/TMDEAH⁺ and C6-amines/TEAH⁺ were 8.8, 30.1 and 17.9, respectively. These values were comparable
177 to the ratio of total amines to total aminiums (14.9) over a mountain site in southern China (Liu et al., 2018a). Except for the
178 three aminiums commonly detected in this study, MMAH⁺ and MEAH⁺ (Liu et al., 2018a; Ho et al., 2015; Shen et al., 2017)
179 were other abundant aminiums detected in the urban site.

180 Aerosols were collected using a MOUDI over Huaniao Island and the YECS. Aminiums in PM_{1.8} of the MOUDI samples were
181 compared to those of PM_{2.5}, since MOUDI does not have the 50% cutoff diameter of 2.5 μm and aminiums in PM_{1.8} accounted
182 for over 60% concentrations of the whole size range of aerosols. Our measurements of aminiums over Huaniao Island and the
183 YECS were comparable to those previously observed over the eastern China seas (Hu et al., 2015; Yu et al., 2016; Xie et al.,
184 2018), but they were apparently higher than many other oceanic regions such as Arabian Sea (Gibb et al., 1999) and Cape
185 Verde (Müller et al., 2009). The high aminiums over the YECS were probably associated with the severe air pollution in eastern
186 China as well as the high ocean productivity in marginal seas.

187 3.2 Environmental factors affecting aminium concentrations

188 3.2.1 Boundary layer height (BLH)

189 The concentrations of PM_{2.5}, NH₄⁺ and three aminiums sampled in Shanghai in 2013 dropped significantly when the BLH
190 increased from 200 m to 500 m and then slowly decreased with the further increase of BLH (Fig. 3a and Fig. S1), due to the
191 enhanced ventilation. Specifically, the concentrations of DMAH⁺, TMDEAH⁺ and TEAH⁺ (58.4, 13.9 and 80.5 ng m⁻³) in
192 Shanghai reached the maximum along with PM_{2.5} (447 μg m⁻³) during the severe haze event between 30 Nov. and 8 Dec. 2013,
193 when the average BLH and wind speed were 298 m and 1.35 m s⁻¹, respectively (Fig. S2). By comparison, the average
194 concentrations of DMAH⁺, TMDEAH⁺ and TEAH⁺ (8.9, 4.0 and 10.1 ng m⁻³) were much lower prior to the haze event (on 26-
195 29 Nov 2018) associated with the higher BLH (636.4 m) and wind speed (2.73 m s⁻¹). Thus, the generally stagnant meteorology
196 in winter (Liu et al., 2013) could cause a substantial accumulation of aerosol aminiums and lead to the seasonal variation of
197 aminiums in Shanghai.

198 3.2.2 Temperature

199 To eliminate the synchronous change of aminiums and NH₄⁺ with PM_{2.5}, the mass ratios of aminiums to PM_{2.5} (aminiums/PM_{2.5})
200 and NH₄⁺ to PM_{2.5} (NH₄⁺/PM_{2.5}) were applied for analysis. These ratios were found to be negatively correlated with air
201 temperature in Shanghai (Fig. 3b). Similar to NH₄⁺, aminiums combined with NO₃⁻, Cl⁻ and organic acids are semi-volatile
202 and can dissociate in the atmosphere (Tao and Murphy, 2018). So the negative correlations may be explained by the movement

203 of gas-particle partitioning equilibrium to the gas phase at higher temperatures (Ge et al., 2011a). This is consistent with the
204 previous observation that the proportion of particles containing aminiums in the urban area of Shanghai was much higher in
205 winter (23.4%) than that in summer (4.4%) (Huang et al., 2012). The seasonal variation of temperature may also lead to the
206 change of concentrations of aerosol aminiums. It should be pointed out that environmental variables like BLH and temperature
207 are constantly changing with time and their impacts on aminium concentrations may vary within the sampling duration (24 or
208 48 hours). However, these variables must be averaged over the same time interval as aminium concentrations. This analysis
209 may eliminate the instant discordance and improve the correlations between environmental variables and aminiums or
210 aminiums/PM_{2.5}, and the results could well explain the seasonal variation of aminiums.

211 3.2.3 Oxidizing capacity

212 As gaseous amines can be oxidized by oxidants such as ·OH, O₃ and NO₃· in the atmosphere before partitioning into the
213 particulate phase (Ge et al., 2011b; Nielsen et al., 2012; Yu and Luo, 2014), aminium concentrations in aerosols may decrease
214 with the enhanced atmospheric oxidizing capacity. Ozone concentration can represent oxidizing capacity of the lower
215 atmosphere (Thompson, 1992). Here the relationship between aminium/NH₄⁺ ratios and O₃ was examined, because the
216 formation of particulate aminiums and NH₄⁺ were both temperature-dependent and using their ratios could avoid the
217 temperature effect to some extent. Besides, the residence time of NH₃ in the atmosphere due to the oxidation reaction is about
218 72.3 days (Ge et al., 2011b), and therefore NH₄⁺ concentrations in aerosols should not be affected by O₃. Negative correlations
219 were found between aminiums/NH₄⁺ and O₃ concentrations in Shanghai during the summer of 2013 (Fig. 3c). In other seasons,
220 the correlations were not obvious, especially in winter when O₃ concentrations were the lowest and neither of aminiums/NH₄⁺
221 was correlated with O₃ (Figure S3). In general, atmospheric oxidizing capacity is the strongest in summer (Logan, 1985; Liu
222 et al., 2010), and the results verified that high oxidizing capacity in summer may reduce the formation of particulate aminiums
223 by oxidizing gaseous amines. It was consistent with the diurnal pattern of gaseous amines with the lowest values at noon and
224 the negative correlations between the concentrations of amines and O₃ observed in Shanghai during the summer of 2015 (Yao
225 et al., 2016). It should be noted that there was no significant variation in temperature and little rainfall during the sampling
226 periods in the summer of 2013. In other seasons, due to the relatively weak photochemistry and more complex sources and
227 meteorology, other factors except oxidizing capacity played more important roles in affecting aerosol aminiums.

228 3.2.4 Relative humidity and fog processing

229 In the spring of 2017 over the YECS, although the sample of 4–5 Apr. was influenced by high Chl-a concentrations and low
230 BLH, the concentrations of DMAH⁺ and TMDEAH⁺ (13.3 and 17.4 ng m⁻³) were about half of those on 7–9 Apr. (Fig. 4). This
231 was probably due to the intense fog event occurred on 7–9 Apr. with relative humidity >90%, which could enhance the gas-to-
232 particle partitioning of amines. The enhancement of TMA gas to particles by cloud and fog processing has been observed in
233 both field and laboratory simulations (Rehbein et al., 2011). It was also found that the number fraction of TMA-containing
234 particles dramatically increased from ~7% in clear days to ~35% in foggy days and number-based size distribution of TMA-
235 containing particles shifted towards larger mode, peaking at the droplet mode (0.5–1.2 μm) in Guangzhou (Zhang et al., 2012).
236 The investigation over the Yellow and Bohai seas in the summer of 2015 found significantly positive correlations between the
237 concentrations of DMAH⁺ and TMAH⁺ and relative humidity (Yu et al., 2016). Therefore, fog and high relative humidity (RH)
238 are also favorable conditions for gas-to-particle conversion of amines.

239 3.3 Size distributions and formation pathways of aerosol aminiums

240 The aminiums were mainly distributed in fine aerosols with diameter less than 1.8 μm, and the mass percentages of DMAH⁺
241 and TMDEAH⁺ in the coarse mode were around 36% in the autumn of 2016 at Huaniao Island and less than 15% in all other
242 campaigns at Huaniao Island and over the YECS (Fig. 5a-d). This is consistent with the previous reports that >70% of aminiums

243 were distributed in fine particles over Shanghai during the summer of 2013 (Tao et al., 2016) and over the western North
244 Pacific and its marginal seas (Xie et al., 2018). The aminiums mostly demonstrated a bimodal distribution in the autumn and
245 early summer campaigns at Huaniao Island with peaks at 0.18–0.32 μm (condensation mode) and 0.56–1.0 μm (droplet mode).
246 This is similar to the size distributions of DMAH⁺ and TMDEAH⁺ observed in Shanghai (Tao et al., 2016) and to NH₄⁺ and
247 non-sea-salt (nss-SO₄²⁻) in all campaigns over Huaniao Island and the YECS (Fig. S4-5). The size distribution suggests that
248 the gas-to-particle condensation (condensation mode) and cloud processing (droplet mode) seem to be major mechanisms for
249 the formation of aminiums and other secondary species NH₄⁺ and nss-SO₄²⁻.

250 In order to compare the contributions between condensation and cloud processing to the formation of specific species, the ratio
251 of its concentrations in droplet mode (0.56–1.0 μm) to condensation mode (0.18–0.32 μm) was calculated (denoted as α). It
252 could be seen that the α values of NH₄⁺ and nss-SO₄²⁻ were significantly greater than 1, especially in the case of high
253 concentrations, indicating that the cloud processing probably determined the concentrations of these species (Fig. 6).
254 Differently, aminiums had α values around 1, suggesting that condensation and cloud processing might be equally important
255 to the formation of aminiums.

256 In late summer at Huaniao Island and the spring cruise over the YECS when air masses were mainly from oceanic regions (see
257 Sect. 3.4.3), the aminiums generally exhibited a unimodal distribution with one wide peak at 0.18–1.0 μm due to the increased
258 concentrations at 0.32–0.56 μm (Fig. 5e-h). The concentrations of NH₄⁺ and nss-SO₄²⁻ also showed a significant elevation in
259 the size range of 0.32–0.56 μm during these periods. The deviation of MOUDI cutoff diameters during the sampling could be
260 ruled out because the concentrations of particulate matter always presented a trimodal distribution with peaks at 0.18–0.32 μm ,
261 0.56–1.8 μm and 3.2–10 μm . The unimodal distributions of aminiums with the peak at 0.18–1.0 μm have been widely reported
262 over the eastern China seas (Hu et al., 2015; Yu et al., 2016; Xie et al., 2018). This suggests that the formation mechanisms of
263 aerosol aminiums over the ocean may be different from that over the land. It was indicated that the high concentration and
264 unique size distribution of TMAH⁺ observed over the oligotrophic western North Pacific were mainly attributed to the primary
265 TMAH⁺ in sea-spray aerosols (Hu et al., 2018). In addition, some studies have demonstrated that artificially generated sea
266 spray aerosols and actual primary marine aerosol both contained amines/aminiums (Bates et al., 2012; Frossard et al., 2014;
267 Dall'Osto et al., 2019). So we speculate that the elevated concentrations of aminiums at 0.32–0.56 μm over the eastern China
268 seas may be also associated with the increased concentration of sea-spray aerosols which contain substantial primary aminiums.
269 In other hand, the heterogeneous formation of secondary aminiums on the surface of sea spray aerosols cannot be ruled out
270 (Yu et al., 2016).

271 3.4 Sources of aerosol aminiums

272 3.4.1 Anthropogenic sources on the land

273 Correlation analysis was carried out between aminiums, other PM_{2.5} components and gaseous pollutants measured in Shanghai
274 (Fig. 7). It can be seen that the secondary inorganic components SO₄²⁻, NO₃⁻ and NH₄⁺ (SNA), PM_{2.5} and DMAH⁺ were
275 significantly correlated with each other with the correlation coefficients above 0.6. This suggests that anthropogenic sources
276 may have a great contribution to the atmospheric DMA in Shanghai, which is consistent with previous findings in Nanjing
277 (Zheng et al., 2015). Considering the unique role of DMA in new particle formation (Almeida et al., 2013), our results re-
278 enforce that the frequent new particle formation events observed in extremely polluted Chinese cities are indeed, at least in
279 part, due to amines (Yao et al., 2018). The correlations between TEAH⁺ and SNA were relatively weak, but TEAH⁺ was found
280 to be significantly correlated with the components mainly from industrial sources (represented by the high concentrations of
281 K, Mn, Cd, Pb, Zn, and Cl⁻) (Tian et al., 2015; Liu et al., 2018b), indicating that the industrial emission could be an important
282 source of TEA. It was consistent with the observation result in a suburban site that gaseous C₄- to C₆-amines had some abrupt
283 and frequent increase in the night and may be caused by some local emissions (You et al., 2014). Compared to the DMAH⁺

284 and TEAH⁺, TMDEAH⁺ showed much weaker correlations with the anthropogenically derived components. Weak correlations
285 were also found between all the aminiums and V, Ni, Al, Mg, Ca and Fe, suggesting that ship emission (traced by V and Ni)
286 and soil dust (represented by Al, Ca and Fe) were not main sources of aminiums in PM_{2.5} over Shanghai.

287 **3.4.2 Marine biogenic source**

288 As discussed in Sect. 3.1, the relatively high concentrations of DMAH⁺ and TMDEAH⁺ over Huaniao Island and the YECS
289 implied that the marine sources contributed substantially to these two aminiums. Accordingly, a spatial variation of aminium
290 concentrations was observed over the YECS during the spring cruise. The concentrations of DMAH⁺ and TMDEAH⁺ increased
291 by a factor of 3–5 in the southern ECS (average 24.4 and 40.3 ng m⁻³ for the samples of 7–11 Apr. respectively) compared to
292 the YS and northern ECS (average 7.0 and 8.4 ng m⁻³ for the samples of 27 Mar.–5 Apr. respectively) (Fig. 8). This is consistent
293 with the noticeable difference of Chl-a concentrations between the southern and northern YECS (2.3 times higher in southern
294 YECS than that in northern YECS, unpublished data). Furthermore, the highest TMDEAH⁺ and lowest NH₄⁺ concentrations
295 observed on 7–11 Apr. corresponded to the air-mass back trajectories originating from the ocean, suggesting that the metabolic
296 activities of surface plankton in the high-productive seas could be a strong source of amines as previously reported (Facchini
297 et al., 2008; Müller et al., 2009; Sorooshian et al., 2009; Hu et al., 2015). Differently, the high concentrations of aminiums
298 observed on 14 Apr. near Qingdao was affected by the air masses transported from eastern China (Fig. 8) and thereby
299 contributed mainly by terrestrial sources.

300 Fine-mode NH₄NO₃ could decompose during its transport from the land to the ocean, and the released HNO₃ gas would react
301 with dust and sea salt aerosols to form coarse-mode NO₃⁻. Therefore, negative correlations were observed between the
302 concentrations of fine-mode NO₃⁻ and alkaline species (Na⁺+Ca²⁺) over the East Asia (Bian et al., 2014; Uno et al., 2017).
303 Since only one dust event was encountered on 12–13 Apr. during the cruise (unpublished data), the coarse-mode NO₃⁻ in this
304 study should be mostly formed by the heterogeneous reaction with sea salts. Therefore, the importance of terrestrial transport
305 to marine aerosols could be roughly estimated by the percentage of NO₃⁻ in the fine mode. For aerosols collected on 29–31
306 Mar., 4–5 Apr., 7–9 Apr. and 9–11 Apr., over 2/3 concentrations of NO₃⁻ were in the coarse mode (>1.8 μm, Fig. 9a). These
307 samples should be less affected by the terrestrial air masses (referred to category 1) compared to other samples (referred to
308 category 2), and the analysis was consistent with the forward directions of air mass trajectories (Fig. S6). Aminiums were
309 negatively correlated with NH₄⁺ for Category 1 samples suggesting that aminiums were probably dominated by marine
310 biogenic sources whereas NH₄⁺ was influenced by terrestrial transport (Fig. 9b). For Category 2 samples, a positive correlation
311 was found between aminiums and NH₄⁺, indicating that terrestrial sources could contribute significantly to aminiums over the
312 YECS in these cases (Fig. 9c).

313 **3.4.3 Source contributions to aminiums over the coastal sea**

314 Huaniao Island is located in the frontline of terrestrial transport to the ECS and influenced by the air masses from the land or
315 ocean depending on the seasonal variation of prevailing winds. Significantly positive correlations were found between the
316 concentrations of aminiums and NH₄⁺ in the autumn but not in the summer of 2016 or in late summer of 2017 (Fig. 10).
317 Accordingly, the majority of backward trajectories pointed to the northern China in autumn whereas air masses predominantly
318 originated from the ECS in summer (Fig. 11). Meanwhile, NO₃⁻ demonstrated a tri-modal distribution with three peaks at
319 0.18–0.32 μm (condensation mode), 0.56–1.0 μm (droplet mode) and 3.2–5.6 μm (coarse mode) in autumn but only one peak
320 at 3.2–5.6 μm in late summer of 2017 (Fig. S7). These implies that terrestrial transport could be a dominant source for aminiums
321 over the coastal ECS in autumn while marine sources were dominant in late summer. In early summer of 2017, the mass ratios
322 of aminiums to NH₄⁺ were significantly lower on 26–28 Jun. than those on other days (Fig. S8), corresponding to different
323 origins and properties of the air masses. Removing the data measured on 26–28 Jun., we found a significantly positive
324 correlation between the concentrations of DMAH⁺ and NH₄⁺ but not between TMDEAH⁺ and NH₄⁺. This suggests that DMAH⁺

325 and TMDEAH⁺ may be predominantly derived from terrestrial and marine sources, respectively.
 326 Good positive correlations were generally found between the concentrations of TMDEAH⁺ and DMAH⁺ over Huaniao Island
 327 and the YECS, and the slope for autumn samples dominated by terrestrial sources was significantly lower than those influenced
 328 primarily by marine air masses (e.g. late summer at Huaniao Island and spring over the YECS, Fig. 12). The highest slope of
 329 TMDEAH⁺ vs. DMAH⁺ (1.98) occurred in the summer of 2016 which was also mainly affected by marine sources. Therefore,
 330 it is speculated that aminiums derived from marine biogenic source might have significantly higher TMDEAH⁺ to DMAH⁺
 331 ratios than those from terrestrial sources. Similarly, Hu et al. (2015) observed a significant correlation between the TMDEAH⁺
 332 and DMAH⁺ concentrations over the Yellow Sea with the slope of 1.27–2.49. In early summer of 2017, the weak correlation
 333 between the DMAH⁺ and TMDEAH⁺ and very low slope (0.29) suggested the mixing of terrestrial and marine influence on
 334 aminiums over Huaniao Island during that period as discussed above.

335 Dimethylsulfide (DMS) produced in seawater by the metabolism of plankton will be released into the atmosphere, and SO₂,
 336 MSA, SO₄²⁻ and other products can be formed through a series of oxidation reactions (Saltzman et al., 1985; Charlson et al.,
 337 1987; Faloon, 2009; Barnes et al., 2006). MSA is often used as a tracer of marine biogenic source to calculate the marine
 338 biogenic contribution to nss-SO₄²⁻ (Yang et al., 2009; Yang et al., 2015). Therefore, the mass ratio of MSA to nss-SO₄²⁻
 339 (MSA/nss-SO₄²⁻) can be used to indicate the contribution of marine sources to relevant aerosol components. A significantly
 340 linear relationship was found between aminium/NH₄⁺ and MSA/nss-SO₄²⁻ for the samples collected in the autumn of 2016 and
 341 summer of 2017 over Huaniao Island (Fig. 13). The value of aminium/NH₄⁺ increased with the increasing contribution of
 342 marine sources to the aerosol aminium. When the marine biogenic source contribution is 0, the corresponding aminium/NH₄⁺
 343 values (*b* in Eq. (3)) represent the average ratios completely contributed by terrestrial sources. By multiplying the ratios by
 344 NH₄⁺ concentrations, the aerosol aminiums contributed by terrestrial sources can be calculated (Eq. (4)). Therefore, the
 345 contributions of terrestrial and marine sources to aerosol aminiums can be quantitatively estimated.

$$346 \left(\frac{[\text{aminium}]}{[\text{NH}_4^+]} \right)_{\text{terrestrial}} = k \times \left(\frac{[\text{MSA}]}{[\text{nss} - \text{SO}_4^{2-}]} \right)_{\text{terrestrial}} + b \quad (3)$$

$$347 [\text{aminium}] = \left(\frac{[\text{aminium}]}{[\text{NH}_4^+]} \right)_{\text{terrestrial}} \times [\text{NH}_4^+] + [\text{aminium}]_{\text{marine}} \quad (4)$$

348 where *k* and *b* are the slope and intercept of the linear fitting equation of $[\text{aminium}]/[\text{NH}_4^+]$ and $[\text{MSA}]/[\text{nss} - \text{SO}_4^{2-}]$,
 349 respectively (Fig. 13).

350 Although most of MSA comes from marine sources, the terrestrial sources may also have a certain contribution (Yuan et al.,
 351 2004). Therefore, MSA/nss-SO₄²⁻=0 was not used as the end member value for calculating the terrestrial contribution. We have
 352 simultaneously measured MSA and nss-SO₄²⁻ in a total of 64 total suspended particle (TSP) samples collected in the autumn
 353 of 2016 and the summer of 2017. The retention percentage of air mass over the land (R_L) was calculated for each sample based
 354 on three-day backward trajectories (Figure S9 and see supplementary text for more information). Samples with the largest 10%
 355 R_L values (n=7, R_L>74%) were considered to be terrestrial-dominant with the average MSA/nss-SO₄²⁻ (± 1 standard deviation)
 356 of 0.0021±0.0013. Therefore, this value was regarded as the end member value of terrestrial MSA/nss-SO₄²⁻ in these seasons.
 357 Substituting it into the previous fitting equation, the values of $([\text{DMAH}^+]/[\text{NH}_4^+])_{\text{terrestrial}}$ and $([\text{TMDEAH}^+]/$
 358 $[\text{NH}_4^+])_{\text{terrestrial}}$ were 0.0068 (0.0038–0.0105) and 0.0034 (0.00005–0.0076), respectively. Then the average contributions
 359 of terrestrial and marine sources to the two aminiums in each campaign were calculated and shown in Table 3. It can be seen
 360 that the average terrestrial contributions to DMAH⁺ and TMDEAH⁺ in aerosols were both more than 60% in autumn, higher
 361 than those in summer. The contributions of marine sources during late summer of 2017 (63.0% for DMAH⁺ and 78.3% for
 362 TMDEAH⁺) were higher than those in early summer (53.3% for DMAH⁺ and 74.2% for TMDEAH⁺), which was consistent
 363 with previous hypothesis. Furthermore, the contribution of marine sources was greater to TMDEAH⁺ than to DMAH⁺ in all
 364 campaigns, which corresponded to the higher ratio of TMDEAH⁺/DMAH⁺ in the samples influenced primarily by marine air
 365 masses (Fig. 12). It should be pointed out that aminium/NH₄⁺ ratios could vary with the chemistry of aerosols due to slightly
 366 different gas-to-particle partitioning of the amines and NH₃ (Chan and Chan, 2013; Pankow, 2015; Xie et al., 2018) and marine
 367 aminiums may also partially originated from primary source as discussed above. Therefore, our discussion is constrained on

368 the source analysis of aerosol aminiums, but not gaseous or total amines (gaseous amines + aerosol aminiums). Although NH_4^+
369 was mainly derived from the land, marine sources may also had a certain contribution (Altieri et al., 2014; Paulot et al., 2015).
370 This was neglected in our calculation and might lead to the overestimate of terrestrial contributions to aminiums. Besides, the
371 relatively small number of data points used in the fitting (25 points) and the treatment of $([\text{aminium}]/[\text{NH}_4^+])_{\text{terrestrial}}$ as a
372 fixed value ignoring its variation would cause uncertainty in the results. Nonetheless, our method is the first attempt to calculate
373 the contributions of marine biogenic and terrestrial sources to aerosol aminiums over the coastal sea, which will provide an
374 insight of sources and roles of amines in the atmosphere.

375 4 Conclusion

376 Amines in the atmosphere play an important role in new particle formation and subsequent particle growth, and studying
377 aerosol aminiums can provide insight into the sources, reaction pathways and environmental effects of amines. An integrated
378 observation was conducted on aerosol aminiums (DMAH^+ , TMDEAH^+ and TEAH^+) in a coastal city (Shanghai), a nearby
379 island (Huaniao) and the marginal seas (the YECS). All three aminiums exhibited significant seasonal variation in Shanghai
380 with their highest concentrations in winter, which was consistent with relatively severe air pollution associated with the winter
381 monsoon (continental winds) and the lowest BLH and temperature in this season. Atmospheric oxidizing capacity and
382 relatively humidity may also influence the concentrations of aerosol aminiums to some extent by oxidizing gaseous amines
383 and enhancing the gas-particle partitioning, respectively. By comparing the ocean sites to Shanghai, similar concentrations of
384 DMAH^+ and 3-fold higher TMDEAH^+ were observed suggesting that these two aminiums may have significant marine sources.
385 By contrast, TEAH^+ was abundant in Shanghai but it was below the detection limit over Huaniao Island and the YECS,
386 implying its terrestrial origin.

387 Aminiums influenced substantially by terrestrial transport showed a bimodal distribution with two peaks at 0.18–0.32 μm
388 (condensation mode) and 0.56–1.0 μm (droplet mode), suggesting that the gas-to-particle condensation and cloud processing
389 were main formation pathways for aerosol aminiums. Nonetheless, aminiums demonstrated a unimodal distribution with a
390 wide peak at 0.18–1.0 μm over the YECS and in late summer of Huaniao Island, and the elevated concentration at 0.32–0.56
391 μm might be related to sea-spray aerosols that either contain primary aminiums or provide surface for heterogeneous reactions
392 to form secondary aminiums. This indicates that aminiums in marine aerosols may undergo different formation pathways from
393 those on the land.

394 We distinguished the contributions of terrestrial and marine sources to aerosol aminiums for the first time by taking the mass
395 ratio of MSA to nss-SO_4^{2-} as an indicator of marine biogenic sources. In the autumn of 2016, the contributions of terrestrial
396 sources to aminiums over Huaniao Island were estimated to be more than 60%. By contrast, marine biogenic sources dominated
397 aminium concentrations especially for TMDEAH^+ (~80%) in the summer of 2017. Our results indicated that marine biogenic
398 emission of amines could not be ignored in the eastern coast of China, especially in summer. Therefore, it is necessary to add
399 this source into the emission inventory of amines and recent modelling of amines over eastern China without marine source
400 (Mao et al., 2018) may result in significant deviations. Besides, the role of amines in new particle formation over the open
401 ocean is likely to be more important due to much less pollutants compared to the coastal area, which should be further studied.

402
403 *Data availability.* Data are available from the corresponding author on request (yingchen@fudan.edu.cn).

404
405 *Author contribution.* SZ, YC and CD conceived the study. SZ, YC and CD wrote the paper. SZ, HL, and JX collected the
406 samples. SZ, TY and JX performed the measurement. All have contributed to review of the manuscript.

407
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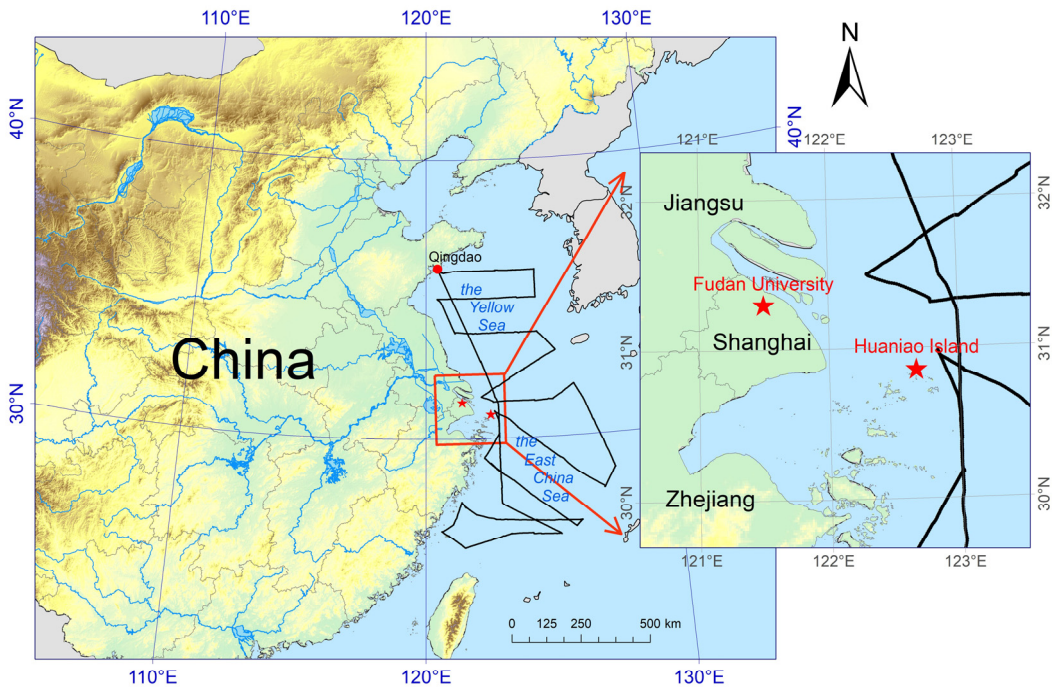
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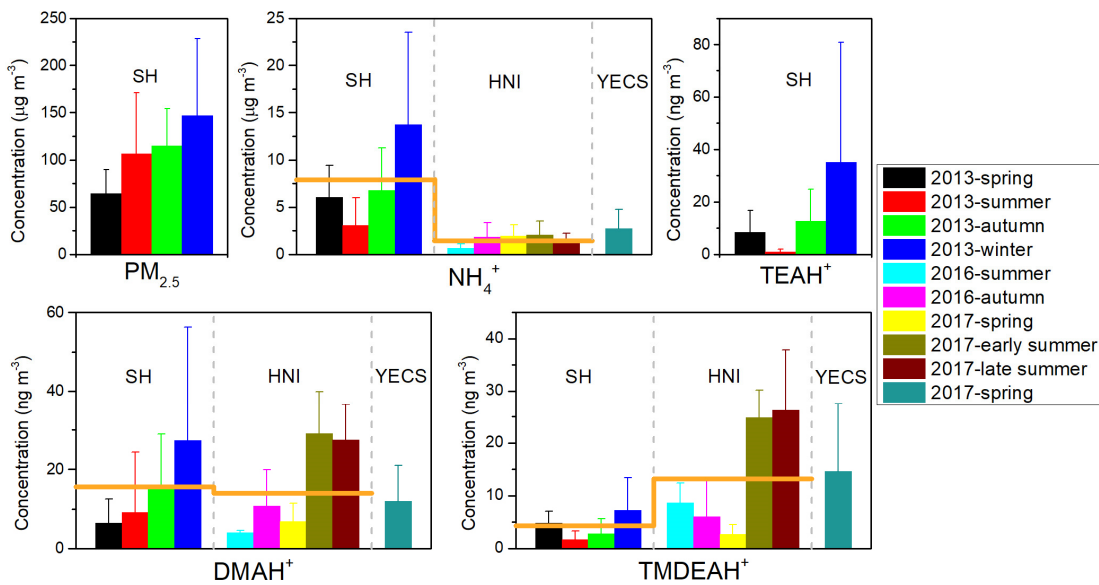
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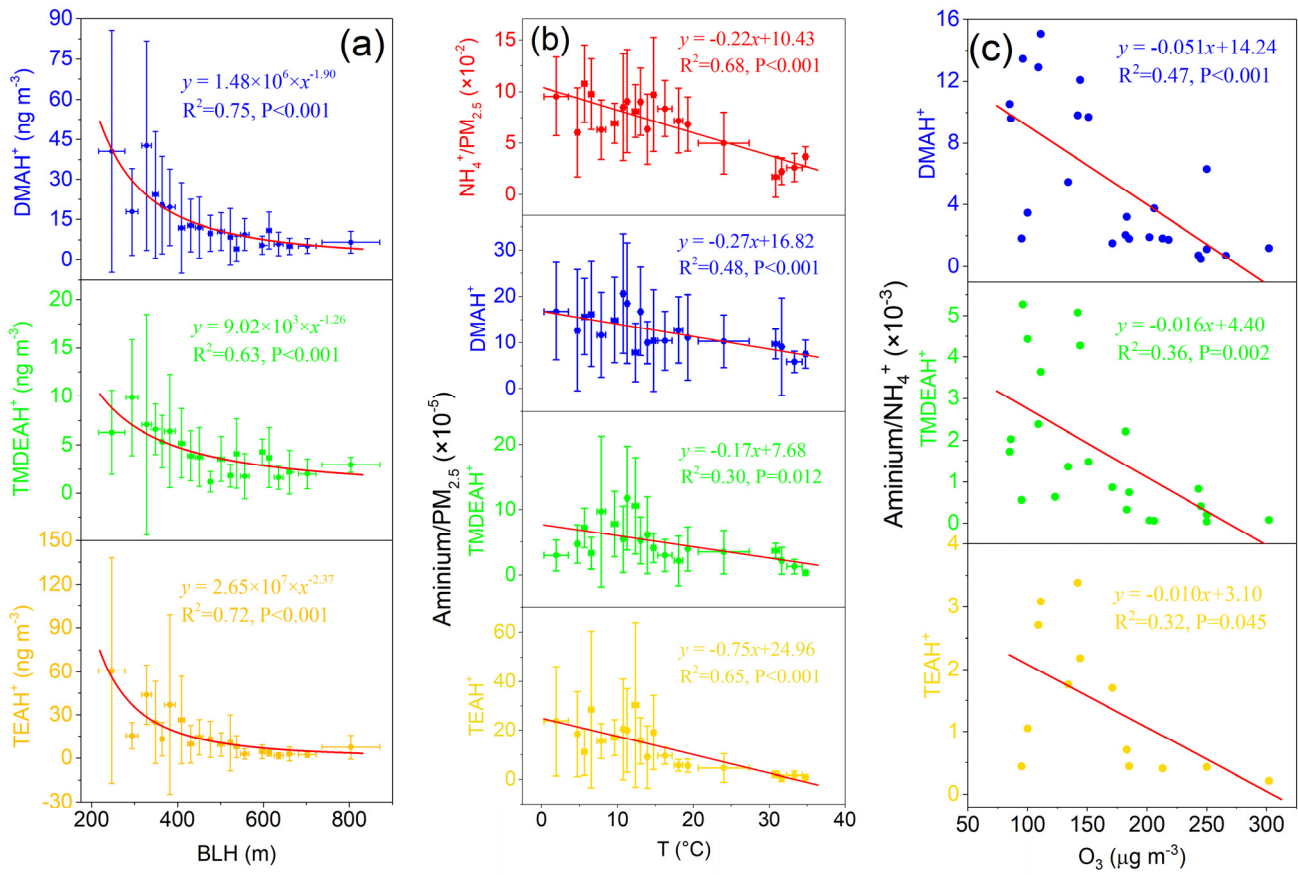
Figure 1. Map of sampling sites and area. The red stars represent the locations of Shanghai (Fudan University) and Huaniao Island, and the black line in the marginal seas represents the cruise track in the spring of 2017.



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Figure 2. The mass concentrations of PM_{2.5}, fine-particle NH₄⁺ and three aminiums (TEAH⁺, DMAH⁺ and TMDEAH⁺) in different campaigns in Shanghai (SH), Huaniao Island (HNI) and the Yellow and East China seas (YECS). The columns and error bars represent average concentrations and standard deviations, respectively. The orange horizontal lines represent the annual average concentrations of aminiums in SH and HNI.



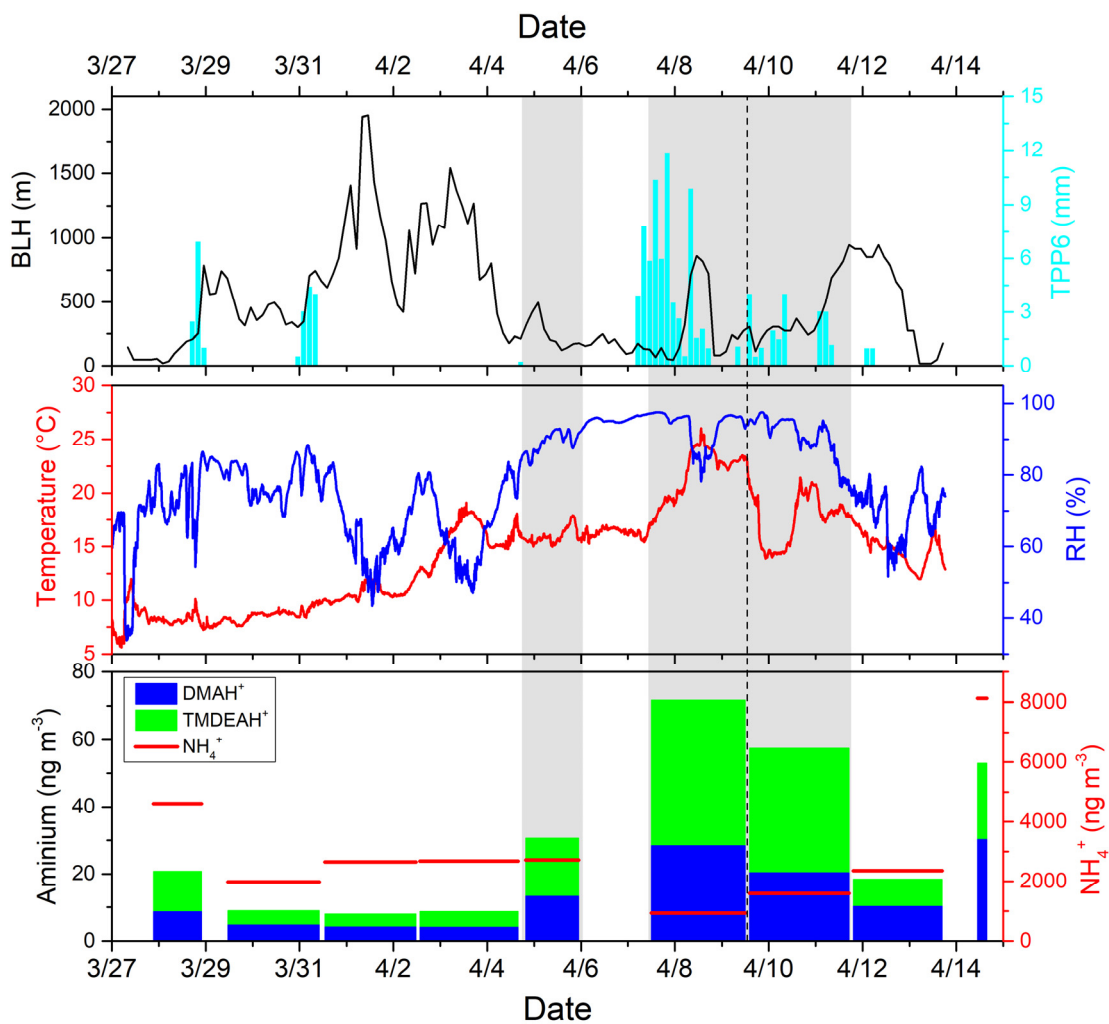
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Figure 3. (a) Relationships between concentrations of aminiums and boundary layer height (BLH) over Shanghai in 2013. (b) Relationships between mass ratios of aminiums and NH₄⁺ to PM_{2.5} and temperature over Shanghai in 2013. (c) Relationships between mass ratios of aminiums to NH₄⁺ and O₃ concentrations over Shanghai during the summer of 2013.

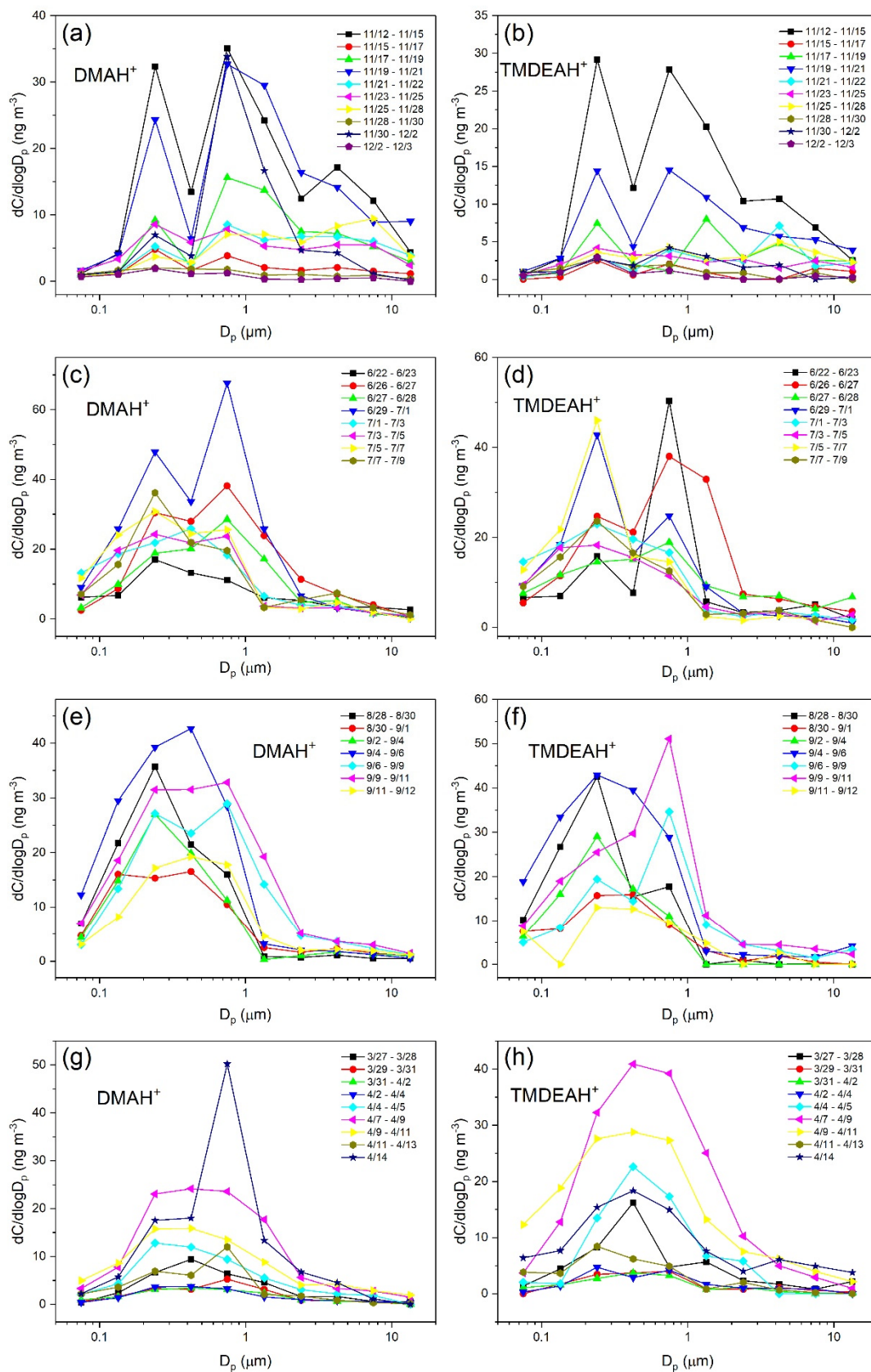


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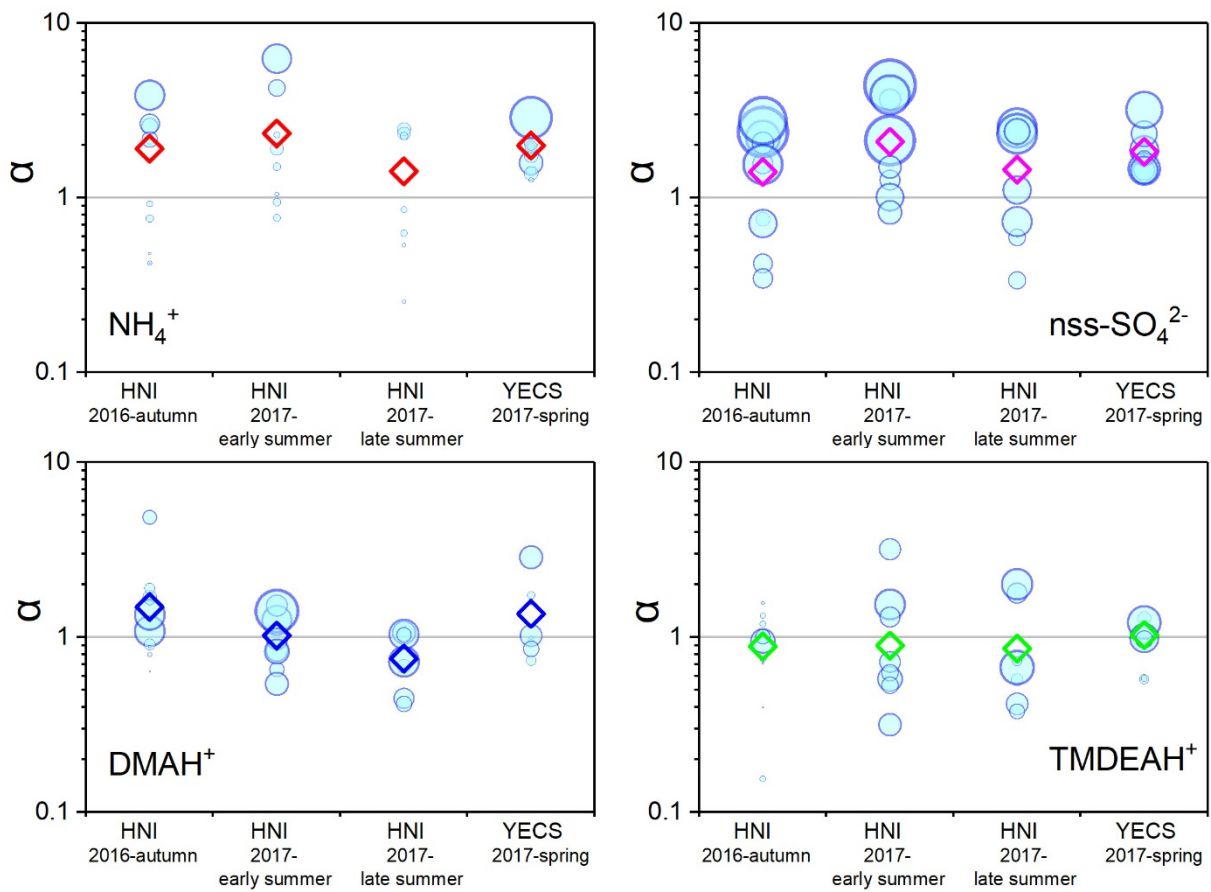
Figure 4. Time series of meteorological parameters and the concentrations of aminiums and NH_4^+ during the cruise of 2017. The time range spanned by the column of each aminium concentration corresponds to the sampling time.



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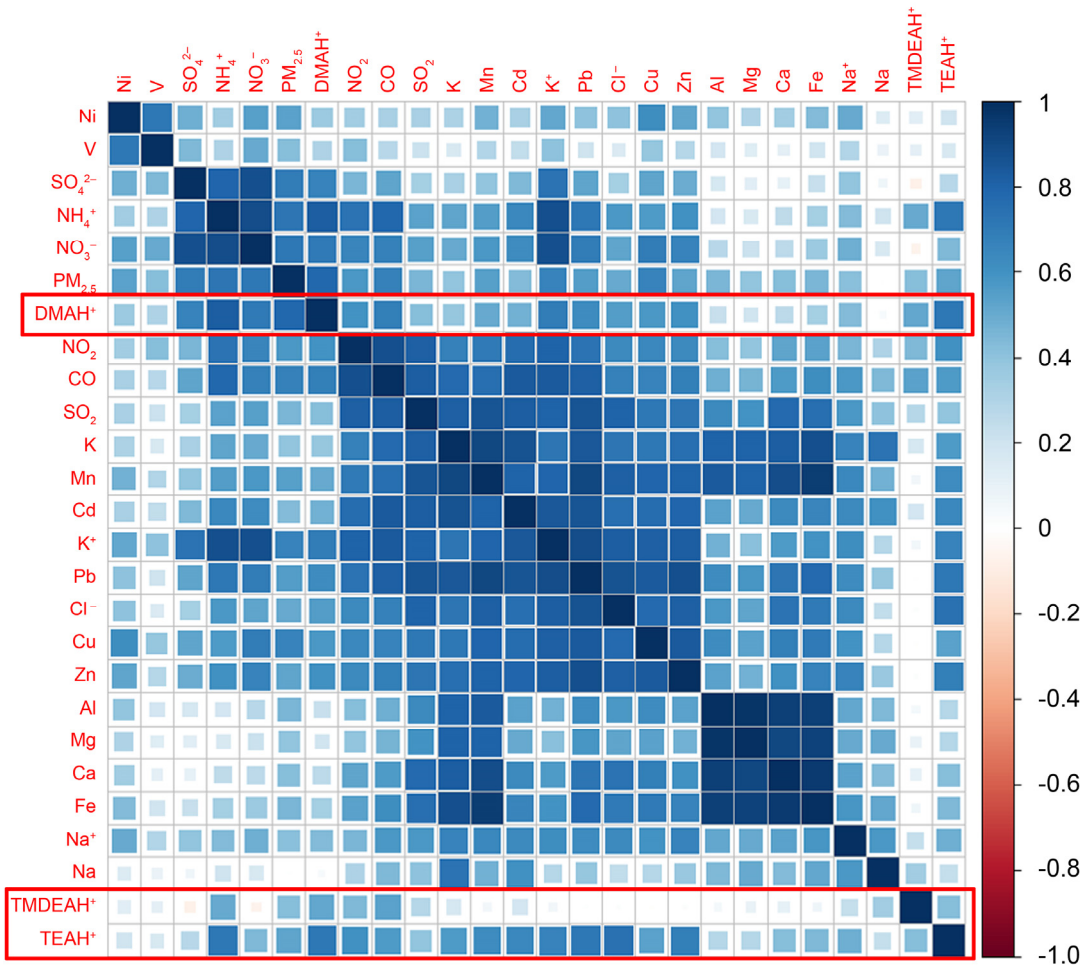
663 **Figure 5.** Size distributions of aminiums during different campaigns. (a-b): in the autumn of 2016 at Huaniao Island, (c-d): in early summer
 664 of 2017 at Huaniao Island, (e-f): in late summer of 2017 at Huaniao Island, (g-h): in 2017 spring cruise over the Yellow and East China seas.

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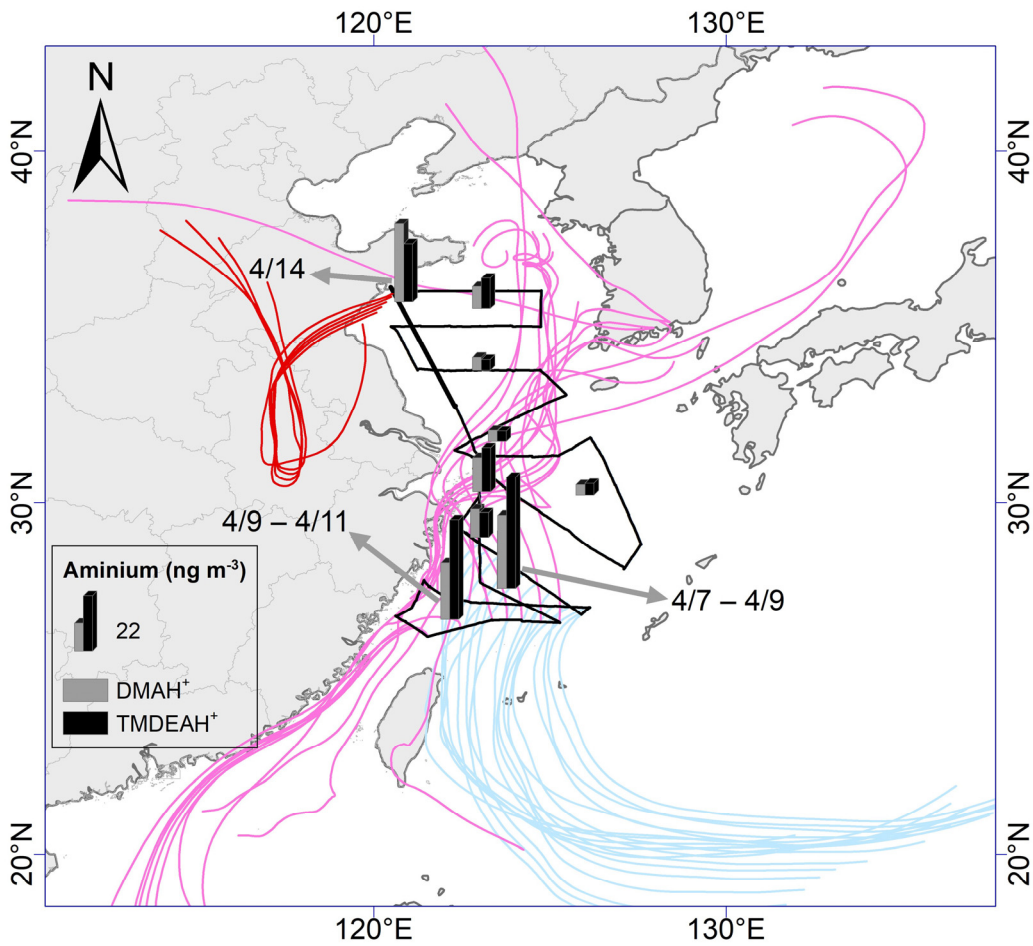
Figure 6. The α values of NH_4^+ , nss-SO_4^{2-} and aminiums in different campaigns. The diameter of the circle is proportional to the concentration and the diamond-shape symbol represents the average value of α for each campaign. It should be noted that the bottom of column is the line of $\alpha=1$.



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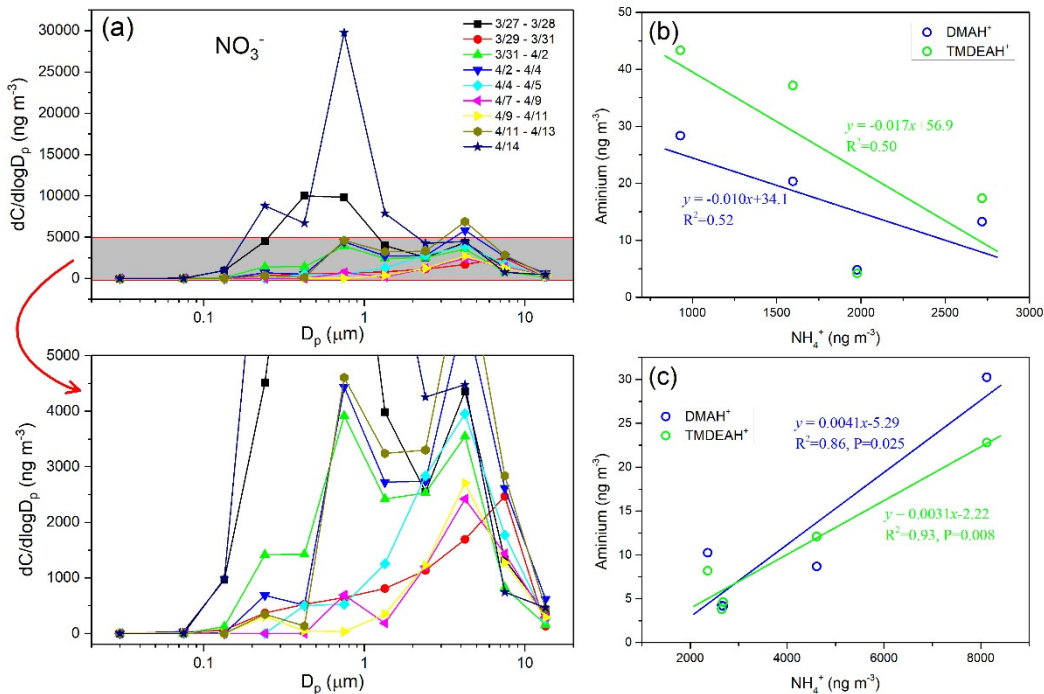
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Figure 7. Correlation coefficient matrix among the concentrations of PM_{2.5} components and gaseous pollutants over Shanghai in 2013.



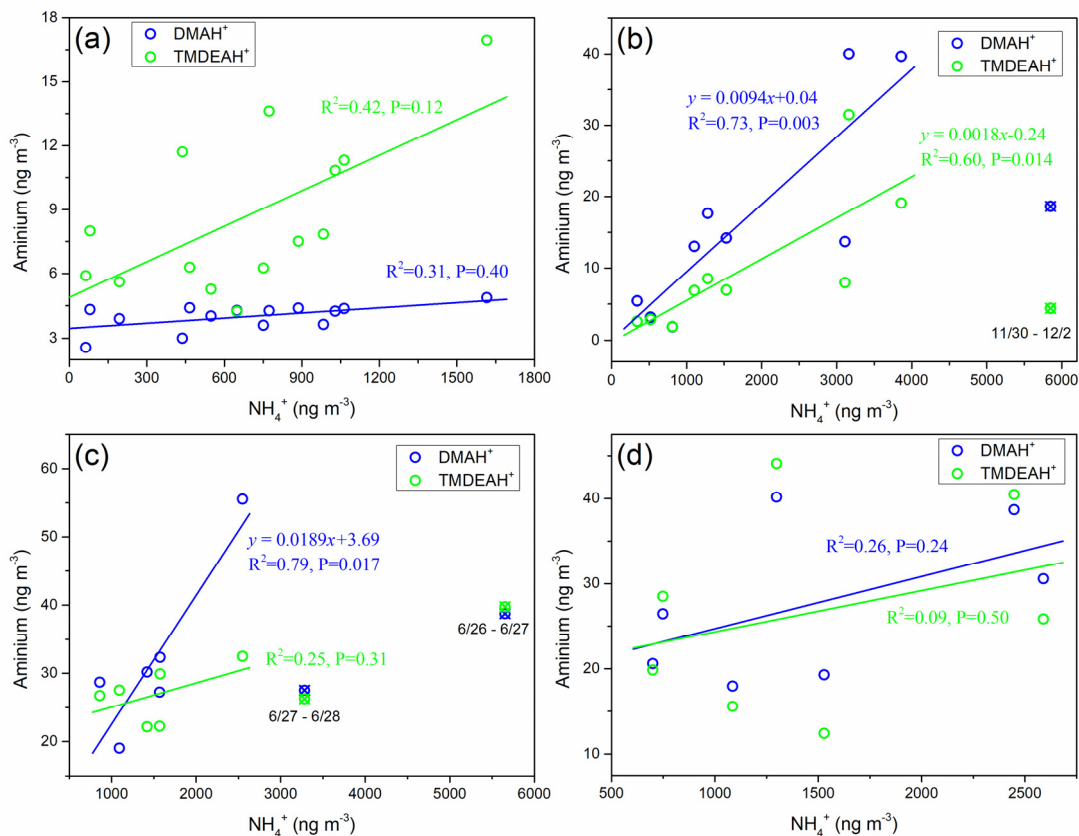
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673 **Figure 8.** The spatial distribution of aminiums over the YECS in the spring of 2017. The ocean color represents the concentration of
 674 chlorophyll a obtained from Kriging interpolation from the observed concentrations. The light blue, pink and red lines represent 72-hour
 675 backward trajectories corresponding to sample sets collected on 7–9 Apr., 9–11 Apr. and 14 Apr., respectively.



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677 **Figure 9.** (a) Size distributions of NO_3^- over the YECS in the spring of 2017. (b) Correlations between concentrations of aminiums and
 678 NH_4^+ for the samples mainly influenced by marine air masses. (c) Correlations between concentrations of aminiums and NH_4^+ for the samples
 679 predominantly influenced by terrestrial transport.

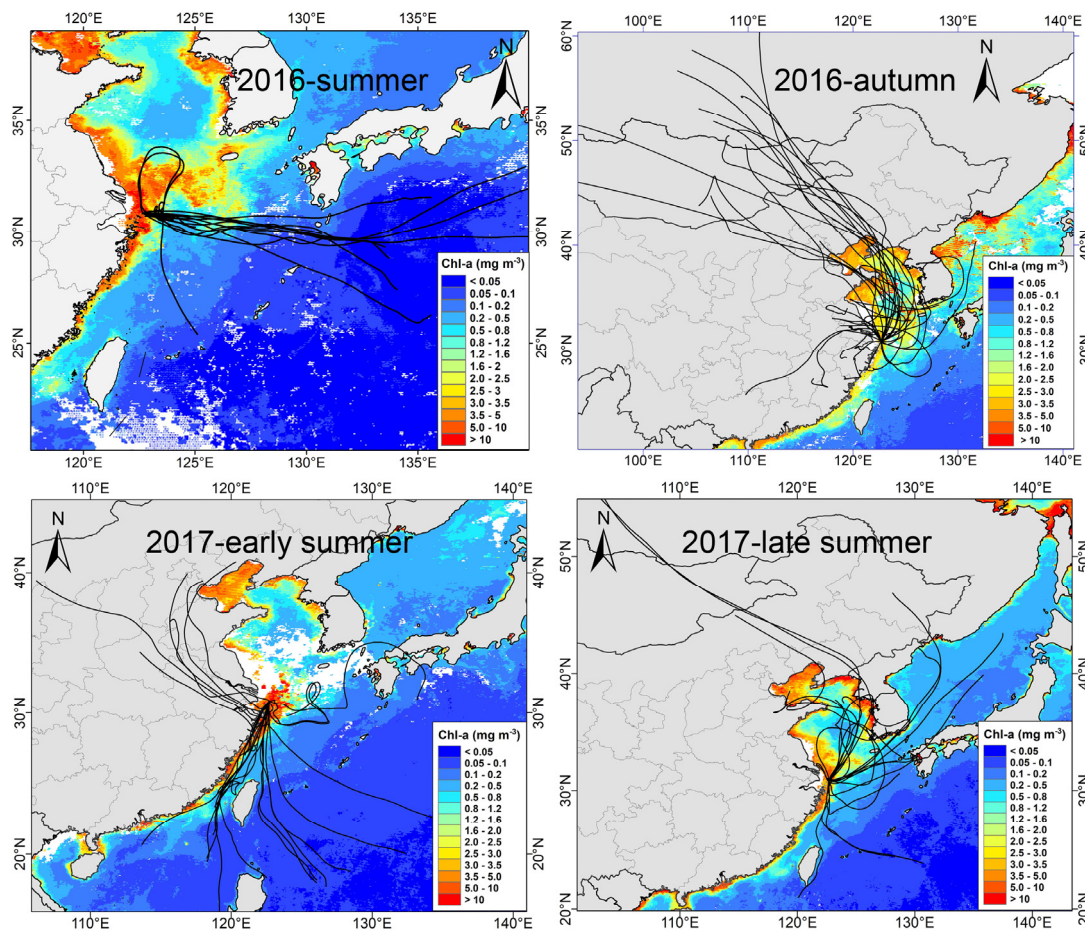


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Figure 10. Correlations between aminiums and NH₄⁺ concentrations over Huaniao Island for each campaign. **(a):** in the summer of 2016, **(b):** in the autumn of 2016, **(c):** in early summer of 2017, **(d):** in late summer of 2017.



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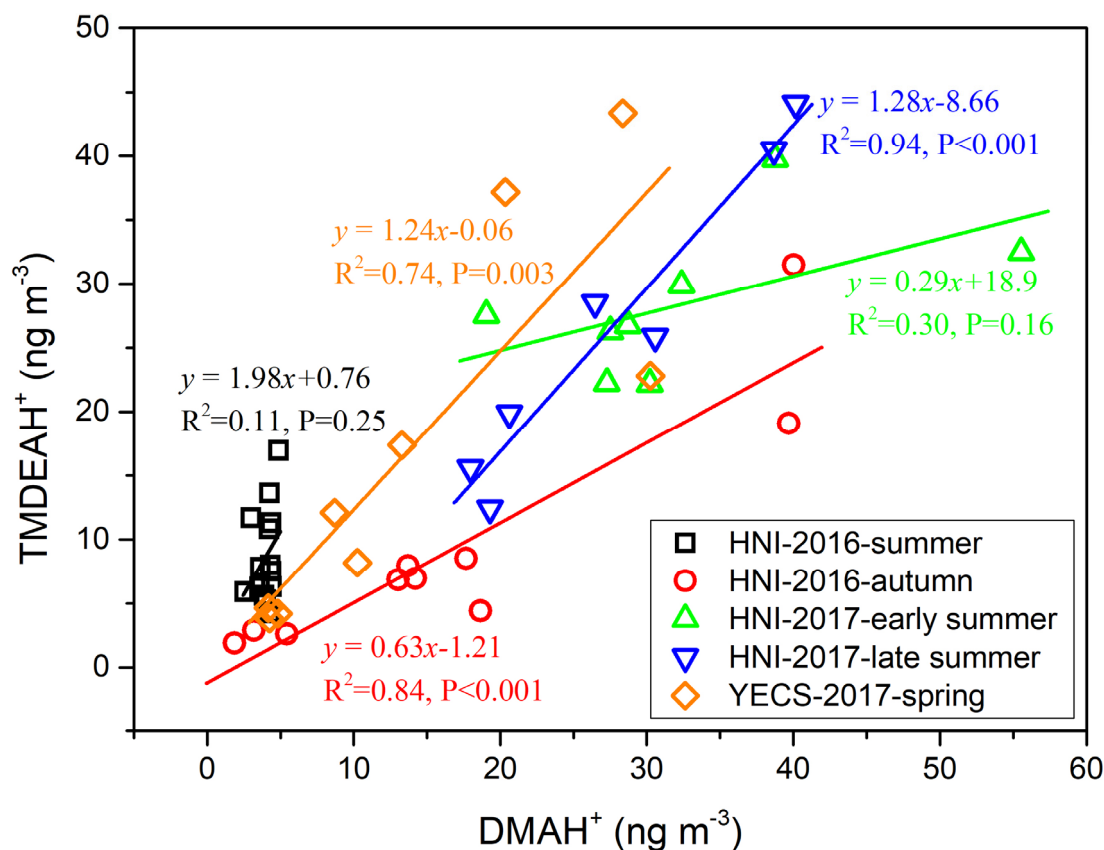
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Figure 11. The 72-hour backward trajectories starting from Huaniao Island and the average chlorophyll a concentration retrieved and combined from aqua- and terra-MODIS during the sampling period. Each sample during the summer of 2016 corresponds to one trajectory

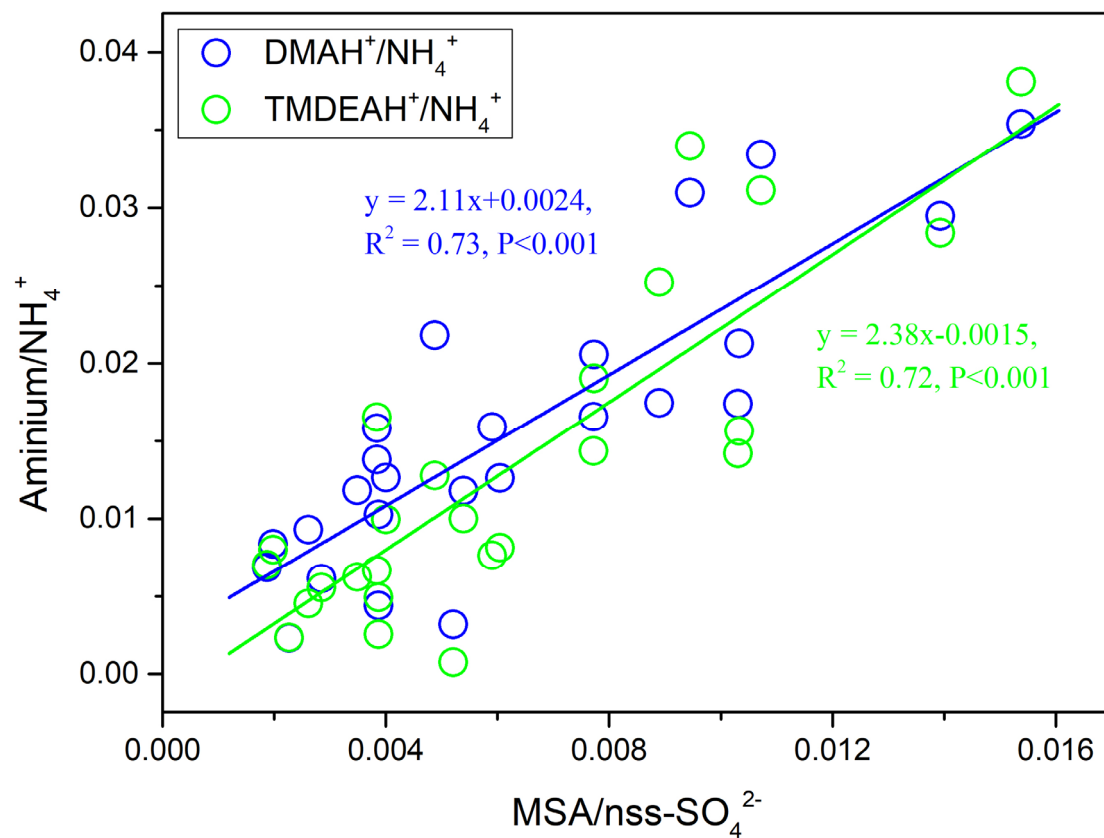
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with a starting time in the middle of sampling period. Each sample set during the autumn of 2016 and the summer of 2017 corresponds to 3 trajectories and the starting times are taken at equal intervals in the sampling period.



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Figure 12. Correlations between DMAH⁺ and TMDEAH⁺ for each campaign over Huaniao Island and the YECS.



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Figure 13. Correlations between ammonium/NH₄⁺ and MSA/nss-SO₄²⁻ over Huaniao Island during the autumn in 2016 and the summer in 2017.

Table 1. Summary of sampling information in different campaigns.

Sampling site	Sampler	Sampling period	Number of samples or sample sets
		25 Mar. 2013–26 Apr. 2013 (spring)	29
Fudan University, Shanghai	Medium-flow PM _{2.5} sampler	16 Jul. 2013–17 Aug. 2013 (summer)	26
		30 Oct. 2013–30 Nov. 2013 (autumn)	29
		1 Dec. 2013–23 Jan. 2014 (winter)	37
Huaniao Island	Medium-flow PM _{2.5} sampler	4 Aug. 2016–18 Aug. 2016 (summer)	14
		12 Nov. 2016–3 Dec. 2016 (autumn)	9
Huaniao Island	MOUDI	11 Mar. 2017–19 Mar. 2017 (spring)	4
		22 Jun. 2017–9 Jul. 2017 (early summer)	8
		27 Aug. 2017–12 Sep. 2017 (late summer)	7
the Yellow Sea and the East China Sea	MOUDI	27 Mar. 2017–14 Apr. 2017 (spring)	9

Table 2. The mass concentrations (mean values ± 1 standard deviation) of NH_4^+ and aminiums over Shanghai, Huaniao Island and the YECS compared to other sites reported in literatures. The values below the detection limits are indicated by < DL.

No.	Site	Site type	Sampling period	Particle size	NH_4^+ ($\mu\text{g m}^{-3}$)	Aminium (ng m^{-3})					Reference
						MMAH ⁺	DMAH ⁺	TMDEAH ⁺	MEAH ⁺	TEAH ⁺	
1	Shanghai, China	urban	Spring (Mar.–Apr. 2013)	PM _{2.5}	6.0±3.4		6.4±6.1	4.8±2.3		8.4±8.4	this study
2			Summer (Jul.–Aug. 2013)	PM _{2.5}	3.1±2.9		9.1±15.2	1.7±1.6		0.9±1.0	
3			Autumn (Nov. 2013)	PM _{2.5}	6.8±4.5		15.5±13.4	2.8±2.9		12.7±12.2	
4			Winter (Dec. 2013–Jan. 2014)	PM _{2.5}	13.7±9.8		27.3±29.0	7.3±6.2		35.2±45.6	
5	Shanghai, China	urban	Jul.–Aug. 2013	PM _{1.8}	2.5±1.3	8.9±6.1	15.7±7.9	38.8±17.0	11.5±17.4		(Tao et al., 2016)
6				PM ₁₀	2.6±1.3	9.9±6.9	20.1±10.7	47.0±19.9	15.7±26.4		
7	Shanghai, China	urban	Jan. 2013	PM _{2.5}		2.4			0.2		(Huang et al., 2016)
8			Jul.–Aug. 2013	PM _{2.5}		3.9			0.3		
9	Yangzhou, China	urban	Nov. 2015–Apr. 2016	PM _{2.5}		4.9±1.9	4.3±2.4		15.4±8.1		(Shen et al., 2017)
10	Nanjing, China	urban	Apr.–May 2016	PM _{2.5}		7.6	4.2		21.7		
11			Aug. 2014	PM _{1.8}		7.2±4.1	18.0±11.7		36.4±18.6		
12	Xi'an, China	urban	Jul. 2008–Aug. 2009	PM _{2.5}		14.4±9.6			3.3±2.4		(Ho et al., 2015)
13	Guangzhou, China	urban	Sep.–Oct. 2014	PM _{0.95}	4.3±1.1	41.8±11.4	14.5±3.2	3.7±0.9	3.2±0.4		(Liu et al., 2017)
14				PM ₃	5.1±1.4	50.4±13.7	17.7±3.6	4.8±1.4	4.0±0.5		
15				PM ₁₀	5.2±1.4	51.8±13.9	19.0±3.8	5.4±1.6	4.2±0.6		
16	Tampa Bay, Florida, USA	urban	Jul.–Sep. 2005	PM _{2.5}	1.4±1.2		31.6±28.3				(Calderón et al., 2007)
17	a traffic site, Milan, Italy	urban	Oct. 2013	TSP	4.2±2.9		90±20			360±20	(Perrone et al., 2016)
18	a limited traffic site, Milan, Italy	urban	Oct. 2013	TSP	4.0±3.0		100±10			420±100	
19	Qingdao, China	semi-urban	May 2013, Nov.–Dec. 2013, Nov.–Dec. 2015	PM _{0.056-10}			6.3	5.8			(Xie et al., 2018)
20	resort beach site of Qingdao, China	coastal, rural	Aug. 2016	PM _{0.056-10}			28.5±23.0	9.0±6.6			
21	Egbert, Toronto, Canada	agricultural and semi-forested	Oct. 2010	PM _{2.5}			0.1±0.2	1±0.6			(VandenBoer et al., 2012)
22	Hyytiälä, southern Finland	boreal forest	Mar. 2015	PM ₁₀	0.4±0.1	6.8	1.5	1.1			(Hemmilä et al., 2018)
23			Apr. 2015	PM ₁₀	0.1±0.1	2.9	3.1	0.7			
24			Jul. 2015	PM ₁₀	0.1±0.1	3.0	8.4±4.9	1.8±1.4	0.4		
25	Nanling, Guangdong, China	forest	Oct. 2016	PM _{2.5}	0.9±0.6	8.8±7.8	2.4±3.2	1.1±1.8			(Liu et al., 2018a)
26			May–Jun. 2017		1.8±1.6	11.9±9.8	5.0±2.2	1.7±1.7			

No.	Site	Site type	Sampling period	Particle size	NH ₄ ⁺ (µg m ⁻³)	Aminium (ng m ⁻³)					Reference
						MMAH ⁺	DMAH ⁺	TMDEAH ⁺	MEA ⁺	TEAH ⁺	
27	Brent, Alabama, USA	forest	Jun.1 – July 15, 2013	submicron	0.52	148*					(You et al., 2014)
28	Huaniao Island, China	marine	Aug. 2016	PM _{2.5}	0.7±0.4		4.0±0.6	8.7±3.7		< DL	this study
29			Nov.–Dec. 2016	PM _{1.8}	1.9±1.5		10.7±9.3	6.0±6.8		< DL	
30				PM ₁₀	2.1±1.8		15.1±12.4	8.4±8.8		< DL	
31			Mar. 2017	PM _{1.8}	2.0±1.2		6.8±4.6	2.7±1.8		< DL	
32				PM ₁₀	2.3±1.4		11.4±11.6	3.1±2.2		< DL	
33			Jun.–Jul. 2017	PM _{1.8}	2.1±1.4		29.0±10.8	24.8±5.4		< DL	
34				PM ₁₀	2.2±1.6		32.2±11.0	27.5±5.7		< DL	
35			Aug.–Sep. 2017	PM _{1.8}	1.4±0.7		25.8±8.7	25.0±11.0		< DL	
36				PM ₁₀	1.5±0.8		27.4±9.1	26.3±11.6		< DL	
37			the Yellow Sea and the East China Sea	marine	Mar.–Apr. 2017	PM _{1.8}	2.8±2.0		11.9±9.0	14.6±12.9	
38	PM ₁₀	3.0±2.2					13.5±10.1	16.6±14.5		< DL	
39	the Yellow Sea and the northwest Pacific	marine	Apr. 2015	PM _{0.056-10}			12.9±10.6	13.2±13.8		(Xie et al., 2018)	
40	the East China Sea	marine	Jan. 2016	PM _{0.056-10}			30.8±9.7	12.0±6.6			
41	the Yellow Sea and the Bohai Sea	marine	Aug. 2015, Jun.–Jul. 2016	PM _{0.056-10}			33.3	19.4			
42	the south Yellow Sea	marine	Nov. 2013	PM _{0.056-10}			18.9±16.6	31.8±19.2			
43	the Yellow Sea and the Bohai Sea	marine	May 2012	PM ₁₁			202±170	432±426		(Hu et al., 2015)	
44	the south Yellow Sea	marine	Nov. 2012	PM ₁₀			13.3±4.6	30.0±12.6		(Yu et al., 2016)	
45	the north Yellow Sea and the Bohai Sea	marine	Nov. 2012	PM ₁₀			-	15.0±6.6			
46	Arabian Sea	marine	Aug.–Oct. 1994	PM _{0.9}	0.04	3.2	2.1	0.3			(Gibb et al., 1999)
47			Nov.–Dec. 1994	PM _{0.9}	0.1	3.7	11.1	0.5			
48	Mace Head, Ireland	marine	Jan.–Dec. 2006	PM ₁			4.7±6.0	7.6±9.4		(Facchini et al., 2008)	
49	Irish West Coast	marine	Jun.–Jul. 2006	PM ₁			14.7±14.3	14.3±8.7			
50	São Vicente, Cape Verde	marine	May–Jun., Dec. 2007	PM _{0.14-0.42}	0.1	0.1	0.4	0.2		(Müller et al., 2009)	
51	off the Central Coast of California, USA	marine	Jul. 2007	PM ₁				22		(Sorooshian et al., 2009)	
52	the Eastern Mediterranean, Greece	marine	2005–2006	PM ₁			9.2±36.8	< DL		(Violaki and Mihalopoulos, 2010)	

* Fourier Transform Infrared spectroscopy (FTIR) measured total primary aminiums (R-NH₃⁺)

700 **Table 3.** Calculated terrestrial and marine source contributions to aerosol aminiums over Huaniao Island (mean (minimum –
 701 maximum)).

Campaign	DMAH ⁺		TMDEAH ⁺	
	Terrestrial contribution (%)	Marine contribution (%)	Terrestrial contribution (%)	Marine contribution (%)
2016-autumn	74.1 (42.5 - 100)	25.9 (0 - 57.5)	69.1 (34.3 - 100)	30.9 (0 - 65.7)
2017-early summer	46.7 (20.3 - 98.8)	53.3 (1.2 - 79.7)	25.8 (11.0 - 48.6)	74.2 (51.4 - 89.0)
2017-late summer	37.0 (19.2 - 57.4)	63.0 (42.6 - 80.8)	21.7 (9.0 - 42.1)	78.3 (57.9 - 91.0)

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