

Exploring the inconsistent variations in atmospheric primary and secondary pollutants during the G20 2016 Summit in Hangzhou, China: implications from observation and model

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Abstract. Complex aerosol and photochemical pollution (ozone and peroxyacetyl nitrate (PAN)) frequently occur in eastern China and mitigation strategies to effectively alleviate both kinds of pollution are urgently needed. Although the effectiveness of powerful control measures implemented by the Chinese State Council has been comprehensively evaluated on reducing atmospheric primary pollutants, the effectiveness on mitigating photochemical pollution is less assessed and therein the underlying mechanisms are still poorly understood. The stringent emission controls implemented from 24 August to 6 September, 2016 during the summit for Group of Twenty Finance Ministers and Central Bank Governors (G20) provides us a unique opportunity to address this issue. Surface concentrations of atmospheric O₃, PAN, and their precursors including volatile organic compounds (VOCs) and nitrogen dioxides (NO_x), in addition to the other trace gases and particulate matter were measured at the National Reference Climatological Station (NRCS) (30.22 °N, 120.17 °E, 41.7 m a.s.l) in urban Hangzhou. We found significant decreases in atmospheric PAN, NO_x, the total VOCs, PM_{2.5}, and sulfur dioxide (SO₂) under the unfavorable meteorological condition during G20 (DG20) relative to the adjacent period

35 before and after G20 (BG20 and AG20), indicating that the powerful control measures have taken into
36 effect on reducing the pollutants emissions in Hangzhou. Unlike with the other pollutants, daily
37 maximum average-8 h (DMA8) O₃ exhibited a slight increase and then decrease from BG20 to AG20,
38 which was mainly attributed to the variation in the solar irradiation intensity and regional transport
39 besides the contribution from the implement of stringent control measures. Results from
40 observation-based chemical model (OBM) indicated that acetaldehyde and methyl glyoxal (MGLY)
41 were the most important second-generation precursors of PAN, accounting for 37.3-51.6% and
42 22.8%-29.5% of the total production rates including the reactions of OVOCs, propagation of other
43 radicals, and the other minor sources. Moreover, we confirmed the productions of PAN and O₃ were both
44 sensitive to VOCs throughout the whole period, specifically dominated by aromatics in BG20 and DG20
45 but alkenes in AG20. These findings suggested that reducing emissions of aromatics, alkenes, and alkanes
46 would mitigate photochemical pollution including PAN and O₃. Source appointment results attributed the
47 reductions of VOCs source and ozone formation potentials (OFP) during G20 to the effective emission
48 controls on traffic (vehicle exhaust) and industrial processes (solvent utilization and industrial
49 manufacturing). However, fuel combustion and biogenic emission both weakened such effect with
50 sizeable contribution on the VOCs mixing ratios (18.8% and 20.9%) and OFPs (25.6% and 17.8%),
51 especially during the latter part of G20 (G20 II) when anthropogenic VOCs were substantially reduced.
52 This study highlights the effectiveness of stringent emission controls in relation to traffic and industrial
53 sources, but a coordinated program related with controlling fuel combustion and biogenic emissions is
54 also required on addressing secondary pollution.

55 **1 Introduction**

56 Complex atmospheric pollution including particulate matter and photochemical pollution (ozone (O₃)
57 and peroxyacetyl nitrate (PAN)) is a pervasive environmental issue in eastern China (Geng et al., 2007;
58 Ding et al., 2013; Mo et al., 2015; Li et al., 2016; Zhang et al., 2018). Numerous mitigation strategies
59 have been released by the Chinese government, such as the nationwide application of flue-gas
60 desulfurization (FGD) devices in power plants after 2006 (Feng et al., 2014) and “Atmospheric
61 Pollution Prevention and Control Action Plan” in 2013 (Zhang et al., 2016). As expected, ambient
62 concentrations of primary gas pollutants such as sulfur dioxide (SO₂) (Koukouli et al., 2016) and
63 nitrogen oxides (NO_x = NO + NO₂) (de Foy et al., 2016) showed good response to emission reductions.
64 However, secondary atmospheric pollutants such as ozone and secondary aerosols frequently exceeded
65 their respective Chinese Grade II standards over urban cities in China (Wang et al., 2014). Severe haze
66 pollution, mainly comprised of PM_{2.5} (particles within 2.5 μm diameter range), still occur in China
67 during wintertime, although it started to decline during the 11th Five-Year Plan period (Huang et al.,
68 2014; Cheng et al., 2016; Miao et al., 2018; Miao and Liu, 2019). Surface O₃ also exhibits a rapid
69 increasing trend over China since 2000 (Verstraeten et al., 2015; Wang et al., 2017), with high levels

(9.5-14.0 ppbv) of PAN often encountered during O₃ pollution events (Shao et al., 2009; Liu et al., 2010; Zhang et al., 2012a; Zhang et al., 2014; Zhang et al., 2015; Xue et al., 2014c). Due to the highly nonlinear response of O₃ and PAN to primary pollutant emissions, in addition to intricate photochemical reactions, the mitigation of secondary photochemical pollution is even more challenging. In the troposphere, O₃ and PAN are both formed in photochemical reactions of VOCs in the presence of NO_x. However, PAN is exclusively formed by the oxidation of a small part of VOCs that can generate peroxy acetyl radical (CH₃C(O)O₂, PA) including oxygenated VOCs (OVOCs) such as acetaldehyde, acetone, methacrolein (MACR), methyl vinyl ketone (MVK), and methyl glyoxal (MGLY) (Williams et al., 2000; LaFranchi et al., 2009), while O₃ formation involves almost all VOCs. Therefore, PAN is considered to be a better indicator for photochemical smog than O₃ (McFadyen and Cape, 2005). In addition, these OVOCs are mainly oxidation products (here referred to secondary precursors of PAN) of a certain class of hydrocarbons (e.g., ethane, propene, isoprene, and some aromatics) by the oxidations of OH/NO₃/O₃. The relative importance of individual precursors to the formation of PAN and O₃ varies from place to place depending on the reactivity and composition of VOCs. Identification of the dominant precursors is the key to effective control of photochemical pollution, which, however, remains poorly characterized in China.

Recently, a series of temporary and stringent emission control measures were implemented in China during several mega-events including the 29th Summer Olympic Games (August 2008), the 21th Asia-Pacific Economic Cooperation (APEC) conference (November 2014), and China Victory Day Parade (Victory Parade 2015) in Beijing (Verstraeten et al., 2015) and the surrounding areas (Xu et al., 2010; Zhang et al., 2012b; Gao et al., 2011; Li et al., 2017). During these events, the effectiveness of a series of emission control measures on reducing atmospheric primary pollutants, in particular to the particulate matter, has been comprehensively evaluated, but less on photochemical pollution.

In September 2016, the Group of Twenty (G20) summit was hosted in Hangzhou, the capital city of Zhejiang Province, which is located along the mid-Yangtze River Delta (YRD) in the eastern part of China. Similar with other major events held in Beijing, rigorous temporal control measures were set to reduce emissions of air pollutants in Hangzhou and the adjacent regions including Zhejiang, Shanghai, Jiangsu, and Anhui province from 24 August to 7 September including Phase I (24-27 August) and Phase II (28 August-6 September). These control measures included restrictions on the number of vehicles, limited production or complete shut-down of industrial enterprises, and temporary cessation of construction activities, and the target sources incorporated vehicles, paint and solvent use, steel factories, chemical factories, power plants. During phase I the government implemented strict emission control measures in industrial source, power plant, and residential and the phase II referred to the additional controls measures as vehicles controls in the Hangzhou and surrounding provinces (including Zhejiang, Jiangsu, Jiangxi, and Anhui).

In this study, to evaluate the effectiveness of emission control measures on reducing pollutant

concentrations, we compared the variations of atmospheric O₃, PAN, particulate matter, VOCs, NO_x, and other trace gases before, during, and after G20, also demonstrating the effect of meteorological conditions by using WRF-Chem model. An observation-based chemical box model (OBM) was used to identify the predominant precursors and key chemical processes in PAN and O₃ formation and to further assess the effect of reducing their respective precursors before, during, and after G20. Positive matrix factorization (PMF) was employed to appoint the corresponding sources of various VOCs and compare their variations and their respective ozone formation potentials (OFPs) before, during, and after G20.

2. Experimental

2.1 Observations

In-situ observations of atmospheric PAN, O₃, and VOCs and a suite of associated chemical species and meteorological parameters, including NO_x, CO, SO₂, fine particulate matter (PM_{2.5}), were conducted at an urban site named as National Reference Climatological Station (NRCS) (30.22°N, 120.17°E, 41.7 m a.s.l) in the center of Hangzhou as shown in Fig. 1. PAN was measured by a modified gas chromatography (Agilent 7890B, USA) equipped with electron capture detector, which has been described in our previous studies in details (Zhang et al., 2012a; Zhang et al., 2014; Zhang et al., 2015). Trace gases including O₃, SO₂, NO_x, and CO were detected by a set of commercial trace gas analyzers (Thermo Environmental Instruments Inc., USA i-series 49i, 43i, 42i, and 48i), respectively (Zhang et al., 2018). All trace gas analyzers were weekly span and daily zero checked. Ambient VOCs were measured by using an on-line gas chromatography (Syntech Spectras Instrument Co., Ltd., The Netherlands) coupled with dual detectors (Photo Ionization Detector (PID) and flame ionization detector (FID) for quantifying C₂-C₅ VOCs (GC955 series 811) and PID for detecting C₆-C₁₂ VOCs (GC955 series 611). Ambient PM_{2.5} samples were collected using co-located Thermo Scientific (formerly R&P) Model 1405D samplers. PM-Coarse and PM_{2.5} particulate, split by a virtual impactor, each accumulate on the system's exchangeable TEOM filters. By maintaining a flow rate of 1.67 L min⁻¹ through the coarse sample flow channel and 3 L min⁻¹ through the PM_{2.5} sample channel, and measuring the total mass accumulated on each of the TEOM filters, the device can calculate the mass concentration of both the PM_{2.5} and PM Coarse sample streams in near real-time.

2.2 Models

2.2.1 WRF-Chem model

To quantify the separate effects of meteorological condition (EMC) and emission control measures (ECC) on observed particulate concentrations, we performed simulations using Weather Research and Forecasting model coupled to Chemistry (WRF-Chem). WRF-Chem V3.9 was used to simulate the variation of PM_{2.5} concentration from Aug. 6 00:00 UTC, 2016 to Sep. 16 00:00 UTC, 2016. Multi-resolution Emission Inventory for China at 0.25° in 2016, developed by Tsinghua University

(<http://www.meicmodel.org/>), was used as input for WRF-Chem. WRF-Chem was configured to have two nested domains, i.e. an outer domain with horizontal resolution of 25 km (140×100 grid points) covering East China and the surrounding areas and an inner domain with 5 km-resolution (101×101 grid points) covering Yangtze River Delta. Hangzhou is located in the center of domain. Vertically, there were a total of 35 full eta levels extending to the model top at 50 hPa, with 16 levels below 2 km. The National Centers for Environmental Prediction (NCEP) Final Operational Global Analysis (FNL) data available at 1°×1° every six hours were used meteorological driving fields. Analysis nudging was used for domain one. RADM2 chemical mechanism and MADE/SORGAM aerosols were used in this study. Here we assumed no significant change in chemical processes (specifically the other reactive gases involved in the chemical reactions with these pollutants) from BG20 to AG20. Thereby, the net contribution (NCC) of emission controls and meteorological conditions primarily results in the difference between observed PM_{2.5} before and during G20, which is represented by the ratio of (observed PM_{2.5} (BG20)-observed PM_{2.5} (DG20 II))/observed PM_{2.5} (BG20). The effect of meteorological conditions (EMC) was quantified by comparing the modeled PM_{2.5} without emission controls before and during G20 under their respective meteorological condition (Equation 1). Thereby, the effect of emission controls (ECC) could be obtained through the difference between NCC and EMC before and during G20 (Equation 2) below

$$EMC = \frac{Modeled\ PM_{2.5}(BG20) - Modeled\ PM_{2.5}(DG20\ II)}{Modeled\ PM_{2.5}(BG20)} \times 100\% \quad (1)$$

$$ECC = (NCC - EMC) \times 100\% \quad (2)$$

In general, the modeled results of PM_{2.5} before and after G20 can reproduce the observation results (mean bias (MB) = 2.46, root mean-square error (RMSE) = 15.5, R = 0.63, p < 0.01), providing the basis of the following comparison.

2.2.2 Backward trajectories analysis

To determine the influence of regional transport on the pollutant concentrations, 24 h air mass back trajectories starting at 300 m from NRCS site were calculated by using the National Oceanic and Atmospheric Administration (NOAA) HYSPLIT-4 model with a 1°×1° grid and the final meteorological database. The 6-hourly final archive data were obtained from the National Center for Environmental Prediction's Global Data Assimilation System (GDAS) wind field reanalysis. GDAS uses a spectral medium-range forecast model. More details can be found at <http://www.arl.noaa.gov/ready/open/hysplit4.html>. The model was run 24 times per day. The method used in trajectory clustering was based on the GIS-based software TrajStat (Wang et al., 2004).

2.2.3 Observation-based chemical box model (OBM)

Here we used OBM model to simulate in situ PAN and O₃ production and their sensitivity to changes in

PAN and O₃ precursors, which has been successfully implied in our previous studies (Xue et al., 2014a; Xue et al., 2014c; Xue et al., 2016; Li et al., 2018). In brief, the model was built on the latest version of the Master Chemical Mechanism (MCM v3.3), an explicit mechanism describing the degradation of 143 primarily emitted VOC, resulting in 17,224 reactions involving 5833 molecular and free radical species (Saunders et al., 2003). Besides the existing reactions in MCM v3.3, the heterogeneous reactions of NO₂, HO₂, NO₃, and N₂O₅ were also incorporated. In addition, we also optimized the model with some physical processes such as the variations of boundary layer height and solar zenith angle, dry deposition, and the dilution of air pollutants within the planetary boundary layer (Xue et al., 2014b). The photolysis frequencies appropriate for Hangzhou are parameterized using a two-stream isotropic-scattering model under clear sky conditions. In this study, all of these reactions were tracked and grouped into a small number of formation pathways, such as acetaldehyde, acetone, MACR, MVK, MGLY, other OVOCs, reactions of O₃ with isoprene and MPAN, and propagation of other radicals to PA. The production rate of PA could be estimated as the sum of these reaction rates. The ozone production rates were calculated through the oxidation of NO by HO₂ and RO₂, and its destruction rates were mainly facilitated by O₃ photolysis and reaction with NO, NO₂, OH, HO₂, and unsaturated VOCs. Moreover, we investigated the sensitivities of PAN and O₃ formation to their respective precursor species by introducing a relative incremental reactivity (RIR) concept which is widely applied in the OBM investigation of PAN and ozone formation (Chameides et al., 1999; Xue et al., 2014c). In this calculation, we performed model calculations during the period of 20 August-10 September, 2016, during which the VOCs measurement were available. The model was run based on the hourly average profiles of PAN, O₃, CO, SO₂, NO, NO₂, C₂-C₁₀ NMHCs, air temperature and pressure, and RH measured at NRCS site. During the simulation, the model was pre-run for three days with constrain of the data of 20-22 August so that it reached a steady state for the unmeasured species (e.g., MACR, MVK, HONO, radicals). More detailed description of this model has been given in previous studies (Jenkin et al., 2003; Xue et al., 2014a; Xue et al., 2014c).

2.2.4 Positive matrix factorization (PMF) Model

Positive matrix factorization (PMF) is an effective source apportionment receptor model based on the fingerprints of the sources that does not require the source profiles prior to analysis and has no limitation on source numbers (Hopke, 2003; Pentti and Unto, 1994). The data used in PMF is of the form of an $i \times j$ matrix X , in which i is the sampling number and j is the number of species. Based on chemical mass balance of the pollutants, the following equation can be derived as:

$$X_{ij} = \sum_{k=1}^p g_{ik}f_{ik} + e_{ij}$$

where p is the number of the sources (i.e., the number of factors), f is the profile of each source, g refers to the contribution of each factor to the total concentration, and e is the residual. Factor contributions

204 and profiles are derived by minimizing the total scaled residual Q:

$$Q = \sum_{i=1}^n \sum_{j=1}^m \left(\frac{e_{ij}}{u_{ij}} \right)^2$$

205 where u is the uncertainty of the sampling data. More details about principles have been found
206 elsewhere (Cai et al., 2010; Zhang et al., 2013; Li et al., 2017; Li et al., 2015). In this study, we used
207 EPA PMF 5.0 model to identify major VOCs sources and their temporal variations. We discarded the
208 species that were below MDL for more than 50% of the time or showed a significantly smaller signal to
209 noise ratio (S/N). The uncertainties for each sample and species were calculated based on the following
210 equation if the concentration is greater than the method detection limit (MDL) provided:

$$\text{Uncertainty} = \sqrt{(0.5 \times \text{DML})^2 + (\text{Error Fraction} \times \text{Concentration})^2}$$

211 Values below the detection limit were replaced by one-half of the MDL and their overall uncertainties
212 were set at five-sixths of the MDL values. In this analysis, different numbers of factors were tested. The
213 robust mode was used to reduce the influence of extreme values on the PMF solution. More than 95% of
214 the residuals were between -3 and 3 for all compounds. The Q values in the robust mode were
215 approximately equal to the degrees of freedom.

216 **3 Results and discussion**

217 In order to comprehensively evaluate air quality during the G20 period, we compared the concentrations
218 of pollutants during G20 with the adjacent time period in 2016, respectively. According to the control
219 measures schemes, we classified the whole period into three episodes: one week before G20 (BG20)
220 (16-23 August, 2016), during G20 (DG20) (24 August-6 September) including Phase I (24-27 August)
221 and Phase II (28 August-6 September), and one week after G20 (AG20) (7-15 September).

222 **3.1 Evolutions of meteorological condition**

223 First, we looked into the day-to-day variations of meteorological parameters and atmospheric pollutants
224 from BG20 to AG20 in Fig. S1 in Supplement (SI). In the period of BG20 and the beginning of DG20 I
225 (16-25 August), subtropical anticyclone dominated the Hangzhou and surrounding area, leading to
226 continuous 10 days with daily mean temperature of 31.5 °C ranged from 29.9-32.5 °C and strong solar
227 irradiation intensity (mean daily maximum value: 369.4 W m⁻²), favorable for the photochemical
228 production of O₃ and PAN. The highest O₃ (113.4 ppbv) occurred at 13:00 LT on 25 August under the
229 maximum air temperature of 35.2 °C. Meanwhile, the mean daily maximum height of mixing boundary
230 layer (MBL) during this period was up to ca. 1895 m, beneficial for the diffusion of atmospheric
231 primary pollutants in the vertical direction. In addition, the prevailing wind was from east (15.1%) with
232 a mean wind speed of 2.9 m s⁻¹. Results from the backward trajectory simulations demonstrated that the
233 air masses from the east originated from the East China Sea and Yellow Sea, bringing in clean marine
234 air (Fig. S2). Thus, meteorological conditions before G20 were favorable for the dispersal of

atmospheric pollutants. On 26 and 27 August, the weather pattern changed to a cold continental high with showery and windy days. The total precipitation and mean wind speed both reached their respective maximums of 14.6 mm and 3.7 m s^{-1} on 26 August. Accordingly, all species except CO significantly decreased by 12.3% for SO_2 , 29.7% for NO_x , 6.7% for $\text{PM}_{2.5}$, 11.9% for daily maximum average-8 h (DMA8) O_3 , and 56.1% for PAN relative to BG20. With respect to the last half of DG20 I and the beginning of DG20 II (28-31 August), the prevailing wind experienced a shift from northwest to west and to southwest. On 28 August, the prevailed wind was from the north with the average daily maximum wind speed of 3.9 m s^{-1} during G20, and the relative humidity rapidly decreased by 26.2% relative to the previous day. As seen in Fig. S3, air masses arrived at Hangzhou from the north passed through all of Jiangsu Province and northern parts of Zhejiang Province, two of the most developed provinces in China, with intense human activities. They carried higher $\text{PM}_{2.5}$, SO_2 , NO_x , and CO loadings than the other clusters (See Table S1). On 1 September, the prevailing wind was from southwest with high wind speeds (3.3 m s^{-1}). Results from back trajectories indicated that the southwesterly air masses originated from northern Jiangxi Province, transported over western Zhejiang Province, and arrived at Hangzhou, with high concentration loadings of SO_2 , particulate matter, O_3 , and PAN. The increased relative humidity (56.5%) relative to 49.5% on 31 August was beneficial for the formation of particulate matter. During 2-4 September, Hangzhou area witnessed a stable meteorological condition with weak wind ($w_s < 2.6 \text{ m/s}$), continuously high air temperature (daily maximum average: 32.2°C), and moderate relative humidity (ca. 60%). Such condition was favorable for the accumulation of particulate matter and the photochemical production of O_3 . It caused significant increases by 25.1% for $\text{PM}_{2.5}$, 16.7% for PM_{10} , and 10.7% for O_3 compared with BG20, in contrast to the large decreases by 56.4% for SO_2 and 27.9% for NO_x due to the implement of emission control measures. Overall, the meteorological condition during G20 II was not favorable for the dispersal of atmospheric primary pollutants but beneficial for producing O_3 . However, with the proceeding of the stringent control measures, the most distinct drops of pollutants concentrations were found on 5 September, with the large reductions of 50.0% for $\text{PM}_{2.5}$, 18.3% for DMA8 O_3 , 55.7% for SO_2 , 41.3% for NO_x , and 65.6% for PAN relative to BG20, respectively. Within AG20, 7 rainy days with mean daily total precipitation of 18.7 mm occurred as well as 6 days with low wind speed (ca. 2.0 m/s) and 8 days with low MBL ($<1000 \text{ m}$). Such meteorological condition was beneficial for scavenging the particulate matter and SO_2 by wet deposition in addition to the accumulation of NO_x . In addition, weak solar irradiation intensity was not favorable for the photochemical formation of O_3 and PAN. On 7 September a moderate showery lasted from 2:00 LT to 11:00 LT with daily total precipitation of 9.5 mm, accompanied by low air temperature (21.5°C) and wind speed (1.8 m/s). Compared with the previous day, significant decreases of DMA8 O_3 (22.6%) was found as expected, while together with a small reduction ratio of $\text{PM}_{2.5}$ (2.7%) and unexpected increases for NO_x (41.1%) and SO_2 (175.1%), indicating that emissions immediately bounced back after lifting the ban on emission controls.

3.2 Evolutions of pollutant concentrations

Statistically, observed daytime concentrations of $\text{PM}_{2.5}$, NO_x , and SO_2 in DG20 II both exhibited significant decreases relative to those in BG20 with the reduction ratios of 11.3%, 17.0%, and 18.0%, respectively (Fig. 2). Furthermore, by using WRF-Chem model we quantified the contributions of the emission control measures (ECC) with 63.5%, 44.1%, and 31.2% to the reductions of $\text{PM}_{2.5}$, SO_2 , and NO_2 in DG20 II relative to BG20, respectively, but for the meteorological conditions it made negative contributions. This evidence well indicated that powerful control measures have taken into effect on reducing pollutant emissions in Hangzhou under the unfavorable meteorological conditions. The large decreases of NO_x and SO_2 reflected the reduction of vehicle exhaust and coal consumption during G20 in Hangzhou and surrounding areas. It is worth noting that CO showed gradual increases (ca. 20.7%) from BG20 to DG20, which was mainly attributed to the weak control in fuel combustion. Specifically, residential usage and liquid natural gas and petroleum gas around YRD regions during this period might account for such unique pattern of CO. The other two types of fuel combustion including straw combustion and coal combustion were both excluded as discussed in Section 3.4. Under the same stringent control measures, the variation of O_3 was not consistent with the primary pollutants. Observed DMA8 O_3 increased by 12.4% in DG20 I relative to BG20, which was attributed to regional transport from the northern provinces and the enhanced solar radiation intensity. Afterwards, DMA8 O_3 decreased by 33.4% from DG20 II to AG20 (Fig. 2), as did the peak values of mean daily O_3 in DG20 II compared to BG20 and DG20 I (Fig. S3). This evidence suggests that additional vehicles controls implemented during DG20 II might have played an important role in reducing atmospheric O_3 pollution in Hangzhou reflected by shaping such unique diurnal variation, which was also confirmed by the decreased OFP from vehicle exhaust below. Elevated O_3 during DG20 rush hours (as shown in Fig. S1 and S2) was attributed to the reduced titration of fresh NO emission under the control measures on vehicle exhaust. Considering such effects, O_x (represented by the sum of $\text{O}_3 + \text{NO}_2$) was used to determine the local photochemical formation. The variation of DMA8 O_x was similar with O_3 , with distinct decreasing trend from DG20 II to AG20. For PAN, it showed different pattern with O_3 . Daytime PAN exhibited significant decrease (ca. 45.4%) found from BG20 to DG20 II and then it sharply built up to similar magnitudes in AG20 with BG20. Thereby, it both indicates the significant effectiveness of emission control measures on reducing local photochemical formation of O_3 and PAN. The underlying formation mechanisms of PAN and O_3 including their respective key precursors and chemical process are elucidated in Sect.3.3.

With respect to VOCs, the mixing ratios of total VOCs also showed significant reduction of 20.0% in DG20 compared with BG20, but increased by 104.1% in AG20 after control (Table S2). Alkanes were the most abundant VOCs group (55.4%) in all periods, and were reduced by 19.8% from BG20 to DG20. On the contrary, alkenes increased by 20.0% in DG20 compared to BG20, among which ethylene

accounted for 63.9%-78.0% during the three periods, although other alkenes decreased to a minor extent. As expected, aromatics were reduced by 49.7% in DG20 compared with BG20. Ambient mixing ratios of specific VOCs at NRCS station are summarized in Table S3. Ethane, ethylene, benzene, and toluene are the four most abundant species during all the periods. Compared with BG20, except ethane, isopentane, and ethylene, the mixing ratios of all species decreased in DG20. Ethylene, as a representative tracer of fuel combustion, showed continuous increase from BG20 to AG20, possibly indicating the ineffectiveness of control measures in this source.

3.3 Identification of the Key Precursors and Chemical Processes for PAN and O₃

To identify the key precursors and chemical processes for PAN, we employed the observation-based model to investigate the daytime average contributions to PA radical production rates directly from individual pathways for these four episodes (Fig. 3). Acetaldehyde (e.g., oxidation of OH and NO₃) and MGLY (e.g., photolysis and oxidation by OH and NO₃) were the most important sources of PA in Hangzhou, accounting for 37.3-51.6% and 22.8%-29.5% of the total production rates. This was in agreement with the findings obtained from the other typical urban areas such as Beijing (Xue et al., 2014c; Liu et al., 2010; Zhang et al., 2015), Tokyo (Kondo et al., 2008), Houston, Nashville (Roberts et al., 2001), and Sacramento (LaFranchi et al., 2009). Reactions of OVOCs and propagation of other radicals to PA (mainly including decomposition of some RO radicals and reactions of some higher acyl peroxy radicals with NO) were also significant sources, with average contributions of 7.1%-9.1% and 18.1%-27.0%, respectively. A minor contribution (~1% in total) was originated from the other pathways of O₃+isoprene, O₃+MPAN, acetone, and MVK. Acetaldehyde and other OVOCs are mainly photooxidation products of hydrocarbons, thus it's necessary to further identify the first-generation precursors of PAN here. We tested the model sensitivity by introducing the concept of relative incremental reactivity (RIR), which is widely used in the OBM study of ozone formation (Chameides et al., 1999). Here RIR is defined as the ratio of decrease in PAN production rates to decrease in precursor concentrations (e.g., 20% reduction is used in this study). A number of sensitivity model runs were performed to calculate the RIRs for NO_x, alkanes, alkenes, and aromatics classes as well as the individual C₂-C₁₀ hydrocarbon species. As shown in Fig. 4a, production of PAN was sensitive to VOCs from BG20 to AG20. Meanwhile, the negative RIR values for NO_x also indicated a VOCs regime of PAN production around the G20 period in urban Hangzhou. In terms of BVOCs, the positive RIRs values for isoprene (0.18-0.38) from BG20 to AG20 implied that in-situ formation of PAN at NRCS was highly sensitive to isoprene. As to AVOCs, alkenes and aromatics were the most important first-generation PAN precursors, with the RIRs range of 0.24-0.37 and 0.26-0.52, respectively. Furthermore, we identified the other specific VOCs controlling PAN production, which were xylenes, trans/cis-2-butenes, trimethylbenzenes, toluene, and propene evidenced by their positive RIRs. Compared with their individual RIRs between control and non-control period, the in-situ production of

PAN was dominated by aromatics in BG20 and DG20 I, but controlled by alkenes in AG20. Besides secondary acetaldehyde formed by the oxidation of ethanol, most aromatics were mainly emitted by vehicle exhaust. The decreased RIRs of aromatics together with the decreased contribution ratios of acetaldehyde to the PA radical formation during G20 both indicated the effectiveness of control measures on vehicle exhaust on reducing atmospheric PAN concentration. Similar with PAN, the daytime average RIRs for major groups of O₃ precursors during the episodes are shown in Fig. 4b. Overall, the in-situ ozone formation was also controlled by VOCs from BG20 to AG20. AVOCs were dominated by alkenes and aromatics, along with their increasing and decreasing RIRs, respectively. With the proceeding of emission control, the RIR for AVOCs showed gradual decrease from BG20 to DG20, but increased after G20. In contrast, BVOCs (mainly as isoprene) exhibited gradual increases for all periods, especially during the phase II in DG20 and AG20 when their RIRs were both higher than those for AVOCs. Thereby, the contribution of BVOCs to the photochemical production of O₃ weakened the effect of stringent control measures on reducing surface O₃. The RIRs for NO_x were negative throughout the period of G20, also indicating a VOC-limited regime for the sensitivity of ozone formation. This suggests that reducing emissions of aromatics, alkenes, and alkanes would alleviate the O₃ formation, yet cutting NO_x emissions may aggravate the local O₃ problems.

3.4 Identification of VOCs sources and quantification of their respective ozone formation potential

To distinguish the various sources of VOCs, we compared the PMF profiles with the reference profiles from the literature as listed below. Seven sources were identified as follows: (1) gasoline evaporation (2) solvent utilization (3) industrial manufacturing (4) industrial chemical feedstock (5) vehicle exhaust (6) fuel combustion (7) biogenic emission. Figure 5 exhibited the modelled source profiles together with the relative contributions of each sources to individual species. The first source is characterized by a significant amount (78.5%) of isopentane which is a typical tracer for gasoline evaporation (Liu et al., 2008). Therefore, this source was identified as gasoline evaporation. The second source was rich in n-pentane and aromatics. Many aromatics such as BTEX are the dominant components of organic paints, and were regarded as chemical tracers of solvent utilization (Watson et al., 2001). Significant amounts of ethylbenzene, xylenes, and n-pentane present in the second source, accounting for 19.2%, 58.8%, and 98.8%, respectively. Thus, the second source was identified as solvent utilization. The third source was characterized by high loading of cyclohexane (54.7%) and BTEX (15.1%-46.2%). These compounds are confirmed to be typical species in the industrial manufacturing in China (Liu et al., 2008). Thus, this source was representative of industrial manufacturing. The fourth source identified as industrial chemical feedstock (shown in Fig. 5) was characterized by a very little contribution to alkanes and aromatics and large amounts of 3-ethyltoluene (29.4%), 3-methylheptane (51.0%), and n-hexane (47.1%), which are typical proxies for industrial chemical feedstock (Liu et al., 2008; Mo et al., 2015).

376 The fifth source was characterized by abundant 2-methylpentane (61.7%) and BTEX, which is a typical
 377 tracer for vehicle exhaust (Liu et al., 2008; Li et al., 2015). In addition, 2, 2, 4-trimethylpentane is a fuel
 378 additive used to gain higher octane ratings (McCarthy et al., 2013) with high abundance of 21.4% in
 379 this source and thus it is identified as vehicle exhaust. The sixth source profile shown in Fig. 5 was in
 380 relation to 48.9% of the total measured ethylene mixing ratios, of which was major species emitted from
 381 fuel combustion process (Li et al., 2015). It was also characterized by significant amounts of ethane,
 382 propane, n-butane, propene, and benzene. Ethane and propane are the tracers of natural gas and liquid
 383 petroleum gas (LPG) usage, respectively, and the source profiles of resident fuel combustion in China
 384 contained alkenes (Wang et al., 2013). Coal combustion can release a large amount of BTEX into the
 385 atmosphere and styrene is a typical indicator of industrial manufacturing in China (Liu et al., 2008; Li et
 386 al., 2015). Thus, this source was believed to be as fuel combustion related with industrial process and
 387 residual usage. The seventh source was distinguished by a significant amount of isoprene, a
 388 representative indicator of biogenic emission. About 93.1% of the total isoprene mixing ratios is
 389 apportioned to this factor (Guenther et al., 1995). There were very small quantities of the other species
 390 such as aromatics (0-1.8%) in this factor. Therefore, it was excluded from biomass burning but mainly
 391 identified as biogenic emission. Figure 6 shows the variation of the seven sources during the four
 392 periods. Clearly, anthropogenic sources such as solvent utilization, industrial manufacturing, vehicle
 393 exhaust, fuel combustion, and industrial chemical feedstock were the predominant sources to the total
 394 VOCs before and after G20, as high as 52.4%-81.7%. Furthermore, anthropogenic emission showed
 395 significant reductions during G20 response to the stringent emission control. In BG20, solvent
 396 utilization was the predominant contributors to VOCs mixing ratios, contributing 1.88 ppbv, followed
 397 by vehicle exhaust (1.77 ppbv, 21.6%), industrial manufacturing (1.55 ppbv, 19.0%), biogenic emission
 398 (1.16 ppbv, 14.1%), gasoline evaporation (0.83 ppbv, 10.1%), and fuel combustion (0.35 ppbv, 4.3%).
 399 The industry-related emission (industrial manufacturing, chemical feedstock, and solvent utilization)
 400 together accounted for 50.0% of the total VOCs mixing ratios. The vehicle-related emission sources
 401 (vehicle exhaust and gasoline evaporation), accounted for 31.7% of the total VOCs mixing ratios. It
 402 indicated that traffic and industry sources were the major VOCs sources before the control period.
 403 Compared with BG20, the contribution of solvent utilization was reduced to the largest extent, with a
 404 magnitude of 1.43 ppbv, followed by industrial manufacturing (0.69 ppbv), and vehicle exhaust (0.38
 405 ppbv), during the first emission control period (DG20 I). According to the control strategy during G20,
 406 the control measures of source emission were mainly on the industry and power plant in DG20 I, and
 407 thus it was responsible for the large reduction of industry-related emission including solvent utilization
 408 (76.0%), industrial manufacturing (44.0%), and vehicle exhaust (21.0%). With the acceleration of
 409 emission control (DG20 II), the contribution of vehicle-related emission was reduced as expected in
 410 vehicle exhaust (66.1%) and gasoline evaporation (61.8%) relative to DG20 I, while significant increase
 411 was also found in fuel combustion with the increment of 0.7 ppbv (152.6%). After G20, the

contributions of vehicle-related emission and industry-related emission both showed bounces due to lifting a ban on industry, power plant, and transport in and around Zhejiang Province. It should be mentioned that biogenic emission also played an indispensable importance in contributing to the VOCs mixing ratios, from 0.81 ppbv to 1.29 ppbv. About 20.9% of the total VOCs mixing ratios could be ascribed to the biogenic emission, acting as the second major source, during the G20 II period. It indicated that biogenic VOCs might make more contribution to the VOCs mixing ratios especially when anthropogenic VOCs were substantially reduced following the process of control measures. Moreover, we quantified their respective ozone formation potential (OFP) before, during, and after G20 by using the latest maximum incremental reactivity (MIR) and the appointed concentration profiles above (See Fig. 7). Overall, the total OFP in DG20 was significantly reduced by the implement of stringent control measures compared with BG20 and AG20. Specifically, the OFPs of solvent utilization, industrial manufacturing, and vehicle exhaust both showed significant decreases (17.3%-77.2%) compared with BG20, while fuel combustion significantly increased by 52.2% with the OFP of 6.9 ppbv, accounting for 25.6% of the total during G20. Thus, it is clear that the high OFP of fuel combustion contributed by ethylene was also responsible for the enhanced concentration of O₃ during G20. Such high OFP from fuel combustion was also elucidated in APEC in Beijing (Li et al., 2015). To classify the specific fuel type, we first examined the fire spots derived from the Fire Inventory NCAR Version-1.5 (FINNV1.5) in eastern China before, during, and after the period of 2016 G20 (See Fig. S4 in SI). Straw combustion was excluded according to the decrease in the number of fire spots in the same time period from BG20 to AG20. As mentioned above, industrial process with coal combustion was strictly limited throughout the whole G20 period. To ensure the clean energy used in 2016 G20, local government accelerated the supply of liquid natural gas during the 13th Five-Year Plan period in Hangzhou. In 2016, the consumption amounts of natural gas and liquid petroleum gas both increased up to 4.55×10⁹ kg (54.4%) and 5.09×10⁸ kg (13.4%) compared with those in 2015, respectively (ZPSY, 2016, 2017). Thus, liquid natural gas and petroleum gas were identified as the major fuel used in the residential usage during G20. After G20, all anthropogenic sources both showed significant increments of OFP, among which the fastest growth of source was vehicle exhaust (17.6 ppbv, 638.4%), followed by fuel combustion (9.4 ppbv, 35.1%), industrial manufacturing (7.7 ppbv, 89.2%), and solvent utilization (7.4 ppbv, 258.1%), respectively.

4 Conclusions

In this study, ground-based concentrations of atmospheric trace gases and particulate matter, together with meteorological parameters, were measured at a NRCS site in urban Hangzhou before, during, and after G20. We found significant decreases in atmospheric VOCs, PM_{2.5}, NO_x, and SO₂ in DG20 relative to BG20 and AG20, respectively, under the unfavorable meteorological conditions (e.g., stable weather pattern and regional transport). This evidence well indicated that the powerful control measures have

447 taken effect in their emissions in Hangzhou. On the contrary, observed DMA8 O₃ increased from BG20
448 to DG20 I, which was attributed to the regional transport from the northern provinces and the enhanced
449 solar radiation intensity, and then decreased from DG20 II to AG20. The decreases in the peak
450 concentration of daily O₃ and the OFP estimated from various VOCs sources both suggested the
451 effectiveness of stringent control measures on reducing atmospheric O₃ concentrations. Unlike O₃, PAN
452 exhibited gradual decrease from BG20 to DG20. With the OBM model, we found acetaldehyde and
453 methyl glyoxal (MGLY) to be the most important second-generation precursors of PAN, accounting for
454 37.3-51.6% and 22.8%-29.5% of the total production rates. Furthermore, we confirmed that the
455 production of PAN was sensitive to anthropogenic and biogenic VOCs (isoprene) throughout the whole
456 period, specifically aromatics in BG20 and DG20 I but alkenes in AG20. Similarly, the sensitivity of
457 ozone formation was also under VOC-limited regime throughout G20 period. These findings suggest
458 that reducing emissions of alkanes, alkenes, and aromatics would mitigate photochemical smog
459 including PAN and O₃ formation. Furthermore, traffic (vehicle exhaust and gasoline evaporation) and
460 industrial sources (solvent utilization, industrial manufacturing, and chemical feedstock) were found to
461 be the major VOCs sources before G20, accounting for ca. 50.0% and 31.7% of the total, respectively,
462 with the ozone formation potential (OFP) of 14.4 ppbv and 16.1 ppbv. Large decreases were found in
463 the sources and OFPs of solvent utilization (74.1% and 17.3%), followed by vehicle exhaust (57.4% and
464 77.2%) and industrial manufacturing (56.0% and 40.3%) response to the stringent control measures
465 during G20. We also appeal to pay attention on controlling fuel combustion and biogenic emission
466 especially when anthropogenic VOCs were substantially reduced following the process of control
467 measures.

468 **Author contributions.** GZ and HX designed research; HW, BQ, RD, and XM performed research, GZ,
469 LX, JH, WX, CL, LL, ZL, KG, YY, and WJ analyzed data; and GZ, HX, LX wrote the paper.

471 **Data availability.** The data in the figures in both the main text and the Supplement are available upon
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474 **Competing interests.** The authors declare that they have no conflict of interest.

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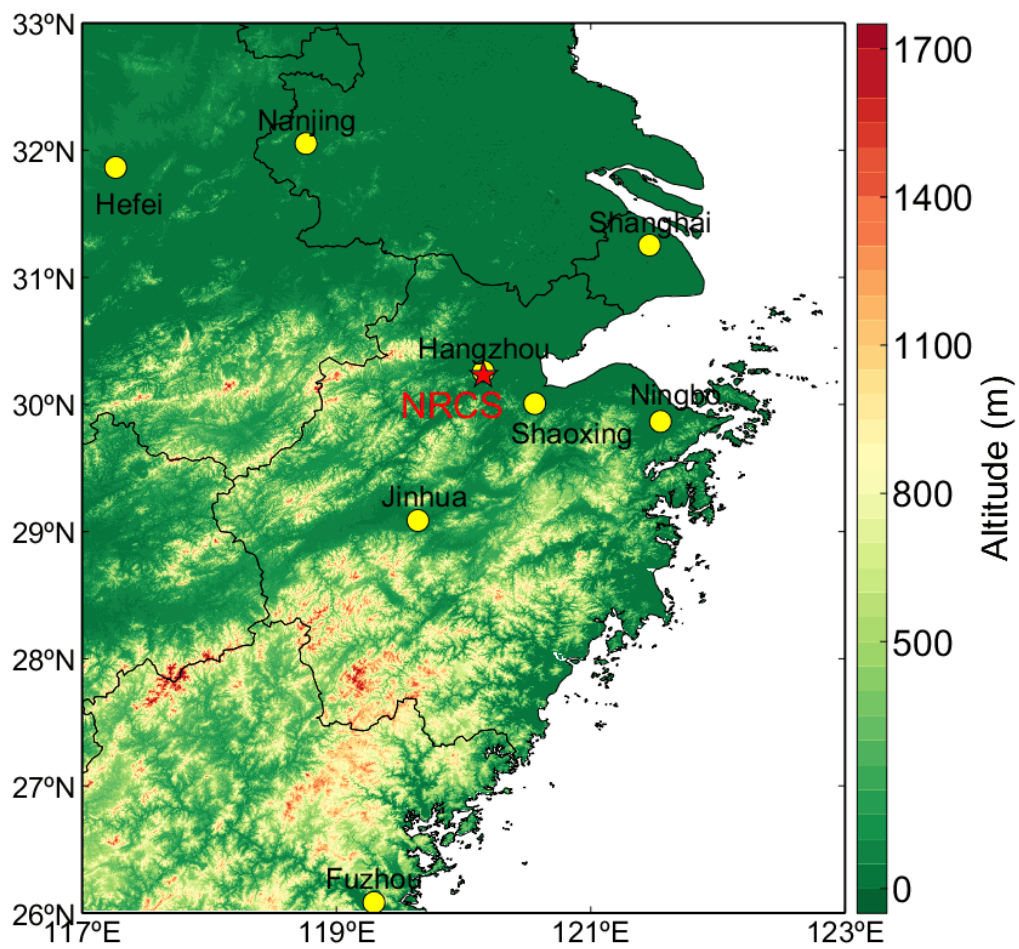
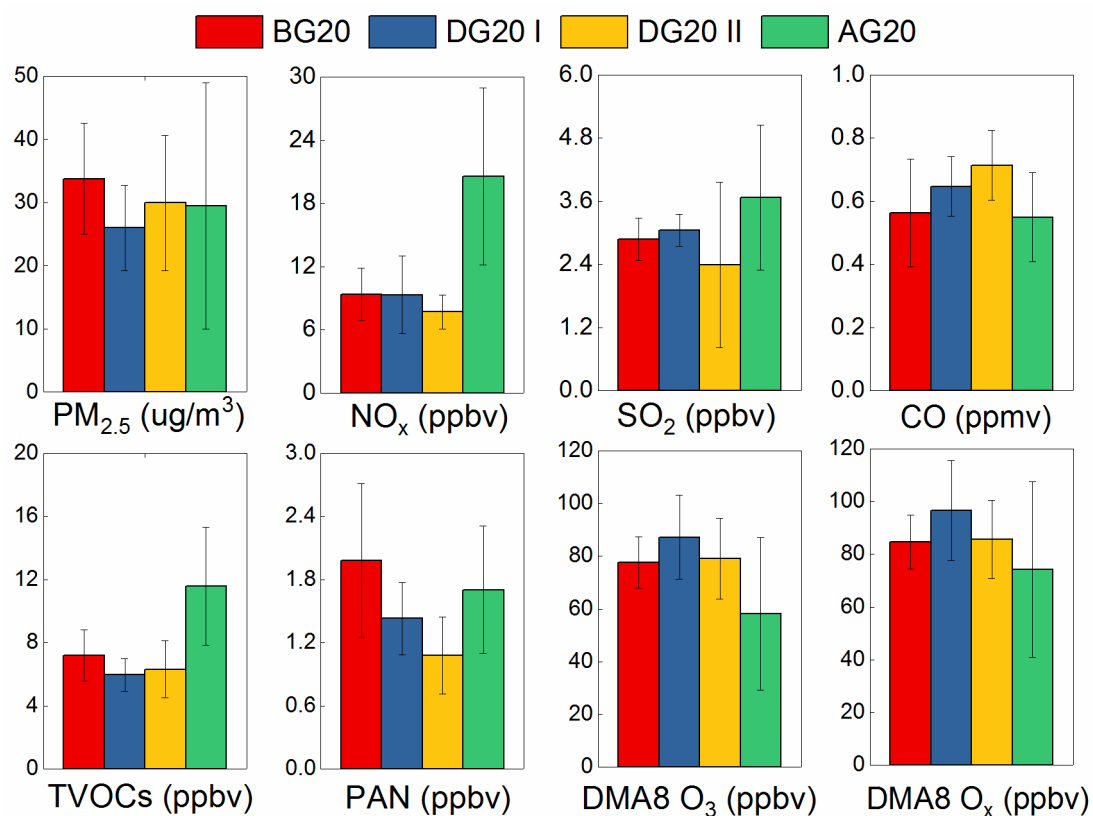


Figure. 1. The topography of National Reference Climatological Station (NRCS) (30.22 °N, 120.17 °E, 41.7 m a.s.l) in Hangzhou, China. The pentagram represents the location of NRCS.



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Figure 2. The comparisons of daytime PM_{2.5}, NO_x, SO₂, CO, TVOCs, PAN, DMA8 O₃, and DMA8 O_x. before, during, and after G20, denoted as BG20, DG20, and AG20, respectively. The error bars represent the standard deviations.

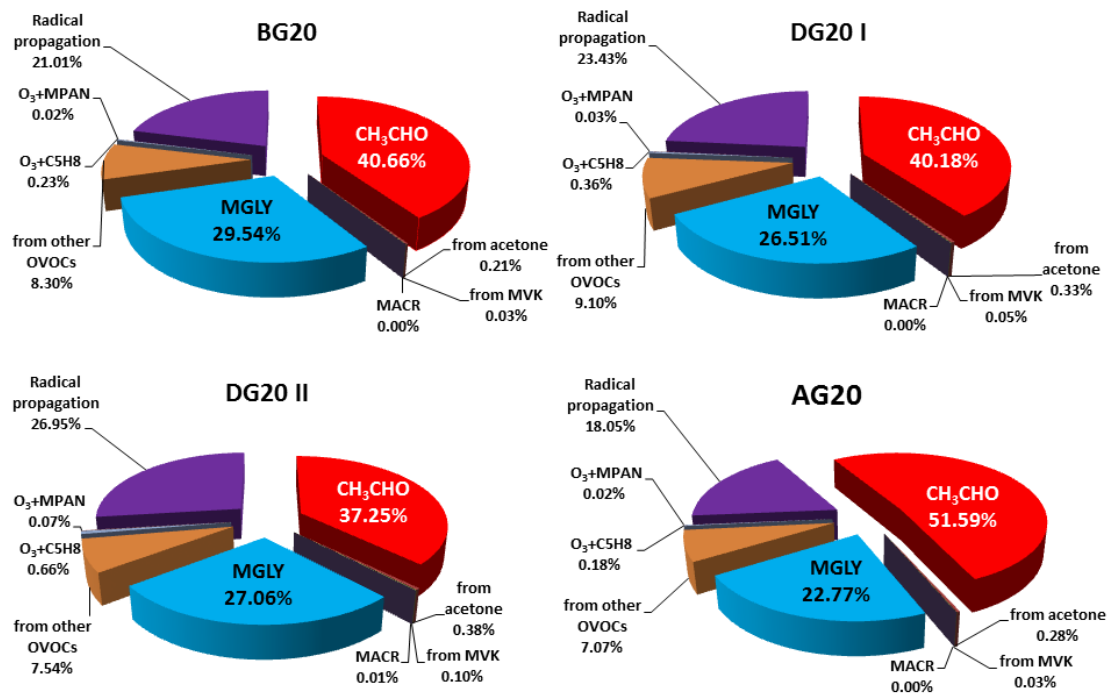


Figure 3. Contributions of individual pathways to PA radical formation during the episodes of BG20, DG20 I, DG20 II, and AG20, respectively.

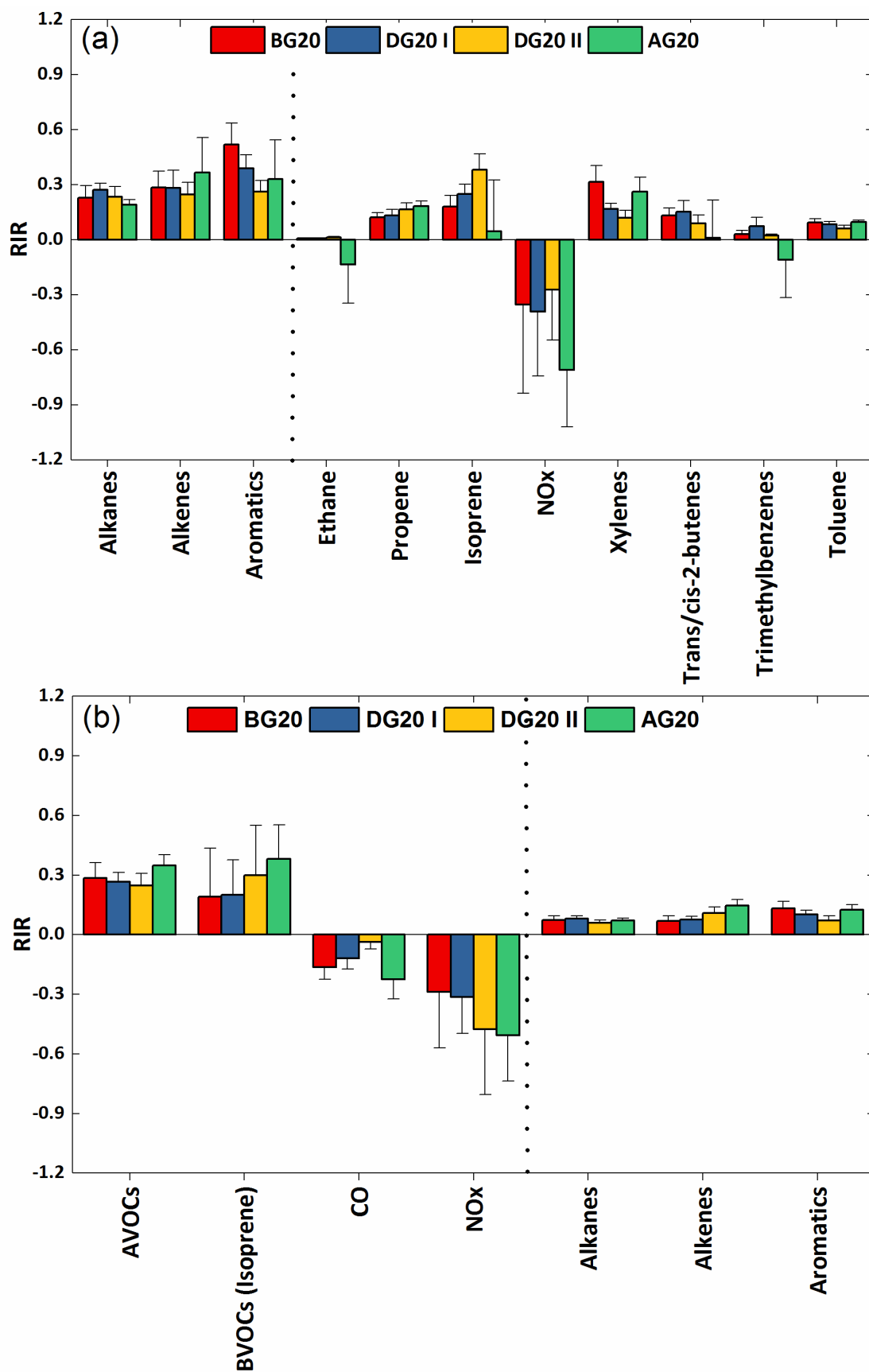


Figure 4. Sensitivity of PAN (a) and O₃ (b) production rate to major precursor groups and individual species (09:00-17:00). Error bars are standard deviations.

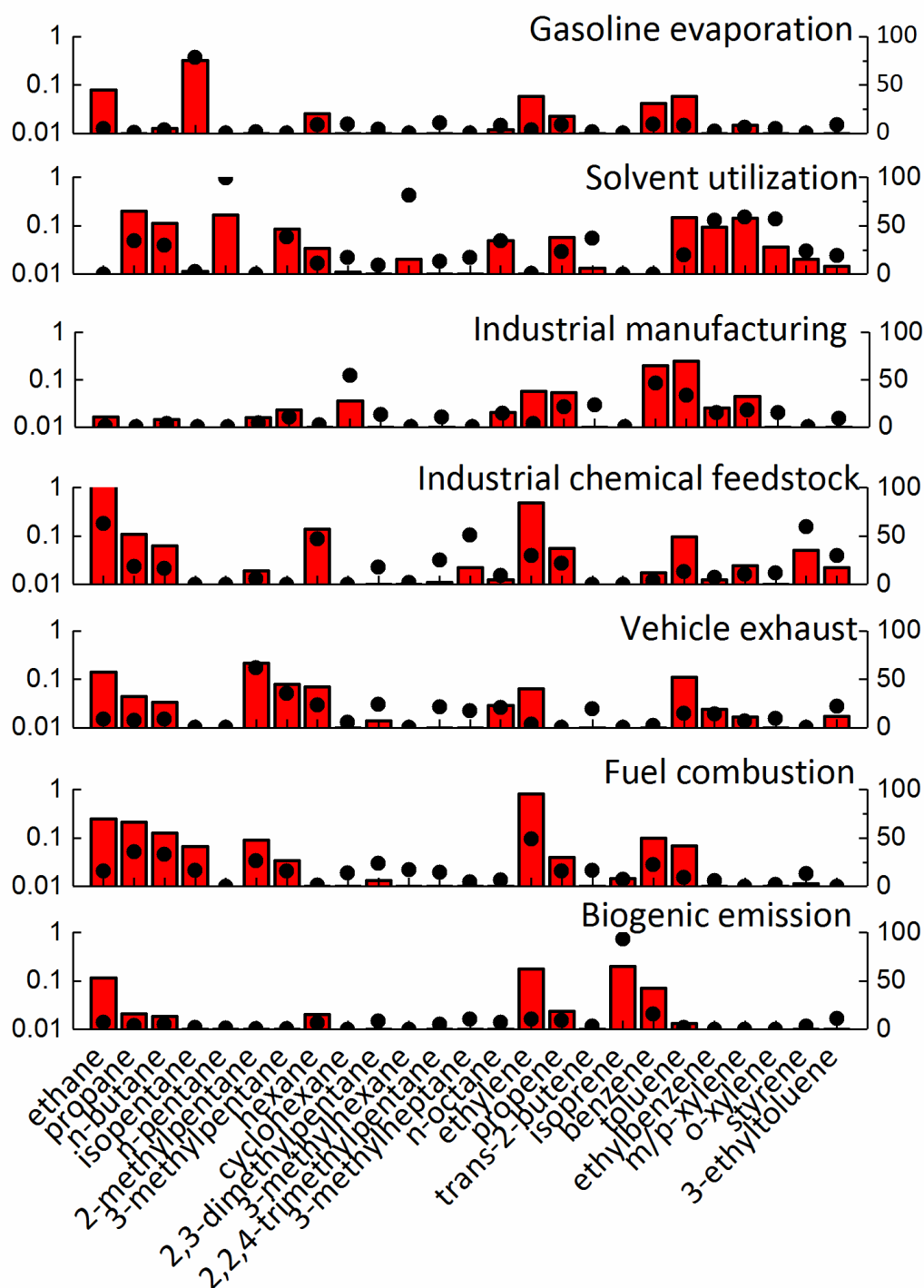


Figure 5. Seven source profiles and their respective contribution resolved from PMF model. The bars are the profiles (ppbv, left axis), and the dots are the percentage contribution (% , right axis) from individual factor.

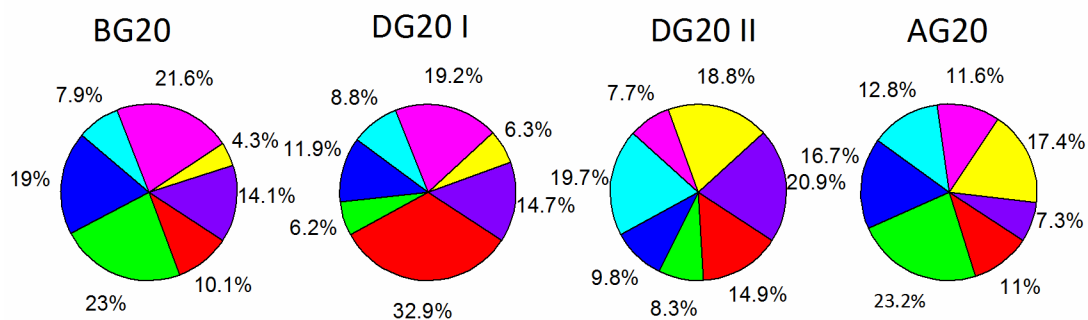
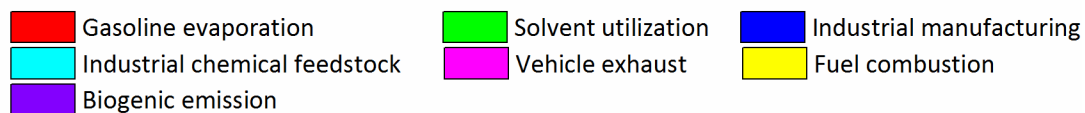


Figure 6. Variation of the sources (percentage) during the four periods

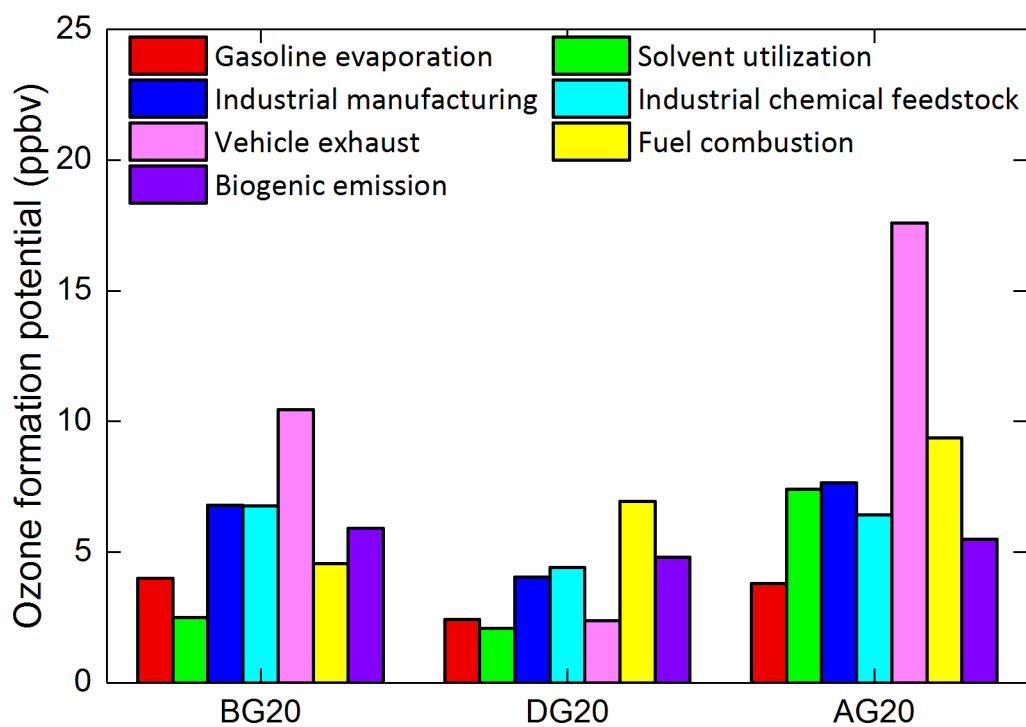


Figure 7. Ozone formation potential (ppbv) of each source before, during, and after the control period during 2016 G20 in China