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Strong Sensitivity of the Isotopic Composition of Methane to the Plausible Range of Tropospheric Chlorine

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Abstract. The ^{13}C isotopic ratio of methane, $\delta^{13}C$ of CH_4, provides additional constraints on the CH_4 budget to complement the constraints from CH_4 observations. The interpretation of $\delta^{13}C$ observations is complicated, however, by uncertainties in the methane sink. The reaction of CH_4 with Cl is highly fractionating, increasing the relative abundance of $^{13}CH_4$, but there is currently no consensus on the strength of the tropospheric Cl sink. We use a set of GEOS global model simulations with different predicted Cl fields to test the sensitivity of the $\delta^{13}C$ of CH_4 to the diversity of Cl output from chemical transport models. We find that $\delta^{13}C$ is highly sensitive to both the amount and geographic distribution of Cl. Simulations with Cl providing 0.28% or 0.66% of the total CH_4 loss bracket the $\delta^{13}C$ observations for a fixed set of emissions. Thus, even when Cl provides only a small fraction of the total CH_4 loss and has a small impact on total CH_4, it provides a strong lever on $\delta^{13}C$. The geographic distribution and seasonal cycle of Cl also impacts the hemispheric gradient and seasonal cycle of $\delta^{13}C$. The large effect of Cl on $\delta^{13}C$ compared to total CH_4 broadens the range of CH_4 source mixtures that can be reconciled with $\delta^{13}C$ observations. Stronger constraints on tropospheric Cl are necessary to improve estimates of CH_4 sources from $\delta^{13}C$ observations.

1. Introduction

The global budget of methane is of great interest due to methane's role as a greenhouse gas, ozone precursor, and sink of the hydroxyl radical. Despite extensive study, major uncertainties in the methane budget remain, with top-down and bottom-up estimates often yielding different results (Kirschke et al., 2013;Saunois et al., 2016;Saunois et al., 2017, and refs therein) for the strength of specific source types. Furthermore, the resumed increase of methane concentrations beginning in 2007 (Dlugokencky et al., 2009; Rigby et al., 2008) can be explained by multiple

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hypotheses including an increase in fossil fuel emissions (Turner et al., 2016;Thompson et al., 2015;Hausmann et al., 2016), an increase in fossil fuel emissions combined with a decrease in biomass burning (Worden et al., 2017), an increase in biogenic sources (Schaefer et al., 2016;Nisbet et al., 2016), or a decrease in hydroxyl concentrations (Turner et al., 2017;Rigby et al., 2017). Variations in hydroxyl concentrations may also be important for the decrease in methane growth from 1999-2006 (McNorton et al., 2016).

Observations and modeling of methane's carbon isotopes provides additional information on methane sources since individual sources differ in their ^{13}C to ^{12}C ratio ($\delta^{13}C$). Isotopic information can be used to better constrain methane sources (e.g. Thompson et al., 2015; Mikaloff Fletcher et al., 2004b, a) and infer how the source mixture changed over glacial (e.g. Hopcroft et al., 2018; Fischer et al., 2008; Bock et al., 2017), millennial (e.g. Ferretti et al., 2005; Houweling et al., 2008), and decadal timescales (e.g. Nisbet et al., 2016; Schaefer et al., 2016; Kai et al., 2011; Schwietzke et al., 2016; Thompson et al., 2018). However, there are considerable uncertainties in the processes that control methane's isotopic composition that may confound source apportionment studies. Many modeling studies use a single value for the isotopic ratio of each source, while in reality sources such as wetlands, biomass burning, and natural gas show large regional or environment-dependent variations in their isotopic signature (Ganesan et al., 2018; Brownlow et al., 2017; Dlugokencky et al., 2011; Schwietzke et al., 2016; Sherwood et al., 2017).

The isotopic composition of atmospheric methane is also sensitive to methane's sinks. Reaction with OH, the principal loss for atmospheric methane, has a kinetic isotope effect (KIE) of −5.4‰ ($\alpha=k_{13}/k_{12}=0.9946$) to −3.9‰ ($\alpha=0.9961$) (Saueressig et al., 2001; Cantrell et al., 1990) and contributes to the interhemispheric gradient of $\delta^{13}C$ (Quay et al., 1991). Mass balance (Lassey et al., 2007) and observations of the seasonal cycle of $\delta^{13}C$ versus methane concentration, however, suggest larger apparent KIE values, which may indicate a role for methane oxidation by chlorine (Cl) in the marine boundary layer (MBL) (Allan et al., 2001;Allan et al., 2007) since Cl has a KIE of -61.9‰ ($\alpha=0.938$) at 297K (Saueressig et al., 1995). Inclusion of the MBL Cl sink alters the source mixture inferred from inverse modeling of $\delta^{13}CH_4$ (Rice et al., 2016). Nisbet et al. (2019) point out that interannual variability in the CH_4 Cl sink could explain some of the variability of $\delta^{13}C$. Cl is also an important methane sink in the stratosphere, and the impact of this sink on surface $\delta^{13}C$ is a source of uncertainty in modeling $\delta^{13}C$ (Ghosh et al., 2015). Reaction with stratospheric Cl contributes approximately 0.23‰ to the $\delta^{13}C$ of surface methane and makes a small contribution to the observed trend in surface $\delta^{13}C$ over the last century (Wang et al., 2002).

The global concentration of Cl in the MBL and its role in the methane budget is still uncertain. Cl concentrations are highly variable and not well constrained by direct observations. Modeling work by Hossaini et al. (2016) and Sherwen et al. (2016) suggests that chlorine provides 2-2.5% of tropospheric methane oxidation. This agrees well with estimates based on the isotopic fractionation, which also suggest Cl provides several percent of the total sink (Allan et al., 2007;Platt et al., 2004). However, Gromov et al. (2018) suggest that these are overestimates as values over 1% are inconsistent with the $\delta^{13}C$ of CO, which is a product of CH_4 oxidation. The recent modeling study of Wang et al. (2019) also suggests a value of 1%. There is thus considerable uncertainty in the role of chlorine in the budget and isotopic composition of methane.

Here, we investigate the sensitivity of $\delta^{13}C$ of CH_4 to tropospheric chlorine concentrations in a global model to better quantify how much uncertainty in the interpretation of $\delta^{13}C$ is imposed by the uncertainty in Cl. Section 2

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describes the modeling framework. We present results for total CH_4 and its isotopic composition compared to surface observations in Section 3, and discuss the implications for the global CH_4 budget in Section 4.

2. Methods

2.1 Model Description

We simulate atmospheric methane with the Goddard Earth Observing System (GEOS) global earth system model (Molod et al., 2015; Nielsen et al., 2017). The model has 72 vertical levels extending from the surface to 1 Pa. We conduct simulations at C90 resolution on the cubed sphere, which corresponds to approximately 100 km horizontal resolution. The simulations' meteorology is constrained to the MERRA-2 reanalysis (Gelaro et al., 2017) using a "replay" method (Orbe et al., 2017). The GEOS replay agrees well with the tropospheric mean age of the GMI CTM (Orbe et al., 2017), which shows reasonable agreement with the age derived from SF_6 observations, albeit with an old bias in the southern hemisphere (Waugh et al., 2013). We thus expect the simulated interhemispheric transport time to be reasonable.

The GEOS CH_4 simulation can be interactively coupled to CO and OH (Elshorbany et al., 2016), or run independently with prescribed OH fields. We take the latter approach in this study, since this approach is able to capture many of the observed variations in atmospheric methane (Elshorbany et al., 2016). We prescribe the OH field following (Spivakovsky et al., 2000), but modify the OH to be approximately 20% higher in the Northern Hemisphere than the Southern Hemisphere, consistent with the OH field produced by many global atmospheric chemistry models (Naik et al., 2013;Strode et al., 2015). We also include stratospheric losses for CH_4 from reaction with OH, Cl, and O^1D. These fields are prescribed from output of the Global Modeling Initiative (GMI) chemical transport model (https://gmi.gsfc.nasa.gov) (Strahan et al., 2007; Duncan et al., 2007).

We implement the CH_4 isotopes in GEOS by separately simulating $^{13}CH_4$ and $^{12}CH_4$ tracers. We then calculate total CH_4 as the sum of the two carbon isotopologues and calculate $\delta^{13}C$ of CH_4 in per mil using the standard definition:

$$\delta^{13}\text{C-CH}_4\ (‰) = ([^{13}CH_4]/[^{12}CH_4]/R_{std} - 1) * 1000 \quad (1)$$

where R_{std}=0.112372 is the peedee belemnite isotopic standard (Craig, 1957). The reaction rates for CH_4+OH, CH_4+Cl, and CH_4+O^1D differ between the $^{12}CH_4$ and $^{13}CH_4$ simulations to account for the kinetic isotope effect (KIE). In particular, we assume α values of 0.987 and 0.938 for CH_4+O^1D and CH_4+Cl, respectively (Saueressig et al., 1995;Saueressig et al., 2001). Our standard simulation uses α_{OH} = 0.9946 (Cantrell et al., 1990).

Methane from different sources is tracked individually using a "tagged tracer" approach, which allows us to simulate the spatial footprint of CH_4 and δ^{13}C-CH_4 from individual sources. We partition each emission source into $^{12}CH_4$ and $^{13}CH_4$ emissions according to a source-specific $\delta^{13}C$ value from the literature, provided in Table 1. The soil sink is applied to each tracer as a fraction of its source, modified to account for faster loss of $^{12}CH_4$ to soil compared to $^{13}CH_4$ (α_{soil} = 0.978). Figure 1 shows the CH_4 and δ^{13}C-CH_4 footprints of the biomass burning, wetland, and coal + other geologic CH_4 sources from the tagged tracers to illustrate the tagged tracer approach. We note that the $\delta^{13}C$ values of the surface methane from each source is heavier (less negative) than the emission value for that source (Table

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1), especially in regions far from the source, because of the fractionating effects of the sinks. Supplemental Fig. S1 shows the corresponding footprints for January.

2.2 Description of Simulations

We simulate the period from 1990 through 2004, and focus our analysis on 2004. We choose 2004 as our endpoint because it lies within the period when methane concentrations remained relatively flat, simplifying our analysis. Ending the simulations in 2004 also avoids much of the uncertainty about the causes of the resumed growth rate in recent years. The isotopic ratios of methane take longer to adjust to a perturbation than total methane (Tans, 1997). Since we wish to begin our simulations with a state that is as close as possible to "spun up", we choose the initial condition for each tagged tracer based on its present-day distribution and proportion of the total CH_4 and scale it back to 1990 levels such that the total CH_4 is consistent with the global mean CH_4 from surface observations for 1990. We then iteratively adjusted the ^{12}C- to ^{13}C-CH_4 tracer ratios at the beginning of 1990 to yield a good match to global mean $\delta^{13}C$-CH_4 observations for 1998, when more $\delta^{13}C$-CH_4 observations are available. The same initial condition is used for the standard and sensitivity simulations.

We use interannually-varying emissions of CH_4 from anthropogenic, biomass burning, and wetland sources. Emissions from anthropogenic sources such as oil and gas, energy production, industrial activities, and livestock come from the EDGAR version 4.2 inventory (European Commission, 2011). Biomass burning emissions come from the MACCity inventory (Granier et al., 2011). We treat forest fires as C3 burning and savannas as C4 burning for partitioning the biomass burning emissions between isotopologues. Wetland and rice emissions come from the Vegetation Integrative Simulator for Trace gases (VISIT) terrestrial ecosystem model (Ito and Inatomi, 2012), scaled by 0.69 and 0.895, respectively, for consistency with the Transcom-CH_4 study (Patra et al., 2011). Ocean (Houweling et al., 1999), termite (Fung et al., 1991), and mud volcano emissions (Etiope and Milkov, 2004) are also from the Transcom study (Patra et al., 2011) and have a seasonal cycle but no interannual variability. Initial tests with these emissions showed a substantial underestimate of the CH_4 growth rate. Consequently, we scale up all the emissions by 10% for 1990-1998, and by 6.8% for 1998-2004. We find the resulting emissions lead to a good simulation of the timeseries of surface CH_4 observations from the NOAA GMD (Dlugokencky et al., 2018), especially towards the end of the period (Fig. 2).

Our standard simulation (SimStd) uses Cl from the GMI CTM for the tropospheric as well as stratospheric loss of CH_4 by reaction with Cl. Tropospheric Cl concentrations are small in GMI since it does not include very short-lived species, and reaction with Cl represents only 0.28% of the total tropospheric CH_4 loss. We also conduct several sensitivity simulations in which we alter the tropospheric Cl fields (Table 2). Sensitivity simulation SimGC uses Cl from the GEOS-Chem chemistry module within GEOS (Long et al., 2015; Hu et al., 2018). GEOS-Chem v11-02f with fully coupled tropospheric and stratospheric chemistry was used for this simulation, with halogen chemistry as described in Sherwen et al. (2016). SimGC has higher values of tropospheric Cl than SimStd (Figs. 3,4) and leads to 0.66% of the total CH_4 loss occurring via Cl. Both SimStd and SimGC are thus below the 1% loss via Cl suggested by (Gromov et al., 2018). We conduct a third sensitivity simulation, SimTom, which uses Cl from the TOMCAT model simulations that include chlorine sources from chlorocarbons (including very short-lived substances), HCl from industry and biomass burning, and very short lived substances (Hossaini et al., 2016). This simulation leads to Cl

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accounting for 2.5% of CH_4 loss. Finally, we conduct a fourth sensitivity simulation, SimMBL, which modifies the Cl over the oceans at altitudes below 900 hPa (Fig. 3d) to reflect the marine boundary layer distribution suggested by (Allan et al., 2007). This Cl field is described by the following equation:

$$Cl\_MBL = 18*10^3 \text{ atoms/cm}^3 * (1 + \tanh(3\lambda)*\sin(2\pi*(t-90)/365)) \quad (2)$$

where λ is latitude in radians and t is the day of the year. Elsewhere SimMBL uses the Cl field from SimStd. This simulation has the highest percent of CH_4 loss occurring via Cl: 3.9%.

We designed the sensitivity experiments to alter the isotopic composition of CH_4 without greatly affecting the total CH_4. Consequently, we reduce the OH concentrations in the SimTom and SimMBL simulations by 2% and 4%, respectively, relative to the SimStd OH to offset the effect of increasing Cl. In addition, the SimTom and SimMBL simulations use α_{OH}=0.9961 (Saueressig et al., 2001) rather than α_{OH}=0.9946 (Cantrell et al., 1990) to avoid too much fractionation from the combined Cl and OH sinks. We conduct an additional sensitivity simulation, SimOHp, that uses the same Cl field as SimStd but does not alter the hemispheric ratio of OH. Table 2 summarizes the standard and sensitivity simulations.

The four Cl distributions differ in their vertical and horizontal spatial distributions as well as their tropospheric mean (Figs. 3 and 4). The SimStd Cl is largest in the tropics, nearly symmetric between hemispheres, and increases with altitude. Both SimGC and SimTom have Cl that is larger in the Northern Hemisphere than the Southern Hemisphere in the annual mean and reaches a minimum in the mid-troposphere. However, the maximum in lower tropospheric Cl occurs in the tropics in SimGC but in the extratropics in SimTom. This mid-latitude Cl maximum arises because SimTom has high Cl values over east Asia, whereas SimGC Cl is highest over ocean regions (Fig. 4). SimMBL has a strong maximum in the MBL compared to the free troposphere and land regions. Its annual mean Cl concentrations are higher in the Southern Hemisphere (Fig. 3) due to the larger ocean area in the Southern Hemisphere. However, SimMBL includes a strong seasonal shift in peak Cl between the hemispheres.

2.3 Observations

We use surface observations from the NOAA GMD Carbon Cycle Cooperative Global Air Sampling Network to evaluate our simulations. We use the monthly mean observations of total CH_4 (Dlugokencky et al, 2018) and $\delta^{13}C$ of CH_4 (White et al., 2018) to compare to the monthly mean simulation results. The isotopic measurements were made at the Institute of Arctic and Alpine Research at the University of Colorado and are referenced to the VPDB scale (Zhang and Li, 1990). The analytical uncertainty of the isotopic measurements is 0.06‰. The variability between measurements taken in a given month may, however, be larger, so we use the maximum of analytical uncertainty and the within-month standard deviation as the uncertainty in the monthly mean.

3. Results and Discussion

3.1 Evaluation of Simulated CH_4

We find good agreement between the SimStd simulation and the GMD observations for CH_4 (Fig. 5) for 2004. The latitudinal distribution is well-reproduced, and the simulation captures the elevated concentrations of CH_4 observed over Europe in January as well as the January versus July differences in concentration. Overall, the spatial

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correlation between SimStd and the observations is 0.93 in January and 0.85 in July. The sensitivity simulations described in Table 2 have little effect on the CH_4 distribution, as shown by the overlapping symbols in Fig. 5c,d.

3.2 Sensitivity of $\delta^{13}C$ to Cl

We next examine the distribution of $\delta^{13}C$ in SimStd compared to observations. Fig. 6a,b shows both meridional and zonal variability in $\delta^{13}C$. Background values are less negative (heavier) in the Southern versus Northern Hemisphere (NH) (Fig. 7), a feature seen more strongly in the observations, but there is also variability due to the different source signatures. Areas of biomass burning, such as Tropical Africa, show up as particularly heavy, while regions with large wetland and rice emissions, such as SE Asia, are particularly light. Another prominent feature is the isotopically heavy region in northern Asia in January, which we attribute to the influence of the geologic (including oil, gas, and coal) source in this region (Supp. Fig. S1). This signal is less evident in July, when greater influence from boreal wetlands lightens the isotopic mix. The spatial correlation (r^2) between the SimStd and observed $\delta^{13}C$ is 0.61 in January and 0.75 in July.

The sensitivity simulations with altered oxidant concentrations alter the global values of $\delta^{13}C$, but the geographic patterns remain similar to that of SimStd. The larger Cl sink in SimGC leads to an overall less negative $\delta^{13}C$, which agrees better than SimStd with observations at Southern Hemisphere (SH) sites but worse in the NH (Figs. 6c,d and 7). The isotopic effect of the larger Cl sink in SimTom is compensated by the lower OH and α_{OH} values used in that simulation, flattening the interhemispheric gradient (Figs. 6e,f and 7). In contrast, the very large MBL Cl concentrations in SimMBL lead to an overestimate (insufficiently negative) of the observed $\delta^{13}C$ (Fig. g,h), but strengthens the interhemispheric gradient. We note that since all simulations began with the same initial conditions but have different sinks, the isotopic composition is not in steady state in 2004 and the results of the sensitivity simulations diverge further with additional years of simulation, with SimMBL becoming clearly inconsistent with observations. We note that while these results highlight the differences in $\delta^{13}C$ imposed by changing Cl, the absolute values of $\delta^{13}C$, and hence their agreement with observations, would be different for CH_4 source mixtures with a different average $\delta^{13}C$.

Figure 7 reveals an underestimate in the interhemispheric gradient of $\delta^{13}C$ in both SimStd and the sensitivity runs compared to the GMD observations. Table 3 presents the observed and simulated $\delta^{13}C$ interhemispheric gradients calculated as the difference between the $\delta^{13}C$ values averaged over all sites south of 30°S and the average over sites north of 30°N. SimStd and SimGC show similar underestimates of the observed gradient, and the underestimate is more severe in SimTom. The gradient is improved in SimMBL in January. The differences between simulations reflect differences in the locations where CH_4 oxidation occurs. Fig. 8 shows that the higher Cl values over the NH, particularly China, in SimTom versus SimStd leads to more CH_4 loss occurring in the NH and higher (heavier) $\delta^{13}C$ in the NH. This effect is particularly pronounced over China and Europe. Less fractionation by the OH sink in SimTom leads to lighter values in the SH. Conversely, SimMBL has more loss occurring over the SH oceans in January, leading to heavier $\delta^{13}C$ in the SH (Fig. 9). This effect is not present in July, when the SimMBL Cl loss shifts

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to the NH (Fig. S2). The reduced hemispheric difference in OH in SimOHp leads to a small improvement in the hemispheric gradient in $\delta^{13}C$.

We further examine the seasonal cycle of $\delta^{13}C$ in Fig. 10. We focus on the seasonal cycle at the South Pole Observatory (SPO) site because it is far from large CH_4 sources and thus the seasonal cycle depends strongly on the seasonality of the CH_4 sinks. While all simulations lie mostly within the error bars of the observations, SimMBL has the largest seasonal cycle amplitude, overestimating the seasonal cycle at of the SPO observations with a $\delta^{13}C$ value that is both too heavy in Feb.-June and too light in Aug.-Nov. In contrast, SimStd and the other sensitivity simulations underestimate the magnitude of the observed seasonal cycle at SPO. Supplemental Fig. S3 shows a large enhancement in the seasonal cycle amplitude between SimMBL and the other simulations for the Cape Grim site in Tasmania (CGO), but only a small change at other sites. This suggests that while MBL Cl is attractive as an explanation for the SH seasonality of $\delta^{13}C$, this explanation may be inconsistent with the inclusion of non-marine Cl sources.

3.3 Sensitivity of $\delta^{13}C$ to the Isotopic Distribution of Sources

Other factors in addition to the Cl distribution likely contribute to the mismatch between the observed and simulated interhemispheric gradients. Fig. 6 shows the impact of the geologic source on the $\delta^{13}C$ values over northern Asia. A bias in either the strength or the isotopic composition of this source will impact the interhemispheric gradient. Another likely contributing factor is our use of a globally uniform isotopic ratio for each source type. Ganesan et al. (2018) developed a global map of the isotopic signatures of wetland emissions. We conduct a sensitivity study, SimWet, that parallels SimStd but uses these spatially varying isotopic ratios for the wetland emissions. SimWet increases the amplitude of the seasonal cycle in $\delta^{13}C$-CH_4 particularly for northern latitudes sites such as ALT, BRW, and MHD (Supplemental Fig. S3). It has little effect on the seasonal cycle at the SH CGO and SPO sites, where SimMBL shows a large effect on the cycle. SimWet results in improved agreement with the observed interhemispheric gradient (Figs. 6,7; Table 3). Including spatially-varying isotopic signature for other sources as well could further modify the simulated interhemispheric gradient, potentially correcting some of the flat gradient of e.g. the SimTom simulation.

4. Conclusions

The role of Cl as a methane sink is a significant uncertainty in the global CH_4 budget, particularly with respect to isotopes. The global distribution of Cl is not well known from observations, and the Cl distributions simulated by global models varies widely from model to model. We investigated the sensitivity of the surface $\delta^{13}C$ distribution of CH_4 using a series of sensitivity studies with a global 3D model. Given the uncertainties in CH_4 sources and their isotopic ratios, it is not possible to conclude from this study which Cl field is best. However, the differences between the simulations provides insight on the strong lever that tropospheric Cl exerts on the $\delta^{13}C$ distribution.

Our standard and sensitivity simulations all reproduce well the geographic distribution of and temporal evolution of CH_4 observed at the GMD surface sites. However, imposing Cl distributions from a range of chemical transport

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models used in the scientific community leads to large differences in the simulated distribution of the $\delta^{13}C$ of CH_4. The CH_4 sinks from Cl in our SimStd and SimGC simulations are both below 1% of the total CH_4 sink, as suggested by Gromov et al. (2018). Yet the SimStd and SimGC simulations underestimate and overestimate, respectively, the observed $\delta^{13}C$ in 2004, despite the fact that both include only a relatively small CH_4 sink from Cl.

Our ability to reproduce the observed latitudinal distribution of $\delta^{13}C$ depends not only on the assumed value of global mean Cl, but also its geographic distribution. The detailed halogen chemistry model (TOMCAT) of Hossaini et al. (2016) places the maximum Cl values in the continental NH, in contrast to the large MBL Cl sink used in Allan et al. (2007) to explain SH observations. We find that the NH Cl maximum acts to flatten the interhemispheric gradient of $\delta^{13}C$, while the MBL Cl sink increases the hemispheric differences in NH winter and also strengthens the seasonal cycle. However, the interhemispheric gradient is also influenced by spatial variation in the isotopic signatures of the sources, complicating this issue.

Two values for the fractionating effect of OH (α_{OH}) on $\delta^{13}C$ (Cantrell et al., 1990; Saueressig et al., 2001) are widely cited in the literature. Combining the TOMCAT Cl fields with the α_{OH} of Saueressig et al. (2001) leads to an underestimate of observed $\delta^{13}C$, but combining it with the Cantrell et al. (1990) α_{OH} would lead to an overestimate. Reducing uncertainty in the fractionating effect of OH would thus improve our ability to constrain the role of Cl.

Observations of the $\delta^{13}C$ of CH_4 provide an important tool for constraining the CH_4 budget. We find that the range of Cl fields available from current global models leads to a wide range of simulated $\delta^{13}C$ values. The choice of Cl field thus strongly impacts what CH_4 source mixture best fits $\delta^{13}C$ observations. Better quantification of the role of Cl in the methane budget is therefore critical for interpreting the $\delta^{13}C$ observations to their fullest potential.

Data Availability

The methane and $\delta^{13}CH_4$ observations are available from the NOAA GMD website: https://www.esrl.noaa.gov/gmd/dv/data/. Output from the GEOS model is on the NASA Center for Climate Simulation (NCCS) system.

Author Contributions

SS designed and conducted simulation, performed analysis, and prepared the manuscript. JSW contributed to model development and experiment design. MM contributed to model development. BD contributed to model development and conceptualization. RH and CK contributed inputs to the simulations. SM and JWCW contributed data and aided in its interpretation. All authors contributed to the editing and revising of the manuscript.

The authors declare no competing interests.

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References

Allan, W., Manning, M. R., Lassey, K. R., Lowe, D. C., and Gomez, A. J.: Modeling the variation of δ13C in atmospheric methane: Phase ellipses and the kinetic isotope effect, Global Biogeochemical Cycles, 15, 467-481, doi:10.1029/2000GB001282, 2001.

Allan, W., Struthers, H., and Lowe, D.: Methane carbon isotope effects caused by atomic chlorine in the marine boundary layer: Global model results compared with Southern Hemisphere measurements, Journal of Geophysical Research: Atmospheres, 112, 2007.

Bock, M., Schmitt, J., Beck, J., Seth, B., Chappellaz, J., and Fischer, H.: Glacial/interglacial wetland, biomass burning, and geologic methane emissions constrained by dual stable isotopic CH_4 ice core records, Proceedings of the National Academy of Sciences, 114, E5778-E5786, 10.1073/pnas.1613883114, 2017.

Brownlow, R., Lowry, D., Fisher, R., France, J., Lanoisellé, M., White, B., Wooster, M., Zhang, T., and Nisbet, E.: Isotopic ratios of tropical methane emissions by atmospheric measurement, Global Biogeochemical Cycles, 31, 1408-1419, 2017.

Cantrell, C. A., Shetter, R. E., McDaniel, A. H., Calvert, J. G., Davidson, J. A., Lowe, D. C., Tyler, S. C., Cicerone, R. J., and Greenberg, J. P.: Carbon kinetic isotope effect in the oxidation of methane by the hydroxyl radical, Journal of Geophysical Research: Atmospheres, 95, 22455-22462, 1990.

Craig, H.: Isotopic Standards for Carbon and Oxygen and Correction Factors for Mass-Spectrometric Analysis OF Carbon Dioxide, Geochimica Et Cosmochimica Acta, 12, 133-149, 10.1016/0016-7037(57)90024-8, 1957.

Dlugokencky, E., Bruhwiler, L., White, J., Emmons, L., Novelli, P., Montzka, S., Masarie, K., Lang, P., Crotwell, A., Miller, J., and Gatti, L.: Observational constraints on recent increases in the atmospheric CH4 burden, Geophysical Research Letters, 36, 10.1029/2009GL039780, 2009.

Dlugokencky, E., Nisbet, E., Fisher, R., and Lowry, D.: Global atmospheric methane: budget, changes and dangers, Philosophical Transactions of the Royal Society a-Mathematical Physical and Engineering Sciences, 369, 2058-2072, 10.1098/rsta.2010.0341, 2011.

Dlugokencky, E.J., P.M. Lang, A.M. Crotwell, J.W. Mund, M.J. Crotwell, and K.W. Thoning, Atmospheric Methane Dry Air Mole Fractions from the NOAA ESRL Carbon Cycle Cooperative Global Air Sampling Network, 1983-2017, Version: 2018-08-01, Path: ftp://aftp.cmdl.noaa.gov/data/trace_gases/ch4/flask/surface/, 2018.

Duncan, B. N., Strahan, S. E., Yoshida, Y., Steenrod, S. D., and Livesey, N.: Model study of the cross-tropopause transport of biomass burning pollution, Atmospheric Chemistry and Physics, 7, 3713-3736, 2007.

Elshorbany, Y. F., Duncan, B. N., Strode, S. A., Wang, J. S., and Kouatchou, J.: The description and validation of the computationally Efficient CH4-CO-OH (ECCOHv1.01) chemistry module for 3-D model applications, Geoscientific Model Development, 9, 799-822, 10.5194/gmd-9-799-2016, 2016.

Etiope, G., and Milkov, A.: A new estimate of global methane flux from onshore and shallow submarine mud volcanoes to the atmosphere, Environmental Geology, 46, 997-1002, 10.1007/s00254-004-1085-1, 2004.

European Commission, Joint Research Center (JRC)/Netherlands Environmental Assessment Agency (PBL). Emission Database for Global Atmospheric Research (EDGAR), http://edgar.jrc.ec.europe.eu.

https://doi.org/10.5194/acp-2019-1047
Preprint. Discussion started: 22 November 2019

331 Ferretti, D. F., Miller, J. B., White, J. W. C., Etheridge, D. M., Lassey, K. R., Lowe, D. C., Meure, C. M. M., Dreier,
332 M. F., Trudinger, C. M., van Ommen, T. D., and Langenfelds, R. L.: Unexpected Changes to the Global Methane
333 Budget over the Past 2000 Years, Science, 309, 1714-1717, 10.1126/science.1115193, 2005.
334 Fischer, H., Behrens, M., Bock, M., Richter, U., Schmitt, J., Loulergue, L., Chappellaz, J., Spahni, R., Blunier, T.,
335 Leuenberger, M., and Stocker, T. F.: Changing boreal methane sources and constant biomass burning during the last
336 termination, Nature, 452, 864, 10.1038/nature06825
337 https://www.nature.com/articles/nature06825#supplementary-information, 2008.
338 Fung, I., John, J., Lerner, J., Matthews, E., Prather, M., Steele, L., and Fraser, P.: 3-Dimensional Model Synthesis of
339 the Global Methane Cycle, Journal of Geophysical Research-Atmospheres, 96, 13033-13065, 10.1029/91JD01247,
340 1991.
341 Ganesan, A., Stell, A., Gedney, N., Comyn-Platt, E., Hayman, G., Rigby, M., Poulter, B., and Hornibrook, E.:
342 Spatially Resolved Isotopic Source Signatures of Wetland Methane Emissions, Geophysical Research Letters, 45,
343 3737-3745, 2018.
344 Gelaro, R., McCarty, W., Suarez, M., Todling, R., Molod, A., Takacs, L., Randles, C., Darmenov, A., Bosilovich,
345 M., Reichle, R., Wargan, K., Coy, L., Cullather, R., Draper, C., Akella, S., Buchard, V., Conaty, A., da Silva, A.,
346 Gu, W., Kim, G., Koster, R., Lucchesi, R., Merkova, D., Nielsen, J., Partyka, G., Pawson, S., Putman, W.,
347 Rienecker, M., Schubert, S., Sienkiewicz, M., and Zhao, B.: The Modern-Era Retrospective Analysis for Research
348 and Applications, Version 2 (MERRA-2), Journal of Climate, 30, 5419-5454, 10.1175/JCLI-D-16-0758.1, 2017.
349 Ghosh, A., Patra, P., Ishijima, K., Umezawa, T., Ito, A., Etheridge, D., Sugawara, S., Kawamura, K., Miller, J., and
350 Dlugokencky, E.: Variations in global methane sources and sinks during 1910–2010, Atmospheric Chemistry and
351 Physics, 15, 2595-2612, 2015.
352 Granier, C., Bessagnet, B., Bond, T., D'Angiola, A., van der Gon, H. D., Frost, G. J., Heil, A., Kaiser, J. W., Kinne,
353 S., Klimont, Z., Kloster, S., Lamarque, J. F., Liousse, C., Masui, T., Meleux, F., Mieville, A., Ohara, T., Raut, J. C.,
354 Riahi, K., Schultz, M. G., Smith, S. J., Thompson, A., van Aardenne, J., van der Werf, G. R., and van Vuuren, D. P.:
355 Evolution of anthropogenic and biomass burning emissions of air pollutants at global and regional scales during the
356 1980-2010 period, Climatic Change, 109, 163-190, 10.1007/s10584-011-0154-1, 2011.
357 Gromov, S., Brenninkmeijer, C. A., and Jöckel, P.: A very limited role of tropospheric chlorine as a sink of the
358 greenhouse gas methane, Atmospheric Chemistry and Physics, 18, 9831-9843, 2018.
359 Hausmann, P., Sussmann, R., and Smale, D.: Contribution of oil and natural gas production to renewed increase in
360 atmospheric methane (2007–2014): top–down estimate from ethane and methane column observations, Atmospheric
361 Chemistry and Physics, 16, 3227-3244, 2016.
362 Hopcroft, P. O., Valdes, P. J., and Kaplan, J. O.: Bayesian Analysis of the Glacial-Interglacial Methane Increase
363 Constrained by Stable Isotopes and Earth System Modeling, Geophysical Research Letters, 45, 3653-3663,
364 doi:10.1002/2018GL077382, 2018.
365 Hossaini, R., Chipperfield, M. P., Saiz-Lopez, A., Fernandez, R., Monks, S., Feng, W., Brauer, P., and von Glasow,
366 R.: A global model of tropospheric chlorine chemistry: Organic versus inorganic sources and impact on methane
367 oxidation, Journal of Geophysical Research: Atmospheres, 121, 14,271-214,297, doi:10.1002/2016JD025756, 2016.

https://doi.org/10.5194/acp-2019-1047
Preprint. Discussion started: 22 November 2019

Atmospheric Chemistry and Physics Discussions

368 Houweling, S., Kaminski, T., Dentener, F., Lelieveld, J., and Heimann, M.: Inverse modeling of methane sources
369 and sinks using the adjoint of a global transport model, Journal of Geophysical Research-Atmospheres, 104, 26137-
370 26160, 10.1029/1999JD900428, 1999.
371 Houweling, S., Dentener, F., and Lelieveld, J.: Simulation of preindustrial atmospheric methane to constrain the
372 global source strength of natural wetlands, Journal of Geophysical Research-Atmospheres, 105, 17243-17255,
373 10.1029/2000JD900193, 2000.
374 Houweling, S., Van der Werf, G., Goldewijk, K. K., Röckmann, T., and Aben, I.: Early anthropogenic CH4
375 emissions and the variation of CH4 and 13CH4 over the last millennium, Global Biogeochemical Cycles, 22, 2008.
376 Hu, L., Keller, C., Long, M., Sherwen, T., Auer, B., Da Silva, A., Nielsen, J., Pawson, S., Thompson, M., Trayanov,
377 A., Travis, K., Grange, S., Evans, M., and Jacob, D.: Global simulation of tropospheric chemistry at 12.5 km
378 resolution: performance and evaluation of the GEOS-Chem chemical module (v10-1) within the NASA GEOS Earth
379 system model (GEOS-5 ESM), Geoscientific Model Development, 11, 4603-4620, 10.5194/gmd-11-4603-2018,
380 2018.
381 Ito, A., and Inatomi, M.: Use of a process-based model for assessing the methane budgets of global terrestrial
382 ecosystems and evaluation of uncertainty, Biogeosciences, 9, 759-773, 10.5194/bg-9-759-2012, 2012.
383 Kai, F., Tyler, S., Randerson, J., and Blake, D.: Reduced methane growth rate explained by decreased Northern
384 Hemisphere microbial sources, Nature, 476, 194-197, 10.1038/nature10259, 2011.
385 Kirschke, S., Bousquet, P., Ciais, P., Saunois, M., Canadell, J. G., Dlugokencky, E. J., Bergamaschi, P., Bergmann,
386 D., Blake, D. R., Bruhwiler, L., Cameron-Smith, P., Castaldi, S., Chevallier, F., Feng, L., Fraser, A., Heimann, M.,
387 Hodson, E. L., Houweling, S., Josse, B., Fraser, P. J., Krummel, P. B., Lamarque, J.-F., Langenfelds, R. L., Le
388 Quere, C., Naik, V., O'Doherty, S., Palmer, P. I., Pison, I., Plummer, D., Poulter, B., Prinn, R. G., Rigby, M.,
389 Ringeval, B., Santini, M., Schmidt, M., Shindell, D. T., Simpson, I. J., Spahni, R., Steele, L. P., Strode, S. A., Sudo,
390 K., Szopa, S., van der Werf, G. R., Voulgarakis, A., van Weele, M., Weiss, R. F., Williams, J. E., and Zeng, G.:
391 Three decades of global methane sources and sinks, Nature Geoscience, 6, 813-823, 10.1038/NGEO1955, 2013.
392 Lassey, K. R., Etheridge, D. M., Lowe, D. C., Smith, A. M., and Ferretti, D. F.: Centennial evolution of the
393 atmospheric methane budget: what do the carbon isotopes tell us?, Atmos. Chem. Phys., 7, 2119-2139, 10.5194/acp-
394 7-2119-2007, 2007.
395 Long, M., Yantosca, R., Nielsen, J., Keller, C., da Silva, A., Sulprizio, M., Pawson, S., and Jacob, D.: Development
396 of a grid-independent GEOS-Chem chemical transport model (v9-02) as an atmospheric chemistry module for Earth
397 system models, Geoscientific Model Development, 8, 595-602, 10.5194/gmd-8-595-2015, 2015.
398 McNorton, J., Chipperfield, M., Gloor, M., Wilson, C., Feng, W., Hayman, G., Rigby, M., Krummel, P., O'Doherty,
399 S., Prinn, R., Weiss, R., Young, D., Dlugokencky, E., and Montzka, S.: Role of OH variability in the stalling of the
400 global atmospheric CH4 growth rate from 1999 to 2006, Atmospheric Chemistry and Physics, 16, 7943-7956,
401 10.5194/acp-16-7943-2016, 2016.
402 Mikaloff Fletcher, S. E., Tans, P. P., Bruhwiler, L. M., Miller, J. B., and Heimann, M.: CH4 sources estimated from
403 atmospheric observations of CH4 and its 13C/12C isotopic ratios: 2. Inverse modeling of CH4 fluxes from
404 geographical regions, Global Biogeochemical Cycles, 18, doi:10.1029/2004GB002224, 2004a.

https://doi.org/10.5194/acp-2019-1047
Preprint. Discussion started: 22 November 2019

Atmospheric Chemistry and Physics Discussions

405 Mikaloff Fletcher, S. E., Tans, P. P., Bruhwiler, L. M., Miller, J. B., and Heimann, M.: CH4 sources estimated from
406 atmospheric observations of CH4 and its 13C/12C isotopic ratios: 1. Inverse modeling of source processes, Global
407 Biogeochemical Cycles, 18, doi:10.1029/2004GB002223, 2004b.
408 Molod, A., Takacs, L., Suarez, M., and Bacmeister, J.: Development of the GEOS-5 atmospheric general circulation
409 model: evolution from MERRA to MERRA2, Geoscientific Model Development, 8, 1339-1356, 10.5194/gmd-8-
410 1339-2015, 2015.
411 Monteil, G., Houweling, S., Dlugockenky, E., Maenhout, G., Vaughn, B., White, J., and Rockmann, T.: Interpreting
412 methane variations in the past two decades using measurements of CH 4 mixing ratio and isotopic composition,
413 Atmospheric chemistry and physics, 11, 9141-9153, 2011.
414 Naik, V., Voulgarakis, A., Fiore, A. M., Horowitz, L. W., Lamarque, J. F., Lin, M., Prather, M. J., Young, P. J.,
415 Bergmann, D., Cameron-Smith, P. J., Cionni, I., Collins, W. J., Dalsoren, S. B., Doherty, R., Eyring, V., Faluvegi,
416 G., Folberth, G. A., Josse, B., Lee, Y. H., MacKenzie, I. A., Nagashima, T., van Noije, T. P. C., Plummer, D. A.,
417 Righi, M., Rumbold, S. T., Skeie, R., Shindell, D. T., Stevenson, D. S., Strode, S., Sudo, K., Szopa, S., and Zeng,
418 G.: Preindustrial to present-day changes in tropospheric hydroxyl radical and methane lifetime from the
419 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP), Atmospheric Chemistry and
420 Physics, 13, 5277-5298, 10.5194/acp-13-5277-2013, 2013.
421 Nielsen, J., Pawson, S., Molod, A., Auer, B., da Silva, A., Douglass, A., Duncan, B., Liang, Q., Manyin, M., Oman,
422 L., Putman, W., Strahan, S., and Wargan, K.: Chemical Mechanisms and Their Applications in the Goddard Earth
423 Observing System (GEOS) Earth System Model, Journal of Advances in Modeling Earth Systems, 9, 3019-3044,
424 10.1002/2017MS001011, 2017.
425 Nisbet, E., Dlugokencky, E., Manning, M., Lowry, D., Fisher, R., France, J., Michel, S., Miller, J., White, J., and
426 Vaughn, B.: Rising atmospheric methane: 2007–2014 growth and isotopic shift, Global Biogeochemical Cycles, 30,
427 1356-1370, 2016.
428 Nisbet, E., Manning, M., Dlugokencky, E., Fisher, R., Lowry, D., Michel, S., Myhre, C., Platt, M., Allen, G.,
429 Bousquet, P., Brownlow, R., Cain, M., France, J., Hermansen, O., Hossaini, R., Jones, A., Levin, I., Manning, A.,
430 Myhre, G., Pyle, J., Vaughn, B., Warwick, N., and White, J.: Very Strong. Atmospheric Methane Growth in the 4
431 Years 2014-2017: Implications for the paris Agreement, Global Biogeochemical Cycles, 33, 318-342,
432 10.1029/2018GB006009, 2019.
433 Orbe, C., Oman, L., Strahan, S., Waugh, D., Pawson, S., Takacs, L., and Molod, A.: Large-Scale Atmospheric
434 Transport in GEOS Replay Simulations, Journal of Advances in Modeling Earth Systems, 9, 2545-2560,
435 10.1002/2017MS001053, 2017.
436 Patra, P., Houweling, S., Krol, M., Bousquet, P., Belikov, D., Bergmann, D., Bian, H., Cameron-Smith, P.,
437 Chipperfield, M., Corbin, K., Fortems-Cheiney, A., Fraser, A., Gloor, E., Hess, P., Ito, A., Kawa, S., Law, R., Loh,
438 Z., Maksyutov, S., Meng, L., Palmer, P., Prinn, R., Rigby, M., Saito, R., and Wilson, C.: TransCom model
439 simulations of CH4 and related species: linking transport, surface flux and chemical loss with CH4 variability in the
440 troposphere and lower stratosphere, Atmospheric Chemistry and Physics, 11, 12813-12837, 10.5194/acp-11-12813-
441 2011, 2011.

https://doi.org/10.5194/acp-2019-1047
Preprint. Discussion started: 22 November 2019

Atmospheric Chemistry and Physics Discussions

442 Platt, U., Allan, W., and Lowe, D.: Hemispheric average Cl atom concentration from $^{13}C/^{12}C$ ratios in atmospheric
443 methane, Atmos. Chem. Phys., 4, 2393-2399, 10.5194/acp-4-2393-2004, 2004.
444 Quay, P., King, S., Stutsman, J., Wilbur, D., Steele, L., Fung, I., Gammon, R., Brown, T., Farwell, G., Grootes, P.,
445 and Schmidt, F.: Carbon Isotopic Composition of Atmospheric CH4: Fossil and Biomass Burning Source Strenghts,
446 Global Biogeochemical Cycles, 5, 25-47, 10.1029/91GB00003, 1991.
447 Rice, A. L., Butenhoff, C. L., Teama, D. G., Röger, F. H., Khalil, M. A. K., and Rasmussen, R. A.: Atmospheric
448 methane isotopic record favors fossil sources flat in 1980s and 1990s with recent increase, Proceedings of the
449 National Academy of Sciences, 113, 10791-10796, 2016.
450 Rigby, M., Prinn, R. G., Fraser, P. J., Simmonds, P. G., Langenfelds, R. L., Huang, J., Cunnold, D. M., Steele, L. P.,
451 Krummel, P. B., Weiss, R. F., O'Doherty, S., Salameh, P. K., Wang, H. J., Harth, C. M., Mühle, J., and Porter, L.
452 W.: Renewed growth of atmospheric methane, Geophysical Research Letters, 35, doi:10.1029/2008GL036037,
453 2008.
454 Rigby, M., Montzka, S. A., Prinn, R. G., White, J. W., Young, D., O'Doherty, S., Lunt, M. F., Ganesan, A. L.,
455 Manning, A. J., and Simmonds, P. G.: Role of atmospheric oxidation in recent methane growth, Proceedings of the
456 National Academy of Sciences, 114, 5373-5377, 2017.
457 Rotman, D., Tannahill, J., Kinnison, D., Connell, P., Bergmann, D., Proctor, D., Rodriguez, J., Lin, S., Rood, R.,
458 Prather, M., Rasch, P., Considine, D., Ramaroson, R., and Kawa, S.: Global Modeling Initiative assessment model:
459 Model description, integration, and testing of the transport shell, Journal of Geophysical Research-Atmospheres,
460 106, 1669-1691, 10.1029/2000JD900463, 2001.
461 Saueressig, G., Bergamaschi, P., Crowley, J., Fischer, H., and Harris, G.: Carbon Kinetic Isotope Effect in the
462 Reaction of CH4 with Cl Atoms, Geophysical Research Letters, 22, 1225-1228, 10.1029/95GL00881, 1995.
463 Saueressig, G., Crowley, J. N., Bergamaschi, P., Brühl, C., Brenninkmeijer, C. A. M., and Fischer, H.: Carbon 13
464 and D kinetic isotope effects in the reactions of CH4 with O(1 D) and OH: New laboratory measurements and their
465 implications for the isotopic composition of stratospheric methane, Journal of Geophysical Research: Atmospheres,
466 106, 23127-23138, doi:10.1029/2000JD000120, 2001.
467 Saunois, M., Bousquet, P., Poulter, B., Peregon, A., Ciais, P., Canadell, J. G., Dlugokencky, E. J., Etiope, G.,
468 Bastviken, D., and Houweling, S.: The global methane budget 2000–2012, Earth System Science Data (Online), 8,
469 2016.
470 Saunois, M., Bousquet, P., Poulter, B., Peregon, A., Ciais, P., Canadell, J., Dlugokencky, E., Etiope, G., Bastviken,
471 D., Houweling, S., Janssens-Maenhout, G., Tubiello, F., Castaldi, S., Jackson, R., Alexe, M., Arora, V., Beerling,
472 D., Bergamaschi, P., Blake, D., Brailsford, G., Bruhwiler, L., Crevoisier, C., Crill, P., Covey, K., Frankenberg, C.,
473 Gedney, N., Hoglund-Isaksson, L., Ishizawa, M., Ito, A., Joos, F., Kim, H., Kleinen, T., Krummel, P., Lamarque, J.,
474 Langenfelds, R., Locatelli, R., Machida, T., Maksyutov, S., Melton, J., Morino, I., Naik, V., O'Doherty, S.,
475 Parmentier, F., Patra, P., Peng, C., Peng, S., Peters, G., Pison, I., Prinn, R., Ramonet, M., Riley, W., Saito, M.,
476 Santini, M., Schroeder, R., Simpson, I., Spahni, R., Takizawa, A., Thornton, B., Tian, H., Tohjima, Y., Viovy, N.,
477 Voulgarakis, A., Weiss, R., Wilton, D., Wiltshire, A., Worthy, D., Wunch, D., Xu, X., Yoshida, Y., Zhang, B.,

https://doi.org/10.5194/acp-2019-1047
Preprint. Discussion started: 22 November 2019

Atmospheric Chemistry and Physics Discussions

478 Zhang, Z., and Zhu, Q.: Variability and quasi-decadal changes in the methane budget over the period 2000-2012,
479 Atmospheric Chemistry and Physics, 17, 11135-11161, 10.5194/acp-17-11135-2017, 2017.
480 Schaefer, H., Fletcher, S. E. M., Veidt, C., Lassey, K. R., Brailsford, G. W., Bromley, T. M., Dlugokencky, E. J.,
481 Michel, S. E., Miller, J. B., Levin, I., Lowe, D. C., Martin, R. J., Vaughn, B. H., and White, J. W. C.: A 21st-century
482 shift from fossil-fuel to biogenic methane emissions indicated by 13CH4, Science, 352, 80-84,
483 10.1126/science.aad2705, 2016.
484 Schwietzke, S., Sherwood, O., Ruhwiler, L., Miller, J., Etiope, G., Dlugokencky, E., Michel, S., Arling, V., Vaughn,
485 B., White, J., and Tans, P.: Upward revision of global fossil fuel methane emissions based on isotope database,
486 Nature, 538, 88-91, 10.1038/nature19797, 2016.
487 Sherwen, T., Schmidt, J. A., Evans, M. J., Carpenter, L. J., Großmann, K., Eastham, S. D., Jacob, D. J., Dix, B.,
488 Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A., Prados-Roman, C., Mahajan, A. S., and
489 Ordóñez, C.: Global impacts of tropospheric halogens (Cl, Br, I) on oxidants and composition in GEOS-Chem,
490 Atmos. Chem. Phys., 16, 12239-12271, 10.5194/acp-16-12239-2016, 2016.
491 Sherwood, O., Schwietzke, S., Arling, V., and Etiope, G.: Global Inventory of Gas Geochemistry Data from Fossil
492 Fuel, Microbial and Burning Sources, version 2017, Earth System Science Data, 9, 639-656, 10.5194/essd-9-639-
493 2017, 2017.
494 Spivakovsky, C. M., Logan, J. A., Montzka, S. A., Balkanski, Y. J., Foreman-Fowler, M., Jones, D. B. A.,
495 Horowitz, L. W., Fusco, A. C., Brenninkmeijer, C. A. M., Prather, M. J., Wofsy, S. C., and McElroy, M. B.: Three-
496 dimensional climatological distribution of tropospheric OH: Update and evaluation, Journal of Geophysical
497 Research-Atmospheres, 105, 8931-8980, 10.1029/1999jd901006, 2000.
498 Strahan, S. E., Duncan, B. N., and Hoor, P.: Observationally derived transport diagnostics for the lowermost
499 stratosphere and their application to the GMI chemistry and transport model, Atmospheric Chemistry and Physics, 7,
500 2435-2445, 2007.
501 Strahan, S. E., Douglass, A. R., and Newman, P. A.: The contributions of chemistry and transport to low arctic
502 ozone in March 2011 derived from Aura MLS observations, Journal of Geophysical Research: Atmospheres, 118,
503 1563-1576, 2013.
504 Strode, S., Duncan, B., Yegorova, E., Kouatchou, J., Ziemke, J., and Douglass, A.: Implications of carbon monoxide
505 bias for methane lifetime and atmospheric composition in chemistry climate models, Atmospheric Chemistry and
506 Physics, 15, 11789-11805, 2015.
507 Tans, P.: A note on isotopic ratios and the global atmospheric methane budget, Global Biogeochemical Cycles, 11,
508 77-81, 10.1029/96GB03940, 1997.
509 Thompson, R., Nisbet, E., Pisso, I., Stohl, A., Blake, D., Dlugokencky, E., Helmig, D., and White, J.: Variability in
510 Atmospheric Methane From Fossil Fuel and Microbial Sources Over the Last Three Decades, Geophysical Research
511 Letters, 45, 11499-11508, 10.1029/2018GL078127, 2018.
512 Thompson, R. L., Stohl, A., Zhou, L. X., Dlugokencky, E., Fukuyama, Y., Tohjima, Y., Kim, S. Y., Lee, H., Nisbet,
513 E. G., and Fisher, R. E.: Methane emissions in East Asia for 2000–2011 estimated using an atmospheric Bayesian
514 inversion, Journal of Geophysical Research: Atmospheres, 120, 4352-4369, 2015.

https://doi.org/10.5194/acp-2019-1047
Preprint. Discussion started: 22 November 2019

Atmospheric Chemistry and Physics Discussions

515 Turner, A., Jacob, D., Benmergui, J., Wofsy, S., Maasakkers, J., Butz, A., Hasekamp, O., and Biraud, S.: A large
516 increase in US methane emissions over the past decade inferred from satellite data and surface observations,
517 Geophysical Research Letters, 43, 2218-2224, 10.1002/2016GL067987, 2016.
518 Turner, A. J., Frankenberg, C., Wennberg, P. O., and Jacob, D. J.: Ambiguity in the causes for decadal trends in
519 atmospheric methane and hydroxyl, Proceedings of the National Academy of Sciences, 114, 5367-5372, 2017.
520 Wang, J. S., McElroy, M. B., Spivakovsky, C. M., and Jones, D. B. A.: On the contribution of anthropogenic Cl to
521 the increase in δ13C of atmospheric methane, Global Biogeochemical Cycles, 16, 20-21-20-11, 2002.
522 Wang, X., Jacob, D., Eastham, S., Sulprizio, M., Zhu, L., Chen, Q., Alexander, B., Sherwen, T., Evans, M., Lee, B.,
523 Haskins, J., Lopez-Hilfiker, F., Thornton, J., Huey, G., and Liao, H.: The role of chlorine in global tropospheric
524 chemistry, Atmospheric Chemistry and Physics, 19, 3981-4003, 10.5194/acp-19-3981-2019, 2019.
525 Waugh, D., Crotwell, A., Dlugokencky, E., Dutton, G., Elkins, J., Hall, B., Hintsa, E., Hurst, D., Montzka, S.,
526 Mondeel, D., Moore, F., Nance, J., Ray, E., Steenrod, S., Strahan, S., and Sweeney, C.: Tropospheric SF6: Age of
527 air from the Northern Hemisphere midlatitude surface, Journal of Geophysical Research-Atmospheres, 118, 11429-
528 11441, 10.1002/jgrd.50848, 2013.
529 White, J.W.C., B.H. Vaughn, and S.E. Michel, University of Colorado, Institute of Arctic and Alpine Research
530 (INSTAAR), Stable Isotopic Composition of Atmospheric Methane (13C) from the NOAA ESRL Carbon Cycle
531 Cooperative Global Air Sampling Network, 1998-2017, Version: 2018-09-24, Path:
532 ftp://aftp.cmdl.noaa.gov/data/trace_gases/ch4c13/flask/, 2018.
533 Worden, J. R., Bloom, A. A., Pandey, S., Jiang, Z., Worden, H. M., Walker, T. W., Houweling, S., and Röckmann,
534 T.: Reduced biomass burning emissions reconcile conflicting estimates of the post-2006 atmospheric methane
535 budget, Nature communications, 8, 2227, 2017.
536 Zhang, Q.-L. and Li, W.-J.: A Calibrated Measurement of the Atomic Weight of Carbon, Chinese Science Bulletin
537 35, 290, doi: 10.1360/sb1990-35-4-290, 1990.
538
539

https://doi.org/10.5194/acp-2019-1047
Preprint. Discussion started: 22 November 2019

Atmospheric Chemistry and Physics Discussions

Table 1: Emission source references and $\delta^{13}C$ values

Source	Reference	IAV	$\delta^{13}C$ (‰)[a]	CH_4 Source (Tg yr^{-1})[b]
Animals (enteric fermentation)	EDGAR	Y	-62	102
C3 Biomass Burning (Forests)	MACCity	Y	-26	16
C4 Biomass Burning (Savannas)	MACCity	Y	-15	10
Coal, energy, and industry	EDGAR	Y	-35	6
Geologic (oil/gas/non-coal fuels, volcanos)	EDGAR, Transcom	Y, except volcanos	-40	124
Waste (solid and animal waste, wastewater)	EDGAR	Y	-55	74
Ocean	Transcom	N	-59	8
Rice	Visit model	Y	-63	44
Termites	Transcom	N	-57	22
Wetlands	Visit model	Y	-60	149

[a]$\delta^{13}C$ values from Dlugokencky et al., 2011;Lassey et al., 2007;Monteil et al., 2011;Houweling et al., 2000 and refs therein
[b]Values for 2004

Table 2: Oxidants for the Standard and sensitivity simulations

Simulation	$[Cl]_{Trop}$[a] (molec cm^{-3})	Cl Model[b]	Cl Reference	OH modification[c]
SimStd	210	GMI	(Strahan et al., 2007; Rotman et al., 2001; Strahan et al., 2013;Duncan et al., 2007)	$\alpha = 0.9946$
SimGC	384	GEOSChem	(Sherwen et al., 2016)	$\alpha = 0.9946$
SimTom	1710	TOMCAT	(Hossaini et al., 2016)	-2% [OH] $\alpha = 0.9961$
SimOHp	210	GMI	See SimStd	Not modified for 20% higher in NH
SimMBL	2810	Tanh function below 900hPa over ocean; GMI elsewhere	(Allan et al., 2007)	-4% [OH] $\alpha = 0.9961$

[a]Concentration of Cl averaged over the troposphere
[b]Name of the model that generated the offline Cl field
[c]Changes to [OH] or α_{OH} compared to SimStd

Table 3: Observed and Simulated Interhemispheric Gradient in $\delta^{13}C$-CH_4

	Jan. Gradient (‰)[a]	July Gradient (‰)[a]
GMD Obs	0.36	0.28
SimStd	0.17	0.11
SimGC	0.17	0.098
SimTom	0.051	0.010
SimMBL	0.30	0.13
SimOHp	0.22	0.15
SimWet	0.28	0.25

[a]Average $\delta^{13}C$-CH_4 at GMD site locations south of 30°S minus average $\delta^{13}C$-CH_4 at locations north of 30°N

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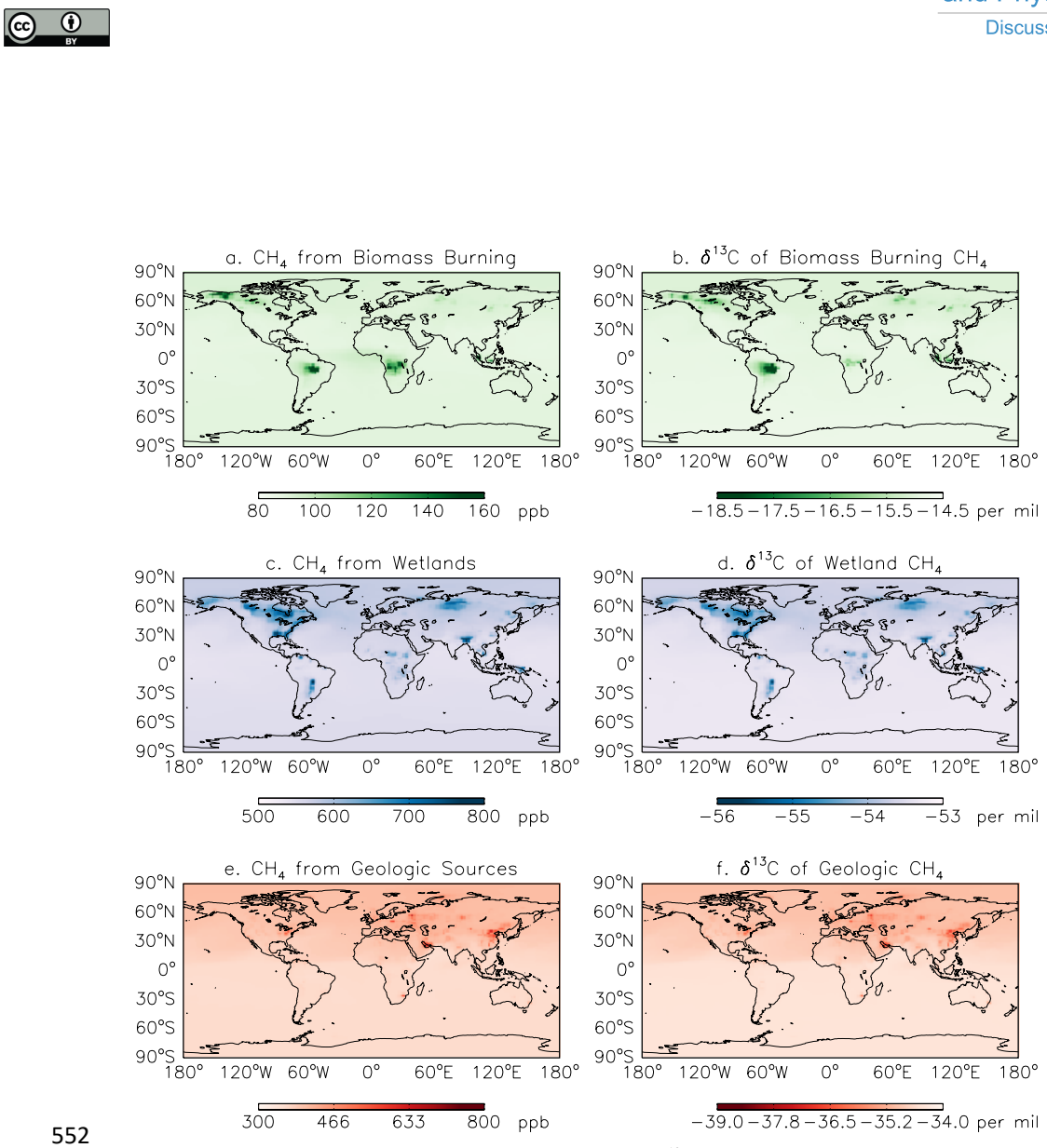

Fig. 1: The July 2004 surface concentration of CH_4 (left) and $\delta^{13}C$-CH_4 (right) from biomass burning (a,b), wetlands (c,d), and geologic+coal sources (e,f) from the GEOS tagged CH_4 tracers.

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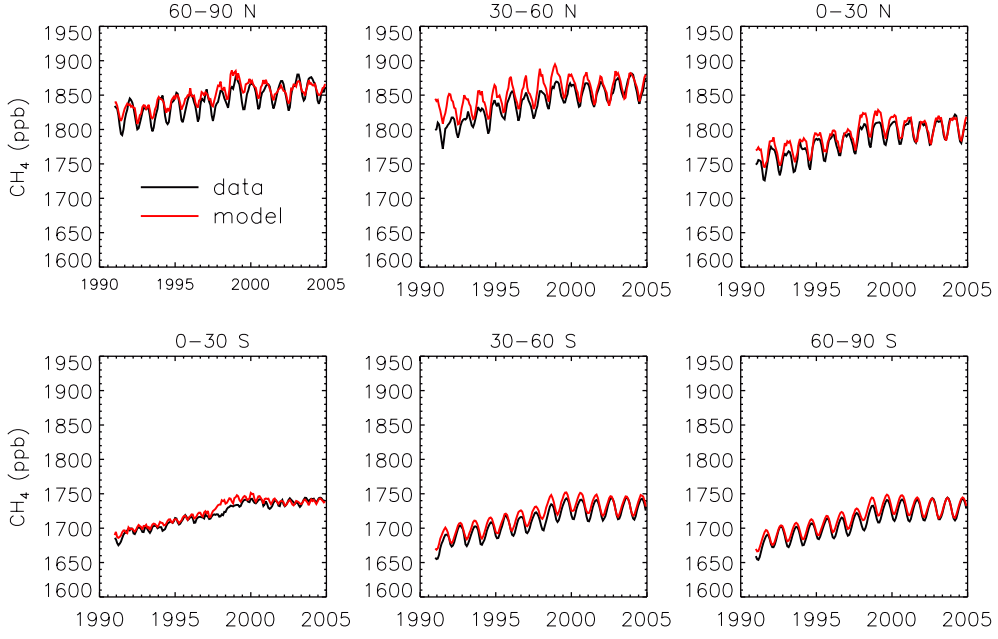

Fig 2: Monthly CH_4 observations from the GMD network (black) and simulated surface concentrations from SimStd (red) averaged over latitude bands

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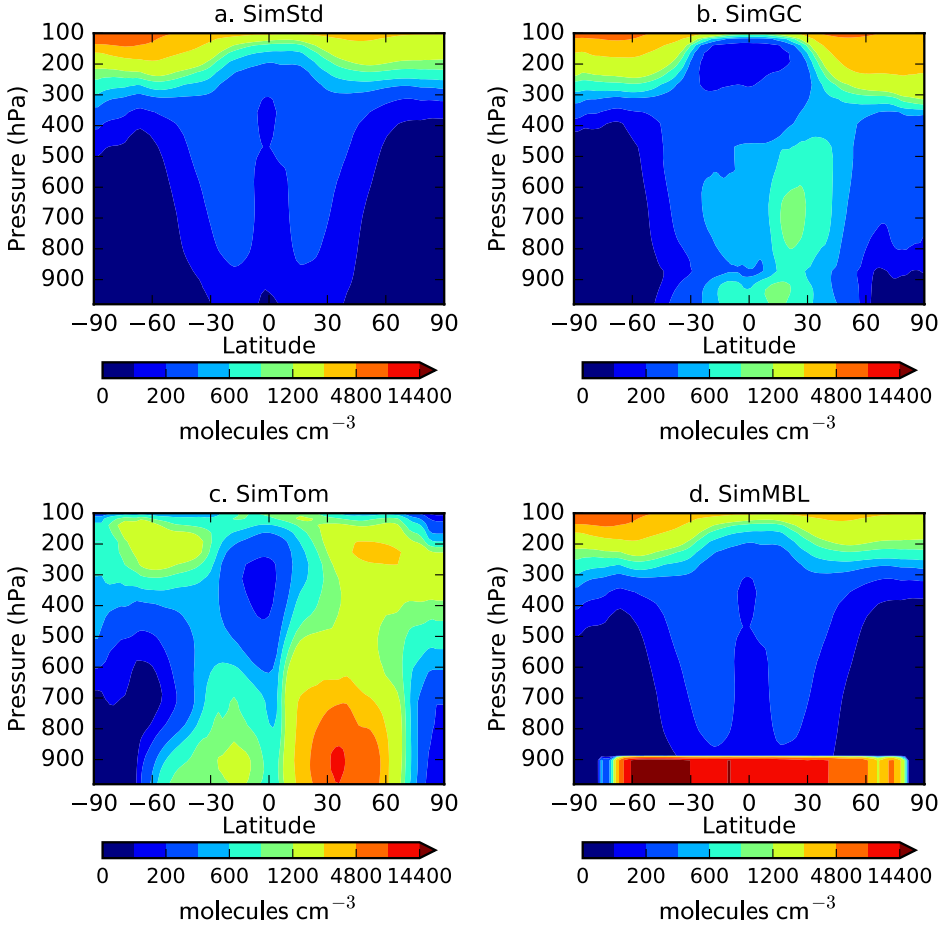

560
561 **Fig. 3**: Annual zonal mean Cl field for a) SimStd, b) SimGC, c) SimTom, and d) SimMBL.
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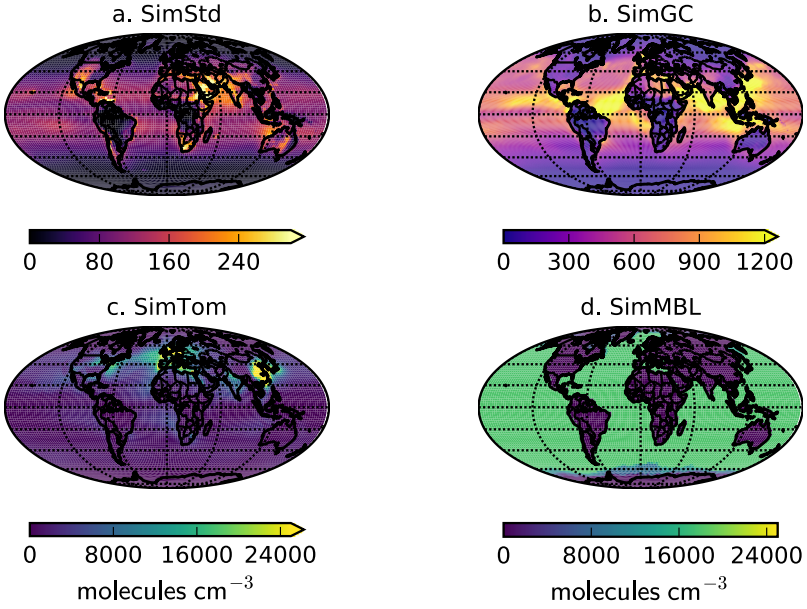

Fig. 4: Annual mean surface concentrations of Cl in a) SimStd, b) SimGC, c) SimTom, and d) SimMBL. Note the different color scales between panels.

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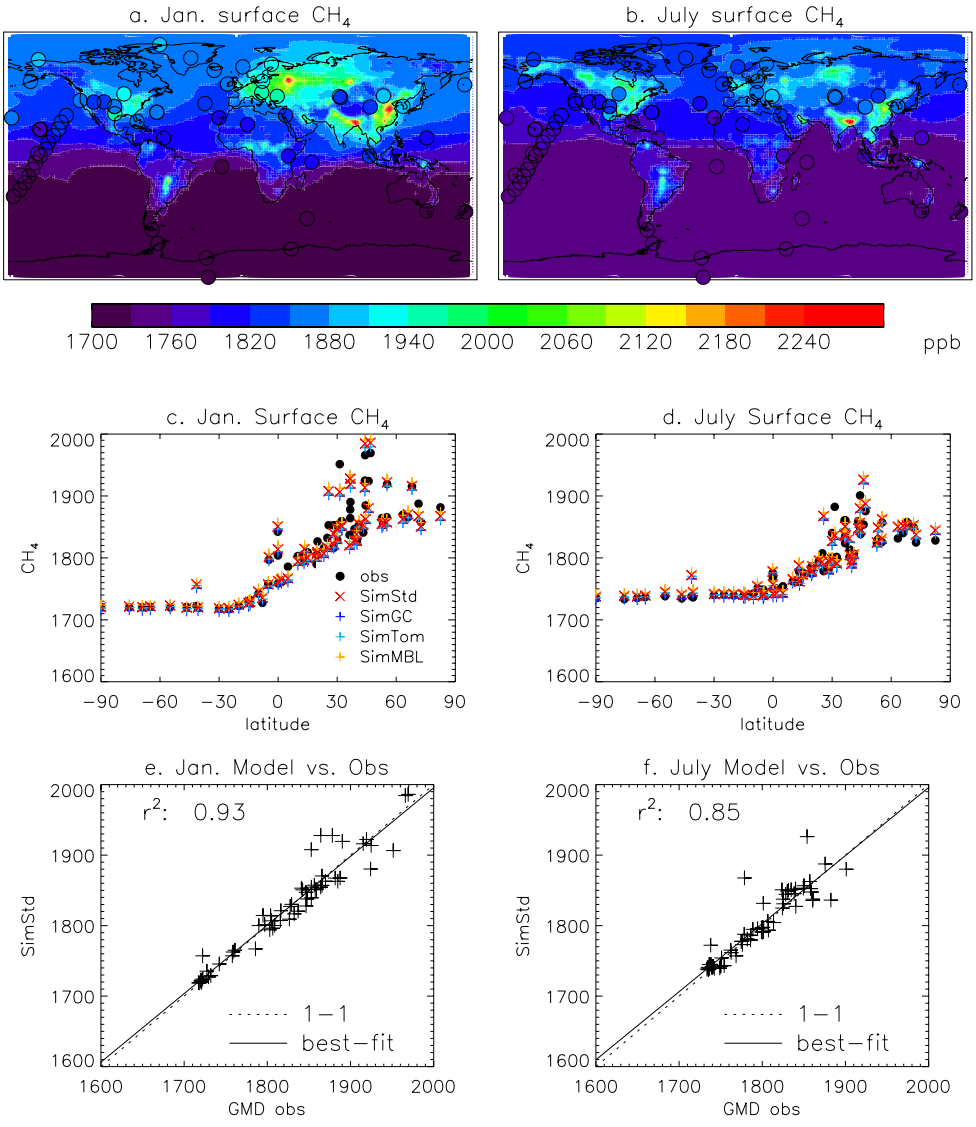

Fig. 5: Comparison of 2004 simulated and observed surface CH_4 concentrations for January (left) and July (right). a,b) Surface concentrations of CH_4 from SimStd are overplotted with the concentrations from the GMD observations in circles. c,d) GMD observations (black circles), SimStd (red x), SimGC (dark blue +), SimTom (light blue +), and SimMBL (orange +) CH_4 as a function of latitude. E,f) SimStd CH_4 (ppb) at the observation locations versus the GMD observations (+ signs) as well as the regression line (solid) and 1 to 1 line (dashed).

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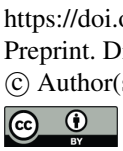

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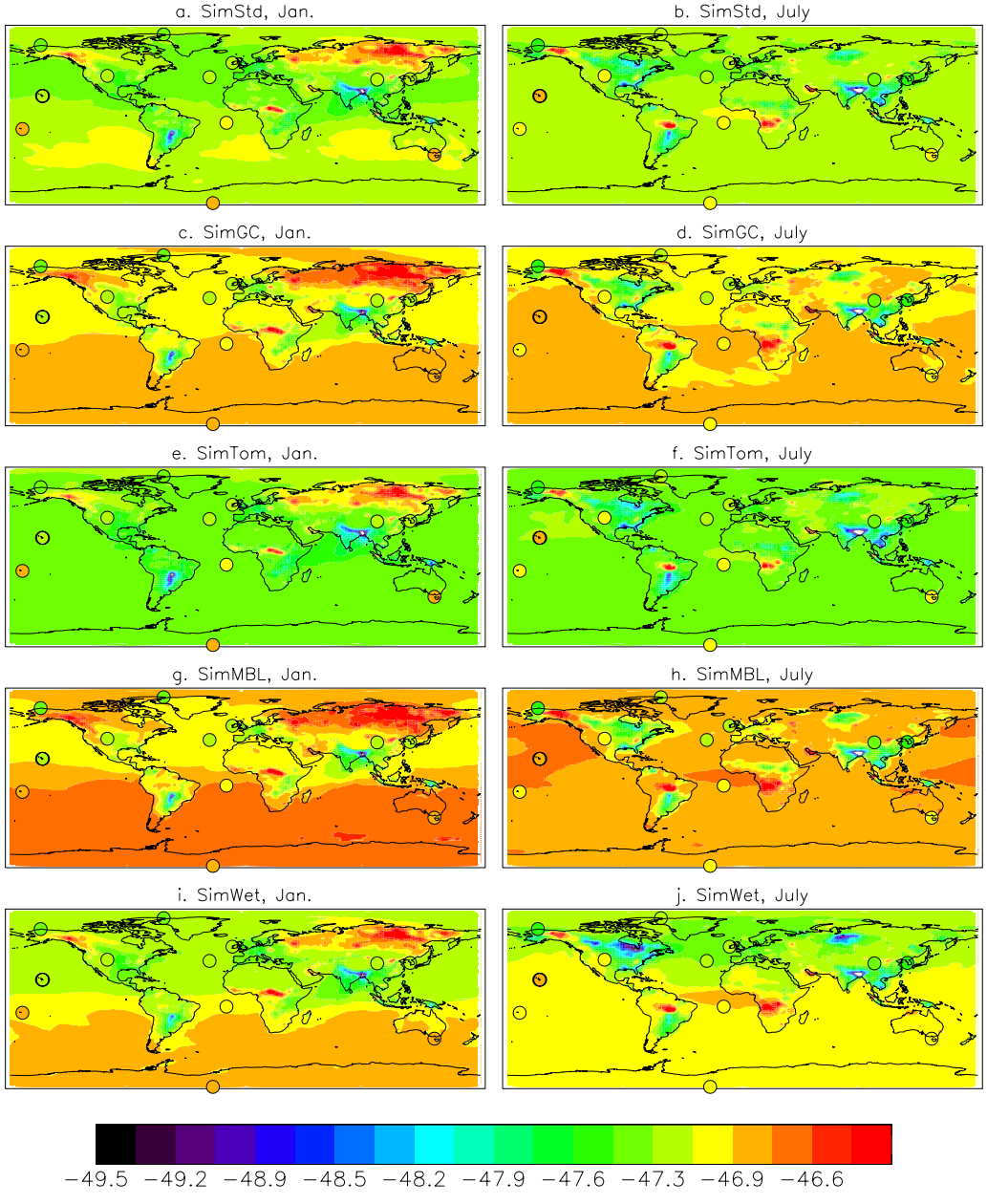

Fig. 6: Maps of the simulated surface $\delta^{13}C$ of CH_4 in per mil for Jan. (left) and July (right) overplotted with observations from the GMD sites (circles). The simulations are (a,b) SimStd, (c,d) SimGC, (e,f) SimTom, (g,h) SimMBL, and (I,j) SimWet.

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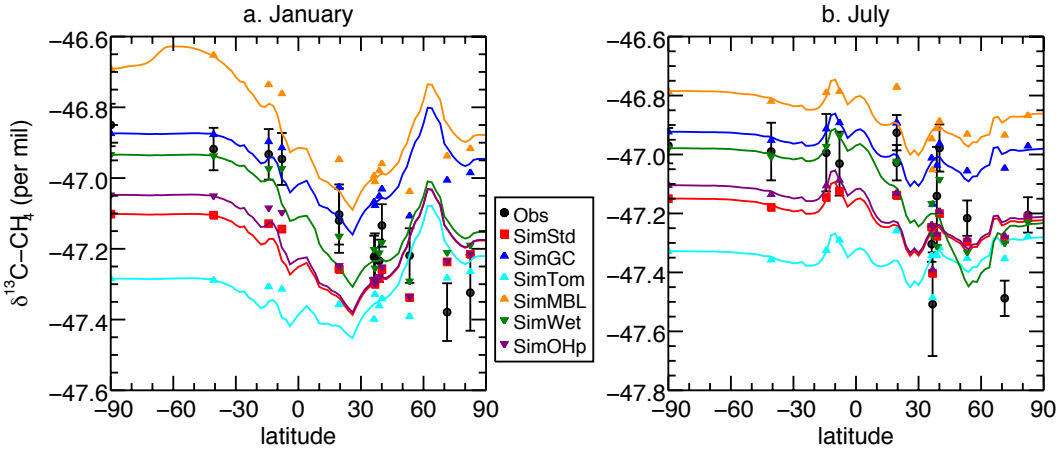

Fig. 7: $\delta^{13}C$ of CH_4 as a function of latitude in a) January and b) July 2004 for the GMD observations (Black circles), SimStd (red), SimGC (dark blue), SimTom (cyan), SimMBL (orange), SimWet (green), and SimOHp (purple). Errorbars represent the maximum of the analytical uncertainty (0.06‰) and the standard deviation of individual measurements in the month for each site. The colored lines represent the simulated zonal mean, while the colored symbols represent the simulation sampled at the location of the GMD observations.

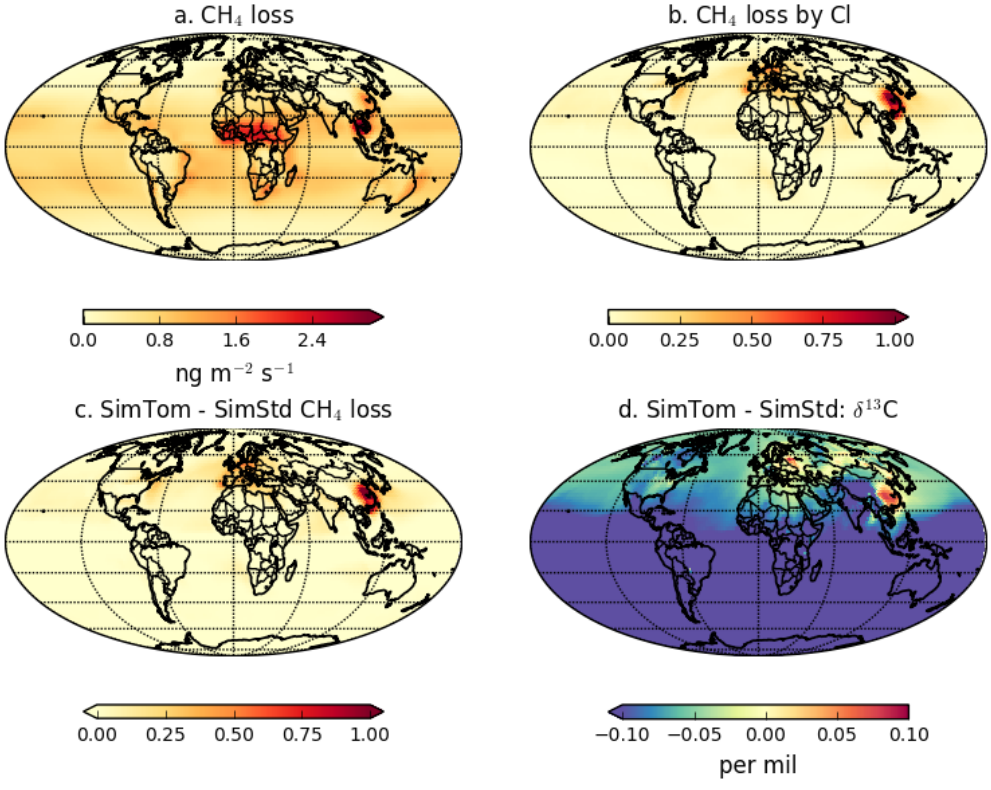

Fig 8: January a) CH_4 loss and b) CH_4 loss by Cl only in the SimTom simulation, as well as the difference in c) CH_4 loss and d) $\delta^{13}C$-CH_4 between the SimTom and SimStd simulations.

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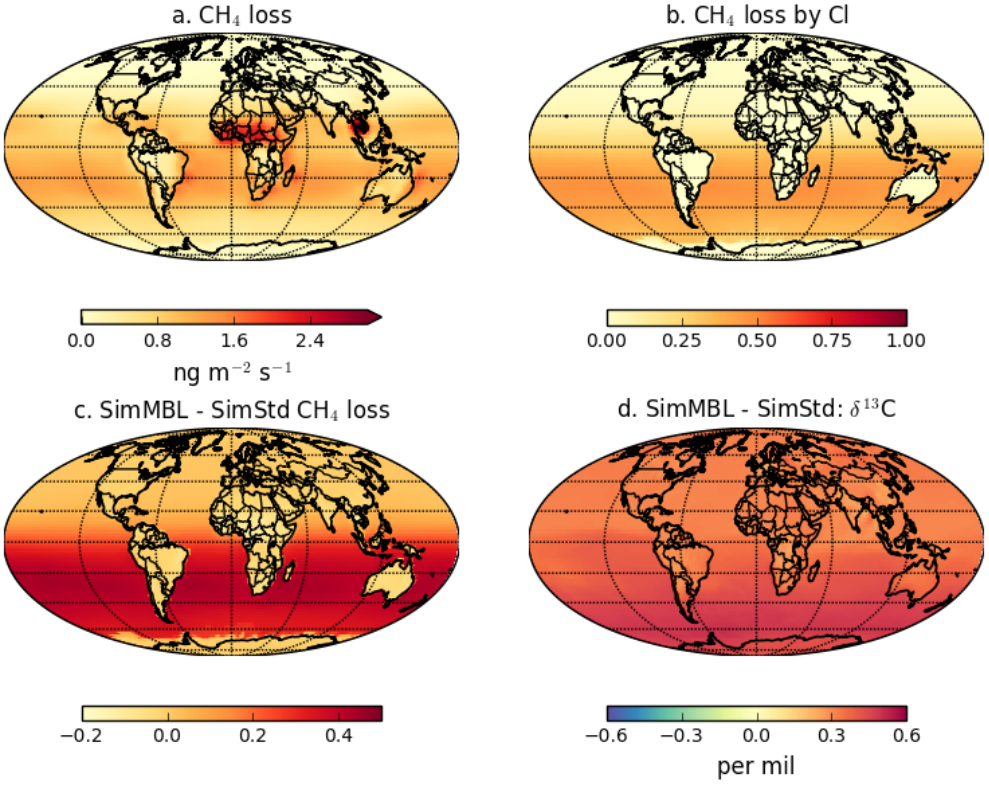

Fig 9: January a) CH_4 loss and b) CH_4 loss by Cl only in the SimMBL simulation, as well as the difference in c) CH_4 loss and d) δ^{13}C-CH_4 between the SimMBL and SimStd simulations.

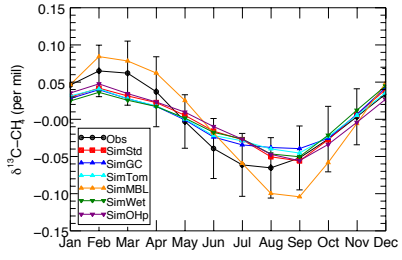

Fig. 10: The seasonal cycle of δ^{13}C of CH_4 at the SPO site with the annual mean removed averaged over 2002-2004 for the GMD observations (black), SimStd (red), SimGC (blue), SimTom (cyan), SimMBL (orange), SimWet (green), and SimOHp (purple). Errorbars represent the standard error, calculated as the maximum of the pooled standard deviation or the analytical uncertainty (0.06‰), divided by the square root of the number of years of observations.