Strong Sensitivity of the Isotopic Composition of Methane to the

Plausible Range of Tropospheric Chlorine

3

2

1

- 4 Sarah A. Strode^{1,2,*}, James S. Wang^{1,2,**}, Michael Manyin^{2,3}, Bryan Duncan², Ryan Hossaini⁴,
- 5 Christoph A. Keller^{1,2}, Sylvia E. Michel⁵, James W. C. White⁵

6

- 7 ¹Universities Space Research Association, Columbia, MD, USA
- 8 ²NASA Goddard Space Flight Center, Greenbelt, MD, USA
- 9 ³SSAI, Lanham, MD, USA
- ⁴Lancaster Environment Centre, Lancaster University, Lancaster, UK
- 11 ⁵Institute of Arctic and Alpine Research, University of Colorado, Boulder, CO, USA
- *correspondence to: sarah.a.strode@nasa.gov
- 13 **Now at the Institute for Advanced Sustainability Studies, Potsdam, Germany

14 15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

from δ^{13} C observations.

Abstract. The 13 C isotopic ratio of methane, δ^{13} C of CH₄, provides additional constraints on the CH₄ budget to complement the constraints from CH₄ observations. The interpretation of δ^{13} C observations is complicated, however, by uncertainties in the methane sink. The reaction of CH₄ with Cl is highly fractionating, increasing the relative abundance of ¹³CH₄, but there is currently no consensus on the strength of the tropospheric Cl sink. Global model simulations of halogen chemistry differ strongly from one another in terms of both the magnitude of tropospheric Cl and its geographic distribution. This study explores the impact of the inter-model diversity in Cl fields on the simulated δ^{13} C of CH₄. We use a set of GEOS global model simulations with different predicted Cl fields to test the sensitivity of the δ^{13} C of CH₄ to the diversity of Cl output from chemical transport models. We find that δ^{13} C is highly sensitive to both the amount and geographic distribution of Cl. Simulations with Cl providing 0.28% or 0.66% of the total CH₄ loss bracket the δ^{13} C observations for a fixed set of emissions. Thus, even when Cl provides only a small fraction of the total CH₄ loss and has a small impact on total CH₄, it provides a strong lever on δ^{13} C. Consequently, it is possible to achieve a good representation of total CH₄ using widely different Cl concentrations, but the partitioning of CH₄ loss between the OH and Cl reactions leads to strong differences in isotopic composition depending on which model's Cl field is used. Comparing multiple simulations, we find that altering the tropospheric Cl field leads to approximately a 0.5% increase in δ^{13} CH₄ for each percent increase in how much CH₄ is oxidized by Cl. The geographic distribution and seasonal cycle of Cl also impacts the hemispheric gradient and seasonal cycle of δ^3 C. The large effect of Cl on δ^{13} C compared to total CH₄ broadens the range of CH₄ source mixtures that can be reconciled with δ^{13} C observations. Stronger constraints on tropospheric Cl are necessary to improve estimates of CH₄ sources

1. Introduction

The global budget of methane is of great interest due to methane's role as a greenhouse gas, ozone precursor, and sink of the hydroxyl radical. Despite extensive study, major uncertainties in the methane budget remain, with top-down and bottom-up estimates often yielding different results (Kirschke et al., 2013;Saunois et al., 2016;Saunois et al., 2017, and refs therein) for the strength of specific source types. Furthermore, the resumed increase of methane concentrations beginning in 2007 (Dlugokencky et al., 2009; Rigby et al., 2008) can be explained by multiple hypotheses including an increase in fossil fuel emissions (Turner et al., 2016;Thompson et al., 2015;Hausmann et al., 2016), an increase in fossil fuel emissions combined with a decrease in biomass burning (Worden et al., 2017), an increase in biogenic sources (Schaefer et al., 2016;Nisbet et al., 2016), or a decrease in hydroxyl concentrations (Turner et al., 2017;Rigby et al., 2017). Variations in hydroxyl concentrations may also be important for the decrease in methane growth from 1999-2006 (McNorton et al., 2016).

Observations and modeling of methane's carbon isotopes provides additional information on methane sources since individual sources differ in their 13 C to 12 C ratio (δ^{13} C). Isotopic information can be used to better constrain methane sources (e.g. Thompson et al., 2015; Mikaloff Fletcher et al., 2004b, a) and infer how the source mixture changed over glacial (e.g. Hopcroft et al., 2018; Fischer et al., 2008; Bock et al., 2017), millennial (e.g. Ferretti et al., 2005; Houweling et al., 2008), and decadal timescales (e.g. Nisbet et al., 2016; Schaefer et al., 2016; Kai et al., 2011; Schwietzke et al., 2016; Thompson et al., 2018). However, there are considerable uncertainties in the processes that control methane's isotopic composition that may confound source apportionment studies. Many modeling studies use a single value for the isotopic ratio of each source, while in reality sources such as wetlands, biomass burning, and natural gas show large regional or environment-dependent variations in their isotopic signature (Ganesan et al., 2018; Brownlow et al., 2017; Dlugokencky et al., 2011; Schwietzke et al., 2016; Sherwood et al., 2017).

The isotopic composition of atmospheric methane is also sensitive to methane's sinks. Reaction with OH, the principal loss for atmospheric methane, has a kinetic isotope effect (KIE) of -5.4% (α =k₁₃/k₁₂=0.9946) to -3.9% (α =0.9961) (Saueressig et al., 2001; Cantrell et al., 1990) and contributes to the interhemispheric gradient of δ^{13} C (Quay et al., 1991). Mass balance (Lassey et al., 2007) and observations of the seasonal cycle of δ^{13} C versus methane concentration, however, suggest larger apparent KIE values, which may indicate a role for methane oxidation by chlorine (CI) in the marine boundary layer (MBL) (Allan et al., 2001;Allan et al., 2007) since Cl has a KIE of -61.9% (α =0.938) at 297K (Saueressig et al., 1995). Inclusion of the MBL Cl sink alters the source mixture inferred from inverse modeling of δ^{13} CH₄ (Rice et al., 2016). Nisbet et al. (2019) point out that interannual variability in the CH₄ Cl sink could explain some of the variability of δ^{13} C. Cl is also an important methane sink in the stratosphere, and the impact of this sink on surface δ^{13} C is a source of uncertainty in modeling δ^{13} C (Ghosh et al., 2015). Reaction with stratospheric Cl contributes approximately 0.23% to the δ^{13} C of surface methane and makes a small contribution to the observed trend in surface δ^{13} C over the last century (Wang et al., 2002).

The global concentration of Cl in the MBL and its role in the methane budget is still uncertain. Cl concentrations are highly variable and not well constrained by direct observations. Modeling work by Hossaini et al. (2016) and Sherwen et al. (2016) suggests that chlorine provides 2-2.5% of tropospheric methane oxidation. This

agrees well with estimates based on the isotopic fractionation, which also suggest Cl provides several percent of the total sink (Allan et al., 2007;Platt et al., 2004). However, Gromov et al. (2018) suggest that these are overestimates as values over 1% are inconsistent with the δ^{13} C of CO, which is a product of CH₄ oxidation. The recent modeling study of Wang et al. (2019) also suggests a value of 1%. There is thus considerable uncertainty in the role of chlorine in the budget and isotopic composition of methane.

Here, we investigate the sensitivity of δ^{13} C of CH₄ to inter-model diversity in tropospheric chlorine concentrations to better quantify how much uncertainty in the interpretation of δ^{13} C is imposed by the uncertainty in Cl. Section 2 describes the modeling framework. We present results for total CH₄ and its isotopic composition compared to surface observations in Section 3, and discuss the implications for the global CH₄ budget in Section 4.

2. Methods

2.1 Model Description

We simulate atmospheric methane with the Goddard Earth Observing System (GEOS) global earth system model (Molod et al., 2015; Nielsen et al., 2017). The model has 72 vertical levels extending from the surface to 1 Pa. We conduct simulations at C90 resolution on the cubed sphere, which corresponds to approximately 100 km horizontal resolution. The simulations' meteorology is constrained to the MERRA-2 reanalysis (Gelaro et al., 2017) using a "replay" method (Orbe et al., 2017). The GEOS replay agrees well with the tropospheric mean age of the Global Modeling Initiative (GMI) chemistry and transport model (CTM) (Orbe et al., 2017), which shows reasonable agreement with the age derived from SF₆ observations, albeit with an old bias in the southern hemisphere (Waugh et al., 2013). We thus expect the simulated interhemispheric transport time to be reasonable.

The GEOS CH₄ simulation can be interactively coupled to CO and OH (Elshorbany et al., 2016), or run independently with prescribed OH fields. We take the latter approach in this study, since this approach is able to capture many of the observed variations in atmospheric methane (Elshorbany et al., 2016). We prescribe the OH field following (Spivakovsky et al., 2000), but modify the OH to be approximately 20% higher in the Northern Hemisphere than the Southern Hemisphere, consistent with the OH field produced by many global atmospheric chemistry models (Naik et al., 2013;Strode et al., 2015). This modification is designed to make our results more applicable to understanding the impacts of inter-model differences in Cl, since it makes our OH distribution more consistent with that produced by many CCMs. The OH field varies monthly but repeats every year. We also include stratospheric losses for CH₄ from reaction with OH, Cl, and O¹D. These fields are prescribed from output of the GMI CTM (https://gmi.gsfc.nasa.gov) (Strahan et al., 2007; Duncan et al., 2007).

We implement the CH₄ isotopes in GEOS by separately simulating 13 CH₄ and 12 CH₄ tracers. We then calculate total CH₄ as the sum of the two carbon isotopologues and calculate δ^{13} C of CH₄ in per mil using the standard definition:

$$\delta^{13}\text{C-CH}_4 (\%) = (\lceil {}^{13}\text{CH}_4 \rceil / \lceil {}^{12}\text{CH}_4 \rceil / R_{\text{std}} - 1) * 1000$$
 (1)

where R_{std} =0.0112372 is the peedee belemnite isotopic standard (Craig, 1957). We partition each emission source into 12 CH₄ and 13 CH₄ emissions according to a source-specific δ^{13} C value from the literature, provided in Table 1. We use the Craig (1957) R_{std} value to partition the sources since it is cited in the literature used in Table 1 (Houweling et

al, 2000; Lassey, 2007), and so for consistency we use the same value in equation 1 to calculate the simulated δ^{13} C of the CH₄ concentrations. We note, however, that the GMD observations now use a slightly different standard, the VPDB value of 0.011183 (Zhang and Li, 1990). A sensitivity study (not shown) confirms that the choice Rstd has little effect on our results as long as the same value is used for the source partitioning as for the calculation of δ^{13} C-CH₄ from simulated [13 CH₄] and [12 CH₄].

The reaction rates for CH₄+OH, CH₄+Cl, and CH₄+O¹D differ between the 12 CH₄ and 13 CH₄ simulations to account for the kinetic isotope effect (KIE). In particular, we assume α values of 0.987 and 0.938 for CH₄+O¹D and CH₄+Cl, respectively (Saueressig et al., 1995;Saueressig et al., 2001). Our standard simulation uses $\alpha_{OH} = 0.9946$ (Cantrell et al., 1990).

Methane from different sources is tracked individually using a "tagged tracer" approach, which allows us to simulate the spatial footprint of CH₄ and δ^{13} C-CH₄ from individual sources. The soil sink is applied to each tracer as a fraction of its source, modified to account for faster loss of ¹²CH₄ to soil compared to ¹³CH₄ (α_{soil} = 0.978) (Tyler et al., 1994). Supplemental figure S1 shows the July 2004 CH₄ and δ^{13} C-CH₄ footprints of the biomass burning, wetland, and coal + other geologic CH₄ sources from the tagged tracers to illustrate the tagged tracer approach. We note that the δ^{13} C values of the surface methane from each source is heavier (less negative) than the emission value for that source (Table 1), especially in regions far from the source, because of the fractionating effects of the sinks. Supplemental Fig. S2 shows the corresponding footprints for January.

2.2 Description of Simulations

We simulate the period from 1990 through 2004, and focus our analysis on 2004. We choose 2004 as our endpoint because it lies within the period when methane concentrations remained relatively flat, simplifying our analysis. Ending the simulations in 2004 also avoids much of the uncertainty about the causes of the resumed growth rate in recent years. The isotopic ratios of methane take longer to adjust to a perturbation than total methane (Tans, 1997). Since we wish to begin our simulations with a state that is as close as possible to "spun up", we choose the initial condition for each tagged tracer based on its present-day distribution and proportion of the total CH₄ and scale it back to 1990 levels such that the total CH₄ is consistent with the global mean CH₄ from surface observations for 1990. We then iteratively adjusted the ¹²C- to ¹³C-CH₄ tracer ratios at the beginning of 1990 to yield a good match to global mean δ¹³C-CH₄ observations for 1998, when more δ¹³C-CH₄ observations are available. The same initial condition is used for the standard and sensitivity simulations.

We use interannually-varying emissions of CH₄ from anthropogenic, biomass burning, and wetland sources. Emissions from anthropogenic sources such as oil and gas, energy production, industrial activities, and livestock come from the EDGAR version 4.2 inventory (European Commission, 2011). Biomass burning emissions come from the MACCity inventory (Granier et al., 2011). We treat forest fires as C3 burning and savannas as C4 burning for partitioning the biomass burning emissions between isotopologues. Wetland and rice emissions come from the Vegetation Integrative Simulator for Trace gases (VISIT) terrestrial ecosystem model (Ito and Inatomi, 2012), scaled by 0.69 and 0.895, respectively, for consistency with the Transcom-CH₄ study (Patra et al., 2011). Ocean (Houweling et al., 1999), termite (Fung et al., 1991), and mud volcano emissions (Etiope and Milkov, 2004) are also from the

Transcom study (Patra et al., 2011) and have a seasonal cycle but no interannual variability. Initial tests with these emissions showed a substantial underestimate of the CH₄ growth rate. Consequently, we scale up all the emissions by 10% for 1990-1998, and by 6.8% for 1998-2004. We find the resulting emissions lead to a good simulation of the timeseries of surface CH₄ observations from the National Oceanic and Atmospheric Administration (NOAA) Global Monitoring Division (GMD) (Dlugokencky et al., 2018), especially towards the end of the period (Fig. 1). The simulation has only a 0.1% mean bias compared to the observations for 2004.

Our standard simulation (SimStd) uses Cl from the GMI CTM for the tropospheric as well as stratospheric loss of CH₄ by reaction with Cl. Tropospheric Cl concentrations are small in GMI since it does not include very short-lived species, and reaction with Cl represents only 0.28% of the total tropospheric CH₄ loss. We also conduct several sensitivity simulations in which we alter the tropospheric and lower stratospheric Cl fields (Table 2). Cl is not altered above 56 hPa. Sensitivity simulation SimGC uses Cl from the GEOS-Chem chemistry module within GEOS (Long et al., 2015; Hu et al., 2018). GEOS-Chem v11-02f with fully coupled tropospheric and stratospheric chemistry was used for this simulation, with halogen chemistry as described in Sherwen et al. (2016). SimGC has higher values of tropospheric Cl than SimStd (Figs. 3,4) and leads to 0.66% of the total CH₄ loss occurring via Cl. Both SimStd and SimGC are thus below the 1% loss via Cl suggested by (Gromov et al., 2018). We conduct a third sensitivity simulation, SimTom, which uses Cl from the TOMCAT model simulations that include chlorine sources from chlorocarbons (including very short-lived substances), HCl from industry and biomass burning, and very short lived substances (Hossaini et al., 2016). This simulation leads to Cl accounting for 2.5% of tropospheric CH₄ loss in our simulation. Finally, we conduct a fourth sensitivity simulation, SimMBL, which modifies the Cl over the oceans at altitudes below 900 hPa (Fig. 2d) to reflect the marine boundary layer distribution suggested by (Allan et al., 2007). This Cl field is described by the following equation:

 $Cl_MBL = 18*10^3 \text{ atoms/cm}^3 * (1 + \tanh(3\lambda)*\sin(2\pi*(t-90)/365))$ (2)

where λ is latitude in radians and t is the day of the year. Elsewhere SimMBL uses the Cl field from SimStd. This simulation has the highest percent of CH₄ loss occurring via Cl: 3.9%. If we consider the loss of methane throughout the atmosphere rather than just the troposphere, then the percent lost via Cl increases to 1.6%, 2.0%, 3.6% and 5.0% for SimStd, SimGC, SimTom, and SimMBL, respectively.

We designed the sensitivity experiments to alter the isotopic composition of CH₄ without greatly affecting the total CH₄. Consequently, we reduce the OH concentrations in the SimTom and SimMBL simulations by 2% and 4%, respectively, relative to the SimStd OH to offset the effect of increasing Cl. These changes are small compared to the uncertainty in global OH (Rigby et al., 2017). In addition, the SimTom and SimMBL simulations use α_{OH} =0.9961 (Saueressig et al., 2001) rather than α_{OH} =0.9946 (Cantrell et al., 1990) to avoid too much fractionation from the combined Cl and OH sinks. While these changes are necessary to maintain consistent total CH₄ and reasonable isotopic ratios, changing multiple factors in addition to Cl makes it difficult to quantify the impact of Cl alone. Consequently, we conduct an additional sensitivity study, called SimTomB, which uses the same Cl field as SimTom but retains the OH and α_{OH} values of SimStd. SimTomB is used in Section 3.3. This simulation becomes too heavy compared to observations, justifying the need to change α_{OH} in the main SimTom simulation. We also

conduct a sensitivity simulation, SimOHp, that uses the same Cl field as SimStd but does not alter the hemispheric ratio of OH. Table 2 summarizes the standard and sensitivity simulations.

The four Cl distributions differ in their vertical and horizontal spatial distributions as well as their tropospheric mean (Figs. 2 and 3). The SimStd Cl is largest in the tropics, nearly symmetric between hemispheres, and increases with altitude. Both SimGC and SimTom have Cl that is larger in the Northern Hemisphere than the Southern Hemisphere in the annual mean and reaches a minimum in the mid-troposphere. However, the maximum in lower tropospheric Cl occurs in the tropics in SimGC but in the extratropics in SimTom. This mid-latitude Cl maximum arises because SimTom has high Cl values over east Asia, whereas SimGC Cl is highest over ocean regions (Fig. 3). SimMBL has a strong maximum in the MBL compared to the free troposphere and land regions. Its annual mean Cl concentrations are higher in the Southern Hemisphere (Fig. 2) due to the larger ocean area in the Southern Hemisphere. However, SimMBL includes a strong seasonal shift in peak Cl between the hemispheres. SimStd and SimGC have more modest seasonal shifts, while Cl in SimTom remains concentrated in the northern hemisphere throughout the year (Fig. S3). All simulations repeat the same Cl field from year to year.

The sensitivity simulations listed above are designed to test the role of the Cl sink. We conduct an additional sensitivity study, SimWet, to illustrate the role of spatial variation in the isotopic source signature. SimWet parallels SimStd, but the isotopic composition of the wetland source uses spatial variation from Ganesan et al (2018). The global mean source signature of the wetland emissions remains -60%.

2.3 Observations

We use surface observations from the NOAA GMD Carbon Cycle Cooperative Global Air Sampling Network to evaluate our simulations. We use the monthly mean observations of total CH₄ (Dlugokencky et al, 2018) and δ^{13} C of CH₄ (White et al., 2018) to compare to the monthly mean simulation results. The isotopic measurements were made at the Institute of Arctic and Alpine Research at the University of Colorado and are referenced to the VPDB scale (Zhang and Li, 1990). The analytical uncertainty of the isotopic measurements is 0.06‰. The variability between measurements taken in a given month may, however, be larger, so we use the maximum of analytical uncertainty and the within-month standard deviation as the uncertainty in the monthly mean. When multiple years are observations are averaged together, we use the pooled variance to calculate the standard error, thus reducing the error based on the number of years. The GMD observations are located at remote sites, shown in Fig. 4 for CH₄ in 2004. Measurements of δ^{13} C of CH₄ are available at a subset of the sites, shown in Fig. 5.

3. Results and Discussion

3.1 Evaluation of Simulated CH₄

We find good agreement between the SimStd simulation and the GMD observations for CH₄ (Fig. 4) for 2004. We focus on these two months to represent the seasonal differences. The latitudinal distribution is well-reproduced, and the simulation captures the elevated concentrations of CH₄ observed over Europe in January as well as the January versus July differences in concentration. Overall, the spatial correlation between SimStd and the observations is 0.93

in January and 0.85 in July. The sensitivity simulations described in Table 2 have little effect on the CH₄ distribution, as shown by the overlapping symbols in Fig. 4c,d.

3.2 Impact of Cl on the δ^{13} C Distribution

227 d

observed and simulated δ^{13} C for 1998-2004 at the 6 GMD sites with δ^{13} C records covering this time period. We begin the figure at 1998 rather than 1990 due to the lack of data availability in the earlier years. The standard and sensitivity simulations overestimate δ^{13} C at the northernmost station, BRW. The observations at the other stations lie within the range of simulations, with most simulations underestimating the observations at the south pole. The differences between the different sensitivity simulations are large compared to the interannual variability in both observed and simulated δ^{13} C. We focus our subsequent analysis on a single year, 2004.

We next examine the distribution of δ^{3} C in SimStd compared to observations. Figure 6 shows the timeseries of

Fig. 5a,b shows both meridional and zonal variability in δ^{13} C. Background values are less negative (heavier) in the Southern versus Northern Hemisphere (NH) (Fig. 7), a feature seen more strongly in the observations, but there is also variability due to the different source signatures. Areas of biomass burning, such as Tropical Africa, show up as particularly heavy, while regions with large wetland and rice emissions, such as SE Asia, are particularly light. Another prominent feature is the isotopically heavy region in northern Eurasia (around 60°N) in January, which we attribute to the influence of the geologic (including oil, gas, and coal) source in this region (Supp. Fig. S2). This signal is less evident in July, when greater influence from boreal wetlands lightens the isotopic mix. The spatial correlation (r^2) between the SimStd and observed δ^{13} C is 0.61 in January and 0.75 in July.

The sensitivity simulations with altered oxidant concentrations alter the global values of δ^{13} C, but the geographic patterns remain similar to that of SimStd. The larger Cl sink in SimGC leads to an overall less negative δ^{13} C, which agrees better than SimStd with observations at Southern Hemisphere (SH) sites but worse in the NH (Figs. 6c,d and 7). The isotopic effect of the larger Cl sink in SimTom is compensated by the lower OH and α_{OH} values used in that simulation, flattening the interhemispheric gradient (Figs. 6e,f and 7). In contrast, the very large MBL Cl concentrations in SimMBL lead to an overestimate (insufficiently negative) of the observed δ^{13} C (Fig. g,h), but strengthens the interhemispheric gradient. We note that since all simulations began with the same initial conditions but have different sinks, the isotopic composition is not in steady state in 2004 and the results of the sensitivity simulations diverge further with additional years of simulation, with SimMBL becoming clearly inconsistent with observations. We note that while these results highlight the differences in δ^{13} C imposed by changing Cl, the absolute values of δ^{13} C, and hence their agreement with observations, would be different for CH₄ source mixtures with a different average δ^{13} C.

Figure 7 reveals an underestimate in the interhemispheric gradient of δ^{13} C in both SimStd and the sensitivity runs compared to the GMD observations. Table 3 presents the observed and simulated δ^{13} C interhemispheric gradients calculated as the difference between the δ^{13} C values averaged over all sites south of 30°S and the average over sites north of 30°N. SimStd and SimGC show similar underestimates of the observed gradient, and the underestimate is more severe in SimTom. The gradient is improved in SimMBL in January. The differences between simulations

reflect differences in the locations where CH₄ oxidation occurs and the amount and location of isotopic fractionation due to Cl versus OH. Fig. 8 shows that the higher Cl values over the NH, particularly China, in SimTom versus SimStd leads to more CH₄ loss occurring in the NH and higher (heavier) δ^{13} C in the NH. This effect is particularly pronounced over China and Europe. Less fractionation by the OH sink in SimTom leads to lighter values in the SH. Conversely, SimMBL has more loss occurring over the SH oceans in January, leading to heavier δ^{13} C in the SH (Fig. 9). This effect is not present in July, when the SimMBL Cl loss shifts to the NH (Fig. S4). The reduced hemispheric difference in OH in SimOHp leads to a small improvement in the hemispheric gradient in δ^{13} C.

We further examine the seasonal cycle of δ^{13} C in Fig. 10. We focus on the seasonal cycle at the South Pole Observatory (SPO) site because it is far from large CH₄ sources and thus the seasonal cycle depends strongly on the seasonality of the CH₄ sinks. While all simulations lie mostly within the error bars of the observations, SimMBL has the largest seasonal cycle amplitude, overestimating the seasonal cycle at of the SPO observations with a δ^{13} C value that is both too heavy in Feb.-June and too light in Aug.-Nov. In contrast, SimStd and the other sensitivity simulations underestimate the magnitude of the observed seasonal cycle at SPO. Supplemental Fig. S5 shows a large enhancement in the seasonal cycle amplitude between SimMBL and the other simulations for the Cape Grim site in Tasmania (CGO), but only a small change at other sites. This suggests that while MBL Cl is attractive as an explanation for the SH seasonality of δ^{13} C, this explanation may be inconsistent with the inclusion of non-marine Cl sources. However, since the seasonal cycle amplitude at SPO lies in between SimMBL and the other simulations, it is possible that at an MBL Cl source similar to that of SimMBL but with a smaller average value could reproduce the amplitude well.

3.3 Quantifying the Sensitivity of δ^{13} C to CH₄ Loss by Cl

Given the substantial range in estimates for how much methane is lost by reaction with tropospheric Cl, it is important to quantify the sensitivity of global mean surface $\delta^{13}C$ to the CH₄ loss by Cl. This analysis summarizes the global impact of the isotopic effect of the Cl differences between simulation discussed above. Fig. 11 shows the global mean, area weighted surface $\delta^{13}C$ in 2004 as a function of the percent of CH₄ oxidized by Cl for SimStd, SimGC, and SimTomB the three simulations with the same OH and emissions but different Cl. A strong linear relationship is evident between the oxidation by Cl and the surface $\delta^{13}C$. The slope of the linear regression line indicates the expected increase in surface $\delta^{13}C$ for a change in the percent of CH₄ oxidized by Cl. Based on this analysis we expect that surface $\delta^{13}C$ will increase by approximately 0.5% for each one % increase in CH₄ loss by Cl.

3.4 Sensitivity of δ^{13} C to the Isotopic Distribution of Sources

Other factors in addition to the Cl distribution likely contribute to the mismatch between the observed and simulated interhemispheric gradients. Fig. 5 shows the impact of the geologic source on the δ^{13} C values over northern Asia. A bias in either the strength or the isotopic composition of this source will impact the interhemispheric gradient. Another likely contributing factor is our use of a globally uniform isotopic ratio for each source type. Ganesan et al.

(2018) developed a global map of the isotopic signatures of wetland emissions. We use this map to impose spatially varying isotopic ratios on our SimWet simulation. SimWet increases the amplitude of the seasonal cycle in δ^{13} C-CH₄ particularly for northern latitudes sites such as ALT, BRW, and MHD (Supplemental Fig. S5). It has little effect on the seasonal cycle at the SH CGO and SPO sites, where SimMBL shows a large effect on the cycle. SimWet results in improved agreement with the observed interhemispheric gradient (Figs. 5,7; Table 3). SimWet is better able to simultaneously match the δ^{13} C-CH₄ observations at both the northernmost (BRW) and southernmost (SPO) sites shown in Fig. 6 than the other simulations, even though all simulations reproduce the latitudinal distribution of CH₄ well (Fig. 4). This highlights the importance of spatially varying isotopic ratios for the δ^{13} C-CH₄ distribution. The size of the effect of including spatially varying ratios in wetland emissions depends on the strength of the wetland emissions as well as the other sources. Including spatially-varying isotopic signature for other sources as well could further modify the simulated interhemispheric gradient, potentially correcting some of the flat gradient of e.g. the SimTom simulation.

4. Conclusions

The role of Cl as a methane sink is a significant uncertainty in the global CH₄ budget, particularly with respect to isotopes. The global distribution of Cl is not well known from observations, and the Cl distributions simulated by global models varies widely from model to model. We investigated the sensitivity of the surface δ^{13} C distribution of CH₄ to the inter-model diversity in tropospheric Cl using a series of sensitivity studies with a global 3D model. Given the uncertainties in CH₄ sources and their isotopic ratios, it is not possible to conclude from this study which Cl field is best. However, the differences between the simulations provides insight on the strong lever that tropospheric Cl exerts on the δ^{13} C distribution.

Our standard and sensitivity simulations all reproduce well the geographic distribution of and temporal evolution of CH₄ observed at the GMD surface sites. However, imposing Cl distributions from a range of chemical transport models used in the scientific community leads to large differences in the simulated distribution of the δ^{13} C of CH₄. The CH₄ sinks from Cl in our SimStd and SimGC simulations are both below 1% of the total CH₄ sink, as suggested by Gromov et al. (2018). Yet the SimStd and SimGC simulations underestimate and overestimate, respectively, the observed δ^{13} C in 2004, despite the fact that both include only a relatively small CH₄ sink from Cl.

Our ability to reproduce the observed latitudinal distribution of δ^{13} C depends not only on the assumed value of global mean Cl, but also its geographic distribution. The detailed halogen chemistry model (TOMCAT) of Hossaini et al. (2016) places the maximum Cl values in the continental NH, in contrast to the large MBL Cl sink used in Allan et al. (2007) to explain SH observations. We find that the strong NH Cl maximum, along with the resulting reduction in OH fractionation required to maintain consistency with observations, acts to flatten the interhemispheric gradient of δ^{13} C, while the MBL Cl sink increases the hemispheric differences in NH winter and also strengthens the seasonal cycle. However, the interhemispheric gradient is also influenced by spatial variation in the isotopic signatures of the sources and uncertainties in the soil sink, complicating this issue.

Two values for the fractionating effect of OH (α_{OH}) on δ^{13} C (Cantrell et al., 1990; Saueressig et al., 2001) are widely cited in the literature. Combining the TOMCAT Cl fields with the α_{OH} of Saueressig et al. (2001) leads to an underestimate of observed δ^{13} C, but combining it with the Cantrell et al. (1990) α_{OH} would lead to an overestimate. Reducing uncertainty in the fractionating effect of OH would thus improve our ability to constrain the role of Cl.

Observations of the δ^{13} C of CH₄ provide an important tool for constraining the CH₄ budget. We find that the range of Cl fields available from current global models leads to a wide range of simulated δ^{13} C values. Each percent increase in the amount of CH₄ loss occurring by reaction with Cl increases global mean surface δ^{13} C of CH₄ by approximately 0.5‰. This relationship can be used to estimate the impact on methane's isotopic values from future model simulations of Cl. The choice of Cl field thus strongly impacts what CH₄ source mixture best fits δ^{13} C observations. Better quantification of the role of Cl in the methane budget and further developing models of tropospheric halogens is therefore critical for interpreting the δ^{13} C observations to their fullest potential.

341342343

331

332

333

334

335

336

337

338

339

340

Data Availability

344 $\delta^{13}\text{CH}_4$ observations The methane and available from the NOAA **GMD** website: are 345 https://www.esrl.noaa.gov/gmd/dv/data/. Output from the GEOS model is on the NASA Center for Climate 346 Simulation (NCCS) system.

347 348

Author Contributions

350 SS designed and conducted simulation, performed analysis, and prepared the manuscript. JSW contributed to model development and experiment design. MM contributed to model development. BD contributed to model development and conceptualization. RH and CK contributed inputs to the simulations. SM and JWCW contributed data and aided in its interpretation. All authors contributed to the editing and revising of the manuscript.

353 354

The authors declare no competing interests.

355 356

Acknowledgements

Computational resources were provided by the NASA Center for Climate Simulation (NCCS). The authors thank Prabir Patra for useful discussions. RH is supported by a NERC Independent Research Fellowship (NE/N014375/1).

359 360

References

- Allan, W., Manning, M. R., Lassey, K. R., Lowe, D. C., and Gomez, A. J.: Modeling the variation of $\delta 13C$ in
- atmospheric methane: Phase ellipses and the kinetic isotope effect, Global Biogeochemical Cycles, 15, 467-481,
- 363 doi:10.1029/2000GB001282, 2001.
- Allan, W., Struthers, H., and Lowe, D.: Methane carbon isotope effects caused by atomic chlorine in the marine
- boundary layer: Global model results compared with Southern Hemisphere measurements, Journal of Geophysical
- Research: Atmospheres, 112, 2007.

- Bock, M., Schmitt, J., Beck, J., Seth, B., Chappellaz, J., and Fischer, H.: Glacial/interglacial wetland, biomass
- burning, and geologic methane emissions constrained by dual stable isotopic CH₄ ice core records, Proceedings of
- 369 the National Academy of Sciences, 114, E5778-E5786, 10.1073/pnas.1613883114, 2017.
- Brownlow, R., Lowry, D., Fisher, R., France, J., Lanoisellé, M., White, B., Wooster, M., Zhang, T., and Nisbet, E.:
- 371 Isotopic ratios of tropical methane emissions by atmospheric measurement, Global Biogeochemical Cycles, 31,
- **372** 1408-1419, 2017.
- Cantrell, C. A., Shetter, R. E., McDaniel, A. H., Calvert, J. G., Davidson, J. A., Lowe, D. C., Tyler, S. C., Cicerone,
- R. J., and Greenberg, J. P.: Carbon kinetic isotope effect in the oxidation of methane by the hydroxyl radical, Journal
- of Geophysical Research: Atmospheres, 95, 22455-22462, 1990.
- 376 Craig, H.: Isotopic Standards for Carbon and Oxygen and Correction Factors for Mass-Spectrometric Analysis OF
- 377 Carbon Dioxide, Geochimica Et Cosmochimica Acta, 12, 133-149, 10.1016/0016-7037(57)90024-8, 1957.
- Dlugokencky, E., Bruhwiler, L., White, J., Emmons, L., Novelli, P., Montzka, S., Masarie, K., Lang, P., Crotwell,
- A., Miller, J., and Gatti, L.: Observational constraints on recent increases in the atmospheric CH4 burden,
- 380 Geophysical Research Letters, 36, 10.1029/2009GL039780, 2009.
- Dlugokencky, E., Nisbet, E., Fisher, R., and Lowry, D.: Global atmospheric methane: budget, changes and dangers,
- Philosophical Transactions of the Royal Society a-Mathematical Physical and Engineering Sciences, 369, 2058-
- 383 2072, 10.1098/rsta.2010.0341, 2011.
- 384 Dlugokencky, E.J., P.M. Lang, A.M. Crotwell, J.W. Mund, M.J. Crotwell, and K.W. Thoning, Atmospheric
- Methane Dry Air Mole Fractions from the NOAA ESRL Carbon Cycle Cooperative Global Air Sampling Network,
- 386 1983-2017, Version: 2018-08-01, Path: ftp://aftp.cmdl.noaa.gov/data/trace_gases/ch4/flask/surface/, 2018.
- Duncan, B. N., Strahan, S. E., Yoshida, Y., Steenrod, S. D., and Livesey, N.: Model study of the cross-tropopause
- transport of biomass burning pollution, Atmospheric Chemistry and Physics, 7, 3713-3736, 2007.
- Elshorbany, Y. F., Duncan, B. N., Strode, S. A., Wang, J. S., and Kouatchou, J.: The description and validation of
- the computationally Efficient CH4-CO-OH (ECCOHv1.01) chemistry module for 3-D model applications,
- 391 Geoscientific Model Development, 9, 799-822, 10.5194/gmd-9-799-2016, 2016.
- Etiope, G., and Milkov, A.: A new estimate of global methane flux from onshore and shallow submarine mud
- 393 volcanoes to the atmosphere, Environmental Geology, 46, 997-1002, 10.1007/s00254-004-1085-1, 2004.
- 394 European Commission, Joint Research Center (JRC)/Netherlands Environmental Assessment Agency (PBL).
- Emission Database for Global Atmospheric Research (EDGAR), http://edgar.jrc.ec.europe.eu.
- Ferretti, D. F., Miller, J. B., White, J. W. C., Etheridge, D. M., Lassey, K. R., Lowe, D. C., Meure, C. M. M., Dreier,
- 397 M. F., Trudinger, C. M., van Ommen, T. D., and Langenfelds, R. L.: Unexpected Changes to the Global Methane
- 398 Budget over the Past 2000 Years, Science, 309, 1714-1717, 10.1126/science.1115193, 2005.
- Fischer, H., Behrens, M., Bock, M., Richter, U., Schmitt, J., Loulergue, L., Chappellaz, J., Spahni, R., Blunier, T.,
- 400 Leuenberger, M., and Stocker, T. F.: Changing boreal methane sources and constant biomass burning during the last
- 401 termination, Nature, 452, 864, 10.1038/nature06825
- 402 https://www.nature.com/articles/nature06825#supplementary-information, 2008.

- 403 Fung, I., John, J., Lerner, J., Matthews, E., Prather, M., Steele, L., and Fraser, P.: 3-Dimensional Model Synthesis of
- the Global Methane Cycle, Journal of Geophysical Research-Atmospheres, 96, 13033-13065, 10.1029/91JD01247,
- **405** 1991.
- Ganesan, A., Stell, A., Gedney, N., Comyn-Platt, E., Hayman, G., Rigby, M., Poulter, B., and Hornibrook, E.:
- 407 Spatially Resolved Isotopic Source Signatures of Wetland Methane Emissions, Geophysical Research Letters, 45,
- **408** 3737-3745, 2018.
- Gelaro, R., McCarty, W., Suarez, M., Todling, R., Molod, A., Takacs, L., Randles, C., Darmenov, A., Bosilovich,
- 410 M., Reichle, R., Wargan, K., Coy, L., Cullather, R., Draper, C., Akella, S., Buchard, V., Conaty, A., da Silva, A.,
- Gu, W., Kim, G., Koster, R., Lucchesi, R., Merkova, D., Nielsen, J., Partyka, G., Pawson, S., Putman, W.,
- Rienecker, M., Schubert, S., Sienkiewicz, M., and Zhao, B.: The Modern-Era Retrospective Analysis for Research
- 413 and Applications, Version 2 (MERRA-2), Journal of Climate, 30, 5419-5454, 10.1175/JCLI-D-16-0758.1, 2017.
- Ghosh, A., Patra, P., Ishijima, K., Umezawa, T., Ito, A., Etheridge, D., Sugawara, S., Kawamura, K., Miller, J., and
- Dlugokencky, E.: Variations in global methane sources and sinks during 1910–2010, Atmospheric Chemistry and
- 416 Physics, 15, 2595-2612, 2015.
- 417 Granier, C., Bessagnet, B., Bond, T., D'Angiola, A., van der Gon, H. D., Frost, G. J., Heil, A., Kaiser, J. W., Kinne,
- 418 S., Klimont, Z., Kloster, S., Lamarque, J. F., Liousse, C., Masui, T., Meleux, F., Mieville, A., Ohara, T., Raut, J. C.,
- Riahi, K., Schultz, M. G., Smith, S. J., Thompson, A., van Aardenne, J., van der Werf, G. R., and van Vuuren, D. P.:
- 420 Evolution of anthropogenic and biomass burning emissions of air pollutants at global and regional scales during the
- 421 1980-2010 period, Climatic Change, 109, 163-190, 10.1007/s10584-011-0154-1, 2011.
- 422 Gromov, S., Brenninkmeijer, C. A., and Jöckel, P.: A very limited role of tropospheric chlorine as a sink of the
- greenhouse gas methane, Atmospheric Chemistry and Physics, 18, 9831-9843, 2018.
- Hausmann, P., Sussmann, R., and Smale, D.: Contribution of oil and natural gas production to renewed increase in
- 425 atmospheric methane (2007–2014): top-down estimate from ethane and methane column observations, Atmospheric
- 426 Chemistry and Physics, 16, 3227-3244, 2016.
- 427 Hopcroft, P. O., Valdes, P. J., and Kaplan, J. O.: Bayesian Analysis of the Glacial-Interglacial Methane Increase
- 428 Constrained by Stable Isotopes and Earth System Modeling, Geophysical Research Letters, 45, 3653-3663,
- 429 doi:10.1002/2018GL077382, 2018.
- Hossaini, R., Chipperfield, M. P., Saiz-Lopez, A., Fernandez, R., Monks, S., Feng, W., Brauer, P., and von Glasow,
- 431 R.: A global model of tropospheric chlorine chemistry: Organic versus inorganic sources and impact on methane
- 432 oxidation, Journal of Geophysical Research: Atmospheres, 121, 14,271-214,297, doi:10.1002/2016JD025756, 2016.
- Houweling, S., Kaminski, T., Dentener, F., Lelieveld, J., and Heimann, M.: Inverse modeling of methane sources
- and sinks using the adjoint of a global transport model, Journal of Geophysical Research-Atmospheres, 104, 26137-
- 435 26160, 10.1029/1999JD900428, 1999.
- Houweling, S., Dentener, F., and Lelieveld, J.: Simulation of preindustrial atmospheric methane to constrain the
- 437 global source strength of natural wetlands, Journal of Geophysical Research-Atmospheres, 105, 17243-17255,
- 438 10.1029/2000JD900193, 2000.

- Houweling, S., Van der Werf, G., Goldewijk, K. K., Röckmann, T., and Aben, I.: Early anthropogenic CH4
- emissions and the variation of CH4 and 13CH4 over the last millennium, Global Biogeochemical Cycles, 22, 2008.
- Hu, L., Keller, C., Long, M., Sherwen, T., Auer, B., Da Silva, A., Nielsen, J., Pawson, S., Thompson, M., Trayanov,
- A., Travis, K., Grange, S., Evans, M., and Jacob, D.: Global simulation of tropospheric chemistry at 12.5 km
- resolution: performance and evaluation of the GEOS-Chem chemical module (v10-1) within the NASA GEOS Earth
- 444 system model (GEOS-5 ESM), Geoscientific Model Development, 11, 4603-4620, 10.5194/gmd-11-4603-2018,
- **445** 2018.
- 446 Ito, A., and Inatomi, M.: Use of a process-based model for assessing the methane budgets of global terrestrial
- ecosystems and evaluation of uncertainty, Biogeosciences, 9, 759-773, 10.5194/bg-9-759-2012, 2012.
- Kai, F., Tyler, S., Randerson, J., and Blake, D.: Reduced methane growth rate explained by decreased Northern
- 449 Hemisphere microbial sources, Nature, 476, 194-197, 10.1038/nature10259, 2011.
- 450 Kirschke, S., Bousquet, P., Ciais, P., Saunois, M., Canadell, J. G., Dlugokencky, E. J., Bergamaschi, P., Bergmann,
- D., Blake, D. R., Bruhwiler, L., Cameron-Smith, P., Castaldi, S., Chevallier, F., Feng, L., Fraser, A., Heimann, M.,
- Hodson, E. L., Houweling, S., Josse, B., Fraser, P. J., Krummel, P. B., Lamarque, J.-F., Langenfelds, R. L., Le
- Quere, C., Naik, V., O'Doherty, S., Palmer, P. I., Pison, I., Plummer, D., Poulter, B., Prinn, R. G., Rigby, M.,
- Ringeval, B., Santini, M., Schmidt, M., Shindell, D. T., Simpson, I. J., Spahni, R., Steele, L. P., Strode, S. A., Sudo,
- 455 K., Szopa, S., van der Werf, G. R., Voulgarakis, A., van Weele, M., Weiss, R. F., Williams, J. E., and Zeng, G.:
- Three decades of global methane sources and sinks, Nature Geoscience, 6, 813-823, 10.1038/NGEO1955, 2013.
- Lassey, K. R., Etheridge, D. M., Lowe, D. C., Smith, A. M., and Ferretti, D. F.: Centennial evolution of the
- 458 atmospheric methane budget: what do the carbon isotopes tell us?, Atmos. Chem. Phys., 7, 2119-2139, 10.5194/acp-
- **459** 7-2119-2007, 2007.
- Long, M., Yantosca, R., Nielsen, J., Keller, C., da Silva, A., Sulprizio, M., Pawson, S., and Jacob, D.: Development
- of a grid-independent GEOS-Chem chemical transport model (v9-02) as an atmospheric chemistry module for Earth
- 462 system models, Geoscientific Model Development, 8, 595-602, 10.5194/gmd-8-595-2015, 2015.
- 463 McNorton, J., Chipperfield, M., Gloor, M., Wilson, C., Feng, W., Hayman, G., Rigby, M., Krummel, P., O'Doherty,
- S., Prinn, R., Weiss, R., Young, D., Dlugokencky, E., and Montzka, S.: Role of OH variability in the stalling of the
- global atmospheric CH4 growth rate from 1999 to 2006, Atmospheric Chemistry and Physics, 16, 7943-7956,
- 466 10.5194/acp-16-7943-2016, 2016.
- Mikaloff Fletcher, S. E., Tans, P. P., Bruhwiler, L. M., Miller, J. B., and Heimann, M.: CH4 sources estimated from
- 468 atmospheric observations of CH4 and its 13C/12C isotopic ratios: 2. Inverse modeling of CH4 fluxes from
- geographical regions, Global Biogeochemical Cycles, 18, doi:10.1029/2004GB002224, 2004a.
- 470 Mikaloff Fletcher, S. E., Tans, P. P., Bruhwiler, L. M., Miller, J. B., and Heimann, M.: CH4 sources estimated from
- 471 atmospheric observations of CH4 and its 13C/12C isotopic ratios: 1. Inverse modeling of source processes, Global
- 472 Biogeochemical Cycles, 18, doi:10.1029/2004GB002223, 2004b.
- 473 Molod, A., Takacs, L., Suarez, M., and Bacmeister, J.: Development of the GEOS-5 atmospheric general circulation
- model: evolution from MERRA to MERRA2, Geoscientific Model Development, 8, 1339-1356, 10.5194/gmd-8-
- **475** 1339-2015, 2015.

- 476 Monteil, G., Houweling, S., Dlugockenky, E., Maenhout, G., Vaughn, B., White, J., and Rockmann, T.: Interpreting
- 477 methane variations in the past two decades using measurements of CH 4 mixing ratio and isotopic composition,
- 478 Atmospheric chemistry and physics, 11, 9141-9153, 2011.
- Naik, V., Voulgarakis, A., Fiore, A. M., Horowitz, L. W., Lamarque, J. F., Lin, M., Prather, M. J., Young, P. J.,
- 480 Bergmann, D., Cameron-Smith, P. J., Cionni, I., Collins, W. J., Dalsoren, S. B., Doherty, R., Eyring, V., Faluvegi,
- 481 G., Folberth, G. A., Josse, B., Lee, Y. H., MacKenzie, I. A., Nagashima, T., van Noije, T. P. C., Plummer, D. A.,
- Righi, M., Rumbold, S. T., Skeie, R., Shindell, D. T., Stevenson, D. S., Strode, S., Sudo, K., Szopa, S., and Zeng,
- 483 G.: Preindustrial to present-day changes in tropospheric hydroxyl radical and methane lifetime from the
- 484 Atmospheric Chemistry and Climate Model Intercomparison Project (ACCMIP), Atmospheric Chemistry and
- 485 Physics, 13, 5277-5298, 10.5194/acp-13-5277-2013, 2013.
- Nielsen, J., Pawson, S., Molod, A., Auer, B., da Silva, A., Douglass, A., Duncan, B., Liang, Q., Manyin, M., Oman,
- 487 L., Putman, W., Strahan, S., and Wargan, K.: Chemical Mechanisms and Their Applications in the Goddard Earth
- Observing System (GEOS) Earth System Model, Journal of Advances in Modeling Earth Systems, 9, 3019-3044,
- 489 10.1002/2017MS001011, 2017.
- Nisbet, E., Dlugokencky, E., Manning, M., Lowry, D., Fisher, R., France, J., Michel, S., Miller, J., White, J., and
- Vaughn, B.: Rising atmospheric methane: 2007–2014 growth and isotopic shift, Global Biogeochemical Cycles, 30,
- **492** 1356-1370, 2016.
- Nisbet, E., Manning, M., Dlugokencky, E., Fisher, R., Lowry, D., Michel, S., Myhre, C., Platt, M., Allen, G.,
- Bousquet, P., Brownlow, R., Cain, M., France, J., Hermansen, O., Hossaini, R., Jones, A., Levin, I., Manning, A.,
- Myhre, G., Pyle, J., Vaughn, B., Warwick, N., and White, J.: Very Strong. Atmospheric Methane Growth in the 4
- 496 Years 2014-2017: Implications for the paris Agreement, Global Biogeochemical Cycles, 33, 318-342,
- 497 10.1029/2018GB006009, 2019.
- 498 Orbe, C., Oman, L., Strahan, S., Waugh, D., Pawson, S., Takacs, L., and Molod, A.: Large-Scale Atmospheric
- Transport in GEOS Replay Simulations, Journal of Advances in Modeling Earth Systems, 9, 2545-2560,
- 500 10.1002/2017MS001053, 2017.
- Patra, P., Houweling, S., Krol, M., Bousquet, P., Belikov, D., Bergmann, D., Bian, H., Cameron-Smith, P.,
- Chipperfield, M., Corbin, K., Fortems-Cheiney, A., Fraser, A., Gloor, E., Hess, P., Ito, A., Kawa, S., Law, R., Loh,
- Z., Maksyutov, S., Meng, L., Palmer, P., Prinn, R., Rigby, M., Saito, R., and Wilson, C.: TransCom model
- simulations of CH4 and related species: linking transport, surface flux and chemical loss with CH4 variability in the
- troposphere and lower stratosphere, Atmospheric Chemistry and Physics, 11, 12813-12837, 10.5194/acp-11-12813-
- **506** 2011, 2011.
- Platt, U., Allan, W., and Lowe, D.: Hemispheric average Cl atom concentration from ¹³C/¹²C ratios in atmospheric
- 508 methane, Atmos. Chem. Phys., 4, 2393-2399, 10.5194/acp-4-2393-2004, 2004.
- Quay, P., King, S., Stutsman, J., Wilbur, D., Steele, L., Fung, I., Gammon, R., Brown, T., Farwell, G., Grootes, P.,
- and Schmidt, F.: Carbon Isotopic Composition of Atmospheric CH4: Fossil and Biomass Burning Source Strenghts,
- 511 Global Biogeochemical Cycles, 5, 25-47, 10.1029/91GB00003, 1991.

- Rice, A. L., Butenhoff, C. L., Teama, D. G., Röger, F. H., Khalil, M. A. K., and Rasmussen, R. A.: Atmospheric
- methane isotopic record favors fossil sources flat in 1980s and 1990s with recent increase, Proceedings of the
- 514 National Academy of Sciences, 113, 10791-10796, 2016.
- Rigby, M., Prinn, R. G., Fraser, P. J., Simmonds, P. G., Langenfelds, R. L., Huang, J., Cunnold, D. M., Steele, L. P.,
- Krummel, P. B., Weiss, R. F., O'Doherty, S., Salameh, P. K., Wang, H. J., Harth, C. M., Mühle, J., and Porter, L.
- W.: Renewed growth of atmospheric methane, Geophysical Research Letters, 35, doi:10.1029/2008GL036037,
- **518** 2008.
- Rigby, M., Montzka, S. A., Prinn, R. G., White, J. W., Young, D., O'Doherty, S., Lunt, M. F., Ganesan, A. L.,
- Manning, A. J., and Simmonds, P. G.: Role of atmospheric oxidation in recent methane growth, Proceedings of the
- 521 National Academy of Sciences, 114, 5373-5377, 2017.
- Rotman, D., Tannahill, J., Kinnison, D., Connell, P., Bergmann, D., Proctor, D., Rodriguez, J., Lin, S., Rood, R.,
- Prather, M., Rasch, P., Considine, D., Ramaroson, R., and Kawa, S.: Global Modeling Initiative assessment model:
- Model description, integration, and testing of the transport shell, Journal of Geophysical Research-Atmospheres,
- **525** 106, 1669-1691, 10.1029/2000JD900463, 2001.
- Saueressig, G., Bergamaschi, P., Crowley, J., Fischer, H., and Harris, G.: Carbon Kinetic Isotope Effect in the
- 527 Reaction of CH4 with Cl Atoms, Geophysical Research Letters, 22, 1225-1228, 10.1029/95GL00881, 1995.
- 528 Saueressig, G., Crowley, J. N., Bergamaschi, P., Brühl, C., Brenninkmeijer, C. A. M., and Fischer, H.: Carbon 13
- and D kinetic isotope effects in the reactions of CH4 with O(1 D) and OH: New laboratory measurements and their
- implications for the isotopic composition of stratospheric methane, Journal of Geophysical Research: Atmospheres,
- 531 106, 23127-23138, doi:10.1029/2000JD000120, 2001.
- 532 Saunois, M., Bousquet, P., Poulter, B., Peregon, A., Ciais, P., Canadell, J. G., Dlugokencky, E. J., Etiope, G.,
- Bastviken, D., and Houweling, S.: The global methane budget 2000–2012, Earth System Science Data (Online), 8,
- **534** 2016.
- Saunois, M., Bousquet, P., Poulter, B., Peregon, A., Ciais, P., Canadell, J., Dlugokencky, E., Etiope, G., Bastviken,
- D., Houweling, S., Janssens-Maenhout, G., Tubiello, F., Castaldi, S., Jackson, R., Alexe, M., Arora, V., Beerling,
- D., Bergamaschi, P., Blake, D., Brailsford, G., Bruhwiler, L., Crevoisier, C., Crill, P., Covey, K., Frankenberg, C.,
- Gedney, N., Hoglund-Isaksson, L., Ishizawa, M., Ito, A., Joos, F., Kim, H., Kleinen, T., Krummel, P., Lamarque, J.,
- Langenfelds, R., Locatelli, R., Machida, T., Maksyutov, S., Melton, J., Morino, I., Naik, V., O'Doherty, S.,
- Parmentier, F., Patra, P., Peng, C., Peng, S., Peters, G., Pison, I., Prinn, R., Ramonet, M., Riley, W., Saito, M.,
- Santini, M., Schroeder, R., Simpson, I., Spahni, R., Takizawa, A., Thornton, B., Tian, H., Tohjima, Y., Viovy, N.,
- Voulgarakis, A., Weiss, R., Wilton, D., Wiltshire, A., Worthy, D., Wunch, D., Xu, X., Yoshida, Y., Zhang, B.,
- Zhang, Z., and Zhu, Q.: Variability and quasi-decadal changes in the methane budget over the period 2000-2012,
- 544 Atmospheric Chemistry and Physics, 17, 11135-11161, 10.5194/acp-17-11135-2017, 2017.
- Schaefer, H., Fletcher, S. E. M., Veidt, C., Lassey, K. R., Brailsford, G. W., Bromley, T. M., Dlugokencky, E. J.,
- Michel, S. E., Miller, J. B., Levin, I., Lowe, D. C., Martin, R. J., Vaughn, B. H., and White, J. W. C.: A 21st-century
- shift from fossil-fuel to biogenic methane emissions indicated by 13CH4, Science, 352, 80-84,
- 548 10.1126/science.aad2705, 2016.

- 549 Schwietzke, S., Sherwood, O., Ruhwiler, L., Miller, J., Etiope, G., Dlugokencky, E., Michel, S., Arling, V., Vaughn,
- B., White, J., and Tans, P.: Upward revision of global fossil fuel methane emissions based on isotope database,
- 551 Nature, 538, 88-91, 10.1038/nature19797, 2016.
- 552 Sherwen, T., Schmidt, J. A., Evans, M. J., Carpenter, L. J., Großmann, K., Eastham, S. D., Jacob, D. J., Dix, B.,
- Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A., Prados-Roman, C., Mahajan, A. S., and
- Ordóñez, C.: Global impacts of tropospheric halogens (Cl, Br, I) on oxidants and composition in GEOS-Chem,
- 555 Atmos. Chem. Phys., 16, 12239-12271, 10.5194/acp-16-12239-2016, 2016.
- 556 Sherwood, O., Schwietzke, S., Arling, V., and Etiope, G.: Global Inventory of Gas Geochemistry Data from Fossil
- Fuel, Microbial and Burning Sources, version 2017, Earth System Science Data, 9, 639-656, 10.5194/essd-9-639-
- **558** 2017, 2017.
- 559 Spivakovsky, C. M., Logan, J. A., Montzka, S. A., Balkanski, Y. J., Foreman-Fowler, M., Jones, D. B. A.,
- Horowitz, L. W., Fusco, A. C., Brenninkmeijer, C. A. M., Prather, M. J., Wofsy, S. C., and McElroy, M. B.: Three-
- dimensional climatological distribution of tropospheric OH: Update and evaluation, Journal of Geophysical
- 562 Research-Atmospheres, 105, 8931-8980, 10.1029/1999jd901006, 2000.
- 563 Strahan, S. E., Duncan, B. N., and Hoor, P.: Observationally derived transport diagnostics for the lowermost
- stratosphere and their application to the GMI chemistry and transport model, Atmospheric Chemistry and Physics, 7,
- **565** 2435-2445, 2007.
- 566 Strahan, S. E., Douglass, A. R., and Newman, P. A.: The contributions of chemistry and transport to low arctic
- ozone in March 2011 derived from Aura MLS observations, Journal of Geophysical Research: Atmospheres, 118,
- **568** 1563-1576, 2013.
- 569 Strode, S., Duncan, B., Yegorova, E., Kouatchou, J., Ziemke, J., and Douglass, A.: Implications of carbon monoxide
- bias for methane lifetime and atmospheric composition in chemistry climate models, Atmospheric Chemistry and
- **571** Physics, 15, 11789-11805, 2015.
- Tans, P.: A note on isotopic ratios and the global atmospheric methane budget, Global Biogeochemical Cycles, 11,
- **573** 77-81, 10.1029/96GB03940, 1997.
- Thompson, R., Nisbet, E., Pisso, I., Stohl, A., Blake, D., Dlugokencky, E., Helmig, D., and White, J.: Variability in
- Atmospheric Methane From Fossil Fuel and Microbial Sources Over the Last Three Decades, Geophysical Research
- 576 Letters, 45, 11499-11508, 10.1029/2018GL078127, 2018.
- Thompson, R. L., Stohl, A., Zhou, L. X., Dlugokencky, E., Fukuyama, Y., Tohjima, Y., Kim, S. Y., Lee, H., Nisbet,
- E. G., and Fisher, R. E.: Methane emissions in East Asia for 2000–2011 estimated using an atmospheric Bayesian
- inversion, Journal of Geophysical Research: Atmospheres, 120, 4352-4369, 2015.
- Turner, A., Jacob, D., Benmergui, J., Wofsy, S., Maasakkers, J., Butz, A., Hasekamp, O., and Biraud, S.: A large
- increase in US methane emissions over the past decade inferred from satellite data and surface observations,
- 582 Geophysical Research Letters, 43, 2218-2224, 10.1002/2016GL067987, 2016.
- Turner, A. J., Frankenberg, C., Wennberg, P. O., and Jacob, D. J.: Ambiguity in the causes for decadal trends in
- atmospheric methane and hydroxyl, Proceedings of the National Academy of Sciences, 114, 5367-5372, 2017.

- Tyler, S. C., Crill, P. M., and Brailsford, G. W.: ¹³C¹²C Fractionation of methane during oxidation in a temperate
- 586 forested soil, Geoschem. Cosmochem. Acta, 58, 1625-1633, 10.1016/0016-7037(94)90564-9, 1994.
- Wang, J. S., McElroy, M. B., Spivakovsky, C. M., and Jones, D. B. A.: On the contribution of anthropogenic Cl to
- 588 the increase in δ13C of atmospheric methane, Global Biogeochemical Cycles, 16, 20-21-20-11, 2002.
- Wang, X., Jacob, D., Eastham, S., Sulprizio, M., Zhu, L., Chen, Q., Alexander, B., Sherwen, T., Evans, M., Lee, B.,
- Haskins, J., Lopez-Hilfiker, F., Thornton, J., Huey, G., and Liao, H.: The role of chlorine in global tropospheric
- 591 chemistry, Atmospheric Chemistry and Physics, 19, 3981-4003, 10.5194/acp-19-3981-2019, 2019.
- Waugh, D., Crotwell, A., Dlugokencky, E., Dutton, G., Elkins, J., Hall, B., Hintsa, E., Hurst, D., Montzka, S.,
- Mondeel, D., Moore, F., Nance, J., Ray, E., Steenrod, S., Strahan, S., and Sweeney, C.: Tropospheric SF6: Age of
- air from the Northern Hemisphere midlatitude surface, Journal of Geophysical Research-Atmospheres, 118, 11429-
- 595 11441, 10.1002/jgrd.50848, 2013.
- White, J.W.C., B.H. Vaughn, and S.E. Michel, University of Colorado, Institute of Arctic and Alpine Research
- 597 (INSTAAR), Stable Isotopic Composition of Atmospheric Methane (13C) from the NOAA ESRL Carbon Cycle
- Cooperative Global Air Sampling Network, 1998-2017, Version: 2018-09-24, Path:
- 599 ftp://aftp.cmdl.noaa.gov/data/trace_gases/ch4c13/flask/, 2018.
- Worden, J. R., Bloom, A. A., Pandey, S., Jiang, Z., Worden, H. M., Walker, T. W., Houweling, S., and Röckmann,
- T.: Reduced biomass burning emissions reconcile conflicting estimates of the post-2006 atmospheric methane
- budget, Nature communications, 8, 2227, 2017.
- Zhang, Q.-L. and Li, W.-J.: A Calibrated Measurement of the Atomic Weight of Carbon, Chinese Science Bulletin
- 604 35, 290, doi: 10.1360/sb1990-35-4-290, 1990.

Table 1: Emission source references and δ^{13} C values

Source	Reference	IAV	δ ¹³ C (‰) ^a	CH ₄ Source (Tg yr ⁻¹) ^b
Animals (enteric fermentation)	EDGAR	Y	-62	102
C3 Biomass Burning (Forests)	MACCity	Y	-26	16
C4 Biomass Burning (Savannas)	MACCity	Y	-15	10
Coal, energy, and industry	EDGAR	Y	-35	6
Geologic (oil/gas/non-coal fuels, volcanos)	EDGAR, Transcom	Y, except volcanos	-40	124
Waste (solid and animal waste, wastewater)	EDGAR	Y	-55	74
Ocean	Transcom	N	-59	8
Rice	Visit model	Y	-63	44
Termites	Transcom	N	-57	22
Wetlands	Visit model	Y	-60	149

^aδ¹³C values from Dlugokencky et al., 2011;Lassey et al., 2007;Monteil et al., 2011;Houweling et al., 2000 and refs therein

^bValues for 2004

607

608 609

610 611 612

Table 2: Oxidants for the Standard and sensitivity simulations

Simulation	[Cl] _{Trop} ^a (molec cm ⁻³)	Cl Model ^b	Cl Reference	OH modification ^c
SimStd	210	GMI	(Strahan et al., 2007; Rotman et al., 2001; Strahan et al., 2013;Duncan et al., 2007)	$\alpha = 0.9946$
SimGC	384	GEOSChem	(Sherwen et al., 2016)	$\alpha = 0.9946$
SimTom	1710	TOMCAT	(Hossaini et al., 2016)	-2% [OH] $\alpha = 0.9961$
SimTomB	1710	TOMCAT	(Hossaini et al., 2016)	$\alpha = 0.9946$
SimOHp	210	GMI	See SimStd	Not modified for 20% higher in NH
SimMBL	2810	Tanh function below 900hPa over ocean; GMI elsewhere	(Allan et al., 2007)	-4% [OH] $\alpha = 0.9961$

^aConcentration of Cl averaged over the troposphere

^bName of the model that generated the offline Cl field

^cChanges to [OH] or α_{OH} compared to SimStd

Table 3: Observed and Simulated Interhemispheric Gradient in δ^{13} C-CH₄

	Jan. Gradient (‰) ^a	July Gradient (‰) ^a	
GMD Obs	0.36	0.28	
SimStd	0.17	0.11	
SimGC	0.17	0.098	
SimTom	0.051	0.010	
SimMBL	0.30	0.13	
SimOHp	0.22	0.15	
SimWet	0.28	0.25	

^aAverage δ¹³C-CH₄ at GMD site locations south of 30°S minus average δ¹³C-CH₄ at locations north of 30°N

618 619

613 614

615

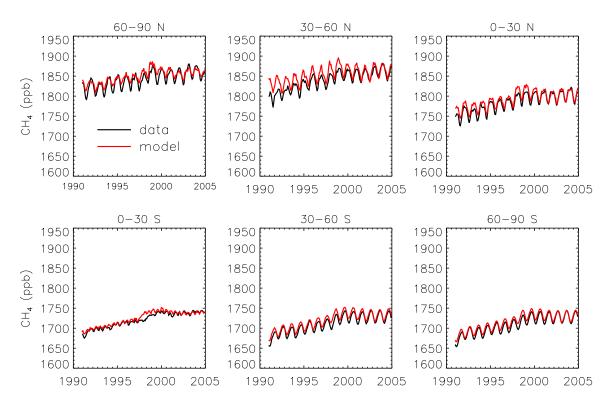


Fig 1: Monthly CH₄ observations from the GMD network (black) and simulated surface concentrations from SimStd (red) averaged over latitude bands

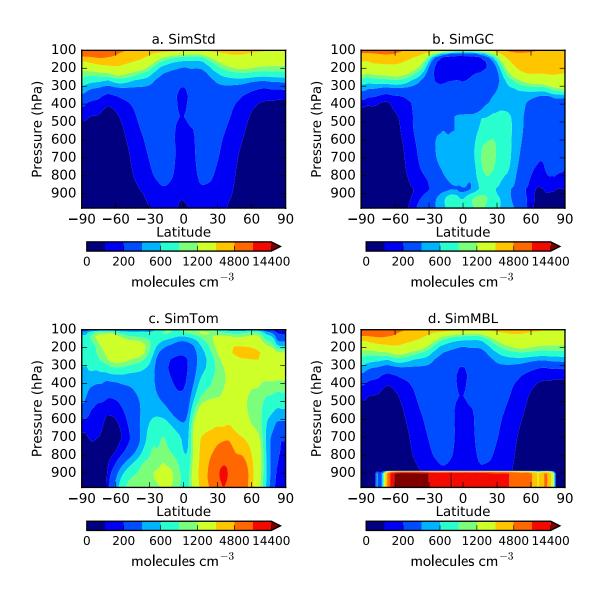


Fig. 2: Annual zonal mean Cl field for a) SimStd, b) SimGC, c) SimTom, and d) SimMBL.

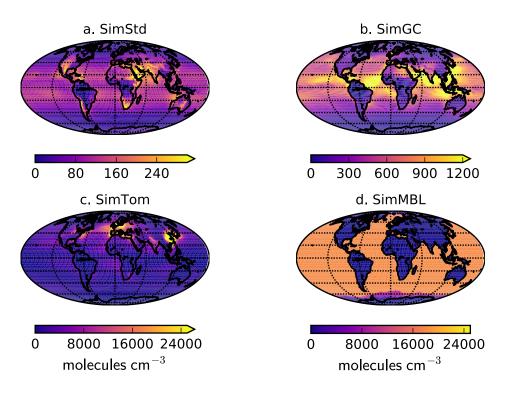


Fig. 3: Annual mean surface concentrations of Cl in a) SimStd, b) SimGC, c) SimTom, and d) SimMBL. Note the different color scales between panels.

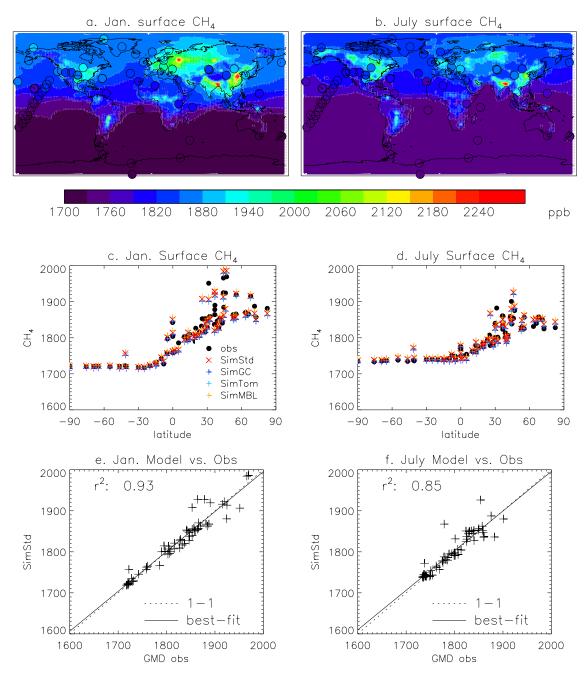


Fig. 4: Comparison of 2004 simulated and observed surface CH₄ concentrations for January (left) and July (right). a,b) Surface concentrations of CH₄ from SimStd are overplotted with the concentrations from the GMD observations in circles. c,d) GMD observations (black circles), SimStd (red x), SimGC (dark blue +), SimTom (light blue +), and SimMBL (orange +) CH₄ as a function of latitude. E,f) SimStd CH₄ (ppb) at the observation locations versus the GMD observations (+ signs) as well as the regression line (solid) and 1 to 1 line (dashed).

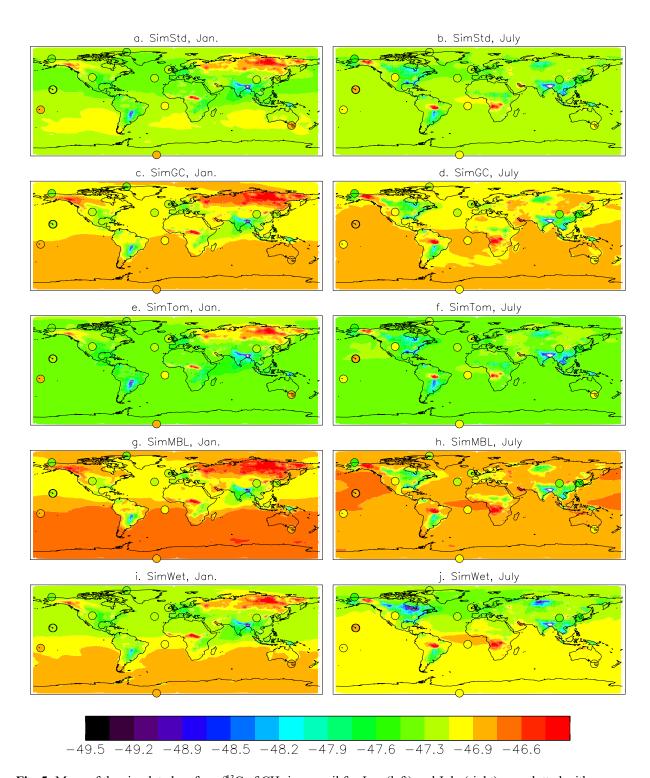


Fig. 5: Maps of the simulated surface $\delta^{13}C$ of CH_4 in per mil for Jan. (left) and July (right) overplotted with observations from the GMD sites (circles). The simulations are (a,b) SimStd, (c,d) SimGC, (e,f) SimTom, (g,h) SimMBL, and (I,j) SimWet.

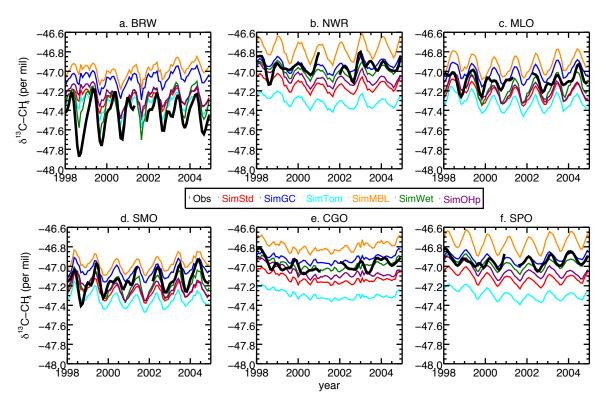


Fig. 6: The timeseries of observed (black) and simulated (colors) δ^{13} CH₄ at the 6 GMD sites with records extending back to 1998. BRW: 71.3°N, 156.6°W; NWR: 40.0°N, 105.6°W; MLO: 19.5°N, 155.6°W; CGO: 40.7°S, 144.7°E and SPO: 90.0°S, 24.8°W.

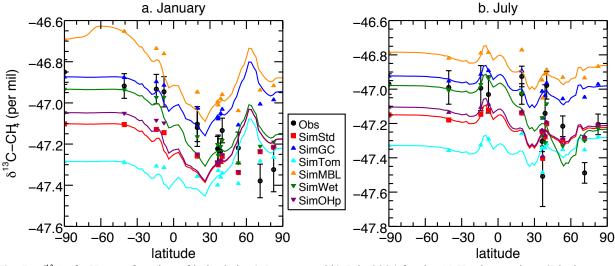


Fig. 7: δ^{13} C of CH₄ as a function of latitude in a) January and b) July 2004 for the GMD observations (Black circles), SimStd (red), SimGC (dark blue), SimTom (cyan), SimMBL (orange), SimWet (green), and SimOHp (purple). Errorbars represent the maximum of the analytical uncertainty (0.06‰) and the standard deviation of individual measurements in the month for each site. The colored lines represent the simulated zonal mean, while the colored symbols represent the simulation sampled at the location of the GMD observations.

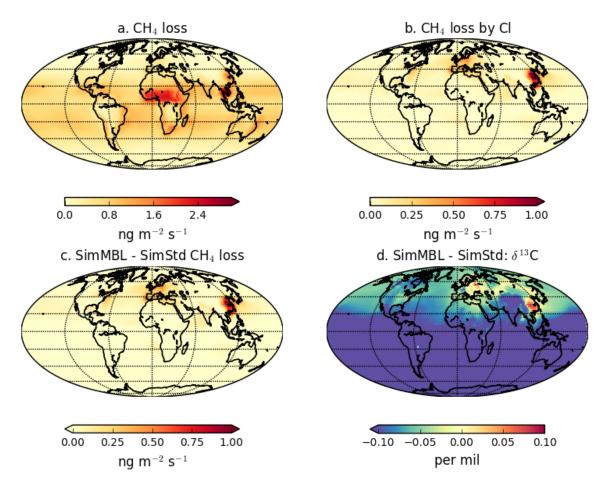


Fig 8: January a) CH₄ loss and b) CH₄ loss by Cl only in the SimTom simulation, as well as the difference in c) CH₄ loss and d) δ^{13} C-CH₄ between the SimTom and SimStd simulations.

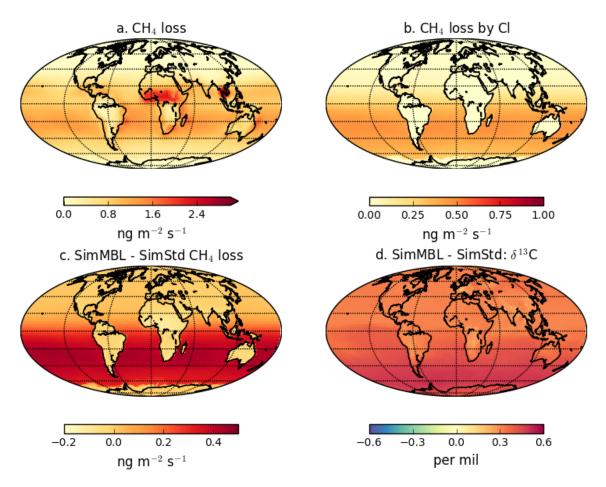


Fig 9: January a) CH₄ loss and b) CH₄ loss by Cl only in the SimMBL simulation, as well as the difference in c) CH₄ loss and d) δ^{13} C-CH₄ between the SimMBL and SimStd simulations.

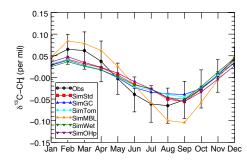


Fig. 10: The seasonal cycle of δ^{13} C of CH₄ at the SPO site with the annual mean removed averaged over 2002-2004 for the GMD observations (black), SimStd (red), SimGC (blue), SimTom (cyan), SimMBL (orange), SimWet (green), and SimOHp (purple). Errorbars represent the standard error, calculated as the maximum of the pooled standard deviation or the analytical uncertainty (0.06‰), divided by the square root of the number of years of observations.

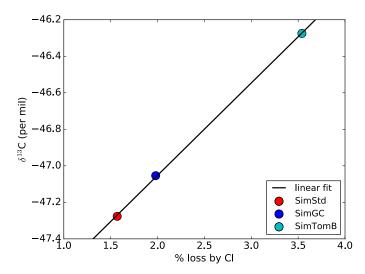


Fig. 11: Area-weighted global mean surface $\delta^{13}C$ for the SimStd (red), SimGC (blue) and SimTomB (cyan) simulations in 2004 as a function of the percent of CH₄ loss occurring by reaction with Cl. The linear best-fit line is shown in black.