Net ozone production and its relationship to NO_x and VOCs in the marine boundary layer around the Arabian Peninsula

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Abstract. Strongly enhanced tropospheric ozone mixing ratios have been reported in the Arabian Basin, a region with intense solar radiation and high concentrations of ozone precursors such as nitrogen oxides and volatile organic compounds. To analyze photochemical ozone production in the marine boundary layer (MBL) around the Arabian Peninsula, we use ship-

- 15 borne observations of NO, NO₂, O₃, OH, HO₂, HCHO, actinic flux, water vapor, pressure and temperature obtained during the summer 2017 Air Quality and Climate in the Arabian Basin (AQABA) campaign, compare them to simulation results of the ECHAM-MESSy atmospheric chemistry (EMAC) general circulation model. Net ozone production rates (NOPR) were greatest with 16 ppb_v day⁻¹ over both the Gulf of Oman and the Northern Red Sea and with 32 ppb_v day⁻¹ over the Arabian Gulf. NOPR over the Mediterranean, the Southern Red Sea and the Arabian Sea did not significantly deviate from zero;
- 20 however, results for the Arabian Sea indicate weak net ozone production of 5 ppb_v day⁻¹, and net ozone destruction over the Mediterranean and the Southern Red Sea with -1 ppb_v day⁻¹ and -4 ppb_v day⁻¹, respectively. Constrained by HCHO/NO₂-ratios, our photochemistry calculations show that net ozone production in the MBL around the Arabian Peninsula occurs mostly in NO_x-limitation regimes with a significant share of ozone production occurring in the transition regime between NO_x- and VOC-limitation over the Mediterranean and more significantly over the Northern Red Sea and Oman Gulf.

25 1 Introduction

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Revenues from exploitation of the great oil reserves in the states of and around the Arabian Peninsula have propelled remarkable economic development associated with industrialization and urbanization. Strong population growth and anthropogenic emissions of gases and particulates in the last few decades have resulted in the Middle East becoming a hotspot for air pollution and associated health effects, while it is also one of the regions worldwide where climate change is particularly

- 30 rapid (Lelieveld et al., 2016a). Unique meteorological conditions such as intense solar radiation, high temperatures and aridity, as well as strong anthropogenic emissions of volatile organic compounds (VOCs) and NO_x (= NO + NO₂) by on- and off-shore petrochemical industries, dense ship traffic, fossil energy production for air conditioning and desalination, and urban development are expected to further intensify in the future and contribute to photochemical ozone production (Lelieveld et al, 2009; Krotkov et al., 2016; Pfannerstill et al., 2019). Understanding the sources and sinks of NO_x and other ozone precursors on and around the Arabian Peninsula is therefore of major importance for atmospheric chemistry studies, including the
- investigation of net ozone production rates (NOPR) (Monks et al., 2015; Reed et al., 2016; Bozem et al., 2017).

NO_x plays a central role in atmospheric photochemistry (Nakamura et al., 2003; Tuzson et al., 2013; Reed et al., 2016). It is the primary precursor for tropospheric ozone (O₃), secondary organic aerosols and photochemical smog in urban areas (Hollaway et al., 2012; Javed et al., 2019). Main ground-based sources of NO and NO₂ are fossil fuel combustion and to a lesser extent bacterial processes in soils, and both lightning and aircraft emissions in the upper troposphere (Nakamura et al., 2003; Miyazaki et al., 2017; Javed et al., 2019). Transport of NO_x in the atmosphere is relatively limited due to its short lifetime of a few hours (Reed et al., 2016). It is removed from the troposphere mainly by conversion to HNO₃ (via reaction with OH) during the day, or the formation of N₂O₅ (in the reaction of NO₂ with NO₃ at night-time), which also leads to formation of nitric acid by heterogeneous hydrolysis on aerosol surfaces (Crutzen, 1973; Liu et al., 2016; Reed et al., 2016). Ultimately, the

- 45 deposition of HNO₃ constitutes the major loss process of NO_x from the atmosphere. Ozone is a secondary pollutant that is photochemically formed in the troposphere from its precursors NO_x and VOCs (Bozem et al., 2017; Jaffe et al., 2018). It is an important greenhouse gas, an atmospheric oxidant and the most important primary precursor for OH (Lelieveld et al., 2004; Monks et al., 2015; Bozem et al., 2017). O₃ in the planetary boundary layer causes health damage, notably respiratory diseases, and reduces crop yields (Monks et al., 2015; Jaffe et al., 2018).
- 50 NO_x and O₃ mixing ratios in the troposphere vary from less than 20 ppt_v and 10 ppb_v, respectively, for pristine conditions such as the remote marine boundary layer (MBL) up to mixing ratios of several hundreds of ppb_v in regions with heavy automobile traffic and in international shipping lanes (for NO_x) and downwind of urbanized areas (for O₃) (Reed et al., 2016; Jaffe et al., 2018). Low NO_x environments such as the clean MBL and the lower free troposphere are considered net ozone destruction regimes whereas the upper troposphere and areas with anthropogenic emissions of ozone precursors are regions of net ozone
- 55 production (Klonecki and Levy, 1997; Bozem et al., 2017). Measurements performed in the Houston Ship Channel revealed NOPR of the order of several tens of ppb h⁻¹ (Chen et al., 2010; Mao et al., 2010; Ren et al., 2013).

In the last decade much effort has been successfully devoted to the mitigation of NO_x emissions over Europe and America, and levels of reactive nitrogen trace gases have decreased (Miyazaki et al., 2017). But in Asia, India and the Middle East, NO_x emissions have substantially increased during the last decade so that the global NO_x burden has essentially remained constant

- 60 (Miyazaki et al., 2017). NO_x emissions by ocean-going vessels have attracted considerable attention as they are reported to account for 15 % of the global NO_x emission burden (Celik et al., 2019). Model calculations suggest that the Arabian Gulf, with an estimated annual NO_x emission density of about one ton km⁻² from ship traffic, is among the regions with highest NO_x emission densities worldwide (Johansson et al., 2017). Although NO_x emissions in the Red Sea and Arabian Sea areas were reported to be three and five times smaller than for the Arabian Gulf, respectively, these values are still 50-100 times larger
- 65 than the emission density reported for the South Pacific Ocean, for example (Johansson et al., 2017).

In the present study, we characterize photochemical NOPR in the MBL around the Arabian Peninsula. In Sect. 2, the campaign, instrument description, data processing and a description of the methods used in this study is presented. In Sect. 3, mixing ratios of nitrogen oxides and ozone around the Arabian Peninsula are reported. Based on concurrent measurements of HO_x , actinic flux, temperature and pressure, noontime (HO_2+RO_2) mixing ratios are estimated and used to calculate NOPR in the

70 different regions around the Arabian Peninsula. Observation-based analysis of HCHO/NO₂-ratios will be used to distinguish between NO_x- or VOC-limited chemistry in the particular regions. A comparison of the results with data retrieved from the 3D global circulation model EMAC is also included.

2 Experimental

2.1 AQABA campaign

- 75 The AQABA ship campaign (<u>A</u>ir <u>Q</u>uality and Climate in the <u>A</u>rabian <u>Ba</u>sin) investigated the chemical composition of the MBL around the Arabian Peninsula. From late June to early September 2017, the *Kommandor Iona* Research and Survey Vessel sailed from Toulon (France) to Kuwait and back in order to perform gas-phase and particle measurements in the region. The gas-phase and aerosol measurement instrumentation was housed in five laboratory containers on the front deck. A 6 m high, 20 cm diameter cylindrical stainless steel common inlet was installed on the front deck of the vessel to sample air at a total mass flow rate of 10,000 SLM. NO and NO₂ chemiluminescence measurements were obtained at a total bypass flow rate of 28.5 SLM sampling air from the common inlet with a residence time in the tubing of ~3 s. HCHO, NO₂ cavity ring-down spectroscopy and O₃ measurements were obtained with similar bypass systems sampling air from the common inlet. H₂O vapor was measured on the top of the ship mast in the front. The OH and HO₂ detection units were placed on the prow to allow for inlets with residence times less than 10 ms.
- 85 The Kommandor Iona left Malta in late June 2017 traversing the Mediterranean Basin, the Suez Canal and the Northern Red Sea. A 3 day stop over at KAUST University (Saudi Arabia) was made from 11 July 2017 to 13 July 2017 before passing the Southern Red Sea area. On 17 July 2017, we briefly stopped at Djibouti port before passing the Gulf of Aden, the Arabian Sea

and the Gulf of Oman. Kuwait at the northern end of the Arabian Gulf marked the turning point of the ship cruise where, during a second 3-day stop-over, scientific staff was exchanged. The *Kommandor Iona* started the second leg on 03 August

90 2017 arriving in Toulon (France) in early September 2017 without any further stops. Figure 1 shows the ship's route subdivided into six different regimes.

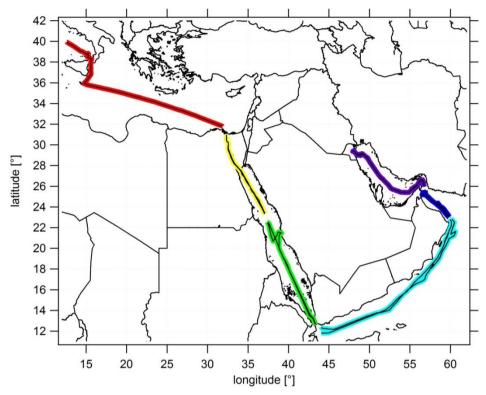


Figure 1: Ship cruises during both legs and color-coded subdivision into six different regimes. The following abbreviations will be used: AG for Arabian Gulf (purple), OG for Oman Gulf (dark blue), AS for Arabian Sea (blue), SRS for Southern Red Sea (green), NRS for Northern Red Sea (yellow), M for Mediterranean (red).

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To enhance the statistical significance of our results and due to comparable signatures of the NO_x and O₃ measurements in the northern part of the Red Sea, the Suez Gulf and the Suez Canal, we have combined these regions which are represented by the 'Northern Red Sea' (NRS). For the same reasons we have merged the Gulf of Aden with the Arabian Sea (AS). See supplementary Table ST1 for the range of latitudinal and longitudinal coordinates of the different regions and supplementary Table ST2 for a detailed day to day description of the route.

2.2 Measurements of nitrogen oxides during AQABA

Chemiluminescent detection of NO and NO₂ is a widely applied method to quantify mixing ratios from the ppm_v down to the

- 105 low ppt_v range (Nakamura et al., 2003; Pollack et al., 2011; Hosaynali Beygi et al., 2011; Reed et al., 2016). During AQABA we deployed a compact, robust and commercially available two-channel chemiluminescence instrument CLD 790 SR (ECO Physics AG, Dürnten, Switzerland) that has been optimized for in situ field measurements during the last decade (Hosaynali Beygi et al., 2011). The measurement principle of the CLD is based on the addition of O₃ to NO to produce stoichiometric quantities of excited state NO₂^{*} that will emit an infrared photon ($\lambda > 600$ nm) forming the chemiluminescent detection
- 110 principle for NO (Drummond et al., 1985; Reed et al., 2016). Both channels feature an identical layout and were operated at a mass flow of 1.5 SLM during AQABA. One channel of the CLD (NO_c-channel) has additionally been equipped with a LED solid state photolytic converter (Droplet Measurement Techniques, Boulder, Colorado) installed upstream of the O₃ addition to selectively photolyze NO₂ to NO, which is subsequently measured. In this section, we will concentrate on modifications made prior to the campaign and especially on operational conditions of the photolytic converter during the campaign. Further
- 115 details on the measurement principle are described elsewhere (Pollack et al., 2011; Hosaynali Beygi et al., 2011; Reed et al., 2016).

During AQABA, the cylindrical photolytic converter (length 14 cm, volume ~ 0.079 l) was operated at a constant pressure of 95 hPa yielding a residence time of ~ 0.3 s. The photolytic NO₂ converter features a set of 200 UV LED units attached to each end of the converter. The emission profile of the UV LED units was characterized in laboratory measurements to peak at 398 nm with a Full Width at Half Maximum (FWHM) of 16 nm. The UV-induced positive bias in the NO₂-measurement due to

- 120 nm with a Full Width at Half Maximum (FWHM) of 16 nm. The UV-induced positive bias in the NO₂-measurement due to photolysis of BrONO₂, HONO, NO₃ and ClNO₂ to produce NO was estimated at 6.1 %, 2.8 %, 2.7 % and 1.2 %, respectively, based on the absorption cross sections from the MPI-Mainz UV/VIS Spectral Atlas of Gaseous Molecules (Keller-Rudek et al., 2013). These values represent upper limits for the interference of the respective NO_y compound as the respective molecular quantum yield was estimated conservatively at 1. Note that the values represent percent interferences if the interferent had the
- 125 same concentration as NO₂. Due to small daytime concentrations of these molecules in the MBL, a UV-induced bias was neglected for the observations in this study. To limit wall loss of NO₂, the inner cavity surface is made of PTFE (polytetrafluoroethylene), which may potentially provide a reservoir (via surface adsorption) for NO_y that can thermally dissociate to increase the background signal of the NO₂ measurement (Reed et al., 2016). The conversion efficiency K_e of the photolytic NO₂ conversion was estimated by gas phase titration (SYCOS K-GPT-DLR, ansyco, Karlsruhe, Germany) several
- 130 times before, during and after the campaign at (29.4 ± 0.9) % allowing the calculation of NO₂ concentrations by $[NO_2] = \frac{[NO_c] [NO]}{\kappa_e}$. To avoid chemical interferences due to adding ozone in excess during a gas phase titration, a small but not vanishing amount of NO has always been left unoxidized during gas phase titrations.

During AQABA, regular dry zero-air measurements as well as NO and NO₂ calibrations were performed autonomously over a 10 minute period every 6 hours to accurately quantify the instrumental background and to correct for sensitivity drifts. An

- 135 autonomous cycle of '2 min zero air measurements 2 min NO calibration 2 min zero air measurement 2 min NO₂ calibration 2 min zero air measurement' was implemented. Continuous flows NO and NO₂ calibration gases were added to the synthetic airflow or directed to a pump by switching solenoid valves. The NO calibration standard (1.954 \pm 0.039 ppm_v NO in N₂, Air Liquide, Germany) used during the campaign was compared to a primary standard (5.004 \pm 0.025) ppm_v (NPL, Teddington, UK) after the campaign yielding an effective NO mixing ratio of (2.060 \pm 0.057) ppm_v in the NO
- 140 calibration gas. Zero air measurements and NO calibrations were performed with a total flow of 3.44 SLM achieving an overflow of 0.44 SLM to guarantee ambient air free standard measurements. The calibration gas was added at 4.5 sccm to the zero air flow. During AQABA, NO calibrations at 2.5 ppb_v were achieved. During the first leg of the campaign, zero air was sampled from a bottle (Westfalen AG, Germany), whereas during the second leg zero air was generated from a zero air generator (Air Purifier CAP 180, acuraLine). Zero air measurements generated with the zero air generator were statistically
- 145 not significantly different from those achieved by a bottle. To correctly account for the photomultiplier background and chemical interferences due to reactions of ozone with ambient alkenes additional pre-chamber measurements were performed every 5 minutes as well as at the beginning of zero air measurements and calibrations for 25 s each. This correction is removing a large fraction of the interference signal from alkenes. However, in regions where alkene concentrations are strongly varying in time and magnitude, the CLD is prone to enhanced backgrounds due to the interference of alkenes with ozone in the
- 150 instrument. A schematic setup of the two-channel CLD instrument is given in Figure 2.

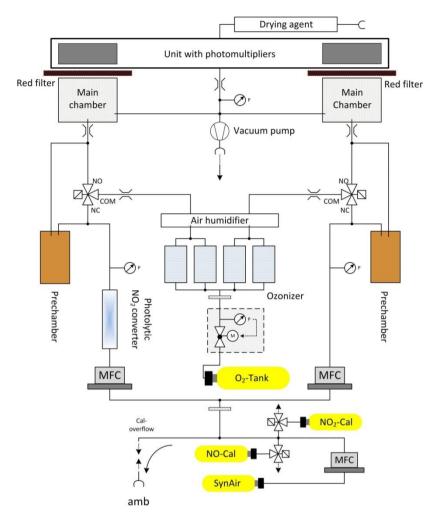


Figure 2: Schematic setup of the two channel CLD instrument in the configuration used during AQABA. NO and NO₂ calibration gases were running continuously and were added to the zero airflow by switching the respective solenoid valves.

155 The total measurement uncertainty (TMU) in NO has been calculated at 6 % at an integration time of 5 minutes and a confidence level of 1σ by adding the precision (5 %) and the error of the calibration gas mixture (3 %) in quadrature: TMU([NO]) = √(5 %)² + (3 %)² ≈ 6 %. Note that the precision is calculated from the reproducibility of all in-field calibrations, which is mainly determined by drifts in the detector sensitivity rather than by statistical Poisson noise of the measured signal. The limit of detection in the NO channel was estimated as the full width at half maximum of the frequency distribution of all zero air measurements obtained during the campaign to be 9 ppt_v at a 5 min integration time and a confidence level of 1σ. The TMU for NO₂ has been estimated as the largest error possible of the statistically dependent variables NO and NO_c.

$$\mathsf{TMU}([\mathsf{NO}_2]) = \frac{1}{[\mathsf{NO}_2]} \cdot \left(\left| \frac{\Delta[\mathsf{NO}_c]}{K_e} \right| + \left| \frac{\Delta[\mathsf{NO}]}{K_e} \right| + \left| \frac{\Delta K_e \cdot ([\mathsf{NO}_c] - [\mathsf{NO}])}{K_e^2} \right| \right)$$

Note that the total measurement uncertainty of the NO_c-channel data has also been calculated at 6 % at an integration time of

- 165 5 minutes and a confidence level of 1σ by adding the precision and the error of the calibration gas mixture in quadrature. Over the course of the campaign the median and the average relative uncertainty of NO₂ are 13 % and 23 %, respectively. The relative uncertainty in NO₂ has been estimated as a conservative upper limit at 23 % as the average of the relative uncertainties of all data points obtained during AQABA. As the zero air measurements in the NO₂ channel produced an increased background affected by memory effects after exposure to high NO_x levels e.g. during measurements of stack emissions, the
- 170 NO₂ raw data were initially processed without converter background subtraction. As we therefore expect the CLD NO₂ data to be offset due to not being initially background corrected, the converter background was estimated at 112 ppt_v from the centre of a Gaussian fit representing the difference of 1-minute averaged CLD NO₂ and concurrent cavity ring-down spectroscopy (CRDS) NO₂ measurements for data points below 10 ppb_v. Setting the threshold for calculating the difference of the two concurrent data sets to 10 ppb_v is somewhat arbitrary, however, changing this limit to 5 ppb_v or 20 ppb_v does not significantly
- 175 vary the estimated offset of the CLD NO₂ data. The offset correction of 112 ppt_v was taken as the ultimate absolute measurement uncertainty of the CLD NO₂ measurement. Further corrections of to the final CLD data include residence time corrections as well as corrections for NO and O₃ losses and the subsequent formation of NO₂ in the sampling line (Ryerson et al., 2000). Both NO and NO₂ CLD data have also been corrected for nonlinearities for concentrations higher than 55 ppb_v, as experienced during probing of stack emissions.

180 **2.3 Further measurements used in this study**

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An extensive set of concurrent measurements providing mixing ratios of O₃, NO₂, HCHO, OH, HO₂, absolute humidity and actinic flux, temperature and pressure data obtained during AQABA was used in this study. Ozone was measured with an absorption photometer (Model 202 Ozone Monitor, 2B Technologies, Boulder, Colorado) based on the well-established absorption of the mercury line in the Hartley band at 254 nm (Viallon et al., 2015). Eliminating water and particle interferences during sampling was achieved via sampling through a nafion tube and a Teflon filter. The ozone monitor was zeroed ten times during the campaign. NO₂ was further measured by cavity ring-down spectroscopy (Sobanski et al., 2016) and used for correcting the instrumental background of the CLD NO₂ data, as described above (the correction was taken as the ultimate

- absolute measurement uncertainty in the CLD NO₂ data). Note that in this study we will use the NO₂ CLD data rather than the NO₂ CRDS data as the temporal coverage of the CLD NO₂ data over the course of the campaign is about 60 % compared to about 35 % for the cavity ring-down measurement. Formaldehyde (HCHO) was measured with an Aerolaser 4021 (AERO-
- LASER GmbH, Garmisch-Partenkirchen, Germany), which is a fully automatized monitor based on the Hantzsch technique (Kormann et al., 2003). H₂O measurements were obtained using a cavity ring-down spectroscopy monitor (PICARRO G2401, Santa Clara, California) supervised by Laboratoire des Sciences du Climat et de l'Environnement (LSCE) (Kwok et al., 2015).

Measurements of OH and HO₂ were performed with the custom-built $\underline{\mathbf{H}}$ ydr $\underline{\mathbf{O}}$ xyl $\underline{\mathbf{R}}$ adical measurement $\underline{\mathbf{U}}$ nit based on 195 fluorescence $\underline{\mathbf{S}}$ pectroscopy (HORUS) instrument based on laser-induced fluorescence (LIF) spectroscopy of the OH molecule

- and NO titration of HO₂ to OH followed by LIF spectroscopy detection of the OH molecule (Martinez et al., 2010; Regelin et al., 2013). HO₂ data used in this study is still preliminary due to not yet corrected interference of organic peroxy radicals RO₂. The largest uncertainty due to interference by contribution of RO₂ is 7 % or 3 ppt_v whichever is higher. The 1 sigma accuracy of both OH and HO₂ is 20 %. The uncertainty in the OH data is here estimated as the 1 sigma accuracy of the data set at 20 %,
- whereas the uncertainty in HO₂ is estimated at $\sqrt{20 \%^2 + 7 \%^2} \approx 21 \%$. As HO₂ data are preliminary in this study, they will be called HO₂*. Wavelength resolved down-welling actinic flux was measured with a spectral radiometer (model CCD Spectroradiometer 85237). The *j*-values for NO₂ and O₃ were not corrected for upwelling UV radiation and were estimated to have a ~ 10 % measurement uncertainty (Meusel et al., 2016). The radiometer was installed 10 m above sea level, respectively 5 m above the front deck surface. Decreases in sensitivity due to sensor contamination with e.g. sea-spray were corrected with
- a linear interpolation between two (daily) cleaning events. Temperature and pressure measurements were performed with the Shipborne <u>Eu</u>ropean <u>C</u>ommon <u>A</u>utomatic <u>W</u>eather <u>S</u>tation (EUCAWS), a weather station specifically designed for ships. The weather station incorporates sensors, processing units, satellite positioning and communication systems in one device and is implemented and coordinated by the European National Meteorological Service EUMETNET. Table 1 lists the measurement methods and the TMU for each observation.
- 210 Table 1: List of observations and gas phase measurements during AQABA. The relative uncertainty associated with each data set is given. Note that for OH and HO₂*, the relative uncertainty is estimated as the 1σ accuracy (and the 7 % interference of RO₂ in the case of HO₂* data). In addition, a reference of the measurement operability are given.

Molecule	Method	Associated relative uncertainty	References
NO	chemiluminescence	6 %	Hosaynali Beygi et al., 2011
NO ₂	photolysis-chemiluminescence	23 %	Hosaynali Beygi et al., 2011
NO ₂	cavity ring-down spectroscopy	7 %	Sobanski et al., 2016
O ₃	UV absorbance	2 %	Viallon et al., 2015
ОН	LIF	20 %	Martinez et al., 2010
HO ₂ *	NO titration / LIF	21 %	Martinez et al., 2010
НСНО	Hantzsch technique	13 %	Kormann et al., 2003
H ₂ O	cavity ring-down spectroscopy	5 %	Kwok et al., 2015
actinic flux	spectral radiometer	10 %	Meusel et al., 2016

The Kommandor Iona Research and Survey Vessel sailed whenever possible with the wind coming from the bow to avoid

- 215 contamination by stack emissions. However, based on the relative wind direction, the variability in NO as well as the temporal evolution of NO_x , SO_2 , and O_3 sections of data in which the air mass was contaminated by the ship's stack were identified. All data used here to calculate (HO_2+RO_2) and NOPR have been filtered to remove contaminated air masses. Altogether, 21 % of the sampling time was potentially contaminated by the ship exhaust of the KI of which 87 % occurred on the first leg. During the second leg the ship sailed against the wind and most of the data was free of stack contamination. Our analysis is based on
- 220 a 5-minute running mean for each data set, whereby only averages that have been calculated at a temporal coverage greater than 30 % have been used. A time series of the NO, NO₂ (both CLD), O₃, OH, HO₂* and $j(NO_2)$ measurements is given in the supplementary Figures S2 and S3.

NO and NO₂ were measured from 03 July 2017 to 31 August 2017, O_3 was measured from 22 June 2017 to 01 September 2017, HCHO from 01 July 2017 to 31 August 2017 and OH and HO₂* from 18 July 2017 to 31 August 2017. For the analysis

of peroxy radicals RO_2 and NOPR around the Arabian Peninsula we have removed data measured during the stop-overs in Jeddah (11 July to 13 July), Kuwait (31 July to 03 August) and during bunkering at Fujairah City (06 August, 07:00 – 15:00 UTC). Due to HO_x data being available from 18 July 2017 onward, we have limited the net ozone production analysis to the period after this date.

230 2.4 Methods

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The so-called NO_x-O₃-null cycle represents a rapid daytime cycling between NO, NO₂ and O₃. Solar UV radiation photolyzes NO₂ to NO and O(³P) (R1) which will reform O₃ in the subsequent reaction with molecular oxygen O₂ (R2) (Leighton, 1961). NO and O₃ react to form NO₂ and O₂ (R3). R1, R2 and R3 constitute a so called null cycle which establishes photostationary steady state (PSS) for both NO_x and O₃ in mid latitudes during noon time on a time scale of ~100 s (Thornton et al., 2002; Mannschreck et al., 2004).

 $NO_2 + h\nu (\lambda < 424 \text{ nm}) \rightarrow NO + O(^3P)$ (R1)

$$O_2 + O(^3P) + M \to O_3 + M$$
 (R2)

$$NO + O_3 \rightarrow NO_2 + O_2 \tag{R3}$$

Under the assumption of PSS, the Leighton Ratio φ is unity (Leighton, 1961)

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$$\varphi = \frac{j(\text{NO}_2) \cdot [\text{NO}_2]}{k_{\text{NO}+\text{O}_3} \cdot [\text{NO}][\text{O}_3]} = 1$$
 (1)

with $j(NO_2)$ being the NO₂ photolysis rate [s⁻¹]. In low NO_x environments (< 100 ppt_v) previous studies have indicated that further NO oxidizing trace gases such as peroxy radicals (HO₂, RO₂) and halogen monoxides (XO) may result in a deviation from unity (Nakamura et al., 2003; Hosaynali Beygi et al., 2011; Reed et al., 2016).

$$NO + HO_2 \rightarrow NO_2 + OH \tag{R4}$$

$$245 \quad NO + RO_2 \rightarrow NO_2 + RO \tag{R5}$$

$$NO + XO \rightarrow NO_2 + X$$
 (R6)

Deviations from expected NO/NO₂-ratios at low NO_x generally refer to missing oxidants converting NO to NO₂ (Hosaynali Beygi et al., 2011; Reed et al., 2016) or to a measurement error due to an instrumental background or a positive interference from thermal labile NO_x reservoir species (Reed et al., 2016; Silvern et al., 2018). In the present study we include HO₂ and R_iO_2 into the production term for NO₂.

$$j(NO_2) \cdot [NO_2] = k_{NO+O_3} \cdot [NO][O_3] + k_{NO+HO_2} \cdot [NO][HO_2] + [NO] \cdot \sum_i k_{NO+R_iO_2} \cdot [R_iO_2]$$
(2)

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Assuming that the temperature-dependent rate coefficient for the reaction of each particular peroxy radical R_iO_2 with NO equals the rate k_{NO+HO_2} for Reaction R4 (Hauglustaine et al., 1996; Cantrell et al., 1997; Thornton et al., 2002), we can combine the sum of all organic peroxy radicals R_iO_2 to the entity RO₂. The sum of HO₂ and RO₂ can be estimated using the steady state equation

$$[HO_2] + [RO_2] = \frac{j(NO_2) \cdot [NO_2] - k_{NO+O_3} \cdot [NO][O_3]}{k_{NO+HO_2} \cdot [NO]}.$$
(3)

However, the steady state assumption is not valid if the sampled air parcel is affected by fresh emissions or fast changes in the actinic flux (Thornton et al., 2002). After sampling a fresh emission e.g. a ship plume, for which NO_x went up typically to values of several tens of ppb_v with simultaneous titration in O₃, we assume that PSS is re-established on a time scale of 2 minutes (Thornton et al., 2002; Mannschreck et al., 2004). To best approximate PSS in our analysis we have restricted the estimation of RO₂ on time frames \pm 2 h around noontime for which we expect the smallest relative changes in the actinic flux. Noontime for each day was determined as the centre of a Gaussian fit that was applied to the actinic flux data. We applied a Gaussian Fit to the actinic flux data as this fitting method is sufficient to estimate the centre of the diurnal actinic flux. To further limit the effect of periods for which PSS is not fulfilled, we use the median instead of the average that is often

265 disproportionately biased by strong NO_x sources nearby. See supplementary Tables ST3, ST5 and ST7 for detailed statistics and a further motivation on regional averages and median values. See supplementary Figure S1 for a detailed illustration of the calculation of the fraction of the noontime integral. A further part of the analysis will be the investigation of NOPR. Ozone production is initiated by reactions that produce HO_x , for which primary production is from the photolysis of ozone, formaldehyde, nitrous acid (HONO) and hydrogen peroxide

(H₂O₂) (Thornton et al., 2002; Lu et al., 2010; Hens et al., 2014; Mallik et al., 2018). The production of ozone can be approximated by the rate of oxidation of NO with HO₂ and RO₂ to form NO₂ that will rapidly form O₃ (R1-R2) (Parrish et al., 1986; Thornton et al., 2002; Bozem et al., 2017).

$$P(O_3) = k_{\rm NO+HO_2}[\rm NO] \cdot ([\rm HO_2] + [\rm RO_2])$$
(4)

Photochemical O₃ loss is mainly due to photolysis ($\lambda < 340$ nm) in the presence of water vapor and the reactions of ozone with OH and HO₂ (Bozem et al., 2017).

 $0_3 + h\nu \ (\lambda < 340 \text{ nm}) \rightarrow 0_2 + 0(1D)$ (R7.1)

$$D(^{1}D) + H_{2}O \rightarrow 20H$$
 (R7.2)

$$0_3 + 0H \rightarrow H0_2 + 0_2 \tag{R8}$$

$$0_3 + HO_2 \rightarrow 0H + 2O_2 \tag{R9}$$

280 α , the fraction of O(¹D) that reacts with H₂O

$$\alpha = \frac{k_{O(1D) + H_2O}[H_2O]}{k_{O(1D) + H_2O}[H_2O] + k_{O(1D) + M}[M]}$$
(5)

was (10.6 \pm 2.2) % during AQABA with a quasi linear dependence on water concentrations. The error in α is mainly determined by the error of H₂O at 5 %. Furthermore, ozone is lost due to reactions with alkenes (R12) and halogen radicals (R13).

285
$$O_3$$
 + alkenes \rightarrow radicals (R12)

$$0_3 + X \to 0_2 + X0 \tag{R13}$$

We find that the loss rate is dominated by the photolysis of ozone with subsequent reaction of $O(^{1}D)$ with $H_{2}O$, was 60 - 80% of the total loss rate, followed by the reaction of O_{3} with HO_{2} , which makes up 10 - 30 % (note that the uncertainty in HO_{2} radical concentrations mentioned above has no significant influence on the total O_{3} loss rate, due to its small contribution). The remaining fraction (10-30 %) is due to the reaction of O_{3} with OH. The reaction of ozone with ethene is on average 0.005 - 0.01 ppb_v h⁻¹ and therefore generally less than 2 % of the total ozone loss rate (Bourtsoukidis et al., 2019). The reaction of O_{3} with all alkenes will hence be neglected. Halogen radicals were not measured during AQABA and will not be incorporated into our study. Based on oxidative pairs, Bourtsoukidis et al. (2019) have classified the majority of their samples collected during AQABA by an OH/Cl-ratio of 200:1. As measured daytime OH concentrations were of the order of $5 \cdot 10^{6}$ molecule

295 cm⁻³, the estimate would yield a Cl concentration of $2.5 \cdot 10^4$ molecule cm⁻³, which would decrease the estimated diurnal net

ozone production rates by roughly $0.2 \text{ ppb}_v \text{ day}^{-1}$ over the Arabian Sea and at most $0.6 \text{ ppb}_v \text{ day}^{-1}$ over the other regions, which does not substantially alter the here presented results. The noontime chemical ozone loss rate can be summarized by

$$L(0_3) = [0_3] \cdot (\alpha \cdot j(0^1 \text{D}) + k_{\text{OH}+0_3} \cdot [\text{OH}] + k_{\text{H}0_2+0_3} \cdot [\text{H}0_2]).$$
(6)

NOPR presented in this study is finally calculated as the difference of Eq. 4 and Eq. 6.

300 NOPR =
$$k_{\text{NO}+\text{HO}_2}[\text{NO}] \cdot ([\text{HO}_2] + [\text{RO}_2]) - [\text{O}_3] \cdot (\alpha \cdot j(\text{O}^1\text{D}) + k_{\text{OH}+\text{O}_3}[\text{OH}] + k_{\text{HO}_2+\text{O}_3}[\text{HO}_2])$$
 (7)

Under the assumption of constant chemical composition for a given day, the NOPR is expected to have a diel cycle following the measured actinic flux. Hence integrating the estimated NOPR over the course of a day based on the particular fractional noontime integral of $j(NO_2)$ will yield a diurnal value for NOPR. A detailed calculation of the diurnal fractional integrals is given in the supplementary Figure S1. Note that all reaction rate constants used are from the IUPAC Task Force on Atmospheric Chemistry Chemical Kinetic Data Evaluation (Atkinson et al., 2004). Indications whether a chemical regime is NO_x -limited or VOC-limited can be derived from the ratio of HCHO to NO_2 . Former studies have derived HCHO/ NO_2 -ratios from satellite measurements to establish whether ozone production is NO_x -limited or VOCs-limited. The results indicate NO_x limitation for HCHO/ $NO_2 > 2$ and prevailing VOC-limitation for HCHO/ $NO_2 < 1$ (Duncan et al., 2010).

2.5 ECHAM/MESSy Atmospheric Chemistry (EMAC) model

305

- 310 EMAC is a 3D general circulation model that includes a variety of sub-models to describe numerous processes in the troposphere, their interaction with oceans and land surfaces and incorporates anthropogenic influences. Here we use the second development cycle of the Modular Earth Submodel System (MESSy2) (Jöckel et al., 2010) and ECHAM5 (Röckner et al., 2006) which is the fifth generation European Centre Hamburg general circulation model in the T106L31 resolution (corresponding to a quadratic grid of roughly 1.1° and 1.1°). The model has 31 vertical pressure levels and involves the complex
- 315 organic chemistry mechanism MOM (Mainz Organic Mechanism) as presented by Sander et al. (2019) that includes further developments of the version used by Lelieveld et al. (2016b). Here we use the lowest pressure level in a terrain following coordinates (equivalent to the surface level) and simulations of NO, NO₂, O₃, OH, HO₂, $j(NO_2)$ and $j(O^1D)$. The sum of peroxy radicals was estimated as the sum of all radicals R_iO₂ with less than four carbon atoms. Net ozone production based on data retrieved from EMAC was estimated as

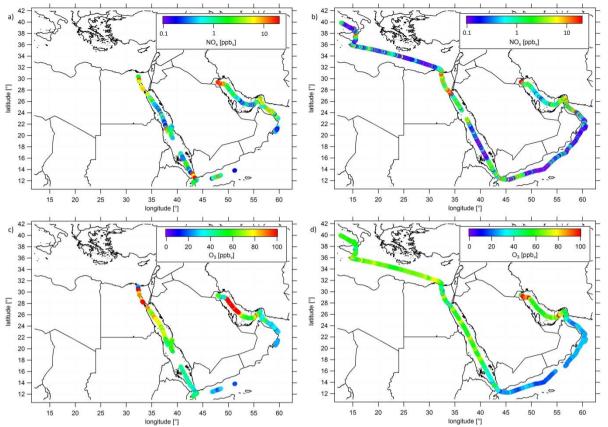
320 NOPR =
$$[NO] \cdot (k_{NO+HO_2}[HO_2] + \sum_i k_{NO+R_iO_2}[R_iO_2]) - [O_3] \cdot (\alpha \cdot j(O^1D) + k_{OH+O_3}[OH] + k_{HO_2+O_3}[HO_2]).$$
 (8)
A list of all included peroxy radicals R_iO_2 for the reaction with NO is given in the supplementary Table ST10.

3 Results and discussions

3.1 NO_x and O_3 in the MBL around the Arabian Peninsula

During AQABA NO_x mixing ratios varied over three orders of magnitude with lowest values of less than 50 ppt_v observed in
relatively pristine regions and highest values of more than 10 ppb_v found in the vicinity of areas with strong anthropogenic influence or nearby passing ships. Ozone mixing ratios ranged from values of less than 20 ppb_v, detected over the Arabian Sea, to more than 150 ppb_v during episodes of severe pollution. Figures 3a) and 3b) show distributions of NO_x measured during the first and second leg of the campaign (range from 0.1 ppb_v to 20 ppb_v) while Figure 3c) and 3d) show corresponding ozone mixing ratios covering a range from 20 ppb_v to 100 ppb_v, respectively. A classification of the different regions based on BoxWhisker-Plots, including the 25-75-percentile interval (box) and whiskers for the 10-90-percentile interval, is shown in Figure

- 4 and Figure 5 for NO_x and O₃, respectively. As average NO_x is often influenced by fresh, localized emissions, we have included the median (black bar) instead of the average in the Box-Whisker-Plot for NO_x, which is less sensitive to extreme values. For O₃, although the difference between median and mean is mostly negligible, we also use the median in Figure 5. NO_x and O₃ averages, medians, standard deviations, 1^{st} and 3^{rd} quantiles and the number of data points quantified per region
- are given in the supplementary Table ST3. See supplementary Figure S4 for OH and HO₂* mixing ratios around the Arabian Peninsula. Supplementary Figure S5 shows that absolute humidity observed during AQABA ranges from lowest values of less than 1 % observed in the Suez Golf during the first leg to about 3 % observed during both legs in the southeastern part of the Arabian Gulf and in the Strait of Hormuz. Although observing highest absolute humidity on both legs in the southeastern part of the Arabian Gulf, absolute humidity was very low on the first leg near Kuwait, where absolute humidity was about 1 %.
- 340 These air masses were brought from the Kuwait/Iraq area into the MBL of the Arabian Gulf on the first leg, whereas a change of wind direction for the second leg resulted in winds coming from Iran area with moister air. For the rest of the cruise, absolute humidity mixing ratio was about 1.5 % with variations being generally less than 0.5 %.



345 Figure 3: Ship cruises with color-scaled NO_x mixing ratios (logarithmic scale) a) during the first and b) the second leg and colorscaled O₃ mixing ratios (linear scale) c) during the first and d) during the second leg. Note that both NO_x and O₃ has been filtered for own stack contamination.

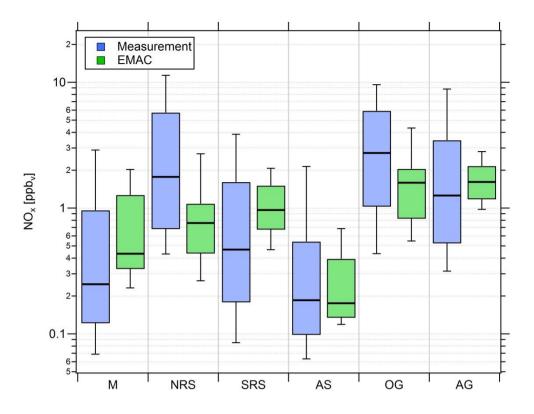


Figure 4: Comparison of measured (blue) and simulated (green) NO_x mixing ratios in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90percentiles.

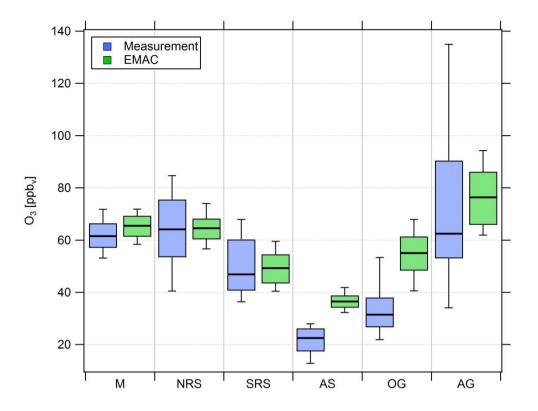


Figure 5: Comparison of measured (blue) and simulated (green) O₃ mixing ratios in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90-355 percentiles.

Overall, we find that NO_x mixing ratios over the Northern Red Sea, the Gulf of Oman and the Arabian Gulf are approximately one order of magnitude higher than in the other three regions (Southern Red Sea, Arabian Sea, Mediterranean). NO_x medians over the Arabian Gulf, the Northern Red Sea and the Gulf of Oman are 1.26 ppb_v, 1.76 ppb_v and 2.74 ppb_v, respectively. Lower
median NO_x mixing ratios were measured over the Southern Red Sea (0.46 ppb_v), the Mediterranean (0.25 ppb_v) and the Arabian Sea (0.19 ppb_v). With respect to observed O₃ mixing ratios, the Arabian Sea is the only region representing remote MBL conditions with lowest median and average O₃ of 21.5 ppb_v and 22.5 ppb_v respectively, followed by the Gulf of Oman where median and mean O₃ were 31.5 ppb_v and 34 ppb_v, respectively. The low O₃ mixing ratios over the Arabian Sea were accompanied by the smallest variability (whisker-interval: 15.1 ppb_v). Although observing highest NO_x over the Oman Gulf, O₃ observed over the Oman Gulf was amongst the lowest detected throughout the whole campaign, which can partly be explained the fact that high NO_x eventually leads to ozone destruction. The immediate vicinity of point sources in this region, which leads to higher NO_x (before it is lost by reaction with OH and deposition to the surface) and titration of O₃ (note the relatively low regional O₃ median of 31.5 ppb_v), may partly explain why NO_x was highest over the Gulf of Oman. However,

a significantly larger whisker-interval of observed ozone of 31.4 ppb_v over the Gulf of Oman indicates increasing amounts of

370 pollution and advection from the Arabian Gulf where extreme events of ozone were observed several times during the

campaign with maximum mixing ratios of up to 170 ppb_v when wind was coming from Kuwait/Iraq. Please note that during the second leg wind was coming from Iran (Pfannerstill et al., 2019). The whisker-interval over the Arabian Gulf was 100.9 ppb_v, more than six times higher than that over the Arabian Sea. Reasons for large variations of both NO_x and O₃ over the Arabian Gulf were a multitude of point sources as well as a change in the observed wind direction with air masses coming

- 375 from Iraq/Kuwait area during the first leg and air masses coming from Iran during the second leg (Pfannerstill et al., 2019). Over the Mediterranean, the Northern Red Sea and the Southern Red Sea, median ozone was 61.5 ppb_v, 64.2 ppb_v and 46.9 ppb_v, respectively. The whisker-intervals over the Northern Red Sea and the Southern Red Sea were 44.2 ppb_v and 31.6 ppb_v, respectively. Air masses over the Mediterranean were characterized as photochemically aged due to their impact by northerly winds (Etesians) which bring processed/oxidized air from eastern Europe (Turkey, Greece) to the Mediterranean area
- (Derstroff et al., 2017; Pfannerstill et al., 2019). This photochemical ageing/oxidation over the Mediterranean leads to a rather small whisker-interval of 18.7 ppb_v in ozone. In summary, median NO_x over the Oman Gulf was 56 % and 117 % higher than over the Northern Red Sea and the Arabian Gulf, respectively. However, the highest NO_x average was measured over the Northern Red Sea at 4.69 ppb_v, similar to the values observed over the Oman Gulf (4.16 ppb_v) and the Arabian Gulf (3.65 ppb_v). Note that highest NO_x mixing ratios over the Oman Gulf and over the Northern Red Sea are not always associated with
- 385 high O_3 mixing ratios. We find that average ozone was highest over the Arabian Gulf with 74 ppb_v followed by the Northern Red Sea region (63.4 ppb_v). The average ozone mixing ratio over the Oman Gulf was 34 ppb_v, which corresponds to 46 % of the value observed over the Arabian Gulf. Photochemically aged air masses over the Mediterranean Basin show an ozone average of 61.6 ppb_v and air masses encountered over the Northern Red Sea (O₃ median of 64.2 ppb_v, O₃ average of 63.4 ppb_v) are comparable to the Arabian Gulf.
- Due to a number of large pollution sources in the region around the Arabian Peninsula such as passing ships, highly urbanized areas as well as on- and off-shore petrochemical processing, NO_x levels were rarely as low as those found in remote locations such as over the South Atlantic (Fischer et al., 2015) where NO_x levels may be under 20 ppt_v. Apart for a few occasions where NO_x was below 50 ppt_v for short periods (Arabian Sea, the Southern Red Sea and the Mediterranean), NO_x levels during AQABA generally ranged from 100 ppt_v up to several ppb_v. The campaign NO_x median of 0.65 ppb_v and mean value of (2.51 ± 5.84) ppb_v is comparable to urban sites (Kleinman et al., 2005). A detailed emission density analysis performed by Johansson et al. (2017) shows that NO_x level in the MBL around the Peninsula (Johansson et al., 2017; Pfannerstill et al., 2019). O₃ mixing ratios measured during AQABA were also very variable with O₃ mixing ratios ranging between less than 20 ppb_v in the remote
- MBL (Fischer et al., 2015) to 60-70 ppbv in the Mediterranean (consistent with previous ship-based measurements in the region
 (Kouvarakis et al., 2002) and as high as 150 ppbv measured over the Arabian Gulf region. The latter are consistent with O₃ mixing ratios reported from regions influenced by oil and gas processing (Pfannerstill et al., 2019) and shipping lanes such as the Houston Ship Channel (Mazzuca et al., 2016).

Figure 4 also shows that the general trend for NO_x mixing ratios in the different regions is widely reproduced by the EMAC model. We find that the median NO_x (model)/ NO_x (measurement)-ratio of all five minute averaged data points of the whole

- 405 campaign is 0.91, indicating that the model underestimates NO_x by roughly 10 %. The average ratio and its standard deviation are significantly larger at 2.57 and 5.71, respectively, indicating that single modeled data points strongly exceed the measurements, especially during periods of low in situ NO_x (see supplementary Figure S6). Particularly over the Arabian Sea and the Southern Red Sea, the model generally simulates NO_x mixing ratios higher than 100 and 200 ppt_v, respectively while the measurements indicate mixing ratios of less than 50 ppt_v for certain periods. Furthermore, as expected, the model is not
- 410 able to reproduce point sources such as passing ships for which we observe a significant underestimation of the measured NO_x . For ozone we find that the median $O_3(model)/O_3(measurement)$ -ratio throughout the campaign is 1.23, indicating that over the course of the campaign the model overestimates O_3 by about 23 %. This could partly be related to the same limitation, i.e. the inability of the model to resolve point sources in which O_3 is locally reduced due to titration by NO. While the model is in rather good agreement with the measurements over the Mediterranean, the Northern Red Sea and Southern Red Sea, large
- 415 deviations are found over the Arabian Sea and the Oman Gulf, where the model overestimation with respect to the regional median is 63 % and 75 %, respectively. A possible explanation for the overestimation of both ozone and NO_x in pristine regions such as over the Arabian Sea and the Oman Gulf could be related to the model resolution of 1.1° x 1.1°. Interpolation of model simulations along the *Kommandor Iona* ship track close to the coast at this resolution will most likely incorporate contributions from nearby land areas, affected by anthropogenic emissions. See supplementary Table ST3 and Table ST4 for further
- 420 information and Figure S6 and S7 for additional scatterplots of measured and simulated regional median NO_x and O_3 , respectively.

3.2 Estimation of (HO₂+RO₂) around the Arabian Peninsula

Noontime (HO₂+RO₂) was estimated based on Eq. 3. As the steady state assumption will not hold for air masses originating from fresh emissions (times to acquire steady state estimated from the inverse sum of the loss and production terms for NO₂ typically ranged from 1-2 minutes during AQABA) and for fast changes in the actinic flux, we have calculated Box-Whisker-Plots for \pm 2 h around noontime for which we expect relatively minor changes in the actinic flux (Figure 6). The noontime of each day was approximated by applying a Gaussian fit routine to the measured *j*(NO₂) values whereas *j*(NO₂) values being less than 10⁻³ s⁻¹ were neglected. Due to the availability of OH and HO₂* data from 18 July 2017 onwards, we have limited the

analysis to this period. Note that there are no noontime (HO₂+RO₂) estimates from 18 July to 21 July due to contamination by
the ship exhaust and on 24 August 2017 due to missing data. The black bar in Figure 6 indicates the median value, with the
Box-interval marking the 25- and 75-percentile and the whisker showing the 10- and 90-percentile. Figure 7 shows summarized
regional trends of the (HO₂+RO₂) estimates for measured and simulated data.

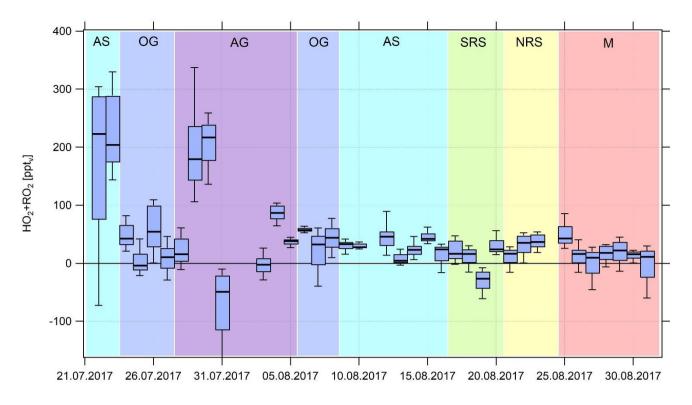


Figure 6: Timeline of median (HO₂+RO₂) noontime estimates from 22 July to 31 August 2017. Due to contamination by the ship exhaust itself, there is no data from 18 July to 21 July 2017. See annotations for the classification of the different regions.

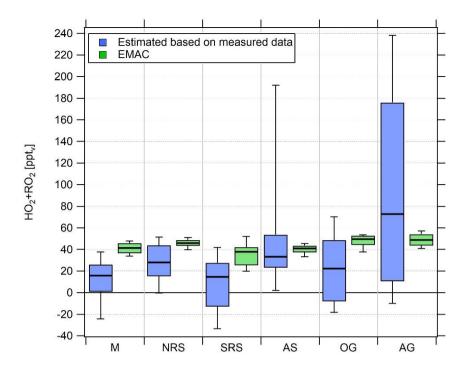


Figure 7: Comparison of Box-Whisker-Plots of the regional estimated noontime (HO_2+RO_2) median based on measured data and simulated (HO_2+RO_2) data for the period from 18 July 2017 onwards.

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The relative uncertainty $R([HO_2] + [RO_2])$ associated with the (HO_2+RO_2) estimate has been calculated by error propagation of Eq. 3.

$$R([HO_2] + [RO_2]) =$$

$$\frac{1}{[\text{HO}_{2}] + [\text{RO}_{2}]} \cdot \sqrt{\left(\frac{\Delta j(\text{NO}_{2}) \cdot [\text{NO}_{2}]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot j(\text{NO}_{2})]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{O}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{NO}]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{O}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{NO}]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{O}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{NO}]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{O}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{NO}]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}] \cdot k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}] \cdot k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{HO}_{2}} \cdot [\text{NO}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}] \cdot k_{\text{NO}+\text{O}_{3}}}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{2}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{NO}_{3}] \cdot k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}{k_{\text{NO}+\text{O}_{3}} \cdot [\text{O}_{3}]}\right)^{2} + \left(\frac{\Delta [\text{O}_{3}] \cdot$$

445 Over the course of the campaign, the median relative (HO₂+RO₂) uncertainty is 74 %. The average is 176 % and heavily biased by single data outliers and therefore not representative. The relative error associated with the (HO₂+RO₂) calculation is hence estimated at 74 %. Note that our calculation assumes that errors in the used rate coefficients are negligible.

We find median noontime (HO_2+RO_2) mixing ratios over the Mediterranean, the Northern Red Sea, the Southern Red Sea, the Arabian Sea and Oman Gulf of 16 ppt_v, 28 ppt_v, 15 ppt_v, 33 ppt_v and 22 ppt_v, respectively, with each respective 75-percentile

450 RO₂ being equal or less than 54 ppt_v. Only over the Arabian Gulf, the RO₂ estimate yields a median noontime mixing ratio of

73 ppt_v accompanied by the largest variations in the box-interval of the whole campaign. While the box-interval of the (HO_2+RO_2) estimate in the other regions is 25-57 ppt_v, the box-interval over the Arabian Gulf is significantly higher at 165 ppt_v. Negative values for all regions are regularly found in the vicinity of fresh emissions and air masses not in photochemical equilibrium. The elevated 90-percentile over the Arabian Sea is due to high (HO_2+RO_2) estimates during the first leg on 22 and 23 July.

Estimated (HO₂+RO₂) mixing ratios based on measured tracer data are in general agreement with previous studies performed in marine boundary layer environments which report maximum mixing ratios between 30 and 55 ppt_v around noontime (Hernandez et al., 2001). As peroxy radicals are short-lived molecules generated from the oxidation of VOCs, enhanced (HO₂+RO₂) concentrations observed over the Arabian Gulf are most likely due to high VOC emissions from intense oil and gas activities in the region (Bourtsoukidis et al., 2019; Pfannerstill et al., 2019). However high HO₂ and RO₂ can also occur in aged air masses with low NO_x and VOCs but still significant O₃ (and perhaps HCHO whose photolysis would then yield peroxy radicals). Bourtsoukidis et al. report that spatial volume mixing ratios of ethane and propane over the Arabian Gulf were about a factor of 10-15 times higher than over the Arabian Sea and the Southern Red Sea (Bourtsoukidis et al., 2019). We find that the median noontime (HO₂+RO₂)(measurement estimate)/HO₂(measurement)-ratio throughout the whole campaign is 1.88.
Note that during single days, HO₂ may be higher than the (HO₂+RO₂) estimate, which is within the uncertainty of the (HO₂+RO₂) estimate.

EMAC modelled, median noontime (HO₂+RO₂) mixing ratios estimated as the sum of simulated HO₂ and all simulated peroxy radicals with less than four carbon molecules are 41 ppt_v, 46 ppt_v, 38 ppt_v, 41 ppt_v, 50 ppt_v and 49 ppt_v over the Mediterranean, the Northern Red Sea, the Southern Red Sea, the Arabian Sea, the Oman Gulf and the Arabian Gulf, respectively. The

470 observation based (HO₂+RO₂) estimate yields 16 ppt_v, 28 ppt_v, 15 ppt_v, 33 ppt_v, 22 ppt_v and 73 ppt_v respectively. We find that the median point by point (HO₂+RO₂)(model)/(HO₂+RO₂)(measurement estimate)-ratio from 18 July onward is 1.05 so that, on average, the model overestimates the measurement by 5 %. Please note that the observational variability is much higher than the modeled one and that the median of 1.05 is accompanied by a larger average (1.84) and a large variability (42.51). See supplementary Table ST5 and ST6 for further information and Figure S8 for an additional scatterplot of measured and simulated regional median (HO₂+RO₂).

3.3 Net ozone production rates around the Arabian Peninsula

In the following, net ozone production rates (at noon) are calculated based on Eq. 7 for the different regions. These noontime values are scaled to diurnal production rates (Figure 8). As photochemical net ozone destruction is in good approximation linear with actinic flux $j(NO_2)$ and as on average (46.1 ± 2.8) % of the total $j(NO_2)$ occurred ± 2h around noon, the median noontime NOPR estimate was multiplied by $4/0.461 \approx 8.68$ to obtain a diurnal value. The error in the total actinic flux located ± 2h around noon is estimated from the standard deviation of the best estimate of 0.461 at $\Delta s \approx 6$ %. Due to contamination

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by the own ship exhaust and due to the availability of OH and HO₂* data only from 18 July 2017 onwards, we have limited the analysis to the period from 22 July 2017 to 31 August 2017. A comparison of NOPR estimated based on measured and simulated data for the different regions is shown in Figure 9. A break-down of the different terms of Eq. 7 in the six regions is included in the supplementary Figures S10-S13.

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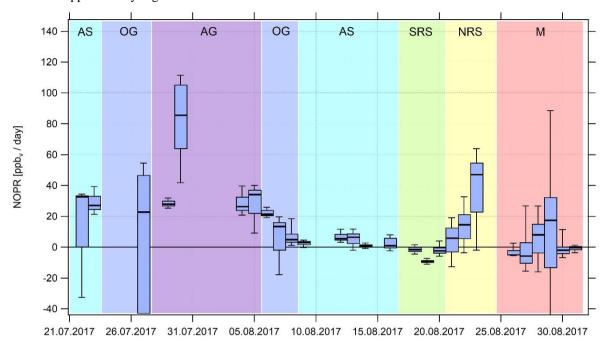


Figure 8: Timeline of the diurnal NOPR from 22 July to 31 August 2017. NOPR calculations are limited to the time period from 22 July onwards due to missing HO_x data and contamination by the ship exhaust itself before this period. See annotations for the classification of the different regions.

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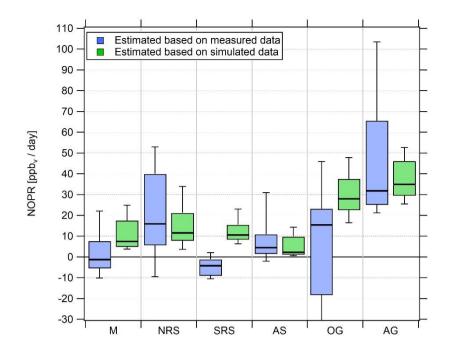


Figure 9: Diurnal net ozone production rates in the different regions. Related to the magnitude of pollution sources, the lower whisker of the NOPR estimate over the Oman Gulf is -324 ppb day⁻¹.

495 The relative uncertainty associated with the NOPR estimate has been calculated by error propagation of Eq. 7.

$$R(\text{NOPR}) = \frac{1}{\text{NOPR}} \cdot \left\{ \begin{pmatrix} k_{\text{NO+HO}_2} \cdot [\text{NO}] \cdot \Delta([\text{HO}_2] + [\text{RO}_2]) \end{pmatrix}^2 + \begin{pmatrix} k_{\text{NO+HO}_2} \cdot \Delta[\text{NO}] \cdot ([\text{HO}_2] + [\text{RO}_2]) \end{pmatrix}^2 + \\ \left(\Delta[0_3] \cdot \left(\alpha \cdot j(0^{1}\text{D}) + k_{0\text{H+O}_3} \cdot [\text{OH}] + k_{\text{HO}_2+\text{O}_3} \cdot [\text{HO}_2] \right) \end{pmatrix}^2 + \\ \left(\Delta j(0^{1}\text{D}) \cdot [0_3] \cdot \alpha)^2 + (j(0^{1}\text{D}) \cdot [0_3] \cdot \Delta \alpha)^2 + \\ \left(\Delta[\text{OH}] \cdot k_{0\text{H+O}_3} \cdot [0_3] \right)^2 + \left(\Delta[\text{HO}_2] \cdot k_{\text{HO}_2+\text{O}_3} \cdot [0_3] \right)^2 \right\}$$

Incorporating a relative error of 74 % associated with (HO_2+RO_2) , the median of the relative NOPR error of all data points obtained during AQABA is 91 %. The average relative uncertainty of NOPR is 21 % and strongly biased by single data outliers, which are in the case of NOPR significantly negative (due to fresh emissions and titration of O₃ by NO). Again the median is a more representative measure for the general uncertainty associated with the NOPR calculations. The relative error associated with the NOPR estimates based on measured data is hence estimated at 91 %.

Over the Mediterranean and the Southern Red Sea, NOPR values do not significantly deviate from zero (production equals

505 loss) within the atmospheric variability. The best estimate indicates slight net ozone destruction for the Mediterranean and Southern Red Sea (- 1 ppb day⁻¹) and (- 4 ppb day⁻¹) respectively, and slight net production for the Arabian Sea (5 ppb day⁻¹), which is significantly positive within the variability of the box-interval. Variations in NOPR calculated as the width of the 2575-percentile-box yield comparable values of 9-11 ppb day⁻¹ for these three regions. Substantial net ozone production was inferred over the Oman Gulf, the Northern Red Sea, and the Arabian Gulf with the median values being 16 ppb day⁻¹, 16 ppb

510 day⁻¹ and 32 ppb day⁻¹, respectively. Especially over the Red Sea we find a strong latitudinal gradient in net ozone production rates with higher values towards the northern end, while slight net ozone destruction of -4 ppb day⁻¹ is reported over the southern part.

NOPR estimates for the Oman Gulf, the Northern Red Sea and the Arabian Gulf are comparable to results reported for dense traffic shipping routes such as the Houston Ship Channel with NOPR of a few tens of ppb h⁻¹ for periods of severe pollution (Zhou et al., 2014). Similar net ozone production rates have been reported for regions of Beijing in summer 2006 (Lu et al.,

515 (Zhou et al., 2014). Similar net ozone production rates have been reported for regions of Beijing in summer 2006 (Lu et al., 2010). For regions with low anthropogenic influence such as the Southern Red Sea and the Arabian Sea we estimate net ozone production that does not differ significantly from zero. This is due to the rather low NO_x mixing ratios in the clean marine boundary layer (Bozem et al., 2017). Note that we calculated net ozone destruction only for a few days over the Southern Red Sea and the Arabian Sea, indicating that the marine boundary layer around the Arabian Peninsula is rarely free from 520 anthropogenic influence owing to the multitude of on- and off-shore anthropogenic activities.

We find that model-calculated estimates of NOPR reproduce the trends observed for NOPR calculated from in situ measurements except over the Mediterranean and the Southern Red Sea. Although EMAC predicts high ozone levels over the Arabian Sea, it also reports the lowest NOPR in this region. On the other side, the large overestimation of the model-calculated estimate NOPR against the one based on measured tracer data over the Mediterranean and over the Southern Red Sea could be linked to NO_x being overestimated in the model in these regions. In the model, pollution emissions, especially over the Oman Gulf and the Arabian Gulf, seem to be averaged over a large (1.1° grid size) region. High background concentrations of ozone precursors hence contribute to net ozone production rates that compare to conditions observed in the Houston case (Zhou et al., 2014). Even in the more pristine regions such as over the Southern Red Sea and the Arabian Sea, the model is not able to reproduce net ozone destruction, which is consistent with the fact the ozone is generally too high and that NO_x levels below 0.1 ppb_v are not found in the model. See supplementary Table ST7 and ST8 for further information and supplementary Figure

S9 for an additional scatterplot of measured and simulated regional NOPR.

Measured OH and HO_2^* as well as (HO_2+RO_2) estimated based on measured data are generally underestimating the concurrent simulated data. Speaking in terms of absolute amounts, we find that the break-down loss and productions terms of Eq. 7 (NOPR) based on measured data are generally underestimating the results based on simulated data. The deviations between

535 measurement and model pretty much represent the differences observed in the noontime concentrations of the mentioned tracers. Largest deviations of the break-down loss terms, associated with reactions of O₃ with OH and HO₂, are found over the OG and AG, where also OH and HO₂ is significantly overestimated in the model. In the case of $j(0^1D) \cdot \alpha \cdot [0_3]$ a slight overestimation by the estimate based on simulated data compared to the estimate based on measured data is observed. This is due to simulated absolute humidity being slightly larger than the concurrent measured data. Also we find that the break-down 540 production term $k_{\text{NO+HO}_2} \cdot [\text{NO}] \cdot ([\text{HO}_2] + [\text{RO}_2])$ estimated based on simulated data is generally larger than the estimate based on measured data. This pretty much reflects that noontime (HO₂+RO₂) is overestimated in the model by a factor of 2, except for the Arabian Gulf where fair agreement is found.

3.4 VOC- and NO_x-sensitivity

- Ozone is photochemically formed when the precursors NO_x and VOCs are abundant in the presence of sunlight (Bozem et al., 2017; Jaffe et al., 2018). In order to determine whether a chemical system is NO_x- or VOC-limited or in a transition between those two regimes, one has to estimate the total amount of OH reactivity towards VOCs and towards NO_x. Therefore the VOC/NO_x-ratio is an important indicator of the behavior of NO_x, VOCs and O₃ in a system. Since it is not feasible to precisely define all ambient VOCs (could be thousands), formaldehyde mixing ratios have been used as a proxy for the OH reactivity towards VOCs since it is a short-lived oxidation product of many VOCs that is often positively correlated with peroxy radicals
 (Sillman et al., 1995; Duncan et al., 2010). Sillman et al. first used afternoon concentrations of indicator species such as HCHO and total reactive nitrogen (NO_y) to determine the sensitivity of ozone production to VOCs or NO_x (Sillman et al., 1995). Their approach was later successfully transferred to space-based satellite observations by using the ratio of tropospheric columns of HCHO and NO₂ to determine the sensitivity of ozone production (Martin et al., 2004). Here we use HCHO/NO₂-ratios (referred to as "Ratio") deduced by Duncan et al. as indicators for the sensitivity of ozone production to NO_x- and VOC-limitations in megacities in the United States with large amounts of anthropogenic NO_x and VOC emissions (Duncan et al., 2010). The Ratio
- is an indicator of surface photochemistry as most of the atmospheric column of HCHO and NO₂ is located in the planetary boundary layer (Duncan et al., 2010). Duncan et al. have derived NO_x-limited ozone production regimes for HCHO/NO₂ > 2 and VOC-limited ozone production for HCHO/NO₂ < 1 (Duncan et al., 2010). For 1 < HCHO/NO₂ < 2 both NO_x and VOC emission reductions may lead to a reduction in ozone. Figure 10 shows the Box-Whisker-Plot classification of the HCHO/NO₂-
- 560 ratio of the different regions during noontime.

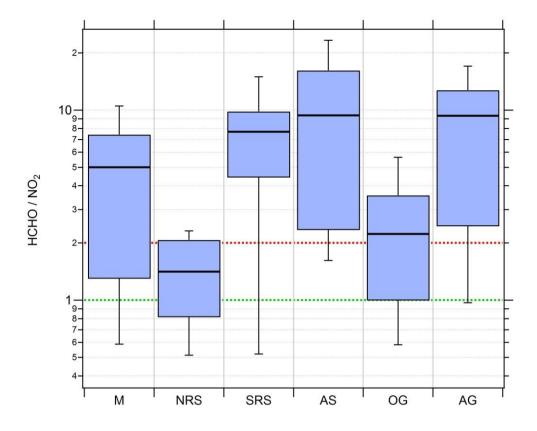


Figure 10: Box-Whisker-Plots of the HCHO/NO₂-ratio for the different regions with the black bar indicating the median value. Red (ratio = 2) and green (ratio = 1) lines indicate the limits for HCHO/NO₂ deduced by Duncan et al. (2010) for NO_x-limitation and VOC-limitation, respectively.

Median HCHO/NO₂-ratios of 5, 7.7, 9.4 and 9.3 over the Mediterranean, the Southern Red Sea, the Arabian Sea and the Arabian Gulf respectively indicate tendencies towards NO_x-limited regimes. In a previous study based on measured OH reactivity, Pfannerstill et al. classified these regions as being mostly in a transition between NO_x- and VOC-limitation, with a tendency towards NO_x-limitation (2019). Median HCHO/NO₂-ratios of 1.4 and 2.2 estimated over the Northern Red Sea and the Oman Gulf signify tendencies towards VOC-limitation. However, none of the medians of the six regions falls below the VOC-limit deduced by Duncan et al. (2010).

Over the Red Sea we find a latitudinal gradient in the HCHO/NO₂-ratio, similar to the gradients for NO_x and NOPR. Due to very low NO_x over the Southern Red Sea, O₃ production is NO_x-limited, changing into a more VOC-limited regime over the Northern Red Sea. Ozone production over the Mediterranean was classified as rather NO_x-limited, however partly being in the transition regime between NO_x- and VOC-limitation, which can be explained by measurements obtained on 29 August 2017 when laying at anchor in front of Malta with a multitude of (NO_x)-emissions from nearby situated vessels. Average noontime NO_x on that particular day was about three times as large as the regional average noontime NO_x observed over the whole

Mediterranean area. NO_x limitation is also inferred for the relatively clean Arabian Sea and the polluted Arabian Gulf 580 atmosphere. Note that a further increase in NO_x -emissions from shipping in the Arabian Gulf may initially lead to higher ozone production. However, a further increase in NO_x might eventually lead to a change from NO_x- to VOC-sensitivity and a decrease in ozone production for this region, as observed for the Oman Gulf (median HCHO/NO₂-ratio of 2.2 and average O_3 of 34 ppb_v). See supplementary Table ST9 for detailed statistics on regional HCHO/NO₂-ratios.

585 **4** Conclusion

In situ observations of NO, NO₂, O₃, HCHO, OH, HO₂, absolute humidity, actinic flux, temperature and pressure were carried out in the marine boundary layer around the Arabian Peninsula during the AQABA ship campaign from late June to early September 2017. Concentration ranges of both NO_x and O_3 clearly showed anthropogenic influence in the MBL. NO_x was highest over the Arabian Gulf, the Northern Red Sea and the Oman Gulf. Lowest NO_x was observed over the Arabian Sea and 590 over the Southern Red Sea during the second leg. O_3 mixing ratios were highest over the Arabian Gulf. We observed a latitudinal gradient in O₃ concentrations with higher values towards the northern part of the Red Sea. Although comparable O₃ averages were measured over the Northern Red Sea and over the Mediterranean, lower variability over the Mediterranean towards the end of August 2017 indicates photochemically more extensively aged air masses. The lowest regional O_3 mixing ratio average was detected over the Arabian Sea, which is broadly comparable to remote marine boundary layer conditions in

Noontime (HO_2+RO_2) estimates based on deviations from the Leighton Ratio yield median values around the Arabian Peninsula amount to 15 - 33 ppt_v for all regions except over the Arabian Gulf where the median is 73 ppt_v. The uncertainty due to the missing up-welling actinic flux portion is expected to be insignificant. Furthermore, we estimated noontime and diurnal NOPR based on Eq. 6 and the integral over the actinic flux. Highest diurnal NOPR were observed over the Oman Gulf, 600 the Northern Red Sea and the Arabian Gulf with median values of 16 ppb_v day⁻¹, 16 ppb_v day⁻¹ and 32 ppb_v day⁻¹, respectively, which is in agreement with previous studies that predicted net photochemical O_3 formation conditions in the region. Net ozone destruction was only observed for a few days with clean conditions over the Arabian Sea and the Southern Red Sea. Based on HCHO/NO₂-ratios our analysis suggests tendencies towards NO_x-limitation over the Mediterranean, the Southern Red Sea. the Arabian Sea and the Arabian Gulf and VOC-limitation over the Northern Red Sea and the Oman Gulf, which reproduces the trends observed by Pfannerstill et al. (2019).

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 NO_x results from the atmospheric chemistry – general circulation model EMAC underestimate the measurement data by 10 % whereas median modeled O₃ overestimates the measurement by 23 %, the latter being related to limitations in model resolution in coastal proximity and near shipping lanes. Although EMAC generally reproduces regional NO_x and O_3 medians, the scatter when comparing both data sets is large. NO_x is generally too low as it does not resolve local point sources and too high for

⁵⁹⁵ the Northern Hemisphere.

610 clean regions. Lowest NO_x of less than 0.1 ppb_v found in the in situ measurements is not reproduced by the model as emissions are averaged over a large area (1.1°). Median noontime (HO₂+RO₂) retrieved from the EMAC model are ~ 5 % higher than (HO₂+RO₂) estimates based on measurement data, however, (HO₂+RO₂) deduced from EMAC is sometimes about a factor of 2 higher than the regional (HO₂+RO₂) estimate based on the Leighton Ratio and measured tracer data. NOPR estimates based on modeled data reproduce the tendencies derived from the measurements very well. However, the model does not reproduce 615 observed net ozone destruction along some clean parts of the ship cruise.

Data availability

Data used in this study is available to all scientists agreeing the AOABA protocol to at https://doi.org/10.5281/zenodo.3693988.

Author contributions

620 IT, HF and JL designed the study. UP and IT performed the CLD NO and NO₂ measurements and processed the data. JC and PE performed the O₃ measurements, JS performed the actinic flux measurements. JS performed cavity ring-down spectroscopy measurements of NO₂. DD and BH performed the HCHO measurements. HH, MM, RR, ST performed the OH and HO₂ measurements. J-DP was responsible for the H₂O measurements. Model simulations were made by AP. All authors have contributed to writing this manuscript

625 Competing interests

The authors declare no conflict of interest.

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635 Appendix: Acronyms and abbreviations

General

	AQABA	<u>A</u> ir <u>O</u> uality and Climate in the <u>A</u> rabian <u>Ba</u> sin campaign
	CyI	The <u>Cy</u> prus <u>I</u> nstitute
	KAUST	King Abdullah University of Science and Technology
640	KISR	<u>K</u> uwait <u>Institute for Scientific R</u> esearch

Regions

	AG	<u>A</u> rabian <u>G</u> ulf
	AS	<u>A</u> rabian <u>S</u> ea
	Μ	<u>M</u> editerranean Sea
645	NRS	<u>N</u> orthern <u>R</u> ed <u>S</u> ea
	OG	<u>O</u> man <u>G</u> ulf
	SRS	<u>S</u> outhern <u>R</u> ed <u>S</u> ea

Scientific

	CLD	<u>Chemiluminescence</u> <u>d</u> etector
650	CRDS	<u>C</u> avity <u>r</u> ing- <u>d</u> own <u>s</u> pectroscopy
	ECHAM5	Fifth generation <u>European</u> Centre <u>Hamburg</u> general circulation model
	EMAC	<u>ECHAM/MESSy Atmospheric Chemistry model</u>
	FWHM	<u>F</u> ull <u>w</u> idth at <u>h</u> alf <u>m</u> aximum
	GC-FID	<u>Gas chromatography – flame ionization detector</u>
655	HORUS	$\underline{\mathbf{H}}$ ydr $\underline{\mathbf{O}}$ xyl $\underline{\mathbf{R}}$ adical measurement $\underline{\mathbf{U}}$ nit based on fluorescence $\underline{\mathbf{S}}$ pectroscopy instrument
	HO _x	$OH + HO_2$
	LED	Light emitting diode
	LIF	Laser induced fluorescence
	MBL	<u>M</u> arine <u>b</u> oundary <u>l</u> ayer
660	MESSy	<u>M</u> odular <u>E</u> arth <u>S</u> ubmodel <u>Sy</u> stem
	NOPR	Net ozone production rate
	NO _x	$NO + NO_2$
	PFA	<u>P</u> er <u>f</u> luoro <u>a</u> lkoxy
	PSS	Photostationary steady state
665	PTFE	Polytetrafluoroethylene
	SLM	<u>S</u> tandard <u>l</u> iter per <u>m</u> inute
	STEAM3	<u>Ship</u> <u>Traffic</u> <u>E</u> mission <u>A</u> ssessment <u>M</u> odel 3
	TMU	<u>T</u> otal <u>m</u> easurement <u>u</u> ncertainty
	VOC	Volatile organic compounds
670	UV	Ultraviolet

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