Reply to Editor comments

In the following the comments of the editor are presented (black) alongside with our replies (in blue) and changes made to the manuscript (in red).

General statement: I am pleased the authors have addressed the referees' comments in detail. With respect to the main comment by referee #2, the treatment of error propagation has been adequately corrected. Nevertheless, I am not fully satisfied with the error estimates.

Dear editor, we appreciate your positive feedback. Below we provide detailed responses to your comments.

- Comment 1: The CLD instrument measures NO and NOc by chemiluminescence. The NO detection involves photon counting and the statistical noise of the signals should follow Poisson statistics. The absolute 1σ precision should scale with the square root of the NO signal for a given integration time, whereas the relative precision (in percent) is expected to decrease with the square root of the NO concentration. In line 156 of their revised manuscript, the authors specify a general constant value for the relative precision (5%). How can that be? Same arguments apply to the measurement of NOc.
- 15 It is true that the detection involves photon counting and that the statistical noise of each signal should follow Poisson statistics. However, the precision, which is used for calculating the total measurement uncertainty of both NO and NO_c, is defined as the reproducibility of all in-field calibrations. As the number of counts obtained for each 5 second calibration data point is large (> 50.000, yielding a relative Poisson uncertainty (noise) of less than 0.5 %), we find that the precision of both channels is mainly determined by drifts in the detector sensitivity (and practically not by Poisson noise of the signal). We have added a sentence after line 157, Page 7: Note that the precision is calculated from the reproducibility of all in-field calibrations, which is mainly determined by drifts in the detector sensitivity rather than by statistical Poisson noise of the measured signal.
- Comment 2: In line 157, the authors calculate a total relative uncertainty of NO from the signal precision (5%) and the error of the calibration gas mixture (3%). The same is done for NOc. In the calculation of the total uncertainty for NO2, the total uncertainties for NO and NOc are treated as statistically independent. This is not correct, because the calibrations of NO and NOc rely on the same calibration standard. The correct procedure to calculate the uncertainty of NO2 would be to calculate first the precision of NO2 from the precisions of NO and NOc. In a second step, the precision of NO2 can be combined with the uncertainty from the calibration mixture.
- We agree that NO and NO_c cannot be considered statistically independent, already by the fact that both channels are sensitive to ambient NO (the NO_c-channel further measures NO₂). However, and this should be clear from our statement to "Comment 1", it is not possible to calculate the precision of NO₂ as described above, simply as the chemiluminescence technique does not involve a direct NO₂ measurement (which could be used to calculate the reproducibility of calibrations associated with drifts in the detector sensitivity). Instead, we can estimate the relative uncertainty of NO₂ by use of the estimation of the largest error possible of statistically dependent variables NO and NO_c.

$$TMU([NO_2]) = \frac{1}{[NO_2]} \cdot \left(\left| \frac{\Delta[NO_c]}{K_e} \right| + \left| \frac{\Delta[NO]}{K_e} \right| + \left| \frac{\Delta K_e \cdot ([NO_c] - [NO])}{K_e^2} \right| \right)$$

This yields a campaign average and median measurement uncertainty of 23 % and 13 %, respectively, instead of 16 % and 8 %, respectively. We have revised the error formula for NO₂ and the given NO₂ uncertainty to 23 % (instead of 16%) in the manuscript. On Page 7, Line 161 now it says: The TMU for NO₂ has been estimated as the largest error possible of the statistically dependent variables NO and NO_c.

$$TMU([NO_2]) = \frac{1}{[NO_2]} \cdot \left(\left| \frac{\Delta[NO_c]}{K_e} \right| + \left| \frac{\Delta[NO]}{K_e} \right| + \left| \frac{\Delta K_e \cdot ([NO_c] - [NO])}{K_e^2} \right| \right)$$

Note that propagating a 23 % uncertainty in NO₂ yields revised median relative uncertainties associated with the (HO₂ + RO₂) and NOPR estimate of 74 % and 91 %, respectively. A 23 % NO₂ relative uncertainty yields average relative uncertainties of (HO₂ + RO₂) and NOPR of 176 % and 21 %, respectively. We have corrected these values in the manuscript on Page 21, L 445: Over the course of the campaign, the median relative (HO₂+RO₂) uncertainty is 74 %. The average is 176 % and heavily biased by single data outliers and therefore not representative. The relative error associated with the (HO₂+RO₂) calculation is hence estimated at 74 %. And on Page 24, L499: Incorporating a relative error of 74 % associated with (HO₂+RO₂), the median of the relative NOPR error of all data points obtained during AQABA is 91 %. The average relative uncertainty of NOPR is 21 % and strongly biased by single data outliers, which are in the case of NOPR significantly negative (due to fresh emissions and titration of O₃ by NO). Again the median is a more representative measure for the general uncertainty associated with the NOPR calculations. The relative error associated with the NOPR estimates based on measured data is hence estimated at 91 %.

55

Comment 3: In line 166, the authors assume that the error of the NO measurement is zero when no NO is present. However, when NO approaches zero, the signal noise will be dominated by the instrumental background (e.g., dark signal) which is responsible for the limit of detection. The role of background noise should be clarified.

This is true. Also since the equation on Page 8, Line 166: $\sqrt{(6\%^2+0\%^2+3\%^2)}\approx 6.7$ % used to calculate the NO₂ uncertainty during nighttime does not apply anymore, the error in NO₂ is instead calculated as the largest error possible from statistically dependent variables. We now compare the median and average to give a conservative upper limit for the NO₂ uncertainty. We have rewritten the following passage on Page 8, L164: "The median is lower than the average as NO is practically zero during nighttime. During nighttime, the relative error of the NO₂ data points is about 6.7 %. Note for convenience that is exactly the value calculated from errors in quadrature (if NO is practically zero during nighttime): $\sqrt{(6\%^2+0\%^2+3\%^2)}\approx 6.7$ %."

To: Over the course of the campaign the median and the average relative uncertainty of NO₂ are 13 % and 23 %, respectively. The relative uncertainty in NO₂ has been estimated as a conservative upper limit at 23 % as the average of the relative uncertainties of all data points obtained during AQABA.

Comment 4: In the paper, HO2 data are called "preliminary" because it has not been corrected for interferences from organic peroxy radicals. Since the discovery of possible RO2 interferences in the detection of HO2 by LIF, uncorrected HO2 is generally called HO2* (e.g., Lu et al., Atmos. Chem. Phys., 12, 1541–1569, 2012). I recommend to adopt this nomenclature. We have adopted the suggested nomenclature in the manuscript (text, tables, legends and captions of figures). Additionally on Page 9, L 200 now it says: As HO2 data are preliminary in this study, they will be called HO2*.

75

Comment 5: In line 199, the possible RO2 interference is treated as a statistical error when calculating the uncertainty of HO2. This is not appropriate. The interference is an additive positive bias. If the interference is 7%, then all HO2* values will be too high by this amount. Please clarify: where does the value of 7% come from? Why don't you correct the HO2* data by this amount?

We agree that the interference is generally an additive uncertainty rather than a statistical uncertainty, however note that 7 % (or 3 ppt, whichever value is larger) represents the <u>largest</u> uncertainty due to not yet corrected interference of organic peroxy radicals RO₂. The real quantitative interference remains unknown at this point.

85

Comment 6: In Table 1, I have the impression that TMU is used either for precision, or accuracy, or a combination of both. I suggest to specify instead the limit of detection and accuracy (usually related to calibration) for each substance in separate columns.

We agree that the TMUs given in Table 1 have been estimated differently, however, it has been non-trivial to compare and cluster the errors and total measurement uncertainties of each data set and observation to a general statement. The values given in Table 1 represent the relative uncertainties which were used in this study to determine the relative uncertainty associated with the (HO₂ + RO₂) and NOPR estimate. Only in the case of OH and HO₂, the uncertainty of the data sets were estimated as the 1 sigma accuracy of the respective measurement as these values give a good representation of the uncertainty associated with the OH and HO₂ data sets. Therefore we have renamed the column "TMU" as Associated relative uncertainty. Also we have rewritten the caption of Table 1 to: List of observations and gas phase measurements during AQABA. The relative uncertainty associated with each data set is given. Note that for OH and HO₂*, the relative uncertainty is estimated as the 1 sigma accuracy (and the 7 % interference of RO₂ in the case of HO₂* data). In addition, a reference of the measurement operability is given.

100

125

Comment 7: Equation (4) in line 271 looks weird. I find the re-definition of RO2 as the total sum of peroxy radicals (including HO2) confusing and inconsistent with other parts of the paper (e.g., line 96, line 239, reactions R4-R6). According to the general nomenclature, "RO2" should be used for the sum of organic peroxy radicals without HO2. I suggest to modify the equation accordingly:

105 P(O3) = kNO+RO2 [NO] [RO2] + kNO+HO2 [NO] [HO2]) = kNO+HO2 [NO] ([RO2] + [HO2])

We also state that RO_2 is used as the sum of organic peroxy radicals without HO_2 . Note that this necessitates a revision of the manuscript as, for instance, the initial RO_2 estimate is now called (HO_2+RO_2) estimate. The manuscript has been revised carefully in the different places.

110 Page 11, Line 252 now says: ..., we can combine the sum of all organic peroxy radicals R_iO₂ to the entitiy RO₂. The sum of HO₂ and RO₂ can be estimated using the steady state equation

$$[HO_2] + [RO_2] = \frac{j(NO_2) \cdot [NO_2] - k_{NO+O_3} \cdot [NO][O_3]}{k_{NO+HO_2} \cdot [NO]}.$$
(3)

We have modified P12, L269 and Eq. 4 accordingly: The production of ozone can be approximated by the rate of oxidation of NO with HO₂ and RO₂ to form NO₂ that will rapidly form O₃ (R1-R2) (Parrish et al., 1986, Thornton et al., 2002; Bozem et al., 2017).

$$P(O_3) = k_{\text{NO+HO}_2}[\text{NO}] \cdot ([\text{HO}_2] + [\text{RO}_2]) \tag{4}$$

120 Also we have modified Eq. 7 to:

$$NOPR = k_{NO+HO_2}[NO] \cdot ([HO_2] + [RO_2]) - [O_3] \cdot (\alpha \cdot j(O^1D) + k_{OH+O_2}[OH] + k_{HO_2+O_2}[HO_2]).$$
(7)

Futhermore, the term "RO₂ estimate" has been replaced by "(HO₂+RO₂) estimate" at various places throughout the whole manuscript and supplements (text, legends, captions). The error formulas for (HO₂+RO₂) and NOPR given in the manuscript have also been revised.

Net ozone production and its relationship to NO_x and VOCs in the marine boundary layer around the Arabian Peninsula

Ivan Tadic¹, John N. Crowley¹, Dirk Dienhart¹, Philipp Eger¹, Hartwig Harder¹, Bettina Hottmann¹, Monica Martinez¹, Uwe Parchatka¹, Jean-Daniel Paris², Andrea Pozzer^{1,4}, Roland Rohloff¹, Jan Schuladen¹, Justin Shenolikar¹, Sebastian Tauer¹, Jos Lelieveld^{1,3}, and Horst Fischer¹

Correspondence to: Ivan Tadic (i.tadic@mpic.de)

Abstract. Strongly enhanced tropospheric ozone mixing ratios have been reported in the Arabian Basin, a region with intense solar radiation and high concentrations of ozone precursors such as nitrogen oxides and volatile organic compounds. To analyze photochemical ozone production in the marine boundary layer (MBL) around the Arabian Peninsula, we use shipborne observations of NO, NO₂, O₃, OH, HO₂, HCHO, actinic flux, water vapor, pressure and temperature obtained during the summer 2017 Air Quality and Climate in the Arabian Basin (AQABA) campaign, compare them to simulation results of the ECHAM-MESSy atmospheric chemistry (EMAC) general circulation model. Net ozone production rates (NOPR) were greatest with 16 ppb_v day⁻¹ over both the Gulf of Oman and the Northern Red Sea and with 32 ppb_v day⁻¹ over the Arabian Gulf. NOPR over the Mediterranean, the Southern Red Sea and the Arabian Sea did not significantly deviate from zero; however, results for the Arabian Sea indicate weak net ozone production of 5 ppb_v day⁻¹, and net ozone destruction over the Mediterranean and the Southern Red Sea with -1 ppb_v day⁻¹ and -4 ppb_v day⁻¹, respectively. Constrained by HCHO/NO₂-ratios, our photochemistry calculations show that net ozone production in the MBL around the Arabian Peninsula occurs mostly in NO_x-limitation regimes with a significant share of ozone production occurring in the transition regime between NO_x- and VOC-limitation over the Mediterranean and more significantly over the Northern Red Sea and Oman Gulf.

¹Atmospheric Chemistry Department, Max Planck Institute for Chemistry, Mainz, Germany

²Laboratoire des Sciences du Climat et de l'Environnement, LSCE/IPSL, CEA-CNRS-UVSQ, Université Paris-Saclay, Gifsur-Yvette, France

^{135 &}lt;sup>3</sup>Energy, Environment and Water Research Center, The Cyprus Institute, Nicosia, Cyprus

⁴International Centre for Theoretical Physics, Trieste, Italy

1 Introduction

Revenues from exploitation of the great oil reserves in the states of and around the Arabian Peninsula have propelled remarkable economic development associated with industrialization and urbanization. Strong population growth and anthropogenic emissions of gases and particulates in the last few decades have resulted in the Middle East becoming a hotspot for air pollution and associated health effects, while it is also one of the regions worldwide where climate change is particularly rapid (Lelieveld et al., 2016a). Unique meteorological conditions such as intense solar radiation, high temperatures and aridity, as well as strong anthropogenic emissions of volatile organic compounds (VOCs) and NO_x (= $NO + NO_2$) by on- and off-shore petrochemical industries, dense ship traffic, fossil energy production for air conditioning and desalination, and urban development are expected to further intensify in the future and contribute to photochemical ozone production (Lelieveld et al, 2009; Krotkov et al., 2016; Pfannerstill et al., 2019). Understanding the sources and sinks of NO_x and other ozone precursors on and around the Arabian Peninsula is therefore of major importance for atmospheric chemistry studies, including the investigation of net ozone production rates (NOPR) (Monks et al., 2015; Reed et al., 2016; Bozem et al., 2017).

 NO_x plays a central role in atmospheric photochemistry (Nakamura et al., 2003; Tuzson et al., 2013; Reed et al., 2016). It is the primary precursor for tropospheric ozone (O_3), secondary organic aerosols and photochemical smog in urban areas (Hollaway et al., 2012; Javed et al., 2019). Main ground-based sources of NO and NO_2 are fossil fuel combustion and to a lesser extent bacterial processes in soils, and both lightning and aircraft emissions in the upper troposphere (Nakamura et al., 2003; Miyazaki et al., 2017; Javed et al., 2019). Transport of NO_x in the atmosphere is relatively limited due to its short lifetime of a few hours (Reed et al., 2016). It is removed from the troposphere mainly by conversion to HNO_3 (via reaction with OH) during the day, or the formation of N_2O_5 (in the reaction of NO_2 with NO_3 at night-time), which also leads to formation of nitric acid by heterogeneous hydrolysis on aerosol surfaces (Crutzen, 1973; Liu et al., 2016; Reed et al., 2016). Ultimately, the deposition of HNO_3 constitutes the major loss process of NO_x from the atmosphere. Ozone is a secondary pollutant that is photochemically formed in the troposphere from its precursors NO_x and VOCs (Bozem et al., 2017; Jaffe et al., 2018). It is an important greenhouse gas, an atmospheric oxidant and the most important primary precursor for OH (Lelieveld et al., 2004; Monks et al., 2015; Bozem et al., 2017). O_3 in the planetary boundary layer causes health damage, notably respiratory diseases, and reduces crop yields (Monks et al., 2015; Jaffe et al., 2018).

 NO_x and O_3 mixing ratios in the troposphere vary from less than 20 ppt_v and 10 ppb_v, respectively, for pristine conditions such as the remote marine boundary layer (MBL) up to mixing ratios of several hundreds of ppb_v in regions with heavy automobile traffic and in international shipping lanes (for NO_x) and downwind of urbanized areas (for O_3) (Reed et al., 2016; Jaffe et al., 2018). Low NO_x environments such as the clean MBL and the lower free troposphere are considered net ozone destruction regimes whereas the upper troposphere and areas with anthropogenic emissions of ozone precursors are regions of net ozone production (Klonecki and Levy, 1997; Bozem et al., 2017). Measurements performed in the Houston Ship Channel revealed NOPR of the order of several tens of ppb h^{-1} (Chen et al., 2010; Mao et al., 2010; Ren et al., 2013).

In the last decade much effort has been successfully devoted to the mitigation of NO_x emissions over Europe and America, and levels of reactive nitrogen trace gases have decreased (Miyazaki et al., 2017). But in Asia, India and the Middle East, NO_x emissions have substantially increased during the last decade so that the global NO_x burden has essentially remained constant (Miyazaki et al., 2017). NO_x emissions by ocean-going vessels have attracted considerable attention as they are reported to account for 15 % of the global NO_x emission burden (Celik et al., 2019). Model calculations suggest that the Arabian Gulf, with an estimated annual NO_x emission density of about one ton km⁻² from ship traffic, is among the regions with highest NO_x emission densities worldwide (Johansson et al., 2017). Although NO_x emissions in the Red Sea and Arabian Sea areas were reported to be three and five times smaller than for the Arabian Gulf, respectively, these values are still 50-100 times larger than the emission density reported for the South Pacific Ocean, for example (Johansson et al., 2017).

In the present study, we characterize photochemical NOPR in the MBL around the Arabian Peninsula. In Sect. 2, the campaign, instrument description, data processing and a description of the methods used in this study is presented. In Sect. 3, mixing ratios of nitrogen oxides and ozone around the Arabian Peninsula are reported. Based on concurrent measurements of HO_x, actinic flux, temperature and pressure, noontime (HO₂+RO₂) mixing ratios are estimated and used to calculate NOPR in the different regions around the Arabian Peninsula. Observation-based analysis of HCHO/NO₂-ratios will be used to distinguish between NO_x- or VOC-limited chemistry in the particular regions. A comparison of the results with data retrieved from the 3D global circulation model EMAC is also included.

2 Experimental

185

190

195

205

210

200 2.1 AQABA campaign

The AQABA ship campaign ($\underline{\mathbf{A}}$ ir $\underline{\mathbf{Q}}$ uality and Climate in the $\underline{\mathbf{A}}$ rabian $\underline{\mathbf{Ba}}$ sin) investigated the chemical composition of the MBL around the Arabian Peninsula. From late June to early September 2017, the *Kommandor Iona* Research and Survey Vessel sailed from Toulon (France) to Kuwait and back in order to perform gas-phase and particle measurements in the region. The gas-phase and aerosol measurement instrumentation was housed in five laboratory containers on the front deck. A 6 m high, 20 cm diameter cylindrical stainless steel common inlet was installed on the front deck of the vessel to sample air at a total mass flow rate of 10,000 SLM. NO and NO₂ chemiluminescence measurements were obtained at a total bypass flow rate of 28.5 SLM sampling air from the common inlet with a residence time in the tubing of ~3 s. HCHO, NO₂ cavity ring-down spectroscopy and O₃ measurements were obtained with similar bypass systems sampling air from the common inlet. H₂O vapor was measured on the top of the ship mast in the front. The OH and HO₂ detection units were placed on the prow to allow for inlets with residence times less than 10 ms.

The *Kommandor Iona* left Malta in late June 2017 traversing the Mediterranean Basin, the Suez Canal and the Northern Red Sea. A 3 day stop over at KAUST University (Saudi Arabia) was made from 11 July 2017 to 13 July 2017 before passing the Southern Red Sea area. On 17 July 2017, we briefly stopped at Djibouti port before passing the Gulf of Aden, the Arabian Sea

and the Gulf of Oman. Kuwait at the northern end of the Arabian Gulf marked the turning point of the ship cruise where, during a second 3-day stop-over, scientific staff was exchanged. The *Kommandor Iona* started the second leg on 03 August 2017 arriving in Toulon (France) in early September 2017 without any further stops. Figure 1 shows the ship's route subdivided into six different regimes.

215

225

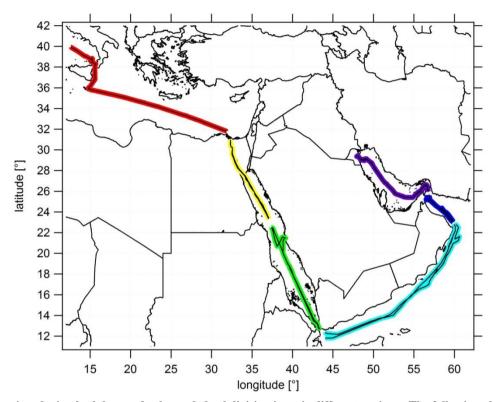


Figure 1: Ship cruises during both legs and color-coded subdivision into six different regimes. The following abbreviations will be used: AG for Arabian Gulf (purple), OG for Oman Gulf (dark blue), AS for Arabian Sea (blue), SRS for Southern Red Sea (green), NRS for Northern Red Sea (yellow), M for Mediterranean (red).

To enhance the statistical significance of our results and due to comparable signatures of the NO_x and O_3 measurements in the northern part of the Red Sea, the Suez Gulf and the Suez Canal, we have combined these regions which are represented by the 'Northern Red Sea' (NRS). For the same reasons we have merged the Gulf of Aden with the Arabian Sea (AS). See supplementary Table ST1 for the range of latitudinal and longitudinal coordinates of the different regions and supplementary Table ST2 for a detailed day to day description of the route.

2.2 Measurements of nitrogen oxides during AQABA

245

250

255

230 Chemiluminescent detection of NO and NO₂ is a widely applied method to quantify mixing ratios from the ppm_y down to the low ppt_v range (Nakamura et al., 2003; Pollack et al., 2011; Hosaynali Beygi et al., 2011; Reed et al., 2016). During AQABA we deployed a compact, robust and commercially available two-channel chemiluminescence instrument CLD 790 SR (ECO Physics AG, Dürnten, Switzerland) that has been optimized for in situ field measurements during the last decade (Hosaynali Beygi et al., 2011). The measurement principle of the CLD is based on the addition of O₃ to NO to produce stoichiometric 235 quantities of excited state NO_2^* that will emit an infrared photon ($\lambda > 600$ nm) forming the chemiluminescent detection principle for NO (Drummond et al., 1985; Reed et al., 2016). Both channels feature an identical layout and were operated at a mass flow of 1.5 SLM during AOABA. One channel of the CLD (NO_c-channel) has additionally been equipped with a LED solid state photolytic converter (Droplet Measurement Techniques, Boulder, Colorado) installed upstream of the O₃ addition to selectively photolyze NO₂ to NO, which is subsequently measured. In this section, we will concentrate on modifications 240 made prior to the campaign and especially on operational conditions of the photolytic converter during the campaign. Further details on the measurement principle are described elsewhere (Pollack et al., 2011; Hosaynali Beygi et al., 2011; Reed et al., 2016).

During AQABA, the cylindrical photolytic converter (length 14 cm, volume ~ 0.079 l) was operated at a constant pressure of 95 hPa yielding a residence time of ~ 0.3 s. The photolytic NO₂ converter features a set of 200 UV LED units attached to each end of the converter. The emission profile of the UV LED units was characterized in laboratory measurements to peak at 398 nm with a Full Width at Half Maximum (FWHM) of 16 nm. The UV-induced positive bias in the NO₂-measurement due to photolysis of BrONO₂, HONO, NO₃ and ClNO₂ to produce NO was estimated at 6.1 %, 2.8 %, 2.7 % and 1.2 %, respectively, based on the absorption cross sections from the MPI-Mainz UV/VIS Spectral Atlas of Gaseous Molecules (Keller-Rudek et al., 2013). These values represent upper limits for the interference of the respective NO_v compound as the respective molecular quantum yield was estimated conservatively at 1. Note that the values represent percent interferences if the interferent had the same concentration as NO₂. Due to small daytime concentrations of these molecules in the MBL, a UV-induced bias was neglected for the observations in this study. To limit wall loss of NO₂, the inner cavity surface is made of PTFE (polytetrafluoroethylene), which may potentially provide a reservoir (via surface adsorption) for NO_v that can thermally dissociate to increase the background signal of the NO_2 measurement (Reed et al., 2016). The conversion efficiency K_e of the photolytic NO₂ conversion was estimated by gas phase titration (SYCOS K-GPT-DLR, ansyco, Karlsruhe, Germany) several times before, during and after the campaign at (29.4 ± 0.9) % allowing the calculation of NO₂ concentrations by $[NO_2]$ = $\frac{[NO_c]-[NO]}{\nu}$. To avoid chemical interferences due to adding ozone in excess during a gas phase titration, a small but not vanishing amount of NO has always been left unoxidized during gas phase titrations.

During AQABA, regular dry zero-air measurements as well as NO and NO₂ calibrations were performed autonomously over a 10 minute period every 6 hours to accurately quantify the instrumental background and to correct for sensitivity drifts. An autonomous cycle of '2 min zero air measurements - 2 min NO calibration - 2 min zero air measurement - 2 min NO₂ calibration – 2 min zero air measurement' was implemented. Continuous flows NO and NO₂ calibration gases were added to the synthetic airflow or directed to a pump by switching solenoid valves. The NO calibration standard (1.954 \pm 0.039 ppm_y NO in N_2 , Air Liquide, Germany) used during the campaign was compared to a primary standard (5.004 \pm 0.025) ppm_y (NPL, Teddington, UK) after the campaign yielding an effective NO mixing ratio of (2.060 ± 0.057) ppm_y in the NO calibration gas. Zero air measurements and NO calibrations were performed with a total flow of 3.44 SLM achieving an overflow of 0.44 SLM to guarantee ambient air free standard measurements. The calibration gas was added at 4.5 sccm to the zero air flow. During AQABA, NO calibrations at 2.5 ppb_v were achieved. During the first leg of the campaign, zero air was sampled from a bottle (Westfalen AG, Germany), whereas during the second leg zero air was generated from a zero air generator (Air Purifier CAP 180, acuraLine). Zero air measurements generated with the zero air generator were statistically not significantly different from those achieved by a bottle. To correctly account for the photomultiplier background and chemical interferences due to reactions of ozone with ambient alkenes additional pre-chamber measurements were performed every 5 minutes as well as at the beginning of zero air measurements and calibrations for 25 s each. This correction is removing a large fraction of the interference signal from alkenes. However, in regions where alkene concentrations are strongly varying in time and magnitude, the CLD is prone to enhanced backgrounds due to the interference of alkenes with ozone in the instrument. A schematic setup of the two-channel CLD instrument is given in Figure 2.

260

265

270

275

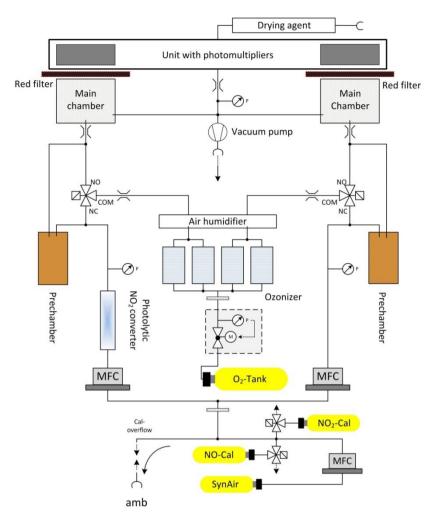


Figure 2: Schematic setup of the two channel CLD instrument in the configuration used during AQABA. NO and NO₂ calibration gases were running continuously and were added to the zero airflow by switching the respective solenoid valves.

280

285

The total measurement uncertainty (TMU) in NO has been calculated at 6 % at an integration time of 5 minutes and a confidence level of 1σ by adding the precision (5 %) and the error of the calibration gas mixture (3 %) in quadrature: TMU([NO]) = $\sqrt{(5\%)^2 + (3\%)^2} \approx 6\%$. Note that the precision is calculated from the reproducibility of all in-field calibrations, which is mainly determined by drifts in the detector sensitivity rather than by statistical Poisson noise of the measured signal. The limit of detection in the NO channel was estimated as the full width at half maximum of the frequency distribution of all zero air measurements obtained during the campaign to be 9 ppt_v at a 5 min integration time and a confidence level of 1σ . The TMU for NO₂ has been estimated as the largest error possible of the statistically dependent variables NO and NO_c.

$$\mathrm{TMU}([\mathrm{NO}_2]) = \frac{1}{[\mathrm{NO}_2]} \cdot \left(\left| \frac{\Delta[\mathrm{NO}_c]}{K_e} \right| + \left| \frac{\Delta[\mathrm{NO}]}{K_e} \right| + \left| \frac{\Delta K_e \cdot ([\mathrm{NO}_c] - [\mathrm{NO}])}{K_e^2} \right| \right)$$

290 Note that the total measurement uncertainty of the NO_c-channel data has also been calculated at 6 % at an integration time of 5 minutes and a confidence level of 1σ by adding the precision and the error of the calibration gas mixture in quadrature. Over the course of the campaign the median and the average relative uncertainty of NO₂ are 13 % and 23 %, respectively. The relative uncertainty in NO₂ has been estimated as a conservative upper limit at 23 % as the average of the relative uncertainties of all data points obtained during AOABA. As the zero air measurements in the NO₂ channel produced an increased 295 background affected by memory effects after exposure to high NO_x levels e.g. during measurements of stack emissions, the NO₂ raw data were initially processed without converter background subtraction. As we therefore expect the CLD NO₂ data to be offset due to not being initially background corrected, the converter background was estimated at 112 ppty from the centre of a Gaussian fit representing the difference of 1-minute averaged CLD NO₂ and concurrent cavity ring-down spectroscopy (CRDS) NO₂ measurements for data points below 10 ppb_y. Setting the threshold for calculating the difference of the two 300 concurrent data sets to 10 ppb_v is somewhat arbitrary, however, changing this limit to 5 ppb_v or 20 ppb_v does not significantly vary the estimated offset of the CLD NO₂ data. The offset correction of 112 ppt_v was taken as the ultimate absolute measurement uncertainty of the CLD NO₂ measurement. Further corrections of to the final CLD data include residence time corrections as well as corrections for NO and O₃ losses and the subsequent formation of NO₂ in the sampling line (Ryerson et al., 2000). Both NO and NO₂ CLD data have also been corrected for nonlinearities for concentrations higher than 55 ppb_v, as 305 experienced during probing of stack emissions.

2.3 Further measurements used in this study

310

315

An extensive set of concurrent measurements providing mixing ratios of O₃, NO₂, HCHO, OH, HO₂, absolute humidity and actinic flux, temperature and pressure data obtained during AQABA was used in this study. Ozone was measured with an absorption photometer (Model 202 Ozone Monitor, 2B Technologies, Boulder, Colorado) based on the well-established absorption of the mercury line in the Hartley band at 254 nm (Viallon et al., 2015). Eliminating water and particle interferences during sampling was achieved via sampling through a nafion tube and a Teflon filter. The ozone monitor was zeroed ten times during the campaign. NO₂ was further measured by cavity ring-down spectroscopy (Sobanski et al., 2016) and used for correcting the instrumental background of the CLD NO₂ data, as described above (the correction was taken as the ultimate absolute measurement uncertainty in the CLD NO₂ data). Note that in this study we will use the NO₂ CLD data rather than the NO₂ CRDS data as the temporal coverage of the CLD NO₂ data over the course of the campaign is about 60 % compared to about 35 % for the cavity ring-down measurement. Formaldehyde (HCHO) was measured with an Aerolaser 4021 (AERO-LASER GmbH, Garmisch-Partenkirchen, Germany), which is a fully automatized monitor based on the Hantzsch technique (Kormann et al., 2003). H₂O measurements were obtained using a cavity ring-down spectroscopy monitor (PICARRO G2401, Santa Clara, California) supervised by Laboratoire des Sciences du Climat et de l'Environnement (LSCE) (Kwok et al., 2015).

320 Measurements of OH and HO2 were performed with the custom-built HydrOxyl Radical measurement Unit based on fluorescence Spectroscopy (HORUS) instrument based on laser-induced fluorescence (LIF) spectroscopy of the OH molecule and NO titration of HO₂ to OH followed by LIF spectroscopy detection of the OH molecule (Martinez et al., 2010; Regelin et al., 2013). HO₂ data used in this study is still preliminary due to not yet corrected interference of organic peroxy radicals RO₂. The largest uncertainty due to interference by contribution of RO₂ is 7 % or 3 ppt_y whichever is higher. The 1 sigma accuracy 325 of both OH and HO₂ is 20 %. The uncertainty in the OH data is here estimated as the 1 sigma accuracy of the data set at 20 %, whereas the uncertainty in HO₂ is estimated at $\sqrt{20\%^2 + 7\%^2} \approx 21\%$. As HO₂ data are preliminary in this study, they will be called HO₂*. Wavelength resolved down-welling actinic flux was measured with a spectral radiometer (model CCD Spectroradiometer 85237). The j-values for NO₂ and O₃ were not corrected for upwelling UV radiation and were estimated to have a ~ 10 % measurement uncertainty (Meusel et al., 2016). The radiometer was installed 10 m above sea level, respectively 330 5 m above the front deck surface. Decreases in sensitivity due to sensor contamination with e.g. sea-spray were corrected with a linear interpolation between two (daily) cleaning events. Temperature and pressure measurements were performed with the Shipborne European Common Automatic Weather Station (EUCAWS), a weather station specifically designed for ships. The weather station incorporates sensors, processing units, satellite positioning and communication systems in one device and is implemented and coordinated by the European National Meteorological Service EUMETNET. Table 1 lists the measurement 335 methods and the TMU for each observation.

Table 1: List of observations and gas phase measurements during AQABA. The relative uncertainty associated with each data set is given. Note that for OH and HO₂*, the relative uncertainty is estimated as the 1σ accuracy (and the 7 % interference of RO₂ in the

case of HO₂* data). In addition, a reference of the measurement operability are given.

Molecule	Method	Associated relative uncertainty	References
NO	chemiluminescence	6 %	Hosaynali Beygi et al., 2011
NO ₂	photolysis-chemiluminescence	23 %	Hosaynali Beygi et al., 2011
NO ₂	cavity ring-down spectroscopy	7 %	Sobanski et al., 2016
O ₃	UV absorbance	2 %	Viallon et al., 2015
ОН	LIF	20 %	Martinez et al., 2010
HO ₂ *	NO titration / LIF	21 %	Martinez et al., 2010
НСНО	Hantzsch technique	13 %	Kormann et al., 2003
H ₂ O	cavity ring-down spectroscopy	5 %	Kwok et al., 2015
actinic flux	spectral radiometer	10 %	Meusel et al., 2016

The *Kommandor Iona* Research and Survey Vessel sailed whenever possible with the wind coming from the bow to avoid contamination by stack emissions. However, based on the relative wind direction, the variability in NO as well as the temporal evolution of NO_x, SO₂, and O₃ sections of data in which the air mass was contaminated by the ship's stack were identified. All data used here to calculate (HO₂+RO₂) and NOPR have been filtered to remove contaminated air masses. Altogether, 21 % of the sampling time was potentially contaminated by the ship exhaust of the KI of which 87 % occurred on the first leg. During the second leg the ship sailed against the wind and most of the data was free of stack contamination. Our analysis is based on a 5-minute running mean for each data set, whereby only averages that have been calculated at a temporal coverage greater than 30 % have been used. A time series of the NO, NO₂ (both CLD), O₃, OH, HO₂* and *j*(NO₂) measurements is given in the supplementary Figures S2 and S3.

NO and NO₂ were measured from 03 July 2017 to 31 August 2017, O₃ was measured from 22 June 2017 to 01 September 2017, HCHO from 01 July 2017 to 31 August 2017 and OH and HO₂* from 18 July 2017 to 31 August 2017. For the analysis of peroxy radicals RO₂ and NOPR around the Arabian Peninsula we have removed data measured during the stop-overs in Jeddah (11 July to 13 July), Kuwait (31 July to 03 August) and during bunkering at Fujairah City (06 August, 07:00 – 15:00 UTC). Due to HO_x data being available from 18 July 2017 onward, we have limited the net ozone production analysis to the period after this date.

2.4 Methods

355

360

The so-called NO_x -O₃-null cycle represents a rapid daytime cycling between NO, NO_2 and O_3 . Solar UV radiation photolyzes NO_2 to NO and $O(^3P)$ (R1) which will reform O_3 in the subsequent reaction with molecular oxygen O_2 (R2) (Leighton, 1961). NO and O_3 react to form NO_2 and O_2 (R3). R1, R2 and R3 constitute a so called null cycle which establishes photostationary steady state (PSS) for both NO_x and O_3 in mid latitudes during noon time on a time scale of ~100 s (Thornton et al., 2002; Mannschreck et al., 2004).

$$NO_2 + h\nu \ (\lambda < 424 \text{ nm}) \rightarrow NO + O(^3P)$$
 (R1)

$$O_2 + O(^3P) + M \rightarrow O_3 + M$$
 (R2)

$$NO + O_3 \rightarrow NO_2 + O_2$$
 (R3)

365 Under the assumption of PSS, the Leighton Ratio φ is unity (Leighton, 1961)

$$\varphi = \frac{j(NO_2) \cdot [NO_2]}{k_{NO+O_3} \cdot [NO][O_3]} = 1 \tag{1}$$

with $j(NO_2)$ being the NO_2 photolysis rate [s⁻¹]. In low NO_x environments (< 100 ppt_v) previous studies have indicated that further NO oxidizing trace gases such as peroxy radicals (HO₂, RO₂) and halogen monoxides (XO) may result in a deviation from unity (Nakamura et al., 2003; Hosaynali Beygi et al., 2011; Reed et al., 2016).

$$370 \text{ NO} + \text{HO}_2 \rightarrow \text{NO}_2 + \text{OH}$$
 (R4)

$$NO + RO_2 \rightarrow NO_2 + RO \tag{R5}$$

$$NO + XO \rightarrow NO_2 + X \tag{R6}$$

Deviations from expected NO/NO₂-ratios at low NO_x generally refer to missing oxidants converting NO to NO₂ (Hosaynali Beygi et al., 2011; Reed et al., 2016) or to a measurement error due to an instrumental background or a positive interference from thermal labile NO_x reservoir species (Reed et al., 2016; Silvern et al., 2018). In the present study we include HO₂ and R_iO₂ into the production term for NO₂.

375

380

385

390

$$j(NO_2) \cdot [NO_2] = k_{NO+O_3} \cdot [NO][O_3] + k_{NO+HO_2} \cdot [NO][HO_2] + [NO] \cdot \sum_i k_{NO+R_iO_2} \cdot [R_iO_2]$$
 (2)

Assuming that the temperature-dependent rate coefficient for the reaction of each particular peroxy radical R_iO_2 with NO equals the rate k_{NO+HO_2} for Reaction R4 (Hauglustaine et al., 1996; Cantrell et al., 1997; Thornton et al., 2002), we can combine the sum of all organic peroxy radicals R_iO_2 to the entity RO_2 . The sum of HO_2 and RO_2 can be estimated using the steady state equation

$$[HO_2] + [RO_2] = \frac{j(NO_2) \cdot [NO_2] - k_{NO+O_3} \cdot [NO][O_3]}{k_{NO+HO_2} \cdot [NO]}.$$
(3)

However, the steady state assumption is not valid if the sampled air parcel is affected by fresh emissions or fast changes in the actinic flux (Thornton et al., 2002). After sampling a fresh emission e.g. a ship plume, for which NO_x went up typically to values of several tens of ppb_v with simultaneous titration in O_3 , we assume that PSS is re-established on a time scale of 2 minutes (Thornton et al., 2002; Mannschreck et al., 2004). To best approximate PSS in our analysis we have restricted the estimation of RO_2 on time frames \pm 2 h around noontime for which we expect the smallest relative changes in the actinic flux. Noontime for each day was determined as the centre of a Gaussian fit that was applied to the actinic flux data. We applied a Gaussian Fit to the actinic flux data as this fitting method is sufficient to estimate the centre of the diurnal actinic flux. To further limit the effect of periods for which PSS is not fulfilled, we use the median instead of the average that is often disproportionately biased by strong NO_x sources nearby. See supplementary Tables ST3, ST5 and ST7 for detailed statistics and a further motivation on regional averages and median values. See supplementary Figure S1 for a detailed illustration of the calculation of the fraction of the noontime integral.

A further part of the analysis will be the investigation of NOPR. Ozone production is initiated by reactions that produce HO_x, for which primary production is from the photolysis of ozone, formaldehyde, nitrous acid (HONO) and hydrogen peroxide (H₂O₂) (Thornton et al., 2002; Lu et al., 2010; Hens et al., 2014; Mallik et al., 2018). The production of ozone can be approximated by the rate of oxidation of NO with HO₂ and RO₂ to form NO₂ that will rapidly form O₃ (R1-R2) (Parrish et al., 1986; Thornton et al., 2002; Bozem et al., 2017).

$$P(0_3) = k_{\text{NO+HO}}, [\text{NO}] \cdot ([\text{HO}_2] + [\text{RO}_2]) \tag{4}$$

400 Photochemical O_3 loss is mainly due to photolysis ($\lambda < 340$ nm) in the presence of water vapor and the reactions of ozone with OH and HO_2 (Bozem et al., 2017).

$$O_3 + h\nu \,(\lambda < 340 \,\text{nm}) \rightarrow O_2 + O(1D)$$
 (R7.1)

$$O(^{1}D) + H_{2}O \rightarrow 2OH$$
 (R7.2)

$$O_3 + OH \rightarrow HO_2 + O_2 \tag{R8}$$

$$405 \quad 0_3 + H0_2 \rightarrow OH + 20_2$$
 (R9)

 α , the fraction of O(¹D) that reacts with H₂O

410

$$\alpha = \frac{k_{0(1D)+H_20}[H_20]}{k_{0(1D)+H_20}[H_20] + k_{0(1D)+M}[M]}$$
(5)

was (10.6 \pm 2.2) % during AQABA with a quasi linear dependence on water concentrations. The error in α is mainly determined by the error of H₂O at 5 %. Furthermore, ozone is lost due to reactions with alkenes (R12) and halogen radicals (R13).

$$O_3$$
 + alkenes \rightarrow radicals (R12)

$$O_3 + X \rightarrow O_2 + XO \tag{R13}$$

We find that the loss rate is dominated by the photolysis of ozone with subsequent reaction of O(¹D) with H₂O, was 60 – 80 % of the total loss rate, followed by the reaction of O₃ with HO₂, which makes up 10 – 30 % (note that the uncertainty in HO₂ radical concentrations mentioned above has no significant influence on the total O₃ loss rate, due to its small contribution). The remaining fraction (10-30 %) is due to the reaction of O₃ with OH. The reaction of ozone with ethene is on average 0.005 – 0.01 ppb_v h⁻¹ and therefore generally less than 2 % of the total ozone loss rate (Bourtsoukidis et al., 2019). The reaction of O₃ with all alkenes will hence be neglected. Halogen radicals were not measured during AQABA and will not be incorporated into our study. Based on oxidative pairs, Bourtsoukidis et al. (2019) have classified the majority of their samples collected during AQABA by an OH/Cl-ratio of 200:1. As measured daytime OH concentrations were of the order of 5 · 10⁶ molecule cm⁻³, the estimate would yield a Cl concentration of 2.5 · 10⁴ molecule cm⁻³, which would decrease the estimated diurnal net

ozone production rates by roughly 0.2 ppb_v day⁻¹ over the Arabian Sea and at most 0.6 ppb_v day⁻¹ over the other regions, which does not substantially alter the here presented results. The noontime chemical ozone loss rate can be summarized by

$$L(O_3) = [O_3] \cdot (\alpha \cdot j(O^1D) + k_{OH+O_3} \cdot [OH] + k_{HO_3+O_3} \cdot [HO_2]).$$
(6)

425 NOPR presented in this study is finally calculated as the difference of Eq. 4 and Eq. 6.

$$NOPR = k_{NO+HO_2}[NO] \cdot ([HO_2] + [RO_2]) - [O_3] \cdot (\alpha \cdot j(O^1D) + k_{OH+O_3}[OH] + k_{HO_2+O_3}[HO_2])$$
(7)

Under the assumption of constant chemical composition for a given day, the NOPR is expected to have a diel cycle following the measured actinic flux. Hence integrating the estimated NOPR over the course of a day based on the particular fractional noontime integral of $j(NO_2)$ will yield a diurnal value for NOPR. A detailed calculation of the diurnal fractional integrals is given in the supplementary Figure S1. Note that all reaction rate constants used are from the IUPAC Task Force on Atmospheric Chemistry Chemical Kinetic Data Evaluation (Atkinson et al., 2004). Indications whether a chemical regime is NO_x -limited or VOC-limited can be derived from the ratio of HCHO to NO_2 . Former studies have derived HCHO/ NO_2 -ratios from satellite measurements to establish whether ozone production is NO_x -limited or VOCs-limited. The results indicate NO_x -limitation for HCHO/ NO_2 > 2 and prevailing VOC-limitation for HCHO/ NO_2 < 1(Duncan et al., 2010).

435 2.5 ECHAM/MESSy Atmospheric Chemistry (EMAC) model

430

440

445

EMAC is a 3D general circulation model that includes a variety of sub-models to describe numerous processes in the troposphere, their interaction with oceans and land surfaces and incorporates anthropogenic influences. Here we use the second development cycle of the Modular Earth Submodel System (MESSy2) (Jöckel et al., 2010) and ECHAM5 (Röckner et al., 2006) which is the fifth generation European Centre Hamburg general circulation model in the T106L31 resolution (corresponding to a quadratic grid of roughly 1.1° and 1.1°). The model has 31 vertical pressure levels and involves the complex organic chemistry mechanism MOM (Mainz Organic Mechanism) as presented by Sander et al. (2019) that includes further developments of the version used by Lelieveld et al. (2016b). Here we use the lowest pressure level in a terrain following coordinates (equivalent to the surface level) and simulations of NO, NO₂, O₃, OH, HO₂, j(NO₂) and j(O¹D). The sum of peroxy radicals was estimated as the sum of all radicals R_iO_2 with less than four carbon atoms. Net ozone production based on data retrieved from EMAC was estimated as

NOPR = [NO] ·
$$(k_{\text{NO+HO}_2}[\text{HO}_2] + \sum_i k_{\text{NO+R}_i\text{O}_2}[\text{R}_i\text{O}_2]) - [\text{O}_3] · (\alpha · j(\text{O}^1\text{D}) + k_{\text{OH+O}_3}[\text{OH}] + k_{\text{HO}_2+\text{O}_3}[\text{HO}_2]).$$
 (8)
A list of all included peroxy radicals R_iO₂ for the reaction with NO is given in the supplementary Table ST10.

3 Results and discussions

3.1 NO_x and O₃ in the MBL around the Arabian Peninsula

450 During AQABA NO_x mixing ratios varied over three orders of magnitude with lowest values of less than 50 ppt_y observed in relatively pristing regions and highest values of more than 10 ppb_y found in the vicinity of areas with strong anthropogenic influence or nearby passing ships. Ozone mixing ratios ranged from values of less than 20 ppb_v, detected over the Arabian Sea, to more than 150 ppb_v during episodes of severe pollution. Figures 3a) and 3b) show distributions of NO_x measured during the first and second leg of the campaign (range from 0.1 ppb_v to 20 ppb_v) while Figure 3c) and 3d) show corresponding ozone 455 mixing ratios covering a range from 20 ppb_y to 100 ppb_y, respectively. A classification of the different regions based on Box-Whisker-Plots, including the 25-75-percentile interval (box) and whiskers for the 10-90-percentile interval, is shown in Figure 4 and Figure 5 for NO_x and O₃, respectively. As average NO_x is often influenced by fresh, localized emissions, we have included the median (black bar) instead of the average in the Box-Whisker-Plot for NO_x, which is less sensitive to extreme values. For O₃, although the difference between median and mean is mostly negligible, we also use the median in Figure 5. NO_x and O₃ averages, medians, standard deviations, 1st and 3rd quantiles and the number of data points quantified per region 460 are given in the supplementary Table ST3. See supplementary Figure S4 for OH and HO₂* mixing ratios around the Arabian Peninsula. Supplementary Figure S5 shows that absolute humidity observed during AQABA ranges from lowest values of less than 1 % observed in the Suez Golf during the first leg to about 3 % observed during both legs in the southeastern part of the Arabian Gulf and in the Strait of Hormuz. Although observing highest absolute humidity on both legs in the southeastern part 465 of the Arabian Gulf, absolute humidity was very low on the first leg near Kuwait, where absolute humidity was about 1 %. These air masses were brought from the Kuwait/Iraq area into the MBL of the Arabian Gulf on the first leg, whereas a change of wind direction for the second leg resulted in winds coming from Iran area with moister air. For the rest of the cruise, absolute humidity mixing ratio was about 1.5 % with variations being generally less than 0.5 %.

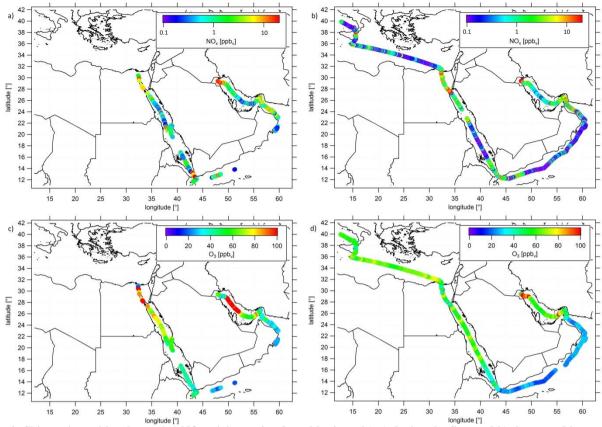


Figure 3: Ship cruises with color-scaled NO_x mixing ratios (logarithmic scale) a) during the first and b) the second leg and color-scaled O_3 mixing ratios (linear scale) c) during the first and d) during the second leg. Note that both NO_x and O_3 has been filtered for own stack contamination.

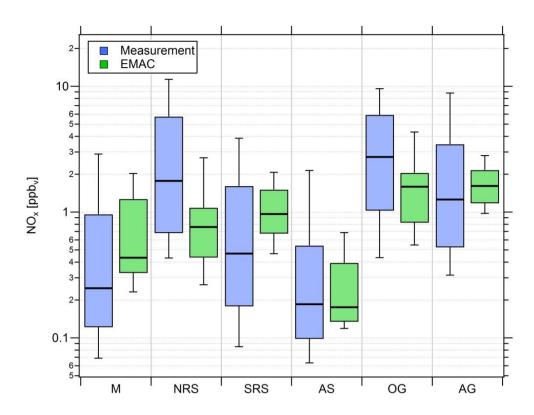


Figure 4: Comparison of measured (blue) and simulated (green) NO_x mixing ratios in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90-percentiles.

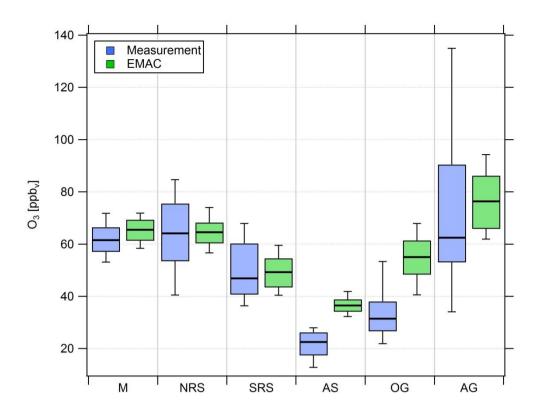


Figure 5: Comparison of measured (blue) and simulated (green) O₃ mixing ratios in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90-percentiles.

Overall, we find that NO_x mixing ratios over the Northern Red Sea, the Gulf of Oman and the Arabian Gulf are approximately one order of magnitude higher than in the other three regions (Southern Red Sea, Arabian Sea, Mediterranean). NO_x medians over the Arabian Gulf, the Northern Red Sea and the Gulf of Oman are 1.26 ppb_y, 1.76 ppb_y and 2.74 ppb_y, respectively. Lower median NO_x mixing ratios were measured over the Southern Red Sea (0.46 ppb_y), the Mediterranean (0.25 ppb_y) and the Arabian Sea (0.19 ppb_y). With respect to observed O₃ mixing ratios, the Arabian Sea is the only region representing remote MBL conditions with lowest median and average O₃ of 21.5 ppb_y and 22.5 ppb_y respectively, followed by the Gulf of Oman where median and mean O₃ were 31.5 ppb_y and 34 ppb_y, respectively. The low O₃ mixing ratios over the Arabian Sea were accompanied by the smallest variability (whisker-interval: 15.1 ppb_y). Although observing highest NO_x over the Oman Gulf, O₃ observed over the Oman Gulf was amongst the lowest detected throughout the whole campaign, which can partly be explained the fact that high NO_x eventually leads to ozone destruction. The immediate vicinity of point sources in this region, which leads to higher NO_x (before it is lost by reaction with OH and deposition to the surface) and titration of O₃ (note the relatively low regional O₃ median of 31.5 ppb_y), may partly explain why NO_x was highest over the Gulf of Oman. However, a significantly larger whisker-interval of observed ozone of 31.4 ppb_y over the Gulf of Oman indicates increasing amounts of pollution and advection from the Arabian Gulf where extreme events of ozone were observed several times during the

campaign with maximum mixing ratios of up to 170 ppb_v when wind was coming from Kuwait/Iraq. Please note that during the second leg wind was coming from Iran (Pfannerstill et al., 2019). The whisker-interval over the Arabian Gulf was 100.9 ppb_v, more than six times higher than that over the Arabian Sea. Reasons for large variations of both NO_x and O₃ over the Arabian Gulf were a multitude of point sources as well as a change in the observed wind direction with air masses coming from Iraq/Kuwait area during the first leg and air masses coming from Iran during the second leg (Pfannerstill et al., 2019). Over the Mediterranean, the Northern Red Sea and the Southern Red Sea, median ozone was 61.5 ppb_y, 64.2 ppb_y and 46.9 ppb_y, respectively. The whisker-intervals over the Northern Red Sea and the Southern Red Sea were 44.2 ppb_y and 31.6 ppb_y. respectively. Air masses over the Mediterranean were characterized as photochemically aged due to their impact by northerly winds (Etesians) which bring processed/oxidized air from eastern Europe (Turkey, Greece) to the Mediterranean area (Derstroff et al., 2017; Pfannerstill et al., 2019). This photochemical ageing/oxidation over the Mediterranean leads to a rather small whisker-interval of 18.7 ppb_v in ozone. In summary, median NO_x over the Oman Gulf was 56 % and 117 % higher than over the Northern Red Sea and the Arabian Gulf, respectively. However, the highest NO_x average was measured over the Northern Red Sea at 4.69 ppb_y, similar to the values observed over the Oman Gulf (4.16 ppb_y) and the Arabian Gulf (3.65 ppb_y). Note that highest NO_x mixing ratios over the Oman Gulf and over the Northern Red Sea are not always associated with high O₃ mixing ratios. We find that average ozone was highest over the Arabian Gulf with 74 ppb_y followed by the Northern Red Sea region (63.4 ppb_v). The average ozone mixing ratio over the Oman Gulf was 34 ppb_v, which corresponds to 46 % of the value observed over the Arabian Gulf. Photochemically aged air masses over the Mediterranean Basin show an ozone average of 61.6 ppb_v and air masses encountered over the Northern Red Sea (O₃ median of 64.2 ppb_v, O₃ average of 63.4 ppb_v) are comparable to the Arabian Gulf.

500

505

510

515

520

525

Due to a number of large pollution sources in the region around the Arabian Peninsula such as passing ships, highly urbanized areas as well as on- and off-shore petrochemical processing, NO_x levels were rarely as low as those found in remote locations such as over the South Atlantic (Fischer et al., 2015) where NO_x levels may be under 20 ppt_v. Apart for a few occasions where NO_x was below 50 ppt_v for short periods (Arabian Sea, the Southern Red Sea and the Mediterranean), NO_x levels during AQABA generally ranged from 100 ppt_v up to several ppb_v. The campaign NO_x median of 0.65 ppb_v and mean value of (2.51 \pm 5.84) ppb_v is comparable to urban sites (Kleinman et al., 2005). A detailed emission density analysis performed by Johansson et al. (2017) shows that NO_x emissions on and around the Arabian Peninsula are amongst the highest worldwide, which could explain the rather high NO_x level in the MBL around the peninsula (Johansson et al., 2017; Pfannerstill et al., 2019). O_3 mixing ratios measured during AQABA were also very variable with O_3 mixing ratios ranging between less than 20 ppb_v in the remote MBL (Fischer et al., 2015) to 60-70 ppb_v in the Mediterranean (consistent with previous ship-based measurements in the region (Kouvarakis et al., 2002) and as high as 150 ppb_v measured over the Arabian Gulf region. The latter are consistent with O_3 mixing ratios reported from regions influenced by oil and gas processing (Pfannerstill et al., 2019) and shipping lanes such as the Houston Ship Channel (Mazzuca et al., 2016).

Figure 4 also shows that the general trend for NO_x mixing ratios in the different regions is widely reproduced by the EMAC model. We find that the median NO_x(model)/NO_x(measurement)-ratio of all five minute averaged data points of the whole campaign is 0.91, indicating that the model underestimates NO_x by roughly 10 %. The average ratio and its standard deviation are significantly larger at 2.57 and 5.71, respectively, indicating that single modeled data points strongly exceed the measurements, especially during periods of low in situ NO_x (see supplementary Figure S6). Particularly over the Arabian Sea and the Southern Red Sea, the model generally simulates NO_x mixing ratios higher than 100 and 200 ppt_y, respectively while the measurements indicate mixing ratios of less than 50 ppt_v for certain periods. Furthermore, as expected, the model is not able to reproduce point sources such as passing ships for which we observe a significant underestimation of the measured NO_x. For ozone we find that the median O₃(model)/O₃(measurement)-ratio throughout the campaign is 1.23, indicating that over the course of the campaign the model overestimates O₃ by about 23 %. This could partly be related to the same limitation, i.e. the inability of the model to resolve point sources in which O₃ is locally reduced due to titration by NO. While the model is in rather good agreement with the measurements over the Mediterranean, the Northern Red Sea and Southern Red Sea, large deviations are found over the Arabian Sea and the Oman Gulf, where the model overestimation with respect to the regional median is 63 % and 75 %, respectively. A possible explanation for the overestimation of both ozone and NO_x in pristine regions such as over the Arabian Sea and the Oman Gulf could be related to the model resolution of 1.1° x 1.1°. Interpolation of model simulations along the Kommandor Iona ship track close to the coast at this resolution will most likely incorporate contributions from nearby land areas, affected by anthropogenic emissions. See supplementary Table ST3 and Table ST4 for further information and Figure S6 and S7 for additional scatterplots of measured and simulated regional median NO_x and O₃, respectively.

3.2 Estimation of (HO₂+RO₂) around the Arabian Peninsula

530

535

540

545

550

555

Noontime (HO_2+RO_2) was estimated based on Eq. 3. As the steady state assumption will not hold for air masses originating from fresh emissions (times to acquire steady state estimated from the inverse sum of the loss and production terms for NO_2 typically ranged from 1-2 minutes during AQABA) and for fast changes in the actinic flux, we have calculated Box-Whisker-Plots for \pm 2 h around noontime for which we expect relatively minor changes in the actinic flux (Figure 6). The noontime of each day was approximated by applying a Gaussian fit routine to the measured $j(NO_2)$ values whereas $j(NO_2)$ values being less than 10^{-3} s⁻¹ were neglected. Due to the availability of OH and HO_2* data from 18 July 2017 onwards, we have limited the analysis to this period. Note that there are no noontime (HO_2+RO_2) estimates from 18 July to 21 July due to contamination by the ship exhaust and on 24 August 2017 due to missing data. The black bar in Figure 6 indicates the median value, with the Box-interval marking the 25- and 75-percentile and the whisker showing the 10- and 90-percentile. Figure 7 shows summarized regional trends of the (HO_2+RO_2) estimates for measured and simulated data.

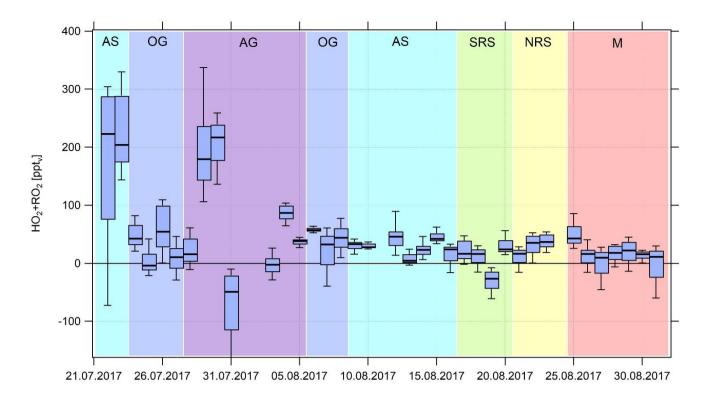


Figure 6: Timeline of median (HO₂+RO₂) noontime estimates from 22 July to 31 August 2017. Due to contamination by the ship exhaust itself, there is no data from 18 July to 21 July 2017. See annotations for the classification of the different regions.

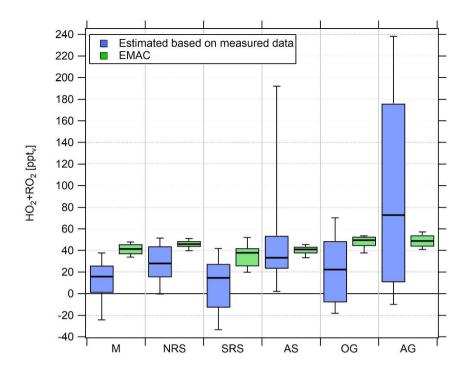


Figure 7: Comparison of Box-Whisker-Plots of the regional estimated noontime (HO₂+RO₂) median based on measured data and simulated (HO₂+RO₂) data for the period from 18 July 2017 onwards.

The relative uncertainty $R([HO_2] + [RO_2])$ associated with the (HO_2+RO_2) estimate has been calculated by error propagation of Eq. 3.

$$R([\mathrm{HO_2}] + [\mathrm{RO_2}]) =$$

$$\frac{1}{[HO_{2}] + [RO_{2}]} \cdot \sqrt{\frac{\left(\frac{\Delta j(NO_{2}) \cdot [NO_{2}]}{k_{NO+HO_{2}} \cdot [NO]}\right)^{2} + \left(\frac{\Delta [NO_{2}] \cdot j(NO_{2})]}{k_{NO+HO_{2}} \cdot [NO]}\right)^{2} + \left(\frac{\Delta [O_{3}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+HO_{2}} \cdot [NO]}\right)^{2} + \left(\frac{\Delta [NO_{2}] \cdot [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+HO_{2}} \cdot [NO] - k_{NO+HO_{2}} \cdot (j(NO_{2}) \cdot [NO_{2}] - k_{NO+O_{3}} \cdot [O_{3}][NO])}\right)^{2}} + \frac{1}{(k_{NO+HO_{2}} \cdot [NO])^{2}} \cdot \left(\frac{\Delta [NO]}{k_{NO+HO_{2}} \cdot [NO]}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{(k_{NO+HO_{2}} \cdot [NO])^{2}}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+O_{3}} \cdot [NO]}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+HO_{2}} \cdot [NO]}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+HO_{2}} \cdot [NO]}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+HO_{2}} \cdot [NO]}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+O_{3}} \cdot [NO]}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+O_{3}} \cdot [NO]}\right)^{2} + \frac{\Delta [NO_{2}] \cdot k_{NO+O_{3}} \cdot [NO]}{k_{NO+O_{3}} \cdot [NO]}$$

Over the course of the campaign, the median relative (HO_2+RO_2) uncertainty is 74 %. The average is 176 % and heavily biased by single data outliers and therefore not representative. The relative error associated with the (HO_2+RO_2) calculation is hence estimated at 74 %. Note that our calculation assumes that errors in the used rate coefficients are negligible.

We find median noontime (HO₂+RO₂) mixing ratios over the Mediterranean, the Northern Red Sea, the Southern Red Sea, the 575 Arabian Sea and Oman Gulf of 16 ppt_v, 28 ppt_v, 15 ppt_v, 33 ppt_v and 22 ppt_v, respectively, with each respective 75-percentile RO₂ being equal or less than 54 ppt_v. Only over the Arabian Gulf, the RO₂ estimate yields a median noontime mixing ratio of

73 ppt_v accompanied by the largest variations in the box-interval of the whole campaign. While the box-interval of the (HO₂+RO₂) estimate in the other regions is 25-57 ppt_v, the box-interval over the Arabian Gulf is significantly higher at 165 ppt_v. Negative values for all regions are regularly found in the vicinity of fresh emissions and air masses not in photochemical equilibrium. The elevated 90-percentile over the Arabian Sea is due to high (HO₂+RO₂) estimates during the first leg on 22 and 23 July.

Estimated (HO_2+RO_2) mixing ratios based on measured tracer data are in general agreement with previous studies performed in marine boundary layer environments which report maximum mixing ratios between 30 and 55 ppt_v around noontime (Hernandez et al., 2001). As peroxy radicals are short-lived molecules generated from the oxidation of VOCs, enhanced (HO_2+RO_2) concentrations observed over the Arabian Gulf are most likely due to high VOC emissions from intense oil and gas activities in the region (Bourtsoukidis et al., 2019; Pfannerstill et al., 2019). However high HO_2 and RO_2 can also occur in aged air masses with low NO_x and VOCs but still significant O_3 (and perhaps HCHO whose photolysis would then yield peroxy radicals). Bourtsoukidis et al. report that spatial volume mixing ratios of ethane and propane over the Arabian Gulf were about a factor of 10-15 times higher than over the Arabian Sea and the Southern Red Sea (Bourtsoukidis et al., 2019). We find that the median noontime (HO_2+RO_2) (measurement estimate)/ HO_2 (measurement)-ratio throughout the whole campaign is 1.88. Note that during single days, HO_2 may be higher than the (HO_2+RO_2) estimate, which is within the uncertainty of the (HO_2+RO_2) estimate.

EMAC modelled, median noontime (HO₂+RO₂) mixing ratios estimated as the sum of simulated HO₂ and all simulated peroxy radicals with less than four carbon molecules are 41 ppt_v, 46 ppt_v, 38 ppt_v, 41 ppt_v, 50 ppt_v and 49 ppt_v over the Mediterranean, the Northern Red Sea, the Southern Red Sea, the Arabian Sea, the Oman Gulf and the Arabian Gulf, respectively. The observation based (HO₂+RO₂) estimate yields 16 ppt_v, 28 ppt_v, 15 ppt_v, 33 ppt_v, 22 ppt_v and 73 ppt_v respectively. We find that the median point by point (HO₂+RO₂)(model)/(HO₂+RO₂)(measurement estimate)-ratio from 18 July onward is 1.05 so that, on average, the model overestimates the measurement by 5 %. Please note that the observational variability is much higher than the modeled one and that the median of 1.05 is accompanied by a larger average (1.84) and a large variability (42.51). See supplementary Table ST5 and ST6 for further information and Figure S8 for an additional scatterplot of measured and simulated regional median (HO₂+RO₂).

3.3 Net ozone production rates around the Arabian Peninsula

580

585

590

595

600

605

In the following, net ozone production rates (at noon) are calculated based on Eq. 7 for the different regions. These noontime values are scaled to diurnal production rates (Figure 8). As photochemical net ozone destruction is in good approximation linear with actinic flux $j(NO_2)$ and as on average (46.1 \pm 2.8) % of the total $j(NO_2)$ occurred \pm 2h around noon, the median noontime NOPR estimate was multiplied by $4/0.461 \approx 8.68$ to obtain a diurnal value. The error in the total actinic flux located \pm 2h around noon is estimated from the standard deviation of the best estimate of 0.461 at $\Delta s \approx 6$ %. Due to contamination

by the own ship exhaust and due to the availability of OH and HO₂* data only from 18 July 2017 onwards, we have limited the analysis to the period from 22 July 2017 to 31 August 2017. A comparison of NOPR estimated based on measured and simulated data for the different regions is shown in Figure 9. A break-down of the different terms of Eq. 7 in the six regions is included in the supplementary Figures S10-S13.

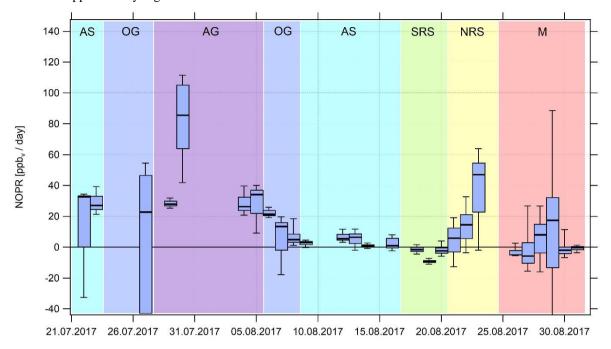


Figure 8: Timeline of the diurnal NOPR from 22 July to 31 August 2017. NOPR calculations are limited to the time period from 22 July onwards due to missing HO_x data and contamination by the ship exhaust itself before this period. See annotations for the classification of the different regions.

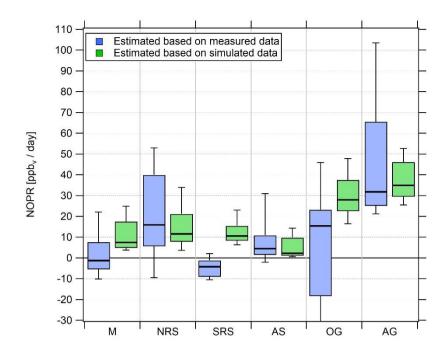


Figure 9: Diurnal net ozone production rates in the different regions. Related to the magnitude of pollution sources, the lower whisker of the NOPR estimate over the Oman Gulf is -324 ppb day⁻¹.

The relative uncertainty associated with the NOPR estimate has been calculated by error propagation of Eq. 7.

620

$$R(\text{NOPR}) = \frac{1}{\text{NOPR}} \cdot \frac{\left(k_{\text{NO+HO}_2} \cdot [\text{NO}] \cdot \Delta([\text{HO}_2] + [\text{RO}_2])\right)^2 + \left(k_{\text{NO+HO}_2} \cdot \Delta[\text{NO}] \cdot ([\text{HO}_2] + [\text{RO}_2])\right)^2 + \left(\Delta[\text{O}_3] \cdot \left(\alpha \cdot j(\text{O}^1\text{D}) + k_{\text{OH+O}_3} \cdot [\text{OH}] + k_{\text{HO}_2 + \text{O}_3} \cdot [\text{HO}_2]\right)\right)^2 + \left(\Delta j(\text{O}^1\text{D}) \cdot [\text{O}_3] \cdot \alpha)^2 + (j(\text{O}^1\text{D}) \cdot [\text{O}_3] \cdot \Delta\alpha)^2 + \left(\Delta[\text{OH}] \cdot k_{\text{OH+O}_3} \cdot [\text{O}_3]\right)^2 + \left(\Delta[\text{HO}_2] \cdot k_{\text{HO}_2 + \text{O}_3} \cdot [\text{O}_3]\right)^2}$$

- Incorporating a relative error of 74 % associated with (HO₂+RO₂), the median of the relative NOPR error of all data points obtained during AQABA is 91 %. The average relative uncertainty of NOPR is 21 % and strongly biased by single data outliers, which are in the case of NOPR significantly negative (due to fresh emissions and titration of O₃ by NO). Again the median is a more representative measure for the general uncertainty associated with the NOPR calculations. The relative error associated with the NOPR estimates based on measured data is hence estimated at 91 %.
- Over the Mediterranean and the Southern Red Sea, NOPR values do not significantly deviate from zero (production equals loss) within the atmospheric variability. The best estimate indicates slight net ozone destruction for the Mediterranean and Southern Red Sea (- 1 ppb day⁻¹) and (- 4 ppb day⁻¹) respectively, and slight net production for the Arabian Sea (5 ppb day⁻¹), which is significantly positive within the variability of the box-interval. Variations in NOPR calculated as the width of the 25-

75-percentile-box yield comparable values of 9-11 ppb day⁻¹ for these three regions. Substantial net ozone production was inferred over the Oman Gulf, the Northern Red Sea, and the Arabian Gulf with the median values being 16 ppb day⁻¹, 16 ppb day⁻¹ and 32 ppb day⁻¹, respectively. Especially over the Red Sea we find a strong latitudinal gradient in net ozone production rates with higher values towards the northern end, while slight net ozone destruction of -4 ppb day⁻¹ is reported over the southern part.

635

650

655

660

665

NOPR estimates for the Oman Gulf, the Northern Red Sea and the Arabian Gulf are comparable to results reported for dense traffic shipping routes such as the Houston Ship Channel with NOPR of a few tens of ppb h⁻¹ for periods of severe pollution (Zhou et al., 2014). Similar net ozone production rates have been reported for regions of Beijing in summer 2006 (Lu et al., 2010). For regions with low anthropogenic influence such as the Southern Red Sea and the Arabian Sea we estimate net ozone production that does not differ significantly from zero. This is due to the rather low NO_x mixing ratios in the clean marine boundary layer (Bozem et al., 2017). Note that we calculated net ozone destruction only for a few days over the Southern Red Sea and the Arabian Sea, indicating that the marine boundary layer around the Arabian Peninsula is rarely free from anthropogenic influence owing to the multitude of on- and off-shore anthropogenic activities.

We find that model-calculated estimates of NOPR reproduce the trends observed for NOPR calculated from in situ measurements except over the Mediterranean and the Southern Red Sea. Although EMAC predicts high ozone levels over the Arabian Sea, it also reports the lowest NOPR in this region. On the other side, the large overestimation of the model-calculated estimate NOPR against the one based on measured tracer data over the Mediterranean and over the Southern Red Sea could be linked to NO_x being overestimated in the model in these regions. In the model, pollution emissions, especially over the Oman Gulf and the Arabian Gulf, seem to be averaged over a large (1.1° grid size) region. High background concentrations of ozone precursors hence contribute to net ozone production rates that compare to conditions observed in the Houston case (Zhou et al., 2014). Even in the more pristine regions such as over the Southern Red Sea and the Arabian Sea, the model is not able to reproduce net ozone destruction, which is consistent with the fact the ozone is generally too high and that NO_x levels below 0.1 ppb_v are not found in the model. See supplementary Table ST7 and ST8 for further information and supplementary Figure S9 for an additional scatterplot of measured and simulated regional NOPR.

Measured OH and HO_2^* as well as (HO_2+RO_2) estimated based on measured data are generally underestimating the concurrent simulated data. Speaking in terms of absolute amounts, we find that the break-down loss and productions terms of Eq. 7 (NOPR) based on measured data are generally underestimating the results based on simulated data. The deviations between measurement and model pretty much represent the differences observed in the noontime concentrations of the mentioned tracers. Largest deviations of the break-down loss terms, associated with reactions of O_3 with OH and O_2 , are found over the OG and AG, where also OH and O_2 is significantly overestimated in the model. In the case of O_3 with OH and O_3 a slight overestimation by the estimate based on simulated data compared to the estimate based on measured data is observed. This is due to simulated absolute humidity being slightly larger than the concurrent measured data. Also we find that the break-down

production term $k_{\text{NO}+\text{HO}_2} \cdot [\text{NO}] \cdot ([\text{HO}_2] + [\text{RO}_2])$ estimated based on simulated data is generally larger than the estimate based on measured data. This pretty much reflects that noontime $(\text{HO}_2+\text{RO}_2)$ is overestimated in the model by a factor of 2, except for the Arabian Gulf where fair agreement is found.

3.4 VOC- and NO_x-sensitivity

670 Ozone is photochemically formed when the precursors NO_x and VOCs are abundant in the presence of sunlight (Bozem et al., 2017: Jaffe et al., 2018). In order to determine whether a chemical system is NO_x- or VOC-limited or in a transition between those two regimes, one has to estimate the total amount of OH reactivity towards VOCs and towards NO_x. Therefore the VOC/NO_x -ratio is an important indicator of the behavior of NO_x , VOCs and O_3 in a system. Since it is not feasible to precisely define all ambient VOCs (could be thousands), formaldehyde mixing ratios have been used as a proxy for the OH reactivity 675 towards VOCs since it is a short-lived oxidation product of many VOCs that is often positively correlated with peroxy radicals (Sillman et al., 1995; Duncan et al., 2010). Sillman et al. first used afternoon concentrations of indicator species such as HCHO and total reactive nitrogen (NO_v) to determine the sensitivity of ozone production to VOCs or NO_x (Sillman et al., 1995). Their approach was later successfully transferred to space-based satellite observations by using the ratio of tropospheric columns of HCHO and NO₂ to determine the sensitivity of ozone production (Martin et al., 2004). Here we use HCHO/NO₂-ratios (referred 680 to as "Ratio") deduced by Duncan et al. as indicators for the sensitivity of ozone production to NO_x- and VOC-limitations in megacities in the United States with large amounts of anthropogenic NO_x and VOC emissions (Duncan et al., 2010). The Ratio is an indicator of surface photochemistry as most of the atmospheric column of HCHO and NO₂ is located in the planetary boundary layer (Duncan et al., 2010). Duncan et al. have derived NO_x-limited ozone production regimes for HCHO/NO₂ > 2 and VOC-limited ozone production for HCHO/NO₂ < 1 (Duncan et al., 2010). For 1 < HCHO/NO₂ < 2 both NO_x and VOC 685 emission reductions may lead to a reduction in ozone. Figure 10 shows the Box-Whisker-Plot classification of the HCHO/NO₂ratio of the different regions during noontime.

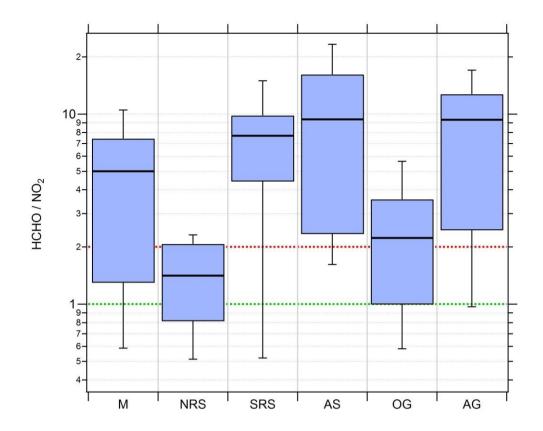


Figure 10: Box-Whisker-Plots of the HCHO/NO₂-ratio for the different regions with the black bar indicating the median value. Red (ratio = 2) and green (ratio = 1) lines indicate the limits for HCHO/NO₂ deduced by Duncan et al. (2010) for NO_x-limitation and VOC-limitation, respectively.

Median HCHO/NO₂-ratios of 5, 7.7, 9.4 and 9.3 over the Mediterranean, the Southern Red Sea, the Arabian Sea and the Arabian Gulf respectively indicate tendencies towards NO_x-limited regimes. In a previous study based on measured OH reactivity, Pfannerstill et al. classified these regions as being mostly in a transition between NO_x- and VOC-limitation, with a tendency towards NO_x-limitation (2019). Median HCHO/NO₂-ratios of 1.4 and 2.2 estimated over the Northern Red Sea and the Oman Gulf signify tendencies towards VOC-limitation. However, none of the medians of the six regions falls below the VOC-limit deduced by Duncan et al. (2010).

695

700

Over the Red Sea we find a latitudinal gradient in the HCHO/NO₂-ratio, similar to the gradients for NO_x and NOPR. Due to very low NO_x over the Southern Red Sea, O_3 production is NO_x -limited, changing into a more VOC-limited regime over the Northern Red Sea. Ozone production over the Mediterranean was classified as rather NO_x -limited, however partly being in the transition regime between NO_x - and VOC-limitation, which can be explained by measurements obtained on 29 August 2017 when laying at anchor in front of Malta with a multitude of (NO_x) -emissions from nearby situated vessels. Average noontime NO_x on that particular day was about three times as large as the regional average noontime NO_x observed over the whole

Mediterranean area. NO_x limitation is also inferred for the relatively clean Arabian Sea and the polluted Arabian Gulf atmosphere. Note that a further increase in NO_x-emissions from shipping in the Arabian Gulf may initially lead to higher ozone production. However, a further increase in NO_x might eventually lead to a change from NO_x- to VOC-sensitivity and a decrease in ozone production for this region, as observed for the Oman Gulf (median HCHO/NO₂-ratio of 2.2 and average O₃ of 34 ppb_v). See supplementary Table ST9 for detailed statistics on regional HCHO/NO₂-ratios.

4 Conclusion

710

715

720

725

730

735

In situ observations of NO, NO₂, O₃, HCHO, OH, HO₂, absolute humidity, actinic flux, temperature and pressure were carried out in the marine boundary layer around the Arabian Peninsula during the AQABA ship campaign from late June to early September 2017. Concentration ranges of both NO_x and O₃ clearly showed anthropogenic influence in the MBL. NO_x was highest over the Arabian Gulf, the Northern Red Sea and the Oman Gulf. Lowest NO_x was observed over the Arabian Sea and over the Southern Red Sea during the second leg. O₃ mixing ratios were highest over the Arabian Gulf. We observed a latitudinal gradient in O₃ concentrations with higher values towards the northern part of the Red Sea. Although comparable O₃ averages were measured over the Northern Red Sea and over the Mediterranean, lower variability over the Mediterranean towards the end of August 2017 indicates photochemically more extensively aged air masses. The lowest regional O₃ mixing ratio average was detected over the Arabian Sea, which is broadly comparable to remote marine boundary layer conditions in the Northern Hemisphere.

Noontime (HO₂+RO₂) estimates based on deviations from the Leighton Ratio yield median values around the Arabian Peninsula amount to 15 – 33 ppt_v for all regions except over the Arabian Gulf where the median is 73 ppt_v. The uncertainty due to the missing up-welling actinic flux portion is expected to be insignificant. Furthermore, we estimated noontime and diurnal NOPR based on Eq. 6 and the integral over the actinic flux. Highest diurnal NOPR were observed over the Oman Gulf, the Northern Red Sea and the Arabian Gulf with median values of 16 ppb_v day⁻¹, 16 ppb_v day⁻¹ and 32 ppb_v day⁻¹, respectively, which is in agreement with previous studies that predicted net photochemical O₃ formation conditions in the region. Net ozone destruction was only observed for a few days with clean conditions over the Arabian Sea and the Southern Red Sea. Based on HCHO/NO₂-ratios our analysis suggests tendencies towards NO_x-limitation over the Mediterranean, the Southern Red Sea, the Arabian Sea and the Arabian Gulf and VOC-limitation over the Northern Red Sea and the Oman Gulf, which reproduces the trends observed by Pfannerstill et al. (2019).

 NO_x results from the atmospheric chemistry – general circulation model EMAC underestimate the measurement data by 10 % whereas median modeled O_3 overestimates the measurement by 23 %, the latter being related to limitations in model resolution in coastal proximity and near shipping lanes. Although EMAC generally reproduces regional NO_x and O_3 medians, the scatter when comparing both data sets is large. NO_x is generally too low as it does not resolve local point sources and too high for

clean regions. Lowest NO_x of less than 0.1 ppb_v found in the in situ measurements is not reproduced by the model as emissions are averaged over a large area (1.1°) . Median noontime (HO_2+RO_2) retrieved from the EMAC model are ~ 5 % higher than (HO_2+RO_2) estimates based on measurement data, however, (HO_2+RO_2) deduced from EMAC is sometimes about a factor of 2 higher than the regional (HO_2+RO_2) estimate based on the Leighton Ratio and measured tracer data. NOPR estimates based on modeled data reproduce the tendencies derived from the measurements very well. However, the model does not reproduce observed net ozone destruction along some clean parts of the ship cruise.

Data availability

740

750

Data used in this study is available to all scientists agreeing to the AQABA protocol at https://doi.org/10.5281/zenodo.3693988.

745 Author contributions

IT, HF and JL designed the study. UP and IT performed the CLD NO and NO₂ measurements and processed the data. JC and PE performed the O₃ measurements, JS performed the actinic flux measurements. JS performed cavity ring-down spectroscopy measurements of NO₂. DD and BH performed the HCHO measurements. HH, MM, RR, ST performed the OH and HO₂ measurements. J-DP was responsible for the H₂O measurements. Model simulations were made by AP. All authors have contributed to writing this manuscript

Competing interests

The authors declare no conflict of interest.

Acknowledgements

We acknowledge the collaborations with King Abdullah University of Science and Technology (KAUST), The Cyprus Institute (CyI) and the Kuwait Institute for Scientific Research (KISR). We would like to thank Marcel Dorf and Claus Koeppel for the organization of the campaign. We would like to thank Hays Ships Ltd. and the ship crew for caring about the physical well-being of the scientific participants and for an unforgettable time on board *Kommandor Iona*. Last but not least we would like to thank the whole AQABA community for a successful campaign.

760

Appendix: Acronyms and abbreviations

General

AQABA <u>Air Quality and Climate in the Arabian Basin campaign</u>

CyI The Cyprus Institute

765 KAUST King Abdullah University of Science and Technology

KISR Kuwait Institute for Scientific Research

Regions

AG
AS
Arabian Gulf
AS
Arabian Sea

770 M
Mediterranean Sea
NRS
OG
Oman Gulf
SRS
Southern Red Sea

Scientific

775 CLD <u>Chemiluminescence detector</u> CRDS <u>Cavity ring-down spectroscopy</u>

ECHAM5 Fifth generation European Centre Hamburg general circulation model

EMAC <u>ECHAM/MESSy Atmospheric Chemistry model</u>

FWHM Full width at half maximum

780 GC-FID $\underline{\mathbf{G}}$ as $\underline{\mathbf{c}}$ hromatography – $\underline{\mathbf{f}}$ lame $\underline{\mathbf{i}}$ onization $\underline{\mathbf{d}}$ etector

HORUS HydrOxyl Radical measurement Unit based on fluorescence Spectroscopy instrument

 HO_x $OH + HO_2$

LED \underline{L} ight \underline{e} mitting \underline{d} iodeLIF \underline{L} aser \underline{i} nduced \underline{f} luorescenceMBL \underline{M} arine \underline{b} oundary \underline{l} ayer

MESSy <u>M</u>odular <u>E</u>arth <u>S</u>ubmodel <u>Sy</u>stem

NOPR NOPR <u>Net ozone production rate</u>

 $\begin{array}{ll} NO_x & NO + NO_2 \\ PFA & \underline{\textbf{P}}er\underline{\textbf{f}}luoro\underline{\textbf{a}}lkoxy \end{array}$

790 PSS <u>Photostationary steady state</u>

PTFE Polytetrafluoroethylene
SLM Standard liter per minute

STEAM3 Ship Traffic Emission Assessment Model 3

TMU <u>T</u>otal <u>measurement uncertainty</u>
VOC Volatile organic compounds

UV <u>U</u>ltra<u>v</u>iolet

795

785

References

825

Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G., Jenkin, M. E., Rossi, M. J., and Troe, J.: Atmos. Chem. Phys., 4, 1461, 2004; IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation, (http://iupac.pole-ether.fr).

Bourtsoukidis, E., Ernle, L., Crowley, J. N., Lelieveld, J., Paris, J.-D., Pozzer, A., Walter, D., and Williams, J.: Non Methane Hydrocarbon (C2-C8) sources and sinks around the Arabian Peninsula, doi:10.5194/acp-2019-92, 2019.

Bozem, H., Butler, T.M., Lawrence, M. G., Harder, H., Martinez, M., Kubistin, D., Lelieveld, J., and Fischer, H: Chemical processes related to net ozone tendencies in the free troposphere, Atmos. Chem. Phys., 17, 10565-10582, doi:10.5194/acp-17-10565-2017, 2017.

810 Cantrell, C. A., Shetter, R. E., Calvert, J. G., Eisele, F. L., Williams, E., Baumann, K., William, H. B., Stevens, P. S., Mather, J., H.: Peroxy radicals from photostationary state deviations and steady state calculations during the Tropospheric OH Photochemistry Experiment at Idaho Hill, Colorado, 1993, J. Geophys. Res., 102, 6369-6378, 1997.

Celik, S., Drewnick, F., Fachinger, F., Brooks, J., Darbyshire, E., Paris, J.-D., Eger, P. G., Schuladen, J., Tadic, I., Friedrich, N., Dienhart, D., Crowley, J. N., Harder, H., and Borrmann, S.: Influence of vessel characteristics and atmospheric processes on the gas and particle phase of ship emission plumes measured in the Mediterranean Sea and around the Arabian Peninsula, Atmos-. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2019-859, in review, 2019.

Chen, S., Ren, X., Mao, J., Chen, Z., Brune, W. H., Lefer, B., Rappenglück, B., Flynn, J., Olson, J., Crawford, J. H.: A comparison of chemical mechanisms based on TRAMP-2006 field data, Atmos. Environ., 44, 4116-4125, doi: 10.1016/j.atmosenv.2009.05.027, 2009.

820 Crutzen, P. J.: Photochemical reactions initiated by and influencing ozone in unpolluted tropospheric air, Tellus, 26, 47–57, doi:10.1111/j.2153-3490.1974.tb01951.x, 1973.

Derstroff B., Hüser, I., Bourtsoukidis, E., Crowley, J. N., Fischer, H., Gromov, S., Harder, H., Janssen, R. H. H., Kesselmeier, J., Lelieveld, J., Mallik, C., Martinez, M., Novelli, A., Parchatka, U., Phillips, G. J., Sander, R., Sauvage, C., Schuladen, J., Stönner, C., Tomsche, L., and Williams, J.: Volatile organic compounds (VOCs) in photochemically aged air from the eastern and western Mediterranean, Atmos. Chem. Phys., 17, 9547-9566, doi:10.5194/acp-17-9547-2017, 2017.

Duncan, B. N., Yoshida, Y., Olson, J. R., Sillman, S., Martin, R. V., Lamsal, L., Hu, Y., Pickering, K. E., Retscher, C., Allen, D. J., Crawford, J. H.: Application of OMI observations to a space-based indicator of NO_x and VOC controls on surface ozone formation, J. Atmos. Env., 44, 2213-2223, doi:10.1016/j.atmosenv.2010.03.010, 2010.

Drummond, J.W., Volz, A., and Ehhalt, D. H.: An optimized chemiluminescence detector for tropospheric NO measurements, 330 J. Atmos. Chem., 2, 287–306, doi:10.1007/BF00051078, 1985.

Fischer, H., Pozzer, A., Schmitt, T., Jöckel, P., Klippel, T., Taraborrelli, D., and Lelieveld, J.: Hydrogen peroxide in the marine boundary layer over the South Atlantic during the OOMPH cruise in March 2007, Atmos. Chem. Phys., 15, 6971-6980, doi:10.5194/acp-15-6971-2015, 2015.

Hauglustaine, D. A., Madronich, S., Ridley, B. A., Walega, J. G., Cantrell, C. A., and Shetter, R. E.: Observed and model calculated photostationary state at Mauna Loa Observatory during MLOPEX 2, J. Geophys. Res., 101, 14681-14696,
 doi:10.1029/95JD03612, 1996.

Hens, K., Novelli, A., Martinez, M., Auld, J., Axinte, R., Bohn, B., Fischer, H., Keronen, P., Kubistin, D., Nölscher, A. C., Oswald, R., Paasonen, P., Petäjä, T., Regelin, E., Sander, R., Sinha, V., Sipilä, M., Taraborrelli, D., Tatum Ernest, C., Williams, J., Lelieveld, J., Harder, H.: Observation and modelling of HO_x radicals in a boreal forest, Atmos. Chem. Phys., 14, 8723-8747, doi:10.5194/acp-14-8723-2014, 2014.

Hernandez, M. D. A., Burkert, J., Reichert, L., Stöbener, D., Meyer-Arnek, J., and Burrows, J. P.: Marine boundary layer peroxy radical chemistry during the AEROSOLS99 campaign: Measurements and analysis, J. Geophys. Res., 106, 20833-20846, doi:10.1029/2001JD900113, 2001.

Hollaway, M. J., Arnold, S. R., Challinor, A. J., and Emberson, L. D.: Intercontinental trans-boundary contributions to ozoneinduced crop yield losses in the Northern Hemisphere, Biogeosciences, 9, 271-292, doi:10.5194/bg-9-271-2012, 2012.

Hosaynali Beygi, Z., Fischer, H., Harder, H. D., Martinez, M., Sander, R., Williams, J., Brookes, D. M., Monks, P. S., and Lelieveld, J.: Oxidation photochemistry in the Southern Atlantic boundary layer: unexpected deviations of photochemical steady state, Atmos. Chem. Phys., 11, 8497-8513, doi:10.5194/acp-11-8497-8513, 2011.

Jaffe, D. A., Cooper, O. R., Fiore, A. M., Henderson, B. H., Tonnesen, G. S., Russell, A. G., Henze, D. K., Langford, A. O., Lin, M., and Moore, T.: Scientific assessment of background ozone over the U.S.: Implications for air quality management, Elem. Sci. Anth., 6, 56, doi:10.1525/elementa.309, 2018.

Javed, U., Kubistin, D., Martinez, M., Pollmann, J., Rudolf, M., Parchatka, U., Reiffs, A., Thieser, J., Schuster, G., Horbanski,
 M., Pöhler, D., Crowley, J. N., Fischer, H., Lelieveld, J., and Harder, H.: Laser-induced fluorescence-based detection of atmospheric nitrogen dioxide and comparison of different techniques during the PARADE 2011 field campaign, Atmos. Meas.
 Tech., 12, 1461-1481, doi:10.5194/amt-12-1461-2019, 2019.

Jöckel, P., Kerkweg A., Pozzer, A., Sander, R., Tost, H., Riede, H., Baumgartner, A., Gromov, S., and Kern, B.: Development of cycle 2 of the Modular Earth Submodel System (MESSy2), Geosci. Model Dev., 3, 717-752, doi:10.5194/gmd-3-717-2010, 2010.

Johansson, L., Jalkanen, J.-P., Kukkonen, J.: Global assessment of shipping emissions in 2015 on a high spatial a temporal resolution, J. Atmos. Env., 167,403-415, doi:10.1016/j.atmosenv.2017.08.042., 2017.

Keller-Rudek, H., Moortgat, G. K., Sander, R., and Sörensen, R.: The MPI-Mainz UV/VIS Spectral Atlas of Gaseous Molecules of Atmospheric Interest, Earth Syst. Sci. Data, 5, 365-373, doi:10.5194/essd-5-365-2013, 2013.

Kleinman, L. I., Daum, P. H., Lee, Y.-N., Nunnermacker, L. J., Springston, S. R. Weinstein-Lloyd, J., and Rudolph, J.: A comparative study of ozone production in five U.S. metropolitan areas, J. Geophys. Res.-Atmos., 110, D02301, doi:10.1029/2004jd005096, 2005.

Klonecki, A. and Levy, H.: Tropospheric chemical ozone tendencies in CO-CH₄-NO_y-H₂O system: Their sensitivity to variations in environmental parameters and their application to a global chemistry transport model study, J. Geophys. Res., 102, 21221–21237, doi:10.1029/97JD01805, 1997.

Kormann, R., Fischer, H., de Reus, M., Lawrence, M., Brühl, Ch., von Kuhlmann, R., Holzinger, R., Williams, J., Lelieveld,
 J., Warneke, C., de Gouw, J., Heland, J., Ziereis, H., and Schlager, H.: Formaldehyde over the eastern Mediterranean during MINOS: Comparison of airborne in-situ measurements with 3D-model results, Atmos. Chem. Phys., 3, 851-861, doi:10.5194/acp-3-851-2003, 2003.

Kouvarakis, G., Vrekoussis, M., Mihailopoulos, N., Kourtidis, K., Rappenglueck, B., Gerasopoulos, E., and Zerefos, C.: Spatial and temporal variability of tropospheric ozone (O3) in the boundary layer above the Aegean Sea (eastern Mediterranean), J. Geophys. Res., 107, D18, 8137, doi:10.1029/2000JD000081, 2002.

Krotkov, N. A., McLinden, C. A., Li, C., Lamsal, L. N., Celarier, E. A., Marchenko, S. V., Swartz, W. H., Bucsela, E. J., Joiner, J., Duncan, B. N., Boersma, K. F., Veefkind, J. P., Levelt, P. F., Fioletov, V. E., Dickerson, R. R., He, H., Lu, Z., and

- Streets, D. G.: Aura OMI observations of regional SO₂ and NO₂ pollution changes from 2005 to 2015, Atmos. Chem. Phys., 16, 4605–4629, doi:10.5194/acp-16-4605-2016, 2016.
- Kwok, C. Y., Laurent, O., Guemri, A., Philippon, C., Wastine, B., Rella, C. W., Vuillemin, C., Truong, F., Delmotte, M., Kazan, V., Darding, M., Lebegue, B., Kaiser, C., Xueref-Remy, I., and Ramonet, M.: Comprehensive laboratory and field testing of cavity ring-down spectroscopy analyzers measuring H₂O, CO₂, CH₄ and CO, Atmos. Chem. Phys., 8, 3867-3892, doi:10.5194/amt-8-3867-2015, 2015.
 - Leighton, P. A.: Photochemistry of air pollution, Phys. Chem., 9, 1961.
- Lelieveld. J., Dentener, F. J., Peters, W., and Krol, M. C.: On the role of hydroxyl radicals in the self-cleansing capacity of the troposphere, Atmos. Chem. Phys., 4, 2337-2344, doi:10.5194/acp-4-2337-2004, 2004.
 - Lelieveld, J., Hoor, P., Jöckel, P., Pozzer, A., Hadjinicolaou, P., Cammas, J.-P., and Beirle, S.: Severe ozone air pollution in the Persian Gulf region, Atmos. Chem. Phys., 9, 1393–1406, doi:10.5194/acp-9-1393-2009, 2009.
- Lelieveld, J.: Strongly increasing heat extremes in the Middle East and North Africa (MENA) in the 21st century, Climate Change, 137, 245-260, doi:10.1007/s10584-016-1665-6, 2016a.
 - Lelieveld, J., Gromov, S., Pozzer, A., and Taraborrelli, D.: Global tropospheric hydroxyl distribution, budget and reactivity, Atmos. Chem. Phys., 16, 12477-12493, doi:10.5194/acp-16-12477-2016, 2016b.
- Liu, F., Beirle, S., Zhang, Q., Dörner, S., He, K., and Wagner, T.: NO_x lifetimes and emissions of cities and power plants in polluted background estimated by satellite observations, Atmos. Chem. Phys., 16, 5823-5298, doi:10.5194/acp-16-5283-2016, 2016.
 - Lu, K., Zhang, Y., Su, H., Brauers, T., Chou, C., Hofzumahaus, A., Liu, S., Kita, K., Kondo, K., Shao, M., Wahner, A., Wang, J., Wang, X., and Zhu, T.: Oxidant (O₃+NO₂) production processes and formation in regimes in Beijing, J. Geophys. Res., 115, D07303, doi:10.1029/2009JD012714, 2010.
- Mallik, C., Tomsche, L., Bourtsoukidis, E., Crowley, J. N., Derstroff, B., Fischer, H., Hafermann, S., Hüser, I., Javed, U., Keßel, S., Lelieveld, J., Martinez, M., Meusel, H., Novelli, A., Phillips, G. J., Pozzer, A., Reiffs, A., Sander, R., Taraborrelli, D., Sauvage, C., Schuladen, J., Su, H., Williams, J., and Harder, H.: Oxidation processes in the eastern Mediterranean atmosphere: evidence from the modelling of HO_x measurements over Cyprus, Atmos. Chem. Phys., 18, 10825-10847, doi:10.5194/acp-18-10825-2018, 2018.

Mannschreck, K., Gilge, S., Plass-Duelmer, C., Fricke, W., and Berresheim, H.: Assessment of the applicability of NO-NO₂-O₃ photostationary state to long-term measurements at the Hohenpeissenberg GAW Station, Germany, Atmos. Chem. Phys., 4, 1265-1277, doi:10.5194/acp-4-1265-2004, 2004.

Mao, J., Ren, X., Chen, S., Brune, W. H., Chen, Z., Martinez, M., Harder, H., Lefer, B., Rappenglück, B., Flynn, J., Leuchner, M.: Atmospheric oxidation capacity in the summer of Houston 2006: Comparison with summer measurements in the metropolitan studies, Atmos. Environ., 44, 4107-4115, doi:10.1016/j.atmosenv.2009.01.013, 2009.

910 Martin R., Fiore, A., Van Donkelaar, A. 2004: Space-based diagnosis of surface ozone sensitivity to anthropogogenic emissions, Geophys. Res. Lett., 31, L06120, doi:10.1029/2004GL019416, 2004.

Martinez, M., Harder, H., Kubistin, D., Rudolf, M., Bozem, H., Eerdekens, G., Fischer, H., Klüpfel, T., Gurk, C., Königstedt, R., Parchatka, U., Schiller, C. L., Stickler, A., Williams, J., and Lelieveld, J.: Hydroxyl radicals in the tropical troposphere over the Suriname rainforest: airborne measurements, Atmos. Chem. Phys., 10, 3759-3773, doi: 10.5194/acp-10-3759-2010, 2010.

915

930

Mazzuca G. M., Ren, X., Loughner, C. P., Estes, M., Crawford, J. H., Pickering, K. E., Weinheimer, A. J., and Dickerson, R. R.: Ozone production and its sensitivity to NOx and VOCs: results from the DISCOVER-AQ field experiment, Houston 2013, Atmos. Chem. Phys., 16, 14463-14474, doi:10.5194/acp-16-14463-2016, 2016.

Meusel, H., Kuhn, U., Reiffs, A., Mallik, C., Harder, H., Martinez, M., Schuladen, J., Bohn, B., Parchatka, U., Crowley, J. N., Fischer, H., Tomsche, L., Novelli, A., Hoffmann, T., Janssen, R. H. H., Hartogensis, O., Pikridas, M., Vrekoussis, M., Bourtsoukidis, E., Weber, B., Lelieveld, J., Williams, J., Pöschl, U., Cheng, Y., and Su, H.: Daytime formation of nitrous acid at a coastal remote site in Cyprus indicating a common ground source of atmospheric HONO and NO, Atmos. Chem. Phys., 16, 14475-14493, doi:10.5194/acp-16-14475-2016, 2016.

Miyazaki, K., Eskes, H., Sudo, K., Folkert Boersma, K., Bowman, K., and Kanaya, Y.: Decadal changes in global surface NO_x emissions from multi-constituent satellite data assimilation, Atmos. Chem. Phys., 17, 807-837, doi:10:5194/acp-17-807-2017, 2017.

Monks, P. S., Archibald, A. T., Colette, A., Cooper, O. Coyle, M., Derwent, R., Fowler, D., Granier, C., Law, K. S., Mills, G. E., Stevenson, D. S., Tarasova, O., Thouret, V., von Schneidemesser, E., Sommariva, R., Wild, O., Williams, M. L.: Tropospheric ozone and its precursors from the urban to the global scale from air quality to short-lived climate forcer, Atmos. Chem. Phys., 15, 8889-8973, doi:10.5194/acp-15-8889-2015, 2015.

- Nakamura, K., Kondo, Y., Chen, G., Crawford, J. H., Takegawa, N., Koike, M., Kita, K., Miyazaki, Y., Shetter, R. E., Lefer, B. L., Avery, M., and Matsumoto J.: Measurement of NO₂ by the photolysis conversion technique during the Transport and Chemical Evolution Over the Pacific (TRACE-P) campaign, J. Geophys. Res., 108, D24, 4752, doi:10.1029/2003JD003712, 2003.
- Parrish, D. D., Trainer, M., Williams, E. J., Fahey D. W., Hübler, G., Eubank, C. S., Liu S. C., Murphy, P. C., Albritton, D. L., Fehsenfeld, F. C.: Measurements of the NO_x-O₃-photostationary steady state at Niwot Ridge, Colorado, J. Geophys. Res., 91,5361-5370, doi:10.1029/JD091iD05p05361, 1986.
 - Pollack, I. B., Lerner, B. M., Ryerson, T. B.: Evaluation of ultraviolet light-emitting diodes for detection of atmospheric NO₂ by photolysis chemiluminescence, J. Atmos. Chem., 65, 111-125, doi:10.1007/s10874-011-9184-3, 2011.
- Pfannerstill, E. Y., Wang, N., Edtbauer, A., Bourtsoukidis, E., Crowley, J. N., Dienhart, D., Eger, P. G., Ernle, L., Fischer, H., Hottmann, B., Paris, J.-D., Stönner, C., Tadic, I., Walter, D., Lelieveld, J., Williams, J.: Shipborne measurements of total OH reactivity around the Arabian Peninsula and its role in ozone chemistry, Atmos. Chem. Phys., 19, 11501-11523, doi:10.5194/acp-19-11501-2019, 2019.
- Reed, C., Evans, M. J., Di Carlo, P., Lee, J. D., and Carpenter, L. J.: Interferences in photolytic NO₂ measurements: explanation for an apparent missing oxidant?, Atmos. Chem. Phys., 16, 4707-4724, doi:10.5194/acp-16-4707-2016, 2016.
 - Regelin, E., Harder, H., Martinez, M., Kubistin, D., Tatum Ernest, C., Bozem, H., Klippel, T., Hosaynali Beygi, Z., Fischer, H., Sander, R., Jöckel, P., Königstedt, R., and Lelieveld, J.: HO_x measurements in the summertime upper troposphere over Europe: a comparison of observations to a box model and a 3-D model, Atmos. Chem. Phys. Phys., 13, 10703-10720, doi:10.5194/acp-13-10703-2013, 2013.
- Ren, X., van Duin, D., Cazorla, M., Chen, S., Mao, J., Zhang, L., Brune, W. H., Flynn, J. H., Grossberg, N., Lefer, B. L., Rappenglück, B., Wong, K. W., Tsai, C., Stutz, J., Dibb, J. E., Jobson, B. T., Luke, W. T., and Kelley, P.: Atmospheric oxidation chemistry and ozone production: Results from SHARP 2009 in Houston, Texas, J. Geophys. Res., 118, 5570-5780, doi:10.1002/jgrd.50342, 2013.
- Roeckner, E., Brokopf, R., Esch, M., Giorgetta, M., Hagemann, S., Kornblueh, L., Manzini, E., Schlese, U., and Schulzweida, U.: Sensitivity of Simulated Climate to Horizontal and Vertical Resolution in the ECHAM5 Atmosphere Model, J. Climate, 19, 3771-3791, doi:10.1175/JCLI3824.1, 2006.

- Ryerson, T. B., Williams, E. J., and Fehsenfeld, F. C.: An efficient photolysis system for fast-response NO₂ measurements, J. Geophys. Res., 105, D21, 26447-26461, doi:10.1029/2000JD900389, 2000.
- Sander, R., Baumgartner, A., Cabrera-Perez, D., Frank, F., Gromov, S., Grooß, J.-U., Harder, H., Huijnen, V., Jöckel, P., Karydis, V. A., Niemeyer, K. E., Pozzer, A., Riede, H., Schultz, M. G., Taraborrelli, D., and Tauer, S.: The community atmospheric chemistry box model CAABA/MECCA-4.0, Geosci. Model Dev., 12, 1365-1385, doi: 10.5194/gmd-12-1365-2019, 2019.
- Sobanski, N., Schuladen, J., Schuster, G., Lelieveld, J., and Crowley, J. N.: A five-channel cavity ring-down spectrometer for the detection of NO₂, NO₃, N₂O₅, total peroxy nitrates and total alkyl nitrates, Atmos. Meas. Techn., 9, 5103-5118, doi:10.5194/amt-9-5103-2016, 2016.
 - Sillman, S., Al-Wali, K., Marsik, F. J., Nowacki, P., Samson, P. J., Rodgers, M. O., Garland, L. J., Martinez, J. E., Stoneking, C., Imhoff, R., Lee, J. H., Newman, L., Weinstein-Lloyd, J., and Aneja, V.: Photochemistry of ozone formation in Atlanta, GA models and measurements, Atmos. Environ., 29, 21, 3055-3066., doi:10.1016/1352-2310(95)00217-M, 1995.
- Thornton, J. A., Wooldridge, P. J., Cohen, R. C., Martinez, M., Harder, H., Brune, W. H., Williams, E. J., Roberts, J. M., Fehsenfeld, F. C., Hall, S. R., Shetter, R. E., Wert, B. P., and Fried, A.: Ozone production rates as a function of NO_x abundances and HO_x production rates in the Nashville urban plume, J. Geophys. Res., 107, D12, doi:10.1029/2001JD000932, 2002.
 - Tuzson, B., Zeyer, K., Steinbacher, M., McManus, J. B., Nelson, D. D., Zahniser, M. S., and Emmenegger, L.: Selective measurements of NO, NO₂ and NO_y in the free troposphere using quantum cascade laser spectroscopy, Atmos. Meas. Tech., 6, 927-936, doi:10/5194/amt-6-927-2013, 2013.
- 975 Viallon, J., Lee, S., Moussay, P., Tworek, K., Petersen, M., and Wielgosz, R. I.: Accurate measurements of ozone absorption cross-sections in the Hartley band, Atmos. Meas. Tech., 8, 1245-1257, doi:10.5194/amt-8-1245-2015, 2015.
 - Velchev, K., Cavalli, F., Hjorth, J., Vignati, E., Dentener, F., and Raes, F.: Ozone over the Western Mediterranean Sea results from two years of shipborne measurements, Atmos. Chem. Phys., 11, 675-688, doi:10.5194/acp-11-675-2011, 2011.
- Zhou, W., Cohan, D. S., and Henderson, B. H.: Slower ozone production in Houston, Texas following emission reductions: evidence from Texas Air Quality Studies in 2000 and 2006, Atmos. Chem. Phys., 14, 2777-2788, doi:10.5194/acp-14-2777-2014, 2014.

Supplement of

1000

Net ozone production and its relationship to NO_x and VOCs in the marine boundary layer around the Arabian Peninsula

Ivan Tadic¹, John N. Crowley¹, Dirk Dienhart¹, Philipp Eger¹, Hartwig Harder¹, Bettina Hottmann¹, Monica Martinez¹, Uwe Parchatka¹, Jean-Daniel Paris², Andrea Pozzer^{1,4}, Roland Rohloff¹, Jan Schuladen¹, Justin Shenolikar¹, Sebastian Tauer¹, Jos Lelieveld^{1,3}, and Horst Fischer¹

Correspondence to: Ivan Tadic (i.tadic@mpic.de)

Figure S1 shows a day to day calculation of the correction factor to scale the fractional noontime integral to a diurnal value. An average \pm standard deviation of (46.1 \pm 2.8) % of the diurnal integral within \pm 2h around noontime was estimated.

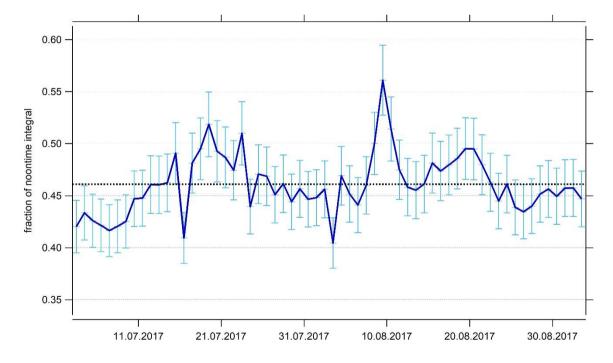


Figure S1: Ratio of the noontime actinic flux (\pm 2h around noon) with regard to the total actinic flux $j(NO_2)$ of that particular day. Dashed line represents the campaign average of (46.1 ± 2.8) %. The errors bars are represented by the relative uncertainty (6 %) of the campaign average.

¹Atmospheric Chemistry Department, Max Planck Institute for Chemistry, Mainz, Germany

^{990 &}lt;sup>2</sup>Laboratoire des Sciences du Climat et de l'Environnement, LSCE/IPSL, CEA-CNRS-UVSQ, Université Paris-Saclay, Gifsur-Yvette, France

³Energy, Environment and Water Research Center, The Cyprus Institute, Nicosia, Cyprus

⁴International Centre for Theoretical Physics, Trieste, Italy

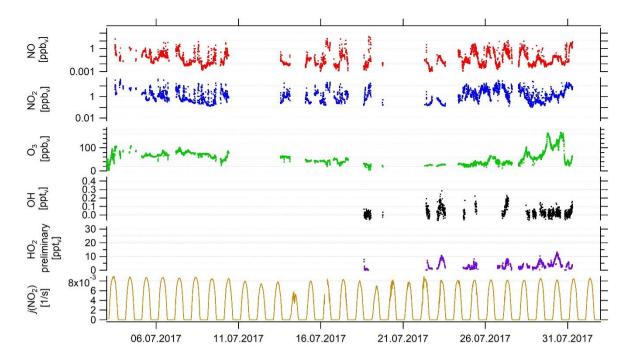


Figure S2: Timeline of NO, NO₂ (both CLD), O₃, OH, HO₂ preliminary ($\frac{\text{HO}_2*}{\text{IO}_2*}$) and $j(\text{NO}_2)$ data during the first leg. See Table ST2 for additional information on the ship cruise. Note that HO₂ data are preliminary.

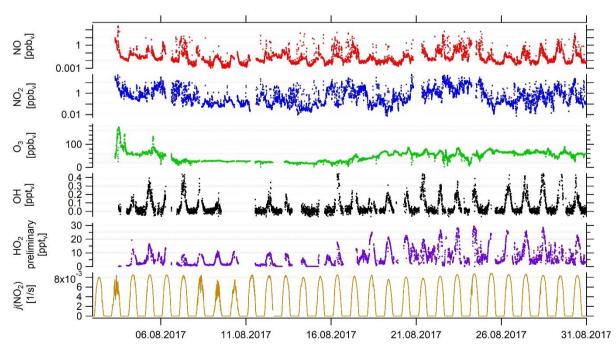
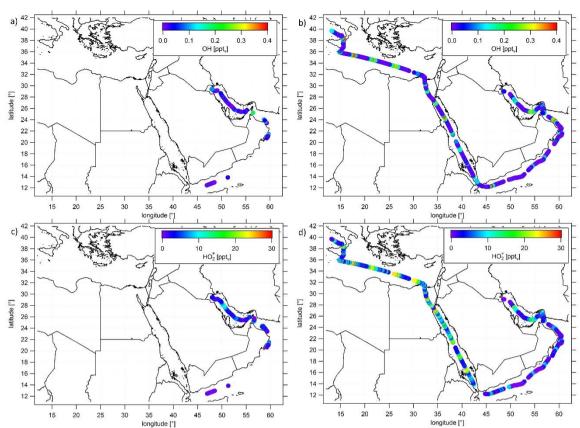


Figure S3: Timeline of NO, NO₂ (both CLD), O₃, OH, HO₂ preliminary ($\frac{\text{HO}_2*}{\text{HO}_2*}$) and $j(\text{NO}_2)$ data during the second leg. See Table ST2 for additional information on the ship cruise. Note that HO₂ data are preliminary.



1010 Figure S4: Ship cruises with color-scaled OH mixing ratios a) during the first and b) the second leg and color-scaled HO₂* mixing ratios c) during the first and d) the second leg. Note that OH and HO₂* data have been filtered for own stack contamination.

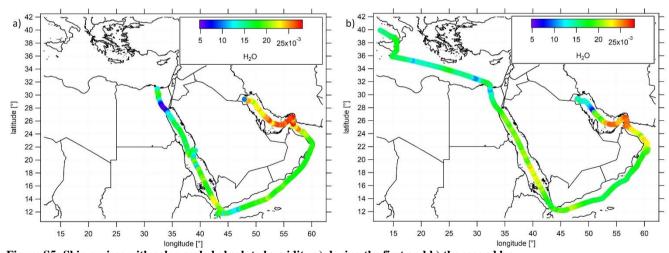


Figure S5: Ship cruises with color-scaled absolute humidity a) during the first and b) the second leg.

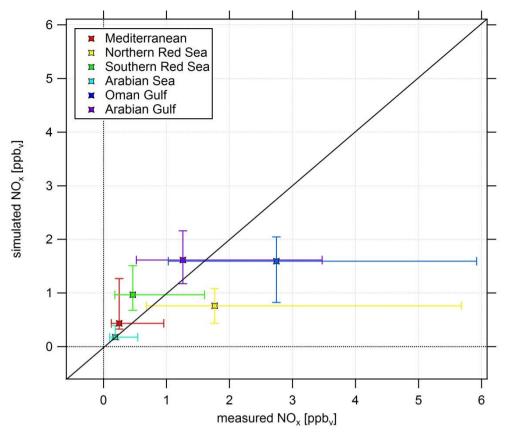
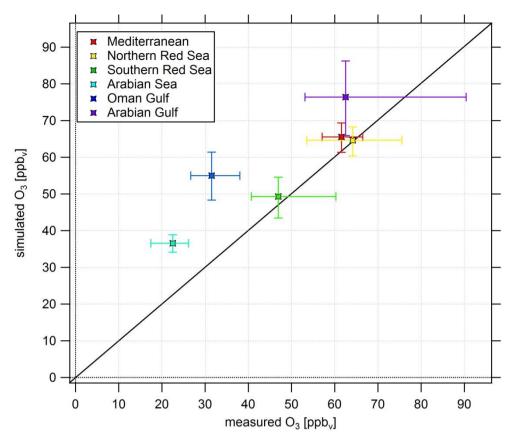


Figure S6: Scatterplot of simulated and measured regional NO_x median in ppb_y. 1:1 line added for orientation. The error bars represent the 25-75-percentile variation.



1020 Figure S7: Scatterplot of simulated and measured regional O₃ median in ppb_v. 1:1 line added for orientation. The error bars represent the 25-75-percentile variation.

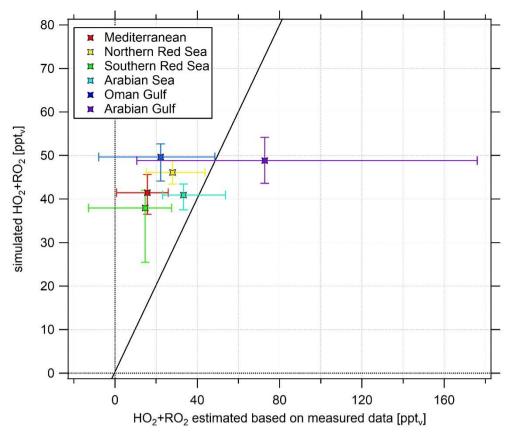


Figure S8: Scatterplot of simulated regional ($\frac{HO_2+RO_2}{RO_2}$) median and regional ($\frac{HO_2+RO_2}{RO_2}$) median estimated based on measured tracer data in ppt_v. 1:1 line added for orientation. The error bars represent the 25-75-percentile variation.

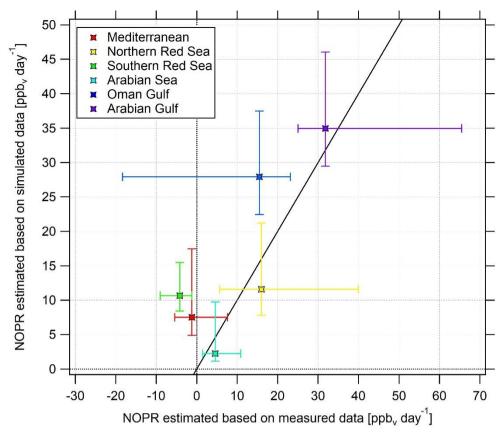


Figure S9: Scatterplot of median NOPR in ppb_v day⁻¹ estimated based on simulated and measured tracer data. 1:1 line added for orientation. The error bars represent the 25-75-percentile variation.

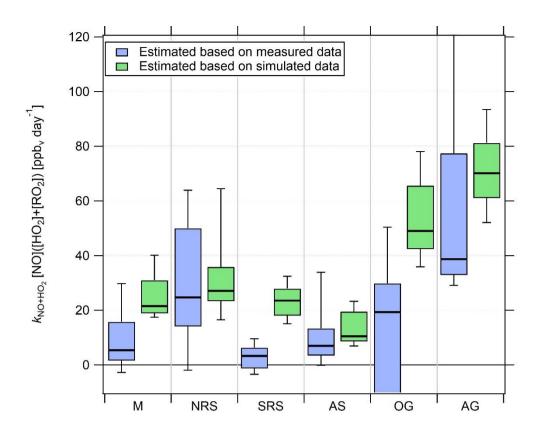


Figure S10: Comparison of the regional, absolute contribution of $k_{\text{NO+HO}_2}[\text{NO}]([\text{HO}_2] + [\text{RO}_2])$ to NOPR in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90-percentiles.

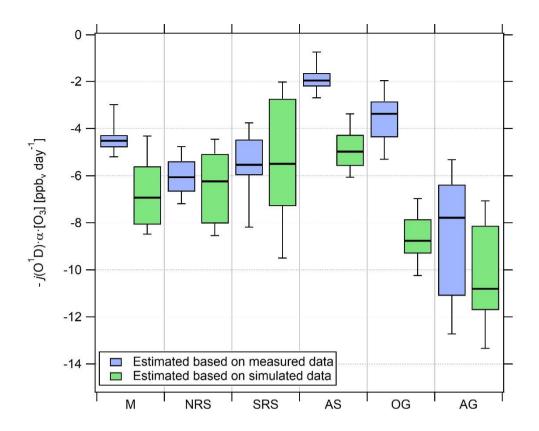


Figure S11: Comparison of the regional, absolute contribution of $-j(0^1D) \cdot \alpha \cdot [0_3]$ to NOPR in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90-percentiles.

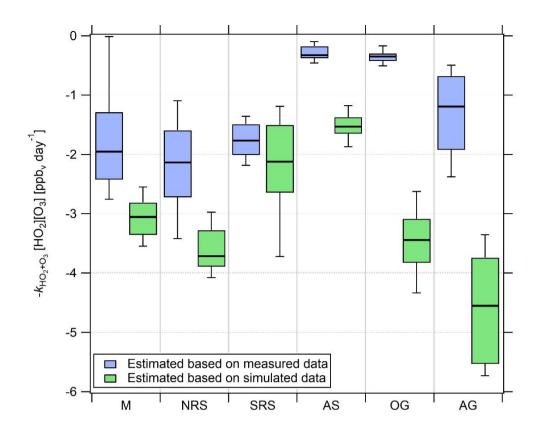


Figure S12: Comparison of the regional, absolute contribution of $-k_{\rm HO_2+O_3}[{\rm HO_2}][{\rm O_3}]$ to NOPR in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90-percentiles.

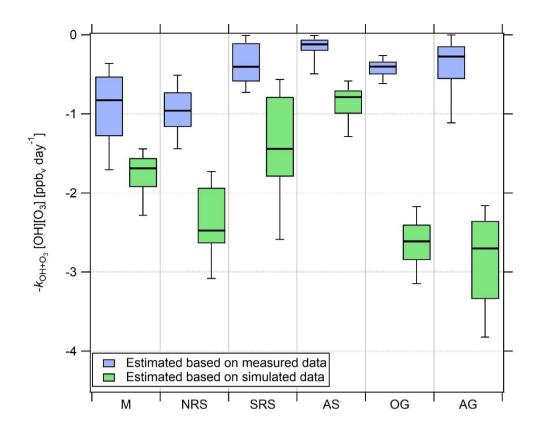


Figure S13: Comparison of the regional, absolute contribution of $-k_{0H+0_3}[0H][0_3]$ to NOPR in the six different regions investigated during AQABA. The horizontal black bar indicates the median value, the box the 25- and 75-percentiles and the whiskers the 10- and 90-percentiles.

1050 Table ST1: Range of latitudinal and longitudinal coordinates and dates during both legs of the different regions.

region (abbreviation)	latitudinal range	longitudinal range	Date (1st leg)	Date (2 nd leg)
Mediterranean (M)	31.810° N-	12.620° E-		25.08.2017 –
Wiediterranean (Wi)	39.923° N	31.850° E		31.08.2017
Northern Red Sea	23.343° N-	32.305° E-	03.07.2017 -	21.08.2017 -
(NRS)	30.986° N	37.085° E	08.07.2017	24.08.2017
Southern Red Sea	12.672° N-	37.411° E-	09.07.2017 –	17.08.2017 –
(SRS)	22.494° N	43.327° E	16.07.2017	20.08.2017
Arabian Sea (AS)	11.797° N-	44.035° E-	18.07.2017 -	07.08.2017 -
Arabian Sea (AS)	22.782° N	60.636° E	24.07.2017	16.08.2017
Oman Gulf (OG)	23.050° N-	56.492° E-	24.07.2017 -	05.08.2017 -
	25.622° N	59.913° E	27.07.2017	07.08.2017
Arabian Gulf (AG)	25.396° N-	47.920° E-	27.07.2017 –	03.08.2017 -
Alabian Gun (AG)	29.425° N	56.772° E	31.07.2017	05.08.2017

Table ST2: Overview of the time spent in the particular regions during AQABA. Red color indicates periods with KI at anchor that are not included in the data analysis. Data measured during bunkering at Fujairah City (06 August, 07:00-15:00 UTC) were also not included in the analysis.

Date	Region	Date	Region
03.07.2017	Northern Red Sea	02.08.2017	Kuwait port
04.07.2017	Northern Red Sea	03.08.2017	Kuwait port/Arabian Gulf
05.07.2017	Northern Red Sea	04.08.2017	Arabian Gulf
06.07.2017	Northern Red Sea	05.08.2017	Arabian Gulf/Oman Gulf
07.07.2017	Northern Red Sea	06.08.2017	Oman Gulf
08.07.2017	Northern Red Sea	07.08.2017	Oman Gulf/Arabian Sea
09.07.2017	Southern Red Sea	08.08.2017	Arabian Sea
10.07.2017	Southern Red Sea	09.08.2017	Arabian Sea
11.07.2017	Southern Red Sea/Jeddah port	10.08.2017	Arabian Sea
12.07.2017	Jeddah port	11.08.2017	Arabian Sea
13.07.2017	Jeddah port/Southern Red Sea	12.08.2017	Arabian Sea
14.07.2017	Southern Red Sea	13.08.2017	Arabian Sea
15.07.2017	Southern Red Sea	14.08.2017	Arabian Sea
16.07.2017	Southern Red Sea	15.08.2017	Arabian Sea
17.07.2017	Djibouti port	16.08.2017	Arabian Sea
18.07.2017	Arabian Sea	17.08.2017	Southern Red Sea
19.07.2017	Arabian Sea	18.08.2017	Southern Red Sea
20.07.2017	Arabian Sea	19.08.2017	Southern Red Sea
21.07.2017	Arabian Sea	20.08.2017	Southern Red Sea
22.07.2017	Arabian Sea	21.08.2017	Northern Red Sea
23.07.2017	Arabian Sea	22.08.2017	Northern Red Sea
24.07.2017	Arabian Sea/Oman Gulf	23.08.2017	Northern Red Sea
25.07.2017	Oman Gulf	24.08.2017	Northern Red Sea
26.07.2017	Oman Gulf	25.08.2017	Mediterranean
27.07.2017	Oman Gulf/Arabian Gulf	26.08.2017	Mediterranean
28.07.2017	Arabian Gulf	27.08.2017	Mediterranean
29.07.2017	Arabian Gulf	28.08.2017	Mediterranean
30.07.2017	Arabian Gulf	29.08.2017	Mediterranean
31.07.2017	Arabian Gulf/Kuwait port	30.08.2017	Mediterranean
01.08.2017	Kuwait port	31.08.2017	Mediterranean

 $Table \ ST3: \ Overview \ of \ measured \ NO_x \ (upper \ table) \ and \ measured \ O_3 \ (lower \ table) \ spatial \ volume \ mixing \ ratio \ average, \ standard$ deviation, 1st quantile, median, 3rd quantile (all in ppb_v) and number of considered data points.

NO _x (upper),	Mediterranea	Northern Red	Southern Red			
O ₃ (lower)	n	Sea	Sea	Arabian Sea	Oman Gulf	Arabian Gulf
data points	1767	1694	1755	2656	1056	1539
average	1.24	4.69	1.62	0.95	4.16	3.65
stdev	3.34	7.9	3.7	3.15	4.33	9.24
1 st quantile	0.12	0.68	0.18	0.10	1.03	0.52
median	0.25	1.76	0.46	0.19	2.74	1.26
3 rd quantile	0.96	5.68	1.6	0.54	5.92	3.47
data points	2010	2717	2307	4130	1249	1809
average	61.56	63.39	50.35	21.53	34.04	73.99
stdev	8.25	18.45	12.96	6.8	11.27	35.68
1 st quantile	57.05	53.51	40.68	17.45	26.66	53.08
median	61.54	64.16	46.93	22.52	31.5	62.5
3 rd quantile	66.48	75.51	60.28	26.19	38.03	90.42

Table ST4: Overview of simulated NO_x (upper table) and simulated O₃ (lower table) spatial volume mixing ratio average, standard deviation, 1st quantile, median, 3rd quantile (all in ppbv) and number of considered data points.

NO _x (upper), O ₃	Mediterranea	Northern Red	Southern Red			
(lower)	n	Sea	Sea	Arabian Sea	Oman Gulf	Arabian Gulf
data points	2012	2719	2310	4464	1253	1810
average	0.84	1.27	1.13	0.31	1.88	1.91
stdev	0.75	1.97	0.62	0.29	1.47	1.37
1 st quantile	0.33	0.43	0.67	0.14	0.82	1.17
median	0.43	0.76	0.97	0.18	1.59	1.61
3 rd quantile	1.27	1.08	1.51	0.39	2.05	2.16
data points	2012	2719	2310	4464	1253	1810
average	65.15	64.76	49.5	36.91	55	76.85
stdev	4.99	7.7	7.21	3.87	9.94	12.8
1 st quantile	61.35	60.34	43.47	34.12	48.37	65.88
median	65.53	64.6	49.33	36.54	55.01	76.39
3 rd quantile	69.33	68.25	54.58	38.85	61.42	86.22

Table ST5: Overview of noontime (HO₂+RO₂) spatial volume mixing ratio average, standard deviation, 1st quantile, median, 3rd quantile (all in ppt_y) estimated based on measured tracer data. Number of considered data points added in the first line.

HO ₂ +RO ₂	Mediterranean	Northern Red	Southern Red	Arabian Sea	Oman Gulf	Arabian Gulf
		Sea	Sea			
data points	288	126	190	338	166	242
average	13	27	7	64	23	94
stdev	24	20	34	83	42	113
1 st quantile	1	15	-13	23	-8	11
median	16	28	15	33	22	73
3 rd quantile	26	44	27	54	49	176

Table ST6: Overview of simulated noontime (HO₂+RO₂) spatial volume mixing ratio average, standard deviation, 1st quantile, median, 3rd quantile (all in ppt_y). Number of considered data points added in the first line.

HO ₂ +RO ₂	Mediterranean	Northern Red	Southern Red	Arabian Sea	Oman Gulf	Arabian Gulf
		Sea	Sea			
data points	336	192	192	720	203	293
average	41	46	36	40	48	49
stdev	5	4	11	5	6	7
1 st quantile	36	43	25	38	44	44
median	41	46	38	41	50	49
3 rd quantile	46	49	42	43	53	54

Table ST7: Overview of NOPR average, standard deviation, 1st quantile, median, 3rd quantile (all in ppb_v day⁻¹) estimated based on measured tracer data. Number of considered data points added in the first line.

		Northern	Southern			
NOPR	Mediterranean	Red Sea	Red Sea	Arabian Sea	Oman Gulf	Arabian Gulf
data points	148	114	89	187	84	111
average	-3	16	-5	-67	-105	-24
stdev	43	39	12	576	362	449
1 st quantile	-5	6	-9	1	-18	25
median	-1	16	-4	5	16	32
3 rd quantile	8	40	-1	11	23	65

Table ST8: Overview of NOPR average, standard deviation, 1^{st} quantile, median, 3^{rd} quantile (all in ppb_v day⁻¹) estimated based on simulated tracer data. Number of considered data points added in the first line.

		Northern	Southern			
NOPR	Mediterranean	Red Sea	Red Sea	Arabian Sea	Oman Gulf	Arabian Gulf
data points	336	192	192	720	203	293
average	11	16	12	5	30	38
stdev	8	12	6	6	11	10
1 st quantile	5	8	8	1	22	30
median	8	12	11	2	28	35
3 rd quantile	17	21	15	10	37	46

1075

Table ST9: Overview of measured HCHO/NO₂-ratio average, standard deviation, 1st quantile, median, 3rd quantile and number of considered data points.

	Mediterranea	Northern Red	Southern Red			
Ratio	n	Sea	Sea	Arabian Sea	Oman Gulf	Arabian Gulf
data points	203	79	48	252	108	122
average	5.4	1.5	8.5	11.1	2.7	9
stdev	4.7	0.7	6.5	8.9	2.1	6.4
1 st quantile	1.3	0.8	4.4	2.3	1	2.5
median	5	1.4	7.7	9.4	2.2	9.3
3 rd quantile	7.4	2.1	9.8	16.1	3.6	12.7

Table ST10: List of included peroxy radicals (with less than four carbon atoms) for the reaction with NO as recommended by Sander et al. (2019).

et al. (2019).
Species
HO_2
CH ₃ O ₂
C ₂ H ₅ O2
C ₂ H ₅ CO ₃
CH ₃ CO ₃
C3DIALO2 (C ₃ H ₃ O ₄)
CH ₃ CHOHO ₂
CH ₃ COCH ₂ O ₂
CH ₃ COCO ₃
CHOCOCH ₂ O ₂
CO ₂ H ₃ CO ₃
HCOCH ₂ CO ₃
HCOCH ₂ O ₂
HCOCO ₃
HCOCOHCO ₃
HOC ₂ H ₄ CO ₃
HOCH ₂ CH ₂ O ₂
HOCH ₂ CO ₃
HOCH ₂ COCH ₂ O ₂
HOCH ₂ O ₂
CH ₃ CHO ₂ CH ₂ OH
IC3H7O2 (isopropylperoxy radical)
NC3H7O2 (propylperoxy radical)
NCCH ₂ O ₂
NO ₃ CH ₂ CO ₃
CH ₃ CHO ₂ CH ₂ ONO ₂