

1 **Novel approaches to improve estimates of short-lived halocarbon emissions during summer**  
2 **from the Southern Ocean using airborne observations**

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17

18

19 **Abstract.**

20 Fluxes of halogenated volatile organic compounds (VOCs) over the Southern Ocean remain  
21 poorly understood, and few atmospheric measurements exist to constrain modeled emissions of  
22 these compounds. We present observations of  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ ,  $\text{CH}_3\text{I}$ ,  $\text{CHClBr}_2$ ,  $\text{CHBrCl}_2$ , and  
23  $\text{CH}_3\text{Br}$  during the  $\text{O}_2/\text{N}_2$  Ratio and  $\text{CO}_2$  Airborne Southern Ocean (ORCAS) study and the 2<sup>nd</sup>  
24 Atmospheric Tomography mission (ATom-2), in January and February of 2016 and 2017. Good  
25 model-measurement correlations were obtained between these observations and simulations from  
26 the Community Earth System Model (CESM) atmospheric component with chemistry (CAM-  
27 Chem) for  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ ,  $\text{CH}_3\text{I}$ , and  $\text{CHClBr}_2$  but all showed significant differences in  
28 model:measurement ratios. The model:measurement comparison for  $\text{CH}_3\text{Br}$  was satisfactory and  
29 for  $\text{CHBrCl}_2$  the low levels present precluded us from making a complete assessment.  
30 Thereafter, we demonstrate two novel approaches to estimate halogenated VOC fluxes; the first  
31 approach takes advantage of the robust relationships that were found between airborne  
32 observations of  $\text{O}_2$  and  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ , and  $\text{CHClBr}_2$ ; we use these linear regressions with  $\text{O}_2$   
33 and modeled  $\text{O}_2$  distributions to infer a biological flux of halogenated VOCs. The second  
34 approach uses the Stochastic Time-Inverted Lagrangian Transport (STILT) particle dispersion  
35 model to explore the relationships between observed mixing ratios and the product of the  
36 upstream surface influence and sea ice, chl *a*, absorption due to detritus, and downward  
37 shortwave radiation at the surface, which in turn relate to various regional hypothesized sources  
38 of halogenated VOCs such as marine phytoplankton, phytoplankton in sea ice brines, and  
39 decomposing organic matter in surface seawater. These relationships can help evaluate the  
40 likelihood of particular halogenated VOC sources, and in the case of statistically significant  
41 correlations, such as was found for  $\text{CH}_3\text{I}$ , may be used to derive an estimated flux field. Our  
42 results are consistent with a biogenic regional source of  $\text{CHBr}_3$ , and both non-biological and  
43 biological sources of  $\text{CH}_3\text{I}$  over these regions.

44

45 **1 Introduction**

46 Emissions of halogenated volatile organic compounds (VOCs) influence regional atmospheric  
47 chemistry and global climate. Through the production of reactive halogen radicals at high  
48 latitudes, halogenated VOCs contribute to tropospheric and stratospheric ozone destruction, and  
49 alter the sulfur, mercury, nitrogen oxide and hydrogen oxide cycles (e.g. Schroeder et al., 1998;  
50 Boucher et al., 2003; Bloss et al., 2005; von Glasow and Crutzen; 2007; Saiz-Lopez et al., 2007;  
51 Obrist et al., 2011; Engel and Rigby, 2018). In the marine boundary layer and lower  
52 troposphere, sea salt is the main source of reactive bromine (Finlayson-Pitts 1982, Simpson et  
53 al., 2015). Yet halogenated VOCs may also be a more important source of inorganic bromine to  
54 the whole atmosphere than previously thought, according to a recent study, which indicates that  
55 sea salt is scarce and insufficient to control the bromine budget in the middle and upper  
56 troposphere (Murphy et al., 2019).

57 Phytoplankton and macroalgae in the ocean are the main sources to the atmosphere of several  
58 very short-lived bromocarbons, including bromoform ( $\text{CHBr}_3$ ), dibromomethane ( $\text{CH}_2\text{Br}_2$ ),

59 dibromochloromethane ( $\text{CHClBr}_2$ ), and bromodichloromethane ( $\text{CHBrCl}_2$ ) (Moore et al., 1996;  
60 Carpenter et al. 2003; Butler et al., 2007; Raimund et al., 2011). Other halogenated VOCs, such  
61 as methyl iodide ( $\text{CH}_3\text{I}$ ), and methyl bromide ( $\text{CH}_3\text{Br}$ ) have many natural sources, such as  
62 coastal macroalgae, phytoplankton, temperate forest soil and litter, and biomass burning (e.g.,  
63 Bell et al., 2002; Sive et al., 2007; Colomb et al. 2008; Drewer et al., 2008).  $\text{CH}_3\text{I}$  is also formed  
64 through non-biological reactions in surface seawater, and  $\text{CH}_3\text{Br}$  is emitted as a result of  
65 quarantine and pre-shipment activities, which are not regulated by the Montreal Protocol (e.g.,  
66 Moore and Zafiriou; 1994, Engel and Rigby, 2018). Over the Southern Ocean specifically,  
67 hypothesized sources of halogenated VOCs include: coastal macroalgae, phytoplankton, sea ice  
68 algae, and photochemical or dust stimulated non-biological production at the sea surface (e.g.,  
69 Abrahamsson et al. 2018, Manley and Dastoor 1998; Moore and Zafiriou 1994; Moore et al.,  
70 1996; Richter and Wallace 2004; Williams et al., 2007; Tokarczyk and Moore 1994; Sturges et  
71 al., 1992).

72 We largely owe our current understanding of marine halogenated VOC emissions over the  
73 Southern Ocean to ship-based field campaigns and laboratory process studies (e.g., Abrahamsson  
74 et al. 2004a,b; Atkinson et al., 2012; Carpenter et al., 2007; Moore et al., 1996; Chuck et al.,  
75 2005; Butler et al., 2007; Raimund et al., 2011; Hughes et al., 2009; Mattsson et al. 2013;  
76 Hughes et al., 2013). These studies have reported surface water and sea-ice halogenated VOC  
77 supersaturation and corresponding elevated levels of halogenated VOCs in the marine boundary  
78 layer (MBL) in summer, and have identified numerous biological and non-biological ocean  
79 sources for these compounds. Mattsson et al. (2013) noted that the ocean also acts as a sink for  
80 halogenated VOCs, when undersaturated surface waters equilibrate with air masses transported  
81 from halogenated VOC source regions. The spatially heterogeneous ocean sources of  $\text{CHBr}_3$  and  
82  $\text{CH}_2\text{Br}_2$  at high latitudes in the Southern Hemisphere are often underestimated in global  
83 atmospheric models (Hossaini et al., 2013; Ordoñez et al., 2012; Ziska et al., 2013). Ship-based  
84 and Lagrangian float observations provide invaluable information on the sources and temporal  
85 variability of compounds in the surface ocean. These methods offer the advantage of  
86 simultaneous measurements of both air and seawater to evaluate the gases' saturation state in the  
87 surface ocean and calculate fluxes. Yet ship-based measurements onboard these slow moving  
88 platforms also have drawbacks: they under sample the spatial variability of halogenated VOCs  
89 (e.g., Butler et al., 2007) and require assumptions about gas-exchange rates to estimate fluxes.

90 To disentangle the roles of atmospheric transport and spatial variability of emissions on  
91 halogenated VOC distributions requires large-scale atmospheric observations. At low latitudes,  
92 large-scale convection at the intertropical convergence zone carries bromocarbons and other  
93 halogenated VOCs into the free troposphere and lower stratosphere (e.g., Liang et al., 2014;  
94 Navarro et al., 2015). Polar regions are characterized by stable boundary layers in summer.  
95 Wind shear, frontal systems, and internal gravity waves create turbulence and control vertical  
96 mixing within and across a stable polar boundary layer (e.g. Anderson et al., 2008), and small,  
97 convective plumes may form over the marginal sea ice zone, related to sea ice leads as well as  
98 winds from ice-covered to open-ocean waters (e.g. Schnell et al., 1989). As a result of limited  
99 vertical transport in these regions, however, air-sea fluxes lead to strong vertical gradients.  
100 Zonal transport from lower latitudes has a large impact on the vertical gradients of trace gas

101 mixing ratios over polar regions (Salawitch 2010). Given their extended photochemical lifetimes  
102 at high latitudes (see Sect. 2.3 for a brief discussion), many halogenated VOC distributions are  
103 particularly sensitive to zonal transport at altitude.

104 Aircraft observations can rapidly map basin-wide vertical distributions, support quantitative flux  
105 estimates, and provide spatial constraints to atmospheric models (e.g. Xiang et al., 2010;  
106 Stephens et al., 2018; Wofsy et al., 2011). Few airborne observations of halogenated VOCs exist  
107 at high latitudes in the Southern Hemisphere. Two earlier aircraft campaigns that have measured  
108 summertime halogenated VOCs in this region are the first Aerosol Characterization Experiment  
109 (ACE-1; Bates et al., 1999) and the first High-performance Instrumented Airborne Platform for  
110 Environmental Research (HIAPER) Pole-to-Pole Observations (HIPPO; Wofsy, 2011)  
111 campaign. For these two aircraft campaigns, whole air samples were collected onboard the  
112 NSF/NCAR C-130 and the NSF/NCAR Gulfstream V (GV) during latitudinal transects over the  
113 Pacific Ocean as far south as 60° S and 67° S, respectively. However, the ACE-1 and HIPPO  
114 campaigns obtained relatively few whole air samples in this region, with  $\leq 100$  samples poleward  
115 of 60° S combined (e.g., Blake et al., 1999; Hossaini et al., 2013). ACE-1 measurements of CH<sub>3</sub>I  
116 in the MBL indicate a strong ocean source between 40° S and 50° S in austral summer, with  
117 mixing ratios above 1.2 pmol below ~1 km (Blake et al., 1999).

118 Halogenated VOC emissions are frequently incorporated into earth system models, using either  
119 climatologies or parameterizations based on satellite observations of chlorophyll and  
120 geographical region and evaluated using mixing ratio comparisons with airborne observations. In  
121 Section 3.1 and 3.2, we report new airborne observations of CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, CH<sub>3</sub>I, CHClBr<sub>2</sub>,  
122 CHBrCl<sub>2</sub>, and CH<sub>3</sub>Br from high latitudes in the Southern Hemisphere, where data are scarce, and  
123 large-scale regional mixing ratio comparisons for halogenated VOCs with the community earth  
124 system model (CESM) atmospheric component with chemistry (CAM-Chem). In section 3.4, we  
125 present two novel approaches to estimate regional fluxes of halogenated VOCs for comparison  
126 with global climate models' parameterizations or climatologies. One approach uses correlations  
127 of halogenated VOCs to marine, oxygen (O<sub>2</sub>) of marine origin, as measured by deviations in the  
128 ratio of O<sub>2</sub> to nitrogen (N<sub>2</sub>) ( $\delta(O_2/N_2)$  see Sect. 2.1.2 and 3.1.2). We exploit robust ratios of  
129 halogenated VOCs to oxygen (O<sub>2</sub>) determined from linear regressions (i.e. the enrichment ratio),  
130 and the ocean flux of O<sub>2</sub> from CESM's ocean component, to estimate the marine biogenic flux of  
131 several halogenated VOCs. The second approach relies on observed halogenated VOC mixing  
132 ratios, the Stochastic Time-Inverted Lagrangian Transport (STILT) particle dispersion model and  
133 geophysical datasets (see Sect. 2.3 and 3.3). We assess contributions from previously  
134 hypothesized regional sources for the Southern Ocean, and estimate halogenated VOC fluxes  
135 based on regressions between upstream influences and observed mixing ratios and distributions  
136 of remotely sensed data.

137

## 138 **2 Methods**

### 139 **2.1 Measurements**

140 Atmospheric measurements for this study were collected at high latitudes in the Southern  
141 Hemisphere as part of the O<sub>2</sub>/N<sub>2</sub> Ratio and CO<sub>2</sub> Airborne Southern Ocean (ORCAS) study

142 (Stephens et al., 2018), and the second NASA Atmospheric Tomography Mission (ATom-2),  
143 near Punta Arenas, Chile (Fig. 1). The ORCAS field campaign took place from Jan. 15 – Feb.  
144 29, 2016 onboard the NSF/NCAR GV. On Feb. 10 and 13, 2017 the sixth and seventh ATom-2  
145 research flights passed over the eastern Pacific sector poleward of 60° S (defined here as Region  
146 1) and over the Patagonian Shelf between 40° S and 55° S and between 70° W and 50° W  
147 (defined here as Region 2), respectively. The two regions for this study are defined based  
148 loosely on dynamic biogeochemical provinces identified using bathymetry, algal biomass, sea  
149 surface temperature and salinity (Reygondeau et al., 2013).

150 Both projects featured en route vertical profiling from near the ocean surface (~ 150 m) to the  
151 upper-troposphere, with 74 ORCAS and seven ATom-2 (during the sixth and seventh flights)  
152 low-altitude level legs in the MBL. These campaigns shared a number of instruments, including  
153 the NCAR Trace Gas Organic Analyzer (TOGA), the NCAR Atmospheric Oxygen (AO2)  
154 instrument, a Picarro cavity ringdown spectrometer operated by NOAA, discussed below. More  
155 information about individual instruments may be found in Stephens et al., 2018 and at  
156 [https://www.eol.ucar.edu/field\\_projects/orcas](https://www.eol.ucar.edu/field_projects/orcas) and <https://espo.nasa.gov/atom/content/ATom>.

157

### 158 **2.1.1 Halogenated VOCs**

159 During ORCAS and ATom-2 TOGA provided mixing ratios of over 60 organic compounds,  
160 including halogenated VOCs. The instrument, described in Apel et al. (2015), continuously  
161 collects and analyzes samples for  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ ,  $\text{CHClBr}_2$ ,  $\text{CHBrCl}_2$ , and  $\text{CH}_3\text{I}$  among other  
162 compounds, with a 35-second sampling period and repeats the cycle every two-minutes using  
163 online fast gas chromatography and mass spectrometry. This study also leverages measurements  
164 of  $\text{CH}_3\text{Br}$  from whole air samples from the U. Miami / NCAR Advanced Whole Air Sampler  
165 (AWAS; Schauffler et al., 1999) onboard the GV during the ORCAS campaign and the UC  
166 Irvine Whole Air Sampler (WAS; Blake et al., 2001) onboard the DC-8 during the ATom-2  
167 campaign. Halogenated VOCs reported here have an overall  $\pm 15\%$  accuracy and  $\pm 3\%$  relative  
168 precision, and detection limits of 0.03 ppt for  $\text{CH}_3\text{I}$ , 0.2 ppt for  $\text{CHBr}_3$ , 0.03 ppt for  $\text{CH}_2\text{Br}_2$ , 0.03  
169 ppt for  $\text{CHClBr}_2$ , 0.05 ppt for  $\text{CHBrCl}_2$ , and 0.2 ppt for  $\text{CH}_3\text{Br}$  – 0.2 ppt. In addition,  
170 comparisons between onboard collected whole air samples and in-flight TOGA measurements,  
171 when sharing over half of their sampling period with TOGA measurements, showed good  
172 correlations for  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ ,  $\text{CH}_3\text{I}$ , and  $\text{CHClBr}_2$ , although there were some calibration  
173 differences (Fig. S1 and Fig. S2). In addition to the comparison between co-located atmospheric  
174 measurements, we also conducted a lab inter-comparison following the campaign between  
175 NOAA's programmable flask package (PFP) and TOGA (Table S1; see supplement for details).

176

### 177 **2.1.2 $\delta(\text{O}_2/\text{N}_2)$ and $\text{CO}_2$**

178 The AO2 instrument measures variations in atmospheric  $\text{O}_2$ , which are reported as relative  
179 deviations in the oxygen to nitrogen ratio ( $\delta(\text{O}_2/\text{N}_2)$ ), following a dilution correction for  $\text{CO}_2$   
180 (Keeling et al., 1998; Stephens et al., 2018). The instrument's precision is  $\pm 2$  per meg units (one

181 in one million relative) for a 5 second measurement (Stephens et al., 2003; Stephens et al.,  
182 manuscript in preparation, 2019). Anthropogenic, biogenic, and oceanic processes introduce O<sub>2</sub>  
183 perturbations that are superimposed on the background concentrations of O<sub>2</sub> in air (XO<sub>2</sub>, in dry  
184 air = 0.2095). Air-sea O<sub>2</sub> fluxes are driven by both biological production and consumption of O<sub>2</sub>  
185 and by heating and cooling of surface waters. O<sub>2</sub> is consumed when fossil fuels are burned and  
186 produced and consumed during terrestrial photosynthesis and respiration. Seasonal changes in  
187 the ocean heat content lead to small changes in atmospheric N<sub>2</sub>. As others have done, we  
188 isolated the air-sea O<sub>2</sub> signal by subtracting model estimates of the terrestrial O<sub>2</sub>, fossil-fuel O<sub>2</sub>,  
189 and air-sea N<sub>2</sub> flux influences from the δ(O<sub>2</sub>/N<sub>2</sub>) measurements (Equation 1; Keeling et al., 1998;  
190 Garcia and Keeling, 2001; Stephens et al., 2018). The difference of the δ(O<sub>2</sub>/N<sub>2</sub>) measurement  
191 and these modeled components is multiplied by XO<sub>2</sub> to convert to ppm equivalents as needed  
192 (ppm eq; Keeling et al., 1998; Equation 1).

$$193 \quad O_{2\text{-ppm-equiv}} = [\delta(O_2/N_2) - \delta(O_2/N_2)_{\text{Land}} - \delta(O_2/N_2)_{\text{Fossil Fuel}} - \delta(O_2/N_2)_{\text{N}_2}] \times XO_2 \quad (1)$$

194 We obtained the modeled δ(O<sub>2</sub>/N<sub>2</sub>) signal terrestrial influences from the land model component  
195 of the CESM, the fossil fuel combustion influences from the Carbon Dioxide Information  
196 Analysis Center (CDIAC; Boden et al. 2017), and the air-sea N<sub>2</sub> influences from the oceanic  
197 component of CESM. These fluxes were all advected through the specified dynamics version of  
198 CESM's atmosphere component, as described below in Sect. 2.2 and in Stephens et al. (2018).

199  
200 CO<sub>2</sub> measurements were provided by NOAA's Picarro G2401-m cavity ring down spectrometer  
201 modified to have a ~1.2 sec measurement interval and a lower cell pressure of 80 Torr, which  
202 enabled the instrument to function at the full range of GV altitudes. (McKain et al. manuscript in  
203 preparation, 2019). Dry-air mole fractions were calculated using empirical corrections to account  
204 for dilution and pressure broadening effects as determined in the laboratory before and after the  
205 campaign deployments, and in-flight calibrations were used to determine an offset correction for  
206 each flight. Corrected CO<sub>2</sub> data have a total average uncertainty of 0.07 ppm (McKain et al.  
207 manuscript in preparation, 2019). To merge them with the TOGA data, these faster O<sub>2</sub> and CO<sub>2</sub>  
208 measurements were arithmetically averaged over TOGA's 35-s sampling periods (Stephens et  
209 al., 2017 and <https://espo.nasa.gov/atom/content/ATom>).

210

## 211 **2.2 CAM-Chem model configuration**

212 The CESM version 1, atmospheric component with chemistry (CAM-Chem) is a global three-  
213 dimensional chemistry climate model that extends from the Earth's surface to the stratopause.  
214 CAM-Chem version 1.2 includes all the physical parameterizations of Neale et al. (2013) and a  
215 finite volume dynamical core (Lin, 2004) for tracer advection. The model has a horizontal  
216 resolution of 0.9° latitude × 1.25° longitude, with 56 vertical hybrid levels and a time-step of 30  
217 minutes. Meteorology is specified using the NASA Global Modeling and Assimilation Office  
218 (GMAO) Goddard Earth Observing System Model, version 5 (GEOS-5; Rienecker et al., 2008)  
219 (GEOS-5), following the specified dynamic procedure described by Lamarque et al. (2012).  
220 Winds, temperatures, surface pressure, surface stress, and latent and sensible heat fluxes are

221 nudged using a 5-hour relaxation timescale to GEOS-5  $1^\circ \times 1^\circ$  meteorology. The sea surface  
222 temperature boundary condition is derived from the Merged Hadley-NOAA Optimal  
223 Interpolation Sea Surface Temperature and Sea-Ice Concentration product (Hurrell et al., 2008).  
224 The model uses chemistry described by Tilmes et al. (2016), biomass burning and biogenic  
225 emissions from the Fire INventory of NCAR (FINN; Wiedinmyer et al., 2011) and MEGAN  
226 (Model of Emissions of Gasses and Aerosols from Nature) 2.1 products (Guenther et al., 2012)  
227 and additional tropospheric halogen chemistry described in Fernandez et al. (2014) and Saiz-  
228 Lopez et al. (2014). These include ocean emissions of  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ ,  $\text{CHBr}_2\text{Cl}$ , and  $\text{CHBrCl}_2$ ,  
229 with parameterized emissions based on chlorophyll *a* (*chl a*) concentrations and scaled by a  
230 factor of 2.5 over coastal regions, as opposed to open ocean regions (Ordoñez et al., 2012). The  
231 model used an existing  $\text{CH}_3\text{I}$  flux climatology (Bell et al., 2002), and  $\text{CH}_3\text{Br}$  was constrained to a  
232 surface lower boundary condition, also described by Ordoñez et al. (2012). This version of the  
233 model was run for the period of the ORCAS field campaign (January and February 2016),  
234 following a 24-month spin-up. To facilitate comparisons to ORCAS observations, output  
235 included vertical profiles of modeled constituents from the two nearest latitude and two nearest  
236 longitude model grid-points (four profiles in total) to the airborne observations at every 30-min  
237 model time-step. Following the run, simulated constituent distributions were linearly interpolated  
238 to the altitude, latitude and longitude along the flight track, yielding co-located modeled  
239 constituents and airborne observations. This version of the model has not yet been run for the  
240 ATom-2 period.

241

### 242 **2.3 STILT model configuration**

243 The Stochastic Time-Inverted Lagrangian Transport (STILT; Lin et al., 2003) particle dispersion  
244 model uses a receptor oriented framework to infer surface sources or sinks of trace gases from  
245 atmospheric observations collected downstream, thus simulating the upstream influences that are  
246 ultimately measured at the receptor site. The model tracks ensembles of particle trajectories  
247 backward in time and the resulting distributions of these particles can be used to define surface  
248 influence maps for each observation. STILT was run using  $0.5^\circ$  Global Data Assimilation System  
249 (GDAS) reanalysis winds to investigate the transport history of air sampled along the flight track  
250 (Stephens et al., 2018). For each TOGA observation, an ensemble of 4,096 particles was  
251 released from the sampling location and followed over a backwards simulation period of seven  
252 days. Particles in the lower half of the simulated MBL are assigned a surface influence value,  
253 which quantitatively links observed mixing ratios to surface sources (Lin et al., 2003). The  
254 average surface influence of all 4,096 particles per sampling location yields an hourly and  
255 spatially gridded surface influence value ( $\text{ppt m}^2 \text{ s pmol}^{-1}$ ) at a spatial resolution of  $0.25^\circ \times 0.25^\circ$   
256 for each sample point.

257 Uncertainty in the surface influence value is strongly influenced by the accuracy of the  
258 underlying meteorological transport, as discussed in Xiang et al. (2010). We evaluated the  
259 GDAS reanalysis winds by comparing model winds interpolated in space and averaged between  
260 corresponding time points and pressure levels to match aircraft observations. By evaluating  
261 observed winds compared with modeled winds along the flight tracks we can estimate

262 uncertainty in the surface influence values. We consider the observation-model differences in  
263 both wind speed and direction to approximate errors in surface influence strength and location.  
264 For wind speed, a small bias may be present, where we find a median difference between  
265 observations and reanalysis of 0.68 m/s, a 5% relative bias. The 1-sigma of the wind speed  
266 difference is 2.3 m/s, corresponding to a 19% 1-sigma uncertainty in wind speed. In its simplest  
267 approximation, the surface influence strength error is perfectly correlated with the wind speed  
268 error, and thus we take 19% as an approximation of the surface influence strength uncertainty.  
269 The uncertainty in surface influence location depends on the error in the wind direction. We find  
270 a 1-sigma error of 14 degrees in wind speed, which corresponds to a possible error of 260  
271 km/day.

272 Finally, we note that photochemical loss during transport is not accounted for in this analysis.  
273 Low OH mixing ratios, cold temperatures, and lower photolysis rates due to angled sunlight at  
274 high latitudes lead to longer than average halogenated VOC lifetimes. For instance, assuming an  
275 average diurnal OH concentration of 0.03 ppt, and average photochemical loss according to the  
276 Tropospheric Ultraviolet and Visible (TUV) radiation model and the Mainz Spectral data site  
277 ([http://satellite.mpic.de/spectral\\_atlas](http://satellite.mpic.de/spectral_atlas)) for Jan. 29 under clear sky conditions at 60° S, CHBr<sub>3</sub> has  
278 a lifetime of 30 days, CH<sub>2</sub>Br<sub>2</sub> has a lifetime of 270 days, CH<sub>3</sub>I has a lifetime of 7 days, and  
279 CHClBr<sub>2</sub> has a lifetime of 63 days. As such, the photochemical lifetimes of these gases are  
280 greater than or equal to the time of our back-trajectory analysis. Moreover, OH concentrations in  
281 this region have large uncertainties, the inclusion of which would lead to more, not less,  
282 uncertainty in surface influence based regression coefficients and estimated fluxes (see Sect. 2.3  
283 and 3.3 for details).

284

### 285 **2.3.1 STILT surface influence functions**

286 For this study, we used STILT surface influence distributions with remotely sensed ocean  
287 surface and reanalysis data (i.e. surface source fields) in linear and multi-linear regressions to  
288 generate empirical STILT influence functions. Surface influence functions can help explain  
289 observed mixing ratios of CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, CH<sub>3</sub>Br and CH<sub>3</sub>I, evaluate the likelihood of particular  
290 halogenated VOC sources, and in the case of statistically significant correlations, may be used to  
291 derive an estimated flux field (See Sect. 3.3 and 3.4.2 for details).

292

293 We tested whether observed mixing ratios ( $Z$ ) could be explained by a linear relationship in  
294 which the predictor variable is a surface influence function, equal to the product of the surface  
295 influence ( $H$ ) and a potential geophysical surface source field(s), such as chl  $a$ , as well as an  
296 intercept ( $b$ ), a slope ( $a$ ), and error term  $\xi$  (Equation 2; Fig. S5). This relationship can be  
297 generalized as a multiple linear regression with multiple surface influence functions ( $H_{s_1}, H_{s_2}, \dots$ )  
298 and slope coefficients ( $a_1, a_2$ ; Equation 3), when multiple sources contribute to observed  
299 halogenated VOC mixing ratios. The multiple linear regression may also include an interaction  
300 term ( $H_{s_2}$ ) between predictor variables (e.g.  $H_{s_1}$  and  $H_{s_2}$ ) with a slope coefficient ( $a_3$ ) to improve  
301 the fit. Statistical correlations between mixing ratios and surface influence functions may be



302 used to support or reject hypothesized sources. A flux ( $\mu\text{mol m}^{-2} \text{s}^{-1}$ ) may then be estimated for  
303 each grid cell based on the product of the slopes ( $a_1, a_2, \dots$ ) and the potential source fields ( $s_1, s_2, \dots$ ).  
304 Grid cell fluxes are averaged over a geographical region to yield the average regional flux. We  
305 used the standard deviation of the regression coefficients and the relative uncertainty in the  
306 surface source, added in quadrature, to estimate the uncertainty in the flux (see Sect. 3.4.2 for  
307 fractional uncertainties). We note that the uncertainty in STILT transport (see Sect. 2.3 for  
308 details) is inherently reflected in the relative uncertainty of the regression coefficients ( $a_1, a_2, \dots$ ).

$$309 \quad Z = aHs + b + \xi \quad (2)$$

$$310 \quad Z = a_1Hs_1 + a_2Hs_2 + (a_3Hs_1Hs_2) \dots + b + \xi \quad (3)$$

311

### 312 **2.3.2 Surface Source Fields**

313 Geophysical surface source fields of remotely sensed and reanalysis data included a combination  
314 of sea ice concentration, chl *a*, absorption due to ocean detrital material, and downward  
315 shortwave radiation at the ocean surface.

316 We used daily sea ice concentration data (<https://nsidc.org/data/nsidc-0081>) at a 25 km x 25 km  
317 spatial resolution between 39.23° S and 90° S, 180° W – 180° E from the NASA National Snow  
318 and Ice Data Center Distributed Active Archive Center (NSIDC; Maslanik et al., 1999). This  
319 data reports the fraction of sea-ice cover, land-ice cover, and open water. Unfortunately, these  
320 data do not provide any information on sea ice thickness, or the presence of brine channels or  
321 melt ponds, which may modulate emissions from sea-ice covered regions. Sea ice concentration  
322 data were calculated using measurements of near-real-time passive microwave brightness  
323 temperature from the Special Sensor Microwave Image/Sounder (SSMIS) on the Defense  
324 Meteorological Satellite Program (DMSP) satellites. NSIDC sea ice concentration data were  
325 arithmetically averaged to yield 0.25° x 0.25° binned sea ice fraction for use with gridded surface  
326 influences.

327 Due to persistent cloud cover over the Southern Ocean, which often precludes the retrieval of  
328 remotely sensed ocean color data, we used 8-day mean composite Aqua MODIS L3 distributions  
329 of chl *a* from the Ocean Color Index (OCI) algorithm and absorption due to gelbstoff and detrital  
330 material at 443 nm from the Generalized Inherent Optical Properties (GIOP) model (NASA  
331 Goddard Space Flight Center, 2014). Absorption due to gelbstoff and detrital material at 443 nm  
332 is used as a proxy for colored dissolved organic matter (CDOM;  
333 <https://oceancolor.gsfc.nasa.gov/atbd/giop/>). CDOM is hypothesized to be an important source of  
334 carbon for the photochemical production of  $\text{CH}_3\text{I}$  (Moore et al., 1994). The GIOP model also  
335 publishes an uncertainty in the absorption due to gelbstoff and detrital material at 443 nm. Raw  
336 4 km x 4 km data were geometrically averaged, based on lognormal probability density  
337 functions, to a spatial resolution of 0.25° x 0.25° for use with gridded surface influences. We  
338 used the ratio of the 0.25° x 0.25° gridded uncertainty in the detrital material absorption to the  
339 absorption as the relative uncertainty for flux calculations (see Sect. 3.4.2).

340 The National Center for Environmental Prediction (NCEP) provides Final Global Data  
341 Assimilation System (GDAS/FNL) global data of downward shortwave radiation at the surface  
342 at 0.25 degree and 6-hour resolution (NCEP, 2015). We chose downward shortwave radiation  
343 for use with gridded surface influences because the photo-production of  $\text{CH}_3\text{I}$  has been observed  
344 at all visible wavelengths (Moore et al., 1994). This reanalysis data is available at a higher  
345 temporal resolution and better spatial coverage than satellite retrievals of photosynthetically  
346 active radiation (PAR) or temperature.

347

### 348 **3 Results and discussion**

#### 349 **3.1 Observed halogenated VOC patterns and relationships**

350 Zonal cross-sections of halogenated VOC data collected on ORCAS and ATom-2 illustrate  
351 unprecedented spatial sampling across our study area between the surface and 12 km (Fig. 2).  
352 Above average mixing ratios of  $\text{CH}_3\text{I}$ ,  $\text{CHBr}_3$ , and  $\text{CHClBr}_2$  typically remain confined to the  
353 lower ~2-4 km of the atmosphere (Fig. 2a, b, d). These compounds have lifetimes of  
354 approximately two months or less. Conversely, weak sources and longer lifetimes ( $\geq 3$  months)  
355 may have contributed to similar concentrations of  $\text{CH}_2\text{Br}_2$  and  $\text{CHBrCl}_2$  throughout the  
356 troposphere and above average mixing ratios as high as 8 km (Fig. 2c, e). Unfortunately, the  
357 availability of data above the detection limit and absence of BL enhancements for  $\text{CHBrCl}_2$   
358 preclude the identification of strong regional sources at this time. Meridional distributions also  
359 indicate lower latitude sources of  $\text{CH}_3\text{I}$  and  $\text{CH}_3\text{Br}$  ( $< 50^\circ \text{S}$ ), potentially resulting from terrestrial  
360 and anthropogenic contributions, and higher latitude sources ( $> 60^\circ \text{S}$ ) of  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ , and  
361  $\text{CHClBr}_2$  (Fig. 2a-d,f).

362

##### 363 **3.1.1 Observed halogenated VOC interrelationships**

364 Across our study area in both 2016 and 2017, we found that  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  exhibit a  
365 consistent enhancement ratio with each other in the bottom 2 km of the atmosphere both in  
366 Region 1 and Region 2, which suggests that these bromocarbon fluxes are closely coupled.  
367 Previous studies have documented co-located source regions of  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  in the  
368 Southern Ocean (e.g. Hughes et al., 2009; Carpenter et al., 2000; Nightingale et al., 1995;  
369 Laturnus et al., 1996), and laboratory studies have demonstrated that phytoplankton and their  
370 associated bacteria cultures, including a cold water diatom isolated from coastal waters along the  
371 Antarctic Peninsula and common to the Southern Ocean, produce both  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$   
372 (Hughes et al., 2013; Tokarczyk and Moore 1994, Sturges et al., 1993). The non-linearity  
373 observed in ratios of these two gases at low  $\text{CHBr}_3$  may reflect the different rates of their  
374 production or loss in seawater, or possibly, the influence of air masses from distant, more  
375 productive low-latitude source regions. Several studies have documented bacterially mediated  
376 loss of  $\text{CH}_2\text{Br}_2$ , but not  $\text{CHBr}_3$ , and report distinct ratios of  $\text{CH}_2\text{Br}_2$  to  $\text{CHBr}_3$  in seawater during  
377 the growth and senescent phases of a phytoplankton bloom (e.g. Carpenter et al., 2009, Hughes  
378 et al., 2013). Although this analysis is restricted to the bottom 2 km of the atmosphere, zonal

379 transport of air masses with lower ratios of  $\text{CH}_2\text{Br}_2$  to  $\text{CHBr}_3$  ratios, as have been observed in the  
380 MBL over productive, low-latitude regions, may also have influenced our observations  
381 (Yokouchi et al. 2005). Mixing ratios of  $\text{CHBr}_3$  and  $\text{CHClBr}_2$  were also correlated (Fig. 3d) in  
382 Region 2, and, a similar, weaker relationship was observed in Region 1 (Fig. 3b).  $\text{CHClBr}_2$  is a  
383 less well-studied compound than  $\text{CH}_2\text{Br}_2$ . Yet these consistent relationships suggest that  $\text{CHBr}_3$   
384 and  $\text{CHClBr}_2$  may either share some of the same sources or have sources that co-vary.

385

### 386 **3.1.2 Observed halogenated VOC relationships to $\delta(\text{O}_2/\text{N}_2)$ and $\text{CO}_2$**

387 We sought to test if the biologically mediated production of bromocarbons and oxygen result in  
388 similar atmospheric distributions. Conversely, we expected halogenated VOC atmospheric  
389 distributions and  $\text{CO}_2$  distributions to anticorrelate because  $\text{CO}_2$  fixation in surface waters is  
390 proportional to the production of oxygen.

391 For these comparisons, both  $\text{O}_2$  and  $\text{CO}_2$  mixing ratios from the upper troposphere (5-7 km) were  
392 subtracted from the data to detrend for seasonal and inter-annual variability (Fig. 4; Fig. S3). To  
393 isolate the contribution of ocean  $\text{O}_2$  fluxes, the ORCAS  $\delta(\text{O}_2/\text{N}_2)$  values reported here represent  
394 the  $\Delta\delta(\text{O}_2/\text{N}_2)$  to observed values between 5-7 km adjusted for CESM  $\text{O}_2$  land and fossil fuel  
395 contributions and the influence of air-sea  $\text{N}_2$  fluxes. In Fig. 4 we present type II major axis  
396 regression fits to data (fits were calculated using data scaled to their full range) between the  
397 ocean surface and the lowest 7 km for bromocarbons with photochemical lifetimes of  $\geq 1$  month  
398 and from the lowest 2 km for  $\text{CH}_3\text{I}$  with a photochemical lifetime of  $\sim 1$  week. We used a type II  
399 major axis regression model to balance the influences of uncorrelated processes and  
400 measurement uncertainty in halogenated VOCs (on the y-axis) and uncorrelated processes and  
401 measurement uncertainty in  $\text{O}_2$  and  $\text{CO}_2$  (on the x-axis) on the regression slope (Ayers et al.,  
402 2001; Glover et al., 2011). As noted by previous studies, simple least squares linear regressions  
403 fail to account for uncertainties in predictor variables (e.g. Cantrell et al., 2008).

404 The robust correlations of  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  with  $\delta(\text{O}_2/\text{N}_2)$ , in both 2016 and 2017 and in  
405 Region 1 and Region 2, provides support for a regional biogenic source of these two halogenated  
406 VOCs (Fig. 4a, b and Fig. 4d, e). The air-sea exchange of  $\text{O}_2$  during summer in the Southern  
407 Ocean is driven by net community production (the excess of photosynthesis over respiration) in  
408 the surface mixed layer, surface warming, and to a lesser extent ocean advection and mixing (e.g.  
409 Stephens et al., 1998; Tortell and Long 2009; Tortell et al., 2014). Note that we adjust for  
410 influences on the  $\delta(\text{O}_2/\text{N}_2)$  from thermal  $\text{N}_2$  fluxes (see Equation 1, Sect. 2.1.2 for details).  
411 Biological  $\text{O}_2$  supersaturation in the surface mixed layer develops quickly in the first several  
412 days of a phytoplankton bloom and diminishes as community respiration increases and air-sea  
413 gas exchange equilibrates the surface layer with the atmosphere on a timescale of  $\sim 1$  week.  
414  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  are emitted from phytoplankton during the exponential growth phase  
415 (Hughes et al., 2013), which often coincides with high net community production and the  
416 accumulation of  $\text{O}_2$  in surface waters. The bulk air-sea equilibration time for an excess of  $\text{CHBr}_3$   
417 and other halogenated VOCs is less than two weeks, although the photochemical loss of

418 halogenated VOCs will alter their ratio over time (see Supplement for details on calculations of  
419 bulk sea air equilibration times).

420 Our observations suggest a biological source for  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  in both Region 1 and Region  
421 2 (Fig. 4). Interestingly, the slope of the regression between  $\text{CHBr}_3$  and  $\text{O}_2$  appears distinct in  
422 Region 1 and Region 2, but between  $\text{CH}_2\text{Br}_2$  is the same. Molar enrichment ratios are  $0.20 \pm$   
423  $0.01$ , and  $0.07 \pm 0.004$  pmol : mol for  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  to  $\text{O}_2$  in Region 1, and  $0.32 \pm 0.02$ , and  
424  $0.07 \pm 0.004$  pmol : mol in Region 2. We observe a weaker relationship between  $\text{CH}_3\text{I}$  and  
425  $\text{CHClBr}_2$  and  $\text{O}_2$  in Region 1 (Fig. 4c, d), consistent with the existence of other, non-biological  
426 sources of  $\text{CH}_3\text{I}$  in this region. Figure 4f illustrates a strong relationship between  $\text{CH}_3\text{I}$  and  $\text{O}_2$ ,  
427 as well as  $\text{CHClBr}_2$  and  $\text{O}_2$ , in Region 2, however, which implies that the dominant sources of  
428  $\text{CH}_3\text{I}$  and  $\text{CHClBr}_2$  emissions over the Patagonian Shelf are biological. The corresponding molar  
429 enrichment ratios of  $\text{CH}_3\text{I}$  to  $\text{O}_2$  and  $\text{CHClBr}_2$  to  $\text{O}_2$  in Region 2 are  $0.38 \pm 0.03$  pmol : mol and  
430  $0.19 \pm 0.04$  pmol : mol, respectively.

431 In contrast to  $\text{O}_2$ , air-sea fluxes of  $\text{CO}_2$  over the Southern Ocean during summer reflect the  
432 balance of opposing thermal and biological drivers (e.g. Stephens et al., 1998; 2018). Ocean  
433 buffering chemistry results in  $\text{CO}_2$  equilibration across the air-sea interface on a timescale of  
434 several months. ORCAS observations showed a depletion of  $\text{CO}_2$  in the MBL, indicating that  
435 uptake driven by net photosynthesis dominated over thermally driven outgassing during the  
436 several months preceding the campaign (Stephens et al., 2018).  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  in the  
437 lowest 7 km were negatively correlated with  $\text{CO}_2$  in both years in Region 1 and Region 2 (Fig.  
438 S3a, b, d, e). Interestingly,  $\text{CH}_3\text{I}$  was not correlated with  $\text{CO}_2$  in Region 1, likely due to the long  
439 air-sea equilibration timescale of  $\text{CO}_2$  compared with a 9-day air-sea equilibration time and a ~7-  
440 day photochemical lifetime for  $\text{CH}_3\text{I}$ . For longer lived species, correlations for halogenated  
441 VOCs to  $\text{CO}_2$  have similar  $r^2$ -values as those for halogenated VOCs to  $\delta(\text{O}_2/\text{N}_2)$ , but model and  
442 climatological estimates of Southern Ocean  $\text{CO}_2$  fluxes are much less certain than for  $\text{O}_2$  (Anav  
443 et al., 2015; Nevison et al., 2016). As a result, we use modeled  $\text{O}_2$  fluxes as the basis for our  
444 halogenated VOC flux estimates (see Sect. 3.4.1 for details).

445

### 446 3.2 Model-observation comparisons

447 The ORCAS dataset provides an exceptional opportunity to evaluate the CAM-Chem  
448 halogenated VOC emission scheme (Ordoñez et al., 2012) at high latitudes in the Southern  
449 Hemisphere. We compared modeled halogenated VOC constituents to corresponding  
450 observations along the ORCAS flight track (Fig. 5; Fig. 6). In these figures, we used type II  
451 major axis regression models to balance the measurement uncertainty (on the y-axis) and the  
452 inherent, yet difficult to quantify representativeness and errors in a global atmospheric chemistry  
453 model (on the x-axis). We note that this comparison may favor constituents with longer  
454 photochemical lifetimes, when transport and mixing dominate over source heterogeneity.

455 In Region 1 and Region 2, both the model and observations indicate that elevated mixing ratios  
456 of  $\text{CH}_3\text{I}$  remain confined to the MBL (Fig. 5a and Fig. 6a), presumably due to its relatively short

457 photochemical lifetime. Modeled and observed CH<sub>3</sub>I are poorly correlated in Region 1 ( $r^2 =$   
458 0.20; Fig. 5b) and better correlated in Region 2 ( $r^2 = 0.70$ ; Fig. 6b). In both regions, the model  
459 most likely under predicts CH<sub>3</sub>I in the upper troposphere and lower stratosphere (UTLS), likely  
460 stemming from the poleward transport of lower latitude air masses, where CAM-Chem also  
461 exhibits a negative bias. Mixing ratio comparisons with CAM-Chem over the tropics (see Figure  
462 10 in Ordoñez et al., 2012) depict similar or larger discrepancies, and have been attributed to  
463 stronger than anticipated convective cells in the tropics. We found strong correlations and  
464 agreement to within a factor of  $\sim 2$  between modeled and observed CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> (Fig. 5c-f  
465 and Fig. 6c-f). Relatively long lifetimes ( $\geq 1$  month) in Region 1 likely enable vertical and zonal  
466 transport of CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> to the mid and upper troposphere (Fig. 5c and e). The model was  
467 biased low with respect to measurements of CH<sub>3</sub>Br by  $\sim 25\%$  in Region 1 and Region 2 (Fig. 5g-  
468 h and Fig. 6g-h), potentially as a result of an incorrect surface lower boundary condition. The  
469 model underpredicted the mean vertical gradient in CHClBr<sub>2</sub>, although it did a reasonable job of  
470 representing the mean vertical gradient in CHBrCl<sub>2</sub>, in both Region 1 and Region 2. In both  
471 cases, however, the model failed to capture the spatial variability in both CHClBr<sub>2</sub> and CHBrCl<sub>2</sub>  
472 observations (Fig. 5i-l and Fig. 6i-l). Region 2 contains stronger sources of halogenated VOCs  
473 than Region 1, which has been documented in numerous ship-based campaigns and archived in  
474 the Halocarbons in the Ocean and Atmosphere database (HalOcat; <https://halocat.geomar.de/>).  
475 Region 2 also has much higher chl *a* (Fig. S4), supporting biogenic sources for these gases.

476

### 477 **3.3 Relationships between STILT surface influence functions and observations**

478 We used the STILT model to explore the relationships between observed mixing ratios and the  
479 upstream surface influence functions (Equations 2-3) of sea ice, chl *a*, absorption due to detritus,  
480 and downward shortwave radiation at the surface, which relate to various regional hypothesized  
481 sources of halogenated VOCs such as marine phytoplankton, phytoplankton in sea ice brines,  
482 and decomposing organic matter in surface seawater (e.g. Moore and Zafiriou 1994; Moore et  
483 al., 1996; Tokarczyk and Moore 1994; Sturges et al., 1992).

484

485 We found no positive relationships between upstream sea-ice influence and any measured  
486 halogenated VOC Region 1 (Fig. 7). We interpret this result to mean that increased summertime  
487 sea ice acts either to reduce the production of halogenated VOCs by blocking sunlight or as a  
488 physical barrier to oceanic emissions of halogenated VOCs from under-ice algae. Both of these  
489 mechanisms are also consistent with a link between enhanced CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> emissions due  
490 to sea-ice retreat and surface sea-ice melt water (Carpenter et al., 2007).

491 In other studies, it has also been proposed that sea ice could be an important source for CHBr<sub>3</sub>  
492 and other halogenated VOCs, since high mixing ratios of CHBr<sub>3</sub> have been observed at the sea-  
493 ice and ice-snow interface in the austral winter (Abrahamsson et al., 2018) and in under-ice algae  
494 in the austral spring (Sturges et al., 1993). At present, CAM-Chem v1.2 with very short-lived  
495 halogen chemistry does not include a regional flux of halogenated VOCs over sea-ice covered  
496 waters in summer, and our results do not indicate a need to include one. Our data, which were

497 collected in January and February, however, cannot assess the importance of sea ice as a source  
498 of halogenated VOCs in other seasons, such as winter or spring (Abrahamsson et al., 2018;  
499 Sturges et al., 1993). More field campaigns are needed to further study the seasonality and  
500 regional strength of sea ice related halogenated VOC emissions.

501 We observed a statistically significant positive correlation between the surface influence function  
502 of 8-day satellite composites of chl *a* concentration, which is widely used as a proxy for near-  
503 surface phytoplankton biomass, and mixing ratios of CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> in Region 1 (Fig. 8a, b).  
504 This finding corroborates previous findings from ship-borne field campaigns and laboratory  
505 studies that have suggested a biogenic source for these two bromocarbons (e.g., Moore et al.,  
506 1996; Hughes et al., 2013), and further substantiates the current CAM-Chem parameterization of  
507 regional bromocarbon emissions using satellite retrievals of chl *a* in polar regions. CH<sub>3</sub>Br  
508 mixing ratios were not significantly correlated with chl *a* surface influence functions (Fig. 8c).  
509 Although potentially suggesting that marine phytoplankton and microalgae were not a strong  
510 regional source of CH<sub>3</sub>Br during ORCAS, it is also possible that the relatively long lifetime of  
511 CH<sub>3</sub>Br precludes a definitive analysis of its origin based on chl *a* using 7-day back-trajectories.  
512 Neither CHClBr<sub>2</sub> nor CHBrCl<sub>2</sub> were significantly correlated with chl *a* composite surface  
513 influence functions (data not shown); however, more observations of these short-lived species in  
514 the remote MBL are needed to substantiate this result.

515 Similar to Lai et al. (2011), we observed a significant correlation between mixing ratios of CH<sub>3</sub>I  
516 and total weekly upstream influence functions of 8-day chl *a* composites (Fig. 8d). Weaker  
517 correlations were observed with upstream influence functions on shorter timescales than seven  
518 days. We found that CH<sub>3</sub>I, particularly in Region 1, was better explained by a multi-linear  
519 regression with two predictors: 1) the influence function of downward shortwave radiation at the  
520 surface (Fig. 9a) and 2) the absorption of light due to detrital material (Fig. 9b), yielding  
521 improved agreement between predicted and observed CH<sub>3</sub>I (Fig. 9c). Several previous studies  
522 have correlated mixing ratios of CH<sub>3</sub>I to satellite retrievals of PAR and surface ocean  
523 temperature, revealing a link to solar radiation (e.g. Happell et al., 1996; Yokouchi et al., 2001).

524 Although certain species of phytoplankton are capable of producing CH<sub>3</sub>I (e.g. Manley and de la  
525 Cuesta 1997; Hughes et al., 2011), several studies also indicate a non-biological source for CH<sub>3</sub>I  
526 in the surface ocean. This non-biological source, though not fully understood, requires light, a  
527 humic like substance at the surface ocean supplying a carbon source and methyl group, and  
528 reactive iodine (Moore and Zarifou 1994; Richter and Wallace 2004). Thus far, two chemical  
529 mechanisms have been proposed for the non-biological production of methyl iodide, one – a  
530 radical recombination of a methyl group and iodine involving UV photolysis (e.g. Moore and  
531 Zarifou 1994), and two, a substitution reaction involving the reduction of an oxidant, such as iron  
532 III (e.g. Williams et al. 2007).

533

### 534 **3.4 Flux estimation**

#### 535 **3.4.1 O<sub>2</sub>-based emission estimates**

536 We present a novel approach that facilitates a basin-wide halogenated VOC flux estimate using  
537 the robust relationship between airborne observations of O<sub>2</sub> and halogenated VOCs combined  
538 with modeled O<sub>2</sub> fluxes. Unlike the existing CAM-Chem halogenated VOC biogenic flux  
539 parameterization, this method does not rely on weekly retrievals of chl *a* at high latitudes, which  
540 are often patchy. In addition, our study indicates that CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, and CHClBr<sub>2</sub> and CH<sub>3</sub>I are  
541 better correlated with marine derived O<sub>2</sub> than the upstream influence of chl *a*.

542 For CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, and CHClBr<sub>2</sub> we construct ocean emission inventories for January and  
543 February using a scaled version of gridded modeled air-sea O<sub>2</sub> fluxes and the slopes (i.e. molar  
544 ratios) of linear correlations between δ(O<sub>2</sub>/N<sub>2</sub>) and halogenated VOC mixing ratios (Fig. 10). O<sub>2</sub>  
545 fluxes were obtained from simulations using a configuration of the CESM model nudged to  
546 reanalysis temperatures and winds as described in Stephens et al. (2018). An earlier free running  
547 version of CESM was one of the best evaluated for reproducing the seasonal cycle of δ(O<sub>2</sub>/N<sub>2</sub>)  
548 over the Southern Ocean (Nevinson et al., 2015; 2016). To date, the north-south gradient in  
549 atmospheric O<sub>2</sub> has not been well reproduced by any models (Resplandy et al., 2016). Vertical  
550 gradients in O<sub>2</sub> on ORCAS indicate that CESM overestimated gradients by 47% on average;  
551 accordingly, O<sub>2</sub> fluxes were adjusted downward by 47% to better match the observations. This is  
552 obviously a very simple adjustment to the modeled fluxes, and the actual air-sea O<sub>2</sub> flux biases in  
553 CESM likely have a great deal of spatial and temporal heterogeneity. We calculated an  
554 uncertainty for the CESM flux using a second, independent estimate of O<sub>2</sub> fluxes based on  
555 dissolved O<sub>2</sub> measurements in surface seawater. The Garcia and Keeling (2001) climatology has  
556 much smoother temporal and spatial patterns than CESM flux estimates but also results in  
557 overestimated atmospheric O<sub>2</sub> spatial gradients. We calculate the relative uncertainty in O<sub>2</sub> flux  
558 as the ratio of the mean absolute difference between gridded Garcia and Keeling values (2001;  
559 also adjusted down by 51 % everywhere to better match ORCAS observations) to the CESM  
560 model flux estimates in Regions 1 and 2 (adjusted down by 47% everywhere). These  
561 disagreements were 7.3 % and 3.4 % for Regions 1 and 2, respectively. Based on the ratios of  
562 halogenated VOC to O<sub>2</sub> mixing ratios in bivariate least squares regressions and these adjusted O<sub>2</sub>  
563 fluxes, we estimate mean emissions of CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> in Region 1 and Region 2. Relative  
564 uncertainty in the slopes (i.e., the standard deviation of the slopes) from these regressions and the  
565 mean relative uncertainties in regional O<sub>2</sub> fluxes were added in quadrature to yield uncertainties  
566 in calculated halogenated VOC emission rates.

567

568 Figure 10 shows the mean emissions for Jan. and Feb. of CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, and CHClBr<sub>2</sub> in  
569 Region 1 and Region 2. Mean regional emissions of CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> and CHClBr<sub>2</sub> are 91 ± 8,  
570 31 ± 17, and 11 ± 4 pmol m<sup>-2</sup> hr<sup>-1</sup> in Region 1 and 329 ± 23, 69 ± 5, and 24 ± 5 pmol m<sup>-2</sup> hr<sup>-1</sup> in  
571 Region 2 (Table 1). The mean flux of CH<sub>3</sub>I in Region 2 is 392 ± 32 (Table 1). Table 1 also lists  
572 the mean Jan. and Feb. CAM-Chem emissions from Region 1 and Region 2, as well as emissions  
573 from several other observational and modeling Antarctic polar studies. Our estimates fall within  
574 the range of these other studies, which span every month of the year and whose estimated fluxes  
575 range from negative (i.e. from the atmosphere into the ocean) to 3500 pmol m<sup>-2</sup> hr<sup>-1</sup> CHBr<sub>3</sub> in a  
576 coastal bay during its peak in primary production. CAM-Chem emissions for all species are

577 significantly lower than our observationally derived values in Region 1, with the exception of  
578 CH<sub>3</sub>I. Conversely, CAM-Chem emissions are significantly higher than our estimated emissions  
579 in Region 2, with the exception of CHClBr<sub>2</sub> in Region 1, which remains underpredicted by the  
580 model (Table 1). We note that in Region 2, CAM-Chem fluxes of CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub>, although  
581 still significantly different, are more similar to our estimated fluxes.

582

### 583 3.4.2 STILT-based emission estimates

584 Similar to our O<sub>2</sub>-based emission estimates, we used the relationship between surface influence  
585 functions and CH<sub>3</sub>I mixing ratios (Fig. 9) to predict a flux field in Region 1 (Fig. 11). We used a  
586 multiple linear regression ( $\pm$  1 standard deviations; Equation 2), where Hs<sub>1</sub> and Hs<sub>2</sub> are the  
587 downward shortwave radiation and detrital absorption surface influence functions, respectively,  
588 with an intercept  $b = 0.19 \pm 0.01$ , and influence coefficients  $a_1 = 3.7E-5 \pm 1.3E-5$ ,  $a_2 = 3.5 \pm 0.74$ ,  
589 and an interaction term with the coefficient  $a_3 = -5.2E-4 \pm 1.5E-4$  (Fig.9c). These regression  
590 coefficients and interaction term were used to estimate an average non-biological flux of CH<sub>3</sub>I  
591 (Fig. 11; Table 1). This method could be used in place of the current Bell et al. (2002)  
592 climatology to update near weekly (~8 day) emissions of CH<sub>3</sub>I in future versions of CAM-Chem.  
593 Our estimated mean CH<sub>3</sub>I flux in Region 1 ( $35 \pm 29$  pmol m<sup>-2</sup> hr<sup>-1</sup>) is significantly lower than the  
594 current CAM-Chem estimated emissions (Table 1). As noted in Sect. 3.2, our observations of  
595 CH<sub>3</sub>I are also much lower than the modeled mixing ratios. As discussed above, the strong  
596 correlations between CH<sub>3</sub>I and O<sub>2</sub> in Region 2 also suggest a dominant biological source for this  
597 compound in this region. As a result, we have not used this relationship to parameterize a flux  
598 for CH<sub>3</sub>I in Region 2 (see Sect. 3.1.2 and 3.4.1 for details). We note that although it would be  
599 possible to provide STILT-based emission estimates for other halogenated VOCs (e.g. CHBr<sub>3</sub>  
600 and CH<sub>2</sub>Br<sub>2</sub>), the correlations these compounds were less strong with surface influence functions  
601 than those with O<sub>2</sub>/N<sub>2</sub>.

602

## 603 4 Conclusions

604 Our work combined TOGA and AWAS halogenated VOC airborne observations from the  
605 ORCAS and ATom-2 campaigns, with coincident measurements of O<sub>2</sub> and CO<sub>2</sub>, geophysical  
606 datasets and numerical models, including the global atmospheric chemistry model CAM-Chem,  
607 and the Lagrangian transport model, STILT. We evaluated model predictions, calculated molar  
608 enrichment ratios, inferred regional sources, and provided novel means of parameterizing ocean  
609 fluxes. We found that the Southern Ocean MBL is enriched in halogenated VOCs, but that these  
610 MBL enhancements are less pronounced at higher latitudes, i.e., poleward of 60° S (Region 1)  
611 than over the productive Patagonian shelf (Region 2).. Overall, our results indicated that the  
612 Southern Ocean is a moderate regional sources of CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, and CH<sub>3</sub>I, and a weak source  
613 of CHClBr<sub>2</sub> and CHBrCl<sub>2</sub> in January and February. Good model-measurement correlations were  
614 obtained between our observations and simulations from the Community Earth System Model  
615 (CESM) atmospheric component with chemistry (CAM-Chem) for CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, CH<sub>3</sub>I, and  
616 CHClBr<sub>2</sub> but all showed significant differences in model:measurement ratios. The



617 model:measurement comparison for  $\text{CH}_3\text{Br}$  was satisfactory and for  $\text{CHBrCl}_2$  the low levels  
618 present precluded us from making a complete assessment.

619  $\text{CHBr}_3$  and  $\text{CH}_2\text{Br}_2$  exhibited strong and robust correlations with each other and with  $\text{O}_2$  and  
620 weaker but statistically significant correlations with the influence of chl  $a$ , which is a proxy for  
621 phytoplankton biomass.  $\text{CHClBr}_2$  and  $\text{CHBr}_3$  were well correlated with one another, particularly  
622 in Region 2. Together, these correlations suggested a biological source for these gases over the  
623 Southern Ocean. We found that  $\text{CH}_3\text{I}$  mixing ratios in Region 1 were best correlated with a non-  
624 biological surface influence function, although biogenic  $\text{CH}_3\text{I}$  emissions appear important in  
625 Region 2.

626

627 Our flux estimates based on the relationship of halogenated VOC mixing ratios to  $\text{O}_2$  and  
628 remotely sensed parameters (for  $\text{CH}_3\text{I}$ ) were compared with those derived from global models  
629 and ship-based studies (Table 1). Our emission estimates of  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ , and  $\text{CHClBr}_2$  are  
630 significantly higher than CAM-Chem's globally prescribed emissions in Region 1, where  
631 halogenated VOC mixing ratios are under predicted (Table 1; Fig. 5). Similarly, our estimate of  
632  $\text{CHClBr}_2$  emissions is also significantly higher than CAM-Chem's in Region 2, where  $\text{CHClBr}_2$   
633 mixing ratios remained underpredicted. Yet, to the best of our knowledge, CAM-Chem's global  
634 parameterization of halogenated VOC fluxes has not been compared with data at high latitudes.  
635 Indeed, our emission estimates of  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ ,  $\text{CH}_3\text{I}$  fall within a range of CAM-Chem's  
636 estimates (on the low end) and most prior estimates based on either other models or localized  
637 studies using seawater-side measurements from the Antarctic polar region in summer (on the  
638 high end). In the case of  $\text{CH}_3\text{I}$ , our estimated emissions suggest that the prescribed emissions in  
639 CAM-Chem may be too high in Region 1 and Region 2. Our parameterizations of the  $\text{CH}_3\text{I}$  flux  
640 could be used to explore inter-annual variability in emissions, which is not captured by the Bell  
641 et al. (2002)  $\text{CH}_3\text{I}$  climatology currently employed in CAM-Chem.

642 To extend these relationships to year-round and global parameterizations for use in global  
643 climate models, they must be studied using airborne observations in other seasons and regions.  
644 These approaches may help parameterize emissions of new species that can be correlated with  
645 surface influence functions or the biological production of oxygen or may improve existing  
646 emissions, where persistent biases exist. Finally, future airborne observations of halogenated  
647 VOCs have the potential to further improve our understanding of air-sea flux rates and their  
648 drivers for these chemically and climatically important gases over the Southern Ocean.

649 *Data Availability.* The ORCAS and ATom-2 datasets are publically available at  
650 <https://doi.org/10.5065/D6SB445X> ; ([www.eol.ucar.edu/field\\_projects/orcas](http://www.eol.ucar.edu/field_projects/orcas)) and  
651 <https://doi.org/10.3334/ORNLDAAAC/1581>.

652 *Author Contributions.* EA is responsible for the bulk of the conceptualization, formal analysis,  
653 writing, review, and editing with contributions from all authors. BBS and ECA were  
654 instrumental in the investigation and supervision related to this manuscript. RSH contributed to  
655 the conceptualization, as well as the investigation and halogenated VOC data curation for this  
656 project. BBS, EJM, and RFK were responsible for the data curation of  $\delta(\text{O}_2/\text{N}_2)$  data and  
657 contributed to formal analysis involving these data. MSHM along with EAK were responsible  
658 for STILT data curation and formal analysis, and the conceptualization and formal analysis of

659 SITLT-based geostatistical influence functions and flux estimates were also informed by these  
660 two. DK, along with ST, JFL and ASL were responsible for constructing CAM halogenated  
661 VOC emissions and conducting CAM runs. MCL was responsible for CESM simulations  
662 yielding O<sub>2</sub> fluxes and comparing this product alongside the Garcia and Keeling O<sub>2</sub> climatology  
663 in CAM. KMC and CM were responsible for the data curation of CO<sub>2</sub> observations. AJH  
664 contributed to the investigation for halogenated VOC data.

665

666 *Acknowledgements.* We would like to thank the ORCAS and ATom-2 science teams and the  
667 NCAR Research Aviation Facility and NASA DC-8 pilots, technicians and mechanics for their  
668 support during the field campaigns. In addition, we appreciate the NCAR EOL staff who have  
669 facilitated computing and data archival. In particular, we thank Tim Newberger for his help in  
670 supporting the NOAA Picarro CO<sub>2</sub> observations and Andrew Watt for his help in supporting the  
671 AO<sub>2</sub> O<sub>2</sub> observations. This work was made possible by grants from NSF Polar Programs  
672 (1501993, 1501997, 1501292, 1502301, 1543457), NSF Atmospheric Chemistry Grants  
673 1535364, 1623745, and 1623748 and NASA funding of the EVS2 Atmospheric Tomography  
674 (ATom) project, as well as the support of the NCAR Advanced Study Program (ASP)  
675 Postdoctoral Fellowship Program and computing support from Yellowstone, provided by  
676 NCAR's Computational and Information Systems Laboratory. The National Center for  
677 Atmospheric Research is sponsored by the National Science Foundation.

678

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995



997 **Tables**

998 Table 1. Mean  $\pm$  uncertainty (see Sect. 3.4.1 and 3.4.2 for details) halogenated VOC emission  
 999 estimates ( $\text{pmol m}^{-2} \text{hr}^{-1}$ ) in Region 1 and Region 2 calculated in this study (with method  
 1000 indicated below each value), from CAM-Chem (Ordoñez et al., 2012) and from several other  
 1001 modeling and ship-based observational studies.

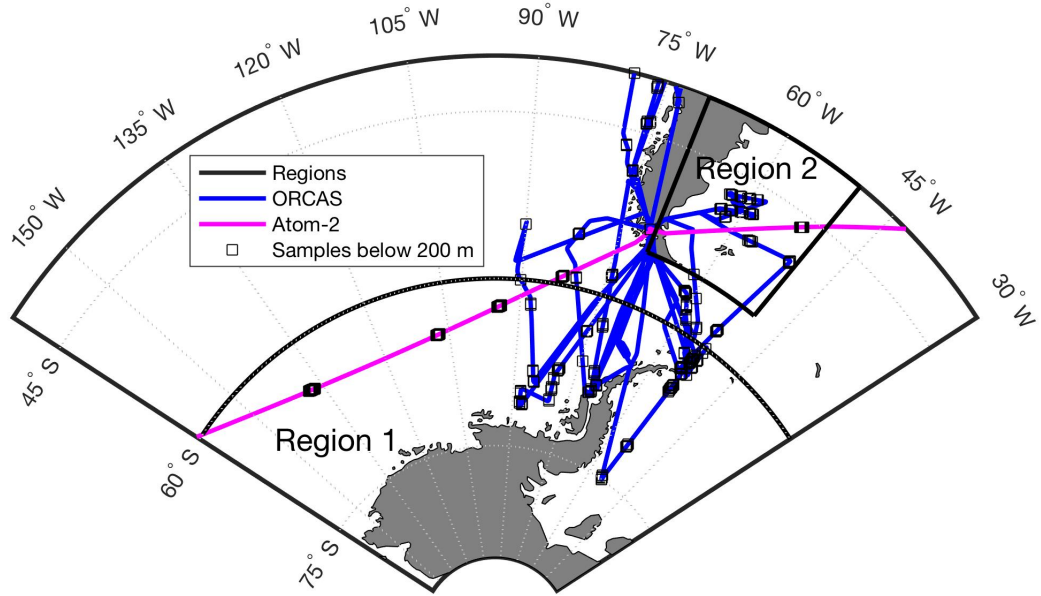
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Region/Months	$\text{CHBr}_3$	$\text{CH}_2\text{Br}_2$	$\text{CH}_3\text{I}$	$\text{CHClBr}_2$	Reference
Region 1 (JF) < 60° S	91 $\pm$ 8 O <sub>2</sub> Repr.	31 $\pm$ 18 O <sub>2</sub> Repr.	35 $\pm$ 29 MLR	11 $\pm$ 4 O <sub>2</sub> Repr.	This Study
Region 2 (JF) >55° S and <40° S	329 $\pm$ 23 O <sub>2</sub> Repr.	69 $\pm$ 5 O <sub>2</sub> Repr.	392 $\pm$ 32 O <sub>2</sub> Repr.	25 $\pm$ 5 O <sub>2</sub> Repr.	This Study
Region 1 (JF)	10	1.9	120	0.38	CAM-Chem
Region 2 (JF)	360	44	800	8.7	CAM-Chem
Southern Ocean ( $\geq 50^\circ\text{S}$ ), (DJ)	200	200	200	-----	Ziska et al. 2013 (model)
Marguerite Bay (DJF)	3500	875	-----	-----	Hughes et al. 2009 (obs)
70°S-72°S Antarctica	1300	-----	-----	-----	Carpenter et al. 2007 (obs)
Southern Ocean ( $\geq 50^\circ\text{S}$ ) (Feb. - April)	225	312	708	-----	Butler et al. 2007 (obs)
40°S-52°S S. Atlantic (Sept.- Feb.)	-1670	-----	250	-----	Chuck et al. 2005
Southern Ocean ( $\geq 50^\circ\text{S}$ ), (DJ)	-330	-----	-----	-----	Mattson et al. 2013 (model)

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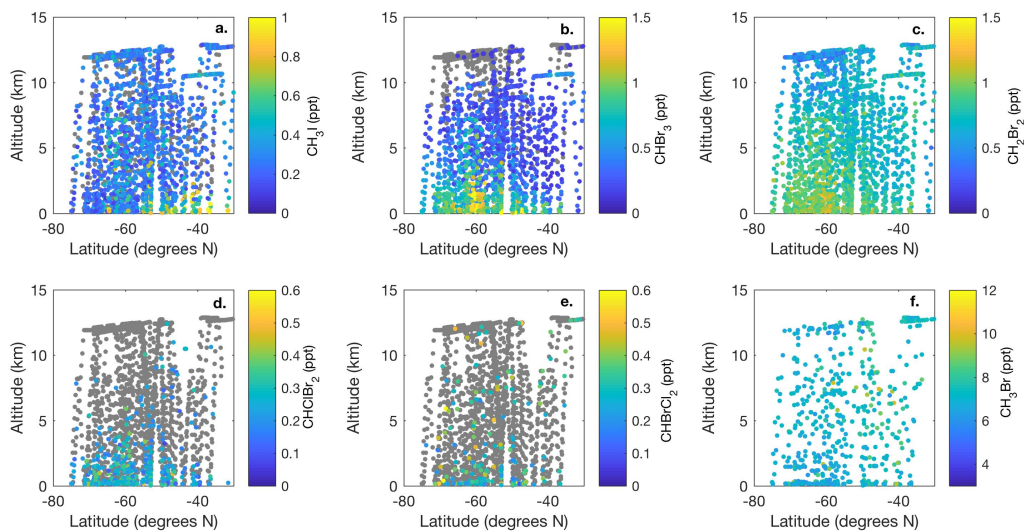


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1008 **Figure 1.** Overview map ORCAS and ATom-2 flight tracks in the study regions: 1) high  
1009 latitudes in the Southern Hemisphere poleward 60° S and 2) the Patagonian Shelf. The ORCAS  
1010 and ATom-2 aircraft flights and dips below 200 m that took place within these regions are also  
1011 shown.

1012

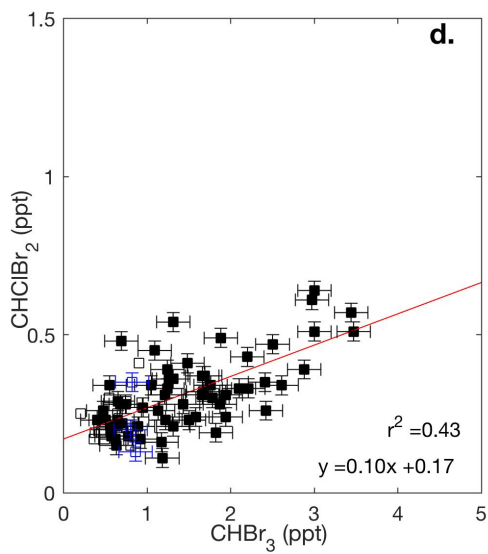
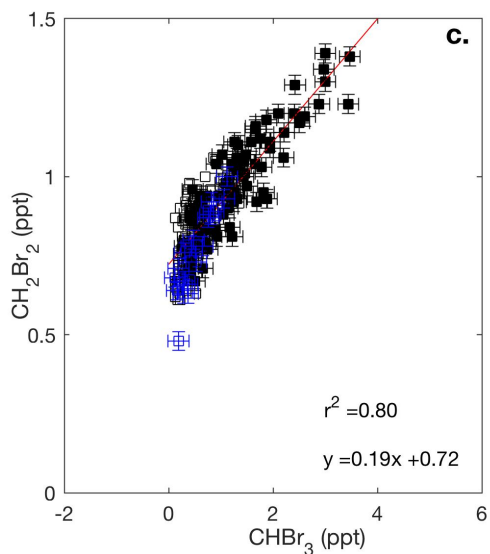
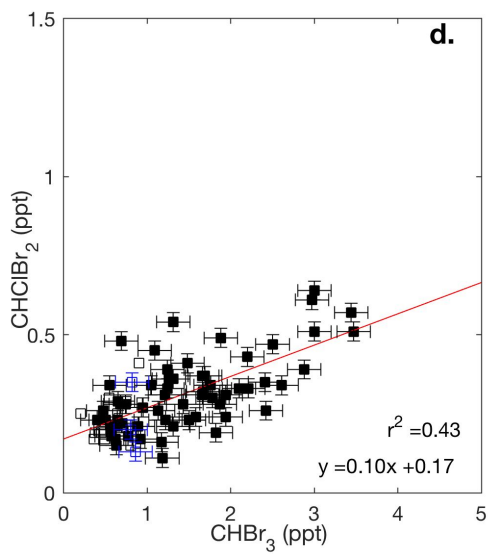
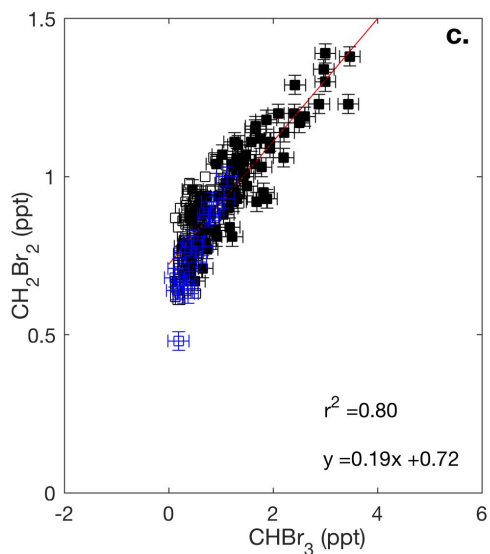
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1015 **Figure 2.** Meridional-altitudinal cross-sections of mixing ratios of a)  $\text{CH}_3\text{I}$ , b)  $\text{CHBr}_3$ , c)  
 1016  $\text{CH}_2\text{Br}_2$ , d)  $\text{CHClBr}_2$ , and e)  $\text{CHBrCl}_2$  from the TOGA and mixing ratios of f)  $\text{CH}_3\text{Br}$  from  
 1017 AWAS and WAS in 2016 and 2017, respectively, during the ORCAS and ATom-2 campaigns  
 1018 over the Southern Ocean in the austral summer. Note the different color bar scales. Gray points  
 1019 denote measurements below the detection limit of each species ( $\text{CH}_3\text{I}$  - 0.03 ppt,  $\text{CHBr}_3$  - 0.2  
 1020 ppt,  $\text{CH}_2\text{Br}_2$  - 0.03 ppt,  $\text{CHClBr}_2$  - 0.03 ppt,  $\text{CHBrCl}_2$  - 0.05 ppt,  $\text{CH}_3\text{Br}$  - 0.2 ppt).

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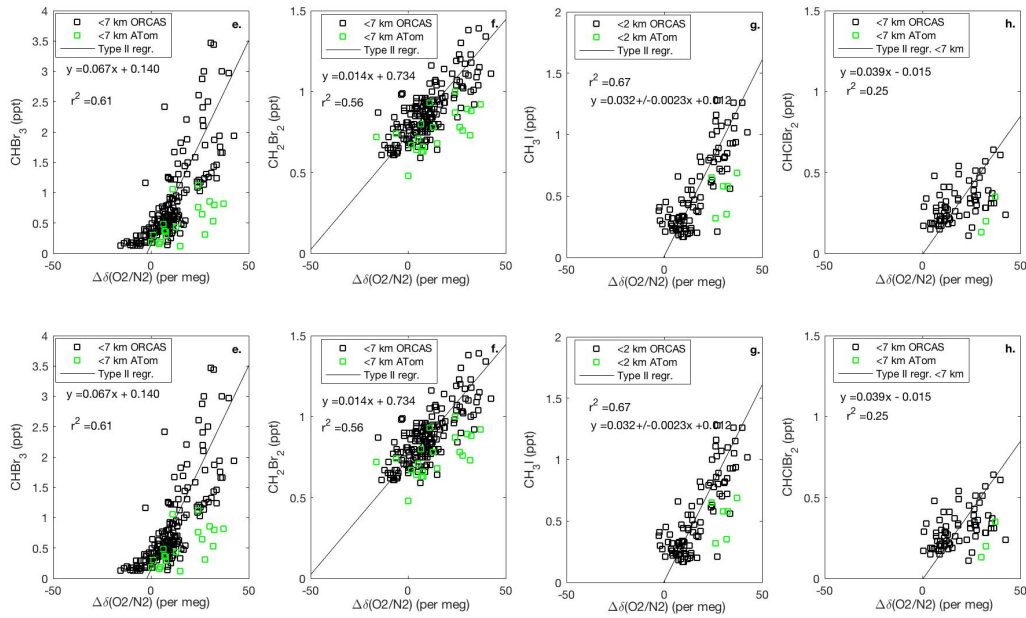
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**Figure 3.** Mixing ratios of  $\text{CHBr}_3$  vs.  $\text{CH}_2\text{Br}_2$  and  $\text{CHClBr}_2$  across the ORCAS and ATom-2 campaigns in Region 1 (Fig.3a,b) and in (Fig.3c,d), respectively. Type II major axis regression model (bivariate least squares regressions) are based on ORCAS data below 2 km and illustrate regional enhancement ratios. Error bars represent the uncertainty in halogenated VOC measurements.

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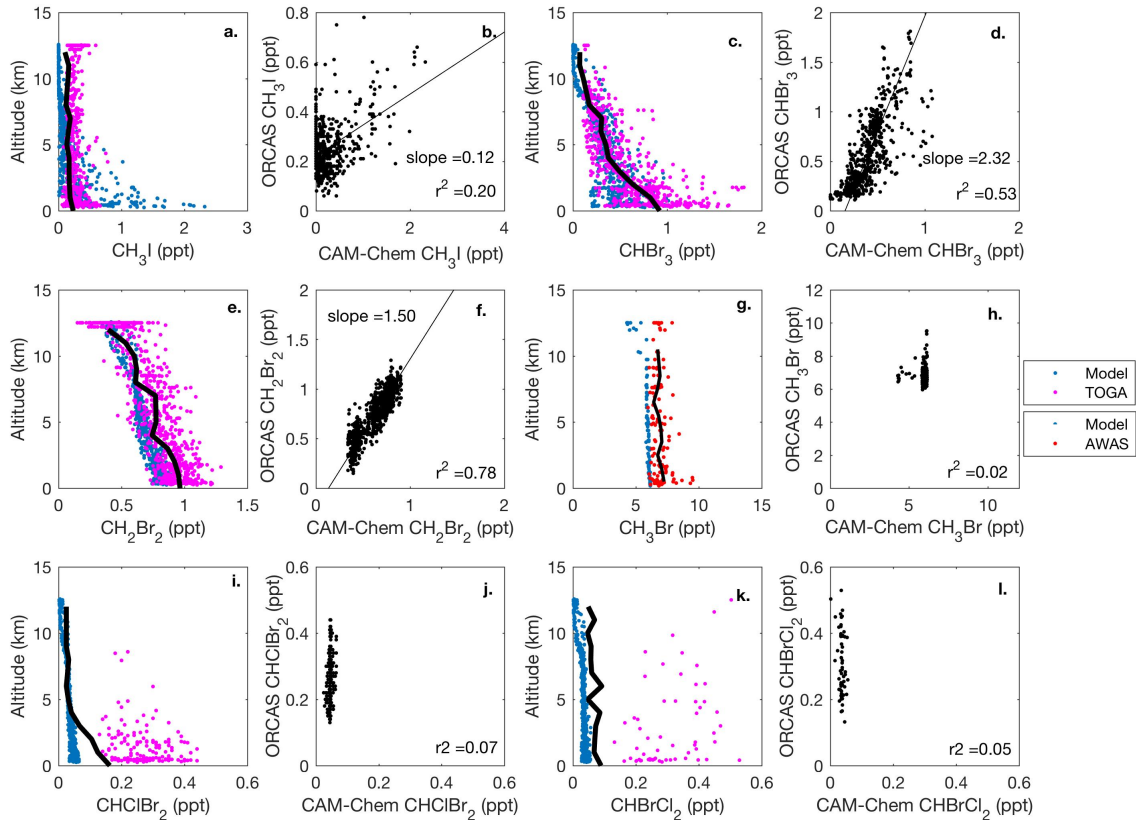
1038 **Figure 4.** Mixing ratios of  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ , and  $\text{CH}_3\text{I}$  vs.  $\text{O}_2$  on ORCAS and ATom-2 in Region  
1039 1, poleward of  $60^\circ \text{S}$  (a-d) and Region 2 over the Patagonian Shelf (e-h). Slopes  $\pm$  standard  
1040 errors from type II major axis regression model (bivariate least squares regression) fits of  
1041 ORCAS data for regressions with  $r^2 > 0.2$  (fits were calculated on variables scaled to their full  
1042 range). The slopes reported in the figure are converted to pmol:mol ratios prior to estimating  
1043 biogenic halogenated VOC fluxes based on modeled CESM  $\text{O}_2$  fluxes. Data from above 7 km  
1044 were excluded due to the influence of air masses transported from further north.

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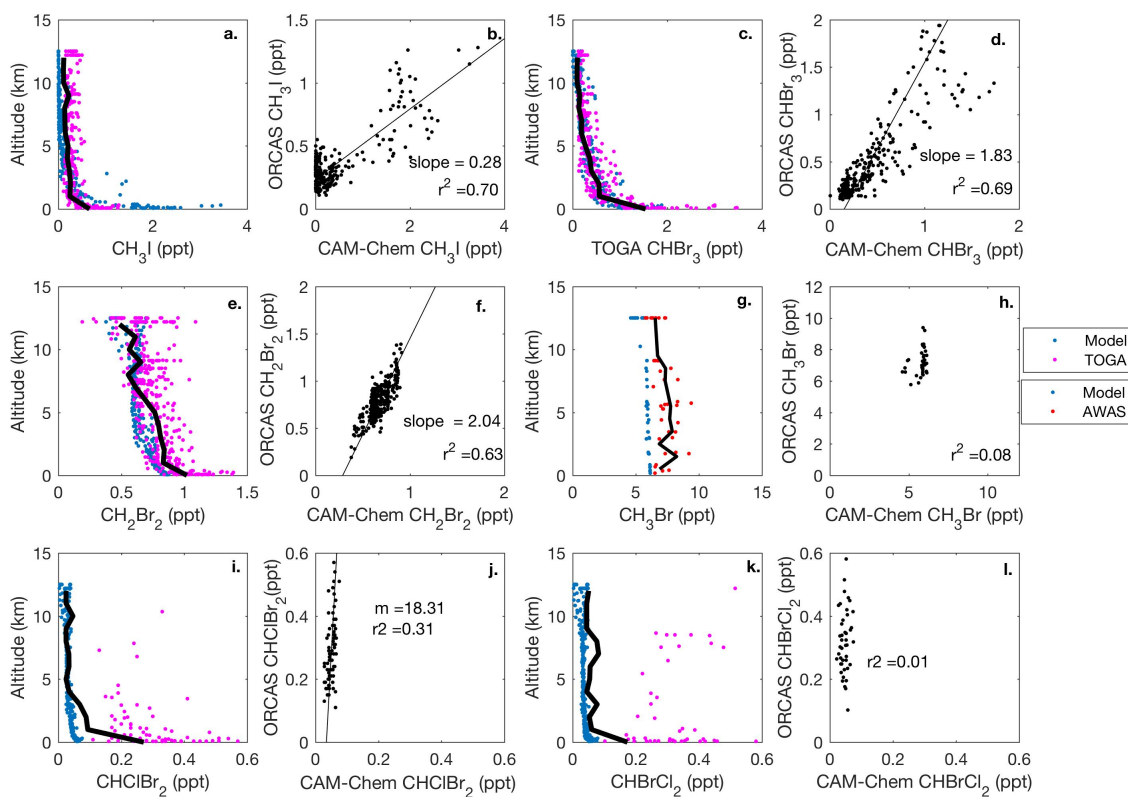
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1049 **Fig 5.** CAM-Chem1.2 model-aircraft measurement comparison during the ORCAS campaign  
1050 between 1-12 km in Region 1, high latitudes in the Southern Hemisphere poleward 60° S. All  
1051 regressions are type II major axis regression models bivariate least squares regressions (slopes  
1052 are shown when the  $r^2 \geq 0.2$ ). The bold, black line in each vertical profile represents the binned  
1053 (mean) mixing ratio of halogenated VOC measurements at that altitude. The binned mean  
1054 includes measurements below the detection limit (DL), which for this calculation are assigned a  
1055 value equal to the DL multiplied by the percentage of data below detection. Modeled values  
1056 include locations where observations were below the DL.

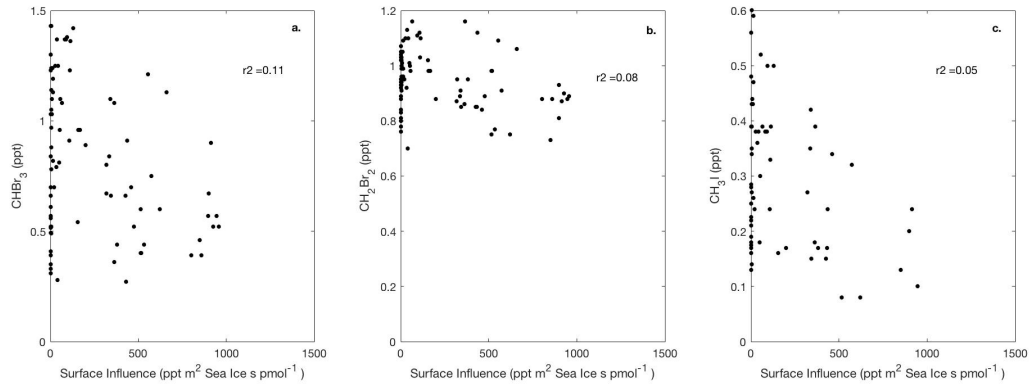


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1058 **Figure 6.** CAM-Chem 1.2 model-aircraft measurement (TOGA and AWAS) comparison during  
 1059 ORCAS campaign between 1-12 km in Region 2, the Patagonian Shelf. All regressions are type  
 1060 II major axis regression models bivariate least squares regressions (slopes are shown when the  $r^2$   
 1061  $\geq 0.2$ ). The bold, black line in each vertical profile represents the binned (mean) mixing ratio of  
 1062 halogenated VOC measurements at that altitude. Again, the binned mean includes measurements  
 1063 below the detection limit (DL), which for this calculation are assigned a value equal to the DL  
 1064 multiplied by the percentage of data below detection. Modeled values include locations where  
 1065 observations were below the DL.

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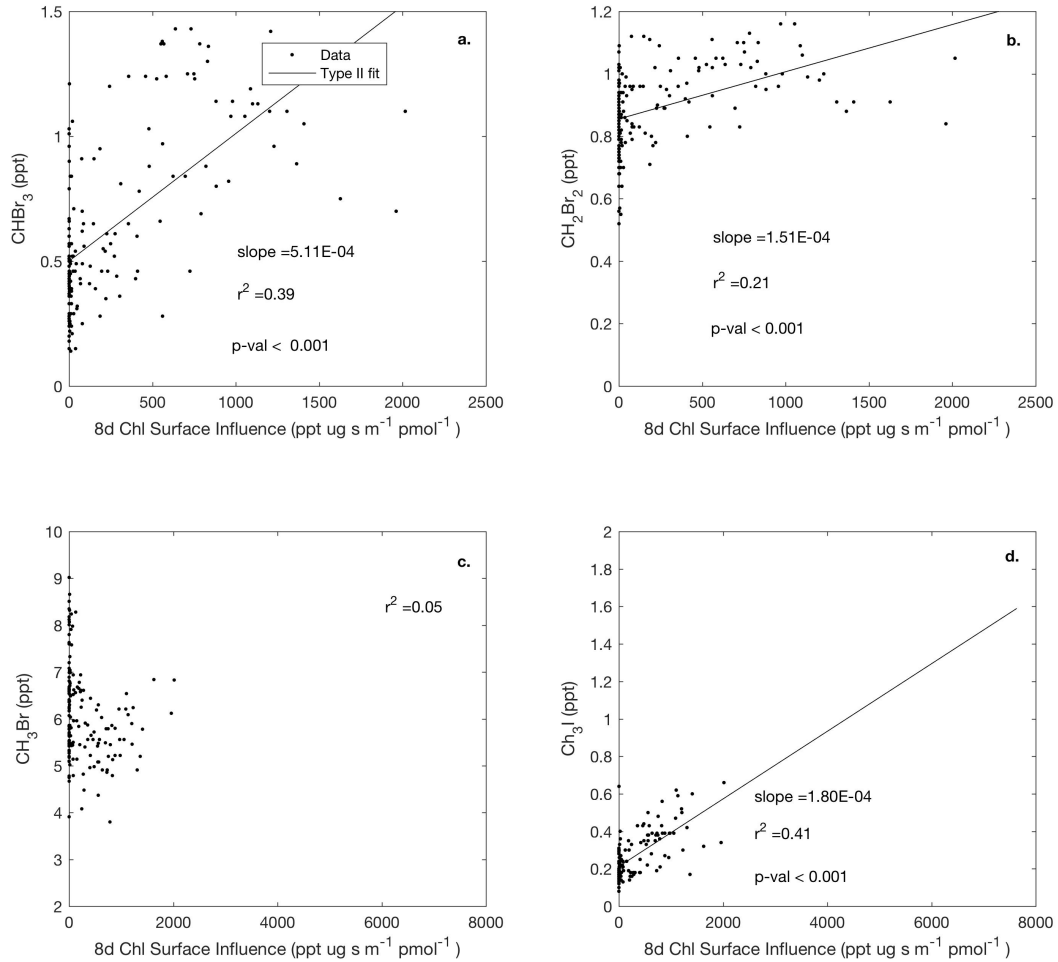
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1069 **Figure 7.** Linear type II regressions between influence functions convolved with sea ice  
 1070 distributions, which exclude land ice, and mixing ratios for  $\text{CHBr}_3$ ,  $\text{CH}_2\text{Br}_2$ , and  $\text{CH}_3\text{I}$  in Region  
 1071 1, poleward of  $60^\circ \text{S}$ . Surface influence ( $\text{ppt m}^2 \text{ s pmol}^{-1}$ ) in each grid cell was multiplied by  
 1072 fractional sea ice concentration surface field, which is unit-less, yielding sea ice surface  
 1073 influence function units of  $\text{ppt m}^2 \text{ s pmol}^{-1}$ , as shown on the x-axis. Linear regression lines are not  
 1074 shown, as  $p \geq 0.001$ .

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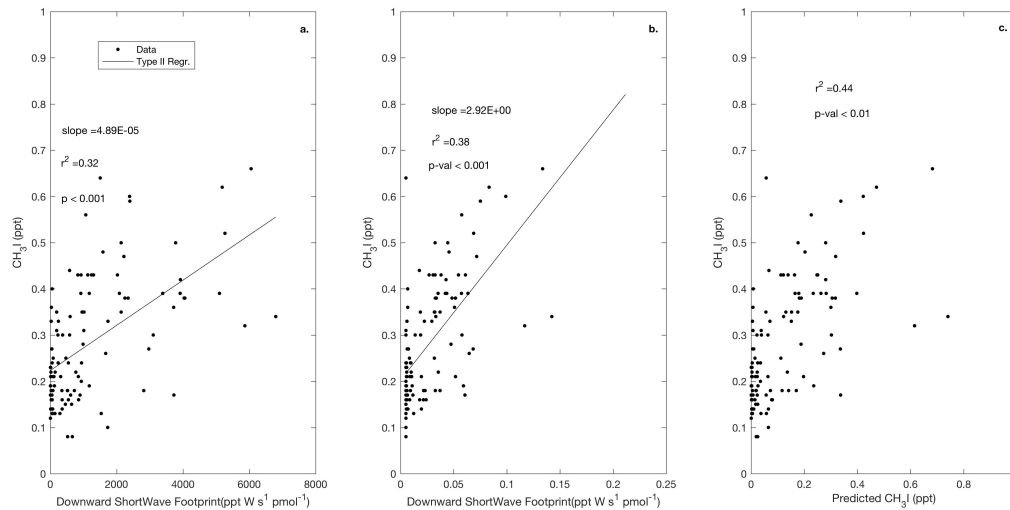


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1078 **Figure 8.** Linear type II regressions between influence functions of eight day composites of chl  
 1079 *a* and mixing ratios of halogenated VOCs (a-d) poleward of 60° S (Region 1). Surface influence  
 1080 (ppt m<sup>2</sup> s pmol<sup>-1</sup>) in each grid cell was multiplied by the chl *a* (μg m<sup>-3</sup>) surface field, resulting in  
 1081 surface influence function units of μg ppt s pmol<sup>-1</sup> m<sup>-1</sup>, shown on the x-axis. Linear regression  
 1082 lines are shown where when p < 0.001.

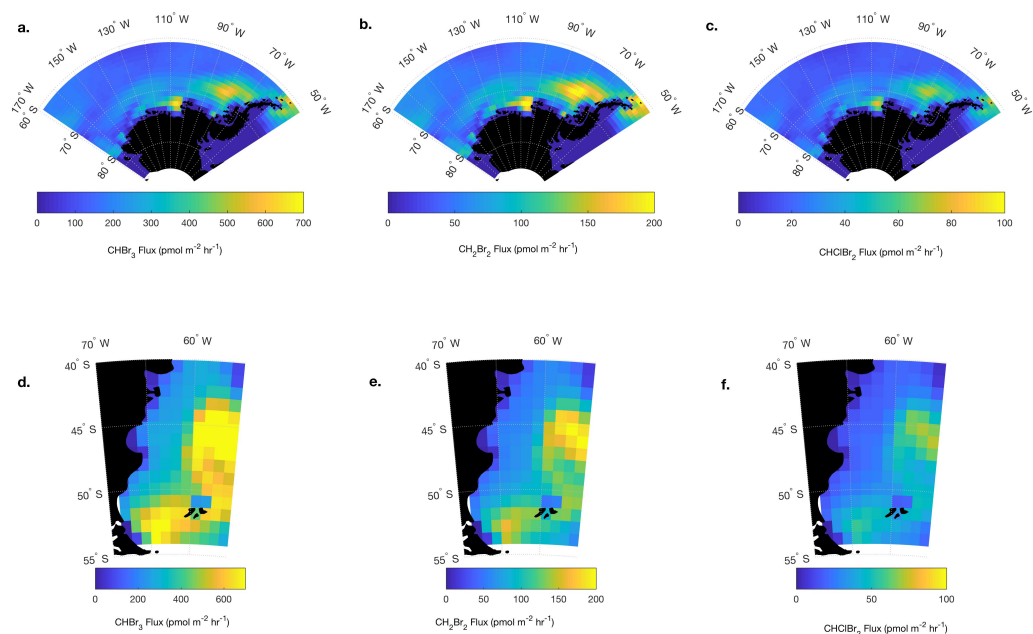
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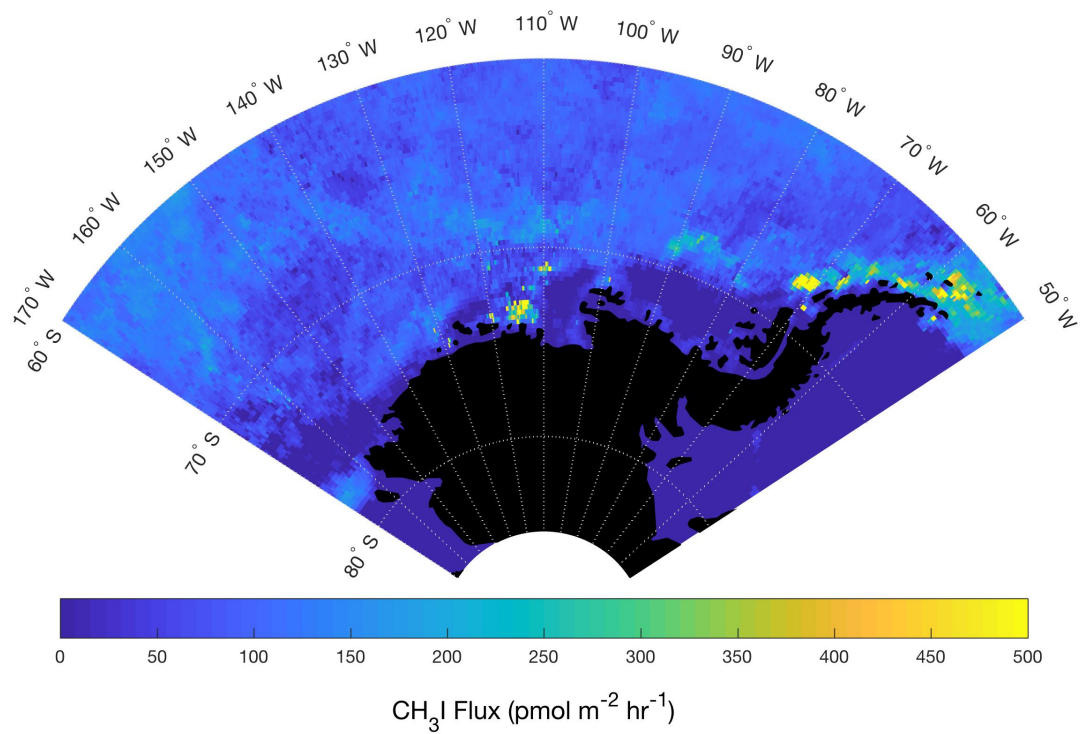
1086 **Figure 9.** Observed CH<sub>3</sub>I plotted against the surface influence functions of downward shortwave  
 1087 radiation (a) and absorption due to detritus (b). Predicted mixing ratios of CH<sub>3</sub>I based on a  
 1088 multiple linear regressions (MLR) using these two predictors in Region 1 are shown in Fig. 9c  
 1089 according to Equation 3. Surface influence (ppt m<sup>2</sup> s pmol<sup>-1</sup>) in each grid cell was multiplied by  
 1090 the surface source field, such as shortwave radiation at the surface (W m<sup>-2</sup>), yielding units of ppt  
 1091 Ws pmol<sup>-1</sup>, and the surface ocean's detrital absorption (m<sup>-1</sup>), yielding units of ppt m s pmol<sup>-1</sup>,  
 1092 shown on the x-axes.

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1094  
 1095 **Figure 10.** Resulting mean Jan. – Feb. 2016 O<sub>2</sub>-based (parameterized) CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub> and  
 1096 CHClBr<sub>2</sub> fluxes (pmol m<sup>-2</sup> s<sup>-1</sup>) in Region 1 (a-c) poleward of 60° S and Region 2 (d-f) over the  
 1097 Patagonian Shelf. CESM modeled O<sub>2</sub> fluxes are scaled by the slope between the oceanic  
 1098 contribution to δ(O<sub>2</sub>/N<sub>2</sub>) and CHBr<sub>3</sub> and CH<sub>2</sub>Br<sub>2</sub>, and CHClBr<sub>2</sub> reported in Fig. 4. Note that these  
 1099 fluxes represent mean estimated biogenic fluxes in Jan. -Feb. 2016 (see Sect. 3.4.1 for details).

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1101  
1102 **Figure 11.** Mean estimated CH<sub>3</sub>I fluxes for Jan. – Feb. The multilinear regression in Fig. 9  
1103 between CH<sub>3</sub>I mixing ratios and geophysical influence functions related to shortwave radiation  
1104 and detrital material at the sea surface was used to derive a mean flux field in Jan.-Feb., 2016 for  
1105 Region 1.  
1106