

## Responses to the comments of Referee #1

It is recommended that the manuscript be edited for English language use prior to publication. There are minor grammatical corrections needed throughout and occasional need for technical revision. One example is on p.6 line 164 where the authors state that "the NO measurements were basically reliable", which is not a sufficient description.

**Reply:** Thank you for the comments. We have made final corrections on grammar errors and some insufficient expressions throughout the main text. Please refer to the main text and supplementary information for detail.

1 **Causes of a continuous summertime O<sub>3</sub> pollution event in Ji'nan, a central**  
2 **city in the North China Plain**

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17

18 **Abstract:** In summer 2017, measurements of ozone (O<sub>3</sub>) and its precursors were carried out at an  
19 urban site in Ji'nan, a central city in the North China Plain (NCP). A continuous O<sub>3</sub> pollution  
20 event was captured during August 4-11, with the maximum hourly O<sub>3</sub> reaching 154.1 ppbv.  
21 Model simulation indicated that local photochemical formation and regional transport  
22 contributed 14.0±2.3 and 18.7±4.0 ppbv/hr to the increase of O<sub>3</sub> during 09:00-15:00 local time  
23 (LT) in this event, respectively. For local O<sub>3</sub> formation, the calculated OH reactivities of volatile  
24 organic compounds (VOCs) and carbon monoxide (CO) were comparable between O<sub>3</sub> episodes  
25 and non-episodes ( $p>0.05$ ), so was the OH reactivity of nitrogen oxides (NO<sub>x</sub>). However, the  
26 ratio of OH reactivity of VOCs and CO to that of NO<sub>x</sub> increased from 2.0±0.4 s<sup>-1</sup>/s<sup>-1</sup> during non-  
27 episodes to 3.7±0.7 s<sup>-1</sup>/s<sup>-1</sup> during O<sub>3</sub> episodes, which resulted in the change of O<sub>3</sub> formation

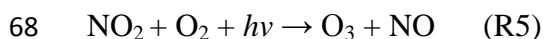
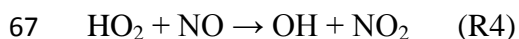
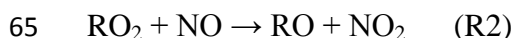
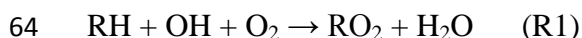
28 mechanism from the VOC-limited regime before the O<sub>3</sub> pollution event to the transitional regime  
29 during the event. Correspondingly, the simulated local O<sub>3</sub> production rate during the event  
30 (maximum: 21.3 ppbv/hr) was markedly higher than that before the event ( $p < 0.05$ ) (maximum:  
31 16.9 ppbv/hr). Given that gasoline and diesel exhaust made large contributions to **the abundance**  
32 **of** O<sub>3</sub> precursors and O<sub>3</sub> production rate, constraint on vehicular emissions is the most effective  
33 strategy to control O<sub>3</sub> pollution in Ji'nan. The NCP has been confirmed as a source region of  
34 tropospheric O<sub>3</sub>, where the shift of regimes controlling O<sub>3</sub> formation like the case presented in  
35 this study can be expected **across the entire region**, due to the substantial reductions of NO<sub>x</sub>  
36 emissions in recent years.

37 **Keywords:** Ozone, local formation, regional transport, volatile organic compound, North China  
38 Plain

## 39 **1 Introduction**

40 Air pollution in the North China Plain (NCP), the largest alluvial plain of China consisting of  
41 Beijing, Tianjin and many cities in Hebei, Shandong, and Henan provinces, has attracted much  
42 attention in recent years. While the annual average concentration of PM<sub>2.5</sub> (particulate matters  
43 with aerodynamic diameter less than or equal to 2.5 μm) has been reduced under concerted  
44 efforts on emission restrictions (Zhang et al., 2015; Lang et al., 2017), the tropospheric ozone  
45 (O<sub>3</sub>) pollution, which is less visible than haze but may be equivalently harmful to human health,  
46 is still severe. At a regional receptor site of the NCP in a mountainous area **to the** north of  
47 Beijing, Wang et al. (2006) reported the maximum hourly O<sub>3</sub> of 286 ppbv. A year-round  
48 observation of O<sub>3</sub> at 10 urban sites in Beijing **indicated frequent O<sub>3</sub> non-attainments (hourly O<sub>3</sub> >**  
49 **100 ppbv)** through May to August of 2013 (Wang et al., 2015a). Hourly O<sub>3</sub> mixing ratio of up to  
50 120 ppbv **was** reported on Mt. Tai, the highest mountain in the NCP (1534 m a.s.l.) (Gao et al.,  
51 2005). **All these studies revealed** the significant photochemical O<sub>3</sub> pollution over the entire NCP.  
52 Moreover, O<sub>3</sub> has been increasing in the NCP during the last decades (Zhang et al., 2014; Zhang  
53 et al., 2015). The increase rate of O<sub>3</sub> at an urban site in Beijing from 2005 to 2011 was quantified  
54 as 2.6 ppbv/year (Zhang et al., 2014), comparable to that (1.7-2.1 ppbv/year) at Mt. Tai in the  
55 summer between 2003 and 2015 (Sun et al., 2016). Overall, the NCP suffers from severe O<sub>3</sub>  
56 pollution, which is even aggravating.

57 Apart from the intrusion of stratospheric O<sub>3</sub> in some places with high elevations (Cooper et al.,  
58 2005; Lin et al., 2015), photochemical formation is the main source of the ground-level O<sub>3</sub>.  
59 Volatile organic compounds (VOCs), carbon monoxide (CO) and nitrogen oxides (NO<sub>x</sub>) are key  
60 precursors of tropospheric O<sub>3</sub> (Crutzen, 1973; Chameides and Walker, 1973; Carter, 1994; Carter  
61 et al., 1995). The general chemical reactions R(1) - R(5) show the production of O<sub>3</sub> from the OH  
62 initiated oxidation of hydrocarbons (RH) (Jenkin et al., 1997; Atkinson, 2000; Jenkin and  
63 Clemitshaw, 2000).



69 The production of O<sub>3</sub> is generally limited by VOCs or NO<sub>x</sub> or co-limited by both VOCs and NO<sub>x</sub>,  
70 depending upon the chemical compositions of the air, particularly the ratio between OH  
71 reactivity of VOCs and NO<sub>x</sub> (OH reactivity is the sum of the products of O<sub>3</sub> precursors  
72 concentrations and the reaction rate constants between O<sub>3</sub> precursors and OH). Xue et al. (2014)  
73 indicated that the formation of O<sub>3</sub> was limited by NO<sub>x</sub> in Lanzhou in summer, consistent with the  
74 findings of Liu et al. (2010) who proved that NO<sub>x</sub>-limited regime dominated O<sub>3</sub> formation in  
75 most areas of northwestern China. In southwestern China, O<sub>3</sub> formation was diagnosed as VOC-  
76 limited in Chengdu, but NO<sub>x</sub>-limited in Pengzhou due to the large amount of emissions from  
77 petrochemical industry (Tan et al., 2018a). Lyu et al. (2016) reported the VOC-limited regime in  
78 Wuhan, central China. The VOC-limited regime has also been repeatedly confirmed for O<sub>3</sub>  
79 formation in Shanghai (Xue et al., 2014; Xing et al., 2017) and Nanjing (Ding et al., 2013),  
80 eastern China. In the Pearl River Delta of southern China, it was found that O<sub>3</sub> formation was  
81 generally limited by VOCs in the southwest, while limited by NO<sub>x</sub> in the northeast (Ye et al.,  
82 2016). In the NCP, both Han et al. (2018) and Xing et al. (2018) summarized that VOCs limited  
83 the production of O<sub>3</sub> in most urban areas. However, in the suburban and rural areas, O<sub>3</sub> formation  
84 was generally in the transitional regime, e.g. Yucheng (Zong et al., 2018), or limited by NO<sub>x</sub>, e.g.  
85 Wangdu (Tan et al., 2018b). From a historical perspective, Jin et al. (2017) pointed out that the  
86 sensitivity of O<sub>3</sub> formation to VOCs increased in most Chinese cities, however decreased in

87 some megacities (such as Beijing and Shanghai) due to the stringent control of NO<sub>x</sub> emissions in  
88 recent years. Different VOCs play non-equivalent roles in O<sub>3</sub> formation. Alkenes, aromatics and  
89 carbonyls can be readily oxidized by oxidative radicals (*e.g.* OH) or photolyzed (applicable for  
90 carbonyls), leading to O<sub>3</sub> formation (Cheng et al., 2010; Guo et al., 2013). Therefore, the sources  
91 with **large quantities of emissions of** these VOCs generally make considerable contributions to  
92 the photochemical production of ground-level O<sub>3</sub>. For example, Cheng et al. (2010) pointed out  
93 that carbonyls increased the peak O<sub>3</sub> production rates at a rural site and **at** a suburban site in  
94 South China by 64% and 47%, respectively. Solvent based industry and paint solvent usage with  
95 **intensive emissions of aromatics** were responsible for more than half of O<sub>3</sub> formation potential in  
96 Shanghai (Cai et al., 2010). Carbonyls and alkenes accounted for 71-85% of the total OH  
97 reactivity of VOCs in Beijing (Shao et al., 2009).

98 In addition to the chemical processes, **meteorological conditions also play significant roles in** the  
99 formation, transport and accumulation of O<sub>3</sub>. Studies (Chan and Chan, 2000; Huang et al., 2005)  
100 indicated that tropical cyclone (typhoon **as the mature form**) and continental anticyclone are the  
101 most common synoptic systems conducive to O<sub>3</sub> pollution in coastal cities of **South** China. Many  
102 O<sub>3</sub> episodes in **East** China occurred under the control of the west Pacific subtropical high  
103 pressure (He et al., 2012; Shu et al., 2016). In the NCP (**North China**), the summertime O<sub>3</sub>  
104 pollution is generally accompanied with weak high pressure **systems** (Wang et al., 2010).  
105 Furthermore, O<sub>3</sub> pollution **is also related to the topography**. For example, the mountains **to the**  
106 north and west of Beijing lead to upslope winds (valley breeze) in daytime, **which transport the**  
107 polluted air masses laden with O<sub>3</sub> **and/or O<sub>3</sub> precursors** from the NCP to Beijing (Lin et al.,  
108 2008). Overall, the causes of O<sub>3</sub> pollution are complicated and need to be analyzed case by case.

109 The NCP is the region with the largest emissions of **many** air pollutants, **such as VOCs and NO<sub>x</sub>,**  
110 in China (Gu et al., 2014; Li et al., 2017), partially accounting for the severe O<sub>3</sub> pollution there.  
111 In addition, O<sub>3</sub> pollution in the NCP is closely related to the synoptic systems and topographic  
112 features (Chen et al., 2009; Zhang et al., 2016). For example, the strong photochemical  
113 production of O<sub>3</sub> in urban plumes of Beijing was found by Wang et al. (2006), while the  
114 contribution of regional transport was revealed by the enhanced O<sub>3</sub> production at a rural site in  
115 the NCP under southerly winds (Lin et al., 2008). Through the review of synoptic systems in the  
116 NCP from 1980 to 2013, Zhang et al. (2016) concluded that the air quality was generally

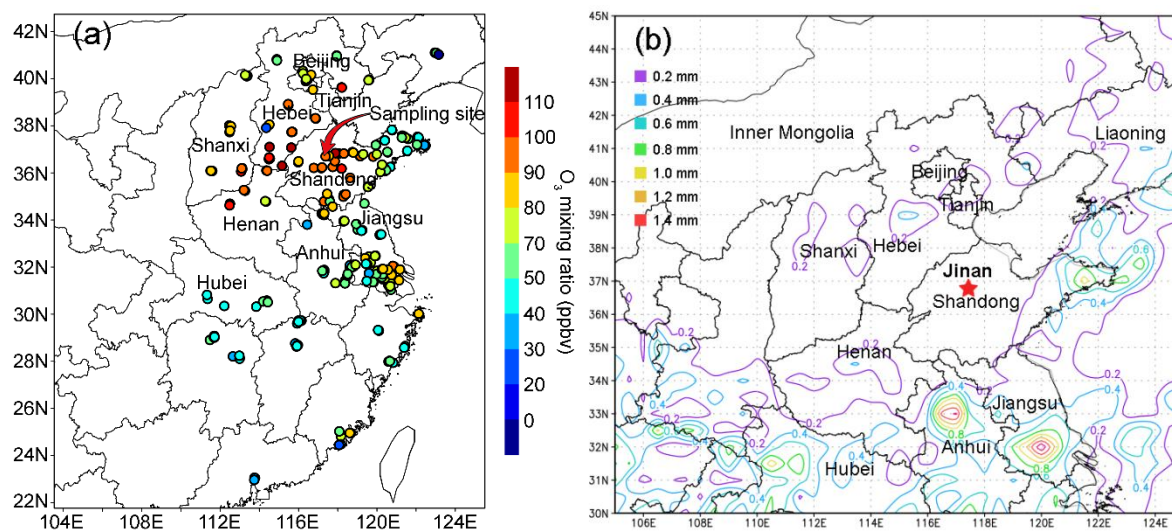
117 unhealthy under weak East Asian Monsoons. Moreover, a decadal statistical analysis indicated  
118 that meteorological factors explained ~50% of the O<sub>3</sub> variations in Beijing (Zhang et al., 2015).  
119 Despite many previous studies, the evolutions of the synoptic and photochemical processes in O<sub>3</sub>  
120 pollution events and their contributions to the non-attainment of O<sub>3</sub> have been seldom looked  
121 into in the NCP. Besides, the local and regional contributions to the elevated O<sub>3</sub> in the NCP are  
122 not unambiguously quantified, limited by the deficiencies in model representation of either  
123 physical or local chemical processes. The situation was even much worse for Ji'nan, the capital  
124 of Shandong province. As early as 2000s, studies (Shan et al., 2008; Yin et al., 2009) reported  
125 the maximum hourly O<sub>3</sub> of 143.8 ppbv and 147.8 ppbv in June 2004 and 2005, respectively.  
126 Even higher O<sub>3</sub> (198 ppbv) was observed at a rural site downwind of Ji'nan in June 2013 (Zong  
127 et al., 2018). However, almost no study was carried out to explore the mechanisms responsible  
128 for high O<sub>3</sub> there, though it has been confirmed that air pollution in the cities like Ji'nan **in the**  
129 **NCP** influenced air quality in Beijing (Lin et al., 2008; Wang et al., 2010). To better understand  
130 O<sub>3</sub> pollution in the NCP, this study investigated the causes of an O<sub>3</sub> episode **event** lasting for  
131 eight days in Ji'nan in summer 2017. The analyses presented here **focused** on the synoptic  
132 systems dominating Shandong Peninsula during this event; the chemical profiles of O<sub>3</sub> and O<sub>3</sub>  
133 precursors; and the simulation of factors contributing to O<sub>3</sub> in Ji'nan with the aid of a chemical  
134 transport model and a photochemical box model. In addition, we **proposed** feasible O<sub>3</sub> control  
135 measures based on the source-resolved OH reactivity of VOCs and NO<sub>x</sub>.

## 136 **2 Methodology**

### 137 **2.1 Site description**

138 The air quality monitoring and sample collection were carried out on the rooftop of a 7-story  
139 building on the campus of Shandong University from July 15 to August 14, 2017. The campus is  
140 located in the urban area of Ji'nan, and the site is about 50 m from a main road (Shanda South  
141 Road) outside the campus. Figure 1 shows the locations of the sampling site (36.68 °N, 117.07 °E,  
142 22 m a.g.l.) and **the** surrounding air quality monitoring stations (AQMSs) set up by China  
143 National Environmental Monitoring Center (CNEMC). Also shown are the observed O<sub>3</sub> and  
144 rainfall averaged over August 4-11, 2017 when **the** O<sub>3</sub> episode **event** occurred in Ji'nan. It is  
145 noteworthy that the days with maximum hourly O<sub>3</sub> exceeding 100 ppbv (Grade II of National  
146 Ambient Air Quality Standard) were defined as O<sub>3</sub> episode days. The hourly O<sub>3</sub> **values** at the

147 AQMSs were obtained from the website of CNEMC (<http://www.cnemc.cn/>). The high O<sub>3</sub> levels  
148 at almost all the AQMSs in the NCP (Figure 1 (a)) indicated a regional O<sub>3</sub> pollution event in this  
149 period. In view of the comparable O<sub>3</sub> mixing ratios observed at our sampling site to those at the  
150 surrounding AQMSs, we believe that the observations at our sampling site to some extent  
151 represented the characteristics of this regional pollution event. This was confirmed by the strong  
152 influences of regional transport on O<sub>3</sub> variations at the site, as discussed in section 3.3.



153  
154 Figure 1 (a) Locations of the sampling site and the CNEMC AQMSs, and the average observed  
155 O<sub>3</sub> at 14:00 LT on August 4-11, 2017 (colored circles). The sampling site is overlapped with the  
156 nearest AQMS in Ji'nan. (b) Rainfall distribution, in millimeters (mm), averaged over August 4-  
157 11, 2017.

## 158 2.2 Air quality monitoring and sample collection

### 159 2.2.1 Continuous monitoring of air pollutants and meteorological parameters

160 O<sub>3</sub>, NO and NO<sub>2</sub> were continuously monitored at the sampling site between July 15 and August  
161 14, 2017. The air was drawn through a 4 m Teflon tube by the built-in pumps of the trace gas  
162 analyzers at the total flow rate of 2 L/min (1.4 L/min for O<sub>3</sub> analyzer and 0.6 L/min for NO<sub>x</sub>  
163 analyzer). The inlet was located ~1 m above the rooftop of the building. O<sub>3</sub> and NO/NO<sub>x</sub> were  
164 detected with a UV photometric based analyzer and a chemiluminescence NO-NO<sub>2</sub>-NO<sub>x</sub>  
165 analyzer, respectively (see Table S1 for the specifications). The lowest NO observed during the  
166 sampling period was 2.4 ppbv, 6 times the detection limit (DL) of the NO<sub>x</sub> analyzer (0.4 ppbv).



167 Since the measurement accuracy of the analyzer was <15%, the DL was low enough to not  
168 influence the accurate measurements of NO in this study. NO<sub>2</sub> was calculated from the  
169 difference between NO and NO<sub>x</sub>. Studies indicated that NO<sub>2</sub> monitored with chemiluminescence  
170 was generally overestimated due to the conversion of the total odd nitrogen (NO<sub>y</sub>) to NO by  
171 molybdenum oxide catalysts (McClenny et al., 2002; Dunlea et al., 2007; Xu et al., 2013). The  
172 positive bias was more significant in more aged air masses, resulting from higher levels of NO<sub>z</sub>  
173 (NO<sub>z</sub> = NO<sub>y</sub> - NO<sub>x</sub>) (Dunlea et al., 2007). The average overestimation of NO<sub>2</sub> was 22% in  
174 Mexico City, which even increased to 50% in the afternoon (Dunlea et al., 2007). Xu et al. (2013)  
175 suggested that the chemiluminescence monitors overestimated NO<sub>2</sub> by less than 10% in urban  
176 areas with fresh emission of NO<sub>x</sub>, but the positive bias went up to 30-50% at the suburban sites.  
177 As described in section 2.1, our sampling site was located in the urban area of Ji'nan and was  
178 only ~50 m to a main road. Therefore, we infer that NO<sub>2</sub> might not be significantly  
179 overestimated in this study. However, larger overestimation could be expected during O<sub>3</sub>  
180 episodes, when the stronger photochemical reactions caused higher production of NO<sub>2</sub>.  
181 According to Xu et al. (2013), we adopted 30% (minimum bias in suburban area) and 10%  
182 (maximum bias in urban area) as the maximum fraction of NO<sub>2</sub> overestimation during episodes  
183 and non-episodes at this urban site, respectively. The influences of the NO<sub>2</sub> measurement  
184 interferences on the results were discussed where necessary.

185 The hourly concentrations of sulfur dioxide (SO<sub>2</sub>) and CO were acquired from a nearest AQMS,  
186 which is ~1 km from our sampling site. Year-round monitoring of inorganic trace gases was  
187 conducted at this AQMS. The air was drawn into the analyzers at a flow of 3 L/min through an  
188 inlet, ~1 m above the rooftop of a 5-story building (~ 16 m a.g.l.). The specifications of the  
189 analyzers deployed at the AQMS are also provided in Table S1. The hourly concentrations of O<sub>3</sub>  
190 and NO<sub>2</sub> measured at the AQMS (NO data was not available at the CNEMC website) agreed  
191 well with those observed at our sampling site, with the slope of 1.04 (R<sup>2</sup> = 0.82) and 1.13 (R<sup>2</sup> =  
192 0.71) for O<sub>3</sub> and NO<sub>2</sub> in the linear least square regressions, respectively (Figure S1). Due to the  
193 differences in analyzers and/or in sources and sinks of air pollutants between the two sites, the  
194 agreements were worse at low mixing ratios for both O<sub>3</sub> and NO<sub>2</sub>. Therefore, we only used SO<sub>2</sub>  
195 and CO monitored at the AQMS in this study, which had lower photochemical reactivity than O<sub>3</sub>  
196 and NO<sub>2</sub>, and were more homogeneous at a larger scale.



197 In addition, the meteorological parameters, including wind speed, wind direction, pressure,  
198 temperature and relative humidity, were monitored at the sampling site by a widely used weather  
199 station (China Huayun group, Model CAWS600-B). The daily total solar radiation was obtained  
200 from the observations at a meteorological station in Ji'nan (36.6 °N, 117.05 °E, 170.3 m a.s.l), 9  
201 km to our sampling site.

### 202 **2.2.2 Sample collection and chemical analysis**

203 The VOC and oxygenated VOC (OVOC) samples were collected on 9 selective days (*i.e.*, July  
204 20 and 30, August 1, 4-7 and 10-11), referred to as VOC sampling days hereafter. The days were  
205 selected to cover the periods with relatively high and **normal** levels of O<sub>3</sub>. The high O<sub>3</sub> days were  
206 forecasted prior to sampling based on the numerical simulations of meteorological conditions  
207 and air quality. In total, 6 out of 9 VOC sampling days were O<sub>3</sub> episode days with the maximum  
208 hourly O<sub>3</sub> **values** ranging from 100.4 to 154.1 ppbv. On each day (regardless of episode or non-  
209 episode), 6 VOC/OVOC samples were collected between 08:00 and 18:00 LT every 2 hours with  
210 the duration of 1 hour for VOC and 2 hours for OVOC samples. VOC samples were collected  
211 with 2 L stainless steel canisters which were cleaned and evacuated before sampling. A flow  
212 restrictor was connected to the inlet of the canister to guarantee 1 hour sampling. OVOC were  
213 sampled with the 2,4-dinitrophenylhydrazine (DNPH) cartridge, in front of which an O<sub>3</sub> scrubber  
214 was interfaced to remove O<sub>3</sub> in the air. A pump **was used to draw the air through the DNPH**  
215 **cartridge** at a flow of 500 mL/min. After sampling, all the DNPH cartridges were stored in a  
216 refrigerator at 4 °C **till** chemical analysis.

217 VOC samples were analyzed with a gas chromatograph-mass selective detector/flame ion  
218 detector/electron capture detector system (Colman et al., 2001). In total, 85 VOCs, including 59  
219 hydrocarbons, 19 halocarbons and 7 alkyl nitrates, were quantified. The overall ranges of the DL,  
220 accuracy and precision for VOCs analysis were 1-154 pptv, 1.2-19.8% and 0.1-17.9%,  
221 respectively. The analysis results given by this system have been compared with those analyzed  
222 by UCI and good agreements were achieved (Figure S2). OVOC samples were eluted with 5 mL  
223 acetonitrile, followed by analysis with the high performance liquid chromatography. The DL,  
224 accuracy and precision for the **detected OVOCs species** were within the range of 3-11 pptv, 0.32-  
225 0.98% and 0.01-1.03%, respectively.

### 226 **2.3 Model configuration**

### 227 2.3.1 Chemical transport model

228 To analyze the processes contributing to high O<sub>3</sub> in Ji'nan, a chemical transport model, **i.e.** the  
229 Weather Research Forecast-Community Multi-scale Air Quality (WRF-CMAQ), was utilized to  
230 simulate O<sub>3</sub> in this study. WRF v3.6.1 was run to provide the offline meteorological field for  
231 CMAQ v5.0.2. A two-nested domain was adopted with the resolution of 36 km (outer domain)  
232 and 12 km (inner domain), respectively. As shown in Figure S3, the outer domain covered the  
233 entire continental area of China aiming to provide sufficient boundary conditions for the inner  
234 domain, which specifically focused on eastern China.

235 We used the 2012-based Multi-resolution Emission Inventory for China (MEIC) to provide  
236 anthropogenic emissions of air pollutants, which was developed by Tsinghua University specific  
237 for China, with the grid resolution of 0.25 °×0.25 ° (Zhang et al., 2007; He, 2012). Five emission  
238 sectors, namely transportation, agriculture, power plant, industry and residence were included in  
239 MEIC. The emission inventory was linearly interpolated to the domains with consideration of the  
240 earth curvature effect. For grids outside China, the air pollutant emissions were derived from  
241 INTEX-B (Intercontinental Chemical Transport Experiment-Phase B) Asian emission inventory  
242 (Zhang et al., 2009). Consistent with many previous studies (Jiang et al., 2010; Wang et al.,  
243 2015b), the Model of Emissions of Gases and Aerosols from Nature (MEGAN) was used to  
244 calculate the biogenic emissions. The physical and chemical parameterizations for WRF-CMAQ  
245 were generally identical to those described in Wang et al. (2015b), with **the following**  
246 improvements. Firstly, the carbon bond v5 with updated toluene chemistry (CB05-TU) was  
247 chosen as the gas phase chemical mechanism (Whitten et al., 2010). Secondly, a single-layer  
248 urban canopy model (Kusaka and Kimura, 2004) was used to model the urban surface-  
249 atmosphere interactions. Thirdly, the default 1990s U.S. Geological Survey data in WRF was  
250 replaced by adopting the 2012-based moderate resolution imaging spectroradiometer (MODIS)  
251 land cover data for eastern China. The substitution was performed to update the simulation of  
252 boundary meteorological conditions (Wang et al., 2007).

253 An integrated process rate (IPR) module incorporated in CMAQ was used to analyze the  
254 processes influencing **the variations of** O<sub>3</sub>. Through solving the mass continuity equation  
255 established between the overall change of O<sub>3</sub> concentration **across** time and the change of O<sub>3</sub>  
256 concentration caused by individual processes, including horizontal diffusion (HDIF), horizontal

257 advection (HADV), vertical diffusion (VDIF), vertical advection (VADV), dry deposition  
258 (DDEP), net effect of chemistry (CHEM) and cloud processes (CLD), **the O<sub>3</sub> variation rates**  
259 **induced by individual processes were determined.** Note that **since** the estimate of CHEM is  
260 influenced by the estimate of O<sub>3</sub> precursor emissions, the simulation of meteorological  
261 conditions and the chemical mechanism, all the three aspects should be taken into account  
262 wherever CHEM is discussed. The IPR analysis has been widely applied in diagnosis of  
263 **processes influencing** O<sub>3</sub> pollution (Huang et al., 2005; Wang et al., 2015b). Since the field  
264 observations were conducted near the surface (~ 22 m a.g.l.), and the box model (**section 2.3.2**)  
265 was constrained by the observations, the modeling results on the ground-level layer were  
266 extracted from WRF-CMAQ for analyses in this study.

### 267 **2.3.2 Photochemical box model**

268 **We also utilized** a Photochemical Box Model incorporating the Master Chemical Mechanism  
269 (PBM-MCM) to study the in situ O<sub>3</sub> chemistry, **thanks to** the detailed (species-based)  
270 descriptions of VOC degradations in the MCM (Saunders et al., 2003; Lam et al., 2013). The  
271 PBM model was localized to be applicable in Ji'nan, with the settings of geographic coordinates,  
272 sunlight duration and photolysis rates. The photolysis rates were calculated by the TUV model  
273 (Madronich and Floke, 1997). Specifically, the geographical coordinates, date and time were  
274 input into the TUV model **to initialize** the calculation of solar radiation with the default aerosol  
275 optical depth (AOD), cloud optical depth (COD), surface albedo and other parameters. Then,  
276 COD was adjusted to make the calculated daily total solar radiation progressively **approach** the  
277 observed value. When the difference between the calculated and observed solar radiation **was**  
278 less than 1%, the input parameters with the adjusted COD were accepted. Based on the settings,  
279 the hourly solar radiations and the photolysis rates of O<sub>3</sub> (J(O<sup>1</sup>D)) and NO<sub>2</sub> (JNO<sub>2</sub>) were  
280 calculated by the TUV model, and applied to **the** PBM-MCM for O<sub>3</sub> chemistry modelling. Table  
281 S2 shows the daily maximum J(O<sup>1</sup>D) and JNO<sub>2</sub> on the VOC sampling days. The MCM v3.2  
282 (<http://mcm.leeds.ac.uk/MCM/>) consists of 17,242 reactions among 5,836 species. The **mixing**  
283 **ratios** of O<sub>3</sub> and its precursors at 00:00 on each day were used as the initial conditions for each  
284 day's modelling. The initial O<sub>3</sub> **therefore represented** O<sub>3</sub> left over from the days before the  
285 modelling day, **and partially** accounted for the primary OH production. Hourly concentrations of  
286 46 VOCs, 4 OVOCs and 4 trace gases (SO<sub>2</sub>, CO, NO and NO<sub>2</sub>), as well as hourly meteorological

287 parameters (temperature and relative humidity) were taken as inputs to constrain the model. O<sub>3</sub>,  
288 as the species to be modeled, was not input except for the setting of initial conditions. The Freon,  
289 cycloalkanes and methyl cycloalkanes with low O<sub>3</sub> formation potentials were not included in  
290 model inputs either. Also excluded were the species whose concentrations were lower than the  
291 DLs in more than 20% samples, such as the methyl hexane and methyl heptane isomers. For the  
292 hours when measurement data were not available, the concentrations were obtained with linear  
293 interpolation. Some secondary species, such as formaldehyde (HCHO), acetaldehyde and  
294 acetone, were input into the model to constrain the simulation. Since other secondary species,  
295 *e.g.*, PAN and HNO<sub>3</sub> were not observed in this study, their concentrations were calculated by the  
296 model. The model simulated dry depositions of all the chemicals, and the deposition velocities  
297 were set identical to those in Lam et al. (2013). Since NO and NO<sub>2</sub> were separately measured,  
298 they were not treated as a whole (*i.e.* NO<sub>x</sub>) in the model. Instead, both NO and NO<sub>2</sub> data were  
299 input into the model so that the partitioning between them was constrained to observations.

300 The simulations were separately performed on all the VOC sampling days. As spin-up, the model  
301 was run 72 hours prior to the simulation on the day of interest, with the same inputs. The model  
302 treated the air pollutants to be well-mixed within the boundary layer, while dilution and transport  
303 were not considered. O<sub>3</sub> in the free troposphere was not considered either, due to the lack of O<sub>3</sub>  
304 observations above the boundary layer over Ji'nan. This might hinder the accurate reproduction  
305 of the observed O<sub>3</sub>, particularly on the days when advection and diffusion were strong. Since the  
306 model mainly described the in situ photochemistry, it was validated through comparison with the  
307 CHEM process simulated by WRF-CMAQ. The simulated O<sub>3</sub> production rates were output every  
308 hour, which were integrated values over every 3600 s in one hour (model resolution: 1 s). More  
309 details about the model configuration can be found in Lam et al. (2003) and Lyu et al. (2017).

### 310 **3. Results and discussion**

#### 311 **3.1 Overall characteristics of O<sub>3</sub> pollution in Ji'nan**

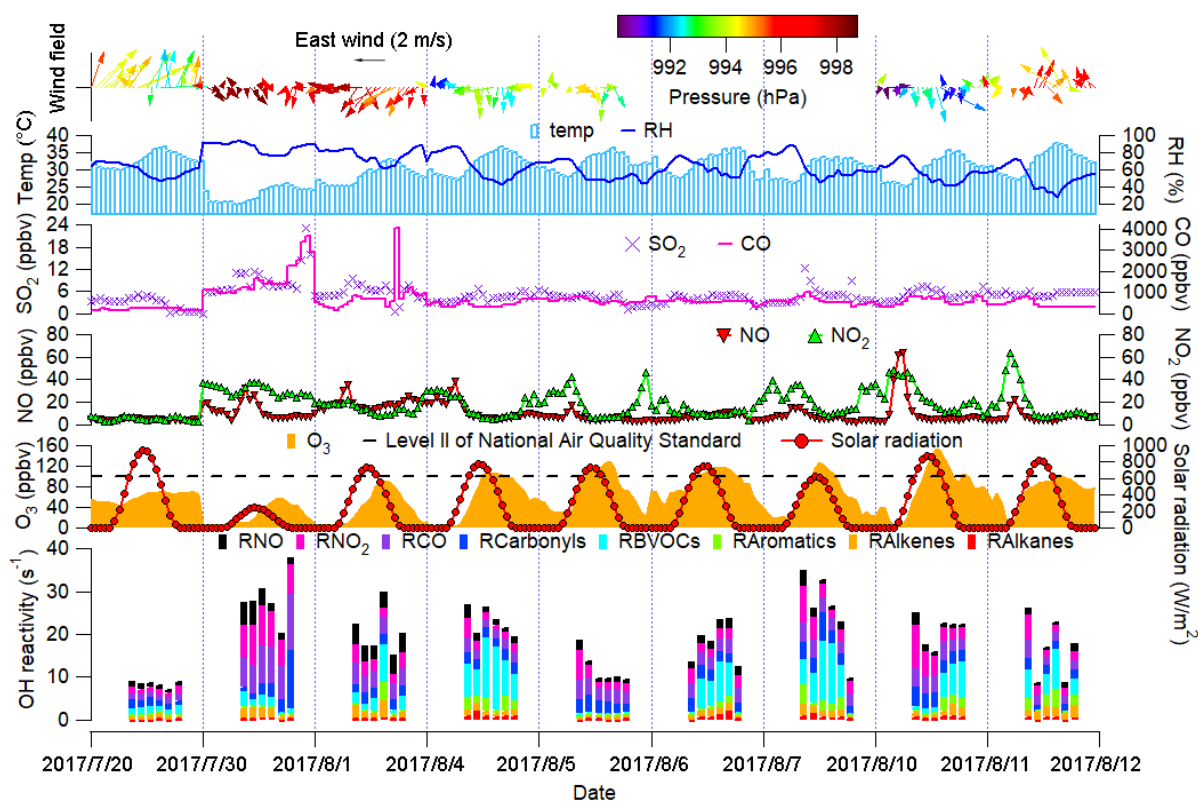
312 Figure 2 shows the time series of trace gases, OH reactivity of VOCs, CO and NO<sub>x</sub>, and  
313 meteorological conditions on the VOC sampling days in Ji'nan (Trace gases in the whole  
314 sampling period are shown in Figure S4). All the OH reactivity values discussed in this study  
315 were calculated rather than observed ones. The OH reactivity of VOCs was categorized into  
316 carbonyls, biogenic VOCs (BVOCs), aromatics, alkenes and alkanes (Table S3 lists the VOCs

317 included in each group). The reaction rate constants between O<sub>3</sub> precursors and OH in  
318 calculation of OH reactivity were adopted from the MCM v3.2. The average total OH reactivity  
319 on all the VOC sampling days (19.4±2.1 s<sup>-1</sup>) was comparable to that reported in New York (19±3  
320 s<sup>-1</sup>, Ren et al., 2003), Houston (9-22 s<sup>-1</sup>, Mao et al., 2010) and Beijing (15-27 s<sup>-1</sup>, Williams et al.,  
321 2016). Consistent with previous studies in urban areas (Ren et al., 2003; Yang et al., 2016 and  
322 references therein), NO<sub>x</sub> was the largest contributor (28.9±1.9%) to the total OH reactivity.  
323 Noticeably, 20.5±4.1% of the total OH reactivity were attributable to BVOCs, which were much  
324 higher than the contributions in urban areas (<10%) reviewed by Yang et al. (2016). The  
325 elevated isoprene levels (2.2±0.6 ppbv during episodes and 0.9±0.3 ppbv during non-episodes)  
326 under high temperature (mean: 31 °C) explained the considerable contribution of BVOCs to the  
327 total OH reactivity in this study.

328 The total OH reactivity of VOCs and CO ( $OH\ reactivity_{VOCs+CO}$ ) was comparable between O<sub>3</sub>  
329 episodes (14.8±2.0 s<sup>-1</sup>) and non-episodes (12.2±3.0 s<sup>-1</sup>), so was the OH reactivity of NO<sub>x</sub>  
330 (4.7±0.8 s<sup>-1</sup> and 6.9±1.9 s<sup>-1</sup> during episodes and non-episodes, respectively). Taking the positive  
331 biases of NO<sub>2</sub> measurement into account (section 2.2.1), we found that the OH reactivity of NO<sub>x</sub>  
332 was overestimated at maximum by 17.5±1.1% and 5.4±0.7% during O<sub>3</sub> episodes and non-  
333 episodes, respectively. In the case of maximum overestimation, the actual OH reactivity of NO<sub>x</sub>  
334 during episodes (4.0±0.7 s<sup>-1</sup>) might be lower ( $p<0.05$ ) than that during non-episodes (6.6±1.9 s<sup>-1</sup>).  
335 The high OH reactivity during non-episodes mainly occurred on July 30 and August 1, due to the  
336 unfavorable meteorological conditions, which were discussed later. Despite the comparable OH  
337 reactivity, we found that the ratio of  $\frac{OH\ reactivity_{VOCs+CO}}{OH\ reactivity_{NOx}}$  during O<sub>3</sub> episodes (3.7±0.7 s<sup>-1</sup>/s<sup>-1</sup>) was  
338 higher than during non-episodes (2.0±0.4 s<sup>-1</sup>/s<sup>-1</sup>) ( $p<0.05$ ). The difference was likely even larger,  
339 due to the more significant overestimation of NO<sub>2</sub> during episodes. This indicated that O<sub>3</sub>  
340 formation was more limited by VOCs during non-episodes than during episodes. Indeed, O<sub>3</sub>  
341 formation in Ji'nan switched from the VOC-limited regime during non-episodes to the  
342 transitional regime during episodes (see section 3.4.2). This partially explained the build-up of  
343 O<sub>3</sub> on episode days, because the transitional regime features the highest O<sub>3</sub> production rates.

344 From the aspect of meteorological conditions, O<sub>3</sub> episodes had relatively stronger solar radiation,  
345 higher temperature, lower relative humidity and weaker winds ( $p<0.05$ ). This is reasonable as O<sub>3</sub>  
346 formation and accumulation are generally enhanced under these weather conditions. In particular,

347 the solar radiation on July 30 was much weaker than those during O<sub>3</sub> episodes, **primarily**  
 348 **accounting for the** low O<sub>3</sub> on this day. Figure S5 shows the COD retrieved from the  
 349 terra/MODIS ([https://ladsweb.modaps.eosdis.nasa.gov/search/imageViewer/1/MOD06\\_L2--](https://ladsweb.modaps.eosdis.nasa.gov/search/imageViewer/1/MOD06_L2--61/2017-08-06/DB/Site:142/2873994172--3)  
 350 [61/2017-08-06/DB/Site:142/2873994172--3](https://ladsweb.modaps.eosdis.nasa.gov/search/imageViewer/1/MOD06_L2--61/2017-08-06/DB/Site:142/2873994172--3)) at 10:00 – 12:00 LT **on all** the VOC sampling days.  
 351 The terra/MODIS image revealed thick cloud cover with high COD over Ji'nan on July 30,  
 352 **which caused** the weak solar radiation. The influences of cloud cover/COD and solar radiation  
 353 on O<sub>3</sub> pollution **were further** discussed in section 3.2. Unlike **our previous understanding** that O<sub>3</sub>  
 354 pollution **is** aggravated **under** high pressure (Chan and Chan, 2000; Zhao et al., 2009), the sea-  
 355 level pressure during O<sub>3</sub> episodes ( $993.4 \pm 0.2$  hPa) was significantly lower than during non-  
 356 episodes ( $996.1 \pm 0.4$  hPa) in this study ( $p < 0.05$ ). When O<sub>3</sub> reached its hourly maximum on  
 357 August 10 (154.1 ppbv), the pressure was at its lowest value (990.2 hPa). **This discrepancy**  
 358 **inspired us to look into the synoptic and chemical processes in this continuous O<sub>3</sub> pollution event.**



359  
 360 Figure 2 Time series of trace gases, OH reactivity of O<sub>3</sub> precursors and meteorological  
 361 parameters. Wind speed and wind direction were not monitored from 17:00 LT on August 5 to  
 362 23:00 LT on August 7 due to malfunction of the weather station. RX in the bottom panel is the  
 363 OH reactivity of species/group X.



### 3.2 Synoptic processes and relationship with O<sub>3</sub> pollution

Figure 3 displays the average weather charts at 14:00 LT during O<sub>3</sub> episodes and non-episodes (weather charts on individual VOC sampling days are shown in Figure S6). Clearly, the temperature in Shandong province was much higher during O<sub>3</sub> episodes than non-episodes, which favored O<sub>3</sub> formation on episode days. Additionally, southerly and southwesterly winds originating from the inland areas (Hubei, Henan, and Anhui provinces) prevailed in central and western Shandong during O<sub>3</sub> episodes. In contrast, the winds were generally from the sea or coastal regions in Jiangsu province during non-episodes. It is more likely that O<sub>3</sub> and O<sub>3</sub> precursors were transported to Ji'nan during episodes. The high concentrations of O<sub>3</sub> precursors on July 30 and August 1 (non-episode days) were mainly caused by the weather conditions (high pressure, low temperature and low solar radiation), as discussed in section 3.1. Further, we also noticed that the winds changed direction from the southwest to the northwest around Ji'nan during O<sub>3</sub> episodes. This meant that there might be a local circulation, hampering the dispersion of air pollutants during episodes. It seemed that the change of wind direction was caused by the convergence of continental air and sea breeze from Bohai Bay, similar to the convergence zone formed over the Pearl River Estuary in South China (Fung et al., 2005; Lo et al., 2006). Overall, the surface winds were more favorable for regional transport and accumulation of air pollutants during O<sub>3</sub> episodes. In addition, Shandong province was under the control of a uniform pressure system with the sea-level pressure of 1000-1001 hPa during O<sub>3</sub> episodes, implying the relatively stagnant weather.

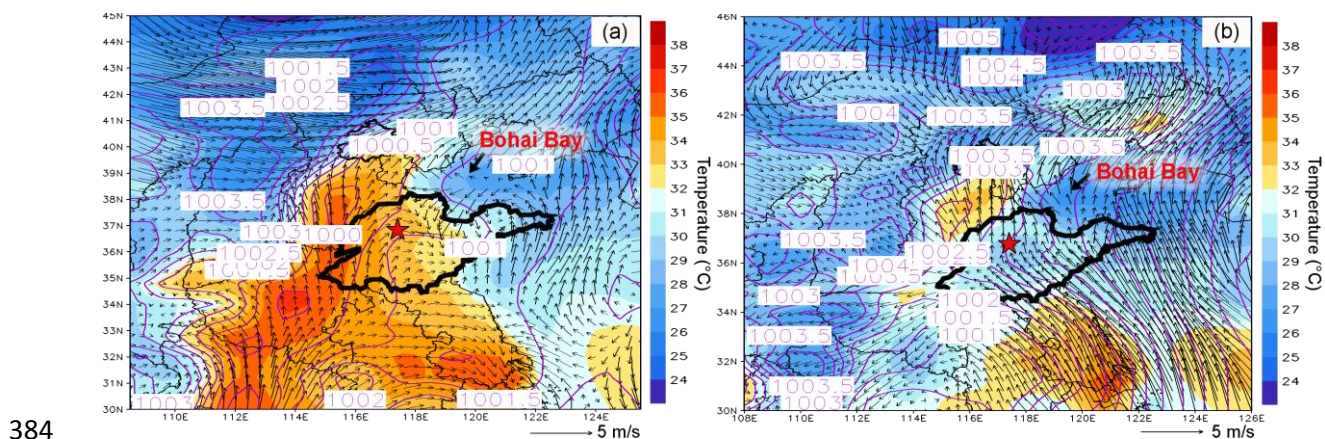


Figure 3 Weather chart at 14:00 LT averaged over (a) O<sub>3</sub> episodes and (b) non-episodes. The red star represents Ji'nan. The dark black line is the boundary of Shandong province. Bohai Bay is



387 located to the northeast of Shandong province. Numbers in the figure are sea-level pressures in  
388 unit of hPa.

389 To better understand the relationship between O<sub>3</sub> pollution and the synoptic systems, Table 1  
390 summarizes the synoptic systems, weather conditions and air mass origins on all the VOC  
391 sampling days. The weather charts at surface level and 500 hPa on August 1, 4, 7, 10 and 13 are  
392 presented in Figures S7-S8, showing the evolution of the synoptic systems. To identify the  
393 origins of air masses, the backward trajectories of air masses are shown in Figure 4. The  
394 trajectories were computed using the Hybrid Single Particle Lagrangian Integrated Trajectory  
395 (HYSPPLIT) Model v 4.9. Each trajectory was calculated for 48 hours and the calculation was  
396 done every 6 hours (4 trajectories each day). Our sampling site (36.68 °N, 117.07 °E) was set as  
397 the **end point of the** trajectories with the height of 500 m a.s.l. The discrepancy between the wind  
398 direction and origin of air masses, *e.g.* on August 1 and 11, was likely due to the air recirculation  
399 at the ground level.

400 It was found that Ji'nan was under the control of the Western Pacific Subtropical High (WPSH)  
401 on July 20 (weather chart on 500 hPa is not shown here), and the air masses arriving in Ji'nan  
402 originated from South China (Figure 4). As anticipated, the WPSH caused high **temperature** and  
403 the intensive solar radiation (Figure 2), which was conducive to O<sub>3</sub> formation. However, the  
404 winds on July 20 were the strongest in the entire VOC sampling period, with the highest hourly  
405 wind speed of 3.9 m/s. The strong winds facilitated the dispersion of O<sub>3</sub> **and its** precursors,  
406 **leading to low O<sub>3</sub> levels on this day**. The WPSH moved southward on the following days and  
407 Ji'nan was **controlled by** a uniform pressure **system**, which was formed in the peripheries of two  
408 low pressure systems (two rain belts as shown in Figure 1), *i.e.* one over Central China and  
409 another over North China (Figure S7). Thus, the pressure in Ji'nan was relatively high  
410 (997.1 ± 0.3 hPa), compared to the south and north regions. This synoptic system lasted for  
411 several days until August 7, covering 2 non-episode days and 4 O<sub>3</sub> episode days. The **low O<sub>3</sub> on**  
412 **2 non-episode days** (July 30 and August 1) were mainly attributable to the weak solar radiation  
413 and low temperature as discussed above.

414 In contrast, continuously strong solar **radiations** with low COD (Figure 2 and Figure S5), high  
415 temperature and continental air masses (Figure 4) were observed on August 4-7. This, in addition  
416 to the shift of O<sub>3</sub> formation mechanism (see sections 3.1 and 3.4.2), explained the prolonged O<sub>3</sub>

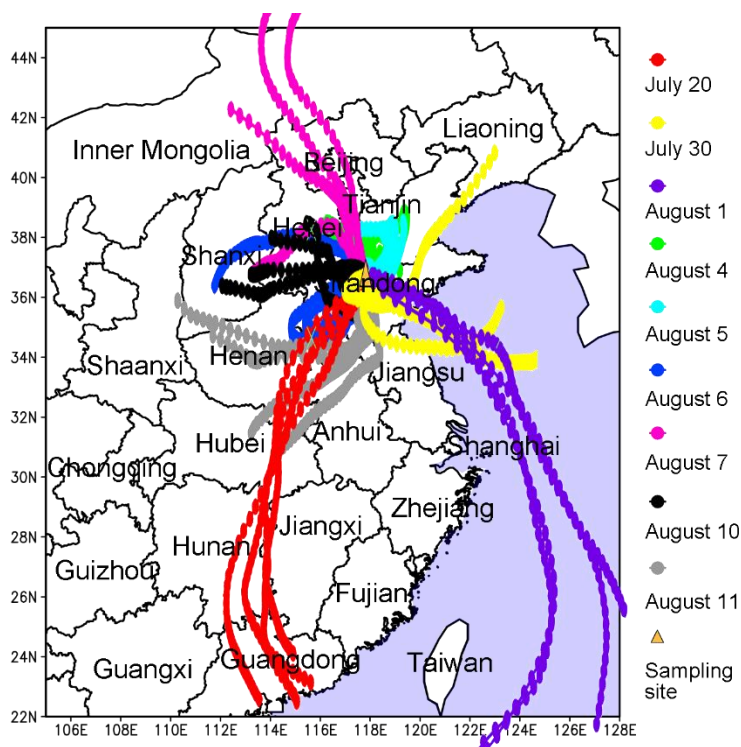
417 pollution event. On August 10, the rain belt over North China moved southward, forming a deep  
 418 low pressure trough over the NCP and Ji'nan was behind the trough (Figure S8 (d)). The low  
 419 pressure trough is a typical synoptic system conducive to O<sub>3</sub> pollution, resulting from the  
 420 intrusion of O<sub>3</sub> in the stratosphere and/or the upper troposphere (Chan and Chan, 2000).  
 421 Moreover, there was nearly no cloud cover over the entire NCP on August 10 (Figure S5).  
 422 Consequently, the highest O<sub>3</sub> (154.1 ppbv) in this sampling campaign was observed. On August  
 423 11, the low pressure system continued to extend to the Yellow Sea. O<sub>3</sub> decreased substantially on  
 424 this day with the disappearance of the low pressure trough and the weakening of solar radiation,  
 425 though the hourly maximum O<sub>3</sub> still reached 100.4 ppbv. On the following days, the  
 426 precipitations relieved the O<sub>3</sub> pollution in Ji'nan.

427 Table 1 Summary of the synoptic systems, weather conditions and air mass origins on VOC  
 428 sampling days.

Date	Maximum hourly O <sub>3</sub> (ppbv)	Episode/non-episode	Synoptic system Weather condition	Air mass origin
July 20	71.0	Non-episode	WPSH, strong southwesterly winds	Continental air masses from South China
July 30	57.6		Uniform pressure field (weak high pressure), rain, fog, calm winds	Marine air masses
August 1	90.6		Uniform pressure field (weak high pressure), northeasterly winds	
August 4	107.5	Episode	Uniform pressure field (weak high pressure), northeasterly winds	Continental air masses from Shandong province
August 5	128.2		Uniform pressure field (weak high pressure), calm winds	
August 6	116.9		Uniform pressure field (weak high pressure), southwesterly winds	
August 7	126.9		Uniform pressure field (weak high pressure), calm winds	Continental air masses from the north
August 10	154.1		Low-pressure trough, calm winds	Continental air masses from the west

August 11	100.4	Subtropical high, southeasterly winds	Continental air masses from the southwest
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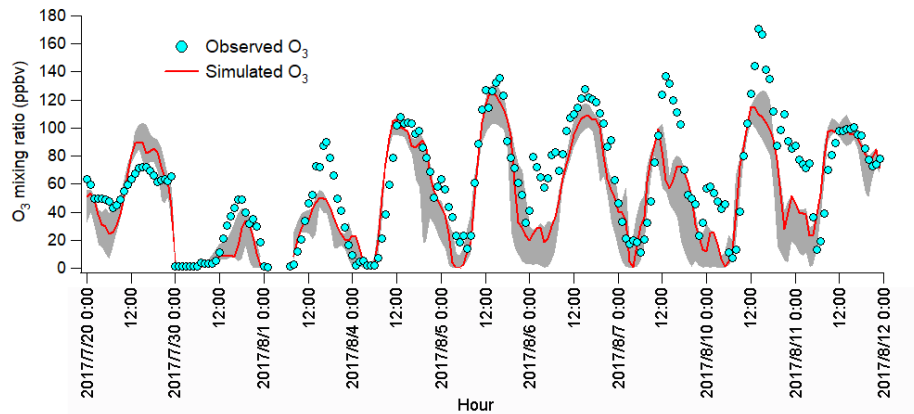


430

431 Figure 4 Forty eight hour backward trajectories calculated every 6 hours, with Ji'nan (36.68 N,  
 432 117.07 E, 500 m a.g.l.) as the ending point. The trajectories are simulated by HYSPLIT v4.9.  
 433 The water areas are highlighted in blue.

### 434 3.3 O<sub>3</sub> simulation and process analysis

435 The observations indicated the likely different regimes controlling local O<sub>3</sub> formation and the  
 436 potential impacts of regional transport. To understand the atmospheric chemistry and dynamics,  
 437 as well as their roles in this O<sub>3</sub> pollution event, the WRF-CMAQ was applied. Figure 5 shows  
 438 the hourly average simulated and observed O<sub>3</sub> on the VOC sampling days in Ji'nan. Overall, the  
 439 model well reproduced the magnitudes and diurnal patterns of the observed O<sub>3</sub>, except for the  
 440 higher simulated O<sub>3</sub> on July 20 and the under-prediction of O<sub>3</sub> on August 1, 7 and 10.  
 441 Discussions on the discrepancies and the model validation were provided in Text S1, Figures S9-  
 442 S11 and Table S4.



443

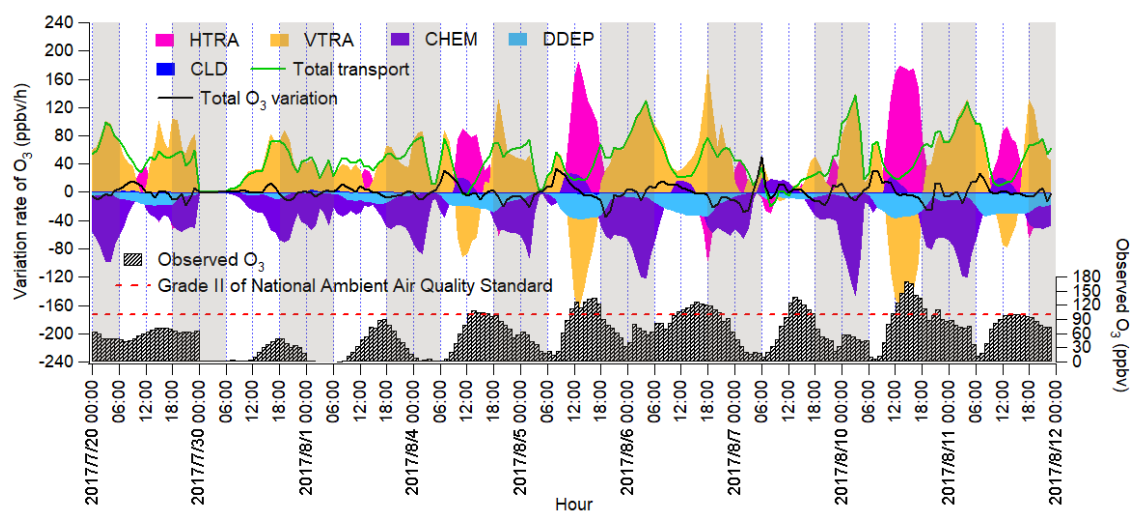
444 Figure 5 Hourly average mixing ratios of the WRF-CMAQ simulated and observed O<sub>3</sub> in Ji'nan.  
 445 The grey area shows the minimum and maximum simulated O<sub>3</sub> at the sampling site and 8  
 446 adjoining grids (12×12 km<sup>2</sup> for each grid).

447 The IPR analysis quantified the O<sub>3</sub> variation rates induced by different processes, as shown in  
 448 Figure 6. HDIF and HADV were summed as horizontal transport (HTRA), and the vertical  
 449 transport (VTRA) was a total representative of VDIF and VADV. It was found that chemical  
 450 reactions generally led to the decrease of O<sub>3</sub> mixing ratio during non-episodes. The negative  
 451 contributions of chemical reactions on July 20 coincided with the very low concentrations of O<sub>3</sub>  
 452 precursors and the flat diurnal cycle of O<sub>3</sub> (Figure 2). The chemical destruction to O<sub>3</sub> on July 30  
 453 and August 1 was most likely related to the weak solar radiation and low temperature, which  
 454 inhibited the photochemical reactions. In fact, the negative chemical effect should be considered  
 455 as the titration of NO to the regionally-transported and/or background O<sub>3</sub> and the depletion of O<sub>3</sub>  
 456 by the freshly emitted NO near the sources (Beck and Grennfelt, 1994; Sillman, 1999).  
 457 Conversely, the combined effect of horizontal and vertical transport was to increase O<sub>3</sub> levels  
 458 during non-episodes.

459 During O<sub>3</sub> episodes, chemical reactions made positive contributions to O<sub>3</sub> production rates  
 460 between 09:00 LT and 15:00 LT, with the average hourly O<sub>3</sub> production rate of 14.0±2.3 ppbv/hr.  
 461 At the same time, O<sub>3</sub> was also elevated by transport at an average rate of 18.7±4.0 ppbv/hr, as a  
 462 combined effect of vertical transport (-40.8±20.2 ppbv/hr) and horizontal transport (59.5±19.8  
 463 ppbv/hr). The negative contribution of vertical transport to O<sub>3</sub> in these hours might be caused by  
 464 the updraft with the increase of temperature in the city. The positive contributions of horizontal  
 465 transport could be explained by the air masses laden with O<sub>3</sub> originating from the west and the

466 **north** (Figure 4 and Figure S10). The much higher O<sub>3</sub> over the NCP than in the surrounding  
467 regions indicated that the NCP was an O<sub>3</sub> source in this case. In fact, the transport of O<sub>3</sub> from the  
468 lower troposphere over the NCP to the free troposphere and further to northeast China was also  
469 presented by Ding et al. (2009).

470 During 16:00-08:00 LT on O<sub>3</sub> episode days, O<sub>3</sub> was titrated and chemically consumed at the rate  
471 of 49.4±6.3 ppbv/hr. This was reasonable, **because the fresh vehicular emissions in the morning**  
472 **and evening rush hours consumed O<sub>3</sub>, particularly the irreversible titration of NO to O<sub>3</sub> in**  
473 **absence of sunlight.** The NO<sub>2</sub> **produced from the titration reaction** was carried over to the other  
474 places by air circulation, and/or oxidized to NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>, which could further react with  
475 aerosol to form HNO<sub>3</sub> and ClNO<sub>2</sub> in the evening. Horizontal and vertical transport dominated O<sub>3</sub>  
476 sources, with the average positive contribution of 5.7±7.0 and 54.5±9.6 ppbv/hr during 16:00-  
477 08:00 LT on August 4-11, respectively. The strong vertical transport coincided with the  
478 downward winds in the evening, which brought the high-altitude O<sub>3</sub> to the ground, as indicated  
479 in Figure S9. However, the sources of O<sub>3</sub> in the upper atmosphere were beyond the scope of this  
480 study.



481  
482 Figure 6 Time series of O<sub>3</sub> variation rate in Ji'nan induced by individual processes calculated  
483 based on the change of O<sub>3</sub> per hour. Total transport is the sum of HTRA and VTRA, and the sum  
484 of O<sub>3</sub> variation rates attributable to all the processes is represented by total O<sub>3</sub> variation rate. The  
485 nighttime (18:00 – 06:00 LT) has been highlighted in grey.

### 486 3.4 Local O<sub>3</sub> formation and control

### 487 3.4.1 Pathway and source contributions to O<sub>3</sub> production

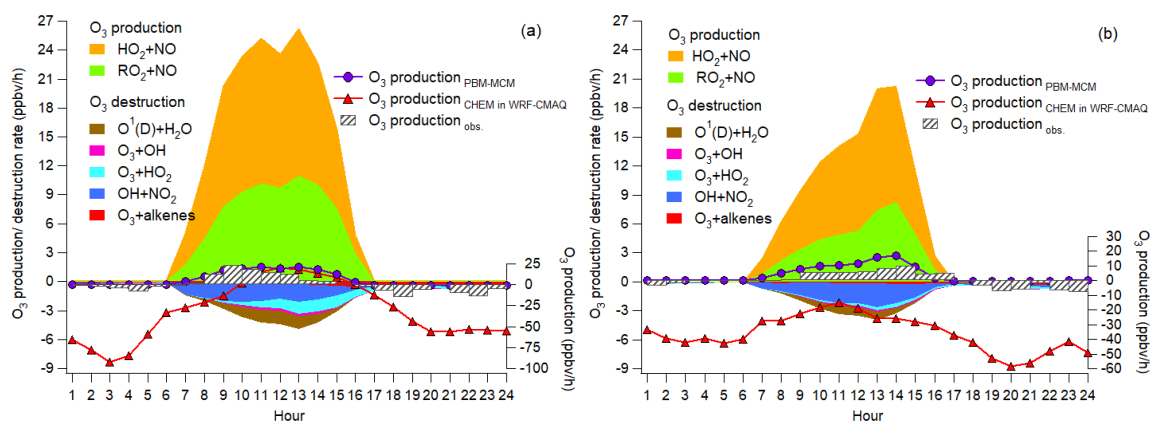
488 The IPR analyses showed that chemical reactions served as an important source of O<sub>3</sub> on episode  
489 days in Ji'nan, particularly during 09:00-15:00 LT when O<sub>3</sub> was at high levels. This process was  
490 further studied through the simulation of the in-situ photochemistry by PBM-MCM. It should be  
491 noted that the simulations were based on the observed concentrations of O<sub>3</sub> precursors, which  
492 could be influenced by both local and regional air. It required cautions to extend the results to all  
493 the situations in Ji'nan, because the regional effect was not always consistent. Table S5 lists the  
494 production and destruction pathways of O<sub>3</sub> (Thornton et al., 2002; Monks, 2005; Kanaya et al.,  
495 2009). Briefly, the oxidation of NO by HO<sub>2</sub> and RO<sub>2</sub> produced NO<sub>2</sub>, which led to O<sub>3</sub> formation  
496 following NO<sub>2</sub> photolysis (R2 and R4-R5 in introduction). Therefore, the reactions between NO  
497 and HO<sub>2</sub>/RO<sub>2</sub> were considered as the production pathways of O<sub>3</sub>. To account for O<sub>3</sub> destruction,  
498 reaction between O<sup>1</sup>(D) and H<sub>2</sub>O denoted the photolysis of O<sub>3</sub>, and reactions of O<sub>3</sub> with OH,  
499 HO<sub>2</sub> and alkenes were also included. Furthermore, since HNO<sub>3</sub> was an important sink of NO<sub>2</sub>,  
500 the reaction between OH and NO<sub>2</sub> was treated to be destructive to O<sub>3</sub>. The titration of O<sub>3</sub> by NO  
501 was not included in O<sub>3</sub> destruction, because NO<sub>2</sub> produced in this reaction was either not  
502 considered as a source of O<sub>3</sub>.

503 Figure 7 (a) and (b) show the average diurnal cycles of the simulated contributions to O<sub>3</sub>  
504 production rates of different pathways. Also shown are the net O<sub>3</sub> production rates simulated by  
505 PBM-MCM (*O<sub>3</sub> production*<sub>PBM-MCM</sub>), those simulated by WRF-CMAQ (*O<sub>3</sub> production*<sub>CHEM</sub>),  
506 and those calculated from the observed hourly O<sub>3</sub> (*O<sub>3</sub> production*<sub>obs.</sub>). Overall, *O<sub>3</sub> production*<sub>PBM-</sub>  
507 *MCM* and *O<sub>3</sub> production*<sub>obs.</sub> were on the same magnitudes, especially during O<sub>3</sub> episodes with more  
508 stagnant weather conditions. This indicated that the PBM-MCM model reasonably reproduced  
509 the in situ O<sub>3</sub> photochemistry. Though obvious discrepancies existed between *O<sub>3</sub> production*<sub>CHEM</sub>  
510 and *O<sub>3</sub> production*<sub>PBM-MCM</sub>, they agreed well with each other during 10:00-15:00 LT on episode  
511 days, consistent with the finding that chemical reactions made great contributions to O<sub>3</sub> in **these**  
512 **hours** (Figure 6). The lower or even negative *O<sub>3</sub> production*<sub>CHEM</sub> resulted from the titration of the  
513 regionally transported and/or local background O<sub>3</sub> by NO and the following depletion of NO<sub>2</sub>  
514 through reacting with OH and/or **dispersion**. Differently, PBM-MCM did not consider the  
515 transport of O<sub>3</sub>, though the transport effect was partially represented by constraining the model to  
516 the observed concentrations of O<sub>3</sub> precursors. In addition, the PBM-MCM was constructed by



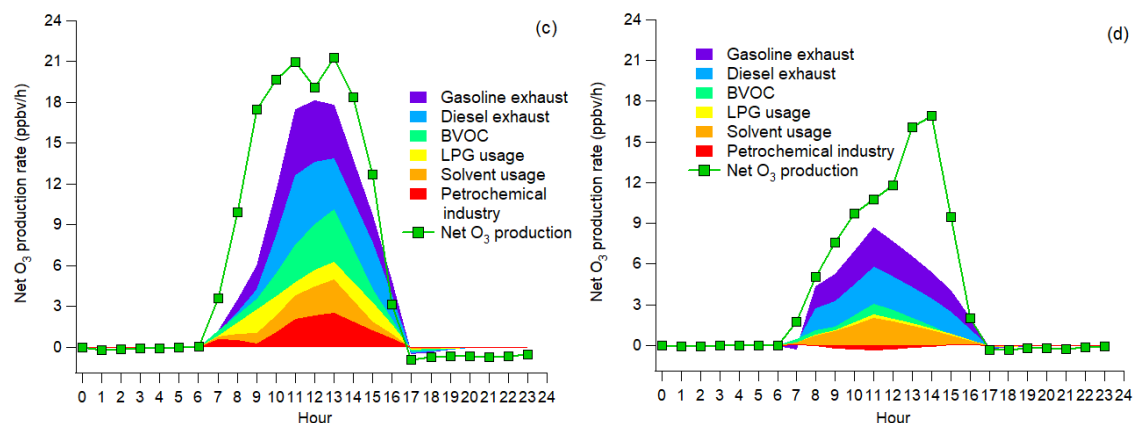
517 the observed air pollutants, which were already subject to chemical reactions before being  
 518 detected by the analytical instruments. This meant that the reaction between NO and O<sub>3</sub> from the  
 519 emission to the detection of NO<sub>x</sub> was not considered in PBM-MCM. However, as an emission-  
 520 based model, WRF-CMAQ performed better in describing the reactions immediately after the  
 521 emissions of air pollutants. Therefore, the chemical destructions of O<sub>3</sub> in the vicinity of NO<sub>x</sub>  
 522 sources also accounted for the aforementioned discrepancy. The obviously higher reaction **rates**  
 523 between NO and O<sub>3</sub> simulated by WRF-CMAQ (Figure S12) confirmed our inferences.

524 During both O<sub>3</sub> episodes and non-episodes, the reaction between HO<sub>2</sub> and NO dominated over  
 525 “RO<sub>2</sub>+NO” in O<sub>3</sub> production, while the O<sub>3</sub> destruction was mainly attributable to the formation  
 526 of HNO<sub>3</sub>, the reaction between O<sub>3</sub> and HO<sub>2</sub> and photolysis of O<sub>3</sub>. The net O<sub>3</sub> production rate  
 527 during O<sub>3</sub> episodes (maximum: 21.3 ppbv/hr) was much (*p*<0.05) higher than during non-  
 528 episodes (maximum: 16.9 ppbv/hr), which partially explained the higher O<sub>3</sub> on episode days. In  
 529 general, “OH+NO<sub>2</sub>” serves as the chain terminating reaction in VOC-limited regime of O<sub>3</sub>  
 530 formation, while the radical-radical reactions take over the role in NO<sub>x</sub>-limited regime  
 531 (Finlayson-Pitts and Pitts, 1993; Kleinman, 2005). Here, we found that the ratio of total reaction  
 532 rates between “HO<sub>2</sub>+RO<sub>2</sub>” and “OH+NO<sub>2</sub>” substantially increased from 0.2±0.1 during non-  
 533 episodes to 1.0±0.3 during O<sub>3</sub> episodes (*p*<0.05). This suggested that O<sub>3</sub> formation during non-  
 534 episodes was limited by VOCs, while it switched to be co-limited by VOCs and NO<sub>x</sub> during O<sub>3</sub>  
 535 episodes in view of the equivalent role of “HO<sub>2</sub>+RO<sub>2</sub>” and “OH+NO<sub>2</sub>” in terminating the chain  
 536 reactions.



537





538

539 Figure 7 Pathway contributions to O<sub>3</sub> production and destruction rate during episodes (a) and  
 540 non-episodes (b). Contributions of O<sub>3</sub> precursor sources to net O<sub>3</sub> production rate during  
 541 episodes (c) and non-episodes (d).

542 Further, the contributions to the net O<sub>3</sub> production rates of different sources of O<sub>3</sub> precursors  
 543 were identified, as presented in Figure 7 (c) and (d). Text S2 and Figure S13 illustrate the source  
 544 apportionment of O<sub>3</sub> precursors and the simulations of the source-specific contributions to O<sub>3</sub>  
 545 production rates. The results are presented in Table 2. Since the source apportionment was  
 546 performed for the ambient O<sub>3</sub> precursors which were already subject to atmospheric processes,  
 547 such as dispersion, deposition and chemical reactions, the results represented the source  
 548 contributions to the steady – state concentrations of O<sub>3</sub> precursors and the corresponding O<sub>3</sub>  
 549 production rates. It was found that gasoline exhaust and diesel exhaust were the largest  
 550 contributors to O<sub>3</sub> production rates regardless of O<sub>3</sub> episodes or non-episodes. Further, the net O<sub>3</sub>  
 551 production rates attributable to gasoline exhaust (diesel exhaust) increased from 1.0±0.3 ppbv/hr  
 552 (1.0±0.3 ppbv/hr) during non-episodes to 1.8±0.6 ppbv/hr (1.7±0.4 ppbv/hr) during O<sub>3</sub> episodes.  
 553 This suggested that vehicular emissions played critical roles in building up ground-level O<sub>3</sub> in  
 554 the O<sub>3</sub> pollution event. If carbonyls were taken into account, the contributions of vehicular  
 555 emissions to O<sub>3</sub> production rates were even higher than the currently simulated values, due to the  
 556 abundances of carbonyls in vehicle exhausts (Grosjean et al., 1990; Granby et al., 1997). In  
 557 addition, the contributions of the other sources to O<sub>3</sub> production rates all increased during O<sub>3</sub>  
 558 episodes except for solvent usage (*p*>0.05), as listed in Table 2. It is not surprising to see the  
 559 synchronous increases, because of the stronger solar radiation and higher temperature during  
 560 episodes.

561 Further insight into the percentage contributions (not shown here) found that the contributions of  
 562 BVOC, LPG usage and petrochemical industry to O<sub>3</sub> production rates increased substantially  
 563 from 9.9±4.2%, 4.3±1.4% and -2.8±1.9% during non-episodes to 19.2±4.3%, 9.1±3.4% and  
 564 12.1±3.1% during O<sub>3</sub> episodes, respectively. The increased O<sub>3</sub> production rates by BVOCs could  
 565 be explained by the increase of isoprene (episodes: 2.2±0.6 ppbv; non-episodes: 0.9±0.3 ppbv),  
 566 under higher temperature and stronger solar radiation during O<sub>3</sub> episodes. The enhancement of  
 567 O<sub>3</sub> production rates driven by petrochemical industry on episode days was likely associated with  
 568 the dominance of continental air (Figure 4) and the extensive petrochemical industries in the  
 569 NCP. For example, the mixing ratio of styrene increased from 54.7±22.0 pptv during non-  
 570 episodes to 162.3±44.7 pptv during O<sub>3</sub> episodes. The reason for elevated O<sub>3</sub> production rates  
 571 resulting from LPG usage during episodes was unknown. It is worth to note that the source  
 572 contributions to O<sub>3</sub> production rates might have some uncertainties due to the limited number of  
 573 samples (54 samples) and O<sub>3</sub> precursors (31 VOCs, CO, NO and NO<sub>2</sub>) applied for source  
 574 apportionment.

575 Table 2 Contributions to VOCs, CO, NO, NO<sub>2</sub> and O<sub>3</sub> production rate by the sources of O<sub>3</sub>  
 576 precursors averaged on the VOC sampling days in Ji'nan (Unit: % unless otherwise specified).

Source	VOCs*	CO	NO	NO <sub>2</sub>	O <sub>3</sub> production rate (ppbv/hr)	
					O <sub>3</sub> episodes	Non-episodes
GE <sup>1</sup>	25.7±3.6	29.9±2.1	30.9±2.4	22.2±2.4	1.8±0.6	1.0±0.3
DE <sup>2</sup>	17.6±2.4	57.3±5.2	52.0±5.8	54.4±5.8	1.7±0.4	1.0±0.3
BVOC	6.1±2.6	0.0±1.7	0.0±2.8	0.0±2.3	1.2±0.5	0.2±0.1
LPG <sup>3</sup>	14.7±2.0	2.2±1.1	9.1±1.6	4.7±0.9	0.8±0.5	0.1±0.1
Solvent <sup>4</sup>	17.1±3.9	3.1±1.8	5.1±3.8	7.8±3.1	0.8±0.5	0.7±0.3
PI <sup>5</sup>	18.8±3.1	7.4±1.9	2.9±1.8	10.9±2.5	1.0±0.3	-0.1±0.1

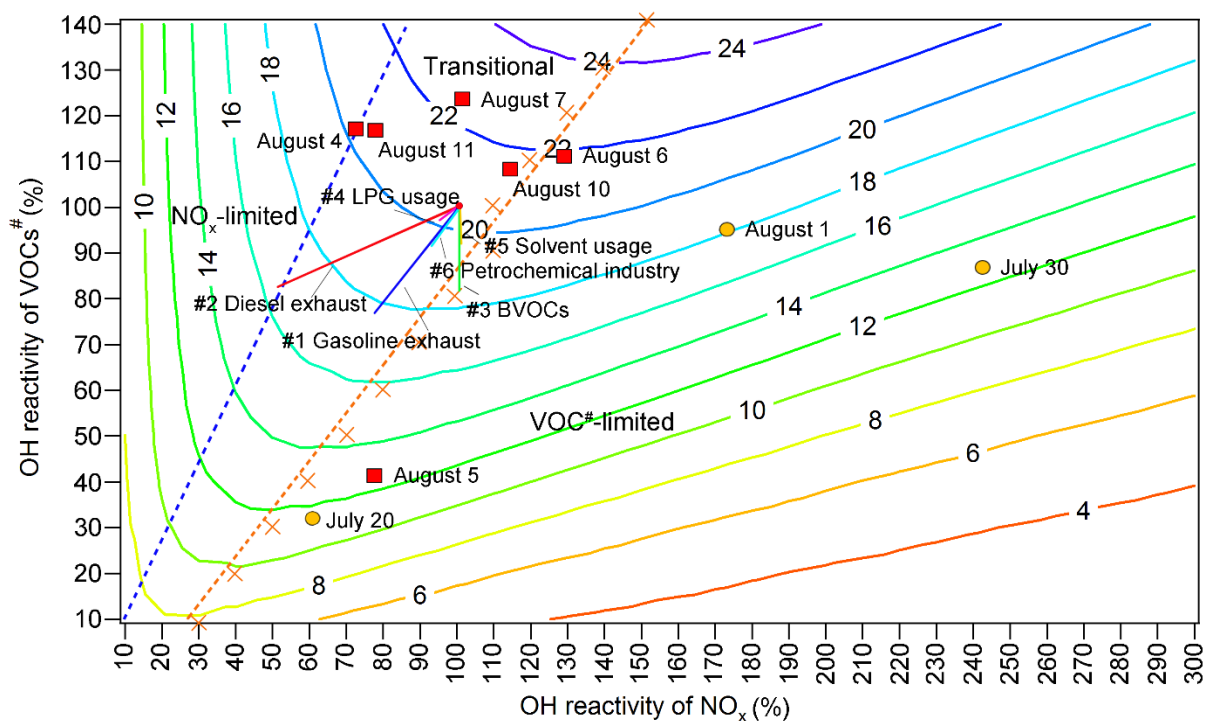
577 VOCs\*: VOCs applied in source apportionment (see Text S2).

578 <sup>1</sup> gasoline exhaust, <sup>2</sup> diesel exhaust, <sup>3</sup> LPG usage, <sup>4</sup> solvent usage and <sup>5</sup> petrochemical industry.

### 579 3.4.2 O<sub>3</sub> control measures

580 Both WRF-CMAQ and PBM-MCM revealed the significant local O<sub>3</sub> formation in the O<sub>3</sub>  
 581 pollution event. The relationships between O<sub>3</sub> and its precursors needed to be clarified, so that  
 582 the science-based control measures could be taken. Throughout the VOC sampling period, the

583 OH reactivity values of VOCs ( $OH\ reactivity_{VOCs}$ ) were within the range of 33-123% of the  
 584 average  $OH\ reactivity_{VOCs}$  during O<sub>3</sub> episodes. For OH reactivity of NO<sub>x</sub> ( $OH\ reactivity_{NOx}$ ),  
 585 the range was 61-242%. The O<sub>3</sub> production rates were simulated in a set of assumed scenarios  
 586 with different  $OH\ reactivity_{VOCs}$  and  $OH\ reactivity_{NOx}$  values. To include the OH reactivity  
 587 of VOCs and NO<sub>x</sub> on all the VOC sampling days, factors from 10% to 140% with the step of 10%  
 588 were applied to the average diurnal profiles of VOCs and CO during O<sub>3</sub> episodes, while the  
 589 factors ranged from 10% to 300% with the step of 10% for NO<sub>x</sub>. The initial concentrations of all  
 590 the air pollutants were also scaled by the factors and the model was constrained to these scaled  
 591 concentrations every hour, except for O<sub>3</sub>. It should be noted that the factors applied to CO were  
 592 exactly the same as those applied to VOCs, therefore we use VOCs<sup>#</sup> to represent the sum of  
 593 VOCs and CO hereafter. The 14 gradients of  $OH\ reactivity_{VOCs}^{\#}$  values and 30 gradients of  
 594  $OH\ reactivity_{NOx}$  values made up 420 scenarios. Meteorological conditions were exactly the  
 595 same for all the scenarios and the clear sky was hypothesized. According to the simulations, the  
 596 maximum O<sub>3</sub> production rates occurred at 12:00 LT. Thus, the simulated O<sub>3</sub> production rates at  
 597 12:00 LT, as a function of percentages of  $OH\ reactivity_{VOCs}$  and  $OH\ reactivity_{NOx}$ , are  
 598 plotted in Figure 8.



599

600 Figure 8 Isopleths of the net O<sub>3</sub> production rate (ppbv/hr) at 12:00 LT as a function of  
601 *OH reactivity*<sub>VOCs#</sub> and *OH reactivity*<sub>NO<sub>x</sub></sub>. The red blocks and orange circles denote the  
602 calculated *OH reactivity*<sub>VOCs#</sub> and *OH reactivity*<sub>NO<sub>x</sub></sub> values at 12:00 LT on O<sub>3</sub> episode and  
603 non-episode days, respectively. Each orange cross represents the *OH reactivity*<sub>VOCs#</sub> and  
604 *OH reactivity*<sub>NO<sub>x</sub></sub> at 12:00 LT in the scenario with highest O<sub>3</sub> production rate at a given  
605 *OH reactivity*<sub>VOCs#</sub>. The orange dashed line and blue dashed line divide O<sub>3</sub> formation into the  
606 VOC-limited regime, transitional regime, and NO<sub>x</sub>-limited regime.

607 Text S3 describes the methods to define the regimes of O<sub>3</sub> formation. Overall, O<sub>3</sub> formation was  
608 mainly limited by VOCs<sup>#</sup> during non-episodes. However, it switched to be co-limited by VOCs<sup>#</sup>  
609 and NO<sub>x</sub> (transitional regime) on episode days with the net O<sub>3</sub> production rates among the  
610 highest, except for August 5 when the strong sea breeze diluted air pollutants in Ji'nan and/or  
611 intercepted the transport of air pollutants from Central China to Ji'nan (Figure S6). In fact, the  
612 sensitivity of O<sub>3</sub> formation to NO<sub>x</sub> might be underemphasized due to the positive biases of NO<sub>2</sub>  
613 measurement (Lu et al., 2010). This effect was expected to be more significant during episodes  
614 when the overestimation of NO<sub>2</sub> was more obvious. However, O<sub>3</sub> formation was not likely only  
615 limited by NO<sub>x</sub> even during O<sub>3</sub> episodes, because NO<sub>2</sub> could not be overestimated by more than  
616 30% according to our inferences (see section 2.2.1). Therefore, O<sub>3</sub> formation was treated to be in  
617 the transitional regime during episodes. This partially explained the increased O<sub>3</sub> during episodes  
618 in Ji'nan, given the higher O<sub>3</sub> production rates in transitional regime (Figure 8). Noticeably, the  
619 change of regimes controlling O<sub>3</sub> formation is consistent with that predicted by the  
620  $\frac{OH\ reactivity_{VOCs\#}}{OH\ reactivity_{NOx}}$  ratio and the ratio of the reaction rates between “HO<sub>2</sub>+RO<sub>2</sub>” and “OH+NO<sub>2</sub>”.

621 The source apportionment of O<sub>3</sub> precursors enabled us to calculate the source-specific  
622 *OH reactivity*<sub>VOCs#</sub> and *OH reactivity*<sub>NO<sub>x</sub></sub> values. Accordingly, the variations of O<sub>3</sub>  
623 production rates induced by the reductions in source emissions are presented in Figure 8 (straight  
624 solid lines #1-#6). The start point of the straight lines corresponded to 100% of the total average  
625 *OH reactivity*<sub>VOCs#</sub> and *OH reactivity*<sub>NO<sub>x</sub></sub> during O<sub>3</sub> episodes. The end points, however,  
626 represented the *OH reactivity*<sub>VOCs#</sub> and *OH reactivity*<sub>NO<sub>x</sub></sub> with the complete removal of  
627 emissions from the individual sources. Therefore, the differences of the O<sub>3</sub> production rates  
628 between the start point and end points were the source contributions to the O<sub>3</sub> production rates,

629 while the lengths of the lines reflected the contributions to the OH reactivity of the sources.  
630 Further, the simulated O<sub>3</sub> production rates on the lines #1-#6, as a response of reductions in  
631 source emissions, are extracted and plotted in Figure S14. Obviously, the highest efficiencies of  
632 O<sub>3</sub> reduction could be achieved by cutting diesel exhaust (0.58 ppbv hr<sup>-1</sup>/10% emission reduction)  
633 and gasoline exhaust (0.47 ppbv hr<sup>-1</sup>/10% emission reduction). In fact, the sensitivities of O<sub>3</sub>  
634 production rates to the vehicle exhausts might be somewhat underestimated, due to the exclusion  
635 of carbonyls in the source apportionment. However, the reductions of O<sub>3</sub> production rates by  
636 cutting 10% of vehicle exhausts were still insignificant, compared to the overall maximum O<sub>3</sub>  
637 production rate of 21.3 ppbv/hr during O<sub>3</sub> episodes. This indicated that by only restraining  
638 emissions from one to two sources, high percentages of emission reductions were required to  
639 sufficiently reduce the overall O<sub>3</sub> production rate. Otherwise, the combined efforts should be  
640 made to control the emissions of O<sub>3</sub> precursors from the diverse sources. In particular, it is  
641 essential to get rid of the transitional regime featuring high O<sub>3</sub> production rates.

#### 642 **4 Implications**

643 This study investigates the causes of a severe O<sub>3</sub> pollution event lasting for eight consecutive  
644 days in the NCP, one of the most densely populated regions in the world. Photochemical  
645 formation in the lower troposphere of the NCP is demonstrated as the main source of O<sub>3</sub>, under  
646 the control of weak high pressure or low pressure trough. Though the emissions of NO<sub>x</sub>, an  
647 important precursor of O<sub>3</sub>, have been significantly reduced in China since 2013 (Duncan et al.,  
648 2016; Liu et al., 2017), O<sub>3</sub> pollution is still severe or even worsening in the NCP, as revealed in  
649 the present and also previous studies (Zhang et al., 2014; Sun et al., 2016). The finding that O<sub>3</sub>  
650 formation shifted from VOC-limited regime on relatively low O<sub>3</sub> days to the transitional regime  
651 on O<sub>3</sub> non-attainment days may elucidate the increase of O<sub>3</sub>, because O<sub>3</sub> production rates in the  
652 transitional regime are the highest. It is unrealistic to expect the continuously linear reduction in  
653 NO<sub>x</sub> emissions in the NCP, after the substantial decreases of emissions from power plants and  
654 industries in recent years. In other words, restraining on VOC emissions is urgent for O<sub>3</sub>  
655 abatement in the NCP. Another important finding in this study is that the NCP served as an O<sub>3</sub>  
656 source. This was ever proposed by Ding et al. (2009), based on the aircraft measurement and  
657 simulation of atmospheric dynamics. We confirm it through the ground-level observation and the  
658 simulation of in-situ photochemistry. It can be expected that organic nitrates are also intensively

659 formed in the NCP as byproducts of O<sub>3</sub> formation. In view of the fact that the NCP locates  
660 within the mid-latitude band of Northern Hemisphere under the dominance of westerlies, we  
661 believe that O<sub>3</sub> and organic nitrates formed in this region may be transported over a long distance  
662 following the uplifting of air masses, which has been confirmed to partially account for the  
663 enhancement of background O<sub>3</sub> in North America and even Europe (Derwent et al., 2015; Lin et  
664 al., 2017). Therefore, the recent air pollution control measures taken in China (including China's  
665 Clean Air Act Plan in force in 2013) are still inadequate to ease the burden of global tropospheric  
666 O<sub>3</sub> in a short period. More effective Action Plans should be implemented for O<sub>3</sub> benefit, with  
667 comprehensive thinking of atmospheric dynamics and chemistry.

#### 668 **Author contribution**

669 The study was designed by the corresponding author, Hai Guo, with the help of Lihui Han and  
670 Ying Zhou. Fei Jiang and Hairong Cheng provided valuable input on the characteristics of  
671 meteorological conditions and emissions of air pollutants in the NCP. Xiaopu Lyu collected the  
672 samples and managed the logistics in the field campaign, when Likun Xue provided sufficient  
673 assistances in field sampling campaign. The solar radiation data in Ji'nan were obtained from  
674 Zhe Cai. Xiaopu Lyu did data analyses and wrote the paper, while the chemical transport  
675 modelling was done by Nan Wang and Yangzong Zeren. Hai Guo revised and finalized the paper  
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686 [https://drive.google.com/open?id=1\\_KeOxOuVsLY83xL74RtcRORsiiyIR\\_8FZ](https://drive.google.com/open?id=1_KeOxOuVsLY83xL74RtcRORsiiyIR_8FZ).

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## Supporting Information for “Causes of a continuous summertime O<sub>3</sub> pollution event in Ji’nan, a central city in the North China Plain”

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### Text S1 Validation of the WRF-CMAQ modeling

Due to **the** inevitable uncertainties inherent in chemical transport models, such as the uncertainties in emission inventory, meteorological simulation and chemical mechanisms (Hu et al., 2016), it is difficult to deduce the reasons **of** the discrepancies between the **simulated and observed O<sub>3</sub> mixing ratios**. However, the **observations indicated** extremely high levels of some VOCs on the three days when O<sub>3</sub> **values were** under-predicted. Specifically, 13.5 ppbv of ethene was observed at 14:00 LT on August 1. On August 7, 5.6 ppbv of isoprene, 16.2 ppbv of HCHO and 2.3 ppbv of hexanal were measured during 12:00 – 14:00 LT. On August 10, 22.7 ppbv of propene and 12.7 ppbv of *i*-butane were recorded at 08:00 and 16:00 LT, respectively. It is noteworthy that these mixing ratios were 5 – 10 times higher than the **average mixing ratios of these VOCs**. Further, most of these VOCs are highly reactive in O<sub>3</sub> photochemistry and may make great contributions to local O<sub>3</sub> production. With the setting of constant emissions of O<sub>3</sub> precursors, WRF-CMAQ did not reproduce these extremely high levels of VOCs, which was a plausible reason for the under-prediction of O<sub>3</sub> on August 1, 7 and 10. Ji’nan was behind a low pressure trough on August 9 – 10. However, vertical transport was simulated to make negative contributions to O<sub>3</sub> between 10:00 LT and 18:00 LT on August 10, according to the process

analysis. In addition, the simulated O<sub>3</sub> in the upper atmosphere on August 10 was relatively low compared to that on August 5, 6 and 9 (Figure S9). Namely, the model might fail to reproduce the O<sub>3</sub> enhancement driven by the low pressure trough on August 10.

The process analysis indicated that horizontal and vertical transport dominated the sources of O<sub>3</sub> at noon (10:00-12:00 LT) and the other **time periods on** July 20, respectively. While vertical transport explained the high O<sub>3</sub> at night (Figure 2), it was not likely that horizontal transport built up O<sub>3</sub> at noon, because the southwesterly **airflows were** originated from South China and passed central China (Figure 4), where O<sub>3</sub> values were relatively low on that day (high O<sub>3</sub> occurred in Hebei province **to** the northwest **of Ji'nan**). Therefore, the overestimate of the transport effect led to the higher simulated O<sub>3</sub> on July 20.

Despite these discrepancies, overall the observed O<sub>3</sub> at the sampling site was well reproduced. In addition, the spatial distribution of the simulated O<sub>3</sub> was highly consistent with the observed O<sub>3</sub> distribution, as shown in Figure S10. The average concentrations of the simulated VOCs were also compared with the observations (Figure S11). While the day-to-day and diurnal variations of the observed VOCs were not well reproduced (not shown), which is a proverbial drawback of the WRF-CMAQ, the model reasonably simulated the magnitudes of VOCs. Moreover, the averages of the observations (Avg. Obs.) and simulations (Avg. Sim.), difference between Avg. Obs. and Avg. Sim. (Diff.), root mean square error (RMSE), normalized mean bias (NMB), normalized mean error (NME) and index of agreement (IOA) were calculated to reflect the agreements between the simulated and observed temperature (Temp.), relative humidity (R.H.), wind speed (W.S.), pressure (Press.), NO<sub>2</sub> and O<sub>3</sub>, as listed in Table S4. Generally, the lower Diff., RMSE, NMB and NME, but higher IOA indicate better agreement between the simulated and observed values (Willmott et al., 1985). The validation of the simulations of air pollutants was carried out at 8 AQMSs of CNEMC in and around Ji'nan, and at the sampling site, while the meteorological parameters monitored at 6 airports in eastern and northern China and at the sampling site were used to validate the simulated meteorological conditions. The **metrics** calculated in this study were well within the ranges of those reported in previous studies involving WRF-CMAQ simulations (Table S4) (Jiang et al., 2010; Wang et al., 2015), suggesting good performance of the model in reproducing the meteorological conditions and air pollutants. Thus, the simulated results were accepted for further analyses.

**Text S2** Source apportionment of O<sub>3</sub> precursors

The positive matrix factorization (PMF) model was employed to identify the sources of O<sub>3</sub> precursors. Details about the operation principles of PMF can be found in Paatero and Tapper (1994). Briefly, the model treats the matrix of input concentrations as the product of two matrixes (*i.e.*, factor contribution and factor profile). Here, hourly concentrations of 31 VOCs, CO, NO and NO<sub>2</sub> in 54 samples were input into the model. The VOCs **applied** for source apportionment (termed as VOCs\* hereafter) **were either** tracers of specific sources (*e.g.*, isoprene for biogenic emissions), **or having** high concentrations (detectable in at least 80% samples). On average, VOCs\* accounted for 79.5±11.7% of the total quantified VOCs. The uncertainties of the input concentrations of O<sub>3</sub> precursors were set as  $\frac{5}{6} \times DL$  and  $\sqrt{(10\% \times concentration)^2 + (0.5 \times DL)^2}$  for the concentrations lower than and higher than **DLs**, respectively.

The model was run for 20 times with a random seed and the best resolution automatically given by the model was accepted. A total of 6 sources of O<sub>3</sub> precursors were resolved by PMF in this study. The number of sources was chosen based on the criteria that the tracers indicating different sources were not allocated in the same source, and all the sources were interpretable according to the tracers. The Bootstrap method integrated in PMF was used to estimate the uncertainties of the modelling results.

Figure S13 shows the profiles of the six sources of O<sub>3</sub> precursors extracted from PMF. The first source contained high levels of *n/i*-pentanes and aromatics, likely representing gasoline exhaust (Ho et al., 2009; Ling and Guo, 2014). The heavy hydrocarbons (C<sub>8</sub>-C<sub>10</sub>) dominated the second source, **with** great abundances of the combustion tracers, such as C<sub>2</sub>-C<sub>3</sub> hydrocarbons, CO, NO and NO<sub>2</sub>. **These are** in line with the features of diesel exhaust (Liu et al., 2008). The third source was assigned as BVOC, due to the exclusively high loading of isoprene (Guenther, 2006). The fourth source was rich in C<sub>4</sub> hydrocarbons, including *n/i*-butanes and 1,3-butadiene. It was defined as liquefied petroleum gas (LPG) usage, since butanes and butenes are present in large quantities in China's LPG (Song et al., 2008 and references therein). Solvent usage was represented by the fifth source, in view of the high loadings of hexane isomers (2,3-dimethylbutane, 2-methylpentane and 3-methylpentane) and moderate loadings of *n*-hexane, toluene, ethylbenzene and xylenes (Guo et al., 2011). At last, most of styrene, benzene, toluene, ethylbenzene and xylenes **were** allocated to the sixth source, which also contained moderate levels of light (C<sub>2</sub>-C<sub>5</sub>) hydrocarbons. Since styrene is a common

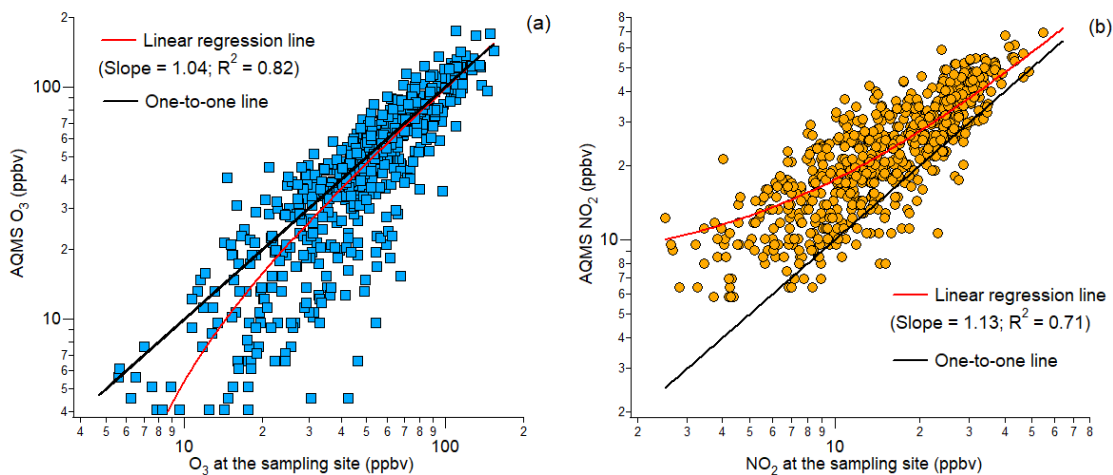
petrochemical product (Jobson et al., 2004; Liu et al., 2008), this source was designated as petrochemical industry.

The source contributions to the O<sub>3</sub> production rates were obtained from the differences in simulated O<sub>3</sub> production rates between a base run and a constrained run. In the base run, the O<sub>3</sub> production rate was simulated with the observed concentrations of air pollutants except for the carbonyls, while the concentrations of air pollutants attributable to a specific source were deducted from the observed concentrations for the input of the constrained run. To account for the influence of primary hydrocarbons on the formation of carbonyls, and the subsequent impact on O<sub>3</sub> production, carbonyls were not constrained to observations in either the base run or the constrained runs. However, the source-specific primary emissions of carbonyls and their contributions to O<sub>3</sub> production were not considered in this approach. Therefore, the source-specific contributions to net O<sub>3</sub> production rates were expected to be underestimated, as carbonyls are generally of high O<sub>3</sub> formation potentials (Cheng et al., 2010; Dong et al., 2014). The method was applied to each of the six sources, derived from the PMF analysis, thereby acquiring the contribution to O<sub>3</sub> production rates of each source.

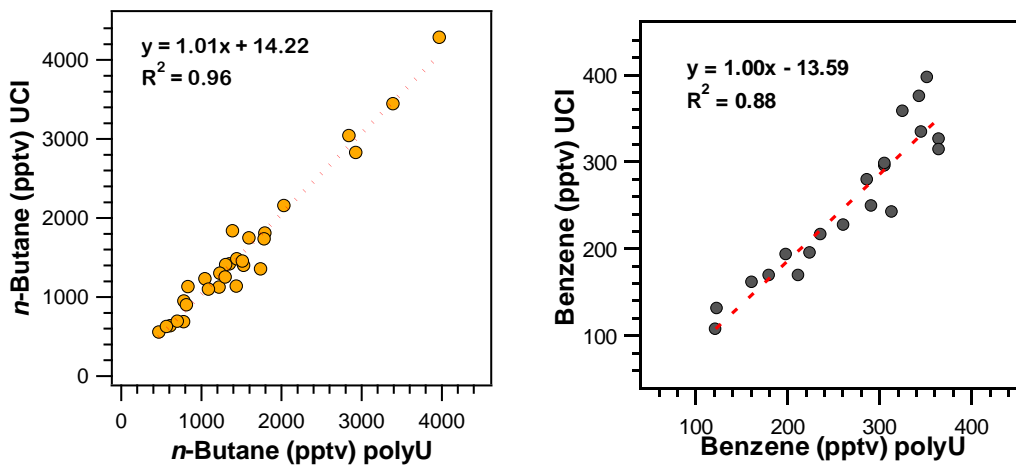
### **Text S3** Definitions of the O<sub>3</sub> formation regimes

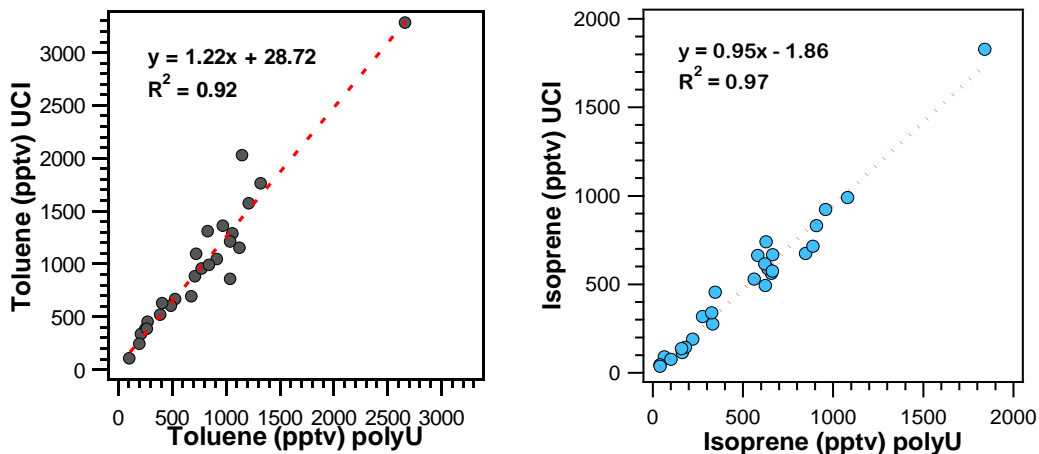
As shown in Figure 8, O<sub>3</sub> formation was divided into VOC<sup>#</sup>-limited regime and NO<sub>x</sub>-limited regime with the method used in Lyu et al. (2017). Briefly, at a given  $OH\text{ reactivity}_{VOCs^{\#}}$  value, the simulated O<sub>3</sub> production rate generally got the maximum at a specific  $OH\text{ reactivity}_{NOx}$  value due to the dual role of NO<sub>x</sub> in O<sub>3</sub> formation. This specific  $OH\text{ reactivity}_{NOx}$  value was treated as the dividing point between NO<sub>x</sub>-limited regime and VOC<sup>#</sup>-limited regime at the given  $OH\text{ reactivity}_{VOCs^{\#}}$  value. Since the scenarios involved 14 gradients of  $OH\text{ reactivity}_{VOCs^{\#}}$  (10% - 140% with the step of 10%), 14 pairs of  $OH\text{ reactivity}_{NOx}$  and  $OH\text{ reactivity}_{VOCs^{\#}}$  were obtained, as shown in the orange crosses in Figure 8. A dividing line was acquired from the linear regression between  $OH\text{ reactivity}_{VOCs^{\#}}$  and  $OH\text{ reactivity}_{NOx}$  in these scenarios (presented in orange dashed line in Figure 8). O<sub>3</sub> formation was limited by VOCs<sup>#</sup> and NO<sub>x</sub> in the lower right and upper left areas of the dividing line, respectively. Since the horizontal and vertical coordinates were the percentages relative to the average  $OH\text{ reactivity}_{VOCs^{\#}}$  and  $OH\text{ reactivity}_{NOx}$  during O<sub>3</sub> episodes, rather than the actual values of OH reactivity, we did not calculate the dividing ratio of  $\frac{OH\text{ reactivity}_{VOCs^{\#}}}{OH\text{ reactivity}_{NOx}}$  here. Further, it was found that the O<sub>3</sub> production rates were also enhanced with the increase of

$OH$  reactivity $_{VOCs^\#}$  in the upper left area close to the dividing line. We defined it as a transitional regime where the  $O_3$  formation was comparably sensitive to  $VOCs^\#$  and  $NO_x$ . Beyond the transitional regime in the upper left area of the dividing line, the sensitivity of  $O_3$  formation to  $NO_x$  was generally ten times higher than to  $VOCs^\#$ , which was designated as  $NO_x$ -limited regime. The transitional regime and the  $NO_x$ -limited regime were divided by the blue dashed line in Figure 8.

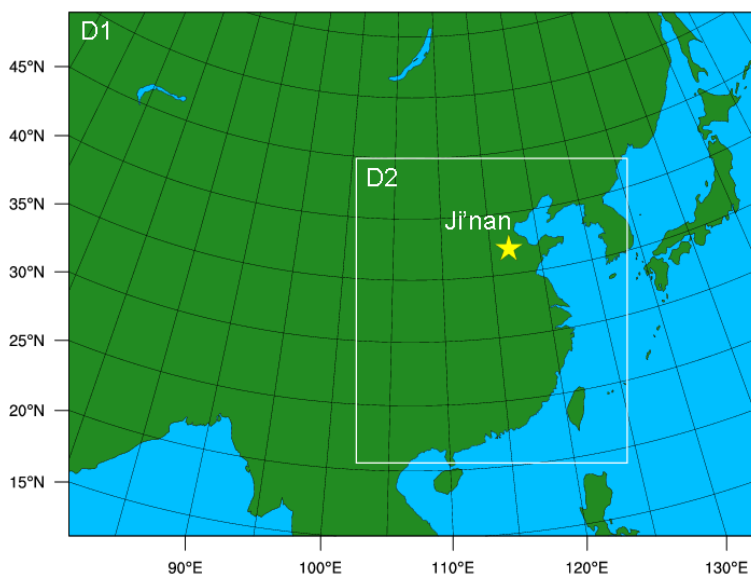


**Figure S1.** Agreement of the hourly (a)  $O_3$  and (b)  $NO_2$  between our observations on the campus of Shandong University and those monitored at the nearest AQMS by CNEMC.



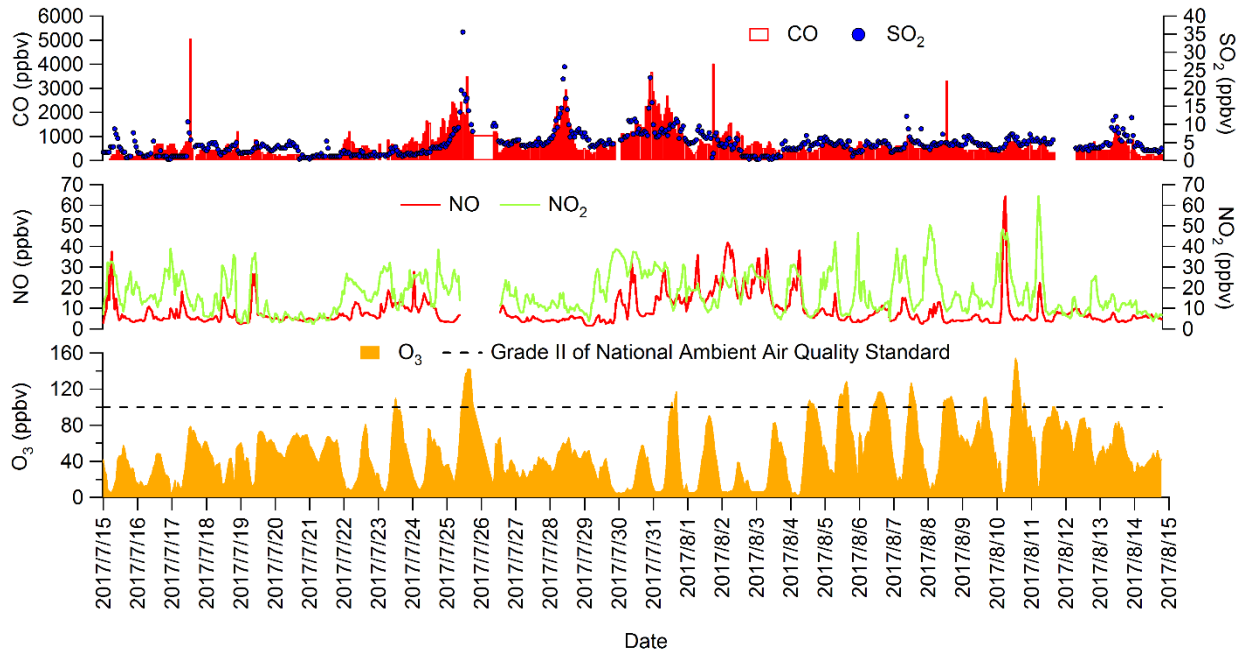


**Figure S2.** Inter-comparison of VOC analysis results between our laboratory (x axis) and Prof. Donald Blake's group (y axis). *n*-butane, benzene, toluene and isoprene are selected as examples. The red dashed line represents the linear regression between VOCs analyzed in two laboratories.

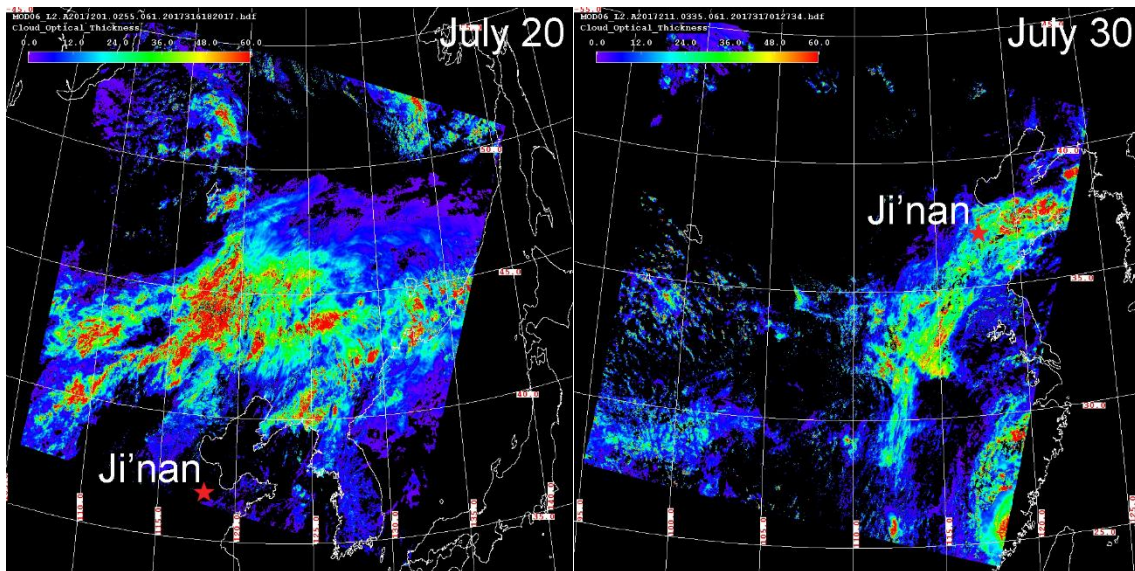


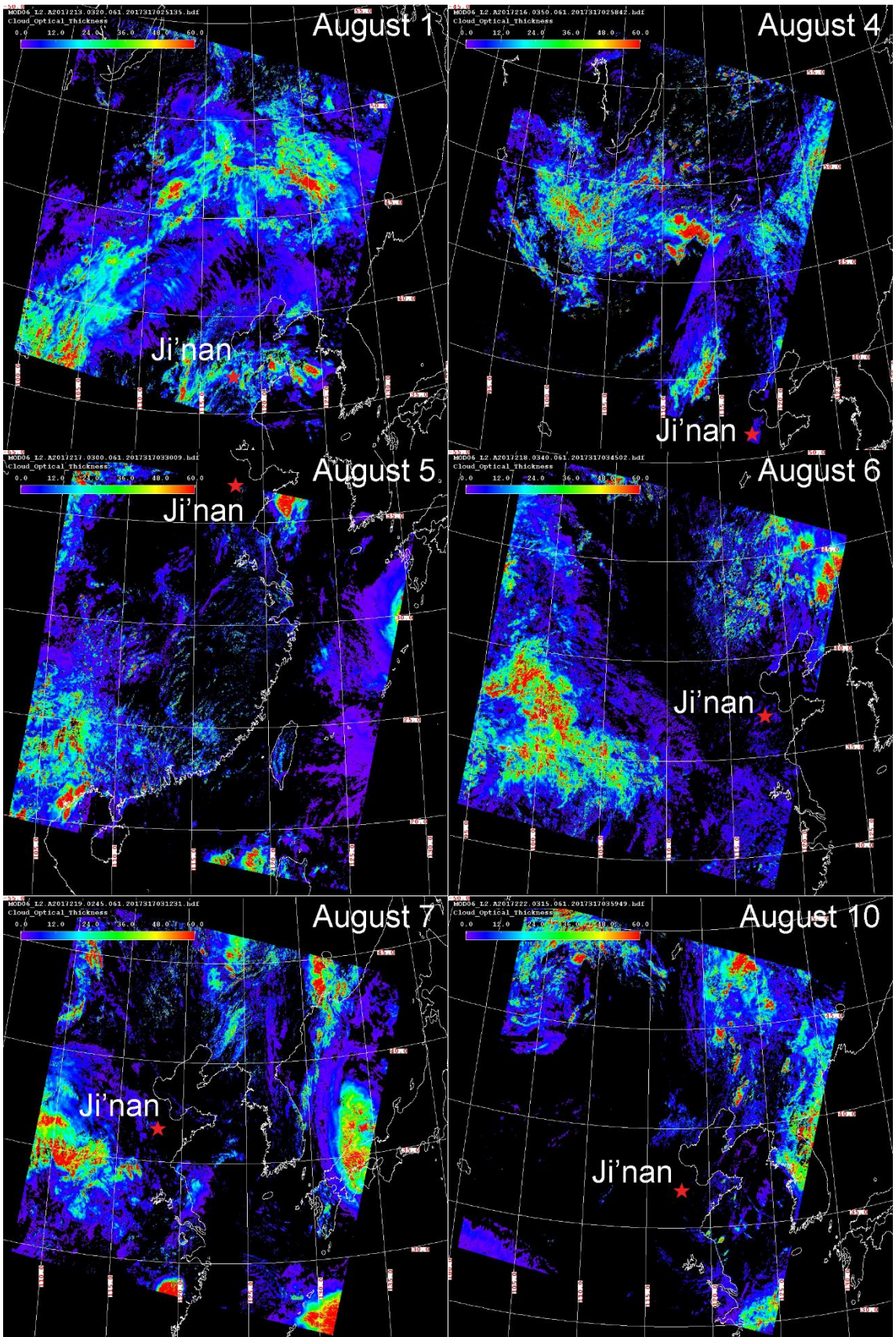
**Figure S3.** Settings of the two-nested domains for the WRF-CMAQ model. D1 and D2 are the outer and inner domains, covering the entire continental area of China and eastern China, respectively. The yellow star represents Ji'nan.



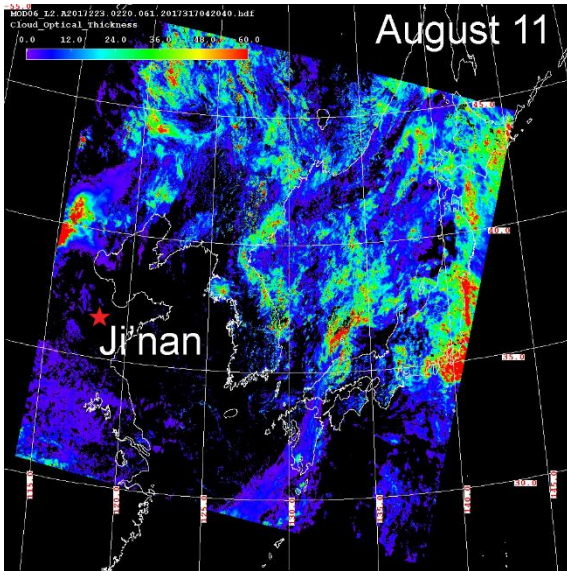


**Figure S4.** Hourly variations of trace gases monitored at the sampling site (O<sub>3</sub>, NO and NO<sub>2</sub>) and at the nearest AQMS (CO and SO<sub>2</sub>) during July 15-August 14, 2017.

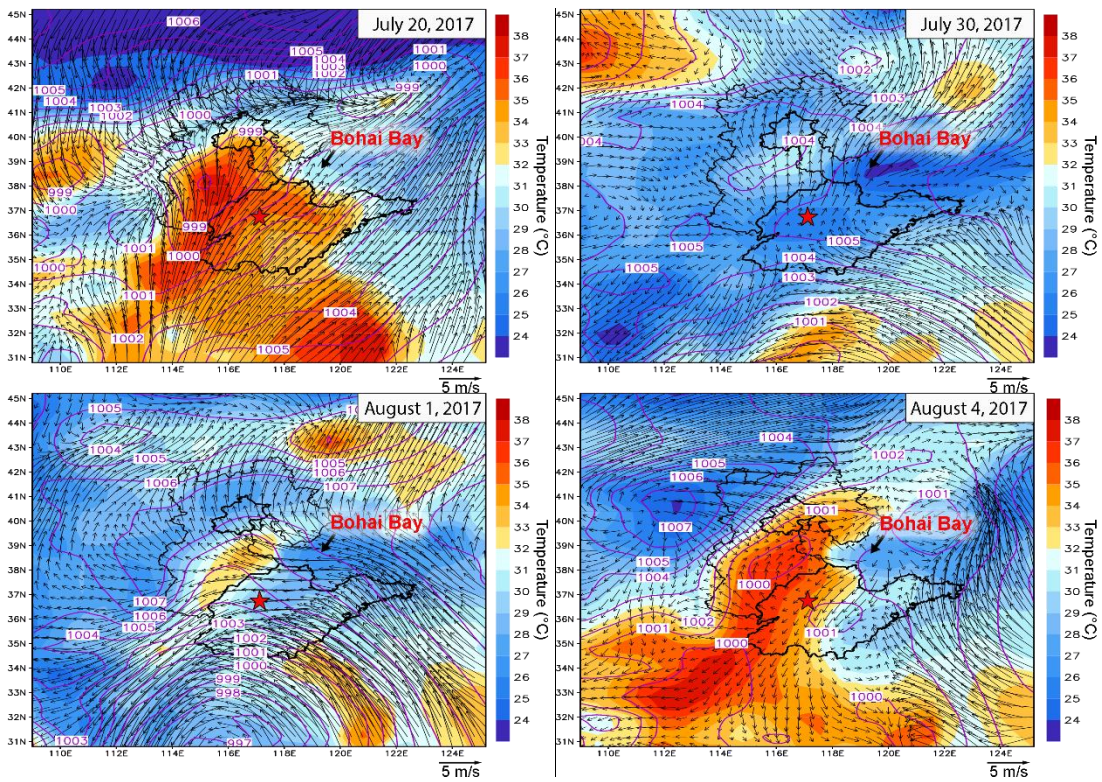




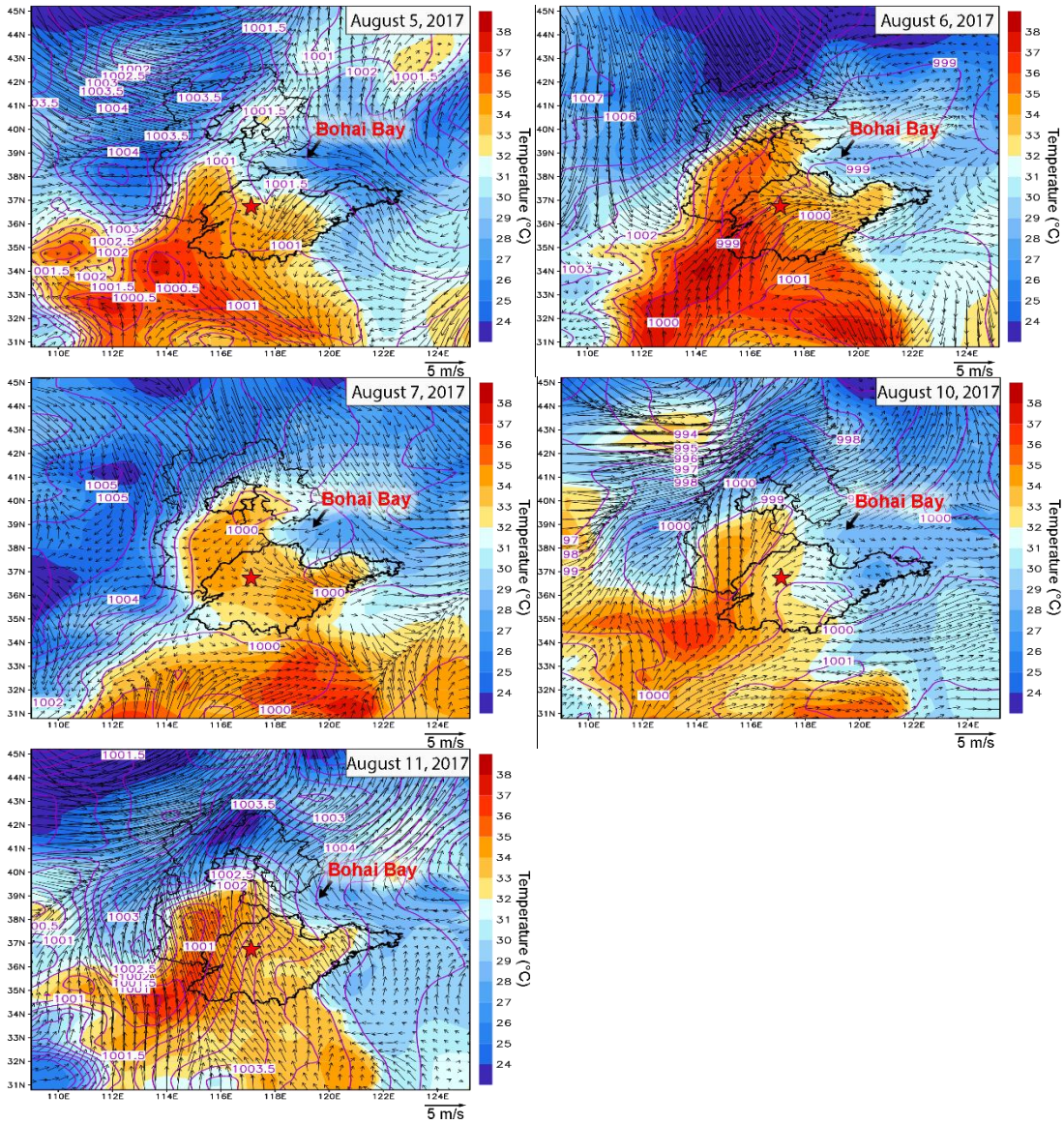




**Figure S5.** Cloud optical depth (COD) retrieved from terra/MODIS at noon (10:30 – 12:00 LT) of the canister sampling days. The color scale denotes for the COD within the range of 0 (purple) to 60 (red). The red star denotes Ji'nan.

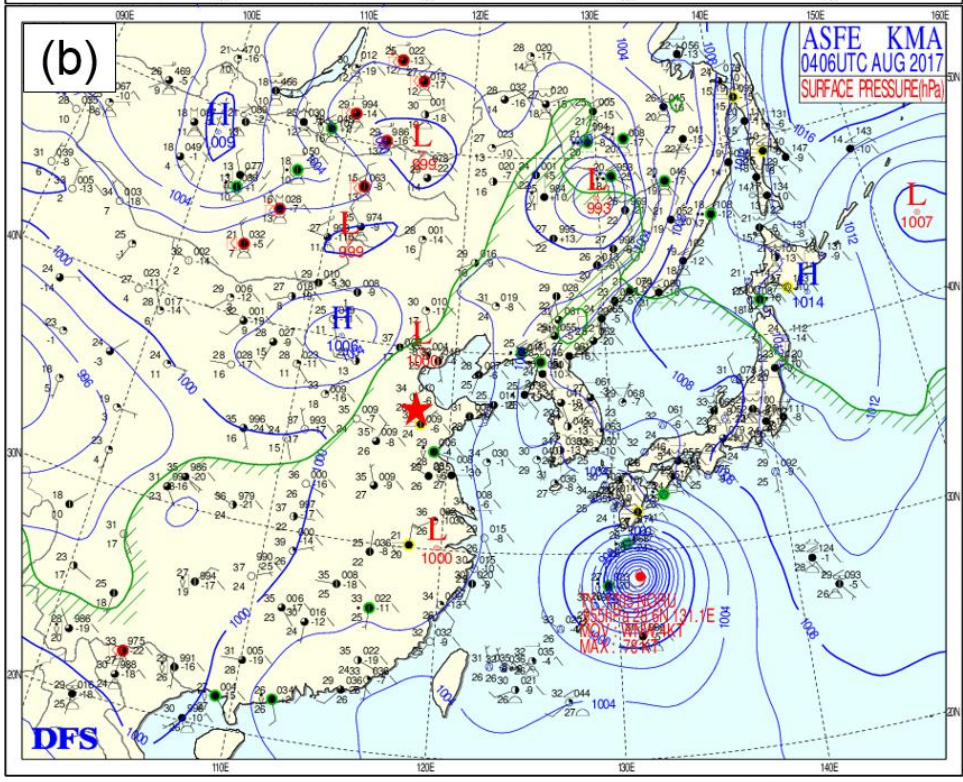
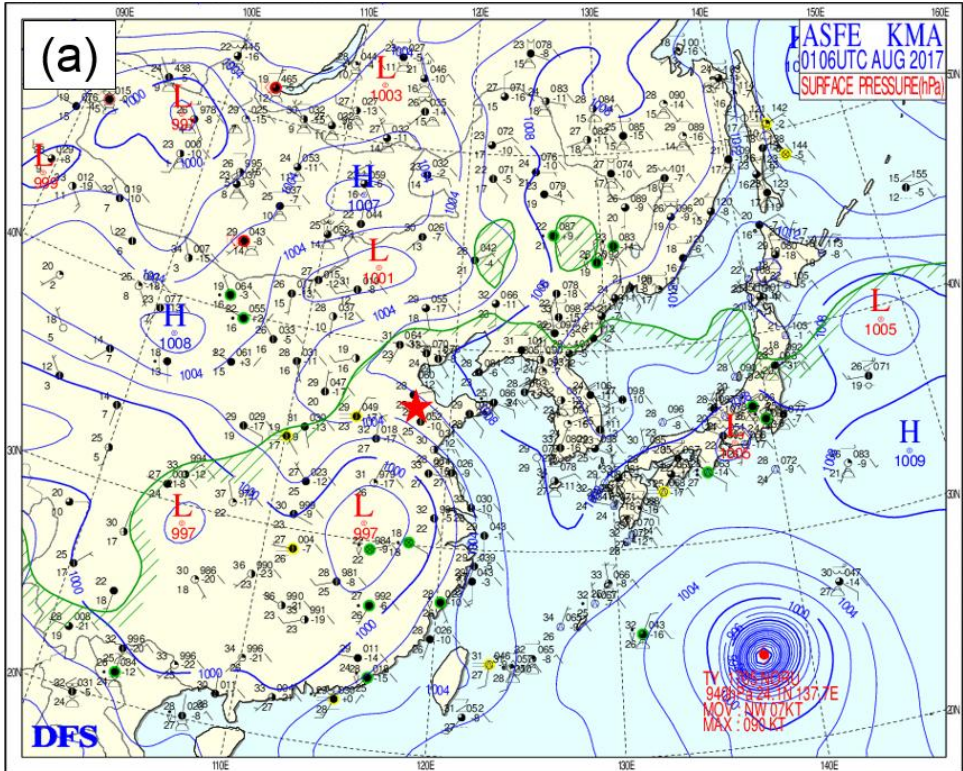




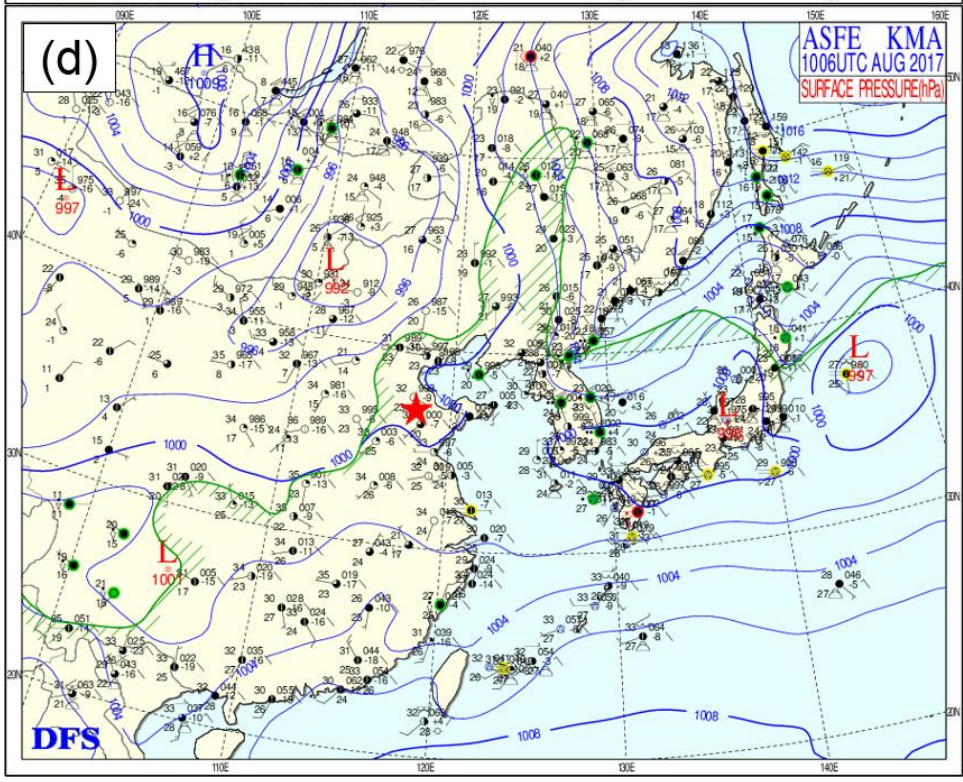
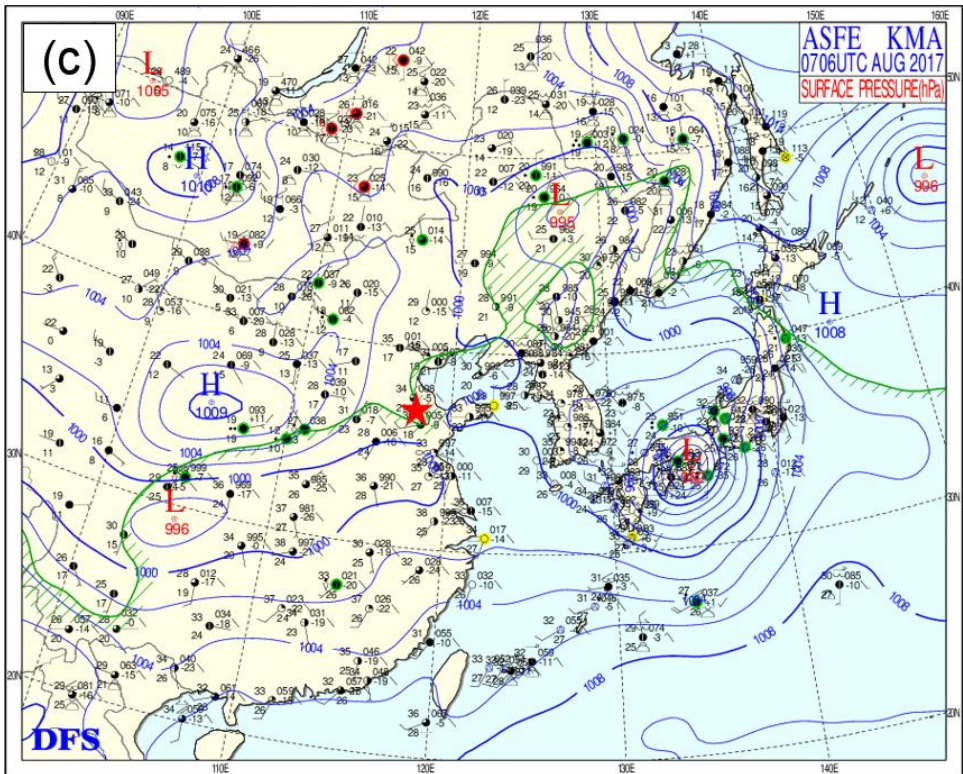


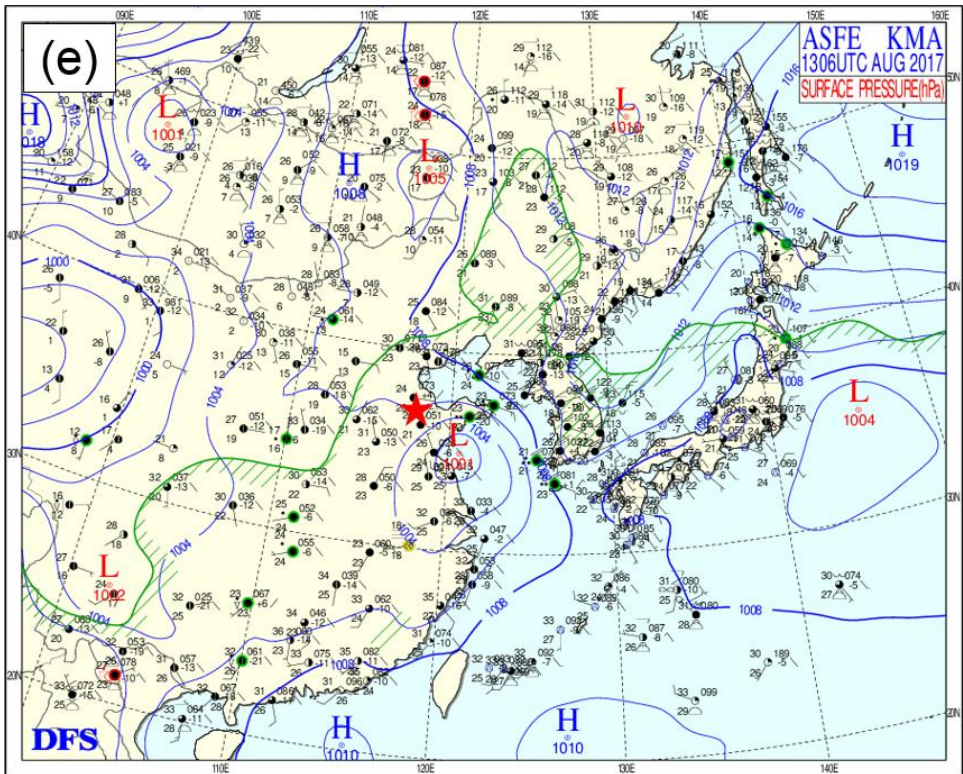
**Figure S6.** Weather charts at 14:00 LT on individual VOC sampling days. The red star denotes for Ji'nan. The dark black line is the boundary of Shandong province. Bohai Bay is located to the northeast of Shandong province. Numbers in the figure are sea-level pressures in unit of hPa.





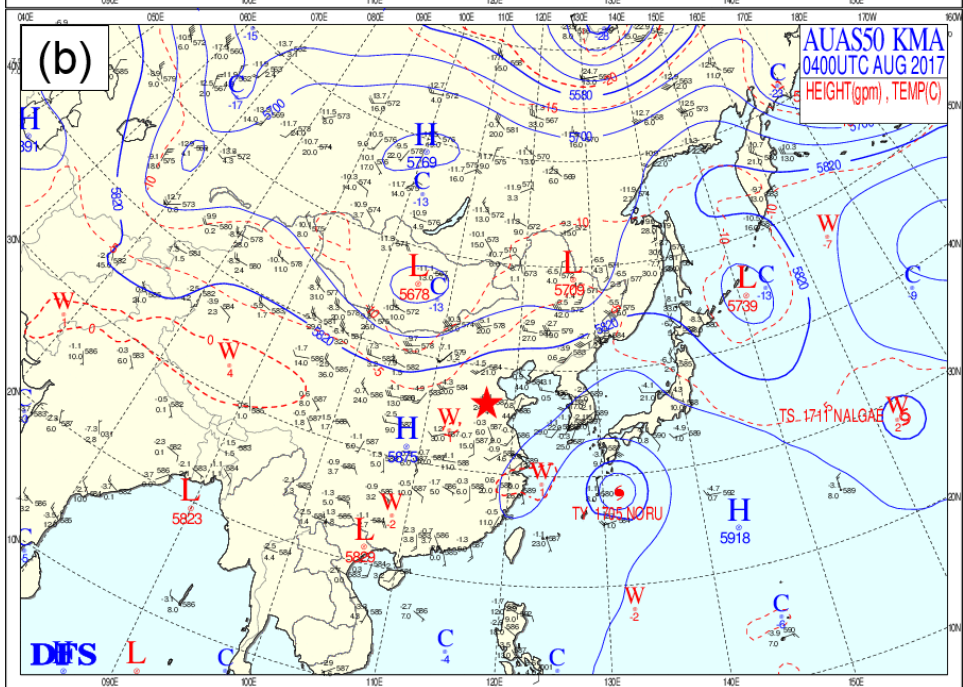
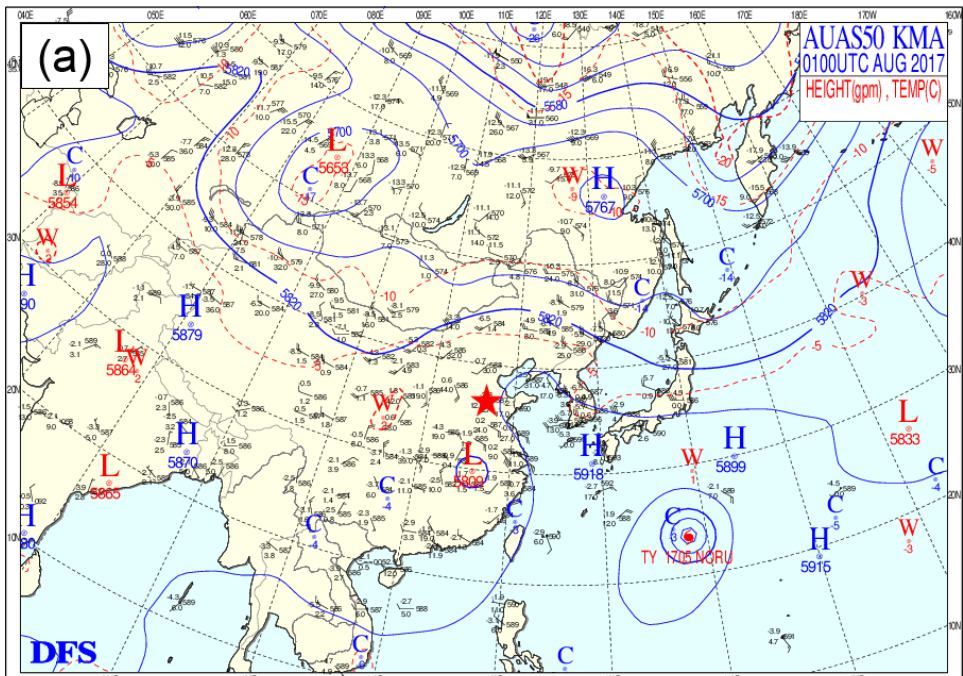


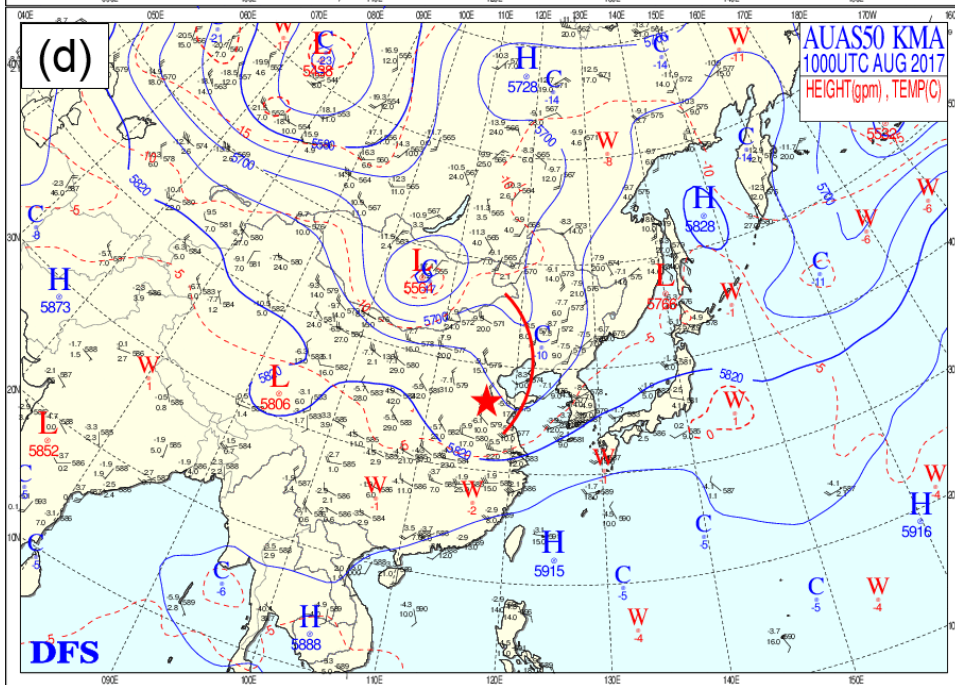
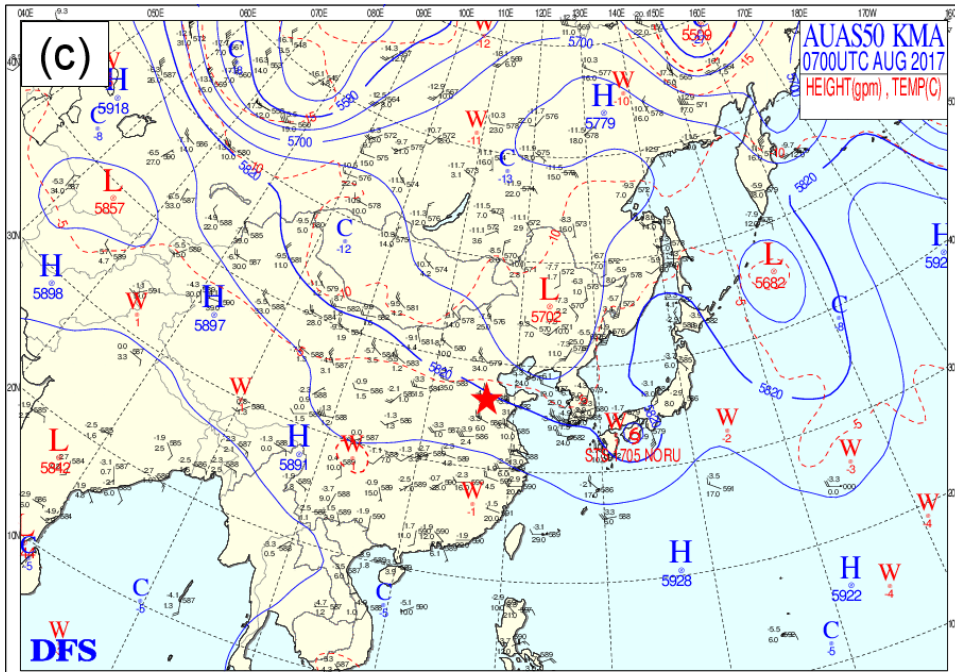


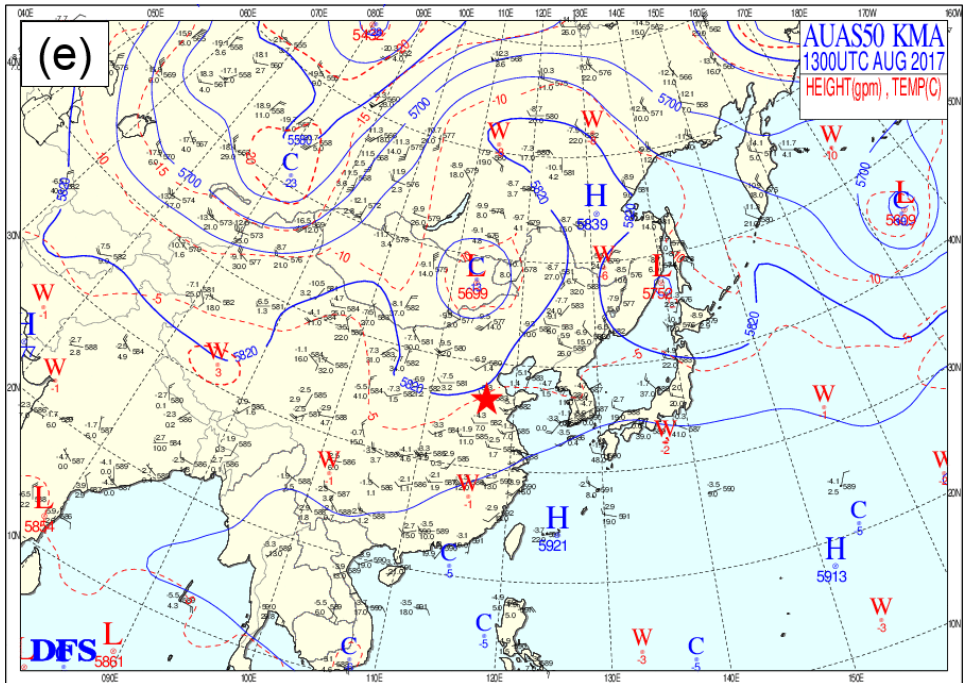


**Figure S7.** Weather chart over the Northeast Asia on (a) August 01, (b) August 04, (c) August 07, (d) August 10 and (e) August 13 at 06:00 UTC (14:00 LT) at surface level. The red star denotes Ji'nan city. The capital letters "H" and "L" represent high pressure center and low pressure center, respectively. Blue lines are the sea level isobars. Green line is the isometric humidity line with the specific humidity of  $\geq 15\text{g/kg}$  on the grid side. All the charts can be accessed through the link: <http://222.195.136.24/forecast.html>.

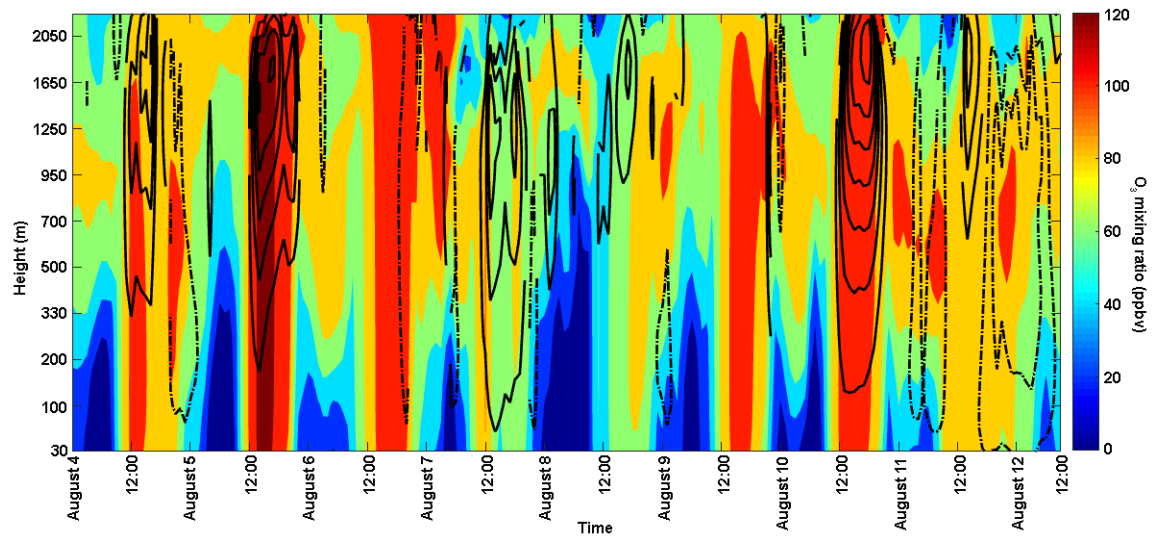




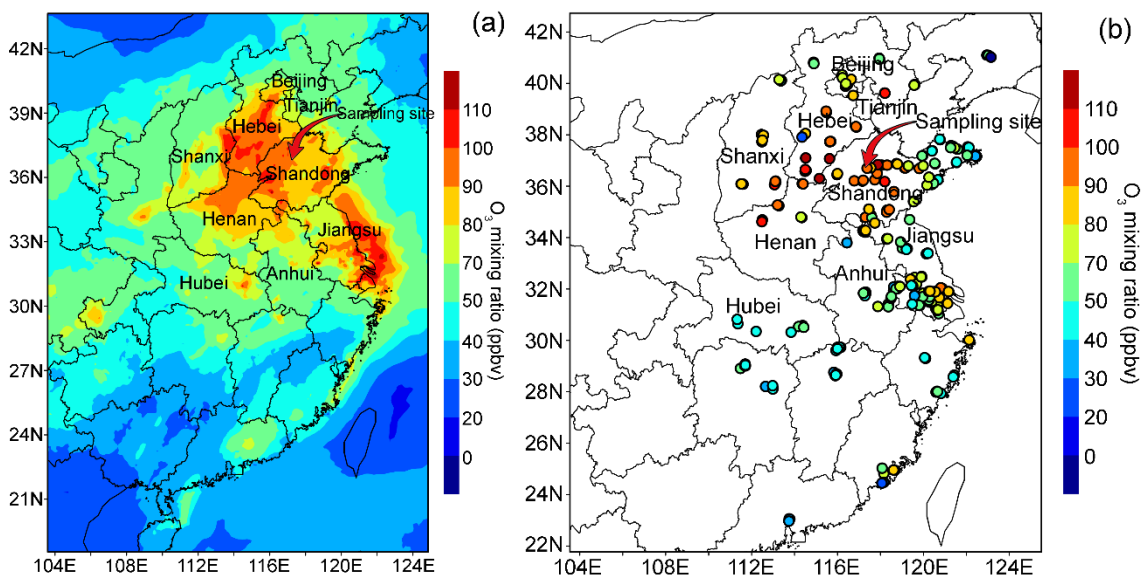




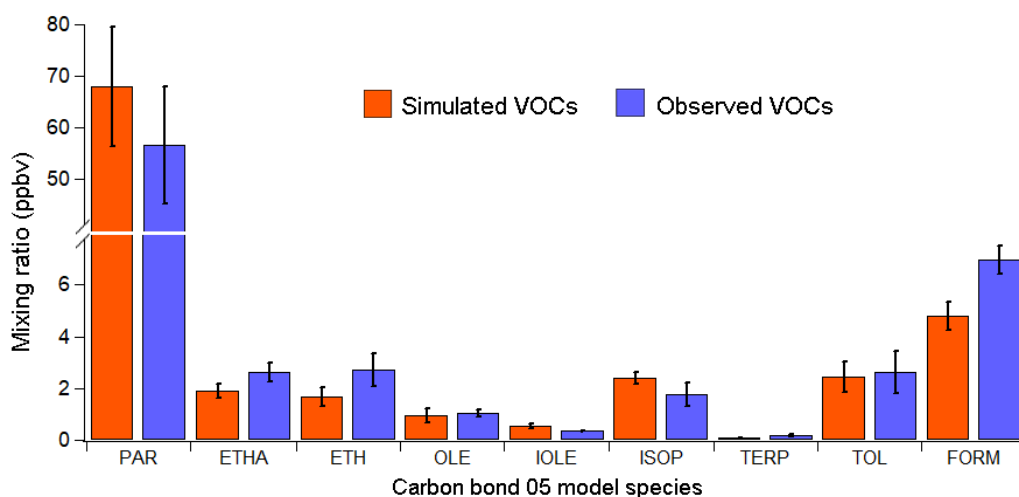
**Figure S8.** Weather chart over the Northeast Asia on (a) August 01, (b) August 04, (c) August 07, (d) August 10 and (e) August 13 at 00:00 UTC (08:00 LT) at altitude of 500 hPa. The red star denotes Ji'nan city. The capital letters "H" and "L" represent high pressure center and low pressure center, respectively. Blue lines are the 500 hPa geopotential height (gpm) lines. The red curve in panel (d) demonstrates the low pressure trough. All the charts can be accessed through the link: <http://222.195.136.24/forecast.html>.



**Figure S9.** Vertical profile of the simulated  $O_3$  over Ji'nan during August 4-11. The black solid and dotted lines represent the updraft and downdraft simulated by WRF-CMAQ, respectively. The areas with no line indicate that there were no simulated winds in vertical direction.

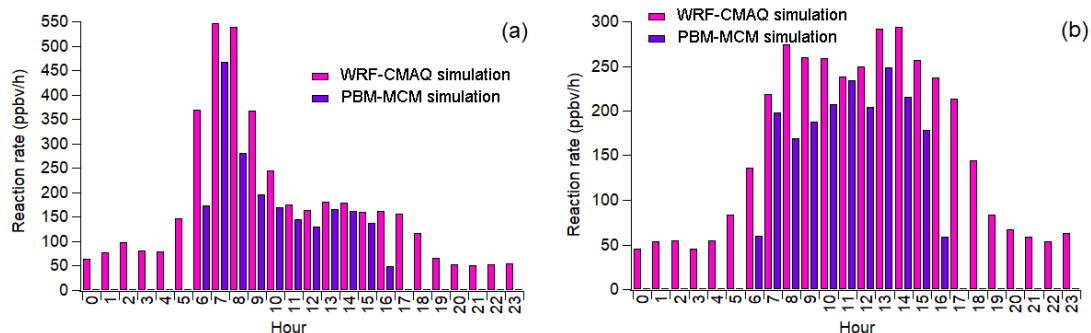


**Figure S10.** Comparison between the spatial distributions of (a) the WRF-CMAQ simulated  $O_3$  and (b) the observed  $O_3$  at 14:00 LT averaged over August 4-11. The observed  $O_3$  is acquired from the AQMSs of CNEMC.

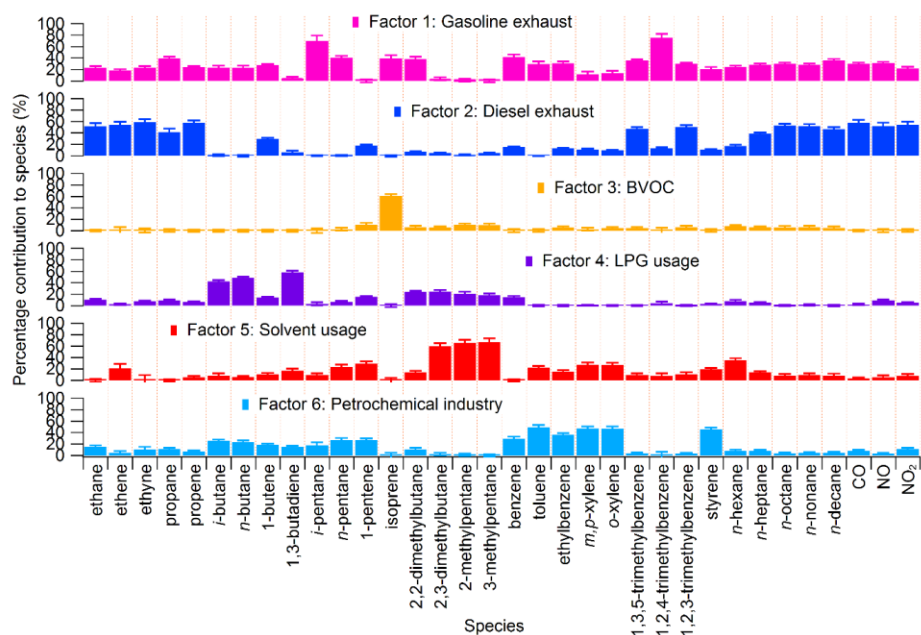


**Figure S11.** Comparison between the averages of the hourly observed and WRF-CMAQ simulated VOCs. PAR: paraffin carbon bond; ETHA: ethane; ETH: ethene; OLE: terminal olefin carbon bond; IOLE: internal olefin carbon bond; ISOP: isoprene; TERP: terpene; TOL: toluene and other monoalkyl aromatics; FORM: formaldehyde. The matrix of assignments from real compounds to carbon bond 05 model species can be found in Yarwood et al. (2005).

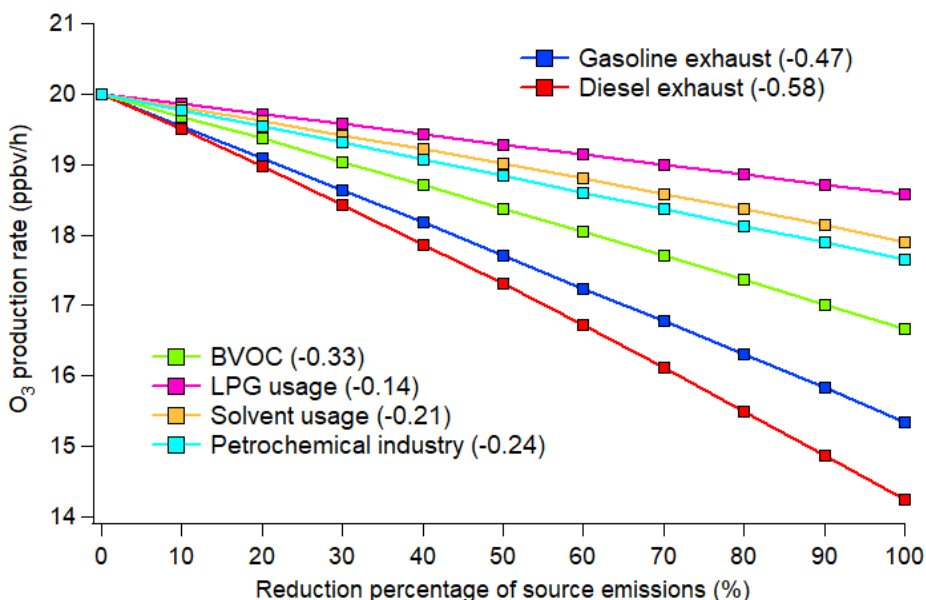




**Figure S12.** Average diurnal cycle of "NO+O<sub>3</sub>" reaction rates simulated by WRF-CMAQ and PBM-MCM during (a) O<sub>3</sub> episodes and (b) non-episodes.



**Figure S13.** Profiles of the six sources of O<sub>3</sub> precursors identified for the samples collected in daytime of the VOC sampling days in Ji'nan.



**Figure S14.** Average O<sub>3</sub> production rate at 12:00 LT during O<sub>3</sub> episodes as a response of the reduction percentages of source emissions. Numbers in the brackets are the average O<sub>3</sub> reduction efficiencies (ppbv/10% reduction in source emissions).

Species	Site	Instrument	Resolution	Accuracy	Precision	Detection limit
SO <sub>2</sub>	AQMS*	API, Model 100 E	20 sec	<20%	0.5% of reading above 50 ppbv	0.4 ppbv
CO	AQMS	API, Model 300 E	10 sec	<20%	0.5% of reading	40 ppbv
NO-NO <sub>2</sub> -NO <sub>x</sub>	AQMS	API, model 200E	20 sec	<20%	0.5% of reading	0.4 ppbv
	Campus site #	Thermo, Model 42C	1 min	<15%	0.4 ppbv	0.4 ppbv
O <sub>3</sub>	AQMS	API, model 400E	10 sec	<20%	<0.5% of reading	0.6 ppbv
	Campus site	Thermo, Model 49C	20 sec	<15%	1.0 ppbv	1.0 ppbv

\* An air quality monitoring station of China National Environmental Monitoring Center closest to our sampling site in the campus of Shandong University; # Our sampling site on the campus of Shandong University.

**Table S1.** Descriptions of the trace gas analyzers used in this study.

Date	Episode/Non-episode	J(O <sup>1</sup> D) (s <sup>-1</sup> )	JNO <sub>2</sub> (s <sup>-1</sup> )
July 20	Non-episode	$3.40 \times 10^{-5}$	$9.27 \times 10^{-3}$
July 30	Non-episode	$1.02 \times 10^{-5}$	$2.73 \times 10^{-3}$

August 1	Non-episode	$2.71 \times 10^{-5}$	$7.50 \times 10^{-3}$
August 4	Episode	$2.85 \times 10^{-5}$	$7.95 \times 10^{-3}$
August 5	Episode	$2.69 \times 10^{-5}$	$7.50 \times 10^{-3}$
August 6	Episode	$2.75 \times 10^{-5}$	$7.70 \times 10^{-3}$
August 7	Episode	$2.34 \times 10^{-5}$	$6.52 \times 10^{-3}$
August 10	Episode	$3.07 \times 10^{-5}$	$8.72 \times 10^{-3}$
August 11	Episode	$2.90 \times 10^{-5}$	$8.25 \times 10^{-3}$

**Table S2.** Daily maximum photolysis rates of O<sub>3</sub> and NO<sub>2</sub> on VOC sampling days in Ji'nan.

OH reactivity of species X	Full name of species/VOC groups	Species included
RNO	Nitric oxide	Nitric oxide
RNO <sub>2</sub>	Nitrogen dioxide	Nitrogen dioxide
RCO	Carbon monoxide	Carbon monoxide
RCarbonyls	Carbonyls	Formaldehyde, acetaldehyde, acetone, hexanal
RBVOCs	Biogenic VOCs	Isoprene, <i>α</i> -pinene, <i>β</i> -pinene
RAromatics	Aromatics	Benzene, toluene, ethylbenzene, <i>m/p</i> -xylenes, <i>o</i> -xylene
RAkenes	Alkenes	Ethene, ethyne, propene, 1- <i>i</i> -butene, 1,3-butadiene, <i>trans</i> -2-butene, <i>cis</i> -2-butene, 1-pentene
RAkanes	Alkanes	Ethane, propane, <i>n/i</i> -butanes, <i>n/i</i> -pentanes

**Table S3.** Full name of inorganic trace gases and VOC species for the calculation of OH reactivity.

Meteorological parameter/Air pollutant	Avg. Obs.	Avg. Sim.	Diff.	RMSE	NMB	NME	IOA
Temp. (°C)	30.0	30.7	0.7	2.4	0.02	0.06	0.89
R.H. (%)	72.7	67.5	-5.2	14.4	-0.06	0.15	0.82
W.S. (m/s)	2.8	3.3	0.5	1.5	0.38	0.56	0.74
Press. (hPa)	1000.5	998.8	-1.7	4.0	-0.002	0.003	0.56
NO <sub>2</sub> (ppbv)	26.7	28.4	1.7	16.7	0.18	0.58	0.73
O <sub>3</sub> (ppbv)	62.8	52.4	-10.4	24.0	-0.07	0.48	0.89

**Table S4.** Statistical comparisons of the WRF-CMAQ simulated and observed meteorological parameters, O<sub>3</sub> and NO<sub>2</sub>. The comparisons are made for the hourly data in 24 hours on all the VOC sampling days.



O <sub>3</sub> production pathway	O <sub>3</sub> destruction pathway
HO <sub>2</sub> + NO	OH + NO <sub>2</sub>
RO <sub>2</sub> + NO	O <sup>1</sup> (D) + H <sub>2</sub> O
	O <sub>3</sub> + OH
	O <sub>3</sub> + HO <sub>2</sub>
	O <sub>3</sub> + alkenes

**Table S5.** Production and destruction pathways of O<sub>3</sub>.

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