



- 1 Arctic marine secondary organic aerosol contributes
- 2 significantly to summertime particle size distributions in the
- 3 Canadian Arctic Archipelago
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#### 34 Abstract

35 36 Summertime Arctic aerosol size distributions are strongly controlled by natural regional 37 emissions. Within this context, we use a chemical transport model with size-resolved 38 aerosol microphysics (GEOS-Chem-TOMAS) to interpret measurements of aerosol size 39 distributions from the Canadian Arctic Archipelago during the summer of 2016, as part of 40 the "NETwork on Climate and Aerosols: addressing key uncertainties in Remote Canadian Environments" (NETCARE). Our simulations suggest that condensation of 41 42 secondary organic aerosol (SOA) from precursor vapors emitted in the Arctic and near Arctic marine (open ocean and coastal) regions plays a key role in particle growth events 43 that shape the aerosol size distributions observed at Alert (82.5° N, 62.3° W), Eureka 44 45 (80.1° N, 86.4° W), and along a NETCARE ship track within the Archipelago. We refer to this SOA as Arctic marine SOA (Arctic MSOA) to reflect the Arctic marine-based and 46 47 likely biogenic sources for the precursors of the condensing organic vapors.

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Arctic MSOA from a simulated flux (500 µg m<sup>-2</sup> d<sup>-1</sup>, north of 50° N) of precursor vapors 49 50 (assumed yield of unity) reduces the summertime particle size distribution modelobservation mean fractional error by 2- to 4-fold, relative to a simulation without this 51 52 Arctic MSOA. Particle growth due to the condensable organic vapor flux contributes 53 strongly (30-50%) to the simulated summertime-mean number of particles with diameters 54 larger than 20 nm in the study region, and couples with ternary particle nucleation 55 (sulfuric acid, ammonia, and water vapor) and biogenic sulfate condensation to account for more than 90% of this simulated particle number, a strong biogenic influence. The 56 57 simulated fit to summertime size-distribution observations is further improved at Eureka 58 and for the ship track by scaling up the nucleation rate by a factor of 100 to account for 59 other particle precursors such as gas-phase iodine and/or amines and/or fragmenting primary particles that could be missing from our simulations. Additionally, the fits to 60 61 observed size distributions and total aerosol number concentrations for particles larger 62 than 4 nm improve with the assumption that the Arctic MSOA contains semi-volatile 63 species; reducing model-observation mean fractional error by 2- to 3-fold for the Alert and ship track size distributions. Arctic MSOA accounts for more than half of the 64 65 simulated total particulate organic matter mass concentrations in the summertime





66 Canadian Arctic Archipelago, and this Arctic MSOA has strong simulated summertime 67 pan-Arctic-mean top-of-the-atmosphere aerosol direct (-0.04 W m<sup>-2</sup>) and cloud-albedo 68 indirect (-0.4 W m<sup>-2</sup>) radiative effects. Future work should focus on further understanding 69 summertime Arctic sources of Arctic MSOA.

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### 71 1. Introduction

Aerosols have an important role in the summertime Arctic climate system. Similar to 73 74 their effects in other regions, aerosols interact directly with incoming solar radiation by scattering and absorption (Charlson et al., 1992; Hegg et al., 1996; Yu et al., 2006; 75 Shindell and Faluvegi, 2009; Yang et al., 2014) and indirectly through modification of 76 77 cloud properties by acting as the seeds for cloud droplet formation (Lohmann and 78 Feichter, 2005; McFarquhar et al., 2011). In the summertime Arctic, efficient wet 79 removal by precipitation and the smaller extent of the polar dome limit transport of pollution from lower latitudes and maintain an atmosphere that is more pristine than in 80 81 the Arctic winter and springtime (Barrie, 1995; Polissar et al., 2001; Quinn et al., 2002; Stohl, 2006; Garrett et al., 2011; Brock et al., 2011; Fisher et al., 2011; Sharma et al., 82 2013; Xu et al., 2017). As a result, natural regional Arctic sources make strong 83 contributions to summertime Arctic aerosol, to the related radiative effects, and to 84 85 associated uncertainties (Korhonen et al., 2008; Leck and Bigg, 2010; Heintzenberg and Leck, 2012; Karl et al., 2013; Carslaw et al., 2013; Heintzenberg et al., 2015; Croft et al., 86 87 2016a; Willis et al., 2016; Burkart et al., 2017a; Mungall et al., 2017; Willis et al., 2017; Dall'Osto et al., 2017; Breider et al., 2017; Dall'Osto et al., 2018b; Leaitch et al., 2018). 88 89

90 Observations indicate that aerosol particle formation and growth events occur frequently

91 in the summertime Canadian Arctic Archipelago region (within 60-130° W and

92 60-85° N) (Chang et al., 2011b; Leaitch et al., 2013; Willis et al., 2016; Willis et al.,

93 2017; Croft et al., 2016a; Burkart et al., 2017a; Burkart et al., 2017b; Collins et al., 2017;

94 Tremblay et al., 2018). These events contribute towards shaping a summertime aerosol 95 number size distribution that is characterized by a dominant Aitken mode (particles with 96 diameters between 10 and 100 nm) in this region (Croft et al., 2016b), similar to 97 observations at other pan-Arctic sites (Tunved et al., 2013; Asmi et al., 2016; Nguyen et





98	al., 2016; Freud et al., 2017; Gunsch et al., 2017; Heintzenberg et al., 2017; Kolesar et al.,
99	2017). Summertime Arctic aerosol size distributions are also characterized by a
100	suppressed accumulation mode (particles with diameters between 100 and 1000 nm) due
101	to the efficient wet removal processes in frequently drizzling low clouds (Browse et al.
102	2014) and the limited transport from lower latitudes (Stohl, 2006; Law and Stohl, 2007;
103	Korhonen et al., 2008)
104	
105	Evidence points to a strong marine biogenic influence on summertime Arctic aerosols
106	(Leck and Bigg, 2010; Chang et al., 2011a; Heintzenberg et al., 2015; Dall'Osto et al.,
107	2018b). The oceans provide the atmosphere with many particle-relevant trace gases
108	(Carpenter et al., 2012; Carpenter and Nightingale, 2015; Ghahremaninezhad et al., 2017;
109	Mungall et al., 2017), as well as primary particles (Gantt and Meskhidze, 2013; Grythe et
110	al., 2014; Wilson et al., 2015). Arctic melt ponds and melting ice are also sources of
111	vapors such as dimethyl sulfide (DMS) (Hayashida et al., 2017; Gourdal et al., 2018),
112	which yield condensable products following oxidation (Barnes et al., 2006) that can form
113	and grow particles (Kirkby et al., 2011). As well, observations suggest a key role for
114	Arctic marine secondary organic aerosol (Arctic MSOA) in the Canadian Arctic
115	Archipelago (Willis et al., 2017; Burkart et al., 2017a; Köllner et al., 2017; Leaitch et al.,
116	2018). The condensing vapors that contribute to particle growth by formation of
117	secondary organics in the Canadian Arctic Archipelago may be more volatile than at
118	lower latitudes because smaller modes (particle diameters around 20 nm) grow somewhat
119	more slowly than larger modes (particle diameters around 90 nm) (Burkart et al., 2017a).
120	However, these vapors are still capable of growing newly formed particles, and the
121	details about the origin and composition of Arctic MSOA precursors are not well
122	understood.
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124	In this study, the terminology Arctic MSOA indicates SOA formed from any organic
125	precursor vapors emitted in open ocean and coastal regions north of 50° N, excluding
126	methane sulfonic acid, which we treat as a separate aerosol component, consistent with
127	most filter-based aerosol species mass measurements. In the Canadian Arctic
128	Archipelago, Arctic MSOA is likely strongly controlled by marine biogenic activity

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- 129 (Willis et al, 2017; Leaitch et al., 2018). Due to the spatial and temporal variability, and
- 130 diversity of organic precursor vapor sources and chemistry, the chemical character of
- 131 Arctic MSOA is not necessarily the same as other types of MSOA arising from
- 132 precursors originating in other marine regions, which could be more strongly influenced
- by shipping and differing types of marine biogenic activity (Facchini et al., 2008; Rinaldi
- 134 et al., 2010). As well, under our definition of Arctic MSOA, the presence of Arctic
- 135 MSOA is not limited to the atmospheric marine boundary layer or marine environment
- 136 due to transport of precursors and Arctic MSOA to continental regions.
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138 There are few measurements of size-resolved aerosol mass concentrations in the 139 summertime Arctic (Zábori et al., 2015; Giamarelou et al., 2016; Tremblay et al., 2018). 140 Such measurements can provide insight about the processes that control the size 141 distributions. Limited observations indicate that growing Aitken-mode particles with 142 diameters between 60 and 100 nm in the Canadian Arctic Archipelago are composed 143 almost entirely of organics, suggesting a strong role for secondary organics (Tremblay et 144 al., 2018). On the other hand, observations from the Svalbard region (within 74-81 °N 145 and 10-35 °E) indicate that the smaller sub-12 nm particles are composed primarily of 146 ammonium sulfate, suggesting that ternary nucleation and early growth involving gas-147 phase water, ammonia (NH<sub>3</sub>) and sulfuric acid (H<sub>2</sub>SO<sub>4</sub>) play a key role in the 148 development of nucleation-mode aerosols (particle diameters smaller than 10 nm) in the 149 region (Giamarelou et al., 2016). In the Canadian Arctic Archipelago, summertime gas-150 phase NH<sub>3</sub> concentrations have been observed to be in the range of a few hundred pptv 151 (Wentworth et al., 2016), levels that could contribute to initial particle formation (Napari 152 et al., 2002; Kirkby et al., 2011; Almeida et al., 2013). Sources for NH<sub>3</sub> in this region are 153 not yet fully understood, but contributors include Arctic seabird colonies, biomass 154 burning, and possibly other terrestrial sources such as tundra ecosystems that can 155 contribute to bi-directional exchange (Skrzypek et al., 2015; Croft et al., 2016a; Lutsch et 156 al., 2016; Wentworth et al., 2016). In addition to  $NH_3$ ,  $H_2SO_4$  and gas-phase water, other 157 components of nucleation mode particles (diameters less than 10 nm) could include, but 158 are not limited to, iodine (Allan et al., 2015; Dall'Osto et al., 2018a), amines (Almeida et





al., 2013) and fragmentation of primary particles as clouds and fog evaporate (Leck and

- 160 Bigg, 2010).
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162 Given the complexity of interacting processes and source-related uncertainties described 163 above, a coupled model-measurement-based approach enables exploration of the role of 164 particles of biogenic origin in development of summertime aerosol size distributions in 165 the Canadian Arctic Archipelago. In this study, we use the GEOS-Chem-TOMAS model 166 (http://geos-chem.org) with size-resolved aerosol microphysics to interpret aerosol 167 measurements taken during the summer of 2016 in the Canadian Arctic Archipelago, at 168 both Alert and Eureka, in Nunavut, Canada, and also along the 2016 CCGS Amundsen 169 ship track. These measurements include aerosol mass loading, total aerosol number and 170 aerosol size distributions, some of which were taken as part of the NETwork on climate 171 and aerosols: addressing key uncertainties in Remote Canadian Environments 172 (NETCARE) (Abbatt et al. (in prep.)). Section 2 describes our methodology, including further details about the observations, a model description, and a summary of simulations. 173 174 Section 3 interprets simulations and observations to explore the contribution of both 175 marine primary organic aerosol (arising from sea spray) and Arctic MSOA in shaping the 176 summertime aerosol size distributions and mass concentrations in the Canadian Arctic 177 Archipelago. We also consider the role of ternary nucleation, the simulated contribution to particle nucleation events and size distributions, and comparison with observations. 178 179 Section 3 also presents sensitivity studies to explore the role of the volatility of the Arctic 180 MSOA during growth events, and in shaping aerosol size distributions. Finally, Sect. 3 181 presents estimates of the contribution of Arctic MSOA to summertime Arctic direct and 182 indirect aerosol effects. Section 4 presents a summary, and highlights key directions for 183 future research.

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## 185 2. Methodology

- 186
- 187 2.1 Aerosol measurements in the Canadian Arctic Archipelago
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189 Figure1 shows the locations of aerosol measurements, taken at both Alert (82.5° N, 62.3° W) and Eureka (80.1° N, 86.4° W), in Nunavut, Canada and along the 2016 ship track of 190 191 the CCGS Amundsen through the Canadian Arctic Archipelago that we interpret using the 192 GEOS-Chem-TOMAS model. The measurements at Alert, Nunavut, Canada are made at 193 the Global Atmospheric Watch (GAW) observatory. Weekly samples of ambient aerosol 194 are collected on Teflon filters, using a cyclone to limit particles to diameters smaller than 195 1 μm (PM1.0), and analyzed by ion chromatography for several inorganic ions including 196 sulfate, ammonium, nitrate, and methane sulfonic acid (MSA). Further details are 197 provided in Leaitch et al. (2013) and Li and Barrie (1993). Weekly samples are also 198 collected on quartz filters behind the cyclone (PM1.0) and analyzed for organic and 199 elemental carbon by thermal volatilization and combustion (Huang et al., 2006), as 200 described in Sharma et al. (2017) and Leaitch et al. (2018). There is a possibility that 201 volatile organic carbon (VOC) vapors could also be absorbed on the quartz filters 202 (Leaitch et al., 2018). As a result, we view these measurements as an upper limit on organic carbon mass concentrations at Alert as discussed further in Sect. 3.1. Since 2011. 203 204 hourly-mean size distributions for particles with diameters from 20 to 500 nm are 205 measured at Alert using a TSI 3034 Scanning Mobility Particle System (SMPS), which is 206 verified for sizing on site using mono-disperse particles of polystyrene latex and of 207 ammonium sulfate generated with a Brechtel Manufacturing Incorporated Scanning 208 Electrical Mobility Spectrometer. Total particle number concentration for particles larger 209 than 10 nm in diameter is measured at Alert with a TSI 3772 Condensation Particle 210 Counter (CPC). The CPC and SMPS agree to within 10% when all particles are large 211 enough to be counted by both instruments. Data that could be influenced by local camp 212 activities are filtered from the data set by removing data 1) when the wind direction was 213 within a  $45^{\circ}$  angle centred on the Alert camp; 2) during line zeroes to check the 214 connections to the instruments and any other repetitive occurrence that might influence 215 the measurements; 3) when logs indicated potential sources nearby (e.g. trucks); and 4) to 216 account for unknown factors, when data spikes remain that lasted two hours or less. 217

At Eureka, Nunavut, Canada, aerosol measurements are taken at the RidgeLab of the
Polar Environment Atmospheric Research Laboratory (PEARL) (Fogal et al., 2013),





- which is located on Ellesmere Island at 610 m above sea level and about 480 km southwest of Alert. Since 2015, size distributions for particles with diameters between 10 and 500 nm have been measured at the RidgeLab using a TSI 3034 SMPS. Data are recorded every three minutes and averaged to hourly means. Further details about the instrument and sampling are presented in Tremblay et al. (2018).
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Figure 1: *CCGS Amundsen* 2016 cruise track through Canadian Arctic Archipelago color coded by the number of days after 24 July 2016. Land is shaded in grey. Location of Alert and Eureka shown by red and cyan stars, respectively.

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233 During the summer of 2016, the research icebreaker *CCGS Amundsen* travelled through

the Canadian Arctic Archipelago as a part of the NETCARE project (Abbatt et al. (in





235 prep)). Figure 1 shows the cruise track for 24 July - 23 August 2016. During the cruise, 236 total number concentrations of aerosols with diameters larger than 4 nm were measured 237 with a TSI 3776 ultrafine condensation particle counter (UCPC). Aerosol size 238 distributions for particles with diameters between 10 nm to 430 nm were measured with a 239 TSI 3080/3087 SMPS. Data collected while the wind direction was less than  $60^{\circ}$  to port 240 and less than 90° to starboard of the ship's orientation were accepted for further analysis. 241 We consider all measurements taken over 23 July 2016 to 24 August 2016, when the ship 242 was north of 66° N. Further details about the instrumental setup and sampling are in 243 Collins et al. (2017). Measurements of  $NH_3$  were also taken during the cruise with a 244 Quantum Cascade Tunable Infrared Laser Differential Absorption Spectroscopy (QC-TILDAS) analyzer (Ellis et al., 2010). The instrument has a fast response time that 245 246 enabled measurements at 1 Hz during the cruise, with measurements taken from 29 July 247 2016 to 23 August 2016. NH<sub>3</sub> data were also filtered for wind direction, ship speed and 248 measured aerosol number concentrations to exclude periods that indicated influence from 249 the ship exhaust.

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251 Tundra samples were collected in triplicate from three sites near Alert, NU on 14 and 15 252 July 2016 to estimate tundra NH<sub>3</sub> emission potential. The sites ranged from 253 approximately 1 to 9 km west of the GAW observatory. Sampling and subsequent analysis for ammonium concentration ([NH<sub>4</sub><sup>+</sup>]) and pH were carried out according to 254 255 Wentworth et al. (2014). From mid June to the end of July 2016, hourly measurements of 256 tundra temperature were recorded adjacent to the GAW observatory using commercially 257 available soil temperatures sensors (iButtons, Maxim Integrated). Tundra [NH4<sup>+</sup>], pH, 258 (both based on the 14 and 15 July 2016 soil samples) and hourly temperature were used 259 to calculate the hourly NH<sub>3</sub> tundra compensation, which reflects the predicted 260 equilibrium NH<sub>3</sub> concentration in the atmosphere above the tundra (Wentworth et al., 261 2014). A tundra-air exchange velocity was calculated using a resistance-in-series scheme 262 with parameterizations from Wesley (1989) and Walker et al. (2014). The average  $NH_3$ 263 emissions at the three sites were then calculated as the mean of the products of the 264 exchange velocity and compensation point, resulting in estimated emission rates of 0.12, 1.4, and 2.2 ng m<sup>-2</sup> s<sup>-1</sup>. Here, we adopt the highest value to provide an upper estimate on 265





the contribution of the tundra to atmospheric NH<sub>3</sub>. It should be noted that extrapolating calculated emissions from discrete tundra samples to the entire Canadian Arctic Archipelago carries a very large degree of uncertainty. However, the paucity of necessary tundra measurements and the lack of any terrestrial Arctic NH<sub>3</sub> fluxes prevent a more rigorous approach.

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### 273 2.2 Model description

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275 The GEOS-Chem (GC) chemical transport model version v10-01 (http://geos-chem.org) 276 coupled to the TwO-Moment Aerosol Sectional (TOMAS) microphysics model (Adams 277 and Seinfeld, 2002; Kodros et al., 2016; Kodros and Pierce, 2017) is employed in this study. Our model version has 47 vertical levels and a 4° x 5° horizontal resolution. The 278 279 GEOS-FP reanalysis (http://gmao.gsfc.nasa.gov) provides the meteorological fields. We 280 use a TOMAS version with 15 size sections, including dry diameters ranging from 3 nm 281 to 10 µm (Lee and Adams, 2012). Tracers within each size bin include particle number 282 and mass of sulfate, black carbon (hydrophobic and hydrophilic), organic carbon 283 (hydrophobic and hydrophilic), sea salt, dust and water. All simulations are for the 284 months of July and August 2016, with a one-month spin-up during June that is not 285 included in our analysis.

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287 2.2.1 TOMAS aerosol microphysics

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289 The TOMAS aerosol microphysics scheme employed in our simulations has 13 290 logarithmically spaced size sections for aerosol dry diameters from approximately 3 nm 291 to 1  $\mu$ m, and 2 additional size sections to represent aerosol dry diameters from 1 to 10  $\mu$ m 292 (Lee and Adams, 2012). Particle formation is treated according to the ternary  $H_2SO_{4-}$ 293 NH<sub>3</sub>-H<sub>2</sub>O nucleation scheme described by Baranizadeh et al. (2016). The formation rate 294 of particles at about 1.2 nm in mass diameter is determined from a full kinetics simulation 295 by Atmospheric Cluster Dynamics Code (ACDC) (Olenius et al., 2013) using particle 296 evaporation rates based on quantum chemistry. The scheme is implemented as a look-up





297 table as a function of gas-phase  $H_2SO_4$  and  $NH_3$  concentrations, relative humidity, 298 temperature and condensation sink for condensable vapors. Growth and loss of particles 299 smaller than 3 nm are approximated with the Kerminen et al. (2004) scheme. 300 Implementation of this scheme is supported by the findings of Giamarelou et al. (2016) 301 that nucleation-mode particles in the summertime Arctic are predominantly ammoniated 302 sulfates. All simulations use the Brownian coagulation scheme of Fuchs (1964) and 303 consider coagulation between all particle sizes, an important sink for particle number, 304 particularly for those particles with diameters smaller than 100 nm. Coagulation between 305 aerosols contained in cloud hydrometeors and interstitial aerosols is parameterized as 306 described in Pierce et al. (2015). An overview of the condensational-growth assumptions 307 follows the discussion of inventories and secondary organic aerosol (SOA) scheme 308 below.

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310 2.2.2 Natural emissions

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312 Several natural emissions inventories and parameterizations are used in our study. 313 Emissions of dimethyl sulfide (DMS) are based on the seawater DMS climatology of 314 Lana et al. (2011) with modifications for the Canadian Arctic Archipelago region as 315 described in Mungall et al. (2016), who found that the climatology seawater DMS was 316 biased low relative to observations from summer 2014. The air-water transfer velocities 317 for DMS are based on the scheme of Johnson (2010). Natural sources of  $NH_{3}$ , along with 318 biofuel and anthropogenic sources are from the Global Emissions InitiAtive (GEIA) 319 (Bouwman et al., 1997). For some of our simulations, as described in Sect. 2.3, Arctic 320 seabird colony NH<sub>3</sub> emissions are implemented following Riddick et al. (2012a), Riddick 321 et al (2012b), Croft et al. (2016a) and Wentworth et al. (2016). Our simulations also implement an NH<sub>3</sub> source from Arctic tundra with a fixed emission rate of 2.2 ng  $m^{-2} s^{-1}$ . 322 323 an upper estimate based on inferred bi-directional exchange fluxes calculated using soil 324 measurements made during the summer of 2016 near Alert, which found the tundra can 325 act as a source of  $NH_3$  to the atmosphere (Murphy et al. (in prep)). Given the uncertainty in the tundra source, this source can be viewed as a surrogate for the missing emissions 326 327 needed to bring the simulated NH<sub>3</sub> mixing ratios to agreement with measurements as





discussed in Sect. 3.1. Additionally, natural sources of NH<sub>3</sub> and organic carbon (OC) aerosol are included in the biomass burning emissions from the 3-hourly Global Fire Emissions Database, version 4 (GFED4) for 2016 (Giglio et al., 2013; Van Der Werf et al., 2017), which is employed in all simulations. Dust emissions employ the Dust Entrainment and Deposition (DEAD) scheme of Zender et al. (2003), developed in GEOS-Chem by Fairlie et al. (2007).

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335 Emissions of sea spray in our simulations are based on the Mårtensson et al. (2003) parameterization. We view this parameterization as yielding an upper bound on ultrafine 336 337 primary marine emissions. Comparisons with the Jaeglé et al. (2011) parameterization, 338 employed in the standard GEOS-Chem-TOMAS model, indicate that the Mårtensson et 339 al. (2003) parameterization yields greater sub-100 nm fluxes by up to a few orders of 340 magnitude (Jaeglé et al., 2011). This choice of emissions inventory enables evaluation of 341 the contribution of sea-spray to simulated ultrafine particle concentration under one of the 342 inventories that is most favorable to ultrafine sea-spray particle production. Additionally, 343 as opposed to assuming that all sea spray is sodium chloride, we emit sea spray particles 344 with diameters smaller than 100 nm as hydrophobic organic carbon aerosol and particles 345 larger than 100 nm as sodium chloride. This modification was introduced based on 346 measurements indicating that sub-100 nm sea spray particles are composed mostly of 347 organics, whereas larger particles have a progressively more dominant salt component 348 (Collins et al., 2013; Prather et al., 2013; Gantt and Meskhidze, 2013; Quinn et al., 2015). 349 However, knowledge gaps remain related to the spatial distribution of sea spray 350 composition and hygroscopicity (Collins et al., 2016).

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352 2.2.3 Anthropogenic emissions

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Our simulations also include global anthropogenic emissions from the Emissions
Database for Global Atmospheric Research
(http://edgar.jrc.ec.europa.eu/archived\_datasets.php) (EDGAR) inventory. The EDGAR
inventory is overwritten by regional inventories for Europe (European Monitoring and
Evaluation Program (EMEP) (Crippa et al., 2016)), Canada (Criteria Air Contaminant





- Inventory), the United States (National Emission Inventory (NEI)), and Asia (MIX
  inventory (Li et al., 2017)). As well, the Bond et al. (2007) inventory overwrites the
- 361 EDGAR fossil-fuel and biofuel emissions for black and organic carbon.
- 362
- 363 2.2.4 Chemical mechanism
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365 The GEOS-Chem-TOMAS chemical mechanism represents the reactions of more than 366 100 gas-phase species, including particle-relevant reactions such as DMS oxidation by 367 the hydroxyl radical (OH) to produce sulfur dioxide (SO<sub>2</sub>) by both the addition and 368 abstraction channels, and also reaction with the nitrate radical (NO<sub>3</sub>) (Chatfield and 369 Crutzen, 1990; Chin et al., 1996; Alexander et al., 2012). SO<sub>2</sub> then undergoes either gas-370 phase reactions with OH to produce  $H_2SO_4$  or aqueous oxidation with either hydrogen 371 peroxide (H<sub>2</sub>O<sub>2</sub>) or ozone (O<sub>3</sub>) to produce particulate sulfate. For the aerosol 372 microphysics scheme, gas-phase H<sub>2</sub>SO<sub>4</sub> can join with water vapor and gas-phase NH<sub>3</sub> for 373 ternary nucleation of nascent particles, and it can also condense to grow pre-existing 374 particles. The sulfate produced by aqueous-phase reactions is added to particles that are 375 large enough to have activated to form cloud droplets, only contributing to the growth of 376 these larger particles. In general, particles with diameters of 50 nm or larger activate in 377 our simulations, although observations from the Canadian Arctic Archipelago indicate 378 that particles as small as 20 nm could activate under some conditions (Leaitch et al., 379 2016). Methane sulfonic acid (MSA) that is produced by the DMS-OH-addition channel 380 contributes to condensational growth of existing particles, as described in Hodshire et al. 381 (in prep), but does not contribute to particle nucleation in our simulations. In this study, 382 we did not include additional chemistry related to production of dimethylsulfoxide 383 (DMSO). As a result, our simulated yields of MSA may be slightly underpredicted, and 384 sulfate formation slightly overpredicted (Breider et al., 2014).

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386 2.2.5 Secondary organic aerosol scheme

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Arctic MSOA is treated with the simplified SOA scheme developed by Kim et al. (2015), which for all simulations includes SOA precursors from non-marine sources associated





390 with terrestrial biogenic, fossil fuel, biofuel, and biomass burning emissions. The scheme 391 introduces two additional tracers, a gas-phase SOA precursor, and a SOA tracer that is 392 available to condense on the particles. The gas-phase SOA precursor oxidizes to form 393 SOA on a fixed timescale of 1-day. For biogenic sources the emissions are distributed 394 between these two tracers with a 50/50 split to represent the fast oxidation timescale of 395 biogenic precursors of typically shorter than 1 day. The model employed for this study 396 does not include aqueous-phase production of SOA. As a part of our sensitivity 397 simulations (described in Sect. 2.3), additional fixed-flux organic vapor emissions 398 yielding Arctic MSOA are implemented in the region north of 50° N over open seawater 399 with a 50/50 split between the gas-phase precursor and the phase that is available for 400 condensation.

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402 While our simulations do not include initial formation of nascent particles by clusters of 403 organic vapors, we do allow growth of particles by condensation of organic vapors, as 404 well as by condensation gas-phase  $H_2SO_4$  and MSA. In the standard model, all vapors 405 condense proportional to the Fuchs-corrected aerosol surface area distribution, behaving 406 like a non-volatile condensing gas (Donahue et al., 2011; Pierce et al., 2011; Riipinen et 407 al., 2011; Liu et al., 2016; Tröstl et al., 2016; Ye et al., 2016). The important role of 408 organic condensation was found at lower latitude continental sites (Riipinen et al., 2011). 409 Our simulations investigate this role for the Arctic. We also conduct additional sensitivity 410 simulations (described in Sect. 2.3), which allow condensation of a fraction of the vapors 411 according to the mass distribution, behaviour like a semi-volatile as opposed to nonvolatile condensing organic. 412

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414 2.2.6 Wet and dry deposition

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416 Removal of simulated aerosol mass and number occurs by both wet and dry deposition.

417 The wet deposition parameterization includes both in-cloud and below-cloud scavenging

418 as developed by Liu et al. (2001) and Wang et al. (2011), with modifications as described

419 in detail in Croft et al. (2016b) to more closely link the wet removal to the meteorological

420 fields for cloud liquid, ice water content, and cloud fraction. To represent the impact of





- 421 drizzle from summertime Arctic low-level clouds, we implemented wet removal from all Arctic clouds below 500 m using a fixed efficiency of 0.01 s<sup>-1</sup>, similar to the approach of 422 Browse et al. (2012). In-cloud wet removal in GEOS-Chem-TOMAS is limited to the size 423 424 range that can activate to form cloud hydrometeors. Size-dependent dry deposition uses 425 the resistance in series approach of Wesley (1989). Simulated gas-phase species are also 426 removed by dry and wet deposition as described in Amos et al. (2012). Removal depends 427 on solubility such that aerosol precursors including ammonia, and sulphur dioxide are 428 removed by precipitation, while SOA precursors and dimethyl sulphide are not.
- 429
- 430 2.2.7 Radiative calculations
- 431

432 The following radiative transfer calculations are conducted off-line using the simulated 433 summertime-mean aerosol mass and number concentrations to examine the effects of 434 organic condensation. For calculation of the direct radiative effect (DRE), aerosol optical 435 depth, single-scattering albedo and asymmetry factor are calculated with Mie code 436 (Bohren and Huffman, 1983) and use refractive indices from the Global Aerosol Dataset 437 (GADS) (Koepke et al., 1997). These optical properties, along with cloud fraction and 438 surface albedo from GEOS-FP assimilated meteorology are input to the Rapid Radiative 439 Transfer Model for Global Climate Models (RRTMG) (Iacono et al., 2008), to determine 440 the change in top-of-the-atmosphere solar flux between two simulations, treating all 441 particles except black carbon as internally mixed and spherical.

442

443 We also calculate the cloud-albedo aerosol indirect effect (AIE) with the method 444 described in Croft et al. (2016a) and Kodros et al. (2016). The cloud droplet number concentration (CDNC) is calculated off-line using the parameterization of Abdul-Razzak 445 446 and Ghan (2002), again using the summertime mean aerosol mass and number concentrations from our simulations. We assume an updraft velocity of 0.5 m s<sup>-1</sup> and treat 447 448 all aerosols as internally mixed to determine the hygroscopicity parameter of Petters and 449 Kreidenweis (2007). For each model grid box, we assume cloud droplet radii (r) of 10  $\mu$ m 450 and perturb this value with the ratio of summertime-mean CDNC from simulations





- 451 (acronyms described in Table 1 and simulations described in Sect. 2.3), following Rap et
- 452 al. (2013), Scott et al. (2014) and Kodros et al. (2016),
- 453

$$454 \quad r_{perturbed} = 10 \left( \frac{CDNC_{BASE+TUNDRA+BIRDS+100xnuc}}{CDNC_{BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv}} \right)^{1/3}. \tag{1}$$

455

456 Then RRTMG is used to determine the change in top-of-the-atmosphere solar flux 457 attributed to the change in effective cloud droplet radii, again using the summertime 458 meteorological data from GEOS-FP. Our AIE calculation is limited to this single aerosol 459 indirect effect; the impact of Arctic MSOA on additional indirect effects (Lohmann and 460 Feichter, 2005) requires further investigation in future studies.

461

## 462 **2.3 Overview of simulations**

463

464 Our simulations are conducted with a focus on interpreting the summertime 2016 aerosol 465 measurements from the Canadian Arctic Archipelago. These simulations are used to 466 explore the role of biogenic sources in shaping the aerosol size distributions by the 467 processes of nucleation of particles from gas-phase molecules followed by growth, with a 468 focus on Arctic MSOA. We also consider the role of marine primary particle emissions.

469

470 Table 1 presents simulation acronyms used in the following discussion. Simulation BASE 471 employs the standard GEOS-Chem-TOMAS model described in Sect. 2.2. We examine 472 the potential contribution of regional terrestrial NH<sub>3</sub> sources to aerosol size distributions 473 with simulations BASE+BIRDS, BASE+TUNDRA, and BASE+TUNDRA+BIRDS. 474 Simulation BASE+BIRDS implements seabird-colony NH<sub>3</sub> emissions of 35 Gg spread 475 uniformly in time between 1 May and 30 September, based on the Riddick et al. (2012a) 476 and Riddick et al. (2012b) inventory with modifications and spatial distribution as described in Croft et al. (2016a) and Wentworth et al. (2016). Simulation 477 BASE+TUNDRA includes a fixed NH<sub>3</sub> emission of 2.2 ng m<sup>-2</sup> s<sup>-1</sup> from all Arctic tundra 478 479 as discussed in Sect. 2.2.2. Simulation BASE+TUNDRA+BIRDS uses both the seabird 480 colony and tundra NH<sub>3</sub> sources.

481





482

483

484

485

Simulation Acronyms	Description
BASE	Base simulation, described in detail in Sect. 2.2.
BIRDS	Seabird-colony ammonia emissions included
TUNDRA	Tundra ammonia emissions included
AMSOAnv	Non-volatile Arctic MSOA
AMSOAnv/sv	30% non- and 70% semi-volatile Arctic MSOA
2xAMSOAnv/sv	Double organic vapor emissions of AMSOAnv/sv
100xnuc	Particle formation rate scaled by 100-fold

486

Table 1: Description of acronyms that are used in the simulation names. Simulations are described in detailby full simulation name in Sect. 2.3.

489 490

491 Simulation BASE+TUNDRA+BIRDS+AMSOAnv adds a source of Arctic MSOA. This 492 source draws upon measurements presented by Mungall et al. (2017) indicating that the marine microlaver is a source of oxygenated volatile organic compounds (OVOCs), key 493 494 precursors to secondary organic aerosol. Furthermore, Willis et al. (2017) identified a 495 positive relationship between the ratio of organic to sulfate aerosol mass concentrations 496 and time spent over open water, suggesting a marine SOA source. Studies from other 497 regions also identified biogenic VOCs of marine origin, but their marine sources are not 498 fully identified (Carpenter and Nightingale, 2015; Tokarek et al., 2017; Chiu et al., 2017). 499 Likewise for the Arctic, the emission rates for these vapors are not well understood 500 (Burkart et al., 2017a). Given this uncertainty, we identified a fixed Arctic MSOA-501 precursor vapor source flux (Arctic MSOA formed with a mass yield of unity) from the 502 ice-free seawater in the Arctic and near Arctic (north of 50° N). The marine organic 503 vapors are emitted into the simplified SOA scheme described in Sect. 2.2.5 with a 50/50 504 split between precursor gas (with a 1-day lifetime before forming SOA) and vapor that 505 instantly condenses to particles in the marine boundary layer. This split represents 506 variability in the VOCs lifetime for aging to sufficiently low volatility that they can 507 condense on the particulate phase. At the point of condensation, we assume the vapors to 508 be effectively non-volatile, with condensation according to the Fuchs-corrected surface area. A top-down estimate of the flux (500  $\mu$ g m<sup>-2</sup> d<sup>-1</sup>; simulated flux magnitude of 468  $\mu$ g 509





510  $m^{-2} d^{-1}$  north of 50° N, reported to one significant figure due to uncertainty) is adopted 511 that best represents observations as shown in the following sections within the context of 512 our simulations, which are an upper limit on primary organic aerosol contribution. To put 513 the implemented flux in context, this is within an order of magnitude of the isoprene flux 514 estimated from a north temperate deep lake (Steinke et al., 2018). Future work should 515 include a bottom-up estimate of the SOA-precursor source flux.

516

Simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv examines the impact of 517 518 particle precursors in addition to H<sub>2</sub>SO<sub>4</sub>, NH<sub>3</sub> and water that could nucleate nascent 519 particle clusters in the Arctic. These precursors could include (but are not limited to) gas-520 phase iodine (Allan et al., 2015; Dall'Osto et al., 2018a), amines (Almeida et al., 2013) 521 and organics (Riccobono et al., 2014). It is unclear if marine biological activity creates 522 organic vapors that participate in particle nucleation. Disintegration of larger particles 523 from evaporating clouds and fog could contribute to the number of nascent particles 524 (Leck and Bigg. 2010). Unfortunately, a nucleation parameterization does not exist that is 525 suitable to include interactions of all these materials simultaneously (Riccobono et al., 526 2014; Dunne et al., 2016; Gordon et al., 2017). To explore these effects, we scale up the 527 ternary nucleation by 100-fold to represent the potential effects of particle precursors 528 with similar spatial origin to those involved in ternary nucleation. Almeida et al. (2013) 529 and Riccobono et al. (2014) observed increases in nucleation rates by about 100-fold 530 above the sulfate-ammonia-water vapor system when amines or monoterpene-oxidation 531 products were added. We treat the 100-fold increase as an estimate of how additional 532 nucleation. We materials could enhance also conduct simulation 533 BASE+TUNDRA+BIRDS+100xnuc, which is otherwise similar to 534 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv, but without the condensable marine 535 organic vapors.

536

Finally, we conduct simulations to examine the impact of Arctic MSOA volatility.
Burkart et al. (2017a) found that condensing gas-phase materials in the Canadian Arctic
Archipelago were surprisingly more volatile than at lower latitudes. Simulation
BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv is similar to simulation

18





541	BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv, except that 30% of the Arctic MSOA
542	behaves as non-volatile compounds and condenses according to Fuchs-corrected surface
543	area, whereas 70% of the Arctic MSOA behaves as idealized semi-volatile compounds
544	and condenses according to the particle mass distribution (quasi-equilibrium
545	condensation). This is a larger fraction of semi-volatile vapors than the 50/50 semi-/non-
546	volatile split employed by Riipinen et al. (2011) for lower latitude continental sites, and
547	consistent with the findings of Burkart et al. (2017a) for the Canadian Archipelago region
548	that the condensing vapors were more semi-volatile than at lower latitudes. Simulation
549	BASE+TUNDRA+BIRDS+100xnuc+2xAMSOAnv/sv is similar to simulation
550	BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv, except that for the former, we
551	double the flux of marine organic vapors to examine the impact of a change in flux since
552	the geographic distribution of the source rate is highly uncertain.

553

554 We calculate the mean fraction error (MFE) (Boylan and Russell, 2006) between our 555 simulations and observed aerosol size distributions using the following equation.

556

557 MFE = 
$$\frac{1}{N} \sum_{i=0}^{i=N} \frac{abs|C_m(i) - C_o(i)|}{(C_m(i) + C_o(i))/2}$$
 (2)

558

where  $C_m(i)$  is the integrated value for the *i*<sup>th</sup> moment of the simulated aerosol size distribution and  $C_o(i)$  is the integrated value for the *i*<sup>th</sup> moment of the observed size distribution, for the *N* moments (zero<sup>th</sup> to third). Fractional bias is calculated likewise, but without the absolute value of the numerator.

563

## 564 **3. Results and Discussion**

565

## 566 **3.1 Summertime aerosol mass concentrations at Alert**

567

Table 2 shows the observed climatological- (data available for years 2011 to 2015), summertime- (July and August) mean mass concentrations for organic carbon (OC),  $NH_4^+$ , sulfate ( $SO_4^=$ ) and MSA from filter measurements at Alert and also the summertime mean for our 9 simulations (all for 2016). These long-term observations at





572 Alert provide a useful climatology for comparison with our simulations, even if not for 573 the same years. An organic matter (OM) to OC ratio of 2.1 was applied to convert the 574 simulated OM to OC. There is uncertainty in the OM/OC ratio, which can range from about 1.4 to 2.4; however, the larger values are generally associated with rural and remote 575 576 regions (Turpin and Lim, 2001; Russell, 2003; Philip et al., 2014). This ratio (2.1) is 577 within 4% of summertime-mean OM/OC ratios from the Organic Functional Groups 578 (OFG) measurements from available years (2012-2014) at Alert that are detailed in 579 Leaitch et al. (2018). Simulation BASE underpredicts the observed annual mean OC 580 mass concentrations by a factor of about 3, and is below the inter-annual range shown in 581 Table 2. The simulated OC is strongly sensitive to Arctic MSOA, increasing by a factor 582 of 2.5 for simulation BASE+TUNDRA+BIRDS+AMSOAnv relative to simulation 583 BASE+TUNDRA+BIRDS, to yield better consistency with the measurement mean. As 584 described in Leaitch et al. (2018), these measurements could also include organics from 585 condensation of volatile organic compounds directly onto the quartz filters, contributing 586 to a factor of 1.1-2.5 increase in concentrations relative to OFG analysis by Fourier 587 transform infrared (FTIR) spectroscopy. This is only one of the potential issues with 588 determining OC based on sampling at such low concentrations over one week. 589 Furthermore, the measurement OC mass concentrations were not available for the year of 590 our simulations, and inter-annual variability in summertime OC is large. As a result of 591 these complexities, we also compare between the year-matched simulated and SMPS 592 volume distributions (Sect. 3.3), which we consider to be a more robust evaluation of the 593 simulated particle mass concentrations.

594

595 The simulated OC mass concentration is also sensitive to the rate of formation of nascent

596 particles in our simulations. With an increase in nucleation rate, as in simulation,

597 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv relative to simulation

598 BASE+TUNDRA+BIRDS+AMSOAnv, the simulated OC mass concentrations increase

599 by about 15% because there are more nucleation- and Aitken-mode particles available to

- 600 'catch' the condensing vapors, and these smaller particles are less efficiently removed by
- 601 wet scavenging. Additionally, the simulations of OC mass concentrations are sensitive to
- the assumed volatility of the Arctic MSOA. For the 30/70 split of non-volatile and semi-





- 603 volatile vapors as in simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv, the
- 604 OC mass concentrations are reduced by about 15% relative to simulation
- 605 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv. The semi-volatile vapors condense to
- larger particles (Pierce et al., 2011; D'Andrea et al., 2013; Hodshire et al. (in prep.)) that
- are more efficiently removed by precipitation. The emissions flux of Arctic MSOA-
- 608 precursor
- vapors has a nonlinear impact on the simulated OC mass concentrations. Table 2 shows
- 610 that the change in mean OC concentrations differs between simulation pairs with the
- 611 same absolute change in Arctic MSOA-precursor emissions (simulations
- 612BASE+TUNDRA+BIRDS+100xnuc+2xAMSOAnvand613BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv,relativeto
- 614 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv and
- 615 BASE+TUNDRA+BIRDS+100xnuc).
- 616

	OC	$\mathrm{NH_4}^+$	$MSA + SO_4^{=}$
Observations	85.0	16.3	58.8
	(37.7, 132.9)	(7.6, 31.2)	(30.8, 97.5)
Simulations			
BASE	28.7	2.8	54.8
Extra Ammonia			
BASE+BIRDS	27.9	12.6	56.6
BASE+TUNDRA	27.0	20.0	58.5
BASE+TUNDRA+BIRDS	33.3	22.6	66.2
Arctic MSOA (non-volatile)			
BASE+TUNDRA+BIRDS+AMSOAnv	72.0	22.0	64.8
Extra Nucleation			
BASE+TUNDRA+BIRDS+100xnuc	33.2	23.5	69.2
BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv	82.0	24.1	72.1
Arctic MSOA Volatility (mix non-/semi-volatile)			
BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv	70.2	23.6	69.7
BASE+TUNDRA+BIRDS+100xnuc+2xAMSOAnv/sv	104.6	23.4	68.7

617

- 622
- 623

624 We also evaluated the simulation of relevant inorganic ions. Simulation BASE strongly

625 underpredicts measured ammonium concentrations by a factor of 8. Implementation of

<sup>618</sup>Table 2: July and August mean aerosol mass concentrations (ng m³) at Alert from measurements (2011-6192015) (brackets show the minimum and maximum annual mean over the measurement period) and GEOS-620Chem-TOMAS simulations (2016) as described in Table 1 and Sect. 2.3 for organic carbon (OC),621ammonium (NH4<sup>+</sup>), and methane sulfonic acid (MSA) and sulfate (SO4<sup>=</sup>).





626 the seabird-colony and tundra NH<sub>3</sub> sources (simulation BASE+TUNDRA+BIRDS) yields 627 closer model-measurement agreement (within about 40%), although this is a challenging 628 simulation given the uncertainties in the NH<sub>3</sub> sources within these remote regions. 629 Relative to measurements taken during the summer 2016 cruise track, simulation BASE 630 also underpredicts grid-box mean NH<sub>3</sub> mixing ratios with a mean fractional bias (MFB, 631 described in Sect. 2.3) (Boylan and Russell, 2006) of -1.98, which is reduced for 632 simulations **BASE+BIRDS** (-1.23),BASE+TUNDRA (-0.22),and 633 BASE+TUNDRA+BIRDS (+0.06).

634

635 Implementation seabird-colony of and tundra NH<sub>3</sub> sources (simulation 636 BASE+TUNDRA+BIRDS) also increases the simulated summertime (July and August) 637 mean sulfate concentrations by about 20% over simulation BASE. The extra free NH<sub>3</sub> 638 (simulation BASE+TUNDRA+BIRDS relative to BASE) promotes particle nucleation, 639 which decreases the mean diameter of the particles available for H<sub>2</sub>SO<sub>4</sub> condensation as 640 sulfate, slowing sulfate wet scavenging as the smaller particles have less efficient 641 removal. Scaling up the nucleation rate further increases the sulfate concentrations by 642 about 10-15% as the concentrations of nucleation- and Aitken-mode particles are further 643 increased, further reducing the mean diameter of the particles available for condensation 644 of H<sub>2</sub>SO<sub>4</sub>, and slowing wet scavenging of sulfate. All simulations are within 20% of the measured sulfate concentrations. As well, we evaluated the model-measurement 645 646 agreement for sodium ions and found agreement within a factor of two for all 647 simulations. These mass-based comparisons offer confidence that the simulations which 648 include Arctic MSOA are reasonable. In the following sections, we consider the impact 649 of Arctic MSOA on total aerosol number concentrations and the moments of the aerosol 650 size distributions.

651

# 3.2 Total aerosol number concentrations along the 2016 ship track and at Alert

Figure 2 shows time series measurements during August 2016 of total particle number concentration for particles with diameters larger than 4 nm conducted from the *CCGS Amundsen* (Collins et al., 2017) and for particles with diameters larger than 10 nm at





Alert. The measurement time series shows episodic bursts of particle number concentration greater than 500 cm<sup>-3</sup>. Interestingly, at several times the elevated number concentrations occur at both Alert and at the ship, such as on August 3, 8, 9, 10, 11, 15 and 16. Both time series also show relatively lower total number concentrations on August 5 and August 13 to 14, when there was precipitation, and when the ship was in the Nares Strait region shown on Fig.1.



- 663
- 664

Figure 2: Time series for August 2016 observed number concentration from condensation particle counter
(CPC) for aerosols with (a) diameters larger than 4 nm (N4) along Amundsen ship track (Fig. 1) and (b)
diameters larger than 10 nm (N10) at Alert (described in Sect. 2.1) and for the simulations as described in
Table 1 and Sect. 2.3 (color coded as shown on legend). MFE: mean fractional error between observations
and simulations, color-coded to match simulation names.

670

Figure 2 also shows the time series of coincidently sampled simulated number concentrations for five of the simulations described in Table 1 and Sect. 2.3. We calculated the model-to-measurement mean fractional error (MFE) (Eq. (2)) for the





674 simulations of total number concentration shown on Fig. 2. The BASE simulation is 675 associated with the greatest MFE values for the ship track (1.44) and Alert (1.84). The 676 simulations better capture the bursts of particle number when including NH<sub>3</sub> sources from 677 seabird colonies and tundra, similar to the findings of Croft et al. (2016a) with MFE 678 values reduced to 0.75 for both the Alert and ship track time series for simulation 679 BASE+TUNDRA+BIRDS.

680

681 Implementation of Arctic MSOA in simulation BASE+TUNDRA+BIRDS+AMSOAnv 682 increases the MFE relative to simulation BASE+TUNDRA+BIRDS, to 1.06 for the ship 683 track and 0.96 for the Alert time series. This MFE increase occurs because more vapors 684 are available to condense on to the particle surface area, building the condensation sink 685 for  $H_2SO_4$ , which reduces the simulated formation of nascent particles by ternary nucleation with H<sub>2</sub>SO<sub>4</sub>. These effects reduce the number of ultrafine particles, similar to 686 that described by D'Andrea et al. (2013), who investigated the impact of 687 688 anthropogenically enhanced SOA for a set of sites distributed around the world.

689

Scaling the nucleation rate by 100-fold reduces the MFE to 0.87 for the ship track and 0.64 at Alert, for simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv. This scaling acts as a surrogate for the parameterization of particle nucleation by materials in addition to the simulated gas-phase  $NH_3$ ,  $H_2SO_4$  and water, as described in Sect. 2.3. This increased nucleation rate enables ultrafine particles to become more numerous, despite the increased condensation sink associated with the implemented Arctic MSOA source.

696

697 Considering both the 2016 ship track and Alert time series together, the lowest MFE pair 698 is found for simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv, 0.72 and 699 0.65, respectively. This simulation treats the Arctic MSOA as a 30/70 mix of non- and 700 semi-volatile species. Higher volatility condensing vapors enable simulated growth of the 701 nascent particles at a slower rate (relative to larger particles) than lower volatility 702 condensing vapors. As a result, the newly formed particles grow more slowly with semi-703 volatile Arctic MSOA, which lowers the condensation and coagulation sinks (relative to 704 non-volatile Arctic MSOA), and increases the total number of particles. As well, there is





relatively more condensation of the semi-volatile Arctic MSOA to larger particles, which contribute proportionately less to surface area and more to aerosol mass. These larger particles are efficiently removed by the frequent low-cloud drizzle of the summertime Arctic in our simulations. As shown on Fig. 2, the net effect is an increase in the number of ultrafine particles that better matches the observed time series of total number concentration.

711

## 712 **3.3** Moments of the aerosol size distribution for Alert, Eureka and ship track

713

Figures 3, 4 and 5 show the 2016 summertime (July and August) median aerosol size distributions from SMPS measurements at Alert (Fig. 3), Eureka (Fig. 4), and for the 2016 ship track (Fig. 5). The figure panels show the zero<sup>th</sup> through third moments of the aerosol size distribution, aerosol number, integrated diameter (length), surface area and volume. We define the aerosol number distribution (zero<sup>th</sup> moment) as

719

720 
$$n_N(D_p) = \frac{dN}{dlog_{10}D_p}.$$
 (3)

721

The aerosol integrated diameter (length) distribution (first moment) is

723

724 
$$n_D(D_p) = \frac{dD}{dlog_{10}D_p} = D \frac{dN}{dlog_{10}D_p}.$$
 (4)

725

The aerosol surface area (second moment) is

727

728 
$$n_{S}(D_{p}) = \frac{dS}{dlog_{10}D_{p}} = \pi D^{2} \frac{dN}{dlog_{10}D_{p}}.$$
 (5)

730 The aerosol volume (third moment) is

731

732 
$$n_V(D_p) = \frac{dV}{dlog_{10}D_p} = \frac{\pi}{6}D^3 \frac{dN}{dlog_{10}D_p}.$$
 (6)

733









Figure 3: July and August 2016 median aerosol size distributions from scanning mobility particle sizer (SMPS) measurements at Alert (82.5° N, 62.3° W) (black) (described in Sect. 2.1) and for five GEOS-Chem-TOMAS simulations (color coded as shown on legend). Grey shading shows SMPS 20<sup>th</sup> to 80<sup>th</sup> percentile. Simulations are described in Table 1 and Sect. 2.3. Panels show aerosol distribution moments
(a) aerosol number, (b) integrated aerosol diameter, (c) aerosol surface area, and (d) aerosol volume distributions. Note the different vertical scale relative to Figs. 4 and 5.

742 743

The observed distributions are similar between the three measurement sets. The number

745 distributions peak in the Aitken mode at the particle diameter of 30-50 nm, which is









Figure 4: July and August 2016 median aerosol size distributions from scanning mobility particle sizer (SMPS) measurements at Eureka (80.1° N, 86.4° W) (black) (described in Sect. 2.1) and for five GEOS-Chem-TOMAS simulations (color coded as shown on legend). Grey shading shows SMPS 20<sup>th</sup> to 80<sup>th</sup> percentile. Simulations are described in Table 1 and Sect. 2.3. Panels show aerosol distribution moments (a) aerosol number, (b) integrated aerosol diameter, (c) aerosol surface area, and (d) aerosol volume distributions. Note the different vertical scale relative to Figs. 3 and 5.

similar to summertime observations at other pan-Arctic sites (Tunved et al., 2013; Asmi

et al., 2016; Nguyen et al., 2016; Freud et al., 2017; Gunsch et al., 2017; Heintzenberg et al., 2017; Kolesar et al., 2017) and also in the central Arctic marine boundary layer (Heintzenberg and Leck, 2012; Karl et al., 2013; Heintzenberg et al., 2015). Interestingly, the value for the mode for the number distributions ( $dN/dlogD_p$ ) has its smallest magnitude of about 200 cm<sup>-3</sup> at the most northerly site (Alert), and increases moving southward to about 300 cm<sup>-3</sup> at Eureka and 400 cm<sup>-3</sup> for the ship track, which includes the









766 Figure 5: July and August 2016 median aerosol size distributions from scanning mobility particle sizer 767 (SMPS) measurements for the CCGS Amundsen 2016 ship track (black) (described in Sect. 2.1) and for five 768 GEOS-Chem-TOMAS simulations (color coded as shown on legend). Grey shading shows SMPS 20th to 769 80<sup>th</sup> percentile. Simulations are described in Table 1 and Sect. 2.3. Panels show aerosol distribution 770 moments (a) aerosol number, (b) integrated aerosol diameter, (c) aerosol surface area, and (d) aerosol 771 volume distributions. Note the different vertical scale relative to Figs. 3 and 4.

772 773

774 most southward extent. This pattern is consistent with the hypothesis of an important role 775 for open water in building summertime aerosol size distributions (Heintzenberg et al., 776 2015; Willis et al., 2017; Dall'Osto et al., 2018b). A similar pattern is noted for the other 777 three moments of the aerosol distribution.

778

779 Figures 3, 4 and 5 also show the simulated moments for the 3 sets of aerosol 780 distributions. Simulation BASE strongly underpredicts all four moments of the 781 distribution relative to all three of the measurement sets. Table 3 shows the MFE (Eq. (2))





between the simulations and measurements, using integrated values from the 4 moments of the distributions, similar to the approach employed by Hodshire et al. (2018). The MFEs are 1.17, 1.36 and 1.34 for Alert, Eureka, and the ship track, respectively, for simulation BASE. Implementation of sources of NH<sub>3</sub> from seabird-colonies (simulation BASE+BIRDS) reduces the MFE for all sites, and additional NH<sub>3</sub> from a tundra source for simulation BASE+TUNDRA+BIRDS further lowers the MFE at all sites (0.53 for Alert, 0.80 for Eureka and 0.97 for the ship track).

- 789 790
- Ship **Mean Fractional Error** Eureka Alert 3-site mean BASE 1.36 1.17 1.29 1.34 Extra Ammonia BASE+BIRDS 1.16 1.13 0.75 1.01 **BASE+TUNDRA** 1.01 0.86 0.66 0.84 BASE+TUNDRA+BIRDS 0.97 0.80 0.53 0.77 Arctic MSOA (non-volatile) BASE+TUNDRA+BIRDS+AMSOAnv 0.43 0.35 0.13 0.30 Extra Nucleation BASE+TUNDRA+BIRDS+100xnuc 0.78 0.30 0.31 0.46 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv 0.22 0.08 0.30 0.20 Arctic MSOA volatility (mix non-/semi-volatile) BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv 0.11 0.24 0.10 0.15 BASE+TUNDRA+BIRDS+100xnuc+2xAMSOAnv/sv 0.22 0.09 0.27 0.19

797 798

799 Figures 3-5 also show that with the NH<sub>3</sub> from seabird colonies and tundra (simulation 800 BASE+TUNDRA+BIRDS), an Aiken mode peak develops around 20-30 nm, but there is an underprediction of the number of aerosols with diameters between 30 nm to 200 nm 801 802 and a strong underprediction of the aerosol diameter, surface area and volume moments. 803 Simulation BASE+TUNDRA+BIRDS also underpredicts the organic carbon mass 804 concentrations as shown in Table 2. This simulation suggests that condensation of  $H_2SO_4$ and MSA alone do not yield sufficient particle growth to match observations from the 805 806 Canadian Arctic Archipelago, which show frequent particle growth events (Willis et al., 807 2016; Collins et al., 2017; Burkart et al., 2017b; Tremblay et al., 2018) and suggest a key

<sup>791</sup> 792

<sup>Table 3: Mean fractional error (MFE) (Eq. (2)) between the nine GEOS-Chem-TOMAS simulations
(described in Table 1 and Sect. 2.3) and the SMPS measurements (described in Sect. 2.1) for summertime(July and August 2016) median aerosol size distributions at Alert, Eureka and during the CCGS Amundsen
cruise shown in Figs. 3, 4, and 5, respectively.</sup> 





role for growth by organic vapor condensation (Burkart et al., 2017a; Willis et al., 2017;
Mungall et al., 2017). Marine primary organic aerosols could contribute to the Aitken
mode as investigated further in the following Sect. 3.5.

811

812 With the implementation of Arctic MSOA (simulation 813 BASE+TUNDRA+BIRDS+AMSOAnv), all four moments of the simulated aerosol 814 distributions are more consistent with the measurements. The MFE is reduced for the ship 815 track (0.43), Eureka (0.35) and Alert (0.13). These additional vapors condense on the 816 simulated particles and build the aerosol diameter, surface area, and volume distributions 817 to better represent the observations. For the ship track and at Eureka, scaling up the 818 reduces the MFE nucleation rate further (simulation 819 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv) by maintaining the number of 820 ultrafine particles despite the increase in the condensation sink that arises with the growth 821 from the Arctic MSOA. This scaling acts as a surrogate for nucleating vapors that could 822 be missing in our simulations such as iodine (Allan et al., 2015; Dall'Osto et al., 2018a) 823 and amines (Almeida et al., 2013), and also possible contribution from primary particle 824 fragmentation (Leck and Bigg, 2010). For Alert, the MFE deteriorates with nucleation 825 scaling suggesting that the standard ternary scheme yields sufficient particle formation 826 for that portion of the Canadian Arctic Archipelago under the assumption of growth by 827 non-volatile vapors.

828

829 The simulation with a 30/70 mix of non- and semi-volatile Arctic MSOA, respectively, 830 (simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv) yielded the lowest 831 MFE for the ship track (0.11) and for Alert (0.10). We find a similarly low MFE for 832 Eureka (0.09) with a doubling of the Arctic MSOA source under the assumption of a 30/70 833 mixed volatility (simulation 834 BASE+TUNDRA+BIRDS+100xnuc+2xAMSOAnv/sv). Given that our simulations 835 employ a fixed flux of Arctic MSOA-precursor vapors, and there is likely spatial 836 variability in their source flux and oxidative aging to become available for condensation 837 that is not well captured in our simulations; these intra-regional differences in the flux 838 strength required to match observations are expected. For the same reasons, the





simulations do not perfectly capture the increase in the magnitude of the mode for the
number, diameter, area and volume distributions between Alert and Eureka. However,
simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv (having lowest 3-sitemean MFE) does capture the larger magnitude of the mode value for all four moments for
the ship track relative to those for Alert and Eureka.

844

845 Our finding that a mixture of non- and semi-volatile Arctic MSOA gives a closer fit 846 between the simulations and observations is in agreement with the measurement-based 847 findings of Burkart et al. (2017a) that the condensing vapors were surprisingly more 848 volatile than at lower latitudes. As discussed by Burkart et al. (2017a), these semi-volatile 849 (as opposed to non-volatile) vapors enable slower growth of the smallest mode of 850 particles with diameters around 20 nm and faster growth of the larger mode with 851 diameters around 90 nm. This larger mode is more efficiently removed by precipitation, 852 maintaining a relatively pristine environment with lower particle mass concentrations that 853 favors particle formation and growth.

854

855 The general improvements of the simulations with the addition of Arctic MSOA offers 856 support for a key role of marine biogenic emissions in shaping the Arctic size 857 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv, distributions. Simulation 858 specifically, yielded the lowest MFE for the total number concentration time series 859 comparisons (Fig. 2), and had close agreement with the observed summertime mean 860 organic carbon mass concentration at Alert (within about 10%, Table 2). As discussed in 861 Sect. 3.1, several uncertainties affect the interpretation of the model-measurement 862 comparisons for the quartz filter OC mass concentrations. The model-measurement 863 fractional bias (FB) for the volume distribution (Table 4) provides a year-matched 864 constraint on the total aerosol mass concentrations in our simulations without the 865 complications associated with the quartz filter measurements such as the possibility of 866 VOC filters. Simulation uptake on the 867 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv has the lowest volume distribution 868 FB for both Alert (+0.07) and the ship track (+0.01), while for Eureka two simulations 869 had the lowest FB, BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv (-0.06) and





870 BASE+TUNDRA+BIRDS+100xnuc+2xAMSOAnv/sv (+0.06). For all three sites,

- 871 implementation of Arctic MSOA reduced the fractional bias for the volume distribution
- 872 relative to an otherwise similar simulation without Arctic MSOA. As shown in Table 2,
- 873 all simulations generally matched the mass of sulfate+ammonium+MSA (and
- 874 contributions of other measured species, e.g. nitrate, were minor) so organic aerosol is
- 875 likely the most uncertain species.
- 876 877
- **Fractional Bias** Ship Eureka Alert Base -0.83 -0.87 -0.54 Extra Ammonia BASE+BIRDS -0.83 -0.84 -0.50 BASE+TUNDRA -0.84 -0.79 -0.45 BASE+TUNDRA+BIRDS -0.79-0.74-0.38 Arctic MSOA (non-volatile) BASE+TUNDRA+BIRDS+AMSOAnv +0.25-0.20 +0.19Extra Nucleation BASE+TUNDRA+BIRDS+100xnuc -0.75 -0.69 -0.32 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv +0.36-0.06 +0.31Arctic MSOA Volatility (mix non-/semi-volatile) BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv +0.01-0.22 +0.07BASE+TUNDRA+BIRDS+100xnuc+2xAMSOAnv/sv +0.38+0.06+0.38
- 878
- 879

Table 4: Fractional bias (as defined in Sect. 2.3) between the nine GEOS-Chem-TOMAS
simulations (described in Table 1 and Sect. 2.3) and the SMPS volume distribution
measurements (described in Sect. 2.1) for summertime- (July and August 2016) median
aerosol size distributions at Alert, Eureka and during the *CCGS Amundsen* cruise shown
in panel (d) on Figs. 3, 4, and 5, respectively

886

# 887 **3.4 Role of Arctic MSOA during a growth event in Canadian Arctic Archipelago**

888

Figure 6 provides an example of a particle growth event from the summer 2016 *CCGS Amundsen* ship track through the Canadian Arctic Archipelago. The observations during 14-15 August 2016 show growth of particles from about 15 nm to about 35 nm over a period of about 10 hours. Collins et al., 2017 and Burkart et al. (2017a) also report growth rates of about 2-4 nm h<sup>-1</sup> for similar size aerosols during other growth events observed from the *CCGS Amundsen* during the 2016 cruise. The top right panel shows





895	that	without	the	source	of	Arctic	MSO	A (s	imulation
896	BASE+T	UNDRA+BI	RDS+100	)xnuc), th	e nascent	particles	do not	exhibit	sufficient

- growth beyond about 15 nm by condensation of  $H_2SO_4$  and MSA alone. The bottom left
- 898 panel shows that with the source of non-volatile



899 900

Figure 6: Time series of size-resolved aerosol number distributions (color contours show dN/dlogD<sub>p</sub>) for the growth event of 14-15 August 2016 as (a) observed along the Amundsen ship track (described in Sect. 2.1) and for the GEOS-Chem-TOMAS simulations along the ship track

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904 (b) BASE+TUNDRA+BIRDS+100xnuc, (c) BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv and
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```
    905 (d) BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv. Simulations are described in Table 1 and Sect.
    906 2.3.
```

907 908

```
Arctic MSOA for simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv, there is
growth from about 10 nm to about 50 nm over 10 hours, a growth rate that is faster than
observed for this event and faster than reported by Burkart et al., 2017a.
```

913 The bottom right panel of Fig. 6 shows for simulation
914 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv, particles grow from about 10 nm to





915 20 nm over about 8 hours, which is closer to the observed rate and slower than the 916 simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv, which assumed non-917 volatile Arctic MSOA. Semi-volatile Arctic MSOA also enables faster growth of the 918 larger mode around 90 nm, in agreement with the observations of Burkart et al. (2017a) 919 that the larger mode grew faster. We note that the timing of the simulated event shown in 920 Fig. 6 is not a perfect match to the observations. Capture of the timing of individual 921 growth events is challenging for global models since sub-grid scale effects, including 922 spatial variations in the meteorological fields over the time frame of few hours, such as 923 cloud cover are not perfectly captured for the measurement site by the model resolution. 924 However, similar to our findings from the previous sections, a source of mixed non- and 925 semi-volatile Arctic MSOA also yields closer agreement with these measurements from 926 the Canadian Arctic Archipelago during the summer of 2016. This key role for semi-927 volatile Arctic MSOA during the frequent summertime growth events in the Canadian 928 Arctic Archipelago is consistent with measurement-based studies for this region (Willis et 929 al., 2017; Collins et al., 2017; Burkart et al., 2017a; Leaitch et al., 2018; Tremblay et al., 930 2018).

931

### 932 3.5 Size-resolved aerosol composition

933

934 Few measurements are available of the composition of the summertime Arctic Aitken 935 mode due to insufficient instrument detection limits to detect the extremely low mass concentrations in this size range (less than 100 ng m<sup>-3</sup>). However the limited information 936 937 available does provide insight into the processes that shape the size distribution. For 938 example, Giamarelou et al. (2016) found that sub-12 nm particles in the Svalbard region 939 were primarily ammoniated sulfates, pointing to the importance of particle formation by 940 ternary nucleation of gas-phase NH<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub> and water and initial growth by lower 941 volatility sulfur-containing vapors.

942

Figure 7 shows the size-resolved mass fractions for the various aerosol components for
simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv. For the simulated sub10 nm particles, the simulated summertime (July and August) mean mass fractions at





946 Alert, Eureka and for the ship track are primarily biogenic sulfate and MSA, which arise 947 from oxidation of DMS, which is released to the atmosphere by marine biological 948 activity. Thus, the simulated composition exhibits similarities with the Svalbard 949 measurements, with the additional identification of a biogenic source. Figure 7 is also 950 consistent with the strong summertime biogenic sulfate component observed in the 951 Canadian Arctic Archipelago by Ghahremaninezhad et al. (2016).

952



954

955Figure 7: Simulated summertime- (July and August 2016) mean size-resolved aerosol component mass956fractions for (a) Alert, (b) Amundsen ship track and (c) Eureka, for the simulation957BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv as described in Table 1 and Sect. 2.3. Other organics958includes all organic aerosol except the Arctic marine secondary organic aerosol (AMSOA). Biogenic959sulfate includes all sulfate derived from the oxidation of dimethyl sulfide (DMS).

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- 961
- 962
- 963





964 Limited measurements of the composition of particles with diameters between 60 to 100 965 nm during growth events at Eureka show that these particles are almost entirely 966 composed of organic compounds, which could also include a minor contribution from 967 MSA (Tremblay et al., 2018). Unfortunately, these measurements were limited to a few 968 growth events and cannot be directly compared with the simulated summertime mean 969 mass fractions shown in Fig. 7. Burkart et al. (2017a) calculated a cloud condensation 970 nuclei (CCN) hygroscopicity parameter (Petters and Kreidenweis, 2007) for the particles 971 during a growth event in the Canadian Arctic Archipelago and found a value also 972 indicating a mostly organic composition for those particles large enough to act as CCN. 973 Figure 7 shows that our simulation captures an increasing contribution of organics with 974 particle diameters towards 50-100 nm (sizes that can act as CCN), reflecting the key role 975 of organics in growth of particles towards sizes that can be climate-relevant by acting as 976 seeds for cloud droplet formation, or directly scattering and absorbing radiation 977 (diameters larger than about 100 nm). Semi-volatile organic vapors have also been shown 978 to have a role in growth of particles after they reach diameters of about 5 nm (Tröstl et 979 al., 2016). However, as noted by Karl et al. (2013) lower volatility vapors are needed for 980 initial growth over the first few nm. Thus, semi-volatile organic vapors are likely only 981 important in later growth beyond 10-20 nm.

982

983 Figure 7 shows that the simulated contribution of organics is greatest for the ship track, 984 reflecting the marine source of the condensable organics in our simulation. The ship track 985 also has the strongest contribution of 'other organics' in the sub-100 nm range, with a 986 peak contribution for particle diameters of 10-30 nm. This sub-100 nm organic 987 contribution (shaded in dark green on Fig. 7) represents the mass-fraction contribution of 988 primary marine organics in our simulation. These primary particles also grow to larger 989 sizes by condensation of Arctic MSOA in our simulations. As described in Sect. 2.2, all 990 sea spray emissions with diameters smaller than 100 nm are treated as hydrophobic 991 organic carbon. We use the Mårtensson et al. (2003) parameterization, which can be 992 considered as the upper limit of sea spray emissions for the sub-100 nm range, relative to 993 other parameterizations. As a result, we consider that our simulation provides an upper 994 limit on the mass fraction of marine primary organic aerosol in the Aitken mode in the





995

996 boundary layer, is climate-relevant as it grows by condensation of Arctic MSOA towards 997 sizes of 50 nm to 100 nm. The dark green shading ('other organics') on Fig. 7 for sizes 998 larger than 100 nm represents contributions to the mass fractions by organics that have 999 been transported from lower latitudes, including those primary and secondary aerosols 1000 from biomass burning and other non-marine lower-latitude sources. 1001 1002 3.6 Impact of Arctic MSOA on climate-relevant aerosol number concentrations, 1003 direct and indirect radiative effects 1004 1005 In this section, we consider the role of Arctic MSOA on the simulated total number 1006 concentration of aerosols with diameter larger than 50 nm (N50) and 100 nm (N100) and 1007 the associated radiative effects using our simulation with the lowest overall model-1008 measurement MFE (simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv) 1009 relative to the simulation without Arctic MSOA (BASE+TUNDRA+BIRDS+100xnuc). 1010 Figure 8 shows the pan-Arctic distribution of the simulated summertime- (July and 1011 August) mean surface-layer N50 and N100 for simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv. In the Canadian Arctic Archipelago 1012 region, the simulated summertime-mean N50 (50 cm<sup>-3</sup> to 100 cm<sup>-3</sup>) and N100 (10 cm<sup>-3</sup> to 1013 1014 30 cm<sup>-3</sup>) ranges are consistent with monthly mean values from observations at Alert 1015 presented in Croft et al. (2016b). The panels in the middle column of Fig. 8 show that the 1016 of MSOA addition Arctic (simulation 1017 BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv relative to 1018 BASE+TUNDRA+BIRDS+100xnuc) yields a N50 increase of about 50-75 cm<sup>-3</sup> and a N100 increase of about 20 cm<sup>-3</sup> in the Canadian Arctic Archipelago. These differences in 1019 1020 the simulated N50 and N100 are attributed to the process of growth by condensation of 1021 Arctic MSOA, and will have climate-relevant impacts on aerosol radiative effects. 1022 1023 1024 1025

Canadian Arctic Archipelago region. The primary aerosol, particularly in the marine









1027

1028 Figure 8: Impact of Arctic MOSA, simulated summertime- (July and August 2016) mean geographic 1029 distribution of surface-layer aerosol number concentrations for (a) particles with diameters larger than 50 1030 nm (N50) for simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv, (b) surface-layer N50 1031 difference for simulation BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv relative to simulation 1032 BASE+TUNDRA+BIRDS+100xnuc, (c) aerosol indirect effect (AIE) at top of the atmosphere 1033 (methodology described in Sect. 2.2) between these two simulations, attributed to Arctic MSOA 1034 (AMSOA), (d) similar to a) but for N100, (e) similar to (b) but for N100 difference, (f) direct aerosol effect 1035 (DRE) at top of the atmosphere (methodology described in Sect. 2.2) between these two simulations, 1036 attributed to Arctic MSOA (AMSOA).

1037 1038

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1039
        Figure 8 also shows the geographic distribution of the top-of-the-atmosphere DRE and
1040
        cloud-albedo AIE (described in Sect. 2.2) for marine SOA (comparing between
1041
                                    BASE+TUNDRA+BIRDS+100xnuc
        simulations
                                                                                        and
        BASE+TUNDRA+BIRDS+100xnuc+AMSOAnv/sv). The pan-Arctic mean DRE
1042
1043
        attributed to condensational growth by Arctic MSOA is -0.04 W m<sup>-2</sup>. The simulated
        Arctic MSOA effect is largest (about -0.1 W m<sup>-2</sup> to -0.2 W m<sup>-2</sup>) over the regions of open
1044
1045
        water such as Baffin Bay, east of Greenland, and the Bering Sea. These are also regions
```





1046 of the largest N100 change since those particles with diameters larger than about 100 nm 1047 contribute strongly to scattering of solar radiation. Although our simulated Arctic 1048 MSOA-precursor vapor flux yielded particle distributions that were in close agreement 1049 with observations from the Canadian Arctic Archipelago (as shown in previous sections), 1050 there is considerable uncertainty about the Arctic MSOA source function such that 1051 terrestrial sources could also be missing from our simulations. These effects could change 1052 the spatial distribution and magnitude of the DRE. Further work is needed to better 1053 understand the source and nature of Arctic MSOA-precursor vapors in the Arctic to 1054 reduce uncertainties in the associated DRE.

1055

The pan-Arctic mean cloud-albedo AIE attributed to Arctic MSOA is about -0.4 W m<sup>-2</sup>. 1056 The AIE shows a similar geographic distribution to the changes in the N50, with largest 1057 values of -1 to -2 W m<sup>-2</sup> in the Canadian Arctic Archipelago and east of Greenland, again 1058 1059 related to the open water regions associated with the Arctic MSOA-precursor vapor flux 1060 implemented in our simulations. As with the DRE, there is a large uncertainty associated 1061 with the spatial distribution of this AIE due to the uncertainty about the source function 1062 for Arctic MSOA. However, we view this calculation as an indication that the impact of 1063 condensational growth by Arctic MSOA is expected to be relevant for the Arctic climate. 1064 Work is needed to better understand the composition and source flux of Arctic MSOA-1065 precursor vapors.

1066

# 1067 4. Conclusions

1068

1069 We used the GEOS-Chem-TOMAS chemical transport model with size-resolved aerosol 1070 microphysics to interpret measurements conducted during the summertime of 2016 in the 1071 Canadian Arctic Archipelago, some as part of the NETwork on Climate and Aerosols: 1072 addressing key uncertainties in Remote Canadian Environments (NETCARE) project 1073 (Abbatt et al. (in prep.)). Three measurement platforms were considered. These platforms 1074 were located at Alert and Eureka, both in Nunavut, Canada and also onboard the CCGS 1075 Amundsen. We focused on examining the key processes that build summertime aerosol 1076 size distributions in this region, particularly the role of Arctic marine secondary organic





aerosol (Arctic MSOA) condensation. The terminology Arctic MSOA was used to
indicate secondary organic aerosol formed from precursors from marine (open ocean and
coastal) sources north of 50° N, excluding MSA, which we treated as a separate aerosol
component. In the Canadian Arctic Archipelago, Arctic MSOA is likely strongly
controlled by emissions from marine biogenic activity (Willis et al., 2017; Leaitch et al.
2018).

1083

1084 We find that Arctic MSOA contributes strongly to the summertime particle size 1085 distributions in the Canadian Arctic Archipelago. Building on measurement-based studies 1086 from the NETCARE project, we implemented a flux of condensable Arctic MSOAprecursor vapors into our GEOS-Chem-TOMAS simulations. This fixed flux of 500 µg 1087 m<sup>-2</sup> d<sup>-1</sup> of Arctic MSOA-precursor vapors (with a yield of unity) emitted from open 1088 1089 seawater in the Arctic and near Arctic (north of  $50^{\circ}$  N) was a crude representation of the 1090 source function because of the lack of knowledge about the nature and source of Arctic 1091 MSOA. However, implementation of condensable Arctic MSOA in our simulation 1092 reduced the model-to-measurement MFE for the summertime median aerosol size 1093 distributions by a factor of 2-4 across the three measurement platforms, indicating a 1094 strong sensitivity of the simulated size distributions to growth by Arctic MSOA. Without 1095 Arctic MSOA, particle growth to diameters of 50 nm to 200 nm was strongly 1096 underpredicted in our simulations. Increasing the particle nucleation rate by 100-fold 1097 further reduced the MFE for Eureka and the ship track, indicating that additional 1098 materials such as (but not limited to) gas-phase iodine, and/or amines and/or possibly 1099 extremely low volatility organics may be participating in nucleation, and/or other 1100 mechanisms such as particle fragmentation, leading to faster rates than our ternary 1101 scheme.

1102

1103 Introduction of a 30/70 non-/semi-volatile split for the simulated Arctic MSOA reduced 1104 by 2- to 3-fold the model-to-measurement MFE for the summertime aerosol size 1105 distributions for Alert (0.10) and the ship track (0.10), and also yielded the lowest MFE 1106 for Eureka (0.09) if the Arctic MSOA-precursor vapor source flux was doubled. These 1107 findings offer support that the condensing Arctic MSOAs contributing to growth of

40





particles with diameters larger than about 20 nm in the Canadian Arctic Archipelagocould contain a large fraction of semi-volatile species.

1110

1111 Size-resolved mass fractions indicated that initial growth of simulated nascent sub-10 nm 1112 particles (arising from ternary nucleation of ammonia sulfuric acid, water vapors) 1113 occurred primarily by condensation involving biogenic sulfate and MSA, both derived 1114 from oxidation of dimethyl sulfide of marine origin. Arctic MSOA contributed about 20-1115 80% to size-resolved particle mass for diameters between 10 nm and 100 nm, with the 1116 largest contributions for the ship track simulation. The simulated contribution of primary 1117 organics of sea-spray origin to sub-100 nm particle mass fractions was largest for the ship 1118 track simulation in the marine boundary layer, with mass fractions approaching 20% for 1119 particles with diameters around 10 nm to 20 nm.

1120

1121 By comparing our best (lowest MFE) simulations with and without the Arctic MSOA 1122 formed from precursors with marine sources north of 50° N, we found that Arctic MSOA 1123 had a strong summertime- and pan-Arctic-mean top-of-the-atmosphere aerosol direct 1124 radiative effect (DRE) of -0.04 W m<sup>-2</sup>, and cloud-albedo aerosol indirect effect (AIE) of -0.4 W m<sup>-2</sup>. The comparison of these simulations with and without Arctic MSOA 1125 1126 suggested a strong sensitivity of climate-relevant effects to Arctic MSOA. However, we 1127 caution that a high level of uncertainty is associated with our quantification of these 1128 effects, due to uncertainty about the composition, and source fluxes for these condensing 1129 vapors. Future studies are needed to reduce these uncertainties.

1130

1131 Many knowledge gaps remain regarding the role of organics within the processes that 1132 shape particle size distributions in the Arctic climate system. For example, Willis et al. 1133 (2017) found that the organics in the aerosol in the summertime Canadian Arctic 1134 Archipelago were not like typical biogenic SOA, having instead a character with a long 1135 hydrocarbon chain, implying a fatty-acid-type precursor, which is a common component 1136 of the marine microlayer. Additionally, Mungall et al. (2017) found that the marine 1137 microlaver in the Canadian Arctic Archipelago was a source of OVOCs, which could also 1138 be related to Arctic MSOA. Further measurements are needed to identify the organic





1139 vapors that yield Arctic MSOA through condensational particle growth, along with their 1140 sources, chemistry, and spatial distribution within the Arctic. Additionally, given the 1141 climate relevance of NH<sub>3</sub> through formation of nascent particles, measurements are 1142 needed to better identify and quantify its sources across the summertime Arctic, and to 1143 further examine the spatial distribution of the subsequent Arctic particle growth events. 1144 Further, size-resolved particle concentrations and composition measurements 1145 (particularly for sulfate and organic aerosol), would constrain the controlling processes 1146 for all sub-micron particle diameters. Such work could also reduce uncertainty related to 1147 aerosol effects within the Arctic climate system. This work will also lay a foundation for 1148 prediction of future aerosol effects within the context of our rapidly changing and 1149 warming Arctic, as sea ice extent, biological and anthropogenic activity are altered.

1150

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1152

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1162

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1170 Change). Colorado State University researchers were supported by the US Department of 1171 Energy's Atmospheric System Research, an Office of Science, Office of Biological and 1172 Environmental Research program, under Grant No. DE-SC0011780, the U.S. National 1173 Science Foundation, Atmospheric Chemistry program, under Grant No. AGS-1559607, 1174 and by the U.S National Oceanic and Atmospheric Administration, an Office of Science, 1175 Office of Atmospheric Chemistry, Carbon Cycle, and Climate Program, under the 1176 cooperative agreement award No. NA17OAR430001. A. Moravek's work was supported 1177 by the NSERC CREATE program IACPES postdoctoral fellowship. 1178 1179 References 1180 1181 Abbatt, J. P. D., Leaitch, W. R., Aliabadi, A. A., Bertram, A. K., Blanchet, J.-P., Bozem, 1182 H., Burkart, J., Chang, R., Chaubey, J.-P., Christensen, R. J., Cirisan, A., Collins, D. B., 1183 Croft, B., Dionne, J., Evans, G., Fletcher, C., Ghahremaninezhad, R., Girard, E., Gong, 1184 W., Gosselin, M., Gourdal, M., Hanna, S., Hayashida, H., Herber, A., Hesaraki, S., Hoor, 1185 P., Huang, L., Hussherr, R., Irish, V. E., Keita, S., Kodros, J. K., Köllner, F., Kolonjari, F., Kunkel, D., Ladino, L. A., Law, K., Levasseur, M., Libois, Q., Liggio, J., Lizotte, M., 1186 1187 Macdonald, K., Mahmood, R., Martin, R. V., Mason, R. H., Miller, L., Moravek, A., 1188 Mortenson, E., Mungall, E., Murphy, J. G., Namazi, M., Norman, A.-L., O'Neill, N., 1189 Pierce, J.R., Russell, L. M., Schneider, J., Schulz, H., Sharma, S., Si, M., Staebler, R. M., 1190 Steiner, N. S., Tapias, M., Thomas, J., von Salzen, K., Wentzell, J., Willis, M. D., 1191 Wentworth, G. R., Xu, J.-W., Yakobi-Hancock, J. D.: New insights into aerosol and 1192 climate in the Arctic (in prep.), 2018. 1193 1194 Abdul-Razzak, H.: A parameterization of aerosol activation 3. Sectional representation, J. 1195 Geophys. Res., 107(D3), 4026, doi:10.1029/2001JD000483, 2002. 1196 1197 Alexander, B., Allman, D. J., Amos, H. M., Fairlie, T. D., Dachs, J., Hegg, D. A. and 1198 Sletten, R. S.: Isotopic constraints on the formation pathways of sulfate aerosol in the 1199 marine boundary layer of the subtropical northeast Atlantic Ocean, J. Geophys. Res. 1200 Atmos., 117(6), 1–17, doi:10.1029/2011JD016773, 2012. 1201 1202 Allan, J. D., Williams, P. I., Najera, J., Whitehead, J. D., Flynn, M. J., Taylor, J. W., Liu, 1203 D., Darbyshire, E., Carpenter, L. J., Chance, R., Andrews, S. J., Hackenberg, S. C. and 1204 McFiggans, G.: Iodine observed in new particle formation events in the Arctic 1205 atmosphere during ACCACIA, Atmos. Chem. Phys., 15(10), 5599–5609, 1206 doi:10.5194/acp-15-5599-2015, 2015. 1207 1208 Almeida, J., Schobesberger, S., Kürten, A., Ortega, I. K., Kupiainen-Määttä, O., Praplan, 1209 A. P., Adamov, A., Amorim, A., Bianchi, F., Breitenlechner, M., David, A., Dommen, J., 1210 Donahue, N. M., Downard, A., Dunne, E., Duplissy, J., Ehrhart, S., Flagan, R. C., 1211 Franchin, A., Guida, R., Hakala, J., Hansel, A., Heinritzi, M., Henschel, H., Jokinen, T.,





- 1212 Junninen, H., Kajos, M., Kangasluoma, J., Keskinen, H., Kupc, A., Kurtén, T., Kvashin,
- 1213 A. N., Laaksonen, A., Lehtipalo, K., Leiminger, M., Leppä, J., Loukonen, V.,
- 1214 Makhmutov, V., Mathot, S., McGrath, M. J., Nieminen, T., Olenius, T., Onnela, A.,
- 1215 Petäjä, T., Riccobono, F., Riipinen, I., Rissanen, M., Rondo, L., Ruuskanen, T., Santos,
- 1216 F. D., Sarnela, N., Schallhart, S., Schnitzhofer, R., Seinfeld, J. H., Simon, M., Sipilä, M.,
- 1217 Stozhkov, Y., Stratmann, F., Tomé, A., Tröstl, J., Tsagkogeorgas, G., Vaattovaara, P.,
- 1218 Viisanen, Y., Virtanen, A., Vrtala, A., Wagner, P. E., Weingartner, E., Wex, H.,
- 1219 Williamson, C., Wimmer, D., Ye, P., Yli-Juuti, T., Carslaw, K. S., Kulmala, M., Curtius,
- 1220 J., Baltensperger, U., Worsnop, D. R., Vehkamäki, H. and Kirkby, J.: Molecular
- 1221 understanding of sulphuric acid-amine particle nucleation in the atmosphere, Nature,
- 1222 502(7471), 359–363, doi:10.1038/nature12663, 2013.
- 1223
- 1224 Amos, H. M., Jacob, D. J., Holmes, C. D., Fisher, J. A., Wang, Q., Yantosca, R. M.,
- 1225 Corbitt, E. S., Galarneau, E., Rutter, A. P., Gustin, M. S., Steffen, A., Schauer, J. J.,
- 1226 Graydon, J. A., St Louis, V. L., Talbot, R. W., Edgerton, E. S., Zhang, Y. and
- Sunderland, E. M.: Gas-particle partitioning of atmospheric Hg(II) and its effect on
   global mercury deposition, Atmos. Chem. Phys., 12(1), 591–603, doi:10.5194/acp-12-
- 591-2012, 2012.
- 1230
- 1231 Asmi, E., Kondratyev, V., Brus, D., Laurila, T., Lihavainen, H., Backman, J., Vakkari,
- 1232 V., Aurela, M., Hatakka, J., Viisanen, Y., Uttal, T., Ivakhov, V. and Makshtas, A.:
- 1233 Aerosol size distribution seasonal characteristics measured in Tiksi, Russian Arctic,
- 1234 Atmos. Chem. Phys., 16, 1271–1287, doi:10.5194/acp-16-1271-2016, 2016.
- 1235
- Baranizadeh, E., Murphy, N. B., Julin, J., Falahat, S., Reddington, L. C., Arola, A., Ahlm,
  L., Mikkonen, S., Fountoukis, C., Patoulias, D., Minikin, A., Hamburger, T., Laaksonen,
  A., Pandis, N. S., Vehkamäki, H., Lehtinen, E. J. K. and Riipinen, I.: Implementation of
  state-of-the-art ternary new-particle formation scheme to the regional chemical transport
  model PMCAMx-UF in Europe, Geosci. Model Dev., 9(8), 2741–2754,
  doi:10.5194/gmd-9-2741-2016, 2016.
- 1241 1242
- Barnes, I., Hjorth, J. and Mihalopoulos, N.: Dimethyl Sulfide and Dimethyl Sulfoxide
- and Their Oxidation in their Atmosphere, Chem. Rev., 106, 940–975,
- 1245 doi:10.1021/cr020529+, 2006.
- 1246
- 1247 Barrie, L. A.: Arctic Aerosols: Composition, Sources and Transport BT Ice Core
- Studies of Global Biogeochemical Cycles, edited by R. J. Delmas, pp. 1–22, Springer
  Berlin Heidelberg, Berlin, Heidelberg., 1995.
- 1250
- Bond, T. C., Bhardwaj, E., Dong, R., Jogani, R., Jung, S., Roden, C., Streets, D. G. and
  Trautmann, N. M.: Historical emissions of black and organic carbon aerosol from energyrelated combustion, 1850-2000, Global Biogeochem. Cycles, 21(2), 1–16,
- 1254 doi:10.1029/2006GB002840, 2007.
- 1255
- 1256 Bouwman, A. F., Lee, D. S., Asman, W. A. H., Dentener, F. J., Hoek, K. W. Van Der,
- 1257 Olivier, J. G. J. and Tg, N.: A global high-resolution emission inventory for ammonia,





- 1258 Global Biogeochem. Cycles, 11(4), 561–587, 1997.
- Boylan, J. W. and Russell, A. G.: PM and light extinction model performance metrics,
- goals, and criteria for three-dimensional air quality models, Atmos. Environ., 40(26),
  4946–4959, doi:10.1016/j.atmosenv.2005.09.087, 2006.
- 1263
- Breider, T. J., Mickley, L. J., Jacob, D. J., Wang, Q. Q., Fisher, J. A., Chang, R. Y. and
  Alexander, B.: Annual distributions and sources of Arctic aerosol components, aerosol
  optical depth, and aerosol absorption, J. Geophys. Res. Atmos., 119, 4107–4124,
  doi:10.1002/2013JD020996, 2014.
- 1268
- 1269 Breider, T. J., Mickley, L. J., Jacob, D. J., Ge, C., Wang, J., Sulprizio, M. P., Croft, B.,
- 1270 Ridley, D. A., Mcconnell, J. R., Sharma, S., Husain, L., Dutkiewicz, V. A., Eleftheriadis,
- 1271 K., Skov, H. and Hopke, P. K.: Multidecadal trends in aerosol radiative forcing over the
- Arctic: Contribution of changes in anthropogenic aerosol to Arctic warming since 1980,
  J. Geophys. Res. Atmos., 122, doi:10.1002/2016JD025321, 2017.
- 1274
- 1275 Brock, C. A., Cozic, J., Bahreini, R., Froyd, K. D., Middlebrook, A. M., McComiskey,
- 1276 A., Brioude, J., Cooper, O. R., Stohl, A., Aikin, K. C., De Gouw, J. A., Fahey, D. W.,
- 1277 Ferrare, R. A., Gao, R. S., Gore, W., Holloway, J. S., Hübler, G., Jefferson, A., Lack, D.
- 1278 A., Lance, S., Moore, R. H., Murphy, D. M., Nenes, A., Novelli, P. C., Nowak, J. B.,
- 1279 Ogren, J. A., Peischl, J., Pierce, R. B., Pilewskie, P., Quinn, P. K., Ryerson, T. B.,
- 1280 Schmidt, K. S., Schwarz, J. P., Sodemann, H., Spackman, J. R., Stark, H., Thomson, D.
- 1281 S., Thornberry, T., Veres, P., Watts, L. A., Warneke, C. and Wollny, A. G.:
- 1282 Characteristics, sources, and transport of aerosols measured in spring 2008 during the
- aerosol, radiation, and cloud processes affecting Arctic Climate (ARCPAC) Project,
- 1284 Atmos. Chem. Phys., 11(6), 2423–2453, doi:10.5194/acp-11-2423-2011, 2011.
- 1285
- Browse, J., Carslaw, K. S., Arnold, S. R., Pringle, K. and Boucher, O.: The scavenging processes controlling the seasonal cycle in Arctic sulphate and black carbon aerosol, Atmos. Chem. Phys. 12(15), 6775, 6708, doi:10.5104/aep.12.6775.2012, 2012.
- 1288 Atmos. Chem. Phys., 12(15), 6775–6798, doi:10.5194/acp-12-6775-2012, 2012. 1289
- Burkart, J., Hodshire, A. L., Mungall, E. L., Pierce, J. R., Collins, D. B., Ladino, L. A.,
  Lee, A. K. Y., Irish, V., Wentzell, J. J. B., Liggio, J., Papakyriakou, T., Murphy, J. and
  Abbatt, J.: Organic Condensation and Particle Growth to CCN Sizes in the Summertime
- 1293 Marine Arctic Is Driven by Materials More Semivolatile Than at Continental Sites,
- 1294 Geophys. Res. Lett., 44(20), 10,725-10,734, doi:10.1002/2017GL075671, 2017a.
- 1295
- 1296 Burkart, J., Willis, M. D., Bozem, H., Thomas, J. L., Law, K., Hoor, P., Aliabadi, A. A.,
- 1297 Köllner, F., Schneider, J., Herber, A., Abbatt, J. P. D. and Richard Leaitch, W.:
- 1298 Summertime observations of elevated levels of ultrafine particles in the high Arctic
- 1299 marine boundary layer, Atmos. Chem. Phys., 17(8), 5515–5535, doi:10.5194/acp-17-
- 1300 5515-2017, 2017b.
- 1301
- Carpenter, L. J. and Nightingale, P. D.: Chemistry and Release of Gases from the Surface
  Ocean, Chem. Rev., 115(10), 4015–4034, doi:10.1021/cr5007123, 2015.

Atmospheric Chemistry and Physics Discussions



1204	
1304	Commuter I. J. Archar G. D. and Darle D. Orangetware three traces are being
1305	Carpenter, L. J., Archer, S. D. and Beale, R.: Ocean-atmosphere trace gas exchange,
1306	Chem. Soc. Rev., $41(19)$ , $64/3-6506$ , $doi:10.1039/c2cs35121h$ , $2012$ .
1307	
1308	Carslaw, K. S., Lee, L. A., Reddington, C. L., Pringle, K. J., Rap, A., Forster, P. M.,
1309	Mann, G. W., Spracklen, D. V, Woodhouse, M. T., Regayre, L. A. and Pierce, J. R.:
1310	Large contribution of natural aerosols to uncertainty in indirect forcing, Nature, 503, 67
1311	[online] Available from: http://dx.doi.org/10.1038/nature12674, 2013.
1312	
1313	Chang, R. Y. W., Leck, C., Graus, M., Müller, M., Paatero, J., Burkhart, J. F., Stohl, A.,
1314	Orr, L. H., Hayden, K., Li, S. M., Hansel, A., Tjernström, M., Leaitch, W. R. and Abbatt,
1315	J. P. D.: Aerosol composition and sources in the central Arctic Ocean during ASCOS,
1316	Atmos. Chem. Phys., 11(20), 10619–10636, doi:10.5194/acp-11-10619-2011, 2011a.
1317	
1318	Chang R Y W Siostedt S J Pierce J R Papakyriakou T N Scarratt M G
1319	Michaud S Levasseur M Leaitch W R and Abbatt I P D Relating atmospheric
1320	and oceanic DMS levels to particle nucleation events in the Canadian Arctic I Geophys
1321	Res Atmos $116(21)$ 1–10 doi:10.1029/2011ID015926 2011b
1322	res. runos., 110(21), 1 10, doi:10.102)/201130010520, 20110.
1323	Charlson R. I. Schwartz, S. F. Hales, I. M. Cess, R. D. Coakley, I. A. Hansen, I. F.
1324	and Hofmann D. I. Climate Forcing by Anthronogenic Aerosols Science (80-)
1324	255(50/3) /23_/30 [online] Available from:
1325	http://science.sciencemag.org/content/255/50/3//23 abstract_1002
1320	http://selence.selencemag.org/content/255/50+5/425.abstract, 1772.
1220	Chatfield P. B. and Crutzen P. I. Are There Interactions of Indine and Sulfur Species in
1220	Marine Air Dhotochemistry I. Coonhyg. Dog. 05(D12), 22210, 22241
1229	$d_{2}$ $d_{2$
1221	doi.10.1029/JD09J1D15p22519, 1990.
1001	Chin M. Jacob D. J. Cardner C. M. Fereman Fewler M. S. Spire D. A. and Savaia
1002	D. L.: A global three dimensional model of transportation gulfate. I. Coophys. Bos.
1000	D. L. A global under undersional model of hopospheric surface, J. Geophys. Res.
1334	Aunos., 101(D15), 18007–18090, doi:10.1029/90JD01221, 1990.
1335	Chie D. Tinel J. Conseler J. Cimere D. Demend F. Course C. and Vellemen D.
1330	Chiu, R., Tinei, L., Gonzalez, L., Ciuraru, R., Bernard, F., George, C. and Volkamer, R.:
1337	Uv photochemistry of carboxylic acids at the air-sea boundary. A relevant source of
1338	glyoxal and other oxygenated VOC in the marine atmosphere, Geophys. Res. Lett., 44(2),
1339	10/9–108/, doi:10.1002/2016GL0/1240, 2017.
1340	
1341	Collins, D. B., Ault, A. P., Moffet, R. C., Ruppel, M. J., Cuadra-Rodriguez, L. A.,
1342	Guasco, T. L., Corrigan, C. E., Pedler, B. E., Azam, F., Aluwihare, L. I., Bertram, T. H.,
1343	Roberts, G. C., Grassian, V. H. and Prather, K. A.: Impact of marine biogeochemistry on
1344	the chemical mixing state and cloud forming ability of nascent sea spray aerosol, J.
1345	Geophys. Res. Atmos., 118(15), 8553–8565, doi:10.1002/jgrd.50598, 2013.
1346	
1347	Collins, D. B., Bertram, T. H., Sultana, C. M., Lee, C., Axson, J. L. and Prather, K. A.:
1348	Phytoplankton blooms weakly influence the cloud forming ability of sea spray aerosol,
1349	Geophys. Res. Lett., 43(18), 9975–9983, doi:10.1002/2016GL069922, 2016.





1350	
1351	Collins, D. B., Burkart, J., Chang, R. YW., Lizotte, M., Boivin-Rioux, A., Blais, M.,
1352	Mungall, E. L., Bover, M., Irish, V. E., Massé, G., Kunkel, D., Tremblay, JÉ.,
1353	Papakyriakou, T., Bertram, A. K., Bozem, H., Gosselin, M., Levasseur, M. and Abbatt, J.
1354	P D Frequent Ultrafine Particle Formation and Growth in the Canadian Arctic Marine
1355	Environment Atmos Chem Phys 17 13119-13138 doi:10.5194/acp-17-13119-2017
1356	2017
1357	
1358	Crinna M. Janssens-Maenhout G. Dentener F. Guizzardi D. Sindelarova K
1359	Muntean M Van Dingenen R and Granier C. Forty years of improvements in
1360	European air quality: Regional policy-industry interactions with global impacts. Atmos
1361	Chem Phys $16(6)$ $3825-3841$ doi:10.5194/acn-16-3825-2016 2016
1362	Chemi 1 1190., 10(0), 5025 5011, doi:10.519 #dep 10 5025 2010, 2010.
1363	Croft B Wentworth G R Martin R V Leaitch W R Murnhy I G Murnhy B
1364	N Kodros I K Abbatt I P D and Pierce I R Contribution of Arctic seabird-colony
1365	ammonia to atmospheric particles and cloud-albedo radiative effect. Nat. Commun. 7. 1–
1366	10 doi:10.1038/ncomms13444_2016a
1367	10, doi.10.1050/neoninis15+++, 2010d.
1368	Croft B Martin R V Richard Leaitch W Tunved P Breider T I D'Andrea S D
1369	and Pierce I R · Processes controlling the annual cycle of Arctic aerosol number and
1370	size distributions Atmos Chem Phys 16(6) 3665–3682 doi:10.5194/acn-16-3665-
1370	2016 2016b
1372	2010, 20100.
1372	D'Andrea S.D. Häkkinen S.A.K. Westervelt D.M. Kuang C. Levin F. I.T.
1374	Kanawade V P Leaitch W R Spracklen D V Rijninen L and Pierce L R
1375	Understanding global secondary organic aerosol amount and size-resolved
1376	condensational behavior. Atmos. Chem. Phys. 13(22), 11519–11534. doi:10.5194/acn-
1377	13-11519-2013 2013
1378	10 11017 2010, 2010.
1379	Dall'Osto M Beddows D C S Tunved P Kreici R Ström I Hansson H C
1380	Yoon Y I Park K T Becagli S Udisti R Onasch T Ódowd C D Simó R and
1381	Harrison R M · Arctic sea ice melt leads to atmospheric new particle formation Sci
1382	Rep. $7(1)$ 1–10 doi:10.1038/s41598-017-03328-1.2017
1383	Rep., 7(1), 1 10, doi:10.1050/011090 017 05520 1, 2017.
1384	Dall'Osto M Simo R Harrison R M Beddows D C S Saiz-Lonez A Lange R
1385	Skov H Naigaard J K Nielsen J E and Massling A : Abiotic and biotic sources
1386	influencing spring new particle formation in North East Greenland Atmos Environ
1387	190(July) 126–134 doi:10.1016/I.ATMOSENV.2018.07.019.2018a
1388	1) ((uil)), 120 15 1, uoi.1010/0.111100021((1.2010.07.01)), 2010u.
1389	Dall'Osto M. Geels C. Beddows D.C. S. Boertmann D. Lange R. Nøigaard I.K.
1390	Harrison R M Simo R Skov H and Massling A Regions of onen water and melting
1391	sea ice drive new particle formation in North Fast Greenland Sci Ren 8(1) 6109
1392	doi:10.1038/s41598-018-24426-8_2018b
1393	di.10.1050/511570 010 21120 0, 20100.
1394	Donahue N.M. Epstein S.A. Pandis S.N. and Robinson A.L.: A two-dimensional
1071	2 channes, 11, 11, 2 provin, 5, 11, 1 and 5, 5, 11, and 100 moon, 11, 2., 11 two dimensional





1396 11(7), 3303–3318, doi:10.5194/acp-11-3303-2011, 2011. 1397 1398 Dunne, E. M., Gordon, H., Kurten, A., Almeida, J., Duplissy, J., Williamson, C., Ortega, 1399 I. K., Pringle, K. J., Adamov, A., Baltensperger, U., Barmet, P., Benduhn, F., Bianchi, F., 1400 Breitenlechner, M., Clarke, A., Curtius, J., Dommen, J., Donahue, N. M., Ehrhart, S., 1401 Flagan, R. C., Franchin, A., Guida, R., Hakala, J., Hansel, A., Heinritzi, M., Jokinen, T., 1402 Kangasluoma, J., Kirkby, J., Kulmala, M., Kupc, A., Lawler, M. J., Lehtipalo, K., 1403 Reddington, C. L. S., Riccobono, F., Richards, N. A. D., Rissanen, M. P., Rondo, L., 1404 Sarnela, N., Schobesberger, S., Sengupta, K., Simon, M., Sipilä, M., Smith, J. N., 1405 Stozkhov, Y., Tomé, A., Tröstl, J., Wagner, P. E., Williamson, C., Wimmer, D., Winkler, 1406 P. M., Yan, C. and Carslaw, K. S. .: Global atmospheric particle formation from CERN 1407 CLOUD measurements., Science (80-.)., 354(6316), 1119-1124, 2016. 1408 1409 Ellis, R. A., Murphy, J. G., Pattey, E., Van Haarlem, R., O'Brien, J. M. and Herndon, S. 1410 C.: Characterizing a Quantum Cascade Tunable Infrared Laser Differential Absorption 1411 Spectrometer (QC-TILDAS) for measurements of atmospheric ammonia, Atmos. Meas. 1412 Tech., 3(2), 397–406, doi:10.5194/amt-3-397-2010, 2010. 1413 1414 Facchini, M. C., Decesari, S., Rinaldi, M., Carbone, C., Finessi, E., Mircea, M., Fuzzi, S., 1415 Moretti, F., Tagliavini, E., Ceburnis, D. and O'Dowd, C. D.: Important Source of Marine 1416 Secondary Organic Aerosol from Biogenic Amines, Environ. Sci. Technol., 42(24), 1417 9116-9121, doi:10.1021/es8018385, 2008. 1418 1419 Fairlie, T. D., Jacob, D. J. and Park, R. J.: The impact of transpacific transport of mineral 1420 dust in the United States, Atmos. Environ., 41(6), 1251–1266, 1421 doi:10.1016/j.atmosenv.2006.09.048, 2007. 1422 1423 Fisher, J. A., Jacob, D. J., Wang, Q., Bahreini, R., Carouge, C. C., Cubison, M. J., Dibb, 1424 J. E., Diehl, T., Jimenez, J. L., Leibensperger, E. M., Lu, Z., Meinders, M. B. J., Pve, H. 1425 O. T., Quinn, P. K., Sharma, S., Streets, D. G., van Donkelaar, A. and Yantosca, R. M.: 1426 Sources, distribution, and acidity of sulfate-ammonium aerosol in the Arctic in winter-1427 spring, Atmos. Environ., 45(39), 7301–7318, doi:10.1016/j.atmosenv.2011.08.030, 2011. 1428 1429 Fogal, P. F., LeBlanc, L. M. and Drummond, J. R.: The Polar Environment Atmospheric 1430 Research Laboratory (PEARL): Sounding the Atmosphere at 80 North, Arctic, 66(3), 1431 377–386 [online] Available from: http://www.jstor.org/stable/23594645, 2013. 1432 1433 Freud, E., Krejci, R., Tunved, P., Leaitch, R., Nguyen, Q. T., Massling, A., Skov, H. and 1434 Barrie, L.: Pan-Arctic aerosol number size distributions: Seasonality and transport 1435 patterns, Atmos. Chem. Phys., 17(13), 8101-8128, doi:10.5194/acp-17-8101-2017, 2017. 1436 1437 Fuchs, N. A.: The mechanics of aerosols. By N. A. Fuchs. Translated by R. E. Daisley 1438 and Marina Fuchs; Edited by C. N. Davies. London (Pergamon Press), 1964. Pp. xiv, 1439 408; 82 Figures; 40 Tables. £6, Q. J. R. Meteorol. Soc., 91(388), 249, 1440 doi:10.1002/qj.49709138822, 1964.

1441





- 1442 Gantt, B. and Meskhidze, N.: The physical and chemical characteristics of marine
- 1443 primary organic aerosol: A review, Atmos. Chem. Phys., 13(8), 3979–3996,
- 1444 doi:10.5194/acp-13-3979-2013, 2013.
- 1445
- Garrett, T. J., Brattström, S., Sharma, S., Worthy, D. E. J. and Novelli, P.: The role of
  scavenging in the seasonal transport of black carbon and sulfate to the Arctic, Geophys.
  Res. Lett., 38(16), 1–6, doi:10.1029/2011GL048221, 2011.
- 1449
  1450 Ghahremaninezhad, R., Norman, A. L., Abbatt, J. P. D., Levasseur, M. and Thomas, J.
  1451 L.: Biogenic, anthropogenic and sea salt sulfate size-segregated aerosols in the Arctic
- summer, Atmos. Chem. Phys., 16(8), 5191–5202, doi:10.5194/acp-16-5191-2016, 2016.
- 1453
- Ghahremaninezhad, R., Norman, A. L., Croft, B., Martin, R. V., Pierce, J. R., Burkart, J.,
  Rempillo, O., Bozem, H., Kunkel, D., Thomas, J. L., Aliabadi, A. A., Wentworth, G. R.,
  Levasseur, M., Staebler, R. M., Sharma, S. and Richard Leaitch, W.: Boundary layer and
  free-Tropospheric dimethyl sulfide in the Arctic spring and summer, Atmos. Chem.
- 1458 Phys., 17(14), 8757–8770, doi:10.5194/acp-17-8757-2017, 2017.
- 1459
- Giamarelou, M., Eleftheriadis, K., Nyeki, S., Tunved, P., Torseth, K. and Biskos, G.:
  Indirect evidence of the composition of nucleation mode particles in the high Arctic, J.
- 1462 Geophys. Res. Atmos., 121, 965–975, doi:10.1002/2015JD023646, 2016.
- 1463

Giglio, L., Randerson, J. T. and Van Der Werf, G. R.: Analysis of daily, monthly, and
annual burned area using the fourth-generation global fire emissions database (GFED4),
J. Geophys. Res. Biogeosciences, 118(1), 317–328, doi:10.1002/jgrg.20042, 2013.

- 1467
- 1468 Gordon, H., Kirkby, J., Baltensperger, U., Bianchi, F., Breitenlechner, M., Curtius, J.,
- 1469 Dias, A., Dommen, J., Donahue, N. M., Dunne, E. M., Duplissy, J., Ehrhart, S., Flagan,
- 1470 R. C., Frege, C., Fuchs, C., Hansel, A., Hoyle, C. R., Kulmala, M., Kürten, A., Lehtipalo,
- 1471 K., Makhmutov, V., Molteni, U., Rissanen, M. P., Stozkhov, Y., Tröstl, J.,
- 1472 Tsagkogeorgas, G., Wagner, R., Williamson, C., Wimmer, D., Winkler, P. M., Yan, C.
- 1473 and Carslaw, K. S.: Causes and importance of new particle formation in the present-day
- and preindustrial atmospheres, J. Geophys. Res. Atmos., 122(16), 8739–8760,
- 1475 doi:10.1002/2017JD026844, 2017.
- 1476
- 1477 Gourdal, M., Lizotte, M., Massé, G., Gosselin, M., Scarratt, M. and Levasseur, M.:
- 1478 Dimethylsulfide dynamics in first-year sea ice melt ponds in the Canadian Arctic
- 1479 Archipelago, Biogeosciences, 15, 3169–3188, doi:10.5194/bg-2017-432, 2018.
- 1480

Grythe, H., Ström, J., Krejci, R., Quinn, P. and Stohl, A.: A review of sea-spray aerosol
source functions using a large global set of sea salt aerosol concentration measurements,

- 1483 Atmos. Chem. Phys., 14(3), 1277–1297, doi:10.5194/acp-14-1277-2014, 2014.
- 1484
- 1485 Gunsch, M. J., Kirpes, R. M., Kolesar, K. R., Barrett, T. E., China, S., Sheesley, R. J.,
- 1486 Laskin, A., Wiedensohler, A., Tuch, T. and Pratt, K. A.: Contributions of transported
- 1487 Prudhoe Bay oil field emissions to the aerosol population in Utqiagvik, Alaska, Atmos.





1488 Chem. Phys., 17(17), 10879–10892, doi:10.5194/acp-17-10879-2017, 2017. 1489 1490 Hayashida, H., Steiner, N., Monahan, A., Galindo, V., Lizotte, M. and Levasseur, M.: 1491 Implications of sea-ice biogeochemistry for oceanic production and emissions of 1492 dimethyl sulfide in the Arctic, , 3129-3155, 2017. 1493 1494 Hegg, D. A., Hobbs, P. V, Gass, S., Nance, J. D. and Rangno, A. L.: Aerosol 1495 measurements in the Arctic relevant to direct and indirect radiative forcing, J. Geophys. 1496 Res., 101(D18), 23,349-23,363, 1996. 1497 1498 Heintzenberg, J. and Leck, C.: The summer aerosol in the central Arctic 1991-2008: Did 1499 it change or not?, Atmos. Chem. Phys., 12(9), 3969-3983, doi:10.5194/acp-12-3969-1500 2012, 2012. 1501 1502 Heintzenberg, J., Leck, C. and Tunved, P.: Potential source regions and processes of 1503 aerosol in the summer Arctic, Atmos. Chem. Phys., 15(11), 6487–6502, doi:10.5194/acp-1504 15-6487-2015, 2015. 1505 1506 Heintzenberg, J., Tunved, P., Galí, M. and Leck, C.: New particle formation in the 1507 Svalbard region 2006-2015, Atmos. Chem. Phys., 17(10), 6153–6175, doi:10.5194/acp-1508 17-6153-2017, 2017. 1509 1510 Hodshire, A. L., Palm, B. B., Alexander, M. L., Bian, Q., Campuzano-Jost, P., Cross, E. 1511 S., Day, D. A., de Sá, S. S., Guenther, A. B., Hansel, A., Hunter, J. F., Jud, W., Karl, T., 1512 Kim, S., Kroll, J. H., Park, J.-H., Peng, Z., Seco, R., Smith, J. N., Jimenez, J. L. and 1513 Pierce, J. R.: Constraining nucleation, condensation, and chemistry in oxidation flow 1514 reactors using size-distribution measurements and aerosol microphysical modelling, 1515 Atmos. Chem. Phys. Discuss., (May), 1–53, doi:10.5194/acp-2018-223, 2018. 1516 1517 Hodshire, A. L., Kodros, J. K., Croft, B., Campuzano-Jost, P., Nault, B. A., Schroder, J. 1518 C., Jimenez, J.-L. and Pierce, J. R.: The potential role of methansulfonic acid (MSA) in 1519 aerosol formation and growth and the associated radiative forcings, (in prep.), 2018. 1520 1521 Huang, L., Brook, J. R., Zhang, W., Li, S. M., Graham, L., Ernst, D., Chivulescu, A. and 1522 Lu, G.: Stable isotope measurements of carbon fractions (OC/EC) in airborne particulate: 1523 A new dimension for source characterization and apportionment, Atmos. Environ., 40, 1524 2690-2705, 2006. 1525 1526 Iacono, M. J., Delamere, J. S., Mlawer, E. J., Shephard, M. W., Clough, S. A. and 1527 Collins, W. D.: Radiative forcing by long-lived greenhouse gases: Calculations with the 1528 AER radiative transfer models, J. Geophys. Res. Atmos., 113(13), 2–9, 1529 doi:10.1029/2008JD009944, 2008. 1530 1531 Jaeglé, L., Quinn, P. K., Bates, T. S., Alexander, B. and Lin, J. T.: Global distribution of 1532 sea salt aerosols: New constraints from in situ and remote sensing observations, Atmos. 1533 Chem. Phys., 11(7), 3137–3157, doi:10.5194/acp-11-3137-2011, 2011.

Atmospheric Chemistry and Physics Discussions



1534 1535

1536

1537 2010, 2010. 1538 1539 Karl, M., Leck, C., Coz, E. and Heintzenberg, J.: Marine nanogels as a source of 1540 atmospheric nanoparticles in the high Arctic, Geophys. Res. Lett., 40(14), 3738–3743, 1541 doi:10.1002/grl.50661, 2013. 1542 1543 Kerminen, V. M., Anttila, T., Lehtinen, K. E. J. and Kulmala, M.: Parameterization for 1544 atmospheric new-particle formation: Application to a system involving sulfuric acid and 1545 condensable water-soluble organic vapors, Aerosol Sci. Technol., 38(10), 1001–1008, 1546 doi:10.1080/027868290519085, 2004. 1547 1548 Kim, P. S., Jacob, D. J., Fisher, J. A., Travis, K., Yu, K., Zhu, L., Yantosca, R. M., 1549 Sulprizio, M. P., Jimenez, J. L., Campuzano-Jost, P., Froyd, K. D., Liao, J., Hair, J. W., 1550 Fenn, M. A., Butler, C. F., Wagner, N. L., Gordon, T. D., Welti, A., Wennberg, P. O., 1551 Crounse, J. D., St. Clair, J. M., Teng, A. P., Millet, D. B., Schwarz, J. P., Markovic, M. Z. 1552 and Perring, A. E.: Sources, seasonality, and trends of Southeast US aerosol: An 1553 integrated analysis of surface, aircraft, and satellite observations with the GEOS-Chem 1554 chemical transport model, Atmos. Chem. Phys., 15, 10411-10433, doi:10.5194/acpd-15-1555 10411-2015, 2015. 1556 Kirkby, J., Curtius, J., Almeida, J., Dunne, E., Duplissy, J., Ehrhart, S., Franchin, A., 1557 1558 Gagné, S., Ickes, L., Kürten, A., Kupc, A., Metzger, A., Riccobono, F., Rondo, L., 1559 Schobesberger, S., Tsagkogeorgas, G., Wimmer, D., Amorim, A., Bianchi, F., 1560 Breitenlechner, M., David, A., Dommen, J., Downard, A., Ehn, M., Flagan, R. C., Haider, 1561 S., Hansel, A., Hauser, D., Jud, W., Junninen, H., Kreissl, F., Kvashin, A., Laaksonen, A., 1562 Lehtipalo, K., Lima, J., Lovejov, E. R., Makhmutov, V., Mathot, S., Mikkilä, J., 1563 Minginette, P., Mogo, S., Nieminen, T., Onnela, A., Pereira, P., Petäjä, T., Schnitzhofer, R., Seinfeld, J. H., Sipilä, M., Stozhkov, Y., Stratmann, F., Tomé, A., Vanhanen, J., 1564 1565 Viisanen, Y., Vrtala, A., Wagner, P. E., Walther, H., Weingartner, E., Wex, H., Winkler, 1566 P. M., Carslaw, K. S., Worsnop, D. R., Baltensperger, U. and Kulmala, M.: Role of 1567 sulphuric acid, ammonia and galactic cosmic rays in atmospheric aerosol nucleation, 1568 Nature, 476(7361), 429-435, doi:10.1038/nature10343, 2011. 1569 1570 Kodros, J. K. and Pierce, J. R.: Important global and regional differences in aerosol 1571 cloud-albedo effect estimates between simulations with and without prognostic aerosol 1572 microphysics, J. Geophys. Res., 122(7), 4003–4018, doi:10.1002/2016JD025886, 2017. 1573 1574 Kodros, J. K., Cucinotta, R., Ridley, D. A., Wiedinmyer, C. and Pierce, J. R.: The aerosol 1575 radiative effects of uncontrolled combustion of domestic waste, Atmos. Chem. Phys., 1576 16(11), 6771–6784, doi:10.5194/acp-16-6771-2016, 2016. 1577 1578 Kolesar, K. R., Cellini, J., Peterson, P. K., Jefferson, A., Tuch, T., Birmili, W., 1579 Wiedensohler, A. and Pratt, K. A.: Effect of Prudhoe Bay emissions on atmospheric

Johnson, M. T.: A numerical scheme to calculate temperature and salinity dependent air-

water transfer velocities for any gas, Ocean Sci., 6(4), 913–932, doi:10.5194/os-6-913-

Atmospheric Chemistry and Physics Discussions



1580	aerosol growth events observed in Utqiagvik (Barrow), Alaska, Atmos. Environ., 152,
1581	146–155, doi:https://doi.org/10.1016/j.atmosenv.2016.12.019, 2017.
1582	Källnen E. Schneiden I. Willie M. Klimeen T. Helleis E. Derem H. Kunkel D.
1503	Konner, F., Schneider, J., Willis, M., Kilmach, I., Heileis, F., Bozem, H., Kunkel, D.,
1504	nooi, F., Buikan, J., Kichalu Leanch, W., Allabaul, A. A., Abbaul, J. P. D., Helbel, A. B.
1505	lower troposphere Atmos Chem Phys. 17(22) 13747 13766 doi:10.5104/acn 17
1500	13747_2017_2017
1588	13/4/-2017, 2017.
1589	Korhonen H. Carslaw K. S. Spracklen D. V. Ridley D. A. and Ström I.: A. global
1590	model study of processes controlling aerosol size distributions in the Arctic spring and
1591	summer I Geophys Res 113(D8) D08211 doi:10.1029/2007ID009114.2008
1592	bulline, v. Goophys. 100., 115(20), 200211, uol.10.102/200702005111, 2000.
1593	Lana, A., Bell, T. G., Simó, R., Vallina, S. M., Ballabrera-Poy, J., Kettle, A. J., Dachs, J.,
1594	Bopp, L., Saltzman, E. S., Stefels, J., Johnson, J. E. and Liss, P. S.: An updated
1595	climatology of surface dimethlysulfide concentrations and emission fluxes in the global
1596	ocean, Global Biogeochem. Cycles, 25(1), 1–17, doi:10.1029/2010GB003850, 2011.
1597	
1598	Law, K. S. and Stohl, A.: Arctic Air Pollution: Orgins and Impacts, Science, 315(March),
1599	1537–1540, 2007.
1600	
1601	Leaitch, R. W., Russell, L. M., Liu, J., Kolonjari, F., Toom, D., Huang, L., Sharma, S.,
1602	Chivulescu, A., Veber, D. and Zhang, W.: Organic functional groups in the submicron
1603	aerosol at 82.5 degrees N, 62.5 degrees W from 2012 to 2014, Atmos. Chem. Phys., 18,
1604	3269–3287, doi:10.5194/acp-18-3269-2018, 2018.
1605	Lecital W. D. Chamme G. Harme L. Terry Granters D. Chingleson A. Mardanald A.
1606	Lealton, W. K., Sharma, S., Huang, L., 100m-Sauntry, D., Chivulescu, A., Macdonald, A.
1607	M., Voli Salzeli, K., Fleice, J. K., Defualii, A. K., Schlodel, J. C., Shahiz, N. C., Chalig, P. V. W. and Norman, A. J.: Dimethyl sulfide control of the clean summertime Arctic
1600	aerosol and cloud Elem Sci Anthr. 1. 000017 doi:10.12952/journal elementa.000017
1610	2013
1611	2013.
1612	Leaitch W R Koroley A Aliabadi A A Burkart J Willis M D Abbatt J P D
1613	Bozem, H., Hoor, P., Köllner, F., Schneider, J., Herber, A., Konrad, C. and Brauner, R.:
1614	Effects of 20-100nm particles on liquid clouds in the clean summertime Arctic, Atmos.
1615	Chem. Phys., 16(17), 11107–11124, doi:10.5194/acp-16-11107-2016, 2016.
1616	
1617	Leck, C. and Bigg, E. K .: New particle formation of marine biological origin, Aerosol
1618	Sci. Technol., 44(7), 570–577, doi:10.1080/02786826.2010.481222, 2010.
1619	
1620	Lee, Y. H. and Adams, P. J.: A fast and efficient version of the TwO-Moment Aerosol
1621	Sectional (TOMAS) global aerosol microphysics model, Aerosol Sci. Technol., 46(6),
1622	678–689, doi:10.1080/02786826.2011.643259, 2012.
1623	
1624	Li, M., Zhang, Q., Kurokawa, J. I., Woo, J. H., He, K., Lu, Z., Ohara, T., Song, Y.,
1625	Streets, D. G., Carmichael, G. K., Cheng, Y., Hong, C., Huo, H., Jiang, X., Kang, S., Liu,





1626

1627 the international collaboration framework of the MICS-Asia and HTAP, Atmos. Chem. 1628 Phys., 17(2), 935–963, doi:10.5194/acp-17-935-2017, 2017. 1629 1630 Li, S.-M. and Barrie, L. a.: Biogenic sulfur aerosol in the Arctic troposphere: 1. 1631 Contributions to total sulfate, J. Geophys. Res., 98(D11), 20613, doi:10.1029/93JD02234, 1632 1993. 1633 1634 Liu, H., Jacob, D. J., Bey, I. and Yantosca, R. M.: Constraints from 210Pb and 7Be on wet 1635 deposition and transport in a global three-dimensional chemical tracer model driven by 1636 assimilated meteorological fields, J. Geophys. Res. Atmos., 106(D11), 12109–12128, 1637 doi:10.1029/2000JD900839, 2001. 1638 1639 Liu, P., Li, Y. J., Wang, Y., Gilles, M. K., Zaveri, R. A., Bertram, A. K. and Martin, S. 1640 T.: Lability of secondary organic particulate matter, Proc. Natl. Acad. Sci., 113(45), 1641 12643–12648, doi:10.1073/pnas.1603138113, 2016. 1642 1643 Lohmann, U. and Feichter, J.: Global indirect aerosol effects: a review, Atmos. Chem. 1644 Phys., 5, 715–737, doi:10.5194/acpd-4-7561-2004, 2005. 1645 1646 Lutsch, E., Dammers, E., Conway, S. and Strong, K.: Long-range Transport of NH<sub>3</sub>, 1647 CO, HCN and C<sub>2</sub> H<sub>6</sub> from the 2014 Canadian Wildfires, Geophys. Res. Lett., 1–12, 1648 doi:10.1002/2016GL070114, 2016. 1649 1650 Mårtensson, E. M., Nilsson, E. D., de Leeuw, G., Cohen, L. H. and Hansson, H.-C.: 1651 Laboratory simulations and parameterization of the primary marine aerosol production, J. 1652 Geophys. Res. Atmos., 108(D9), n/a-n/a, doi:10.1029/2002JD002263, 2003. 1653 1654 McFarquhar, G. M., Ghan, S., Verlinde, J., Korolev, A., Strapp, J. W., Schmid, B., 1655 Tomlinson, J. M., Wolde, M., Brooks, S. D., Cziczo, D., Dubey, M. K., Fan, J., Flynn, C., 1656 Gultepe, I., Hubbe, J., Gilles, M. K., Laskin, A., Lawson, P., Leaitch, W. R., Liu, P., Liu, 1657 X., Lubin, D., Mazzoleni, C., MacDonald, A. M., Moffet, R. C., Morrison, H., 1658 Ovchinnikov, M., Shupe, M. D., Turner, D. D., Xie, S., Zelenyuk, A., Bae, K., Freer, M. 1659 and Glen, A.: Indirect and semi-direct aerosol campaign: The impact of arctic aerosols on 1660 clouds, Bull. Am. Meteorol. Soc., 92(2), 183–201, doi:10.1175/2010BAMS2935.1, 2011. 1661 1662 Mungall, E. L., Croft, B., Lizotte, M., Thomas, J. L., Murphy, J. G., Levasseur, M., 1663 Martin, R. V., Wentzell, J. J. B., Liggio, J. and Abbatt, J. P. D.: Dimethyl sulfide in the 1664 summertime Arctic atmosphere: Measurements and source sensitivity simulations, 1665 Atmos. Chem. Phys., 16(11), 6665–6680, doi:10.5194/acp-16-6665-2016, 2016. 1666 1667 Mungall, E. L., Abbatt, J. P. D., Wentzell, J. J. B., Lee, A. K. Y., Thomas, J. L., Blais, 1668 M., Gosselin, M., Miller, L. A., Papakyriakou, T., Willis, M. D. and Liggio, J.: 1669 Microlayer source of oxygenated volatile organic compounds in the summertime marine 1670 Arctic boundary layer, Proc. Natl. Acad. Sci., 114(24), 6203-6208,

F., Su, H. and Zheng, B.: MIX: A mosaic Asian anthropogenic emission inventory under

1671 doi:10.1073/pnas.1620571114, 2017.

Atmospheric Chemistry and Physics Discussions



1672 1673 Murphy, J.G., Moravek, A., Wentworth, G.R., et al.: Observational constraints on the 1674 atmospheric ammonia budget in the Canadian Arctic Archipelago, (in prep.), 2018. 1675 1676 Napari, I., Noppel, M., Vehkamaki, H. and Kulmala, M.: Parametrization of ternary 1677 nucleation rates for H2SO4-NH3-H2O vapors, J. Geophys. Res. Atmos., 107(19), 2-7, 1678 doi:10.1029/2002JD002132, 2002. 1679 1680 Nguyen, Q. T., Glasius, M., Sørensen, L. L., Jensen, B., Skov, H., Birmili, W., 1681 Wiedensohler, A., Kristensson, A., Nøjgaard, J. K. and Massling, A.: Seasonal variation 1682 of atmospheric particle number concentrations, new particle formation and atmospheric 1683 oxidation capacity at the high Arctic site Villum Research Station, Station Nord, Atmos. 1684 Chem. Phys., 16(17), 11319–11336, doi:10.5194/acp-16-11319-2016, 2016. 1685 1686 Olenius, T., Kupiainen-Määttä, O., Ortega, I. K., Kurtén, T. and Vehkamäki, H.: Free 1687 energy barrier in the growth of sulfuric acid-ammonia and sulfuric acid-dimethylamine 1688 clusters, J. Chem. Phys., 139(8), doi:10.1063/1.4819024, 2013. 1689 1690 Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic 1691 growth and cloud condensation nucleus activity-Part 3: Including surfactant partitioning, 1692 Atmos. Chem. Phys., 7, 1961–1971, doi:10.5194/acp-13-1081-2013, 2007. 1693 1694 Philip, S., Martin, R. V., Pierce, J. R., Jimenez, J. L., Zhang, Q., Canagaratna, M. R., 1695 Spracklen, D. V., Nowlan, C. R., Lamsal, L. N., Cooper, M. J. and Krotkov, N. A .: 1696 Spatially and seasonally resolved estimate of the ratio of organic mass to organic carbon, 1697 Atmos. Environ., 87, 34-40, doi:10.1016/j.atmosenv.2013.11.065, 2014. 1698 1699 Pierce, J. R., Riipinen, I., Kulmala, M., Ehn, M., Petäjä, T., Junninen, H., Worsnop, D. R. 1700 and Donahue, N. M.: Quantification of the volatility of secondary organic compounds in 1701 ultrafine particles during nucleation events, Atmos. Chem. Phys., 11(17), 9019–9036, 1702 doi:10.5194/acp-11-9019-2011, 2011. 1703 1704 Pierce, J. R., Croft, B., Kodros, J. K., D'Andrea, S. D. and Martin, R. V.: The importance 1705 of interstitial particle scavenging by cloud droplets in shaping the remote aerosol size 1706 distribution and global aerosol-climate effects, Atmos. Chem. Phys., 15(11), 6147–6158, 1707 doi:10.5194/acp-15-6147-2015, 2015. 1708 1709 Polissar, A. V, Hopke, P. K. and Harris, J. M.: Source Regions for Atmospheric Aerosol 1710 Measured at Barrow, Alaska, Environ. Sci. Technol., 35(21), 4214–4226, 1711 doi:10.1021/es0107529, 2001. 1712 1713 Prather, K. A., Bertram, T. H., Grassian, V. H., Deane, G. B., Stokes, M. D., DeMott, P. 1714 J., Aluwihare, L. I., Palenik, B. P., Azam, F., Seinfeld, J. H., Moffet, R. C., Molina, M. J., 1715 Cappa, C. D., Geiger, F. M., Roberts, G. C., Russell, L. M., Ault, A. P., Baltrusaitis, J., 1716 Collins, D. B., Corrigan, C. E., Cuadra-Rodriguez, L. A., Ebben, C. J., Forestieri, S. D., 1717 Guasco, T. L., Hersey, S. P., Kim, M. J., Lambert, W. F., Modini, R. L., Mui, W., Pedler,





- 1718 B. E., Ruppel, M. J., Ryder, O. S., Schoepp, N. G., Sullivan, R. C. and Zhao, D.:
- 1719 Bringing the ocean into the laboratory to probe the chemical complexity of sea spray 1720 aerosol, Proc. Natl. Acad. Sci., 110(19), 7550–7555, doi:10.1073/pnas.1300262110,
- 1721

2013.

- 1722
- Quinn, P. K., Miller, T. L., Bates, T. S., Ogren, J. A., Andrews, E. and Shaw, G. E.: A 3 year record of simultaneously measured aerosol chemical and optical properties at
- 1725 Barrow, Alaska, J. Geophys. Res. Atmos., 107(D11), doi:10.1029/2001JD001248, 2002.
- 1726
- 1727 Quinn, P. K., Collins, D. B., Grassian, V. H., Prather, K. A. and Bates, T. S.: Chemistry
- and Related Properties of Freshly Emitted Sea Spray Aerosol, Chem. Rev., 115(10),
- 1729 4383–4399, doi:10.1021/cr500713g, 2015.
- 1730
- 1731 Rap, A., Scott, C. E., Spracklen, D. V., Bellouin, N., Forster, P. M., Carslaw, K. S.,

Schmidt, A. and Mann, G.: Natural aerosol direct and indirect radiative effects, Geophys.
Res. Lett., 40(12), 3297–3301, doi:10.1002/grl.50441, 2013.

- 1734
- 1735 Riccobono, F., Schobesberger, S., Scott, C. E., Dommen, J., Ortega, I. K., Rondo, L.,
- 1736 Almeida, J., Amorim, A., Bianchi, F., Breitenlechner, M., David, A., Downard, A.,
- 1737 Dunne, E. M., Duplissy, J., Ehrhart, S., Flagan, R. C., Franchin, A., Hansel, A., Junninen,
- 1738 H., Kajos, M., Keskinen, H., Kupc, A., Kupiainen, O., Kürten, A., Kurtén, T., Kvashin,
- 1739 A. N., Laaksonen, A., Lehtipalo, K., Makhmutov, V., Mathot, S., Nieminen, T., Olenius,
- 1740 T., Onnela, A., Petäjä, T., Praplan, A. P., Santos, F. D., Schallhart, S., Seinfeld, J. H.,
- 1741 Sipilä, M., Spracklen, D. V, Stozhkov, Y., Stratmann, F., Tomé, A., Tsagkogeorgas, G.,
- 1742 Vaattovaara, P., Vehkamäki, H., Viisanen, Y., Vrtala, A., Wagner, P. E., Weingartner, E.,
- 1743 Wex, H., Wimmer, D., Carslaw, K. S., Curtius, J., Donahue, N. M., Kirkby, J., Kulmala,
- 1744 M., Worsnop, D. R., Baltensperger, U. U. ., Schobesberger, S., Scott, C. E., Dommen, J.,
- 1745 Ortega, I. K., Rondo, L., Almeida, J., Amorim, A., Bianchi, F., Breitenlechner, M.,
- 1746 David, A., Downard, A., Dunne, E. M., Duplissy, J., Ehrhart, S., Flagan, R. C., Franchin,
- 1747 A., Hansel, A., Junninen, H., Kajos, M., Keskinen, H., Kupc, A., Kürten, A., Kvashin, A.
- 1748 N., Laaksonen, A., Lehtipalo, K., Makhmutov, V., Mathot, S., Nieminen, T., Onnela, A.,
- 1749 Petäjä, T., Praplan, A. P., Santos, F. D., Schallhart, S., Seinfeld, J. H., Sipilä, M.,
- Spracklen, D. V, Stozhkov, Y., Stratmann, F., Tomé, A., Tsagkogeorgas, G., et al.:
  Oxidation Products of Biogenic Emissions Contribute to Nucleation of Atmospheric
- 1751 Oxidation Products of Biogenic Emissions Contribute to Nucleation of Atr 1752 Particles, Science, 344(May), 717–721 [online] Available from:
- Particles, Science, 344(May), 717–721 [online] Available from:
   http://www.sciencemag.org/content/344/6185/717.abstract, 2014.
- 1754

1755 Riddick, S. N., Dragosits, U., Blackall, T. D., Daunt, F., Wanless, S. and Sutton, M. A.:

- 1756 The global distribution of ammonia emissions from seabird colonies, Atmos. Environ.,
- 1757 55, 319–327, doi:10.1016/j.atmosenv.2012.02.052, 2012a.
- 1758

1759 Riddick, S. N. Dragosits, U., Blackall, T.D., Daunt, F., Wanless, S., Sutton, M.A.: Global

- ammonia emissions from seabirds. NERC Environmental Information Data
- 1761 Centre, https://doi.org/10.5285/c9e802b3-43c8-4b36-a3a3-8861d9da8ea9, 2012b.
- 1762
- 1763 Riipinen, I., Pierce, J. R., Yli-Juuti, T., Nieminen, T., Häkkinen, S., Ehn, M., Junninen,





H., Lehtipalo, K., Petäjä, T., Slowik, J., Chang, R., Shantz, N. C., Abbatt, J., Leaitch, W. 1764 1765 R., Kerminen, V. M., Worsnop, D. R., Pandis, S. N., Donahue, N. M. and Kulmala, M.: 1766 Organic condensation: A vital link connecting aerosol formation to cloud condensation 1767 nuclei (CCN) concentrations, Atmos. Chem. Phys., 11(8), 3865-3878, doi:10.5194/acp-1768 11-3865-2011, 2011. 1769 1770 Rinaldi, M., Decesari, S., Finessi, E., Giulianelli, L., Carbone, C., Fuzzi, S., O'Dowd, C. 1771 D., Ceburnis, D. and Facchini, M. C.: Primary and Secondary Organic Marine Aerosol 1772 and Oceanic Biological Activity: Recent Results and New Perspectives for Future 1773 Studies, Adv. Meteorol., 2010, 1–10, doi:10.1155/2010/310682, 2010. 1774 1775 Russell, L. M.: Aerosol organic-mass-to-organic-carbon ratio measurements, Environ. 1776 Sci. Technol., 37(13), 2982–2987, doi:10.1021/es026123w, 2003. 1777 1778 Scott, C. E., Rap, A., Spracklen, D. V., Forster, P. M., Carslaw, K. S., Mann, G. W., 1779 Pringle, K. J., Kivekäs, N., Kulmala, M., Lihavainen, H. and Tunved, P.: The direct and 1780 indirect radiative effects of biogenic secondary organic aerosol, Atmos. Chem. Phys., 1781 14(1), 447-470, doi:10.5194/acp-14-447-2014, 2014. 1782 1783 Sharma, S., Ishizawa, M., Chan, D., Lavoué, D., Andrews, E., Eleftheriadis, K. and 1784 Maksyutov, S.: 16-year simulation of arctic black carbon: Transport, source contribution, 1785 and sensitivity analysis on deposition, J. Geophys. Res. Atmos., 118(2), 943–964. 1786 doi:10.1029/2012JD017774, 2013. 1787 1788 Sharma, S., Richard Leaitch, W., Huang, L., Veber, D., Kolonjari, F., Zhang, W., Hanna, 1789 S. J., Bertram, A. K. and Ogren, J. A.: An evaluation of three methods for measuring 1790 black carbon in Alert, Canada, Atmos. Chem. Phys., 17(24), 15225–15243, 1791 doi:10.5194/acp-17-15225-2017, 2017. 1792 Shindell, D. and Faluvegi, G.: Climate response to regional radiative forcing during the 1793 1794 twentieth century, Nat. Geosci., 2(4), 294–300, doi:10.1038/ngeo473, 2009. 1795 1796 Skrzypek, G., Wojtuń, B., Richter, D., Jakubas, D., Wojczulanis-Jakubas, K. and 1797 Samecka-Cymerman, A.: Diversification of nitrogen sources in various tundra vegetation 1798 types in the high arctic, PLoS One, 10(9), 1–21, doi:10.1371/journal.pone.0136536, 2015. 1799 1800 Steinke, M., Hodapp, B., Subhan, R., Bell, T. G. and Martin-Creuzburg, D.: Flux of the 1801 biogenic volatiles isoprene and dimethyl sulfide from an oligotrophic lake, Sci. Rep., 1802 8(1), 1-10, doi:10.1038/s41598-017-18923-5, 2018. 1803 1804 Stohl, A.: Characteristics of atmospheric transport into the Arctic troposphere, J. 1805 Geophys. Res. Atmos., 111(11), 1-17, doi:10.1029/2005JD006888, 2006. 1806 1807 Tokarek, T. W., Brownsey, D. K., Jordan, N., Garner, N. M., Ye, C. Z., Assad, F. V., 1808 Peace, A., Schiller, C. L., Mason, R. H., Vingarzan, R. and Osthoff, H. D.: Biogenic 1809 Emissions and Nocturnal Ozone Depletion Events at the Amphitrite Point Observatory on

Atmospheric Chemistry and Physics Discussions



- 1810 Vancouver Island, Atmos. Ocean, 55(2), 121–132,
- 1811 doi:10.1080/07055900.2017.1306687, 2017.
- 1812
- 1813 Tremblay, S., Picard, J.-C., Bachelder, J. O., Lutsch, E., Strong, K., Fogal, P., Leaitch,
- 1814 W. R., Sharma, S., Kolonjari, F., Cox, C. J., Chang, R. Y.-W. and Hayes, P. L.:
- 1815 Characterization of aerosol growth events over Ellesmere Island during summers of 2015
- and 2016, Atmos. Chem. Phys. Discuss, 5194(May), 2018–428, doi:10.5194/acp-2018428, 2018.
- 1818
- 1819 Tröstl, J., Chuang, W. K., Gordon, H., Heinritzi, M., Yan, C., Molteni, U., Ahlm, L.,
- 1820 Frege, C., Bianchi, F., Wagner, R., Simon, M., Lehtipalo, K., Williamson, C., Craven, J.
- 1821 S., Duplissy, J., Adamov, A., Almeida, J., Bernhammer, A. K., Breitenlechner, M.,
- 1822 Brilke, S., Dias, A., Ehrhart, S., Flagan, R. C., Franchin, A., Fuchs, C., Guida, R., Gysel,
- 1823 M., Hansel, A., Hoyle, C. R., Jokinen, T., Junninen, H., Kangasluoma, J., Keskinen, H.,
- 1824 Kim, J., Krapf, M., Kürten, A., Laaksonen, A., Lawler, M., Leiminger, M., Mathot, S.,
- 1825 Möhler, O., Nieminen, T., Onnela, A., Petäjä, T., Piel, F. M., Miettinen, P., Rissanen, M.
- 1826 P., Rondo, L., Sarnela, N., Schobesberger, S., Sengupta, K., Sipilä, M., Smith, J. N.,
- 1827 Steiner, G., Tomè, A., Virtanen, A., Wagner, A. C., Weingartner, E., Wimmer, D.,
- 1828 Winkler, P. M., Ye, P., Carslaw, K. S., Curtius, J., Dommen, J., Kirkby, J., Kulmala, M.,
- 1829 Riipinen, I., Worsnop, D. R., Donahue, N. M. and Baltensperger, U.: The role of low-
- 1830 volatility organic compounds in initial particle growth in the atmosphere, Nature,
- 1831 533(7604), 527–531, doi:10.1038/nature18271, 2016.
- 1832
- Tunved, P., Ström, J. and Krejci, R.: Arctic aerosol life cycle: Linking aerosol size
  distributions observed between 2000 and 2010 with air mass transport and precipitation at
- 1835 Zeppelin station, Ny-Ålesund, Svalbard, Atmos. Chem. Phys., 13(7), 3643–3660,
- 1836 doi:10.5194/acp-13-3643-2013, 2013.
- 1837
- 1838 Turpin, B. J. and Lim, H.-J.: Species Contributions to PM2 . 5 Mass Concentrations :
- 1839 Revisiting Common Assumptions for Estimating Organic Mass Species Contributions to
   1840 PM2 . 5 Mass Concentrations : Revisiting Common Assumptions for Estimating Organic
- 1841 Mass, Aerosol Sci. Technol., 35, 602–610, doi:10.1080/02786820119445, 2001.
- 1842
- Walker, J. T., Robarge, W. P. and Austin, R.: Modeling of ammonia dry deposition to a
  pocosin landscape downwind of a large poultry facility, Agric. Ecosyst. Environ., 185,
  161–175, doi:https://doi.org/10.1016/j.agee.2013.10.029, 2014.
- 1846
- 1847 Wang, Q., Jacob, D. J., Fisher, J. A., Mao, J., Leibensperger, E. M., Carouge, C. C., Le
- 1848 Sager, P., Kondo, Y., Jimenez, J. L., Cubison, M. J. and Doherty, S. J.: Sources of
- 1849 carbonaceous aerosols and deposited black carbon in the Arctic in winter-spring:
- 1850 Implications for radiative forcing, Atmos. Chem. Phys., 11(23), 12453–12473,
- 1851 doi:10.5194/acp-11-12453-2011, 2011.
- 1852
- 1853 Wentworth, G. R., Murphy, J. G., Gregoire, P. K., Cheyne, C. A. L., Tevlin, A. G. and
- 1854 Hems, R.: Soil-atmosphere exchange of ammonia in a non-fertilized grassland: Measured





- 1856 doi:10.5194/bg-11-5675-2014, 2014.
- 18571858 Wentworth, G. R., Murphy, J. G., Croft, B., Martin, R. V., Pierce, J. R., Côté, J. S.,
- 1859 Courchesne, I., Tremblay, J. É., Gagnon, J., Thomas, J. L., Sharma, S., Toom-Sauntry,
- 1860 D., Chivulescu, A., Levasseur, M. and Abbatt, J. P. D.: Ammonia in the summertime
- 1861 Arctic marine boundary layer: Sources, sinks, and implications, Atmos. Chem. Phys.,
- 1862 16(4), 1937–1953, doi:10.5194/acp-16-1937-2016, 2016.
- 1863 1864 Van Der Werf, G. R., Randerson, J. T., Giglio, L., Van Leeuwen, T. T., Chen, Y., Rogers,
- 1865 B. M., Mu, M., Van Marle, M. J. E., Morton, D. C., Collatz, G. J., Yokelson, R. J. and
- 1866 Kasibhatla, P. S.: Global fire emissions estimates during 1997-2016, Earth Syst. Sci.
- 1867 Data, 9(2), 697–720, doi:10.5194/essd-9-697-2017, 2017.
- 1868
- 1869 Wesely, M. L.: Parameterization of surface resistances to gaseous dry deposition in 1870 regional-scale numerical models, Atmos. Environ., 23, 1293–1304, 1989.
- 1871
- 1872 Willis, M. D., Burkart, J., Thomas, J. L., Köllner, F., Schneider, J., Bozem, H., Hoor, P.
- 1873 M., Aliabadi, A. A., Schulz, H., Herber, A. B., Leaitch, W. R. and Abbatt, J. P. D.:
- 1874 Growth of nucleation mode particles in the summertime Arctic: A case study, Atmos.
- 1875 Chem. Phys., 16(12), 7663–7679, doi:10.5194/acp-16-7663-2016, 2016.
- 1876
- Willis, M. D., Köllner, F., Burkart, J., Bozem, H., Thomas, J. L., Schneider, J., Aliabadi,
  A. A., Hoor, P. M., Schulz, H., Herber, A. B., Leaitch, W. R. and Abbatt, J. P. D.:
  Evidence for marine biogenic influence on summertime Arctic aerosol, Geophys. Res.
- 1880 Lett., 44(12), 6460–6470, doi:10.1002/2017GL073359, 2017.
- 1881
- Wilson, T. W., Ladino, L. A., Alpert, P. A., Breckels, M. N., Brooks, I. M., Browse, J.,
  Burrows, S. M., Carslaw, K. S., Huffman, J. A., Judd, C., Kilthau, W. P., Mason, R. H.,
  McFiggans, G., Miller, L. A., Najera, J. J., Polishchuk, E., Rae, S., Schiller, C. L., Si, M.,
  Temprado, J. V., Whale, T. F., Wong, J. P. S., Wurl, O., Yakobi-Hancock, J. D., Abbatt,
  J. P. D., Aller, J. Y., Bertram, A. K., Knopf, D. A. and Murray, B. J.: A marine biogenic
  source of atmospheric ice-nucleating particles, Nature, 525(7568), 234–238,
  doi:10.1038/nature14986, 2015.
- 1889
- 1890 Xu, J. W., Martin, R. V., Morrow, A., Sharma, S., Huang, L., Richard Leaitch, W.,
- Burkart, J., Schulz, H., Zanatta, M., Willis, M. D., Henze, D. K., Lee, C. J., Herber, A. B.
  and Abbatt, J. P. D.: Source attribution of Arctic black carbon constrained by aircraft and
  surface measurements, Atmos. Chem. Phys., 17(19), 11971–11989, doi:10.5194/acp-1711971-2017, 2017.
- 1895
- Yang, Q., Bitz, C. M. and Doherty, S. J.: Offsetting effects of aerosols on Arctic and
  global climate in the late 20th century, Atmos. Chem. Phys., 14(8), 3969–3975,
  doi:10.5194/acp-14-3969-2014, 2014.
- 1899
- Ye, Q., Robinson, E. S., Ding, X., Ye, P., Sullivan, R. C. and Donahue, N. M.: Mixing of secondary organic aerosols versus relative humidity, Proc. Natl. Acad. Sci., 113(45),





- 1902 12649–12654, doi:10.1073/pnas.1604536113, 2016.
- 1903
- 1904 Yu, H., Kaufman, Y. J., Chin, M., Feingold, G., Remer, L. A., Anderson, T. L.,
- 1905 Balkanski, Y., Bellouin, N., Boucher, O., Christopher, S., DeCola, P., Kahn, R., Koch,
- 1906 D., Loeb, N., Reddy, M. S., Schulz, M., Takemura, T. and Zhou, M.: A review of
- 1907 measurement-based assessments of the aerosol direct radiative effect and forcing, Atmos.
- 1908 Chem. Phys., 6(3), 613–666, doi:10.5194/acp-6-613-2006, 2006.
- 19091910 Zábori, J., Rastak, N., Yoon, Y. J., Riipinen, I. and Ström, J.: Size-resolved cloud
- 1910 condensation nuclei concentration measurements in the Arctic: Two case studies from the
- summer of 2008, Atmos. Chem. Phys., 15(23), 13803–13817, doi:10.5194/acp-15-13803-
- 1913 2015, 2015.
- 1914
- 1915 Zender, C. S.: Mineral Dust Entrainment and Deposition (DEAD) model: Description and
- 1916 1990s dust climatology, J. Geophys. Res., 108(D14), 4416, doi:10.1029/2002JD002775,
- 1917 2003.
- 1918