$\frac{1}{2}$ 

Answer to the comments by #1 Referee

General Comment: More quantitative discussion based on a statistics analysis recommended,
 because it seems that model intercomparison is the main purpose of this paper. In
 addition, more detailed information about observation and model's differences is
 necessary.

Since the purpose of this paper is to elucidate the essential element causing the different output for  $O_3$ , and not to discuss the superiority or inferiority of the models, quantitative comparison among the models and with observation are not made. Such quantitative discussions have been made in the overall intercomparison paper for  $O_3$  by Li et al. to be submitted in this special issue. However, some more information on the difference with observation has been added according to the suggestion of the reviewer.

13

L65: What is the difference between "mixing ratios of surface ozone" and "concentration of
ozone"? Both of "mixing ratio" and "concentration" are mixed in the manuscript.

In principle, ppb, ppm etc. should be called "mixing ratio" and µg/m<sup>3</sup>, mol/L, etc, should be called "concentration". However, the quantity in ppb unit is sometimes customary called "concentration". In this paper, we use only the unit of "ppbv", so that we unified the terminology to "mixing ratio" in order to void mixed use.

20

L79-82: Why the vertical structures are different among the three models, nevertheless the
 meteorological fields were derived from the same WRF output?

Although WRF fields are common to the three models, convective model for boundary
layer is different between CMAQ and NAQM, and some parameters within the module are
different between the CMAQ v. 5.0.2 and v.4.7.1, which causes the different vertical
structure of O<sub>3</sub>.

27

L87-88: It is introduced that every model adopted MIX. However, the predicted O<sub>3</sub> seems to
be affected if the ratio of NO/NO<sub>2</sub> in NO<sub>x</sub> emission and the speciation of VOCs emission
were not unified in the model-ready emission input.

<b>91</b>	Among the three models. NOANO motion in NO amining and an and a second back
31	Among the three models, $NO/NO_2$ ratios in $NO_x$ emission are set commonly as 0.9.
32	Speciation of VOC is the same for the CMAQ v.5.0.2 and v.4.7.1 since they used the
33	common chemistry sub-model, SAPRC99, but it is different form NAQM which employed
34	CMB-Z. Yes, it is true that the difference in speciation and reaction mechanism causes the
35	difference in predicted O <sub>3</sub> as discussed in the first part of Discussion.
36	
37	L91: What is the reason for using two global models for the intercomparison study?
38	Each modeler wants to use accustomed global model for providing the boundary for
39	easiness. Reflecting such request of participating modelers, the project accepted to use
40	either of the two global models after confirming their output does not differ much in East
41	Asian region (page 3, Lines 88-90).
42	
43	L105: I guess CMAQ v4.7.1 does not include AERO6. It may be up to AERO5.
44	Yes, the reviewer is right. CMAQ v.4.7.1 includes AERO5. Line 102 has been corrected.
45	
46	L115-116: Only monthly averaged diurnal variations are introduced for a model evaluation. A
47	table of statistics for hourly comparison (including mean observation, mean simulation,
48	normalized mean bias, correlation coefficient in different season and whole observation
49	period) seems to be necessary, and the discussion should be more quantitative because
50	model intercomparison is the main purpose of this paper.
51	As noted above in the answer to the general comments, the purpose of this paper is not for
52	the so-called model intercomparison per se, but for trying to identify the possible model
53	elements that causes much different outputs of O3 concentration even though using the
54	common emissions, meteorological fields and boundary conditions. The quantitative
55	statistical analyses have been made in the overall intercomparison paper by Li et al. for the
56	submitted all models including the three models in this paper.
57	
58	L139-140: More details about the observation conducted by IAP are necessary (e.g.
59	monitoring equipment, height of IAP tower and its location).
60	More detailed information on the observational sites and equipments by IAP has been
61	added on page 5 (Lines 140-147) as follows.

 $\mathbf{2}$ 

62 "The  $O_3$  and  $NO_x$  instrument at IAP site (39.9N, 116.3E) were on the rooftop of a building 63 (8 m above the ground), which is a urban site surrounded by residential infrastructure and 64 freeway in the east (360m). Yangfang (40.2N, 116.1E) is a suburban site in the north of 65 Beijing, 40 km away from IAP. The instruments were 10 m above the ground on the 66 campus of a university with little influence by local sources and sinks. The  $O_3$  and  $NO_x$ 67 instruments were an ultraviolet photometric analyzer (Model 49i, Thermo Fisher Scientific 68 (Thermo)) and a chemiluminescence analyzer (Model 42i TL, Thermo), respectively."

69

70 71

72

L158-159: Since information about the location of each site in Beijing are not introduced, it cannot be determined that the predicted results can be compared with the observation or not.

- We agree that more detailed information of the referred sites in Beijing is necessary.
  We added explanation of the sites in page 5 (Lines 147-153) as follows.
- "One of the referred Beijing data is the monthly averaged daily maximum
  concentration of O<sub>3</sub> in April and July in 2014-2015 averaged over two suburban sites,
  Daxing (39.7N, 116.4E) and Shunyi (40.1N, 116.7E) presented by Chen et al. (2015).
  Another data is the averaged diurnal variation at three urban/suburban sites, Fengtai,
  Shunyi and Baolian, in July and August in 2007, which are given in the paper by Xu et
  al. (2011). All the denoted observational sites in Beijing are located within the
  selected nine model grids shown in Fig.1."
- 82
- L237: What is the definition of "net chemical production of O<sub>3</sub>"? How did you calculate it? If it
  is just the difference from the concentration in the previous time, I guess transported O<sub>3</sub>
  is also included in the net chemical production.

86 We added the definition of "net chemical production of  $O_3$ " explicitly in the text as follow 87 on page 8 (Lines 251-256).

88 "Here, the net chemical production,  $N(O_3)$ , is calculated by the equation,  $N(O_3) = 89$ 89  $F(O_3)-D(O_3) = \{k_1[HO_2][NO] + k_2[RO_2][NO]\} - \{k_3[O(^1D)][H_2O] + k_4[OH][O_3] + k_4[HO_2][O_3]$ 90  $+k_5[O_3][olefin]\}$  in NAQM. The CMAQ models give the net chemical production as the 91 difference of O<sub>3</sub> mixing ratio between the calculation steps of chemistry module with a

process analysis package. The net chemical production was calculated in each grid and then average was taken for all the selected grids."

93 94

95 **L321:** Is "process analysis" also used for the calculation of "net chemical production of  $O_3$ "?

- 96 In this paper, the term of "process analysis" includes the evaluation of vertical and97 horizontal transport, surface deposition and net chemical production.
- 98

**L322:** It is recommend to prove "the horizontal transport has been found to be nearly the same
for the three models" in this manuscript.

101 The horizontal transport at each region in spring and summer has been evaluated, and it has 102 been found there is not much difference among the three models. This is possibly due to 103 the fact that horizontal transport is mainly controlled by the WRF fields, which are 104 common to all three models. In order to reduce the number of figures, the graphs for the 105 comparison of horizontal transport was not included in the paper. The text is slightly 106 modified as follows for clarification (page 11, Lines 341-343).

107 Original version: Since the horizontal transport has been found to be nearly the same for
108 the three models, only the vertical transport will be discussed here for discussion."

109**Revised Version:** "Since it has been found that there is not much difference in horizontal110transport and surface deposition, and the chemical mechanisms of CMAQ 5.0.2 and111CMAQ 4.7.1 are the same, the difference in the model performance must be ascribed to the112difference in vertical transport processes."

- 113
- L333: I am afraid, but the discussion is unclear about a reason why only CMAQv5.0.2
  reproduced largest downward fluxes of O<sub>3</sub>.

116The difference between CMAQ 5.0.2 and 4.7.1 in the treatment of vertical transport of O3117has been clearly stated in the Methods, and included in the discussion. The added sentences118are as follows.

"For the computation of the vertical transport for advection, CMAQ 5.0.2 used PPM
(piecewise parabolic method) scheme, as compared to CMAQ 4.7.1 and NAQM which
used the vertical velocity directly from WRF." (page 4, Lines 109-111)

"Here, it should be noted that the vertical transport was computed in the PPM scheme in
CMAQ 5.0.2 instead of the direct extraction from WRF in CMAQ 4.7.1 as described in the
section of Methods. The PPM method has been known to introduce more downward flux of
O<sub>3</sub> from higher layer to the surface layer. " (page 12, Lines 361-365)

126

L349-350: Why could you conclude that "the difference in the model performance must be
 ascribed to the difference in transport processes"? Sorry, but discussion is unclear on it.

The difference in the two models (CMAQ v.5.0.2 and CMAQ v.4.7.1) should be due to either of the processes: vertical transport, horizontal transport, surface deposition, and chemical processes. Among them, as noted above, there is not much difference in horizontal transport and surface deposition, and the chemistry model is the same for these models. Therefore, we concluded that the difference in the model performance must be mainly ascribed to in the vertical transport process. The sentence is slightly modified for clarification as follows (page 11, Lines 341-344).

Original version: "Since the horizontal transports in the selected regions and seasons have
been found to be nearly the same for all the three models, only the difference in the vertical
transport will be discussed here."

Revised version: "Since it has been found that there is not much difference in horizontal transport and surface deposition, and the chemical mechanisms of CMAQ 5.0.2 and CMAQ 4.7.1 are the same, the difference in the model performance must be ascribed to the difference in vertical transport processes."

143

144 **Fig. 2 (a):** *Please modify "Mixing Raio" to "Mixing Ratio".* 

- 145 Thank you for alerting. We corrected the miss spelling.
- 146

147 **Fig. 5 (b):** *Please modify "ppb/hr" to "ppbv/hr"*.

- 148 Thank you for alerting. We corrected the miss spelling.
- 149

150 **Fig. 7&8:** What is the difference between "mixing ratio" and "concentration"?

151 Thank you for alerting. We unified them to "mixing ratio".

152 Answer to the comments by #2 Referee

153

154 1) The employed horizontal resolution is 45 km for all the models, and the highest height and 155 number of vertical layers are 45 km and 40 layers for the CMAQ models and 20 km and 156 20 layers for the NAQM so that the vertical resolution in the troposphere is about the 157 same. The top height of 45 km sounds really high. What is the top pressure of the CMAQ 158 model?

- 159 The top pressure of the CMAQ is ca. 52-53 hPa.
- 160
- 161 2) Line 105: CMAQ v.4.7.1 and v.5.0.2 included AERO6 There was no AERO6 mechanism in
   162 CMAQ v4.7.1. 3
- 163 We corrected AERO6 to AERO5 for CMAQ5.7.1.
- 164
- 3) The quality of the figures might be improved: 1) Figure 2: It should be better if the labels
  were added inside each plot. For example, O<sub>3</sub> in April, O<sub>3</sub> in July, NO in April, NO in
  July can be added on the top corner of each panel
- 168 Thank you for the comment. We improved the quality of Figures checking all of them.
- 169
- 170 4) Line 197: A morning peaks (grammar issue)
- 171 Thank you for your alert. We corrected.
- 172

5) Line 237: Figures 5 (a)-(d) show the net chemical production of O3 in Beijing and Tokyo in
April and July calculated in this study. More details need to be described. For example,
how did the authors calculate the net chemical production? Did the authors add some
diagnostic equations or use some internal diagnostic packages to get the net chemical
production? Did the authors calculate the production in each grid and did average of all
the grids at the end?

We agree with the reviewer comments and we added the definition of "net chemical production of  $O_3$ " explicitly in the text. We calculated the production in each grid and did average of all the grids at the end. The added sentences are on page 8 (Line 251-256). 182 "Here, the net chemical production,  $N(O_3)$ , is calculated by the equation,  $N(O_3) =$ 

183  $F(O_3)-D(O_3)=\{k_1[HO_2][NO]+k_2[RO_2][NO]\}-\{k_3[O(^1D)][H_2O]+k_4[OH][O_3]+k_4[HO_2][O_3]$ 

184  $k_5[O_3][olefin]\}$  in NAQM. The CMAQ models give the net chemical production as the 185 difference of O<sub>3</sub> mixing ratio between the calculation steps of chemistry module with a 186 process analysis package. The net chemical production was calculated in each grid and 187 then average was taken for all the selected grids."

- 188
- 189 6) Line 284: observational Typo: observational

190 We corrected.

191

1927) Line 250-253: The authors tried to explain the overestimation in Fig. 2b and Fig. 3b for193NAQM. The peak in Fig. 5b,d seems to support the overestimation. However, I feel the194evidence is not strong. In Fig. 5d, the net reaction is negative, I am not sure how the195negative production contributes to the ozone overestimation. In addition, the morning196peak is obvious in Fig. 5a,c as well, why is there no overestimation in Fig. 2a and Fig. 3a197if the early morning peak may result in the over shooting of ozone?

We agree with the reviewer that the explanation of  $O_3$  overestimation by NAQM in early morning by the morning peaks of net-chemical production is not very strong since no quantitative sensitivity check has been done in this study. We modified the expression as follows (page 9, Lines 270-276).

202Original version: "It can be noted that net O3 production of NAQM shows a second peak203in early morning after breaking of dawn in both Beijing and Tokyo in July, which would be204a cause of overestimate of O3 in the morning by NAQM simulation as seen in Fig. 2(b) and205Fig. 3(b). The cause of the early morning peak of net O3 production in NAQM might be206due to the photolysis of higher HONO that is produced by the heterogeneous reaction of207NO2, although it has not been quantified in the present study."

Revised version: "It can be noted that net O<sub>3</sub> production of NAQM shows a peak in early
morning after breaking of dawn in both Beijing and Tokyo, which could be a cause of
overestimate or earlier rise of O<sub>3</sub> in the morning by NAQM simulation as seen in Fig. 2(a),
(b) and Fig. 3(a), (b) although the effect is marginal in the case of Beijing in April. The

214

cause of the early morning peak of net  $O_3$  production in NAQM might be due to the photolysis of higher HONO that is produced by the heterogeneous reaction of NO<sub>2</sub>. More quantitative sensitivity analysis should be performed to confirm these effects. "

- 215
- 8) Line 257-259: In April, net chemical production of O<sub>3</sub> is in general negative for all the
  models both in Beijing and Tokyo except for CMAQ 4.7.1 around midday and NAQM in
  early morning showing slight positive values. I feel the descriptions are not accurate. In
  April (Fig. 5a,c), both CMAQ5.0.2 and CMAQ 4.7.1 shows substantial positive net
  chemical production of O3 in Tokyo. Please double check the statement.

We appreciated the reviewer's check. We made a mistake of Fig. 5(c) and (d) were placed interchanged. Thus, in April in Tokyo net chemical production is negative for all the models.

- 224
- 2259) Line 348-350: Since the chemical mechanisms of CMAQ 5.0.2 and CMAQ 4.7.1 are the 226 same, the difference in the model performance must be ascribed to the difference in 227transport processes. The authors concluded that the chemical mechanism of CMAQ 5.0.2 228and CMAQ 4.7.1 are the same, then why is there large differences in the O3 chemical 229production based on Fig. 5? The section of "Comparison of Chemical Mechanism 230Sub-Modules" mainly compared the mechanism between SAPRC99 (CMAO 5.0.2 and 231CMAQ 4.7.1) and CBM-Z (in NAQM), but discussed relatively little about the chemical 232production differences between CMAQ 5.0.2 and CMAQ 4.7.1 (Fig. 5). Any 233explanations?

Net chemical O<sub>3</sub> production reflects not only chemical reaction mechanism but also
concentrations of each relevant compound, which is affected by transport processes as well.
Therefore, we think the differences of net chemical O<sub>3</sub> production between CMAQ 5.0.2
and 4.7.1 as shown in Fig. 5 are due to the difference of concentrations of relevant species.

238

239 10) Fig. 2b: There is a line with yellow line, which should be the red line. Please double check.

240 We appreciated the alert. We modified the Figure.

241

242 11) Line 382: "course" should be "coarse"

243	We corrected.
244	
245	12) Line 383: "it would not enough" should be "it would not be enough"
246	We corrected.
247	
248	over
249	
250	
251	
252	
253	

254		
255		Comparison of Surface Ozone Simulation among Selected Regional Models
256		in MICS-Asia III
257		- Effects of Chemistry and Vertical Transport for the Causes of Difference-
258		
259		Hajime Akimoto <sup>1</sup> , Tatsuya Nagashima <sup>1</sup> , Li Jie <sup>2</sup> , Joshua S. Fu <sup>3</sup> , Dongsheng Ji <sup>2</sup>
260		Jiani Tan <sup>3</sup> , and Zifa Wang <sup>2</sup>
261		
262		
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267		
268	Ab	ostract
269		In order to clarify the causes of variability among the model outputs for surface ozone in the
270	Mo	odel Intercomparison Study Asia Phase III (MICS-Asia III), three regional models, CMAQ
271	v.5	.0.2, CMAQ v.4.7.1 and NAQPMS (abbreviated as NAQM in this paper) have been selected.
272	De	tailed analyses of monthly averaged diurnal variation have been performed for selected grids
273	cov	vering the metropolitan areas of Beijing and Tokyo and at a remote oceanic site, Oki. The
274	che	emical reaction mechanism, SAPRC99, used in the CMAQ models tended to give a higher
275	net	chemical ozone production than CBM-Z used in NAQM, agreeing with previous studies.
276	Inc	lusion of the heterogeneous "renoxification" reaction of HNO <sub>3</sub> (on soot surface) $\rightarrow$ NO +
277	NC	$D_2$ only in NAQM would give a higher NO concentration resulting in a better agreement with
278	obs	servational data for NO and nighttime O3 mixing ratios. In addition to chemical processes, the
279	dif	ference in the vertical transport of $O_3$ was found to affect the simulated results significantly.
280	Par	ticularly, the increase in downward $O_3$ flux from the upper layer to the surface after dawn
281	wa	s found to be substantially different among the models. Larger early morning vertical
282	tra	nsport of $O_3$ simulated by CMAQ 5.0.2 is thought to be the reason for higher daytime $O_3$ in
283	Jul	y in this model. All the three models overestimated the daytime ozone by ca. 20 ppbv at the
284	ren	note site Oki in July, where in situ photochemical activity is minimal.
285		

## 286 Introduction

287In the Model Intercomparison Study Asia Phase III (MICS-Asia Phase III), one of the 288targets was to narrow down the difference in the model simulation results by using common key 289input parameters such as precursor emissions, meteorological fields, and boundary conditions to 290allow a more focused discussion on the causes of the difference among model outputs. In most 291of the past model intercomparison studies for chemical transport models (CTM) for air quality, 292such key parameters were not common to all the models, which made the discussion of the 293causes of the differences among the model outputs difficult, and the results often demonstrated 294that the ensemble mean of simulated mixing ratios agreed reasonably well with observations 295even though the disagreement among the models were often significantly large (for example, 296Han et al., 2008; Fiore et al., 2009).

297 In order to improve the situation of model intercomparison studies, participants of the 298MICS-Asia III studies agreed to use common emission data (Li, M. et al., 2017), meteorological 299fields (specified Weather Research and Forecasting Model (WRF)) and boundary conditions by 300 either of two global CTM (GEOS-Chem and CHASER) provided within the project (Wang et al. 301 overview paper; Li, J. et al. to be published in this special issue). The following 12 regional 302 models have been submitted to the MICS-Asia III using the designated common emissions, 303 meteorological fields and boundary conditions: six WRF-CMAQ (Community Multiscale Air 304 Quality Modeling System, two v.5.0.2, one v.5.0.1 and three v.4.7.1), four WRF-Chem 305 (Weather Research and Forecasting (WRF) model coupled with Chemistry), one WRF-NHM 306 (JMA NonHydrostatic Model)/Chem and one WRF-NAQPMS (Nested Air Quality Prediction 307 Modeling System, which is abbreviated to NAQM in this paper hereafter for simplicity). It 308 turned out, however, that even though these 12 models used the specified common key input 309 components, large variabilities in the spatial distribution and absolute mixing ratios among the 310 models were found for ozone  $(O_3)$  (Li, J. et al., in this special issue).

311 In the present study, three regional models, two WRF-CMAQ, v.5.02 and v.4.7.1, and 312 WRF-NAQM were selected among the 12 above mentioned models to elucidate the causes of 313 differences, and detailed comparisons were made for selected grids covering the metropolitan 314 areas of Beijing and Tokyo, and at a remote oceanic site at Oki in April and July in 2010. We 315 selected the two models of CMAQ because CMAQ models have been widely used to assess the 316 air quality for ozone in Asia (e.g. Yamaji et al., 2008; Kurokawa et al., 2009; Fu et al., 2012), 317 and the difference in the simulated results between different versions (v.5.02 and v.4.7.1) is of 318 concern. Furthermore, we selected WRF-NAQM because this is one of the regional CTMs

319 developed in Asia, giving substantially lower mixing ratios of surface ozone as compared to 320 most of WRF-CMAQ including the selected two models (Li, J. et al., in this special issue). The 321 metropolitan areas of the two megacities of Beijing and Tokyo have been selected for the 322comparison to test whether regional models can be applied to the mitigation policy of urban 323 ozone pollution. Oki, an EANET (Acid Deposition Monitoring Network in East Asia) 324 monitoring station located in the southern part of the Sea of Japan, was selected as a remote 325 reference site located between the two megacities, as in situ photochemical production of O<sub>3</sub> is 326 known to be minimal there (Jaffe et al., 1996).

327

## 328 Models

329 Basic features and the simulated whole domain of the regional models, CMAQ v.4.7 (Foley 330 et al., 2010), v.5.0 (CMAS, 2011), and NAQM (Li, J. et al., 2016) used in this study, are given 331 elsewhere in this issue (Li, J. et al., in this special issue). The employed horizontal resolution 332 was 45 km for all the models, and the maximum height and number of vertical layers were 45 333 km and 40 layers, respectively, in the CMAQ models and 20 km and 20 layers, respectively, in 334 the NAQM, so that the vertical resolution in the troposphere was about the same. The lowest 335layer for which the simulated data of ozone were extracted in this paper was 50 m from the 336 ground. Model calculations by the CMAQ v.5.0.2, v.4.7.1, and NAQM were conducted at the 337 University of Tennessee, USA, National Institute for Environmental Studies, Japan, and 338 Institute of Atmospheric Physics, China, respectively. All the models used the common 339 meteorological fields from WRF simulation and common emissions of MIX  $(0.25^{\circ} \times 0.25^{\circ})$  for 340 2010 (Li, M. et al., 2017) developed in the MICS-Asia III project. The initial and boundary 341conditions were supplied by global models, CHASER for CMAQ v.4.7.1 and NAQM, and 342 GEOS-Chem for CMAQ v.5.0.2. It was agreed that either CHASER or GEOS-Chem may be 343 used in the MICS-Asia III, since they were confirmed to give reasonably good agreement for 344 the O<sub>3</sub> field in the Asian domain.

Other than these three key components (emissions, meteorological field, and boundary conditions), the three models employed different sub-models and parameters for e.g. the gas-phase and aerosol chemistry module, dry deposition parameters, boundary layer scheme, etc. As for the gas-phase chemistry, CMAQ v.4.7.1 and v.5.0.2 incorporated SAPRC99 (Carter, 2000), and NAQM employed CBM-Z (Zaveri and Peters, 1999). In the CMAQ v.4.7.1 (Foley et al., 2010), major upgrades were made on the aerosol treatment from the previous version: (a) updates to the heterogeneous  $N_2O_5$  parameterization, (b) improvement in the treatment of 352 secondary organic aerosol (SOA), (c) inclusion of dynamic mass transfer for coarse-mode 353 aerosol, and (d) revisions to the cloud model. The NAQM and CMAQ v.4.7.1 employed 354 ISORROPIA v.1.7 (Nenes et al, 1998), and CMAQ v.5.0.2 incorporated ISORROPIA v.2.1 355(Fountoukis and Nenes, 2007) for inorganic aerosol chemistry modules. In addition, CMAQ 356 v.4.7.1 and v.5.0.2 included AERO5 and AERO6 (Binkowski and Rosselle, 2003), respectively, 357as an organic aerosol chemistry module. The sub-modules for dry deposition and wet deposition 358 employed in the three models were essentially the same. The Asymmetric Convective Model 359 version 2 (ACM2) for the planetary boundary layer (PBL) (Pleim, 2007) was employed in both 360 CMAQ v.4.7.1 and v.5.0.2. The Yonsei University (YSU) Boundary Layer (BL) scheme was 361 used for calculating BL height for NAQM (Li, J. et al., 2016). As for the advection module, the 362 models by Yamartino (1993), and Walcek and Aleksic (1998) were used for CMAQ (v.4.7.1 363 and v.5.0.2) and NAQM, respectively. For the computation of the vertical transport for 364 advection, CMAQ v.5.0.2 used the PPM (piecewise parabolic method) scheme, as compared to 365 CMAQ v.4.7.1, which used the vertical velocity directly from WRF.

366

## 367 Comparison Domain and Observational Data

368 All the comparisons between the model simulations and the model using observational data 369 were made for monthly averaged diurnal variations in the mixing ratios of O<sub>3</sub> and NO in April 370and July. April and July were chosen here because in situ photochemical build-up of  $O_3$  in April 371 is insignificant but the daytime maximum mixing ratio of O<sub>3</sub> is relatively high reflecting the 372 well-known spring maximum of  $O_3$  for the background in the northern hemisphere including 373 East Asia (Monks, 2000; Pochanart et al., 2003), while in July, a much higher in situ 374 photochemical build-up of O<sub>3</sub> is observed in urban areas in East Asia. Two representative 375megacities, Beijing and Tokyo, were selected as urban areas for the comparison. As a remote 376 reference site, Oki, an EANET site situated between Beijing and Tokyo, was selected. The Oki 377 site is located on a cliff of an island where the local emissions of NO<sub>x</sub> and VOCs are 378 insignificant so that in situ production of  $O_3$  is also minimal (Jaffe et al., 1996; Pochanart et al., 379 2002). Since the NO levels at Oki are too low to get any meaningful data using the conventional 380chemiluminescence NO<sub>x</sub> monitoring instrument, comparison with modeling results was made in 381 this study only for  $O_3$  at this site. All the calculations were conducted for the whole year of 2010 382 using the meteorological field and emission data for this year.

The domains of Beijing, Tokyo and Oki site were centered at 39.9°N, 116.3°E; 36.0°N, 139.3°E and 36.3°N and 133.1°E, respectively. The selected domains for Beijing and Tokyo 385 consisted of 9 ( $3\times3$ ), and 3 (2+1) grids, respectively, covering the metropolitan areas of the 386 cities as shown in Fig. 1. Data of a single grid covering the island was used for the Oki site. The 387 observational data used for Tokyo were 1-hr averaged values in 2010 of the average of 118 (for 388 O<sub>3</sub>) and 126 (for NO) non-roadside monitoring stations within the selected grid (Fig. 1). The 389 data were obtained from Atmospheric Environment Monitoring Data Files in the Environmental 390 Information Database stored at the National Institute for Environmental Studies (NIES), Japan. 391 In Beijing, unfortunately, no routine monitoring data of 1-hr averaged values of  $O_3$  in 2010 are 392 open to the public. Therefore, unpublished data from two sites (IAP tower campus and 393 Yangfang) obtained by IAP, and literature values published in Xu et al. (2011) and Chen et al. 394 (2015) have been referred to in this work. The  $O_3$  and  $NO_x$  instruments at the IAP site (39.9N, 395 116.3E), which is an urban site surrounded by residential infrastructure and freeway in the east 396 (ca 200 m), were on the rooftop of a building (10 m above the ground). Yangfang (39.5N, 397 116.7E) is a suburban site in the north of Beijing, ca 40 km away from IAP. The instruments 398 were 10 m above the ground on the campus of a university with little influence from local 399 sources and sinks. The  $O_3$  and  $NO_8$  instruments were an ultraviolet photometric analyzer (Model 400 49i, Thermo Fisher Scientific (Thermo), USA) and a chemiluminescence analyzer (Model 42i 401 TL, Thermo, USA), respectively. One of the referred Beijing data is the monthly averaged daily 402maximum concentration of  $O_3$  in April and July in 2014-2015 averaged over two suburban sites, 403 Daxing (39.7N, 116.4E) and Shunyi (40.1N, 116.7E) presented by Chen et al. (2015). Another 404 data is the averaged diurnal variation at three urban/suburban sites, Fengtai, Shunyi and Baolian, 405 in July and August in 2007, which are given in the paper by Xu et al. (2011). All the denoted 406observational sites in Beijing are located within the selected nine model grids shown in Fig.1.

407The observational data for Oki is the 1-hr averaged EANET data in 2010 provided on408request by the Network Center, Asia Center for Air Pollution Research (ACAP)409(http://www.acap.asia).

410

## 411 **Results**

Figures 2(a)-(d) depict the simulated and observed mixing ratios of the monthly averaged diurnal variations of the O<sub>3</sub> and NO concentrations in April and July in Beijing, and Figs. 3(a)-(d) show similar results in Tokyo. The comparisons of the values simulated by the CMAQ 5.0.2 and 4.7.1 (hereafter, "v." for version will be omitted for simplicity) and NAQM are plotted in each figure together with the observational data.

Fig. 2 Fig. 3 417 In Beijing, observational data of surface ozone at the routine monitoring stations managed 418 by the Beijing municipal government were, unfortunately, not available until 2013 (Chen et al., 419 2015). The average of two observational data sets obtained by IAP in 2010 is marked by the 420 dashed lines with filled circles in Figs. 2(a) and (b) for O<sub>3</sub> and in Figs. 2(c) and (d) for NO. 421Other published observational data of diurnal variation of  $O_3$  in Beijing in April are available by 422Xu et al. (2011) at four sites, two urban (Fengtai and Baolian), one suburban (Shunyi), and one 423 rural (Shangdianzi) in summer (21 June-12 September) in 2007. Since the diurnal variation of 424 the urban and suburban sites is consistent, the average of these three sites is plotted in Fig. 2(b), 425marked by a dashed line with triangles. No monthly average diurnal variation of O<sub>3</sub> is available 426 for April in Beijing in the literature. Chen et al (2015) reported the monthly averaged daily 427maximum mixing ratio of  $O_3$  to be ca. 60 ppbv at an urban site (Dongsi), and ca. 75 and ca. 65 428ppbv at two suburban sites (Daxing and Shunyi, respectively) within the selected grids in this 429 study. If we simply take the average of these three values, the daily maximum mixing ratio is ca. 430 65 ppbv (not shown in Fig. 2(a)). Only the IAP data are plotted for NO with solid lines in Figs. 431 2(c) and (d).

432 As can be seen in Figs. 2(a), (b), and Figs. 3(a), (b), the diurnal pattern of the simulated 433 surface ozone shows a maximum in the late afternoon around 14-16 o'clock local time in both 434 Beijing and Tokyo, agreeing well with the observations. The simulated mixing ratios of  $O_3$  by 435CMAQ 4.7.1 are the highest, and those simulated by NAQM are the lowest both in Beijing and 436 in Tokyo in both April and July. The diurnal variations of O<sub>3</sub> simulated by CMAQ 4.7.1 are in 437 parallel with the NAQM values for whole days in all cases, but the predicted mixing ratios by 438CMAQ 4.7.1 are by ca. 20 ppbv and ca. 40 ppbv higher than those predicted by NAQM in April 439and July, respectively, both in Beijing and in Tokyo. The  $O_3$  mixing ratios predicted by CMAQ 4405.0.2 have a peculiar seasonal characteristics, i.e., the mixing ratio is slightly higher but close to 441 that predicted by NAQM within 10 ppbv both in Beijing and in Tokyo in April, whereas in July 442the daytime O<sub>3</sub> maximum predicted by CMAQ 5.0.2 is very close to that predicted by CMAQ 443 4.7.1, much higher than the value by NAQM. In Tokyo, the simulated mixing ratios of CMAQ 444 5.0.2 and NAQM are closer to the observations in April, and NAQM gives a closer matching 445with observations in July, while CMAQ 4.7.1 overestimates the values in both months as shown 446 in Figs. 2 (b) and 3(b). A comparison with the observations will be discussed later including the 447 uncertainty of the observational data in Beijing.

The observed mixing ratios of NO show a peak value at around 7 AM, a decrease during morning followed by a slow decay in the afternoon, and they start to build up during nighttime 450both in April and in July, in both Beijing and Tokyo. The peak values of the mixing ratios in the

451

morning are ca. 13-14 and 6 ppbv in April and ca. 11 and 5-6 ppbv in July in Beijing and Tokyo, 452respectively. The minimum mixing ratios in the evening are ca. 1.7 and 1.4 ppbv in April and 4532.3 and 1.3 ppbv in July in Beijing and Tokyo, respectively. Thus, it can be noted that the NO 454mixing ratios in Beijing are nearly the double of those in Tokyo.

455The simulated mixing ratios of NO are generally in the order of NAQM > CMAQ 5.0.2 > 456CMAQ 4.7.1., but they vary considerably among the models. In April, CMAQ 5.0.2 gives 457morning peak values of 13-14ppbv in Beijing and ca. 5 ppbv in Tokyo, which agrees well with 458the observations. NAQM overpredicts the NO mixing ratio in April in Beijing but gives a 459reasonable agreement with the observations in Tokyo as shown in Figs 2(c) and 3(c). In contrast, 460 CMAQ 4.7.1 gives a broad daytime peak of only ca. 2 ppbv in Beijing and ca. 1 ppbv in Tokyo 461in April, which is quite different from other models, and it underpredicts considerably the 462 observational data. In July, only NAQM gives a morning peak mixing ratio of ca. 8 ppbv in 463 Beijing and 5.5 ppbv in Tokyo, agreeing fairly well with the observations including diurnal 464 variation (Figs. 2(d) and 3(d)). In contrast, both CMAQ 5.0.2 and 4.7.1 give morning peaks as 465low as 1-2 ppbv and nearly zero mixing ratio during nighttime, which are significantly lower 466 than the observational values.

467 It can be noted that the simulated and observed levels of  $O_3$  are highly anti-correlated with 468 those of NO. For example, the reasonably good agreements of  $O_3$  simulated by CMAQ 5.0.2 469 and NAQM in April, and by NAQM in July in Tokyo correspond to the reasonably good 470agreement of NO in each case. Much higher overestimates of O<sub>3</sub> by CMAQ 4.7.1 in April and 471by both CMAQ 5.0.2 and 4.7.1 in July correspond to the substantial underestimates of NO.

472Figures 4(a) and (b) show the monthly averaged diurnal variation of  $O_3$  mixing ratios at Oki 473in April and July, respectively. As shown in Fig. 4(a), all the three models give consistent 474mixing ratios of O<sub>3</sub> at 60-65 ppbv in April, agreeing well with observations within 10 ppbv. In 475July, although the simulated mixing ratios of O<sub>3</sub> agree well with each other within 10 ppbv, they 476 are in the range of 50-70 ppbv as compared to the observational level of 35-45 ppbv. Thus, all 477the three models overestimate the O<sub>3</sub> mixing ratio by nearly 20 ppbv. Although the 478characteristics of remote sites showing only a slight daytime build-up of  $O_3$  is well reproduced 479by the models, the substantial overestimate of the simulated  $O_3$  mixing ratio in July compared to 480 the observational values should be of concern.

481

482Discussion The causes of the differences in the simulated results among the three models mentioned above must be due to either chemical or transport processes incorporated in the models. Here, possible causes of differences of those processes are discussed individually.

## 486 Comparison of Chemical Mechanism Sub-Modules

487 One of the differences in the three models in this study is the chemical reaction mechanism 488 sub-module. CMAQ 5.0.2 and 4.7.1 incorporate SAPRC99 while NAQM employs CBM-Z. It 489 has been well known that different photochemical mechanisms used in the regional chemical 490 transport models produce different results in the prediction of O<sub>3</sub>. Jimenez et al. (2003) 491 compared seven photochemical mechanisms including CBM-IV (Gery et al., 1989) and 492SAPRC99 using a box model. Comparisons of CBM-IV, CBM-V (Sarwar et al., 2008) and 493 SAPRC99 incorporated into regional chemical transport models have been made by Faraji et al. 494(2007) and Luecken et al. (2008). The main differences among these mechanisms have been 495noted to be the lumping technique describing organic compounds into surrogate groups 496 (Jimenez et al., 2003), the differences in the products of the reaction of aromatics with OH 497 radical, and the overall branching ratio of radical generation and termination reactions (Faraji et 498al., 2007). The results of these studies gave a consistent picture that SAPRC99 gives higher 499concentrations of  $O_3$  than CBM-IV both in the box model calculation and in regional model 500simulation over the United States. The O<sub>3</sub> concentration obtained by CBM-V is reported to be 501between the CBM-IV and SAPRC99 values (Luecken et al., 2008). The reason for the higher 502concentration of  $O_3$  by SAPRC99 has been deduced to be due to the more efficient peroxy 503radical production in the photochemical reaction scheme of SAPR99 compared to those of the 504 CBM modules.

505Figures 5 (a)-(d) show the net chemical production of  $O_3$  in Beijing and Tokyo in April and 506 July calculated in this study. Here, the net chemical production,  $N(O_3)$ , was calculated by the 507 equation,  $N(O_3) = F(O_3) - D(O_3) = \{k_1 [HO_2] [NO] + k_2 [RO_2] [NO] \} - \{k_3 [O(^1D)] [H_2O] + k_4 [OH] [O_3] + k_4 [OH] [O_3] \}$ 508 $k_4$ [HO<sub>2</sub>][O<sub>3</sub>]+  $k_5$ [O<sub>3</sub>][olefin]} in NAQM. The CMAQ models give the net chemical production 509 as the difference in the O<sub>3</sub> mixing ratio between the calculation steps of the chemistry module 510with a process analysis package. The net chemical production was calculated in each grid and 511then the average was taken for all the grids. As revealed in Figure 5, the CMAQ models gave 512higher net ozone productions than the NAQM models did, which is consistent with the results of 513earlier studies, showing that the photochemical reaction scheme of SAPRC99 gives a higher  $O_3$ 514production than do the CBM modules. The reaction scheme of CBM-Z is the revision of

Fig. 5

515CBM-IV, and the major revision is described as (1) inclusion of revised inorganic chemistry, (2) 516explicit treatment of lesser reactive paraffins, (3) revised parameterization for reactive paraffin, 517olefin, and aromatic reactions, (4) inclusion of alkyl and acyl peroxy radical interactions and 518their reaction with NO<sub>3</sub>, (5) inclusion of organic nitrates and hydroperoxides, and (6) refined 519isoprene chemistry. Although any intercomparison including CBM-Z has not been reported, the 520overall photochemical reactivity would be assumed to be similar to CBM-V, which gives a 521higher O<sub>3</sub> value than CBM-IV and a lower value than SAPRC99. Thus, the maximum values of 522daytime net O<sub>3</sub> production in CMAQ 5.0.2 and 4.7.1 in July are ca.10 and 7-9 ppbv  $hr^{-1}$  as 523compared to ca. 6 and ca. 2 ppbv hr<sup>-1</sup> in NAQM in Beijing and Tokyo, respectively, showing 524substantially larger values for CMAQ than for NAQM.

- It can be noted that the net  $O_3$  production in NAQM shows a peak in the early morning after breaking of dawn in both Beijing and Tokyo, which could be a cause of overestimation or earlier rise of  $O_3$  in the morning by the NAQM simulation as seen in Figs. 2(a), (b) and Figs. 3(a), (b) although the effect is marginal in the case of Beijing in April. The cause of the early morning peak of net  $O_3$  production in NAQM might be due to the photolysis of higher HONO that is produced by the heterogeneous reaction of NO<sub>2</sub>. More quantitative sensitivity analyses should be performed to confirm these effects.
- 532In April, the net chemical production of  $O_3$  is, in general, negative in all the models for both 533Beijing and Tokyo, except for that in CMAQ 4.7.1 around midday and that in NAQM in early 534morning, showing slightly positive values. A tendency of higher net O<sub>3</sub> production is seen 535particularly for CMAQ 4.7.1, which may be the main cause of higher O<sub>3</sub> by this model both in 536Beijing and in Tokyo in April (Fig. 2(a) and Fig. 3(a)). The daytime net  $O_3$  production simulated by CMAQ 5.0.2 is similar to that simulated by CMAQ 4.7.1 in July but is 537538substantially lower in April. Since the chemistry mechanism of SAPRC99 is used in both 539 CMAQ versions, the difference may be related to the vertical transport of some relevant species.

# 540 Effects of heterogeneous "renoxification" reaction of HNO<sub>3</sub>

Figures 2 and 3 show the common feature of anti-correlation of  $O_3$  and NO concentrations as noted above. This feature is most clearly seen for the comparison of  $O_3$  and NO concentrations in July in both cities, demonstrating a large overestimation of  $O_3$  and a large underestimation of NO by CMAQ 4.7.1 and 5.0.2, while much lower  $O_3$  and much higher NO are estimated by NAQM. The situation in April also confirms this finding. 546 It should be noted that the rate constants of the most sensitive gas-phase reactions affecting 547 the balance of  $O_3$  and NO (Finlayson-Pitts and Pitts, 2000; Akimoto, 2016) such as,

- 548  $\operatorname{NO} + \operatorname{O}_3 \rightarrow \operatorname{NO}_2 + \operatorname{O}_2$  (1)
- 549  $NO + HO_2 \rightarrow NO_2 + OH$  (2)
- 550  $NO + RO_2 \rightarrow NO_2 + RO$  (3)

have been well established (Burkholder et al., 2015 and earlier evaluations of the series) and more or less the same reaction rates are employed in both of SAPRC99 and CBM-Z. As for the heterogeneous processes affecting NO<sub>x</sub>, the reaction,

554  $N_2O_5 + H_2O$  (on particle)  $\rightarrow 2 \text{ HNO}_3$ 

is included in common in the heterogeneous inorganic chemistry sub-module, ISORROPIA andemployed in the CMAQ and NAQM models.

(4)

It has been noted that the simulated gaseous  $HNO_3$  concentration and  $HNO_3/NO_x$  ratio were found to be 2-10 times higher when using global and regional chemical transport models than the observational data during the PEM-West (Singh et al., 1996), TRACE-P (Talbot et al., 2003), and PEM-Tropics A and SONEX (Brunner et al., 2005) aircraft campaigns over the Pacific and Atlantic Ocean. The same result has also been reported by ground observations in the remote troposphere at Mauna Loa (Hauglustaine et al., 1996) and in the polluted boundary layer of Beijing-Tianjin-Hebei region (Li, Y. et al., 2015).

564 Another concern regarding recent NO<sub>x</sub> chemistry has been focused on the high 565concentration of HONO in the urban atmosphere, which is thought to be produced by the 566 heterogeneous reaction of NO<sub>2</sub> and H<sub>2</sub>O on the aerosol and ground surface (for example, Li, Y. 567et al., 2011; Gonçalves et al., 2012; Wong et al., 2013). Inclusion of the additional 568heterogeneous source of HONO not only affects the photochemical O<sub>3</sub> formation due to the 569increase of OH radicals but also increases  $HNO_3$  due to the increase of the reaction,  $OH + NO_2$ 570  $+ M \rightarrow HNO_3 + M$ . Li, Y. et al. (2015) have shown that the inclusion of the heterogeneous 571formation of HONO gives more HNO<sub>3</sub>, which tends to give a larger overestimation of gaseous 572HNO<sub>3</sub> in the Beijing-Tianjin-Hebei region.

In order to solve the problem of overestimation of  $HNO_3$ , the heterogeneous reaction of HNO<sub>3</sub> on soot surface to reproduce NO and NO<sub>2</sub> has been proposed as "renoxification" process early by Lary et al. (1997) in the analysis of the above aircraft observation data. The heterogeneous reaction of  $HNO_3$  on soot surface to produce  $NO/NO_2$  has been confirmed experimentally in laboratory studies (Disselkamp et al., 2000; Muñoz and Rossi, 2002), although the product ratio and reaction mechanism are not well established yet. The steady state 579uptake coefficient  $\gamma_{ss}$  of this reaction has been reported to be  $(4.6 \pm 1.6) \times 10^{-3}$  for black soot 580using geometric surface area (Muñoz and Rossi, 2002).

581Only NAQM among the three models studied here incorporates the following heterogeneous 582non-stoichiometric reactions on the surface of soot (Li, J. et al., 2015).

583 
$$HNO_3 + soot \rightarrow NO + NO_2$$

(5)  $NO_2 + \text{soot} \rightarrow \rightarrow 0.5 \text{ HONO} + 0.5 \text{ HNO}_3$ . (6)

with  $\gamma_{\rm HNO3} = 3.0 \times 10^{-3}$  for Reaction (5) and  $\gamma_{\rm HONO} = 1.0 \times 10^{-4}$  for Reaction (6). The 585586 "renoxification" by Reaction (5) could have contributed to the increase of NO in Figs. 2(c) and 587 (d) and Figs. 3(c) and (d) resulting in a better agreement with the observation. The increase of 588NO could decrease  $O_3$  by the titration reaction (Reaction (1)), which may also give a better 589agreement for O<sub>3</sub> with the observation, particularly during nighttime. However, no quantitative 590sensitivity analysis has been made in the present study, and it is highly recommended that 591verification of the importance of such heterogeneous renoxification reaction in model 592simulation be made against accurate measurements of gaseous HNO<sub>3</sub> together with other NO<sub>y</sub> in 593 the polluted urban atmosphere.

#### 594**Effects of Vertical Transport**

595Other than the difference in chemical reaction mechanisms, the difference in transport 596module could give rise to differences in the output of O<sub>3</sub> concentrations. In order to analyze the 597 effects of transport, process analysis of horizontal and vertical transport of  $O_3$  has been 598conducted. Since it has been found that there is not much difference in horizontal transport and 599surface deposition, and the chemical mechanisms of CMAQ 5.0.2 and CMAQ 4.7.1 are the 600 same, the difference in model performance must be ascribed to the difference in vertical 601 transport processes.

602 Figures 6(a) and (b) show the comparison of vertical O<sub>3</sub> transport among the three models in 603 Beijing in April and July, respectively, and Figs. 6(c) and (d) show similar plots for Tokyo. The 604 davtime downward vertical flux of O<sub>3</sub> for both CMAQ models in Beijing are nearly the same (22-25 ppbv hr<sup>-1</sup>) in July, and much larger than the values (ca. 6 ppbv hr<sup>-1</sup>) in April. In contrast, 605 the values of NAQM are ca. 10 ppbv hr<sup>-1</sup> both in April and in July, which is larger than the 606 607 values of CMAQ in April, but smaller than those of CMAQ by a factor of two in July. The 608 diurnal variation of vertical O<sub>3</sub> flux in Tokyo is quite different from that in Beijing in July; 609 downward O<sub>3</sub> flux is positive only in the morning till noon and nearly zero or negative in the 610 afternoon. Such characteristics is common for all the three models. The maximum downward

Fig. 6

611 fluxes of  $O_3$  in the morning in Tokyo in CMAQ 5.0.2 (ca. 17 ppbv hr<sup>-1</sup>) and CMAQ 4.7.1 (ca. 612 13 ppbv hr<sup>-1</sup>) are much higher than those in NAQM (< 5 ppbv hr<sup>-1</sup>). Thus, it is concluded that at 613 least a part of much higher  $O_3$  concentrations estimated by CMAQ 5.0.2 and 4.7.1 as compared 614 to NAQM shown in Fig. 2(b) and Fig. 3(b) in Beijing and Tokyo in July can be ascribed to the 615 higher downward  $O_3$  flux estimated by the CMAQ models compared to NAQM.

- 616 A peculiar feature of vertical  $O_3$  flux in the CMAQ 5.0.2 shown in Fig. 6 is the strong 617 positive morning peak at around 7 and 6 am in Beijing in April and July, respectively, and also
- 618 at 6-7 am in Tokyo in April. Here, it should be noted that the vertical transport was computed in
- 619 the PPM scheme in CMAQ 5.0.2 instead of the direct extraction from WRF in CMAQ 4.7.1 as

described in the method section. The PPM method has been known to introduce more downward flux of  $O_3$  from higher layers to the surface layer. Another point to be noted is the delayed rise of vertical downward  $O_3$  flux by nearly 2 hours in NAQM both in April and July in Beijing and Tokyo. Although this feature is not scrutinized in this study, it should be noted here that the vertical transport treatment affects significantly the simulated results of  $O_3$  in regional chemical transport models.

#### 626 Comparison of the Transport Process in CMAQ v. 5.0.2 and v. 4.7.1

627 As seen in Fig. 2 and Fig. 3, CMAQ 5.0.2 gives relatively low O<sub>3</sub> and relatively high NO 628 mixing ratios, closer to the values in NAQM in April, but relatively high O<sub>3</sub> and low NO closer 629 to those in CMAQ 4.7.1 in July both in Beijing and in Tokyo. Since the chemical mechanisms 630 of CMAO 5.0.2 and CMAO 4.7.1 are the same, the difference in the model performance must 631 be ascribed to the difference in transport processes. Figures 7(a) and (b) show the comparison of 632  $O_3$  mixing ratios and the change in hourly  $O_3$  mixing ratios between CMAQ 5.0.2 and the 633 observations in Beijing in April, and similar plots for July are shown in Figs. 7(c) and (d). As 634 for the observational values, data provided by IAP are used for the plots. The large rise in the 635change of  $O_3$  concentrations at 7-8 am shown in Figs. 7(b) and (d) clearly correspond to the 636 early morning peak of downward transport flux of  $O_3$  at 6-7 am in Figs. 6(a) and (b). Such a 637 sharp rise at 7 am is not seen in CMAQ 4.7.1 although a small peak is discernable in April. This 638 implies that such a feature is due to the characteristics of the vertical transport module of 639 CMAQ 5.0.2. Similar plots for NO are shown in Figs. 8(a)-(d). In April, the NO mixing ratio by 640 CMAQ 5.0.2 rises in early morning, which corresponds well with the observation. The cause of 641 such an early morning rise of NO mixing ratio and change in hourly mixing ratio is assumed to 642 be the increase of traffic in the morning. In July however, although the observation of NO

Fig. 7

643 mixing ratio and hourly change shows a similar morning peak in April, the CMAQ 5.0.2 644 simulation doesn't give any such morning peak, which would correspond to a very low NO 645 mixing ratio simulated by CMAQ 5.0.2 together with CMAQ 4.7.1 as seen in Fig. 2(d). 646 Although the phenomena could be caused by rapid oxidation of NO to  $NO_2$  in summer, the 647 reason is unknown at this stage.

648 It should be noted that after the large rise at 7-8 am, the hourly change of  $O_3$  mixing ratio 649 simulated by CMAQ 5.0.2 agrees well with the observed  $O_3$  change in the late morning and 650 afternoon as shown in Figs. 7(b) and (d). This implies that the large morning surge gives a much 651 earlier rise of O<sub>3</sub> compared to the observation. It can be noted, however, that the morning surge 652at 7 am in July is ca. 15 ppby, which is not much higher than the ca.10 ppby in April. Thus, 653 although the morning surge is larger in July than in April, this would not be the main cause of 654the much higher predicted O<sub>3</sub> concentration in the morning in July as compared to April. A 655large difference in the simulated concentration of nighttime  $O_3$  can be seen between April and 656 July in CMAQ 5.0.2, and also between CMAQ 5.0.2 and CMAQ 4.7.1 in April. The nighttime 657 O<sub>3</sub> is as low as 10-20 ppbv in Beijing, and 20-30 ppbv in Tokyo in both CMAQ 5.0.2 and 658NAQM in April, agreeing with observation. However, the nighttime O<sub>3</sub> simulated by CMAQ 659 4.7.1 is as high as 30 and 45 ppbv in April in Beijing and Tokyo, respectively. In July, the 660 nighttime O<sub>3</sub> is 20-30 ppbv in Beijing and ca. 20 ppbv in Tokyo in NAQM, which is close to the 661 observation, while both CMAQ models give 40-50 ppbv both in Beijing and in Tokyo, which is 662 substantially higher than the observation. The high nighttime O<sub>3</sub> simulated by the CMAQ 663 models would contribute at least partly to the high daytime O<sub>3</sub> in July. Although the coarse 664 resolution of 45 km grid tends to give a higher nighttime  $O_3$  due to less effective NO titration, it 665 would not be enough to explain such a high nighttime O<sub>3</sub> in CMAQ 4.7.1 both for April and for 666 July, and CMAQ 5.0.2 for July, since the NAQM simulation with the same grid size reproduces 667 the nighttime  $O_3$  as low as 20 ppb, agreeing better with the observation. It would be important to 668 quantify the effect of the heterogeneous production of nighttime NO from HNO<sub>3</sub> to evaluate its 669 impact on nighttime O<sub>3</sub>.

### 670 Comparison of simulations with the observational data of O<sub>3</sub> in Beijing and Tokyo

Both CMAQ 5.0.2 and NAQM give reasonably good agreement of  $O_3$  mixing ratios with the observational data in April in Tokyo. It can be noted that both CMAQ 5.0.2 and NAQM give higher mixing ratios by 10-15 ppb after dawn. For CMAQ 5.0.2, as mentioned above, the overestimate could be caused by the peak of downward  $O_3$  flux in the early morning. NAQM 675 gives a similar overestimate of the O<sub>3</sub> mixing ratio by ca. 10 ppbv in the early morning, but this phenomenon could be caused by the peak of the net chemical ozone production (Fig. 5) rather than the vertical transport. Although the cause of the early morning peak of the net O<sub>3</sub> production has not been elucidated in this study, it may be related to the photolysis of HONO accumulated during nighttime, since the heterogeneous production of HONO (Eq. 6) is included in NAQM.

681 In July, NAQM is the sole model giving a good agreement with the observation in Tokyo. It 682 can be noted, however, that the calculated concentration is higher than the observation by ca. 10 683 ppbv in early morning similar to April. Such a higher rise of the  $O_3$  mixing ratio in the early 684 morning is discernible in July in Tokyo in CMAQ 5.0.2. The same phenomenon can also be 685 seen in July in Tokyo, and the cause is assumed to be the early morning peaks of downward 686 flux of O<sub>3</sub> and net O<sub>3</sub> production in CMAQ5.0.2 and NAQM, respectively. It should be noted 687 that the enhanced mixing ratios of  $O_3$  in early morning are persistent at least till noon, giving 688 higher values of simulated mixing ratios.

- Substantially higher simulated  $O_3$  mixing ratios in CMAQ 4.7.1 than the observation both in April and in July, and in CMAQ 5.0.2 in July in Tokyo (Figs. 3(a) and (b)), may at least partially be caused by the higher nighttime mixing ratios of  $O_3$ , which would contribute to the baseline mixing ratio for the whole day. It would be expected that if the nighttime  $O_3$  could be reduced to the observational level, a better agreement of  $O_3$  with observation would be expected for the whole day.
- 695 As for the observational data in Beijing, the daily maximum of  $O_3$  mixing ratio in July is ca. 696 90 ppbv in Xu et al. (2011) and ca. 60 ppbv by the IAP data, while the nighttime minimums are 697 both 10-20 ppbv consistently. Since the maximum  $O_3$  mixing ratio in summer is expected to be 698 higher in Beijing than in Tokyo (ca. 60 ppbv) due to higher NO (see Figs. 2(d) and 3(d)) and 699  $NO_2$  levels by a factor of  $\sim 2$  (not shown), the higher observational data in Beijing than in 700 Tokyo in Fig. 2(c) could be more representative for the average of the calculated grids in 701 Beijing. Although there still is a large uncertainty in the monthly averaged observational data of 702O<sub>3</sub> in Beijing in 2010, a tendency of overestimation by CMAQ 5.0.2 and 4.7.1, and 703 underestimation by NAQM in Beijing in July can be suggested.
- In April in Beijing, Chen et al. (2015) reported the daily maximum mixing ratio of  $O_3$  at ca. 65 ppbv in 2014-2015, which is substantially higher than the IAP data of ca. 40 ppbv in 2010. An increase in surface ozone has been reported in Beijing at the rural sites of Shangdianzi during the period of 2004-2015 with regard to the maximum daily average 8 hr mixing ratios

(MDA8) (Ma et al., 2016). Although the long-term increasing trend indicates an average rate of 1.13  $\pm$  0.01 ppb yr<sup>-1</sup>, no monthly data was reported, and the year-by-year variability is substantial. If we assume that the monthly averaged MDA8 values in April in 2010 are lower by 10 ppbv than those in 2015, the uncertainty of the daily maximum observational value in April in Beijing would be in the range of 40-55 ppbv. Thus, within the uncertainty range, a tendency of overestimation by CMAQ 4.7.1 and an underestimation by NAQM could be suggested.

As for the discussion of reproducibility of the model simulation, a comparison of three-year averaged values in more recent years after 2013, when routine monitoring data at considerably more sites within the targeted grids are available, would be highly desirable, particularly in Beijing.

## 718 Overestimation of O<sub>3</sub> at Oki, a remote oceanic site

719 At the remote site of Oki, an overestimation by ca. 20 ppb for daytime  $O_3$  has been seen in 720 July in Fig. 4(b) by all the three models with a spatial resolution of 45 km. Such an 721 overestimation of summertime O<sub>3</sub> at Oki by the CMAQ models has been reported by Lin et al. 722 (2009) with MM5-CMAQ v. 4.6 (27 km  $\times$  27 km), while a much better agreement with the 723 observation was previously reported by Yamaji et al. (2006) (80 km × 80 km) using 724RAMS-CMAQ v. 4.4, and Li, J. et al. (2007) using NAQM (81 km × 81 km). The seasonal 725variation of O<sub>3</sub> at remote sites around Japan has shown a springtime monthly maximum of ca. 726 60 ppb and a summertime monthly minimum of 35-40 ppb (Pochanart et al., 1999, 2002), which 727 is consistent with the observational data shown in Fig. 4. The summertime minimum at Oki and 728 other remote islands in this region are well established to be due to prevailing clean marine air 729(Pochanart et al., 2002; Yamaji et al., 2006).

730Since the overestimation does not depend on the spatial resolution of the model, as noted 731 above, and the daytime build-up of  $O_3$  due to local photochemical activity is <10 ppbv in the 732 observation and 5-15 ppbv in the simulation as shown in Fig. 4(b), the overestimation of  $O_3$ 733 concentration in July by all the three models cannot be ascribed to the direct influence of nearby 734terrestrial emissions of precursors in mainland Japan. The overestimation could be either due to 735a more frequent influence of terrestrial air masses by WRF compared to the real meteorology, or 736 higher O<sub>3</sub> concentration in the oceanic air around this area affected by the influence of 737 non-episodic terrestrial emissions including long-range transport. The reproduction of observed 738 concentrations by models at Oki would be important for the analysis of air quality in Japan, 739 since air masses passing through Oki provide a flowing-in background mixing ratio to mainland

740 Japan.

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# 742 Summary

In order to identify the causes of the substantial variability among the simulated modelling results for surface ozone in MICS-Asia III even though using the same emissions, meteorological field and boundary conditions, three regional models, namely CMAQ 5.0.2, 4.7.1 and NAQM were selected and a detailed comparison was made in the selected grids covering the metropolitan areas in Beijing and Tokyo and at the remote oceanic site of Oki. The analyses were made for the monthly averaged diurnal change of surface ozone in April and July in 2010.

The simulated  $O_3$  concentration was the highest in CMAQ 4.7.1, followed by those in CMAQ 5.0.2 and NAQM both in Beijing and in Tokyo in April, while both CMAQ models gave much higher  $O_3$  values than NAQM did in July. At Oki, the simulations for  $O_3$  by all the three models agree well with each other and with the observation in April. In July, however, all the models overestimated daytime  $O_3$  by ca. 20 ppb compared to the observation.

Three causes for the difference among model outputs have been identified and discussed.

(1) The chemistry mechanism sub-module, SAPRC99 used in the CMAQ was found to give
 higher net ozone production values than CBM-Z in NAQM, agreeing with previous studies.

- (2) Higher NO concentrations have been predicted by NAQM than by CMAQ, possibly due to the inclusion of a heterogeneous "renoxification" reaction of HNO<sub>3</sub> (on soot surface)  $\rightarrow$  NO + NO<sub>2</sub>, which gave a better agreement with observational concentration particularly for nighttime NO and O<sub>3</sub>.
- 762 (3) A vertical downward  $O_3$  flux was found to affect substantially the diurnal pattern and 763 mixing ratios of  $O_3$ .
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## 765 Acknowledgements

his research was supported by the Environment Research and Technology Development
Fund (S12-1) of the Ministry of the Environment, Japan, and by the Natural Science Foundation
of China (41620104008).

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# 770 References

771 Akimoto, H., Atmospheric Reaction Chemistry, Springer Japan, Tokyo, 2016.

- Binkowski, F. S. and S. J. Roselle, Models-3 Community Multiscale Air Quality(CMAQ)
  model aerosol component 1. Model description, J. Geophys. Res., 108, D6, 4183,
  doi:10.1029/2001JD001409, 2003.
- 775 Brunner, D., J. Staehelin, H. L. Rogers, M. O. Köhler, J. A. Pyle, D. A. Hauglustaine, L.
- Jourdain, T. K. Berntsen, M. Gauss, I. S. A. Isaksen, E. Meijer, P. van Velthoven, G. Pitari,
- E. Mancini, V. Grewe, and R. Sausen, An evaluation of the performance of chemistry
- transport models Part 2: Detailed comparison with two selected campaigns, Atmos. Chem.
  Phys., 5, 107–129, 2005.
- Burkholder, J. B., S. P. Sander, J. P. D. Abatt, J. R. Barker, R. E. Huie, C. E. Kolb, M. J. Kurylo,
  V. L. Orkin, D. M. Wilmouth, and P. H. Wine, Chemical Kinetics and Photochemical Data
  for Use in Atmospheric Studies, Evaluation Number 18, JPL Publication 15-10, Pasadena,
  California, http://jpldataeval.jpl.nasa.gov/, 2015.
- Carter, W. L., Implementation of the SAPRC-99 chemical mechanism into the Models-3
   framework, Report to the United States Environmental Protection Agency, January 29, 2000.
- Chen, W., H. Tang, and H. Zhao, Diurnal, weekly and monthly spatial variations of air
  pollutants and air quality of Beijing, Atmos. Environ., 119, 21-34, 2015.
- CMAS, Operational Guidance for the Community Multiscale Air Quality (CMAQ) Modeling
  System: Version 5.0, https://www.airqualitymodeling.org/index.php/CMAQ\_version\_5.0\_
  (February 2010 release) OGD, 2011.
- Disselkamp, R. S., M. A. Carpenter, J. P. Cowin, A chamber investigation of nitric acid-soot
  aerosol chemistry at 298 K, J. Atmos. Chem., 37, 113-123, 2000.
- Faraji, M., Y. Kimura, and D. Allen, Comparison of the Carbon Bond and SAPRC
  photochemical mechanisms under conditions relevant to southeast Texas, Atmos. Environ.,
  42, 5821-5836, 2007.
- Finlayson-Pitts, B. J. and J. N. Pitts, Jr., *Chemistry of the Upper and Lower Atmosphere*,
  Academic Press, 2000.
- Fiore, A. M. et al. (46 other co-authors), Multimodel estimates of intercontinental sourcereceptor relationships for ozone pollution, J. Geophys. Res., 114, D04301,
  doi:10.1029/2008JD010816, 2009.
- Foley, K. M., S. J. Roselle, K. W. Appel, P. V. Bhave, J. E. Pleim, T. L. Otte, R. Mathur, G.
  Sarwar, J. O. Young, R. C. Gilliam, C. G. Nolte, J. T. Kelly, A. B. Gilliland, and J. O. Bash,
  Incremental testing of the Community Multiscale Air Quality (CMAQ) modeling system
- 804 version 4.7, Geosci. Model Dev., 3, 205–226, 2010.

- Fountoukis, C. and A. Nenes, ISORROPIA II: a computationally efficient thermodynamic equilibrium model for  $K^+$ -  $Ca^{2+}$ -  $Mg^{2+}$ -  $NH_4^+$ -  $Na^+$ -  $SO_4^{-4}$ -  $NO_3^{-}$ -  $Cl^-$ -  $H_2O$  aerosols, Atmos. Chem. Phys., 7, 4639–4659, 2007.
- Fu, J. S., X. Dong, Y. Gao, D. C. Wong, and Y. F. Lam, Sensitivity and linearity analysis of
  ozone in East Asia: The effects of domestic emission and intercontinental transport, J. Air
  Waste Manag. Assoc., 62, 1102–1114, 2012.
- Gery, M. W., G. Z. Whitten, J. P. Killus, and M. C. Dodge, A photochemical kinetics
  mechanism for urban and regional scale computer modeling, J. Geophys. Res., 94,
  12,925-12,956, 1989.
- Gonçalves, M., D. Dabdub, W. L. Chang, O. Jorba, and J. M. Baldasano, Impact of HONO
  sources on the performance of mesoscale air quality models, Atmos. Environ., 54, 168-176,
  2012.
- Han, Z., T. Sakurai, H. Ueda, G. R. Carmichael, D. Streets, H. Hayami, Z. Wang, T. Holloway,
  M. Engardt, Y. Hozumi, S. U. Park, M. Kajino, K. Sartelet, C. Fung, C. Bennet, N.
  Thongboonchoo, Y. Tang, A. Chang, K. Matsuda, M. Amann, MICS-Asia II: Model
  intercomparison and evaluation of ozone and relevant species, Atmos. Environ. 42, 3491–
  3509, 2008.
- Hauglustaine, D. A., B. A. Ridley, S. Solomon, P. G. Hess, S. Madronich, HNO<sub>3</sub>/NO<sub>x</sub> ratio in
  the remote troposphere During MLOPEX 2: Evidence for nitric acid reduction on
  carbonaceous aerosols?, Geophys Res. Lett., 23, 2609–2612, 1996.
- Jaffe, D. A., R. E. Honrath, L. Zhang, H. Akimoto, A. Shimizu, H. Mukai, K. Murano, S.
  Hatakeyama, and J. Merrill, Measurements of NO, NO<sub>y</sub>, CO and O<sub>3</sub> and estimation of the
  ozone production rate at Oki Island, Japan, during PEM-West, J. Geophys. Res., 101,
  2037-2048, 1996.
- Jimenez, P., Jose M. Baldasano, and D. Dabdub, Comparison of photochemical mechanisms for
  air quality modeling, Atmos. Environ., 37, 4179–4194, 2003.
- 831 Kurokawa, K., T. Ohara, I. Uno, M. Hayasaki, and H. Tanimoto, Influence of meteorological
- variability on interannual variations of springtime boundary layer ozone over Japan during
  1981–2005, Atmos. Chem. Phys., 9, 6287–6304, 2009.
- Lary, D. J., A.M. Lee, R. Toumi, M. J. Newchurch, M. Pirre, and J. B. Renard, Carbon aerosols
  and atmospheric photochemistry, J. Geophys. Res., 102, 3671-3682, 1997.
- Li, J., Z. Wang, H. Akimoto, C. Gao, P. Pochanart, and X. Wang, Modeling study of ozone seasonal cycle in lower troposphere over east Asia, J. Geophys. Res., 112, D22S25,

- doi:10.1029/2006JD008209, 2007.
- Li, J., H. Dong, L. Zeng, Y. Zhang, M. Shao, Z. Wang, Y. Sun, and P. Fu, Exploring Possible
  Missing Sinks of Nitrate and Its Precursors in Current Air Quality Models—A Case
  Simulation in the Pearl River Delta, China, Using an Observation-Based Box Model, SOLA,
- 842 11, 124–128, 2015.
- Li, J., W. Yang, Z. Wang, H. Chen, B. Hu, J. Li, Y. Sun, P. Fu, Y. Zhang, Modeling study of surface ozone source-receptor relationships in East Asia, Atmos. Res., 167, 77–88, 2016.
- Li, J., Z. Wang, G. Zhuang, G. Luo, Y. Sun, and Q. Wang, Mixing of Asian mineral dust with
  anthropogenic pollutants over East Asia: a model case study of a super-duststorm in March
  2010, Atmos. Chem. Phys., 12, 7591–7607, 2012. over east asia. Sci. Total Environ., 622,
  1327-1342, 2018.
- Li, J., X. Chen, Z. Wang, H. Du, W. Yang, Y. Sun, B. Hu, J. Li, W. Wang, T. Wang, P. Fu, and
  H. Huang, Radiative and heterogeneous chemical effects of aerosols on ozone and inorganic
  aerosols over East Asia, Sci. Total Environ. 622–623, 1327–1342, 2018.
- Li, J., T. Nagashima, L. Kong, B. Ge, K. Yamaji, J. S. Fu, X. Wang, S. Itahashi, H.-J. Lee, C.-H.
  Kim, C.-Y. Lin, M. Zhang, Z. Tao, M. Kajino, H. Liao, M. Li, J.-H. Woo, J. Kurokawa, Y.
  Pan, Q. Wu, H. Akimoto, G. R. Carmichael, and Z. Wang Model evaluation and
  inter-comparison of surface-level ozone and relevant species in East Asia in the context of
  MICS-ASIA phase III, Atmos. Chem. Phys., to be published.
- Li, M., Zhang, Q., Kurokawa, J., Woo, J.-H., He, K. B., Lu, Z., Ohara, T., Song, Y., Streets, D.
- G., Carmichael, G. R., Cheng, Y. F., Hong, C. P., Huo, H., Jiang, X. J., Kang, S. C., Liu, F.,
  Su, H., and Zheng, B., MIX: a mosaic Asian anthropogenic emission inventory for the
  MICS-Asia and the HTAP projects, Atmos. Chem. Phys. 17, 935-963, 2017.
- Li, Y., J. An, M, Min, W. Zhang, F. Wang, and P. Xie, Impacts of HONO sources on the air
  quality in Beijing, Tianjin and Hebei Province of China, Atmos. Environ., 45, 4735–4744,
  2011.
- Li, Y., J. An, M. Kajino, J. Li, and Y. Qu, Impacts of additional HONO Sources on
  concentrations and deposition of NO<sub>y</sub> in the Beijing-Tianjin-Hebei Region of China,
  SOLA, 11, 36-42, 2015.
- Lin, M., T. Holloway, T. Oki, D. G. Streets, and A. Richter, Multi-scale model analysis of
  boundary layer ozone over East Asia, Atmos. Chem. Phys., 9, 3277–3301, 2009.
- 869 Luecken, D. L., S. Phillips, G. Sarwar and C. Jang, Effects of using the CB05 vs. SAPRC99 vs.
- 870 CB4 chemical mechanism on model predictions: Ozone and gas-phase photochemical

- precursor concentrations, Atmos. Environ., 42, 5805-5820, 2008.
- Ma, Z., J. Xu, W. Quan, Z. Zhang, W. Lin, and X. Xu, Significant increase of surface ozone at a
  rural site, north of eastern China, Atmos. Chem. Phys., 16, 3969–3977, 2016.
- 874 Monks, P. S., A review of the observations and origins of the spring ozone maximum, Atmos.
- Environ., 34, 3545-3561, 2000.
- Muñoz, M. S. S. and M. J. Rossi, Heterogeneous reactions of HNO<sub>3</sub> with flame soot generated
  under different combustion conditions. Reaction mechanism and kinetics, Phys. Chem.
- 878 Chem. Phys., 4, 5110–5118, 2002.
- Nenes, A., S. N. Pandis, and C. Pilinis, ISORROPIA: A new thermodynamic equilibrium model
  for multiphase multicomponent inorganic, Aerosols, Aquat. Geochem., 4, 123-152, 1998.
- 881 Pleim, J. E., A combined local and nonlocal closure model for the atmospheric boundary layer.
- Part I: Model description and testing, J. Appl. Meteor. Climatol., 46, 1383–1395, 2007.
- Pochanart, P., J. Hirokawa, Y. Kajii, and H. Akimoto, Influence of regional-scale anthropogenic
  activity in northeast Asia observed at Oki, Japan, J. Geophys. Res., 104, 3621-3631, 1999.
- Pochanart, P., H. Akimoto, Y. Kinjo, H. Tanimoto, Surface ozone at four remote island sites
  and the preliminary assessment of the exceedances of its critical level in Japan, Atmos.
  Environ., 36, 4235–4250, 2002.
- Pochanart, P., H. Akimoto, Y. Kajii, V. M. Potemkin, and T. V. Khodzher, Regional
  background ozone and carbon monoxide variations in remote Siberia/East Asia, J. Geophys.
  Res., 108, 4028, doi:10.1029/2001JD001412, 2003.
- Sarwar, G., D. Luecken, G. Yarwood, G. Whitten, and W. P. L. Carter, Impact of an updated
  Carbon Bond mechanism on predictions from the CMAQ modeling system: preliminary
  assessment, J. Appl. Meteor. Climat. 47, 3-14, 2008.
- Singh, H. B., D. Herlth, R. Kolyer, L. Salas, J. D. Bradshaw, S. T. Sandholm, D. D. Davis J.
  Crawford, Y. Kondo, M. Koike, R. Talbot, G. L. Gregory, G. W. Sachse, E. Browell, D. R.
- Blake, F. S. Rowland, R. Newell, J. Merrill, B. Heikes, S. C. Liu, P. J. Crutzen, M.
- 897 Kanakidou, Reactive nitrogen and ozone over the western Pacific: Distribution, partitioning,
- and sources, J. Geophys Res., 101, 1793–1808, 1996.
- Talbot, R., J. Dibb, E. Scheuer, G. Seid, R. Russo, S. Sandholm, D. Tan, H. Singh, D. Blake, N.
  Blake, E. Atlas, G. Sachse, C. Jordan, M. Avery, Reactive nitrogen in Asian continental
  outflow over the western Pacific: Results from the NASA Transport and Chemical Evolution
  over the Pacific (TRACE-P) airborne mission, J. Geophys Res., 108, D20,
  doi: 10.1029/2002JD003129, 2003.

- 904 Walcek, C. J. and N. M. Aleksic, A simple but accurate mass conservative, peak-preserving, 905 mixing ratio bounded advection algorithm with FORTRAN code, Atmos. Environ., 32, 906 3863-3880, 1998.
- 907 Wang, T., X. L. Wei, A. J. Ding, C. N. Poon, K. S. Lam, Y. S. Li, L. Y. Chan, and M. Anson, 908 Increasing surface ozone concentrations in the background atmosphere of Southern China, 909
- 1994-2007. Atmos. Chem. Phys. 9, 6217-6227, 2009.
- 910 Wang, Z. F. et al. overview paper in this special issue, Atmos. Chem. Phys. to be published.
- 911 Wong, K. W., C. Tsai, B. Lefer, N. Grossberg, and J. Stutz, Modeling of daytime HONO 912 vertical gradients during SHARP 2009, Atmos. Chem. Phys., 13, 3587-3601, 2013.
- 913 Xu, J., J. Z. Ma, X. L. Zhang, X. B. Xu, X. F. Xu, W. L. Lin, Y. Wang, W. Meng, and Z. Q. Ma, 914 Measurements of ozone and its precursors in Beijing during summertime: impact of urban
- 915 plumes on ozone pollution in downwind rural areas, Atmos. Chem. Phys., 11, 12241–12252, 916 2011.
- 917 Yamaji, K., T. Ohara, I. Uno, H. Tanimoto, J. Kurokawa, and H. Akimoto, Analysis of the 918 seasonal variation of ozone in the boundary layer in East Asia using the Community 919 Multi-scale Air Quality model: What controls surface ozone levels over Japan?, Atmos. 920 Environ., 40, 1856-1868, 2006.
- 921 Yamaji, K., T. Ohara, I. Uno, J. Kurokawa, P. Pochanart, and H. Akimoto, Future prediction of 922 surface ozone over east Asia using Models-3 Community Multiscale Air Quality Modeling 923 System and Regional Emission Inventory in Asia, J. Geophys. Res., 113, D08306, 924 doi:10.1029/2007JD008663, 2008.
- 925 Yamartino, R. J., Nonnegative, conserved scalar transport using grid-cell-centered, spectrally 926 constrained Blackman cubics for applications on a variable-thickness mesh, Mon. Weather 927 Rev. 121, 753-763, 1993.
- 928 Zaveri, R. A. and L. K. Peters, A new lumped structure photochemical mechanism for 929 large-scale applications, J. Geophys. Res., 104, 30, 387-30, 415, 1999.
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931	Figure Captions
932	
933	Fig. 1 Grids for comparison of the model simulation and observation; Beijing and Tokyo
934	metropolitan areas and Oki EANET site
935	
936	Fig. 2 Monthly averaged diurnal variation in Beijing, (a) O <sub>3</sub> in April, (b) O <sub>3</sub> in July, (c) NO in
937	April, and (d) NO in July.
938	
939	Fig. 3 Monthly averaged diurnal variation in Tokyo, (a) O <sub>3</sub> in April, (b) O <sub>3</sub> in July, (c) NO in
940	April, and (d) NO in July.
941	
942	Fig. 4 Monthly averaged diurnal variation of $O_3$ at Oki (a) in April, (b) in July.
943	
944	Fig. 5 Comparison of net chemical O <sub>3</sub> production in (a) Beijing in April, (b) Beijing in July,
945	(c) Tokyo in April, and (d) Tokyo in July.
946	
947	Fig. 6 Comparison of vertical transport of O <sub>3</sub> in (a) Beijing in April, (b) Beijing in July, (c)
948	Tokyo in April, and (d) Tokyo in July.
949	
950	Fig. 7 Monthly averaged diurnal variation of (a) O <sub>3</sub> concentration in April, (b) hourly O <sub>3</sub>
951	concentration change in April, (c) O <sub>3</sub> concentration in July and (d) hourly O <sub>3</sub> concentration
952	change in July in Beijing.
953	
954	Fig. 8 Monthly averaged diurnal variation of (a) NO concentration in April, (b) hourly NO
955	concentration change in April, (c) NO concentration in July and (d) hourly NO concentration
956	change in July in Beijing.
957	







Fig. 4

Fig. 5





Fig. 6







