

Polyols and glucose particulate species as tracers of primary biogenic organic aerosols at 28 French sites

Abdoulaye Samake¹, Jean-Luc Jaffrezo¹, Olivier Favez², Samuël Weber¹, Véronique Jacob¹, Alexandre Albinet², Véronique Riffault³, Esperanza Perdrix³, Antoine Waked^{1*}, Benjamin Golly¹, Dalia Salameh^{1†}, Florie Chevrier¹, Diogo Miguel Oliveira^{2,3}, Jean-Luc Besombes⁴, Jean M.F Martins¹, Sébastien Conil⁵, Géraldine Guillaud⁶, Boualem Mesbah⁷, Benoit Rocq⁸, Pierre-Yves Robic⁹, Agnès Hulin¹⁰, Sébastien Le Meur¹¹, Maxence Descheemaeker¹², Eve Chretien¹³, and Gaëlle Uzu¹.

¹Univ. Grenoble Alpes, CNRS, IRD, INP-G, IGE (UMR 5001), F-38000 Grenoble, France.

²INERIS, Parc Technologique Alata, BP 2, 60550 Verneuil-en-Halatte, France

³IMT Lille Douai, Univ. Lille, SAGE – Département Sciences de l'Atmosphère et Génie de l'Environnement, F-59000 Lille, France

⁴Univ. Savoie Mont-Blanc, LCME, F-73000 Chambéry, France

⁵ANDRA DRD/GES Observatoire Pérenne de l'Environnement, F-55290 Bure, France

⁶Atmo AuRA, F-38400 Grenoble, France

⁷Air PACA, F-03040, France

⁸Atmo Hauts de France, F-59000, France

⁹Atmo Occitanie, F-31330 Toulouse, France

¹⁰Atmo Nouvelle Aquitaine, F-33000, France

¹¹Atmo Normandie, F-76000, France

¹²Lig'Air, F-45590 Saint-Cyr-en-Val, France

¹³Atmo Grand Est, F-16034 Strasbourg, France

*Now at IMT Lille Douai, Univ. Lille, SAGE – Département Sciences de l'Atmosphère et Génie de l'Environnement, F-59000 Lille, France

†Now at Airport pollution control authority (ACNUSA), 75007 Paris, France

Corresponding author(s): A Samaké (abdoulaye.samake2@univ-grenoble-alpes.fr) and JL Jaffrezo (Jean-luc.Jaffrezo@univ-grenoble-alpes.fr)

33 **Abstract.** A growing number of studies are using specific primary sugar species, such as sugar alcohols or primary
34 saccharides, as marker compounds to characterize and apportion primary biogenic organic aerosols (PBOA) in the
35 atmosphere. To better understand their annual cycles, as well as their spatio-temporal abundance in terms of
36 concentrations and sources, we conducted a large study focusing on three major atmospheric primary sugar
37 compounds (i.e. arabitol, mannitol and glucose) measured in various environmental conditions on about 5,300
38 filter samples collected at 28 sites in France. Our results show significant atmospheric concentrations of polyols
39 (defined here as the sum of arabitol and mannitol) and glucose at each sampling location, highlighting their
40 ubiquity. Results also confirm that polyols and glucose are mainly associated with the coarse rather than the fine
41 aerosol mode. At nearly all sites, atmospheric concentrations of polyols and glucose display a well-marked
42 seasonal pattern, with maximum concentrations from late spring to early autumn, followed by an abrupt decrease
43 in late autumn, and a minimum concentration during wintertime. Such seasonal patterns support biogenic
44 emissions associated with higher biological metabolic activities (e.g. sporulation, growth, etc.) during warmer
45 periods. Results from a previous comprehensive study using Positive Matrix Factorization (PMF) based on an
46 extended aerosol chemical composition dataset of up to 130 species for 16 of the same sample series has also been
47 used in the present work. Results show that PBOA are significant sources of total OM in PM₁₀ (13±4 % on a yearly
48 average, and up to 40 % in some environments in summer) at most of the investigated sites. The mean PBOA
49 chemical profile is clearly dominated by contribution from OM (78±9 % of the mass of the PBOA PMF factor on
50 average), and only a minor contribution from dust class (3±4 %), suggesting that ambient polyols are most likely
51 associated with biological particle emissions (e.g. active spore discharge) rather than soil dust resuspension.

52 **1. Introduction**

53 Airborne particles (or particulate matter, PM) are of major concern due to their multiple effects on climate and
54 adverse human health impacts (Boucher et al., 2013; Cho et al., 2005; Ntziachristos et al., 2007). The diversity of
55 PM impacts is closely linked to their complex and highly variable nature: size distribution, concentration and
56 chemical composition, or specific surface properties. PM consists of inorganic and elemental substances, and a
57 large fraction made of carbonaceous matter (organic carbon (OC) and elemental carbon (EC)) (Franke et al., 2017;
58 Putaud et al., 2004a; Yttri et al., 2007a). Substantial amounts of atmospheric organic matter (OM) remain
59 unidentified and uncharacterized at the molecular level. In most studies, a maximum of only 20 % of particulate
60 OM mass can generally be speciated and quantified (Alfarra et al., 2007; Fortenberry et al., 2018; Liang et al.,
61 2017; Nozière et al., 2015). This detailed composition of OM and its spatial and seasonal distribution can give
62 important insights on the adverse effects of PM. So far, the majority of air pollution studies have focused on
63 organic atmospheric particles associated with anthropogenic and secondary sources, whereas a significant fraction
64 of OM is also associated with primary emissions from biogenic sources (Bauer et al., 2008a; Jaenicke, 2005; Liang
65 et al., 2016). Therefore, the characterization of primary OM biogenic sources at the molecular level is still limited
66 (Fuzzi et al., 2006; Liang et al., 2017; Zhu et al., 2015), and should be further investigated for a better understanding
67 of aerosol sources and formation processes.

68 Primary biogenic organic aerosols (PBOA) are emitted directly from the biosphere to the atmosphere where they
69 are ubiquitous and participate in many atmospheric processes (Elbert et al., 2007; Fröhlich-Nowoisky et al., 2016).
70 Additionally, their inhalation has long been associated with human respiratory impairments (e.g. asthma,

71 aspergillosis, etc.) (Després et al., 2012; Morris et al., 2011). PBOA comprises living and dead microorganisms
72 such as bacteria, fungi, viruses, bacterial and fungal spores, and microbial fragments, endotoxins, mycotoxins, or
73 pollens (Elbert et al., 2007; Jaenicke, 2005; Morris et al., 2011). In most semi-urban European site, PBOA can
74 account for up to 25 % of the atmospheric aerosol mass, in the size range of 0.2 to 50 μm (Fröhlich-Nowoisky et
75 al., 2016; Jaenicke, 2005; Huffman et al., 2012; Manninen et al., 2014; Morris et al., 2011). However, their sources
76 and contribution to total airborne particles are still poorly documented, partly because of the difficulty to recognize
77 them by conventional microbiological methods (cells culture, microscopic examination, etc.) (Di Filippo et al.,
78 2013; Heald and Spracklen, 2009; Jia et al., 2010a).

79 Several specific chemical components, such as primary sugar compounds (i.e. primary saccharides and sugar
80 alcohols) emitted persistently from biogenic sources, have long been suggested as powerful and unique biomarkers
81 in tracing sources, and abundances of PBOA as well. For instance, ambient concentrations of glucose have been
82 used as markers for plant materials (such as pollen, leaves, and their fragments) or soil emissions from several
83 areas in the world (Fu et al., 2012; Jia et al., 2010a, 2010b; Pietrogrande et al., 2014; Rathnayake et al., 2017).
84 Many studies indicated that glucose is the most abundant monosaccharide in vascular plants, where it serves as
85 the common energy materials, and an important source of carbon for soil active microorganisms (such as bacteria
86 or fungi) (Jia et al., 2010a; Medeiros et al., 2006; Pietrogrande et al., 2014; Zhu et al., 2015). Additionally, sugar
87 alcohols (also called polyols) including arabitol and mannitol have been proposed as markers for airborne fungi,
88 and are widely used to quantify their contributions to PBOA mass (Bauer et al., 2008a, 2008b; Golly et al., 2018;
89 Srivastava et al., 2018; Zhang et al., 2010). These sugar alcohols have also been found to correlate very well with
90 fluorescent PBOA in UV-APS and WIBS-3 online studies, particularly in rainy periods (Gosselin et al., 2016),
91 favoring microbial sporulation (such as fungi belonging to Ascomycota and Basidiomycota phyla) (China et al.,
92 2016; Elbert et al., 2007; Jones and Harrison, 2004). Polyols are produced in large amounts by many fungi and
93 bacteria, and several functions have been described for these compounds, such as common energy storage
94 materials, intracellular protectants against stressful conditions (e.g. heat or drought), storage or transport of
95 carbohydrates, quencher of oxygenated reactive species, or regulators of intracellular pH by acting as a sink or
96 source of protons (Jennings et al., 1998; Medeiros et al., 2006; Véléz et al., 2007). Hence, polyols, especially
97 arabitol and mannitol, may represent a significant fraction of the dry weight of fungi, and mannitol can contribute
98 between 20 to 50 % of the mycelium dry weight (Ruijter et al., 2003; Véléz et al., 2007). However, polyols are also
99 often identified in the lower plants (leaves, pollens) and green algal lichens (Medeiros et al., 2006; Véléz et al.,
100 2007; Yang et al., 2012). The primary sugar compounds (defined as polyols and primary saccharide species) are
101 actually thought to be relatively stable in the atmosphere (Wang et al., 2018). However, studies investigating their
102 atmospheric lifetime are quite limited. One previous laboratory study has been conducted by the US-EPA to
103 evaluate the stability of these chemicals on filter material exposed to gaseous oxidants as well as in aqueous
104 solutions (simulating clouds and fog droplet chemistry). Findings of this former study have shown that primary
105 sugar compounds remain quite stable up to 7 days (the extent of the testing period), pointing out their suitability
106 for use as tracers of atmospheric transport (Fraser, 2010). With all of this information, the use of primary sugar
107 compounds (such as mannitol, arabitol, glucose, etc.) as suitable tracers of PBOA is generally acknowledged (Jia
108 and Fraser, 2011; Zhu et al., 2015, 2016).

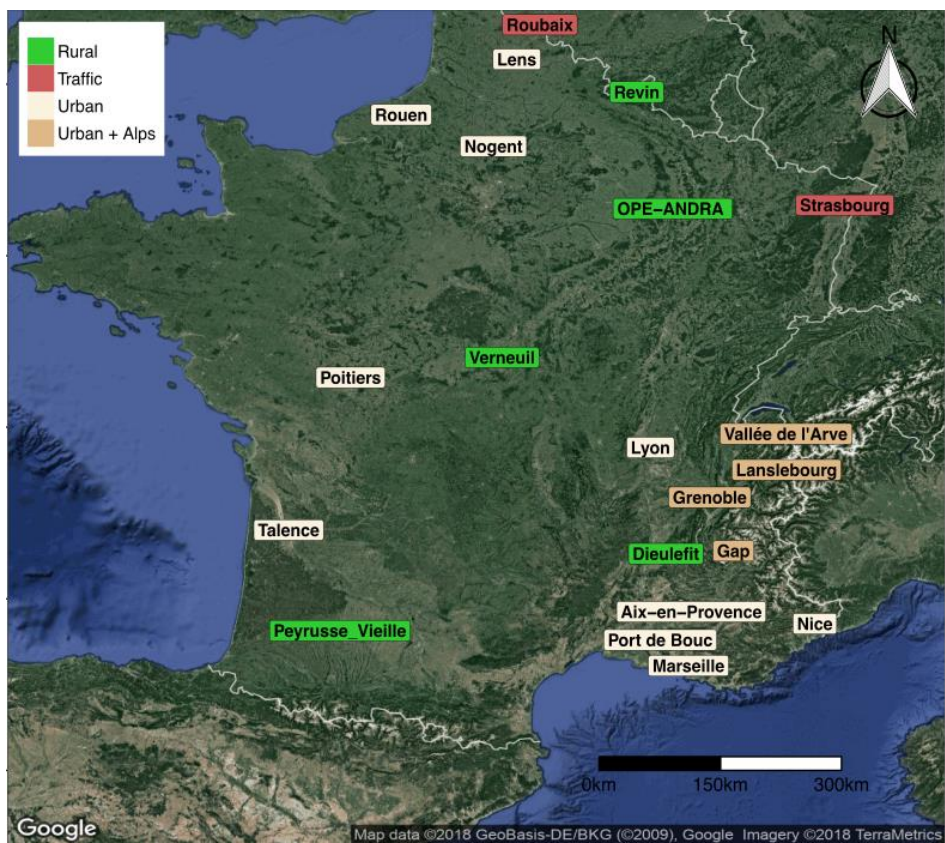
109 Although atmospheric concentrations of polyols, including arabitol and mannitol, as well as that of some primary
110 monosaccharides (e.g. glucose), have been previously quantified as part of several studies in various environments

111 including urban/suburban, rural, rainforest, mountain, and marine areas (Fu et al., 2012; Graham et al., 2003; Jia
112 et al., 2010a, 2010b; Liang et al., 2016; Pietrogrande et al., 2014; Simoneit, 2004; Verma et al., 2018; Yttri et al.,
113 2007b; Zhu et al., 2015), large datasets investigating their annual cycles and spatial distributions are still limited.
114 Such information could give important insights on environmental factors influencing their atmospheric levels such
115 as climate and biotope, and therefore help to elucidate patterns regarding their major sources and atmospheric
116 emission pathways (i.e. atmospheric input process). Even if numerous sources and emission mechanisms have
117 been widely proposed, including among others, metabolic active microbial wet emissions, entrainment of farmland
118 or natural soils and associated microbiota (Elbert et al., 2007; Fu et al., 2013; Gosselin et al., 2016; Jia et al., 2010a,
119 2010b; Medeiros et al., 2006; Pietrogrande et al., 2014; Simoneit et al., 2004; Verma et al., 2018; Yttri et al.,
120 2007b), the dominant atmospheric input processes have not been yet sufficiently elucidated.
121 In this context, the present study was designed to provide a large overview of the spatial and seasonal variations
122 of polyols and glucose mass concentrations, as well as their contribution to the aerosol organic mass fraction, in
123 France. To do so, data was collected at many sites in different environments (rural, traffic, urban), in order to
124 represent various sampling conditions in terms of site typologies and meteorological conditions. Thanks to the
125 availability of results from an extended Positive Matrix Factorization (PMF) analysis performed for the
126 corresponding datasets, the overall contributions of the main polyols and glucose emission sources could also be
127 investigated in light of their spatial patterns. To the best of our knowledge, this is the first study providing such an
128 extended phenomenology of these compounds over multiple sites with different typologies. This paper is the first
129 part of on-going work on the concentrations of ambient particulate polyols and glucose in France. A second part
130 will examine the short-term (daily) variations of these chemicals at multiple spatial scale (local to nation-wide)
131 and their potential processes for entering the atmosphere. A third part will investigate the links between the
132 atmospheric concentrations of polyols and glucose and airborne microbial fingerprint up to the genus level.

133 **2. Material and methods**

134 **2.1 Aerosol sampling**

135 Ambient aerosol samples considered in the present work come from different research and monitoring programs,
136 conducted over the last 5 years in France (Figure 1). Each program includes at least one-year of field sampling,
137 providing a total number of 5,343 daily filter samples available for the sake of the present study. These sites offer
138 diverse conditions in terms of typologies (i.e. rural, traffic, urban sites, Alpine valley environments, etc.), local
139 climate and vegetation types and were selected in order to cover the complex and variable national environmental
140 conditions. These sites are assumed to represent typical environmental conditions in France, and our
141 observations/and general tendency could therefore be extrapolated to neighboring western European countries
142 presenting quite homogeneous environmental conditions.



143
 144 **Figure 1: Map of sampling site locations in France. Green: rural background, red: traffic, wheat: urban background**
 145 **and dark wheat: urban background in Alpine valley sites. The area of Grenoble (Grenoble_LF, Grenoble_CB and**
 146 **Grenoble_VIF) and Vallée de L'Arve (Marnaz, Passy and Chamonix) include 3 sites each. The area of Marseille**
 147 **includes four sites: Marseille, Mallet, Meyreuil and Gardanne.**

148 The site of OPE-ANDRA is a specific monitoring observatory in a rural environment, without any village or
 149 industry within several kilometers (description available from: <http://www.andra.fr>). All other sites correspond to
 150 stations of regional air quality monitoring networks (AASQA). The availability of filter samples was variable from
 151 one site to the other one, depending on the sampling frequency (typically every third or sixth day). Filter collection
 152 was conducted within the PM₁₀ or the PM_{2.5} aerosol size fraction, depending on the investigated site (Table SI-1).
 153 Moreover, co-located and simultaneous PM₁₀ and PM_{2.5} samplings were conducted at OPE-ANDRA and Revin,
 154 allowing to investigate the distribution of primary sugar compounds between the fine and the coarse aerosol size
 155 fractions at these two sites.

156 Ambient aerosols were collected onto quartz fiber filters (Tissu-quartz PALL QAT-UP 2500 150 mm diameter),
 157 preheated at 500 °C for 4 h minimum before use. After collection, all filter samples were wrapped in aluminum
 158 foils, sealed in zipper plastic bags and stored at <4°C until further chemical analysis. Field blank filters were also
 159 collected, at least once a month, using the same handling procedure than for PM samples. More detailed
 160 information on the sampling periods, air sampler, number of filters and nature of PM samples are provided in
 161 Table S1 and Fig. S1.

162 **2.2 Chemical analyses**

163 PM samples were analyzed for various chemical species using sub-sampled fractions of the collection filters. In
 164 the frame of the present study, the carbonaceous matter (organic carbon (OC) and elementary carbon (EC)) was
 165 analyzed using a thermo-optical method on a Sunset Lab analyzer (Birch and Cary, 1996) as described by Aymoz

166 et al. (2007), using the EUSAAR2 temperature program (Cavalli et al., 2010), except for the five sites of Northern
167 France where the NIOSH870 protocol was employed (Birch and Cary, 1996). Particulate organic matter (OM)
168 contents were then estimated by multiplying the organic carbon mass concentrations by a fixed factor, with $OM =$
169 $1.8 \times OC$. This OM-to-OC ratio value of 1.8 was chosen based on previous studies performed in France (Favez et
170 al., 2010; Petit et al., 2015 and reference therein) and around the world. (e.g., Aiken et al., 2008; Li et al., 2018;
171 Ruthenburg et al., 2014; Vlachou et al., 2018), with a typical range of 1.2-2.4 values.

172 For the analysis of anhydrosugars, sugar alcohols, and primary saccharides, filter punches (typically of about 10
173 cm²) were first extracted into ultrapure water, then filtered using a 0.22 μm Acrodisc filter. Depending on the site,
174 analyses were conducted either by IGE (Institut des Géosciences de l'Environnement) or by LSCE (Laboratoire
175 des Sciences du Climat et de l'Environnement) (Table SI-1). At IGE, extraction was performed during 20 min in
176 a vortex shaker and analyses were achieved using an HPLC with Pulsed Amperometric Detection. A first set of
177 equipment was used until March 2016, consisting of a Dionex DX500 equipped with three columns Metrosep
178 (Carb 1-Guard + A Supp 15-150 + Carb 1-150), the analytical run being isocratic with 70 mM sodium hydroxide
179 eluent, followed by a gradient cleaning step with a 120 mM NaOH eluent. This analytical technique enables to
180 detect anhydrous saccharides (levoglucosan, mannosan, galactosan), polyols (arabitol, sorbitol, mannitol), and
181 glucose (Waked et al., 2014). A second set of equipment was used after this date, with a Thermo-Fisher ICS 5000+
182 HPLC equipped with 4 mm diameter Metrosep Carb 2 × 150 mm column and 50 mm pre-column. The analytical
183 run is isocratic with 15 % of an eluent of sodium hydroxide (200 mM) and sodium acetate (4 mM) and 85 % water,
184 at 1 mL min⁻¹. This method allows for additional separation and quantification of erythritol, xylitol, and threulose.
185 At LSCE, extraction was performed during 45 min by sonication and analyses were achieved using an IC (DX600,
186 Dionex) with Pulsed Amperometric Detection (ICS3000, Dionex). A CarboPAC MA1 columns has be used
187 (4 × 250 mm, Dionex), the analytical run being isocratic with 480 mM sodium hydroxide eluent. This analytical
188 technique enable to detect anhydrous saccharides (levoglucosan, mannosan, galactosan), polyols (arabitol,
189 mannitol), and glucose.

190 Field blank filters were handled as real samples for quality assurance. The present data were corrected with field
191 blanks. The reproducibility of the analysis of primary sugar species (polyols, glucose), estimated from the analysis
192 of extracts of 10 punches from the same filters are generally in the range of 10-15 %.

193 Additional chemical analyses were conducted for most of the sites, allowing to quantify up to 130 different
194 chemical species (Calas et al., 2018). 30- 35 chemical species were then selected in order to achieve PMF analyses
195 as discussed hereafter.

196 **2.3 Statistical analysis**

197 Species concentration measurements were first analyzed for normality using Shapiro-Wilk's method with the
198 statistical program *R studio interface* (version 3.4.1). Since data were generally not distributed normally, we used
199 non-parametric statistical methods. The strength of the relationship between species concentrations was
200 investigated using the non-parametric Spearman rank correlation method. Multiple mean comparison analyses
201 were performed with the Kruskal-Wallis test method. Statistical significance was set at $p < 0.05$.

202 Positive Matrix Factorization for the source apportionment of the PM was previously performed at several sites of
203 this study, as part of the [SOURCES](#) (Favez et al., 2016; Salameh et al., in prep.) and [DECOMBIO](#) (Chevrier, 2017)
204 projects. We used the US EPA PMF 5.0 software (US EPA, 2015), following the general recommendation

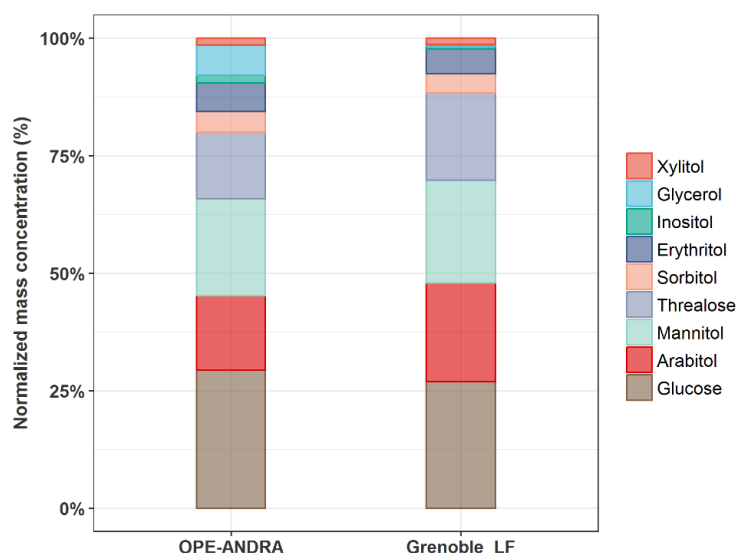
205 guidelines of the European Joint Research Centre (JRC) (Belis et al., 2014). Briefly, the SOURCES program aimed
206 at performing source apportionment at 15 sites using a harmonized methodology, i.e. using the same chemical
207 species, uncertainties, constraints, and criteria for factor identification. The PMF conducted within SOURCES
208 project uses about 30 different species (Table S6), including carbonaceous fraction (OC, EC), ions (Cl^- , NO_3^- , SO_4^{2-}
209 , NH_4^+ , K^+ , Mg^{2+} , Ca^{2+}), organic markers (polyols i.e. sum of arabitol, mannitol and sorbitol; levoglucosan;
210 mannosan) and metals (Al, As, Ba, Cd, Co, Cs, Cu, Fe, La, Mn, Mo, Ni, Pb, Rb, Sb, Se, Sn, Sr, Ti, V, Zn). The
211 PMF conducted within the DECOMBIO project, for the sites of Marnaz, Chamonix, and Passy, used aethalometer
212 (AE 33) measurements instead of EC (Chevrier, 2016). This complementary measure gives the total black carbon
213 (BC), thus enabling the deconvolution of BC concentrations into its two main constituents: wood-burning BC
214 (BC_{wb}) and fossil-fuel BC (BC_{ff}) (Sandradewi et al., 2008). For graphical simplicity, BC_{wb} and BC_{ff} were summed
215 up and labeled as EC in the following Figures. PMF modelling was performed separately for each site. Statistical
216 significance was validated with bootstrap higher than 80 % for each factor. Detailed methodology and results about
217 these studies are given in their respective papers (Chevrier, 2017; Favez et al., 2016; Salameh et al., in prep.). It
218 should be noted that glucose was not included in the final solution for any of these PMF, since it generally produced
219 statistical instability of the solutions (this point is further discussed in Sect. 3.2).
220 The PMF analysis took advantage of the ME-2 algorithm to add constraints to different chemical profiles (see
221 Tables SI-3 and SI-4 for details). Mainly soft constraints were applied in order to add some prior knowledge about
222 the emission sources and “clean” the different profiles without forcing the model toward an explicit solution. In
223 particular, the polyol concentrations were “pulled up maximally”, while levoglucosan and mannosan were set to
224 zero, and EC was “pulled down maximally” in this factor in the PBOA factor. This was achieved to avoid mixing
225 with the biomass burning factor as well as possible influences of unrealistic high contributions of EC to PBOA.
226 Other constraints were added parsimoniously to other factors, targeting specific proxies of sources (Table SI-4).
227 As for the general results of this large PMF study, we identified some well-known sources for almost all the sites
228 (biomass-burning, road traffic, secondary inorganics, dust and sea salt). Two other less-common factors were
229 identified for all sites: secondary biogenic aerosols (probably from marine origin), traced mainly by the presence
230 of MSA, and PBOA, traced by the presence of more than 90% of the polyols total mass in the factor. Table SI-5
231 and Fig. SI-4 present more detailed description of the chemical tracers in each factor, together with their yearly
232 average contribution for each site, respectively. Hereafter, only the PBOA chemical profile will be extensively
233 investigated. The uncertainties of this PBOA factor are discussed below and its stability is presented in Fig. SI-5.
234 Bootstrap analysis based on 100 resampling runs evidenced the very high stability of this PBOA factor since the
235 PBOA initial constrained factor was mapped to PBOA bootstrap factor (BF) more than 99% of the time.

236 **3. Results and discussion**

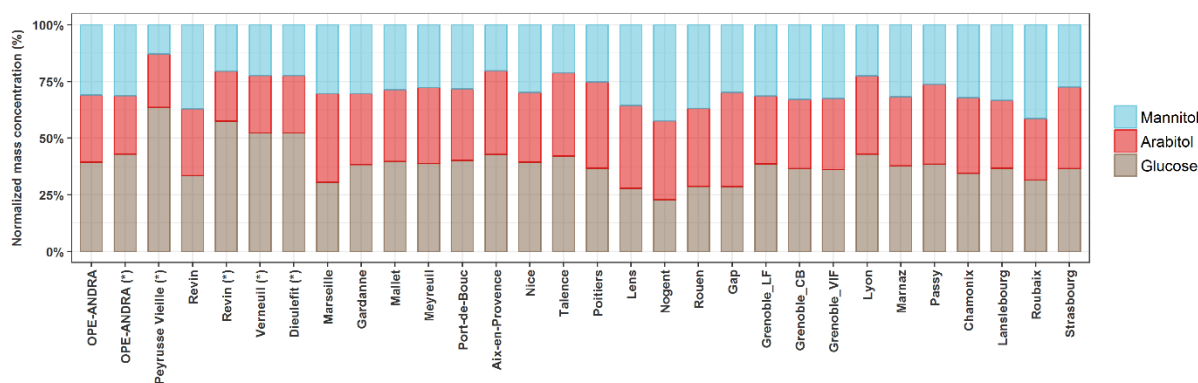
237 **3.1 Relative distribution between sugar alcohols and glucose**

238 Figure 2 presents an overview of the relative mass concentration distributions of individual chemical species
239 quantified at two sites with very different characteristics, an urban site in Grenoble and the rural site of OPE-
240 ANDRA. Data are presented for the warmer season (e.g. during summer and fall), when concentrations were at
241 their maximum (see Sect. 3.4). Glucose is the most abundant species measured (average $37.6 \pm 26.4 \text{ ng m}^{-3}$),
242 accounting on average for 25 % of primary sugar compounds (SC) total mass at both sites. Mannitol

243 (37.3±24.6 ng m⁻³) and arabitol (32.0±22.2 ng m⁻³), are the second and third most abundant species, accounting
 244 respectively for 25 and 23 % of SC mass. Threalose is relatively abundant in samples from these two sites
 245 (20.1±16.2 ng m⁻³), accounting for 14 % of SC mass, but in general its concentration is frequently below the limit
 246 of quantification for samples from other sites in France. The other identified polyols (i.e. erythritol, inositol,
 247 glycerol, sorbitol, and xylitol) present lower concentration levels (4.9±2.1 ng m⁻³), corresponding altogether to
 248 13 % of SC total mass.
 249 Such ambient mass concentration distribution patterns are similar (but with variable intensities) to those previously
 250 reported for aerosol samples collected at various locations around the world. For example, Verma et al. (2018)
 251 found that glucose, and arabitol together with mannitol, contributed to 16.7 and 48.1 %, respectively, of total
 252 primary sugar compounds in aerosols from Chichijima Island. Similarly, Yttri et al. (2007b) showed that glucose
 253 and the pair arabitol-mannitol were the main contributors of total primary monosaccharides and sugar alcohols in
 254 aerosols collected from four various background sites in Norway. In addition, Carvalho et al. (2003) reported that
 255 arabitol, mannitol and glucose as the most dominant primary sugar compounds in aerosols from rural background
 256 and boreal forest sites in Germany and Finland, respectively.



257
 258 **Figure 2: Overview of relative mass distributions of individual primary sugar alcohols and saccharide compounds**
 259 **quantified in PM₁₀ samples at two sites over summer and autumn periods (June to November) corresponding to**
 260 **maximal atmospheric concentrations of sugar alcohols/saccharide compounds.**

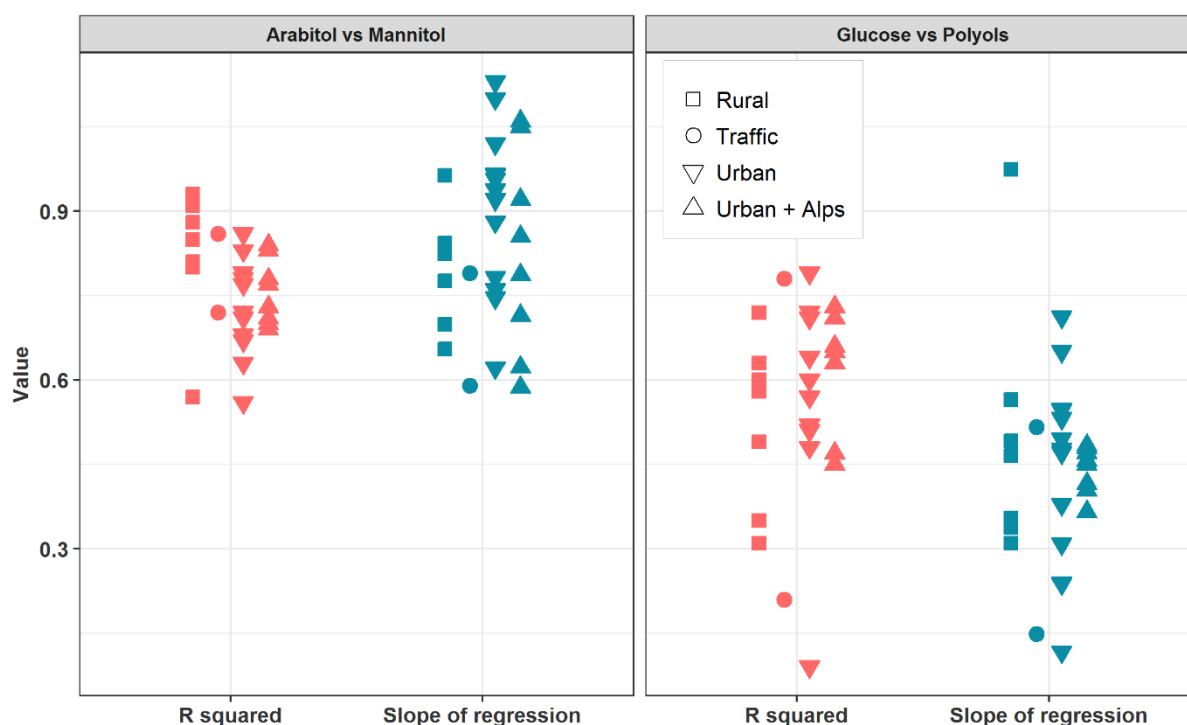


262
 263 **Figure 3: Average mass concentration distributions of arabitol, mannitol, and glucose. Symbol (*) indicates PM_{2.5}**
 264 **aerosol samples. Only data for warmer season (June to November), corresponding to maximal atmospheric**
 265 **concentration of polyols and glucose are shown.**

266 Although various primary sugar alcohols and saccharides have been detected and quantified for most of the
267 investigated sites, the following study focuses only on the three major and ubiquitous species, namely arabitol,
268 mannitol and glucose. Figure 3 presents their average relative contributions at all sites, for the warmer period,
269 displaying very similar features at a first glance. However, discrepancies could be observed from site to site, as
270 discussed in following sections.

271 **3.2 Relationships between selected primary sugar compounds**

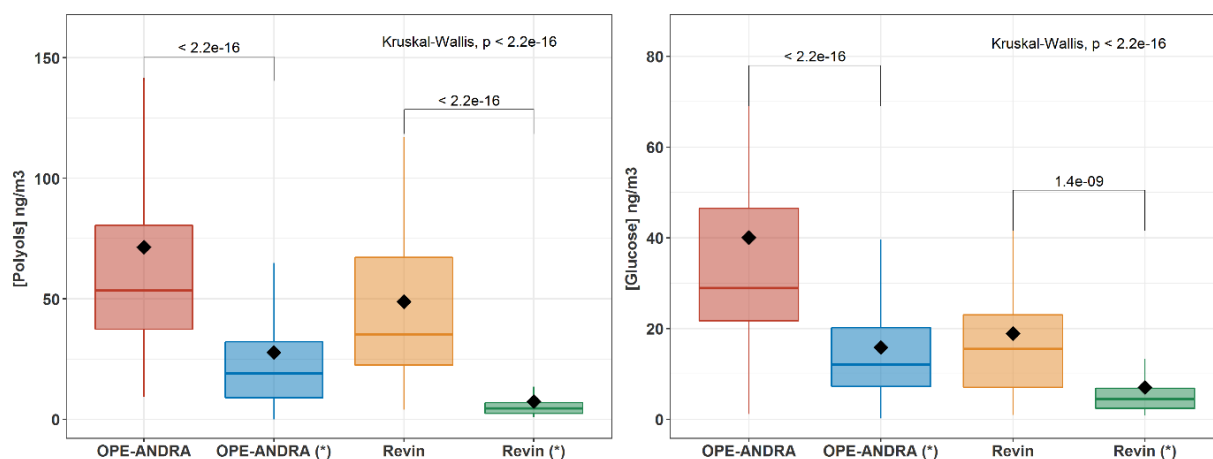
272 Figure 4 summarizes linear correlations obtained between arabitol and mannitol concentrations at each site during
273 the warmer period. Medium to very high correlation coefficients could be observed ($0.58 \leq R^2 \leq 0.93$; $30 \leq n \leq$
274 143 or $45 \leq n \leq 341$ for $PM_{2.5}$ and PM_{10} series, respectively), with slopes in a rather narrow range (between 0.59
275 and 1.10), and quite low intercepts (always below 9 ng m^{-3}). Such covariations indicate that both species are most
276 probably co-emitted, by one or several type(s) of sources, at each site during the summer-autumn period. These
277 observations are in agreement with previous studies also showing strong covariations between arabitol and
278 mannitol (Kang et al., 2018; Verma et al., 2018; Zhu et al., 2015). Therefore, it seemed reasonable to consider both
279 species together, so that their concentrations are summed up and labelled as "polyols" in the following sections.
280 Conversely, linear correlations between glucose and polyols concentrations are generally weaker
281 ($0.10 < R^2 \leq 0.78$), with slopes varying over a much larger range (between 0.12 and 0.94), and variable intercepts
282 (between -5.6 and 16.4 ng m^{-3}). This suggests that glucose concentrations might follow a different pattern
283 compared to that of polyols, either due to different emission sources, or different chemical stability in the
284 atmosphere. It is therefore reasonable to keep glucose as a separate chemical species in the following discussion.
285 It should be emphasized that the variability in the slope of the regressions between the chemical concentrations
286 are most probably related to the emissions and atmospheric processing. Particularly in the case of mannitol and
287 arabitol, they may be influenced by biogenic or biotope characteristics. Nevertheless, no evident relationship
288 between the slope values and the typology or the geographical location of the sites could be observed (Figure 4).



289
 290 **Figure 4: Linear regression analysis between selected primary sugar compounds mass concentrations (i.e. arabitol,**
 291 **mannitol and glucose) during summer and autumn seasons (June to November), for all the sites considered in this study.**

292 **3.3 Relative distributions between PM₁₀ and PM_{2.5}**

293 Figure 5 shows the average PM₁₀ and PM_{2.5} concentrations of polyols and glucose at OPE-ANDRA and Revin
 294 during the summer and autumn seasons. The polyols mass concentrations ranged from 7.5±10.9 to 27.8±33.3 ng m⁻³
 295 ³ in PM_{2.5}, and from 48.9±38.2 to 73.5±61.8 ng m⁻³ in PM₁₀, in Revin and OPE-ANDRA sites, respectively. PM₁₀-
 296 to-PM_{2.5} ratios were then on average of about 3 to 5. Similar size distribution patterns, with variable intensity, were
 297 observed for glucose (Fig. 5). These results indicate that polyols and glucose are mainly associated with the coarse
 298 PM fraction. This observation is in good agreement with several previous investigations where polyols (especially
 299 arabitol and mannitol), together with glucose, were prevalent in the coarse fraction (Fu et al., 2012; Fuzzi et al.,
 300 2007; Pio et al., 2008; Yttri et al., 2007b). However, Carvalho et al. (2003) reported different size distributions for
 301 polyols and glucose, with variable fine or coarse mode maxima depending upon sampling location. For instance,
 302 maximum atmospheric concentrations of mannitol were associated to fine and coarse aerosols from boreal forest
 303 (Finland) and rural background sites (Germany), respectively. The authors hypothesized that these observations
 304 are due to different assemblages of dominant fungal biota (with variable aerodynamic characteristics) at different
 305 sites. Some other previous studies showed aerodynamic diameters typically ranging from 2 to 10 μm, even though
 306 a few airborne bacterial/fungal spores could exceed that size (Bauer et al., 2008a; Elbert et al., 2007; Huffman et
 307 al., 2012; Zhang et al., 2015).



308
 309 **Figure 5: Box plots of mass concentrations of polyols (left) and glucose (right) in PM₁₀ and PM_{2.5} (with symbol (*)**
 310 **samples). Black markers inside each boxplot indicate the mean concentration value, while the top, middle and bottom**
 311 **lines of the box represent the 75th, median and 25th percentile, respectively. The whiskers at the top and bottom of the**
 312 **box extend from the 95th to the 5th percentile. Number of samples were N = 123 for OPE-ANDRA and N = 87 for Revin,**
 313 **respectively. Statistical differences between average mass concentrations were analyzed with the Kruskal-Wallis**
 314 **methods (p < 0.05).**

315 Hence, although if the precise mechanisms of atmospheric emission of particulate polyols and glucose are not fully
 316 resolved, our observations are in good agreement with ambient mass concentrations of polyols and glucose being
 317 likely associated with biological particles, as already suggested elsewhere (Fu et al., 2012; Verma et al., 2018;
 318 Zhang et al., 2015). These species could enter into the atmosphere through either natural or anthropogenic
 319 resuspension of surface soils and associated bacterial/fungal spores (containing polyols and primary sugar
 320 compounds), or via a direct input resulting from microbial activities (e.g. sporulation). Another hypothesis would
 321 be the abrasion of leaves and the subsequent release of microbial organisms and plant debris (Fu et al., 2012;
 322 Medeiros et al., 2006; Simoneit et al., 2004).

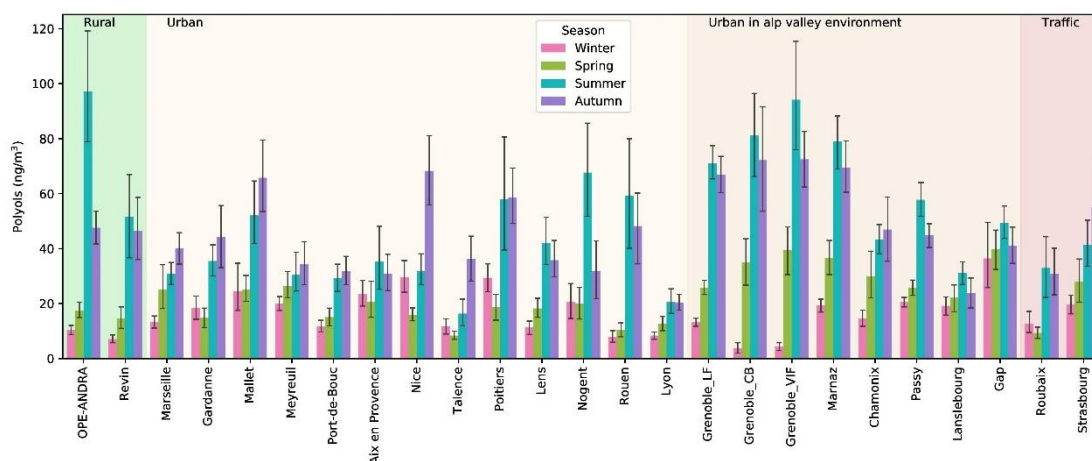
323 3.4 Spatial and seasonal distribution of atmospheric concentrations

324 3.4.1 Spatial and seasonal patterns of polyol concentrations

325 As illustrated in Fig. 6, significant concentrations of polyols were measured at each investigated site, evidencing
 326 the ubiquity of these organic compounds. The annual average concentration levels of polyols measured in PM₁₀
 327 aerosols at all sites (33.2±33.5 ng m⁻³; see Table SI-2) are within the range previously reported for urban and rural
 328 sites across Europe (Burshtein et al., 2011; Di Filippo et al., 2013; Pietrogrande et al., 2014; Yttri et al., 2007,
 329 2011). Additionally, polyols mass concentrations clearly exhibit seasonal trends, with variable intensity according
 330 to the sampling sites. On a seasonal average, polyols are more abundant in summer (46.8±43.6 ng m⁻³) and autumn
 331 (43.0±36.7 ng m⁻³), followed by spring (19.0±13.6 ng m⁻³) and winter (16.2±11.5 ng m⁻³). The average
 332 concentrations of polyols are then at least 2 to 3 times higher during summer or autumn months than during the
 333 cold months, with a ratio that can be as high as 8 to 10.

334 Previous studies also reported similar seasonal variation pattern for urban and rural aerosol samples collected at
 335 various locations. For example, Pashynska et al. (2002) measured higher atmospheric polyol (arabitol, mannitol)
 336 contents during late summer and autumn, in Belgium. Several other studies reported higher concentrations of
 337 polyols in summer than spring and winter time, in aerosols collected from Texas, USA and Jeju Island, respectively

338 (Fu et al., 2012; Jia et al., 2010a, 2010b). More recently, Liang et al. (2016) and Verma et al. (2018) also reported
 339 similar seasonal distributions for aerosols sampled at Beijing, China and north-western Pacific, respectively.
 340 The higher atmospheric polyols concentrations observed are likely due to the increased contribution from
 341 metabolically active microbial derived sources (fungi, bacteria, green algal lichens) as a result of external stressors
 342 such as heat, drought and relative moisture. Indeed, fungal and prokaryotic cells activities, including their growth
 343 and sporulation, are promoted by high temperature and humid conditions occurring in summer and autumn (China
 344 et al., 2016; Elbert et al., 2007; Jones and Harrison, 2004; Rathnayake et al., 2017).
 345 As also evidenced from Fig. 6, atmospheric polyols concentrations do not present any significant seasonal
 346 differences related to the site typology (rural, traffic, urban sites with/without Alpine influences), or latitude. There
 347 is some tendency toward higher concentrations in summer in Alpine environments, but some other sites (like the
 348 rural site of OPE-ANDRA, in the North-East of France) can reach the same levels of concentrations. We tested
 349 several types of hierarchical classifications, including variables like monthly or seasonal mean polyols
 350 concentrations, the ratio arabitol-to-mannitol, or linear regression parameters (slope, R square) but none of them
 351 led to a simple clustering of the sites that would explain the variability of the concentrations.

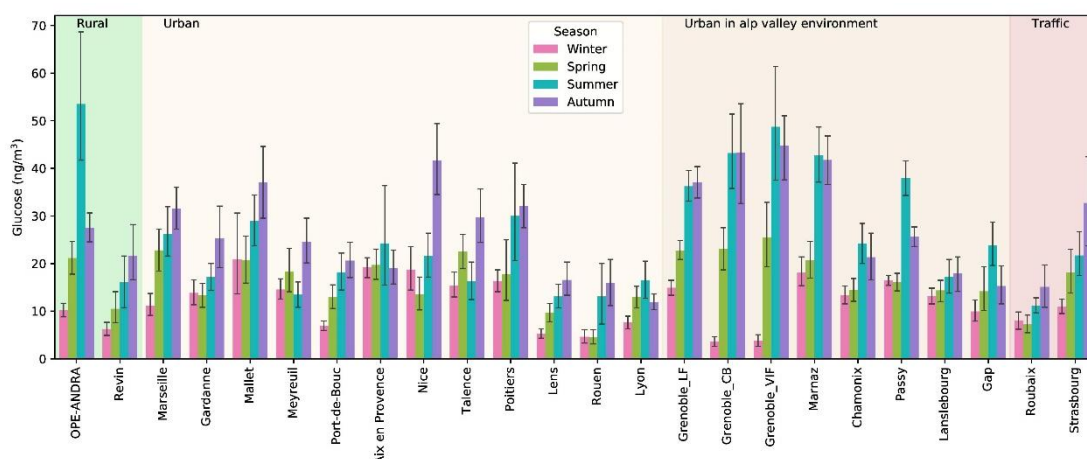


352
 353 **Figure 6: Spatial and seasonal distributions of atmospheric polyol average concentrations (ng m^{-3}) for various types of**
 354 **sites in France. Error bars correspond to standard deviations calculated with seasonal concentrations. The seasons were**
 355 **defined as follows: Winter = December to February, Spring = March to May, Summer = June to August, and Autumn**
 356 **= September to November.**

357 3.4.2 Spatial and seasonal patterns of glucose concentrations

358 The annual average concentrations of glucose measured in PM_{10} aerosols at all sites ($20.4 \pm 15.6 \text{ ng m}^{-3}$; see Table
 359 SI-2) are comparable to those previously reported for various sites across Europe (Alves et al., 2006; Theodosi et
 360 al., 2018; Yttri et al., 2007, 2011). Likewise polyols, the atmospheric concentrations of glucose also display
 361 seasonal and site-to-site variations (Fig. 7). The ambient seasonal mean concentrations (with standard deviations)
 362 of glucose are maximum in summer ($25.0 \pm 24.2 \text{ ng m}^{-3}$) and autumn ($24.6 \pm 19.8 \text{ ng m}^{-3}$), followed by spring
 363 ($15.8 \pm 12.4 \text{ ng m}^{-3}$) and winter ($12.6 \pm 10.2 \text{ ng m}^{-3}$). The summer / winter ratio for glucose seems generally lower
 364 than that of polyols, with higher ratios in the Alpine areas than in other parts of France. However, as for polyols,
 365 it remains difficult to classify the sites according to any criteria linked to site typology or latitude.
 366 The seasonal trend of glucose concentrations in the present work is similar to that recently observed for aerosols
 367 (PM_{10} or total suspended particles) collected at various environmental background (suburban, urban and coastal)

368 sites around the world (Liang et al., 2016; Srithawirat and Brimblecombe, 2015; Verma et al., 2018). On average,
 369 a wide range of daily glucose concentrations (expressed as min-max, mean) in PM₁₀ (0.1-297.2 ng m⁻³,
 370 20.6±18.7 ng m⁻³) were observed in the present study. These values are comparable to those in PM₁₀ (8.4-93.0,
 371 47.0 ng m⁻³) reported from an urban site in Belgium (Yttri et al., 2007b). More recently, Liang et al. (2016) also
 372 reported similar concentrations in PM₁₀ (3.1-343.6, 46.2±27.5 ng m⁻³) from Beijing (China).
 373 The sources and formation processes of glucose in the atmosphere are not currently well known and are rarely
 374 discussed. Glucose is an important carbon source for soil metabolic active microbiota, and it is commonly present
 375 in vascular plants. Additionally, cellulose (a linear polymer made of glucose subunits linked by β-1,4 bonds) is
 376 one of the most important form of organic compounds in terrestrial ecosystems and a major plant structural
 377 polymer (Boex-Fontvieille et al., 2014). It can also be quite abundant in the atmosphere (Puxbaum and Tenze-
 378 Kunit, 2003). Hence, it is hypothesized that ambient glucose could be formed through active microbial (i.e.
 379 bacteria, fungi etc.) enzymatic hydrolysis of cellulose in plant debris. Consistent with these observations, glucose
 380 could be released into the atmosphere from both vascular plant materials (e.g. leaves, fruits, pollens, etc.) growing
 381 in spring and decomposing in autumn/summer, and soil microbiota, as already suggested elsewhere (Di Filippo et
 382 al., 2013; Jia et al., 2010a; Medeiros et al., 2006; Verma et al., 2018; Zhu et al., 2015).



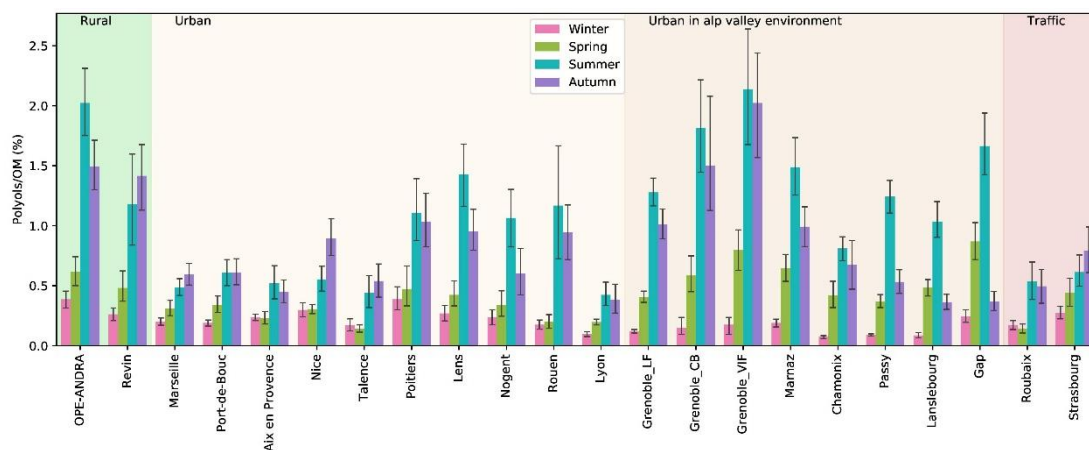
383
 384 **Figure 7: Spatial and seasonal distributions of atmospheric glucose levels (ng m⁻³) for various types of sites in France**
 385 **(except the site of Nogent, which presented too many missing values. Error bars correspond to standard deviations**
 386 **calculated with seasonal concentrations.**

387 3.4.3 Relative contributions to aerosol organic matter concentrations

388 The average contribution of polyols to the organic matter content (OM) of PM clearly displayed a seasonal
 389 behavior, as shown in Fig. 8. Here again, contributions are 2 to 10 times higher during summer and autumn
 390 compared to winter and spring, consistent with the assumption of higher emissions during these periods. The
 391 seasonal mean contribution of polyols to OM fluctuates from site to site, and accounts for 0.1 to 2.1 % of overall
 392 OM for these French sites (Fig. 8). Similarly, the seasonal mean concentrations of polyols together with glucose
 393 represent between 0.2 to 3.1 % of total OM at these sites (Fig. S2). However, on a daily basis (Samake et al., in
 394 prep.), atmospheric polyols mass concentrations can represent up to 6.3 % of total OM in PM₁₀, indicating that
 395 polyols can be amongst the major molecular species identified in aerosol organic matter (Fig. S3). Again, we could
 396 not find any simple way to group the sites according to their characteristics (typology or latitude, or climatic

397 region), in order to better understand the drivers behind the variability of this mass fraction. Further studies are
 398 currently conducted using multi criterion examinations.

399 The seasonal polyols-to-OM distribution patterns in this study are comparable to those found for different urban
 400 or rural sites in Europe (around 0.2 to 2.5 % of OM) (Pashynska et al., 2002; Yttri et al., 2007b). Zhu et al. (2015)
 401 also reported similar seasonal polyols-to-OM contribution trend for aerosols sampled at Cape Hedo (coastal site,
 402 Japan).



403
 404 **Figure 8: Spatial and seasonal distributions of mean contributions (in %) of polyols to the organic matter content of**
 405 **PM for various types of sites in France. Daily time-series of organic carbon (OC) were not available for the following**
 406 **sites: Gardanne, Mallet, and Meyreuil. Error bars correspond to standard deviations calculated with seasonal**
 407 **concentrations.**

408 3.5 Primary biogenic factor in PMF studies

409 The sum of polyols (arabitol + mannitol) represent only a small fraction of the total OM. However, as proxies of
 410 PBOA, they are most probably emitted with other chemical species. Emission from biological particles is a
 411 complex topic since it may include a wide variety of compounds, both organic and inorganic (Elbert et al., 2007;
 412 Zhang et al., 2015). Moreover, it is not clear if polyols are mainly emitted directly in the atmosphere or are linked
 413 to other materials, for example with soil dust during resuspension processes. To investigate the relationship
 414 between polyols and other species, it would have been possible to perform simple correlation analysis with
 415 individual chemical species. This approach has the disadvantage of being a one-to-one relation and thus highly
 416 sensitive to the dynamics of all PM emission sources, not only the one we are interested in. Alternatively, another
 417 way is to use a PMF approach, also based on correlations but including much more information on the temporal
 418 variations of the different sources influencing the PM chemistry at a given receptor site.

419 As mentioned in Sect. 2.3, the PMF results used in this study include sites of different typologies (rural, traffic,
 420 urban sites in Alpine valley environments, etc.) for 16 different locations spread over France and part of the current
 421 dataset. At each site, the PMF studies allowed to identify a PBOA factor, characterized by the presence of more
 422 than 90 % of the total polyols content (sum of arabitol, mannitol and sorbitol), as presented in table SI-5 and
 423 Fig. SI-6. Moreover, the sensitivity of this factor to random noise in the data was investigated thanks to randomly
 424 re-sampling the input matrix of observation. In PMF analysis, this is done via the bootstrap method (Paatero et al.,
 425 2014) in the constrained run. The PBOA factor was always mapped to itself for 13 of the sites and quasi-always
 426 (97%) for the last three ones. It means that the PBOA factor does have a very high statistical stability since it never
 427 swaps with another factor (see Fig. SI-5). Hence, the chemical composition of this factor may be informative to

428 investigate the PBOA source components (Table S6), and to evaluate the importance of PBOA emissions in term
429 of OM mass apportionment.

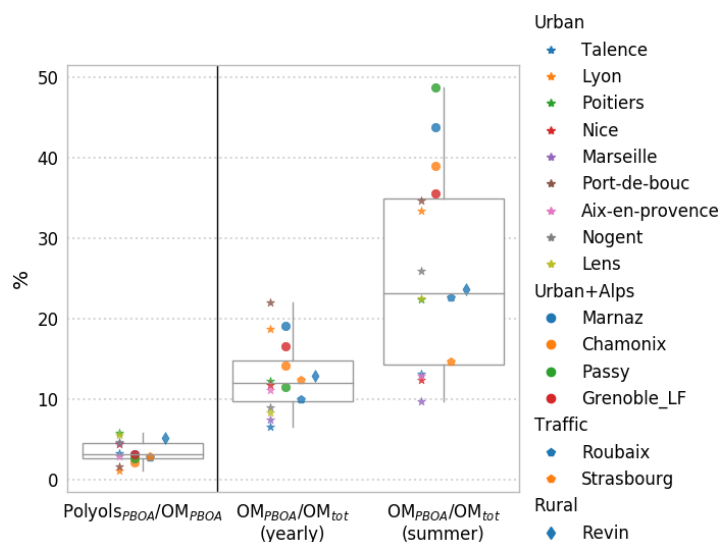
430 **3.5.1. Contributions of PBOA to OM and polyols to PBOA**

431 Altogether, the results from these 16 sites highlight the importance of the PBOA source contribution to total OM.
432 As shown in Fig. 9, the OM apportioned by the PBOA factor represents a significant fraction of the total OM mass
433 on a yearly average (range 6–28 %; average 13 ± 6 %). When considering only the summer period (June-July-
434 August), this contribution is even larger and can exceed 40 % of the total OM at sites in the Alpine area (Marnaz,
435 Passy, Chamonix, Grenoble_LF) which are partially protected from large regional influences due to the local
436 topography. This result may be nuanced, in particular during summer, since some extent of mixing between PBOA
437 and Biogenic Secondary Organic Aerosols (BSOA) cannot be entirely excluded. However, several evidences tend
438 in favor of a non-significant mixing between BSOA and PBOA. First, the ratio of polyols-to- OC_{PBOA} shows a low
439 variability from site to site, while it is unlikely that such a secondary process led to the same amount of OC for all
440 sites since they present different meteorology, sunshine duration etc. Second, the bootstrap analysis do not show
441 any “swap” between factors for the PBOA profile for all sites, indicative of a well-defined factor (see Fig. SI-5).
442 Finally, the OC_{PBOA} -to-polyols ratio in this work (about 16) is in the range of ratio expected for fungal spores (12
443 -27, when arabitol and mannitol are considered together) (Bauer et al., 2008; Yttri et al., 2011).

444 Interestingly, some previous work using the same samples from the sites in the Arve valley (Passy, Chamonix)
445 showed that about 90 % of the OM is from modern origin (using ^{14}C measurements) during summer, with no
446 apparent correlation between this modern carbon and polyols concentrations (Bonvalot et al., 2016). Hence, despite
447 being an important contributing source, PBOA is not the major biogenic source in this type of environment.

448 Interestingly, opposite to the case of the Alpine valleys where this proportion is the highest, the ratios OM_{PBOA} -to-
449 OM_{total} are amongst the lowest for coastal environments (Talance, Marseille, Nice), a possible indication that the
450 marine environment is not a large emitter for these species. Recently, much lower concentrations of polyols in
451 aerosols from marine environments than those in terrestrially influenced sites were also reported off the coast of
452 Japan, also suggesting a higher contribution from terrestrial sources (Kang et al., 2018).

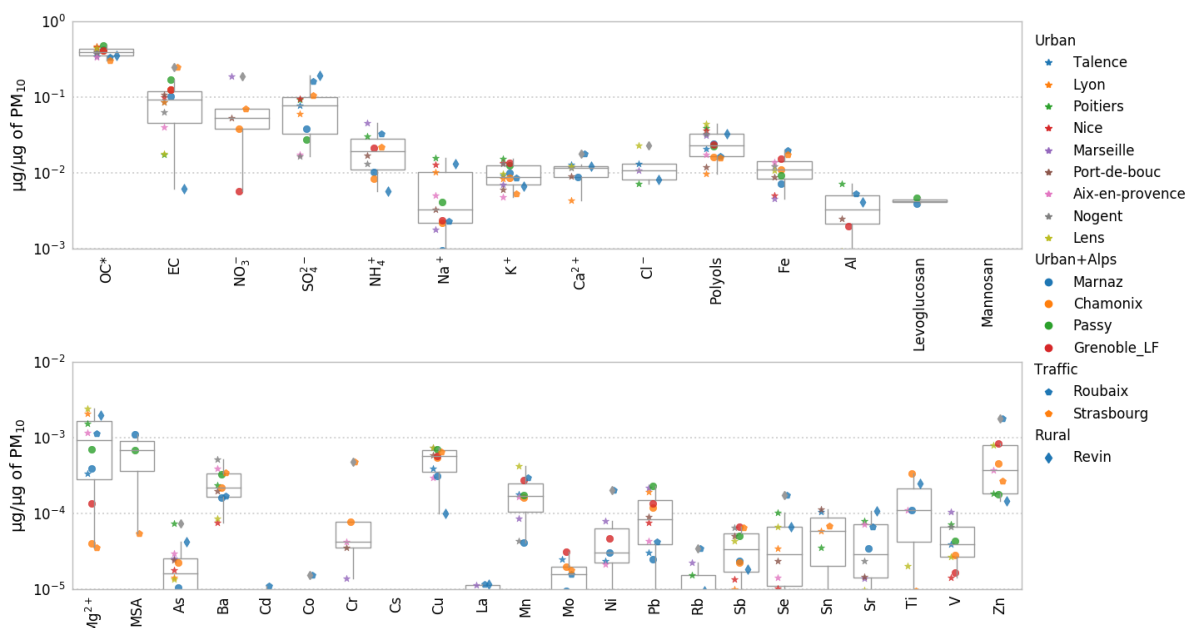
453 As illustrated in Fig. 9, polyols represent only a small fraction of the OM apportioned in the PBOA factor (1.2 %–
454 6.0 %; average 3.0 ± 1.5 %) for the 16 studied sites. This variability is indeed rather small, considering the wide
455 range of sites and the diversity of other potential sources (on average 8 to 10 PMF factors were obtained for the
456 different sites). Indeed, this narrow range of the polyols fraction to the OM_{PBOA} highlights the stability of the
457 chemical profile of this source over a large regional scale. It indicates also that, if polyols are good proxies of the
458 PBOA sources, a large amount of other organic species are co-emitted, that still remain unknown.



459
 460 **Figure 9: Mass contribution of polyols to OM in the PBOA factor, and relative contributions of the OM_{PBOA} factor to**
 461 **the total OM in PM for the 16 studied sites where PMF model was run, over the year and summertime only. Stars and**
 462 **circle refer to urban sites without/and with Alpine valley influence, respectively. Pentagon corresponds to traffic sites**
 463 **and diamond to rural sites.**

464 3.5.2 PBOA profile constituents and emission process

465 Figure 10 shows the contribution (in μg of species per μg of PM in the PBOA factor profile) of each chemical
 466 species included in the averaged PBOA factor from the 16 PMF studies. The principal contributors are OC and
 467 EC, and significant fractions of crustal material also appear (Na^+ , K^+ , Ca^{2+} , Al, Ba, Cu, Fe, Mn, Ti, Zn) as well as
 468 secondary elements such as nitrate and sulfate. However, EC appears to be highly variable both within and between
 469 sites under consideration. The reader may refer to figure SI-7 for an estimation of the EC mass uncertainties in the
 470 different sites. On average, the PBOA factor does not comprise a large fraction of metals and trace elements, most
 471 of them being below $1 \text{ pg } \mu\text{g}^{-1}$. Here again, the low variability of the PBOA chemical profile encountered across
 472 a large array of sites is remarkable.



473
 474 **Figure 10: PMF chemical profile of the PBOA factor in the DECOMBIO and SOURCES programs expressed as a**
 475 **fraction of the PM mass. Values lower than a few $\text{pg } \mu\text{g}^{-1}$ are not displayed on purpose. For each boxplot, the top, middle**

476 and bottom lines represent the 75th, median and 25th percentile, respectively. The whiskers at the top and bottom of the
477 box extend from the maximum to the minimum. OC* corresponds to the bulk organic carbon fraction minus individual
478 molecular weight of characterized organic species.

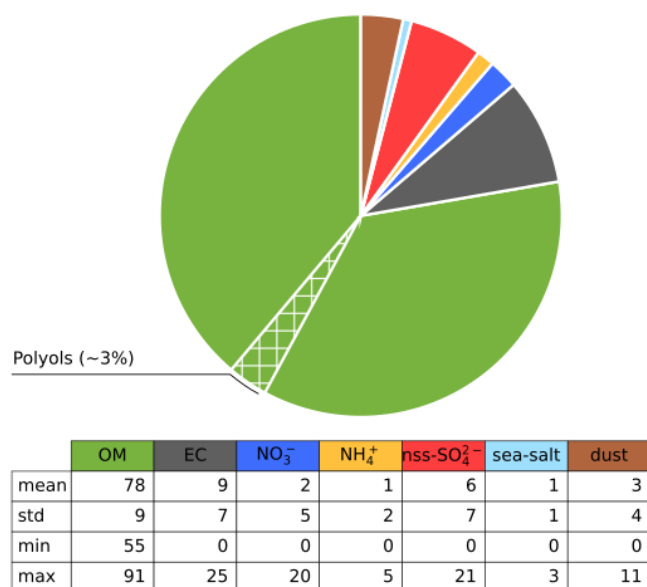
479 The contribution from some crustal material could agree with the coarse mode distributions of polyols (Sect. 3.3)
480 and could be indicative of an emission process with the entrainment of spores with soil dust resuspension. To
481 investigate the importance of mineral dust in the PBOA factor, we clustered the chemical components of PM from
482 PBOA into 7 classes: OM ($= 1.8 \times \text{OC}$), EC, NO_3^- , NH_4^+ , non-sea-salt sulfate (nss- SO_4), sea-salt, and dust. nss-
483 SO_4^{2-} is calculated from the measured SO_4^{2-} minus the sea-salt fraction of SO_4^{2-} ($\text{nssSO}_4^{2-} = \text{SO}_4^{2-} - \text{ssSO}_4^{2-}$
484 where $\text{ssSO}_4^{2-} = 0.252 \times \text{Na}^+$) according to Seinfeld and Pandis (1997). The sea-salt fraction is calculated
485 according to Putaud et al. (2010): $\text{sea} - \text{salt} = \text{Cl}^- + 1.47 \times \text{Na}^+$. Finally the dust fraction is estimated from
486 Putaud et al. (2004b) as: $\text{dust} = (\text{nss} - \text{Ca}^{2+}) \times 5.6$ with nss- Ca^{2+} stands for non-sea salt Ca^{2+} and is computed
487 thanks to $\text{nss} - \text{Ca}^{2+} = \text{Ca}^{2+} - \text{Na}^+ / 26$.

488 Figure 11 presents the normalized average contributions of these 7 classes to the PBOA mass for the 16 sites with
489 PMF modelling. It clearly revealed that the PBOA factor is dominated by contributions from OM (78±9 %),
490 followed by EC (9±7 %), and only a minor contribution from the dust class (3±4 %).

491 The large value for the contribution of EC is driven by two high values obtained at the sites of Strasbourg (that
492 reaches 25%) and Chamonix (18%) both influenced by direct and indirect traffic emissions. However, 6 other sites
493 present no EC in PBOA. Moreover, the uncertainties of EC in the PBOA profile of Strasbourg and Chamonix is
494 rather high (between 5 to 30% of PM mass at Strasbourg, see SI-7). On a yearly average, EC apportioned by this
495 factor (0 to 400 ng m⁻³ depending on the site) is close to the rural EC background in France of about 300 ng m⁻³
496 (Golly et al., 2018).

497 This result on the general chemical profile of the PBOA factor tends to infirm the hypothesis of an emission process
498 associating PBOA material with mineral dust resuspension. Indeed, our findings rather suggest that a main part of
499 PBOA (and polyols) are most likely associated with biological particle direct emissions. It leaves only a minor
500 fraction that could be linked to the mechanical resuspension of PBOA with crustal elements. Some minor fraction
501 of EC in this factor could come from resuspended EC-containing dust particles being accumulated in topsoil as
502 demonstrated in previous works (Forbes et al., 2006; Hammes et al., 2007; Zhan et al., 2016). Hence, the origin of
503 the larger fraction of the contribution of EC remains unknown. Our conclusions are in good agreement with those
504 made by Jia and Fraser (2011), based on the concentrations of these chemicals in different types of samples: i.e.
505 size-fractionated (equivalent to PM_{2.5} and PM₁₀) soil, plant, fungi, atmospheric PM_{2.5} and PM₁₀. They found that
506 the ambient concentrations of primary saccharide compounds at the suburban site of Higley (USA) are typically
507 dominated by contributions of biological materials rather than resuspension of soil dust particles and associated
508 microbiota.

509



510
 511 **Figure 11: Average contribution (%) of species in the PBOA factor for the sites in SOURCES and DECOMBIO. The**
 512 **hatched area represents the proportion of the OM apportioned by the polyols (see text for reconstruction method).**

513 **4. Conclusion**

514 The contribution of primary biogenic organic aerosols to PM is barely documented in the scientific literature. The
 515 present study aimed at providing a large overview of the spatial and temporal evolution of concentrations and
 516 contributions to aerosol organic matter of dominant primary sugar alcohols and saccharide compounds, for a large
 517 selection of environmental conditions in France. With 28 sites and more than 5,340 samples, it is to our knowledge
 518 the most comprehensive dataset for these compounds. The main results obtained indicate that:

- 519 • among the identified polyols, arabitol together with mannitol are the major species by mass, with lesser
 520 amounts of others polyols (e.g., erythritol, inositol, glycerol, sorbitol, and xylitol). Glucose is the
 521 dominant primary monosaccharide and its relative abundance is comparable to the sum of arabitol and
 522 mannitol;
- 523 • the two main polyols (arabitol and mannitol) together with glucose are mainly present within the coarse
 524 aerosol mode;
- 525 • at nearly all sites, ambient levels of the main polyols and glucose displayed clear seasonal variation cycles,
 526 with a gradual increase from spring and maximum in summer and autumn aerosols, followed by a sudden
 527 decrease in late autumn, and a winter minimum;
- 528 • atmospheric concentrations of the main polyols and glucose fluctuate according to site and season, and
 529 account each for between 0.1 to 2.1 % of OM on a seasonal average basis at these French sites;
- 530 • ambient mass concentrations of arabitol and mannitol are comparable. Meanwhile, they display very good
 531 temporal covariation, with ratios varying between sites. Conversely, linear correlations between the main
 532 polyols and glucose concentrations are much lower, suggesting different atmospheric sources, or
 533 atmospheric processes;

- 534 • arabitol and mannitol are efficient organic markers for PBOA. PMF studies of the yearly series from 16
535 sites give contributions of the primary biogenic emission (traced with the main polyols) to the total OM
536 around $13\pm 6\%$ on a yearly average and $26\pm 12\%$ during summer, thereby showing that PBOA is an
537 important source of total OM in PM_{10} for all sites across France. Furthermore, the average PBOA
538 chemical source profile is made out of a very large fraction of OM ($78\pm 9\%$ of the total PBOA mass on
539 average), suggesting it is mainly related to direct biogenic emissions from biological particles.
540 Noteworthy, the presence of BSOA within the PBOA factor, particularly during summer could not be
541 fully ruled out and further works using additional organic tracers (such as 3-MBTCA, pinic acid, and/or
542 cellulose) are still needed to solve this issue. Additionally, the low crustal fraction indicates that this factor
543 is weakly linked to soil dust resuspension associated with biological material;
- 544 • however, the PBOA source remains chemically poorly characterized as the main polyols represent only
545 a small fraction of its total OM mass ($3.0\pm 1.5\%$ on average);
- 546 • despite comparable high concentrations in the atmosphere, the sources and processes leading to glucose
547 concentrations and seasonal evolutions are still elusive. Indeed, the different PMF performed with glucose
548 in input variable do not lead to statistically stable solution;

549 Further investigations of the emission pathways and chemical characterization of the PBOA source associated with
550 polyols are on-going, which may improve our understanding of its dynamic at various geographical scales, for a
551 potential implementation in emission models in the future.

552

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565

566 **Author contributions:** JLJ was the supervisor for the PhD for AS, FC, SW, and for the post-doc of DS. He
567 directed all the personnel who performed the analysis at IGE. He was coordinator or PI for the programs that
568 generated the data for 18 of the 28 sites in this study (OPE-ANDRA, Part'Aera, CAMERA, SRN 2013, 3 Villes
569 PACA, DECOMBIO, QAMECS) and co-PI for programs for 5 other sites. He is the coordinator for the CNRS
570 LEFE-EC2CO CAREMBIOS program that is funding the work of AS. GU was the co-supervisor for the PhD of
571 AS and SW. OF is the coordinator of the CARA program, (co-)funding and supervising the filter sampling and
572 chemical analyses at 12 of the 28 sites. EP, OF, and VR supervised the PhD of DMO who investigated the 5 sites
573 in northern France. Finally, JLB was the coordinator (program Lanslebourg) or partner of several programs whose

574 data were used in this study (OPE-ANDRA, Part'Aera, 3 Villes PACA, DECOMBIO), and OF was the coordinator
575 of the SOURCES program, which includes the work of DS as a post-doctoral fellow under the supervision of JLJ
576 to gather and prepare most of the datasets used in the present studies.
577 All authors from the ANDRA (#5) and AASQA (#6 to 13) are representatives for each network that conducted the
578 sample collection and the general supervision of the sampling sites.
579 FC and DS ran the PMF analysis. AS, SW and JLJ processed the data and wrote up the manuscript. All authors
580 reviewed and commented on the manuscript.

581 **Competing interests:** The authors declare that they have no conflict of interest.

582

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