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We thank the editors and anonymous referees for their constructive corrections and comments that greatly improved the present manuscript. We have considered carefully the different corrections suggested by both editors and anonymous referees, and made corrections accordingly in the manuscript (*red color*). The detailed answers to the specific questions are given below, point by point *in blue color*.

Answer to Anonymous Referee #4 comments

L.47: Please add “organic matter (OM)”

This information has been added (line 51).

L. 48-51: I find the current ending of the abstract somewhat confusing, going into details on a minor contribution of dust. My suggestion would be instead to provide some “useful” results from the analysis, which will certainly increase impact and relevancy of the work. One aspect that comes to my mind would be the estimate of PBOA mass concentration based on polyols (or individual arabitol or mannitol), a number which you can provide with uncertainties based on PMF analysis of 1-year sampling from 16 sites (no small feat). And then finalizing with a well designed sentence on the goal and relevancy of this study would wrap up much better the abstract, in my opinion.

We agree with the reviewer that providing the contribution of Polyols to PM_{PBOA} mass can bring added value. Here we provide some statistics concerning the Polyols-to- PM_{PBOA} ratio. We modified the end of the manuscript by adding the following text (lines 47-49):

“The Polyols-to- PM_{PBOA} ratio is about 0.024 ± 0.010 on average for all sites, with no clear distinction between traffic, urban or rural typology. Overall, even if the exact origin of the PBOA source is still under investigation, it appears to be an important source of PM, especially during summertime”.

L.64: ...”whereas a significant fraction of OM can also be associated with...”

This has been corrected accordingly (line 67).

L.80-81: “...have long been suggested as tracers of PBOA (refs here). For instance...”

Thanks for your attentive review. References are now provided (lines 84-85)

L.91: As the reader may not be familiar with specialized fluorescent based instrumentation, I’d suggest the following replacement: “with real-time detectors of fluorescent PBOA (such as UV-APS and WIBS-3), particularly in rainy periods...”

As also suggested by the editor, we are now providing the definition of both UV-APS and WIBS-3 (lines 94-95).

L.102: remove “actually”

This has been removed (line 105).

L.103: “...(Wang et al., 2018) , although studies...”

Thank for this suggestion, it has been changed accordingly (line 106).

L.117: please remove “(i.e. atmospheric input processes)

This has been removed (line 121).

L.123: Please remove the comma prior “in France”.

The comma has been removed.

L.130-133: The reader needs to be guided through the results, and not be “explained to” why some results are not being shown. Unless referring to a companion, such references to future publications do not seem well-suited and I find that your manuscript would be better without it.

We agree with the reviewer and removed the corresponding phrases (lines 133-137).

Figure 4: Does the spread of the slope arabitol vs Mannitol makes more sense if separated by biome (or coarsely by region?).

We have also tried to investigate if sites can be grouped by biome (regionally) or typology when focusing on the spreading of the slope between Arabitol vs Mannitol, but we did not observe any clear pattern.

L.334: please remove “into”.

This has been removed.

Section 3.4.1 and so forth: Please include the caveat that different sampling sites do not correspond to the same time period (you could be biased an “urban” site by a rainy summer, for example).

Thank you for this suggestion. We included the following caveat in the main text (lines 362-363)

“Year of PM sampling campaigns are not concurrent at all sites (see Fig. SI-1)”.

L.413-416: This needs to be rewritten, or removed.

This has been rephrased (line 421).

Answer to Anonymous Referee #3 comments

Line 73: change “site” to “sites”

Site has been changed to sites.

Lines 214-216: does not seem correct to label as EC to the sum of BC_{wb} with BC_{ff}. Why not BC?

We agree that EC and BC do not refer exactly to the same species. However, this simplification was used to uniform labelling of graphs (i.e. readability and clarity). Moreover, the ratio between (BC_{wb}+BC_{ff}) to EC is about 1.4±0.4 for the 3 sites (Chamonix, Marnaz and Passy). So we may overestimate the value given in Figure 10 by a factor 1.4, it may thus indeed explain the higher “EC” values for the 3 alpin valley. Nevertheless, it does not change the general conclusion

Line 176 and others: the tables and figures in the annex are called sometimes as “SI” other times as “S” in the main text and in the annex section. Please make uniform labelling.

The supplemental Tables and Figures are now indexed as SI.

Line 506: I have doubts about the correctness of using Putaud et al (2004b) coefficient (5.6) to estimate the dust fraction. The Figure in Putaud et al (2004b) (Figure 2) from where this coefficient was taken reveals a very large dispersion (R²=0.31) and is highly influenced by one only value. If this sample was removed the coefficient would be much more similar to the value of 15 obtained during the outbreak of Sahara dust. The Ca average fraction in the continental crust gives coefficient values of 28-35 (Mason B. and Moore C. B.: Principles of Geochemistry, 4 Edn., Wiley & Sons, New York, 1982. Wedepohl, K. H.: The composition of the continental crust, Geochim. Cosmochim. Ac., 59, 1217–1232, 1995). Comparison of relative ratio values between Al, Fe, Ca²⁺ and Ti, in table S6 and in the average crust, shows that in the PBOA factor the proportions of these soil tracers are quite different from the crust, which makes difficult the calculation of a soil contribution in PBOA.

Indeed, the dust concentration may vary with the different reconstruction methods found in the literature (Putaud et al. 2004; Malm et al. 1994; Querol et al. 2002; Perez et al. 2008, or the one you provide). Since we do not have all elements for every sites, we decided to choose Putaud et al, 2004

as a first approximation. Even if we agree that the relationship between dust and Ca seems highly influenced by one extreme point. It seems that without this point, the relationship would be closer to a $[\text{dust}] = 10 \times [\text{Ca}]$, i.e. twice the amount we have, leading to 6% on average for the dust fraction in the PBOA factor, which is still very low.

Besides, we should also be cautious when trying to interpret the other metals in the PBOA source factor since they are potentially associated with important uncertainties. The metals in this factor result in a mixing of different other sources, not only the dust one. It is then expected that the ratio do not fit the one of the average crust.

We decided not to change the value, since it is hard to discuss Putaud et al. without the exact values and detailed explanation of the authors. However, we added the following sentence (lines 496-497):

“We note that the conversion coefficient provided by Putaud et al. (2004b) may be influenced by an extreme value and then gives only a low estimate of dust resuspension”.

1 Polyols and glucose particulate species as tracers of primary 2 biogenic organic aerosols at 28 French sites

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33 **Abstract.** A growing number of studies ~~is~~^{are} using specific primary sugar species, such as sugar alcohols or
34 primary saccharides, as marker compounds to characterize and apportion primary biogenic organic aerosols
35 (PBOA) in the atmosphere. To better understand their annual cycles, as well as their spatio-temporal abundance
36 in terms of concentrations and sources, we conducted a large study focusing on three major atmospheric primary
37 sugar compounds (i.e., arabitol, mannitol and glucose) measured in various environmental conditions on about
38 5,300 filter samples collected at 28 sites in France. Our results show significant atmospheric concentrations of
39 polyols (defined here as the sum of arabitol and mannitol) and glucose at each sampling location, highlighting
40 their ubiquity. Results also confirm that polyols and glucose are mainly associated with the coarse rather than the
41 fine aerosol mode. At nearly all sites, atmospheric concentrations of polyols and glucose display a well-marked
42 seasonal pattern, with maximum concentrations from late spring to early autumn, followed by an abrupt decrease
43 in late autumn, and a minimum concentration during wintertime. Such seasonal patterns support biogenic
44 emissions associated with higher biological metabolic activities (e.g., sporulation, growth, etc.) during warmer
45 periods. Results from a previous comprehensive study using Positive Matrix Factorization (PMF) based on an
46 extended aerosol chemical composition dataset of up to 130 species for 16 of the same sample series has also been
47 used in the present work. The Polyols-to-PM_{PBOA} ratio is about 0.024±0.010 on average for all sites, with no clear
48 distinction between traffic, urban or rural typology. Overall, even if the exact origin of the PBOA source is still
49 under investigation, it appears to be an important source of PM, especially during summertime. Results also show
50 that PBOA are significant sources of total OM in PM₁₀ (13±4 % on a yearly average, and up to 40 % in some
51 environments in summer) at most of the investigated sites. The mean PBOA chemical profile is clearly dominated
52 by contribution from organic matter (OM) (78±9 % of the mass of the PBOA PMF factor on average), and only a
53 minor contribution from dust class (3±4 %), suggesting that ambient polyols are most likely associated with
54 biological particle emissions (e.g., active spore discharge) rather than soil dust resuspension.

55 **1. Introduction**

56 Airborne particles (or particulate matter, PM) are of major concern due to their multiple effects on climate and
57 adverse human health impacts (Boucher et al., 2013; Cho et al., 2005; Ntziachristos et al., 2007). The diversity of
58 PM impacts is closely linked to their complex and highly variable nature: size distribution, concentration and
59 chemical composition, or specific surface properties. PM consists of inorganic and elemental substances, and a
60 large fraction made of carbonaceous matter (organic carbon (OC) and elemental carbon (EC)) (Franke et al., 2017;
61 Putaud et al., 2004a; Yttri et al., 2007a). Substantial amounts of atmospheric organic matter (OM) remain
62 unidentified and uncharacterized at the molecular level. In most studies, a maximum of only 20 % of particulate
63 OM mass can generally be speciated and quantified (Alfarra et al., 2007; Fortenberry et al., 2018; Liang et al.,
64 2017; Nozière et al., 2015). This detailed composition of OM and its spatial and seasonal distribution can give
65 important insights on the adverse effects of PM. So far, the majority of air pollution studies have focused on
66 organic atmospheric particles associated with anthropogenic and secondary sources, whereas a significant fraction
67 of OM ~~is~~^{can} also ~~be~~ associated with primary emissions from biogenic sources (Bauer et al., 2008a; Jaenicke,
68 2005; Liang et al., 2016). Therefore, the characterization of primary OM biogenic sources at the molecular level is
69 still limited (Fuzzi et al., 2006; Liang et al., 2017; Zhu et al., 2015), and should be further investigated for a better
70 understanding of aerosol sources and formation processes.

71 Primary biogenic organic aerosols (PBOA) are emitted directly from the biosphere to the atmosphere where they
72 are ubiquitous and participate in many atmospheric processes (Elbert et al., 2007; Fröhlich-Nowoisky et al., 2016).
73 Additionally, their inhalation has long been associated with human respiratory impairments (e.g., asthma,
74 aspergillosis, etc.) (Després et al., 2012; Morris et al., 2011). PBOA comprises living and dead microorganisms
75 such as bacteria, fungi, viruses, bacterial and fungal spores, and microbial fragments, endotoxins, mycotoxins, or
76 pollens (Elbert et al., 2007; Jaenicke, 2005; Morris et al., 2011). In most semi-urban European sites, PBOA can
77 account for up to 25 % of the atmospheric aerosol mass, in the size range of 0.2 to 50 μm (Fröhlich-Nowoisky et
78 al., 2016; Jaenicke, 2005; Huffman et al., 2012; Manninen et al., 2014; Morris et al., 2011). However, their sources
79 and contribution to total airborne particles are still poorly documented, partly because of the difficulty to recognize
80 them by conventional microbiological methods (cells culture, microscopic examination, etc.) (Di Filippo et al.,
81 2013; Heald and Spracklen, 2009; Jia et al., 2010a).

82 Several specific chemical components, such as primary sugar compounds (i.e., primary saccharides and sugar
83 alcohols) emitted persistently from biogenic sources, have long been suggested as powerful and unique biomarkers
84 in tracing sources, and abundances of PBOA as well ([Bauer et al., 2008a](#); [Medeiros et al., 2006](#); [Simoneit et al.,](#)
85 [2004b](#); [Zhang et al., 2010](#); [Zhu et al., 2016](#)). For instance, ambient concentrations of glucose have been used as
86 markers for plant materials (such as pollen, leaves, and their fragments) or soil emissions from several areas in the
87 world (Fu et al., 2012; Jia et al., 2010a, 2010b; Pietrogrande et al., 2014; Rathnayake et al., 2017). Many studies
88 indicated that glucose is the most abundant monosaccharide in vascular plants, where it serves as the common
89 energy material, and an important source of carbon for soil active microorganisms (such as bacteria or fungi) (Jia
90 et al., 2010a; Medeiros et al., 2006; Pietrogrande et al., 2014; Zhu et al., 2015). Additionally, sugar alcohols (also
91 called polyols) including arabitol and mannitol have been proposed as markers for airborne fungi, and are widely
92 used to quantify their contributions to PBOA mass (Bauer et al., 2008a, 2008b; Golly et al., 2018; Srivastava et
93 al., 2018; Zhang et al., 2010). These sugar alcohols have also been found to correlate very well with fluorescent
94 PBOA in [the ultraviolet aerodynamic particle sizer UV-APS\(UV-APS\)](#) and [wideband integrated bioaerosol sensor](#)
95 [\(WIBS-3\)](#) online studies, particularly in rainy periods (Gosselin et al., 2016), favoring microbial sporulation (such
96 as fungi belonging to Ascomycota and Basidiomycota phyla) (China et al., 2016; Elbert et al., 2007; Jones and
97 Harrison, 2004). Polyols are produced in large amounts by many fungi and bacteria, and several functions have
98 been described for these compounds, such as common energy storage materials, intracellular protectants against
99 stressful conditions (e.g., heat or drought), storage or transport of carbohydrates, quencher of oxygenated reactive
100 species, or regulators of intracellular pH by acting as a sink or source of protons (Jennings et al., 1998; Medeiros
101 et al., 2006; Vélèz et al., 2007). Hence, polyols, especially arabitol and mannitol, may represent a significant
102 fraction of the dry weight of fungi, and mannitol can contribute between 20 to 50 % of the mycelium dry weight
103 (Ruijter et al., 2003; Vélèz et al., 2007). However, polyols are also often identified in the lower plants (leaves,
104 pollens) and green algal lichens (Medeiros et al., 2006; Vélèz et al., 2007; Yang et al., 2012). The primary sugar
105 compounds (defined as polyols and primary saccharide species) are ~~actually~~ thought to be relatively stable in the
106 atmosphere (Wang et al., 2018). ~~However,~~ [although](#) studies investigating their atmospheric lifetime are quite
107 limited. One previous laboratory study has been conducted by the US-EPA to evaluate the stability of these
108 chemicals on filter material exposed to gaseous oxidants as well as in aqueous solutions (simulating clouds and
109 fog droplet chemistry). Findings of this former study have shown that primary sugar compounds remain quite

110 stable up to 7 days (the extent of the testing period), pointing out their suitability for use as tracers of atmospheric
111 transport (Fraser, 2010). With all of this information, the use of primary sugar compounds (such as mannitol,
112 arabitol, glucose, etc.) as suitable tracers of PBOA is generally acknowledged (Jia and Fraser, 2011; Zhu et al.,
113 2015, 2016).

114 Although atmospheric concentrations of polyols, including arabitol and mannitol, as well as that of some primary
115 monosaccharides (e.g., glucose), have been previously quantified as part of several studies in various environments
116 including urban/suburban, rural, rainforest, mountain, and marine areas (Fu et al., 2012; Graham et al., 2003; Jia
117 et al., 2010a, 2010b; Liang et al., 2016; Pietrogrande et al., 2014; Simoneit, 2004a,2004b; Verma et al., 2018; Yttri
118 et al., 2007b; Zhu et al., 2015), large datasets investigating their annual cycles and spatial distributions are still
119 limited. Such information could give important insights on environmental factors influencing their atmospheric
120 levels such as climate and biotope, and therefore help to elucidate patterns regarding their major sources and
121 atmospheric emission pathways ~~(i.e. atmospheric input process)~~. Even if numerous sources and emission
122 mechanisms have been widely proposed, including among others, metabolic active microbial wet emissions,
123 entrainment of farmland or natural soils and associated microbiota (Elbert et al., 2007; Fu et al., 2013; Gosselin et
124 al., 2016; Jia et al., 2010a, 2010b; Medeiros et al., 2006; Pietrogrande et al., 2014; Simoneit et al., 2004a, 2004b;
125 Verma et al., 2018; Yttri et al., 2007b), the dominant atmospheric input processes have not been yet sufficiently
126 elucidated.

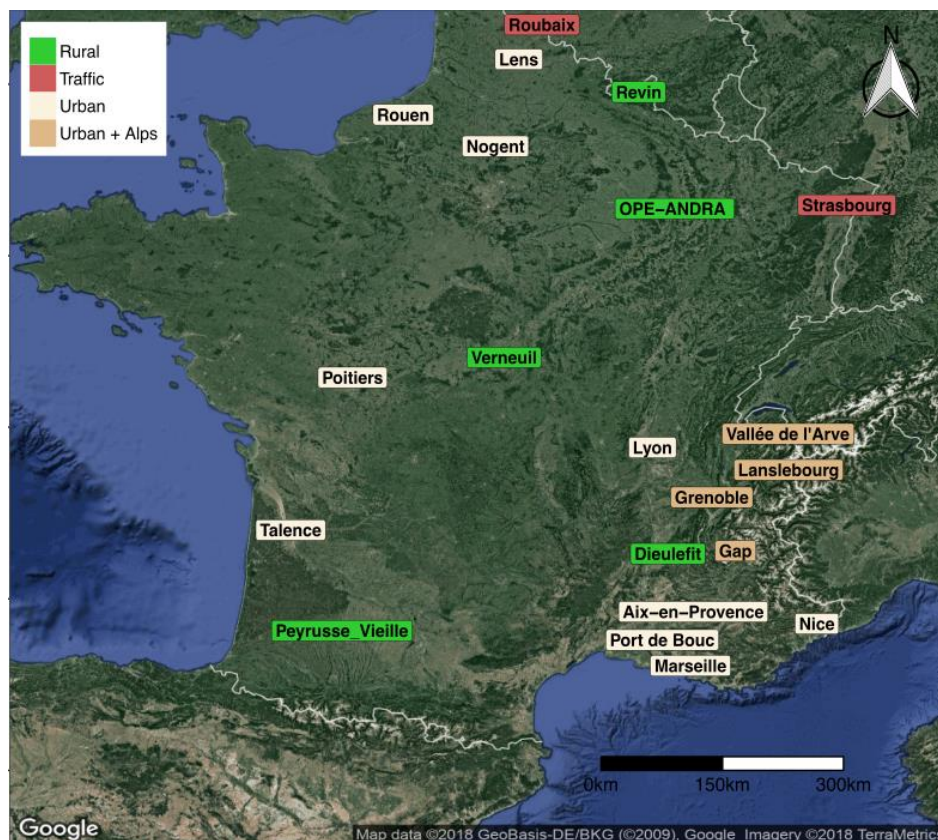
127 In this context, the present study was designed to provide a large overview of the spatial and seasonal variations
128 of polyols and glucose mass concentrations, as well as their contribution to the aerosol organic mass fraction, in
129 France. To do so, data was collected at many sites in different environments (rural, traffic, urban), in order to
130 represent various sampling conditions in terms of site typologies and meteorological conditions. Thanks to the
131 availability of results from an extended Positive Matrix Factorization (PMF) analysis performed for the
132 corresponding datasets, the overall contributions of the main polyols and glucose emission sources could also be
133 investigated in light of their spatial patterns. To the best of our knowledge, this is the first study providing such an
134 extended phenomenology of these compounds over multiple sites with different typologies. ~~This paper is the first
135 part of on-going work on the concentrations of ambient particulate polyols and glucose in France. A second part
136 will examine the short term (daily) variations of these chemicals at multiple spatial scale (local to nation wide)
137 and their potential processes for entering the atmosphere. A third part will investigate the links between the
138 atmospheric concentrations of polyols and glucose and airborne microbial fingerprint up to the genus level.~~

139 **2. Material and methods**

140 **2.1 Aerosol sampling**

141 Ambient aerosol samples considered in the present work come from different research and monitoring programs,
142 conducted over the last 5 years in France (Figure 1). Each program includes at least one-year of field sampling,
143 providing a total number of 5,343 daily filter samples available for the sake of the present study. These sites offer
144 diverse conditions in terms of typologies (i.e., rural, traffic, urban sites, Alpine valley environments, etc.), local
145 climate and vegetation types and were selected in order to cover the complex and variable national environmental
146 conditions. These sites are assumed to represent typical environmental conditions in France, and our

147 observations/and general tendency could therefore be extrapolated to neighboring western European countries
148 presenting quite homogeneous environmental conditions.



149
150 **Figure 1: Map of sampling site locations in France. Green: rural background, red: traffic, wheat: urban background**
151 **and dark wheat: urban background in Alpine valley sites. The areas of Grenoble (Grenoble_LF, Grenoble_CB and**
152 **Grenoble_VIF) and Vallée de L'Arve (Marnaz, Passy and Chamonix) include 3 sites each. The area of Marseille**
153 **includes four sites: Marseille, Mallet, Meyreuil and Gardanne.**

154 The site of OPE-ANDRA is a specific monitoring observatory in a rural environment, without any village or
155 industry within several kilometers (description available from: <http://www.andra.fr>). All other sites correspond to
156 stations of regional air quality monitoring networks (AASQA). The availability of filter samples was variable from
157 one site to the other one, depending on the sampling frequency (typically every third or sixth day). Filter collection
158 was conducted within the PM₁₀ or the PM_{2.5} aerosol size fraction, depending on the investigated site (Table SI-1).
159 Moreover, co-located and simultaneous PM₁₀ and PM_{2.5} samplings were conducted at OPE-ANDRA and Revin,
160 allowing to investigate the distribution of primary sugar compounds between the fine and the coarse aerosol size
161 fractions at these two sites.

162 Ambient aerosols were collected onto quartz fiber filters (Tissu-quartz PALL QAT-UP 2500 150 mm diameter),
163 preheated at 500 °C for 4 h minimum before use. After collection, all filter samples were wrapped in aluminum
164 foils, sealed in zipper plastic bags and stored at <4°C until further chemical analysis. Field blank filters were also
165 collected, at least once a month, using the same handling procedure ~~than~~ as for PM samples. More detailed
166 information on the sampling periods, air sampler, number of filters and nature of PM samples are provided in
167 Table SI-1 and Fig. SI-1.

168 2.2 Chemical analyses

169 PM samples were analyzed for various chemical species using sub-sampled fractions of the collection filters. In
170 the frame of the present study, the carbonaceous matter (organic carbon (OC) and elementary carbon (EC)) was
171 analyzed using a thermo-optical method on a Sunset Lab analyzer (Birch and Cary, 1996) as described by Aymoz
172 et al. (2007), using the EUSAAR2 temperature program (Cavalli et al., 2010), except for the five sites of Northern
173 France where the NIOSH870 protocol was employed (Birch and Cary, 1996). ~~Particulate organic matter (OM)~~
174 contents were then estimated by multiplying the organic carbon mass concentrations by a fixed factor, with $OM =$
175 $1.8 \times OC$. This OM-to-OC ratio value of 1.8 was chosen based on previous studies performed in France (Favez et
176 al., 2010; Petit et al., 2015 and reference therein) and around the world. (e.g., Aiken et al., 2008; Li et al., 2018;
177 Ruthenburg et al., 2014; Vlachou et al., 2018), with a typical range of 1.2-2.4 values.

178 For the analysis of anhydrosugars, sugar alcohols, and primary saccharides, filter punches (typically of about 10
179 cm²) were first extracted into ultrapure water, then filtered using a 0.22 μm Acrodisc filter. Depending on the site,
180 analyses were conducted either by IGE (Institut des Géosciences de l'Environnement) or by LSCE (Laboratoire
181 des Sciences du Climat et de l'Environnement) (Table SI-1). At IGE, extraction was performed during 20 min in
182 a vortex shaker and analyses were achieved using ~~an~~ **High-Performance Liquid Chromatography (HPLC)**
183 with Pulsed Amperometric Detection. A first set of equipment was used until March 2016, consisting of a Dionex
184 DX500 equipped with three columns Metrosep (Carb 1-Guard + A Supp 15-150 + Carb 1-150), the analytical run
185 being isocratic with 70 mM sodium hydroxide eluent, followed by a gradient cleaning step with a 120 mM NaOH
186 eluent. This analytical technique enables to detect anhydrous saccharides (levoglucosan, mannosan, galactosan),
187 polyols (arabitol, sorbitol, mannitol), and glucose (Waked et al., 2014). A second set of equipment was used after
188 this date, with a Thermo-Fisher ICS 5000+ HPLC equipped with 4 mm diameter Metrosep Carb 2 × 150 mm
189 column and 50 mm pre-column. The analytical run is isocratic with 15 % of an eluent of sodium hydroxide
190 (200 mM) and sodium acetate (4 mM) and 85 % water, at 1 mL min⁻¹. This method allows for additional separation
191 and quantification of erythritol, xylitol, and threulose. At LSCE, extraction was performed during 45 min by
192 sonication and analyses were achieved using an **ion chromatography (IC) instrument** (DX600, Dionex) with
193 Pulsed Amperometric Detection (ICS3000, Dionex). A CarboPAC MA1 column ~~s~~ has been used (4 × 250 mm,
194 Dionex), the analytical run being isocratic with 480 mM sodium hydroxide eluent. This analytical technique
195 enables to detect anhydrous saccharides (levoglucosan, mannosan, galactosan), polyols (arabitol, mannitol), and
196 glucose.

197 Field blank filters were handled as real samples for quality assurance. The present data were corrected with field
198 blanks. The reproducibility of the analysis of primary sugar species (polyols, glucose), estimated from the analysis
199 of extracts of 10 punches from the same filters ~~is~~ generally in the range of 10-15 %.

200 Additional chemical analyses were conducted for most of the sites, allowing to quantify up to 130 different
201 chemical species (Calas et al., 2018). 30- 35 chemical species were then selected in order to achieve PMF analyses
202 as discussed hereafter.

203 2.3 Statistical analysis

204 Species concentration measurements were first analyzed for normality using Shapiro-Wilk's method with the
205 statistical program *R studio interface* (version 3.4.1). Since data were generally not distributed normally, we used
206 non-parametric statistical methods. The strength of the relationship between species concentrations was

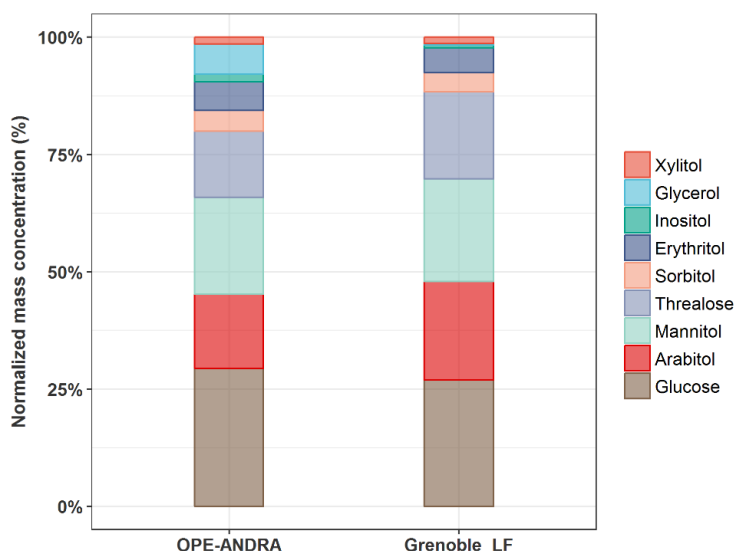
207 investigated using the non-parametric Spearman rank correlation method. Multiple mean comparison analyses
208 were performed with the Kruskal-Wallis test method. Statistical significance was set at $p < 0.05$.
209 Positive Matrix Factorization for the source apportionment of the PM was previously performed at several sites of
210 this study, as part of the [SOURCES](#) (Favez et al., 2016; Salameh et al., in prep.) and [DECOMBIO](#) (Chevrier, 2017)
211 projects. We used the US EPA PMF 5.0 software (US EPA, 2015), following the general recommendation
212 guidelines of the European Joint Research Centre (JRC) (Belis et al., 2014). Briefly, the SOURCES program aimed
213 at performing source apportionment at 15 sites using a harmonized methodology, i.e., using the same chemical
214 species, uncertainties, constraints, and criteria for factor identification. The PMF conducted within SOURCES
215 project uses about 30 different species (Table SI-6), including carbonaceous fraction (OC, EC), ions (Cl^- , NO_3^- ,
216 SO_4^{2-} , NH_4^+ , K^+ , Mg^{2+} , Ca^{2+}), organic markers (polyols i.e., sum of arabitol, mannitol and sorbitol; levoglucosan;
217 mannosan) and metals (Al, As, Ba, Cd, Co, Cs, Cu, Fe, La, Mn, Mo, Ni, Pb, Rb, Sb, Se, Sn, Sr, Ti, V, Zn). The
218 PMF conducted within the DECOMBIO project, for the sites of Marnaz, Chamonix, and Passy, used aethalometer
219 (AE 33) measurements instead of EC (Chevrier, 2016). This complementary measure gives the total black carbon
220 (BC), thus enabling the deconvolution of BC concentrations into its two main constituents: wood-burning BC
221 (BC_{wb}) and fossil-fuel BC (BC_{ff}) (Sandradewi et al., 2008). For graphical simplicity, BC_{wb} and BC_{ff} were summed
222 up and labeled as EC in the following Figures. PMF modelling was performed separately for each site. Statistical
223 significance was validated with bootstrap higher than 80 % for each factor. Detailed methodology and results about
224 these studies are given in their respective papers (Chevrier, 2017; Favez et al., 2016; Salameh et al., in prep.). It
225 should be noted that glucose was not included in the final solution for any of these PMF, since it generally produced
226 statistical instability of the solutions (this point is further discussed in Sect. 3.2).
227 The PMF analysis took advantage of the ME-2 algorithm to add constraints to different chemical profiles (see
228 Tables SI-3 and SI-4 for details). Mainly soft constraints were applied in order to add some prior knowledge about
229 the emission sources and “clean” the different profiles without forcing the model toward an explicit solution. In
230 particular, the polyol concentrations were “pulled up maximally”, while levoglucosan and mannosan were set to
231 zero, and EC was “pulled down maximally” ~~in this factor~~ in the PBOA factor. This was achieved to avoid mixing
232 with the biomass burning factor as well as possible influences of unrealistic high contributions of EC to PBOA.
233 Other constraints were added parsimoniously to other factors, targeting specific proxies of sources (Table SI-4).
234 As for the general results of this large PMF study, we identified some well-known sources for almost all the sites
235 (biomass-burning, road traffic, secondary inorganics, dust and sea salt). Two other less-common factors were
236 identified for all sites: secondary biogenic aerosols (probably from marine origin), traced mainly by the presence
237 of MSA, and PBOA, traced by the presence of more than 90% of the polyols total mass in the factor. Table SI-5
238 and Fig. SI-4 present more detailed description of the chemical tracers in each factor, together with their yearly
239 average contribution for each site, respectively. Hereafter, only the PBOA chemical profile will be extensively
240 investigated. The uncertainties of this PBOA factor are discussed below and its stability is presented in Fig. SI-5.
241 Bootstrap analysis based on 100 resampling runs evidenced the very high stability of this PBOA factor since the
242 PBOA initial constrained factor was mapped to PBOA bootstrap factor (BF) more than 99% of the time.

243 **3. Results and discussion**

244 **3.1 Relative distribution between sugar alcohols and glucose**

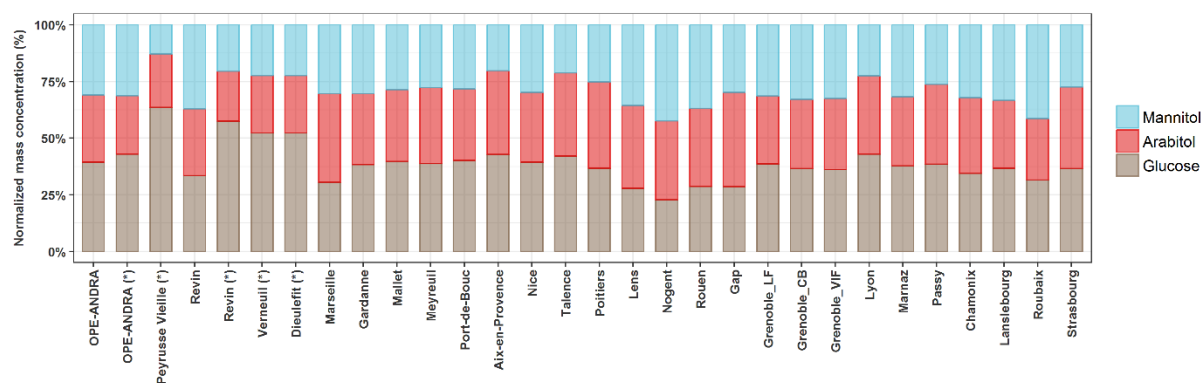
245 Figure 2 presents an overview of the relative mass concentration distributions of individual chemical species
246 quantified at two sites with very different characteristics, an urban site in Grenoble and the rural site of OPE-
247 ANDRA. Data are presented for the warmer season (e.g., during summer and fall), when concentrations were at
248 their maximum (see Sect. 3.4). Glucose is the most abundant species measured (average $37.6 \pm 26.4 \text{ ng m}^{-3}$),
249 accounting on average for 25 % of primary sugar compounds (SC) total mass at both sites. Mannitol
250 ($37.3 \pm 24.6 \text{ ng m}^{-3}$) and arabitol ($32.0 \pm 22.2 \text{ ng m}^{-3}$), are the second and third most abundant species, accounting
251 respectively for 25 and 23 % of SC mass. Threalose is relatively abundant in samples from these two sites
252 ($20.1 \pm 16.2 \text{ ng m}^{-3}$), accounting for 14 % of SC mass, but in general its concentration is frequently below the limit
253 of quantification for samples from other sites in France. The other identified polyols (i.e., erythritol, inositol,
254 glycerol, sorbitol, and xylitol) present lower concentration levels ($4.9 \pm 2.1 \text{ ng m}^{-3}$), corresponding altogether to
255 13 % of SC total mass.

256 Such ambient mass concentration distribution patterns are similar (but with variable intensities) to those previously
257 reported for aerosol samples collected at various locations around the world. For example, Verma et al. (2018)
258 found that glucose, and arabitol together with mannitol, contributed to 16.7 and 48.1 %, respectively, of total
259 primary sugar compounds in aerosols from Chichijima Island. Similarly, Yttri et al. (2007b) showed that glucose
260 and the pair arabitol-mannitol were the main contributors of total primary monosaccharides and sugar alcohols in
261 aerosols collected from four various background sites in Norway. In addition, Carvalho et al. (2003) reported that
262 arabitol, mannitol and glucose are the most dominant primary sugar compounds in aerosols from rural background
263 and boreal forest sites in Germany and Finland, respectively.



264 **Figure 2: Overview of relative mass distributions of individual primary sugar alcohols and saccharide compounds**
265 **quantified in PM₁₀ samples at two sites over summer and autumn periods (June to November) corresponding to**
266 **maximal atmospheric concentrations of sugar alcohols/saccharide compounds.**

268



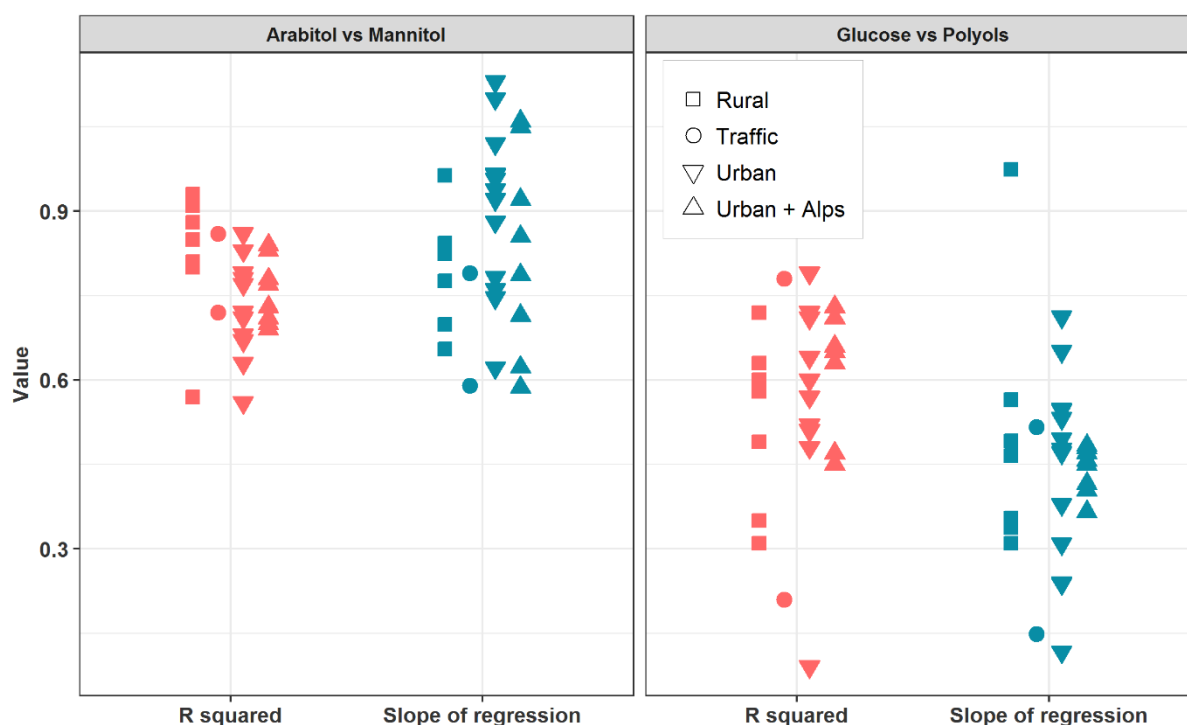
269
 270 **Figure 3: Average mass concentration distributions of arabitol, mannitol, and glucose. Symbol (*) indicates PM_{2.5}**
 271 **aerosol samples. Only data for warmer season (June to November), corresponding to maximal atmospheric**
 272 **concentration of polyols and glucose are shown.**

273 Although various primary sugar alcohols and saccharides have been detected and quantified for most of the
 274 investigated sites, the following study focuses only on the three major and ubiquitous species, namely arabitol,
 275 mannitol and glucose. Figure 3 presents their average relative contributions at all sites, for the warmer period,
 276 displaying very similar features at a first glance. However, discrepancies could be observed from site to site, as
 277 discussed in following sections.

278 3.2 Relationships between selected primary sugar compounds

279 Figure 4 summarizes linear correlations obtained between arabitol and mannitol concentrations at each site during
 280 the warmer period. Medium to very high ~~correlation~~ coefficients of determination could be observed
 281 ($0.58 \leq R^2 \leq 0.93$; $30 \leq n \leq 143$ or $45 \leq n \leq 341$ for PM_{2.5} and PM₁₀ series, respectively), with slopes in a rather
 282 narrow range (between 0.59 and 1.10), and quite low intercepts (always below 9 ng m⁻³). Such covariations
 283 indicate that both species are most probably co-emitted, by one or several type(s) of sources, at each site during
 284 the summer-autumn period. These observations are in agreement with previous studies also showing strong
 285 covariations between arabitol and mannitol (Kang et al., 2018; Verma et al., 2018; Zhu et al., 2015). Therefore, it
 286 seemed reasonable to consider both species together, so that their concentrations are summed up and labelled as
 287 "polyols" in the following sections.

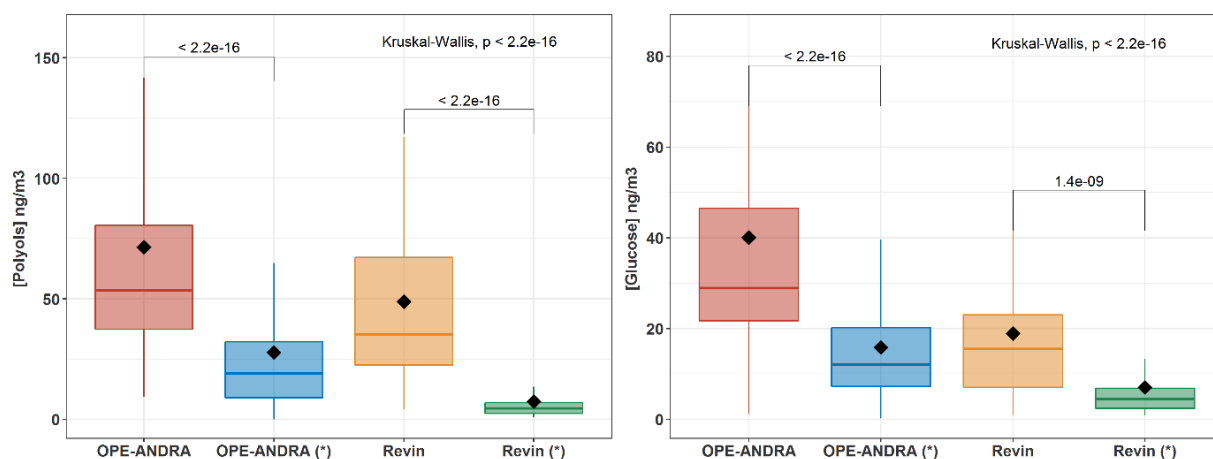
288 Conversely, linear correlations between glucose and polyols concentrations are generally weaker
 289 ($0.10 < R^2 \leq 0.78$), with slopes varying over a much larger range (between 0.12 and 0.94), and variable intercepts
 290 (between -5.6 and 16.4 ng m⁻³). This suggests that glucose concentrations might follow a different pattern
 291 compared to that of polyols, either due to different emission sources, or different chemical stability in the
 292 atmosphere. It is therefore reasonable to keep glucose as a separate chemical species in the following discussion.
 293 It should be emphasized that the variability in the slope of the regressions between the chemical concentrations
 294 isare most probably related to the emissions and atmospheric processing. Particularly in the case of mannitol and
 295 arabitol, they may be influenced by biogenic or biotope characteristics. Nevertheless, no evident relationship
 296 between the slope values and the typology or the geographical location of the sites could be observed (Figure 4).



297
 298 **Figure 4: Linear regression analysis between selected primary sugar compounds mass concentrations (i.e., arabitol,**
 299 **mannitol and glucose) during summer and autumn seasons (June to November), for all the sites considered in this study.**

300 3.3 Relative distributions between PM₁₀ and PM_{2.5}

301 Figure 5 shows the average PM₁₀ and PM_{2.5} concentrations of polyols and glucose at OPE-ANDRA and Revin
 302 during the summer and autumn seasons. The polyols mass concentrations ranged from 7.5±10.9 to 27.8±33.3 ng m⁻³
 303 in PM_{2.5}, and from 48.9±38.2 to 73.5±61.8 ng m⁻³ in PM₁₀, in Revin and OPE-ANDRA sites, respectively. PM₁₀-
 304 to-PM_{2.5} ratios were then on average of about 3 to 5. Similar size distribution patterns, with variable intensity, were
 305 observed for glucose (Fig. 5). These results indicate that polyols and glucose are mainly associated with the coarse
 306 PM fraction. This observation is in good agreement with several previous investigations where polyols (especially
 307 arabitol and mannitol), together with glucose, were prevalent in the coarse fraction (Fu et al., 2012; Fuzzi et al.,
 308 2007; Pio et al., 2008; Yttri et al., 2007b). However, Carvalho et al. (2003) reported different size distributions for
 309 polyols and glucose, with variable fine or coarse mode maxima depending upon sampling location. For instance,
 310 maximum atmospheric concentrations of mannitol were associated to fine and coarse aerosols from boreal forest
 311 (Finland) and rural background sites (Germany), respectively. The authors hypothesized that these observations
 312 are due to different assemblages of dominant fungal biota (with variable aerodynamic characteristics) at different
 313 sites. Some other previous studies showed aerodynamic diameters typically ranging from 2 to 10 μm, even though
 314 a few airborne bacterial/fungal spores could exceed that size (Bauer et al., 2008a; Elbert et al., 2007; Huffman et
 315 al., 2012; Zhang et al., 2015).



316
 317 **Figure 5: Box plots of mass concentrations of polyols (left) and glucose (right) in PM₁₀ and PM_{2.5} (with symbol (*)**
 318 **samples). Black markers inside each boxplot indicate the mean concentration value, while the top, middle and bottom**
 319 **lines of the box represent the 75th, median and 25th percentile, respectively. The whiskers at the top and bottom of the**
 320 **box extend from the 95th to the 5th percentile. Number of samples were N = 123 for OPE-ANDRA and N = 87 for Revin,**
 321 **respectively. Statistical differences between average mass concentrations were analyzed with the Kruskal-Wallis**
 322 **methods (p < 0.05).**

323 Hence, although if the precise mechanisms of atmospheric emission of particulate polyols and glucose are not fully
 324 resolved, our observations are in good agreement with ambient mass concentrations of polyols and glucose being
 325 likely associated with biological particles, as already suggested elsewhere (Fu et al., 2012; Verma et al., 2018;
 326 Zhang et al., 2015). These species could enter into the atmosphere through either natural or anthropogenic
 327 resuspension of surface soils and associated bacterial/fungal spores (containing polyols and primary sugar
 328 compounds), or via a direct input resulting from microbial activities (e.g., sporulation). Another hypothesis would
 329 be the abrasion of leaves and the subsequent release of microbial organisms and plant debris (Fu et al., 2012;
 330 Medeiros et al., 2006; Simoneit et al., 2004**b**).

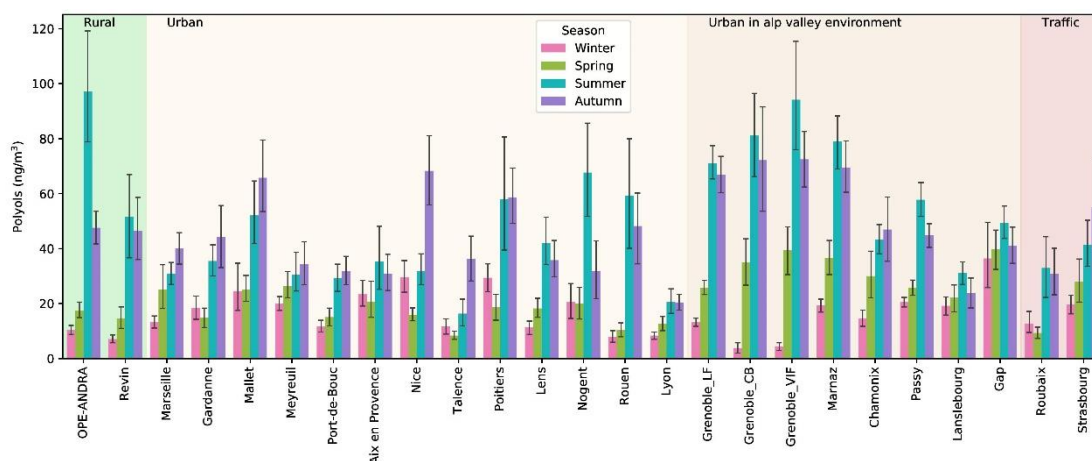
331 3.4 Spatial and seasonal distribution of atmospheric concentrations

332 3.4.1 Spatial and seasonal patterns of polyol concentrations

333 As illustrated in Fig. 6, significant concentrations of polyols were measured at each investigated site, evidencing
 334 the ubiquity of these organic compounds. The annual average concentration levels of polyols measured in PM₁₀
 335 aerosols at all sites (33.2±33.5 ng m⁻³; see Table SI-2) are within the range previously reported for urban and rural
 336 sites across Europe (Burshtein et al., 2011; Di Filippo et al., 2013; Pietrogrande et al., 2014; Yttri et al., 2007**b**,
 337 2011). Additionally, polyols mass concentrations clearly exhibit seasonal trends, with variable intensity according
 338 to the sampling sites. On a seasonal average, polyols are more abundant in summer (46.8±43.6 ng m⁻³) and autumn
 339 (43.0±36.7 ng m⁻³), followed by spring (19.0±13.6 ng m⁻³) and winter (16.2±11.5 ng m⁻³). The average
 340 concentrations of polyols are ~~then~~ at least 2 to 3 times higher during summer or autumn months than during the
 341 cold months, with a ratio that can be as high as 8 to 10.

342 Previous studies also reported similar seasonal variation pattern for urban and rural aerosol samples collected at
 343 various locations. For example, Pashynska et al. (2002) measured higher atmospheric polyol (arabitol, mannitol)
 344 contents during late summer and autumn, in Belgium. Several other studies reported higher concentrations of
 345 polyols in summer than spring and winter time, in aerosols collected from Texas, USA and Jeju Island, respectively

346 (Fu et al., 2012; Jia et al., 2010a, 2010b). More recently, Liang et al. (2016) and Verma et al. (2018) also reported
 347 similar seasonal distributions for aerosols sampled [in](#) Beijing, China and north-western Pacific, respectively.
 348 The higher atmospheric polyols concentrations observed are likely due to the increased contribution from
 349 metabolically active microbial derived sources (fungi, bacteria, green algal lichens) as a result of external stressors
 350 such as heat, drought and relative moisture. Indeed, fungal and prokaryotic cells activities, including their growth
 351 and sporulation, are promoted by high temperature and humid conditions occurring in summer and autumn (China
 352 et al., 2016; Elbert et al., 2007**b**; Jones and Harrison, 2004; Rathnayake et al., 2017).
 353 As also evidenced from Fig. 6, atmospheric polyols concentrations do not present any significant seasonal
 354 differences related to the site typology (rural, traffic, urban sites with/without Alpine influences), or latitude. There
 355 is some tendency toward higher concentrations in summer in Alpine environments, but some other sites (like the
 356 rural site of OPE-ANDRA, in the North-East of France) can reach the same levels of concentrations. We tested
 357 several types of hierarchical classifications, including variables like monthly or seasonal mean polyols
 358 concentrations, the ratio arabitol-to-mannitol, or linear regression parameters (slope, R square) but none of them
 359 led to a simple clustering of the sites that would explain the variability of the concentrations.

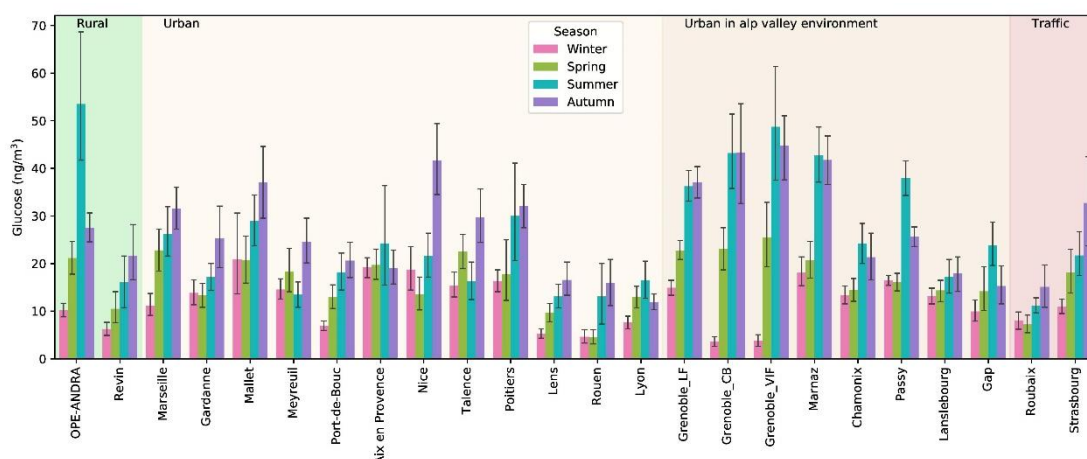


360
 361 **Figure 6: Spatial and seasonal distributions of atmospheric polyol average concentrations (ng m^{-3}) for various types of**
 362 **sites in France. Error bars correspond to standard deviations calculated with seasonal concentrations. Year of PM**
 363 **sampling campaigns are not concurrent at all sites (see Fig. SI-1).** The seasons were defined as follows: Winter =
 364 **December to February, Spring = March to May, Summer = June to August, and Autumn = September to November.**

365 3.4.2 Spatial and seasonal patterns of glucose concentrations

366 The annual average concentrations of glucose measured in PM_{10} aerosols at all sites ($20.4 \pm 15.6 \text{ ng m}^{-3}$; see Table
 367 SI-2) are comparable to those previously reported for various sites across Europe (Alves et al., 2006; Theodosi et
 368 al., 2018; Yttri et al., 2007, 2011). Likewise polyols, the atmospheric concentrations of glucose also display
 369 seasonal and site-to-site variations (Fig. 7). The ambient seasonal mean concentrations (with standard deviations)
 370 of glucose are maximum in summer ($25.0 \pm 24.2 \text{ ng m}^{-3}$) and autumn ($24.6 \pm 19.8 \text{ ng m}^{-3}$), followed by spring
 371 ($15.8 \pm 12.4 \text{ ng m}^{-3}$) and winter ($12.6 \pm 10.2 \text{ ng m}^{-3}$). The summer / winter ratio for glucose seems generally lower
 372 than that of polyols, with higher ratios in the Alpine areas than in other parts of France. However, as for polyols,
 373 it remains difficult to classify the sites according to any criteria linked to site typology or latitude.
 374 The seasonal trend of glucose concentrations in the present work is similar to that recently observed for aerosols
 375 (PM_{10} or total suspended particles) collected at various environmental background (suburban, urban and coastal)

376 sites around the world (Liang et al., 2016; Srithawirat and Brimblecombe, 2015; Verma et al., 2018). On average,
 377 a wide range of daily glucose concentrations (expressed as min-max, mean) in PM₁₀ (0.1-297.2 ng m⁻³,
 378 20.4±15.6 ng m⁻³) were observed in the present study. These values are comparable to those in PM₁₀ (8.4-93.0,
 379 47.0 ng m⁻³) reported from an urban site in NorwayBelgium (Yttri et al., 2007b). More recently, Liang et al. (2016)
 380 also reported similar concentrations in PM₁₀ (3.1-343.6, 46.2±27.5 ng m⁻³) from Beijing (China).
 381 The sources and formation processes of glucose in the atmosphere are not currently well known and are rarely
 382 discussed. Glucose is an important carbon source for soil metabolic active microbiota, and it is commonly present
 383 in vascular plants. Additionally, cellulose (a linear polymer made of glucose subunits linked by β-1,4 bonds) is
 384 one of the most important form of organic compounds in terrestrial ecosystems and a major plant structural
 385 polymer (Boex-Fontvieille et al., 2014). It can also be quite abundant in the atmosphere (Puxbaum and Tenze-
 386 Kunit, 2003). Hence, it is hypothesized that ambient glucose could be formed through active microbial (i.e.,
 387 bacteria, fungi, etc.) enzymatic hydrolysis of cellulose in plant debris. Consistent with these observations, glucose
 388 could be released into the atmosphere from both vascular plant materials (e.g., leaves, fruits, pollens, etc.) growing
 389 in spring and decomposing in autumn/summer, and soil microbiota, as already suggested elsewhere (Di Filippo et
 390 al., 2013; Jia et al., 2010a; Medeiros et al., 2006; Verma et al., 2018; Zhu et al., 2015).



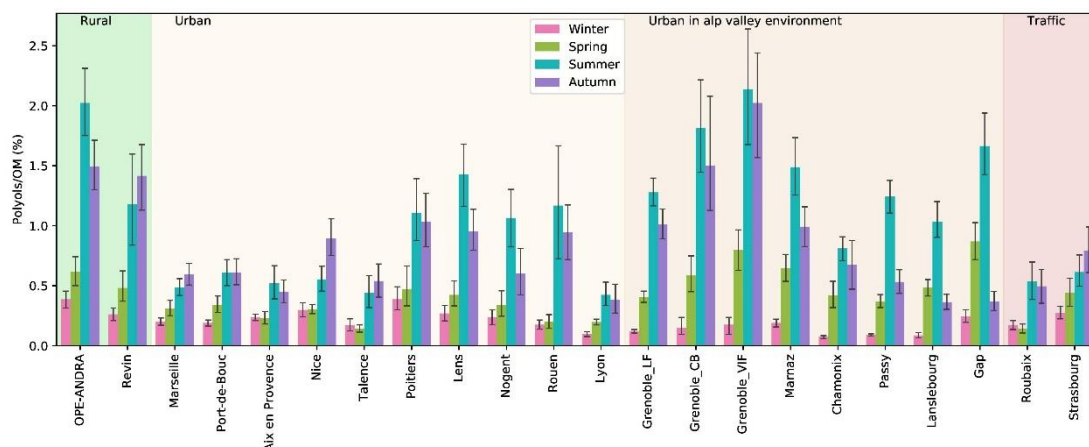
391
 392 **Figure 7: Spatial and seasonal distributions of atmospheric glucose levels (ng m⁻³) for various types of sites in France**
 393 **(except the site of Nogent, which presented too many missing values. Error bars correspond to standard deviations**
 394 **calculated with seasonal concentrations.**

395 3.4.3 Relative contributions to aerosol organic matter concentrations

396 The average contribution of polyols to the OM content~~organic matter content (OM)~~ of PM clearly displayed a
 397 seasonal behavior, as shown in Fig. 8. Here again, contributions are 2 to 10 times higher during summer and
 398 autumn compared to winter and spring, consistent with the assumption of higher emissions during these periods.
 399 The seasonal mean contribution of polyols to OM fluctuates from site to site, and accounts for 0.1 to 2.1 % of
 400 overall OM for these French sites (Fig. 8). Similarly, the seasonal mean concentrations of polyols together with
 401 glucose represent between 0.2 to 3.1 % of total OM at these sites (Fig. SI-2). However, on a daily basis (Samake
 402 et al., in prep.), atmospheric polyols mass concentrations can represent up to 6.3 % of total OM in PM₁₀, indicating
 403 that polyols can be amongst the major molecular species identified in aerosol organic matter (Fig. SI-3). Again,
 404 we could not find any simple way to group the sites according to their characteristics (typology or latitude, or

405 climatic region), in order to better understand the drivers behind the variability of this mass fraction. Further studies
 406 are currently conducted using multi criterion examinations.

407 The seasonal polyols-to-OM distribution patterns in this study are comparable to those found for different urban
 408 or rural sites in Europe (around 0.2 to 2.5 % of OM) (Pashynska et al., 2002; Yttri et al., 2007b). Zhu et al. (2015)
 409 also reported similar seasonal polyols-to-OM contribution trend for aerosols sampled at Cape Hedo (coastal site,
 410 Japan).



411
 412 **Figure 8: Spatial and seasonal distributions of mean contributions (in %) of polyols to the organic matter content of**
 413 **PM for various types of sites in France. Daily time-series of organic carbon (OC) were not available for the following**
 414 **sites: Gardanne, Mallet, and Meyreuil. Error bars correspond to standard deviations calculated with seasonal**
 415 **concentrations.**

416 3.5 Primary biogenic factor in PMF studies

417 The sum of polyols (arabitol + mannitol) represents only a small fraction of the total OM. However, as proxies of
 418 PBOA, they are most probably emitted with other chemical species. Emission from biological particles is a
 419 complex topic since it may include a wide variety of compounds, both organic and inorganic (Elbert et al., 2007;
 420 Zhang et al., 2015). Moreover, it is not clear if polyols are mainly emitted directly in the atmosphere or are linked
 421 to other materials, for example with soil dust during resuspension processes. To investigate the relationship
 422 between polyols and other molecular tracers of emission sources species, it would have been possible to perform
 423 simple correlation analysis with individual chemical species. This approach has the disadvantage of being a one-
 424 to-one relation and thus highly sensitive to the dynamics of all PM emission sources, not only the one we are
 425 interested in. Alternatively, another way is to use a PMF approach, also based on correlations but including much
 426 more information on the temporal variations of the different sources influencing the PM chemistry at a given
 427 receptor site.

428 As mentioned in Sect. 2.3, the PMF results used in this study include sites of different typologies (rural, traffic,
 429 urban sites in Alpine valley environments, etc.) for 16 different locations spread over France and part of the current
 430 dataset. At each site, the PMF studies allowed to identify a PBOA factor, characterized by the presence of more
 431 than 90 % of the total polyols content (sum of arabitol, mannitol and sorbitol), as presented in table SI-5 and
 432 Fig. SI-6. Moreover, the sensitivity of this factor to random noise in the data was investigated thanks to randomly
 433 re-sampling the input matrix of observation. In PMF analysis, this is done via the bootstrap method (Paatero et al.,
 434 2014) in the constrained run. The PBOA factor was always mapped to itself for 13 of the sites and quasi-always
 435 (97%) for the last three ones. It means that the PBOA factor does have a very high statistical stability since it never

436 swaps with another factor (see Fig. SI-5). Hence, the chemical composition of this factor may be informative to
437 investigate the PBOA source components (Table SI-6), and to evaluate the importance of PBOA emissions in
438 terms of OM mass apportionment.

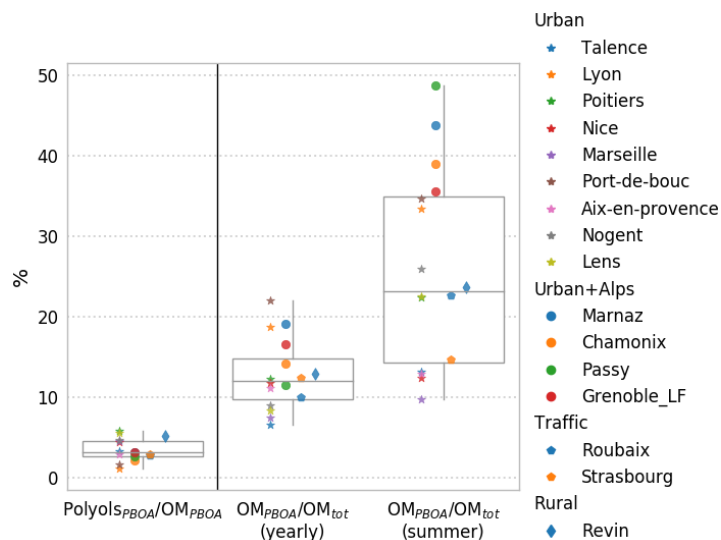
439 3.5.1. Contributions of PBOA to OM and polyols to PBOA

440 Altogether, the results from ~~the~~these 16 sites highlight the importance of the PBOA source contribution to total
441 OM. As shown in Fig. 9, the OM apportioned by the PBOA factor represents a significant fraction of the total OM
442 mass on a yearly average (range 6–28 %; average 13 ± 6 %). When considering only the summer period (June-July-
443 August), this contribution is even larger and can exceed 40 % of the total OM at sites in the Alpine area (Marnaz,
444 Passy, Chamonix, Grenoble_LF) which are partially protected from large regional influences due to the local
445 topography. This result may be nuanced, in particular during summer, since some extent of mixing between PBOA
446 and Biogenic Secondary Organic Aerosols (BSOA) cannot be entirely excluded. However, several evidences tend
447 in favor of a non-significant mixing between BSOA and PBOA. First, the ratio of polyols-to- OC_{PBOA} shows a low
448 variability from site to site, while it is unlikely that such a secondary process led to the same amount of OC for all
449 sites since they present different meteorology, sunshine duration, etc. Second, the bootstrap analysis ~~do~~es not show
450 any “swap” between factors for the PBOA profile for all sites, indicative of a well-defined factor (see Fig. SI-5).
451 Finally, the OC_{PBOA} -to-polyols ratio in this work (about 16) is in the range of ratio expected for fungal spores (12
452 -27, when arabitol and mannitol are considered together) (Bauer et al., 2008a; Yttri et al., 2011).

453 Interestingly, some previous work using the same samples from the sites in the Arve valley (Passy, Chamonix)
454 showed that about 90 % of the OM is from modern origin (using ^{14}C measurements) during summer, with no
455 apparent correlation between this modern carbon and polyols concentrations (Bonvalot et al., 2016). Hence, despite
456 being an important contributing source, PBOA is not the major biogenic source in this type of environment.

457 Interestingly, opposite to the case of the Alpine valleys where this proportion is the highest, the ratios OM_{PBOA} -to-
458 OM_{total} are amongst the lowest for coastal environments (Talence, Marseille, Nice), a possible indication that the
459 marine environment is not a large emitter for these species. Recently, much lower concentrations of polyols in
460 aerosols from marine environments than those in terrestrially influenced sites were also reported off the coast of
461 Japan, also suggesting a higher contribution from terrestrial sources (Kang et al., 2018).

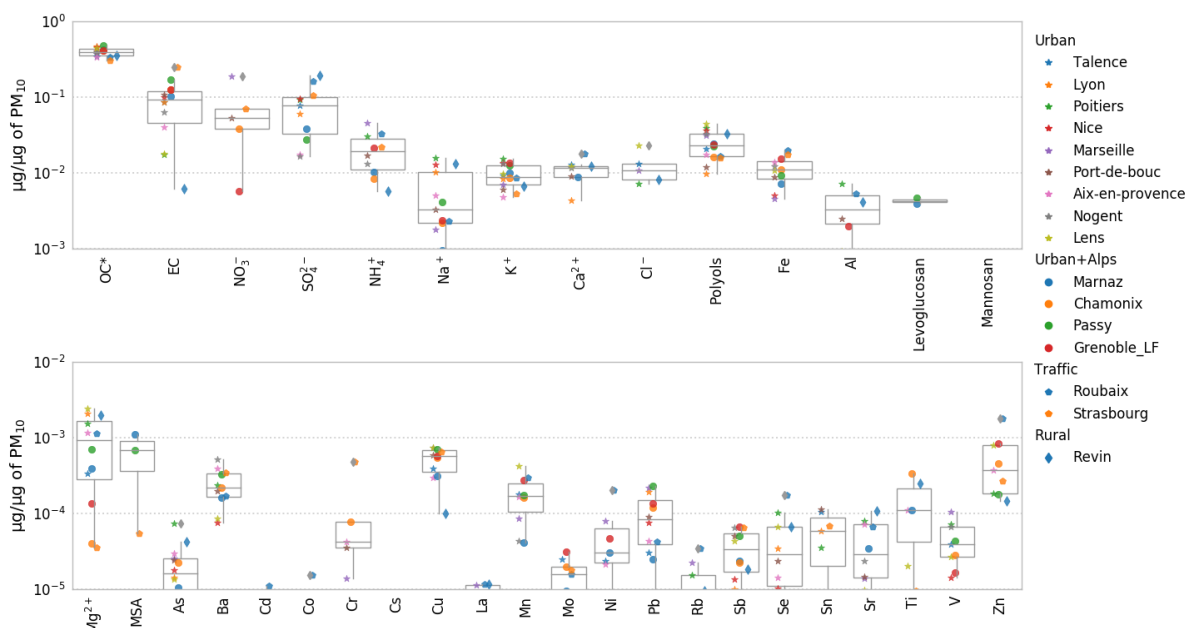
462 As illustrated in Fig. 9, polyols represent only a small fraction of the OM apportioned in the PBOA factor (1.2 %–
463 6.0 %; average 3.0 ± 1.5 %) for the 16 studied sites. This variability is indeed rather small, considering the wide
464 range of sites and the diversity of other potential sources (on average 8 to 10 PMF factors were obtained for the
465 different sites). Indeed, this narrow range of the polyols fraction to the OM_{PBOA} highlights the stability of the
466 chemical profile of this source over a large regional scale. It indicates also that, if polyols are good proxies of the
467 PBOA sources, a large amount of other organic species are co-emitted, that still remain unknown.



468
 469 **Figure 9: Mass contribution of polyols to OM in the PBOA factor, and relative contributions of the OM_{PBOA} factor to**
 470 **the total OM in PM for the 16 studied sites where PMF model was run, over the year and summertime only. Stars and**
 471 **circle refer to urban sites without/and with Alpine valley influence, respectively. Pentagon corresponds to traffic sites**
 472 **and diamond to rural sites.**

473 **3.5.2 PBOA profile constituents and emission process**

474 Figure 10 shows the contribution (in μg of species per μg of PM in the PBOA factor profile) of each chemical
 475 species included in the averaged PBOA factor from the 16 PMF studies. The principal contributors are OC and
 476 EC, and significant fractions of crustal material also appear (Na^+ , K^+ , Ca^{2+} , Al, Ba, Cu, Fe, Mn, Ti, Zn) as well as
 477 secondary elements such as nitrate and sulfate. However, EC appears to be highly variable both within and between
 478 sites under consideration. The reader may refer to figure SI-7 for an estimation of the EC mass uncertainties in the
 479 different sites. On average, the PBOA factor does not comprise a large fraction of metals and trace elements, most
 480 of them being below $1 \text{ pg } \mu\text{g}^{-1}$. Here again, the low variability of the PBOA chemical profile encountered across
 481 a large array of sites is remarkable.



482
 483 **Figure 10: PMF chemical profile of the PBOA factor in the DECOMBIO and SOURCES programs expressed as a**
 484 **fraction of the PM mass. Values lower than a few $\text{pg } \mu\text{g}^{-1}$ are not displayed on purpose. For each boxplot, the top, middle**

485 and bottom lines represent the 75th, median and 25th percentile, respectively. The whiskers at the top and bottom of the
486 box extend from the maximum to the minimum. OC* corresponds to the bulk organic carbon fraction minus the carbon
487 individual molecular weight of in the characterized organic species.

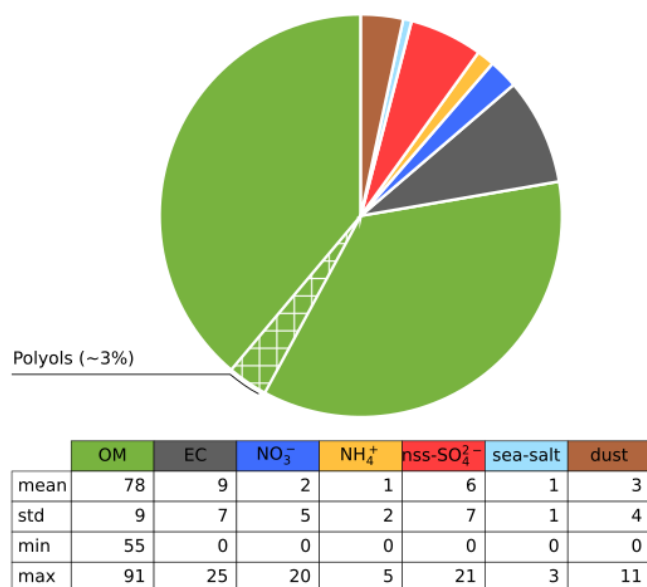
488 The contribution from some crustal material could agree with the coarse mode distributions of polyols (Sect. 3.3)
489 and could be indicative of an emission process with the entrainment of spores with soil dust resuspension. To
490 investigate the importance of mineral dust in the PBOA factor, we clustered the chemical components of PM from
491 PBOA into 7 classes: OM ($= 1.8 \times \text{OC}$), EC, NO_3^- , NH_4^+ , non-sea-salt sulfate (nss- SO_4), sea-salt, and dust. nss-
492 SO_4^{2-} is calculated from the measured SO_4^{2-} minus the sea-salt fraction of SO_4^{2-} ($\text{nssSO}_4^{2-} = \text{SO}_4^{2-} - \text{ssSO}_4^{2-}$
493 where $\text{ssSO}_4^{2-} = 0.252 \times \text{Na}^+$) according to Seinfeld and Pandis (1997). The sea-salt fraction is calculated
494 according to Putaud et al. (2010): $\text{sea} - \text{salt} = \text{Cl}^- + 1.47 \times \text{Na}^+$. Finally the dust fraction is estimated ~~from~~
495 thanks to Putaud et al. (2004b) as: $\text{dust} = (\text{nss} - \text{Ca}^{2+}) \times 5.6$ with nss- Ca^{2+} stands for non-sea salt Ca^{2+} and is
496 computed thanks to $\text{nss} - \text{Ca}^{2+} = \text{Ca}^{2+} - \text{Na}^+ / 26$. We note that the conversion coefficient provided by Putaud
497 et al. (2004b) may be influenced by an extreme value and then gives only a low estimate of dust resuspension.

498 Figure 11 presents the normalized average contributions of these 7 classes to the PBOA mass for the 16 sites with
499 PMF modelling. It clearly reveals ~~see~~ that the PBOA factor is dominated by contributions from OM ($78 \pm 9\%$),
500 followed by EC ($9 \pm 7\%$), and only a minor contribution from the dust class ($3 \pm 4\%$).

501 The large value for the contribution of EC is driven by two high values obtained at the sites of Strasbourg (that
502 reaches 25%) and Chamonix (18%) both influenced by direct and indirect traffic emissions. However, 6 other sites
503 present no EC in PBOA. Moreover, the uncertainties of EC in the PBOA profile of Strasbourg and Chamonix is
504 rather high (between 5 to 30% of PM mass at Strasbourg, see SI-7). On a yearly average, EC apportioned by this
505 factor (0 to 400 ng m^{-3} depending on the site) is close to the rural EC background in France of about 300 ng m^{-3}
506 (Golly et al., 2018).

507 This result on the general chemical profile of the PBOA factor, with a low crustal fraction, tends to infirm the
508 hypothesis of an emission process associating PBOA material with mineral dust resuspension. Indeed, our findings
509 rather suggest that a main part of PBOA (and polyols) ~~is~~ are most likely associated with biological particle direct
510 emissions. It leaves only a minor fraction that could be linked to the mechanical resuspension of PBOA with crustal
511 elements. Some minor fraction of EC in this factor could come from resuspended EC-containing dust particles
512 being accumulated in topsoil as demonstrated in previous works (Forbes et al., 2006; Hammes et al., 2007; Zhan
513 et al., 2016). Hence, the origin of the larger fraction of the contribution of EC remains unknown. Our conclusions
514 are in good agreement with those made by Jia and Fraser (2011), based on the concentrations of these chemicals
515 in different types of samples: i.e., size-fractionated (equivalent to $\text{PM}_{2.5}$ and PM_{10}) soil, plant, fungi, atmospheric
516 $\text{PM}_{2.5}$ and PM_{10} . They found that the ambient concentrations of primary saccharide compounds at the suburban
517 site of Higley (USA) are typically dominated by contributions of biological materials rather than resuspension of
518 soil dust particles and associated microbiota.

519



520

521 **Figure 11: Average contribution (%) of species in the PBOA factor for the sites in SOURCES and DECOMBIO. The**
 522 **hatched area represents the proportion of the OM apportioned by the polyols (see text for reconstruction method).**

523 **4. Conclusion**

524 The contribution of primary biogenic organic aerosols to PM is barely documented in the scientific literature. The
 525 present study aimed at providing a large overview of the spatial and temporal evolution of concentrations and
 526 contributions to aerosol organic matter of dominant primary sugar alcohols and saccharide compounds, for a large
 527 selection of environmental conditions in France. With 28 sites and more than 5,340 samples, it is to our knowledge
 528 the most comprehensive dataset for these compounds. The main results obtained indicate that:

- 529 • among the identified polyols, arabitol together with mannitol are the major species by mass, with lesser
 530 amounts of others polyols (e.g., erythritol, inositol, glycerol, sorbitol, and xylitol). Glucose is the
 531 dominant primary monosaccharide and its relative abundance is comparable to the sum of arabitol and
 532 mannitol;
- 533 • the two main polyols (arabitol and mannitol) together with glucose are mainly present within the coarse
 534 aerosol mode;
- 535 • at nearly all sites, ambient levels of the main polyols and glucose displayed clear seasonal variation cycles,
 536 with a gradual increase from spring and maximum in summer and autumn aerosols, followed by a sudden
 537 decrease in late autumn, and a winter minimum;
- 538 • atmospheric concentrations of the main polyols and glucose fluctuate according to site and season, and
 539 account each for between 0.1 to 2.1 % of OM on a seasonal average basis at these French sites;
- 540 • ambient mass concentrations of arabitol and mannitol are comparable. Meanwhile, they display very good
 541 temporal covariation, with ratios varying between sites. Conversely, linear correlations between the main
 542 polyols and glucose concentrations are much lower, suggesting different atmospheric sources, or
 543 atmospheric processes;

- 544 • arabitol and mannitol are efficient organic markers for PBOA. PMF studies of the yearly series from 16
545 sites give contributions of the primary biogenic emission (traced with the main polyols) to the total OM
546 around $13\pm 6\%$ on a yearly average and $26\pm 12\%$ during summer, thereby showing that PBOA is an
547 important source of total OM in PM_{10} for all sites across France. Furthermore, the average PBOA
548 chemical source profile is made out of a very large fraction of OM ($78\pm 9\%$ of the total PBOA mass on
549 average), suggesting it is mainly related to direct biogenic emissions from biological particles.
550 Noteworthy, the presence of BSOA within the PBOA factor, particularly during summer could not be
551 fully ruled out and further works using additional organic tracers (such as 3-methylbutanecarboxylic acid,
552 3-MBTCA, pinic acid, and/or cellulose) are still needed to solve this issue. Additionally, the low crustal
553 fraction indicates that this factor is weakly linked to soil dust resuspension associated with biological
554 material;
- 555 • however, the PBOA source remains chemically poorly characterized as the main polyols represent only
556 a small fraction of its total OM mass ($3.0\pm 1.5\%$ on average);
- 557 • despite comparable high concentrations in the atmosphere, the sources and processes leading to glucose
558 concentrations and seasonal evolutions are still elusive. Indeed, the different PMF performed with glucose
559 in input variable do not lead to a statistically stable solution;

560 Further investigations of the emission pathways and chemical characterization of the PBOA source associated with
561 polyols are on-going, which may improve our understanding of its dynamic at various geographical scales, for a
562 potential implementation in emission models in the future.

563

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577

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580 of PI for the programs that generated the data for 18 of the 28 sites in this study (OPE-ANDRA, Part'Aera,
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588 doctoral fellow under the supervision of JLJ to gather and prepare most of the datasets used in the present studies.
589 All authors from the ANDRA (#5) and AASQA (#6 to 13) are representatives for each network that conducted the
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591 FC and DS ran the PMF analysis. AS, SW and JLJ processed the data and wrote up the manuscript. All authors
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594

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