

Aerosol hygroscopic growth, contributing factors, and impact on haze events in a severely polluted region in northern China

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1 **Abstract:**

2 This study investigates the impact of the aerosol hygroscopic growth effect on haze
3 events in Xingtai, a heavily polluted city in the central part of the North China Plain,
4 using a large array of instruments measuring aerosol optical, physical, and chemical
5 properties. Key instruments used and measurements made include the Raman lidar for
6 atmospheric water vapor content and aerosol optical profiles, the PC-3016A GrayWolf
7 six-channel handheld particle/mass meter for atmospheric total particulate matter (PM)
8 that have diameters less than 1 μm and 2.5 μm (PM_{10} and $\text{PM}_{2.5}$, respectively), the
9 aerosol chemical speciation monitor (ACSM) for chemical components in PM_{10} , and the
10 hygroscopic tandem differential mobility analyzer (H-TDMA) for aerosol
11 hygroscopicity. The changes in PM_{10} and $\text{PM}_{2.5}$ agreed well with that of the water vapor
12 content due to the aerosol hygroscopic growth effect. Two cases were selected to
13 further analyze the effects of aerosol hygroscopic growth on haze events. The lidar-
14 estimated hygroscopic enhancement factor for the aerosol backscattering coefficient
15 during a relatively clean period (Case I) was lower than that during a pollution event
16 (Case II) with similar relative humidity (RH) levels of 80–91%. The Kasten model was
17 used to fit the aerosol particle hygroscopic growth factor whose parameter b differed
18 considerably between the two cases, i.e., 0.1000 (Case I) versus 0.9346 (Case II). The
19 aerosol acidity value calculated from ACSM data for Case I (1.35) was less than that
20 for Case II (1.50) due to different amounts of inorganics such as NH_4NO_3 , NH_4HSO_4 ,
21 and $(\text{NH}_4)_2\text{SO}_4$. Model results based on H-TDMA data showed that aerosol hygroscopic
22 growth factors in each size category (40, 80, 110, 150, and 200 nm) at different RH

1 levels (80–91%) for Case I were lower than those for Case II. For similar ambient RH
2 levels, the high content of nitrate facilitates the hygroscopic growth of aerosols, which
3 may be a major factor contributing to heavy haze episodes in Xingtai.

4

5 **Key words:** Raman lidar; aerosol hygroscopic growth; water content; haze; remote
6 sensing

1 **1. Introduction**

2 Aerosols, as solid or liquid particles suspended in the air, help regulate Earth's climate
3 mainly by directly scattering or absorbing incoming radiation, or indirectly changing cloud
4 optical and microphysical properties (IPCC, 2013). Many studies suggest that aerosols have a
5 direct impact on human health (Araujo et al., 2008; Anenberg et al., 2010; Liao et al., 2015; Li
6 et al., 2017). For example, exposure to fine airborne particulates is linked to increased
7 respiratory and cardiovascular diseases (Hu et al., 2015). Atmospheric aerosols can also reduce
8 visibility. Poor visibility is not only detrimental to human health but also hazardous to all means
9 of transportation (Zhang et al., 2010; Zhang et al., 2018).

10 Poor visibility is caused by the presence of atmospheric aerosols whose loading depends
11 on both emission and meteorology. The increase in anthropogenic emissions directly affects
12 the formation of haze, such as biomass burning, and factory and vehicle emissions (Watson,
13 2002; Sun et al., 2006; Q. Liu et al., 2016; Qu et al., 2018). During some major events like the
14 2008 Summer Olympic Games, drastic measures were taken to reduce emissions which led to
15 a significant improvement in air quality (Huang et al., 2014; Shi et al., 2016; Y.-Y. Wang et al.,
16 2017). This attests to the major role of emissions in air quality. Surface solar radiation and
17 weather such as wind conditions also affect aerosol pollution (Yang et al., 2015). It has been
18 widely known that aerosols interact with the planetary boundary layer (PBL; Quan et al., 2013;
19 Li et al., 2017; Qu et al., 2018; Su et al., 2018). More aerosols reduce surface solar radiation,
20 resulting in a more stable PBL which enhances the accumulation of pollutants within the PBL.
21 Numerous studies have highlighted that the diurnal evolution of the PBL is crucial to the
22 formation of air pollution episodes (Tie et al., 2015; Amil et al., 2016; Kusumaningtyas and

1 Aldrian, 2016; Li et al., 2017; Qu et al., 2018). Besides feedbacks, the stability of the PBL
2 affects the dispersion of pollutants.

3 Aerosol hygroscopicity also significantly affects visibility due to the swelling of aerosols
4 (Jeong et al., 2007; Wang et al., 2014). A number of studies have shown that aerosol
5 hygroscopic growth can accelerate the formation and evolution of haze pollution in the North
6 China Plain (NCP; e.g., Quan et al., 2011; Liu et al., 2013; Wang et al., 2014; Yang et al., 2015).
7 There are many ways to measure aerosol hygroscopicity. A widely used parameter, the aerosol
8 particle size hygroscopic growth factor (GF), is defined as the ratio of the wet particle diameter
9 ($D_{p,wet}$) at a high relative humidity (RH) to the corresponding dry diameter ($D_{p,dry}$). The GF at
10 a certain particle size can be detected by a hygroscopicity tandem differential mobility analyzer
11 (H-TDMA; e.g., Liu et al., 1978; Swietlicki et al., 2008; Y.-Y. Wang et al., 2017). In general,
12 the H-TDMA system mainly consists of two differential mobility analyzer (DMA) systems and
13 one condensation particle counter (CPC). The DMA is first used to select particles at a specific
14 size, and the second DMA and the CPC are used to measure the size distribution of humidified
15 particles. Another instrument known as the differential aerosol sizing and hygroscopicity
16 spectrometer probe (DASH-SP) can also measure the GF at different RHs (Sorooshian et al.,
17 2008). The DASH-SP couples one DMA and an optical particle size spectrometer (OPSS). The
18 dry size-dependent particles are selected by the DMA, then exposed to different RH
19 environments and finally measured in the OPSS (Sorooshian et al., 2008; Rosati et al., 2015).

20 The aerosol optical hygroscopic enhancement factor [$f(RH)$] has also been employed to
21 investigate aerosol hygroscopicity, which is defined as the ratio of aerosol optical properties
22 (aerosol extinction, scattering or backscattering coefficients) between wet and dry conditions

1 (Kotchenruther et al., 1998). Two tandem nephelometers are used to measure $f(\text{RH})$ (e.g.,
2 Covert et al., 1972; Feingold and Morley, 2003; Titos et al., 2018). One nephelometer measures
3 the aerosol optical properties of dry ambient aerosols at $\text{RH} < 40\%$, and another measures that
4 of wet aerosols at different RHs adjusted by a humidifier placed between them (Koloutsou-
5 Vakakis et al., 2001; Titos et al., 2018). MacKinnon (1969) was the first to find that the lidar
6 backscattering signal is affected by environmental RHs. Later studies have demonstrated the
7 possibility of using the lidar to observe aerosol hygroscopic growth (Tardif et al., 2002; Pahlow
8 et al., 2006; Veselovskii et al., 2009; Di Girolamo et al., 2012; Fernández et al., 2015;
9 Granados-Muñoz et al., 2015; Lv et al., 2017; Bedoya-Velásquez et al., 2018). Compared with
10 tandem nephelometers, lidar technology allows for measurements under unmodified ambient
11 atmospheric conditions without drying ambient aerosols. Actual aerosol properties are not as
12 affected when measured this way (Lv et al., 2017; Bedoya-Velásquez et al., 2018). The lidar
13 also provides an opportunity to study the vertical characterization of aerosol hygroscopicity.
14 Many ground-based Raman lidar systems have been operated around the world for measuring
15 both atmospheric water vapor and aerosol profiles at higher spatial and temporal resolutions
16 (Leblanc et al., 2012; Froidevaux et al., 2013; Wang et al., 2015; Bedoya-Velásquez et al.,
17 2018). These measurements are useful for examining the effects of aerosol hygroscopic growth
18 on pollution events (e.g., Y.-F. Wang et al., 2012, 2017; Su et al., 2017). Many studies on aerosol
19 hygroscopic growth are based on the surface measurements, but few studies have investigated
20 the vertical characterization of aerosol hygroscopicity.

21 Xingtai as a city with a high density of heavy industries was ranked as one of the most
22 polluted cities in central NCP. A joint field campaign was carried out in this region in the

1 summer of 2016. Some studies based on this campaign have been done for understanding the
2 causes and evolution of pollution events in this region (Y.-Y. Wang et al., 2018; Zhang et al.,
3 2018). These studies have shown that aerosols in Xintai are highly aged and internally mixed
4 due to strong secondary formation. The goal of this study is to further investigate how aerosol
5 hygroscopic growth affects haze events and what are the controlling factors by combining
6 surface and vertical measurements of aerosol optical, physical, and chemical properties.

7 The following section describes the instruments and methodology. Section 3 presents the
8 results and discussion. Section 4 provides a brief summary of the study.

9

10 **2. Instruments and methodology**

11 **2.1 Instruments**

12 A Raman lidar was used to analyze the relationship between atmospheric water vapor
13 content and PM₁ or PM_{2.5} mass concentrations, and to explore the atmospheric aerosol
14 hygroscopic growth effect on haze events. The lidar is an automated system that retrieves
15 atmospheric water vapor mixing ratios (W) and aerosol optical property profiles (aerosol
16 extinction and backscattering coefficients, Ångström exponent (AE), and the depolarization
17 ratio) throughout the day. The system employs a pulsed neodymium-doped yttrium aluminum
18 garnet laser as a light source and emits three laser beams simultaneously at 355, 532, and 1064
19 nm with a time resolution of 15 min and a range resolution of 7.5 m based on its original factory
20 settings. The lidar sends 5,000 laser beams in the first four minutes and ten seconds of the 15-
21 minute cycle, then the mean value of the received 5,000 signals are stored as the signal profile

1 to enhance the signal-to-noise ratio. When a laser beam is sent into the atmosphere, the received
2 backscattering signal generally includes Mie scattering by aerosols, Rayleigh scattering by
3 atmospheric molecules, and Raman scattering caused by the rotation and vibration of the
4 molecules. The size of many molecules and atoms in the atmosphere are typically much smaller
5 than the wavelength of the laser, so Rayleigh scattering occurs when they interact (Strutt, 1871).
6 Mie scattering describes the interaction between large particles (mainly atmospheric aerosols)
7 and laser beams. As for the optical receiving unit of this lidar system, optical fiber (OF),
8 dichroic beam splitter (DBS), and ultra-narrowband filters following an ultraviolet telescope
9 divides atmospheric Mie scattering signals and vibrational Raman scattering signals from H₂O
10 and N₂ molecules (at 355, 386, and 407 nm, respectively). Atmospheric Mie scattering signals
11 at 532 and 1064 nm are divided by OF, DBS and ultra-narrowband filters after a visible infrared
12 telescope. Based on the perpendicular and parallel components at 532 nm received by the lidar
13 system, the aerosol depolarization ratio, a parameter that measures the shapes of aerosols, can
14 be calculated. In general, the more irregular the particle shape, the larger the value of the
15 depolarization ratio (Chen et al., 2002; Baars et al., 2016). The AE can also be calculated using
16 lidar signals at 532 and 1064 nm, which is inversely related to the average size of the aerosols
17 (Ångström, 1964; Tiwari et al., 2016).

18 Co-located radiosondes were launched twice a day, i.e., at ~0715 and ~1915 Beijing Time
19 (BJT), during the field campaign. The GTS1 detector collected profiles of atmospheric RH,
20 temperature, and pressure at a resolution of 1.0%, 0.1°C, and 0.1 hPa, respectively. The
21 radiosonde ascension velocity was typically ~5–6 m s⁻¹.

22 A co-located Doppler lidar system (TWP3-M) was also in operation at Xingtai. This

1 system emits electromagnetic beams in different directions to the upper air, then directly
2 receives the backscattering signals after those beams interact with atmospheric turbulence.
3 Based on the Doppler effect, this system can derive time series of horizontal wind velocity and
4 direction at a time resolution of 5 min and a range resolution of 60 m below 1 km and 120 m
5 above 1 km. The root-mean-square errors (RMSEs) of the Doppler lidar-retrieved wind speed
6 and direction are typically $\leq 1.5 \text{ m s}^{-1}$ and $\leq 10^\circ$, respectively. The maximum and minimum
7 detection distances of this system are 3–5 km and 0.1 km, respectively.

8 A GrayWolf six-channel handheld particle/mass meter (model PC-3016A) was used to
9 directly monitor the total mass concentrations of $\text{PM}_{2.5}$ and PM_{10} in the actual atmosphere (Yan
10 et al., 2017). The minimum detection particle size is $0.3 \mu\text{m}$, the counting efficiency for 0.3--
11 μm particles is 50%, and 100% for particle sizes greater than $0.45 \mu\text{m}$. The non-refractory PM_{10}
12 (NR- PM_{10}) chemical components including organics, sulfate, nitrate, ammonium, and chloride
13 were measured in situ by an aerodyne quadrupole aerosol chemical speciation monitor (ACSM)
14 at a time resolution of five minutes. Detailed information about the operation of the ACSM and
15 its application in this campaign can be found elsewhere (Zhang et al., 2018). Briefly, aerosols
16 with vacuum aerodynamic diameters of $\sim 40\text{--}1000 \text{ nm}$ are sampled into the ACSM through a
17 100-mm critical orifice mounted at the inlet of an aerodynamic lens. The particles are then
18 directed onto a resistively heated surface ($\sim 600^\circ\text{C}$) where NR- PM_{10} components are flash
19 vaporized and ionized by a 70-eV electron impact. The ions are then analyzed by a commercial
20 quadrupole mass spectrometer. Mass spectra are the raw data collected by the ACSM, and
21 standard analysis software offered by Aerodyne Inc. is provided to derive mass concentrations
22 of each chemical component. In this study, the ACSM was calibrated with pure ammonium

1 nitrate following the procedure detailed by Ng et al. (2011) to determine its ionization
2 efficiency. The aerosol aerodynamic particle size was determined by an aerodynamic lens. The
3 uncertainties of ACSM-derived quantities are insignificant (Ng et al. 2011).

4 The aerosol GF probability distribution function (GF-PDF) at RH = 85% was measured
5 by an in situ H-TDMA. The H-TDMA system mainly consists of a Nafion dryer, a bipolar
6 neutralizer, two DMAs, a CPC, and a Nafion humidifier. The first DMA is used to select
7 monodispersed aerosols with a set mobility size (40, 80, 110, 150, and 200 nm in this study)
8 after the sample is dried and neutralized by the Nafion dryer and the bipolar neutralizer. The
9 selected particles are then humidified when passing through a Nafion humidifier with
10 controlled RH (85%). The second DMA and the CPC are responsible for measuring the number
11 size distribution of the humidified particles. Finally, the TDMA-fit algorithm is used to retrieve
12 GF-PDF (Stolzenburg and McMurry, 2008). Uncertainties of these retrieved parameters are
13 insignificant. More detailed descriptions about the H-TDMA system are given by Tan et al.
14 (2013) and Y.-Y. Wang et al. (2017, 2018). All data are reported in BJT in this study.

15 **2.2 Methodology**

16 **2.2.1 Water vapor retrieval**

17 Using the ratio of the Raman signals of H₂O (P_H) and N₂ (P_N), W is calculated as follows
18 (Melfi, 1972; Leblanc et al., 2012; Su et al., 2017):

$$19 \quad W(z) = C_w \Delta q \frac{P_H(z)}{P_N(z)}, \quad (1)$$

$$\Delta q = \frac{\exp[-\int_0^z (\alpha_N^m + \alpha_N^p) dz]}{\exp[-\int_0^z (\alpha_H^m + \alpha_H^p) dz]}, \quad (2)$$

where C_w is the Raman lidar calibration constant which can be calculated using co-located radiosonde data (Melfi, 1972; Sherlock et al., 1999). The parameters α_N^m and α_H^m are the molecular extinction coefficients at 386 and 407 nm, respectively. These can also be calculated using temperature and pressure profiles from radiosonde measurements (Bucholtz, 1995). The parameters α_N^p and α_H^p are the aerosol extinction coefficients (AECs) at 386 and 407 nm, respectively. Here, we use the Fernald method to retrieve AECs (Fernald et al., 1972; Fernald, 1984), which is an analytic solution to the following basic lidar equation for Mie scattering:

$$P_s(z) = ECZ^{-2}[\beta_1(z) + \beta_2(z)]T_1^2(z)T_2^2(z), \quad (3)$$

where $P_s(z)$ is the return signal, E is the energy emitted by the laser, C is the calibration constant of the lidar system, and $\beta_1(z)$ and $\beta_2(z)$ are the backscattering cross-sections of atmospheric aerosols and molecules at altitude z , respectively. $T_1(z)$ and $T_2(z)$ are the transmittances of aerosols and air molecules at height z . Note that during the daytime, the height of the retrieved W profile is limited because the Raman signal is affected by radiation (Tobin et al., 2012).

We can also calculate the vertical distribution of RH based on the vertical profile of W retrieved from Raman lidar measurements and the temperature and pressure profiles provided by radiosonde data. The following equations are used to retrieve the RH profile:

$$RH(z) = \left[\frac{e(z)}{e_s(z)} \right] \times 100\%, \quad (4)$$

$$e(z) = \frac{W(z)p(z)}{0.622 + W(z)}, \quad (5)$$

$$e_s(z) = 6.1078 \exp \left[\frac{17.13 [T(z) - 273.16]}{T(z) - 38} \right], \quad (6)$$

where $e(z)$ and $e_s(z)$ are the vertical profiles of water vapor pressure (in hPa) and saturation vapor pressure (in hPa) at a certain temperature, respectively, $W(z)$ is the W profile obtained from the Raman lidar, $p(z)$ is the pressure profile (in hPa), and $T(z)$ is the temperature profile (in K) provided by radiosonde data.

To assess the accuracy of the retrieval algorithm, Raman lidar- and radiosonde-derived W and RH profiles at 0515 BJT on 24 May 2016 and their differences are shown in Fig. 1. The W profiles agree reasonably well with an absolute error between them less than 0.5 g kg^{-1} . Absolute errors between Raman lidar- and radiosonde-derived RH profiles are generally less than 5%. The same inversion results for a relatively wet case on 23 May 2016 are given in Fig. 2. In general, large absolute errors tend to occur at the inflection points. Figures 1 and 2 suggest that the retrieval algorithm can produce reasonable results.

2.2.2 Selection of aerosol hygroscopic cases and their optical properties

How aerosol hygroscopic growth cases were chosen is described here. First, atmospheric mixing conditions were examined using radiosonde-based vertical potential temperature (θ) and W profiles. Cases with near-constant values of θ and W in the analyzed layer (variations less than 2°C and 2 g kg^{-1} , respectively) represent good atmospheric mixing conditions (Granados-Muñoz et al., 2015). Then aerosol backscattering coefficient profiles at 532 nm were calculated using the Fernald method (see details in section 2.2.1).

A simultaneous increase in atmospheric RH and the aerosol backscattering coefficient is also needed, which might indicate aerosol hygroscopic growth (Bedoya-Velásquez et al., 2018).

1 Based on the above criteria, individual cases with the same ambient humidity and different
 2 pollution conditions were selected for studying the influence of aerosol hygroscopicity on haze
 3 events. Aerosol hygroscopic properties of the selected cases were investigated in terms of the
 4 hygroscopic enhancement factor for the aerosol backscattering coefficient defined as follows:

$$5 \quad f_{\beta}(RH, \lambda) = \frac{\beta(RH, \lambda)}{\beta(RH_{ref}, \lambda)}, \quad (7)$$

6 where $\beta(RH, \lambda)$ and $\beta(RH_{ref}, \lambda)$ represent aerosol backscattering coefficients at a
 7 certain RH value and at a reference RH value, respectively, at wavelength λ . In this study,
 8 we selected $RH_{ref} = 80\%$ which is the lowest RH in the layer.

9 Finally, a relationship between $f_{\beta}(RH)$ and RH was established. The most commonly
 10 used equations are the single-parameter fit equation (e.g., Hänel, 1980; Kotchenruther and
 11 Hobbs, 1998; Gassó et al., 2000) and the dual-parameter fit equation (e.g., Hänel, 1980; Carrico,
 12 2003; Zieger et al., 2011). The single-parameter fit equation introduced by Hänel (1976) is

$$13 \quad f_{\beta}(RH) = \left(\frac{1 - RH}{1 - RH_{ref}} \right)^{-\gamma}, \quad (8)$$

14 where γ is an empirical parameter. Larger γ values in this formulation denote a stronger
 15 hygroscopic growth.

16 The dual-parameter fit equation is (Fernández et al., 2015)

$$17 \quad f_{\beta}(RH) = a(1 - RH)^{-b}. \quad (9)$$

18 The single- and dual-parameter fit equations are similar, but with an additional scale factor
 19 parameter, a , in the case of the dual-parameter fit equation. The parameter b is also an
 20 empirical parameter with larger values of b indicating particles with stronger hygroscopicities.
 21 In this study, both parameterized equations are used to verify the consistency of the results. The

1 equation that fits the measurement data best is selected.

2

3 **2.2.3 Calculation of aerosol acidity**

4 Aerosol acidity is associated with aerosol hygroscopic growth (e.g. Sun et al., 2009; Fu et
5 al., 2015; Zhang et al., 2015; Lv et al., 2017). When atmospheric aerosols are acidic, they have
6 stronger hygroscopicities than when in their neutralized forms (Zhang et al., 2015). The
7 swelling of aerosols due to hygroscopic growth enhances their ability to scatter solar radiation.
8 We examined acidity by comparing the measured NH_4^+ mass concentration with the needed
9 amount to fully neutralize sulfate, nitrate, and chloride ions ($NH_{4\text{ predicted}}^+$) detected by the
10 ACSM (Sun et al., 2009; Zhang et al., 2015; Lv et al., 2017):

$$11 \quad NH_{4\text{ predicted}}^+ = (2 \times SO_4^{2-} / 96 + NO_3^- / 62 + Cl^- / 35.5) \times 18, \quad (10)$$

12 where SO_4^{2-} , NO_3^- , and Cl^- represent the mass concentrations (in $\mu\text{g m}^{-3}$) of the three
13 species measured by the ACSM. The molecular weights of SO_4^{2-} , NO_3^- , Cl^- , and NH_4^+ are
14 96, 62, 35.5, and 18, respectively. Aerosols are considered “more acidic” if the measured NH_4^+
15 mass concentration is significantly lower than that of $NH_{4\text{ predicted}}^+$. Aerosols are considered
16 “bulk neutralized” if the two values are similar (Zhang et al., 2007; Sun et al., 2009; Zhang et
17 al., 2015; Lv et al., 2017).

18 The acidity of aerosols can be quantified by a parameter called the acid value (AV) (Zhang
19 et al., 2007):

$$20 \quad AV = (2 \times SO_4^{2-} / 96 + NO_3^- / 62 + Cl^- / 35.5) / (NH_4^+ / 18). \quad (11)$$

21 The chemical formula and numbers after the equal sign have the same meanings as in Eq. (10).

1 Aerosols are considered “bulk neutralized” if $AV = 1$ and “strongly acidic” if $AV > 1.25$. When
 2 $AV = 1.25$, 50% of the total sulfate ions in the atmosphere consist of NH_4HSO_4 , and the other
 3 50% consist of $(\text{NH}_4)_2\text{SO}_4$.

4 **2.2.4 Aerosol chemical ion-pairing scheme**

5 The magnitude of $f(\text{RH})$ is correlated with the inorganic mass fraction (Zieger et al., 2014).
 6 However, GFs differ with different inorganic salts. To examine the mass fractions of neutral
 7 inorganic salts, ACSM measurements were used to calculate their mass concentrations and
 8 volume fractions (Gysel et al., 2007). This approach is based on the ion-pairing scheme
 9 introduced by Reilly and Wood (1969). The ACSM mainly measures the mass concentrations
 10 of SO_4^{2-} , NO_3^- , NH_4^+ , Cl^- , and organics. The chlorine ion was not considered here because
 11 its concentration is low. The aerosol chemical ion combination scheme is given by the
 12 following equations:

$$\begin{aligned}
 n_{\text{NH}_4\text{NO}_3} &= n_{\text{NO}_3^-} \\
 n_{\text{NH}_4\text{HSO}_4} &= \min(2n_{\text{SO}_4^{2-}} - n_{\text{NH}_4^+} + n_{\text{NO}_3^-}, n_{\text{NH}_4^+} - n_{\text{NO}_3^-}) \\
 n_{(\text{NH}_4)_2\text{SO}_4} &= \max(n_{\text{NH}_4^+} - n_{\text{NO}_3^-} - n_{\text{SO}_4^{2-}}, 0) \quad , \\
 n_{\text{H}_2\text{SO}_4} &= \max(0, n_{\text{SO}_4^{2-}} - n_{\text{NH}_4^+} + n_{\text{NO}_3^-}) \\
 n_{\text{HNO}_3} &= 0
 \end{aligned} \tag{12}$$

14 where n donates the mole numbers, and “min” and “max” are minimum and maximum values
 15 (Gysel et al., 2007). The volume fractions of inorganic salts can be calculated based on the ion
 16 combination scheme and the parameters in Table 1. Furthermore, for a multicomponent particle,
 17 the Zdanovskii-Stocks-Robinson mixing rule (Zdanovskii, 1948; Stokes and Robinson, 1966)
 18 can be applied to calculate the hygroscopicity parameter κ :

$$\kappa = \sum_i \varepsilon_i \kappa_i \quad , \tag{13}$$

1 where κ_i is the hygroscopicity parameter of each individual component. The parameter ε_i
2 is the volume fraction of each component.

3 **3. Results and discussion**

4 **3.1 Observations of W and mass concentrations of PM₁ and PM_{2.5}**

5 Figure 3a shows the time series of the lidar-derived W at Xingtai from 19–31 May 2016.
6 The height of the retrieved W profile was limited because of solar radiation during the daytime
7 (e.g., Tobin et al., 2012). Overall, W is generally less than 6 g kg⁻¹ between 0.3–4 km with a
8 strong daily variability during the period analyzed. Figures 3b and 3c show the simultaneous
9 time series of the surface mass concentrations of PM₁ and PM_{2.5}, and W and RH, respectively.
10 The variabilities in PM₁ and PM_{2.5} mass concentrations are strongly coupled with that in W at
11 the surface and in the lower atmospheric layer. Others have also found the same relationship
12 between W in the lower atmospheric layer and the surface mass concentration of PM_{2.5} (e.g.,
13 Y.-F. Wang et al., 2012, 2017; Su et al., 2017). Su et al. (2017) suggested that this is due to the
14 aerosol hygroscopic growth effect. The aerosol hygroscopicity is related to aerosol chemical
15 composition over the North China Plain (Zou et al., 2018). Figure 3d shows simultaneous mass
16 fractions of the chemical species comprising PM₁. As W in the lower atmospheric layer and the
17 surface mass concentrations of PM₁ and PM_{2.5} increases, the proportion of organic aerosols
18 decreases (highlighted as shaded grey areas in Fig. 3), suggesting that the proportion of
19 hygroscopic aerosols increased. This shows that strong aerosol hygroscopicity may aggravate
20 air pollution conditions over Xingtai.

21 Two instances when this relationship was not seen (highlighted as shaded grey areas in

1 Fig. 3) are shown by the black triangles in Fig. 3d and marked with grey lines across Fig. 3. In
2 the evening of 21 May 2016 (the leftmost triangle and grey line), W and the mass fractions of
3 organics are comparable to those in the evening of 23 May (the rightmost triangle and grey line
4 in Fig. 3). However, the mass concentrations of PM_{10} and $PM_{2.5}$ at that time indicated by the
5 leftmost grey line (in the evening of 21 May 2016) are significantly less than that in the evening
6 of 23 May (indicated by the rightmost grey line). Su et al. (2017) and Y.-F. Wang et al. (2012,
7 2017) have studied the relationship between atmospheric water vapor and haze events over
8 Beijing and Xi'an, respectively, using Raman lidar measurements. Their analyses showed a
9 positive correlation between W , and PM_{10} and $PM_{2.5}$ mass concentrations, but they ignored
10 some unexpected cases behind this positive correlation. The two unexpected cases that
11 occurred on 21 May 2016 (Case I) and 23 May 2016 (Case II) were selected for further study.

12 **3.2 Cases studies of aerosol hygroscopic growth**

13 **3.2.1 Lidar-estimated hygroscopic measurements**

14 Two cases were selected: one on 21 May 2016 (Case I) and the other on 23 May 2016
15 (Case II) at the closest time of the radiosonde launch time at 1915 BJT. Figure 4 shows the
16 vertical distributions of W , θ , the aerosol backscattering coefficient at 532 nm (β_{532}), the
17 backscatter-related Ångström exponent (AE) based on measurements at 532 and 1064 nm, and
18 the particle linear depolarization ratio at 532 nm for Case I and Case II. The altitude ranges are
19 1642.5–1905.5 m for Case I and 1680.0–2130.0 m for Case II. W and θ calculated from
20 radiosonde-measured temperature and RH profiles were used to examine the atmospheric
21 mixing conditions in the individual layers. Table 2 lists the gradients of the variables within

1 each layer. The gradient in W changes little within the layer of interest, decreasing
2 monotonically with altitude at a rate of $-0.34 \text{ g kg}^{-1} \text{ km}^{-1}$ and $-1.42 \text{ g kg}^{-1} \text{ km}^{-1}$ for Case I and
3 Case II, respectively. The gradient in θ shows a monotonic increase within the layers of interest
4 ($0.27^\circ\text{C km}^{-1}$ for Case I and $0.96^\circ\text{C km}^{-1}$ for Case II). Overall, W and θ variations are less than
5 2 g kg^{-1} and 2°C , respectively, showing that good mixing atmospheric conditions were present
6 in both cases (Granados-Muñoz et al., 2015). This confirms that aerosols within the analyzed
7 layer of each case were well mixed.

8 Figure 4 shows the time series of the horizontal wind velocity and direction retrieved from
9 the co-located Doppler lidar system. From 1830–2030 BJT, Case I (Fig. 4c) and Case II (Fig.
10 4d) winds within their respective layers are mainly from the north and northwest, respectively,
11 and have relatively low speeds ($< 5 \text{ m s}^{-1}$, Fig. 4a and 4b). This suggests that the aerosols in
12 each case were transported into their respective layers at low speeds from almost the same
13 direction. In other words, there is no change in the aerosol type of both cases within the region
14 of interest.

15 The RH and β_{332} simultaneously increase with altitude in the Case I (Fig. 5c and 5d) and
16 Case II (Fig. 5i and 5j) layers of interest. The AE and depolarization ratio were retrieved in
17 order to differentiate the fine/coarse mode predominance and shape of the aerosols (Fig. 5e, f,
18 k, and l). A decrease in AE and the depolarization ratio suggests that there is an increase in the
19 predominance of coarse-mode particles and an increase in the sphericity of particles due to
20 water uptake, respectively (Granados-Muñoz et al., 2015; Lv et al., 2017; Bedoya-Velásquez
21 et al., 2018).

1 Based on the β_{532} and RH profiles retrieved from Raman lidar measurements, the
2 enhancement factor for the backscattering coefficient at 532 nm, $f_{\beta}(RH)$, is calculated for
3 both cases using Eq. (7). The reference RH value was set to 80% in this study, the lowest RH
4 recorded in the layers of interest of both cases. This study applies the single-parameter Hänel
5 model [Eq. (8)] and the dual-parameter Kasten model [Eq. (9)]. Table 3 lists the parameterized
6 results of each model for each case, and Fig. 6 shows the best-fit lines. The $f_{\beta}(RH)$ for Case
7 II is greater than that for Case I. β_{532} increases by a factor of 1.094 (Case I) and 1.794 (Case
8 II) as RH changes from 80% to 91%. The magnitudes of $f_{\beta}(85\%)$ for Case I and Case II are
9 1.0283 and 1.0770, respectively. The b value from the Kasten parameterization is much larger
10 in Case II (0.9346) than in Case I (0.1000), and the γ value from Hänel parameterization for
11 Case II (0.6538) is also much larger than that for Case I (0.09895). Chen et al. (2014) studied
12 the aerosol hygroscopicity parameter derived from light-scattering enhancement factor
13 [$f(RH)$] measurements made in the NCP and showed that $f(RH)$ for polluted cases is
14 higher than that for clean periods at a specific RH. This is consistent with the results of this
15 study where the mass concentrations of PM_{1} and $PM_{2.5}$ during Case II ($69.36 \mu\text{g m}^{-3}$ for PM_{1}
16 and $94.88 \mu\text{g m}^{-3}$ for $PM_{2.5}$) were greater than those during Case I ($34.08 \mu\text{g m}^{-3}$ for PM_{1} and
17 $45.00 \mu\text{g m}^{-3}$ for $PM_{2.5}$). An observational study of the influence of aerosol hygroscopic growth
18 on the scattering coefficient at a rural area near Beijing also demonstrated that aerosols had
19 relatively strong water-absorbing properties during urban pollution periods (Pan et al., 2009).

20 **3.2.2 The influences of chemical composition inferred from ACSM measurements**

21 Liu et al. (2014) have pointed out that inorganics are the primary aerosol component
22 contributing to aerosol hygroscopicity especially in the size range of 150–1000 nm. The acidity

1 of aerosols is a key parameter affecting aerosol hygroscopic growth (Sun et al., 2009; Lv et al.,
2 2017). The dominant form of inorganics can be examined by comparing measured NH_4^+ and
3 predicted NH_4^+ (Lv et al., 2017; see section 2.2.3 for details).

4 Figure 7 shows the relationships between ACSM-measured NH_4^+ and predicted NH_4^+
5 based on PM_{10} chemical species information collected during the full day of each case. The
6 slopes of the linear regression best-fit lines are 0.72 and 0.68 on 21 May 2016 (Case I) and 23
7 May 2016 (Case II), respectively. The RMSEs of the linear regression best-fit lines are 0.63 and
8 0.48 on 21 May 2016 and 23 May 2016, respectively. The parameter AV for Case I is 1.35 and
9 for Case II is 1.50. These values suggest that there was insufficient NH_3 in the atmosphere to
10 neutralize H_2SO_4 , HNO_3 , and HCl in each case and that the dominant form of inorganics was
11 NH_4NO_3 , NH_4HSO_4 , and $(NH_4)_2SO_4$. The acidity of aerosols in Case II is greater than that in
12 Case I, suggesting that aerosols in Case II were more hygroscopic than those in Case I. This is
13 consistent with the results presented in section 3.2.1.

14 A hygroscopicity parameter, κ , was developed by Petters and Kreidenweis (2007). κ can
15 be calculated using the chemical composition information from Eq. (13) (Gysel et al., 2007;
16 Y.-C. Liu et al., 2016; see section 2.3.4). To further confirm the effect of aerosol hygroscopic
17 growth on haze events, κ is computed for each case based on the dominant form of the
18 inorganics mentioned above.

19 Figure 8 shows the chemical species obtained from ground-based ACSM measurements
20 of PM_{10} around the times of the cases. In Case I (Fig. 8a), PM_{10} was mainly made up of organic
21 particles (39%) and sulfate (39%), followed by nitrate (8%), ammonium (13%), and chloride

1 (1%). In Case II (Fig. 8b), PM₁ was made up of 37% organics, 25% sulfate, 22% nitrate, 12%
2 ammonium, and 1% chloride. Based on the aerosol chemical ion-pairing scheme introduced in
3 Section 2.2.4 and the aerosol properties shown in Table 1, chloride and organics were neglected
4 because of their relatively small contents and comparatively low hygroscopicities (Gysel et al.,
5 2007; Petters and Kreidenweis, 2013). Table 4 lists the mass concentrations and volume
6 fractions of NH₄NO₃, NH₄HSO₄, and (NH₄)₂SO₄ for each case as well as κ computed using
7 Eq. (13). The mass concentration of H₂SO₄ is equal to zero. Liu et al. (2014) have shown that
8 κ for NH₄NO₃, NH₄HSO₄, and (NH₄)₂SO₄ is equal to 0.68, 0.56, and 0.60, respectively. The
9 parameter κ for Case I (0.557) is less than that for Case II (0.610). This suggests that the aerosol
10 hygroscopicity for Case II was higher than that for Case I. It also suggests that under the same
11 ambient RH conditions, the nitrate content in aerosols can cause differences in the
12 hygroscopicity of aerosols.

13 **3.2.3 Comparison with H-TDMA measurements**

14 In the last decade, many studies have compared remotely sensed and in situ aerosol
15 scattering enhancement factor measurements using a humidified tandem nephelometer and
16 have shown positive results (Zieger et al., 2011, 2012; Sheridan et al., 2012; Tesche et al., 2014;
17 Lv et al., 2017). The H-TDMA is also a reliable instrument for measuring the aerosol diameter
18 hygroscopicity due to water uptake (Liu et al., 1978). Aerosol GFs observed by the ground-
19 based H-TDMA at times nearest to the times of each case are examined next.

20 Based on H-TDMA-derived aerosol GFs at an RH level of 85% for different particle sizes
21 (40, 80, 110, 150, and 200 nm), GFs for different aerosol sizes in both cases were extrapolated

1 to different RH levels using Eq. (3) from (2009) who used the κ model introduced by Petters
2 and Kreidenwies (2007). Figure 9 shows that Case II aerosol GFs at each RH level (80–91%)
3 are higher than those of Case I. Although the $f_{\beta}(RH)$ and GF are completely different
4 parameters for calculating the hygroscopicity of aerosols and are difficult to compare
5 quantitatively, the H-TDMA results offer a sense of confidence that aerosol hygroscopicity has
6 an important influence on the formation of heavy haze.

7 In general, both the lidar-estimated aerosol backscattering hygroscopic enhancement
8 factor and the ACSM and H-TDMA measurements support the proposed hypothesis that the
9 different hygroscopic properties of aerosols are mainly responsible for the strong coupling
10 between the variability in PM_{1} and $PM_{2.5}$ mass concentrations and the variability in W .

11 **4. Conclusions**

12 During late May 2016, the water vapor mixing ratio in the 0.3–4 km layer over Xingtai
13 was generally less than 6 g kg^{-1} with a strong daily variability. Overall, the simultaneous
14 temporal changes in the mass concentrations of PM_{1} and $PM_{2.5}$ were strongly associated with
15 that of the atmospheric water vapor content due to the hygroscopicity of aerosols. Two cases
16 where this relationship was not seen were identified and further examined. Case I represents a
17 relatively clean case, and Case II represents a polluted case. The lidar-estimated aerosol
18 backscattering coefficient hygroscopic enhancement factor [$f_{\beta}(RH)$] for Case II is greater
19 than that for Case I. The γ and b values from the Hanel and Kasten parameterizations,
20 respectively, for Case II were larger than those for Case I. A key parameter affecting the
21 hygroscopicity of aerosols, namely, the acid value (AV), was examined by comparing measured

1 NH_4^+ and predicted NH_4^+ based on data obtained by the ACSM. The AV for Case I (1.35) was
2 less than that for Case II (1.50) and the main form of inorganics was NH_4NO_3 , NH_4HSO_4 , and
3 $(NH_4)_2SO_4$. The aerosol chemical composition determined by the ACSM showed that the
4 aerosol hygroscopicity parameter κ for Case II (0.610) was greater than that for Case I (0.577)
5 due to the greater mass fraction of nitrate salt. Based on H-TDMA measurements, model results
6 showed that the aerosol size hygroscopic growth factor (GF) in each particle size category (40,
7 80, 110, 150, and 200 nm) for Case II was greater than that for Case I.

8 The $f_\beta(RH)$, GF, AV , and κ are completely different quantities for calculating the
9 hygroscopicity of aerosols and are difficult to compare quantitatively. The lidar-estimated
10 $f_\beta(RH)$ and ACSM and H-TDMA measurements show that the hygroscopic growth of
11 aerosols has a strong influence on the process of air pollution. Under the same atmospheric
12 relative humidity conditions, the stronger the hygroscopicity of aerosols, the more likely they
13 cause severe air pollution. The mass fraction of the nitrate ion in aerosols was one of the main
14 factors that determined the hygroscopic ability of aerosols in the study area (Xingtai). These
15 findings not only reveal why haze events in Xingtai can be severe, but they also provide
16 scientific evidence that may be used to persuade the local government to prevent and control
17 environmental contamination in this heavily polluted part of China.

18

19 **Author contributions**

20 ZL and JC determined the main goal of this study. JC carried it out, analyzed the data and
21 prepared the paper with contributions from all co-authors. YW provided technical guidance for
22 related instruments.

1 **Data availability**

2 Data used in this study are available from the first author upon request
3 (jchen@mail.bnu.edu.cn).

4

5 **Competing interests**

6 The authors declare that they have no conflict of interest.

7

8 **Special issue statement**

9 This article is part of the special issue “Regional transport and transformation of air pollution
10 in eastern China”. It is not associated with a conference.

11

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Table 1. Aerosol properties of selected compounds used for the calculation of the hygroscopicity parameter κ , i.e., the density (ρ_i) and (κ_i) of each compound.

species	NH ₄ NO ₃	NH ₄ HSO ₄	(NH ₄) ₂ SO ₄	H ₂ SO ₄
density ^a	1.725	1.78	1.76	1.83
κ ^b	0.68	0.56	0.52	0.91

(a) Tang and Munkelwitz (1994); Carrico et al. (2010);

(b) Fountoukis and Nenes (2007); Carrico et al. (2010); Liu et al. (2014).

Table 2. Range of values and gradient values over the analyzed layer for the water vapor mixing

ratio (W), the potential temperature (θ), the backscattering coefficient at 532 nm (β_{532}), the Ångström exponent [AE (532–1064 nm)], and the depolarization ratio at 532 nm for Cases I and

II.

	Case I			Case II		
	Range		Gradient (km^{-1})	Range		Gradient (km^{-1})
Altitude (m)	1642.5	1905.0	—	1680.0	2130.0	—
W (g kg^{-1})	7.65	7.56	-0.34	6.42	5.78	-1.42
θ ($^{\circ}\text{C}$)	26.93	27.00	0.27	25.18	25.61	0.96
RH (%)	80	91	—	80	91	—
$\beta_{532\text{nm}}$ ($\text{km}^{-1} \text{sr}^{-1}$)	0.01379	0.01535	—	0.003711	0.006762	—
AE (532–1064 nm)	0.74	0.68	-0.23	0.42	0.35	-0.16
Depolarization ratio	0.046	0.044	-0.0076	0.041	0.039	-0.0044

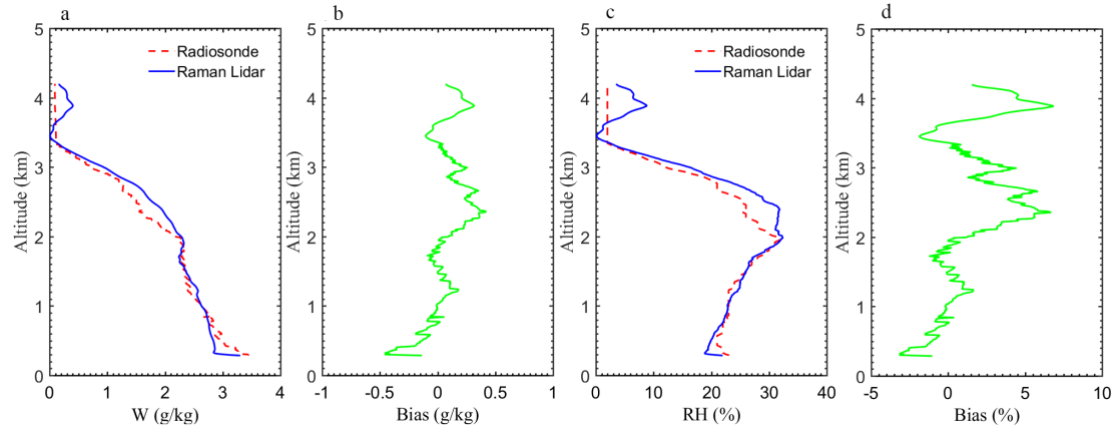
Table 3. The fitting parameters and R^2 of the fits for the Kasten and Hänel models.

	Case I			Case II		
	a	b	R^2	a	b	R^2
Kasten model	0.8508	0.1000	0.97	0.1916	0.9346	0.95
	γ	R^2	γ	R^2		
Hänel model	0.09895±0.0047		0.97	0.6538±0.0662		0.84

Table 4. Calculated mass concentrations and volume fractions of NH_4NO_3 , NH_4HSO_4 , and $(\text{NH}_4)_2\text{SO}_4$, and the hygroscopicity parameter (κ) for Case I and Case II.

	Case I			Case II		
	NH_4NO_3	NH_4HSO_4	$(\text{NH}_4)_2\text{SO}_4$	NH_4NO_3	NH_4HSO_4	$(\text{NH}_4)_2\text{SO}_4$
mass conc. ($\mu\text{g m}^{-3}$)	3.60	8.31	8.30	12.2979	10.3795	3.0616
volume fraction	0.18	0.41	0.41	0.48	0.40	0.12
κ		0.557			0.610	

Fig. 1. (a, c) Water vapor mixing ratio (W) and relative humidity (RH) profiles at 0515 BJT 24 May



2016 retrieved by the Raman lidar (blue line) and the radiosonde (red dashed line), respectively, and (b, d) the absolute error in W and RH between the lidar and radiosonde retrievals (lidar minus radiosonde), respectively.

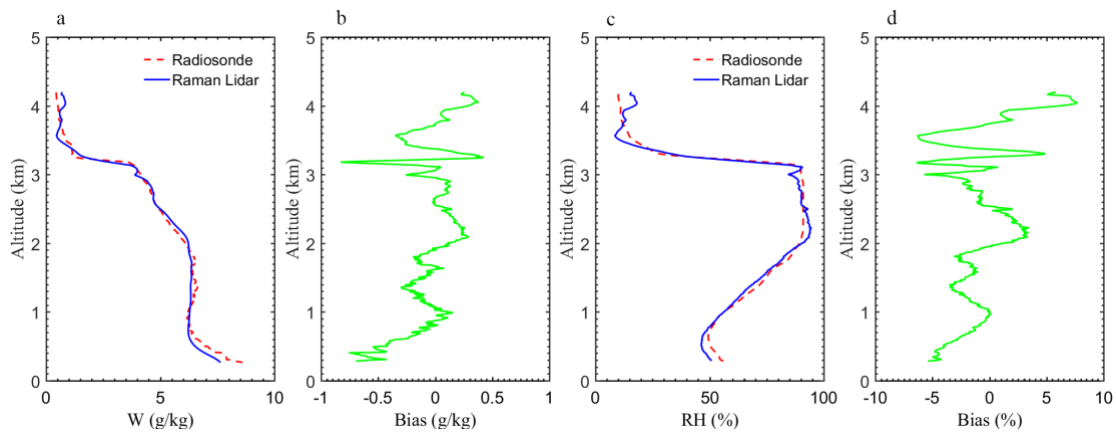


Fig. 2. (a, c) Water vapor mixing ratio (W) and relative humidity (RH) profiles at 2000 BJT 23 May 2016 retrieved by the Raman lidar (blue line) and the radiosonde (red dashed line), respectively, and (b, d) the absolute error in W and RH between the lidar and radiosonde retrievals (lidar minus radiosonde), respectively.

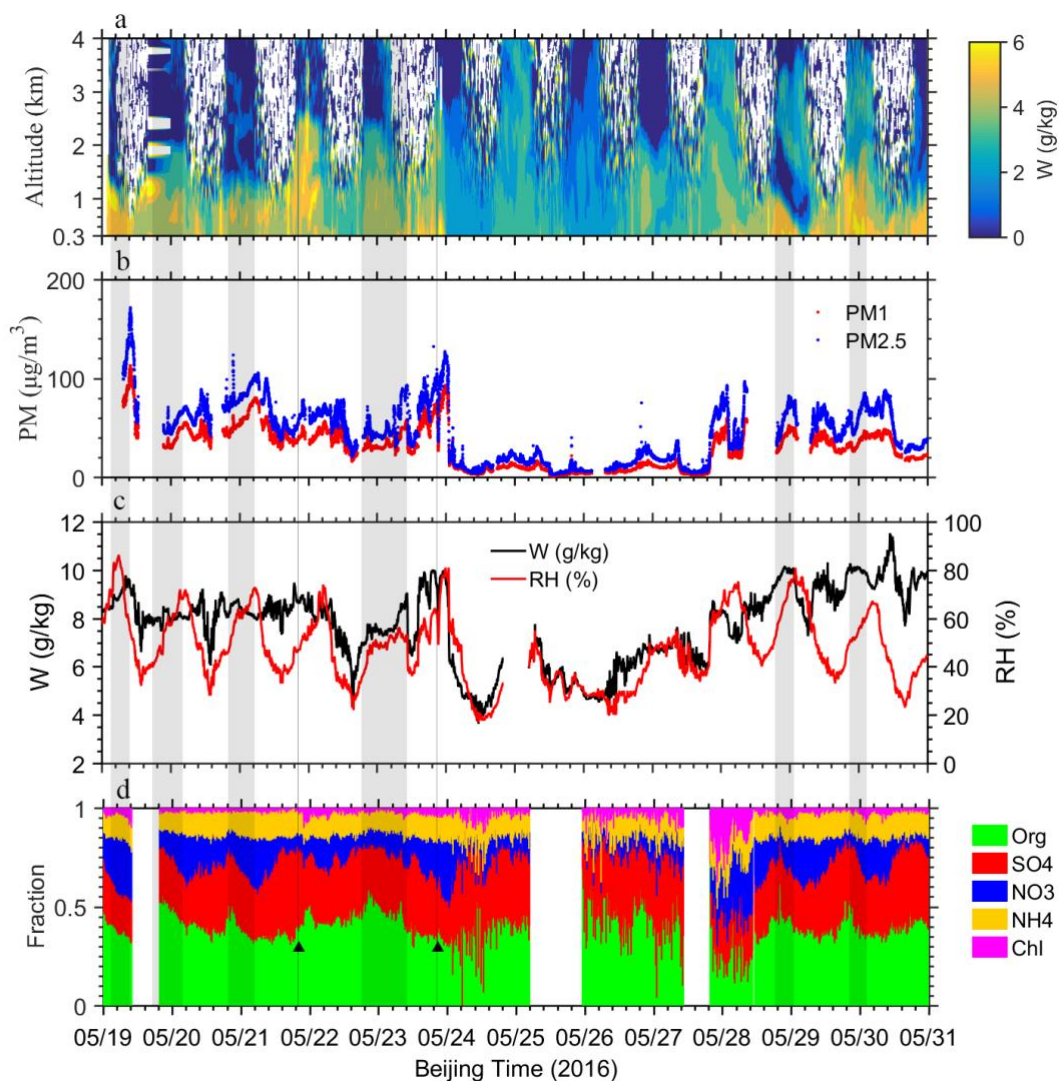


Fig. 3. Time series of (a) water vapor mixing ratio (W) profiles measured by the Raman lidar, (b) mass concentrations of PM_{1} (red dots) and $PM_{2.5}$ (blue dots), (c) surface W (black line) and relative humidity (RH, red line), and (d) chemical species mass fractions of PM_{1} measured by the ACSM. Data are from 19–31 May 2016 at Xingtai. The shaded grey areas are to enhance the readability of the article. The black triangles in (d) and grey lines in (a, b, c, d) represent the two cases chosen for further examination. Blank parts of the data are missing due to uncontrollable factors such as power supply.

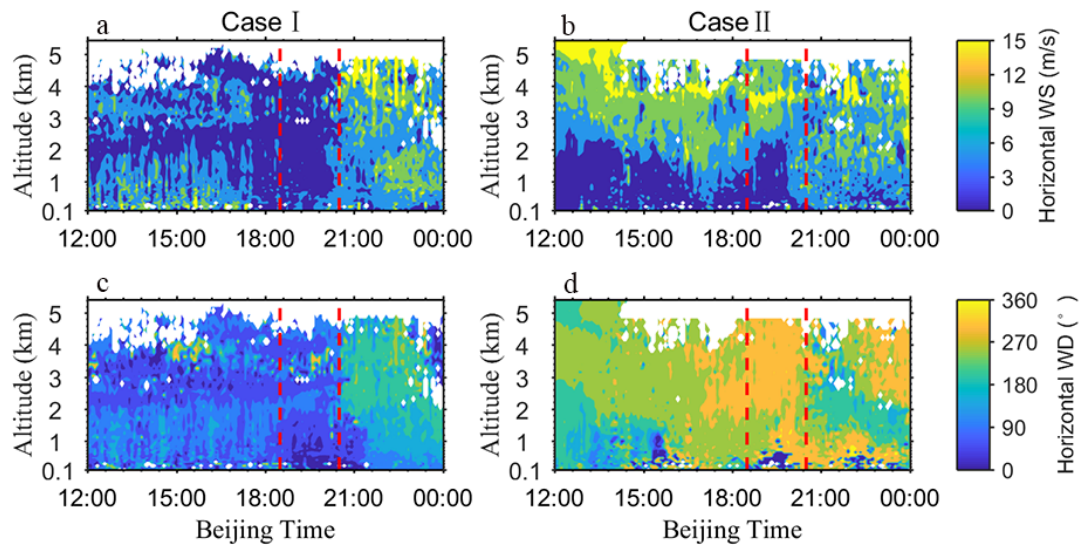


Fig. 4. Time series of Doppler-lidar-retrieved (a, b) horizontal wind speed and (c, d) horizontal wind direction on 21 May 2016 (Case I, left-hand panels) and 23 May 2016 (Case II, right-hand panels). Red dashed lines outline the time range 1830–2030 BJT. The analyzed layers are 1642.5–1905.0 m for Case I and 1680.0–2130.0 m for Case II.

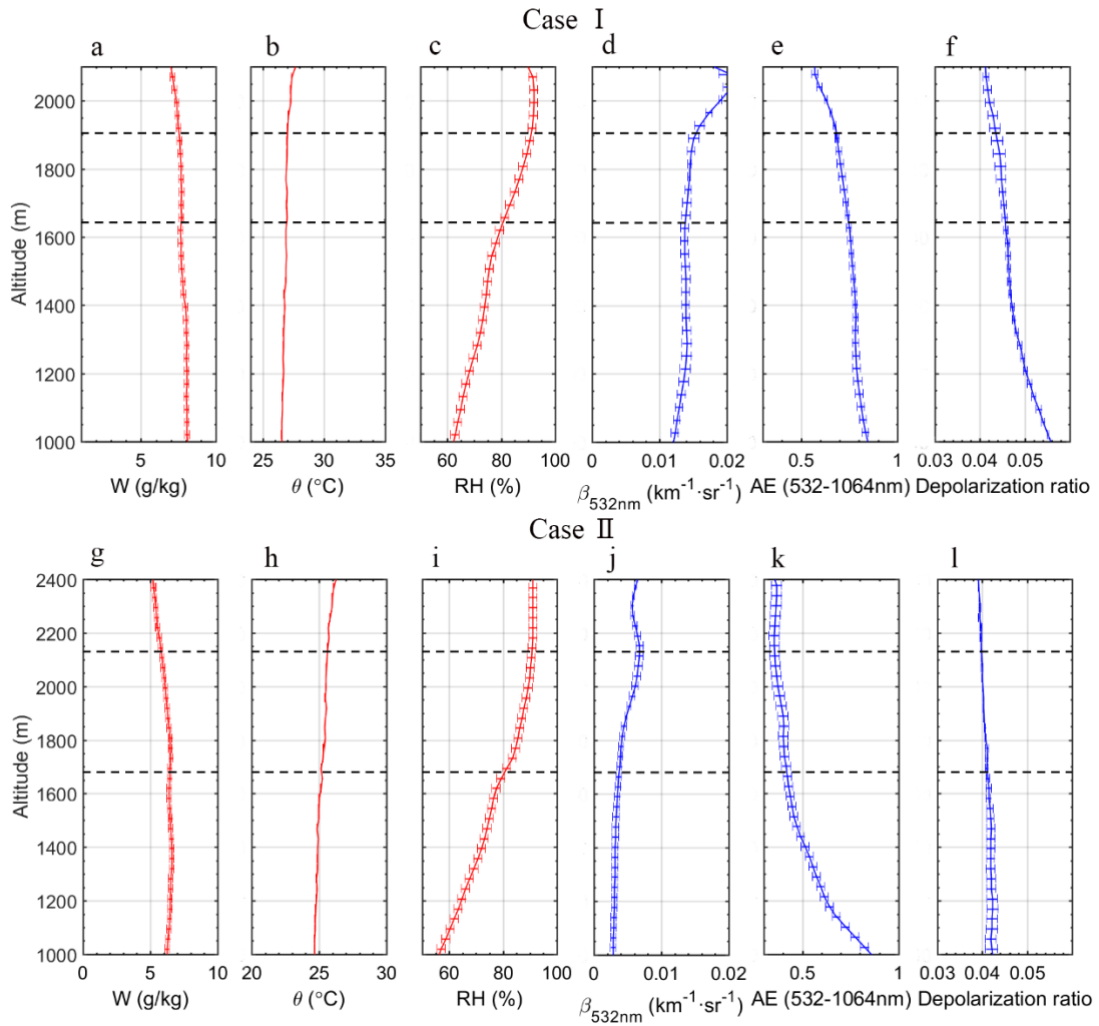


Fig. 5. The vertical profiles of (a, g) water vapor mixing ratio (W), (b, h) potential temperature (θ), (c, i) relative humidity (RH) calculated from radiosonde data, (d, j) backscattering coefficient at 532 nm (β_{532}), (e, k) the Ångström exponent [AE (532-1064nm)], (f, l) depolarization ratio retrieved from Raman lidar data for Case I (top panels) and Case II (bottom panels). Horizontal dashed lines show the upper and lower boundaries of the layer under analysis (1642.5–1905.0 m for Case I and 1680.0–2130.0 m for Case II). Horizontal error bars denote the uncertainty of each property.

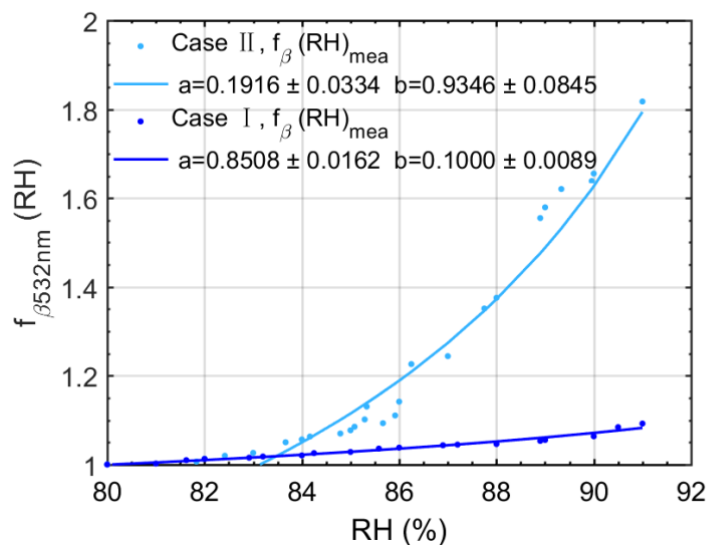


Fig. 6. $f_{\beta}(RH)$ at 532 nm retrieved on 21 May 2016 in the 1642.5–1905.0 m layer (Case I, dark blue points) and 23 May 2016 in the 1680.0–2130.0 m layer (Case II, light blue points). The best-fit lines through the points are shown. The reference RH is 80 %.

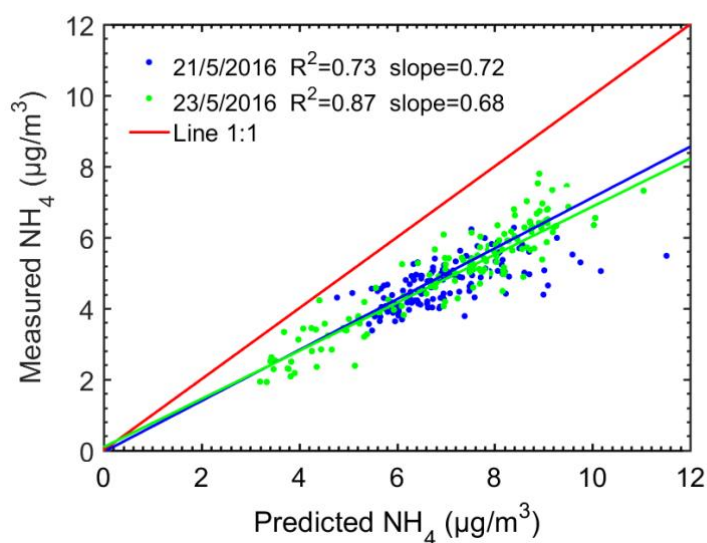


Fig. 7. Mass concentrations of measured ammonium (NH_4) versus predicted ammonium assuming full neutralization of sulfate, nitrate and chloride on the whole day of 21 May 2016 (blue dots, Case I) and 23 May 2016 (green dots, Case II). The solid blue and green lines are the least-squares regression lines for each day. The 1:1 line is shown in red.

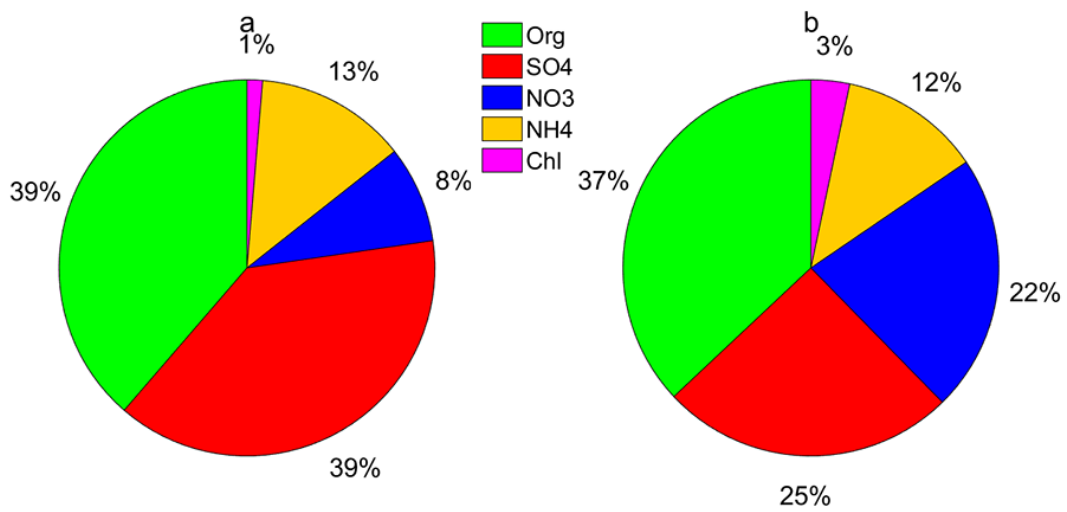


Fig. 8. Aerosol mass fractions of PM₁ measured by the ACSM for (a) Case I and (b) Case II.

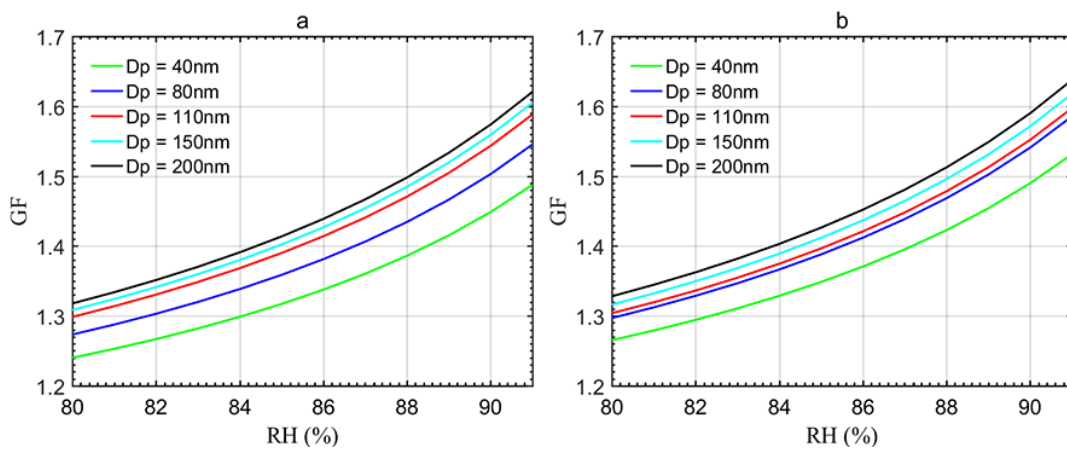


Fig. 9. Aerosols size hygroscopic growth factor (GF) as a function of relative humidity (RH) for (a) Case I and (b) Case II. The different colors represent different particle sizes (Dp). These are the results of a model based on Eq. 3 from Gysel et al. (2009).