Surface–atmosphere exchange of inorganic water–soluble gases and associated ions in bulk aerosol above agricultural grassland pre– and post– fertilisation

Robbie Ramsay[1,2], Chiara F. Di Marco[1], Mathew R. Heal[2], Marsailidh M. Twigg[1], Nicholas Cowan[1], Matthew R. Jones[1], Sarah R. Leeson[1], William J. Bloss[3], Louisa J. Kramer[3], Leigh Crilley[3], Matthias Sörgel[4], Meinrat Andreae[4,5], Eiko Nemitz[1]

[1] Centre for Ecology & Hydrology (CEH), Edinburgh, EH26 0QB, United Kingdom
[2] School of Chemistry, University of Edinburgh, West Mains Road, EH9 3JJ, United Kingdom
[3] School of Geography, Earth and Environmental Sciences, University of Birmingham, Birmingham, B15 2TT, United Kingdom
[4] Max Plank Institute for Chemistry, Biogeochemistry Department, Mainz, P.O. Box 3060, Germany
[5] Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA 92093, USA

Correspondence to: Eiko Nemitz (en@ceh.ac.uk)

Abstract. The increasing use of intensive agricultural practices can lead to damaging consequences for the atmosphere through enhanced emissions of air pollutants. However, there are few direct measurements of the surface-atmosphere exchange of trace gases and water–soluble aerosols over agricultural grassland, particularly of reactive nitrogen compounds. In this study, we present measurements of the concentrations, fluxes and deposition velocities of the trace gases HCl, HONO, HNO_3, SO_2 and NH_3, and their associated water-soluble aerosol counterparts Cl^-, NO_2^-, NO_3^-, SO_4^{2-}, NH_4^+ as determined hourly for one month in May–June 2016 over agricultural grassland near Edinburgh, UK, pre- and post- fertilisation. Measurements were made using the Gradient of Aerosols and Gases Online Registration (GRAEGOR) wet–chemical two–point gradient instrument. Emissions of NH_3 peaked at 1460 ng m^{-2} s^{-1} three hours after fertilisation, with an emission of HONO peaking at 4.92 ng m^{-2} s^{-1} occurring five hours after fertilisation. Apparent emissions of NO_3^- aerosol were observed after fertilisation which, coupled with a divergence of HNO_3 deposition velocity (V_d) from its theoretical maximum value, suggested the reaction of emitted NH_3 with atmospheric HNO_3 to form ammonium nitrate aerosol. The use of the conservative exchange fluxes of tot-NH_4^+ and tot-NO_3^- indicated net emission of tot-NO_3^-, implying a ground source of HNO_3 after fertilisation. Daytime concentrations of HONO remained above the detection limit (30 ng m^{-3}) throughout the campaign, suggesting a daytime source for HONO at the site. Whilst the mean V_d of NH_4^+ was with 0.93 mm/s in the range expected for the accumulation mode, the larger average V_d for Cl^- (3.65 mm/s), NO_3^- (1.97 mm/s), SO_4^{2-} (1.89 mm/s) reflected the contribution of a super-micron fraction and decreased with increasing $PM_{2.5}$/PM_{10} ratio (a proxy measurement for aerosol size), providing evidence – although limited by the use of a proxy for aerosol size – of a size-dependence of aerosol deposition velocity for aerosol chemical compounds, which has been suggested from process orientated models of aerosol deposition.

1 Introduction

As the demand for food production grows in line with an increasing global population, so too does the development of intensive agricultural practices. These can have deleterious impacts on the environment and human health (Godfray et al., 2010; Foley et al., 2011), particularly through the emission of trace gases and the formation of airborne particles generated by their reactive chemistry. It is therefore important that measurements be made of the surface-atmosphere exchange of trace gases and associated aerosol compounds to quantify the emissions from – and deposition to - land used for agriculture, in order to quantify the impact of agricultural activities on the atmosphere and environment. Such studies can inform the development of abatement strategies and legislation designed to control emissions from agricultural activities. This also provides important process understanding to represent better the dry deposition processes in chemistry and transport models used to predict air quality and climate change.

Of particular importance to the surface-atmosphere exchange over agricultural land is the formation of atmospheric reactive nitrogen (N_r) compounds. These include the gases ammonia (NH_3), and nitrous acid (HONO), the latter of which, together with nitric acid (HNO_3), also derives from the oxidation of nitrogen oxides (NO_x) emitted by combustion sources. NH_3 and HNO_3, and their aerosol equivalents NH_4^+ and NO_3^-, are the primary contributors to atmospheric reactive nitrogen (N_r) dry deposition (Andersen and Hovmand, 1999). The majority of NH_3 emissions originate from agricultural sources, either from direct point sources from the application of N-containing fertilisers, or from long-term sources from livestock (Behera et al., 2013). The use of urea as a fertiliser is associated with particularly large losses of NH_3 after application, due to the action of the urease enzyme present in soil, which leads to NH_3 volatilization (Suter et al., 2013). Ferm et al. (1998) estimate that fertiliser losses as NH_3 average 14% of the N applied. Nitrogen losses from animal waste present on grassland used for sheep grazing has also been observed (Cowan et al., 2015). While NH_3 is predominantly deposited close to source, resulting NH_4^+ aerosol can be transported over large distances.

HNO_3 is primarily formed from the oxidation of nitrogen oxides (NO_x), which are principally anthropogenic in origin but also have a soil biogenic origin (Pilegaard, 2013). HNO_3 is extremely water soluble and is rapidly removed from the atmosphere through deposition or by gas-particle interactions, leading to a high deposition velocity. The gas-phase equilibrium reaction of HNO_3 with NH_3, which is dependent upon temperature and relative humidity (Mozurkewich, 1993, Robertson et al., 2013), gives rise to ammonium nitrate (R1).

$$NH_3 + HNO_3 \rightleftharpoons NH_4NO_3 \tag{R1}$$

The associated condensed-phase components of ammonium (NH_4^+) and nitrate (NO_3^-) exist in equilibrium (as ammonium nitrate (NH_4NO_3)) with NH_3 and HNO_3. Higher temperatures and humidity favour the decomposition of NH_4NO_3. The emission of these N_r species and their subsequent deposition by washout (wet deposition) or uptake on the surface (dry deposition)

have high spatial and temporal variability, and can have critical impacts on terrestrial and aquatic ecosystems, especially those which are nitrogen limited (Galloway et al., 2003; Fowler et al., 2013).

The interaction of NH_3 with HNO_3 can also lead to over estimation of the HNO_3 deposition rate, as the additional sink for HNO_3 deposition provided for by the reaction violates the theoretical deposition rate modelled on a zero surface resistance model for HNO_3. The dissociation of NH_4NO_3 over vegetation can induce an opposite effect, with apparent emissions of HNO_3 occurring with associated high deposition rates for NO_3^- and NH_4^+ (Nemitz and Sutton, 2004). The sums of the total ammonium (tot-NH_4^+ = NH_3 + NH_4^+) and of total nitrate (tot-NO_3^- = HNO_3 + NO_3^-), however, are conservative quantities (Kramm and Dlugi, 1994), and the use of them in the measurement of exchange fluxes can help to account for the NH_3-HNO_3-NH_4NO_3 triad on overall deposition rates.

HONO is similar to HNO_3 in that it can derive from oxidation of NO_x precursors. Although it can be formed homogeneously in the atmosphere by the reaction of the hydroxyl radical OH with NO (R2) (Pagsberg et al., 1997), the rate of this reaction is too slow to account for measured concentrations of HONO. Similarly, the heterogeneous reaction involving the reaction of NO_2 with H_2O on terrestrial surfaces, while potentially a contributory source to atmospheric HONO, has also been found to be too slow to account for measured concentrations (Kleffman, 2007). HONO is photolized during daytime, being a primary source of OH-radicals depending on the source and sink mechanisms that govern its abundance (Sörgel et al., 2015). However, a growing number of field measurements of non-zero HONO concentrations during the day points to the presence of daytime sources (Acker et al., 2006), including the emissions of HONO from soils (Su et al., 2011; Oswald et al., 2013; Scharko et al., 2015).

$$NO + OH \longrightarrow HONO \tag{R2}$$

As the primary basic gas in the atmosphere, NH_3 also reacts with trace acidic gases, such as hydrogen chloride (HCl) and sulfuric acid (H_2SO_4). The products of these reactions give rise to the aerosols ammonium chloride (NH_4Cl) and ammonium sulfate ((NH_4)$_2SO_4$), which along with NH_4NO_3 act as scattering aerosols that alter the Earth's total albedo and contribute significantly to regional and global climate (Fiore et al., 2015). The ammonium salts make a significant contribution to inhalable particulate matter (PM) associated with human health impacts, with NH_4NO_3 often dominating PM pollution events in northern Europe (Vieno et al., 2014). Ammonium sulfate is particularly long lived, and its transport and subsequent deposition to surfaces such as agricultural soils can affect plant health (van der Eerden et al., 1992) and lower soil pH (Elliott et al., 2008).

The dry deposition of the acidic gases themselves can also induce soil acidification, which on agricultural soils can limit the growth of crops through perturbing the uptake of nutrients. HCl, like HNO_3, is highly water soluble, is deposited quickly to the surface, and consequently has a high deposition velocity. It can be formed by the reaction of other acidic gases, such as

HNO$_3$ and SO$_2$, with sodium chloride found in sea spray (Pio and Harrison, 1983) SO$_2$, which is the precursor for H$_2$SO$_4$ in the atmosphere, is primarily anthropogenic in origin, being emitted via the burning of fossil fuels that contain sulfur.

Measurements of trace gases and associated aerosols are, however, restricted by the availability of appropriate instrumentation,
5 complications in their measurement due to their reactivity and water solubility, as well as the potential interference of gas-particle interactions.

Techniques to measure concentrations and fluxes of these trace gas and associated aerosol components require multispecies quantification, low detection limits and fast temporal resolution. Eddy covariance, the most direct micrometeorological tech-
10 nique for the measurement of trace gas fluxes, requires fast-response sensors that are not available for some species (such as HNO$_3$) or are limited by the time-response and potential for chemical interferences of the inlet (Neuman et al., 1999). While eddy covariance has been used to measure NH$_3$ concentrations using laser absorption spectroscopy, such as through the use of quantum cascade lasers (QCL) (Famulari et al., 2004, Zöll et al., 2016), intercomparisons with more established techniques are still lacking.

The aerodynamic gradient method derives fluxes of a tracer from its vertical concentration gradient, which can be obtained from concentration measurements at two or more heights, avoiding the requirement for fast response measurement. Developments in automated wet chemistry instrumentation have in turn led to the development of the Gradient of Aerosols and Gases Online Registrator (GRAEGOR), a two-point gradient system that measures the concentrations of HCl, HONO, HNO$_3$, SO$_2$
20 and NH$_3$, and their associated aerosol counterparts Cl$^-$, NO$_2^-$, NO$_3^-$, SO$_4^{2-}$ and NH$_4^+$ (Thomas et al., 2009). One of the advantages of the modified aerodynamic gradient method is the ability to determine the deposition velocities (V_d) of chemical tracers, provided the flux and concentration at a reference height have been calculated. With the use of the GRAEGOR, which takes measurements of tracers at two heights over one hour, high-resolution time scale measurements of deposition velocities can be acquired.

Other wet chemistry instruments have also been developed to measure individual species at one height, such as the Long Path Absorption Photometer (LOPAP), which measures concentrations of HONO with fewer artefacts than the GRAEGOR (Heland et al., 2001). A comparison study between LOPAP HONO measurements and the Gas and Aerosol Collector (GAC) - an instrument which uses similar measurement techniques to the GRAEGOR – was conducted by Dong et al. (2012), but there
30 has not yet been a published comparison between the LOPAP and GRAEGOR in measurements of HONO. Similarly, measurements of trace gases and aerosols above agricultural grassland using the GRAEGOR are limited, and previous studies above these land systems have been restricted to measurements of a limited number of species within a limited particle size range.

The aim of this study was to use the GRAEGOR to measure concentrations and fluxes of the trace gases HCl, HONO, HNO$_3$, SO$_2$ and NH$_3$ and their water-soluble aerosol counterparts Cl$^-$, NO$_2^-$, NO$_3^-$, SO$_4^{2-}$ and NH$_4^+$ over agricultural grassland in Scotland during a period in early summer (May–June 2016) that included a fertilisation event using urea pellets. The possible formation of NH$_4$NO$_3$ post fertilisation, a link between aerosol deposition velocity and size (specifically, a proxy for size based on the PM$_{2.5}$/PM$_{10}$ ratio from measurements nearby), and the potential ground source formation of HONO are discussed. A further aim of this study was to undertake intercomparisons between the measurements of HONO by the GRAEGOR and two LOPAP instruments, and between measurements of NH$_3$ recorded by a parallel quantum cascade laser eddy covariance system.

2 Methodology

2.1 Easter Bush Site Description

The campaign was conducted during the late spring/summer 2016 (21st May – 24th June) at the Easter Bush measurement site (3°12'W, 55°52'N, 190 m above sea level), located 10 km south of Edinburgh, UK. Measurements were made at a 3 m tower situated on the boundary of two intensively managed grassland fields (hereafter referred to as North and South Field) of 16 ha total area, composed principally of *Lolium perenne* (perennial Rye grass) (Fig. 1). Due to the presence of the Pentland Hills close by to the west, local wind direction is channelled such that SW winds – the predominant wind direction at the site – yield flux footprints over the South field, while NE winds produces flux footprints over the North field.

Both fields are used for year-round (although not continuous) sheep grazing, in rotation with adjacent fields, but the South Field also typically has an annual cutting for silage. Mineral fertilisation is carried out twice a year on both fields. During this study, fertilisation of the two fields occurred between 08:00 – 09:00 on the 13th June, using urea mineral fertiliser at a rate of 69.9 kg N ha^{-1}. In preparation for this application, sheep that had been present in the fields since April were removed from the South Field on the 2nd June and removed from the North Field on the 9th June. Sheep were reintroduced to the North Field on the 21st June.

Over the years the Easter Bush field site has hosted several long-term measurements of CO$_2$, CH$_4$ and NO$_2$, and has participated in a number of international projects, such as GRAMINAE (GRassland AMmonia INteractions Across Europe) (Sutton et al., 2009), Greengrass (Soussana et al., 2007) and NitroEurope (Sutton et al., 2007). It has also supported several individual campaigns of trace gas measurements (Di Marco et al., 2004; Famulari et al., 2004; Jones et al., 2017). In particular, fluxes of NH$_3$ were measured over an 18-month period (Milford et al., 2000) and the GRAEGOR was operated during a period of manure application (Twigg et al., 2011).

2.2 Instrumentation

2.2.1 Gradient of Aerosols and Gases Online Registrator

The GRAEGOR (Energy Research Centre of the Netherlands) is a wet chemistry instrument that measures the concentrations of reactive trace gases (HCl, HONO, HNO_3, SO_2 and NH_3) and water-soluble aerosols (Cl^-, NO_2^-, NO_3^-, SO_4^{2-}, NH_4^+) contin-

5 uously, semi-autonomously, and with online analysis at hourly resolution (Thomas et al., 2009; Wolff et al., 2010). The instrument consists of two sampling boxes placed at two heights (during this campaign, $z_1 = 0.6$ m, $z_2 = 2.4$ m), from which concentration gradients and hence fluxes can be derived.

Each sample box contains a horizontal wet rotating annular denuder (WRD) (Keuken, 1988) and a steam jet aerosol collector

10 (SJAC) (Khylstov et al., 1995; Slanina et al., 2001) connected in series. Air is drawn through each sample box simultaneously by an air pump at a rate of 16.7 L min^{-1}, passing first through the WRD, which is continuously coated with a feeding solution of double-deionized water (DDI) of 18.2 MΩ resistance. Trace gases within the laminar air flow are absorbed into the sorption solution which is then fed from the sample box to a detection unit located at ground level. The trace-gas-free air then passes through the SJAC, where particles within the air flow are mixed with steam generated from the DDI water feeding solution,

15 precipitating a supersaturation event causing the water–soluble particles to grow into droplets. The enlarged droplets are separated out of the air stream by a cyclone and fed as a liquid sample to the detection unit. Liquid samples from the SJAC and WRD of each sample box are analysed for NH_3/NH_4^+ using flow injection analysis (FIA) (Wyers, 1993; Norman et al., 2009). An ion chromatography (IC) unit equipped with a Dionex AS12 column, quantifies the concentration of $HONO/NO_2^-$, HNO_3/NO_3^- and SO_2/SO_4^{2-} based on the measured conductivity of the respective anions within the liquid sample compared to

20 a reference standard of 50 ppb Br^- added to the sample solution. Analysis by FIA and IC is carried out over 15 minutes, and using a flow control scheme, a half-hourly averaged concentration of trace gases and water-soluble aerosols is generated for each height every hour.

A high-density polyethylene tube (0.3 m length, and 1/3" outer diameter) with a HDPE filter is placed at the inlet of the WRD

25 in order to minimise the loss of HNO_3 and NH_3 and to ensure a particle diameter cutoff of 0.2 nm. A biocide of 0.6 mL of hydrogen peroxide (30%) is added to every 1 L of the DDI water feeding solution to prevent biological contamination in the WRD of each sample box. Air flow is controlled using a critical orifice downstream of the SJAC.

Autonomous calibration of the FIA system was carried out 24 h after the beginning of the campaign, and every 72 h thereafter,

30 giving a total of 5 internal calibrations of this system. Calibration was conducted using three liquid NH_4^+ standards of 0, 50 and 500 ppb concentration. The IC unit is continuously checked for analytical performance by the addition of a liquid Br^- internal standard (50 ppb concentration) to each column injection. Calibration of the IC unit was conducted twice during the

campaign (23rd May and the 28th June, prior to and after the campaign respectively) using a mixed ionic liquid standard consisting of 25 ppb SO_4^{2-}, 20 ppb NO_3^- and 20 ppb Cl^-.

Measurements of the air flow into the sample boxes were conducted using an independent device (TSI Mass Flowmeter 4140) once every fortnight during the campaign. Additional checks of the field performance of the instrument included daily checks of the WRD tubes and sample box air inlets for signs of visible contamination.

The GRAEGOR sampling boxes have very short inlets with no size-selection. Consequently, the aerosol concentration reflects water-soluble total suspended particulate (TSP). It detects any compound that dissociates to form the measured ions and therefore has a number of artefacts. These include interferences in HONO measurements through NO_2, particularly during periods of high SO_2 concentrations (Spindler et al., 2003); the inclusion of dinitrogen pentoxide (N_2O_5) concentrations in measurements of HNO_3 during the night-time measurement periods, though the magnitude of this unclear in rural environments (Phillips et al., 2013); and the potential for organic chloride compounds to be included in measurements of overall Cl^- aerosol (Nemitz et al., 2000a).

The GRAEGOR has been demonstrated to be capable of measuring fluxes in a number of studies both in identical form to the one used here (Wolff et al., 2010, 2011; Twigg et al., 2011) or in related variants (Nemitz et al, 2004; Rumsey and Walker, 2016). Ammonia-specific instruments based on the same technology (AMANDA, GRAHAM, ECN Petten, NL; Wyers et al., 1993) represent the most commonly used instrument for the automated measurement of ammonia fluxes.

2.2.2 Supplementary Measurements

Vertical profiles of temperature were measured at the tower using fine-thread, custom-made thermocouples set at the same heights as the GRAEGOR sample boxes. Located 0.4 m from the tower, an eddy covariance system (Gill Anemometer R01012 with LI-COR-7000) at a height of 2.6 m measured three-dimensional wind speed, sensible heat flux (H), frictional velocity (u_*) and wind direction. Ongoing, long-term measurements of relative humidity (RH) (Vaisla 50/Y humitter), global radiation (Skye Instruments SKS 110 pyranometer), and total rainfall (Campbell Scientific ARG110 tipping bucket rain gauge) were also available at the site for the campaign period. Measurements of HONO taken by a LOPAP (QUMA Electronik &Analytik, Wuppertal, Germany) and NH_3 measurements taken by a Quantum Cascade laser (QCL) (Aerodyne Research Inc., Billerica, USA) during the campaign period were used for comparison studies with GRAEGOR measurements. Measurements of NO_2 concertation, used in Section 4.4.1 to quantify an artefact in GRAGOR HONO measurements, were recorded by a chemiluminescence NO_2 detector (200E, Teledyne API, San Diego, California, USA) located 300 m south-east of the Easter Bush site.

2.3 Micrometeorological Theory

2.3.1 Aerodynamic Gradient Method

The aerodynamic gradient method (AGM), based upon flux-gradient similarity theory, calculates the flux of a tracer (χ, such as a gas or aerosol species) based on its vertical concentration gradient coupled with turbulence parameters (Fokken, 2008). In this paper a hybrid version of the AGM is used, in which the flux is calculated as (Flechard, 1998):

$$F_\chi = -u * \kappa \frac{\frac{\partial \chi}{}}{ln\left(\frac{z_2 - d}{z_1 - d}\right) - \psi_H\left(\frac{z_2 - d}{L}\right) + \psi_H\left(\frac{z_1 - d}{L}\right)} \tag{1}$$

where the friction velocity (u_*) is derived from eddy-covariance measurements with a sonic anemometer; κ is the von Karman constant ($\kappa = 0.41$); z_2 and z_1 are the heights of the sample boxes; d is the displacement height; and ζ is a dimensionless stability parameter expressing the ratio $(z-d)/L$, where L is the Obukhov length, a measure for atmospheric stability. The parameter Ψ_H, an integrated form of the heat stability correction term, accounts for deviations from the log-linear profile under non-neutral stratification. By convention, negative and positive flux values denote deposition and emission, respectively.

2.3.2 Choice of displacement height, *d*, value

A temperature gradient profile for the campaign was derived from measurements of air temperature at the two heights at which concentrations were measured (0.6 m and 2.4 m). Sensible heat flux (*H*) was calculated from the temperature gradient as per Wang and Bras (1998):

$$H = -\rho_{air}\, c_p\, K_H(z - d)\frac{dT}{dz}, \tag{2}$$

where c_p is the heat capacity of air, ρ_{air} is the density of air, and K_H is the eddy diffusivity constant for heat. K_H can be calculated as

$$K_H(z - d) = \frac{(z-d)u_*}{\phi_H\left(\frac{z-d}{L}\right)}, \tag{3}$$

where z is the absolute height above ground, d is the displacement height, u_* is the friction velocity, and Φ_H is the stability correction for sensible heat. Sensible heat flux and, by extension, the flux of the trace gas and aerosol species, are dependent upon the value of d. In order to ensure that the correct displacement height was chosen, the sensible heat flux based upon the temperature gradient developed from thermocouple measurements was calculated using a variety of different values for displacement height. The resulting values for the sensible heat flux were then compared through linear regression to the value for the sensible heat flux recorded by the eddy covariance system also present. A displacement height value of 0.14 m gave the closest agreement between the sensible heat fluxes derived by the aerodynamic gradient approach and eddy-covariance, with a linear regression slope of 0.997 and $R^2 = 0.945$.

2.3.3 Determination of Dry Deposition Velocities

The dry deposition velocity (V_d) of a tracer is the negative ratio of its flux to its concentration (χ) at height z – d

$$V_d(z-d) = -\frac{F_\chi}{\chi_{z(z-d)}}, \tag{4}$$

The V_d for gas species may also be expressed as the reciprocal of the total resistance for deposition, which is composed of R_a
5   (the aerodynamic resistance), R_b (the quasi-laminar boundary layer resistance) and R_c (the canopy resistance) as per the resistance analogy for dry deposition (Fowler and Unsworth, 1979; Wesley, 1989). R_a and R_b were calculated from (5) and (6) using meteorological measurements taken at the site using (Garland, 1977)

$$R_a(z-d) = \frac{u(z-d)}{u_*{}^2} - \frac{\psi_H(\zeta)-\psi_M(\zeta)}{\kappa u_*}, \tag{5}$$

$$R_b = (Bu_*)^{-1}, \tag{6}$$

10   where B^{-1}, B being the Stanton number, is parametrized by the turbulent Reynold's number, R_{e*} (the ratio of the frictional force to the kinematic velocity of air) and the Schmidt number, S_e, (the ratio of kinematic velocity of air to the molecular diffusivity coefficient of the gas species) :

$$B^{-1} = 1.45 R_{e_*}{}^{0.24} S_c{}^{0.8}, \tag{7}$$

If R_a and R_b are calculated from measurements, R_c can be inferred via:

15   $$R_c = \frac{1}{V_d(z-d)} - R_a(z-d) - R_b, \tag{8}$$

For gases, a theoretical maximum deposition velocity can be calculated when it is assumed that the gas is completely absorbed by the canopy (i.e. for $R_c = 0$):

$$V_{max} = \frac{1}{(R_a+R_b)}, \tag{9}$$

The canopy resistance approach can only describe deposition and fails when the exchange of a gas is bi-directional, such as
20   often the case with NH_3. In this case, the canopy compensation point model can be adopted, which considers the surface interaction of NH_3 in terms of parallel resistance pathways, composed of individual resistances such as stomatal resistance and cuticular resistance (Nemitz et al., 2000b; Flechard et al., 1999).

The gradient technique is only applicable for inert species whose flux is constant with height. Most studies of surface-exchange
25   fluxes of reactive compounds do not have the information to assess whether chemical reactions might interfere with the flux measurement, but in this study the behaviour of HNO_3 and HCl allows us to draw conclusions on flux divergence (Section 4.2.1). Following precedence in the literature (e.g. Nemitz et al., 2004) we initially evaluate fluxes assuming that chemistry can be ignored, and then discuss the validity of this discussion based on the results.

2.3.4 Limits of detection and estimation of uncertainties in concentration measurements and flux calculations

The concentration limit of detection (LOD) of the instrument for each of the species measured was quantified from a field blank test. The field blank test was carried out prior to the campaign on the 20th March over 24 hours by switching off the sample box air pump and sealing the air inlets, but leaving the rest of the system unaltered, as per Thomas (2009). Limits of detection were then calculated as three standard deviations from the average background signal. Results from this test are presented in Table 1, expressed as LOD values for each trace gas and corresponding water–soluble aerosol species.

The minimum detectable flux for each aerosol and gas species measured by the GRAEGOR is dependent upon atmospheric stability and the ambient concentration of the given trace gas or aerosol species. Based on the method described by Thomas et al. (2009), median minimum detectable fluxes (F_{LOD}) were calculated for each trace gas and aerosol species measured and are detailed in Tables 2 and 3 respectively.

When calculating the flux of a species using the aerodynamic gradient method, it is apparent that errors in individual concentration measurements propagate into an error in the concentration differences, and subsequently, affect the accuracy of the calculated vertical concentration gradient. Some errors systematically affect both heights and therefore affect the gradient to a lesser extent than systemic errors in sampling efficiency at a single height, such as difference in capture efficiency of the WRD tubes or slight differences in air flow caused by differences in the critical orifices, may impact on the accuracy of concentration measurements and resultantly affect the precision in the error of concentration difference.

The overall random error in the measurements of the trace gas and water–soluble aerosol concentrations (σ_m) can be determined using a Gaussian error propagation approach, in which the concentration error is expressed as a product of several individual measurement errors with the mixing ratio, m (Trebs et al., 2004) (eq. 9) –

$$\sigma_m = m \sqrt{(\frac{\sigma_{m_{liq}}}{m_{liq}})^2 + (\frac{\sigma_{Br_{(std)}}}{Br_{(std)}})^2 + (\frac{\sigma_{Q_{Br}}}{Q_{Br}})^2 + (\frac{\sigma_{m_{Br}}}{m_{Br}})^2 + (\frac{\sigma_{Q_{air}}}{Q_{air}})^2} \qquad (10)$$

Here, m_{liq} is the mixing ratio of the compounds found in the analysed liquid sample in ppb, $Br_{(std)}$ the stated mixing ratio of the internal Br^- standard, Q_{Br} the flow rate of the internal Br^- standard, m_{Br} the analysed Br^- mixing ratio and Q_{air} the air mass flow through the system. All values have an associated standard deviation, σ_x. This formulation holds strictly for the species measured by ion chromatography; for NH_3 and NH_4^+, the equation is altered by omitting the factor relating to Br^- addition and substituting the factor for Q_{Br} and its associated standard deviation with the term Q_S, the flow of the analysed liquid sample of NH_3 or NH_4^+.

Uncertainties for the trace gases and water–soluble aerosols measured calculated by error propagation ranged from 8% - 18% (3σ) throughout the campaign, varying primarily due to fluctuations in the measured flow rate and analysed concentration of the internal Br⁻ standard.

5   The error in the concentration difference ($\sigma_{\Delta c}$) can be characterised experimentally, by placing both sample boxes can at one height, or – provided that the absolute difference between sample heights is small – by using one common air inlet at a specified height, with the instrument operated normally. From this side-by-side measurement, linear regression analysis accompanied by orthogonal best of fit between the concentrations measured by each sample box can be conducted, with deviation from a 1:1 fit between sample heights defined as a systemic error. Using the calculated orthogonal fit equation, corrections in the

10   concentrations can then be applied, accounting for the systemic bias (Wolff et al., 2010b). After correction using the orthogonal fit, the remaining scatter – termed the residuals – was used to determine the error in the concentration difference. During this campaign, one side-by-side measurement was conducted on the 8th June for 16 hours by connecting a common air inlet set at $z = 1.2$ m between each sample box. From the results obtained, it was found that for the gases NH_3, HCl, $HONO$, HNO_3 and SO_2 that deviation from the 1:1 fit resulted in a precision of measurements <4% (3σ). For the aerosol species Cl^-, NO_3^- and

15   SO_4^{2-}, precision was calculated as <8% (3σ), while for NH_4^+ was calculated as <9% (3σ).

Errors in flux calculations can similarly be determined through the Gaussian error propagation method applied to Eq. (1). Wolff et al. (2010b), using an analogous form of this equation, showed that total error in the flux is composed of ($\sigma_{\Delta c}$) and the error in the flux-gradient relationship (expressed as a transport velocity by Wolff et al., 2010b), which is dominated by the

20   error in u_* (σ_{u*}).

$$\sigma_F = F \sqrt{\left(\frac{\sigma_{vu_*}}{u_*}\right)^2 + \left(\frac{\sigma_{\Delta c}}{\Delta_c}\right)^2} \qquad (11)$$

This simplification neglects the detailed secondary errors associated with the stability correction which to quantify fully is

25   beyond the scope of this paper.

σ_{u*} is dependent upon the sonic anemometer used and whether conditions are neutral or non-neutral (Foken, 2008; Nemitz et al., 2009a). For neutral conditions, and based on the sonic anemometer used, σ_{u*} was estimated at $\leq 10\%$. For non-neutral conditions, σ_{u*} was estimated at 12% median, which, in combination with $\sigma_{\Delta c}$, was used to calculated σ_F.

Throughout this paper, stated errors for concentration measurements are derived from the measurement uncertainty as calculated by Eq. (10), while stated errors for flux calculations are derived from the flux uncertainty as calculated by Eq. (11). Calculated errors for the uncertainty in concentration measurements, the error in the concentration difference, and the error in

the calculated fluxes for all species measured are similar to values determined by previous studies which have used the GRAEGOR successfully to measure flux gradients (Thomas et al., 2009; Wolff et al., 2010; Twigg et al., 2011).

2.3.5 Data Post Processing

Concentrations that were less than five times the limit of detection as calculated before the campaign began (20th March) were discarded. Calculated fluxes were filtered according to a standard protocol. Fluxes were not calculated for periods of low wind speed ($u < 1$ m s^{-1}), low friction velocity ($u < 0.15$ m s^{-1}), and very stable conditions as indicated by the Obukhov length absolute value ($|L| < 5$ m). Fluxes were also discarded for periods when the wind was obstructed by the measurement cabin and other towers ($270° >$ wd $< 320°$, and $120° >$ wd $< 160°$). Calculated fluxes which were below the minimum detectable flux value for their respective trace gas or aerosol species were discarded.

3 Results

3.1 Meteorology

Figure 2 shows time series of the rainfall, radiation, relative humidity, air temperature and wind speed and direction measured during the campaign. The meteorology splits into two episodes. From 24th May to 5th June 2016, the dominant prevailing wind direction was north easterly, accompanied by dry and sunny conditions with air temperature displaying a characteristic diel cycle that increased each day. Following a period of cloudier conditions from 6th to 10th June, the prevailing wind direction shifted to south westerly for the remainder of the measurement period. Conditions became wetter and the diel air temperature amplitude was reduced. Relative humidity remained high throughout the campaign, with only occasional periods <70%, such as 3rd – 4th June and the 21st – 23rd June. Wind speed was variable throughout, ranging between 0.05 and 5.87 m s^{-1}, with a median value of 2.16 m s^{-1}. During the fertilisation period, the prevailing wind direction was from the SW, and therefore over the South Field, with no precipitation but high (>90%) relative humidity.

3.2 Concentrations of trace gases and water-soluble aerosols

Summary statistics for the concentrations of the trace gas and water-soluble aerosol species measured at 2.4 m during the campaign are presented in Table 1. Median values for the concentrations of water-soluble aerosol species were similar to those measured in PM$_{10}$ at the nearby rural background monitoring site of Auchencorth Moss (Twigg et al., 2015). The time series of the measured aerosol and trace gas concentrations are displayed in Figures 3 and 4, respectively. Data gaps in the time series are due to in-field calibrations, poor chromatograms, or instability in liquid or air flow.

Mean concentrations of NO$_3^-$ were 1.53 µg m^{-3} (2.4 m), whereas its gaseous counterpart, HNO$_3$, had mean concentrations of 0.19 µg m^{-3} (2.4 m). The mean particulate NO$_3^-$ concentrations were therefore almost 6 times greater than the gaseous HNO$_3$ counterpart. The same dominance of particulate SO$_4^{2-}$ concentrations over gaseous SO$_2$ concentrations was also observed.

Median concentrations of particulate Cl^- were 0.37 µg m^{-3} and 0.36 µg m^{-3} at 0.6 m and 2.4 m, respectively. The mean concentrations of Cl^- were also similar at both heights at 0.89 µg m^{-3} and 0.91 µg m^{-3}, respectively. Variation in HCl concentrations at each height was more pronounced, with a mean value of 0.16 µg m^{-3} at 0.6 m and 0.20 µg m^{-3} at 2.4 m, and a median value of 0.12 µg m^{-3} at 0.6 m and 0.15 µg m^{-3} at 2.4 m. As for particulate NO_3^- and gaseous HNO_3, measured particulate Cl^- concentrations were greater than those of gaseous HCl, by about a factor of 2 at each height.

In contrast, NH_3 concentrations were larger than those of particulate NH_4^+; median concentrations of NH_3 were 1.15 µg m^{-3} (2.4 m), while median concentrations of NH_4^+ were 0.64 µg m^{-3} (2.4 m). The average concentrations of NH_3 were similar to those reported previously at the same site for the same time of year (Milford, 2004). Similarly, the average concentrations of HONO are higher than those of its particulate counterpart, NO_2^-, with median concentrations for HONO of 0.04 µg m^{-3} (2.4 m) and corresponding concentrations for NO_2^- 0.02 µg m^{-3} (2.4 m) respectively.

Maximum concentrations for NH_3 and HONO at 0.6 m were 21.4 µg m^{-3} and 0.15 µg m^{-3}. At 2.4 m, the maximum concentration for NH_3 and HONO were 13.8 µg m^{-3} and 0.12 µg m^{-3}. The maximum values at each height occurred at 11:00 on the 13th June for NH_3, one hour after fertilisation of the South Field, and at 13:00 on the 13th June for HONO, four hours after fertilisation of the South Field.

The time series of measurements presented in Figures 3 and 4 show that both aerosol and trace gas concentrations are affected by prevailing meteorological conditions, with larger concentrations for each species during the drier, warmer period of 28th May to 6th June, followed by decreased concentrations from 6th to 10th June when precipitation increased and temperature decreased. Concentrations were lower – except for the peaks in NH_3 and HONO after fertilisation on the 13th June - during the period from 10th June to the end of the campaign, concurrent with the change in prevailing wind direction from the NE to the SW.

The concentrations of HNO_3 and SO_2 showed a strong diel cycle (Figure 4) from the 26th May to the 9th June, with maxima at both measurement heights occurring between 11:00 and 14:00, and minima occurring at night between 03:00 and 06:00. A similar, but weaker, inverted pattern was exhibited by their particulate counterparts, with NO_3^- concentrations at both heights (Figure 3) having maxima between 02:00 and 04:00, and minima between 12:00 and 15:00.

Figure 5 shows the median diel concentrations of NH_3, HCl, HONO, HNO_3 and SO_2 at 2.4 m prior to fertilisation. The median concentrations of HONO remained above the detection limits of the instrument even during daytime, contrary to its expected photochemistry. While concentrations of HONO peaked during night-time and decreased during the day as incoming solar radiation increases, there remained a detectable concentration of HONO at both heights even for the measurement minima at

15:00. The median diel concentrations for HCl, HNO$_3$ and SO$_2$ show a shared pattern, with concentrations peaking during the day to reach a maximum between 11:00 to 14:00, followed by a decrease during the night, reaching minima between 02:00 and 04:00. The concentrations of NH$_3$ showed little variation across the day. Figure 6 shows the median diel concentrations of NH$_4^+$, Cl$^-$, NO$_3^-$ and SO$_4^{2-}$ at 2.4 m prior to fertilisation. The median diel concentrations of NH$_4^+$ reach a minimum at 16:00, with a maximum at 02:00. The concentrations of NO$_3^-$ show a similar pattern of early morning median maxima (04:00) and afternoon minima (13:00). The median diel SO$_4^{2-}$ concentrations had maxima at midnight and a minimum at 16:00. The Cl$^-$ concentrations reached a maximum at 03:00 and a minimum at 13:00; however, the upper quartile range was high across all hours, with the maximum concentration of 7.88 µg m^{-3} recorded at 03:00 (median at this time is 0.5 µg m^{-3}).

3.3 Fluxes, Deposition Velocities, and Canopy Resistance

3.3.1 Fluxes of trace gases

Figure 7 shows the time series of the fluxes for the traces gases measured during the campaign. Data gaps are due to either absent data points (unpaired concentrations), or periods where data were filtered (refer to Section 2.3.4).

Bi-directional fluxes were present for both NH$_3$ and HONO, with emission events for each gas occurring during the period of fertilisation of the South Field. For the other trace gases – HCl, HNO$_3$ and SO$_2$ – the flux was uni-directional, with deposition occurring throughout the campaign. The deposition for HCl, HNO$_3$ and SO$_2$ varied, with larger deposition fluxes occurring during the warmer, drier periods, particularly during the period 1st - 8th June, and smaller deposition fluxes close to zero during the colder, wetter period at the end of the campaign (15th - 24th June).

Summary statistics for the trace gas fluxes, deposition velocities, theoretical maximum deposition velocities, and canopy resistance values are presented in Table 2. The maximum NH$_3$ flux was +1460 ng m^{-2} s^{-1}, recorded at 12:00 on the 13th June, three hours after fertilisation. The mean flux for NH$_3$ was +15.24 ng m^{-2} s^{-1}, suggesting that emission was the predominant flux for NH$_3$ during this campaign. For all other gases, the mean flux values were negative, suggesting that deposition was the net flux process overall. However, a maximum flux for HONO of +4.92 ng m^{-2} s^{-1}, recorded five hours after fertilisation, highlights the bi-directional flux pattern for HONO during the campaign. The maximum HONO flux measured here was particularly large. Nitrous acid emissions have previously been reported post fertilisation of grassland using cattle slurry at the same field site ranging from +1.0 to +1.5 ng m^{-2} s^{-1} (Twigg et al. 2011). Table 2 also shows the median relative flux error, the typical flux detection limits (F_{LOD}) and the fraction of 60-minute flux values that exceed the F_{LOD} of that period, based on actual concentration and turbulence. It should be noted that the uncertainty of the campaign averages is much smaller as random uncertainty reduces with the square root of the number of observation that enter the calculation of the ensemble average (e.g. Langford et al., 2015).

Median diel cycles for the deposition velocity and calculated theoretical maximum deposition velocity for the trace gases HCl, HONO, HNO$_3$ and SO$_2$ are shown in Figure 8. The calculation of median diel values for trace gas deposition velocities and canopy resistances excludes the period of flux divergence which occurred during fertilisation. The diurnal deposition velocities for HCl and HNO$_3$ were very close to the calculated maximum deposition velocities, which is expected as a result of their reactivity and high water solubility. The deposition velocity for SO$_2$ is near the theoretical maximum during night–time but is lower during daytime. The deposition velocity for HONO was consistently lower than its theoretical maximum throughout the entire day. While median values for the V_d for HNO$_3$ are close to the values for V_{max}, deposition velocities were recorded that exceeded their corresponding theoretical maximum. While most exceedances fall within the uncertainty range of the measurement, a maximum deposition velocity of 56.8 mm s^{-1} was recorded at 14:00 on the 13th June, four hours after fertilisation.

3.3.2 Fluxes of water-soluble aerosol components

The measured surface fluxes of the aerosol species Cl$^-$, NO$_3^-$, SO$_4^{2-}$ and NH$_4^+$ are shown in Figure 9, as well as the summary statistics for the fluxes and deposition velocities in Table 3. A large data gap in NH$_4^+$ fluxes from 31st May to 10th June 2016 was due to NH$_4^+$ only being measured at one height on account of unreliable data for NH$_4^+$ at the lower height of 0.6 m.

Pre-fertilisation, all aerosol species exhibited deposition fluxes. The deposition fluxes were larger during the drier, warmer period from 31st May to 6th June, and close to zero during the wetter conditions at the end of the campaign. An important exception was the emission of NH$_4^+$ and NO$_3^-$ from 13:00 on the 13th June to 02:00 on the 14th June, starting 4 hours after fertilisation of the South Field.

Summary statistics for the fluxes and deposition velocities for the aerosol species measured are shown in Table 3. As for the trace gases, the median deposition velocities for the aerosol species excludes the period of flux divergence which occurred during fertilisation. The maximum flux for NH$_4^+$ of +18.16 ng m^{-2} s^{-1} was recorded at 16:00 on the 13th June, seven hours after fertilisation of the South Field. Similarly, the maximum flux for NO$_3^-$ (+31.84 ng m^{-2} s^{-1}) was also recorded soon after fertilisation, at 18:00 on the 13th June. Overall, however, the mean fluxes for all aerosol species were negative, confirming a predominant net deposition to the surface.

3.4 HONO and NH$_3$ GRAEGOR measurement comparisons with LOPAP and QCL

3.4.1 HONO Comparison Study between GRAEGOR and LOPAP

A comparison of HONO measurements from the GRAEGOR and two LOPAP instruments was conducted from the 26th May to the 6th June to investigate the potential artefacts in the WRD method used by the GRAEGOR. The LOPAPs were part of a study to investigate the mechanisms controlling HONO fluxes over managed grassland, including investigating the potential ground sources of HONO, details of which are presented in Di Marco *et al.* (in preparation). A series of simple linear regression

analyses was conducted to determine the level of agreement between the concentrations of HONO measured by each sample box of the GRAEGOR and each of the LOPAPs. The two LOPAP instruments were operated at the two heights of 0.6 m and 2.0 m (hereafter referred to as LOPAP (0.6 m) and LOPAP (2.0m) respectively). In all comparisons, the GRAEGOR recorded a higher concentration of HONO than either of the LOPAPs. The linear regressions suggest that there is a consistent offset in all GRAEGOR concentrations, varying between 0.01 µg m^{-3} and 0.02 µg m^{-3}. In comparisons between the GRAEGOR Sample Box 1 at 0.6 m and both LOPAPs, the linear concentration relation for HONO varies from 0.92 to 0.97. The comparisons between the GRAEGOR Sample Box 2 (2.4 m) and the LOPAPs suggest that the linear concentration relation for HONO is 1.06 and 1.01 for LOPAP (2.0 m) and LOPAP (0.6m) respectively. In all comparisons, however, there exists a constant concentration offset, which results in a constant higher concentration recorded by both GRAEGOR sample boxes. The closest agreement is between GRAEGOR Sample Box 2 (set at height 2.4 m) and LOPAP (2.0 m), where the HONO concentration recorded by the GRAEGOR Sample Box 2 (2.4 m) is 1.06 that of LOPAP(2.0 m). This comparison also has the best statistical agreement, with an R^2 value of 0.67, suggesting a reasonable agreement between the GRAEGOR Sample Box 2 and LOPAP (2.0 m) measurements.

3.4.2 NH$_3$ Comparison Study between GRAEGOR and QCL

On the 7th June, a QCL with inlet at height 1.6 m was installed at the Easter Bush site and took measurements of NH$_3$ from 19th June to 7th August. Three days of concurrent NH$_3$ measurements taken by the GRAEGOR and the QCL were recorded in the period 21st – 24th June. The time series of the NH$_3$ measurements by each instrument are shown in Figure 10(a). An averaged NH$_3$ concentration at 1.0 m (χ (1 m)) taken by the GRAEGOR was compared with the NH$_3$ concentrations taken by the QCL in a simple linear regression analysis, displayed in Figure 10(b). The linear regression shows that the GRAEGOR recorded a factor 1.22 higher concentrations of NH$_3$ than the QCL, with an associated R^2 value of 0.76. However, the number of concurrent measurements is small, with only 41 shared hourly measurement values across three days and a period of 19 continuous hours missing between 02:00 and 23:00 of the 23rd June.

4 Discussion

4.1 Ion Balance

The ion balance for the hourly-measured cation (NH$_4^+$) and anion (NO$_3^-$ and SO$_4^{2-}$) aerosol species pre-fertilisation is shown in Figure 11. Values are shown as molar equivalent concentration, derived from aerosol mass concentrations converted to molar concentrations and subsequently multiplied by their charge. Cl$^-$ charge was not included, under the assumption that it would be entirely associated, in the form of sea salt, with Na$^+$ which was not measured by the GRAEGOR. While the correlation between cation and anion species is very good ($R^2 = 0.71$), the linear regression suggests a deficit of NH$_4^+$, suggesting that some of the NO$_3^-$ and/or SO$_4^{2-}$ was balanced by ions other than NH$_4^+$. A likely candidate is Na$^+$: some of the SO$_4^{2-}$ is likely to

have represented sea-salt SO_4^{2-} and some $NaNO_3$ is formed by reaction of NaCl with HNO_3. Figure 11 is coloured by Cl^- concentration, and periods of anion excess tend to be associated with elevated Cl^- concentrations.

The formation of $NaNO_3$ through the reaction of HNO_3 or NO_x with sea salt has been previously observed in coastal sites (Andreae et al., 1999, 2000; Bardouki et al., 2003; Dasgupta et al., 2007)(Kutsuna & Ibusuki, 1994), and within the UK and Ireland, where the interaction with marine air with polluted air masses at coastal sites was shown to significantly shift the aerosol NO_3^- to the coarse mode (Yeatman et al., 2001; Twigg et al., 2015). Scavenging of atmospheric H_2SO_4, formed from SO_2 (O'Dowd and de Leeuw., 2007), by sea salt may also be occurring, which would also shift some of the SO_4^{2-} from the fine to the coarse mode.

4.2 Deposition velocities and fluxes of water-soluble aerosol and trace gas species

4.2.1 Fluxes of water-soluble aerosols and trace gases

Fluxes of SO_4^{2-} and Cl^- throughout the campaign were unidirectional deposition. However, during the fertilisation period of the South Field, bidirectional fluxes of NH_4^+ and NO_3^- were observed. Prior to fertilisation these species were deposited to the site. An apparent emission flux of NO_3^- is consistent with the possibility of NH_4NO_3 formation above grassland suggested by the divergence of HNO_3 V_d from V_{max} (Nemitz et al., 2009b) in the presence of high concentrations of NH_3 near the surface. Concentrations of NH_3 peak at 21.4 µg m^{-3} on the 13th June, 11:00, which occurs three hours before peak HNO_3 V_d and 7 hours prior to the apparent peak in emissions of NO_3^- at 18:00.

Fluxes for the trace gases were bi-directional for NH_3 and HONO, with deposition for all other species measured. Emissions of NH_3 and HONO occurred throughout the campaign, with HONO emissions particularly present during the early morning. Both species reached peak emissions soon after fertilisation. Increases in atmospheric NH_3 concentration and emissions of NH_3 resulting from the application of solid urea fertiliser has been previously established (Akiyama et al., 2004; Sommer and Hutchings, 2001), with losses from volatilisation increased if the urea pellets are poorly mixed into the soil and if conditions are dry and warm. While conditions prior to the fertilisation event were cool, temperatures increased quickly throughout the day, peaking at 19.2 °C at 13:00, four hours after fertilisation. Volatilisation was likely exacerbated by the dry conditions throughout the 13th June. The increase in concentration and upward flux of NH_3 provides the source for the formation of NH_4NO_3 in the presence of HNO_3. The mechanisms of the HONO emission fluxes are not discussed here but can be found in Di Marco *et al. (*in preparation*).

4.2.2 Aerosol deposition velocities

Deposition velocities for NO_3^- reached a maximum value of 9.8 mm s^{-1} during daytime, and a minimum of 0.2 mm s^{-1} outside the period of apparent emission fluxes at night. A similar pattern was observed for sulfate, which reached a maximum value

of 9.5 mm s^{-1} during daytime and a minimum value, outside of apparent emission events, of 0.15 mm s^{-1} during night. Median V_d values for NO$_3^-$ and SO$_4^{2-}$ were 1.52 mm s^{-1} and 1.45 mm s^{-1}, respectively. For Cl$^-$, the median V_d was 3.14 mm s^{-1}. The deposition velocities for SO$_4^{2-}$ where larger than those previously observed and derived for accumulation mode particles from theoretical considerations (Petroff et al., 2008). For sulfate in the fine (<0.1 µm diameter) range, Allen et al. (1991) recorded

5   a mean value of 1 mm s^{-1} for deposition velocity over short grass, similar to observations made by Gallagher et al. (2002) who reported a mean value of 0.9 mm s^{-1}.

The dry deposition of particles can be modelled using a process-orientated approach, which describes the deposition velocity as a function of particle size based on removal mechanisms acting within the vegetation canopy, such as Brownian diffusion,

10   impaction, interception and sedimentation (Slinn and Slinn, 1980; Davidson et al., 1982; Slinn, 1982). The models predict that for particles >0.1 µm in diameter deposition velocity increases with increasing particle size. Vong et al. (2004) recorded deposition velocities of greater than 2 mm s^{-1} for PM$_{10}$ particles over grassland. If the sulfate and chloride were in particularly coarse particles, deposition velocities would potentially be skewed towards a higher deposition velocity.

15   Secondary ammonium compounds are typically found in the accumulation mode (0.1 to 1 µm), while seasalt is found in super-micron particles (Myhre et al., 2006). Although measurements of particle size were not made during this campaign, measurements of aerosol species (including Cl$^-$ and SO$_4^{2-}$) in the PM$_{2.5}$ and PM$_{10}$ size fractions were taken by a two-channel Monitor for Aerosols and Gases in Ambient Air (MARGA, Applikon B.V, The Netherlands) instrument located at Auchencorth Moss, 12 km south west of Easter Bush. Aerosol concentration data was taken from an online database of MARGA measurements

20   (DEFRA, 2018). Agreement between MARGA and GRAEGOR aerosol concentrations were excellent (with correlations for SO$_4^{2-}$ with $R^2 = 0.95$, and for Cl$^-$ with $R^2 = 0.91$ between MARGA PM$_{10}$ and GRAEGOR TSP). As proxy for a particle size measurement, the proportion of PM$_{2.5}$ to PM$_{10}$ was used, with a lower proportion of PM$_{2.5}$ indicating a greater proportion of coarse aerosol, and a corresponding larger deposition velocity based on process-orientated modelling. To a first–order approximation, particle deposition velocities scale with u_* (Pryor et al, 2008). Figure 12 shows plots of the normalised deposition

25   velocities (V_d / u_*) against the fraction of the PM$_{10}$ mass contained in PM$_{2.5}$ at Auchencorth Moss ($f_{PM2.5} = $ PM$_{2.5}$/PM$_{10}$) for nitrate (a), sulfate (b) and chloride (c).

While the dynamic range of $f_{PM2.5}$ varied between compounds, third-order polynomial curves consistently describe the relation between the proportion of PM$_{2.5}$ to overall PM and the normalised V_d for nitrate, sulfate and chloride, suggesting – in line with

30   Slinn (1982) – that deposition velocity increases strongly with increasing particle size above 0.1 µm particle diameter. However, the relationship – although statistically significant – shows significant variability, which may be due to measurement uncertainty, but might also reflect the additional effect of atmospheric stability on particle fluxes (e.g. Wesely et al., 1985; Petroff et al., 2008) or differences in the size distribution between the Auchencorth and Easter Bush measurement sites. It must

be stressed that the proportion of $PM_{2.5}$ to PM_{10} is a proxy measurement for particle size and can only differentiate the proportions of aerosol of diameter less than or greater than 2.5 µm.

By contrast, the median deposition velocity of 0.37 mm s^{-1} for NH_4^+ was much smaller and within the range of previous measurements of dry deposition velocities of accumulation mode particles to grassland. The average $f_{PM2.5}$ for NH_4^+ recorded was 96%, compared to 78% for NO_3^- and 86% for SO_4^{2-}, suggesting that virtually all of the NH_4^+ measured was contained in fine particles, within the measurement error. The average normalised deposition velocity (V_d/u_*) of NH_4^+ of 0.04 was in the range of the values for the other compounds evaluated at $f_{PM2.5} = 100\%$.

Thus, the relatively high deposition velocities for Cl^-, NO_3^- and SO_4^{2-} (compared with NH_4^+) are a result of some of these compounds being contained in coarse aerosol. This is consistent with the ion balance (Fig. 11), which suggests that some of these compounds are balanced by seasalt Na^+, which is found mostly in the coarse fraction.

It should be noted that the increase in V_d with increasing contribution of coarse aerosol only accounts for the size-dependence of the processes of impaction and interception. As a non-turbulent process, gravitational sedimentation is not reflected in micrometeorological flux measurements and the sedimentation velocity would need to be added to the deposition velocity derived here.

4.2.3 Trace gas deposition velocities

Median diel deposition velocities for HNO_3 and HCl closely matched the theoretical maximum deposition velocities within the uncertainty of the measurement (Fig. 8), which closely conforms to their expected physico-chemical behaviour. Both HNO_3 and HCl are reactive and highly water soluble, and consequently it is expected that their deposition velocities should equal the theoretical maximum, and that the canopy resistance for these species should be equal to zero (Dollard et al., 1987; Muller et al., 1993). The close agreement between V_d and V_{max} during most of the campaign suggests that chemistry is not affecting the fluxes and that the standard aerodynamic gradient method is therefore applicable. However, significant deviations of the calculated deposition velocity from the theoretical maximum for HNO_3 exist: R_c values for HNO_3 were particularly large 40 hours after fertilisation, from the 15th June to the 16th June, when the mean R_c value was 14.8 s m^{-1}. Conversely, there were periods when the V_d for HNO_3 exceeded the V_{max}, such as on the 13th June at 13:00 hours, when V_d for HNO_3 was recorded as 56.8 mm s^{-1} compared with a calculated maximum of 17.5 mm s^{-1}.

Reductions in V_d for HNO_3 (or in other words a non-zero R_c) have been linked to ground-level sources or non-zero vapour pressures of HNO_3 over nitrate-containing aerosol (particularly, NH_4NO_3), which may evaporate from aerosol within the air space below the measurements or previously deposited to leaf surfaces (Brost et al., 1988; Kramm and Dlugi, 1994; Nemitz et al., 2000a, 2004). By contrast, values of V_d for HNO_3 that exceed the theoretical maximum could suggest the presence of an

additional sink for HNO_3, which would potentially arise as the result of NH_3 reactions with HNO_3 to form NH_4NO_3 (Nemitz et al., 2000b; van Oss et al., 1998). The higher V_d values for HNO_3 during the fertilisation period, followed by a higher R_c value 40 hours afterwards, could suggest the formation of NH_4NO_3 immediately following fertilisation followed by its volatilisation soon after. Indeed, the exceedance of V_{max} coincided with upward fluxes of NH_4^+ and NO_3^- (Fig. 9) and this suggests

5 that during the period after fertilisation, the increase in NH_3 concentration lead to an exceedance of the equilibrium vapour pressures of NH_4NO_3 near the ground, resulting in partitioning of NH_3 and HNO_3 into the aerosol phase. This would have constituted an additional airborne sink for HNO_3 ($V_d > V_{max}$) as well as a source (apparent emission) for NH_4^+ and NO_3^- as previously reported by Nemitz et al. (2009).

10 It should be noted that during this period the aerodynamic gradient method does not derive accurate fluxes because the condition of flux conservation is not met (Wolff et al., 2010), and this period has therefore not been included in the diel cycles and summary statistics presented above.

By contrast, fluxes of total ammonium (tot-NH_4^+ =NH_4^+ + NH_3) and total nitrate (tot-NO_3^- = NO_3^- + HNO_3) would be con-
15 served, as the effect of gas-particle interactions are not considered, and their assessment provides additional information on the processes occurring during periods when fluxes are not conserved with height.

The time series for tot-NO_3^- and tot-NH_4^+ fluxes are shown in Figure 13. Prior to the fertilisation event on the 13th June, the fluxes for tot-NO_3^- were universally depositional to the surface, while fluxes of tot-NH_4^+ were bi-directional with significant
20 variation. However, six hours after fertilisation, a significant emission event of tot-NO_3^- was observed lasting for six hours. Interestingly, this indicates that the apparent NO_3^- emission during this period (Fig. 9) exceeded the measured deposition of HNO_3, and that there must have been a net source of NO_3^- at the surface during this period. Upward fluxes have previously been reported in the literature where it was attributed to the volatilisation of NH_4NO_3 from leaf surfaces (Neftel et al., 1996) or alkyl nitrate chemistry (Farmer and Cohen, 2008). Primary emissions of HNO_3 could arise from the heterogeneous reaction
25 of NO_2 with water (Harrison, 1996):

$$2NO_2 + H_2O \longrightarrow HONO + HNO_3, \hspace{5cm} (R3)$$

Kleffman (2007) suggests that HNO_3 could be formed by the reduction of NO_2 on organic sources of humic acid, a process that would also lead to the production of HONO. The formation of HNO_3 inferred from observations coincided with emissions of HONO post–fertilisation. However, as discussed previously, this reaction is slow, and while possibly contributing to some
30 of the observed HONO emission, may not be able to account for the majority of observed emissions.

A second potential pathway is the emission of HONO from the soil. As described by Scharko et al. (2015), the oxidation of ammonium by microbes in soils with high nitrification rates can lead to biogenic emissions of HONO. The addition of urea to

the agricultural soil at Easter Bush would lead to an increase in soil NH_4^+ concentrations and subsequently, through oxidation by soil microbes, the observed emission of HONO. Further discussion of the sources of HONO emissions at Easter Bush will be described in a future paper by Di Marco *et al.* (in preparation).

4.3 Daytime Source of HONO

As shown in Figure 5, the median diel concentrations for HONO recorded by the GRAEGOR at 2.4 m do not drop below the detection limits of the instrument, determined to be 30 ng m^{-3} from calibrations carried out during the campaign. This is contrary to what would be expected based solely on the photolysis rate of HONO, which would suggest that, after accumulation of HONO during night–time, rapid photolysis should reduce concentrations to below the detectable levels for measurement during early morning (Pagsberg et al., 1997). As measurement approaches have improved over the past 10 years, a growing number of measurements have revealed non-negligible HONO daytime concentrations at rural (Acker et al., 2006; Su et al., 2008; Sörgel et al., 2011), agricultural (Laufs et al., 2017) and urban (Lee et al., 2016) sites, including previous studies at the Easter Bush site (Twigg et al., 2011). Details on the discussion of a potential daytime source of HONO are further discussed in Di Marco *et al.* (in preperation).

4.4 Comparison of GRAGOR with other instrumentation

4.4.1 Comparison of nitrous acid measurement between GRAEGOR and LOPAP

The comparison between the LOPAPs and the GRAEGOR revealed that both sample boxes of the GRAEGOR measured higher HONO concentrations than the LOPAP, principally due to the presence of a constant concentration offset of 0.01 to 0.02 µg m^{-3} of HONO. Previous comparisons of measurements of HONO have been between the wet annular rotating denuder (WRD), as used in the GRAEGOR, and optical absorption techniques, primarily differential optical absorption spectroscopy (DOAS) instruments. In those comparisons, it has been found that HONO measurements by WRD, particularly during daytime and at low concentrations, tend to be significantly higher than DOAS measurements (Appel et al., 1990). By comparison, the LOPAP shows good agreement in HONO measurements with the DOAS (Kleffman et al., 2006), as the DOAS method is a molecule specific method and the LOPAP method measures any potential NOx artefact.

The higher concentrations recorded by the GRAEGOR can be explained by the presence of chemical interferences that occur on the inlet, at the air/liquid interface and within the sampling solution. As the WRD uses a liquid film to sample HONO, and as HONO can form heterogeneously on such surfaces, overestimation of HONO can occur. Furthermore, interferences by chemical reactions of NO_2 with hydrocarbons within the sampler can lead to a further interference (Gutzwiller et al., 2002), particularly in proximity to diesel emissions. It has also been shown that in high-alkalinity sampling solutions, mixtures of SO_2 and NO_2 can add a further interference to measurements (Spindler et al., 2003). Finally, photolytically induced artefacts can be introduced in the sampling lines that connect the GRAEGOR sampling box to the detector unit (Kleffman and Wiesen,

2008). The LOPAP, which is also a wet chemistry-based instrument, is designed to minimise the chemical interferences and artefacts that can be introduced in other wet chemistry instruments.

A comparison between daytime (06:00 to 18:00) GRAEGOR HONO concentrations and LOPAP HONO concentrations found only a slightly greater difference than the comparison between night time (19:00 to 05:00) concentrations recorded by the GRAEGOR and LOPAP. While previous comparisons between the DOAS and the WRD found that daytime concentrations measured by the WRD were higher than the DOAS compared to night-time measurements, these studies were generally conducted in urban areas where both HONO and NO_x concentrations were high (Febo et al, 1996), in contrast to the low concentrations at Easter Bush. The implementation of thermal insulation material around the GRAEGOR sampling lines may have also reduced the influence of photolytic artefacts in exposed sampling lines during the day, which would have elevated daytime HONO measurements recorded by the GRAEGOR.

Spindler et al. (2003) developed the following quantification of the chemical artefact produced by the mixing of NO_2 and SO_2 in highly alkaline sampling solutions for HONO measurements in their investigation of SO_2 and NO_2 chemical interference, with all concentrations measured in ppb.

$$[HONO]_{artefact} = 0.0056[NO_2] + 0.0022 \text{ ppb}^{-1}[NO_2][SO_2] \tag{12}$$

The first term describes the heterogeneous formation of NO_2 with water alone, and the second describes the aqueous-phase reaction of NO_2 and SO_2. Using measurements of SO_2 and NO_2 concentrations, the HONO artefact for the period of the GRAEGOR-LOPAP comparison was calculated and subtracted from the HONO concentrations recorded by the GRAEGOR. A linear regression between the concentrations recorded by GRAEGOR Sample Box 2 and LOPAP (2.0 m), which had the best agreement without artefact reduction, indicated better agreement after the correction for the artefact (GRAEGOR$_{artefact}$(2.4 m) = 1.02*HONO(HONO(LOPAP(2.0 m)), intercept = 5 x 10^{-3} µg m^{-3}, $R^2 = 0.72$). Coefficient values were altered to produce the best possible agreement between GRAEGOR and LOPAP HONO values, arriving at a final artefact quantification of:

$$[HONO]_{artefact} = 0.0090[NO_2] + 0.0034 \text{ ppb}^{-1}[NO_2][SO_2] \tag{13}$$

Use of these altered coefficients further reduced the offset in GRAEGOR HONO measurements, but also reduced the statistical agreement between GRAEGOR and LOPAP HONO measurements (GRAEGOR$_{artefact}$(2.4 m) = 0.98*HONO(HONO(LOPAP(2.0 m)), intercept = 2 x 10^{-3} µg m^{-3}, $R^2 = 0.57$). Figure 14 shows the results of these analyses, with the linear regression between GRAEGOR Sample Box 2 (2.4 m) and LOPAP (2.0 m) without the artefact reduction applied to GRAEGOR Sample Box 2 (2.4 m) HONO concentrations, with Spindler's artefact reduction, and with the modified Spindler's artefact reduction. While the agreement between the LOPAP and GRAEGOR is improved by the introduction of an artefact reduction value, it does not fully close the gap even with altered coefficient values, with a constant concentration

offset still present in measurements. The possibility that a further artefact is introduced from NO_2 mixing with hydrocarbons would require further investigation, with concurrent measurements of hydrocarbons.

To determine if the HONO concentration offset in the GRAEGOR measurements impacted upon the measurements of HONO flux, a comparison between the HONO flux values derived from GRAEGOR and LOPAP measurements was conducted. Concurrent fluxes of HONO derived from LOPAP and GRAEGOR measurements exist for 72 hourly measurements, from the 26th May – 6th June. Figure 15 shows (a) the full time series of concurrent HONO flux values derived from GRAEGOR and LOPAP measurements and (b) a scatter plot of GRAEGOR against LOPAP HONO flux values. Overall, GRAEGOR HONO fluxes are biased towards deposition, with greater deposition values and lesser emission values compared to concurrent LOPAP values. This pattern would be consistent with the concept of an artefact formation dependent upon SO_2 and NO_2. SO_2 fluxes were unidirectionally depositional at Easter Bush during the campaign as measured by the GRAEGOR. Deposition of SO_2 would lead to greater formation of artefact within the sample box set at a higher height, which is consistent with comparisons of HONO concentrations between each sample box of the GRAEGOR and each LOPAP instrument. In turn, this would lead to a bias in HONO flux values, resulting in a skew towards deposition. It should be noted that the sample size of concurrent measurements of HONO flux from GRAEGOR and LOPAP measurements is limited (n = 72).

4.4.2 Comparison of ammonia measurements with GRAEGOR and QCL

The comparison between the GRAEGOR and QCL found that, while there was reasonable agreement between the instruments, the GRAEGOR measured somewhat higher NH_3 concentrations than the QCL, by a factor of 1.2. Due to lack of ancillary micrometeorological data during this campaign, the short overlap in measurements, and necessary filtering of unreliable data, there are too few concurrent measurements (15 hours) of flux between the QCL and the GRAEGOR for a reliable comparison. There are also only 41 hours of concurrent concentration measurements between the two instruments, which overlapped with a period of low NH_3 concentrations.

A similar comparison between a WRD system (the Ammonia Measurement by Annular Denuder with Online Analysis, AMANDA) and the QCL system was conducted at the same site in 2004 and 2005 by Whitehead et al. (2008). This comparison also found that the WRD system measured higher concentrations of NH_3 compared to the QCL, but at a far greater factor of 1.67. This difference was particularly pronounced during periods of low NH_3 concentrations, with better agreement recorded during a fertilisation and cutting event that occurred during that study. The older (pumped) QCL used during the earlier campaign did not derive its concentrations from first principles, in contrast to the QCL used during the comparison with the GRAEGOR reported here, which should be within 3% of the absolute value without further calibration, according to the manufacturer. An inter-comparison between eleven different measurement techniques for NH_3 – including the AMANDA and two QCL instruments (the DUAL-QCLAS and the compact-QCLAS) - was conducted at the Easter Bush site in 2008 (von Bobrutzki et al., 2011). While good statistical agreement was found in linear regression between the AMANDA and both QCL

instruments for NH_3 concentrations throughout the entirety of the campaign ($R^2 = 0.92$ and $R^2 = 0.97$ for the compact-QCL and DUAL-QCLAS, respectively), there was less agreement between the instruments during periods of low (<10 ppb) NH_3 concentrations ($R^2 = 0.81$ and $R^2 = 0.52$ for the compact-QCL and DUAL-QCLAS respectively). During periods of low concentration, the QCL systems also underestimated NH_3 concentrations compared to the AMANDA.

Any errors in the GRAEGOR's internal NH_3 calibration system are unlikely to have an effect at low NH_3 concentrations. As a test, the calibration values obtained from all the internal calibration checks which were carried out through the campaign (total calibrations = 5) were used to calculate the NH_3 concentrations during the period of QCL measurements. No significant concentration difference was found between the concentrations obtained by different calibration values, due to no systematic

10   difference in agreement between the different calibration periods.

While there remain significant differences in measured NH_3 concentrations between the GRAEGOR and QCL, the improved agreement between those concentrations, particularly at low values, compared with the results from 2004 and 2005 suggests an improved methodology in use by the QCL system in place at Easter Bush. Further measurements, particularly of fluxes and

15   during periods of high NH_3 concentrations, would be required for a more detailed analysis.

5 Conclusion

In this paper, we have presented for the first time simultaneous measurements of the trace gases HCl, HONO, HNO_3, SO_2 and NH_3, and their associated water-soluble aerosols counterparts Cl^-, NO_2^-, NO_3^-, SO_4^{2-}, NH_4^+, before and after urea fertilisation of an agricultural grassland. The main findings for this study are:

1. Simultaneous measurements of the components of the NH_3-NO_3-NH_4NO_3 triad suggested formation of ammonium nitrate post fertilisation. The use of the conservative exchange fluxes tot-NH_4^+ and tot-NO_3^- indicates the presence of a ground source of HNO_3 post fertilisation, which would be rapidly scavenged by high post-fertilisation concentrations of NH_3 to form NH_4NO_3. Through this mechanism, use of urea fertiliser becomes a source of regional, rather

25   than local, pollution.

2. The deposition velocities measured for the aerosol compounds Cl^-, NO_3^- and SO_4^{2-} were significantly larger than those measured for NH_4^+. After normalisation by turbulence, the measurements suggested a clear relationship between deposition velocity and particle size for Cl^-, NO_3^- and SO_4^{2-}, as parameterised using the proxy of compound in

30      $PM_{2.5}$/PM_{10}, although the relationship shows significant variability. Therefore, the high deposition velocities for aerosol compounds recorded at the site are a result of a fraction of the compounds being contained in super-micron aerosol, such as sea–salt sulphate and sodium nitrate.

3. Evidence for a HONO daytime source at the site throughout the campaign adds to the growing body of past measurements that has found evidence for HONO daytime formation in rural, urban and agricultural areas. There is also evidence for the emission of HONO post fertilisation at the site.

This also appears to be the first time a comparison between measurements of HONO concentrations determined by the LOPAP and the GRAEGOR instruments has been documented. While good linear agreement exists between HONO measurements taken by GRAEGOR and LOPAP at both measurement heights, a consistent offset in GRAEGOR HONO measurements suggest the presence of chemically induced artefacts within the GRAEGOR system. This is potentially linked to atmospheric SO_x

10 and NO_x concentrations.

Furthermore, this paper presents a comparison between measurements of NH_3 concentration determined by the GRAEGOR and a QCL system. While changes to the QCL operation system compared to previous studies conducted at the site have resulted in better agreement between the GRAEGOR and QCL, particularly for low NH_3 concentrations, there still remain

15 significant differences in NH_3 concentrations with larger values reported by the denuder system.

Future measurements of aerosol deposition velocities should aim to investigate the effect of particle size upon deposition velocity, using a more robust measurement of particle size than used here. In addition, the ability of urea pellets to act as a potential surface on which heterogeneous formation of HONO and HNO_3 occurs should be investigated, particularly as the

20 formation of these compounds can give rise to the formation of the regional pollutant NH_4NO_3.

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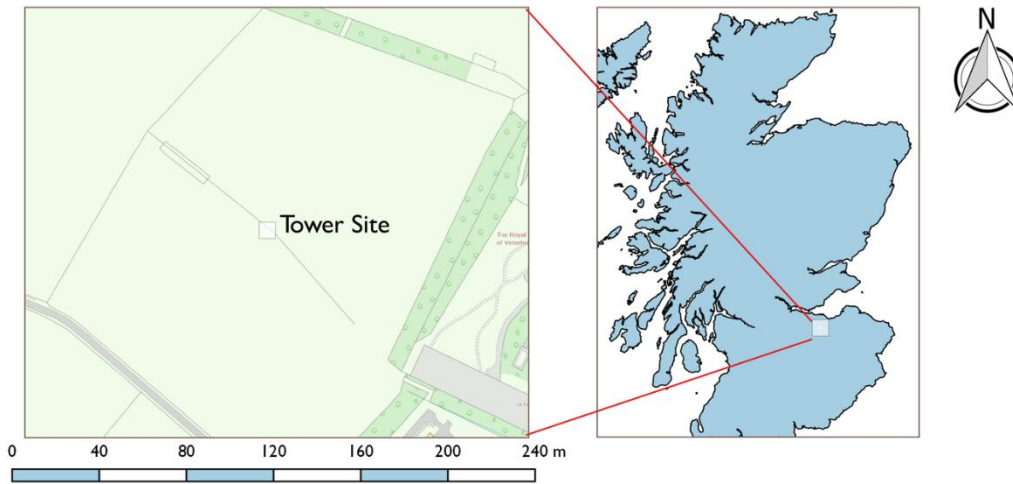
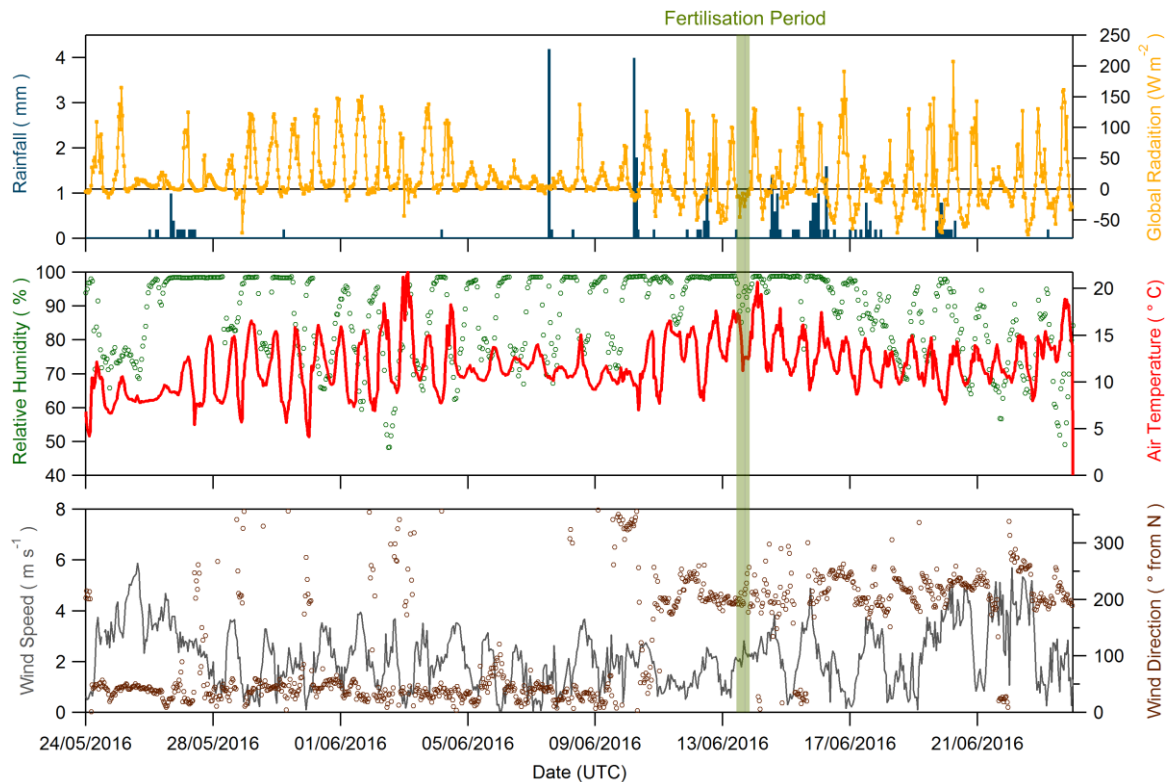


Figure 1: Location of the Bush Tower Site (3°12'W, 55°52'N) in relation to surrounding agricultural land and within Scotland, UK.

5   **Figure 2: Global radiation (orange line), rainfall (blue bars), relative humidity (green dots), air temperature (red line), wind direction (brown circles) and wind speed (grey line) recorded during the Easter Bush Campaign, May to June 2016. The fertilisation period was 08:00 – 09:00 on 13th June and is highlighted in green.**

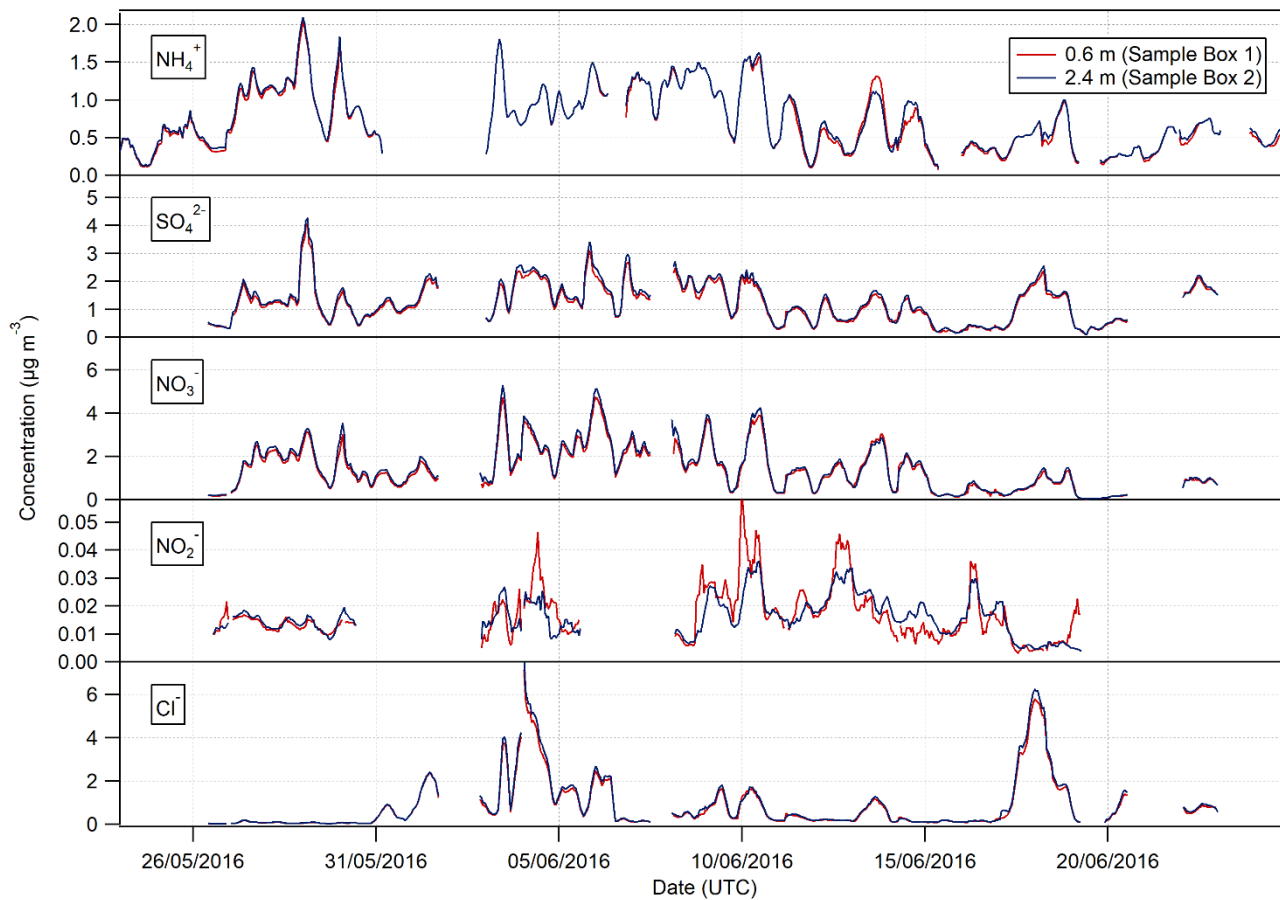



Figure 3: Time series of hourly concentrations of the water-soluble aerosol species measured during the Easter Bush campaign. Results smoothed using a 5-hour moving point average.

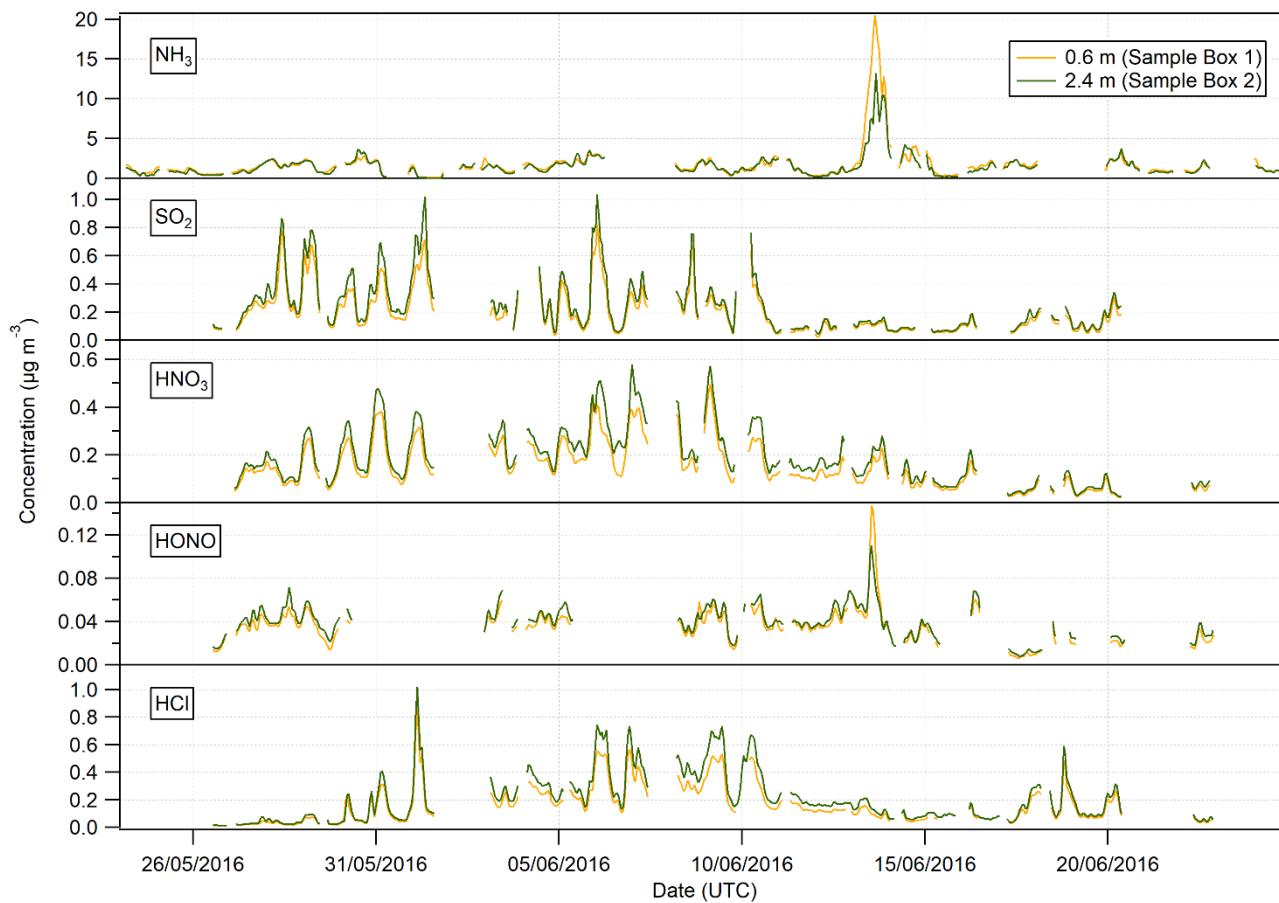



Figure 4: Time series of hourly concentrations of the gaseous species measured during the Easter Bush campaign. Results smoothed using a 5-hour moving point average.

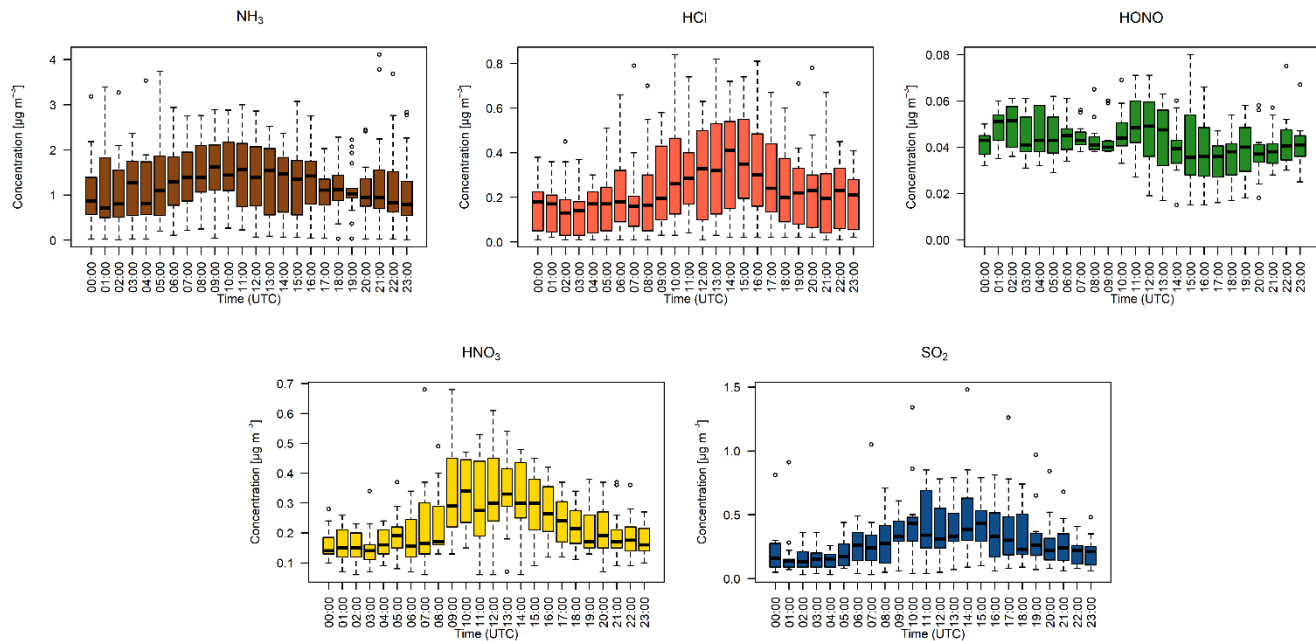



Figure 5: Hourly median diel trace gas concentrations measured by the GRAEGOR at 2.4 m. Boxes show the lower and upper quartiles and whiskers the 5% to 95% range, with outliers shown as circles.

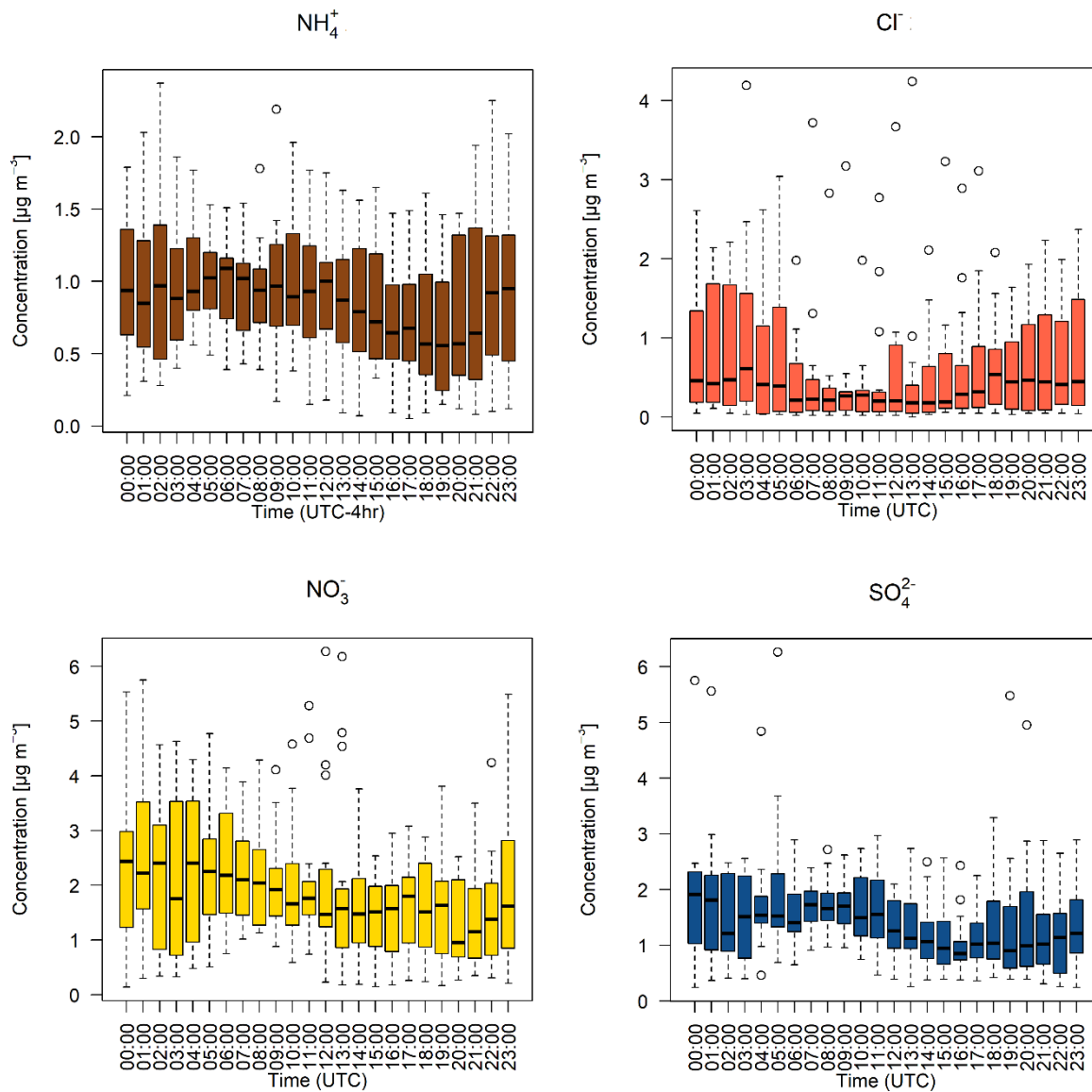



Figure 6: Hourly median diel water–soluble aerosol concentrations measured by the GRAEGOR at 2.4 m. Boxes show the lower and upper quartiles and whiskers the 5% to 95% range, with outliers shown as circles.

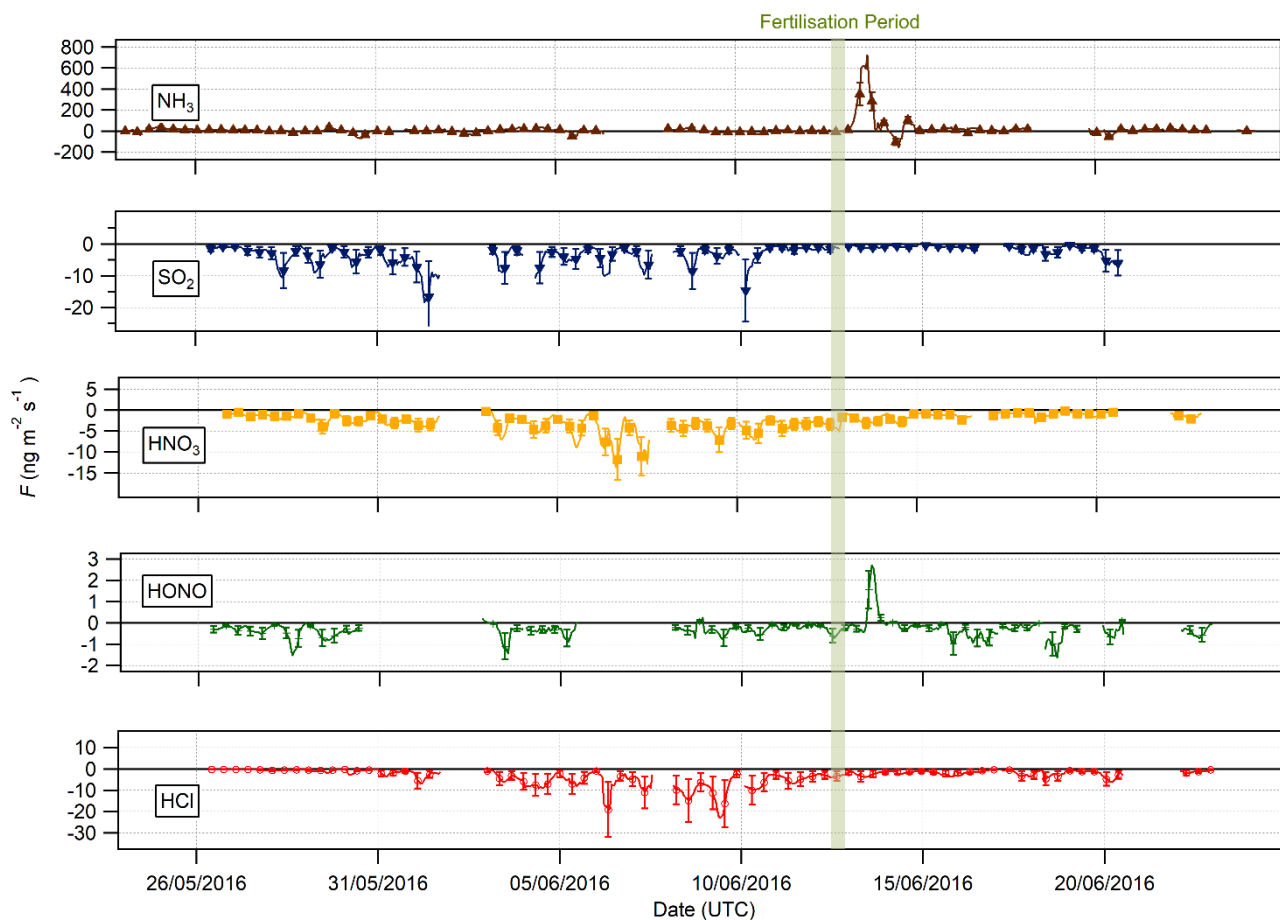



Figure 7: Time series of hourly trace gas fluxes measured during the Easter Bush campaign. Results smoothed using a 5-point moving point average. The fertilisation period was 08:00 – 09:00 on 13th June, and is highlighted in green. Flux uncertainties for each trace gas are included as error bars.

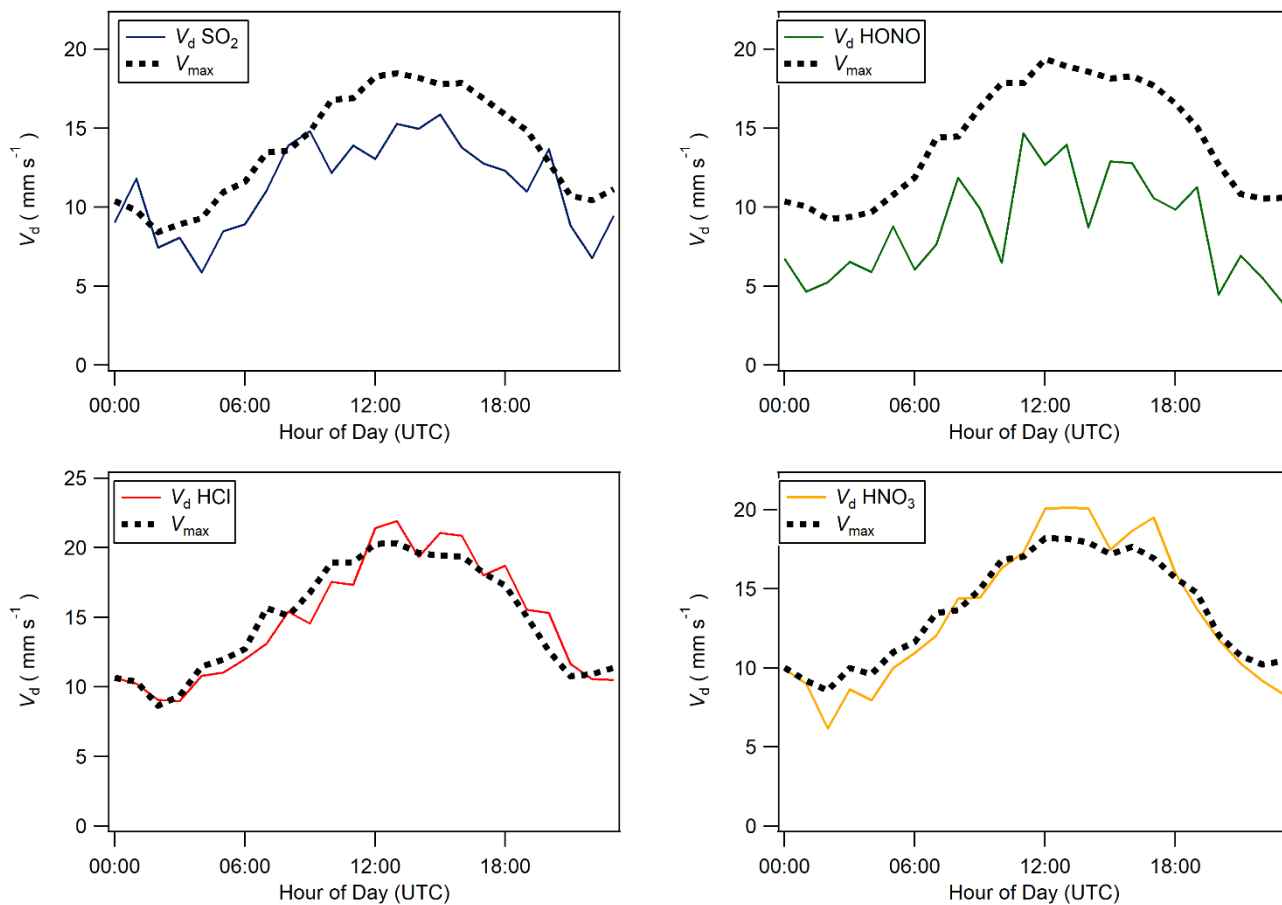



Figure 8: Median diel cycles for deposition velocity (V_d) and maximum deposition velocity (V_{max}) for (from top left clockwise) SO_2, HONO, HNO_3 and HCl.

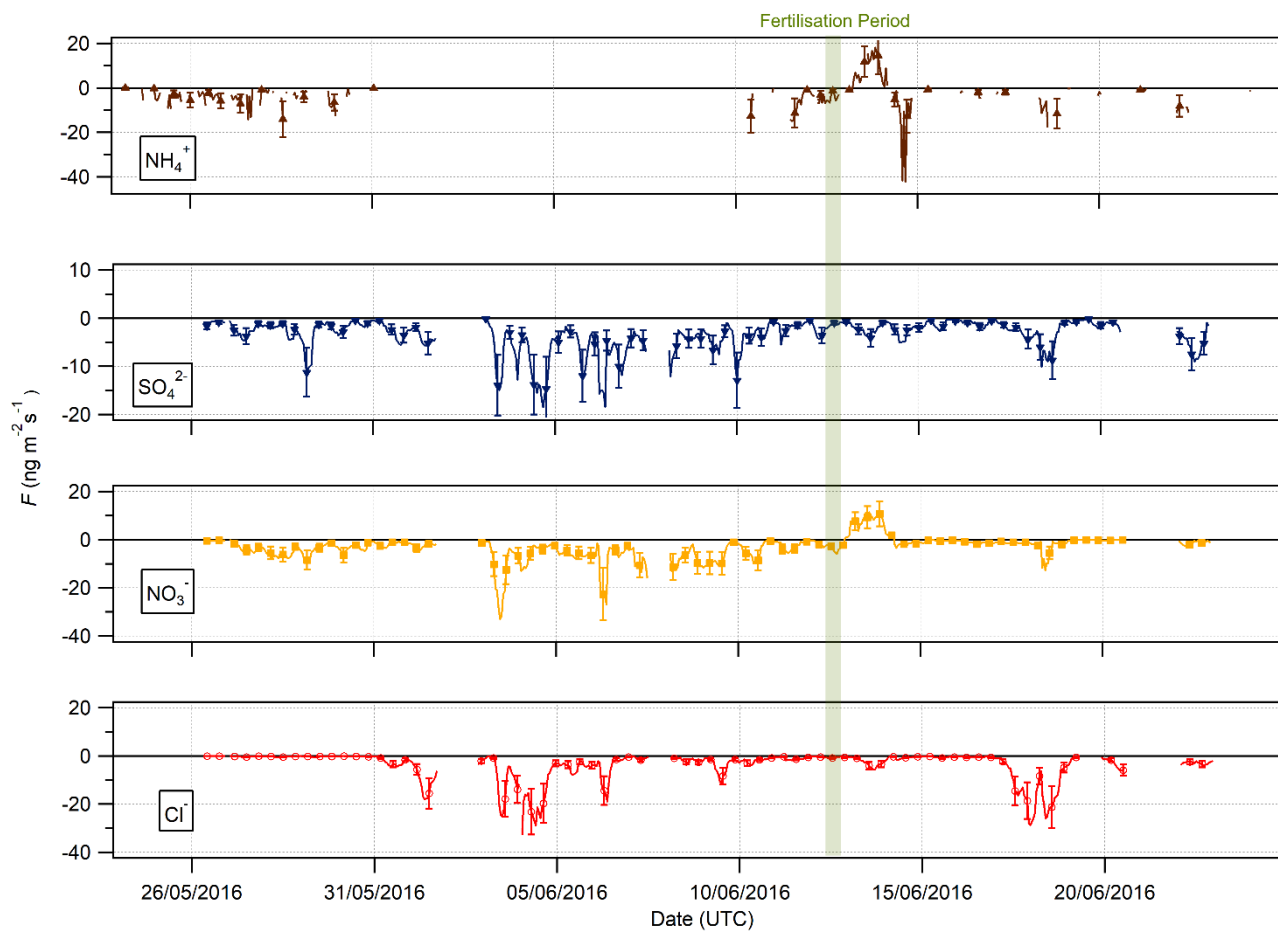



Figure 9: Time series of hourly fluxes of water-soluble aerosol species measured during the Easter Bush campaign. Results smoothed using a 5-point moving point average. The fertilisation period was 10.00 on 13th June, and is highlighted in green. Flux uncertainties for each aerosol are included as error bars.

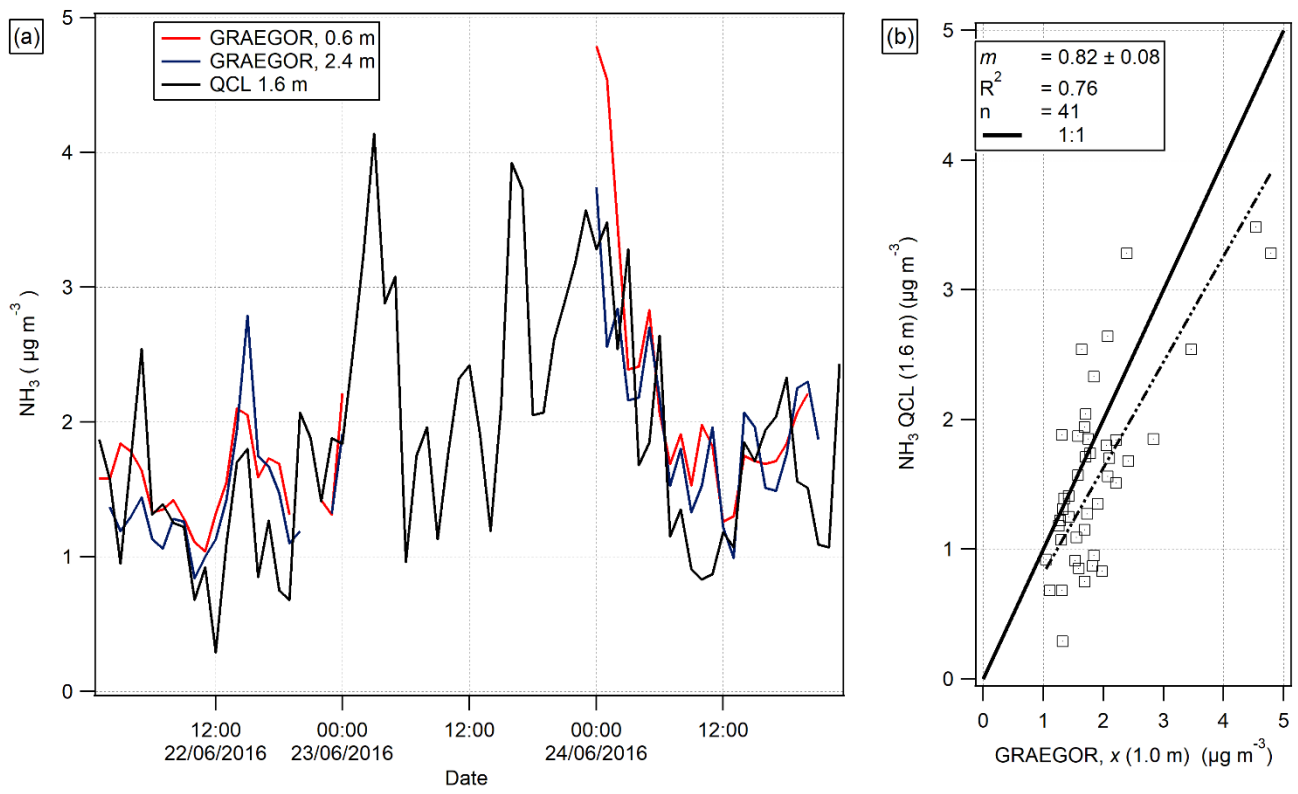



Figure 10 : (a). Time series of hourly averages of NH₃ measurements recorded by GRAEGOR (0.6 m and 2.4 m) and QCL. (b). Linear regression analysis between QCL NH₃ measurements and GRAEGOR (derived averaged concentration at 1.0 m) NH₃ measurements.

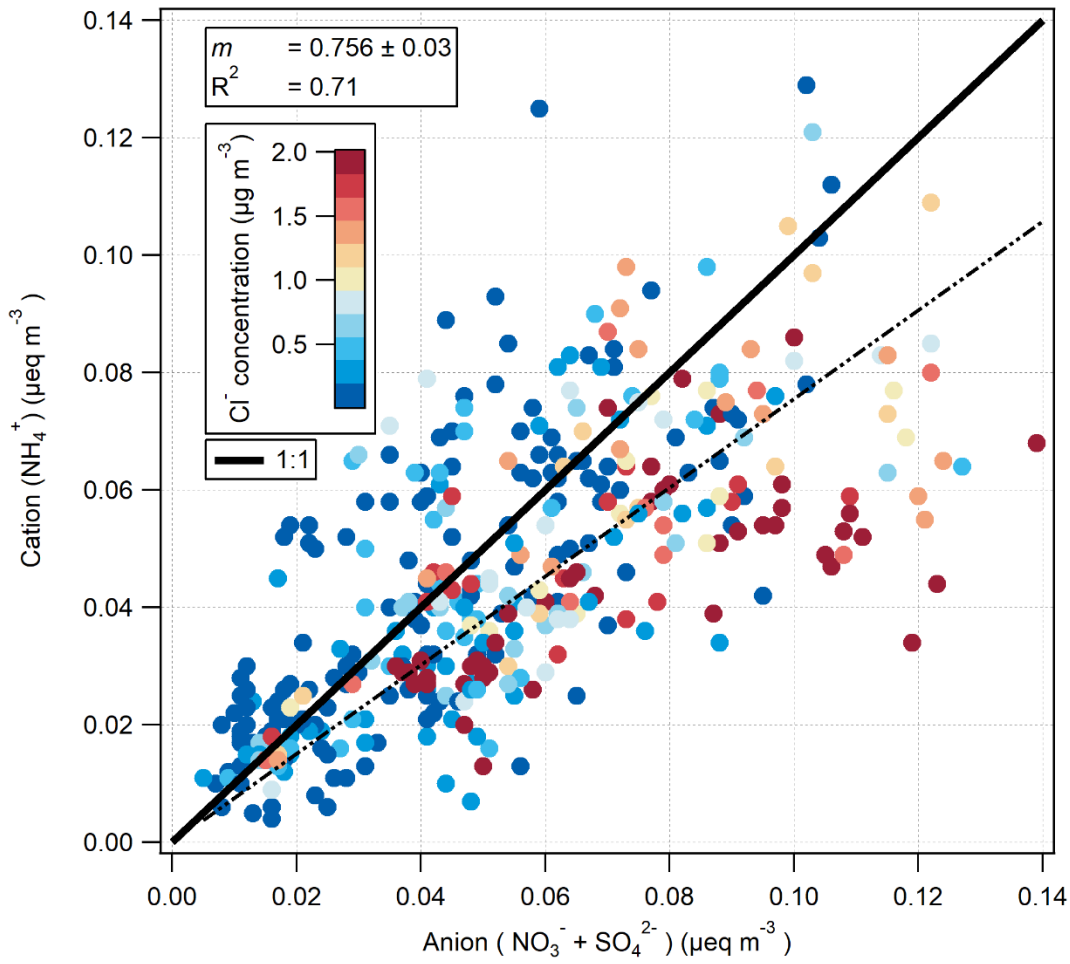



Figure 11: The ion balance of measured selected anions (NO₃⁻ + SO₄²⁻) and measured cations (NH₄⁺) in μeq m⁻³. The colour scale is capped at 2 μeq m⁻³ Cl⁻ to highlight the association of anion excess with periods of sea salt influence.

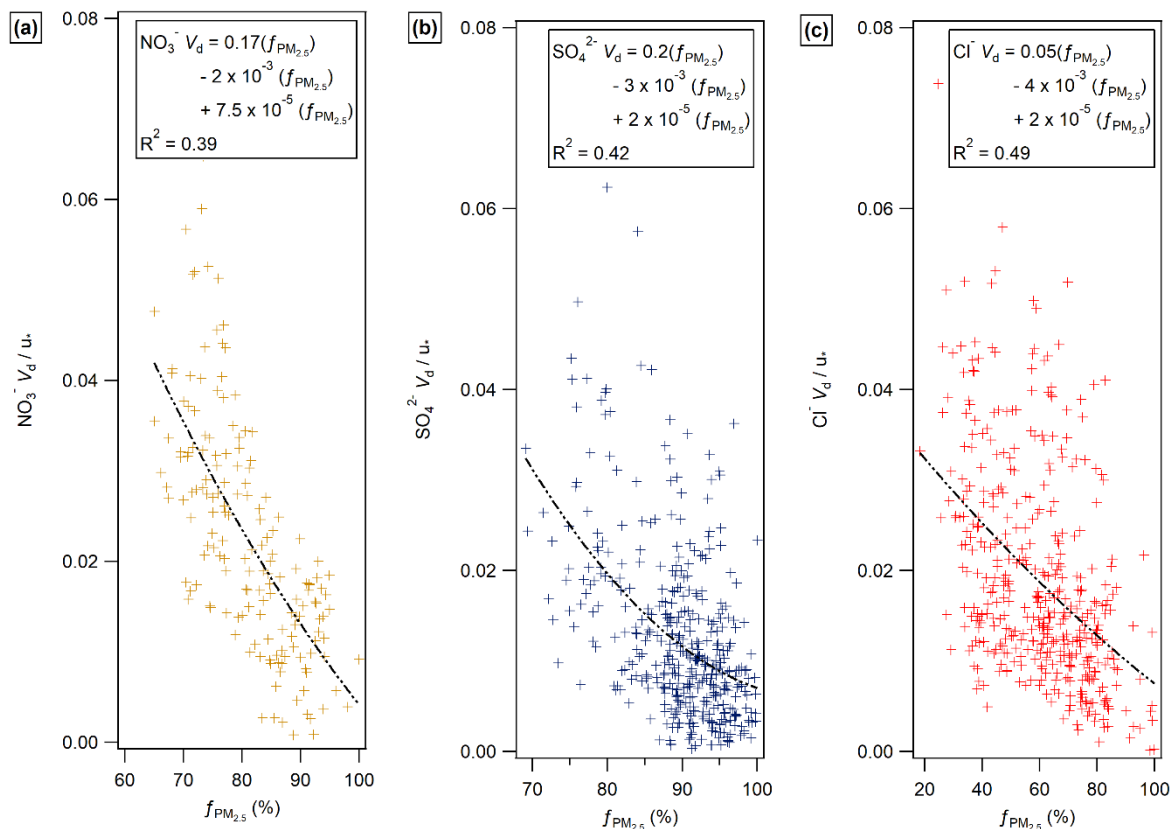



Figure 12: The normalised deposition velocity as a function of $f_{PM2.5}$ (expressed as a %) for (a) nitrate, (b) sulfate and (c) chloride, derived from the MARGA measurements at Auchencorth Moss.

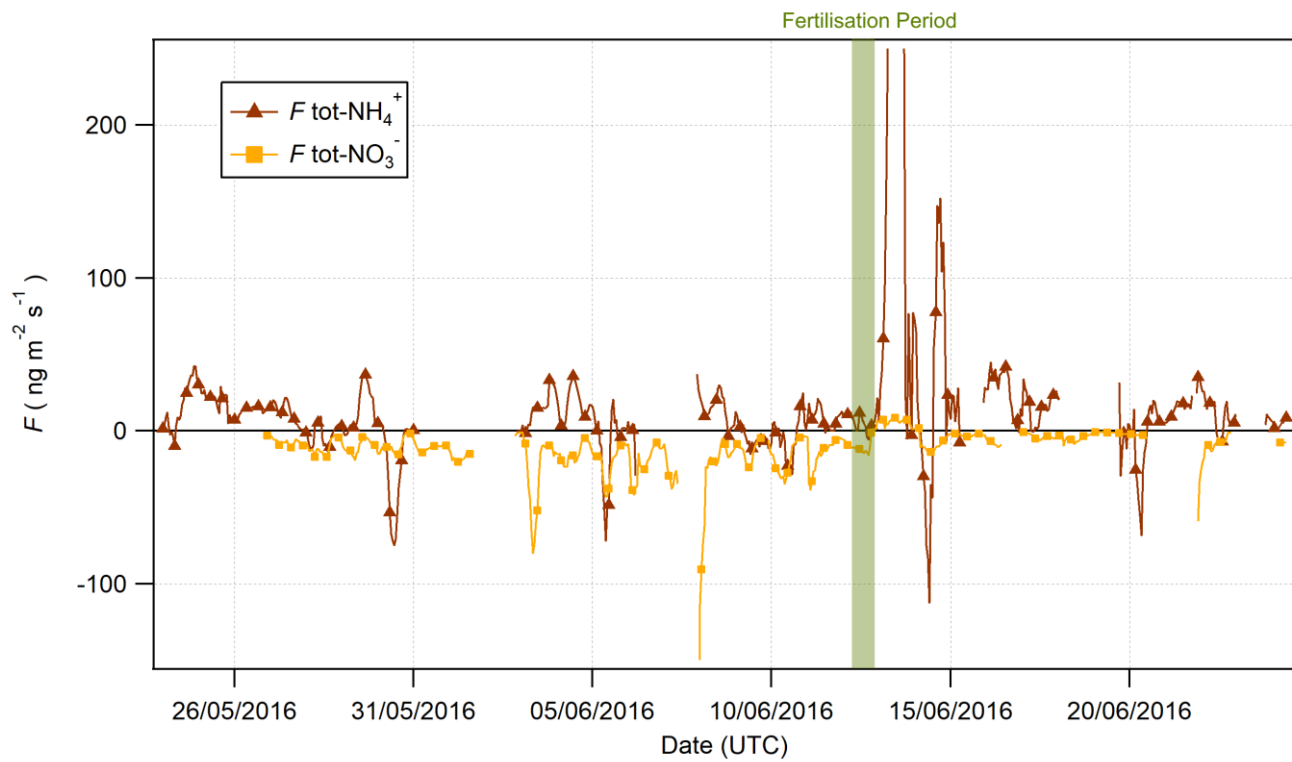



Figure 13: Fluxes of tot-NO$_3^-$ and tot-NH$_4^+$ pre-and post-fertilisation on the 13th June 2016 at 09:00 (marked in green).

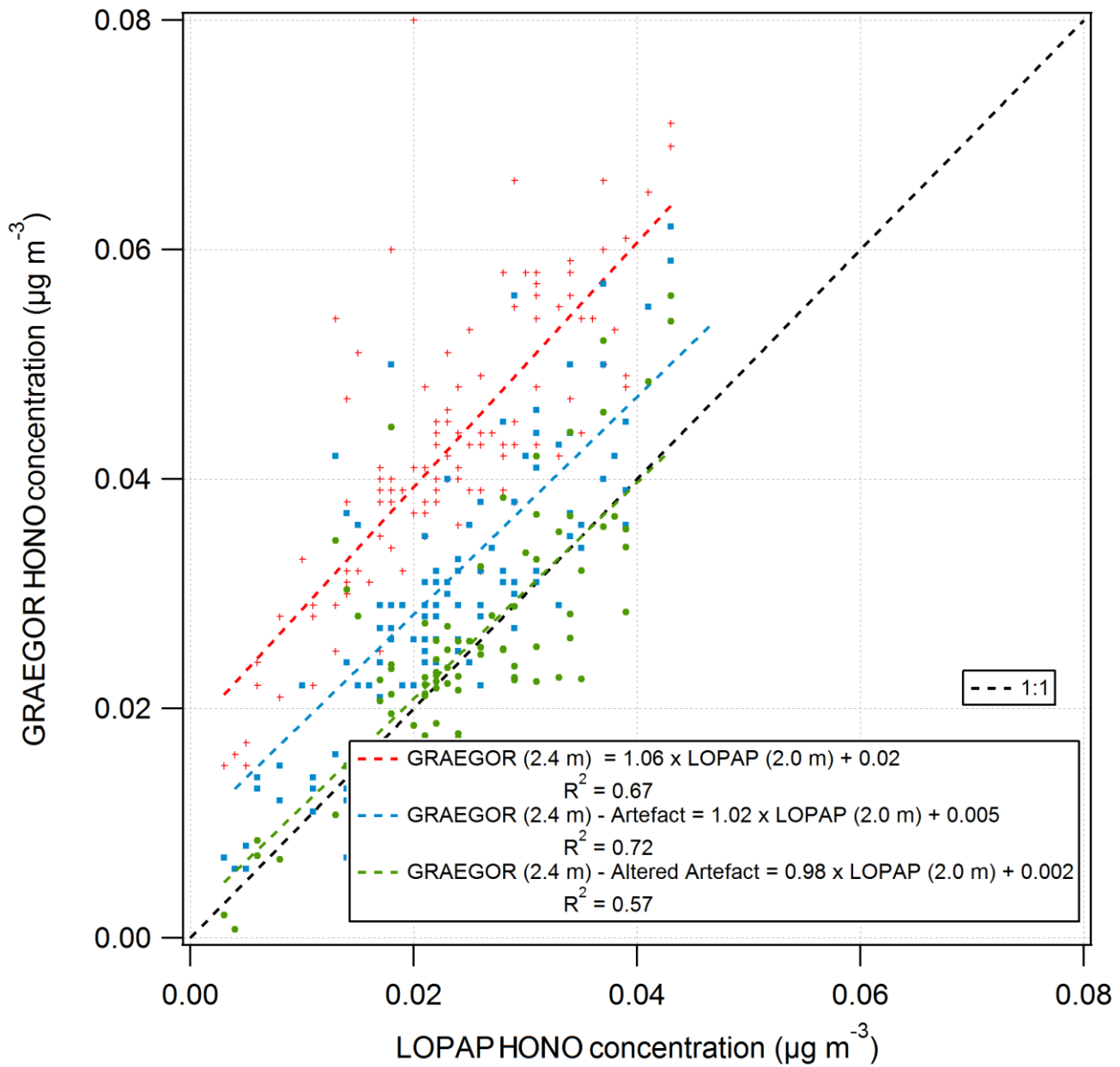



Figure 14: Simple linear regression analyses between GRAEGOR (2.4 m) and LOPAP (2.0 m) without artefact reduction (red), with Spindler's artefact reduction (blue), and with modified Spindler's artefact reduction (green) applied to GRAEGOR (2.4 m) HONO concentration.

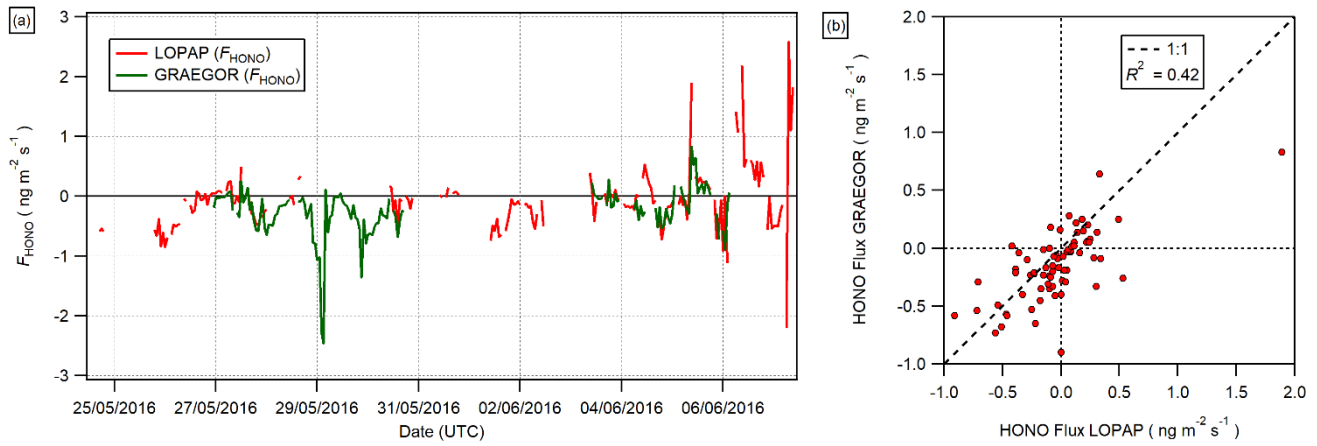



Figure 15: (a) Time series of concurrent flux measurements of HONO derived from LOPAP (red) and GRAEGOR (green) measurements. **(b)** Scatter plot of GRAEGOR HONO flux values against LOPAP HONO flux values.

Table 1: Limits of detection (LOD, determined as three standard deviations from average baseline signal), mean (μ_A), median (μ_M), min, max, and arithmetic standard deviation (σ_A) for concentrations measured at 2.4 m for trace gas and water-soluble aerosols measured during Easter Bush campaign, calculated from hourly data. Number of measurements (N) for each compound is also shown.

(2.4 m)	LOD ng m^{-3}	μ_A µg m^{-3}	μ_M µg m^{-3}	Min µg m^{-3}	Max µg m^{-3}	σ_A µg m^{-3}	N
NH_4^+	190	0.74	0.64	<LOD	2.33	0.43	580
Cl^-	15	0.91	0.36	<LOD	7.88	1.31	515
NO_2^-	17	0.02	0.02	<LOD	0.05	0.01	373
NO_3^-	47	1.53	1.32	<LOD	6.27	1.18	538
SO_4^{2-}	109	1.29	1.22	<LOD	6.26	0.83	540
NH_3	172	1.48	1.15	<LOD	13.79	1.5	602
HCl	67	0.2	0.15	<LOD	1.4	0.18	544
HONO	30	0.04	0.04	<LOD	0.12	0.02	410
HNO_3	97	0.19	0.16	<LOD	0.68	0.12	509
SO_2	120	0.24	0.18	<LOD	1.48	0.21	480

Table 2: Mean (µA), median (µM), minimum and maximum vales for flux, deposition velocity (V_d), maximum deposition velocity (V_{max}), and canopy resistances (R_c) for trace gases measured during Easter Bush campaign, based on hourly values. Also shown are the median relative standard error (σ_F), the flux limit of detection (F_{LOD}) evaluated for typical conditions (median u_* and median concentration) as well as the fraction of the hourly flux value that exceed the flux detection limit evaluated for that hour (f_{LOD}).

		NH_3	HCl	$HONO$	HNO_3	SO_2
Flux (ng m^{-2} s^{-1})	μ_A	15.24	-3.51	-0.3	-2.66	-3.04
	μ_M	5.65	-1.98	-0.29	-1.99	-1.68
	Min	-324.5	-61.24	-2.46	-18.57	-35.57
	Max	1460	-0.03	4.92	0.82	-0.03
	No. of measurements	577	506	384	500	465
	σ_F (%)	32	58	56	42	67
	F_{LOD}	1.28	0.75	0.18	0.89	0.97
	f_{LOD} (%)	94	84	78	87	89
V_d (mm s^{-1})	μ_A	-8.99	15.1	8.8	13.61	11.69
	μ_M	-6.1	14.49	7.69	12.87	10.00
	Min	-215.3	0.01	-55.6	-4.72	0.34
	Max	92.90	52.83	59.81	56.78	55.38
V_{max} (mm s^{-1})	μ_A	19.46	15.33	14.12	13.91	14.22
	μ_M	18.8	15.26	13.99	13.75	14.02
	Min	1.7	0.04	0.04	0.04	0.45
	Max	57	40.41	36.99	36.93	36.99
R_c (s m^{-1})	μ_A	0	33.75	331.5	23.29	49.20
	μ_M	0	1.82	13.07	5.71	27.61

Table 3: Mean (μA), median (μM), minimum and maximum vales for flux and deposition velocity (V$_d$) for water soluble aerosols measured during Easter Bush campaign. Also shown are the median relative standard error (σ$_F$), the flux limit of detection (F$_{LOD}$) evaluated for typical conditions (median u_* and median concentration) as well as the fraction of the hourly flux value that exceed the flux detection limit evaluated for that hour (f$_{LOD}$).

		NH$_4^+$	Cl$^-$	NO$_3^-$	SO$_4^{2-}$
Flux (ng m^{-2} s^{-1})	μ$_A$	-3.55	-4	-3.34	-3.56
	μ$_M$	-2.97	-1.11	-1.76	-2.19
	Min	-42.23	-60.04	-89.32	-59.69
	Max	18.15	-1.06	31.84	-0.95
	No. of meas-urements	224	484	477	482
	σ$_F$ (%)	58	41	48	45
	F$_{LOD}$	2.21	0.85	1.28	1.78
	f$_{LOD}$ (%)	91	81	84	87
V$_d$ (mm s^{-1})	μ$_A$	0.93	3.65	1.97	1.89
	μ$_M$	0.37	3.14	1.52	1.45
	Min	-0.04	-0.92	-9.43	-2.48
	Max	7.57	21.26	9.8	9.53