

## Interactive comment on "Satellite evidence of substantial rain-induced soil emissions of ammonia across the Sahel" by Jonathan E. Hickman et al.

## Anonymous Referee #1

Received and published: 2 August 2018

This manuscript examines pulses of NH3 observed by the IASI-A instrument on MetOp-A over the Sahel. These pulses occur at the start of the rainy season and appear to correspond to increased biochemical activity in soils induced by rainfall. The paper is well written, the analysis is clear, and the results are very interesting. My main concern is that the timing of the enhancements in OMI NO2 (May-June), IASI NH3 (March-April), and surface NH3 (May-June) appear to be different. I elaborate on this below.

The timing of the NH3 peaks seems to be different than those of NO2. Several studies (for example Jaeglé et al., 2004; Hudman et al., 2012) have examined satellite NO2

C1

pulses over the Sahel, finding that the pulses take place sometime in May-June, corresponding to the arrival of the first rains after the dry season over various regions of the Sahel. Figure 2, shows the same pattern of the largest OMI NO2 enhancement taking place in June. For NH3, the authors focus mostly on March and April. Some discussion of why the timing would be different in terms of pulses of NO2 and NH3 would be useful to include in the manuscript (for example in section 3.2.3 discussing the coemissions of NH3 and NO2). Does it have to do with different population dynamics of the various bacteria and their response to subsequent wetting events? Interestingly, Figure 3 shows that surface observations do indicate maximum NH3 mixing ratios in May-June (at least for sites b, c, f), consistent with the OMI NO2 pulses. Why would surface observations of NH3 show a different seasonality than the satellite observations of NH3? Is the increasing cloud cover affecting the number and quality of the retrievals in May-June? Is the later June NH3 pulse masked by clouds?

Minor comments. - Throughout the manuscript (including the abstract). The authors refer to NO2 as 'nitric dioxide'. It should be nitrogen dioxide.

- Equation 1. It isn't clear how this is used for low concentrations. Does the equation mean that even if the error is above 100%, if the concentrations are low enough then the retrievals are kept? Also, it would be useful to have units after 5x1015 (I assume it is molec/cm2)

- Equation 3. There is an extra space before the tx (effective lifetime), also, x should be subscripted.

Interactive comment on Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2018-550, 2018.