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- 1 The impacts of regional shipping emissions on the chemical characteristics of coastal
- 2 submicron aerosols near Houston, TX
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- 16 Abstract
- 17 The air quality of the Texas Gulf Coast region historically has been influenced heavily by
- 18 regional shipping emissions. However, the effects of the recently established North American
- 19 Emissions Control Area on aerosol concentrations and properties in this region are presently
- 20 unknown. In order to better understand the current sources and processing mechanisms
- 21 influencing coastal aerosol near Houston, a high-resolution time-of-flight aerosol mass

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spectrometer (HR-ToF-AMS) was deployed for three weeks at a coastal location during May-June 2016. Total mass loadings of organic and inorganic non-refractory aerosol components during onshore flow periods were similar to those published before establishment of the regulations. Based on estimated methanesulfonic acid (MSA) mass loadings and published biogenic MSA to non-sea-salt-sulfate (nss-SO₄) ratios, an average of over 75% of the observed nss-SO₄ was from anthropogenic sources, predominantly shipping emissions. Mass spectral analysis indicated that for periods with similar backward-trajectory-averaged meteorological conditions, air masses influenced by shipping emissions had an increased mass fraction of ions related to carboxylic acids and larger oxygen-to-carbon ratios than those that avoided shipping lanes, suggesting that shipping emissions increase marine organic aerosol (OA) oxidation state. Amine fragment mass loadings were correlated positively with anthropogenic nss-SO₄ during onshore flow, implying anthropogenic-biogenic interaction in marine OA production. Model calculations also suggest that advection of shipping-derived aerosol may enhance inland aqueous-phase secondary OA production. These results emphasize the continuing role of shipping emissions on aerosol properties over the Gulf of Mexico and suggest that further regulation of shipping fuel sulfur content will reduce coastal submicron aerosol mass loadings near Houston.

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1. Introduction

Seaborne trade is a relatively inexpensive and efficient mechanism to transport goods across the globe (IMO, 2012). As a result, such transportation is thought to account for more than 90% of global trade volume (Eyring et al., 2010; IMO, 2012) and has been growing rapidly in the past two decades (Lack et al., 2009; Eyring et al., 2010; Tournadre, 2014; Johansson et al., 2017). As

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international waters, they frequently burn low-quality residual fuel oils, leading to considerable 46 emissions of sulfur dioxide (SO₂), nitrogen oxides (NO_x), and particulate matter (PM) (Lack et 47 al., 2009; Murphy et al., 2009; Czech et al., 2017). Recently, increasing attention has been paid 48 to the impact of these emissions on ambient PM mass loadings in coastal areas, with notable 49 contributions in Europe (Viana et al., 2014 and references therein; Aksoyoglu et al., 2016), Asia 50 (Zhao et al., 2012; Liu et al., 2016), and the United States (Vutukuru and Dabdub, 2008; 51 52 Agrawal et al., 2009). Coastal populations exposed to these emissions are subsequently affected by numerous negative health impacts. Corbett et al. (2007) estimated that shipping activity was 53 responsible for 60,000 global premature mortalities annually. More recent studies have 54 confirmed links between shipping emissions and increased hospitalizations (Tian et al., 2013). 55 56 The Port of Houston is the second largest in the United States (U.S.) by tonnage (Port of Houston, 2017), and the Gulf of Mexico has a high density of marine vessel emissions relative to 57 many other marine locations (Tournadre, 2014; Johansson et al., 2017); however, relatively little 58 research has aimed to characterize the impact of shipping emissions on Houston air quality. 59 During the Texas Air Quality Study and Gulf of Mexico Atmospheric Composition and Climate 60 61 Study 2006, measurements onboard the R/V Brown were used to characterize aerosol sources over the Gulf of Mexico (Bates et al., 2008; Russell et al., 2009). Measured submicron aerosol 62 sulfate (SO₄) mass loadings during periods of onshore flow were significantly larger than 63 expected for a marine environment, leading Bates et al. (2008) to conclude that shipping 64 65 emissions contributed heavily to total submicron aerosol mass. Russell et al. (2009) further determined that an "oil combustion/refining" organic factor accounted for 33-68% of organic 66 aerosol (OA) mass during onshore flow periods. Using a large-scale three dimensional air quality 67

large commercial shipping vessels historically have had little or inconsistent regulation in

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model, Caiazzo et al. (2013) calculated that in 2005, marine vessel emissions increased annual 68 average PM mass loadings across the Texas Gulf Coast by ~0.5 to 1 µg m⁻³, leading to 645 69 estimated premature mortalities in Texas. 70 71 Recent concerns over the health impacts of marine vessel emissions led to the establishment of the North American Emissions Control Area (ECA; U.S. Environmental Protection Agency 72 (EPA), 2010). Prior to establishment of the ECA, multiple studies demonstrated that shipping 73 74 emissions of PM were related to fuel sulfur content (FSC) (Kasper et al., 2007; Lack et al., 2009 75 and references therein), leading to the requirement that shipping vessels within 200 nautical miles of the U.S. and Canadian coast reduce their FSC from the commonly utilized 3-4% (by 76 77 mass) to only 1%. In 2015, the limit was reduced to 0.1% (Zetterdahl et al., 2017). In order to comply with these regulations, marine vessels typically switch from low-grade heavy fuel oil to 78 79 marine gas oil or marine diesel oil at the ECA boundary; however, low-FSC residual fuels have also recently become available (Wan et al., 2016; Czech et al., 2017). Numerous studies have 80 demonstrated that such fuel switching dramatically reduces emissions of SO₂, SO₄, primary OA 81 (POA), and black carbon (Lack et al., 2011; Browning et al., 2012; Zetterdahl et al., 2017). 82 Using the U.S. Interagency Monitoring of Protected Visual Environments network and 83 84 positive matrix factorization (PMF) modeling, Kotchenruther (2016) determined that the average decrease in annual PM_{2.5} (that with diameters less than or equal to 2.5 µm) from residual fuel 85 86 combustion (i.e., shipping emissions) in U.S. coastal locations due to establishment of the ECA (i.e., pre-2012 to 2016) was 74.1%. However, at three sites along the Gulf Coast (located in 87 Louisiana and Florida), the average reduction was only 35-50% (Kotchenruther, 2016). While 88 the reason for the difference between the Gulf sites and the rest of the country is currently 89 unclear, it is nevertheless evident that the implementation of the ECA may have changed 90

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91 drastically the speciation and total mass loading of aerosol over the Gulf of Mexico, presenting

92 the need for further research on this source.

Shipping emissions also may have numerous secondary effects on marine aerosol. Models indicate that shipping-related NO_x emissions likely elevate hydroxyl radical (OH) concentrations within the marine boundary layer (MBL) (Chen et al., 2005; Kim et al., 2009; Kim et al., 2013), potentially impacting the oxidation state of marine OA. Furthermore, production of the two most commonly identified components of marine secondary OA (SOA), methanesulfonic acid (MSA) and dimethyl/diethylamines (Facchini et al., 2008; Claeys et al., 2009; Rinaldi et al., 2010), may be enhanced in the presence of shipping emissions (Gaston et al., 2010; Sorooshian et al., 2015). Finally, shipping-related SO₄ should increase submicron mass loadings of aerosol liquid water (ALW), which may subsequently impact aqueous processing of water-soluble organics (Carlton and Turpin, 2013). These effects are difficult to model on a global scale due to the complexities of accurately simulating the photochemistry and physical transport of shipping plumes (Kim et al., 2009), making field measurements useful to evaluate these hypotheses.

In the present study, three weeks of coastal air measurements were performed near Houston, TX, to investigate the impact of marine vessel emissions on ambient aerosol mass and composition. Specific focus was placed on apportioning anthropogenic and biogenic sources of SO₄, attributing anthropogenic SO₄ to marine vessel emissions, investigating links between marine vessel emissions and measured OA, and exploring whether these emissions appear to influence OA composition, amine/MSA aerosol formation, or ALW.

2. Experimental Methods

2.1 Sampling Site Characterization

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Atmospheric measurements were conducted May 24 - June 14, 2016, at a private coastal home southwest of Galveston, Texas (29.074°N, 95.125°W). Figure 1 presents an overview of the sampling location. The site is approximately 75 km directly south of the Houston Ship Channel (HSC) and is therefore a similar distance from Houston's urban core. In addition, the primary inlet to Galveston Bay used for commercial shipping is about 45 km to the northeast. The nearest road, Highway 257 just north of the site, connects the cities of Galveston and Freeport, TX, and receives relatively little traffic. As a result, this location is likely to be less influenced by primary anthropogenic emissions than recent campaigns in Houston that took place closer to the urban core (Cleveland et al., 2012; Bean et al., 2016; Leong et al., 2017; Wallace et al., 2018). Instruments including a high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS, Aerodyne, Inc.) and those measuring traces gases and meteorological parameters were housed inside the University of Houston/Rice University Mobile Air Quality Laboratory (MAQL), which was stationed outside of the private home and has been described previously (Leong et al., 2017).

2.2 HR-ToF-AMS Operation

The chemical composition of non-refractory submicron PM (NR-PM₁) was determined through the use of a HR-ToF-AMS (DeCarlo et al., 2006). Numerous detailed descriptions of HR-ToF-AMS operation can be found elsewhere (DeCarlo et al., 2006; Canagaratna et al., 2007). Air flow was drawn into the HR-ToF-AMS through a 2.5-µm cut diameter Teflon®-coated cyclone located on top of the MAQL mast approximately 6 m above ground level. Incoming air is transmitted through a 100-µm critical orifice, after which particles are focused into a beam through the use of an aerodynamic lens and accelerated under high vacuum (10⁻⁵ Torr) into the sizing chamber. After passing the sizing chamber, non-refractory chemical

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components are flash vaporized at approximately 600°C and ionized at 70 eV. Ionized mass fragments are then directed into the time-of-flight mass spectral detection region. For this study, the HR-ToF-AMS was operated in V-mode (higher signal, less mass-to-charge (m/z) resolution compared to the alternative W-mode), and data were collected over 80-s intervals. A nafion dryer was placed upstream of the HR-ToF-AMS inlet to maintain a sampling line relative humidity (RH) below 40%.

2.3 HR-ToF-AMS Data Analysis

The HR-ToF-AMS data were analyzed with the SQUIRREL v 1.57I and PiKA v 1.16I (D. Sueper, University of Colorado-Boulder) software packages within Igor Pro (Wavemetrics, Inc.). The collection efficiency (CE) of the HR-ToF-AMS, which is influenced by sampling line RH as well as particle composition, was determined using the composition-dependent calculator within the SQUIRREL and PiKA software packages (Middlebrook et al., 2011). This method produced a CE of 0.5 for the majority of the campaign (89% of the time). High-resolution analysis was performed on each ion in the m/z range 10-125, and elemental analysis of organic composition was performed using the Improved-Ambient method (Canagaratna et al., 2015). The ionization efficiency of the HR-ToF-AMS with respect to nitrate (NO₃) was calibrated before and after the campaign using 350-nm ammonium nitrate (NH₄NO₃) particles following standard procedures. In order to calculate campaign-averaged detection limits, filtered air was sampled every two days for approximately 30 minutes at a time, and the detection limit was calculated as three times the standard deviation of the filter measurements. Detection limits are provided in Table S1 in the supplemental information (SI).

2.4 Positive Matrix Factorization

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Positive matrix factorization analysis (Paatero and Tapper, 1994) was performed on the high-resolution HR-ToF-AMS mass spectral dataset in order to further investigate potential sources and transformation processes of measured OA. The PMF technique has been applied extensively in urban (Ulbrich et al., 2009; Ng et al., 2010), rural/downwind (Crippa et al., 2014 and references therein), and coastal locations (Hildebrandt et al., 2010; Hildebrandt et al., 2011; Schmale et al., 2013) to characterize classes of compounds that constitute OA. The PMF model assumes that the time series of organic mass spectra can be divided into a number of temporally unvarying components. These components, defined by their fixed mass spectra, contribute varying amounts of organic mass to the total organic signal at each time. Details on PMF and the resulting factors are included in the SI.

2.5 HYSPLIT Backward Trajectory Calculation

Analysis of air mass history is often a useful tool for characterizing likely sources and processes affecting measured aerosol composition. As a result, 120-hr backward trajectories were calculated at heights of 100, 200, 300, 400, and 500 m using the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model (Draxler and Rolph, 2003) for every hour during the campaign. Meteorological data at a resolution of 1° (latitude-longitude) were obtained from the Global Data Assimilation System archive (http:www.arl.noaa.gov/ss/transport/archives.html) through the HYSPLIT software. Recorded meteorological parameters such as solar flux (W/m²), mixing layer depth, and precipitation were averaged for each trajectory to provide insight into influences of photochemistry, mixing, and possible wet deposition during transport. In addition, the overall length of each five-day trajectory was used to represent an average wind speed, as knowledge of historical wind speed is

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important for predictions concerning the influence of POA particles in ocean environments (Zorn et al., 2008; Russell et al., 2010; Ovadnevaite et al., 2011).

2.6 Weighted Potential Source Contribution Function

In order to provide further insight into likely aerosol source regions during the campaign, the weighted potential source contribution function (WPSCF) was applied to the dataset. The WPSCF combines measured mass loadings of atmospheric species with air mass trajectories to determine probable source locations. The WPSCF method has been used to study regional sources of air pollutants at different receptor sites (Hopke et al., 1995; Zhu et al., 2011; Guo et al., 2014). For this study, the WPSCF analysis utilized HYSPLIT trajectories described previously. The spatial area covered by the trajectories was divided into a grid of 0.25° x 0.25° cells, and the number of trajectory segment endpoints located in each cell for five different starting heights (100, 200, 300, 400, and 500 m) was determined. Incorporation of multiple starting heights accounts for the general clockwise rotation of air mass backward trajectories with altitude. While these cell sizes are slightly smaller than those often used (~0.5°-2°), this study was particularly focused on attribution of measured SO₄ to specific locations within the Gulf of Mexico (i.e., shipping lanes), which requires a small cell size.

In order to calculate the WPSCF value for each cell, the total potential source contribution function (TPSCF) value is first calculated and then weighted. The value of the TPSCF function for a specific grid cell (i, j) is calculated using (Hopke et al., 1995; Guo et al., 2014):

$$TPSCF_{i,j} = \frac{\sum m_{i,j}^k}{\sum n_{i,j}^k} \tag{1}$$

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where $n_{i,j}^k$ represents the total number of trajectory segment endpoints located within cell i,j for height k, while $m_{i,j}^k$ represents the number of these endpoints that also correspond to measured values of a specific species above a critical value, in this case the 75th percentile (Hopke et al., 1995; Guo et al., 2014).

In the case of highly variable air mass trajectories or strong local sources during a campaign, distant grid cells that were intersected by only a small number of trajectories may be incorrectly assumed to represent likely sources. To prevent this, a weighting function is applied to TPSCF values based on the $n_{i,j}^k$ value, with higher weight given to cells that were intersected by more trajectories. The weighting method, based on the power of the number of trajectories at a specific height, is (Guo et al., 2014)

$$W(\sum n_{i,j}^{k}) = \begin{cases} 1, & T^{0.7} < \sum n_{i,j}^{k} \\ 0.7, & T^{0.56} < \sum n_{i,j}^{k} \le T^{0.7} \\ 0.42, & T^{0.42} < \sum n_{i,j}^{k} \le T^{0.56} \\ 0.17, & \sum n_{i,j}^{k} \le T^{0.42} \end{cases}$$
(2)

212 where T represents the total number of trajectories calculated at each specific height. The

213 WPSCF value is then calculated by applying the relevant weights to each cell.

$$WPSCF_{i,j} = W_{i,j} \times TPSCF_{i,j}$$
 (3)

2.7 MSA Calibration

Methanesulfonic acid is widely regarded as a robust indicator of SOA production from marine sources (Facchini et al., 2008; Crippa et al., 2013; Schmale et al., 2013; Ovadnevaite et al., 2014). In addition, MSA is often the most abundant identifiable component of marine OA (Facchini et al., 2008; Claeys et al., 2009). Recent work has identified that MSA mass loadings

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can be quantified in near-real time by the HR-ToF-AMS provided that accurate instrument-220 221 specific calibrations are performed (Zorn et al., 2008; Ovadenvaite et al., 2014; Huang et al., 222 2017). As MSA fragments into both organic and inorganic SO₄-containing ions within the HR-ToF-AMS, accurate mass prediction requires reconstruction of the compound based on 223 knowledge of the fragmentation pattern in the specific instrument being used (Zorn et al., 2008). 224 225 As such, calibrations were performed following the procedure of Ovadnevaite et al. (2014). 226 A 0.02% aqueous solution of MSA (Sigma-Aldrich, >99.0% purity) was nebulized by a TSI, Inc. atomizer (model 3076) and passed through a differential mobility analyzer (BMI, Inc.) 227 to size select particles 300-nm in mobility diameter. These particles were then measured by the 228 229 HR-ToF-AMS. Mass spectra from two separate calibrations are provided in Figure S13 While MSA fragments into a variety of ions within the HR-ToF-AMS (CH₃⁺, CHS⁺, CH₃SO₂⁺, SO⁺, 230 231 SO₂⁺, etc.), the CH₃SO₂⁺ ion is thought to originate almost exclusively from MSA, as other organosulfate standards measured by the HR-ToF-AMS show negligible contributions to 232 CH₃SO₂⁺ (Huang et al., 2015). Therefore, MSA mass loadings during the campaign were 233 calculated based on the ratio of this ion to the total MSA mass measured during the calibrations 234 (Huang et al., 2015; Huang et al., 2017). The average ratio measured during the two calibrations 235 236 (18.18), was similar to that determined from the calibration of Huang et al. (2017) (23.81). 237 2.8 Distinction Between Anthropogenic and Biogenic nss-SO₄ 238 The MSA measurements also allow estimation of the relative contributions of biogenic and 239 anthropogenic (primarily due to shipping) sources of non-sea-salt (nss)-SO₄ in marine 240 environments. In many studies attempting to apportion the impact of shipping emissions on 241 measured aerosol mass, ratios of trace metals specific to heavy fuel oil combustion are used as tracers (Zhao et al., 2012; Viana et al., 2014; Kotchenruther, 2016). However, in cases where 242

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sulfide (DMS), should produce a latitude-specific biogenic MSA/nss-SO₄ ratio, presenting a metric to apportion biogenic and anthropogenic nss-SO₄ (Jung et al., 2014). Specifically, DMS oxidation, which primarily occurs through initial reaction with OH, can either proceed through an abstraction or addition pathway (Hynes et al., 1986). The addition pathway, which is favored at lower temperatures prevalent at higher latitudes, mainly produces dimethylsulfoxide and MSA. The abstraction pathway, favored in higher temperatures, primarily produces SO₂ and therefore eventually nss-SO₄ (Hynes et al., 1986; Jung et al., 2014). As a result, previous long distance remote trans-oceanic cruises have observed significant latitudinal gradients in the MSA/nss-SO₄ ratio in both the Atlantic and Pacific Oceans (Jung et al., 2014; Huang et al., 2017), with consistently larger values at high latitudes. As nss-SO4 measured in marine environments is often produced by a combination of anthropogenic and biogenic sources, multiple linear regression (MLR) analysis is often used to extract the biogenic MSA/nss-SO4 ratio from ambient marine aerosol data. The MLR technique assumes that marine nss-SO₄ is produced from a biogenic source, which can be traced with MSA mass loadings (used as one predictor variable), and an anthropogenic source, which can be traced using concentrations of heavy metals emitted by shipping vessels (e.g., antimony) (used as the second predictor variable) (Savoie et al., 2002). Previously published agreement between measured and predicted nss-SO₄ using the MLR method was robust ($R^2 > 0.7$) (Savoie et al., 2002). For this study, the biogenic MSA/nss-SO₄ ratio (0.053) determined by Savoie et al. (2002) using multiple linear regression (MLR) at Bermuda was utilized to apportion biogenic versus anthropogenic sources of nss-SO₄, as Bermuda is the closest location to our sampling site in terms of latitude (32°N at Bermuda versus 29°N at our sampling site). In addition to being

such data are unavailable, biogenic sulfur sources, based on the oxidation chemistry of dimethyl

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collected at the closest location to our sampling site, the ratio extracted at Bermuda is the lowest 267 reported ratio in the literature, to the authors' knowledge. A more recent study by Lin et al. 268 (2012) quantified the biogenic MSA/nss-SO₄ ratio of submicron marine aerosol using sulfur isotopic data at varying latitudes across the Atlantic Ocean. The authors reported ratios similar to 269 270 or larger than 0.053 at all sampled latitudes. As a result, the application of other published ratios will only increase the fraction of nss-SO₄ attributed to anthropogenic sources. 271 272 However, as the biogenic MSA/nss-SO₄ ratio was originally determined using samples of 273 total suspended particulate matter (i.e., no size-cutoff) application of the ratio to PM1 data should 274 produce an upper-limit estimate of the anthropogenic fraction of marine nss-SO₄. Briefly, as 275 previous field and laboratory studies have noted that MSA solubility decreases with solution 276 acidity (Kerminen et al., 1997; Jung et al., 2014), the presence of acidic sulfate aerosol can shift 277 the size distribution of MSA towards larger, more alkaline particles relative to sulfate (a stronger 278 acid) (Jung et al., 2014). This effect, if substantial, could cause the HR-ToF-AMS to report a lower observed MSA/nss-SO₄ ratio than would be observed by an instrument measuring both 279 submicron and super-micron PM. However, Saltzman et al. (1983) found that the size 280 281 distributions of MSA and SO4 measured in the Gulf of Mexico were quite similar, with around 75% of MSA and approximately 87% of nss-SO₄ contained within sub-micron particles, 282 suggesting that the overall uncertainty resulting from this effect is small. 283 284 Quantification of anthropogenic nss-SO₄ also requires that the contribution of sea salt (ss)-SO₄ be determined. Using laboratory calibrations, the ss-SO₄ mass loading measured by the 285 HR-ToF-AMS when sampling a sea-salt standard (Lake Products Co., ASTM D1141), is 286 287 approximately $26 \pm 2\%$ of the corresponding chloride mass loading. Using this ratio and measured chloride mass loadings during the campaign, ss-SO₄ contributed only $0.4 \pm 0.4\%$ of the 288

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total SO₄ mass loading. Therefore, to produce an estimate of anthropogenic nss-SO₄, the HR-ToF-AMS estimate of MSA is divided by the biogenic MSA/nss-SO₄ ratio published by Savoie et al. (2002) to produce a "biogenic" mass loading of nss-SO₄. This amount is subtracted from the total nss-SO₄ measured by the HR-ToF-AMS, and the remaining nss-SO₄ is assumed to be anthropogenic. Multiple lines of evidence described in Section 3.2 support the use of this technique.

2.9 Ancillary Measurements

A variety of trace gases and meteorological parameters were measured during the campaign. All trace gas and meteorological data were measured with a 5-minute averaging time. Individual NO_x species (nitric oxide and nitrogen dioxide (NO₂)) and total reactive gas-phase nitrogen were measured using high sensitivity chemiluminescence monitors (AQD, Inc.). Ozone (O₃) mixing ratios were measured with an ultraviolet absorption instrument (2BTech, Inc., model 205), and carbon monoxide mixing ratios were measured using high-resolution cavity enhanced direct-absorption spectroscopy (Los Gatos Research, Inc.). Sulfur dioxide was measured with a pulsed fluorescence analyzer (ThermoFischer Scientific, model 43i). Ambient temperature, pressure, wind speed, and wind direction were measured using an RM Young meteorological station.

3. Results and Discussion

3.1 Campaign Overview

Figure 2 displays the speciated aerosol mass loadings, PMF factor contributions, important trace gas concentrations, and meteorological conditions encountered during the campaign. Overall, the average NR-PM₁ mass loading was 4.66 ± 3.17 (one standard deviation) $\mu g m^{-3}$ and was dominated at times by either SO₄ (44% on average) or OA (42%). As the measurements

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were performed in the early summer in close proximity to the coast, RH was relatively high 311 312 (average 81%), conditions were generally sunny, and temperatures were warm (average 27.3°C). Examination of the aerosol time series data reveals three distinct period types. Marine periods 313 are characterized by consistent on-shore flow, while continental periods are characterized by off-314 shore flow or daily land- and sea-breeze circulation patterns. A three-day period influenced by 315 the passage of two cold fronts and a low pressure (LP) system that produced heavy cloud cover, 316 317 intermittent rain, and a distinct aerosol diurnal profile was termed "Frontal/LP." Each of these periods contained a unique dominant PMF factor resembling low-volatility 318 oxygenated organic aerosol (OOA) (Ng et al., 2010), denoted as OOA-1, OOA-2, or OOA-3 319 320 (Figure 3). An overview of the average aerosol and trace gas characteristics during each period is provided in Table 1, and a comparison to previous campaigns in the Houston region is shown in 321 322 Figure S14. While the extracted PMF factors are highlighted briefly below and summarized in Table 2, more detailed factor descriptions are included in the SI. 323 324 The majority of the campaign (~12 days total), characterized by onshore flow conditions with 325 wind directions generally between 120° and 240°, was classified as "marine." During these 326 periods, which encompass 5/24-6/1 and 6/10-6/14, aerosol mass loadings were relatively stable 327 from day-to-day. Interestingly, average observed mass loadings were much larger in the first portion of the marine period $(4.69 \mu g \text{ m}^{-3}) (5/24-6/1)$ than in the second $(2.71 \mu g \text{ m}^{-3}) (6/10-6/1)$ 328 6/14), despite similar local wind direction, O₃, and meteorological conditions, implying that air 329 mass history has a large influence on marine aerosol loadings. The observation of SO₄ mass 330 loadings much larger than 1 µg m⁻³, which is generally the maximum observed in remote marine 331 locations, even during periods of high biological activity (Zorn et al., 2008; Rinaldi et al., 2010; 332 Schmale et al., 2013; Ovadnevaite et al., 2014), supports a major anthropogenic aerosol source in 333

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measured during onshore flow by Bates et al. (2008) (3 µg m⁻³), despite recent regulations on shipping emissions, while measured OA mass loadings were larger during this study (0.72 µg m⁻¹ ³ versus 0.38 µg m⁻³ from Bates et al. (2008)). The OA, which constituted 21% of total mass, was highly oxidized (average oxygen to carbon ratio, O:C = 0.73), consistent with previous measurements of marine aerosols (Russell et al. 2009; Chang et al., 2011; Schmale et al., 2013). The average mass fraction of m/z 44 (f₄₄), a metric used to describe the extent of OA oxidation, was 0.15, a value very similar to that observed by Russell et al. (2009) during marine flow conditions (0.16), suggesting that on average, the oxidation state of marine OA over the Gulf of Mexico has not changed substantially since ECA implementation. The light winds observed during the campaign suggest that little of the measured marine OA was the result of organic-enriched sea spray, as this production pathway generally requires significant white-cap coverage, which is typically only observed above wind speeds of 7-8 m/s (Gantt et al., 2011; Shank et al., 2012; Ovadnevaite et al., 2011; Schmale et al., 2013; Frossard et al., 2014). Local wind speeds were virtually never above 8 m/s (Figure 2), and 5-day averaged wind speeds calculated using total trajectory lengths were only >8 m/s for 4% of the marine period. Potential major sources of OA therefore include secondary production through processing of biogenic volatile organic compounds (VOCs), as well as primary and secondary production from shipping emissions (Lack et al., 2009; Coggon et al., 2012). This hypothesis is supported by the fact that marine OA composition was dominated by a highly oxidized PMF factor, OOA-3 (O:C = 0.77; 55% of OA on average) (Figure S5), that was moderately correlated with SO_4 ($R^2 = 0.55$) and displayed little diurnal variation.

the Gulf of Mexico. The average mass loading of SO₄ plus NH₄ (3.04 µg m⁻³) was similar to that

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Two storms occurred during the sampling campaign. The first (5/27) caused a loss of power to the HR-ToF-AMS and the data gap shown in Figure 2; the second, denoting the beginning of the "Frontal/LP" period, caused a rapid reduction in aerosol mass that was followed by three days of markedly different aerosol characteristics, despite initially similar wind directions. Diurnal profiles of virtually all NR-PM₁ species during the frontal/LP period are distinct from the preceding marine period (Figure 3) and show maximum concentrations at night. Satellite images of the area show the arrival of a large-scale frontal system on 6/2 and the presence of heavy cloud cover through 6/5 (Figure S15). The O:C ratio during this period is the highest of the campaign, which, combined with the strong correlation between diurnal trends of OA and SO_4 ($R^2 = 0.78$) suggests measured OA represents regional background OA that is diluted with the rise of the boundary layer in the morning. The dominant PMF factor extracted during this period (OOA-1) had an O:C ratio (1.15) similar to the most aged OA observed in urban areas (Hayes et al., 2013) implying an influence of either extensive atmospheric processing during transport (Ortega et al., 2016), aqueous processing of highly oxidized water soluble organics (e.g., glyoxal O:C = 1) (Chhabra et al., 2010), or some combination of the two. The third identified period, which occurred from 6/6-6/9, shows evidence of continentally influenced air masses and a multi-day increase of NR-PM₁ following passage of the frontal system. The organic to SO₄ ratio shifts from a value of 0.34 during the marine period, typical of marine environments (Coggon et al., 2012), to an average value of 3.08, highlighting the predominance of OA sources within the Houston region. Local wind direction measured from 6/6-6/8 appears to show a land-sea breeze type circulation pattern, and midday O₃ concentrations during this period reach the highest levels of the campaign (Figure 2). Diurnal profiles of NR-PM₁ species highlight the influence of local photochemistry on aerosol formation (Figure 3).

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OOA-2, the dominant PMF factor during this period (72%), displays a photochemical dependence similar to previously extracted OOA factors in Houston's urban core (Cleveland et al., 2012) but is much more oxidized (O:C = 0.79 in this study versus 0.46 in Cleveland et al. (2012)), highlighting the effect of aging during transport. Plotting mass loadings of OOA-2 against ambient CO concentrations produces a slope of ~150 µg m⁻³/ppmv during this period. This value is similar to previous aircraft measurements of aged industrial plumes in Houston (Bahreini et al., 2009; Wood et al., 2010). Modeling results have suggested that biogenic VOCs contribute little OA during Houston industrial plume transport (Bahreini et al., 2009) except in the case of advection into the forested north of Houston (Brown et al., 2013), which suggests a likely anthropogenic origin of OOA-2.

3.2 Analysis of MSA Mass Loadings

The time series of calculated MSA mass loadings is shown in Figure 4, as are concentrations determined for the three distinct periods described previously and comparisons with literature values. Overall during the marine period, MSA mass ranged from ~0 to 0.07 µg m⁻³ and showed moderate correlation with nss-SO₄ ($R^2 = 0.46$) and weak correlation with OA ($R^2 = 0.12$), suggesting major additional sources of both nss-SO₄ and OA over the MBL. While previous MSA measurements in the Gulf of Mexico are sparse, Saltzman et al. (1983) recorded submicron mass loadings of 0.022-0.066 µg m⁻³ in Miami, in agreement with our results. In addition, our results align well with previous submicron measurements taken at lower latitudes in both the Atlantic and Pacific Oceans, as well as with measurements taken at higher latitudes while sampling tropical air masses (Figure 4) (Zorn et al., 2008; Ovadnevaite et al., 2014; Huang et al., 2017).

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On average, MSA accounts for only 3.2% of submicron OA during marine periods, a value 401 402 much lower than observed in previous coastal measurements with a HR-ToF-AMS. For instance, 403 Crippa et al. (2013) reported that MSA accounted for approximately 20% of submicron OA in Paris when air masses traveled from marine locations. At Mace Head, Ireland, MSA represented 404 12.5-18% of submicron OA during May and June when air masses traveled from the tropics 405 (Ovadnevaite et al., 2014). However, before establishment of the ECA, Russell et al. (2009) 406 407 found that during onshore flow in the Gulf of Mexico, between 52 and 89% of organic mass 408 could be attributed to oil combustion/refining and wood smoke-related sources. Therefore, the small MSA mass fraction observed here is likely the result of strong remaining anthropogenic 409 OA sources over the Gulf. This hypothesis is supported by the relatively weak correlation 410 between the dominant marine PMF factor (OOA-3) and MSA ($R^2 = 0.41$). For comparison, the 411 distinctly biogenic marine PMF factor extracted by Crippa et al. (2013) in Paris correlated 412 strongly with MSA ($R^2 = 0.84$). 413 Quantification of the MSA/OA ratio permits a rough calculation of the contribution of 414 biogenic sources to total marine OA. Using an assumption that MSA should only represent 5-415 10% of total biogenic OA under pristine conditions over the Gulf of Mexico (a low estimate 416 based on the previous observations discussed), calculation of the mass fraction of biogenic OA 417 based on this ratio (MSA/Bio. OA = 0.05-0.1) and the measured MSA/Total OA ratio 418 419 (MSA/Total OA = 0.032) implies biogenic sources only produce ~32-64% of total measured OA. 420 In addition, vanadium, an element common to shipping emissions, is thought to act as a catalyst to MSA formation (Gaston et al., 2010). As oil combustion emissions were responsible for a 421 major fraction of OA over the Gulf of Mexico in the past (Russell et al., 2009), this type of 422 423 catalytic process may be enhancing MSA production relative to more pristine locations at similar

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latitudes. This further suggests that the assumption that MSA accounts for only 5-10% of biogenic OA is likely a low estimate.

MSA mass loadings were positively, though only slightly, correlated with trajectory-averaged solar flux ($R^2 = 0.12$) and negatively correlated with trajectory length (i.e., wind speed) ($R^2 = 0.16$). The lack of a strong correlation with these parameters is partly due to the fact that DMS is emitted primarily in regions with high concentrations of biological organisms in the seasurface layer, which typically occur close to the coast. Therefore, emissions are not uniform across the Gulf of Mexico (Sorooshian et al., 2009). Often, high MSA mass loadings are linked to specific locations of high biological activity through analysis of backward trajectories and comparison to chlorophyll-a levels (Sorooshian et al., 2009; Gaston et al., 2010; Schmale et al., 2013; Sorooshian et al., 2015; Huang et al., 2017). While the accuracy of satellite-derived measures of chlorophyll-a as an indicator of DMS production potential is still under debate (Sorooshian et al., 2009; Huang et al., 2017), the data here support a link between oceanic chlorophyll-a and MSA mass loadings, as a peak in MSA mass is observed on 6/11, when backward trajectory analysis indicates air masses slowly traveled over the nutrient-rich waters close to the coast and near the mouth of the Rio Grande River (Figure S16).

3.3. Quantifying Anthropogenic Contributions to Marine Aerosol Mass

The average MSA/nss-SO₄ ratio measured during the marine period was 0.012. Applying the biogenic MSA/nss-SO₄ ratio determined by Savoie et al. (2002) indicates that an average of 77% of nss-SO₄ (1.8 μ g m⁻³) is the result of anthropogenic sources during onshore flow (Figure 4c). This value likely represents an upper limit (see Section 2.8). Furthermore, as the partitioning of gaseous ammonia (NH₃) to the aerosol phase is driven by the neutralization of acidic SO₄, mass loadings of nss-SO₄ and NH₄ are highly correlated during the marine period (R² = 0.97). As a

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NH₄/nss-SO₄ ratio and the calculated anthropogenic fraction of nss-SO₄. The classification of a fraction of NH₄ as "anthropogenic" therefore refers to the necessity of an anthropogenic species (in this case nss-SO₄) for the production of NH₄ aerosol, rather than anthropogenic NH₃ emissions. By applying this method and combining anthropogenic NH₄ and nss-SO₄ mass loadings, anthropogenic sources contribute 73% of total inorganic NR-PM₁ (2.3 µg m⁻³) on average during marine flow conditions. Multiple lines of evidence support the use of MSA measurements coupled with the biogenic MSA/nss-SO₄ ratio to apportion anthropogenic and biogenic nss-SO₄. Using measurements of chlorophyll-a concentrations, wind speeds measured onboard the R.V. Brown, and the wind speed/transfer velocity relationship determined by Nightingale et al. (2000), Bates et al. (2008) estimated that the DMS flux from the Gulf of Mexico was capable of producing between 0.2 and 0.4 µg m⁻³ of biogenic nss-SO₄. For comparison, the average mass loading of biogenic nss-SO₄ calculated using the biogenic MSA/nss-SO₄ ratio during marine periods in our study is 0.54 µg m⁻³, in relatively good agreement with those results. Furthermore, Figure 5 shows the WPSCF analysis of anthropogenic nss-SO₄ and MSA, the Automated Mutual Assistance Vessel Rescue System (AMVER) shipping spatial proxy map (Wang et al., 2008), and chlorophyll-a levels observed by the MODIS satellite. The use of the biogenic MSA/nss-SO₄ ratio is qualitatively supported by the relatively distinct WPSCF results of anthropogenic nss-SO₄ and MSA and by the agreement between the anthropogenic nss-SO₄ WPSCF map and the region of high shipping traffic indicated by the AMVER inventory. The high probability region of anthropogenic nss-SO₄ is located predominately outside of the ECA boundary where shipping lanes converge, while

result, an estimate of an "anthropogenic" mass loading of NH₄ can be calculated based on the

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the MSA high probability region is largely within the ECA where surface chlorophyll-a 469 470 concentrations are elevated (Figure 5). 471 While point-source emissions in Florida or long-range transport could contribute to the 472 anthropogenic nss-SO₄ measured during this study, further analysis suggests these sources are minor in comparison to marine vessel emissions. According to the National Emissions Inventory 473 (NEI), ~160,000 tons of SO₂ were emitted in Florida in 2014 (U.S. EPA, 2014). However, only 474 475 ~30,000 tons (~19%) were emitted in the southern peninsular region indicated as a potential source by the WPSCF analysis (south of 28°N) (Figure S17). While point-source distributed NEI 476 477 data are not yet available for 2016, EPA statewide average data suggest that Florida SO2 478 emissions were approximately half of those in 2014 (~80,000 tons), with the change almost entirely due to a 75% reduction in emissions from electricity generating stations (U.S. EPA, 479 480 2017). If emissions from individual electricity generating stations south of 28°N have been similarly reduced, only ~20,000 tons of SO₂ were emitted in the southern peninsular region in 481 482 2016. For comparison, recent emissions inventories predict that marine vessels emit as much as 75,000 tons of SO₂ annually in the Gulf of Mexico after accounting for the ECA, nearly four 483 times as much as the geographically relevant Florida emissions (Johansson et al., 2017). 484 485 In terms of the contribution from long-range transport, air masses that originated in Europe 486 or Africa required 15 days of transit or more to reach the measurement site based on HYSPLIT modeling. Assuming that sulfur compounds have a lifetime of ~5-7 days in the MBL (Faloona, 487 2009), 89-95% of the original sulfur in these air masses would be lost prior to measurement. This 488 agrees with the finding by Bates et al. (2008) that only a small fraction of SO₄ measured in the 489 Gulf of Mexico was contributed by African dust during measurements in 2006. 490

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Volcanic sources of SO₂ have occasionally contributed significantly to nss-SO₄ in marine regions during previous campaigns (Jung et al., 2014). Any nss-SO₄ produced by volcanic emissions would be apportioned to anthropogenic sources due to the apportionment technique used (i.e., volcanoes would not be expected to produce substantial MSA, leading to a depression of the MSA/nss-SO₄ ratio). Therefore, the influence of volcanic emissions would incorrectly increase the fraction of measured nss-SO₄ attributed to anthropogenic sources. However, the only relevant volcanoes in the area are along the Caribbean islands, and backward trajectory analysis reveals that the largest measured mass loadings of nss-SO₄ correspond to air masses that passed far north of them. It therefore appears that the vast majority of measured anthropogenic nss-SO₄ was emitted by marine vessels rather than other sources.

These results contrast with those from the previous model study of Lauer et al. (2007), who predicted using a global model that shipping contributes only ~30% of sub-micron SO₄ over the Gulf of Mexico using the AMVER-distributed shipping inventory from Eyring et al. (2005) and as little as 15% or less using the International Comprehensive Ocean-Atmospheric Dataset-distributed inventory from Corbett and Kohler (2003) on an annual basis. Multiple lines of evidence suggest that the discrepancies observed between our results and previous modeling results are not simply due to the timing of our measurements. For instance, while less shipping-related SO₄ is likely produced in the fall/winter due to the reduction in photochemical activity during that time, conversion of biologically-emitted SO₂ into nss-SO₄ should have the same photochemical dependence. Furthermore, the SO₂ yield from DMS oxidation, the major biological nss-SO₄ production pathway, is reduced in the winter due to the temperature dependence of DMS oxidation chemistry, as previously explained (Section 2.8) (Jung et al., 2014). Finally, data from the Port of Houston suggests that shipping traffic (estimated by the

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number of twenty-foot equivalent cargo units (TEUs) processed at the port) is only reduced by 514 ~10% in the winter (Figure S18) (Port of Houston, 2017). However, a portion of this discrepancy 515 516 is likely attributable to the fact that the marine period of our study only encompasses onshore flow conditions, whereas the annual average calculated by Lauer et al. (2007) also incorporates 517 periods of offshore flow, when continental emissions act as source of nss-SO₄ to the MBL over 518 519 the Gulf of Mexico. 520 Quantification of anthropogenic nss-SO₄ allows for a more detailed apportionment of marine OA than was possible based on MSA alone. While the correlation between total nss-SO4 and 521 OOA-3 (the dominant marine OA factor) is moderate ($R^2 = 0.55$), anthropogenic nss-SO₄ is 522 strongly correlated with OOA-3 ($R^2 \ge 0.78$) (Figure S12), suggesting OOA-3 is coupled to 523 shipping emissions either directly (e.g., SOA from marine vessel VOCs) or indirectly (e.g., 524 525 increased uptake of water soluble gases over the MBL due to increased ALW). Substantial processing of OOA-3 during transport leads to the removal of major mass spectral tracers; 526 however, there is some evidence for a contribution from naphthalene OA (discussed in the SI), 527 which is the dominant commonly-measured VOC emitted by major commercial shipping vessels 528 (Agrawal et al., 2008; Murphy et al., 2009; Czech et al., 2017). Assuming, as a strictly upper 529 530 bound estimate, that OOA-3 production is entirely dependent on shipping emissions, anthropogenic sources contributed 71% of total NR-PM₁ (2.7 µg m⁻³) on average during the 531 marine period. 532 Comparing the submicron mass loadings of nss-SO₄ and NH₄ measured during marine 533 periods by Bates et al. (2008) (pre-ECA) (3 µg m⁻³) to those measured during our study (3.04 µg 534 535 m⁻³) suggests ECA implementation has had a negligible effect on aerosol mass. However, the 536 amount of shipping traffic within the Gulf of Mexico, estimated with the total number of loaded

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TEUs processed at the ports of Houston, TX, Galveston, TX, Freeport, TX, New Orleans, LA, and Mobile, AL has increased by approximately 42% since 2006 (Figure S19) (U.S. Army Corps of Engineers, Navigation Data Center, 2016), suggesting that emissions reductions per vessel within the ECA boundary may have been offset by increased traffic. On a yearly basis, the estimated increase in shipping traffic since 2006 (4.6% per year) is similar to the annual growth in seaborne trade observed between 2002 and 2007 (5.2%) and is within the range of growth predicted through 2050 (3.6-5.9%) (Corbett et al., 2007; Eyring et al., 2010). In support of these rapid growth estimates, Tournadre (2014) recently concluded that shipping traffic in the Atlantic Ocean nearly doubled between 2006 and 2012, corresponding to an average annual increase of ~8%. While it is also possible that the specific meteorological conditions encountered during this campaign (i.e., air mass trajectories, average wind speeds, etc.) were more conducive to the accumulation of anthropogenic nss-SO₄ than during the study of Bates et al. (2008), this is unlikely to be the dominant reason for the little change observed since ECA implementation. Therefore, our results suggest that the ECA has reduced shipping emissions on a per vessel basis, as there has been little change in shipping-related aerosol despite significant growth in the shipping trade. However, these results also provide justification for further limits on FSC, which are expected to be implemented in 2020 and require a reduction of FSC to a maximum of 0.5% globally (Kotchenruther, 2016).

3.4 Relationship Between Shipping Emissions and OA Oxidation State

In order to obtain a quantitative measure of the difference in OA composition between air masses influenced by shipping emissions and those lacking such influence, a 12-hour period was isolated on 6/10-6/11 when the site encountered air masses that had been inside the ECA boundary but over the ocean (i.e., within 200 nautical miles of the coast) for virtually their entire

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five-day history (Figure S20). Assuming ECA compliance, these air masses should receive only a small fraction of the particulate, SO₂, and NO_x emissions encountered by those originating outside the boundary (Lack et al., 2009; Lack et al., 2011; Browning et al., 2012). Based on air mass history and the accompanying mass spectral analysis described below, we classified OA measured during this period as "marine-biogenic." For comparison, we distinguished a second 24-hour period (5/30) that had similar 5-day backward-trajectory-averaged meteorological conditions to the biogenic period (faster average wind speed and comparable average solar flux) but had trajectories that originated outside the ECA boundary and passed through the high intensity shipping region. The shipping-influenced period had notably larger mass loadings of anthropogenic nss-SO₄ (2.24 versus 1.09 µg m⁻³) and OA (1.04 versus 0.288 µg m⁻³). Figure 6 presents the average OA mass spectra determined for each of these periods. In the shipping-influenced air masses, measured OA is highly processed, with a much larger f₄₄ (0.20) (a marker of carboxylic acids) than is typical of marine biogenic OA (~0.08-0.14) (Chang et al., 2011; Coggon et al., 2012; Crippa et al., 2013; Coggon et al., 2014) and a composition dominated by oxygenated species (66%). In contrast, OA measured during the period of minor shipping influence is notably less aged and contains numerous indicators of a marine biogenic source. For instance, prominent non-oxygenated spectral fragments are observed at m/z 27, 39, 41, 43, 55, and 67 (Figure S21) implying the presence of alkenes, cycloalkenes, cycloalkanes, and dienes, in agreement with Ovadnevaite et al. (2011, 2014) for marine OA measured at Mace Head, Ireland and by Bates et al. (2012) in physically generated sea spray aerosol. A relatively significant contribution from m/z 79 (CH₃SO₂) (~1%) is also apparent, and as a result MSA contributes 9.3% of total OA, a value three times larger than the average during the marine period, and in closer agreement with previous measurements in remote marine regions

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(Ovadnevaite et al., 2014). Furthermore, a prominent signal from the CHO+ ion, an aldehyde tracer, is observed (~7%), which is uncharacteristic of aged urban emissions (Ng et al., 2010) but has been observed in the mass spectra of numerous biogenic SOAs from both chamber experiments and ambient measurements (Shilling et al., 2009; Chhabra et al., 2010; Slowik et al., 2010; Setyan et al., 2012) and from marine biogenic OA specifically (Chang et al., 2011; Crippa et al., 2013; Coggon et al., 2014). Ultimately, the biogenic period spectra correlates well with the marine biogenic factor extracted by Chang et al. (2011) over the Arctic Ocean ($R^2 = 0.78$) as well as with the marine OA factor extracted by Crippa et al. (2013) in Paris ($R^2 = 0.68$), while the shipping-influenced period spectra correlates extremely well with the continental factor extracted by Chang et al. (2011) ($R^2 = 0.95$). The mass spectra from the shipping-influenced period has notably larger signals from m/z 44 and m/z 28 than the biogenic period, suggesting a larger amount of atmospheric processing that converted OA components into organic acids (Chhabra et al., 2011). Numerous remote marine studies have shown that on average, the oxidation state of marine aerosol varies only slightly in the absence of anthropogenic influences (Gantt and Meskhidze, 2013; Wozniak et al., 2014). In this case, the absolute difference in the O:C ratio between the two scenarios is 0.29 (0.90 for the shipping-influenced period versus 0.61 for marine-biogenic), implying a major impact of shipping on related OA chemical and potentially physical properties. While primary marine aerosol particles can have high O:C ratios (~1) due to the significant mass fraction of carbohydrate components in dissolved organic matter (Russell et al., 2010), the low trajectoryaveraged wind speeds and high f₄₄ suggest that OA measured during the shipping-influenced period is not primary (Frossard et al., 2014).

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Using the function developed by Duplissy et al. (2011) to describe the relationship between OA oxidation state (represented by the mass fraction of m/z 44) and hygroscopicity, the calculated hygroscopic growth factor (Korg) for the shipping-influenced period is three times larger (0.31 versus 0.101) than that calculated for the marine-biogenic period. Therefore, despite the fact that freshly emitted in-plume shipping aerosol is thought to have a suppressed hygroscopic growth factor relative to background marine aerosol (Murphy et al., 2009), our results suggest that extensive aging during transport near shipping lanes (presumably due to increased oxidant levels) may lead to an eventual increase in bulk marine OA hygroscopicity relative to aerosol unaffected by shipping emissions. This hypothesis is supported by the relatively strong correlation observed between daily anthropogenic nss-SO₄ and the organic hygroscopicity factor ($R^2 = 0.64$) calculated using the Duplissy et al. (2011) method during the marine period (Figure 7). Figure 8 displays marine OA plotted on the f₄₄ versus f₄₃ triangle diagram (Ng et al., 2010) to describe OA aging. Less oxidized OA typically occupies a wide space at the bottom of the plot, indicative of variable ambient OA mass spectra, while aging causes movement diagonally upward, as mass spectra become more similar with age (Ng et al., 2010). Figure 8 highlights that OA oxidation is greatly influenced by a combination of physical air mass history and meteorology. Three specific days demonstrate these influences particularly well. On 5/24, backward trajectory analysis reveals that air masses passed directly over the region of major shipping influence, resulting in a substantial amount of nss-SO₄ aerosol and highly oxidized OA. In contrast, on 6/11, despite the fact that trajectory-averaged wind speeds were lower and solar flux was comparable, suggesting meteorological conditions were more conducive to OA processing and elevated aerosol mass loadings, air masses largely missed the high intensity

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629 oxidized OA. On 6/13, arriving air masses had faster average wind speeds and avoided shipping lanes, resulting in an extreme case of very little nss-SO₄ and only minor processing. 630 631 There are multiple ways in which the presence of shipping emissions could increase the rate of OA processing. While peak daytime concentrations of OH of 6 x 10⁶ - 1 x 10⁷ mol cm⁻³ are 632 relatively consistent throughout the clean MBL (Raper et al., 2001; Vaughan et al., 2012), 633 634 modeling results by Chen et al. (2005) and Kim et al. (2013) indicate that within individual shipping plumes, OH concentrations are elevated by a factor of 1.2 to 2.7, and OH 635 concentrations can remain elevated up to 140 km behind an individual shipping vessel. 636 637 Significant NO₂ levels within the plume also increase concentrations of nitrate radical to several pptv, even during the daytime, which would hypothetically result in rapid oxidation of any 638 639 unsaturated VOCs or components of primary marine OA (Myriokefalitakis et al., 2010; Bates et al., 2012; Kim et al., 2013). Additionally, elevated production of nss-SO₄ aerosol increases 640 641 ambient ALW mass, increasing the partitioning medium available to small, water-soluble organic gases (WSOG) produced from both biogenic and anthropogenic sources (i.e., glyoxal, 642 methylglyoxal, acetaldehyde, etc.) and processed in the aqueous phase into highly oxidized 643 644 species (such as glyoxylic acid/glyoxylate, O:C = 1.5, or oxalic acid/oxalate, O:C = 2) (Ervens et al., 2011; Ge et al., 2012). 645 3.5 Relationship Between Shipping Emissions and Major Marine OA Components: 646 647 **Amines and MSA** 648 While MSA and alkyl-amines, such as dimethyl-amine (DMA) and diethyl-amine 649 specifically, are frequently observed over the MBL and are linked to biogenic emissions (Murphy et al., 2007; Facchini et al., 2008; Sorooshian et al., 2009), the partitioning dynamics of 650

shipping region (and remained largely within the ECA), resulting in less nss-SO₄ and less-

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each are influenced by shipping emissions. For instance, recent single particle measurements in California reveal a possible catalytic role of vanadium in MSA formation (Gaston et al., 2010), while gaseous alkyl-amines typically undergo neutralization reactions with sulfuric or nitric acids to form aminium salts (Murphy et al., 2007). However, previous studies have produced conflicting results about whether biogenic marine SOA mass is maximized in clean or polluted environments. For instance, Sorooshian et al. (2009) and Facchini et al. (2008) both noted that mass loadings of amines and MSA were largest in clean rather than polluted air masses, supporting their attribution to biogenic sources; however, Sorooshian et al. (2015) observed similar size distributions of MSA and vanadium along the California coast, while Youn et al. (2015) reported noticeable long-term correlations between amines and SO₄. Myriokefalitakis et al. (2010) suggested that on a global basis, modeled marine SOA originates almost entirely from either DMS oxidation (i.e., MSA-related) (~78%) or formation of dialkyl amine salts (~21%), highlighting the importance of understanding anthropogenic influences on their production in areas influenced heavily by ship traffic. To quantify a lower-bound ambient amine signal from this coastal dataset, individual mass spectral fragments typical of alkyl amines identified in previous HR-ToF-AMS studies, specifically those at m/z 27 (CHN), 30 (CH₄N), 44 (C₂H₆N), 56 (C₃H₆N), 58 (C₃H₈N), and 72 (CH₄N₄), were combined (Murphy et al. 2007; Hildebrandt et al., 2011; Sun et al., 2011). Figure 9 highlights that hourly-averaged amine mass loadings correlate well with anthropogenic nss-SO₄ ($R^2 = 0.63$) while MSA mass loadings show a noticeably weaker relationship ($R^2 = 0.30$). A strong correlation between MSA and anthropogenic nss-SO₄ would indicate that either the biogenic nss-SO₄ fraction had been under-predicted or that a strong catalytic effect on MSA production was occurring.

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The correlation between anthropogenic nss-SO₄ and amines is consistent with those observed by Youn et al. (2015) for DMA and SO₄ in Tucson, AZ, in 2013 ($r \ge 0.72$). Amines also display a positive relationship with NH_4 ($R^2 = 0.61$, similarly to nss-SO₄), in agreement with the fact that throughout the campaign, NR-PM1 was never fully neutralized by the small ammonia sources that exist over the MBL. This is highlighted by the fact that the average neutralization ratio, the molar ratio of ammonium to the sum of sulfate and nitrate ($[NH_4^+]/(2 \times [SO_4^{2-}] + [NO_3^-])$, was only 0.74, resulting in a consistent pathway for amine SOA formation through aqueous dissolution and partial neutralization of the acidic nss-SO4 aerosol. Furthermore, the correlation between amines and anthropogenic nss-SO₄ is much stronger than correlations with average wind speed, solar flux, and mixing layer depth ($R^2 = 0.06$, 0.17, and 0.06 respectively), suggesting that anthropogenic emissions play a larger role in amine aerosol formation than meteorology. The link between shipping emissions and amine formation also is supported by the high nitrogen to carbon ratio (N:C) of the dominant marine PMF factor, OOA-3 (N:C = 0.074), a value larger than that observed in aged marine OA (N:C ~0.04) (Schmale et al., 2013) and amine-related urban PMF factors extracted in Pasadena, CA (N:C = 0.052) (Hayes et al., 2013) and New York City (N:C = 0.053) (Sun et al., 2011), but similar to a biogenic MSA-related factor extracted at Bird Island near Antarctica (N:C = 0.08) (Schmale et al., 2013). It is likely that this anthropogenic-biogenic link can be extrapolated to other areas where marine biogenic and shipping emissions coexist. As a result, amines measured in heavily trafficked marine environments should not be interpreted exclusively as products of a purely biogenic SOA formation pathway.

3.6 Anthropogenic ALW and Potential Influences on SOA formation

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Despite the fact that shipping emissions produce substantial amounts of hygroscopic SO₄ 697 698 aerosol, there have been few measurement-based predictions of the role of shipping on the 699 production of ALW in coastal marine environments. In addition, organic gases capable of 700 partitioning to ALW are present throughout the MBL (Sinreich et al., 2010), have elevated 701 concentrations near the coasts (Fu et al., 2008; Fu et al., 2011), and contribute significantly to aerosol mass in both rural and urban areas within the southeastern United States (Li et al., 2015). 702 703 As a result, measurement-based modeling of ALW is needed to inform understanding how future 704 changes to shipping sulfur emissions may influence SOA formation in coastal environments. On average, ALW mass loadings of $5.21 \pm 4.62 \,\mu g \, m^{-3}$ are modeled using ISORROPIA II 705 706 (Fountoukis and Nenes, 2007) during the marine period, representing on average 58% of total 707 NR-PM₁ particle mass. This value is slightly larger than the average determined for the HSC region in September-October 2006 by Nguyen et al. (2016) (4.6 µg m⁻³), presumably because of 708 higher average RH along the coast and similar total inorganic mass loadings, and is larger than 709 710 the average values reported for every major city in North America analyzed by Nguyen et al. (2016). If anthropogenic nss-SO₄ were eliminated completely, average ALW mass loadings 711 associated with NR-PM₁ aerosol would ultimately be reduced by 66.4%. As a result, the majority 712 713 of NR-PM₁-associated ALW over the Gulf of Mexico appears to be controllable. While concentrations of WSOG over the MBL are relatively small (Sinreich et al., 2010), advection of 714 this ALW inland may have large impacts on nearby SOA formation where precursor sources are 715 more prevalent. 716 717 Multiple modeling studies have suggested that small WSOG, specifically glyoxal, methylglyoxal, and isoprene epoxides (IEPOX), contribute heavily to SOA mass in the Houston 718 region (Li et al., 2015; Ying et al., 2015). Li et al. (2015) found that these three compounds were 719

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responsible for nearly 80% of total SOA mass loadings at a downtown site during a simulated period in 2006, with the largest fraction (~30-50%) contributed by IEPOX. The authors also showed that these species dominate SOA mass across the Gulf Coast, with significant total loadings (~4 µg m⁻³ or greater) from the southern end of Texas to the Florida panhandle. Ying et al. (2015) used the Community Multi-scale Air Quality model (CMAQ) to characterize biogenic and anthropogenic contributions to glyoxal and methylglyoxal SOA and found that isoprene is a major contributor to both (47% of glyoxal and 82% of methylglyoxal SOA, specifically). As these compounds are precursors to aqueous SOA (aqSOA) formation, anthropogenic impacts on ALW represent another potential anthropogenic-biogenic link in SOA production (Carlton and Turpin, 2013).

The aqSOA formation from these WSOG ultimately depends on both uptake into ALW and subsequent reactions to produce low-volatility organic acids or high-molecular weight oligomeric products (McNeill, 2015). However, as uptake of OH, the dominant aqueous phase oxidant, is typically surface-limited (Ervens et al., 2014), large scale models often simplify this process by assuming that aqSOA formation is irreversible and surface-controlled (Li et al., 2015; Ying et al., 2015), representing SOA production rate by

$$\frac{dM_{a,i}}{dt} = \frac{1}{4}\gamma_i v_i A M_i \tag{4}$$

where $M_{a,i}$ is the aerosol-phase mass concentration of species i (µg m⁻³), γ_i is its reactive uptake coefficient, v_i is its gas-phase thermal velocity (m s⁻¹), A is the ambient aerosol surface area concentration (m² m⁻³), and M_i is the mass concentration of the species in the gas phase (µg m⁻³). As a fraction of WSOG partitioning is reversible (Chhabra et al., 2010; Wong et al., 2015), and SOA formation may be more dependent on the particle-phase reaction rate than simply the

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743 represents an upper limit; however, published CMAO results for OA mass loadings in the 744 Houston area calculated in this manner agree well with observations (Li et al., 2015; Ying et al., 2015). 745 In order to approximate the effect of anthropogenic marine aerosol (AMA) on WSOG 746 agSOA production, we modeled agSOA formation from isoprene-derived glyoxal, 747 748 methylglyoxal, and IEPOX in the Houston area using the equation above and a previously 749 developed 0-D model including a semi-explicit isoprene oxidation mechanism (Schulze et al., 750 2017). This model assumes that air masses rich in AMA advect over the urban core of Houston, 751 where the added SA due to anthropogenic emissions over the ocean increases SA-dependent aqSOA production rates. Total model SA was quantified by combining the dry mass size 752 753 distribution measured by the HR-ToF-AMS at the coastal site during onshore flow and the mass 754 added by NR-PM₁-associated ALW; however, a correction was applied to account for SA loss due to deposition and ALW evaporation during transport to the HSC. A detailed description of 755 all model assumptions (i.e., boundary layer height, aerosol deposition and ALW evaporation 756 757 during transport, etc.) is provided in the SI. Average diurnal isoprene, O₃, and NO_x concentrations measured by five monitors within the HSC during the marine period were used as 758 model constraints (Figure S22). Diurnal OH concentrations were taken from measurements in 759 760 downtown Houston during the SHARP 2009 campaign (Ren et al., 2013). 761 A diurnal model run was first performed using the total corrected marine aerosol SA to predict aqSOA formation in the HSC. This procedure isolates aqSOA production due to marine 762 763 aerosol SA specifically. In order to produce an upper bound estimate of the effect of AMA, a second run was performed with all SA contributions from anthropogenic species removed (i.e., 764

particle surface area (SA) (Budisulistiorini et al., 2017), this estimated production rate likely

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anthropogenic nss-SO₄, NH₄, and ALW), and the difference in aqSOA was calculated. A lower bound estimate was calculated with a model run that only removed the SA contribution of anthropogenic ALW. In order to ensure conservative results, OA was assumed to be entirely biogenic for the purposes of this calculation.

To compare this effect with SOA production from locally-emitted anthropogenic VOCs (AVOC-SOA) in Houston, gas-phase AVOC data measured during the marine period (concentrations of 16 alkanes, 7 alkenes, and 9 aromatics) was obtained from the same monitoring sites around the HSC (Figure S22). Estimates of SOA production rates from these 32 VOCs were calculated using the volatility basis set approach utilized in Tsimpidi et al. (2010). In this mechanism, organic condensable gases produced from initial VOC oxidation are allowed to undergo further aging to produce lower volatility products (Tsimpidi et al., 2010; Hayes et al., 2015). A more detailed description of this process is provided in the SI.

Figure 10 shows that on a daily basis, aqSOA production attributable to isoprene WSOG reactive uptake is primarily due to methylglyoxal rather than IEPOX, implying "high-NO_x" rather than "low-NO_x" ambient conditions (Budisulistiorini et al., 2017). Assuming high-NO_x conditions, the modeled effect of AMA on aqSOA production in the HSC is equivalent to 6-23% of potential daily SOA production from AVOCs measured locally. Using data from the monitor with the highest isoprene concentrations (Haden Road; Figure S24), we predict that the AMA effect may constitute as much as 11-43% of total AVOC-SOA production, implying strong spatial variability in the relative contribution of this effect. Modeled AVOC-SOA production peaks in the early afternoon, consistent with the fact that the aging of condensable gases formed by measured VOCs produces the majority (~80%) of modeled AVOC-SOA.

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Recent studies have revealed that a substantial fraction of SOA formation in urban environments may be produced by primary anthropogenic semi-volatile/intermediate volatility VOCs (P-S/IVOCs) co-emitted with typical VOCs or evaporated during POA dilution but not typically measured (Hayes et al., 2015 and references therein). In Los Angeles, for instance, Hayes et al. (2015) predicted that P-S/IVOCs comprise between 44% and 92% of total modeled SOA depending on the specific SOA formation mechanism used. As a result, the relative magnitude of the AMA effect may be somewhat overestimated here. Still, the AMA effect is responsible for 0.2-0.35 µg m³ of ambient aqSOA according to the model calculations, which represents 4-6% of ambient OA measured by Cleveland et al. (2012) near downtown and ~10-17% of average OA measured in Houston's urban core by Leong et al. (2017). Furthermore, as AVOCs such as benzene and acetylene are known to produce glyoxal and methylglyoxal with high yields (Fu et al., 2008), the total OA mass attributable to AMA through this pathway (on an absolute rather than relative basis) may actually be larger than predicted here. Our results therefore suggest that future reductions in marine nss-SO4 may reduce aqSOA formation in both urban (e.g., Houston) and forested regions across the Gulf Coast.

4 Conclusions

Three weeks of continuous measurements with an HR-ToF-AMS at a coastal location near Houston, TX were used to gain further insight into the impact of shipping emissions on coastal aerosol properties. Measured mass loadings of inorganic NR-PM₁ components were similar to those reported before establishment of the ECA within the Gulf of Mexico; however, data from nearby ports suggests that this is the result of growth in the shipping trade rather than regulatory ineffectiveness on a per vessel basis. Using MSA calibrations and published biogenic

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MSA/nss-SO₄ ratios, we predict that over 70% of inorganic marine NR-PM₁ is anthropogenic rather than biogenic. Source apportionment using PMF revealed that the dominant marine OA factor (OOA-3) is highly correlated with calculated anthropogenic nss-SO₄ ($R^2 \ge 0.78$), supporting a link between shipping emissions and SOA production. Assuming, as an upper bound estimate, that OOA-3 production is entirely dependent on shipping emissions, anthropogenic sources contribute over 70% of total measured NR-PM₁ during onshore flow, despite the regulations. This indicates that the proposed future global decrease in shipping FSC (decrease to 0.5%) should substantially reduce PM levels over the Gulf of Mexico.

Shipping emissions were also found to have numerous secondary effects on OA composition. Detailed backward trajectory and mass spectral analysis revealed that air mass transit within shipping lanes leads to more processed (i.e., oxidized) OA than is encountered in "clean" marine air masses, and calculations suggest that this aging increases OA hygroscopicity. In addition, marine alkyl amine aerosol formation in the Gulf of Mexico appears to depend on ambient anthropogenic nss-SO4 mass, implying that marine amine aerosol cannot be viewed as purely biogenic in heavily trafficked marine environments. OOA-3 was found to have a larger N:C ratio than is typical of aged marine components, supporting this link. Finally, modeling suggests that inland advection of shipping-related nss-SO4 and related ALW may enhance aqSOA formation and produce 4 to 17% of OA in the urban core of Houston during marine flow for the conditions considered. More detailed 3-D modeling studies are warranted to better quantify this effect.

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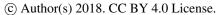




Data Availability: 832 The compiled datasets used to produce each figure within this manuscript are available as 833 834 Igor Pro files upon request. 835 836 **Author Contribution:** All listed authors contributed to data collection during the field campaign. B.C.S. 837 838 performed data analysis and wrote the manuscript. R.J.G assisted heavily with manuscript development and editing. H.W.W, A.B., Q.D., M.H.E., J.H.F., S.A., S.U., and R.S. provided 839 helpful comments and edits. 840 841 842 Acknowledgements: 843 844 The authors would like to acknowledge J. Stutz and K. Tuite (UCLA) for assistance and 845 846 helpful discussions during the field campaign. This material is based upon work supported by the National Science Foundation Graduate Research Fellowship under Grant No. DGE# 1450681. 847

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018







848 References

849

Agrawal, H., Welch, W. A., Miller, J. W., and Cocker, D. R.: Emission measurements from a crude oil tanker at sea, Environ. Sci. Technol., 42, 7098-7103, doi: 10.1021/es703102y, 2008.

853

Agrawal, H., Eden, R., Zhang, X., Fine, P. M., Katzenstein, A., Miller, J. W., Ospital, J.,
 Teffera, S., and Cocker, D.: Primary particulate matter from ocean-going engines in the
 Southern California Air Basin, Environ. Sci. Technol., 43, 5398-5402, doi:
 10.1021/es8035016, 2009.

858

3) Aksoyoglu, S., Baltensperger, U., and Prevot, A. S.: Contribution of ship emissions to the concentration and deposition of air pollutants in Europe, Atmos. Chem. Phys., 16, 1895-1906, doi: 10.5194/acp-16-1895-2016, 2016.

862 863

4) Bahreini, R., Ervens, B., Middlebrook, A.M., Warneke, C., de Gouw, J.A., DeCarlo, P.F.,
Jimenez, J.L., Brock, C.A., Neuman, J.A., Ryerson, T.B., Stark, H., Atlas, E., Brioude, J.,
Fried, A., Holloway, J.S., Peischl, J., Richter, D., Walega, J., Weibring, P., Wollny, A.G.,
and Fehsenfeld, F.C.: Organic aerosol formation in urban and industrial plumes near Houston
and Dallas, Texas, J. Geophys. Res.-Atmos. 114, 17, doi: 10.1029/2008JD011493, 2009.

867 868 869

870

871

5) Bates, T.S., Quinn, P.K., Coffman, D., Schulz, K., Covert, D.S., Johnson, J.E., Williams, E.J., Lerner, B.M., Angevine, W.M., Tucker, S.C., Brewer, W.A., and Stohl, A.: Boundary layer aerosol chemistry during TexAQS/GoMACCS 2006: Insights into aerosol sources and transformation processes. J. Geophys. Res-Atmos. 113, 18, doi: 10.1029/2008JD010023, 2008.

872 873 874

875 6) Bates, T. S., Quinn, P. K., Frossard, A. A., Russell, L. M., Hakala J., Petaj a, T., Kulmala,
876 M., Covert, D. S., Cappa, C. D., Li, S.-M., Hayden, K. L., Nuaaman, I., McLaren, R.,
877 Massoli, P., Canagaratna, M. R, Onasch, T. B., Sueper, D., Worsnop, D. R., and Keene, W.
878 C.: Measurements of ocean derived aerosol off the coast of California, J. Geophys. Res., 117,
879 D00V15, doi: 10.1029/2012JD017588, 2012.

880

Bean, J. K., Faxon, C. B., Leong, Y. J., Wallace, H. W., Cevik, B. K., Ortiz, S., Canagaratna,
 M. R., Usenko, S., Sheesley, R. J., Griffin, R. J., and Hildebrandt, L: Composition and
 sources of particulate matter measured near Houston, TX; Anthropogenic-biogenic
 interactions, Atmos., 5, 73, doi: 10.3390/atmos7050073, 2016.

885

88 Brown, S. S., Dubé, W. P., Bahreini, R., Middlebrook, A. M., Brock, C. A., Warneke, C., de Gouw, J. A., Washenfelder, R. A., Atlas, E., Peischl, J., Ryerson, T. B., Holloway, J. S., Schwarz, J. P., Spackman, R., Trainer, M., Parrish, D. D., Fehshenfeld, F. C., and Ravishankara, A. R.: Biogenic VOC oxidation and organic aerosol formation in an urban nocturnal boundary layer: aircraft vertical profiles in Houston, TX, Atmos. Chem. Phys., 13, 11317–11337, doi: 10.5194/acp-13-11317-2013, 2013.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.





Browning, L., Hartley, S., Bandemehr, A., Gathright, K., and Miller, W.: Demonstration of fuel switching on oceangoing vessels in the Gulf of Mexico, J. Air. Waste Manag. Assoc.,
 62(9):1093-1101, doi: 10.1080/10962247.2012.697974, 2012.

 10) Budisulistiorini, S. H., Nenes, A., Carlton, A. G., Surratt, J. D., McNeill, V. F., and Pye, H. O. T.: Simulating aqueous phase isoprene epoxydiol (IEPOX) secondary organic aerosol production during the 2013 Southern Oxidant and Aerosol Study (SOAS), Environ. Sci. Tech., 51, 5026-5034, doi: 10.1021/acs.est.6b05750, 2017.

11) Caiazzo, F., Ashok, A., Waitz, I. A., Yim, S. H. L., and Barrett, S. R. H.: Air pollution and early deaths in the United States: Part I: Quantifying the impact of major sectors in 2005, Atmos. Environ., 79, 198-208, doi: 10.1016/j.atmosenv.2013.05.081, 2013.

 12) Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q., Onasch, T. B., Drewnick, F., Coe, H., Middlebrook, A., Delia, A., Williams, L. R., Trimborn, A. M., Northway, M. J., DeCarlo, P. F., Kolb, C. E., Davidovits, P., and Worsnop, D. R.: Chemical and microphysical characterization of ambient aerosols with the aerodyne aerosol mass spectrometer, Mass Spectrom. Rev., 26, 185–222, doi: 10.1002/mas.20115, 2007.

 13) Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P., Hildebrandt Ruiz, L., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue, N. M., Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of organic compounds using aerosol mass spectrometry: characterization, improved calibration, and implications, Atmos. Chem. Phys., 15, 253-272, https://doi.org/10.5194/acp-15-253-2015, 2015.

14) Carlton, A. G. and Turpin, B. J.: Particle partitioning potential of organic compounds is highest in the Eastern U.S. and driven by anthropogenic water, Atmos. Chem. Phys., 13, 10203–10214, doi: 10.5194/acp-13-10203-2013, 2013.

 15) Chang, R. Y.-W., Leck, C., Graus, M., Müller, M., Paatero, J., Burkhart, J. F., Stohl, A., Orr, L. H., Hayden, K., Li, S.-M., Hansel, A., Tjernström, M., Leaitch, W. R., and Abbatt, J. P. D.: Aerosol composition and sources in the central Arctic Ocean during ASCOS, Atmos. Chem. Phys., 11, 10619–10636, doi: 10.5194/acp-11-10619-2011, 2011.

16) Chen, G., Huey, L. G., Trainer, M., Nicks, D., Corbett, J., Ryerson, T., Parrish, D., Neuman, J. A., Nowak, J., Tanner, D., Holloway, J., Brock, C., Crawford, J., Olson, J. R., Sullivan, A., Weber, R., Schauffler, S., Donnelly, S., Atlas, E., Roberts, J., Flocke, F., Hubler, G., and Fehsenfeld, F.: An investigation of the chemistry of ship emission plumes during ITCT 2002, J. Geophys. Res., 110, D10S90, doi: 10.1029/2004JD005236, 2005.

17) Chhabra, P. S., Flagan, R. C., and Seinfeld, J. H.: Elemental analysis of chamber organic
 aerosol using an aerodyne high-resolution aerosol mass spectrometer, Atmos. Chem. Phys.,
 10, 4111–4131, doi: 10.5194/acp-10-4111-2010, 2010.

18) Chhabra, P. S., Ng, N. L., Canagaratna, M. R., Corrigan, A. L., Russell, L. M., Worsnop, D.
 R., Flagan, R. C., and Seinfeld, J. H.: Elemental composition and oxidation of chamber
 organic aerosol, Atmos. Chem. Phys., 11, 8827–8845, doi: 10.5194/acp-11-8827-2011, 2011.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.





19) Claeys, M. B., Wang, W., Vermeylen, R., Kourtchev, I., Chi, X., Farhat, Y.J., Surratt, D.,
 Gomez-Gonzalez, Y., Sciare, J., and Maenhaut, W.: Chemical characterization of marine
 aerosol at Amsterdam Island during the austral summer of 2006–2007, J. Aero. Sci., 41(1),
 13–22, doi: 10.1016/j.jaerosci.2009.08.003, 2009.

20) Cleveland, M. J., Ziemba, L. D., Griffin, R. J., Dibb, J. E., Anderson C. H., Lefer, B., and Rappengluck, B.: Characterization of urban aerosol using aerosol mass spectrometry and proton nuclear magnetic resonance spectroscopy, Atmos. Environ., 511-518, 54, doi: 10.1016/j.atmosenv.2012.02.074, 2012.

21) Coggon, M. M., Sorooshian, A., Wang, Z., Metcalf, A. R., Frossard, A. A., Lin, J. J., Craven,
 J. S., Nenes, A., Jonsson, H. H., Russell, L. M., Flagan, R. C., and Seinfeld, J. H.: Ship
 impacts on the marine atmosphere: insights into the contribution of shipping emissions to the
 properties of marine aerosol and clouds, Atmos. Chem. Phys., 12, 8439–8458, doi:
 10.5194/acp-12-8439-2012, 2012.

957 22) Coggon, M.M., Sorooshian, A., Wang, Z., Craven, J.S., Metcalf, A.R., Lin, J.J., Nenes, A.,
 958 Jonsson, H.H., Flagan, R.C., and Seinfeld, J.H.: Observations of continental biogenic impacts
 959 on marine aerosol and clouds off the coast of California. J. Geophys. Res. 119, 6724-6748,
 960 doi: 10.1002/2013JD021228, 2014.

23) Corbett, J.J., and Kohler, H.W.: Updated emissions from ocean shipping. J. Geophys. Res., 108. doi: 10.1029/2003JD003751, 2003.

24) Corbett, J. J., Winebrake, J. J., Green, E. H., Kasibhatla, P., Eyring, V., and Lauer, A.: Mortality from ship emissions: A global assessment, Environ. Sci. Technol., 41, 8512–8518, doi: 10.1021/es071686z, 2007.

25) Crippa, M., El Haddad, I., Slowik, J. G., DeCarlo, P. F., Mohr, C., Heringa, M. F., Chirico, R., Marchand, N., Sciare, J., Baltensperger, U., and Prévôt, A. S. H.: Identification of marine and continental aerosol sources in Paris using high resolution aerosol mass spectrometry, J. Geophys. Res.-Atmos., 118, 1950–1963, doi: 10.1002/jgrd.50151, 2013.

26) Crippa, M., Canonaco, F., Lanz, V. A., Äijälä, M., Allan, J. D., Carbone, S., Capes, G., Ceburnis, D., Dall'Osto, M., Day, D. A., De-Carlo, P. F., Ehn, M., Eriksson, A., Freney, E., Hildebrandt Ruiz, L., Hillamo, R., Jimenez, J. L., Junninen, H., Kiendler-Scharr, A., Kortelainen, A.-M., Kulmala, M., Laaksonen, A., Mensah, A. A., Mohr, C., Nemitz, E., O'Dowd, C., Ovadnevaite, J., Pandis, S. N., Petäjä, T., Poulain, L., Saarikoski, S., Sellegri, K., Swietlicki, E., Tiitta, P., Worsnop, D. R., Baltensperger, U., and Prévôt, A. S. H.: Organic aerosol components derived from 25 AMS data sets across Europe using a consistent ME-2 based source apportionment approach, Atmos. Chem. Phys., 14, 6159–6176, doi: 10.5194/acp-14-6159-2014, 2014.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.



990

995

1000

1006

1014

1018

1023



- 27) Czech, H., Stengel, B., Adam, T., Sklorz, M., Streibel, T., and Zimmerman, R.: A
 chemometric investigation of aromatic emission profiles from a marine engine in comparison
 with residential wood combustion and road traffic: Implications for source apportionment
 inside and outside sulphur emission control areas, Atmos. Environ., 167 212-222, doi:
 10/1016/j.atmosenv.2017.08.022, 2017.
- 28) DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C.,
 Gonin, M., Fuhrer, K., Horvath, T., Docherty, K. S., Worsnop, D. R., and Jimenez, J. L.:
 Field-deployable, high-resolution, time-of-flight aerosol mass spectrometer, Anal. Chem., 78,
 8281–8289, doi: 10.1021/ac061249n, 2006.
- 29) Draxler, R. R., and G. D. Rolph: HYSPLIT (Hybrid Single-Particle Lagrangian Integrated
 Trajectory) Model, access via NOAA ARL READY Website, NOAA Air Resources
 Laboratory, Silver Spring, MD., 2003. Available at http://
 www.arl.noaa.gov/ready/hysplit4.html.
- 30) Duplissy, J., DeCarlo, P. F., Dommen, J., Alfarra, M. R., Metzger, A., Barmpadimos, I.,
 Prevot, A. S. H., Weingartner, E., Tritscher, T., Gysel, M., Aiken, A. C., Jimenez, J. L.,
 Canagaratna, M. R., Worsnop, D. R., Collins, D. R., Tomlinson, J., and Baltensperger, U.:
 Relating hygroscopicity and composition of organic aerosol particulate matter, Atmos.
 Chem. Phys., 11, 1155–1165, doi: 10.5194/acp-11-1155-2011, 2011.
- 31) Ervens, B., Turpin, B. J., and Weber, R. J.: Secondary organic aerosol formation in cloud droplets and aqueous particles (aqSOA): a review of laboratory, field and model studies, Atmos. Chem. Phys., 11, 11069–11102, doi: 10.5194/acp-11-11069-2011, 2011.
- 32) Ervens, B., Sorooshian, A., Lim, Y. B., and Turpin, B.: Key parameters controlling OH initiated formation of secondary organic aerosol in the aqueous phase (aqSOA), J. Geophys.
 Res. Atmos., 119, 3997-4016, doi: 10.1002/2013JD021021, 2014.
- 33) Eyring, V., H. W. Kohler, A. Lauer, and Lemper, B.: Emissions from international shipping:
 2. Impact of future technologies on scenarios until 2050, J. Geophys. Res., 110, D17301, doi: 10.1029/2004JD005620, 2005.
- 34) Eyring, V., Isaksen, I. S. A., Berntsen, T., Collins, W. J., Corbett, J. J., Endresen, O.,
 Grainger, R. G., Moldanova, J., Schlager, H., and Stevenson, D. S.: Transport impacts on
 atmosphere and climate: Shipping, Atmos. Environ., 44, 4735–4771, doi:
 10.1016/j.atmosenv.2009.04.059, 2010.
- 35) Facchini, M. C., Decesari, S., Rinaldi, M., Carbone, C., Finessi, E., Mircea, M., Fuzzi, S.,
 Moretti, F., Tagliavini, E., Ceburnis, D., and O'Dowd, C. D.: Important source of marine
 secondary organic aerosol from biogenic amines, Environ. Sci. Technol., 42, 9116–9121, doi:
 10.1021/es8018385, 2008.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.





36) Faloona, I.: Sulfur processing in the marine atmospheric boundary layer: A review and critical assessment of modeling uncertainties, Atmos. Environ., 43, 2841-2854, doi: 10.1016/j.atmosenv.2009.02.043, 2009.

1032

37) Fountoukis, C. and Nenes, A.: ISORROPIA II: a computationally efficient thermodynamic equilibrium model for K⁺-Ca²⁺-Mg²⁺-NH₄⁺-Na⁺-SO₄²⁻-NO₃⁻-Cl⁻-H₂O aerosols, Atmos. Chem. Phys., 7, 4639-4659, https://doi.org/10.5194/acp-7-4639-2007, 2007.

1036

38) Frossard, A. A., Russell, L. M., Burrows, S. M., Elliott, S. M., Bates, T. S., and Quinn, P. K.: Sources and composition of submicron organic mass in marine aerosol particles, J. Geophys. Res.- Atmos., 119, 12977–13003, doi: 10.1002/2014JD021913, 2014.

1040

39) Fu, T., Jacob, D. J., Wittrock, F., Burrows, J. P., Vrekoussis, M., and Henze, D. K.: Global
 budgets of atmospheric glyoxal and methylglyoxal, and implications for formation of
 secondary organic aerosols, J. Geophys. Res. Atmos., 113, 15 303, doi:
 10.1029/2007JD009505, 2008.

1045

40) Fu, P., Kawamura, K., and Miura, K.: Molecular characterization of marine organic aerosols
 collected during a round-the-world cruise, J. Geophys. Res. Atmos., 116, D13, doi:
 10.1029/2011JD015604, 2011.

1049

41) Gantt, B., Meskhidze, N., Facchini, M. C., Rinaldi, M., Ceburnis, D., and O'Dowd, C. D.:
 Wind speed dependent size-resolved parameterization for the organic mass fraction of sea
 spray aerosol, Atmos. Chem. Phys., 11, 8777–8790, doi: 10.5194/acp-11-8777-2011, 2011.

1053 1054

42) Gantt, B. and Meskhidze, N.: The physical and chemical characteristics of marine primary organic aerosol: a review, Atmos. Chem. Phys., 13, 3979-3996, https://doi.org/10.5194/acp-13-3979-2013, 2013.

1057

43) Gaston, C. J., Pratt, K. A., Qin, X., and Prather, K. A.: Real-time detection and mixing state
 of methanesulfonate in single particles at an inland urban location during a phytoplankton
 bloom, Environ. Sci. Technol., 44, 1566–1572, doi: 10.1021/es902069d, 2010.

1061

44) Ge, X., Zhang, Q., Sun, Y., Ruehl, C. R., and Setyan, A.: Effect of aqueous-phase processing
 on aerosol chemistry and size distributions in Fresno, California, during wintertime, Environ.
 Chem., 9, 221–235. http://dx.doi.org/10.1071/en11168, 2012.

1065

45) Guo, Q., Hu, M., Guo, S., Wu, Z., Hu, W., Peng, J., Hu, W., Wu, Y., Yuan, B., Zhang, Q., and Song, Y: The identification of source regions of black carbon at a receptor site off the eastern coast of China, Atmos. Environ., 100, 78-84, doi: 10.1016/j.atmosenv.2014.10.053, 2014.

- 46) Hayes, P. L., Ortega, A. M., Cubison, M. J., Froyd, K. D., Zhao, Y., Cliff, S. S., Hu, W. W.,
 Toohey, D. W., Flynn, J. H., Lefer, B. L., Grossberg, N., Alvarez, S., Rappenglück, B.,
- Taylor, J. W., Allan, J. D., Holloway, J. S., Gilman, J. B., Kuster, W. C., de Gouw, J. A.,
- Massoli, P., Zhang, X., Liu, J., Weber, R. J., Corrigan, A. L., Russell, L. M., Isaacman, G.,

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.





Worton, D. R., Kreisberg, N. M., Goldstein, A. H., Thalman, R., Waxman, E. M., Volkamer,
R., Lin, Y. H., Surratt, J. D., Kleindienst, T. E., Offenberg, J. H., Dusanter, S., Griffith, S.,
Stevens, P. S., Brioude, J., Angevine, W. M., and Jimenez, J. L.: Organic aerosol
composition and sources in Pasadena, California, during the 2010 CalNex campaign, J.
Geophys. Res.-Atmos., 118, 9233–9257, doi: 10.1002/jgrd.50530, 2013.

1080 1081

47) Hayes, P. L., Carlton, A. G., Baker, K. R., Ahmadov, R., Washenfelder, R. A., Alvarez, S.,
Rappenglück, B., Gilman, J. B., Kuster, W. C., de Gouw, J. A., Zotter, P., Prévôt, A. S. H.,
Szidat, S., Kleindienst, T. E., Offenberg, J. H., Ma, P. K., and Jimenez, J. L.: Modeling the
formation and aging of secondary organic aerosols in Los Angeles during CalNex 2010,
Atmos. Chem. Phys., 15, 5773-5801, https://doi.org/10.5194/acp-15-5773-2015, 2015.

1086

48) Hildebrandt, L., Engelhart, G. J., Mohr, C., Kostenidou, E., Lanz, V. A., Bougiatioti, A.,
 DeCarlo, P. F., Prevot, A. S. H., Baltensperger, U., Mihalopoulos, N., Donahue, N. M., and
 Pandis, S. N.: Aged organic aerosol in the Eastern Mediterranean: the Finokalia Aerosol
 Measurement Experiment – 2008, Atmos. Chem. Phys., 10, 4167-4186,
 https://doi.org/10.5194/acp-10-4167-2010, 2010.

109210931094

1095

1096

49) Hildebrandt, L., Kostenidou, E., Lanz, V. A., Prevot, A. S. H., Baltensperger, U., Mihalopoulos, N., Laaksonen, A., Donahue, N. M., and Pandis, S. N.: Sources and atmospheric processing of organic aerosol in the Mediterranean: insights from aerosol mass spectrometer factor analysis, Atmos. Chem. Phys., 11, 12499-12515, https://doi.org/10.5194/acp-11-12499-2011, 2011.

1097 1098 1099

1100

50) Hopke, P.K., Barrie, L.A., Li, S.M., Cheng, M.D., Li, C., and Xie, Y.: Possible sources and preferred pathways for biogenic and non-sea-salt sulfur for the high arctic. J. Geophys. Res. Atmos. 100 (D8), 16595-16603, doi: 10.1029/95JD01712, 1995.

1101 1102

51) Huang, D. D., Li, Y. J., Lee, B. P., and Chan, C. K.: Analysis of organic sulfur compounds in atmospheric aerosols at the HKUST supersite in Hong Kong using HR-ToF-AMS, Environ.
 Sci. Technol., 49, 3672–3679, doi: 10.1021/es5056269, 2015.

1106

52) Huang, S., Poulain, L., Pinxteren, D. V., Pinxteren, M. V., Wu, Z., Herrmann, H., and
 Wiedensohler, A.: Latitudinal and seasonal distribution of particulate MSA over the Atlantic
 using a validated quantification method with HR-ToF-AMS, Environ. Sci. Technol., 51, 418 426, doi: 10.1021/acs.est.6b03186, 2017.

1111

53) Hynes, A.J., Wine, P.H., and Semmes, D.H.: Kinetics and mechanism of hydroxyl reactions with organic sulfide, J. Phys. Chem. 90, 4148-4156, doi: 10.1021/j100408a062, 1986.

1114

54) IMO: International Shipping Facts and Figures – Information Resources on Trade, Safety,
 Security, Environment. International Maritime Organization, 2012.

1117

55) Johansson, L., Jalkanen, J.-P., and Kukkonen, J.: Global assessment of shipping emissions in 2015 on a high spatial and temporal resolution, Atmos. Environ., 167, 403-415, doi: 10.1016/j.atmosenv.2017.08.042, 2017.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.





1121

56) Jung, J., Furutani, H., Uematsu, M., and Park, J.: Distributions of atmospheric non-sea-salt sulfate and methanesulfonic acid over the Pacific Ocean between 48°N and 55°S during summer, Atmos. Environ., 99, 374-384, doi: 10.1016/j.atmosenv.2014.10.009, 2014.

1125

57) Kasper, A., S. Aufdenblatten, A. Forss, M. Mohr, and Burtscher, H.: Particulate emissions from a low-speed marine diesel engine, Aerosol Sci. Technol., 41, 24 – 32, doi: 10.1080/02786820601055392, 2007.

1129

58) Kerminen, V.-M., Aurela, M., Hillamo, R.E., and Virkkula, A.: Formation of particulate MSA: deductions from size distribution measurements in the Finnish Arctic. Tellus, 49B, 159-171, doi: 10.3402/tellusb.v49i2.15959, 1997.

1133

59) Kim, H. S., Song, C. H., Park, R. S., Huey, G., and Ryu, J. Y.: Investigation of ship-plume chemistry using a newly-developed photochemical/dynamic ship-plume model, Atmos. Chem. Phys., 9, 7531–7550, doi: 10.5194/acp-9-7531-2009, 2009.

1137

60) Kim, H. S., Kim, Y. H., and Song, C. H.: Ship-plume sulfur chemistry: ITCT 2K2 case study,
 Sci. Tot. Environ., 450-451:178-87, doi:10.1016/j.scitotenv.2013.01.099, 2013.

1140

61) Kotchenruther, R.: The effects of marine vessel fuel sulfur regulations on ambient PM_{2.5} at coastal and near coastal monitoring sites in the U.S., Atmos. Environ., 151, 52-61, doi: 10.1016/j.atmosenv.2016.12.012, 2016.

1144

62) Lack, D. A., Corbett, J. J., Onasch, T., Lerner, B., Massoli, P., Quinn, P. K., Bates, T. S.,
 Covert, D. S., Coffman, D., Sierau, B., et al.: Particulate emissions from commercial
 shipping: Chemical, physical, and optical properties, J. Geophys. Res., 114, D00F04, doi:
 10.1029/2008JD011300, 2009.

1149

63) Lack, D. A., Cappa, C. D., Langridge, J., Bahreini, R., Buffaloe, G., Brock, C. A., Cerully,
K., Hayden, K., Holloway, J. S., Lerner, B., Li, S. M., McLaren, R., Middlebrook, A.,
Moore, R., Nenes, A., Nuaanman, I., Peischl, J., Perring, A., Quinn, P. K., Ryerson, T. B.,
Schwarz, J. P., Spackman, J. R., and Williams, E. J.: Impact of fuel quality regulation and
speed reductions on shipping emissions: Implications for climate and air quality, Environ.
Sci. Technol., 45, 9052–9060, doi: 10.1021/es2013424, 2011.

1156

1157 64) Lauer, A., Eyring, V., Hendricks, J., Jockel, P., and Lohmann, U.: Effects of oceangoing shipping on aerosols and clouds. Atmos. Chem. Phys. 7, 5061–5079, doi: 10.5194/acp-7-5061-2007, 2007.

1160

65) Leong, Y. J., Sanchez, N. P., Wallace, H. W., Cevik, K., Hernandez, C. S., Han, Y., Flynn, J.
 H., Massoli, P., Floerchinger, C., Fortner E. C., Herndon, S., Bean, J. K., Hildebrandt Ruiz,
 L., Jeon, W., Choi, Y., Lefer, B., and Griffin, R. J.: Overview of surface measurements and
 spatial characterization of submicrometer particulate matter during the DISCOVER-AQ 2013
 campaign in Houston, J. Air Waste Manage. Assoc., 28, 1-19, doi:

1166 10.1080/10962247.2017.1296502, 2017.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.





1168 66) Li, J., Cleveland, M., Ziemba, L. D., Griffin, R. J., Barsanti, K. C., Pankow, J. F., Ying, Q.:
 1169 Modeling regional secondary organic aerosol using the Master Chemical Mechanism, Atmos.
 1170 Environ., 102, 52-61, doi: 10.1016/j.atmosenv.2014.11.054, 2015.

1172 67) Lin, C. T., Baker, A. R., Jickells, T. D., Kelly, S., and Lesworth, T.: An assessment of the significance of sulphate sources over the Atlantic Ocean based on sulphur isotope data, Atmos. Environ., 62, 615-621, doi: 10.1016/j.atmosenv.2012.08.052, 2012.

68) Liu, H., Fu, M., Jin, X., Shang, Y., Shindell, D., Faluvegi, G., Shindell, C., and He, K.:. Health and climate impacts of ocean-going vessels in East Asia, Nat. Clim. Change Adv., 6, 1037-1041, doi: 10.1038/nclimate3083, 2016.

1180 69) McNeill, V. F.: Aqueous organic chemistry in the atmosphere: sources and chemical 1181 processing of organic aerosols, Environ. Sci. Technol., 49, 1237-1244, doi: 10.1021/es5043707, 2015.

70) Middlebrook, A. M., Bahreini, R., Jimenez, J. L., and Canagaratna, M. R.: Evaluation of composition-dependent collection efficiencies for the Aerodyne aerosol mass spectrometer using field data, Aerosol. Sci. Technol., 46, 258–271, doi: 10.1080/02786826.2011.620041, 2011.

71) Murphy, S. M., Sorooshian, A., Kroll, J. H., Ng, N. L., Chhabra, P., Tong, C., Surratt, J. D., Knipping, E., Flagan, R. C., and Seinfeld, J. H.: Secondary aerosol formation from atmospheric reactions of aliphatic amines, Atmos. Chem. Phys., 7, 2313-2337, https://doi.org/10.5194/acp-7-2313-2007, 2007.

72) Murphy, S. M., Agrawal, H., Sorooshian, A., Padro, L. T., Gates, H., Hersey, S., Welch, W.
A., Jung, H., Miller, J. W., Cocker, D. R., Nenes, A., Jonsson, H. H., Flagan, R. C., and
Seinfeld, J. H.: Comprehensive simultaneous shipboard and airborne characterization of
exhaust from a modern container ship at sea, Environ. Sci. Technol., 43, 4626–4640, doi:
10.1021/es802413j, 2009.

73) Myriokefalitakis, S., Vignati, E., Tsigaridis, K., Papadimas, C., Sciare, J., Mihalopoulos, N.,
 Facchini, M. C., Rinaldi, M., Dentener, F. J., Ceburnis, D., Hatzianastasiou, N., O'Dowd, C.
 D., van Weele, M., and Kanakidou, M.: Global modelling of the oceanic source of organic
 aerosols, Adv. Meteorol., 2010, 939171, doi: 10.1155/2010/939171, 2010.

74) Ng, N. L., Canagaratna, M. R., Zhang, Q., Jimenez, J. L., Tian, J., Ulbrich, I. M., Kroll, J. H.,
Docherty, K. S., Chhabra, P. S., Bahreini, R., Murphy, S. M., Seinfeld, J. H., Hildebrandt, L.,
Donahue, N. M., DeCarlo, P. F., Lanz, V. A., Prevot, A. S. H., Dinar, E., Rudich, Y., and
Worsnop, D. R.: Organic aerosol components observed in Northern Hemispheric datasets
from Aerosol Mass Spectrometry, Atmos. Chem. Phys., 10, 4625–4641, doi: 10.5194/acp-10-4625-2010, 2010.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.



1215

1219

1225

1229

1234

1238

1241

1246

1252



- 75) Nguyen, T. K. V., Zhang, Q., Jimenez, J. L., Pike, M., and Carlton, A. G.: Liquid water: ubiquitous contributor to aerosol mass, Environ. Sci. Technol. Lett., 3, 257–263, doi: 10.1021/acs.estlett.6b00167, 2016.
- 76) Nightingale, P. D., Liss, P. S., and Schlosser, P.: Measurements of air-sea gas transfer during an open ocean algal bloom, J. Geophys. Res. Lett., 27, 2117–2120, doi: 10.1029/2000GL011541, 2000.
- 77) Ortega, A. M., Hayes, P. L., Peng, Z., Palm, B. B., Hu, W., Day, D. A., Li, R., Cubison, M.
 J., Brune, W. H., Graus, M., Warneke, C., Gilman, J. B., Kuster, W. C., de Gouw, J.,
 Gutiérrez-Montes, C., and Jimenez, J. L.: Real-time measurements of secondary organic
 aerosol formation and aging from ambient air in an oxidation flow reactor in the Los Angeles
 area, Atmos. Chem. Phys., 16, 7411-7433, https://doi.org/10.5194/acp-16-7411-2016, 2016.
- 78) Ovadnevaite, J., O'Dowd, C., Dall'Osto, M., Ceburnis, D., Worsnop, D. R., and Berresheim,
 H.: Detecting high contributions of primary organic matter to marine aerosol: A case study,
 Geophys. Res. Lett., 38, L02807, doi: 10.1029/2010GL046083, 2011.
- 79) Ovadnevaite, J., Ceburnis, D., Leinert, S., Dall'Osto, M., Canagaratna, M., O'Doherty, S.,
 Berresheim, H., and O'Dowd, C.: Submicron NE Atlantic marine aerosol chemical
 composition and abundance: Seasonal trends and air mass categorization, J. Geophys. Res.
 Atmos., 119, 11850–11863, doi: 10.1002/2013JD021330, 2014.
- 80) Paatero, P., and Tapper, U.: Positive matrix factorization A nonnegative factor model with
 optimal utilization of error-estimates of data values, Environmetrics, 5, 111-126, doi:
 10.1002/env.3170050203, 1994.
- 1239 81) "Port Houston Statistics" Port Houston. http://porthouston.com/portweb/about-us/statistics/, Accessed August 9, 2017.
- 82) Raper, J. L., Kleb, M. M., Jacob, D. J., Davis, D., Newell, R. E., Fuelberg, H. E., Bendura, R. J., Hoell, J. M., and McNeal, R. J.: Pacific Exploratory Mission in the Tropical Pacific: PEM Tropics B, March–April 1999, J. Geophys. Res.-Atmos., 106, 32401–32425, doi: 10.1029/2000JD900833, 2001.
- 83) Ren, X., van Duin, D., Cazorla, M., Chen, S., Mao, J., Zhang, L., Brune, W. H., Flynn, J.,
 Grossberg, N., Lefer, B. L., Rappengluck, B., Wong, K. W., Tsai, C., Stutz, J., Dibb, J. E.,
 Jobson, B. T., Luke, W. T., and Kelley, P.: Atmospheric oxidation chemistry and ozone
 production: Results from SHARP 2009 in Houston, Texas, J. Geophys. Res. Atmos., 118,
 5770-5780, doi: 10.1002/jgrd.50342, 2013.
- 84) Rinaldi, M., Decesari, S., Finessi, E., Giulianelli, L., Carbone, C., Fuzzi, S., O'Dowd, C. D.,
 Ceburnis, D., and Facchini, M. C.: Primary and secondary organic marine aerosol and
 oceanic biological activity: Recent results and new perspectives for future studies, Adv.
 Meteorol., 2010, 310–682, doi: 10.1155/2010/310682, 2010.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.



1262

1267

1271

1276

1281

1286



- 85) Russell, L. M., Takahama, S., Liu, S., Hawkins, L. N., Covert, D. S., Quinn, P. K., and Bates,
 T. S.: Oxygenated fraction and mass of organic aerosol from direct emission and atmospheric
 processing measured on the R/V Ronald Brown during TexAQS/GoMACCS 2006, J.
 Geophys. Res. Atmos., 114, D00F05, doi: 10.1029/2008JD011275, 2009.
- 86) Russell, L. M., Hawkins, L. N., Frossard, A. A., Quinn, P. K., and Bates, T. S.:
 Carbohydrate-like composition of submicron atmospheric particles and their production from ocean bubble bursting, P. Natl. Acad. Sci. USA, 107, 6652–6657, doi: 10.1073/pnas.0908905107, 2010.
- 1268 87) Saltzman, E. S., D. L. Savoie, R. G. Zika, and J. M. Prospero: Methane sulfonic acid in the
 1269 marine atmosphere, J. Geophys. Res., 88, 10,897–10,902, doi: 10.1029/JC088iC15p10897,
 1270 1983.
- 88) Savoie, D. L., Arimoto, R., Keene, W. C., Prospero, J. M., Duce, R. A., and Galloway, J. N.:
 Marine biogenic and anthropogenic contributions to non-sea-salt sulfate in the marine
 boundary layer over the North Atlantic Ocean, J. Geophys. Res., 107, 4356, doi:
 10.1029/2001JD000970, 2002.
- 89) Schmale, J., Schneider, J., Nemitz, E., Tang, Y. S., Dragosits, U., Blackall, T. D., Trathan, P.
 N., Phillips, G. J., Sutton, M., and Braban, C. F.: Sub-Antarctic marine aerosol: dominant contributions from biogenic sources, Atmos. Chem. Phys., 13, 8669–8694, doi: 10.5194/acp-13-8669-2013, 2013.
- 90) Schulze, B. C., Wallace, H. W., Flynn, J. H., Lefer, B. L., Erickson, M. H., Jobson, B. T.,
 Dusanter, S., Griffith, S. M., Hansen, R. F., Stevens, P. S., VanReken, T., and Griffin, R. J.:
 Differences in BVOC oxidation and SOA formation above and below the forest canopy,
 Atmos. Chem. Phys., 17, 1805-1828, https://doi.org/10.5194/acp-17-1805-2017, 2017.
- 91) Setyan, A., Zhang, Q., Merkel, M., Knighton, W. B., Sun, Y., Song, C., Shilling, J. E.,
 Onasch, T. B., Herndon, S. C., Worsnop, D. R., Fast, J. D., Zaveri, R. A., Berg, L. K.,
 Wiedensohler, A., Flowers, B. A., Dubey, M. K., and Subramanian, R.: Characterization of submicron particles influenced by mixed biogenic and anthropogenic emissions using high-resolution aerosol mass spectrometry: results from CARES, Atmos. Chem. Phys., 12, 8131-8156, doi: 10.5194/acp-12-8131-2012, 2012.
- 92) Shank, L. M., Howell, S., Clarke, A. D., Freitag, S., Brekhovskikh, V., Kapustin, V.,
 McNaughton, C., Campos, T., and Wood, R.: Organic matter and non-refractory aerosol over
 the remote Southeast Pacific: oceanic and combustion sources, Atmos. Chem. Phys., 12, 557 576, https://doi.org/10.5194/acp-12-557-2012, 2012.
- 93) Shilling, J. E., Chen, Q., King, S. M., Rosenoern, T., Kroll, J. H., Worsnop, D. R., DeCarlo,
 P. F., Aiken, A. C., Sueper, D., Jimenez, J. L., and Martin, S. T.: Loading-dependent
 elemental composition of alpha-pinene SOA particles, Atmos. Chem. Phys., 9, 771–782, doi:
 10.5194/acp-9-771-2009, 2009.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

© Author(s) 2018. CC BY 4.0 License.



1307

1314

1320

1335

1344



- 94) Sinreich, R., Coburn, S., Dix, B., and Volkamer, R.: Ship-based detection of glyoxal over the
 remote tropical Pacific Ocean, Atmos. Chem. Phys., 10, 11359–11371, doi: 10.5194/acp-10-1306
 11359-2010, 2010.
- 95) Slowik, J. G., Stroud, C., Bottenheim, J. W., Brickell, P. C., Chang, R. Y.-W., Liggio, J.,
 Makar, P. A., Martin, R. V., Moran, M. D., Shantz, N. C., Sjostedt, S. J., van Donkelaar, A.,
 Vlasenko, A., Wiebe, H. A., Xia, A. G., Zhang, J., Leaitch, W. R., and Abbatt, J. P. D.:
 Characterization of a large biogenic secondary organic aerosol event from eastern Canadian
 forests, Atmos. Chem. Phys., 10, 2825-2845, https://doi.org/10.5194/acp-10-2825-2010,
 2010.
- 96) Sorooshian, A., Padro, L. T., Nenes, A., Feingold, G., McComiskey, A., Hersey, S. P., Gates,
 H., Jonsson, H. H., Miller, S. D., Stephens, G. L., Flagan, R. C., and Seinfeld, J. H.: On the
 link between ocean biota emissions, aerosol, and maritime clouds: Airborne, ground, and
 satellite measurements off the coast of California, Global Biogeochem. Cy., 23, Gb4007, doi:
 10.1029/2009gb003464, 2009.
- 97) Sorooshian, A., Crosbie, E., Maudlin, L. C., Youn, J. S., Wang, Z., Shingler, T., Ortega, A.
 M., Hersey, S., and Woods, R. K.: Surface and airborne measurements of organosulfur and methanesulfonate over the western United States and coastal areas, J. Geophys. Res., 120, 8535–8548, doi: 10.1002/2015JD023822, 2015.
- 98) Sun, Y.-L., Zhang, Q., Schwab, J. J., Demerjian, K. L., Chen, W.-N., Bae, M.-S., Hung, H.-M., Hogrefe, O., Frank, B., Rattigan, O. V., and Lin, Y.-C.: Characterization of the sources and processes of organic and inorganic aerosols in New York city with a high-resolution time-of-flight aerosol mass spectrometer, Atmos. Chem. Phys., 11, 1581-1602, https://doi.org/10.5194/acp-11-1581-2011, 2011.
- 99) Tian, L., Ho, K.F., Louie, P.K.K., Qiu, H., Pun, V.C., Kan, H., Yu, I.T.S., Wong, T.W.:
 Shipping emissions associated with increased cardiovascular hospitalizations, Atmos.
 Environ., 74, 320-325, doi: 10.1016/j.atmosenv.2013.04.014, 2013.
- 1336 100) Tournadre, J.: Anthropogenic pressure on the open ocean: The growth of ship traffic
 1337 revealed by altimeter analysis, Geophys. Res. Lett., 41, 7924-7932, doi:
 1338 10.1002/2014GL061786, 2014.
- 1340 101) Tsimpidi, A. P., Karydis, V. A., Zavala, M., Lei, W., Molina, L., Ulbrich, I. M., Jimenez, J. L., and Pandis, S. N.: Evaluation of the volatility basis-set approach for the simulation of organic aerosol formation in the Mexico City metropolitan area, Atmos. Chem. Phys., 10, 525–546, doi: 10.5194/acp-10-525-2010, 2010.
- 1345 102) Ulbrich, I. M., Canagaratna, M. R., Zhang, Q., Worsnop, D. R., and Jimenez, J. L.:
 1346 Interpretation of organic components from Positive Matrix Factorization of aerosol mass
 1347 spectrometric data, Atmos. Chem. Phys., 9, 2891–2918, doi: 10.5194/acp-9-2891-2009,
 1348 2009.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

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- 1350 103) U.S. Army Corps of Engineers, Navigation Data Center, 2016: Waterborne Commerce
 1351 Statistics Center, U.S. Waterborne Container Traffic in 2015 2003, available at:
 1352 http://www.navigationdatacenter.us/wcsc/containers.htm, Accessed August 9, 2017
 1353
- 1354 104) U.S. Environmental Protection Agency, 2010: Designation of North American Emission 1355 Control Area to Reduce Emissions from Ships: Regulatory Announcement. EPA-420-F-10-1356 015.
- 1358 105) U.S. Environmental Protection Agency, 2014: National Emissions Inventory (NEI) Data,
 1359 available at: https://www.epa.gov/air-emissions-inventories/2014-national-emissions-inventory-nei-data., Accessed November 10, 2017
- 1362 106) U.S. Environmental Protection Agency, 2017: Air Pollutants Emissions Trends Data,
 1363 State Average Annual Emissions Trends, available at: https://www.epa.gov/air-emissions-inventories/air-pollutant-emissions-trends-data., Accessed November 10, 2017
- 1366 107) Vaughan, S., Ingham, T., Whalley, L. K., Stone, D., Evans, M. J., Read, K. A., Lee, J. D.,
 1367 Moller, S. J., Carpenter, L. J., Lewis, A. C., Fleming, Z. L., and Heard, D. E.: Seasonal
 1368 observations of OH and HO₂ in the remote tropical marine boundary layer, Atmos. Chem.
 1369 Phys., 12, 2149–2172, doi: 10.5194/acp-12-2149- 2012, 2012.
- 1371 108) Viana, M., Hammingh, P., Colette, A., Querol, X., Degraeuwe, B., de Vlieger, I., van 1372 Aardenne, J.: Impact of maritime transport emissions on coastal air quality in Europe, Atmos. 1373 Environ., 90, 96–105, doi: 10.1016/j.atmosenv.2014.03.046, 2014.
- 1375 109) Vutukuru, S., and Dabdub, D.: Modeling the effects of ship emissions on coastal air quality: A case study of southern California, Atmos. Environ., 42, 3751-3764, doi: 10.1016/j.atmosenv.2007.12.073, 2008.
- 1379 110) Wallace, H. W., Sanchez, N. P., Flynn, J. H., Erickson, M. H., Lefer, B. L., and Griffin, R. J.: Source apportionment of particulate matter and trace gases near a major refinery near the Houston Ship Channel, Atmos. Env., 173, 16-29, doi: 10.1016/j.atmosenv.2017.10.049, 2018.
- 1384 111) Wan, Z., Zhu, M., Chen, S., and Sperling, D.: Pollution: three steps to a green shipping industry. Nature, 530, 275-277, doi: 10.1038/530275a, 2016.
- 1387 112) Wang, C., Corbett, J. J., and Firestone, J.: Improving spatial representation of global ship emissions inventories, Environ. Sci. Tech., 42, 193-199, doi: 10.1021/es0700799, 2008.
- 1390 Wong, J. P. S., Lee, A. K. Y., and Abbatt, J. P. D: Impacts of sulfate seed acidity and water content on isoprene secondary organic aerosol formation, Environ. Sci. Tech., 49, 13215-13221, doi: 10.1021/acs.est.5b02686, 2015.
- 1394 114) Wood, E. C., Canagaratna, M. R., Herndon, S. C., Onasch, T. B., Kolb, C. E., Worsnop,
 1395 D. R., Kroll, J. H., Knighton, W. B., Seila, R., Zavala, M., Molina, L. T., DeCarlo, P. F.,
 1396 Jimenez, J. L., Weinheimer, A. J., Knapp, D. J., Jobson, B. T., Stutz, J., Kuster, W. C., and

Discussion started: 12 June 2018

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Williams, E. J.: Investigation of the correlation between odd oxygen and secondary organic aerosol in Mexico City and Houston, Atmos. Chem. Phys., 10, 8947–8968, doi: 10.5194/acp-10-8947-2010, 2010.

115) Wozniak, A. S., Willoughby, A. S., Gurganus, S. C., and Hatcher, P. G.: Distinguishing molecular characteristics of aerosol water soluble organic matter from the 2011 trans-North Atlantic US GEOTRACES cruise, Atmos. Chem. Phys., 14, 8419-8434, doi: 10.5194/acp-14-8419-2014, 2014.

116) Ying, Q., Li, J., and Kota, S. H.: Significant contributions of isoprene to summertime secondary organic aerosol in eastern United States, Environ. Sci. Tech., 49, 7834-7842, doi: 10.1021/acs.est.5b02514, 2015.

117) Youn, J.-S., Crosbie, E., Maudlin, L. C., Wang, Z., and Sorooshian, A.: Dimethylamine as a major alkyl amine species in particles and cloud water: Observations in semi-arid and coastal regions, Atmos. Environ., 122, 250–258, doi: 10.1016/j.atmosenv.2015.09.061, 2015.

118) Zetterdahl, M., Salo, K., Fridell, E., and Sjoblom, J.: Impact of aromatic concentration in marine fuels on particle emissions, J. Mar. Sci. App., 16, 3, 352-361, doi: 10.1007/s11804-017-1417-7, 2017.

119) Zhao, M., Zhang, Y., Ma, W., Fu, Q., Yang, X., Li, C., Zhou, B., Yu, Q., and Chen, L: Characteristics and ship traffic source identification of air pollutants in China's largest port, Atmos. Environ., 64, 277-286, doi: 10.1016/j.atmosenv.2012.10.007, 2012.

120) Zhu, L., Huang, X., Shi, H., Cai, X., and Song, Y.: Transport pathways and potential sources of PM₁₀ in Beijing, Atmos. Environ. 45 (3), 594-604, doi: 10.1016/j.atmosenv.2010.10.040, 2011.

121) Zorn, S. R., Drewnick, F., Schott, M., Hoffmann, T., and Borrmann, S.: Characterization of the South Atlantic marine boundary layer aerosol using an aerodyne aerosol mass spectrometer, Atmos. Chem. Phys., 8, 4711–4728, doi: 10.5194/acp-8-4711-2008, 2008.

Manuscript under review for journal Atmos. Chem. Phys.

Discussion started: 12 June 2018

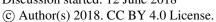






Table 1: Aerosol (μ g m⁻³) and trace gas (ppbv) characteristics (avg. \pm std. dev.) measured during each distinct period type of the campaign.

Period	NR-PM ₁	OA	SO ₄	NH ₄	NO ₃	Chl.	O ₃	NO _x	СО
Marine	3.8±2.0	0.7±0.8	2.4±1.1	0.7±0.3	0.02±0.01	0.02±0.01	31.1±11.9	0.4±1.2	111.5±16.5
Frontal/LP	2.6±2.1	1.0±0.9	1.3±1.2	0.3 ± 0.4	0.04±0.03	0.02±0.01	43.8±11.1	1.0±1.3	107.2±14.4
Continental	9.9±2.9	7.2±2.8	1.9±0.7	0.6±0.2	0.1±0.1	0.02±0.01	52.7±12.7	1.3±1.8	141.3±26.2

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Table 2: Elemental composition of each PMF factor and average contributions to total OA during each period of the campaign.

	Elemental Analysis			Marine Period		Frontal/LP Period		Continental Period	
	O:C	Н:С	N:C	%	$\mu g \ m^{\text{-}3}$	%	$\mu g m^{-3}$	%	$\mu g m^{-3}$
OOA – 1	1.16	1.29	0.013	21	0.14	65	0.63	15	1.10
OOA - 2	0.79	1.41	0.007	11	0.08	6	0.06	72	5.21
OOA - 3	0.76	1.44	0.077	55	0.40	11	0.11	2	0.15
SV – OOA	0.43	1.77	0.013	3	0.02	9	0.09	8	0.60
HOA	0.08	1.89	0.002	7	0.05	9	0.09	2	0.16

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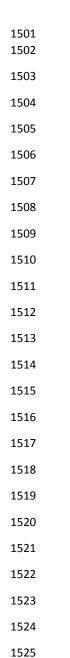
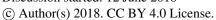




Figure 1: Map depicting the Houston region, the coastal study site (star), and the location of recent stationary campaigns that characterized aerosol dynamics in Houston.

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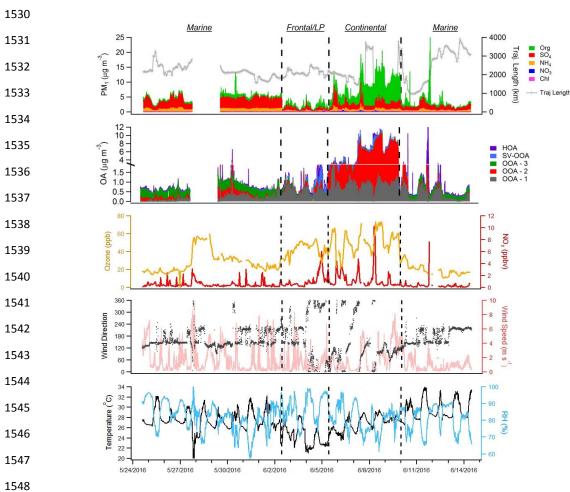


Figure 2: From top to bottom, time series of major NR-PM₁ species and 5-day backward trajectory lengths, extracted PMF factors, O₃ and NO₂, and meteorological variables (wind direction, wind speed, RH, and temperature) measured during the campaign. Dotted lines distinguish distinct time period types



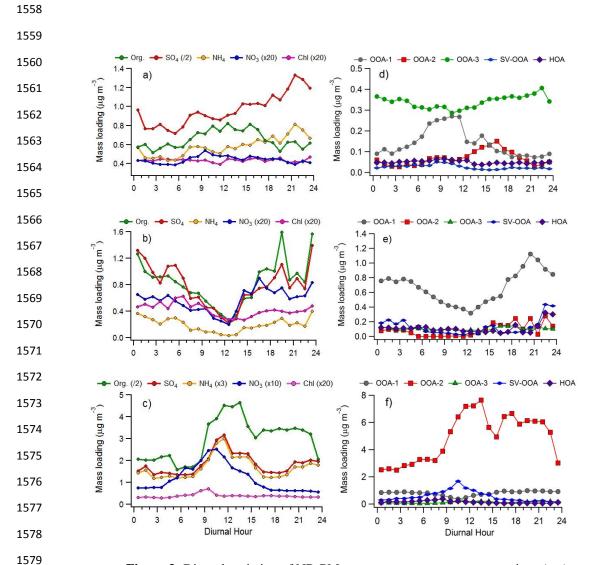


Figure 3: Diurnal variation of NR-PM₁ component average concentrations (a-c) and PMF factors (d-f) during the marine period (a & d), the frontal/LP period (b & e), and the continental period (c & f). The legends above a-c describe how mass loadings of specific components were adjusted to fit the figure. Standard deviations are not included to aid visual distinction of individual NR-PM₁ and OA components. Note the different y-axis ranges applicable to each period type.

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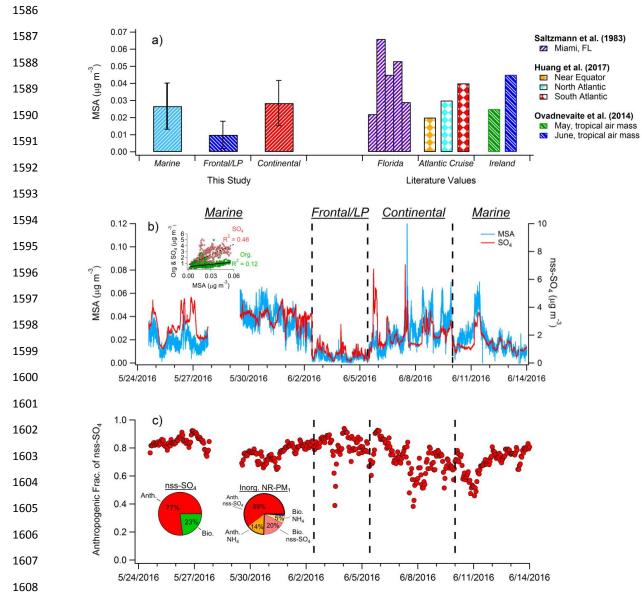


Figure 4: (a) Comparison of average MSA mass loadings measured during each of the three major periods of the campaign with previously published values measured in Florida, on an Atlantic Cruise, and at Mace Head, Ireland. (b) Time series of MSA and nss-SO₄. Black dashed lines denote boundaries of distinct time period types. Inset graph shows the correlation of total OA and SO₄ with MSA during the marine period. (c) Hourly averages of the estimated fraction of nss-SO₄ attributed to anthropogenic sources. Inset pie charts depict anthropogenic and biogenic contributions to nss-SO₄ (left) and total inorganic NR-PM₁ (right) during the marine period. Mass loadings of nitrate and chloride comprise less than 2% of total inorganic aerosol

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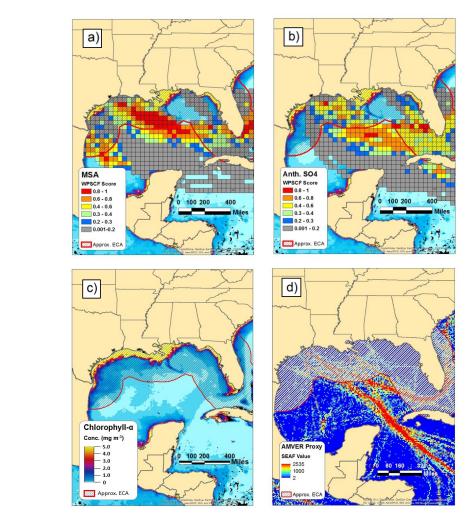


Figure 5: WPSCF plots of MSA (a) and anthropogenic nss-SO₄ (b) for the marine period of the study, along with the chlorophyll-a concentration observed by the NASA MODIS satellite (c) and AMVER shipping spatial proxy map (d). Warmer colors in the WPSCF grid cells indicate higher source probability. The color of each 0.1° x 0.1° grid cell in the AMVER map is based on the corresponding "shipping emissions allocation factor" (SEAF) value (Wang et al., 2008). The hatched region extending from the coasts in each panel represents the approximate area encompassed by the ECA (i.e., 200 nautical miles from the coastal U.S.).

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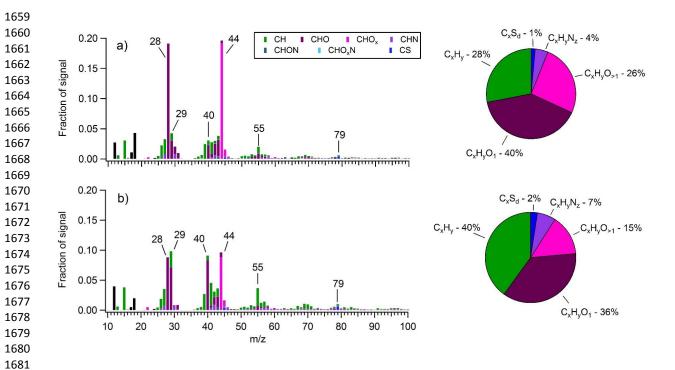


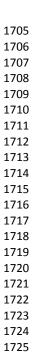
Figure 6: Average OA mass spectra measured during (a) a period heavily influenced by shipping emissions (5/30/2016) and (b) the "marine-biogenic" period when air masses traveled within the ECA for their entire 5-day history. The overall organic fragment composition measured during each period is shown in the corresponding pie charts.

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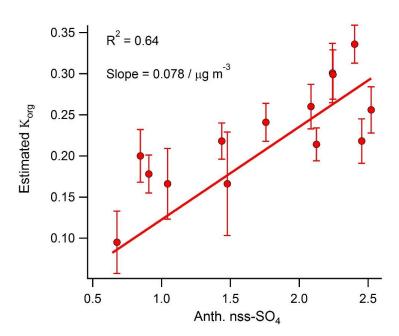


Figure 7: Correlation between daily-averaged anthropogenic nss-SO₄ and the OA hygroscopicity factor calculated using the method developed by Duplissy et al. (2011).

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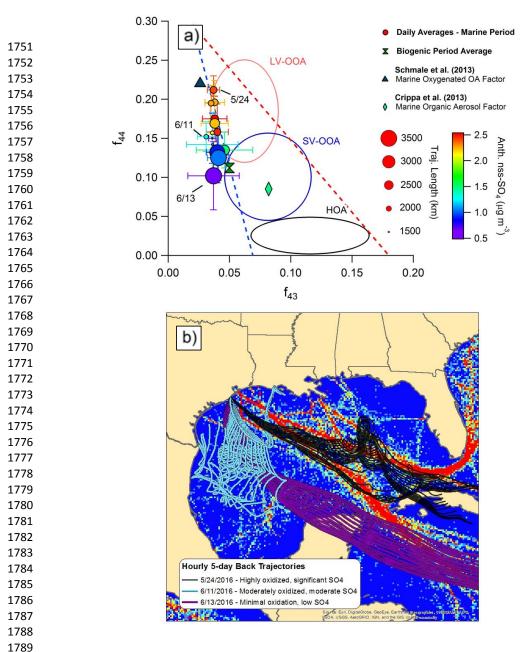


Figure 8: (a) The f₄₄:f₄₃ diagram highlighting different influences on OA oxidation state during the marine period. Circles represent daily average values and bars indicate standard deviations. Periods with HOA mass loadings greater than twice the median value during the marine period were removed from the analysis. Two additional values from published marine studies are shown as a reference. (b) Map of Gulf of Mexico showing hourly 5-day back trajectories calculated for each of the three days identified in the f₄₄:f₄₃ diagram. The AMVER shipping emissions spatial proxy map (Wang et al., 2008) is shown for reference.

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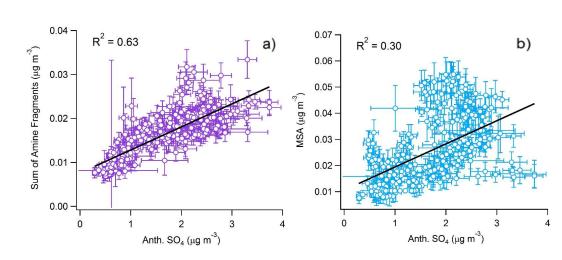


Figure 9: Observed correlation between anthropogenic nss-SO₄ and (a) the sum of measured alkyl amine fragments and (b) MSA.

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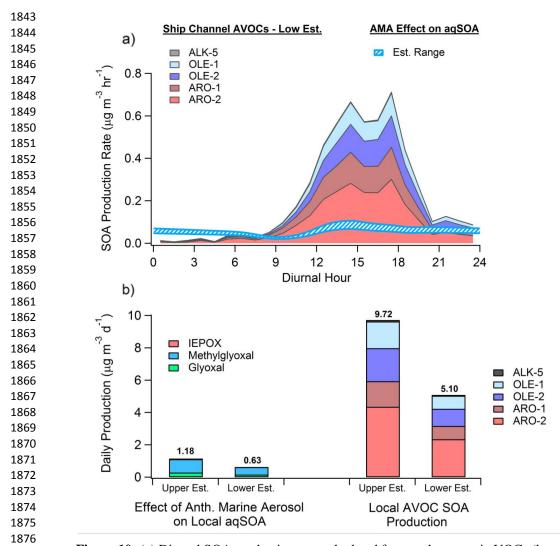


Figure 10: (a) Diurnal SOA production rate calculated from anthropogenic VOCs (low estimate) measured by monitors in the HSC (filled) and the modeled-estimated production rate from WSOGs (glyoxal, methylglyoxal, IEPOX) attributable to the effect of anthropogenic marine aerosol (AMA) (hatched). (b) Total daily SOA production from WSOGs (upper and lower estimates) and HSC AVOCs (upper and lower estimates). WSOG aqSOA production is characterized by individual species. AVOC SOA production is characterized by lumped VOC species defined in Tsimpidi et al. (2010). These species are further described in the SI. Upper and lower estimates of AVOC SOA production are based on the assumed background OA mass loading, as described in the SI. High-NO_x product yields were used for both AVOC estimates.