Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2018-385-RC1, 2018 © Author(s) 2018. This work is distributed under the Creative Commons Attribution 4.0 License.



## Interactive comment on "Nitrogen isotope fractionation during gas-particle conversion of $NO_x$ to $NO_3^-$ in the atmosphere – implications for isotope-based $NO_x$ source apportionment" by Yunhua Chang et al.

## **Anonymous Referee #1**

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Current application of stable isotopes in atmospheric particulate nitrate to partition NOx source contributions generally presupposes that nitrogen isotopic fractionation during the conversion of NOx to NO3- is minor. Here Chang et al. present a comprehensive evaluation of the nitrogen isotope fractionation during gas-to-particle conversion of NOx to NO3-. The computational quantum chemistry is applied to calculate the net N isotope effect (ÔŚN) associated with the conversion between NOx and NO3-, and validated through a source-specific monitoring campaign. The applicability of this method to atmospheric aerosol samples from a megacity shows satisfactory results,

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which are in line to atmospheric chemistry modeling and to what one can expect in terms of source impact in a traffic-intensive environment. The source apportionment model to calculate nitrate fractions of different NOx sources is presented in a clear and concise way and is easily applicable by other researchers for similar studies. Great benefit with the method compared to other  $\delta$ 15N-based source apportionment studies of atmospheric nitrate is the fact that coal combustion may be substantively overestimated in previous studies when the N isotope fractionation during atmospheric nitrate formation is neglected. This makes the study with more profound implications. I recommend this manuscript to be published in ACP with minor revision. 1. Title: replace "gas-particle" by "gas-to-particle" 2. L54: delete "nationwide," 3. L103: add relevant reference 4. L154-155: to my understanding, the source apportionment study of pNO3was only performed in Nanjing 5. L190: it is a bit awkward to use "heartland" here 6. L203-207: I didn't find the data of dicarboxylic acids and related compounds. No need to mention the method here 7. L317: enough credits should be given to previous researchers 8. L319-321: although described in the SI, relevant references should be added 9. Figure 2b: what "MSA" stands for 10. Figure 6: replace "ÔŚN"

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