

Interactive comment on “Effectiveness of Ammonia Reduction on Control of Fine Particle Nitrate” by Hongyu Guo et al.

Anonymous Referee #2

Received and published: 2 June 2018

There is growing interest in controlling ammonia emissions to reduce N deposition and NH_4NO_3 . This study focuses on the non-linearities in the response of particulate matter to changes in ammonia concentration. In particular, they introduce a simple conceptual framework (S-curve) to help explain differences in the sensitivity of PM to NH_3 emissions at different locations and for different seasons. Such non-linearities have often been overlooked in previous studies. This study is within the scope of ACP. It is well written and the methods are sound. I only have few comments for the authors to address before I can recommend this study for publication.

Comments

1) The authors introduce a new conceptual framework to explain seasonal and regional differences in the sensitivity of particulate matter to ammonia emissions. This has

potential policy implications and it would be useful for the authors to compare with other techniques that have been used previously to highlight potential differences.

In particular, previous studies have used the gas-ratio from Ansari and Pandis to interpret global model results (see for instance Pinder et al. (2007, 2008), Paulot (2016), Pozzer et al., 2017) $GR = (TNH_4 - 2 \cdot TSO_4) / TNO_3$ with $0 < GR < 1$ indicating sensitivity to NH_x and $GR > 1$ indicating sensitivity to NH_3 .

Obviously, this cannot directly address variations associated with seasonality. However, based on the information provided in Table S1, $GR \sim < 1$ only for SE US, Virginia, and Pasadena. In other words the weak sensitivity of nitrate to ammonia emissions at the other sites could be inferred simply from concentrations, which is consistent with the findings of the studies mentioned earlier.

In addition, many global models do not use ISORROPIA but simpler (cheaper) aerosol thermodynamic models (see for instance Bellouin et al (2011), Hauglustaine (2014)). Such schemes, which do not explicitly account for aerosol pH, will also simulate a non-linear response of ammonium nitrate to changes in ammonia emissions (see equation A8 in Bellouin et al (2011)). It would be useful for the authors to show how different the response of nitrate and ammonium to changes in ammonia/ NO_x emissions (i.e., Fig 5) would be using such approach.

In particular, this would help strength the case for thinking in terms of aerosol pH rather than simply in terms of concentrations.

2)

I am not convinced by the current discussion of the impact of NH_3 emissions controls on nitrogen deposition. The authors argue that lowering aerosol pH (via lower NH_3 emissions) will modify the ratio of reduced to oxidized nitrogen deposition. However, it is unclear why this is important (no reference is given), especially considering the benefits of lower NH_x deposition and the existence of other removal pathways (wet

[Printer-friendly version](#)[Discussion paper](#)

deposition) that may not exhibit the same sensitivity to the NH₄/NH₃ partitioning. A longer discussion is needed given that this conclusion is highlighted in the abstract.

3) the authors focus on seasonal averages. It would be interesting to discuss whether the sensitivity of particulate matter to NH₃ emissions is different depending on the concentration of NO₃ and whether this would affect the probability distribution of PM under the different emission reduction scenarios shown in Fig. 5. This may be important for policy makers as some standards are based on 24hr averages (https://www3.epa.gov/ttn/naaqs/standards/pm/s_pm_history.html)

Technical comments

4) p4 line 5 NH₃ can also enhance the in-cloud oxidation of SO₂ by O₃. See for instance Wang (2011) or Paulot (2017)

5) p17 line 15 I believe livestock emissions are likely to dominate ammonia emissions in summer.

6) dash black line Fig. 4 not defined

Wang, S., J. Xing, C. Jang, Y. Zhu, J. S. Fu, and J. Hao (2011), Impact assessment of ammonia emissions on inorganic aerosols in east China using response surface modeling technique, *Environ. Sci. Technol.*, 45, 9293–9300.

Paulot, F., S. Fan, and L. W. Horowitz (2017), Contrasting seasonal responses of sulfate aerosols to declining SO₂ emissions in the Eastern U.S.: Implications for the efficacy of SO₂ emission controls, *Geophys. Res. Lett.*, 44, 455–464, doi:10.1002/2016GL070695.

Ansari, A. S. and Pandis, S. N.: Response of Inorganic PM to Pre-cursor Concentrations, *Environ. Sci. Technol.*, 32, 2706–2714, 1998.

Hauglustaine, D. A.; Balkanski, Y. & Schulz, M. A global model simulation of present and future nitrate aerosols and their direct radiative forcing of climate *Atmos. Chem.*

Printer-friendly version

Discussion paper



Phys., 2014, 14, 11031-11063

Pinder et al. Observable indicators of the sensitivity of PM_{2.5} nitrate to emission reductions—Part I: Derivation of the adjusted gas ratio and applicability at regulatory-relevant time scales *Atmospheric Environment* 42 (2008) 1275–1286

Pinder, R. W.; Gilliland, A. B. & Dennis, R. L. Environmental impact of atmospheric NH₃ emissions under present and future conditions in the eastern United States *Geophys. Res. Lett.*, 2008, 35

Pinder, R. W.; Adams, P. J. & Pandis, S. N. Ammonia emission controls as a cost-effective strategy for reducing atmospheric particulate matter in the Eastern United States *Environ. Sci. Technol.*, 2007, 41, 380-386

Paulot, F., Ginoux, P., Cooke, W. F., Donner, L. J., Fan, S., Lin, M.-Y., Mao, J., Naik, V., and Horowitz, L. W.: Sensitivity of nitrate aerosols to ammonia emissions and to nitrate chemistry: implications for present and future nitrate optical depth, *Atmos. Chem. Phys.*, 16, 1459-1477, <https://doi.org/10.5194/acp-16-1459-2016>, 2016.

Aerosol forcing in the Climate Model Intercomparison Project (CMIP5) simulations by HadGEM2-RES and the role of ammonium nitrate Nicolas Bellouin; Jamie Rae; Andy Jones; Colin Johnson; Jim Haywood; Olivier Boucher *Journal of Geophysical Research: Atmospheres* (1984–2012). 2011, DOI: 10.1029/2011JD016074

Interactive comment on *Atmos. Chem. Phys. Discuss.*, <https://doi.org/10.5194/acp-2018-378>, 2018.

Printer-friendly version

Discussion paper

