



## Sources and processes that control the submicron organic aerosol in an urban Mediterranean environment (Athens) using high temporal resolution chemical composition measurements.

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15 **Abstract.** Submicron aerosol chemical composition has been studied during a year-long period (26/07/2016-31/07/2017) and two winter-time intensive campaigns (18/12/2013 – 21/02/2014 and 23/12/2015 – 17/02/2016), at a central site in Athens, Greece, using an Aerosol Chemical Speciation Monitor (ACSM). Concurrent measurements include a Particle-Into-Liquid Sampler (PILS-IC), a Scanning Mobility Particle Sizer (SMPS), an AE-33 Aethalometer and Ion  
20 Chromatography analysis on 24 or 12 hour filter samples. Quality of the ACSM data was assured by comparison versus the above mentioned measurements. The aim of the study was to characterize the seasonal variability of the main fine aerosol constituents and decipher the sources of organic aerosol (OA). Organics were found to contribute almost half of the submicron mass, with concentrations during wintertime reaching up to 200  $\mu\text{g m}^{-3}$ , on occasions. During this season,  
25 the primary sources contribute about 34% of the organic fraction, comprising of biomass burning (10%), fossil fuel combustion (16%) and cooking (8%), while the remaining 66% is attributed to secondary aerosol. The semi-volatile component of the oxidized organic aerosol (SV-OOA; 31%) was found to be clearly linked to combustion sources and in particular biomass burning, and even  
30 a part of the very oxidized, low-volatility component (LV-OOA; 35%) could also be attributed to the oxidation of emissions from these primary combustion sources. These results highlight the



rising importance of biomass burning in urban environments during wintertime, as revealed through this characteristic example of Athens, Greece, where the economic recessions led to an abrupt shift to biomass burning for heating purposes in winter. During summer, when concentrations of fine aerosols are considerably lower, more than 80% of the organic fraction is attributed to secondary aerosol (SV-OOA 30% and LV-OOA 53%). In contrast to winter, SV-OOA appears to result from a well-mixed type of aerosol, linked to fast photochemical processes and the oxidation of primary traffic and biogenic emissions. Finally, LV-OOA presents a more regional character in summer, owing to the oxidation, within a few days, of organic aerosol.

## 40 1. Introduction

The Greater Athens Area (GAA) has over 4 million inhabitants, gathering more than 40% of the total population of Greece in a basin on the west coast of the Attica Peninsula (450 km<sup>2</sup>) (Kanakidou et al., 2011; Grivas et al., 2012). Being surrounded by mountains on three sides, the topography of the area is quite unfavorable to dispersion of air pollutants and ventilation takes place mostly under northeasterly flow (Kanakidou et al., 2011; Grivas et al., 2012; Pateraki et al., 2014). Under low wind speed conditions and the absence of removal via wet deposition, high levels of atmospheric pollutants are observed, significantly deteriorating the air quality of the city (Fourtziou et al., 2017).

The GAA is influenced by long-range transport of atmospheric particulate matter, as well as by significant local sources, namely traffic, industrial activities and combustion of fossil fuel and biomass, the latter especially during wintertime. Particulate matter is considered as the highest priority pollutant to mitigate with respect to population exposure risks, as severe exceedances of the PM<sub>10</sub> air quality standards and elevated concentrations are frequently encountered, especially during the cold period (Chaloulakou et al., 2005).

The economic recession in Greece, making its trace notable on population during the past decade, has initially caused deceleration of industrial activity and reduction of vehicular circulation (Vrekoussis et al., 2013; Paraskevopoulou et al., 2014). At a later stage, the crisis compelled the residents to switch from fossil fuel combustion for domestic heating to sometimes uncontrolled burning of wood and biomass, leading to significant air quality deterioration of an episodic nature (Saffari et al., 2013, Fourtziou et al., 2017, Gratsea et al., 2017). During winter, wood combustion is one of the major sources of organic aerosol (OA) in Europe (Puxbaum et al., 2007; Kalogridis



et al., 2017) and besides primary particles, that consist of a complex mixture of soot, inorganic and organic matter (OM), a considerable amount of volatile organic compounds (VOCs) are also emitted from wood burning, some of which are known to lead to secondary organic aerosol (SOA) formation (Heringa et al., 2012).  
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The emissions and properties of the smoke from burning activities can be highly variable, depending significantly on the type of biomass burned (Lathem et al., 2013). The chemical aging of biomass burning (BB) primary organic aerosol (POA) via photochemistry that leads to the formation of oxygenated organic aerosol (OOA) has been observed during both chamber (Jimenez et al., 2009) and field studies (DeCarlo et al., 2010; Cubison et al., 2011). Moreover, it has been recently observed that even within a few hours of atmospheric processing, freshly emitted biomass burning organic aerosol (BBOA) is transformed to more oxidized OA (Lathem et al., 2013; Bougiatioti et al., 2014), and this processing can also take place during night-time, in the absence of sunlight (Bougiatioti et al., 2014). Existing measurements of aerosol chemical composition in Athens have mainly been performed using 24-h filter samples (or 12-h during intensive campaigns) (e.g Theodosi et al., 2011; Paraskevopoulou et al., 2014). Respective measurements using high resolution techniques are scarce and limited in time (Florou et al., 2017).  
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In this study we present, for the first time, long-term results on the sources of submicron organic aerosols in Athens from high temporal resolution measurements during a year-long period, complemented by two intensive winter campaigns. For the collection of data we deployed an ACSM, a PILS coupled with ion chromatography and auxiliary aerosol and gas phase measurements. The main objectives of this are (i) to characterize the non-refractory part of the submicron aerosol and its variability using high temporal resolution, (ii) to quantify the sources of the organic aerosol and their seasonal variability (via PMF analysis) and (iii) to study the year-to-year changes of aerosol sources during winter time, with special emphasis on wood burning.  
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## 2. Experimental Methods

### 2.1 Sampling site and period

The measurements exploited in this study were conducted, at the urban background site of the National Observatory of Athens at Thissio (37.97N, 23.72E), as representative of the mean population exposure over Athens metropolitan area (Fourtziou et al., 2017). The site stands at an  
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elevation of 105 m above a moderately populated area, where the influence of direct local emissions is limited.

The measurement period lasted for an entire year, from July 2016 to July 2017. Additionally,  
95 two intensive winter campaigns took place at the same site, the first from mid-December 2013 to mid-February 2014 and the second from 23 December 2015 to 17 February 2016. These intensive campaigns aimed at studying the year-to-year variability of biomass burning on the air quality of the city of Athens during wintertime.

## 100 2.2 Instruments and Methods

Measurements were performed with an Aerosol Chemical Speciation Monitor (ACSM, s/n 140-139) by Aerodyne Research (Ng et al. 2011a), measuring the non-refractory PM<sub>1</sub> (NR-PM<sub>1</sub>) aerosol mass and chemical composition in near real-time (30-minute temporal resolution). The instrument was sampling through a BGI Inc. SCC 1.197 Sharp Cut Cyclone operated at 3 L min<sup>-1</sup>,  
105 yielding a cut off diameter of approximately 2 μm. The ACSM measures mass and provides real time chemical composition (organics, sulfate, nitrate, ammonium and chloride) for non-refractory aerosol particles. The design of the instrument limits these measurements in the submicron range; more specifically the particles that reach the final stage of the instrument lie in the size range from 40 to 700 nm. Practically the ACSM operates following a similar principle as the Aerosol Mass Spectrometer (AMS) (Jayne et al., 2000) where ambient air is drawn through a critical orifice to a particle focusing aerodynamic lens; the resulting particle beam is flash-vaporized at 600°C, ionized  
110 via electron impact ionization and guided through a quadrupole mass spectrometer. The instrument has participated in an intercomparison study of 15 Q-ACSM instruments organized by the European Center for Aerosol Calibration (ECAC) at the Aerosol Chemical Monitor Calibration Center (ACMCC) at Site Instrumental de Recherche par Télédétection Atmosphérique (SIRTA, Paris) during March 2016 ([www.actris-ecac.eu/acsm-2016-1.html](http://www.actris-ecac.eu/acsm-2016-1.html)). It showed excellent agreement (well below ±20%) with the ACMCC's reference instrument, in both pre- and post-calibration periods of the intercomparison study, while the relative ionization efficiencies (RIE) for ammonium and sulfate, obtained by the on-site calibration, were very close to the ones previously  
115 used, indicating a stable performance of the instrument. The detection limits for the ACSM provided by Ng et al. (2011a) are: 0.284 μg m<sup>-3</sup> for ammonium, 0.148 μg m<sup>-3</sup> for organics, 0.024  
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$\mu\text{g m}^{-3}$  for sulfate,  $0.012 \mu\text{g m}^{-3}$  for nitrate, and  $0.011 \mu\text{g m}^{-3}$  for chloride. Mass concentrations are calculated with the recommended collection efficiency of 0.5 for all constituents.

Parallel measurements were performed for biomass burning identification, but also for quality control purposes. In this context, a Metrohm ADI 2081 Particle Into Liquid Sampler (Orsini et al., 2003) coupled with Ion Chromatography (Dionex ICS-1500) was used, which was sampling ambient air from a different, but adjacent to the ACSM's,  $\text{PM}_{10}$  inlet. The ion Chromatograph was set to measure cations such as ammonium and potassium at a time resolution of 15 minutes. The resulting concentrations from the ACSM were tested against filter measurements and the concentrations provided by the PILS. For the PILS, the detection limit was calculated at 1 ppb for  $\text{Na}^+$ ,  $\text{NH}_4^+$ ,  $\text{Mg}^{+2}$  and  $\text{Ca}^{+2}$  and 2 ppb for  $\text{K}^+$ . Reported concentrations were blank corrected.

Two different absorption photometers were monitoring Black Carbon (BC) concentrations. A 7-wavelength Magee Scientific AE-42 portable aethalometer was used for the 2013-14 and 2015-16 winter campaigns, providing 5-min resolution measurements. For the year-long period a dual spot, 7-wavelength Magee Scientific AE-33 aethalometer (Drinovec et al., 2015) was used, operating at 1-min resolution. Standard gas analyzers monitoring, like  $\text{O}_3$  (Thermo Electron Co., model: 49i),  $\text{CO}$ ,  $\text{SO}_2$  and  $\text{NO}_x$  concentrations (HORIBA, 360 series), a  $\text{PM}_{10}$  beta-attenuation monitor for aerosol mass concentration (Eberline, FH 62 I-R) and a Scanning Mobility Particle Sizer for  $\text{PM}_{10}$  size distributions (SMPS 3034, TSI Inc.) were also operating at the sampling site, as part of its routine measurements. Wavelength dependent source apportionment of the BC load was performed providing a fossil fuel ( $\text{BC}_{\text{ff}}$ ) and a wood combustion ( $\text{BC}_{\text{wb}}$ ) generated factor. An absorption exponent of 1 for fossil fuel combustion and 1.9 for pure wood burning based on the approach of Sandradewi et al. (2008) was used, very close to the respective values of 0.9 and 2, used in a suburban site in Athens (Kalogridis et al., 2017). Necessary meteorological parameters for the study were taken from the historic actinometric meteorological station of NOAA, at Thissio (Kazadzis et al., 2018). All measurements were averaged to 1-hour intervals in order to synchronize the different data sets.

Four-day back trajectories were calculated using the HYbrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT\_4) model (Draxler and Hess, 1998) developed by the ARL/NOAA, and 1-degree GDAS (NCEP) meteorological data. Trajectories were computed every 3-h, for air masses arriving at Athens at a height of 1000 m. The selected height is considered suitable to capture transport at a representative upper limit of the boundary layer in Athens



(Markou and Kassomenos, 2010). Trajectory clustering was performed using the TrajStat plugin (Sirois and Bottenheim, 1995; Wang et al., 2009) of the MeteInfo GIS software  
155 (<http://www.meteothinker.com/products/meteoinfo.html>). The change of the total space variance for decreasing number of clusters was examined as a criterion for cluster number selection. The analysis was performed separately for summer and winter, resulting in 5 clusters for each period.

### 2.3 Source apportionment using Positive Matrix Factorization analysis

160 Positive Matrix Factorization (Paatero and Tapper, 1994) was performed on the organic spectra obtained by the ACSM. The graphic interface SoFi (Source Finder) version 6.1, developed at PSI, Zurich (Canonaco et al., 2013) was used. SoFi implements the multilinear engine algorithm ME-2 (Paatero and Hopke, 2003), analyzing the acquired mass spectral timeseries matrix into a linear combination of source profile and time series sub-matrices.

165 Briefly, the organic mass spectra measured by the ACSM are represented as a matrix X, with columns representing the different m/z's and rows the time series of the mass spectrum. Then the measured matrix X is approximated by the product of F and G, added to E ( $X=GF+E$ ) where each column of matrix G represents the time series of a factor and each row of F represents the profile (mass spectrum) of this factor and E is the model residual. The entries in F and G are then fitted  
170 using a least squares algorithm that iteratively minimizes the quantity  $Q^m$ , which is the sum of the squared residuals weighted by their uncertainties. Finally, at each step of the solution process, outliers are defined based on the ratio of residuals to uncertainties (Canonaco et al., 2013).

The input organics and organics' error matrices are derived automatically from the ACSM data analysis software, using a simple automated procedure. Several PMF runs were realized, following  
175 the methodology proposed by Crippa et al. (2014). In brief, initially unconstrained runs provided insight on the potential number of factors. For the following steps, primary OA factors were deconvolved by introducing constrains to known factor profiles, implementing the  $\alpha$  value approach, while potential secondary oxygenated components were left unconstrained. The  $\alpha$  value ranges between 0 and 1 and is a measure of how much the resolved factors are allowed to vary  
180 from the input ones. Next, the model's residuals were analyzed in search of structures that could indicate underestimation or overestimation of the number of separated factors. The year-long data series was divided into a cold (November 2016 to March 2017) and warm (August to September 2016 and May to July 2017) period as sources of organics and thus mass spectra and intensities



are expected to differ considerably. A detailed description on the source apportionment method  
185 applied can be found in the following section.

### 3. Results and Discussion

#### 3.1 Comparison of ACSM data with ancillary measurements

As a first quality control/quality assurance of the ACSM data, the ammonium concentrations  
190 are compared to the respective ones derived from the PILS, on an hourly basis for winter 2016-17.  
A very good agreement is found (squared Pearson correlation coefficient  $R^2=0.83$ , slope of 0.98).  
The sulfate and nitrate concentrations for the winter 2016-2017 period are compared to the  
respective ones from the ion chromatography analysis ( $PM_{2.5}$  filters), on a daily basis ( $R^2=0.69$ ,  
slope of 0.95 and  $R^2=0.81$ , slope of 0.95, respectively). The organics concentrations are compared  
195 to the organic carbon concentrations of the  $PM_{2.5}$  filters determined by the thermal-optical method  
(SUNSET Laboratory Inc.) using the EUSAAR-2 protocol (Cavalli et al., 2010). A very good  
agreement is found (squared Pearson correlation coefficient  $R^2=0.94$ , slope of 1.77) with the slope  
being close to values reported for urban areas (Petit et al., 2015) and OM:OC calculations from  
AMS measurements in polluted environments (Saarikoski et al., 2012). The results from the  
200 aforementioned comparisons are provided in the Supplementary material (SI.1.1).

During the intensive winter 2015-2016 campaign, the concentrations of the ACSM components  
are compared to those determined from the ion chromatography, based on concurrent filter samples  
collected at the same site, twice per day, (06:00 - 18:00 pm and t 18:00 - 06:00 local time). Results  
indicate an excellent agreement for sulfate ( $R^2=0.88$ , slope of 0.92), ammonium ( $R^2=0.81$ , slope  
205 of 1.04), and nitrate ( $R^2=0.87$ , slope of 1.08) (Figure SI.1.2). During the intensive winter 2013-  
2014 campaign, the ammonium concentrations from the ACSM showed significant correlation  
with the respective ones from the PILS ( $R^2=0.81$ , slope of 0.97).

Finally, the sum of the ACSM component concentrations plus BC, measured with the 7-  
wavelength aethalometer was compared with the mass concentrations determined by a Scanning  
210 Mobility Particle Sizer (SMPS; TSI 3034) measuring particle number concentrations in the range  
of 10.4-469.8 nm (since February 2017 at Thissio). The results obtained using a constant collection  
efficiency of 0.5, are portrayed in Figure 1 and indicate strong correlation (squared Pearson  
correlation coefficient  $R^2=0.88$ ) a slope of 0.90 and an insignificant intercept of 0.01. The slight  
underestimation of the SMPS-derived submicron mass can be attributed to the instrument's upper



215 limit diameter (469.8 nm versus 700 nm for the ACSM). It should also be noted that this  
discrepancy could be attributed to the uncertainty in the estimation of aerosol density used for the  
conversion of volume to mass concentration of the SMPS (Bougiatioti et al., 2014).

### 3.2 PM<sub>1</sub> average chemical composition and seasonality

#### 220 3.2.1 Chemical composition and characteristics

The time series of the main submicron aerosol components measured by the ACSM and the  
black carbon concentrations are presented in the upper panel of Figure 2 (one complete year  
period). The average cumulative concentration of the ACSM components and BC was  $15.1 \pm 16.7$   
 $\mu\text{g m}^{-3}$ . The highest concentrations were measured during winter (average  $20.9 \pm 26.4 \mu\text{g m}^{-3}$ ) and  
225 the lowest during summer (average  $9.1 \pm 6.2 \mu\text{g m}^{-3}$ ). On an annual basis, the most abundant  
component was organic aerosol, followed by sulfate, contributing 46.1 and 29.1% to the total  
submicron mass, respectively, while BC contribution was calculated at 12%, ammonium 8.5% and  
nitrate 4.3%. In the middle and bottom panels of Figure 2 the respective time series of the main  
submicron aerosol components during the two intensive 2-month winter campaigns are presented.  
230 During winter 2013-14 the average mass concentration of the ACSM components (plus BC  
concentrations) was  $25.1 \pm 29.9 \mu\text{g m}^{-3}$ , with organics and sulfate contributing 53.5 and 15.9% to  
the total submicron mass, respectively, followed by BC (12.9%). During winter 2015-16 the  
average concentration was  $19.5 \pm 25.3 \mu\text{g m}^{-3}$ , with organics and BC contributing 51.2 and 17.9%  
to the total submicron mass, respectively, followed by sulfate (17.6%), nitrate (6.7%) and  
235 ammonium (6.6%). It is clearly deduced that during the last winters, organics constitute half or  
even more of the total PM<sub>1</sub> mass, sulfate around 20% and BC around 14%.

What is also striking is the fact that during wintertime, PM<sub>1</sub> concentration spikes can reach up  
to  $240 \mu\text{g m}^{-3}$  hourly values, with organics taking up most of the mass. Maxima are recorded during  
night-time and mostly during meteorological conditions favoring pollutants emission and  
240 accumulation, such as low wind speed and low temperature (Fourtziou et al., 2017). There are on  
average 8 such incidents occurring during each winter (10 in 2013-14, 7 in 2015-16 and 8 in 2016-  
17), with organic levels being higher than  $100 \mu\text{g m}^{-3}$ . Such levels are the highest reported for  
Europe during wintertime and highlight the impact of local emissions and especially those related  
to heating/wood burning (see below), to the levels of organics and consequently PM<sub>1</sub>. These  
245 observations are in accordance with Florou et al. (2017, same site from 10 January until 9 February





2013), whereas organics concentration alone reached up to  $125 \mu\text{g m}^{-3}$  and maxima of  $8 \mu\text{g m}^{-3}$  for BC and up to  $5 \mu\text{g m}^{-3}$  for nitrate, were recorded. Similarly, wintertime pollution events with increased local character and elevated organics concentrations (around  $100 \mu\text{g m}^{-3}$ , average of  $22.6 \mu\text{g m}^{-3}$ ) have been reported at a regional background site, just outside of Paris, during February 250 2012 (Petit et al., 2015).

### 3.2.2 Seasonal variability

The seasonal variability of the main measured species, along with the average  $\text{PM}_{10}$  concentration ( $\mu\text{g m}^{-3}$ ), as calculated from the ACSM+BC measurements is shown in Figure 3 and 255 the basic statistics are included in Table 1. Organics contribute 43.9% to the total submicron aerosol mass in summer, followed by sulfate (33.1%), ammonium (12.5%) and BC (7.6%), while in winter, organics and sulfate contribute 49.1 and 24.1%, respectively, followed by BC (12.1%), ammonium (6.9%) and nitrate (6.7%).

The mass concentrations of organics, nitrate, chloride and BC exhibit a clear annual cycle, with 260 minimum during summer and maximum in winter. This pattern seems to be due to a combination of three simultaneous processes. At first, the additional primary emissions from heating play an important role. A second reason could be the decreased boundary layer depth during winter. According to Kassomenos et al., 1995, daytime PBL depth shows a clear annual cycle, with maxima during the warm months (June to September) and exhibiting a two-fold decrease during 265 wintertime. Finally, the effect of temperature to the partitioning of the semi-volatile inorganics can also contribute to the processes leading to the observed pattern. In support of the above, larger standard deviation is found in winter, demonstrating the frequency and magnitude of the observed pollution events due to the increased need for heating purposes (Fourtziou et al., 2017). Independently of the year, it can be seen that winter concentrations of organics, nitrate, chloride 270 and BC are very similar and more than twice the respective ones during the rest of the seasons (Table 1).

Organics concentration are consistently high during all studied winters (from December to February), while the higher nitrate values can be also attributed to the increased local sources combined with the overall lower temperatures, which favor the stability of ammonium nitrate (Park 275 et al., 2005; Mariani and de Mello, 2007). Ammonium and sulfate exhibit the opposite seasonal cycle, with maximum values in summer and minimum during winter and spring. The higher



summer sulfate levels are the result of enhanced photochemistry associated with more intense insolation, combined with less precipitation, favoring the regional transport of polluted air masses (Cusack et al., 2012). The seasonal variation of concentrations is in agreement with that observed  
280 in Athens, during prior long-term measurement campaigns based on analysis of daily filter samples (Theodosi et al. 2011, Paraskevopoulou et al., 2014; 2015).

### 3.2.3 Diurnal variability

When investigating the diurnal patterns of the measured species (Figure 4), it is observed that  
285 during wintertime, ammonium and sulfate do not exhibit any significant variability, which is due to the more regional character of ammonium sulfate (Seinfeld and Pandis, 2016). In order to quantify the extent of this variability we calculated the normalized diurnal pattern by dividing each hourly value with the respective mean concentration. More specifically, sulfate varies by 12.5% around the mean value while ammonium varies by 33%. On the other hand, organics, BC and  
290 nitrate vary significantly during the day (83.4%, 79.8% and 74.3% respectively). These species clearly double their concentrations during night-time, caused by the additional primary emissions. Furthermore, BC and nitrate also exhibit a second maximum during early morning hours, which should be attributed to the primary emissions during the morning traffic rush-hour.

During summer, all concentrations are significantly lower, especially organics (note scale  
295 change) which exhibit a 5-fold decrease of their mean maximum concentration during night-time. Normalizing the diurnals, as mentioned above, reveals a much less pronounced variability for organics (46.9%), implying a more regional character, while BC and nitrate exhibit the highest variability (67.7% and 57.6% respectively) in accordance to their local nature. The night-time maxima of BC vanishes, nitrate shows much lower concentrations, due to nitrate partitioning  
300 between gas and aerosol phase, favoring the vaporization of ammonium nitrate. BC still exhibits only one maximum during early morning hours owing to traffic emissions. Ammonium and sulfate diurnal profile follows expected photochemistry patterns, with peaking concentrations around 14:00 LT (UTC+2), consistent with secondary aerosol formation and increased vertical mixing with regional aerosol from aloft due to the evolution of the convective boundary layer which  
305 exhibits a bell shaped diurnal structure ranging from a few hundred meters to above one kilometer, with maximum heights during early afternoon (Asimakopoulos et al., 2004; Tombrou et al., 2007). Sulfate concentrations exhibit lower night-time background, a concurrent (to winter) primary



maximum (8:00-10:00 LT), and a secondary significant increase later in the afternoon (twice the night-time background). Finally, organics concentrations are somewhat higher during early night  
310 which could possibly be associated with biogenic/vegetation sources either local or regional that produce volatile compounds and condense on the particulate phase during night when temperatures are lower. Furthermore organic variation also follows the morning peak related to traffic and the late afternoon peak also observed for ammonium and sulfate. Condensation of the particulate phase could also apply for nitrate, which also exhibits higher concentrations during night-time (almost  
315 double).

### 3.3 Source apportionment of organic aerosol

A source apportionment analysis of the organic aerosol data was carried out, separately for the cold (1 November to 31 March) and the warm period (July-August-September 2016 and May-July  
320 2017). The separation served to better characterize the different profiles of the sources, as it is expected that the different  $m/z$  intensities vary throughout the year. Several runs were performed, following the technical guidelines and methodology proposed by Crippa et al. (2014). Initially, an unconstrained run with the number of factors ranging from 2 to 8 was performed, allowing for a first estimate of the number of potential deconvolved primary and secondary OA spectra. After  
325 identifying the presence of primary OA, namely hydrocarbon-like organic aerosol (HOA), cooking organic aerosol (COA) for both periods and BBOA for the winter period, several constrained runs were performed using the  $\alpha$  value approach.

In this context, the HOA mass spectrum was first constrained with a low  $\alpha$  value, ranging from 0.05 to 0.1, using as reference the average HOA mass spectrum from Ng et al. (2011b). COA was  
330 also constrained with an  $\alpha$  value ranging from 0.1 to 0.2, using as reference the mass spectrum from Crippa et al. (2013) reported for Paris. Finally, for the three wintertime periods, BBOA was also constrained with higher  $\alpha$  values, (between 0.3 and 0.5) using as reference the average BBOA mass spectrum from Ng et al. (2011b). Only  $m/z \leq 120$  were used in order to avoid interferences from the naphthalene signal ( $m/z$  127, 128 and 129). Weak signals, with signal-to-noise ratio (S/N)  
335 below 0.2 were downweighted by a factor of 10, and those with S/N between 0.2 and 1 were downweighted by a factor of 2 (Ulbrich et al., 2009), using the aforementioned customized software. Solutions were chosen based on specific indicators, such as the  $Q/Q_{\text{exp}}$  ratio, residual analysis, reproducibility of factors for different model seeds, the affinity of the deconvolved



spectra with spectra found in the AMS mass spectral database ([http://cires.colorado.edu/jimenez-](http://cires.colorado.edu/jimenez-group/AMSsd/)  
340 [group/AMSsd/](http://cires.colorado.edu/jimenez-group/AMSsd/)) and also the correlation of the time series with external time series (such as BC, NO, CO and nss-K<sup>+</sup>). The profiles of all solutions were inter-compared as means for sensitivity analysis and the results are provided in the Supplementary material (SI.2). The affinity between spectra is expressed using the  $\theta$  angle approach (Kostenidou et al., 2009). In this context each deconvolved mass spectrum represents a vector. The cosine of the  $\theta$  angle between two such  
345 vectors (e.g. MS<sub>a</sub> and MS<sub>b</sub>) is equal to the correlation coefficient  $R$  between them. The  $\theta$  angle is calculated using the dot product formula:

$$\theta = \cos^{-1} \left( \frac{MS_a MS_b}{|MS_a| |MS_b|} \right)$$

Calculated angles less than 15° correspond to  $R > 0.96$  thus indicate spectra which are similar to each other, angles between 15° and 30° correspond to  $0.96 > R > 0.86$  and indicate some similarity  
350 but also some important differences between the compared spectra, while angles larger than 30° correspond to correlation coefficient  $R < 0.86$  and thus are considered to indicate spectra that do not compare well.

*Warm period:* In this period, the selected solution is a two factor constrained run (HOA using  $\alpha = 0.05$  and COA using  $\alpha = 0.1$ ) and consists of four factors: HOA (hydrocarbon-like OA), COA-  
355 like (cooking-like OA.), SV-OOA (semi-volatile oxygenated OA) and LV-OOA (low-volatility oxygenated OA). The two summer periods have also been treated separately, but the derived spectra were almost identical ( $\theta$  angles between 1.6 and 11.9 for all derived factors, see SI). The time series of the four identified sources during summer 2017 is shown in Figure 5 along with their diurnal variability and the respective average daily contribution. The mass spectra of the selected  
360 solution are also provided in the supplementary material. No primary biomass burning aerosol could be identified, which is justified by the absence of fresh emissions over the city center during the warm period. In the summer periods HOA makes up 5.5% of the total organic fraction, COA around 12% on average (10.9 and 13% for 2016 and 17, respectively). In summer 2016 SV-OOA made up 30.5% and the rest 53% is LV-OOA. In summer 2017, SV-OOA contributes 46% to the  
365 total organic fraction while LV-OOA 35%. The dominance of secondary influence (SV-OOA & LV-OOA) is apparent, and accounts for the majority of the organic aerosol. This finding is in accordance with Kostenidou et al. (2015), who reported that 65% of the sampled aerosol during summer can attributed to SOA (SV-OOA & LV-OOA), at a suburban site in Athens.



Since a strong constraint was used for COA ( $\alpha=0.1$ ), the derived COA mass spectrum is almost  
370 identical to the mass spectrum of wintertime COA in Paris (Crippa et al., 2013) with which the  
constraint was made, having a  $\theta$  angle of  $7^\circ$ . COA also shares many similarities with the mass  
spectrum of fresh organic aerosol emissions from meat charbroiling (Kaltsonoudis et al., 2017)  
( $\theta=14.5^\circ$ ). When calculating the O:C ratio in COA following the study of Canagaratna et al. (2015)  
we find a ratio of 0.19, which is comparable to the value of 0.24 obtained for COA during summer  
375 at a suburban site in Athens (Kostenidou et al., 2015). Similarly, due to the constraint ( $\alpha=0.05$ )  
HOA is identical ( $\theta=2.5^\circ$ ) with the average HOA spectrum (Ng et al., 2011b), obtained by  
averaging together different HOA mass spectra, but is also very similar to HOA reported by Crippa  
et al. (2014) ( $\theta=4.5^\circ$ ) and HOA from Paris ( $\theta=6.5^\circ$ ) (Crippa et al., 2013). SV-OOA is similar to  
the SV-OOA reported by Crippa et al. (2014) ( $\theta=14^\circ$ ), and also has many similarities with the  
380 average SV-OOA ( $\theta=16.5^\circ$ ) (Ng et al., 2011b) as well as with the laboratory-generated SOA from  
isoprene via the reactive uptake of epoxydiols (IEPOX) ( $\theta=16.2^\circ$ ) (Budisulistiorini et al., 2013).  
This type of aerosol is the oxidation product of isoprene, denoting a possible link of SV-OOA with  
biogenic aerosol. This association is further strengthened by considering its similarities with SOA  
from biogenic precursors, such as  $\alpha$ - and  $\beta$ -pinene ( $\theta=20^\circ$  and  $18^\circ$ , respectively) (Bahreini et al.,  
385 2005), which are maximum during night similarly to SV-OOA. On the other hand, the derived SV-  
OOA shares some similarities with SOA from diesel exhaust after 4 h of photochemical ageing  
(Sage et al. 2008). Finally, SV-OOA does not compare well with the mass spectrum from aged  
organic aerosol emissions from meat charbroiling (Kaltsonoudis et al., 2017) ( $32^\circ < \theta < 37^\circ$ ). This  
indicates that during summer, it is not linked to the oxidation of primary COA, but rather to SOA  
390 formation from the oxidation of VOCs from both biogenic and traffic sources. Finally, LV-OOA  
is identical ( $\theta=3.6^\circ$ ) with the very oxidized regional OOA found in the area (Finokalia, Crete)  
(Bougiatioti et al., 2014) and has many similarities with the LV-OOA reported by Crippa et al.  
(2014) ( $\theta=11.7^\circ$ ) and with the average LV-OOA ( $\theta=18.2^\circ$ ) (Ng et al., 2011b). When calculating  
the elemental ratios based on the study of Canagaratna et al. (2015), the O:C ratio for LV-OOA is  
395 1.2, which is identical to the value of OOA obtained at Finokalia (Bougiatioti et al. 2014).

HOA correlates significantly with nitrate ( $R^2=0.63$ ) as well as with BC ( $R^2=0.52$ ) while COA  
shows moderate correlation with CO ( $R^2=0.32$ ) and nitrate ( $R^2=0.36$ ). SV-OOA is highly  
correlated with nitrate ( $R^2=0.89$ ), implying common mechanisms in their variability, possibly  
linked with the partitioning between the gas and particulate phases. The moderate correlation with



400 CO ( $R^2=0.39$ ) and BC ( $R^2=0.39$ ) implies that SV-OOA may, to some extent, partially originate  
from a combustion source. LV-OOA shows moderate correlation with sulfate ( $R^2=0.47$ ) and  
ammonium ( $R^2=0.44$ ), consistent with the regional character of this factor. Results from the  
trajectory cluster analysis (Figure 8) show that enhanced LV-OOA levels are related to air masses  
405 originating from Eastern Europe and the Black Sea region, which have both been identified as the  
main areas of influence for secondary aerosols that are regionally processed and transported to  
Athens (Gerasopoulos et al., 2011; Grivas et al., 2018).

Primary fossil fuel emissions (HOA) are very low during summer, as in July and August most  
of the Athenians leave for their summer vacations, thus reducing local traffic. Concentrations peak  
around 7:00 and after 19:00 LT that corresponds to the early morning and evening rush hours in  
410 downtown Athens. COA exhibits a slight hump during lunch hours (12-15:00 LT), also seen in the  
relative contribution of the factor, while a large night-time peak is present at around 22:00 LT.  
This late peak is consistent with the late dinner hours and operation of tavernas (typical restaurants)  
in Athens during the touristic season. SV-OOA exhibits two-fold higher concentrations during  
night-time, which apart from boundary layer dynamics may also be attributed to the condensation  
415 of semi-volatile compounds, as also implied by the excellent correlation of SV-OOA with nitrates.  
Finally, LV-OOA exhibits a peak during mid-day, consistent with increased photochemical  
processes that lead to further organic aerosol oxidation. In coincidence to that a moderate hump is  
also observed for SV-OOA.

In summary, during the warm period, the vast majority (more than 80%) of organic aerosol in  
420 the area is linked to secondary organic aerosol formation. The semi-volatile product is of mixed  
origin, linked to quick atmospheric processes, within a few hours, such as photochemistry of  
primary sources, like biogenic emissions from vegetation, traffic emissions, or to a lesser extent  
regional biomass burning. On the contrary, the low-volatility product is the result of more  
extensive oxidation of organic aerosol in the area, within a few days, and has, thus, a more regional  
425 character.

*Cold period:* In this period, the selected solution is a three factor constrained run (HOA using  
 $\alpha =0.1$ , COA using  $\alpha =0.2$ , BBOA using  $\alpha =0.4$ ) and consists of five factors: BBOA (biomass  
burning OA), HOA, COA, SV-OOA and LV-OOA. The HOA spectra are identical for all studied  
winter periods ( $1.2^\circ < \theta < 5.2^\circ$ ) and so are the COA ( $3.3^\circ < \theta < 3.9^\circ$ ) and LV-OOA spectra  
430 ( $3.2^\circ < \theta < 3.6^\circ$ ). BBOA spectra still have many similarities between them ( $9.7^\circ < \theta < 13.3^\circ$ ) while SV-



OOA also exhibit similarities ( $12.9^\circ < \theta < 17.1^\circ$ ). Indicatively, the solution for winter 2016-17 is presented (Fig. 6), while the respective solutions for winter 2013-14 and 2015-16 are provided in the supplementary material (Fig. SI.2.3). The time series of the five PMF factors for winter 2016-17 are shown in Figure 6 along with their diurnal variability and the hourly contribution of each factor.

HOA during wintertime 2016-17, is almost identical to the average HOA spectrum (Ng. et al., 2011b), as expected due to the constraint, having a  $\theta$  angle of  $5.1^\circ$ , HOA from Crippa et al. 2014 ( $\theta=5^\circ$ ) and HOA from Paris during winter (Crippa et al., 2013) ( $\theta=5.8^\circ$ ) and from Hyytiälä (Äijälä et al., 2017) ( $\theta=9.6^\circ$ ). COA is also almost identical to COA from Paris during winter (Crippa et al., 2013) having a  $\theta$  angle of  $6.2^\circ$ . It is also very similar to fresh organic aerosol emissions from meat charbroiling (Kaltsonoudis et al., 2017) ( $\theta=13.2^\circ$ ) and it shares similarities with COA from Hyytiälä (Äijälä et al., 2017) ( $\theta=16.9^\circ$ ). When calculating the elemental ratios based on the study of Canagaratna et al. (2015) the O:C ratio for COA is 0.18, which is in accordance with the value of 0.11 derived for COA at the same site by Florou et al. (2017). BBOA is very similar to the average BBOA spectrum (Ng. et al., 2011b) having a  $\theta$  angle of  $11.5^\circ$ , sharing similarities with biomass burning aerosol from oak ( $\theta=15.7^\circ$ ), spruce ( $\theta=19.8^\circ$ ) (Schneider et al. 2006), BBOA from Crippa et al. (2014) ( $\theta=20^\circ$ ) and with BBOA during wintertime in Zurich (Lanz et al., 2008) ( $\theta=23.4^\circ$ ). Once more, the calculated O:C ratio for BBOA is 0.25, which is in accordance with the value of 0.27 derived for BBOA at the same site by Florou et al. (2017). SV-OOA spectrum shares similarities with the average SV-OOA from Ng et al. (2011b) ( $\theta=21.3^\circ$ ), as well as with the IEPOX-OA from Budisulistiorini et al. (2013) ( $\theta=22.8^\circ$ ), as isoprene main oxidation products such as methyl vinyl ketone and methacrolein are often used as biomass burning tracers (Santos et al., 2017). Similar correlation is also found with IEPOX-OA and SV-OOA during the winter 2015-16 campaign (Supplementary material, SI.2). It shares some similarities with SV-OOA from wintertime in Paris (Crippa et al., 2013) ( $\theta=25.2^\circ$ ) and SV-OOA from Hyytiälä (Äijälä et al., 2017) ( $\theta=26.2^\circ$ ). Finally, LV-OOA is almost identical ( $\theta=8.3^\circ$ ) with the LV-OOA from Crippa et al. (2014), the average LV-OOA from Ng et al. (2011b) ( $\theta=8.4^\circ$ ), LV-OOA from Zurich during winter (Lanz et al., 2008;  $\theta=9.3^\circ$ ) while it also shares many similarities ( $\theta=15.1^\circ$ ) with the oxidized OOA found in the extended area (Finokalia Crete) (Bougiatioti et al., 2014).

Since the identification of BBOA is mainly based on the two fragments of  $m/z$  60 and 73, considered as the “fingerprint” fragments of levoglucosan and biomass burning tracers, BBOA



exhibits indeed excellent correlation with these two fragments ( $R^2=0.94$  and  $0.9$ , respectively).  $\text{Nss-K}^+$  is also proposed as a very good tracer for biomass burning and as is reported by Fourtziou et al. (2017), it shows a significant correlation with BC coming from wood burning ( $\text{BC}_{\text{wb}}$ ), during  
465 wintertime in Athens. Consequently, the time series of  $\text{nss-K}^+$  provided by PILS-IC and  $m/z$  60 are studied together. It appears that during both winters (2013-14 and 2016-17) for which  $\text{nss-K}^+$  data is available,  $m/z$  60 is in very good agreement with  $\text{nss-K}^+$  ( $R^2=0.85$ ) (Figure 7a). Furthermore BBOA correlates well with  $\text{BC}_{\text{wb}}$  ( $R^2=0.78$ ),  $\text{nss-K}^+$  ( $R^2=0.62$ ) and with CO ( $R^2=0.52$ ). SV-OOA correlates significantly with both wood burning “fingerprint” fragments of  $m/z$  60 and 73 ( $R^2=0.99$   
470 for both),  $\text{BC}_{\text{wb}}$  ( $R^2=0.82$ ), CO ( $R^2=0.61$ ) (Figure 7b) and  $\text{nss-K}^+$  ( $R^2=0.61$ ), demonstrating the direct link between SV-OOA and primary combustion sources (mainly biomass burning). It can be seen in Figure 8, that increased concentrations of both BBOA and SV-OOA are linked to air masses originating from Northern and Eastern Europe. During wintertime, these flow categories are associated with the prevalence of synoptic-scale northern winds and a decline in temperature  
475 in the area, leading to the appearance of PM episodes due to local combustion for residential heating (Paschalidou et al., 2015).

HOA correlates very well with BC ( $R^2=0.53$ ), CO ( $R^2=0.54$ ) and with NO ( $R^2=0.55$ ). COA correlates moderately with  $\text{nss-K}^+$  ( $R^2=0.32$ ) and chloride ( $R^2=0.27$ ) while low correlation with BC and CO (both  $R^2=0.22$ ) has also been previously reported at the same site (Florou et al., 2017).  
480 Finally, LV-OOA showed a good correlation with ammonium ( $R^2=0.62$ ), nitrate ( $R^2=0.63$ ),  $\text{nss-K}^+$  ( $R^2=0.52$ ) and  $m/z$  73 ( $R^2=0.51$ ), demonstrating that part of the very oxidized OA during wintertime may also originate from combustion sources as well.

Therefore, during the cold period, the organic aerosol in the area linked to secondary organic aerosol formation contributes around 65% to the total organic fraction. In contrast to summer, the  
485 semi-volatile product has a very clear origin, linked to the fast oxidation of primary combustion sources (BBOA and HOA), which is also reflected on its diurnal variability as will be discussed shortly (Fig. 6). Its affinity to biomass burning tracers points out that the largest part of it originates from the fast oxidation of BBOA. The low-volatility product is in this case of more local than long-range transport nature, as also highlighted by the almost two-fold higher values during night-  
490 time.

The diurnal cycles of the five factors are shown in Figure 6 (bottom left panel). HOA, originating from fossil fuel combustion, exhibits maximum values during night, associated with





combustion from central heating, and presets a secondary peak during early morning traffic rush  
hour. COA has similar winter and summer diurnal profiles t with a moderate hump during lunch  
495 hours (12-15:00 LT) and a large night-time peak (approx. 22:00 LT), controlled by the decrease  
of the planetary boundary layer. BBOA is characterized by a pronounced diurnal cycle with  
peaking values during night-time, associated with the production of this component in the evening  
by combustion for heating purposes. SV-OOA exhibits the largest diurnal amplitude, with night-  
time values being almost 6-fold higher compared to daytime. A moderate peak during the morning  
500 traffic rush hour (partly masked by the high night values), demonstrates its provenance from the  
oxidation of freshly-emitted primary combustion organic aerosol (both BBOA and HOA). Finally,  
LV-OOA also exhibits 2-fold higher values during night compared to daytime. It has a secondary  
peak during the morning traffic rush hour, showing once more that part of the low volatility OA  
may also originate from the fast oxidation of primary combustion source, as also implied by its  
505 correlation with combustion tracers.

Table 2 sums up the contribution of each one of the 5 identified factors during the three studied  
winters. Overall, during wintertime BBOA constitutes around 10% of the total organic fraction.  
Based on the diurnal variability of this factor, its contribution is more pronounced during night-  
time, when concentrations are 4-fold or higher than the daytime ones, matching emissions from  
510 fossil fuel combustion (HOA; traffic plus heating). Taking into account that a large part of the  
semi-volatile OOA (SV-OOA) comes from the rapid oxidation of freshly emitted BBOA, justified  
via the excellent correlations with biomass burning tracers, the overall contribution of biomass  
burning becomes even more significant. Given that SV-OOA contributes around 30% to the  
organic mass, it is evident that during wintertime, biomass burning may contribute almost half of  
515 the total organic aerosol, with this contribution maximizing during night-time. More specifically,  
for BBOA the lowest contribution during daytime is 2.6% reaching a maximum of 19.4% during  
night (Figure 6, bottom right panel). The same applies to SV-OOA with daytime minimum  
contribution of 13.5% and night-time maximum of 35.7%. What is also very important is the fact  
that even though the winter and summer mass spectra of SV-OOA have some similarities  
520 ( $R^2=0.85$ ,  $\theta=21.4^\circ$ ), there are also differences, especially in the origin of this component, as during  
winter the majority is linked to the oxidation of primary combustion sources, while during summer  
the absence of a significant correlation with BC or nss- $K^+$  implies the presence of different sources,  
both anthropogenic (but not biomass burning) and possibly biogenic.



#### 525 4. Summary and conclusions

High temporal-resolution measurements were conducted for an entire year (plus two, two-month duration, intensive measurement campaigns during wintertime) at an urban background site in Athens, using an ACSM, a PILS-IC system and an aethalometer, in excess to routine pollution measurements. During the 16 month measurement period, several pollution events with PM<sub>1</sub> concentrations reaching as high as 240 µg m<sup>-3</sup> were recorded, all encountered during wintertime nights. In these cases, organics contributed the largest fraction to the submicron particulate mass, with overall contribution during wintertime reaching 50%, followed by sulfate (~20%) and BC (~14%). Within a typical winter day, organics, BC and nitrate double their concentrations during night-time, as a result of the additional primary combustion emissions for heating purposes. During summer, all concentrations are significantly lower, and organics are once more the main aerosol constituent contributing by 44%, followed by sulfate (33%), ammonium (13%) and BC (8%). Within a typical summer day, ammonium and sulfate concentrations peak at about 14:00 LT (UTC+2), consistent with secondary aerosol formation.

Organics, nitrate, chloride and BC exhibited a clear seasonal cycle with maximum during winter and minimum during summer. Sulfate and ammonium exhibited the opposite cycle, as a result of enhanced photochemistry, limited precipitation and higher regional transport.

Based on the source apportionment of the organic aerosol, four factors were identified during summer, namely hydrocarbon-like OA (HOA), cooking-like OA (COA), semi-volatile oxygenated OA (SV-OOA) and low-volatility OA (LV-OOA), and five factors during winter, the same as in summer with the addition of primary biomass burning emissions (BBOA). During summer, HOA makes up 5.7% of the total organic fraction, COA around 12%, and the rest is linked to secondary organics (SV-OOA and LV-OOA). HOA has peaking values during the morning traffic rush hour, and COA mainly during night-time. SV-OOA exhibits two-fold higher concentrations during night-time while LV-OOA exhibits a peak during mid-day, consistent with photochemical processes. The semi-volatile product is clearly of mixed origin, linked to quick atmospheric processing within a few hours, of VOCs emitted from primary sources like vegetation, traffic and to some limited extent to processed regional biomass burning. The low-volatility product, on the other hand, is the result of more excessive oxidation, in the order of several days, having thus a more regional character.



555 Combining the results from the three different winter campaigns, HOA accounts for almost  
16% of the organic fraction, COA around 8%, BBOA 10%, SV-OOA 31% and LV-OOA 35%.  
All constituents exhibit significantly higher concentrations during night-time, with HOA being  
affected by combustion from central heating and presenting a secondary peak during the morning  
traffic rush hour. COA has a similar diurnal profile to the one observed during summer. BBOA is  
560 also characterized by a pronounced diurnal cycle with peaking values during night from  
combustion for heating purposes. SV-OOA has almost 6-fold higher concentrations during night,  
consistent with its link to the oxidation of primary combustion sources, while even LV-OOA  
exhibits almost 2-fold higher concentrations during night. In contrast to summer, the semi-volatile  
product during winter has a very clear origin, linked to the fast oxidation of primary combustion  
565 sources (HOA and BBOA) with BBOA being the major source, due to the affinity of SV-OOA  
with biomass burning tracers. Part of the LV-OOA, as well, could originate from the extensive  
oxidation of the local primary combustion sources, showing that LV-OOA during winter is of more  
local than regional character.

Concluding, it is clear that organic aerosol constitutes a large fraction of submicron aerosol  
570 throughout the year, in the urban environment of Athens. During wintertime, a large part of this  
OA, as high as 50%, originates from combustion sources for heating purposes, such as central  
heating and biomass burning, causing significant air quality deterioration. Night-time contribution  
of BBOA is 7-fold higher than the one during day, while the respective contribution of SV-OOA  
is increased by a factor of 2.6. Given that during wintertime, fine PM concentrations reach up to  
575  $240 \mu\text{g m}^{-3}$ , the significance of these sources contribution becomes even more striking,  
demonstrating the necessity for strategic, long-term mitigation actions.

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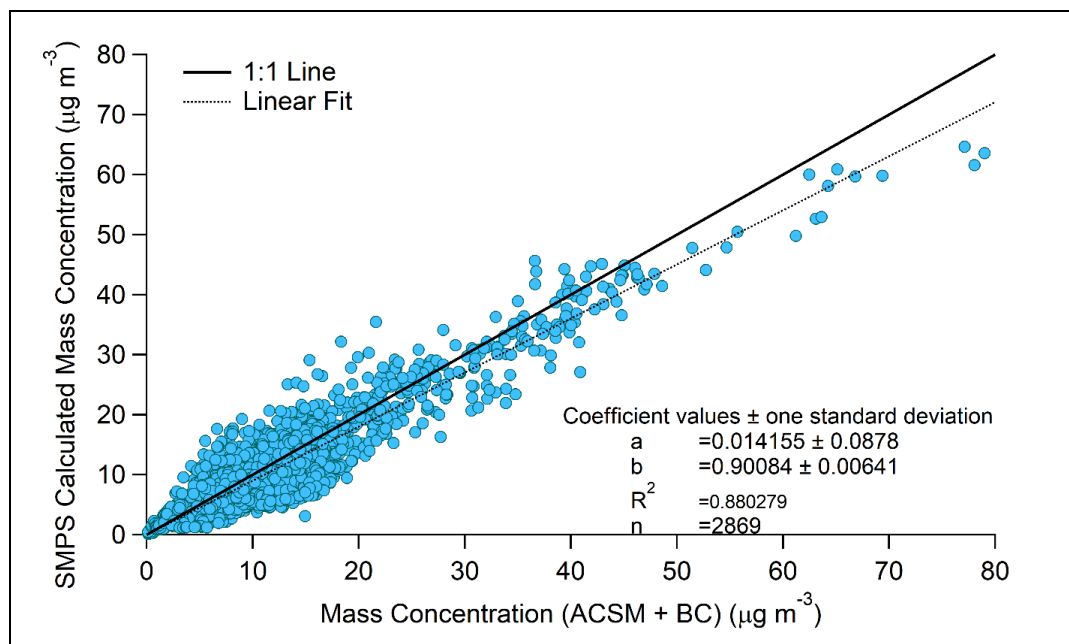


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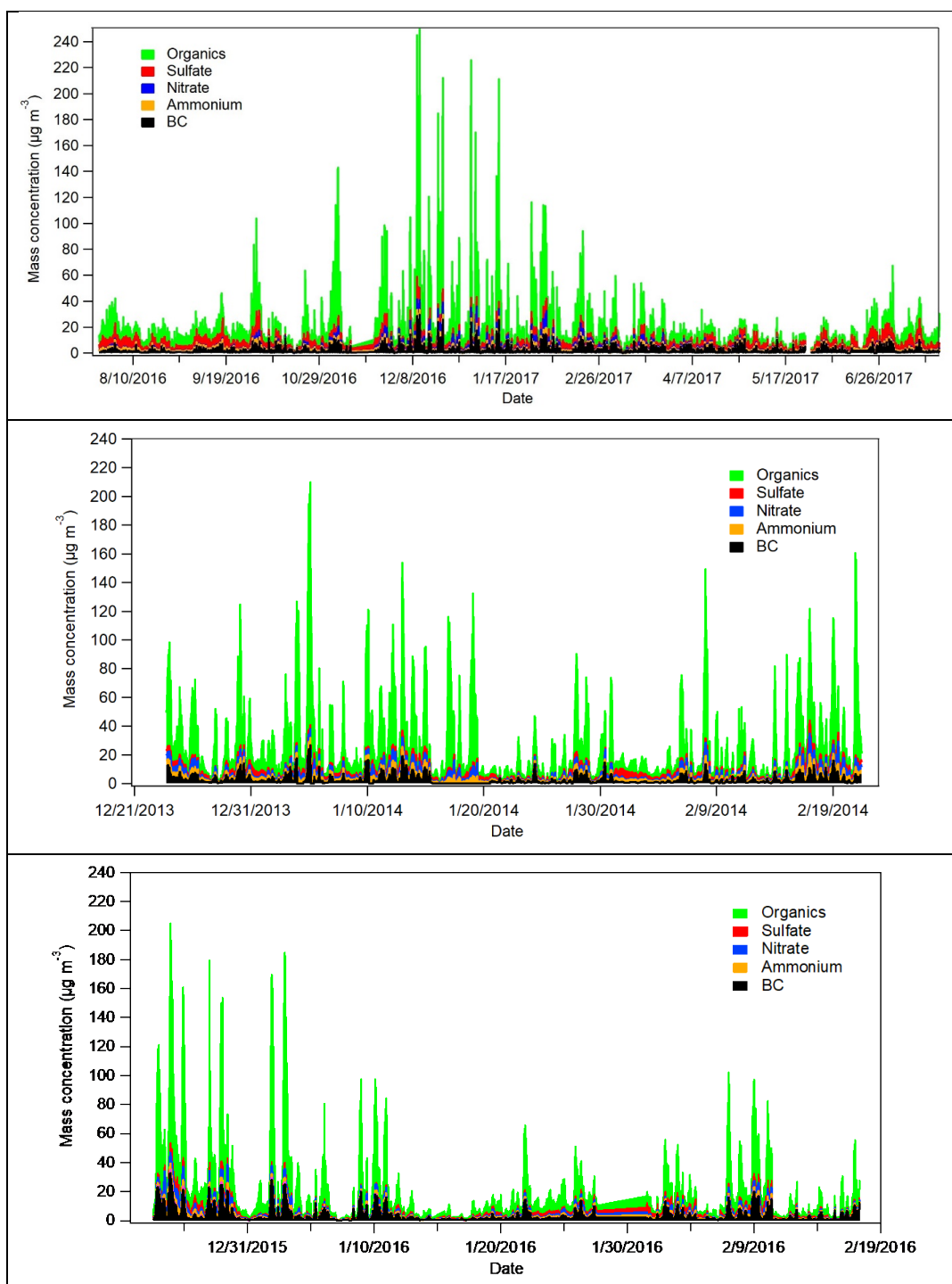
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**Figure 1:** Correlation between ACSM+BC vs. SMPS-derived mass concentrations for the 2016-17 measurement period.



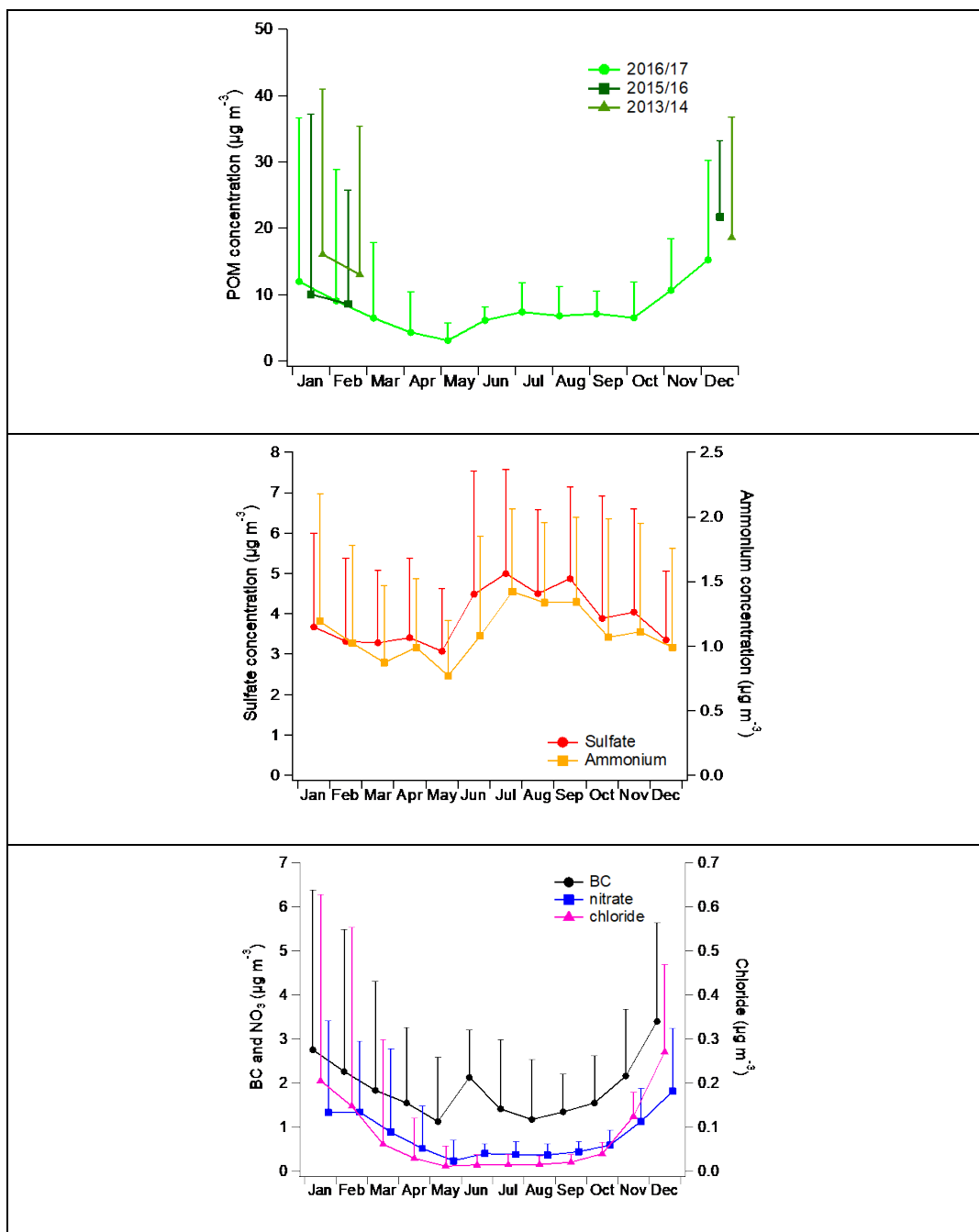
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**Figure 2:** Time series of the main submicron aerosol components. On the top panel the one-year period starting on 26 July 2016 and ending on 31 July 2017, on the middle panel the 2013-2014 winter campaign (18 December-21 February), and on the bottom panel, the 2015-2016 winter campaign (23 December-17 February).

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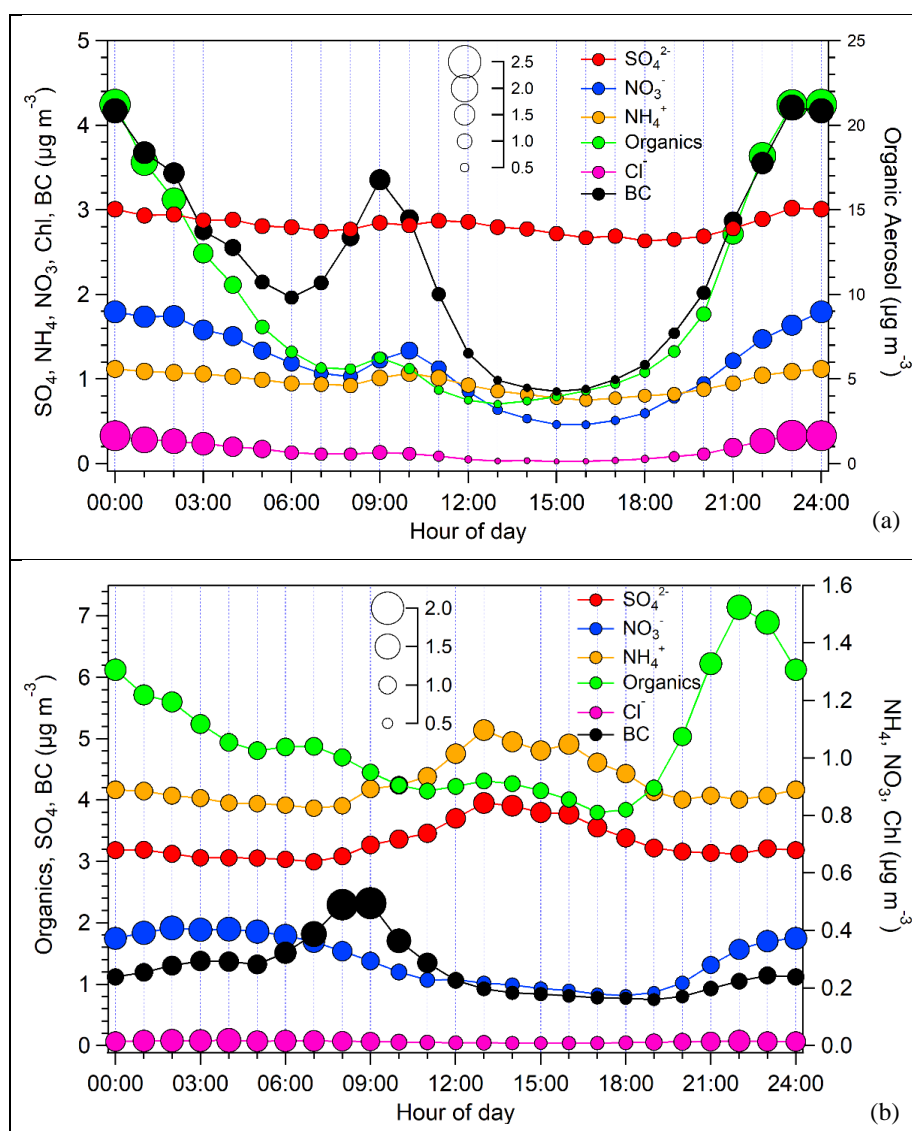


**Figure 3:** Monthly average concentrations of the main aerosol constituents. Organics are shown on the top panel for the one year 2016-2017 period as well as the 2013-2014 and 2015-2016 winter periods, while sulfate and ammonium on the middle panel, and BC, nitrate and chloride on the bottom panel

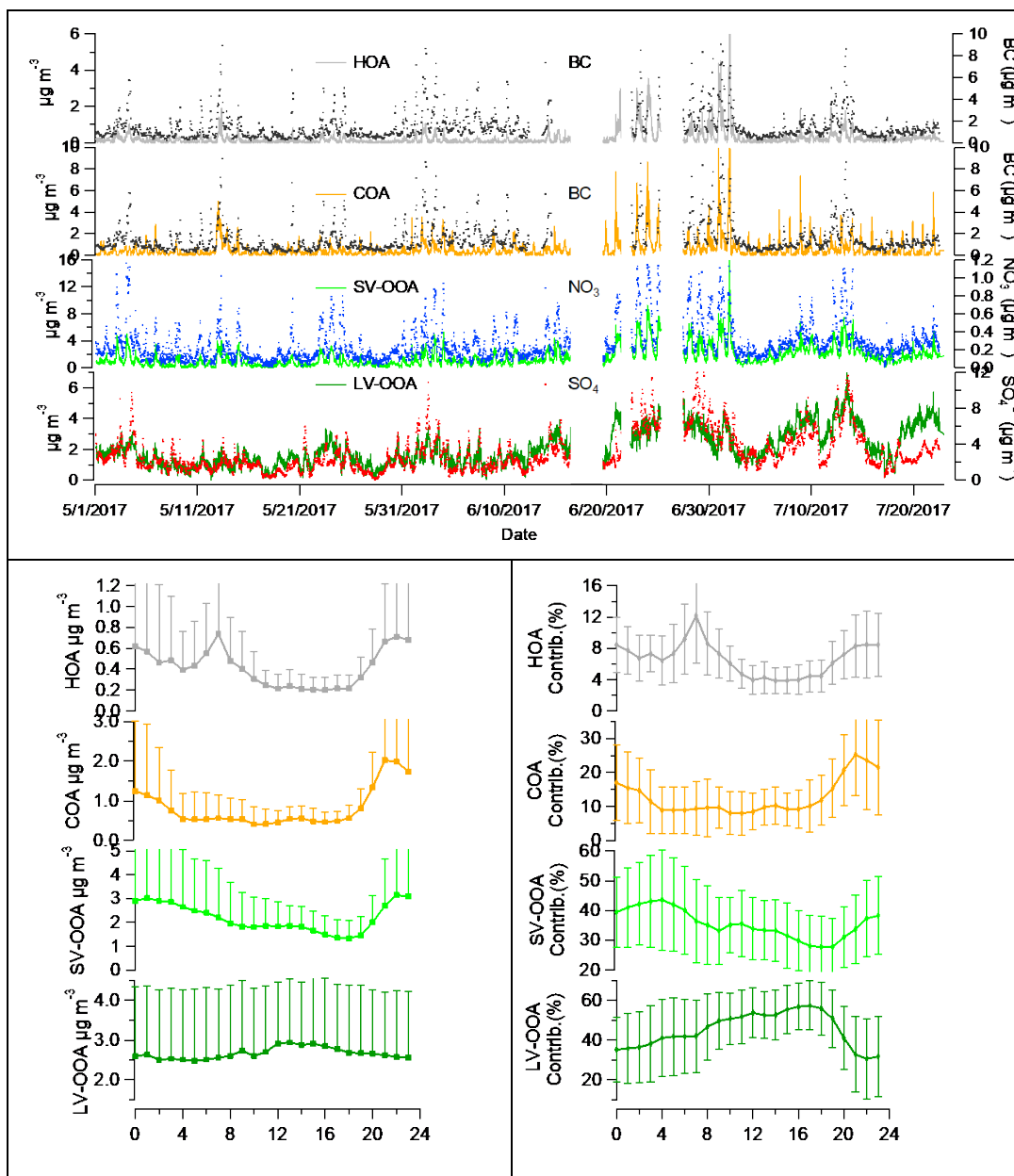




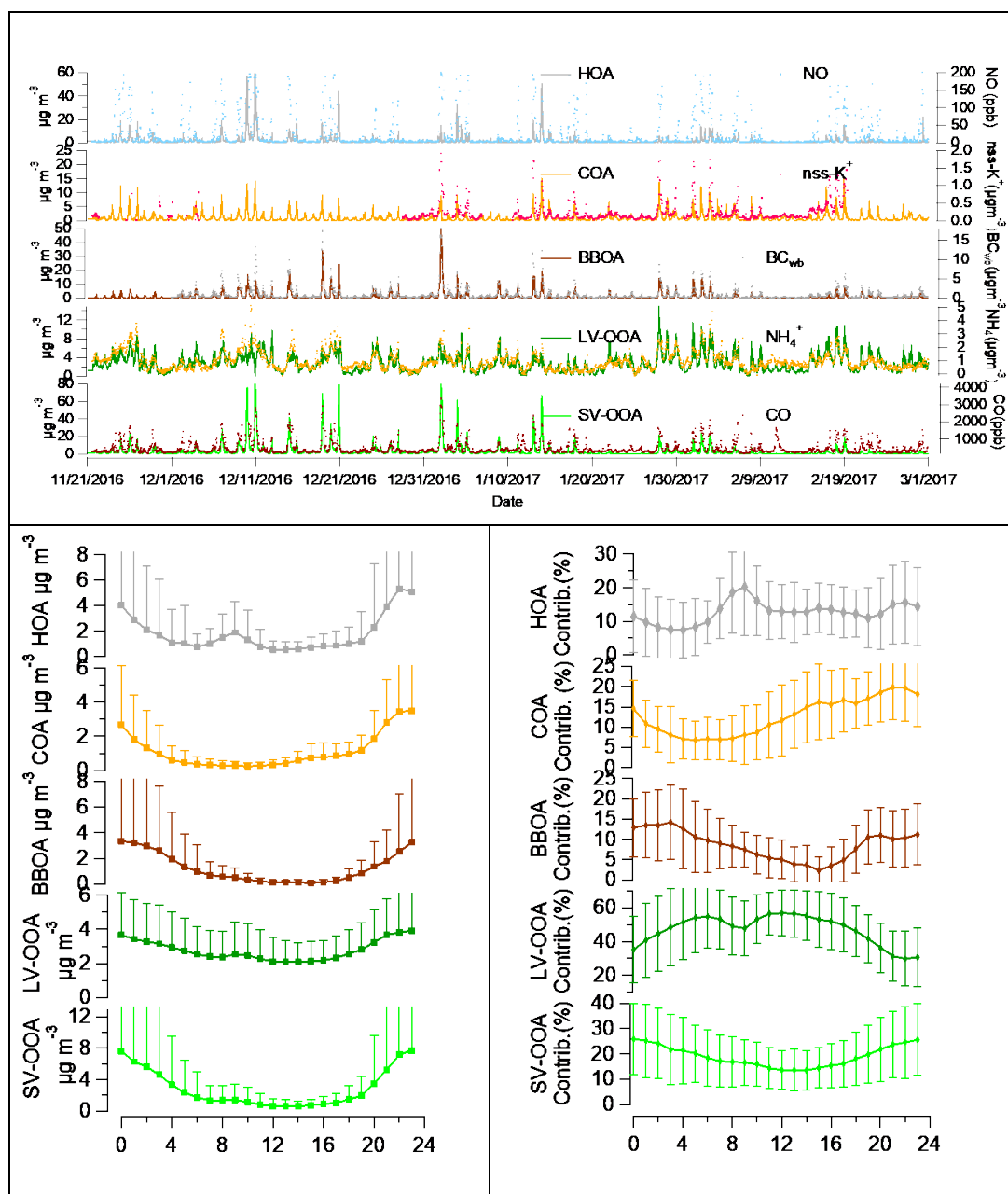
850 shown for the one year 2016-2017 period. Standard deviation is also depicted (error bars; only the positive part is shown for plot's clarity issues).



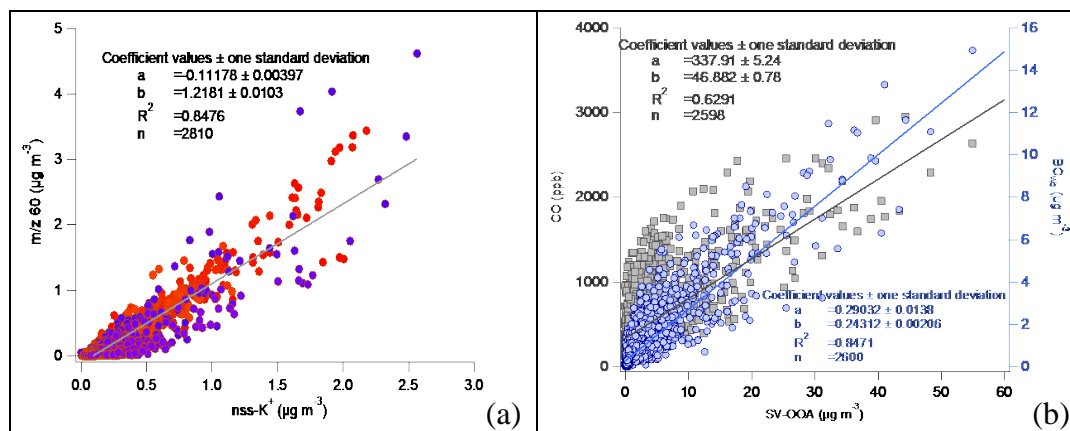
855 **Figure 4:** Average daily cycle of the main submicron aerosol constituents for the cold period 2016-17 on the top panel and the warm period of 2017 on the bottom panel. The size of the markers indicates the normalized values relative to each species' daily mean value.



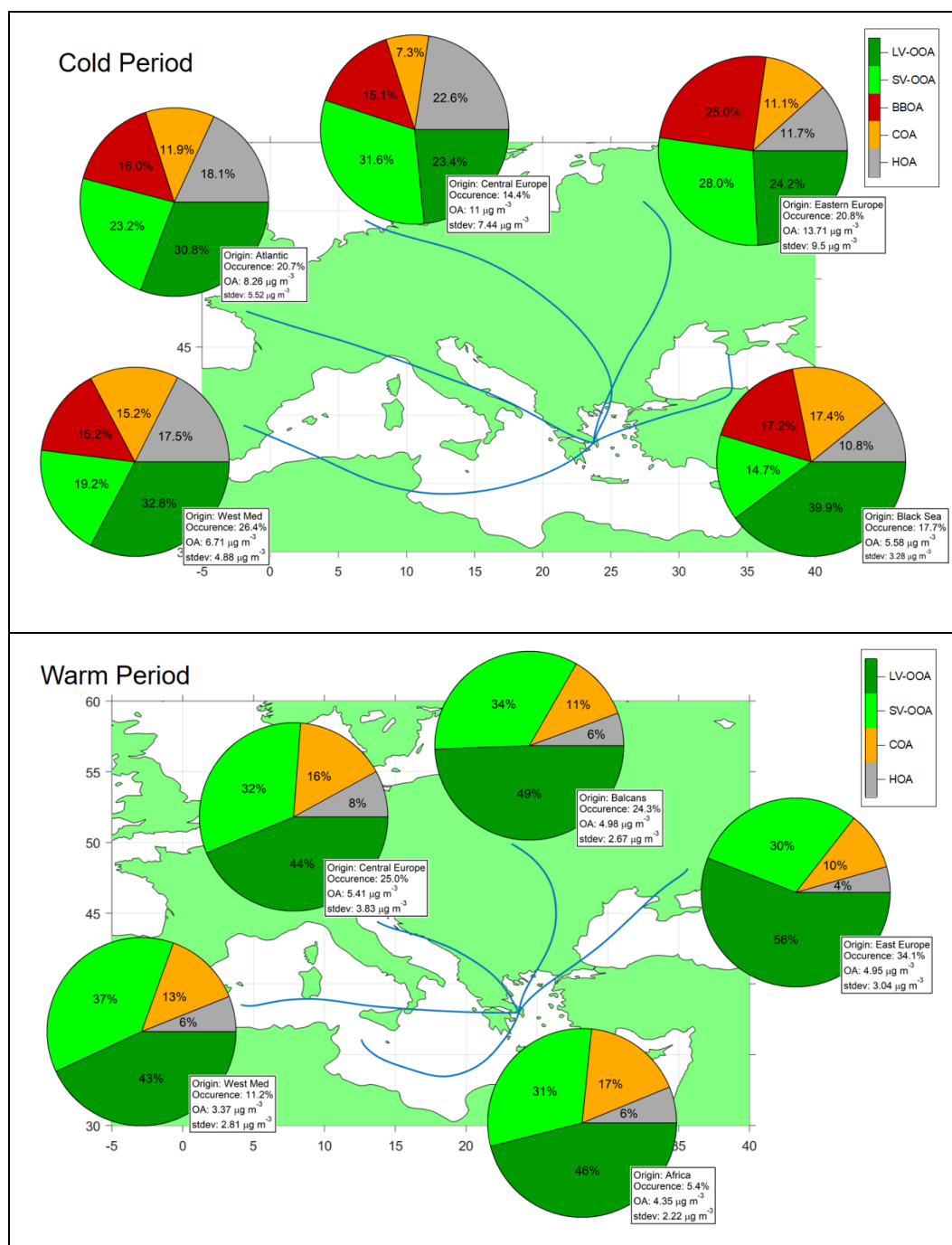
**Figure 5:** Time series of the contribution of the different factors identified by PMF between 1 May – 31 July 2017 (top) along with their average diurnal cycles (bottom left) and the respective hourly average contributions (bottom right).



**Figure 6:** Time series of the contribution of the different factors identified by PMF between 21 Nov. 2016 – 1 March 2017 (top) along with their average diurnal cycle (bottom left) and respective hourly contribution (bottom right).



865 **Figure 7:** (a) Correlation of  $m/z\ 60$  with  $nss\text{-}K^+$  for 2013-14 (red) and 2016-17 (blue), and (b) Correlation of SV-OOA with CO (grey) and BC (blue) for 2016-17.



**Figure 8:** Average compositional pie-charts of OA fractions for each trajectory cluster, separately for cold (November 2016 – March 2017) and warm (Aug-Sep 2016 and May-Jul 2017) periods on the top and



870 bottom panel respectively. Displaying also frequency of occurrence and average OA concentrations for each cluster.

	<b>MAM 2017</b>	<b>JA 2016&amp; JJ 2017</b>	<b>SON 2016</b>	<b>DJF 2016- 17</b>	<b>DJF 2013- 14</b>	<b>DJF 2015- 16</b>
Organics	3.5 ± 3.3 (0.3-35.9)	5.3 ± 3.4 (0.4-44.8)	6.3 ± 7.9 (0.3-98.6)	9.9 ± 16.1 (0.2-200.5)	15.7 ± 21.8 (0.5-189.4)	11.5 ± 18.4 (0.7-151.4)
Ammonium	0.6 ± 0.5 (0.4-2.9)	1.0 ± 0.6 (0.2-3.9)	1.1 ± 0.7 (0.4-5.1)	1.0 ± 0.7 (0.2-5.1)	1.9 ± 1.3 (0.2-9.9)	1.0 ± 0.9 (0.3-6.2)
Sulfate	2.8 ± 1.4 (0.2-9.7)	3.6 ± 2.0 (0.2-14.5)	3.5 ± 2.2 (0.2-15.4)	2.7 ± 1.7 (0.2-14.8)	2.3 ± 1.3 (0.3-12.3)	2.2 ± 1.5 (0.4-9.3)
Nitrate	0.4 ± 0.5 (0.05-4.8)	0.3 ± 0.2 (0-1.4)	0.6 ± 0.8 (0.1-6.5)	1.2 ± 1.5 (0.05-11.1)	2.1 ± 2.0 (0.09-15.4)	1.4 ± 1.8 (0.07-9.9)
Chloride	0.03 ± 0.05 (0-0.9)	0.02 ± 0.02 (0.04-0.2)	0.05 ± 0.1 (0.08-2.0)	0.17 ± 0.33 (0-3.7)	0.14 ± 0.22 (0.08-6.8)	0.11 ± 0.24 (0-2.5)
BC	1.5 ± 1.4 (0.1-14.6)	1.2 ± 0.8 (0.2-10.5)	1.7 ± 1.6 (0.1-12.4)	2.4 ± 3.4 (0.1-29.6)	2.7 ± 3.2 (0.2-26.8)	3.4 ± 4.6 (0.2-32.3)
PM1	10.2 ± 7.1 (0.6-115.6)	9.6 ± 6.2 (0.5-67.4)	15.8 ± 13.2 (0.9-143.9)	20.7 ± 26.4 (0.8-251.8)	25.1 ± 29.9 (1.4-213.1)	19.5 ± 25.3 (1.7-202.5)

**Table 1:** Seasonal average concentrations ± standard deviation (range) and total mass of the main submicron aerosol components for the one-year study period and the two winter campaigns.

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	<b>Winter 2013-14</b> 18/12/13 – 21/02/14	<b>Winter 2015-16</b> 23/12/15 – 17/02/16	<b>Cold 2016-17</b> 01/11/16-18/03/17
<b>BBOA</b>	8.3%	9%	10.3%
<b>HOA</b>	16%	12.5%	16.5%
<b>COA</b>	7.4%	7.2%	11.4%
<b>SV-OOA</b>	33.1%	27.5%	30.8%
<b>LV-OOA</b>	35.2%	43.8%	31%

**Table 2:** Contribution of the five organic aerosol components to the total organic fraction during the three individual winter campaigns.