



# **Using CALIOP to constrain blowing snow emissions of sea salt aerosols over Arctic and Antarctic sea ice**

Jiayue Huang<sup>1</sup>, Lyatt Jaeglé<sup>1</sup>, Viral Shah<sup>1</sup>

<sup>1</sup>Department of Atmospheric Sciences, University of Washington, Seattle, WA

5 *Correspondence to*: jaegle@uw.edu

**Abstract.** Sea salt aerosols (SSA) produced on sea ice surfaces by blowing snow events or lifting of frost flower crystals have been suggested as important sources of SSA during winter over polar regions. The magnitude and relative contribution of blowing snow and frost flower SSA sources, however, remain uncertain. In this study, we use 2007–2009 aerosol extinction coefficients from the Cloud-Aerosol Lidar with Orthogonal Polarization (CALIOP) instrument onboard the

- 10 Cloud-Aerosol Lidar and Infrared Pathfinder Satellite Observation (CALIPSO) satellite and the GEOS-Chem global chemical transport model to constrain sources of SSA over Arctic and Antarctic sea ice. CALIOP retrievals show elevated levels of aerosol extinctions (10–20 Mm<sup>-1</sup>) in the lower troposphere (0–2 km) over polar regions during cold months. The standard GEOS-Chem model underestimates the CALIOP aerosol extinctions by 50–70%. Adding frost flower emissions of SSA fails to explain the CALIOP observations. With blowing snow SSA emissions, the model captures the overall spatial
- 15 and seasonal variation of CALIOP aerosol extinctions over the polar regions, but overestimates springtime aerosol extinctions over Arctic sea ice and winter-spring extinctions over Antarctic sea ice. We reduce the surface snow salinity over multi-year sea ice and infer the monthly FYI snow salinity required to minimize the discrepancy between CALIOP extinctions and the GEOS-Chem simulation. The empirically-derived snow salinity shows a decreasing trend in between fall and spring. The optimized blowing snow model with inferred snow salinities generally agrees with CALIOP extinction
- 20 observations to within 10% over sea ice, but underestimates aerosol extinctions over the regions where frost flowers are expected to have a large influence. Frost flowers could thus contribute indirectly to SSA production by increasing the local surface snow salinity and, therefore, the SSA production from blowing snow. We carry out a case study of an Arctic blowing snow SSA feature predicted by GEOS-Chem and sampled by CALIOP. Using backtrajectories, we link this feature to a blowing snow event which occurred 2 days earlier over first-year sea ice and was also detected by CALIOP.

# 25 **1 Introduction**

Sea salt aerosols (SSA) are produced via wave breaking in the open ocean (Lewis and Schwartz, 2004; De Leeuw et al., 2011 and references therein). Over polar regions, SSA can also be generated via sublimation of saline blowing snow (Simpson et al., 2007; Yang et al., 2008) and wind-blown frost flower crystals (Rankin et al. 2000; Domine et al., 2004; Xu et al. 2013). SSA production from blowing snow events requires strong winds  $(27 \text{ m/s})$  and depends on the salinity of snow

30 cover on sea ice (Yang et al., 2008). Frost flowers grow over new sea ice formed from open leads under low ambient





temperature  $( $-20^{\circ}$ C)$  (Kaleschke et al., 2004). These two sea ice sources of SSA have been proposed to help explain polar observations of wintertime maxima in SSA mass concentrations (Wagenbach et al., 1998; Weller et al., 2008; Jourdain et al., 2008; Udisti et al., 2012; Huang and Jaeglé, 2017), the depletion of sulfate-to-sodium mass ratio in winter SSA relative to bulk sea water at sites in Antarctica (Wagenbach et al., 1998; Rankin et al., 2000; Jourdain et al., 2008; Hara et al., 2012)

and some sites in the Arctic (Jacobi et al., 2012; Seguin et al., 2014), as well as the increase in Na<sup>+</sup> deposition fluxes during glacial periods relative to interglacial periods (Wolff et al. 2006; Fischer et al., 2007; Abram et al., 2013).

In a previous study (Huang and Jaeglé, 2017), we used the GEOS-Chem chemical transport model to examine the relative roles of blowing snow and frost flowers as sources of polar SSA during winter. Our study was based on the blowing snow parameterization developed by Yang et al. (2008) and the frost flower parameterization of Xu et al. (2013). We compared

- 10 our simulations to in situ observations of SSA mass concentrations at three surface sites in the Arctic and two sites in coastal Antarctica, showing that blowing snow appeared to be the dominant source of polar SSA during winter. Here, we further constrain the spatial and temporal distribution of polar sources of SSA by using observations of aerosol extinction coefficients from the Cloud-Aerosol Lidar with Orthogonal Polarization (CALIOP) instrument onboard the Cloud-Aerosol Lidar and Infrared Pathfinder Satellite Observations (CALIPSO) satellite together with the GEOS-Chem model.
- 15 One of the main uncertainties in estimating blowing snow emissions is the salinity of snow on sea ice. In our previous work (Huang and Jaeglé, 2017), we assumed constant surface snow salinity over Arctic (0.1 practical salinity unit, or psu) and Antarctic (0.03 psu) sea ice. In reality, surface snow salinity is highly variable in time and space. The sources of sea salt in snow over sea ice include upward migration of brine from the sea ice surface, incorporation of frost flowers and SSA deposition from the nearby open ocean (Domine et al., 2004). Initial sea ice formation is accompanied by upward salt
- 20 transport, such that first-year sea ice (FYI) has a high surface salinity reaching 20–100 psu (Weeks and Lee, 1958; Martin, 1979; Weeks and Ackley, 1986). Nakawo and Sinha (1981) found that sea ice salinity decreases rapidly within the first week of sea ice formation in fall, and then decreases more slowly between December and May in the Canadian Arctic. As snow accumulates on FYI throughout winter, the brine is wicked upward resulting in brine-wetted snow (Barber et al., 1995). Snow salinity is highest in the first 10 cm above FYI, with values of 1–20 psu (Geldsetzer et al., 2009) and then decreases
- 25 rapidly as the snow cover gets thicker, with low salinities on the surface of thick snow (Nandan et al., 2017). Older and thicker multiyear sea ice (MYI), is desalinated by flushing and gravity drainage during repeated summer melt cycles, such that the overlaying snow has very low salinity. Krnavek et al. (2012) reported that surface snow salinity sampled on MYI was 0.01 psu, compared to 0.1 psu for snow on thick FYI and 0.8 psu on recently frozen thin FYI in March near Barrow, Alaska.
- 30 The role of frost flowers as a direct source of SSA remains subject to debate. Some studies have shown that strong winds inhibit frost flower formation and bury existing frost flowers with snow (Perovich and Richeter-Menge, 1994; Rankin et al., 2000), while field experiments show that frost flowers are difficult to break (Domine et al., 2005; Alvarez-Avilez et al.,





2008; Obbard et al. 2009). In addition, laboratory experiments show that evaporating frost flowers form a cohesive chunk of salt (Yang et al., 2017), which is unlikely to be a direct source of SSA even when exposed to large wind speeds (Roscoe et al., 2011).

In this study, we use 3 years (2007–2009) of CALIOP aerosol extinctions to constrain the spatial and temporal distribution of 5 polar SSA emissions with the GEOS-Chem model. Satellite observations and GEOS-Chem simulations are described in section 2. In section 3, we evaluate the model's ability to reproduce observed aerosol extinctions over the Arctic and Antarctic sea ice regions with and without sea ice sources of SSA. In section 4, we develop an empirical parameterization of seasonally-varying surface snow salinity on FYI. In section 5, we conduct a case study of an Arctic blowing snow event and the resulting SSA observed by CALIOP.

# 10 **2 Observations and model simulations**

#### **2.1 CALIOP observations of aerosol extinction coefficients**

The CALIOP Lidar measures backscatter signals of optical pulses at 532 and 1064 nm (Winker et al., 2009). CALIOP samples the optical properties of clouds and aerosols during daytime and nighttime with a 16-day repeat cycle. It has a sampling resolution of 335 m horizontally and 30 m vertically below 30–40 km altitude. In this study, we use vertical

- 15 profiles of 532 nm aerosol extinctions from CALIOP Level 2 (L2) version 4.10 profile data for 2007–2009 (Winker, 2016). L2 data is retrieved with a set of algorithms, which identify the cloud and aerosol layers and classify their feature types (Liu et al., 2009). The layer detection algorithm is performed downward for single shots and profiles are averaged horizontally at 1, 5, 20 and 80 km to achieve good signal-to-noise-ratio (SNR) for aerosol retrievals (Winker et al., 2009). The estimated detection sensitivity of the CALIOP 532-nm channel varies with different horizontal averaging, with better sensitivity at
- 20 larger horizontal averaging (80 km). Nighttime CALIOP extinction coefficients have better sensitivity than daytime observations, which are affected by noise from solar radiation scattering (Winker et al., 2009).

We average monthly CALIOP L2 aerosol extinction profiles poleward of 60° over a 2° latitude by 5° longitude horizontal grid in 60 m vertical bins, using the same approach as Winker et al. (2013). When no aerosols are detected and the layer is classified as "clear air", we assign it an extinction coefficient value of 0.0  $km<sup>-1</sup>$ . Following Winker et al. (2013), we exclude

25 the following aerosol layers from our gridded averages: (1) all aerosol layers within 60 m of the surface to avoid surface contamination, (2) layers with a Cloud Aerosol Discrimination (CAD) score falling outside the range of −100 to −20, (3) uncertainty flags of 99.9 km<sup>-1</sup> when the retrieval is considered failed for aerosol extinctions and the layers beneath, (4) extinction QC flags indicating possible large errors, (5) "clear air" under the lowest detected aerosol layer with base below 250 m to avoid low bias for undetected surface-attached aerosol. In addition, we exclude very high values of aerosol





extinctions ( $> 0.35$  km<sup>-1</sup>) below 2 km poleward of 60° during cold months (September–May for the Arctic and March– November for the Antarctic), as they are likely related to diamond dust misclassified as aerosol (Di Pierro et al., 2013).

The CALIOP nighttime retrievals over polar regions are limited during summer months (May–July in the Arctic, November– January in the Antarctic), with maximum latitudinal extents of 55° latitude beyond which only daytime retrievals are

- 5 available. Daytime retrievals have higher detection thresholds and can only detect aerosol layers with relatively high extinctions (Fig. S1–2). Therefore, at a given latitude fewer aerosol layers are detected in the daytime retrievals, and the average daytime extinction coefficients are lower than the nighttime ones over polar regions (Fig. S3). Most of our work is based on analysis of nighttime CALIOP retrievals during winter over polar regions. However, to reconstruct the full seasonal cycle of aerosol extinctions over polar regions, we calculate nighttime equivalent aerosol extinction profiles by combining
- 10 both daytime and nighttime CALIOP extinctions coefficients, following the empirical algorithm derived by Di Pierro et al. (2013). This approach provides an empirical correction for the differences in detection sensitivity and aerosol extinctions in the daytime CALIOP retrievals (more details are given in the Supplement).

#### **2.2 The GEOS-Chem chemical transport model**

15 We use the GEOS-Chem (v10-01) 3-D global chemical transport model (Bey et al., 2001) driven by meteorological fields from the Modern Era Retrospective-Analysis for Research and Applications (MERRA; Rienecker et al., 2011). The MERRA assimilated meteorological fields have a native horizontal resolution of 0.5° latitude by 0.666° longitude with 72 vertical levels, which we regrid to  $2^{\circ} \times 2.5^{\circ}$  horizontal resolution and 47 vertical levels with merged levels above 80 hPa.

We conduct a three-year (2007–2009) global simulation of tropospheric aerosol-oxidant chemistry. The model is initialized 20 with 1-year spin up. Global anthropogenic emissions are from EDGAR v4.2 (Emissions Database for Global Atmospheric

- Research, Olivier and Berdowski, 2001) for 1970–2008. For the years after 2008, anthropogenic emissions are scaled relative to year 2008, based on government statistics for different countries/regions (van Donkelaar et al., 2008). Over North America, the anthropogenic emissions are from the 2011 National Emissions Inventory (NEI2011; https://www.epa.gov/airemissions-inventories/2011-national-emissions-inventory-nei-data) produced by the US Environmental Protection Agency
- 25 (EPA), with annual scaling factors from the EPA for other years. Over Asia, the anthropogenic emissions are from the MIX emission inventory (Li et al., 2017). Over Europe, we use the Co-operative programme for monitoring and evaluation of long range transmission of air pollutants in Europe (EMEP) anthropogenic emissions (http://www.ceip.at/review-of-inventories/). Monthly biomass burning emissions are from the Global Fire Emissions Database version 4 (GFEDv4, van der Werf et al., 2010). Black carbon (BC) and organic carbon (OC) emissions are based on the Bond et al. (2007) monthly emission
- 30 inventory, including sources from fossil fuel and biofuel. Dust emissions are based on the dust entrainment and deposition





scheme from Zender et al. (2003). Biogenic emissions of volatile organic compounds (VOCs) are from the Model of Emissions of Gases and Aerosols from Nature version 2.1 (MEGAN 2.1, Guenther et al., 2012). The  $HO<sub>x</sub>-NO<sub>x</sub>-VOC-O<sub>3</sub>$ - $BrO<sub>x</sub>$  tropospheric chemistry chemical mechanism is described in Mao et al. (2010, 2013) with recent updates in biogenic VOC chemistry (Fisher et al., 2016; Travis et al., 2016). The gas-particle partitioning of  $SO_4^2$ -NO<sub>3</sub> -NH<sub>4</sub><sup>+</sup> aerosol is

5 computed with the ISORROPIA II thermodynamic module (Fountoukis and Nenes, 2007), as implemented by Pye et al. (2009).

Advection is based on the Lin and Rood (1996) advection algorithm and boundary layer mixing is computed using the nonlocal scheme in Lin and McElroy (2011). Dry deposition in the GEOS-Chem follows a standard resistance-in-series scheme based on Wesely (1989) as described by Wang et al. (1998). Dry deposition of SSA in the model follows the Zhang et al.

10 (2001) size-dependent scheme over land, and is calculated based on the Slinn and Slinn (1980) deposition model over ocean. The dry deposition velocity for aerosols over snow and ice is assumed to be  $0.03 \text{ cm s}^{-1}$  in the model. Wet deposition of aerosol includes convective updraft, washout and rainout from precipitation (Liu et al., 2001), as well as snow scavenging (Wang et al., 2011).

The open-ocean emissions of SSA are a function of wind speed and sea surface temperature (SST) as described in Jaeglé et 15 al. (2011). In Huang and Jaeglé (2017), we inferred that wave-breaking SSA emissions are suppressed during summer at coastal polar sites with cold waters (SST  $\leq$  5°C). As in Huang and Jaeglé (2017), we reduce SSA emissions for these cold waters.

The implementation of blowing snow and frost flower SSA emissions in GEOS-Chem is described in detail in Huang and Jaeglé (2017). Briefly, the blowing snow parameterization follows Yang et al. (2008). We assume a uniform surface snow

- 20 salinity of 0.1 psu over Arctic sea ice based on observations of surface snow salinity (Mundy et al., 2005; Krnavek et al., 2012). Over Antarctic sea ice, we assume 0.03 psu for surface snow as more frequent snowfall likely dilutes the salinity (Yang et al., 2008). We calculate a mean snow age of 3 days for the Arctic and 1.5 days for the Antarctic from MERRA. In addition, we assume that 5 SSA particles are produced per snowflake (N=5) based on a comparison against observations of submicron SSA mass concentrations at Barrow, Alaska. Frost flower SSA emissions follows the emission scheme of Xu et
- 25 al. (2013), which follows the empirical wind-dependence of Shaw et al. (2010) and the potential frost flower (PFF) coverage of Kaleschke et al. (2004). The PFF is a function of ambient air temperature, and frost flowers are formed on very new and young sea ice once the ambient air temperature is cold enough ( $\leq$  about  $-20^{\circ}$ C). We set a threshold of 10 cm for the thickness of newly formed sea ice beyond which we assume that frost flowers do not form due to inefficient brine transport through thicker sea ice.
- 30 The sea ice concentration boundary conditions in MERRA are derived from the weekly product of Reynolds et al. (2002), which is based on Special Sensor Microwave Imager (SSMI) instruments on Defense Meteorological Satellite Program





(DMSP) satellites. The weekly products are linearly interpolated in time to each model time step. For each year, we use the preceding summertime minimum sea ice extent in MERRA (September in the Arctic and February in the Antarctic) to infer the location of MYI. FYI extent is calculated by subtracting MYI extent from total sea ice extent (Fig. S4).

Our standard simulation (STD) includes tropospheric aerosol-oxidant chemistry and SSA emissions from the open ocean.

5 The STD+Snow simulation is the STD simulation to which we add SSA emissions from blowing snow as in Huang and Jaeglé (2017), with uniform surface snow salinities (0.1 psu over the Arctic and 0.03 psu over the Antarctic). The STD+FF simulation is the STD simulation with SSA emissions from frost flowers. In Section 4, we develop an optimized blowing snow simulation (STD+Opt. Snow), with seasonally varying snow salinity on FYI.

The GEOS-Chem simulations are sampled at the time and location of the CALIOP overpasses and averaged over the same 10 horizontal and vertical grid. For comparison to CALIOP observations, we apply the CALIOP nighttime detection threshold

to the model, setting the modeled backscatter coefficients to  $0 \text{ Mm}^{-1} \text{ sr}^{-1}$  for backscatter values lower than 0.2  $\text{Mm}^{-1} \text{ sr}^{-1}$ .

## **3 Model evaluation with CALIOP observations**

#### **3.1 Arctic**

The Arctic cold season (November–April) CALIOP extinction coefficients in the lower troposphere (0–2 km above sea 15 level, asl) display values of 10-30 Mm<sup>-1</sup> (Fig. 1a). The largest extinction coefficients occur over the open-ocean regions of the Greenland and Barents Seas. In addition, significant aerosol extinction coefficients  $(10-20 \text{ Mm}^{-1})$  are seen over the sea ice covered Chukchi Sea, East Siberian Ocean, Laptev Sea, Kara Sea and Canadian Arctic Archipelago (Fig. 1a). While the STD GEOS-Chem simulation reproduces the pattern of extinctions over the open ocean regions, it fails to capture the enhancements over sea ice, underestimating aerosol extinction coefficients by  $\sim$ 10 Mm<sup>-1</sup> over the central Arctic (Fig. 1b).

- 20 The normalized mean bias (NMB=100×( $\overline{Model}/\overline{Obs}$  1), with  $\overline{Model}$  and  $\overline{Obs}$  representing mean observed and modeled values) is –55% over FYI, and –68% over MYI (Fig. 2a). GEOS-Chem also underestimates CALIOP aerosol extinctions over Northern Russia, which could be due to missing sources of aerosols and their precursors from gas flaring in the region (Li et al., 2016; Klimont et al., 2017; Xu et al., 2017).
- 25 In the STD simulation, aerosol extinction coefficients during the cold season are dominated by SSA over the high latitude open ocean and by long-range transport of sulfate aerosols over sea ice (Fig. 1f and g). Adding blowing snow emissions of SSA increases the simulated aerosol extinctions by  $\sim$ 10 Mm<sup>-1</sup> over the central Arctic, bringing the STD+Snow simulation in better agreement with CALIOP observations (Fig. 1c and h) and reducing the bias to +16–18% (Fig. 2b). Inclusion of frost flower emissions of SSA in the STD+FF simulation has the largest influence over Canadian Arctic Archipelago where cold





temperature and open leads co-exist (Fig. 1j), but the overall magnitude of the increase cannot explain the CALIOP extinctions.

- Figure 3 compares the vertical and seasonal distribution of aerosol extinction coefficients in the lower troposphere (0–2 km 5 altitude) over FYI, MYI, and the Canadian Arctic Archipelago, where frost flowers are expected to have their largest influence (Huang and Jaeglé, 2017). Over all three regions, the STD simulation underestimates cold season CALIOP extinctions by factors of 3–6 (Fig. 3d and f). The STD+FF simulation reduces the negative model bias, but the modeled extinctions remain too low by 20–40% (Fig. 3d–f). Furthermore, the STD+FF simulation does not capture the rapid increase in CALIOP extinctions in October–December (Fig. 3g–i). We find that applying a single scaling factor to the frost flower
- 10 emissions cannot address the seasonally-varying model discrepancy. In comparison, the STD+Snow simulation displays the best agreement with the CALIOP observations, reproducing both the vertical profile and seasonal cycle of CALIOP extinctions. The STD+Snow simulation, however, overestimates the CALIOP cold season aerosol extinctions by up to 8 Mm<sup>-1</sup> below 1 km altitude (Fig. 3d and e). In addition, it predicts a maximum in aerosol extinction during April, while CALIOP observations display their largest concentrations in January–March for FYI, and in March over MYI.

15

#### **3.2 Antarctic**

During the Austral cold season (May–October), CALIOP aerosol extinction coefficients decrease with increasing latitudes, ranging from 20–30 Mm<sup>-1</sup> at 60°S to 5–10 Mm<sup>-1</sup> near coastal Antarctica (Fig. 4a). Over Antarctic FYI (Fig. 5a), CALIOP aerosol extinctions display values of  $10-14$  Mm<sup>-1</sup> in July–October in the lower troposphere (Fig. 5g), with aerosol

20 extinctions attaining 30 Mm<sup>-1</sup> near the surface (Fig. 5d). Over MYI sea ice offshore of the Ronne ice-shelf, the CALIOP extinctions are somewhat smaller, reaching 20 Mm<sup>-1</sup> near the surface (Fig. 5e).

Poleward of 70°S, open ocean SSA dominate aerosol extinction coefficients in the STD simulation (Fig. 4f), accounting for 80% of the extinction, with the remaining 20% due to the combined contributions from sulfate, black carbon and organic aerosols. The STD simulation underestimates CALIOP observations by 5–10 Mm<sup>-1</sup> (Fig. 5g–i), with a –53% bias over FYI

25 and a –64% bias over MYI (Fig. 2d). The inclusion of frost flowers in the STD+FF simulation leads to a  $\sim$ 2 Mm<sup>-1</sup> increase in extinction coefficients near the Ross and Ronne Ice-shelves where cold temperatures and open leads persist (Fig. 4i). This increase is insufficient to explain CALIOP observations (Fig. 4e and 5).

The STD+Snow simulation increases aerosol extinction coefficients by  $10-20$  Mm<sup>-1</sup> in the Indian Ocean (0–100°) and Pacific Ocean (180–270°) sectors (Fig. 4g), where strong winds persist. The inclusion of blowing snow SSA emissions results in a

30 53% overestimate of CALIOP extinctions over FYI sea ice (Fig. 2e) and too strong a seasonal increase in extinctions





between May and October (Fig. 5g). Over the smaller MYI region, the positive bias of the STD+Snow simulation is +39% (Fig. 2e, 5e and h).

#### **4 Blowing snow simulation with optimized snow salinity**

While the inclusion of blowing snow leads to improved agreement with CALIOP, we hypothesize that the remaining 5 discrepancies in the magnitude and seasonal cycle of aerosol extinction coefficients are due to our simplifying assumption of a uniform surface snow salinity over sea ice. As discussed in Section 1, the surface snow salinity is highest over thin FYI with little snow cover early in the cold season, declining in the ensuing months as a result of thickening sea ice and increasing snowpack cover (Barber et al., 1995; Krnavek et al., 2012; Weeks and Lee, 1958; Weeks and Ackley, 1986; Nakawo and Sinha, 1981). In addition, snow salinity is very low over MYI as a result of repeated summer melt drainage of

10 brine from the sea ice (e.g., Martin, 1979).

As no systematic observations of surface snow salinity are available over sea ice, our approach is to find the monthly salinity of snow on FYI required to minimize the discrepancy between CALIOP extinctions and the GEOS-Chem simulation. For Arctic MYI, we assume a low salinity of 0.01 psu based on observations of Krnavek et al. (2012). Over Antarctic MYI we assume a salinity of 0.003 psu, assuming that frequent snowfall further dilutes the snow salinity.

15 By using a linear regression between the GEOS-Chem simulation and CALIOP monthly extinctions over Arctic sea ice, we derive a surface snow salinity on FYI of 0.9 psu in September, decreasing to 0.09 psu in April (with values of 0.36, 0.26, 0.19, 0.16, 0.16, and 0.14, between October and March). The inferred snow salinities decrease with time and are generally consistent with observations near Alaska reported by Krnavek et al. (2012): 0.8 psu for snow over recently frozen thin FYI and 0.1 psu over thick FYI. For Antarctic FYI, we infer snow salinities of 0.05 psu in April, 0.02 in May–June and 0.018 in 20 July–September. For the rest of the year the salinity is 0.015 psu.

We use these empirically-derived monthly snow salinities to conduct an optimized blowing snow simulation (STD+Opt. Snow). Over Arctic FYI, the model bias decreases from +16% (STD+Snow) to +8% (STD+Opt. Snow), and for MYI the positive bias of +18% decreases to –2% (Fig. 2b and c). Over Antarctic FYI, the model overestimate decreases from +53% (STD+Snow) to +4% (STD+Opt. Snow). Similarly, the model bias over MYI decreases to –10% (Fig. 2e and f). The

25 STD+Opt. Snow simulation displays cold season extinction profiles that are within 5–10% of CALIOP observations over sea ice (Fig. 3d–e and 5d–e). The seasonal cycles are in better agreement with observations (Fig. 3g–h and 5g–h). The STD+Opt. Snow simulation shows improved agreement with the CALIOP observations over the Beaufort Sea, due to a reduced contribution from blowing snow emissions for MYI (Fig. 1d).





The low snow salinity over MYI, however, leads to an underestimate of aerosol extinctions over the Canadian Arctic Archipelago (Fig. 3f). Combining the STD+Opt. Snow and frost flower emissions could help improve the agreement in that region, but it would also lead to substantial overestimates over FYI and MYI for the rest of the Arctic. One possibility is that snow-covered frost flowers in the Canadian Archipelago increase the local surface snow salinity (Domine et al., 2004). The

- 5 recent study of Hara et al. (2017) in northwestern Greenland proposed that snowfall buries frost flowers and the associated slush layer on new FYI. The resulting brine migrates vertically enriching the surface snow layer, which can be mobilized under strong winds. We estimate that increasing the salinity of snow over the Canadian Archipelago to a value of 3 psu would help reconcile the optimized blowing snow simulation with CALIOP observations. Direct measurements of snow salinity in this region would help confirm this estimate. Similarly, we find that increasing the snow salinity near the Ross ice-
- 10 shelf region, where frost flowers are expected to occur, would improve the agreement with CALIOP observations.

Figure 6 evaluates the performance of the STD+Opt. Snow simulation against independent observations of SSA mass concentrations at Barrow, Alaska (71.3ºN, 156.6ºW); Alert, Nunavut, Canada (82.5ºN, 62.5ºW); Zeppelin, Svalbard (78.9ºN, 11.9ºE); Neumayer, Antarctica (70.7°S, 8.3°W) and Dumont d'Urville, Antarctica (66.7ºS, 140ºE). Descriptions of the in situ observations are provided in Huang and Jaeglé (2017). At Barrow, the optimized simulation improves the agreement with

15 observed SSA mass concentrations in November–May. In particular, the enhanced salinity in October–November brings the model closer to the observations. At Zeppelin and Dumont d'Urville, the model bias in the STD+Opt. Snow simulation is reduced relative to the STD+Snow simulation. However, the model bias worsens at Neumayer (STD+Snow: –11%, STD+Opt. Snow: –48%) and Alert (STD+Snow: +24%, STD+Opt. Snow: –32%). As Neumayer and Alert are close to the frost flower producing regions, this underestimate may be related to an underestimate in the snow salinity in those regions.

#### 20 **5 A case study of a blowing snow event over the Arctic**

Figure 7 shows a case study of a blowing snow SSA event, which occurred on November 6, 2008 over the Arctic. The STD+Snow and STD+Opt. Snow simulations show enhanced extinction coefficients (40–80 Mm<sup>-1</sup>) over the Barents Sea along the 60ºE longitude line extending from the North Pole to 70ºN (Fig. 7b and c). This feature is due to blowing snow SSA and it is not seen in the STD simulation (Fig. 7a) or the frost flower simulation (not shown).

- 25 The CALIPSO 00:58–01:12 UTC overpass on November 6, 2008 transected this feature, with CALIOP aerosol extinction coefficients of 50–150 Mm<sup>-1</sup> between points A (78°N; 52.5°E) and B (82° N; 110° E) labelled in Fig. 7a–c. The cross section along the CALIOP overpass shows that the large extinctions are confined between the surface and  $1-2$  km altitude (Fig. 7e). This overpass region is mostly covered by FYI (Fig. 7d). We sampled the GEOS-Chem simulations along the same cross section, finding a very good correspondence between CALIOP and the STD+Snow and STD+Opt. Snow simulations (Fig.
- $30 \text{ g}$ –i). The optimized blowing snow model simulates slightly higher aerosol extinctions due to larger surface snow salinity on FYI in November.





We use the FLEXPART particle dispersion model (Stohl et al., 1998, 2005; Seibert and Frank, 2004) with meteorological data from ERA-Interim with a horizontal resolution of 0.5° to track the origin of this feature. We release 100,000 particles between points A and B at 0.01–2km and track them back in time over 2 days. Figure 8a shows that air along the AB transect originated from the 120–140°E sector at 60°–90°N near the surface (0–100 m), on November 4, 2008. This region is

- 5 covered by both MYI and FYI (Fig. 7d). The 01:11–01:24 UTC CALIOP overpass across this area on that day displays elevated 532 nm attenuated perpendicular backscatter and depolarization ratio below 200 m (Fig. 8c and d), which are colocated with strong surface winds ( $>5$  m s<sup>-1</sup>). The pattern of elevated backscatter, surface winds and depolarization ratio satisfies the requirements of the CALIOP blowing snow detection algorithm described in Palm et al. (2011, 2017), who defined blowing snow events as layers with high color ratio  $(>1)$ , high depolarization ratio  $(>0.25)$ , strong surface winds  $(>4)$
- 10 m s<sup>-1</sup>), and enhanced backscatter signals below 300 m (ranging between  $2.5\times10^{-2}$  km<sup>-1</sup> sr<sup>-1</sup> and 0.2 km<sup>-1</sup> sr<sup>-1</sup>). The FLEXPART-predicted source region and CALIOP blowing snow feature are co-located with enhanced blowing snow emissions in the STD+Snow simulation (Fig. 8b). This case study thus shows that CALIOP can detect not only the blowing snow event (Palm et al., 2011) but also the resulting SSA produced after sublimation.

## **6 Discussion and conclusions**

- 15 In this work, we used the GEOS-Chem chemical transport model to assess the ability of the CALIOP Lidar onboard the CALIPSO satellite to provide constraints on sea ice sources of SSA. We find that mean CALIOP aerosol extinction coefficients below 2 km altitude reach  $10-15$  Mm<sup>-1</sup> over sea ice covered regions during the 6-month polar cold season. The enhanced extinctions are located below 2 km, with the largest values  $(20-35 \text{ Mm}^{-1})$  occurring near the surface. We find that a standard GEOS-Chem simulation without sea ice sources of SSA underestimates CALIOP extinctions by 50-70% over
- 20 Arctic and Antarctic sea ice. A simulation with frost flower SSA emissions is unable to explain the spatial and temporal distribution of CALIOP aerosol extinctions. Adding a blowing snow SSA source results in improved agreement over the Arctic (NMB = +16% for first year sea ice and +18% over multi-year sea ice), but yields a 50% overestimate of CALIOP extinctions over Antarctic sea ice. Additionally, the simulation including blowing snow SSA tends to overestimate CALIOP observations towards the end of the cold season (April–May over the Arctic and September–October over the Antarctic).
- 25 We hypothesize that our assumption of constant surface snow salinity (0.1 psu over the Arctic and 0.03 psu over the Antarctic) in the blowing snow simulation could explain this overestimate. Given the paucity of snow salinity observations, we infer the monthly FYI snow salinity required to minimize the discrepancy between CALIOP extinctions and the GEOS-Chem simulation. The resulting snow salinities decrease progressively between the beginning and end of the cold season (from 0.9 to 0.09 psu for Arctic FYI; 0.05 to 0.018 psu over Antarctic FYI). This decrease is consistent with the seasonally
- 30 increasing sea ice thickness and accumulating snow depth. We also prescribe a 0.01 psu snow salinity on Arctic MYI and 0.003 psu snow salinity on Antarctic MYI. The optimized blowing snow model using the monthly-varying snow salinities





shows improved agreement with CALIOP observations and in situ observations of SSA mass concentrations at five surface sites. However, the optimized blowing snow model tends to underestimate the aerosol extinctions over the Canadian Arctic Archipelago and off the Ross Ice-shelf. Both regions are predicted to favor frost flower growth, which could locally increase the salinity of snow when frost flowers are buried under snow. We find that increasing the Canadian Arctic Archipelago FYI

5 snow salinity to 3 psu would help reconcile our simulation with CALIOP and in situ observations. Systematic observations of surface snow salinity over multiple sea ice locations and times would help further constrain snow salinities in the Arctic and Antarctic.

We conduct a case study of a blowing snow SSA event over the Arctic which was detected by CALIOP on November 6, 2008 and predicted by our blowing snow simulation. Using FLEXPART, we find that the observed aerosol extinction layer

10 originated 2 days earlier over sea ice below 100m altitude. We demonstrate that CALIOP detects this blowing snow event with enhanced extinction below 200 m and large depolarization ratio, co-located with surface high winds,

Our work suggests that blowing snow emissions are the dominant source of SSA over sea ice covered regions during cold months. As SSA can act as a source of halogens, inclusion of blowing snow in chemical transport models is important to understand springtime bromine explosions and the resulting ozone and mercury depletion events (Schroeder et al., 1998;

- 15 Simpson et al., 2007; Steffen et al., 2008; Gilman et al., 2010; Yang et al., 2010). Furthermore, these sea-ice sources of SSA can act as ice nuclei for cloud formation, and may increase the downward longwave radiative forcing (Xu et al., 2013). Arctic sea ice has been rapidly changing over the past 30 years, with decreasing sea ice extent and thickness (e.g. Kwok and Rothrock, 2009), a shift towards less MYI and more FYI (Fowler et al., 2004; Maslanik et al., 2007), and a thinning of snow depth in spring (Renner et al., 2014; Blanchard-Wrigglesworth et al., 2015). All these factors are likely to have resulted in an
- 20 increase in snow salinity on sea ice, and hence increasing SSA emissions from blowing snow. In the Southern Hemisphere, sea ice extent is FYI-dominant, and it annual/seasonal trend is more complex, varying in space. Sea ice extent has been increasing over the Ross Sea, but decreasing over the Amundsen-Bellingshausen Sea over the past decades (Turner et al., 2009; Parkinson and Cavalieri, 2012; Stammerjohn et al., 2012). Consequently, this may have resulted in a shift in the spatial pattern of blowing snow SSA emissions, with increased influence over the Ross Sea and reduced influence over the
- 25 Amundsen-Bellingshausen Sea.

#### **Acknowledgements.**

This work was supported by funding from the NASA Atmospheric Composition Modeling and Analysis Program under award NNX15AE32G. The CALIOP data was obtained from the NASA Langley Research Center Atmospheric Science Data Center.

30





## **References**

Abram, N. J., Wolff, E. W., and Curran, M. A. J.: A review of sea ice proxy information from polar ice cores, Quaternary Sci. Rev., 79, 168–183, doi:10.1016/j.quascirev.2013.01.011, 2013.

Alvarez-Aviles, L., Simpson, W. R., Douglas, T. A., Sturm, M., Perovich, D., and Domine, F.: Frost flower chemical

5 composition during growth and its implications for aerosol production and bromine activation, J. Geophys. Res., 113, D21304, doi:10.1029/2008JD010277, 2008.

Barber, D. G., Reddan, S. P., and Ledrew, E. F.: Statistical Characterization of the Geophysical and Electrical-Properties of Snow on Landfast First-Year Sea-Ice, J. Geophys. Res.-Oceans, 100, 2673–2686, doi:10.1029/94jc02200, 1995.

Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q., Liu, H. Y., Mickley, L. J., and Schultz,

10 M. G.: Global modeling of tropospheric chemistry with assimilated meteorology: Model description and evaluation, J. Geophys. Res.-Atmos., 106, 23073–23095, doi:10.1029/2001jd000807, 2001.

Blanchard-Wrigglesworth, E., Farrell, S.L., Newman, T. and Bitz, C.M.: Snow cover on Arctic sea ice in observations and an Earth System Model. Geo. Res. Lett., 42(23), 10,342–10,348, doi:10.1002/2015GL066049, 2015.

Bond, T. C., Bhardwaj, E., Dong, R., Jogani, R., Jung, S. K., Roden, C., Streets, D. G., and Trautmann, N. M.: Historical

15 emissions of black and organic carbon aerosol from energy-related combustion, 1850–2000, Global Biogeochem. Cy., 21, GB2018, doi:10.1029/2006gb002840, 2007.

de Leeuw, G., Andreas, E. L., Anguelova, M. D., Fairall, C. W., Lewis, E. R., O'Dowd, C., Schulz, M., and Schwartz, S. E.: Production flux of sea-spray aerosol, Rev. Geophys., 49, RG2001, doi:10.1029/2010RG000349, 2011.

Di Pierro, M., Jaeglé, L., Eloranta, E. W., and Sharma, S.: Spatial and seasonal distribution of Arctic aerosols observed by 20 the CALIOP satellite instrument (2006–2012), Atmos. Chem. Phys., 13, 7075-7095, doi:10.5194/acp-13-7075-2013, 2013.

Domine, F., Sparapani, R., Ianniello, A., and Beine, H. J.: The origin of sea salt in snow on Arctic sea ice and in coastal regions, Atmos. Chem. Phys., 4, 2259-2271, doi:10.5194/acp-4-2259-2004, 2004.

Domine, F., Taillandier, A. S., Simpson, W. R., and Severin, K.: Specific surface area, density and microstructure of frost flowers, Geophys. Res. Lett., 32, L13502, doi:10.1029/2005GL023245, 2005.

25 Fischer, H., Siggaard-Andersen, M.L., Ruth, U., Röthlisberger, R., and Wolff, E.: Glacial/interglacial changes in mineral dust and sea-salt records in polar ice cores: Sources, transport, and deposition, Rev. Geophys., 45, RG1002, doi:10.1029/2005rg000192, 2007.

Fisher, J. A., Jacob, D. J., Travis, K. R., Kim, P. S., Marais, E. A., Chan Miller, C., Yu, K., Zhu, L., Yantosca, R. M., Sulprizio, M. P., Mao, J., Wennberg, P. O., Crounse, J. D., Teng, A. P., Nguyen, T. B., St. Clair, J. M., Cohen, R. C., Romer, P.,

30 Nault, B. A., Wooldridge, P. J., Jimenez, J. L., Campuzano-Jost, P., Day, D. A., Hu, W., Shepson, P. B., Xiong, F., Blake, D.





R., Goldstein, A. H., Misztal, P. K., Hanisco, T. F., Wolfe, G. M., Ryerson, T. B., Wisthaler, A., and Mikoviny, T.: Organic nitrate chemistry and its implications for nitrogen budgets in an isoprene- and monoterpene-rich atmosphere: constraints from aircraft (SEAC4RS) and ground-based (SOAS) observations in the Southeast US, Atmos. Chem. Phys., 16, 5969–5991, doi:10.5194/acp-16-5969-2016, 2016.

5 Fountoukis, C. and Nenes, A.: ISORROPIA II: a computationally efficient thermodynamic equilibrium model for  $K^+$ -Ca<sup>2+</sup>- $Mg^{2+}$ –NH<sub>4</sub><sup>+</sup>–Na<sup>+</sup>–SO<sub>4</sub><sup>2–</sup>–NO<sub>3</sub><sup>–</sup>–Cl<sup>–</sup>–H<sub>2</sub>O aerosols, Atmos. Chem. Phys., 7, 4639-4659, doi:10.5194/acp-7-4639-2007, 2007.

Fowler, C., Emery, W. J., Maslanik, J.: Satellite-derived evolution of Arctic sea ice age: October 1978 to March 2003. IEEE Geosci Remote Sens Soc Lett 1(2):71–74, doi:10.1109/LGRS.2004.824741, 2004.

Guenther, A. B., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L. K., and Wang, X.: The Model of 10 Emissions of Gases and Aerosols from Nature version 2.1 (MEGAN2.1): an extended and updated framework for modeling

biogenic emissions, Geosci. Model Dev., 5, 1471-1492, doi:10.5194/gmd-5-1471-2012, 2012.

Geldsetzer, T., Langlois, A., and Yackel, J.: Dielectric properties of brine-wetted snow on first-year sea ice, Cold Re.g. Sci. Technol., 58, 47–56, doi:10.1016/j.coldregions.2009.03.009, 2009.

Gilman, J. B., Burkhart, J. F., Lerner, B. M., Williams, E. J., Kuster, W. C., Goldan, P. D., Murphy, P. C., Warneke, C.,

15 Fowler, C., Montzka, S. A., Miller, B. R., Miller, L., Oltmans, S. J., Ryerson, T. B., Cooper, O. R., Stohl, A., and de Gouw, J. A.: Ozone variability and halogen oxidation within the Arctic and sub-Arctic springtime boundary layer, Atmos. Chem. Phys., 10, 10223-10236, doi:10.5194/acp-10-10223-2010, 2010.

Hara, K., Osada, K., Yabuki, M., and Yamanouchi, T.: Seasonal variation of fractionated sea-salt particles on the Antarctic coast, Geophys. Res. Lett., 39, L18801, doi:10.1029/2012GL052761, 2012.

20 Hara, K., Matoba, S., Hirabayashi, M., and Yamasaki, T.: Frost flowers and sea-salt aerosols over seasonal sea-ice areas in northwestern Greenland during winter–spring, Atmos. Chem. Phys., 17, 8577-8598, doi:10.5194/acp-17-8577-2017, 2017.

Huang, J. and Jaeglé, L.: Wintertime enhancements of sea salt aerosol in polar regions consistent with a sea ice source from blowing snow, Atmos. Chem. Phys., 17, 3699-3712, doi:10.5194/acp-17-3699-2017, 2017.

Jacobi, H. W., Voisin, D., Jaffrezo, J. L., Cozic, J., and Douglas, T. A.: Chemical composition of the snowpack during the 30 25 OASIS spring campaign 2009 at Barrow, Alaska, J. Geophys. Res., 117, D00R13, doi:10.1029/2011jd016654, 2012.

Jaeglé, L., Quinn, P. K., Bates, T. S., Alexander, B., and Lin, J.-T.: Global distribution of sea salt aerosols: new constraints from in situ and remote sensing observations, Atmos. Chem. Phys., 11, 3137-3157, doi:10.5194/acp-11-3137-2011, 2011.

Jourdain, B., Preunkert, S., Cerri, O., Castebrunet, H., Udisti, R., and Legrand, M.: Year-round record of size segregated aerosol composition in central Antarctica (Concordia station): Implications for the degree of fractionation of sea-salt

30 particles, J. Geo-phys. Res., 113, D14308, doi:10.1029/2007JD009584, 2008.





Kaleschke, L., Richter, A., Burrows, J., Afe, O., Heygster, G., Notholt, J., Rankin, A. M., Roscoe, H. K., Hollwedel, J., Wagner, T., and Jacobi, H.-W.: Frost flowers on sea ice as a source of sea salt and their influence on tropospheric halogen chemistry, Geophys. Res. Lett., 31, L16114, doi:10.1029/2004GL020655, 2004.

Klimont, Z., Kupiainen, K., Heyes, C., Purohit, P., Cofala, J., Rafaj, P., Borken-Kleefeld, J., and Schöpp, W.: Global 5 anthropogenic emissions of particulate matter including black carbon, Atmos. Chem. Phys., 17, 8681–8723, doi:10.5194/acp-17-8681-2017, 2017.

Krnavek, L., Simpson, W. R., Carlson, D., Dominé, F., Douglas, T. A., and Sturm, M.: The chemical composition of surface snow in the Arctic: examining marine, terrestrial, and atmospheric influence, Atmos. Environ., 50, 349–359, doi:10.1016/j.atmosenv.2011.11.033, 2012.

10 Kwok, R., and D. Rothrock: Decline in Arctic sea ice thickness from submarine and ICESat records: 1958–2008, Geophys. Res. Lett., 36, L15501, doi:10.1029/2009GL039035, 2009.

Lewis, E. R. and Schwartz, S. E.: Sea Salt Aerosol Production: Mechanisms, Methods, Measurements, and Models: A Critical Review, American Geophysical Union, Washington, D.C., 2004.

Li, C., Hsu, N. C., Sayer, A. M., Krotkov, N. A., Fu, J. S., Lamsal, L. N., Lee, J., and Tsay, S.-C.: Satellite observation of

15 pollutant emissions from gas flaring activities near the Arctic, Atmos. Environ., 133, 1–11, doi:10.1016/j.atmosenv.2016.03.019, 2016.

Li, M., Zhang, Q., Kurokawa, J.-I., Woo, J.-H., He, K., Lu, Z., Ohara, T., Song, Y., Streets, D. G., Carmichael, G. R., Cheng, Y., Hong, C., Huo, H., Jiang, X., Kang, S., Liu, F., Su, H., and Zheng, B.: MIX: a mosaic Asian anthropogenic emission inventory under the international collaboration framework of the MICS-Asia and HTAP, Atmos. Chem. Phys., 17, 935-963,

20 doi:10.5194/acp-17-935-2017, 2017.

Lin, J.-T. and McElroy, M. B.: Detection from space of a reduction in anthropogenic emissions of nitrogen oxides during the Chinese economic downturn, Atmos. Chem. Phys., 11, 8171-8188, doi:10.5194/acp-11-8171-2011, 2011.

Lin, S.-J. and Rood, R. B.: Multidimensinal flux-form semi-Lagrangian transport schemes, Mon. Weather Rev., 124, 2046– 2070, doi:10.1175/1520-0493(1996)124<2046:MFFSLT>2.0.CO;2, 1996.

25 Liu, H., Jacob, D. J., Bey, I., and Yantosca, R. M.: Constraints from 210Pb and 7Be on wet deposition and transport in a global three-dimensional chemical tracer model driven by assimilated meteorological fields, J. Geophys. Res., 106(D11), 12 109–12 128, doi:10.1029/2000JD900839, 2001.

Liu, Z. Y., Vaughan, M., Winker, D., Kittaka, C., Getzewich, B., Kuehn, R., Omar, A., Powell, K., Trepte, C., and Hostetler, C.: The CALIPSO Lidar Cloud and Aerosol Discrimination: Version 2 Algorithm and Initial Assessment of Performance, J.

30 Atmos. Ocean. Tech., 26, 1198–1213, doi:10.1175/2009JTECHA1229.1, 2009.





Mao, J., Jacob, D. J., Evans, M. J., Olson, J. R., Ren, X., Brune, W. H., Clair, J. M. St., Crounse, J. D., Spencer, K. M., Beaver, M. R., Wennberg, P. O., Cubison, M. J., Jimenez, J. L., Fried, A., Weibring, P., Walega, J. G., Hall, S. R., Weinheimer, A. J., Cohen, R. C., Chen, G., Crawford, J. H., McNaughton, C., Clarke, A. D., Jaeglé, L., Fisher, J. A., Yantosca, R. M., Le Sager, P., and Carouge, C.: Chemistry of hydrogen oxide radicals (HOx) in the Arctic troposphere in 5 spring, Atmos. Chem. Phys., 10, 5823-5838, doi:10.5194/acp-10-5823-2010, 2010.

Mao, J., Fan, S., Jacob, D. J., and Travis, K. R.: Radical loss in the atmosphere from Cu-Fe redox coupling in aerosols,

Atmos. Chem. Phys., 13, 509-519, doi:10.5194/acp-13-509-2013, 2013.

Martin, S.: A field study of brine drainage and oil entrapment in first-year sea ice., J. of Glaciol, 22 (88), 473–502, 1979.

Maslanik, J. A., Fowler. C., Stroeve, J., Drobot, S., Zwally, H. J., Yi, D., Emery, W.J.: A younger, thinner ice cover: 10 increased potential for rapid, extensive ice loss. Geophys. Res. Lett. 34: L24501. doi:10.1029/2007GL032043, 2007.

Mundy, C.J., Barber, D.G. and Michel, C.: Variability of snow and ice thermal, physical and optical properties pertinent to sea ice algae biomass during spring, J. Mar. Syst., 58, 107–120, doi:10.1016/j.jmarsys.2005.07.003, 2005.

Nandan, V., Geldsetzer, T., Yackel, J., Mahmud, M., Scharien, R., Howell, S., King, J., Ricker, R. and Else, B.: Effect of Snow Salinity on CryoSat-2 Arctic First-Year Sea Ice Freeboard Measurements, Geophys. Res. Lett., 44(20), 15 doi:10.1002/2017GL074506, 2017.

Nakawo, M. and Sinha, N.K.: Growth rate and salinity profile of first-year sea ice in the high Arctic. J. Glaciol, 27(96), pp.315-330, doi: DOI: 10.1017/ S0022143000015409, 1981.

Rankin, A. M., Auld, V., and Wolff, E. W.: Frost flowers as a source of fractionated sea salt aerosol in the polar regions, Geophys. Res. Lett., 27, 3469–3472, doi:10.1029/2000GL011771, 2000.

20 Renner, A.H., Gerland, S., Haas, C., Spreen, G., Beckers, J.F., Hansen, E., Nicolaus, M. and Goodwin, H.: Evidence of Arctic sea ice thinning from direct observations. Geo. Res. Lett., 41(14), pp.5029-5036, doi:10.1002/2014GL060369, 2014. Reynolds, R. W., Rayner, N. A., Smith, T. M., Stokes, D. C., and Wang, W.: An improved in situ and satellite SST analysis

for climate, J. Clim., 15, 1609–1625, doi:10.1175/1520-0442(2002)015<1609:AIISAS>2.0.CO;2, 2002.

Rienecker, M. M., Suarez, M. J., Gelaro, R., Todling, R., Bacmeister, J., Liu, E., Bosilovich, M. G., Schubert, S. D., Takacs,

25 L., Kim, G.-K., Bloom, S., Junye, C., Collins, D., Conaty, A., da Silva, A., Gu, W., Joiner, J., Koster, R. D., Lucchesi, R., Molod, A., Owens, T., Pawson, S., Pegion, P., Redder, C. R., Reichle, R., Robertson, F. R., Ruddick, A. G., Sienkiewicz, M., and Woollen, J.: MERRA: NASA's modern-era retrospective analysis for research and applications, J. Climate, 24, 3624– 3648,doi:10.1175/JCLI-D-11-00015.1, 2011.





Roscoe, H. K., Brooks, B., Jackson, A. V., Smith, M. H., Walker, S. J., Obbard, R. W., and Wolff, E. W.: Frost flowers in the laboratory: Growth, characteristics, aerosol, and the underlying sea ice, J. Geophys. Res., 116, D12301, doi:10.1029/2010JD015144, 2011.

Schroeder, W. H., Anlauf, K. G., Barrie, L. A., Lu, J. Y., Steffen, A., Schneeberger, D. R., and Berg, T.: Arctic springtime 5 depletion of mercury, Nature, 394, 331–332, doi:10.1038/28530, 1998.

Seguin, A.M., Norman, A.L. and Barrie, L.: Evidence of sea ice source in aerosol sulfate loading and size distribution in the Canadian High Arctic from isotopic analysis, J. Geophys. Res. Atmos., 119, 1087–1096, doi:10.1002/2013JD020461, 2014.

Seibert, P. and Frank, A.: Source-receptor matrix calculation with a Lagrangian particle dispersion model in backward mode, Atmos. Chem. Phys., 4, 51-63, doi:10.5194/acp-4-51-2004, 2004.

10 Shaw, P. M., Russell, L. M., Jefferson, A., and Quinn, P. K.: Arctic organic aerosol measurements show particles from mixed combustion in spring haze and from frost flowers in winter, Geophys. Res. Lett., 37, L10803, doi:10.1029/2010GL042831, 2010.

Simpson, W. R., Carlson, D., Hönninger, G., Douglas, T. A., Sturm, M., Perovich, D., and Platt, U.: First-year sea-ice contact predicts bromine monoxide (BrO) levels at Barrow, Alaska better than potential frost flower contact, Atmos. Chem. 15 Phys., 7, 621-627, doi:10.5194/acp-7-621-2007, 2007.

Slinn, S.A. and Slinn, W.G.N.: Predictions for particle deposition on natural waters. Atmospheric Environment (1967), 14(9), pp.1013-1016, doi:10.1016/0004-6981(80)90032-3, 1980.

Stammerjohn, S., Massom, R., Rind, D. and Martinson, D.: Regions of rapid sea ice change: An inter-hemispheric seasonal comparison. Geo. Res. Lett., 39(6), doi:10.1029/2012GL050874, 2012.

20 Steffen, A., Douglas, T., Amyot, M., Ariya, P., Aspmo, K., Berg, T., Bottenheim, J., Brooks, S., Cobbett, F., Dastoor, A., Dommergue, A., Ebinghaus, R., Ferrari, C., Gardfeldt, K., Goodsite, M. E., Lean, D., Poulain, A. J., Scherz, C., Skov, H., Sommar, J., and Temme, C.: A synthesis of atmospheric mercury depletion event chemistry in the atmosphere and snow, Atmos. Chem. Phys., 8, 1445-1482, doi:10.5194/acp-8-1445-2008, 2008.

Stohl, A., Hittenberger, M., and Wotawa, G.: Validation of the Lagrangian particle dispersion model FLEXPART against 25 large scale tracer experiment data, Atmos. Environ., 32, 4245–4264, doi:10.1016/S1352-2310(98)00184-8, 1998.

Stohl, A., Forster, C., Frank, A., Seibert, P., and Wotawa, G.: Technical note: The Lagrangian particle dispersion model FLEXPART version 6.2, Atmos. Chem. Phys., 5, 2461-2474, doi:10.5194/acp-5-2461-2005, 2005.

Travis, K. R., Jacob, D. J., Fisher, J. A., Kim, P. S., Marais, E. A., Zhu, L., Yu, K., Miller, C. C., Yantosca, R. M., Sulprizio, M. P., Thompson, A. M., Wennberg, P. O., Crounse, J. D., St. Clair, J. M., Cohen, R. C., Laughner, J. L., Dibb, J. E., Hall, S.





R., Ullmann, K., Wolfe, G. M., Pollack, I. B., Peischl, J., Neuman, J. A., and Zhou, X.: Why do models overestimate surface ozone in the Southeast United States?, Atmos. Chem. Phys., 16, 13561– 13577, doi:10.5194/acp-16-13561-2016, 2016.

Turner, J., Comiso, J.C., Marshall, G.J., Lachlan-Cope, T.A., Bracegirdle, T., Maksym, T., Meredith, M.P., Wang, Z. and Orr, A.: Non-annular atmospheric circulation change induced by stratospheric ozone depletion and its role in the recent 5 increase of Antarctic sea ice extent. Geo., Res. Lett., 36(8), doi:10.1029/2009GL037524, 2009.

Obbard, R. W., Roscoe, H.K., Wolff, E. W., and Atkinson, H. M.: Frost flower surface area and chemistry as a function of salinity and temperature, J. Geophys. Res., 114, D20305, doi:10.1029/2009JD012481, 2009.

Olivier, J. G. J. and Berdowski, J. J. M.: Global emissions sources and sinks, in: The Climate System, Berdowski, J., Guicherit, R. and B. J. Heij (eds.), A.A. Balkema Publishers/Swets & Zeitlinger Publishers, Lisse, The Netherlands, ISBN 90 10 5809 255 0, 33-78, 2001.

Palm, S. P., Yang, Y. K., Spinhirne, J. D., and Marshak, A.: Satellite remote sensing of blowing snow proper- ties over Antarctica, J. Geophys. Res.-Atmos., 116, 1–16, doi:10.1029/2011jd015828, 2011.

Palm, S. P., Kayetha, V., Yang, Y., and Pauly, R.: Blowing snow sublimation and transport over Antarctica from 11 years of CALIPSO observations, The Cryosphere, 11, 2555-2569, doi:10.5194/tc-11-2555-2017, 2017.

15 Parkinson, C. L. and Cavalieri, D. J.: Antarctic sea ice variability and trends, 1979–2010, The Cryosphere, 6, 871-880, doi:10.5194/tc-6-871-2012, 2012.

Perovich, D. K. and Richter-Menge, J. A.: Surface characteristics of lead ice, J. Geophys. Res., 99, 16341–16350, doi:10.1029/94JC01194, 1994.

Pye, H. O. T., Liao, H., Wu, S., Mickley, L. J., Jacob, D. J., Henze, D. K., and Seinfeld, J. H.: Effect of changes in cli- mate 20 and emissions on future sulfate-nitrate-ammonium aerosol levels in the United States, J. Geophys. Res., 114, D01205, doi:10.1029/2008JD010701, 2009.

Udisti, R., Dayan, U., Becagli, S., Busetto, M., Frosini, D., Legrand, M., Lucarelli, F., Preunkert, S., Severi, M., Traversi, R., and Vitale, V.: Sea spray aerosol in central Antarctica. Present atmospheric behaviour and implications for paleoclimatic reconstructions, Atmos. Environ., 52, 109–120, doi:10.1016/j.atmosenv.2011.10.018, 2012.

25 van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla, P. S., Morton, D. C., DeFries, R. S., Jin, Y., and van Leeuwen, T. T.: Global fire emissions and the contribution of deforestation, savanna, forest, agricultural, and peat fires (1997–2009), Atmos. Chem. Phys., 10, 11707-11735, doi:10.5194/acp-10-11707-2010, 2010.

van Donkelaar, A., Martin, R. V., Leaitch, W. R., Macdonald, A. M., Walker, T. W., Streets, D. G., Zhang, Q., Dunlea, E. J., Jimenez, J. L., Dibb, J. E., Huey, L. G., Weber, R., and Andreae, M. O.: Analysis of aircraft and satellite measurements from

30 the Intercontinental Chemical Transport Experiment (INTEX-B) to quantify long-range transport of East Asian sulfur to





Canada, Atmos. Chem. Phys., 8, 2999-3014, doi:10.5194/acp-8-2999-2008, 2008.

Wagenbach, D., Ducroz, F., Mulvaney, R., Keck, L., Minikin, A., Legrand, M., Hall, J. S., and Wolff, E. W.: Sea-salt aerosol in coastal Antarctic regions, J. Geophys. Res., 103, 10 961–10 974, doi:10.1029/97JD01804, 1998.

Wang, Y., D.J. Jacob, and J.A. Logan, Global simulation of tropospheric O3-NOx-hydrocarbon chemistry, 1. Model 5 formulation, J. Geophys. Res., 103, D9,10,713-10,726, doi:10.1029/98JD00158, 1998.

Wang, Q., Jacob, D. J., Fisher, J. A., Mao, J., Leibensperger, E. M., Carouge, C. C., Le Sager, P., Kondo, Y., Jimenez, J. L., Cubison, M. J., and Doherty, S. J.: Sources of carbonaceous aerosols and deposited black carbon in the Arctic in winterspring: implications for radiative forcing, Atmos. Chem. Phys., 11, 12453-12473, doi:10.5194/acp-11-12453-2011, 2011.

Weeks, W. F., and Lee O. S.: Observations on the physical properties of sea-ice at Hopedale, Labrador, Arctic, 11(3), 135– 10 155, doi: 10.14430/arctic3740, 1958.

Weeks, W. F. and Ackley, S. F.: The growth, structure and properties of sea ice, 9–164, Plenum Press, New York, 1986.

Weller, R., Woltjen, J., Piel, C., Resenberg, R., Wagenbach, D., Konig-Langlo, G., and Kriews, M.: Seasonal variability of crustal and marine trace elements in the aerosol at Neumayer station, Antarctica, Tellus, 60, 742–752, doi:10.1111/j.1600- 0889.2008.00372.x, 2008.

15 Wesely, M. L., Parameterization of surface resistance to gaseous dry deposition in regional-scale numerical models, Atmos. Environ., 23, 1293-1304, doi:10.1016/0004-6981(89)90153-4, 1989.

Winker, D. M., Vaughan, M. A., Omar, A., Hu, Y., and Powell, J. A.: Overview of the CALIPSO Mission and CALIOP Data Processing Algorithms, J. Atmos. Ocean. Tech., 26, 2310–2323, doi:10.1175/2009JTECHA1281.1, 2009.

Winker, D. M., Tackett, J. L., Getzewich, B. J., Liu, Z., Vaughan, M. A., and Rogers, R. R.: The global 3-D distribution of

20 tropospheric aerosols as characterized by CALIOP, Atmos. Chem. Phys., 13, 3345-3361, doi:10.5194/acp-13-3345-2013, 2013.

Winker, D.: CALIPSO LID L2 Standard HDF File -Version 4.10. NASA Langley Research Center Atmospheric Science Data Center DAAC. https://doi.org/10.5067/CALIOP/CALIPSO/LID\_L2\_05kmAPro-Standard-V4-10, Last Access February 2018, 2016

25 Wolff, E. W., Fischer, H., Fundel, F., Ruth, U., Twarloh, B., Littot, G. C., Mulvaney, R., Rothlisberger, R., de Angelis, M., Boutron, C. F., Hansson, M., Jonsell, U., Hutterli, M. A., Bigler, M., Lambeck, K., Kaufmann, P., Stauffer, B., Stocker, T. F., Steffensen, J. P., Siggaard-Andersen, M. L., Udisti, R., Becagli, S., Castellano, E., Severi, M., Wagenbach, D., Barbante, C., Gabrielli, P., and Gaspari, V.: Southern Ocean sea-ice extent, productivity and iron flux over the past eight glacial cycles, Nature, 440, 491–496, doi:10.1038/nature04614, 2006.





Xu, L., Russell, L. M., Somerville, R. C. J., and Quinn, P. K.: Frost flower aerosol effects on Arctic wintertime longwave cloud radiative forcing, J. Geophys. Res.-Atmos., 118, 13282–13291, doi:10.1002/2013JD020554, 2013.

Xu, J.-W., Martin, R. V., Morrow, A., Sharma, S., Huang, L., Leaitch, W. R., Burkart, J., Schulz, H., Zanatta, M., Willis, M. D., Henze, D. K., Lee, C. J., Herber, A. B., and Abbatt, J. P. D.: Source attribution of Arctic black carbon constrained by 5 aircraft and surface measurements, Atmos. Chem. Phys., 17, 11971-11989, doi:10.5194/acp-17-11971-2017, 2017.

Yang, X., Pyle, J. A., and Cox, R. A.: Sea salt aerosol production and bromine release: Role of snow on sea ice, Geophys. Res. Lett., 35, L16815, doi:10.1029/2008GL034536, 2008.

Yang, X., Pyle, J. A., Cox, R. A., Theys, N., and Van Roozendael, M.: Snow-sourced bromine and its implications for polar tropospheric ozone, Atmos. Chem. Phys., 10, 7763-7773, doi:10.5194/acp-10-7763-2010, 2010.

10 Yang, X., Neděla, V., Runštuk, J., Ondrušková, G., Krausko, J., Vetráková, Ľ., and Heger, D.: Evaporating brine from frost flowers with electron microscopy and implications for atmospheric chemistry and sea-salt aerosol formation, Atmos. Chem. Phys., 17, 6291-6303, doi:10.5194/acp-17-6291-2017, 2017.

Zender, C. S.: Mineral Dust Entrainment and Deposition (DEAD) model: Description and 1990s dust climatology, J. Geophys. Res., 108, doi:10.1029/2002JD002775, 2003.

15 Zhang, L., Gong, S., Padro, J., and Barrie, L.: A size-segregated particle dry deposition scheme for an atmospheric aerosol module, Atmos. Environ., 35, 549–560, doi: /10.1016/S1352-2310(00)00326-5, 2001.

 $\odot$  Author(s) 2018. CC BY 4.0 License. Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2018-298 Manuscript under review for journal Atmos. Chem. Phys. Discussion started: 7 May 2018







**Figure 1: Spatial distribution of mean aerosol extinction coefficients (0-2 km) during the 2007-2009 Arctic cold season (November-**April) observed by (a) CALIOP and calculated with the GEOS-Chem model in (b) a Standard simulation (STD), (c) a simulation **and a** simulation of the standard simulation of the standard simulation of the standard simulatio 5 **including blowing snow SSA emissions (STD+Snow), (d) an optimized blowing snow simulation (STD+Opt. Snow), and (e) a simulation including frost flower SSA emissions (STD+FF). The simulated extinctions are sampled at the time and location of the CALIOP overpasses, and the CALIOP sensitivity threshold is applied. The bottom panels show the extinction coefficients of individual aerosol components in the GEOS-Chem simulations: (f) sulfate aerosol, (g) open ocean SSA, (h) blowing snow SSA, (i) Opt. blowing snow SSA and (j) frost flower SSA. Note the different colorbar scales for the top and bottom rows.**

10







**Figure 2. Scatter plot of CALIOP and GEOS-Chem aerosol extinction coefficients over FYI (orange circles) and MYI (purple circles) for the Arctic cold season (November-April, top panels) and Antarctic cold season (May-October, bottom panels). Each symbol represents the monthly aerosol extinction coefficients for individual grid boxes (2**°**x5**°**, 0-2 km) over sea ice. The dashed gray line is the 1:1 line. The purple and orange lines are the linear fit for the points over MYI and FYI, respectively. The slope of**  the regression line, correlation coefficient (r) and normalized mean bias,  $NMB = 100 \times (\text{Model}/\text{Obs} - 1)$ , are given in the insert of **each figure.** 

 $\circled{c}$  Author(s) 2018. CC BY 4.0 License. Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2018-298 Manuscript under review for journal Atmos. Chem. Phys. Discussion started: 7 May 2018







**Figure 3: Top row: 2007-2009 CALIOP mean aerosol extinction coefficients (0-2 km) in the Arctic cold season (November-April)**  1.0 **over (a) first-year sea ice (FYI), (b) multi-year sea ice (MYI) and (c) the Canadian Arctic Archipelago (CAA). Middle row: Vertical profiles of Arctic cold season mean aerosol extinction coefficients over (d) FYI, (e) MYI and (f) CAA for CALIOP (black**  0.5 5 **dots with horizontal lines indicating standard deviations) and GEOS-Chem model simulations (STD: black lines, STD+Snow: red solid lines, STD+Opt. Snow: red dashed lines; STD+FF: green lines). Bottom row: Seasonal cycle of 0-2 km monthly aerosol**  0.0 extinction coefficients averaged over (g) FYI, (h) MYI and (f) CAA. CALIOP observations are shown as black circles and vertical lines indicate the interannual standard deviation. The four GEOS-Chem model simulations are also shown (STD: black lines, **STD+Snow: red solid lines, STD+Opt. Snow: red dashed lines, STD+FF: green lines).**  $\begin{bmatrix} 3 \\ 1 \end{bmatrix}$ 

Manuscript under review for journal Atmos. Chem. Phys.  $\odot$  Author(s) 2018. CC BY 4.0 License. Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2018-298 Discussion started: 7 May 2018







Figure 4. Spatial distributions of mean aerosol extinction coefficients (0-2 km) during the 2007-2009 Antarctic cold season (May-**October) observed by (a) CALIOP and calculated with the GEOS-Chem (b. STD, c. STD+Snow, d. STD+Snow-2, e. STD+FF). The**  bottom panels show extinctions of individual aerosol components in the GEOS-Chem simulations: (f) open ocean SSA, (g) blowing 5 **snow SSA, (h) Opt. blowing snow SSA and (i) frost flower SSA. Note the different colorbar scales for the top and bottom rows.**

 $\circled{c}$  Author(s) 2018. CC BY 4.0 License. Atmos. Chem. Phys. Discuss., https://doi.org/10.5194/acp-2018-298 Manuscript under review for journal Atmos. Chem. Phys. Discussion started: 7 May 2018







**Figure 5. Same as Figure 3, but for Antarctic aerosol extinction coefficients during Austral winter (May-October) over (a) FYI**  1.0 **(excluding offshore Ross Ice-shelf), (b) MYI and (c) offshore Ross Ice-shelf. As shown in (g), the monthly average aerosol extinction coefficients are not available over FYI during Antarctic summer (January-March) due to the limited FYI extent.** 0.5 e<br>Id







**Figure 6. Monthly mean SSA mass concentrations at Arctic sites (a. Barrow, b. Alert, c. Zeppelin) and Antarctic sites (d. Dumont d'Urville, e. Neumayer). All observations and model results are for 2001-2008 except at Neumayer (2001-2007). The observed mean concentrations are indicated with filled black circles, and the lines are for the GEOS-Chem simulations (STD: black line,**  5 **STD+Snow: red line, STD+Opt, Snow-2: red dashed line; STD-FF: green line). The black vertical lines are the standard deviations of monthly means for the observation years. For each individual panel, the insert lists mean concentrations and standard deviations, as well as the NMB for the cold season only (Arctic: November-April; Antarctic: May-October).**







**Figure 7. November 6, 2008 case study of a blowing snow SSA feature over the Arctic. Top panels: Spatial distribution of mean aerosol extinctions below 2 km altitude for the (a) STD, (b) STD+Snow, and (c) STD+Opt. Snow GEOS-Chem simulations. The CALIOP nighttime overpass at 00:58–01:12 UTC is displayed on the top panels, with filled circles color-coded according to** 

- 5 **observed mean extinction coefficients below 2 km. The overpass intercepts the SSA blowing snow feature between points A and B. Panel (d) shows the MERRA sea ice coverage on November 6, 2008, with light gray shading indicating FYI and black shading for MYI. Panels (e-h) show the observed and simulated vertical cross-sections of aerosol extinction coefficients along the CALIOP overpass. The light gray shading in (e) indicates that no valid data is available for CALIOP. The dark gray shading in (f-h) shows the local topography in the model. Panel (i) shows the 0-2 km CALIOP mean aerosol extinctions and the contributions of different**
- 10 **aerosol types in the GEOS-Chem simulations.**







**Figure 8. (a) The November 4, 2008 FLEXPART footprint below 100m (in seconds) for particles initialized at 0-2 km over the black squares on November 6, 2008, near the blowing snow feature observed by CALIOP (Figure 7). (b) November 4, 2008 blowing snow SSA emissions from the STD+Snow simulation, and the CALIOP overpass at 01:11–01:24 UTC on that day. The**  5 **bottom two panels display CALIOP cross-sections between points C and D for (c) the 532nm perpendicular attenuated backscatter (km-1 sr-1 ) and (d) attenuated depolarization ratio. The surface wind speeds (m s-1 ) are shown with a red line in panel c.**