## Chemical composition of isoprene SOA under acidic and non-acidic conditions: Effect of relative humidity

## **Supplementary Information**

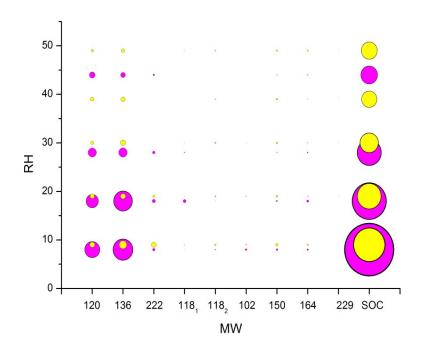
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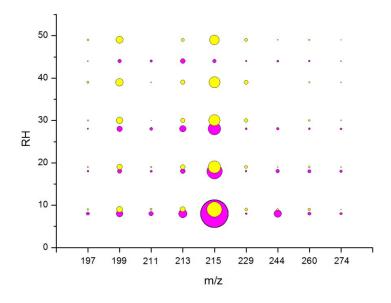
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**Figure S1.** Relative amounts of aerosol components detected with GC-MS acidic seed (pink) and non-acidic seed (yellow) experiments (the areas of the circles are proportional to the estimated mass concentrations of compounds).

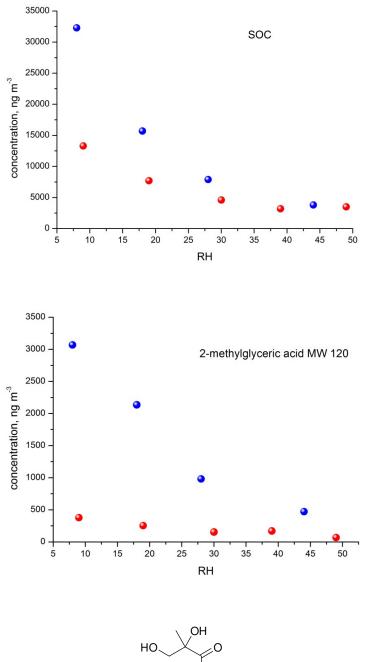


**Figure S2.** Relative intensities of aerosol components detected with LC-MS in acidic seed (pink) and non-acidic seed (yellow) experiments (the areas of the circles are proportional to relative intensities of compounds detected).

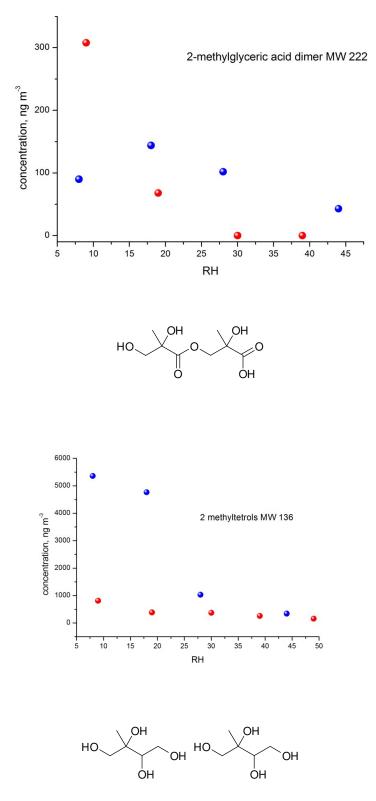
Product	MW	m/z	RH = 8	RH = 18	RH = 28	RH = 39 –
			- 9	- 20	- 30	49
2-methylglyceric acid	120		>			
2-methyltetrol OS		244	>			
2-methylthreonic acid OS		274	>			
furanone OS		211	>			
2-methyltetrols	136			> =		
2-methyltartaric acid	164			> =		
2-methyltetrol OS		260		> =		
furanetriol OS		213	>		=	
2-methyltetrol OS		215	>		=	
IEPOX-1	118		=	>	=	
dimer of 2-methylglyceric	222		<		>	
acid						
C5-diol	102		>	<		=
IEPOX OS		197	>	=	= <	
2-methylthreonic acid	150		<	=	<	
2-methylglyceric acid OS		199	=		<	
IEPOX-2	118		<			
2-methylthreonic acid OS +		229	<			

**Table S1.** Comparison of product yields in acidic seed experiments vs. non-acidic seed experiments at various RH levels ( > higher, = equal and lower < )

Figure S3. Concentrations of some compounds in acidic seed experiments (blue) and non-acidic seed experiments (red) – influence of Relative Humidity



OH



2-methylthreitol (left) and 2-methylerythritol (right)

